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DEPARTMENT OF THE INTERIOR
UNITED STATES GEOLOGICAL SURVEY

CHARLES D. WALCOTT, DIRECTOR

MINERAL ANALYSES

FROM THE

LABORATORIES OF THE UNITED STATES GEOLOGICAL SURVEY

1880 to 1903

TABULATED BY

F. W. CLARKE, Chief Chemist



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LETTER OF TRANSMITTAL.

DEPARTMENT OF THE INTERIOR,
UNITED STATES GEOLOGICAL SURVEY,
Washington, D. C., October 3, 1903.

SIR: I have the honor to transmit herewith a paper entitled "Mineral Analyses from the Laboratories of the United States Geological Survey, 1880-1903, tabulated by F. W. Clarke, Chief Chemist." Much of the material of this paper is unpublished, and it is very desirable that the whole should be printed in a conveniently accessible form.

Very respectfully,

GEORGE F. BECKER,
*Geologist in Charge,
Division of Physical and Chemical Research.*

Hon. CHARLES D. WALCOTT,
Director United States Geological Survey.

MINERAL ANALYSES FROM LABORATORIES OF UNITED STATES GEOLOGICAL SURVEY, 1880-1903.

Tabulated by F. W. CLARKE, *Chief Chemist.*

INTRODUCTION.

In the present bulletin I have brought together and classified 507 analyses of minerals from the laboratory records of the United States Geological Survey. A large proportion of these analyses have been already published elsewhere, but they are so scattered that their collection in one place seemed to be desirable. Over 150 distinct species are represented, and among them a considerable number were originally described by chemists connected with the Survey. These species are josephinite, cuprobismutite, warrenite, guitermanite, elpasolite, zunyite, ptilolite, hydronephelite, lucasite, antlerite, knoxvillite, redingtonite, plumbojarosite, emmonsite, and powellite. In other cases minerals which were imperfectly described have been more sharply characterized and their true composition made known. Natrojarosite, for example, is here definitely recognized as a species.

In general the order of Dana's classification has been followed, but with some small variations. Thus the tellurides are put in a group by themselves, the borosilicates are brought together, while the phosphates, vanadates, and arsenates are given as three separate classes. On purely chemical grounds these changes are warranted; on morphological grounds the usual mineralogical classification may be better.

Detailed descriptions of the various minerals have been purposely omitted. Neither are the analyses discussed. These deficiencies, however, are mitigated by the literature references to descriptions published elsewhere. Such references as "annual report," "bulletin," or "monograph" relate to the regular publications of this Survey; all other references are easily recognizable.

I. NATIVE ELEMENTS.

GOLD.

From Persia; exact locality unknown. Analyzed by Charles Catlett.

Au	93.24
Ag	6.65
Fe11
Cu	none
	<hr/> 100.00

JOSEPHINITE.

A nickel-iron alloy found in placer gravels in Jackson and Josephine counties, Oreg. Described as a new species by W. H. Melville, Am. Jour. Sci., 3d ser., vol. 43, 1892, p. 509. Waterworn pebbles. Specific gravity, 6.204.

Ni	60.45
Co55
Fe	23.22
Pyrrhotite55
Chromite12
Magnetite	
Cu50
As23
Cl04
Silicates (anhydrous)	12.26
H ₂ O below 100°81
H ₂ O above 100°	1.12
Volatile matter70
CO ₂	trace

100.55

The silicate admixture, including the water, amounts to 13.38 per cent, of which 12.88 per cent, soluble in hydrochloric acid, is serpentine. The insoluble portion may be bronzite. Analyses of the silicates are as follows:

	Total.	Insoluble portion.	Soluble portion.
SiO ₂	5.14	0.23	4.91
Al ₂ O ₃33	.03	.30
Fe ₂ O ₃	2.08	.04	2.04
NiO, CoO32	trace	.32
CaO	1.62	.06	1.56
MgO	2.69	.14	2.55
Na ₂ O08		.08
H ₂ O above 100°	1.12		1.12
	<hr/> 13.38	<hr/> .50	<hr/> 12.88

For analyses of nickel-iron of meteoric origin, see Bulletin 168.

II. SULPHIDES AND ARSENIDES.

BISMUTHINITE.

From the Rosario Mining District, Sinaloa, Mexico. Analysis by W. H. Melville. Described by him in Bulletin 90.

Bi	72.90
Pb	6.03
Cu	1.67
Fe35
S	18.11
Quartz63
	<hr/>
	99.69

METACINNABARITE.

Crystallized material from New Almaden, Cal. Analysis by W. H. Melville. Not free from admixtures.

Hg	78.01
S	13.68
Fe61
Co	trace
Zn90
Mn15
CaCO ₃71
Quartz	4.57
Organic matter63
	<hr/>
	99.26

Another specimen of metacinnabarite from Knoxville, Cal., gave Melville the following figures:

HgS	98.48
FeS69
SiO ₂71
	<hr/>
	99.88

The latter mineral was described by Melville and Lindgren in Bulletin 61.

COVELLITE.

From the East Greyrock mine, Butte, Mont. Collected by G. W. Tower. Analysis by W. F. Hillebrand. Color, indigo-blue. Massive. Specific gravity, 4.76 at 26°.

Cu	66.06
S	33.87
Fe14
Insoluble11
	<hr/>
	100.18

POLYDYMITE.

A massive ore from the mine of the Canadian Copper Company, Sudbury, district of Algoma, Ontario. Specific gravity, 4.541. Analysis by Charles Catlett. Described by Clarke and Catlett in Am. Jour. Sci., 3d ser., vol. 37, 1889, p. 372. Composition nearly Ni_3FeS_5 .

A. Actual analysis.

B. Analysis corrected by deduction of quartz and chalcopyrite.

	A.	B.
Ni	41.96	43.18
Fe	15.57	15.47
S	40.80	41.35
Cu62
SiO_2	1.02
	99.97	100.00

Another nickel-iron sulphide, from the Worthington mine, Sault branch of the Canadian Pacific Railway, 25 miles west of Sudbury, has been analyzed by W. F. Hillebrand. Grayish, with a cast of yellow. Not pyrrhotite. Possibly a mixture of polydymite and pyrite.

Fe	38.36
Ni	4.57
Mn10
S	45.11
SO_395
CO_2	^a 1.49
CaO	1.91
MgO41
H_2O at 105°55
Insoluble	4.80
	98.25

STROMEYERITE.

From the Silver King mine, Calico, San Bernardino County, Cal. Specific gravity, 6.28. Analysis by W. H. Melville. Described by Melville and Lindgren in Bulletin 61.

Ag	53.96
Cu	28.58
Fe26
S	15.51
Residue	1.55
	99.86

SULPHIDE OF SILVER, COPPER, AND ZINC.

Massive, resembling bornite. Apparently homogeneous, but may be a mixture. Specific gravity, 5.407 at 20° . Analyzed by W. F.

^a Calculated to saturate CaO .

Hillebrand and described in Bulletin 55. From the Gagnon mine, Butte, Mont.

Cu	40.24
Ag	21.80
Pb	1.46
Zn	12.83
Fe	1.98
S	20.88
	<hr/>
	99.19

LÖLLINGITE.

From Teocalli Mountain, Brush Creek, Gunnison County, Colo. Specific gravity, 7.400 at 14.5°, corrected for impurity. Analyzed by W. F. Hillebrand and described in Bulletin 20.

As	71.18
S56
Bi08
Cu39
Fe	22.96
Co	4.37
Ni21
	<hr/>
	99.75

A doubtful arsenide of nickel and cobalt has also been examined by Hillebrand and described in Proc. Colorado Sci. Soc., vol. 3, pt. 1, p. 46. From the Rose mine, Grant County, N. Mex. Specific gravity, 6.644 at 20°. Probably a mixture. Ni: Co=3:1, approximately.

As	74.04
S13
Ag	4.78
Cu04
Pb03
Fe44
Ni, Co	19.52
CaO09
MgO05
	<hr/>
	99.12

ENARGITE.

From the Rarus mine, Butte, Mont. Collected by G. W. Tower. Analysis by W. F. Hillebrand.

Cu	48.67
Fe33
Zn10
As	17.91
Sb	1.76
S	31.44
Insoluble11
	<hr/>
	100.32

CUPROBISMUTITE.

New species, discovered by W. F. Hillebrand and described by him in Bulletin 20. Named by Dana. Specific gravity, 6.680 at 15°, corrected for impurities. From the Missouri mine, Halls Valley, Park County, Colo. Analyses, by Hillebrand, of three different samples.

	I.	II.	III.
Bi	60.74	63.36	62.45
Ag89	4.09	9.89
Cu	15.96	12.65	6.68
Pb			2.74
Fe	2.13	.59	.10
Zn10	.07	.07
S	^a 19.94	^a 18.83	17.90
	99.76	99.59	99.83

^a Calculated.

ZINKENITE.

From the Brobdignag mine, Red Mountain, San Juan County, Colo. Specific gravity, 5.21 at 18°. Analyzed by W. F. Hillebrand and described by him in Bulletin 20.

	I.	II.
Sb	35.00	
As	5.64	5.59
Pb	32.77	32.79
Cu	1.20	
Ag23	
Fe02	
CaO31	
Alkalies45	
S	22.50	22.50
Gangue59	
	98.71	

WARRENITE.

New species, described by L. G. Eakins in Am. Jour. Sci., 3d ser., vol. 36, 1888, p. 450. From the Domingo mine, Gunnison County, Colo. Occurs in matted, fibrous masses; known locally as "mineral wool." Analysis by Eakins.

Pb	39.33
Sb	36.34
S	21.19
Ag	trace
Fe	1.77
Mn	trace
Insoluble52
	<hr/>
	99.15

COSALITE.

From the Comstock mine, near Parrott City, La Plata County, Colo.
Described by W. F. Hillebrand in Bulletin 20.

Bi	42.93
Ag	8.43
Cu	7.50
Pb	22.49
Fe70
Zn	trace
S	17.11
	<hr/>
	99.16

FREIESLEBENITE.

From Augusta Mountain, Gunnison County, Colo. Known locally as "mineral wool." Remarkable for its freedom from silver. Analyzed by L. G. Eakins and described by him in Am. Jour. Sci., 3d ser., vol. 36, 1888, p. 452.

Pb	55.52
Sb	25.99
S (calculated)	18.98
Ag	trace
Fe	trace
Zn	trace
	<hr/>
	100.49

GUITERMANITE.

New species, discovered by W. F. Hillebrand and described by him in Bulletin 20. From the Zufi mine, Anvil Mountain, near Silverton, San Juan County, Colo. Forms the matrix of zunyite. Corrected specific gravity, 5.94 at 17.5°. Analyses by Hillebrand.

	I.	II.
Admixed zunyite	1.77	3.82
As	13.40	13.00
Pb	63.60	61.63
Cu17	.17
Ag02	.02
Fe43	.88
S	19.67	19.56
O55
	<hr/>	
	99.06	99.63

III. TELLURIDES.

HESSITE.

From San Sebastian, Jalisco, Mexico. Specific gravity, 8.24 at 26°. Analysis by W. F. Hillebrand.

Ag	61.16
Te	36.11
Pb	1.90
S, Fe, Zn	undet.
	99.17

PETZITE.

From the Norwegian mine, Calaveras County, Cal. Collected by F. L. Ransome. Specific gravity, 8.925 at 23°. Analysis by W. F. Hillebrand. Formula, $\text{Au}_2\text{Te} \cdot 3\text{Ag}_2\text{Te}$.

Ag	41.87
Au	25.16
Te	33.21
Se	trace
Mo08
	100.32

CALAVERITE.

From Cripple Creek, Colo. Collected by R. A. F. Penrose, jr. Analyzed by W. F. Hillebrand and described by him in Bulletin 167.

A. From the Prince Albert mine. Corrected specific gravity, 9.0 at 24°.

B. Raven mine.

C. C. O. D. mine.

	A.	B.	C
Te	57.27	47.69	53.89
Au	38.95	33.93	39.31
Ag	3.21	1.47	.85
Insoluble matter33	5.80	.91
Fe_2O_312		
Fe		5.41	1.67
S		^a 6.17	1.58
Mn			^b .23
Ca51
Mg10
O, F, and soluble SiO_2 by difference95
	99.88	100.47	100.00

^aCalculated from Fe to make FeS_2 .

^bAs MnO_3 ?

MELONITE.

From the Melonés mine, Carson Hill, Calaveras County, Cal. Collected by F. L. Ransome. Analyzed by W. F. Hillebrand and described by him in Bulletin 167. Three samples. Specific gravity of B, 7.72 at 22.5°, which is probably too high for the pure NiTe_2 . Sample C is the purest.

	A.	B.	C.
Te	75.29	77.72	80.75
Ni	15.71	17.16	18.31
Co	}	.10	}
Ag		5.09	.86
	99.44	100.07	99.92

IV. CHLORIDES AND FLUORIDES.

HALITE.

Rock salt from Salton, Cal. Analysis by E. T. Allen.

NaCl	94.54
KCl31
Na_2SO_4	3.53
$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$79
Moisture14
Insoluble residue50
	99.81

EMBOLITE.

From Broken Hill, Australia. Analysis by L. G. Eakins.

AgCl	59.97
AgBr	39.22
H_2O63
	99.82

CRYOLITE.

From St. Peters Dome, Pikes Peak district, Colorado: Described by Cross and Hillebrand in Bulletin 20. Massive. Pinkish in color. Specific gravity, 2.972 at 24°. Analysis by W. F. Hillebrand.

Al	12.81
Na	32.40
Ca28
F	53.55
Fe_2O_340
H_2O30
	99.74

PACHNOLITE.

From St. Peters Dome, Pikes Peak district, Colorado. Described by Cross and Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

A. Compact bluish variety. Specific gravity, 2.980 at 22°.

B. Crystalline coating.

C. Fresh, transparent, crystalline material.

D. Carefully selected crystals. Specific gravity, 2.965 at 17°; 2.962 at 22°.

	A.	B.	C.	D.
Al	11.94	12.93, 12.92	12.14	12.27
Ca	19.32	15.27, 15.17	18.06	18.04
Mg13	1.53	-----	-----
Na	10.43	10.28	10.23	10.25
K	-----	.13	-----	-----
H ₂ O	7.87, 7.95	8.64, 8.79	8.10, 8.11	8.05
F	-----	-----	51.33, 51.28	^a 51.39
			99.86	100.00

^a By difference.

ELPASOLITE.

From St. Peters Dome, Pikes Peak district, Colorado. Described as a new species by Cross and Hillebrand in Bulletin 20. Incomplete analysis on insufficient material, by W. F. Hillebrand. The fluorine was deduced by calculation on the assumption that the metals are fully combined with it.

Al	11.32
Ca72
Mg22
K	28.94
Na	9.90
F	46.98
	98.08

GEARKSUTITE.

From St. Peters Dome, Pikes Peak district, Colorado. Described by Cross and Hillebrand in Bulletin 20. Analysis by W. F. Hillebrand.

Al	15.20
Ca	22.30
Na10
K04
H ₂ O	15.46
F	42.07
	95.17
Oxygen (loss)	4.83
	100.00

PROSOPITE.

Two samples of prosopite have been analyzed by W. F. Hillebrand, as follows:

A. From St. Peters Dome, Pikes Peak district, Colorado. Described by Cross and Hillebrand in Bulletin 20. Specific gravity, 2.880 at 23°. Mean of four analyses.

B. Pale green, massive variety, from the Dugway mining district, Tooele County, Utah. Specific gravity, 2.87 at 21°. Described by Hillebrand in Bulletin 167.

	A.	B.
Al	22.02	20.08
Ca	17.28	17.55
Mg17	trace
K12
Na48	.32
Cu17
F	33.18	28.00
H ₂ O	13.46	14.24
	86.59	80.48
Oxygen (loss)	13.41	^a 19.52
	100.00	100.00

^a Including a little quartz; undetermined.

TYSONITE.

From Cheyenne Mountain, near Pikes Peak, Colorado. Analyzed by W. F. Hillebrand and described by him in Bulletin 167. Corrected specific gravity, 6.14 at 28°.

Ce ₂ O ₃ (includes 0.13 ThO ₂ ?)	42.89
La ₂ O ₃ group (at. wt. 139.7)	39.31
F	28.71
CO ₂53
CaO18
Na ₂ O30
K ₂ O and Li ₂ O	traces
Fe ₂ O ₃11
	112.03
Less O equivalent to F	12.08
	99.95

V. OXIDES AND HYDROXIDES.

SPINEL.

Variety pleonaste. Separated from a pyroxenite found between South Meadow and Moore creeks, Madison County, Mont. Rock described by Merrill in Proc. U. S. Nat. Mus., vol. 17, p. 659. Specific gravity of spinel, 3.89 at 32°.3. Analysis by L. G. Eakins.

Al ₂ O ₃	62.09
FeO.....	17.56
MgO.....	15.61
Cr ₂ O ₃	2.62
Fe ₂ O ₃	2.10
MnO.....	trace
CaO.....	.16
SiO ₂55
	<hr/>
	100.69

GAHNITE.

From Gilmore's mica mine, Montgomery County, Md., about 12 miles north of Washington, D. C. Color, bottle-green. Specific gravity, 4.59. Analyzed by T. M. Chatard and described by him in Bulletin 9.

Al ₂ O ₃	55.46
Fe ₂ O ₃	2.77
ZnO.....	40.07
MgO.....	.59
CuO.....	undet.
SiO ₂57
Ignition.....	.30
	<hr/>
	99.76

MAGNETITE.

