

Microscopic Determination of the Nonopaque Minerals

U.S. GEOLOGICAL SURVEY BULLETIN 1627



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**By MICHAEL FLEISCHER, RAY E. WILCOX, and
JOHN J. MATZKO**

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**A revision of Bulletin 848 by Larsen and Berman (1934),
including new optical data and new methods of determination**

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PREFACE TO THE THIRD EDITION

The first edition of "The Microscopic Determination of the Nonopaque Minerals" by Esper S. Larsen, Jr., was published in 1921 as U.S. Geological Survey Bulletin 679 after more than 10 years of preparation. Preparation involved not only compilation of data from the literature, but also new determinations of optical properties on about 500 samples. These new descriptions accounted for 126 of the 294 pages of the 1921 edition.

The second edition, by Larsen and Harry Berman, both then at Harvard University, published in 1934 as U.S. Geological Survey Bulletin 848, increased the data in Bulletin 679 by more than 50 percent. The second edition was republished in Russia in 1965 (Petrov, 1965) with insertions of newer data from the literature.

Assembly of new data on optical properties of minerals was begun by Fleischer and Matzko in 1950, and compilation in the present tabular format was begun in 1965. Insofar as possible, data from the literature up to August 1, 1983 have been included in the tables. The introductory text on methods was written by Wilcox, who also critically reviewed and suggested minor modifications of the tables as prepared by Fleischer.

The present edition, in holding to the objective of mineral identifications, omits discussion of statistics of mineral optics, as covered in Chapter 3 of the 1934 edition. Modern treatments of the Gladstone-Dale relation and other aspects of the relations of refractive index to density and chemical composition can be found in Allen (1956), Jaffe (1956), Batsanov (1961), Anderson (1975), Pabst (1975), Mandarino (1976, 1978, 1979, 1981) and Bloss and others (1983). The Commission on New Minerals and Mineral Names, International Mineralogical Association (American Mineralogist, v. 67, p. 191-192), states for new minerals, "It is recommended that the relationship between chemical composition, density, and refractive indices, be checked by the Gladstone-Dale rule."

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Microscopic Determination of the Nonopaque Minerals

By Michael Fleischer, Ray E. Wilcox, and John J. Matzko

CHAPTER I. INTRODUCTION

Identification of an unknown mineral in former times was a haphazard operation, involving determination of a few optical, physical, and chemical properties for comparison with a woefully incomplete set of properties of known minerals. Today, not only can the diagnostic optical properties be determined more reliably and the chemical composition more rapidly, but several X-ray techniques are available for determination of diagnostic crystal structure.

Compilation of X-ray data in a form useful for mineral identification has proceeded apace and has far outstripped the compilation of new optical data, with the unfortunate result that the use of the polarizing microscope in mineralogic and petrologic investigations has decreased markedly; in addition, the time allotted in the curriculum of many colleges for training in optical crystallography has been reduced. The polarizing microscope nevertheless remains a powerful and sometimes indispensable tool in mineral identification, and to help restore its usefulness the authors have taken on the task of bringing up to date the tables compiled by Larsen and Berman (1934).

The tables, which appear in chapters 3 and 4 herein, are a compilation of properties of known minerals. They have been updated by adding optical data that have appeared in the literature since the 1934 edition and up until August 1, 1983. The range of optical properties of previously listed minerals and isomorphous groups has been corrected or extended, and the names and data for newly discovered minerals have been inserted. Methods for the measurement of optical properties are outlined in chapter 2.

Identification of an unknown by its optical properties proceeds most effectively in three stages: (1) measurement under the microscope of selected key

properties; (2) elimination from further consideration minerals not having these properties; and (3) determination of selected additional diagnostic properties to eliminate all but one or a very few candidates. For confirmation of that one mineral or a final choice between remaining candidates, one may then turn to X-ray or chemical analysis.

It is important that this process of elimination using the tables be given the same care and attention as the measurements of the basic optical properties, for to overlook a possible candidate at this stage can jeopardize the correctness of the eventual identification. For the final stage, one should consult published encyclopedic descriptive mineralogies (for example, Deer and others, 1962, 1963, 1978; Tröger, 1967, 1979; Winchell and Winchell, 1951) then return to the microscope to determine the needed additional diagnostic characteristics.

Scope of the Tables

For each known mineral, tables 3 to 7 list optical properties readily determinable by the polarizing microscope and its accessories, and they also note certain nonmicroscopic but useful identifying properties such as bulk color, fusibility, hardness, and chemical behavior. The minerals are grouped in separate tables according to basic optical character (isotropic, uniaxial, or biaxial) and according to optic sign (positive or negative). For convenience in entering the tables by means of optical data routinely obtained by the so-called immersion method (see chap. 2), the minerals in each table are arranged in order of increasing refractive index. The tables may also be entered with refractive indices estimated from thin sections, although a much greater number of listed minerals will usually have to be considered as possible candidates due to the uncertainty of the estimate.

Need for Additional Data

Although the tables in chapters 3 and 4 have been updated, they contain many imperfections and cannot be regarded either as complete or entirely correct. The reliability of the individual listings varies widely and for some minerals cannot be estimated, inasmuch as the source literature does not supply adequate descriptions of methods used. Indeed, in too many cases it is painfully evident that no set of minimum standards was adhered to in carrying out the original optical determination, much less in the preparation and purification of the minerals for chemical analysis. Still remaining to be filled are gaps in the knowledge of the ranges of optical properties of rare minerals, many of which, judging from their determined chemical compositions, are members of isomorphous series. Finally, there is the certainty that some naturally occurring minerals still wait to be discovered and described optically.

Optical methods of mineral identification are useful not only as an adjunct to the powerful methods of X-ray and electron microprobe but in many situations also as the most reliable and rapid means of identification. However, the gaps and flaws mentioned above in available optical data of minerals are a matter of concern particularly because identification rests on the process of elimination. One therefore must keep in mind the possibility, remote as it might be, that the unknown mineral under examination is not listed in the part of the table involved. The use of optical methods in the modern laboratory will not reach full effectiveness until these shortcomings are corrected. Fortunately, basic optical data can now be obtained with greater reliability and with considerably less tedium than formerly, thanks to the availability of more documented mineral specimens and more advanced development of instrumentation and methods. Also, many of the difficulties due to limited sample size that plagued the early workers in descriptive mineralogy are now avoidable because optical, X-ray, and chemical analyses can

be carried out on extremely small samples, using the spindle stage, the single-crystal X-ray goniometer, and the electron microprobe.

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CHAPTER II. DETERMINATION OF OPTICAL PROPERTIES OF TRANSPARENT MINERALS BY THE IMMERSION METHOD

Whereas the textures and interrelations of minerals in rocks, so important in petrologic interpretations, are well seen in thin section, measurements of principal refractive indices and other optical properties necessary for reliable identification are better accomplished on individual fragments by the so-called immersion method. Advantages of this method include the ability to manipulate crystal fragments individually without the handicap of interference from adjacent crystals.

For the immersion method, the sample is crushed to separate the indi-

vidual crystal units and expose fresh surfaces, so that fragments can be chosen and their refractive indices can be compared under the microscope with successive liquids of known refractive indices until a match is found. In the same mounts, birefringence, optic angle, optic sign, pleochroism, and the relationships of optical directions to crystallographic directions (extinction angle and sign of elongation, and so on) may be measured or estimated.

It is possible here to give only a brief review of established procedures and techniques; emphasis will be on several practical techniques not yet fully covered in most textbooks. For details of the conventional techniques, the reader is referred to textbooks such as Bloss (1961), Wahlstrom (1969), Hartshorne and Stuart (1969, 1970), Burri (1950), Tatarskii (1965), Stoiber and Morse (1972), Hutchison (1974, especially p. 78-93, 247-263), Schumann and Kornder (1973), McKie and McKie (1974), and el-Hinnawi (1966). Several useful works deal with problems encountered in special fields: industrial chemistry and product control (Hartshorne and Stuart, 1970; Chamot and Mason, 1958; McCrone and Delly, 1973; Lehr and others, 1967), ceramics (Insley and Frechette, 1955; Dietzel and K  ppen, 1974), identification and characterization of synthetic inorganic substances (Winchell and Winchell, 1964; K  rdes, 1960; Lehr and others, 1967), and organic compounds (Winchell, 1954).

Refractive Index

Three techniques are especially useful for determination of refractive index, the primary basis for entry into the tables: (1) central illumination ("Becke line"), (2) inclined illumination ("Schroeder van der Kolk"), and (3) dispersion coloration ("dispersion staining"). All three techniques depend on the same basic optic phenomenon: the smaller the difference of index between fragment and liquid, the less strongly diverted the light rays passing through an inclined interface. The techniques

differ only in the manner of illumination and in the resulting observed effects. Because the immersion liquids are so important for the correctness of the determination, their calibration and use will be dealt with first.

Immersion Liquids

Sets of liquids ranging in index from about 1.46 to 1.80 and separated by regular index intervals, say 0.005 or 0.002, are available commercially. In choosing the interval between liquids, it may be kept in mind that the greater initial cost of a set having the smaller interval may be more than compensated by the saving of time and the greater precision of results. The label on each bottle in the set should show not only the index of refraction for the D-line of the spectrum, but also its coefficients of temperature and dispersion. With some care and effort, sets of liquids can be made up in the laboratory from end members, as described by Harrington and Buerger (1931), Butler (1933), and Meyrowitz (1955, 1956). Useful general discussions of possible end-member liquids are given by Johannsen (1918, p. 259), Burri (1950, p. 221), Wilcox (1964), and Hartshorne and Stuart (1970, p. 265).

The refractive index of a typical organic immersion liquid decreases significantly with increasing temperature--on the order of 0.0004 to 0.0005 per degree Celsius; the corresponding change of index of the immersed crystal fragment is an order of magnitude smaller, thus negligible in the present application. A bottle label reading $n_{\text{D}}^{25} = 1.534$ indicates that the refractive index of its contained liquid is 1.534 at 25°C in light of the D-line (wavelength 589.3 nm). A further notation,

$$-\frac{dn}{dt} = 0.00039/\text{°C},$$

indicates the magnitude of the inverse change of index with temperature. Thus, after reaching the match between fragment and liquid, it is necessary to make

an index correction for the difference between the temperature of the immersion mount itself and the temperature at which it was calibrated. If, for instance, the above liquid was found to match the index of a fragment at 28°C, the correction to be made would be 3x 0.00039, or 0.00117, to be subtracted from 1.534 to give 1.53283, rounded for the present application to 1.533. This is the refractive index of the matching liquid under these conditions and thus of the index of the fragment.

The refractive index of each liquid in the set must be rechecked from time to time, depending on the stability of the liquid and on the accuracy required in use. Adequate for most mineralogical laboratories is the Abbe-type refractometer for indices up to about 1.70 with a precision about ± 0.0002 . Although a prism of special glass may be substituted to permit measurements to higher than 1.80, the glass commonly used for this purpose is a high-lead glass that is vulnerable to corrosion from most liquids used in this higher range. Less expensive refractometers, such as that designed by E.E. Jelley (1934; see also Edwards and Otto, 1938), extend to 1.90, are not damaged by liquids above 1.70, and are precise enough (about ± 0.001) for much work. Whatever the refractometer, it too should be calibrated carefully according to the directions of the manufacturer (see also Fisher, 1958).

In some applications it is necessary to know the dispersion strength of the immersion liquids. This property may be expressed as the difference between refractive indices at two standard wavelengths, such as the F- and C-lines. A common expression of the dispersive power of the liquid is the coefficient:

$$v = \frac{n_D - 1}{n_F - n_C}$$

Dispersion can be determined by measurement of refractive indices of the liquid at the wavelengths of the F-, D-, and C-lines by minimum deviation on a goniometer using a prism-cell, or on an Abbe

refractometer making corrections for the dispersion of the glass of its reference prism. Considerably more convenient, however, is use of an Abbe refractometer equipped with Amici prisms, whereby a measure of the dispersion is obtained in the course of determination of n_D in white light.

Central Illumination ("Becke Line") Technique

The so-called Becke line test of the refractive index difference between fragment and immersion liquid is carried out in illumination consisting of a narrow cone of light symmetrically about the axis of the microscope. The fragment in liquid is brought into sharp focus. (If anisotropic, the fragment should be placed in a position of extinction, then the upper polar is withdrawn.) As the focus is raised slightly, that is, as distance between objective lens and object is increased, a bright-line (the Becke line) forms at the edge of the fragment and moves toward the substance of higher index. Thus, if refractive index of the fragment is higher, the Becke line moves inward; if lower, it moves outward. As the focus is raised further, the line fades and the image blurs. The basis for this fading and blurring has been variously ascribed to the crude lens effect of the solid fragment and to total reflection of rays by the very steep interfaces at the edge of the fragment.

Determination of the refractive index of the fragment becomes a matter of finding the matching liquid in a set of graduated index liquids. As this matching liquid is approached in successive liquids, the Becke line noticeably weakens but, up to a point, may be made more visible by stopping down the substage aperture iris diaphragm. In a liquid near the match and with white light for illumination, the edges of the fragment take on faint colors. Attention to the quality of these faint dispersion colors enables one to recognize a close approximation of the true match (Emmons and Gates, 1948). A more exact

match may be established using monochromatic light, such as that of the standard sodium line (589.3-nm wavelength); when the index of the solid and liquid are the same in monochromatic light, the Becke line disappears completely, and the fragment becomes invisible except for flaws and inclusions.

If the host fragment is crowded with inclusions, precise results are not attainable by the Becke line method, for as the matching liquid is approached in successive mounts, the still strong Becke lines of the crowded inclusions coalesce to form a single bright line that prevents recognition of the fading and disappearance of the Becke line of the host. A further difficulty, possibly related, arises from the so-called "spurious Becke line," a light line that moves opposite to the true Becke line. This situation is often encountered with fragments having appreciable light absorption (body color) or unfavorable fragment contour. Here at best one finds only a broad minimum in the intensity of the light lines through a series of adjacent immersion liquids, leaving in question the refractive index of the true match.

check the behavior of a particular microscope and illuminating system in this respect by use of a fragment of known index in a liquid of known direction of mismatch.

In white light, as the indices of liquid and fragment become more nearly equal in successive mounts, the apparent relief of the fragment decreases. At match, one edge of the fragment is bluish and the opposite edge reddish. One learns to recognize the diagnostic shades and intensities of colors that are obtained on either edge when a match has been reached, keeping in mind that the quality of the colors is affected to some extent by the natural body color of the fragment as well as the spectrum of the light source. The match point is more precisely defined with monochromatic illumination, for at match the fragment completely disappears except for flaws and inclusions. In order to cancel disturbing effects due to differences in shape of opposite edges of the fragment, it is helpful to observe the fragment at the one position of the microscope stage, then rotate the fragment 180° on the stage and observe it again to confirm that the relief effect is in the same relation to the lighter side of the field.

Oblique Illumination (Schroeder van der Kolk) Technique

The oblique illumination technique, details of which are given in most textbooks and in the excellent discussion by Wright (1911, p. 92-95), is capable of greater sensitivity than the Becke line method. Briefly, to obtain oblique illumination, the entering cone of light is blocked on one side at the proper level in the substage or in the microscope tube to produce a sidelighting effect that enhances the appearance of relief (mismatch) of the fragment in respect to the immersion liquid: one edge of the fragment is highlighted while the opposite edge is shaded. As a general rule, the index of the fragment is higher than the liquid if its brighter side is toward the more brightly illuminated portion of the field of view. It is advisable, however, to

Dispersion Coloration Technique Using Central Focal Masking

This method is rapid and in large part free of misleading effects such as the "spurious Becke line" encountered in the central illumination method. Although the precision of this method is somewhat less than that of oblique monochromatic illumination, it is quite adequate for purposes of crystal identification. In addition to its value for determination of refractive indices, the method enables rapid distinction between different mineral species in a fragmental mixture and determination of the kind and amount of undesired constituents in a product at successive stages

in a purification process. Details of the method and its variants may be found in descriptions by Cherkasov (1960), Wilcox (1962, 1983), Brown and McCrone (1963), Brown and others (1963), and Hartshorne and Stuart (1970, p. 437). Similar color effects are obtained in the "Grenzdunkelfeld" technique described by Schmidt and Heidermanns (1958) and in the early "dispersion staining" technique of Crossmon (1948). In the two latter techniques, however, some light rays traverse the fragment at an appreciable angle to the axis of the microscope, leading to possible inaccuracies when working with strongly anisotropic substances.

The immersion mount is illuminated by a narrow beam of ordinary white light, produced by severely restricting the substage aperture diaphragm. With a small opaque dot mask installed in the objective near the focal point, the image of the fragment appears on the dark field in a color that indicates the degree of match or mismatch with the immersion liquid. These useful color effects result from the appreciably greater dispersion of the liquid over that of the solid of similar refractive index. Thus in figure 1, the refractive index of the glass matches that of the liquid for light of wavelength near that of the D-line (5893 Å or 589.3 nm) but is progressively less than that of the liquid at shorter wavelengths and greater at longer wavelengths.

The critical color effects begin to appear as the index of the immersion liquid approaches to within about 0.01 of that of the fragment, depending on the relative dispersive strengths of fragment and liquid. In liquids of index outside this range in either direction, the fragment edges are imaged brightly in the color of the source light. To determine whether to go to a liquid of higher or lower index, one simply observes the movement of the single strong Becke line as the focus is raised.

Under the microscope a fragment immersed in a matching liquid and illu-

minated by a thin beam of white light refracts the various wavelengths differently (fig. 2). Light beams of a wave-

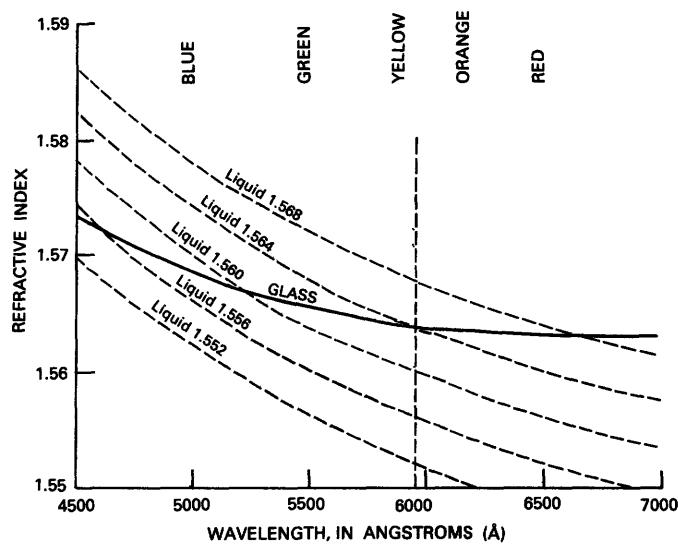


Figure 1. Dispersion of the refractive index of a typical inorganic solid compared to dispersions of successive members of a typical set of organic immersion liquids.

length close to the standard yellow wavelength, 589.3 nm, are blocked by the opaque dot at the focal point of the objective, while beams of shorter wavelengths (more blue) and of longer wavelengths (more red) bypass the opaque dot and combine again to form an image of the fragment edges in deep violet color. This violet image is the combined effect of the spectral colors remaining after the band near the yellows has been subtracted. The field of view is otherwise dark, because beams not passing through inclined surfaces of the fragment edge are also blocked by the opaque dot. By raising the focus slightly, a red outline of the image moves inward and a blue outline moves outward. (These, of course, are the Becke lines for light of the respective wavelengths.)

From figure 1 it can be further deduced that, in a liquid of slightly lower index, say by 0.002, the edges of the fragment would be seen as noticeably richer in reds; in this case a band in the blue-green would have been subtracted. Similarly the edges of the fragment in a liquid higher by 0.002 would be seen as richer in blue.

Table 1 gives word descriptors of the colors seen in different situations. Each person should learn the color of the match as seen by eye, however. To this end, one may make a mount of a solid in a liquid as closely matched as can be judged by inclined illumination (described above) using monochromatic sodium-D illumination, then change to central focal masking illumination to observe and fix in mind the quality of color seen at the match.

Techniques of Higher Precision

Very precise determination of refractive indices beyond the third decimal place is seldom required for identification of unknown minerals. Should greater precision be desired, for instance in the characterization of newly discovered minerals or in applications to crystal structures, a number of techniques are available, each of which requires special auxiliary equipment. Perhaps the most rapid of these is the wavelength-variation method introduced by Wherry (1918), Posnjak and Merwin (1922), and extended by Emmons (1929, 1943). A precision of $\pm .0001$ refractive index is estimated for this method, as refined by Morse (1968) using a convenient substage variable interference filter and immersion liquids for which dispersion strengths have previously been determined. A further extension of the method enables determination of members of an isomorphous series, such as the plagioclase series, for which dispersion characteristics are also known.

Sueno (1933) and Micheelson (1957) use wavelength-variation techniques, adding glass powders of precisely known index to the immersion mount of the unknown substance to act in the role of refractometer, thus eliminating the necessity of temperature correction. In the double-variation method of Emmons (1929, 1943), both wavelength and temperature are changed at will to match solid and liquid at several wavelengths, enabling one to determine not only the refractive index of the solid at the D-

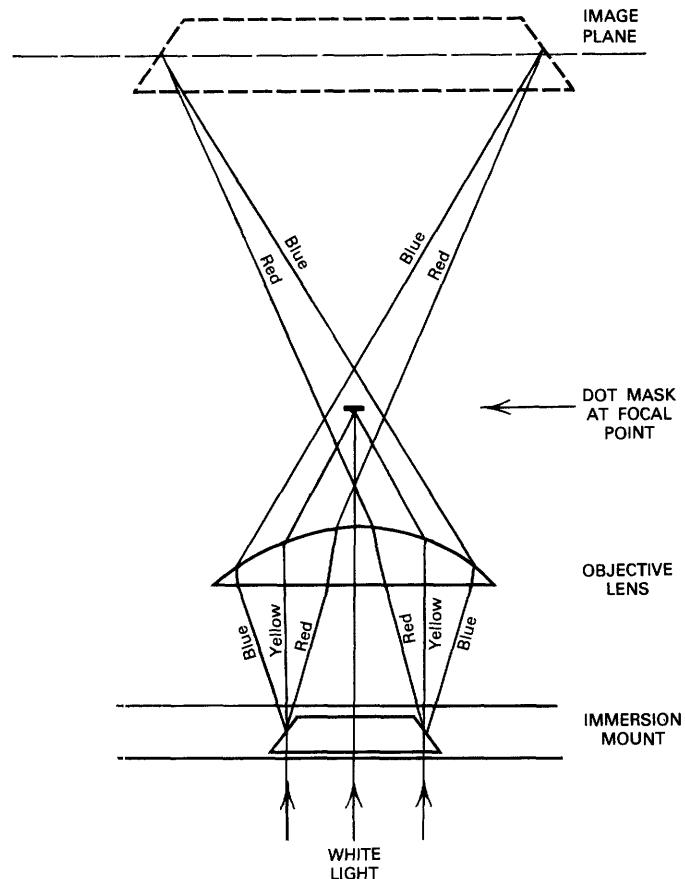


Figure 2. Formation of violet image (spectrum minus yellow band) of a fragment whose refractive index matches the liquid for the yellow band of wavelengths of light.

line, but also the dispersion curve. For greater sensitivity in determination of the index match between liquid and immersed solid, Saylor (1935) developed a double-diaphragm mask for the microscope objective and substage. Used with an ultrasensitive stage refractometer (Saylor, 1966), this mask enables precision to the fifth decimal place in determining the refractive index. Chao (1976) attains precision to the fourth place by interference microscopy.

Determination of Refractive Index of Optically Isotropic Substances

Light in an optically isotropic substance can vibrate in all directions in a plane perpendicular to its direction of transmission, and the refractive

Table 1. Dispersion colors observed at edges of fragments on dark field with central focal masking technique
 [Modified from Cherkasov, 1960]

Matching wavelength (nm)	Observed color			Refractive index n_D of solid is	
	At focus	Raising focus			
		Moving in	Moving out		
660	bright greenish blue	(none)	bright greenish blue	less	
625	sky blue	(none)	sky blue	than	
600	blue	faint dark red	greenish blue	liquid	
<u>589</u>	DEEP VIOLET	WEAK RED	STRONG BLUE	SAME AS LIQUID	
575	purple	red	blue		
540	reddish purple	orange-red	bluish-violet	greater	
505	orange-red	orange	weak blue violet	than	
480	orange	yellow	weak violet	liquid	
465	bright gold	bright gold	(none)		

index is the same regardless of the directions of passage and vibration. Hence the refractive index of such a substance may be determined in a straightforward manner by making a series of immersion mounts of randomly oriented crushed fragments to find the matching liquid. Having found the liquid that matches the fragments, an appropriate correction for the temperature may be applied, as outlined in the section entitled "Immersion Liquids."

Determination of Refractive Indices of Anisotropic Substances

Light traveling in an anisotropic substance is polarized, that is, its vibrations are confined to two directions that are perpendicular both to each other and to the direction of transmission. Refractive indices depend on vibration directions and may vary between limits set by the extreme values of the principal refractive indices, determinations of which are necessarily more involved than for an isotropic substance because the fragment first must be properly oriented.

To aid in visualizing the behavior of light in anisotropic crystals, one may make use of the optical indicatrix, an ellipsoid in which the length of a given radius represents the refractive index of the light wave vibrating in the direction of the radius. Thus, the indicatrix of an isotropic substance is a sphere. Anisotropic crystals may be divided into two classes: uniaxial and biaxial. The indicatrix of a uniaxial crystal is an ellipsoid of revolution (fig. 3A), having its axis (called the optic axis) coincident with the crystallographic symmetry axis c , perpendicular to which is the circular section. Light traveling in the uniaxial crystal at an angle to the optic axis is polarized. One wave (called the ordinary or O wave) vibrates in the plane of the circular section and has a refractive index ω , which is represented by the radius of the circular section. The other wave (called the extraordinary or E wave) vibrates at right angles to that of ω

(fig. 3A) and has a refractive index ϵ' which is represented by the radius of the ellipsoid in vibration. The index ω is called the ordinary principal refractive index, and ϵ' is called the extraordinary refractive index. The extreme value of ϵ' , obtained when light travels perpendicular to the optic axis, is designated as ϵ , the extraordinary principal refractive index.¹ Note that, in this case, the other polarized wave again has ω as its refractive index. Light traveling parallel to the optic axis is not polarized and has the ordinary principal refractive index ω . Crystals in which ϵ is greater than ω are said to be optically positive (the indicatrix is a prolate ellipsoid, such as that of fig. 3A). Where ϵ is the lesser index (an oblate ellipsoid), the crystal is said to be optically negative.

The optical indicatrix of a biaxial crystal (fig. 3B) is an ellipsoid in which the three unequal symmetry axes represent the principal vibration directions X, Y, and Z of light for the three principal refractive indices: α , the lowest; β , intermediate; and γ , the highest.² Such an ellipsoid has two circular sections, the normals to which here are called the optic axes. These axes define the optic plane, and the angle between them is the optic angle $2V$. The optic plane includes the principal vibration directions X and Z as bisectrices of the optic angle. The crystal is said to be optically positive when $2V$ is acute about Z and optically negative when $2V$ is acute about X.

¹Other symbols for refractive indices used in some publications are, for ω : n_w , n_0 , n^0 , or merely 0,

for ϵ : n_ϵ , n_E , n^E , or E.

²Other symbols used in some publications are,

for α : n_α , n_x , n_p , or merely X,

for β : n_β , n_y , n_m , or Y,

for γ : n_γ , n_z , n_g , or Z.

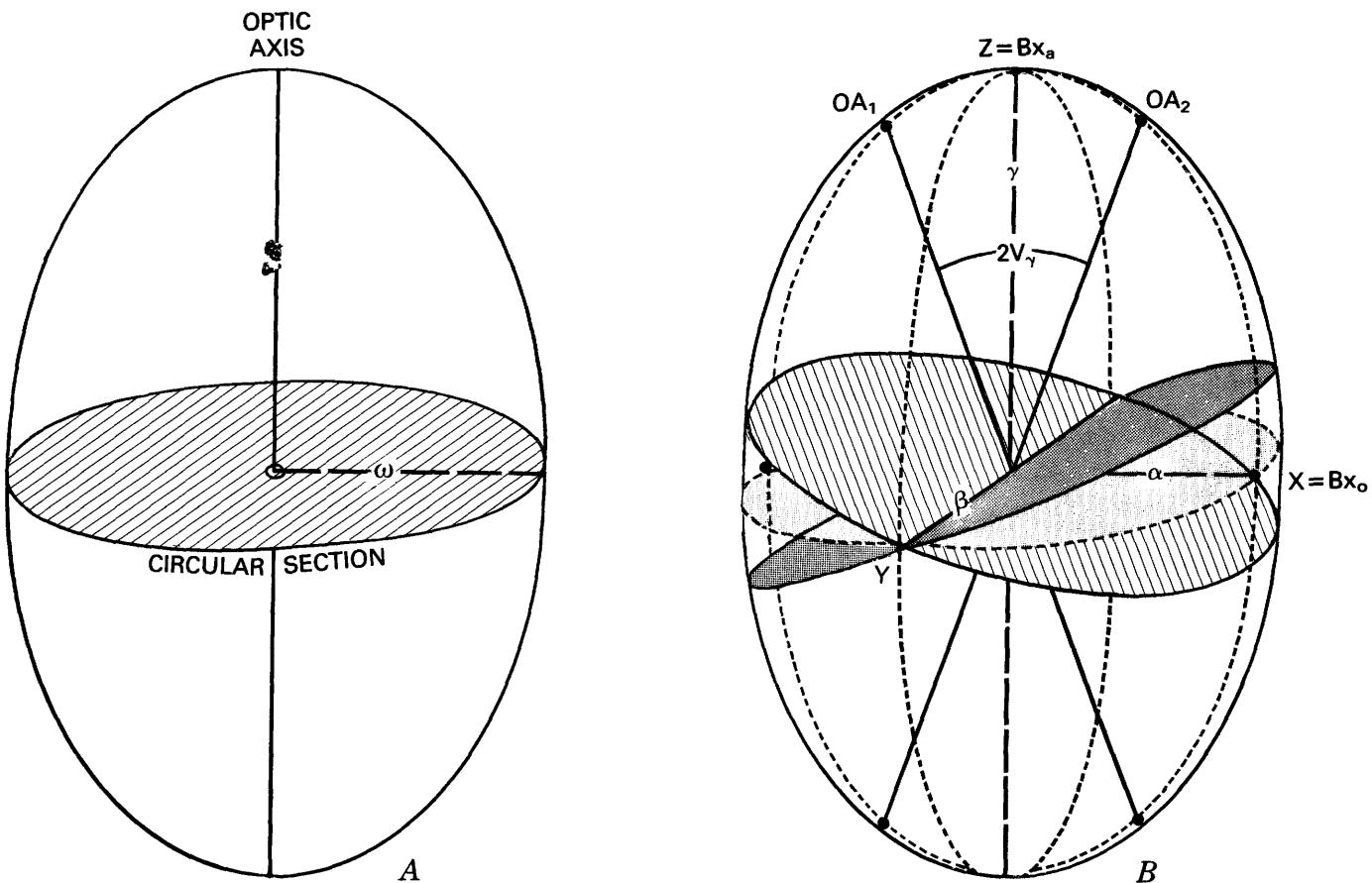


Figure 3. Indicatrices of anisotropic crystals. *A*, Uniaxial positive (a prolate ellipsoid of revolution, $\omega < \epsilon$). *B*, Biaxial positive ($2V_z < 90^\circ$).

Light traveling parallel to an optic axis vibrates in the circular section, is not polarized, and has β as its refractive index. (Although optic axes are customarily shown in the indicatrix, as in figure 3, it should be kept in mind that they really represent directions of travel and that the corresponding fundamental vibrational elements are the circular sections.) Light traveling at an angle to the optic axes is polarized into two waves vibrating at right angles to each other. One wave vibrates in the dihedral angle of the circular sections that includes the bisectrix X , and in the general case it has refractive index α' , a nonprincipal index between α and β . The other wave vibrates in the other dihedral angle that includes the bisectrix Z and has refractive index γ' between γ and β . Light traveling parallel to a symmetry plane of the indicatrix, XY , XZ , or YZ , will

have one wave with a principal refractive index, γ , β , or α , and generally ally the other wave with a nonprincipal index. Light traveling parallel to a symmetry axis X , Y , or Z will have both waves with principal refractive indices.

As can be seen from the foregoing discussion, to determine a principal refractive index of an anisotropic crystal by the immersion method, the crystal fragment must be oriented so that the plane-polarized light from the polarizer of the microscope vibrates parallel to the corresponding principal vibration direction in the crystal. In practice, one has a choice of (1) the "method of random grain mounts," in which properly oriented fragments are sought in successive immersion mounts of multiple fragments, (2) the coated-slide technique, in which one seeks properly oriented fragments among the scattered fragments held on a thin film of adhe-

sive on a slide and then changes liquids as necessary to reach a match, or (3) a method wherein single fragments are mounted on a spindle for orientation so that each principal refractive index can be determined in its respective position.

The method of random grain mounts.--This method, described in varying detail in textbooks, is carried out using loose-grain mounts in a chosen succession of immersion liquids. It needs no special auxiliary equipment but does require a thorough and logical procedure to ensure useful results in a reasonable time. The first step is to examine the behavior of the interference figures of a series of grains in a preliminary immersion mount to ascertain whether the unknown is uniaxial or biaxial. If uniaxial, every grain should show the ω -index at one or the other extinction position; one may, therefore, make a series of mounts in liquids leading to a match of ω . In another series of mounts, the other principal index, ϵ , is to be sought from grains showing highest interference colors, the correctness of the orientation being confirmed finally by the interference figure, which should be a symmetrically behaving flash figure.

For a biaxial substance, successive mounts may be made to match the index of grains that show very little or no interference color on the rotation between crossed nicols. This index is β , confirmed by a centered optic axis figure. Proceeding next through a series of mounts of progressively lower index and observing fragments showing relatively high interference colors, the extreme low index is sought; this will be the principal index, α , confirmed by an interference figure symmetrical about the crosshair that is perpendicular to the vibration direction of the polarizer. Helpful details in the use of such symmetrically behaving interference figures in this procedure are given by Slawson and Peck (1936) and by Stoiber and Morse (1972, p. 188-198, 211-214). (See also discussion in sec-

tion on "Spindle Stage," below). Similarly, the highest index, γ , is sought in grains showing high interference colors in mounts of index successively greater than β , again confirmed by the appropriately symmetrical interference figure.

If the unknown has marked cleavage oblique to the principal vibration directions, the chances are much reduced of finding fragments oriented suitably for determination of all principal refractive indices. Further, if the crushed sample contains two or more substances of overlapping optical properties, the unsophisticated or impatient investigator may settle on a set of totally misleading principal refractive indices and birefringence for the unknown "substance."

The coated slide technique.--In this technique, fragments of the unknown are scattered over a glass slide thinly coated with tacky material, such as gelatin (Olcott, 1960), thermal-setting epoxy (Langford, 1962), thick water glass (Lindberg, 1944), quick setting epoxy, or other colorless adhesive not affected by immersion liquids. Having once located a suitably oriented fragment by use of interference figures, the liquids can be changed to find the match for the respective principal refractive index without loss or disturbance of the fragment. In many cases, this is a considerable improvement over the method of random loose-grain mounts, in that new fragments need not be sought for each change of liquid. But this technique suffers also from the hazards and uncertainties arising when the unknown is composed of two or more substances of overlapping optical properties.

Spindle Stage

Most satisfactory for entry into the identification tables is a complete set of basic optical data from the unknown crystal fragment. For their determination, the spindle stage is well suited, enabling determination of all principal refractive indices as well as

optic angle and relation of optical symmetry axes to cleavage.³ A succession of fragments should of course be examined to establish the range of optical properties of the unknown mineral and whether the sample is composed of more than one mineral species. Procedures for the use of the spindle stage are given by Rosenfeld (1950), Wilcox (1959), Wilcox and Izett (1968), Hartshorne and Stuart (1970), Bloss and Light (1973), and Bloss (1981). The essentials will be outlined in the following.

To carry out the determination on the spindle stage, the crystal fragment is cemented on the tip of the spindle and placed on the base plate (fig. 4) attached to the stage of the microscope. With the fragment in the transparent immersion cell, orientation is accomplished by rotation about the horizontal (spindle) axis and vertical (microscope) axis. At the orientation for each principal refractive index, immersion liquids are changed successively to arrive at the match.

The adhesive for mounting the fragment must be insoluble in the immersion liquids, and ordinary water-soluble carpenter's glue works well for this if mixed with an appropriate amount of molasses, corn syrup, or similar substance to lengthen its period of tackiness. The mineral fragments, crushed to a convenient size (usually less than 0.2-mm diameter), are viewed under a stereomicroscope, and the desired fragment is isolated. The tip of the spindle is touched to a drop of the adhesive, then brought into the field of the stereomicroscope, holding the spindle as nearly vertical as practical and barely touching it to the fragment (fig. 5A).

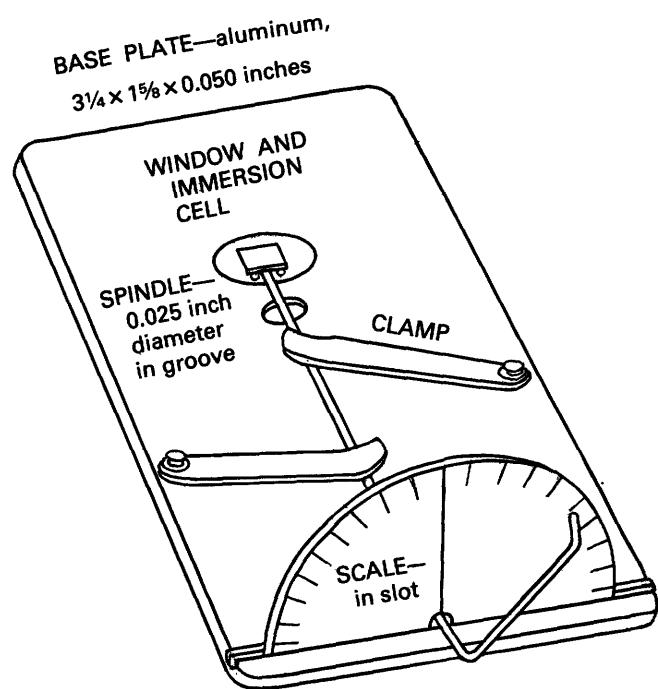


Figure 4. Spindal stage (modified from Wilcox, 1959).

With the fragment clinging to the spindle tip, the mount may be breathed on gently, whereupon the moisture-sensitive adhesive flows onto the fragment to hold it more securely. (Care must be taken, however, not to moisten it so much that the adhesive completely coats the fragment.) If the fragment is out of line, as in figure 5B, it may be gently nudged with a dissecting needle while viewed under the stereomicroscope to move it into the axis of rotation of the spindle (fig. 5C).

After allowing a minute or two for the adhesive to set, the spindle with fragment thus randomly mounted is clamped in the groove of the base plate, then pushed forward to insert the fragment into the immersion cell, and the small scale for the spindle arm is put in place. A trial immersion liquid is introduced at the front end of the cell, into which it moves by capillary action. Under low power, the image of the fragment is brought to the crosshair intersection by shifting the base plate. Liquids can be changed by holding a tab of blotting paper at the rear of the cell along the shank of the spindle (fig. 6). After the old liquid is

³The universal stage (Emmons, 1943) also enables determination of these diagnostic properties to a high precision but requires an array of peripheral equipment as well as a much greater expenditure of time.

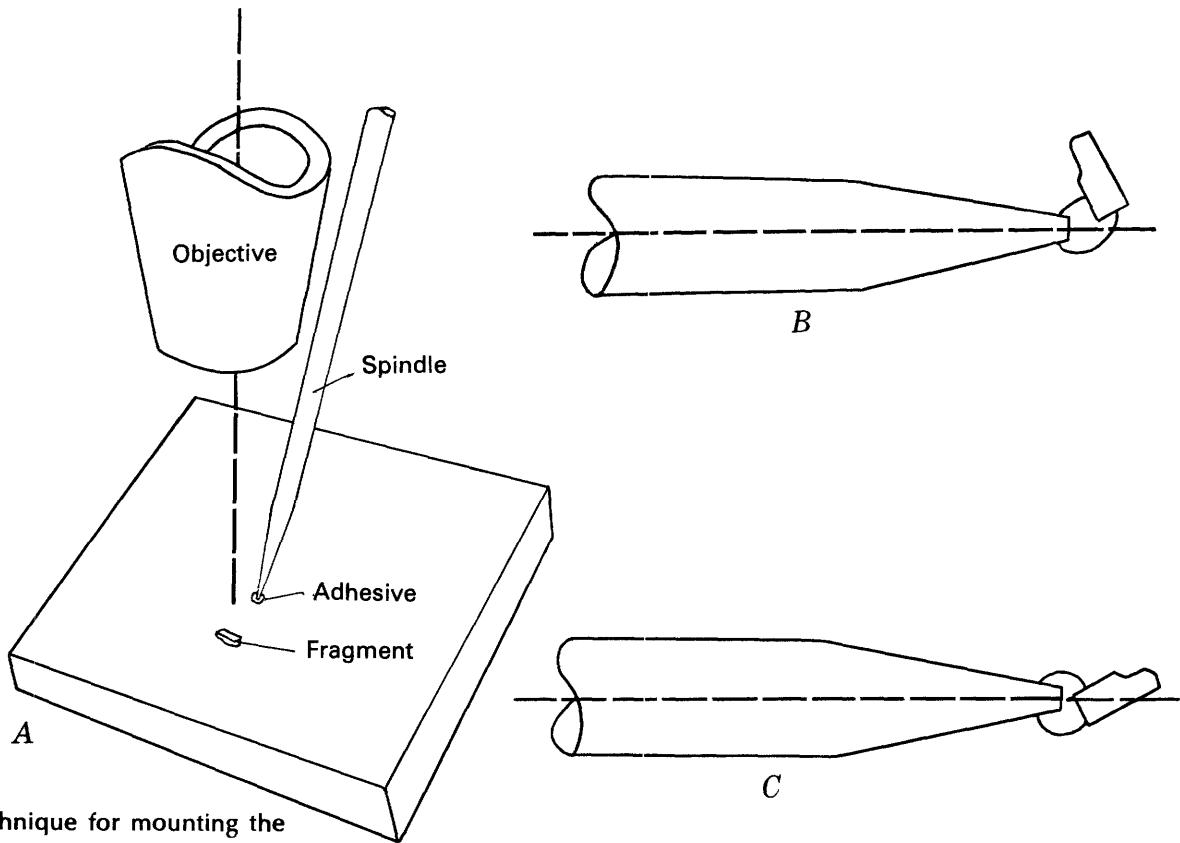


Figure 5. Technique for mounting the fragment on spindle: *A*, The spindle with adhesive is touched to the fragment. *B*, Fragment out of line. *C*, Fragment nudged into line with the spindle axis.

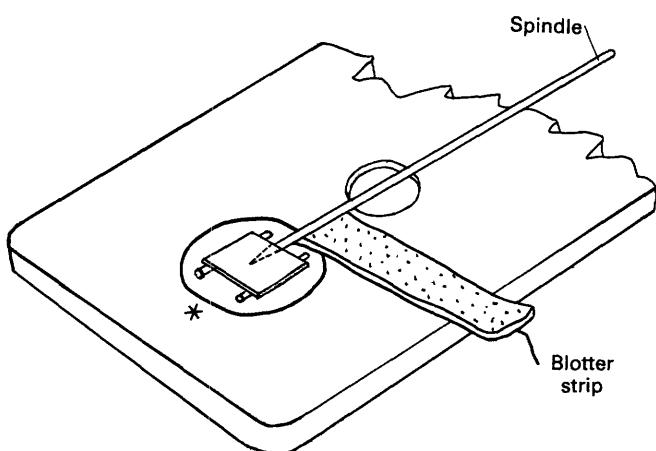


Figure 6. Withdrawal of liquid from immersion cell using absorbent paper.

removed, the new liquid is introduced at the front of the cell as before. This operation should be repeated once to flush out old liquid trapped in the cell and once again if the index change is large.

Orientation of the fragment for index determination may be carried out conoscopically (Wilcox, 1959, p. 1277), enabling at the same time an overview of other optical properties such as optic angle and optic sign. If the fragment is too small or if interference figures cannot be obtained, the somewhat more time-consuming orthoscopic procedure (Wilcox, 1959, p. 1281; Bloss and Riess, 1973; Bloss, 1981) can be used.

For a biaxial crystal, the correct orientation for index comparison is recognized conoscopically by an interference figure having bilateral symmetry along the crosshair perpendicular to the plane of the polarizer, as pointed out by Slawson and Peck (1936). Typical examples are shown in figure 7, in which the polarizer is NS. For any randomly mounted fragment, there are three orientations (within the upper hemisphere of rotation) that satisfy this criterion and at which the respective principal refractive indices are to be determined. It is advantageous to start with the orientation for β , that is, the orientation where the optic plane isogyre is symmetrically disposed along the EW crosshair. This isogyre is recognized, even before reaching the symmetrical setting, by its "counterrotating" behavior (Stoiber and Morse, 1972, p. 122, 172) or by the presence of the melatope itself, about which the isogyre pivots. With this isogyre symmetrically along the EW crosshair, the settings of spindle stage arm and microscope stage, at which β is to be determined, are recorded.

One then proceeds toward the setting for the next principal index by small rotations about the spindle axis followed by compensating rotations on the microscope stage to keep the one isogyre always well within the field of view. Eventually another symmetrical interference figure must be encountered, with an isogyre lying either along the EW crosshair, which is the desired setting, or along the NS crosshair, in which case the desired setting is made by rotating 90° on the microscope stage. The accessory plate may be used to establish whether it is the vibration direction of α or γ that is now NS, and the readings of spindle arm and microscope stage are recorded. One proceeds in the same manner to establish the setting for the third principal index. Converting now to orthoscopic illumination and a lower power objective, the index of the liquid is compared with each of the three principal indices at its respective setting by means of one of the index comparison techniques out-

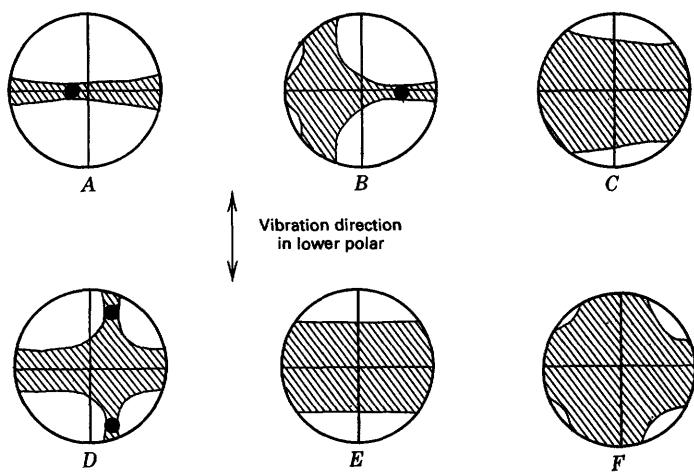


Figure 7. Examples of typical isogyre configurations to be seen when a principal vibration direction is horizontal EW; that is, when the crystal is oriented for comparison of the corresponding principal refractive index with the immersion liquid (polarizer NS).

lined in earlier sections. Liquids are changed accordingly to find the matching liquid for each index and, after correcting for temperature, the principal refractive indices are recorded.

The procedure for a uniaxial crystal is somewhat simpler, once having established from an optic axis figure that it is indeed uniaxial. Here one simply places an isogyre along the EW crosshair and, by incremental rotations of spindle and stage, "follows" this isogyre to the flash figure. The vibration direction of ω is now NS and, converting to orthoscopic illumination at this setting, ω may be compared with the liquid. Rotating 90° to the other extinction position similarly permits ϵ to be compared with the liquid.

Birefringence

The difference between the extreme principal refractive indices, α and γ , or ω and ϵ , ordinarily referred to as the birefringence of the mineral, is very helpful in identification. Birefringence is especially useful for identification of an unknown that is a member of an isomorphous series, for birefringence usually remains within relatively narrow limits while the refractive indices may have a wide range.

In routine immersion work, birefringence is usually taken as the difference between the already measured values of γ and α , or of ω and ϵ . Lacking these measurements, it can be estimated from the order of interference color observed between crossed polars, giving due consideration to the orientation and the probable thickness of the fragment. (This approach, of course, is the basis for rapid recognition of many minerals in thin section, where the thickness is known.) Maximum birefringence may also be determined on the spindle stage, following the suggestion of Fisher (1960, p. 87), by mounting the fragment on the spindle so that the optic plane can be made horizontal (flash figure), in which position the maximum retardation, R , is determined at the point where thicknesses is greatest using a compensator.⁴ That thickness, t , is measured with the ocular micrometer after rotating the fragment 90° about the spindle axis to show it in profile. Birefringence, B , is then $B = R/t$.

Optic Angle and Optic Sign

The optic angle, usually designated as $2V$, is the angle between the two optic axes (see fig. 3). A statement of the numerical value of the angle is not complete without an indication of optic sign. A positive optic angle may be shown, for example, as $(+)2V=47^\circ$, $2V_z=47^\circ$, or $2V\gamma=47^\circ$. The size of the optic angle can be calculated from the relationships among the three principal indices of refraction or can be measured independently, thus offering a means of checking on the consistency of the measured values of principal refractive indices.

Calculation of Optic Angle

The relation of optic angle to the principal refractive indices may be expressed as

$$2V_z = 2 \operatorname{arc} \cos \frac{\alpha}{\beta} \sqrt{\frac{\gamma^2 - \beta^2}{\gamma^2 - \alpha^2}} \quad (1)$$

where $2V_z$ is the angle between the two optic axes across the bisectrix Z . A simpler approximation,

$$2V_z = 2 \operatorname{arc} \cos \sqrt{\frac{\gamma - \beta}{\gamma - \alpha}} \quad (2)$$

⁴If the optic plane of the randomly mounted fragment is oblique to the spindle axis, it cannot be made horizontal by simple rotation about the spindle axis. With practice, however, it is not difficult to correct the position of the fragment by (1) rotating the optic plane to verticality, (2) noting the angle of the optic plane with the spindle, (3) withdrawing the spindle, (4) dipping the spindle tip in solvent to remove the oil, (5) breathing gently on the mount to soften the adhesive, and finally (6) under the stereomicroscope with a dissecting needle, nudging the fragment the estimated amount necessary to place the optic plane in line with the spindle.

is accurate within a degree or two if the optic angle or birefringence is not too high (see also Wright, 1951; Tröger, 1952).

Convenient nomograms based on the exact equation (Mertie, 1941) are reproduced by Bloss (1961, p. 158) and by Hartshorne and Stuart (1970, p. 393, and folding plate). Knowing any three terms of equation (1), the fourth term may quickly be determined from the nomogram. Thus, it is easy to determine not only the nominal value of $2V$, but also the range of uncertainty due to the estimated inaccuracies in the refractive index determination, as illustrated by the example in table 2 (see also

Table 2. Range of calculated optic angles, taking into account a possible $\pm .001$ error of refractive index determinations

	α	β	γ	2V by exact formula	2V by approx. formula	2V by nomogram
Measured refractive indices ($\pm .001$)	1.534	1.545	1.560	(+)81° 46'	(+)81° 10'	(+)82°
Lowest worst case	1.535	1.544	1.561	(+)72° 40'	(+)72° 10'	(+)73°
Highest worst case	1.533	1.546	1.559	(+)89° 24'	90° 00'	(+)89°
Range, lowest to highest				16° 44'	17° 50'	16°

Willard, 1961). Obviously it is more realistic and informative to state the calculated value as $(+)2V_{\text{calc}}=82^\circ \pm 8^\circ$ than to imply an unjustifiable precision, such as $(+)2V_{\text{calc}}=81^\circ 46'$.

An approximate value of optic angle, quite adequate for much work, may be obtained from figure 8A. This nomogram is based on equation (2) using the form proposed by Tröger (1952). Like Tröger's nomogram, its error is considerably less than the error resulting from the usual uncertainties in the determination of the principal refractive indices themselves, especially for minerals of low or moderate birefringence. Should high birefringence justify refinement, a correction, to be added algebraically to the approximate value, may be obtained from figure 8B.

As an example, using the following measurements of a specimen of augite:

$$\begin{aligned}\alpha &= 1.680 \pm 0.001 & \gamma - \beta &= 0.019 \\ \beta &= 1.686 \pm 0.001 & \gamma - \alpha &= 0.025 \\ 2V_z & \text{(measured)} = 61^\circ \pm 2^\circ\end{aligned}$$

$$\gamma = 1.705 \pm 0.001 \quad \frac{\gamma - \beta}{\gamma - \alpha} = 0.760$$

From fig. 8A:
approximate $2V'_z = (+)58.7^\circ$

From fig. 8B:

correction $+ 0.6$
Corrected calculated
optic angle, $2V_z = 59.3^\circ$,
compared to value obtained from equation (1): $2V_z \text{ (calc)} = 59.2^\circ \pm 11^\circ$.

Similarly, for a specimen of hastingsite:

$$\begin{aligned}\alpha &= 1.653 \pm 0.001 & \gamma - \beta &= 0.010 \\ \beta &= 1.665 \pm 0.001 & \gamma - \alpha &= 0.022 \\ 2V_x & \text{(measured)} = 80^\circ \pm 2^\circ\end{aligned}$$

$$\gamma = 1.675 \pm 0.001 \quad \frac{\gamma - \beta}{\gamma - \alpha} = 0.455$$

From fig. 8A:
approximate $2V'_x = (-)84.8^\circ$
From fig. 8B:
correction $+ 0.6$
Corrected calculated
optic angle, $2V_x = 84.2^\circ$,
compared to value obtained from equation (1): $2V_x \text{ (calc)} = 84.2^\circ \pm 10.5^\circ$.

This relation between optic angle and the three principal refractive indices provides a useful test of the internal consistency of the measured results. If the value of optic angle calculated from measured α , β , and γ is not in reasonable agreement with the optic angle, there must be an error in one or more of the measured values, and a recheck is in order. For tables 6 and

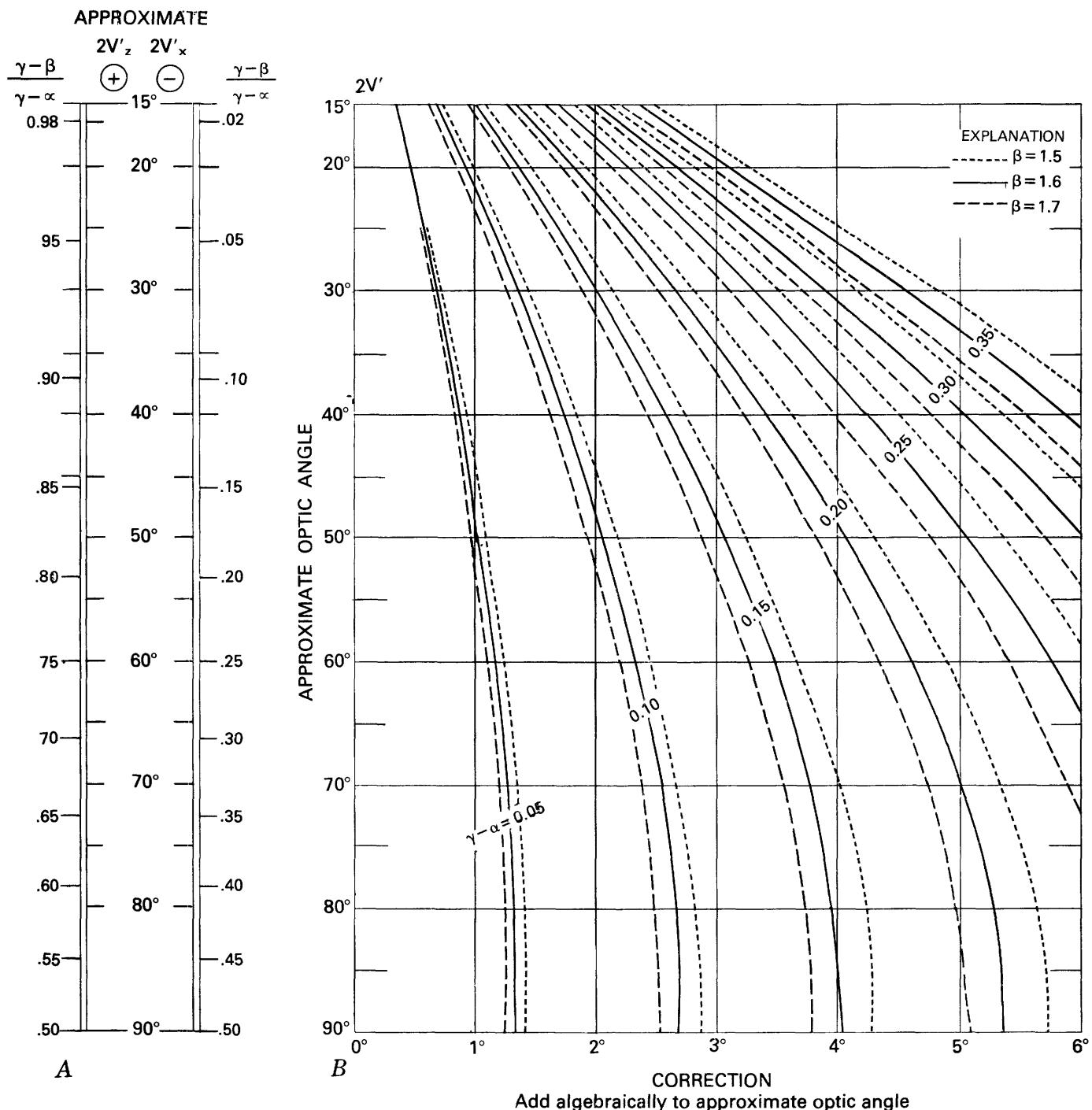


Figure 8. Nomogram for determination of optic angle from ratio of partial birefringences, $\frac{\gamma-\beta}{\gamma-\alpha}$: A, Approximate optic angle. B, Correction to be added algebraically.

7, this test has been run on the data as reported; where agreement is poor, the calculated value of optic angle has been inserted in parentheses below the reported optic angle as a cautionary signal. Also shown in the parentheses is a tolerance angle based on assumed "worst case" errors of ± 1 significant

figure in the values reported for principal refractive indices.⁵

⁵Unfortunately few data in the literature are accompanied by estimates of error. The "worst case" assumed here may err on the side of charity for many of the reported data and may be too pessimistic for others.

Measurement of Optic Angle

Among several means available to measure optic angles of crystal fragments in immersion mounts are the following:

(1) direct measurement conoscopically on the universal stage or spindle stage by placing the optic plane perpendicular to the rotation axis, then rotating from one melatope to the other across an identified bisectrix. Somewhat less reliably (Fairbairn and Podolsky, 1951), the optic angle may be determined by doubling the angle of rotation from one melatope to a bisectrix;

(2) measurement of the angle between stereographically plotted optic axis positions, as determined conoscopically by an extension of Mallard's method on the spindle stage (Wilcox and Izett, 1968; Noble, 1968);

(3) stereographic plotting of a series of orthoscopic extinction positions, determined on the spindle stage, from which the unique positions of the optic axes may be inferred (Wilcox, 1960; Tocher, 1962; Bloss and Reiss, 1973; Bloss, 1981). This procedure is analogous to the Berek-Dodge Method for universal stage (Emmons, 1943, p. 28), without requiring corrections of rotations.

Whatever the method of determining the optic angle, the identity of the acute bisectrix, X or Z, must also be determined to establish the optic sign. Direct measurement of optic angle by rotation from one optic axis to the other is of course to be preferred when the situation permits. On the universal stage, the optic plane may be brought directly into position for the rotational measurement, to which are applied corrections for index difference between mineral and hemispheres. On the spindle stage, the fragment may be nudged into position after the adhesive has been softened, and no correction to the rotational measurement is required. All three methods provide greater accuracy than can usually be obtained by calculation of optic angle from principal refractive indices, particularly for

substances of low or moderate birefringence.

Estimation of Optic Angle and Determination of Sign

For rough work, one may use the method developed by Wright (1911, p. 168) to obtain an estimate of optic angle. This requires an optic axis interference figure, that is, a melatope near the center of the conoscopic field, and judgment of the curvature of the isogyre when the optic plane is at 45° to the polars. The dark band of the isogyre is distinctly curved for small optic angles and straight diagonally across the field for optic angles approaching 90°. Estimates of optic angle to within about 10° can be made using figure 9, modified from Wright's

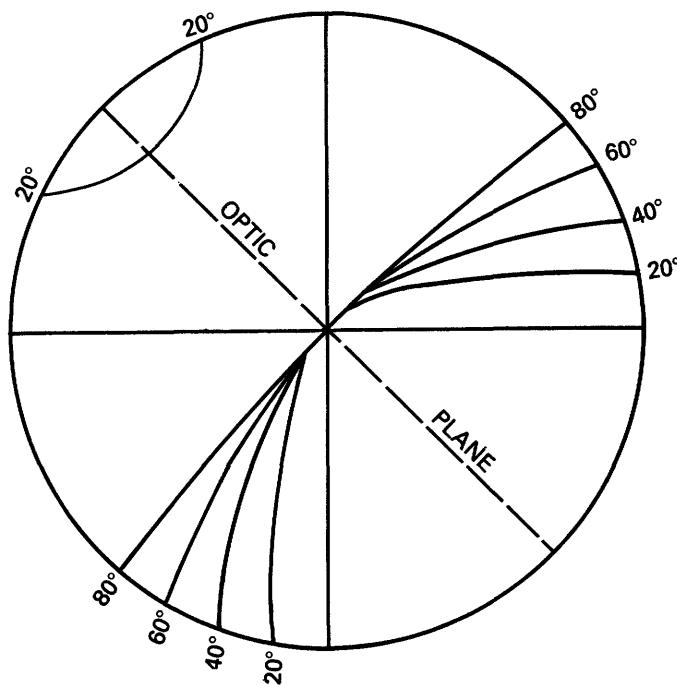


Figure 9. Curvatures of superimposed isogyres for centered optic axis figure of optic angles 20°, 40°, and 80° (after Wright, 1911).

(1911) composite of superposed isogyes, keeping in mind that objectives of higher numerical aperture will display a greater length of isogyre (see Winchell, 1965, p. 11). Determination of optic sign is made on the same interference figure by placing the optic plane in the diagonal position parallel to the vibration direction of the slow ray (high index) of the inserted accessory plate. For an optically positive crystal, insertion of the plate raises the retardation color on the concave side of the isogyre at the melatope.⁶

When both melatopes lie symmetrically disposed in the conoscopic field, one may use Mallard's method sensu stricto, on the basis of separation of the melatopes (Johannsen, 1918). When the melatopes are not symmetrically disposed, one may still estimate the size of the optic angle by comparison of their separation with the total diameter of the conoscopic field of known angular aperture. With one melatope and a bisectrix in the conoscopic field, the half-optic angle can be estimated in a similar manner but with reduced precision. When both melatopes lie outside the conoscopic field of a near-centered bisectrix figure or flash figure, the optic angle may be estimated roughly by the amount of rotation of the microscope stage that is required to move the isogyes from the crossed position to a certain radial distance representing a known angular aperture (Johannsen, 1918, p. 472; Kamb, 1958). Inasmuch as it is the darkest band of the diffuse and

varying isogyre that must be set at the reference radial distance, the latter is best chosen well within the field rather than at the extreme outer edge, as done in the theoretical treatment by Kamb (1958). A useful discussion of this and other conoscopic methods is given by el-Hinnawi (1966, p. 114-136).

Dispersion of the Optic Axes

If the directions of the optic axes change with wavelength, the mineral is said to have dispersion of the optic axes, a characteristic often helpful in distinguishing it from otherwise similar minerals. This property is included in the tables together with the nominal value of $2V$ for the standard yellow light. For example, for the biaxial positive minerals (table 6), wavellite at $\beta=1.524$ shows $2V_z$ as "60°, $r > v$ wk," meaning that the optic angle for red is slightly greater and for blue slightly less than the 60° for yellow.

Dispersion of an optic axis may be recognized in white light illumination using a near-centered optic axis figure with the optic plane 45° to the polars. Because the melatope for each wavelength (color) is a point of no illumination and because these melatopes are now slightly spread out along the trace of the optic plane, the melatope for the red wavelengths is represented by a bluish color, while that for the blue wavelengths is represented by a reddish or, more commonly, brownish color. Thus, if the concave side of the isogyre is seen as bluish and the convex side reddish or brownish, it is concluded that the optic angle for red is greater than that for blue wavelengths, $r > v$.

⁶To avoid misreading the direction of curvature, especially for a crystal of large optic angle, a setting is first made with the isogyre parallel to (not necessarily on) the NS crosshair. The stage is then rotated exactly 45° to read the curvature of the isogyre. The farther the melatope from the center of the field, of course, the smaller is the length of isogyre available on which to estimate curvature and, in general, the less precise is the estimate.

Relation of Optical to Crystallographic Directions

The orientation of the indicatrix with respect to crystallographic elements, such as crystal axes, faces, cleavages, and twin planes, can be of much help in identification of an unknown, particularly after the number of

candidates has been reduced to a manageable few on the basis of the observed principal refractive indices, optic angle, and sign. Firm establishment of the relation of indicatrix axes to crystallographic axes would require coordinated optical and X-ray structural investigation of the same crystal fragment (see for instance Chao and Minkin, 1970). For many unknowns, however, much can be inferred from the behavior of interference figures of fragments lying on a prominent cleavage (see for instance Ehlers, 1980) and from such readily determinable properties as extinction angle and sign of elongation.

Extinction Angle

The extinction angle is the angle through which the microscope stage must be rotated to change it from a position in which a specified crystallographic direction is parallel to the crosshair to the extinction position in which the specified vibration direction is parallel to the crosshair. The vibration direction may be a principal symmetry axis of the indicatrix, such as X or Z, or, when the symmetry planes of the indicatrix are oblique to the section, merely an extinction position expressed then as X' or Z'. If the extinction angle is not given as such in tables 6 and 7, it often can be inferred from the basic information provided in the columns on "Optical orientation," "System," and "Cleavage." Helpful in visualizing these relationships are the admirable perspective sketches of typical crystals in Tröger (1979).

Extinction angles serve to distinguish between certain important minerals or mineral groups, for instance, between clinopyroxenes and clinoamphiboles using the angle Z:c. Extinction angles serve further to distinguish between members in an isomorphous series, such as the plagioclase series using X':a. On the spindle stage the crystallographic directions of a fragment may be inferred by orienting and plotting the bounding cleavage surfaces, from which the specified extinction angle can be read off

the stereogram. Alternatively the fragment can be mounted with a recognizable crystallographic zone axis parallel to the spindle axis so that the extinction angle can be read directly.

Sign of Elongation

The sign of elongation is positive if a characteristic direction of elongation of the fragment or crystal is less than 45° to the extinction position of the slow wave (high index). A mineral with two directions of good or excellent cleavage will generally break into elongate fragments. Thus, cleavage fragments of most amphiboles show positive elongation, since Z is at a low angle to c, defined by the intersection of the excellent prismatic cleavages. For most micas, both Y and Z lie in or very near the plane of the one perfect cleavage, so that cleavage flakes, viewed edge-on in thin section or on the spindle stage, show positive elongation. In random immersion mounts, of course, this relation is generally not seen because the flakes lie on their cleavage.

Color and Pleochroism

The color of the hand specimen, together with its luster, can be helpful in sight-identification of some minerals. Caution is required here, however, because the color can be affected appreciably by foreign inclusions, internal flaws, fractures, or by variations of certain minor chemical constituents. Under the stereo-microscope with standard conditions of illumination, small fragments may show characteristic colors and lusters, enabling approximate modal counts and removal of certain species by hand picking.

In plane-polarized transmitted light, the colors of small fragments are diagnostic of certain mineral species. Pleochroism, the change in color with the change in direction of vibration of the plane-polarized light, is especially useful in sight identification. Pleochroism is customarily described in

terms of colors observed for light vibrating parallel to the respective axes of the indicatrix: O and E for uniaxial crystals and X, Y, and Z for biaxial crystals. (Strictly speaking for monoclinic and triclinic crystals, however, the principal axes of the indicatrix do not necessarily coincide with the principal axes of the "absorption ovaloid." Clinopyroxenes provide an example, for which a more consistent convention was proposed by Hess (1949, p. 632).)

Depth of color for light transmitted through fragments in immersion mounts increases with the thickness of the fragment, so that the intensities of the observed colors may not correspond strictly to the pleochroic colors listed in tables 4-7. An added complication may arise in immersion mounts due to changes in dispersion colors or relief with rotation of the stage, for either may simulate pleochroism or mask a faint body color. These disturbing effects can largely be eliminated by swinging in the substage auxiliary condenser to flood the crystal with light. One must keep in mind, however, the differences in the spectral distribution of the source illumination from one microscope to another. In addition, one must be aware of the subjective nature of the descriptive terms used in the tables; these terms might be interpreted differently by different observers.

CHAPTER III. IDENTIFICATION TABLES (TABLES 3-7)

List of Abbreviations

a-----	crystallographic axis a
Ab-----	albite
abs-----	absorption
acic-----	acicular
Ak-----	Akermanite
Al-----	Almandine
alk-----	alkaline
amor-----	amorphous
An-----	Anorthite, Andradite
anom-----	anomalous
b-----	crystallographic axis b
bb-----	before the blowpipe

biax-----	biaxial
biref-----	birefringence, birefringent
c-----	crystallographic axis c
Ca-----	Calderite
calc-----	calculated
char-----	character
clv-----	cleavage
Cn-----	Celsian
cols-----	colorless
comp-----	composition
compd-----	compound
concd-----	concentrated
conch-----	conchoidal
cryptocryst-----	cryptocrystalline
cryst-----	crystalline
crystn-----	crystallization
cub-----	cubic
dec-----	decomposed
dehyd-----	dehydrates
deliq-----	deliquesces, deliquescent
detd-----	determined
diff-----	difficult, difficulty
dil-----	dilute
disp-----	dispersion
diss-----	dissolved
dist-----	distinct
dod-----	dodecahedral
E-----	extraordinary wave
eff-----	effervesces, effervescence
effl-----	efflorescent
el-----	elongation
En-----	Enstatite
ext-----	extinction
extr-----	extreme
F-----	fusibility
Fa-----	Fayalite
fib-----	fibers, fibrous
fluor-----	fluoresces
Fo-----	Forsterite
fr-----	fracture
Fs-----	Ferrosilite
fus-----	fusible
G-----	specific gravity
Ge-----	Gehlenite
gel-----	gelatinizes
Go-----	Goldmanite
Gr-----	Grossularite
grp-----	group
H-----	hardness
hex-----	hexagonal
hyg-----	hygroscopic
imperf-----	imperfect
insol-----	insoluble
isomor-----	isomorphous
isot-----	isotropic

Kn-----	Knorrtingite
lam-----	lamellar, lamellae
mass-----	massive
mcl-----	monoclinic
Me-----	Meionite
med-----	medium
mic-----	micaceous
mod-----	moderate
n-----	refractive index
neg-----	negative
O-----	ordinary wave
oct-----	octahedral
opt-----	optical, optically
Or-----	Orthoclase
org-----	organic
orth-----	orthorhombic
pct-----	percent
penet-----	penetration
perc-----	perceptible
perf-----	perfect
pl-----	plane
pleoc-----	pleochroic, pleochroism
poly-----	polysynthetic
pos-----	positive
pris-----	prismatic
ps-----	pseudo
Py-----	Pyrope
pyram-----	pyramidal
r-----	optic angle in red light
rect-----	rectangular
sepn-----	separation
ser-----	mineral series
sol-----	soluble
soln-----	solution
Sp-----	Spessartite
spher-----	spherulites, spherulitic
sq-----	square
str-----	strong
symm-----	symmetrical
synth-----	synthetic
tab-----	tabular
tcl-----	triclinic
tet-----	tetragonal
tetrah-----	tetrahedral
tr-----	trace
trig-----	trigonal
tw-----	twinning
u-----	usually
uniax-----	uniaxial
unk-----	unknown
Uv-----	Uvarovite
UV-----	ultra-violet
v-----	optic angle in blue light
var-----	variety
volat-----	volatilizes
wk-----	weak
X, Y, Z-----	vibration directions for α , β , γ , respectively
2V _X -----	optic angle about X
2V _Z -----	optic angle about Z
⊥-----	perpendicular to
-----	parallel to
>-----	greater than
<-----	less than
~-----	near, approximately
±-----	plus or minus
001, 001̄1 -----	Miller symbols for all faces of a form (conventional braces have been omitted)
: -----	angle with α , β , γ
lowest, intermediate, and highest principal refractive indices of biaxial crystals, respectively	
ω , ϵ -----	ordinary and extraordinary principal refractive indices of uniaxial crystals

Arrangement of Data

The minerals are divided among five tables according to their basic optical characteristics: isotropic minerals, including mineraloids and natural glasses, are in table 3; uniaxial positive minerals are in table 4; uniaxial negative minerals are in table 5; biaxial and positive minerals are in table 6; biaxial negative minerals are in table 7. The mineral names generally follow those of Fleischer (1983), and the chemical formulas attempt to convey the type of compounds and interchangeable elements without necessarily indicating lattice positions or other structural aspects.

In table 3 the isotropic substances are listed in order of increasing refractive index, n , shown in the second column. In the first column, titled "Other entries," arrows with values of refractive index indicate the next higher or lower listing of the particular mineral. Since the determined

value of refractive index of the unknown will generally fall between those of the representative entries of that mineral in the table, it is essential to give close attention to the indications of index variation shown in this column. Known variations of index too small to justify separate entries are indicated by square bracket symbols, with or without numerical values. The lack of a variability notation in this column opposite a mineral name, however, does not mean that variability does not exist, rather merely that none was found in published descriptions, a point to be kept in mind especially for the lesser known minerals of complex composition.

Next in table 3 are columns for crystal "System and habit" and for "Cleavage or fracture." Here note that the conventional enclosing braces denoting forms are omitted from the Miller symbols because of space limitations. The "Color" column indicates the color of fragments in reflected light. "Hardness, specific gravity, and fusibility" are shown in the next column, and finally in the "Remarks" column are given various additional helpful properties, such as optical anomalies, weight percentages of selected oxides of the chemical analysis, and behavior in chemical tests. Dilute acids are understood unless otherwise specified.

In tables 4 and 5 the uniaxial minerals are listed in order of increasing values of the ordinary principal refractive index, ω , with the extraordinary principal index, ϵ , given in the adjacent column. In addition to the properties already mentioned for table 3, the "Remarks" column of tables 4 and 5 gives information on twinning, sign of elongation, and pleochroism.

In tables 6 and 7, the biaxial minerals are listed in order of increasing intermediate principal refractive index, β , with the values of α and γ in adjacent columns and with birefringence in another column. (Parentheses around an index or birefringence indicate a value calculated from the reported value of optic angle). The column for "2V" gives the reported value of optic angle, and if this value does not appear to be

consistent with the reported values of α , β , and γ , the calculated value of optic angle is given in parentheses, along with a plus-or-minus tolerance based on possible errors of one in the last significant figure of the reported indices. (No calculated values are shown for minerals of low birefringence, since here normal errors in index would lead to wide variations in calculated optic angle). Also shown is the reported dispersion of the optic angle. The column for "Optic orientation" gives diagnostic relationships between indicatrix symmetry axes X, Y, and Z, and crystallographic axes, a , b , and c , as well as extinction angle and sign of elongation. The statement "el clv pos," for instance, indicates positive elongation as seen with steeply dipping cleavage.

Because of the variations of optical character, some minerals are listed in more than one table. For example, a normally isotropic mineral that is weakly anisotropic in some occurrences may be listed in both table 3 (isotropic minerals) and in the appropriate table for anisotropic minerals. Furthermore, in some isomorphous series the optic sign may reverse due to progressively changing relations between the principal refractive indices. The melilite group, for instance, passes from optically uniaxial negative gehlenite through "isotropic" melilite to uniaxial positive akermanite and thus is listed in three tables. In the olivine group, the magnesium-rich members are optically positive, but as the Fe/Mg ratio increases the optic angle passes through 90° near Fa₁₅ and the remainder of the series is optically negative. In table 6 (biaxial positive minerals), therefore, an olivine group mineral is entered at β -values of 1.651, 1.664, 1.674, and finally at 1.680, at which point a switch to table 7 (biaxial negative minerals) is indicated by the notation 1.680 neg. Due to thermal history or other factors, the cross-over in an isomorphous series may not take place precisely at the same value of β -index in every case; so that, with an unknown of observed optic angle near 90°, likely candidates should be sought

in both tables near the measured value of β .

Use of the Tables for Identification

In all cases, a systematic and thorough procedure must be carried out, and the following approach is recommended: For an isotropic unknown, one enters the table at the determined value of refractive index and then examines the other properties given for minerals of index within a reasonable range above and below the determined refractive index. Unfortunately for isotropic crystals there are no other optical-crystallographic parameters easily measured under the microscope, so one must depend heavily on cleavage, hardness, solubility, and so on. Those minerals that correspond to the unknown in these respects should be listed and consideration then given as to whether X-ray, chemical analysis, or another approach will be most effective in making the final assignment of identity.

For an anisotropic unknown, after having determined the values of principal indices, birefringence, optic angle, and optic sign, the following steps are recommended:

(1) enter the appropriate table at the determined value of ω or β .

(2) proceed both above and below in the table for a distance of at least 0.020 difference in index, listing the names of those minerals whose values of birefringence and optic angle are in reasonable agreement with those of the unknown.

(3) for each mineral so listed, check the remaining columns for distinguishing properties, returning to the microscope or the original sample as necessary to make diagnostic tests and crossing out each nonconforming mineral.

(4) for the remaining candidate or candidates, consult encyclopedic optical descriptions for further distinguishing properties.

(5) confirm or complete identification if necessary by X-ray and chemical analyses.

Examples

Example No. 1.--is crushed material from clusters of clear, colorless crystals in small lithophysae in rhyolite. Fragments mounted on the spindle stage show:

$$\begin{array}{ll} \alpha 1.611 \pm .001 & \gamma - \alpha 0.011 \pm .002 \\ \beta 1.614 \pm .001 & 2V_z 65^\circ \pm 2^\circ \text{ measured} \\ \gamma 1.622 \pm .001 & \end{array}$$

Entering the table for biaxial positive minerals (table 6) at $\beta = 1.614$ and proceeding toward lower and higher values of β , the following minerals seem reasonable possibilities (keeping in mind that, due to chemical complexity, some of them may be expected to vary in index even though it is not so indicated in the "Other entries" column).

Looking at the characteristics listed in other columns of table 6, tuhualite can be eliminated at once on the basis of its "dark blue to black" color; probably uranopilite ("bright yellow, straw yellow") can also be eliminated. In a simple hardness test, particles of the unknown leave scratches when rubbed between two glass slides, showing that it has a hardness of 5 or greater; this finding confirms the elimination of tuhualite and uranopilite, and also eliminates crandallite, uralborite, and buchwaldite. Under the microscope, the unknown appears to show one perfect cleavage, casting doubt on stokesite, which should show two cleavages. As a test of density, particles of the unknown in a drop of methylene iodide (sp gr 3.3) on a microscope slide are seen to rest at the bottom of the liquid; this eliminates stokesite (sp gr 3.19), afwillite (sp gr 2.62) and cuspidine (sp gr 2.86-3.05), all of which should float. The unknown is found to be insoluble eliminating hydrophilite, and is unaffected by acids, eliminating latiumite, afwillite, cuspidine, and uranopilite. This leaves only topaz which possesses all the observed characters of the unknown. According to the variation diagram given by Winchell and Winchell (1951, p. 509), the optical properties would imply a high fluorine variety of topaz.

	Other entries	β	Birefringence	$2V_z$
Cuspidine		1.595	.012	62°
Hydrophilite		1.605	.012	med
Latiumite		1.606	.014	83°-90°
Crandallite		1.607	.013	70°-75°
Uralborite		1.609	.011	85°
Topaz		1.610	.011	67°
Buchwaldite		1.610	.009	(?)
Tuhualite		1.612	.013	60°-70°
Stokesite		1.613	.010	70°
(UNKNOWN IN QUESTION)		1.614 ± .001	.011 ± .002	65° ± 2°
Topaz		1.618	.009	61°
Afwillite		1.619	.015	55°
Stokesite		1.621	.012	70°
Uranopilite		1.624	.011	small to large
Topaz		1.631	.099	48°

Example No. 2.--In a crushed sample of a coarse igneous rock as seen under the stereomicroscope, fragments of the unknown are clear and faintly honey-brown in color. Mounted on the spindle stage they show the following optical properties:

α $1.682 \pm .001$
 β $1.691 \pm .001$
 γ $1.696 \pm .001$

$\gamma - \alpha$ $.014 \pm .002$
 $2V_x$ $70^\circ \pm 2^\circ$ measured
 $(73^\circ \pm 17^\circ$ calc)

Biaxial negative minerals of similar optical properties in table 7 within about 0.020 of this β -index form an extensive list as follows:

	Other entries	β	Birefringence	$2V_x$
Bustamite		1.675	.015	53°
Gerstmannite		1.675	.013	$50^\circ - 60^\circ$
Ferroaxinite		1.676	.011	69°
Bronzite, orthopyroxene		1.680	.011	79°
Tinzenite		1.681	.014	large
Manganaxinite		1.687	.014	75°
Lavenite		1.690	.016	68°
Chlorophoenicite		1.690	.015	83°
(UNKNOWN IN QUESTION)		$1.691 \pm .001$	$.014 \pm .002$	$70^\circ \pm 2^\circ$
Tinzenite		1.693	.010	med
Bronzite, orthopyroxene		1.695	.015	69°
Bustamite		1.695	.015	43°
Arfvedsonite		1.695	.015	69°
Kempite		1.695	.014	med
Ferroaxinite		1.695	.011	70°
Ferro-richterite		1.699	.016	35° ($68^\circ \pm 15^\circ$)
Tinzenite		1.701	.011	63°
Barylite		1.702	.013	70°
Bustamite		1.708	.015	34°

At the outset it would appear doubtful that the unknown could be bustamite, due to the fairly large discrepancy in optic angle. This doubt is confirmed on fragments of the unknown on the spindle stage. There they are seen to have two good cleavages, whereas bustamite has three cleavages, two good and one perfect. Members of the axinitite group (ferroaxinitite, manganesexinitite, and tinzenite) with layenite and gerstmannite are similarly eliminated, inasmuch as they have only one cleavage. Cleavage fragments of the unknown show parallel extinction and positive elongation, eliminating arfvedsonite and ferroricterite, which have inclined extinction. Although barylite fragments should also show two

good cleavages (001 and 100) and parallel extinction, the elongation would be negative, since $X=b$. A scratch test of fragments of the unknown between glass slides shows their hardness to be between 5 and 6, eliminating kempite and chlorophoenicite, having hardness 3.5, and confirming the elimination of gerstmannite, having hardness 4.5.

This leaves only orthopyroxene as a viable candidate; and, indeed, under the polarizing microscope the fragments show the pleochroism common in many orthopyroxenes: Z faint green, X faint reddish brown. An orthopyroxene with these refractive indices would have composition near En_{73} (Deer, Howie, and Zussman, v. 2A, 1978, p. 109).

TABLES 3–38

Table 3. Isotropic minerals and mineraloids

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.309		ICE H_2O	HEX	Conch	Col s	H 1.5 G 0.92	Melts at 0°C. Biref v low, pos.
1.326		VILLIAUMITE NaF	CUB u mass	001 perf	Carmine-red to col s	H 2-3.5 G 2.79 F easy	Sol in H_2O . Also uniax neg, biref v low. Pleoc., O carmine-red, E golden-yellow.
1.333		WATER H_2O	Liquid	---	Col s	G 1.00	---
1.339		CRYOLITHIONITE $Na_3LiAl_2F_{12}$	CUB	011 dist	Col s	H 2.5-3 G 2.77 F easy	Diss by acids.
1.340		HIERATITE K_2SiF_6	CUB cubo-oct	111 perf	Col s, gray	H ~ 2.5 G 2.67	Sol in hot water.
1.362		CAROBBIITE KF	CUB	001	Col s	G 2.50 (synth)	Sol in H_2O .
1.364		NEIGHBORITE $NaMgF_3$	ORTH ps cub	Uneven	Col s to brownish	H 4.5 G 3.03	Slowly diss by hot acids. Biref 0.002.
1.369		CRYPTOHALITE $(NH_4)_2SiF_6$	CUB cubo-oct	111 perf	Col s	H ~ 2.5 G 2.03 Volut	Sol in H_2O .
1.376		ELPASOLITE K_2NaAlF_6	CUB	Uneven	Col s	H 2.5 G 2.99 F 3	Slowly diss by concd H_2SO_4 .
1.38	v	Unnamed fluoride near Ralstonite (Ca,Na)(Mg,Al) ₂ (F,OH) ₆ • H_2O	CUB oct	---	Yellow	---	Diss by acids. Am. Mineral., 28, 283-284 (1943).
1.427	v	OPAL $SiO_2 \cdot xH_2O$	Amor	Conch	Col s, white, gray, green	H ~ 6 G 1.9-2.3	Insol in acids, diss by KOH soln.
1.434	v	MELANOPHLOGITE $SiO_2 + C, H, O, S$	TET ps cub	---	Col s, brownish	H 6.5-7 G 1.99 Infus	Insol in acids. Turns black when heated (organic matter). Biref wk.
1.467	v	RALSTONITE $Na_xMg_xAl_{2-x}(F,OH)_6 \cdot H_2O$	CUB	111 poor	Col s to yellowish	H 4.5 G 2.52-2.62 Infus	Dec by H_2SO_4 . Opt anom, divided into oct biref sectors.
1.38	v	FLUORITE CaF_2	CUB	111 perf	Col s, green, violet, yellow	H 4 G 3.18 F 1.5	Diss by acids.
1.440	v						

1.406 1.46	1.434 1.46	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols	H ~ 6 G 1.86	
	1.439	SODIUM ALUM $\text{NaAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Cols	H ~ 3 G 1.67	Sol in H_2O .
1.443	1.440	CHUKHROVITE $\text{Ca}_3(\text{Y}, \text{Ce})\text{Al}_2(\text{SO}_4)\text{F}_{13} \cdot 10\text{H}_2\text{O}$	CUB oct "Porcelain-like"	111 indist uneven	Cols, white	H 3 G 2.27-2.40 F easy	Diss by acids. Anom biref.
1.434 1.457	1.440	FLUORITE, yttrian (Ca, Y) F_{2-3}	CUB u mass	111	Cols, green, violet	H 4 G 3.41 F 2	Slowly diss by HCl . Contains 6.9% Y_{2}O_3 .
1.440 1.51	1.443	CHUKHROVITE-(Ce) $\text{Ca}_3(\text{Ce}, \text{Y})\text{Al}_2(\text{SO}_4)\text{F}_{13} \cdot 10\text{H}_2\text{O}$	CUB	--	--	--	--
1.445	1.44	HISINGERITE ($\text{Fe}^{+3}, \text{Mg}, \text{Fe}^{+2}$) $_{2-} \text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3.5 G 2.5 Infus	Dec by acids.
1.465	1.456	EVANSITE approx. $\text{Al}_3(\text{PO}_4, \text{Si}_4\text{O}_10)(\text{OH})_6 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols, yellowish	H 3-4 G 1.87-2.13	Diss by hot H_2SO_4 .
1.462	1.450	NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2\text{F} \cdot 19\text{H}_2\text{O}$	CUB	Conch	Cols to white	H 2.5 G 1.72	--
1.455	1.455	SULPHOHALITE $\text{Na}_6(\text{SO}_4)_2\text{FCl}$	CUB dod	Conch	Cols	H 3.5 G 2.50	Slowly sol in H_2O .
1.456	1.457	POTASSIUM ALUM $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Cols	H 2-2.5 G 1.76 F 1	Sol in H_2O .
1.440	1.458	FLUORITE, yttrian (Ca, Y) F_{2-3}	CUB	111 imperf	Yellowish	H 4.5 G 3.55 fus	Diss by acids.
1.434 1.46	1.46	TSCHERMIGITE $(\text{NH}_4)\text{Al}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	White	H 1.5 G 1.65 F 1	Sol in H_2O . Opt anom.
1.48	1.46	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols, white, gray, green	H ~ 6 G ~ 2.2 infus	Insol in acids, diss by KOH soln.
1.450	1.462	Glass, "Lechatelierite" mainly SiO_2	Amor	Conch	Cols	H 6 G 2.19 infus	Natural fused silica from fulgurite. SiO_2 99%.

1.406 1.46	1.434 1.46	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Conch	Cols	H ~ 6 G 1.86	Insol in acids, diss by KOH soln. Contains 12.6% H_2O .
	1.439	SODIUM ALUM $\text{NaAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	H ~ 3 G 1.67	Sol in H_2O .
1.443	1.440	CHUKHOVITE $\text{Ca}_3(\text{Y}, \text{Ge})\text{Al}_2(\text{SO}_4)\text{F}_{13} \cdot 10\text{H}_2\text{O}$	CUB oct "Porcelain-like"	111 indist uneven	Cols, white	H 3 G 2.27-2.40 F easy
1.434 1.457	1.440	FLUORITE, yttrian (Ca, Y) F_{2-3}	CUB u mass	111	Cols, green, violet	Slowly diss by HCl. Contains 6.9% Y_{2}O_3 .
1.440 1.443	1.44	CHUKHOVITE-(Ce) $\text{Ca}_3(\text{Ce}, \text{Y})\text{Al}_2(\text{SO}_4)\text{F}_{13} \cdot 10\text{H}_2\text{O}$	CUB	---	---	---
1.51	1.445	HISINGERITE ($\text{Fe}^{+3}, \text{Mg}, \text{Fe}^{+2}$) $\cdot 2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	Dec by acids.
1.465	1.450	EVANSITE approx. $\text{Al}_3(\text{PO}_4, \text{Si}_4\text{O}_10)(\text{OH})_6 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols, yellowish	Diss by hot H_2SO_4 .
1.462	1.455	NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2\text{F} \cdot 19\text{H}_2\text{O}$	CUB	Conch	Cols to white	H 3-4 G 1.87-2.13
1.456	1.456	SULFOHALITE $\text{Na}_6(\text{SO}_4)_2\text{FCl}$	CUB	Conch	Cols	H 2.5 G 1.72
1.440	1.457	POTASSIUM ALUM $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Cols	Slowly sol in H_2O .
1.458	1.46	FLUORITE, yttrian (Ca, Y) F_{2-3}	CUB	111 imperf	Yellowish	Diss by acids.
1.434 1.48	1.46	TSCHERMIGITE $(\text{NH}_4)\text{Al}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	White	H 4.5 G 3.55 fus
1.450	1.462	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	H 1.5 G 1.65 F 1	Sol in H_2O . Opt anom.
		Glass, "Lechatelierite" mainly SiO_2	Amor	Conch	Cols, white, gray, green	Insol in acids, diss by KOH soln.
		NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2\text{F} \cdot 19\text{H}_2\text{O}$	CUB	Conch	Cols	Natural fused silica from fulgurite. SiO_2 99%.

1.471 ^	1.48	FAUJASITE (Zeolite grp) (Na ₂ ,Ca)(Al ₂ Si ₄) ₁₂ ·8H ₂ O	CUB oct	111 dist	White	H 5 G 1.92 F 3	
	1.482	Unnamed calcium silicate	---	---	White, silky fibers	Soft	Dec by HCl. On losing H ₂ O, becomes uniax, pos, in 8 sectors.
v 1.487	1.483	SODALITE (Sodalite grp) Na ₈ Al ₆ Si ₆ ·24Cl ₂	CUB dod	110 poor	Gray, white, blue	H 6 G 2.30 F 3.5-4	An. Mineral., 26, 375 (1941).
	1.486	CRISTOBALITE SiO ₂	TET ps cub, oct	---	White	H 6-7 G 2.3 infus	Gel with acids. Cl 7.1, H ₂ O 2.5%.
1.483 ^	1.487	SODALITE, var Hackmanite (Sodalite grp) Na ₈ Al ₆ Si ₆ (O,S)·24Cl ₂	CUB dod	110 poor	Reddish-violet, blue	H 5 G 2.2-2.3 F 4	Insol in acids. Intricate tw. Biref very low.
v 1.489	1.489	D'ANSITE Na ₁₈ Mg(SO ₄) ₁₀ ·3NaCl	CUB	---	Col's, yellow	G 2.66 F easy	Gel with acids. Color fades in sunlight.
v 1.49	1.49	VISHNEVITE (Cancrinite grp) (Na,Ca) ₆₋₈ (Al ₆ Si ₆ O ₂₄) ₂ (SO ₄ ,CO ₃ ,Cl) ₂ ·1-2H ₂ O	HEX	10T0 perf	Col's	H 5 G 2.32 F 2	Sol in H ₂ O.
33 1.47 1.52	1.490	ALLOPHANE Hydrous aluminum silicate KCl	Amor	---	White, blue, green	H 3 G 1.86 infus	Gel with acids. Also uniax pos and neg. Biref .002.
1.493 ^ 1.505	1.493	SYLVITE KCl	CUB	perf	Col's, white, red, yellow	H 2 G 1.99 F 1.5	Sol in H ₂ O. Tastes bitter.
1.479 ^ 1.509	1.494	ANALCIMITE (Zeolite grp) NaAlSi ₂ O ₆ ·H ₂ O	CUB 211	100 poor	Col's, white	H 5 G 2.25 F 3.5	Gel with HCl.
	1.495	HAUYNITE (Sodalite grp) (Na,Ca)·8Al ₆ Si ₆ (O,S)·24 (SO ₄ ,Cl) ₂ ·1-2	CUB dod	110 poor	Blue, white	H 6 G 2.40 fus	Gel with acids. SO ₃ 9.8, Cl 1.3, Na ₂ O 19.9, CaO 4.9%.
1.470 ^	1.498	NOSEAN (Sodalite grp) Na ₈ Al ₆ Si ₆ ·24(SO ₄) ₅ (OH) ₂₆ ·20H ₂ O	CUB dod	110 poor	Blue, white	H 6 G 2.3-2.4 F 4.5	Diss by HNO ₃ . Blackens bb.
1.50		ZAHERITE Al ₁₂ (SO ₄) ₅ (OH) ₂₆ ·20H ₂ O	Mass, fine-grained	One clv	White	H 3.5 G 2.01 infus	Biref < .001.
		ROSIERESTE Hydrous phosphate of Pb, Cu, Al	Amor compact	---	Yellow, brown	G 2.2 infus	

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.48 1.53	1.50	Glass rhylitic to dacitic comp	Amor	Conch	Col s, gray	H 5-6 G ~ 2.4	Insol in acids.
[]	1.50	STEVENITE (Smectite grp) $Mg_3Si_4Al_{10}(OH)_2 \cdot xH_2O$	MCL mass	---	Buff, pink infus	H 2.5 G 2.15-2.57	Dec by HCl. Bioref 0 to .02.
1.465	1.50	EVANSITE $Al_3PO_4(OH)_6 \cdot 6H_2O$	Amor	Conch	Col s, brown, yellow	H 3-4 G 1.9-2.1	Diss by hot H_2SO_4 .
1.522	1.500	LAZURITE (Sodalite grp) $(Na,Ca)_7 \cdot 8(Al,Si)_{12}(O,S)_{24}$ [$SO_4, Cl_2, (OH)_2$]	CUB dod	110 poor	Azure blue	H 6 G 2.4 F 3.5	Gel with acids, gives off H_2S .
1.51	1.50	BOLIVARITE $Al_2(Po_4)(OH)_3 \cdot 4 \cdot 5H_2O$	Amor	Splintery to conch	Greenish-yellow to green	H 2.5 G 2.05	---
	1.502	NASTROPHITE $Na(Sn,Ba)Po_4 \cdot 9H_2O$	CUB	Conch	Col s	H 2 G 2.05	Dec by H_2O , diss by HCl.
	1.504	NABAPHITE $NaBaPo_4 \cdot 9H_2O$	CUB	100 dist	Col s	H ~ 2 G 2.3	Dec by H_2O , diss by acids.
1.48 1.505	1.505	VASEGYITE $Al_4(Po_4)_3(OH)_3 \cdot 13H_2O$ (?)	CUB	ORTH (?) cryptocryst	White, green, yellowish	H 3.5 G 1.98 infus	Diss by acids.
1.493 1.507	1.505	ANALCIME, cesian (Zeolite grp) $(Na,Cs)AlSi_2O_6 \cdot H_2O$	CUB	100 poor	Col s, white	H 5 G 2.60 F 4.5	Diff dec by acids. Cs 20 14.9%.
1.505 1.525	1.507	POLLUCITE (Zeolite grp) $(Cs,Na)AlSi_2O_6 \cdot H_2O$	CUB u mass	Uneven	Col s, white	H 6.5 G 2.85 F diff	Dec by acids.
1.550	1.508	TYCHITE $Na_6Mg_2(CO_3)_4(SO_4)$	CUB oct	Conch	Col s	H 3.5-4 G 2.46-2.59 F 1	Slightly sol in H_2O , diss by acids with eff. Compare Northupite.
1.509		LEUCITE $KAlSi_2O_6$	TET ps cub, 211	110 poor	Col s	H 5.5 G 2.47 infus	Dec by acids. In part cryptocryst.

1.494 Λ	1.509	HAUNYE (Sodalite grp) (Na,Ca)4-8Al ₆ Si ₆ (O,S) ₂₄	CUB dodecagonal	110 poor	Blue, white	H 6 G 2.51 fus	Gel with acids. Na ₂ O 13.2, K ₂ O 2.8, CaO 10.1%.
1.44 Δ 1.57	1.51	HISINGERITE Fe ⁺³ Si ₂ Si ₅ (OH) ₄ ·2H ₂ O	MCL cryptocryst	Conch	Brownish-black	H 3 G 2.5-3.0 infus	Dec by acids. In part cryptocryst. ---
1.51	1.51	MOTUKOREAITE Na ₂ Mg ₃₈ Al ₂₄ (CO ₃) ₁₃ (SO ₄) ₈ (OH) ₁₀₈ ·56H ₂ O	HEX cryptocryst	---	White to pale yellowish-green	H 1-1.5 G 1.48-1.53	---
1.513	1.513	CADVALADERITE Al(OH) ₂ Cl·4H ₂ O	Amor (?)	Conch	Lemon-yellow	G 1.6	---
1.550 Δ	1.514	NORTHUPITE Na ₆ Mg ₂ (CO ₃) ₄ Cl ₂	CUB octahedron	Conch	Colts	H 3.5-4 G 2.38 F 1	Slightly sol in H ₂ O, diss by acids with eff. Tw common. Compare Tychite.
1.517 Δ	1.517	PLANERITE (Ca,Cu)Al ₆ (PO ₄) ₄ (OH) ₈ ·4H ₂ O (?)	Cryptocryst	---	Green	H 5 G 2.65	Slightly sol in acids. In part birefr.
1.517 Δ	1.517	SEPiolite, var Meerschaum Mg ₄ Si ₆ O ₁₅ (OH) ₂ ·6H ₂ O	ORTHO cryptocryst	---	White	H 1-2 G 2.0	Biref 0 to 0.02.
1.49 Δ	1.52	ALLOPHANE Hydrous aluminum silicate	Amor	---	White, brown	H 3 G 1.9 infus	Gel with acids. Fe ₂ O ₃ 5-6%.
1.500	1.522	LAZORITE (Sodalite grp) (Na,Ca) ₇₋₈ (Al,Si) ₁₂ (O,S) ₂₄ [SiO ₄ .Cl] ₂ ·(OH) ₂	CUB dodecagonal	110 poor	Azure blue	H 6 G 2.4 F 3.5	Gel with acids, gives off H ₂ S.
1.507	1.525	POLLUCITE (Zoelite grp) (Cs,Na)AlSi ₂ O ₆ ·H ₂ O	CUB unmassive	Uneven	Colts, white	H 6.5 G 2.90 F diff	Dec by acids.
1.50 Δ 1.57	1.53	Glass andesitic comp	Amor	Conch	Colts, gray	H 5-6 G ~ 2.5	Insol in acids.
1.53	1.53	KEHODEITE (Zn,Ca)Al ₂ (PO ₄) ₂ (OH) ₂ ·5H ₂ O	Amor (?) mass	---	White	H 2-34 infus	Diss by acids. Status of mineral in doubt.
1.530	1.530	VISEITE NaCa ₅ Al ₁₀ (SiO ₄) ₃ (PO ₄) ₅ (OH) ₁₄ ·10H ₂ O (?)	CUB massive	---	White, bluish, yellowish	H 3-4 G 2.2 F easy	---
1.534 Δ	1.572	LANGBEINITE K ₂ Mg ₂ (SO ₄) ₃	CUB tetrahedron	Conch	Colts, yellow	H 3.5-4 G 2.83 F 2	Slowly sol in H ₂ O.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.530 1.54	1.535	SUCCINITE (Amber) a hydrocarbon	Amor	Conch	Amber-yellow	Soft 6 1.07 F easy	---
1.537	1.537	TACHARANITE $\text{Ca}_{12}(\text{Mg},\text{Al})_2\text{Si}_{18}\text{O}_{51} \cdot 18\text{H}_2\text{O}$ (?)	MCL mass	001 good	White	G 2.36	Biref low.
1.54	1.54	NEOTOCITE $\text{MnO}_2 \cdot \text{SiO}_2 \cdot \text{H}_2\text{O}$	Amor	Conch	Brown to black	H 4 G 2.7 F diff	Dec by acids.
1.47 1.47	1.54	Hydrated basaltic glass ("Chlorophaeite")	Amor	Conch	Yellow, brown, orange-brown	H 1.5-3 G 1.8-2.2	Color may darken on exposure.
1.541	1.541	AJKAITITE a resin	Amor	----	Pale yellow to red-brown	H 2.5 G 1.0 F easy and burns	Gives off H_2S when heated. C 80, H 10, 20 9, S 1%.
1.542	1.542	TELEGDITE a resin	Amor	----	----	H 2.5 G 1.09	Partly diss by alcohol. H 10.2, 0 11.2, S 1.7%.
1.555 1.555	1.542	HALLOYSITE $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	MCL u crypto-cryst	----	White, blue	H 2 G 2.6 infus	Insol in acids. When dried at 60°C, $\underline{\text{H}}$ increases to 1.555.
1.544	1.544	HALITE NaCl	CUB	perf	Cols, yellow, red	H 2 G 2.17 F 1.5	Sol in H_2O , salty taste.
1.514	1.550	NORTHUPITE $\text{Na}_6(\text{Mg},\text{Fe})_2(\text{CO}_3)_4\text{Cl}_2$	CUB oct	Conch	Bright green	H 3.5-4 G 2.52 F 1	Slightly sol in H_2O , diss by acids with eff. FeO up to 7.6%.
1.508	1.550	FERROTYCHITE $\text{Na}_6(\text{Fe}^{+2},\text{Mg},\text{Mn})(\text{SO}_4)(\text{CO}_3)_3$	CUB	Conch	Cols to light yellow	H 4 G 2.79	---
1.542 1.570	1.555	HALLOYSITE $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	MCL u crypto-cryst	----	White, blue	H 2 G 2.6 infus	Insol in acids.
1.558	1.558	SERPENTINE $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	MCL or ORTH crypto-cryst	----	Green	H 4 G 2.5 F 6	Dec by acids. Biref low. Contains NiO 2.6%. Species not detd.

1.59	ZARATITE $\text{Ni}_3(\text{CO}_3)(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	CUB	Conch	H 3-3.5 G 2.57-2.69 Infus	Diss by acids with eff.
1.560	GUTSEVICHITE $(\text{Al},\text{Fe})_3(\text{PO}_4)_2(\text{OH})_3 \cdot 8\text{H}_2\text{O}$ (?)		Crusts, concretions	--	Diss by acids.
1.606	Ni analogue of Serpentine ("Garnierite") $(\text{Ni},\text{Mg})_3\text{Si}_2\text{O}_5(\text{OH})_4$		MCL (?) cryptocryst	--	Dec by acids. (= Pecorite or Nepouite?)
1.59	ZUNYITE $\text{Al}_{13}\text{Si}_5\text{O}_{20}(\text{OH},\text{F})_{18}\text{Cl}$	CUB tetrah	111	Col s	Insol in acids. F 12.5, Cl 2.2, H_2O 6.5%.
1.592					--
1.563					
1.565	ALUMOPHARMACOSIDERITE $\text{KAl}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6.5\text{H}_2\text{O}$	CUB	--	White	G 2.68
1.676	CARBONATE-FLUORAPATITE ("Colliphane" (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3(\text{CO}_3)_3(\text{F},\text{OH})$)		HEX cryptocryst	--	White, brown H 3.5 G ~2.6 F diff
1.59					Diss by acids.
1.569	HISINGERITE $\text{Fe}^{+3}2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL microcrys	Conch	Brownish-black	H 3.5 G 2.5-3.0 Infus
1.51					Dec by acids.
1.59					
1.53	Glass andesitic to latitic comp	Amor	Conch	Pale brown, yellowish	H 5 G ~ 2.06
1.61					Insol in acids.
1.558	SERPENTINE, zincian $(\text{Mg},\text{Zn})_3\text{Si}_2\text{O}_5(\text{OH})_4$		MCL or ORTH cryptocryst	--	Diss by acids. ZnO 7.25%.
1.570					
1.57	BORICKITE Hydrous Ca ferric phosphate	Amor	--	Reddish-brown	H 3.5 G ~ 2.7 F easy
1.60					Diss by acids.
1.571	NITROBARITE $\text{Ba}(\text{NO}_3)_2$	CUB oct	--	Col s	Sol in H_2O .
1.572	MANGANOLANGBEINITE $\text{K}_2\text{Mn}_2(\text{SO}_4)_3$	CUB tetrah	--	Rose-red	G 3.02
1.534					Sol in H_2O .
1.573	MILLOSEVICHITE $(\text{Al},\text{Fe})_2(\text{SO}_4)_3$	--	--	Cherry-red	Diss by acids. Fe_2O_3 8.8%.
1.577	NAMUWITE $(\text{Zn},\text{Cu})_4(\text{SO}_4)(\text{OH})_6 \cdot 4\text{H}_2\text{O}$	HEX	0001 perf	Pale sea-green	ZnO 37.8, CuO 22%.
1.58	KONINCKITE $\text{FePO}_4 \cdot 3\text{H}_2\text{O}$	TET spherulitic	--	Yellow	Diss by hot acids.
1.65					

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[]	1.582	CHINGLUSUITE $\text{Na}_4\text{Mn}_5\text{Ti}_3\text{Si}_{14}\text{O}_{41} \cdot 9\text{H}_2\text{O}$	Amor	Uneven	Black	H 2-3 G 2.15 F easy	Gel with acids.
1.57 △ 1.66	1.59	HISINGERITE $\text{Fe}^{+3} \text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3 G 3.0 infus	Dec by acids.
1.56 △ 1.61	1.59	ZARATITE $\text{Ni}_3(\text{CO}_3)(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	CUB	Conch	Emerald-green	H 3-3.5 G 2.6 infus	Diss by acids with eff.
1.569 △ 1.63	1.59	CARBONATE-FLUORAPATITE ("Collophane") (Apatite grp) $\text{Ca}_5(\text{PO}_4, \text{CO}_3)_3(\text{F}, \text{OH})$	HEX cryptocryst	---	White, brown	H 3.5 G 2.6-2.7 F diff	Diss by acids.
1.563 △ 1.667	1.59	Ni analogue of Serpentine ("Garnierite") $(\text{Ni}_1, \text{Mg})_3\text{Si}_2\text{O}_5(\text{OH})_4$	MCL (?) cryptocryst	---	Green	H 2 G ~ 2.6 infus	Dec by acids. (= Pecorite or Nepouite?)
1.563 △ 1.600	1.59	SCHOENFLIESITE $\text{MgSn}(\text{OH})_6$	CUB mass	---	Brown	H 3.48 G 2.89 infus	Diss by HCl, slowly by NaOH soln.
1.567 △ 1.608	1.592	ZUNYITE $\text{Al}_{13}\text{Si}_5\text{O}_{20}(\text{OH}, \text{F})_{18}\text{Cl}$	CUB tetrah	111	Col s	H 7 G 2.89 infus	Insol in acids. F 5.5, Cl 2.6. H_2O 10.0%.
1.593 △ 1.595	1.593	VOLTITE $\text{K}_2\text{Fe}^{+2} \text{Fe}^{+3}_5(\text{SO}_4)_4 \cdot 12 \cdot 18\text{H}_2\text{O}$	CUB	Conch	Dull oil-green, brown, black	H 3 G 2.6-2.8	Diss by acids. Oil-green in thin section.
1.593 △ 1.595	1.593	GEORGITE $\text{Cu}_5(\text{CO}_3)_3(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	Amor	---	Light blue	Soft G 2.55	---
1.595 △ 1.595	1.595	COESITE SiO_2	MCL ps hex	Subconch	Col s	G 2.92 infus	Insol in acids. Biref .003 - .005.
1.64 []	1.596	VUDYAVRITE Hydrous silicate of Ce, La, Ti	Amor	Conch	Brown, cream, yellow	H ~ 3 G 2.40-2.52	Dec by acids.
1.598	1.598	STURITE $(\text{Mn}, \text{Mg})_6\text{Fe}^{+3} \cdot 2\text{Si}_8\text{O}_{25} \cdot 23\text{H}_2\text{O}$ (?)	Amor	Conch	Brownish-black	H 3 G 2.05 F diff	Dec by acids.
		COMBEITE $\text{Na}_2\text{Ca}_2\text{Si}_3\text{O}_9$	TRIG	---	Col s	G 2.84	Diss by hot dilute HCl. Biref very low.

1.57 1.67	1.60 Hydrous Ca ferric phosphate	Amor	---	Reddish-brown	H 3.5 G ~ 2.7 F easy	Diss by acids.
1.60 1.70	STIBICONITE (Stibiconite grp) Sb ⁺³ Sb ⁺⁵ O ₆ (0,0H)	CUB u mass	Conch	Gray, white, yellow	H 4-6 G 5.1-5.6 infus	Insol in acids. ± highly variable.
1.592 1.600	ZUNYITE Al ₁₃ Si ₅ O ₂₀ (OH,F) ₁₈ Cl	CUB tetrah	111	Cols	H 7 G 2.88 infus	Insol in acids. F 0.5, Cl 3.5, H ₂ O 11.4%.
[]	EUDIALYTE Na ₄ (Ca,Ce) ₂ (Fe ⁺² ,Mn ⁺²)ZrSi ₈ O ₂₂ (OH,Cl) ₂	TRIG	0001 dist	Yellow, pink, brown	H 5-5.5 G 2.9-3.1 F 3	Gel with acids. Uniax, pos or neg.
1.560 1.606	GUTSEVICHITE (Al,Fe) ₃ (PO ₄) ₂ (OH) ₃ •8H ₂ O (?)	Crusts, concretions	---	Dark green to brown	H 2.5 G 1.9-2.0	Diss by acids.
[]	CALCIUM FERRI-PHOSPHATE CaFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ •9H ₂ O (?)	Amor	---	Light brown	H 2-3	Diss by acids.
1.593 1.608	VOLTAINITE K ₂ Fe ⁺² ₅ Fe ⁺³ ₄ (SO ₄) ₁₂ •18H ₂ O	CUB	Conch	Dull oil- green, brown, black	H 3 G 2.6-2.8	Diss by acids. thin section.
1.57 1.65	1.61 Glass basaltic comp	Amor	Conch	Brown	H 5 G 2.8	----
1.59 1.61	ZARATITE Ni ₃ (CO ₃) ₂ (OH) ₄ •4H ₂ O	CUB	Conch	Emerald-green	H 3 G ~ 2.6 infus	Diss by acids.
[]	DIADODCHITE Fe ₂ (PO ₄) ₂ (OH) ₄ •5H ₂ O	TCL u cryptocryst	Uneven to conch	Brown, yellow	H 3-4 G 2.0-2.4 F easy	Diss by HCl.
1.61 1.68	ALLANITE (Epidote grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	MCL metamict	---	Brown, black	H 6 G 2.96 F 3	Gel with acids. In part biref.
1.613	HSIANGHUALITE Ca ₃ Li ₂ Be ₃ (SiO ₄) ₃ F ₂	CUB dode	---	White	H 6.5 G 2.98	----
1.614 1.643	MAYENITE Ca ₁₂ Al ₁₄ O ₃₃	CUB mass	---	White	H 2.85 infus	----
1.616 1.635	PITTICITE Hydrous ferric arsenate-sulfate	Amor	---	Brown, yellow	H 2-3 G 2.2-2.5 F easy	Diss by HCl. Comp widely variable.
1.618	METABORITE HBO ₂	CUB tetrah	Conch	Cols to brownish	H 5 G 2.47 F 1	Slowly sol in H ₂ O.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[]	1.62	ZIRRESITE $Zr_3Fe_2Si_4O_{17} \cdot 15H_2O$ (?)	Amor	---	Pale yellow	G 2.70	Gel with acids.
	1.625	BICCHULITE $Ca_2Al_2Si_6(OH)_2$	CUB	---	Col s	G 2.75 (synth)	---
	1.629	KAMISHILITE $Ca_2Al_2Si_6(OH)_2$	TET	---	Col s	---	---
1.59 △	1.63	CARBONATE-FLUORAPATITE ("Colliphane") (Apatite grp) $Ca_5(PO_4, CO_3)_3(F, OH)$	HEX cryptocryst	---	White, brown	H 3.5 G 2.7 F diff	Diss by acids.
1.66 γ	1.63	GRIPHITE $Na_4Ca_6(Mn, Fe^{+2}, Mg)_{19}Li_2Al_8$ $(PO_4)_{24}(F, OH)_8$	CUB u mass	Conch	Brown	H 5.5 G 3.4 F easy	Diss by HCl.
1.685 γ	1.63	TRITOMITE-(Y) ("Spencite") $(Y, Ca, La, Fe^{+2})_5(Si, B, Al)_3$ $(O, OH, F)_{13}$ (?)	TRIG (?) metamict	---	Reddish-brown	H 3.5 G 3.1 F 3	---
	1.633	BURTTITE $CaSr(OH)_6$	CUB	001 good	Col s	H ~ 3 G 3.28	Diss by HCl.
1.616 △	1.635	PITTICITE Hydrous ferric arsenate-sulfate	Amor		Brown, yellow	H 2-3 G 2.2-2.5 F easy	Comp widely variable.
	1.639	SALAMMONIAC NH_4Cl	CUB	111 imperf	Col s	H 1.5-2 G 1.53 F 1 volat	Sol in H_2O .
1.653 γ	1.639	HARERITE $Ca_{24}Mg_8Al_2(SiO_4)_8(BO_3)_6(CO_3)_{10} \cdot 2H_2O$	CUB oct	---	Col s	H 4 G 2.95	Diss by HCl. Anom biref.
	1.639	TOMBARTHITE $Y_4(Si, H_4)_4(OH)_{12-x}(OH)_{4+2x}$	MCL metamict	Conch	Brown, black	H 5-6 G 3.51-3.65	---
1.63- 1.66	-	THOROSTEENSTRUPINE $(Ca, Th, Mn)_3Si_4O_{12} \cdot 6H_2O$	Amor metamict	Conch	Dark brown	H 4 G 3.02	ThO_2 35.7%. Weakly magnetic.
1.64 γ	1.685	THORITE $ThSiO_4 (+ xH_2O)$	TET metamict	---	Brown to yellow	H 2-4 G 3.8-4.3	Gel with warm acids.

1.64	$\frac{v}{x}$	PICRITE Hydrous ferric phosphate	Amor	---	Dark brown	H 3-4 G 2.8 F easy	Diss by acids.	
1.64	[]	HOMMELITE (altered) $\text{Ca}_2(\text{Fe}^{+2}, \text{Mg})_2\text{Si}_2\text{O}_{10}$	MCL mass	Conch	Black	H 3 G 3.35	Gel with acids.	
1.64	$\frac{v}{x}$	UMBROZERITE $\text{Na}_3\text{Sr}_4\text{ThSi}_8(0, \text{OH})_{24}$	Amor metamict	Conch	Bottle-green to greenish-brown	H ~ 5 G 3.60	---	
1.64	$\frac{v}{x}$	COFFINITE $\text{U}(\text{Si}_4)_1-x(\text{OH})_{4x}$	TET metamict	---	Yellow, brown	G 3.6-3.9	---	
1.74	$\frac{v}{x}$	VUDYAVRITTE Hydrous silicate of Ce, La, Ti	Amor	Conch	Brown, cream, yellow	H ~ 3 G 2.40-2.52	Dec by acids.	
1.595	$\frac{v}{x}$	SAKHAITE $\text{Ca}_3\text{Mg}(\text{BO}_3)_2\text{CO}_3 \cdot x\text{H}_2\text{O}$ ($x < 1$)	CUB	---	Gray, white	H 5 G 2.78-2.84	Diss by HCl with eff.	
1.614	$\frac{v}{x}$	MAVENITE $\text{Ca}_{12}\text{Al}_{14}\text{O}_{33}$	CUB mass	---	Colts	G 2.85 Infus	Insol in dil acids.	
1.58	$\frac{v}{x}$	KONINCKITE $\text{FePO}_4 \cdot 3\text{H}_2\text{O}$	TET spherulitic	---	Yellow	H 3.5 G 2.3 F 3	Diss by hot acids.	
1.65	[]	EQUELITE Hydrous Ca ferric phosphate	Amor	---	Yellow-brown	G 2.60 F easy	Diss by acids.	
1.65	[]	GREENALITE $(\text{Fe}^{+2}, \text{Fe}^{+3})_{2-3}\text{Si}_2\text{O}_5(\text{OH})_4$	MCL microcryst	---	Green to brown	G 2.8-3.0	Dec by HCl.	
1.65	$\frac{v}{x}$	Glass basaltic comp	Amor	Conch	Brown	H ~ 5 G ~ 3	---	
1.65	$\frac{v}{x}$	MELANOCERITE $(\text{Ca}, \text{Ce})_5(\text{Si}, \text{B})_3\text{O}_{12}(\text{OH}, \text{F}) \cdot x\text{H}_2\text{O}$ (?)	HEX metamict	Conch	Dark brown to black	H 5-6 G 4.1 Infus	Dec by HCl. Apatite structural type.	
1.648	$\frac{v}{x}$	MELLILITE (Mellilite grp) $(\text{Ca}, \text{Na})_2(\text{Mg}, \text{Al})(\text{Si}, \text{Al})_2\text{O}_7$	TET	001 dist	Colts to brown	H 5 G 3.0 F 3	Gel with acids. Comp close to Gehlenite 50, Akermanite 50. Biref .001.	
1.653	$\frac{v}{x}$	HARKERITE $\text{Ca}_2\text{Mg}_8\text{Al}_2(\text{Si}_4\text{O}_4)_8(\text{BO}_3)_6(\text{CO}_3)_{10} \cdot 2\text{H}_2\text{O}$	CUB oct	---	Colts	H 4 G 2.95	Diss by HCl. Anom biref.	
1.655	$\frac{v}{x}$	ZIRCON $\text{ZrSiO}_4 (+ x\text{H}_2\text{O})$	TET metamict	---	Yellow, brown	G 2.9 infus	Partly diss by acids.	

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.63 △	1.66	GRIPHITE $\text{Na}_4\text{Ca}_6(\text{Mn},\text{Fe},\text{Mg})_{19}\text{Li}_2\text{Al}_8(\text{PO}_4)_{24}(\text{F},\text{OH})_8$	CUB u mass	Conch	Brown	H 5.5 G 3.4 F easy	Diss by HCl.
1.59 △	1.66	HISINGERITE $\text{Fe}^{+3}\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3 G 3.0 F diff to infus	Dec by acids.
□	1.66	VESUVIANITE $\text{Ca}_{10}(\text{Mg},\text{Fe})_2\text{Al}_4(\text{Si}_2\text{O}_7)_2(\text{Si}_4\text{O}_4)_5(\text{OH},\text{F})_4$	TET metamict	---	Brown	H 6 G 3.3 F 3	Insol in acids.
1.59 △ 1.705	1.667	SCHOENFLIESITE $(\text{Mg},\text{Mn})\text{Sn}(\text{OH})_6$	CUB mass	---	Brown	G 3.49 infus	Diss by HCl.
1.60 △	1.67	BORICKITE Hydrous Ca ferric phosphate	Amor	---	Reddish-brown	H 3.5 G ~ 2.7 F 3-4	Diss by acids.
1.702 ▽	1.670	HYDROGLOSSULAR (Garnet grp) $\text{Ca}_3\text{Al}_2(\text{Si}_4\text{O}_8)_{3-x}(\text{OH})_{4x}$	CUB ddd	---	Col s	H 6-6.5 G 3.1 infus	Gel with acids. Some samples divided into biref sectors. H_2O 8-9%.
1.65 △	1.675	GREENALITE $(\text{Fe}^{+2},\text{Fe}^{+3})_2\text{Si}_2\text{O}_5(\text{OH})_4$	MCL microcryst	---	Green to brown	G 3.0	Dec by HCl.
1.565 △ 1.704	1.676	PHARMACOSIDERITE $\text{KFe}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	CUB cubes, tetrah	100 imperf	Brown, green	H 2.5-3 F 1 G 2.8-3.0	Diss by HCl. Biref 0 to 0.01.
1.64 △	1.68	PICITE Hydrous ferric phosphate	Amor	---	Dark brown	H 3-4 G 2.8 F easy	Diss by acids.
1.67 △ 1.71	1.680	HUTTONITE $\text{ThSiO}_4 (+ x\text{H}_2\text{O})$	MCL metamict	---	Col s to yellow	---	---
1.73	1.68	DELVAUXITE $\text{CaFe}^{+3}_3(\text{PO}_4,\text{SO}_4)_2(\text{OH})_8 \cdot 4-6\text{H}_2\text{O}$ (?)	Amor (?)	---	Brown	H 2.5-4 G 2.0-2.8	Diss by acids.

1.61 1.72	1.68	ALLANITE (Epidote grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	MCL metamict	---	Brown, black	H 6 G 3.4-3.7 F 3	In part biref. May gel with acids.
1.68	1.68	SATPAEVITE Al ₁₂ V ⁺⁴ ₂ V ⁺⁵ ₆ O ₃₇ ·30H ₂ O	ORTH (?)	One pina- coidal, perf	Canary-to saffron-yellow	H 1.5 G 2.4	Diss by acids. In section greenish-yellow to olive. U biaxial, pos.
1.63 1.763	1.685	TRITOMITE-(Y) ("Spencite") (Y,Ca,La,Fe ⁺²) ₅ (Si,B,Al) ₃ (O,OH,F) ₁₃	TRIG metamict	---	Reddish-brown	H 3.5 G 3.3 F 3	---
1.64 1.715	1.685	THORITE ThSiO ₄ (+ xH ₂ O)	TET metamict	---	Brown, yellow	H 2-4 G 4.4	Gel with acids. H ₂ O 9%.
1.694 1.70	1.694	RHODIZITE (Cs,K)Al ₄ Be ₄ B ₁₁ O ₂₅ (OH) ₄	CUB dod	111, 111 diff	Col s	H 8 G 3.31-3.44	Insol in acids. Opt anom.
1.78	1.70	GADDOLINITE (Y,Ca) ₂ FeBe ₂ Si ₂ 10	MCL metamict	Conch	Greenish-black	H 7 G 4.0 infus	Gel with acids. Pale green in section.
1.60 1./4	1.70	STIBICONITE (Stibiconite grp) Sb ⁺³ Sb ⁺⁵ ₂ 6(0,OH)	CUB u mass	Conch	Gray, white, yellow	H 4-6 G 5.1-5.6 infus	Insol in acids. n highly variable.
1.670 1.734	1.702	HYDROGROSSULAR (Garnet grp) Ca ₃ Al ₂ (SiO ₄) _{3-x} (OH) _{4x}	CUB dod	---	Col s	H 6-6.5 G 3.35 infus	Diff sol in acids. May gel.
1.676 1.712	1.704	PHARMACOSIDERITE KFe ₄ (AsO ₄) ₃ (OH) ₄ ·6H ₂ O	CUB cubes, tetrah	100 imperf	Brown, green	H 2.5-3 G 2.8-3.0 F 1	Diss by acids. Biref 0 to 0.01.
1.726	1.704	ROWLANDITE Y ₃ (SiO ₄) ₂ (OH,F) (?)	Amor metamict	Conch	Green, red	H 6-7 G 4.4-4.5 infus	Gel with acids.
1.667 1.748	1.705	WICKMANITE MnSn(OH) ₆	CUB oct	001 good	Brownish-to honey-yellow	H 3.5 G 3.89	---
1.655 1.79	1.71	BERZELIITE (Ca,Na) ₃ (Mg,Mn) ₂ (AsO ₄) ₃	CUB	Subconch to uneven	Yellow, orange	H 5 G 4.08 F 3	Diss by acids. MnO 1.3%. Garnet-type structure.
1.704	1.712	ZIRCON ZrSiO ₄ (+ xH ₂ O)	TET metamict	---	Yellow, brown	G 3.0-3.5 infus	Nearly insol in acids.
		BARIUM-PHARMACOSIDERITE Ba(Fe,Al) ₄ (AsO ₄) ₃ (OH) ₅ ·5H ₂ O (?)	TET (?)	100 good	Yellow-brown	H 2-3 G 3.00	Diss by acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
v 1.741	1.714	PYROPE (Garnet grp) $Mg_3Al_2(SiO_4)_3$	CUB 110, 211	---	Col s	H 7 G 3.58 F 4	Insol in acids. Data for synth end-member.
v 1.738	1.714	SPINEL (Spinel grp) $MgAl_2O_4$	CUB oct	111 imperf	Col s, red, green	H 8 G 3.58 infus	Insol in acids. Data for synth end-member.
v 1.685 1.804	1.715	THORITE (Th,U)SiO ₄ (+ xH ₂ O)	TET metamicrt	---	Brown, green, yellow	H 4 G 3.8-4.4	Gel with acids.
v 1.68 1.75	1.72	ALLANITE (Epidotite grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	MCL metamicrt	---	Brown, black	H 6 G 3.4-3.7 F 3	In part birefr. May gel with acids.
v 1.75	1.72	BRITHOLITE (Ce,Ca) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX metamicrt	---	Yellow, brown	H 5 G 3.85 infus	Gel with HCl. Yellow in section. Apatite structural type.
v 1.79	1.72	BINDHEIMITE (Stibiconite grp) $Pb_2Sb_2O_6(O,OH)$	CUB u mass	Conch	Yellow, brown, gray	H 4-4.5 G 4.6-5.6 F 3-4	Dec by HCl.
v 1.725	1.72	CALCIOURANOITE (Ca,Ba,Pb)U ₂ O ₇ •5H ₂ O	Amor	---	Brown to orange	H 4 G 4.62	Weakly anisotropic.
v 1.726	1.725	DZHALINDITE In(OH) ₃	CUB	---	Yellow-brown	G 4.34 infus	Pale yellow in section.
v 1.704	1.726	ROWLANDITE $Y_3(SiO_4)_2(F,OH)$ (?)	Amor metamict	Conch	Green, red	H 6-7 G 4.4-4.5 infus	Gel with acids.
v 1.68	1.73	DELVAUXITE $CaFe^{+3}_3(Po_4,So_4)_2(OH)_8$ 4-6H ₂ O (?)	Amor (?)	---	Brown	H 2.5-4 G 2.0-2.8	Diss by acids.
v 1.738	1.732	HELVITE (Helvite grp) $(Mn,Fe,Zn)_4Be_3(SiO_4)_3S$	CUB tetrab	111 imperf	Yellow	H 6 G 3.22 F 3	Gel with acids. MnO 45.5, FeO 2.2, ZnO 1.2%.
v 1.702 1.752	1.734	GROSSULAR (Garnet grp) $Ca_3Al_2(SiO_4)_3$	CUB 110, 211	Uneven	White, green, pink, brown	H 6-7 G 3.59 F 3	Insol in acids. Opt anom. Data for synth end member.

ν	1.735	PERICLASE MgO	CUB 100, 111	100 perf; Parting 110	Col s	H 5.5 G 3.56 infus	Diss by acids.		
	1.735	VISMARNOVITE $ZnSn(OH)_6$	CUB	---	Pale yellow	H 4 G 4.04	Diss by HCl.		
	1.736	SÖHNGEITE $Ga(OH)_3$	CUB	---	Light brown	H 4-4.5 G 3.84 infus	---		
	1.738	HELVITE (Helvite grp) (Mn, Fe, Zn) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	111 imperf	Brown	H 6 G 3.4 F 3	Gel with acids. MnO 29.9, FeO 10.1, ZnO 5.2%.		
	1.744	SPINEL (Spinel grp) (Mg, Fe) Al_2O_4	CUB oct	111 imperf	Green, black	H 8 G 3.62 infus	Insol in acids. FeO 5.6, Fe ₂ O ₃ 1.5, MnO 0.1%.		
	1.714	COFFINITE $U(SiO_4)_{1-x}(OH)_{4x}$	TET metamict	---	Yellow, brown	G 3.6-3.9	---		
	1.747	CARYOCERITE Borosilicate of Y, Ce, Ca, Th	ps trig, metamict	Conch	Brown	H 5-6 G 4.30	Dec by hot HCl. Swells without fusing hb. Compare Melano- cerite.		
	1.64	THORGUMMITTE (Th,U)(Si, P) _{4-x} (OH) _{4x}	TET metamict	---	Brown to black	H 3 G ~ 4.6 infus	Gel with acids. Compare Thorite.		
	1.92	STIBICONITE (Stibiconite grp) $Sb^{+3}Sb^{+5}Al_2O_6(O, OH)$	CUB u mass	Conch	Gray, brown, yellow	H 4-5.5 G 5.2-5.6 infus	Insol in acids.		
	1.7	PYROPE-ALMANDINE (Garnet grp) (Mg, Fe) ₃ $Al_2(SiO_4)$ ₃	CUB 110, 211	---	Red, brown	H 7 G 3.72 F 3.5	Insol in acids. Py 66, Al 20, Sp 1, Gr 11, An 2.		
	1.70	GENTHELVITE (Helvite grp) (Zn, Fe) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	---	Brown, red	H 6 G 3.66 F 4	Gel with acids. MnO 1.5%.		
	1.80	DANALITE (Helvite grp) (Fe, Zn, Mn) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	---	Pink, brown	H 6 G 3.44 F 3.5	Gel with acids. ZnO 46.0, FeO 20.0, MnO 8.15%.		
	1.714	SPINEL, zincian (Spinel grp) (Mg, Fe, Zn) Al_2O_4	CUB oct	---	Greenish-black	H 8 G 3.97 infus	Insol in acids. ZnO 18.2, MgO 16.8, FeO 1.9%.		
	1.744	HELVITE (Helvite grp) (Mn, Fe, Zn) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	---	Brown	H 7 G 3.31	Gel with acids. MnO 28.0, FeO 15.8, ZnO 7.8%.		
	1.738	SPINEL, zincian (Spinel grp) (Mg, Fe, Zn) Al_2O_4	CUB oct	---					
	1.776	HELVITE (Helvite grp) (Mn, Fe, Zn) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	---					
	1.745	HELVITE (Helvite grp) (Mn, Fe, Zn) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	---					
	1.747	HELVITE (Helvite grp) (Mn, Fe, Zn) ₄ $Be_3(SiO_4)$ ₃ S	CUB tetrah	---					

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.707 1.777	1.748	BERZELIITE (Ca,Mn) ₃ (Mg,Mn) ₂ (AsO ₄) ₃	CUB	Subconch to uneven	Yellow, orange	H 5 G 4-27 F 3	Diss by acids. MnO 8.8%. Garnet-type structure.
1.72 1.81	1.75	YTTRIALITE (Y,Th) ₂ Si ₂ O ₇	HEX (?) metamic	Conch	Olive-green	H 6.5 G 4.3-4.7 infus	Gel with acids.
1.72 1.81	1.75	HELLANDITE (Ca,Y) ₆ (Al,Fe)Si ₄ B ₄ O ₂₀ (OH) ₄	MCL metamic	---	Brown	H 5.5 G 3.70 F 2-3	Gel with acids.
1.72 1.75	1.75	BRITHOLITE (Ce,a,Ca) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX metamic	---	Yellow, brown	H 5 G 3.85 infus	Gel with HCl. Apatite-like structure.
1.72 1.75	1.75	ALLANITE (Epidote grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	MCL metamic	---	Brown, black	H 6 G 3.4-3.7	In part biref. May gel with acids.
1.734 1.773	1.752	GROSSULAR (Garnet grp) Ca ₃ (Al,Fe) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Brown, white, green	H 6-7 G 3.57 F 3	Insol in acids. Gr 85, An 13, Py 1.
1.755	1.755	ARSENOLITE As ₂ O ₃	CUB oct	111	White	H 1.5 G 3.80 F 1	Sol in H ₂ O. Volat when heated.
1.755	1.756	NATANITE Fe ⁺² Sn(OH) ₆	CUB	---	Greenish-brown	H 5 G 4.07	Diss by HCl.
1.735	1.756	PERICLASE (Mg,Fe)O	CUB	100 perf parting 110	Col s	H 5.5 G 3.56 infus	Diss by acids. FeO 5.8%.
1.758	1.758	AZOVSKITE Fe ₃ (PO ₄) ₂ (OH) ₆	Anor (?)	---	Dark brown	H 4 G 2.5	Diss by acids. Faintly biref.
1.747	1.760	DANALITE (Hevitite grp) (Fe,Mn,Zn) ₄ Be ₃ (SiO ₄) ₃ S	CUB tetrah	---	Dark red	H 6 G 3.41 F 3	Gel with acids. FeO 38.4, MnO 7.1, ZnO 6.8%.
1.685	1.763	TRITOMITE (Ce,Y,La,Th) ₅ (Si,B) ₃ (O,OH,F) ₁₃	TRIG metamic	---	Reddish-brown	H 4-5.5 G 4.2 F 3	Gel with acids

1.741 1.782	1.766 1.782	CUB CUB	110, 211 (Mg, Fe, Ca) ₃ (Al, Fe) ₂ (SiO ₄) ₃ RINGWOODITE (Mg, Fe) ₂ SiO ₄	Uneven ---	Red, brown Purple	H 7 G 3.88 F diff	Insol acids. Py 43, Al 41, Sp 2, Gr 8, An 6.
1.72 1.798	1.768	CUB	HEX metamict	Conch	Dark brown to black	H 5-6 G 4.1 infus	Structure of spinel type.
1.65 1. 1.77	1.77	HEX metamict	(Ca, Ce) ₅ (Si, B) ₃ O ₁₂ (OH, F)•xH ₂ O (?)	---	Black	H 5.5 G 5.4 infus	Dec by HCl. Apatite structural type.
1.74 1. 1.77	1.77	TET square prisms metamict	ThO _{4-x} (OH) _{4x}	---	Peach-tan	H 7 G 3.80 F 4	Diff sol in acids, gel. Clouded in section.
1.752 1.796	1.773 1.796	CUB	GROSSULAR-SPESSARTINE (Garnet grp) (Ca, Mn, Fe) ₃ Al ₂ (SiO ₄) ₃	---	Greenish-black	H 8 G 4.20 infus	Insol in acids. Deep green in section. FeO 23.3, MgO 10.8, Fe ₂ O ₃ 11.7%.
1.747 1.798	1.776 1.798	CUB oct	HERCYNITE, magnesian ferrian ("Pneonaste") (Spinel grp) (Fe, Mg)(Al, Fe) ₂ O ₄	Uneven	Yellow, orange	H 5 G 4.45 F 3	Diss by acids. Data for synth end-member. Garnet-type structure.
1.748 1. 1.777	1.777	CUB	MANGANBERZELIITE Ca ₃ Mn ₂ (AsO ₄) ₃	Subconch to uneven	---	Bright yellow	Dec by acids.
1.78	1.78	CUB	HAWLEYITE CdS	---	Black to greenish-black	H 6.5-7 G 3.9-4.3 infus	Gel with acids. Heated material is brief with <u>n</u> = 1.82.
1.70 1.81	1.78 1.81	MCL metamict	GADOLINITE Y ₂ FeBe ₂ Si ₂ O ₁₀	Conch	Red to brown	H 7 G 4.02 F 4	Insol in acids. Al 55, Py 27, Gr 14, Sp 3.
1.766 1.797	1.782	CUB 110, 211	ALMANDINE-PYROPE (Garnet grp) (Fe, Mg, Ca) ₃ Al ₂ (SiO ₄) ₃	Uneven	Gray, green, black	H 8 G 4.38 infus	Insol in acids. ZnO 31.4, FeO 8.5, MgO 2.4%.
1.776 1.805	1.782	CUB oct	GAHNITE (Spinel grp) (Zn, Fe, Mg)Al ₂ O ₄	1111 imperf	---	Yellow, brown	Insol in acids.
1.71 1.82	1.79	TET metamict	ZIRCON ZrSiO ₄ (+x H ₂ O (?)	---	---	H 6 G 3.9-4.0 infus	Dec by HCl.
1.72 1.84	1.79	CUB u mass	BINDEHIMITE (Stibiconite grp) Pb ₂ Sb ₂ O ₆ (O, OH)	Conch	Yellow, brown, gray	H 4-4.5 G 4.6-5.6 F 3-4	---
1.790	1.790	TRIG	PHILIPSBORNITE (Crandalite grp) PbAl ₃ (AsO ₄) ₂ (OH) ₃ •H ₂ O	Conch	Grayish-green	H ~ 4.5 G (4.33)	---

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.773 1.800	1.796	GROSSULAR-UVAROVITE (Garnet grp) $\text{Ca}_3(\text{Al},\text{Cr},\text{Fe})_2(\text{SiO}_4)_3$	CUB ddd	Uneven	Green	H 7 G 3.68 F 4	Insol in acids. Gr 52, W 40, An 8.
1.782 1.805	1.797	ALMANDINE (Garnet grp) (Fe,Mg,Mn,Ca) $_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Reddish-brown	H 7 G 4.10 F 4	Insol in acids. Al 60, Py 18, Sp 15, Gr 7.
1.776 1.833	1.798	SPINEL, chromian (Spinel grp) (Mg,Fe)(Al,Cr,Fe) $_2\text{O}_4$	CUB oct	Uneven	Black	H 8 G 3.94 infus	Insol in acids. MgO 19.5, FeO 9.8, Al ₂ O ₃ 48.3, Cr ₂ O ₃ 19.6%.
1.74 1.90	1.80	STIBICONITE (Stibiconite grp) Sb ⁺³ Sb ⁺⁵ O ₆ (OH) ₂	CUB u mass	Conch	Gray, brown, yellow	H 4-5.5 G 5.2-5.6 infus	Insol in acids.
1.796 1.801	>1.80	MANGANOSTEENSTRUPINE (Ce,La,Th,Ca)MnSiO ₃ (OH) ₂ ·2H ₂ O	Amor	Conch	Black	H 5.5-6 G 3.29	Brownish-red in section.
1.800	1.800	SPESSARTINE (Garnet grp) Mn ₃ Al ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red to brown	H 7 G 4.18 F 3	Insol in acids. Synth end-member.
1.800 1.805	1.800	YAFSOANITE (Zn,Ca,Pb) ₃ TeO ₆	CUB	---	Light to dark brown	---	Diss by acids.
1.766 1.855	1.801	GROSSULAR-ANDRADITE (Garnet grp) (Ca,Fe,Mn) ₃ (Al,Fe) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Green, red, brown	H 7 G 3.71 F 4.5	Insol in acids. Opt. anom. Gr 52, An 42, Al 4, Sp 2.
1.715 1.782 1.818	1.803	KNORRINGITE (Garnet grp) (Mg,Ca,Fe) ₃ (Cr,Al) ₂ (SiO ₄) ₃	CUB u mass	Uneven	Blue-green	H 7 G 3.76	Insol in acids. Kn 53, G 19, Al 14, Py 10, An 3.
1.800 1.814	1.804	THORITE (Th,U)SiO ₄ (+ xH ₂ O)	TET metamict	---	Brown, green, yellow	H 4 G 4.8-6.4	Gel with acids.
1.797 1.814	1.805	GAHNITE (Spinel grp) ZnAl ₂ O ₄	CUB oct	---	Black	H 8 G 4.61 infus	Insol in acids. Synth end-member.
1.800	1.805	SPESSARTINE (Garnet grp) (Mn,Fe) ₃ Al ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown	H 7 G 4.20 F 3.5	Sp 80, Al 17, Py 2.
1.797 1.814	1.805	ALMANDINE (Garnet grp) (Fe,Mg,Ca) ₃ Al ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown	H 7 G 4.15 F 4	Insol in acids. Al 74, Py 11, Gr 11, Sp 4.

1.78 △	1.81	GADOLINITE $\gamma\text{FeBe}_2\text{Si}_2\text{O}_{10}$	MCL metamict	Conch	H 6-5-7 G 3.9-4.3 infus	Gel with acids.
1.75 △	1.81	BRITHOLITE (Ce,La,Ca) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX metamict	---	Yellow, brown	H 5 G 4.1 infus
1.82 △	1.81	ANTHONITE Al ₂ W ₀ ₉ •3H ₂ O	MCL (?) mass	---	White, chalky	H 1 G 4.6
1.805 △	1.814	SPESSARTINE-ALMANDINE (Garnet grp) (Mn,Fe) ₃ Al ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown	H 7 G 4.23 F 3.5
1.818 △	1.817	ROMEITE (Stibiconite grp) (Ca,Na) ₂ Sb ₂ O ₆ (O,OH)	CUB oct	Conch	Yellow	H 5.5-6.5 G 4.7-5.4 F diff
1.87 △	1.818	GAHNITE (Spinel grp) (Zn,Fe)(Al,Fe) ₂ O ₄	CUB oct	---	Black	H 8 G 4.60 infus
1.805 △	1.818	ALMANDINE (Garnet grp) (Fe,Mn) ₃ Al ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown	H 7 G 4.25 F 4
1.830 △	1.82	ZIRCON, var Malacocon (Naegite) (Zr,Th,U)SiO ₄ (+ xH ₂ O)	TET metamict	Conch	Brown, green	H 3-7 G 4.0-4.3 infus
49	1.79 △	GOLDMANITE (Garrett grp) Ca ₃ (V,Al,Fe) ₂ (SiO ₄) ₃	CUB dod mass	---	Deep green to dull green	G 3.74
1.801 △	1.821	UVAROVITE-GROSSULAR (Garnet grp) Ca ₃ (Cr,Al) ₂ (SiO ₄) ₃	CUB dod	---	Green	H 7.5 G 3.81 infus
1.834 △	1.825	ANDRADITE-GROSSULAR (Garnet grp) Ca ₃ (Fe,Al) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown, green	H 7 G 3.77 F 4
1.801 △	1.827	LIANDRATITE U ⁺⁶ (Nb,Ta) ₂ O ₈	HEX u metamict	Conch	Yellow to yellow-brown	H 3.5 G 6.8
1.863 △	1.83	ALMANDINE (Garnet grp) Fe ₃ Al ₂ (SiO ₄) ₃	CUB	Uneven	Red	H 7 G 4.32 F 3
1.818 △	1.830	YTTRIOPROCHLORE ("Obreuchievite") (Pyrochlore grp) (Y,Ca,Na,U) ₁₋₂ (Nb,Ta,Ti) ₂ (O,OH) ₇	CUB	Conch	Dark brown	H 4.5-5 G ~ 3.7
50	1.83					Insol in acids.

Slowly diss by acids, diss by strong KOH soln.

Insol in acids. Sp 53, Al 45, Py 1, Gr 1.

Insol in acids. Opt anom with low brief.

Insol in acids. FeO 1.7, Fe₂O₃ 2.55%.

Insol in acids. Al 74, Sp 20, Py 5, Gr 11.

Partly diss by acids.

Weakly biref. V₂O₃ 18.3, Fe₂O₃ 5.4, Al₂O₃ 4.9%.

Insol in acids. n is for green light. Uv 72, Gr 26, An 1, Py 1.

Insol in acids. An 57, Gr 32, Al 5, Py 3, Sp 3.

Insol in acids. Synth end-member.

Insol in acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.798 1.835	1.833	SPINEL, chromian (Spinel grp) (Mg,Fe)(Al,Cr,Fe) ₂ O ₄	CUB oct	Uneven	Black	H 7 G 3.93 Infus	Insol in acids. MgO 19.3, FeO 11.7, Al ₂ O ₃ 45.4, Cr ₂ O ₃ 16.85, Fe ₂ O ₃ 5.8%.
1.821 1.855	1.834	GOLDMANITE (Garnet grp) Ca ₃ V ₂ (SiO ₄) ₃	CUB	---	Green	G 3.765	Data for synth compd.
1.833 1.855 1.90	1.835	HERCYNITE (Spinel grp) FeAl ₂ O ₄	CUB oct	Uneven	Greenish-black	H 8 G 4.40 Infus	Insol in acids. Data calc for end-member.
1.838	1.838	LIME CaO	CUB	100 perf	Col s	H 3-4 G 3.32 Infus	Diss by acids, slightly sol in H ₂ O. Rapidly alters on exposure to moist air.
1.79 1.94	1.84	BINDEHIMITE (Stibiconite grp) Pb ₂ Sb ₂ O ₆ (OH)	CUB u mass	Conch	Yellow, gray, yellow	H 4-4.5 G 4.6-5.6 F 3-4	Dec by HCl.
1.825	1.855	UVAROVITE (Garnet grp) Ca ₃ (Cr,Al) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Deep green	H 7.5 G 3.77 Infus	Insol in acids. Opt anom. Uv 91, Gr 6, Py 2, An 1.
1.834	1.855	GOLDMANITE (Garnet grp) (Ca,Mn) ₃ V ₂ (SiO ₄) ₃	CUB	---	Dark green	G 3.91	V ₂ O ₃ 24.9, Al ₂ O ₃ 2.0, Fe ₂ O ₃ 1.1, MnO 15.9%.
1.923	>1.86	GALAXITE (Spinel grp) (Mn,Fe)(Al,Fe) ₂ O ₄	CUB oct	Conch	Black	H 7.5 Infus	Insol in acids. MnO 33.1, FeO 2.3, Fe ₂ O ₃ 6.4%.
1.827 1.887	1.863	ARSENOSOBISMITE Bi ₂ (AsO ₄)(OH) ₃	CUB (?) ocherous masses	---	Yellow-brown to green	G ~ 5.7	---
1.817 2.09	1.87	ANDRADITE (Garnet grp) (Ca,Fe,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown, green	H 7 G 3.72 F 3.5	Gel imperfectly with HCl. Opt anom. An 78.5, Gr 12, Al 5.5, Sp 4.
1.915	1.87	ROMEITE (Stibiconite grp) (Ca,Mn) ₂ Sb ₂ O ₆ (OH)	CUB oct	Conch	Yellow	H 5.5-6.5 G 4.7-5.4 F 4.5	Insol in acids. Opt anom. MnO 6.3%.
		PYROCHLORINE (Pyrochlorine grp) (Na,Ca) ₂ (Nb,Ta) ₂ O ₆ (OH,F)	CUB oct	Conch	Brown, yellow	H 5-6 G 3.77 Infus	Dec by H ₂ SO ₄ .

2.00	1.88 ✓	CHEVINKINITE (Ca,Ce,Th) ₄ (Fe,Mg) ₂ (Ti,Fe) ₃ Si ₄ O ₂₂	MCL metamict	Conch	Black to reddish-brown	H 5 G 4-3-4.6 F 4	Gel with acids. Red-brown in section. In part brief.		
1.863 △	1.887 (Garnet grp) Ca ₃ Fe ₂ (SiO ₄) ₃	ANDRADITE	CUB dod	Red	H 7 G 3.86 F 3.5	Gel imperfectly with acids. Data for synth end-member.			
1.890 △	1.890 CALDERITE-ANDRADITE (Garnet grp) (Mn ⁺² ,Ca) ₃ Fe ⁺³ ₂ (SiO ₄) ₃		CUB	Conch	H 7 G 4.07	Fe ₂ O ₃ 29.0, MnO 19.3%. (Ca 47, An 45, Gr 8%) (Min. Abs. 32, 310.)			
1.887 △	1.893 ANDRADITE-SPESSARTINE (Garnet grp) (Ca,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃		CUB	---	Dark green	H 7 G 3.98 F 3.5	Nearly insol in acids. An 71, Sp 28, Al 1.		
1.890 △	1.893 ANDRADITE-SPESSARTINE (Garnet grp) (Ca,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃		CUB	110, 211	Yellow, gray, brown	H 4.5-5.5 G 5.2-5.6 infus	Insol in acids.		
1.890 △	1.893 ANDRADITE-SPESSARTINE (Garnet grp) (Ca,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃		CUB	Conch	Yellow, gray, brown	H 4.5-5.5 G 5.2-5.6 infus	Insol in acids. Data for synth end-member.		
1.890 △	1.893 ANDRADITE-SPESSARTINE (Garnet grp) (Ca,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃		CUB	---	Dark green	H 7 G 3.98 F 3.5	Nearly insol in acids. An 71, Sp 28, Al 1.		
1.835 △	1.90 STIBICONITE (Stibiconite grp)		CUB	u mass	Yellow, gray, brown	H 7 G 3.98 F 3.5	Insol in acids.		
2.05 △	1.90 STIBICONITE (Stibiconite grp)		CUB	---	Yellow, gray, brown	H 7 G 3.98 F 3.5	Insol in acids.		
1.90 △	1.90 MAGNESIOCHROMITE (Spinel grp) MgCrO ₄		CUB	oct	Uneven	Gray-green	Dec by H ₂ SO ₄ .		
1.97 △	1.90 MAGNESIOCHROMITE (Spinel grp) MgCrO ₄		CUB	---	Conch	Greenish-black			
1.87 △	1.915 BETAITE (Pyrochlore grp)		CUB	oct	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.925 △	1.915 (Ca,Na,U) ₂ (Ti,Nb,Ta) ₂ O ₆ (OH)		TET	---	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.74 △	1.92 COFFINITE U(SiO ₄) _{1-x} (OH) _{4x}		CUB	oct	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.856 △	1.923 GALAXITE (Spinel grp) (Mn,Fe)Al ₂ O ₄		CUB	oct	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.915 △	1.925 MICROLITE (Pyrochlore grp)		CUB	oct	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.94 △	1.93 NANTOKITE CuCl		CUB	011	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.985 △	1.93 KALIPYROCHLORITE (Pyrochlore grp)		CUB	---	Conch	Greenish-black	Dec by H ₂ SO ₄ .		
1.84 △	1.94 BINDHEIMITE (Stibiconite grp) Pb ₂ Si ₂ O ₆ (OH)		CUB	u mass	Conch	Gray, brown, green	H 4-4.5 G 5.0-5.6 F 3-4	Dec by HCl.	
1.893 △	1.94 ANDRADITE (Garnet grp) Ca ₃ (Fe,Ti) ₂ (SiO ₄) ₃		CUB	110, 211	Conch	Black	H 7 G 3.79 F 4	Gel with acids. Ti ₀₂ 9.4%.	

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.94	1.94	MELANITE $\text{Ca}_3(\text{Fe}^{+3}, \text{Ti})_2(\text{SiO}_4)_3$	---	---	---	G 3.79	TiO ₂ 9.4%.
1.94	1.94	KIMMAYITE (Garnet grp) $\text{Ca}_3(\text{Zr}, \text{Ti})_2(\text{Al}_2\text{Si})_0_{12}$	CUB	---	Brown	H 7 G 3.94	ZrO ₂ 29.9, TiO ₂ 5.6%.
1.925 1.97	1.94	BETAFITE (Pyrochlore grp) $(\text{Ca}, \text{Na}, \text{U})_2(\text{Ti}, \text{Nb}, \text{Ta})_2^0\text{O}_6(\text{OH}, \text{F})$	CUB oct	Conch	Black, brown, yellow	H 5 G 5.2 infus	Dec by H ₂ SO ₄ .
2.11	1.95	BRANNERITE $(\text{U}, \text{Th}, \text{Ce})\text{Ti}_2\text{O}_6$	MCL metamict	Conch	Black, brown, yellow	H 5-6 G 4.0-4.3 infus	Diss by acids.
1.95	1.95	STEFELDTITE (Stibiconite grp) $\text{Ag}_2\text{Sb}_2(\text{O}, \text{OH})_7$ (?)	CUB mass	---	Black, brown	H 3.5-4.5 G 4.1-4.6	---
1.90 2.10	1.97	MAGNESIOTACHYLITE (Spinel grp) $(\text{Mg}, \text{Fe})(\text{Cr}, \text{Al}, \text{Fe})_2\text{O}_4$	CUB oct	Uneven	Black	H 6 G 4.67 infus	Insol in acids. MgO 16.0, FeO 11.1, Cr ₂ O ₃ 43.6, Al ₂ O ₃ 24.4, Fe ₂ O ₃ 4%.
1.94 2.00	1.97	URANMICROLITE ("Djalmaitite") (Pyrochlore grp) $(\text{Ca}, \text{Na}, \text{U})_2(\text{Ta}, \text{Nb}, \text{Ti})_2^0\text{O}_6(\text{OH})$	CUB oct	Uneven	Brown, black	H 5.5 G 5.8	Dec by H ₂ SO ₄ .
1.98	1.97	MOLURANITE $\text{H}_4\text{U}^{+4}(\text{UO}_2)_3(\text{MoO}_4)_7 \cdot 18\text{H}_2\text{O}$	Amor	Conch	Black	H 3-4 G ~ 4	Diss by acids.
1.98	1.98	URANPYROCHLORITE ("Hatchettolite") (Pyrochlore grp) $(\text{U}, \text{Ca}, \text{Ce})_2(\text{Nb}, \text{Ta})_2^0\text{O}_6(\text{OH}, \text{F})$	CUB oct	Conch	Brown	H 5 G ~ 4.8 infus	Insol in acids.
1.94	1.98	SCHORBLOMITE (Garnet grp) $\text{Ca}_3(\text{Fe}, \text{Ti})_2(\text{Si}, \text{Ti})_3\text{O}_{12}$	CUB 110, 211	Conch	Black	H 7 G 3.85 F 4	Gel with acids. Opt anom. TiO ₂ 16.9%.
1.983	1.985	MAGNUSSONITE $\text{Mn}_5(\text{AsO}_3)_3(\text{OH}, \text{Cl})$	CUB and TET	---	Grass-to-emerald-green	H 3.5-4.0 G 4.30	---
1.93	1.988	KALIPYROCHLORITE (Pyrochlore grp) $(\text{K}, \text{Sr})\text{Nb}_2^0(\text{O}, \text{OH}) \cdot \text{xH}_2\text{O}$	CUB	---	Greenish	H 4-4.5 G 3.48	---
1.88	2.00	CHEVKINITE (Ca, Ce, Th) ₄ (Fe, Mg) ₂ (Ti, Fe)Si ₄ O ₂₂ metamict	MCL	Conch	Black	H 5 G 4.3-4.7 F 4	Gel with acids. Red-brown in section. In part biref.

1.97 2.02	2.00	PYROCHLORE (Pyrochlore grp) (Ca, Na) ₂ (Nb, Ta) ₂	CUB oct	111 fr conch	Brown, red	H 5 G 4.1-4.3 infus	Dec by H ₂ SO ₄ . had n 2.23.	Heated material
[]	>2.00	BITSMUTOMICROLITE ("Westgrenite") (Pyrochlore grp) (Bi, Ca)(Ta, Nb) ₂ O ₆ (OH)	CUB	Uneven	Yellow, pink, brown	H 5 G 6.5-7.2 infus	Insol in acids.	
[]	>2.0	CAFARSITE Ca ₈ (Ti, Fe ⁺² , Fe ⁺³ , Mn) ₆ -7 (As ₃₊) ₁₂ •4H ₂ O	CUB	Conch	Dark brown	H 5.5-6 G 3.90	---	
2.00 2.023	2.02	BETAFITE (Pyrochlore grp) (Ca, Na, U) ₂ (Ti, Nb, Ta) ₂ O ₆ (OH, F)	CUB oct	Conch	Black	H 4.5 G 4.76 infus	Dec by H ₂ SO ₄ .	
2.02 2.07	2.023	MICROLITE (Pyrochlore grp) (Ca, Na) ₂ Ta ₂ O ₆ (O, OH, F)	CUB oct	Conch	Brown, yellow	H 6 G 6.42 infus	---	
2.04	2.04	CHLORALITE PbCu(TeO ₃) ₂ •H ₂ O	CUB	Uneven	Deep green	H 3 G 6.4	---	
1.90 []	2.05	STIBICONITE (Stibiconite grp) Sb ⁺³ Sb ⁺⁵ ₂ O ₆ (O, OH)	CUB u mass	Conch	Yellow, brown	H 4-4.5 G 5.2-5.6 infus	Insol in acids.	
[]	2.05	GOETHITE ("Limonite") FeO(OH)	ORTH cryptocryst	---	Ocher-yellow	H 4 G ~ 3.8 infus	Diss by HCl.	
2.05	2.05	PERCYLITE PbGuCl ₂ (OH) ₂	CUB	100	Sky-blue	H 2.5 G 5.25 F 1	Diss by HNO ₃ . Sky-blue in section. A doubtful mineral.	
2.05	2.05	EULYTITE Bi ₄ Si ₃ O ₁₂	CUB tetrah	110 imperf	Brown, yellow, gray	H 4.5 G 6.11 F 2	Gel with acids. Opt anom.	
2.05	2.05	FERGUSONITE (Y, U)(Nb, Ta) ₄	TET metamict	Conch	Dark brown	H 5 G 4.2-5.8 infus	Diss by hot concd H ₂ SO ₄ . Reddish-brown in section.	
2.19	2.06	ZIRKELITE (Ca, Th, Ce)Zr(Ti, Nb) ₂ O ₇	MCL ps cub	Uneven to conch	Brown to black	H 6 G 4.02 infus	Dec by hot HCl.	
2.18	2.06	LORANSKITE (Y, Ce, Ca)ZrTaO ₆ (?)	ORTH (?) metamict	Conch	Black, brown yellow	H 5 G 4.2-4.6	Status doubtful, perhaps = Euxenite.	
[]	2.06	EUXENITE (Y, Ca, Ce, U, Th)(Nb, Ta, Ti) ₂ O ₆	ORTH metamict	Conch	Brownish-black	H 6.5 G 4.6-5.5 infus	Insol in acids. Reddish-brown in section. After ignition, n = 2.22.	
[]	2.24							

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
["]	2.065	MOSSITE Hg ₂ N(C ₁ ,SO ₄ ,MoO ₄)·H ₂ O	CUB oct	Uneven	Lemon- to canary-yellow	H 3 G 7.72 volat	In part anom biref.
2.023 2.18 2.07	2.07	MICROLITE (Pyrochlore grp) (Ca,Na) ₂ Ta ₂ O ₆ (O,OH,F)	CUB oct	Conch	Brown, yellow	H 6 G 5.95 infus	Insol in acids.
2.15	2.071	CHLORARGYRITE Ag(Cl,Br)	CUB u mass	Sectile	Gray, yellowish	H 2.5 G 5.56 F 1	Diss by NH ₄ OH.
2.07 ["] 2.11	2.08	BAROPYROCHLORE ("Pandaite") (Pyrochlore grp) (Ba,Sr) ₂ (Nb,Ti) ₂ (O,OH) ₇	CUB	111 poor fr conch	Yellow, gray	H 4-4.5 G 3.7-4.0 infus	Insol in HCl.
["]	2.08	PLUMBOPYROCHLORE (Pyrochlore grp) (Pb,Y,U,Ca) ₂ Nb ₂ O ₆ (OH)	CUB	---	Greenish-yellow	G 6.34	---
2.085	2.087	GIANELLAITE Hg ₄ N ₂ (SO ₄) ₂	CUB oct	111 poor	Straw-yellow	H ~ 3 G 7.19 volat	Dec by concd HCl.
["]	2.09	SENAHMONTITE Sb ₂ O ₃	CUB	---	Col s	H 2 G 5.50 F 1.5 volat	Diss by HCl. Anom biref.
1.87 2.20 2.05 2.20	2.09 ~2.1	BISMUTOSTIBICONITE (Stibiconite grp) (Bi,Fe)Sb ₂ O ₇	CUB	---	Yellow to yellow-brown	G 7.38	---
1.97 2.12 2.20	2.10 2.10	ROMEITE, var Schneebergite (Stibiconite grp) (Ca,Fe,Mn,Na) ₂ (Sb,Ti) ₂ O ₆ (O,OH,F)	CUB	111 dist	Honey-yellow	H 6.5 G 5.41 F diff	Insol in acids. Opt anom, low biref.
[""]	2.10	FERRITUNGSTITE (W,Fe ⁺³)(O,OH) ₃ (?)	CUB	---	Yellow	G 4.4-5.2	Dec by HCl.
[""]	2.10	CHROMITE (Spinel grp) (Mn,Fe,Mg)(Cr,Al) ₂ O ₄	CUB oct	Uneven	Brownish-black	H 6 G 4.65 infus	Insol in acids. Cr ₂ O ₃ 49.7, Al ₂ O ₃ 17.6, Fe ₂ O ₃ 8.1, MnO 9.3, FeO 8.7, MgO 2.4, ZnO 4.1%.
[""]	2.20	THORIANITE (Th,U)O ₂	CUB	001 poor fr uneven	Black to brown	H 6 G 9.2 infus	Slowly diss by HNO ₃ or H ₂ SO ₄ .

