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Literature Review of the Potential Effects of Hydrogen Peroxide on Nitrogen Oxidation Efficiency of the Biofilters of Recirculating Aquaculture Systems (RAS) for Freshwater Finfish

By Kim T. Fredricks

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Conversion Factors

International System of Units to Inch/Pound

Multiply	By	To obtain
Area		
square meter (m ²)	10.7639	square foot (ft ²)
square meter (m ²)	1.19599	square yard (yd ²)
Length		
Micrometer (μm)	3.937 x 10 ⁻⁵	inch (in.)
Volume		
liter (L)	33.814	ounce, fluid (fl. oz)
liter (L)	2.11338	pint (pt)
liter (L)	1.05669	quart (qt)
liter (L)	0.264172	gallon (gal)
cubic meter (m ³)	264.172	gallon (gal)
Flow rate		
centimeters per second (cm/s)	0.0328	foot per second (ft/s)
Mass		
milligram (mg)	3.5274 x 10 ⁻⁵	ounce, avoirdupois (oz)
kilogram (kg)	2.20462	pound, avoirdupois (lb)
metric ton (tonnes)	0.0984207	pound, avoirdupois (lb)
metric ton (tonnes)	1.016	ton, long (2,240 lb)
Density		
kilogram per cubic meter (kg/m ³)	0.062428	pound per cubic foot (lb/ft ³)
kilogram per cubic meter (kg/m ³)	0.008345	pound per US gallon (lb/gal)
kilogram per square meter (kg/m ²)	0.20482	pound per square foot (lb/ft ²)

Temperature in degrees Celsius (°C) may be converted to degrees Fahrenheit (°F) as °F = (1.8 × °C) + 32.

Supplemental Information

Concentrations of chemical constituents in water are given in milligrams per liter (mg/L).

Abbreviations

CaCO ₃	calcium carbonate
C/N	carbon nitrogen ratio
CO ₂	carbon dioxide
DO	dissolved oxygen
FDA	U.S. Food and Drug Administration
HP	hydrogen peroxide
NH ₃ -N	un-ionized ammonia-nitrogen
NO ₃ -N	nitrate-nitrogen
PAA	peracetic acid
PVC	polyvinyl chloride
RAS	recirculating aquaculture systems
rRNA	ribosomal ribonucleic acid
TAN	total ammonia nitrogen
UMESC	Upper Midwest Environmental Sciences Center
USGS	U.S. Geological Survey

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Summary

A comprehensive literature review was done for the effects of hydrogen peroxide (HP) on biofilter function in recirculating aquaculture systems (RAS) using these databases: ISI/Web of Knowledge, Scopus, and Pubmed. Inclusion and exclusion criteria were developed as the literature review was conducted. The initial search produced 5,748 potential citations. Once the literature search was complete, these 5,748 titles were screened for applicable papers using the inclusion and exclusion criteria. If the title contained any of the inclusion terms, it was retained. Titles of the remaining papers were then screened for exclusion terms. If the title contained one or more of the exclusion terms, it was eliminated from further consideration. This refined search produced 1,405 papers.

After the initial screening, the remaining 1,405 papers underwent a second screening. Titles and abstracts (when available) were again read to verify that the topic of the paper was related to RAS. During the second screening, a second person verified that the papers proposed for elimination were not related to RAS. A combined reference list of the 512 remaining papers was created and submitted to the U.S. Geological Survey (USGS) Upper Midwest Environmental Sciences Center (UMESC) librarian in order to obtain the actual papers; electronic copies of those citations were obtained and reviewed. The UMESC librarian also received weekly updates from Scopus (a bibliographic database containing abstracts and citations for academic journal articles) using the search terms. Any resulting papers from those updates were screened using the inclusion criteria and relevant papers were requested. From those, 86 were cited in the literature review. An additional 11 papers from other search methods (e.g., mining references lists) also were obtained.

I. Overview of Aquaculture

Aquaculture may be defined as cultivation of saleable aquatic products under controlled conditions using the highest stocking density possible, the highest quality

feeds, and active water-quality management (Ebeling and others, 2006). Aquaculture accounts for about 50 percent of the world's food fish (Food and Agriculture Organization of the United Nations, 2014), particularly in Asia and developing countries (Håstein and others, 2006; Crab and others, 2007). By 2030, an estimated 62 percent of fish for human consumption will come from aquaculture (Food and Agriculture Organization of the United Nations, 2014). In 2004, 17.3 million tonnes of carp (*Cyprinus carpio*), 1.2 million tonnes of tilapia (*Tilapia spp.*), 1.1 million tonnes of salmon (*Oncorhynchus sp.*), and 0.5 million tonnes each of rainbow trout (*Oncorhynchus mykiss*) and shrimp were produced using aquaculture methods (Håstein and others, 2006). Salmon farming is the main economic activity in Chile, valued at about US\$ 224.4 million (Bravo and others, 2013). Sustained and enhanced productivity are major goals of aquaculture, and current technologies have led to the expansion of aquaculture production (Panigrahi and Azad, 2007).