From the Gallatin Range, between Middle and Bozeman creeks, near Bozeman, Mont. Analysis by T. M. Chatard.

Insoluble.....	0.16
Fe ₃ O ₄	96.70
Al ₂ O ₃04
MnO.....	.93
CaO.....	trace
MgO.....	.07
TiO ₂	2.71
P ₂ O ₅012
S.....	.171
	<hr/>
	100.793
Less O equivalent to S.....	.06
	<hr/>
	100.733

CHROMITE.

From Corundum Hill, North Carolina. Analysis by T. M. Chatard.
 Fe_2O_3 not separately determined.

Cr_2O_3	45.94
FeO	42.90
Al_2O_3	2.51
MnO84
NiO, CoO16
CaO	1.40
MgO	2.81
CuO40
SiO_2	3.20
TiO_236
P_2O_512
	<hr/> 100.64

RUTILE.

From near St. Peters Dome, Pikes Peak district, Colorado. Specific gravity, 4.288 at 19°. Analysis by L. G. Eakins.

TiO_2	94.93
FeO	3.77
SiO_2	1.37
H_2O71
	<hr/> 100.78

DIASPORE.

From Mount Robinson, Custer County, Colo. Analysis by L. G. Eakins.

Al_2O_3	83.97
H_2O	15.43
	<hr/> 99.40

BAUXITE.

Two samples from Jacksonville, Calhoun County, Ala. Alkalies, lime, and magnesia not looked for. Analyses by W. F. Hillebrand.

	Red.	White.
Al_2O_3	41.00	48.92
Fe_2O_3	25.25	2.14
H_2O at 100°.....	.65	.45
H_2O by ignition.....	20.43	23.41
SiO_2	10.25	21.08
TiO_2	2.53	2.52
P_2O_5	trace	trace
	<hr/> 100.11	<hr/> 98.52

BRUCITE.

From Texas, Lancaster County, Pa. Analysis by E. A. Schneider:

MgO	67.97
MnO97
Fe ₂ O ₃39
H ₂ O at 105°18
H ₂ O, ignition	30.63
	<hr/> 100.14

PSILOMELANE.

From a prospect hole on plain south of Round Mountain, Silver Cliff, Colo. Analysis by W. F. Hillebrand. There were strong spectroscopic reactions for strontia and lithia, but these bases were not estimated. A little insoluble gangue is included with the silica.

MnO ₂	76.18
MnO	5.71
H ₂ O below 120°	1.41
H ₂ O above 120°	3.94
Al ₂ O ₃	1.81
Fe ₂ O ₃34
CoO	trace
ZnO	2.80
CaO83
MgO29
K ₂ O	3.46
Na ₂ O81
Sb ₂ O ₅12
SiO ₂	2.30
	<hr/> 100.00

VI. CARBONATES.

DOLOMITE.

A. Dolomite marble, New York Quarry Company, Tuckahoe, Westchester County, N. Y. Analysis by W. F. Hillebrand.

B. Dolomite marble, Cockeysville, Md. Analysis by J. E. Whitfield.

C. Same as B. Analysis by E. A. Schneider. Another sample.

D. Dolomite marble, Webster, Mass. Analysis by H. N. Stokes.

E. Pink, crystallized dolomite, Joplin, Mo. Analysis by E. T. Allen.

	A.	B.	C.	D.	E.
Insoluble.....	0.62	-----	5.57	-----	-----
SiO ₂71	0.44	-----	1.01	-----
Al ₂ O ₃	-----	1.22	} .40	.17	-----
Fe ₂ O ₃	} .21	-----		none	-----
FeO.....		trace	-----	.37	0.90
MnO.....	-----	-----	-----	.08	.06
MgO.....	20.71	20.87	20.30	21.35	19.65
CaO.....	30.68	30.73	29.08	30.82	32.05
CO ₂	46.66	45.85	44.26	45.84	46.82
H ₂ O.....	.16	1.22	-----	.09	.48
K ₂ O.....	-----	-----	-----	.10	-----
Na ₂ O.....	-----	-----	-----	.01	-----
P ₂ O ₅	-----	-----	-----	.06	-----
TiO ₂	-----	-----	-----	trace	-----
S.....	-----	-----	-----	trace	-----
	99.75	100.33	99.61	99.90	99.96

SMITHSONITE.

From Marion County, Ark. Bright yellow variety, known locally as "turkey-fat ore." Analysis by H. N. Stokes.

SiO ₂	0.06
ZnO.....	64.12
CdO.....	.63
CdS.....	.25
CO ₂	34.68
CuO.....	trace
FeO.....	.14
CaO.....	.38
	100.26

BASTNÄSITE.

Associated with the tysonite of Cheyenne Mountain, near Pikes Peak, Colorado. Specific gravity, 5.12 at 27°. Analyzed by W. F. Hillebrand and described by him in Bulletin 167.

Ce ₂ O ₃	^a 37.71
La ₂ O ₃ group	36.29
Fe ₂ O ₃22
CO ₂	20.03
Na ₂ O.....	.18
K ₂ O, Li ₂ O	traces
H ₂ O.....	.08
F.....	7.83
	102.34
O equivalent to F.....	3.30
	99.04

URAO.

Deposited from the waters of Owens Lake, California. Analyses by T. M. Chatard. Described by Chatard in Bulletin 60. Five samples analyzed, as follows:

- A. The best material. Specific gravity, 2.1473 at 21.7°.
- B. Crude urao.
- C. Urao deposited upon a grass root.
- D. From a small lagoon.
- E. From a vat dug on the beach of Owens Lake.

	A.	B.	C.	D.	E.
Insoluble, inorganic.....	0.02	0.22	2.92	0.40	4.10
Insoluble, organic.....			.14	.12	.27
SiO ₂10	.05	.09	.04
Cl.....	.19	1.57	2.73	.21	1.83
SO ₃70	.79	.76	.63	.84
CO ₂	38.13	37.00	35.24	37.50	35.10
Na ₂ O.....	41.00	41.26	40.22	40.08	39.86
K ₂ O.....				trace	
CaO.....				.06	
MgO.....				.02	
H ₂ O.....	20.07	19.62	18.31	19.94	18.58
	100.11	100.56	100.37	99.05	100.12
O equivalent to Cl.....	.04	.35	.61	.05	.41
	100.07	100.21	99.76	99.00	99.71

^a Includes 0.10 ThO₂.

VII. SILICATES.

PETALITE.

From Peru, Me., associated with spodumene. Analysis by F. W. Clarke.

Ignition	1.03
SiO ₂	77.29
Al ₂ O ₃	16.95
Fe ₂ O ₃	trace
MnO	trace
Li ₂ O	2.62
Na ₂ O	2.39
K ₂ O	trace
	100.28

ORTHOCLASE AND MICROCLINE.

A. Orthoclase from Silver City, Idaho. Occurs as a gangue mineral. Partial analysis by W. F. Hillebrand.

B. Orthoclase from Mitchell County, N. C. Large cleavage mass. Analysis by E. T. Allen.

C. Feldspar crystal from the nevadite of Chalk Mountain, Summit County, Colo. Analysis by W. F. Hillebrand.

D. Pink orthoclase crystals, from gray porphyry, Johnson Gulch, near Leadville, Colo. Described by Cross in Monograph XII, Appendix A. Analysis by W. F. Hillebrand.

E. Feldspar separated from gabbro. East side of North Fowl Lake, Minnesota. Analysis by W. F. Hillebrand.

F. Feldspar from the elæolite-syenite of Litchfield, Me. Described by Bayley in Bulletin 150, p. 201. Analysis by W. H. Melville.

	A.	B.	C.	D.	E.	F.
SiO ₂	66.28	65.09	65.04	62.22	62.71	65.14
TiO ₂					trace	
Al ₂ O ₃	17.93	18.95	20.40	20.33	19.20	18.19
Fe ₂ O ₃36			1.08	
FeO93	.25
MnO					trace	
CaO42	.79	2.95	.44	.33
SrO					trace	
MgO		none	none		.81	.16
Na ₂ O25	2.29	4.11	3.45	2.96	1.68
K ₂ O	15.12	12.95	9.74	8.31	10.41	14.14
Li ₂ O			trace		none	
H ₂ O at 100°29	1.90	.23	.17
H ₂ O above 100°92	
	99.58	100.06	100.37	99.16	99.69	100.06

G. Flesh-colored microcline from the pegmatite of Jones Falls, Baltimore, Md.

H. Greenish microcline, same locality as G. G and H described by S. L. Powell in Johns Hopkins Univ. Circular, vol. 12, p. 49. Analyses by W. F. Hillebrand.

I. Anorthoclase from the clæolite-syenite of Peaked Butte, Crazy Mountains, Montana. See Wolff and Tarr in Bull. Mus. Comp. Zoöl., Harvard Coll., vol. 16, No. 12, 1893. Analysis by W. F. Hillebrand.

J. Anorthoclase from red soda-granite, Pigeon Point, Minnesota. Described by Bayley in Bulletin 109. Analysis by J. E. Whitfield.

K. Anorthoclase from keratophyre, Marblehead Neck, Massachusetts. Described by Sears, Bull. Mus. Comp. Zoöl., Harvard Coll., vol. 16, No. 9, 1893. Analysis by T. M. Chatard.

	G.	H.	I.	J.	K.
SiO ₂	65.06	68.48	62.31	65.00	65.66
Al ₂ O ₃	18.41	16.11	22.63	18.22	20.05
Fe ₂ O ₃	trace	.20	}	2.64	trace
FeO17			trace
MnO13
MgO04	.03		.06	.18
CaO26	.23	.63	1.06	.67
SrO	trace	trace	.57		
BaO13	.05	.77		
K ₂ O	14.30	12.99	4.79	4.18	6.56
Na ₂ O	1.60	1.27	7.68	8.40	6.98
Li ₂ O	trace	trace			
H ₂ O at 100°04	.06	.16	} .46	.04
H ₂ O above 100°26	.26	.72		.37
	100.10	99.85	100.26	100.02	100.64

ALBITE.

A. From feldspathic schist, central shaft of the Hoosac Tunnel, Berkshire County, Mass. Described by Wolff in Monograph XXIII, pp. 60-187. Analysis by R. B. Riggs.

B, C. From the porphyritic mica-schist of Greylock Mountain, Massachusetts. Described by Wolff, loc. cit. Analyses by R. B. Riggs.

D. From the clæolite-syenite of Litchfield, Me. Described by Bayley in Bulletin 150, p. 201. Specific gravity, 2.622. Analysis by W. H. Melville.

E. From the pegmatite of Jones Falls, Baltimore, Md. Described by S. L. Powell in Johns Hopkins Univ. Circular, vol. 12, p. 49. Analysis by W. F. Hillebrand.

	A.	B.	C.	D.	E.
SiO ₂	69.69	68.08	67.83	68.28	63.72
Al ₂ O ₃	18.60	20.11	19.92	19.62	22.26
Fe ₂ O ₃					
FeO.....				.23	
MgO.....	.20	(?)	(?)	.09	.06
CaO.....	trace	trace	trace	.31	3.58
SrO.....					trace
Na ₂ O.....	10.28	11.00	11.65	10.81	8.98
K ₂ O.....	.40	.36	.25	.39	.76
Li ₂ O.....					trace
MnO.....		trace	trace		
H ₂ O below 100°.....	.42	.31	.12	.09	.09
H ₂ O above 100°.....					.43
	99.59	99.86	99.77	99.82	99.88

ANORTHITE.

A. From Raymond, Me. Associated with idocrase, garnet, pyroxene, and scapolite. Analysis by W. H. Melville.

B. From Phippsburg, Me. Occurrence similar to A. Incomplete analysis by George Steiger.

C. Separated from "hyperite changing to diorite," near Whitaker's ore pit, Wilmington, Del. Described by Chester in Bulletin 59. Analysis by R. B. Riggs.

	A.	B.	C.
SiO ₂	43.13	45.62	44.09
Al ₂ O ₃	30.95	35.29	35.41
Fe ₂ O ₃	1.04		.51
FeO.....	trace		
MnO.....	trace		
CaO.....	19.71	17.31	18.47
MgO.....	.31		none
K ₂ O.....	1.29		.19
Na ₂ O.....	.69		.99
Li ₂ O.....	trace		
H ₂ O at 100°.....	.22		.35
H ₂ O above 100°.....	2.80		
	100.14	98.22	100.01

SODA-LIME FELDSPARS.

A. Transparent oligoclase, from Bakersville, N. C. Analysis by F. W. Clarke.

B. Feldspar separated from porphyrite; Sugar Loaf, northwest of Elk Mountain, Tenmile district, Summit County, Colo. Analysis by W. F. Hillebrand. Much altered.

C. Feldspar separated from andesite; mesa northwest of Parkdale, Colo. Analysis by W. F. Hillebrand.

D. Feldspar separated from gabbro; Brandywine Creek, Wilmington, Del. Specific gravity, 2.592-2.877. Described by Chester in Bulletin 59. Analysis by R. B. Riggs.

E. Feldspar separated from the gabbro of Ashland County, Wis. Analysis by W. F. Hillebrand.

F. Feldspar separated from diabase; near SE. corner of sec. 13, T. 47 N., R. 46 W., Michigan. Described by Van Hise in Monograph XIX. Analysis by T. M. Chatard.

	A.	B.	C.	D.	E.	F.
SiO ₂	62.92	62.96	63.02	70.37	53.30	51.18
TiO ₂					trace	-----
Al ₂ O ₃	25.32	21.51	23.05	18.36	29.03	27.00
Fe ₂ O ₃	trace			.58	.55	3.19
FeO.....		.32		undet.	.23	undet.
CaO.....	4.03	4.00	3.39	5.08	11.40	11.70
SrO.....		.13	trace		trace	-----
BaO.....					trace	-----
MgO.....		.30	trace	.04	.13	1.92
MnO.....	trace			trace	none	.17
K ₂ O.....	.96	1.60	3.92	.63	.40	.41
Na ₂ O.....	6.18	6.15	6.76	4.32	4.87	3.48
Li ₂ O.....		trace	none		none	-----
P ₂ O ₅					trace	-----
H ₂ O.....	.25	2.78	.26	.45	.23	1.19
	99.66	99.75	100.40	99.83	100.14	100.24

G, H. Feldspars separated from olivine-diabase; NE. $\frac{1}{4}$ sec. 13, T. 45 N., R. 1 W., Wisconsin. See Van Hise, Monograph XIX. Analyses by T. M. Chatard.

I. Feldspar from gabbro; southern half of sec. 14, T. 44 N., R. 4 W., Wisconsin. See Van Hise, loc. cit. Analysis by T. M. Chatard.

The following feldspars were separated from Minnesota gabbros for W. S. Bayley and analyzed by W. F. Hillebrand:

J. From average gabbro, south quarter post, sec. 35, T. 61 N., R. 12 W.

K. From gabbro, NW. $\frac{1}{4}$ of SE. $\frac{1}{4}$ sec. 23, T. 62 N., R. 10.

L. From gabbro, center of sec. 25, T. 64 N., R. 8.

M. From gabbro, Duluth and Iron Range Railroad.

N. O. Two feldspars separated from the amphibolite of Palmer Center, Mass. Analyses by W. F. Hillebrand. Specific gravity of N, 2.667 at 24°; of O, 2.677 at 22°. In N, calculation gives about 7.4 and in O 7.6 per cent of admixed quartz.

	G.	H.	I.	J.	K.
SiO ₂	61.65	56.15	51.99	51.89	52.50
Al ₂ O ₃	19.91	26.05	29.32	29.68	30.15
Fe ₂ O ₃	2.28	1.98	1.23	.32	.47
FeO	undet.	undet.		.37	.15
MnO	trace	.13	trace	-----	-----
CaO	4.12	8.70	12.60	12.62	12.82
MgO61	.54	.63	.38	.10
Na ₂ O	4.74	4.79	2.91	3.87	3.72
K ₂ O	5.72	1.56	.28	.50	.53
H ₂ O at 100°95	.13	.03	.07	.25
H ₂ O above 100°64	.54	.39	
TiO ₂	-----	-----	-----	-----	trace
	99.98	100.67	99.53	100.09	100.69

	L.	M.	N.	O.
SiO ₂	52.61	53.45	62.91	60.90
Al ₂ O ₃	29.80	29.77	23.37	24.97
Fe ₂ O ₃57	.33	trace	trace
FeO23	.15	-----	-----
CaO	12.25	11.33	5.83	7.85
MgO20	.11	-----	-----
Na ₂ O	3.80	4.33	7.78	6.26
K ₂ O53	.68	.20	.16
H ₂ O at 100°29	.23	.42	.48
H ₂ O above 100°				
TiO ₂	trace	trace	trace	trace
MnO	trace	-----	-----	-----
	100.23	100.38	100.51	100.62

LEUCITE.

From Mount Vesuvius. A fine crystal. Analysis by George Steiger.

SiO ₂	55.40
Al ₂ O ₃	23.69
CaO16
K ₂ O	19.54
Na ₂ O	1.25
H ₂ O24
	100.28

ENSTATITE.

A. Enstatite from Granville, Mass. Slightly altered. Described by Emerson in Monograph XXIX. Analysis by W. F. Hillebrand.

B. Enstatite separated from the San Emigdio meteorite, found in San Bernardino County, Cal. Analysis by J. E. Whitfield.

C. White, fibrous mineral, near enstatite, from seams in chrysolite rock, Corundum Hill, North Carolina.

D. Altered enstatite, Corundum Hill. Analyses C and D by T. M. Chatard. (See Bulletin 42.) Specific gravity, 2.872.

E. Bronzite separated from the websterite of Hebbville, Md. Described by Williams in Am. Geologist, vol. 6, p. 35. Analysis by T. M. Chatard.