	2.10	THORUTITE $(\text{Th},\text{U})\text{Ti}_2\text{O}_6$	MCL metamict	Conch	Brownish-black	H 5-6 G 5.82 infus	Compare Brannerite.
Y	2.105	LOPARITE (Perovskite grp) $(\text{Ce},\text{Na},\text{Ca})_2(\text{Ti},\text{Nb})_2\text{O}_6$	ORTH metamict	---	Reddish-brown	H 5-5.5 G 4.48 infus	Insol in acids.
2.38	2.11	BRANNERITE $(\text{U},\text{Th},\text{Ce})\text{Ti}_2\text{O}_6$	MCL metamict	Conch	Black	H 4.5 G 4.2-4.6 infus	Diss with diff by hot HNO_3 .
1.95 V 2.49	2.11	DAVIDITE $(\text{La},\text{Ce})(\text{Y},\text{U},\text{Fe}^{+2})(\text{Ti},\text{Fe}^{+3})_{20}$ $(\text{O},\text{OH})_{38}$	TRIG metamict	---	Brownish-black	H 6-7 G 4.4-4.6 infus	Diss with diff by hot H_2SO_4 .
V 2.3	2.11	CLIFFORDITE $\text{U}\text{Te}_{3.0}$	CUB oct	---	Bright yellow	H 4 G 6.6 (synth)	Diss by concd HCl .
>2.11	2.12	CHROMITE (Spinel grp) FeCr_2O_4	CUB oct	Uneven	Black	H 6 G 5.22 infus	Insol in acids. Data for synth compd.
2.10 A	2.12	BISMUTITE $\text{Bi}_2(\text{CO}_3)_2$	TET u earthy mass	001 dist	Yellow, brown	H 2.5-4 G 6.7-7.4 F 1.5	Diss by acids with eff.
V 2.26	2.12	BORNITE Cu_5FeS_4	CUB u mass	Uneven	Red-brown, tarnishes purple	H 3 G 5.07 F 2	Dec by HNO_3 .
2.13 Na	2.13	MURATAITE $(\text{Na},\text{Y})_4(\text{Zn},\text{Fe})_3(\text{Ti},\text{Nb})_6\text{O}_{18}$ $(\text{F},\text{OH})_4$	CUB	Conch	Black	H 4.50-4.69	Insol in acids except HF.
2.11	2.13	SAMARKITE $(\text{Y},\text{U},\text{Ca},\text{Fe})(\text{Nb},\text{Ta},\text{Ti})_2\text{O}_6$	MCL metamict	Conch	Black	H 4 G 4.0-4.3 F 4.5	Diss by hot concd acids. Reddish-brown in section.
2.25	2.13	YTTROCRASITE $(\text{Y},\text{Th},\text{U},\text{Ca})\text{Ti}_2(\text{O},\text{OH})_6$	ORTH metamict	Conch	Black	H 4-6 G 4.80 infus	Diss by hot H_2SO_4 . Amber in section. In part birefringent.
2.12 V 2.15	2.14	OLDHAMITE $(\text{Ca},\text{Mn})\text{S}$	CUB	100	Pale brown	H 4 G 2.61 infus	Diss by acids. Rapidly dec in air. Meteorite mineral.
2.14	2.14	FORMANITE $(\text{Y},\text{U})(\text{Ta},\text{Nb})\text{O}_4$	TET metamict	Conch	Black	H 5.5-6.5 G 6.17 infus	Diss by hot concd H_2SO_4 .
2.12 V 2.18	2.14	AESCHYNITE-(Y), var Priorite $(\text{Y},\text{Ce},\text{Ca},\text{Th})(\text{Ti},\text{Nb})_2\text{O}_6$	ORTH metamict	Conch	Brown to black	H 5-6 G 4.7-5.0 infus	Diss by hot concd acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[]	2.15	YTTRIOTANTALITE (Y,U,Fe)(Ta,Nb)O ₄	ORTH metamict	010 indist fr conch	Black, brown, yellow	H 5 G 5.7-6.1 infus	Insol in acids. Reddish-brown in section.
2.071 ^ 2.20	2.15	BROMARGYRITE, var Embolite Ag(Br,Cl)	CUB u mass	Sectile	Cols, yellow	H 1-1.5 G 5.6-6.0 F 1	Diss by NH ₄ OH.
[]	2.155	FERGUSONITE-BETA-(Ce) (Ce,La,Nd)NbO ₄	MCL metamict	---	Reddish-brown	H 6 G 5.34-5.44 infus	Dec by H ₂ SO ₄ *
2.16 Li		BLAKEITE Ferric tellurite	Mass	---	Reddish-brown	H > 2 G > 3.1	Diss by acids.
2.17		MANGANOSITE MnO	CUB oct, u mass	001 fair	Emerald-green, turns black	H 5-6 G 5.36 infus	Diss by acids. Emerald-green in powder and section.
56	2.18	PYROCHLORE (Pyrochlore grp) (Ca,Na,Ce) ₂ Nb ₂ O ₆ (OH,F)	CUB oct	111 poor fr conch	Brownish-red	H 5 G 4.1-4.3 infus	Dec by H ₂ SO ₄ *
2.07 ^ 2.2	2.18	ZIRKELITE (Ca,Th,Ce)Zr(Ti,Nb) ₂ O ₇	MCL ps cub	Conch	Black to dark brown	H 5-6 G 4.3-4.7 infus	Insol in acids. Reddish-brown in section.
2.06 ^ 2.28	2.19	FERGUSONITE (Y,U,Ce)(Nb,Ta)O ₄	TET metamict	Conch	Black	H 6 G ~ 5.8 infus	Dec by H ₂ SO ₄ * After ignition, <u>n</u> = 2.14, biref.
2.05 ^	2.192	BIDEAUXITE Pb ₂ AgCl ₃ (F,OH) ₂	CUB	Conch	Cols to pale lavender	H 3 G 6.26 F 1	Diss by HNO ₃ , dec by HCl or by warm NH ₄ OH.
2.10 ^	2.20	THORIANITE (Th,U)O ₂	CUB	001 poor fr uneven	Black	H 6 G 9.32 infus	Slowly diss by HNO ₃ or H ₂ SO ₄ . Nearly opaque.
2.15 ^ 2.253	2.20	BROMARGYRITE, var Iodobromite Ag(Br,Cl,I)	CUB	Sectile	Cols to yellow	H 2 G 6.0-6.2 F 1	Diss by NH ₄ OH.
2.20		MERSITE (Ag,Cu)I	CUB	011 perf sectile	Yellow	H 2.5 G 5-6.4 F 1	Diss by NH ₄ OH. Tw pl 11.1.

2.09 ^	2.20	ROMEITE, var Lewisite (Sr,biconite grp) (Ca,Fe) ₂ (Sb,Ti) ₂ ⁰ ₆ (O,OH) ₆	CUB oct	111 good	Honey-yellow to brown	H 5.5 G 4.95 fus	Insol in acids. Opt anom.
2.205	2.205	KOBELITE (Y,U)(Ti,Nb) ₂ (O,OH) ₆ (?)	Amor	----	Dark brown	----	----
2.18 ^	2.2 calc	STIBIOBETAITE (Pyrochlore grp) (Ca,Sb ⁺³) ₂ (Ti,Nb,Ta) ₂ (O,OH) ₇	CUB	----	Brownish-black	H ~ 5 G 5.30	Nb ₂ ⁰ 21.6, Ta ₂ ⁰ 19.3, Ti ₀ ² 16.5, Sb ₂ ⁰ 23.2, CaO 14.5%.
[]	2.22	POLYMIGNITE (Ca,Ce,Th,Fe)(Nb,Ti,Ta,Zr)O ₄	ORTH metamict	Conch	Black	H 6.5 G 4.8 infus	Reddish-brown in section.
2.19 [] 2.26	2.23	AESCHYNITE (Ce,Ca,Fe,Th)(Ti,Nb) ₂ (O,OH) ₆	ORTH metamict	Conch	Brownish-black	H 5.5 G 4.9-5.2 infus	Insol acids. Reddish-brown in section. Brief after igni- tion.
2.06 ^	2.24	EUXENITE (Y,Ca,Ce,U,Th)(Nb,Ta,Ti) ₂ O ₆	ORTH metamict	Conch	Brownish-black	H 6.5 G 4.6-5.5 infus	Insol in acids. Reddish-brown in section.
2.13 ^	2.25	SAMARSKITE (Y,U,Ca,Fe)(Nb,Ta,Ti) ₂ O ₆	MCL metamict	Conch	Black	H 4-6 G 5.6-5.7 F 5	Diss sol hot acids. In section deep brown to opaque.
[]	2.25	POLYCRASE (Y,Ca,U,Th)(Ti,Nb) ₂ O ₆	ORTH metamict	Conch	Black	H 5-6 G 4.7-4.9 infus	Insol in acids. Reddish-brown in section. Ti analogue of Euxenite.
2.20 ^	2.25	TALHERANITE (Zn,Ca,Ti)O ₂	CUB	Uneven	Orange to red	H 7.5 G 5.01 infus	----
2.12 ^	2.26	BISMUTITE Bi ₂ (CO ₃)O ₂	CUB	Sectile	Yellow, green	H 2.5 G 6.3 F 1	Diss by NH ₄ OH.
[]	2.262	JIXIANITE Pb(W,Fe ⁺³) ₂ (O,OH) ₇	TET ^u mass earthy	001 dist	Yellow, brown	H 2.5-4 G 6.7-7.4 F 1.5	Diss by acids with eff.
2.315	2.265	PEROVSKITE, var Knopite and Dysanalyte (Perovskite grp) (Ca,Ce) ₂ (Ti,Nb) ₂ O ₆	CUB	----	Brownish-black	H ~ 5 G 5.30	----
2.40	2.28	ZIRKELITE (Ca,Th,Ce)Zr(Ti,Nb) ₂ O ₇	ORTH ps cub	100	Black	H 5 G 4.1-4.2 infus	Dec by hot H ₂ SO ₄ . In part brief.
2.18 ^			MCL ps cub	Conch	Black to dark brown	H 5-6 G 4.3-4.7 infus	Insol in acids. Reddish-brown in section.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.11 Δ	2.3	DAVIDITE (La, ³⁺ Fe ⁺²)(Ti,Fe ⁺³) ₂₀ (O,OH) ₃₈	TRIG metamict	---	Brownish-black	H 6-7 G 4.4-4.6 infus	Diss by hot H ₂ SO ₄ .
2.3 (?)		TREVORITE (Spinel grp) NiFe ₂ O ₄	CUB oct	Uneven	Black	H 5 G 5.2 infus	Nearly opaque. Strongly magnetic. $\underline{n} = 2.39$ calc.
2.30 γ		JACOBITE (Spinel grp) (Mn,Fe ⁺² ,Mg)Fe ⁺³ ₂₀ TiTe ₃ O ₈	CUB oct	Uneven	Brownish-black	H 5.5-6.5 G 4.75-4.93 infus	Diss by HCl. Nearly opaque. Weakly magnetic.
2.585		WINSTANLEYITE TiTe ₃ O ₈	CUB	---	Yellow, tan	H 4 G 5.57	Diss slowly in warm HNO ₃ .
2.34		MARSHITE CuI	CUB tetrah	011 perf	Yellow, brown, red	H 2.5 G 5.68 F 1.5	Diss by acids. Disp exceeds that of diamond.
2.346 Na 2.313 Li		FRANKLINITE (Spinel grp) (Zn,Mn,Fe ⁺²)(Mn,Fe ⁺³) ₂ O ₄	CUB oct	Uneven	Iron-black	H 6 G 5.2 infus	Diss by HCl. Reddish-brown in section. Weakly magnetic.
2.36 Li		BUNSENITE NiO	CUB oct	---	Dark green	H 5.5 G 6.90 infus	Diff sol in acids.
2.37 Li		SPHALERITE ZnS	CUB tetrah	110 perf	Col's to pale yellow	H 3.5 G 4.1 F 5	Diss by HCl. Luster resinous. Data for pure ZnS.
2.40 γ	2.37 Na 2.34 Li	MAGNETOSTOFERRITE (Spinel grp) (Mg,Fe)Fe ₂ O ₄	CUB oct	Uneven	Black	H 6-6.5 G 4.5-4.7 infus	Diss by HCl. Nearly opaque to dark red in section. Strongly magnetic.
2.42 Δ	2.38 Na 2.34 Li	LOPARITE (Perovskite grp) (Ce,Na,Ca) ₂ (Ti,Nb) ₂ O ₆	ORTH metamict	---	Black to brown	H 5-5.5 G 4.7-4.8 infus	Insol in acids.
2.105 Δ	2.38	PEROVSKITE (Perovskite grp) Ca ₂ Ti ₂ O ₆	ORTH ps cub	100	Brown to black	H 5-5.5 G 4.1-4.48 infus	Diss by hot H ₂ SO ₄ . In part biref.
2.265 Δ	2.40	ROMANECHITE BaMn ⁺² Mn ⁺⁴ O ₁₆ (OH) ₄	MCL ps orth	---	Black	H 5-6 G 4.4-4.7 infus	Diss by HCl with evolution of Cl.

2.37 2.43	2.40	SPHALERITE (Zn,Fe)S	CUB tetrah	110 perf	Brown	H 3.5-4 G 4.0-4.1 F 5	Diss by HCl. Luster resinous.
2.418	2.418	DIAMOND C	CUB tetrah	111 highly perf	Cols, yellow, blue	H 10 G 3.51 infus	Insol in acids. Tw pl 111. Disp extr.
2.38 2.42	2.42	MAGNETITE (Spinel grp) FeFe ₂ O ₄	CUB oct	Parting 111 fr uneven	Black	H 6 G 5.17 infus	Diss by HCl. Nearly opaque. Strongly magnetic.
>2.42		SILLENITE Bi ₁₂ SiO ₂₀	CUB	---	Green	H soft G 8.8 calc	---
2.40 2.47	2.43	SPHALERITE (Zn,Fe)S	CUB tetrah	110 perf	Brown to black	H 3.5-4 G 4.0 F 5	Diss by acids. Luster resinous. Fe 10.4%.
[]	2.45	BITBYITE (Mn,Fe) ₂ O ₃	CUB	111 in traces	Black	H 6-6.5 G 4.95-5.05 F 4 to infus	Slowly diss by HCl.
2.36 2.43	2.45 2.41 Li	FRANKLINITE (Spinel grp) (Zn,Mn,Fe ⁺²)(Mn,Fe ⁺³) ₂ O ₄	CUB oct	Uneven	Iron-black	H 6 G 5.2 infus	Diss by HCl. Reddish-brown in section. Weakly magnetic.
59	2.47	SPHALERITE (Zn,Fe)S	CUB tetrah	110 perf	Brown to black	H 3.5-4 G 3.9 F 4.5-5	Diss by acids. Luster resinous. Fe 17.1%.
2.11 2.49	2.49	BRANNERITE (U,Ca,Ce)Ti ₂ O ₆	MCL metamict	Conch	Black	H 4.5 G 4.4-5.0 infus	Diff sol in acids. Light reddish-brown in section.
2.49 Li		CADMIUM OXIDE CdO	CUB	111	Black	H 3 G 8.1-8.2 infus	Diss by HCl.
2.5		STILLEITE ZnSe	CUB	---	---	H 4-5 G 5.29 calc F 5	---
>2.5		CORDEROITE Hg ₃ S ₂ Cl ₂	CUB	---	Orange-pink	G (6.89)	---
2.58 2.49 Li	2.58 Na 2.49 Li	EGLESTONITE Hg ₆ Cl ₃ O ₂ H	CUB	---	Yellow	H 2-3 G 8.33-8.55	Dec by acids. Volat when heated. Anom biref. Darkens on exposure.
2.30	2.585 Na 2.535 Li	JACOBSSITE (Spinel grp) (Mn,Mg,Fe ⁺²)Fe ₂ O ₄	CUB oct	Uneven	Black to brownish-black	H 5.5-6.5 G 4.76 infus	Diss by HCl. Weakly to moderately magnetic.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.6	MAGHEMITE $\gamma\text{-Fe}_2\text{O}_3$	CUB	---	Brownish-black	H 5 G ~ 4.8 infus	Diss by HCl. Strongly magnetic.	
2.60	CRYPTOMELANE $K(\text{Mn}^{+4}, \text{Mn}^{+2})_8\text{S}_{16}$	MCL ps tet u mass botryoidal	---	Steel-gray to black	H 6 G 4.2-4.4 infus	Diss by HCl. Streak brownish-black.	
2.68 Na 2.62 Li	HAUERITE MnS_2	CUB pyrito- hedral	001 perf	Brownish-black	H 4 G 3.46 F 3	Diss by HCl. Streak reddish.	
2.69 Na 2.68 Li	METACINNABAR HgS	CUB tetrah	Subconch to uneven	Grayish-black	H 3 G 7.65 volat	---	
2.78 Na 2.73 Li	ALABANDITE MnS	CUB	100, 110	Iron-black	H 3.5-4 G 4.0 F 3	Diss by HCl. Streak green.	
2.85 Li	CUPRITE Cu_2^0	CUB gyroidal	111 inter- rupted	Red	H 3.5-4 G 6.14 F 3	Diss by HCl. Streak crimson.	
2.93 Na 3.09 Li	DOMEYKITE Cu_3As	CUB u mass	Uneven	Tin-white to steel-gray	H 3-3.5 G 7.2-7.9 F 2	Dec by HNO_3 .	
3.02 Na 2.97 Li	TENNANTITE $(\text{Cu}, \text{Fe})_{12}(\text{As}, \text{Sb})_4\text{S}_{13}$	CUB u mass	---	Gray, iron- black	H 3-4.5 G 4.6-5.1 F 1.5	Dec by HNO_3 . In section bright red to opaque. Tw pl III.	
3.24 Na 3.22 Li	TIEMANNITE HgSe	CUB tetrah	Uneven to conch	Steel- to blackish-gray	H 2.5 G 8.30-8.47 volat	Dec by HNO_3 .	
3.30 Na 4.16 Li	ALTITE PbTe	CUB u mass	001 perf	Tin-white	H 3 G 8.15 F 1.5	Dec by HNO_3 .	
3.45 Na 3.37 Li	PETZITE Ag_3AuTe_2	CUB u mass	001, fr subconch	Steel-gray	H 2.5 G 9.13 F 1.5	Dec by HNO_3 .	
3.50 Na 3.79 Li	SPERRYLITE PtAs_2	CUB	001 indist fr conch	Tin-white	H 6-7 G 10.58 fus	---	

3.80 4.27	Na Li	LINNAEITE $(\text{Co}, \text{Ni})_3\text{S}_4$	CUB oct	001. imperf fr uneven	Steel-gray	H 4.5-5.5 G 4.85 F 2	Dec by HNO_3 .
3.89 3.87	Na Li	GALENA PbS	CUB	001. perf	Lead-gray	H 2.5-3 G 7.58 F 2	Dec by HNO_3 .
4.50 4.18	Na Li	PYRITE FeS_2	CUB pyrito- hedral	Conch to uneven	Pale brass- yellow	H 6-6.5 G 5.02 F 2.5-3	Dec by HNO_3 .

Table 4. Uniaxial positive minerals

Other entries	Refractive index ω	Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.309	1.312	.003	ICE H_2O	HEX mass	Conch	Col s	H 1.5, G 0.92	Melts at 0°C.
1.378	1.390	.012	SELLAITE MgF_2	TET prism	010, 110, perf	Col s	H 5, G 3.16, F 5	Diss by concd H_2SO_4 . F with intumescence. El pos.
1.40	---	mod	"CHRYSOCOLLA" $Cu_2H_2Si_2O_5(OH)_4 \cdot xH_2O$	Fib $\frac{C}{\bar{C}}$ opal-like	---	Green	H 2+, G 2, infus	Dec by acids. Pleo faint, 0 nearly col s, E pale bluish-green. Perhaps stained opal.
1.439	1.442	.0029	KOGARKOITE $Na_3(SO_4)F$	MCL ps trig	---	Col s	H 3.5, G 2.67, F 1.5	Sol in H_2O . Biref determined on synth crystal.
1.443	1.445	.0012	SCHAIRERITE $Na_{21}(SO_4)_7F_6Cl$	TRIG steep rhombs	Conch	Col s	H 3.5, G 2.62, F 1.5	Sol in H_2O .
1.447	1.449	.002	GALEITE $Na_{15}(SO_4)_5F_4Cl$	TRIG	---	White	G 2.61, F 1.5	Sol in H_2O .
1.0 red 1.05 Na 1.80 Ti	---	---	COVELLITE CuS	HEX plates	0001 perf	Indigo-blue	H 1.5-2, G 4.6-4.76, F 2	Diss by HNO_3 . Translucent only in thinnest plates. In transmitted light, green and pleo c.
1.46	1.57	.11	"CHRYSOCOLLA" $(Cu,Al)_2H_2Si_2O_5(OH)_4 \cdot xH_2O$	MCL fib	---	Bluish-green	H 3, G 2.4, infus	Disp extr.
1.460	1.478	.018	LANNONITE $HCa_4Mg_2Al_4(SO_4)_8F \cdot 3H_2O$	TET	---	White	G 2.22	Diss by acids.
1.461	1.474	.013	TINCALCONITE $Na_2B_4O_7 \cdot 5H_2O$	Hackly	Col s, white	H 2.5, G 1.88, F 2	Sol in H_2O . Data for synth compd.	
1.461	1.465	.004	CHABAZITE (Zeolite grp) $(Ca,Na_2)Al_2Si_4O_12 \cdot 6H_2O$	TRIG rhombs	10T1 dist	White, red	H 4.5, G 2.05, F 3	Dec by HCl with sepn of slimy silica. 2V small to 0°.
1.468	1.473	.005	ERIONITE (Zeolite grp) $(K_2, Ca, Na_2)_2Al_4Si_14O_36 \cdot 15H_2O$	HEX fib	---	White	G 2.02, F easy	Diff dec by HCl with sepn of silica. El pos.

1.47	1.50	.03	HATCHETTITE Ca ₄ H ₈₂	ORTH (?)	001 good	White	H 1, G. 0.9, F 1	Sol in org liquids. Melts at 65° and burns.
1.475	1.474	.004	GMELINITE (Zeolite grp) (Na ₂ ,Ca)Al ₂ Si ₄ ⁰ 1 ₂ •6H ₂ O	HEX	10TO dist	White	H 4.5, G 2.0-2.1, F 3	Dec by HCl. 2V small to 0°. Tw axis c. Dec by HCl.
1.476	1.474	.003	HERSCHELITE (Zeolite grp) (Na ₂ ,Ca)Al ₂ Si ₄ ⁰ 1 ₂ •6H ₂ O	TRIG	---	White	H 4.5, G 2.1, F 3	Dec by HCl.
1.475	1.492	.020	GAGARINITE NaCaY(F,C ₁) ₆	HEX	Prism, fair	Creamy, yellow	H 3-4.5, G 4.1-4.2	Dec by acids. Anom biax, 2V up to 20°.
1.468	1.477	.003	ERIONITE (Zeolite grp) (K ₂ ,Ca,Na ₂) ₂ Al ₄ Si ₁₄ ⁰ 36. 15H ₂ O	HEX fib	---	White	G 2.08, F easy	Diff dec by HCl with sepn of silica. E1 pos.
1.478	1.484	.006	BENTONITE Ca ₆ (Cr,Al) ₂ (Si ₂) ₃ 10OH) ₁₂ 26H ₂ O	HEX	10TO perf 0001 dist	Bright violet	H 2 2.03	Cr ₂ O ₃ 7.5, Al ₂ O ₃ 1.0%. Pleoc. E pale violet, 0 nearly col.
1.48	---	wk	FAUJASITE (Zeolite grp) (Na ₂ ,Ca)Al ₂ Si ₄ ⁰ 1 ₂ •8H ₂ O	CUB oct	111 dist	White	H 5, C 1.92, F 3	Dec by HCl. Uniax in 8 segments.
1.480	1.493	.013	TETRANATROLITE (Zeolite grp) Na ₂ Al ₂ Si ₃ O ₁₀ •2H ₂ O	TET	---	White	H 5, G 2.25, F 2	Gel with HCl.
1.481	1.483	.002	LEVYNE (Zeolite grp) CaAl ₂ Si ₄ ⁰ 1 ₂ •6H ₂ O	TRIG	0221 dist	White	H 4, G 2.14, F 2-2.5	Gel with acids.
1.484	1.603	.119	UREA CO(NH ₂) ₂	TET pyram	---	Yellow, brown	G 1.33	Diss by H ₂ O.
1.487	1.492	.005	APHTHITALITE (K,Na) ₃ Na(SO ₄) ₂	TRIG tab, rhombs	10TO fair 0001 poor	White	H 3, G 2.68, F 1.5	Slowly sol in H ₂ O.
1.498	1.487	.002	CHABAZITE (Zeolite grp) (Ca,Na ₂)Al ₂ Si ₄ ⁰ 1 ₂ •6H ₂ O	TRIG rhombs	10T1 dist	White, red	H 4.5, G 2.08, F 3	Dec by HCl with sepn of silica. 2V small to 0°.
1.461	1.489	.012	DOUGLASITE K ₂ FeCl ₄ •2H ₂ O	MCL u mass	---	Light green, alters to reddish	G 2.16	Nearly uniax. Sol in H ₂ O. Tends to lie on face normal to opt. axes.
1.503	1.488	1.500	DOUGLASITE K ₂ FeCl ₄ •2H ₂ O	ORTH fib	110 perf 010 imperf	White	H 5, G 2.26, F 2-3	Gel with acids. E1 pos. Submicroscopic inter- growths of fibers give 2V near 0°.
1.490	1.502	.012	NATROLITE (Zeolite grp) (Na ₂ ,Ca)Al ₂ Si ₃ ⁰ 1 ₀ •2H ₂ O	ORTH fib	---	Pale violet- blue	H 6-7, G 2.35	---
1.491	1.507	.016	GIUSEPPIITE (Canocrinite grp) (Na,K,Ca) ₇₋₈ (Si,Al) ₁₂ O ₂₄ (SO ₄ ,Cl) ₁₋₂	HEX	---			

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence ω	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.504	1.496	1.542	.046	PARAFFIN (mixt. of hydrocarbons)	HEX and ORTH plates and fib	---	White, cols	H 1, G 0.9, F very easy	Insol in acids.
1.492	1.496	1.502	.006	TUGUTIPITE $\text{Na}_4\text{BeAlSi}_4\text{O}_1\text{Cl}$	TET	Pyramid, conch	Rose, gray, bluish	H 4, G 2.34	---
1.487	1.498	1.503	.005	APHTHITALITE (K, Na, Cu) ₃ Na(SO ₄) ₂	TRIG	10TO fair 0001 poor	White, blue, yellowish	H 3, G 2.68, F 1.5	Slowly sol in H ₂ O.
1.505	1.500	1.502	.002	GARRDONITE (Zeolite grp) $\text{Na}_2\text{Ca}_5\text{Al}_{12}\text{Si}_{20}\text{O}_{64} \cdot 27\text{H}_2\text{O}$	ORTH ps tet, radiating	2 at 90°, poor	Col's	G 2.15, H 4	---
1.487	1.503	1.507	.004	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na}_2)\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG rhombs	10T1 dist	White, red	H 4.5, G 2.09, F 3	Dec by HCl with sepn of silica. 2V small to 0°.
1.496	1.504	1.550	.046	PARAFFIN (mixt. of hydrocarbons)	HEX (?) and ORTH, plates and fib	---	White, cols	H 1, G 0.9, F very easy	Insol in acids. Lies on base when crushed below cover glass.
1.508	1.509	.001	LEUCITE KAlSi_2O_6	TET ps cub, 211	110 poor	Col's	H 6, G 2.5, infus	Dec by acids. Inclusions are characteristic. 2V small to 0°. Poly tw.	
1.513	1.510	1.515	.005	FRANCONELLITE $\text{H}_6(\text{K}, \text{NH}_4)_3\text{Al}_5(\text{PO}_4)_8 \cdot 18\text{H}_2\text{O}$	TRIG nodular, pulverulent	---	Yellow, white	Very soft, G 2.27	Diss by acids.
1.510	1.512	.002	FRANZNITE (Cancrinite grp) $(\text{Na}, \text{Ca})_7(\text{Si}, \text{Al})_{12}\text{O}_{24}$ $(\text{SO}_4, \text{CO}_3, \text{OH}, \text{Cl})_3 \cdot \text{H}_2\text{O}$	HEX pris	---	White	H 5, G 2.49	---	
1.511	1.518	.007	LEIFITE $\text{Na}_2(\text{Si}, \text{Al}, \text{Be})_7(\text{O}, \text{OH}, \text{F})_{14}$	TRIG pris	Pris dist	Col's	H 6, G 2.5, F easy	Insol in HCl. El pos.	
1.514	1.522	.008	NITROCALCITE $\text{Ca}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$	Fib, efflor	One perf	Col's	G 1.90	Readily sol in H ₂ O, hygroscopic. Data on natural material (synth material biax neg, $\beta = 1.498$).	
1.515	1.54	.025	OZOCERITE (hydrocarbon)	ORTH (?) fib	---	Pale yellow	H 2.5, G 1.9-2.1, F very easy	Insol in acids. Burns. El neg. Lies on base.	

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.537	1.542	.005	ROEDDERITE (Osumilite grp) (Na,K) ₂ Mg ₅ Si ₁₂ O ₃₀	HEX	---	Col s	G 2.6	---	
1.540	1.560	.020	CHLORMAGALUMINITE (Mg,Fe ⁺²) ₄ Al ₂ (OH) ₁₂ (Cl, ₅ CO ₃)•2H ₂ O	HEX	0001 perf	Col s to greenish-brown	Soft, G 2.0-2.1, infus	Diss by acids.	
1.540	1.546	.006	OSUMILITE-(Mg) (Osumilite grp) (K,Na)(Mg,Fe) ₂ (Al,Fe) ₃ (Si,Al) ₁₂ O ₃₀ •H ₂ O	HEX	---	Light blue	G 2.63	Pleoc wk, 0 col s, E pale blue.	
1.546	1.542	.001	HYDROXYAPPHYLLITE KCa ₄ Si ₈ O ₂₀ (OH,F)•8H ₂ O	TET	001 perf	Col s	H 4.5-5, G 2.37, F 2	Dec by HCl with sepn of slimy silica. Opt anom.	
1.536 ^λ	1.543	.0025	EIFELITE (Osumilite grp) K ₂ Na ₄ Mg ₉ Si ₂₄ O ₆₀	HEX	---	---	---	Stated to be negative.	
1.543	1.545	.0025	QUARTZ SiO ₂	TRIG	Conch	Col s	H 7, G 2.66, infus	Insol in acids, dec by Hf.	
1.544	1.553	.009	OSUMILITE (K,Na)(Mg,Fe) ₂ (Al,Fe) ₃ (Si,Al) ₁₂ O ₃₀ •H ₂ O	HEX	hex prisms and pyramids	Col s, blue, red	G 2.62	Pleoc wk, 0 light blue, E col s.	
1.540 ^λ	1.550	.004	BASSANITE 2CaSO ₄ •H ₂ O	HEX	---	White	G 2.7	Diss slowly by acids.	
1.547	1.575	.028	VATERITE CaCO ₃	HEX	ps hex, fib	Col s	G 2.64, infus	Pleoc pos.	
1.550	1.650	.100	COQUIMBITE Fe ₂ (SO ₄) ₃ •9H ₂ O	TRIG	1011 imperf	Col s to violet	H 2.5, G 2.1, F 4.5-5	Sol in H ₂ O.	
1.536 ^λ	1.552	.006	KOENENITE Na ₄ (Mg,Ca) ₉ Al ₄ (OH) ₂₂ C ₁₂	TRIG	0001 mic	Yellow to red	H 1.5, G 2.15	Dec by H ₂ O, diss by acids. Sphere aggregates give lower ns down to 1.538.	
1.52 ^λ	1.585	.030	JULIENITE Na ₂ CO ₃ (SCN) ₄ •8H ₂ O	TET	001	Blue	G 1.65	Sol in H ₂ O. An artifact?	
1.556	1.645	.089	FERRINATRITE Na ₃ Fe(SO ₄) ₃ •3H ₂ O	TRIG	1010 perf	White, greenish	H 2.5, G 2.57,	Sol in H ₂ O.	
1.558	1.613	.055	FERRINATRITE Na ₃ Fe(SO ₄) ₃ •3H ₂ O	TRIG	1120 less so	greenish	F 1.5		

1.547	1.558	1.586	.028	BASSANITE 2CaSO ₄ •H ₂ O	HEX ps hex, fib	---	White	G 2.7
1.592	1.559	---	Low med	MERRIHUEITE (Osumilite grp) (K,Na) ₂ (Fe ⁺² ,Mg) ₅ Si ₁₂ O ₃₀	HEX	---	Greenish-blue	G 2.87 (calc)
1.59	1.559	1.580	.021	BRUCITE Mg(OH) ₂	TRIG tab to fib	0001 mic	White	H 2.5, G 2.39, infus
1.576	1.562	1.576	.014	CLINOCHLORE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al)O ₁₀ (OH) ₈	MCL ps hex, plates	0001 mic	White	Diss by acids. Flexible. Luster pearly on base. Opt anom. El neg.
1.564	1.564	1.577	.013	PERHAMITE Ca ₃ Al ₇ (Si ₄ O ₈) ₃ (PO ₄) ₄ (OH) ₃ • 16H ₂ O	HEX	0001 perf	Brown, white	Diff dec by acids. El neg. Fe ₂ O ₃ 1.7%.
1.565	1.565	1.575	.010	PINNOSTE MgB ₂ O ₄ •3H ₂ O	TET	Uneven	Yellowish	H near 5, G 2.64
1.569	1.581	1.581	.012	SCHAURITE Ca ₃ Ge(SO ₄) ₂ (OH) ₆ •3H ₂ O	HEX acic	---	White	H 2.5-3, G 2.5, F diff
1.571	1.571	1.590	.019	NEFEDOVITE Na ₅ Ca ₄ (PO ₄) ₄ F	TCL ps tet	Conch	Col's	H 4.5, G 3.01
1.572	1.572	1.592	.020	ALUNITE (Alunite grp) KAl ₃ (SO ₄) ₂ (OH) ₆	TRIG tab 0001, u mass	0001 dist	White, yellow, reddish	H 3.5-4, G 2.6, infus
1.572	1.572	1.586	.014	EUCRYPTITE LiAlSiO ₄	TRIG	Conch	Col's	H 5.5-6.5, G 2.66
1.573	1.573	1.599	.026	TUNISITE NaCa ₂ Al ₄ (CO ₃) ₄ (OH) ₈ Cl	TET	0001 perf pris good	Col's	H 4.5, G 2.51, infus
1.574	1.574	1.590	.016	NATROALUNITE (Alunite grp) NaAl ₃ (SO ₄) ₂ (OH) ₆	TRIG tab 0001, u mass	0001 dist	White, yellow	H 4, G 2.78, infus
1.603	1.576	1.576	1.579	CLINOCHLORE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al)O ₁₀ (OH) ₈	MCL ps hex, plates	001 perf	Green, white	H 2, G 2.7, F 5-5.5
1.562 1.587	1.577	1.577	---	NAMUWITE (Zn,Ca) ₄ (SO ₄)(OH) ₆ •4H ₂ O	HEX	0001 perf	Pale sea-green	Dec by H ₂ SO ₄ . El neg. Abnormal blue interf color. Pleoc, X, Y green, Z col's.
1.580	1.580	1.588	.008	COERULEOLACTITE (Ca,Cu)Al ₆ (PO ₄) ₄ (OH) ₈ • 4-5H ₂ O	TCL (?) fib crusts	---	Milk-white, light blue	Diss by acids. Related to turquoise.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.572 1.620	1.583 1.656	1.595 .071	.012	ALUNITE (Alunite grp) (K, Na)(Al, Fe) ₃ (SO ₄) ₂ (OH) ₆	TRIG tab 0001 u mass	0001 dist	White, red, yellow	H 4, G 2.7, infus	Insol in acids, but diss by acids after gentle heating. El neg.
1.575 1.604	1.585 1.586	1.656 1.595	.071 .009	CACOGENITE Fe ⁺³ 9(Po ₄) ₄ (OH) ₁₅ ·18H ₂ O	HEX needles c	---	Yellow to brownish	H 3-4, G 2.3, F 2.5-3	Diss by acids. Pleoc, 0 cols to pale yellow, E orange- to canary-yellow.
1.594				WARDITE NaAl ₃ (PO ₄) ₂ (OH) ₄ ·2H ₂ O	TET	001 perf	Col to pale green	H 5, G 2.87, F 3	Slowly diss by acids. Opt anom, biax in sectors. El neg.
1.576 1.598	1.587 1.588	1.594 1.593	.007 .005	CLINOCLORE (Chlorite grp) (Mg, Fe)Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex, plates	001 perf	Greenish- gray	H 2.5, G 2.67, F 4.5	Slowly dec by acids. 2V 0-5°. El neg.
1.604				EUDIALYTE Na ₄ (Ca, Ce) ₂ Fe ₂ ZnSi ₈ O ₂₂ (OH, Cl) ₂	TRIG	0001 dist 1020 poor	Yellow, pink, brown	H 5-5.5, G 2.85, F 3.0	Gel with acids. Opt anom, biax. Pleoc wk in pink and yellow.
1.588	1.589	1.589	.001	RINNEITE K ₃ NaFeCl ₆	TRIG	1120 good	Col, rose, yellow	H 3, G 2.35, F easy	Sol in H ₂ O. Anom biax. Abnormal interef colors in ultrablue.
1.59	1.60	.01		BRUCITE (Mg, Mn)(OH) ₂	TRIG	0001 perf	Light brown	H 2.5, G 2.5, infus	Diss by HCl. El neg. MnO 18.1, ZnO 3.7%.
1.559 1.707	1.59	---		CHLORMANGANOKALITE K ₄ MnCl ₆	TRIG	Conch	Yellow	H 2.5, G 2.31, F easy	Sol in H ₂ O. Deliq.
1.559 1.592	1.592	---	very wk	MERRIHUEITE (Osumilite grp)	HEX	---	Greenish-blue	---	Opt char unk. MgO 4.4 FeO 24.9%.
1.586	1.604	.010	low to med	WARDITE NaAl ₃ (PO ₄) ₂ (OH) ₄ ·2H ₂ O	TET	001 perf	Col to green	H 5, G 2.81, F 3	Slowly diss by acids. Opt anom, biax in sectors. El neg.
1.597	1.612	.015		AMESITE (Kaolinite- Serpentine grp) (Mg, Fe)Al ₂ (Al ₂ Si ₂) ₁₀ (OH) ₈	MCL ps hex, plates	001 mic	Pale bluish-green	H 2-3, G 2.77, infus	Slowly dec by HCl.
1.587 1.618	1.598 1.606	.008		CLINOCLORE (Chlorite grp) (Mg, Fe)Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex, plates	001 mic	Green	H 2, G 2.80, F 5	Diff dec by HCl. 2V 0- 20°. El neg. FeO 9.1, Fe ₂ O ₃ 1.9%.

Gel with hot HCl.

Gel

TRIG

COMBEITE

Na₂Ca₂Si₃O₉

NARSARSKITE

Na₂(Ti,Fe)Si₄(O,F)₁₁

TET

tab

110 good

010 dist

0001 dist

0001 perf

0001 perf

Honey-yellow

to reddish

white,

yellow,

infus

H 4, G 2.78,

H 4, G 2.75,

F easy

F

G 2.84

Insol in acids. Pleoc
wk, 0 yellow, E cols.

Col s

NATROALUNITE

(Alunite grp)

(Na,K)Al₃(SO₄,PO₄)₂(OH)₆

TRIG

tab 0001,

u mass

H 6-7, G 2.75,

F easy

H 4, G 2.78,

infus

H 4, G 2.78,

H 4, G 2.75,

F

Gel with hot HCl.

Insol in acids but diss in
acids after being heated
gently. Na₂O 3.9,
K₂O 2.3, P₂O₅ 5.1%.

Col s

CALCIUM CATALPILITE

CaZrSi₃O₉·2H₂O

CACOXENITE

Fe⁺³(PO₄)₄(OH)₁₅·18H₂O

TRIG

hex

needles

c

Col s

COMBEITE

Na₂Ca₂Si₃O₉

NARSARSKITE

Na₂(Ti,Fe)Si₄(O,F)₁₁

TET

tab

110 good

010 dist

0001 dist

0001 perf

Honey-yellow

to reddish

white,

yellow,

infus

H 4, G 2.78,

H 4, G 2.75,

F

Gel with hot HCl.

Col s

CALCIUM CATALPILITE

CaZrSi₃O₉·2H₂O

CACOXENITE

Fe⁺³(PO₄)₄(OH)₁₅·18H₂O

TRIG

hex

needles

c

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence ω	Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ								
1.598	1.618	1.621	.003		CLINOCHLORITE (Chlorite grp) (Mg,Fe)5Al(Si3Al)O10(OH)8	MCL ps hex	001 perf	Green	H 2, G 2.8	Pleoc wk in green. FeO 13.7, Fe2O3 1.9%.
1.583	1.620	1.641	.021		ALUNITE (Alunite grp) K(Al,Fe)(SO4)2(OH)6	TRIG u mass	0001 dist	White to reddish	H 4, G 2.81, infus	Insol in acids, but after gentle heating diss by acids. El neg. Fe2O3 5.0%.
1.640	1.620	1.630	.010		GOYAZITE (Crandallite grp) SrAl3(Po4)2(OH)5•H2O	TRIG tab 001	0001 perf	Cols to brown	H 4.5-5, G 3.15-3.26, F 4	Slowly diss by acids. Zoned. Anom bix on basal sections. Pleoc, 0 red-brown, E yellow.
1.613	1.623	1.634	.011		CRANDALLITE (Crandallite grp) CaAl3(Po4)2(OH)5•H2O	TRIG u fib	0001 perf	Cols to yellow	H 5, G 2.8, F 3	Diff sol in acids. Fuses with intumescence. El neg.
1.627					WADEITE K2CaZrSi4O12	HEX	2 dist fr conch	Col, rose, lilac	H 6, G 3.11, F diff	Insol in acids. El pos.
1.624	1.624	1.673	.049		WADEITE K2CaZrSi4O12	HEX	2 dist fr conch	Col, rose, lilac	H 6, G 3.10, F diff	Insol in acids. Anom bix. El pos.
1.624	1.624	1.653	.029		EUDIALYTE (OH,C1)2	TRIG	0001 dist IT20 poor	Yellow, pink, brown	H 5-5.5, G 2.94, F 4-5	Gel with acids. Opt anom, bix. Pleoc wk in pink and yellow.
1.604	1.624	1.629	.005		GORCEIXITE (Ba,Ca,Ce)Al3(Po4)2(OH)5•H2O	TRIG u mass	---	Brown, white	H 6, G 3.09-3.32	---
1.610	1.625	--			SYNBERGITE (Beudantite grp) SrAl3(Po4)5(OH)6	TRIG ps cub	0001 dist	Col, yellow, brown	H 4.5, G 2.98-3.22, F diff	Diss with diff by acids. Basal section may show 6 bix sectors.
1.639	1.626	1.640	.014		METATORBERNITE (Meta-autunite grp) Cu(UO2)2(Po4)2•8H2O	TET tab 001	001 mic	Green	H 2.5, G 3.6-3.8, F 3	Diss by acids. Pleoc wk, 0 bright green. E cols. Not fluor in UV.
1.62	1.626	1.627	.001		CRANDALLITE (Crandallite grp) (Ca,Sr,Ce)Al3(Po4)2(OH)5•H2O	TRIG u fib	0001 perf	Col to yellow	H 5, G 3.2, F 4	Diff sol in acids. SrO 5.1, RE2O3 10.5%.

1.629	---	KAMAIISHILITE Ca ₂ Al ₂ Si ₆ (OH) ₂	TET	---	Cols	---			
1.629	1.632	.003 BABEFFHITE BaBe(PO ₄)(O,F)	TET tab	---	White	G 4.31			
1.630	1.638	.008 HAUCKITE Zn ₁₈ (Mg,Mn) ₂₄ Fe ⁺³ (SO ₄) ₄ (CO ₃) ₂ (OH) ₈₁ (?)	HEX 001 perf	H 2-3, G 3.10 Orange to yellow			Insol in acids, diss by aqua regia or HF.		
1.630	1.637	.007 MELILITE (Meilielite grp) (Ca,Na) ₂ (Mg,Al) ₂ Si ₄ O ₁₀	TET	001, 110	Cols	H 5, G 2.95- 3.05, F 3	Pleoc., 0 golden-brown, E pale yellow. ZnO 36.0, MnO 17.1%.		
1.631	1.652	.021 FAHEYITE (Mn,Mg)Fe ⁺³ Be ₂ (PO ₄) ₄ · 6H ₂ O	HEX fib Pris perf		White to brownish	G 2.66	Gel with acids. Fe ₂ O ₃ 1.9, FeO 3.1, Na ₂ O 3.75%.	Slowly diss by hot acids. E1 pos.	
1.630	1.632	.007 AKERMANITE (Meilielite grp) Ca ₂ MgSi ₂ O ₇	TET	001, 110	Cols	H 5, G 2.95	Gel with acids. Data for synth compd.		
1.633	---	Low BURITITE CaSn(OH) ₆	TRIG 001 good			H ~ 3, G 3.28	Anom biax.		
1.636	1.647	.011 WOODHOUSEITE (Beudantite grp) CaAl ₃ (PO ₄)(SO ₄)(OH) ₆	TRIG ps cub	0001 good	Cols	H 4.5, G 3.01	Diff diss by acids. Basal section may show biax sectors.		
1.662	1.639	.007 SVANBERGITE (Beudantite grp) SrAl ₃ (PO ₄)(SO ₄)(OH) ₆	TRIG tab 0001 dist		Cols, yellow, brown	H 4.5, G 2.98- 3.22, F diff	Diff diss by acids. Basal section may show 6 biax sectors. SrO 21.2, CaO 1.0%.		
1.626	1.640	.011 GOYALITE (Grandalellite grp) SrAl ₃ (PO ₄) ₂ (OH) ₅ H ₂ O	TRIG tab 0001	0001 perf	Cols to brown	H 4.5-5, G 3.15, F 4	Diff diss by acids. Anom biax. 2V 0-20°.		
1.620	1.643	1.755 .112 PARISITE-(Y) ("ittrotroparisite") Ca(Y,Ce) ₂ (CO ₃) ₃ ²	HEX 0001		Yellow, brown	---	No analysis, needs confirmation.		
1.670	1.643	1.73 .087 SYNCHYSITE-(Y) ("boverite") CaY(CO ₃) ₂ F	ORTH ps hex, u mass	001	Red-brown	H 4.5-5, G 3.15, F 4	Diff diss by acids.		
1.650	1.644	1.744 .100 SYNCHYSITE CaCe(CO ₃) ₂ F	ps hex	---	Red-brown	H 4.5, G 3.9	Diff diss by acids.		
1.674	1.664	1.664 .020 TRISTRAMITE (Ca, U ⁺⁴ , Fe ⁺³)(PO ₄ , SO ₄) ₂ H ₂ O	HEX fib	---	Paler to greenish-yellow	G 3.8-4.2	Not fluor in UV light.		
1.646	1.647	.001 BRAITSCHITE (Ca,Na ₂) ₇ (Ce,La) ₂ B ₂₂ O ₄₃ [*]	HEX mass	---	White to pink	G 2.90	---		

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index ω	Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.632 1.653 neg	1.648	1.649	.001 MELILITE (Mellilitte grp) $\text{Ca}_2(\text{Mg},\text{Al})(\text{Si},\text{Al})_2\text{O}_7$	TET	001, 110 ---	Col s	H 5, G 2.9-3.1, F 3	Gel with acids. Data for synth Ak 60, Ge 40.
1.680	1.653	1.675	.022 PLUMBOGUMMITE (Crandalite grp) $\text{PbAl}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG gum-like	---	Yellow, brown	H 4-5, G 4.0, F 4-5 (?)	Diss with diff by HNO_3 .
1.678	1.653	1.661	.008 FLORENCITE (Crandalite grp) $\text{CeAl}_3(\text{PO}_4)_2(\text{OH})_6 \cdot \text{H}_2\text{O}$	TRIG	0001	Pink	H 5, G 3.46, infus	Diss with diff by acids.
1.675	1.654	1.703	.049 RHABDOPHANE (Ce, La) $\text{PO}_4 \cdot \text{H}_2\text{O}$	HEX fib, mass	Uneven	Brown, white, reddish	H 3-4, G 3.5- 4.0, infus	Diss by acids.
1.654	1.670	.016 PHENAKITE Be_2SiO_4	TRIG pris	1120 dist 1011 imperf fr conch	Col s, yellow	Col s	H 7.5, G 2.97, infus	Insol in acids. Tw p1 T010 penet.
1.655	1.656	.001 CAHNITE $\text{Ca}_2\text{BaSiO}_4(\text{OH})_4$	TET	110 perf	Col s	H 3, G 3.06, F 3	H 3, G 3.28, infus	Diss by acids. Abnormal interf colors in brown.
1.662	1.656	1.708	.052 DIOPTASE $\text{CuSiO}_2(\text{OH})_2$	TRIG	1011 perf	Emerald-green	H 2.5, G 2.46	Gel with acids. Anom biax. Abs in thick sections 0 > E.
1.652 1.658	1.656	1.682	.026 DESPUJOLSITE $\text{Ca}_3\text{Mn}^{+4}(\text{SO}_4)_2(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	HEX	Conch	Lemon-yellow	Col s to light yellow	Diss by HCl with evolution of chlorine. Pleoc wk, 0 pale yellow, E yellow.
1.660	1.756	.096 RÖNTGENITE $\text{Ca}_2\text{Ce}_3(\text{CO}_3)_5\text{F}_3$	TRIG	---	---	---	G 4.2	Diff diss by acids.
1.636	1.662	1.669	.007 WOODHOUSEITE (Beudartite grp) $\text{CaAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001 good	Col s, gray, titanic	H 4-4.5, G 2.9- 3.0	Diff diss by acids. Basal sector may show biax sectors.
1.664	1.672	.008 MELIPHANITE var Gugiaite (Ca, Na) $_2\text{Be}(\text{Si}, \text{Al})_2$ (O, OH, F) ₇	TET thin tab	010 perf 001 dist	Col s	H 5, G 3.03	Gel with acids.	
1.665	1.685	.020 THOROGUMMITE (Th, Ce) $[\text{SiO}_4]_{1-x}(\text{OH})_{4x}$	TET u mass earthy	Conch	Yellow, brown	H 3, G 4.0-4.7, infus	Gel with acids.	

1.666	1.747	.081	PETERSITE (Y,Ce, ^{La} ,Ca)Cu ₆ (PO ₄) ₃ (OH) ₆ ·3H ₂ O	HEX	---	Yellow-green	G 3.41	
1.643 ^ 1.678	1.670 ^ 1.678	.098	PARISITE Ca(Ce, ^{La}) ₂ (CO ₃) ₂ F	TRIG	0001	Yellow	H 4.5, G 4.0-4.4, infus	
1.67	1.70	.03	HINSDALITE (Beudantite grp) PbAl ₃ (PO ₄)(SO ₄)(OH) ₆	TRIG	0001 perf	Col s	H 4.5, G 3.65, infus	
1.688	1.673	1.678	.005	FILLLOWITE Na ₂ Ca(Mn,Fe) ₇ (PO ₄) ₆	MCL ps hex	001 perf	Yellow, col s	H 4.5, G 3.43, F 1.5
1.644 ^	1.674	1.770	.096	SYNCHYSITE CaCe(CO ₃) ₂ F	ps hex	---	Red-brown	H 4.5, G 3.9, infus
1.654 ^ 1.692	1.675	1.705	.030	RHABDOPHANE (Ce, ^{La})PO ₄ ·H ₂ O	HEX fib, mass	Uneven	Brown, gray-green	H 3-4, G 3.3- 4.0, infus
1.681	1.675	1.685	.010	BAZIRITE BaZrSi ₃ O ₉	HEX pris	0001	Col s	G 3.82 calc
1.670 ^	1.678	1.767	.089	PARISITE Ca(Ce, ^{La}) ₂ (CO ₃) ₃ F ₂	TRIG	0001	Yellow, brown	H 4-5, G 4.0- 4.4, infus
1.653 ^ 1.695	1.678	1.684	.006	FLORENCITE (Crandaillite grp) CeAl ₃ (PO ₄) ₂ (OH) ₆	TRIG	0001	Brown	H 5-6, G 3.46, infus
1.653 ^ 1.680	1.680	1.698	.018	PLUMBOGUMMITE (Crandaillite grp) PbAl ₃ (PO ₄) ₂ (OH) ₅ ·H ₂ O	TRIG gum-like		Yellow, gray	H 4.5, G 4.0- 4.9, F 4-5 (?)
1.680	1.695	.015	BRUCKITE (Ca, Th, La)(PO ₄) ₃	HEX mass, earthy		Reddish- brown to yellow	G 3.9	
1.675 ^	1.681	1.691	.010	BAZIRITE BaZrSi ₃ O ₉	HEX pris	0001	Col s	G 3.82 calc
1.688	1.698	.010	WEILERITE (Beudantite grp) BaAl ₃ (AsO ₄)(SO ₄)(OH) ₆ (?)	TRIG crusts	---	White	---	
1.67	1.688	1.697	.009	HINSDALITE (Beudantite grp) PbAl ₃ (PO ₄)(SO ₄)(OH) ₆	TRIG	0001 perf	Col s	H 4.5, G 3.65, infus
1.689	1.695	.006	ILIMAASSITE Ba ₂ Na ₄ CeFe ⁺³ Nb ₂ Si ₈ O ₂₈ · 5H ₂ O	HEX Conch		Brownish- yellow	H ~ 4, G 3.6	

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index ω	Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
1.720	1.690	1.760	XENOTIME YPO_4	TET	100 perf	Yellow, brown	H 4-5, G 4.7- 5.1, infus	Insol in acids. Pleoc wk. 0 pale yellow, E greenish-yellow.	
1.714	1.691	1.719	.028 WILLEMITE Zn_2SiO_4	TRIG	0001, 1120	Green, white, red, brown	H 5.5, G 4.05- 4.20, infus	Gel with acids. Fluor bright green in UV. MnO 0.12%.	
1.675 1.721	1.692	1.732	.040 RHABDOPHANE-(La) (La,Ce)PO ₄ •H ₂ O	HEX mass	Uneven	Brown, gray	H 3-4, G 4.4, infus	Diss by acids.	
1.665 1.78(?)	1.692	1.710	.018 THOROGUMMITE (Th,Ce)(Si,P)(O,OH) ₄	TET u mass, earthy	Conch	Yellow, brown	H 3, G 4.0-4.7, infus	Gel with acids. Forms series with Thorite?	
1.694	1.701	.007 FLORENCITE-(La)	TRIG	Splintery	Col s to pale yellow	H 5, G 3.52	Diff diss by acids.		
1.695	1.785	.090 BASTNAESITE (Ce,La)CO ₃ F	HEX pris	Parting 0001	Yellow, brown, reddish	H 4, G 4.9-5.1, infus	Diff diss by acids. Pleoc wk, abs E > 0.		
1.717	1.695	1.705	.010 FLORENCITE (Crandalite grp) CeAl ₃ (PO ₄) ₂ (OH) ₆	TRIG	0001	Col s, brown	H 6, G 3.69, infus	Diff diss by acids.	
1.678 1.713	1.697	1.704	.007 MUJIRITE $\text{Ba}_{10}\text{Ca}_2\text{MnTiSi}_{10}^0\text{Si}_{30}$ (OH,Cl,F) ₁₀	TET	001, 100 +indist	Orange	H 2.5, G 3.86, F 3	Dec by acids. Anom purplish-brown interf colors.	---
1.701	1.707	.006 KEMMLITZITE (Beudantite grp) SrAl ₃ (AsO ₄) ₃ (SO ₄) ₂ (OH) ₆	TRIG	0001	Grayish- brown	H 5.5, G 3.63	---		
1.702	1.706	.004 VESUVIANITE $\text{Ca}_{10}\text{Mg}_2\text{Al}_4(\text{Si}_{20})_2(\text{SiO}_4)_5$ (OH,F) ₄	TET pris	110 poor	Col s, green, brown, yellow	H 6-6.5, G 3.37, F 3	Insol in acids. Pleoc wk. Anom biax. Abnormal interf colors in blue and brown.		
1.704	1.765	.061 GOUDIYITE $\text{Cu}_6(\text{Al},\text{V})(\text{AsO}_4)_3(\text{OH})_6$ • $3\text{H}_2\text{O}$	HEX pris c	---	Yellow-green	H 3-4, G 3.50	Pleoc, 0 pale yellow- green, E green. Compare Mixite.		
1.722	1.705	.072 AGARDITE $(\text{Y,Ca})\text{Cu}_6(\text{AsO}_4)_3(\text{OH})_6$ • $3\text{H}_2\text{O}$	HEX	---	Blue-green	G 3.72	Diss by acids.		

1.59 △	1.707	1.722	.015	AMAKINITE (Brucite ser) (Fe,Mg)(OH) ₂	TRIG	0001 fr uneven	Greenish, turns brown	H 3.5-4, G 2.98, infus	Diss by acids. Fe ₀ ⁰ 31.6, FeO 30.4, MnO 3.6%. Compare Brucite at $\omega = 1.59$.
1.730 γ	1.71	(1.715)	.005	HIDALGOITE (Bendallite grp) PbAl ₃ (AsO ₄)(SO ₄)(OH) ₆	TRIG mass	---	White	H 4.5, G 3.96	Insol in acids.
1.695 △	1.713	1.719	.006	FLORENCITE (Crandallite grp) CeAl ₃ (PO ₄) ₂ (OH) ₆	TRIG	0001	Cols, brown	H 6, G 3.4-3.6, infus	Diff diss by acids.
1.691 △	1.714	1.732	.018	WILLEMITE (Zn,Mn) ₂ SiO ₄	TRIG	0001, 1120	Reddish- brown	H 5, G 4.05- 4.20, infus	Gel with acids. MnO 6.8, FeO 0.2%.
1.75 γ	1.714	1.731	.017	OSARIZAWAITE PbCuAl ₂ (SO ₄) ₂ (OH) ₆	TRIG mass	---	Green	G 4.04	Diss by hot HNO ₃ . Anom biax.
1.715	1.728	.013	JASMUNDITE Ca ₁₁ (SiO ₄) ₄ O ₂ S	TET	---	Dark brown	G 3.03	---	---
1.695 △ 1.730 γ	1.717	1.818	.101	BASTNAESITE (Ce,La)CO ₃ F	HEX pris	Parting 0001	Yellow, brown, reddish	H 4, G 4.9-5.1, infus	Diff diss by acids. Pleoc wk, abs E > 0.
1.719	1.733	.014	BROMELLITE BeO	HEX	10TO good 1120, 0001 poor	Col s	H 8.5-9, G 3.02, infus	Insol in acids.	
1.72 △	---	very wk	ZIRCON "Malacon" ZrSiO ₄	TET	Uneven	Brown	H 6-7, G ~ 4	Metamict.	Insol in acids.
1.690 △ 1.753 △	1.720	1.820	.100	XENOTIME YPO ₄	TET	100 perf	Yellow to reddish- brown	H 4-5, G 4.7-5.1, infus	Insol in acids. Pleoc wk, 0 pink to pale yellow, E pale greenish-yellow.
1.692 △	1.721	1.748	.027	RHABDOPHANE (Ce,La)PO ₄ ·H ₂ O	HEX u mass	Uneven	Greenish- yellow, brown	H 3-4, G 3.3-4.0, infus	Diss by acids.
1.705 △	1.722	1.815	.093	AGARDITE (Y,Ca)Cu ₆ (AsO ₄) ₃ (OH) ₆ 3H ₂ O	HEX	---	Blue-green	G 3.72	Diss by acids. Compare Mixite at $\omega = 1.730$.
1.702 △	1.722	1.725	.003	VESUVIANITE Ca ₁₀ Mg ₂ Al ₄ (Si ₂ O ₇) ₂ (SiO ₄) ₅ (OH,F) ₄	TET pris	110 poor	Cols, green, brown, yellow	H 6-6.5, G 3.4, F 3	Insol in acids. Pleoc wk. Anom biax. Abnormal interf colors in blue and brown.
1.724 △	1.746	.022	CONNELLITE Cu ₁₉ (SO ₄)Cl ₄ (OH) ₃₂ ·3H ₂ O	HEX acic	---	Fine blue, greenish- blue	H 3, G 3.36, F 2.5	Diss by acids.	

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index ω	Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.780	1.728	1.730	.002 BRITHOLITE-(Y) ("Abukumalite") (Y,Ca,Ce) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX	0001, 1010, imperf 0001	Reddish-brown	H 6-6.5, G 4.25	Dec by acids. Apatite structural type.
1.717 1.757	1.730	1.830	.100 BASTNAESITE (Ce,La)CO ₃ F	HEX pris	parting 001	Yellow, brown, reddish	H 4, G 4.9-5.1, infus	Diff diss by acids.
1.749	1.730	1.810	.080 MIXITE BiCu ₆ (AsO ₄) ₃ (OH) ₆ •3H ₂ O	HEX acic	One perf	Emerald-to bluish-green	H 3-4, G 3.8, F 2	Diss by acids. Pleoc very wk in shades of green.
1.71 ^	1.730	1.735	.005 HIDALGOITE (Beudantite grp) PbAl ₃ (AsO ₄)(SO ₄)(OH) ₆	TRIG	---	Pale green to emerald-green	H 4.5, G 3.95	Insol in acids. MoO ₃ 3.5%.
1.724 ^	1.736	1.756	.020 CONNELLITE Cu ₁₉ (SO ₄)C ₁₄ (OH) ₃₂ •3H ₂ O	HEX acic	---	Fine blue, greenish-blue	H 3, G 3.36, F 2.5	Diss by acids.
	1.738	1.752	.014 BUTTGENBACHITE Cu ₁₉ (NO ₃) ₂ C ₁₄ (OH) ₃₂ •2H ₂ O	HEX acic	---	Azure-blue	H 3, G 3.33, F 2	Diss by acids or by NH ₄ OH.
	1.747	1.776	.029 FLEISCHERITE Pb ₃ Ge(SO ₄) ₂ (OH) ₆ •3H ₂ O	HEX fib	---	White to pale rose	H soft, G 4.2-4.4	---
1.730 ^	1.749	1.83	.081 MIXITE BiCu ₆ (AsO ₄) ₃ (OH) ₆ •3H ₂ O	HEX acic	One perf	Blue-green	H 3-4, G 3.8, F 2	Diss by acids.
1.714 ^	1.75	~1.76	~.01 OSARIZAWAITE Pb(Cu,AI) ₃ (SO ₄) ₂ (OH) ₆	TRIG	---	Green	---	Diss by hot HNO ₃ . Anom biax.
1.720 ^	1.753	~1.84	~.09 CHERNOVITE (Xenotime ser) Y(As,P)O ₄	TET pris	100 perf	Brown, yellow	H 4-5, G 4.8, infus	Insol in acids. Forms a series with Xenotime.
1.756	1.802	.046 BENITOITE BaTiSi ₃ O ₉	HEX pyram or tab	1011 imperf fr conch	Blue	H 6, G 3.65, F 3	Reddish-brown, yellow	Insol in acids. Pleoc, 0 cols, E purplish-blue.
1.730 ^	1.757	1.864	.107 HYDROXYLBASTNAESITE (Ce,La)CO ₃ (OH,F)	HEX pris	1120 imperf	H 4, G 4.7, infus	Slowly diss by acids.	E1 neg.
1.754 ^	1.757	---	very wk	MCGOVERNITE (Mn,Mg,Zn) ₂₂ (AsO ₃)(AsO ₄) ₃ (SiO ₄) ₃ (OH) ₂	TRIG 0001 mic	H 3, G 3.72	Reddish-bronze	

		TRIG	0001 perf		
1.759	.001	THEOPHRASITE $\text{Ni}(\text{OH})_2$	TET	Emerald-green	H 3.5, G 4.00
1.760	.035	HEMITERMITEITE (Garnet grp) $\text{Ce}_3(\text{Mn},\text{Al})_2(\text{SiO}_4)_2(\text{OH})_4$	TRIG	Clove-brown	Dec by HCl . Anom biax. Pleoc wk, 0 pale yellow, E lemon-yellow.
1.765	1.800	STILLWELLITE ($\text{Ce},\text{La},\text{Ca}$) $\text{BSi}_0.5$	TRIG	Pinkish-gray, brown	Anom biax, 2V 0-6°.
1.765	1.782	LUSUNGITE (Sr,Pb) $\text{Fe}^{+3}(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG	Dark brown	Pleoc wk.
1.766	1.77	- - -	- - -	- - -	In section green, pleoc wk. U biax.
1.778	1.801	CONICHALCITE $\text{CaCu}(\text{AsO}_4)(\text{OH})$	ORTH fib	Uneven	In section green, pleoc wk. U biax.
1.778	1.78	THORITE ThSiO_4	TET	110 dist	Pistachio-to emerald-green
1.780	1.79	.01	HEX	Brown, yellow	H 4-5, G 4.13, F 3
1.780	1.783	BRITHOLITE-(Y) ("Abukumaite") (Y,Ca,Ce) $5(\text{SiO}_4)_3(\text{PO}_4)_3(\text{OH},\text{F})$	HEX	100 perf, 10 TO_0 , imperf	H 5, G 4.4-5.7, infus
1.783	1.879	.096	CHERNOVITE YAsO_4	100 perf	H 6, G 4.35
1.783	1.783	.003	MCL pris	Reddish-brown	Dec by acids. Apatite structural type.
1.794	1.803	.009	MCL u mass	Yellow	H 4-5, G 4.87
1.800	1.845	.045	TET	Brown	H 3-4, G 4.25, F 2-3
1.805	1.81	.01	STISHOVITE SiO_2	Col	Insol in acids. Orange to red in section, not pleoc.
1.806	1.808	- - -	TRIG	Dark brown to yellow	Pleoc wk, 0 cols, E pale pinkish-yellow.
1.806	1.808	wk	HEX	Col	Diss by acids. Orange to red in section, not pleoc.
1.806	1.808	.002	TRIG	Deep coppery brown	Insol in acids.
1.806	1.808	.002	TRIG	Brown, reddish	Brittle.
1.806	1.808	.002	TRIG	H 5.5, G 4.75, infus	Gel with acids. Commonly biax, 2V 0-25°.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	specific gravity, and fusibility	Remarks	
	ω	ϵ								
1.81 1.72 1.878	2.02	.21	GAUDEFROYITE $\text{Ca}_4\text{Mn}^{+3}_{3-x}(\text{BO}_3)_3(\text{CO}_3)(\text{O},\text{OH})_3$	HEX pris	Pris	Black	H 6, G 3.35	Pleoc, 0 reddish-brown, E orange-red.		
1.81 1.85	1.86	.05	ZIRCON ZrSiO_4	TET pris	110 fr uneven	Brown, yellow	H 7.5, G 4.56, infus	Insol in acids. El pos. Partly metamict.		
1.81	1.84	.03	CHROMATITE CaCrO_4	TET	Conch	Lemon-yellow	---	Diss by acids.		
---	---	mod	DOWNHITE SeO_2	TET	---	Cols to yellow	G calc 4.15	Sol in H_2O , very hygroscopic. Mean $n_{\text{calc}} = 1.84$.		
1.78	1.837	1.898	.061	THORITE (Th, U)SiO ₄	TET	110	Green, brown infus	GeL with acids.		
1.842	1.848	.006	PARATACAMITE $\text{Cu}_2\text{Cl}(\text{OH})_3$	TRIG	10T1 good	Green	H 3, G 3.74, F easy	Diss by acids or NH_4OH . Tw 10T1, poly.		
1.85	1.93	.08	ZEMANNITE $(\text{Zn}, \text{Fe})_2(\text{TeO}_3)_3\text{Na}_x\text{H}_{2-x}\text{yH}_2$	HEX pris	---	Light to dark brown	Soft, G 4.36	Pleoc, E yellow-brown, 0 reddish brown, abs 0 > E.		
1.81 1.77	1.88	.03	CHROMATITE CaCrO_4	TET	Pris poor, fr conch	Lemon-yellow	---	Diss by acids.		
1.855	---	.03-.04	LUSUNGITE (Crandallite grp) $(\text{Sr}, \text{Pb})_3\text{Fe}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG	---	Dark brown	---	Pleoc wk.		
1.908	1.860	.005	BYSTROMITE MgSb_2O_6	TET u mass	---	Blue-gray	H 7, G 5.5 calc			
1.864	1.88	.016	WELINITE $(\text{Mn}^{+4}, \text{V})^{+2}_{1-x}(\text{Mn}^{+2}, \text{H}, \text{Mg})_3\text{ySi}(\text{O}, \text{OH})_7$	HEX	0001 poor to dist	Red-brown to reddish black	H 4, G 4.47	Dec by HCl. Blood-red in section. Anom biax with small 2v.		
1.81 1.925	1.895	.017	ZIRCON ZrSiO_4	TET pris	110, fr uneven	Brown, yellow	H 7, G 4.62, infus	Insol in acids. El pos.		

1.880	1.884	.004 MORELANDITE (Apatite grp) $\text{Ba}_5(\text{AsO}_4)_3\text{Cl}$	0001 poor	Light yellow to gray	H 4.5, G 5.33	Diss by cold HNO_3 .
1.89	2.00	.11 DENNINGITE (Mn,Zn) Fe_2O_5	001 perf fr conch	Col's to pale green	H 4, G 5.05, F easy	Diss by cold HCl , insol in HNO_3 . Anom biax, 2v 0- 15°.
1.898	1.915	.017 HUEGELITE $\text{Pb}(\text{UO}_2)_3(\text{AsO}_4)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	MCL tab	100 good	G 5.0	Diss by HNO_3 . Pleoc, 0 yellow, E col's to pale yellow. Anom interf color.
1.90	2.12	.22 TRIPPEKITE CuAs_2O_4	TET	100 perf 110 good	Bluish- green	Diss by acids. Breaks into flexible pieces. Bluish-green in section, not pleoc.
1.855 ^Y	1.908	.007 BYSTROMITE MgSb_2O_6	TET u mass	---	Blue-gray	Diss by concd HCl + KI . Opt char unk.
1.910	1.945	.035 GANOMALITE $\text{Pb}_6(\text{Ca},\text{Mn})_4(\text{Si}_2\text{O}_7)_3(\text{OH})_2$	HEX tab 0001	0001, 1010 perf	Gray	Gel with acids. Anom biax. Bioref variable.
1.912	1.925	.013 SCHEELITE CaW_4	TET oct or tab	101 dist	White, yellow, gray, brown	Dec by HCl with sepn of yellow WO_3 . Fluor bright blue (purple) to yellow in UV (purple when Mo is present).
1.913	1.923	.010 NASONITE $\text{Pb}_6\text{Ca}_4(\text{Si}_2\text{O}_7)_3\text{Cl}_2$	HEX pris	0001, 1010 imperf	White	---
1.945	1.925	.058 ZIRCON ZrSiO_4	TET pris	110, fr uneven	Brown, col's, yellow, green, purple	Insol in acids. EI pos.
1.878 ^Y	1.983	.058 ZIRCON ZrSiO_4	TET oct or tab	101 dist	White, brown	---
1.930 ^Y	1.947	.017 SCHEELITE $\text{Ca}(\text{W},\text{Mo})_4$	TET oct or tab	110, fr uneven	White, col's, yellow, green, purple	Dec by HCl with sepn of yellow WO_3 . Fluor yellow in UV.
1.912 ^Y	1.951	.22 BAOTITE $\text{Ba}_4(\text{Ti},\text{Nb})_8\text{Si}_4\text{O}_{28}\text{Cl}$	TET	2 clvs	Light brown to black	Pleoc, 0 col's to brown, E greenish-yellow to dark red-brown.
1.94	2.16	.22 BAOTITE $\text{Ba}_4(\text{Ti},\text{Nb})_8\text{Si}_4\text{O}_{28}\text{Cl}$	HEX pris	0001, 1010 imperf	White	---
1.913 ^Y	1.945	.025 NASONITE $\text{Pb}_6\text{Ca}_4(\text{Si}_2\text{O}_7)_3\text{Cl}_2$	HEX mass, fib	2 clvs	White	H 4, G 5.7, F 1
1.948	1.958	.010 HEDYPHANE (Apatite grp) $(\text{Pb},\text{Ca})_5(\text{AsO}_4)_3\text{Cl}$	TET pyram	---	White, yellow	Diss by HNO_3 .
1.930 ^Y	1.951	.016 POWELLITE (Scheelite ser) $\text{Ca}(\text{Mo},\text{W})_4$	---	---	---	Dec by HCl with sepn of yellow WO_3 . Data for synth. CaMoO_4 60%, CaW_4 40%.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.925	1.960	2.015	.055	ZIRCON ZrSiO ₄	TET pris	110 fr uneven	Brown, yellow, col s	H 7.5, G 4.65 infus	Insol in acids. El pos.
	1.96	---	wk	DIXENITE Cu ⁺¹ Mn ₁₄ Fe ⁺³ (AsO ₃) ₅ (AsO ₄) ₆ (SiO ₄) ₂ (OH) ₆	TRIG plates	0001 mic	Dark brown	H 3-4, G 4.20-4.36	Diss by HCl. Glowing red in transmitted light.
1.973Na 1.956Li	2.656Na 2.601Li	.683 .645	Calomel Hg ₂ C ₁ ₂	TET	110 good 011 imperf sectile	Col s, gray	H 1.5, G 6.8-7.15, volat	---	
1.974	1.984	.010	POWELLITE (Scheelite ser) CaMoO ₄	TET pyram	Uneven to conch	White	H 4, G 4.23, F 5	Diss by HCl. Data for synth. CaMoO ₄ . Fluor yellow in UV.	
1.993 2.008	2.001	2.098	.097	CASSITERITE SnO ₂	TET pyram	100 imperf fr subconch to uneven	Brown, black, white	H 6-7, G 6.98- 7.02, infus	Insol in acids. Tw 011 poly. Sometimes dichroic in brown.
2.013Na 1.990Li	2.029Na 2.005Li	.016 .015	ZINCITE ZnO	HEX	1010 good 1120, 0001 poor	Deep red	H 4, G 5.6, infus	Diss by acids. In sec- tion deep red, not pleoc. MnO 0.27%.	
2.01 2.05	2.03	---	high	VOLTZITE ZnS + Zn-salt of organic acid	HEX spherical globules	---	Yellow, brown, black	H 4-4.5, G 3.1- 3.7, infus	Diss by acids, leaving an organic residue.
>2.0	>>1.74	---	high	BEHIERITE (Ta, Nb)BO ₄	TET pyram	110, 010 dist	Grayish- pink	H 7-7.5, G 7.86, infus	Insol in acids. Mean \underline{n} calc = 2.12.
2.10	2.26	.16	SCHAFAZIKITE FeSb ₂ O ₄	TET pyram	110 perf 100 good	Red to reddish- brown	H 3.5, G 5.5 calc	Pleoc, 0 straw yellow, E brownish yellow. Mean \underline{n} calc = 2.10.	
2.116	2.143	.027	CALIZIRITE CaZr ₃ TiO ₉	TET pris	Conch	Yellow-to dark-brown	H 6-7, G 5.07, infus	Partly diss by concd acids. Not pleoc.	
2.13	2.21	.08	PHOSGENITE Pb ₂ CO ₃ Cl ₂	TET pris tab	001, 110 dist	White, gray, yellow	H 2-3, G 6.13, F 1	Diss with eff by HNO ₃ . Pleoc wk, 0 reddish, E greenish.	
>2.1	---	---	PENFIELDITE Pb ₂ C ₁ ₃ (OH) ₆	HEX pris	001 dist	White	G 5.9-6.6, F 1	Diss by HNO ₃ . Tw.	
			PRIDERITE (K,Ba)(Ti,Fe) ₈ O ₁₆	TET pris	001 perf pris fair	Reddish- black	G 3.86, infus	Pleoc, 0 deep reddish- brown, E deep reddish- brown to black.	

2.15	2.275	.125	RANCIEITE $(\text{Ca}_3\text{Mn}^{+2})\text{Mn}^{+4} \cdot 4\text{O}_9 \cdot 3\text{H}_2\text{O}$	HEX (?) mass	---	Dark brown, streak purpleish	Soft. G 3.3	Diss by HCl. Opt char unk.
2.18	---	---	RUSSELLITE Bi_2W_6	TET mass	---	Yellow	H 3.5, G 7.37	Opt char unk.
2.18	2.19	.01	IODARGYRITE AgI	HEX thin plates	0001 perf sectile	Yellow, green, brown	H 1-1.5, G 5.6- 5.7, F 1	Becomes isotropic at 146°C. Abnormal green interf colors. Anom biax, 2V small. El neg.
2.19	2.21	.02	KLEINITE $\text{Hg}_2\text{N}(\text{Cl},\text{SO}_4) \cdot x\text{H}_2\text{O}$	HEX short prisms	0001 easy uneven, 1010 imperf	Yellow, orange	H 3.5-4, G ~ 8, volat	Diss by acids. Uniax above 130°C, biax neg at room temp.
2.19Li	2.21Li	.02	MACKAYITE $\text{FeTe}_2\text{O}_5(\text{OH})$	TET	Subconch	Green to yellow-green	H 4.5, G 4.86, fus	Pleoc, 0 green, E yellow.
>2.0	---	---	ORDONEZITE ZnSb_2O_6	TET tw	Conch	Pale to dark brown	H 6.5, G 6.64	Insol in acids. Tw pl (013). Mean n calc = 2.28.
2.21	---	wk	ZAVARITSKITE BiOF	TET	---	Gray	G (9.2)	Opt sign unk.
2.27	2.43	.16	PARATELLURITE TeO_2	TET pyram	102 perf 110 dist	Grayish- yellow	H 3, G 5.60, fus	---
2.27Li	2.42Li	.15	TAPIOLITE $\text{Fe}(\text{Ta},\text{Nb})_2\text{O}_6$	TET	---	Black	H 6, G 7.3-8.2, infus	Red-brown in section. Pleoc str, 0 yellow to reddish-brown, E nearly opaque.
2.35	2.27	2.36	CALZIRITE $\text{CaZr}_3\text{TiO}_9$	TET pris	Conch	Yellow- to dark-brown	H 6-7, G 5.0, infus	Partly diss by concd acids.
2.35	2.35	2.47	TAPIOLITE $\text{Fe}(\text{Ta},\text{Nb})_2\text{O}_6$	TET	---	Black	H 6-7, G 7.49, infus	---
2.374	2.36Na 2.33Li	2.38Na 2.35Li	WURTZITE $(\text{Zn},\text{Fe})\text{S}$	HEX pris	1120 good 0001, 1010 poor	Brownish- black	H 3.5-4, G 3.98, F diff	Diss by acids. Pleoc wk.
2.37Li	2.42Li	.05	FREUDENBERGITE $\text{Na}_2(\text{Ti},\text{Fe})_8\text{O}_{16}$	MCL ps hex	2 good	Brownish- black	G 4.3	Insol in acids. Pleoc, 0 dark brown, E yellow- brown.
2.36 2.39	2.374Li 2.39	2.395Li	WURTZITE $(\text{Zn},\text{Cd})\text{S}$	HEX pris	1120 good 0001 poor	Green	H 3.5-4, G 4.44, F diss	Diss by acids. Data for 41.5 mol % CdS. Fluor yellow in UV.
2.374 2.50	2.39Li	2.41Li	GREENOCKITE $(\text{Cd},\text{Zn})\text{S}$	HEX pris	1010 good 1120, 0001 poor	Orange	H 3-3.5, G 4.53, infus	Diss by acids. Data for 57.6 mol % CdS. Fluor orange-red in UV.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index ω	Birefringence ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.39 [~]	2.50Na 2.43Li	2.53Na 2.45Li	.024 .024	GREENOCKITE CdS	HEX pris	1010 good 1120, 0001 poor	Yellow, green	H 3-3.5, G 4.9- 5.0, infus
2.45	2.51Li	.06	DERBYLITE $\text{Fe}_4\text{Ti}_3\text{SbO}_{13}(\text{OH})$	MCL pris	Conch		H 5, G 4.53, infus	Insol in acids. Not pleoc.
2.476	2.485	.009	CHANGBAILITE PbNb_2O_6	TRIG	0001 perf		H 5-5.5, G 6.48	---
2.65 ^v	2.61Na 2.56Li	2.90Na 2.84Li	.287 .28	RUTILE TiO_2	TET pris	110 dist 100 less so	Brown, red, yellow	H 6-6.5, G 4.22, infus abs 0 < E. E1 pos.
2.61 [~]	2.65	2.80	.15	RUTILE, var Ilmenorutile (niobian) ($\text{Ti}, \text{Nb}, \text{Fe} \text{O}_2$)	TET pris	110 dist 100 less so	Black	H 6-6.5, G 4.36, infus
	2.65Na 2.63Li	2.69Na 2.67Li	.043 .040	MOISSANITE SiC	HEX plates	0001	Conch	Insol in acids. Pleoc, 0 light blue, E indigo blue. Data for synth compd.
	2.85Na 2.82Li	3.20Na 3.15Li	.347Na .327Li	CINNABAR HgS	TRIG		1010 perf	Streak scarlet. Disp very high. Circular polarization.
							Cochineal - red	H 2-2.5, G 8.09, volat

Table 5. Uniaxial negative minerals

Other entries	Refractive index ω	Biref ϵ	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.312	1.309	.003	MALLADRITE Na_2SiF_6	TRIG	---	Cols, rose	G 2.71 F 1	Sol in hot H_2O .
1.327	---	very wk	VILLIAUMITE NaF	CUB ps tet, u mass	001 perf	Carmine-red	H 3.5 G 2.79 F 1.5	Sol in H_2O . Pleoc., 0 carmine-red to pink, E yellow.
1.349	1.342	.007	CHIOLITE $\text{Na}_5\text{Al}_3\text{F}_{14}$	TET sq prisms	001 perf 011 dist	White	H 3.5-4 G 3.00 F 1.5	Diss by acids.
1.388	1.385	.003	COLQUHUIITE CaLiAlF_6	TRIG	Conch	Cols, white	H 4 G 2.94	---
1.406	1.309	.097	BARARITE $(\text{NH}_4)_2\text{SiF}_6$	HEX	0001 perf	White	H 2.5 G 2.15 volat	Sol in H_2O .
1.458	1.432	.026	MENDOZITE $\text{NaAl}(\text{SO}_4)_2 \cdot 11\text{H}_2\text{O}$	MCL fib	100 good	Cols	H 3 G 1.75 F 1	Sol in H_2O .
1.462	1.460	.002	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG rhombs	10T1 dist	White, red	H 4.5 G 1.97 F 3	Dec by HCl with sepn of slimy silica. Anom biax.
1.489	1.464	1.458	ETTRINGITE $\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12} \cdot 26\text{H}_2\text{O}$	HEX acic	10T0 perf	Cols, white	H 2-2.5 G 1.77 F 3	Diss by acids. Indices increase on dehydration.
1.491	1.436	.038	HUMBERTONITE $\text{K}_3\text{Na}_7\text{Mg}_2(\text{SO}_4)_6(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$	TRIG platy	0001	0001 perf	Cols	6 2.25
1.474	1.470	.004	HERSCHELITE (Zeolite grp) $(\text{Na}, \text{Ca}, \text{K})\text{AlSi}_2\text{O}_6 \cdot 3\text{H}_2\text{O}$	TRIG plates	---	White	H 4 G 2.06 F 3	Diss by H_2O . El pos.
1.485	1.475	.002	GMELINITE (Zeolite grp) $(\text{Na}, \text{Ca})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	HEX rhombs	10T0 good 0001	Cols, rose	H 4.5 G 2.04 F 3	Dec by acids. Anom biax with small 2V. Tw axis <u>C</u> .
1.465 1.484	1.408	.071	ZHEMCZUZHNIKOVITE $\text{NaMg}(\text{Al}, \text{Fe})(\text{C}_2\text{O}_4)_3 \cdot 8\text{H}_2\text{O}$	TRIG acic	0001 fair	Smoky green	H 2 G 1.69 infus	An oxalate. Sol in H_2O . Pleoc., 0 greenish-yellow, E reddish-violet.
1.481	1.461	.020	HANKSITE $\text{KNa}_{22}(\text{SO}_4)_9(\text{CO}_3)_2\text{Cl}$	HEX short prisms	0001 good	White	H 3-3.5 G 2.56 F 1.5	Easily sol in H_2O .

1.487	1.484	.003	CRISTOBALITE SiO ₂	TET ps cub, oct	---	Col's	H 6-7 G 2.33 infus	
1.479 1.493	1.487	.001	ANALCIME (Zeolite grp) NaAlSi ₂ ₀ ⁺ H ₂ ⁰	CUB, 211	001 poor	Col's, white	H 5 G 2.25 F 3.5	Dec by acids. Commonly biax. Lam tw on 001 and 110.
1.495	1.489	.003	OFFRETTITE (Zeolite grp) (K ₂ ,Ca)Al ₁₀ Si ₂₆ O ₇₂ •30H ₂ O	HEX pris	0001	Col's, white	G 2.13 F 3	Dec by acids. El neg.
1.462 1.518	1.489	.003	CHABAZITE (Zeolite grp) (Ca,Na)Al ₂ Si ₄ O ₁₂ •6H ₂ O	TRIG rhombs	10T1 dist	White, red	H 4.5 G 2.06 F 3	Dec by HCl with sepn of slimy silica. Anom biax.
1.510	1.489	.002	LEVYNE (Zeolite grp) CaAl ₂ Si ₄ O ₁₂ •6H ₂ O	TRIG	0221 dist	White	H 4 G 2.14 F 2-2.5	Gel with acids.
1.490	1.471	.019	LOEWEITE Na ₁₂ Mg ₇ (SO ₄) ₁₃ •15H ₂ O	TRIG	Conch	White, yellow	H 2.5-3 G 2.40 F 1.5	Sol in H ₂ O. Opt anom.
1.490	1.476	.014	WARDSMITHITE Ca ₅ MgB ₂ O ₄₂ •30H ₂ O	HEX platy	10T0 perf 0001 (?) good	White	H 2.5 G 1.88 F easy	Diss by acids.
1.491	1.470	.021	ETTRINGITE Ca ₆ Al ₂ (SO ₄) ₃ (OH) ₁₂ •26H ₂ O	HEX acidic	---	Col's, white	H 2-2.5 G 1.77 F 3	Diss by acids. Indices increase on dehydration.
1.492	1.475	.017	CHARLESITE Ca ₆ (Al, Si) ₂ (SO ₄) ₂ B(OH, O) ₁₆ • 26H ₂ O	HEX	10T0 perf	Col's	H 2.5 G 1.77	---
1.493	1.482	.011	WERMLANDITE	HEX	0001 perf	Greenish- gray	H 1.5 G 1.93	Diss by HCl. In part biax, 2V 3-5°.
1.494	1.467	.027	Ca ₂ Mg ₁₄ {Fe ⁺³ , Al} ₄ (CO ₃) ₄ (OH) ₄₂ • 14-15H ₂ O	HEX fib columnar	---	White	H 3.5 G 1.87 infus	Dec by acids with eff.
1.494	1.489	.005	THAUMASITE Ca ₃ Si(OH) ₆ (CO ₃)(SO ₄)•12H ₂ O	TET fib	---	White	G 2.19	El neg.
1.495	1.460	.035	GOBBINSITE (Zeolite grp) Na ₄ (Ca, K) ₂ Al ₆ Si ₁₀ O ₃₂ •12H ₂ O	TRIG	Conch	Col's to grey	H 1-1.5 G 1.17 F easy	Diss by H ₂ O.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.490 V 1.515	1.499	1.493	.006	VISHNEVITE (Cancrinite grp) $\text{Na}_6\text{Ca}_2\text{Al}_6\text{Si}_{12}^{0-24}$ $(\text{SO}_4, \text{CO}_3, \text{Cl})_{2-4}$	HEX	1010 perf	Col s	H 5 G 2.40 F 2	Gel with acids. SO_3 4.7, CO_2 2.2%. Some samples are isotropic, some unax pos.
1.502	1.449	.053		UNGEMACHITE $\text{K}_3\text{Na}_8\text{Fe}^{+3}(\text{SO}_4)_6(\text{OH})_2 \cdot 10\text{H}_2\text{O}$	TRIG	0001 perf	Col s, pale yellow	H 2.5 G 2.29	Diss by acids. El cl v pos.
1.504	1.49	.01		NYERERITE $(\text{Na}, \text{K})_2\text{Ca}(\text{CO}_3)_2$	ORTH ps hex, tab	---	---	G 2.41-2.54 fus	Diss by acids with eff. Tw poly.
1.505	1.486	.019		SACROFANITE (Cancrinite grp) $(\text{Na}, \text{Ca}, \text{K})(\text{Si}, \text{Al})_{12}^{0-24}$ $[(\text{OH})_2, \text{SO}_4, \text{CO}_3, \text{Cl}]_{2-3} \cdot x\text{H}_2\text{O}$	HEX	0001, 0110 perf	Col s	H 5.5-6 G 2.42	---
1.506	1.499	.007		MAZZITTE (Zeolite grp) $\text{K}_2\text{CaMg}_2(\text{Al}, \text{Si})_{36}^{0-72} \cdot 28\text{H}_2\text{O}$	HEX acidic	---	Col s	G 2.11	---
1.507	1.464	.043		M CALLISTERITE $\text{Mg}_2\text{B}_{12}^{14}(\text{OH})_{12} \cdot 9\text{H}_2\text{O}$	TRIG	0001, 0112 good	Col s	H 2.5 G 1.87	Diss by acids.
1.494 A	1.507	1.468	.039	THAUMASITE $\text{Ca}_3\text{Si}(\text{OH})_6(\text{CO}_3)(\text{SO}_4) \cdot 12\text{H}_2\text{O}$	HEX fib columnar	---	White	H 3.5 G 1.87 infus	Dec by acids with eff. El neg.
1.507	1.502	.005		TARANAKITE $\text{KA}_1\text{Al}_3(\text{PO}_4)_3(\text{OH}) \cdot 9\text{H}_2\text{O}$	TRIG u mass	---	White	Soft G 2.09-2.15 F easy	Diss by acids.
1.507 V	1.505	.002		GARRONITE (Zeolite grp) $\text{Na}_2\text{Ca}_5\text{Al}_{12}\text{Si}_{20}^{0-64} \cdot 27\text{H}_2\text{O}$	ORTH ps tet radiating	2 at 90°	Col s	G 2.15	---
1.510	1.486	.024		RETGERSITE $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$	TET tab	001 perf	Emerald-green	H 2.5 G 2.05	Sol in H_2O .
1.510 V	1.486	.024		FLUOBORITE $\text{Mg}_3\text{BO}_3(\text{F}, \text{OH})_3$	HEX fib	0001 indist	White	H 3.5 G 2.96	Diff diss by acids. near F end member.
1.489 A	1.510	1.502	.008	LEVYNE (Zeolite grp) $\text{CaAl}_2\text{Si}_{12}^{0-6}\text{H}_2\text{O}$	TRIG	0221 dist	White	H 4 G 2.14 F 2-2.5	Gel with acids.

1.52	1.511	1.495	.016	HYDROTALCITE $Mg_6Al_2CO_3(OH)_16 \cdot 4H_2O$	TRIG plates	0001 perf	White, luster pearly	H 2 G 2.06 infus.	Diss by acids. El pos. Anom biax.
	1.513	1.470	.043	ARCHERITE $(K, NH_4)H_2PO_4$	TET	---	Col s	Soft G 2.23	Sol in H_2O .
	1.515	1.417	.098	STEPANOVITE $NaMgFe^{+3}(C_2O_4)_3 \cdot 9H_2O$	TRIG	----	Yellow-green	H 3 G 1.61	An oxalate. Diss by H_2O . Pleoc, E col s, 0 yellow-green.
	1.499	1.515	.019	CANCINITITE (Cancrinite grp) $Na_6Ca_2Al_6Si_6O_{24}(CO_3)_2$	HEX	10T0 perf	Col s	H 5 G 2.42 F 2	Gel with acids. CO_2 7.0, SO_3 1.4, Cl 0.4%.
	1.518 pos V	1.528				2 at 90°	Col s	G 2.15	----
	1.507	1.515	.003	GARRONITE (Zeolite grp) $Na_2Ca_5Al_{12}Si_{20}O_{64} \cdot 27H_2O$	ORTH ps tet radiating	001 mic	White, brown, yellow	H 1.5 G 2.6 infus	2V 0-16°. El clv pos. Indices increase on standing in certain oils.
	1.516	1.470	.046	BEIDELLIITE (Smectite grp) $(Na, Ca)_0.33Al_2(Si, Al)_4O_{10}$ $(OH)_2 \cdot xH_2O$	MCL u mass	Col s			
	1.533								
	1.517	~1.501	~.016	MELIXNERITE $Mg_6Al_2(OH)_18 \cdot 4H_2O$	TRIG	0001 perf	Col s	6 1.9 infus	Diss by acids.
	1.489	1.518	1.515	CHABAZITE (Zeolite grp) $(Ca, Na)Al_2Si_4O_{12} \cdot 6H_2O$	TRIG rhombs	10T1 dist	White, red	H 4.5 G 2.18 F 3	Dec by HCl with sepn of slimy silica. Anom biax.
	1.519	1.503	.016	TEEPLEITE $Na_2B(OH)_4Cl$	TET tab	Uneven	Col s	H 3-3.5 G 2.08 F 1	Diss by H_2O .
	1.52	1.51	.01	MOUNTKEITHITE $(Mg, Ni)_{11}(Fe^{+3}, Cr, Al)_3(OH)_{24}$ $(SO_4, CO_3)_{3.5} \cdot 11H_2O$	HEX	0001 perf	Pale pink to white	H soft G 2.12	Diss by acids. Pleoc pale pink to col s.
	1.520	1.512	.008	TACHYHYDRITE $CaMg_2Cl_6 \cdot 12H_2O$	TRIG	10T1 perf	Max-to honey-yellow	H 2 G 1.67 F 1	Easily sol in H_2O . Deliq.
	1.521	1.517	.004	CARLETONITE $KNa_4Ca_4Si_8O_{18}(CO_3)_4(F, OH)_2$	TET	001 perf 110 good	Pink to blue	H 4-4.5 G 2.45	Blue var pleoc, 0 pale blue, E pale pinkish-brown.
	1.524	1.510	.014	MANASESITE $Mg_6Al_2CO_3(OH)_16 \cdot 4H_2O$	HEX foliated	0001 perf	White	H 2 G 2.05 infus	Diss by acids. El clv pos.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.520 (mean η)	.005-.006	VIRGILITE $\text{Li}_x\text{Al}_x\text{Si}_{3-x}\text{O}_6$ ($x = 0.5$ to 1.0)	HEX	ORTH (?) ps hex	One perf 001 (?)	White	H 2.5	Poly tw. In part biax. Nyerereite (?), $\omega = 1.504$.	
1.525	1.459	NATROFAIRCHILDITE $\text{Na}_2\text{Ca}(\text{CO}_3)_2$	TET	---	White to dark brown	G 2.04 volat	Diss in H_2O .		
1.525	1.480	BIPHOSPHAMMITE $(\text{NH}_4, \text{K})_2\text{P}_2\text{O}_4$	TRIG tab	---	Yellow-green	G 1.95	Diss by acids. Pleoc, 0 lemon-yellow, E pale yellow to cols.		
1.526	1.495	SLAVIKITE $\text{NaMg}_2\text{Fe}_5(\text{SO}_4)_7(\text{OH})_6 \cdot 33\text{H}_2\text{O}$	HEX	10TO perf	Col s	H 5 G 2.48 F 2	Gel with acids. El clv neg. CO_2 6.2, SO_3 0.2, Cl 0.1%.		
1.537	1.528	CANCINITITE (Cancrinite grp) $\text{Na}_6\text{Ca}_2\text{Al}_6\text{Si}_6\text{O}_{24}(\text{CO}_3)_2$	HEX fib	0001 indist	White	H 3.5 G 2.95	Diff diss by acids. F 20.9, H_2O 3.2%.		
1.515 ^	1.503	FLUOBORITE $\text{Mg}_3\text{BO}_3(\text{F}, \text{OH})_3$	HEX flat prisms	---	Col s	H 5 G 2.56	S0 ₃ 8.7, Cl 2.6, CO ₂ 2.1%		
88 1.547	1.530 1.507	LIOTITE (Cancrinite grp) $(\text{Ca}, \text{Na}, \text{K})_8(\text{Si}, \text{Al})_{12-24}(\text{SO}_4, \text{CO}_3, \text{Cl}, \text{OH})_4 \cdot \text{H}_2\text{O}$	HEX	10TO dist 0001 imperf	Col s	H 6 G 2.58 F 3.5	Gel with acids. Luster greasy.		
1.528 ^	1.530	NEPHELINE (Na, K)AlSi ₄ O ₈	HEX	10TO dist 110 less so	Col s	H 6 G 2.55 F 3	Insol in acids. Na ₂ O 7.1, CaO 6.3, K ₂ O 4.3, Cl 1.0, CO ₂ 2.0%.		
1.531	1.527	MARIALITE (Scapolite grp) $3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf	Col s	H 6 G 2.60 F 3.5	Gel with acids.		
1.539	1.532	KALIOPHILITE $\text{KA}_1\text{Si}_3\text{O}_4$	HEX pris	10TO dist	Col s	H 6 G 2.60 F 3.5	Insol in acids. Basal section shows 6 biax segments.		
1.546	1.532	MILARITE (Osumilite grp) $\text{K}_2\text{Ca}_4\text{Al}_2\text{Be}_4\text{Si}_{24-60} \cdot \text{H}_2\text{O}$	HEX	0001 indist	Col s	H 5-5.5 G 2.57 F 3	Diss by acids with eff.		
1.530 1.536	1.532	FAIRCHILDITE $\text{K}_2\text{Ca}(\text{CO}_3)_2$	HEX plates	0001 good	Col s	H 2.45 F 2	El pos.		
1.553	1.498								

1.516 1.549	1.533	1.502	.031	BEIDELITE (Smectite grp) $\text{Na}_0.33(\text{Al},\text{Fe}^{+3})_2(\text{Al},\text{Si})_4\text{H}_{10}$ $(\text{OH})_2 \cdot \text{xH}_2\text{O}$	MCL u mass	001 perf	White, brown, yellow	H 1.5 G 2.6 infus	
1.534	---	---	---	STRAETLINGITE $\text{Ca}_2\text{Al}_2\text{Si}_10_7 \cdot 8\text{H}_2\text{O}$	TRIG	001 perf	Cols to light green	G 1.9	
1.534	1.514	.020	ZINCALUMINITE $\text{Zn}_6\text{Al}_6(\text{SO}_4)_2(\text{OH})_{26} \cdot 5\text{H}_2\text{O}$	HEX or ps hex, plates	---	White	H 2.5-3 G 2.26 infus	Diss by acids. In part biax with small 2V.	
1.535	1.531	.004	KALSILITE (K, Na)(Al, Fe)SiO ₄	HEX	---	Col s	G 2.65 F 3.5	Gel with acids.	
1.535	1.537	0.490	SCHROECKINGERITE $\text{NaCa}_3(\text{UO}_2)(\text{CO}_3)_3(\text{SO}_4)_2\text{F} \cdot 10\text{H}_2\text{O}$	TCL ps hex	001	Greenish- yellow	H 2.5 G 2.54	Diss by H ₂ O. 2V 0-18°. Fluor yellow-green in UV.	
1.526 ^	1.537	1.498	.039	SLAVIKITE $\text{NaMg}_2\text{Fe}_5(\text{SO}_4)_7(\text{OH})_6 \cdot 33\text{H}_2\text{O}$	TRIG tab	---	Yellow-green	G 1.45	Diss by acids. Pleoc, 0 lemon-yellow, E pale yellow to col s.
1.538	1.535	.003	INDIALITE $\text{Mg}_2\text{Al}_4\text{Si}_5\text{O}_18$	HEX	---	Col s to blue	H 6-7 G 2.6 infus	Insol in acids. Dimorph with Cordierite.	
89	1.531 1.546	1.536	.003	NEPHELINE (Na, K)AlSiO ₄	HEX	10T0 dist 0001 imperf	Col s	H 6 G 2.60 F 3.5	Gel with acids. Luster greasy. K ₂ O 5.7, CaO 1.45%.
1.562	1.540	1.510	.030	BRUGNATELLITE $\text{Mg}_6\text{FeCO}_3(\text{OH})_{13} \cdot 4\text{H}_2\text{O}$	HEX tab	0001 mic	White, brownish	H 2 G 2.14 infus	Diss by acids. El pos. Pleoc, 0 yellow-red, E cols.
1.540	1.512	.028	MELLITE $\text{Al}_2[\text{C}_6(\text{COO})_6] \cdot 16\text{H}_2\text{O}$	TET	011 indist	Honey- yellow, white	H 2-2.5 G 1.64	Diss by HNO ₃ , dec by hot H ₂ O. In part biax with small 2V. Pleoc wk, E cols, 0 pale yellow.	
1.540	1.535	.005	TETRAKASILITE (K, Na)AlSiO ₄	HEX	---	Col s	G 2.59 F 3.5	Gel with acids.	
1.540	1.517	.023	VERMICULITE $(\text{Mg},\text{Fe},\text{Al})_3(\text{Al},\text{Si})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	MCL ps hex	001 mic	Green, white	H 1.5 G 2.3	Dec by acids. Exfoliates when heated. El clv pos.	
1.542	1.533	.009	IONAITE $\text{Mg}_4\text{Fe}^{+3}(\text{OH})_8\text{OCl} \cdot 2\text{-}4\text{H}_2\text{O}$	HEX	0001 perf	Bluish- green	H 1.5 G 2.11	Diss by acids. Anom biax. El clv pos.	
1.537 ^	1.545	1.496	.049	SCHROECKINGERITE $\text{NaCa}_3(\text{UO}_2)(\text{CO}_3)_3(\text{SO}_4)_2\text{F} \cdot 10\text{H}_2\text{O}$	TCL ps hex	001	Greenish- yellow	H 2.5 G 2.55	Diss by H ₂ O. 2V 0-18°. Fluor yellow-green in UV.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.542 1.553	1.545	1.518	.027	STICHTITE $Mg_6Cr_2CO_3(OH)_{16} \cdot 4H_2O$	TRIG plates 0001	0001 perf	Lilac	H 1.5-2 G 2.16 infus	Diss by acids. Anom biax. El pos. Pleoc wk, abs 0 > E.
1.532 1.561	1.546	1.542	.004	MARIALITE (Scapolite grp) $3NaAlSi_3O_8 \cdot NaCl$	TET	100 perf 110 less so	Cols	H 6 G 2.62 F 3	Insol in acids. Na ₂ O 10.5, CaO 4.8, K ₂ O 1.2, Cl 3.0, CO ₂ 1.1% (Me19).
1.539 1.546	1.546	1.542	.004	NEPHELINE (Na, K)AlSiO ₄	HEX	10T0 dist 0001 imperf	Cols	H 6 G 2.65 F 3.5	Gel with acids. Luster greasy. K ₂ O 8.7, CaO 3.7%.
1.530 1.570	1.547	1.522	.025	FLUOBORITE $Mg_3BO_3(F, OH)_3$	HEX fib	0001 indist	White	H 3.5 G 2.92	Diff diss by acids. F 17.6%.
1.545 1.549	1.547	1.536	.011	GYROLITE $Ca_2Si_3O_7(OH)_2 \cdot H_2O$	HEX fib	0001 mic	White	H 3.4 G 2.39 F diff	Diss by acids. El clv pos.
1.533 1.585	1.549	1.517	.032	BEIDELITE (Smeectite grp) $Na_0.33(Al, Fe)_2(Al, Si)_4O_{10}$ (OH) ₂ • xH ₂ O	MCL u mass	001. mic	Brown, yellow	H 1.5 G 2.6 infus	2V small. El clv pos.
1.549	1.522	.027		TRUSCOTTITE $(Ca, Mn)_{14}Si_{24}O_{58}(OH)_8 \cdot 2H_2O$	HEX	One mic	White	G 2.47	Dec by HCl. Commonly biax, 2V small.
1.550	1.495	.055		ANTARCTICITE $CaCl_2 \cdot 6H_2O$	TRIG	0001 and pris, perf	Cols	G 1.7	Readily sol in H ₂ O.
1.550 1.570	1.522	.028		TAENIOLITE (Mica grp) KLiMg ₂ Si ₄ O ₁₀ F ₂	MCL ps hex	001 perf	Cols, yellowish	H 3 G 2.82 F 3	Nearly insol in acids. Biax, 2V small. El clv pos.
1.532 1.553	1.553	1.549	.004	MILARITE (Osumilite grp) $K_2Ca_2Al_2Be_4Si_{24}O_{60} \cdot H_2O$	HEX	0001 indist	Cols, pale green, pink	H 5.5-6 G 2.57 F 3	Insol in acids. Basal section may show 6 biax segments.
1.535 1.556	1.554	1.550	.004	KALSILITE (K, Na)(Al, Fe)SiO ₄	HEX	---	Cols	G 2.66 F 3.5	Gel with acids.
		1.540	.016	JOURAVSKITE $Ca_3Mn^{+4}(SO_4, CO_3)_2(OH)_6 \cdot 12H_2O$	HEX	Pris, good	Greenish-yellow, orange	H 2.5 G 1.95	In section bright yellow. pleoc wk. Abnormal interf colors.

1.557	1.529	.028	BARBERTONITE $Mg_6Cr_2CO_3(OH)_{16} \cdot 4H_2O$	HEX u mass	0001 perf	Lilac, rose-pink	H 1.5-2 G 2.10 infus.	Diss by acids.
1.592	1.559	---	wk to med	MERRIHEUITE (Osumilite grp) (K, Na) ₂ (Fe ⁺² , Mg) ₅ Si ₁₂ O ₃₀	HEX	---	G 2.87 calc	opt char unk. FeO 10.5%.
	1.559	1.557	.002	HOLTEDAHLITE Mg_2PO_4	HEX	---	H 4.5-5 G 2.94	---
	1.560	1.507	.053	CHLORALUMINITE $AlCl_3 \cdot 6H_2O$.	TRIG	---	Col s	Sol in H ₂ O. Deliq.
1.593	1.560	1.540	.020	SAPONITE (Smectite grp) (Na, Ca)0.33(Mg, Fe) ₃ (Si, Al) ₄	MCL ps hex	001 perf	Col s to yellowish	Ubiax, 2V small. El clv pos. Al ₂ O ₃ 7.2, Fe ₂ O ₃ 1.0%.
	1.560	1.540	.020	VERMICULITE (Mg, Fe, Al) ₃ (Al, Si) ₄ Al ₁₀ (OH) ₂ 4H ₂ O	MCL ps hex	001 mic	White, green	Dec by HCl. Exfoliates when heated. El clv pos. Fe ₂ O ₃ 4.2, FeO 1.8%.
	1.583	1.540	.020	PYROAURITE $Mg_6Fe_2CO_3(OH)_{16} \cdot 4H_2O$	TRIG tab 00001 fib	0001 mic	Green	H 1.5 G 2.3
	1.572	1.560	.015		TET	100 perf 110 less so	White, yellowish	Diss by acids. El pos. Pleoc, 0 yellow-red, E cols. Dimorph with Sjogrenite.
	1.546	1.561	.015	SCAPOLITE (Scapolite grp) 3NaAlSi ₃ O ₈ ·NaCl - 3CaAl ₂ Si ₂ O ₈ ·CaCO ₃	TET	100 perf 110 less so	Col s, brown	Insol in acids. Na ₂ O 8.6, CaO 8.3, K ₂ O 1.1, Cl 2.2, CO ₂ 1.7, SO ₃ 0.4% (Me ₃₃).
	1.574	1.545	.012	LOVOZERITE $Na_3ZrSi_6(O, OH)_{18}$	TRIG	Uneven to conch	Dark brown to black	Insol in acids. Tw.
	1.578	1.561	.012	BRUGNATELLITE $Mg_6(Fe,Cr)CO_3(OH)_{13} \cdot 4H_2O$	HEX tab 00001	0001 mic	Brown	H 6 G 2.63 F 3
	1.540	1.562	.019		MCL ps hex, platy	001 perf	Dark brown to black	Insol in acids. Cr ₂ O ₃ 4.9-6.2%.
	1.584	1.564	.029	PHLOGOPITE (Mica grp) K(Mg, Fe) ₃ AlSi ₃ O ₁₀ (F, OH) ₂	TRIG	0001 perf conch	Yellow brown	Insol in acids. 2V 0-10°. El pos.
	1.565	1.535	.030	ZAKHAROVITE $Na_4(Mn, Fe)_5Si_{10}O_{24}(OH)_6 \cdot 5H_2O$	TCL ps trig	001 perf	Yellow	Strongly electromagnetic.
	1.565	1.560	.005	ZEOPHYLLITE $Ca_4Si_{3.8}(OH, F)_4 \cdot 2H_2O$			Col s, white	Gel with acids. Biax, 2V 0-27°. El clv pos.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.567	1.566	.001		BRANNOCKITE (Osumilite grp) KL <i>i</i> ₃ SnSi ₁₂ O ₃₀	HEX	---	---	6.2.98	Fluor bluish white in UV. K ₂ 3.7-4.5, Li ₂ 3.75, SnO ₂ 28.2%.
1.568	1.563	.005		REYERITE (Na,K) ₄ Ca ₁₄ (Si,Al) ₂₄ O ₆₀ (OH) ₅ •5H ₂ O	trig radiating	0001 perf	Col s	6.2.55	Dec by HCl. El clv pos.
1.568	1.564	.004		BERYL Be ₃ Al ₂ Si ₆ O ₁₈	HEX pris	0001 imperf	Green, blue, col s	H 8 6.2.66 infus	Insol in acids. Pleoc variable. El neg. Data for nearly alkali-free mineral.
1.569	1.547	.022		DESAUTELITE Mg ₆ Mn ₂ ⁺³ (CO ₃) ₂ (OH) ₁₆ •4H ₂ O	TRIG tab	0001 perf	Orange	6.2.13	Diss by acids.
1.569	1.549	.020	wk	KALISTRONTITE K ₂ Sr(SO ₄) ₂	TRIG	0001 perf	Col s	H 2 6.3.20	Dec by dil acids.
1.57(?)	---			LAWRENCE (Fe,Ni)Cl ₂	HEX tab	0001 imperf	Green, brown	6.3.16	A meteorite mineral. Diss in H ₂ O. Alters readily. El pos.
1.547	1.570	.036		FLUOBORITE Mg ₃ BO ₃ (F,OH) ₂	HEX fib	0001 indist	White	H 3.5 6.2.8	Diff diss by acids. F 14.8%.
1.550	1.570	.030		TAENIOLITE (Mica grp) KL <i>i</i> Mg ₂ Si ₄ O ₁₀ F ₂	MCL ps hex	001 perf	Brown	H 3 6.2.82 F 3	Nearly insol in acids. Biax, 2V ~ 5°. El clv pos. TiO ₂ 2.0, FeO 1.9%.
1.570	1.564	.006		CESANITE Ca ₂ Na ₃ (SO ₄) ₃ (OH)	HEX	---	Col s	H 2.3 6.2.79	---
1.560	1.572	1.549	.023	PYROAURITE Mg ₆ Fe ₂ CO ₃ (OH) ₁₆ •4H ₂ O	TRIG tab	0001 mic	White, yellow	H 2.5 6.2.12 infus	Diss by acids. El pos. Pleoc, 0 yellow-red, E cols. Dimorph with Sjögrenite.
1.573	1.572	.001		STEACYITE Th(Ca,Na) ₂ K _{1-x} Si ₈ O ₂₀ (x = 0.2-0.4)	TET	---	Dark brown	H 5 G (3.02)	---
1.573	1.550	.023		SØÐGREENITE Mg ₆ Fe ₂ CO ₃ (OH) ₁₆ •4H ₂ O	HEX plates	0001 perf	Yellowish	H 2.5 6.2.11 infus	Diss by acids. El pos. Dimorph with Pyroaurite.

1.609 Y	1.573	1.572	.001	EKANITE (Th,U)(Ca,Fe,Pb) ₂ Si ₈ H ₂₀	TET	---	Cols, green, brown	G 3.3	---
1.574	1.547	.027		PORLANDITE Ca(OH) ₂	HEX	0001 perf	Cols	H 2-2.3 G 2.23 infus	Diss by acids. El cl v pos. Luster pearly.
1.561 Y 1.587	1.574	1.549	.025	SCAPOLITE (Scapolite grp) 3NaAlSi ₃ O ₈ ·NaCl - 3CaAl ₂ Si ₂ O ₈ ·CaCO ₃	TET	100 perf 110 less so	Cols, brown	H 5-6 G 2.68 F 4	Insol in acids. Na ₂ O 5.4, CaO 12.5, K ₂ O 1.0, CO ₂ 2.9, Cl 1.4%.
1.574	1.552	.022		Unnamed P analogue of Troegerite (Autunite grp) (UO ₂) ₃ (PO ₄) ₂ ·8H ₂ O	TET	001 perf	Yellow to greenish	H 3 G 3.76	Diss by acids. El cl v pos.
1.571 -	1.574	1.559	.015	SALEITE (Autunite grp) Mg(UO ₂) ₂ (PO ₄) ₂ ·10H ₂ O	MCL ps tet tab	001 perf	Yellow	H 2-3 G 3.27	Diss by acids. El pos. Pleoc, 0 pale yellow, E cols. Fluor yellow in UV. Blaix in part.
1.575 -	1.575	1.57	.005	CALCIOTOFFRITE Ca ₂ Fe ₂ (PO ₄) ₃ (OH)·7H ₂ O	MCL	0001 perf	Yellow-green	H 2.5 G 2.53 F easy	Diss by HCl. El cl v pos.
1.575 Y	1.575	1.572	.003	AUTUNITE (Autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ ·10-12H ₂ O	TET tab	001 perf 100 indist	Yellow to yellow-green	H 2-2.5 G 3.2 F 3	Diss by acids. Fluor yellow in UV. U biax, 2V 0-30°. El pos.
1.576 93	1.576	1.546	.030	PARSETTENSITE (K,Na,Ca)(Mn,Al) ₇ Si ₈ H ₂₀	MCL ps hex	001	Copper-red	G 2.59-2.68 fus	Dec by hot HCl. Pleoc, 0 greenish-yellow, E cols.
1.568 Y 1.584	1.576	1.573	.003	BERYL Be ₃ Al ₂ Si ₆ O ₁₈	HEX pris	0001 imperf	Pale green	H 8 G 2.71 infus	Insol in acids. El reg. Pleoc variable. Fe ₂ O ₃ 0.7, total alkalies 0.7%.
1.561 Y	1.578	1.538	.040	LOVOZERITE Na ₃ ZrSi ₆ (O,OH) ₁₈	TRIG	Uneven to conch	Dark brown to black	H 5 G 2.63 F 2	Insol in acids. Tw.
1.578 Y	1.548	.030		NOVACEKITE (Autunite grp) Mg(UO ₂) ₂ (AsO ₄) ₂ ·12H ₂ O	TET tab	001 perf 100 dist	Yellow	H 2-2.5 G 3.23	Diss by acids. Fluor yellow in UV. Commonly biax, 2V 0-40°.
1.578	1.559	.019		SODIUM AUTUNITE (Autunite grp) Na ₂ (UO ₂) ₂ (PO ₄) ₂ ·8H ₂ O	TET plates	001 perf 100 less so	Lemon-yellow	H 2-2.5 G 3.58	Diss by acids. El cl v pos. Pleoc, 0 light yellow, E pale yellow. Fluor yellow- green in UV.
1.578	1.577	.001		ARCTITE Na ₂ Ca ₄ (PO ₄) ₃ F	TRIG	001 perf	Cols	H 5 G 3.13	Diss by acids.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.579	1.577	.002	CLINOCHLORITE, var Penninite (Chlorite grp) (Mg,Fe)Al ₅ Al ₁₀ (OH) ₈	MCL ps trig. plates	001 perf	Green, brown, cols	H 2 G 2.7 F diff	Dec by H ₂ SO ₄ . Biax, 2V small. El clv pos. Pleoc. 0 green, E nearly col. Abnormal blue interf. colors.	
---	1.577	very low	NAMUWITE (Zn,Cu) ₄ (SO ₄)(OH) ₆ ·4H ₂ O	HEX	001 perf	Pale sea- green	G 2.77	opt sign unk.	
1.580	1.575	.005	DARAPIOSITE (Osumilite grp) KNa ₂ Li(Mn,Zn)ZrSi ₁₂ Si ₃₀	HEX	---	---	H 5 G 2.92	Na ₂ O 3.0, K ₂ O 5.1, ZrO ₂ 5.0, MnO 8.2, ZnO 7.8%.	
1.580	1.485	.095	BERBORITE Be ₂ (BO ₃)(OH,F)·H ₂ O	TRIG	0001 perf	cols	H 3 G 2.20 infus	Insol in acids. El clv pos.	
1.580	1.550	.030	ZINNWADITE (Mica grp) KLiFe ⁺² Al ₃ Si ₃ Al ₁₀ (F,OH) ₂	MCL ps trig	001 perf	Brown	H 2.5-4 G 3.01 F 2.5	Insol in acids. 2V 0-30°.	
1.581	---	---	META-ANKOLEITE (Meta-autunite grp) (K ₂ ,Ba,Ca)(UO ₂) ₂ (PO ₄) ₂ ·6H ₂ O	TET thin plates	001 perf 100 dist	Yellow	G 3.54	Diss by acids. El clv pos. Fluor yellow-green in UV.	
1.582	1.564	.018	SABUGALITE (Autunite grp) HA1(UO ₂) ₄ (PO ₄) ₄ ·16H ₂ O	TET plates	001 perf	Yellow	H 2.5 G 3.20	Diss by acids. El clv pos. Fluor bright yellow in UV. Commonly biax, 2V 0°-med.	
1.583	1.554	.029	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) ₁₀ Si ₁₂ O ₃₀ (0,OH) ₁₂	MCL and TCL ps hex	001 mic	Brown	H 3.5 G 2.74 F 3.5 2.1%	Diss by acids. El clv pos. Pleoc, E pale yellow, 0 green. FeO 26.7, Fe ₂ O ₃ 2.1%.	
1.583	1.583	1.564	VERMICULITE (Mg,Fe) ₃ (Al,Si) ₄ O ₁₀ (OH) ₂ 4H ₂ O	MCL ps hex	001 mic	Green	H 1.5 G 2.67	Dec by HCl. Exfoliates when heated. El clv pos. Fe ₂ O ₃ 6.2, FeO 2.6, MgO 27.3%.	
1.584	1.545	.039	TALC Mg ₃ Si ₄ O ₁₀ (OH,F) ₂	TCL ps hex	001 perf	White to green	H 1 G 2.7 infus	Tric. analogue of monoclinic Talc. 2V 0-10°. Pearly luster, greasy feel. El clv pos.	
1.599	1.599	1.555	PHLOGOPITE (Biotite ser., Mica grp) K(Mg,Fe) ₃ Si ₃ Al ₁₀ (F,OH) ₂	MCL ps hex, plates	001 perf	Brown	H 2-2.5 G 2.86 F diff	Insol in acids. 2V 0-30°. El pos. FeO 4.2, Fe ₂ O ₃ 3.7%.	
1.584	1.584	1.598							

1.576 ^ 1.595	1.584	1.577	.007	BERYL $\text{Be}_3\text{Al}_2\text{Si}_6\text{O}_{18}$	MCL u mass	001 mic	Brown	H 8 G 2.72 infus	Insol in acids. El neg. Pleoc variable. Data for mineral with 1-2% alkalis.
1.549 ^	1.585	1.559	.026	BEDELLITE (Smectite grp) $\text{Na}_0.33(\text{Al},\text{Fe})_2(\text{Al},\text{Si})_4\text{O}_{10}$ $(\text{OH})_2 \cdot x\text{H}_2\text{O}$	MCL ps hex	001 perf	Cols, blue, green	H 1.5 G 2.7 infus	Biax, 2V 0-30°. El clv pos.
1.600 ^	1.586	1.559	.027	GLAUCONITE, var Skolite (Mica grp) $(\text{K},\text{Na})(\text{Al},\text{Fe},\text{Mg})_2(\text{Si},\text{Al})_4\text{O}_{10}$ $(\text{OH})_2$	TET	001 perf	Yellow	H 2-3 G 2.6 fus	Dec by acids. 2V small. El clv pos. Fe_2O_3 6.4, Al_2O_3 18.2, FeO 2.6%.
1.586 ^	1.586	1.560	.026	URANOSPINITE (Autunite grp) $\text{Ca}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 10\text{H}_2\text{O}$	TET	001 perf	Yellow to yellow-green	H 2-3.5 G 3.45 fus	Diss by acids. Commonly biax. El clv pos. Pleoc, 0 pale yellow, E col. S. Fluor bright yellow in UV.
1.575 ^	1.586	1.578	.008	AUTUNITE (Autunite grp) $\text{Ca}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 10\text{H}_2\text{O}$	TET tab	001 perf	Yellow to yellow-green	H 2-2.5 G 3.2 F 3	Diss by acids. Commonly biax, 2V 0-30°. El clv pos. Fluor yellow in UV.
1.587	1.336	.251		NITRATITE (Soda niter) NaNO_3	TRIG	10T1 perf	White	H 1.5-2 G 2.27 F 1	Sol in H_2O . Taste cooling. Deflagrates when heated.
1.587 ^ 1.607	1.587	1.555	.032	MELONITE (Scapolite grp) $3\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot \text{CaCO}_3$ - $3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf 110 less so	Cols to brown	H 5-6 G 2.70 F 4-5	Insol in acids. CaO 15.8, Na_2O 2.4, K_2O 2.4, SO_3 0.9, CO_2 2.9% (Me70).
1.587	1.570	.017		SHAFRANOVSKITE $(\text{Na},\text{K})_6(\text{Mn},\text{Fe})_3\text{Si}_9\text{O}_{24} \cdot 6\text{H}_2\text{O}$	TRIG	0001 perf	Dark green	H 2-3 G 2.77	Magnetic.
1.588	1.560	.028		COALINGITE $\text{Mg}_{10}\text{Fe}_2\text{CO}_3(\text{OH})_{24} \cdot 2\text{H}_2\text{O}$	TRIG plates	0001	Reddish- brown	H 2.33 G infus	Diss by acids. Pleoc, 0 golden-brown, E col. Anom biax, 2V 5-20°. El clv pos.
1.606 ^	1.589	1.583	.006	KAEMMERERITE (Chlorite grp) $(\text{Mg},\text{Al},\text{Cr})_6(\text{Si}_3\text{Al})_{10}(\text{OH})_8$	MCL ps hex	001 perf	Rose, lilac	H 2.5 G 2.72 F diff	Insol in HCl. El clv pos. FeO 2.0, Cr_2O_3 7.9%. 0-8°.
1.594 ^	1.590	1.543	.047	MONOHYDROCALCITE $\text{CaCO}_3 \cdot \text{H}_2\text{O}$	HEX	---	White	Infus	Diss by acids with eff.
1.59	1.56	.03		"CONNARITE" $(\text{Mg},\text{Ni})_3\text{Si}_2\text{O}_5(\text{OH})_4$ (?)	MCL ps hex (?)	001 perf	Yellow-green	H 2.5-3 G 2.5 infus	Pleoc wk. El clv pos (interlayered vermiculite- chlorite).
1.590	1.585	.005		IRAAQITE $(\text{K},\text{Ce},\text{La},\text{Th})(\text{Ca},\text{La},\text{Na})_2\text{Si}_8\text{O}_{20}$	TET	2 good fr uneven	Pale green- ish yellow	H 4.5 G 3.27	In part biax, 2V up to 7°. Compare Ekanite.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.588 1.596	1.591	1.573	.018	METAVOLTINE $K_2Na_6Fe^{+2}Fe^{+3}_2(SO_4)_{12}O_2\cdot 18H_2O$	HEX scales	0001 perf	Yellow-brown, greenish-brown	H 2.5 G 2.53 F 5	Diss by hot H_2O , diss by acids. El cleav pos. Pleoc., E pale green, 0 dark yellow, E pale yellow.
1.559	1.592	---	wk to med	MERRIHUEITE (0sumilite grp) $(K, Na)_2(Fe^{+2}, Mg)_5Si_{12}O_8$	HEX	---	---	---	Opt char unk. MgO 4.4, FeO 29.7%.
1.592	1.582	.010		TORBERNITE (Autunite grp) $Cu(UO_2)_2(Po_4)_2\cdot 10\cdot 12H_2O$	TET tab	001 perf 100 indist	Green	H 2-2.5 G 3.22 F 3	Diss by acids. El cleav pos. Pleoc., E pale green, 0 dark green to blue. Not fluor in UV.
1.560	1.593	1.573	.020	SAPONITE (Smectite grp) $Na_0.33(Mg, Fe, Fe^{+3})_3(Si, Al)_4\cdot 0_{10}(OH)_2\cdot 4H_2O$	MCL ps hex	001 perf	White, green	H 1-2 G 2.39 F 4	U biax, 2V small. FeO 19.9, FeO 5.3, Al 2.3 5.7%.
1.593	1.585	.008		MACHATSCHKITE $(Ca, Na)_6(AsO_4)_4(AsO_3OH)_3\cdot (PO_4, SO_4)\cdot 15H_2O$	TRIG	Conch	Colts	H 2-3 G 2.5-2.6	Diss by acids.
1.595	1.455	.140		BUETSCHLILITE $K_2Ca(CO_3)_2$	TRIG	---	Colts	G 2.60	Diss by acids with eff.
1.605 1.608	1.595	1.587	.008	BERYL $Cs_x[Al]_{2-x}Li_xBe_3Si_6O_{18}$	HEX pris	0001 imperf	Pink	H 8 G 2.79 infus	Insol in acids. El neg. Na 0 1.6, Li 0 1.7, (Cs, Rb, K) 0 1.7%.
1.595	1.589	.006		WENKITE $Ba_4Ca_6(Si, Al)_{20}O_{39}(OH)_2\cdot (SO_4)_3\cdot xH_2O$ (?)	HEX pris	Pris. poor	White, gray	H 6 G 3.19	In part biax, 2V 0-10°. Cancrinite grp?
1.597	1.570	.027		ABERNATHYITE (Meta-autunite grp) $K_2(UO_2)_2(AsO_4)_2\cdot 4H_2O$	TET tab	001 perf	Yellow	H 2-3 G 3.74	Diss by acids. El pos. Pleoc., 0 yellow, E pale yellow. Fluor yellow-green in UV. In part biax.
1.584 1.640	1.598	1.558	.040	BIOTITE (Mica grp) $K(Mg, Fe)_3(Al, Fe)Si_3O_{10}(OH, F)_2$	MCL ps hex, plates	001 mic	Dark brown to green	H 2-2.5 G 2.95 F 3-4	2V 0-13°. El pos. FeO 13.85, Fe ₂ O ₃ 2.1, MnO 0.4%.
1.598	1.598	<.001		COMBEITE $Na_2Ca_2Si_3O_9$	TRIG pris	---	Colts	G 2.84	Gel with hot HCl.

1.584 ^A	1.599	.045	TALC (Mg,Fe) ₃ Si ₄ O ₁₀ (OH) ₂	001 perf	Green	H 1 G 2.87 infus
1.586 ^V 1.618 ^A	1.600	.015	GLAUCONITE (Mica grp) (K,Na) ^{Al} ,Fe,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	MCL ps hex	001 perf	Green, blue, cols
1.611 ^V	1.600	.002	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ •6H ₂ O	TET tab	001 perf	Yellow to greenish-yellow
1.601	1.480	.121	GRIMSELIITE K ₂ Na(UO ₂)(CO ₃) ₃ •H ₂ O	HEX prism	Conch	Yellow
1.601	1.591	.010	DAVISONITE Ca ₃ Al(Po ₄) ₂ (OH) ₃ •H ₂ O	HEX fib, botryoidal crusts	0001 perf	cols, white
1.598 ^V 1.634 ^A	1.602	.008	TAKOVITE Ni ₆ Al ₂ (OH) ₁₆ (CO ₃ ,OH)•4H ₂ O	TRIG	---	Yellow-green to blue-green
1.583 ^V 1.634 ^A	1.603	.048	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₁₂	MCL and TCL ps hex	001 mic	Brown
97	1.622 ^V	.005	CARBONATE-HYDROXYLAPATITE (Apatite grp) Ca ₅ (PO ₄ ,CO ₃) ₃ (OH,F)	HEX u mass	---	Col, gray, brown
1.605	1.450	.155	EITELITE Na ₂ Mg(CO ₃) ₂	TRIG	0001	Col
1.595 ^A	1.605	.152	BUETSCHLILITE K ₂ Ca(CO ₃) ₂	TRIG	---	Col
1.605	1.574	.031	HEINRICHITE (Autunite grp) Ba(UO ₂) ₂ (AsO ₄) ₂ •10-12H ₂ O	TET	001 perf 100 dist	Yellow to green
1.606	1.608	.002	SOGDIANITE (Osumilite grp) (K,Na) ₂ Li ₂ (Li,Fe ⁺³ ,Al) ₂ Zr Si ₁₂ O ₃₀	HEX	---	Violet
1.588 ^V 1.634 ^A	1.606	.031	COALINGITE Mg ₁₀ Fe ₂ O ₃ (OH) ₂₄ •2H ₂ O	TRIG platy	0001	Reddish- brown

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula ^a	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.630	1.606	1.602	.004	EUDIALYTE $\text{Na}_4(\text{Ca},\text{Ce})_2(\text{Fe}^{+2},\text{Mn}^{+2})_2\text{ZrSi}_8\text{O}_{22}(\text{OH},\text{Cl})_2$	TRIG	0001 dist 1020 poor	Yellow, pink, brown	H 5-5.5 G 2.85 F 3	Gel with acids. Pleoc wk in pink and yellow. Anom biax. Also uniax pos.
1.583 1.648	1.607	1.560	.047	VERMICULITE $(\text{Mg},\text{Fe},\text{Al})_3(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	MCL ps hex	001 perf	Green, brown	H 1.5 G 2.77 F 5	Dec by HCl. El clv pos. Exfoliates when heated. Fe ₂ O ₃ 4.5%.
1.587	1.607	1.571	.036	MEIONITE (Scapolite grp) $3\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot \text{CaCO}_3 - 3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf 110 less so	Col to brown	H 5-6 G 2.7 F 4-5	Insol in acids (Me ₈₄).
[]	1.607	1.602	.005	JAMBORITE $(\text{Ni}^{+2},\text{Ni}^{+3},\text{Co},\text{Fe})(\text{OH})_2$ $(\text{OH},\text{S},\text{H}_2\text{O})$ (?)	HEX fib, platy	---	Green	G 2.67 infus	Slowly diss by cold HCl.
1.607	1.604	.003		WHITLOCKITE, carbonatian, var Martinite $\text{Ca}_9(\text{Mg},\text{Fe})_4(\text{PO}_4,\text{CO}_3)_7$	TRIG u mass	---	Col s	H 5 G 3.12 infus	Diss by acids.
98	1.623	1.608	.009	BERYL $\text{Cs}_x[\text{Al}_{2-x}(\text{Li},\text{Fe})_x]$ $(\text{Be}_{3-x}\text{Fe}_x)\text{Si}_6\text{O}_{18}$	HEX pris	0001 imperf	Blue-green	H 8 G 2.92 infus	Insol in acids. El neg. FeO 2.2, Fe ₂ O ₃ 2.0. Total alkalies 8.2%.
1.595	1.608	1.599		SOGDIANITE (Osunilite grp) ⁺³ $(\text{K},\text{Na})_2\text{Li}_2(\text{Li},\text{Fe},\text{Al})_2\text{ZrSi}_{12}\text{O}_{30}$	HEX	0001 perf	Violet	H 7 G 2.90	---
1.609	1.596	.013		HILAIRITE $\text{Na}_2\text{ZrSi}_3\text{O}_9 \cdot 3\text{H}_2\text{O}$	TRIG	Conch	Pale brown	H > 4 G 2.72	Slightly attacked by HCl at room temp.
1.573	1.609	1.604	.005	EKANITE $(\text{Th},\text{U})(\text{Ca},\text{Fe},\text{Pb})_2\text{Si}_8\text{O}_{20}$	TET	---	Green, brown, col s	G 3.3	---
1.610	1.605	.005		ZIRINALITE $\text{Na}_6(\text{Ca},\text{Mn},\text{Fe})\text{ZrSi}_6\text{O}_{18}$	TRIG	Uneven to conch	Col s	H 5 G 2.90 F 3	Dec by warm HCl.
1.610	1.607	.003		SUGILITE (Osunilite grp) ⁺³ $(\text{K},\text{Na})(\text{Na},\text{Fe})_2(\text{Li},\text{Fe})\text{Si}_12\text{O}_{30}$	HEX	0001 poor	Light brownish-yellow	H 6-6.5 G 2.74	---

1.611	1.592	.019	MELIPHANTITE (Ca, Na) ₂ Be(Si, Al) ₂ (O, F) ₇	TET pyram	001 perf	Yellow	H 5-6 G 3.01 fus	Insol in acids. Anom biax. El clv pos.
1.600	1.611	1.594	.017	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ ·6H ₂ O	TET tab	001 perf	Yellow to greenish	H 2-2.5 G 3.5 F 3
	1.612	1.585	.027	SODIUM URANOSPINITE (Meta-autunite grp) (Na ₂ , Ca)(UO ₂) ₂ (AsO ₄) ₂ ·5H ₂ O	TET tab, fib	001 perf 010, 100 dist	Yellow, yellow- green	H 2.5 G 3.85
1.610 1.616	1.613	1.585	.028	ZEUNERITE (Autunite grp) Cu(UO ₂) ₂ (AsO ₄) ₂ ·10-16H ₂ O	TET tab	001 perf 010 dist	Green	Diss by acids. El clv pos. Does not fluor in UV.
1.614	1.595	.019	SEMENOVITE (Ca, Ce, La, Na) ₁₀₋₁₂ (Fe, Mn) (Si, Be) ₂₀ (O, OH, F)48	ORTH ps tet	Uneven	Col s	H 2-2.5 G 3.4 fus	Diss by acids. El clv pos. Does not fluor in UV.
1.609 1.618	1.614	1.608	.006	FLUOCERITE (Ce, La)F ₃	HEX	0001 dist	Wax-yellow to brown	Insol in HCl or HNO ₃ , diss by H ₂ SO ₄ . Anom biax.
1.598 1.640	1.617	1.575	.042	BIOTITE, var Manganophyllite (Mica grp) K(Mg, Fe, Mn) ₃ (Al, Fe)Si ₃ O ₁₀ (OH, F) ₂	MCL ps hex	001 perf	Reddish- brown	2V 0-30°. El clv pos. Pleoc, 0 dark brown, E brown. Fe ₂ O ₃ 5.8, FeO 1.2, MnO 6.2%.
99	1.617	1.595	.022	KARNASURITITE (La, Ce, Th)(Ti, Nb)(Al, Fe) (Si, P) ₂₇ (OH) ₄ ·3H ₂ O (?)	HEX (?)	One good	Yellow	Anom biax.
	1.618	1.552	.066	CHALCOPHYLLITE Cu ₁₈ Al ₂ (AsO ₄) ₃ (SO ₄) ₃ (OH) ₂₇ 33H ₂ O	TRIG tab	0001 perf	Deep emerald- green	Diss by acids or NH ₄ OH. El clv pos. Pleoc, 0 blue- green, E nearly col s. Indices increase with loss of H ₂ O.
	1.632	1.618	.021	GLAUCONITE (Mica grp) (K, Na)(Al, Fe, Mg) ₂ (Si, Al) ₄ O ₁₀ (OH) ₂	MCL ps hex	001 perf	Green	Dec by acids. El clv pos. Pleoc, E yellow, 0 bluish- green. Fe ₂ O ₃ 18.8, Al ₂ O ₃ 8.5, FeO 4.0, MgO 3.6%.
	1.600 1.634	1.618	.012	BURBANKITE (Na, Ca) ₃ (Sr, Ba, Ce) ₃ (CO ₃) ₅	HEX u mass	Pris, 0001	Yellow, pink	Diss by acids with eff.
	1.632	1.618	.012	CLINOCHLORE (Chlorite grp) (Mg, Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex	001 perf	Olive-to dark-green	Dec by acids. El clv pos. Pleoc, E pale green, 0 green. 2V 0-5°.
	1.589 1.619	1.605	.014	CYRRITE Ba ₂ Si ₂ (O, OH) ₈ ·H ₂ O	MCL ps hex	001 perf pris imperf	Col s	Slightly sol in acids. El clv pos.
1.611 1.622	1.611	.008					H 2-3 G 3.45	

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.619	1.618	.001	GILLESPITE BaFeSi ₄ O ₁₀	TET platy	001 mic	Rose-red	H 3 G 3.33 F easy	Dec by HCl. El pos. Pleoc str, 0 pale pink, E deep rose-red.	
1.621	1.609	.012	KHANNESHITE (Na,Ca) ₃ (Ba,Sr,Ce,Ca) ₃ (CO ₃) ₅	HEX	Prism indist	Pale yellow	H soft G 3.8-3.9	Diss by HCl.	
1.622	1.606	.016	BAZZITE Be ₃ (Sc,Al,Fe) ₂ Si ₆ O ₁₈	HEX barrel-shaped	---	Bright blue	H 7 G 2.78 infus	Insol in acids. Pleoc, E azur-blue, 0 pale yellow to colrs.	
1.636	1.622	.008	CARBONATE-FLUORAPATITE (Apatite grp) Ca ₅ (PO ₄ ,CO ₃) ₃ (F,OH)	HEX pris, fib	0001 perf	Gray to greenish	H 5 G 3.05 F 3	Diss by acids. El cle pos. Basal section may show biax segments.	
1.603 1.640	1.614	.007	HUNTIITE CaMg ₂ (CO ₃) ₄	TRIG u mass	---	White	H 2-2.5 G 2.70 infus	Diss by HCl with eff.	
1.622	1.615	.007	WHITLOCKITE ("Merillite") (Ca,Na) ₉ (Mg,Fe)H(PO ₄) ₇	TRIG	---	Colrs	H 5 G 3.1 infus	Diss by acids. Meteorite mineral.	
1.607 1.630	1.623	.003	OLGITE Na(Sr,Ba)PO ₄	HEX	---	Bright blue, greenish-blue	H 4.5 G 3.94	---	
1.623	1.619	.004	'	TRIG	Uneven to conch	Yellow to orange	H 5 G 2.67	---	
1.624	1.591	.033	TISINALITE (Lovozerite grp) Na ₃ H ₃ (Mn,Ca,Fe)TiSi ₆ (O,OH) ₁₈ •2H ₂ O						
1.624 1.630	1.582	.045	TROEGERITE (Autunite grp) (UO ₂) ₃ (AsO ₄) ₂ •12H ₂ O (?)	TET tab 001	001 perf 100 good	Lemon-yellow	H 2-3 G 3.3 F 2.5	Diss by acids. Fluor Fluor yellow in UV.	
1.629	---	very low	KAMAISHILITE Ca ₂ Al ₂ SiO ₆ (OH) ₂	TET	---	Colrs	---	Opt sign unk.	
1.629	1.624	.005	MELILITE (Mellilite grp) (Ca,Na) ₂ (Mg,Al)(Si,Al) ₂ O ₇	TET	001 dist	Colrs	H 5 G 2.98 F 3	Gel with acids. Mg 0.8.2, Fe 2.1, Fe ₂ O ₃ 1.2, Al ₂ O ₃ 6.9, Na ₂ O 3.2%.	
1.653									

1.63	1.59	.04	HYDROHONESTITE Ni ₆ Fe ⁺³ (SO ₄) ₂ (OH) ₁₆ 7H ₂ O	HEX	---	Bright yellow	---	
1.606 ^	1.630	.010	EUDIALYTE Na ₄ (Ca,Ce) ₂ (Fe ⁺² ,Mn)ZrSi ₈ O ₂₂ (OH,Cl) ₂	TRIG	0001 dist 10T0 poor	Yellow, pink brown, pink	H 5-5.5 G 3.03 F 3	
1.652 ^	1.630	.003	WHITLOCKITE Ca ₉ (Mg,Fe)H(PO ₄) ₇	TRIG u mass	---	cols	H 5 G 3.12 infus	
1.623 ^	1.631	.003	METATORBERNITE (Meta-autunite grp) Cu(UO ₂) ₂ (PO ₄) ₂ •4-8H ₂ O	TET tab	001 perf 100 indist	Green	H 2.5 G 3.7-3.8 F 3	
1.625 ^	1.628	.003	CHALCOPHYLLITE Cu ₁₈ Al ₂ (AsO ₄) ₃ (SO ₄) ₃ (OH) ₂₇ 33H ₂ O	TRIG tab	0001 perf	Deep emerald-green	H 2.69 G 2.69 F 2-2.5	
1.618 ^	1.632	.057	BEMENTITE Mn ₈ Si ₆ O ₁₅ (OH) ₁₀	MCL fib	001 perf	Brown	H 4-6 G 2.7-3.1 F 3	
1.650 ^	1.632	.030	BURBANKITE (Na,Ca) ₃ (Sr,Ba,Ce) ₃ (CO ₃) ₅	HEX u mass	Pris., 0001	Yellow, pink	H 4.5-5 G 3.39-3.60 infus	
101	1.618 ^	1.620	.012	TET	001 perf	Blue	H 5 G 3.08	
	1.633	1.590	.043	HEX pris	0001 imperf	cols, green, brown	H 5 G 3.2 F 5	
	1.633	1.630	.003	FLUORAPATITE (Apatite grp) Ca ₅ (PO ₄) ₃ F	MCL and TCL ps hex	Brown	H 3.5 G 2.84 F 3.5	
	1.638 ^	1.634	1.575	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₂	TRIG platy	0001	reddish-brown	H 2.32 infus
	1.661 ^	1.634	1.590	COALINGITE Mg ₁₀ Fe ₂ CO ₃ (OH) ₂₄ •2H ₂ O	MCL ps hex	001 perf	Green	H 2 G 2.74 infus
	1.606 ^	1.634	.044	GLAUCONITE (Mica grp) (K,Na)(Fe ⁺³ ,Al,Mg,Fe ⁺²) ₂ (Si,Al) ₄ 10(OH) ₂				
	1.618 ^	1.634	.024					
	1.643(?)							

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Brief	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
v 1.638	1.634	1.612	.022	DRAVITE (Tourmaline grp) NaMg ₃ Al ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH) ₄	TRIG pris	1120, 10T1 very poor	Brown	H 7 G 3.03 infus	Insol in acids. El neg. Pleoc, abs 0 > E. MgO 8.4, FeO 0.2, Na ₂ 1.9%.
v 1.637	1.635	1.618	.017	ELBAITE (Tourmaline grp) Na(Al,Li) ₃ Al ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH) ₄	TRIG pris	1120, 10T1 very poor	Red	H 7 G 3.02 infus	Insol in acids. El neg. Pleoc, abs 0 > E. Fe ₂ O ₃ 0.3, Li ₂ O 1.9, CaO 0.4, Na ₂ O 2.4%.
[]	1.635	---	wk	EYLETTERSITE (Crandallite grp) (Tn,Pb,Ba)Al ₃ (PO ₄ ,SiO ₄) ₂ (OH) ₆	TRIG	---	White to cream	G 3.38-3.44	Fluor wk greenish-yellow in short-wave UV.
1.622 ^	1.636	1.618	.018	BAZZITE Be ₃ (Sc,Fe) ₂ Si ₆ O ₁₈	HEX barrel-shaped	---	Blue	H 7 G 2.78 infus	Insol in acids. Pleoc, E azure blue, 0 pale yellow to col.
1.637	1.615		.022	MITTSCHERLICHITE K ₂ CuCl ₄ ·2H ₂ O	TET tw on 001	---	Green to blue	H 2.5 G 2.40 fus	Sol in H ₂ O. Pleoc, E sky-blue, 0 grass-green. Anom biax. Data for synth compd.
1.637	1.615		.022	CHAMOSITE (Chlorite grp) (Mg,Fe ⁺² ,Fe ⁺³) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex	001 mic	Green	H 2.5 G 3.0 F diff	Diss by acids. Pleoc, E olive-green, 0 brownish-yellow. FeO 22.8, Fe ₂ O ₃ 9.1, Al ₂ O ₃ 22.1%.
[]	1.637	1.620	.017	METANOVACEKITITE (Meta-autunite grp) Mg(UO ₂) ₂ (AsO ₄) ₂ ·8H ₂ O	TET tab	001 perf 100 dist	Yellow	H 2-2.5 G 3.66	Diss by acids. Fluor yellow in UV. Commonly biax, 2V 0-20%.
1.635 ^ 1.646	1.637	1.621	.016	LIDDICOATITE (Tourmaline grp) Ca(Li,Al) ₃ Al ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH,F) ₄	TRIG pris	1120, 10T1 very poor	Brown, pink, green	H 7.5 G 3.02 infus	Insol in acids. Pleoc, 0 dark brown, E light brown. El neg. CaO 4.2, Na ₂ O 0.9, Li ₂ O 2.5, F 1.7%.
1.634 ^ 1.653	1.638	1.619	.019	UVITE (Tourmaline grp) (Ca,Na)(Mg,Fe) ₃ Al ₅ Mg(BO ₃) ₃ Si ₆ O ₁₈ (OH,F) ₄	TRIG pris	1120, 10T1 very poor	Brown	H 7 G 3.08 F 5	Insol in acids. El neg. Pleoc, abs 0 > E. MgO 15.4, FeO 0.5, CaO 5.5, Na ₂ O 0.2%.
1.633 ^ 1.644	1.638	1.634	.004	FLUORAPATITE (Apatite grp) (Ca,Sr) ₅ (PO ₄) ₃ F	HEX pris	0001 imperf	Col s	H 5 G 3.32 F 5	Diss by acids. El neg. SrO 11.6%.

1.617 1.652	1.640 1.640	1.590 1.633	.050 .007	SIDEROPHYLLITE (Biotite ser., Mica grp) KFe ₂ Al(Al ₂ Si ₂) ₁₀ (F,OH) ₂	MCL ps hex	001 mic	Black, dark brown	H 3 G 3.12 F 4	E1 clv pos. Pleoc, X brown, Y and Z dark brown. FeO 30.2, MnO 1.0%.
1.622 1.651	1.641 1.641	1.633 1.609	.032 .032	CARBONATE-FLUORAPATITE ("Dehnenite") (Apatite grp) (Ca,Na,K) ₅ (PO ₄ ,CO ₃) ₃ (F,OH)	HEX pris, fib	0001 good	Gray to greenish	H 5 G 3.05 F 3	Diss by acids. E1 clv pos, el crystal neg.
1.642	1.608	1.609	.034	METAHEINRICHITE (Meta-autunite grp) Ba(UO ₂) ₂ (AsO ₄) ₂ •8H ₂ O	TET	001 perf 100 dist	Yellow to green	H 2.5 G 4.04	Diss by acids. Pleoc, 0 pale yellow, E cols. Fluor yellow-green in UV.
1.643 1.643 1.663	1.612 1.623 1.643	1.612 1.623 1.644	.031 .020 .102	METAKAHLERITE (Meta-autunite grp) Fe(UO ₂) ₂ (AsO ₄) ₂ •8H ₂ O	TET	001 perf 100 good	Sulfur- yellow	---	Diss by acids. Does not fluor in UV. Anom biax.
1.644	1.644	1.542	.027	CELADONITE (Mica grp) (K,Na)(Fe ⁺³ ,Al,Fe ⁺² ,Mg) ₂ (Si,Al) ₄ Al ₁₀ (OH) ₂	MCL ps hex	001 perf	Bright green	H 2.8 G 2.8 F 3	Dec by HCl. Pleoc, E yellow-green, 0 green. Fe ₂ O ₃ 24.3, Al ₂ O ₃ 1.0, FeO 9.8, MgO 3.7%.
1.644	1.644	1.644	.006	ZUSSMANITE K(Fe,Mg,Mn) ₁₃ (Al,Si) ₁₈ 42 (OH) ₁₄	TRIG tab	0001 perf	Pale green	G 3.15	E1 clv pos. Pleoc wk, pale green, E cols.
1.644	1.644	1.644	.006	EWALDITE Ba(Ca,Y,Na,K)(CO ₃) ₂	HEX	---	Blue-green, brick red	H 3 G 3.42	---
1.644	1.644	1.644	.006	METAKIRCHHEIMERITE (Meta-autunite grp) Ca(UO ₂) ₂ (AsO ₄) ₂ •8H ₂ O	ORTH ps tet (?) plates	001 perf	Rose	H 2-2.5 G > 3.33	Diss by acids. E1 clv pos. Not fluor in UV.
1.644 1.651	1.646 1.646	1.638 1.625	.006 .021	FLUORAPATITE (Apatite grp) (Ca,Mn) ₅ (PO ₄) ₃ F	HEX pris	0001 imperf	Blue	H 5 G 3.26 F 5	Diss in acids. E1 neg. Fluor orange in UV. MnO 4.7%.
1.646 1.646	1.646 1.646	1.635 1.635	.011 .011	ELBAITE (Tourmaline grp) Na(Al,Li,Fe) ₃ Al ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH) ₄	TRIG pris	1120, 10T1 very poor	Blue-green	H 7 G 3.09 infus	Insol in acids. E1 neg. Na ₂ O 2.5, Li ₂ O 1.2, FeO 4.9, MnO 0.7%.
1.646	1.646	1.646	.011	ZAPATALITE Cu ₃ Al ₄ (PO ₄) ₃ (OH) ₉ •4H ₂ O	TET	001 good	Pale blue	H 1.5 G 3.02	Pleoc, wk in green, abs E > 0. Diss by acids.
1.646	1.646	1.639	.007	KELLYITE (Kaoinite-serpentine grp) (Mn,Al,Mg) ₃ (Si,Al) ₂ Si ₅ (OH) ₄	TRIG and HEX granular	0001 perf	Yellow	G 3.07	Anom biax. Pleoc wk, E cols, 0 pale yellow.
1.647	1.647	1.637	.010	AMINOFFITE Ca ₂ (B ₂ ,Al)(Si,Al) ₂ Si ₇ (OH) ₁₀ •H ₂ O	TET pyram	001 poor fr conch	Col s	H 5.5 G 2.94	Anom biax, 2V 0-15°.
1.648 1.607	1.648 1.607	1.622 1.622	.026 .026	VERMICULITE (Mg,Ni,Fe) ₃ (Si,Al) ₄ Si ₁₀ (OH) ₂ 4H ₂ O	MCL ps hex	001 perf	Brownish- black	H 1.5 G 2.8 F 5	Dec by HCl. E1 clv pos. Exfoliates when heated. Fe ₂ O ₃ 19.2, FeO 5.0, MgO 13.9, NiO 8.6%.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.643 1.651	1.648	1.624	.024	METAZEUNERITE (Meta-autunite grp) $Cu(UO_2)_2(AsO_4)_2 \cdot 8H_2O$	TET tab	001 perf 010 dist	Green	H 2.5 G 3.67	Diss by acids. Does not fluor in UV.
1.648	1.625	.023		KAZAKOVITE $Na_6H_2TiSi_6O_{18}$	TRIG	Uneven to subconch	Pale yellow	H 4 G 2.84 F 1.5	Tw poly.
1.637 1.685	1.649	1.643	.006	CHAMOSITE (Chlorite grp) $(Fe^{+2}, Mg, Fe^{+3})_5Al(Si_3Al)_10(OH, O)_8$	MCL ps hex	001 perf	Dark green	H 2 G 3.20	Dec by hot HCl. El cl v pos. Pleoc, 0 olive-green, E pale yellow.
1.649	1.641	.008		JEREMEJEVITE $Al_6B_5O_15(OH)_3$	HEX pris	Conch	Col to yellow-brown	H 6.5 G 3.28 infus	After ignition, diss by conc H_2SO_4 or KOH. El neg. Pleoc, 0 yellowish, E light blue. Anom biax.
1.649	1.641	.008		PYROCHROITE $Mn(OH)_2$	TRIG	0001 perf	Col to brown	H 2-2.5 G 3.1-3.2 infus	Diss by acids. Rapidly darkens in air. El cl v pos.
1.65	1.60	.05		BENNETITE $Mn_8Si_6O_15(OH)_{10}$	MCL fib	001 perf	Brown	H 4-6 G 2.7-3.1 F 3	Pleoc wk, 0 yellow, E nearly col's.
1.632 1.650	1.650	1.624	.026	STRONTIUM-APATITE (Apatite grp) $(Sr, Ca, Ba)_5(Po_4)_3(F, OH)$	HEX	---	Col to green	H 5 G 3.84	Diss by acids. El neg. Sr 0.46, Ca 0.10.8, Ba 0.2.7%.
1.644 1.660	1.651	1.637	.014	HYDROXYLAPATITE (Apatite grp) $Ca_5(Po_4)_3(OH, F)$	HEX	---	Col	H 5 G 3.21 infus	Diss by acids. El neg. F 0.16, H 2O 1.86%.
1.640 1.652	1.651	1.644	.007	BIOTITE (Mica grp) $K(Fe, Mg)_3(Al, Fe)Si_3O_{10}(OH, F)_2$	MCL ps hex	001 perf	Dark brown	H 2.5-3 G 3.05 F 4	El cl v pos. Pleoc, 0 dark brown, E brown. Fe 0.19.9, Mn 0.4, Fe 2O 3 3.2%.
1.640 1.672	1.652	1.640	.012	EUDIALYTE $Na_4(Ca, Ce)_2(Fe^{+2}, Mn)ZrSi_8O_22(OH, Cl)_2$	TRIG	0001 dist 1010 poor	Pink	H 5-5.5 G 2.87 F 3	Gel with acids. Pleoc wk in pink. Mn 0.10.8, Fe 2O 3 1.9%.
1.647 1.655	1.649	.003	"WILKEITE" (Apatite grp) $Ca_5(SiO_4, Po_4, SO_4)_3(O, OH, F)$	HEX u mass	0001 imperf	Yellow, reddish	H 5 G 3.1 F diff	Dec by acids. P 2O 5 15.9, SO 3 13.5, SiO 12.0%. = Silicatian Sulfatian Apatite or Phosphatian Fluorelltestadite.	

1.638 1.660	1.637	.016	DRAVITE (Tourmaline grp) $\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris 1120, 10T1 very poor	Dark brown	H 7 G 3.25 F 5
1.653	1.642	.011	CHANTALITE $\text{CaAl}_2\text{SiO}_4(\text{OH})_4$	TET ---	Cols, white	G 2.8-2.9
1.653	1.652	.001	MELILITE (Mellilite grp) $(\text{Ca}, \text{Na})_2(\text{Mg}, \text{Al})(\text{SiAl})_0{}_7$	TET 001 dist	Cols	H 5 G 3.0 F 4
1.629 1.672	1.654	.029	FRIEDELITE (Mn, Fe) ₈ Si ₆ O ₁₅ (OH, Cl) ₁₀	MCL ps trig tab 0001	pink, brown	H 4.5 G 3.17 fus
1.681	1.655	1.629	KONINCKITE $\text{FePO}_4 \cdot 3\text{H}_2\text{O}$	0001 perf 10T0 imperf ---	Yellow-green	H 3 G 2.4 F 2.5-3
1.655 1.703	1.655	1.645	ELLESTADITE (Apatite grp) $\text{Ca}_5(\text{SiO}_4, \text{P}_2\text{O}_7)_3(\text{OH}, \text{Cl}, \text{F})$	HEX TET fib	cols, pale rose, purple	H 5 G 3.1-3.2 F diff
1.654 1.660	1.655	1.650	CALCITE (Calcite grp) CaCO_3	TRIG 10T1 perf	cols, pink, brown	H 3 G 2.71 infus
1.658	1.658	1.486	CHLORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3(\text{Cl}, \text{F}, \text{OH})$	MCL ps hex 001 imperf	greenish- yellow	H 5 G 3.18 F 5
1.672	1.658	1.653	MCKELVEYITE $\text{Ba}_3\text{Na}(\text{Ca}, \text{U})\text{Y}(\text{CO}_3)_6 \cdot 3\text{H}_2\text{O}$	TCL ps trig tab	dark green	H 3.5-3.6
1.666	1.666	1.57	BELOVITE (Apatite grp) $(\text{Sr}, \text{Ce}, \text{Na}, \text{Ca})_5(\text{PO}_4)_3(\text{OH})$	HEX pris pris, pinacoidal, imperf	honey- yellow	H 5 G 4.19 F diff
1.651 1.661	1.660	1.640	FERMORITE (Apatite grp) $(\text{Ca}, \text{Sr})_5(\text{PO}_4, \text{AsO}_4)_3(\text{OH}, \text{F})$	HEX pyram ---	pink to white	H 5 G 3.52 F diff
1.66	---	wk	UVITE (Tourmaline grp) $\text{Ca}(\text{Mg}, \text{Fe})_3\text{Al}_5\text{Mg}(\text{BO}_3)_3\text{Si}_6\text{O}_{18}$	TRIG pris 1120, 10T1 very poor	dark brown	H 7 G 3.16 F 5
1.653 1.661	1.660	1.639	STILPNOMELANE $\text{K}(\text{Fe}^{+3}, \text{Fe}^{+2}, \text{Mg})_{10}\text{Si}_{12}\text{O}_{30}$ (0, OH) ₁₂	MCL and TCL 001 perf ps hex	brown	H 3.5 G 2.80 F 3.5

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.660 \downarrow 1.675	1.661	1.632	.029	SCHORL (Tourmaline grp) $\text{NaFe}_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	11 $\overline{2}0$, 10 $\overline{1}1$ very poor	Black	H 7 G 3.17 F diff	Insol in acids. El neg. Pleoc, abs 0 > E. FeO 8.7, MgO 5.6, Fe_2O_3 0.8, Na_2 2.3%.
1.667	1.661	1.646	.015	PENNANTITE (Chlorite grp) $\text{Mn}_5\text{Al}(\text{Si}_3\text{Al})_8(\text{OH})_{10}$	MCL ps hex, plates	001 perf	Orange- brown	H 2.5 G 3.06	Pleoc in orange. El clv pos. MnO 38.9, Al_2O_3 18.6, Fe_2O_3 4.4%.
1.643 \wedge	1.663	1.644	.019	CELAONITE (Mica grp) (K, Na)(Al, Fe, Mg) ₂ (Si, Al) ₄ O ₁₀ (OH) ₂	MCL ps hex	001 perf	Bright green	H 2 G 2.8 F 3	Dec by HCl. El clv pos. Pleoc, E yellow-green, 0 emerald-green. FeO 0, Al_2O_3 26.9, Al_2O_3 2.1, FeO 9.3%.
1.669	1.664	1.634	.030	PYROSMALITE (Fe, Mn) ₈ Si ₆ O ₁₅ (OH, Cl) ₁₀	TRIG	0001 perf	Brown	H 4.5-5 G 3.11 fus	Dec by HCl. El clv pos. Pleoc, 0 light brown, E dark brown.
1.664	1.646	1.634	.018	GONYERITE (Chlorite grp) (Mn, Mg) ₅ Fe(Si ₃ Fe)O ₁₀ (OH) ₈	ORTH (?) ps hex	001 perf	Deep brown	H 2.5 G 3.01	---
1.666	1.653	1.653	.013	TIENSHANITE $\text{BaNa}_2\text{MnTi}_1\text{B}_2\text{S}_1\text{O}_6\text{Cl}_2\text{O}_2$	HEX	0001 dist	Pistachio- green	H 6-6.5 G 3.29 F 2	Diss by acids. El clv pos. Data for synth compd.
1.658 \wedge	1.666	1.660	.006	CHLORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3\text{Cl}$	MCL ps hex	001 imperf	Cols	H 4.5 G 3.2 F 5	Pleoc, 0 very dark brown, E reddish-brown.
1.661 \downarrow 1.673	1.667	1.658	.009	PENNANTITE ("Grovesite") (Chlorite grp) (Mn, Fe)Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex	001	Blackish- brown	G 3.15	Pleoc wk in brown and yellow.
1.664 \square 1.671	1.667	1.662	.005	STEENSTRUPINE (Ce, La, Na, Mn) ₆ (Si, P) ₆ O ₁₈ (OH) ₈	---	---	Brown, red, yellow	H 5 G 3.5 infus	Dec by HCl.
1.668	1.641	1.662	.027	MCGILLITE (Mn, Fe ⁺² , Mg) ₈ Si ₆ O ₁₅ (OH) ₈ Cl ₂	MCL ps trig	0001 good	Pink	G 2.98	Dec by HCl. El clv pos.
1.664 \wedge 1.682	1.669	1.631	.038	MANGANPYROSMALITE (Mn, Fe) ₈ Si ₆ O ₁₅ (OH, Cl) ₁₀	HEX	0001 perf	Brown	H 4.5-5 G 3.13 fus	Pleoc, 0 greenish-yellow, cols.