Aquaculture systems can be classified as one of three types: extensive systems (which require little daily care by a fish culturist), intensive systems (in which fish survival depends completely on the intervention by a fish culturist), and intensive recirculating systems (Ebeling and others, 2006). RAS, in particular, are intensive because of the additional need to manage water quality in addition to feeding. This brings on unique challenges as well as benefits. This review will focus on recirculation systems, particularly those that use freshwater.

II. Recirculating Aquaculture Systems (RAS)

A. Overview of RAS

RAS are controlled environments used to intensively rear fish and other aquatic organisms in ecologically sustainable ways (Summerfelt and others, 2001; Wik and others, 2009). RAS technology is designed for intensive fish farming where water availability is limited (Avnimelech, 2006; Ebeling and others, 2006; Badiola and others, 2012) or where biosecurity concerns limit water intake (Avnimelech, 2006). Intensive recirculating systems recycle 90 to 99 percent of the water used to grow fish or other aquatic organisms (Summerfelt and others, 2001; Ebeling and others, 2006; Park and others, 2008; Badiola and others, 2012). With greater operator control over water quality, an operator can optimize growth conditions for the species being reared (Badiola and others, 2012; Carrera and others, 2013). Maintaining water quality for growout systems may provide greater economic output for the producer. Critical water-quality variables in RAS include dissolved oxygen, carbon dioxide, ammonia, nitrite, nitrate, pheromones, endocrine disrupters that may affect fish maturation, and other toxic chemicals that may leach from rubber or plastics used in building the RAS (Colt and others, 2006). RAS typically have a unit for growing fish, a mechanical filter to remove larger particles before biofiltration, an aerobic biological nitrification area to remove potentially toxic nitrogenous compounds, and sometimes an anaerobic denitrification filter (Barbu and others, 2008).

Biofilters are used to help control the accumulation of ammonia-nitrogen in RAS. The main driver of water quality in RAS is the feeding rate. As fish metabolize protein in feed, they release ammonia as a waste product. Oxygen levels within the culture tank are reduced as a result of bacterial metabolism of nitrogenous compounds ($NH_4^+ + O_2 \rightarrow$

NO_3^-). Oxygenation of culture tanks had a significant impact on reducing the ammonia concentration that reached the biofilters (Caraman and others, 2010). Nitrifying bacteria are well known for their ability to secrete and embed in a lipopolysaccharide matrix that helps create a stable biofilm (Hagopian and Riley, 1998). Fast growing heterotrophic bacteria form the outer layers of biofilms where they can use available oxygen. Oxygen use by heterotrophic bacteria, however, can limit the nitrification processes because oxygen is required for ammonia oxidation (Chen and others, 2006). In one study, nitrification was limited to the top 50 micrometers (μm) of the biofilm (Schramm and others, 1996). Slow growing nitrifying bacteria are protected under the outer layer (Michaud and others, 2006).

Attached biofilms on biofilter material are advantageous in that there is increased efficiency from a large surface area and stability that reduces the chance of biofilms being washed out. *Nitrosomanas* form dense clusters in the upper part of the biofilm and *Nitrobacter* cluster around the *Nitrosomanas* clusters, though less densely. Nitrifiers in the deeper anoxic layers are relatively uncommon (Schramm and others, 1996). Nitrification processes also can be inhibited by heterotrophic processes when organic carbon is present. Nitrification decreased about 70 percent due to heterotrophic processes when organic carbon was present in RAS in a carbon to nitrogen (C/N) ratio of 1 or more (Zhu and Chen, 2001). In production, biofilters may operate under an organic load that is high enough to inhibit the nitrification efficiency so it is beneficial for RAS operators to remove organic matter to maximize nitrification processes. (A more complete overview of RAS can be found in the Engineering Assessment by Holder Timmons Engineering and Julie Bebak, written commun., 2014).

B. Advantages of RAS

There are several advantages to RAS. Recirculation of water helps reduce use. Compared with a flow-through system, use of a rotating biological contactor in a RAS unit reduced water consumption by 90 percent with no negative effects on specific growth or condition factors of rainbow trout (d'Orbecastel and others, 2009a; Marin and others, 2011). Within the effluent of a non-RAS facility are high levels of phosphorus, carbon, and nitrogen from feed waste, which can have a negative environmental impact. Food conversion can be increased in RAS, thereby reducing the potential environmental impact. The lower water-exchange rates in RAS also allow for temperature control, which can lead to year-round production (Gutierrez-Wing and Malone, 2006; Lyssenko and Wheaton, 2006) and help reduce energy costs associated with maintaining a particular temperature (Summerfelt and others, 2001; Avnimelech, 2006; Gutierrez-Wing and Malone, 2006). RAS also allow for greater biosecurity and independence in location of production facilities (Summerfelt and others, 2001; Cancino-Madariaga and others, 2011). Hybrid striped bass, for example, have been reared in the deserts of California (Gutierrez-Wing and Malone, 2006). Lastly, RAS allow for higher fish-production rates and a higher density of fish per rearing tank (Lyssenko and Wheaton, 2006; Good and others, 2009; Gullian-Klanian and Arámburu-Adame, 2013). For example, in one study, freshwater eel (*Abguilla japonica*) were raised at a density of 2.6 percent (compared to 0.6 percent traditionally) in a nearly closed system that only replaced water lost to evaporation (Suzuki and others, 2003).