	A.	B.	C.	D.	E.
SiO ₂	54.04	54.42	56.39	56.58	54.53
TiO ₂	none		none	none	undet.
Al ₂ O ₃52		2.31	1.74	1.93
Fe ₂ O ₃	1.51		.16	1.89	1.70
FeO	3.90	14.03	1.96	3.67	8.92
MnO11			.21	.28
NiO23				
Cr ₂ O ₃14			.24	.30
MgO	34.40	29.11	34.57	30.34	29.51
CaO	none	2.46	.04	.59	2.25
Alkalies08			.17	
P ₂ O ₅	none				trace
CO ₂	1.32				
H ₂ O at 100°70		} 4.32	} 4.55	} 1.14
H ₂ O above 100°	3.07				
	100.02	100.02	99.75	99.98	100.56

HYPERSTHENE.

A. From the basalt of Mount Thielsen, Oregon. Incomplete analysis by T. M. Chatard.

B. From gabbro, SE. $\frac{1}{4}$ sec. 20, T. 65 N., R. 4 W., Minnesota. Described by Bayley in Jour. Geol., vol. 3, p. 1. Analysis by E. A. Schneider.

C. From the augite-andesite of the Tokajer-Berg, Hungary. Analysis by W. F. Hillebrand. Specific gravity, 3.495 at 25°.

D, E, F. Three specimens separated from the hypersthene-andesite of the Buffalo Peaks, Colorado. Described by Cross in Bulletin 1. Analyses by W. F. Hillebrand. In D and E alkalis were disregarded. In E and F all the iron is given as FeO. Specific gravity of F, 3.307 at 23°.

	A.	B.	C.	D.	E.	F.
SiO ₂ -----	53.31	48.44	51.44	51.70	51.16	50.04
TiO ₂ -----		undet.	.73			
Al ₂ O ₃ -----	5.99	7.91	.60	1.72	2.15	2.91
Fe ₂ O ₃ -----		.33	2.28	.30		
FeO-----	13.43	20.88	20.77	18.00	18.36	17.81
MnO-----		.92	.88	.36	.36	.12
MgO-----	21.69	19.35	19.93	25.09	24.25	21.74
CaO-----	3.69	1.44	3.80	2.87	3.81	6.70
Na ₂ O-----						.27
H ₂ O at 100°-----		.08				
H ₂ O above 100°-----		none				
	98.11	99.35	100.43	100.04	100.09	99.59

PYROXENES.

A. Diopside separated from the websterite of Hebbville, Md. Described by G. H. Williams, Am. Geologist, vol. 6, p. 35. Analysis by T. M. Chatard.

B. Diopside from the leucite rocks of the Leucite Hills, Wyoming. Described by Cross, Am. Jour. Sci., 4th ser., vol. 4, p. 115. Specific gravity, 3.290 at 20°. Analysis by W. F. Hillebrand.

C. Pyroxene from Moriah, N. Y. The source of associated serpentine. Analysis by Charles Catlett.

D. Dark-gray pyroxene, Montville, N. J. The source of associated serpentine. Analysis by Charles Catlett.

E. Diallage from the gabbro of Ashland County, Wis. Analysis by W. F. Hillebrand.

F. Jeffersonite from Franklin Furnace, N. J. Analyzed by W. F. Hillebrand and described by him in Bulletin 167.

	A.	B.	C.	D.	E.	F.
SiO ₂ -----	51.80	50.86	55.36	51.45	49.80	51.70
TiO ₂ -----	.13	3.03			1.29	
Al ₂ O ₃ -----	2.21		.22	2.94	2.86	.36
Cr ₂ O ₃ -----	.51		none			
Fe ₂ O ₃ -----	1.29	1.19	.18	1.06	2.48	.37
FeO -----	3.50	1.82	.57	.96	10.82	
MnO -----	trace	.03	trace	trace	.37	7.43
ZnO -----						3.31
MgO -----	17.76	17.42	19.53	18.43	15.33	12.57
CaO -----	20.99	23.32	24.48	24.02	16.50	23.68
Na ₂ O -----	undet.	.76			.51	.12
K ₂ O -----	undet.	.42			.12	trace
P ₂ O ₅ -----	trace				trace	
H ₂ O -----	.65	.31		1.08	.33	.65
	98.84	99.16	100.34	99.94	100.41	100.19

G. Augite from nepheline-basalt, Black Mountain, Uvalde quadrangle, Texas. Violet colored. Analysis by W. F. Hillebrand.

H. Augite from dolerite dike, near Valmont, Colo. See Cross, Monograph XXVII. Analysis by L. G. Eakins.

I. Augite from tinguaita, Two Buttes, Colo. Specific gravity, 3.43 at 28°. Analysis by W. F. Hillebrand.

J. Pyroxene from syenitic lamprophyre, Two Buttes, Colo. Specific gravity, 3.45 at 25°. Analysis by W. F. Hillebrand.

K. Augite from granite, north end of Blue Mountains, Silver Cliff, Colo. Specific gravity, 3.225 at 18°. Analysis by L. G. Eakins.

L. Augite from the Golden King dike, Silver Cliff, Colo. Specific gravity, 3.281 at 13°. Analysis by L. G. Eakins.

M. Pyroxene from norite, dike east of Sugar Loaf, Boulder County, Colo. Analysis by L. G. Eakins.

	G.	H.	I.	J.	K.	L.	M.
SiO ₂	45.23	49.10	47.54	51.27	48.72	54.87	47.32
TiO ₂	4.28	-----	3.00	.70	-----	-----	-----
ZrO ₂	-----	-----	none	none	-----	-----	-----
Al ₂ O ₃	7.73	7.95	4.14	3.05	9.27	6.34	6.37
Cr ₂ O ₃	-----	-----	trace?	none	-----	-----	-----
Fe ₂ O ₃	2.95	-----	5.64	3.08	3.77	2.88	2.56
FeO	4.07	8.30	6.42	4.34	6.34	4.61	14.40
MnO07	-----	.36	.28	.34	.14	-----
MgO	12.25	12.37	10.05	14.21	14.67	14.47	13.43
CaO	23.37	22.54	21.57	22.58	16.79	15.87	16.08
Na ₂ O47	trace	1.38	.67	.19	.28	-----
K ₂ O12	trace	.12	.06	-----	-----	-----
Li ₂ O	trace	-----	trace	-----	-----	-----	-----
BaO	none	-----	none	none	-----	-----	-----
SrO	none	-----	none	none	-----	-----	-----
NiO05	-----	trace	.03	-----	-----	-----
P ₂ O ₅	none	-----	-----	-----	-----	-----	-----
H ₂ O37	-----	undet.	undet.	.18	.31	-----
	100.96	100.26	100.22	100.27	100.27	99.77	100.16

N. Pyroxene from Italian Mountain, Gunnison County, Colo. Associated with idocrase, scapolite, garnet, epidote, etc. Analysis by L. G. Eakins. Specific gravity, 3.312 at 16.7°.

O. Augite from basalt, 6 miles northeast of Grants, Mount Taylor region, New Mexico. Analysis by T. M. Chatard.

P. Pyroxene from peridotite, east of Fort Ellis, Mont. Partial analysis by F. W. Clarke.

Q. Pyroxene from basalt, east side of Bozeman Creek, 2½ miles southeast of Bozeman, Mont. Described by Merrill, Proc. U. S. Nat. Mus., vol. 17, p. 637. Analysis by L. G. Eakins.

	N.	O.	P.	Q.
SiO ₂	47.53	47.06	51.95	52.50
TiO ₂		1.82		
Al ₂ O ₃	9.88	7.77	4.21	2.26
Cr ₂ O ₃		trace		1.07
Fe ₂ O ₃	1.79	1.30		2.05
FeO.....	.91	8.15	5.90	2.47
MgO.....	trace	.20	undet.	trace
NiO,CoO.....		trace		
MgO.....	14.43	13.52	13.81	17.11
CaO.....	25.46	19.33	23.32	21.70
Na ₂ O.....	trace	.33		.35
K ₂ O.....		.11		.07
P ₂ O ₅06		
H ₂ O.....	.30	.20	undet.	.64
	100.30	99.85	99.19	100.22

ÆGIRITE.

From Magnet Cove, Arkansas. Analysis by George Steiger.

SiO ₂	50.45
Al ₂ O ₃	2.76
Fe ₂ O ₃	23.42
FeO.....	5.26
MnO.....	.10
MgO.....	1.43
CaO.....	5.92
Na ₂ O.....	9.84
K ₂ O.....	.24
H ₂ O at 100°.....	.15
H ₂ O above 100°.....	.40

100.02

JADEITE.

The following analyses of jadeite, all by F. W. Clarke, were discussed by Clarke and Merrill in the Proceedings of the United States National Museum for 1888. They all represent worked material.

A. Light-colored bead, mottled with emerald green, from State of Oaxaca, Mexico. Specific gravity, 3.007, determined by William Hallock.

B. Carved head, light green, from Zaachita, Oaxaca. Specific gravity, 3.190, Hallock.

C. Fragment from Sardinal, Costa Rica; pale green, translucent. Specific gravity, 3.32, Clarke.

D. Fragment from Culebra, Costa Rica; light green, granular, opaque; quite impure. Specific gravity, 3.27, Clarke.

	A.	B.	C.	D.
SiO ₂	58.88	58.18	59.18	58.33
Al ₂ O ₃	25.93	23.53	22.96	21.63
Cr ₂ O ₃12	-----	-----	-----
Fe ₂ O ₃	} .24	} 1.67	} 1.87	1.71
FeO73
MgO36	1.72	.67	3.09
CaO40	2.35	1.52	4.92
Na ₂ O	11.64	11.81	12.71	8.13
K ₂ O63	.77	trace	.22
Ignition	1.81	.53	.90	.93
	100.01	100.56	99.81	99.69

WOLLASTONITE.

Compact variety, from Diana, N. Y. Partial analysis by E. A. Schneider:

SiO ₂	50.05
Al ₂ O ₃	} 1.13
Fe ₂ O ₃	
CaO.....	47.10
MgO.....	.09
H ₂ O.....	.45
	<hr/> 98.82

PECTOLITE.

A. Stone hammer, at first supposed to be jade, collected among the Eskimo of Point Barrow, Alaska. Analyzed by F. W. Clarke, and described by him in Bulletin 9. Pale apple-green, tough, compact, highly polished. Specific gravity, 2.873.

B. Radiated pectolite from Bergen Hill, New Jersey. Analyzed by E. A. Schneider.

C. Another sample from Bergen Hill. Analysis by George Steiger.

	A.	B.	C.
SiO ₂	53.94	53.11	53.34
Al ₂ O ₃ +Fe ₂ O ₃58	.40	.33
MnO.....		.81	.45
CaO.....	32.21	33.88	33.23
MgO.....	1.43		
Na ₂ O.....	8.57	8.62	9.11
CO ₂67
H ₂ O at 105°.....	} 4.09	.04	.27
H ₂ O, 250°-300°.....		.14	} 2.70
H ₂ O, red heat.....		2.86	
	<hr/> 100.82	<hr/> 99.86	<hr/> 100.10

AMPHIBOLE.

A. Tremolite, pseudomorphous after sahlite; northeast slope of Canaan Mountain, Connecticut. Analysis by W. F. Hillebrand.

B. Tremolite, found in the serpentine of Easton, Pa. Analysis by L. G. Eakins.

C. Actinolite (?), Corundum Hill, North Carolina. Specific gravity, 3.062. Analyzed by T. M. Chatard and described by him in Bulletin 42.

D. Nephrite, New Zealand. Fragment from a dark-green boulder. Analysis by F. W. Clarke.

E. Nephrite, Robenhausen, Lake Pfäffikon, Switzerland. Part of a green, compact, highly polished jade implement. Specific gravity, 3.015; determined by William Hallock. Analysis by F. W. Clarke.

	A.	B.	C.	D.	E.
SiO ₂	57.97	58.27	55.23	56.73	56.87
TiO ₂	none	-----	none	-----	-----
Al ₂ O ₃09	.33	3.04	3.22	1.50
Cr ₂ O ₃	-----	-----	.19	-----	-----
Fe ₂ O ₃11	trace	1.88	} 5.96	6.33
FeO18	-----	2.51		
MnO	trace	.08	.26	trace	trace
NiO, CoO	-----	-----	trace	-----	-----
CaO	15.05	11.90	13.36	13.24	13.45
SrO	trace	-----	-----	-----	-----
MgO	22.45	25.93	22.31	19.42	21.06
K ₂ O12	.42	} .58	undet.	undet.
Na ₂ O20	1.25		undet.	undet.
H ₂ O at 100°03	} 1.22	.04	} .83	} .63
H ₂ O above 100°	2.57		.52		
CO ₂	1.69	-----	-----	-----	-----
	100.46	99.40	99.92	99.40	99.84

F. Dark-green nephrite implement, from the Eskimo of Point Barrow, Alaska. Specific gravity, 3.012. Analysis by F. W. Clarke.

G. Nephrite adze, Point Barrow. Nearly black, with grayish-green patches. Specific gravity, 2.922, Hallock. Analysis by F. W. Clarke.

H. Nephrite adze, Cape Prince of Wales, Alaska. Dark green, laminated in two shades, opaque. Specific gravity, 2.989, Hallock. Analysis by F. W. Clarke.

I. Worked nephrite, St. Michael, Alaska. Dull apple green, fairly uniform in tint, semitranslucent at edges. Specific gravity, 3.006, Hallock. Analysis by F. W. Clarke.

J. Jade implement, Diomed Island, Alaska. Dark green, laminated in two shades, opaque. Specific gravity, 3.010, Hallock. Analysis by F. W. Clarke.

	F.	G.	H.	I.	J.
SiO ₂	57.01	57.11	56.01	56.12	53.08
Al ₂ O ₃42	2.57	1.98	.63	1.01
FeO.....	6.95	5.15	6.34	7.45	7.67
CaO.....	12.75	11.54	12.54	12.72	13.35
MgO.....	21.36	21.38	21.57	20.92	19.96
MnO.....		trace	trace	trace	trace
Ignition.....	1.41	2.06	1.91	1.42	2.03
Alkalies.....	undet.	undet.	undet.	undet.	undet.
	99.90	99.81	100.35	99.26	100.10

K, L, M, N. Four samples of nephrite, found in situ by Lieutenant Stoney, U. S. Navy, near the Kowak River, Alaska. K, greenish gray, splintery-lamellar. L, like K, but more granular. M, paler, nearly white, closer grained. N, brownish, highly foliated. Analyses by F. W. Clarke. Ferrous iron determinations by R. B. Riggs. O, Nephrite from Jordansmühl, Silesia; analysis by George Steiger; typical material from an old locality. For details concerning nephrites D to N, see the memoir by Clarke and Merrill in Proc. U. S. Nat. Mus. for 1888.

	K.	L.	M.	N.	O.
SiO ₂ -----	58.11	55.87	56.85	57.38	56.39
TiO ₂ -----					none
Al ₂ O ₃ -----	.24	2.07	.88	.19	1.63
Fe ₂ O ₃ -----	5.44	5.79	4.33	4.43	1.72
FeO-----	.38	.38	1.45	1.25	3.70
MnO-----	trace	trace	trace	trace	.26
NiO-----					.13
CaO-----	12.01	12.43	13.09	12.14	7.92
MgO-----	21.97	21.62	21.56	22.71	24.63
Alkalies-----	undet.	undet.	undet.	undet.	none
P ₂ O ₅ -----					trace
H ₂ O at 100°-----	} 1.78	} 1.38	} 1.76	} 1.73	.65
H ₂ O above 100°-----					3.42
	99.93	99.54	99.92	99.83	100.45

P. Brown hornblende, Pierrepont, N. Y. Analysis by T. M. Chatard.

Q. Hornblende separated from gabbro, east shaft of waterworks extension, Washington, D. C. Analysis by R. B. Riggs.

R. Amphibole from the serpentine of Montville, N. J. Analysis by L. G. Eakins.

S. Hornblende separation, from south of soapstone quarry, Warwick, Mass. Analysis by E. A. Schneider.

T. Hornblende from amphibolite dike, Palmer Center, Mass. Specific gravity, 3.220 at 31.5°. Analysis by W. F. Hillebrand.

U. Hornblende from amphibolite bed, same locality and analyst. Specific gravity, 3.217 at 29°.

	P.	Q.	R.	S.	T.	U.
SiO ₂	56.44	52.43	43.31	47.86	43.11	44.09
TiO ₂11	-----	-----	.63	1.32	1.73
Al ₂ O ₃	1.77	3.15	17.41	14.09	11.10	10.68
V ₂ O ₃	-----	-----	-----	-----	.07	undet.
Cr ₂ O ₃	-----	-----	-----	-----	trace	trace
Fe ₂ O ₃84	3.60	.71	.33	4.97	2.72
FeO73	8.36	.59	13.41	13.04	12.96
MnO11	.11	.14	.14	.43	.32
NiO	-----	-----	-----	-----	trace	trace
CaO	11.83	14.33	12.84	.57	11.76	11.58
MgO	22.98	15.85	19.39	19.89	9.35	10.75
K ₂ O75	.12	1.36	.06	1.27	.88
Na ₂ O	2.13	1.16	2.23	.93	1.18	1.19
H ₂ O at 100°05	} 1.25	} 1.17	none	.16	.21
H ₂ O above 100°	2.41			2.46	1.92	1.91
P ₂ O ₅	trace	-----	-----	.05	.10	.10
	100.15	100.35	99.15	100.42	99.78	99.12

^aAdmixed rutile.