1.669	1.657	.012	HARDYSTONITE (Mellilite grp) Ca ₂ ZnSi ₂ O ₇	TET granular	001 good 100, 110 rare	White	H 3 G 3.4 F diff	Gel with acids.
1.669	1.658	.011	GEHLENITE (Mellilite grp) Ca ₂ Al(AlSi ₁ O ₇) ₂	TET	001 imperf	Col s	H 6 G 3.04 F 6	Gel with acids. Data for synth compd.
1.653 ^ 1.672	1.672	.171	CALCITE (Calcite grp) (Ca, Mn, Mg)CO ₃	TRIG	10T1 perf	White, pink	H 3 G 2.82 infus	Diss by acids with eff. Mn 0 4.2, FeO 2.1, MgO 1.3%.
1.658 ^ 1.710	1.672	.048	ANNITE (Biotite ser, Mica grp) KFe ₃ AlSi ₃ O ₁₀ (OH,F) ₂	MCL ps hex	001 perf	Dark brown	H 2.5-3 G 3.0 F 4	Pleoc., 0 dark brown, E brown. FeO 32.1, Fe ₂ O ₃ 3.1, TiO ₂ 3.6%.
1.652 ^	1.672	.011	MELLILITE (Mellilite grp) (Na, Ca) ₂ (Mg, Fe, Al)(Si, Al) ₂ O ₇	TET	001 dist	Brown	H 5 G 3.1 F 4	Gel with acids. Fe ₂ O ₃ 8.4, Al ₂ O ₃ 8.4, FeO 1.1, MgO 4.7, Na ₂ O 3.8%.
1.669 ^	1.672	.145	PARALSTONITE (Ba, Sr)Ca(CO ₃) ₂	TRIG	---	White, col s	H 4-4.5 G 3.60 infus	Fluor pale to bright orange in long-wave UV.
1.672	1.664	.009	PENNANTITE (Chlorite grp) (Mn, Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex	001 perf	Brown	H 2.5 G 3.04	Pleoc. in browns. El cl v pos.
1.672	1.664	.085	CHLOROMAGNESITE MgCl ₂	HEX plates	---	Col s	H soft G 2.33 F 2	Sol in H ₂ O, deliq.
1.667 ^	1.675	.032	DRAVITE (Tourmaline grp) NaMg ₃ (Al, V) ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH) ₄	TRIG pris	11T0, 10T1 very poor	Green	H 7 G 3.13 F 5	Insol in acids. El neg. Pleoc., E pale green, 0 red-brown. MgO 6.3, FeO 1.3, V ₂ O ₃ 8.0%.
1.661 ^ 1.693	1.675	.048	SPANGOLITE Cu ₆ Al(SO ₄) ₂ (OH) ₁₂ Cl ₁ ·3H ₂ O	TRIG tab	0001 perf	Dark- to bluish-green	H 2-3 G 3.14 F 3	Diss by acids. El pos. Pleoc., 0 green, E bluish-green.
1.686 ^	1.678	.179	DOLOMITE (Dolomite grp) CaMg(CO ₃) ₂	TRIG	10T1 perf	White, buff	H 3.5-4 G 2.865 infus	Diss with eff in warm HCl. FeO 0.5%.
1.694 ^	1.680	.025	SINCOSITE CaV ⁺⁴ ₂ (PO ₄) ₂ (OH) ₄ ·3H ₂ O	MCL ps tet, tab	001 good 100, 110 poor	Leek-green	H soft G 2.84 fus	Diss by acids to a blue soln. Pleoc., 0 gray-green, E nearly colrs.
1.654 ^	1.681	.038	FRIEDELITE (Mn, Fe) ₃ Si ₆ O ₁₅ (OH, Cl) ₁₀	MCL ps trig tab 0001	0001 perf 10T0 imperf	Brown	H 4.5 G 3.3 fus	Dec by HCl. El pos. Pleoc. in browns.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.669 ^v	1.682	1.650	.032	PYROSMALITE (Fe,Mn) ₈ Si ₆ O ₁₅ (OH,Cl) ₁₀	HEX	0001 perf	Brown	H 4.5-5 G 3.1 fus	Diss by HCl. E1 clv pos.
1.683	1.672	.011		VERPLANCKITE Ba ₂ (Mn,Fe ⁺² ,Ti)Si ₂ O ₆ (O,OH,Cl,F) ₂ ·3H ₂ O	HEX	1120 good	Orange to yellow	H 2.5-3 G 3.52	Pleoc., 0 orange-yellow, E cols.
1.684	1.672	.012		SVABITE (Apatite grp) Ca ₅ (AsO ₄ ,PO ₄) ₃ (F,OH)	HEX	---	Green	H 4-4.5 G 3.54 F 5	Diss by acids. P ₂ O ₅ 12.5, F 1.4%.
1.685 ^v	1.595	.090		STILPNOMELANE K(Fe ⁺³ ,Fe ⁺² ,Mg) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₂	MCL and TCL ps hex	001 mic	Brownish-black	H 3.5 G 2.85 F 3	Diss by acids. E1 clv pos. Pleoc., 0 dark olive-brown, E yellow. FeO 13.7, Fe2O ₃ 22.0%.
1.661 ^v	1.685	1.595		CHAMOSITE (Chlorite grp) (Fe ⁺² ,Fe ⁺³ ,Mg) ₆ Al(Si ₃ Al) ₁₀ (OH,O) ₈	MCL ps hex	001 perf	Brown	H 6 G 2.96	Diss by hot HCl. E1 clv pos. Pleoc., 0 dark green, E yellow.
1.649 ^v	1.685	1.670	.015	PABSTITE Ba(Sn,Ti)Si ₃ O ₉	HEX	---	Col., white	H 6 G 4.03	Anom blue-violet and yellow interf colors. Fluor bluish white in short-wave UV.
1.685	1.674	.011		SPANGOLITE Cu ₆ Al(SO ₄) ₂ Cl(OH) ₁₂ ·3H ₂ O	TRIG tab	0001 perf	Dark-to bluish-green	H 2-3 G 3.14 F 3	Diss by acids. E1 clv pos. Pleoc., 0 green, E bluish-green.
1.678 ^v	1.686	1.635	.051	JOHNBAUMITE (Apatite grp) Ca ₅ (AsO ₄) ₃ (OH)	HEX	100 dist	Col., white	H 4.5 G 3.68	Fluor med pink-orange in short-wave UV.
1.687	1.684	.003		COMBLAINITE (Ni ⁺² ,Co ⁺³)(OH) ₂ (CO ₃)·xH ₂ O	TRIG	---	Turquoise	G 3.05	----
1.690	1.684	.006		BENSTONITE (Ni ⁺² ,Co ⁺³)(OH) ₂ (CO ₃)·xH ₂ O	TRIG	101 very good	White	H 3-4 G 3.60 infus	Diss by acids with eff. Fluor red in UV.
1.691 ^v	1.527	.164		BANDYLITE (Ba,Sr) ₆ (Ca,Mn) ₆ Mg(CO ₃) ₁₃	TET tab	001 perf	Deep blue	H 2.5 G 2.81 F 2	E1 clv pos. Pleoc., 0 deep blue, E pale greenish-yellow.
1.691	1.641	.050		BANDYLITE Ca ₂ (OH) ₄					

1.692	1.648	.044	WICKENBURGITE $Pb_3CaAl_2Si_{10}O_{24}(OH)_6$	HEX tab	0001 indist	White, pink	H 5 G 3.85 F 3	Insol in acids. Fluor dull orange in UV.
1.675 △ 1.710	1.693	.034	SCHORL (Tourmaline grp) $NaFe_3Al_6(BO_3)_3Si_6O_{18}(OH)_4$	TRIG pris	1120, 1011 very poor	Black	H 7 G 3.2 F 5	Insol in acids. El neg. Pleoc., abs 0 > E. MgO 7.3, FeO 0.6, Fe_{23} 9.2, CaO 1.5, Na_2O 2.6, TiO_2 2.2%.
1.680 △ 1.700	1.694	.184	DOLOMITE (Dolomite grp) $Ca(Mg,Fe)(CO_3)_2$	TRIG	1011 perf	White, brown	H 3.5-4 G 2.92 infus	Diss by warm HCl. FeO 8.4, MnO 1.1, SrO 0.4%.
1.687 △ 1.712	1.694	.175	NORSETHITE (Dolomite grp) $BaMg(CO_3)_2$	TRIG	1011 perf	White to yellow- green	H 3.5 G 3.74-3.84 infus	Diss by acids.
1.696	1.694	.002	ALFORSITE (Apatite grp) $Ba_5(Po_4)_3Cl$	HEX	---	Cols	G 4.80	Data for synth.
1.699	1.694	.041	PHOSPHURANYLITE $Ca(UO_2)_4(Po_4)_2(OH)_4 \cdot 7H_2O$	ORTH plates	100 good 010	Lemon- yellow	H 2.5-3 G 4.1 infus	Diss by acids. Pleoc., E pale yellow, 0 golden- yellow. U biax, 2V 0-20°. Not fluor in UV.
1.724	1.700	.1.21	DOLOMITE, zincian (Dolomite grp) $Ca(Mg,Zn)(CO_3)_2$	TRIG	1011 perf	White, brown	H 3.5-4 G 2.95 infus	Diss by warm HCl. Fluor red in UV. FeO 2.2, ZnO 8.3, PbO 1.1%.
1.694 △ 1.710	1.700	.179	MAGNESITE (Calcite grp) $MgCO_3$	TRIG u mass	1011 perf	Cols, white	H 3-4 G 2.96 infus	Diss by warm acids. Data for pure $MgCO_3$.
1.711	1.700	.191	VESUVIANITE $Ca_{10}(Mg,Fe)_2Al_4(Si_{20})_2$ $(SiO_4)_5(OH,F)_4$	TET	110 poor	Brown, blue	H 5.5-6 G 3.40 F 4	Nearly insol in HCl. FeO 2.0, Fe_{23} 2.3, TiO_2 1.7%.
1.714	1.702	.002	APATITE-BRITHOLITE (Apatite grp) $(Ca,Ce)_5(Po_4,SiO_4)_3F$	HEX mass	---	Reddish- brown	H 4.5 G 3.83 infus	Dec by HCl. Po_5 17.3, SiO_2 12.4, Ce_{23} 12.9%.
1.655 △ 1.752	1.703	1.699	SCHALLERITE $(Mn,Fe)_{16}Si_{12}As^{+3}O_{36}(OH)_{17}$	TRIG	0001 perf	Reddish- brown	H 4.5 G 3.37 F diff	Slowly dec by HCl. El clv pos. MnO 44.7, FeO 2.1%.
1.691 △	1.704	1.679	STILPNOELANE $K(Fe^{+3},Fe^{+2},Mg)_{10}Si_{12}O_{30}$ $(O,OH)_2$	TRIG	1011 very good	White	H 3-4 G 3.58 infus	Diss by HCl with eff. BaO 44.2, CaO 12.3, PbO 2.8, MgO 4.7, MnO 1.6%.
1.685 △ 1.735	1.705	1.596	MCL and TCL ps hex	001 mic	Brownish- black	H 3.5 G 2.85 F 3	Dec by acids. El clv pos. Pleoc., E golden-yellow, 0 dark red-brown. Fe_{23} 23.9, FeO 13.2%.	

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.700 \downarrow 1.711	1.710	1.519	.191	KUTNOHORITE (Dolomite grp) Ca(Mn,Mg)(CO ₃) ₂	TRIG	10̄I perf	Pink, brown	H 3-4 G 3.00 infus	Diss by hot acids. MnO 17.6, MgO 10.0, FeO 1.3%.
1.672 \downarrow 1.713	1.710	1.523	.189	CALCITE, cadmian (Calcite grp) (Ca,Cd)CO ₃	TRIG	10̄I perf	Col to yellow	G 2.9 infus	CaCO ₃ 73, CdCO ₃ 27%.
1.693 \downarrow 1.735	1.710	1.664	.046	DRAVITE, chromian (Tourmaline grp) NaMg ₃ (Al,Cr) ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH) ₄	TRIG pris	11̄I0, 10̄I poor	Green	H 7 G 3.13 infus	Insol in acids. El neg. Pleoc, abs 0 > E. Cr ₂ O ₃ 17.8, Fe ₂ O ₃ 1.0, FeO 1.8%.
1.700 \downarrow 1.726	1.711	1.519	.192	MAGNESITE, nickelocaloan (Calcite grp) (Mg,Ni)CO ₃	TRIG	10̄I perf	Green	H 3-4 G 3.15 infus	Diss by hot acids. NiO 13.55%.
1.710 \downarrow 1.740	1.711	1.520	.191	DOLOMITE, ferroan (Dolomite grp) Ca(Mg,Fe)(CO ₃) ₂	TRIG	10̄I perf	Brown	H 3.5-4 G 3.01 infus	Diss by warm HCl. FeO 12.6, MnO 1.2%.
1.694 \wedge	1.712	1.512	.200	NORSETHITE (Dolomite grp) Ba(Mg,Fe)(CO ₃) ₂	TRIG	10̄I perf	White to yellow-green	H 3.5 G 3.85 infus	Diss by acids. FeO 3.6, MnO 0.8%.
1.712	---	str		PALMIERITE (K,Na) ₂ Pb(SO ₄) ₂	TRIG plates	---	Col	G 4.5 F easy	Dec by boiling H ₂ O, diss by HNO ₃ . Luster pearly.
1.710 \downarrow 1.731	1.713	1.519	.194	CALCITE, manganesean (Calcite grp) (Ca,Mn)CO ₃	TRIG	10̄I perf	Pink	H 3 G 3.02 infus	Diss by acids with eff. MnO 20.1, ZnO 0.4%.
1.714	1.702	.012		TRASKITE Ba ₉ Fe ₂ Ti ₂ Si ₁₂ O ₃₆ (OH,Cl,F) ₆ ·6H ₂ O	HEX	Conch	Brownish-red	H 5 G 3.71 F 3.5	Insol in acids. Pleoc, 0 brown-red, E col to straw-yellow.
1.702 \downarrow 1.742	1.714	1.709	.005	VESUVIANITE Ca ₁₀ Mg ₂ Al ₄ (Si ₂ O ₇) ₂ (SiO ₄) ₅ (OH,F) ₄	TET	110 poor	Brown	H 5 G 3.38 F 4	Nearly insol in HCl. Fe ₂ O ₃ 3.3, MnO 3.3, BeO 1.6%.
1.684 \wedge	1.716	1.698	.018	SVABITE (Apatite grp) Ca ₅ (AsO ₄) ₃ (Cl,F)	HEX	---	Green, brown	H 4-4.5 G 3.5 F 5	Diss by acids. Cl 1.9, F 0.45%.
1.721	1.719	.002		SATTERLYITE (Fe ⁺² ,Mg,Fe ⁺³) ₂ (PO ₄) ₂ (OH)	HEX	---	Pale brown, pale yellow	H 4.5-5 G 3.68 fus	Diss by acids. Pleoc, 0 pale yellow, E brownish-yellow, abs E > 0.

1.722	1.718	.004	TAAFFEITE $Mg_3Al_8BeO_{16}$	---	Mauve	G 3.61	---		
1.723	1.716	.007	HATURRITE Ca_3SiO_5	ps hex	Col s	---	Dec by H_2O .		
1.699	1.724	.050	PHOSPHURANYLITE $Ca(UO_2)_4(PO_4)_2(OH)_4 \cdot 7H_2O$	ORTH plates	100 good 010	Lemon-yellow	H 2.5-3 6 4.1 fus	Diss by acids. Pleoc, E pale yellow, 0 golden-yellow. U biax, 2V 0-20°.	
1.65	1.724	.042	PYROCHROITE $Mn(OH)_2$	TRIG	0001 perf	White to brown	H 1.5-2 6 3.1-3.25 infus	Diss by acids. El cleav pos. Darkens in sunlight. Abs 0 > E.	
1.711	1.726	.198	MAGNESITE (Calcite grp) $(Mg,Fe)CO_3$	TRIG	10T1 perf	White to brown	H 3-4 6 3.10 infus	Diss by hot acids. FeO 9.5, CaO 0.4%.	
1.788	1.727	---	KITTATINNYITE $Ca_4Mn^{+3}O_4Si_4O_{16}(OH)_8 \cdot 18H_2O$	HEX	0001 perf	Bright yellow	H ~4 6 2.61	Pleoc wk, abs E > 0.	
1.728	---	---	WALLKILDELLITE $Ca_4Mn^{+2}As^{+5}_4O_{16}(OH)_8 \cdot 18H_2O$	HEX	0001 perf	Dark red	G 2.85	Pleoc, 0 reddish-orange, E orange-pink, abs 0 > E.	
1.73	1.69	.04	BIRNESSITE $Na_4Mn^{+2}O_2 \cdot 9H_2O$	ORTH mass	---	Brownish-black	H 1.5 6 3.0 infus	Diss by HCl. Nearly opaque.	
~1.73	1.72	.01	MELANOCRITERE (Ca,Ce,Y) ₅ (Si,B) ₃ O ₁₂ (OH,F)• xH_2O	HEX tab	Conch	Deep brown to black	H 5-6 6 4.13 infus	Dec by acids. Pale yellow in section.	
1.713	1.731	(1.55)	.18	RHOODOCHROSITE (Calcite grp) $(Mn,Ca)CO_3$	TRIG	10T1 perf	Pink	H 4 6 3.05 infus	Diss by hot acids. MnO 33.4, CaO 21.0, FeO 2.7, MgO 2.2%.
1.753	[]	1.733	1.714	.019	HEMATOLITE $(Mn,Mg,Al)_{15}(AsO_4)_2AsO_3(OH)_{23}$	TRIG	0001 perf	Brown to red	Diss by acids. El cleav pos. In section yellow to brown. Anom biax, small 2V.
1.750	v	1.734	1.592	.192	MIRECORDITE (Dolomite grp) $Ca(Zn,Mg)(CO_3)_2$	TRIG	10T4 perf	Col s	Slowly diss by cold HCl.
1.705	1.735	1.625	.110	STILLPONOMELANE $K(Fe,Mg,Al)_{10}Si_{12}O_{30}(OH)_{12}$	MCL and TCL ps hex	001 mic	Brownish-black	Dec by acids. El cleav pos. Pleoc, E golden yellow, 0 deep red-brown. Fe ₂ O ₃ 31.7, FeO 1.3, MnO 2.6%.	
1.72	1.735	(~1.650)	~.085	REEVESITE $Ni_6Fe_2(OH)_{16}CO_3 \cdot 4H_2O$	TRIG	---	Yellow to greenish-yellow	Diss by acids.	
							6 2.80-2.88 infus		

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.710 _A	1.735	1.655	.080	BUERGERITE (Tourmaline grp) $\text{NaFe}_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{21}\text{F}$	TRIG pris	Prism, dist	Dark brown	H 7 G 3.31. F diff	Insol in acids. El neg. Pleoc, 0 yellow-brown, E pale yellow.
1.738	1.728	.010	STOTTITE $\text{FeGe}(\text{OH})_6$	TET ps oct	100, 010 good, 001	Brown	H 4-5 G 3.60	Luster greasy. Anom biax.	
1.711(?) _V 1.741	1.740	1.547	.193	KUTNOHORITE (Dolomite grp) $\text{CaMn}(\text{CO}_3)_2$	TRIG	10T1 perf	Pink, brown	H 3-4 G 3.14 infus	Diss by hot acids. MnO 31.8%, ZnO 1.4%, FeO 0.8%, MgO 0.5%.
1.740 _A	1.741	1.536	.205	ANKERITE (Dolomite grp) $\text{Ca}(\text{Fe,Mg})(\text{CO}_3)_2$	TRIG	10T1 perf	Brown	H 3.5-4 G 3.12 infus	Diss by hot HCl. FeO 24.0, MnO 1.6%.
1.714 _A 1.762	1.742	1.736	.006	VESUVIANITE $\text{Ca}_{10}(\text{Mg,Fe,Ti})_2\text{Al}_4(\text{Si}_2\text{O}_7)_2$ (SiO_4) ₅ (OH) ₄	TET	110 poor	Brown	H 5 G 3.41 F 4	Insol or partly dec by HCl. Fe ₂ O ₃ 4.4, FeO 1.6%, TiO ₂ 4.7%.
1.745	1.565	.180	ZHONGHUACERITE $\text{Ba}_2\text{Ce}(\text{CO}_3)_3\text{F}$	TRIG	---	Yellow	H 4.5-5 G 4.3	Diss by acids.	
1.747	1.741	.006	AKDALAITITE $4\text{Al}_2\text{O}_3 \cdot \text{H}_2\text{O}$	HEX	Fr irreg	White	H 7 G 3.68	---	
1.748	1.645	.103	LAVENDULAN $\text{NaCaCu}_5(\text{AsO}_4)_4\text{Cl} \cdot 5\text{H}_2\text{O}$	ORTH ps hex	010 good 100, 001 dist	Greenish-blue	H 2.5 G 3.54 F easy	Diss by acids. Pleoc, E blue to greenish-blue, 0 pale blue to pale greenish- blue, abs E > 0.	
1.734 _A	1.750	1.550	.200	MINRECORDITE (Dolomite grp) $\text{CaZn}(\text{CO}_3)_2$	TRIG	10T4 perf	Milk-white	H 3.5 G 3.45	Slowly diss by cold HCl.
1.75	1.728	.02	JASMUNDITE $\text{Ca}_{11}(\text{SiO}_4)_2\text{O}_2\text{S}$	TET	---	---	G 3.03	---	
1.751	1.645	.106	CLARALITE $(\text{Cu,Zn})_3(\text{CO}_3)(\text{OH}) \cdot 4\text{H}_2\text{O}$	TCL (?) ps hex spherulitic	10T0 perf	Bluish-green	H ~ 2 G 3.33	r > v.	
1.751	1.748	.003	CAPPELENITE $(\text{Ba,Ca,Na})(\text{Y,La})_6\text{B}_6\text{Si}_3$ (OH) ₂₇	HEX	---	Green to brown	H 6-6.5 G 4.41 F diff	Diss by HCl, gel.	
1.703 _A 1.823	1.752	1.748	.004	BRITHOLITE $(\text{Ce,Ca,Y})_5(\text{SiO}_4,\text{PO}_4)_3(\text{OH},\text{F})$	HEX	---	Brown	H 5 G 4.1-4.4 infus	Gel with acids. Apatite structural type.

1.753	1.740	.004	OKANOGANITE (Na,Ca) ₃ (Y,Ce,Nd,La) ₁₂ Si ₆ B ₂ O ₇ F ₄	TRIG	---	Pale pink	H 4 G 4.35	Tw.						
1.753	1.560	.193	RHODOCHROSITE (Ca,Cite grp) (Mn,Ca)CO ₃	TRIG	10T1 perf	Pink	H 4 G 3.3 infus	Diss by hot acids. MnO 44.3, CaO 13.4, MgO 2.3, FeO 0.8%.						
1.753	1.717	.036	WISERITE Mn ₄ B ₂ O ₅ (OH,Cl) ₄	TET fib	Clv across fibers, perf	Dark brown	G 3.42	Diss by HCl.						
1.755	1.731	.024	CONGOLITE (Fe,Mn,Mg) ₃ B ₇ O ₁₃ Cl	TRIG	---	Pale red	G 3.58	Diss by acids.						
1.756	1.680	.076	NATISITE Na ₂ TiSiO ₅	TET	001 perf 100 less so	Yellow- green to greenish- gray	H 3-4 G 3.15 calc F easy	Insol in dilute acids.						
1.760	1.577	.183	CORDYLITE Ba(Ce,La) ₂ (CO ₃) ₃ F ₂	HEX pyram	0001 dist	Wax-yellow	H 4.5 G 4.10 infus	Diss by acids. Pleoc, 0 green-yellow, E brownish- yellow.						
1.762	1.750	.012	VESUVIANITE (Ca,Ce) ₁₀ (Mg,Fe) ₂ Al ₄ (Si ₂ O ₇) ₂ (Si ₁ O ₄) ₅ (OH) ₄	TET	110 poor	Dark brown	H 5 G 3.60 F 4	Insol in acids. (Ce,La) ₂ O ₃ 16.7, Fe ₂ O ₃ 5.4, FeO 5.5%.						
1.765	1.735	.030	SHERWOODITE Ca ₉ Al ₂ V ⁺⁵ ₂₄ V ⁺⁴ ₄ O ₈₀ ·56H ₂ O	TET	Conch to uneven	Blue-black	H 2 G 2.8	Diss by acids. Pleoc, 0 green, E blue.						
1.766	1.585	.178	HANGHOTITE Ba(Ce,La)(CO ₃) ₂ F	HEX	0001	Honey- yellow to yellow-green	H 4.5-5 G 4.58 infus	Diss by acids. Pleoc wk in yellow-green.						
1.766	1.758	.008	CORUNDUM Al ₂ O ₃	TRIG	Parting 0001, 00T1 sometimes perf	Col, red, blue, brown	H 9 G 4.00 infus	Insol in acids. Colored var pleoc. Anom biax, 2V 0-58°.						
1.772	1.770	.002	SWEDENborgite NaBe ₄ SnO ₇	HEX pris	0001 dist	Col, to wine- yellow	H 8 G 4.29	Insol in acids. E1 crystal neg.						
1.775	1.765	.010	FRESNOITE Ba ₂ TiSi ₂ O ₈	TET	001 fair	Lemon- to canary- yellow	H 3-4 G 4.43 F 4	Insol in cold, dec by hot HCl. Pleoc, 0 cols, E yellow.						
1.778	1.660	.118	NORDENSKIÖLDINE CaSnB ₂ O ₆	TRIG tab	0001 perf fr conch	Col to yellow	H 5.5-6 G 4.20 infus	Diff diss by acids. E1 neg.						

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.846	1.778	1.772	.006	CHROMDRAVITE (Tourmaline grp) Na ⁺ (Mg, V, Al) ₃ (Cr, Fe ⁺³) ₆ B ₃ Si ₆ O ₂₇ (OH) ₄	TRIG	---	Dark green	G 3.40	Pleoc, 0 dark green, E yellow-green, abs 0 > E.
v	1.78	1.75	.03	HIBONITE (Ca, Ce)(Al, Ti, Fe, Mg) ₁₂ O ₁₉	HEX	0001 easy, parting 10T0	Brownish-black	H 7.5-8 G 3.84 infus	Insol in acids. Pleoc, 0 cols, E blue.
1.782	1.780	0.002		CLARINGBULLITE Cu ₈ C ₁₂ (OH) ₁₄ •H ₂ O	HEX plates	0001 perf 10T0, 1120 dist	Blue	G 3.9	Abnormal intert colors dark blue to yellow-green.
v	1.788	1.570	.218	MAGNESITE-SIDERITE (Calcite grp) (Mg, Fe)CO ₃	TRIG	10T1 perf	White to brown	H 4 G 3.43 infus	Diss by hot acids. Calc for Mg:Fe = 1:1, FeO 35.9, MgO 20.1%.
v	1.816	1.791	.086	JAROSITE (Alunite grp) KFe ₃ (SO ₄) ₂ (OH) ₆	TRIG tab, rhombs	0001 dist	Yellow to brown	H 2.5-3.5 G 2.9-3.2 F 4.5	Diss by HCl. Pleoc, 0 yellow to brown, E paler to col. Basal section may be divided into biax segments.
v	1.820	1.794	.009	CORUNDUM (Al, Fe) ₂ O ₃	TRIG	Parting 0001, 00T1 sometimes perf	Brown	H 9 G 4.1 infus	Insol in acids. Fe ₂ O ₃ 9.2, TiO ₂ 0.4%.
v	1.766	1.785	.020	VESUVIANITE (Ca, Sb) ₁₀ (Mg, Fe) ₂ Al ₄ (Si ₂ O ₇) ₂ (SiO ₄) ₅ (OH) ₄	TET pris	110 poor	Pale green	H 5 F 4	Insol in acids. El neg. E cols, 0 pale yellow-green. Sb ₂ O ₃ 15.7, Fe ₂ O ₃ 3.3, FeO 4.1%.
v	1.762	1.775	.020	CRONSTEDTITE (Kaolinite-Serpentinite grp) (Fe ⁺² , Mg)Fe ⁺³ (SiFe) ₀ (OH)	HEX, TRIG, and MCL polytypes	0001 perf	Black	H 3.5 G 3.34 F 4	Gel with acids. Pleoc, E dark reddish-brown, 0 nearly opaque.
v	1.80	---		FERRIDRAVITE (Tourmaline grp) (Na, K)(Mg, Fe ⁺²) ₃ Fe ⁺³ B ₃ Si ₆ O ₂₇ (OH) ₄	TRIG	Uneven	Black	G 3.26	Streak brown. Insol in acids.
v	1.800	1.743	.057	AMMONIOJAROSITE (Alunite grp) (NH ₄)Fe ₃ (SO ₄) ₂ (OH) ₆	TRIG tab	0001 dist	Pale yellow	G 3.11	Diss by HCl.
v	1.800	1.750	.050						

1.80	1.795	.005	NIGERITE $(\text{Zn}, \text{Mg}, \text{Fe}) (\text{Sn}, \text{Zn})_2 (\text{Al}, \text{Fe})_{12}$ $0.22 (\text{OH})_2$	TRIG	---	Dark brown	H 8-9 G 4.25
1.802	1.740	.062	QUETZALCOATLITE $\text{Zn}_8 \text{Cu}_4 (\text{TeO}_3)_3 (\text{OH})_{18}$	HEX	10T0 fair	Blue	H 3 G 6.05
1.803	1.584	.219	RHOODOCHROSITE (Calcite grp) $(\text{Mn}, \text{Ca}) \text{CO}_3$	TRIG	10T1 perf	Pink	H 4 G 3.65 infus
1.753 1.816	1.804	.031	CYRILOVITE $\text{NaFe}_3 (\text{PO}_4)_2 (\text{OH})_4 \cdot 2\text{H}_2\text{O}$	TET	---	Yellow, brown, yellow-green	G 3.08 F 2-3
v 1.851	1.805	1.783	HÜGBOMITE $(\text{Mg}, \text{Fe})_2 (\text{Al}, \text{Ti})_5 \text{O}_{10}$	HEX and TRIG polytypess	10T0 good parting 0001	Brown, yellow	H 6.5 G 3.70 infus
1.711 1.84	1.81	.23	GASPEITE (Calcite grp) $(\text{Ni}, \text{Mg}) \text{CO}_3$	TRIG	10T1 perf	Green	H 3-4 G 3.6 infus
v 1.832	1.815	1.740	WATROJAROSITE (Alunite grp) $\text{NaFe}_3 (\text{SO}_4)_2 (\text{OH})_6$	TRIG plates	0001 perf	Yellow, brown	H 3 G 3.18 F 4.5
1.850	1.815	1.601	SMITHSONITE (Calcite grp) $(\text{Zn}, \text{Mg}) \text{CO}_3$	TRIG	10T1 perf	Milky white	H 4.5 infus
1.850	1.815	1.761	MOLYBDOPHYLLITE $\text{Pb}_2 \text{Mg}_2 \text{Si}_2 \text{O}_7 (\text{OH})_2$	TRIG lam	0001 perf	Cols to pale green	H 3-4 G 4.72 F diff
1.788 1.836	1.816	1.592	SIDERITE (Calcite grp) $(\text{Fe}, \text{Mg}, \text{Mn}) \text{CO}_3$	TRIG	10T1 perf	Brown	H 4 G 3.59 infus
1.803 1.836	1.816	1.597	RHOODOCHROSITE (Calcite grp) MnCO_3	TRIG	10T1 perf	Pink	H 4 G 3.70 infus
1.816	1.728	.088	HYDRONIUM JAROSITE (Alunite grp) $(\text{H}_3\text{O})\text{Fe}^{+3} (\text{SO}_4)_3 (\text{OH})_6$	TRIG	0001	Yellow, brown	H 4-4.5 G 2.5-2.9, 3.17 calc
1.816	1.788	.028	PAINITE $\text{CaZrBAI}_9 \text{O}_{18}$	HEX	---	Garnet-red	H 8 G 4.00
Insol in acids. Pleoc. 0 yellow-brown, E yellow. U opt pos.							
Diss by acids. Pleoc. 0 blue-green, E nearly col.							
Diss by hot acids. FeO 3.8, MgO 0.6, CaO 1.3%.							
Diss by hot acids. Pleoc faint, abs 0 > E.							
Insol in acids. Pleoc. 0 deep yellow, E yellow. TiO ₂ 9.1, FeO 7.8%. Anom biax.							
Diss by hot acids. NiO 36.8, MgO 14.8, FeO 4.3%.							
Diss by HC1. E1 pos. Pleoc in yellows, abs 0 > E. Anom biax. Basal sec- tions divided into seg- ments.							
Diss by hot acids. ZnO 53.2, MgO 7.6, CaO 1.1%.							
Col s in section. E1 pos.							
Diss by warm acids. MgO 11.8, MnO 5.9, CaO 1.4%.							
Diss by warm acids. Calc for pure end member.							
Diss by HCl.							
Insol in acids.							

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.791	1.820	1.715	.105	JAROSITE (Alunite grp) $KFe_3(SO_4)_2(OH)_6$	TRIG tab, rhombs	0001 dist	Yellow to brown	H 2.5-3.5 G 3.2 F 4.5	Diss by HCl. Pleoc in brown and yellow, abs 0 > E. Anom biax, with basal segments.
1.82	1.78	.04		NANLINGITE $CaMg_4(AsO_3)_2F_4$	TRIG	One imperf	Brownish-red	G 3.93	---
1.752	1.823	1.817	.006	BIRTHOLITE (Ce,Ca,Y) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX	---	Brown	H 5 G 4.77 infus	Gel with acids. Apatite structural type.
1.82- 1.83	---	---	ZAIRITE (Crandalite grp) $Bi(Fe,Al)_3(PO_4)_2(OH)_6$	TRIG	---	Greenish	G 4.37	---	
1.816 1.836	1.830 1.836	1.605 1.750	.225	OTAVITE (Calcite grp) $(Cd,Ca)CO_3$	10T1 perf	White, yellow	G ~ 4.9 infus	Diss by acids with eff. $CdCO_3$ 91, $CaCO_3$ 9%.	
1.815	1.832	1.750	.082	NATROJAROSITE (Alunite grp) $NaFe_3(SO_4)_2(OH)_6$	TRIG plates	0001 perf	Yellow, brown	H 3 G 3.18 F 4.5	Diss by HCl. El pos. Pleoc in yellow, abs 0 > E. Anom biax with basal seg- ments.
1.816 1.840	1.836 (1.610)	1.610	.226	RHODOCHROSITE (Calcite grp) $(Mn,Zn,Fe)CO_3$	TRIG	10T1 perf	Pink	H 4 G 3.76 infus	Diss by warm acids. MnO 29.8, FeO 13.9, ZnO 14.9, CaO 3.1%.
1.837	1.833	1.833	.004	JEANBANDYITE (Fe^{+3} ,Mn ⁺²) $Sn(OH)_6$	TET ps cub	001 and 100 fair	Brown-orange	H 3.5 G 3.81	Diss slowly by cold HCl.
1.836 1.849	1.840 1.849	1.615 (1.62)	.225	RHODOCHROSITE, ferroan (Calcite grp) $(Mn,Fe)CO_3$	TRIG	10T1 perf	Brown	H 4 G 3.72 infus	Diss by warm acids. MnO 35.4, FeO 26.1%.
1.81	1.84	1.84	.22	GASPEITE (Calcite grp) $NiCO_3$	TRIG	10T1 perf	Emerald-green	H 3-4 G 3.7 infus	Diss by hot acids. NiO 55.5, MgO 3.3, FeO 0.1%.
1.78	1.846	1.804	.042	HIBONITE (Ca,Ce)(Al,Ti,Fe) ₁₂ O ₁₉	HEX	0001 easy, parting 10T0	Brownish-black	H 7.5-8 G 3.88 infus	Insol in acids. Pleoc, abs E > 0.
1.840 1.860	1.849 1.860	1.615	.234	SIDERITE, manganeseoan (Calcite grp) $(Fe,Mn)CO_3$	TRIG	10T1 perf	Brown	H 3.5 G 3.81 infus	Diss by warm acids. FeO 47.9, MnO 9.5, MgO 3.2, CaO 0.2%.

1.815 Δ	1.850	1.623	.227	SMITHSONITE (Calcite grp) $ZnCO_3$	TRIG	10T1 perf	Cols, green	H 4.5 G 4.42 infus	Diss by acids. FeO 0.4, MnO 0.5%.
1.83 Δ	1.85	(~1.81)	~.04	BEAVERITE (Alunite grp) $Pb(Cu,Fe,Al)_3(SO_4)_2(OH)_6$	TRIG plates, u earthy	---	Canary-yellow	G 4.08-4.36	Diss by acids.
1.805 Δ	1.851	1.802	.049	HÜGBOMITE (Mg,Fe) ₂ (Al,Ti) ₅ O ₁₀	HEX and TRIG (polypytes)	10T0 good parting 0001	Brown, yellow	H 6.5 G 3.8. infus	Insol in acids. Pleoc, 0 dark brown, E med-brown. Anom biax.
1.855	1.60	.255		SPHAEROCCOBALTITE (Calcite grp) $CoCO_3$	TRIG	10T1 perf	Rose-red	H 3-4 G 4.1 infus	Diss by acids. Pleoc, 0 violet-red, E rose red.
1.849 Δ	1.860 1.875	1.625	.235	SIDERITE, zincian (Calcite grp) $(Fe,Mn,Zn)CO_3$	TRIG	10T1 perf	Brown	H 4 G 3.92 infus	Diss by hot acids. MnO 14.65, ZnO 11.55, MgO 0.3, FeO 36.4%.
1.86	1.860	1.83	.03	ASPECASITE $Ca_3(Ti,Si)As_6Be_2Si_2O_{20}$	TRIG	10T1 perf	Lemon-yellow	H 6.5-7 G 3.70	Anom biax, 2V 0-17°.
1.915	1.860	1.855	.005	BYSTROMITE $MgSb_2O_6$	TET u mass	---	Blue-gray	H 4 G 5.7	Insol in dil acids, diss by concd HCl + KI. Opt char unk.
1.870 Δ	1.898	1.792	.078	ARSENIOSIDERITE $Ca_3Fe_4(AsO_4)_4(OH)_6 \cdot 3H_2O$	MCL ps tet, fib	001 perf	Yellow, brown	H 4.5 G 3.5-3.6 F 3	Diss by acids. Pleoc, 0 brownish-red, E pale yellow.
1.870	1.845		.025	DUSSELTITE (Crandalite grp) $BaFe_3(AsO_4)_2(OH)_5 \cdot H_2O$	TRIG	0001	Yellow, green	H 3.5 G 3.75	Diss by acids. Anom biax, 2V 0-20°.
1.860 Δ	1.875	1.633	.242	SIDERITE (Calcite grp) $FeCO_3$	TRIG	10T1 perf	Brown	H 4 G 3.89 infus	Diss by hot acids. Calc for pure end-member.
1.875	1.785		.090	PLUMBOJAROSITE (Alunite grp) $PbFe_6(SO_4)_4(OH)_12$	TRIG plates	10T4 fair	Brown	H soft G 3.63	Diss by HCl. Pleoc, 0 brown-red, E pale golden-yellow.
1.882	1.785		.097	ARGENTOJAROSITE (Alunite grp) $AgFe_3(SO_4)_2(OH)_6$	TRIG plates	0001	Yellow, brown	G 3.66	Diss by HNO_3 . E1 poss. Pleoc in yellow, abs 0 > E.
1.870 Δ	1.898	1.815	.083	ARSENIOSIDERITE $Ca_3Fe_4(AsO_4)_4(OH)_6 \cdot 3H_2O$	MCL ps tet, fib	001 perf	Yellow, brown	H 4 G 3.5-3.6 F 3	Diss by acids. Pleoc, 0 brownish-red, E pale yellow.
1.91	(1.90)		.01	DAUBREEITE $Bio(OH,Cl)_1$	TET	001 perf	Yellow, brown	H 2-2.5 G 6.4-7.5 F easy	Diss by HCl. E1 clv poss.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.860	1.915	1.908	.007	BYSTROMITE $MgSb_2O_6$	TET u mass	---	Blue-gray	H 7 G 5.7	Insol in dil acids, diss by concd HCl + KI. Opt char unk.
1.957	1.916	1.909	.007	BEUDANTITE (Beudantite grp) $PbFe_3(AsO_4)(SO_4)(OH)_6$	TRIG	0001 easy	Olive, brown, yellow	H 4 G 4.0-4.3 F 3.5	Diss by acids. Pleoc, 0 Yellow, E pale yellow. Anom biax. As:S = 4:1.
	1.93	---	wk	CORKITE (Beudantite grp) $PbFe_3(PO_4)(SO_4)(OH)_6$	TRIG	0001 easy	Yellow, brown, greenish	H 4 G 4.2-4.3 F 4-5	Diss by HNO_3 . Abnormal green inter ¹ colors. Anom biax.
1.916	1.957	1.943	.014	BEUDANTITE (Beudantite grp) $PbFe_3(AsO_4)(SO_4)(OH)_6$	TRIG	0001 easy	Yellow, brown, black	H 4 G 4.0-4.3 F 3.5	Diss by HNO_3 . Pleoc, 0 yellow to red-brown, E cols to yellow. Anom biax. As:S = 1:1.
1.98	1.97	---	wk	DIXENITE $Cu^{+1}Mn_{14}Fe^{+3}(As^{+3}O_3)_5$ ($As^{+5}O_4$) $(SiO_4)_2(OH)_6$	TRIG plates	0001 mic	Black	H 3-4 G 4.20	Dec by HCl. Glowing red in transmitted light. El pos.
	1.977	1.967	.010	DUGGANITE $Pb_3(Zn,Cu)_3(Te^{+6}O_6)AsO_4(OH)_3$	HEX	---	Col to green	H 3 G 6.33	---
	1.98	1.85	.13	DIABOLEITE $Pb_2CuCl_2(OH)_4$	TET tab	001 perf	Sky-blue	H 2.5 G 5.42 F easy	Diss by HNO_3 . Deep blue, E nearly cols. El pos.
	1.99	---	mod	Unnamed (Crandaillite grp) $PbFe_3(AsO_4)_2(OH)_5H_2O$	TRIG	0001	Brown	---	Diss by HNO_3 . Anom biax. 2V 0-50°.
2.0 red	---	---	extr	GRAPHITE C	HEX and TRIG	0001 perf	Black	H 1-2 G 2.2 infus	Insol in acids. Thinnest flakes are deep blue in transmitted light. Strongly pleoc. El pos.
	2.0	(1.8)	.025	JAGOITE $Pb_3FeSi_3O_{10}(OH,Cl)$	TRIG	0001 perf	Yellow-green	H 3 G 5.43	El clv pos.
	2.00	2.14	.14	WAKEFIELDITE YVO_4	TET	---	Pale tan	H 5 G (4.26)	---
	2.01	1.99	.02	ARMANGITE $Mn_{26}As_{18}O_{50}(OH)_4CO_3$	HEX	0001 fair	Black	H 4 G 4.43 F easy	Diss by HCl. Streak brown.

2. ^v 2.041	2.026	1.965	.061	CUMENGTITE $Pb_4Cu_4Cl_8(OH)_8 \cdot H_2O$	TET ---	101 good 110 dist	Indigo-blue H 2.5 G 4.67 F 1
	>2.0	>2.0	str	KUSUITE (Ce ⁺³ , Pb ⁺² , Pb ⁺⁴)V ₀₄	TET ---	Black	H 4.5 G 5.30 calc
>2.0	>2.0	---	---	DREYERITE BiV ₀₄	TET plates ---	Orange- to brownish-yellow	Pleoc., 0 bright yellow, E brownish-yellow.
2.026	2.010	.016	HEDYPHANE (Apatite grp) (Ca, Pb) ₅ (AsO ₄) ₃ Cl	HEX (mc1, ps hex?)	10T1 ---	White	H 4.5 G 5.82 F 2
2.03	2.00	.03	PSEUDOBOLEITE $Pb_5Cu_4Cl_{10}(OH)_8 \cdot 2H_2O$	TET ps cub ---	001, 101 perf	Indigo-blue H 2.5 G 4.85 F 1	Diss by warm HNO ₃ * luster on cleav. Pearly
2.033	2.015	.018	BARYSTILITE $Pb_8Mn(Si_2O_7)_3$	TRIG tab ---	0001 dist	Gray, white H 3 G 6.72 F 2.5	Gel with acids. MnO 3.3, MgO 0.8, ZnO 0.3%.
2. ^v 7	2.04	2.00	.04	SIMPSONITE Al ₄ (Ta, Nb) ₃ (O, OH, F) ₁₄	TRIG ---	0001 white, yellow	H 7 G 6.6-7.3 infus
2. ^v 6 [^]	2.041	1.926	.115	CUMENGTITE $Pb_4Cu_4Cl_8(OH)_8 \cdot H_2O$	TET ---	101 good 110 dist	Indigo-blue H 2.5 G 4.67 F 1
2.05	2.03	.02	BOLEITE $Pb_{26}Ag_{10}Cu_{24}Cl_{62}(OH)_48 \cdot 3H_2O$	TET ps cub ---	001 perf 101 good	Indigo-blue H 3-3.5 G 5.05 F 1	Diss by warm HNO ₃ * Pearly luster on cleav. Trillings on 001.
2.05	---	very wk	EULYTITE $Bi_4Si_3O_{12}$	CUB tetrah ---	110 imperf	Gray, brown H 4-4.5 G 6.62 F 2	Gel with acids.
2.055	1.975	.080	GRIMALDITE CrO(OH)	TRIG tab ---	Deep red	G 4.12	Data for synth compd.
2.05	2.057	2.046	.011	PYROMORPHITE (Apatite grp) $Pb_5(PO_4)_3Cl$	HEX prts ---	Green, yellow, brown	H 3.5-4 G 7.0-7.1 F 1.5
>2.05	---	---	---	TET plates ---	Brown to yellow	H "not too high" G 5.80	---
2.06	2.05	.01	KETTNERITE $CaBiO(CO_3)_F$	---	Dark yellow	H 3 G 3.9	---

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.033 ^Λ	2.07	2.05	.02	BARYSILITE $Pb_8Mn(Si_{27})_3$	TRIG tab	0001 dist	Gray, white	H 3 G 6.72 F 2.5	Gel with acids.
2.09		1.94	.15	HYDROGERUSSITE $Pb_3(CO_3)_2(OH)_2$	TRIG plates	0001 perf	Col	H 3.5 G 6.80 F 1.5	Diss by HNO_3 . Data for Synth compd. El pos.
2.125		2.059	.074	YEDLINITE $Pb_6Cr(O,OH)_8Cl_6$	TRIG	1120 dist	Red-violet	H 2.5 G 5.85	Pleoc, 0 pale cobalt blue, E lavender.
2.11 ^Λ 2.15	2.13	1.99	.14	BEYERITE $(Ca,Pb)Bi_2(CO_3)_2O_2$	TET tab	Conch	Yellow, green	H 2-3 G 6.56 F 1.5	Diss by acids.
2.057 ^Λ 2.18	2.135	2.12	.015	MIMETITE (Pyromorphite grp) $Pb_5(AsO_4)_3Cl$	MCL ps hex	10T1 imperf	Yellow, brown, gray	H 3.5-4.5 G 7.1-7.24 F 1	Diss by HNO_3 . Commonly biax. El clv pos.
2.145		2.006	.14	MATLOCKITE $PbFCI$	TET tab	001 perf	Yellow, greenish	H 2.5-3 G 7.12 F 1	Anom Diss by warm HNO_3 . El clv pos.
2.15		--	str	BISMOCOLITE $BiOCI$	TET	001 perf	Yellow, brown	H 2-2.5 G 7.4-7.6 F 1	Diss by acids. El clv pos.
2.155		1.975	.180	GRIMALDIITE $CrO(OH)$	TRIG	---	Deep red	H 4.12 synth infus	---
2.155		2.120	.035	PARAKHINITE $PbCu_3TeO_4(OH)_6$	HEX	---	Dark green	H 3.5 G 6.7 calc	---
>2.10		>2.10	mod to str	SCHUETTEITE $Hg_3(SO_4)_2$	HEX	---	Yellow	H 3 G 8.2 volat	Diss by acids. Pleoc wk, 0 greenish-yellow, E orange yellow.
2.18		--	--	RUSSELLITE Bi_2WO_6	TET mass	---	Yellow	H 3.5 G 7.37	Opt char unk.
2.21		--	wk	ZAVARITSKITE $BiOF$	TET	---	Gray	G (9.2)	Opt sign unk.
2.25 ^ν	2.20		.05	VANADINITE, arsenian (Apatite grp) $Pb_5(VO_4)_3Cl$	HEX prs	---	Yellow	H 3 G 7.0 F 1.5	Diss by HNO_3 . El neg.

2.26	2.10	.16	HYDROHETAEROLITE $Zn_2Mn_4O_8 \cdot H_2O$	TET fib	001 perf	Brownish-black	H 5-6 G 4.6 infus	Diss by HCl. Pleoc wk. O red-brown, E nearly opaque.
2.27	2.18	.09	STOLZITE Pb_3WO_4	TET pyram	001 imperf	Green, brown	H 2.5-3 G 7.9-8.4 F 2	Dec by HCl. Pleoc, O pale greenish-yellow, E bright yellow.
2.30	2.15	.125	RANCIEITE $(Ca, Mn^{+2})Mn^{+4}O_9 \cdot 3H_2O$	HEX (?) mass	---	Dark brown, streak purplish		Diss by HCl. Opt char unk.
2.295	2.285	.01	FINNEMANITE $Pb_5(AsO_3)_3Cl$	HEX pris	10T1 dist	Olive-green, gray, black	H 2-3 G 7.27 F 2	Diss by HNO_3 . El neg.
2.30	2.07	.23	HETAEROLITE $ZnMn_2O_4$	TET	001 indist	Black	H 6 G 5.2 infus	Diss by HCl. Red-brown in section. Pleoc wk, abs O > E.
2.35	2.30 Li	.07	WULFENITE, tungstenian $Pb(Mo,W)O_4$	TET	011 dist	Yellow	H 3 G 6.6 F 2	Dec by acids. MoO ₃ 21%. W/Mo = 0.7.
2.27	2.30 Li	2.23 Li	PLATTNERITE PbO_2	TET	110	Black	H 5-5.5 G 8.5-9.5 F 2	Diss by HCl with evolution Cl ₂ . Nearly opaque. Basal section shows biax segments.
2.35	2.30 Li	---	---	---	---	Straw-yellow	H ~ 4	---
2.32	2.12	.20	SANTANAITE $9PbO \cdot 2PbO_2 \cdot CrO_3$	HEX	0001 perf 1210	Yellow, green	H 2.5-3 G 7.14 F 1.5	Diss by HNO_3 . In part biax.
2.32 Li	2.25 Li	.07	ECDEMITE $Pb_6As_2O_7Cl_4$	TET tab	001 dist	Brownish-black	H 6 G 3.8-4.1 infus	Diff diss by HCl. Pleoc wk in red-brown, red, or purple.
2.33	1.96	.37	GEIKIELITE $(Mg,Fe)TiO_3$	TRIG rhombs	10T1 perf			Diss by HCl. Red-brown in section. Pleoc wk, abs O > E.
2.31 2.35	2.33	2.12	HETAEROLITE $ZnMn_2O_4$	TET	001 indist	Black	H 6 G 5.18 infus	Diss by HNO_3 . UO ₃ 11.6%.
2.30	2.35	.23	WULFENITE, uranian $Pb(Mo,U)O_4$	TET tab	011 dist	Yellow	H 3 G 6.6 F 2	Diss by HNO_3 . UO ₃ 11.6%.
2.30 2.47	2.35 Li	2.33 Li	.06 "LORETTITOITE" $Pb_7O_6Cl_2$	TET (?) mass	001 perf	Orange-yellow	H 3 G 7.95 F 1	Diss by HNO_3 . Probably synth material (?).
2.40	2.25 Li	.11	SCHWARTZEMBERGITE $Pb_5(TiO_3)Cl_3O_3$	TET or ps tet	001 dist	Honey-to straw-yellow, reddish	H 2.5 G 7.39 F 1	Diss by HNO_3 . Shows biax sectors.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index ω	Refractive index ϵ	Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.36 Li	2.31 Li	.05	wk	LANGBANIITE $(\text{Mn}^{+2}, \text{Sb}, \text{Ca})_4 (\text{Mn}^{+4}, \text{Fe})_3 \text{SiO}_{12}$	IRIG	---	Iron-black	H 6.5 G 4.6-4.8 infus	Diss by HCl. Pleoc wk in reddish-brown, abs 0 > E.
2.41 Li	---			MINIUM Pb_3O_4	TET u mass	---	Brownish-red	H 2.5 G 9.1 F 2	Diss by acids. Abn green interf colors. Pleoc, deep red-brown to nearly cols.
2.43	2.37	.06		VANADINITE (Apatite grp) $\text{Pb}_5(\text{VO}_4)_3\text{Cl}$	HEX pris	---	Red, yellow, brown	H 3 G 6.65-6.98	Diss by HNO_3 . El neg.
2.45	2.33	.12		TAPIOLITE $\text{Fe}(\text{Ta}, \text{Nb})_2\text{O}_6$	TET	---	Black	H 6-7 G 7.11 infus	Pleoc str, dark red to black. U opt pos.
2.46 Li 2.40	2.15 Li 2.15	.31		HAUSMANNITE $\text{Mn}^{+2}\text{Mn}^{+3}$ Mn_2^{+4}	TET oct	001 good	Brownish-black	H 5.5 G 4.84 infus	Diss by HCl. In section, brownish-red, not pleoc.
2.35	2.47	.21		WULFENITE PbMoO_4	TET tab	011 dist 001, 013 less so	Yellow, orange	H 3 G 6.7 F 2	Diss by HNO_3 .
2.48 Na 2.46	2.21 Na 2.07	.27 .39	wk to mod	PYROPHANTITE MnTiO_3	TRIG	0221 perf 1012 less so	Deep blood-red	H 5 G 4.54 infus	Nearly insol in HCl. In section yellow-red.
2.50 Li	---			SENAITE $\text{Pb}(\text{Ti}, \text{Fe}, \text{Mn})_{21} \text{O}_{38}$	TRIG	---	Black	H 6 G 5.30	Nearly opaque. Not pleoc. Tw pl 1120 common.
2.55 2.60	2.49	.07		ANATASE TiO_2	TET oct	001, 011 perf	Brown to black	H 5.5-6 G 3.90 infus	Insol in acids. Pleoc in thick sections, 0 pale blue to yellow, E dark blue to orange.
2.6 Li	---		extr	TRECHMANNITE AgAsS_2	TRIG	1011 good 0001 dist	Scarlet- vermillion	H 1.5-2 F 1	Streak scarlet. Pleoc, 0 pale red, E col.
2.60	2.50	.10		BRAUNITE $3\text{Mn}_2^{+3} \cdot \text{MnSiO}_3$	TET pyram	112 perf	Brownish-black	H 6-6.5 G 4.2-4.83 infus	Dec by HCl.
2.665Li	2.535Li	.130		LITHARGE PbO	TET tab	110 perf	Red	H 2 G 9.13 F 1.5	Diss by HNO_3 . Fuses to a yellow bead. El clv neg.

2.70	1.75	.95	CHALCOPHANITE $(\text{Zn}, \text{Mn})\text{Mn}_3\text{O}_7 \cdot 3\text{H}_2\text{O}$	TCL	001 perf	Brownish-black	H 2.5 G 4.0 infus	Diss by HCl. Pleoc str, 0 nearly opaque, E deep red.
2.98 Li 3.09 Na	2.71 Li 2.79 Na	.27 .30	PROUSTITE Ag_3AsS_3	TRIG	10T1 dist	Scarlet	H 2-2.5 G 5.57 F 1	Dec by HNO_3 . Streak scarlet. Bright red in section, pleoc wk, 0 blood- red, E cochineal-red.
3.01 Li 3.22 Na	2.78 Li 2.94 Na	.23 .28	HEMATITE Fe_2O_3	TRIG	Parting 0001	Red to black	H 5-6 G 5.26 infus	Diss by HCl. Streak red. Abs 0 > E.
3.08 Li	2.88 Li	.20	PYRARGYRITE Ag_3SbS_3	TRIG	10T1 dist	Dark red	H 2.5 G 5.85 F 1	Dec by HNO_3 . Streak purpleish-red. Red in section. Tw p1 1120.
3.25	2.40	.85	CREDNERITE CuMnO_2	MCL ps hex	001, 100 perf III good	Iron-black	H 4 G 5.0 F diff	Diss by HCl. Poly tw.
3.30	2.90	.40	HOLLANDITE $\text{Ba}(\text{Mn}^{+4}, \text{Mn}^{+2})_8\text{O}_{16}$	MCL ps tet	---	Black, gray	H 6.95 G 2.95 infus	Diss by HCl.
3.78	3.75	.03	GRATONITE $\text{Pb}_9\text{As}_4\text{S}_{15}$	TRIG	Conch	Lead-gray	H 2.5 G 6.22 F 1	---
4.34 Li	2.03 Li	2.31	MOLYBDENITE MoS_2	HEX	0001 perf	Lead-gray	H 1-1.5 G 4.7 infus	Diss by HNO_3 . E1 clv pos.

Table 6. Biaxial positive minerals

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_z$ calc ($2V_z$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.301	<u>1.301</u>	1.307	.006	FERRUCITE NaBF_4	11°	$X = \frac{b}{c}$ $Z = \frac{v}{c}$	ORTH tab	100, 010 001	cols	H 3 G 2.50 F 1
1.338	<u>1.338</u>	1.339	.001	CRYOLITE Na_3AlF_6	43° $r < v$	$X = \frac{b}{c}$ $Z:c = -44^\circ$	MCL	Parting 110, 001	White, reddish, brownish	H 2.5 G 2.97 F 2
1.346	<u>1.348</u>	1.350	.004	WEBERITE $\text{Na}_2\text{MgAlF}_7$	80°	$X = \frac{a}{b}$ $y = \frac{v}{b}$	ORTH	101 poor 010	Gray	H 3.5 G 2.96
1.409	<u>1.412</u>	1.419	.010	PACHNOLITE $\text{NaCaAlF}_6 \cdot \text{H}_2\text{O}$	76° $r < v \text{ wk}$	$X = \frac{b}{c}$ $Z:c = 53^\circ$ -69° disp str	MCL	001 fair	White	H 3 G 2.98, 2.97 calc
1.411	<u>1.416</u>	1.422	.011	CARLHINTZITE $\text{Ca}_2\text{AlF}_7 \cdot \text{H}_2\text{O}$	77°	---	TCL pris	---	White, cols	TW [010].
1.413	<u>(1.417)</u>	1.423	.010	YAROSLAVITE $\text{Ca}_3\text{Al}_2\text{F}_{10}(\text{OH})_2 \cdot \text{H}_2\text{O}$	74°	---	ORTH fib	White	G 3.09	---
1.425	<u>1.428</u>	1.438	.013	CALCJARLITE $(\text{Na}, \text{K})\text{Ca}_3\text{Al}_3(\text{F}, \text{OH})_{16}$	72°	$Z:c = 10^\circ$ -15°	MCL	Uneven	cols	H 4 G 3.51
1.429	<u>1.433</u>	1.435	.006	JARLITE $\text{NaSr}_3\text{Al}_3(\text{F}, \text{OH})_{16}$	~ 90°	$Y = \frac{b}{c}$ $X:c = -6^\circ$	MCL tab	---	cols	H 4-4.5 G 3.78-3.93
1.422	<u>1.435</u>	1.480	.058	SANTITE $\text{KB}_5\text{O}_6(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	70° $(58+5^\circ)$	$X = \frac{c}{b}$ $y = \frac{v}{b}$	ORTH	010 perf 100 dist	cols	H 2.5 G 1.74
1.431	<u>1.438</u>	1.488	.057	SBORGITE $\text{NaB}_5\text{O}_6(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	35° $(42+6^\circ)$ $r > v \text{ wk}$	---	MCL	---	cols	H 3.5 G 1.71
1.439	<u>1.442</u>	1.469	.030	STERCORITE $\text{HNa}(\text{NH}_4)\text{PO}_4 \cdot 4\text{H}_2\text{O}$	36° $r > v$ rather str	$Z = \frac{b}{c}$ $y:c = 30^\circ$ disp str	TCL	---	cols	H 2 G 1.6 F 1
1.441	<u>1.442</u>	1.444	.003	USOVITE $\text{Ba}_2\text{CaMgAl}_2\text{F}_{14}$	70°	---	MCL	One, perf	Brown	H 3.5 F easy

Pleoc., X and Y
brownish-yellow, Z
pale yellow.

1.445	1.460	1.491	.046	MERCALLITE KHSO ₄	56° (71+5°) r < v wk	X = b Z = $\frac{a}{d}$	---	G 2.31 F 1	Sol in H ₂ O, giving an acid solution.
1.456	1.460	1.480	.024	CHESSEXITE Na ₄ Ca ₂ (Mg,Zn) ₃ Al ₈ (SO ₄) ₁₀ (SiO ₄) ₂ (OH) ₁₀ • 40H ₂ O	47°	X = $\frac{a}{d}$ Z = $\frac{b}{d}$	ORTH 010	White	Diss by HCl.
1.460	1.461	1.470	.010	ALUNOGEN Al ₂ (SO ₄) ₃ •17H ₂ O	31°	X ~ b Z:c = 42°	010 perf tab or fib c	White	H 1.5-2 G 1.77 infus
1.454	1.461	1.471	.017	DORFMANITE Na ₂ HPO ₄ •2H ₂ O	65° (80+14°) r > v wk	---	ORTH 010	Uneven to subconch	Sol in H ₂ O. In closed tube, melts in its water of crystn.
1.458	1.461	1.471	---	---	---	---	ORTH	White	Sol in H ₂ O, gives an alk reaction.
1.461	1.461	1.471	---	---	---	Y = b X:a = -1°	20T perf	White	Dec by HCl. Opt sign unk.
1.461	1.463	1.476	.015	SILHYDRITE 3SiO ₂ •H ₂ O	48° r > v	---	MCL fib	White	H 2.5 G 2.10 F 2.
1.461	1.463	1.476	.015	PICROMERITE K ₂ Mg(SO ₄) ₂ •6H ₂ O	---	---	MCL fib	White	H 1-2 G 1.66-1.8 infus
1.459	1.464	1.470	.011	ALUMINITE Al ₂ SO ₄ (OH) ₄ •7H ₂ O	Med to large (85+21°)	X = e!	---	White	Diss by acids.
1.462	1.465	1.474	.012	MALLARDITE (Melanterite grp) MnSO ₄ •7H ₂ O	Large	Z:c = 44°	MCL 001 dist	Rose	Sol in H ₂ O.
1.462	1.466	1.469	.007	BOGGILDITE Na ₂ Sr ₂ Al ₂ (PO ₄)F ₉	79°	---	MCL ---	Flesh-red	H 4-5 G 3.66 F 1-2
1.464	1.468	1.474	.010	KONYAITE Na ₂ Mg(SO ₄) ₂ •5H ₂ O	106° (79+24°)	---	MCL ---	Cols	H 2.5 G 2.097
1.456	1.469	1.508	.052	LANSFORDITE MgCO ₃ •5H ₂ O	60°	X = b Z ~ $\frac{c}{e}$	MCL tab	001 perf less so	Diss by acids. Alters in air to Nesquehonite.
1.468	1.470	1.473	.005	ERIONITE (Zeolite grp) (K ₂ ,Ca,Na ₂) ₂ Al ₄ Si ₁₄ 0.36•15H ₂ O	---	eI pos	HEX fib	---	G 2.05 F easy
1.468	1.470	1.474	.006	TRIDYMITE SiO ₂	40°	X = b Z = $\frac{c}{e}$ el neg	MCL ps hex, tab	Cols	Insol in acids, diss by hot Na ₂ CO ₃ soln. Tw, trillings, fourlings.
1.484	V	1.484							

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks		
	α	β	γ										
1.478	1.467	1.471	1.476	.009	MELANTERITE (Fe,Mg)SO ₄ ·7H ₂ O	Large r > v	$Y = \frac{b}{a}$ $Z:C = 77^\circ$	MCL el c	T01 perf 111 poor	Greenish	H 2 G 1.76 F easy	Sol in H ₂ O. MgO 7.45, CuO 0.3, ZnO 0.5%.	
1.491(?)	1.470	1.471	1.479	.009	BOUSSINGAULTITE (NH ₄) ₂ Mg(SO ₄) ₂ ·6H ₂ O	51° r > v perc	$Y = \frac{b}{a}$ $Z:C = 12^\circ$	MCL	20T perf	Cols	H 2 G 1.72 F 1	Sol in H ₂ O. Con- tinuous Series to Mohrite?	
	1.469	1.473	1.491	.022	META-ALUNOGEN Al ₄ (SO ₄) ₆ ·27H ₂ O	Large (51+12°)	---	---	---	White	---	---	
	1.458	1.473	1.501	.043	BARRINGTONITE MgCO ₃ ·2H ₂ O(?)	(74+6°)	el pos	TCL fib	---	cols	G 2.82 (calc)	Optics are discordant for compn given.	
	1.466	1.474	1.495	.029	CARNALLITE KMgCl ₃ ·6H ₂ O	70° r < v	$Z = \frac{a}{c}$ $X = \frac{c}{a}$	ORTH	Conch	White, reddish	H 2.5 G 1.62 F 1-1.5	Sol in H ₂ O. Deliq.	
	1.473 1.477	1.469	1.475	.015	THEWARDITE Na ₂ SO ₄	83° r > v wk	$X = \frac{c}{b}$ $Y = \frac{a}{b}$	ORTH	010 perf 101 fair	White	H 2.5-3 G 2.66 F 2	Sol in H ₂ O.	
	---	1.47- 1.48	---	wk	HYDROBASALUMINITE Al ₄ SO ₄ (OH) ₁₀ ·12-36H ₂ O	---	---	---	cols to yellowish	---	G 1.86	Diss by acids. Opt char unk.	
	1.472	---	1.487	.015	MAKAITE Na ₂ Si ₄ O ₉ ·5H ₂ O	---	---	---	White	---	G 2.07	Dec by acids. Opt char unk.	
	1.472	1.475	1.477	.005	MORDENITE (Zeolite grp) (Ca,K ₂ ,Na ₂)Al ₂ Si ₁₀ ²⁴ · 7H ₂ O	80°	$X = \frac{c}{a}$ $Y = \frac{a}{c}$	ORTH or MCL spher	---	cols to pink, yellow	---	H 3-4 G 2.12 F 4-5	Partly dec by HC1.
	1.482	1.471 1.483	1.486	.015	MELANTERITE FeSO ₄ ·7H ₂ O	85° r > v wk	$Y = \frac{b}{a}$ $Z:C = 77^\circ$	MCL el c	T01 perf 111 poor	Green	H 2 G 1.90 F easy	Sol in H ₂ O.	
	1.461	1.475	1.482	.010	ALUNOGEN Al ₂ (SO ₄) ₃ ·17H ₂ O	75°	$X \sim \frac{b}{a}$ $Z:C = 42^\circ$	TCL tab or fib c	010 perf 100, 3T3	White	H 1.5-2 G 1.79 infus	In closed tube, melts in its water of crystn.	
	1.489	1.478	1.479	.004	FERRIERITE (Zeolite grp) (Na,K) ₂ MgAl ₃ Si ₁₅ O ₃₆ (OH)·9H ₂ O	50°	$X = \frac{a}{b}$ $Y = \frac{b}{a}$ el pos	ORTH blades 100	100 perf	White	H 3 G 2.15 F 3-4	Insol in HCl.	

1.478	1.479	1.481	1.481	.003	CLINOPTILOLITE (Zeolite grp) (Na, K, Ca) ₂₋₃ Al ₃ (Al, Si) ₂ Si ₁₃ O ₃₆ •12H ₂ O	40° r > v	X = b el cTw pos	MCL 010 perf	Cols 010 perf	H 4 G 2.11 F 2	
1.47	1.48	1.49	1.49	.02	BOOTHITE CuSO ₄ •7H ₂ O	Large	Y = b x ~ c el pos	MCL fib c imperf	Blue	H 2-2.5 G 2.1 F 3	
1.475	1.480	1.488	1.488	.013	DIETRICHITE (Zn, Fe, Mn)Al ₂ (SO ₄) ₄ • 22H ₂ O	Large	X = b Z:c = 29°	MCL fib c imperf	White	H 2 infus	
1.477	1.480	1.490	1.490	.013	NATROLITE (Zeolite grp) Na ₂ Al ₂ Si ₃ O ₁₀ •2H ₂ O	63° (71+16°)	X = a Z = c el pos	ORTH acic c imperf	110 perf 010 imperf	White	H 5 G 2.22 F 2
1.475	1.480	1.490	1.490	.015	MAKATITE Na ₂ Si ₄ O ₉ •5H ₂ O	Large	Z = b X:c = 29°	ORTH spher	White	H 4.5 G 2.07	
1.475	1.480	1.487	1.487	.012	MISENITE K ₂ SO ₄ •6KHSO ₄ (?)	~ 0°	---	MCL fib c imperf	010 dist	White	G 2.32 F 1.5
---	1.480	---	---	---	FAUJASITE (Zeolite grp) (Na ₂ , Ca)Al ₂ Si ₄ O ₁₂ •8H ₂ O	~ 0°	---	CUB oct	111 dist	White	H 5 G 1.92 F 3
1.480	(1.480)	1.485	1.485	.005	CHABAZITE (Zeolite grp) (Ca, Na ₂)Al ₂ Si ₄ O ₁₂ •6H ₂ O	Small	---	TRIG rhombs	1011 dist	White, red	H 4.5 G 2.06 F 3
1.485	1.486	1.481	1.485	.005	DACHIARDITE (Zeolite grp) (K ₂ , Na ₂ , Ca) ₅ Al ₁₀ Si ₃₈ 0.96•25H ₂ O	55-60°	X = b Z:c = 35°	MCL	100, 001 perf	White	H 4-4.5 G 2.16 fus
1.480	1.481	1.481	1.481	.027	RIVADAVITE Na ₆ MgB ₂₄ O ₄₀ •22H ₂ O	80° r > v	Y = b Z:a = -32°	MCL el b poor	100, 101 perf	White	H 3.5 G 1.91 F 1.5
1.470	1.481	1.497	1.497	.023	NICKEL-HEXYHYDRITE (Ni, Mg, Fe)SO ₄ •6H ₂ O	---	Z:c = 45°	MCL crusts, fib	010 perf 100	Blue- green	---
1.470	---	1.493	1.493	.023	GMELINITE (Zeolite grp) (Na ₂ , Ca)Al ₂ Si ₄ O ₁₂ •6H ₂ O	~ 0°	---	HEX	1010 dist	White	H 4.5 G 2.0-2.1
---	1.482	---	1.482	.002- .008	MORDENITE var. Ashtonite (Zeolite grp) (Ca, Na ₂)Al ₂ Si ₁₀ O ₂₄ •7H ₂ O	Large	X = c Y = a	ORTH radiating	100 good 010 fair	White to red	H 3-4 G 2.12 F 4-5
1.481	1.481	1.486	1.486	.005	BAYLISSITE K ₂ Mg(CO ₃) ₂ •4H ₂ O	64°	---	MCL	Conch	Cols	H 2-3 G 2.01
1.462	(1.483)	1.531	1.531	.069							Diss by acids with eff. Dec by H ₂ O.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.478 ^Λ	1.479	1.483	1.488	.009	ZINC-COPPER MELANTERITE (Zn, Cu)SO ₄ ·7H ₂ O	Large disp wk	---	MCL fib	---	Light-blue-green	H 2 G 2.02 F easy
1.480	1.483	1.486	1.486	.006	LEONITE K ₂ Mg(SO ₄) ₂ ·4H ₂ O	~ 90°	$\gamma = b$ $z:a$ small	MCL prts	Conch	Cols, yellow	H 2.5-3 G 2.20 F easy
1.468	1.484	1.515	1.515	.083	CAICLACITE CaCl ₂ ·Ca(C ₂ H ₃ O ₂) ₂ ·10H ₂ O	80° (72+5°)	---	MCL or TCL	---	Cols	G 1.5
1.470 ^Λ	1.482	1.484	1.486	.004	TRIDYMITE SiO ₂	86°	$\chi = b$ $z:c$	MCL ps hex, tab	---	Cols	H 6.5 G 2.25 infus
1.503	1.483	1.485	1.487	.004	PHILLIPSITE (Zeolite grp) (K ₂ , Ca, Na ₂) ₁₋₂ (Al, Si) ₈	63° $r < v$ mod	$\chi = b$ $z:c$ = 19° el pos	MCL fib a	100, 010 dist	White	H 4 G 2.19 F 3
(1.480) 1.492 ^Λ	1.485	1.485	1.488	.003	CHABAZITE (Zeolite grp) (Ca, Na ₂) ₂ Al ₂ Si ₄ O ₁₂ ·6H ₂ O	Med to large	---	TRIG	10T dist	Cols, white, red	H 4.5-5 G 2.1 F 3
1.472	---	1.485	---	---	MAGADITE NaSi ₇ O ₁₃ (OH) ₃ ·3H ₂ O	low	el pos	MCL	---	White	---
1.484	1.486	1.502	1.502	.018	INDIGIRITE Mg ₂ Al ₂ (CO ₃) ₄ (OH) ₂ ·15H ₂ O	---	el pos	fib	---	Snow-white	H 2 G 1.6
1.483 ^Λ	1.486	1.495	1.495	.012	CYANOMORPHITE K ₂ Cu(SO ₄) ₂ ·6H ₂ O	47° $r < v$ str	$\gamma = b$ $x:c$ = 19°	MCL crusts	20T perf	Greenish-blue	G 2.22
1.470	1.487	1.540	1.540	.070	NATROLITE (Zeolite grp) Na ₂ Al ₂ Si ₃ O ₁₀ ·2H ₂ O	Med (60+20°) $r < v$ wk	$x = a$ $z = c$ el pos	ORTH acetic c 010 poor	110 perf white	H 5 G 2.22 F 2	
					AMMONIOMORBITE (NH ₄) ₂ B ₁₀ O ₁₆ ·5H ₂ O	59° $r < v$ wk	$y = b$ $z:c$ = 7°	MCL	---	White	Soft G 1.77 volat

1.476	---	1.500	.024	UKLONSKOVITE $\text{Na}_2\text{Mg}_2(\text{SO}_4)_2(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	---	---	MCL Y = $\frac{b}{c}$ X:c = 5°	T01, T11 good 010 perf	ColS 010 perf	H 3 G 2.07 fus	Diss in H_2O . Poly tw.	
1.484	<u>1.487</u>	1.496	.012	TAMARUGITE $\text{NaAl}(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$	60° disp wk	---	MCL Z = $\frac{b}{c}$ el cTV neg	010 perf	White	H 4 G 2.14 F 2	Dec by HCl. SiO_2 61.8, CaO 5.5, Na_2O 1.1, K_2O 0.3%.	
1.488	<u>1.488</u>	1.489	.001	HEULANDITE (Zeolite grp) (Ca, Na_2) ₂₋₃ $\text{Al}_3(\text{Al}, \text{Si})_2$ $\text{Si}_{13}0_{36} \cdot 12\text{H}_2\text{O}$	70° $r > v$	---	MCL X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el pos	ORTH blades 100	100 perf	White	H 3 G 2.21 F 3-4	Insol in HCl.
1.479	<u>v</u> <u>1.489</u>	1.489	.001	FERRERITE (Zeolite grp) (Na, K) ₂ $\text{MgAl}_3\text{Si}_{15}0_{36}$ (OH) _{9H_2O}	Small	---	ORTH blades 100	100 perf	White	H 3 G 2.21 F 3-4	Insol in acids.	
1.489	<u>1.491</u> <u>1.496</u>	1.491	.029	FLUELLITE $\text{Al}_2(\text{PO}_4)\text{F}_2(\text{OH}) \cdot 7\text{H}_2\text{O}$	85° $r < v$	---	ORTH X = $\frac{b}{c}$ Y = $\frac{b}{c}$	010, 111 indist	White	H 3 G 2.16 infus	Insol in acids.	
1.480	1.491	1.509	.029	INDERITERITE $\text{MgB}_3(\text{OH})_5 \cdot 5\text{H}_2\text{O}$	37°	Z:c = 9°	MCL ---	110 good	ColS 010	H 2.5 G 1.79 F 1	Diss by acids.	
1.488	1.491	1.505	.017	MOHRITE $(\text{NH}_4)_2\text{Fe}(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$	~75°	Y = $\frac{b}{c}$	MCL ---	20T good	Pale bluish-green	G 1.86	Sol in H_2O . Con- tinuous Series to Boussingaultite?	
1.471(?)	<u>1.487</u>	1.491	.012	CHABAZITE (Zeolite grp) (Ca, Na_2) $\text{Al}_2\text{Si}_40_{12} \cdot 6\text{H}_2\text{O}$	58°	---	TRIG X = $\frac{b}{c}$ el cTV pos	10T1 dist	ColS, red, white	H 4-5 G 2.1 F 3	Dec by HCl. Tw pl 10T1.	
1.485	<u>1.491</u>	1.492	.003	CLINOPTILOLITE (Zeolite grp) ($\text{Na}_2, \text{Ca}, \text{K}_2$) ₂₋₃ Al_3 $(\text{Al}, \text{Si})_2\text{Si}_{13}0_{36} \cdot 12\text{H}_2\text{O}$	72° $r > v$	---	MCL X = $\frac{b}{c}$ el cTV pos	010 perf	ColS 010	H 4 G 2.1 F 2	Dec by HCl. SiO_2 59.7, CaO 3.4, Na_2O 1.3, K_2O 4.2%.	
1.479	<u>v</u> <u>1.491</u>	1.493	.006	CANAVESITE $\text{Mg}_2(\text{CO}_3)(\text{HBO}_3) \cdot 5\text{H}_2\text{O}$	Very large disp wk	Z = $\frac{b}{c}$	MCL fib	---	Milky white	G 1.8	---	
1.485	1.494	1.505	.020	ARCANITE K_2SO_4	67° $r > v$ mod	X = $\frac{b}{c}$ Z = $\frac{c}{d}$	ORTH X = $\frac{b}{c}$ Z = $\frac{c}{d}$	010, 001 good	White	G 2.66 F 3	Sol in H_2O . Tw common.	
1.494	<u>1.495</u>	1.497	.003	STRUVITE $(\text{NH}_4)\text{MgPO}_4 \cdot 6\text{H}_2\text{O}$	37° $r < v$ str	---	ORTH X = $\frac{b}{c}$ Z = $\frac{c}{d}$	001 good 100 poor	ColS to yellow	H 2 G 1.71	Diss by acids.	
1.495	<u>1.496</u>	1.504	.009	DACHIARDITE (Zeolite grp) ($\text{K}_2, \text{Na}_2, \text{Ca}$) ₅ $\text{Al}_{10}\text{Si}_{38}$ $0_{96} \cdot 25\text{H}_2\text{O}$	69°	X = $\frac{b}{c}$ Z:c = 35°	MCL ---	100, 001 perf	White	H 4-4.5 G 2.17 fus	Dec by acids. Magnetic tw 001.	
1.481	<u>v</u>	1.493	.007									

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula ^a	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ	Biref								
1.504	1.495	1.497	1.504	.009	YUGAWARALITE (Zeolite grp) $\text{CaAl}_2\text{Si}_6^0\text{Al}^4\text{H}_2^0$	78° $r > v \text{ wk}$	$Z = \frac{b}{c} = 7^\circ$ $y:c =$	MCL	Parting 010	H 4.5 G 2.23	Insol in acids. U opt neg.	
	1.484	1.498	1.523	.039	ARISTARAINITE $\text{Na}_2\text{MgB}_{12}\text{Al}^4\text{H}_2^0$	70° $r > v \text{ wk}$	$X = \frac{b}{c} = -38^\circ$ $y:c =$	MCL	001, 100 perf 110	H 3.5 G 2.03 fus	Diss by HCl.	
1.510	1.497	1.498	1.499	.002	GONNARDITE (Zeolite grp) $\text{Na}_2\text{CaAl}_4\text{Si}_6^0\text{Al}^4\text{H}_2^0$	Ned to large	---	ORTH fib	---	White	G 2.26	---
1.488 1.507	1.496	1.498	1.504	.008	HEULANDITE (Zeolite grp) $(\text{Ca}, \text{K}_2)_2\text{Al}_3(\text{Al}, \text{Si})_2$ $\text{Si}_{13}0_{36}\cdot 12\text{H}_2^0$	35°	$Z = \frac{b}{c}$ $e_l c T_v \text{ neg}$	MCL	010 perf	White	H 4 G 2.20 F 2	Dec by HCl. SiO 57.4, CaO 7.0, SF 1.55, Na ₂ O 0.1, K ₂ O 1.4%.
	1.499	1.500	1.502	.003	WAIRAKITE (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{Al}^4\text{H}_2^0$	75° $r > v \text{ wk}$	$X = \frac{b}{c}$ $Z \sim \frac{c}{c}$	MCL	---	White	H 5-5.6 G 2.26 F 2.5	Gel with acids.
	1.484	1.501	1.550	.066	GOMERITE $\text{CaB}_6\text{Al}^4\text{H}_2^0$	$(63+4^\circ)$ $r > v \text{ wk}$	$Y = \frac{b}{c}$ $Z:c = 27^\circ$	MCL	001 perf	White	H 3 G 2.00 fus	Diss by acids.
1.498 1.503	1.500	1.501	1.504	.004	WELLSITE (Zeolite grp) $(\text{K}_2, \text{Ca}, \text{Ba})\text{Al}_2\text{Si}_6^0\text{Al}^4\text{H}_2^0$	39°	$Z = \frac{b}{c}$ $X:c = -52^\circ$	MCL	---	Col s	H 4-5 G 2.3 F 3	Dec by acids. Complex tw. Perhaps a barian Phillipsite.
	1.497	1.502	1.539	.042	LIEBIGITE $\text{Ca}_2(\text{UO}_2)(\text{CO}_3)_3\cdot 10\text{H}_2^0$ 6H_2^0	40° $r > v \text{ med}$	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	ORTH crusts	100	Green to yellow-green	H 2.5-3 G 2.41 infus	Diss by acids. Pleoc, X nearly col, Y and Z pale yellow-green. Fluor green in UV.
	1.498	1.503	1.510	.012	HEXAHYDROBORITE $\text{Ca}[\text{B}(\text{OH})_4]_2\cdot 2\text{H}_2^0$	83° $r > v$ very str	$X:c = 14^\circ$ $Z:c =$	MCL pris	---	Col s, white	H 2.5 G 1.87	Diss by acids.
1.485	1.500	1.503	1.510	.010	PHILLIPSITE (Zeolite grp) $(\text{K}_2, \text{Ca}, \text{Na}_3)_1\cdot 2(\text{Al}, \text{Si})_8$ $0_{16}\cdot 6\text{H}_2^0$	Med $r < v \text{ mod}$	$X = \frac{b}{c}$ $Z:c = 19^\circ$ $e_l \text{ pos}$	MCL fib a dist	100, 010	White	H 4 G 2.15 F 3	Gel with acids. Tw pl 001 and 011, penet, giving orth forms.

1.501	<u>1.503</u>	1.510	.009	PROSOPITE CaAl ₂ (F,OH) ₈	63° r > v str	$\frac{Y = b}{Z:c} = -35^\circ$	MCL tab 010	111 perf	cols	H 4.5 G 2.38 infus	---
1.503	(1.504)	1.535	.032	SVEITE KA1 ₇ (NO ₃) ₄ Cl ₂ (OH) ₁₆ • 8H ₂ O	Small	---	MCL	001 perf	White	Very soft G 2.0	Diss by H ₂ O. Sol in acids.
1.504	<u>1.504</u>	1.553	.049	EVENKITE (n-tetracosene) C ₂₁ H ₄₄	Small	---	MCL	One good	cols	H 1 G 0.92	M.p. 50°C. Hydro- carbon. Poly tw.
1.501	<u>1.504</u>	1.520	.019	HYDROCHLORBORITE Ca ₂ B ₄ O ₄ Cl(OH) ₇ •7H ₂ O	46° r < v	$\frac{Y = b}{X:c} = 25^\circ$ $X:a = 65^\circ$	MCL mass	001 good conch	cols	H 2.5 G 1.85	Diss by hot H ₂ O.
1.501	<u>1.504</u>	1.509	.008	HARMOTOME (Zeolite grp) (Ba,K) ₂ 1-2(Al,Si) ₈ 0 ₁₆ • 6H ₂ O	78° disp crossed	$Z = \frac{b}{c} = 65^\circ$ $X:a = 65^\circ$	MCL pris	010 good 001 less so	White	H 4.5 G 2.38 F 3.5	Dec by HCl. Cruciform tw, tw pl 001.
1.500	<u>1.509</u>	1.529	.036	ULEXITE NaCaB ₅ 9•8H ₂ O	70-73° (84-9°)	$X = \frac{b}{c} \sim 20^\circ$ $Y:c \sim 20^\circ$	TRIC fib	010 perf 120 good	White	H 2.5 G 1.96 F 1	Diss by acids. Partly dec by hot H ₂ O.
1.493	<u>1.505</u>	1.525	.025	LOUGHLINITE Na ₂ Mg ₃ Si ₆ 16•8H ₂ O	60°	---	ORTH fib	---	White	G 2.165	Dec by HCl, slowly dec by H ₂ O.
1.500	<u>1.505</u>	1.514	.012	RHODESITE (Ca,Na ₂ ,K ₂) ₈ Si ₁₆ 0 ₄₀ • 11H ₂ O	Small (60-20°)	$X = \frac{b}{a}$ $Y = \frac{b}{c}$	ORTH fib	100 good	White	G 2.36 fus	---
1.502	<u>1.505</u>	1.507	.003	MESOLITE (Zeolite grp) Na ₂ Ca ₂ Al ₆ Si ₉ 30•8H ₂ O	Large r > v str	$Y = \frac{b}{c} \sim \frac{1}{2}$ $X \sim \frac{1}{2}$	MCL el = b	101, T01 perf	White, gray	H 2.5 G 2.26 F easy	Gel with acids. Tw pl 100, 2V may vary with temp.
1.504	<u>1.506</u>	1.528	.034	BISCHOFITE MgCl ₂ •6H ₂ O	79° r > v wk	$X = \frac{b}{c} = 9.5^\circ$ $Y:c = 9.5^\circ$	MCL fib	---	cols	H 1.5 G 1.60 fus	Sol in H ₂ O. Deliq.
1.494	<u>1.507</u>	1.515	.011	HEULANDITE (Zeolite grp) (Ca,Na,Sr) ₂ -3Al ₃ (Al, Si) ₂ Si ₁₃ 36•12H ₂ O	10-40° (63-22°) r > v	$Z = \frac{b}{c} \sim \frac{1}{2}$ neg	MCL	010 perf	White	H 4 G 2.20 F 2	Dec by HCl. SiO ₂ 55.8, CaO 3.0, SiO ₂ 4.4, BaO 2.4, Na ₂ O 1.5, K ₂ O 1.3%.
1.504	<u>1.507</u>	1.543	.039	USSINGITE Na ₂ AlSi ₃ 8(OH)	32-39°	$Z:b = 3^\circ$	TCL	001 perf 110 fair	Rose-violet	H 6 G 2.48 F easy	Gel with acids.
1.498	<u>1.512</u>	1.525	.023	RABBITTITE Ca ₃ Mg ₂ (UO ₂) ₂ (CO ₃) ₆ (OH) ₄ •18H ₂ O	(62+11°)	$Y = \frac{b}{c} = 15^\circ$	MCL fib	001 and 2 pris	Pale green	H 2.5 G 2.6	Diss by acids, slowly dec by H ₂ O. Weakly fluor in UV.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
---	1.508	---	.001	LEUCITE $KAlSi_3O_8$	Small	$Z = \frac{a}{c}$	TET ps cub	110 poor	Col s	H 6 G 2.5 infus	Dec by acids. tw.	
1.493	1.509	1.561	.068	LARDERELLITE (NH_4) $B_5O_6(OH)_4$	60° $r < v$	$X = \frac{b}{c} = 30^\circ$	MCL tab	001, 010 001 less so	Col s	Soft, G 1.91 volat	Diss by acids. Rhombs with angle 66°.	
1.504	1.505	1.509	.008	HARMOTOME (Zeolite grp) (Ba, K_2) $Al_2(Si, Al)_8O_16$ · $6H_2O$	~ 90° disp crossed	$Z = \frac{b}{c} = 68^\circ$	MCL pris	010 good 001 less so	White	H 4.5 G 2.44 F 3.5	Dec by HCl. Tw 001 cruciform.	
1.510	1.510	1.513	.002	OLYMPITE Na_3PO_4	46°	---	ORTH	Conch	Col s	H 4 G 2.8	Sol in H_2O , gives an alk reaction.	
1.510	1.510	1.512	.002	OLYMPITE Na_3PO_4	85°	---	MCL plates	---	Col s	---	Diss by acids.	
1.470	1.510	1.579	.109	STRONTIOBORITE $SrB_8O_{11}(OH)_4$	30° $r < v \text{ wk}$	$X = \frac{a}{c}$ $Z = \frac{b}{c}$	ORTH pris or tab	Conch	Col s	H 3-3.5 G 2.36 F 2-2.5	Diss by acids with eff. Dec by hot H_2O .	
1.504	1.510	1.573	.069	PIRSSONITE $Na_2Ca(CO_3)_2 \cdot 2H_2O$	26° $r < v$	---	ORTH (?)	001 good	Yellow	H 1 G 0.95 F 80°C	A paraffin hydro- carbon.	
1.508	1.510	1.569	.061	HATCHETTITE $Ca_4H_8O_8$ (?)	Very small $r > v \text{ wk}$	$Z:c = 2.5^\circ$ el pos	MCL	---	Col s	G 2.51	Diss by HCl.	
1.510	1.510	1.545	.035	SOLONGOITE $Ca_2B_3O_4Cl(OH)_4$	Med to large	$Y = \frac{b}{c} = 18^\circ$	MCL fib	---	White	G 2.36	---	
1.504	1.510	1.519	.015	MOUNTAINITE (Ca, Na_2, K_2) Si_4O_{10} · $3H_2O$	---	---	Globu- lar aggre- gates	---	White	6 2.15 calc	Diss by HCl with eff. Fluor light blue in UV.	
1.513	1.510	1.516	.008	DYPINGITE $Mg_5(CO_3)_4(OH)_2 \cdot 5H_2O$	Med to large	ORTH fib	---	White	G 2.27	---		
1.498	1.508	1.510	1.513	.005	GONNARDITE (Zeolite grp) $Na_2CaAl_4Si_4O_{10} \cdot 7H_2O$	$Z = \frac{b}{c} = 22^\circ$ el clv neg	MCL el c	010 perf	White	H 5 G 2.32	Dec by acids.	
1.508	1.511	1.523	.015	BREWSTERITE (Zeolite grp) (Sr, Ba, Ca) $Al_2Si_6O_16$ · $5H_2O$								

1.505	<u>1.511</u>	1.516	.011	PETALITE LiAlSi ₄ O ₁₀	83° r > v wk		Z = <u>b</u> X: <u>a</u> = -5°	MCL	001 perf 201 good	H 6 G 2.39				
1.505	<u>1.512</u>	1.524	.019	FLAGSTAFFITE Ca ₁₀ H ₂₂ O ₃	77° r > v		Z = <u>a</u> X = <u>c</u>	ORTH pris	110 imperf	Soft. G 1.09 F 119°C	Sol. in warm alcohol (= cis-terpin hydrate).			
	1.501	<u>1.513</u>	1.536	.035	BOBIEIRITE Mg ₃ (PO ₄) ₂ •8H ₂ O	73° r < v wk	Y = <u>b</u> Z: <u>c</u> = 27°	MCL acic	010 perf	Col s	H 2-2.5 G 2.20 fus	Diss by acids. Data for synth compound.		
	^v 1.553	1.511	<u>1.513</u>	1.518	.007	THOMSONITE (Zeolite grp) NaCa ₂ Al ₅ Si ₅ O ₂₀ •6H ₂ O	75° r > v	X = <u>a</u> Z = <u>b</u>	ORTH pris fib <u>c</u>	010 perf 100 good	White	H 5 G 2.33 F 2	Col with acids.	
	1.525	1.510	<u>1.513</u>	1.517	.007	MONTEREGIANITE (Na,K) ₆ (Y,Ca) ₂ Si ₁₆ O ₃₈ • 10H ₂ O	87°	X = <u>c</u> Y = <u>a</u>	ORTH acic, tab <u>a</u>	010 perf 001 good 100 fair	White, cols	H 3.5 G 2.42	Dec by acids.	
						---	---	TCL	---	White	G 2.2-2.3	Opt. char unk.		
						---	---	mass	---					
						---	---	ORTH	---	Col s	G 1.83	Dec by hot H ₂ O, diss by acids.		
						---	---	ORTH pris	100, 010, 001 dist	Col s to yellowish	H 5-6 G 2.33 F 3	Insol in acids.		
						---	---	MCL	001 perf neg ps trig	Pale yellow, pale green	H 2.5 G 1.97 F 5	Dec by hot H ₂ O, diss by acids. Pleoc., X and Y col s, Z pale yellow.		
						---	---	ORTH tab 100 001 imperf	010 perf	White	H 3-3.5 G 2.11 infus	Diss by acids.		
						---	---	ORTH(?)	001 perf 100, 010	Col s	H 1.5 G 2.35 infus	Diss by acids. Col fragments lath- shaped.		
						---	---	MCL	---	---	G 1.47 F 234°C	= Phthalimide.		
						---	---	ORTH	---	Deep blue	H 1-1.5 G 1.95	Sol. in cold H ₂ O, Pleoc., X and Y blue, Z pale blue.		
						---	---	MCL	100 perf	Col s	H 3 G 2.09	Diss by acids.		
						---	---	el neg	Mass, compact	White	G 2.1	Diss by HC1. Opt. char unk.		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index		MINERAL NAME and formula		$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ	Biref							
1.510	1.518	1.523	1.588	.070	HATCHETTITE $\text{Ca}_{40}\text{H}_{82}$ (?)	33° $r < v$	$Z = c$	001 good	White	H 1 F 0.95 80°C	A paraffin hydro-carbon.
1.520	1.523	1.533	1.533	.013	GYPSUM $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$	58° $r > v$ perc	$y = b$ $x:c = 37^\circ$ $z:c = 37^\circ$	010 perf 011, 010 100 poor	White	H 1.5-2 G 2.32 F 2.5-3,	Diss by HCl. Tw p1 100.
1.521	1.523	1.533	1.533	.012	MASCAGNITE $(\text{NH}_4)_2\text{SO}_4$	52° $r < v$ wk	$x = c$ $y = \frac{c}{b}$	001 dist	ColS, yellowish	H 2-2.5 G 177 F 1	Sol in H_2O . Volat when heated. Tw p1 110.
1.518	---	1.530	1.533	.012	TACHARANITE $\text{Ca}_{12}\text{Al}_2\text{Si}_{18}^0\text{Al}^{51} \cdot 18\text{H}_2\text{O}$	---	$x = fib$	001 good	ColS	G 2.35	Opt char unk.
1.518	1.524	1.544	1.544	.026	WAVELLITE $\text{Al}_3(\text{PO}_4)_2(\text{OH},\text{F})_3 \cdot 5\text{H}_2\text{O}$	60° $r > v$ wk	$x = b$ $y = \frac{a}{b}$ $z:c = 37^\circ$	010 perf 010 101 good 010 dist	White, yellow, green	H 3.5-4 G 2.36 infus	Diss by acids. Colored vars pleoc, X greenish, Z yellowish, abs X > Z.
1.534	1.524	1.530	1.530	.012	MACDONALDITE $\text{BaCa}_4\text{Si}_{16}^0\text{Al}^{36}(\text{OH})_2 \cdot 10\text{H}_2\text{O}$	90°	$x = c$ $y = \frac{a}{b}$	010 perf 001 good	ColS, white	H 3.5-4 G 2.27 F 5.5	Insol in cold acids, dec by boiling HCl (1:1).
1.508	1.525	1.586	1.586	.078	SIDERONATRITE $\text{Na}_2^+\text{Fe}(\text{SO}_4)_2(\text{OH}) \cdot 3\text{H}_2\text{O}$	58° $r > v$ str	$x = a$ $y = \frac{a}{b}$	001 perf	Orange to straw-yellow	H 2 G 2.28 F 2	Dec by hot H_2O , diss by HCl. Pleoc, X colS, Y pale yellow, Z pale amber-yellow.
1.513	1.525	1.577	1.577	.064	STRONTIOLIGNORITE $(\text{Sr},\text{Ca})_2\text{B}_{14}^{10}23 \cdot 8\text{H}_2\text{O}$	50°	$y = b$ $z:c = 40^\circ$	010 perf	White	H 3 G 2.25 fus	Diss by acids. Sr:Ca = 2:1 to 1:1.
1.517	1.525	1.579	1.579	.062	GINORITE $\text{Ca}_2\text{B}_{14}^{10}23 \cdot 8\text{H}_2\text{O}$	42° inclined disp	$y = b$ $z:c = 39^\circ$	010 perf	White	H 3.5 G 2.08 fus	Diss by acids.
1.515	1.525	1.544	1.544	.029	PROBERTITE $\text{NaCaB}_5^0(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	73° $r > v$	$y = b$ $z:c = 12^\circ$ $el pos$	110 perf	White	H 3.5 G 2.14 F 2	Diss by acids.
1.513	1.525	1.542	1.542	.029	VOGLITE $\text{Ca}_2^+\text{Cu}(\text{UO}_2)(\text{CO}_3)_4 \cdot 6\text{H}_2\text{O}$ (?)	90° $(81+8)$	$x = b$ $y = \frac{c}{b}$ $el c\bar{t}v$ pos	010 perf	Emerald-green	Soft infus	Diss by acids. Pleoc, X deep blue, Y bluish-green, Z yellow. Poly tw.

1.521	<u>1.525</u>	1.545	.024	SPADATE $MgSiO_2(OH)_2 \cdot H_2O(?)$	Small to med (49+12°)	Ext ~ , el pos	ORTH(?) feated, column- nar	---	Cream to pink	G 2.2	F 5	Dec by HCl .	
1.523	<u>1.525</u>	1.532	.009	CHALCOALUMITE $CuAl_4(SO_4)(OH)_{12} \cdot 3H_2O$	Small r > v str	Ext 40° on flat face	MCL fib crusts	Several perf	Turquoise green to blue	H 2.5	F 5	Diss by hot acids. Tw pl el.	
1.513 v 1.533	<u>1.523</u>	1.532	.009	THOMSONITE (Zeolite grp) $NaCa_2Al_5Si_5O_{20} \cdot 6H_2O$	48° r > v	X = $\frac{a}{b}$ Z = $\frac{c}{b}$	ORTH pris, fib \underline{c}	010 perf 100 good	White	H 5	F 2	Gel with acids.	
1.508	<u>1.527</u>	1.550	.042	KALIBORITE $KMgB_{12}O_{16}(OH)_{10} \cdot 4H_2O$	81°	Y = $\frac{b}{c}$ Z:c = 65°	MCL pris	001, T01 perf 100 good	White	H 4-4.5	F 1	Diss by acids.	
1.523 v 1.537	<u>1.527</u>	1.545	.022	HYDROMAGNESITE $Mg_5(CO_3)_4(OH)_2 \cdot 4H_2O$	50°	Z = $\frac{b}{c}$ X:c = 47° el clv neg fib \underline{c}	MCL ps orth parting 100	010 perf 100	White	H 4-4.5	F 2.22	Diss by hot acids. Tw pl 100, poly common.	
1.526	<u>1.528</u>	1.551	.025	TERUGGITE $Ca_4MgAs_2B_{12}O_{22}(OH)_{12} \cdot$ 12H ₂ O	33° r > v wk	Z = $\frac{b}{c}$ X:c = -26°	MCL pris	001 good 110 fair	Col's, white	H 2.5	G 2.15	Diss by acids.	
1.525	<u>1.528</u>	1.544	.019	SAZHINITE $Na_2CeSi_6O_{14}(OH) \cdot 5H_2O$	47°	Z = $\frac{a}{b}$ Y = $\frac{c}{b}$	ORTH ps tet tab	100, 010, 001 perf	Gray, white, cream	H 2-3	G 2.61	Diss by acids.	
1.527	---	1.531	.004	BIRUNITE $Ca_{36}(SiO_3)_{17}(CO_3)_{17}$ $(SO_4)_2 \cdot 30H_2O$ (?)	----	----	ORTH(?) fib	One perf	White	H 2	G 2.36	Gel with acids.	
1.507	<u>1.529</u>	1.573	.067	MAGNESIOTIOPIAPISTE $(Mg, Fe)Fe^{+3}_4(SO_4)_6$ $(OH)_2 \cdot 20H_2O$	73° r > v str	X ~ $\frac{b}{c}$ Y:c ~ 38°	TCL tab	010 perf T01 imperf	Sulfur- to golden- yellow	H 2.5-3	G 2.13	Diss by acids.	
1.535	<u>1.522</u>	1.577	.055	BOTRYOGEN $MgFe^{+3}(SO_4)_2(OH) \cdot 7H_2O$	41°	X = $\frac{b}{c}$ Z:c = 12°	MCL reni- form	010 perf 110 good	Red to orange	H 2-2.5	G 2.05-2.14	Dec by hot H ₂ O, diss by acids.	
1.548										Pleoc., X yellow- green, Y pale yellow- Z yellow to yellow- green.	Pleoc., X cols to brown, Y pale brown, Z golden- to orange- red.	F 4.5-5	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.483	1.530	1.576	.093	CHALCONATRONITE $\text{Na}_2\text{Cu}(\text{CO}_3)_2 \cdot 3\text{H}_2\text{O}$	Large (93°-2°)	$\gamma = \epsilon_1$ $z:c$ small	MCL ps hex	---	Greenish-blue	Soft G 2.27	Dec by H_2O , diss by acids. Fe^{2+} bleo., X nearly cols, Y pale blue, Z blue.	
1.515	1.530	1.580	.065	EARLANDITE $\text{Ca}_3(\text{C}_6\text{H}_5\text{O}_7)_2 \cdot 4\text{H}_2\text{O}$	60°	---	MCL	---	White to pale yellow	G 1.95 6 2.15	Calcium citrate.	
---	---	1.542	---	GLAUOKERITE (Zn,Cu) $10\text{Al}_4(\text{SO}_4)_3 \cdot 2\text{H}_2\text{O}$ (?)	---	$z = \epsilon_1$	Radiating, fib.	---	Sky-blue	H 1 6 2.15	Concentric color banding. Opt. char unk.	
1.525	1.53	1.550	.025	ROESSLERITE $\text{MgHAsO}_4 \cdot 7\text{H}_2\text{O}$	Small	$z = b$ $x:a = 14^\circ$	MCL	110 imperf	Col s	H 2-3 G 1.94	Diss by acids.	
1.496	1.531	1.579	.083	CHALLANTITE $6\text{Fe}_2(\text{SO}_4)_3 \cdot \text{Fe}_2\text{O}_3 \cdot 6\text{H}_2\text{O}$ $63\text{H}_2\text{O}$	80°	---	tab	---	Yellow	---	Slowly sol in H_2O , diss by HCl .	
1.530	1.531	1.534	.004	BUDDINGTONITE (Fe)dspar grp) $(\text{NH}_4)_2\text{Al}_2\text{Si}_6\text{O}_{16} \cdot \text{H}_2\text{O}$	---	$z = b$ $x:a = 4^\circ$	MCL	001 good 010 dist	Col s	H 5.5 G 2.32 F 5	Insol in acids.	
1.5285	1.531	1.532	.004	Unnamed silicate (K, Na) $4\text{Ca}_2\text{AlSi}_6$ (O, OH, F) $_{18} \cdot \text{NaCl}$	29°	el clv neg	ORTH	One perf	Col s	H 4 G 2.58 F easy	Gel with acids (Am. Min. 44, 90).	
1.528	1.532	1.539	.011	ALBITE plagioclase, plutic (Fe)dspar grp) $\text{NaAlSi}_3\text{O}_8$	74°	$x':a$ on $001 = 3^\circ$ $010 = 20^\circ$	TCL	001 perf 010 good	Col s, white	H 6 G 2.61 F 4	End member, An_0 . Insol in acids. Poly tw pl 010 almost universal. Other tw laws common.	
1.541	1.527	1.532	1.538	CORDIERITE $\text{Mg}_2\text{Al}_4\text{Si}_5\text{O}_{18}$	88°	$x = c$ $z = \frac{c}{b}$	ORTH	010 poor	Blue, cols	H 7 G 2.59 F 5.5	Nearly insol in acids. Fe^{2+} 1.5, Fe^{2+} 2.2, Na^+ 0.2%.	
1.542	1.520	1.533	1.584	KIESERITE $\text{MgSO}_4 \cdot \text{H}_2\text{O}$	55°	$y = b$ $z:c = -77^\circ$	MCL perf T11, 101, 011 imperf disp dist	111, 110	White	H 3.5 G 2.57 F 2-3	Slowly sol in H_2O , diss by acids. Poly tw.	
1.590	1.520	1.533	1.584	KIESERITE $\text{MgSO}_4 \cdot \text{H}_2\text{O}$	55°	$r > v$ mod	MCL	111, 110	White	H 3.5 G 2.57 F 2-3	Slowly sol in H_2O , diss by acids. Poly tw.	

1.525 ^	<u>1.533</u>	1.542	.012	THOMSONITE (Zeolite grp) $\text{Na}(\text{Ca}, \text{Sr})_2\text{Al}_5\text{Si}_5\text{O}_{20} \cdot 6\text{H}_2\text{O}$	52° $r > v$, dist	$X = \frac{a}{b}$ $Z = \frac{c}{b}$	010 perf 100 good	White	H 5 G 2.37
1.519	<u>1.534</u>	1.569	.050	TUNELLITE $\text{SrB}_6\text{O}_{10} \cdot 4\text{H}_2\text{O}$	(68+5°) $r > v$ wk	$Y = \frac{b}{c}$ $Z:c = -5^\circ$	MCL 100 perf 001 dist	Cols	H 2.5 G 2.40
1.522	<u>1.534</u>	1.569	.047	HYDROBORACITE $\text{CaMgB}_6\text{O}_8(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	63° $r < v$ perc	$Y = \frac{b}{c}$ $X:c = 33^\circ$ el neg	MCL tab 010	White less	H 2.3 G 2.17
1.518 ^	<u>1.533</u>	1.575	.042	FIBROFERRITE $\text{Fe}(\text{SO}_4)_2(\text{OH})_6 \cdot 5\text{H}_2\text{O}$	Small	$Z = c$	MCL ps trig	001 perf	Pale yellow
1.520	---	1.558	.038	CHELKARITE $\text{CaMgB}_2\text{O}_4\text{Cl}_2 \cdot 7\text{H}_2\text{O}$ (?)	---	---	ORTH pris	Perf el	White pale rose
1.524 ^	<u>1.534</u>	1.553	.028	WAVELLITE $\text{Al}_3(\text{PO}_4)_2(\text{OH}, \text{F})_3 \cdot 5\text{H}_2\text{O}$	72° $r > v$ wk	$X = \frac{b}{c}$ $Y = \frac{a}{c}$ el pos	ORTH fib, radiating	110 perf 101 good 010 dist	White, yellow, green
1.543	1.531	1.534	.007	MINYULITE $\text{KAl}_2(\text{PO}_4)_2(\text{OH}, \text{F}) \cdot 4\text{H}_2\text{O}$	Large	$X = \frac{c}{a}$ $Y = \frac{c}{a}$	ORTH acic	---	Cols
1.527 ^	1.537	1.52	1.535	COPIAPITE $\text{Fe}^{+2}\text{Fe}^{+3}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20\text{H}_2\text{O}$	70° (63+4°) $r > v$ str	$X \sim \frac{b}{c}$ $Y:c \sim 38^\circ$	TCL tab	010 perf T01	Sulfur- to golden- yellow
1.529 ^	1.540	1.578	.06	NEKOITE $\text{Ca}_3\text{Si}_6\text{O}_{12}(\text{OH})_6 \cdot 5\text{H}_2\text{O}$	70°	X:needles $Z \sim \perp$ 100	TCL acic	100 good	Cols
1.531	<u>1.536</u>	1.544	.013	PENTAHYDROBORATE $\text{CaB}_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	73°	---	TCL u mass	---	Cols
1.536	<u>1.536</u>	1.543	.007	ROEDDERITE (osumilite grp) $(\text{Na}, \text{K})_2(\text{Mg}, \text{Fe})_5\text{Si}_{12}\text{O}_{30}$	5-8°	---	HEX	---	Cols
1.535	<u>1.536</u>	1.538	.003	FLUORAPOPHYLLITE $\text{KCa}_4\text{Si}_8\text{O}_{20}\text{F} \cdot \text{H}_2\text{O}$	18-42°	$Z = c$	TET	001 perf 110 poor 110 poor	White pale yellow
1.527 ^	<u>1.537</u>	1.550	.023	HYDROMAGNESITE $\text{Mg}_5(\text{CO}_3)_4(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	83°	$Z = b$ $X:c = 47^\circ$	MCL ps orth, fib	010 perf parting 100	White H 4-4.5 G 2.17 infus

Gel with acids.
Diss by acids.
Diss by acids.
Diss by acids.
Opt char unk.

Diss by acids.
Colored vars are
paleoc., X greenish,
Z yellowish, abs X > Z.

Diss by warm acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Diss by acids.

Diss by acids.
Colored vars are
paleoc., X pale
yellow, Z yellow.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.535	<u>1.537</u>	1.545	.010	ASHCROFTINE $K_{10}CaY_2Si_6O_{12}(OH)_{10} \cdot 4H_2O$	Small		$Z = \underline{c}$	TET acic	100 perf 001 good	White, pink	H 5 6 2.61	---
1.536	<u>1.538</u>	1.544	.008	NATROAPOPHYLLITE $NaCa_4Si_8O_{20}F \cdot 8H_2O$	32°		---	ORTH	001 perf	Cols to yellow	6 2.50	Slightly dec by HCl.
1.536	<u>1.539</u>	1.603	.067	VOLKOVSKITE $(Ca, Sr)_8O_{16} \cdot 3H_2O$	24°		$Y = \frac{b}{a}$ $Z : a = 31^\circ$	MCL	010 perf 001 good	Cols	G 2.32	---
1.535	<u>1.540</u>	1.590	.065	ALUMINOCOPIAPITE $Al(Fe^{+3})_4(SO_4)_6(OH) \cdot 20H_2O$	Mod		---	TCL	010 perf	Yellow	---	Pleoc in yellow.
$\hat{1.545}$	<u>1.541</u>	1.547	.006	OSUMILITE-(Mg) (Ossumilite grp) $(K, Na)(Mg, Fe)_2(Al, Fe)_3$ $(Si, Al)_{12}O_{30} \cdot H_2O$	5-15°		---	HEX	---	Col s	6 2.63	---
1.536	<u>1.541</u>	1.546	.010	OLIGOCLASE plagioclase, plutonic (Feldspar grp) $(Na, Ca)Al(Si, Al)Si_2O_8$	~ 90°		$X : a$ on 001 = 1° 010 = 7°	TCL	001 perf 010 good	Col s	H 6 G 2.64	Data for An_{18} . Insol in acids. Tw poly on 010.
$\hat{1.541}$ neg	<u>1.532</u>	1.542	1.556	PHAUNOUXITE $Ca_3(AsO_4)_2 \cdot 11H_2O$	80°		---	TCL	100 perf 011 good	Col s	G 2.28	Diss by acids.
1.540	<u>1.542</u>	1.550	.010	RAITE $Na_4Mn_4Si_8(O, OH)_{24} \cdot 8-10H_2O$	(~ 50°)		$Z : c = 0-30^\circ$	ORTH acic	100, 010, 001 perf	Gold to brown	H 3 6 2.39 F easy	Dec by acids. Partly dec by acids. Dec by acids. Diss by acids.
$\hat{1.550}$	<u>1.538</u>	1.547	.009	CORDIERITE $(Mg, Fe)_2Al_4Si_5O_18$	84°		$X = \underline{c}$ $Z = \underline{\overline{b}}$	ORTH	010 dist	Blue	H 7 G 2.64 F 5	Fe 8.0, Fe ₂ O ₃ 0.6, Na 20 1.0%.
$\hat{1.550}$	<u>1.535</u>	1.543	1.561	WAVELLINE $Al_3(Po_4)_2(OH, F)_3 \cdot 5H_2O$	65°		$X = \frac{b}{a}$ $Y = \frac{a}{\overline{b}}$ el pos	ORTH radiating, fib	110 perf 101 good 010 dist	White, yellow, green	H 3.5-4 G 2.30 infus	Colored var pleoc, X greenish, Z yellowish, abs X > Z.
1.534	<u>1.543</u>	1.558	.024	GORDONITE $MgAl_2(Po_4)_2(OH)_2 \cdot 8H_2O$	73°		$X \perp c$ $Z : c = 30^\circ$	TCL laths	010 perf	Col s	H 3.5 G 2.23 F 3	Diss by acids.

1.536	---	1.550	.014 NaCa ₅ T ₄ Si ₁₆ H ₄₀ (OH) ₁₁ • 8H ₂ O	LOUDOUNITE Na ₂ Bes ₂ Si ₂ 6	---	---	fib	---	Green to white	H 5 G 2.48	Opt sign unk. Ext wavy.
v 1.550	1.543	1.549	.009	CHKALOVITE Na ₂ Bes ₂ Si ₂ 6	78° r > v	y = c	ORTH	001 good 111 poor	White	H 6 G 2.66 F easy	Dec by acids with sepn of silica.
1.540	1.544	1.559	.019 (Mg,Fe) ₃ (PO ₄) ₂	FARRINGTONITE Ca(VO)Si ₄ •4H ₂ O	55°	Z:c = 16°	MCL	100, 010 fair to good	Cols to yellowish	G ~ 2.80	FeO 3.7-5.4% Meteorite mineral.
1.542	1.544	1.551	.009	CAVANSITE NaBeSi ₃ 7(OH)	52° r < v extr	X = b Z = a ---	ORTH plates	010 good	Greenish- blue	G 2.21-2.31	Diff diss by acids. Pleoc, X and Z cols, Y blue.
1.543	1.544	1.546	.003	EPIDIDYMITE NaBeSi ₃ 7(OH)	22-30° r > v	X = a Z = b ---	ORTH plates	010, 001 perf	Col s	H 5.5 G 2.58	Insol in acids. Tw.
1.540 v 1.550	1.545	1.595	.070	MAGNESIOCOPAPITE (Mg,Fe)Fe ⁺³ (SO ₄) ₆ (OH) ₂ •20H ₂ O	Med r > v str	X ~ b el clv pos	TCL tab	010 perf	Yellow	H 2.5-3 G 2.1 F 4.5-5	Diss by acids. Pleoc, X and Y nearly cols, Z deep yellow.
1.532	1.545	1.572	.040	HALURGITE Mg ₂ B ₈ 10(OH) ₈ •H ₂ O	(71+6°)	X = b Z ~ c	MCL	---	Col s	H 2.5-3 G 2.19	Diss by boiling H ₂ O.
1.539	1.545	1.551	.012	BRUSHITE CaHPO ₄ •2H ₂ O	87° r > v crossed	Z = b X:c = -30°	MCL el c	010, 001 perf	White	H 2.5 G 2.30	Diss by acids.
1.545 (1.569)	1.546	1.553	.008	CHRYSOTILE (Serpentine grp) Mg ₃ Si ₂ 0 ₅ (OH) ₄	50°	---	MCL fib	---	Pale green	G 2.55 infus	Insol in acids. FeO 2.3, Al ₂ O ₃ 1.2, FeO 1.6%
1.531	1.546	1.562	.031	BRASSITE MgHAsO ₄ •4H ₂ O	Very large	Z = c y = a	ORTH crusts	001 perf	White	G 2.28 infus	Diss by acids.
1.545	1.546	1.549	.004	EUDIDYMITE NaBeSi ₃ 7(OH)	23-30° r > v	Y = b Z:c = -55°	MCL tab	001 perf	White	H 6 G 2.58 F 3	Insol in acids. Tw pl 001, lam, always present.
1.541	1.547	1.564	.023	VOGLITE Ca ₂ Cu(UO ₂)(CO ₃) ₄ • 6H ₂ O (?)	60°	X = b y = c	MCL	010 perf	Emerald- green	Soft infus	Poly tw lam. Pleo- str, X and Y deep bluish-green, Z yellow.
1.525	1.544 v 1.551	1.548	1.572	.028 MgFe ⁺³ (SO ₄) ₂ (OH) ₇ H ₂ O	40° r > v str	X = b Z:c = 12° el clv pos	MCL reniform	010 perf 110 good	Red to orange	H 2-2.5 G 2.08 F 4.5-5	Diss by acids. Pleoc, X yellow, Y red, Z orange-red.
1.542	1.548	1.566	.024	SCHODERITE Al ₂ (VO ₄)(PO ₄) ₂ •8H ₂ O	(61+11°)	Y = b Z:c = 29°	MCL tab	---	Yellow- orange	H 2 G 1.88	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.547	1.549	1.559	.012	PARTHEITE $\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot 2\text{H}_2\text{O}$	48° $r > v$ mod	$X:e = 23^\circ$ 30°	$Z = e$	MCL fib	clv e	White	6 2.39	---
1.549	1.549	1.680	.131	STUDTITE $\text{UO}_4 \cdot 4\text{H}_2\text{O}$	Small				---	Yellow	---	Slowly sol in cold dil HCl. Hydrous U-peroxide.
1.555	1.530	1.550	.062	FERRICOPJAPITE $\text{Fe}^{+3}\text{Fe}^{+3}(\text{SO}_4)_6\text{O}(\text{OH}) \cdot 20\text{H}_2\text{O}$	69° $r > v$	$X \sim b$	TCL	010 perf	Yellow	H 2 G 2.2	Diss by acids. Pleoc in yellow.	
1.545 1.554	1.544	1.556	.012	CORDIERITE $(\text{Mg}, \text{Fe})_2\text{Al}_4\text{Si}_5\text{O}_{18}$	85°	$X = c$ $Z = \frac{c}{b}$	ORTH	010 dist	Blue	H 7 G 2.63 F 5	Partly dec by acids. FeO 8.4, Fe ₂ O ₃ 0.6%.	
1.542 1.558	1.544	1.556	.012	ANDESINE plagioclase, plutonic (Feldspar grp) $(\text{Na}, \text{Ca})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	90°	$X':a = 0^\circ$ $001 = 0^\circ$ $010 = -5^\circ$	TCL	001 perf 010 good	Col s	H 6 G 2.66 F 5.5	Insol in acids. Poly tw 010. Data for An ₃₃ .	
1.550 neg 1.558	1.546	1.550	.008	CHALCOCITE $\text{Na}_2\text{BeSi}_2\text{O}_6$	81° $r > v$	$Y = c$	ORTH	001 good 111 poor	White	H 6 G 2.66 F easy	Dec by acids with sepn of silica.	
1.543 1.549	1.550	1.554	.004	VEATCHITE-A $\text{Sr}_2\text{B}_{11}\text{O}_{16}(\text{OH})_5 \cdot \text{H}_2\text{O}$	25°	$X = b$ $Z = \frac{c}{b}$	TCL	100 perf 011, 011 good	Col s	G 2.73	Diss by acids.	
1.548 1.540	1.551	1.621	.072	ZINCOBOTRYGEN $(\text{Zn}, \text{Mg})\text{Fe}^{+3}(\text{SO}_4)_2(\text{OH}) \cdot 7\text{H}_2\text{O}$	(54+6°) $r > v$		MCL pris	010 good	Orange-red	H 2.5 G 2.20	Diss by acids. Pleoc str, X yellow, Y and Z red.	
1.540	1.552	1.570	.030	RAUENTHALITE $\text{Ca}_3(\text{AsO}_4)_2 \cdot 10\text{H}_2\text{O}$	85°	$Y:e = 5^\circ$ 10°	TCL spher	---	Snow-white	G 2.36	---	
1.548	1.552	1.570	.022	NORBERGITE (Humite grp) $\text{Mg}_3\text{Si}_4(\text{F}, \text{OH})_2$	33° (51+13°)	$X = a$ $Z = \frac{a}{b}$	ORTH	---	Col s	H 5 G 3.2 infus	Gel with acids. Data for synth F compd.	
1.552 neg 1.562	1.549	1.552	.007	ANDESINE plagioclase, volcanic (Feldspar grp) $(\text{Na}, \text{Ca})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	~ 90°	$X':a = 0^\circ$ $001 = 2^\circ$ $010 = -4^\circ$	TCL	001 perf 010 good	Col s	H 6 G 2.67 F 5.5	Insol in acids. Poly tw 010. Data for An ₃₆ .	
1.534	1.553	1.638	.104	RHOMBOLASE $\text{HFe}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	27° (53+3°)	$X = c$ $Y = \frac{c}{a}$	ORTH tab 001	001 perf 110 good	Col s to yellowish fus	H 2 G 2.23 fus	Slowly sol in H ₂ O, diss by acids.	

1.543	<u>1.553</u>	1.620	.077	VEATCHITE $\text{Sr}_2\text{B}_{11}\text{O}_{16}(\text{OH})_5 \cdot \text{H}_2\text{O}$	44° r > v wk	$Z = \frac{b}{c} = -30^\circ$ $X:C = -30^\circ$	MCL	100, 011 white perf	H 2 G 2.69 F 1.5	Diss by acids.				
1.551	<u>1.553</u>	1.621	.070	P-VEATCHITE $\text{Sr}_2\text{B}_{11}\text{O}_{16}(\text{OH})_5 \cdot \text{H}_2\text{O}$	37° (20+10°)	$Y = \underline{b}$ $Z:c = 27^\circ$	MCL	cols	H 2 G 2.60-2.69 F 1.5	Diss by acids.				
1.513	<u>1.553</u>	1.582	.035	BOBIEIRITE (Mg,Fe) ₃ (PO ₄) ₂ ·8H ₂ O	57° r < v	$Y = \underline{b}$ $Z:c = 27^\circ$	MCL acic	010 perf	cols	H 2-2.5 G 2.3 fus	Diss by acids. FeO 15.1, MnO 2.95%.			
1.547	<u>1.553</u>	1.559	.009	CHAROTITE K(Ca,Na) ₂ Si ₄ Al ₁₀ (OH,F)·H ₂ O	29°	$X = \underline{b}$ $Z:c = 5^\circ$ el pos	MCL	3 c/v	Lilac to violet	G 2.54	In thick fragments, pleoc, X rose, Z cols.			
1.550	<u>1.553</u>	1.559	.030	LEMONYNITE (Na,K)CaZr ₂ Si ₁₀ Al ₂₆ 5-6H ₂ O	80° r < v wk	$Y = \underline{b}$ $Z:a = 5^\circ$ el pos	MCL	100	cols	H 4 G 2.29	---			
1.540	<u>1.553</u>	1.570	.030	ZINOCOPAPISTE ZnFe ₄ (SO ₄) ₆ (OH) ₂ ·18H ₂ O	(78+5°) r < v str	$X \sim \underline{b}$ el c/v pos	TCL tab	010 perf	Yellow-green	H 2 G 2.18 F 5	Diss by acids. Pleoc str in shades of yellow.			
1.550	<u>1.554</u>	1.586	.052	GLUCINE Ca ₂ Be ₈ (PO ₄) ₄ (OH) ₈ ·H ₂ O	---	---	---	---	cols	H ~ 5 G 2.23-2.40	Diss by acids. Opt char unk.			
1.575	<u>1.554</u>	1.571	.024	GROTHINE Silicate of Ca, Al, Fe (?)	Med r < v	el pos, ext	ORTH tab 010	Fine needles	---	cols	G 3.09 infus	Dec by H ₂ SO ₄ .		
1.547	---	1.554	---	CLINOCHLORE var. Corundophyllite (Chlorite grp) (Mg,Fe), ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	50° r < v	$Z \sim \underline{c}$	MCL	001 mic	Green	H 3 G 2.6 F diff	Insol in acids.			
1.553	<u>1.554</u>	1.558	.005	WHEWELLITE (oxalate) CaC ₂ O ₄ ·H ₂ O	83° r < v wk	$X = \underline{b}$ $Z:c = 30^\circ$	MCL	101 good 001, 010 dist	cols	H 2.5-3 G 2.21 infus	Diss by acids. Tw pl 101.			
1.571	v	1.553	1.554	STUDITITE UO ₄ ·4H ₂ O	Small	$Z = \underline{eI}$	MCL fib	---	Yellow	---	Slowly sol in cold dil HCl. Hydrous U-peroxide.			
1.491	<u>1.555</u>	1.654	.163	AMARILLITE NaFe(SO ₄) ₂ ·6H ₂ O	Large r < v	$Y = \underline{b}$ $Z:c = 51^\circ$	MCL thick tab	110 good	Pale greenish-yellow	H 2.5-3 G 2.19	Sol in H ₂ O.			
1.545	<u>1.555</u>	1.680	.135	HARTITE C ₁₉ H ₃₂ (?)	Med (57+7°) r < v perc	---	TCL	---	White	H < 1 G 1.06 F 1	A hydrocarbon. Diss by alcohol or ether.			
1.532	<u>1.555</u>	1.591	.059	WOODWARDITE Cu ₄ Al ₂ (SO ₄) ₂ (OH) ₁₂ 3H ₂ O (?)	(58+19°)	---	ORTH(?) spher	---	Greenish-blue	G 2.38	---			
1.546	<u>1.555</u>	1.587	.041											
1.552	<u>1.555</u>	1.565	.013											
~1.571														

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.551	1.555	1.562	.011	VAUXITE $\text{FeAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	32° (74+21°) $r > v$ str	$Z \sim \perp$ 001	TCL tab 010	---	Sky blue to vene- tian blue	H 3.5 G 2.39	Pleoc str, X and Z cols, Y blue.
1.550	1.557	1.567	.017	LITHIOPHOSPHATE Li_3PO_4	69-88°	$X = \underline{a}$ $Y = \underline{b}$	ORTH u mass	100 perf 110 dist	Col s to rose	H 4 G 2.46	Slowly sol in H_2O , diss by acids.
1.551	1.558	1.582	.031	METAVARISCITE $\text{AlPO}_4 \cdot 2\text{H}_2\text{O}$	55° $r < v$ wk	$Y = \underline{b}$	MCL	010	Green	H 3.5 G 2.54 infus	Diss by HCl after being heated gently (color changes to lavender). Pleoc wk, X cols, Y and Z pale green.
1.551	1.554	1.565	.020	Unnamed phosphate	~ 90°	ext large	MCL (?)	---	Col s	---	Occurs with Lacroixite (Am. Mineral., 57, 1914 (1972)).
1.554	1.558	1.573	.019	PARAVAUXITE $\text{FeAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Med	---	TCL prts	010 perf	Col s	H 3 G 2.36	---
1.555	1.558	1.562	.007	LABRADORITE plagioclase, plutonic (Feldspar grp) $(\text{Ca}, \text{Na})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	78°	$X': \underline{a}$ on 001 = -6° 010 = -17°	TCL	001 perf 010 good	Col s	H 6 G 2.69	Insol in acids. Poly tw 010. Data for An_{50} .
1.536	1.559	1.697	.161	ZELLERITE $\text{Ca}(\text{UO}_2)(\text{CO}_3)_2 \cdot 5\text{H}_2\text{O}$	30-40° (47+2°) $r < v$ wk	$Z = \underline{c}$	ORTH fib	---	Light lemon- yellow	Soft G 3.25	Pleoc, X and Y cols, Z light green. Fluor green in UV.
1.559	1.560	1.584	.025	BASSANITE $2\text{CaSO}_4 \cdot \text{H}_2\text{O}$	10-15°	$Z \parallel$ \underline{x} el pos	HEX	---	White	G 2.7 F 3	Diss by HCl. Tw pl 100.
1.557	1.561	1.569	.012	CLINOPHOSINATE $\text{Na}_3\text{CaPSi}_3\text{O}_7$	80°	MCL	Conch	Pale lilac	Pale lilac	H 4 G 2.85	Dec by HCl.
1.494	1.561	1.692	.198	HUMBOLDTINE (oxalate) $\text{FeC}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$	Large	---	MCL ps orth	---	Yellow	H 1.5-2 G 2.28 infus	Diss by acids. Pleoc, X pale yellow-green, Y pale greenish-yellow, Z intense yellow.
1.550	1.561	1.582	.032	METAVAUXITE $\text{FeAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Large	$X = \underline{b}$ $Z: \underline{c} = -17^\circ$	MCL actic	---	Col s	H 3 G 2.34	---

1.559	<u>1.562</u>	(1.572)	.013	MCNEARITE NaCa ₅ H ₄ (AsO ₄) ₅ ·4H ₂ O	66° disp str	One perf TCL fib	White	G 2.60
1.552 1.569	<u>1.568</u>	1.566	.008	LABRADORITE plagioclase, volcanic (Feldspar grp) (Ca,Na)Al(Si ₂ Al)Si ₂ O ₈	76°	X':a on 001 = -15° 010 = -26° el pos	Col s 001 perf 010 good	H 6 G 2.69 infus
1.561	<u>1.563</u>	1.567	.006	DICKITE Al ₂ Si ₂ O ₅ (OH) ₄	55-80° r > v str	Z = b Y:a Z:c = 14- 20°	001 perf MCL ps hex plates	H 2 G 2.62 infus
1.554	<u>1.564</u>	1.595	.041	BARICITE (Mg,Fe) ₃ (PO ₄) ₂ ·8H ₂ O	59° r < v wk	X = b Z:c = 32°	010 perf MCL plates	H 1.5-2 G 2.42
1.547	<u>1.566</u>	1.594	.047	QUENSTEDTITE Fe ₂ (SO ₄) ₃ ·10H ₂ O	70° (80+5°) r < v str horizontal	---	TCL	Col s 100 good
1.562	<u>1.566</u>	1.587	.025	SENEGALITE Al ₂ (PO ₄)(OH) ₃ ·H ₂ O	53° r > v	Z = a Y = c el pos	ORTH TCL fib	Violet H 2.5 G 2.15
1.561	<u>1.568</u>	1.585	.024	PICROPHARMACOLITE H ₂ Ca ₄ Mg(AsO ₄) ₄ ·11H ₂ O	Med large	X = b Z:c = 8- 15° el pos	One good	White
1.571	<u>1.571</u>	1.568	.019	GIBBSITE Al(OH) ₃	0-5° r < v or r > v	X = b Y:c = 69°	001 perf MCL ps hex, tab 001	White, brownish H 3 G 2.3-2.4 infus
1.579	<u>1.579</u>	1.568	.019	ELPIDITE Na ₂ ZrSi ₆ O ₁₅ ·3H ₂ O	84° r < v dist	X = c Z = a el neg	110 dist el c	White, red, brown H 5-7 G 2.63
1.565 1.570	<u>1.562</u>	1.568	.013	BYTOMNITE plagioclase, plutonic (Feldspar grp) (Ca,Na)Al(Si ₂ Al)Si ₂ O ₈	~ 90°	X':a on 001 = -16° 010 = -29° el pos	001 perf 010 good	Col s H 6 G 2.71 infus
1.558 1.568 neg	<u>1.563</u>	1.568	.010	GÖRGÉYITE K ₂ Ca ₅ (SO ₄) ₆ ·H ₂ O	80°	---	TCL tab	Col s, yellowish H 3.5 G 2.8-2.9
1.560	<u>1.569</u>	1.584	.024	SUDOITE (Chlorite grp) Mg ₂ (Al,Fe) ₃ Si ₃ Al ₁₀ (OH) ₈	54° (71+13°)	Z:c = 7° el clv neg	001 perf MCL	White H 2.63
1.563	<u>1.569</u>	1.581	.018	LIZARDITE (Serpentine grp) Mg ₃ Si ₂ O ₅ (OH) ₄	Small	---	TRIG and HEX fib	Green G 2.58 infus
1.582	<u>1.568</u>	1.569	.013					Nearly insol in acids. Fe ₂ O ₃ 3.6, FeO 1.1%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.562 \wedge 1.569 neg	1.565	1.569	1.574	.009	BYTOMITE plagioclase, volcanic (Feldspar grp) (Ca, Na)Al(Si ₂ Al)Si ₂ O ₈	~ 90°	X':a on 001 = -23° 010 = -33°	TCL	001 perf 010 good	H 6 G 2.72 infus	Insol in acids. Poly tw on 010. Data for An75.
(1.546) \wedge	1.569	(1.569)	1.570	.001	CHRYSOTILE (Serpentine grp) Mg ₃ Si ₂ O ₅ (OH) ₄	Med	---	MCL fib	---	Dark green	Insol in acids. Fe ₂ O ₃ 1.4, Al ₂ O ₃ 0.7, FeO 1.3%.
1.555	1.57		1.585	.030	TENGFRITE CaY ₃ (CO ₃) ₄ (OH) ₃ ·3H ₂ O	Large	X = e1	ORTH fib	---	White	Infus G 2.8
1.628			1.591	.026	NORBERGITE (Humite grp) Mg ₃ Si ₁₀ O ₄ (F, OH) ₂	49°	X = $\frac{a}{b}$ Z = $\frac{c}{d}$	ORTH	---	Yellow to brown	Gel with acids.
1.552 \wedge	1.565	1.570		1.582	WAGNERITE (Mg, Fe) ₂ (PO ₄)F	27°	Y = $\frac{b}{c}$ Z:c = -22°	MCL	100, 120, 010 imperf	Col., flesh, yellow	H 5-5.5 G 3.15 F 4
1.569	1.570				BUKOVSKYITE Fe ₂ (AsO ₄) ₂ (SO ₄)(OH)· 7H ₂ O	---	ext at 18°	MCL(?) needles	---	Yellow to gray-green	Diss by HCl. Opt char unk.
1.582	---	1.570			SHABYNITE Mg ₅ (BO ₃) ₂ (Cl, OH) ₂ (OH) ₅ · 4H ₂ O	---	X e1	MCL fib	---	Snow-white	H 3 G 2.32
1.543	1.571		1.577	.043	HOERNESITE Mg ₃ (AsO ₄) ₂ ·8H ₂ O	---					Diss by acids.
1.563 \wedge	1.571		1.596	.033	BRUCITE Mg(OH) ₂	60°	X = $\frac{b}{c}$ Z:c = 31° e1 clv pos	MCL tab 010	010 perf	White	G 2.73 F 2-3
1.565	1.571		1.584	.019	NEFEDOVITE Na ₅ Ca ₄ (PO ₄) ₄ F	30-70°	X = e1	TRIG fib	0001 perf	Greenish- white	Diss by acids. Anom biax. Anom interf colors in some specimens.
1.571	1.571		1.590	.019	BANASITE BaNa ₂ Al ₄ Si ₄ O ₁₆	~ 0°	---	TCL ps tet	Conch	Col.s	H 4.5 G 3.01
1.570	1.571		1.578	.008		41°	X = $\frac{c}{a}$ Y = $\frac{c}{d}$	ORTH good	110, 001	White	H 6 G 3.06
											--

1.554 1.585	<u>1.571</u>	1.576	.005	CLINOCHLORITE (Chlorite grp) (Mg, Fe)5Al(Si3Al)O10 (OH)8	Small $r < v$	MCL $Z \sim c$ el neg	001 perf	Green	H 3 G 2.66 F diff				
1.555 ^	1.571	---	1.576	.005	WOODWARDITE $Cu_4Al_2(SO_4)(OH)_{12} \cdot 3H_2O$ (?)	ORTH(?) spher	---	Greenish-blue	G 2.38	----	Diss by HCl.		
	1.562	<u>1.572</u>	1.585	.023	FERRARISITE $Ca_5H_2(AsO_4)_4 \cdot 9H_2O$	TCL 83°	001 perf	Col s	G 2.63	----	Diss by HCl.		
1.570 1.575	1.572	1.594	.022	PREOBRAZHENSKITE $Mg_3B_{11}O_{15}(OH)_9$	0-20°	ORTH	---	Lemon-yellow, cols	H 4.5-5 G 2.44	----			
	1.562	<u>(1.572)</u>	1.583	.021	Unnamed hydrous calcium arsenate	X = $\frac{a}{c}$ Y = $\frac{b}{c}$	MCL ---	Col s	---	Opt char unk (Am. Mineral., 58, 561 (1973)).			
	1.570	<u>1.572</u>	1.582	.012	FROLOVITE $CaB_2(OH)_8$	Z = $\frac{b}{c}$ Y:z = 8°	TCL ---	White	H 3.5 G 2.14	----	Diss by acids. Ext may be undulatory.		
	1.572	<u>(1.572)</u>	1.582	.010	BAYERITE $Al(OH)_3$	MCL 75°	---	White	----	Diss by hot alkalies.			
	1.543	<u>1.575</u>	1.634	.091	METASSIDERONATRITE $Na_2Fe^{+3}(SO_4)_2(OH) \cdot H_2O$	Small X = $\frac{a}{c}$ Y = $\frac{b}{c}$	ORTH ---	001 mic 100, 010 perf	Golden- to straw- yellow	H 2.5 G 2.68	Diss by acids.	Pleoc in green, abs X and Z > Y.	
	1.558	<u>1.575</u>	1.620	.062	CUPROCOPIAPITE $CuFe_4(SO_4)_6(OH)_2 \cdot 20H_2O$	60° $r > v$ str (75+3°)	TCL pos tab	010 perf	Greenish-yellow	H 2.5 G 2.1 F 5	Diss by acids. Pleoc in green, abs X and Z > Y.		
1.554 ^	1.573	<u>1.575</u>	1.588	.015	AUGELITE $Al_2(Po_4)(OH)_3$	X ~ b el ctv pos Z:c = 35°	MCL tab 001	110 perf 201 good	Greenish-yellow	H 4.5-5 G 2.69 infus	Nearly insol in acids.		
	1.570	<u>1.576</u>	1.614	.044	ANHYDRITE $CaSO_4$	X = $\frac{a}{c}$ Z = $\frac{b}{c}$	ORTH ---	010 perf 100 less so 001 good	Col s, bluish	H 3.5 G 2.98 F 3	Diss by HCl.		
	1.573	<u>1.576</u>	1.579	.006	ILMAJOKITE $(Na, Ca, Ba)_2TiSi_3O_5$ (OH)10 · xH ₂ O	44° $r < v$	MCL ---	2 at 72° perf	Bright yellow	H 1 G 2.20 fus	Dec by acids.		
	1.554	<u>1.577</u>	1.618	.064	BONATTITE $CuSO_4 \cdot 3H_2O$	(75+4°)	---	MCL	010 dist	Pale blue	G 2.66 F 3	Sol in H ₂ O. Hydrates to chalcanthite. Tw pl 100 common.	
1.54	<u>1.578</u>	~1.60	.06	FICHTELITE $C_{19}H_{34}$	87°	opt pl Z:c = 13°	001 perf Tol dist	White	H 1 G 1.03 F 46°C	001 perf Tol dist	Sol in ether. Tw pl 001. Dimethyl- isopropyl-perhydro- phenanthrene.		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, fusibility	Remarks
α	β	γ										
1.575	1.578	1.584	.009	NIFONTOVITE $\text{Ca}_3\text{B}_6\text{O}_6(\text{OH})_{12} \cdot 2\text{H}_2\text{O}$	76° $r > v$ str	e1 pos	MCL	Poor, e1	Col s	H 3.5 G 2.36	Diss by acids. Anom interf colors.	
1.576	1.579	1.597	.031	COOKITE (Chlorite grp) $\text{LiAl}_4(\text{Si}_3\text{Al})_0(\text{OH})_8$	0-80° most ~ 50° (45-14°)	$Z \sim \frac{c}{x}$ edge e1 neg	MCL tab 001	001 mic	White, pink	H 2.5 G 2.67	Insol in acids. Exfoliates when heated. Base shows 6 biax segments.	
1.568	1.578	1.579	.012	GIBBSITE $\text{Al}(\text{OH})_3$	0-5° $r < v$ or $r > v$	$Y = \frac{b}{c}$ $Z:c = -30^\circ$	MCL	001 perf ps hex	White, brownish	H 3 G 2.3-2.4 infus	Diss by hot concd H_2SO_4 or by hot alkalies.	
1.588	1.578	1.583	.005	BAVENITE $\text{Ca}_4\text{Be}_2\text{Al}_2\text{Si}_9\text{O}_{26}(\text{OH})_2$	26-60°	$X = \frac{c}{b}$ $Z = \frac{a}{b}$ e1 neg	ORTH fib	100 good 001 fair	Col s, pink	H 5.5 G 2.71	Apparent tw pl 100.	
1.591	1.543	1.580	.074	HAMBURGITE $\text{Be}_2\text{B}_3\text{O}_3(\text{OH},\text{F})$	90° $r > v$ wk	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH pris c	010 perf 100 good	White	H 7.5 G 2.37 infus	Insol in acids except HF.	
1.583	1.580	1.596	.016	NORDSTRANDITE $\text{Al}(\text{OH})_3$	Small	$Z:c = 33^\circ$ e1 neg	TCL	---	White	G 2.43	Poly tw.	
1.567	1.581	1.638	.071	KORNELITE $\text{Fe}_2(\text{SO}_4)_3 \cdot 7\text{H}_2\text{O}$	49-62° $r > v$ perc	$Z = \frac{b}{c}$ $X:c = 17^\circ$	MCL pris	010 good	Pale rose, violet	G 2.3	Sol in H_2O . Poly tw pl 100.	
1.573	1.581	1.585	.012	SLAWSONITE (Feldspar grp) $(\text{Sr},\text{Ca})\text{Al}_2\text{Si}_2\text{O}_8$	82° $r < v$	$Z = \frac{b}{c}$ $X:a = 11^\circ$	MCL pris	001 good 100 fair	Light gray	H 5.5 G 3.12	----	
1.570	---	1.582	>.05	BUKOVSKYITE $\text{Fe}_2(\text{AsO}_4)(\text{OH}) \cdot 7\text{H}_2\text{O}$	---	ext 18°	MCL (?) needles	---	Yellow to gray-green	G 2.34	Diss by HCl. Opt char unk.	
1.570	1.577	1.582	.018	WAGNERITE $(\text{Mg},\text{Fe})_2(\text{PO}_4)\text{F}$	32° (64+14°) $r > v$ perc	$Y = \frac{b}{c}$ $Z:c = -22^\circ$	MCL	100, 120, 010 imperf	Col s, flesh, yellow	H 5-5.5 G 3.23 F 4	Diss by acids. FeO 9.9, MnO 2.0%.	
1.569	1.578	1.582	.018	SUDOITE (Chlorite grp) $\text{Mg}_2(\text{Al},\text{Fe})_3\text{Si}_3\text{Al}_{10}\text{O}_{10}$ (OH) ₈	55°	$Z:c = 7^\circ$ e1 neg	MCL	001 perf	White	G 2.64	Di-octahedral chlorite.	
1.582	1.582	1.592	.010	XONOTLITE $\text{Ca}_6\text{Si}_6\text{O}_{17}(\text{OH})_2$	0-5°	$X = \frac{b}{c}$ $Z = \frac{c}{b}$ e1 pos	MCL and TCL fib	010 perf	Col s, pink	H 6 G 2.69 F 2.5	Dec by HCl with sepn of silica.	

1.560	<u>1.583</u>	.091	AJOITE (K-Na)Cu ₇ AlSi ₉ O ₂₄ (OH) ₆ 3H ₂ O	80° (76+3°)	Z:c on 010 = 15°	TCL	---	Bluish-green.	G 2.96	Dec by acids. Pleoc. x light bluish-green, y and z brilliant bluish-green. ---			
1.580	<u>1.583</u>	.022	NORDSTRANDITE Al(OH) ₃	24° (44+16°)	e1 neg	TCL	110 perf	Cols, white	H ~ 3 G 2.42				
1.580	<u>1.583</u>	.010	WHITEITE Ca(Fe ⁺² ,Mn) ₂ Mg ₂ (PO ₄) ₄ Al ₂ (OH) ₂ •8H ₂ O	40-50° r < v perc	X = <u>b</u> Y = <u>a</u>	MCL	---	Tan	G 2.58	---			
1.580	<u>(1.583)</u>	.010	MISERITE K(Ca,Ce) ₄ Si ₅ O ₁₃ (OH) ₃	60-78°	X = <u>c</u> Z = <u>a</u> el neg	TCL fib	100 good 010 imperf	Pink to violet	H 2.5 G 2.84	Insol in acids.			
1.593	<u>1.585</u>	.008	CLINOPHLORE (Chlorite grp) (Mg,Fe) ₃ Al(Si ₃ Al) ₁₀ (OH) ₈	0-15° r < v	Z ~ <u>c</u> el neg	MCL	001 perf	Cols to green	H 3 F 5	Insol in acids. FeO 3.3, Fe ₂ O ₃ 2.1%.			
1.571 ^	<u>1.585</u>	.008	SIGLOITE (Fe ⁺² ,Fe ⁺³)Al ₂ (PO ₄) ₂ (0,OH) ₂ •8H ₂ O	76° (81+4°) r < v str	---	TCL	010 perf 001 good	Straw- yellow	H 3 G 2.35	Diss by acids. Oxidation product of chilidrenite.			
1.601	<u>1.586</u>	.056	EAKERITE Ca ₂ SnAl ₂ Si ₆ O ₁₈ (OH) ₂ 2H ₂ O	35° r > v str	X = <u>b</u> Y:c = 23.5°	MCL	Conch	Cols to white	H 5.5 G 2.93	Insol in acids.			
1.584	<u>1.586</u>	.016	SCHOLZITE CaZn ₂ (PO ₄) ₂ •2H ₂ O	70° r < v	X = <u>a</u> Y = <u>b</u>	ORTH	100 fair	Cols	H 3-3.5 G 3.11	Diss by acids.			
1.581	<u>1.586</u>	.015	COOKITE (Chlorite grp) LiAl ₄ (Si ₃ Al) ₁₀ (OH) ₈	60°	Z ~ <u>c</u> X edge el neg	MCL tab 001	001 mic	White, pink	H 2.5 G 2.61	Insol in acids. Ex- foliates when heated. Base shows 6 biax segments.			
1.593	<u>1.588</u>	.014	BAVENITE Ca ₄ Be ₂ Al ₂ Si ₉ O ₂₆ (OH) ₂	60°	X = <u>c</u> Z = <u>b</u> el neg	ORTH fib	100 good 001 fair	Cols, pink	H 4.5-5 G 2.73	Apparent tw pl 100.			
1.579 ^	<u>1.584</u>	.014	PARASCHOIZITE CaZn ₂ (PO ₄) ₂ •2H ₂ O	25° r > v	X = <u>b</u> Z:c = 13°	MCL	Parting 100	Cols to white	H 4 G 3.12	Tw on 100.			
1.579 ^	<u>1.588</u>	.007	MANGANESE- HOERENESITE (Mn,Mg) ₃ (AsO ₄) ₂ •8H ₂ O	65-70° (71+8°)	X = <u>b</u> Z:c = 31° el pos	MCL acic	010 perf	White	G 3.12	Diss by acids.			
1.580	<u>1.589</u>	.016	HEIDORNITE Na ₂ Ca ₃ B ₅ O ₈ (OH) ₂ (SO ₄) ₂ C ₁	63-77° r < v	Y = <u>b</u> Z:a = 23°	MCL spear-like	001 perf	Cols	H 4-5 G 2.75	Diss by acids.			
1.589 neg ^	<u>1.584</u>	.010	CELSIAN (Feldspar grp) BaAl ₂ Si ₂ O ₈	86°	Y = <u>b</u> Z:a = 28°	MCL 010 good	001 perf 010 good	Cols	H 6 G 3.37 infus	Gel with HCl.			

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	Brief	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
[]	1.583	<u>1.589</u> (1.601)	(.018)	SUDOITE (Chlorite grp) $Mg_2(Al,Fe^+3)_3Si_3Al_{10}(OH)_8$	70°	---	MCL	001 good	Pale green	Mg 14.1, Al 0.2, Fe 0.2, Mn 0.3%.	
1.533 1.558 1.623	1.590	1.629	.071	SZOMOLNOKITE (Kieserite grp) $(Fe,Mg)Si_4H_2O$	Large $r > v$	Y = b	MCL	---	Yellow	Slowly sol in H_2O . Fe:Mg = 3:2.	
1.587	<u>1.590</u>	1.597	.010	BULTFONTEINITE $Ca_2Si_2(OH,F)_4$	70° $r > v$	Z:c on $01\bar{0} = 28^\circ$ $100 = 47^\circ$	TCL spher	010, 001 fairly good	Pink	Gel with HCl. Poly tw.	
1.580	1.560	<u>1.591</u>	1.631	.071	HAMBURGITE $Be_2B_3(OH,F)$	87° $r > v$ wk	X = $\frac{a}{b}$ Y = $\frac{c}{d}$	010 perf 100 good	White	H 2.5 G 2.8 fus	
1.590	1.590	1.592	.037	CATAPLEITE $(Na_2,Ca)ZrSi_3O_9 \cdot 2H_2O$	Small $r < v$ wk	---	HEX	1010 perf 100 good	Yellow, brown	Insol in acids except HF.	
1.586	1.592	1.614	.028	COLEMANITE $Ca_2B_6O_11 \cdot 5H_2O$	56° $r > v$ wk	X = b Y:c = -6°	MCL	010 perf 001 dist	Col s	Gel with acids. Complex tw. Unax when heated to 200°C.	
1.588	1.592	1.598	.010	ALTHAUSITE $Mg_2(Po_4)(OH,F,O)$	~ 70°	X = c Y:c = -6° el neg	ORTH mass	010 perf 100 dist	Gray	Diss by hot HCl.	
1.586	1.588	<u>1.593</u>	1.602	.014	SCHOLZITE $CaZn_2(Po_4)_2 \cdot 2H_2O$	65° $r < v$	X = a Y = d	ORTH	100 fair	Col s	Diss by acids.
(1.583)	1.586	<u>1.593</u>	1.600	.014	MISERITE $K(Ca,Ce)_4Si_5O_{13}(OH)_3$	52-78°	X = c Z = a el neg	TCL fib	100 good 010 imperf	Col s	H 3-3.5 G 3.11
1.562	1.595	1.632	.070	SIMIKITE $MnSi_4H_2O$	~ 90°	Z = b	MCL fib or platy	One perf	White to rose	Insol in acids.	
1.590	1.595	1.602	.012	CUSPIDINE $Ca_4Si_2O_7(F,OH)_2$	62° $r > v$	Y = b Z:c = 7° disp perc	MCL spear-shaped	001 dist	Pale rose, col s	So l in H_2O .	
1.594	1.595	1.599	.005	COESITE SiO_2	54-64° $r < v$ wk	X = b Z:c = 5°	MCL ps hex	Subconch	Col s	Gel with acids. Poly tw.	
1.600										Insol in acids.	

Gel with acids.