C. Challenges

A significant challenge with RAS is that, unlike flow-through systems, nitrogenous and organic-waste products accumulate in the rearing tank. RAS thus require biofilters to remove the nitrogenous wastes and settling tanks to help remove organic wastes. If fish are stocked before the biofilter is fully functional, fish can develop methemoglobinemia (Svobodová and others, 2005). Establishing biofilters is a challenge because the bacteria that oxidize ammonia and nitrite to less toxic nitrate are slower growing. Allowing a natural population of bacteria to colonize a biofilter can take 4 to 8 weeks (Kuhn and others, 2010). One reason that biofilters take time to establish is that the heterotrophic bacteria can out-compete the nitrifying bacteria for space and oxygen because of their faster growth rates (Cancino-Madariaga and others, 2011; Michaud and others, 2006; Brambilla and others, 2008; Blancheton and others, 2013). Using a starter culture can significantly decrease the time needed to establish a biofilter (Gross and others, 2003; Kuhn and others, 2010).

Other technologies are being investigated as a means to remove ammonia from RAS that do not have the same challenges as establishing a biofilter. For example, reverse osmosis membranes also may help with ammonia removal in RAS (Cancino-Madariaga and others, 2011). Ion-exchange resin also can help remove ammonia from RAS. These resins are cheaper to use, require a smaller volume than a typical biofilter, do not require a long start-up, are not temperature sensitive, and can be turned on and off at will. Additionally, the resins convert ammonia directly to nitrogen gas (Gendel and Lahav, 2013).

A second challenge with RAS is that the higher rearing density may stress fish, reducing their immune function, which may then provide greater opportunity for pathogens to infect fish. If a pathogenic organism is introduced into the system, either from addition of new stock or replacement water, it may persist in the biofilms within the system (King and others, 2008; Jacobs and Chenia, 2011). This can result in recurring fish mortality and economic losses to the facility. Many of the pathogens are opportunistic and cause disease only if the fish are compromised or if their environment is suboptimal. The organic peroxide peracetic acid (PAA) had the greatest effect on inhibiting growth of pathogenic bacteria in biofilms of RAS (King and others, 2008). However, there is always a concern that use of disinfectants may impair the nitrifying bacteria in the biofilters, resulting in accumulation of ammonia or nitrite, which is toxic to fish (Summerfelt and others, 2001; Pedersen and others, 2009).

A third challenge is off-flavors in fish that can result in economic loss for the producers. These off-flavors are produced by bacteria *Nocardia cummidelens*, *N. fluminea*, *Streptomyces roseoflavus*, *Streptomyces luridiscabiei*, and *Streptomyces cf. albidoflavus* that were isolated from RAS biofilters (Guttman and van Rijn, 2008; Schrader and Summerfelt, 2010; Auffret and others, 2013) and heat exchangers (Schrader and Summerfelt, 2010).

Biofilter operation and solids management were identified as main issues in RAS in a survey conducted in Europe (Badiola and others, 2012). Integration of the current technology to manage biofilters and solids within commercial scale RAS is a challenge. One reason for this is that much of the research has been at laboratory scale, which does not translate well to commercial-scale operations (Badiola and others, 2012). Dissolved organic carbon from feed waste or improper mechanical filtration can reduce nitrification

because the heterotrophic bacteria feed on the carbon and overgrow the autotrophic nitrifying bacteria within the biofilms (Guerdat and others, 2011).

Other challenges with aquaculture, both freshwater and marine, include food-safety concerns. Rearing fish in crowded tanks can increase the risk of disease transmission, and may require the use of antibiotics or other therapeutic agents to prevent loss of the crop. For example, *Streptococcus iniae* may cause chronic septicemia in fish. It is commonly isolated from cultured species including rainbow trout, tilapia, channel catfish (*Ictalurus nebulosus*), and Japanese flounder (*Paralichthys olivenceus*) (Håstein and others, 2006). Antibiotic use, particularly in countries where regulation is not as stringent as it is in the U.S., is a concern because it can lead to the development of resistant bacteria. Those resistance determinants may then be transferred to zoonotic pathogens which could impact humans or other commercially important animals. For example, molecular evidence exists that *Aeromonas* can transmit determinants for resistance to antibiotics with the human pathogen *Escherichia coli* (Cabello, 2006). Horizontal gene transfer of bacterial resistance has been shown from aquaculture to humans (Heuer and others, 2009). Antimicrobial use in developed countries is more restricted to try to prevent resistance in human pathogens; however, in other countries, their use is not as restricted. In 2007, for example, Canada used 0.175 kilograms (kg) of antimicrobials/metric ton of salmon produced whereas Chile used 1.4 kg per metric ton of salmon produced (Cabello and others, 2013). Other countries are reducing their use of antibiotics to reduce the potential for resistance. Norway reduced antibiotic use by 37 percent from 1992 to 1996 and by 96 percent in fish that were farmed, even though biomass of farmed fish increased by 100 percent (Grave and others, 1999).