V. Hornblende near barkevikite. Separated from the sodalite-syenite of Square Butte, Highwood Mountains, Montana. Described by Lindgren in Am. Jour. Sci., 3d ser., vol. 45, p. 286. Analysis by W. H. Melville.

W. Amphibole separated from quartz-diorite, south of Table Mountain, on ridge between Butte and Plumas counties, Cal. Described by Turner in Seventeenth Annual Report, Part I, p. 521. Analysis by William Valentine. Chromium determination by Hillebrand.

X. Amphibole separated from amphibole gabbro, Beaver Creek, Big Trees quadrangle, California. Analysis by William Valentine. See Bulletin 168, p. 206.

Y. Amphibole separated from quartz-monzonite, Tioga road, south-east of Mount Hoffman, Mariposa County, Cal. Specific gravity, 3.203 at 21.5°. Analysis by W. F. Hillebrand. See Bulletin 168, p. 208.

Z. Amphibole separated from the granite of Butte, Mont. Incomplete analysis for lack of material. Analysis by H. N. Stokes.

	V.	W.	X.	Y.	Z.
SiO ₂	38.41	50.08	46.08	47.49	45.73
TiO ₂76	.77	1.21	1.43
Al ₂ O ₃	17.65	7.97	10.56	7.07	6.77
V ₂ O ₃04	
Cr ₂ O ₃16		none	none
Fe ₂ O ₃	3.75	2.69	2.81	4.88	4.94
FeO.....	21.75	6.71	8.30	10.69	10.39
MnO.....	.15	.49	.15	.51	.54
NiO.....	trace			.02	
CaO.....	10.52	11.21	12.64	11.92	11.25
MgO.....	2.54	16.31	14.40	13.06	12.32
Na ₂ O.....	2.95	1.22	1.62	.75	.77
K ₂ O.....	1.95	.46	.34	.49	1.22
Li ₂ O.....			none	trace	trace
H ₂ O at 100°.....			.17		.49
H ₂ O above 100°.....	.24	1.40	1.97	1.86	2.29
P ₂ O ₅		trace	.18	none	.35
F.....		undet.	none	.06	.28
	99.91	99.46	99.99	100.05	98.77
O equivalent to F.....				.02	.12
				100.03	98.65

BERYL.

A. White beryl from the tin mine at Winslow, Me. Opaque, milky, associated with mica, fluorite, calcite, tinstone, etc. Specific gravity, 2.707 at 27°. Analysis by W. F. Hillebrand.

B. Green beryl from near Home post-office in eastern Tennessee. Analysis by F. W. Clarke.

	A.	B.
SiO ₂	65.21	65.39
TiO ₂	trace
Al ₂ O ₃	18.50	19.10
Fe ₂ O ₃33	
FeO.....	}	
GlO.....		13.03
K ₂ O, Cs ₂ O.....	.14
Na ₂ O.....	.87
Li ₂ O.....	.16
MgO.....	.09
H ₂ O.....	1.80	1.76
	100.13	99.60

NEPHELITE.

A. Elæolite from Litchfield, Me. Dark-gray cleavable masses, of greasy luster. Described by F. W. Clarke in Bulletin 42.

B. Another sample from Litchfield. Analysis by George Steiger.

C. Nephelite extracted by solution from the elæolite-syenite of Red Hill, N. H. Described by Bayley in Bull. Geol. Soc. America, vol. 3, p. 231. Analysis by W. F. Hillebrand.

	A.	B.	C.
SiO ₂	43.74	45.91	45.31
Al ₂ O ₃	34.48	31.14	32.67
Fe ₂ O ₃34
FeO.....		.23
MgO.....			.16
CaO.....	trace	.33	2.00
Na ₂ O.....	16.62	14.60	12.60
K ₂ O.....	4.55	5.60	5.70
H ₂ O at 100°.....	.86	.47
H ₂ O above 100°.....		.93	1.56
CO ₂40
	100.25	99.95	100.00

CANCRINITE.

All from Litchfield, Me. See description by Clarke in Bulletin 42.

A. Pale yellow, granular.

B. Bright orange-yellow, cleavable, transparent in thin fragments.

C. Dingy pale yellow, otherwise like B.

D. Average yellow sample.

E. Flesh-colored, cleavable mixture of elæolite and cancrinite.

Analysis D by George Steiger; the others by F. W. Clarke. CO₂ determinations in A, B, C, E by R. B. Riggs.

	A.	B.	C.	D.	E.
SiO ₂	37.22	36.29	35.83	36.19	38.93
Al ₂ O ₃	28.32	30.12	29.45	29.24	32.52
Fe ₂ O ₃	trace	trace	trace	trace	-----
MnO	trace	trace	trace	-----	trace
CaO	4.40	4.27	5.12	4.72	2.47
MgO07	-----	-----	-----	none
Na ₂ O	19.43	19.56	19.33	19.20	17.02
K ₂ O18	.18	.09	.14	3.23
H ₂ O	3.86	2.98	3.79	4.15	2.83
CO ₂	6.22	6.96	6.50	6.11	2.95
	99.70	100.36	100.11	99.75	99.95

SODALITE.

A. Blue sodalite, from Litchfield, Me. Analysis by F. W. Clarke. See description in Bulletin 42.

B. Blue sodalite, from Ice River, near Kicking Horse Pass, British Columbia. Analysis by George Steiger.

C. Sodalite separated from the sodalite-syenite of Square Butte, Highwood Mountains, Montana. See Lindgren in Am. Jour. Sci., 3d ser., vol. 45, p. 286. Specific gravity, 2.265. Analysis by W. H. Melville.

	A.	B.	C.
SiO ₂	37.33	39.66	41.56
Al ₂ O ₃	31.87	30.09	29.48
Fe ₂ O ₃31	
FeO49
MgO15
CaO18	.49
Na ₂ O	24.56	22.60	19.21
K ₂ O10	1.14	.91
H ₂ O at 100°	1.07	.17	.45
H ₂ O above 100°79	3.73
Cl	6.83	6.12	4.79
	101.76	101.06	101.26
O equivalent to Cl	1.54	1.39	1.08
	100.22	99.67	100.18

ZUNYITE.

From the Zuñi mine, Anvil Mountain, near Silverton, Colo. Analyzed by W. F. Hillebrand, and described by him as a new species in Bulletin 20. Specific gravity, 2.875 at 15°.

SiO ₂	24.33
Fe ₂ O ₃20
Al ₂ O ₃	57.88
K ₂ O10
Na ₂ O24
Li ₂ O	trace
H ₂ O	10.89
P ₂ O ₅60
F	5.61
Cl	2.91
	102.76
O equivalent to F and Cl	3.02
	99.74

GARNET.

A. Grossularite. Large waterworn pebble, pale green, very compact. At first, thought to be jade. From Eltoro, 40 miles south of Los Angeles, Cal. Described by Clarke in *Am. Jour. Sci.*, 3d ser., vol. 50; p. 76. Analysis by George Steiger. Specific gravity, 3.485.

B, C, D. Three samples of garnet from Italian Mountain, Gunnison County, Colo. Specific gravities: A, 3.72 at 16°; B, 3.629 at 23°; C, 3.721 at 17.2°. Analyses by L. G. Eakins.

E. Pyrope from the peridotite dike of Elliott County, Ky. Described by Diller in *Bulletin* 38. Analysis by T. M. Chatard.

F. Garnets from Hawkes's quarry, Goshen, Mass. Almandite. Analysis by George Steiger.

	A.	B.	C.	D.	E.	F.
SiO ₂	37.54	37.89	39.26	36.88	41.32	37.30
TiO ₂	trace	-----	-----	-----	.16	.24
Al ₂ O ₃	22.84	7.90	19.63	10.34	21.21	21.84
Cr ₂ O ₃	-----	-----	-----	-----	.91	-----
Fe ₂ O ₃79	16.43	4.48	17.51	4.21	.98
FeO.....	.26	-----	-----	-----	7.93	32.62
MnO.....	-----	-----	trace	-----	.34	1.86
CaO.....	36.66	35.43	36.61	34.85	4.94	3.19
MgO.....	.44	.59	trace	.43	19.32	2.50
Na ₂ O.....	.13	1.10	.16	trace	.07	-----
K ₂ O.....	-----	trace	trace	-----	-----	-----
H ₂ O.....	1.74	.36	.08	.21	.17	-----
P ₂ O ₅	trace	-----	-----	-----	none	-----
	100.40	99.70	100.22	100.22	100.58	100.53

G. Spessartite from Amelia Court House, Amelia County, Va. Pale brown, crystalline masses. Analyzed by F. W. Clarke and described in Bulletin 60.

H. Spessartite from Llano County, Tex. Yellow, granular. Analyzed by W. H. Melville and described in Bulletin 90. FeO could not be separately determined.

I. Spessartite from cavities in rhyolite, Nathrop, Colo. Brilliant crystals. Specific gravity, 4.23 at 18°. Described by Cross in Am. Jour. Sci., 3d ser., vol. 31, p. 432. Analysis by L. G. Eakins.

J. Garnet from the Peacock mining claim, Seven Devils mining district, Idaho. Associated with bornite and powellite. Analyzed by W. H. Melville and described in Bulletin 90.

K, L. Two samples andradite from Clifton, Ariz. Analyses by George Steiger.

	G.	H.	I.	J.	K.	L.
SiO ₂	35.35	35.93	35.66	38.67	36.26	42.63
TiO ₂		trace			none	none
Al ₂ O ₃	20.41	18.08	18.55	10.08	.78	1.53
Fe ₂ O ₃	2.75	4.60	.32	16.00	32.43	31.41
FeO	1.75		14.25	.91	.32	.30
MnO	38.70	31.77	29.48		.27	.43
CaO94	8.48	1.15	33.35	29.67	23.37
BaO		trace				
MgO	none	.69		.77	none	none
Na ₂ O17	.21		none	
K ₂ O27		none	
CuO				trace		
P ₂ O ₅		none			.06	trace
H ₂ O at 100°27	.03	.44	.06	.13	
H ₂ O above 100°36			.44	
	100.17	100.11	100.33	99.84	100.36	99.67

CHRYSolITE.

A. Transparent green pebbles from near Fort Wingate, N. Mex. Analysis by E. A. Schneider.

B. Transparent olivine from the meteorite of Kiowa County, Kans. Described in Bulletin 78. Specific gravity, 3.376 at 23.2°. Analysis by L. G. Eakins.

C. Olivine from the peridotite dike of Elliott County, Ky. Described by Diller in Bulletin 38. Analysis by T. M. Chatard.

D. Olivine from the peridotite at Riddle, Oreg. The rock is the matrix of nickel silicates. Described by Diller and Clarke in Bulletin 60. Analysis by F. W. Clarke.

E. Olivine from olivine-gabbro, west side of Birch Lake, Minnesota. Described by Bayley, Jour. Geol., vol. 1, p. 688. Analysis by W. F. Hillebrand.

	A.	B.	C.	D.	E.
SiO ₂	41.98	40.70	40.05	42.81	35.58
TiO ₂07		1.22
Al ₂ O ₃		trace?	.39		.92
Cr ₂ O ₃24	.79	trace
Fe ₂ O ₃51	.18	2.36	2.61	} 33.91
FeO.....	5.71	10.79	7.14	7.20	
MnO.....	.10	.14	.20		.35
NiO.....	.42	.02		.26	} .20
CoO.....			trace		
MgO.....	51.11	48.02	46.68	45.12	26.86
CaO.....			1.16	none	.90
Na ₂ O.....			.08		
K ₂ O.....			.21		
H ₂ O at 100°.....	.05		.14	} .57	.11
H ₂ O above 100°.....	.23		.66		.20
P ₂ O ₅04		
	100.11	99.85	99.42	99.36	100.25

MIZZONITE.

From a gulch on the side of Italian Mountain, Gunnison County, Colo. Analysis by L. A. Eakins.

SiO ₂	57.55
Al ₂ O ₃	21.53
Fe ₂ O ₃	trace
CaO.....	6.18
K ₂ O.....	1.64
Na ₂ O.....	7.43
H ₂ O.....	3.23
Cl.....	2.82
	<hr/> 100.38
O equivalent to Cl.....	.63
	<hr/> 99.75

VESUVIANITE.

A. Finely crystallized material from Italian Mountain, Gunnison County, Colo. Analysis by L. G. Eakins. Specific gravity, 3.394 at 20°.

B. From Nevada, 24 kilometers northeast of Silver Peak. Analysis by George Steiger.

C. Pale green, massive. South Fork of Indian Creek, 12 miles from Happy Camp, Siskiyou County, Cal. Analysis by George Steiger.

	A.	B.	C.
SiO ₂	37.11	36.80	35.86
TiO ₂66	.10
Al ₂ O ₃	19.30	17.53	18.35
Fe ₂ O ₃	3.31	1.56	1.67
FeO.....		3.27	.39
MnO.....		.48	.05
MgO.....	3.89	1.23	5.43
CaO.....	36.24	35.00	33.51
Alkalies.....		.13	none
H ₂ O at 100°.....	.06	.10	.29
H ₂ O above 100°.....		1.56	4.18
P ₂ O ₅07	.02
CO ₂65	none
F.....	.58	.88	none
	<hr/> 100.49	<hr/> 99.92	<hr/> 99.85
O equivalent to F.....	.24	.36	-----
	<hr/> 100.25	<hr/> 99.56	-----

ZOISITE.

A. Zoisite from gabbro, east shaft of waterworks extension, Washington, D. C. Analysis by F. W. Clarke; iron determination by R. B. Riggs.

B. Rose-red zoisite, James's mica mine, Yancey County, N. C. Specific gravity, 3.352 at 27°. Analysis by L. G. Eakins.

C. Saussurite from gabbro, Sacramento River road, 37 miles north of Pit River ferry, Shasta County, Cal. Specific gravity, 3.148. Analysis by F. W. Clarke. See Bulletin 9.

D. Saussurite from the Saas Valley, Switzerland. Pale greenish. Specific gravity, 3.37. Analysis by F. W. Clarke; iron determination by R. B. Riggs.

	A.	B.	C.	D.
SiO ₂	45.12	38.98	42.79	48.29
Al ₂ O ₃	30.53	31.02	29.43	27.65
Fe ₂ O ₃		4.15		
FeO.....	1.90		3.65	1.45
MnO.....		.23		
MgO.....	.42		1.40	5.36
CaO.....	17.34	23.80	18.13	12.95
Na ₂ O.....	2.02		2.51	3.57
K ₂ O.....	1.09			trace
H ₂ O.....	.74	2.03	2.42	.54
	99.16	100.21	100.33	99.81

EPIDOTE.

A. Dark gray, brilliant crystals from Phippsburg, Me. Analyzed by W. F. Hillebrand and described in Bulletin 167.

B, C. Epidote from Italian Mountain, Gunnison County, Colo. Specific gravities: B, 3.448 at 25°; C, 3.452 at 17°. Analyses by L. G. Eakins.

	A.	B.	C.
SiO ₂	38.54	38.21	37.22
Al ₂ O ₃	28.39	28.70	24.09
Fe ₂ O ₃	6.89	8.16	12.80
FeO.....	.50		.79
MnO.....			.11
MgO.....	trace		trace
CaO.....	24.12	24.30	23.36
Na ₂ O.....		.21	.06
H ₂ O.....	2.26	.10	1.61
F.....	none	.35	.06
	100.70	100.03	100.10
O equivalent to F.....		.15	.02
		99.88	100.08

PIEDMONTITE.

From the rhyolite of Pine Mountain, near Monterey Station, Maryland. Contains a little admixed quartz. Analysis by W. F. Hillebrand.

SiO ₂	47.37
Al ₂ O ₃	18.55
Ce ₂ O ₃75
Other rare earths, mol. wt. at 295.....	1.28
Fe ₂ O ₃	4.02
Mn ₂ O ₃	6.85
MnO.....	1.92
PbO.....	.14
CuO.....	.11
CaO.....	15.82
MgO.....	.25
K ₂ O.....	.68
Na ₂ O.....	.23
Li ₂ O.....	trace
H ₂ O at 100°.....	.14
H ₂ O above 100°.....	1.94
P ₂ O ₅	trace
	100.05

ALLANITE.

From Platte Mountain, Douglas County, Colo. Specific gravity, 3.52 at 29°. Analysis by L. G. Eakins. See Proc. Colorado Sci. Soc., vol. 2, p. 32.

SiO ₂	31.13
Al ₂ O ₃	11.44
Fe ₂ O ₃	6.24
Ce ₂ O ₃	12.50
(LaDi) ₂ O ₃	10.98
FeO	13.59
BaO27?
MnO61
CaO	9.44
MgO16
K ₂ O	trace
Na ₂ O56
H ₂ O	2.78
CO ₂21
P ₂ O ₅	trace
	<hr/> 99.91

PREHNITE.

A. From Paterson, N. J. Analysis by George Steiger.

B. From Fassa, Tyrol. Analysis by E. A. Schneider.

	A.	B.
SiO ₂	42.31	43.32
Al ₂ O ₃	19.95	25.50
Fe ₂ O ₃	6.20	trace
FeO	none	-----
CaO	26.63	26.49
MgO	none	-----
H ₂ O at 100°21	.17
H ₂ O at 250°-300°	} 4.81	.14
H ₂ O by ignition		4.70
	<hr/> 100.11	<hr/> 100.32

TOPAZ.