1.594	<u>1.595</u>	1.598	.004	FOSHAGITE Ca ₄ Si ₃ O ₉ (OH) ₂	---	el pos	TCL fib	---	White	H 3 G 2.36 infus		
1.590	<u>1.596</u>	1.616	.026	PANASQUEIRAITE CaMg(Po ₄)(OH,F)	51°	Z = b X:c = 22°	MCL	010 poor	Pink	H 5 G 3.27	---	
1.592	<u>1.597</u>	1.616	.039	JOHANNITE Cu(HO ₂) ₂ (SO ₄) ₂ (OH) ₂ • 8H ₂ O	~ 90° r < v str	X ~ b disp str	TCL pris tab 001	100 good	Emerald- to apple-green	H 2-2.5 G 3.32 infus	Diss by acids. Poly tw. Pleoc str, X cols, Y pale yellow, Z greenish- or canary-yellow.	
											Diss by acids.	
1.593	<u>1.597</u>	1.619	.026	ISOKITE CaMg(Po ₄)F	39-51° r > v	Z = b X:c = 32° el neg	MCL fib	010 very good	Cols to buff	H 5 G 3.27		
1.596	<u>1.598</u>	1.632	.036	PETARASITE Na ₅ Zr ₂ Si ₆ O ₁₈ (Cl,OH)• 2H ₂ O	29° r < v wk	X = b Z:c = 41.5°	MCL	110 perf 001 dist	Greenish-yellow	H 5-5.5 G 2.88	Pleoc, X cols, Y and Z pale greenish-yellow.	
1.580	<u>1.598</u>	1.627	.047	VIVIANITE Fe ₃ (Po ₄) ₂ •8H ₂ O	Large r < v wk	X = b Z:c = 29° disp str el clv pos	MCL	010 perf	Pale blue to bluish-green	H 1.5-2 G 2.66 F 1.5	Diss by HCl. Nearly cols fresh, oxidizes to deep blue, then pleoc, X deep blue, Y pale blue, Z olive to brown.	
v	1.607											
1.595	<u>1.598</u>	1.610	.015	KOSTYLEVITE K ₂ ZrSi ₃ O ₉ •H ₂ O	48°	X = b Z:c = 45°	MCL	110 perf	Col s	G 2.74	Dec by HCl.	
1.598	<u>1.598</u>	1.605	.007	TAKOVITE Ni ₆ Al ₂ (OH) ₁₆ (CO ₃ ,OH)• 4H ₂ O	Small	el pos	TRIG sph	---	Blue-green	---	Diss by HCl.	
v	1.597	1.599	1.615	.018	AMESITE (Kaolinite-serpentine grp) Mg ₂ Al(SiAl) ₅ (OH) ₄	18° r < v	Z:c = 0- 20° el neg	MCL or TCL ps hex plates	001 mic	Pale green	H 2-3 G 2.8 infus	Slowly dec by HCl. FeO ~ 8%.
1.589	<u>1.593</u>	1.599	1.608	.015	CELSIAN (feldspar grp) BaAl ₂ Si ₂ O ₈	Y = b X:a on 010 = 65°	MCL	001 perf 010 good	Col s	H 6 G 3.35 infus	Gel with HCl. Fe ₂ O ₃ 2.2, BaO 38.6, K ₂ O 0.5%.	
v	1.593	---	1.608	.015	Unnamed silicate of K, Ca, Zr	---	MCL ps hex	---	Brown	---	(Am. Mineral. 50, 533 (1965)).	
1.590	<u>1.600</u>	1.634	.044	HAIDINGERITE CaHAsO ₄ •H ₂ O	58° r > v wk	X = b Z = c el pos	ORTH tab 010	010 perf	Col s	H 2-2.5 G 2.85 F 2.5	Diss by acids.	
1.595	<u>1.600</u>	1.631	.036	PETARASITE Na ₅ Zr ₂ Si ₆ O ₁₈ (Cl,OH)• 2H ₂ O	43° r < v	X = b Z:c = 41.5°	MCL	110 perf 010 very good	Greenish-yellow	H 5-5.5 G 2.88	Pleoc, X cols, Y and Z pale greenish-yellow.	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, fusibility	Remarks
	α	β	γ									
1.597 1.61	1.595 1.60	1.628	.033	CEBOLLITE (Ca,Na ₂) ₄ Al ₂ Si ₃ O ₁₂ (OH) ₂ (?)	58°	---	ORT Fib	ORT	001 perf 100, 010 good	White	H 5 G 2.96 F 5	Gel with acids. Alteration product of Mellite.
1.596	1.600	1.618	.022	CHIAVENNITE CaMnBe ₂ Si ₅ O ₁₃ (OH) ₂ 2H ₂ O	50° r > v	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el neg	ORT	001 perf 100, 010	Orange	H ~ 3 G 2.24	---	
---	~1.6	---	.014	MANANDITE LiAl ₄ BSi ₃ O ₁₀ (OH) ₈	17°	Z ~ $\frac{c}{b}$ el cTw neg	MCL	001 mic	Col s	G 2.89 F easy	Dec by H ₂ SO ₄ . Basal section divided into 6 biax segments.	
1.599 1.620	1.601 1.609	1.609	.010	CLINOCHLORITE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	12-22° r < v	Z ~ $\frac{c}{b}$ el cTw neg	MCL	001 perf	Col s to green	H 3 G 2.78 F 4	Nearly insol in acids. FeO 11.4, Fe ₂ O ₃ 4.3%.	
1.582	1.602	1.629	.047	GOLDICHITE KFe(SO ₄) ₂ ·4H ₂ O	82°	X = $\frac{b}{c}$ Z:C = 11°	MCL	---	Pale green	G 2.43	Sol in H ₂ O.	
1.579	1.603	1.629	.050	METAVIVIANITE Fe ⁺² _{3-x} Fe ⁺³ _x (PO ₄) ₂ (OH) _x ·(8-x)H ₂ O	85°	X ⊥ 110	TCL pris el c	110 perf	Leek-green	G 2.69 calc F 1.5	Diss by HCl. Pleoc. X light blue to blue- green, Y yellow-green to light green, Z yellow, abs X > Z > Y.	
1.593	1.603	1.623	.030	CHONDRDITE (Hunitite grp) (Mg,Fe) ₅ (SiO ₄) ₂ (F,OH) ₂	71° r > v wk	Z = $\frac{b}{c}$ X:C = 27°	MCL	100 poor	Yellow to red	H 6 G 3.18 infus	Gel with acids. Poly tw 001. Pleoc. X yellow, Y and Z nearly col s. FeO 2.8, F 7.9%.	
1.617	1.603	1.615	.021	NATROMONTEBRASITE (Na,Li)AlPO ₄ (OH,F)	Very large lam = 29°	TCL	2 dist	White	White	H 5.5 G 3.09 F easy	Diff diss by acids. Poly tw in 2 directions.	
1.594	1.603	1.630	.053	GUNNINGITE (Zn,Mn)SO ₄ ·H ₂ O	---	---	MCL fine-grained	---	White	H 2.5 G 3.2 infus	Sol in H ₂ O.	
1.608	---	1.630	.053	NATROMONTEBRASITE (Na,Li)AlPO ₄ (OH,F)	---	---	TCL acid b	100, 001 perf	Col s	H 4.5-5 G 2.90 F 2	Partly dec by HCl.	
1.599 1.636	1.604	1.638	.039	PECTOLITE NaCa ₂ Si ₃ O ₈ (OH)	59° r > v wk	---	---	---	---	---	---	
1.598	1.604	1.626	.028	METASSCHODERITE Al ₂ (PO ₄)(VO ₄)·6H ₂ O	(56+10°) Y:C = -20°	Z = $\frac{b}{c}$ Y:C = -20°	MCL	---	Yellow-orange	---	---	

1.580	<u>1.605</u>	.064	EUCHLORINE (K, Na) ₈ Cu ₉ (SO ₄) ₁₀ (OH) ₆ (?)	Med large (73+4°)	---	ORTH tab	2 clv	Emerald-green	---	Partly diss by H ₂ O. Pleoc, X pale green, Y green, Z yellow-green.
1.593	<u>1.605</u>	.034	FUKALITE Ca ₄ Si ₂ O ₆ (OH) ₂ (CO ₃)	(74+7°)	---	ORTH	---	H ~ 4 G 2.77	Eff with acids.	
1.600	<u>1.605</u>	.013	HYDROPHILITE CaCl ₂ (?)	Med	---	ORTH ps tet	Pris, perf	White	G 2.2 F 2	Sol in H ₂ O, deliq. Lam tw p ^f 110°. Data on synth compd.
1.598	<u>1.606</u>	.032	HUMITE (Humite grp) Mg ₇ (SiO ₄) ₃ (F, OH) ₂	59° r > v wk	X = <u>a</u> Z = <u>b</u>	ORTH tab	001 perf	Col s	H 6 G 3.20 infus	Gel with acids. Data for synth Mg-F compd.
1.596	<u>1.606</u>	.025	SCAWITTE Ca ₇ Si ₆ (CO ₃) ₁₈ ·2H ₂ O	74°	Y = <u>b</u> Z: <u>a</u> = 30°	MCL	001 perf 010 fair	Col s	H 4.5-5 G 2.74 infus	Gel and gel with acids.
1.600	<u>1.606</u>	.014	LATTIUMITE (Ca, K) ₈ (Al, Mg, Fe) (Si, Al) ₁₀ ·25(SO ₄)	83-90° r > v dist	Z = <u>b</u> X: <u>c</u> = 16- 28°	MCL tab	100 perf	White	G 2.93	---
1.587	<u>1.607</u>	.044	VIVIANITE Fe ₃ (PO ₄) ₂ ·8H ₂ O	Large r < v wk	X = <u>b</u> Z: <u>c</u> = 27° dist str el clv pos	MCL	010 perf	Blue	H 1.5-2 G 2.66 F 1.5 FeO 29.3, Fe ₂ O ₃ 14.0%.	Diss by HCl. Pleoc, X dark blue, Y light blue, Z nearly colorless.
1.598	<u>1.624</u>	.013	GRANDALLITE (?) CaAl ₃ (PO ₄) ₂ (OH) ₅ ·H ₂ O	70-75°	---	TCL (?) ps trig	Basal, perf	White	H 3 G 2.50 F 2.5	Diss with diff in HCl (Am. Mineral., 48, 1144 (1963)).
1.602	<u>1.607</u>	.013	ALDZHANITE CaMgB ₂ O ₄ Cl·7H ₂ O (?)	---	---	ORTH diopra- mida	---	Col s to rose	G 2.21	Opt char unk.
1.600	---	.020	CHURCHITE YPO ₄ ·2H ₂ O	Med disp dist	X = <u>b</u> Z: <u>c</u> = 35°	MCL laths el c	010, 100, 001	White, yellow	H 3 G 3.26 infus	Diss by hot acids.
1.600	<u>1.608</u>	.045	IMANDRITE Na ₁₂ Ca ₃ Fe ⁺³ ₂ Si ₁₂ O ₃₆	75°	---	ORTH	---	Honey-yellow	H 4 G 2.93	---
1.605	<u>1.608</u>	.007	MONTEBRASITE (Li, Na)AlPO ₄ (OH, F)	Large r < v	---	TCL	100 perf 100 good 011 dist	White	H 6 G 3.04 F 2	Diss by H ₂ SO ₄ . Poly tw in 2 directions at 90°. Li ₂ O 7.5%, Na ₂ O 3.5, F 1.4%.
1.603	<u>1.624</u>	.035	BRAZILIANITE NaAl ₃ (PO ₄) ₂ (OH) ₄	75° r < v wk	X = <u>b</u> Z: <u>c</u> = -20°	MCL	010 good fr conch	Yellow	H 5.5 G 2.98 F 4.5	Insol in HCl.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ	Biref								
1.618	1.604	1.609	1.615	.011	URALBORITE $\text{CaB}_2\text{O}_2(\text{OH})_4$	85° $r > v$ str	---	MCL pris	Parallel to el dist	H 4 G 2.60	Diss by HCl. Anom interf colors in blue and brown.	
	1.607	1.610	1.618	.011	TOPAZ $\text{Al}_2\text{SiO}_4(\text{F}, \text{OH})_2$	67° $r > v$ dist	$X = \frac{a}{c}$ $Z = \frac{c}{a}$ el pos el clv neg	001 perf	Col s, yellow	H 8 G 3.57 infus	Insol in acids. F 20.4%.	
	1.607	1.610	1.616	.009	BUCHWALDITE NaCaPO_4	---	el pos	0RTH el c	One clv	Col s	H < 3 G 3.21	Reported as opt neg.
	---	-1.61	---	wk	HARBORITE $\text{Al}_3(\text{PO}_4)_2(\text{OH})_3 \cdot 3\text{H}_2\text{O}$ (?)	---	---	Spher	---	White to brown	H 5-5.5 G 2.80	---
	1.610	1.610	1.611	.001	FOGGITE $\text{CaAl}(\text{PO}_4)(\text{OH})_2 \cdot \text{H}_2\text{O}$	40-45°	$X = \frac{c}{b}$ $Y = \frac{a}{b}$	0RTH plates	010 perf 100 good	Col s, white	H 4 G 2.78	---
	1.610	1.611	1.654	.044	CYCLOWOLLASTONITE CaSiO_3	Very small	---	TCL	---	Col s	H 5 G 2.91 F 4	Dec by acids. Also called Pseudowollastonite.
	1.608	1.612	1.621	.013	TUHALITE $(\text{Na}, \text{K})\text{Fe}^{+2}\text{Fe}^{+3}\text{Si}_6\text{O}_{15}$	60-70° $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{a}{b}$ el pos	0RTH	100, 010, 001 good	Dark blue to black	H 3-4 G 2.89	Insol in acids. Pleoc str, X pale pink, Y violet or lavender, Z deep purplish-blue.
	1.602	1.613	1.649	.047	ANAPITE $\text{Ca}_2\text{Fe}(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	54° $r > v$ perc	on 100, ext:c = 15°	TCL radiating	001 perf 010 dist	Greenish-white	H 3.5 G 2.81	Diss by acids.
	1.609	1.613	1.619	.010	STOKESITE $\text{Ca}_3\text{Si}_3\text{O}_9 \cdot 2\text{H}_2\text{O}$	70° $r < v$	$Y = \frac{b}{c}$ $X = \frac{a}{c}$	0RTH pyram	110 perf 010 imperf	Col s	H 6 G 3.19	---
	1.587	1.615	1.640	.053	MONETITE CaHPo_4	Large $r > v$ wk	---	TCL rhombs	3 pinacoidal, indist.	Yellowish-white	H 3.5 G 2.93 infus	Diss by acids.
1.649	1.608	1.615	1.630	.022	MAGNIOTRIPOLITE $(\text{Mg}, \text{Fe}, \text{Mn})_2\text{PO}_4(\text{F}, \text{OH})$	66° $r > v$ perc	$Y = \frac{b}{c}$ $Z:c = 53^{\circ}$	MCL	100 perf	Yellow	H 5.5 G 3.47	Diss by acids. Pleoc, X pale yellow, Y and Z deep yellow. $\text{Mg} 0.27.9$, $\text{Fe} 0.13.4$, $\text{Mn} 0.7.7\%$.

1.598	<u>1.617</u>	1.655	.067	CYANOTRICHITE $\text{Cu}_4\text{Al}_2(\text{SO}_4)(\text{OH})_{12} \cdot 2\text{H}_2\text{O}$	82° r < v str	$Z = \frac{c}{el pos}$	---	Bright blue	G 2.8 F 3	
1.603 1. 1.632	1.607 <u>v</u> 1.632	1.639	.032	CHONDROLITE (Humite grp) $(\text{Mg}, \text{Fe})_5(\text{SiO}_4)_2(\text{F}, \text{OH})_2$	80° (69+8°) r > v wk	$Z = \frac{b}{c}$ $X:c = 28^\circ$	MCL	100 poor	Yellow to red	
1.614	<u>1.617</u>	1.636	.022	HEMIMORPHITE $\text{Zn}_4\text{Si}_2\text{O}_7(\text{OH})_2 \cdot \text{H}_2\text{O}$	46° r > v str	$X = \frac{b}{c}$ $Z = \frac{c}{c}$	ORTH e1 c	110 perf 101 less so	H 6 G 3.26 infus	
1.616	<u>1.617</u>	1.622	.006	KURUMASAKTE (Zn, Ni, Cu)Al ₈ V ₂ Si ₅ O ₃₅ 27H ₂ O (?)	35°	---	fib	---	H 5 G 3.45 F 6	
1.618	<u>1.618</u>	1.670	.052	DIADODCHITE $\text{Fe}_2(\text{PO}_4)(\text{SO}_4)(\text{OH}) \cdot 5\text{H}_2\text{O}$	Small r > v str	---	TCL u mass	Uneven to conch	Greenish-yellow	
1.618	<u>(1.618)</u>	1.638	.032	KOETTIGITE (Zn, Mg) ₃ (AsO ₄) ₂ ·8H ₂ O	Large r < v	$X = \frac{b}{el pos}$	MCL fib c	010 perf	Yellow to brown	
1.627	<u>v</u>	1.606	<u>1.618</u>	1.618	.028	CLINOHUMITE (Humite grp) $\text{Mg}_9(\text{SiO}_4)_4(\text{F}, \text{OH})_2$	59° (74+8°) r < v	$Z = \frac{b}{X:c}$ $= 9^\circ$	White	H 3 G 2.0-2.4 fus
1.683	<u>v</u>	1.608	<u>1.618</u>	1.636	.028	PREHNITE $\text{Ca}_2\text{Al}_2\text{Si}_3\text{O}_{10}(\text{OH})_2$	69° r > v	ORTH	001 good	Diss by acids.
153	<u>v</u>	1.643	<u>1.610</u>	1.618	.025	PARGASITE (Amphibole grp) $\text{NaCa}_2\text{Mg}_4\text{Al}(\text{Si}_6\text{Al}_2)_0\text{O}_{22}$ 0.22(OH, F) ₂	61°	MCL pris c	001 perf at 124°	ZnO 27.7, MgO 7.2%.
1.634	<u>v</u>	1.613	<u>1.618</u>	1.635	.022	TOPAZ $\text{Al}_2\text{Si}_3\text{O}_4(\text{F}, \text{OH})_2$	61° r > v dist	$Y = \frac{b}{Z:c}$ $= 26^\circ$	Col s to brown	Diss by acids.
1.610 1. 1.631	1.616 <u>v</u> 1.631	1.618	<u>1.625</u>	.009	AFWILLITE $\text{Ca}_3\text{Si}_2\text{O}_4(\text{OH})_6$	55° r < v perc	ORTH e1 c clv neg	001 perf 100 good	Al ₂ O ₃ 11.1, Fe ₂ O ₃ 0.7, FeO 1.7, MgO 20.6, CaO 12.5, Na ₂ O 2.5, K ₂ O 1.2%.	
1.616	<u>1.619</u>	1.631	.015	AFTWILLITE $\text{Ca}_3\text{Si}_2\text{O}_4(\text{OH})_6$	55° r < v perc	$Y = \frac{b}{X:c}$ $= 30^\circ$ diSp str	MCL pris b	001 perf 100 good	Insol in acids.	
1.61	<u>1.62</u>	1.71	.10	BISBEITE $\text{CuSiO}_3 \cdot \text{H}_2\text{O}$ (?)	Small	e1 pos	ORTH thin laths	---	G 2.62 F diff	
									Bluish-white	
									Soft	
									Abs Z > (X, Y). Status of species in doubt.	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ $(2V_z \text{ calc})$	Optical orientation ($2V_z$ disp)	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.590	1.620	1.648	.058	NICKELBISCHOFITE $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$	87°	$Y = b$ $X:C = 8^\circ$	MCL	001 perf	Emerald-green	Soft	Sol H_2O , delid. Pleoc., X pale green, Z green.	
1.60	1.62	1.65	.05	CHAVESITE Hydrous phosphate of Ca, Mn	Large	ext:tw pl ~ 30°	TCL	2 good ~ 90°	Col s	H ~ 3	Poly tw (Am., Mineral., 43, 1148 (1985)).	
1.605	(1.620)	1.632	.027	CUMMINGTONITE (Amphibole grp) $(\text{Mg}, \text{Fe})_7\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	89°	Z:c = 14°	MCL	110 perf at 125°	---	H 4-4.5 G 3.04	$\text{FeO} 6.5$, $\text{MnO} 0.2$, $\text{MgO} 28.1$, $\text{CaO} 1.3$, $\text{Na}_2\text{O} 2.4\%$.	
1.640	1.620	1.648	.036	TURQUOISE $\text{CuAl}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 5\text{H}_2\text{O}$	40° (57+7°) r < v str	ext on IT0 = 12° 001 = 5° 010 = 34° disp dist	TCL u mass	001 perf 010 good	Sky-blue, bluish-green	H 5 G 2.84 infus	Diss by acids only after gentle heating. Pleoc., X col s, Z pale bluish.	
1.612	1.620	1.648	.036	METAHAWEEITE $\text{Ca}(\text{UO}_2)_2\text{Si}_6\text{O}_{15} \cdot x\text{H}_2\text{O}$	(63+7°)	---	---	---	Yellow to greenish-yellow	---	---	
1.611	1.620	1.645	.034	METAHAWEEITE $\text{Ca}(\text{UO}_2)_2\text{Si}_6\text{O}_{15} \cdot x\text{H}_2\text{O}$	5° r < v	Z ~ C	MCL	001 perf	Green	H 3 G 2.90 F 3.5	Nearly insol in acids. $\text{FeO} 19.1$, $\text{Fe}_2\text{O}_3 6.6\%$.	
1.601 1.630	1.620	1.625	.005	CHAMOSITE (Chlorite grp) $(\text{Fe}, \text{Mg}, \text{Al})_6(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_8$	5°	---	---	---	---	H 6 G 2.82-2.90	Insol in acids. Pleoc., X and Y col s, Z pale orange-yellow.	
1.627	1.593	1.621	1.666	.073	TINAKSITE $\text{K}_2\text{Na}(\text{Ca}, \text{Mn})_2\text{TiSi}_7\text{O}_{19}(\text{OH})$	76° disp str	---	TCL pris	010 perf 110 imperf	Pale yellow to pink	---	---
1.613 1.647	1.616	1.621	1.628	.012	STOKESITE $\text{CaSnSi}_3\text{O}_9 \cdot 2\text{H}_2\text{O}$	70° r < v	X = c Y = $\frac{c}{b}$	ORTH pyram	110 perf 010 imperf	Col s	H 6 G 3.19	Diss by acids. Pleoc.
1.618	1.622	1.658	.040	VESZELYITE $(\text{Cu}, \text{Zn})_3(\text{PO}_4)_3 \cdot 2\text{H}_2\text{O}$	39° r < v str	Y = b Z:c = 36°	MCL pris	001, 110	Dark bluish-green	H 3.5 G 3.09 fus	Slowly sol in H_2O .	
1.590	1.591	1.663	.072	SZOMONOKITE (Kieserite grp) $\text{FeSO}_4 \cdot \text{H}_2\text{O}$	80° r > v str	Y = b X:c = -26°	MCL pyram	Conch to uneven	Yellow to brown	H 2.5 G 3.05 fus	Slowly sol in H_2O .	

1.606 v 1.638	<u>1.623</u>	1.643	.036	HUMITE (Humite grp) (Mg,Fe) ₇ (SiO ₄) ₃ (OH,F) ₂	81° r > v wk	X = $\frac{a}{b}$ Z = $\frac{b}{d}$	001 perf 010	Yellow, brown, cols	H 6 6 3.24 infus
1.608 ^	<u>1.624</u>	1.644	.030	MONTEBRASITE LiAlPO ₄ (OH,F)	70° r > v str	---	TCL	100 perf 110 good	White
1.607 ^	<u>1.624</u>	1.650	.050	VIVIANITE Fe ₃ (PO ₄) ₂ •8H ₂ O	Large r < v wk	X = $\frac{b}{d}$ Z:c = 27° disp str	MCL	010 perf	Blue
1.744	<u>1.624</u>	1.642	.020	MANGFIELDITE (Variscite grp) Al(AsO ₄) ₂ •2H ₂ O	30° r > v str	X = $\frac{a}{b}$ Z = $\frac{b}{d}$	ORTH	201 imperf	White to green
1.622	<u>1.624</u>	1.633	.011	URANOPLITE (UO ₂) ₆ (SO ₄) ₂ (OH) ₁₀ •12H ₂ O	Small to large r > v str	Z = $\frac{b}{c}$ Y:c = 20°	NCL	010 perf	Bright yellow, straw yellow
1.622	<u>1.624</u>	1.631	.009	CELESTITE (Barite grp) SrSO ₄	51° r < v	Y = $\frac{b}{d}$ Z = $\frac{c}{e}$	ORTH Tab 001	001 perf 210 good	Cols to pale blue
1.637	<u>1.625</u>	1.637	.023	PARAHOPEITE Zn ₃ (PO ₄) ₂ •4H ₂ O	~90° r < v percr	X ~ $\frac{a}{b}$ Y:c = 30° on 100	TCL	010 perf	Col's F easy
1.614	<u>1.625</u>	---	.007	SMOLIANINOVITE (Co,Ni,Mg,Ca) ₃ (Fe ⁺³ ,Al) ₂ (AsO ₄) ₄ •11H ₂ O	---	el pos	ORTH fib	---	Yellow
---	<u>1.625</u>	---	.008	UDUMINELITE Ca ₃ Al ₈ (PO ₄) ₆ (OH) ₄ ₄	77°	---	ORTH acic	110	White
1.623	<u>1.626</u>	1.631	.008	FLUCKITE CaMnH ₂ (AsO ₄) ₂ •2H ₂ O	Large	---	TCL	010 perf 100 easy	Deep to pale rose
1.618 ^	<u>1.627</u>	1.642	.024	DIADOCHITE Fe ₂ (PO ₄) ₂ (OH) ₅ H ₂ O	Small (57+5°) r < v str	---	TCL	Uneven to conch	Yellow to brown
1.615	<u>1.627</u>	1.670	.055	ANTHOPHYLLITE (Mg,Fe) ₇ Si ₈ O ₂₂ (OH) ₂	79° (88+9°)	Y = $\frac{b}{d}$ Z = $\frac{c}{e}$ el pos	ORTH pris c at 126°	210 perf ---	Brown
1.616 ^	<u>1.628</u>	1.641	.025	TENTERITE CaY ₃ (CO ₃) ₄ (OH) ₃ •3H ₂ O	Large	---	ORTH fib	---	White
1.649	<u>1.622</u>	---	.020						Infus G 2.8
1.57									Diss by acids with eff.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ calc ($2V_z$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.618 1.647	1.625	1.629	1.655	.030	PREHNITE $\text{Ca}_2\text{Al}(\text{AlSi}_3)_0\text{Si}_{10}(\text{OH})_2$	63° (43+10°) $r > v$	$X = \frac{a}{b}$ $Y = \frac{v}{b}$	ORTH	001 good	Green	H 6 6.93 F 2	Dec by HCl with sepn of silica. Fe_2O_3 3.5, FeO 0.5%.
1.616 (1.63)	1.677	1.677	.061		CARBONATE-CYANOTRICHITE (Cu, Zn) ₄ Al ₂ (CO ₃ , SO ₄) ₂ (OH) ₁₂ • $2\text{H}_2\text{O}$	55-60° $r > v$ str	el neg	ORTH fib	---	Pale blue to azure-blue	H ~ 2 G 2.66 fus	Diss by acids. Pleoc, X and Y pale yellow, Z greenish-blue.
1.623	1.630	1.684	.061		GUILDITE $\text{CuFe}(\text{SO}_4)_2(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	Small to med (40+6°)	$Y = \frac{c}{c}$	MCL	100, 001 perf	Chestnut-brown	H 2.5 6.70	Diss by acids. Pleoc, X and Y pale yellow, Z greenish-yellow.
1.65	---	1.63	---	.02	HOMILITE (altered) $\text{Ca}_2(\text{Fe}, \text{Mg})\text{B}_2\text{Si}_{10}\text{H}_2\text{O}$	Med large, $r < v$ very str	---	MCL	One imperf	Yellow	H 5 G 3 F 2	Gel with acids.
1.620 λ	1.630	1.630	1.635	.005	CHAMOSITE (Chlorite grp) (Fe ⁺² , Mg, Fe ⁺³) ₅ Al (Si ₃ Al) ₁₀ (OH) ₈	Small	$Z \sim \frac{c}{c}$ el cl v neg	MCL	001 perf	Green	H 3 G 2.9	Nearly insol in acids. Pleoc, X and Y deep green, Z greenish brown. FeO 28.7, Al ₂ O ₃ 22.9%.
1.608 λ	1.623	1.631	1.657	.034	CHURCHITE $\text{YPO}_4 \cdot 2\text{H}_2\text{O}$	Small (59+8°) dis p̄ dist	$X = \frac{b}{c}$ $Z:c = 31^\circ$ 31°	MCL el c	010, 100, 001	White, yellow	H 3 G 3.2 infus	Diss by hot acids.
1.618 λ	1.629	1.631	1.638	.009	TOPAZ $\text{Al}_2\text{Si}_4(\text{F}, \text{OH})_2$	48° $r > v$	$X = \frac{a}{c}$ $Z = \frac{c}{c}$ el pos el cl v neg	ORTH el c	001 perf	Gray, brown, yellow	H 8 G 3.50 infus	Insol in acids. F 13.2%.
1.617 λ	1.619	1.632	1.653	.034	CHONDRODITE (Humite grp) (Mg, Fe) ₅ (SiO ₄) ₂ (F, OH) ₂	80° $r > v$ wk	$Z = \frac{b}{c}$ $X:c = 27^\circ$	MCL	100 poor	Yellow, red	H 6 G 3.28 infus	Gel with acids. Poly tw 001. Pleoc, X yellow, Y and Z pale yellow. FeO 10.5, MnO 1.2, F 5.4%.
1.605	1.633	1.703	.098		ROKJHNITE $\text{FeCl}_2 \cdot 2\text{H}_2\text{O}$	64° $r < v$	$X = \frac{b}{c}$ $Z:c = -49^\circ$	MCL	110 very good	Light green	---	Sol in H ₂ O.
1.618 1.632 1.651	1.634	1.651	.019		PARGASITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{Al}_2(\text{Si}_6\text{Al}_2)_0\text{Si}_{22}(\text{OH}, \text{F})_2$	64° (38+20°)	$Y = \frac{b}{c}$ $Z:c = 17^\circ$ el pos	MCL pris c at 124°	110 perf	Col s, brown	H 6 G 3.1	Insol in acids. FeO 4.0, Al ₂ O ₃ 13.3, CaO 10.2, Na ₂ O 2.6%.

1.628	<u>1.635</u>	1.698	.070	SARMIENTITE $\text{Fe}_2(\text{AsO}_4)(\text{SO}_4)(\text{OH}) \cdot 5\text{H}_2\text{O}$	38°	$\frac{Y = b}{Z:c = 12^\circ}$	MCL	---	Yellow-orange	G 2.58	Diss by acids.									
1.615	<u>1.635</u>	1.656	.041	SKLODOWSKITE $\text{Mg}(\text{UO}_2)_2\text{Si}_2\text{O}_6(\text{OH})_2 \cdot 5\text{H}_2\text{O}$	~90° r < v str	$\frac{Y = b}{Z:c = 12^\circ}$	MCL el b	100 perf	Lemon-yellow	H 2-3 G 3.64	Gel with acids. U opt. neg. Pleoc., X cols, Y pale yellow, Z yellow.									
1.632		1.617	<u>1.635</u>	TILLEYITE $\text{Ca}_5\text{Si}_{20}(\text{CO}_3)_2$	88° r < v perf	$\frac{X = b}{Z = \frac{b}{a}}$	MCL	100 perf another at 42°	White	G 2.84 infus	Eff and gel with acids.									
		1.618	<u>1.635</u>	1.650	.032	SEGELERITE $\text{CaMgFe}^{+3}(\text{PO}_4)_2(\text{OH}) \cdot 4\text{H}_2\text{O}$	Large	ORTH	Pale green	H 4 G 2.67	Pleoc., X and Y cols, Z yellow.									
		1.630	<u>1.636</u>	1.664	.034	HILGARDITE $\text{Ca}_2\text{B}_5\text{C}_{10}\text{O}_8(\text{OH})_2$	35° (50+8°) r > v	$\frac{Y = b}{Z:c = 1.5^\circ}$	MCL tab	010 perf 100 less so	H 5 G 2.71 F 2	Diss by acids.								
		1.630	<u>1.636</u>	1.664	.034	PARAHILGARDITE $\text{Ca}_2\text{B}_5\text{C}_{10}\text{O}_8(\text{OH})_2$	35° (50+8°) r > v	---	TCL	010, 100 perf	H 5 G 2.71 F 2	Diss by acids.								
		(1.639)		1.636	1.643	.011	ANDALUSITE Al_2SiO_5	Large r < v str	$\frac{X = a}{Z = \frac{c}{b}}$ el pos	010 perf	Cols, pink	H 7 G 3.15 infus	Insol in acids. FeO 1.5%.							
				1.632	<u>1.636</u>	1.660	.029	PECTOLITE $\text{Na}(\text{Ca}, \text{Mn})_2\text{Si}_3\text{O}_8(\text{OH})$	47° r < v str	$\frac{Z \sim b}{Y:a = 9^\circ}$ el pos	110 perf	Light red	H 5 G 2.97-3.13 F 2	Partly dec by acids. MnO 19.5, FeO 2.5%.						
					1.631	<u>1.636</u>	1.660	BARITE BaSO_4	37° r < v wk	$\frac{X = c}{Y = \frac{b}{d}}$ el pos	001 perf 210 less so	White, blue, yellow	H 5 G 4.50 F 3	Insol in acids.						
					1.636	<u>1.637</u>	1.648	.012	IRHTEMITE $\text{Ca}_4\text{MgH}_2(\text{AsO}_4)_4 \cdot 4\text{H}_2\text{O}$	---	$\frac{\text{ext. angle}}{25^\circ}$ el neg	---	White to pale rose	H 3-3.5 G 3.09	Diss by acids.					
						1.634	---	1.642	.008	---	MCL spher	---								
						1.624	<u>1.636</u>	<u>1.697</u>	1.648	0.012	---	001 perf	Yellow, brown	H 6 G 3.28 infus	Gel with acids. Pleoc., X and Z pale yellow, Y nearly cols. FeO 7.9, FeO 1.0, MnO 1.7, F 5.0%.					
							1.628	<u>1.638</u>	1.655	.027	HUMITE (Humite grp) (Mg, Fe) ₇ (SiO ₄) ₃ (OH, F) ₂	76° r > v wk	$\frac{X = a}{Z = \frac{b}{d}}$ el neg	010 perf 010	Pink	H < 4 G 3.30	---			
							1.623	<u>1.653</u>	1.638	.066	KRAUTITE $\text{MnAs}^{+5}\text{O}_3(\text{OH}) \cdot \text{H}_2\text{O}$	65+5°	X = b el cTV pos	MCL	010 perf T01 good	H 5 G 2.9 F 2	Diss by acids.			
							1.636	<u>(1.639)</u>	1.670	.032	STRONTIOHILGAROITE $(\text{Sr}, \text{Ca})_8\text{B}_{18}\text{O}_{33}\text{Cl}_4 \cdot 4\text{H}_2\text{O}$	19-46° r > v	---	TCL	010, 100 perf					
							1.625	<u>1.640</u>	1.696	.071	STRUNZITE $\text{MnFe}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	(56+4°)	Z:c = 10°	TCL ps mcl laths	---	Straw-to brownish-yellow	G 2.47-2.56 Diss by acids. Pleoc wk in yellow, abs Z > X, Y.			

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.637	<u>1.640</u>	1.662	.025	PENKVILKSITE $\text{Na}_4\text{Ti}_2\text{Si}_8\text{O}_{22} \cdot 5\text{H}_2\text{O}$	(41+13°)	---	MCL or ORTH	001 perf	White	H 5 6 2.58 fus	Gel with acids. Clotted mass of fibers.	
(1.620) 1.651	1.630 1.651	<u>1.640</u>	1.655	.025	CUMMINGTONITE (Amphibole grp) (Mg, Fe) $\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	78° $r > v$	$Y = \frac{b}{c}$ $Z:c = 20^\circ$ el pos	MCL pris \underline{c}	110 perf at 125°	Brown, green	H 6 G 3.15 fus	Insol in acids. Pleoc wk, X col, Y and Z pale green, FeO 16.8, FeO 1.9, Al ₂ O ₃ 2.8, MnO 30.8%.
1.64	<u>1.64</u>	1.66	.02	ROEBLINGITE $\text{Pb}_2(\text{Ca}, \text{Sn})_6(\text{Mn}, \text{Ca})$ $(\text{SO}_4)_2\text{Si}_6^{14}(\text{OH})_{10}$	Small	---	MCL u mass	001 perf 100 imperf	White	H 3 6 3.43 F 3	Gel with acids.	
1.640	<u>1.640</u>	1.657	.017	SARCOLITE (Ca, Na) ₇₋₈ Al ₄ Si ₆ O ₂₄ (OH) ₂ (?)	Small $r > v$ str	---	TET cubo-oct	---	Reddish-white,	H 6 G 2.7 F 3 (?)	Gel with acids.	
1.64	---	wk	NINGYOITE (U, Ca, Ce) ₂ (PO ₄) ₂ ·1·2H ₂ O	---	el pos	ORTH ps hex	---	Brown to brownish-green	---	Slightly pleoc in brown. Opt char unk.		
1.615	---	1.685	.070	LIKASITE $\text{Cu}_6(\text{PO}_4)_2(\text{NO}_3)_2(\text{OH})_7$	---	$X = \frac{a}{b}$ $Z = \frac{c}{d}$	ORTH ps hex	001 perf	Blue	G 2.97	Diss by acids. Pleoc, X greenish-blue, Z pale blue. Opt char unk.	
1.633	<u>1.641</u>	1.652	.019	FAIRFIELDITE $\text{Ca}_2\text{Mn}(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	Large $r > v$ med	ext: c on $100 = 40^\circ$, on 010 = 10° disp perc	TCL pris foliated	001 perf 010 good 110 dist	White, yellowish	H 3.5 G 3.05 F 4	Diss by acids. MnO 19.7, FeO 1.0%.	
1.650	<u>1.641</u>	1.651	.015	ROSCHERITE $\text{Ca}(\text{Al}, \text{Fe}, \text{Mn})_3\text{Be}_2(\text{PO}_4)_3$ $(\text{OH})_3 \cdot 2\text{H}_2\text{O}$ (?)	Med $r > v$ str	$X = \frac{b}{c}$ $Y:c = -15^\circ$	MCL and TCL pris	001 perf 010 good	Brown to olive-green	H 4.5 G 2.93 fus	Diss by acids	
1.637	<u>1.642</u>	1.670	.033	TRETSKITE $\text{Ca}_2\text{B}_5\text{O}_8(\text{OH})_2(\text{OH}, \text{Cl})$	46°	---	TCL fib	---	White to brownish	G 2.19 (?)	---	
1.632	<u>1.642</u>	1.657	.025	COLLINSITE $\text{Ca}_2(\text{Mg}, \text{Fe})(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	80°	el pos	TCL bladed	001, 010 fair	Light brown	H 3.5 G 2.99 fus	Diss by acids.	

1.657	1.638	1.642	1.653	.015	MULLITE Al ₆ Si ₂ O ₁₃	20-60° (63+16°) r > v perc.	Y = b Z = c el pos	ORTH acic c fib	010 perf	Col s, gray	H 6-7 G 3.23 infus	Insol in acids. Data for synth compd.
1.640	1.642	1.647	.007	JUANITE Ca ₁₀ Mg ₄ Al ₂ Si ₁₁ O ₃₉ 4H ₂ O (?)	50°	el pos	ORTH(?)	---	White, brownish	G 3.01 F 3	Dec by acids. Alteration product of Melilite.	
1.631	1.643	1.695	.064	RANSOMITE CuFe ₂ (SO ₄) ₄ •6H ₂ O	Med	---	MCL radiating el c	010 perf	Sky-blue	H 2.5 G 2.63	Diss by acids.	
1.618 1.679	1.632	1.643	1.664	.032	CLINOHUMITE (Humite grp) (Mg, Fe) ₉ (SiO ₄) ₄ (F, OH) ₂	74° r < v	Y = b X ~ c el pos	MCL	---	Yellow to brown	H 6 G 3.28 infus	Gel with acids. FeO 6.6, MnO 1.7, F 3.2%.
1.638	1.643	1.650	.012	MAGNESIO-ARFVEDSONITE (Amphibole grp)	80°	Y = b Z:c = 39° el pos	MCL pris c	110 perf at 124°	Blue-green	H 5-6 G 3.17	Insol in acids. FeO 8.3, MnO 6.0%.	
1.620	(1.644)	1.674	.054	ZIRCOSULFATE Zr(SO ₄) ₂ •4H ₂ O	75°	---	ORTH	---	White	H 2.5-3 G 2.95	Sol in H ₂ O.	
1.640	1.644	1.650	.010	RANKINITE Ca ₃ Si ₂ O ₇	64°	Y = b X:c = 15° el pos	MCL	---	Col s infus	G 2.96 infus	Gel with acids.	
1.618	1.645	1.679	.061	ERYTHRITE (Co, Ni, Mg) ₃ (AsO ₄) ₂ •8H ₂ O	Large r < v wk	X = b Z:c = 39° el pos	MCL laths 010	010 perf	Crimson, peach-color	H 2 G 3.1 F 2	Diss by acids. Pleoc, X pale pink, violet, Z red. CoO 14.2, NiO 9.2, MgO 3.2%.	
1.641	1.645	1.672	.031	KURGANITAITE (Sr, Ca) ₂ B ₄ O ₈ •H ₂ O (?)	Small (43+11°)	el pos	TCL(?) mass	---	White	H ~ 6 G 2.9-3.0	Diss by acids. Perhaps = strontio-hilgardite.	
1.640	1.645	1.652	.012	JADEITE (Pyroxene grp) Na(Al, Fe ⁺³)Si ₂ O ₆	67° r > v mod	Y = b Z:c = 40° el pos	MCL fib c	110 good at 87°	Col s to green	H 6 G 3.25 F 2.5	Insol in acids. FeO 0.3, FeO 0.2, CaO 1.4%.	
1.667	1.643	1.645	1.651	.008	MOSANDRITE var. Rinkolite (Na, Ca, Ce) ₃ Ti(SiO ₄) ₂ F	45-80° r > v str	Y = b X:c = 3° el pos	MCL el c	100 dist 010 poor	Yellow to reddish-brown	H 4-4.5 G 3.0-3.4 F 3	Gel with acids. Pleoc wk, abs Z > Y > X.
1.643	1.645	1.649	.006	PELLYLITE Ba ₂ Ca(Fe ⁺² , Mg) ₂ Si ₆ O ₁₇	47° r > v very str	---	ORTH	Conch prism poor	Conch pale-yellow	H 6 G 3.51	Dec by HCl.	
1.638 1.658	1.647	1.682	.044	VESZELYITE (Cu, Zn) ₃ (PO ₄) ₃ •2H ₂ O	56° r < v str	Y = b Z:c = 36° el pos	MCL pris	001, 110	Dark bluish-green	H 3.5 G 3.43 fus	Diss by acids. Pleoc, X blue, yellow-green.	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ								
1.629	1.637	<u>1.647</u>	1.668	.031	PREHNITE $\text{Ca}_2\text{Al}_2\text{Si}_3\text{O}_{10}(\text{OH})_2$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	001 good	Dark green	H 6 G 2.97 F 2	
	1.642	<u>1.647</u>	1.672	.030	KORZHINSKITE $\text{CaB}_2\text{O}_4\text{H}_2\text{O}$	44°	ORTH(?) pris el pos	One cleavage el	Col s	---	
1.64 1.66	1.642 <u>(1.647)</u>	1.653	.011		HJORTDAHLITE $(\text{Ca}, \text{Na})_3\text{ZrSi}_2\text{O}_7$ $(\text{F}, \text{OH}, \text{O})_2$	~ 90° r > v perh	opt pl ~ 111 Ext. on 100 = 65°	TCL tab 100 dist	Yellow to brown	H 5.5 G 3.25 fus	
	1.637	<u>1.648</u>	1.676	.039	LOSEYITE $(\text{Mn}, \text{Zn})_7(\text{CO}_3)_2(\text{OH})_{10}$	64°	$Y = \frac{b}{c}$	MCL el b	---	Bluish-white	
	1.643	<u>1.648</u>	1.674	.031	REDDINGITE $(\text{Mn}, \text{Fe})_3(\text{PO}_4)_2 \cdot 3\text{H}_2\text{O}$	(48+9°) r > v dist	$Y = \frac{b}{c}$ $X = \frac{a}{d}$	ORTH oct, tab	Pink fr uneven	H 3-3.5 G 3.27 fus	
160	1.662				MAGNIOTRIPHITE $(\text{Mg}, \text{Mn}, \text{Fe})_2\text{P}_2\text{O}_4(\text{F}, \text{OH})$	60° (79+12°) r > v perh	MCL	100 perf	Yellow to brown	H 5.5 G 3.57 fus	
	1.615 1.652	1.641	<u>1.649</u>	1.661	.020		$Y = \frac{b}{c}$ $X \cdot a = 18^\circ$			Diss by acids. Pleoc., X and Y light yellow, Z pale yellow.	
	1.659	1.644	<u>1.649</u>	1.663	.019	MESSELITE $\text{Ca}_2(\text{Fe}, \text{Mn})(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	Small (62+13°)	ext. on 100 23° to c	COLS to brownish	H 3-3.5 G 3.1	
	1.628 1.655	1.645	<u>1.649</u>	1.661	.016	ANTHOPHYLLITE (Amphibole grp) $(\text{Mg}, \text{Fe})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	59°	$Y = \frac{b}{c}$ $Z = \frac{c}{a}$ el pos	ORTH pris c at 126°	Brown	H 6 G 3.24
	1.657	1.649	<u>1.649</u>	1.655	.006	BOEHMITE $\text{Al}_2\text{O}_3(\text{OH})$	Small	---	ORTH 100 less so	G 2.98 infus	
	1.588		<u>1.650</u>	1.722	.134	KRAUSITE $\text{KFe}(\text{SO}_4)_2 \cdot \text{H}_2\text{O}$	Large	$Z = \frac{b}{c}$ $Y \cdot c = -35^\circ$	MCL pris or flat tab	H 2.5 G 2.84	
								001 perf 100 good	Yellowish-green	Diss by acids. Pleoc. wk, X col s, Y and Z pale yellow.	

1.600	<u>1.650</u>	1.722	.122	BLJVVOETITE (Y,Dy,Gd) ₂ (UO ₂) ₄ (CO ₃) ₄ (OH) ₆ •11H ₂ O	84°	X = c Y ~ <u>a</u>	ORTH tab	One good	Yellow	G 3.9	Pleoc. X colts, pale yellow, Z dark yellow.	
1.626	<u>1.650</u>	1.686	.060	PSEUDOLAUITE MnFe ₂ (PO ₄) ₂ (OH) ₂ •8H ₂ O	80°	Z = b Y:a = 12°	MCL pris and thick tab	---	Orange- yellow	H 3 G 2.46	Diss by acids.	
1.64- 1.65	---	1.67- 1.68	.03	CASSIDYITE Ca ₂ (Ni,Mg)(PO ₄) ₂ •2H ₂ O	---	---	TCL fib	---	Pale- to bright- green	6 3.1-3.2	Opt char unk.	
1.633	---	1.672	.039	SHARPITE (UO ₂)(CO ₃)•H ₂ O (?)	---	el pos	ORTH(?)	---	Yellow- green	H 2 G > 3.3 infus	Diss by acids. Pleoc faint. Not fluor in UV.	
1.640	<u>1.650</u>	1.660	.020	FAIRFIELDITE (Fairfieldite grp) Ca ₂ (Mn,Fe)(PO ₄) ₂ •2H ₂ O	86° r > v med	ext:c on 100 = 40° 010 = 10° disp perc	TCL pris, foli- ated	001 perf 010 good T10 dist	White, yellowish	H 3.5 G 3.02 F 4	Diss by acids. MnO 14.8, FeO 4.75%.	
1.641	---	1.65	---	HOMILITE (altered) Ca ₂ FeB ₂ Si ₂ 10 ^x H ₂ O	Small r > v or r < v str	---	MCL mass	---	Yellow	H 5 G 3.4 F 2	Gel with acids.	
1.63	---	1.65	---	SAMUELSONITE (Ca,Ba)Ca ₈ (Fe,Mn) ₄ Al ₂ (PO ₄) ₁₀ (OH) ₂	Large	X:c = 22°	MCL pris	001 fair	Col s	H 5 G 3.27-3.33		
1.725	<u>v</u> <u>v</u>	1.650	1.655	PARASYMPLESITE Fe ₃ (AsO ₄) ₂ •8H ₂ O	83°	X = b Z:c = 31° el pos	MCL tab 010	010 perf	Greenish- blue	H 2 G 3.07	Diss by acids. Pleoc, X bluish- green, Z yellowish.	
1.648	<u>1.651</u>	1.687	.065	VUONNEMITE Na ₄ TiNb ₂ Si ₄ O ₁₇ •2Na ₃ PO ₄	53° (66-6°)	Y:c = 4°	TCL	001 and 2 others perf	Light yellow	H 2-3 G 3.13	Dec by H ₂ O.	
1.660	1.639	<u>1.651</u>	1.681	FORSTERITE (Olivine grp) Mg ₂ SiO ₄	85° r < v	X = b Y = <u>c</u>	ORTH equant	010, 001 poor	Col s to green	H 7 G 3.22 infus	Gel with acids. Data for synth compd (Fo ₁₀₀).	
1.664	<u>v</u>	1.651	1.669	FORSTERITE (Olivine grp) Mg ₂ SiO ₄	89° r > v	Y = b Z:c = 19° el pos	MCL pris c	110 perf at 125°	Brown, green	H 6 G 3.10 fus	Insol in acids. Pleoc wk, X colts, Y and Z brown-green. FeO 11.1, MnO 13.2%.	
1.640	1.638	<u>v</u> 1.664	1.665	TIRODITE (Cummingtonite ser., Amphibole grp) Mn ⁺² (Mg,Fe ⁺²) ₅ Si ₈ O ₂₂ (OH) ₂	89°							

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.634 $\hat{\wedge}$ 1.652 neg	1.641 <u>1.651</u>	1.664 γ	PARGASITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ Al (Si ₆ Al ₂) ₂₂ (OH,F) ₂	82°	$y = \frac{b}{c}$ $z:c = 18^\circ$ el pos	MCL pris c	110 perf at 124°	Brown	H 6 G 3.18	Insol in acids. Pleoc wk, X col s to yellow, Y and Z brown to green. FeO 8.9, Fe ₂ O ₃ 1.1, Al ₂ O ₃ 15.3, CaO 12.2%.
1.632 <u>1.652</u>	1.693 <u>1.652</u>	.061 .024	KORITNIGITE Zn(AsO ₃)(OH)•H ₂ O	70°	---	TCL	010 perf 001, 100	Col s to white	H 2 G 3.54	Diss by acids.
1.649 $\hat{\wedge}$ 1.673	1.648 <u>1.652</u>	1.672 <u>(1.652)</u>	TRIPLITE (Mn,Fe,Mg,Ca) ₂ P0 ₄ (F,OH)	40°	$y = \frac{b}{c}$ $x:a = 18^\circ$	MCL	100 perf 010 poor	Brown	H 5.5-6 G 3.6 F 3	Diss by acids. Pleoc in yellow. MnO 31.5, FeO 11.4, MgO 9.9%.
1.681	1.643 \wedge	1.675 <u>1.653</u>	NAGELSCHMIDTITE Ca ₃₋₄ (Si,P) ₂ O ₈	0-20°	---	MCL (?)	001 good 110 fair	Col s	G 3.04	---
1.638 \wedge	1.653 <u>1.653</u>	.032	HUMITE (Humite grp) (Mg,Fe) ₇ (SiO ₄) ₃ (OH,F) ₂	68° r > v wk	$x = \frac{a}{b}$ $z = \frac{c}{b}$	ORTH tab	010 perf 001	Brown	H 6 G 3.27 infus	Gel with acids. Pleoc, Y pale yellow, X and Z golden-yellow. FeO 7.8, Fe ₂ O ₃ 2.7, MnO 0.7%.
1.652 $\hat{\wedge}$	1.653 <u>1.653</u>	1.674 .022	KOTOITE Mg ₃ B ₂ O ₆	21° r > v	$x = \frac{a}{b}$ $y = \frac{a}{b}$	ORTH mass	110 perf	Col s	H 6.5 G 3.10 infus	Diss by warm acids.
1.665	1.651 <u>1.653</u>	1.659 .008	GÖTZENITE (Ca,Na) ₃ (Ti,Al)Si ₂ O ₇ (F,OH) ₂	~ 60° r > v str	$x \sim c$	TCL	100 perf 001 good	Col s, pink	H 6 G 3.11 F 3	Gel with acids. Poly tw.
1.669	1.650 <u>1.653</u>	1.658 .008	ENSTATITE (Orthopyroxene ser, Pyroxene grp) Mg ₂ Si ₂ O ₆	59° r < v	$x = \frac{b}{c}$ $y = \frac{a}{c}$ el pos	ORTH pris c	210 good at 88°	Col s	H 5-6 G 3.18 F 6	Insol in acids. Tw 100. Data for synth compd En 100.
1.652 $\hat{\wedge}$ 1.663	1.62 <u>1.654</u>	1.689 .069	ANNABERGITE Ni ₃ (AsO ₄) ₂ •8H ₂ O	~ 90°	$x = \frac{b}{c}$ disp dist el clv pos	MCL	010 perf	Apple-green to white	H 2.5-3 G 3.0-3.1 F 4	Diss by acids. U neg.
1.636 $\hat{\wedge}$ 1.671	1.650 <u>1.654</u>	1.670 .020	ANDALUSITE, manganian (Al,Mn) ₂ SiO ₅	70° (54+13°) r < v str	$x = \frac{a}{c}$ $z = \frac{c}{c}$ el pos	ORTH fib c	110 perf	Green	H 7 G 3.17 infus	Insol in acids. Pleoc, X yellow-green, Y emerald green, Z golden yellow. Mn 20.3, Fe ₂ O ₃ 3.7%.

1.650	<u>1.654</u> ·678	1.661	.011	VLADMIRITE $H_2Ca_5(AsO_4)_4 \cdot 5H_2O$	70° r > v str	Z:c = 36°	MCL	---	Col s	H 3.5 6 3.14 F diff	Diss by acids. Reported as biax neg.		
1.651	<u>1.654</u> ·678	1.660	.009	CLINENSTATITE (Pyroxene grp) $Mg_2Si_2O_6$	54°	X = $\frac{b}{d}$ Z:c = 22°	MCL el c	110 good at 87°	Col s	H 5-6 6 3.28 infus	Insol in acids. Poly tw 100 character- istic. Synth compd.		
1.652	---	1.659	.007	YUKSPORITE (Na,K) ₄ (Ca,Sr,Ba) ₄ (Ti,Al,Fe) ₃ Si ₈ O ₁₆ (F,Cl) ₂ ·4H ₂ O (?)	---	el neg	MCL(?) fib, platy	---	Pink to red	---	---	---	
1.648	<u>1.655</u> ·662	1.670	.022	Unnamed phosphate of Ca, Ba, Mn, Fe, Al	80°	---	fib	---	Greenish-	---	Al 20.3 8.0, Fe 0.3 7.6, Mn 8.6, CaO 24.2, BaO 3.5, H ₂ O 3.3%.		
1.648	1.655	1.662	.014	GEDRITE (Amphibole grp) (Mg,Fe) ₅ Al ₂ (Si ₆ Al ₂) ₀ ₂₂ (OH) ₂	87°	Y = $\frac{b}{d}$ Z = $\frac{c}{pos}$ el pos	ORTH pris c	210 perf at 126°	Brown	H 6 G 3.26	Insol in acids. FeO 14.6, Fe ₂ O ₃ 1.3, Al ₂ O ₃ 13.3%.		
1.648	<u>1.655</u> ·662	1.662	.014	DICKINSONITE (K,Ra)(Na,Ca) ₅ (Mn,Fe,Mg) ₁₄ Al(Po ₄) ₁₂ (OH,F) ₂	~ 90° r > v str	X = $\frac{b}{d}$ Y:c = 15° el neg	MCL tab, foli- ated 001	001 perf	Olive- green	H 3.3.5 6 3.40 F easy	Diss by acids. Pleoc wk in green.		
1.649	<u>1.656</u> ·662	1.6714	.065	NATROCHALCITE $NaCu_2(SO_4)_2(OH) \cdot H_2O$	37° r < v str inclined	Y = $\frac{b}{d}$ X:c = -12° el clv pos	MCL el c	001 perf	Emerald- green	H 4.5 G 3.49 F easy	Diss by acids.		
1.649	1.656	---	.052	DIOPHASE $CuSiO_2(OH)_2$	43°	---	TRIG	1011 perf	Emerald- green	H 5 G 3.28 infus	Gel with acids. Pleoc wk, abs X > Y and Z.		
1.652	<u>1.656</u> ·662	1.672	.020	EUCLASE BeAlSiO ₄ (OH)	48° r > v	Y = $\frac{b}{d}$ Z:c = 42°	MCL el a	010 perf 100, 001 poor	Col s to brown	H 6-6.5 G 3.07 F diff	Insol in acids.		
1.643	<u>1.657</u> ·664	1.681	.038	LEUCOSPHENITE $BaNa_4Ti_2B_2Si_{10}O_{30}$	78° r > v	Z = $\frac{b}{d}$ Y ~ $\frac{c}{cTV}$ el cTV neg	MCL el a	010 perf 001 fair	Col s to brown	H 6-6.5 G 3.07 F diff	Insol in acids. Tw pl 001.		
1.648	<u>1.657</u> ·664	1.668	.020	BOEHMITE Al ₁₀ (OH) ₆	79°	---	ORTH	010 perf 100 less so	Col s	G 2.98 infus	Diss by hot NaOH soln.		
1.653	<u>1.657</u> ·664	1.671	.018	MULLILITE $Al_6Si_2O_13$	20-60° (57+14°) r > v perf	Y = $\frac{b}{d}$ Z = $\frac{c}{pos}$ el pos	ORTH acid c	010 perf	Gray, lilac	H 6-7 G 3.23 infus	Insol in acids. Fe ₂ O ₃ 0.9, TiO ₂ 2.3%.		
1.654	<u>1.657</u> ·670	1.666	.012	JADEITE (Pyroxene grp) Na(A,Fe)Si ₂ O ₆	70° r > v mod	Y = $\frac{b}{d}$ Z:c = 34°	MCL fib c	110 good at 87°	Col s to green	H 6 G 3.34 F 2.5	Insol in acids. Fe ₂ O ₃ 0.45, CaO 0.1, Na ₂ O 13.4%.		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ	Brief							
1.647 \wedge	1.640	<u>1.658</u>	1.760	.120 $\text{UO}_4 \cdot 2\text{H}_2\text{O}$	METASTUDTITE $(\text{Cu}, \text{Zn})_3(\text{PO}_4)_3 \cdot 2\text{H}_2\text{O}$	48°	---	ORTH fib	---	Yellow	6.4.67 calc Sol in hot HCl. Does not fluor in UV.
1.647 \wedge	1.640	<u>1.658</u>	1.695	.055 $(\text{Cu}, \text{Zn})_3(\text{PO}_4)_3 \cdot 2\text{H}_2\text{O}$	VESZELYITE $\text{LiAlSi}_2\text{O}_6$	71° $r < v$	$\gamma = \frac{b}{c} = 36^\circ$ $z:c = el$	MCL pris	001, 110	Dark bluish-green	Diss by acids. Pleoc., X blue. Z yellow-green.
1.671 \vee	1.653	<u>1.659</u>	1.677	.024 (Pyroxene grp)	SPODUMENE $\text{LiAlSi}_2\text{O}_6$	66° $r < v$	$\gamma = \frac{b}{c} = 25^\circ$ $z:c = el$	MCL pris \underline{c}	110 perf at 87°	Pink, gray, green	Insol in acids. Abs X > Y > Z in thick sections.
1.649 \wedge	1.653	<u>1.659</u>	1.676	.023 $\text{Ca}_2(\text{Fe}, \text{Mn})(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	MESSELITE $(62+11)^\circ$	25° $(62+2^\circ)$	ext on 100 20° to \underline{c}	TCL tab 100	One good	Col to brownish	Diss by acids. FeO 20.9, Fe ₂ O ₃ 0.8, MnO 3.9, MgO 1.4%.
1.651 \wedge	1.628	<u>1.660</u>	1.705	.077 $\text{Fe}_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	PARASYMPLESITE $\text{Fe}_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	~90° $(82+2^\circ)$	$\chi = \frac{b}{c} = 31^\circ$ $z:c = el$	MCL tab 010	010 perf	Greenish-blue	Diss by acids. Pleoc., X bluish-green, Y yellow, Z brownish-yellow.
1.718 \vee	1.645	<u>1.660</u>	1.715	.070 $\text{Cu}_8\text{Si}_8O_{22}(\text{OH})_4 \cdot \text{H}_2\text{O}$	PLANCHEITE $\text{Cu}_8\text{Si}_8O_{22}(\text{OH})_4 \cdot \text{H}_2\text{O}$	Large $(57+4)^\circ$	$el \sim z$	ORTH fib	---	Blue	H 5. G 3.4
1.592 neg \wedge	1.650	<u>1.660</u>	1.680	.030 $(\text{Fe}, \text{Al})\text{PO}_4 \cdot 2\text{H}_2\text{O}$	STRENGITE, aluminian $(\text{Fe}, \text{Al})\text{PO}_4 \cdot 2\text{H}_2\text{O}$	Med Large $r > v$	$\chi = \frac{a}{b}$ $y = \frac{c}{d}$	ORTH fib \underline{c}	---	Gray, brown, red	Diss by acids. Al ₂ O ₃ 12.5%.
1.710 \wedge	1.659	<u>1.660</u>	1.680	.021 Al_2SiO_5	SILLIMANITE Al_2SiO_5	20° $r > v$	$\gamma = \frac{b}{c} = el$	ORTH acidic \underline{c}	010 perf	White, brown	Insol in acids. Some vars pleoc., col to blue.
1.664 \hookrightarrow	1.656	<u>1.660</u>	1.668	.012 $\text{Ca}_2\text{MgAl}_2\text{BSi}_4\text{O}_15(\text{OH})$	MAGNESIOAXINITE $\text{Ca}_2\text{MgAl}_2\text{BSi}_4\text{O}_15(\text{OH})$	Large	---	TCL	---	Pale blue	H 6.5. G 3.18 fus
1.641	<u>1.660</u>	1.682	.041 $\text{Ba}_4(\text{K}, \text{Na})_2\text{Ti}_4\text{Al}_2\text{Si}_{10}O_{36} \cdot 6\text{H}_2\text{O}$	JONESITE $\text{Ba}_4(\text{K}, \text{Na})_2\text{Ti}_4\text{Al}_2\text{Si}_{10}O_{36} \cdot 6\text{H}_2\text{O}$	77°	$\chi = \frac{b}{c} = el$	ORTH bladed	010	cols	H 3-4. G 3.25	Insol in HCl. Fluor orange in short-wave UV.
1.676 \vee	1.660	<u>(1.661)</u>	1.663	.003 $\text{KFe}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6 \cdot 7\text{H}_2\text{O}$	PHARMACOSIDERITE $\text{KFe}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6 \cdot 7\text{H}_2\text{O}$	Large $r < v$	---	CUB	100 imperf	Olive-green to brown	Diss by acids. Poly tw.
1.629	<u>1.662</u>	1.729	.098 $\text{H}_2\text{Cu}_5(\text{AsO}_4)_4 \cdot 8\text{H}_2\text{O}$	LINDACKERITE $\text{H}_2\text{Cu}_5(\text{AsO}_4)_4 \cdot 8\text{H}_2\text{O}$	73° $r < v$	$\gamma = \frac{b}{c} = 26^\circ$ $x:el = 26^\circ$	MCL fib	010 perf	Apple-green	H 2-2.5. G 2-3 (?)	Diss by acids.

1.648 1.655	<u>1.662</u>	1.683	.028	REDDINGITE (Mn,Fe) ₃ (PO ₄) ₂ •3H ₂ O	65° r > v perc	X = $\frac{b}{a}$ Y = $\frac{b}{c}$	010 poor fr uneven	Pink	H 3-3.5 G 3.14 fus	Diss by acids. Pleoc, X col., Y pinkish-brown, Z yellow. MnO 38.4, FeO 12.7%.			
1.655 <u>1.667</u>	<u>1.662</u>	1.671	.013	DICKINSONITE (K,Ba)(Na,Ca) ₅ (Mn,Fe,Mg) ₁₄ Al(PO ₄) ₁₂ (OH,F) ₂	Med r > v str	X = $\frac{b}{a}$ Y:c = 15° el neg	001 perf uneven	0live-green	H 3.5-4 G 3.40 F easy	Diss by acids. Pleoc wk in green.			
1.658	1.668	1.668	.010	BORACITE Mg ₃ B ₇ O ₁₃ Cl	Large	X = $\frac{c}{a}$ Y = $\frac{b}{c}$	Conch to uneven	White to greenish	H 7-7.5 G 2.95 F 2	Slowly diss by HCl. Inverts to cub at 265°C.			
1.658	<u>1.662</u>	1.668	.010	PARABUTLERITE Fe(SO ₄)(OH)•2H ₂ O	44-87° (90+2°) r > v mod	X = $\frac{b}{a}$ Y = $\frac{b}{c}$	120 poor fr conch	Orange to brown	H 2.5 G 2.55	Diss by acids.			
1.658	1.663	1.737	.139	ERYTHRITE Co ₃ (AsO ₄) ₂ •8H ₂ O	~ 90° r < v wk	X = $\frac{b}{a}$ Z:c = 32° el clv pos	010 perf	Crimson	H 2 G 3.06 F 2	Pleoc, X pink, Y violet, Z red. CoO 33.4, FeO 4.0%.			
1.645 <u>1.667</u>	1.663	1.701	.072	LEUCOSPHENITE BaNa ₄ Ti ₂ B ₂ Si ₁₀ O ₃₀	76° r > v perc	Z = $\frac{b}{a}$ Y ~ $\frac{b}{c}$ el neg	010 perf 001 fair	Cols to brown	H 6-6.5 G 3.09 F diff	Insol in acids. Tw pl 001.			
1.645	<u>1.664</u>	1.691	.042	WILHELMVIERLINGITE CaMnFe ⁺³ (PO ₄) ₂ (OH)•2H ₂ O	(90+4°)	X = $\frac{b}{a}$ Y = $\frac{b}{c}$	010 ORTH	---	Yellow	H 4 G 2.58	Reported to be neg with 2V x yellow. X and Y light yellow.		
1.657 <u>1.667</u>	<u>1.664</u>	1.692	.055	FORSTERITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	87° (71+7°) r < v	X = $\frac{b}{a}$ Y = $\frac{b}{c}$	010, 001 equant	Cols to green	H 7 G 3.35 infus	Gel with acids. FeO 8.6, Fe ₂ O ₃ 0.4, MnO 0.2% (F693).			
1.657	1.664	1.686	.033	SERANDITE Na ₂ (Mn,Ca) ₂ Si ₃ O ₈ (OH)	36° (45+11°)	Z = $\frac{b}{a}$ X:c = 57°	TCL el b	100, 001 Rose-red	G 2.32	MnO 29.0, Fe ₂ O ₃ 1.3%.			
1.653 <u>1.674</u>	<u>1.664</u>	1.688	.028	CUMMINGTONITE (Amphibole grp) (Fe,Mg) ₇ Si ₈ O ₂₂ (OH) ₂	86° r > v	Y = $\frac{b}{a}$ Z:c = 17° el pos	MCL pris c	110 perf at 125°	Brown, green	H 7 G 3.27 fus	Insol in acids. Pleoc in green. FeO 28.0, MnO 1.0, Al ₂ O ₃ 2.4%.		
1.651 <u>1.677</u>	<u>1.664</u>	1.678	.027	ATTAKOLITE (Ca,Mn,Sr) ₃ Al ₆ (PO ₄ ,SiO ₄) ₇ •3H ₂ O	~ 84° r > v	---	ORTH mass	---	Pink	G 3.24 F easy	Partly dec by acids.		
1.655	<u>1.664</u>	1.675	.020	JUNITOITE CaZn ₂ Si ₂ O ₇ •H ₂ O	(90+14°) r < v, wk	---	ORTH hemi-morphic	010 good 100, 101 poor	Cols to milky-white	H 4.5 G 3.5 fus	Ge1 with acids.		
1.656	<u>1.664</u>	1.672	.016										

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ calc.	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ	Biref	($2V_z$ disp)							
1.687	1.657 (1.664)	1.675	.018	ROSENBUSCHITE (Ca, Na) ₃ (Zr, Ti)Si ₂ O ₈ F	80°	X:c = 9° el neg	TCL	010 perf	Yellow, brown	H 5-6 G 2.97 F easy	Gel with acids. Pleoc wk, X colis, Z yellow.	
	1.663	1.664	1.675	.012	GIANNETITE Silicate of Na, Ca, Mn, Zr, Ti	30°	Y:c = 22°	TCL prts	100 perf	Cols to pale yellow	---	Validity dubious.
(1.639)	1.665	1.704	.065	TRIANGULITE Al ₃ (UO ₂) ₄ (PO ₄) ₄ (OH) ₅ ·5H ₂ O	80°	---	TCL	---	Yellow	G 3.7	---	
1.653	1.662	1.665	.010	GÖTZENITE (Ca, Na) ₃ (Ti, Al)Si ₂ O ₇ (F, OH) ₂	62°	X ~ c	TCL	100 perf 001 good	Pink, yellow	H 6 G 3.17 F 3	Gel with acids. Poly tw.	
	1.663	1.666	1.673	.010	LITHIOPHILITE Li(Mn, Fe)PO ₄	60°	X = c Y = a el neg	ORTH el c	100 perf 010 less so	Salmon-pink	H 5 G 3.43 F 1.5	Diss by acids. Pleoc, X deep pink, Y pale yellow, Z pale pink. MnO 42.6, FeO 2.9%.
1.679	1.650	1.667	1.688	.038	LUDLAMITE (Fe, Mg, Mn) ₃ (PO ₄) ₂ ·4H ₂ O	83°	Y = b Z:c = -67°	MCL tab	001 perf 100 dist	Bright green	H 3.5-4 G 3.2 F 2.2-5	Diss by acids.
1.675	1.657	1.667	1.678	.021	FERRO-GEDRITE (Amphibole grp) (Fe, Mg) ₅ Al ₂ (Si ₆ Al ₂) ₂₂ (OH) ₂	87°	Y = b Z = c el pos	ORTH prts c	210 perf at 126°	Gray, brown	H 6 G 3.28	Insol in acids. FeO 18.3, Fe ₂ O ₃ 1.0, Al ₂ O ₃ 17.8%.
1.655 1.681	1.662	1.667	1.681	.019	MOSANDRITITE (Na, Ca, Ce) ₃ TiSi ₂ O ₇ (O, OH, F) ₂	45-80° (62+13°) r > v	Y = b X:c = 3°	MCL el c	100 dist 010 poor	Yellow to reddish-brown	H 4-4.5 G 3.3-3.4 F 3	Gel with acids. Fe Pleoc wk, abs Z > Y > X.
1.645	1.646	1.668	1.705	.059	COBALTKORITNIGITE (Co, Zn)(As ⁺⁵ O ₃) ₂ ·H ₂ O	78°	X:b = 3° Y:a =	TCL	010 perf 100 good	Deep purple	---	Pleoc, X deep violet, Y reddish-violet, Z bluish-violet.
	1.665	1.669	1.768	.103	NENADKEVICHITE (Na, Ca, K)(Nb, Ti)Si ₂ O ₆ (O, OH) ₂ ·2H ₂ O	38° (24+5°)	X = a Z =	ORTH	001 poor	Yellow, rose	H 5 G 2.76 F diff	Diss by H ₂ SO ₄ ·1s, Y pale yellow, Z pale rose.
1.686												

1.653 1.677	<u>1.669</u>	1.674	.009	ENSTATITE (Orthopyroxene ser, Pyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	80° r < v	$\frac{Y = \frac{a}{c}}{Z = \frac{c}{a}}$ el pos	210 good at 88°	Green	H 5.5 6 3.2 F 5.5	Insol in acids. FeO 7.0% Commonly faintly pleoc.	
1.660	<u>1.670</u>	1.770	.110	LABUNTSONITE (K,Ba,Na)(Ti,Nb) (Si,Al) ₂ (OH) ₇ ·H ₂ O	35° r > v	$\frac{Y = \frac{b}{c}}{Z:c}$ = 63°	MCL	T02 perf	Dark brown G 2.87	Pleoc wk, X colts, Z yellow.	
1.698	1.660	<u>1.670</u>	1.720	.101.	STRUNZITE MnFe ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	(93+2°)	$Z:c = 10^{\circ}$	TCL ps mc laths	---	Diss by acids. Pleoc wk in yellow, abs Z > X, Y.	
1.640 1.670 neg	1.619	<u>1.670</u>	1.685	.021.	OMPHACITE (Pyroxene grp) (Ca,Na)(Mg,Fe,Al)Si ₂ O ₆	66° r < v mod	$\frac{Y = \frac{b}{c}}{Z:c}$ = 35°	MCL pris c	110 good at 87°	Straw- to brownish- yellow Green	2.47-2.56 H 5-6 G 3.40 F 5
1.681	1.664	<u>1.670</u>	1.685	.021.	JADEITE (Pyroxene grp) (Na,Ca)(Al,Mg,Fe)Si ₂ O ₆	70° r > v mod	$\frac{Y = \frac{b}{c}}{Z:c}$ = 35°	MCL pris c	110 good at 87°	Green	6 3.35 G 3.35 F 2
1.657 1.681	1.662	<u>1.670</u>	1.682	.020	PUMPELLYITE Ca ₂ (Mg,Fe)Al ₂ (Si ₀ 4) (Si ₂ O ₇)(OH) ₂ ·H ₂ O	15° (64+14°) r < v str r > v	$\frac{Y = \frac{b}{c}}{X:c} = 15^{\circ}$	MCL	001 good 100 imperf	Green, bluish- green, brown	5.5-7 H 5.5-7 G 3.18 F 2
1.688	1.665	<u>1.670</u>	1.683	.018	ANDALUSITE, ferrian (Al,Mn,Fe) ₂ Si ₁₀ O ₅	71° r < v str	el pos	ORTH fib c	110 perf	Green	7 3.22 H 7 G 3.22 infus
1.654 1.722	1.662	<u>1.671</u>	1.691	.029	FERROTSCHERMAKITE (Amphibole grp) Ca ₂ (Fe ⁺² ,Mg) ₃ (Al,Fe ⁺³) ₂ Si ₆ Al ₂ O ₂₂ (OH,F) ₂	78° r > v	$\frac{Y = \frac{b}{c}}{Z:c} = 18^{\circ}$	MCL pris	110 good at 124°	Green, brown	6 3.2 H 6 G 3.2
1.669 neg 1.693 neg	1.660	<u>1.671</u>	1.687	.027	HINSDALITE (Beudantite grp) PbAl ₃ (PO ₄)(SO ₄)(OH) ₆	0-30°	Z = c	TRIG	0001 perf	white, gray	4.5 H 4.5 G 3.65 infus
1.670	1.668	<u>1.671</u>	1.689	.019	SPODUMENE (Pyroxene grp) LiAlSi ₂ O ₆	65° r < v	$\frac{Y = \frac{b}{c}}{Z:c} = 24^{\circ}$	MCL pris c	110 good at 87°	Pink, gray, green	6.5 H 6.5 G 3.19 F 3.5
1.659	1.668	<u>1.671</u>	1.682	.014	DIOPSIDE (Pyroxene grp) CaMgSi ₂ O ₆	57° r > v wk	$\frac{Y = \frac{b}{c}}{Z:c} = 39^{\circ}$	MCL pris c	110 good at 87°	Cols	5-6 H 5-6 G 3.28 infus
1.678	1.665	<u>1.672</u>	1.695	.030	MAGNESIUM CHLOROPHOENICITE (Mg,Mn) ₃ Zn ₂ (AsO ₄) ₆ (OH,O) ₆	r < v str	y el	MCL fib	perf	White	3-3.5 H 3-3.5 G 3.57
1.690	1.669	<u>1.672</u>	1.667	.008						Diss by acids.	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula ^a	$2V_z$ ($2V_z$ calc.)	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.672	<u>1.672</u>	1.676	.004	FILLLOWITE $\text{Na}_2\text{Ca}(\text{Mn},\text{Fe})_7(\text{PO}_4)_6$	Small $r < v$	$X \sim C$	MCL ps trig	001 very good	Yellow, brown, cols	H 4.5 G 3.43 F 1.5	Diss by acids.
1.672	<u>1.672</u>	1.673	.001	BARATOVITE $\text{KC}_8\text{Li}_2\text{Si}_{12}\text{O}_{37}\text{F}$	60° $r > v$ str	$X:C = 50^\circ$	MCL	001 perf	White	H 3.5 G 2.92	---
1.669	<u>1.673</u>	1.692	.023	GOEDKENITE $(\text{Sr},\text{Ca})_2\text{Al}(\text{PO}_4)_2(\text{OH})$	45-50°	$X = b$	MCL	100 fair	Col's to pale yellow	H 5 G 3.83 calc	---
1.665	<u>1.673</u>	1.682	.017	TRIPOLITE $(\text{Mn},\text{Fe},\text{Mg},\text{Ca})_2\text{PO}_4(\text{F},\text{OH})$	~90°	$Y = \frac{b}{a} = 42^\circ$	MCL	100 perf 010 poor	Salmon-pink	H 3.5 G 3.7 F 2.5	Diss by acids. Pleoc., X pale red-brown, Y cols. MnO 52.4, FeO 5.0, MgO 0.6, CaO 3.2%.
1.652 1.683	<u>1.674</u>	1.695	.039	CHRYSOLITE (Olivine grp) $(\text{Mg},\text{Fe})_2\text{SiO}_4$	89° $r < v$	$X = \frac{b}{c}$ $Y = \frac{c}{a}$	ORTH equant	010, 001 poor	Green	H 7 G 3.33 infus	Gel with acids. FeO 10.8, Fe_2O_3 1.7, MnO 0.2% (Fo_{88}).
1.664 1.680	<u>1.674</u>	1.699	.036	SPODIOSITE $\text{Ca}_2(\text{PO}_4)\text{F}$ (?)	69°	---	ORTH (?)	010 dist 001 indist	Ash-gray to brown	H 5 G 2.94 F diff	Diss by acids.
1.663	<u>1.674</u>	1.699	.036	LAWSONITE $\text{CaAl}_2\text{Si}_2\text{O}_7(\text{OH})_2 \cdot \text{H}_2\text{O}$	82° $r > v$ str	$X = \frac{a}{b}$ $Y = \frac{c}{d}$	ORTH tab	001 010 good	Col's, bluish	H 8 G 3.08 F 4	Insol in acids, but after ignition gel with acids. Pleoc. wk.
1.665	<u>1.674</u>	1.686	.021	NATROPHILITE NaMnPO_4	82° $r > v$	$X = \frac{a}{b}$ $Y = \frac{c}{d}$	ORTH	100 good 010 good	Deep wine-yellow	H 4.5-5 G 3.41 F 2-2.5	Diss by acids.
1.671	<u>1.674</u>	1.684	.013	AKROCHORDITE $\text{Mn}_4\text{Mg}(\text{AsO}_4)_2(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	75° $r < v$ str	$X = \frac{b}{c}$ $Y:C = 45^\circ$	MCL radial aggre-gates	Two	Yellow-brown	H 3.5 G 3.26	Diss by acids.
1.672	<u>1.676</u>	1.683	.011	PHARMACOSIDERITE $\text{KFe}^{+3}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	Med $r < v$	---	CUB	100 imperf	Olive-green to brown	H 2.5 G 2.8-2.9 F 2	Diss by acids. Poly tw divided into segments.
(1.661) (1.700)	<u>1.676</u>	---	wk	PHARMACOSIDERITE $\text{KFe}^{+3}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	Large $r < v$ str	---	MCL pris c	110 perf at 125°	Green to brown	H 7 G 3.47 F fus	Insol in acids. FeO 31.2, Fe_2O_3 3.4%.
1.664	<u>1.677</u>	1.693	.033	CUMMINGTONITE (Amphibole group) $(\text{Fe},\text{Mg})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	Large $r > v$	$Y = b$	---				
1.677 neg											

1.669 1.677 neg	<u>1.677</u>	1.682	.010	BRONZITE (Orthopyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	~ 90°	$Y = \frac{a}{c}$ $Z = \frac{c}{e}$ el pos	ORTH pris c	210 good at 88°	Green	H 5.5 G 3.3 F 5
1.672 1.683	<u>1.678</u>	1.712	.040	MAPIMITE Zn ₂ Fe ⁺³ (AsO ₄) ₄ ·4H ₂ O	50° r < v str	$Y = \frac{b}{c}$ $Z:c = 13°$	MCL	---	Blue to green	H 3 G 2.95
1.672 1.675	<u>1.678</u>	1.705	.032	DIOPSIDE (Pyroxene grp) Ca(Mg,Fe)Si ₂ O ₆	48°	$Y = \frac{b}{c}$ $Z:c = 42°$	MCL	110 good at 87°	Green to brown	H 6 G 3.24
1.654 1.668	<u>1.678</u>	1.688	.013	CLINOHYPERSTHENNE (Pyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	57°	$X = \frac{b}{c}$ $Z:c = 36°$	MCL el c	110 good at 87°	Green	H 5-6 G 3.3-3.4 infus
1.643 1.714	<u>1.679</u>	1.700	.032	CLINOHUMITE (Humite grp) (Mg,Fe) ₉ (SiO ₄) ₄ (F,OH) ₂	Med (73+7°) r < v	$Z = \frac{b}{c}$ $X \sim \frac{c}{e}$	MCL	---	Yellow to red	H 6 G 3.3 infus
1.666 1.689	<u>1.679</u>	1.688	.013	LITHIOPHILITE Li(Mn,Fe)PO ₄	60° r < v str	$X = \frac{c}{e}$ $Y = \frac{a}{c}$ el neg	ORTH el c	100 perf 010 less so	Light brown	H 6 G 3.48 infus
1.674 1.680 neg	<u>1.661</u>	1.697	.036	CHRYSOLITE (Olivine grp) (Mg,Fe) ₂ Si ₁₀ O ₄	~ 90°	$X = \frac{b}{c}$ $Y = \frac{a}{c}$	ORTH	010, 001 poor	Green	H 5 G 3.45 infus
1.662	<u>1.680</u>	1.700	.028	PHOSPHOFERRITE (Fe,Mn) ₃ (PO ₄) ₂ ·3H ₂ O	68° r > v dist	$X = \frac{a}{b}$ $Y = \frac{a}{b}$	ORTH pris c	010 poor fr uneven	Green	H 4 G 3.3 infus
1.678	<u>1.688</u>	1.683	.003	HARSTIGITE Ca ₆ (Mg,Mn)Be ₄ Si ₆ (0,OH) ₂₄	52° r < v wk	$Y = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH pris c	---	Col s	H 5.5 G 3.16
1.670 1.699	<u>1.681</u>	1.695	.022	OMPHACITE (Pyroxene grp) (Ca,Na)(Mg,Fe,Al)Si ₂ O ₆	70° r < v mod	$Y = \frac{b}{c}$ $Z:c = 41°$	MCL pris c	110 good at 87°	Green	H 5-6 G 3.34 F 5
(1.652)	<u>1.680</u>	(1.681)	1.698	NAGELSCHMIDTITE Ca ₃₋₄ (Si,P) ₂ O ₈	0-20°	---	MCL(?)	001 good 110 fair	Col s	G 3.04
1.667 1.688	<u>1.674</u>	1.691	.017	FERRO-GEUDRITE (Amphibole grp) (Fe,Mg) ₅ Al ₂ (Si ₆ Al ₂) ₀ 22 (OH) ₂	80°	$Y = \frac{b}{c}$ $Z = \frac{c}{e}$	ORTH pris c	210 perf at 126°	Brown	H 6 G 3.36 F 6
1.676	(1.681)	1.690	.014	SATPAVITE 6Al ₂ O ₃ ·V ₂ O ₅ ·30H ₂ O	~ 70°	---	ORTH(?) Pina- micro- cryst	Canary- to saffron- yellow	Diss by acids. wk, abs Z > X.	H 1.5 G 2.4

Nearly insol in acids. Faint pleoc common.

Pleoc, X pale yellow,
Y deep blue, Z greenish-yellow.Insol in acids. FeO 8.2, Fe₂O₃ 1.4,
Al₂O₃ 4.5%.Insol in acids. Poly
tw 100. FeO 12.6,
MnO 0.5, Al₂O₃ 0.5%.Gel with acids. FeO 9.9, MnO 0.3,
Fe₂O₃ 1.0, F 1.8%.

Diss by acids. MnO 31.9, FeO 11.0%.

Gel with acids (Fo₈₆).

Diss by acids.

Insol in acids, but
gel after being heated.Insol in acids.
Pleoc wk in green.
FeO 2.7, Fe₂O₃ 1.3,
Na₂O 4.6, CaO 14.7%.

Insol in acids.
Pleoc X, Y yellow-green,
Z blue-green.
FeO 23.0, MnO 1.7,
Al₂O₃ 19.2%.Diss by acids. Pleoc
wk,

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	MINERAL NAME and formula	$2V_z$ (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
a	β	Biref							
		γ							
1.670 \wedge	1.679 $\underline{1.681}$	1.685 .006	JADEITE (Pyroxene grp) (Na,Ca)(Al,Fe)Si ₂ O ₆	$\gamma = \frac{b}{c} = 64^\circ$ $r > v$ mod	MCL prts \underline{c}	110 good at 87°	Green	H 6 G 3.4 F 2	Insol in HCl. FeO 1.1, Fe ₂ O ₃ 5.7, Na ₂ O 13.0, CaO 2.7%.
(1.618) \wedge	1.662 $\underline{1.683}$	1.717 .055	KOETTIGITE Zn ₃ (AsO ₄) ₂ •8H ₂ O	$\gamma = \frac{b}{c} = 74^\circ$ $r < v$ str	MCL fib \underline{c}	010 perf	Carmine-red	H 2.5-3 G 3.33 F 3	Diss by acids. Pale pink in section, not pleoc.
1.678 \wedge 1.685 \wedge	1.676 $\underline{1.683}$	1.705 .029	DIOPSIDE, manganeseoan (Pyroxene grp) Ca(Mg,Mn,Zn)Si ₂ O ₆	$\gamma = \frac{b}{c} = 60^\circ$ $r > v$	MCL prts \underline{c}	110 good at 87°	Brown	H 5.5 G 3.39 F 6	Insol in acids. MnO 7.4, ZnO 3.3, Fe ₂ O ₃ 0.4%.
1.673 \wedge 1.704 \wedge	1.675 $\underline{1.683}$	1.692 .017	TRIPOLITE (Mn,Fe,Mg,Ca) ₂ P ₄ (F,OH)	$\gamma = \frac{b}{c} = 80^\circ$ $r > v$	MCL	100 perf 010 poor	Reddish-brown	H 4.5 G 3.87 F 2.5	Diss by acids. Pleoc in brown, abs X > Y > Z. MnO 34.8, FeO 23.5, CaO 3.5%.
1.698	1.684 $\underline{1.684}$	1.707 .023	PIGEONITE (Pyroxene grp) (Mg,Fe,Ca)(Mg,Fe)Si ₂ O ₆	$\gamma = \frac{a}{c} = 18-26^\circ$	MCL	110 good at 87°	Cols to green	H 5-6 G 3.4 infus	Insol in acids. FeO 20.1, Fe ₂ O ₃ 3.5, CaO 5.5%.
1.65	---	1.72 .07	OTWAYITE Ni ₂ (CO ₃)(OH) ₂ •H ₂ O	---	X = fib	ORTH fib ros-settes	---	Bright green	Diss slowly by cold dil HCl. Pleoc wk, abs Z > X. Opt char unk.
1.646	1.685 $\underline{1.685}$	1.745 .099	ERIOCHALCITE CuCl ₂ •2H ₂ O	$\gamma = \frac{b}{c} = 75^\circ$ (80+2°) $r < v$ str	ORTH	110 perf 001 good	Bluish-green	H 2.5 G 2.47 F easy	Sol in H ₂ O.
1.683 \wedge 1.692 \wedge	1.677 $\underline{1.685}$	1.708 .031	DIOPSIDE (Pyroxene grp) Ca(Mg,Fe)Si ₂ O ₆	$\gamma = \frac{b}{c} = 59^\circ$ $r > v$ wk	MCL prts \underline{c}	110 good at 87°	Cols to green	H 5-6 G 3.28 infus	Insol in acids. FeO 5.6, MnO 0.5, Fe ₂ O ₃ 3.2, Al ₂ O ₃ 2.2%.
1.669 \wedge	1.659	1.785 .126	NENADKEVICHITE (Na,Ca,K)(Nb,Ti)Si ₂ O ₆ (0,OH)•2H ₂ O	$\gamma = \frac{b}{c} = 46^\circ$ (58+2°)	ORTH	001 poor	Rose to brown	H 5 G 2.86 F diff	Diss by H ₂ SO ₄ . Pleoc wk, X col's, Y pale yellow, Z pale rose.
1.683 \wedge	1.687 $\underline{1.687}$	1.718 .035	BORNEMANITE BaNa ₄ Ti ₂ NbSi ₄ O ₁₇ (F,OH) Na ₃ PO ₄	$\gamma = \frac{c}{a}$ el pos	ORTH plates	001 perf	Pale yellow	H 3.5-4.5 G 3.48	Dec by acids with sep of silica. Pleoc wk, X and Y col's, Z brownish.
1.695 \wedge									

1.678	<u>1.687</u>	1.705	.027	ROSENBUSCHITE (Ca, Na) ₃ (Zr, Ti)Si ₂ O ₈ F	78°	X ~ c el pos	TCL	010 perf	Yellow, brown	H 5-6 G 3.35 F easy			
1.684	<u>1.688</u>	1.705	.021	PENKISITE Ba(Mg, Fe) ₂ Al ₂ (PO ₄) ₃ (OH) ₃	56° r > v str	X ~ b Z:c = -6°	TCL	010, 100 fair to good	Blue to green	H 4 G 3.79	Pleoc. X grass-green, Y blue-green, Z pale pink, abs X ~ Y > Z.		
1.680	<u>1.688</u>	1.700	.020	FERRO-GEHRITE (Amphibole grp) (Fe, Mg) 5Al ₂ (Si ₆ Al ₂)O ₂₂ (OH) ₂	83°	Y = b Z:c = c el pos	0RTH pris c	210 perf at 126°	Brown	H 6 G 3.37 F 6	Insol in acids. Pleoc. X light brown, Z pale blue-green. FeO 26.2, Fe ₂ O ₃ 3.6, Al 2.3 17.7%.		
1.686	<u>1.688</u>	1.699	.013	PUMPELLITE Ca ₂ (Mg, Fe)Al ₂ (Si ₄ O ₄) (Si ₂ O ₇)(OH) ₂ •H ₂ O	50° r < v str	Z:c = 32°	MCL	001 good 100 imperf	Bluish-green	H 5.5-7 G 3.15	Insol in acids. Pleoc. X and Z pale yellow-green, Y green. FeO 3.0, Fe ₂ O ₃ 1.7%.		
1.688	<u>1.689</u>	1.695	.007	TRIPHYLITE Li(Fe, Mn)PO ₄	Small r < v str	Z = b	0RTH	100 perf 0.0 less so	Grayish-green	H 5 G 3.33 F 2	Diss by acids. FeO 29.6, MnO 14.6, MgO 1.1%.		
1.675	<u>1.690</u>	1.735	.060	LEGENDITE Zn ₂ (AsO ₄) ₂ (OH)•H ₂ O	36° (61+4°) r < v dist	X = b Z:c = -38°	MCL pris el b	100 fair	Yellow, cols	H 5 G 3.98	Diss by acids. Pleoc. faint in yellow.		
1.682	<u>1.690</u>	1.698	.016	CHLOROPHOENICITE (Mn, Mg) 3Zn ₂ (AsO ₄) ₆ (OH, O) ₆	83° r > v str	Y = b	MCL el b	100 good	Gray-green	H 3.5 G 3.46 F diff	Diss by acids. Color pink to purplish-red in str artificial light.		
1.663	<u>1.691</u>	1.716	.053	ASTROPHYLITE (K, Na) ₃ (Fe, Mn) ₇ (Ti, Nb) ₂ Si ₈ O ₂₄ (OH) ₇	Large r > v str	X ~ b Z:c = 3- 19°	TCL	001 perf	Yellow, brown	H 3 G 3.2 F 3	Dec by acids. Pleoc in yellow, abs X > Y > Z.		
1.686	<u>1.691</u>	1.708	.022	MARSTURITE Na ₂ Ca ₂ Mn ₆ S ₁₀ O ₂₈ (OH) ₂	60°	---	TCL pris	110, 001 imperf	White to pale pink	H 6 G 3.46	MnO 34%.		
1.688	<u>1.691</u>	1.696	.008	FERRONYLLITE (Na, Ca, Mn)(Fe ⁺² , Mn) (Fe ⁺² , Fe ⁺³ , Mg)Al(Po ₄) ₃	50° r < v str	---	MCL	010 perf T01 dist	Bluish-green	H 4	Pleoc. X smoky blue- grey, Y bluish-green, Z green.		
1.687	<u>1.692</u>	1.715	.028	DIOPSIDE (Pyroxene grp) Ca(Mg, Fe)(Si, Al) ₂ O ₆	51°	Y = b Z:c = 42°	MCL	110 good at 87°	Green, brown	H 6 G 3.33	Insol in acids. FeO 9.5, MnO 0.2, Fe ₂ O ₃ 1.6, Al ₂ O ₃ 2.3%.		
1.691	<u>1.692</u>	1.713	.022	YODERITE (Mg, Al) ₈ Si ₄ (O, OH) ₂₀	25-30°	X:a = 9°	MCL	Parting 001	Purple, emerald green	H 6 G 3.39	Pleoc. X pale blue, Y indigo, Z light olive-green. Y and yellow, X green.		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index α	β	γ	Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orienta- tion $X = \frac{c}{b}$ $Y = \frac{c}{a}$	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.703	1.691	1.692	1.700	.009	ZOISITE (Epidote grp) $(\text{Ca}_2\text{Al}_3\text{Si}_3\text{O}_{12})(\text{OH})$	54° r > v str	$X = \frac{c}{b}$ $Y = \frac{c}{a}$	0RTH	010, 100 good but diff, fr conch	Violet- blue to cols	H 6 G 3.5 F 3.5	Insol in acids, but gel after being heated. Abn interf colors. Pleoc u wk, X violet, Y deep blue, Z yellow-green.
1.687	(1.692)	1.698	.011	TINZENITE (Axinite grp) $(\text{Ca}, \text{Mn}, \text{Fe})_3\text{Al}_2\text{BSi}_4\text{O}_{15}$ (OH)	75-85° r < v	----	----	TCL	100 good	Yellow	H 7 G 3.3 F 2	Insol in acids. Anom disp, deep blue, red- brown. Ca0 17.0, Mn0 14.8, Fe2O3 2.8, MgO 0.7%.
1.672	1.693	1.710	.038	JAGOWERITE $\text{BaAl}_2(\text{PO}_4)_2(\text{OH})_2$	80° (97+6°)	----	----	TCL	100, 0T1 good 021 fair	Light green	H 4.5 G 4.01	Insol in acids. Fluor greenish- white in short-wave UV.
1.686	1.693	1.714	.028	FASSAITE (Pyroxene grp) $\text{Ca}(\text{Mg}, \text{Fe}^{+3}, \text{Al})$ $(\text{Si}, \text{Al})_{2-6}$	56° r > v mod	$Y = \frac{b}{c} = \frac{43}{43}$	----	MCL pris	110 good at 87°	Green	H 6 G 3.33	Insol in acids. Fe0 3.1, Fe2O3 2.1, Al2O3 13.4, TiO2 0.5%.
1.719	1.691	1.719	.028	MANGAN-NEPTUNITE $\text{KNa}_2\text{Li}(\text{Mn}, \text{Fe})_2\text{Ti}_2\text{Si}_8$ 0.24	36° r < v str	$Y = \frac{b}{c} = \frac{16}{20}$ $Z: \frac{c}{a} = 20^{\circ}$ el pos	----	MCL pris C	110 perf at 80°	Black to deep red	H 5-6 G 3.20 F 2.5	Insol in acids. Pleoc str, X pale yellow, Y brownish, Z orange-red, abs Z > Y > X.
1.700	1.691	1.719	.018	BJAREBYITE $(\text{Ba}, \text{Sr})(\text{Mn}, \text{Fe}, \text{Mg})_2\text{Al}_2$ $(\text{PO}_4)_3(\text{OH})_3$	~35° r > v str	----	----	MCL spears	010, 100 perf	Emerald- green	H 4+ G 3.95	Pleoc wk, gray-tan to pale yellow-green. Mn0 9.6, Fe0 8.7%.
1.727	1.692	1.710	.018	BORNEMANITE $\text{BaNa}_4\text{Ti}_2\text{NbSi}_4\text{O}_{17}(\text{F}, \text{OH})$ $\cdot \text{Na}_3\text{PO}_4$	66°	$X = \frac{c}{b}$ $Z = \frac{c}{a}$	----	0RTH platy	001 perf	Pale yellow	H 3.5-4.5 G 3.48	Dec by acids with sepn of silica. Pleoc wk, X and Y cols, Z brownish.
1.687	1.682	1.695	1.720	BEUSITE $(\text{Mn}, \text{Fe}, \text{Ca}, \text{Mg})_3(\text{PO}_4)_2$	25° r > v str	$X = \frac{b}{c} = -36^{\circ}$ $Z: \frac{c}{a} = -36^{\circ}$	----	MCL	010 good 100 fair	Salmon- pink to reddish- brown	H 5 G 3.7 F 2	Diss by acids. Mn0 24.8, Fe0 16.8, Ca0 12.2, Mg0 2.4%.
1.711	1.695	1.715	.020	BARYLITE $\text{BaBe}_2\text{Si}_2\text{O}_7$	60-81° r > v wk	$X = \frac{b}{c}$ $Y = \frac{b}{a}$	----	0RTH	100 perf 210, 010	Col s, pink	H 6-7 G 4.02 infus	Insol in acids.
1.692	1.691	1.703	.012	----	----	----	----	----	----	----	----	----
1.702	1.691	1.703	.012	----	----	----	----	----	----	----	----	----

1.691	<u>1.696</u>	1.725	.034	BRABANTITE CaTh(PO ₄) ₂	44°								
1.637 ^	1.693 (1.697)	1.718	.025	BARITE (plumboglan) (Ba,Pb)SO ₄	~ 50° r < v	X = c Y = b Z:c = 63°	ORTH so	001 perf 210 less	White	H 3 G ~ 5 F 2.5	Insol in acids.		Diss by HC1. Compare Cheralite, Monazite.
1.883 ^		1.825	.138	LABUNTSEVITE (K,Ba,Na)(Ti,Nb) (Si,Al) ₂ (O,OH) ₇ H ₂ O	20° (35+3°) r > v	MCL	T02 perf	Dark brown	H 5 G 2.87	Pleoc wk, X and Y cols, Z pale yellow.			
1.670 ^	<u>1.698</u>	1.733	.038	EUCHROITE Cu ₂ (AsO ₄)(OH)·3H ₂ O	29° r > v mod	X = c Z = b Z:c = 41°	ORTH pris c	110, 101	Emerald- to leek-green	H 3.5-4 G 3.44 F 2-2.5	Diss by acids. Bright bluish-green in section, pleoc faint or absent.		
1.695	<u>1.698</u>	1.721	.025	PIGEONITE (Pyroxene grp) (Mg,Fe,Ca)(Mg,Fe)Si ₂ O ₆	25°	MCL	110 good at 87°	Cols to green	H 5-6 G 3.38 infus	Insol in acids. Pleoc wk. FeO 21.4, Fe ₂ O 0.8, MnO 0.4, CaO 3.5%.			
1.684 ^	1.696 1.714	1.721	.025	DIOPSIDE, manganese (Pyroxene grp) Ca(Mg,Mn)Si ₂ O ₆	60° r > v wk	Y = b Z:c = 43°	MCL pris c	110 good at 87°	Brown	H 6 G 3.3 infus	Insol in acids. MnO 10.0, FeO 1.65, Fe ₂ O 1.5%.		
1.692 ^	1.690 1.707	1.721	.031	OMPHACITE (Pyroxene grp) (Ca,Na)(Mg,Fe,Al)Si ₂ O ₆	64° (73+8°) r < v mod	Y = b Z:c = 41°	MCL pris c	110 good at 87°	Green	H 6 G 3.35 F 5	Insol in acids. Pleoc wk in green. FeO 5.9, Fe ₂ O 3.8, Al ₂ O ₃ 6.1, CaO 20.6, Na ₂ O 1.7%.		
1.681 ^	1.689 1.699	1.718	.029	GRAFTONITE (Fe,Mn,Ca) ₃ (PO ₄) ₂	50° r > v str	X = b Z:c = -36°	MCL u mass	010 good 100 fair	Salmon- pink to red-brown	H 5 G 3.7 F 2	Diss by acids. FeO 28.8, MnO 16.0, CaO 12.8%.		
				PENNAITE Silicate of Na, Ca, Fe, Ti, Zr	25°	Y:c = 13°	TCL(?) acic	---	Yellow to light brown	---	Pleoc, X dark yellow, Y grayish-yellow, Z bright-yellow.		
				MANGAN-NEPTUNITE KNa ₂ Li(Mn,Fe) ₂ Ti ₂ Si ₈ O ₂₄	36° r < v str	Y = b Z:c = 16- 20° el pos	MCL pris c	110 perf at 80°	Red to black	H 5.5-6 G 3.2 F 2.5	Insol in acids. Pleoc str, X yellow, Y orange-red, Z deep red.		
				GADOLINITE Y ₂ Fe ₂ Si ₂ O ₁₀	(52+11°)	X = b Z:c = 4- 13°	MCL	Conch	Black to dark brown	H 7 G 4.2 infus	Gel in part. Pale green, not pleoc in section.		
				PHARMACOSIDERITE KFe ⁺³ ₄ (AsO ₄) ₃ (OH) ₄ · 6-7H ₂ O	Large r < v str	---	CUB	100 imperf	Olive- green to brown	H 2.5 G 2.8-2.9 F 2	Diss by acids. Poly tw. Abnorm interf colors.		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ calc. ($2V_z$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, fusibility	Remarks
	α	β	γ									
1.692 χ	1.700	1.703	1.708	.008	ZOISITE (Epidote grp) $\text{Ca}_2\text{Al}_3(\text{SiO}_4)_3(\text{OH})$	71° $r < v$ or $r > v$	$X = \frac{c}{b}$ $Y = \frac{c}{b}$	ORTH	010 perf but diff 100	Col s., green	H 6-7 G 3.35 F 3.5	Insol in acids, but gel after ignition. Abnorm interf colors. Fe_2O_3 3.1, FeO 1.3, MnO 0.8%.
1.717 χ	1.700	1.703	1.706	.006	CLINOZOISITE (Epidote grp) $\text{Ca}_2\text{Al}_3(\text{SiO}_4)_3(\text{OH})$	~90° $r < v$ str	$Y = \frac{b}{c}$ $X \sim \frac{c}{b}$	MCL	001 perf	Col s.	H 6.5 G 3.28 F 3	Insol in acids. Abnorm interf colors. Fe_2O_3 3.2, FeO 0.6%.
(1.741) χ	1.701	1.703	1.706	.005	SERENDIBITE $\text{Ca}_2(\text{Mg},\text{Al})_6(\text{Si},\text{Al},\text{B})_6$ O_{20}	~90° $r < v$ str	ext:tw lam ~ 40°	TCL platy	---	Stay-blue to indigo-blue	H 6.5 G 3.42 infus	Insol in acids. Poly tw. Pleoc, X pale green, Y nearly cols, Z deep blue.
1.728 χ	1.694	1.704	1.719	.025	ROSELINEITE $\text{Ca}_2(\text{Co},\text{Mg})(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	75° $r < v$ perc	$Y = \frac{b}{c}$ $X:C = 0-4^\circ$	MCL	010 perf	Pale rose	H 3.5 G 3.69 F 3	Diss by acids. Pleoc wk in rose, abs X > Y > Z.
1.698	1.704	1.720	1.720	.022	AKATOREITE $\text{Mn}_9(\text{Si},\text{Al})_{10}O_{23}(\text{OH})_9$	65.5°	---	TCL pris	010 perf 012 imperf	Orange-brown	H 6 G 3.48	Pleoc, X cols, Y pale yellow, Z light canary-yellow.
1.683 χ	1.696	1.704	1.713	.017	ZWIESELITE (Triplite ser) $(\text{Fe},\text{Mn},\text{Mg},\text{Ca})_2\text{PO}_4(\text{OH},\text{F})$	87° $r > v$	$Y = \frac{b}{c}$ $X:A = 42^\circ$	MCL	100 perf 010 poor	Dark brown	H 4.5 G 3.97 F 2.5	Diss by acids. Pleoc in brown, abs X > Y > Z. FeO 42.0, MnO 18.4, MgO 0.8, CaO 1.7%.
(1.727) χ	1.703	1.704	1.710	.007	SAPPHIRINE $(\text{Mg},\text{Al})_8(\text{Al},\text{Si})_6O_{20}$	Med	$Y = \frac{b}{c}$	MCL	010, 001, 100 poor	Pale blue	H 7.5 G 3.44 infus	Insol in acids. Pleoc in blue, abs Z > Y > X.
1.703 χ	1.705	1.723	.020	KULANITE $\text{Ba}(\text{Fe},\text{Mn},\text{Mg})_2\text{Al}_2(\text{PO}_4)_3$ $(\text{OH})_3$	32° $r > v$ str	$Y = \frac{b}{c}$ $Z:c = -8^\circ$	TCL plates	010 100 fair to good	Blue to green	H 4 G 3.91	Pleoc, X brownish-green, Y green, Z pale brown.	
1.688 χ	1.698	1.705	1.715	.017	PUMPELLYITE $\text{Ca}_2(\text{Mg},\text{Fe})\text{Al}_2(\text{SiO}_4)$ $(\text{Si}_{27})(\text{OH})_2 \cdot \text{H}_2\text{O}$	68° $r < v$ str rarely r > v	$Y = \frac{b}{c}$ $Z:c = 5-$ -22°	MCL	001 good 100 imperf	Bluish-green	H 5.5-7 G 3.2	Insol in acids. Pleoc, X and Z yellow-green, Y green. Fe_2O_3 3.6, FeO 2.9%.
1.699 χ	1.700	1.707	1.724	.024	SALITE (Diopside ser, Pyroxene grp) $\text{Ca}(\text{Mg},\text{Fe})(\text{Si},\text{Al})_2O_6$	52° $(66+10^\circ)$	$Y = \frac{b}{c}$ $Z:c = 44^\circ$	MCL pris	110 good at 87°	Green, brown	H 6 G 3.42	Insol in acids. FeO 9.8, MnO 0.3, Fe_2O_3 1.8, Al_{23} 2.8%.
1.714 χ												

				Zn:Mg = 1:1.
1.689	<u>1.707</u>	.038	GAITITE $\text{Ca}_2(\text{Zn},\text{Mg})(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	
1.703	<u>1.708</u>	.026	OXYCHILDRENITE $(\text{Mn}^{+2},\text{Mg})(\text{Fe}^{+3},\text{Mn}^{+3})_4$ $\text{Al}_4(\text{PO}_4)_4(\text{OH})_{14}$	
1.704	<u>1.708</u>	.010	ALLANITE (Epidote grp) $(\text{Ca},\text{Ce})_2(\text{Al},\text{Fe})_3(\text{SiO}_4)_3$ (OH)	
1.702	<u>1.709</u>	.038	LEGENDRITE $\text{Zn}_2(\text{AsO}_4)(\text{OH}) \cdot \text{H}_2\text{O}$	
1.708	<u>1.710</u>	.038	STRANGITE, manganese $(\text{Fe}^{+3},\text{Mn}^{+3})_3(\text{PO}_4) \cdot 2\text{H}_2\text{O}$	
1.700	<u>1.710</u>	.033	JOHANNSENITE (Pyroxene grp) $\text{Ca}(\text{Mn},\text{Mg},\text{Fe})\text{Si}_2\text{O}_6$	
1.707	<u>1.710</u>	.023	NAMBULITE $\text{NaLi}(\text{Mn},\text{Mg})_8\text{Si}_{10}\text{O}_{28}$ (OH) ₂	
1.702	<u>1.710</u>	.021	MERWINITE $\text{Ca}_3\text{Mg}(\text{SiO}_4)_2$	
1.711	<u>1.711</u>	.015	BEUSITE $(\text{Mn},\text{Fe},\text{Ca})_3(\text{PO}_4)_2$	
1.708	<u>1.711</u>	.015	BRANDTITE $\text{Ca}_2\text{Mn}(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	
1.709	<u>1.711</u>	.015	MANGANUMITE (Humite grp) $(\text{Mn},\text{Mg})_7(\text{SiO}_4)_3(\text{OH})_2$	
1.707	<u>1.712</u>	.025	OTTRELITE $(\text{Mn},\text{Fe},\text{Mg})_2\text{Al}_4\text{Si}_2\text{O}_{10}$ (OH) ₄	
1.709	<u>1.712</u>	.007		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index α	β	Biref γ	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation $X = \frac{a}{b}$ $Y = \frac{c}{d}$	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.703	1.713	1.722	.019	GERHARDITE $Cu_2(No_3)(OH)_3$	Large $r < v$, very str	---	ORTH horizontal vicinal striations	001 perf 110 good	Emerald-green	H 2 G 3.42 F 2	Diss by acids. Pleoc., X and Y green, Z blue.
(1.731)	1.708	(1.713)	.014	VARULITE (Na,Ca)Mn(Mn,Fe ⁺² , Fe ⁺³) ₂ (PO ₄) ₃	Large $r > v$	---	MCL	001, 010	Olive-green to brown	H 5 G 3.5-3.6 Fus	Diss by acids. Pleoc., X yellow-green, Z grass-green.
1.707 1.732	1.708	1.714	.028	FERROSALITE (Diopsid de ser., Pyroxene grp) Ca(Fe,Mg)Si ₂ 0 ₆	70° (56+10°) $r > v$ wk	$Y = \frac{b}{c} = 43^\circ$ $Z:c = 43^\circ$	MCL pris c	110 good at 87°	Dark green	H 6 G 3.41 F 4	Insol in acids. FeO 18.6, MnO 0.2, Fe ₂ O ₃ 0.5, Al ₂ O ₃ 1.4%.
1.698	1.714	1.714	.028	PIGEONITE (Pyroxene grp) (Mg,Fe,Ca)(Mg,Fe)Si ₂ 0 ₆	0-12°	$X = \frac{a}{c} = 41^\circ$ $Z:c = 41^\circ$	MCL	110 good at 87°	Green	H 5-6 G 3.44 F 5-6	Insol in acids. FeO 27.8, MnO 1.0, CaO 3.8, Fe ₂ O ₃ 1.7%.
1.679	1.707	1.714	.027	CLINOHUMITE (Humite grp) (Mg,Fe) ₉ (SiO ₄) ₄ (OH,F) ₂	56° $r < v$	$Z = \frac{b}{c} \sim \frac{c}{a}$	MCL	---	Yellow to red	H 6 G 3.36 Infus	Gel with acids. FeO 10.2, Fe ₂ O ₃ 1.2, TiO ₂ 5.1, F 0.1%.
1.728	1.712	1.714	.014	CHLORITOID (Fe,Mg,Mn) ₂ Al ₄ Si ₂ 10 (OH) ₄	51° $r > v$ str	$X = \frac{b}{c} = 11^\circ$ $e \parallel neg$	MCL and TCL	001 perf	Greenish-gray	H 6.5 G 3.54 F diff	Dec by H ₂ SO ₄ . Pleo X green, Y slate blue, Z yellowish FeO 21.2, Fe ₂ O ₃ 1.6, MnO 0.4%.
v 1.722	1.700	1.715	1.740	DIASPORE A10(OH)	84° $r < v$ wk	$Y = \frac{b}{c} = \frac{a}{d}$	ORTH blades 010	010 perf 110 less so	Col s	H 6.5-7 G 3.44 Infus	Insol in acids. Luster pearly. Fe ₂ O ₃ 0.1, TiO ₂ 0.2%.
v 1.725	1.700	1.715	1.732	EVEITE $Mn_2(AsO_4)(OH)$	65° (87+7°) $r < v$ mod	---	ORTH tab 100	011 fair	Apple-green	H 4 G 3.67	Diss by acids.
v 1.711	1.707	1.715	1.730	LARNITE Ca_2SiO_4	Mod large	$Z = \frac{b}{c} = 14^\circ$	MCL	100	White	---	Gel with acids. Poly tw on 100.
v 1.725	1.711	1.715	1.724	RHODONITE (Mn,Ca,Mg,Fe)SiO ₃	69° $Z:c$ on $010 = 9^\circ$ $110 = 30^\circ$	TCL	100, 001 perf 010 less so	Pink	H 6 G 3.56 F 3	Slightly attached by acids. MnO 40.2, CaO 6.7, MgO 4.6%.	

1.707 1.723	<u>1.716</u>	1.736	.026	FERROAUGITE (Pyroxene grp) (Ca, Na)(Fe, Mg, Al, Ti) (Si, Al) ₂ 6	52°	$\frac{Y = b}{Z:c} = 49^\circ$	MCL	110 good at 87°	Green, brown	H 6 G 3.49
1.712	<u>1.716</u>	1.725	.013	BREDIGITE $\text{Ca}_7\text{Mg}(\text{SiO}_4)_4$	10-34° (68+18°)	---	ORTH	130	White	G 3.38
^v 1.707	<u>(1.717)</u>	1.778	.071	JOAQUINITE $\text{Ba}_2\text{NaCe}_2\text{Fe}(\text{Ti}, \text{Nb})_2\text{Si}_8$ $0_{26}(\text{OH}, \text{F}) \cdot \text{H}_2\text{O}$	45° $r < v$	$Z = \frac{c}{a}$ $X = \frac{a}{d}$	MCL ps orth	---	Brown	H 5.5 G 3.62 F 2.5
1.709	<u>1.717</u>	1.729	.020	STRINGHAMITE $\text{CaCuSi}_4 \cdot 2\text{H}_2\text{O}$	80°	$X = \frac{b}{c}$ $Y:c = 2.5^\circ$	MCL	---	Deep azure- blue	G 3.67 calc Pleoc, X light gray- blue, Y light blue, Z dark blue.
^v 1.705	<u>1.711</u>	1.727	.016	PUMPELLYITE $\text{Ca}_2(\text{Mg}, \text{Fe})\text{Al}_2(\text{Si}_4\text{O}_4)$ (Si ₂ O ₇) ₂ •H ₂ O	Large $r < v$ str	$Y = \frac{b}{c}$ $Z:c = 25^\circ$	MCL	001 good 100 imperf	Bluish- green	H 5.5-7 G 3.25
1.715	<u>1.717</u>	1.728	.013	KANOITE (Pyroxene grp) (Mn, Mg) ₂ Si ₂ O ₆	41°	$Y = \frac{b}{c}$ $Z:c = 42^\circ$	MCL	110 perf	Light pinkish- brown	H 6 G 3.66
1.713	<u>1.717</u>	1.723	.010	JOHACHIDOLITE CaAlB_{37}	75-80° $r > v$ str	$X = \frac{a}{c}$ $Y = \frac{a}{c}$	ORTH mass	---	White	H 7.5 G 3.44 fus
1.703 1.720 neg 1.733	<u>1.715</u>	1.721	.006	CLINOZOISITE (Epidoite grp) $\text{Ca}_2\text{Al}_3\text{Si}_3\text{O}_{12}(\text{OH})$	~90° $r < v$ str	$Y = \frac{b}{c}$ $X \sim \frac{c}{b}$ el c _{IV} pos	MCL	001 perf	Cols to green	H 6.5 G 3.21
1.710	<u>1.718</u>	1.780	.070	STRONTIOTIOQUINITE $\text{Sr}_2\text{Ba}_2(\text{Na}, \text{Fe}^{+2})_2\text{Ti}_2\text{Si}_8$ $0_{24}(\text{OH}, \text{H}_2\text{O})_2 \cdot \text{H}_2\text{O}$	35-45° $r > v$ str	---	MCL	001 good	Green to yellow- green	H 5.5 G 3.68
^v 1.660	<u>1.697</u>	1.741	.044	PLANCHEITE $\text{Cu}_8\text{Si}_8\text{O}_{22}(\text{OH})_4 \cdot \text{H}_2\text{O}$	~90°	$X = \frac{c}{b}$ $Y = \frac{b}{d}$	ORTH fib	---	Blue	H 5 G 3.65-3.80
1.709	<u>1.718</u>	1.734	.025	METAHOMANNITE $\text{Fe}_2(\text{SO}_4)_2(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	(94+9°)	---	Mass, powdery	---	Orange	---
1.713	<u>1.718</u>	1.728	.015	WICKSITE $\text{NaCa}_2(\text{Fe}^{+2}, \text{Mn})_4\text{MgFe}^{+3}$ (PO ₄) ₆ •2H ₂ O	(71+16°)	---	ORTH	010 good	Dark blue	H 4.5-5 G 3.54

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ	Biref								
1.710 $\hat{\vee}$ 1.728	1.710	1.719	1.738	.028	JOHANNSENITE (Pyroxene grp) CaMnSi ₂ O ₆	70° $Y = \frac{b}{c} = 48^\circ$ $Z:c = 48^\circ$	MCL pris \underline{c}	110 good at 87°	Brown to green	H 6 G 3.45 F 5	Insol in acids. MnO 26.8, FeO 1.0, MgO 1.0, Fe ₂ O ₃ 0.3%.	
1.693 $\hat{\vee}$ 1.736	1.712	1.719	1.736	.024	FASSAITE (Pyroxene grp) Ca(Mg,Fe,Al)(Al,Si) ₂ O ₆	55° $(66+10^\circ)$ $r > v$	MCL pris \underline{c} $Y = \frac{b}{c} = 46^\circ$ $Z:c = 46^\circ$	110 good at 87°	Green	H 6 G 3.34	Insol in acids. FeO 0.2, Fe ₂ O ₃ 6.1, Al ₂ O ₃ 15.75, TiO ₂ 0.8%.	
1.710 \wedge	1.714	1.720	1.730	.016	MERMINITE Ca ₃ Mg(SiO ₄) ₂	68° $r > v$ wk	MCL pris \underline{c} $Z:c = 36^\circ$ $X:c = 16-20^\circ$ el pos	010 poor	Col s	H 6 G 3.32 F 6	Gel with acids. Poly tw very common, 2 sets with angle 43° between the sets.	
1.700 \wedge	1.711	1.721	1.744	.033	NEPTUNITE KNa ₂ Li(Fe,Mn) ₂ Ti ₂ Si ₈ 0.24	62° $r < v$ str	MCL pris \underline{c} $Y = \frac{b}{c} = 16-$ 20° el pos	110 perf	Red	H 5.5-6 G 3.2 F 2.5	Insol in acids. Pleoc str, X gold- orange, Y orange-red, Z deep red.	
1.707	1.721	1.739	1.739	.032	LEUCOPHOSPHITE KFe ₂ (PO ₄) ₂ (OH)-2H ₂ O	84° $r < v$ str	X = $\frac{b}{c} = 26^\circ$ $Z:c = 26^\circ$	MCL	001 perf	Buff	6 2.95	Diss by acids. Pleoc, cols to greenish-yellow.
1.713	1.721	1.734	.021		HARADAITE Sr _v +4Si ₂ O ₇	105° $(77+12^\circ)$ $r < v$ str	ORTH $X = \frac{a}{b}$ $y = \frac{a}{b}$	010 perf 100, 001 dist	Bright green	H 4.5 G 3.80	Pleoc, X cols to pale green, Y cols to light yellow-green, Z bluish-green. (Re- ported as opt neg.)	
1.712	1.721	1.731	.019		ADELITE CaMg(AsO ₄)(OH,F)	$\sim 90^\circ$ $r < v$ $perc$	ORTH u mass	Uneven to conch	Col s, gray	H 5 G 3.73 F easy	Diss by acids.	
1.691 $\hat{\vee}$ 1.746	(1.694)	1.722	.063)		ASTROPHYLLITE (K,Na) ₃ (Fe,Mn) ₇ Ti ₂ Si ₈ 0.24(0,0H) ₇	85° $r > v$ str	X ~ $\frac{b}{c}$ $Z:c = 3-$ el clv neg	TCL	001 perf	Red-brown	H 3 G 3.3 F 3	Dec by acids. Pleoc, X red-orange, Y yellow, Z lemon- yellow, abs X > Y > Z.
1.715 \wedge	1.702	1.722	1.750	.048	DIASPORE Al ₁₀ (OH)	82° $r < v$ wk	$Y = \frac{b}{c}$ $Z = \frac{a}{c}$	010 perf blades 010, el \underline{c}	Col s	H 6.5-7 G 3.44 infus	Insol in acids. Luster pearly.	

1.671 1.730	(1.71) 1.722	1.74 (.03)	ANDALLUSITE, manganeseoan (Al,Mn,Fe) ₂ SiO ₅	80° r < v	$X = \frac{a}{b}$ Y = $\frac{d}{e}$ el pos	ORTH pris	110 good	Green	H 7 G 3.3 infus
1.70 1.774	1.715 (1.722)	1.740 0.025	GADOLINITE Y ₂ FeBe ₂ Si ₂ O ₁₀	65°	$X = \frac{b}{c}$ $Z:c = 4-$ 13°	MCL	Conch	Greenish-black	H 6.5 G 4.2 infus
1.716 1.730	1.714 1.723	1.774 .060	AUGITE, zincian (Pyroxene grp) (Ca, Na)(Mg, Mn, Zn, Fe) Si ₂ O ₆	74° (47±5°)	$Y = \frac{b}{c}$ $Z:c = 55°$	MCL pris	110 good at 87°	Greenish-brown	H 6 G 3.55 F 5
1.699 1.712	1.718 1.723	1.745 .027	GRAFTONITE (Fe, Mn, Ca) ₃ (PO ₄) ₂	52° r > v str	$X = \frac{b}{c}$ $Z:c = -36°$	MCL u mass	010 good 100 fair	Reddish-brown	H 5 G 3.75 F 2
			KRINOVITE NaMg ₂ CrSi ₃ O ₁₀	61°	$X = \frac{b}{c}$ disp str	TCL	---	Deep emerald-green	H 5.5-7 G 3.38
			HOMILITE Ca ₂ (Fe, Mg)B ₂ Si ₂ O ₁₀	80° r > v rather str	$Z = \frac{b}{c}$ Y ~ $\frac{c}{c}$	MCL tab 001	Black to dark brown	H 5 G 3.36 F 2	
1.65 1.722	1.715 1.725	1.738 .023	RHODONITE (Mn, Ca, Mg, Fe)SiO ₃	70°	---	TCL	100, 001 perf 010 less so	Pink	H 6 G 3.54 F 3
1.715 1.728	1.722 (1.708)	1.725 1.758	CESTUM KUPLETSKITE (Cs, K, Na) ₃ (Mn, Fe ⁺²) ₇ (Ti, Nb) ₂ Si ₈ O ₂₄ (O, OH, F) ₇	75°	$Z = \frac{a}{b}$ Y:b = 10°	TCL	001 perf	Gold-brown	H 4 G 3.68
1.725 1.749	1.726 1.749	1.730 .005	TRIPODITE (Mn, Fe) ₂ (PO ₄)(OH)	Med r > v extr	$X = \frac{b}{c}$ $Z:c = 4°$ disp dist	MCL pris	010 good 120 fair	Yellowish-to reddish-brown	H 4.5-5.5 G 3.70 F 1.5
1.715 1.730	1.727 1.730	1.749 .034	BABINGTONITE Ca ₂ (Fe ⁺² , Mn)Fe ⁺³ Si ₅ O ₁₄ (OH)	76° r > v str	ext: c on 100 = 44° 010 = 31°	TCL pris c	001 perf 010, 100 less so	Greenish-brownish-black	H 5.5-6 G 3.34 F 3

Insol in acids.
Pleoc, X yellow-green, Y deep green,
Z golden. Mn 0.0%
21.4, Fe₂O₃ 4.5%.

Gel in part with
acids.

Insol in acids. MgO
5.9, ZnO 7.7, MnO
7.2, Fe₂O₃ 4.2%.

Diss by acids. FeO
34.3, MnO 20.1, CaO
5.1%.

Pleoc str, X yellow-green, Y blue-green,
Z greenish-black to dark reddish-brown.
Meteorite mineral.

Gel with acids.
Pleoc, X bluish-green, Y brownish-red,
Z deep smoky gray to brownish-yellow.

Nearly insol in
acids. MnO 41.4,
CaO 6.6, MgO 2.3,
FeO 1.9%.

Pleoc, X yellow-green, Y yellow to
brown, Z brown.

Diss by acids. Cols
in section. Mn:Fe
about 3:1.

Insol in acids.
Pleoc str, X deep green, Y lilac-brown,
Z pale to deep brown.
Fe₂O₃ 15.6, FeO 7.8,
MnO 1.8%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	α	β	γ	Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$) calc disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.695	1.724	<u>1.727</u>	1.749	.025	B-JAREBYITE (Ba, Sr) (Mn, Fe, Mg) ₂ Al ₂ (PO ₄) ₃ (OH) ₃	38° r > v str	Z:c = 8° el clv pos	MCL spears	010, 100 perf	Emerald-green	H 4+ G 3.90	Pleoc, wk, X apple-to olive-green, Y brownish-green, Z col to pale yellow. MnO 16.0, FeO 5.9, Fe ₂ O ₃ 3.3%.	
1.704	1.725	<u>(1.727)</u>	1.732	.007	SAPPHİRINE (Mg, Al) ₈ (Al, Si) ₆ O ₂₀	66°	Y = b	MCL	010, 001, 100 poor	Yellow	H 7.5 G 3.4 infus	Insol in acids. Pleoc, X pale yellow, Y green, Z pale orange.	
1.704	1.725	<u>1.728</u>	1.755	.030	ROSELITE Ca ₂ (Co, Mg)(AsO ₄) ₂ ·2H ₂ O	60° (37+11°) r < v	Y = b X:c = 0-4° Z:c = 48°	MCL pris c	010 perf	Dark rose	H 3.5 G 3.7 F 3	Diss by acids. Pleoc, X dark rose, Y pale rose, Z paler rose.	
1.719	1.716	<u>1.728</u>	1.745	.029	JOHANNSENITE, ferroan (Pyroxene grp) Ca(Mn, Fe, Mg)Si ₂ O ₆	64° (81+8°)	Y = b X:c = 48°	MCL pris c	110 good at 87°	Pale green	H 6 G 3.55	Insol in acids. MnO 14.1, FeO 13.4, MgO 2.2%.	
1.725	1.725	<u>1.728</u>	1.737	.012	RHODONITE (Mn, Fe, Ca, Mg)Si ₃ O ₆	70°	---	TCL	100, 001 perf 010 less so	Brownish-pink	H 4 G 3.65 F 3	Nearly insol in acids. MnO 29.2, FeO 14.5, CaO 6.6, MgO 1.9%.	
1.714	1.727	<u>1.728</u>	1.733	.006	CHLORITOID (Fe, Mg, Mn) ₂ Al ₄ Si ₂ O ₁₀ (OH) ₄	53° r > v str	X = b Z:c = 12° el neg	MCL and TCL	001 perf	Greenish-gray	H 6.5 G 3.5 F diff	Dec by H ₂ SO ₄ . Pleoc, X green, Y blue, Z yellowish. FeO 27.1, MnO 0.2, Fe ₂ O ₃ 1.2%.	
1.696	1.730	<u>1.730</u>	1.798	.102	OJUELAITE ZnFe ⁺³ ₂ (AsO ₄) ₂ (OH) ₂ 4H ₂ O	(73+2°)	X = b Z = c	MCL f1b	---	Chartreuse	H 3 G 3.39	Sol in acids.	
1.715	1.730	<u>1.795</u>	.080	RUTHERFORDINE (UO ₂)CO ₃	53°	X = b Y = c	ORTH	010 perf	Yellow	G 5.7 infus	Diss by acids with eff. Pleoc, X nearly col, Y pale yellow, Z pale greenish-yellow.		
1.727	1.716	<u>1.746</u>	.030	MANGANABINGTONITE Ca ₂ (Mn, Fe) ₂ Fe ⁺³ Si ₅ O ₁₄ (OH)	80° r > v str	Z = b Y ~ c	TCL pris c	001 perf 010, 100 less so	Reddish-brown	H 6 G 3.45 F 3	Insol in acids. Pleoc, X green, Y pale rose, Z rose-brown. MnO 7.9, FeO 4.5, Fe ₂ O ₃ 12.3%.		
1.734													

1.723 1.741	<u>1.730</u>	1.750	.025	AUGITE, titanian (Pyroxene grp) (Ca, Na) (Mg, Fe, Al, Ti) (Si, Al) ₂ 6	40° (54+11°)	MCL $Y = \frac{b}{c}$ $Z:c = 45^\circ$	H 6 G 3.42	Brown	110 good at 87°	Insol in acids. FeO 5.0, MnO 0.2, Fe ₂ O ₃ 4.4, Al ₂ O ₃ 7.35, TiO ₂ 5.5%.
1.722 1.741	<u>1.730</u>	1.823	.121	KANONAITE (Andalusite ser) (Mn ⁺³ , Al) AlSiO ₅	53° (60+2°) disp str	ORTH $X = \frac{a}{D}$ $Y = \frac{b}{D}$	H 6.5 G 3.40	Greenish-black	110 poor	Pleoc., X yellow-green, Y bluish-yellow. Z deep golden-yellow. Mn ₂ O ₃ 32.2%.
1.707 1.741	<u>1.730</u>	1.748	.035	GAIITTE Ca ₂ Zn(AsO ₄) ₂ •2H ₂ O	88°	TCL ---	010, 001, 011 good	White	H 5 G 3.81	ZnO 15.3, MgO 1.2%.
1.713 1.741	<u>1.730</u>	1.746	.016	MONAZITE, sulfatite (Ce, La, Th)(PO ₄ , SO ₄)	5°	MCL $X = \frac{b}{c}$	100 dist	Reddish-brown	G 4.54	SO ₃ 3.1%.
1.778 1.741	<u>1.730</u>	1.730	.008	CARBOIRITE FeAl ₂ GeO ₅ (OH) ₂	55-70°	TCL $Z:c = 4-$ $\frac{14}{14}^\circ$ r > v	001 perf	Pale to bright green	---	Poly tw. Pleoc., X blue-green, Y light blue, Z cols to yellowish.
1.727 1.741	<u>1.730</u>	1.735	.008	ALLUAUDITE (Varulite ser) (Na, Ca) ₄ Fe ⁺² ₄ (Mn, Fe ⁺² , Fe ⁺³ , Mg) ₈ (PO ₄) ₁₂	Large r > v	MCL ---	001	Greenish-black	H 5 G 3.5 fus	Diss by acids. Pleoc. in green. "Huehner-kobelite."
(1.713) 1.741 (1.766)	<u>1.731</u>	1.738	.011	LOPEZITE K ₂ Cr ₂ O ₇	50° (55+3°) r > v med	TCL ---	010 perf 100, 001 dist	Orange-red	H 2.5 G 2.69	SoI in H ₂ O. Pleoc., X reddish-yellow, Y yellow, Z greenish-yellow.
1.727 1.741	<u>1.732</u>	1.805	.091	HEDENBERGITE (Diopsidite ser, Pyroxene grp) Ca(Fe, Mg)Si ₂ O ₆	49° r > v wk	MCL $Y = \frac{b}{c}$ $Z:c = 44^\circ$	110 good at 87°	Green	H 6 G 3.48	Insol in acids. Pleoc., X pale green, Z dark green. FeO 24.7, MnO 0.0., Fe ₂ O ₃ 1.9%.
1.726 1.753	<u>1.732</u>	1.753	.027	PYROMANGITE (Mn, Fe, Mg)SiO ₃	42° (57+14°)	TCL ---	---	Pink to dark brown	H 6 G 3.61	Insol in acids. dec by NH ₄ OH. Pleoc., X and Y pale yellow, Z dirty gray to yellow-green.
1.728 1.742	<u>1.732</u>	1.746	.018	FERRIMOLYBDITE Fe ₂ (MoO ₄) ₃ •8H ₂ O (?)	28° r < v str	ORTH fib c	001 dist	Sulfur-yellow	H 1-2 G 2.99	Insol in acids. dec by NH ₄ OH. Pleoc., X cols, Y yellow-brown, Z reddish-brown.
v 1.79	<u>1.733</u>	1.935	.215	BATISITE BaNa ₂ Ti ₂ (Si ₂ O ₇) ₂	8° (34+6°) r < v str	ORTH fib c	100 fair	Dark brown	H 6 G 3.43 F easy	Insol in acids. Pleoc., X cols, Y yellow-brown, Z reddish-brown.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ	Biref								
1.710 \wedge	1.730	<u>1.733</u>	1.763	.033	STRENGITE $\text{Fe}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$	$X = \frac{a}{b}$ $Y = \frac{c}{b}$ $r < \frac{v}{s} \text{ str}$	ORTH	---	Rose to violet	H 4-5 G 2.8 fus	Diss by acids. Pleoc wk, \wedge in rose, abs Z $>$ X.	
1.717 \wedge	1.723	<u>1.733</u>	1.755	.032	MUKHINITE (Epidote grp) $\text{Ca}_2\text{Al}_2\text{V}(\text{SiO}_4)_3(\text{OH})$	$Y = \frac{b}{c}$ $Z:a = 32^\circ$	MCL	001 perf 100 less so	Brownish-black	H 8	Diss with diff by acids. Pleoc str, X pale olive-green, Y pale red-brown, Z reddish-brown.	
1.557	<u>1.734</u>	2.07	.513	IDRIALITE Hydrocarbon, near $\text{C}_{22}\text{H}_{14}$	84° $r > v \text{ wk}$	$X = \frac{b}{c}$ $Z = \frac{c}{a}$ $e1 c \overline{t} v \text{ neg}$	ORTH	001 perf 100 poor	Yellow, greenish	H 1-1.5 G 1.22 F 1	Dimethylbenzphenanthrene. Pleoc, X pale yellow, Y and Z yellow.	
1.719 \vee	<u>(1.734)</u>	1.778	.059	SEEDOZERITE $(\text{Na}, \text{Ca})_2(\text{Zr}, \text{Ti}, \text{Mn})_2$ $\text{Si}_2\text{O}_7(0, \text{F})_2$	62° $r > v \text{ str}$	$X = \frac{b}{c}$ $Y:c = 13^\circ$	MCL	001 perf	Dark yellow to reddish-brown	H 4-5 G 3.43 F easy	Diff diss by HCl. Pleoc, X dark red, Y red, Z yellow.	
1.758	<u>1.734</u>	1.770	.059	BABINGTONITE $\text{Ca}_2(\text{Fe}^{+2}, \text{Mn})\text{Fe}^{+3}\text{Si}_5$ $0.14(\text{OH})$	88° $r > v \text{ str}$	---	TCL pris c	001 perf 010, 100 less so	Greenish-black	H 5.5-6 G 3.48 F 3	Insol in acids. Pleoc, X green, Z brown. Feo 7.9, Mn 3.0, Fe ₂ O ₃ 13.6%.	
1.730 \wedge	1.719	<u>1.734</u>	1.750	.031	CARBOIRITE $\text{FeAl}_2\text{GeO}_5(\text{OH})_2$	$Z:c = 7^\circ$	TCL	001 perf	Pale to bright green	---	Poly tw. Pleoc, X blue-green, Y light blue, Z col's to yellowish.	
1.730 \wedge	1.731	<u>1.735</u>	1.740	.009	FASSAITE (Pyroxene grp) $\text{Ca}(\text{Mg}, \text{Fe}, \text{Al})(\text{Si}, \text{Al})_2\text{O}_6$	$60-75^\circ$ $r > v$	---	MCL pris	110 good at 87°	Green	H 6 G 3.37	Insol in acids. Feo 1.2, Fe ₂ O ₃ 6.2, Al ₂ O ₃ 12.3, TiO ₂ 3.0%.
1.719	<u>1.736</u>	1.751	.024	RHODONITE $(\text{Mn}, \text{Fe}, \text{Mg}, \text{Ca})\text{SiO}_3$	49° $(75+10^\circ)$ $r > v$	$Y = \frac{b}{c}$ $Z:c = 48^\circ$	---	TCL	100, 001 perf 010 less so	Brownish-pink	H 4 G 3.65 F 3	Insol in acids. MnO 49.2, Feo 0.6, MgO 0.8, CaO 2.6, Fe ₂ O ₃ 0.3%.
1.728 \wedge <u>1.748</u>	1.733	<u>1.736</u>	1.744	.011	INNELLITE $\text{Na}_2(\text{Ba}, \text{K})_4(\text{Ca}, \text{Mg}, \text{Fe})\text{Si}_4\text{O}_18(\text{OH}, \text{F})_{1.5}$ (SO_4)	68°	---	TCL plates	010, 110, 101 perf 001	Yellow-brown	H 5 G 3.96	Nearly insol in acids. Poly tw. Anom blue interf colors. Pleoc, X and Y light yellow, Z pale brownish-yellow.
1.726	<u>1.737</u>	1.766	.040		82° $(64+6^\circ)$ $r > v \text{ str}$	$Z \sim \frac{a}{c}$ $\frac{1}{100} \text{ str}$						
						$\frac{X}{Y} = 77^\circ$						
						$\frac{Y}{Z} = 12^\circ$						
						$\frac{Z}{X} = 88^\circ$						

1.735	<u>1.737</u>	1.80	.065	BARIO-ORTHOJOAQUINITE $\text{Ba}_2(\text{Ba}, \text{Sr})_2\text{Fe}^{+2}\text{Ti}_2\text{Si}_8$	10-15°	$X = \frac{a}{c}$ $Z = \frac{a}{c}$	Yellow-brown	G 3.96
1.732	<u>1.737</u>	1.744	.012	CHAMBERSITE $\text{Mn}_3\text{B}_7\text{O}_{13}\text{Cl}$	83°	ORTH ps cub	Conch to uneven	H 7 G 3.49 F 3-4
1.588	<u>1.739</u>	1.898	.310	URICITE $\text{C}_5\text{H}_4\text{N}_4\text{O}_3$	84° (96+1°)	---	cols to deep purple	G 1.84 F 1
								= Uric acid (2-, 6-, 8-trihydroxyxypurine). Insol in H_2O , diss by alkalis.
1.726	<u>1.739</u>	1.788	.062	ANTLERITE $\text{Cu}_3(\text{SO}_4)(\text{OH})_4$	55° r < v very str	$X = \frac{b}{c}$ $Y = \frac{c}{b}$ el pos	010 perf tab 010 100 poor	H 3.5 G 3.88 F 2.5
1.730	(1.739)	1.748	.018	SUZUKIITE $\text{Ba}_2\text{V}^{+4}2^0\text{Si}_4\text{O}_{12}$	~ 90° r < v very str	$X = \frac{a}{b}$ $Y = \frac{a}{b}$	010 perf 001, 100 dist	H 4-4.5 G 4.03
								Pleoc., X pale green, Y yellow-green, Z blue-green.
1.733	<u>1.740</u>	1.769	.036	LAMPROPHYLLITE $(\text{Sr}, \text{Ba})_2\text{Na}_2\text{Ti}_3(\text{SiO}_4)_4$ $(\text{OH}, \text{F})_2$	32° (58+8°) r > v str	$X = \frac{b}{c}$ $Z:C = 0-8^\circ$	MCL 100 perf 011 good 010 dist	H 4-5 G 3.50
								Poly tw 100. Pleoc wk, X and Y light yellow, Z brownish- yellow, Y brown- yellow, Z brown.
1.741								
1.739	<u>1.740</u>	1.760	.021	ARDENNIITE $\text{Mn}_4(\text{Al}, \text{Mg})_6[(\text{V}, \text{As})_0_4]$ $\text{Si}_5\text{O}_{18}(\text{OH})_6$	Small r > v or r < v str	$Z = \frac{a}{c}$ $X \text{ or } Y = \frac{b}{c}$	010 perf prts c 110 dist	H 6.5 G 3.65 F 2.5
								Nearly insol in acids. Pleoc. wk to str., X brownish yellow, Y golden yellow, Z pale yellow.
1.753								
1.740	<u>1.740</u>	1.760	.020	ANDREMEYERITE $\text{BaFe}^{+2}\text{Si}_2\text{O}_7$	0-80° disp extr	$Z = \frac{b}{c}$	MCL prts 100, 010 perf	H 3-3.5 G 2.41
								Tw on (100). 2V 0° at 490 nm, to 40° at 540, to 0° at 585, to 80° at 640 nm. X:c. 2° at 670 nm to 61° at 470 nm.
1.736	<u>1.740</u>	1.745	.009	STAUROLITE $(\text{Fe}, \text{Mg}, \text{Zn})_2\text{Al}_9(\text{Si}, \text{Al})_4$ $(\text{Si}, \text{Al})_4\text{O}_{22}(\text{OH})_2$	83° r > v wk	$Z = \frac{c}{b}$ $X \sim \frac{b}{c}$ el pos	010 dist MCL ps orth	H 7 G 3.75 infus
1.746								Insol in acids. Pleoc Cruciform tw.
1.735	<u>1.741</u>	1.767	.032	BARYTOLAMPROPHYLLITE $(\text{Na}, \text{K})_2(\text{Ba}, \text{Ca}, \text{Sr})_2$ $(\text{Ti}, \text{Fe})_3(\text{SiO}_4)_4(0, \text{OH})_2$	55° r > v str	$X = \frac{b}{c}$ $Z:C = -3^\circ$	100 perf 011 good	H 4-5 G 3.50
1.740	$\frac{\wedge}{\vee}$ 1.754							Pleoc. str., X light yellow, Z brown, abs Z > Y > X.
1.730	$\frac{\wedge}{\vee}$ 1.751	1.761	.026	FERROAUGITE (Pyroxene grp) $\text{Ca}(\text{Fe}, \text{Mg})(\text{Si}, \text{Al})\text{Si}_2\text{O}_6$	51°	$Y = \frac{b}{c}$ $Z:C = 40^\circ$	110 good MCL at 87°	H 6 G 3.42
								Insol in acids. FeO 24.0, Fe_2O_3 1.2, Al_2O_3 4.1, MnO 0.4%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index		Brief	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.703	1.738	(1.741)	1.743	.005 SERENDIBITE $\text{Ca}_2(\text{Mg},\text{Al})_6(\text{Si},\text{Al},\text{B})_6$ 0 ₂₀	~90° r < v str	Ext:tw lam = 40°	TCL plates	---	Sky-blue to indigo-blue	H 6.5 G 3.52 infus	Insol in acids. Poly tw, Pleoc, X pale yellow-green, Y nearly col's, Z deep blue.
1.810	1.722	<u>1.742</u>	1.763	.041 ADAMITE $\text{Zn}_2(\text{AsO}_4)(\text{OH})$	88° r < v str	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	ORTH	101 good 010 poor	Yellow, green, rose	H 3.5 G 4.44	Diss by acids. U opt neg. May contain Cu or Co.
1.732	1.738	<u>1.742</u>	1.754	.016 PYROMANGITE (Mn,Fe,Mg,Ca)Si ₁₀ O ₃	39° (60-16°)	Z':c = 45° on T00	TCL	---	Pink to brown	H 6 G 3.76	Insol in acids. MnO 29.3, FeO 19.1, MgO 2.0, CaO 2.9%. (Series with Pyroferrite.)
1.750											
1.715	1.743	1.783	.068 JOHILLERITE $\text{Na}(\text{Ng},\text{Zn})\text{Cu}(\text{AsO}_4)_3$	80°	Z = $\frac{b}{c}$ X:c = 16°	MCL	010 perf 100, 001 good	Violet	H 3 G 4.15	Pleoc, X violet red, Y bluish violet, Z greenish-blue.	
1.741	1.744	1.768	.027 SCORODITE (Variscite grp) $\text{FeAsO}_4 \cdot 2\text{H}_2\text{O}$	40° r > v str	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	ORTH pyram	201 imperf	Green to brown	H 3.5-4 G 3.14 F easy	Diss by acids. FeO 25.8, Al ₂ O ₃ 4.7%.	
1.624	1.796	1.744	1.753	.015 ALLANITE (Epidote grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	70-80° r > v	Y = $\frac{b}{c}$ X:c = 5-40°	MCL	001 imperf	Brownish-black	H 5-6 G 3.69 F 3	Slowly attacked by HCl, gel. U pleoc in browns and reds. G and n's decrease with alteration. FeO = 9.2%.
1.708	1.738	<u>1.744</u>									
1.768											
1.698	1.745	1.793	.095 LIBETHENITE $\text{Cu}_2(\text{PO}_4)(\text{OH})_4$	90° r > v str	X = $\frac{b}{c}$ Y = $\frac{c}{c}$	ORTH	100, 010 poor	Olive-to blackish-green	H 4 G 3.97 F 2	Diss by acids. Pleoc in yellow and green.	
1.75	--	<u>1.74</u>	--	--	--	ORTH(?) earthy masses	---	Creamy white	H 2.5 G 2.8	---	
1.722	1.740	<u>1.746</u>	1.765	.025 ASTROPHYLLITE (K,Na) ₃ (Fe,Mn) ₇ Ti ₂ Si ₈ O ₂₄ (OH) ₇	80° (59+10°) r > v str	TINTICITE $\text{Fe}_6(\text{PO}_4)_4(\text{OH})_6 \cdot 7\text{H}_2\text{O}$	TCL	001 perf	Brown	H 3 G 3.24 F 3	Dec by acids. Pleoc, yellow to orange, abs X > Y > Z. FeO 20.7, MnO 1.4, Fe ₂ O ₃ 11.2%.

1.740 △ 1.751 neg 1.753	1.739	<u>1.746</u>	1.753	.014	STAUROLITE, cobaltian (Fe,Mg,Co) ₂ Al ₉ (Si,Al) ₄ 0 ₂₂ (OH) ₂	~ 90° r > v wk	Z = $\frac{c}{b}$ X ~ $\frac{b}{d}$ el pos	MCL ps orth	010 dist	Blue	H 7 G 3.76 infus	
1.737	1.747	1.768	.031	CREASEYITE Pb ₂ Cu ₂ Fe ₂ Si ₅ ⁰ 17•6H ₂ O 0 ₂₂ (OH) ₂	69° r < v wk	X = $\frac{a}{c}$ Z = $\frac{c}{e}$	ORTH fib	---	Green	H 2.5 G 4.1 fus	Dec by hot HNO ₃ . Pleoc., X blue, yellow-green, Y clear green, abs Z = X > Y.	
1.746	<u>1.748</u>	1.768	.022	PROSPERITE CaZn ₂ (AsO ₄) ₂ •H ₂ O	34° r > v str	Y = $\frac{b}{c}$ Z:c = 27°	MCL prism	---	Col s, white	H 4.5 G 4.31	---	
1.736 △	1.739	<u>1.748</u>	1.760	RHODONITE, ferroan (Mn,Fe,Mg,Ca)SiO ₃	75°	---	TCL	100, 001 perf 010 less so	Brownish- pink	H 4 G 3.76 F 3	Insol in acids. MnO 29.3, FeO 19.1, MgO 2.0, CaO 2.4%.	
1.746	1.748	1.756	.010	CHRYSOBERYL BeAl ₂ O ₄	70° r > v	X = $\frac{c}{b}$ Y = $\frac{c}{d}$	ORTH tab 100 ps hex tw	110 dist 010 imperf	Green, red, yellow	H 8.5 G 3.75 infus	Insol in acids. Tw pl 031. Pleoc., X red, Y orange to yellow, Z emerald- green.	
1.748	<u>1.749</u>	1.753	.005	WOLFEITE (Triploidite ser) (Fe,Mn) ₂ (PO ₄) ₂ (OH)	Med r > v extr	X = $\frac{b}{d}$ disp dist	MCL pris	010 good 120 fair	Reddish- brown	H 4.5-5 G 3.8 fus	Diss by acids.	
1.763	1.715	<u>1.75</u>	1.80	ERYTHROSIDERITE K ₂ FeCl ₅ •H ₂ O	62° r > v str	X = $\frac{a}{b}$ Z = $\frac{c}{d}$	ORTH tab 100	210, 011 perf	Red	G 2.37	Sol in H ₂ O. Yellow in thin section.	
1.772	1.732	<u>1.750</u>	1.778	PIEMONITITE (Epidote grp) Ca ₂ (Al,Mn,Fe) ₃ (SiO ₄) ₃ (OH)	Large	Y = $\frac{b}{c}$ X:c = -5° el-c v pos	MCL pris	001 perf	Reddish- brown	H 6 G 3.45	Insol in acids. Pleoc str, X buff, Y deep lavender, Z pink. Mn ₂ O ₃ 11.8, Fe ₂ O ₃ 3.8%.	
1.748	1.750	1.764	.016	PYROXEROITE (Fe,Mn,Ca)SiO ₃	30°	---	TCL	---	Brown, yellow	H 6 G 3.80	Insol in acids. FeO 28.3, MnO 20.6, CaO 1.9%. (Series with Pyroxmangite.)	
1.740 neg △ 1.741	1.741	<u>1.751</u>	1.774	ACMITE (Pyroxene grp) Na(Fe,Zn,Mn)Si ₂ O ₆	Med (68+7°)	Y = $\frac{b}{c}$ Z:c = 60°	MCL	110 good at 87°	Reddish- brown	H 6 G 3.59 F 3	Insol in acids. FeO 12.9, FeO 1.4, ZnO 8.8, MnO 7.2, Na ₂ O 5.0%.	
1.745	<u>1.751</u>	1.760	.015	TIRAGALLOITE Mn ₄ As ⁺⁵ Si ₃ O ₁₂ (OH)	38-46° (79+16°)	X = $\frac{a}{b}$ Y = $\frac{c}{d}$ Z ~ $\frac{c}{e}$ el pos	MCL	100 good	Orange	G 3.84	Insol in HCl.	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.864	1.750	1.751	1.761	.011	SYNADELPHITE (Mn,Mg,Ca,Pb) ₉ (AsO ₃) ₂ (OH) ₉ •2H ₂ O	37° r > v	---	TCL ps orth imperf	Col s	H 4.5 G 3.64 F 2	Diss by acids. Pleoc wk, X and Y cols, Z light brown.	
1.740	1.752	(1.80)	(.06)		Unnamed ferric phosphate Fe ₃ (PO ₄) ₂ (OH) ₃ •3-6H ₂ O	Med r < v str	---	---	Brown	G 2.55	Pleoc, X pale brown, Z reddish brown.	
1.732	1.745	1.753	1.771	.026	HEDENBERGITE (Diopside ser., Pyroxene grp.) Ca ₂ Si ₂ O ₆	52° (68+9°) r > v wk	Y = b Z:c = 41°	MCL pris c	110 good at 87°	Greenish-black	H 6 G 3.65 fus	Insol in acids. FeO 29.7, MnO 0.9, Fe ₂ O ₃ 1.4%.
1.745	1.753	1.778		.033	O'DANIELITE Na(Zn,Mg) ₃ (AsO ₄) ₃	60°	Z = b Y:c = 18°	MCL	010, 100 perf, .001	Pale violet	H 3 G (4.49)	---
1.750	1.753	1.780		.030	NAGASHIMALITE Ba ₄ (V ⁺³ , Ti) ₄ B ₂ Si ₈ O ₂₇ Cl (0, OH) ₂	30° r > v str	X = a Y = c	ORTH	---	Greenish-black	G 4.08	Pleoc, X greenish-yellow, Y green, Z greenish-brown, abs Z > Y > X.
1.753	1.753	1.770		.017	ARDENNITE Mn ₄ Al ₆ [(V, As) ₄]Si ₅ O ₁₈ (OH) ₆	Smal l r > v or r < v str	Z = a X or Y = b	ORTH pris c	010 perf 110 dist	Yellow to brown	H 6.5 G 3.65 F 2.5	Nearly insol in acids. Pleoc in yellow, wk to str.
1.740	1.753	1.753			STAUROLITE (Fe,Mg,Zn) ₂ Al ₉ (Si,Al) ₄ O ₂₂ (OH) ₂	80° r > v wk	Z = c X ~ d	MCL ps orth	010 dist	Reddish-brown	H 7 G 3.77 infus	Insol in acids. Cruciform tw. Pleoc wk, X and Y light yellow, Z orange-yellow.
1.746	1.747	1.753		.014								
1.751 neg												
1.741	1.743	1.754	1.777	.034	BARYTOLAMPROPHYLLITE (Na,K) ₂ (Ba,Ca,Sr) ₂ (Ti,Fe) ₃ (SiO ₄) ₄ (O,OH) ₂	30° (70+7°) r > v str	X = b Z:c = -7°	MCL	100 perf 011 good 010 dist	Dark brown	H 4-5 G 3.50	Dec by warm HCl. Pleoc, X blue-green, Y yellow, Z brown, abs X > Z > Y.
1.775	1.725	1.755	1.815	.090	OLMSTEADITE KFe ⁺² ₂ (Nb,Ta)(PO ₄) ₂ O ₂ 2H ₂ O	~ 60° (73+3°)	X = c Y = a	ORTH	100, 001 good	Deep brown to black	H 4 G 3.31	AUSTINITE CaZn(AsO ₄)(OH)
1.774	1.752	1.756	1.779	.027		41°	X = a Y = c	ORTH	011 good	Col s	H 4-4.5 G 4.13	

1.730	<u>1.758</u>	1.838	.108	AZURITE $\text{Cu}_3(\text{CO}_3)_2(\text{OH})_2$	67° (64+2°) $r > v$ str	X = $\frac{b}{c}$ $\frac{z:c}{disp} = -13°$ disp dist	MCL	011 perf inter- rupted, 100 fair	Azure- blue	H 3.5-4 G 3.77 F 3	Diss by acids with eff. Pleoc in blue, Z > Y > X.	
(1.734)	1.725	<u>1.758</u>	1.830	.105	SEIDOZERITE (Na, Ca) ₂ (Zr, Ti, Mn) ₂ Si ₂ O ₇ (O, F) ₂	68° (71+2°) $r > v$ str	X = $\frac{b}{c}$ $y:c = 13°$	MCL	001 perf	Brownish- red to yellow	H 4-5 G 3.47 F easy	Diff diss by HCl. Pleoc, X dark red, Y red, Z yellow.
1.750	<u>1.756</u>	<u>1.758</u>	1.768	.012	PYROXEROITE (Fe, Ca, Mn)SiO ₃	40°	---	TCL	---	Yellow, brown	H 6 G 3.80	Insol in acids.
1.754	<u>1.749</u>	<u>1.760</u>	1.779	.030	LAMPROMPHYLITE (Sr, Ba) ₂ Na ₂ Ti ₃ (SiO ₄) ₄ (O, F) ₂	32° (75+8°) $r > v$ str	X = $\frac{b}{c}$ $z:c = 1.8°$	MCL	100 perf 011 good	Yellow- brown	H 4-5 G 3.5	Poly tw 100. Pleoc wk, X and Y light yellow, Z brownish- yellow.
1.75	<u>1.733</u>	<u>1.763</u>	1.807	.016	KREMERSITE (Erythrosiderite ser) (NH ₄ , K) ₂ FeCl ₅ ·H ₂ O	70°	X = $\frac{a}{b}$ $z = \frac{b}{c}$	ORTH ps cub	---	Ruby red	G 2.18	Sol in H ₂ O. Data calc from end- members.
1.762 neg △ 1.772	1.755	<u>1.763</u>	1.773	.018	EULITE (Orthopyroxene ser, Pyroxene grp) (Fe, Mg, Mn) ₂ Si ₂ O ₆	84° $r > v$	X = $\frac{a}{c}$ $z = \frac{c}{c}$	ORTH pr's c	210 good at 87°	Dark green	H 5.5 G 3.88 F 4	Insol in acids. FeO 41.7, MnO 5.0, Fe ₂ O ₃ 0.3% (En ₁₂).
1.765	<u>1.765</u>	<u>1.765</u>	1.800	.035	HENRITERMIERITE (Garnet grp) Ca ₃ (Mn, Al) ₂ (SiO ₄) ₂ (OH) ₄	Small	---	TET	Conch	Clove- to apricot- brown	G 3.34 fus	Dec by warm HCl. Pleoc wk, pale yellow, E lemon- yellow.
1.747	<u>1.765</u>	<u>1.765</u>	1.78	.033	JOESMITHITE (Ca, Pb) ₃ (Mg, Fe ⁺² , Fe ⁺³) ₅ Si ₆ Be ₂ O ₂₂ (OH) ₂	60-70°	---	MCL	110 perf	Lustrous black	H 5.5 G 3.83	Insol in cold acids. Pleoc, X and Z olive, Y olive-brown, abs Y > X, Z.
1.750	<u>1.766</u>	<u>1.766</u>	1.85	.10	KOMAROVITE (H, Ca) ₂ Nb ₂ Si ₂ O ₁₀ (OH, F) ₂ ·H ₂ O	48°	X = $\frac{a}{b}$ $z = \frac{b}{c}$	ORTH	001 fair	Pale rose	H 1.5-2	---
1.765	---	1.775	.010	NATRODUFRENITE	---	el pos	MCL spheru- litic	---	Bronze- green	G 3.20	Pleoc, X yellow, Z green.	
1.763	---	1.769	.006	ROWLANDITE Y ₃ (SiO ₄) ₂ (OH, F) (?)	---	---	TCL(?)	Conch	Green, brown	H 6-6.5 G 4.85	Gel with acids. Com- monly metamict.	
(1.731) △ 1.802	<u>1.760</u>	<u>(1.766)</u>	1.775	.015	ALLUAUDITE (Na, Ca)Fe ⁺² (Mn, Fe ⁺² , Fe ⁺³ , Mg) ₂ (PO ₄) ₃	Large $r > v$ mod	Z = $\frac{b}{c}$ $y:c = 18°$	MCL 2 others u mass	T01 perf	Greenish- black	H 5.5-5 G 3.5 F 3	Diss by acids.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
(1.717) \wedge	1.753	<u>1.767</u>	1.822	.069	JOAQUINITE $Ba_2NaCe_2Fe(Ti,Nb)_2Si_8$ $0_{26}(OH,F) \cdot H_2O$	$X = \frac{a}{c}$ $Z = \frac{v}{c}$	MCL ps orth	---	Brown to black	H 5 G 3.98 F 2.5	Insol in acids. Pleoc wk, X and Y cols, Z light yellow.	
1.736	<u>1.767</u>	1.796	.060	ARTHURITE $CuFe_2(AsO_4,PO_4,SO_4)_2$ $(0,OH)_2 \cdot 4H_2O$	$\sim 90^\circ$	$Y = \frac{b}{c}$ $Z:C = 10^\circ$	MCL	---	Apple-green	G 3.2	Pleoc, X cols to pale green, Y grass green, Z olive-green, abs Z > Y > X.	
1.744 \wedge	1.763	<u>1.768</u>	1.788	.025	ALLANITE (Epidote grp) $(Ca,Ce,Y)_2(Al,Fe)_3Si_3$ $O_{12}(OH)$	57° $r > v$	MCL	001 imperf	Dark brown	H 5.5 G 3.8 fus	Slowly attacked by HCl, gel. U pleoc in browns and reds. FeO 10.4, FeO ₃ 3.8%.	
1.710	<u>1.770</u>	1.840	.130	ROSSITE $CaV_2O_6 \cdot 4H_2O$	Large disp str	$Z \sim \frac{c}{b} = 45^\circ$ $Y:b = 45^\circ$	TCL pris tab	010 good	Yellow	H 2-3 G 2.45 F easy	Slowly sol in H ₂ O. Tw on 100.	
1.769	<u>1.770</u>	1.785	.016	HOLDENITE $(Mn,Mg)_6Zn_3(AsO_4)_2SiO_4$ $(OH)_8$	30° $r > v$ perc	$X = \frac{c}{D}$ $Y = \frac{v}{D}$	ORTH tab	010 poor subconch	Pink to red	H 4 G 4.11	Diss by acids.	
1.750 \wedge 1.788	1.754	<u>1.772</u>	1.795	.041	PIEMONITITE (Epidote grp) $Ca_2(Mn,Al,Fe)_3Si_3O_12$ $(OH)_8$	85° $r < v$	$Y = \frac{b}{c}$ $X:c = -6^\circ$ $el c\bar{v}$ pos	MCL pris	001 perf	Reddish-brown	H 6 G 3.47	Insol in acids. Pleoc str, X yellow, Y pink, Z crimson. Mn 22.0, Fe 23.6.4%.
1.763	1.768	<u>1.772</u>	1.790	.022	ORTHOFERROSTILITE (Orthopyroxene ser., Pyroxene grp) $(Fe,Mg)_2Si_2O_6$	$\sim 55^\circ$ $r < v$	$X = \frac{b}{c}$ $Z = \frac{v}{c}$	ORTH	210 good at 87°	Dark green	H 5.5 F 4	Insol in acids. Extrapolated value for En ₀ .
1.712 \wedge	(1.761)	<u>1.772</u>	1.781	.020	MANGANHUMITE (Humite grp) $Mn_7(SiO_4)_3(OH)_2$	84°	$X = \frac{a}{D}$ $Z = \frac{v}{D}$	ORTH	010	Orange-brown	H 4 G (4.05)	GeI with acids, MnO ~ 69, FeO ~ 1.0, MgO ~ 0.8%.
1.752	<u>1.773</u>	1.796	.044	LUETHEITE $Cu_2Al_2(AsO_4)_2(OH)_4 \cdot H_2O$	$(88+5^\circ)$ $r < v$ mod	$X = \frac{b}{c}$ $Z:C = 10^\circ$	MCL tab	100 fair to good	Blue to greenish-blue	H 3 G 4.28 F easy	Diss by hot HCl. Pleoc wk in pale blue, abs Z = Y > X.	
1.772 \vee	(1.773)	1.792	.020	WARRICKITE $(Mg,Ti,Fe,Al)_2(BO_3)_0$	Small	$X = \frac{c}{D}$ $Y = \frac{v}{D}$ $el c\bar{v}$ neg	ORTH pris	100 perf	Dark brown to black	H 3.5-4 G 3.35 infus	Dec by H ₂ SO ₄ . Pleoc X brown, Y and Z light brown.	
1.813												

1.770	<u>1.774</u>	1.83	.06	TARAMELLITE Ba ₄ (Fe,Ti,Mg,V) ₄ B ₂ Si ₆ O ₂₉ Cl	40° r > v str	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el pos	100 perf	Brownish-red	H 5.5 G 3.92 F easy									Insol in acids.	
(1.722) 1.812	<u>1.774</u>	1.787	.022	GADOLINITE (Y,Ca)FeBe ₂ Si ₂ O ₁₀	Med (80+6°) r < v	X = $\frac{b}{c}$ Z:c = 4- 13°	MCL	Conch	Greenish-to-brownish-black	H 6.5 G 4.1-4.5 infus								Diff diss by acids, gel.	
1.770	<u>1.774</u>	1.783	.013	AUSTINITE Ca(Zn,Cu)(AsO ₄) ₂ (OH)	Med r > v wk	X = $\frac{a}{c}$ Y = $\frac{b}{c}$	011 good	Cols to green	H 4-4.5 G 4.1									Diss by acids.	
1.775 1.795	<u>1.775</u>	1.800	.035	OGDENSBURGITE Ca ₄ Fe ⁺³ ₆ (AsO ₄) ₅ (OH) ₁₁ · 5H ₂ O	25° (65+7°)	---	Thin plates	One perf	Dark reddish-brown	H ~ 2 G 2.92								Pleoc moderate, X < Y = Z.	
1.765	<u>1.775</u>	1.835	.070	OLMSTEADITE KFe ⁺² ₂ {Nb,Ta}(PO ₄) ₂ O ₂ · 2H ₂ O	~ 60° (46+5°)	X = $\frac{c}{a}$ Y = $\frac{b}{a}$	ORTH	100, 001 good	Deep brown to black	H 4 G 3.36								Dec by warm HCl. Pleoc, X dark blue, Y light brown, Z dark brown, abs X > Z > Y.	
1.755 1.763	<u>1.777</u>	1.785	.022	YOSHIMURAITE (Ba,Sr) ₂ TiMn ₂ Si ₂ O ₈ (PO ₄ ,SO ₄)(OH,C1)	80-90° (106-111°) r > v	---	TCL	010 perf 101, 10T dist	Brown	H 4.5 G 4.13								Pleoc, X bright yellow, Y orange- brown, Z brown, abs Z > Y > X. Poly tw on 010.	
(1.730) 1.789	<u>1.777</u>	1.823	.046	MONAZITE (Ce,La,Nd,Th)PO ₄	14.5°	X = $\frac{b}{c}$ Z:c = 2°	MCL	100 dist	Green	H 5-5.5 G 5.3 infus							Diff sol in HCl. Pleoc wk. UO ₂ 1.5, ThO ₂ 11.3%.		
1.76	<u>1.78</u>	>>1.85	>>.07	KARPATITE C ₂₄ H ₁₂	Small r < v extr	X = $\frac{b}{c}$ Z:c = 21°	MCL	001, 100, 201 perf	Pale	H < 1 G 1.35 F 1							A hydrocarbon, Coronene. Sol in org solvents.		
1.774	<u>1.780</u>	1.813	.039	BERAUNITE Fe ⁺² Fe ⁺³ ₅ (PO ₄) ₄ (OH) ₅ · 4H ₂ O	Med large (47+8°) r > v str	Z = $\frac{b}{c}$ X:c = 6°	MCL tab	100 good	Reddish-brown	H 2 G 2.9-3.1 F 3							Diss by acids. Tw. Pleoc str, X and Y pale yellow to flesh, Z carnelian red.		
1.779	<u>1.780</u>	1.816	.037	CHERALITE (Ce,La,Th,Ca)(PO ₄ ,SiO ₄) ₄	18°	X = $\frac{b}{c}$ Z:c = 7°	MCL	010 dist	Dark to pale green	H 5 G 5.28							Pleoc wk, X and Y green, Z yellow-green. Compare Monazite.		
1.77	<u>1.78</u>	1.80	.03	VANDENBRANDEITE Cu(UO ₂)(OH) ₄	Large	Opt ax ~ (001)	TCL	110 perf 2 others	Dark green	H 4 G 5.0							Diss by hot HCl. Pleoc, yellow-green to blue-green.		
1.77	<u>1.78</u>	1.80	.03	MEDAITITE Mn ₆ V ⁺⁵ Si ₅ O ₁₈ (OH)	---	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el pos	MCL	100 good	---	G 3.70							---		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.794 ^y	1.776	1.780	1.805	.029	CARYINITE (Na,Ca,Pb) ₃ (Mg,Mn, Fe ⁺³) ₄ (AsO ₄) ₄ (?)	X = $\frac{c}{c}$ Y = $\frac{a}{d}$	MCL u mass	110, 010	Nut brown	H 4 G 4.27 F 2.5	Diss by HNO ₃ . Orange to red in section, not pleoc.	
1.793	1.782	1.815	.062	SHATTUCKITE Cu ₅ (SiO ₃) ₄ (OH) ₂	~ 90°	X = $\frac{c}{d}$ Y = $\frac{b}{d}$	ORTH	010, 100	Blue	H 3.5 G 3.9-4.1	Pleoc., pale blue to deep blue.	
1.774	1.782	1.798	.028	RETZIAN-(Nd) Mn ₂ (Nd,Ce,La)(AsO ₄) ₄ (OH) ₄	69°	X = $\frac{c}{d}$ Y = $\frac{b}{d}$	ORTH	---	Brown	---	Pleoc., X yellow, reddish-brown, Z brown.	
1.772 ^z	1.756	1.788	1.829	.073	PIEMONITITE (Epidote grp) Ca ₂ (Mn,Al,Fe) ₃ (SiO ₄) ₃ (OH)	86° r < v	Y = $\frac{b}{c}$ X:c = -9° el cl v pos	MCL pris	001 perf	Reddish-brown	H 6 G 3.47	Insol in acids. Pleoc str, X yellow, Y violet, Z red. Mn ₂ O ₃ 22.0, Al 20.3 12.4%, Fe ₂ O ₃ 6.4%.
1.777	1.788	1.800	.023	RETZIAN Mn ₂ Ce(AsO ₄)(OH) ₄	Large r < v wk	Y = $\frac{b}{c}$ X = $\frac{c}{c}$	ORTH	---	Chocolate to chestnut-brown	H 4 G 4.15 F diff	Diss by acids. Pleoc, X col s, Y pale yellow-brown, Z reddish-brown.	
1.778 ^v 1.801	1.787	1.839	.052	MONAZITE (Ce,La,Nd)PO ₄	10-16° r < v wk	X = $\frac{b}{c}$ Z:c = -4°	MCL	110 dist 010 diff	Brown to red	H 5-5.5 G 5.26 infus	Diss sol in HCl, dec by NH ₄ OH. Pleoc, X and Y pale yellow, Z dark yellow, Z greenish-yellow.	
1.733 ^v 1.827	1.78	1.79	.26	FERRIMOLYBDITE Fe ₂ (MoO ₄) ₃ •7-10H ₂ O	Small r < v dist	Y = $\frac{a}{c}$ Z = $\frac{c}{c}$	ORTH fib c	001 dist	Sulfur-yellow	H 1-2 G 2.99 F easy	Diss by HCl, dec by NH ₄ OH. Pleoc, X and Y pale yellow, Z yellow-green.	
1.75	1.79	1.85	.10	Hydrous U sulfate (?)	Med r < v str	Y = $\frac{b}{c}$ Z = $\frac{c}{c}$ el pos	ORTH 100 laths el c	---	Lemon-yellow	Soft	Diss by acids. Comp unknown.	
1.792	1.794	1.821	.029	JIMBOITE Mn ₃ B ₂ O ₆	35° r > v	Y = $\frac{b}{c}$ X = $\frac{a}{d}$ el pos	ORTH	110 perf parting 101	Light purple-brown	H 5.5 G 3.98	Diss by acids. Tw pl 101.	
1.780 ^z	1.794	1.803	.009	CARYINITE (Na,Ca,Pb) ₃ (Mg,Mn, Fe ⁺³) ₄ (AsO ₄) ₄ (?)	Small r > v wk	Y = $\frac{c}{d}$ Y = $\frac{a}{d}$	MCL u mass	110, 010	Brownish-red	H 3-4 G 4.3 F 2.5	Diss by HNO ₃ . Orange to red in section, not pleoc.	