Aquaculture produces substantial nutrient and organic matter that must be treated. Nutrients can increase phytoplankton (i.e., eutrophication) and potentially decrease water quality. Excess organic matter can limit dissolved oxygen, which could negatively affect other aquatic life in the area near the discharge of wastewater (Ghaly and others, 2005). Higher levels of nitrogen and phosphorus released with wastewater can accelerate eutrophication (Hussenot, 2003). Mats created from waste feed and discharged fecal matter may decrease benthic biodiversity and increase biological oxygen demand (Droppo and others, 2007). Some of these environmental concerns can be mitigated. For example, phytotreatment can be used to help remove nitrogen and phosphorus from water before it is discharged (Porrello and others, 2003). The plants that are produced can be used as feed, which may provide another source of income. Rye, barley, and oats were found to be the best plants at removing nitrogen and phosphorus in one study (Ghaly and others, 2005).

Finally, investment costs are higher with RAS compared to flow-through systems. In 1998, it cost about US\$0.90 per pound (lb) of annual production in a pond system compared to US\$1.00–4.00 for RAS (Gutierrez-Wing and Malone, 2006). The payback period also is longer than other systems, on average about 8 years (Badiola and others, 2012). Overall energy costs are typically higher with RAS compared to flow-through as well (Singh and Marsh, 1996; d'Orbcastel and others, 2009b).

D. Current Work with Biofilters

With increased biomass in RAS, water quality often becomes a limiting factor for growth, mostly due to increased ammonia and nitrites produced as fish metabolize

protein in feed (Hagopian and Riley, 1998). Water quality can be controlled by replacement, which often requires high exchange rates, or by recycling water through a biological filter (Avnimelech, 2006). Biological filters, or biofilters, are an important component of RAS and are used to reduce total ammonia (ionized and un-ionized ammonia) through nitrification processes that oxidize ammonia to nitrite then nitrate. Biofilters can be made of a variety of materials but are often divided into two main categories. Fixed film biofilters have media for microorganism attachment and growth whereas suspended growth biofilters maintain the microorganisms in suspension. Suspended growth biofilters require more management and are not commonly used in RAS (Gutierrez-Wing and Malone, 2006). Effective biofilters have high populations of nitrifying bacteria that help reduce water required for tank exchanges in RAS (Guerdat and others, 2010). However, if organic load in the RAS is high, some heterotrophic bacteria can migrate to the biofilter and impair function of the nitrifying bacteria (Barbu, 2012). Carbon dioxide (CO₂) is produced by biofilters through respiration of the nitrifying bacteria. Summerfelt and Sharrer (2004) estimated that the biofilter accounted for 37 percent of the total CO₂ produced in a salmonid RAS and suggested that a CO₂ stripping unit be placed immediately after the biofilter, where CO₂ is at its highest level, to minimize the CO₂ fish are exposed to as the water recirculates. Currently (2015), there is no standardized method to evaluate biofilter performance, though several authors have suggested the need for developing standards for the industry to help plan for media and volume of media used in RAS (Colt and others, 2006; Malone and Pfeiffer, 2006).

Fixed film biofilters used in recirculating systems include rotating biological contractors (Brazil, 2006; Marin and others, 2011), trickling filters (Kamstra and others, 1998; Eding and others, 2006; Barbu, 2012), sequencing batch reactors (Fontenot and others, 2007; De Schryver and Verstraete, 2009; Luo and others, 2013), and fluidized bed biofilters (Skjølstrup and others, 1998; Malone and Beecher, 2000; Timmons and others, 2006; Weaver, 2006; Davidson and others, 2008). Various media and flow rates have been studied, and one study indicated that media characteristics may be more important than flow rates at affecting the efficacy of the filter (Yang and others, 2001). The filter material should have a surface capable of developing biofilms and capturing organic solids (Yang and others, 2001). Media size and surface area often influences maximal feed load, which affects the amount of total ammonia nitrogen (TAN) that can be removed (Summerfelt, 2006; Pfeiffer and Wills, 2011). Media can vary between commercially available media (Pfeiffer and Wills, 2011) to something as simple as polyvinyl chloride (PVC) shavings (Prinsloo and others, 1999). Other low cost materials that have been used in effective biofilters include lava stones and oyster shells (Ogunlela and Ogunlana, 2011), Styrofoam beads (Peng and others, 2003), wood chips and wheat straw (Saliling and others, 2007), polystyrene microbeads (Timmons and others, 2006), and scrub pads and plastic hair rollers (Al-Hafedh and others, 2003). More surface area does not necessarily mean more TAN removal. In one study, commercially available Siporax glass beads with a water contact area of 32,000 square meters (m²) were only 93 percent efficient at converting ammonia nitrogen (NH₃-N) to nitrate nitrogen (NO₃-N), while the PVC shavings with a water contact area of 1,220 m² were 96 percent efficient (Prinsloo and others, 1999).