A. White, opaque topaz from Stoneham, Me. Specific gravity, 3.51.

B. Alteration product of A, greenish. Specific gravity, 3.42. Analyses A and B by J. E. Whitfield. See discussion by Clarke in Bulletin 27.

C. Topaz from Florissant, Colo. Specific gravity, 3.578 at 22°. Analyzed by W. F. Hillebrand, and described in Bulletin 20.

	A.	B.	C.
SiO ₂	31.92	35.15	33.15
Al ₂ O ₃	57.38	53.18	57.01
CaO		1.32	
MgO17	
K ₂ O15	1.52	
Na ₂ O	1.33	1.28	
H ₂ O20	.90	
F	16.99	12.88	16.04
	107.97	106.40	103.20
O equivalent to F	7.16	5.42	6.75
	100.81	100.98	99.45

The final alteration of the Stoneham topaz is into muscovite, q. v. The alteration product here given represents the beginning of the process.

SILLIMANITE.

Fibrolite adze from Brittany. Specific gravity, 3.147, determined by William Hallock. Analysis by F. W. Clarke. See Bulletin 60.

SiO ₂	34.66
Al ₂ O ₃	63.24
Fe ₂ O ₃	trace
MgO37
Ignition	1.31
	99.58

KYANITE.

Pale green variety, associated with the dumortierite of Clip, Ariz. Analyzed by W. F. Hillebrand. Titanic oxide is present in appreciable amounts, but was not separated from alumina. The mineral contained a few black, nonmagnetic grains, which may have carried the titanium. Specific gravity, 3.656 at 18.5°.

SiO ₂	36.30
Al ₂ O ₃ (TiO ₂).....	62.51
Fe ₂ O ₃70
FeO.....	undet.
CuO.....	trace
Ignition.....	.40
	<hr/>
	99.91

ILVAITE.

From Golconda mine, South Mountain, Owyhee County, Idaho.
Analysis by W. F. Hillebrand. Specific gravity, 4.059 at 31°.

SiO ₂	29.16
Al ₂ O ₃52
Fe ₂ O ₃	20.40
FeO.....	29.14
MnO.....	5.15
CaO.....	13.02
MgO.....	.15
Na ₂ O.....	.08
H ₂ O at 105°.....	.15
H ₂ O above 105°.....	2.64
	<hr/>
	100.41

CALAMINE.

White, highly crystalline. From Sterling, N. J. Analysis by
George Steiger.

SiO ₂	24.15
Al ₂ O ₃	} .19
Fe ₂ O ₃	
ZnO.....	67.55
CaO.....	.12
H ₂ O at 100°.....	.27
H ₂ O above 100°.....	7.68
	<hr/>
	99.96

LAWSONITE.

From Tiburon Peninsula, California. Collected by F. L. Ransome.
Analysis by W. F. Hillebrand.

SiO ₂	38.45
TiO ₂38
Al ₂ O ₃	31.35
Fe ₂ O ₃86
FeO.....	.10
MnO.....	trace
CaO.....	17.52
MgO.....	.17
K ₂ O.....	.23
Na ₂ O.....	.06
Ignition.....	11.21
	<hr/>
	100.33

STAUROLITE.

Altered staurolite from Liberty Grove, Cecil County, Md. About two-thirds muscovite. Analysis by George Steiger.

SiO ₂	50.17
TiO ₂55
Al ₂ O ₃	27.97
Fe ₂ O ₃	6.13
FeO.....	1.18
MgO.....	1.15
K ₂ O.....	7.77
Na ₂ O.....	.48
H ₂ O at 100°.....	.42
H ₂ O above 100°.....	3.94
P ₂ O ₅06
	<hr/> 99.82

GADOLINITE.

Analyses by L. G. Eakins.

A. From Llano County, Tex. Specific gravity, 4.239 at 17.4°.

B, C. From Devils Head Mountain, Douglas County, Colo. Specific gravities: B, 4.56 at 17°; C, 4.59 at 25.5°. Described in Proc. Colorado Sci. Soc., vol. 2, pt. 1, p. 32.

	A.	B.	C.
SiO ₂	23.79	22.13	21.86
ThO ₂58	.89	.81
Al ₂ O ₃		2.34	.54
Fe ₂ O ₃96	1.13	3.59
Ce ₂ O ₃	2.62	11.10	6.87
(La Di) ₂ O ₃	5.22	21.23	19.10
Y ₂ O ₃	^a 41.55	^b 9.50	^b 12.63
Er ₂ O ₃		^b 12.74	^b 15.80
FeO.....	12.42	10.43	11.36
MnO.....	trace		.11
GdO.....	11.33	7.19	5.46
CaO.....	.74	.34	.47
MgO.....	trace	.14	.16
K ₂ O.....	trace	.18	.20
Na ₂ O.....	trace	.28	.32
H ₂ O.....	1.03	.86	.74
P ₂ O ₅05		
	<hr/> 100.29	<hr/> 100.48	<hr/> 100.02

^a Molecular weight, 260.

^b Molecular weight Y, Er group: in B, 296; in C, 294.

YTTRIALITE.

From the Barringer mine, Llano County, Tex. Analyzed by W. F. Hillebrand, and discussed in Amer. Jour. Sci., 4th ser., vol. 13, p. 145.

SiO ₂	29.63
TiO ₂	.05
ThO ₂	10.85
UO ₂	1.64
Ce ₂ O ₃	3.07
La ₂ O ₃ group, mol. wt., 335.6	5.18
Y ₂ O ₃ group, mol. wt., 265.8	43.45
Fe ₂ O ₃	.76
FeO	1.96
MnO	.88
PbO	.80
CaO	.67
MgO	.16
H ₂ O at 105°	.32
H ₂ O above 105°	.04
CO ₂	.11
P ₂ O ₅	.12
A, He, F, and alkalis by difference	.31

100.00

ROWLANDITE.

From Llano County, Tex. Analysis by W. F. Hillebrand, with discussion in Bulletin 113. Specific gravity, 4.513 at 15.5°.

SiO ₂	26.04
X ^a	.39
ThO ₂	.59
Ce ₂ O ₃	5.06
La ₂ O ₃ group, mol. wt., 336.8	9.34
Yt ₂ O ₃ group, mol. wt., 266.2	47.70
Fe ₂ O ₃	.09
FeO	4.39
MnO	.67
CaO	.50
MgO	1.62
Alkalies	.28
H ₂ O	.24
CO ₂	.34
F	3.87
P ₂ O ₅	trace

101.12

O equivalent to F 1.63

99.49

MACKINTOSHITE.

From Llano County, Tex. Analyzed by W. F. Hillebrand and discussed in Bulletin 113. Specific gravity, 5.43 at 21.4°. Only nine-tenths of a gram available for analysis.

^aA mixture of indefinable earths with some uranium and a trace of titanium.

SiO ₂	13.90
UO ₂	22.40
ZrO ₂ ?88
ThO ₂	} 45.30
Ce ₂ O ₃	
La ₂ O ₃ group	} 1.86
Yt ₂ O ₃ group	
PbO	3.74
FeO	1.15
CaO59
MgO10
K ₂ O42
(NaLi) ₂ O68
P ₂ O ₅67
H ₂ O at 100°50
H ₂ O above 100°	4.31
	96.50

CYRTOLITE.

A doubtful mineral from Devils Head Mountain, Douglas County, Colo. Analyses by W. F. Hillebrand, with description in Proc. Colorado Sci. Soc., vol. 3, pt. 1, p. 44. Brown crystalline growths of irregular form. Specific gravities: A, 3.70; B, 3.60; C, 3.64. The material may be a mixture of cyrtolite or some analogous alteration of zircon with limonite and a phosphate.

	A.	B.	C.
SiO ₂	20.06	20.64	19.21
Ta ₂ O ₅	} 47.99	.71	} 51.00
SnO ₂03	
ZrO ₂		47.81	
ThO ₂	1.16	} 1.20	} .60
Ce ₂ O ₃06		
(LaDi) ₂ O ₃19		
Er ₂ O ₃	4.77	4.76	4.55
Y ₂ O ₃	2.27	2.48	3.13
Fe ₂ O ₃	5.53	5.97	4.86
MnO47	.57	.33
CaO	1.99	1.93	2.15
MgO13	.11
K ₂ O20	.10	.17
Na ₂ O46	.50	.42
H ₂ O	12.87	12.00	12.97
P ₂ O ₅	1.64	1.75	.93
F25	.42	.42
	100.04	100.98	100.74

DANBURITE.

From Russell, N. Y. Analysis by J. E. Whitfield. See Bulletin 55.

SiO ₂	49.70
B ₂ O ₃	25.80
CaO	23.26
Fe ₂ O ₃ , Al ₂ O ₃	1.02
Ignition20
	<hr/> 99.98

DATOLITE.

From Bergen Hill, N. J. Analysis by J. E. Whitfield. See Bulletin 55.

SiO ₂	35.74
B ₂ O ₃	22.60
CaO	35.14
FeO31
H ₂ O	6.14
	<hr/> 99.93

AXINITE.

Analyses by J. E. Whitfield, with discussion in Bulletin 55.

A. Clove brown, from Cornwall, England. Translucent.

B. Pearl gray, from Bourg d'Oisans, Dauphiny. Transparent.

	A.	B.
SiO ₂	42.10	41.53
Al ₂ O ₃	17.40	17.90
Fe ₂ O ₃	3.06	3.90
FeO	5.84	4.02
CaO	20.53	21.66
MnO	4.63	3.79
MgO66	.74
B ₂ O ₃	4.64	4.62
H ₂ O	1.80	2.16
	<hr/> 100.66	<hr/> 100.32

DUMORTIERITE.

A. Harlem, N. Y. Analysis by R. B. Riggs. See Am. Jour. Sci., 3d ser., vol. 34, p. 406.

B. Harlem, N. Y. Analysis by J. E. Whitfield.

C. Near Clip, Yuma County, Ariz. Analysis by J. E. Whitfield.

D. Purified sample from Clip. Analysis by Whitfield. Specific gravity a little over 3.265.

See note by Whitfield in Bulletin 60, and paper by Diller and Whitfield in Bulletin 64.

	A. ^a	B.	C.	D.
SiO ₂	34.82	31.44	31.52	27.99
Al ₂ O ₃	55.30	68.91	63.66	64.49
Fe ₂ O ₃	trace			
CaO	trace		trace	
MgO57		.52	trace
Na ₂ O	1.76		.37	
K ₂ O	1.04		.11	
Li ₂ O	trace			
B ₂ O ₃	4.07	trace	2.62	4.95
P ₂ O ₅20
Ignition	2.96		1.34	1.72
	100.32	100.35	100.14	99.35

^a Sample A contaminated by tourmaline.

TOURMALINE.

Analyses A to V, inclusive, by R. B. Riggs. Discussed by Riggs in Bulletin 55.

A. Pink center of crystals having a green margin, from Calhao, province of Minas-Geraes, Brazil. Specific gravity, 3.028.

B. Pale green, border of A.

C. Olive-green, also from Calhao, Brazil.

D. Rose-colored, massive, from Black Mountain, Rumford, Me. Specific gravity, 2.997.

E. Dark green, massive, same locality as D.

	A.	B.	C.	D.	E.
SiO ₂	37.19	37.39	36.91	38.07	36.53
Al ₂ O ₃	42.43	39.65	38.13	42.24	38.10
Fe ₂ O ₃	none	.15	.31	-----	none
FeO.....	.52	2.29	3.19	.26	6.43
MnO.....	.79	1.47	2.22	.35	.32
MgO.....	none	none	.04	.07	none
CaO.....	.57	.49	.38	.56	.34
Li ₂ O.....	1.73	1.71	1.61	1.59	.95
Na ₂ O.....	2.24	2.42	2.70	2.18	2.86
K ₂ O.....	.23	.25	.28	.44	.38
P ₂ O ₅	none	trace	.11	none	trace
B ₂ O ₃	10.06	10.29	9.87	9.99	10.22
H ₂ O.....	3.90	3.63	3.64	4.26	3.52
F.....	trace?	.32	.14	.28	.16
	99.66	100.06	99.53	100.29	99.81
O equivalent to F.....	-----	.13	.06	.12	.07
	-----	99.93	99.47	100.17	99.74

F. Alteration product of the Rumford tourmaline.

G. Alteration product of rubellite, from Hebron, Me.

H. Colorless to very pale pinkish or greenish crystals, Auburn, Me.
Specific gravity, 3.07.

I. Light-green crystals, Auburn, Me.

J. Dark green, massive tourmaline, Auburn, Me.

	F.	G.	H.	I.	J.
SiO ₂	53.03	43.90	38.14	37.85	36.26
Al ₂ O ₃	31.67	38.71	39.60	37.73	36.68
Fe ₂ O ₃51	.58	.30	.42	.15
FeO.....		.25	1.38	3.88	7.07
MnO.....	trace	.04	1.38	.51	.72
MgO.....	trace	.05	trace	.04	.16
CaO.....	trace	.41	.43	.49	.17
Li ₂ O.....	.26		1.34	1.34	1.05
Na ₂ O.....	.54	1.05	2.36	2.16	2.88
K ₂ O.....	9.44	10.92	.27	.62	.44
B ₂ O ₃	trace	trace	10.25	10.55	9.94
P ₂ O ₅			trace	trace	trace
H ₂ O.....	4.80	4.25	4.16	4.18	4.05
F.....	trace?	none	.62	.62	.71
	100.25	100.16	100.23	100.39	100.28
O equivalent to F.....			.26	.26	.30
			99.97	100.13	99.98

K. White to light-brown crystals, Dekalb, St. Lawrence County, N. Y. Specific gravity, 3.085.

L. Brown tourmaline, Gouverneur, St. Lawrence County, N. Y.

M. Cinnamon-brown crystals, Hamburg, N. J.

N. Brilliant black crystals, Pierrepont, St. Lawrence County, N. Y. Specific gravity, 3.08.

O. Dark-brown crystals, Orford, N. H.

P. Dark-brown crystals, Monroe, Conn.

	K.	L.	M.	N.	O.	P.
SiO ₂ -----	36.88	37.39	35.25	35.61	36.66	36.41
TiO ₂ -----	.12	1.19	.65	.55	.23	1.61
Al ₂ O ₃ -----	28.87	27.79	28.49	25.29	32.84	31.27
Fe ₂ O ₃ -----		.10	none	.44	none	none
FeO-----	.52	.64	.86	8.19	2.50	3.80
MnO-----	none	none	none	trace	trace	trace
MgO-----	14.53	14.09	14.58	11.07	10.35	9.47
CaO-----	3.70	2.78	5.09	3.31	1.35	.98
SrO-----	trace	trace	trace	none	trace	trace
Li ₂ O-----	trace	trace	trace	trace	trace	none
Na ₂ O-----	1.39	1.72	.94	1.51	2.42	2.68
K ₂ O-----	.18	.16	.18	.20	.22	.21
B ₂ O ₃ -----	10.58	10.73	10.45	10.15	10.07	9.65
P ₂ O ₅ -----	undet.	none	trace	trace	none	trace
H ₂ O-----	3.56	3.83	3.10	3.34	3.78	3.79
F-----	.50	trace?	.78	.27	trace?	none
	100.83	100.42	100.37	99.93	100.42	99.87
O equivalent to F-----	.21	-----	.33	.11	-----	-----
	100.62	-----	100.04	99.82	-----	-----

Q. Massive, black tourmaline, Auburn, Me. Specific gravity, 3.19.

R. Massive, black tourmaline, Paris, Me.

S. Black tourmaline, Calhao, Brazil. Specific gravity, 3.20.

T. Black tourmaline, Haddam, Conn.

	Q.	R.	S.	T.
SiO ₂	34.99	35.03	34.63	34.95
TiO ₂	none			.57
Al ₂ O ₃	33.96	34.44	32.70	31.11
Fe ₂ O ₃	none	1.13	.31	.50
FeO	14.23	12.10	13.69	11.87
MnO06	.08	.12	.09
MgO	1.01	1.81	2.13	4.45
CaO15	.24	.33	.81
Li ₂ O	trace	.07	.08	trace
Na ₂ O	2.01	2.03	2.11	2.22
K ₂ O34	.25	.24	.24
B ₂ O ₃	9.63	9.02	9.63	9.92
P ₂ O ₅	trace	trace	none	trace
H ₂ O	3.62	3.69	3.49	3.62
F	none	none	.06	none
	100.00	99.89	99.52	100.35
O equivalent to F02	
			99.50	

U. Black crystal, Nantic Gulf, Baffin Land. Specific gravity, 3.095.

V. Black crystal, Stony Point, Alexander County, N. C. Specific gravity, 3.13.

W. Brown tourmaline from near Colfax, Nevada County, Cal. Fine, radiating needles. Specific gravity, 3.065. Incomplete analysis by W. H. Melville. See Bulletin 90.

X. Chrome tourmaline from near Etchison, Montgomery County, Md. Analysis by T. M. Chatard. See Bulletin 64.

	U.	V.	W.	X.
SiO ₂ -----	35.34	35.56	36.40	36.57
TiO ₂ -----	.40	.55	-----	.09
Al ₂ O ₃ -----	30.49	33.38	33.64	32.58
Cr ₂ O ₃ -----	-----	-----	-----	4.32
Fe ₂ O ₃ -----	none	none	} 3.13 }	.79
FeO-----	8.22	8.49		
MnO-----	trace	.04	-----	trace
NiO-----	-----	-----	-----	.05
MgO-----	7.76	5.44	10.01	9.47
CaO-----	2.32	.53	1.51	.75
SrO-----	trace	none	-----	-----
Li ₂ O-----	trace	trace	-----	trace
Na ₂ O-----	1.76	2.16	2.49	2.22
K ₂ O-----	.15	.24	.12	.13
B ₂ O ₃ -----	10.45	10.40	^a 8.74	8.90
P ₂ O ₅ -----	none	(?)	-----	.04
H ₂ O-----	3.60	3.63	3.53	3.74
F-----	none	none	.74	.06
	100.49	100.42	100.31	99.71
O equivalent to F-----	-----	-----	.31	.02
	-----	-----	100.00	99.69

^aBy difference.