1.774	<u>1.795</u>	1.815	.035	CONICHALCITE $\text{Ca}(\text{Cu}(\text{AsO}_4)(\text{OH})$	~ 90° r < v mod	---	ORTH equant	011 uneven	Grass- green	H 3 G 4.1 fus
1.774	<u>1.796</u>	1.814	.029	SCRODITE $\text{Fe}(\text{AsO}_4 \cdot 2\text{H}_2\text{O}$	75° r > v	$X = \frac{a}{c}$ $Y = \frac{c}{a}$	ORTH pyram	201 imperf	Green to brown	H 3.5 G 3.29 F 2-2.5
1.79	(1.79)	1.82	.03	KRZHANOVSITE $\text{MnFe}^{+3} \cdot 2(\text{PO}_4)_2 \cdot (\text{OH})_2 \cdot \text{H}_2\text{O}$	40-45° r < v str	---	ORTH prism	001 perf	Brown to greenish- brown	Diss by acids. Pleoc str, X wine-yellow, Y blue, Z red. Al 0 1.8%.
1.743	<u>1.80</u>	1.88	.137	TUNDRITE $\text{Na}_3(\text{Ce}, \text{La})_4(\text{Ti}, \text{Nb})_2\text{Si}_2\text{O}_8(\text{CO}_3)_3\text{O}_4(\text{OH}) \cdot 2\text{H}_2\text{O}$	Large	Z:c = 14°	TCL	010 perf	Brownish- to greenish- yellow	Pleoc wk, X pale yellow, Z greenish- yellow.
1.72	<u>1.80</u>	1.98	.26	HELMUTWINKLERITE $\text{PbZn}_2(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	70° r < v dist	X:c = 50° Z:D = 30°	TCL	---	Col to light blue	H 4.5 G 5.3
1.775	<u>1.800</u>	1.846	.071	BOTALLACKITE $\text{Cu}_2(\text{OH})_3\text{Cl}$	Med large r > v str	X ⊥ clv	MCL	One direction	Bluish- green	Diss by acids. Slightly pleoc.
1.792	<u>1.801</u>	1.852	.060	JINSHAJIANGITE $(\text{Na}, \text{K})_5(\text{Ba}, \text{Ca})_4(\text{Fe}, \text{Mn})_{15}(\text{Ti}, \text{Fe}^{+3}, \text{Nb})_8(\text{SiO}_4)_{15}(\text{F}, \text{OH})_{10}$	72° (47+6°) r < v	---	MCL tab	010, 100 perf	blackish- red to golden-red	G ~ 3.6 (H 4.30 kg/sq mm)
1.783	<u>1.801</u>	1.834	.051	FLINKITE $\text{Mn}^{+2}\text{Mn}^{+3}(\text{AsO}_4)(\text{OH})_4$	Large (74-5°) disp wk	X = b Z = $\frac{a}{c}$	ORTH tab	001	Greenish- brown	Diss by acids. Pleoc, X yellow-to brownish-green, Y yellow-green, Z orange-brown.
1.789	<u>1.801</u> (1.821)	1.849	.049	MONAZITE $(\text{Ce}, \text{La}, \text{Nd}, \text{Th})\text{PO}_4$	11° r < v wk	X = b Z:c = -10°	MCL	100 dist 010 diff	Yellow, red, brown	Diss sol in acids. Pleoc wk in yellow. ThO ₂ 11.2%. (Solid soil with Huttonite.)
1.766	<u>1.802</u>	1.835	.053	ALLUAUDITE $(\text{Na}, \text{Ca})_4\text{Fe}^{+2}(\text{Mn}, \text{Fe}^{+2}, \text{Fe}^{+3}, \text{Mg})_8(\text{PO}_4)_{12}$	79° r > v mod	Z = b Y:c = 18°	MCL u mass	T01 perf 2 others	Greenish- black	H 5-5.5 G 3.58 F 3
1.793	<u>1.804</u>	1.87	.08	ENIGMATICITE $\text{Na}_2\text{Fe}_5\text{TiSi}_6\text{O}_{20}$	Small	Z:c = 40°	TCL prism	010, 100	Black	H 5.5 G 3.8 F 3
1.82	<u>1.82</u>									

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.832	1.805	1.805	1.915	.110	LUDWIGITE (Mg, Fe) ₂ Fe ⁺³ Be ₅ O ₅	$X = \frac{a}{d}$ $Y = \frac{b}{d}$ Small r < v extr	ORTH	Uneven, parting 001	Greenish-black	H 5 G 3.8 F diff	Diss by acids. Pleoc, X and Y dark green, Z brown.	
~1.81	(~1.812)	~1.83	~.02		WELSHITE Ca ₂ Mg ₄ Fe ⁺³ Sb ⁺⁵ Be ₂ Si ₄ O ₂₀	(~25°)	---	TCL pris	Reddish-black	H 6 G 3.77	Opt char uncertain. 2E ~ 45°.	
1.742	1.772	1.810	1.863	.091	OLIVENITE Cu ₂ (AsO ₄) ₂ (OH)	$X = \frac{b}{c}$ $Y = \frac{c}{c}$ ~90° r < v str	ORTH acic	101 conch	Olive-green	H 3 G 4.46 F 3	Diss by acids. In thin-section pale green, not pleoc.	
~	1.795	1.810	1.835	.040	TAVORITE LiFe(Po ₄) ₂ (OH)	Large r > v str	---	TCL	Yellow-green	G 3.29 fus	Diss by acids. Pleoc, X and Y yellow-green, Z col.	
1.808	1.812	1.838	~.030		TOERNEBOHMITE (Ce, La, Si ₂ O ₈ (OH)) ₂	20-30° (43+10°) r < v dist	---	MCL ps hex	---	H 4.5 G 4.94	Slowly diss by hot conc. acids. Pleoc, X rose to greenish-yellow, Y bluish-green, Z rose, abs Y > X = Z.	
1.852	1.80	---	1.87	.07	KEYSITE (Cu, Zn, Cd) ₃ (AsO ₄) ₂	Disp str $Y = \frac{b}{c}$ $X: \underline{c} = 10^\circ$	MCL	001 good	Deep sky-blue	---	Pleoc, X pale blue, Y greenish blue, Z deep blue. Opt char unk.	
1.774	1.801	1.812	1.824	.023	GADOLINITE Y ₂ Fe ⁺² Be ₂ Si ₂ O ₁₀	85° r < v	MCL	Conch	Greenish-to-brownish-black	H 6.5 G 4.33 infus	Gel with acids. Pleoc, olive to green.	
~	1.810	1.813	1.855	.045	DUFRENITE Fe ⁺² Fe ⁺³ (PO ₄) ₃ (OH) ₅ [*] 2H ₂ O	Small r > v extr	Z = b el neg	MCL fib	010 perf	Dark green	H 3.5-4 G 3.2-3.4 F 2.5	Diss by acids. Pleoc str, X yellow-brown, Y deep blue, Z deep greenish-blue, abs Z > Y > X.
(1.773)	1.808	1.813	1.827	.019	WARWICKITE (Mg, Fe ⁺³ , Ti, Al) ₂ (BO ₃) ₂	55°	X = c Y = $\frac{c}{d}$ el cTV neg	ORTH pris	100 perf	Dark brown to black	H 3.5-4 G 3.35 infus	Dec by H ₂ SO ₄ . Pleoc, X and Z dark brown, Y red-brown.
1.81	1.815	~1.85	.04		CORNWALLITE Cu ₅ (AsO ₄) ₂ (OH) ₄ •H ₂ O	Small	---	MCL fib	Conch	Blackish-green	H 4.5 G 4.5 F 2-2.5	Diss by acids.

1.806	1.817	1.818	.004	CERITE (Ce,Ca) ₉ (Mg,Fe)Si ₇ (O,OH,F) ₂₈	Small r < v very str	---	TRIG	Uneven	Clove- brown, reddish- gray	H 5.5 G 4.8-4.9 infus	Gel with acids.			
1.715	1.820	1.880	.165	DOLEROPHANITE Cu ₂ (SO ₄) ₀	85° (110+1°) r > v very str crossed	Y = b Z:c = -10°	MCL	T01 perf	Chestnut- to dark brown	H 3 G 4.17 fus	Diss by acids. Pleoc, X deep brown, Y brownish-yellow, Z lemon-yellow. Opt char neg?			
1.804	1.81	1.82	.09	AENIGMATITE Na ₂ Fe ₅ TiSi ₆ 20	30-50°	Z:c = 40°	TCL pris	010, 100	Black	H 5.5 G 3.8 F 3	Insol acids. Tw p1 T10. Pleoc, X yellow-brown, Y red- brown, Z dark brown to black.			
1.85	1.81	---	.03	DEERITE (Fe ⁺² ,Mn) ₆ (Fe ⁺³ ,Al) ₃ Si ₆ 20(OH) ₅	---	Z = c el pos	MCL acic	110 good	Black	G 3.84	Opt char not stated.			
1.813	1.820	1.830	.017	ANDREWITE (Cu,Fe)Fe ⁺³ ₃ (PO ₄) ₃ (OH) ₂	Large r > v extr crossed	---	ORTH fib	2 cl v	Dark-to bluish-green	H 4 G 3.3-3.4	Diss by acids.			
1.809	1.821	1.857	.048	BETPARDALITE CaFe ₂ H ₈ (AsO ₄) ₂ (MoO ₄) ₅ [*] 10H ₂ O	60°	Y = b X:c = 12°	MCL pris	---	Lemon- yellow	H 3 G 3.0 F 2	Diss by HCl. Pleoc, X pale yellow, Y greenish-yellow, Z bluish-green, abs Z > V > X.			
1.801	1.821	(1.821)	1.825	004	HUTTONITE cerian phosphatian (Th,Ce)(SiO ₄ ,PO ₄)	21° r < v wk	Y = b	MCL	---	Yellow to reddish- brown	H 5 G 5.06 infus	(Solid soln with Monazite.)		
1.900	1.799	1.822	1.855	.056	AZPROLITE (Mg,Fe ⁺²) ₂ (Fe ⁺³ ,Ti,Mg) BO ₅	(81+14°)	X = b Y = a	ORTH pris	010 good 001 less so, conch	Black	H 5.5 G 3.63	Diss by HCl. Pleoc, X pale green, Y dark green, Z reddish- brown.		
1.79	1.806	1.827	2.005	.178	FERRIMOLYBDITE Fe ₂ (MoO ₄) ₃ •7-10H ₂ O	Small (41+2°) r < v dist	Y = a Z = c	ORTH fib	001 dist	Sulfur- yellow	H 1-2 G 2.99 F easy	Diss by HCl. Pleoc, X and Y pale yellow, Z yellow-green.		
1.880	1.82	1.83	1.88	.06	ROCKBRIDGEITE (Fe ⁺² ,Mn)Fe ⁺³ ₄ (PO ₄) ₃ (OH) ₅	Med r > v very str	X = c	ORTH fib	100 perf 010 fair	Dark green	H 4.5 G 3.51 F easy	Diss by HCl. Pleoc, X yellow-brown, Y bluish-green, Z deep blue, abs Z > X > Y.		
1.805	1.825	1.832	1.940	.115	LUDWIGITE (Mg,Fe) ₂ Fe ⁺³ BO ₅	25-40° r < v str	X = a Y = b	ORTH	Uneven, parting 001	Greenish- black	H 5 G 3.9 F diff	Diss by acids. Pleoc, X blue-green, Y dark green, Z dark brown, abs Z > X > Y.		

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2N_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
---	1.833	---	wk	RUSAKOVITE $(\text{Fe}, \text{Al})_5(\text{VO}_4, \text{PO}_4)_2$ $(\text{OH})_9 \cdot 3\text{H}_2\text{O}$	$\sim 50^\circ$ $r > v$ str	---	Cryptocryst	---	Yellow-orange	H 1.5-2 G 2.76	Diss by acids. Opt char unk.	
2.02	1.820	1.835	1.920	.100	VOLBORTHITE $\text{Cu}_3(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	63° $(47+3^\circ)$ $r < v$ str	---	MCL	001 perf	Yellow-green	H 3-4 G 3.4-3.6	Diss by acids. Pleoc, X and Y yellow, Z yellow-green.
1.813 1.855	1.832	1.837	1.890	.058	DUFRENITE $\text{Fe}^{+2}\text{Fe}^{+3}(\text{PO}_4)_3(\text{OH})_5 \cdot 2\text{H}_2\text{O}$	Small $r < v$ extr	$Z = \frac{b}{e} \text{ clv neg}$	MCL fib	010 perf	Dark green to red-brown	H 3.5-4 G 3.2-3.4 F 2.5	Diss by acids. Pleoc str, X deep blue-green, Y yellow-brown, Z deep olive-brown.
1.792	1.840	1.888	.096	LAUTARITE $\text{Ca}(\text{IO}_3)_2$	$\sim 90^\circ$ $r > v$ mod	$Y = \frac{b}{e} = 25^\circ$	MCL pris	011 good	Col's to light yellow	H 3.5-4 G 4.59 F 1.5	Slightly sol in H_2O , diss by HCl.	
1.833	1.802	1.840	1.888	.086	ORTHOERICSONITE $\text{BaMn}_2(\text{Fe}^{+3}\text{O})\text{Si}_2\text{O}_7(\text{OH})$	$\sim 50^\circ$ $(85+3^\circ)$ $r > v$ str	$X = \frac{b}{e} = 25^\circ$ $Y = \frac{c}{e} \text{ clv neg}$	0RTH	100 perf	Brownish-black	H 5.5 G 4.22	Pleoc, X yellow-brown, Y reddish brown, Z dark brown, abs Z > Y > X.
1.842	1.842	1.848	.006	PARATACAMITE $\text{Cu}_2(\text{OH})_3\text{Cl}$	$0-50^\circ$ $r > v$	---	TRIG	10Tl good	Dark green	H 3 G 3.74	Diss by acids. Poly tw 10Tl.	
---	~1.84	---	---	ITOITE $\text{Pb}_3\text{Ge}(\text{SO}_4)_2\text{O}_2(\text{OH})_2$	---	---	0RTH acic	---	White	G 6.67 calc	Opt char unk.	
1.840	1.847	1.892	.052	LAUBMANNITE $\text{Fe}^{+2}\text{Fe}^{+3}(\text{PO}_4)_4(\text{OH})_{12}$	Med $r < v$ extr	---	0RTH fib	---	Gray-green, yellow-green	H 3.5 G 3.33	Diss by acids. Pleoc, X buff, Y greenish-brown to olive-green, Z reddish-brown, abs Z > Y > X.	
1.840	1.840	1.870	.030	DEERITE $(\text{Fe}^{+2}, \text{Mn})_6(\text{Fe}^{+3}, \text{Al})_3$ $9\text{H}_2\text{Si}_6\text{O}_{20}(\text{OH})_5$	---	$Z = \frac{c}{e} \text{ pos}$	MCL acic	110 good	Black	G 3.84	Opt char unk.	
1.85	1.85	1.88	.03	PARWELITE $(\text{Mn}, \text{Mg})_5\text{Sb}(\text{As}, \text{S})_2\text{O}_{12}$	27° $r > v$ str	---	MCL pris	010 fair	Yellowish brown	H 5.5 G 4.62	Insol in dil HCl.	

v 1.89	1.85 <u>1.85</u>	1.87 .02	DUMONTITE $Pb_2(OH)_3(PO_4)_2(OH)_4 \cdot 3H_2O$	Small $r < v$ str	$X = \frac{b}{c}$ $Y = \frac{b}{c}$	MCL ---	---	Yellow	G 5.7	Diss by acids.			
1.812 ^	1.845 <u>1.852</u>	1.878 .033	TOERNEBOHMITE $(Ce, La)_2AlSi_2O_8(OH)$	20-40° $r < v$ dist	---	MCL ps hex	---	Light green to olive	H 4.5 G 4.94	Slowly diss by hot conc acids. Pleoc, X rose to greenish-yellow, Y bluish-green, Z rose.			
1.854 neg ^	1.820 <u>1.854</u>	1.888 .068	LIEBENBERGITE (011 vine grp) $(Ni, Mg, Fe, Co)_2SiO_4$	92° $r < v$	$X = \frac{b}{c}$ $Z = \frac{a}{a}$	ORTH	010 poor to fair 100 poor	Yellowish green	H 6-6.5 G 4.60	Pleoc, X and Y yellow-green, Z emerald-green.			
1.852	(1.854)	1.867 .015	LUDDENLITE $Pb_2Cu_2Si_5O_14 \cdot 14H_2O$	40°	---	MCL	---	Green	H 4 G 4.45 F easy	Diss by acids. Pleoc str, X pale yellow-brown, Y yellow-brown, Z red-brown.			
1.845 ^	1.855 <u>1.855</u>	1.890 .045	DUFRENITE $Fe^{+2}Fe^{+3}_4(Po_4)_3(OH)_5 \cdot 2H_2O$	Small (57+6°) $r < v$ str	$Z = \frac{b}{c}$ $e1 \bar{c}lv$ neg fib	MCL ps hex fib	010 perf	Reddish-brown	H 3.5-4 G 3.2-3.4 F 2.5	Diss by acids. Pleoc str, X pale yellow-brown, Y yellow-brown, Z red-brown.			
1.837 ^	---	1.888 --	ANANDITE (Mica grp) $(Ba, K)(Fe, Mg)_3(Si, Al, Fe)_4O_{10}(0, OH)_2$	Disp str	$Y = \frac{b}{c}$ $Z:a = 12^\circ$ $e1 \bar{c}lv$ pos plates	MCL ps hex plates	001 perf	Black	H 3-4 G 3.94	Nearly opaque. Pleoc, Y green, Z brown.			
1.85 ^	1.86 <u>1.86</u>	1.92 .07	PURPURITE $(Mn^{+3}, Fe^{+3})PO_4$	Med	$X = \frac{a}{a}$	ORTH	100 good 010 less so	Deep red or purple	H 4-4.5 G 3.4 F easy	Diss by acids. Pleoc str, X dark brownish-gray, Y scarlet, Z purple.			
1.92	1.835 ---	1.910 .075	DEMESMAEKERITE $Pb_2Cu_5(UO_2)_2(SeO_3)_6(OH)_6 \cdot 2H_2O$	---	---	TCL $e1 \bar{l}$ 100	---	Bottle-green	H 3-4 G 5.28	Diss by dilute HNO ₃ .			
1.751 ^	1.851 <u>1.88</u>	1.894 .043	SYNADELPHITE $(Mn, Mg, Ca, Pb)_9(AsO_3)_6(AsO_4)_2(OH)_9 \cdot 2H_2O$	40° $r > v$	---	TCL ps orth	010 imperf	Cols to red-brown	H 4.5 G 3.8 F 2	Diss by HNO ₃ . Pleoc, X light brown, Y dark reddish-brown.			
1.832 ^	1.850 <u>1.865</u>	1.985 .135	LUDWIGITE $(Mg, Fe)_2Fe^{+3}B_0_5$	25-40° $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{d}$	ORTH	Uneven, parting 001	Greenish-black	H 5 G 4.0 F diff	Nearly opaque, X and Y dark green, Z brownish-black.			
1.907 ^	1.843 <u>1.870</u>	1.943 .100	TITANITE $(Ca, Y)TiSiO_4(O, OH)$	38° $r > v$ str	$Y = \frac{b}{c}$ $Z:c = 40^\circ$	MCL wedges	110 good	Yellow to reddish-brown	H 4.5 G 3.56 F 3	Dec by hot H ₂ O ₄ . Pleoc, X pale yellow, Y pale greenish, Z pink to reddish-brown.			

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.817	<u>1.879</u>	2.057	.240	UVANITE (U,Ca) ₂ V ₆ O ₂₁ ·15H ₂ O (?)	52° (66+1°)	---	0RTH(?)	2 clv-pina-coidal	Brownish-yellow	---	Diss by NH ₄ OH. Pleoc, X light brown to yellow, Y dark brown, Z greenish yellow. Not fluor in UV.
1.841	(~1.88)	1.935	.094	PERETTAITE CaSb ₄ O ₄ (OH) ₂ ·2H ₂ O	Large	---	---	---	Col	6 4.06	---
1.87	<u>1.880</u>	1.98	.11	LEITEITE ZnAs ₂ O ₄	26° r < v str	$Y = \frac{b}{c} = 10^\circ$	MCL	100 perf	Col to brown	6 4.61	Lam flexible, ineastic.
1.877	<u>1.880</u>	1.935	.058	KASOLITE Pb(UO ₂)SiO ₄ ·H ₂ O	43° (27+8°)	$X = \frac{b}{c} = 1^\circ$ el clv neg	MCL prs	001 perf 100, 010 indist	Ocher-yellow to amber	H 4.5 G 5.8-6.5	Gel with acids. Pleoc wk or absent.
1.910	---	---	---	CHENEVIXITE Cu ₂ Fe ₂ (AsO ₄) ₂ (OH) ₄ ·H ₂ O	---	$X = \frac{b}{c}$ $Z \sim \frac{c}{2}$	MCL cryptocryst	---	Dark green to greenish-yellow	H 3.5-4.5 G 3.93 F 2.5	Diss by acids.
1.96	1.88	---	high	ROCKBRIDGEITE (Fe,Mn)Fe ⁺³ ₄ (PO ₄) ₃ (OH) ₅	Med r > v very str	$X = \underline{c}$	0RTH fib	100 perf 010 fair	Dark green	H 4.5 G 3.5	Diss by HCl. Pleoc, X yellow-brown, Y bluish-green, Z dark bluish-green.
1.83	<u>1.880</u>	1.897	.022	STIBIVANITE Sb ⁺³ ₂ V ⁺⁴ ₀ ₅	85° r > v str	$X = \frac{b}{c}$ $Z = \underline{c}$	MCL fib	---	Yellow-green	---	Opt sign unk. Pleoc, X and Y emerald-green, Z olive-green.
<	>1.87	(1.88)	<1.89	RAUVITE Ca(UO ₂) ₂ V ₁₀ O ₂₈ ·16H ₂ O	---	---	Mass	---	Purplish-black	---	Opt char unk.
<	---	1.88	---	ANGLE SITE (Barite grp) PbSO ₄	68-75° r < v str	$Y = \frac{b}{c}$ $X = \underline{c}$	0RTH	001 good 210 dist	Col	H 2.5-3 G 6.38 F 1.5	Slowly diss by HNO ₃ .
(1.697)	1.877	<u>1.883</u>	1.894	DUMONTITE Pb ₂ (UO ₂) ₃ (PO ₄) ₂ (OH) ₄ ·3H ₂ O	Large r < v str	$X = \frac{b}{c}$ $Y = \underline{c}$	MCL	---	Yellow	G 5.7	Diss by acids.
<	1.88	<u>1.89</u>	1.90	DENNINGITE (Mn,Zn)Te ₂ O ₅	0-15°	---	TET mass	001 perf	Pale green	H 4 G 5.05 F easy	Diss by cold HCl, insol in HNO ₃ .

1.882 <u>(1.895)</u>	1.915	.033	HALLIMONDITE $Pb_2(UO_2)(AsO_4)_2$	80° r > v	$Z':c = 11^\circ$ on $\overline{1}00$, $X':b = 9^\circ$ on $\overline{0}01$	TCL tab	Conch	Yellow	H ~ 3	Diss by HNO_3 .
1.888	1.895	1.915	.027	Unknown	62° r > v str	---	---	Pale green	---	Labeled "scorodite," Kiura, Japan. Larsen (1921).
1.898 <u>(1.898)</u>	1.915	.017	HUEGELITE $Pb_2(UO_2)_3(AsO_4)_2(OH)_4 \cdot 3H_2O$	Small	---	MCL tab	100 good	Orange-yellow to brown	---	Pleoc, X yellow, Y orange-yellow, Z cols to pale yellow. Anom interf colors.
1.898 <u>(1.899)</u>	1.915	.017	HUEGELITE $Pb_2(UO_2)_3(AsO_4)_2(OH)_4 \cdot 3H_2O$	25°	---	MCL	100 very good, conch	Brown to orange	H 2.5	Pleoc, X yellow, Y yellow to orange, Z cols to pale yellow.
<1.89	<u>1.90</u>	1.95	>.06	LAMMERITE $Cu_3(AsO_4)_2$	$54+5^\circ$ $X = \frac{b}{c}$ $Z:c = 40^\circ$	MCL tab	010 perf 100 good	Dark green	H 3.5-4 G 5.18	Pleoc, X bright blue, Y sky blue, Z bluish-green.
1.810	<u>1.900</u>	>2.01	>.20	DUTTONITE $VO(OH)_2$	60° r < v mod	$X = \frac{a}{c}$ $Y = \frac{a}{c}$	MCL	---	H 2.5 G 3.24	Pleoc, X pale pinkish-brown, Z pale brown.
1.890	<u>1.90</u>	1.99	.10	BELLINGERITE $Cu_3(TiO_3)_6 \cdot 2H_2O$	Med r > v str	---	TCL pinacoidal	Subconch	H 4 G 4.89	Diss by HCl. Gives iodine vapor when heated in closed tube.
(1.821) <u>1.898</u>	1.900	1.922	.024	HUTTONITE $ThSiO_4$	25° r < v mod	$Y = \frac{b}{c}$ $Z \sim \frac{a}{c}$	MCL	---	G 7.1 infus	---
1.899 <u>(1.901)</u>	1.903	.004	QUEITITE $Pb_4Zn^2(SO_4)(SiO_4)$ (Si_2O_7)	$\sim 90^\circ$ r < v str	$X = \frac{b}{c}$ $Z:a \text{ varies}$	MCL tab	---	Col to pale cream	H 4 G (6.07)	Diff diss by HNO_3 . $Z:a = 40^\circ$ at 405 nm, 0° at 592 nm.
1.870 <u>1.907</u> 1.970	1.907	2.034	.134	TITANITE $CaTiSiO_5$	27° r > v str	$Y = \frac{b}{c}$ $Z:c = 40-50^\circ$	MCL	110 good	H 5 G 3.55 F 4	Dec by hot H_2SO_4 . Pleoc, X pale yellow, Y greenish, Z red to brown.
1.85	<u>1.91</u>	(2.20)	.35	SPIROFFITE $(Mn,Zn)_2Te_3O_8$	55°	---	MCL mass	2 clv, conch	H 3.5 G 5.01 F easy	Diss by HCl.
1.890	<u>1.91</u>	1.977	.087	SCHULLENITE $Pb(HAsO_4)_2$	58° disp str	$X = \frac{b}{c}$ $Z:c = 66^\circ$ el pos	MCL thin plates, 010	010 good	H 2.5 G 5.94	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.380 \wedge	1.395	1.910	1.950	.055	KASOLITE $Pb(UO_2)SiO_4 \cdot H_2O$	43° ($64+5^\circ$)	$X = b$ $Z:c \approx 1^\circ$ $\epsilon\perp clv neg$	MCL pris	001 perf 100, 010 indist	Ocher-yellow to amber-brown	H 4.5 G 5.8-6.5	Gel with acids. Pleoc wk or absent.
---	>1.9 <2.0	---	---	.110	PAULMOOREITE $Pb_2As^{+3}O_5$	65° $r > v$ very str	$Y = b$ $Z:c \approx 10^\circ$ $\epsilon\perp clv neg$	MCL	001 perf	Cols to light orange	H ~ 3 G 7.0:Luster adamantine.	Diss by 1:1 HNO_3 .
1.91	1.91	1.945	.035		GANOMALITE $Pb_6Ca_4Si_6O_{21}(OH)_2$	Small	---	HEX tab 0001	1010 perf 0001 less so	Gray	H 3 G 5.74 F 3	Gel with HNO_3 .
1.871	1.92	2.01	.139		CLAUDETITE As_2O_3	58° $r < v$ str	$Y = b$ $Z:c \approx 5^\circ$	MCL thin tab 010	010 perf	Cols	H 2.5 G 4.15 F volat	Slightly sol in hot H_2O . Tw on 100, penet.
1.89	1.92	1.95	.06		TSUMEITE $Pb_2Cu(Po_4)(SO_4)(OH)$	88° $r < v$ str	---	MCL tab	Uneven	Emerald-green	H 3.5 G 6.1 F fus	Diss by HNO_3 . Pleoc, X and Y pale blue, Z robin's egg blue.
1.96 \wedge	---	1.92	---	.04	PURPURITE $(Mn^{+3}, Fe^{+3})Po_4$	38°	$X = a$	ORTH	100 good 010 less so	Deep red or purple	H 4-4.5 G 3.2-3.4 F easy	Diss by acids. Pleoc str, X dark brownish-gray, Y scarlet, Z purple.
1.920 \wedge	(1.921)	1.943	.023		OIJISUMITE $PbGe_4O_9$	20°	$X:c = 3-5^\circ$	TCL	001 in traces	Cols to white	H 3 G (5.77)	Diss by hot HNO_3 .
1.880	1.928	2.029	.149		CESBRONITE $Cu_5(TeO_3)_2(OH)_6 \cdot 2H_2O$	72° $r > v$ mod	$X = a$ $Y = c$	ORTH el a	021 good 010 poor	Green	H 3 G 4.45 F easy	Diss by acids. Pleoc, X pale bluish-green, Y yellow-green, Z deep emerald-green.
[]	~1.93	~1.93	1.97	.04	ZIRCON $ZrSiO_4$	0-10°	---	TET pris	110	Brown, pink, cols	H 7.5 G 4.7 infus	Insol in acids. Abnorm biax.
1.920	1.960	2.20	.28		GRAEMITE $CuTeO_3 \cdot H_2O$	48°	---	ORTH	010 good parting 100	Blue-green	H 3-3.5 G 4.13 F easy	Diss by acids. Pleoc, X yellow-green, Y and Z blue-green.
1.88 \wedge	1.92	1.96	2.04	.12	CHENEVIXITE $Cu_2^2Fe_2(AsO_4)_2(OH)_4 \cdot 4H_2O$	($73+10^\circ$)	$X = b$ $Z \sim c$	MCL	---	Dark green	H 4 G 3.5 F 2.5	Diss by acids.

1.963	<u>1.963</u>	1.966	.003	HYALOTEKITE $PbBa_2Ca_2B_2(Si,Be)_2$ $Si_8O_{28}F$	Small $r < v$ str	Opt pl \perp c_{1v}	TCI 0RTH ps hex	2 cly at 90°	Col's H 5-5.5 G 3.80 F 3 (?)	Insol in acids.
1.952	---	2.002	.052	KURANAKHITE $PbMn^+4Te^+6O_6$	---	---	---	Brown to black	H 4-5	Diss by HCl with evolution of chlorine. Opt sign unk.
1.950	<u>1.970</u>	2.092	.142	TITANITE Ca_3Si_5	18° (46+2°) $r > v$ str	$Y = \frac{b}{c}$ $Z: \frac{c}{c} = 40-$ $-50°$	MCL	110 good	Yellow, brown, cols	H 5 G 3.52 F 4
1.907	<u>1.97</u>	1.99	.04	BAYDONITE $Pb(Cu,Zn)_3(AsO_4)_2(OH)_2$	Large $r < v$ str	$X = \frac{b}{c}$ $Y: \frac{e}{e} = 45°$	MCL fib c	---	Grass- to dark-green	H 4.5 G 5.6
1.95	<u>1.98</u>	2.06	.10	URANOSPHARITE $Bi_2U_0g \cdot 3H_2O$	56° $r < v$ str	$Y = \frac{b}{c}$ $Z = \frac{c}{c}$	MCL el c	100	Reddish-orange, yellow	H 2-3 G 6.9
1.96	<u>1.98</u>	2.06	.10	URANOSPHARITE $Bi_2U_0g \cdot 3H_2O$	Large $r > v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	0RTH	010, 001	Greenish-yellow	H 3-4 G 3.75 F 2 (?)
2.00	<u>2.01</u>	2.02	.02	CALCIOVOLBORTHITE $CaCuV_4(OH)$	Large	---	TCL	One perf	Dull yellow	Soft F easy
2.05	<u>2.05</u>	2.04	.02	METAROSSITE $CaV_2O_6 \cdot 2H_2O$	Large $r < v$ str	---	MCL	001 perf 010	Dark green to yellow-green	H 3-4 G 3.42-3.62
1.84	<u>>1.84</u>	>>1.84	high	VOLBORTHITE $Cu_3(VO_4)_2 \cdot 3H_2O$	Large $r < v$ str	---	0RTH prts	001, 110, 111 im-perf	Yellow	H 2 G 2.07 F 1
2.01	<u>2.02</u>	2.06	.05	SULFUR	69° $r < v$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	0RTH prts	Uneven subconch	Dark yellow, brown, black	H 4-5 G 4.7-4.9 infus
1.958	<u>2.037</u>	2.245	.287	FERSMITE $(Ca,Ce,Na)(Nb,Ta,Ti)_2$ $(O,OH,F)_6$	Med (82+15°)	---	0RTH	---	Nearly insol in acids.	Pleoc, red-brown to yellow.
1.98	<u>2.040</u>	2.13	.15	GAMAGARITE $Ba_4(Fe,Mn)_2V_4O_{15}(OH)_2$	47-62° $r < v$ str	$Y = \frac{b}{c}$	MCL	2 cly	Dark brown	H 4-5-5 G 4.63
2.016	<u>2.040</u>	2.130	.114	MANGANITE $MnO(OH)$	Small (71+5°) $r > v$ str	$X \sim \frac{a}{b}$ $Y \sim \frac{b}{c}$	MCL pris	010 perf 110, 001 good	Steel-gray to iron-black	Diss by HCl. Pleoc wk, abs Z > X and Y.
1.90	<u>2.05</u>	2.50	.60	FERNANDINITE $CaV^{+4}V^{+5} \cdot 10O_{30} \cdot$ $14H_2O$ (?)	---	---	Cryptocryst	---	Dull green	---
2.25	---	<u>2.05</u>	---	---	---	---	---	---	---	Opt char unk.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula	$\frac{2V_z}{(2V_z \text{ calc})}$	Optical orientation ($2V_z$ calc) disp	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ								
2.01	2.01	<u>2.05</u>	2.09	.08	CALCIOVO-BORTHITE $\text{CaCu}_4(\text{OH})_4$	83° r > v str	X = $\frac{a}{b}$ Y = $\frac{d}{e}$	0RTH	010, 001	Greenish-yellow	H 3-4 G 3.75 F 2 (?)
	2.05	<u>2.05</u>	2.07	.02	CARMINITE $\text{PbFe}_2(\text{AsO}_4)_2(\text{OH})_2$	Med r < v str	X = c	0RTH acic	Pri's	Carmine to lilac	H 3.5 G 5.22 F easy
2.07					LUDLOCKITE $(\text{Fe}^{+2}, \text{Pb})\text{As}_2\text{O}_6$	---	Z ~ $\frac{a}{e}$ el pos	TCL tw	011 perf	Red	H 1.5-2 G 4.40
1.96	<u>2.055</u>		>2.11	>.15	GERSTLEYITE $\text{Na}_2(\text{Sb}, \text{As})_8\text{S}_{13} \cdot 2\text{H}_2\text{O}$	Large (?)	---	MCL	010, T01 perf	Cinnabar-blackish-red	H 2.5 G 3.62 F 2
					SUNDIVSITE $\text{Pb}_{10}(\text{SO}_4)\text{Cl}_2\text{O}_8$	---	---	MCL	100 perf	Col's to white	H 3 G 7.0
					SENAHMONTITE Sb_2O_3	---	---	CUB oct	111 in traces	Col's	H 2 G 5.50 F 1.5 volat
					RYNERSONITE $\text{Ca}(\text{Ta}, \text{Nb})_2\text{O}_6$	---	X = c Y = $\frac{d}{e}$	0RTH fib	Uneven	Creamy white to pink	H ~ 4.5 G 6.40
						---	---	0RTH	One dist	Pale yellow	H 3 G 7.35
						80°	---	0RTH	Uneven, subconch	Brown to black	H 4-5 G 4.7-4.9 infus
							---	0RTH	---	---	Nearly insol in acids.
2.040					BLIXITE $\text{Pb}_2\text{Cl}(\text{O}, \text{OH})_2$						Diss by HNO_3 .
					FERSMITE $(\text{Ca}, \text{Ce}, \text{Na})(\text{Nb}, \text{Ta}, \text{Ti})_2(\text{O}, \text{OH}, \text{F})_6$	Med					
					DUHAMELITE $\text{Pb}_2\text{Cu}_4\text{Bi}(\text{VO}_4)(\text{OH})_3 \cdot 8\text{H}_2\text{O}$	---	Z:c = 3°	0RTH	---	Yellow-green	Opt. sign unk. wk in yellow. Diss by HNO_3 .
					KHINITE $\text{PbCu}_3\text{Te}^{+6}\text{O}_4(\text{OH})_6$	20°	---	0RTH	001 fair	Dark green	H 3.5 G 6.7
2.08	<u>2.11</u>										
2.110	<u>2.112</u>			.055							

2.10	---	2.30	.20	ROOSEVELTITE BiAsO ₄	---	MCL	Conch	Gray to yellow	H 4-4.5 G 6.9-7.1 F 2-3	Diss by acids. Opt char unk.	
>2.11	---	<2.13	<.02	SCHNEIDERHOEHNITE Fe ⁺² As ⁺³ 10 ^{0.23}	---	TCL	100 perf 2 others	Dark brown	H 3 G 4.3	Pleoc, reddish-brown to bright yellow, abs X < Y < Z.	
2.115	<u>2.135</u>	2.26	.125	RAJITE CuTe ⁺⁴ 2 ^{0.5}	40° disp wk	MCL	010	Green	H 4 G 5.75 fus	Diss by cold acids. Pleoc wk in green, abs Z > Y > X.	
2.01	---	---	---	SANTAFEITE Na ₂ (Mn,Ca,Sr) ₆ Mn ⁺⁴ ₃ (V,As) ₆ O ₂₈ •8H ₂ O	---	X = <u>C</u>	010 perf 110 dist	Brownish-black	G 3.38 F easy	Diss by HCl. Pleoc yellow-brown to dark red-brown, abs X > Y > Z. Opt char unk.	
2.14	---	2.15	.01	EZTLITE Pb ₂ Fe ⁺³ 6(Te ⁺⁴ O ₃) ₃ (Te ⁺⁶ O ₆)(OH)10•8H ₂ O	---	Z:c = 3°	MCL	001	Deep orange	---	Sign unk. Not pleoc.
---	<u>2.15</u>	2.15	---	CUPROTUNGSTITE Cu ₂ (WO ₄)(OH) ₂	---	Crypt-octyst	---	Green	Fus	Dec by HCl. Opt char unk.	
2.14	<u>2.15</u>	2.18	.04	ATELESTITE Bi ₈ (AsO ₄) ₃ 0 ₅ (OH) ₅	44° r < v str	MCL	010 indist	Sulfur-yellow	H 4.5-5 G 6.8-7.1 F 1.5	Diss by HCl.	
>2.09	>2.09	>2.09	---	MOUNANAITE PbFe ₂ (VO ₄) ₂ (OH) ₂	---	TCL el c	---	Brownish-red	G 4.85	Tw common. Pleoc brownish-red to brownish-yellow. Opt char unk.	
2.130	<u>2.16</u>	2.195	.065	PREISINGERITE Bi ₃ (AsO ₄) ₂ 0(OH)	90°	TCL tab	---	White to gray	G (7.24)	---	
2.12	<u>2.17</u>	2.31	.19	MELANOTEKITE Pb ₂ Fe ₂ Si ₂ 9	67° r > v str	ORTH pris	2 clv	Black	H 6.5 G 5.73 F 2-2.5	Dec by HNO ₃ . Pleoc str, X nearly col, Y pale red-brown, Z deep reddish-brown.	
2.15	<u>2.17</u>	2.25	.10	MANGANOTANTALITE Mn(Ta,Nd) ₂ 6	r < v disp extr	X = <u>a</u> Y = <u>b</u> Z = <u>c</u>	010 perf	Deep red to black	H 5-6 G 7.5 infus	Insol in acids. Pleoc, X and Y deep red, Z pink.	
2.17	<u>2.17</u>	2.18	.01	GEORGIADESITE Pb ₈ (AsO ₄) ₂ C ₁₇ (OH) ₃	Large r < v str	MCL	---	White	H 3.5 G 7.1 F easy	Diss by HNO ₃ .	
2.11	<u>2.18</u>	2.22	.11	BALYAKINITE CuTeO ₃	80°	X = <u>a</u> Y = <u>b</u> Z = <u>c</u>	ORTH	Gray-green to bluish-green	G 5.6	---	
2.14	---	2.24	.10	FORNACITITE (Pb,Cu) ₃ [Cr,As]O ₄] ₂ (OH)	---	MCL	---	Olive-green	G 6.2	Slightly pleoc. Opt char uncertain.	

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.00 Li	<u>2.18</u> _L i	2.35 Li	.35 TELLURITE TeO_2	~ 90° r < v med	$X = \frac{b}{c}$ $Z = \frac{c}{a}$ el pos	ORTH acic \underline{c}	010 perf	Col	H 2 G 5.90 F 2	Diss by HCl.
2.14	---	2.23	.09 WODGINITE (Ta, Nb, Sn, Mn, Fe) ₁₆ O ₃₂	---	Z: \underline{c} = 26°	MCL	---	Red-brown	H 6 G 7.6 infus	Insol in acids. Pleoc, light yellow to reddish-brown.
2.14	---	2.315	.18 VIGEZZITE (Ca, Ce)(Nb, Ta, Ti) ₂ O ₆	Large	$X = \frac{c}{a}$	ORTH prism	100 dist	Orange-yellow	G (5.34)	Sign unk. Not pleoc.
1.94	<u>2.20</u>	2.51	.57 LEPIDOCROCITE $\text{FeO}(\text{OH})$	83° (96+4°)	$X = \frac{b}{c}$ $Y = \frac{c}{a}$	ORTH blades 100	010 perf 100 less 010 fair	Red	H 5 G 4.09 infus	Slowly diss by HCl. Pleoc str, X yellow, Y red, Z deeper red.
2.13	<u>2.20</u>	2.40	.27 ANGELELLITE $\text{Fe}_4(\text{AsO}_4)_2\text{O}_3$	Med large	---	TCL	001	Blackish-brown	H 5.5 G 4.86	Pleoc str, yellow to deep red.
2.10	<u>2.20</u>	2.31	.21 KENTROLITE $\text{Pb}_2\text{Mn}_2\text{Si}_2\text{O}_9$	88° r < v str	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH pris	110 dist	Dark reddish-brown	H 5 G 6.19 F 2-2.5	Dec by HNO_3 . Pleoc, X pink, Z brownish-red, abs Z > Y > X.
2.19	<u>2.20</u>	2.33	.14 TRIPHYHITE FeSb_2O_6	Small r < v very str	---	TET mass	---	Greenish-yellow, yellow-brown	H ~ 6 G 5.6-5.8 F 4-5 (?)	Insol in acids.
2.18	(<u>2.21</u>)	2.26	.08 EUXENITE (Y, Ca, Ce, U, Th) ₂ O ₆	70°	$Y = \frac{a}{b}$ $Z = \frac{c}{d}$	ORTH	Conch	Brownish-black	H 5-5.5 G 5.0-5.4 infus	Dec by not concd acids.
2.17 2.32	<u>2.21</u>	2.25	.08 MANGANOTANTALITE (Mn, Fe)(Ta, Nb) ₂ O ₆	Large r < v str	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH pris	101 dist 100 poor	Red to reddish-brown	H 6 G 7.52 infus	Insol in acids. Pleoc, X and Y deep red, Z pink, abs X > Y > Z.
2.30	2.19	<u>2.21</u>	.06 AESCHYNITE-(Y) (Y, Ca, Fe, Th) ₂ (Ti, Nb) ₆ (OH) ₆	74°	el neg	ORTH	100, 010, 001 perf uneven	Yellow	H 5 G 4.4 infus	Insol in HCl, diss by H_2SO_4 .
2.30	---	<u>2.21</u>	---	low	---	CUB oct	111 imperf fr uneven	Yellow	H 6.5 G 5.0 infus	Insol in acids. Abnorm violet interf colors.
2.185	<u>2.219</u>	2.266	.081 HEVITE $\text{Pb}_5\text{Fe}^{+2}(\text{VO}_4)_2\text{O}_4$	89° (82+3°)	$Y = \frac{b}{c}$ $X:\underline{c} = 36^\circ$	MCL	Uneven	Yellow-orange	H 4 G 6.3	Tw 110.

2.36	<u>2.22</u>	2.32	.15	HUEBNERITE (Wo,Tramite ser.) (Mn,Fe)W ₄	73°	X = <u>b</u> Z:c = 18° el cIv pos	MCL bladed, tab 100	010 perf	Brownish-red	H 5-5.5 G 7.35 F 4	Diss by HCl with sepn of yellow WO ₃ , Pleoc, X pale yellow, Y yellow-brown, Z green.
2.17	<u>2.22</u>	2.26	.06	COTUNNITE PbCl ₂	67°	X = <u>b</u> Z = <u>c</u> el cIv pos	ORTH acic a	010 perf	White, yellow	H 2.5 G 5.8 F 1	Diss by hot water.
2.20	<u>2.23</u>	2.35	.16	PLUMBOTELLURITE PbTeO ₃	~ 50°	---	ORTH	---	Gray to brown	Soft G 7.2	---
2.19	2.20	---	.07	WODGINITE (Ta,Nb,Sn,Mn,Fe) ₁₆ O ₃₂	---	---	MCL	---	Red-brown, dark brown	H 6 G 7.2-7.4 infus	Insol in acids..
~2.18	~2.20	---	---	MANGANITE MnO(OH)	Small r > v str	X ~ <u>a</u> Y ~ <u>b</u>	MCL pris	010 perf 110, 001 less so	Steel- gray to black, streak brownish	H 4 G 4.33 infus	Diss by HCl • Pleoc wk, abs Z > X and Y.
2.05	<u>2.24</u> _{L1}	2.53	Li .29	OBOYERITE Pb ₆ H ₆ (TeO ₃) ₃ (TeO ₆) ₂ • 2H ₂ O	---	---	TCL fib	---	White	H 1.5 G 6.46	Opt sign unk.
2.24	---	2.26	.02	DESCLOIZITE Pb(Cu,Zn)V ₄ (OH)	~ 90° r < v str	X = <u>c</u> Y = <u>b</u>	ORTH	Conch	Brownish- red, brown, black	H 3.5 G 5.9-6.2 F 1.5	Diss by HNO ₃ . Pleoc, X and Y yellow, Z brownish.
2.18	<u>2.26</u>	2.35	.17	MOTTRAMITE Pb(Cu,Zn)V ₄ (OH)	~ 90° r < v str	X = <u>c</u> Y = <u>b</u>	ORTH	Conch	Reddish- brown, olive- green	H 3.5 G 5.9-6.2 F 1.5	Diss by HNO ₃ . Pleoc, X and Y greenish- yellow, Z brownish- yellow.
2.17	<u>2.26</u>	2.34	.17	SCHUMACHERITE Bi ₃ (VO ₄) ₂ (OH)	Large	X:c = 22°	TCL tab b	Conch	Yellow	H 3 G 6.90	---
2.20	(~2.27)	2.42	.22	MENDIPITE Pb ₃ O ₂ Cl ₂	~ 90° r < v str	X = <u>a</u> Y = <u>b</u> el pos	ORTH fib	110 perf 100, 010 less so	White	H 2.5 G 7.24 F 1	Diss by HNO ₃ .
2.24	<u>2.27</u>	2.31	.07	RASPIITE PbWO ₄	Very small	Y = <u>b</u> Z:c = 30°	MCL tab 001	100 perf	Brownish- yellow	H 2.5-3 G 8.46 F 2.5-3	Dec by HNO ₃ : Z > X and Y.
2.27	<u>2.27</u>	2.30	.03	AESCHYNITE (Ce,Ca,Fe,Th)(Ti,Nb) ₂ (O,OH) ₆	80°	---	ORTH	Conch	Brownish- black	H 5.5 G 5.2 infus	Partly dec by hot acids. Commonly metamict.
2.21	<u>2.28</u> 2.32 neg	2.34	.06	PEROVSKITE (Ca,Ce,Na) ₂ (Ti,Nb) ₂ O ₆	~ 90°	X = <u>a</u> Z = <u>b</u>	ORTH ps cub	100, 110	Brownish- black,	H 5.5-6 G 4.0-4.2	Insol in acids. Pleoc wk, X light green, Z darker green.
2.40	~2.30	---	---	---	---	---	---	---	---	---	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
2.21 \wedge	2.26	2.32	2.43	.17	TANTALITE (Fe,Mn)(Ta,Nb) ₂ 6	Large $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH pris	010 dist 100 less so	Black	H 6 G 6.6-8.0 infus	Insol in acids. Nearby opaque. Pleoc str, X pale yellow, Y red-brown, Z dark red-brown.
2.31 \wedge	2.35	2.40	.09		ASIANITE (Nb,Ta,U,Fe,Mn)4O8	----	$X = \frac{b}{a}$ $Y = \frac{c}{a}$	ORTH	010 perf 010	Brown to black	H 6-6.5	Pleoc, X light brownish-red, Z dark brownish-red.
2.30 L_i	2.35 L_i	2.40 L_i	.10		NADORITE PbSbO ₂ Cl	Large $r > v$ str	$X = \frac{b}{a}$ $Z = \frac{c}{a}$ el pos	ORTH tab	010 perf 010	Brown to yellow	H 3.5-4 G 7.02 F 1.5	Diss by HNO ₃ . Tw on 101.
2.39 (2.36)	2.52	.13			THOREAULITE SnTa ₂ O ₆	30-35° $r > v$ str	$Y = \frac{b}{c} = 58^\circ$ $Z:c = 58^\circ$	MCL acic	100 good 011 less so	Brown	H 5.5-6 G 7.5-7.9	---
2.28 L_i 2.28 Na	2.36 L_i 2.38 Na	2.48 L_i 2.49 Na	.20 .21		BRACKEBUSCHITE Pb ₂ (Mn,Fe)(VO ₄) ₂ •H ₂ O	Large $r > v$ rather str	----	----	----	Dark brown to black	G 6.05 F 2 (?)	Pleoc str, X nearly cols, Y dark reddish- brown, Z clear reddish-brown.
2.22 \wedge	2.31 L_i	2.36 L_i	.15		WOLFRAMITE (Fe,Mn)WO ₄	60-70° $r < v$	$X = \frac{b}{c} = 26^\circ$ el clv pos	MCL tab	100	Brownish-black	H 5-5.5 G 7.4 F 3	Dec by hot HCl with sep of yellow WO ₃ . Nearly opaque. Pleoc str, abs X > Y > Z.
2.31 L_i	2.37 L_i	2.66 L_i	.35		CROCOITE PbCrO ₄	55°(Na) $r > v$ very str., inclined	$Y = \frac{b}{c} = 5^\circ$	MCL	110 dist	Hyacinth-red, orange	H 2.5-3 G 6.0-6.1 F 1.5	---
2.34 L_i 2.38 Na	2.38 L_i 2.44 Na	2.65 L_i 2.65 Na	.31 .27		PHENNICOTHRITE 8Pb ₂ (CrO ₄) ₀	58°(Na) $r > v$ mod	$X = \frac{b}{c} = -2^\circ$ Y:c = 5°	MCL tab	20T good	Cochineal-, hyacinth-red	H 2.5-3 G 7.01 F easy	Diss by HNO ₃ .
2.38 L_i	2.39 L_i	2.42 L_i	.04		PSEUDOBROOKITE Fe ₂ TiO ₅	~ 50° $r < v$	$X = \frac{b}{c}$ $Z = \frac{a}{c}$	ORTH tab	100	Dark brown to black	H 6 G 4.3-4.4 infus	Insol in acids. Pleoc wk in reddish- brown, abs Z > Y > X.
2.37	2.40	2.46	.09		STIBIOTANTALITE Sb(Ta,Nb) ₂ O ₄	75° $r < v$ str	$X = \frac{a}{b}$ $Z = \frac{c}{b}$	ORTH	110 good	Dark brown, yellow- brown	H 5.5 G 6.2-7.9 F 4	Insol in acids. Data for mineral with G 6.8.

2.39	$\frac{2.40}{\text{Li}}$	2.43	.04	BISMUTOTANTALITE $\text{Bi}(\text{Ta}, \text{Nb})_0 4$	80° $r < v$ str	$X = \frac{a}{b}$ $Z = \frac{c}{b}$	110 good	Dark-brown, yellow-brown	H 5 G 8.1-8.8
2.30	$\frac{2.40}{\text{Li}}$	---	low	PEROVSKITE CaTiO_3	~ 90°	$X = \frac{a}{b}$ $Z = \frac{c}{b}$	100, 110	Brownish-black to yellow	H 5-6 G 4.0-4.2 infus
2.40	$\frac{2.42}{\text{Li}}$	2.46	.06	STIBIOCOLUMBITE $\text{Sb}(\text{Nb}, \text{Ta})_0 4$	73° $r > v$ str	$X = \frac{a}{b}$ $Z = \frac{c}{b}$	110 good	Dark brown, yellow-brown	H 5.5 G 5.6-6.6 F 4
>2.42	---	>2.42	---	BISMITE Bi_2O_3	Disp str	---	---	Gray-green, yellow	H 4.5 G 8.6
2.37	$\frac{2.45}{\text{Li}}$	2.65	Li .28	MONTRONDITE HgO	Large	$Z = \frac{a}{b}$ (?) $Y = \frac{c}{b}$	010 perf	Deep red, orange, brown	H 2-3 G 11.2 volat
2.45	$\frac{2.45}{\text{Li}}$	2.51	Li .06	DERBYLITE $\text{Fe}_4\text{Ti}_3\text{Sb}_{13}(\text{OH})$	Small	---	MCL pris	Black	H 5 G 4.53 infus
2.58	$\frac{2.59}{\text{Li}}$	2.70	.12	BROOKITE TiO_2	15-30° (Na), ~ 0° for green, disp very str	$Y = \frac{a}{b}$ $X = \frac{d}{b}$ (red) $X = \frac{c}{b}$ (blue)	ORTH tab 120 indist	Brown to black	H 5.5-6 G 4.14 infus
2.57	$\frac{2.61}{\text{Li}}$	2.71	Li .14	MASSICOT PbO	~ 90° (67+11°) disp str	$Y = \frac{a}{b}$ (?)	ORTH tab 100	010 perf Yellow	Soft G 9.6 F 1.5 calc
				CLINOBISVANITE BiVO_4	Disp str	---	MCL	010 perf Orange to yellow	Opt char unk.
		2.63	---	TENORITE CuO	Large $r < v$	$Y = \frac{b}{c}$ $Z \sim \frac{c}{c}$	MCL laths	TTI, T01 Steel-gray to black	Diss by HCl. T_w common on 011. Pleoc in brown. Opt char unk.
		2.63	---	LORANDITE TiAsS_2	large $r > v$ str	$Z = \frac{b}{c}$ $X \sim \frac{c}{c}$	MCL 100 perf 20T, 001 good	Coch- ineal-red	Diss by HNO_3 . Deep red in thin-section. Pleoc wk, Y purple- red, Z orange-red.
		3.17	---	GETCHELLITE AsSbS_3	80° $r > v$ str crossed	$Z = \frac{b}{c}$ $Y: \underline{a} = 15^\circ$	MCL 001 perf sectile	Dark red G 3.92	---
		>2.72	Li	---	very high				

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
---	$>2.72_i$	---	---	very high	VRBAITE $Tl_4Hg_3Sb_2As_8S_{20}$	Large $r > v$ str	---	ORTH	010 easy	Brownish-gray to reddish	H 3.5 G 5.30 F easy	Diss by HCl.
---	$>2.72_i$	---	---	very high	DUFRENOLITE $Pb_2As_2S_5$	---	---	MCL	010 perf	Lead-gray	H 3 G 5.53 fus	Opt char unk.
---	$>2.72_i$	---	---	very high	MIARGYRITE Ag_3SbS_2	Med	---	MCL	010 traces	Iron-black	H 2-2.5 G 5.2 F 1	Dec by HNO_3 . In thin-section blood-red.
---	---	---	extr	PYROSTILPNITE Ag_3SbS_3	---	$Y = \frac{b}{c} = 8^\circ$ $X : c = \frac{a}{b}$	---	MCL	010 perf	Hyacinth-red	H 2 G 5.94 F 1	Tw pl 100.
2.82 _{Na}	<u>2.93</u> _{Na}	<u>3.14</u> _{Na}	.32	ENARGITE Cu_3AsS_4	($76+8^\circ$)	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH tab 001	110 perf 100, 010 dist	Gray-black, iron-black	---	Tw pl 320.	
2.96 _{Na}	<u>3.09</u> _{Na}	<u>3.42</u> _{Na}	.46	KERMESITE Sb_2S_2O	Med ($70\pm6^\circ$)	$Z = \frac{b}{c}$	TCL pris	001 perf 100 less so	Cherry-red	H 1-1.5 G 4.68 volat	---	
3.14	3.17 (3820 nm)	3.28	.14	BOURNONITE $PbCuSbS_3$	52°	---	ORTH pris tab	010 imperf	Steel-gray	H 2.5-3 G 5.83 fus	Dec by HNO_3 . Tw on 110 common.	

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Table 7. Biaxial negative minerals

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, fusibility	Remarks
1.324	1.3245	1.325	.001 AVOGADRITE (K, Cs)BF ₄	90°	X = $\frac{c}{b}$ Y = $\frac{c}{b}$	ORTH tab	---	Yellow to reddish white	6.2.62	Slightly sol in H ₂ O, diss by acids.
1.393	1.395	1.397	.004 MIRABILITE Na ₂ SO ₄ ·10H ₂ O	76° r < v str crossed	X = $\frac{b}{c}$ Z:c = 30°	MCL pris	100 perf 001, 010, 011	White	H 1.5-2 G 1.49 F 1.5	Sol in H ₂ O, effloresces.
1.407	1.414	1.415	.008 THOMSENOLITE NaCaAlF ₆ H ₂ O	50° r < v wk	Z = $\frac{b}{c}$ X:c = -52°	MCL cube-like	001 perf 110 dist	Col s	H 2.48 G 2.48 F 1.5	Diss by H ₂ SO ₄ .
1.405	1.425	1.440	.035 NATRON Na ₂ CO ₃ ·10H ₂ O	71° (81+7°) r < v perc crossed	X = $\frac{b}{c}$ Z:c = 41°	MCL	001 good 010 im-perf	White	H 1-1.5 G 1.48 F 1	Sol in H ₂ O, effloresces.
1.429	1.433	1.436	.007 JARLITE NaSr ₃ Al ₃ (F, OH) ₁₆ (?)	81-90°	Y = $\frac{b}{c}$ X:c = -6°	MCL tab	---	Col s	H 4-4.5 G 3.78-3.93 F easy	---
1.435						TCL	---	Col s, white	H 2 G 1.58	Sol in H ₂ O.
1.424	1.436	1.438	.012 WILCOXITE MgAl(SO ₄) ₂ F·18H ₂ O	48°	---	MCL fib c	---	White	H 2-2.5 G 1.75 F 1	Sol in H ₂ O.
1.430	1.452	1.458	.028 KALINITE KAl(SO ₄) ₂ ·11H ₂ O (?)	52° disp wk	Z = $\frac{b}{c}$ Y:c = 13°	ORTH pris	011	Col s	H 2-2.5 G 1.74 F 1	Sol in H ₂ O.
1.440	1.453	1.454	.014 LECONTITE (K, NH ₄)Na(SO ₄)·2H ₂ O	30° r < v rather str	Y = $\frac{a}{b}$ Z = $\frac{a}{b}$	ORTH pris	---	White, chalky	H 2 G 2.74	Diss by acids.
1.448	1.454	1.456	.008 GEARKSUTITE CaAl(F, OH) ₅ ·H ₂ O	Small to med	X = $\frac{b}{c}$ Y:c Large el pos	MCL acidic c	---	White, chalky	H 2 G 2.74	Diss by acids.
1.433	1.455	1.461	.028 EPSONITE MgSO ₄ ·7H ₂ O	52° r < v wk	X = $\frac{b}{c}$ Z = $\frac{a}{c}$	ORTH el c	010 good 101 dist	White	H 2 G 1.68 infus	Sol in H ₂ O, taste bitter.
1.470						MCL (?) acidic	---	Col s	G 1.81 F diff	Dec by H ₂ O with sepn of gypsum, diss by acids.
1.435	1.455	1.459	.024 WATTEVILLE Na ₂ Ca(SO ₄) ₂ ·4H ₂ O	48°	---	TCL tab 001	001 perf	White, pearly	H 1 G 1.48	Sol in H ₂ O. Tw pl 001.
1.448 1.461	1.456	1.459	.119 SASSOLITE B(OH) ₃	5-14° b ~ opt pl	X:c = 84-88°	---				

1.452	<u>1.456</u>	1.458	.006	TIKHONENKOVITE SrAlF ₄ (OH)·H ₂ O	70°							---
(1.405)	<u>1.460</u>	1.487	(.082)	ROSTITE Al ₂ SO ₄ (OH)·5H ₂ O	68°	X = $\frac{c}{b}$	ORTH	---	Col s	6 1.93	Diss by acids. Reported α given as "1.44 calc."	
1.455 <u>1.466</u>	<u>1.461</u>	1.463	.014	MENDOZITE NaAl(SO ₄) ₂ ·11H ₂ O	56°	X = $\frac{b}{c}$ Y:c = 30°	MCL Laths c	100 good	White	H 3 F 1 6 1.77	Sol in H ₂ O. Data on synth compd.	
<u>1.466</u>	---	<u>1.461</u>	---	SILHYDRITE 3SiO ₂ ·H ₂ O	---		ORTH	fr uneven to subconch	White	6 2.14	Dec by HC1. Opt sign unk.	
1.438	<u>1.463</u>	1.465	.027	HEXAHYDRITE MgSO ₄ ·6H ₂ O	29°	Y = $\frac{b}{c}$ X:c = -25°	MCL	---	Col s	G 1.8 infus	Sol in H ₂ O. Data on synth compd.	
----	<u>1.465</u>	---	---	ZINCFAUSERITE (Mn,Mg,Zn)SO ₄ ·7H ₂ O (?)	Large	---	ORTH	010 good	Pale rose	H 2.5 6 2.00 infus	Sol in H ₂ O, efflo- resces.	
1.455 <u>1.480</u>	1.447	<u>(1.468)</u>	1.470	.023	GOSLARITE, magnesian (Epsomite grp) (Zn,Mg)SO ₄ ·7H ₂ O	Small	X = $\frac{a}{b}$ Z = $\frac{c}{b}$ el cTv neg	ORTH	010 perf	White	H 2. G 1.8 infus	Sol in H ₂ O.
1.447	<u>1.469</u>	1.472	.025	BORAX Na ₂ B ₄ O ₇ ·10H ₂ O	40°	X = $\frac{b}{c}$ Z:c = -55°	MCL pris c tab 100	100 perf 110 good	White	H 2.2-5 G 1.71 F 1-1.5	Sol in H ₂ O. Tw pl 100 rare.	
1.451	<u>1.470</u>	1.478	.027	KANEMITE NaHSi ₂ O ₄ (OH) ₂ ·2H ₂ O	46°	X = $\frac{b}{a}$	ORTH	010 perf	Col s, white	H 4 6 1.93	Dec by acids.	
1.447	<u>1.470</u>	1.473	.026	EPSONITE, nickel loan (Mg, Ni)SO ₄ ·7H ₂ O	47°	X = $\frac{b}{a}$ Z = $\frac{c}{b}$ c1v pos	ORTH	010 fair	Pale blue-green	H 2 G 1.8 infus	Sol in H ₂ O. Ni O 12.2%.	
1.463	<u>1.470</u>	1.471	.008	PARALUMINITE Al ₄ (SO ₄) ₅ ·15H ₂ O	Small	X = e1	Fib, mass	---	White, chalky	Soft	---	
1.460	<u>1.471</u>	1.474	.014	STARKEVITE MgSO ₄ ·4H ₂ O	(55±18°)	---	MCL	---	White	G ~ 1.8 infus	Sol in H ₂ O. (Synth compd is biax pos, β 1.491.)	
1.455	<u>1.472</u>	1.487	.032	KERNITE Na ₂ B ₄ O ₇ ·4H ₂ O	80°	Z = $\frac{b}{a}$ X:a = 38°	MCL	001, 100 perf 201 fair	White, col s	H 2.5 G 1.90 F 1-1.5		
1.459	<u>1.473</u>	1.483	.024	JURBANITE Al ₂ SO ₄ (OH)·5H ₂ O	80°	Y = $\frac{b}{a}$ Z:a = -5°	MCL	---	Col s	H 2.5 G 1.79		
1.465	<u>1.473</u>	1.477	.012	SASAITE (Al,Fe) ₁₄ (PO ₄) ₁₁ (OH) ₇ (SO ₄)·83H ₂ O (?)	---	X = $\frac{c}{b}$ el cTv pos	ORTH (?)	001 perf	White, chalky	G 1.75 infus	Diss by acids. Fe ₂ O ₃ 1.1%.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ calc ($2V_x$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.465 1.484	---	<u>1.474</u>	---	.002-.008	GMELINEITE (Zeolite grp) (Na,Ca)Al ₂ Si ₄ O ₁₂ ·6H ₂ O	Small	----	HEX	1010 dist	White	H 4.5 G 2.0-2.1 F 3	Dec by HCl. Tw axis \underline{c} .
1.580	---	<u>1.474</u>	---	.008	HISINGERITE Fe ⁺³ Si ₂ O ₅ (OH) ₄ ·2H ₂ O	Small	----	MCL u mass	Conch	Dark brown	H 3.5 G ~ 2.5 infus	Dec by acids.
1.482	---	<u>1.47-</u> <u>1.48</u>	---	wk	HYDROBASALUMINITE Al ₄ SO ₄ (OH) ₁₀ ·12-35H ₂ O	---	----	----	----	Col to yellowish	G 1.86	Diss by acids. Opt char unk.
1.471	<u>1.471</u>	<u>1.475</u>	1.476	.005	SODIUM DACHHARDITE (Zeolite grp) (Na,Ca,K)4-Al ₈ Si ₄₀ O ₉₆ ·28H ₂ O	88-92°	Z:c = 18°	MCL	----	Col s	G 2.17	----
1.480	1.472	<u>1.475</u>	1.477	.005	MORDENITE (Zeolite grp) (Ca,Na,K)Al ₂ Si ₁₀ O ₂₄ ·7H ₂ O	80°	X = $\frac{a}{c}$ Z = $\frac{c}{c}$	ORTH	100 perf 010 good	Col s	H 5 G 2.12 F 3-4	Some samples appear to be monoclinic.
1.461	<u>1.478</u>	1.484	.023	CREEDITE Ca ₃ Al ₂ (SO ₄) ₂ (F,OH) ₁₀ ·2H ₂ O	64° r > v wk	Y = $\frac{b}{c}$ Z:c = 41°	MCL pris c	100 perf	Col s	H 4 G 2.72 F diff	Diss by acids.	
1.358	<u>1.479</u>	1.530	.172	BARENTSITE Na ₇ Al(CO ₃) ₂ (HCO ₃) ₂ F ₄	62° r < v wk	X = \underline{c}	TCL ps hex	001, 100 perf	Col s	H ~ 3 G 2.56	Eff in acids, slowly dec by H ₂ O.	
1.476	<u>1.479</u>	1.481	.005	TWEITITE Ca _{1-x} (Y,Ce) _x F _{2+x} (x ~ 0.3)	34°	----	MCL ps cub	----	White to pale yellow	----	Poly tw. Fluor faint yellow-orange in short-wave UV.	
1.476	<u>1.479</u>	1.479	.003	CLINOPTILOLITE (Zeolite grp) (Na,K,Ca)Al ₂ -3Al ₃ (Al,Si) ₂ Si ₁₃ O ₃₆ ·12H ₂ O	Small r < v str	X = $\frac{b}{c}$ Z:a = 15° el pos (?)	MCL tab 010	010 perf	Col s	H 3.5-4 G 2.2 F 2	Dec by HCl.	
(1.468)	1.457	<u>1.480</u>	1.484	.027	GOSLARITE (Epsomite grp) ZnSO ₄ ·7H ₂ O	46° r > v wk	X = $\frac{a}{c}$ Z = $\frac{b}{c}$ el cTw pos	ORTH acid c	010 perf	Col s, yellowish	H 2-2.5 G 1.98 infus	Sol in H ₂ O.
1.47	<u>1.48</u>	1.49	.02	BOOTHITE CuSO ₄ ·7H ₂ O	Large	Y = $\frac{b}{c}$ X ~ \underline{c}	MCL fib c	001 imperf	White, sticky	H 2-2.5 G 2.1 fus	Sol in H ₂ O. Dehydrates to Chalcantinite.	

1.472	---	1.487	.015	MAKATITE $\text{Na}_2\text{Si}_4\text{O}_9 \cdot 5\text{H}_2\text{O}$	---	---	ORTHO([?]) spher- ulitic	---	White	G 2.07	Dec by HCl. unk.	Opt char	
1.391	<u>1.481</u>	1.486	.095	DARAPSKITE $\text{Na}_3(\text{SO}_4)(\text{NO}_3) \cdot \text{H}_2\text{O}$	27° $r > v$ rather str	X = <u>b</u> $Z:c = 12^\circ$ el-cl v pos	MCL 010 perf 100	Col s	H 2-3 G 2.20 F 1-2	Sol in H_2O . 100.	Poly tw		
1.380	<u>1.482</u>	1.573	.193	KALICINITE KHCO_3	81° (83±1°)	Y = <u>b</u> $X:c = 30^\circ$ el <u>b</u>	MCL 100, 001, 101	White	G 2.16 F easy	Sol in H_2O .			
1.462	<u>1.482</u>	1.495	.033	AUBERTITE $\text{CuAl}(\text{SO}_4)_2\text{Cl} \cdot 14\text{H}_2\text{O}$	71° $r > v$ mod	---	TCL	010 perf	G 1.82	Sol in H_2O .			
1.462	<u>1.482</u>	1.490	.028	MORAESITE $\text{Be}_2\text{P}_0_4(\text{OH}) \cdot 4\text{H}_2\text{O}$	65°	Z = <u>b</u> $Y:c = 11-23^\circ$	MCL 010, 001 good	White	G 1.81	Diss by acids.			
1.470	---	1.493	.023	NICKEL-HEXAHYDRITE $(\text{Ni}, \text{Mg}, \text{Fe})\text{SO}_4 \cdot 6\text{H}_2\text{O}$	---	Z:c = 45°	MCL crusts 010 perf 100	Blue- green	---	Sol in H_2O . MgO 2.8%. Opt char unk.	NiO 22.6%.		
1.477	<u>1.482</u>	1.484	.007	PICKERINGITE (Halotrichite grp) $\text{MgAl}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Med	Y = <u>b</u> $Z:c = 30^\circ$	MCL fib	Col s to yellow	H 2-2.5 G 1.8 infus	Sol in H_2O , taste astringent. In closed tube melts in H_2O of crystn.			
1.479	<u>1.488</u>	1.477	.021	REVDITE $\text{Na}_2\text{Si}_2\text{O}_5 \cdot 5\text{H}_2\text{O}$	75°	Y:c = 0- -10°	TCL	100, 010 perf	Col s	H 2 G 1.94	Sol in H_2O , gives an alk soin.		
1.469	<u>1.482</u>	1.490	.004	APOHONITE (Halotrichite grp) $(\text{Mn}, \text{Mg})\text{Al}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Small	Y = <u>b</u> $Z:c = 29^\circ$	MCL fib <u>c</u>	---	White, silky	H 1.5 G 1.78 infus	Sol in H_2O . In closed tube melts in H_2O of crystn.		
1.478	<u>1.482</u>	1.482	.004	BIEBERITE $\text{Co}_3\text{O}_4 \cdot 7\text{H}_2\text{O}$	88° disp wk	Y = <u>b</u> $Z:c \sim 30^\circ$	MCL 001 perf 110 fair	Rose- to carmine- red	H 2 G 1.95 infus	Sol in H_2O . Loses H_2O on exposure.			
1.477	<u>1.483</u>	1.489	.012	CLINOPTILOLITE (Zeolite grp) $(\text{Na}, \text{K}, \text{Ca})_2\text{Al}_3$ (Al, Si) ₂ $\text{Si}_{13}^{10}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	Large r < v str	X = <u>b</u> $Z:a = 15^\circ$ el pos (?)	MCL 010 perf	Col s	H 3.5-4 G 2.2 F 2	Dec by HCl.			
1.479	<u>1.484</u>	1.486	.005	HUNGCHAOLITE $\text{MgB}_4\text{O}_5(\text{OH})_4 \cdot 7\text{H}_2\text{O}$	36°	---	TCL ps hex	---	White	---	Diss by acids.		
1.442	<u>1.485</u>	1.490	.048	HUNGRERITE (Zeolite grp) $(\text{Na}, \text{K}, \text{Ca})_2\text{Al}_2\text{Si}_7\text{O}_{18} \cdot$ $6\text{-H}_2\text{O}$	---	X = <u>a</u> Y = <u>b</u>	ORTH 0RTH	---	Col s	G 2.13 fus	Sodium analogue of Stellerite.		
1.477	<u>1.485</u>	1.486	.009	PHOSPHOROESSLERITE $\text{MgHPo}_4 \cdot 7\text{H}_2\text{O}$	38° r > v	X = <u>b</u> $Z:c = 7^\circ$	MCL Conch	Col s to yellow	H 2.5 G 1.73 infus	Diss by acids.			

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
---	1.485	---	---	LOW	MAGADIITE $\text{NaSi}_7\text{O}_{13}(\text{OH})_3 \cdot 3\text{H}_2\text{O}$	---	MCL pos	---	White	---	Dec by acids. Sign unk.
1.483	1.486	1.487	.004	BLOEDITE $\text{Na}_2\text{Mg}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	71° $r < v$ str	$y = \frac{b}{c}$ $x:c = 37^\circ$	MCL tab 001	Conch	Col s	H 2.5-3 G 2.25 F 1.5	Sol in H_2O .
1.484	1.486	1.487	.003	PHILLIPSITE (Zeolite grp) (K, Ca, Na) ₁₋₂ (Al, Si) ₈ $0.16 \cdot 6\text{H}_2\text{O}$	Med $r > v$	---	MCL	010, 100 good	Col s	H 4-4.5 G 2.2 F 3	Gel with acids. Penet tw.
1.483	1.487	1.490	.007	LEONITE $\text{K}_2\text{Mg}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	~90° $r < v$	$y = \frac{b}{c}$ $z \sim \frac{a}{2}$	MCL	Conch	Col s	H 2.5-3 G 2.20 F easy	Sol in H_2O . Poly tw.
1.483	1.487	1.490	.007	ANALCIME (Zeolite grp) $\text{NaAlSi}_2\text{O}_6 \cdot \text{H}_2\text{O}$	Very small	---	CUB trapezo-dra	100 imperf	Col s	H 4-4.5 G 2.25 F 3.5	Dec by HC1. Taste grating.
1.480	---	1.487	---	ANALCIME (Zeolite grp) $\text{NaAlSi}_2\text{O}_6 \cdot \text{H}_2\text{O}$	Very small	---	CUB trapezo-dra	100 imperf	Col s	H 4-4.5 G 2.25 F 3.5	Dec by HC1. Taste grating.
1.493	1.493	1.488	1.490	HALOTRICHITE $\text{FeAl}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Med	$y = \frac{b}{c}$ $z:c = 38^\circ$	MCL fib c	010 poor fr conch	White, silky	H 2 G 1.89 F 4.5-5	Sol in H_2O . Taste astringent.
1.482	1.480	1.488	1.490	HALOTRICHITE $\text{FeAl}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Med	$y = \frac{b}{c}$ $z:c = 38^\circ$	MCL fib c	010 poor fr conch	White, silky	H 2 G 1.89 F 4.5-5	Sol in H_2O . Taste astringent.
1.486	1.488	1.489	.003	VANTHOFFITE $\text{Na}_6\text{Mg}(\text{SO}_4)_4$	83° $r < v$ wk	---	MCL	---	Col s	H 3.5 G 2.69 F 3	Sol in H_2O .
1.484	1.488	1.490	.006	CHABAZITE (Zeolite grp) (Ca, Na) $\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	60-70°	---	TRIG	10Tl poor	White, reddish	H 4 G 2.05- 2.10 F 3	Dec by HC1. 10Tl, 0001.
1.448	1.489	1.493	.045	BURKEITE $\text{Na}_6(\text{CO}_3)(\text{SO}_4)_2$	34° $r > v$ dist	$x = \frac{c}{b}$ $z = \frac{a}{b}$	ORTH	Conch	White	H 3.5 G 2.57 F 1.5	Sol in H_2O .
1.473	(1.489)	1.490	.017	Unnamed Na-Mg-U sulfate-borate	17-27° ext 28°	TCL	---	---	Pale yellow	---	Taste yellow-green in UV.
1.482	1.489	1.496	.014	STILBITE (Zeolite grp) $\text{NaCa}_2\text{Al}_5\text{Si}_{13}\text{O}_{36} \cdot 14\text{H}_2\text{O}$	43° (90±17°)	$y = \frac{b}{c}$ $x:a = 2.9$	MCL acic 3	010 perf	White, pink, yellow	H 4 G 2.15 F 3	Dec by HC1. Cruciform, penet. Var with high Na.
1.502	1.502	1.502	1.502								Am. Mineral., 62, 1261 (1977).

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.495	1.498	1.502	.007	GOOSECREEKITE (Zeolite grp) $\text{CaAl}_2\text{Si}_6\text{O}_{16}\cdot 5\text{H}_2\text{O}$	82°	$Y = b$ $Z:c = 46^\circ$	MCL	010 perf	Col s	H 4.5 G 2.21	Insol in HCl.	
1.490	1.498	1.501	.011	HYDROGLAUBERITE $\text{Na}_4\text{Ca}(\text{SO}_4)_3\cdot 2\text{H}_2\text{O}$	60°	$Z:c = 3.5^\circ$ el pos	MCL acic	010	White	G 2.40	Dec by H_2O , diss by HCl.	
1.493	1.499	1.503	.010	YUGAWARALITE (Zeolite grp) $\text{CaAl}_2\text{Si}_6\text{O}_{16}\cdot 4\text{H}_2\text{O}$	65-71° $r < v$	$Z = b$ $X:c = -9^\circ$	MCL	Parting 010	Col s	H 4.5 G 2.23	Insol in acids.	
1.499	1.500	1.501	.002	MERLINITE (Zeolite grp) $(\text{K}, \text{Ca}, \text{Na})_7\text{Si}_{23}\text{Al}_9\text{O}_{64}\cdot 23\text{H}_2\text{O}$	56° $r > v$	$X = b$ $Z = \bar{a}$ el pos	ORTH	---	White	G 2.14- 2.27	---	
1.442	(1.500)	1.504	.062	ADMONTITE $\text{Mg}_2\text{B}_{12}\text{O}_{20}\cdot 15\text{H}_2\text{O}$	30° $r < v$	$X:c = 45^\circ$	MCL tab 100	Conch	Col s	H 2-3 G 1.82	---	
---	1.500	---	wk	BITINITE (Halotrichite grp) $\text{Fe}^{+2}\text{Fe}^{+3}_2(\text{SO}_4)_4\cdot 22\text{H}_2\text{O}$	---	ext 35-39°	MCL radiating, fib	---	White to yellowish	H 2 G 1.88	Diss in H_2O .	
1.442	(1.500)	1.504	.062	Unnamed magnesium borate	~ 30° $r < v$	---	ORTH	---	Col s	---	Am. Mineral., 62, 1261 (1977).	
1.377	1.501	1.583	.206	NAHCOLITE NaHC_3	75° $(72\pm 1^\circ)$ $r > v$ wk	$Y = b$ $Z:c = 27^\circ$	MCL pris	101 perf 111 good	White	H 2.5 G 2.21 F 1	Sol in H_2O , gives an alk soln.	
1.490	---	1.502	.012	BEARSITE $\text{Be}_2\text{AsO}_4(\text{OH})\cdot 4\text{H}_2\text{O}$	---	$Z:c = 9^\circ$	MCL fib	---	White	G 2.2	---	
1.498	(1.501)	1.502	.004	WATRAKITE (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{O}_{12}\cdot 2\text{H}_2\text{O}$	75° $r > v$	$X \sim b$ $Z \sim \bar{c}$	MCL	---	White	H 5.5-6 G 2.26 F 2.5	Gel with acids. U biax pos.	
1.494	1.502	1.507	.013	STILBITE (Zeolite grp) $\text{NaCa}_2\text{Al}_5\text{Si}_{13}\text{O}_{36}\cdot 14\text{H}_2\text{O}$	30-50° $(75\pm 18^\circ)$	$Y = b$ $X:\bar{a} = 2.9^\circ$	MCL acic \bar{a}	010 perf	White, pink, yellow	H 4 G 2.15 F 3	Dec by HCl. Tw 001, cruciform, penet.	
1.414	1.503	1.527	.113	NESQUEHONITE $\text{Mg}(\text{HC}_3)(\text{OH})\cdot 2\text{H}_2\text{O}$	53° $r < v$ wk	$X = \bar{a}$ $Z = \bar{b}$	ORTH el \bar{c}	110 perf 001 poor	Col s	H 2.5 G 1.85 infus	Diss by cold acids with eff.	

1.455	<u>1.503</u>	.094	HELLYERITE $\text{NiCO}_3 \cdot 6\text{H}_2\text{O}$	85°	X = $\frac{c}{b}$ Z = $\frac{d}{b}$	ORTH	100, 101 cols, gray 010 good 001 fair	H 2.5 G 1.97 infus.	Diss by acids with eff. Faintly pleoc.
1.472	<u>1.503</u>	1.526	.054	AKSALITE $\text{MgB}_6\text{O} \cdot 5\text{H}_2\text{O}$	80°	Z = $\frac{b}{c}$ Y:c = 28°	MCL	H 2.5 G 2.0	Diss by acids.
1.498	<u>1.503</u>	1.506	.008	HARMOTOME (Zeolite grp) $(\text{Ba}, \text{K})_1 \cdot 2(\text{Si}, \text{Al})_8\text{H}_2\text{O}$	75-82°	X = e1 Z = $\frac{c}{b}$ Y:c = 36°	ORTH	H 4.5 G 2.44 F 3.5	Dec by HCl. Tw cruciform, penet on 001.
1.426	<u>(1.504)</u>	1.508	.082	Unnamed carbonate of K, Ca	~ 25°	X = e1 Z = $\frac{c}{b}$ Y:c = 13°	ORTH	---	Pale rose --- Am. Mineral., 58, 139 (1973).
1.332	<u>1.504</u>	1.504	.172	NITER KNO_3	5-7° r < v wk	X = $\frac{c}{b}$ Z = $\frac{d}{b}$	ORTH	H 2.11 G 2.11 F 1	Sol in H_2O .
1.492	<u>1.505</u>	1.517	.025	INYOITE $\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 13\text{H}_2\text{O}$	80° r < v wk	Y = $\frac{b}{c}$ X:c = 36°	MCL tab	H 2.5-3 G 2.13 F 2	Sol in hot H_2O .
1.494	<u>1.505</u>	1.516	.022	KAINITE $\text{MgSO}_4 \cdot \text{KC1} \cdot 3\text{H}_2\text{O}$	85° r > v wk	Y = $\frac{b}{c}$ Z:c = 13°	MCL tab	H 2.5-3 G 2.13 F 2	Sol in H_2O .
1.344	<u>1.506</u>	1.506	.162	NITROMAGNE SITE $\text{Mg}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$	5° r < v perc	---	MCL pris	H 1.58 G 2.13 F 2	Sol in H_2O , taste bitter.
1.420	<u>1.506</u>	1.524	.104	THERMONATRITE $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$	48° r < v wk	X = $\frac{b}{d}$ Z = $\frac{a}{d}$	ORTH flat-tended 001 or 010	H 1-1.5 G 2.26 F 1.5	Sol in H_2O , gives an alk soln.
1.470	<u>1.506</u>	1.527	.057	EZCURRITE $\text{Na}_4\text{B}_10\text{O}_{17} \cdot 7\text{H}_2\text{O}$	73° r > v	---	TCL	H 3-3.5 G 2.1 F easy	Sol in H_2O .
1.498	<u>1.506</u>	1.508	.010	SEPIOLITE $\text{Mg}_4\text{Si}_6\text{O}_{15}(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	50°	Z = e1 ORTH fib	---	White to yellow	Dec by HCl with sepn of silica.
1.525	<u>1.504</u>	1.509	.005	NICKELBLOEDITE $\text{Na}_2(\text{Ni}, \text{Mg})(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	60-70°	Y = b MCL	---	Light green	Sol in H_2O . Ni 8.8, Mg 3.0, Cu 0.6, Co 0.3%.
1.486	<u>1.504</u>	1.528	.055	AKSALITE $\text{MgB}_6\text{O} \cdot 5\text{H}_2\text{O}$	(73+4°)	X = $\frac{c}{b}$ Z = $\frac{d}{b}$	ORTH	H 2.5 G 2.0	Diss by acids.
1.518	<u>1.503</u>	1.473	<u>1.508</u>						

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.486 Δ	1.505	1.509	1.511	.006	PHILLIPSITE (Zeolite grp) $K, Ca, Na)_1\cdot 2(Al, Si)_8$ $0\cdot 16\cdot 6H_2O$	Med $r > v$	---	MCL	010, 100 good	Col s	H 4-4.5 G 2.2 F 3	Gel with acids. Penet. tw.
1.515	1.505	1.509	1.509	.004	COWLESITE (Zeolite grp) $CaAl_2Si_3O_10\cdot 5\cdot 6H_2O$	30° $X = \frac{a}{b}$ $y = \frac{b}{e_1} pos$	ORTH bladed	010 perf	Col s, white	H 5-5.5 G 2.13	---	
1.470	1.510	1.579	.109	STRONTIOTOBORITE $SrB_8O_{11}(OH)_4$	80-100° (102+2°) (40°)	---	MCL plates	---	Col s	G 2.81	Diss by acids.	
1.465	1.51	1.546	.075	SMARTZITE $CaMg(UO_2)(CO_3)_3\cdot 12H_2O$	---	MCL pris	---	Green	G 2.3	Diss by acids. Pleoc., X col s, Y and Z green. Fluor bright green in UV.		
1.490	1.510	1.524	.034	KURNAKOVITE $MgB_3O_3(OH)_5\cdot 5H_2O$	80° $r > v$	Y ~ b $Z:c = -22^\circ$	TCL	010 perf 110 good	Col s	H 3 G 1.86 F 2	Diss by acids.	
1.479	1.510	1.511	.032	SAPONITE (Smeectite grp) $(Na, Ca)_0.33Mg_3(Al, Si)_4$ $0\cdot 10(OH)_2\cdot 4H_2O$	20-30° $x \perp 01$ $e_1 pos$	MCL scales	001 perf	White, waxy	H 1-2 G 2.2 F diff	Dec by acids.		
1.531 γ	1.479	1.510	1.522	URANOSPATHITE $HAI(UO_2)_4(Po_4)_4\cdot 4OH_2O$	69-76° $r > v$	X = c $y = \frac{b}{d}$	ORTH plates	001 perf 100, 010 good	Yellow to pale green	H 2-2.5 G 2.50	Diss by acids. Pleoc., X pale yellow, Y and Z yellow. Fluor wk yellow-green in UV.	
1.538 γ	1.492	1.510	.030	JOKOKUITE $MnSO_4\cdot 5H_2O$	70-80° Smal l	---	TCL	No cleavage	Pale pink	H 2.5 G 2.03	So1 in H ₂ O.	
1.498	1.510	1.517	.019	---	---	---	---	---	Yellow	---	Dec by H ₂ O. Pleoc., X col s, Z yellow. Fluor bright yellow-green in UV. Am. Mineral., 62, 1261 (1977).	
1.496	---	1.512	.016	Unnamed Na-Mg-U- borate-sulfate A	---	---	---	---	---	---	Dec by acids.	
1.500	1.511	1.513	.013	EPISTILBITITE (Zeolite grp) $CaAl_2Si_6O_16\cdot 5H_2O$	40° $r > v$ str	$y = \frac{b}{d}$ $Z:c = 9^\circ$	MCL col s	010 perf	H 4 G 2.25 F 3	Dec by acids.		

1.508	<u>1.511</u>	1.512	.004	GARRONITE (Zeo1ite grp) Na ₂ Ca ₅ Al ₁₂ Si ₂₀ H ₆₄ 27H ₂ O	0-30°	---	ORTH ps tet	2 poor at 90°	Cols 100 good fr conch	H 4.5 G 2.15		
1.521	1.483	<u>1.512</u>	1.530	.047	INDERBORITE CaMg[B ₃ O ₃ (OH) ₅] ₂ •6H ₂ O	77°	Z = <u>b</u> X:C = 2°	MCL	---	Diss by acids.		
	1.494	<u>1.512</u>	1.524	.030	NASINITE Na ₂ B ₅ O ₈ (OH) ₂ •2H ₂ O	67° (78+7°)	Y = <u>b</u> Z:a = 7°	MCL	---	Sol in H ₂ O.		
	1.492	<u>1.512</u> (1.526)	1.518	.023	PENTAHYDRITE (Chaicanthite grp) (Mg,Cu)SO ₄ •5H ₂ O	55° r < v	X ~ <u>b</u>	TCL	---	Sol in H ₂ O. CuO 9.0, ZnO 5.6, FeO 1.4%.		
	1.534	1.494	<u>1.512</u>	1.512	.018	MONTMORILLONITE (Smectite grp) (Na,Ca) _{0.33} (Al,Mg,Fe) ₂	Small	Y = <u>b</u> X ~ <u>c</u>	MCL u mass	001 perf	Dec by acids.	
	1.497	<u>1.512</u>	1.513	.016	META-ALUMINITE Al ₂ SO ₄ (OH) ₄ •5H ₂ O	Si ₄ O ₁₀ (OH) ₂ •xH ₂ O	Z = <u>b</u> Y:el = 43°	MCL Laths	---	---		
	1.504	<u>1.512</u>	1.516	.012	LAUMONTITE var Leonhardite (Zeolite grp) CaAl ₂ Si ₄ O ₁₂ •4H ₂ O	44° (70+19°) r < v str	Y = <u>b</u> Z:c = 30°	MCL	010, 110 good	White	Gel with acids. H ₂ O 12.8%.	
	1.524	<u>1.512</u>	1.514	.003	OKENITE CaSi ₂ O ₄ (OH) ₂ •H ₂ O	Large	Z ~ <u>c</u>	TCL fib c mass	010 perf	White	H 3.5-4 G 2.3 F 2.5-3	
	1.535	<u>1.512</u>	1.514	---	KINGITE Al ₃ (PO ₄) ₂ (OH,F) ₃ •9H ₂ O	---	TCL (?) mass	---	White	White	G 2.2-2.3 Opt char unk.	
	1.508	<u>1.515</u>	1.520	.012	NACAPHITE Na ₂ Ca(Po ₄)F	80° r > v wk	---	ORTH	Conch	Cols 001 diff	Diss by HCl.	
	1.445	<u>1.515</u>	1.523	.078	GAYLUSITE Na ₂ Ca(CO ₃) ₂ •5H ₂ O	35° r < v str crossed	X = <u>b</u> Z:c = -15° disp str	MCL el a	110 perf 001 diff	cols white	H 2.5-3 G 1.99 F 1.5	
	1.509	<u>1.512</u> ^	1.515	1.517	.005	COWLESITE (Zeo1ite grp) CaAl ₂ Si ₃ O ₁₀ •5-6H ₂ O	44-53°	X = <u>a</u> Y = <u>b</u> el pos	ORTH bladed	010 perf	cols white	H 5-5.5 G 2.13
	1.501	<u>1.516</u>	1.525	.024	LETOVICITE (NH ₄) ₃ H(SO ₄) ₂	75°	Z = <u>b</u> Y:C = -12°	MCL tab 001	001 dist	Cols white	Sol in H ₂ O. Tw lam.	
	1.512	<u>1.516</u>	1.518	.006	GISMONDINE (Zeo1ite grp) (Ca,Na)Al ₂ Si ₂ O ₈ •4H ₂ O	Large r < v wk	Y = <u>b</u> Z:c = 42°	MCL ps tet	---	Cols white	H 4.5 G 2.2	
	1.528	<u>1.516</u>	1.518	---						Gel with acids. Tw.		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.513	1.516	1.518	1.518	.005 LOUDARITE (Na ₃ K,Ca) ₂ (Be,Al)Si ₃ 0 ₈ •2H ₂ O	~90°	---	ORTH pris	100, 010, 001	Cols to yellowish	H 6.5 G 2.33 F 3.5	Insol in acids.
1.501	1.517	1.518	1.518	.017 SYNGENITE K ₂ Ca(SO ₄) ₂ •H ₂ O	28° r < v str	Z = b X:c = -2° el neg	MCL laths el c	110, 100 perf 010	White	H 2.5 G 2.60 F 1.5-2	Dec by H ₂ O, diss by acids. Tw pl 100 common.
1.507	1.517	1.521	1.521	.014 NATROSILITE Na ₂ Si ₂ O ₅	55° r < v	Y ~ c	MCL ps hex thick tab	100 perf 001 dist 011 poor	Cols	G 2.48 F 2	Dec by acids. Anom blue interf colors.
1.508	1.518	1.522	1.522	.014 ILESITE (Mn,Zn,Fe)SO ₄ •4H ₂ O	Med	Y = b Z:c = 5°	MCL tab 100	---	Clear green	G 2.25	Sol in H ₂ O. Data on synth compd.
[]	1.512	1.518	1.519	.007 SCOLECITE (Zeo1ite grp) CaAl ₂ Si ₃ 10•3H ₂ O	35° r < v str	Z = b X:c = 18° el neg	MCL el c	110 perf 100	White	H 5 G 2.27 F 2	Gel with acids. Tw 100 common
1.507 ^h	1.513	1.518	1.520	.007 NICKELBLOEDITE Na ₂ Ni(SO ₄) ₂ •4H ₂ O	60-70°	Y = b	MCL	---	Light green	G 2.43	Sol in H ₂ O. Ni 13.3, MgO 1.0, FeO 0.9%.
1.433	1.519	1.528	1.528	.095 WEGSCHEIDERITE Na ₅ (CO ₃) ₂ (HC0 ₃) ₃	(34+4°)	---	TCL	---	Cols	G 2.34 F easy	Sol in H ₂ O, gives an alk soln.
1.51 ^[] 1.53	---	1.52	---	.005-BASALUMINITE Al ₄ SO ₄ (OH) ₁₀ •5H ₂ O	---	el neg	Mass compact	Conch	White	G 2.1	Diss by HCl. Opt char unk.
---	1.52	---	Wk	CHLOROCALCITE KCaCl ₃	---	ORTH (?) ps cub	Three clv	Cols	H 2.5-3	Sol in H ₂ O, deliq. Poly tw. Validity dubious.	
1.512 ^h	1.496	1.521	1.539- 1.544	.043- .048 INDERBORITE CaMg[B ₃ O ₃ (OH) ₅] ₂ •6H ₂ O	80-86°	Z = b X:c = 2.5°	MCL fr conch	100 good	Cols	H 3.5 G 2.00 fus	Diss by acids.
1.510	1.521	1.523	0.013	BIKITAITE LiAlSi ₂ O ₆ •H ₂ O	45° r < v	Z = b X:c = 28°	MCL	001, 100 poor	Cols	H 6 G 2.30	---
1.504	1.522	1.539	.035	HANNAYITE (NH ₄) ₂ H ₄ Mg ₃ (PO ₄) ₄ •8H ₂ O	~90° r < v wk	---	TCL tab 100	III com- plete T00, OT0 poor	Cols to yellowish T00, OT0	G 2.03 fus	Diss by acids. Data on synth compd.

1.518	1.522	.006	MICROCLINE (Feldspar grp) KAlSi ₃ O ₈	X':a on 001 = 0- 10° 010 = 5°	60-70° r > v	TCL 001 perf 010 good	H 6 G 2.56 F 4	
1.519	1.522	.005	ANORTHOCLADE (Feldspar grp) (K,Na)AlSi ₃ O ₈	X':a on 001 = 0-3° 010 = 8-9°	51° r > v wk	TCL 001 perf 010 good	H 6 G 2.57 F 4	
1.539	v 1.523	1.531	IVANOVITE Chloride-borate of Ca	X = b Z:c = 20- 26° el clv pos	58-72° r < v	MCL ps hex mass	H 6 G 2.57 F 4	
1.504	1.523			X ~ c y = a	29-45°	MCL mass	H 6 G 2.57 F 4	
(1.492)	(1.524)	1.526	(.034)	SWINEFORDITE (Smectite grp) (Li,Ca,Na)(Al,Li,Mg)4 (Si,A1)8 ₂₀ (OH,F)4	90°	ORTH acic 010 good	H 6 G 2.57 F 4	
1.518	1.524	1.530	.012	MACDONALDITE BaCa ₄ Si ₁₆ O ₃₆ (OH) ₂ • 10H ₂ O	25° r < v str	MCL 010 good	H 6 G 2.56 F 5.5	
1.513	1.524	1.525	.012	LAUMONTITE (Zeolite grp) CaAl ₂ Si ₄ O ₁₂ •4H ₂ O	46° r > v wk	MCL 010 good	H 6 G 2.56 F 5.5	
1.512	^	1.519	1.524	.006	ORTHOCLASE (Feldspar grp) KAlSi ₃ O ₈	Y or Z = b Z:c = 3- 12° r < v	MCL 010 good	H 6 G 2.57 F 4
1.522	^	1.525	1.525	.005	SANIDINE (Feldspar grp) (K,Na)AlSi ₃ O ₈	24° 010 = 6°	MCL 010 good	H 6 G 2.56 F 4
1.544	v 1.529	1.520		URALOLITE CaBe ₃ (PO ₄) ₂ (OH) ₂ •4H ₂ O (80+9°)	Z:el = 20°	MCL fib aggre- gates and needles	H 6 G 2.5 F 4	
1.510	1.526	1.536	.026		X:el = 0°	MCL el indist	H 6 G 2.5 F 4	
1.518	---	1.530	.012	TACHARANITE Ca ₁₂ Al ₂ Si ₁₈ O ₅₁ •18H ₂ O	---	001 good	G 2.36	
1.506	^ 1.516	1.525	.010	SEPIOLITE Mg ₄ Si ₆ 15(OH) ₂ •6H ₂ O	Med	001 good	Opt char unk.	
1.553						001 good	H 2 G 2.3 infus	
						001 good	Dec by HCl with sepn of silica. Fe ₂ O ₃ 2.5%.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ										
1.512 1.537	1.513 <u>1.526</u>	1.534 <u>1.526</u>	1.528 <u>1.526</u>	.021	SIDEROTIL (Chalcocite grp) (Fe,Cu)SO ₄ ·5H ₂ O	60°	---	TCL	---	Pale green to blue	H 2-3 G 2.1-2.2	Sol in H ₂ O.	
1.535	1.521	1.526	1.528	.007	CORDIERITE Mg ₂ Al ₄ Si ₅ O ₁₈	64°	X = $\frac{c}{b}$ Y = $\frac{c}{d}$	ORTH ps hex	010 fair	Col s	H 7 G 2.51 F 5.5	Partly dec by acids. Data for synth compd.	
1.452	1.527	1.538	1.538	.086	STENONITE (Sr,Ba,Na) ₂ Al(CO ₃)F ₅	43°	X = $\frac{b}{c}$ Z: $\frac{c}{c} = -32^\circ$	MCL	001, 120	Col s, white	H 3.5 G 3.86	Diss by acids.	
1.429	1.528	1.538	1.109	.109	AMEGHINITE NaB ₃ O ₃ (OH) ₄	33° r < v	Z = $\frac{b}{c}$ X: $\frac{c}{c} = 9^\circ$	MCL	100 good 010, 001 poor	Col s	H 2.5 G 2.03 F easy	Sol in H ₂ O. Fluor pale blue in UV.	
1.522 1.543	1.528	1.530	1.530	.008	GISMONDINE (Zeolite grp) (Ca,Na)Al ₂ Si ₂ O ₈ ·4H ₂ O	15-90° r < v wk	Y = $\frac{b}{c}$ Z: $\frac{c}{c} = 42^\circ$	MCL ps tet	---	Col s	H 4.5 G 2.22 F 3	Gel with acids. Tw pl 110, 001.	
1.525 1.534	1.523 <u>1.529</u>	1.529	1.530	.007	SANIDINE (Feldspar grp) (Na,K)AlSi ₃ O ₈	33°	---	MCL	001 perf 010 good	Col s	H 6 G 2.56	Insol in acids.	
1.365	1.530	1.595	1.265	.265	GLUSHINSKITE (an oxalate) MgC ₂ O ₄ ·2H ₂ O	(58+1°)	---	MCL	---	---	---	Diss by HCl.	
1.518	1.530	1.542	1.024	.024	MIMASAGRITE VO(SO ₄)·5H ₂ O	Med large (89-9°)	X = $\frac{b}{c}$ Z ~ $\frac{c}{c}$	MCL pris	---	Blue	H 2.03 G 1.85 F easy	Sol in H ₂ O. Pleo str, x deep blue, y blue, Z nearly col s, abs x > y > z.	
1.536	1.522 <u>1.530</u>	1.533	1.011	.011	PALYGORSKITE MgAlSi ₄ O ₁₀ (OH)·4H ₂ O	61°	---	ORTH fib	---	White	H 2.13 - G 2.27 infus	Insol in acids.	
1.546	1.522 <u>1.530</u>	1.531	1.009	.009	FEDORITE (Na,K)Ca(Si,Al) ₄ (O,OH) ₁₀ ·1.5H ₂ O	32° r < v	X ~ $\frac{c}{b}$ Y ~ $\frac{c}{d}$	TCL	001 perf	Col s to pale red	G 2.58 F easy	Insol in acids.	
1.510 1.565	1.490 <u>1.531</u>	1.534	.044	.044	SAPONITE (Smectite grp) (Ca,Na)O ₃₃ (Mg,Fe) ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	Med (30-8°)	X \perp 001 Z = e1	MCL plates	001 perf	White, pink, gray, green	H 1.2 G 2.3 F diff	Dec by acids.	
1.522	(1.531)	1.536	.014	.014	BOYLEITE (Zn,Mg)SO ₄ ·4H ₂ O	70°	---	MCL	---	Uneven	White	G 2.41 infus	Sol in H ₂ O.

1.524	<u>1.536</u>	.012	KOKTAITE $(\text{NH}_4)_2\text{Ca}(\text{SO}_4)_2 \cdot \text{H}_2\text{O}$	72°	$y = 2^\circ b$ $z:c = 2\frac{1}{2}$ e ₁ pos	MCL acic	---	Col s	G 2.09	Dec by H_2O , diss by acids. Tw 100.	
1.529	<u>1.532</u>	1.533	.004	DELHAVELITE $(\text{Na},\text{K})_{10}\text{Ca}_5\text{Al}_6\text{Si}_{32}^0\text{O}_{80}$ $(\text{Cl},\text{F},\text{SO}_4)_3 \cdot 18\text{H}_2\text{O}$	~ 90°	$x = \frac{a}{c}$ $y = \frac{c}{a}$	ORTH platy	010 dist	Col s	H ~ 4 G 2.60 F 3	Diss by acids. Wavy ext, anomalous blue-gray b.
1.511	<u>1.533</u>	1.535	.024	NYEREREITE $(\text{Na},\text{K})_2\text{Ca}(\text{CO}_3)_2$	29°	$x = \frac{c}{a}$ $y = \frac{a}{c}$	ORTH tab	---	Col s	G 2.54 fus	Diss by acids with eff. Tw.
1.515	<u>1.533</u>	1.535	.020	SEARLESITE $\text{NaBSi}_2^0(\text{OH})_2$	Med	$z = \frac{b}{c} = 30^\circ$ $x:c = 30^\circ$	MCL pris	100 perf	White	Soft G 2.45 F easy	Diss by acids.
1.529	<u>1.534</u>	1.556	.068	ARTINITE $\text{Mg}_2\text{CO}_3(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	70°	$y = \frac{b}{c} = 30^\circ$ $z:c = 30^\circ$	MCL acic, radiating, botryoidal	100 perf 001 good	White, silky	H 2.5 G 2.02	Diss by acids with eff.
1.529	<u>1.534</u>	1.488									
1.512	<u>1.534</u>	1.534	.031	MONTMORILLONITE (Smectite grp) $(\text{Na},\text{Ca})_{0.33}(\text{Al},\text{Mg},\text{Fe})_2$ $\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	Small	$y = \frac{b}{c} \sim \frac{1}{2}$ $x \sim \frac{1}{2}$	001 perf ps hex earthy	001 perf	White, gray, pink	H 1-2 G 2.06- 2.23	Dec by acids.
1.529	<u>1.543</u>	1.527	1.534	ALBITITE plagioclase, volcanic (Felspar grp) $(\text{Na},\text{Ca},\text{K})\text{AlSi}_3\text{O}_8$	~ 50°	$x':a = 3^\circ$ $001 = 3^\circ$ $010 = 23^\circ$	TCL	001 perf 010 good	Col s	H 6 G 2.60 F 4	Insol in acids. Forms series with Anorthite and Sanidine. Data for end member Ab ₁₀₀ An ₀ O ₀ .
(1.543)	<u>1.528</u>	(1.534)	1.536	APLOWITE $(\text{Co},\text{Mn},\text{Ni})\text{SO}_4 \cdot 4\text{H}_2\text{O}$	Med	---	MCL	---	Pink	H 3 G 2.33 infus	Sol in H_2O .
1.410	<u>1.535</u>	1.543	.133	NATRITE Na_2CO_3	28°	$y \sim b$	MCL	001 perf 010, 100 less so	---	H 3.5 G 2.54	Sol in H_2O . Poly tw.
1.500	<u>1.535</u>	1.560	.060	MEYERHOFFERITE $\text{Ca}_2\text{B}_6^0\text{Al}_1 \cdot 7\text{H}_2\text{O}$	78°	$z':c = 12^\circ$ $r > v$	TCL tab	100	Col s, white	H 2 G 2.12 F easy	Diss by acids.
1.515	<u>1.535</u>	1.536	.021	GLAUBERITE $\text{Na}_2\text{Ca}(\text{SO}_4)_2$	7°	$z = \frac{b}{c} = 12^\circ$ el clv pos	MCL tab	001 110 in- dist brittle	Gray, col s, yellow- red	H 2.5-3 G 2.75- 2.85 F 1.5	Dec by hot H_2O , diss by acids.
1.514	<u>1.535</u>	1.540	.010	OKENITE $\text{CaSi}_2^0\text{Al}_4(\text{OH})_2 \cdot \text{H}_2\text{O}$	Large	$z \sim c$	TCL fib c	010 perf	White	H 5 G 2.3 F 2.5	Gel with acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.531	1.535	1.541	.010	VERTUMNITE $\text{Ca}_4\text{Al}_4\text{Si}_4\text{O}_6(\text{OH})_24 \cdot 3\text{H}_2\text{O}$	62°	$X = \frac{c}{Z:b} = 16^\circ$	MCL	Conch brittle	Cols	H 5 G 2.15	Gel with acids.
1.530	1.535	1.538	.008	CORDIERITE $(\text{Mg},\text{Fe})_2\text{Al}_4\text{Si}_5\text{O}_8$	84°	$X = \frac{c}{Y} = \frac{b}{d}$	ORT ps hex	010 fair 001 poor	Blue, green, cols	H 7 G 2.59 F 5.5	Partly dec by acids. Tw sectors, also lam. Pleoc, X cols to pale yellow, Y pale to dark blue, Z pale blue. FeO 2.1%.
1.536	1.536	1.554	.131	TESCHEMACHERITE $(\text{NH}_4)\text{HCO}_3$	49° $(41+2^\circ)$ $r < v$	$X = \frac{a}{b}$ $y = \frac{b}{v}$	ORTH mass	110 perf	Cols to yellowish	H 1.5 G 1.57 volat	Sol in H_2O , eff with acids.
1.494	1.536	1.536	.042	BEIDELLIITE (Montmorillonite grp) $(\text{Na},\text{Ca})_0.33\text{Al}_2(\text{Si},\text{Al})_4$ $0.10(\text{OH})_2 \cdot \text{xH}_2\text{O}$	9-16°	$X = \frac{c}{Z} = \frac{a}{c}$ $v = \frac{c}{v}$	MCL earthy el ctv pos	001 perf	White, reddish, gray	G 2.0 infus	Refractive indices increase when immersed in certain oils.
1.513	1.536	1.545	.032	MINASAGRITE $\text{Mn}(\text{SO}_4) \cdot 5\text{H}_2\text{O}$	Med large	$X = \frac{b}{Z} \sim \frac{c}{v}$	MCL pris	---	Blue	G 2.03 F easy	Sol in H_2O . Pleoc str, X deep blue, Y blue, Z nearly cols, abs X > Z.
1.530	1.536	1.541	.014	ROZENITE $\text{FeSO}_4 \cdot 4\text{H}_2\text{O}$	Large $r < v$ wk	---	MCL	---	Cols to greenish	G 2.20	Sol in H_2O .
1.527	1.536	1.542	---	GLAUOKERINITE $(\text{Zn},\text{Cu})_{10}\text{Al}_4\text{SO}_4$ $(\text{OH})_{30} \cdot 2\text{H}_2\text{O}$ (?)	---	Z = el	---	Radiating, fib	Sky-blue	H 1 G 2.15	Concentric color banding. Opt char unk.
1.490	1.537	1.538	.048	SCHRÖCKINGERITE $\text{NaCa}_3(\text{UO}_2)(\text{CO}_3)_3(\text{SO}_4)_4$ $\text{F} \cdot 10\text{H}_2\text{O}$	18°	$X = \frac{c}{Y} = \frac{b}{d}$	TCL ps hex	001	Greenish-yellow	H 5 G 2.54	Diss by H_2O . Fluor yellow-green in UV.
(1.545)	1.537	1.539	.038	CHALCANTHITE $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$	56° $r < v$ perc	---	TCL pris	110 im-perf conch	Dark blue	H 2.5 G 2.29	Sol in H_2O .
1.501	1.537	1.545	.090	IKAITITE $\text{CaCO}_3 \cdot 6\text{H}_2\text{O}$	~ 45° $(31+5^\circ)$ disp inclined	$Y = \frac{b}{Z:c} = 17^\circ$	MCL	---	Cols	G 1.77	Dec by acids.
1.455	1.538	1.545									

1.516	<u>1.538</u>	1.547	.031	KORSHUNOVSKITE $Mg_2Cl(OH)_3 \cdot 3.5H_2O$	62°									Sol in dil acids.			
1.510	(1.522)	1.538	1.542	(.020)	ARSENURANOSPATHITE HAI $(UO_2)_4(AsO_4)_4 \cdot 4OH_2$	52° r > v	X = c el pos	TCL pris	---	Cols	H ~ 2 G 1.80			Sol in acids. Fluor wk in UV.			
		1.534	1.538	1.543	.009	CANASITE $(Na,K)_6Ca_5Si_{12}O_{30}(OH,F)_4$	58°	Y = b Z:a = 5°	TET or ORTH ps tet	001 perf 100, 010 good	Pale yellow	H 2 G 2.54					
		1.496	1.539	1.557	.061	BIRNBUCCITE $Na_4B_{10}O_{16}(OH)_2 \cdot 2H_2O$	63°		MCL acic	2 perf at 118°	Greenish-yellow	G 2.71 F easy		Dec by acids. Poly tw.			
		1.522	1.536	1.539	1.541	.005	ANORTHOCLASE (Feldspar grp) (K, Na, Ca)AlSi ₃ O ₈	62° r > v wk	ext on 010 = 3°	TCL	010, 001 perf	White, pink, etc	H 6 G 2.62 F 4		Insol acids. Tw pl 010, 100, giving very fine grating. Na_0 K_2O 3.3, CaO 3.7%.		
			1.522	<u>1.540</u>	1.552	.030	SULFOBORITE $Mg_3B_2(SO_4)(OH)_{10}$	70-88°	Y = b X = a	ORTH pris	110 good 001 fair	Cols	H 4-4.5 G 2.42 fus		Diss by acids.		
			1.541	pos	1.536	<u>1.541</u>	1.546	.010	OLIGOCLASE plagioclase, plutonic (Feldspar grp) (Na, Ca)AlSi ₂ Si ₂ O ₈	~ 90° r < v wk	X':a on 001 = 1° 010 = 7°	TCL	001 perf 010 good	White	H 6 G 2.64 F 4-5		Insol in acids. Tw pl 010 poly, almost universal. Data for An ₁₈ .
															Diss by acids.		
			1.520	<u>1.541</u>	1.545	.025	LUENEBURGITE $Mg_3B_2(Po_4)_2(OH)_6 \cdot 5H_2O$	62° (47±15°)	---	MCL fib	Pris	Cols to brownish	H 2 G 2.05 infus				
															Diss by acids with eff.		
			1.537	1.466	<u>1.542</u>	1.596	.130	DANSONITE $NaAlCO_3(OH)_2$	77° r < v wk	X = a Y = c	ORTH bladed acic	110 perf	Cols, white	H 3 G 2.44 infus			
			1.528	<u>1.542</u>	1.549	.021	KOYDORSKITE $Mg_{10}(Po_4)_4(CO_3)_2(OH)_4 \cdot 9H_2O$	81° r > v wk	Z:c = 1-3° el pos	MCL	Conch to uneven	Pale rose	H 4 G 2.60				
			1.535	<u>1.542</u>	1.549	.014	FOSHALLASITE $Ca_3Si_2O_7 \cdot 3H_2O$	12-18° (90±16°)	---	MCL (?) tab	100 perf	Snow-white	H 2.5-3 G 2.5				
			1.528	<u>1.538</u>	<u>1.543</u>	1.548	.010	GISMONDINE (Zeolite grp) (Ca, Na)Al ₂ Si ₂ O ₈ · 4H ₂ O	15-90° r < v wk	Y = b Z:c = 42°	MCL ps tet	---	Cols, white	H 4.5 G 2.22 F 3		Gel with acids. Tw common, tw pl 110, 001.	
			1.536	---	1.550	.014	LOUDOUNITE $NaCa_5Zr_4Si_{16}O_{40}(OH)_1 \cdot 8H_2O$	---	fib	---	Green to white	H 5 G 2.48		Sign unk. Ext wavy.			
			1.534	<u>1.543</u>	1.546	.008	OLIGOCLASE plagioclase, volcanic (Feldspar grp) (Na, Ca)AlSi ₂ O ₈	65°	X':a on 001 = 3° 010 = 4°	TCL	001 perf 010 good	White, cols	H 6 G 2.63 F 4-5		Insol in acids. Tw pl 010 poly. Data for An ₂₀ .		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
(1.534) \wedge	1.533	(1.543)	1.546	.013	APLOWITE (Co,Mn,Ni)SO ₄ ·4H ₂ O	Med	---	MCL	---	Pink	H 3 G 2.33 infus	Sol in H ₂ O.
1.539	1.543	1.544	1.544	.005	EPIDIDYMITE NaBeSi ₃ O ₇ (OH)	2-3°	X = $\frac{a}{b}$ Z = $\frac{d}{b}$	ORTH	001 good 010 perf	Col s	H 5.5 G 2.61 F easy	Insol in acids. Multiple tw.
1.522	1.544	1.552	1.552	.030	MUNDRABILLAITE (NH ₄) ₂ Ca(HPO ₄) ₂ ·H ₂ O	(61+9°)	X = $\frac{b}{c}$ Y:c = 26°	---	---	Col s	Soft G 2.05	Sol in H ₂ O.
1.533	1.544	1.548	1.548	.015	BEHOITE Be(OH) ₂	82°	Y = $\frac{a}{c}$ Z = $\frac{c}{c}$	ORTH	---	White	H 4 G 1.92	Diss by acids.
1.533	1.544	1.547	1.547	.014	PENTAGONITE Ca(VO)Si ₄ O ₁₀ ·4H ₂ O	50°	X = $\frac{b}{a}$ Y = $\frac{c}{a}$	ORTH	010 good	Greenish-blue	H 3-4 G 2.33	Pleoc str, X and Z cols, Y blue.
1.524 \wedge 1.568	1.544	1.546	1.546	.004	HYALOPHANE (Feldspar grp) (K,Ba)Al(Si,Al) ₃ O ₈	75°	Z = $\frac{b}{a}$ X:a = -18°	MCL	010, 001 perf	Col s, white	H 6 G 2.79 F diff	Insol in acids. BaO 6.9% (Cn 37).
1.525 \vee 1.560	1.545	1.545	1.545	.020	VERMICULITE (Mg,Fe,Al) ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	~ 0°	X \perp plates	MCL	001 perf	Greenish, brownish	H 1.5 G 2.28 F 5	Dec by HCl. Exfoliates when heated. FeO 0.7, Fe ₂ O ₃ 4.2%.
1.533	1.545	1.547	1.547	.014	MOOREITE (Mg,Zn,Mn)Si ₄ (OH) ₁₄ · 3H ₂ O	50°	X = $\frac{b}{c}$ Z:c = 44° el clv pos	MCL	010 perf	Col s, white	H 3 G 2.47 infus	Diss by acids. Mg:Zn:Mn = 4:2:1.
1.541 \wedge 1.550	1.546	1.550	1.550	.009	OLIGOCLASE Plagioclase, plutonic (Feldspar grp) (Na,Ca)(Al,Si) ₂ Si ₂ O ₈	~ 78°	X: $\frac{a}{c}$ on 001 = 1° 010 = 2°	TCL	001 perf 010 good	Col s, white	H 6 G 2.65 F 4-5	Insol in acids. Tw pl 010, poly. Data for An 25.
1.487	1.546	1.560	1.560	.073	BRADLEYITE Na ₃ Mg(Po ₄)(CO ₃) ₂	49°	X = $\frac{b}{c}$ Y:c = 7°	MCL flat- tened 001	Conch	White, gray	H 3.5 G 2.72	Dec by H ₂ O, diss by acids with eff.
1.507	1.546	1.569	1.569	.062	CARBOBORITE Ca ₂ Mg(CO ₃)B ₂ (OH) ₈ ·4H ₂ O	75°	Y = b Z:c = 12° el pos	MCL pris 001	100 perf	Col s	H 2 G 2.12	Fluor white in UV, pale green phosphorescence.
1.530 \wedge (~1.57)	1.546	1.548	1.548	.020	PALYGORSKITE (Mg,Al,Fe) ₂ Si ₄ O ₁₀ (OH)· 4H ₂ O	30-35°	---	ORTH fib	---	White, greenish	G 2.4-2.5 infus	Insol in acids. Fe ₂ O ₃ 6.6%.

1.535 1.560	<u>1.546</u>	1.548	.008	CORDIERITE (Mg,Fe) ₂ Al ₄ Si ₅ O ₁₈	68°	x = c y = <u>b</u>	010 fair 001 poor	Blue, green, cols	H 7 G 2.61 F 5.5
1.438	<u>1.547</u>	1.595	.157	OXAMMITE (NH ₄) ₂ C ₂ O ₄ •H ₂ O	62° r < v dist	x = c y = <u>a</u> el c ₁ v pos	001 dist	cols to yellowish	G 1.5 F volat
1.537	<u>1.547</u>	1.549	.012	GYROLITE Ca ₂ Si ₃ O ₇ (OH) ₂ •H ₂ O	Small	el clv pos	0001 perf	cols	H 2.5 G 2.40 F diff
1.522	<u>(1.548)</u>	1.549	.027	PHLOGOPITE (Mica grp) KMg ₃ Si ₃ Al ₁₀ (F,OH)	14° r < v	y = b el c ₁ v pos	MCL ps hex	001 perf	H 2-2.5 G 2.88
1.564	<u>1.548</u>	1.548	.027	LEPIDOLITE (Mica grp) K(Li,Al) ₃ (Si,Al) ₄ O ₁₀ (F,OH) ₂	39° r > v	Z':c ₁ v = 0-7° el clv pos	MCL ps hex	001 perf	Insol in acids. FeO 0.1, Fe ₂ O ₃ 0.2, MnO 0.7, Li ₂ O 5.6%.
1.555	<u>1.548</u>	1.553	.024	TRUSCOTTITE (Ca,Mn)Al ₁₄ Si ₂₄ O ₅₈ (OH) ₈ •2H ₂ O	Small	---	HEX	cols, pink, gray	H 2.5-4 G 2.81 F 2.5
1.522	<u>1.549</u>	1.549	.027	"RECTORITE" Na(Mg,Al) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂ •2H ₂ O	0-39° (56-7°)	---	MCL ps hex	001 perf	Dec by HCl.
1.519	<u>1.550</u>	1.559	.040	CHELIKARITE CaNgB ₂ O ₄ Cl ₂ •7H ₂ O (?)	---	---	ORT pris	cols to pale rose	Interstratified Pyrophyllite- Vermiculite.
1.520	---	1.558	.038	SAUCONITE (Smeectite grp) Na _{0.33} Zn ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ •4H ₂ O	---	---	MCL ps hex	001 perf	Opt char unk.
1.530	<u>(1.550)</u>	1.551	.021	NAUJAKASITE Na ₆ (Fe,Mn)(Al,Fe) ₄ Si ₈ O ₂₆	0-20°	x ⊥ clv y = b el c ₁ v pos	MCL ps hex plates	white, yellow, brown	Dec by acids. ZnO 35.7, MgO 2.4, Al ₂ O ₃ 0.4%.
1.537	<u>1.550</u>	1.556	.019	NAUJAKASITE Na ₆ (Fe,Mn)(Al,Fe) ₄ Si ₈ O ₂₆	52-71°	---	MCL ps hex plates	001 perf	Gel with HCl.
1.546 1.550 pos	<u>1.550</u>	1.554	.008	ANDESINE plagioclase, plutonic (Feldspar grp) (Na,Ca)(Al,Si) ₂ Si ₂ O ₈	~ 90	x' : a on 001 = 0° 010 = -5°	TCL 010 good	silvery, gray	H 2-3 G 2.62 F 3
1.530	---	1.559	.029	YOFORTIERITE (Mn,Mg)Si ₈ O ₂₀ (OH) ₂ • 8-9H ₂ O	---	x:fib = 8°	ORTH or mon fib	---	Insol in acids. Poly tw 010. Data for An ₃₃ .
1.519	<u>1.552</u>	1.561	.042	BALAVINSKITE Sr ₂ B ₆ Al ₁₁ •4H ₂ O	Large (54±6°)	---	---	pink to violet	Opt char unk. Pleoc.
						---	---	---	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.543 ↓ 1.552 pos	1.535	1.552	1.553	.018	SATIMOLITE $KNaAl_4B_6O_{15}Cl_3 \cdot 13H_2O$	Small	---	ORTH	---	White	G 2.1	---
	1.549	1.552	1.556	.007	ANDESINE plagioclase, volcanic (Feldspar grp) (Na,Ca)(Al,Si) ₂ Si ₂ O ₈	~ 90°	$X':a$ on 001 = 2° 010 = -4°	TCL	001 perf 010 good	Col s	H 6 G 2.66 F 4.5	Insol in acids. Tw pl 010 poly. Data for An ₃₈ .
1.567 ^	1.485	1.553	1.570	.085	ALUMOHYDROCALCITE $CaAl_2(CO_3)_2(OH)_4 \cdot 3H_2O$	64° (51+3°)	$X = \frac{b}{c}$ ext:el = 7- 10°	MCL	100 perf	Chalky to pale blue	H 2.5 G 2.23 infus	Diss by hot H ₂ O, diss by acids with eff.
	1.550	1.553	1.554	.004	CHAROITE $(K,Ba,Sr)(Ca,Na)_2Si_4O_{10}(OH,F) \cdot H_2O$	30°	Three clv el c ₁ v pos	Lilac to violet	---	G 2.54	In thick fragments pleoc. X rose, Z col. U pos.	
1.570 ^	1.522	1.553	1.553	.031	TAENIOLITE (Mica grp) $KLiMg_2Si_4O_{10}F_2$	0-5° r > v	$Y = \frac{b}{c}$ el pos	MCL ORTH fib	001 perf	Col s to brown	H 2.5-4 G 2.83	Insol in acids.
1.525 ^	(1.522)	1.553	1.579	(.057)	SEPIOLITE (ferrrian) $(Mg,Fe)_4Si_6O_{15}(OH)_2 \cdot 6H_2O$	83°	el pos	001, 010	Brown	---	Dec by HCl. Fe ₂ O ₃ 10.0, FeO 1.2%.	
	1.535	1.553	1.557	.022	HYDROCALUMITE $Ca_4Al(OH)_7 \cdot 3H_2O$	25° (50+12°) r < v	$Y = \frac{b}{c}$ el pos	MCL ps hex	001 perf	Col s to light green	H 3 G 2.15 infus	Diss by acids. Becomes unax at about 90°C.
	1.541	1.553	1.560	.019	BERYLLOLITE $Be_3Si_4(OH)_2 \cdot H_2O$	Small (74+12°)	el pos	Fib	---	White	Soft G 2.20	---
1.549 ^	1.541	1.553	1.557	.016	EDINGTONITE (Zeolite grp) $BaAl_2Si_3O_{10} \cdot 4H_2O$	54° r < v wk	$X = \frac{c}{a}$ Z = $\frac{a}{c}$ el neg	ORTH ps tet	110 perf 110	Col s	H 4 G 2.75 F 5	Gel with acids.
	1.498	1.554	1.594	.096	MINGUZZITE $K_3Fe(C_2O_4)_3 \cdot 3H_2O$	78°	---	MCL tab	010 perf	Green to yellow-green	G 2.03	An oxalate. Sol in H ₂ O. Pleoc. X yellow green, Z bright emerald-green.
	1.545	1.554	1.565	.020	Unnamed phosphate	~ 90° disp wk	---	MCL	---	Pale yellow, pale green	H 4.5 G 3.13	Diss by acids.

1.531	<u>1.555</u>	1.570	.039	SHORTITE $\text{Na}_2\text{Ca}_2(\text{CO}_3)_3$	75° $r < v \text{ mod}$	$X = \frac{c}{z} = \frac{d}{b}$ $Z = \frac{b}{d}$	010 dist	Cols to yellowish	H 3 G 2.6 F 1	Dec by H_2O , diss by acids with eff.	
1.548 \hat{v}	<u>1.555</u>	1.559	.022	LEPIDOLITE (Mica grp) $\text{K}(\text{Li},\text{Al})_3(\text{Si},\text{Al})_4\text{O}_{10}$ (F,OH) ₂	38° $r > v$	$Z':\text{clv} = 0-7^\circ$ el clv pos	001 perf	Cols, pink, lilac	H 2.5-4 G 2.86 F 2.5	Insol in acids. Fe_2O_3 1.1, FeO 0.7, MnO 0.9, Li_2O 5.5%.	
1.567 v	<u>1.555</u>	1.555	.010	LIZARDITE (Serpentine grp) $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	Very small	$y = \frac{b}{z} = \frac{d}{a}$	001 perf	Green	G 2.55 infus	Slowly attacked by acids. Fe_2O_3 2.2%.	
<u>v</u>	<u>1.568</u>	<u>1.556</u>	<u>1.559</u>	.007	HALLSY SITE, chromian (Al,Cr) ₂ $\text{Si}_2\text{O}_5(\text{OH})_4$	$\sim 90^\circ$	---	---	Pale greenish-blue	Insol in acids.	
<u>1.57</u>	1.545	---	1.56	.015	"MN-PALYGORSKITE" $\text{NaMgMn}(\text{Fe},\text{Al})_3\text{Si}_7\text{O}_{20}$ (OH) ₂ •10 H_2O (?)	Large	$Z = \underline{c}$	Spherulitic	---	Rose to red	Pleoc., X yellow, Z reddish brown, abs Z > X. Am. Mineral., 55, 2139 (1970).
1.552	<u>1.558</u>	1.561	.009	BERYLLONITE NaBePO_4	68° $r < v \text{ wk}$	$X = \frac{b}{z} = \frac{d}{c}$	MCL ps orth tab	010 perf 100 good	Cols to pale yellow	G 2.62	Mineral., 55, 2139 (1970).
1.514	<u>1.559</u>	1.562	.048	ZINC SILITE $\text{Zn}_3\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$ (?)	$0-22^\circ$	$Z:c = 3^\circ$	MCL	001 perf	White to bluish	H 5.5-6 G 2.81 F 3	Slowly diss by acids.
1.551	<u>1.559</u>	1.562	.011	ARMENITE (Osumilite grp) $\text{BaCa}_2\text{Al}_6\text{Si}_9\text{O}_{30} \cdot 2\text{H}_2\text{O}$	60°	---	HEX	---	---	---	---
1.541	<u>1.560</u>	1.567	.026	FENAKSITE $(\text{K},\text{Na},\text{Ca})_4(\text{Fe}^{+2},\text{Fe}^{+3},\text{Mn})_2\text{Si}_8\text{O}_{20}(\text{OH},\text{F})$	77° $(65 \pm 10^\circ)$	$Z = \underline{b}$	TCL	001 perf	Pale rose	H 5-5.5 G 2.74 F easy	Dec by acids.
1.540 \hat{v}	<u>1.560</u>	1.560	.020	VERMICULITE $(\text{Mg},\text{Fe},\text{Al})_3(\text{Si},\text{Al})_4$ $0_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	$6-8^\circ$	$X \perp 001$ el clv pos	MCL	001 perf	Green to brown	H 1.5 G 2.29 F 5	Exfoliates when heated. Dec by HCl. Fe_2O_3 4.2, FeO 1.8%.
1.545 \hat{v}	<u>1.560</u>	1.561	.016	LOVOZERITE $(\text{Na},\text{Ca})_3(\text{Zr},\text{Ti})\text{Si}_6$ (O,OH) ₁₈	Small	$X \sim \underline{c}$	TRIG	---	Dark brown to black	H 5 G 2.38 F easy	Insol in acids. Poly tw.
1.546 \hat{v}	<u>1.551</u>	1.562	.011	CORDIERITE $(\text{Mg},\text{Fe})_2\text{Al}_4\text{Si}_5\text{O}_{18}$	42°	$X = \frac{c}{z} = \frac{d}{b}$	ORTH ps hex	010 fair	Blue, green, cols	H 7 G 2.66 F 5	Partly dec by acids. Tw sectors, also lam.
1.569	<u>1.5605</u>	1.561	.009	KULKEITE $\text{Mg}_8\text{Al}(\text{Si}_7\text{Al})_2\text{O}_{20}(\text{OH})_{10}$	24°	$Y = \frac{a}{z} = \frac{d}{b}$	MCL	001 perf	cols	H 2 G 2.70	Pleoc., X col, Y dark blue. FeO 7.1, Fe_2O_3 0.2%.
1.552	<u>1.5605</u>	1.561	.009	KULKEITE $\text{Mg}_8\text{Al}(\text{Si}_7\text{Al})_2\text{O}_{20}(\text{OH})_{10}$	24°	$Y = \frac{a}{z} = \frac{d}{b}$	MCL	001 perf	cols	H 2 G 2.70	Interlayered talc-chlorite.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Bioref	MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.536 \downarrow 1.588	1.526	<u>1.561</u>	1.561	.035	BEIDELLITE (Smectite grp) Na _{0.33} Al ₂ (Si,Al) ₄ · ₁₀ (OH) ₂ · _x H ₂ O	47° (small)	$X = \frac{c}{a}$ $Z = \frac{c}{\overline{c}v}$ pos	MCL earthy	001 perf	White, reddish, gray	G 2.0 infus	Refractive indices increase when immersed in certain oils.
~1.585	1.553	---	1.569	.016	KARPINSKITE (Mg,Ni) ₂ Si ₂ · ₅ (OH) ₂ (?)	---	$Z:c = 12^\circ$	MCL (?)	---	Col to greenish- blue	H 2.5-3 G 2.6	---
	1.547	---	1.571	.024	GLUCINE Ca ₂ Be ₈ (PO ₄) ₄ (OH) ₈ ·H ₂ O	---	el pos ext	Fine needles	---	Col	H ~ 5 G 2.23- 2.40	Diss by acids. Opt char unk.
	1.547	<u>1.562</u>	1.567	.020	POLYHALITE K ₂ Ca ₂ Mg(SO ₄) ₄ ·2H ₂ O	62-70°	$Z':tw$ on 10T = 28°	TCL fib b tab	10T perf parting 010	Red, yellow, cols	H 3.5 G 2.78 F 1.5	Dec by H ₂ O, diss by acids. Poly tw.
	1.557	<u>1.562</u>	1.563	.006	NACRITE Al ₂ Si ₂ · ₅ (OH) ₄	40-90° r > v wk	$Z = \frac{b}{a}$ $X:c = 12^\circ$	MCL tab wedge- shaped	001 perf	White	G 2.60 infus	Insol in acids. Luster pearly.
	1.521	<u>1.563</u>	1.585	.064	SIDOROVITE Na ₃ Mn(PO ₄)(CO ₃) (70+4%)	63°	el b	MCL ps orth	100, 010 perf	Pale rose	H ~ 2 G 2.90	Diss by acids.
	1.541	<u>1.563</u>	1.564	.023	TANCOITE HNa ₂ LiAl(PO ₄) ₂ (OH)	23° r < v wk	$X = \frac{a}{b}$ $Y = \frac{a}{\overline{c}}$	MCL	010, 001 fair	Pale pink	H 4-4.5 G 2.75	Diss by acids.
	1.560	<u>1.563</u>	1.515	.005	SCHODERITE Al ₂ (PO ₄) ₂ (VO ₄)·8H ₂ O	42°	$X:c = 25^\circ$	MCL	---	Yellow- orange	H 2 G 1.92	Pleoc in yellow, abs Y > Z > X.
(1.548) \downarrow 1.584	1.535	<u>1.564</u>	1.565	.030	PHLOGOPITE (Mica grp) K(Mg,Fe) ₃ Si ₃ Al ₁₀ (F,OH) ₂	10° r < v wk	$Y = \frac{b}{a}$ $X \sim \frac{c}{\overline{c}}$ el ctv pos	MCL platy	001 perf	Yellow, brown	H 2-2.5 G 2.15 F diff	Insol in acids. Pleoc, X col's, Y = Z brown, abs X < Y < Z.
	1.552	<u>1.564</u>	1.571	.019	JENNITE Ca ₉ H ₂ Si ₆ · ₁₈ (OH) ₈ ·6H ₂ O	74°	$Y: el =$ 35-40°	TCL bladed fib	001 perf	White	G 2.32	---
	1.495	<u>1.565</u>	1.572	.087	SAPONITE (Smectite grp) (Ca,Na)0.33(Mg,Fe) ₃ (Si,Al) ₄ · ₁₀ (OH) ₂ ·4H ₂ O	Small (32+5°)	$X \perp 001$	MCL plates	001 perf	Dark green	H 1.5 G 2.31 F 4	Gel with acids. Pleoc, X pale yellow, Y olive-green, Z brown- green. FeO 7.8, Fe ₂ O ₃ 7.3, Al ₂ O ₃ 9.1%.
1.531 \wedge												

1.536	<u>1.565</u>	1.570	.034	FLUOBORITE Mg ₃ B ₃ (F,OH) ₃	Small	---	HEX fib	0001 indist	H 3.5 G 2.82
v 1.550	<u>1.565</u>	1.570	.020	VARISCITE AlPO ₄ ·2H ₂ O	Med r < v wk	X = $\frac{a}{b}$ Z = $\frac{b}{c}$ el cTv neg	0RTH	010 good 001 poor	H 4.5 G 2.57- infus
1.554	<u>1.565</u>	1.573	.019	REEDEMGERNITE (Fe ¹ dispgr grp) NaBSi ₃ O ₈	80°	---	TCL	001 perf	H 6-6.5 G 2.70
1.553	<u>1.565</u>	1.567	.014	MORINITE NaCa ₂ Al ₂ (PO ₄) ₂ (OH,F) ₅ ·2H ₂ O	40° r < v wk	Y = b	MCL	100 perf	Wine-red, cols
1.557	<u>1.565</u>	1.571	.014	VITANIEMITE Na(Ca,Mn)Al(PO ₄) ₃ (F,OH) ₃	81°	Y = b	MCL tab	10T good	Gray, white
1.560	<u>1.565</u>	1.566	.006	KAOLINITE Al ₂ Si ₂ O ₅ (OH) ₄	24-50° r > v	Z = b x:c = 1:4° u	MCL earthly masses	001 perf	White to brownish
1.559 v 1.569	<u>1.565</u>	1.566	.006	ZEOPHYLLITE Ca ₄ Si ₃ O ₈ (OH,F) ₄ ·2H ₂ O	Small	Z = b el cTv pos	TRIG	0001 perf	Col's, white
v 1.560	<u>1.565</u>	1.566	.006	FURONGITE Al ₂ (OH ₂)(PO ₄) ₂ (OH) ₂ · 8H ₂ O	65°	---	TCL	3 perf	Bright yellow to lemon- yellow
v 1.577	<u>1.566</u>	1.573	.027	TOBERMORITE Ca ₅ Si ₆ O ₁₆ (OH) ₂ ·4H ₂ O	26-33°	---	ORTH	---	Col's
1.546	<u>1.566</u>	1.573	.027	ANTIGORITE (Serpentine grp) (Mg,Fe) ₃ Si ₂ O ₅ (OH) ₄	48° r > v	X = c el cTv pos	MCL	001 perf	Green to white
1.554	<u>1.566</u>	1.567	.013	ALUMOHYDROCALCITE CaAl ₂ (CO ₃) ₂ (OH) ₄ ·3H ₂ O	46-70°	X = b ext 7-10°	MCL	100 perf	White
v 1.561	<u>1.566</u>	1.567	.006	LEPIDOLITE (Mica grp) K(Li,Fe,Al) ₃ (Si,Al) ₄ 0 ₁₀ (F,OH) ₂	34° r > v	X ~ c el cTv pos	MCL ps hex	001 perf	Col's, pink, brown
v ~1.587	<u>1.567</u>	1.575	.071	URANOSPINITE (Autunite grp) Ca(UO ₂) ₂ (AsO ₄) ₂ ·10H ₂ O	0-62° r > v mod	X = c	TET tab	001 perf 100 dist	Lemon- yellow to green
1.555 v 1.577	<u>1.567</u>	1.570	.028						H 2.5-4 G 2.95 F 2.5
1.542	<u>1.567</u>	1.570	.022						Diss by acids. Pleoc, X col's, Y and Z pale yellow.
1.55	<u>1.567</u>	1.572	.022						
1.592	<u>1.567</u>	1.572	.022						

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.557	<u>1.567</u>	1.570	.013	OLSHANSKYITE $\text{Ca}_3\text{B}_4(\text{OH})_{18}$	54° $r > v$ wk (68+3°)		ext at 30°	MCL (?) fib	---	Col s	H 4	Diss by acids. Poly tw.
1.520	<u>1.568</u>	1.591	.071	BONSHEDTITE $\text{NaFe}^{+2}(\text{PO}_4)(\text{CO}_3)$		$X = \frac{b}{c}$	MCL ps orth tab	010, 100	Col s, rose	H 4 6 3.16	Dec by acids with eff.	
1.541 (1.568)	1.570	.029	PSEUDO-AUTUNITE $(\text{H}_3\text{O})_4\text{Ca}_2(\text{UO}_2)_2(\text{PO}_4)_4 \cdot 5\text{H}_2\text{O}$ (?)	32° $r > v$ dist	$Y = \frac{c}{a}$	ORTH ps hex plates	001 perf	Pale yellow	G 3.28	Diss by acids. Fluor bright greenish-yellow in UV.		
1.562	<u>1.568</u>	1.574	.012	ELPIDITE $\text{Na}_2\text{ZrSi}_6\text{O}_{15} \cdot 3\text{H}_2\text{O}$	89°	$Z = \underline{a}$	ORTH fib	110 dist	White to tan	H 6 6 2.63	Insol in acids. U bias pos.	
1.563 pos 1.577	<u>1.568</u>	1.573	.010	BYTOMNITE plagioclase, Plutonic (Feldspar grp) $(\text{Ca}, \text{Na})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	~90°	$X':a$ on 001 = -16° 010 = -29°	TCL	001 perf 010 good	White, gray	H 6 6 2.71	Insol in acids. Tw pl 010 poly. Data for An70.	
1.544 1.583	1.564	<u>1.568</u>	.008	CELSIAN var Kesoite (Feldspar grp) $(\text{Ba}, \text{K})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	80°	$X:c$ on $\frac{c}{a} = 10^{-1.3}$ $010 = 2-3^{\circ}$	MCL 001 perf 010 dist		Col s	H 5.5 6 3.00	Not tw. Ba0 25.0% (Cn65).	
1.555 1.638	1.562	<u>1.568</u>	.008	LIZARDITE (Serpentine grp) $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	45°	---	MCL fib		Green	G 2.58 infus	Insol in acids. Pleoc wk, X yellow, Y and Z green.	
1.560 1.572	1.559	<u>1.569</u>	.014	SEKANINAITE (Corderite grp) $(\text{Fe}, \text{Mg})_2\text{Al}_4\text{Si}_5\text{O}_18$	70°	$X = \frac{c}{b}$	ORTH ps hex	100 fair	Blue to violet	H 7 6 2.77 F 5	Partly dec by acids. Tw pl 110, 310. Pleoc, X col, Y and Z light blue, abs Z > Y > X. Fe0 17.8%.	
1.563	<u>1.569</u>	1.573	.010	ARMSTRONGITE $\text{CaZrSi}_6\text{O}_{18} \cdot 2.5\text{H}_2\text{O}$	$r < v$	$Z = \frac{b}{c} = 6^{\circ}$	MCL	001 perf 100 fair	Brown	H 4.5 6 2.58	Insol in acids.	
1.561 1.586	1.560	<u>1.569</u>	(1.570)	NONTROLITE var Volchonskoite (Smectite grp) $\text{Na}_0.33(\text{Fe}, \text{Cr})_2(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	Small	$Y = \underline{b}$	MCL earthy	001 perf	Bright green	H 2 6 2.36 infus	Dec by acids. Cr2O3 22.1%.	

1.569 pos	<u>1.565</u>	1.574	.009	BYTOMNITE plagioclase, volcanic (Fe dispar grp) (Ca,Na)Al(Si,Al)Si ₂ O ₈	~ 90°	X':a on 001 = -23° 010 = -33°	TCL	001 perf 010 good	Cols, white	H 6 G 2.72 infus	
1.577	<u>1.569</u>	1.570	.003	PHOSINATE Na ₃ H ₂ (Ca,Ce)(Si ₄ O ₈) (PO ₄)	69°	X = b Y = c Z:a = 3° el-cl v neg	ORTH columnar	100 perf 010, 110 imperf	Cols to rose	H 3.5 G 2.62, F 3.00	
1.572	<u>1.567</u>	1.570	1.571	.035	MASUTOMILITE (Mica grp) K(Li,Mn,Al) ₃ (Si,Al) ₄ 0.10(F,OH) ₂	28° r > v wk	MCL	001 perf 100 dist	Pale purplish-pink	H 2.5 G 2.90	
	1.536	<u>1.570</u>	1.571	.034	NOVACEKITE (Autunite grp) Mg(UO ₂) ₂ (AsO ₄) ₂ ·12H ₂ O	0-40°	X = c	TET tab	001 perf 100 dist	Yellow	H 2-2.5 G 3.23
	1.543	<u>1.570</u>	1.577	.034	TAENIOLITE (Mica grp) KLiMg ₂ Si ₄ O ₁₀ F ₂	5° r < v	Y = b el c ₁ v pos	MCL ps hex	001 perf ---	Cols to brown	H 2.5-4 G 2.87
	1.540	<u>1.570</u>	1.570	.030	BUKOVSKYITE Fe ₂ (AsO ₄) ₂ (SO ₄) ₂ (OH) ₇ H ₂ O	---	ext at 18°	MCL (?) acidic	---	Yellow to gray-green	G 2.34
	1.553	<u>1.570</u>	1.626	---	SALEITE (Autunite grp) Mg(UO ₂) ₂ (PO ₄) ₂ ·10H ₂ O	0-65° r < v str	X = c	TET tab	001 perf 010, 110 indist	Lemon-yellow	H 2-3 G 3.27
	1.582	<u>1.570</u>	1.574	.015	TOSUDITE (interlayered- Montmorillonite-Chlorite)	59°	el pos	MCL	001 perf 010	Col s	H ~ 1 G 2.83
	1.559	<u>1.570</u>	1.574	.010	SHABYNITE Mg ₅ (BO ₃)(Cl,OH) ₂ (OH) ₅ ·4H ₂ O	(49+9°)	X c ₁ v	MCL fib	---	White	H 3 G 2.32
	1.564	<u>1.570</u>	1.574	.034	ROEMERITE Fe ⁺² Fe ⁺³ ₂ (SO ₄) ₄ ·14H ₂ O	51° r > v str crossed	ext on c ₁ v = 33°	TCL tab	010 perf 001 good	Brown to yellow	H 3-3.5 G 2.17 F 4.5-5
	1.543	<u>1.571</u>	1.577	.034	MANGANOPALYGORSKITE (Mn,Mg,Al) ₂ Si ₄ O ₁₀ (OH)·4H ₂ O	Med large	Z = c	ORTH fib	---	Rose to red	G 2.62
	1.524	<u>1.571</u>	1.583	.059	DEFERRITE Ca ₃ (CO ₃) ₂ (OH,Cl) ₈ ·H ₂ O	42° r < v mod	ORTH	---	---	Col s	G 2.5
	1.566	<u>1.56</u>	(~1.57)	1.58	.02	MANGANOPALYGORSKITE (Mn,Mg,Al) ₂ Si ₄ O ₁₀ (OH)·4H ₂ O	18° r > v str	Y = b Y = c Y = d	MCL spherulitic	Pale yellow	H 3.5 G 3.35
	1.546	<u>1.572</u>	1.576	.030	HATIMEITE Ca(UO ₂) ₂ Si ₆ O ₁₅ ·5H ₂ O	42° r < v mod	ORTH	---	---	Diss by cold HCl with eff.	
	1.533)	<u>1.572</u>	1.573	.040		18° r > v str	---	---	---	Pleoc wk, Y pale yellow, Z col s. Fluor pale green in UV.	
	1.580	<u>1.582</u>									

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.562 <u>1.572</u>	1.583	.021	Unnamed hydrous calcium arsenate		~ 90°	$Z = \frac{b}{c}$ $Y:c = 8^\circ$	MCL (?)	---	Col s	---	Opt. char unk. Am. Mineral., 53, 561 (1968).
1.566 <u>1.570</u> 1.581	1.572	1.573	.003	ENGLISHITE $K_4 Na^2 Ca_9 Al_{18} (PO_4)_6$ $(PO_3 OH)_{12} (OH)_{36} \cdot 8H_2O$	Small	$X = \frac{c}{b}$ $Z \sim \frac{b}{c}$ el cTv pos	MCL (?) plates	001 perf	Col s, white	H 2 G 2.65	---
1.568	1.574	1.580	.012	OVERITE $CaMgAl (PO_4)_2 (OH) \cdot 4H_2O$	65-85° r > v wk	$X = \frac{c}{b}$ $Y = \frac{c}{b}$	ORTH	010 perf	Pale green, col s	H 4 G 2.53 F 2	Diss by hot acids.
1.57 1.610	1.574	1.580	.010	BASSETITE (Meta-autunite grp) $Fe(UO_2)_2 (PO_4)_2 \cdot 8H_2O$	62° r > v str	$X = \frac{b}{c}$ $Z:c = -4^\circ$	MCL ps tet	010 perf	01ive-green, olive-brown	H 2.5 G 3.4	Pleoc., X pale yellow, Y and Z deep yellow. Does not fluor in UV.
1.548	1.574	1.582	.034	LITIDIONITE $KNaCuSi_4^{10}$	56°	----	TCL	----	Blue	G 2.75	----
1.546	1.575	1.579	.033	BRAMMALLITE (Mica grp) $(Na, H_3 O)(Al, Mg, Fe)$ $(Si, Al_4)^{10} [OH]_2,$ $[H_2O]$	Small variable	X ~ c el cTv pos	MCL	001 perf	White	H 1-2 G 2.69	Dec by acids.
1.555 1.555	1.575	1.581	.026	TOBELITE (Mica grp) $(NH_4, K)Al_2 (Si_3 Al)_10$ $(OH)_2$	28° (57+10°)	el pos	MCL	001 perf	White	----	----
1.555 <u>1.586</u>	1.575	1.577	.022	AUTUNITE (Autunite grp) $Ca(UO_2)_2 (PO_4)_2 \cdot$ 10-12H ₂ O	0-53° u 10-30° r > v str	X = c	TET tab	001 perf	Lemon- to sulfur-yellow to greenish	H 2-2.5 G 3.1-3.2 fus	Diss by acids. Pleoc., X cols to pale yellow, Y and Z deep yellow. Fluor yellow-green in UV.
1.567	1.576	1.579	.012	PANETHITE (Na, Ca, K)(Mg, Fe, Mn)PO ₄	51°	----	MCL	----	Pale amber	G 2.95	Tw. FeO 5.3, MnO 1.7%.
1.558	1.576	1.593	.035	XIANGJIANGITE $(Fe^{+3}, Al)(UO_2)_4 (PO_4)_2$ $(SO_4)_2 (OH) \cdot 22H_2O$	(87+7°)	----	ORTH ps tet	----	Yellow	H 1-2 G 2.9-3.1	Diss by acids. Not fluor in UV. Pleoc. wk in yellows. Meteorite mineral.

(1.558)	<u>1.576</u>	1.582	(.024)	MATUJALITE CaAl ₁₈ (PO ₄) ₁₂ (OH) ₂₀ • 20H ₂ O	60° r < v very str	$y = \frac{b}{c}$ $z:c = 8^\circ$	MCL	100 perf	H 1 G 2.33
1.567 v	<u>1.577</u>	1.579	.031	ZINNWALDITE (Mica grp) KLiFe ⁺² Al(Si ₃ Al)O ₁₀ (F, OH) ₂	30° r > v	X ~ $\frac{c}{el}$ cTv pos	MCL ps hex	001 perf	Brown H 2.5-4 G 3.01 F 2.5
1.574	<u>(1.577)</u>	1.577	.003	EKATERINITE Ca ₂ Ba ₇ (Cl, OH) ₂ •2H ₂ O	Very small	---	HEX (?)	---	White, rose H ~ 1 G 2.44
1.568 v	<u>1.571</u>	1.577	1.583	BYTOWNITE plagioclase, plutonic (Feldspar grp) (Ca, Na)Al(Al, Si)Si ₂ O ₈	79°	X':a on 001 = -30° 010 = -37°	TCL	001 perf 010 good	White, gray, cols H 6 G 2.74 infus
1.584	1.572	<u>1.577</u>	1.583	BYTOWNITE plagioclase, volcanic (Feldspar grp) (Ca, Na)Al(Al, Si)Si ₂ O ₈	81°	X':a on 001 = -34° 010 = -36°	TCL	001 perf 010 good	White H 6 G 2.74 infus
1.569 v	1.567	<u>1.577</u>	1.578	ANTIGORITE (Serpentine grp) (Mg, Fe, Cr)Si ₂ O ₅ (OH) ₄	58° r < v str	X = c	MCL laths	001 perf	Lavender H 4 G 2.6 diff
1.584	1.566 v	<u>1.578</u>	1.601	KROEHNICKITE Na ₂ Cu(SO ₄) ₂ •2H ₂ O	78° r < v inclined	$y = \frac{b}{c}$ $x:c = 48^\circ$	MCL pris	010 perf fr conch	Blue to greenish- blue H 2 G 2.90 F easy
1.603	1.544	<u>1.578</u>	1.586	TARASOVITE (Mica grp) (Ca, Na) _{0.42} KNa(H ₃ O)Al ₈ (Si, Al) ₁₆ Al ₄₀ (OH) ₈ •2H ₂ O	23° (51+6°)	---	MCL scales	---	White H 2.36
1.576 v	<u>1.583</u>	1.544	<u>1.578</u>	URANOSILITE UO ₃ •7SiO ₂	---	el pos	ORTH acic	---	Yellow- white G 3.25
1.570	---	---	1.584	0.014	---	Z = b $x:c = 60^\circ$ el cTv neg	MCL laths	010 perf	Deep green to H 4 G 2.53
1.572	<u>1.578</u>	1.582	.010	MONTGOMERYITE Ca ₄ MgAl ₄ (PO ₄) ₆ (OH) ₄ • 12H ₂ O	75° r < v wk	---	---	---	Pleoc. X cols to pale green, Y and Z cols.
v 1.588	1.576	<u>1.578</u>	1.578	CLINOCHLORITE (Chlorite grp) (Mg, Fe) ₅ Al(Si ₃ Al)O ₁₀ (OH) ₈	~ 0° r > v wk	X ~ $\frac{c}{el}$ cTv pos	MCL ps hex	001 perf	Green H 2.5 G 2.7 F diff
1.567	<u>1.579</u>	1.581	.014	AGRELLITE NaCa ₂ Si ₄ O ₁₀ (F, OH)	47°	---	TCL	110, 110 perf 010 poor	White H 5.5 G 2.90

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.569	1.579	1.583	.014	SUDOITE (Chlorite grp) $Mg_2(Al,Fe)_3Si_3Al_10(OH)_8$	78° $r < v$ str	$Z \sim \frac{1}{2} c$ el pos (?)	MCL plates	001 perf	Col to green	H 2-3 G 2.68	Partly dec by acids. Diocataedral chlorite.	
1.572	1.56	1.580	.021	HAIMEITE $Ca(UO_2)_2Si_6O_{15} \cdot 5H_2O$	16-20° $r > v$ str	$Y = b$	MCL spherulitic	110 good	Yellow	H 3.5 G 3.35	Pleoc wk, Y pale yellow, Z cols. Fluor faint green in UV.	
1.474	1.562	1.580	.020	HISINGERITE $Fe^{+3} 2Si_2O_5(OH)_4 \cdot 2H_2O$	Small $r > v$ wk	$Z = c$	MCL u mass	---	Brown	H 3.5 G 2.3-2.6	---	
1.560	1.561	1.581	.020	VERMICULITE (Mg, Fe, Al) ₃ (Si, Al) ₄ $O_{10}(OH)_2 \cdot 4H_2O$	~ 0°	$X \perp$ plates el clv pos	MCL	001 perf	Green, brown	H 1.5 G 2.6 F 5	Dec by HCl. Exfoliates when heated. Fe _{0.3} 7.4, Fe 0 1.1%.	
1.607	1.575	1.581	.008	KINGSMOUNTITE $(Ca, Mn^{+2})_4(Fe^{+2}, Mn)Al_4(Po_4)_6(OH)_4 \cdot 12H_2O$	62°	Z:el = 35°	MCL	---	Col s	H 2.5 G 2.51	Mn 0 8.1, Fe 0 4.1%.	
1.576	1.581	1.584	.008	SORENSENITE $Na_4SnBe_2Si_6O_{16}(OH)_4$	0-75° str, inclined	---	MCL	2 at 63°	Col to pink	G 2.90	Anom yellow-brown and blue interference colors.	
1.585	---	1.631	---	BUKOVSKYITE $Fe_2(AsO_4)(SO_4)(OH) \cdot 7H_2O$	---	ext at 18°	MCL (?) acic	---	Yellow to gray-green	G 2.34	Diss by HCl. Opt char unk.	
1.570	---	1.582	1.631	LAUSENITE $(Fe, Al)_2(SO_4)_3 \cdot 6H_2O$	Large	---	MCL	---	Col s	---	Fe/Al = 1.03.	
1.628	1.548	1.600	.052	MUSCOVITE (Mica grp) $KAl_2(Si_3Al)_0_{10}(OH)_2$	30-47° $r > v$	$Z = b$ $X:c \sim 0^{\circ}$ el clv pos	MCL ps hex	001 perf	Pale brown, pale green	H 2.5-3 G 2.77 F 5.5	Insol in acids. Tw pl 001.	
1.590	1.552	1.582	.035	SALEITE (Autunite grp) $Mg(UO_2)(PO_4)_2 \cdot 10H_2O$	0-65° $r < v$ str	$X = c$	TET tab	001 perf 010, 100 indist	Lemon-yellow G 3.27	Diss by acids. Pleoc, X cols, Z greenish-yellow. Fluor lemon-yellow in UV.		