In rotating biological contractors, media (typically molded disks with a high surface area) is attached to a central horizontal shaft. The rotation of the filter partly

submerges some of the media and rotation allows for shearing of excess biofilm growth (Brazil, 2006). Trickling biofilters, on the other hand, use a fixed media through which pre-filtered water is trickled down the height of the filter over a thin aerobic biofilm. Metabolites from the biofilm are picked up and carried out with the wastewater. Trickling allows for oxygenation of the biofilm and degassing of carbon dioxide. Advantages of trickling filters include high process stability due to constant high oxygen levels; carbon dioxide (CO₂) removal by degassing; water cooling in summertime; and simplicity of design, construction, operation, and management. Some disadvantages of trickling filters are the relatively low volumetric removal rates (with consequently large sized biofilters), biofilm shedding, and risk of clogging when not properly designed and operated. For certain fish species, additional solids removal is necessary (Eding and others, 2006).

Sequencing batch reactors incorporate alternating aerobic and anaerobic periods to achieve nitrification and denitrification in a single container. The aerobic phase is followed by an anaerobic phase, then a second aerobic phase, and a settling phase (Fontenot and others, 2007). Lastly, fluidized bed biofilters have a higher surface area per unit volume than other fixed film biofilters (Weaver, 2006) and can be filled with microbeads (Malone and Beecher, 2000; Timmons and others, 2006) or sand (Weaver, 2006; Davidson and others, 2008). Fluidized sand biofilters are efficient, compact, cost competitive, and relatively easy to manage (Sandu and others, 2002; Summerfelt, 2006; Weaver, 2006). Providing adequate oxygen is key to their function as about 60 percent of the dissolved oxygen used in a fluidized sand biofilter went toward nitrification (Summerfelt and Sharrer, 2004). Biofilm shearing can be done by pumping the top particles to the bottom. The released biofilm particles are less dense than the sand and simply wash out with the effluent (Davidson and others, 2008). An advantage to sand is that it has a high surface area for a relatively low cost. These biofilters can be configured to fit the space available and can treat a high volume of water with a relatively small footprint, removing 86 to 88 percent of TAN (Davidson and others, 2008). Additionally, Sandu and others (2002) found that TAN removal rate increased as column diameter increased. They also suggested that the ideal fluidized bed filter would be filled to 40–50 percent capacity with plastic medium, and upflow velocity would be 1–2 centimeters per second (cm/s) at the desired flow rate (Sandu and others, 2002). Microbead filters are a low-cost alternative to fluidized sand filters because they can be scaled to large production systems. A key advantage of microbead filters is that their cost of operation is less than a fluidized sand biofilter because of the ability to use low head high volume pumps for their operation (Timmons and others, 2006).

In a study designed to assess how three different filter types performed under production-scale tilapia culture, the floating bead and fluidized sand filters outperformed the moving bed bioreactor in terms of ammonia removal, though nitrification rates in fluidized bed biofilters were more variable. Nitrite levels were in acceptable ranges for the study duration, with the fluidized bed and moving bed bioreactor filters showing variability. The floating bead biofilter showed a net nitrite production that was attributed to capture and degradation of suspended organic solids at the bottom of the filter. This type of filter serves both as a biofilter and a mechanical filter, and captured solids are degraded between back washings, which were every 2 hours during the study (Guerdat and others, 2010). Another study showed that floating media used less energy to maintain

flow compared to sand filters, which was seen as an advantage, particularly when both filter types had similar solids removal and turbidity reduction (Steicke and others, 2007).

Short-term shocks to biofilters, such as a change in pH, TAN, or temperature, did not negatively affect biofilter operation, and nitrification rates returned to normal within 2 hours. Trickling filters seemed more sensitive to disturbances, perhaps because of a more immediate contact of the water on biofilm. The rate and magnitude of change may be important variables in biofilter performance, especially if change is rapid (Lyssenko and Wheaton, 2006).

Biofilms on biofilter media form layers, and movement of ammonia or nitrite into the biofilm is diffusion controlled (Rittmann and McCarty, 1980). The nitrification rate is in equilibrium between demand for ammonia or nitrite created by growth of the bacterial biomass and the rate of supply of ammonia and nitrite that can be affected by more than 20 physical, chemical, and biological factors; many of which can be optimized in RAS design. Ammonia concentration should be the primary consideration in biofilter design because ammonia is toxic to many fish species at low levels. For example, toxic levels of ammonia are 0.32 milligrams per liter (mg/L) for rainbow trout and 0.05–0.2 mg/L for other salmonids (Summerfelt and others, 2001). The highest water quality that can be maintained in RAS is defined by the minimum amount of ammonia that can sustain the nitrifier population in the biofilter. Dissolved oxygen should be maintained at >2.3 mg/L to sustain nitrification within biofilters (Chen and others, 2006), and a DO/TAN of >2.5 is recommended (Chen and others, 2006; Summerfelt, 2006). Dissolved oxygen may drop rapidly in the biofilm despite sufficient dissolved oxygen in the culture water and may become the limiting factor to nitrification if it falls below 2 mg/L. Turbulence can impair mass transfer into the biofilm and has the greatest effect on nutrient transfer into the biofilm. An alkalinity of >200 mg/L is recommended to maintain optimum nitrification rates when water exchange is minimal. Alkalinity ensures adequate buffering capacity to prevent pH changes from the H⁺ produced during nitrification (Chen and others, 2006). Nitrification can consume from 6.0 to 7.4 milligrams (mg) of alkalinity (as CaCO₃) and about 4 mg of oxygen for every 1.0 mg of ammonia nitrogen that is converted to nitrate-nitrogen (Summerfelt and others, 2001).