GYROLITE.

Associated with the apophyllite of New Almaden, Cal. Fibrous.
Analyzed by F. W. Clarke and described in Bulletin 64.

SiO ₂	52.54
Al ₂ O ₃71
Fe ₂ O ₃	
CaO	29.97
K ₂ O	1.56
Na ₂ O27
F65
Ignition	14.60
	100.30
O equivalent to F27
	100.03

APOPHYLLITE.

From Table Mountain, Golden, Colo. Described by Cross and
Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

A. Apophyllite.

B. Decomposition product of apophyllite.

	A.	B.
SiO ₂	51.89	67.96
Al ₂ O ₃	1.54	8.48
Fe ₂ O ₃13	1.04
CaO	24.51	5.47
MgO53
K ₂ O	3.81	1.23
Na ₂ O59	a .74
H ₂ O	16.52	14.55
F	1.70	none
O equivalent to F	100.69	100.00
	.72	
	99.97	

a By difference.

PTILOLITE.

Described as a new species by Cross and Eakins in Am. Jour. Sci., 3d ser., vol. 32, p. 117, and vol. 44, p. 96. Analyses by L. G. Eakins.

A. From Green Mountain, Jefferson County, Colo.

B. From 3 miles south of Silver Cliff, Custer County, Colo.

	A.	B.
SiO ₂	70.35	67.83
Al ₂ O ₃	11.90	11.44
CaO.....	3.87	3.30
K ₂ O.....	2.83	.64
Na ₂ O.....	.77	2.63
H ₂ O at 110°.....	10.18	3.62
H ₂ O at 125°.....		1.31
H ₂ O at 300°.....		5.41
H ₂ O at redness.....		3.10
	99.90	99.28

HEULANDITE.

A. From Green Mountain, Jefferson County, Colo. Analysis by L. G. Eakins.

B. From Anthracite Creek, Gunnison County, Colo. Analysis by L. G. Eakins. Specific gravity, 2.24 at 20.1°.

C. From Berufjörd, Iceland. Analysis by George Steiger.

	A.	B.	C.
SiO ₂	59.17	57.38	57.10
Al ₂ O ₃	16.80	17.18	16.82
Fe ₂ O ₃		trace	
MgO.....			.07
CaO.....	7.10	8.07	6.95
SrO.....			.46
Na ₂ O.....	1.37	.82	1.25
K ₂ O.....	.34	.40	.42
H ₂ O at 110°.....	15.45	2.57	3.61
H ₂ O at 125°.....		1.10	13.00
H ₂ O at 300°.....		11.70	
H ₂ O at redness.....		.90	
	100.23	100.12	99.68

STILBITE.

A. From Wassons Bluff, near Parrsboro, Nova Scotia. Analysis by George Steiger.

B. From Italian Mountain, Gunnison County, Colo. Analysis by L. G. Eakins.

C. From Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analysis by W. F. Hillebrand.

	A.	B.	C.
SiO ₂	55.41	57.75	54.67
Al ₂ O ₃	16.85	16.64	16.78
Fe ₂ O ₃18	-----	-----
MgO.....	.05	-----	-----
CaO.....	7.78	8.58	7.98
Na ₂ O.....	1.23	trace	1.47
H ₂ O at 100°.....	3.60	} 17.17	} 19.16
H ₂ O above 100°.....	15.41		
	100.51	100.14	100.06

LAUMONTITE.

From Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

A. Yellow grains.

B. White crystals.

	A.	B.
SiO ₂	51.43	52.07
Al ₂ O ₃	21.52	21.30
Fe ₂ O ₃94	-----
CaO.....	11.88	11.24
K ₂ O.....	.35	.42
Na ₂ O.....	.19	.48
H ₂ O.....	13.81	14.58
	100.12	100.09

CHABAZITE.

A. From Wassons Bluff, near Parrsboro, Nova Scotia. Analysis by George Steiger.

B, C. From Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

	A.	D.	C.
SiO ₂	50.78	47.86	47.18
Al ₂ O ₃	17.18	19.30	19.67
Fe ₂ O ₃40	.12	-----
MgO.....	.04	-----	-----
CaO.....	7.84	} 9.94	9.74
SrO.....	-----		.43
Na ₂ O.....	1.28	.52	.51
K ₂ O.....	.73	.35	.37
H ₂ O at 100°.....	5.22	} 22.07	4.76
H ₂ O above 100°.....	16.63		17.39
	100.10	100.16	100.05

LEVYNITE.

From Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

A. Crystals of levynite.

B. Associated fibrous variety.

	A.	B.
SiO ₂	46.76	46.97
Al ₂ O ₃	21.91	22.39
CaO.....	11.12	10.85
K ₂ O.....	.21	1.17
Na ₂ O.....	1.34	.79
H ₂ O.....	18.65	18.03
	99.99	100.20

ANALCITE.

A. From Wassons Bluff, near Parrsboro, Nova Scotia. Analysis by George Steiger. See Bulletin 207.

B. From Table Mountain, Golden, Colo. Analysis by George Steiger. See Bulletin 207.

C, D. From Table Mountain. Described by Cross and Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

	A.	B.	C.	D.
SiO ₂	57.06	55.72	55.82	55.80
Al ₂ O ₃	21.48	23.06	22.42	22.45
Fe ₂ O ₃13		-----	-----
CaO.....	.16	.17	-----	-----
Na ₂ O.....	12.20	12.46	13.48	13.45
H ₂ O at 100°.....	.58	.13	8.38	8.35
H ₂ O above 100°.....	8.38	8.26		
	99.99	99.80	100.10	100.05

NATROLITE.

A. From Bergen Hill, New Jersey. Analysis by George Steiger. See Bulletin 207.

B. From Magnet Cove, Arkansas. Analysis by W. H. Melville. Described in Bulletin 90. Specific gravity, 2.261.

C. From South Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analysis by W. F. Hillebrand.

	A.	B.	C.
SiO ₂	46.62	47.56	43.66
Al ₂ O ₃	26.04	26.82	24.89
Fe ₂ O ₃	none	.23	-----
CaO.....	1.48	.13	4.87
MgO.....	none	.09	-----
Na ₂ O.....	15.67	15.40	14.66
H ₂ O at 100°.....	.39	.07	^a 8.09
H ₂ O above 100°.....	10.18	9.56	-----
CO ₂	-----	-----	^b 3.83
	100.38	99.86	100.00

^a By difference. Too little material for complete analysis.

^b Calculated to satisfy total line.

SCOLECITE.

A. From Whale Cove, Grand Manan, New Brunswick. Analysis by George Steiger. See Bulletin 207.

B. From Italian Mountain, Gunnison County, Colo. Analysis by L. G. Eakins. Specific gravity, 2.247 at 17.2°.

C. From Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analysis by W. F. Hillebrand.

	A.	B.	C.
SiO ₂	45.86	45.90	46.08
Al ₂ O ₃	25.78	26.51	25.28
Fe ₂ O ₃27
CaO.....	13.92	14.17	12.77
MgO.....		trace	
Na ₂ O.....	.41	trace	1.04
K ₂ O.....			.13
H ₂ O at 100°.....	.40		} ^a 14.48
H ₂ O above 100°.....	13.65	13.79	
	100.02	100.37	100.00

^a By difference.

MESOLITE.

From North Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analysis by W. F. Hillebrand.

SiO ₂	46.17
Al ₂ O ₃	26.88
CaO.....	8.77
Na ₂ O.....	6.19
H ₂ O.....	12.16
	<hr/> 100.17

THOMSONITE.

From Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analyses A, B, C, D, E by W. F. Hillebrand. Analysis F by George Steiger; published in Bulletin 207. A represents reddish spherules; F, a mass of fibrous structure.

	A.	B.	C.	D.	E.	F.
SiO ₂	40.52	40.88	40.68	41.21	42.66	41.13
Al ₂ O ₃	29.22	29.68	30.12	29.71	29.25	29.58
Fe ₂ O ₃79					
CaO	12.43	11.88	11.92	11.34	10.90	11.25
Na ₂ O	4.31	4.72	4.44	5.62	4.92	5.31
H ₂ O at 100°	} 12.79	} 12.91	} 12.86	} 12.20	} 12.28	1.01
H ₂ O above 100°						12.12
	100.06	100.07	100.02	100.08	100.01	100.40

HYDRONEPHELITE.

From the elæolite-syenite of Litchfield, Me. Described by Clarke as a new species in Bulletin 42. Analyses by F. W. Clarke. Specific gravity, 2.263, determined by J. S. Diller.

A, B. Slightly impure. Two distinct samples.

C. Carefully purified material dried at 100°.

	A.	B.	C.
SiO ₂	38.90	39.24	38.99
Al ₂ O ₃	33.98	33.16	33.62
Fe ₂ O ₃	trace	trace	
CaO05	trace	.07
Na ₂ O	13.21	13.07	13.07
K ₂ O	1.01	.88	1.12
H ₂ O	13.12	13.30	12.98
	100.27	99.65	99.85

MUSCOVITE.

A. From the Hatch farm, Auburn, Me. Broad plates, bordered by lepidolite. Analysis by R. B. Riggs.

B. Greenish muscovite, Auburn, Me. Analysis by E. A. Schneider.

C. Altered muscovite, Mount Mica, Paris, Me. Occurs as a white enamel on ordinary mica. Analysis by F. W. Clarke.

D. Well-crystallized muscovite from the hiddenite mine, Stony Point, Alexander County, N. C. Analysis by F. W. Clarke.

E. White muscovite from Miask, Ural Mountains, Siberia. Analysis by E. A. Schneider.

F. Mica separated from quartz-schist. Shoemaker's quarry, near Stevenson Station, Green Spring Valley, Maryland. Analysis by E. A. Schneider.

G. Mica from the Eureka tunnel, St. Peters Dome, near Pikes Peak, Colorado. Analysis by W. F. Hillebränd. See Bulletin 20.

	A.	B.	C.	D.	E.	F.	G.
SiO ₂	44.39	46.54	46.61	45.40	44.17	44.93	52.59
TiO ₂		none		1.10		1.05	
Al ₂ O ₃	35.70	34.96	35.61	33.66	37.35	29.81	29.72
Fe ₂ O ₃	1.09	1.59		2.36	1.29	6.10	1.40
FeO.....	1.07				.20		
MnO.....	trace		trace		.10	trace	
MgO.....		.32		1.86		1.16	2.12
CaO.....	.10		trace				.26
Li ₂ O.....				trace		trace	
Na ₂ O.....	2.41	.41	1.76	1.41	1.14	.50	.50
K ₂ O.....	9.77	10.38	8.86	8.33	10.00	10.28	8.33
H ₂ O at 100°.....	5.88	.71	6.50	5.46	1.06	1.38	4.39
H ₂ O above 100°.....		4.72			4.67	4.88	
F.....	.72	none		.69	.90	.22	
	101.13	99.63	99.34	100.27	100.88	100.31	99.31
O equivalent to F.....	.30			.29	.37	.09	
	100.83			99.98	100.51	100.22	

H. Fuchsite. Etchison post-office, Montgomery County, Md. Analysis by T. M. Chatard. See Bulletin 64.

I. Grayish-green, compact mica from Stoneham, Me. Structure subfibrous. Analysis by T. M. Chatard.

J. Same locality as I. Specimen broadly foliated, micaceous. Analysis by T. M. Chatard. For analyses I and J see Bulletin 9.

K. Stoneham, Me. From alteration of topaz, the outer portion of an altered crystal. Specific gravity, 2.82. Analysis by J. E. Whitfield. Discussed by Clarke in Bulletin 27.

L. Pseudomorph from the Rochelle mine, on Running Water River, Wyoming. Near liebenerite. Specific gravity, 2.831 at 12.5°. Analyzed by W. F. Hillebrand and described in Bulletin 20.

M. Mariposite. From vein of the Josephine gold quartz mine, Bear Valley, Mariposa County, Cal. Color, green. Specific gravity, 2.817 at 29.5°.

N. White mariposite, same locality as M. Specific gravity, 2.787 at 28.5°. Analyses M and N by W. F. Hillebrand, and described in Bulletin 167.

	H.	I.	J.	K.	L.	M.	N.
SiO ₂ -----	42.21	45.19	45.34	44.52	45.54	55.35	56.79
TiO ₂ -----						.18	
Al ₂ O ₃ -----	34.55	33.32	33.96	46.19	37.15	25.62	25.29
Cr ₂ O ₃ -----	2.03				.79	.18	none
Fe ₂ O ₃ -----	1.03					.63	1.59
FeO-----		4.25	3.96			.92	
MnO-----	trace	.58	.51	.21			
CaO-----	.47	trace	.22	.30		.07	.07
MgO-----	3.13	.36	.10	.14	.38	3.25	3.29
Li ₂ O-----						trace	trace
Na ₂ O-----	.82	1.57	1.49	2.82	.90	.12	.17
K ₂ O-----	9.16	11.06	10.73	2.30	10.70	9.29	8.92
H ₂ O at 110°-----	.20	4.48	4.78	3.74	4.80	4.52	4.72
H ₂ O above 110°-----	6.57						
F-----				.40			
	100.17	100.81	101.09	100.62	100.26	100.13	100.84
O equivalent to F-----				.16			
				100.46			

LEPIDOLITE.

From various localities in Maine. Analyses by R. B. Riggs. Discussion by F. W. Clarke in Bulletin 42.

A. From Black Mountain, Rumford. Lilac-purple, granular.

B. From Mount Mica, Paris. Broadly foliated.

C. From Hebron. Purple, coarsely granular.

D. From Auburn. Purple border on plates of muscovite.

E. From Auburn. Purple, coarsely granular.

F. From Norway. White, coarsely granular.

G. From Norway. Brownish, finely granular.

	A.	B.	C.	D.	E.	F.	G.
SiO ₂ -----	51.52	50.92	48.80	49.62	51.11	49.52	50.17
Al ₂ O ₃ -----	25.96	24.99	28.80	27.30	25.26	28.80	25.40
Fe ₂ O ₃ -----	.31	.30	.29	.31	.20	.40	.87
FeO-----	undet.	.23	.09	.07	.07	.24	.45
MnO-----	.20	trace	.08	.55	.17	.07	.23
CaO-----	.16	trace	.10	-----	.12	.13	undet.
MgO-----	.02	trace	.07	-----	.01	.02	undet.
Li ₂ O-----	4.90	4.20	4.49	4.34	4.98	3.87	4.03
Na ₂ O-----	1.06	2.11	.74	2.17	1.43	.13	13.40
K ₂ O-----	11.01	11.38	12.21	8.03	10.51	8.82	
Rb ₂ O-----	-----	trace		2.44	1.29	3.73	
Cs ₂ O-----	-----	trace		.72	.45	.08	
H ₂ O-----	.95	1.96	1.73	1.52	.94	1.72	2.02
F-----	5.80	6.29	4.96	5.45	6.57	5.18	5.05
O equivalent to F ..	101.89	102.38	101.86	102.52	103.11	102.71	101.62
	2.44	2.64	2.02	2.29	2.76	2.18	2.13
	99.45	99.74	99.84	100.23	100.35	100.53	99.49

CRYOPHYLLITE.

From Rockport, Mass. Analyses by R. B. Riggs. Discussed by Clarke in Bulletin 42.

- A. Brilliant, broadly foliated, blackish-green variety.
- B. Paler, dull green, less lustrous, probably somewhat altered.
- C. Granular, resembling chlorite.

	A.	B.	C.
SiO ₂	51.96	51.46	52.17
Al ₂ O ₃	16.89	16.22	16.39
Fe ₂ O ₃	2.63	2.21	4.11
FeO.....	6.35	7.66	6.08
MnO.....	.24	.06	.32
CaO.....	.12	trace	trace
MgO.....	.03	.17	trace
Li ₂ O.....	4.93	4.83	5.03
Na ₂ O.....	.92	.95	.60
K ₂ O.....	10.66	10.65	10.54
H ₂ O.....	1.22	1.06	1.43
F.....	6.78	7.44	7.02
	102.73	102.71	103.69
O equivalent to F.....	2.86	3.11	2.95
	99.87	99.60	100.74

BIOTITE.

A. From Merrow Ledge, Auburn, Me. Black. Analysis by F. W. Clarke and R. B. Riggs.

B. Bronze-mica, Laurel Creek corundum mine, Rabun County, Ga. Analysis by E. A. Schneider. The loss is due to undetermined volatile matter.

C. Brown mica from the granite of the Yosemite Valley, California. Analysis by William Valentine. Described by Turner, Am. Jour. Sci., 4th ser., vol. 7, p. 294.

D. Mica from quartz-monzonite, Tioga road, southeast of Mount Hoffman, Mariposa County, Cal. Specific gravity, 3.05 at 21°. Analysis by W. F. Hillebrand. Described by Turner, loc. cit.

E. Black mica from quartz-monzonite. About 1 kilometer south of Bloods Station, Alpine County, Cal. Analysis by William Valentine. Described by Turner, loc. cit.

F. Reddish-brown mica from pyroxenic gneiss. North fork of Mokelumne River, above mouth of Bear River, Amador County, Cal. Analysis by William Valentine. Described by Turner, loc. cit.