1.564	<u>1.584</u>	.020	SABUGALITE (Autunite grp) HAl(UO ₂) ₄ (PO ₄) ₄ •16H ₂ O	35°	el pos	TET plates	001 perf	Bright yellow	H 2.5 G 3.20	Diss by acids. Pleoc. X cols, Y and Z pale yellow. Fluor bright lemon-yellow in UV.
1.576	<u>1.582</u>	1.584	TERSKITE Na ₄ ZrSi ₆ O ₁₆ •2H ₂ O	53° r > v wk	---	ORTH ps tet	---	Pale lilac	H 5 G 2.71	Insol in acids. Bright green in UV.
1.574	<u>1.582</u>	1.582	HOPITE Zn ₃ (PO ₄) ₂ •4H ₂ O	Small r < v wk	X = <u>a</u> Y = <u>c</u>	ORTH tab pris	010 perf 100 good	Cols to yellow	H 3.5 G 3.05	Diss by acids.
1.576	<u>1.582</u>	1.584	GUERINITE Ca ₅ H ₂ (AsO ₄) ₄ •9H ₂ O	7-15° r > v str also r < v	Z = el	MCL acic	Three clv	cols	H 1.5 G 2.68	Diss by acids.
1.514	<u>1.583</u>	1.595	STRONTIOTIORESERITE (Sr,Ca)Al ₂ (CO ₃) ₂ (OH) ₄ • H ₂ O	42° Y = el X ⊥ el	ORTH laths	---	White	G 2.71	Diss by HCl with eff.	
1.554	<u>1.583</u>	1.583	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) ₁₀ Si ₁₂ O ₃₀ (OH) ₁₂	~ 0°	X = b Y ~ <u>c</u>	MCL also TCL	001 perf	Golden-red reddish-brown	H 3-4 G 2.74 F 4.5	Dec by acids. Pleoc, X pale yellow-brown, Y and Z dark brown. FeO 26.7, Fe ₂ O ₃ 2.1, MnO 0.4%.
(1.573)	<u>1.583</u>	1.583	THREADGOLDITE Al(UO ₂) ₂ (PO ₄) ₂ (OH)• 8H ₂ O	70°	Y = b Z:c = 4° el neg	MCL tab	001, 100 010, 012	Greenish-yellow	G 3.4	Fluor green in long- wave UV.
1.574	<u>1.583</u>	1.588	URANOCIRITE Ba(UO ₂) ₂ (PO ₄) ₂ •12H ₂ O	60-70°	X = <u>c</u>	TET tab	001 perf	Yellow-green	H 2 G 3.8	Diss by acids. Pleoc wk, X nearly col., Y and Z pale yellow. Fluor green in UV.
1.580	<u>1.583</u>	1.586	CELSIAN (Feldspar grp) (Ba,K)Al(Si ₁ ,Al)Si ₂ O ₈	74°	Z:c on 100 = 33°	MCL	001 perf 010 good	cols	H 5.5 G 3.30	Not tw (Cn 85).
1.545	<u>1.584</u>	1.584	TALC Mg ₃ Si ₄ O ₁₀ (OH,F) ₂	0-10° r > v	X ~ <u>c</u> Z = <u>b</u> el pos	TCL plates	001 perf	White to green	H 1 G 2.7 infus	Tcl polymorph. Pearly luster, greasy feel.
1.555	<u>1.584</u>	1.588	PHLOGOPITE (Mica grp) K(Mg,Fe) ₃ Si ₃ Al ₁₀ ₁₀ (F,OH) ₂	23-30° r < v wk	Y = b X ~ <u>c</u> el pos	MCL plates	001 perf	Brown	H 2-2.5 G 2.86 F diff	Insol in acids. Pleoc. X cols to pale yellow, Y = Z brown, abs X < Y < Z. FeO 4.2, Fe ₂ O ₃ 3.7%.
1.568	<u>1.584</u>	1.585	LAPLANDITE Na ₄ CeTiPSi ₇ O ₂₂ •5H ₂ O	Small	---	ORTH fib radiating	Splintery	Gray to yellowish	H 2-3 G 2.83 F 2	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.570	1.584	1.585	.015	TORREYITE (Mg,Mn)5Zn2(SO4) (OH)12·4H2O	40°	X = <u>b</u>	MCL	010 good	White	G 2.66	Diss by acids. Poly tw.
1.577 [^]	1.575	1.584	.014	ANORTHITE plagioclase, plutonic, and volcanic (Feil disper grp) CaAl2Si2O8	77° X : a on 001 = -40° 010 = -39°	TCL	001 perf 010 good	Cols, white, pink	H 6-6.5 G 2.76	Gel with acids. Tw poly 010, 001 almost universal, other tw laws common. Data for An100.	
1.582	1.584	1.584	.002	ZEKTZERITE NaLiZrSi6O15	~ 0° r > v wk	0RH	100, 010 perf	Cols to pink	H 6 G 2.79	Fluor light yellow in short-wave UV.	
1.575	1.585	1.595	.020	WHITEITE-(Mn) (Mn,Ca)Mg2(Fe,Mn)Al2 (PO4)4(OH)2·8H2O	80-90°	MCL	001 good	Tan to brown	H 3-4 G 2.67	Diss by acids. Tw 001. FeO 7.9, MnO 7.6, MgO 10.1%.	
1.564	1.585	1.585	.021	URAMPHITE (Meta-autunite grp) (NH4)(UO2)PO4·3H2O	0-3°	---	---	2 dist	Bottle- green to pale green	G 3.7	Diss by acids. Fluor med yellow-green in UV. Pleoc, X col's, Y and Z pale green.
~1.562 [^]	1.570	---	1.594	KARPINSKITE (Mg,Ni)2Si2O5(OH)2	---	Z:c = 0- 12° el pos	MCL (?)	---	Blue	H 2.5-3 G 2.6	Cryptocryst. NiO 2.1%.
1.611	1.545	1.586	.044	BANNISTERITE (Na,K)(Mn,Fe,Al)5 (Si,Al)6·15(OH)5·2H2O	Med (30+9°) r < v wk	Z ~ <u>a</u>	MCL	---	Brown	G 2.84	Pleoc, X nearly col's, Y and Z brown.
1.569 [^] 1.600 [^]	1.545	1.586	.044	NONTROLITE (Smectite grp) Na0.33Fe2+(Si,Al)4 0·10(OH)2·xH2O	Small to med r < v	Y = <u>b</u> Z ~ <u>c</u>	MCL u mass	001 perf	Dark brown to olive- green	H 1-2 G 2.50	Dec by acids. Pleoc, X pale yellow, Y olive- green, Z yellow green.
(1.600)	1.559	1.586	.027	GLAUCONITE var Skelite (Mica grp) (K,Na)(Al,Fe,Mg)2 (Si,Al)4·10(OH)2	Small r < v	Y = <u>b</u> X:c = 10°	MCL	001 perf	Cols, blue, green	H 2 G 2.6 fus	Dec by acids. FeO ₂ 6.4, Al ₂ O ₃ 18.2, FeO 2.6%.

1.567 ^	1.540	---	1.592	.052	ALUMOHYDROCALCITE ("Karpovichite") Ca(Al,Cr) ₂ (CO ₃) ₂ (OH) ₄ •3H ₂ O	---	X = b ext 7°-10°	MCL fib	100 perf	Pink	H 2.5 G 2.23 infus	Diss by acids with eff. Cr ₂ O ₃ 8.8%.
1.590 ^	1.52	1.587	1.613	.093	LANTHANITE (La,Ce,Nd) ₂ (CO ₃) ₃ •8H ₂ O	62° r < v wk	X = b Y = $\frac{b}{c}$ el clv pos	ORTH platy	010 perf	Cols, pink	H 2.5-3 G 2.69 infus	Diss by acids with eff. Tw and comp pl 101.
1.575 ^	1.560	1.587	1.595	.035	TOBELITE (Mica grp) (NH ₄ ,K)Al ₂ Si ₃ Al ₁₀ 10 (OH) ₂	30° (56+8°)	el pos	MCL	001 perf	Yellow- green	---	---
1.571 ^	1.572 1.590	1.587	1.600	.028	ILLITE (Mica grp) (K,H ₃ O)(Al,Mg,Fe) ₂ (Si,Al) ₄ 10[(OH) ₂ ,H ₂ O] (OH) ₂	Small to med (85+9°)	X ~ c el clv pos	MCL	001 perf	Cols, brown, green	H 1-2 G 2.7	Slightly dec by acids.
1.574	1.587	1.599	1.600	.025	KRAUSKOPFITE BaSi ₂ 4(OH) ₂ •2H ₂ O	88° r > v dist	X = b Y:a = 6°	MCL	---	Cols	G 3.14 F 5	Dec by acids.
1.578	1.587	1.595	1.600	.017	LEIGHTONITE K ₂ Ca ₂ Cu(SO ₄) ₄ •2H ₂ O	60° (86+13°) r > v str	X ~ b Y ~ $\frac{c}{a}$ Z ~ $\frac{a}{c}$	TCL ps orth pris	---	Pale blue	H 3 G 2.95	Diss by acids. Multiple tw on 100 and 010.
1.548	1.588	1.600	1.600	.052	GORDONITE MgAl ₂ (PO ₄) ₂ (OH) ₂ •8H ₂ O	Med r > v	X ~ b	TCL tab	010 perf 100 fair	White to pale green	H 3.5 G 2.23 F 3	Diss by acids. Fe ⁺³ present.
1.550	1.588	1.590	1.590	.040	COCONINOITE Fe ₂ Al ₂ (UO ₂) ₂ (PO ₄) ₄ (SO ₄)(OH) ₂ •20H ₂ O	40° (19+8°)	---	MCL aggregates	---	Creamy yellow	Soft G 2.70	Diss by acids. Pleoc., X col., Y and Z pale yellow. Not fluor in UV.
1.578 ^	1.619 ^	1.583	1.588	.006	KAEMMERITE (Chlorite grp) (Mg,Al,Cr) ₆ (Si,Al) ₄ 10 (OH) ₈	8° r > v wk	X ~ c el pos	MCL	001 perf	Rose, Tilac	H 2.5 G 2.72 F diff	Insol in HCl. FeO 2.0, Cr ₂ O ₃ 7.9%.
1.561 ^	1.600	1.559	1.588	.029	BEIDELITE (Smectite grp) Na _{0.33} (Al,Fe) ₂ (Si,Al) ₄ 0 ₁₀ (OH) ₂ •xH ₂ O	Small to med	X = c Z = $\frac{a}{c}$	MCL earthy	001 perf	White to red	G 2.1 infus	Dec by acids.
1.556	1.589	1.601	1.601	.045	PYROPHYLITE Al ₂ Si ₄ 10(OH) ₂	62° r > v wk	Z = b X:c = 10° el clv pos	MCL	001 perf	White, brown, yellow	H 1-2 G 2.7-2.9 infus	Nearly insol in acids.
1.583	1.589	1.594	1.594	.011	PHARMACOLITE CaHAsO ₄ •2H ₂ O	79° r > v	Z = b X:c = -29°	MCL acic	010 perf	Cols	H 2-2.5 G 2.5-2.7 F 2.5	Diss by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ calc ($2V_x$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.583 1.589 pos	1.583	1.589	1.593	.010	CELSIAN (Feldspar grp) $\text{BaAl}_2\text{Si}_2\text{O}_8$	88°	Z:c on 100° = 26°	MCL	001 perf 010 good	Col s	H 5.5 G 3.35	Not tw (Cnq).
1.582 1.595	1.563	1.590	1.594	.031	MUSCOVITE var Phencite (Mica grp) $\text{KAl}_2(\text{Si}_3\text{Al})_0\text{Al}_{10}(\text{OH})_2$	33° r > v	$Z = \frac{b}{c}$ $X \sim \frac{c}{e}$ el pos	MCL ps hex tab	001 perf	Pale brown, green	H 2.5-3 G 2.82 F 5.5	Insol in acids. FeO 1.5, FeO 1.9, MgO 4.0%.
1.525	1.590	1.593	.068		BRENKITE $\text{Ca}_2\text{CO}_3\text{F}_2$	27°	X = $\frac{c}{a}$ Y = $\frac{b}{a}$ el cTv pos	ORTH laths	---	Col s	H 5 G 3.10	Diss by HCl with eff.
(1.550) (1.614)	1.57	1.59	1.60	.03	SAUCONITE (Smectite grp) $\text{Na}_0.33\text{Zn}_3(\text{Si},\text{Al})_4\text{O}_{10}$ $(\text{OH})_2\cdot 4\text{H}_2\text{O}$	0-20°	Y = $\frac{b}{a}$ el cTv pos	MCL mass	001 perf	Col s	H 2 G 2.7	Dec by acids.
1.572	1.590	1.601	.029		KOLBECKITE $\text{ScPO}_4\cdot 2\text{H}_2\text{O}$	60° (75+8°) r > v wk	X = $\frac{c}{a}$ Z = $\frac{c}{a}$	MCL pris	010 good fr conch	Blue-gray	H 3.5-4 G 2.39 F easy	Diss by acids.
1.587	1.532	1.590	1.614	.082	LANTHANITE-(ND) (Nd,La) ₂ (CO ₃) ₃ ·8H ₂ O	61°	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH	010 perf 101 good	Col s	G 2.8 infus	Diss by HCl with eff.
1.575	1.590	1.601	.026		DALYITE $\text{K}_2\text{ZrSi}_6\text{O}_{15}$	72° r > v wk	X:c = 7°	TCL pris	101, 010 good	Col s	H 7.5 G 2.66 F 3	Gel with acids.
1.630	1.575	1.590	1.594	.019	CARPHOLITE (Mn,Fe)Al ₂ Si ₂ O ₆ (OH) ₄	50° r > v str	X = $\frac{b}{c}$ Z = $\frac{c}{a}$ el pos	ORTH fib c	Pris at 68°	Pale yellow	H 5-6 G 3.04 F 3.5	Nearly insol in acids. Pleoc, X and Y pale yellow, Z cols. MnO 16.1, FeO 0.6, Fe ₂ O ₃ 2.4%.
1.581	1.590	1.591	.010		TUSCANITE $\text{K}(\text{Ca},\text{Na})_6(\text{Si},\text{Al})_{10}\text{O}_{22}$ $[\text{SO}_4,\text{CO}_3,(\text{OH})_2\cdot \text{H}_2\text{O}$	40°	Z = $\frac{b}{c}$ X:c = 40°	MCL platy	100 dist	Col s	H 5.5-6 G 2.83	Compare Latiumite.
1.582 1.609	1.590	1.591	.009		LATTIMITE (Ca,K) ₈ (Al,Mg,Fe) (Si,Al) ₁₀ O ₂₅ (SO ₄)	40° r > v str	Z = $\frac{b}{c}$ X:c = 16- 28°	MCL tab	100 perf	White	H 5.5-6 G 2.93 F easy	Gel with acids. Tw pl 100. Mottled extinction.
1.585	1.572	1.591	1.598	.026	PRICEITE $\text{Ca}_4\text{B}_{10}\text{O}_{19}\cdot 7\text{H}_2\text{O}$	32° (65±10°)	X:plates = 65°	TCL (?) rhombic plates	110, 001 perf	White, chalky	H 3-3.5 G 2.42 F 2-3	Diss by acids.

1.569	<u>1.592</u>	1.620	.051	LOKKAITITE (Y,Ca) ₂ (CO ₃) ₃ •2H ₂ O	(94+4°)	Z = $\frac{c}{e}$ el pos	ORTH fib	---	White	Infus
1.567	<u>1.592</u>	1.596	.030	URANOSPINITE (Autunite grp) Ca(UO ₂) ₂ (AsO ₄) ₂ •10H ₂ O	25-35° (42+10°) r > v mod	X = $\frac{c}{e}$ el pos	TET plates	001 perf 100 dist	Lemon-yellow	H 2-3 G 3.45 fus
1.573	<u>1.592</u>	1.599	.026	GALDONNAVITE Na ₂ ZrSi ₃ O ₉ •2H ₂ O	59°	X = $\frac{a}{d}$ Y = $\frac{b}{d}$ r < v wk	ORTH	Conch	Col s	H 6 F 3 G 2.67
1.565	<u>1.592</u>	1.597	.025	VARISCITE, ferrian (Al,Fe)PO ₄ •2H ₂ O	Med large (55+10°)	X = $\frac{a}{d}$ Z = $\frac{b}{d}$ r < v wk	ORTH	010 good 001 poor	White, green	H 4.5 G 2.57- 2.75 infus
1.586	<u>1.592</u>	1.592	.011	TORBERNITE (Autunite grp) Cu(UO ₂) ₂ (PO ₄) ₂ •8-12H ₂ O	Small r > v	X = $\frac{c}{e}$ el pos	TET tab	001 perf	Emerald-green	H 2-2.5 G 3.22 F easy
1.581	<u>1.592</u>	1.592	.011	TORBERNITE (Autunite grp) Cu(UO ₂) ₂ (PO ₄) ₂ •8-12H ₂ O	Small r > v	X = $\frac{c}{e}$ el pos	MCL	001 perf	Yellow, brown	H 4 F 3.8
1.563	<u>1.593</u>	1.593	.030	GANOPHYLLITE (Na, K)(Mn, Fe, Al) ₅ •2H ₂ O (Si, Al) ₆ •15(OH) ₅	~ 0° r < v	X = $\frac{c}{e}$ el pos	TCL ps orth tab 001	001 perf 100 dist	White to pale yellow	H 3.5-4 G 2.96 F 3
1.570	<u>1.593</u>	1.595	.025	LEUCOPHANITE (Ca, Na) ₂ BeSi ₂ (O, OH, F) ₇	4°	X = $\frac{c}{e}$ Z = $\frac{b}{d}$ el pos	MCL	001 perf 100 dist	White to pale yellow	H 3.5-4 G 2.96 F 3
1.581	<u>1.593</u>	1.602	.021	TREMOLITE (Amphibole grp) Ca ₂ Mg ₅ Si ₈ O ₂₂ (OH, F) ₂	86°	Y = $\frac{b}{e}$ Z:c = 21° el pos	MCL	110 good	Col s	H 5.5 G 3.02
1.520	<u>1.594</u>	1.595	.075	HYDRODRESSERITE BaAl ₂ (CO ₃) ₂ (OH) ₄ •3H ₂ O	17°	el pos	TCL radiating spherulitic	010, 210 perf	White	H 3-4 G 2.80
1.572	<u>1.595</u>	1.614	.042	JOHANNITE Cu(UO ₂) ₂ (SO ₄) ₂ (OH) ₂ • 6H ₂ O	90° (83+6°) r > v str	X ~ b Y:c = 5-8° disp str el c	TCL tab	100 good	Green to greenish-yellow	H 2.5 G 3.32 infus
1.590	<u>1.595</u>	1.609	.040	MUSCOVITE var Phengite (Mica grp) K(Al, Mg, Fe) ₂ (Si ₃ Al) ₀ 10(OH) ₂	40° (71+6°) r > v	Z = $\frac{b}{e}$ X ~ $\frac{c}{e}$ el pos	MCL ps hex tab	001 perf	Emerald-green	H 2.5-3 G 2.82 F 5.5
1.581	<u>1.596</u>	1.601	.020	BALIPHOLITE BaMg ₂ LiAl ₃ Si ₄ O ₁₂ (OH) ₈	70°	X = $\frac{b}{e}$ Y = $\frac{c}{e}$	ORTH fib c	010 perf 100, 110 dist	Yellow- ish-white	G 3.4

										Pleoc., X cols, Y and Z yellow. Fluor bright lemon-yellow in UV.
										Gel with acids.
										Diss by acids. Pleoc., X cols to pale green, Y and Z pale to dark green. Not fluor in UV.
										Gel with acids. Pleoc. near X yellow, Y and Z
										tw 110. Luminesces violet-blue in long- wave UV.
										Insol in acids. Data for synth compnd.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.560	<u>1.597</u>	1.600	<u>γ</u>		THADEUITE $\text{Ca}(\text{Fe}^{+2}, \text{Mn})_2\text{Mg}(\text{PO}_4)_2$ (OH, F) ₂	33°	$X = \frac{c}{b}$ $Y = \frac{c}{d}$	ORTH	010 perf	Yellow-orange	H 4 G 3.25	---
1.575	<u>1.597</u>	1.598	.023		CHRYSOCOLLA $(\text{Cu}, \text{Al})_2\text{H}_2\text{Si}_2\text{O}_5(\text{OH})_4$ $\times \text{H}_2\text{O}$	Small	---	MCL fib u mass	---	Bluish-green, sky-blue	H 2.4 G 2.4 infus	Dec by acids. Optics highly variable.
1.596	<u>1.597</u>	1.597	.001		EMELEUSITE (Ossumilite grp)	0-30° $r > v$ str	$X = \frac{b}{a}$ $Y = \frac{b}{a}$	ORTH tab b	---	Col s	H 5-6 G 2.76	---
1.516	<u>1.598</u>	1.621	.105		AMARANTITE $\text{FeSO}_4(\text{OH}) \cdot 3\text{H}_2\text{O}$	30° $r < v$ str	$X \sim \perp 100$ $Y:z = 47^\circ$	TCL acidic columnar	010, 100 perf	Red to orange-red	H 2.5 G 2.19 F 4.5-5	Dec by H_2O , diss by acids. Pleoc., X cols, Y pale yellow Z reddish-brown.
1.605	<u>1.598</u>	1.598	.041		BIOTITE (Mica grp)	10-13° $r < v$	$Y = \frac{b}{c}$ $X \sim \frac{c}{e}$ el pos	MCL plates	001 perf	Dark brown to green	H 2-2.5 G 2.95 F 3-4	Pleoc., X yellow, Y brown, Z dark brown, abs X < Y < Z. FeO 13.8, Fe_2O_3 2.1, Mn 0 0.4%.
1.584 1.617	<u>1.598</u>	1.599	.035		K(Mg, Fe) ₃ (Al, Fe) ₃ $\text{Si}_3\text{O}_{10}(\text{OH}, \text{F})_2$	38° $r > v$	$Z = \frac{b}{c}$ $X \sim \frac{c}{e}$ el pos	MCL ps hex tab	001 perf	Pale brown, greenish, red	H 3-4 G 2.87 F diff	Insol in acids. FeO 0.9, Fe_2O_3 2.8%.
1.595 1.608	<u>1.598</u>	1.600	.035		MUSCOVITE (Mica grp) $\text{KAl}_2(\text{Si}_3\text{Al})_0\text{Al}_1(\text{OH})_2$	77°	---	TCL	100 perf 110 good	White	H 5.5-6 G 3.10 F easy	Diff diss by acids. Li_2O 9.6, Na_2O 1.2, F 3.2%.
1.585 1.607	<u>1.598</u>	1.606	.021		MONTEBRASITE $\text{LiAlPO}_4(\text{OH}, \text{F})$	---			---	White	H 3.5 G 2.58 F 2	Insol in acids.
1.595 1.600	<u>1.598</u>	1.605	.019		HOWLITE $\text{Ca}_2\text{B}_5\text{Si}_1\text{O}_9(\text{OH})_5$	Large (74+12°)	$X = \frac{b}{d}$ $Z:a = 55^\circ$	MCL tab 100	---			Diss by acids.
1.584	<u>1.598</u>	1.602	.018		MILLISITE $(\text{Na}, \text{K})\text{CaAl}_6(\text{PO}_4)_4$ $(\text{OH})_9 \cdot 3\text{H}_2\text{O}$	Med	$X = \frac{c}{e}$ el neg	MCL fib c	---	White, gray	H 5.5 G 2.83 F 3.5	
1.582	<u>1.598</u>	1.599	.010		HOPEITE $\text{Zn}_3(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	37° $r < v$ wk	$X = \frac{a}{c}$ $Y = \frac{a}{c}$	ORTH tab pris	010 perf 100 good	Col s, White	H 3.5 G 3.05- 3.12 F easy	Diss by acids.

1.584 ~1.613	<u>1.554</u>	<u>1.599</u>	1.602	.048	TALC (Mg,Fe) ₃ Si ₄ O ₁₀ (OH) ₂	X ~ c Z = b el pos	MCL	001 perf	Green	H 1 G 2.87 infus	Insol in acids. Luster pearly. FeO 13.4, Fe ₂ O ₃ 1.7%.
1.586 ~1.618	1.589	<u>1.600</u>	1.610	.021	NONTRONITE (Smectite grp)	40° r < v	MCL	001 perf	Dark brown to green	H 1-2 G 2.27	Dec by acids. Pleoc in brown or green.
1.586 ~1.618	1.585	<u>(1.600)</u>	1.600	.015	GLAUCONITE (Mica grp) (K,Na)(Al,Fe,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	Small r < v	MCL	001 perf	Green, blue, cols	H 2 G 2.59 fus	Dec by acids. Fe ₂ O ₃ 14.1, Al ₂ O ₃ 8.9, FeO 1.6%.
1.518	1.588	<u>1.601</u>	1.601	.083	DRESSERITE Ba ₂ Al ₄ (CO ₃) ₄ (OH) ₈ •3H ₂ O	30-40°	X = a Z = c el pos	---	White	H 2.5 G 2.96 infus	Diss in acids with eff.
1.588	1.601	1.610	1.610	.022	PARAUMBITITE K ₃ Zr ₂ HSi ₆ 18•7H ₂ O	82°	X = c Y = d el pos	0RTH fib	---	---	
1.595	1.601	1.604	1.604	.009	HURLBUTITE CaBe ₂ (PO ₄) ₂	70° r > v wk	X = b Y = c el pos	0RTH pris	010 mic 100 perf	H ~ 4.5 G 2.59	Slowly diss by acids. Striated on 110.
1.526	1.586	<u>1.602</u>	1.602	.076	ANTHONYITE Cu(OH,Cl) ₂ •3H ₂ O	3°	Y = b Z:c = 13° el pos	MCL pris	100 good	Lavender sectile	Diss by acids. Pleoc, X lavender, Y and Z deep smoky blue, abs Z = Y > X.
1.586	1.602	1.606	1.606	.020	SPENCERITE Zn ₄ (PO ₄) ₂ (OH) ₂ •3H ₂ O	49° r > v mod	Z = b X ~ a el pos	MCL tab	100 perf 010 good 001 fair	H 3 G 3.14 F easy	Diss by acids. Tw and comp pl 100, poly.
1.596	1.602	1.603	1.603	.007	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ •4-6H ₂ O	0-15°	Z = c el pos	TET tab	---	H 2-2.5 G 3.5 fus	Diss by acids. Pleoc. X cols, Y and Z yellow. Fluor yellow-green in UV.
1.622	1.683 ~1.634	<u>1.603</u>	1.655	.048	STILPNOLEMITE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) 10- Si ₁₂ O ₃₀ (OH) ₁₂	~ 0°	X = b Y ~ c el pos	MCL also TCL	001 perf	H 3-4 G 2.78 F 4-5	Dec by acids. Pleoc, X pale yellow, Y and Z olive-brown. FeO 21.6, Fe ₂ O ₃ 6.4, MnO 1.4%.
1.598 ~1.605	1.571	<u>1.603</u>	1.606	.035	PARAGONITE (Mica grp) NaAl ₂ (Si ₃ Al) ₁₀ (OH) ₂	30° r > v	Z = b el pos	MCL ps hex plates	001 perf	H 2.5 G 2.87 F diff	Insol in acids.
1.598 ~1.607	1.587	<u>1.603</u>	1.612	.025	BERTRANDITE Be ₄ Si ₂ O ₇ (OH) ₂	70-75° r < v wk	X = a Z = c el pos	0RTH col um- nar	110 perf 100, 010, 001 good	H 6.5 G 2.60 infus	Insol in acids. Tw 011.
1.585	1.603	1.604	1.604	.019	WIGHTMANITE Mg ₅ (BO ₃) ₂ (OH) ₅ •2H ₂ O	33° r < v	Z:b = 5° el pos	MCL ps hex pris	101 perf T01 good	H 5.5 G 2.59 infus	Diff sol in cold dil acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula		$2V_x$ calc ($2V_x$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ	Biref								
1.596	1.603	1.606	1.606	.010	WEEKSITE $K_2(UO_2)_2Si_6O_{15} \cdot 4H_2O$	60° r > v str	$X = \frac{b}{c}$ $Y = \frac{c}{a}$ el cTw pos	ORTH ps tet	010 perf	Pale yellow-green	H 1.5 G 4.1	Pleoc, X cols, Y pale yellow-green, Z yellow-green. Not fluor in UV.
1.598	1.603	1.607	1.607	.009	HILLEBRANDITE $Ca_2Si_3(OH)_2$	Med r < v str	Z = $\frac{c}{a}$ el pos	MCL fib	110	White	H 5-6 G 2.69 F diff	Gel with acids. Ultrablu interf colors.
1.577	~1.595	1.603	1.604	~.009	ANTIGORITE (Serpentine grp) (Mg, Fe) ₃ Si ₂ O ₅ (OH) ₄	Small	X = $\frac{c}{a}$ el cTw pos	MCL	001 perf	Brown	H 3 G 2.4-2.6 F diff	Dec by HCl. FeO 21.0, Fe ₂ O ₃ 0.5, MnO 2.5%.
1.582	1.604	1.609	1.609	.027	NIAHITE (NH ₄) ₂ (Mn ⁺² , Mg)PO ₄ · H ₂ O	(50±11°)	---	ORTH	---	Pale orange	Soft G 2.39	---
1.585	1.604	1.612	1.612	.027	NAKAURITE Cu ₈ (SO ₄) ₄ (CO ₃)(OH) ₆ · 4H ₂ O	el pos	ORTH fib	---	---	Sky-blue	G 2.39	Pleoc, X cols, Y light greenish blue, Z pale sky-blue to light blue.
1.51	1.605	1.611	1.611	.101	AMARANTITE FeSO ₄ · 3H ₂ O	28° r < v str	$Y:c = 47^\circ$ $X \perp 100$ acic columnar	010, 100	Red, orange-red	H 2.5 G 2.11 F 4.5-5	Dec by H ₂ O, diss by acids.	
1.598	1.605	1.611	1.611	.031	HEINRICHITE (Autunite grp) Ba(UO ₂) ₂ (AsO ₄) ₂ · 10-12H ₂ O	0-20°	---	TET	001 perf 100 dist	Yellow to green	Pleoc, X cols, Y pale yellow. Fluor green to yellow in UV.	
1.574	1.605	1.605	1.605	.031	MONTDORITE (Mica grp)+2 (K, Na) ₂ (Fe ⁺² , Mn, Mg) ₅ Si ₈ O ₂₀ (F, OH) ₄	0-3°	---	MCL	001 perf	Green to brownish-green	---	
1.580	1.605	1.605	1.605	.025	MAGNESIO-ANTHOPHYLLITE (Amphibole grp) Mg ₂ Si ₈ O ₂₂ (OH) ₂	65° (75±12°)	$Y = \frac{b}{c}$ el pos	ORTH	210 perf	cols to brown	H 5.5-6 G 3.0	Insol in acids. Fe ₂ O ₃ 0.25, Al ₂ O ₃ 1.6%.
1.625	1.593	1.605	1.613	.020	BRIANITE Na ₂ CaMg(Po ₄) ₂	64°	---	MCL	---	cols	H 4-5 G 3.1	Poly tw, ext angle between lam 2-3°. Meteorite mineral.
1.598	1.605	1.608	1.608	.010	VERMICULITE (Mg, Fe, Al) ₃ (Si, Al) ₄ O ₁₀ · (OH) ₂ · 4H ₂ O	Small	X \perp plates el pos	MCL	001 perf	Green, brown	H 1.5 G 2.77 F 5	Dec by HCl. Exfoliates when heated. Fe ₂ O ₃ 4.5%.
1.581	1.560	1.607	1.607	.047								
1.648												

1.598 1.612	<u>1.607</u>	1.614	.021	AMBLYGONITE (Li, Na)AlPO ₄ (F, OH)	61-70° r > v	TCL	100 perf 110 good	White	H 5.5-6 6 3.05 F easy
1.583	<u>1.608</u>	1.633	.050	CHUDDOBAITE (Mg, Zn)H ₂ (AsO ₄) ₄ • 10H ₂ O	79° (89±5°)	TCL	010, 100 good	Pink	H 2.5-3 6 2.94
1.544	<u>1.608</u>	1.608	.064	WONESITE (Mica grp) (Na, K)(Mg, Fe) ₆ (Si, Al) ₈ O ₂₀ (OH, F) ₄	0-5°	MCL	001 perf	Cols to brown	---
1.598 1.609	<u>1.608</u>	1.611	.040	MUSCOVITE (Mica grp) K(Al, V) ₂ (Si ₃ Al) ₁₀ (OH) ₂	34° r > v	MCL	001 mic ps hex	Green	H 3 6 2.93 F 5
1.590	<u>1.608</u>	1.611	.021	LAWSONBAUERITE (Mn, Mg) ₉ Zn ₄ (SO ₄) ₂ (OH) ₂₂ •8H ₂ O	42° r > v str	MCL	---	Col to white	H 4-5 6 2.87
---	<u>1.608</u>	---	.011	MONSMEDITE K ₂ O•Ti ₂ O ₃ •8SiO ₃ •15H ₂ O	~ 52°	ORTH ps cub	cub 2	Dark green to black	6 3.00 H 2
1.608 1.620	<u>1.609</u>	1.611	.038	MUSCOVITE var Phenogite (Mica grp) K(Al, Mg) ₂ (Si ₃ Al) ₁₀ (OH) ₂	31° r > v	MCL ps hex	001 perf	Pale brown	H 3 6 2.88 F 5
1.597	<u>1.609</u>	1.615	.018	MAGBASITE KBa(Al, Sc)(Mg, Fe) ₆ Si ₆ O ₂₀ F ₂	70°	Z:c = 10°	Acic	---	Col, rose- violet
1.590	<u>1.603</u>	1.615	.012	LATIUMITE (Ca, K) ₈ (Al, Mg, Fe) (Si, Al) ₁₀ O ₂₅ (SO ₄)	72-83° r > v str	MCL tab	100 perf	White	H 5-6 6 2.93 F easy
1.596	<u>(1.610)</u>	1.619	.023	UMBITE K ₂ ZrSi ₃ O ₉ •H ₂ O	80°	ORTH platy	100 perf	Col to yellowish	H 4.5 6 2.79
v 1.626	<u>1.600</u>	1.610	.016	SAINFELDITE Ca ₅ H ₂ (AsO ₄) ₄ •4H ₂ O	80°	X = b Y:c = 20°	---	Col, light pink	6 3.04
1.574	<u>1.603</u>	1.617	.014	BASSETITE (Meta-autunite grp) Fe(OH) ₂ (PO ₄) ₂ •8H ₂ O	~ 90° r > v str	X = b Y:c = 18.5° el clv pos	010 perf MCL ps tet	Olive- green to brownish	H 2.5 6 3.4

Diss by acids. Pleoc,
X and Y yellow, Z dark
olive-brown. Not fluor
in UV.

Diss by acids. Pleoc,
X and Y yellow, Z pale blue.
V₂O₃ 3.4, Cr₂O₃ 0.2,
FeO 0.5, MgO 1.1%.

Insol in acids. Pleoc,
X cols, Z pale blue.
V₂O₃ 1.0%,
TiO₂ 1.0%.

Insol in acids. FeO
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Gel with acids. Tw p1
100. Mottled extinction.

Pleoc, X and Y cols, Z
lilac.

Insol in acids. FeO
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Insol in acids. Pleo
2.8, MgO 4.1, Fe₂O₃
4.1, TiO₂ 1.0%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Brief	MINERAL and formula	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.603 \wedge	1.607	<u>1.610</u>	1.616	.009	BUCHWALDITE NaCaP ₀ ₄	---	el pos	ORTH	One cle	Col s	H < 3 G 3.21	Biax pos (?) .
1.605 \wedge	1.605	<u>1.61</u>	1.612	.007	HILLEBRANDITE Ca ₂ Si _{0.3} (OH) ₂	51° r < v str	Z = $\frac{c}{b}$ el pos	MCL fib	110	White	H 5-6 G 2.69 F diff	Gel with acids. Shows ultra-blue interf colors.
1.593 \wedge	1.571	<u>1.611</u>	1.612	.041	GANOPHYLLITE (Na, K)(Mn, Fe, Al) ₅ (Si, Al) ₆ 15(OH) ₅ •2H ₂ O	~ 0° r < v	X = $\frac{c}{b}$ el pos	MCL	001 perf	Cinnamon-brown	H 4 G 2.8 F 3	Gel with acids. Pleoc, X dark yellow-brown, Y and Z pale yellow-brown.
1.586 \wedge	1.574	<u>1.611</u>	1.612	.038	BANNISTERITE (Na, K)(Mn, Fe, Al) ₅ (Si, Al) ₆ 15(OH) ₅ •2H ₂ O	22° r < v wk	Z ~ a	MCL	---	Brown	G 2.84	Pleoc wk, X cols, Y and Z pale yellow.
1.609 $\overline{\wedge}$ 1.620	1.592	<u>1.612</u>	1.621	.029	HERDERITE CaBeP ₀ ₄ (F, OH)	67-75° r > v inclined	Y = $\frac{b}{c}$ Z:c = -3°	MCL pris	110 poor	Col s to yellowish	H 5-5.5 G 3.01- 3.08 F diff	Diss by acids. Tw pl 001 penet.
1.585	<u>1.612</u>	1.612	.027	SODIUM URANOSPINITE (Meta-autunite grp) (Na ₂ , Ca)(UO ₂) ₂ (AsO ₄) ₂ • 5H ₂ O	Small	---	TET tab	001 perf 010, 100 dist	Yellow to yellow	H 2.5 G 3.85	Diss by acids. Pleoc, X cols, Y and Z yellow. Fluor yellow-green in UV.	
1.586 \wedge	1.586	<u>1.612</u>	1.613	.027	MARGARITE Sodium analogue (Mica grp) CaAl ₂ (Si ₂ Al ₁) ₁₀ (OH) ₂	50° r < v	Z = $\frac{b}{a}$ Y:a = 7° el pos	MCL	001 perf	Col s	H 4 G 3.06 F fus	Insol in acids. Na ₀ 5.6, Ca0 3.3%, Li not detd.
1.607 \wedge	1.598	<u>1.612</u>	1.621	.023	MONTEBRASITE LiAlP ₀ ₄ (OH, F)	82° r > v	---	TCL	100 perf 110 good	White	H 5.5-6 G 3.05 F easy	Diff diss by acids. Li 20 9.2, Na ₂ 0 0.6, F 4.4%.
1.606 $\overline{\wedge}$ 1.615	1.596	<u>1.612</u>	1.615	.019	PHOSPHOPHYLLITE Zn ₂ (Fe, Mn)(PO ₄) ₂ •4H ₂ O	40-50° r > v	Z = $\frac{b}{c}$ Y:c = 50°	MCL thick tab	001 perf 010, T02	Col s to bluish-green	H 3-3.5 G 3.13 F easy	Diss in acids. Tw on 100 common, sometimes poly. Fluor violet in short-wave UV.
1.571	<u>1.613</u>	1.623	.052	KTENASITE (Cu, Zn) ₅ (SO ₄) ₂ (OH) ₆ • 6H ₂ O	51°	Z = $\frac{b}{c}$ X ~ $\frac{c}{a}$	MCL	---	Blue-green	H 2-2.5 G 2.97	Diss by acids and NH ₄ OH. Pleoc, X cols, Y bluish-green, Z light green.	

1.599 ~1.630	1.580	---	1.615	.035	MINNESOTAITE (Fe ⁺² ,Mg) ₃ Si ₄ O ₁₀ (OH) ₂	Small	el pos	MCL	001 perf	Cols to yellow-green	G 3.01	Pleoc, X pale yellow, Y and Z pale blue-green.
1.560	1.614	1.615	1.615	.055	PREISWERKITE (Mica grp) NaMg ₂ Al(Al ₂ Si ₂) ₁₀ (OH) ₂	5-7°	---	MCL	001 perf	Pale green	H 2.5 G 2.96	---
1.585	1.614	1.614	1.614	.029	VIMSITE CaB ₂ O ₂ (OH) ₄	28°	el pos	MCL	One clv	Cols	H 4 G 2.54	Diss by acids.
1.586	1.614	1.621	1.621	.035	CAYSICHITE (Y,Ca) ₄ Si ₄ O ₁₀ (CO ₃) ₃ 4H ₂ O	53-73°	X = b Y = a el pos	ORTH radiating	---	Cols to yellow	H 4.5 G 3.03	Diss by cold HCl with eff.
1.59	1.575	(1.614)	1.615	.040	SAUCONITE (Smectite grp) Na _{0.33} Zn ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	0-20°	Y = b	MCL mass	001 perf	Cols, white, yellow	H 2 G 2.7	Dec by acids. ZnO 23.1, Al ₂ O ₃ 17.0%.
1.595	1.614	1.614	1.614	.019	SEMENOVITE (Ca,Ce,La,Na) ₁₀₋₁₂ (Fe,Mn)(Si,Be) ₂₀ (0,OH,F) ₄₈	0-40°	---	ORTH ps tet	Uneven	Cols	H 3.5-4 G 3.14	Tw 110.
1.600	1.615	1.629	1.629	.029	LEHILITE (Na,K) ₂ Ca ₅ Al ₈ (PO ₄) ₈ (OH) ₁₂ ·6H ₂ O	Large (67+8°)	el neg	MCL fib	---	White	H 5.5 G 2.89	Diss by acids.
1.593 ~1.632	1.600	1.615	1.627	.027	TREMOLITE (Actinolite ser, Amphibole grp) Ca ₂ (Mg,Fe)Si ₈ O ₂₂ (OH) ₂	79° (63+8°) r < v wk	Y = b Z:c = 15° el pos	MCL	110 perf at 124°	Cols to green	H 5.5-6 G 2.98	Insol in acids. FeO 0.6%.
(1.627)	1.605	1.615	1.622	.017	RICHTERITE (Amphibole grp) Na ₂ Ca(Mg,Ti,Al) ₅ Si ₈ O ₂₂ (OH) ₂	Med large r < v wk	Y = b Z:c = 25° el pos	MCL	110 perf at 129°	White	H 5.5-6 G 3.07	Insol in acids.
(1.559)	1.616	1.624	1.624	(.065)	PHARALUMITE Al ₂ (UO ₂) ₃ (PO ₄) ₂ (OH) ₂ · 10H ₂ O	40°	X = b Y ~ c el pos	MCL pris	---	Lemon-yellow	H 3 G 3.5	Pleoc, X cols, Y and Z pale yellow. Not fluor in UV.
1.624	1.597	1.616	1.624	.027	SANBORNITE BaSi ₂ O ₅	66°	Z ~ c	TCL platy	001 perf 010 imperf	White	H 5 G 4.19	Dec by acids. Poly tw on 010.
1.598 ~1.640	1.575	1.617	1.621	.046	BIOTITE, var Manganophyllite (Mica grp) K(Mg,Fe ⁺² ,Mn) ₃ (Al,Fe ⁺³)Si ₃ O ₁₀ (OH,F) ₂	30° r < v	Y = b el pos	MCL plates	001 perf	Reddish-brown	H 2.5-3 G 2.98 F 3-4	Pleoc, X brown, Y and Z dark brown. Fe ₂ O ₃ 5.8, FeO 1.2, MnO 6.2%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.600	---	1.620	.020	ALDZHANITE $\text{CaMgB}_2\text{O}_4\text{Cl} \cdot 7\text{H}_2\text{O}$ (?)	---	---	ORTH dipyram	MCL	001 dist	Light pink	H 5-6 G 2.84	Opt char unk.
1.605	<u>1.617</u>	1.620	.014	REINHARDBAUNSTEITE $\text{Ca}_5(\text{SiO}_4)_2(\text{OH},\text{F})_2$	44-50° $r > v$ dist	$Z = \frac{b}{c}$ $X:\underline{c} = 18^\circ$	MCL	001 perf	Green	H 2 G 2.7 fus	---	
(1.600) 1.634	<u>1.618</u>	1.619	.020	GLAUCONITE (Mica grp) (K, Na)(Al, Fe, Mg) ₂	20° $r < v$	$Y = \frac{b}{c}$ $X:\underline{c} = 10^\circ$	MCL	100 perf	Green	Dec by acids. X yellow, Y and Z bluish-green. Fe_2O_3 18.8, Al_2O_3 8.5, FeO 4.0, MgO 3.0%.	Dec by acids. Pleoc., X yellow-green. Fe_2O_3 18.8, Al_2O_3 8.5, FeO 4.0, MgO 3.0%.	
1.636	<u>1.619</u>	1.620	.040	GRANDIERITE $(\text{Mg},\text{Fe})\text{Al}_3(\text{BO}_4)(\text{SiO}_4)_2$	24° $r < v$	$X = \frac{a}{c}$ $Y = \frac{c}{e}$ $\underline{c}:\underline{v}$ pos	ORTH	100 perf 010 poor	Blue	6 2.91 infus	Insol in acids. Pleoc., X pale blue, Y yellow-brown, Z deep blue, abs Z > X > Y.	Insol in acids. Pleoc., X pale blue, Y yellow-brown, Z deep blue, abs Z > X > Y.
1.588 1.637	1.605	<u>1.619</u>	1.619	CLINOCHLORE (Chlorite grp) $(\text{Mg},\text{Fe})_5\text{Al}(\text{Si}_3\text{Al})_10(\text{OH})_8$	~0° $r > v$	$X \sim \frac{c}{e}$ $\underline{c}:\underline{v}$ pos	MCL	001 perf	Olive-green	H 2-3 G 2.8 F 5	Dec by acids. Pleoc., X pale green, Y and Z green.	Dec by acids. Pleoc., X pale green, Y and Z green.
1.611	<u>1.619</u>	1.621	.010	CYMRITE $\text{BaAl}_2\text{Si}_2(0,\text{OH})_8 \cdot \text{H}_2\text{O}$	0-5°	---	MCL ps hex	---	Colts, dark green, brown	H 2-3 G 3.41- 3.45	Slight sol in HCl. Exfoliates when heated.	Slight sol in HCl. Exfoliates when heated.
1.583	<u>1.62</u>	1.633	.050	Unnamed Cu-Zn arsenate	---	---	TCL	2 pinacoidal	Light rose	---	Diss by HCl. Am. Mineral., 44, 1323 (1959).	Diss by HCl. Am. Mineral., 44, 1323 (1959).
1.609 [^]	<u>1.620</u>	1.623	.043	MUSCOVITE, var Phenogite (Mica grp) $(\text{K},\text{Al},\text{Fe},\text{Mg})(\text{Si}_3\text{Al})_10(\text{OH})_2$	35° $r > v$	$Z = \frac{b}{c}$ $X:\underline{c} = 0-5^\circ$ $\underline{e}:\underline{v}$ pos	MCL ps hex plates	001 perf	Colts pale green	H 5 G 2.88	Insol in acids. Pleoc., X col., Y and Z turquoise-blue.	Insol in acids. Pleoc., X col., Y and Z turquoise-blue.
1.584	<u>1.620</u>	1.621	.037	NISSONITE $\text{Cu}_2\text{Mg}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 5\text{H}_2\text{O}$	19° $r > v$ str	$Z = \frac{b}{c}$ $\underline{Y}:\underline{c} = 6^\circ$	MCL tab	100 fair	Bluish-green	H 2.5 G 2.73	Diss by acids. Pleoc., X col., Y and Z turquoise-blue.	Diss by acids. Pleoc., X col., Y and Z turquoise-blue.
1.612 [^]	<u>1.620</u>	1.629	.025	HYDROXYLHERDERITE $\text{CaBePO}_4(\text{OH},\text{F})$	74° $r < v$ inclined	$Y = \underline{b}$	MCL pris	110 poor	Colts	H 5-5.5 G 3.08	Diss by acids. Tw pl 001 penet.	Abnormal dark gray to yellowish interf. colors.
1.610	<u>1.620</u>	1.623	.013	SUOLUNITE $\text{Ca}_2\text{Si}_2\text{O}_5(\text{OH})_2 \cdot \text{H}_2\text{O}$	30-35° (57+19°)	$X = \frac{a}{c}$ $\underline{Y} = \underline{c}$	ORTH	---	White	G 2.68	Abnormal dark gray to yellowish interf. colors.	Abnormal dark gray to yellowish interf. colors.

1.627	<u>1.621</u>	1.622	.010	CARBONATE-FLUORAPATITE [Small ("Dohrnite") (Apatite grp) (Ca, Na, K) ₅ (PO ₄ , CO ₃) ₃ F	~ 0°	X = <u>c</u> el c ₁ v pos	HEX 0001 perf	Col s, yellow	H 5 G 3.04 fus
1.612	<u>1.621</u>	1.622	.004	METATORBERNITE Cu(UO ₂) ₂ (PO ₄) ₂ •8H ₂ O	0-15°	X = <u>c</u> el c ₁ v pos	TET square tab	001 perf Emerald-green	H 2.5 G 3.6 F easy
1.632	<u>1.621</u>	1.622	.026	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ •4-6H ₂ O	40-56° r > v dist	Z = <u>c</u> el c ₁ v pos	TET tab	---	Yellow
1.602	<u>1.604</u>	1.630	.026	VLASOVITE Na ₂ ZrSi ₄ Al ₁₁	0-20° r > v	Opt p1 010	MCL ps tet	010 good Col s, brownish	H 2-2.5 G 3.5 fus
1.607	<u>1.623</u>	1.628	.021	META-UANOCIRRITE (Meta-autunite grp) Ba(UO ₂) ₂ (PO ₄) ₂ •8H ₂ O	18° r < v	X = <u>c</u> el pos	001 perf 100 dist	Yellow-green	H 2-2.5 G 4.08 F 3
1.612	<u>1.623</u>	1.623	.013	EPHESTITE (Margarite ser., Mica grp) NaLiAl ₂ (Si ₂ Al ₂) ₁₀	15-20°	Z = <u>b</u> Y:a = 7° el pos	MCL plates	001 perf pink to brown	H 5-7 G 2.98
1.610	<u>1.623</u>	1.623	.013	FRAIPONTITE (OH, F) ₂	---	el pos	MCL fib	---	Yellow- ish-white
1.612	<u>1.624</u> <u>1.643</u>	1.625	.033	(Zn, Al) ₃ (Si ₄ Al ₂) ₁₀	---	---	Radi- ating	---	Soft fus
---	<u>1.624</u>	1.624	---	Unnamed hydrous copper aluminum sulfate	---	---	TCL	010 perf Col s	Sky-blue
---	<u>1.625</u>	1.645	---	Rather large	---	---	---	---	---
1.614	<u>1.625</u>	1.637	.023	PARAHOPITE Zn ₃ (PO ₄) ₂ •4H ₂ O	~ 90° r < v perc	X ~ <u>a</u>	---	---	H 3.5-4 G 3.31 F easy
1.605	<u>1.615</u> <u>1.630</u>	1.625	.019	GEDRITE (Mg, Fe)Al ₂ (Si ₆ Al ₂) ₂₂	71° (86+12°)	Y = <u>b</u> Z = <u>c</u> el pos	0RTH	210 perf Col s to brown	H 5.5-6 G 2.87
---	<u>1.625</u>	---	.007	SMOLIANINOVITE (Co, Ni, Mg, Ca) ₃ (Fe ⁺³ , Al) ₂ (AsO ₄) ₄ •11H ₂ O	---	el pos	0RTH fib	---	Yellow
1.605	<u>1.626</u>	1.633	.028	JIMTHOMPSONITE (Mg, Fe) ₅ Si ₆ O ₁₆ (OH) ₂	62° r > v wk	X = <u>a</u> Y = <u>b</u> el at 38°	0RTH	210 perf Col s to light pinkish- brown	---
									FeO 12.2, MnO 0.7%.
									Diss by acids. Base divided into biax segments.
									Diss by acids. Anom interf colors. Not fluor in UV.
									Diss by acids. Pleoc, X col s, Y and Z yellow. Fluor yellow-green in UV.

									Pleoc str. Am. Mineral., 57, 1004 (1972).
									Diss by acids. Poly tw common on 100.
									Opt sign unk.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.610 \wedge	1.616	<u>1.626</u>	1.634	.018	SAINFELDITE $\text{Ca}_5\text{H}_2(\text{AsO}_4)_4 \cdot 4\text{H}_2\text{O}$	83°	$X = \frac{b}{c} = 20^\circ$ $Y:C = \underline{\underline{}}$	MCL flat-tened 100	---	Col s, pink	G 3.04	Diss by acids.
1.589	1.627	1.628	.039	CUPRORIVALTE $\text{CaCuSi}_4\text{O}_{10}$	14°	---	TET tab	001 perf	Azure-blue	H 5 G 3.08	Insol in acids. Pale rose, Z blue.	
1.610	<u>1.627</u>	1.633	.023	CLINOHOLMQUISTITE (Amphibole grp) $\text{Li}_2(\text{Mg},\text{Fe}^{+2})_3\text{Al}_2\text{Si}_8\text{O}_{22}(\text{OH},\text{F})_2$	58°	$Y = \frac{b}{c} = 16^\circ$ $Z = \frac{c}{a} = \underline{\underline{}}$ $X:a = \underline{\underline{}}$	MCL pris el pos	110 perf	Blue	G 3.00	---	
1.615 \wedge 1.631	1.616 (1.627)	1.632	.016	RICHTERITE, var Magnophorite (Amphibole grp) $\text{Na}_2\text{Ca}(\text{Mg},\text{Fe}^{+2},\text{Ti})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	70°	$Z:c = 26^\circ$ el pos	MCL pris	110 perf	Col s to brown	H 5-6 G 3.12	Insol in acids. Fe_2O_3 0.6, Al_2O_3 1.7, FeO 0.6, TiO_2 3.5%.	
1.640 \vee	1.619	<u>1.627</u>	1.629	.012	FEDOROVSKITE (Rowellite ser) $\text{Ca}_2(\text{Mg},\text{Mn})_2\text{B}_4\text{O}_7(\text{OH})_6$	Small $r < v$ str	$Z = \frac{b}{c} = \underline{\underline{}}$ $Y = \frac{c}{b} = \underline{\underline{}}$ el cTV pos	ORTH fib	100 perf	Col s to pink	H 4.5 G 2.65	Diss by acids. Pleoc., X col s, Z yellow. MnO 5%.
1.621 \wedge	1.622	<u>1.627</u>	1.627	.005	CARBONATE-FLUORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3(\text{F},\text{OH})$	0-25°	---	HEX u mass	0001 poor	Col s to brown	H 5 G 3.1-3.2	Diss by acids with slight eff.
(1.58) \wedge	1.598	<u>1.628</u>	1.654	.056	LAUSENITE $\text{Fe}_2(\text{SO}_4)_3 \cdot 6\text{H}_2\text{O}$	Large (64±4°)	$X:c = 27^\circ$ el pos	MCL fib $\underline{\underline{}}$	---	Col s	---	Sol in H_2O .
1.602	<u>1.628</u>	1.632	.030	SWITZERITE (Mn, Fe) ₃ (PO ₄) ₂ · 4H ₂ O	42°	$Y = \frac{b}{c} = 10^\circ$ $Z:c = \underline{\underline{}}$	MCL bladed	100 perf 010 fair	Pink, brown	H 2.5 G 2.98	---	
1.618 \vee	1.628	1.631	.013	WOLLASTONITE CaSiO_3	38° $r > v$	$Y:b = 3-5^\circ$	TCL	100 perf 001, T02 good	White	H 5 G 2.92 F 4	Dec by HCl. Fe_2O_3 0.2, FeO 0.1, MnO 0.1%.	
1.644 \vee	1.585	<u>1.630</u>	1.630	.045	TROEGERITE (Autunite grp) $(\text{UO}_2)_3(\text{AsO}_4)_2 \cdot 12\text{H}_2\text{O}$	$\sim 0^\circ$ $r > v$ mod	$X = \frac{c}{b} = 13^\circ$ $Z:a = \underline{\underline{}}$	TET tab	001 perf 100 good	Lemon-yellow	H 2-3 G 3.3 F 2.5	Diss by acids. Not pleoc. Fluor yellow in UV.
~1.613 \wedge 1.652	1.592	---	1.632	.040	MINNESOTAITE (Talc ser) $(\text{Fe}^{+2},\text{Mg})_3\text{Si}_4\text{O}_{10}(\text{OH})_2$	Small	el pos	MCL	001 perf	Gray-green	G 3.03	Pleoc., X pale green, Y and Z bluish-green.

1.610 <u>(1.63)</u>	1.639	.029	GATUMBAITE CaAl ₂ (PO ₄) ₂ (OH) ₂ •H ₂ O	65°					Diss by acids. Luster pearly.
1.606 <u>(1.63)</u>	1.632	.026	CELADONITE (Mica grp) K(Mg,Fe ⁺²)(Fe ⁺³ ,Al)Si ₄ O ₁₀ (OH) ₂	Small r > v					Dec by HCl. Pleoc, X yellow-green. Y and Z green. Fe ₂ O ₃ 12.0, Al ₂ O ₃ 6.7, FeO 2.1, MgO 5.8%.
1.643 1.625 1.655	1.630 1.616	1.641	.025	ANTHOPHYLLITE (Amphibole grp) (Mg,Fe) ₇ Si ₈ O ₂₂ (OH) ₂	88°				One long- itudinal
1.590 1.636	1.630	1.632	.019	CARPHOLITE (Mn,Fe)Al ₂ Si ₂ O ₆ (OH) ₄	50°	r > v str			MCL fib
1.620 1.644 (~1.66)	1.630	1.635	.015	MAGNESIO-CUMMINGTONITE (Amphibole grp) (Mg,Mn,Fe) ₇ Si ₈ O ₂₂ (OH) ₂	74°				Y = <u>b</u> X ~ <u>c</u> el cTv pos
1.613 1.608	1.631 1.646	1.645	.032	SODIUM BOLTWOODITE (H ₃)(Na,K)(UO ₂) ₂ (SiO ₄)•H ₂ O	Large				Y = <u>b</u> Z = <u>c</u> el pos
(1.627) 1.699	1.631	1.642	.034	LAZULITE (Mg,Fe)Al ₂ (PO ₄) ₂ (OH) ₂	70°	r < v perc			Y = <u>b</u> X ~ <u>c</u> el cTv neg
1.617 1.624 1.644	1.631	1.637	.020	RICHTERITE (Amphibole grp) Na ₂ Ca(Mg,Mn,Fe) ₅ Si ₈ O ₂₂ (OH) ₂	68°				Y = <u>b</u> Z:c = 18° el pos
1.620	1.631	1.633	.013	ECKERMANNITE (Amphibole grp) Na ₃ (Mg,Fe) ₄ AlSi ₈ O ₂₂ (OH) ₂	72°				X:c = 49°
1.589	1.632	1.634	.045	WOLLASTONITE-2M, Parawollastonite CaSiO ₃	44°	r > v			Y = <u>b</u> X:c = 34°
1.603	1.632	1.632	.029	KARLITE Mg ₇ (BO ₃) ₃ (OH,Cl) ₅	24°				X = <u>c</u> Y = <u>b</u>
				CARYOPILITE (Serpentine grp) (Mn,Mg) ₃ Si ₂ O ₅ (OH) ₄	~ 0°				MCL fib
									Dec by acids.
									Insol in cold dil HCl.
									Dec by acids.
									H 5.5 G 2.8-2.9 F easy

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral Name and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation $X = \frac{a}{\underline{b}}$ $Y = \frac{c}{\underline{d}}$	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.617	<u>1.632</u>	1.640	.023	CHESTERITE (Mg, Fe ⁺²) ₁₇ Si ₂₀ O ₃₄ (OH) ₆	71° r > v wk		ORTH	110 perf at 45°	Col to light pinkish-brown	---	FeO 14.1, MnO 1.0%.	
1.615 1.643	1.622 ---	1.632 <u>1.632</u>	1.642 1.634	.020 ---	ACTINOLITE (Amphibole grp) Ca ₂ (Mg, Fe) ₅ Si ₈ O ₂₂ (OH) ₂	80° r < v	$Y = \frac{b}{\underline{c}}$ $Z:c = 26^\circ$ el pos	MCL	110 perf at 124°	Green	H 5.5-6 G 3.18	Insol in acids. Pleoc. FeO 7.2, Fe ₂ O ₃ 2.8, Al ₂ O ₃ 2.2%.
					KAHLERITE (Autunite grp) Fe(10 ₂)(AsO ₄) ₂ •xH ₂ O	9-33°	---	TET tab	001 perf	Yellow-green	---	Diss by acids. Not fluor in UV.
					GARRELSITE Ba ₃ NaSi ₂ B ₇ O ₁₆ (OH) ₄	72°	$Z = \frac{b}{\underline{c}}$	MCL bipyram	---	Col s	G 3.8	---
1.620	<u>1.633</u>	1.640	.020	KINOSHITALITE (Mica grp) (Ba ⁺ , Na)(Mg, Mn, Al) ₃ Si ₂ Al ₂ 10(OH, F) ₂	23°	el pos	MCL	001 perf	Yellow-brown	H 2.5-3 G 3.23-3.30	Pleoc wk, X very light yellow, Y and Z yellow. BaO 17.9, K ₂ O 3.3, MnO 7.4, Mn ₂ O ₃ 3.2%.	
1.630	1.619	<u>1.633</u>	1.635	.016	LIBERITE Li ₂ Bes ₁₀ O ₄	66°	$Z:c = 41^\circ$	MCL	010 perf 100, 001 dist	Yellow to brown	H 7 G 2.69	---
					DANBURITE CaB ₂ Si ₂ O ₈	88° r < v str	$X = \frac{a}{\underline{b}}$ $Z = \frac{c}{\underline{d}}$	ORTH el c	001 poor	Col to yellowish	H 7 G 2.94 F diff	Insol in acids.
					ROSCOLELITE (Mica grp) K(V, Al, Mg) ₂ AlSi ₃ O ₁₀ (OH) ₂	Med r < v	$Z = \frac{b}{\underline{c}}$ $X \sim \frac{c}{\underline{d}}$ el pos	MCL	001 perf	Brownish-green	H 3 G 2.8 F 3	Insol in acids. V ₂ O ₃ 17.4, Al ₂ O ₃ 21.9, FeO 1.7%.
1.685	1.59	<u>-1.63</u>	<u>-1.64</u>	<u>.05</u>	STILPNOLEMANE	~ 0°	$X = \frac{b}{\underline{c}}$ $Y \sim \frac{c}{\underline{d}}$	MCL also tric	001 perf	Brown	H 3-4 G 2.84 F 4-5	Dec by acids. Pleoc, X pale brown, Y and Z dark brown. FeO 22.7, Fe ₂ O ₃ 12.2%.
1.603 1.661	1.575 1.634	<u>1.634</u>	1.634	.059	K(Fe ⁺² , Mg, Fe ⁺³ , Al) ₁₀ Si ₁₂ O ₃₀ (OH) ₁₂	56° (78+8°) r > v wk		ORTH pris	210 perf	Violet	H 5.5-6 G 3.04	Pleoc, X and Y col s, Z litac. FeO 7.3, Fe ₂ O ₃ 1.0, Al ₂ O ₃ 14.3, Li ₂ O 3.2%.
1.645	1.616	<u>1.634</u>	1.646	.030	HOLMQUISTITE (Amphibole grp) Li ₂ (Mg, Fe ⁺²) ₃ Al ₂ Si ₈ O ₂₂ (OH) ₂		$Z = \frac{c}{\underline{d}}$ el pos					

1.618 ^	<u>1.634</u>	1.634	.024	GLAUCONITE (Mica grp)	10° r < v	$y = \frac{b}{c}$ $x \sim \frac{c}{e}$	MCL ---	001 perf	Green	H 2 G 2.74 fus
1.610	<u>1.635</u>	1.635	.075	GILALITE $Cu_5Si_6O_{17} \cdot 7H_2O$	Small	$y = \frac{b}{c}$ el	MCL fib	Green	H 2 G 2.82	
1.560	<u>1.635</u>	1.635	.041	SKLODOWSKITE $Mg(OH)_2Si_2O_6(OH)_2 \cdot 5H_2O$	~ 90° r > v str	$y = \frac{b}{c}$ el	MCL pris b tab	100 perf	Yellow	H 2-3 G 3.64
1.615	<u>1.635</u>	1.656	.041							Pleoc, X cols, Y yellow, Z pale yellow. Not fluor in UV.
1.632 ^	<u>1.635</u>	1.652	.035	TILLEVITE $Ca_5Si_2O_7(CO_3)_2$	92° r > v perc	$y = \frac{b}{c}$ $x:a = 18^\circ$	MCL ---	100 perf	White	G 2.84
1.611	<u>1.635</u>	1.643	.032	BURNGAITE $(Na,Ca)_2(Fe,Mg)_2Al_{10}(PO_4)_8(OH,0)_{12}$	58° r > v	$z = \frac{b}{c}$ $x:c = 11^\circ$	MCL pris	100 perf	Bluish to bluish-green	H 5 G 3.05
1.624	<u>1.635</u>	1.654	.030	BAKERITE $Ca_4B_4(BO_4)_3(SiO_4)_3(OH)_3 \cdot H_2O$	85° (105+8°)	$y = \frac{b}{c}$ $x:c = 44^\circ$	MCL pris tab	---	White	H 4.5 G 2.88
1.651 ^	<u>1.635</u>	1.635	.023	METALODEVITE (Meta-autunite grp) $Zn(UO_2)_2(AsO_4)_2 \cdot 10H_2O$	27-37°	---	ORTH tab	---	Pale yellow to olive	G 4.00
1.615	<u>1.635</u>	1.638	.023	NORDITE $(La,Ce)(Sn,Ca)(Na,Mn)_3(Zn,Mg)Si_6O_{17}$	31° (66+10°)	$x = \frac{a}{c}$ $y = \frac{b}{d}$	ORTH	100 good	Light brown	H 5-6 G 3.43 F easy
1.630 ^	<u>1.635</u>	1.642	.023	NORDITE $(La,Ce)(Sn,Ca)(Na,Mn)_3(Zn,Mg)Si_6O_{17}$	82° r < v	$x = \frac{c}{e}$ $y = \frac{b}{d}$ el neg	ORTH	110 good 100 poor	Pink, white	H 6-6.5 G 3.14 infus
1.640	<u>1.635</u>	1.639	.012	ANDALUSITE $Al_2Si_5O_8$	31° (66+10°)	---	MCL ---	---	Green	H 2 G 2.72
1.644 ^	<u>1.635</u>	1.635	.075	GILALITE $Cu_5Si_6O_{17} \cdot 7H_2O$	Small	$x = \frac{a}{c}$ $y = \frac{b}{d}$	ORTH crusts	100	Lemon-yellow	H 1-2 G 4.55 calc
1.560	<u>1.635</u>	1.635	.041	JOLIOTITE $(UO_2)(CO_3) \cdot xH_2O$ (x < 2)	30° r < v str	$x = \frac{a}{c}$ $y = \frac{b}{d}$	ORTH el cTV pos	100 perf 010 poor	Blue-green	G 2.99 infus
~1.650 ^	<u>1.636</u>	1.637	.041	GRANDIERITE $(Mg,Fe)Al_3(BO_4)(SiO_4)_2$	30° r < v str	---	---	---	Pleoc, X cols, Z yellow. Fluor wk in UV.	Pleoc, X cols, Z yellow. Fluor wk in UV.
1.619 ^	<u>1.636</u>	1.639	.037							Insol in acids. Pleoc, X greenish-blue, Y cols, Z pale bluish-green.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.630 $\frac{\Delta}{\Delta}$ 1.644	1.621	<u>1.636</u>	1.640	.019	FERROCARPHOLITE (Fe,Mg)Al ₂ Si ₂ 6(OH) ₄	$X = \frac{b}{c}$ $Z = \frac{c}{e}$ el pos	ORT Fib \underline{c}	Pris at 68°	Yellow to green	H 5.5-6 G 3.0 F 3	Insol in acids. Pleoc, X and Y yellow-green, Z blue-green. FeO 15.5, MnO 0.1, Fe ₂ O ₃ 3.9%.
1.624	1.636	1.642	.018	BERTOSSAITE (Li,Na) ₂ CaAl ₄ (PO ₄) ₄ (OH,F) ₄	Med $r < v$ str	$X = \frac{a}{c}$ $y = \frac{c}{e}$	ORT	100 good	Pink	H 6 G 3.10	Slowly diss by HNO ₃ .
1.584	<u>1.637</u>	1.670	.086	USHKOVITE MgFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ 8H ₂ O	50°	ext at 26°	TCL	010 good	Light yellow	G 3.38	Diss by acids.
1.610	<u>1.637</u>	1.652	.042	FABIANITE CaB ₃ O ₅ (OH)	66° $r < v$ wk	$Y = \frac{b}{c}$ $X:A = 22^{\circ}$	MCL pris	110	Col s	H 6 G 2.77	Diss by acids. Fluor brownish-yellow in UV.
1.615	<u>1.637</u>	1.638	.023	CHAMOSITE (Chlorite grp) (Mg,Fe ⁺² ,Fe ⁺³) ₅ Al (Si ₃ Al) ₁₀ (OH,0) ₈	15°	$X \perp c$	MCL	001 perf	Green	H 2.5 G 3.0 F diff	Dec by acids. Pleoc, X olive-green, Y and Z brownish-yellow. FeO 22.8, Fe ₂ O ₃ 9.1, Al ₂ O ₃ 22.1%.
1.619 $\frac{\Delta}{\Delta}$ 1.649	1.637	1.642	.022	GLAUCOPHANE (Amphibole grp) Na ₂ (Mg,Fe) ₃ Al ₂ Si ₈ O ₂₂ (OH) ₂	51° $r < v$ str	$Y = \frac{b}{c} = 5^{\circ}$ $Z:c = 5^{\circ}$ el pos	MCL pris	110 perf at 121°	Blue	H 6 G 3.13 F 3.5	Insol in acids. Pleoc, X col s, Y lavender, Z blue. FeO 10.6, Fe ₂ O ₃ 2.0, Al ₂ O ₃ 10.1%.
1.620	<u>1.637</u>	1.642	.022	NEPOUITE (Serpentine grp) (Ni,Mg) ₃ Si ₂ O ₅ (OH) ₄	---	---	MCL fib	001 perf	Dark green	H 6 G 3.2 infus	Insol in acids. Pleoc wk, X dark green, Z yellow-green. NiO 43.7, MgO 8.2%.
1.623	---	1.640	.017	SUANITE Mg ₂ B ₂ O ₅	---	---	MCL fib	010	White	H 5.5 G 2.91 F diff	Slowly diss by HCl.
1.596	<u>1.639</u>	1.670	.074	INESITE Ca ₂ Mn ₇ Si ₁₀ O ₂₈ (OH) ₂ 5H ₂ O	70° (70+3°) $r > v$ wk	$X = \frac{b}{c}$ $y:c = 23^{\circ}$	TCL pris	010 perf 100 good	Rose-red to flesh	H 5.5 G 3.02 F 3	Dec by HCl with sepn of silica.
1.618	<u>1.639</u>	1.653	.035	TROLLEITE Al ₄ (PO ₄) ₃ (OH) ₃	75° $r > v$ wk	Ext from clev on 001 = 14°	TCL pris	010 perf 100 good	---	H 5.5 G 3.09 infus	Nearly insol in acids. Fe 2%.
1.636 $\frac{\Delta}{\Delta}$ 1.642	1.639	1.643	.024	---	49° $r > v$ wk	---	MCL mass	---	Pale green	H 5.5 G 3.09 infus	

1.624	1.639	1.643	.019	ROSCHEIRITE $\text{Ca}(\text{Al},\text{Fe},\text{Mn})_3\text{Be}_2(\text{PO}_4)_3$ $(\text{OH})_3 \cdot 2\text{H}_2\text{O}$ (?)	Large $r > v$ str crossed	$X = \frac{b}{c}$ $y:c = -15^\circ$	MCL pris	001 good 010 fair	Dark brown	H 4.5 G 2.93 fus	Diss by acids. Pleoc., X yellow to olive, Y yellow-brown, Z chest- nut brown. Abnormal interf colors.
1.617 1.652	1.590 1.640	1.640	.050	SIDEROPHYLLITE (Biotite ser, Mica grp) $\text{KFe}_2\text{Al}(\text{Al}_2\text{Si}_2)\text{O}_{10}$ $(\text{F}, \text{OH})_2$	Small $r < v$	$Y = \frac{b}{c}$ el pos	MCL plates	001 perf	Dark brown	H 2.5-3 G 3.12 F 4	Pleoc., X brown, Y and Z dark brown. FeO 30.2, MnO 1.0%.
1.624	1.640	1.650	.026	OURSINITE $(\text{Co},\text{Mg})(\text{UO}_2)_2\text{Si}_2\text{O}_7$ $6\text{H}_2\text{O}$	(76+9°)	$Y = \frac{c}{c}$	ORTH	---	Pale yellow	G (3.67)	Does not fluor UV.
1.632	---	1.646	.014	ZYKAITE $\text{Fe}^{+3}(\text{AsO}_4)_3\text{SO}_4$ $15\text{H}_2\text{O}$	Large	el pos	ORTH	Uneven	Gray-white, yellow-green	G 2.50	---
1.631 1.660	1.640	1.641	.010	ROMEITE $\text{Ca}_2(\text{Mn},\text{Mg})_2\text{B}_4\text{O}_7(\text{OH})_6$ $15\text{H}_2\text{O}$	20° $r < v$ str	$Z = \frac{b}{c}$ $y:c = -15^\circ$	ORTH tab	101 poor	Light brown	H 5 G 2.73 F 2	Diss by acids. MnO 17.9, MgO 7.6, FeO 1.2%.
1.627 1.660	1.640	1.645	.012	Unnamed Ca ₂ SiO ₄	57°	---	ORTH fib	c1v ⊥ e1	White	G 2.97 calc infus	Gel with acids. Am. Mineral., 51, 171 (1966).
1.633	1.640	1.645	.012	NINGYOITE $(\text{U},\text{Ca},\text{Ce})_2(\text{PO}_4)_2$ $1-2\text{H}_2\text{O}$	---	el pos	ORTH ps hex	---	Brown to brownish-green	---	Pleoc. wk in brown. Opt char unk.
---	1.64	---	---	Wk	---	---	---	---	---	---	---
1.583	1.641	1.648	.065	SERPIERITE $\text{Ca}(\text{Cu},\text{Zn})_4(\text{SO}_4)_2(\text{OH})_6$ $3\text{H}_2\text{O}$	(37+6) $r > v$ str	$Y = \frac{b}{c}$ $x:a = 24^\circ$	MCL laths	100 perf	Sky-blue	G 3.07	Diss by acids. Pleoc., X pale green, Y and Z bluish-green.
1.607	1.641	1.672	.065	PAPAGOITE $\text{CaCuAlSi}_2\text{O}_6(\text{OH})_3$ $3\text{H}_2\text{O}$	78° $(86+4^\circ)$ $r > v$ wk	$Z = \frac{b}{c}$ $x:c = 44^\circ$	MCL tab	100 good	Blue	H 5-5.5 G 3.25	Pleoc., X cols to pale greenish-blue, Y blue, Z deep greenish-blue.
1.609	1.641	1.641	.032	METAHEINRICHITE (Meta-autunite grp) $\text{Ba}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	0-20°	---	TET tab	001 perf 100 dist	Yellow to green	H 2.5 G 4.04	Diss by acids. Pleoc., X cols, Y and Z pale yellow. Fluor yellow- green in UV.
1.624	1.641	1.647	.023	ROSENHAHNITE $\text{Ca}_3\text{Si}_3\text{O}_8(\text{OH})_2$	60-68°	On 100 $z:c = 36^\circ$	TCL laths	001 perf 100, 010 good	Col s	H 4.5-5 G 2.89 F 3	Slightly attacked by acids.
1.635 1.646	(1.642)	1.642	.007	KILLALAITA $2\text{Ca}_3\text{Si}_2\text{O}_7 \cdot \text{H}_2\text{O}$	26°	$Y = \frac{b}{c}$ $z:c = 16^\circ$	MCL	100 perf 010, 001 good	Col s	G 2.88 calc	Complex penet tw.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
ν 1.653	1.618	<u>1.642</u>	1.652	.034	SOUZALITE (Mg, Fe) ₃ (Al, Fe) ₄ (PO ₄) ₄ (OH) ₆ •2H ₂ O	68° r > v str	$\chi = \frac{a}{e}$ el pos	MCL (?) fib	100 perf	Green	H 5.5-6 G 3.09 infus	Diss by HCl. Poly tw. Pleoc., X green, Y blue, Z yellow.
ν 1.645	1.633	<u>1.642</u>	1.652	.019	MAGNESIO-HORNBLENDE (Amphibole grp) (Ca, Na) ₂ (Mg, Fe)Al (Si ₇ Al) ₂₂ (OH, F) ₂	88°	$\gamma = \frac{b}{c}$ $z:c = 16^\circ$ el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.10	Insol in acids. Pleoc., X pale yellow, Y green. FeO 8.7, Fe ₂ O ₃ 2.7, Al ₂ O ₃ 5.0%.
ν 1.627	1.642	<u>1.644</u>	1.644	.017	PALERMOITE (Li, Na) ₂ (Sr, Ca)Al ₄ (PO ₄) ₄ (OH) ₄	~ 20° r < v mod	$\gamma = \frac{a}{c}$ $x = \frac{c}{e}$ el neg	ORTH pris, el° c	100 perf 001 fair	Col s, white	H 5.5 G 3.22	---
ν 1.555	<u>1.643</u>	1.658	.103	SIBIRSKITE CaHBO ₃	43°	---	MCL	---	Col s	---	---	Diss by acids.
ν 1.559	<u>1.643</u>	1.665	.096	HOHMANNITE Fe ₂ (SO ₄) ₂ (OH) ₂ •7H ₂ O	40° r > v str	$\gamma:c = 23^\circ$	TCL pris	010 perf	Amaranth- red to brown	H 3 G 2.2	Diss by acids. Pleoc., X pale yellow, Y pale greenish-yellow, Z dark greenish-brown.	
ν 1.650	<u>1.650</u>	1.663	.096	HOHMANNITE Fe ₂ (SO ₄) ₂ (OH) ₂ •7H ₂ O	40° r > v str	---	MCL	001 perf	Bright green	H 2.8 F 3	Dec by HCl. Pleoc., X yellow-green, Y and Z green. Fe ₂ O ₃ 29.3, Al ₂ O ₃ 1.0, FeO 9.8, MgO 3.7%.	
ν (1.63) Δ (1.662)	1.612	<u>1.643</u>	1.643	.031	CELADONITE (Mica grp) K(Mg, Fe ⁺²)(Fe ⁺³ , Al)Si ₄ O ₁₀ (OH) ₂	Small r > v	$\gamma = \frac{b}{c}$ $x \sim \frac{c}{e}$ el cTV pos	MCL pris	110 perf at 124°	Green	H 5.5-6 G 3.17	Insol in acids. Pleoc., FeO 8.3, Fe ₂ O ₃ 2.4%.
ν 254	<u>1.629</u>	<u>1.643</u>	1.650	.021	ACTINOLITE (Amphibole grp) Ca ₂ (Mg, Fe) ₅ Si ₈ O ₂₂ (OH) ₂	73° r < v	$\gamma = \frac{b}{c}$ $z:c \sim 20^\circ$ el pos	MCL pris	110 perf at 124°	Green	H 5.5-6 G 3.17	---
ν 1.663	1.632	<u>1.643</u>	1.646	.014	MARGARITE (Mica grp) CaAl ₂ (Al ₂ Si ₂) ₁₀ (OH) ₂	63° r < v	$z = \frac{b}{c}$ $y:c = 7^\circ$	MCL	001 perf	Col s	H 4 G 3.06 F diff	BeO 1.8 to 3.3%.
ν 1.652	1.624	<u>1.643</u>	1.648	.011	KOASHWHITE Na ₆ (Ca, Mn)(Ti, Fe)Si ₆ O ₁₈ •H ₂ O	83° r > v wk	---	ORTH	Conch	Pale yellow	H 6 G 3.00 F 3	---
ν 1.658	1.639	<u>1.643</u>	1.646	.007	HIORTDHALITE (Ca, Na) ₃ ZrSi ₂ O ₇ (O, OH, F) ₂	83° r < v	Ext on 100 = 65°	TCL tab	Pris at about 90°	Yellow, brown	H 5.5 G 3.26 F 3	Gel with acids. Poly tw., comp pl 100. Pleoc. wk, X col s, Y and Z yellow.

1.630 1.665	<u>1.644</u>	1.652	.022	TIRODITE (Cummingtonite ser., Amphibole grp) $Mn_2(Mg,Fe^{+2})_5Si_8O_{22}$ (OH) ₂	73°	$Z:c = 20^\circ$ $y = \frac{b}{\overline{el pos}}$	MCL pris	110 perf at 102 good	Pink to greenish G 3.10	Insol in acids. MnO 16.6, FeO 4.5%.	
1.628 1. (1.652)	1.631	<u>1.644</u>	1.646	.015	WOLLASTONITE (Ca, Mn)SiO ₃	45° r > v	$\overline{Y} \sim \frac{b}{x:c} \sim 40^\circ$	TCL	100 perf at 102 good	Dec by HCl. FeO 0.7%.	
1.635 1. 1.656	1.637	<u>1.644</u>	1.650	.013	ANDALUSITE Al ₂ SiO ₅	73° r < v	$x = \frac{c}{y}$ $y = \frac{b}{\overline{el neg}}$	ORTH	110 good 100 poor	Insol in acids. Pleoc. wk, x rose, y and Z yellow.	
1.631 1. 1.656	1.636	<u>1.644</u>	1.649	.013	ECKERMANNITE (Arfvedsonite ser., Amphibole grp) $Na_3(Mg,Fe)_4AlSi_8O_{22}$ (OH) ₂	74°	$x:c = 25^\circ$ $\overline{el neg}$	MCL pris	110 good	Insol in acids. X bluish-green, Y bright blue-green, Z yellow-green. Na ₂ O 9.8, FeO 2.7, Fe ₂ O ₃ 8.0, MgO 9.1%.	
1.653	1.637	<u>1.644</u>	1.645	.008	JEREMEJEVITE Al ₆ B ₅ O ₁₅ (F, OH) ₃	18° r > v dist	$x = \frac{c}{z}$	HEX pris	Bluish- green G 3.16	Insol in acids.	
255	1.589	<u>1.645</u>	1.659	.056	CAMPIGLIAITE Cu ₄ Mn(SO ₄) ₂ (OH) ₆ ·4H ₂ O	52° r < v wk	$z = \frac{b}{x}$ $x \sim \frac{a}{d}$	MCL el b	White H 5-6 G 3.03 F 4	Dec by HCl. FeO 0.7%.	
1.634 1. 1.664	1.624	<u>1.645</u>	1.651	.027	HOLMQVISTITE (Amphibole grp) $Li_2(Mg,Fe^{+2})_3Al_2Si_8O_{22}$ (OH) ₂	50° r > v wk	$z = \frac{c}{y}$ $\overline{el pos}$	ORTH pris	110 perf at 124°	White H 5-6 G 3.10	Diss by acids. Pleoc. x col s, y pale pink to yellow, Z pink to brown.
1.642 1. 1.664	1.634	<u>1.645</u>	1.658	.024	EDENITE (Horriblende ser., Amphibole grp) $NaCa_2(Mg,Fe)_5Si_7Al_1O_{22}$ (OH) ₂	74° (94+10°)	$y = \frac{b}{\overline{el pos}}$	MCL pris	Conch	Col s to brown H 6.5 G 3.28 infus	Diss by acids with eff.
1. 1.655	1.637	<u>1.645</u>	1.649	.012	HUREAULITE $Mn_5(Po_4)[Po_3(OH)]_2$ · 4H ₂ O	75° r < v very str	$x = \frac{b}{z}$ $z:c = 65^\circ$	MCL tab	100 good	Pink to brown H 3.5 G 3.19 F 2	Diss by acids. Pleoc., x col s, y pale pink to yellow, Z pink to brown.
1.556	1.646	<u>1.652</u>	.096	DONNAYITE Sr ₃ Ca ₂ Y(CO ₃) ₆ ·3H ₂ O	0-20° u 5-10°	---	TCL	001 indist to fair	Yellow H 3 G 3.35	Diss by acids with eff.	
1.558	<u>1.646</u>	1.648	.090	WELOGANITE Sr ₃ Na ₂ Zr(CO ₃) ₆ ·3H ₂ O	0-15°	---	TCL ps hex el c	001 perf fr conch	Lemon- yellow G 3.20	Diss by acids with eff.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
(1.689)	1.575	1.646	1.650	.075	SZAIBELYITE (Suzesite ser) $MgBO_2(OH)$	~ 25° $r > v$	---	MCL fib	110 perf	White	H 3 G 2.65 F 3	Diss by acids. Poly tw.
1.631 1.674	1.620	1.646	1.656	.036	LAZULITE (Mg, Fe)Al ₂ (PO ₄) ₂ (OH) ₂	67° $r < v$ perf	$Y = \frac{b}{c}$ $X \sim \frac{c}{c}$	MCL	110 good 101	Blue	H 5-6 G 3.19 infus	Insol in acids. Pleo, X cols, Y and Z blue. FeO 8.9, Fe ₂ O ₃ 0.9%.
1.640	(1.646)	1.651	.011	RUSTUMITE $Ca_{10}(Si_{20})_2SiO_4Cl_2$ (OH) ₂	80°	$Y = \frac{b}{c}$ $X: \frac{c}{c} = 5^\circ$	MCL tab	100 poor	cols	G 2.9	Lam tw on 100.	
1.639	1.646	1.646	.007	KELLYITE (Kaoelite-Serpentine grp) (Mn, Mg, Al) ₃ (Si, Al) ₂ Si ₄ (OH) ₄	5-30° $r > v$ mod	$X = \frac{c}{c}$ el cTv pos	TRIG and HEX	0001 perf	Yellow	G 3.07	Pleo, X cols to greenish-yellow, Y and Z pale yellow to reddish-brown.	
1.637	1.647	1.647	.010	NIMITE (Chlorite grp) (Ni, Mg, Fe) ₅ Al(Si ₃ Al) ₀ 10(OH) ₈	15°	---	MCL	001 good	Yellow-green	H 3 G 3.16	Pleo wk, X greenish-yellow, Z apple green. NiO 29.5, FeO 2.8, Fe ₂ O ₃ 4.4%.	
1.607 1.659	1.622	1.648	1.651	VERMICULITE (Mg, Ni, Fe, Al) ₃ (Si, Al) ₄ 0 ₁₀ (OH) ₂ ·4H ₂ O	Small	$X \perp$ plates el pos	MCL	001 perf	Brownish-black	H 1.5 G 2.8 F 5	Dec by HCl. Exfoliates when heated. Fe ₂ O ₃ 19.2, FeO 5.0, MgO 13.9, NiO 8.6%.	
1.637 1.659	1.634	1.648	1.653	CROSSITE (Glauconite ser, Amphibole grp) $Na_2(Mg, Fe^{+2})_3$ (Al, Fe ⁺³) ₂ Si ₈ O ₂₂ (OH) ₂	Small (61±13°)	$X:c = 8^\circ$ $Y = \frac{b}{c}$ el neg	MCL pris	110 perf	Blue	H 5 G 3.14	Insol in acids. Pleo, X yellowish, Y lavender, Z blue. FeO 7.5, Fe ₂ O ₃ 6.4, Al ₂ O ₃ 8.4%.	
1.646	1.646	1.650	.004	KILCHOANITE $Ca_3Si_2O_7$	46-54° $r > v$ str	---	ORTH	---	---	G 2.99	Gel with acids. Abnormal ultrablue and brown interf colors.	
1.585	1.649	1.660	.075	DEVILLINE $CaCu_4(SO_4)_2(OH)_6 \cdot 3H_2O$	39° (44±4°)	$Z = \frac{b}{c}$ $Y \sim \frac{c}{c}$	MCL tab	001 perf 110, 101	Emerald-blue dist	H 2.5 G 3.13 F 3.5	Diss by acids. Tw 010. Pleo, X pale green, Y Venice green, Z turquoise.	

1.667	<u>1.649</u>	1.660	.033	URANOPHANE Ca(UO ₂) ₂ Si ₂ O ₇ •5H ₂ O	60° r > v (U r < v)				
1.664	<u>1.641</u>	1.655	.014	MONTICELLINE Ca(Mg,Fe)SiO ₄	80° r > v	X = $\frac{b}{d}$ $Z = \frac{a}{c}$	010 poor	White, gray	H 5.5 G 3.06 F 6
1.667	<u>1.642</u>	<u>1.649</u>	.007	CHLORAPATITE (Apatite grp) Ca ₅ (PO ₄) ₃ (Cl,F,OH)	Small 5-10°	---	MCL ps hex	ColS	H 5 G 3.18 F 5
1.637	<u>1.643</u>	<u>1.649</u>	.006	CHAMOSITE (Chlorite grp) (Fe ⁺² ,Mg,Fe ⁺³) ₅ Al (Si ₃ Al) ₁₀ (OH,0) ₈	~ 0°	X = $\frac{c}{e}$ el pos	MCL	001 perf	Dark green
1.666				VITUSITE Na ₃ (Ce,La)(PO ₄) ₂	30°	---	ORTH	---	Pale pink, gray
1.646	1.602	<u>1.650</u>	1.654	.052	EPISTOLITE Na ₂ (Nb,Ti) ₂ Si ₂ O ₉ •xH ₂ O	80° r < v perc	TCL plates	001 perf 110 dist	White, gray, yellow
1.610	<u>1.650</u>	1.682	.072	FERRUBUSTAMITE Ca(Fe ⁺² ,Ca,Mn)Si ₂ O ₆	60°	X:c = 44°	TCL	100 perf 110, 110 good	H 1-1.5 G 2.89
1.640	<u>(1.650)</u>	1.653	.013	JOLIOTTITE (UO ₂)(CO ₃)•xH ₂ O (x = 2?)	Small	X = $\frac{a}{d}$ Y = $\frac{b}{e}$	ORTH crusts	100	Pink to brown
(1.636)	<u>1.604</u>	---	1.651	.047	NONTRONITE (Smectite grp) Na _{0.33} Fe ⁺³ ₂ (Si,Al) ₄ O ₁₀ (OH) ₂ •xH ₂ O	33° r < v	X = $\frac{b}{d}$ Y ~ $\frac{c}{e}$	MCL u mass	Green to yellow
1.600	<u>1.625</u>	<u>1.65</u>	.030	EOSPHORITE (Chl dendrite ser) (Mn,Fe)AlPO ₄ (OH) ₂ •H ₂ O	45° r < v	X = $\frac{b}{d}$ Z:c = 5°	MCL ps orth	001 poor	H 1-2 G 2.27
1.664	<u>1.629</u>	<u>1.650</u>	1.658	.029	FRIEDELITE (Mn,Fe) ₈ Si ₆ O ₁₅ (OH,Cl) ₁₀	Small	el clv pos	HEX	Rose-red
1.623	<u>1.650</u>	1.651	.028	APACHITE Cu ₉ Si ₁₀ O ₂₈ •11H ₂ O	Small	---	MCL	---	Blue
1.610	<u>1.650</u>	1.650	.040	BEMENTITE Mn ₈ Si ₆ O ₁₅ (OH) ₁₀	~ 0°	X = $\frac{c}{e}$ el cTv pos	MCL	001 perf	Brown
1.624	<u>1.650</u>	1.650	.026						H 5 G 2.98 F 3
									Gel with acids. Pleoc., wk in yellows. Ab- normal blue interef- colors.
									Gel with acids. FeO 1.4%.
									Diss by acids. Cl 1.31, F 0.45, H ₂ O 0.7%.
									Dec by HCl. Pleoc., X pale yellow, Y and Z olive-green.
									Diss by HCl. Tw.
									Diss by acids. MnO 15.4, MnO 1.6%.
									Pleoc., X cols, Z yellow. Fluor wk in UV.
									Pleoc., X cols, Z yellow to greenish- yellow.
									Diss by acids. MnO 29.9, FeO 1.4%.
									Dec by HCl. Pleoc., X cols, Y and Z greenish- yellow.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.590	1.651	1.657	.067	BORGARITE $\text{Ca}_4\text{MgB}_4\text{O}_6(\text{OH})_6(\text{CO}_3)_2$	30° $r < v$	$Z = \frac{b}{c} = 28^\circ$ $Y:c = 67^\circ$	MCL	100 perf	Cols, greenish-blue	H 4 G 2.77	Diss by acids.	
1.635 ^Δ	1.625	1.651	1.664	.039	BAKERITE $\text{Ca}_4\text{B}_4(\text{BO}_4)(\text{SiO}_4)_3(\text{OH})_3 \cdot \text{H}_2\text{O}$	$Y = \frac{b}{c} = 44^\circ$ $X:c = 67^\circ$	MCL pris tab spher- ulitic	---	White	H 4-5 G 2.88	---	
1.618	1.652	1.682	.064	SCHOONERITE $\text{Fe}^{+2}\text{MnZnFe}^{+3}(\text{PO}_4)_3(\text{OH})_2 \cdot 9\text{H}_2\text{O}$	70-80° (85+3°)	$X = \frac{b}{c} = 72^\circ$ $Y = \frac{b}{c} = 70^\circ$ $Z:a = 25^\circ$	ORTH tab el 100	010 perf 001 good	Brown to red-brown	H 4 G 2.89	Diss by cold acids. Pleoc, X yellow, Y pale brown, Z brown.	
1.612	1.652	1.675	.063	LIROCONITE $\text{Cu}_2\text{Al}_3\text{AsO}_4(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	72° $r < v$ mod	$Y = \frac{b}{c} = 25^\circ$ $Z:a = 25^\circ$	MCL pris	110, 011 poor fr conch	Blue, green	H 2.5 G 2.96 F easy	Diss by acids.	
1.640 ^Δ 1.672	1.595	1.652	1.656	.061	BIOTITE (Mica grp) $K(\text{Fe}, \text{Mg})_3(\text{Al}, \text{Fe}^{+3})\text{Si}_3\text{O}_{10}(\text{OH}, \text{F})_2$	Small $r < v$	$Y = \frac{b}{c} = 25^\circ$ $Z:c = 25^\circ$ el pos	MCL plates	001 perf	Dark brown	H 2.5-3 G 3.05 F 4	Pleoc, X brown, Y and Z dark brown. FeO 19.9, MnO 0.4, Fe ₂ O ₃ 3.2%.
1.599 ^Δ	1.600	1.652	(1.655)	(.055)	WILLEMSITE (Talc ser) $(\text{Ni}, \text{Mg})_3\text{Si}_4\text{O}_{10}(\text{OH})_2$	27°	---	MCL	001 perf	Light to apple-green	H 2 G 3.28	Pleoc, X yellowish-green, Z bluish-green. NiO 34.5, MgO 7.1%.
1.651 pos ^Δ 1.664	1.642	1.652	1.660	.018	PARGASITE (Hastingsite ser., Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{Al}(\text{Si}_6\text{Al}_2)_2(\text{OH})_2$	88°	$Y = \frac{b}{c} = 28^\circ$ $Z:c = 28^\circ$ el pos	MCL pris	110 perf at 124°	Brown	H 5-6 G 3.14	Insol in acids. Pleoc. Al ₂ 3, Fe ₂ O ₃ 0.9, FeO 6.1, MgO 14.3%.
1.644 ^Δ	1.639	(1.652)	1.656	.017	WOLLASTONITE (Ca, Fe)SiO ₃	59°	$Y \sim \frac{b}{c} = 3^\circ$ $X:c = 3^\circ$	TCL	100 perf 001, T02 good	Brown	H 5 G 3.07 F 4	Dec by HCl. FeO 8.7, MnO 2.7, Fe ₂ O ₃ 0.6%.
1.690 ^Δ	1.645	1.652	1.656	.011	LAVENITE $(\text{Na}, \text{Ca})_3\text{ZrSi}_2\text{O}_7(0, \text{OH}, \text{F})_2$	86° $r > v$ wk	el pos	ORTH (?) tab 100	100	Light yellow	H 6 G 3.25	Poly tw 100. Dimorph of Lavenite at $\beta = 1.690$ (?)

1.643 1.655	1.643 <u>1.652</u>	1.654	.011	BITYITE (Margarite ser., Mica grp) CaLi ₂ (AlBeSi ₂) ₁₀ (OH) ₂	55° r > v str	MCL pyramid perf	111, 111	Azure- blue	H 3.5 G 3.05 F easy	Insol in HCl. Basal section divided into 6 segments.			
1.559	1.653	1.680	.121	CALLAGHANITE Cu ₂ Mg ₂ CO ₃ (OH) ₆ ·2H ₂ O	Z:c = 18°	MCL	---	Col s	H 3.3-5 G 2.71	Diss by acids with eff. Pleoc in blue, abs Z > Y > X.			
1.649 1. 1.658	1.625 <u>1.653</u>	1.670	.045	DATOLITE CaB ₅ O ₄ (OH)	73° r > v wk	Y = b Z:a = 1-3°	MCL el c	---	---	Pleoc, X and Z col s, Y blue. FeO 14.7%.			
1.642 1. 1.644	1.619 <u>1.653</u>	1.660	.041	GORMANITE (Fe ⁺² , Mg) ₃ Al ₄ (PO ₄) ₄ (OH) ₆ ·2H ₂ O	53° r > v	---	TCL bladed	001 parting	Blue- green	Gel with acids.			
1.640	1.640	1.653	.013	JEREMEJEVITE Al ₆ B ₅ Si ₁₅ (OH) ₃	Variable 0-33°	X = c	HEX pris	Conch	Col s to brown	H 6.5 G 3.28	Insol in acids. Basal section divided into 6 sections.		
1.567	1.654	1.722	.155	TATARSKITE Ca ₆ Mg ₂ (SO ₄) ₂ (CO ₃) ₂ C ₁ ₄ (OH) ₄ ·7H ₂ O	(79±1°)	el pos	ORTH	---	Col s, yellow	H 2.5 G 2.34			
1.618	1.654	1.655	.037	KIVUITTE (Th, Ca, Pb)H ₂ (UO ₂) ₄ (PO ₄) ₂ (OH) ₈ ·7H ₂ O (?)	0-5° r > v	---	ORTH(?)	---	Yellow	---	Dec by HNO ₃ . Pleoc, X col s, Y and Z greenish-yellow.		
1.595	1.654	1.670	.075	CALCIBORITE CaB ₂ O ₄	54°	Ext:el = 22°	MCL radi- ating	Conch	White	H 3.5 G 2.88	Diss by acids.		
1.650	1.654	1.661	.011	VLADIMIRITE H ₂ Ca ₅ (AsO ₄) ₄ ·5H ₂ O	70° (106+21°) r > v str	Z:c = 36°	MCL	---	Col s	H 3.5 G 3.14 F diff	Diss by acids.		
1.630 1. 1.667	1.642 <u>1.655</u>	1.661	.019	GEDRITE (Amphibole grp) (Mg, Fe ⁺²) ₅ Al ₂ Si ₆ Al ₂ 0 ₂₂ (OH) ₂	Large r > v	Y = b Z = c el pos c	ORTH pris c	210 perf	Brown, gray	H 6 G 3.18 F 4	Insol in acids. FeO 9.2, Fe ₂ O ₃ 0.2, Al ₂ O ₃ 23.8%.		
1.645 1. 1.667	1.649 <u>1.655</u>	1.659	.010	HUREAULITE (Mn, Fe) ₅ (PO ₄) ₂ [PO ₃ (OH)] ₂ ·4H ₂ O	75° r < v str	X = b Z:c = 55°	MCL tab	100 good	Pink to brown	H 3.5 G 3.18 F 2	Diss by acids. Pleoc, X col s, Y pale pink or yellow, Z pink or brown.		
1.644 1. 1.683	1.645 <u>1.656</u>	1.661	.016	MAGNESIUM-AREVEDSONITE (Amphibole grp) (Na, Ca) ₃ (Mg, Fe ⁺²) ₄ Fe ⁺³ Si ₈ O ₂₂ (OH, F) ₂	57°	Y = b Z:c = 56°	MCL	110 perf at 124°	Bluish- green	H 6 G 3.23	Insol in acids. Pleoc, X blue-green, Y green, Z pale green. Fe ₂ O ₃ 8.4, FeO 13.4, MnO 1.5, MgO 7.8, F 3.3%.		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.569	1.657	1.686	.117	CALKINISITE (Ce, La) ₂ (CO ₃) ₃ ·4H ₂ O	54° r < v	$Y = \frac{c}{Z} = \frac{1}{4}$	ORTH platy	010 perf 101 fair	Pale yellow	H 2.5 G 3.28	Diss by acids with eff. Commonly tw 101. Pleoc wk.
1.650	(1.657)	1.660	.010	DELLAITE Ca ₆ Si ₃ O ₁₁ (OH) ₂	65°	Z:e1 = 20°	---	---	Col s	---	---
1.612	1.658	1.682	.070	LAUETTE MnFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ · 8H ₂ O	50° (0+3°) r < v str	Z:c ~ 30°	TCL	010 perf brittle	Brown	H 3 G 2.45	Diss by acids. Pleoc wk.
1.615	---	1.685	.070	LIKASITE Cu ₆ (PO ₄)(NO ₃) ₂ (OH) ₇	---	X = $\frac{a}{b}$ Z = $\frac{c}{d}$	ORTH ps hex	001 perf	Blue	G 2.97	Diss by acids. Pleoc, x green-blue, Z pale blue. Opt. char unk.
(1.640)	1.658	1.664	(.024)	JUNGITE Ca ₂ Zn ₄ Fe ⁺³ ₈ (PO ₄) ₉ (OH) ₉ ·16H ₂ O	~ 60° r < v str	Y = $\frac{a}{c}$ Z = $\frac{b}{d}$	ORTH tab	010 perf	Yellow to yellow-green	H 1 G 2.84	---
1.622	1.658	1.687	.065	ANNABERGITE (Erythrite ser) Ni ₃ (AsO ₄) ₂ ·8H ₂ O	84° r > v str crossed	X = $\frac{b}{c}$ Z:c = 36° el clv pos	MCL pris acic	010 perf	Apple-green	H 2.5 G 3.07 F 4	Diss by acid. With increase of Co, becomes gray, then pink.
1.598	1.658	1.660	.062	HENDRICKSITE (Mica grp) K(Zn,Mn) ₃ Si ₃ Al ₁₀ (OH) ₂	8°	---	MCL	001 perf	Dark reddish-brown	H 3 G 3.4	Dec by acids. ZnO 19.8, MnO 12.5, Fe ₂ O ₃ 4.85%.
1.660	1.639	1.658	1.670	JAHNSITE CaMn(Mg, Fe ⁺²) ₂ ⁺³ (PO ₄) ₄ (OH) ₂ ·8H ₂ O	Large	Z = $\frac{b}{c}$ X:c = 18° el pos	MCL pris tab	001 good	Brown to yellow	H 4 G 2.72	Diss by acids. Tw 001. Pleoc, x pale purple to brown, y red-brown, Z cols to yellow, abs Y > Z > X.
1.686	1.639	1.658	1.66	STEWARTITE MnFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ · 8H ₂ O	Large r < v str	X:c = 56°	TCL	010 good	Brownish-yellow	G 2.94	Pleoc wk.
(~1.640)	1.658	1.664	(~.024)	JUNGITE Ca ₂ Zn ₄ Fe ⁺³ ₈ (PO ₄) ₉ (OH) ₉ ·16H ₂ O	~ 60° r < v str	Y = $\frac{a}{c}$ Z = $\frac{b}{d}$	ORTH tab	010 perf	Yellow to yellow-green	H 1 G 2.84	---
1.652	1.658	1.665	.013	HJORTDAHLITE (Ca, Na) ₂ Si ₂ O ₇ (O, OH, F) ₂	~ 90° r < v	Ext on 100 = 65°	TCL tab	Pris at ~ 90°	Yellow, brown	H 5.5 G 3.25 F 3	Gel with acids. Poly tw pl 100. Pleoc wk, x cols, y and Z pale brownish-yellow.

1.648 ^	1.659	1.662	.021	FERRRO-GLAUCOPHANE (Amphibole grp) $\text{Na}_2(\text{Fe}^{+2}, \text{Mg})_3\text{Al}_2\text{Si}_8\text{O}_{22}$ (OH) ₂	42° r < v str	$y = \frac{b}{c}$ $z:c = 6$ el pos	MCL pris	110 perf at 121°	Blue	
1.641	1.659	1.660	.012	CLINTONITE (Mica grp) $\text{Ca}(\text{Mg}, \text{Al})_3(\text{Al}_3\text{Si})_{10}$ (OH) ₂	5-33°	$z = \frac{b}{c}$ el pos	001 perf	Reddish-brown	H 6 G 3.23 F 3	
1.648	1.659	1.660	.071	ERYTHRITE $\text{Co}_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	87° r > v wk	$x = \frac{b}{c} = 31^\circ$ el pos	010 perf	Pink to red	H 5 G 3.07 infus	
1.658	1.660	1.694	.071	TILASITE $\text{CaMg}(\text{AsO}_4)\text{F}$	83° r < v wk	$z = \frac{b}{c} \sim 30^\circ$ X:C ~ 30°	MCL	10T good	H 3 G 3.06 F 2	
1.640	1.660	1.675	.035	ROWETITE $\text{Ca}_2(\text{Mn}, \text{Zn})_2\text{B}_4\text{O}_7(\text{OH})_6$	15° (53+18°) r < v str	$z = \frac{b}{c}$ Y = $\frac{c}{a}$	ORTH tab	10I poor	H 5 G 2.92 F 1	
1.640	1.660	1.663	.015	SODIUM BOLTWOODITE $(\text{H}_3\text{O})(\text{Na}, \text{K})(\text{UO}_2)\text{SiO}_4 \cdot \text{H}_2\text{O}$	Large	---	ORTH	010 perf 001 imperf	H 5 G 3.77 F easy	
(1.63)	1.645	(~1.66)	.1.672	.027	STILPNOLEMITE $\text{K}(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3}, \text{Al})_{10}$ $\text{Si}_{12}(\text{OH})_{30}$	~ 0°	$y = \frac{b}{c}$ $x \sim \frac{c}{a}$ el ctv pos	TCL also MCL	001 perf	H 5 G 2.80 F 4.5
1.634 ^ V	1.584	1.661	.077	SWAMBOITE $\text{U}_3\text{H}_6(\text{UO}_2)_6(\text{SiO}_4)_6 \cdot 30\text{H}_2\text{O}$	34°	---	MCL acic	---	Pale yellow	
1.685	1.661	1.661	.023	PENNANTITE (Chlorite grp) $\text{Mn}_5\text{Al}(\text{Si}_3\text{Al})_0\text{Al}_{10}(\text{CH})_8$	~ 0° r < v str	---	MCL	001 perf	Orange-brown	
1.640	1.661	1.663	.015	VARYNENITE $\text{MnBeO}_4(\text{OH}, \text{F})$	46° r > v mod	$y = \frac{b}{c}$ $z:c = -31^\circ$	MCL pris	Rose	H 5 G 3.18	
1.667	1.667	1.667	.027	CELADONITE (Mica grp) $\text{K}(\text{Mg}, \text{Fe}^{+2})_2(\text{Fe}^{+3}, \text{Al})\text{Si}_4\text{O}_{10}(\text{OH})_2$	Small r > v	$y = \frac{b}{c}$ $x \sim \frac{c}{a}$ el ctv pos	MCL	001 perf	Bright green	
1.640	1.662	1.667	.027	MAGNESIO-RIEBECKITE (Amphibole grp) $(\text{Na}, \text{Ca})_2(\text{Mg}, \text{Fe}^{+2})_3$ $\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH})_2$	Med large	$y = \frac{b}{c} = 9-$ $x:c = 20^\circ$ el neg	MCL pris	110 perf at 124°	Dark blue H 5 G 3.2	
1.643 ^	1.644	(1.662)	1.663	.019						
1.654	1.662	1.668	.014							
1.697										

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ	Brief			MCL	100 dist	Brown	H 4.5 G 3.3 F 2-3		
1.656	1.662	1.668	.012	HELLANDITE (Ca, ⁺) ₆ (Al,Fe ⁺³) ₄ Si ₄ B ₄ O ₂₀ (OH) ₄	87°	X = <u>b</u>	MCL	100 dist	Brown	H 4.5 G 3.3 F 2-3	Gel with acids. Tw 100.	
1.640	1.663	1.665	.025	SEAMANITE Mn ₃ BPO ₄ (OH) ₆	40° r < v	Y = <u>b</u> Z = <u>c</u>	ORTH acic	001 dist	Pale yellow	H 4 G 3.08 F easy	Diss by acids.	
1.643 ^{1.67}	1.650	1.663	1.670	ACTINOLITE (Amphibole grp) Ca ₂ (Mg,Fe)Si ₈ O ₂₂ (OH) ₂	65° r < v	Y = <u>b</u> Z:c = 16° el pos	MCL pris	110 perf at 124°	Green	H 5.5-6 G 3.17	Insol in acids. Pleoc. FeO 14.9, MnO 0.3, Fe ₂ O ₃ 3.9%.	
1.637	1.664	1.692	.055	WILHELMVIERLINGITE CaMnFe ⁺³ (PO ₄) ₂ (OH)• 2H ₂ O	(90±4°)	X = <u>b</u> Y = <u>a</u>	ORTH	---	Yellow	H 4 G 2.58	Pleoc., X and Y light yellow, Z yellow. Re- ported to have 2V = 45° calc.	
1.640	1.664	1.675	.035	CYANOPHILLITE Cu ₁₀ Al ₄ Si ₆ O ₂₅ •25H ₂ O	(67±7°)	X = <u>c</u>	ORTH spher- ulitic	001 perf	Greenish- blue	H 2 G 3.10	---	
1.646	1.664	1.676	.030	TANEYAMALITE Na(Mn ⁺² ,Mg,Fe ⁺³) ₁₂ Si ₁₂ (0,OH) ₄₄	70°	e1 pos	TCL	010 perf	Greenish- gray- yellow	H 5 G 3.30 calc	Pleoc., X and Y nearly cols., Z pale yellow.	
1.643	1.664	1.670	.027	RANUNCULITE HAl(OH ₂)(PO ₄)(OH) ₃ • 4H ₂ O	(56±10°)	---	MCL ps orth	---	Gold- yellow	G 3.4	Pleoc., X pale greenish yellow, Y pale yellow.	
1.650 ^{1.681}	1.640	1.664	1.670	.030	CHILDRENITE (Fe,Mn)AlPO ₄ (OH) ₂ •H ₂ O	X = <u>b</u> Z:c = 4°	MCL ps orth	001 poor	Pink to brown	H 5 G 3.14 F 4	Diss by acids. FeO 14.6, MnO 13.9%.	
1.652 ^{1.669}	1.650	1.664	1.671	.021	TSCHERMAKITE (Amphibole grp) Ca ₂ (Mg,Fe ⁺²) ₃ Al ₂ (Si ₆ Al ₂) ₂₂ (OH) ₂	81°	Y = <u>b</u> ext in- clined el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.15	Insol in acids. Pleoc. FeO 9.5, Fe ₂ O ₃ 2.4%, TiO ₂ 2.4%.
1.652 ^{1.687}	1.652	1.664	1.672	.020	MAGNESIO-HASTINGSITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ Fe ⁺³ Si ₆ Al ₂ O ₂₂ (OH) ₂	80°	Y = <u>b</u> Z:c = 19° el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.18	Insol in acids. Pleoc., X pale brown, Y dark brown, Z green-brown. FeO 8.7, Fe ₂ O ₃ 2.6%, Al ₂ O ₃ 14.3, TiO ₂ 1.3%.

1.649 1. ^Y 1.694	1.654 1. <u>Y</u>	1.674 .020	MONTICELLITE Ca(Mg,Fe)SiO ₄	82° r > v	X = b Z = <u>a</u>	ORTH el clv pos	010 poor 001 perf	White, gray	H 5.5 6 3.2 F 6	
[]	1.646	1.664 1. <u>Y</u>	.018	GONYERITE (Chlorite grp?) (Mn,Mg) ₅ Fe ⁺³ (Si ₃ Fe ⁺³) ₁₀ (OH) ₈	~ 0° r < v str	ORTH(?) el clv pos	001 perf	Dark brown	H 2.5 G 3.01	
1.654	1.664 1. <u>Y</u>	1.666 .012	CUPROSKLODOWSKITE Cu(002) ₂ (Si ₁₀ 3) ₂ (OH) ₂ • 5H ₂ O	Med r > v	X ~ <u>a</u> / <u>b</u> Y ~ v	TCL fib	100	Yellow-green	H 3.85	
1.638	1.665 1. <u>Y</u>	1.676 .038	KINNOLITE Ca ₂ Cu ₂ Si ₁₀ 8(OH) ₄	68° r < v	X = b Z ~ <u>c</u>	MCL tab	010 perf 100, 001	Azure-blue	H 5 G 3.16	
1.644 1. ^Y 1.677	1.650 1. <u>Y</u>	1.679 .029	GRUNERITE (Cumingtonite ser., Amphibole grp) (Fe ⁺² ,Mg,Mn) ₇ Si ₈ O ₂₂ (OH) ₂	87° r < v	Y = b Z:c = 16° el pos	MCL pris	110 perf at 124°	Gray, brown	H 6 G 3.31 fus	
1.654 1. ^Y 1.665	1.654 1. <u>Y</u>	1.668 .009	MAGNESIOAXINITE Ca ₂ (Mg,Fe,Mn)Al ₂ BSi ₄ 0.15(OH)	76° r < v	X ~ <u>l</u> T11	TCL	100 good	Brown	H 6.5 G 3.19 F 2	
1.638	1.666 1. <u>Y</u>	1.682 .044	LEPERSONNITE CaO•(Gd,Dy,Y) ₂ O ₃ 2400 ₃ •8CO ₂ •4SiO ₂ •60H ₂ O	73°	---	ORTH acic	---	Yellow	G 3.97	
1.649 1. ^Y	1.666 1. <u>Y</u>	1.676 .027	UPALITE Al(UO ₂) ₃ (PO ₄) ₂ (OH) ₃	(74+9°)	---	ORTH acic	010 dist	Amber-yellow	G 3.5	
1.656 1. ^Y 1.685	1.666 1. <u>Y</u>	1.676 .010	CHAMOSITE (Chlorite grp) (Fe ⁺² ,Fe ⁺³ ,Mg) ₅ Al (Si ₃ Al) ₁₀ (OH, <u>Y</u>) ₈	~ 6°	X ~ <u>c</u> el pos	MCL plates	001 perf	Dark green	H 2.5 G 2.96 fus	
1.649 1. ^Y	1.520	1.667 1. <u>Y</u>	1.668 .148	STRONTIANITE (Aragonite grp) SrCO ₃	7° r < v wk	X = <u>c</u> Z = <u>a</u>	ORTH pris	110 good	White, gray	H 3.5 G 3.68 infus
1.649 1. ^Y	1.642	1.669 1. <u>Y</u>	1.669 .027	URANOPHANE Ca(UO ₂) ₂ Si ₂ O ₇ •6H ₂ O	32° r < v str	Z = <u>b</u>	MCL acic	100 perf	Lemon-to pale-yellow	H 2.5 G 3.8 infus
									Diss by HCl with eff.	
									Gel with acids. Pleoc wk, X cols, Y pale yellow, Z canary	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.655 \hat{v} 1.681	1.656	<u>1.667</u>	1.672	.016	GEOTRITE (Amphibole grp) $(\text{Mg}, \text{Fe}^{+2})_5\text{Al}_2(\text{Si}_6\text{Al}_2)_0$ $_{22}(\text{OH})_2$	57° $r > v$	$y = \frac{b}{c}$ $z = \frac{c}{e}$ el pos	ORTH pris c	210 perf	Brown, gray	H 6 G 3.22 F 4	Insol in acids. FeO 17.1, Fe ₂ O ₃ 4.3, Al ₂ O ₃ 10.3%.
1.661 $\hat{\Delta}$ 1.673	1.658	<u>1.667</u>	1.667	.009	PENNANTITE (Chlorite grp) $\text{Mn}_5\text{Al}(\text{Si}_3\text{Al})_0$ $_{10}(\text{OH})_8$	Small	---	MCL ps hex rosettes	---	Dark brown	G 3.15	"Grovesite." Pleoc in browns and reds.
1.662	1.662	<u>1.667</u>	1.669	.007	CLINOHEDRITE $\text{CaZnSiO}_3(\text{OH})_2$	Med $r > v$	$x = \frac{b}{c}$ $y:c = -28^\circ$ el neg	MCL	010 perf	Cols	H 5.5 G 3.33 F 4	Gel with acids.
1.649 \wedge	1.665	<u>1.667</u>	1.667	.002	CHLORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3(\text{Cl}, \text{F}, \text{OH})$	10°	---	MCL ps hex	---	Cols	H 5 G 3.18 F 5	Diss by acids. Cl 6.2, F 0.1, H ₂ O 0.9%.
1.655 \wedge	1.657	<u>1.667</u>	1.671	.014	HUREAUITE (Mn, Fe) ₅ (PO ₄) ₂ [PO ₃ (OH)] ₂ ·4H ₂ O	61°	$x = \frac{b}{c}$ $z:c = 61^\circ$	MCL tab	100 good	Pink to brown	H 3.5 F 2	Diss by acids. MnO 22.4, FeO 20.7%.
1.635	1.668	<u>1.702</u>	.067		SYMPLESITE $\text{Fe}_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	87° $r > v$ str	$x \perp 110$ $z:c = 32^\circ$ el cl v pos	TCL pris	110 perf	Green to blue	H 2.5 G 3.01 infus	Diss by acids. Pleoc, X deep blue, Y cols, Z yellow-green.
1.662	1.668	<u>1.672</u>	.010		ARROJADITE (K, Ba)(Na, Ca) ₅ (Fe, Mn, Mg)Al ₁₄ (PO ₄) ₁₂ (OH, F)	80° $r < v$ str	$x = \frac{b}{c}$ $y:c = 21^\circ$ el mass	MCL u mass	001 good 201 dist	Dark green	H 5 G 3.55	Diss by acids. Pleoc, X cols, Y pale green, Z pale yellow-green.
1.658	1.669	<u>1.670</u>	.012		ZINKOSITE ZnSO_4	Small $r < v$ str	$x = \frac{b}{c}$ $z = \frac{c}{e}$	ORTH plates	---	White	G 3.7	Diss by H ₂ O. Alters on exposure. Natural occurrence doubtful.
1.665 \vee	1.669	<u>1.672</u>	.007		LITHIOPHILITE (Triphyllite grp) $\text{Li}(\text{Mn}, \text{Fe})\text{PO}_4$	80° $r < v$ str	$x = \frac{c}{e}$ $y = \frac{a}{c}$	ORTH pris	100 perf 010 less so	Yellow- ish-brown	H 4-5 G 3.45	Diss by acids. MnO 22.6, FeO 12.8, MgO 6.3%.
1.664 \hat{v} 1.693	1.658	<u>1.669</u>	1.676	.018	FERROTSCHERMAKITE (Amphibole grp) $(\text{Ca}, \text{Na})_2(\text{Fe}^{+2}, \text{Mg})_3\text{Al}_2$ $(\text{Si}_6\text{Al}_2)_0$ $_{22}(\text{OH})_2$	(77+13°)	$y = \frac{b}{c}$ $z:c = 7^\circ$ el pos	MCL pris	110 perf at 124°	Green	H 5-6	Pleoc, X tan, Y green, Z bluish green.

1.619 70 pos	1.670	1.720	.101	STRUNZITE $MnFe^{+3}_{2}(PO_4)_2(OH)_2 \cdot 8H_2O$	78°	Med (8°+2°)	Z:c = 10°	MCL	Straw-yellow	6 2.52	Pleoc wk in yellows, abs Z > X > Y.		
1.657 78	1.670	1.679	.022	MAGNESIO-HORRIBLE (Amphibole grp) $Ca_2(Mg,Fe^{+2})_4Al(Si_7Al)_2(O_{22}(OH)_2$	78°	Y = $\frac{b}{el}$ pos	MCL prts	Brown	H 5-6 G 3.23	Insol in acids. Pleoc. Fe0 13.2, Fe2O3 4.7, Mn0 0.3, Mg0 10.6%.			
1.526 52	1.671	1.672	.146	ALSTONITE $BaCa(CO_3)_2$	6° r > v wk	X = $\frac{c}{Z}$	ORTH el c	Col s	H 4.5 G 3.71 F diff	Diss by acids with eff.			
1.624 1.672	1.672	1.672	.048	ANNITE (Biotite ser., Mica grp) $KFe^{+2}_3AlSi_3O_{10}(OH,F)_2$	5° r < v	Y = $\frac{b}{el}$ pos	MCL plates	Dark brown	H 2.5-3 G 3.0 F 4'	Pleoc, X brown, Y and Z dark brown. Fe0 32.1, Fe2O3 3.1, TiO2 3.2%.			
1.632 1.672	1.672	1.672	.040	Serpentine grp mineral (Mg, Ni, Fe, Al)3(Si, Al)2 $O_5(OH)_4$	0-10°	---	MCL	Dark green	---	Species not identified. Pleoc, X yellow-green, Y and Z dark green.			
1.650 1.672	1.672	1.677	.027	PARASPURRITE $Ca_5(SiO_4)_2(CO_3)_3$	77° (50+11°)	X = $\frac{b}{Z:c} \sim$ 30°	MCL	Col s	6 3.03 infus	Dec by HCl with eff. Poly tw.			
1.634 1.673	1.673	1.685	.051	DURANGITE $NaAl(AsO_4)_F$	45° (57+5°)	Y = $\frac{b}{X:c} =$ -25°	MCL prts	Orange-red	H 5 G 4.0 F 2	Pleoc, X orange-yellow, Y pale orange, Z cols.			
1.662 82	1.673	1.684	.022	TRIPPLITE $(Mn,Fe,Mg,Ca)_2P_0_4(F,OH)$	88°	Y = $\frac{b}{Z:a} =$ 42°	MCL u mass	Dark brown	H 5-5.5 G 3.58 F easy	Diss by acids. Mn0 53.8, Fe0 6.7, Ca0 2.2%.			
1.604 65	1.674	1.731	.127	BUTLERITE $Fe^{+3}SO_4(OH) \cdot 2H_2O$	Large	---	MCL prts	Deep orange	H 2.5 G 2.55	Pleoc, X nearly cols., Y and Z pale yellow.			
1.643 81	1.674	1.701	.058	CLINOKURCHATOVITE $Ca(Mg,Fe,Mn)BO_5$	85°	---	MCL	Col s	H 4.5 G 3.07	Poly tw. Diss by acids.			
1.642 70	1.674	1.699	.057	KURCHATOVITE $Ca(Mg,Mn,Fe)BO_5$	82° r > v	X = $\frac{b}{Z}$	ORTH mass	One perf 2 imperf	H 4.5 G 3.07	Diss by warm HCl.			
1.640 76	1.674	1.679	.039	SPURRITE $Ca_5(SiO_4)_2(CO_3)$	40° r > v wk	X = $\frac{b}{Z \sim a}$	MCL prts	Col s	H 5 G 3.01 infus	Dec by HCl with eff. Poly tw 001.			
1.645 46	1.674	1.680	.035	SCORZALITE (Lazulite ser.) $(Fe,Mg)Al_2(PO_4)_2(OH)_2$	64° (48+8°) r < v	Y = $\frac{b}{X \sim c}$	MCL	Dark blue	H 6 G 3.36 infus	Insol in acids. Pleoc, X cols, Y and Z dark blue. Fe0 17.7, Fe2O3 3.0, Mn0 0.5%.			

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orient- ation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.696	1.660	1.674	1.675	.015	KORNERUPINE $Mg_3Al_6(Si,Al,B)_5O_{21}$ (OH)	20° $r > v$ wk	$X = \frac{c}{b}$ $Z = \frac{a}{b}$ el neg	ORTH pris	110 good	Green to brownish-green	H 6.5 G 3.28 F 6	Insol in acids. Pleoc variable, X yellow to green, Y cols to pale brown, Z pale to dark green.
1.691	1.655	1.675	1.685	.030	DUMORTIERITE $Al_7(BO_3)(SiO_4)_3O_3$	Med $r < v$	$X = \frac{c}{b}$ $Z = \frac{a}{b}$ el neg	ORTH acic c	100 dist	Blue	H 7 G 3.25 infus	Insol in acids. Pleoc str, X deep blue, Y yellow, Z cols.
(1.650) 1.695	1.664	1.675	1.679	.015	BUSTAMITE $(Ca,Mn)_3Si_3O_9$	50-55°	$X:a \sim 15^\circ$ $y:b \sim 35^\circ$ $Z:c \sim 35^\circ$	TCL	100 perf 110, 110 good	Pink	H 6 G 3.30	MnO 27.7, CaO 24.9%
1.665	1.675	1.678	.013	GERSTMANNITE $(Mg,Mn)_2ZnSiO_4(OH)_2$	50-60°	$X = \frac{b}{c}$ $Y = \frac{a}{c}$	ORTH pris	010 good	White to pale pink	H 4.5 G 3.68	---	
1.529	1.676	1.677	.148	WITHERITE (Aragonite grp) $BaCO_3$	16° $r > v$ wk	$X = \frac{c}{b}$ $Z = \frac{a}{b}$ el neg	ORTH el c	010 dist 110 imperf	Cols, white	H 3-3.5 G 4.29 F 2.5-3	Diss by HCl. Tw pl 110 universal.	
1.665 1.681	1.670	1.676	1.681	.011	FERROAXINITE $Ca_2(Fe,Mg,Mn)Al_2BSi_4O_15(OH)$	69°	$X \sim \perp T_{11}$	TCL	100 good	Violet	H 6.5 G 3.26 F 2	Insol in acids. CaO 19.6, FeO 8.1, MgO 2.1, MnO 1.8, Fe ₂ O ₃ 0.9%.
1.660	1.676	1.686	.026	FERRIPYROPHYLITE $Fe^{+3}Si_4O_{10}(OH)_2$	(76+9°)	---	MCL	---	Brownish-yellow	H 1.5-2 G 2.99	---	
1.629	1.677	1.679	.050	SAMPLEITE $NaCaCu_5(Po_4)_4Cl \cdot 5H_2O$	23° $r > v$	$X = \frac{b}{a}$ $Y = \frac{a}{b}$	ORTH	010 perf 110, 001 good	Blue	H 4 G 3.20 F 2	Diss by acids. Pleoc, X blue-green, Y and Z turquoise-blue, abs Z = Y > X.	
1.665 1.677 pos 1.682	1.660	1.677	1.693	.033	GRUNERITE (Cummingtonite ser, Amphibole grp) $(Fe^{+2},Mg)_7Si_8O_{22}(OH)_2$	87°	$Y = b$ ext inclined el pos	MCL pris c	110 perf at 110°	Green to brown	H 7 G 3.47	Insol in acids. Pleoc, X pale yellow, Z brown. FeO 31.2, Fe ₂ O ₃ 3.4%.
1.677 pos 1.680	1.672	1.677	1.682	.010	BRONZITE (Orthopyroxene grp) $(Mg,Fe)_2Si_2O_6$	~ 90°	$Y = \frac{a}{c}$ $Z = \frac{c}{b}$ el pos	ORTH pris	210 good at 87°	Green, brown	H 5.5 G 3.3 F 5	Nearly insol in acids. Faint pleoc common.

1.670 1.685	<u>1.678</u>	1.684	.019	EDENITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₅ Si ₇ Al 0 ₂₂ (OH) ₂	65°	MCL $y = \frac{b}{c}$ $z:c = 19^\circ$ el pos	110 perf at 124°	Light green	H 5.5 G 3.15	
1.665	<u>1.665</u>	1.679	1.708	.139	CARBOCERNAITE (Ca,Fe,Sr)CO ₃	X = $\frac{b}{d}$ Y = $\frac{a}{d}$	ORTH ---	cols to yellow	H 3 G 3.66	
1.569	1.679	1.705	.081	POSNIAKITE Cu ₄ Si ₄ (OH) ₆ ·H ₂ O	56° (51+2°) r > v	MCL ---	Dark blue	H 2-3 G 3.32 F easy		
1.624	<u>1.679</u>	1.705	.081	POSNIAKITE Cu ₄ Si ₄ (OH) ₆ ·H ₂ O	57° (67+3°)	MCL ---	cols	H 3-4 G 2.94 infus		
1.530	<u>1.680</u>	1.685	.155	ARAGONITE (Aragonite grp) CaCO ₃	18° r < v wk	ORTH acid c el neg	010 dist	G (3.03)		
1.648	1.680	1.680	.068	METAKOETTIGITE (Zn,Fe ⁺³ ,Fe ⁺²) ₃ (AsO ₄) ₂ ·8(H ₂ O,OH)	87°	TCL tab	110 perf	Bluish-gray		
1.695	<u>1.695</u>	1.680	1.716	CHRYSOLITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	90°	ORTH tab	010, 001 poor	Pale green		
1.680 pos	1.6661	<u>1.680</u>	1.697	0.036	ALLEGHANYITE (Mn,Zn) ₅ (SiO ₄) ₂ (OH) ₂	X = $\frac{b}{d}$ Z = $\frac{a}{d}$	MCL fr conch	Brownish to red-dish-pink	H 7 G 3.15 infus	
1.704	<u>1.704</u>	1.67	<u>1.680</u>	1.703	.03	Z = $\frac{b}{c} \approx 30^\circ$	MCL ---	Brownish to red-dish-pink	H 5 G 3.70 F 3.5	
1.780	<u>1.780</u>	1.667	<u>1.680</u>	1.687	.020	PARGASITE (Amphibole grp) NaCa ₂ (Mg,Fe) ₄ Al (Si ₆ Al ₂) ₂₂ (OH) ₂	74°	MCL el pos	110 perf at 124°	Brown, green
1.652	<u>1.667</u>	1.680	1.69	.02	METAVANMEERSCHETTE U(UO ₂) ₃ (PO ₄) ₂ (OH) ₆ · 2H ₂ O	83° (92+3°)	ORTH tab	010 good 100 less so	H 5.5 G 3.22	
(1.668)	<u>1.668</u>	1.686	(.018)	DAVRUXITE Mn ₂ Al ₁₂ (SiO ₄) ₇ ·(OH) ₆	70°	el pos	MCL fib	Yellow	G (4.49)	
1.674	<u>1.674</u>	1.685	.011	BRONZITE (Orthopyroxene ser., Pyroxene grp) (Mg,Fe) ₂ SiO ₆	79° r > v	ORTH pris c el pos	210 good at 87°	Creamy white to pale rose	H 5.5 G 3.3 F 5	
1.695	---	1.680	---	BALANGEROITE (Mg,Fe ⁺² ,Fe ⁺³ ,Mn) ₄ Si ₁₅ (O,OH) ₉₀	weak	ORTH fib	1 or more good	Brown	G 2.98	
1.677	<u>1.677</u>	1.685	.011	BRONZITE (Orthopyroxene ser., Pyroxene grp) (Mg,Fe) ₂ SiO ₆	79° r > v	ORTH pris c el pos	Nearly insol in acids. Faint pleoc common. FeO 11.1, MnO 0.5%.	Nearly insol in acids. Faint pleoc common. FeO 11.1, MnO 0.5%.		
1.695	---	1.680	---	BALANGEROITE (Mg,Fe ⁺² ,Fe ⁺³ ,Mn) ₄ Si ₁₅ (O,OH) ₉₀	weak	ORTH fib	Pleoc, yellow brown 001, dark brown 001. Am. Mineral., 68, 214-219 (1983).	Pleoc, yellow brown 001, dark brown 001. Am. Mineral., 68, 214-219 (1983).		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.674 \wedge	1.635	1.681	1.698	.063	KURCHATOVITE $\text{Ca}(\text{Mg}, \text{Mn}, \text{Fe})\text{B}_3\text{O}_5$	66° $r > v$ wk	$X = b$ $Z = c$	ORTH tab mass	One perf 2 imperf	Gray to cols	H 4.5 G 3.02	Diss by acids. Fluor bright violet in long-wave UV.
1.664 \wedge	1.646	1.681	1.687	.041	CHILDRENITE $(\text{Fe}, \text{Mn})\text{AlPO}_4(\text{OH})_2 \cdot \text{H}_2\text{O}$	50° $r > v$	$X = b$ $Y:c = 8^\circ$	MCL ps orth	001 poor	Pink to brown	H 5 G 3.19 F 4	Diss by acids. FeO 28.6, MnO 3.1%.
1.667 \wedge 1.710	1.671	1.681	1.690	.019	FERRO-GEDRITE (Amphibole grp) $(\text{Fe}^{+2}, \text{Mg})_5\text{Al}_2(\text{Si}_6\text{Al}_2)_0$ $_{22}(\text{OH})_2$	75° $r > v$ 12°	$Y = b$ $Z = c$ el pos	ORTH pris \perp	210 perf	Brown, gray	H 6 G 3.33 F 4	Insol in acids. Pleoc. FeO 24.4, Fe ₂ O ₃ 0.9, Al ₂ O ₃ 13.7%.
1.676 \wedge 1.687	1.674	1.681	1.688	.014	TINZENITE (Axinite grp) $(\text{Ca}, \text{Mn}, \text{Fe})_3\text{Al}_2\text{Si}_4\text{O}_{15}$ (OH)	Large $r < v$	$x \perp T_{11}$	TCL	100 good	Yellow-green	H 6.5 G 3.28 F 2	Insol in acids. CaO 15.0, MnO 15.8, MgO 1.3, FeO 0.4%.
(1.62)					MUNDITE $\text{Al}(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_3 \cdot 5.5\text{H}_2\text{O}$	33°	el neg	ORTH	001, 100, 010 good	Pale yellow	G (4.30)	Pleoc wk.
1.637	1.682	1.688	(.070)		WROEWOLFITE $\text{Cu}_4(\text{SO}_4)(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	53°	---	MCL	010, 100, 001 perf	Deep greenish-blue	H 2.5 G 3.27	Pleoc, X light blue, Y deep greenish-blue, Z med greenish-blue, abs Y > Z > X.
1.677 \wedge 1.700	1.666	1.682	1.694	.057	DANNEMOREITE (Cummingtonite ser., Amphibole grp) $\text{Mn}_2(\text{Fe}^{+2}, \text{Mg})_5\text{Si}_8\text{O}_{22}$ $(\text{OH})_2$	88° $r < v$	$Y = b$ $Z:c = 15^\circ$ el pos	MCL	110 perf at 125°	Gray	H 5.5-6 G 3.50	Insol in acids. Pleoc, X red-violet, Y pale violet, Z blue. FeO 22.6, MnO 15.6%.
1.673 \wedge	1.671	1.682	1.690	.019	TRIPOLITE $(\text{Mn}, \text{Fe}, \text{Mg}, \text{Ca})_2\text{P}_3\text{O}_4$ (F, OH)	Large $r > v$	$Y = b$ $Z:a = 42^\circ$	MCL	001 good 010 less so	Brown	H 5-5.5 G 3.8	Diss by acids.
1.656 \wedge 1.695	1.667	1.683	1.692	.025	MAGNESIO-ARFVEDSONITE (manganosite) (Amphibole grp) $(\text{Na}, \text{Ca})_3(\text{Mg}, \text{Fe}^{+2})_4\text{Fe}^{+3}$ $\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	62° $(73-9^\circ)$	$Y = b$ $Z:c = 3^\circ$	MCL	110 perf at 124°	Dark bluish-green	H 5.5-6 G 3.27	Insol in acids. Pleoc, X and Y blue-green, Z yellow-green. FeO 1.0, Fe ₂ O ₃ 9.1, MgO 11.3, MnO 7.7%.

1.525	<u>1.684</u>	1.686	.161	BARYTOCALCITE $\text{BaCa}(\text{CO}_3)_2$	15° r > v wk	$Z = \frac{b}{c}$ $x:c = 64^\circ$	MCL el c	110 poor cols, white	H 4 G 3.65 F diff	Diss by HCl with eff.	
1.593	<u>1.684</u>	1.698	.105	YAVAPALITE $\text{KFe}(\text{SiO}_4)_2$	30° (41+3°) r > v str	$Z = \frac{b}{c}$ $x:c = 6^\circ$	MCL perf 110 dist	001, 100 pink to cols	H 2.5-3 G 2.88	Dec by hot H_2O , diss by HCl.	
1.668	<u>1.684</u>	1.685	.017	WALSTROMITE $\text{BaCa}_2\text{Si}_3\text{O}_9$	30°	---	TCL	---	G 3.67	---	
1.669	<u>1.675</u>	1.685	.010	TRIOPHYLLITE $\text{Li}(\text{Fe}, \text{Mg}, \text{Mn})\text{PO}_4$	25° r < v str	$X = \frac{c}{a}$ $y = \frac{c}{a}$	ORTH pris	100 perf 50 less	H 4-5 G 3.44	Diss by acids. FeO 32.9%, MnO 3.1, MgO 7.4%.	
1.700	<u>1.675</u>	1.685	.017	WALSTROMITE $\text{BaCa}_2\text{Si}_3\text{O}_9$	30°	---	TCL	---	G 3.67	---	
-1.63	<u>1.610</u>	1.685	.094	ROSCOEILITE (Mica grp) $\text{K}(\text{V}, \text{Al}, \text{Mg})_2\text{AlSi}_3\text{O}_{10}$ (OH) ₂	10-15° (51+3°) r < v str	$Z = \frac{b}{c}$ $x \sim \frac{c}{a}$ el pos	MCL plates	001 perf	Green	H 3 G 2.97 F 3	Slowly dec by HCl. Pleoc, X olive-green, Z greenish-brown. Char- acteristic green interf- color.
1.661	<u>1.595</u>	1.685	.090	STILPNOCELINE $\text{K}(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3})_{10}$ $\text{Si}_{12-30}(\text{OH})_{12}$	~ 0°	$Y = \frac{b}{c}$ $x \sim \frac{c}{a}$ el cTv pos	TCL and MCL	001 perf	Dark brown	H 3-4 G 2.84 F 4	Dec by acids. Pleoc, X pale yellow, Y and Z dark olive-brown. FeO 13.7, Fe ₂ O ₃ 22.0%.
1.705	<u>1.650</u>	1.685	.062	SODDITE $(\text{UO}_2)_5\text{Si}_2\text{O}_9 \cdot 6\text{H}_2\text{O}$	84°	$X = \frac{c}{b}$ $y = \frac{c}{b}$	ORTH platy	001 perf 111 good	Greenish-yellow	H 3-4 G 4.70 infus	Gel with acids. Not fluor in UV.
1.674	<u>1.659</u>	1.685	.043	TARBUTITE $\text{Zn}_2(\text{PO}_4)(\text{OH})$	50° (77+5°)	Disp str	TCL pris	001 perf	Pale green	H 3.5-4 G 4.15 F easy	Diss by acids.
1.705	<u>1.659</u>	1.685	.043	TARBUTITE $\text{Zn}_2(\text{PO}_4)(\text{OH})$	50° (77+5°)	---	MCL tab	010 good 100 poor	Yellow to yellow-green	H 2-3 G 3.8-3.9	Gel with acids. Pleoc wk, X col, Y and Z light yellow.
1.680	<u>1.664</u>	1.685	.030	URANOPHANE-BETA $\text{Ca}(\text{UO}_2)_2\text{Si}_2\text{O}_7 \cdot 6\text{H}_2\text{O}$	63° r > v str crossed	$X = \frac{b}{c}$ $z:c = 26^\circ$	MCL tab	010 good 100 poor	Yellow to yellow-green	H 2-3 G 3.8-3.9	"Arsenate-belowite." Diss by acids.
1.694	<u>1.672</u>	1.685	.026	TALMESSITE $\text{Ca}_2\text{Mg}(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	~ 90°	---	TCL	---	Col s	H 5 G 3.42	Insol in acids. Pleoc, X greenish-yellow, Y olive-green, Z bluish- green. FeO 18.1, Fe ₂ O ₃ 5.4, Al ₂ O ₃ 13.9%.
1.710	<u>1.670</u>	1.693	.023	FERRO-HORNBLEND (Amphibole grp) $\text{Ca}_2(\text{Fe}^{+2}, \text{Mg})_4\text{Al}$ $(\text{Si}_7\text{Al})_0 22(\text{OH}, \text{F})_2$	61° (72+10°)	$Y = \frac{b}{c}$ $z:c = 18^\circ$ el pos	MCL pris	110 perf at 124°	Greenish- brown	H 5-6 G 3.2	Dec by acids.
1.678	<u>1.670</u>	1.685	.015	CHAMOSITE (Chlorite grp) $(\text{Fe}^{+2}, \text{Fe}^{+3}, \text{Mg})_5\text{Al}$ $(\text{Si}_3\text{Al})_0 10(0, \text{OH})_8$	~ 0°	$X \sim \frac{c}{a}$ el pos	MCL plates	001 perf	Dark brown	H 2.5 G 2.96 fus	Dec by acids. Partly dec by H_2O , alk react on.
1.666	<u>1.670</u>	1.685	.015	CHAMOSITE (Chlorite grp) $(\text{Fe}^{+2}, \text{Fe}^{+3}, \text{Mg})_5\text{Al}$ $(\text{Si}_3\text{Al})_0 10(0, \text{OH})_8$	~ 0°	$Y = \frac{b}{c}$ $x:c = 32^\circ$	MCL	110 perf	Brown	H 4.5-5 G 3.03	Dec by acids. Partly dec by H_2O , alk react on.
1.627	<u>1.686</u>	1.690	.063	SOBOLEVITE $\text{Na}_2\text{Ca}_2^{\text{?}}\text{MnTi}_3\text{Si}_4\text{O}_{18}$ $4\text{Na}_3\text{PO}_4$	29°						

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.666	1.686	1.694	.028	CLINTYROLITE $\text{Ca}_2\text{Cu}_9[(\text{As},\text{S})\text{O}_4]_4$ (0,OH) ₁₀ ·10H ₂ O	66°	$\gamma = \frac{b}{c} = 7\text{-}8^\circ$ $x:c = 7\text{-}8^\circ$	MCL silky	---	Emerald-green	G 3.22	Diss by acids.	
1.667 ⁺	1.525	1.686	1.690	.165 STRONTIANITE (Aragonite grp) SrCO_3	8° (17 - 4°) $r < \sqrt{v}$ wk	$X = \frac{c}{a}$ $Z = \frac{a}{e}$ el neg	ORTH pris	110 good	White, gray	H 3.5 G 3.81 infus	Diss by acids with eff. CaO 1.0, BaO 3.3%.	
1.658 ⁺	1.624	1.686	1.686	.062 HENDRICKSITE (Mica grp) $\text{K}(\text{Zn},\text{Mn})_3\text{Si}_3\text{AlO}_{10}(\text{OH})_2$	2-5°	---	MCL	001 perf	Dark red-brown	H 3 G 3.41	Dec by acids. ZnO 21.4, MnO 13.9, FeO 0.7, Fe ₂ O ₃ 4.85%.	
1.640	1.686	1.702	.062	CHERNYKHITE (Mica grp) (Ba,Mn)(V ⁺³ ,Al) ₂ (Si,Al)O ₁₀ (OH) ₂	11-12° (60 - 4°)	---	MCL	001 perf	Olive-to dark-green	H 2.5-4 G 3.15	$V_{2/3}$ 18.6, $V_{2/4}$ 5.3, BaO 9.5%.	
1.691	1.662	1.686	.030	KAINOSITE $\text{Ca}_2(\text{Y,Ce})_2(\text{SiO}_4)_3$ (CO ₃)·H ₂ O	40° (53 - 9°) $r < \sqrt{v}$ dist	$X = \frac{c}{b}$ $\gamma = \frac{a}{d}$	ORTH	2 clv at 90°	cols, rose, brown	H 5.5 G 3.51-3.65	Diss by acids with slight eff.	
1.682 ⁺ 1.689	1.676	1.687	.019	MAGNESIO-HASTINGSITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ (Fe ⁺³ , Ti)Si ₆ Al ₂ O ₂₂ (OH) ₂	80°	$\gamma = \frac{b}{c}$ $Z:c = 15^\circ$ el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.22	Insol in acids. Pleoc. FeO 6.9, Fe ₂ O ₃ 4.8, Al ₂ O ₃ 13.9, TiO ₂ 4.4%.	
1.664 ⁺ 1.709	1.678	1.687	.014	MANGANAXINITE $\text{Ca}_2(\text{Mn,Fe})\text{Al}_2\text{BSi}_4\text{O}_{15}$ (OH)	75°	$x \perp T_{11}$	TCL	100 good	Yellow-brown	H 6.5 G 3.31 F 2	Insol in acids. MnO 11.7, FeO 3.3, MgO 0.3, CaO 18.2%.	
1.681 ⁺ 1.693	1.658	(1.687)	1.692	.052 MAGNESIUM ASTROPHYLLITE (K,Na) ₄ Mg ₂ (Fe,Mn) ₅ Ti ₂ Si ₈ O ₂₄ (O,OH,F) ₇	82°	$\gamma = \frac{b}{c}$ $Z:c = -6^\circ$	TCL	100, 010 perf	Straw yellow	---	Pleoc, X bright yellow, Z gray, Z > Y > X.	
1.632	1.688	1.738	.106	SODIUM-ZIPPEITE $\text{Na}_4(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot (84\text{-}2^\circ)$ 4H ₂ O	80°	$Z = \frac{c}{c}$	ORTH plates	010 perf	Yellow to H 2 orange	Diss by acids. Tw common. Pleoc, X coils, Y pale yellow, Z yellow. Fluor bright yellow in UV.		

1.658	(1.688)	1.714	.056	ALVANITE $\text{Al}_6(\text{VO}_4)_2(\text{OH})_{12} \cdot 5\text{H}_2\text{O}$	80°-85° r < v str	X:b = 14° el pos and neg	MCL 010 perf	Bluish-green to bluish-black	H 3-3.5 G 2.41	Diss by hot acids. Poly tw.	
1.664	1.688	(1.707)	(.043)	WEILITE CaHfSiO_4	82°	Z:c = 20° Y: $\frac{c}{z} = 27^{\circ}$	TCL ---	White	G 3.48	Diss by acids.	
1.641	1.690	1.705	.064	LANGITE $\text{Cu}_4\text{SiO}_4(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	56-70°	X = $\frac{c}{b}$ Y = \underline{z}	ORTH 001 good 010 dist	Sky-blue to blue-green	H 3 G 3.31	Diss by acids. Poly tw 110 common. Pleoc wk, abs Y > Z > X.	
1.713	v	1.640	1.690	SHUBNIKOVITE $\text{Ca}_2\text{Cu}_8(\text{AsO}_4)_6\text{Cl}(\text{OH})_7\text{H}_2\text{O}$ (?)	Small r < v	---	ORTH(?) platy	---	---	---	
1.660	1.690	1.698	.038	BERGENITE $(\text{Ba}, \text{Ca})_2(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_4 \cdot 5\text{H}_2\text{O}$	> 45° (54+7°)	---	MCL tab	Light blue	H 2 F easy	Fluor pale green in short-wave UV.	
1.673	1.690	1.705	.032	TYROLITE $\text{CaCu}_5(\text{AsO}_4)_2(\text{CO}_3)_2(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	70° r > v str	X = $\frac{b}{a}$ Z = \underline{a}	ORTH laths	001 perf	Green to blue	H 1.5 G 3.1-3.2 F 2.5	Diss by HNO_3 or NH_4OH . Pleoc, X cols, Y and Z green.
1.726	v	1.666	1.690	CALUMETITE $\text{Cu}(\text{Cl}, \text{OH})_2 \cdot 2\text{H}_2\text{O}$	2°	X = $\frac{c}{a}$ Y = \underline{a}	ORTH el ctw pos	001 good	Azure-blue	H 2	Diss by acids. Pleoc wk, in blue, abs Z > Y > X.
1.675	1.690	1.693	.018	SINCOSITE $\text{CaV}_{4+}^4(\text{PO}_4)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	Med r > v str	---	MCL ps tet tab	001 good 100, 110 poor	Emerald-green to golden	H 3 G 2.84 F fus	Diss by acids. Pleoc, X nearly cols, Z gray-green.
1.679	1.690	1.695	.016	LAVENITE $(\text{Na}, \text{Ca})_3\text{ZrSi}_2\text{O}_7(\text{O}, \text{OH}, \text{F})_2$	68° r < v wk	Y = $\frac{b}{a}$ X:c = -20°	MCL tab on 100	100 good	Yellow, brown	H 6 G 3.41 F fus	Diff diss by acids. Tw pl 100, lam.
1.652	v 1.723	1.682	1.690	CHLOROPHENICITE $(\text{Mn}, \text{Zn})_5\text{AsO}_4(\text{OH})_7$	83° r > v str	Y = \underline{b}	MCL pris	100 perf	Gray-green	H 3-3.5 G 3.55 F diff	Diss by acids. Color pink, purple-red in str artificial light.
(1.720)	v	1.653	(1.691)	FERRI-ANNITE (Mica grp) $\text{K}(\text{Fe}^{+2}, \text{Mg})_3(\text{Fe}^{+3}, \text{Al})\text{Si}_3\text{O}_1(\text{OH})_2$	0-10°	---	MCL	001 perf	Reddish-brown	---	Pleoc, X reddish-brown, Y and Z pale yellow-green. Tw common.
1.670	v 1.722	1.691	1.692	DUMORTIERITE $\text{Al}_7(\text{BO}_3)(\text{SiO}_4)_3\text{O}_3$	Small r < v	Z = $\frac{a}{c}$ X = \underline{c}	ORTH acid c	100 dist	Blue	H 7 G 3.3 infus	Insol in acids. Pleoc, X pale blue-violet, Y and Z nearly cols.
1.670	1.692	1.713	.043	PARAKELDYSHITE $\text{Na}_2\text{ZrSi}_2\text{O}_7$	84°	X ~ $\frac{b}{c}$ Y ~ \underline{c}	TCL	001 perf 110, 110 good	Col s	H 5.5-6 G 3.39	Fluor cream in short-wave UV.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index α	<u>β</u>	γ	Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.680	<u>1.692</u>	1.720	.040	VYUNTSIAHKHITE $\text{Y}_4\text{Al}_2\text{AlSi}_5\text{O}_18(\text{OH})_5$	68°	---	MCL pris	---	Col s	H 6-7 G 4.02	Insol in HCl.	
1.710	<u>1.692</u>	1.701	.031	KAERSUTITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{TiSi}_6$ $\text{Al}_2\text{O}_2(\text{OH})_2$	81° (64+8°)	$\text{Y} = \frac{\text{b}}{\text{c}}$ $\text{Z} = \frac{\text{c}}{\text{el pos}}$	MCL pris	110 perf at 124°	Dark brown	H 6 G 3.3	Insol in acids. Pleoc., X yellow-brown, Y red- brown, Z dark red- brown. FeO 5.7, Fe_2O_3 3.3, Al_2O_3 14.2, TiO_2 5.7%.	
				KECKITE $\text{Ca}(\text{Mn}, \text{Zn})_2\text{Fe}^{+3}_3(\text{PO}_4)_4$ $(\text{OH})_3\cdot 2\text{H}_2\text{O}$	---	X:c = 15- 22° Z = <u>b</u>	MCL pris	001, 100	Brown	H 4.5 G 2.6	Pleoc., X red-brown, Y yellow, Z bright yellow, abs X > Y > Z.	
				JAGOWERITE $\text{BaAl}_2(\text{PO}_4)_2(\text{OH})_2$	100° (83+6°)	---	TCL	100, 011 good 021 fair	Light green	H 4.5 G 4.01	Insol in acids. Fluor green-white in UV. (Reported as opt pos.)	
1.672	<u>1.692</u>	1.710	.038	JAGOWERITE $\text{BaAl}_2(\text{PO}_4)_2(\text{OH})_2$	100° (83+6°)	---	TCL	100, 011 good 021 fair	Green	H 5.5 G 3.2	Insol in acids. Pleoc. FeO 20.4, Fe_2O_3 5.5, TiO_2 1.8%.	
				FERRO-FERRI- TSCHERMAKITE (Hornblende ser., Amphibole grp)	62°	---	MCL pris	110 perf at 124°				
				$\text{Ca}_2(\text{Fe}^{+2}, \text{Mg})_3\text{Fe}^{+3}_2\text{Si}_6$ $\text{Al}_2\text{O}_2(\text{OH})_2$	---	$\text{Y} = \frac{\text{b}}{\text{el pos}}$						
				TINZENITE (Axinite grp) $(\text{Ca}, \text{Fe}, \text{Mn})_3\text{Al}_2\text{BSi}_4\text{O}_15$ (OH)	Med	x ⊥ T11	TCL	100 good	Yellow	H 6.5 G 3.39	Insol in acids. MnO 18.9, FeO 0.3, CaO 13.2%.	
						---	MCL ps cub	100 poor	Yellow to brown	H 2.5 G 3.0	Diss by HCl. Abnormal interf. colors in blue and brown. Crystals zoned.	
1.687	<u>1.693</u>	1.697	.010	PHARMACOSIDERITE $\text{KFe}_4(\text{AsO}_4)_3(\text{OH})_4$ 6-7 H_2O	Large r > v	---	0RTH ps cub	010 poor	Col s	H 5-5.5 G 3.31 F 6	Gel with acids. FeO 19.3, MnO 1.1, Fe_2O_3 1.4%.	
						$\text{X} = \frac{\text{b}}{\text{c}}$ $\text{Y} = \frac{\text{v}}{\text{c}}$						
1.664	<u>1.694</u>	1.706	.032	KIRSCHSTEINITE (Monticellite ser.) $\text{Ca}(\text{Fe}, \text{Mg})\text{SiO}_4$	65° (75+7°) r > v	0RTH	010 poor	Col s				
1.680	<u>1.695</u>	1.703	.163	ARAGONITE (Aragonite grp) $(\text{Ca}, \text{Pb})\text{CO}_3$	23° r < v wk	0RTH actic	010 dist	Col s				
1.718	<u>1.695</u>	1.730	.110	ANCYLITE $\text{SrCe}(\text{CO}_3)_2(\text{OH})\cdot \text{H}_2\text{O}$	70° (66+2°) r < v	0RTH ps oct	---	Rose, pale yellow		H 4.5 G 3.82	Diss by acids with eff.	