Additional research on biofilters focused on use of other nitrogen oxidizing organisms and immobilizing bacteria on various substrates to help start the nitrification processes. *Aspergillus niger*, a fungus, was introduced into a laboratory-scale (1 liter [L]) fixed slab reactor and was found to use ammonia at low temperatures (22 degrees Celsius [°C]), but switched to carbon at higher temperatures (35 °C). If used in a production scale biofilter, the operator would likely have to know the C/N ratio of waste produced (Hwang and others, 2004). A similar study examined the ammonia removal rates in a continuous stirred tank reactor in a RAS. *Aspergillus niger* was the dominant species that removed ammonia at a daily rate of 35 mg TAN/L. It required only 20 hours to reach a constant removal rate; however, this system also required ozone treatment (Hwang and others, 2007).

A study by Watanabe and others (2004) demonstrated that immobilized ammonia oxidizing bacteria, *Nitrosomonas europaea*, helped remove ammonia from rearing water. This technology may help start the nitrification process since the bacteria could be present in high amounts from the start. However, they also found that the ammonia oxidizers were damaged by light during the time the gel-plate solidified (Watanabe and

others, 2004). Immobilizing bacteria in a fluidized bed biofilter also was investigated as a way to increase fish resistance to disease. The bacteria *Rhodopseudomonas palustris*, which can inhibit and destroy reproduction of viruses, was immobilized on alginate gels covered with polyvinyl alcohol film. The resulting beads were then placed in a fluidized bed before rainbow trout or common carp (*Cyprinus carpio*) were added. During the study, fish growth and development was normal and survival rate was 98 percent. Maximum fish load was 45 ± 3 kilograms per cubic meter (kg/m^3), ammonia nitrogen removal rate was 80–95 percent, and nitrite N removal rate was <80 percent (Peirong and Wei, 2013).

E. Bacteria That Process Nitrogen

Two distinct groups of bacteria process nitrogen: (1) the ammonia-oxidizing bacteria of the genera *Nitrosomonas*, *Nitrosococcus*, *Nitrospira*, *Nitrosolobus*, and *Nitrosovibrio*; and (2) the nitrite-oxidizing bacteria of the genera *Nitrobacter*, *Nitrococcus*, *Nitrospira*, and *Nitrospina*. Although growth is slow, doubling time is shorter for the ammonia-oxidizing bacteria (about 26 hours) than for nitrite-oxidizing bacteria (about 60 hours). Optimal temperature for nitrification is about 25 °C, and optimal pH is 7.8 (Hagopian and Riley, 1998).

Additionally, differences are seen in bacterial populations in RAS biofilters and culture water and because each fish species introduces a unique microbiological flora, diversity varies from RAS to RAS. Diversity also is affected by use of disinfection (i.e., ozone treatments) that can affect culture water but does not affect the deeper biofilm layers. Diversity may be measured by determining the 16S rRNA sequences of bacteria in the system. Ammonia oxidation is associated with *Nitrosomonas* sp. in freshwater systems. Nitrite and ammonia oxidation occur together in nitrification biofilters, with *Nitrospira* sp. being the dominant nitrite oxidizer (Schreier and others, 2010). In anaerobic areas, such as the deeper layers of the biofilm, high nitrate/nitrite levels and organic carbon wastes promote heterotrophic denitrification. *Pseudomonas* and *Comamonas* spp. are likely heterotrophs that participate in denitrification in freshwater systems (Schreier and others, 2010).

Anaerobic ammonium-oxidation or anammox bacteria in the taxon Planctomyces were discovered in the late 1990s. They convert ammonia and nitrite to nitrogen gas and water in the absence of oxygen. These bacteria are slow growing organisms (doubling times >11 days, Schreier and others, 2010), whose growth is inhibited by oxygen so they are likely found in the low oxygen areas of the biofilm on biofilters. The likely source of these anammox bacteria is fish feces (van Kessel and others, 2011). In a RAS culturing *Scortum barcoo*, the *Nitrospira* formed only a small part of the overall bacterial community and could not account for all the oxidation of nitrite to nitrate. It was thought that some of the *Planctomycetales*-related clones were participating in the ammonia removal process after the biofilter stabilized (Zhu and others, 2012). In another study, anammox-like bacteria were detected by 16S rRNA analysis in a RAS containing Koi carp (*Cyprinus carpio Koi*) but anammox activity was not sufficient to account for significant ammonia removal (van Kessel and others, 2010).