G. Biotite from the granite of Butte, Mont. See Weed, Jour. Geol., vol. 7, p. 737. Analysis by H. N. Stokes.

	A.	B.	C.	D.	E.	F.	G.
SiO ₂	34.67	36.12	35.64	35.75	35.62	36.62	35.79
TiO ₂		1.18	1.12	3.16	2.61	3.03	3.51
Al ₂ O ₃	30.09	20.49	18.62	14.70	15.24	14.37	13.70
V ₂ O ₃05			
Cr ₂ O ₃				trace			
Fe ₂ O ₃	2.42	3.29	5.54	4.65	4.69	4.04	5.22
FeO.....	16.14	5.17	14.60	14.08	13.67	17.09	13.72
MnO.....	.85	.17	.79	.45	.74	.40	.19
NiO.....		.34		.02			
CoO.....							
CaO.....	none	.28	.90	.17	.95	1.48	.05
SrO.....				(?)	trace	trace	
BaO.....		.09	trace	.12	.26	.33	.13
MgO.....	1.98	19.61	9.72	12.37	12.70	9.68	12.13
Li ₂ O.....		trace	trace		trace	trace	trace
Na ₂ O.....	1.67	.78	.38	.32	.50	.45	.15
K ₂ O.....	7.55	8.14	9.22	9.19	7.72	8.20	9.09
H ₂ O at 105°.....	4.64	1.58	.48	1.03	.94	.90	1.21
H ₂ O above 105°.....		.82	2.54	3.64	4.36	3.26	3.64
P ₂ O ₅20	.03	none	none	.10
F.....	.28	.60	.26	.17	none	.10	.76
Cl.....							.20
	100.29	98.66	100.01	99.90	100.00	99.95	99.59
Less O.....	.12	.26	.11	.07		.04	.37
	100.17	98.40	99.90	99.83		99.91	99.22

PHLOGOPITE.

A. From Edwards, St. Lawrence County, N. Y. Outwardly resembles talc. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 78.

B. From Burgess, Canada. Brown. Analysis by E. A. Schneider. See Clarke and Schneider, Bulletin 78.

C. Phlogopite separated from the wyomingite of the Leucite Hills, Wyoming. Described by Cross in Am. Jour. Sci., 4th ser., vol. 4, p. 115. Analysis by W. F. Hillebrand.

The following micas of doubtful character occur in serpentine. Analyses by Charles Catlett. See Bulletin 64.

D. Brown mica, from the serpentine of Montville, N. J.

E. Yellowish mica, same locality as D.

F. White mica, same locality.

G. White mica from near Easton, Pa.

E, F, and G are perhaps to be called vermiculites rather than micas.

	A.	B.	C.	D.	E.	F.	G.
SiO ₂	45.05	39.66	42.56	39.38	32.52	39.14	41.82
TiO ₂56	2.09				
Al ₂ O ₃	11.25	17.00	12.18	15.92	18.14	15.70	11.12
Cr ₂ O ₃73				
Fe ₂ O ₃27	2.73	.71	2.30	1.68	2.68
FeO.....	.14	.20	.90	.80			
CaO.....		none	.20	.28	1.44	5.24	
SrO.....			trace				
BaO.....		.62	1.00				
MgO.....	29.38	26.49	22.40	26.85	29.26	25.44	29.82
Li ₂ O.....	.07		trace				
Na ₂ O.....	.45	.60	.44	.62	1.38	.64	.36
K ₂ O.....	8.52	9.97	10.70	9.84	2.78	2.06	6.08
H ₂ O at 105°.....	5.37	.66		.38	.76	1.10	.94
H ₂ O above 105°.....		2.33	2.35	4.69	10.12	9.10	7.10
CO ₂30	1.80		
P ₂ O ₅		trace	.06				
F.....		2.24	2.46	.62			
O equivalent to F.....	100.23	100.60	100.80	100.39	100.50	100.10	99.92
		.94	1.03	.26			
		99.66	99.77	100.13			

LEPIDOMELANE.

A. From Litchfield, Me. Analyzed by F. W. Clarke, and described in Bulletin 42.

B. Same as A. Later analysis by Clarke. See Bulletin 55.

C. Annite. From Rockport, Mass. Analysis by R. B. Riggs. Discussion by Clarke in Bulletin 42.

D. From Port Henry, N. Y. Analysis by E. A. Schneider. See paper by Clarke and Schneider in Bulletin 78.

E. From Baltimore, Md. Analysis by F. W. Clarke, and discussed in Bulletin 55.

F. From near Pikes Peak, Colorado. Siderophyllite? The slightly altered margin of a large crystal.

G. Much altered central portion of specimen F. Analyses F and G by F. W. Clarke and R. B. Riggs. Described by Clarke in Bulletin 55. Classed by Dana under biotite.

	A.	B.	C.	D.	E.	F.	G.
SiO ₂	32.09	32.35	32.03	34.52	35.78	34.21	34.63
TiO ₂		trace	3.42	2.70			
Al ₂ O ₃	18.52	17.47	11.92	13.22	16.39	16.53	17.95
Fe ₂ O ₃	19.49	24.22	8.00	7.80	14.55	20.15	31.25
FeO	14.10	13.11	30.41	22.27	11.02	14.17	3.01
MnO	1.42	1.02	.21	.41	1.08	.91	.34
CaO		none	.23	none	none	.48	.81
MgO	1.01	.89	.06	5.82	8.67	1.34	1.08
NiO, CoO30			
Li ₂ O			trace	.04			
Na ₂ O	1.55	.70	1.54	.16	.56	1.43	.89
K ₂ O	8.12	6.40	8.46	8.59	7.76	6.50	1.96
H ₂ O at 105°	4.62	4.67	4.19	.57	4.48	4.54	7.82
H ₂ O above 105°				3.82			
P ₂ O ₅				trace			
F	none	none	trace	.34	none	.08	.54
O equivalent to F	100.92	100.83	100.47	100.56	100.29	100.34	100.28
				.14		.03	.22
				100.42		100.31	100.06

ROSCOELITE.

Analyses by W. F. Hillebrand.

A. From the Stockslager mine, near Lotus, Eldorado County, Cal. Described by Hillebrand in Bulletin 167.

B. Soluble portion of a vanadiferous sandstone from Placerville, Colo. Described by Hillebrand in Am. Jour. Sci., 4th ser., vol. 10, p. 130.

	A.	B.
SiO ₂	45.17	46.06
TiO ₂78
Al ₂ O ₃	11.54	22.55
V ₂ O ₃	24.01	12.84
Fe ₂ O ₃73
FeO.....	1.60
CaO.....44
BaO.....	1.35
MgO.....	1.64	.92
Li ₂ O.....	trace
Na ₂ O.....	trace	.22
K ₂ O.....	10.37	8.84
H ₂ O at 100°.....	.40	1.98
H ₂ O above 100°.....	4.29	4.07
	99.80	100.00

MARGARITE.

A. Brownish yellow, from Iredell County, N. C. Analysis by T. M. Chatard. See Bulletin 9.

B. Bright green, associated with pink corundum. From Gainesville, Ga. Specific gravity, 3.00. Analysis by T. M. Chatard. See Bulletin 9.

C. Inclusion in diorite, Crugers Station, near Peekskill, N. Y. Analysis by T. M. Chatard.

	A.	B.	C.
SiO ₂	31.15	31.72	32.73
Al ₂ O ₃	49.51	50.03	46.58
FeO.....	trace	trace	5.12
MgO.....	.45	.12	1.00
CaO.....	11.13	11.57	11.04
Na ₂ O.....	2.74	2.26
H ₂ O at 110°.....	5.68	4.88	.12
H ₂ O above 110°.....			4.49
	100.66	100.58	101.08

CHLORITOID.

From a phyllite rock near Liberty, Md. Analysis by L. G. Eakins.
See Bulletin 168, p. 50.

SiO ₂	23.40
Al ₂ O ₃	39.31
Fe ₂ O ₃	5.14
FeO	21.94
MgO	2.18
CaO	trace
Na ₂ O20
K ₂ O20
H ₂ O	6.81
TiO ₂	1.19
MnO	trace
P ₂ O ₅	trace
	<hr/>
	100.37

XANTHOPHYLLITE.

Variety waluewite, from the Nikolai-Maximilian mine, district of Slatoust, Urals, Siberia. Analysis by E. A. Schneider. Discussion by Clarke and Schneider in Bulletin 113.

SiO ₂	16.85
TiO ₂	trace
Al ₂ O ₃	42.33
Fe ₂ O ₃	2.35
FeO20
CaO	13.30
MgO	20.77
H ₂ O at 105°04
H ₂ O above 105°	4.56
	<hr/>
	100.40

THE VERMICULITES.

A. Jefferisite, from Westchester, Pa. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 78.

B. Vermiculite, near jefferisite or culsageeite, from Corundum Hill, North Carolina. Analyzed by T. M. Chatard and described by him in Bulletin 42.

C. Altered biotite from the zircon mine, Green River, Henderson County, N. C. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 90.

D. Kerrite, from near Franklin, Macon County, N. C. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 78.

E. Lucasite. Described as a new species by Chatard in Bulletin 42. Analysis by T. M. Chatard. From Corundum Hill, North Carolina. Specific gravity, 2.613 at 25.5°.

	A.	B.	C.	D.	E.
SiO ₂	34.20	37.96	38.18	38.13	39.81
TiO ₂			1.68		
Al ₂ O ₃	16.58	22.53	14.02	11.22	12.99
Cr ₂ O ₃54
Fe ₂ O ₃	7.41	11.12	13.02	2.28	5.29
FeO.....	1.13	.30	2.22	.18	.11
MnO.....		.12	.38		.05
NiO.....				.48	
CoO.....				trace	
CaO.....		none	.17		.14
BaO.....			.06		
MgO.....	20.41	15.46	14.62	27.39	24.83
Na ₂ O.....		undet.	.48		.20
K ₂ O.....		undet.	5.40		5.76
H ₂ O at 105°.....	} 21.14	} 12.63	3.20	} 20.47	3.78
H ₂ O above 105°.....			7.32		6.98
	100.87	100.12	100.75	100.15	100.48

F. Hallite from Nottingham, Chester County, Pa.

G. White lennilite from Lenni, Delaware County, Pa.

H. Brown lennilite, same locality.

I. Green lennilite, same locality.

Analyses F to I by E. A. Schneider. Discussion by Clarke and Schneider in Bulletin 90.

	F.	G.	H.	I.
SiO ₂	35.54	36.72	35.09	34.90
TiO ₂	undet.	.18	.58	.10
Al ₂ O ₃	9.74	10.06	12.05	10.60
Fe ₂ O ₃	9.07	5.37	6.67	8.57
Cr ₂ O ₃26	.46	.23
FeO.....	.28	.12	.11	.22
MnO.....	.25	.31	.27	.17
NiO.....	.16	.20	.20	.19
MgO.....	30.05	29.40	27.62	28.21
BaO.....			trace	
H ₂ O at 105°.....	2.64	6.40	5.70	4.99
H ₂ O above 105°.....	12.14	11.37	11.20	11.48
	99.87	100.39	99.95	99.66

J. A vermiculite from Newlin, Chester County, Pa.

K. Painterite from Middleton, Delaware County, Pa.

L. Another sample of painterite.

Analyses J, K, L by E. A. Schneider. Discussion by Clarke and Schneider in Bulletin 90.

M. Hydromica from Rocky Hill, N. J. Analysis by George Steiger. Described by Clarke and Darton in Bulletin 167.

N. Analysis M corrected by deduction of calcite, union of like radicals, and recalculation to 100 per cent.

O. Protovermiculite from Magnet Cove, Arkansas. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 90.

	J.	K.	L.	M.	N.	O.
SiO ₂	31.23	34.86	33.95	32.72	40.24	34.03
TiO ₂		trace	trace	.24		undet.
Al ₂ O ₃	17.52	11.64	12.52	8.41	10.34	14.49
Cr ₂ O ₃14					
Fe ₂ O ₃	4.70	3.78	4.40	19.99	24.57	7.71
FeO.....	1.20	.20	.20	4.24	5.21	.14
MnO.....	.20					.09
NiO.....	.33	.14	.23			
CaO.....		.07	none	10.30		1.88
MgO.....	31.36	31.32	30.56	5.51	6.78	20.89
Na ₂ O.....				.63		
K ₂ O.....				.85	2.20	
H ₂ O at 105°.....	1.08	1.64	1.56	2.47	3.03	11.23
H ₂ O above 105°.....	12.55	16.78	17.05	6.22	7.63	9.96
CO ₂				8.21		
	100.31	100.43	100.47	99.79	100.00	100.42

P. Chloritic vermiculite from Corundum Hill, North Carolina.

Q. Altered chlorite from Corundum Hill. Analyses P, Q by T. M. Chatard. Description in Bulletin 42.

	P.	Q.
SiO ₂	35.88	32.97
Al ₂ O ₃	20.90	17.88
Fe ₂ O ₃	6.55	4.76
FeO.....	3.68	.57
MnO.....		trace
CaO.....	.14	none
MgO.....	19.90	22.36
Alkalies.....	traces	undet.
H ₂ O at 110°.....	12.71	11.42
H ₂ O above 110°.....		10.05
	99.76	100.01

CLINOCHLORE.

A. From Westchester, Pa. Broadly foliated. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 78.

B. From the Nikolai-Maximilian mine, district of Slatoust, Urals, Siberia. Analysis by E. A. Schneider. Discussed by Clarke and Schneider in Bulletin 113.

C. Leuchtenbergite from Slatoust. Analyst and reference as under B.

D. Kotschubeite from Green Valley, American River Canyon, California. Specific gravity, 2.69. Analysis by W. H. Melville. Described by Melville and Lindgren in Bulletin 61.

	A.	B.	C.	D.
SiO ₂	29.87	30.84	30.00	31.74
Al ₂ O ₃	14.48	18.31	20.43	6.74
Co ₂ O ₃	1.56			11.39
Fe ₂ O ₃	5.52	1.94	1.68	
FeO.....	1.93	1.08	.14	1.23
NiO.....	.17			.49
CaO.....			.21	.18
MgO.....	33.06	34.38	34.26	35.18
H ₂ O at 105°.....	13.60	.55	.55	.37
H ₂ O above 105°.....		13.33	13.20	12.68
	100.19	100.43	100.47	100.00

PROCHLORITE.

From the Aqueduct Tunnel, Washington, D. C. Analysis by E. A. Schneider. Described by Clarke and Schneider in Bulletin 78.

SiO ₂	25.40
Al ₂ O ₃	22.80
Fe ₂ O ₃	2.86
FeO	17.77
MnO25
MgO	19.09
H ₂ O at 105°80
H ₂ O above 105°	11.41
F	trace
	100.38

SERPENTINE.

A. From Newburyport, Mass. Analysis by F. W. Clarke.

B. Same locality as A. Analysis by E. A. Schneider. Discussion by Clarke and Schneider in Bulletin 78.

C. Serpentine derived from sahlite, Osburn's soapstone quarry, Blandford, Mass. Analysis by W. F. Hillebrand.

D. Dark-green, oily serpentine, Middlefield, Mass. Analysis by W. F. Hillebrand.

E. Serpentine derived from enstatite, Granville, Mass.

Analysis by George Steiger. Serpentine C, D, E are described by Emerson in Monograph XXIX.

	A.	B.	C.	D.	E.
SiO ₂	41.32	41.47	40.77	38.62	37.82
TiO ₂			none	none	trace
Al ₂ O ₃			1.16	.35	.61
Cr ₂ O ₃28	.39	.19
Fe ₂ O ₃		1.73	3.56	3.44	7.92
FeO	2.36	.09	1.47	3.99	1.15
MnO09	.10	
NiO17	.21	.45
CoO05
CaO		none	none	.40	none
MgO	41.49	41.70	39.37	40.61	37.94
Li ₂ O			trace	trace	
Na ₂ O14	.10	trace
K ₂ O10	.08	trace
H ₂ O at 110°	} 14.54	1.20	.49	.36	.75
H ₂ O above 110°		13.86	12.48	10.91	12.50
P ₂ O ₅			trace	trace	trace
CO ₂			none	.52	
	99.71	100.05	100.08	100.08	99.38

F. Rich, dark-green serpentine, Rowe, Mass.

G. Black serpentine containing marmolite, Atwater's quarry, Russell, Mass.

H. Blackish-green serpentine, containing much chromite, North Blandford, Mass.

I. Gray, splintery serpentine, Chester, Mass. Analyses F to I by George Steiger. Description by Emerson in Monograph XXIX.

J. Black, serpentized boltonite, Stow, Mass. Analysis by W. F. Hillebrand.

	F.	G.	H.	I.	J.
SiO ₂	40.42	36.94	39.14	33.87	36.92
TiO ₂	none	trace	none	none	none
Al ₂ O ₃	1.86	.50	1.18	.77	.10
Cr ₂ O ₃28	.33	.33	.38	none
Fe ₂ O ₃	2.75	6.04	4.46	2.81	1.19
FeO.....	4.27	1.94	3.14	4.25	.87
MnO.....	trace	trace	none	.04	trace
NiO.....	.53	.40	.47	.33	-----
CoO.....	trace	none	trace		-----
CaO.....	.66	none	none	none	.59
MgO.....	35.95	38.35	41.45	38.57	43.99
Na ₂ O.....	.16	none	none	none	.05
K ₂ O.....		none	none	none	
H ₂ O at 100°.....	.21	.71	.34	.38	.72
H ₂ O above 100°.....	10.51	12.07	9.48	7.00	14.70
P ₂ O ₅	trace	trace	.02	trace	trace
CO ₂	1.44	1.85	none	10.82	.90
SO ₃	trace	.20	none	.20	-----
FeS ₂43	-----	-----	-----	-----
	99.47	99.33	100.01	99.42	100.03

K. From Moriah, N. Y. Analysis by Charles Catlett. Described by Merrill, Proc. U. S. Nat. Mus., vol. 12, p. 596.

L. From the aqueduct shaft, New York City. Analysis by Charles Catlett. See Merrill, loc. cit., p. 598.

M. From Montville, N. J., light yellowish-green.

N. From Montville, darker green. Analyses M, N, by Charles Catlett. See Merrill, Proc., U. S. Nat. Mus., vol. 11, p. 105.