1.65	---	1.72	.07	OTWAYITE $\text{Ni}_2\text{CO}_3(\text{OH})_2 \cdot \text{H}_2\text{O}$	---	X fib el neg	---	Bright green	G 3.41 infus	Diss slowly in dil HCl. Pleoc wk, abs Z > X. Opt char unk.	
1.668	<u>1.695</u>	1.698	.030	BOLTWOODITE $(\text{H}_3\text{O})\text{K}(\text{UO}_2)\text{SiO}_4 \cdot \text{H}_2\text{O}$	Large (36+11°)	MCL ps orth fib b	010 perf 001 poor	Pale yellow	H 3.5-4 G 3.6	Abnormal blue interf colors. Fluor dull green in UV.	
^Λ	<u>1.682</u>	<u>1.695</u>	.025	JAHNSITE $\text{CaMn}(\text{Mg},\text{Fe}^{+2})\text{Fe}^{+3}\text{Si}_2$ $(\text{PO}_4)_4(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Large	Z = $\frac{b}{a}$ el pos	MCL pris tab	001 good	Brown to yellow	H 4 G 2.72	Diss by acids. Tw 001. Pleoc, X brown, Y red- brown, Z yellow, abs Y > Z > X. Fe ⁰ 22.8, MnO 10.2, CaO 2.6%.
1.658	<u>1.695</u>	1.707	.025	MARICITE NaFePO_4	43° r > v wk	X = $\frac{a}{b}$ Y = $\frac{c}{b}$ el pos	ORTH	---	Cols to gray to pale brown	H 4-4.5 G 3.66	---
1.67	<u>1.695</u>	1.698	.02	BRONZITE (Orthopyroxene ser., $(\text{Mg},\text{Fe})_2\text{Si}_2\text{O}_6$)	69° r > v	Y = $\frac{a}{c}$ Z = $\frac{c}{b}$ el pos	ORTH pris c	210 good at 87°	Green, brown	H 5.5 G 3.43 F 5	Nearly insol in acids. Faint pleoc common, FeO 17.1, Fe ²⁺ 0.95, MnO 0.4% (En 73%).
1.680	<u>1.695</u>	1.700	.015	BUSTAMITE $(\text{Mn},\text{Ca})_3\text{Si}_3\text{O}_9$	43°	X:c ~ 15° Y:b ~ 35° Z:c ~ 35° el neg	TCL	100 perf 110, 110 good	Pink	H 6 G 3.39	MnO 31.7, CaO 18.2, FeO 0.5, MgO 1.2%.
^Λ	<u>1.713</u>	<u>1.695</u>	1.697	ARFVEDSONITE (Amphibole grp.) $(\text{Na},\text{Ca})_3(\text{Fe}^{+2},\text{Mg})_4$ $\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH},\text{F})_2$	69°	X:c = 29° el neg	MCL	110 perf at 124°	Dark green	H 5-6 G 3.38	Insol in acids. Pleoc, X and Z dark green, Y brown. FeO 20.6, Fe ²⁺ 9.1, MnO 2.3, Na ₂ O 6.8, K ₂ O 3.6%.
1.675	<u>1.682</u>	1.695	.015	KEMPITE $\text{Mn}_2\text{Cl}(\text{OH})_3$	Med	X = $\frac{c}{b}$ Y = $\frac{c}{b}$	ORTH pris	---	Emerald- green	H 3.5 G 2.94	Diss by HCl.
1.683	<u>1.685</u>	1.695	.013	FERROAXINITE $\text{Ca}_2(\text{Fe},\text{Mn},\text{Mg})\text{Al}_2\text{BSi}_4$ 015(OH)	70°	X ~ \perp T11	TCL	100 good	Brownish- blue	H 6.5 G 3.28 F 2	Insol in acids. CaO 19.4, FeO 7.7, MnO 2.3, MgO 11.7, Fe ²⁺ 2.26.
1.684	<u>1.695</u>	1.698	.014	KORNERUPINE $\text{Mg}_3\text{Al}_6(\text{Si},\text{Al},\text{B})_5\text{O}_{21}$ (OH)	48° r > v wk	X = $\frac{c}{b}$ Z = $\frac{c}{b}$ el neg	ORTH pris	110 good	Greenish- black	H 6.5-7 G 3.45 F 6	Insol in acids. Pleoc, X cols to green, Y yellow, Z dark green.
1.693	<u>1.687</u>	<u>1.695</u>	.011	RIEBECKITE (Amphibole grp.) $\text{Na}_2(\text{Fe}^{+2},\text{Mg})_3\text{Fe}^{+3}$ $0_{(2\text{OH})_2}$	81°	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el neg	MCL pris	110 perf at 124°	Dark blue	H 5 G 2.94	Insol in acids. Pleoc, X and Y dark blue, Z yellow-green. FeO 9.5, Fe ²⁺ 0.3 18.3, MnO 6.0%.
^Λ	<u>1.701</u>	<u>1.682</u>	1.696	SINHALITE MgAlBO_4	56° disp str	---	ORTH	---	Brown to yellow	H 7 G 3.50	Insol in acids. Pleoc, X dark brown, Y pale brown.
^Λ	<u>1.700</u>	<u>1.694</u>	<u>1.697</u>								
^Λ	<u>1.704</u>	<u>1.669</u>	<u>1.698</u>								

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
1.656	<u>1.699</u>	1.731	.075	KUPLETTSKITE (K, Na) ₃ (Mn, Fe) ₇ (Ti, Nb) ₂ Si ₈ O ₂₄ (OH) ₇	79° r > v str	e1 pos	TCL	001 perf	Dark brown	H 3 G 3.22 F easy	Gel with acids. Pleoc, X orange-yellow, Z brown, abs Z > X. MnO 23.6, FeO 5.4, Fe ₂ O ₃ 7.8%.	
1.639	<u>(1.699)</u>	1.704	.065	SUSSEXITE (Mn, Mg)BO ₂ (OH)	Small r > v	e1 neg	ORTH fib c	---	Brown	H 3 G 3.12 F 3	Slowly diss by acids. MnO 37.6, ZnO 3.9%.	
1.646 △ 1.713	1.699	1.704	.016	FERRO-RICHTERITE (Amphibole grp) Na ₂ Ca(Fe ⁺² , Mg) ₅ Si ₈ O ₂₂ (OH) ₂	35° (68+15°) r < v	Y = b ext Inclined el pos	MCL	110 perf at 124°	Brown to dark green	H 5-6 G 3.4	Insol in acids. Pleoc. FeO 27.3, Fe ₂ O ₃ 6.2, MnO 1.1, CaO 6.1%.	
1.631 ^	1.688	1.699	1.704	PHOSPHURANYLITE Ca(UO ₂) ₃ (PO ₄) ₂ (OH) ₂ · 6H ₂ O	5-20° r > v str	Z = c X = <u>a</u>	ORTH laths	100 perf 010 good	Golden-yellow	H 2.5 G 4.1	Pleoc, X cols to pale yellow, Y and Z golden-yellow. Not fluor in UV.	
1.660 ^	1.700	1.701	.041	GRUNERITE (Cummingtonite ser., Amphibole grp) (Fe, Mg) ₇ Si ₈ O ₂₂ (OH) ₂	86° r < v	Y = b Z:c = 14° el pos	MCL pris	110 perf at 124°	Greenish-black	H 6 G 3.4	Pleoc, X and Y cols, Z yellow to brown. FeO 45.0, MnO 0.4%.	
1.724	1.700	1.719	.040	TARAMITE, var Mbozite (Amphibole grp) Na ₂ Ca(Fe ⁺² , Mg) ₃ Al ₂ (Si ₆ Al ₂) ₂₂ (OH) ₂	(46±15°)	Y = b	MCL	110 perf at 124°	Blue-green	---	Pleoc, X light yellow, Y violet gray, Z blue.	
1.682 △	1.679	1.700	.019	LITHIOPHILITE (Triphyllite grp) Li(Mn, Fe)PO ₄	~ 70° r < v	X = c Y = <u>a</u>	ORTH	100 perf	Brown, salmon	H 4-5 G 3.55 F 1.5	Diss by acids. MnO 37.3, FeO 8.6, MgO 0.7%.	
1.713 ^	1.684	1.700	1.703	.006	TINZENITE (Axinite grp) (Ca, Mn, Fe) ₃ Al ₂ BSi ₄ O ₁₅ (OH)	x ⊥ T11	TCL	100 good	Yellow	H 6.5-7 G 3.29	Insol in acids. MnO 21.2, CaO 12.5, Fe ₂ O ₃ 1.6%.	
1.684 ^	1.696	1.700	1.702	.011	BARYLITE BaBe ₂ Si ₂ O ₇	70°	X = b Y = <u>c</u>	ORTH	001, 100 good	White	H 7 G 4.07 infus	Insol in acids. Fluor bright blue in UV.
1.693 ^	1.693	1.701	1.704	.013								

1.680 1. 1.720	1.683 <u><u>1.704</u></u>	1.722	.039	CHRYSLITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	85°	X = b Z = <u>a</u>	ORTH	010 fair	Green	H 7 G 3.53 infus
1. 1.720	1.680 <u><u>1.704</u></u>	1.712	.032	PARNAUITE Cu ₉ (AsO ₄) ₂ (OH) ₁₀ •7H ₂ O	60°	X = b Y = <u>a</u>	ORTH blades on 010	---	Green, blue, blue-green	H 2 G 3.09
1. 1.735	1.596 <u><u>1.705</u></u>	1.705	.109	STILPNOMELANE K(Fe ⁺² ,Fe ⁺³ ,Mg,Al) ₁₀ Si ₁₂ O ₃₀ (OH) ₁₂	~ 0°	Y = b X ~ c el cT _v pos	MCL or TCL	001 perf	Dark brown	H 3-4 G 2.82 F 4
1. 1.685	1.660 <u><u>1.705</u></u>	1.713	.053	TARBUTTITE Zn ₂ (PO ₄)(OH)	50°	Disp str	TCL pris	001 perf	Col's, yellow	H 4 G 4.12 F easy
1. 1.675	1.654 <u><u>1.705</u></u>	1.705	.030	KIVUIITE (Th,Ca,Pb)H ₂ (UO ₂) ₄ (PO ₄) ₂ (OH) ₈ •7H ₂ O	0-5° r > v	---	ORTH(?)	---	Yellow	---
1. 1.690	1.705 <u><u>1.711</u></u>	1.711	.021	YFTISITE (Y,Dy,Er) ₄ (Ti,Sn)O (SiO ₄) ₂ (F,OH) ₆	Large r > v dist	---	ORTH	---	Yellowish	H 3.5-4 G 3.96
1. 1.678	1.706 <u><u>1.721</u></u>	1.721	.043	ERNSTITE (Mn ⁺² ,Fe ⁺³ ,x)Al(PO ₄) ₂ (OH) _{2-x} ⁰ _x	74° r > v	Z = b Y: <u>c</u> = -4°	MCL radiating	010, 100 good	Yellow- brown	H 3-3.5 G 3.07
1. 1.688	1.707 <u><u>1.725</u></u>	1.725	.037	LERMONTOVITE U ⁺⁴ (PO ₄)(OH) ₂ SAPPHIRINE	(88+6°)	Z = <u>c</u>	ORTH fib	---	Gray- green	---
1. 1.704	1.723 <u><u>1.707</u></u>	1.710	.006	(Mg,Al) ₈ (Si,Al) ₆ • ₂₀	85° r < v str	Y = b Z: <u>c</u> = 12°	MCL tab	010, 100 poor	Pale blue to green	H 7.5 G 3.50 infus
1. 1.707 pos	1.689 <u><u>1.707</u></u>	1.727	.038	GAIHITE Ca ₂ (Zn,Mg)(AsO ₄) ₂ • 2H ₂ O	85° (92+6°)	---	TCL	010, 001, 011 good	Col's, white	H ~ 5
1. 1.695	1.695 <u><u>1.708</u></u>	1.710	.015	BUSTAMITE (Mn,Ca,Fe) ₃ Si ₃ O ₉	34° r < v	X: <u>a</u> ~ 15° Y: <u>b</u> ~ 35° Z: <u>c</u> ~ 35°	TCL	100 perf 110, 1T0 good	Pink	Mn 0 30.4, Ca 0 12.8, Fe 0 6.9, Mg 0 1.2%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula ^a	$2V_x^{calc}$ ($2V_x$ disp)	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.687 1.731	1.688 <u>1.709</u>	<u>1.714</u>	.026 HASTINGSITE (Amphibole grp) NaCa ₂ (Fe ⁺² ,Mg) ₄ Fe ⁺³ Si ₆ Al ₂ O ₂₂ (OH) ₂	51°	$\gamma = \frac{b}{c} = 12^\circ$ el pos	MCL pris	110 perf at 124°	Dark green	H 5-6 G 3.45	Insol in acids. Pleoc, X greenish-yellow, Y dark olive, Z dark green. FeO 22.9, Fe ₂ 3 5.9, Al ₂ O ₃ 11.5, TiO ₂ 1.7%.
1.692 1.730	1.687 <u>1.710</u>	<u>1.725</u>	.038 KAERSUTITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ TiSi ₆ Al ₂ O ₂₂ (OH) ₂	77°	$\gamma = \frac{b}{c} = 4^\circ$ el por	MCL pris	110 perf at 124°	Brownish-black	H 6 G 3.3	Insol in acids. Pleoc, X pale yellow-brown, Y red-brown, Z dark reddish-brown. FeO 4.9, Fe ₂ 3 9.3, Al ₂ O ₃ 15.5, TiO ₂ 5.9%.
1.685 1.694	1.695 <u>1.710</u>	<u>1.725</u>	.030 TALMESSITE Ca ₂ (Mg,Co,Ni)(AsO ₄) ₂ 2H ₂ O	Large	---	TCL	---	Pink, green	H 5 G 3.57	"Arsenate-helovite." Diss in acids. Pink var is pleoc: cols to rose. Up to 8.5% FeO or NiO.
1.681	1.694 <u>1.710</u>	<u>1.722</u>	.028 FERRO-GEDRITE (Amphibole grp) (Fe ⁺² ,Mg) ₅ Al ₂ (Si ₆ Al ₂) O ₂₂ (OH) ₂	82° r > v	$\gamma = \frac{b}{c} = \frac{1}{2}$ el pos	ORTH pris c	210 perf	Green	H 6 G 3.57 F 4	Insol in acids. Pleoc, X pale green, Y brownish-green, Z bluish-green. FeO 2.3, Al ₂ O ₃ 19.7, FeO 35.5, MnO 2.3%.
1.729	1.705 <u>1.711</u>	<u>1.715</u>	.010 CHLORITOID (Fe,Mg,Mn) ₂ Al ₄ Si ₂ O ₁₀ (OH) ₄	60°	$\chi = \frac{b}{c} = 12^\circ$ el clv pos	MCL	001 perf	Greenish-gray	H 6.5 G 3.5 F diff	Dec by H ₂ SO ₄ . Pleoc, X green, Y blue, Z yellow. FeO 19.9, Fe ₂ O ₃ 4.6%.
1.708 pos 1.738	1.706 <u>1.711</u>	<u>1.714</u>	.008 ALLANITE (Epidote grp) (Ce,Ca,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	61°	$\gamma = b$	MCL	001, 100, 110 poor	Brown to black	H 6 G 3.8 F 3	Slowly attacked by HCl, gel. FeO 4.9, FeO 9.1, ThO ₂ 1.7%.
1.725	1.670 <u>1.712</u>	<u>1.750</u>	.080 WHITMOREITE (Fe ⁺² ,Mn)Fe ⁺³ (OH) ₂ ·4H ₂ O ₂ (PO ₄) ₂	Rather large	$\gamma = \frac{b}{c} \sim \frac{1}{2}$	MCL	100 fair	Brown to greenish-brown	H 3 G 2.88	Tw 100. Pleoc wk, X cols, Y pale yellow, Z yellow-green.
(1.699) 1.728	1.642 <u>1.713</u>	<u>1.721</u>	.079 SUSSEXITE (Mn,Mg)BO ₂ (OH)	37°	el neg	ORTH fib c	---	Brown	H 5 G 3.05 F 3	Slowly diss by HCl. MnO 40.4%.

1.690 ^	<u>1.654</u>	<u>1.713</u>	1.722	.068	LANGITE $\text{Cu}_4\text{Si}_4(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	$\sim 70^\circ$ (42+5°)	$X = \frac{c}{b}$ $Y = \frac{c}{b}$	001 good 010 dist	H 3 G 3.31
1.703	<u>1.713</u>	1.722	.019	GERHARDTITE $\text{Cu}_2\text{NO}_3(\text{OH})_3$	Large $r > v$ str	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	001 perf 100 less	Emerald-green	H 2 G 3.40
1.695 ^ v	1.703 ^	<u>1.713</u>	1.717	.014	HYPERTHENE (Orthopyroxene ser., Pyroxene grp) $(\text{Mg}, \text{Fe})_2\text{Si}_2\text{O}_6$	53°	$Y = \frac{a}{c}$ $Z = \frac{c}{c}$ el pos	001 perf 100 less	F 2
1.700 ^	1.705 ^	<u>1.713</u>	1.715	.010	TARAMITE, var. Mboziite (Amphibole grp) $\text{Na}_2\text{Ca}(\text{Fe}^{+2}, \text{Mg})_3\text{Al}_2$ $(\text{Si}_6\text{Al}_2)_2(\text{OH})_2$	54°	$Y = \frac{b}{c}$ $Z:c = 11^\circ$ el pos	MCL	---
1.740 ^	1.690	<u>1.714</u>	1.735	.045	SCHOEPITE $\text{UO}_3 \cdot 2\text{H}_2\text{O}$	85° $r > v$ str	$X = \frac{c}{b}$ $Y = \frac{b}{b}$ el pos	ORTH tab	---
1.701	<u>1.714</u>	1.720	.019	NIOCALITE $\text{Ca}_4\text{NbSi}_2\text{O}_{10}(0, \text{F})$	56°	$X = \frac{b}{c}$ $Z:c = 12^\circ$	MCL	001 perf	
1.663	<u>1.715</u>	1.734	.071	RUIZITE $\text{CaMn}^+3\text{Si}_2\text{O}_6(\text{OH}) \cdot 2\text{H}_2\text{O}$	60° $r > v$ str inclined	$Y = \frac{b}{c}$ $Z:c = 44^\circ$	MCL el b	---	
1.667 (1.715)	1.737	.070	Unnamed arsenate-sulfate of Cu	Med large	---	ORTH(?)	---	Conch	
1.665	<u>1.715</u>	1.715	.050	KHIBINSKITE $\text{K}_2\text{ZrSi}_2\text{O}_7$	6-16°	$Z = \frac{b}{c}$ $X:c = 34^\circ$	MCL mass	Lemon-yellow	
1.704	<u>1.715</u>	1.724	.020	XANTHOXENITE $\text{Ca}_4\text{Fe}_2(\text{PO}_4)_4(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	Large $r < v$ str	---	Plates	Orange to brown	
1.704	<u>1.715</u>	1.718	.014	VANMEERSCHEITE $\text{U}(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_6 \cdot 4\text{H}_2\text{O}$	56°	---	ORTH tab	Yellow	
1.603	<u>1.716</u>	1.742	.139	DUNDASITE $\text{PbAl}_2(\text{CO}_3)_2(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	Large $(48+2^\circ)$	$X = \frac{a}{b}$ $Y = \frac{b}{b}$	ORTH aggregates	White	
									H 2 G 3.55 infus
									Diss by acids with eff.
									Diss by acids. Poly tw Z > X.
									Diss by acids. Pleoc, X and Y green, Z blue.
									Pleoc common. FeO 25.7, Fe ₂ O ₃ 1.0% (En 57).
									Pleoc, X yellow, Y blue-green, Z deep blue. FeO 18.7, Fe ₂ O ₃ 13.1, Al ₂ O ₃ 10.3, MnO 1.9%.
									Insol in acids. Faint pleoc common. FeO 25.7, Fe ₂ O ₃ 1.0% (En 57).
									Diss by acids. Pleoc, X nearly cols, Y and Z lemon-yellow to bright yellow. Fluor pale green in UV.
									Tw.
									Diss by warm acids. Tw common 100. Pleoc, X pale orange, Y orange, Z yellow.
									Pleoc, X cols, Y and Z pale green to bluish-green.

									Pleoc, X cols, Z yellow. Abnormal blue inter colors. FeO 23.1, MnO 4.5%.
									Pleoc in yellow.
									Diss by acids with eff.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.722	1.692	<u>1.716</u>	1.729	.047	GLAUCOCHROITE CaMnSiO_4	$X = \frac{b}{a}$ $r > v$	ORTH pris	---	Pale pink to violet	H 6 F 3	Gel with acids. MnO 32.7, FeO 4.2, MgO 2.1%.	
1.729	1.695	<u>1.716</u>	1.725	.030	SONOLITE, zincian (Mn,Zn) ₉ (SiO ₄) ₄ (OH,F) ₂	Large	---	MCL	Dark brown to brownish-black	H 5.5 G 3.8	Gel with acids. Tw common 001.	
1.700	1.716	<u>1.726</u>	.026	WOEHLERITE $\text{NaCa}_2(\text{Zr,Nb})\text{Si}_2\text{O}_8$ (0,OH,F)	71-79° $r < v$ dist	$Z = \frac{b}{c}$ el cTV neg	MCL tab	010 dist	Yellow-brown	H 6 G 3.42 F 3.5	Gel with HCl. Pleoc wk, X and Y cols, Z yellow.	
1.703	1.716	<u>1.721</u>	.018	JULGOLDITE (Pumpellyite grp) $\text{Ca}_2(\text{Fe}^{+2},\text{Mg})(\text{Fe}^{+3})_2$ (SiO ₄)(Si ₂ O ₇)(OH) ₂ ·H ₂ O	Med	---	MCL	100, 001 good	Green	H 4.5 G 3.3	Gel with hot HCl. Pleoc, X pale green, Y blue-green. Tw 001.	
1.748	1.709	<u>1.716</u>	1.723	.014	THALENITE $\text{Y}_3\text{Si}_3\text{O}_{10}(\text{OH})$ (?)	70° $r < v$ (90+17°)	Y:c small	MCL pris	110 perf	Reddish-black to black	H 6 G 4.16	---
1.739	1.657	<u>1.717</u>	1.767	.110	ZIPPEITE $\text{K}_4(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 4\text{H}_2\text{O}$	Large	$Z = \frac{c}{b}$ el cTV pos	ORTH plates	010 perf	Yellow to orange	H 2 G 3.66	Diss by acids. Tw common. Pleoc, X cols, Y pale yellow, Z dark yellow. Fluor bright yellow in UV.
1.710 1.731	1.685	<u>1.717</u>	1.720	.035	KOZULITE (Amphibole grp) (Na,K) ₃ (Mn,Mg) ₄ (Fe ⁺³ ,Al)Si ₈ O ₂₂ (OH) ₂	35° $r > v$ wk	$Y = \frac{b}{c} = 25^\circ$ X:c = 25° el neg	MCL pris	110 perf	Reddish-black to black	H 5 G 3.30	Pleoc, X yellow-brown, y red-brown, Z dark brown, abs Z > Y > X. MnO 2.8, Fe ₂ O ₃ 2.9%.
1.695 1.734	1.621	<u>1.718</u>	1.745	.124	ANCYLITE $\text{SrCe}(\text{CO}_3)_2(\text{OH}) \cdot \text{H}_2\text{O}$	70° $r < v$ wk (53+2°)	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	ORTH	---	Yellow, pink, brown	H 4.5 G 4.00 infus	Diss by acids with eff.
1.709	1.718	<u>1.734</u>	.025	METAHOHMANNITE $\text{Fe}_2(\text{SO}_4)_2(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	(86+9°)	---	Mass, powdery	---	Orange	---	Diss by HCl. Pleoc, X pale yellow, Y reddish-yellow, Z reddish-brown. Reported as biax pos.	

1.705 1.73	<u>1.719</u>	1.720	.005	VESUVIANITE $\text{Ca}_{10}\text{Mg}_2\text{Al}_4(\text{Si}_{10})_5$ $(\text{Si}_{20})_2(\text{OH})_4$	30-60°	---	110 poor	Yellow, brown, green	H 6.5 G 3.4 F 3
1.660	<u>1.720</u>	1.728	.068	HYDROASTROPHYLLITE $(\text{H}_3\text{O}, \text{K}, \text{Ca})(\text{Fe}, \text{Mn})_{5-6}$ $\text{Ti}_2\text{Si}_6(\text{O}, \text{OH})_{31}$	40°	---	TCL	2 sets	Dark brown
(1.691) Δ	<u>(1.720)</u>	1.721	.044	FERRI-ANNITE (Mica grp) $\text{K}[\text{Fe}^{+2}, \text{Mg}]_3(\text{Fe}^{+3}, \text{Al})$ $\text{Si}_{30}(\text{OH})_2$	Sma 11	---	MCL	001 perf	Reddish-brown
1.694 1.734	<u>1.689</u>	1.720	.039	KIRSCHSTEINITE (Monticellite ser) $\text{Ca}(\text{Fe}, \text{Mg})\text{SiO}_4$	51° r > v	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH	010 poor	Gray
1.704 1.733	<u>1.698</u>	1.720	.038	HYALOSIDERITE (Olivine grp) $(\text{Mg}, \text{Fe})_2\text{SiO}_4$	80°	X = $\frac{b}{c}$ Z = $\frac{a}{c}$	ORTH	010 fair	Green
1.701	<u>1.720</u>	1.734	.033	HOWEITE $\text{Na}(\text{Fe}, \text{Mn})_{10}(\text{Fe}^{+3}, \text{Al})_2$ $\text{Si}_{12}(\text{OH})_{31}$	65° r < v str	---	TCL	010 good	Dark green
1.717 pos Δ 1.736	<u>1.713</u>	1.720	.014	EPIDOTE (Epidotite grp) $\text{Ca}_2(\text{Al}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	89° r > v str	X ~ $\frac{c}{b}$ el cTV pos	MCL el b	001 perf	Light green
1.704 Δ	<u>1.667</u>	1.720	.070	PARNASITE $\text{Cu}_9(\text{AsO}_4)_2(\text{SO}_4)(\text{OH})_{10}$ 7H ₂ O	60° r < v	X = $\frac{b}{c}$ Y = $\frac{a}{a}$	ORTH blades	---	Green, blue
1.715	<u>1.720</u>	1.725	.010	TRIMERITE (Ca, Mn)BeSiO ₄	83°	Opt pl and X ~ <u>001</u> ps hex tab	TCL el pos	001 dist	Cols
1.713	<u>1.721</u>	1.734	.021	HARADAITTE SFVSi ₂ O ₇	75° ($103+12^\circ$) r < v str	X = $\frac{a}{b}$ Y = $\frac{c}{b}$	ORTH	010 perf 100, 001 dist	Bright green
1.717 Δ 1.724	<u>1.713</u>	1.728	.015	KYANITE Al_2SiO_5	83° r > v wk	Z:c = -30° on 100 el pos	TCL bladed	100 perf 010 good	Bluish, greenish, cols
1.687	<u>1.722</u>	1.731	.044	TARAPACITE K_2CrO_4	52° r > v wk	X = $\frac{b}{a}$ Y = $\frac{c}{a}$	ORTH ps hex	010 perf 001 good	Bright yellow

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.691 \wedge	1.686	<u>1.722</u>	1.723	.037	DUMORTIERITE $Al_7(BO_3)(Si_4O_3)_3$	13° $r > v$ wk	$Z = \frac{a}{c}$ $X = \frac{c}{c}$	ORTH acitic c	100 dist	Brown	H 7 G 3.41 infus
1.716	1.716	<u>(1.723)</u>	1.723	.007	HATRURITE Ca_3SiO_5	Small	---	ps hex	---	Col s	---
1.675	1.675	<u>1.723</u>	1.765	.090	CLINOCHALCOMENITE $CuSiO_3 \cdot 2H_2O$	78° $(84+3^\circ)$ $r < v$	$Y = \frac{b}{c}$ $Z:c = 10^\circ$ el pos	MCL	110 perf	Bluish-green	H 2 G 3.35
1.690 \wedge	1.698	<u>1.723</u>	1.745	.047	LAVENITE $(Na,Ca)_3ZrSi_2O_7$ $(O,OH,F)_2$	80° $r < v$	$Y = \frac{b}{c}$ $X:c = -20^\circ$	MCL tab on 100	100 good	Yellow, brown	H 6 G 3.5 fus
1.746											
1.707 \wedge	1.720	<u>1.723</u>	1.725	.005	SAPPHIRINE $(Mg,Al)_8(Si,Al)_6O_{20}$	52° $r < v$ str	$Y = \frac{b}{c}$ $Z:c = 7^\circ$	MCL tab	010, 100 poor	Pale blue to green	H 7.5 G 3.58 infus
1.741											
1.700 \wedge	1.674	<u>1.724</u>	1.724	.050	PHOSPHURANYLITE $Ca(UO_2)_3(Po_4)_2(OH)_2 \cdot 6H_2O$	$5-10^\circ$ $r > v$ str	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	ORTH laths	100 perf 010 good	Golden-yellow	H 2.5 G 4.1
1.712 \wedge	1.676	<u>1.725</u>	1.745	.069	WHITMOREITE $(Fe^{+2},Mn)Fe^{+3}_2(Po_4)_2(OH)_2 \cdot 4H_2O$	$60-65^\circ$ $r > v$ str	$Y = \frac{b}{c}$ $Z ~ \frac{c}{c}$	MCL	100 fair	Brown to greenish-brown	H 3 G 2.87
1.723 \wedge	1.692	<u>1.725</u>	1.738	.046	PHOSPHOSIDERITE $FePo_4 \cdot 2H_2O$	62° $r > v$ str	$Y = \frac{b}{c}$ $X:c = 4^\circ$	MCL tab 010	010 good	Reddish-violet	H 3.5-4 G 2.76 F easy
1.728											
1.700	1.700	<u>1.725</u>	1.730	.030	VUGNATTITE $CaAlSiO_4(OH)$	48° $r < v$ str	$X = \frac{c}{c}$ $Y = \frac{b}{c}$ el neg	ORTH laths	---	White	G 3.22- 3.42
1.690 \wedge	1.694	<u>1.726</u>	1.730	.036	TYROLITE $CaCu_5(AsO_4)_2(CO_3)$ $(OH)_4 \cdot 6H_2O$	36° $r > v$ str	$X = \frac{b}{c}$ $Z = \frac{a}{c}$	ORTH laths	001 perf	Pale green to sky-blue	H 2 G 3.27 F easy
1.711 \vee	1.711	<u>1.727</u>	1.740	.029	TEPHROITE $(Olivine grp)$ $(Mn,Mg)_2SiO_4$	85° $r > v$	$X = \frac{b}{c}$ $Y = \frac{c}{c}$	ORTH 010 fair	001 good	Reddish-gray	H 5 G 3.72 F 3.5
1.766											

1.718	(~1.727)	1.728	.010	BARIUM-PHARMACOSIDERITE Ba(Fe,Al) ₄ (AsO ₄) ₃ (OH) ₅ ·5H ₂ O (?)	0-39°	---	TET (?)	100 good	Yellow-brown	H 2-3 G 3.00	Diss by acids.
1.681	1.728	1.769	.088	COBALTOMENITE CoSeO ₃ ·2H ₂ O	83° r < v	Z:c = 13°	MCL	Conch	Pink	H 2.5 G 3.42	Data for synth compd. Pleoc., X and Y pink, Z red.
1.670	1.728	1.732	.062	SUSSEXITE MnBO ₂ (OH)	30°	el neg	ORTH fib c	---	Brown	H 5 G 3.30	Calc for pure end members.
1.713	1.728	1.728	.062	SARCOPSIDE (Fe ⁺² ,Mn,Mg) ₃ (PO ₄) ₂	28° r > v	---	MCL	001, 100 good 010 poor	Gray to brown	H 4 G 3.79	Diss by acids. Poly tw 001.
1.670	1.728	1.732	.062	HOLTITE Al ₆ (Ta,Sb,Li)[(Si,As) O] ₄ ·3(BO ₃) ₂ ·OH ₃	27° r < v st.r	X = c Y = $\frac{c}{b}$	ORTH pris	100 good	Buff to greenish	H 8.5 G 3.90 infus	Insol in acids. Tw 110. Not pleoc.
1.705	1.728	1.730	.025	FERROHYPERSTHENITE (Orthopyroxene ser., Pyroxene grp.) (Mg,Fe) ₂ Si ₂ ·6	51°	Y = a Z = $\frac{c}{c}$ el pris	ORTH	210 good	Green	H 5-6 G 3.60 F 5	Insol in acids. Faint pleoc common. FeO 29.4, MnO 0.2, Fe ₂ O ₃ 2.1% (En 45).
1.758	1.715	1.728	.016	LANDESITE Mn ⁺² ·3Fe ⁺³ (PO ₄) ₂ ·(OH) ₃ 3H ₂ O (?)	Large	el pos	ORTH	010 perf	Reddish- brown	H 3-3.5 G 3.03	Diss by acids. Pleoc, X dark brown, Y light brown, Z yellow.
1.713 1.747	1.728	1.731	.016	BERMANITE Mn ⁺² ·3Fe ⁺³ (PO ₄) ₂ ·(OH) ₂ (OH) ₂ ·4H ₂ O	72° r < v str	X = b Y:c = -36°	MCL ps orth	001 perf 010	Reddish- brown	H 3.5 G 2.85 F easy	Diss by acids. Poly tw. Pleoc, X light red, Y pale yellow, Z deep red. Mn ₂ O ₃ 30.6, Fe ₂ O ₃ 3.2, MnO 12.8%.
1.720	1.728	1.735	.015	CHLORITOID (Fe,Mg,Mn)Al ₄ Si ₂ ·10 (OH) ₄	56°	X = b Z:c = 20° el clv neg	MCL	001 perf	Greenish- gray	H 6.5 G 3.5 F diff	Dec by H ₂ SO ₄ . Pleoc, X green, Y blue, Z yellow. FeO 22.2, Fe ₂ O ₃ 4.9, TiO ₂ 2.7%.
1.725	1.690	1.729	.060	Unnamed barium uranyl arsenate	Small	---	ORTH(?) ac ric	---	Yellow	---	Pleoc, X cols, Z yellow. Not fluor in UV. Mn. Mineral., 58, 561 (1973).
1.711	1.725	1.730	.005	MCGUINNESSITE (Mg,Cu) ₂ (CO ₃)(OH) ₂	(very small)	X:c = 11°	MCL fib	---	Blue- green to blue	H 2.5 G 3.02- 3.22	Diss by cold acids. Pleoc, X pale green, Y and Z light bluish- green.
1.605	(1.73)	1.738	.133								
1.602	1.730	1.732	.130								

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.710 ^Δ	1.694	1.730	1.757	.063	KAERSUTITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ TiSi ₆ Al ₂ O ₂ (OH) ₂	80°	$y = \frac{b}{c}$ $z \sim \frac{c}{a}$ el pos	MCL pris	110 perf at 124°	Brownish-black	H 6 G 3.3	Insol in acids. Pleoc., X yellow, Y red-brown, Z dark brown. FeO 7.7, Fe ₂ O ₃ 12.4, TiO ₂ 6.0%.
(1.690)	1.730	1.749	(.059)	PHURCALITE Ca ₂ (UO ₂) ₃ (PO ₄) ₂ (OH) ₄ 4H ₂ O	68°	el pos	ORTH	001, 010 perf 100	Yellow	H 3 G 4.14	Diss by acids. Pleoc., X bright yellow, Y pale yellow. Not fluor in UV.	
1.717 ^Δ	1.674	1.731	1.754	.080	ZIPPEITE K ₄ (UO ₂) ₆ (SO ₄) ₃ (OH) ₁₀ 4H ₂ O	45° r > v str	$z = \frac{c}{b}$ el cTv pos	ORTH plates	010 perf	Yellow to orange	H 2 G 3.61	Diss by acids. Pleoc., X col., Y and Z yellow. Fluor bright yellow in UV.
1.709 ^Δ	1.705	1.731	1.732	.027	HASTINGSITE (Amphibole grp) NaCa ₂ (Fe ⁺² ,Mg) ₄ Fe ⁺³ Si ₆ Al ₂ O ₂ (OH) ₂	25°	$y = \frac{b}{c}$ $z:c = 13^{\circ}$ el pos	MCL pris	110 perf at 124°	Dark green	H 5-6 G 3.43	Insol in acids. Pleoc. FeO 22.0, Fe ₂ O ₃ 11.6, Al ₂ O ₃ 11.5, MnO 0.6%.
1.711	1.731	1.732	.021	CHALCOMENITE CuSeO ₃ ·2H ₂ O	Small r < v or r > v str	X = $\frac{a}{c}$ Y = $\frac{a}{b}$ el cTv neg	ORTH	None	Clear blue	H 2.5 G 3.36 fus	Diss by acids. Pleoc., X light blue, Y and Z darker blue.	
1.680	1.732	1.775	.095	Unnamed uranyl phosphate	80°	X = $\frac{a}{c}$ Y = $\frac{a}{b}$ el cTv neg	ORTH	001 good	Yellow	---	Pleoc., X col., Y yellow, Z yellow-gold. Fluor green in UV. Am. Mineral., 59, 212 (1974).	
1.720 ^Δ 1.750	1.710	1.733	1.750	.040	HYALOSIDERITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	79°	$x = \frac{b}{c}$ $z = \frac{a}{a}$	ORTH	010 fair	Green	H 5 G 3.69 F 6	Gel with acids. FeO 32.0, Fe ₂ O ₃ 2.0, MnO 0.5% (Fo ₆₀).
1.724	1.724	1.733	1.739	.015	CHALCOCYANITE CuSO ₄	Large r > v	X = $\frac{b}{c}$ Y = $\frac{a}{c}$	ORTH tab	None	Pale green to blue	H 3.5 G 3.65	Diss by H ₂ O.
1.718 ^Δ	1.656	1.734	1.776	.120	ANCYLITE SrCe(CO ₃) ₂ (OH)·H ₂ O	70° r < v wk	X = $\frac{a}{c}$ Z = $\frac{a}{c}$	ORTH	---	Yellow, brown, pink	H 4.5 G 4.0 infus	Diss by acids with eff.
1.679	1.734	1.742	.063	HUEMULITE Na ₄ MgV ₁₀ Si ₂₈ ·24H ₂ O	28° (41+5°) r > v str	---	TCL	001 perf 010 less so	Yellow-orange	H 2.5 G 2.39	Diss by cold H ₂ O. Not fluor in UV. Pleoc., X light yellow, Y gold, Z yellow-orange.	

1.768	<u>1.734</u>	1.758	.050	ADAMITE $Zn_2(AsO_4)(OH)$	87° r > v str	X = $\frac{a}{b}$ Z = $\frac{b}{d}$	ORTH	101 good	Cols, green, pink	H 3.5 G 4.43	Diss by acids. CuO 0.6%.	
1.723	<u>1.734</u>	1.736	.013	GAGEITE $(Mn,Mg,Zn)_8Si_3O_{10}(OH)_8$	Med r < v str	Z = $\frac{c}{e}$ el pos	ORTH acid \underline{c}	---	Pink, brown	H 3 G 3.58 infus	Dec by HCl.	
1.705 $\frac{1}{\Delta}$ 1.745	<u>1.735</u>	1.735	.110	STILPNOMELANE $K(Fe^{+2},Fe^{+3},Al)_10Si_{12}O_{30}(OH)_{12}$	~ 0°	Y = $\frac{b}{c}$ X ~ $\frac{c}{e}$ el pos	MCL or TCL	001 perf	Dark brown	H 3-4 G 2.83 F 3	Dec by acids. Pleoc, X golden yellow, Y and Z deep reddish-brown. FeO 1.3, Fe 0.31.7, MnO 2.6%.	
1.707	<u>1.735</u>	1.738	.031	BERAUNITE $Fe^{+2}Fe^{+3}_5(Po_4)_4(OH)_5 \cdot 4H_2O$	25° r < v	X:c = 4-8°	MCL	100 perf	Green	H 3.5-4 G 2.9-3.1	Diss by acids. Pleoc, X blue-green, Y pale olive-green, Z olive- green.	
1.715	<u>1.735</u>	1.745	.030	SICKLERITE $Li(Mn^{+2},Fe^{+3})PO_4$	Med large r > v very str	X = $\frac{a}{e}$ el cTv pos	ORTH	100 good	Yellow- ish-to- dark brown	H 4 G 3.3 F easy	Diss by acids. Pleoc, X deep red, Y pale red, Z very pale red. MnO 33.6, MnO 2.1, FeO ₃ 11.3%.	
1.771												---
1.720	<u>1.736</u>	1.738	.018	GITTINSITE $Ca_2ZrSi_2O_7$	30°	X:c = 5- 10°	MCL fib \underline{c}	---	White	H 3.5-4 G 3.6		
1.640	<u>1.736</u>	1.750	.110	HYDROZINCITE $Zn_5(CO_3)_2(OH)_6$	40° r < v str	X = $\frac{b}{c}$ Z:c = 40°	MCL laths	100 perf	White	H 2.5 G 4.0 infus	Diss by acids with eff. Fluor pale blue in UV.	
1.654	<u>1.736</u>	1.764	.110	LOMONOSOVITE $Na_2Ti_2Si_2O_9 \cdot Na_3PO_4$	60°	el pos	TCL	100 perf	Yellow to dark brown	H 3-4 G 3.15	Dec by H ₂ O. Pleoc, X yellow, Y orange- yellow, Z dark brown.	
1.721	<u>1.736</u>	1.739	.024	REWARDITE $Pb(UO_2)_4(Po_4)_2(OH)_4 \cdot 7H_2O$	~ 40° r > v	X = $\frac{a}{c}$ Y = $\frac{b}{c}$ el cTv pos	ORTH pris	100 perf	Yellow	H 3.5 G 4.35 F easy	Pleoc, X cols, Y and Z yellow. Not fluor in UV.	
1.720 $\frac{1}{\Delta}$ 1.754	<u>1.736</u>	1.740	.019	EPIDOTE $(Epidote grp)Ca_2(Al,Fe)_3Si_3O_{12}(OH)$	78° (54+14°) r > v str	Y = $\frac{b}{c}$ X ~ $\frac{c}{e}$ el cTv pos	MCL el b	001 perf	Green	H 7 G 3.44 F 4	Insol in acids. Fe ₂ O ₃ 9.5, Fe 0.8%.	
1.723	<u>1.737</u>	1.756	.043	ROSELITE-BETA $Ca_2Si_2(AsO_4)_2 \cdot 2H_2O$	80-90° r < v	---	TCL	010 perf	Dark rose-red	H 3.5-4 G 3.71	Pleoc, X pink, Y pale pink, Z cols.	
1.708	<u>1.738</u>	1.747	.039	ZIRCONPHYLITE $(K,Na,Ca)_3(Mn,Fe)_7$	62° r > v str	Y = $\frac{a}{b}$ X:b = 10° el neg	TCL	001 perf	Dark brown	H 4-5 G 3.34	Pleoc, X and Y dark yellow, Z brown.	
1.711 $\frac{1}{\Delta}$ 1.755	<u>1.738</u>	1.742	.021	ALLANITE $(Epidote grp)(Ce,Ca,Y)_2(Al,Fe)_3Si_3O_{12}(OH)$	Large (51+13°) r > v str	Y = \underline{b}	MCL	001, 100, 110 poor	Brown, black	H 6 G 4.0 F 3	Gel with acids. Pleoc. Fe ₂ O ₃ 17.0, MnO 0.6, ThO 0.9%.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.716	1.719	1.739	1.748	.029	THALENITE $Y_3Si_3O_{10}(OH)_2$ (?)	67° $r < v$	$y \sim c$	MCL tab 100	110 perf	Reddish to black	H 6-6.5 G 4.40	---
	1.730	(1.739)	1.748	.018	SUZUKITE $Ba_2V^{+4}O_2Si_4O_{12}$	~90° $r < v$ str	$x = \frac{a}{d}$ $y = \frac{b}{c}$	ORTH	010 perf 100, 001 dist	Bright green	H 4-4.5 G 4.03	Pleoc, X pale green, Y yellowish-green, Z bluish-green, abs $x < y$ $< z$.
	1.722	(1.739- 1.740)	1.745	.023	STEIGERITE $AlVO_4 \cdot 3H_2O$	55-60°	---	MCL	010 good	Canary yellow	---	Diss by acids.
	1.683	1.740	1.755	.072	SCHOEPITE $UO_3 \cdot 2H_2O$	Med $r > v$ str	$x = \frac{c}{b}$ $y = \frac{a}{c}$ $z = \frac{b}{a}$	ORTH tab	001 perf	Yellow	H 2.5 G 4.8-5.0	Diss by acids. Pleoc, X col's, Y yellow, Z lemon-yellow. Fluor pale green in UV.
	1.714	1.720	1.740	.037	ACMITTE-AUGITE (Pyroxene grp) (Na, Ca)(Fe ⁺³ , Al, Mg, Fe ⁺²)Si ₂ O ₆	82-88° $r < v$	$y = \frac{b}{c}$ $x:c = 15^\circ$ $el \text{ neg}$	MCL	110 good at 87°	Dark green	H 6 G 3.40	Insol in acids. Pleoc, X bright green, Y and Z greenish-yellow. FeO 7.5, Fe ₂ O ₃ 14.0, Na ₂ O 5.3%.
	1.751 pos	1.756	1.740	.021	STRASHIMIRITE $Cu_8(AsO_4)_4(OH)_4 \cdot 5H_2O$	70°	Z:el = 5°	MCL spherulites	---	Green	G 3.8 calc	Pleoc wk, Y very pale yellow-green, Z yellow-green.
	1.726	(1.740)	1.747	.021	SAPPHİRINE $(Mg, Al, Fe^{+3})_8(Si, Al)_6O_{20}$	40° $r < v$ str	$y = \frac{b}{c}$ $z \sim \frac{c}{a}$	MCL tab	010, 100 poor	Blue	H 7.5 G 3.60 infus	Insol in acids. Pleoc, X pale brown, Y and Z deep blue. FeO 3.0, Fe ₂ O ₃ 10.4%.
	1.723	1.731	1.741	.012	HODGKINSONITE $MnZn_2SiO_5 \cdot H_2O$	52° $r > v$ str	$y = \frac{b}{c}$ $z:c = 38^\circ$	MCL pyram	001 good	Pale reddish-brown	H 5 G 4.01 F easy	Gel with acids. Pleoc, X and Z lavender, Y pale green.
	1.724	1.742	1.746	.022	HAGENDORFITE $(Na, Ca, Mn^{+2}, Fe^{+3}, Mg)_2(PO_4)_3$	68°	$z = \frac{b}{c}$ $y:c = -22^\circ$	MCL	---	Green	H 3.5 G 3.71	Diss by acids. Pleoc, X yellow brown, Y green, Z blue-green.
	1.735	1.742	1.745	.010	SANERDITE	44°	el pos	TCL	2 clv perf	Deep orange	G 3.47	Pleoc, X deep orange, Y lemon-yellow, Z yellow-orange.
	1.720	1.743	1.748	.028	Na ₂ (Mn ⁺² , Mn ⁺³) ₁₀ Si ₁₁ VO ₃₄ (OH) ₄	68° disp str	$y = \frac{b}{c}$ $z:c = 44^\circ$	MCL plates	010 perf	Blue	G > 3.3	---
	1.738	1.743	1.746	.008	SURINAMITE (Al, Mg, Fe) ₃ (Si, Al) ₂ (O, OH) ₈							

1. ^y ₇₀	<u>1.744</u>	1.782	.112	MURMANITE $\text{Na}_2(\text{Ti},\text{Nb})_2\text{Si}_2\text{O}_9 \cdot \text{xH}_2\text{O}$ $(68\pm 2^\circ)$ $r > v$	X: ^a small $Z = \frac{b}{c}$ el c _v pos	TCL	100 perf	Pink to yellow-brown	H 2.8 G 2.8	
1. ^y ₇₅₂	1.703	<u>1.744</u>	1.786	.083	AHLFELDITE $\text{NiSe}_3 \cdot 2\text{H}_2\text{O}$	Z:c = 12°	MCL	Light green	H 2.5 G 3.51	
1. ^y ₇₅₂	---	<u>1.74-</u> <u>1.75</u>	---	---	TINTICITE $\text{Fe}_6(\text{PO}_4)_4(\text{OH})_6 \cdot 7\text{H}_2\text{O}$	---	ORTH(?) earthy masses	---	---	
1. ^y ₇₀₇	<u>1.745</u>	1.776	.069	SHCERBAKOVITE $(\text{K},\text{Na},\text{Ba})_3(\text{Ti},\text{Nb})_2\text{Si}_4$ 0_{14}	82° r < v dist	X = $\frac{b}{a}$ Y = $\frac{c}{b}$ el pos	ORTH	Brown to dark brown	H 6.5 G 3.21	
1. ^y ₇₃₈ <u>1.752</u>	1.658	<u>1.746</u>	1.751	.093	AURICHALCITE $(\text{Zn},\text{Cu})(\text{CO}_3)_2(\text{OH})_6$	Small r < v str	ORTH fib C	010 perf	Pale green or blue	
1. ^y ₇₀₃	1.746	1.789	.086	LIBETHENITE $\text{Cu}_2\text{PO}_4(\text{OH})$	85-90° r > v str	X = $\frac{b}{c}$ Y = $\frac{c}{b}$	ORTH short pris	100, 010 poor	H 4-4.5 G 3.97 F 2.5	
1. ^y ₇₂₀	<u>1.746</u>	1.760	.040	LAVENITE $(\text{Na},\text{Ca})_2\text{rSi}_2\text{O}_7$ $(\text{O},\text{OH},\text{F})_2$	73° r < v	Y = $\frac{b}{a}$ X:c = -20°	MCL tab	100 good	Yellow to brown	
1. ^y ₇₃₆	<u>1.747</u>	1.752	.016	EULITE (Orthopyroxene ser., Pyroxene grp. $(\text{Fe},\text{Mg})_2\text{Si}_2\text{O}_6$	63°	Y = $\frac{a}{c}$ Z = $\frac{c}{b}$ el pos	ORTH	210 good	Green	
1. ^y ₇₂₈ <u>1.762</u>	1.728	<u>1.748</u>	1.754	JULGOLDITE (Pumpellyite grp.) $\text{Ca}_2(\text{Fe}^{+2},\text{Mg})(\text{Fe}^{+3},\text{Al})_2$ $(\text{Si}_1\text{O}_4)(\text{Si}_2\text{O}_7)(\text{OH})_2 \cdot \text{H}_2\text{O}$	30-60°	---	MCL	100, 001 good	Greenish-black	
1. ^y ₇₁₆ <u>1.79</u>	1.739	<u>1.749</u>	1.752	PRZHEVALSKITE $\text{Pb}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	30° (57±19°)	el neg	ORTH tab	001 good	Bright yellow	
1. ^y ₇₃₆	1.670	<u>1.750</u>	1.778	.108	LOMONOSOVITE $\text{Na}_2\text{Ti}_2\text{Si}_2\text{O}_9 \cdot \text{Na}_3\text{PO}_4$	~ 60°	el pos	TCL	100 perf	Dark brownish-yellow
1. ^y ₇₂	1.75	1.82	.10	MAGNESIUM-ZIPPELITE $\text{Mg}_2(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 16\text{H}_2\text{O}$	Large	Z = C	ORTH plates	010 perf	Yellow to orange	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.733 \downarrow 1.770	1.721 <u>1.750</u>	<u>1.750</u>	1.765	.044	HYALOSIDERITE (011vne grp) (Mg,Fe) ₂ SiO ₄	$X = b$ $Z = \frac{b}{a}$	ORTH	010 fair	Green	H 5 G 3.7 F 5.5	Gel with acids. FeO 36.8, MnO 0.5% (FeO ₅₂).
1.776	1.733 \downarrow	<u>1.750</u>	1.762	.029	PIEMONITE (Epidote grp) Ca ₂ (Al,Mn,Fe) ₃ Si ₃ O ₁₂ (OH)	$Y = b$	MCL el b	001 perf	Reddish-brown	H 6.5 G 3.40 F 3	Pleoc., X yellow, Y pale amethyst, Z deep purplish-red. FeO 3 11.3, Mn ₂ O ₃ 1.0%.
1.746 pos \downarrow 1.753 pos	1.743 <u>1.751</u>	<u>1.751</u>	1.758	.015	STAUROLITE (Fe,Mn,Zn) ₂ Al ₉ (Si,Al) ₄ 0 ₂₂ (OH) ₂	$Z = c$ $X \sim \frac{c}{b}$ el clv pos	MCL ps orth	010 good	Reddish-brown	H 7 G 3.74 infus	Insol in acids. Pleoc., X cols, Y pale yellow, Z golden yellow. FeO 11.2, FeO ₃ 2.9%. U opt pos.
1.735	<u>1.753</u>	<u>1.753</u>	1.767	.032	SURSASSITE Mn ₅ Al ₄ Si ₅ O ₂₁ ·3H ₂ O (?)	$Y = b$ $X:C = 55^\circ$ el clv	MCL el b	001	Brown to reddish-brown	G 3.26	Pleoc., X and Z cols to pale yellow, Y golden to reddish-brown. May show abnormal interf. colors.
1.736 \downarrow 1.768	1.729 <u>1.754</u>	<u>1.754</u>	1.776	.047	EPIDOTE (Epidote grp) Ca ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	73° $(85+5^\circ)$ r > v str	MCL el clv pos	001 perf	Green	H 7 G 3.45 F 4	Insol in acids. Pleoc., X pale yellow, Y pale greenish-yellow, Z greenish-yellow. FeO 14.0, FeO 1.2%.
1.743	<u>1.754</u>	<u>1.764</u>	1.764	.021	CARACOLITE Na ₃ Pb ₂ (SO ₄) ₃ C1	---	MCL ps hex	---	cols	H 4.5 F 2	Insol in H ₂ O, dec by HC1. Complex tw.
(1.710)	<u>1.755</u>	<u>1.775</u>	(.065)		SCHUITINGITE PbCu(Nd,Gd)(CO ₃) ₃ (OH)· 1.5H ₂ O	Very large r > v str	Ext 0-10°	ORTH	el	Turquoise-to-azure-blue	Diss by HNO ₃ with eff.
1.731	<u>1.755</u>	1.755	.024		ERICAITIE (Fe,Mn,Mg) ₃ B ₇ O ₁₃ C1	Small	---	ORTH ps trig	---	Pale red	G 3.5
1.738 \downarrow 1.784	1.740 <u>1.755</u>	<u>1.760</u>	.020		ALLANITE (Epidote grp) (Ce,Ca,V) ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	Large r > v str	Y = b	MCL 001, 100, 110 poor	Brownish-black	H 6 G 3.90 F 3	Gel with acids. Pleoc., FeO 3.2, FeO 10.7, MnO 1.9, TiO ₂ 1.7, ThO ₂ 0.9%.

1.740 1.767	<u>1.732</u>	<u>1.756</u>	1.770	.038	ACMITE, manganese, (Pyroxene grp) (Na,Ca)(Fe ⁺³ ,Mg,Mn) Si ₂ O ₆	r > v 80°	$\frac{Y = b}{X:c} = 9^{\circ}$ el neg	MCL 110 good at 87°	Pink	H 6 G 3.28
1. 1.765	1.740	<u>1.756</u>	1.762	.022	UREYITE (Pyroxene grp) NaCrSi ₂ O ₆	65°	$\frac{Y = b}{X:c} = 14^{\circ}$	MCL 110 good ---	Emerald-green	---
1. 1.728	1.692	<u>1.757</u>	1.800	.108	MUNIRITE NaV ⁺⁵ O ₃ ·2H ₂ O	75°	$X = \frac{c}{a}$ Y = $\frac{b}{d}$	ORTH ---	Greenish-white	6 2.43
1. 1.745	1.745	<u>1.758</u>	1.760	.015	HOLTITE Al ₆ (Ta,Sb,Li)[(Si,As) O ₄] ₃ (BO ₃) ₂ O ₁₀ H) ₃	52° r < v	$X = \frac{c}{b}$ Y = $\frac{b}{d}$	ORTH pris	Buff, greenish to brown	H 8.5 G 3.90 infus
1. 1.705	1.760	<u>1.770</u>	1.770	.065	PARASCHOEPITE UO ₃ ·2H ₂ O (?)	46° r > v	$X = \frac{c}{b}$ Y = $\frac{b}{d}$	ORTH ---	Yellow	H 2
1. 1.724	1.760	<u>(1.772)</u>	(1.772)	(.048)	NIOBOPHYLLITE (K,Na) ₃ (Fe,Mn) ₆ (Nb,Ti) ₂ Si ₈ (0,OH,F) ₃₁	60° X:b = 13°	$Y = \frac{a}{b}$	TCL 001 perf	Brown	G 3.43
1. 1.72	1. 1.763 pos	<u>1.76</u>	1.76	.04	META-UANOPILITE (UO ₂) ₆ (SO ₄) ₁₀ ·5H ₂ O (Fe,Mg) ₂ Si ₁₀ O ₂₆	Small 87°	$X = \frac{b}{c}$ $Y = \frac{a}{c}$ el pris	ORTH ---	Yellow	---
1. 1.747	1.752	<u>1.762</u>	1.772	.020	EULITE (Orthopyroxene ser., Pyroxene grp) (Fe,Mg) ₂ Si ₁₀ O ₂₆	87°	$Y = \frac{a}{c}$ $Z = \frac{c}{d}$	210 good	Dark brown	H 5-6 G 3.84 F 4
1. 1.756	1.742	<u>1.765</u>	1.799	.101	RICHELSDORFITE Ca ₂ Cu ₅ Sb(AsO ₄) ₄ Cl	69° r > v	$Y = \frac{a}{b}$ $Z = \frac{b}{d}$	MCL 001 very good	Sky-blue to turquois	G 3.20
1. 1.727	1.750	<u>1.765</u>	1.767	.025	UREYITE (Pyroxene grp) NaCrSi ₂ O ₆	34°	$X = \frac{b}{c}$ $Y = \frac{b}{d}$	MCL 110 good	Emerald-green	---
1. 1.802	1.705	<u>1.767</u>	1.779	.029	TEPHROITE (Olivine grp) (Mn,Mg) ₂ SiO ₄	67° (83+8°) r > v	$Y = \frac{b}{c}$	ORTH 001	Brown, gray	H 6 G 3.87 F 3.5
1. 1.736	1.767	<u>1.796</u>	1.769	.064	SIMPLOTITE CaV ₄ ₉ ·5H ₂ O	25° r > v wk	$X = \frac{b}{c}$ $Z:c = -58^{\circ}$	MCL 010 perf ps tet	Dark green	H 1 G 2.64
					ARTHURITE CuFe ₂ (AsO ₄) ₂ ·PO ₄ ·SO ₄) ₂ (O,OH) ₂ ·4H ₂ O	~ 90°	$Y = \frac{b}{c}$ $Z:c = 10^{\circ}$	MCL ---	Apple green	G 3.2

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_X$ ($2V_X$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ							
1.756 Δ 1.786	1.741 <u>1.767</u>	1.789	.048 ACMITE-AUGITE (Pyroxene grp) (Na,Ca)(Fe ⁺³ ,Mg)Si ₂ O ₆	85° $r > v$	$y = b$ $x:c = 12^\circ$ $e:l$ neg	MCL	110 good at 87°	Green	H 6 G 3.56	Insol in acids. Pleoc, X yellow green, Y brown-green, abs X > Y = Z. Fe ₂ O ₃ 19.1, FeO 5.2, MnO 0.5%.
1.729	<u>1.767</u>	1.772	.043 SHUISKITE (Pumpellyite grp) Ca ₂ (Mg,Al)(Cr,Al) (SiO ₄)(Si ₂ O ₇)(OH) ₂ H ₂ O	40-50° $r < v$ str	e:l pos	MCL fib	001 perf	Dark brown	H 6 G 3.24	Pleoc, X violet blue, Y yellow green, Z dark violet. Cr ₂ O ₃ 19.3, Fe ₂ O ₃ 1.7%.
1.761 <u>(1.767)</u>	1.772	.011 TADZHIKITE Ca ₃ (Ce,Y) ₂ (Ti,Al,Fe)B ₄ Si ₄ O ₂₂	88°	---	MCL	100 good	Grayish-brown	H 6 G 3.73	---	
288 1.754 Δ 1.784	1.740 <u>1.768</u>	1.787	.047 EPIDOTE (Epidote grp) Ca ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	74° $r > v$ str	$y = b$ $x:c = 13^\circ$ $e:l$ clv pos	MCL	001 perf	Green	H 7 G 3.48 F 4	Insol in acids. Pleoc wk. Fe ₂ O ₃ 17.9, FeO 0.4, MnO 0.5%.
1.734 Δ 1.810	1.742 <u>1.768</u>	1.773	.031 ADAMITE (Olivenite grp) (Zn,Cu) ₂ (AsO ₄)(OH)	47° $r > v$ str	$x = a$ $z = \bar{b}$	ORTH	101 good	Sea-green	H 3.5 G 4.32	Diss by acids. Pleoc, X blue-green, Z yellow-green. ZnO 31.9, CuO 23.4, CoO 0.5%.
1.761	<u>1.768</u>	1.769	.008 CORUNDUM Al ₂ O ₃	17°	$x = c$	HEX	Parting 0001	Blue, gray, brown	H 9 G 4.02 infus	Anom biax. Tw common 1011. Pleoc wk in blue.
1.730 Δ 1.831	<u>1.769</u>	1.771	.041 CONICHALCITE CaCuAsO ₄ (OH)	25° $r > v$ very str	$z = e:l$	ORTH	011 fr uneven dist	Pista- chito-to emerald- green	H 4.5 G 3.96 F 3	Diss by acids. Pleoc, X yellow, Y pale green, Z pale blue-green.
1.760	<u>1.769</u>	1.769	.009 ESPERITE (Ca,Pb)ZnSiO ₄	5-40°	---	MCL	010, 100 dist	White	H 5 G 4.28	Gel with acids. Fluor bright yellow in short-wave UV.
1.744	1.682	<u>1.770</u>	.13 MURMANITE Na ₂ (Ti,Nb) ₂ Si ₂ O ₉ ·xH ₂ O	57-64° $r > v$	$x:a$ small $z = b$ $e:l$ pos	TCL	100 perf	Pink to yellow- brown	H 2 G 2.8	Pleoc, X light brown, Y brown to pink, Z dark brown.

1.72	<u>1.77</u>	1.81	.09	ZINC-ZIPPELITE $\text{Zn}_2(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 16\text{H}_2\text{O}$	Large disp str	$X = \frac{c}{d}$ $Z = \frac{d}{a}$	ORTH plates	010 perf	Yellow	H 2	Diss by acids. Pleoc, X cols, Y pale yellow, Z yellow. Fluor yellow in UV. Data for synth compd.		
1.720	<u>1.770</u>	1.800	.080	MELONJOSEPHITE $\text{CaFe}^{+2}\text{Fe}^{+3}(\text{PO}_4)_2(\text{OH})$	80-85°	$X = \frac{c}{d}$ $Z = \frac{d}{a}$	ORTH fib	el	Dark green	H < 5 G 3.65	Diss by HCl. Pleoc, X deep brown, Y greenish-brown, Z yellow-gold.		
1.750 1.786 1.786	<u>1.770</u>	1.786	.044	HORTONOLITE (011vine grp) $(\text{Fe}, \text{Mg})_2\text{SiO}_4$	69°	$X = \frac{b}{d}$ $Z = \frac{a}{d}$	ORTH	010 fair	Green	H 5 G 3.9 F 5	Gel with acids (FeO_{43}).		
1.734	<u>1.770</u>	1.773	.039	VINOGRADOVITE $(\text{Na}, \text{Ca}, \text{K}_4)\text{Ti}_4\text{AlSi}_6\text{O}_{23}(\text{OH}) \cdot 2\text{H}_2\text{O}$	41° $r > v$	$Z:c = 7^\circ$	MCL	010 perf	Col s	H 4 G 2.88	Gel with acids.		
1.728	<u>1.771</u>	1.800	.072	BROCHANTITE $\text{Cu}_4\text{SiO}_4(\text{OH})_6$	75° $r < v$ med	$Y = \frac{b}{d}$ $X \sim \frac{a}{c} v$ el ct v pos	MCL pris	100 perf	Emerald-green	H 3.5-4 G 3.97 F 3.5	Diss by acids or NH_4OH . Tw 100 common. Pleoc in blue-greens.		
1.727	<u>1.771</u>	1.798	.071	MARGAROSANITE $\text{Pb}(\text{Ca}, \text{Mn})_2\text{Si}_3\text{O}_9$	78° $r < v$	$X':010 = 42^\circ$	TCL lam	010 perf 100 good 001 fair	Col s	H 2.5-3 G 4.33	Dec by HNO_3 with sepn of silica. Fluor pink or blue in UV. F 2 in reducing flame.		
1.726	<u>1.771</u>	1.780	.054	PARADAMITE $\text{Zn}_2\text{AsO}_4(\text{OH})$	50°	---	TCL	010 perf	Pale yellow	G 4.55 F 3	Diss by acids.		
1.751	<u>1.771</u>	1.782	.031	LEUCOPHOENICITE $\text{Mn}_7(\text{SiO}_4)_3(\text{OH})_2$	74° $r > v$ wk	$X \perp c v$	MCL ps orth	---	Light to deep pink	H 5.5 G 3.85 F 3	Gel with acids. Poly tw on 001.		
1.750 1.805 1.779	<u>1.771</u>	1.776	.026	FERRISICKLERITE $\text{Li}(\text{Fe}^{+3}, \text{Mn}^{+2})\text{PO}_4$	52° $r > v$	$X = \frac{a}{d}$ el ct v pos	ORTH	100 good 010	Dark brown	H 4.5 G 3.30 F easy	Diss by acids.		
1.735 1.775	<u>1.772</u>	1.774	.019	ALLACTITE $\text{Mn}_7(\text{AsO}_4)_2(\text{OH})_8$	Small $r > v$ str	$Y = \frac{b}{d}$ $X:c = 51^\circ$	MCL pris	001 dist	Purplish-red	H 6.5 G 3.83 F diff	Diss by HCl. Pleoc, X blood red, Y pale yellow, Z sea-green.		
1.772	<u>1.775</u>	1.777	.005	BRITHOLITE $(\text{Ce}, \text{Ca})_5(\text{SiO}_4, \text{PO}_4)_3(\text{OH}, \text{F})$	0-44° $r > v$	$X = \frac{c}{d}$ $Y = \frac{a}{d}$	HEX pris	---	Brown	H 5-5.5 G 4.0-4.5	Apatite-type structure.		
1.679	<u>1.776</u>	1.807	.128	SAHANALITE $(\text{Mg}, \text{Fe})\text{Ce}_2(\text{CO}_3)_4$	57° $r < v$	$Y = \frac{b}{d}$ $Z:c = 29^\circ$	MCL tab	010 poor	Col s	G 4.30 infus	Diss by concd HCl with eff.		
1.750 1.803	<u>1.776</u>	1.793	.047	PIEMONTITE (Epitote grp) $\text{Ca}_2(\text{Al}, \text{Mn}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	87° $(73+5^\circ)$ $r > v$ str	$Y = \frac{b}{d}$ el b	MCL el b	001 perf	Purplish-red	H 4.5 G 3.44 F 3	Pleoc, X yellow, Y pink, Z deep pink. Fe_2O_3 11.9, Mn_2O_3 6.8%.		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.745	<u>1.777</u>	1.84	.095	NICKEL-ZIPPEITE $\text{Ni}(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 16\text{H}_2\text{O}$	Large	$Z = \underline{c}$	ORTH plates	010 perf	Yellow to orange	H 2	Diss by acids. Pleoc., X col., Y pale yellow, Z yellow. Fluor yellow in UV. Data for synth compd.	
1.740	<u>1.777</u>	1.780	.040	BEIYANKINITE $\text{Ca}(\text{Ti}, \text{Zr}, \text{Nb})_5\text{O}_{12} \cdot 9\text{H}_2\text{O}$ (?)	23°	$Y = \frac{b}{c}$ $X = \frac{a}{c}$	ORTH	One perf		H 2.3 G 2.36 infus	Dec by warm acids. Pleoc., X dark brown, Y and Z light brown.	
1.756	<u>1.777</u>	1.794	.038	ORIENTITE $\text{Ca}_2\text{Mn}^{+2}\text{Mn}^{+3}\text{Si}_3\text{O}_{10} \cdot (\text{OH})_4$	80° $r < v$ str	$X = \frac{b}{c}$ $Y = \frac{a}{c}$	ORTH	001, 120 imperf	Deep brown	G 3.05	Dec by HCl. Pleoc., X red-brown, Y yellow, Z yellow-brown.	
1.763	<u>1.777</u>	1.785	.022	YOSHIMURITE $(\text{Ba}, \text{Sr})_2\text{TiMn}_2(\text{SiO}_4)_2 \cdot (\text{PO}_4, \text{SO}_4)(\text{OH}, \text{Cl})$	90°-95° $r < v$	---	TCL	010 perf 101, 101 dist	Brown	H 4.5 G 4.13	Poly tw on 010. Pleoc., X bright yellow, Y orange-brown, Z brown, abs Z > Y > X. Reported as biax pos.	
1.747	<u>1.779</u>	1.84	.09	COBALT-ZIPPEITE $\text{Co}_2(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 16\text{H}_2\text{O}$	Large	$Z = \underline{c}$	ORTH plates	010 perf	Yellow, orange	H 2	Diss by acids. Pleoc., X col., Y pale yellow, Z yellow. Fluor yellow in UV. Data for synth compd.	
1.716	1.763	<u>1.779</u>	1.793	SONOLITE $\text{Mn}_9(\text{SiO}_4)_4(\text{OH}, \text{F})_2$	78° $r > v$	$X : \underline{c} = 9^\circ$	MCL pris	---	Reddish-brown	H 5.5 G 3.82	Gel with acids. Tw on 001 common.	
1.67		<u>1.78</u>	1.78	.11	NULLAGINITE $\text{Ni}_2(\text{OH})_2\text{CO}_3$	$X : \underline{c} = 6^\circ$	MCL fib	---	Bright green	G 3.56	---	
1.832		<u>1.725</u>	1.780	1.790	.065	BILLIETITE $\text{BaU}_6\text{O}_9 \cdot 11\text{H}_2\text{O}$	$X = \frac{c}{a}$ el ctw pos	001 perf	Golden-yellow	G 5.32	Tw on 111 and 110 common. Pleoc., X pale yellow, Y and Z deep golden-yellow.	
1.680		<u>1.756</u>	<u>1.780</u>	1.792	.036	ALLEGHANYITE $\text{Mn}_5(\text{SiO}_4)_2(\text{OH})_2$	$Z = \frac{b}{a}$ Ext: \overline{tw} pl = 22°	MCL	Conch	Pink to reddish-brown	H 5.5 G 4.02 F 2.5	Gel with acids. Poly tw on 001 common.
1.767		<u>1.782</u>	1.791	.024	TEINEITE $\text{CuTeO}_3 \cdot 2\text{H}_2\text{O}$	36° ($75 \pm 10^\circ$)	$X = \frac{a}{b}$ $Y = \frac{a}{b}$	ORTH	010 good 001, 100 poor	Blue	H 2.5 G 3.80 F 2	Pleoc., X greenish-blue, Y blue, Z indigo blue, abs Z > Y > X.

1.720	<u>1.783</u>	1.842	.122	ROSSITE Ca ₂ V ₆ O ₄ H ₂ O	85°	TCL pris	010 good	Yellow	H 2.5 G 2.45 F easy						
8	1.751	<u>1.784</u>	1.797	.046	EPIDOTE (Epidote grp) Ca ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	64° r > v str	MCL	001 perf	Green	H 7 G 3.48 F 4	Insol in acids. Fe ₂ O ₃ 17.2, FeO 0.5%.				
	1.765	<u>1.784</u>	1.799	.034	MALAYAITE CaSrSiO ₅	85°	X = b X ~ c el cIV pos	---	Pale yellow	H 4 G 4.55	Fluor yellow-green in short-wave UV.				
5	1.769	<u>1.784</u>	1.791	.022	ALLANITE (Epidote grp) (Ce,Ca,Y) ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	60° r > v str	MCL	001, 100, 110 poor	Brownish-black	H 6 G 3.87 F 3	Slowly attacked by HCl, gel. Pleoc, X coils to red-brown, Y yellow- brown to pale green, Z green-brown to dark red-brown.				
	1.77	<u>1.785</u>	1.815	.045	BARBOSALITE Fe ⁺² Fe ⁺³ ₂ (PO ₄) ₂ (OH) ₂	64° r > v str	MCL ps tet tab	100	Dark blue- green	H 2 G 3.65	Slowly diss by hot HCl. Pleoc, X and Y dark blue-green, Z dark olive-green.				
	1.755	<u>1.785</u>	1.785	.030	THEISITE Cu ₂ Zn ₅ (As, Sb) ₂ O ₈ (OH) ₁₄	~ 0°	ORTH ps hex	001	Pale blue- green	H 1.5 G 4.3	Not pleoc.				
7	1.751	<u>1.786</u>	1.800	.049	ACMITE (Pyroxene grp) (Na,Ca)(Fe ⁺³ ,Mg,Fe ⁺²) Si ₂ O ₆	69° r > v	Y = b X:C = 1 el neg	MCL	110 good at 87°	Green	H 6 G 3.55	Pleoc, X and Y deep green, Z yellow-brown. Fe ₂ O ₃ 25.4, FeO 3.7%.			
2	1.758	<u>1.786</u>	1.804	.046	HORTONOLITE manganian, zincian (Olivine grp) (Fe,Mn,Mg,Zn) ₂ Si ₁₀ O ₄	77° r > v	ORTH	010 fair	Reddish-brown	H 6 G 4.1 F 3.5	Gel with acids. Pleoc wk. FeO 35.5, MnO 16.9, ZnO 10.7, MgO 5.6%.				
0	1.779	<u>1.786</u>	1.790	.011	KOLICITE Mn ₇ Zn ₄ (AsO ₄) ₂ (SiO ₄) ₂ (OH) ₈	78° r < v str	Y = c Z = a	ORTH	---	Yellow-orange	H 4.5 G 4.17	Pleoc, X coils to pale yellow, Y yellow- orange, Z light yellow, abs V = Z > X.			
0	1.749	<u>1.790</u>	1.821	.072	REINERITE Zn ₃ (AsO ₃) ₂	(80±3°)	---	ORTH	110 good 011, 111	Blue to greenish-yellow	H 5-5.5 G 4.27	Diss by cold HCl.			
	1.70	<u>1.79</u>	1.793	.09	EPI-IANTHINITE UO ₃ •xH ₂ O (?)	Small	X = c Y = $\frac{c}{D}$	ORTH	---	Yellow	---	Pleoc, X pale yellow, Y yellow, Z deep yellow. Stated to be Schoelite, optics are different.			
8	1.776	<u>1.79</u>	1.820	.044	JULGOLDITE (Pumpellyite grp) Ca ₂ (Fe,Mg)(Fe,Al) ₂ (SiO ₄)(Si ₂ O ₇)(OH) ₂ •H ₂ O	80°	MCL	100, 001 good	Greenish-black	H 4.5 G 3.4	Gel with hot HCl. Pleoc, X and Y pale greenish-yellow, Z olive-green. Tw 001.				

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ										
1.786 1.818	1.761 1.790	1.790 1.790	1.806 1.796	.045 .024	HORTONOLITE (01ivine grp) (Fe,Mg) ₂ SiO ₄	65° (72+5°) $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{c} \text{Tr pos}$	ORTH	010 fair	Green to brown	H 5 G 3.90 F 5	Gel with acids (Fo ₃₃).	
1.835	1.772	1.790	1.796	.024	PSEUDOMALACHITE Cu ₅ (PO ₄) ₂ (OH) ₄ •H ₂ O	50°	X:c = 21°	MCL pris	010 good	Dark green	H 4.5-5 G 4.35 F 2	Diss by acids. Tw 100. Pleoc wk.	
1.765 1.794 1.798	1.792 1.793 1.795	(1.82) (1.82)	(1.82) 1.809 1.800	(.055) .053 .048	VANDENBRANDEITE Cu(UO ₂)(OH) ₄	~ 90°	---	TCL tab	110 perf 2 others	Dark green	H 4 G 5.0	Diss by acids. Not fluor in UV.	
					THORTVEITITE (Sc,Y) ₂ Si ₂ O ₇	50-70° $r > v$ dist	Y = b	MCL pris	110 good	Gray-green	H 6.5 G 3.58 F diff	Insol in acids. Poly tw.	
					PUMPELLYITE-(Mn) (Pumpeleyite grp)	(37+8°)	---	MCL	001 perf	Pink to brownish-pink	H ~ 5 G 3.34	Mn ₂ O ₃ 77, Fe ₂ O ₃ 24, MnO 13%. Pleoc, X pale pink, Y and Z brownish-pink.	
					Ca ₂ (Mn ⁺² ,Mg)(Al,Mn ⁺³) ₂ (SiO ₄)(Si ₂ O ₇)(OH) ₂ •H ₂ O	(38+6°)	---	MCL	001 perf	Red-brown, maroon	H > 5 G 3.43		
					MACFALLITE Ca ₂ (Mn ⁺³ ,Al) ₃ (SiO ₄) ₃ (Si ₂ O ₇)(OH) ₃	Z = b	---	ORTH pris	---	Bright red	G 3.66		
					JAROSEWICHITE Mn ⁺² •3Mn ⁺³ (AsO ₄) ₆	(78+9°)	X = $\frac{a}{c}$ Y = $\frac{a}{b}$	---	MCL (?) spherulitic	---	Blue	H 1 seetile	Streak reddish-orange. Pleoc, X med brownish-red, Z dark brownish-red.
					TLALOCITE (Cu,Zn) ₁₆ (TeO ₃) ₂ Cl(OH) ₂₅ •27H ₂ O	64°	---	---	---	Blue	H 1 seetile	Diss by HNO ₃ . Pleoc, X yellow-green, Y and Z blue-green, abs Z > Y > X.	
					BRIEGGENITE Ca(I0 ₃) ₂ •2H ₂ O	88° (79+6°) $r > v$ mod to str	Z = $\frac{b}{a}$ X:a = g° el neg	MCL	Conch	Cols to bright yellow	H 3.5 G 4.24	Slightly sol in H ₂ O.	
					GUILLEMINTITE Ba(UO ₂) ₃ (SeO ₄) ₂ (OH) ₄ •3H ₂ O	35° $r > v$ str	X = c Y = $\frac{b}{a}$ el cTr neg	ORTH	100 perf 010 good	Canary-yellow	G 4.88	Pleoc str, X bright yellow, Y yellow, Z cols.	
1.720	1.798	1.805	.085										

1.775	<u>1.798</u>	1.800	.025	BOSTWICKITE CaMn ⁺³ Si ₃ O ₁₆ ·7H ₂ O	25°	r < v str	---	---	Dark red	H 1 G 2.93
<1.79	<u>1.798</u>	1.802	>.01	COMPREIGNACITE K ₂ U ₆ ·19·11H ₂ O	10-15°	X = $\frac{c}{a}$ el cTw pos	001 perf	Yellow	G 5.03	
---	<u>1.80</u>	---	Large	CRONSTEDTITE (Fe ⁺² ,Mg) ₂ Fe ⁺³ (SiFe ⁺³) _{5(OH)} ₄	~ 0°	---	MCL ps hex	Blackish-brown	H 3.5 G 3.34 F 4	
1.74	<u>1.80</u>	1.85	.11	SABINAITE Na ₉ Zr ₄ Ti ₂ g(CO ₃) ₈	85°	X ~ <u>—</u> plates el pos	001 perf	White	G 3.31	
1.780	---	1.810	.030	Unnamed manganese silicate (Mn,Fe ⁺²) ₃ Si ₂ 7	60°	---	MCL ps hex plates el pos	Rose-red	---	
1.787	<u>1.800</u>	1.805	.018	KIDWELLITE NaFe ⁺³ g(PO ₄) ₆ (OH) ₁₀ · 5H ₂ O	Large	Y = <u>b</u> —	100 perf	Pale green to yellow	H 3 G > 3.3	
1.715	<u>1.801</u>	1.87	.155	MANDARINOITE Fe ⁺³ Se ₃ O ₉ ·4H ₂ O	85°	X = <u>b</u> Z:c = 2° el pos	MCL pris c	---	Light green	
1.766	<u>1.782</u>	1.812	.030	TEPHROITE (Olivine grp) Mn ₂ SiO ₄	65°	X = <u>b</u> Z = <u>a</u> —	010, 001 good	Reddish-brown	H 6 G 4.03	
1.776	<u>1.768</u>	1.803	.075	PIEMONITITE (Epidotite grp) Ca ₂ (Al,Mn,Fe) ₃ Si ₃ O ₁₂ (OH)	68°	Y = <u>b</u> —	MCL	001 perf	Purplish-red	H 6.5 G 3.52 F 3
1.775	<u>1.803</u>	1.812	.037	GINNITE Fe ⁺² Fe ⁺³ 4(Po ₄) ₄ (OH) ₂ · 2H ₂ O	55°	X = <u>c</u> Y = <u>b</u> —	001 perf	Conch	Blackish-green to blackish-brown	H 3-4 G 3.41
1.793	<u>1.803</u>	1.808	.015	PERLOFFITE Ba(Mn,Fe ⁺²) ₂ Fe ⁺³ (Po ₄) ₃ (OH) ₃	70-80°	X = <u>b</u> Y:c = 42°	MCL spear-shaped	100 perf	Dark brown	H 5 G 4.00
(1.563)	<u>1.805</u>	(1.851)	(.288)	TYUYAMUNITE Ca(UO ₂) ₂ (VO ₄) ₂ ·5-8H ₂ O	42°	X = <u>c</u> Y = <u>a</u> —	001 perf 010, good	Canary-yellow	H 2 G 3.6 F easy	
1.835	<u>1.805</u>	1.820	.090	BECCQUERELITE CaU ₆ ·19·11H ₂ O	30°	X = <u>c</u> el cTw pos	001 perf	Golden-yellow	H 2.5 G 5.3	
1.730	<u>1.805</u>	1.820	.090	(47+3°)	r > v					
1.825	<u>1.835</u>	1.825								

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.771	1.790	<u>1.805</u>	1.820	.025	FERRISICKLERITE $\text{Li}(\text{Fe}^{+3}, \text{Mn}^{+2})\text{PO}_4$	Very large	$X = \frac{c}{a}$ $Y = \frac{c}{a}$	ORTH	010 perf 100 good	Dark brown	H 4.5 G 3.41 F easy	Diss by acids. Pleoc wk, X orange-yellow, Z gold-yellow.
---	---	<u>1.80-</u> <u>1.81</u>	---	---	MUSKOXITE $\text{Mg}_7\text{Fe}^{+3}_4\text{Al}_{13}\cdot 10\text{H}_2\text{O}$	10-40°	el pos	TRIG platy 0001	0001 perf	Dark reddish- brown	H 3 G 3.15	Not pleoc.
1.793	1.793	<u>1.807</u>	1.809	.016	SARKINITE $\text{Mn}_2(\text{AsO}_4)_2(\text{OH})$	83° (41-18°)	$Y = \frac{b}{c}$ $X:c = -54^\circ$	MCL tab	100 good	Flesh-red to yellow	H 3 G 3.15	---
1.768	1.772	<u>1.810</u>	1.863	.091	OLIVENITE $\text{Cu}_2(\text{AsO}_4)_2(\text{OH})$	76° (97-3°) $r > v$ str	$X = \frac{a}{b}$ $Z = \frac{a}{b}$	ORTH	101 fr conch	Olive- green	H 3 G 4.4 F easy	Diss by acids or NH_4OH . Pleoc, X pale green, Y bright green, Z pale green.
1.74	1.74	<u>1.81</u>	1.81	.07	GERASIMOVSKITE $(\text{Mn}, \text{Ca})(\text{Nb}, \text{Ti})_5\text{S}_2\text{H}_2\text{O}$ (?)	18°	el pos	MCL (?) platy	One cly	Brown, gray	H 2 G 2.55	---
1.785	1.785	<u>1.810</u>	1.820	.035	VANDENDRIESCHEITE $\text{PbU}_0.22\text{LiH}_2\text{O}$	Med $r > v$ str	$X = \frac{c}{a}$ $Z = \frac{c}{a}$	ORTH tab	001 perf	Orange to yellow- brown	H 3 G 4.5-5.5	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
~1.86	1.787	1.810	1.816	.029	ARSENOCLASITE $\text{Mn}_5(\text{AsO}_4)_2(\text{OH})_4$	53°	$X = \frac{b}{c}$ $Z = \frac{c}{a}$	ORTH u mass el cly pos	010 perf	Red	H 5.5 G 4.16	---
~1.795	1.795	---	1.815	.020	PARSONSITE $\text{Pb}_2(\text{UO}_2)(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	---	el cly pos	TCL	010 poor fr conch	Pale lemon- yellow	H 2.5-3 G 5.73	Not fluor in UV. Opt char unk.
1.786	1.770	<u>1.812</u>	1.830	.060	ACMITE (Pyroxene grp) $(\text{Na}, \text{Ca})(\text{Fe}^{+3}, \text{Mg})\text{Si}_2\text{O}_6$	58° (65-4°) $r > v$	$Y = \frac{b}{c}$ $Z:c = 6^\circ$ el neg	MCL	110 good at 87°	Green	H 6 G 3.56	Insol in HCl. Pleoc, X pale yellow-green, Y yellow-green, Z emer- ald-green. Fe_2O_3 31.4, Al_2O_3 1.9, FeO 0.8%.
1.835	(1.786)	<u>1.813</u>	1.852	(.066)	BAERTSITITE $\text{Ba}(\text{Fe}, \text{Mn})_2\text{TiSi}_2\text{O}_7$ (0, OH) ₂	80°	$Y = \frac{b}{c}$ $Z:c = 6^\circ$ el cly neg	MCL	001 perf	Orange to red	H 5 G 3.97- 4.25	Pleoc, X yellowish-red, Z pale yellow. FeO 10.8, MnO 12.8, Fe_2O_3 3.7%.

1.79	<u>1.814</u>	1.836	.060	JULGOLDITE (Pumpellyite grp) Ca ₂ (Fe,Mg)(Fe,Al) ₂ (SiO ₄)(Si ₂ O ₇)(OH) ₂ ·H ₂ O	50-70°	---	MCL	100, good	Greenish-black	H 4.5 G 3.60
1.775	<u>1.815</u>	1.825	.050	PASCOITE Ca ₃ V ₁₀ O ₂₈ ·17H ₂ O	50° r > v very str	X = b Y = b X:c = 20°	MCL	010 poor	Dark red-orange	H 2.5 G 2.46 F easy
1.784	<u>1.815</u> 1.857	1.822	.031	ALLANITE (Epidote grp) (Ce,Ca,Y) ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	40° (56+8°) r > v str	---	MCL	001, 100, 110 poor	Brownish-black	H 6 G 3.95 F 3
1.791	<u>1.817</u>	1.820	.105	JAROSITE (Alunite grp) KFe ₃ (SO ₄) ₂ (OH) ₆	Small	---	TRIG	0001 good	Amber yellow	H 3 G 3.2
1.790	<u>1.818</u> 1.828	1.828	.051	FERRDORTONOLITE, mangananoan (Olivine grp) (Fe,Mn) ₂ SiO ₄	52° r > v	X = b Z = a	ORTH	010 fair	Reddish-brown	H 6 G 4.21 F 3.5
1.715	<u>1.820</u>	1.820	.050	MITRIDATITE Ca ₃ Fe ⁺³ ·4(Po ₄) ₄ (OH) ₆ · 3H ₂ O	Small	X ~ l 100	MCL	---	Brownish-green	H 3.24
1.777	<u>1.822</u>	1.820	.20	IRIGINITE (UO ₂)Mo ₂ O ₇ ·3H ₂ O	Large	Y = b Z = c	MCL	---	Canary yellow	H 4-5 G 3.94
1.770	<u>1.822</u>	1.820	.165	DOLEROPHANITE Cu ₂ (SO ₄) ₀	105° (?) (70+2°) r > v very str, crossed	Y = b Z:c = -10°	MCL	T01 perf	Chestnut-to dark brown	H 3 G 4.17 fus
1.85	<u>1.820</u>	1.820	.165	CORNETITE Cu ₃ (PO ₄)(OH) ₃	35° r < v str	X = a Z = b	ORTH	---	Light to dark blue	H 5 G 4.10
1.889	<u>1.820</u>	1.880	.090	CALCURNOLITE Ca(UO ₂) ₃ (MoO ₄) ₃ (OH) ₂ · 11H ₂ O	(80+3°)	---	ORTH pris	---	Honey-yellow	---
1.762	<u>1.820</u>	1.825	.063	LIMORITE Y ₂ (SiO ₄)(CO ₃)	31°	---	TCL	011 good	Gray to tan	H 5.5-6 G 4.47
1.770	<u>1.821</u>	1.860	.090	BECQUERELITE CaU ₆ O ₁₉ ·11H ₂ O	30° r > v	X = c Y = b el cTv pos	ORTH pris hex	001 perf	Golden-to lemon-yellow	H 2.5 G 5.1
1.753	<u>1.824</u>	1.830	.077	---		---		---		
1.730	<u>1.825</u>	1.830	.100	---		---		---		
1.805	<u>1.825</u>	1.830	.100	---		---		---		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral Name and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks	
	α	β	γ									
1.81	1.788	<u>1.825</u>	1.830	.042	HANCOCKITE (Epidote grp) (Pb,Ca,Sr) ₂ (Al,Fe) ₃ Si ₃ 012.(OH)	38° $r > v$	$\gamma = b$	MCL	001	Brownish-red	H 6.5 G 4.03 F 3	
1.818 1.843	1.788	<u>1.828</u>	1.840	.052	FERROHORTONOLITE (Olivine grp) (Fe,Mg)SiO ₄	58° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010 fair	Reddish-brown	H 6 G 4.15 F 3.5	
1.672	<u>1.83</u>	1.83	.16	ROSASITE (Cu,Zn) ₂ (CO ₃)(OH) ₂	Small $r < v$ str	$X = e1$	MCL	---	Bluish-green	G 4.09 fus	Diss by acids with eff. Pleoc., X cols, Y and Z blue.	
1.85	1.69	<u>1.83</u>	1.83	.14	GLAUKOSPHAERITE (Cu,Ni) ₂ (CO ₃)(OH) ₂	Small $X:C = 7^\circ$	MCL fib-spherulitic	001 good	Green	H 3-4 G 3.78-3.96	Diss by acids with eff. Pleoc., X deep green, Y and Z apple-green.	
1.66	<u>1.831</u>	1.915	.255	UMOHOHITE (UO ₂)MoO ₄ ·4H ₂ O	65° $r > v$ str	$\gamma = b$ $X \sim \frac{a}{d}$	MCL tab	---	Blue-black	H 2 G 4.6	Pleoc., X dark blue, Y light blue, Z olive-green.	
(1.769)	1.800	<u>1.831</u>	1.846	.046	CONICHALCITE CaCuAsO ₄ (OH)	Large $r > v$ str	$X = \frac{a}{D}$ $Y = \frac{a}{D}$	ORTH pris	011 fr uneven	Dark-to-yellow-green	H 4-5 G 4.33 F 3	Diss by acids. Pleoc., X green, Y yellow-green, Z bluish-green.
1.780	1.733	<u>1.832</u>	1.839	.106	BILLIETITE BaU ₆ O ₁₉ ·11H ₂ O	35° $(28-4^\circ)$ $r > v$ str	$X = \frac{c}{a}$ $Y = \frac{c}{a}$ $e1 cTv$	ORTH ps hex pos	001 perf	Golden-yellow	G 5.32	Tw common on 111 and 110. Pleoc., X and Z pale green, Y bright green.
1.815	1.750	<u>1.832</u>	1.832	.082	NATROJAROSITE (Alunite grp) NaFe ₃ (SO ₄) ₂ (OH) ₆	Very small	$X = \frac{c}{a}$ $Y = \frac{c}{a}$ $e1 cTv$	TRIG ps orth pos	0001 perf	Yellow-brown	H 3 G 3.2 F 3	Diss by HCl. Pleoc., X nearly cols, Y and Z pale yellow.
1.783	<u>1.834</u>	1.866	.083	W	RUSAKOVITE (Fe,Al) ₅ (VO ₄ ,PO ₄) ₂ (OH) ₉ ·3H ₂ O	---	Crypto-cryst	---	Yellow-orange	H 1.5-2 G 2.76	Diss by acids. Opt char unk.	
					DELRIOTITE SrCaW ₂ O ₆ (OH) ₂ ·3H ₂ O	Med to large	MCL acic	---	Yellow-green	H 2 G 3.1	Diss by H ₂ O. Tw common 100. Pleoc., X cols, Y pale yellow, Z deeper yellow.	

1.805	(1.664)	<u>1.835</u>	1.865	(.201)	TYUYAMUNITE Ca ₂ (UO ₂) ₂ (VO ₄) ₂ •5-8H ₂ O	45° r < v wk	X = $\frac{c}{a}$ Y = $\frac{b}{a}$	001 perf 010, 100 good	Canary- greenish- yellow	H 2 G 3.85	Pleoc wk, X col's, Y and Z greenish-yellow. Not fluor in UV.
1.813	1.805	<u>1.835</u>	1.862	.057	BAFERTISITE Ba(Fe,Mn) ₂ TiSi ₂ 7 (0,OH) ₂	86°	Y = $\frac{b}{a}$ Z:c = 6° el clv neg	MCL	001 perf Orange to red	H 5 G 3.97- 4.25	Poly tw 001. Pleoc, X reddish-brown, Y yellow, Z greenish- yellow.
1.790	1.789	<u>1.835</u>	1.845	.056	PSEUDOMALACHITE Cu ₅ (PO ₄) ₂ (OH) ₄ •H ₂ O	50° r < v str	X:c = 21°	MCL pris	010 good Dark green	H 4.5-5 G 4.35 F 2	Diss by acids. Tw 100. Pleoc wk.
1.86	1.809	<u>1.838</u>	1.859	.050	LIMARITE PbCu(SO ₄)(OH) ₂	80° r < v str	Z = $\frac{b}{a}$ X:c = -24° el	MCL el 010	100 perf Deep azur- blue	H 2.5 G 5.35 F easy	Diss by HNO ₃ . Tw com- mon 100, less so 001. Pleoc, X pale blue, Y clear blue, Z Prussian blue.
1.81	---	1.84	.03	DEERITE (Fe ⁺² ,Mn) ₆ (Fe ⁺³ ,Al) ₃ Si ₆ 0 ₂₀ (OH) ₅	---	Z = $\frac{c}{a}$ el pos	MCL or ORTH acic	110 good Black	G 3.84	Opt char unk.	
1.80	---	1.87	.07	KEYTTE (Cu,Zn,Cd) ₃ (AsO ₄) ₂	Disp str	X:c = 10° Y = $\frac{b}{a}$	MCL	001 good Deep sky- blue	---		Diss by conc acids. Pleoc, X pale blue, Y greenish blue, Z deep blue. Opt char unk.
1.760	1.840	1.870	.110	CALCIOURANOITE (Ca,Ba,Pb)U ₂ O ₇ • 5H ₂ O	(61+2°)	X = $\frac{c}{a}$ el neg	ORTH(?)	---	Brown to orange	H 4 G 4.02- 4.28	Pleoc, X and Y lemon- yellow, Z col's.
1.775	1.840	1.844	.069	CHALCOSTIDERITE CuFe ₆ (PO ₄) ₄ (OH) ₈ •4H ₂ O	22° r > v str crossed	---	TCL pris	001 perf 010 good	Siskin- green	H 4.5 G 3.22	Fe-analogue of Turquoise. Pleoc wk.
1.795	1.842	1.874	.079	STRANSKIITE Zn ₂ Cu(AsO ₄) ₂	80°	---	TCL	010 perf 100 good	Blue	H 4 G 5.23	---
1.825	1.842	1.857	.032	DIEZZEITE Ca ₂ (IO ₃) ₂ (CrO ₄)	86° r < v very str	Y = $\frac{b}{a}$ Z:c = 6°	MCL tab	100 poor fr conch	Golden- yellow	H 3.5 G 3.62	Diss by hot H ₂ O.
1.803	1.843	1.851	.048	FAYALITE, manganese (0)ivine grp (Fe,Mn) ₂ SiO ₄	48° r > v wk	X = $\frac{b}{a}$ Z = $\frac{c}{a}$	ORTH	010 good Reddish- brown	Reddish- brown	H 6 G 4.3 F 3.5	Gel with acids. FeO 56.9, MnO 9.7, MgO 1.1, Fe ₂ O ₃ 1.2%.
1.828	1.850	1.84-	---	ITOITE Pb ₃ Ge(SO ₄) ₂ •2(OH) ₂	---	---	ORTH acic	---	White	G 6.7 calc	Opt char unk.
1.65	1.85	1.90	.25	VANURALITE Al(UO ₂) ₂ (VO ₄) ₂ (OH) ₂ 11H ₂ O	44° r < v wk	Z = $\frac{b}{a}$ X ~ $\frac{c}{a}$	MCL	001 easy Lemon- yellow	H 2 G 3.62	Tw 001. Pleoc str, X col's, Y and Z yellow.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.77	<u>1.85</u>	1.89	.12	DERRIKSITE $\text{Cu}_4(\text{UO}_2)(\text{SeO}_3)_2(\text{OH})_6 \cdot \text{H}_2\text{O}$	(68±19°)	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	010 good	Green	G 4.72	----	
1.740	---	1.855	.115	SINOITE $\text{Si}_2\text{N}_2\text{O}$	Small	$Z = \frac{c}{c}$	ORTH	---	Gray	G 2.82	---	
1.79	<u>1.85</u>	1.89	.10	MROSITE $\text{CaTe}^{+4}(\text{CO}_3)_2$	74° $r < v$ str	$X = \frac{a}{c}$ $Y = \frac{c}{c}$	ORTH	---	Cols to white	H 4 G 4.35	Diss by cold acids with eff.	
1.810 ^	<u>1.850</u>	1.860	.080	VANDENDRIESSCHEITE $\text{PbU}_7\text{O}_{22} \cdot 12\text{H}_2\text{O}$	60° (40±4°) $r > v$ str	$X = \frac{c}{a}$ $Z = \frac{a}{c}$ el cTv pos	ORTH tab	001 perf	Orange-yellow	H 3 G 4.5-5.5	Pleoc., X cols, Y and Z yellow. Not fluor in UV.	
(1.82) ^	<u>1.85</u>	1.85	.065	MIRIDATITE $\text{Ca}_3\text{Fe}^{+3}(\text{PO}_4)_4(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	5-10°	$X \sim \perp 100$	MCL	---	yellow-brown	G 3.24	Pleoc., X pale greenish-yellow, Y and Z deep greenish-brown.	
1.82	<u>1.85</u>	1.85	.03	BURKHARDTITE $\text{Pb}_2(\text{Fe}^{+3}, \text{Mn}^{+3})\text{Te}^{+4}\text{AlSi}_3\text{O}_12(\text{OH})_2 \cdot \text{H}_2\text{O}$	Small	$X = \frac{c}{c}$	MCL ps hex	001 perf	Carmine-to-violet-red	H 2 G 4.96	Insol in acids. Pleoc., X pale magenta, Y and Z carmine-red.	
1.843 ^ 1.854	1.816 1.850	1.863	.047	FAYALITE $(\text{Fe}, \text{Mg})_2\text{SiO}_4$	60° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{a}$	ORTH	010 fair	Dark brown	H 6 G 4.36 F 3.5	Gel with acids (Fo ₇).	
1.847	<u>1.850</u>	1.850	.003	ROMEITE (Stibiconite grp) $(\text{Ca}, \text{Fe}, \text{Mn}, \text{Na})_2(\text{Sb}, \text{Ti})_2_6(0, \text{OH}, \text{F})$	Med	---	CUB oct	Conch	Brown	H 6 G 5.07 infus	Insol in acids. Opt anom. Segmented. Poly tw. Abnormal interf colors.	
1.850 1.854 pos 1.869	1.820 1.854 1.869	1.888	.068	LIEBENBERGITE (01ivine grp) $(\text{Ni}, \text{Mg}, \text{Fe}, \text{Co})_2\text{SiO}_4$	88° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{a}$	ORTH	010 poor to fair 100 poor	Yellowish-green	H 6-6.5 G 4.60 calc	Pleoc., X and Y cols to pale green, Z greenish yellow. Ni 0.56.3, Mg 6.5, Fe 9.4, Co 0.1.8%.	
1.907	1.674	<u>1.855</u>	.206	STRELKINITE $\text{Na}(\text{UO}_2)(\text{VO}_4) \cdot 3\text{H}_2\text{O}$	Med	$X = \frac{c}{b}$ $Y = \frac{b}{c}$ el cTv pos	ORTH plates	001 perf	Gold-to-canary-yellow	H 2-2.5 G 3.70-3.91	Diss by acids. Pleoc. wk, Y yellow, Z pale yellow. Fluor wk, dirty green in UV.	

1.815	1.813	1.857	.078	ALLANITE (Epidote grp) (Ce,Ca,Y)(Al,Fe) ₃ Si ₃ 0 ₁₂ (OH)	80° r > v str	Y = $\frac{b}{c}$ X:c = 34°	MCL	001, 100 poor	Brownish- black	H 6 G 4.13 F 3	
-1.81	1.85	---	1.862	.01	PARSONSITE Pb ₂ (UO ₂)(PO ₄) ₂ ·2H ₂ O	---	Y = $\frac{b}{c}$ Z:c = 2°	TCL	010 poor fr conch	Pale yellow	H 2.5-3 G 5.73
1.835	1.80	1.86	1.88	.080	PSEUDOMALACHITE Cu ₅ (PO ₄) ₂ (OH) ₄ ·H ₂ O	46° r < v str	X:c = 23°	MCL pris	010 good	Dark green	H 4.5-5 G 4.35 F 2
	1.820	1.86	1.88	.06	CORNWALLITE Cu ₅ (AsO ₄) ₂ (OH) ₄ ·H ₂ O	Small r < v med	Y = el Z \perp clv	MCL pris	One perf	Emerald- green	H 4.5 G 4.52 F 2.5
	1.831	1.861	1.880	.049	ATACAMITE Cu ₂ Cl(OH) ₃	75° r < v str	X = $\frac{b}{c}$ Y = $\frac{a}{c}$	ORTH pris tab	010 perf 101 fair	Bright green	Diss by acids. Pleoc, X pale green, Y yellow- green, Z grass-green.
	1.818	1.866	1.909	.091	CALEDONITE Pb ₅ Cu ₂ (CO ₃)(SO ₄) ₃ (OH) ₆	~ 85° r < v wk	X = $\frac{c}{a}$ Y = $\frac{a}{c}$	ORTH	010 perf	Bluish- green	Commonly striated.
	1.827	1.869	1.879	.052	FAYALITE (Olivine grp) Fe ₂ SiO ₄	48° r > v	X = $\frac{b}{c}$ Z = $\frac{a}{c}$	ORTH	010 fair	Dark brown	H 3.5 G 3.76 F 3.5
	1.854	1.675	1.870	.210	METATUYAMUNITE Ca(UO ₂) ₂ (VO ₄) ₂ ·3·5H ₂ O	30-45° r < v	X = $\frac{c}{a}$ Y = $\frac{a}{c}$	ORTH	001 perf 010, 100 good	Canary- to greenish- yellow	Pleoc wk, X col, Y and Z pale yellow. Not fluor in UV.
	1.727	1.870	1.883	.156	ILVAITE Ca(Fe ⁺² ,Mn) ₂ Fe ⁺³ (SiO ₄) ₂ (OH) ₂	20-30° r < v str	X = $\frac{c}{b}$ Y = $\frac{d}{c}$	ORTH	010 good 001 dist	Brownish- black	Gel with acids. Pleoc, X dark green, Y yellow- brown to dark brown, Z dark brown, abs X > Y > Z.
(1.92)	1.93	1.87	1.92	.11	BAHIANITE Al ₅ Sb ₃ O ₁₄ (OH) ₂	Large r > v	---	MCL ps orth	---	Tan, brown, col	G 4.9-5.46 (5.26 calc)
	1.792	1.870	1.870	.078	ARSENIOSIDERITE Ca ₃ Fe ⁺³ ₄ (AsO ₄) ₄ (OH) ₆ 3H ₂ O	~ 0°	el pos	MCL ps tet fib	001 perf	Yellow brown	Diss by acids. Pleoc, X col, Y and Z dark reddish-brown.
	1.756	1.874	1.896	.140	CLINOCLASE Cu ₃ AsO ₄ (OH) ₃	50° (44+2°) el \perp very str	Y = $\frac{b}{c}$ Z ~ $\frac{a}{c}$	MCL pris	001 perf	Greenish- blue	Diss by acids. Pleoc, X pale blue-green, Y light blue-green, Z blue-green.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index		Brief	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.655	<u>1.875</u>	1.909	.254	MALACHITE $\text{Cu}_2\text{CO}_3(\text{OH})_2$	44° $r < v$ str	$\chi = \frac{b}{c} = 23^\circ$	MCL	201 perf 010 good	Bright green	H 3.5-4 G 4.05 F 2	Diss by acids with eff. Pleoc, χ nearly cols, γ yellow-green, Z deep green.
1.786	<u>1.875</u>	1.875	.089	PLUMBOJAROSITE (Alunite grp) $\text{PbFe}_6(\text{SO}_4)_4(\text{OH})_{12}$	Small	$\chi \perp$ plates	TRIG plates	1014 fair	Brown	G 3.63	Diss by HCl. Pleoc, χ golden-yellow, γ and Z brownish-red. Basal plates divided into sectors.
1.810	<u>1.880</u>	1.925	.115	Unknown, labeled "Scorodite"	~ 90° (75+2°)	Med	---	---	Bright green	---	Data of Larsen, 1921, p. 132.
---	<u>1.88</u>	---	.10	CHEVYINITE (Ca, Ga, Th)4(Fe, Mg)2 (Ti, Fe ⁺³) ₃ Si ₄ O ₂₂	$\chi = \frac{b}{c} = 11-$ -26°	MCL	Fr conch	H 5 G 4.3-4.6 F 4	Velvet-black	---	Gel with acids. Pleoc, χ nearly cols, γ pale red-brown, Z dark red-brown. In part isotropic, metamictic.
2.02											
300											
1.860	<u>1.880</u>	1.893	.033	FRONDELITE $\text{MnFe}^{+3}4(\text{PO}_4)_3(\text{OH})_5$	Med $r > v$ str	$\chi = \frac{c}{a}$	ORTH fib	100 perf 010 good 001 poor	Brown	H 4.5 G 3.48	Pleoc, χ pale yellow-brown, γ and Z orange-brown, abs $Z > \gamma > \chi$. Fe_2O_3 48.8, MnO 7.7%.
1.927	<u>1.88</u>	---	---	RAUVITE $\text{Ca}(\text{OH})_2\text{V}^{+5}10\text{O}_{28}$	---	---	Mass	---	Purplish-black	---	Opt. char unk.
>1.87	---	<1.89	---	STIBIVANITE $\text{Sb}^{+3}\text{V}^{+4}\text{O}_5$	85° $r > v$ str	$\chi = \frac{b}{c}$ $Z = \frac{c}{a}$	MCL fib	---	Yellow-green	---	Pleoc str, χ emerald-green with olive tint, γ emerald-green, Z olive-green. Sgn unk.
1.863	<u>1.885</u>	1.890	.027	FOURMARIERITE $\text{PbU}_4\text{O}_{13} \cdot 4\text{H}_2\text{O}$	50° $r > v$	$\chi = \frac{c}{a}$ $\gamma = \frac{a}{c}$	ORTH tab	001 perf 100 good	Reddish-orange	H 3.5 G 5.74	Pleoc, χ cols, γ pale yellow, Z amber.
1.92	[1.86]	<u>1.887</u>	1.91	.05	HETEROHITE (Fe ⁺³ , Mn ⁺³) PO_4	37°	$\chi = \frac{a}{c}$	ORTH	100 good 010 poor	Deep rose H 4 G 3.3	Diss by acids. Pleoc, χ greenish-gray, γ and Z deep red to purple, abs $Z = \gamma > \chi$. Abnormal green interferences near optic axis.
1.820	1.764	<u>1.889</u>	1.936	.172	IRIGINITE (UO ₂) $\text{MoO}_2 \cdot 3\text{H}_2\text{O}$	60°	$\chi = \frac{b}{c}$ $Z = \frac{c}{a}$	MCL	---	Canary yellow	Diss by warm HCl. Not fluor in UV.

(1.80)	<u>1.89</u>	1.91	.11	WYARITITE $\text{Ca}_3\text{U}^{+4}(\text{UO}_2)_6(\text{CO}_3)_2$ (OH) ₁₈ ·3·5H ₂ O	48°	X = $\frac{c}{b}$ Y = $\frac{d}{b}$	e1 pos	ORTH	001 perf and 010	Black to violet-black	H 3-4 G 4.69	Pleoc, X gray, Y violet, Z lavender - blue.
1.870 Λ	<u>1.815</u>	<u>1.898</u>	.083	ARSENOSIDERITE $\text{Ca}_3\text{Fe}^{+3}\text{As}_4(\text{AsO}_4)_4(\text{OH})_6$ 3H ₂ O	~ 0°	X = $\frac{c}{b}$ Y = $\frac{d}{b}$ Z:a = 14°	MCL ps tet fib	001 perf	Yellow to brown	H 4.5 G 3.60 F 3	Diss by acids. Pleoc, X cols, Y and Z dark red-brown.	
1.925	1.76	1.90	.16	CARNOTITE $\text{K}_2(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	50° r < v	X = $\frac{c}{b}$ Y = $\frac{d}{b}$	MCL	001 perf 010, 110 poor	Lemon- to bright-yellow	H soft G 4.70 F diff	Pleoc, X cols to gray, Y and Z canary- to lemon-yellow. Not fluor in UV.	
1.775	<u>1.900</u>	1.920	.145	XOCOMEGLITITE $\text{Cu}_3(\text{TeO}_4)(\text{OH})_4$	41+2°	---	ORTH(?) fib	---	Rich green	H 4 G 4.65	Diss by HNO ₃ . Pleoc in bluish-green, abs Z > X = Y.	
1.899	<u>1.901</u>	1.903	.004	QUEITITE $\text{Pb}_4\text{Zn}_2(\text{SiO}_4)(\text{Si}_{2,7})$ (SO ₄)	~ 90° r < v str	---	MCL tab	010, 001 poor	Pale yellow to cols	H 4 G 6.07	Diff sol in HNO ₃ .	
1.803	<u>1.905</u>	1.945	.142	WIDENMANNITE $\text{Pb}_2(\text{UO}_2)(\text{CO}_3)_3$	63°	X = $\frac{b}{a}$ Y = $\frac{d}{a}$ el pos	ORTH tab	010 010	Yellow	H 2 G 6.89 calc	Diss by HNO ₃ with eff. Not pleoc. Not fluor in UV.	
1.873	<u>1.905</u>	1.910	.037	YEATMANITE $\text{Mn}_7\text{Zn}_8\text{Sb}_2\text{Si}_4\text{O}_{28}$	49° r < v med	---	TCL	100 perf	Deep brown	H 4 G 5.02 F 4	Dec by HCl. Tw on 010.	
1.785	<u>1.906</u>	1.917	.132	MASUYITE $\text{UO}_2 \cdot 2\text{H}_2\text{O}$ (?)	50° (32+3°) r > v	X = $\frac{c}{b}$ Y = $\frac{d}{b}$	ORTH tab	001 perf 010 good	reddish- brownish- orange	H 5.08	Tw 110 and 130 common. Pleoc, X pale yellow, Y and Z deep golden. Not fluor in UV.	
1.855 Λ	<u>1.770</u>	<u>1.907</u>	1.915	STRELKINITE $\text{Na}_2(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 6\text{H}_2\text{O}$	Ned (26+3°)	X = $\frac{c}{b}$ Y = $\frac{d}{b}$ el pos	ORTH plates	001 perf	Gold- to canary- yellow	H 2-2.5 G 3.7-3.91	Diss by acids. Pleoc wk, Y yellow, Z pale yellow. Fluor wk, dirty green in UV.	
1.685	<u>1.91</u>	1.93	.245	IANTHINITE $\text{UO}_2 \cdot 5\text{UO}_3 \cdot 10\text{H}_2\text{O}$	Large (30+15°)	X = $\frac{c}{b}$ Y = $\frac{d}{b}$	ORTH acitic	001 perf 100 good	Black to violet	H 2.5	Pleoc, X cols, Y purple, Z violet.	
1.750 Υ	<u>1.910</u>	1.945	.195	FRANCEVILLITE $\text{Ba}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 5\text{H}_2\text{O}$	46°	X = $\frac{c}{b}$ Y = $\frac{d}{b}$ el pos	ORTH	001 perf	Yellow	H 3 G 4.0	Pleoc, X cols, Y and Z yellow. Not fluor in UV.	
1.885	<u>1.910</u>	1.913	.028	BARTELKEITE $\text{PbFe}^{+2}\text{Ge}_3\text{O}_8$	35° r < v	Z = b	MCL	101 dist	cols to pale green	H ~ 4 G (4.97)	Diss by hot HCl.	
1.820	<u>1.920</u>	1.955	.135	LAZARENKOITE $(\text{Ca}, \text{Fe}^{+2})\text{Fe}^{+3}\text{As}^{+3}$ 3H ₂ O	~ 30°	e1 pos	ORTH fib	---	Orange	H 1 G 3.45	Diss by HCl. Pleoc, X pale yellow, Y pale brown, Z rose-brown.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Mineral name and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	System and habit	Cleavage or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ								
1.870	1.830	(1.92)	1.925	.095 ILVATE $\text{Ca}(\text{Fe}, \text{Mn})_2\text{Fe}^{+3}(\text{SiO}_4)_2$ (OH)	20-30° $r < v$ str	$X = \frac{c}{b}$ $Y = \frac{c}{a}$	ORTH	010 good 001 dist	Brownish-black	H 5.5-6 G 3.98 F 2.5	Gel with acids. Pleoc, X dark green, Y and Z dark brown, abs X > Y > Z.
1.885	1.85	1.92	1.94	.09 FOURMARIERITE $\text{PbU}_4\text{O}_{13} \cdot 4\text{H}_2\text{O}$	Large $r > v$	$X = \frac{c}{b}$ $Y = \frac{c}{a}$	ORTH tab	001 perf 100 good	Reddish-orange	H 3.5 G 5.74	Pleoc, X pale yellow, Y and Z deeper yellow.
1.810	1.810	1.923	1.933	.123 KOLFANITE $\text{Ca}_2\text{Fe}^{+3}(\text{AsO}_4)_3 \cdot 2 \cdot 2\text{H}_2\text{O}$	5-7° (24+4°)	---	MCL	0ne	Red to yellow	G 3.3	Pleoc, X pale yellow, Y and Z dark orange.
1.90	1.750	1.925	1.950	.200 CARNOTITE $\text{K}_2(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	43° $r < v$ (38+2°)	$X = \frac{c}{b}$ $Y = \frac{c}{a}$ $Z:a = 14^\circ$	MCL	001 perf 010, 110 poor	Lemon-to bright yellow	H soft G 4.70 F diff	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
	---	1.926	---	.04 Unnamed lead-zinc arsenate	~90° $r > v$	---	MCL tw	---	Yellow-to cols	---	Am. Mineral., 47, 418 (1962).
1.880	1.915	1.927	1.939	.024 ROCKBRIDGEITE, oxidized (Frondeelite ser) $(\text{Fe}^{+2}, \text{Mn})\text{Fe}^{+3}(\text{PO}_4)_3$ (OH) ₅	Large $r < v$	---	ORTH	100 perf 010 good	Green to brown	H 4.5 G 3.48	Diss by acids. Pleoc, X pale yellow-brown, Y and Z orange-brown. Fe ₂ O ₃ 55.2, MnO 4.1%.
1.870	1.77	1.93	1.97	.20 METATUYAMUNITE $\text{Ca}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3 \cdot 5\text{H}_2\text{O}$ (0, OH, F) ₅	40-55° $r < v$	$X = \frac{c}{b}$ $Y = \frac{c}{a}$	ORTH	001 perf 010, 100 good	Canary-to greenish-yellow	H 2 G 3.85 F easy	Pleoc, X nearly cols, Y canary yellow, Z darker yellow. Not fluor in UV.
1.886	1.886	1.930	1.939	.053 FERSMANITE $(\text{Ca}, \text{Ce}, \text{Na})(\text{Nb}, \text{Ta}, \text{Ti})_2$ (0, OH, F)	0-7° (48+6°)	$Y = \frac{b}{c}$ $X \sim \frac{c}{a}$	MCL	---	Yellow to brown	H 5.5 G 3.44	---
1.87	--	1.93	---	wk CORKITE $\text{PbFe}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	Med $r < v$ very str	$X = \frac{c}{a}$	TRIG	0001	Olive-green to brown	H 4 G 4.20 F 4-5	Basal section divided into biax segments. Abnormal interf colors green, blue, brown.
1.922	1.933	1.938	.016	PLUMBOTUMITE $\text{Pb}_5\text{Si}_4\text{O}_8(\text{OH})_{10}$	32° $r < v$ (68+15°)	---	ORTH ps hex	001 perf	cols	H 2 G 5.6	Diff sol hot HNO ₃ .

1.77	1.935	1.960	.190	SENGIERITE $\text{Cu}_2(\text{UO}_2)_2(\text{VO}_4)_2(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	38° r < v str	X = c Y = b Z ~ $\frac{c}{a}$ el neg	MCL plates	001 perf	Greenish-yellow H 2.5 G 4.41	Diss by acids. Pleoc, X pale greenish-yellow, Y and Z greenish-yellow.
---	1.94	1.95	---	SAYRITE $\text{Pb}_2(\text{UO}_2)_5\text{O}_6(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	Large	X = $\frac{b}{c}$ Y = $\frac{c}{a}$ Z ~ $\frac{a}{c}$	MCL	102 dist	Yellowish to reddish-orange H 2 G (6.76)	---
1.897	1.940	1.942	.045	SCHIEFFELINITE $\text{Pb}(\text{Te}, \text{S})_4 \cdot \text{H}_2\text{O}$	24° r < v wk	X = b Y = $\frac{c}{a}$	ORTH	010 easy	Cols, white H 2 G 4.98	---
(1.79)	1.94	1.96	(.17)	Unnamed Ca-Sr-U oxide	38°	---	ORTH ps hex	001 good	Red-orange H 2 G 5.29	Am. Mineral., 45, 254 (1960).
1.70	1.95	2.04	.34	HYDROTUNGSTITE $\text{H}_2\text{WO}_4 \cdot \text{H}_2\text{O}$	52° r < v	Y = b Y = $\frac{c}{a}$	MCL plates	010 poor	Dark-to yellow-green H 2 G 4.60	Insol in acids, diss by NH_4OH . Pleoc, X cols, Y and Z yellow-green to dark green, abs Z > Y > X.
---	1.95	1.985	---	KLEBELSBERGITE $\text{Sb}_4(\text{OH})_2\text{SO}_4$	~ 70°	X = b Y = $\frac{c}{a}$	ORTH	---	Yellow H 2 G 4.62	Not pleoc.
(1.75)	1.95	1.97	.22	RAMEAUITE $\text{K}_2\text{CaU}_6^{20} \cdot 9\text{H}_2\text{O}$	32°	X = $\frac{b}{c}$ Z:c = 5° el clv pos	MCL ps hex	010 good	Orange H 2 G 5.60	Tw 100.
1.89	1.95	2.02	.13	CEROTUNGSTITE $\text{Ce}_2\text{W}_2\text{O}_6(\text{OH})_3$	(Large)	Z = b X ~ $\frac{a}{c}$	MCL	001 perf	Orange-yellow H 1 G 5.60	Tw 001.
1.85	1.95	1.96	.11	CHAPMANITE $\text{SbFe}_2(\text{SiO}_4)_2(\text{OH})$	Small	Z = el X \perp flat face	ORTH laths	Conch	Olive-green, deep yellow H 2.5 G 3.75	Insol in acids.
1.92	1.95	1.96	.04	LARSENITE PbZnSiO_4	80° (?)	X = a	ORTH pr's	120 good	White H 3 G 5.90	Luster greasy.
1.92	1.95	1.96	.04	MANGANOSTIBITE $(\text{Mn}, \text{Fe})_7\text{SbAsO}_{12}$	Small	---	ORTH fib	---	Black to reddish-brown H 5 G 4.95	Dec by HCl . Pleoc, X reddish-brown, Z nearly opaque.
1.92	1.95	1.95	.03	KATOPRITE $(\text{Mn}, \text{Mg}, \text{Fe})_{13}(\text{Al}, \text{Fe})_4$	Small r > v str inclined	Y = b Z:a = -3°	MCL pris	001 perf	Jet black, red in thin splinters H 5.5 G 4.56	Insol in acids. Pleoc str, reddish-yellow to reddish-brown.
---	1.95	-1.985	---	KLEBELSBERGITE $\text{Sb}_4(\text{OH})_2(\text{SO}_4)$	70°	X = b Y = $\frac{c}{a}$	ORTH powdery	---	Yellow to orange-yellow H 2 G 4.62	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Brief	MINERAL NAME and formula ^a	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.910 \downarrow >2	1.785	1.952	2.002	.217	FRANCEVILLITE (Ba,Pb)(UO ₂) ₂ (VO ₄) ₂ 5H ₂ O	52°	$X = \frac{c}{b}$ $Y = \frac{b}{a}$ el cTv pos	ORTH	001 perf	Yellow	H 3 G 4.55	Pleoc., X cols, Y and Z yellow. Not fluor in UV.
1.94 $\overline{\downarrow}$ 1.985	1.702	1.955	1.965	.263	POUGHITE Fe ₂ (TeO ₃) ₂ (SO ₄)·3H ₂ O	22° r > v very str	$X = \frac{b}{a}$ Y = $\frac{c}{a}$	ORTH	010 perf 101 good	Greenish-yellow	H 2.5 G 3.76 fus	Diss by acids. Pleoc., X nearly cols, Y pale greenish-yellow, Z pale sulfur-yellow.
1.946	1.958	1.964	.018	ALAMOSITE PbSiO ₃	65° r < v very str	$Y = \frac{b}{c} = 8^\circ$ Z:c = 8°	MCL Fib	010 perf	Col s	H 4-5 G 6.49 fus	Gel with acids.	
2.115	1.815	1.960	.145	TLAPALLITE H ₆ (Ca,Pb) ₂ (Cu,Zn) ₃ (SO ₄)(Te ⁺⁴ O ₃)(Te ⁺⁶ O ₆) Med	Small	---	MCL	---	Green	H 3 G 4.7 (5.05 calc)	CuO 15.6, CaO 8.3, ZnO 0.8%.	
1.73	1.96	1.98	.25	MELANOVANADITE Ca ₂ V ⁺⁴ ₄ V ⁺⁵ ₆ 25·xH ₂ O	Z = b Y:c = 15°	TCL pris	One perf	Black, brown	H 2.5 G 3.48 F easy	Diss by acids. Pleoc., X yellow-brown, Y dark red-brown, Z dark reddish-brown to opaque.		
1.92 $\overline{1.97}$	1.94	1.96	1.96	.02	BEUDANTITE PbFe ₃ (AsO ₄) ₂ (OH) ₆	0-50° r > v	TRIG pos	0001 good	Yellow, brown, green	H 4 G 4.1 F 3.5	Diss by hot HNO ₃ . Complex tw. Base divided into biax segments. Abnormal interf colors.	
1.945	1.966	1.983	.038	OLSACHERITE Pb ₂ (SO ₄) ₂ (SeO ₄)	80°	$X = \frac{a}{c}$ Y = $\frac{c}{a}$	ORTH	101 good 010 poor	Col s	H 3-3.5 G 6.55 calc	---	
1.93	1.97	2.01	.08	BISMUTOFERRITE BiFe ₂ (SiO ₄) ₂ (OH)	---	ORTH mass	---	Yellow-green	G 4.47	---		
1.871	1.975	2.005	.134	WALPURGITITE (BiO) ₄ (UO ₂)(AsO ₄) ₂ O ₄ 3H ₂ O	52°	$Y:c = 8^\circ$	TCL tab	010 perf	Yellow to yellow-green	H 5 G 5.95 F 1.5	Tw 010. Pleoc., X pale yellow, Y and Z green.	
1.956	1.978	1.980	.024	WOELSENDORFITE (Pb,Ca)U ₂ O ₇ ·2H ₂ O	Large	---	ORTH	001 good	Orange-red	H 5 G 5.37	PbO 15.5, CaO 3.2%.	
\downarrow ~2.09	1.78	1.98	2.1	KARAPITTE (= Coronene) C ₂₄ H ₁₂	(69+7°)	---	MCL needles	One clv	Yellow	H 1 F 1	Sol in benzene.	

1.89	<u>1.98</u>	2.02	.13	YTTROTUNGSTITE $\text{Y}_2\text{O}_6(\text{OH})_3$	68°	$Z = \frac{b}{c}$ $X:c = 26^\circ$	MCL ps ortho laths	010 perf 101 poor	Yellow	G 5.96	
v >2.0	(1.97)	<u>1.98</u>	1.99	(.02)	RICHETITE Oxide of Pb and U	Large	---	MCL ps hex tab	010 perf	Black	G 7.25
1.955 ^	1.72	<u>1.985</u>	1.990	.27	POUGHITE $\text{Fe}_2(\text{TeO}_3)_2(\text{SO}_4) \cdot 3\text{H}_2\text{O}$	15-20° r > v str	$X = \frac{b}{c}$ $Y = \frac{a}{d}$	ORTH	010 perf 101 good	Greenish-yellow	H 2.5 G 3.76 fus
1.952	---	2.002	.05	KURANAKHITE $\text{PbMn}^{+4}\text{Te}^{+6}\text{O}_6$	---	---	ORTH ps hex	---	Reddish-brown to black	H 4.5	
1.970	<u>1.992</u>	2.011	.041	Unnamed arsenate analogue of Tsunemite $\text{Pb}_2\text{Cu}(\text{AsO}_4)(\text{SO}_4)(\text{OH})$	88°	---	MCL	---	Green	---	
1.990	<u>1.993</u>	1.994	.004	ELYITE $\text{Pb}_4\text{Cu}(\text{SO}_4)(\text{OH})_8$	66°	$Y = \frac{b}{c}$ $X:c = 45^\circ$	MCL pris tab	---	Lavender	H 2 G 6.3	
1.960	<u>1.995</u>	2.020	.060	JAMESITE $\text{Pb}_2\text{Zn}_2\text{Fe}^{+3}\text{O}_4(\text{AsO}_4)_5$	75° r > v	$Y:a$ on (001) = 5°	TCL tab	---	Reddish-brown	H ~ 3	
305	1.952 ^	---	>2	CURIENITE $\text{Pb}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 5\text{H}_2\text{O}$	66°	$X = \frac{c}{c}$ $Y = \frac{d}{b}$	ORTH	---	Canary yellow	G 4.88	
1.82	>2.0	>2.0	>1.8	GRANTSITE $\text{Na}_4\text{Ca}_X\text{V}^{+4} \cdot 2\text{X} \cdot \text{V}^{+5}$ $0_{32} \cdot 8\text{H}_2\text{O}$	---	$Z = \frac{b}{c}$	MCL	---	Olive-green	H 1 G 2.94	
1.87	2.00	2.01	.14	LEADHILLITE $\text{Pb}_4(\text{SO}_4)(\text{CO}_3)_2(\text{OH})_2$	10° r < v str	$Z = \frac{b}{c}$ $X:c = 5^\circ$ el-clv pos	MCL ps hex el-clv pos	001 perf fr conch	Col, gray, blue, green	H 2.5-3 G 6.55 F 1.5	
1.98 ^	>2.0	<2.07	---	RICHETITE Hydrous oxide of Pb, U	Large	---	MCL (?)	010 perf	Black	G 7.25	
1.797	>2.0	>2.0	>.2	BARNESITE $\text{Na}_2\text{V}_6^{16} \cdot 3\text{H}_2\text{O}$	---	$Z = \frac{b}{c}$ $X:c = 5^\circ$	MCL bladed	---	Dark red	G 3.15 F 2	
1.986	<u>1.930</u>	<u>2.002</u>	2.020	.090	LINDGRENITE $\text{Cu}_3(\text{MoO}_4)_2(\text{OH})_2$	71° (52+3°) r > v	$Z = \frac{b}{c}$ $X:c = -7^\circ$ el-clv neg	MCL tab	Apple-green	H 4.5 G 4.26 F easy	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.99	1.928	<u>2.007</u>	2.036	.108	LANARKITE $Pb_2(SO_4)_0$	60° $r > v$ str inclined	$y = \frac{b}{c}$ $z:c = 30^\circ$	MCL el b	201 perf	Gray, greenish	H 2.5 6 6.92	Diss by KOH or warm HNO_3 . Poly tw 010 rare. Fluor yellow in uv.
(1.85)		<u>2.01</u>	2.06	(.21)	AGRINIERITE ($K_2, Ca, Sr)U_3Si_10 \cdot 4H_2O$	55°	$x = \frac{a}{b}$ $y = \frac{a}{c}$ $z:c = 30^\circ$	ORTH tab	001 good	Orange	G 5.7	Sector tw.
2.04	1.92	<u>2.01</u>	2.02	.10	LORENZENITE $Na_2(Ti, Zr)_2Si_2O_9$	25-49° $r > v$	$x = \frac{a}{b}$ $y = \frac{a}{c}$	ORTH	100 perf 110 good	Brown to black	H 6 6 3.37	Insol in acids. Pleoc, X and Y reddish-yellow, Z pale yellow.
	1.942	<u>2.010</u>	2.024	.082	WHERRYITE $Pb_4Cu(CO_3)(SO_4)_2$ (Cl, OH) ₂	50°	Wavy ext	MCL	---	Light green	G 6.45 (7.2 calc) fus	Diss by HNO_3 .
1.96	<u>2.01</u>	2.04	.08		BROWN MILLERITE $Ca_2(Al, Fe)_2O_5$	75°	---	ORTH	---	Reddish-brown	G 3.76 infus	Data for synth compd. Pleoc, X and Y yellow-brown, Z dark brown.
2.00	<u>2.01</u>	2.02	.02		TURANITE $Cu_5(VO_4)_2(OH)_4$ (?)	Med $r > v$ str	---	Reniform, radiating, fib	---	Olive-green	H 5	Pleoc, X and Y brown, Z green.
2.00	<u>2.01</u>	2.02	.02		CALCIOVOLBORTHITE $Ca_3Cu(VO_4)_2(OH)$	~ 90° $r < v$	$x = \frac{a}{b}$ $y = \frac{a}{c}$	ORTH	010 perf 001 good	Yellow, green, brown	H 3.5 6 3.75	Diss by HCl. Pleoc wk.
<2.01	>2.01	>2.01	---		HENDERSONITE $Ca_2V^{4+}V^{5+}_8O_{24} \cdot 8H_2O$	Med $r > v$	$x = \frac{a}{b}$ $y = \frac{a}{c}$	ORTH fib	---	Dark greenish-black	H 2.5 6 2.79	Diss by alkalis. Pleoc, X salmon-red, Y green, Z brown. Mean n calc 2.10.
>2.01	>2.01	>2.01	high		GERSTLEYITE $Na_2(Sb, As)_8S_13 \cdot 2H_2O$	Large (?)	---	MCL (?)	010, T01 perf	Cinnabar-red to reddish-black	H 2.5 G 3.62 F 2	Opt char unk.
1.905	<u>2.02</u>	>2.02	>.115		NAVAJOITE $V_2O_5 \cdot 3H_2O$	---	$z = \frac{b}{c}$	MCL fib b	---	Dark brown	H 1 G 2.56	Pleoc, X and Y yellow-brown, Z dark brown.

<u>2.01</u>	<u>1.96</u>	<u>2.06</u>	.10	PERRIERITE $(\text{Ca}, \text{Ce}, \text{Th})_4(\text{Mg}, \text{Fe})_2$	63° $r > v$ str	$Z = \frac{b}{a} = 24^\circ$ $X: \underline{a} = 24^\circ$	MCL	---	Brownish-black	H 6 G 4.3-4.45	Pleoc., X yellow-brown, Y deep reddish-brown, Z dark brown to opaque, abs Z > Y > X.	
<u>1.88</u>	<u>1.97</u>	<u>2.02</u>	.08	CHEVKINITE $(\text{Ca}, \text{Ce}, \text{Th})_4(\text{Fe}, \text{Mg})_2$	Large	$Z = \frac{b}{c} = 11-$ $X: \underline{c} = 26^\circ$	MCL	Conch	Brown to black	H 5.5 G 4.63	Tw 001. Pleoc., X dark brown, Y and Z nearly opaque.	
<u>2.018</u>	<u>2.023</u>	<u>2.025</u>	.007	SONORITAITE $\text{FeTe}_3(\text{OH}) \cdot \text{H}_2\text{O}$	23°	---	MCL plates	---	Yellowish green	H 3 G 3.95	Diss by acids.	
<u>2.24</u>	<u>1.82</u>	<u>(2.03)</u>	.22	TUNGSTITE $\text{WO}_3 \cdot \text{H}_2\text{O}$	27° $r < v$	$X = \frac{c}{b}$ $y = \frac{a}{d}$ el cTV pos	ORTH	001 perf	Dark to light yellow	H 1.5 G 5.52	Diss by NH_4OH , insol in HCl .	
<u>1.990</u>	<u>2.030</u>	<u>2.035</u>	.045	ARSENDESCOLIZITE $\text{PbZn}(\text{AsO}_4)(\text{OH})$	~30° $r > v$	$X = \frac{b}{d}$ $y = \frac{a}{c}$	ORTH tab 001	---	Pale yellow	H ~4 G (6.57)	---	
<u>1.98</u>	<u>2.04</u>	<u>2.10</u>	.12	FIEDLERITE $\text{Pb}_3\text{Cl}_4(\text{OH})_2$	Large $r < v$	$Z = \frac{b}{c} = -34^\circ$ $Y: \underline{c} = -34^\circ$	MCL	100 good	cols	H 3.5 G 5.88	Diss by HNO_3 . Tw on 100 common.	
<u>2.01</u>	<u>1.95</u>	<u>2.04</u>	.11	LORENZENITE $\text{Na}_2\text{Ti}_2\text{Si}_2\text{O}_9$	37-41° $r > v$	$X = \frac{a}{d}$ $y = \frac{b}{c}$	ORTH	100 perf 110 good	Brown to black	H 6 G 3.43	Insol in acids. Pleoc., X and Y pale reddish- yellow, Z pale yellow.	
<u>2.129</u>	<u>2.01</u>	<u>2.04</u>	.07	VESTIGANITE $\text{BaCu}_3(\text{VO}_4)_2(\text{OH})_2$	60°	Ext:t:w p1 = 10°	MCL ps hex	001 good	Yellow-to olive-green	H 3.5 G 4.05	Poly tw.	
<u>2.01</u>	<u>2.01</u>	<u>2.04</u>	.06	VOLBORTHITE $\text{Cu}_3(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	Large $r < v$ str	---	MCL	001 perf	Yellow to dark green	H 3.8 G 3.42	Tw. Diss by acids. Pleoc., X and Y yellow, Z yellow-green.	
<u>1.908</u>	<u>(1.91)</u>	<u>2.05</u>	.215	(.24)	SCHMITTERITE $(\text{UO}_2)\text{TeO}_3$	75°	$X = \frac{c}{b}$ $y = \frac{a}{d}$	ORTH	100	Light yellow	H 1 G 6.9	---
<u>2.042</u>	<u>2.050</u>	<u>2.065</u>	.157	PINAKIOLITE $(\text{Mg}, \text{Mn}^{+2})_2\text{Mn}^{+3}\text{BO}_5$	32°	$Z = \frac{b}{a}$ el cTV pos	MCL	010 perf	Black	H 6 G 3.88	Diss by HCl. Pleoc in red-browns.	
<u>2.061</u>	<u>1.96</u>	<u>---</u>	<u>2.05</u>	---	FERNANDINITE $\text{CaV}^{+4} \text{V}^{+5} \text{O}_{10} \text{O}_{30} \cdot 14\text{H}_2\text{O}$	---	---	Cryptocryst	---	Dull green	---	Diss by acids. Not pleoc. Opt char unk.
<u>2.03</u>	<u>2.06</u>	<u>2.08</u>	.05	KARIKIBITE $\text{Fe}^{+3} \text{As}^{+3} \text{O}_4(\text{OH})_9$	>14	$X = \underline{c}$	HEX pris	10TO	Green, brown, yellow	H 4 G 7.05 F 1.5	Diss by HNO_3 . Pleoc., X greenish-yellow, Z green.	
<u>2.08</u>				DUFTITE $\text{PbCuAsO}_4(\text{OH})$	Large $r > v$	---	ORTH fib	---	Brownish-yellow	Soft G 4.07	Pleoc wk, straw-yellow to brownish-yellow.	
							ORTH	X = <u>c</u> Y = <u>d</u>	Canary yellow	G 4.88	---	

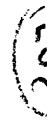


Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.786	<u>2.070</u>	2.075	2.075	.289	SALESITE $\text{Cu}(\text{IO}_3)_2(\text{OH})$	0-5° (12+2°) $r > v$ extr	$X = \frac{a}{b}$ $Y = \frac{c}{b}$ $e1 \text{ pos}$	ORTH	110 perf	Green	H 3 G 4.77	Diss by HNO_3 . Pleoc, X cols, Y light bluish-green, Z bluish-green.
1.804	<u>2.074</u>	2.076	2.076	.272	CERUSSITE (Aragonite grp) PbCO_3	9° $r > v$ str	$X = \frac{c}{b}$ $Y = \frac{a}{b}$	ORTH	110, 021 good	Cols, white	H 3.5 G 6.55 F easy	Diss in dil HNO_3 with eff. Tw on 110 common.
1.95	<u>2.08</u>	2.11	.16		CAFETITE $\text{Ca}(\text{Fe}, \text{Al})_2\text{Ti}_4\text{O}_{12} \cdot 4\text{H}_2\text{O}$	Med	$Z:c = 3^\circ$ $e1 \text{ pos}$	MCL f1b c	Pris	Pale yellow to cols	H 4.5 G 3.28 F easy	Insol in acids.
1.96	<u>2.09</u>	2.10	.14		EMMONSITE $\text{Fe}_2\text{Te}_3\text{g} \cdot 2\text{H}_2\text{O}$	20° $r > v$ str	$Y = \frac{b}{c}$	TCL fib	010 perf	Yellow-green	H 5 G 4.59	Tw. Pleoc, X light green, Z darker green.
1.978	2.05	---	2.09	.04	WOELSENDORFITE $\text{PbU}_2\text{O}_7 \cdot 2\text{H}_2\text{O}$	----	----	ORTH	001 good	Orange-red	G 6.08	PbO 23.1, CaO 1.2%.
	---	---	<u>2.09</u>	---	MONTANITE $\text{Bi}_2\text{TeO}_6 \cdot 2\text{H}_2\text{O}$	Small $r < v$ extr	----	MCL (?) fib	----	Yellowish white to grayish-yellow	Soft G 3.79 F 1.5	Diss by acids. Abnormal green interf colors.
	1.982	<u>2.095</u>	2.19	.208	CARLRIESITE $\text{CaTe}^{+4} \text{Te}^{+6} \text{O}_8$	80°	$Y = \frac{b}{c}$ $X:c = 28^\circ$	MCL	010 fair	Bright yellow	H 3.5 G 6.3	Diss by hot dil HCl. Pleoc wk in yellows, abs Z > X = Y.
	2.01	---	---	---	SANTAFEITE $\text{Na}_2(\text{Mn}, \text{Ca}, \text{Sr})_6\text{Mn}^{+4}_3$ (V, As) $^{0.28} \cdot 8\text{H}_2\text{O}$	----	$X = \frac{c}{b}$	ORTH ro-settes	010 perf 110 dist	Brownish-black	G 3.38 F easy	Diss by HCl. Pleoc, yellow-brown to dark red-brown, abs X > Y > Z. Opt char unk.
	1.70 Li	<u>2.10</u> _{Lj}	2.23 _{Lj}	.53	METAHEWETTITE $\text{CaV}_6\text{O}_{16} \cdot 9\text{H}_2\text{O}$	52°	$Z = \frac{b}{c}$	MCL laths	----	Deep red	Soft G 2.51- 2.94	Diss by acids. Pleoc, X light orange-yellow, Y and Z deep red.
	>2.10	>2.10	---	---	NAMIBITE CuBi_2V_6	Mod for red, small for blue light	$Z = \frac{b}{c}$ $X:a = 12^\circ$	MCL tw	100 good	Dark green	H 4.5-5 G 6.86	Diss by acids. Pleoc, X yellow-green, Y pistachio green, Z dark green, abs Z > Y > X.
	>2.09	>2.09	---	---	MOUNANAITE $\text{PbFe}_2(\text{VO}_4)_2(\text{OH})_2$	Ext: $c = 38^\circ$ on (0.01)	TCL e1 c	----	----	Brownish-red	G 4.85	Tw common. Pleoc, brownish-yellow to brownish-red. Opt char unk.

2.00	<u>2.10</u>	2.11	.11	CLARKEITE (Na,Ca,Pb) ₂ U ₂ (OH) ₇	40°	r < v	MCL	010 perf	Bright yellow to	Dark brown	H 4.5	G 6.3	Pleoc wk in deep orange.	
2.07	<u>2.10</u>	2.12	.05	TRIGONITE Pb ₃ Mn(AsO ₃) ₂ (AsO ₂ OH)	Large	r < v	MCL domatic	101 good	brownish	orange.	H 2.5	G 6.5	Diss by acids.	
2.06	<u>2.11</u>	2.15	.09	CURITE Pb ₂ U ₅ O ₁₇ •4H ₂ O	Large	r > v str	ORTH	100 perf	Deep orange-red	Dark brown	H 4.5	G 7.26	Diss by acids.	
2.07	<u>2.11</u>	---	---	DUHAMELITE Pb ₂ Cu ₄ Bi(VO ₄) ₃ •8H ₂ O	---	X = $\frac{b}{a}$	ORTH	---	Yellow-green	Dark brown	H 4.5	G 7.26	Pleoc, x pale yellow, Y orange, Z red-brown. Not fluor in UV.	
2.08	<u>2.11</u>	---	---	TLAPALLITE H ₆ (Ca,Pb) ₂ (Cu,Zn) ₃ SO ₄ (Te ⁺⁴ O ₃)(Te ⁺⁶ O ₆)	---	---	MCL	---	Green	Dark brown	H 3	G 5.80	Sign unk. Pleoc wk in yellow. Diss by acids.	
1.960	<u>1.915</u>	<u>2.115</u>	2.115	.200	LAURIONITE PbCl(OH)	Large	X = $\frac{c}{a}$	ORTH tab	101 perf	Cols, white	Dark brown	H 3	G 5.38	PbO 13.6, CaO 4.3, CuO 15.8, ZnO 0.7%.
2.08	<u>2.12</u>	2.16	.08	VESIGNIEITE BaCu ₃ (VO ₄) ₂ (OH) ₂	---	Ext:tw pl = 10°	MCL	001 good	Yellow, olive	Dark brown	H 3.5	G 6.14	Diss by HNO ₃ .	
2.04	<u>2.129</u>	2.133	.08	KASSITE CaTi ₂ O ₄ (OH) ₂	58°	r > v very str	ORTH	010 perf 101 dist	Pale yellow	Dark brown	H 3.5	G 4.3-4.6	Poly tw.	
1.95	<u>2.13</u>	2.21	.26	MIMETITE (Apatite grp) Pb ₅ (AsO ₄) ₃ Cl	29°	X = $\frac{c}{d}$	MCL ps hex	---	Yellow, green	Dark brown	H 3.5	G 7.1	Pleoc wk, cols to pale yellow.	
2.12	<u>2.13</u>	2.135	.015	YITROCRASITE (Y, Th, Ca, U)(Ti, Fe) ₂ (OH) ₆	60-70°	---	ORTH pris	---	Greenish brown	Dark brown	H 5.5-6	G 5.32	Diss by HNO ₃ . Basal section in 6 segments. Tw.	
2.13	<u>2.137</u>	2.142	.012	Unnamed lead oxychloride	Large (?)	Ext:tw = 30° el pos	MCL (?)	010 dist	White	Dark brown	H 3	---	Poly tw. Am. Mineral., 59, 211 (1974).	
2.06	<u>2.14</u>	2.17	.11	MOLYBDOMENITE PbSeO ₃	~ 80°	el pos	MCL	001 perf	Cols to yellowish	Dark brown	H 3.5	G 7.07	Diss by HNO ₃ .	
2.12	<u>2.14</u>	~2.15	~.03	EZTLITE Pb ₂ Fe ⁺³ ₆ (Te ⁺⁴ O ₃) ₃ (Te ⁺⁶ O ₆)(OH) ₁₀ •8H ₂ O	---	Z:c = 3°	---	001	Deep orange	Dark brown	---	---	Sign unk. Not pleoc.	
		2.14	---	CUPROTUNGSTITE Cu ₂ WO ₄ (OH) ₂	---	---	Crypto-cryst	---	Green	Dark brown	fus	Dec by HCl. Opt char unk.		

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
2.05	2.15	2.20	.15		PARALAURIONITE $PbCl(OH)$	Med large $r < v$ str	$\gamma = b$ $Z:c = 25^\circ$	MCL tab	001 perf	Col, white	Soft 6.12 F 1	Diss by HNO_3 . Tw 100 very common.
2.01	2.15	2.15	.14		MATLOCKITE $PbFCI$	$\sim 0^\circ$	$X = c$ $\text{el } c\bar{t}v \text{ pos}$	TET	001 perf	Yellow	H 3.7.12 F 1	Diss by HNO_3 . Anom biax.
2.13	2.16	2.195	.065		PREISINGERITE $Bi_3(AsO_4)_2(OH)$	90°	---	TCL tab	---	White to gray	G (7.24)	---
---	2.16	---	.05		BISMUTITE $Bi_2(CO_3)_2+xH_2O$	Small to med	$X = c$ $\text{el } \bar{p}os$	ORTH platy	---	Yellow, dark gray	H 3.5 G 7.0 F 1.5	Diss by acids with eff. Striated 100.
v	2.36	---	---		RODALQUILARITE $H_3Fe^{+3}(TeO_3)_4Cl$	38°	---	TCL	---	Emerald-green	H 2-3 G 5.1	Not pleoc.
>2.1	---	<2.2	<0.1		BALYAKINITE $CuTeO_3$	100° (72+22°)	$X = a$ $Y = \frac{a}{b}$	ORTH	---	Gray-green, bluish-green	G 5.6	Reported as opt pos.
2.11	2.18	2.22	.11		HEMETTITE $CaV_6O_{16}\cdot 9H_2O$	Med (55+4) _L i	$Z = b$	MCL	---	Deep red	---	Diss by acids. Pleoc., X and Y light orange-yellow, Z dark red.
2.04	---	2.18	.14 (?)		TELLURITE TeO_2	$\sim 90^\circ$ $r > v$ mod	$X = b$ $Z = \frac{b}{c}$ $\text{el } c\bar{t}v \text{ pos}$	ORTH	010 perf	White	H 2 G 5.90 fus	Diss by HCl . Flexible.
1.77 Li	2.18 Li	2.35 Li	.58 Li		KLEINITE $Hg_2N(Cl,SO_4)\cdot xH_2O$	Small $r < v$ very str	---	HEX pris	0001 good	Yellow to orange	H 3.5-4 G 8.0 volat	Diss by acids. Uniax pos above 130°C. Poly tw.
2.00 Li	2.18 Li	2.35 Li	.35 Li		BADDELEYITE ZrO_2	30° $r > v$ str	$\gamma = b$ $X:c = 13^\circ$	MCL pris	001 perf 010 good	Col, brown, black	H 6.5 G 5.74 infus	Insol in acids. Poly tw on 100, 110. Pleoc. wk, abs X > Y > Z.
2.16	2.18	2.18	.02		LEPIDOCROCITE $FeO(OH)$	83°	$X = b$ $Y = \frac{b}{c}$	ORTH tab	010 perf 100 good 001 fair	Ruby red	H 5 G 4.09 infus	Diss by HCl. Pleoc., X yellow, Y reddish-orange, Z dark red orange, abs Z > Y > X.
2.13	2.19	2.20	.07		VIGEZZITE $(Ca,Ce)(Nb,Ta,Ti)_2O_6$	Large	$X = c$ $Z = \frac{c}{a}$	ORTH prism	100 dist	Orange-yellow	G (5.54)	Sign unk. Not pleoc.
1.94	2.20	2.51	.57									
2.14	---	2.315	.175									

2.12	<u>2.20</u>	2.23	.11	NATRONIOPHITE NaNbO ₃	10-30° (61+22°) r < v	X':c = 10- 15°	MCL	---	Yellow to black	H 5.5-6 G 4.40	---
2.11	<u>2.22</u>	2.22	.11	VANQUELINITE Pb ₂ Cu(CrO ₄)(PO ₄)(OH)	~ 0°	X = eI	MCL wedge-shaped	Uneven	Green, brown	H 2.5-3 G 6.06 F 2	Diss by HNO ₃ . Pleoc, X pale green, Y and Z pale brown.
2.15	<u>2.22</u>	2.23	.08	GOETHITE Fe(OH)	Small r > v str	X = b Y = c	0RTH	010 perf 100 good	Yellow- brown, reddish- brown	H 5 G 3.8-4.2 infus	Diss by HCl. Pleoc, X clear yellow, Y brown- ish-yellow, Z orange- yellow, abs Z > Y > X. Diss by acids.
2.41											
2.186	<u>2.222</u>	2.224	.038	FERVANITE Fe ⁺³ (VO ₄) ₄ •5H ₂ O	~ 0°	---	MCL (?) fib c	---	Golden- brown	---	Diss by acids.
2.10				ROOSEVELTITE Bi ₂ AsO ₄	.20		MCL	Conch	Gray to yellow	H 4-4.5 G 6.9-7.1 F 2-3	Opt char unk.
---				ONORATOITE Sr ₈ O ₁₁ Cl ₂	---	X:c = 12° Y:a = 8°	TCL ps mcl acic	---	White	G 5.3	Diss by HCl.
>2.11				MOCTEZUMITE Pb(UO ₂)(TeO ₃) ₂	---		MCL	100 perf	Bright orange	H 3 G 5.73	Diss by HCl or NaOH.
				TUNGSTITE WO ₃ •H ₂ O	5-10°	---	0RTH	001 perf 010 good	Light- greenish- yellow	to H 1.5 G 5.52	Insol in acids, diss by NH ₄ OH.
				CHLOROXIPHITE Pb ₃ CuCl ₂ (OH) ₂ ₂	27° r < v	X = c Y = b	MCL bladed	101 perf 100 good	Olive- green	H 2.5 G 6.93 F easy	Diss by HNO ₃ . Pleoc, X pale yellow-brown, Y brown, Z green.
				OBOYERITE Pb ₆ ₀ ₆ (TeO ₃) ₃ (TeO ₆) ₂ 2H ₂ O	---		TCL fib	---	White	H 1.5 G 6.46	Opt sign unk.
				FERGUSONITE-BETA YNbO ₄	Small to 34°	---	MCL ps tet	Conch	Black	H 5.6 G 5.2-5.8 infus	Dec by H ₂ SO ₄ .
2.19	<u>2.24</u>	2.24	.09	MOTTRAMITE Pb ₂ (Cu,Zn)(VO ₄)(OH)	73° r > v str	X = c Y = b	0RTH	Conch	Brownish- red to black,	H 3.5 G 5.9 F 1.5	Diss by HNO ₃ . Pleoc, X pale yellow, Y and Z deep reddish-brown.
2.28											
2.17	<u>2.26</u>	2.32	.15	DESCLOIZITE Pb(Zn,Cu)(VO ₄)(OH)	~ 90° r > v str	X = c Y = b	0RTH	Conch	Red, brown, black	H 3.5 G 6.2 F 1.5	Diss by HNO ₃ . Pleoc, X yellow, Y and Z deep reddish-brown.
2.21	<u>Li</u>	<u>2.31</u> _{Li}	2.33 _{Li}								
2.18	<u>2.27</u>	2.35	.17	IRANITE Pb ₁₀ Cu(CrO ₄) ₆ (SiO ₄) ₂ (F,OH) ₂	---	Ext:eI = 5°	TCL	---	Saffron- yellow	---	Pleoc, brownish-orange to yellow-orange.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Brief MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orienta- tion	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific grav ity, and fusibility	Remarks
	α	β	γ								
γ (2.37)	---	<u>2.29</u>	---	str (Fe,Mn)(Nb,Ta) ₂ 6	82°	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH	010 perf 100 good	Black	H 6 G 5.3 infus	Insol in acids. Tw pl 201.
2.27 [2] 2.35	---	<u>2.30</u>	---	up to .025 LUESHITE NaNbO ₃	46-80°	---	ORTH ps cub	---	Reddish- brown	H 5.5 G 4.35- 4.52	Insol in acids.
2.28	<u>2.31</u>	2.34	.06	MACQUARTITE Pb ₃ Cu(CrO ₄) ₆ SiO ₃ (OH) ₄ · 2H ₂ O	85°	$Y = \frac{b}{c} = +35°$	MCL	100 good	Orange	H 3.5 G 5.49	Dec by concd HCl.
2.10	<u>2.32</u>	2.65	.55	HEMIHEDRITE Pb ₁₀ 7n(CrO ₄) ₆ (SiO ₄) ₂ F ₂ horizontal disp	88°	X:110 = 30°	TCL tab	110 poor	Bright orange to brown	H 3-3.5 G 6.42	Diss by HNO ₃ . Tw common 223 by reflec- tion, less common 010.
2.12	<u>2.32</u>	2.32	.20	SEELIGERITE Pb ₃ (TiO ₃)Cl ₃ O	4°	---	ORTH ps tet plates	001 perf 110 good	Bright yellow	G 6.8	---
2.30 pos ^	<u>2.27</u>	2.34	.07	NIOBO-AESCHYNITE (Ce,Ca,Th)(Nb,Ti) ₂ (O,OH) ₆	85°	$X = \frac{a}{b}$ $Y = \frac{c}{b}$	ORTH	Conch	Black	H 5-5.5 G 5.04	---
2.30 Li	<u>2.34</u> Li	2.36 Li	.06	NADORITE PbSbO ₂ Cl	~ 90°	$X = \frac{b}{c}$ $Y = \frac{c}{a}$ el ctv pos	ORTH tab	010 perf	Brown to yellow	H 3.5-4 G 7.02 F 1.5	Diss by HNO ₃ . Tw pl 101.
2.31	<u>2.35</u>	2.4	.1	ASHANITE (Nb,Ta,U,Fe,Mn) ₄ O ₈	70-75°	$X = \frac{b}{c}$ $Y = \frac{a}{b}$ el pos	ORTH pris	Conch	Brownish- black	G 6.61	Pleoc, brownish-yellow to reddish-brown, abs Z > X.
2.18	<u>2.35</u>	2.35	.17	VALENTINITE Sb ₂ O ₃	Very small r < v	---	ORTH ps tet	110 perf 010 poor	Cols to yellow	H 2.5 G 5.76 F 1.5	Diss by HCl. Prism faces commonly striated 001.
2.25 Li	<u>2.35</u> Li	2.36 Li	.11 Li	SCHWARTZEMBERGERITE Pb ₆ (TiO ₃) ₂ Cl ₄ O ₂ (OH) ₂	Small disp wk	---	001 good	Honey- yellow to reddish- brown	H 2 G 7.39 F 1	Diss by acids. Crystals divided into biax sectors.	
2.20	<u>2.36</u>	2.36	.16	EMBREYITE Pb ₅ (CrO ₄) ₂ (PO ₄) ₂ ·H ₂ O	0-11°	$Y = \frac{b}{c}$	MCL tab	Conch	Orange	H 3.5 G 6.45	Pleoc, X honey-yellow, Y and Z amber.
2.16 ^	<u>2.36</u>	2.38	.08	BISMUTITE Bi ₂ (CO ₃) ₂	45°	$X = \frac{c}{a}$ el pos	ORTH platy	---	Yellow to dark gray	H 3.5 G 8.11 F 1.5	Diss by acids. Striated 100.

2.32	2.37	.05	PYROBELONITE PbMn(VO ₄) ₂ (OH)	29° r > v	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	Conch	Bright red	H 3.5 G 6.45
2.29 ^ (2.41)	2.34 (2.37)	2.40	.06	MAGNOCOLUMBITE (Mg,Fe,Mn)(Nb,Ti) ₂ 6	80°	X = $\frac{a}{B}$ Y = $\frac{c}{c}$	ORTH 010, 100	Black to brownish-black
2.19 ^	2.21 2.38	2.41	.20	BADDELEYITE ZrO ₂	31° r > v str	Y = $\frac{b}{c}$ X:c = 13°	MCL 001 perf 010 good	H 6 G 5.2 infus
2.37	(2.38)	2.39	.02	LANDAUITE NaMnZn ₂ (Ti,Fe) ₆ Ti ₁₂ O ₃₈	60° r > v wk	---	Conch	Black to greenish-brown
2.27 ^ 2.40	---	2.38	---	PEROVSKITE CaTiO ₃	~ 90° r > v	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	010 poor fr sub-conch	H 7.5 G 4.42 infus
2.25	2.382	2.41	.16	CHERVETITE Pb ₂ V ₂ O ₇	50° disp wk inclined	---	MCL 100, 010 (?)	100 poor fr sub-conch
2.22 ^	2.275	2.41	.140	GOTHHITE FeO(OH)	Small r > v str	X = $\frac{b}{c}$ Y = $\frac{c}{c}$	010 perf 100 good	H 5.5 G 4.0-4.3 infus
(2.37)	2.36 ^	(2.41)	2.44	.08	MANGANOCOLUMBITE (Mn,Mg,Fe)(Nb,Ta) ₂ 6	80°	010 perf 100 good	Diss by HCl. Pleoc, X clear yellow, Y brownish-yellow, Z orange-yellow, abs Z > Y > X.
>2.42	---	>2.42	---	BISMITE Bi ₂ O ₃	Disp str	---	010 perf 100 good	Dec by hot H ₂ SO ₄ . Penet or poly tw on 001.
2.44	2.47	2.48	.04	GIRDITE Pb ₃ H ₂ (TeO ₃)TeO ₆	70° r > v str	Z = $\frac{b}{c}$ X:c = 34°	MCL mass	H 6 G 5.4 infus
<1.83	2.49	>2.70	>.87	MARGARITASITE (Cs,K,H ₃ O) ₂ (VO ₂) ₂	45.5°	---	MCL spher	---
2.41Li	2.50Li	2.51Li	.10Li	(VO ₄) ₂ •H ₂ O PUCHERITE BiVO ₄	19° r < v	X = $\frac{c}{c}$ Y = $\frac{a}{el cTV}$	010 perf 010 tab pos	H 4 G 6.57 F 2
2.45Li	2.55Li	2.55Li	.10Li	HEMATITE, var Turgite (impure) Fe ₂ O ₃ +xH ₂ O	~ 0°	---	TRIG mass compact	Diff diss by HCl. Pleoc wk in reds.
								Diss by acids.
								Data for synth Cs compd.
								Diss by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index	Biref	MINERAL NAME and formula	$2V_x$ calc)	Optical orientation ($2V_x$ disp)	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.52 Li	<u>2.61</u> Li	2.67 Li	.15 Li	KOECHLINITE Bi_2Mo_6	$X = \frac{c}{a}$ $Y = \frac{b}{c}$ el clv neg	010 perf ORTH plates	Greenish-yellow	6.8.3 fus	Diss by HCl. Contact and penet tw 101.	
---	<u>2.63</u> calc	---	---	CLINOBISvanITE BiVO_4	Disp str	010 perf	Orange to yellow	6.95 calc	Opt char unk.	
---	<u>2.63</u> red	---	str	TENORITE CuO	$Y = \frac{b}{c}$ $Z = \frac{a}{c}$	MCL In zones perp 011, 011.	Steel-gray to black	H 3.5 G 6.45	Diss by HCl. Tw common on 011. Pleoc in browns. Opt char unk.	
---	<u>3.17</u> blue	---	---	TERLINGUAITE Hg_2Cl_6	20° (Li) r < v	MCL pris	Sulfur- to greenish-yellow	H 2.5 G 8.73	Volut when heated.	
2.35 Li	<u>2.66</u> Li	2.66 Li	.31 Li	REALGAR AsS	$Y = \frac{b}{c}$ $X: \frac{c}{a} = -11^\circ$	MCL 010 good 101, 100 120 poor	Red to orange-yellow	H 2 G 3.51 F 1	Dec by HNO_3 . Contact tw 100. Striated 001.	
2.538	<u>2.684</u>	2.704	.166	REALGAR AsS	41° r < v	MCL 010 perf	Lead gray	H 3 G 5.53 fus	Opt char unk.	
2.46 Li	<u>2.59</u> Li	2.61 Li	.15 Li	DUFRENOYSITE $\text{Pb}_2\text{As}_2\text{S}_5$	---	MCL 001 poor fr uneven	Black	H 2.5 G 6.31 F 1	Dec by HNO_3 . Tw pl 110.	
---	<u>>2.72</u> Li	---	very str	POLYBASITE $(\text{Ag}, \text{Cu})_{16}\text{Sb}_2\text{S}_{11}$	22°	MCL 001 poor ps hex tab	Black	H 2.5 G 5.25 fus	Red in thin splinters.	
---	<u>>2.72</u>	---	very str	MIARGYRITE AgSb_2S_2	Med	MCL 010 poor	Black	Lemon-to brownish-yellow	Tw pl 100. Pleoc, Y yellow, Z greenish-yellow.	
2.4 Li	<u>2.81</u> Li	3.02 Li	.62 Li	OPIMENT As_2S_3	76° r > v str	X = $\frac{b}{c}$ $Z: \frac{c}{a} = 2^\circ$ el clv pos	010 perf ps orth sectile	H 2 G 3.49 fus	Dec by HNO_3 . Tw 100.	
---	<u>~3</u>	---	extr	XANTHOCONITE $\text{Ag}_3\text{As}_2\text{S}_3$	34° r < v	MCL 100 perf ps orth	Red, orange, yellow	H 2.5 G 5.5 F 1	Volut when heated. Pleoc wk in reds.	
---	<u>~3</u>	---	extr	LIVINGSTONITE HgSb_4S_8	---	MCL acic	Lead gray	H 2 G 4.81 F 1		

2.78 _{Li}	<u>3.06</u> _{Li}	3.07 _{Li}	.29 _{Li}	HUTCHINSONITE $(\text{Pb}, \text{Tl})_2\text{As}_5\text{S}_9$
3.08 _{Na}	<u>3.18</u> _{Na}	3.19 _{Na}	.11 _{Na}	
---	<u>3.27</u>	---	---	SMITHITE AgAsS_2
3.19 _{Li}	<u>4.05</u> _{Li}	4.30 _{Li}	1.11 _{Li}	STIBNITE Sb_2S_3

20° Li	37° Na	010 good fr conch	H 2
r < v extr			G 4.6 fus
65°		MCL	
SMITHITE AgAsS_2		100 perf	H 2
very str			G 4.88
el clv pos			F 1
26° Li (46+2°)	X = $\frac{a}{c}$ Y = $\frac{c}{a}$ el clv neg	ORTH pris	
		010 perf	H 2
		100, 110 poor	G 4.63
			F 1

CHAPTER IV. MINERAL GROUPS (TABLES 8-38)

For these tables, the minerals are arranged in groups according to the chemical and physical characteristics they hold in common, following the terminology of Fleischer (1983). For each group, a brief statement of these common characteristics is given, followed by representative examples in tabular form.

These are arranged in order of increasing refractive index, similar to the arrangement in the identification tables above but omitting in most cases those properties that are the same or nearly the same for all members of the group.

Not included are several groups, notably the chlorite, kaolinite-serpentine, and montmorillonite groups, mainly because of difficulties of nomenclature and lack of reliable criteria for distinguishing between the members under the microscope.

Table 8. Alunite group

The minerals of this group have the general formula $AB_6(SO_4)_4(OH)_{12}$, in which A may be K_2 , Na_2 , $(NH_4)_2$, Ag_2 , $(H_3O)_2$, Ca (the hydronium ion), and Pb; B may be Al and Fe^{+3} . In beaverite and osarizawaite, A = Pb, B = Cu_2Al_4 or $Cu_2(Fe^{+3}, Al)_4$.

Alunite	$KAl_3(SO_4)_2(OH)_6$
Alunite	$(K, Na)Al_3(SO_4)_2(OH)_6$
Ammoniojarosite	$(NH_4)Fe_3(SO_4)_2(OH)_6$
Argentojarosite	$AgFe_3(SO_4)_2(OH)_6$
Beaverite	$PbCu(Fe, Al)_2(SO_4)_2(OH)_6$
Hydronium Jarosite	$(H_3O)Fe_3(SO_4)_2(OH)_6$
Jarosite	$KFe_3(SO_4)_2(OH)_6$
Natroalunite	$NaAl_3(SO_4)_2(OH)_6$
Natroalunite	$(Na, K)Al_3(SO_4)_2(OH)_6$
Natrojarosite	$NaFe_3(SO_4)_2(OH)_6$
Osarizawaite	$PbCuAl_2(SO_4)_2(OH)_6$
Plumbojarosite	$PbFe_6(SO_4)_4(OH)_12$

These minerals are trigonal (pseudo-cubic), space group $R\bar{3}m$ (except plumbogummite, $\bar{R}\bar{3}m$). Distinct to perfect basal cleavage is characteristic, with hardness between 2 and 4. Optically, they are uniaxial, although some samples of jarosite are anomalously biaxial. The minerals of this group are closely related to those of the beudantite and plumbogummite groups.