III. Diseases

Bacterial, viral, fungal, and parasitic pathogens, both opportunistic and obligate, cause disease in RAS. Obligate pathogens require a host to survive and do not live long outside of the fish. The major fish pathogens include *Flavobacteria*, *Saprolegnia*, *Aeromonas*, *Vibrio*, and *Yersinia* (Blancheton and others, 2013), and these can potentially infect fish cultured in RAS. In one study, sea bass and mullet reared in recirculation systems showed signs of infection by several bacterial species including *Aeromonas* ssp, *Myxobacteria*, and *Saprolegnia*. Parasites also infested the same species including *Ichthyophthirius* sp, *Chilodonella cephalus*, and *Diplectanum aequans* (Athanasopoulou and others, 2004). Furunculosis was found in a recirculation system growing Senegalese sole (*Solea senegalensis*) held at a stocking density of 25 kilograms per square meter (kg/m²) (Magariños and others, 2011). In another study, Actinomycetes were isolated from a drum filter and heat exchanger in a RAS (Schrader and Summerfelt, 2010).

Heterotrophic bacteria are common in RAS and may cause diseases and out-compete autotrophic bacteria. They also produce significant quantities of bacterial biomass that may clog biofilters and limit nitrification (Michaud and others, 2006). Foam fractionation had variable success in removing heterotrophic bacteria (Brambilla and others, 2008).

Although it may seem that RAS would be at higher risks for disease outbreak, there were no papers found that indicated well-managed RAS had higher incidences of disease. RAS tend to be biosecure; water replacement rates are low, which may limit invasions by pathogens; and they may include a disinfection unit that reduces the risk of disease (Blancheton and others, 2013). However, if water quality is not maintained, fish may develop infections. Nile tilapia showed a higher incidence of parasitic infestation by *Trichodina* and monogeneans when nitrogen compounds in the RAS increased (Jiménez-García and others, 2012).

IV. Use of Hydrogen Peroxide in RAS

A. Current Label

At time of publication, 35 % Perox-Aid® (Western Chemical, Ferndale, Washington) is the only HP product approved by the U.S. Food and Drug Administration (FDA) for use in U.S. aquaculture. It is specifically approved to control mortality of freshwater finfish eggs due to saprolegniasis (fungi of the family Saprolegniaceae). It also is approved to control bacterial gill disease in all freshwater-reared salmonids, and to control external columnaris disease on all coolwater finfish and channel catfish. Detailed information can be found on their Web site:

<http://www.accessdata.fda.gov/scripts/animaldrugsatfda/details.cfm?dn=141-255>.

B. Application of Hydrogen Peroxide in RAS—Effects on Biofilter Bacteria

With antimicrobial resistance becoming an increasing concern in human and veterinary clinical practice, associated with the use of antimicrobials in agriculture (including aquaculture; Defoirdt and others, 2007; Guglielmetti and others, 2009; Fernández-Alarcón and others, 2010; Dang and others, 2011; Defoirdt and others, 2011), other treatments to control bacterial infections must be developed. HP has been approved

for use in flow-through systems and can be used in RAS if the user contacts the Western Chemical for proper procedures regarding flushing of the drug from the RAS. Some drugs have been shown to inhibit the nitrification process in RAS, including ampicillin, chloramine T, and erythromycin (Nimenya and others, 1999). Additionally, there are concerns about worker safety with the use of formalin, or release of higher saline water if sodium chloride is used (Pedersen and Pedersen, 2012).

HP is an oxidizing agent that is approved as a fish therapeutant. It decomposes to water and oxygen and does not accumulate (Pedersen and others, 2006). HP also conforms to many of the principles of green chemistry: the reduction or elimination of the use of hazardous materials in the design, manufacture, and use of chemicals (Pedersen and Pedersen, 2012). A summary of some key variables in RAS exposed to HP is summarized in table 1.

Table 1. Summary of papers that tested the effects of hydrogen peroxide on freshwater recirculating aquaculture systems. [kg/m³, kilogram per cubic meter; °C, degrees Celsius; mg/L, milligrams per liter]

Citation	Scale	Fish biomass (kg/m ³)	Water temperature (°C)	Hydrogen peroxide dose (mg/L)	Duration (hours)	Biofilter type
Pedersen and others, 2006	Pilot	7.5	15.5	13	Single addition	Submerged upflow
Pedersen and others, 2006	Pilot	7.5	15.5	26	Single addition	Submerged upflow
Pedersen and others, 2006	Pilot	7.5	15.5	39	Single addition	Submerged upflow
Møller and others, 2010	Pilot	20–30	20 ± 1	15	Single addition	Submerged
Møller and others, 2010	Pilot	20–30	20 ± 1	10–13	3	Submerged
Pedersen and Pedersen, 2012	Commercial	20–23.3 (calculated)	16.5	64	Single addition	Not reported
Pedersen and Pedersen, 2012	Commercial	20–23.3 (calculated)	16.5	45	Single addition	Not reported
Pedersen and Pedersen, 2012	Commercial	20–23.3 (calculated)	16.5	20	3	Not reported
Pedersen and others, 2012	Pilot	24	18	15	Single addition	Submerged upflow
Pedersen and others, 2012	Pilot	92	18	15	Single addition	Submerged upflow

When HP was applied to active biofilters in RAS, the HP decay followed first order kinetics (Pedersen and others, 2006; Pedersen and others, 2009; Møller and others, 2010). Decomposition of HP varied with the amount of organic matter and the initial dose of HP. If loading rates into a RAS exceed the microbial degradation rate, then severe negative effects could occur within the biofilter microbial community (Pedersen and others, 2006).