Table 8. Alunite group (continued)

Refractive index		Biref	NAME	G
ω	ϵ			
1.572	1.592	(+).020	ALUNITE	2.6
1.574	1.590	(+).016	NATROALUNITE	2.78
1.583	1.595	(+).012	ALUNITE	2.7
1.603	1.611	(+).008	NATROALUNITE	2.8
1.714	1.731	(+).017	OSARIZAWAITE	4.04
1.791	1.705	(-).086	JAROSITE	2.9-3.2
1.800	1.750	(-).050	AMMONIOJAROSITE	3.11
1.815	1.740	(-).075	NATROJAROSITE	3.18
1.816	1.728	(-).088	HYDRONIUM JAROSITE	3.1-3.2 calc.
1.820	1.715	(-).105	JAROSITE	3.2
1.832	1.750	(-).082	NATROJAROSITE	3.18
1.85	(1.81)	(-).04	BEAVERITE	4.08-4.36
1.875	1.785	(-).090	PLUMBOJAROSITE	3.63
1.882	1.785	(-).097	ARGENTOJAROSITE	3.66

Table 9. Amphibole group

The amphiboles form a complex group of double-chain silicates having the general formula $A_{0-1}B_2C_5T_8O_{22}(OH,F,Cl)_2$, in which A may be Ca, Na, and K; B may be Fe⁺², Mg, Mn, and Li; C may be Al, Cr, Fe⁺², Fe⁺³, Mg, and Mn; T may be mainly Si and Al but may have Cr, Fe⁺³, and Ti. The nomenclature used here generally follows that recommended by the International Mineralogical Association (Leake, 1978).

A. <u>Fe-Mg-Mn AMPHIBOLES</u>		Mg
1. Orthorhombic forms		Mg + Fe ⁺²
Anthophyllite	$\} (Mg,Fe^{+2})_7Si_8O_{22}(OH)_2$	0.1-0.9
Magnesio-anthophyllite		0.9-1.0
Gedrite		0.1-0.9
Magnesio-gedrite	$\} (Mg,Fe^{+2})_5Al_2Si_6Al_2O_{22}(OH)_2$	{
Ferro-gedrite		0.9-1.0
Holmquistite	$Li_2(Mg,Fe^{+2})_3Al_2Si_8O_{22}(OH,F)_2$	0.1-0.9
2. Monoclinic forms		
Cummingtonite	$\} (Mg,Fe^{+2})_7Si_8O_{22}(OH)_2$	0.3-0.7
Grunerite		0.0-0.3
Tirodite	$\} Mn_2^{+2}(Mg,Fe^{+2})_5Si_8O_{22}(OH)_2$	0.5-1.0
Dannemorite		0.0-0.5
Clinoholmquistite	$Li_2(Mg,Fe^{+2})_3Al_2Si_8O_{22}(OH)_2$	0.1-0.9
B. <u>CALCIC AMPHIBOLES</u>		Mg + Fe ⁺²
Tremolite		0.9-1.0
Actinolite	$\} Ca_2(Mg,Fe^{+2})_5Si_8O_{22}(OH)_2$	{ 0.5-0.9
Ferro-actinolite		0.0-0.5
Edenite	$NaCa_2(Mg,Fe^{+2})_5Si_7Al_2O_{22}(OH)_2$	0.5-1.0

Table 9. Amphibole group (continued)

		<u>Mg</u>
B. <u>CALCIC AMPHIBOLES</u> (continued)		
Hastingsite		0.0-0.3
Magnesio-hastingsite	$\} \text{NaCa}_2(\text{Mg},\text{Fe}^{+2})_4\text{Fe}^{+3}\text{Si}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	{ 0.7-1.0
(Hornblende)		
Magnesio-hornblende		0.5-1.0
Ferro-hornblende	$\} \text{Ca}_2(\text{Mg},\text{Fe}^{+2})_4\text{AlSi}_7\text{AlO}_{22}(\text{OH},\text{F})_2$	{ 0.0-0.5
Kaersutite	$\text{NaCa}_2(\text{Mg},\text{Fe}^{+2})_4\text{TiSi}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	0.5-1.0
Pargasite	$\text{NaCa}_2(\text{Mg},\text{Fe}^{+2})_4\text{AlSi}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	0.3-1.0
Tschermakite		0.5-1.0
Ferrotschermakite	$\} \text{Ca}_2(\text{Mg},\text{Fe}^{+2})_3\text{Al}_2\text{Si}_6\text{O}_{22}(\text{OH})_2$	{ 0.0-0.5
Ferro-ferry-tschermakite	$\text{Ca}_2(\text{Fe}^{+2},\text{Mg})_5\text{Fe}^{+3}\text{Si}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	
C. <u>SODIC-CALCIC AMPHIBOLES</u>		
Richterite		0.5-1.0
Ferro-richterite	$\} \text{Na}_2\text{Ca}(\text{Mg},\text{Fe}^{+2})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	{ 0.0-0.5
Taramite	$\text{Na}_2\text{Ca}(\text{Fe}^{+2},\text{Mg})_3\text{Al}_2\text{Si}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	0.0-0.5
D. <u>ALKALI AMPHIBOLES</u>		<u>Mg + Fe⁺²</u>
Arfvedsonite		0.0-0.5
Magnesio-arfvedsonite	$\} \text{Na}_3(\text{Mg},\text{Fe}^{+2})_4\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH},\text{F})_2$	{ 0.5-1.0

Table 9. Amphibole group (continued)

		Mg
D. <u>ALKALI AMPHIBOLES</u> (continued)		
Eckermannite		
Kozulite	$\} \text{Na}_3(\text{Mg},\text{Li},\text{Fe}^{+2})_4(\text{Al},\text{Fe}^{+3})\text{Si}_8\text{O}_{22}(\text{OH})_2$	0.5-1.0
Glaucophane		
Ferro-glaucophane	$\} \text{Na}_2(\text{Mg},\text{Fe}^{+2})_3\text{Al}_2\text{Si}_8\text{O}_{22}(\text{OH})_2$	{ 0.5-1.0 0.0-0.5
Crossite	$\text{Na}_2(\text{Mg},\text{Fe}^{+2})_3(\text{Al},\text{Fe}^{+3})_2\text{Si}_8\text{O}_{22}(\text{OH})_2$	
Riebeckite		
Magnesio-riebeckite	$\} \text{Na}_2(\text{Mg},\text{Fe}^{+2})_3\text{Fe}^{+3}_2\text{Si}_8\text{O}_{22}(\text{OH})_2$	{ 0.0-0.5 0.5-0.1

Members of this group may be orthorhombic, space group Pnma, or monoclinic, space group C2/m (also P2₁/m). They are generally prismatic with perfect cleavage on {210} at 124-126°. The orthorhombic amphiboles have orientation Y=b and Z=c; most monoclinic amphiboles have Y=b and Z:c=4-26°, thus with positive elongation. Some alkali amphiboles, such as riebeckite, eckermannite, and crossite, have Z:c > 45°, therefore with negative elongation. All amphiboles have hardness from 4.5 to 6.5 and are insoluble in all acids except HF. In general, the refractive indices increase with increasing content of iron and titanium, but the substitutions are so complex that it is difficult to deduce chemical composition from refractive indices alone.

Table 9. Amphibole group (continued)

A. Fe-Mg-Mn AMPHIBOLES

1. Orthorhombic forms

Refractive index	NAME			2V	Wt. pct. selected oxides	Remarks	
	α	β	γ		Al ₂ O ₃	Fe ₂ O ₃	
1.593 <u>1.605</u>	1.613	MAGNESIO-ANTHOPHYLLITE		(-)65°	1.6	0.3	---
1.613 <u>1.625</u>	1.634	MAGNESIO-GEDRITE		(-)71° (86+12°)	16.1	0.3	1.5
1.616 <u>1.628</u>	1.641	ANTHOPHYLLITE		(+)79° (88+9°)	1.9	---	11.1
1.616 <u>1.630</u>	1.641	ANTHOPHYLLITE		(-)88°	0.2	1.0	11.7
1.616 <u>1.634</u>	1.646	HOLMQUISTITE		(-)56° r > v wk	14.3	1.0	7.3 Li ₂ O 3.2%. Pleoc, X and Y cols, Z lilac.
1.624 <u>1.645</u>	1.651	HOLMQUISTITE		(-)50° r > v wk	13.0	2.2	8.9 Li ₂ O 2.4%. Pleoc, X grayish-yellow, Y lavender, Z blue.
1.645 <u>1.649</u>	1.661	ANTHOPHYLLITE		(+)59°	1.8	1.8	20.5
1.648 <u>1.655</u>	1.662	GEDRITE		(+)87°	13.3	1.3	14.6
1.654 <u>1.660</u>	1.667	ANTHOPHYLLITE		(+)81°	8.1	2.2	18.4
1.657 <u>1.667</u>	1.678	GEDRITE		(+)87°	17.8	1.0	18.3

Table 9. Amphibole group (continued)

A. Fe-Mg-Mn AMPHIBOLES (continued)

1. Orthorhombic forms (continued)

Refractive index	NAME			2V	Wt. pct. selected oxides			Remarks
	α	β	γ		Al ₂ O ₃	Fe ₂ O ₃	FeO	
1.671 <u>1.681</u>	1.690	FERRO-GEDRITE		(-)75°	18.7	0.9	24.4	
1.680 <u>1.688</u>	1.700	FERRO-GEDRITE		(+)83°	17.7	3.6	26.2	
1.694 <u>1.710</u>	1.722	FERRO-GEDRITE		(-)82°	19.7	2.3	33.5	MnO 2.3%, Pleoc, X pale green, Y brown-green, Z bluish green.

2. Monoclinic forms

Refractive index	NAME			2V	Z:C	Wt. pct. selected oxides			Remarks
	α	β	γ			Al ₂ O ₃	Fe ₂ O ₃	FeO	
1.610 <u>1.627</u>	1.633	CLINOHOLMQUISTITE		(-)58°	(X:a=16°)	13.5	0.4	5.9	0.5
1.630 <u>1.640</u>	1.655	CUMMINGTONITE	(+) r > v	78°	20°	2.8	1.9	16.8	0.8
1.630 <u>1.644</u>	1.652	TIRODITE	(+)	73°	20°	---	---	5.0	16.6

Table 9. Amphibole group (continued)

A. Fe-Mg-Mn AMPHIBOLES (continued)

2. Monoclinic forms (continued)

Refractive index	NAME			2V	Z:C	Wt. pct. selected oxides			Remarks	
	α	β	γ			Al ₂ O ₃	Fe ₂ O ₃	FeO		
1.638	<u>1.651</u>	1.665	TIRODITE	(+)89° r > v	19°	---	---	11.1	13.2	Pleoc wk in green.
1.651	<u>1.664</u>	1.678	CUMMINGTONITE	(+)86° r > v	17°	2.4	---	28.0	1.0	Pleoc in green.
1.650	<u>1.665</u>	1.679	GRUNERITE	(-)87° r < v	16°	---	0.6	21.9	8.0	Pleoc, X and Y yellow, Z brownish yellow.
1.660	<u>1.677</u>	1.693	GRUNERITE	(-)large r < v	---	---	3.4	31.2	---	Pleoc, X pale yellow, Z brown.
1.666	<u>1.682</u>	1.698	DANNEMORITE	(-)88° r < v	15°	---	---	22.6	15.6	Pleoc, X red-violet, Y pale violet, Z blue.
1.679	<u>1.700</u>	1.719	GRUNERITE	(-)86° r < v	14°	---	---	45.0	0.4	Pleoc, X and Y cols, Z yellow to brown.

Table 9. Amphibole group (continued)

B. CALCIC AMPHIBOLES

Refractive index				NAME	2V	Z:C	Wt. pct. selected oxides					Remarks
α	β	γ					Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	
1.581	<u>1.593</u>	1.602	TREMOLITE	(-)86°	21°	---	---	---	---	---	---	Synthetic fluor-tremorite.
1.600	<u>1.615</u>	1.627	TREMOLITE	(-)79° (63+8°) r < v wk	15°	---	---	0.6	---	---	---	K ₂ O 1.2%.
1.613	<u>1.618</u>	1.635	PARGASITE	(+)61°	26°	11.1	0.7	1.7	---	12.5	2.5	
1.622	<u>1.632</u>	1.642	ACTINOLITE	(-)80°	26°	2.2	2.8	7.2	---	---	---	
1.632	<u>1.634</u>	1.651	PARGASITE	(+)64°	17°	13.3	---	4.0	---	10.2	2.6	
1.633	<u>1.642</u>	1.652	MAGNESIO-HORNBLERDE	(-)88°	16°	5.0	2.7	8.7	---	---	---	Pleoc., X yellow, Y and Z green.
324	<u>1.643</u>	1.650	ACTINOLITE	(-)73° r < v	---	---	2.4	0.8	---	---	---	
	<u>1.645</u>	1.658	EDENITE	(-)74° (94+10°)	---	---	3.4	9.0	0.3	---	---	
	<u>1.651</u>	1.664	PARGASITE	(+)82°	18°	15.3	1.1	8.9	---	12.2	---	

Table 9. Amphibole group (continued)

B. CALCIC AMPHIBOLES (continued)

Refractive index				NAME		2V	Z:c	Wt. pct. selected oxides					Remarks
α	β	γ						Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	
1.642	<u>1.652</u>	1.660	PARGASITE	(-)88°	---	19.3	0.9	6.1	---	---	---	---	MgO 14.3%.
1.650	<u>1.663</u>	1.670	ACTINOLITE	(-)65°	---	---	3.9	14.9	0.3	---	---	---	---
1.650	<u>1.664</u>	1.671	TSCHERMAKITE	(-)81°	---	---	2.4	9.5	---	---	---	---	TiO ₂ 2.4%.
1.652	<u>1.664</u>	1.672	MAGNESIO-HASTINGSITE	(-)80°	19°	14.3	2.6	8.7	---	---	---	---	TiO ₂ 1.3%.
													Pleoc., X pale brown, Y dark brown, Z greenish-brown.
1.657	<u>1.670</u>	1.679	MAGNESIO-HORNBLENDE	(-)78°	---	---	4.7	13.2	0.3	---	---	---	MgO 10.6%.
1.660	<u>1.671</u>	1.687	FERROTSCHERMAKITE	(+)78°	18°	11.2	9.3	15.8	---	10.2	1.3	---	---
1.665	<u>1.678</u>	1.684	EDENITE	(-)65°	19°	7.8	6.2	13.4	---	---	---	---	Pleoc., X yellow, Y pale green, Z dark green.
1.667	<u>1.680</u>	1.687	PARGASITE	(-)74°	---	---	4.9	9.9	---	---	---	---	MgO 11.6%, TiO ₂ 1.9%.
1.670	<u>1.685</u>	1.693	FERRO-HORNBLENDE (72+10°)	(-)61°	18°	13.9	5.4	18.1	---	---	---	---	Pleoc., X greenish-yellow, Y olive-green, Z bluish-green.

Table 9. Amphibole group (continued)

B. CALCIC AMPHIBOLES (continued)

Refractive index				NAME		2V	Z:C	Wt. pct. selected oxides					Remarks
α	β	γ						Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	
1.676	<u>1.687</u>	1.695		MAGNESIO-HASTINGSITE	(-)80°	15°	13.9	4.8	6.9	---	---	---	TiO ₂ 4.4%.
1.670	<u>1.692</u>	1.701		KAERSUTITE	(-)81° (64+8°)	9°	14.2	3.3	5.7	---	---	---	TiO ₂ 5.7%. Pleoc., X yellow-brown, Y red-brown, Z dark red-brown.
326	1.674	<u>1.693</u>	1.700	FERRO-FERRI-TSCHERMAKITE	(-)62°	---	---	5.5	20.4	---	---	---	Ti ₂ 1.8%.
	1.688	<u>1.709</u>	1.714	HASTINGSITE	(-)51°	12°	11.5	5.9	22.9	---	---	---	TiO ₂ 1.7%. Pleoc., X greenish-yellow, Y dark olive, Z dark green.
1.687	<u>1.710</u>	1.725		KAERSUTITE	(-)77°	4°	15.5	9.3	4.9	---	---	---	TiO ₂ 5.9%.
1.694	<u>1.730</u>	1.757		KAERSUTITE	(-)80°	small	---	12.4	7.7	---	---	---	TiO ₂ 6.0%.
1.705	<u>1.731</u>	1.732		HASTINGSITE	(-)25°	13°	11.5	11.6	22.0	0.6	---	---	

Table 9. Amphibole group (continued)

C. SODIC-CALCIC AMPHIBOLES

Refractive index	NAME			2V	Z:C	Wt. pct. selected oxides				Remarks
	α	β	γ			Al_2O_3	Fe_2O_3	FeO	MnO	
1.605	<u>1.615</u>	1.622	RICHTERITE	med large $r < v$ wk	25°	---	---	---	---	---
1.616	(1.627)	1.632	RICHTERITE	(-)70°	26°	1.7	0.6	0.6	---	TiO ₂ 3.5%.
1.617	<u>1.631</u>	1.637	RICHTERITE	(-)68°	18°	3.0	3.7	0.6	5.4	---
1.688	<u>1.699</u>	1.704	FERRO-RICHTERITE	(-)35° $r < v$ (68+15°)	---	---	6.2	27.3	1.1	6.1
1.684	<u>1.700</u>	1.703	TARAMITE	(-) (46+15°)	---	15.8	9.6	13.0	---	Pleoc, X greenish-yellow, Y dark olive, Z dark green.
1.705	<u>1.713</u>	1.715	TARAMITE	(-)54°	11°	10.3	13.1	18.7	1.9	---
										Pleoc, X yellow, Y blue-green, Z deep blue.

Table 9. Amphibole group (continued)

D. ALKALI AMPHIBOLES

Refractive index	α	β	γ	NAME	2V	Z:C	Wt. pct. selected oxides					Remarks
							Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	
1.624	<u>1.631</u>	1.637	ECKERMANNITE	(-)72°	41°	1.2	8.2	---	1.2	---	---	TiO ₂ 0.3%. Pleoc, X rose, Y lilac, Z pale blue.
1.620	<u>1.637</u>	1.642	GLAUCOPHANE	(-)51° r < v str	5°	10.1	2.0	10.6	---	---	---	Pleoc, X col., Y lavender, Z blue.
1.638	<u>1.643</u>	1.650	MAGNESIO-ARFEDSONITE	(+)80°	39°	---	---	8.3	6.0	---	---	
328	<u>1.636</u>	<u>1.644</u>	1.649	ECKERMANNITE	(-)74°	65°	---	9.5	1.9	---	---	Pleoc, X and Y bluish-green, yellow-green.
1.634	<u>1.648</u>	1.653	CROSSITE	(-)small (61+13°)	82°	8.4	6.4	7.5	---	---	---	Pleoc, X yellow, Y lavender, Z blue.
1.645	<u>1.656</u>	1.661	MAGNESIO-ARFEDSONITE	(-)57°	56°	---	8.4	13.4	1.5	---	---	MgO 7.8, F 3.3%.
			FERRO-GLAUCOPHANE	(-)42° r < v str	6°	10.1	4.0	17.9	---	---	---	Pleoc, X blue-green, Y green, Z pale green.
			MAGNESIO-RIEBECKITE	(-)med	81°	0.4	15.9	3.6	0.1	---	4.5	Pleoc, X dark blue, Y indigo blue, Z yellow-green.
	<u>1.659</u>	1.662		large								
	<u>1.662</u>	1.668										

Table 9. Amphibole group (continued)

D. ALKALI AMPHIBOLES (continued)

Refractive index			NAME	2V	Z:C	Wt. pct. selected oxides					Remarks	
α	β	γ				A1 ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO		
1.667	<u>1.683</u>	1.692	MANGANOAN MAGNESIO- ARFEDSONITE	(-)6° (72±9°)	3°	---	9.1	1.0	7.7	---	9.0	Pleoc., X and Y blue-green, Z yellow-green.
1.694	<u>1.697</u>	1.698	RIEBECKITE	(-)81°	X~C	---	18.3	9.5	6.0	---	---	Pleoc., X and Y dark blue, Z yellow-green.
1.685	<u>1.717</u>	1.720	KOZULITE	(-)35° r > v wk	65°	1.7	2.9	0.0	28.0	1.1	8.4	Pleoc., X yellow- brown, Y reddish brown, Z dark brown.

Table 10. Apatite group

This group includes phosphates, arsenates, and vanadates of general formula $A_5(XO_4)_3(F,Cl,OH)$, in which A may be Ba, Ca, Ce, K, Na, Pb, Sr, and Y; X may be As, C, P, S, Si, and V. Whereas the silicates britholite, britholite-(Y), ellestadite, and hydroxylellestadite are not, strictly speaking, members of the apatite group, they are included here because they intergrade with the phosphates of the apatite group.

Alforsite	$Ba_5(PO_4)_3Cl$
Apatite-britholite	$(Ca,Ce)_5(PO_4,SiO_4)_3F$
Belovite	$(Sr,Ce,Na,Ca)_5(PO_4)_3(OH)$
Britholite	$(Ce,Ca,Y)_5(SiO_4)_3(OH,F)$
Britholite-(Y)	$(Y,Ca,Ce)_5(SiO_4,PO_4)_3(OH,F)$
Carbonate-fluorapatite	$Ca_5(PO_4,CO_3)(F,OH)$
Carbonate-hydroxylapatite	$Ca_5(PO_4,CO_3)_3(OH,F)$
Chlorapatite	$Ca_5(PO_4)_3(Cl,F,OH)$
Ellestадite	$Ca_5[(Si,S,P)O_4]_3(F,OH,Cl)$
Fermorite	$(Ca,Sr)_5(PO_4,AsO_4)(OH,F)$
Fluorapatite	$Ca_5(PO_4)_3F$
Hedyphane	$(Ca,Pb)_5(AsO_4)_3Cl$
Hydroxylapatite	$Ca_5(PO_4)_3(OH,F)$
Hydroxyl-ellestadite	$Ca_5[(Si,S)O_4]_5(OH,Cl,F)$
Johnbaumite	$Ca_5(AsO_4)_3(OH)$
Mimetite	$Pb_5(AsO_4)_3Cl$
Morelandite	$(Ba,Ca,Pb)_5(AsO_4)_3Cl$
Pyromorphite	$Pb_5(PO_4)_5Cl$
Strontium-apatite	$(Sr,Ca,Ba)_5(PO_4)_3(F,OH)$
Svabite	$Ca_5(AsO_4,PO_4)_3(F,OH)$
Vanadinite	$Pb_5(VO_4)_3Cl$

Table 10. Apatite group (continued)

These minerals are hexagonal or monoclinic (pseudohexagonal). They are infusible to difficultly fusible, except for the lead members, which fuse easily. All are readily attacked by HNO_3 ; britholite gels and britholite-(Y) is decomposed. Optically they are uniaxial or biaxial with small optic angle. All except vanadinite have low birefringence.

Refractive index			Biref	NAME	2V	G	Remarks
α or ϵ	β or ω	γ or ϵ					
	<u>1.569</u>			CARBONATE-FLUORAPATITE		2.6	Isotropic.
	<u>1.59</u>			CARBONATE-FLUORAPATITE		2.6-2.7	Isotropic.
1.598	<u>1.603</u>		.005	CARBONATE-HYDROXYL-APATITE	(-)0°	2.9	
1.612	<u>1.621</u>	1.622	.010	CARBONATE-FLUORAPATITE	(-)small	3.04	
1.614	<u>1.622</u>		.008	CARBONATE-FLUORAPATITE	(-)0°	3.05	
1.622	<u>1.627</u>	1.627	.005	CARBONATE-FLUORAPATITE	(-)0-25°	3.1-3.2	
	<u>1.63</u>			CARBONATE-FLUORAPATITE		2.7	Isotropic.
1.630	<u>1.633</u>		.003	FLUORAPATITE	(-)0°	3.2	
1.634	<u>1.638</u>		.004	FLUORAPATITE	(-)0°	3.32	
1.633	<u>1.640</u>		.007	CARBONATE-FLUORAPATITE	(-)0°	3.05	
1.638	<u>1.644</u>		.006	FLUORAPATITE	(-)0°	3.26	
1.642	<u>1.649</u>	1.649	.007	CHLORAPATITE	(-)5-10°	3.18	
1.637	<u>1.651</u>		.014	STRONTIUM-APATITE	(-)0°	3.84	
1.644	<u>1.651</u>		.007	HYDROXYL-APATITE	(-)0°	3.21	F 0.16, H ₂ O 1.86%.

Table 10. Apatite group (continued)

Refractive index			Biref	NAME	2V	G	Remarks
α or ϵ	β or ω	γ or ϵ					
1.650	<u>1.654</u>		.004	HYDROXYL-ELLESTADITE	(-)0°	3.02	
1.650	<u>1.655</u>		.005	ELLESTADITE	(-)0°	3.1-3.2	SiO_2 17.3, SO_3 20.7, P_2O_5 3.1%.
1.653	<u>1.658</u>		.005	CHLORAPATITE	(-)0°	3.18	
1.640	<u>1.660</u>		.020	BELOVITE	(-)0°	4.18	
	<u>1.66</u>		wk	FERMORITE	(-)0°	3.52	
1.660	<u>1.666</u>		.006	CHLORAPATITE	(-)0°	3.2	
1.665	<u>1.667</u>	1.667	.002	CHLORAPATITE	(-)10°	3.18	
1.672	<u>1.684</u>		.012	SVABITE	(-)0°	3.54	P_2O_5 12.5, F 1.4%.
1.684	<u>1.687</u>		.003	JOHNBAUMITE	(-)0°	3.68	
1.694	<u>1.696</u>		.002	ALFORSITE	(-)0°	4.80	
1.699	<u>1.703</u>		.004	APATITE-BRITHOLITE	(-)0°	3.83	P_2O_5 17.3, SiO_2 12.9, Ce_2O_3 12.9%.
	<u>1.72</u>		.000	BRITHOLITE	---	3.85	Isotropic.
	<u>1.728</u>	1.730	.002	BRITHOLITE-(Y)	(+)0°	4.25	
	<u>1.75</u>		.000	BRITHOLITE	---	3.85	Isotropic.
1.748	<u>1.752</u>		.004	BRITHOLITE	(-)0°	4.1-4.4	
1.772	<u>1.775</u>	1.777	.005	BRITHOLITE	(-)0-44°	4.0-4.5	
	<u>1.780</u>	1.783	.003	BRITHOLITE-(Y)	(+)0°	4.35	
	<u>1.81</u>		.000	BRITHOLITE	---	4.1	Isotropic.

Table 10. Apatite group (continued)

Refractive index			Biref	NAME	2V	G	Remarks
α or ϵ	β or ω	γ or ϵ					
	<u>1.880</u>	1.884	.004	MORELANDITE	(+)0°	5.33	
	<u>1.948</u>	1.958	.010	HEDYPHANE	(+)0°	5.7	
2.010	<u>2.026</u>		.016	HEDYPHANE	(-)0°	5.82	
2.042	<u>2.050</u>	2.050	.008	PYROMORPHITE	(-)small	7.05	
2.046	<u>2.057</u>		.011	PYROMORPHITE	(-)0°	7.0-7.1	Anom biax. Pleoc, O green, E greenish yellow.
2.12	<u>2.13</u>	2.135	.015	MIMETITE	(-)29°	7.1	
2.12	<u>2.135</u>		.01	MIMETITE	(-)0°	7.1-7.24	
2.20	<u>2.25</u>		.05	VANADINITE	(-)0°	7.0	Arsenatian var.
2.37	<u>2.43</u>		.06	VANADINITE	(-)0°	6.65- 6.98	

Table 11. Aragonite group

Minerals of this group have the general formula RCO_3 , in which R may be Ca, Sr, Ba, and Pb. They are orthorhombic, space group Pmcn, with twinning on {110} very common, giving pseudo-hexagonal aggregates. Hardness is 3 to 4. The minerals are soluble in HCl with effervescence. All are biaxial negative, with very high birefringence and small optic angle.

Refractive index			Biref	NAME and formula	(-)2V	G	Remarks
α	β	γ					
1.520	<u>1.667</u>	1.668	.148	STRONTIANITE $SrCO_3$	7°	3.68	
1.529	<u>1.676</u>	1.677	.148	WITHERITE $BaCO_3$	16°	4.29	
1.530	<u>1.680</u>	1.685	.155	ARAGONITE $CaCO_3$	18°	3.32	
1.525	<u>1.686</u>	1.690	.165	STRONTIANITE $(Sr,Ba,Ca)CO_3$	8°	3.81	BaO 3.3, CaO 1.0%.
1.540	<u>1.695</u>	1.703	.163	ARAGONITE $(Ca,Pb)CO_3$	23°	3.05	Pb 5.2%.
1.804	<u>2.074</u>	2.076	.272	CERUSSITE $PbCO_3$	9°	6.55	

Table 12. Autunite group

The minerals of this group have the general formula $A(UO_2)_2(XO_4)_2 \cdot 8-12H_2O$, in which A may be Ba, Ca, Cu, Fe^{+2} , 1/2 (HAI), Mg, Mn, Na_2 , and (UO_2) , and X may be As, P, and V. The water is of zeolitic character, and the minerals are readily transformed reversibly into the respective lower hydrates of the meta-autunite group (which see).

Autunite	$Ca(UO_2)_2(PO_4)_2 \cdot 10-12H_2O$
Heinrichite	$Ba(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$
Kahlerit	$Fe(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$
Novacekite	$Mg(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$
Sabugalite	$HAI(UO_2)_4(PO_4)_4 \cdot 16H_2O$
Saleeite	$Mg(UO_2)_2(PO_4)_2 \cdot 10H_2O$
Sodium autunite	$Na_2(UO_2)_2(PO_4)_2 \cdot 8H_2O$
Torbernite	$Cu(UO_2)_2(PO_4)_2 \cdot 10-12H_2O$
Troegerite	$(UO_2)(UO_2)_2(AsO_4)_2 \cdot 12H_2O$
Unnamed analogue of Troegerite	
Uranocircite	$Ba(UO_2)(PO_4)_2 \cdot 12H_2O$
Uranospinite	$Ba(UO_2)(AsO_4)_2 \cdot 8H_2O$
Zeunerite	$Cu(UO_2)(AsO_4)_2 \cdot 10-12H_2O$

These minerals are tetragonal, space group I4/mmm. Crystals are typically square tabular in form with yellow to green color. Optically the members have negative sign, mostly uniaxial, but often showing anomalous biaxial character. Most show moderate to weak birefringence with pleochroism in shades of yellow or green. All except those containing Fe or Cu fluoresce under ultraviolet light.

Refractive index			Biref	NAME	(-)2V	G
α or ϵ	β or ω	γ				
1.55	<u>1.567</u>	1.572	.022	URANOSPINITE	62°	3.45
1.543	<u>1.570</u>	1.577	.034	NOVACEKITE	40°	3.23
1.559	<u>1.570</u>	1.574	.015	SALEEITE	65°	3.27

Table 12. Alunite group (continued)

Refractive index			Biref	NAME	(-)2V	G
α or ϵ	β or ω	γ				
1.552	<u>1.574</u>		.022	Unnamed analogue of TROEGERITE	0°	3.76
1.559	<u>1.574</u>		.015	SALEEITE	0°	3.27
1.572	<u>1.575</u>		.003	AUTUNITE	0°	3.2
1.555	<u>1.575</u>	1.577	.022	AUTUNITE	53°	3.1-3.2
1.548	<u>1.578</u>		.030	NOVACEKITE	0°	3.23
1.559	<u>1.578</u>		.019	SODIUM AUTUNITE	0°	3.58
1.564	<u>1.582</u>		.018	SABUGALITE	0°	3.20
1.565	<u>1.582</u>	1.585	.020	SALEEITE	65°	3.27
1.564	<u>1.582</u>	1.584	.020	SABUGALITE	35°	3.20
1.574	<u>1.583</u>	1.588	.014	URANOCIRCITE	60-70°	---
1.560 <i>e</i>	<u>1.586</u>		.026	URANOSPINITE	0°	3.45
1.578	<u>1.586</u>		.008	AUTUNITE	0°	3.2
1.582	<u>1.592</u>		.010	TORBERNITE	0°	3.22
1.566	<u>1.592</u>	1.596	.030	URANOSPINITE	25-35°	3.45
1.581	<u>1.592</u>	1.592	.011	TORBERNITE	small	3.22
1.574	<u>1.605</u>		.031	HEINRICHITE	0°	---
1.585	<u>1.613</u>		.028	ZEUNERITE	0°	3.4
1.582	<u>1.627</u>		.045	TROEGERITE	0°	3.3
---	<u>1.632</u>	1.634	---	KAHLERITE	9-33°	---

Table 13. Barite group

Members of this group are sulfates with the general formula XSO_4 , in which X may be Ba, Sr, and Pb. They are orthorhombic, space group Pnma, with perfect cleavage on {001}. Optically they are biaxial positive, with $\alpha=c$, $\beta=b$, dispersion $r < v$.

Refractive index			Biref	NAME and formula	$(+)^2V$	G
α	β	γ				
1.622	<u>1.624</u>	1.631	.009	CELESTITE $SrSO_4$	51°	3.96
1.636	<u>1.637</u>	1.648	.012	BARITE $BaSO_4$	37°	4.50
1.693	(<u>1.697</u>)	1.718	.025	BARITE $(Ba,Pb)SO_4$	~ 50°	~ 5
1.877	<u>1.883</u>	1.894	.017	ANGLESITE $PbSO_4$	68-75°	6.38

Table 14. Beudantite group

The minerals of this group have the general formula $AB_3(SO_4)(XO_4)(OH)_6$, in which A may be Ba, Ca, Pb, and Sr; B may be Al and Fe⁺³; X may be P and As, with the ratio $SO_4:XO_4$ varying somewhat from 1. The minerals of this group are closely related to the sulfates of the alunite group and the phosphates of the plumbogummite group.

Beudantite	$PbFe_3(AsO_4, SO_4)_2(OH)_6$
Corkite	$PbFe_3(PO_4)(SO_4)(OH)_6$
Hidalgoite	$PbAl_3(AsO_4)(SO_4)(OH)_6$
Hinsdalite	$PbAl_3(PO_4)(SO_4)(OH)_6$
Kemmlitzite	$(Sr,Ce)Al_3(AsO_4)(SO_4, PO_4)(OH)_6$
Svanbergite	$SrAl_3(PO_4)(SO_4)(OH)_6$
Weilerite	$BaAl_3(AsO_4)(SO_4)(OH)_6$
Woodhousite	$CaAl_3(PO_4)(SO_4)(OH)_6$

Table 14. Beudantite group (continued)

They are trigonal, pseudo-cubic, space group $\overline{R}3m$. Distinct to perfect basal cleavage and low birefringence are characteristic, with anomalous biaxial samples not uncommon.

Refractive index			Biref	NAME	2V	G
α or ϵ	β or ω	γ or ϵ				
	<u>1.626</u>	1.640	.014	SVANBERGITE	(+)0°	2.98-3.22
	<u>1.636</u>	1.647	.011	WOODHOUSEITE	(+)0°	3.01
	<u>1.639</u>	1.646	.007	SVANBERGITE	(+)0°	2.98-3.22
	<u>1.662</u>	1.669	.007	WOODHOUSEITE	(+)0°	2.9-3.0
	<u>1.67</u>	1.70	.03	HINSDALITE	(+)0°	3.65
	<u>1.671</u>	1.689	.019	HINSDALITE	(+)0-30°	3.65
	<u>1.688</u>	1.698	.010	WEILERITE	(+)0°	---
	<u>1.688</u>	1.697	.009	HINSDALITE	(+)0°	3.65
	<u>1.701</u>	1.707	.006	KEMMLITZITE	(+)0°	3.63
	<u>1.71</u>	---	.005	HIDALGOITE	(+)0°	3.96
	<u>1.730</u>	1.735	.005	HIDALGOITE	(+)0°	3.95
1.909	<u>1.916</u>		.007	BEUDANTITE	(-)0°	4.0-4.3
---	<u>1.93</u>		wk	CORKITE	(-)0°	4.2-4.3
---	<u>1.93</u>	---	wk	CORKITE	(-)med	4.20
1.943	<u>1.957</u>		.014	BEUDANTITE	(-)0°	4.0-4.3
1.94	<u>1.96</u>	1.96	.02	BEUDANTITE	(-)6-50°	4.1

Table 15. Calcite group

The calcite group minerals have the general formula RCO_3 , in which R may be Ca, Cd, Co, Fe^{+2} , Mg, Mn, Ni, and Zn. Complete solid solution exists between Ca and Cd, Ca and Mn, Mn and Fe, and Mg and Fe, probably also between (Mg,Fe^{+2}) and (Zn,Co) , although not many representatives are known.

Calcite	$CaCO_3$
Gaspeite	$(Ni,Mg)CO_3$
Magnesite	$(Mg,Fe)CO_3$
Otavite	$(Cd,Ca)CO_3$
Rhodochrosite	$(Mn,Ca)CO_3$
Siderite	$(Fe,Mg)CO_3$
Smithsonite	$(Zn,Mg)CO_3$
Sphaerocobaltite	$CoCO_3$

These minerals are trigonal, space group $\bar{R}\bar{3}c$, with perfect rhombohedral cleavage, hardness 3 to 5. Optically they are uniaxial negative with very high birefringence (compare the related dolomite group).

Refractive index		Biref (-)	NAME	G	Weight %			
ω	ϵ				MnO	FeO	MgO	Other
1.658	1.486	.172	CALCITE	2.71	---	---	---	end member
1.672	1.501	.171	CALCITE	2.82	4.2	2.1	1.3	
1.700	1.509	.191	MAGNESITE	2.96	---	---	47.8	end member
1.710	1.523	.189	CALCITE	---	---	---	---	$CdO\ 20.1$
1.711	1.519	.192	MAGNESITE	3.15	---	---	---	$NiO\ 13.55$
1.713	1.519	.194	CALCITE	3.02	20.1	---	---	$ZnO\ 0.4$
1.726	1.528	.198	MAGNESITE	3.10	---	9.5	---	
1.731	(1.55)	.18	RHODOCHROSITE	3.05	33.4	2.7	2.2	$CaO\ 21.0$
1.753	1.560	.193	RHODOCHROSITE	3.3	44.3	0.8	2.3	$CaO\ 13.4$

Table 15. Calcite group (continued)

Refractive index		Biref (-)	NAME	G	Weight %			
w	e				MnO	FeO	MgO	Other
1.788	1.570	.218	MAGNESITE	3.43	---	(35.9)	(20.1)	Mg/Fe = 1
1.803	1.584	.219	RHODOCHROSITE	3.65	---	3.8	0.6	CaO 1.3
1.81	1.58	.23	GASPEITE	3.6	---	4.3	14.8	NiO 36.8
1.815	1.601	.214	SMITHSONITE	---	---	---	7.6	ZnO 53.2 CaO 1.1
1.816	1.592	.224	SIDERITE	3.59	5.9	---	11.8	CaO 1.4
1.816	1.597	.219	RHODOCHROSITE	3.70	61.7	---	---	end member
1.830	1.605	.225	OTAVITE	~4.8	---	---	---	CdO 67.8
1.836	(1.61)	.226	RHODOCHROSITE	3.76	29.8	13.9	---	{ZnO 14.9 CaO 3.1
1.840	1.615	.225	RHODOCHROSITE	3.72	35.4	26.1	---	
1.84	(1.62)	.22	GASPEITE	3.7	---	0.1	3.3	NiO 55.5
1.849	1.615	.234	SIDERITE	3.81	9.5	47.9	3.2	
1.850	1.623	.227	SMITHSONITE	4.42	0.5	0.4	---	
1.855	1.60	.225	SPHAERO-COBALTITE	4.1	---	---	---	
1.860	1.625	.235	SIDERITE	3.92	14.7	36.4	0.3	ZnO 11.55
1.875	1.633	.242	SIDERITE	3.89	---	62.0	---	end member

Table 16. Cancrinite group

The cancrinite group consists of hexagonal silicates of Ca, Na, and K, also containing carbonate (cancrinite), sulfate (vishnevite), or chloride. Davyne, microsommite, and afghanite are polytypes with nearly the same composition but with different unit cells. They are characterized by low to moderate birefringence and perfect prismatic cleavage.

Afghanite	$(\text{Na}, \text{Ca}, \text{K})_8(\text{Si}, \text{Al})_{12}\text{O}_{24}(\text{Cl}, \text{SO}_4, \text{CO}_3)_3 \cdot \text{H}_2\text{O}$
Cancrinite	$(\text{Na}, \text{Ca})_{7-8}(\text{Si}_6\text{Al}_6)\text{O}_{24}(\text{CO}_3, \text{SO}_4)_{1-2} \cdot 2\text{H}_2\text{O}$
Davyne	$(\text{Na}, \text{Ca}, \text{K})_{7-8}(\text{Si}, \text{Al})_{12}\text{O}_{24}(\text{Cl}, \text{SO}_4, \text{CO}_3)_{2-3}$
Franzinite	$(\text{Na}, \text{Ca})_7(\text{Si}, \text{Al})_{12}\text{O}_{24}(\text{SO}_4, \text{CO}_3)_3 \cdot \text{H}_2\text{O}$
Giuseppettite	$(\text{Na}, \text{K}, \text{Ca})_{7-8}(\text{Si}, \text{Al})_{12}\text{O}_{24}(\text{SO}_4, \text{Cl})_{1-2}$
Liottite	$(\text{Ca}, \text{Na}, \text{K})_8(\text{Si}, \text{Al})_{12}\text{O}_{24}(\text{SO}_4, \text{CO}_3, \text{OH}, \text{Cl})_4 \cdot \text{H}_2\text{O}$
Microsommite	$(\text{Na}, \text{Ca}, \text{K})_{7-8}(\text{Si}, \text{Al})_{12}\text{O}_{24}(\text{Cl}, \text{SO}_4, \text{CO}_3)_{2-3}$
Sacrofanite	$(\text{Na}, \text{Ca}, \text{K})_9(\text{Si}, \text{Al})_{12}\text{O}_{24}[(\text{OH})_2, \text{SO}_4, \text{CO}_3, \text{Cl}_2]_3 \cdot x\text{H}_2\text{O}$
Vishnevite	$(\text{Na}, \text{Ca}, \text{K})_{6-7}\text{Si}_6\text{Al}_6\text{O}_{24}(\text{SO}_4, \text{CO}_3)_{1-2} \cdot \text{H}_2\text{O}$

Refractive index		Biref	NAME	Weight %					
ω	ϵ			SO_3	CO_2	Cl	Na_2O	K_2O	CaO
1.491	1.507	(+).016	GIUSEPPETTITE	---	---	---	---	---	---
1.499	1.493	(-).006	VISHNEVITE	4.7	2.2	0.4	20.8	1.0	1.1
1.505	1.486	(-).019	SACROFANITE	---	---	---	---	---	---
1.510	1.512	(+).002	FRANZINITE	10.7	1.5	0.4	11.5	4.2	12.1
1.515	1.496	(-).019	CANCRI NITE	1.4	7.0	0.4	18.7	0.6	4.8
1.518	1.521	(+).003	DAVYNE	---	---	---	---	---	---
1.521	1.529	(+).008	MICROSOMMITE	---	---	---	---	---	---
1.523	1.529	(+).006	AFGHANITE	8.5	0.4	4.6	12.6	2.7	16.5
1.528	1.533	(+).005	AFGHANITE	9.7	0.8	4.5	12.4	0.9	12.3

Table 16. Cancrinite group (continued)

Refractive index		Biref	NAME	Weight %					
ω	ϵ			SO_3	CO_2	Cl	Na_2O	K_2O	CaO
1.528	1.503	(-).025	CANCIRNITE	0.2	6.2	0.1	9.8	1.2	11.7
1.530	1.528	(-).002	LIOTTITE	8.7	2.1	2.6	8.0	5.0	16.7

Table 17. Chalcanthite group

The minerals of this group are hydrous sulfates of the general formula $\text{ASO}_4 \cdot 5\text{H}_2\text{O}$, in which A may be Cu, Fe^{+2} , Mg, Mn, and Zn. They are triclinic, optically biaxial negative, with moderate $2V$, $r < v$. Hardness and specific gravity are low. All are soluble in water and on heating lose water but do not fuse.

Refractive index			Biref	NAME and formula	$(-)\text{2V}$	Color	Remarks
α	β	γ					
1.482	<u>1.492</u>	1.493	.011	PENTAHYDRITE $\text{MgSO}_4 \cdot 5\text{H}_2\text{O}$	45°	Col s	
1.498	<u>1.510</u>	1.517	.019	JOKOKUITE $\text{MnSO}_4 \cdot 5\text{H}_2\text{O}$	70-80°	Pale pink	
1.495	<u>1.512</u>	1.518	.023	PENTAHYDRITE $(\text{Mg}, \text{Cu}, \text{Zn})\text{SO}_4 \cdot 5\text{H}_2\text{O}$	55°	Blue	$\text{CuO } 9.0,$ $\text{ZnO } 5.6,$ $\text{FeO } 1.4\%$.
1.513	(<u>1.526</u>)	1.534	.021	SIDEROTIL $(\text{Fe}, \text{Cu})\text{SO}_4 \cdot 5\text{H}_2\text{O}$	60°	Pale green to blue	
1.501	<u>1.537</u>	1.539	.038	CHALCANTHITE $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$	56°	Dark blue	

Table 18. Copiapite group

Minerals of this group are sulfates having the general formula $AFe_4^{+3}(SO_4)_6 \cdot 20H_2O$ or $BFe_4^{+3}(SO_4)_6(OH) \cdot 20H_2O$, in which A may be Ca, Cu, Fe⁺², Mg, and Zn; B may be Al and Fe⁺³. All are triclinic with perfect {010} cleavage, low hardness and density. They are optically biaxial positive with moderate 2V, $r > v$ strong, X near b, and pleochroic in yellows or greens. Optical data are not available for calcio-copiapite. All are dissolved by acids.

Refractive index			Biref	NAME and formula	(+)2V	Pleochroism
α	β	γ				
1.507	<u>1.529</u>	1.573	.067	MAGNESIOCOPIAPITE $MgFe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	73°	X yellow-green, Y yellow, Z yellow to yellow-green.
1.520	<u>1.535</u>	1.578	.058	COPIAPITE $Fe^{+2}Fe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	70°	X, Y pale yellow, Z yellow.
1.525	<u>1.540</u>	1.590	.065	ALUMINOCOPIAPITE $AlFe_4^{+3}(SO_4)_6(OH) \cdot 20H_2O$	med	Yellows.
1.525	<u>1.545</u>	1.595	.070	MAGNESIOCOPIAPITE $(Mg, Fe^{+2})Fe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	66°	X, Y very pale yellow, Z deep yellow.
1.530	<u>1.550</u>	1.592	.062	FERRICOPIAPITE $Fe^{+3}Fe_4^{+3}(SO_4)_6(OH) \cdot 20H_2O$	69°	Yellows.
1.534	<u>1.554</u>	1.586	.052	ZINCOCOPIAPITE $ZnFe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	(78 \pm 5°)	Strong in shades of yellow.
1.558	<u>1.575</u>	1.620	.062	CUPROCOPIAPITE $CuFe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	63°	Greens, abs X and Z > Y.

Table 19. Crandallite (Plumbogummite) group

The minerals of this group have the general formula $AB_3(XO_4)_2(OH)_5 \cdot H_2O$ or $AB_3(XO_4)_2(OH)_6$, in which A may be Ba, Ca, Ce, Nd, Pb, Sr, and Th; B may be Al and Fe^{+3} ; X may be P and As. Minerals of this group are closely related to those of the alunite and beudantite groups.

Arsenocrandallite	$(Ca, Sr)Al_3[(As, P)O_4]_2(OH)_5 \cdot H_2O$
Crandallite	$CaAl_3(PO_4)_2(OH)_5 \cdot H_2O$
Dussertite	$BaFe^{+3}(AsO_4)_2(OH)_5 \cdot H_2O$
Eylettersite	$(Th, Pb)Al_3(PO_4)_2(OH)_6$
Florencite	$CeAl_3(PO_4)_2(OH)_6$
Gorceixite	$(Ba, Ca, Ce)Al_3(PO_4)_2(OH)_5 \cdot H_2O$
Goyazite	$SrAl_3(PO_4)_2(OH)_5 \cdot H_2O$
Lusungite	$(Sr, Pb)Fe^{+3}(PO_4)_2(OH)_5 \cdot H_2O (?)$
Philipsbornite	$PbAl_3(AsO_4)_2(OH)_5 \cdot H_2O$
Plumbogummite	$PbAl_3(PO_4)_2(OH)_5 \cdot H_2O$
Unnamed	$PbFe^{+3}(AsO_4)_2(OH)_5 \cdot H_2O$
Zairite	$Bi(Fe^{+3}, Al)_3(PO_4)_2(OH)_6$

These minerals are trigonal, pseudocubic, space group R3m or $\bar{R}3m$. Distinct to perfect basal cleavage and low birefringence (except for Lusungite) are characteristic. An unnamed triclinic (?) dimorph of crandallite with $\beta=1.607$ (see table 7) is also included in this group. Optical data on waylandite have not been recorded, other than that it is uniaxial negative.

Refractive index		Biref	NAME	H	G	Remarks
ω	ϵ					
1.610	1.620	(+).010	GORCEIXITE	6	3.1-3.3	
1.613	1.622	(+).009	CRANDALLITE	5	2.8	

Table 19. Crandallite (Plumbogummite) group (continued)

Refractive index		Biref	NAME	H	G	Remarks
ω	ϵ					
1.620	1.630	(+).010	GOYAZITE	4.5-5	3.15-3.26	Anom biax. Pleoc, O red-brown, E yellow.
1.623	1.634	(+).011	CRANDALLITE	5	2.8	
1.625	---	low	ARSENO-CRANDALLITE	5.5	3.25	
1.625	---	(+) low	GORCEIXITE	6	3.09-3.32	
1.635	---	(-) low	EYLETTERSITE	---	3.4	Fluor wk greenish-yellow in UV.
1.640	1.651	(+).011	GOYAZITE	4.5-5	3.15	Anom biax, 2V 0-20°.
1.653	1.675	(+).022	PLUMBOGUMMITE	4-5	4.0	
1.653	1.661	(+).008	FLORENCITE	5	3.46	
1.678	1.684	(+).006	FLORENCITE	5-6	3.46	
1.680	1.698	(+).018	PLUMBOGUMMITE	4.5	4.0-4.9	
1.695	1.705	(+).010	FLORENCITE	6	3.69	
1.713	1.719	(+).006	FLORENCITE	6	3.4-3.6	
1.77	---	(+).03-.04	LUSUNGITE	---	---	
1.790	---	Isotropic	PHILIPSBORNITE	4.5	4.33	
1.82- 1.83	---	---	ZAIRITE	---	4.37	
1.855	---	(+).03-.04	LUSUNGITE	---	---	
1.870	1.845	(-).025	DUSSERTITE	3.5	3.75	
1.99	---	(-)mod	Unnamed	---	---	Anom biax, 2V 0-50°.

Table 20. Dolomite group

The minerals of this group have the general formula $AB(CO_3)_2$, in which A may be Ca and Ba, and B may be Mg, Fe^{+2} , Mn^{+2} , and Zn. Dolomite, Ankerite, and Kutnohorite probably form a complete series of solid solutions.

Ankerite	$CaFe(CO_3)_2$
Dolomite	$CaMg(CO_3)_2$
Kutnohorite	$CaMn(CO_3)_2$
Minrecordite	$CaZn(CO_3)_2$
Norsethite	$BaMg(CO_3)_2$

The minerals are trigonal, space group R3. All are optically uniaxial negative and, like the members of the calcite group, are characterized by very high birefringence and perfect rhombohedral cleavage. All are infusible and difficultly soluble in dilute acids (compare the related calcite group).

Refractive index		Biref	NAME	G	Weight %		
ω	ϵ				FeO	MnO	Others
1.680	1.501	.179	DOLOMITE	2.865	0.5	---	
1.694	1.510	.184	DOLOMITE	2.92	8.4	1.1	SrO 0.4
1.694	1.519	.175	NORSETHITE	3.74- 3.84	---	---	
1.700	1.521	.179	DOLOMITE	2.95	2.2	---	ZnO 8.3, PbO 1.1
1.710	1.519	.191	KUTNOHORITE	3.00	1.3	17.6	MgO 10.0
1.711	1.520	.191	DOLOMITE	3.01	12.6	1.2	
1.712	1.512	.200	NORSETHITE	3.88	3.6	0.8	
1.734	1.542	.192	MINRECORDITE	3.32			
1.740	1.547	.193	KUTNOHORITE	3.14	0.8	31.8	ZnO 1.4, MgO 0.5
1.741	1.536	.205	ANKERITE	3.12	24.0	1.6	
1.750	1.550	.200	MINRECORDITE	3.45			

Table 21. Epidote group

The minerals of the epidote group are silicates having Al₁₀⁶⁻ and Al₁₀(OH)₂-chains linked by independent SiO₄- and Si₂O₇-groups. The general formula is A₂B₃(Si₂O₇)(SiO₄)₂(OH) or A₂B₃(Si₃O₁₂)(OH), in which A may be Ca, Ce, La, Y, Pb, and Sr, and B may be Al, Fe⁺³, and V.

Allanite	(Ca,Ce,La,Y) ₂ (Al,Fe ⁺³ ,Fe ⁺²) ₃ Si ₃ O ₁₂ (OH)
Clinozoisite	Ca ₂ Al ₃ (Si ₃ O ₁₂)(OH)
Epidote	Ca ₂ (Al,Fe) ₃ (Si ₃ O ₁₂)(OH)
Hancockite	(Pb,Ca,Sr) ₂ (Al,Fe) ₃ (Si ₃ O ₁₂)(OH)
Mukhinite	Ca ₂ (Al,V) ₃ (Si ₃ O ₁₂)(OH)
Piemontite	Ca ₂ (Al,Mn ⁺³ ,Fe ⁺³) ₃ (Si ₃ O ₁₂)(OH)
Zoisite	Ca ₂ Al ₃ (Si ₃ O ₁₂)(OH)

All are monoclinic, except orthorhombic zoisite, and all have perfect {001} cleavage, except allanite and zoisite. They are optically biaxial with moderate to large optic angle, most with strong dispersion and X = or ~ c, Y=b. Refractive indices and birefringence increase with increase in iron or manganese content. Pleochroism is marked in piemontite, mukhinite, and in the higher-iron epidotes.

Many allanites are isotropic metamict because of the presence of U and Th. In this state, refractive index and density are highly variable: n 1.61 to 1.75; G 2.96 to 3.74. Data for such metamict varieties are not listed here but are included in the table for isotropic minerals (table 3).

Refractive index			Biref	NAME	2V	Orientation	Remarks
α	β	γ					
1.691	<u>1.692</u>	1.700	.009	ZOISITE	(+)54° r > v str	X = <u>c</u>	Abn interf colors. Pleoc u wk, X violet, Y deep blue, Z yellow-green.
1.700	<u>1.703</u>	1.708	.008	ZOISITE	(+)71° r < v or r > v	X = <u>c</u>	Abn interf colors. Fe ₂ O ₃ 3.1, FeO 1.3, MnO 0.3%.
1.700	<u>1.703</u>	1.706	.006	CLINO- ZOISITE	~ 90° r < v str	X ~ <u>c</u>	Abn interf colors. Fe ₂ O ₃ 3.2, FeO 0.6%.

Table 21. Epidote group (continued)

Refractive index			Biref	NAME	2V	Orientation	Remarks
α	β	γ					
1.704	<u>1.708</u>	1.714	.006	ALLANITE	(+)70-80° r > v	X:c = 5-40°	U neg. Slowly gel in HCl.
1.706	<u>1.711</u>	1.714	.008	ALLANITE	(-)61°		Slowly gel in HCl. Fe ₂ O ₃ 4.9, FeO 9.1, ThO ₂ 1.7%.
1.715	<u>1.717</u>	1.721	.006	CLINOZOISITE	~ 90° r < v str	X ~ c	Abn interf colors. Fe ₂ O ₃ 3.0, FeO 0.8%.
1.713	<u>1.720</u>	1.727	.014	EPIDOTE	(-)89° r > v str		Fe ₂ O ₃ 6.6, FeO 0.4%.
1.723	<u>1.733</u>	1.755	.032	MUKHINITE	(+)88° (69+8°)	Z:a = 32°	Pleoc str, X
1.721	<u>1.736</u>	1.740	.019	EPIDOTE	(-)78° (54+14°) r > v str	X ~ c	Fe ₂ O ₃ 9.5, FeO 0.8%.
1.721	<u>1.738</u>	1.742	.021	ALLANITE	(-)Large (51+13°) r > v str		Gel with acids. Fe ₂ O ₃ 17.0, MnO 0.6, ThO ₂ 0.9%.
1.738	<u>1.744</u>	1.753	.015	ALLANITE	(+)70-80° r > v	X ~ c = 5-40°	Pleoc in brown and red.
1.732	<u>1.750</u>	1.778	.046	PIEMONTITE	(+)Large	X:c = -5°	Pleoc str, X buff, Y deep lavender, Z pink. Fe ₂ O ₃ 3.8, Mn ₂ O ₃ 11.8%.

Table 21. Epidote group (continued)

Refractive index			Biref	NAME	2V	Orientation	Remarks
α	β	γ					
1.733	<u>1.750</u>	1.762	.029	PIEMONTITE	(-)78° r > v str		Pleoc, X yellow, Y pale amethyst, Z deep purplish red. Fe_2O_3 11.3, Mn_2O_3 1.0%.
1.729	<u>1.754</u>	1.776	.047	EPIDOTE	(-)73° (85°+5°) r > v str	X:c = 8°	Pleoc, X pale yellow, Y pale greenish yellow, Z greenish yellow. Fe_2O_3 14.0, FeO 1.2%.
1.740	<u>1.755</u>	1.760	.020	ALLANITE	(-)Large r > v str		Gel with acids. Fe_2O_3 3.2, FeO 10.7, MnO 1.9, TiO_2 1.7, ThO_2 0.9%.
1.763	<u>1.768</u>	1.788	.025	ALLANITE	(+)57° r > v	X:c = 5-40°	Slowly gel with HCl. Fe_2O_3 3.8, FeO 10.4%.
1.740	<u>1.768</u>	1.787	.047	EPIDOTE	(-)74° r > v str	X:c = 13°	Fe_2O_3 17.9, FeO 0.4, MnO 0.5%.
1.754	<u>1.772</u>	1.795	.041	PIEMONTITE	(+)85° r < v	X:c = -6°	Pleoc str, X yellow, Y violet, Z red.
1.746	<u>1.776</u>	1.793	.047	PIEMONTITE	(-)87° r > v str		Pleoc, X yellow, Y pink, Z deep pink. Fe_2O_3 11.9, Mn_2O_3 6.8%.
1.751	<u>1.784</u>	1.797	.046	EPIDOTE	(-)64° r > v str		Fe_2O_3 17.2, FeO 0.5%.

Table 21. Epidote group (continued)

Refractive index			Biref	NAME	2V	Orientation	Remarks
α	β	γ					
1.769	<u>1.784</u>	1.791	.022	ALLANITE	(-)60° r > v str		Slowly gel with HCl.
1.756	<u>1.788</u>	1.829	.073	PIEMONTITE	(+)86° r < v	X:c = -9°	Pleoc str, X yellow, Y violet, Z red.
1.788	<u>1.825</u>	1.830	.042	HANCOCKITE	38° r > v	$\gamma = \underline{b}$	Abs Z > X.

Table 22. Feldspar group

Minerals of the feldspar group are framework silicates of the general formula $XAlSi_3O_8$ or $Y(Z, Si)Si_2O_8$, in which X may be K and Na, less commonly NH_4 ; Y may be Ca, less commonly Ba and Sr; Z may be Al, less commonly B and Fe^{+3} . Two subgroups of wide occurrence and importance in the classification of igneous rocks are the alkali feldspars and the plagioclases. The barium, strontium, boron, and NH_4 feldspars occur only infrequently or rarely.

Alkali feldspars	$(K, Na)AlSi_3O_8$
Microcline	
Orthoclase	
Anorthoclase	
Sanidine-High Albite	
Plagioclase	$(Na, Ca)Al(Al, Si)Si_2O_8$
Albite	An 0-10
Oligoclase	An 10-30
Andesine	An 30-50
Labradorite	An 50-70

Table 22. Feldspar group (continued)

Bytownite	An 70-90
Anorthite	An 90-100
Barium feldspars	
Banalsite	$\text{BaNa}_2\text{Al}_4\text{Si}_4\text{O}_{16}$
Celsian	$(\text{Ba}, \text{K})\text{Al}(\text{Al}, \text{Si})\text{Si}_2\text{O}_8$
Hyalophane	$(\text{K}, \text{Ba})\text{Al}(\text{Al}, \text{Si})\text{Si}_2\text{O}_8$
Buddingtonite	$(\text{NH}_4)\text{AlSi}_3\text{O}_8$
Reedmergnerite	NaBSi_3O_8
Slawsonite	$(\text{Sr}, \text{Ca})\text{Al}_2\text{Si}_2\text{O}_8$

Minerals of the feldspar group may also be divided on the basis of thermal history. Alkali feldspars that have been rapidly quenched from magmatic temperatures form the sanidine-high albite series. Those that have cooled slowly (annealed) under plutonic conditions or have formed at low temperatures include microcline and orthoclase. Anorthoclases may be an intermediate type. Similarly, the plagioclase series may be divided into quenched types (labeled "volcanic" in tables 6 and 7), annealed types ("plutonic"), and intermediate types. If the aim is to infer thermal history, advantage may be taken of the differences in lattice structure seen in optical orientation or in X-ray. If, on the other hand, the aim is only to infer anorthite content, a convenient approach is to determine the α -refractive index on cleavage flakes (Tsuboi, 1923; Morse, 1968). Useful mineralogical and petrographic discussions of the feldspars are given by Deer, Howie, and Zussman (1963, v. 4), Tröger (1967, pt. 2), and Smith (1974).

The feldspars are monoclinic or triclinic. The space groups are microcline and anorthoclase, C1; orthoclase and sanidine-high albite, C2/m; low albite, C1; anorthite, P1; and celsian, I2/c. They have good to perfect {010} and {001} cleavages, and hardness 5.5 to 6. Simple twins are common, with polysynthetic twinning especially prominent in the plutonic feldspars, less so in the volcanic plagioclases. Except for those high in barium and calcium, the feldspars are essentially unaffected by HCl. All are dissolved by HF.

Table 22. Feldspar group (continued)

Refractive index			Biref	NAME and composition	2V	X' : a on		Poly tw	System
α	β	γ				{001}	{010}		
1.519	<u>1.522</u>	1.524	.005	ANORTHOCLADE Or 21, Ab 73, An 6 (?)	(-)51° r > v wk	2°	8°	fine grating	TCL
1.519	<u>1.523</u>	1.525	.006	MICROCLINE Or 85, Ab 14, An 1	(-)76°	10-15°	7.5°	coarse grating	TCL
1.519	<u>1.524</u>	1.525	.006	ORTHOCLASE Or 84, Ab 14, An 2	(-)46°	0°	6°	---	MCL
1.520	<u>1.525</u>	1.525	.005	SANIDINE Or 78, Ab 22	(-)24°	0°	5-8°	---	MCL
1.523	<u>1.529</u>	1.530	.007	SANIDINE Or 41, Ab 58, An 1	(-)33°	0°	---	---	MCL
1.530	<u>1.531</u>	1.534	.004	BUDDINGTONITE $(\text{NH}_4)\text{AlSi}_3\text{O}_8$	(+)	0°	4°	---	MCL
1.525	<u>1.531</u>	1.532	.007	ANORTHOCLADE Or 22, Ab 76, An 2	(-)47°	---	---	fine grating	TCL
1.528	<u>1.532</u>	1.539	.011	ALBITE, plutonic plagioclase, An 0	(+)74° r < v wk	3°	20°	{010}	TCL
1.527	<u>1.534</u>	1.535	.008	ALBITE, volcanic plagioclase, An 0	(-)50°	3°	23°	less common	TCL
1.536	<u>1.539</u>	1.541	.005	ANORTHOCLADE Or 19, Ab 63, An 18	(-)62°	---	3°	fine grating	TCL
1.536	<u>1.541</u>	1.546	.010	OLIGOCLASE, plutonic plagioclase, An 18	~ 90°	1°	7°	{010}	TCL

Table 22. Feldspar group (continued)

Refractive index			Biref	NAME and composition	2V	<u>X'</u> : <u>a</u> on		Poly tw	System
α	β	γ				{001}	{010}		
1.538	<u>1.543</u>	1.546	.008	OLIGOCLASE, volcanic plagioclase, An 20	(-)65°	3°	4°	less common	TCL
1.542	<u>1.544</u>	1.546	.004	HYALOPHANE Cn 37	(-)75°	0°	-18°	---	MCL
1.541	<u>1.546</u>	1.550	.009	OLIGOCLASE, plutonic plagioclase, An 25	(-)78°	1°	2°	{010}	TCL
1.546	<u>1.550</u>	1.554	.008	ANDESINE, plutonic plagioclase, An 33	~ 90°	0°	-5°	{010}	TCL
1.549	<u>1.552</u>	1.556	.00	ANDESINE, volcanic plagioclase, An 36	~ 90°	2°	-4°	less common	TCL
1.555	<u>1.558</u>	1.562	.007	LABRADORITE, plutonic plagioclase, An 50	(+)78°	-6°	-17°	{010}	TCL
1.558	<u>1.562</u>	1.566	.008	LABRADORITE, volcanic plagioclase, An 58	(+)76°	-15°	-26°	less common	TCL
1.570	<u>1.571</u>	1.578	.008	BANAL SITE	41°				ORTH
1.554	<u>1.565</u>	1.573	.019	REEDMERGNERITE	(-)80°	---	---	---	TCL
1.564	<u>1.568</u>	1.572	.008	CELSIAN, var Kasoite, Cn 35	(-)80°	10-13°	2-3°	untw	MCL

Table 22. Feldspar group (continued)

Refractive index			Biref	NAME and composition	2V	X' : <u>a</u> on		Poly tw	System
α	β	γ				{001}	{010}		
1.563	<u>1.568</u>	1.573	.010	BYTOWNITE, plutonic plagioclase, An 70	$\sim 90^\circ$	-16°	-29°	{010}, peri- cline	TCL
1.565	<u>1.569</u>	1.574	.009	BYTOWNITE, volcanic plagioclase, An 75	$\sim 90^\circ$	-23°	-33°	less common	TCL
1.571	<u>1.577</u>	1.583	.012	BYTOWNITE, plutonic plagioclase, An 88	(-)79°	-30°	-37°	{010}, peri- cline	TCL
1.572	<u>1.577</u>	1.583	.011	BYTOWNITE, volcanic plagioclase, An 90	(-)81°	-34°	-36°	less common	TCL
1.573	<u>1.581</u>	1.585	.012	SLAWSONITE (Sr,Ca)Al ₂ Si ₂ O ₈	(+)82° $r < v$	Z = <u>b</u> X: <u>c</u> = 11°		---	MCL
1.580	<u>1.583</u>	1.586	.006	CELSIAN, Cn 85	(-)74°	Z:c on {100} = 33°		untw	MCL
1.575	<u>1.584</u>	1.589	.014	ANORTHITE, plagioclase, An 100	(-)77°	-40°	-39°	{010}, peri- cline	TCL
1.583	<u>1.589</u>	1.593	.010	CELSIAN, Cn 94	(-)88°	Z:c on {100} = 26°		untw	MCL
1.593	<u>1.599</u>	1.608	.015	CELSIAN, Cn 94	(+)85°	Y = <u>b</u>		---	MCL

Table 23. Garnet group

The minerals of this group are orthosilicates having the general formula $X_3Y_2(SiO_4)_3$, in which X may be Fe^{+2} , Mn^{+2} , and Mg; Y may be Al, Fe^{+3} , Cr, less commonly V, Ti, Mn^{+3} , and Zr. Substitution is common among the end members listed below, with complete solid solution between some.

Almandine	(Al)	$Fe_3^{+2}Al_2(SiO_4)_3$
Andradite	(An)	$Ca_3Fe_2^{+3}(SiO_4)_3$
Melanite		$Ca_3(Fe^{+3},Ti)_2(SiO_4)_3$
Schorlomite		$Ca_3(Fe^{+3},Ti)_2[(Si,Ti)O_4]_3$
Calderite	(Ca)	$Mn_3Fe_2^{+3}(SiO_4)_3$
Goldmanite	(Go)	$Ca_3V_2(SiO_4)_3$
Grossular	(Gr)	$Ca_3Al_2(SiO_4)_3$
Henrictierite		$Ca_3(Mn^{+3},Al)_2(SiO_4)_2(OH)_4$
Hydrogrossular		$Ca_3Al_2(SiO_4)_{3-x}(OH)_{4x}$
Kimzeyite		$Ca_3(Zr,Ti)_2Al_2SiO_12$
Knorringite	(Kn)	$Mg_3Cr_2(SiO_4)_3$
Pyrope	(Py)	$Mg_3Al_2(SiO_4)_3$
Spessartine	(Sp)	$Mn_3Al_2(SiO_4)_3$
Uvarovite	(Uv)	$Ca_3Cr_2(SiO_4)_3$

The garnets are cubic, space group Ia3d, except for henrictierite, which is tetragonal. Their fracture is subconchoidal, hardness 6 to 7.5, specific gravity 3.35 to 4.32; and, except for hydrogrossular and henrictierite, they are nearly insoluble in HCl and only slowly attacked by cold HF.

Optically garnets have very high refractive index and are usually isotropic, but some, especially the andradites, grossulars, and uvarovites, may show low birefringence with complex grating twinning. Because of the many possible substitutions in the solid solution series, refractive index alone is not a sufficient basis on which to infer chemical composition but is helpful together with density and the length of a_0 of the unit cell. The overlaps in the following tabulation dictate caution in deducing composition from physical properties. If a garnet is known to be in the series Gr-An, however, the composition may be inferred from refractive index, specific gravity, or the length of the cell edge. In the Al-Sp-Py series, refractive index and Mn-content will usually give an approximate determination of the composition.

Relations between composition and physical properties were discussed by Ford (1915), Fleischer (1937), and Winchell (1958), and properties of synthesized

Table 23. Garnet group (continued)

garnets are described by Skinner (1956), whose data are included in the tabulation below. Graphs showing these relations are given by Sriramadas (1957). Types of occurrence are discussed by Wright (1938), and Deer, Howie, and Zussman (1962, v. 1, p. 77-112).

Refractive index	NAME	G	Cell edge a_0 (Å)	Composition					Remarks
				Al	An	Gr	Py	Sp	
1.670	HYDROGROSSULAR	---	---	---	---	---	---	---	H ₂ O 9.3%.
1.702	HYDROGROSSULAR	3.35	---	---	---	---	---	---	H ₂ O 4.6%.
1.714	PYROPE	3.582	11.459	---	---	---	100	---	Synth.
1.734	GROSSULAR	3.594	11.851	---	---	100	---	---	Synth.
1.737	GROSSULAR	---	---	27	---	53	20	---	
1.741	PYROPE-ALMANDINE	3.72	---	20	2	11	66	0.6	
1.752	GROSSULAR-ANDRADITE	3.57	---	---	13	85	1	---	
1.766	PYROPE-ALMANDINE	3.88	11.580	41	6	8	43	1.5	
1.773	GROSSULAR-SPESSARTINE	3.80	---	11	7	56	---	25	
1.782	ALMANDINE-PYROPE	4.02	---	55	---	14	27	3	
1.796	GROSSULAR-UVAROVITE	3.68	---	---	8	52	1	---	Uv 39.
1.797	ALMANDINE	4.10	---	60	---	7	18	15	
1.800	SPESSARTINE	4.190	11.621	---	---	---	---	100	Synth.
1.801	GROSSULAR-ANDRADITE	3.71	11.94	4	42	52	---	2	
1.803	KNORRINGITE	3.85	---	14	3	19	10	1	Kn 53.
1.805	SPESSARTINE	4.20	---	17	---	---	2	80	
1.808	ALMANDINE	4.15	---	74	---	11	11	5	

Table 23. Garnet group (continued)

Refractive index	NAME	G	Cell edge a_0 (Å)	Composition					Remarks
				Al	An	Gr	Py	Sp	
1.814	SPESSARTINE-ALMANDINE	4.23	---	45	---	1	1	53	
1.818	ALMANDINE-SPESSARTINE	4.25	---	74	---	1	5	20	
1.821	GOLDMANITE	3.74	12.011	---	17	20	3	---	Go 60.
1.825	UVAROVITE-GROSSULAR	3.81	---	---	1	26	1	---	Uv 72.
1.827	ANDRADITE-GROSSULAR	3.77	---	57	5	32	3	3	
1.830	ALMANDINE	4.318	11.526	100	---	---	---	---	Synth.
1.834	GOLDMANITE	3.765	---	---	---	---	---	---	Go 100, synth.
1.855	UVAROVITE	3.77	---	1	---	6	2	---	Uv 91.
1.855	GOLDMANITE	3.91	11.974	---	---	---	---	---	V_2O_3 24.9%.
1.863	ANDRADITE	3.72	11.996	5	79	12	---	4	
1.875	CALDERITE-GROSSULAR	4.05	---	---	---	---	---	---	
1.887	ANDRADITE	3.859	12.048	---	100	---	---	---	Synth.
1.890	CALDERITE-ANDRADITE	4.07	---	---	---	---	---	---	
1.893	ANDRADITE-SPESSARTINE	3.98	---	1	71	---	---	28	
1.94	MELANITE	3.79	---	---	---	---	---	---	TiO_2 9.4%.
1.94	KIMZEYITE	3.94	12.46	---	---	---	---	---	TiO_2 5.6, ZrO_2 29.9%.
1.98	SCHORLOMITE	3.85	---	---	---	---	---	---	TiO_2 16.9%.

Table 24. Humite group

Minerals of this group are morphotrophic orthosilicates with the general formula $nXSiO_4 \cdot Y(OH, F)_2$, in which n may be 1, 2, 3, or 4; X may be Mg, Fe, and Mn; Y may be Mg and Mn. The site of the Ti, present in some clinohumites, has not been clearly established.

Alleghanyite	$2Mn_2SiO_4 \cdot Mn(OH)_2$
Chondrodite	$2(Mg, Fe)_2SiO_4 \cdot Mg(OH, F)_2$
Clinohumite	$4(Mg, Fe)_2SiO_4 \cdot Mg(OH, F)_2$
Humite	$3(Mg, Fe)_2SiO_4 \cdot Mg(OH, F)_2$
Leucophoenicite	$3Mn_2SiO_4 \cdot Mn(OH)_2$
Manganhumite	$3Mn_2SiO_4 \cdot Mn(OH)_2$
Norbergite	$(Mg, Fe)_2SiO_4 \cdot Mg(OH, F)_2$
Sonolite	$4Mn_2SiO_4 \cdot Mn(OH, F)_2$

The orthorhombic members have space group Pbnm; the monoclinic members, P_{21}/c , commonly with polysynthetic twinning on {001}. Cleavage of the orthorhombic members is perfect {001}; that of the monoclinic members is poor or lacking. Hardness is 5 to 6; specific gravity, 3.2 to 4.0. All gelatinize in HCl.

Optically the members of this group are biaxial with moderately strong birefringence and Z parallel to the b crystallographic axis. The Mg-Fe members are optically positive, colorless to yellow; the Mn members (except Manganhumite) are optically negative, pink to reddish brown.

Table 24. Humite group (continued)

Refractive index			Biref	NAME	2V	System	Orientation	Remarks
α	β	γ						
1.548	<u>1.552</u>	1.570	.022	NORBERGITE	(+)33° (51+13°)	ORTH	$X = \underline{a}$	Synth Mg-F compd.
1.565	<u>1.570</u>	1.591	.026	NORBERGITE	(+)49°	ORTH	$X = \underline{a}$	
1.593	<u>1.603</u>	1.623	.030	CHONDRODITE	(+)71° r > v wk	MCL	$X:C = 27^\circ$	Poly tw {001}. Pleoc, X yellow, Y and Z nearly col. FeO 2.8%, F 7.9%.
359	<u>1.606</u>	1.630	.032	HUMITE	(+)59° r > v wk	ORTH	$X = \underline{a}$	Synth Mg-F compd.
1.607	<u>1.611</u>	1.639	.032	CHONDRODITE	(+)80° (69+8°) r > v wk	MCL	$X:C = 28^\circ$	Poly tw {001}. Pleoc, X yellow, Y and Z pale yellow. FeO 5.0%, Fe2O3 0.8%.
1.608	<u>1.618</u>	1.636	.028	CLINOHUMITE	(+)59° (79+8°) r < v	MCL	$X:C = 9^\circ$	Synth Mg-F compd.
1.607	<u>1.623</u>	1.643	.036	HUMITE	(+)81° r > v wk	ORTH	$X = \underline{a}$	Pleoc, X and Z pale yellow, Y nearly col.
1.619	<u>1.632</u>	1.653	.034	CHONDRODITE	(+)80° r > v wk	MCL	$X:C = 27^\circ$	Pleoc, X yellow, Y and Z pale yellow. FeO 10.5%, MnO 1.2, F 5.4%.

Table 24. Humite group (continued)

Refractive index			Biref	NAME	2V	System	Orientation	Remarks
α	β	γ						
1.628	<u>1.638</u>	1.655	.027	HUMITE	(+)76° r > v wk	ORTH	$X = \underline{a}$	Pleoc, X and Z pale yellow, Y nearly col. FeO 7.9, Fe ₂ O ₃ 1.0, MnO 1.7, F 5.0%.
1.632	<u>1.643</u>	1.664	.032	CLINOHUMITE	(+)74° r < v	MCL	$X \sim \underline{c}$	FeO 6.6, MnO 1.7, F 3.2%.
1.643	<u>1.653</u>	1.675	.032	HUMITE	(+)68° r > v wk	ORTH	$X = \underline{a}$	Pleoc, X and Z golden yellow, Y pale yellow. FeO 7.8, MnO 0.7, Fe ₂ O ₃ 2.7%.
1.668	<u>1.679</u>	1.700	.032	CLINOHUMITE	(+)Med r < v	MCL	$X \sim \underline{c}$	FeO 9.9, MnO 0.3, Fe ₂ O ₃ 1.0, F 1.8%.
1.67	<u>1.680</u>	1.703	.033	ALLEGHANYITE	(-)66-86° r > v	MCL	$X : \underline{a} = \sim 30^\circ$	
1.707	<u>1.712</u>	1.732	.025	MANGANHUMITE	(+)37° r > v	ORTH	$X = \underline{a}$	MnO 57.1, MgO 14.2, FeO 1.0%.
1.707	<u>1.714</u>	1.734	.027	CLINOHUMITE	(+)56° r < v	MCL	$X \sim \underline{c}$	FeO 10.2, Fe ₂ O ₃ 1.2, TiO ₂ 5.1, F 0.1%.
1.695	<u>1.716</u>	1.725	.030	SONOLITE	(-)Large	MCL	---	Tw {001} common zincian var.
1.751	<u>1.771</u>	1.782	.031	LEUCOPHOENICITE	(-)74° r > v wk	MCL, ps orth	$X \perp c\bar{v}$	Poly tw on {001}.

Table 24. Humite group (continued)

Refractive index			Biref	NAME	2V	System	Orientation	Remarks
α	$\underline{\beta}$	γ						
(1.761)	<u>1.772</u>	1.781	.020	MANGANHUMITE	(+)84°	ORTH	$X = \underline{a}$	MnO ~ 69, Fe ~ 1.0, MgO ~ 0.8%.
1.763	<u>1.779</u>	1.793	.030	SONOLITE	(-)78° r > v	MCL, pris	$X:\underline{c} = 9^\circ$	Tw {001} common.
1.756	<u>1.780</u>	1.792	.036	ALLEGHANYITE	(-)72° r > v	MCL	Ext to tw plane = 22°	Poly tw on {001} common.

Table 25. Melilite group

This group consists of silicates with the general formula $A_2YZ_2(O,OH,F)_7$, in which A may be Ca and Na; Y may be Mg, Al, Fe, Be, and Zn; and Z may be Si and Al. The mineral melilite is commonly formulated in terms of its end members akermanite (Ak) and gehlenite (Ge), neglecting the presence of Na.

Akermanite	(Ak)	$Ca_2MgSi_2O_7$
Gehlenite	(Ge)	$Ca_2Al(SiAl)O_7$
Hardystonite		$Ca_2ZnSi_2O_7$
Melilite		$(Ca,Na)(Mg,Al,Fe)(Si,Al)_2O_7$
Meliphanite		$(Ca,Na)Be(Si,Al)_2(O,OH,F)_7$

These minerals are tetragonal, space group $P\bar{4}2_1m$, with distinct basal cleavage and hardness 5 to 6. They are optically uniaxial and, except for meliphanite, have low birefringence, ranging from optically positive for akermanite to negative for gehlenite.

A related mineral, leucophanite, $(Ca,Na)_2BeSi_2(O,OH,F)_7$, is triclinic, pseudo-orthorhombic. It is biaxial negative with $\beta=1.593$ (table 7).

Refractive index		Biref	NAME	G	Remarks
ω	ϵ				
1.611	1.592	(-).019	MELIPHANITE	3.01	
1.629	1.624	(-).005	MELILITE	2.98	Al_2O_3 6.9, Fe_2O_3 1.2, FeO 2.1, MgO 8.2, Na_2O 3.2%.
1.630	1.637	(+).007	MELILITE	2.96-3.05	Al_2O_3 6.9, Fe_2O_3 1.9, FeO 3.1, MgO 8.2, Na_2O 3.2%.
1.632	1.639	(+).007	AKERMANITE	2.961	Synth, Ak 100.
1.648	1.649	(+).001	MELILITE	2.9-3.1	Synth, Ak 60, Ge 40.
1.65	1.65	.00	MELILITE	3.0	Ak 50, Ge 50.
1.653	1.652	(-).001	MELILITE	3.0	Synth, Ak 50, Ge 50.
1.664	1.672	(+).019	MELIPHANITE	3.03	

Table 25. Melilite group (continued)

Refractive index		Biref	NAME	G	Remarks
ω	ϵ				
1.669	1.657	(-).012	HARDYSTONITE	3.4	
1.669	1.658	(-).011	GEHLENITE	3.04	
1.672	1.661	(-).011	MELILITE	3.1	Al_2O_3 8.4, Fe_2O_3 8.4, FeO 1.1, MgO 4.7, Na_2O 3.8%.

Table 26. Meta-autunite group

The minerals of this group, many of which are lower hydrates of respective members of the autunite group, have the general formula $A(\text{UO}_2)_2(\text{XO}_4)_2 \cdot 4-8\text{H}_2\text{O}$, in which A may be K_2 , $(\text{NH}_4)_2$, Ca, Ba, Mg, Fe^{+2} , Co, Cu, and Zn; X may be P and As.

Abernathyite	$\text{K}_2(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$
Bassetite	$\text{Fe}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$
Meta-ankoleite	$(\text{K}, \text{Ba}, \text{Ca})(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 6\text{H}_2\text{O}$
Meta-autunite	$\text{Ca}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 6\text{H}_2\text{O}$
Metaheinrichite	$\text{Ba}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$
Metakahlerite	$\text{Fe}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$
Metakirchheimerite	$\text{Co}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$
Metalodevite	$\text{Zn}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 10\text{H}_2\text{O}$
Metanovacekite	$\text{Mg}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$
Metatorbernite	$\text{Cu}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 4-8\text{H}_2\text{O}$
Meta-uranocircite	$\text{Ba}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 6\text{H}_2\text{O}$
Metazeunerite	$\text{Cu}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$
Sodium Uranospinitite	$(\text{Na}_2, \text{Ca})(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 5\text{H}_2\text{O}$
Uramphite	$(\text{NH}_4)_2(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 6\text{H}_2\text{O}$

Table 26. Meta-autunite group (continued)

These minerals are tetragonal, orthorhombic, or monoclinic, with square tabular form and perfect basal cleavage. Optically they are uniaxial or biaxial, mostly negative, with low to medium birefringence, weakly to moderately pleochroic in yellow or green. All except those containing Fe, Cu, or Co fluoresce under UV light.

Refractive index			Biref	NAME	2V	System	G
α or ϵ	β or ω	γ or ϵ					
1.57	<u>1.574</u>	1.580	.006	BASSETITE	(-)62°	MCL	3.4
---	<u>1.581</u>		---	META-ANKOLEITE	(-)0°	TET	3.54
1.564	<u>1.585</u>	1.585	.021	URAMPHITE	(-)3°	ORTH(?)	3.7
1.570	<u>1.597</u>		.027	ABERNATHYITE	(-)0°	TET	3.74
1.598	<u>1.600</u>		.002	META-AUTUNITE	(-)0°	TET	3.5
1.596	<u>1.602</u>	1.603	.007	META-AUTUNITE	(-)15°	TET	3.5
1.603	<u>1.610</u>	1.617	.014	BASSETITE	~ 90°	MCL	3.4
1.594	<u>1.611</u>		.017	META-AUTUNITE	(-)0°	TET	3.5
1.585	<u>1.612</u>		.027	SODIUM URANOSPINITE	(-)0°	TET	3.85
1.618	<u>1.621</u>	1.622	.004	METATORBERNITE	(-)0°	TET	3.6
1.604	<u>1.622</u>	1.630	.026	META-AUTUNITE	(-)15°	TET	3.5
1.603	<u>1.622</u>		.019	META-URANOCIRCITE	(-)0°	TET	4.0
1.610	<u>1.623</u>	1.623	.013	META-URANOCIRCITE	(-)20°	ORTH	4.08
	<u>1.626</u>	1.627	.001	METATORBERNITE	(+)0°	TET	3.6-3.8
1.628	<u>1.631</u>		.003	METATORBERNITE	(-)0°	TET	3.7-3.8
1.619	<u>1.635</u>	1.638	.019	METALODEVITE	(-)27-37°	ORTH	(4.0)
1.620	<u>1.637</u>		.017	METANOVACEKITE	(-)0°	TET	3.66

Table 26. Meta-autunite group (continued)

Refractive index			Biref	NAME	2V	System	G
α or ϵ	β or ω	γ or ϵ					
1.609	<u>1.641</u>		.032	METAHEINRICHITE	(-)0°	TET	4.04
1.608	<u>1.642</u>		.034	METAKAHLERITE	(-)0°	TET	---
1.617	<u>1.644</u>		.027	METAKIRCH-HEIMERITE	(-)0°	TET	---
1.624	<u>1.648</u>		.024	METAZEUNERITE	(-)0°	TET	3.67

Table 27. Mica group

The minerals of this group are silicates having the general formula $XY_{2-3}Z_4O_{10}(OH,F)_2$, in which X may be K, Na, Ca, Ba, and H_3O ; Y may be Al, Fe^{+3} , Cr, V^{+3} , Mg, Fe^{+2} , Mn, Zn, and Li; Z may be Si, Al, and Be. The so-called brittle micas and the hydromicas are included here.

Anandite	$(Ba,K)(Fe,Mg)_3(Si,Al,Fe)4O_{10}(O,OH)_2$
Annite	$K(Fe^{+2},Fe^{+3},Ti)_3(Si,Al)4O_{10}(OH,F)_2$
Biotite	$K(Mg,Fe)_3(Si_3Al)O_{10}(OH,F)_2$
Bityite	$CaLiAl_2(AlBeSi_2)O_{10}(OH)_2$
Brammallite	$(Na,H_3O)(Al,Mg)_2(Si,Al)4O_{10}(OH)_2 \cdot nH_2O$
Celadonite	$(K,Na)(Al,Fe,Mg)_2(Si,Al)4O_{10}(OH)_2$
Chernykhite	$(Ba,Na)(V,Al)_2(Si,Al)4O_{10}(OH)_2$
Clintonite	$Ca(Mg,Al)_3(Si,Al)4O_{10}(OH)_2$
Ephesite	$Na(LiAl_2)(Si_2Al_2)O_{10}(OH,F)_2$
Ferri-annite	$K(Fe^{+2},Mg)_3(Fe^{+3},Al)Si_3O_{10}(OH)_2$
Glauconite	$(K,Na)(Al,Fe,Mg)_2(Si,Al)4O_{10}(OH)_2$
Hendricksite	$K(Zn,Mn)_3(Si_3Al)O_{10}(OH,F)_2$
Illite	$(K,H_3O)(Al,Mg,Fe)_2(Si,Al)4O_{10}(OH)_2 \cdot nH_2O$

Table 27. Mica group (continued)

Kinoshitalite	$(Ba, K, Na)(Mg, Mn, Al)_3(Si, Al)_{40}O_{10}(OH, F)_2$
Lepidolite	$K(Li, Fe, Al)_3(Si, Al)_{40}O_{10}(F, OH)_2$
Margarite	$CaAl_2(Al_2Si_2)O_{10}(OH)_2$
Masutomilite	$K(Li, Mn, Al, Fe)_3(Si, Al)_{40}O_{10}(F, OH)_2$
Montdorite	$(K, Na)_2(Fe^{+2}, Mn, Mg)_5(Si, Al)_8O_{20}(F, OH)_4$
Muscovite	$KAl_2(Si_3Al)O_{10}(OH, F)_2$
Paragonite	$NaAl_2(Si_3Al)O_{10}(OH, F)_2$
Phlogopite	$K(Mg, Fe)_3(Si_3Al)O_{10}(F, OH)_2$
Preiswerkite	$NaMg_2Al(Al_2Si_2)O_{10}(OH)_2$
Roscoelite	$K(V, Al)_2(Si_3Al)O_{10}(OH, F)_2$
Siderophyllite	$K(Fe, Mg)_3(Si_3Al)O_{10}(F, OH)_2$
Sodium phlogopite	$NaMg_3(Si_3Al)O_{10}(OH)_2$
Taeniolite	$KLiMg_2Si_4O_{10}F_2$
Tarasovite	$(K, Na, H_3O)Al_2(Si, Al)_{40}O_{10}(OH)_2 \cdot nH_2O$
Tobelite	$(NH_4, K)Al_2(Si_3Al)O_{10}(OH)_2$
Wonesite	$(Na, K)(Mg, Fe)_6(Si, Al)_{80}O_{20}(OH, F)_4$
Zinnwaldite	$K(Li, Fe^{+2}, Al)_3(Si_3Al)O_{10}(F, OH)_2$

These minerals are monoclinic but show complex polytypic relations. Nearly all are biaxial negative, but optic angle may be near 0° , so that some have been listed also in table 5 (uniaxial negative) as well as table 7. The acute bisectric (X) is nearly normal to the plane of the basal cleavage, which is perfect ("micaeous"). $Y=b$ in most of the group, but $Z=b$ in muscovite, paragonite, and in the brittle micas margarite, ephesite, and clintonite. The birefringence is mostly 0.03-0.04, less for glauconite, celadonite, and the brittle micas, higher for those of high index, the latter also showing greater pleochroism.

Table 27. Mica group (continued)

Refractive index			Biref	NAME	(-)2V	Weight %			
α	β	γ				FeO	MnO	Fe_2O_3	Other
1.522	(1.548)	1.549	.027	PHLOGOPITE, synth F	14°	---	---	---	
1.529	1.548	1.553	.024	LEPIDOLITE	39°	0.1	0.7	0.2	
1.522	1.553	1.553	.031	TAENIOLITE	0-5°	---	---	---	
1.537	1.555	1.559	.022	LEPIDOLITE	38°	.07	.09	1.1	
1.535	1.564	1.565	.030	PHLOGOPITE	10°	---	---	---	
1.542	1.567	1.570	.028	LEPIDOLITE	34°	2.0	2.0	0.8	
1.534	1.568	1.570	.036	MASUTOMILITE	30°	1.5	8.1	0.4	$Li_2O\ 4.5$
1.540	1.570	1.570	.030	TAENIOLITE	5°	---	---	---	
1.546	1.575	1.579	.033	BRAMMALITE	small	---	---	---	
1.555	1.575	1.581	.026	TOBELITE	28°		0.6		$(NH_4)_2O$ 3.5, K_2O 2.3
1.548	1.577	1.579	.031	ZINNWALDITE	30°	10.4	1.7	0.6	
1.544	1.578	1.586	.042	TARASOVITE	23°	---	---	---	
1.552	1.582	1.587	.035	MUSCOVITE	30-47°	---	---	---	
1.555	1.584	1.588	.033	PHLOGOPITE	23-30°	4.2	---	3.7	$TiO_2\ 0.4$
1.559	1.586	1.586	.027	GLAUCONITE, var Skolite	small	2.6	---	6.4	Al_2O_3 18.2
1.560	1.587	1.595	.035	TOBELITE	30°		1.0		$(NH_4)_2O$ 3.85, K_2O 3.25
1.572	1.587	1.600	.028	ILLITE	small to med	---	---	---	
1.555	1.590	1.590	.035	MUSCOVITE	0-15°	---	---	---	
1.563	1.590	1.594	.031	MUSCOVITE, var Phengite	33°	1.9	---	1.5	$MgO\ 4.0$

Table 27. Mica group (continued)

Refractive index			Biref	NAME	(-)2V	Weight %			
α	β	γ				FeO	MnO	Fe ₂ O ₃	Other
1.569	<u>1.595</u>	1.609	.040	MUSCOVITE, var Phengite	40°	3.4	3.9	3.1	MgO 3.1, Cr ₂ O ₃ 3.6
1.558	<u>1.598</u>	1.599	.041	BIOTITE	10-13°	13.9	0.4	2.1	
1.565	<u>1.598</u>	1.600	.035	MUSCOVITE	38°	0.9	---	2.8	
1.585	(~1.60)	1.600	.015	GLAUCONITE	small	1.6	---	14.1	Al ₂ O ₃ 8.9
1.571	<u>1.603</u>	1.606	.035	PARAGONITE	30°	---	---	---	
1.580	<u>1.605</u>	1.605	.025	MONTDORITE	3°	17.1	9.9	---	MgO 4.3, Al ₂ O ₃ 5.0
1.544	<u>1.608</u>	1.611	.040	WONESITE	0-5°	---	---	---	
1.571	<u>1.608</u>	1.611	.040	MUSCOVITE	34°	0.5	---	---	MgO 1.1, Cr ₂ O ₃ 0.2
1.573	<u>1.609</u>	1.615	.042	MUSCOVITE, var Phengite	31°	2.8	---	4.1	MgO 4.1, TiO ₂ 1.0
1.586	<u>1.612</u>	1.613	.027	MARGARITE (Sodium analogue)	50°	---	---	---	
1.560	<u>1.614</u>	1.615	.055	PREISWERKITE	5-7°	---	---	---	
1.575	<u>1.617</u>	1.621	.046	BIOTITE, var Mangano- phyllite	30°	1.2	6.2	5.8	
1.597	<u>1.618</u>	1.619	.020	GLAUCONITE	20°	4.0	---	18.8	Al ₂ O ₃ 8.5, MgO 3.6
1.580	<u>1.620</u>	1.623	.043	MUSCOVITE, var Phengite	35°	2.5	---	8.0	MgO 1.9
1.592	<u>1.624</u>	1.625	.033	EPHESITE	18°	---	---	---	Na ₂ O 7.4, Li ₂ O 3.5
1.606	(<u>1.630</u>)	1.632	.026	CELADONITE	small	2.1	---	12.4	MgO 5.8, Al ₂ O ₃ 6.7
1.60	<u>1.63</u>	1.63	.03	MUSCOVITE, var Mariposite	~ 0°	---	---	---	

Table 27. Mica group (continued)

Refractive index			Biref	NAME	(-)2V	Weight %			
α	β	γ				FeO	MnO	Fe_2O_3	Other
1.619	<u>1.633</u>	1.635	.016	KINOSHITALITE	23°	---	7.4	0.7	BaO 17.8, Mn_2O_3 3.2
1.59	> <u>1.63</u>	>1.64	>.05	ROSCOELITE	med	1.7	---	---	Al_2O_3 21.9, V_2O_3 17.4
1.610	<u>1.634</u>	1.634	.024	GLAUCONITE	10°	2.9	---	24.9	Al_2O_3 7.3
1.590	<u>1.640</u>	1.640	.050	SIDEROPHYLLITE	small	30.2	1.0	---	
1.612	<u>1.643</u>	1.643	.031	CELADONITE	small	9.8	---	24.3	MgO 3.7
1.632	<u>1.643</u>	1.646	.014	MARGARITE	63°	---	---	---	
1.595	<u>1.652</u>	1.656	.061	BIOTITE	small	19.9	0.4	3.2	
1.643	<u>1.652</u>	1.654	.011	BITYITE	med	---	---	---	
1.598	<u>1.658</u>	1.660	.062	HENDRICKSITE	8°	---	12.5	4.8	ZnO 19.8
1.648	<u>1.659</u>	1.660	.012	CLINTONITE	5-33°	---	---	---	
1.644	---	1.663	.019	CELADONITE	small	9.2	---	26.9	Al_2O_3 2.1
1.624	<u>1.672</u>	1.672	.048	ANNITE	5°	32.1	---	3.1	TiO_2 3.6
1.610	<u>1.676</u>	1.676	.066	PHLOGOPITE (titanian)	19°	4.7	0.2	6.7	TiO_2 4.4
1.610	<u>1.685</u>	1.704	.094	ROSCOELITE	10-15°	---	---	---	
1.624	<u>1.686</u>	1.686	.062	HENDRICKSITE	2-5°	0.7	13.9	4.8	ZnO 21.4
1.640	<u>1.686</u>	1.702	.062	CHERNYKHITE	11-12°	---	---	---	
1.653	(1.691)	1.691	.038	FERRI-ANNITE	0-10°	24.5	---	8	Al_2O_3 5.5
1.660	(1.720)	1.728	.068	FERRI-ANNITE	40°	29.9	---	13.2	Al_2O_3 1.4
---	<u>1.855</u>	>1.88	>.025	ANANDITE	(+)?	33.1	0.7	7.0	BaO 20.4

Table 28. Monticellite group

This group, sometimes considered to be a subgroup of the olivine group, includes minerals having the general formula CaXSiO_4 , in which X may be Mg, Fe^{+2} , and Mn. All members are orthorhombic, space group Pmcn. Optically they are biaxial negative, with medium to large optic angle, $r > v$, $X=b$, $Z=a$. All gelatinize with HCl.

Refractive index			Biref	NAME and formula	(-)2V	Weight %		
α	β	γ				FeO	Fe_2O_3	MnO
1.641	<u>1.649</u>	1.655	.014	MONTICELLITE CaMgSiO_4	80°	1.4	---	---
1.654	<u>1.664</u>	1.674	.020	MONTICELLITE $\text{Ca}(\text{Mg},\text{Fe})\text{SiO}_4$	82°	7.6	---	0.2
1.674	<u>1.694</u>	1.706	.032	KIRSCHSTEINITE $\text{Ca}(\text{Fe},\text{Mg})\text{SiO}_4$	65°	19.3	1.4	1.1
1.695	<u>1.716</u>	1.725	.030	GLAUCOCHROITE $\text{Ca}(\text{Mn},\text{Fe})\text{SiO}_4$	61°	4.2	---	32.7
1.689	<u>1.720</u>	1.728	.039	KIRSCHSTEINITE $\text{Ca}(\text{Fe},\text{Mg})\text{SiO}_4$	51°	29.3	0.7	1.6

Table 29. Olivine group

Minerals of this group have the general formula X_2SiO_4 , in which X may be Mg, Fe, Mn, Ni, and rarely Zn. There is a complete solid solution series from Fe to Mg, referred to as the olivine series.

Liebenbergite	$(Ni,Mg,Fe)_2SiO_4$
Olivine series	$(Mg,Fe)_2SiO_4$
Forsterite	Fo 90-100
Chrysolite	Fo 70-90
Hyalosiderite	Fo 50-70
Hortonolite	Fo 30-50
Ferrohortonolite	Fo 10-30
Fayalite	Fo 0-10
Tephroite	$(Mn,Mg)_2SiO_4$

The members are orthorhombic, space group $Pmcn$ (except liebenbergite, possibly $Pbnm$), with distinct {010} cleavage, hardness 6.5 to 7, high specific gravity. Optically they are biaxial, with $X=b$, $Z=a$, moderate to high birefringence, refractive indices increasing with Fe-content (compare the monticellite group).

Refractive index			Biref	NAME	2V	Weight %			
α	β	γ				FeO	Fe_2O_3	MnO	MgO
1.636	<u>1.651</u>	1.669	.033	FORSTERITE (synthetic) Fo 100	(+) 85°	---	---	---	57.3
1.653	<u>1.664</u>	1.686	.033	FORSTERITE Fo 93	(+) 87°	8.6	0.4	0.2	---
1.656	<u>1.674</u>	1.695	.039	CHRY SOLITE Fo 88	(+) 89°	10.8	1.7	0.2	46.5
1.661	<u>1.680</u>	1.697	.036	CHRY SOLITE Fo 86	$\sim 90^\circ$	---	---	---	---
1.683	<u>1.704</u>	1.722	.039	CHRY SOLITE Fo 76	(-) 85°	21.6	1.6	0.3	---

Table 29. Olivine group (continued)

Refractive index			Biref	NAME	2V	Weight %			
α	β	γ				FeO	Fe_2O_3	MnO	MgO
1.698	<u>1.720</u>	1.736	.038	HYALOSIDERITE Fo 65	(-)80°	27.8	2.1	0.4	---
1.711	<u>1.727</u>	1.740	.029	TEPHROITE	(-)85°	---	0.3	47.1	18.1
1.710	<u>1.733</u>	1.750	.040	HYALOSIDERITE Fo 63	(-)79°	32.0	2.0	0.5	31.0
1.721	<u>1.750</u>	1.765	.044	HYALOSIDERITE Fo 60	(-)85°	36.8	---	0.5	27.0
1.750	<u>1.766</u>	1.779	.029	TEPHROITE	(-)67°	0.2	---	60.7	6.6
1.742	<u>1.770</u>	1.786	.044	HORTONOLITE Fo 43	(-)69°	---	---	---	---
1.758	<u>1.786</u>	1.804	.046	HORTONOLITE, zincian	(-)77°	35.5	(ZnO 10.7)	16.9	5.6
1.761	<u>1.790</u>	1.806	.045	HORTONOLITE Fo 33	(-)65°	---	---	---	---
1.782	<u>1.802</u>	1.812	.030	TEPHROITE	(-)65°	0.9	---	65.1	2.5
1.777	<u>1.818</u>	1.828	.051	FERRO- HORTONOLITE, manganan	(-)52°	41.5	0.7	21.4	4.5
1.788	<u>1.828</u>	1.840	.052	FERRO- HORTONOLITE Fo 20	(-)58°	58.6	---	0.8	---
1.803	<u>1.843</u>	1.851	.048	FERRO- HORTONOLITE, manganan	(-)48°	56.9	1.2	9.7	1.1
1.816	<u>1.850</u>	1.863	.047	FAYALITE Fo 7	(-)60°	---	---	---	---
1.820	<u>1.854</u>	1.888	.068	LIEBENBERGITE	(-)88°	4.4	(NiO 56.3)	(CaO 1.8)	6.5
1.827	<u>1.869</u>	1.879	.052	FAYALITE	(-)48°	65.0	0.4	1.0	---

Table 30. Osumilite group

This group consists of silicates with double rings of $(\text{Si}, \text{Al})_{12}\text{O}_{30}$. The general formula is $\text{A}_{1-2}\text{B}_{2-3}\text{C}_3\text{Z}_{12}\text{O}_{30} \cdot n\text{H}_2\text{O}$, in which A may be Ba, Ca, K, and Na; B may be Fe^{+2} , Li, Mg, Mn, and Na; C may be Al, Be, Fe^{+2} , Fe^{+3} , Li, Mg, Sn, Ti, Zn, and Zr; Z may be Al and Si. The minerals are hexagonal, space group P6/mcc, or pseudohexagonal. Optically they are uniaxial or nearly so, with low birefringence.

Brannockite	$\text{KLi}_2\text{Sn}_3\text{Si}_{12}\text{O}_{30}$
Darapiosite	$\text{KNa}_2\text{Li}(\text{Mn}, \text{Zn})_2\text{ZrSi}_{12}\text{O}_{30}$
Eifelite	$\text{KNa}_4\text{Mg}_9\text{Si}_{24}\text{O}_{60}$
Emeleusite	$\text{Na}_4\text{Li}_2\text{Fe}^{+3}_2\text{Si}_{12}\text{O}_{30}$
Merrihueite	$(\text{K}, \text{Na})_2(\text{Fe}^{+2}, \text{Mg})_5\text{Si}_{12}\text{O}_{30}$
Milarite	$\text{KCa}_2\text{AlBe}_2\text{Si}_{12}\text{O}_{30} \cdot 0.5\text{H}_2\text{O}$
Osumilite	$(\text{K}, \text{Na})(\text{Fe}^{+2}, \text{Mg})_2(\text{Al}, \text{Fe})_3(\text{Si}, \text{Al})_{12}\text{O}_{30}$
Osumilite-(Mg)	$(\text{K}, \text{Na})(\text{Mg}, \text{Fe}^{+2})_2(\text{Al}, \text{Fe})_3(\text{Si}, \text{Al})_{12}\text{O}_{30} \cdot \text{H}_2\text{O}$
Roedderite	$(\text{Na}, \text{K})_2(\text{Mg}, \text{Fe}^{+2})_5\text{Si}_{12}\text{O}_{30}$
Sogdianite	$(\text{K}, \text{Na})\text{Li}_2(\text{Li}, \text{Fe}^{+2})_2\text{ZrSi}_{12}\text{O}_{30}$
Sugilite	$(\text{K}, \text{Na})(\text{Na}, \text{Fe}^{+3})_2(\text{Li}_2\text{Fe}^{+3})\text{Si}_{12}\text{O}_{30}$
Yagiite	$(\text{Na}, \text{K})_{1-2}\text{Mg}_2(\text{Al}, \text{Mg})_3(\text{Si}, \text{Al})_{12}\text{O}_{30}$
(Armenite, $\text{BaCa}_2\text{Al}_6\text{Si}_9\text{O}_{30} \cdot \text{H}_2\text{O}$, may belong to this group.)	

Refractive index		Biref	NAME	Weight %				Remarks
				MgO	FeO	Na_2O	K_2O	
ω	ϵ							
1.532	1.529	(-).003	MILARITE	---	---	---	---	
1.536	1.543	(+).006	ROEDDERITE	19.0	0.4	5.3	3.8	$2V_Z = 5-8^\circ$.
1.536	1.544	(+).008	YAGIITE	10.5	2.4	3.7	1.4	
1.537	1.542	(+).005	ROEDDERITE	19.5	2.0	4.0	3.3	
1.540	1.546	(+).006	OSUMILITE-Mg	9.5	2.2	0.3	4.0	

Table 30. Osumilite group (continued)

Refractive index		Biref	NAME	Weight %				Remarks
				MgO	FeO	Na ₂ O	K ₂ O	
ω	ϵ							
1.541	1.547	(+).006	OSUMILITE-Mg	7.5	4.0	---	4.1	2V _Z = 5-15°.
1.543	1.545	(+).002	EIFELITE	---	---	---	---	
1.546	1.550	(+).004	OSUMILITE	6.7	10.0	1.0	1.4	Fe ₂ O ₃ 2.0%.
1.553	1.549	(-).004	MILARITE	---	---	---	---	
1.559	---	---	MERRIHUEITE	---	10.5	---	---	
1.567	1.566	(-).001	BRANNOCKITE	---	---	0.7	3.5-4.5	SnO ₂ 28.2, Li ₂ O ₃ 3.7%. Fluor in UV.
1.580	1.575	(-).005	DARAPIOSITE	---	---	3.0	5.1	2V = 5°, ZrO ₂ 5.0, MnO 8.2, ZnO 7.8%.
1.592	---	---	MERRIHUEITE	---	24.9	---	---	
1.597	1.596	(-).001	EMELEUSITE	0.1	---	12.0	---	2V = 0-30°, r > v str. Fe ₂ O ₃ 12.1, Li ₂ O 2.8%.
1.606	1.608	(+).002	SOGDIANITE	---	1.2	2.8	4.8	TiO ₂ 2.9, Fe ₂ O ₃ 4.6, ZrO ₂ 9.8, Li ₂ O 3.7%. Violet.
1.610	1.607	(-).003	SUGILITE	---	0.2	4.4	3.8	Fe ₂ O ₃ 12.8, Li ₂ O 3.1%.

Table 31. Pyrochlore group

The minerals of this group are complex oxides having the general formula $A_{1-2}B_2O_6(O,OH,F)$, in which A may be Ba, Bi, Ca, Ce, K, Na, Pb, Sb⁺³, Sn, Sr, Th, U, Y, and Zr; B may be Nb, Ta, and Ti. Jixianite is a related mineral with W and Fe⁺³.

In the Pyrochlore subgroup, Nb > Ta, and Nb + Ta > 2Ti.

In the Microlite subgroup, Ta > Nb, and Nb + Ta > 2Ti.

In the Betafite subgroup, Nb + Ta < 2Ti.

Bariopyrochlore	$(Ba,Sr)_2(Nb,Ti)_2(O,OH)_7$
Betafite	$(Ca,Na,U)_2(Ti,Nb,Ta)_2O_6(OH)$
Bismutomicrolite	$(Bi,Ca)(Ta,Nb)_2O_6(OH)$
Jixianite	$Pb(W,Fe^{+3})_2(O,OH)_7$
Microlite	$(Ca,Na)_2(Ta,Nb)_2O_6(OH)$
Plumbopyrochlore	$(Pb,Y)(Nb,Ta)_2(O,OH)_7$
Pyrochlore	$(Na,Ca)_2(Nb,Ta)_2O_6(OH,F)$
Stibiobetafite	$(Ca,Sb^{+3})(Ti,Nb,Ta)_2(O,OH)_7$
Uranmicrolite	$(Ca,Na,U)_2(Ta,Nb,Ti)_2O_6(OH)$
Uranpyrochlore	$(Ca,Na,U)_2(Nb,Ta,Ti)_2(O,OH)_7$
Yttropyrochlore	$(Y,Na,Ca,U)_{1-2}(Nb,Ta,Ti)_2(O,OH)_7$
Zirkelite	$(Ca,Zr,Th)_2(Ti,Nb)_2O_7$

These minerals are cubic, space group $Fd\bar{3}m$, commonly occurring as octahedra, with conchoidal fracture. Nearly all are infusible. Optically they are isotropic with high refractive index.

Refractive index	NAME	H	G	Color	Action in acid
1.83	YTTROPYROCHLORE	4.5-5	3.7	Dark brown	Insol.
1.87	PYROCHLORE	5-6	3.77	Brown, yellow	Dec by H_2SO_4 .
1.915	BETAFITE	5	4.0	Greenish black	Dec by H_2SO_4 .
1.925	MICROLITE	5.5	5.5	Yellow, brown	Dec by H_2SO_4 .

Table 31. Pyrochlore group (continued)

Refractive index	NAME	H	G	Color	Action in acid
1.94	BETAFITE	5	5.2	Black, brown, yellow	Dec by H ₂ SO ₄ .
1.97	URANMICROLITE	5.5	5.8	Brown, black	Dec by H ₂ SO ₄ .
1.98	URANPYROCHLORE	5	4.8	Brown	Insol.
2.0	PYROCHLORE	5	4.1-4.3	Brown, red	Dec by H ₂ SO ₄ .
>2.00	BISMUTOMICROLITE	5	6.5-7.2	Yellow, brown	Insol.
2.02	BETAFITE	4.5	4.76	Black	Dec by H ₂ SO ₄ .
2.023	MICROLITE	6	6.42	Brown, yellow	
2.06	ZIRKELITE	6	4.02	Brown, black	Dec by hot HCl.
2.07	MICROLITE	6	5.95	Brown, yellow	Insol.
2.08	BARIOPYROCHLORE	4-4.5	3.7-4.0	Yellow, gray	Insol.
2.08	PLUMBOPYROCHLORE	---	6.34	Green-yellow to brown	---
2.18	PYROCHLORE	5	4.1-4.3	Brownish red	Dec by H ₂ SO ₄ .
2.18	ZIRKELITE	5-6	4.3-4.7	Black to dark brown	Insol.
2.2	STIBIOBETAFITE	~ 5	5.30	Brownish black	---
2.26	JIXIANITE	5	5.30	Brownish black	---
2.28	ZIRKELITE	5-6	4.3-4.7	Black to dark brown	Insol.
2.315	JIXIANITE	5	5.30	Brownish black	---

Table 32. Pyroxene group

The minerals of the pyroxene group are single-chain silicates with the general formula $(W)_{1-p}(X,Y)_{1+p}Z_2O_6$. One subgroup is orthorhombic, in which $p=1$ and X may be mainly Mg and Fe^{+2} . Another subgroup includes the monoclinic members, in which p ranges from 0 to 1, and W may be Ca and Na; X may be Mg, Fe^{+2} , Mn, Ni, and Li; Y may be Al, Fe^{+3} , Cr, and Ti; Z may be Si and Al. A discussion of nomenclature is given by Deer, Howie, and Zussman (1978, v. 2A).

<u>Orthopyroxenes</u>	$(Mg,Fe)_2Si_2O_6$
Enstatite	En 100-87
Bronzite	En 87-70
Hypersthene	En 70-50
Ferrohypersthene	En 50-30
Eulite	En 30-10
(Orthoferrosilite)	En 10-0)
<u>Clinopyroxenes</u>	
Acmite	$NaFe^{+3}Si_2O_6$
Aegirine-augite	$(Na,Ca)(Fe^{+3},Mg,Al)(Si,Al)_2O_6$
Augite	$(Ca,Na)(Mg,Fe,Al,Ti)(Si,Al)_2O_6$
Clinoenstatite	$Mg_2Si_2O_6$
Clinohypersthene	$(Mg,Fe)_2Si_2O_6$
Diopside	$CaMgSi_2O_6$
Fassaite	$Ca(Mg,Fe^{+3},Al)(Si,Al)_2O_6$
Ferroaugite	$(Ca,Na)(Fe,Mg,Al,Ti)(Si,Al)_2O_6$
Ferrosalite	$Ca(Fe,Mg)Si_2O_6$
Hedenbergite	$CaFe^{+2}Si_2O_6$
Jadeite	$NaAlSi_2O_6$
Johannsenite	$CaMnSi_2O_6$
Kanoite	$(Mn,Mg)_2Si_2O_6$
Omphacite	$(Ca,Na)(Mg,Fe^{+2},Fe^{+3},Al)Si_2O_6$

Table 32. Pyroxene group (continued)

Clinopyroxenes (continued)

Pigeonite	$(\text{Mg}, \text{Fe}^{+2}, \text{Ca})(\text{Mg}, \text{Fe}^{+2})\text{Si}_2\text{O}_6$
Salite	$\text{Ca}(\text{Mg}, \text{Fe})\text{Si}_2\text{O}_6$
Spodumene	$\text{LiAlSi}_2\text{O}_6$
Ureyite	$\text{NaCrSi}_2\text{O}_6$

The orthorhombic pyroxenes, space group Pbca, form a solid solution series from enstatite through eulite, with no natural occurrences of the theoretical iron end member, orthoferrosilite. Crystals are commonly prismatic, cleavage is good {210} at 87-88°, hardness 5 to 6. All members are insoluble in or only slightly attacked by acids. Birefringence is low, refractive indices increase with increasing iron content, with optic sign changing from positive to negative at about β 1.68 and back to positive at about β 1.76. All members of the series have $X=b$, $Z=c$, positive elongation. Although the enstatite end member is not pleochroic, many intermediate members show weak pleochroism, X pink, Y yellow, Z pale green.

The monoclinic pyroxenes have space group P2/c (except pigeonite, P21/c), good cleavage on {110} at about 87°, hardness 5 to 6. All are insoluble in or only very slightly attacked by acids. Optically, they are biaxial, with refractive indices increasing with increasing iron and titanium content; $Y=b$ (except clinoenstatite, clinohypersthene, and some pigeonites, where $X=b$). Minerals of the series diopside-hedenbergite-johannsenite along with the augites are optically positive, with moderate 2V and birefringence, $Z:c=40-50^\circ$. Pigeonite, a Ca-poor member, is characterized by low 2V. The series jadeite to aegirine to aegirine-augite has high 2V (60-90°), with indices and birefringence increasing with increase in Fe-content. For aegirines, $Z:c=75-89^\circ$.

Table 32. Pyroxene group (continued)

Refractive index	α	β	γ	Biref	NAME	2V	Opt orientation	System	Weight %			Remarks
									FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃
1.640	<u>1.645</u>	1.652	.012	JADEITE	(+)	67°	$Y = \frac{b}{Z:C} = 22^\circ$	MCL	---	---	---	---
1.650	<u>1.653</u>	1.658	.008	ENSTATITE En 100	(+)	59° $r < v$	$X = \frac{b}{Z} = \frac{b}{C}$	ORTH	---	---	---	Synth.
1.651	<u>1.654</u>	1.660	.009	CLINO- ENSTATITE En 100	(+)	54°	$X = \frac{b}{Z:C} = 22^\circ$	MCL	---	---	---	Synth.
1.654	<u>1.657</u>	1.666	.012	JADEITE	(+)	70° $r > v$ med	$Y = \frac{b}{Z:C} = 34^\circ$	MCL	---	0.4	---	
1.653	<u>1.659</u>	1.677	.024	SPODUMENE	(+)	66° $r < v$	$Y = \frac{b}{Z:C} = 25^\circ$	MCL	---	---	---	
1.665	<u>1.669</u>	1.674	.009	ENSTATITE En 91	(+)	80° $r < v$ wk	$X = \frac{b}{Z} = \frac{b}{C}$	ORTH	7.0	---	---	
1.664	<u>1.670</u>	1.685	.021	OMPHACITE	(+)	66° $r < v$ med	$Y = \frac{b}{Z:C} = 35^\circ$	MCL	2.8	---	1.8	---
1.662	<u>1.670</u>	1.682	.020	JADEITE	(+)	70° $r > v$ med	$Y = \frac{b}{Z:C} = 35^\circ$	MCL	0.5	---	3.2	---
1.668	<u>1.671</u>	1.682	.014	SPODUMENE	(+)	65° $r < v$	$Y = \frac{b}{Z:C} = 24^\circ$	MCL	---	---	0.7	---

Table 32. Pyroxene group (continued)

α	β	γ	Biref	NAME	2V	Opt orientation	System	Weight %			Remarks
								FeO	MnO	Fe ₂ O ₃	
1.665	<u>1.672</u>	1.695	.030	DIOPSIDE	(+)57° $r > v$ wk	$Y = \frac{b}{Z:C} = 39^\circ$	MCL	1.5	0.1	0.6	---
1.673	<u>1.678</u>	1.705	.032	DIOPSIDE	(+)48° $r > v$	$Y = \frac{b}{Z:C} = 42^\circ$	MCL	8.2	0.3	1.4	4.5
1.675	<u>1.678</u>	1.688	.013	CLINO-HYPERSTHENE	(+)57°	$X = \frac{b}{Z:C} = 36^\circ$	MCL	12.6	0.5	---	0.5
1.674	<u>1.680</u>	1.685	.011	BRONZITE	(-)79°	$X = \frac{b}{Z} = \frac{b}{C}$	ORTH	11.1	0.5	---	---
1.673	<u>1.681</u>	1.695	.022	OMPHACITE	(+)70° med	$Y = \frac{b}{Z:C} = 41^\circ$	MCL	2.7	---	1.3	8.3
1.679	<u>1.681</u>	1.685	.006	JADEITE	(+)64° $r > v$ med	$Y = \frac{b}{Z:C} = 38^\circ$	MCL	1.1	---	5.7	---
1.676	<u>1.683</u>	1.705	.029	DIOPSIDE, manganan	(+)1.0° $r > v$	$Y = \frac{b}{Z:C} = 5^\circ$	MCL	---	7.4	0.4	0.4
1.684	<u>1.684</u>	1.707	.023	PIGEONITE	(+)18- 26°	$Y = \frac{b}{Z:C} = 40^\circ$	MCL	20.1	0.4	3.5	---
1.677	<u>1.685</u>	1.708	.031	DIOPSIDE	(+)59° $r > v$ wk	$Y = \frac{b}{Z:C} = 39^\circ$	MCL	5.6	0.5	3.2	2.2

Table 32. Pyroxene group (continued)

Refractive index	α	β	γ	Biref	NAME	2V	Opt-orientation	System	Weight %				Remarks
									FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.687	<u>1.692</u>	1.715	.028	DIOPSIDE	(+) 51°	$Y = \frac{b}{c} = 42^\circ$ $Z:c = 42^\circ$	MCL	9.5	0.2	1.6	2.3		
1.686	<u>1.693</u>	1.719	.028	FASSAITE	(+) 56° $r > v$ med	$Y = \frac{b}{c} = 43^\circ$ $Z:c = 43^\circ$	MCL	3.1	---	2.1	13.4	Ti0 ₂ 0.5%.	
1.685	<u>1.695</u>	1.700	.015	BRONZITE En 73	(-) 69° $r > v$	$X = \frac{b}{c}$ $Z = \underline{c}$	ORTH	17.1	0.4	1.0	---		
1.696	<u>1.698</u>	1.721	.025	PIGEONITE	(+) 25°	$Y = \frac{b}{c} = 41^\circ$ $Z:c = 41^\circ$	MCL	21.4	0.4	0.8	---	CaO 5.5%.	
1.690	<u>1.699</u>	1.721	.031	DIOPSIDE, manganan	(+) 60° $r > v$ wk	$Y = \frac{b}{c} = 43^\circ$ $Z:c = 43^\circ$	MCL	1.7	10.0	1.5	0.3		
1.689	<u>1.699</u>	1.718	.029	OMPHACITE	(+) 64° $r < v$ med	$Y = \frac{b}{c} = 40^\circ$ $Z:c = 40^\circ$	MCL	5.9	---	3.8	6.1		
1.695	<u>1.700</u>	1.720	.025	AUGITE	(+) 50°	$Y = \frac{b}{c} = 43^\circ$ $Z:c = 43^\circ$	MCL	14.1	0.2	1.3	2.9	MgO 10.5%.	
1.700	<u>1.707</u>	1.724	.024	SALITE	(+) 52°	$Y = \frac{b}{c} = 44^\circ$ $Z:c = 44^\circ$	MCL	9.8	0.3	1.8	2.8	Ti0 ₂ 1.3%.	
1.700	<u>1.710</u>	1.733	.033	JOHANNSENITE	(+) 60° $r > v$	$Y = \frac{b}{c} = 49^\circ$ $Z:c = 49^\circ$	MCL	1.5	22.6	0.8	---		
1.703	<u>1.713</u>	1.717	.014	HYPERTHENE En 57	(-) 53°	$X = \frac{b}{c}$ $Z = \underline{c}$	ORTH	25.7	---	1.0	---		

Table 32. Pyroxene group (continued)

<u>Refractive index</u>				NAME	2V	Opt orientation	System	Weight %				Remarks
	α	β	γ					FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.708 <u>1.714</u>	1.736	.028		FERROSALITE wk	(+) 70° $r > v$	$y = \frac{b}{c}$	MCL	18.6	0.2	0.5	1.1	
1.714 <u>1.714</u>	1.742	.028		PIGEONITE	(+) $0-$ 12°	$y = \frac{b}{c} = 41^\circ$	MCL	27.8	1.0	1.7	0.9	CaO 3.8%.
1.710 <u>1.716</u>	1.736	.028		FERROAUGITE	(+) 52°	$y = \frac{b}{c} = 49^\circ$	MCL	20.2	1.1	3.5	0.9	TiO ₂ 1.2%.
1.715 <u>1.717</u>	1.728	.013		KANOITE	(+) 41°	$y = \frac{b}{c} = 48^\circ$	MCL	3.1	31.2	---	---	MgO 15.1%.
1.710 <u>1.719</u>	1.738	.028		JOHANNSENITE	(+) 70°	$y = \frac{b}{c} = 48^\circ$	MCL	1.0	26.8	0.3	---	
1.712 <u>1.719</u>	1.736	.024		FASSAITE	(+) 55° $r > v$	$y = \frac{b}{c} = 46^\circ$	MCL	0.2	---	6.1	15.8	CaO 25.3%.
1.714 <u>1.723</u>	1.774	.060		AUGITE, zincian	(+) 74° ($47+5^\circ$)	$y = \frac{b}{c} = 55^\circ$	MCL	3.7	7.2	4.2	1.2	ZnO 7.7%.
1.715 <u>1.728</u>	1.731	.016		FERROHYPER- STHENE En 45	(-) 51°	$x = \frac{b}{c}$	ORTH	29.4	0.2	2.1	---	
1.725 <u>1.730</u>	1.750	.025		AUGITE, titanian	(+) 40°	$y = \frac{b}{c} = 45^\circ$	MCL	5.0	0.2	4.4	7.4	TiO ₂ 5.5%.
1.726 <u>1.732</u>	1.753	.027		HEDENBERGITE wk	(+) 49° $r > v$	$y = \frac{b}{c} = 44^\circ$	MCL	24.7	0.7	1.9	---	Pleoc, X pale green, Z dark green.

Table 32. Pyroxene group (continued)

Refractive index	Biref			NAME	2V	Opt-orientation	System	Weight %			Remarks	
	α	β	γ					FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.727 <u>1.736</u>	1.751	.024		FASSAITE	(+)49° $r > v$	$y = \frac{b}{c} = 48^\circ$ $Z:c =$	MCL	1.2	---	6.2	12.9	TiO ₂ 3.0%.
1.720 <u>1.740</u>	1.757	.037		AEGIRINE-AUGITE	(-)85° $r < v$	$y = \frac{b}{c} = 75^\circ$ $Z:c =$	MCL	7.5	0.6	14.0	2.2	Pleoc., X bright green, Y and Z greenish yellow. Na ₂ O 5.3%.
1.735 <u>1.741</u>	1.761	.026		FERROAUGITE	(+)51°	$y = \frac{b}{c} = 40^\circ$ $Z:c =$	MCL	24.0	0.4	1.2	4.1	TiO ₂ 1.6%.
383 1.736 <u>1.747</u>	1.752	.016		EULITE En 30	(-)63°	$x = \frac{b}{c}$ $Z =$	ORTH	39.4	0.9	---	---	
1.741 <u>1.751</u>	1.774	.033		ACMITTE	(+)mod	$y = \frac{b}{c} = 60^\circ$ $Z:c =$	MCL	1.4	7.2	12.9	---	Zn 8.8%.
1.745 <u>1.753</u>	1.771	.026		HEDENBERGITE	(+)52° $r > v$ wk	$y = \frac{b}{c} = 41^\circ$ $Z:c =$	MCL	29.7	0.9	1.4	---	
1.732 <u>1.756</u>	1.770	.038		ACMITTE	(-)80° $r > v$	$y = \frac{b}{c} = 81^\circ$ $Z:c =$	MCL	1.1	4.7	18.2	5.9	Pleoc., X pink, Y pale blue, Z blue. TiO ₂ 0.6%.
1.740 <u>1.756</u>	1.762	.022		UREYITE	(-)65°	$y = \frac{b}{c} = 76^\circ$ $Z:c =$	MCL	---	---	---	---	
1.752 <u>1.762</u>	1.772	.020		EULITE En ~ 11	(-)87° $r < v$ str	$x = \frac{b}{c}$ $Z =$	ORTH	44.5	---	1.9	---	Dark brown.

Table 32. Pyroxene group (continued)

<u>Refractive index</u>				NAME	2V	Opt orientation	System	Weight %				Remarks
	α	β	γ					FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.741	<u>1.767</u>	1.789	.048	AEGIRINE-AUGITE	(-)85° $r > v$	$y = \frac{b}{c} = 78^\circ$ $z:c =$	MCL	5.2	0.5	19.1	1.3	Pleoc., X deep green, Y yellow-green, Z brown-green, abs X > Y = Z. Na ₂ O 7.2%, TiO ₂ 1.6%.
1.751	1.786	1.800	.049	ACMITTE	(-)69° $r > v$	$y = \frac{b}{c} = 89^\circ$ $z:c =$	MCL	3.7	0.2	25.4	2.9	Pleoc., X and Y deep green, Z yellow-brown. Na ₂ O 11.7%, TiO ₂ 0.7%.
1.770	1.812	1.830	.060	ACMITTE	(-)58° $r > v$	$y = \frac{b}{c} = 84^\circ$ $z:c =$	MCL	0.8	---	31.4	1.9	Pleoc., X pale yellow-green, Y yellow-green, Z emerald green, TiO ₂ 0.8%.

Table 33. Scapolite group

The minerals of this group are solid solutions of the end-members marialite, $\text{Na}_8(\text{AlSi}_3\text{O}_8)_6(\text{Cl}_2,\text{SO}_4,\text{CO}_3)$, and meionite, $\text{Ca}_8(\text{Al}_2\text{Si}_2\text{O}_8)_6(\text{CO}_3,\text{Cl}_2,\text{SO}_4)$, which themselves are not found as such in nature. The composition and paragenetic relations of the scapolites have been reviewed in detail by Shaw (1960).

Minerals of this group are tetragonal, space group $\text{P}4/\text{m}$, with cleavages {100} perfect, {110} less so, hardness 5 to 6. Optically they are uniaxial negative, refractive indices and birefringence increasing with Ca-content.

Refractive index		Biref	NAME	Weight %					
				Na_2O	K_2O	CaO	Cl	CO_2	SO_3
ω	ϵ								
1.532	1.522	(-).010	MARIALITE	7.1	4.3	6.3	1.0	2.0	---
1.546	1.542	(-).004	MARIALITE Me 19	10.5	1.2	4.8	3.0	1.1	0.03
1.561	1.546	(-).015	SCAPOLITE Me 33	8.6	1.1	8.3	2.2	1.7	0.4
1.574	1.549	(-).025	SCAPOLITE Me 54	5.4	1.0	12.5	1.4	2.9	0.1
1.587	1.555	(-).032	MEIONITE Me 70	2.4	2.4	15.8	0.1	2.9	0.9
1.607	1.571	(-).036	MEIONITE Me 84	2.0	0.3	20.4	0.2	4.1	0.3

Table 34. Sodalite group

The minerals of this group are sodium- and calcium-aluminosilicates having the general formula $(\text{Na}, \text{Ca})_{6-8}\text{Al}_6\text{Si}_6(\text{O}, \text{S})_{24}[\text{SO}_4, \text{Cl}_2, (\text{OH})_2]_{1-2}$.

Hauyne	$(\text{Na}, \text{Ca})_{6-8}\text{Al}_6\text{Si}_6(\text{O}, \text{S})_{24}(\text{SO}_4, \text{Cl}_2)_{1-2}$
Lazurite	$(\text{Na}, \text{Ca})_{7-8}(\text{Al}, \text{Si})_{12}(\text{O}, \text{S})_{24}[\text{SO}_4, \text{Cl}_2(\text{OH})_2]$
Nosean	$\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}(\text{SO}_4)$
Sodalite	$\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}\text{Cl}_2$

These minerals are cubic, space group $\text{P}43m$ or $\text{P}43n$, habit dodecahedral, color blue or white, cleavage {110} poor, hardness 5 to 6. They fuse at 3.5-4.5, gelatinize with HCl. Optically they are isotropic, with refractive index increasing with Ca-content. The group is discussed by Taylor (1967) and by Hogarth and Griffin (1976).

Refractive index	NAME	G	Weight %				
			Na_2O	CaO	SO_3	Cl	Other
1.470	NOSEAN	2.32	20.7	1.7	8.8	0.7	
1.483	SODALITE	2.30	---	---	---	7.1	H_2O 2.5%.
1.487	SODALITE var. Hackmanite	2.2-2.3	---	---	---	---	
1.494	HAUYN	2.40	19.9	4.9	9.8	1.3	
1.495	NOSEAN	2.3-2.4	---	---	---	---	
1.500	LAZURITE	2.4	---	---	---	---	
1.509	HAUYN	2.51	13.2	10.1	---	---	K_2O 2.8%.
1.522	LAZURITE	2.4	---	---	---	---	

Table 35. Spinel group

Minerals of this group have the general formula AB_2O_4 , in which A may be Mg, Fe^{+2} , Zn, Mn^{+2} , rarely Ni, Cu, Co, Ti, and Ge; B may be Al, Fe^{+3} , Cr, Mn^{+3} , rarely Fe^{+2} and V $^{+3}$. Common end members are spinel (Mg-Al), hercynite (Fe-Al), gahnite (Zn-Al), magnesiochromite (Mg-Cr), chromite (Fe-Cr), magnesioferrite (Mg-Fe), magnetite (Fe-Fe), and jacobsite (Mn-Fe).

Chromite	$(Fe^{+2}, Mg)(Cr, Al, Fe^{+3})_2O_4$
Franklinite	$(Zn, Mn^{+2}, Fe^{+2})(Mn^{+3}, Fe^{+3})_2O_4$
Gahnite	$Zn(Al, Fe^{+3})_2O_4$
Galaxite	$(Mn, Fe^{+2})(Al, Fe^{+3})_2O_4$
Hercynite	$(Fe^{+2}, Mg)(Al, Fe^{+3})_2O_4$
Jacobsite	$(Mn, Fe^{+2}, Mg)Fe_2^{+3}O_4$
Magnesiochromite	$(Mg, Fe^{+2})(Cr, Al, Fe^{+3})O_4$
Magnesioferrite	$(Mg, Fe^{+2})Fe^{+3}O_4$
Magnetite	$Fe^{+2}Fe^{+3}O_4$
Spinel	$(Mg, Fe^{+2})Al_2O_4$
Trevorite	$NiFe_2O_4$

Rare minerals of the group include coulsonite (Fe-V), cuprospinel (Cu-Fe), ulvöspinel ($TiFe_2^{+2}O_4$), and its germanium analogue brunogeierite, $GeFe_2O_4$. Ringwoodite, $(Mg, Fe)_2SiO_4$, has spinel-type structure ($n=1.768$).

These minerals are cubic, commonly octahedral, space group $Fd\bar{3}m$. They are characterized by high refractive index and specific gravity, lack of cleavage, hardness 7.5 to 8, and infusibility. Those members high in Fe^{+3} or V $^{+3}$ (magnetite, coulsonite, ulvöspinel, brunogeierite) are opaque or nearly so. The aluminum and chromium end-members are insoluble in acids; those with dominant Fe^{+3} are dissolved by HCl.

Table 35. Spinel group (continued)

Refractive index	NAME	G	Weight %					Remarks
			Cr ₂ O ₃	Fe ₂ O ₃	FeO	MnO	MgO	
1.714	SPINEL	3.58	---	---	---	---	---	Synth end member.
1.738	SPINEL	3.62	---	1.5	5.6	0.1	---	
1.747	SPINEL, zincian	3.97	---	---	1.9	---	16.8	ZnO 18.2.
1.776	HERCYNITE	4.20	---	11.7	23.3	---	10.8	"Pleonaste"
1.782	GAHNITE	4.38	---	---	8.5	---	2.4	ZnO 31.4.
1.798	SPINEL, chromian	3.94	19.6	---	9.8	---	19.5	Al ₂ O ₃ 48.3.
1.805	GAHNITE	4.61	---	---	---	---	---	Synth end member.
1.818	GAHNITE	4.60	---	2.9	1.7	---	---	
1.833	SPINEL, chromian	3.93	16.8	5.8	11.7	---	19.3	Al ₂ O ₃ 45.4.
1.835	HERCYNITE	4.40	---	---	---	---	---	Synth end member.
1.856	GALAXITE	---	---	6.4	2.3	33.1	---	
1.90	MAGNESIO- CHROMITE	4.39	---	---	---	---	---	Synth end member.
1.923	GALAXITE	4.23	---	---	16.4	34.0	1.5	
1.97	MAGNESIO- CHROMITE	4.67	43.6	4.0	11.1	---	16.0	Al ₂ O ₃ 24.4.
2.10	CHROMITE	4.65	49.7	8.1	8.7	9.3	2.4	Al ₂ O ₃ 17.6, ZnO 4.1.
2.12	CHROMITE	5.22	---	---	---	---	---	Synth end member.
2.30	JACOB SITE	4.84	---	---	---	---	---	Wk magn.

Table 35. Spinel group (continued)

Refractive index	NAME	G	Weight %					Remarks
			Cr ₂ O ₃	Fe ₂ O ₃	FeO	MnO	MgO	
2.3 (?)	TREVORITE	5.2	---	---	---	---	---	Str magn, n(calc) = 2.39.
2.38 Na 2.34 Li	MAGNESIO- FERRITE	4.5- 4.7	---	---	---	---	---	Str magn.
2.36 Li	FRANKLINITE	5.2	---	---	---	---	---	Wk magn.
2.42 Na	MAGNETITE	5.17	---	---	---	---	---	Str magn.
2.45 Na 2.41 Li	FRANKLINITE	5.2	---	---	---	---	---	Wk magn.
2.585 Na 2.535 Li	JACOB SITE	4	---	---	---	---	---	Wk-mod magn.

Table 36. Stibiconite group

Minerals of this group have the general formula A₁₋₂B₂O₆(O,OH,F), in which A may be Bi⁺³, Ca, Na, Pb, Sb⁺³, Ag, K, Fe, and Mn; B may be Sb⁺⁵ and Ti.

Bindheimite	Pb ₂ Sb ₂ O ₆ (O,OH)
Bismutostibiconite	(Bi,Fe)Sb ₂ O ₇
Romeite	(Ca,Na,Fe,Mn) ₂ (Sb,Ti) ₂ O ₆ (O,OH)
Stetefeldite	Ag ₂ Sb ₂ (O,OH) ₇ (?)
Stibiconite	SbSb ₂ O ₆ (O,OH)

These minerals are cubic, space group Fd3m, isostructural with those of the pyrochlore group. They are commonly massive, fine-grained, and highly variable in physical properties. Specific gravity is high; optically isotropic.

Table 36. Stibiconite group (continued)

Refractive index	NAME	G	Remarks
1.60	STIBICONITE	5.1-5.6	Insol in acids. Infus.
1.70	STIBICONITE	5.1-5.6	Insol in acids. Infus.
1.72	BINDHEIMITE	4.6-5.6	Dec by HCl. F 3-4.
1.74	STIBICONITE	5.2-5.6	Insol in acids. Infus.
1.79	BINDHEIMITE	4.6-5.6	Dec by HCl. F 3-4.
1.80	STIBICONITE	5.2-5.6	Insol in acids. Infus.
1.817	ROMEITE	4.7-5.4	Insol in acids. F diff.
1.84	BINDHEIMITE	4.6-5.4	Dec by HCl. F 3-4.
1.87	ROMEITE	4.7-5.4	Insol in acids. F 4.5.
1.90	STIBICONITE	5.2-5.6	Insol in acids. Infus.
1.94	BINDHEIMITE	5.0-5.6	Dec by HCl. F 3-4.
1.95	STETEFELDITE	4.1-4.6	
2.05	STIBICONITE	5.2-5.6	Insol in acids. Infus.
2.09	BISMUTOSTIBICONITE	7.38	
2.09	ROMEITE, var Schneebergite	5.41	Insol in acids. F diff. Anom biref.
2.20	ROMEITE, var Lewisite	4.95	Insol in acids. Fus. Anom biref.

Table 37. Tourmaline group

Minerals of this group are borosilicates having the general formula $AX_3Y_6(BO_3)_3Si_6O_{18}(OH,F)_4$, in which A may be Ca, Na, and K; X may be Al, Fe⁺², Fe⁺³, Li, Mg, Mn⁺², and V⁺³; Y is mainly Al, but may also be Cr⁺³, Fe⁺³, and V⁺³.

Buergerite	$NaFe^{+3}Al_6(BO_3)_3Si_6O_{18}(O,OH,F)_4$
Chromdravite	$Na(Mg,V,Al)_3(Cr,Fe^{+3})_6(BO_3)_3Si_6O_{18}(OH)_4$
Dravite	$(Na,Ca)(Mg,Fe)_3Al_6(BO_3)_3Si_6O_{18}(OH,F)_4$
Elbaite	$Na(Al,Li,Fe)_3Al_6(BO_3)_3Si_6O_{18}(OH,F)_4$
Ferridravite	$(Na,K)(Mg,Fe^{+2})_3(Fe^{+3},Al)_6(BO_3)_3Si_6O_{18}(O,OH)_3$
Liddicoatite	$Ca(Li,Al)_3Al_6(BO_3)_3Si_6O_{18}(O,OH,F)_4$
Schorl	$(Na,Ca)(Mg,Fe)_3(Al,Fe^{+3})_6(BO_3)_3Si_6O_{18}(OH)_4$
Uvite	$Ca(Mg,Fe)_3Al_5Mg(BO_3)_3Si_6O_{18}(OH)_4$

These minerals are trigonal, space group R3m, with characteristic prismatic habit, lack of cleavage (except buergerite), specific gravite 3.0-3.3. All have hardness of 7 and are insoluble in acids. Optically they are uniaxial negative with moderate to strong birefringence and negative elongation. Absorption in the pleochroic members is O > E.

Table 37. Tourmaline group (continued)

Refractive index		NAME	Pleochroism			Weight %					
ω	ϵ		Biref (-)	0	E	Fe_2O_3	FeO	MgO	CaO	Na_2O	Other
1.634	1.612	.022	DRAVITE	---	---	---	.02	8.4	---	1.9	
1.635	1.618	.017	ELBAITE	---	---	.03	---	---	0.4	2.4	Li_2O 1.9.
1.637	1.621	.016	LIDDICOATITE	Dark brown	Pale brown	---	---	---	4.2	0.9	Li_2O 2.5, F 1.7.
1.638	1.619	.019	UVITE	Dark	Light	---	0.5	15.4	5.5	0.2	
1.646	1.625	.021	ELBAITE	---	---	---	4.9	---	---	2.5	Li_2O 1.2, MnO 0.7.
1.653	1.637	.016	DRAVITE	Dark olive green	Pale brown	2.2	3.6	9.3	1.6	1.9	
1.660	1.639	.021	UVITE	Dark	Light	2.7	5.8	11.0	7.9	---	
1.661	1.632	.029	SCHORL	---	---	0.8	8.7	5.6	---	2.3	
1.675	1.643	.032	DRAVITE	Red-brown	Pale green	---	1.3	6.3	---	---	V_2O_3 8.0.
1.693	1.659	.034	SCHORL	---	---	9.2	0.6	7.3	1.5	2.6	TiO_2 2.2.
1.710	1.664	.046	DRAVITE, chromian	---	---	1.8	1.8	---	---	---	Cr_2O_3 17.8.
1.735	1.625	.110	BUERGERITE	Yellow-brown	Pale yellow	17.6	1.3	---	0.7	2.5	

Table 37. Tourmaline group (continued)

Refractive index	ω	ϵ	Brief (-)	NAME	Pleochroism			Weight %				
					0	E	Fe_2O_3	FeO	MgO	CaO	Na_2O	Other
1.778	1.772	.006	CHROMDRAWITE	Dark green	Yellow-green	7.65	---	9.05	0.16	2.66	Cr_2O_3	31.6,
1.800	1.743	.057	FERRIDRAWITE	Very dark brown	Light brown	38.4	7.5	5.6	---	2.2	K_2O	1.0.

Table 38. Zeolite group

The zeolite minerals form a complex group of hydrated aluminosilicates with ratio O:(Si+Al)=2. They have the general formula $A(XO_2)_n \cdot mH_2O$, in which A may be Na, Ca, and K, and less commonly Ba, Sr, Mg, Li(7), and Cs; X may be Si and Al. They are characterized by ready loss of their water, which can be restored, at least in part, and by their ability to exchange basic cations.

Amicite	$K_2Na_2Al_4Si_4O_{16} \cdot 5H_2O$
Analcime	$(Na, K, Cs)AlSi_2O_6 \cdot H_2O$
Barrerite	$(Na, K, Ca)_2Al_2Si_7O_{18} \cdot 7H_2O$
Brewsterite	$(Sr, Ba, Ca)Al_2Si_6O_{16} \cdot 5H_2O$
Chabazite	$(Ca, Na_2)Al_2Si_4O_{12} \cdot 6H_2O$
Clinoptilolite	$(Na_2, K_2, Ca)_{2-3}Al_3(Al, Si)_2Si_{13}O_{36} \cdot 12H_2O$
Cowlesite	$CaAl_2Si_3O_{10} \cdot 6H_2O$
Dachiardite	$(Ca, Na_2, K_2)_5Al_{10}Si_{38}O_{96} \cdot 25H_2O$
Edingtonite	$BaAl_2Si_3O_{10} \cdot 4H_2O$
Epistilbite	$CaAl_2Si_6O_{16} \cdot 5H_2O$
Erionite	$(Ca, Na_2, K_2)Al_2Si_6O_{16} \cdot 6H_2O$
Faujasite	$Na_2CaAl_4Si_{10}O_{28} \cdot 20H_2O$
Ferrierite	$(Na, K)_2MgAl_3Si_{15}O_{36}(OH) \cdot 9H_2O$
Garronite	$Na_2Ca_5Al_{12}Si_{20}O_{64} \cdot 27H_2O$
Gismondine	$CaAl_2Si_2O_8 \cdot 4H_2O$
Gmelinite	$(Na_2, Ca)Al_2Si_4O_{12} \cdot 6H_2O$
Gobbinsite	$Na_4(Ca, K_2)Al_6Si_{10}O_{32} \cdot 12H_2O$
Gonnardite	$Na_2CaAl_4Si_6O_{20} \cdot 7H_2O$
Goosecreekite	$CeAl_2Si_6O_{16} \cdot 5H_2O$
Harmotome	$(Ba, K_2)_{1-2}(Al, Si)_1O_{16} \cdot 6H_2O$
Herschelite	$(Na_2, Ca, K_2)AlSi_2O_6 \cdot 3H_2O$
Heulandite	$(Na, K)_{2-3}Al_3(Al, Si)_2Si_{13}O_{36} \cdot 12H_2O$
Laumontite	$CaAl_2Si_4O_{12} \cdot 4H_2O$

Table 38. Zeolite group (continued)

Levyne	$\text{CaAl}_2\text{Si}_3\text{O}_{10} \cdot 5\text{H}_2\text{O}$
Mazzite	$(\text{K}_2, \text{Na}_2, \text{Ca})_6\text{Mg}_3\text{Al}_{13}\text{Si}_{27}\text{O}_{80} \cdot 16\text{H}_2\text{O}$ (?)
Merlinoite	$(\text{K}, \text{Ca}, \text{Na})_7\text{Si}_{23}\text{Al}_9\text{O}_{64} \cdot 23\text{H}_2\text{O}$
Mesolite	$\text{Na}_2\text{CaAl}_6\text{Si}_9\text{O}_{30} \cdot 8\text{H}_2\text{O}$
Mordenite	$(\text{Ca}, \text{Na}_2, \text{K}_2)\text{Al}_2\text{Si}_{10}\text{O}_{24} \cdot 7\text{H}_2\text{O}$
Natrolite	$(\text{Na}_2, \text{Ca})\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 2\text{H}_2\text{O}$
Offretite	$(\text{K}_2, \text{Ca})_3\text{Al}_5\text{Si}_{13}\text{O}_{36} \cdot 14\text{H}_2\text{O}$
Paulingite	
Phillipsite	$(\text{K}_2, \text{Na}_2, \text{Ca})_{1-2}(\text{Al}, \text{Si})_8\text{O}_{16} \cdot 6\text{H}_2\text{O}$
Pollucite	$(\text{Cs}, \text{Na})\text{AlSi}_2\text{O}_6 \cdot \text{mH}_2\text{O}$
Scolecite	$\text{CaAl}_2\text{Si}_3\text{O}_{10} \cdot 3\text{H}_2\text{O}$
Sodium dachiardite	$(\text{Na}_2, \text{Ca}, \text{K})_{4-5}\text{Al}_8\text{Si}_{40}\text{O}_{96} \cdot 26\text{H}_2\text{O}$
Stellerite	$\text{CaAl}_2\text{Si}_9\text{O}_{18} \cdot 7\text{H}_2\text{O}$
Stilbite	$\text{NaCa}_2\text{Al}_5\text{Si}_{13}\text{O}_{36} \cdot 14\text{H}_2\text{O}$
Tetranatrolite	$\text{Na}_2\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 2\text{H}_2\text{O}$
Thomsonite	$\text{NaCa}_2\text{Al}_5\text{Si}_5\text{O}_{20} \cdot 6\text{H}_2\text{O}$
Wairakite	$\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 2\text{H}_2\text{O}$
Wellsite	$(\text{Ba}, \text{Ca}, \text{K}_2)\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 3\text{H}_2\text{O}$
Yugawaralite	$\text{CaAl}_2\text{Si}_6\text{O}_{16} \cdot 4\text{H}_2\text{O}$

The basic structure of zeolite minerals is an open framework of $(\text{Si}, \text{Al})\text{O}_4$ tetrahedra, which accounts for their cation exchange and absorption capacity and for the ability of water molecules to move out of and into the framework. In the zeolite group are representatives of every crystal system except triclinic. Hardness is 3.5 to 5, specific gravity is low (mostly 2.0-2.3), cleavage is good to perfect. Most are gelatinized or otherwise decomposed by acids.

These minerals vary in optical character but typically have low refractive index and low to very low birefringence. Depending on the degree of dehydration of a given species, the optic angle may vary widely, even to changing optic sign and orientation. This, together with the overlap of properties and the lack of adequate documentation of some of the examples, makes it difficult to distinguish between members of this group by optical means.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	β , ω , or n	γ or ϵ					
---	<u>1.461</u>	1.465	.004	CHABAZITE	(+)0°	TRIG	
1.460	<u>1.462</u>	---	.001	CHABAZITE	(-)0°	TRIG	Anom biax.
---	<u>1.468</u>	1.473	.008	ERIONITE	(+)0°	HEX	Diff dec by acids. El pos.
1.468	<u>1.470</u>	1.473	.005	ERIONITE	(+)0°	HEX	Diff dec by acids.
---	<u>1.470</u>	1.474	.004	GMELINITE	(+)0°	TRIG	Tw. 2V 0° to small.
---	<u>1.471</u>	1.474	.003	HERSCHELITE	(+)0°	TRIG	
---	<u>1.473</u>	---	.000	PAULINGITE	---	CUB	
1.470	<u>1.474</u>	---	.004	HERSCHELITE	(-)0°	TRIG	
1.471	<u>1.475</u>	1.476	.005	SODIUM- DACHIARDITE	(-)88- 92°	MCL	Z:c = 18°.
1.472	<u>1.475</u>	1.477	.005	MORDENITE	(-)80°	ORTH	X = <u>a</u> , Z = <u>c</u>
---	<u>1.477</u>	1.480	.003	ERIONITE	(+)0°	HEX	Diff dec by HCl. El pos.
1.475	<u>1.477</u>	---	.002	GMELINITE	(-)0°	TRIG	Anom biax. Tw.
1.476	<u>1.479</u>	1.479	.003	CLINOPTILOLITE	(-)small	MCL	X = <u>b</u> ?, Z:a = 15°.
1.478	<u>1.479</u>	1.481	.003	CLINOPTILOLITE	(+)40°	MCL	X = <u>b</u> , el c1v pos.
1.478	<u>1.479</u>	1.482	.004	FERRIERITE	(+)50°	ORTH	Insol in HCl. X = <u>a</u> , Y = <u>b</u> .

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\beta, \omega,$ or n	γ or ϵ					
---	<u>1.479</u>	---	.000	ANALCIME	---	CUB	Opt anom.
---	<u>1.480</u>	1.493	.013	TETRANATROLITE	(+)0°	TET	
1.477	<u>1.480</u>	1.490	.013	NATROLITE	(+)63°	ORTH	X = <u>a</u> , Z = <u>c</u> , el pos.
---	<u>1.48</u>	---	wk	FAUJASITE	---	CUB	On losing H ₂ O, becomes uniax pos.
1.480	(<u>1.480</u>)	1.485	.005	CHABAZITE	(+)small	TRIG	
---	<u>1.481</u>	1.483	.002	LEVYNE	(+)0°	TRIG	
---	<u>1.482</u>	(1.484- 1.490)	.002- .008	GMELINITE	(+) ~ 0°	HEX	
1.481	<u>1.482</u>	1.486	.005	MORDENITE	(+)large	ORTH	Partly dec by HCl. X = <u>c</u> , Y = <u>a</u> .
1.481	<u>1.484</u>	1.486	.005	CLINOPTILOLITE	(-)large	MCL	X = <u>b</u> (?), Z: <u>c</u> = 15°.
1.483	<u>1.485</u>	1.487	.004	PHILLIPSITE	(+)63°	MCL	X = <u>b</u> , Z: <u>c</u> = 19°
1.479	<u>1.485</u>	1.489	.010	BARRERITE	(-78°)	ORTH	X = <u>a</u> , Y = <u>b</u> .
1.485	<u>1.485</u>	1.488	.003	CHABAZITE	(+)med	TRIG	Tw.
1.483	<u>1.486</u>	1.495	.012	NATROLITE	(+)med	ORTH	X = <u>a</u> , Z = <u>c</u> , tw.
1.484	<u>1.486</u>	1.487	.003	PHILLIPSITE	(-)med	MCL	Penet tw.
---	<u>1.487</u>	1.489	.002	CHABAZITE	(+)0°	TRIG	Anom biax.
1.486	<u>1.487</u>	---	.001	ANALCIME	(-)small	CUB	Anom biax, lam tw.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	β , ω , or n	γ or ϵ					
1.488	<u>1.488</u>	1.489	.001	HEULANDITE	(+)70°	MCL	EI clv neg.
1.484	(<u>1.488</u>)	1.490	.006	CHABAZITE	(-)60-70°	TRIG	Anom biax, tw.
1.482	<u>1.489</u>	1.496	.014	STILBITE	(-)43°	MCL	$Y = b$, $X:a$ $= 2-9°$, eI neg. Cruciform penet tw.
1.489	<u>1.489</u>	1.492	.003	FERRIERITE	(+)small	ORTH	Insol in HCl. EI pos.
1.486	<u>1.489</u>	---	.003	OFFRETTITE	(-)0°	HEX	EI pos.
1.486	<u>1.489</u>	---	.003	CHABAZITE	(-)0°	TRIG	Anom biax.
1.487	<u>1.489</u>	---	.002	LEVYNE	(-)0°	TRIG	
1.490	<u>1.490</u>	1.502	.012	NATROLITE	(+)0°	ORTH	EI pos.
1.485	<u>1.490</u>	1.494	.009	AMICITE	(-)82°	MCL	
1.491	<u>1.492</u>	1.494	.003	CHABAZITE	(+)58°	TRIG	Tw.
1.491	<u>1.493</u>	1.497	.006	CLINOPTIOLITE	(+)72°	MCL	EI clv pos.
---	<u>1.493</u>	---	.000	ANALCIME	---	CUB	
1.486	<u>1.494</u>	1.496	.010	SELLERITE	(-)38-48°	ORTH	$X = a$, $Y = c$, eI clv neg.
1.489	<u>1.494</u>	---	.005	GOBBINSITE	(-)0°	TET	EI neg.
1.493	<u>1.496</u>	1.500	.007	DACHIARDITE	(+)69°	MCL	Mimetic tw.
1.495	<u>1.497</u>	1.504	.009	YUGAWARALITE	(+)78°	MCL	Insol in acids. U opt neg.
1.495	<u>1.498</u>	1.502	.007	GOOSECREEKITE	(-)82°	MCL	$Y = b$, $Z:c$ $= 46°$.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	β , ω , or \underline{n}	γ or ϵ					
1.497	(1.498)	1.499	.002	GONNARDITE	(+)med to large	ORTH	
1.496	1.498	1.504	.008	HEULANDITE	(+)35°	MCL	El clv neg.
1.499	1.500	1.502	.003	WAIRAKITE	(+)med	MCL	
---	1.500	1.502	.002	GARRONITE	(+)0°	TET	
1.499	1.500	1.501	.002	MERLINOITE	(-)56°	ORTH	X = b, Z = <u>a</u> , e \bar{t} pos.
1.498	(1.501)	1.502	.004	WAIRAKITE	(-)75°	MCL	X ~ b, Z ~ c. U biax pos.
1.500	1.501	1.504	.004	WELLSITE	(+)39°	MCL	Complex tw.
1.494	1.502	1.507	.013	STILBITE	(-)30- 50°	MCL	Y = b, X:a = 2-9°, e \bar{t} neg. Cruciform penet tw.
1.498	1.503	1.506	.008	HARMOTOME	(-)75- 82°	MCL	Z = b, Y:c = 28°
---	1.503	1.507	.004	CHABAZITE	(+)0°	TRIG	2V small
1.500	1.503	1.510	.010	PHILLIPSITE	(+)med	MCL	Tw.
1.501	1.504	1.509	.008	HARMOTOME	(+)78°	MCL	Cruciform tw.
1.504	1.505	1.507	.003	MESOLITE	(+)large	MCL	
---	1.505	---	.000	ANALCIME	---	CUB	Diff dec by acids. Cs_2O 14.9%.
1.499	1.506	---	.007	MAZZITE	(-)0°	HEX	
1.505	1.507	1.507	.002	GARRONITE	(-)0°	ORTH, ps tet	

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	β , ω , or n	γ or ϵ					
1.504	<u>1.507</u>	1.515	.011	HEULANDITE	(-)10-40°	MCL	E1 clv neg.
---	<u>1.507</u>	---	.000	POLLUCITE	---	CUB	F diff.
1.505	<u>1.509</u>	1.509	.004	COWLESITE	(-)30°	ORTH	X = <u>a</u> , Y = <u>b</u> , e \bar{t} pos.
1.505	<u>1.509</u>	1.513	.008	HARMOTOME	~ 90°	MCL	Cruciform tw.
1.505	<u>1.509</u>	1.511	.006	PHILLIPSITE	(-)med	MCL	Penet tw.
1.508	<u>1.510</u>	1.513	.005	GONNARDITE	(+)large	ORTH	
1.502	<u>1.510</u>	---	.008	LEVYNE	(-)0°	TRIG	
1.508	<u>1.511</u>	1.523	.015	BREWSTERITE	(+)55°	MCL	E1 clv neg.
1.508	<u>1.511</u>	1.512	.004	GARRONITE	(-)0-30°	ORTH	
1.500	<u>1.511</u>	1.513	.013	EPISTILBITE	(-)40°	MCL	X = <u>b</u> , Z: <u>c</u> = 9°
1.504	<u>1.512</u>	1.516	.012	LAUMONTITE	(-)44°	MCL	Y = <u>b</u> , Z: <u>c</u> = 30°
1.511	<u>1.513</u>	1.518	.007	THOMSONITE	(+)75°	ORTH	X = <u>a</u> , Z = <u>b</u>
1.512	<u>1.515</u>	1.517	.005	COWLESITE	(-)44-53°	ORTH	X = <u>a</u> , Z = <u>c</u> , e \bar{t} pos.
1.512	<u>1.515</u>	1.515	.003	GARRONITE	(-)0°	ORTH, ps tet	
1.512	<u>1.516</u>	1.518	.006	GISMONDINE	(-)large	MCL	X = <u>b</u> , Y ~ <u>c</u> . Tw.
1.512	<u>1.518</u>	1.519	.007	SCOLOCITE	(-)35°	MCL	Z = <u>b</u> , X: <u>c</u> = 18°, e \bar{l} neg.
1.515	<u>1.518</u>	---	.003	CHABAZITE	(-)0°	TRIG	Anom biax.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta}{\alpha}$, $\frac{\omega}{\alpha}$, or $\frac{n}{\alpha}$	γ or ϵ					
1.513	<u>1.524</u>	1.525	.012	LAUMONTITE	(-)25°	MCL	$\gamma = b$, $Z:c$ = 20-30°.
1.523	<u>1.525</u>	1.532	.009	THOMSONITE	(+)48°	ORTH	$X = a$, $Z = \underline{b}$
---	<u>1.525</u>	---	.000	POLLUCITE	---	CUB	
1.522	<u>1.528</u>	1.530	.008	GISMONDINE	(-)15-90°	MCL	$X = b$, $\gamma \sim c$. $\underline{T}w.$
1.530	<u>1.533</u>	1.542	.012	THOMSONITE	(+)52°	ORTH	$X = a$, $Z = \underline{b}$
1.538	<u>1.543</u>	1.548	.010	GISMONDINE	(-)15-90°	MCL	$X = b$, $\gamma \sim c$. $\underline{T}w.$
1.541	<u>1.553</u>	1.557	.016	EDINGTONITE	(-)54°	ORTH	$X = c$, $Z = a$, el neg.

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