When biofilters from two pilot-scale RAS were exposed to 0, 5, 10, 20, 25, or 30 mg/L HP, the decay followed first order kinetics, and the nitrogen oxidation processes recovered quickly after a single dose of up to 30 mg/L. An average dose of 10–13 mg/L over 3 hours of repeated HP spiking had a moderate inhibitory effect on nitrification in a pilot-scale biofilter with a high organic load but had a severe inhibitory effect on a pilot-scale biofilter with a low organic load. The biofilters of the RAS with the low organic load had not recovered by 24 hours post-exposure indicating potential long-term perturbations of the biofilter (Møller and others, 2010).

HP also was applied to a production-scale freshwater RAS that produced rainbow trout. Normal aquaculture practices were followed during the study. Throughout the experimental time period, the farm had about 30–35 metric tonnes of rainbow trout. Two studies were performed that evaluated biofilter performance: (1) high dose (64mg/L), single application to a closed biofilter section, and (2) multiple doses to the raceways with evaluation of biofilter response. Water temperature was about 16.5 °C during the experiments. Results from part 1 showed an increase in oxygen, indicating that heterotrophic and nitrifying bacteria were immediately inhibited. Ammonia oxidation rates were significantly less than controls for 7 days after exposure. Nitrite oxidation also was negatively affected, but to a lesser extent. Addition of HP resulted in liberation of a large amount of organic matter, which was controlled by enclosing the biofilter section and redirecting COD-enriched water to the sludge compartment. In the second study, 15 mg/L HP was applied using the standard practice of adding the chemical to one end of the raceway. The concentration was maintained at about 15 mg/L for 3 hours by addition of HP. Fish were observed aggregating downstream of the HP pulse. The half-life of HP was found to be 1.5 hours. The biofilter elements were initially by-passed (which was not standard practice) for 3 hours then exposed to residual HP for 45 minutes, resulting in a slight decrease in TAN removal rates compared to control sections. No negative effects on fish were seen during the 3-hour treatment (Pedersen and Pedersen, 2012).

In another pilot-scale RAS study using rainbow trout, the effects of low dose (10–20 mg/L) HP on biofilter function was investigated under these conditions: low intensity (24 kg/m³) and high intensity (92 kg/m³) culture conditions. Water temperature in the six 1,700 L units was maintained at about 18 °C. Water renewal rate was fixed at 80 liters per day (L/d) for both groups. Before HP was added, the inflow and outflow from the culture tanks were stopped, resulting in two closed systems: the rearing unit, and the biofilter unit to achieve a target of 15 mg/L HP. HP was then added to either the culture tank or to the biofilter. Water samples were analyzed over 4 hours for HP. Half-life for HP added to the biofilter of high intensity units was 13 minutes, and half-life for the low intensity group was 63 minutes. HP decay also was different when it was added to the culture tanks. About 95 percent of the added HP had decayed in 1.5 hours from the high intensity culture tanks; HP concentration was essentially 0 by 2 hours. Only 25 percent was removed in 1.5 hours in the low intensity culture tanks, and concentration

leveled out at about 10 mg/L for the remainder of the experiment. Feed rate was the same for all culture tanks until the day after the exposure when feed rate to the low intensity culture RAS tanks was increased by 4 times (to the same level as the high intensity tanks). Nitrite concentration increased after the increase in feed rate and remained elevated for about 8 days in the low intensity RAS, after which nitrite levels stabilized at low levels. Because nitrite went over 1 mg nitrite-nitrogen per liter, feed rate was reduced on day 2 post-exposure. Conversely, nitrite levels were not changed in the high intensity RAS units. Ammonia concentration was not affected in either type of RAS (Pedersen and others, 2012).

V. Conclusions

- HP may be a useful tool to help mitigate disease occurrence in RAS, particularly at lower doses of HP.
- In a system with a lower organic load, the nitrite oxidation step may be negatively impacted for a prolonged time (Møller and others, 2010; Pedersen and others, 2012).
- In systems where biofilters are exposed to prolonged higher levels of HP, the nitrite oxidation step may be negatively impacted for a prolonged time (Møller and others, 2010; Pedersen and others, 2012).
- Because of the rapid decay of HP in culture water, it may be possible to treat fish if the biofilter can be bypassed temporarily, or if the biofilter can be bypassed such that it receives only a low residual concentration of HP. A potential negative of bypassing the biofilter is the potential for it to become a parasite reservoir (Pedersen and others, 2012).
- Based on the studies reviewed for this literature review, it is likely that pilot-scale data would be very different from commercial-scale data. Commercial systems have higher carbon loads and mature bacterial populations in the biofilter, which may result in even faster rates of decay.

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