O. From Montville, massive.

P. From Montville, chrysotile. Analyses O, P, by E. A. Schneider. Discussion by Clarke and Schneider in Bulletin 78.

	K.	L.	M.	N.	O.	P.
SiO ₂ -----	39.96	39.92	42.38	40.23	42.05	42.42
Al ₂ O ₃ -----	1.07	.08	.07	2.18	.30	.63
Fe ₂ O ₃ -----	3.53	.50	.97	4.02		.62
FeO -----	3.85	none	.17	trace?	.10	undet.
MnO -----	trace					none
NiO -----	none					.23
CaO -----	none	.90		trace	.05	trace
MgO -----	37.61	42.52	42.14	39.46	42.57	41.01
H ₂ O at 105° -----	} 13.65	1.36	} 14.12	} 14.24	.96	2.04
H ₂ O above 105° -----		13.26			13.70	13.60
CO ₂ -----		1.64				
	99.67	100.18	99.85	100.13	99.73	100.55

Q. From Easton, Pa. Resembles a vermiculite. Analysis by E. A. Schneider. See Clarke and Schneider, Bulletin 90.

R. Grayish-green serpentine from Corundum Hill, North Carolina.

S. Picrolite from Buck Creek, Clay County, N. C. Analyses R and S by E. A. Schneider. See Clarke and Schneider, Bulletin 78.

T. From the river Poldnewaja, district of Syskert, Urals, Siberia. Analysis by E. A. Schneider. See Clarke and Schneider, Bulletin 113.

U. From Greenville, Plumas County, Cal. Analysis by W. H. Melville. Described by Diller in Bulletin 150.

	Q.	R.	S.	T.	U.
SiO ₂ -----	43.71	41.90	42.94	42.55	39.14
Al ₂ O ₃ -----	3.59	.71	5.05	1.25	2.08
Fe ₂ O ₃ -----	.90	.91		1.56	4.27
FeO-----		undet.	1.88	1.52	2.04
NiO-----		.10	.61		
CaO-----				none	trace
MgO-----	38.58	40.16	36.53	40.05	39.84
Na ₂ O-----	.13				
K ₂ O-----	2.22				
H ₂ O at 105°-----	.46	2.26	1.53	.21	12.70
H ₂ O above 105°-----	10.79	13.90	11.68	12.26	
Chromite-----				.37	.11
	100.38	99.94	100.22	99.77	100.18

V. Light-green marmolitic serpentine, New Idria, Cal.

W. Black serpentine, Sulphur Bank, California.

X. Light-green serpentine, Sulphur Bank.

Analyses V, W, X by W. H. Melville. Described by Becker in Monograph XIII. Other analyses by Melville of impure serpentines from near Mount Diablo, California, are given in Bulletin 168.

Y. From Three Brothers, Mount Stuart quadrangle, Washington. Analysis by W. F. Hillebrand.

	V.	W.	X.	Y.
SiO ₂	41.54	39.64	41.86	39.00
TiO ₂				trace
Al ₂ O ₃	2.48	1.30	.69	1.75
Cr ₂ O ₃29	.24	.47
Fe ₂ O ₃				5.16
FeO.....	1.37	7.76	4.15	1.71
MnO.....		.12	.20	.15
NiO.....	.04	.33	trace	.10
CaO.....				trace
MgO.....	40.42	37.13	38.63	38.00
Alkalies.....				.10
H ₂ O at 110°.....	14.18	13.81	14.16	1.31
H ₂ O above 110°.....				12.43
P ₂ O ₅				trace
FeS ₂03
	100.03	100.38	99.93	100.21

GENTHITE.

From Riddle, Douglas County, Oreg. Analyzed by F. W. Clarke, and described by Diller and Clarke in Bulletin 60.

SiO ₂	44.73
Al ₂ O ₃	1.18
Fe ₂ O ₃	
MgO.....	10.56
NiO.....	27.57
H ₂ O at 110°.....	8.87
H ₂ O above 110°.....	6.99
	99.90

TALC.

Apple-green, beautifully foliated. From Huntersville, Fairfax County, Va. Analysis by E. A. Schneider.

SiO ₂	62.27
Al ₂ O ₃15
Fe ₂ O ₃95
FeO.....	.85
MgO.....	30.95
MnO.....	trace
H ₂ O at 105°.....	.07
H ₂ O, ignition.....	4.84
	<hr/>
	100.08

GLAUCONITE.

From Big Goose Canyon, 15 miles southwest of Sheridan, Big Horn Mountains, Wyoming. Analysis by George Steiger. Specific gravity, 2.73.

SiO ₂	49.23
Al ₂ O ₃	7.11
Fe ₂ O ₃	20.89
FeO.....	3.06
MnO.....	trace
CaO.....	trace
MgO.....	3.44
Na ₂ O.....	.11
K ₂ O.....	8.51
H ₂ O at 100°.....	1.83
H ₂ O above 100°.....	4.88
	<hr/>
	99.06

KAOLINITE.

A. From Hockessen, Del. Analysis by George Steiger.

B. From Aiken, S. C. Analysis by F. W. Clarke.

C. From Talladega, Ala. Analysis by Charles Catlett.

D. From the Eureka vein, St. Peter's Dome, near Pikes Peak, Colorado. Analysis by W. F. Hillebrand. Described by Cross and Hillebrand in Bulletin 20.

E. From New Discovery mine, Leadville, Colo. Analysis by W. F. Hillebrand.

F. From National Belle mine, Ouray County, Colo. Analysis by W. F. Hillebrand. Specific gravity, 2.611 at 18.5°. Described in Bulletin 20.

	A.	B.	C.	D.	E.	F.
SiO ₂	48.73	45.56	43.21	46.06	43.66	46.35
Al ₂ O ₃	37.02	40.25	37.27	39.63	37.78	39.59
Fe ₂ O ₃79					.11
CaO16		.11		.22	
MgO11		.10		.30	
Na ₂ O04		.40		trace	
K ₂ O41		.28		trace	
H ₂ O at 100°52	} 14.10	5.02	} 13.77	} 17.95	} 13.93
H ₂ O above 100°	12.83		13.48			
P ₂ O ₅03					
TiO ₂17					
SO ₃					trace	
F15
CaF ₂68		
	100.81	99.91	99.87	100.14	99.91	100.13
O equivalent to F06
						100.07

HALLOYSITE.

A. From the Detroit copper mine, near Mono Lake, California. Analyzed by F. W. Clarke, and described in Bulletin 9.

B. From the Logan mine, Rico, Colo. Analysis by W. F. Hillebrand.

	A.	B.
SiO ₂	42.91	38.65
Al ₂ O ₃	38.13	33.27
Fe ₂ O ₃22
H ₂ O at 100°	} 18.95	13.70
H ₂ O above 100°		14.34
	99.99	100.18

CIMOLITE.

Rose-red, earthy variety, from Norway, Mc. Analysis by R. B. Riggs. Described by Clarke in Bulletin 42.

SiO ₂	66.86
Al ₂ O ₃	22.23
Fe ₂ O ₃47
FeO18
MnO07
CaO42
MgO33
Li ₂ O29
Na ₂ O46
K ₂ O18
H ₂ O	8.26
F06
	<hr/>
	99.81

PYROPHYLLITE.

Compact, white. From Deep River, North Carolina. Analysis by George Steiger.

SiO ₂	64.73
TiO ₂73
Al ₂ O ₃	29.16
Fe ₂ O ₃49
MgO	trace
Ignition	5.35
	<hr/>
	100.46

BOLE.

From South Table Mountain, Golden, Colo. Described by Cross and Hillebrand in Bulletin 20. Analyses by W. F. Hillebrand.

A. Dark brown.

B. Light brown.

	A.	B.
SiO ₂	42.63	46.17
Al ₂ O ₃	18.76	22.03
Fe ₂ O ₃	11.88	4.64
CaO	2.59	2.30
MgO	3.39	2.42
K ₂ O35	} ^a 2.06
Na ₂ O24	
H ₂ O	20.21	20.38
	<hr/>	<hr/>
	100.05	100.00

^a By difference.

VIII. TITANATES AND TITANO-SILICATES.

ILMENITE.

From the peridotite dike of Elliott County, Ky. Described by Diller in Bulletin 38. Analysis by T. M. Chatard.

TiO ₂	49.32
SiO ₂76
Al ₂ O ₃	2.84
Cr ₂ O ₃74
Fe ₂ O ₃	9.13
FeO.....	27.81
MnO.....	.20
CaO.....	.23
MgO.....	8.68
Alkalies.....	.19
P ₂ O ₅	trace
Ignitium.....	.20
	<hr/>
	100.10

XANTHITANE.

From Green River, Henderson County, N. C. Alteration product of titanite. Analyzed by L. G. Eakins, and described in Bulletin 60. Specific gravity, 2.941 at 24°. Material dried at 100°.

TiO ₂	61.54
SiO ₂	1.76
Al ₂ O ₃	17.59
Fe ₂ O ₃	4.46
CaO.....	.90
MgO.....	trace
P ₂ O ₅	4.17
H ₂ O.....	9.92
	<hr/>
	100.34

TITANITE.

Pale-yellow, semi-translucent. From the waterworks tunnel, Washington, D. C. Occurs embedded in prochlorite. Analysis by F. W. Clarke. Specific gravity, 3.452.

TiO ₂	40.82
SiO ₂	30.10
MnO.....	trace
CaO.....	28.08
MgO.....	.40
Ignition.....	.54
	<hr/>
	99.94

TSCHEFFKINITE.

From Bedford County, Va. Analyzed by L. G. Eakins and described by him in Bulletin 90.

A. Lustrous portion. Specific gravity, 4.33 at 27°.

B. Dull portion. Specific gravity, 4.38 at 22°.2.

	A.	B.
TiO ₂	18.78	18.99
SiO ₂	20.21	21.49
Ta ₂ O ₅08	.08
ZrO ₂	trace?	trace?
ThO ₂85	.75
(Y, Er) ₂ O ₃	1.82	1.64
(La, Di) ₂ O ₃	19.72	17.16
Ce ₂ O ₃	20.05	19.08
Al ₂ O ₃	3.60	3.65
Fe ₂ O ₃	1.88	2.89
FeO	6.91	5.92
CaO	4.05	5.24
MgO55	.48
Na ₂ O06	.04
H ₂ O94	2.06
	99.50	99.47

ASTROPHYLLITE.

From St. Peter's Dome, near Pikes Peak, Colorado. Analyzed by L. G. Eakins and described by him in Bulletin 90.

TiO ₂	11.40
SiO ₂	35.23
ZrO ₂	1.21
Ta ₂ O ₅34
Fe ₂ O ₃	3.73
Al ₂ O ₃	trace
FeO	29.02
MnO	5.52
CaO22
MgO13
K ₂ O	5.42
Na ₂ O	3.63
H ₂ O	4.18
	100.03

IX. COLUMBO-TANTALATES.

SAMARSKITE.

Mineral near samarskite, from Devils Head Mountain, near Pikes Peak, Colorado. Analyzed by W. F. Hillebrand and described in Bulletin 55.

A. Pitch-black variety, streak dirty brown. Specific gravity, 6.18 at 22°.

B. Black, streak salmon colored. Specific gravity, 6.12 at 25°.

C. Altered variety: Specific gravity, 5.45 at 16°.

	A.	B.	C.
Ta ₂ O ₅	27.03	28.11	19.34
Cb ₂ O ₅	27.77	26.16	27.56
WO ₃	2.25	2.08	5.51
SnO ₂95	1.09	.82
ZrO ₂	2.29	2.60	^a 3.10
UO ₂	4.02	4.22
UO ₃	6.20
ThO ₂	3.64	3.60	3.19
Ce ₂ O ₃54	.49	.41
(La, Di) ₂ O ₃	1.80	2.12	1.44
Er ₂ O ₃	10.71	10.70	9.82
Y ₂ O ₃	6.41	5.96	5.64
Fe ₂ O ₃	8.77	8.72	8.90
FeO32	.35	^b .39
MnO78	.75	} .77
ZnO05	.07	
PbO72	.80	
CaO27	.33	1.61
MgO11
K ₂ O17	.13	} .36
(Na, Li) ₂ O24	.17	
H ₂ O	1.58	1.30	3.94
F	(?)	(?)	(?)
	100.31	99.75	100.18

^a With TiO₂.

^b Or 0.74 UO₂.

X. BORATES.

COLEMANITE.

A. Transparent crystal, ordinary type, Death Valley, California.

B, C. Blade-like crystals, Death Valley.

D. Priceite, Curry County, Oreg.

D. Pandermite, Island of Panderna, Black Sea.

Analyses by J. E. Whitfield, with details in Bulletin 55.

	A.	B.	C.	D.	E.
B ₂ O ₃	50.70	49.56	49.62	48.44	48.63
CaO.....	27.31	27.36	27.40	32.15	32.16
MgO.....	.10	.25	.26	-----	-----
H ₂ O.....	21.87	22.66	22.70	19.42	19.40
SiO ₂	-----	.44	.47	-----	-----
	99.98	100.27	100.45	100.01	100.19

ULEXITE.

From Rhodes' Marsh, Esmeralda County, Nev. Analysis by J. E. Whitfield, with description in Bulletin 55.

B ₂ O ₃	43.20
CaO.....	14.52
Na ₂ O.....	10.20
K ₂ O.....	.44
H ₂ O.....	29.46
SiO ₂04
SO ₃28
Cl.....	2.38
	100.52
O equivalent to Cl.....	.53
	99.99

LUDWIGITE.

From Morawitzza, Banat, Hungary. Analysis by J. E. Whitfield, with description in Bulletin 55.

B ₂ O ₃	12.04
Fe ₂ O ₃	37.93
FeO.....	15.78
MgO.....	30.57
MnO.....	.16
H ₂ O.....	3.62
	100.10

WARWICKITE.

From Edenville, Orange County, N. Y. Somewhat contaminated by spinel, from which the warwickite could not be entirely freed. Analysis by J. E. Whitfield. See Bulletin 64.

B ₂ O ₃	18.96
TiO ₂	18.68
SiO ₂	1.16
Al ₂ O ₃	9.44
FeO.....	14.23
CaO.....	.38
MgO.....	34.41
H ₂ O.....	2.80
	<hr/> 100.06

XI. NITRATES.

SODA NITER.

From the Leucite Hills, Wyoming. Analysis by L. G. Eakins. The N₂O₅ was not determined directly, but calculated to satisfy Na₂O+K₂O.

Na ₂ O.....	32.09
K ₂ O.....	4.97
N ₂ O ₅	61.58
CaO.....	.24
SO ₃33
Cl.....	trace
H ₂ O.....	.68
	<hr/> 99.89

NITER.

A. From Utah, exact locality unknown. Analysis by T. M. Chatard.

B. From the Leucite Hills, Wyoming. Analysis by L. G. Eakins, with the N₂O₅ calculated to satisfy the alkalies.

	A.	B.
Insoluble matter.....	12.12	-----
K ₂ O.....	38.38	44.91
Na ₂ O.....	.12	.07
N ₂ O ₅	44.30	51.49
SiO ₂20	-----
Al ₂ O ₃		-----
CaO.....	1.43	1.09
MgO.....	trace	-----
NaCl.....	.08	-----
Cl.....	-----	.09
SO ₃	2.05	1.59
H ₂ O.....	1.24	.62
	<hr/> 99.92	<hr/> 99.86

XII. PHOSPHATES.

XENOTIME.

From the gold washings at Brindletown, N. C. Analyses by L. G. Eakins.

A. Green, specific gravity, 4.68 at 24.2°.

B. Brown, specific gravity, 4.46 at 24.4°.

	A.	B.
SiO ₂	3.46	3.56
ZrO ₂	1.95	2.19
UO ₂	4.13	1.73
ThO ₂	trace	trace
Al ₂ O ₃77	1.57
Fe ₂ O ₃65	2.79
(La,Di) ₂ O ₃93	.77
(Y,Er) ₂ O ₃ ^a	56.81	55.43
CaO.....	.21	.19
P ₂ O ₅	30.31	29.78
F.....	.06	.56
H ₂ O.....	.57	1.49
	99.85	100.06

^a Molecular weight, 260.

APATITE.

Dark-green, massive apatite from the topaz locality at Stoneham, Me. Analysis by J. E. Whitfield. See Bulletin 27. Specific gravity, 3.27.

P ₂ O ₅	40.36
CaO.....	47.60
MgO.....	6.08
FeO.....	1.44
H ₂ O.....	.11
Cl.....	.29
F.....	6.84
	102.72
Excess O.....	2.94
	99.78

TRIPLITE.

From a tin mine near Rapid City, S. Dak. Probably identical with Headden's "griphite." Analysis by L. G. Eakins, with description in Bulletin 60.

P ₂ O ₅	39.68
SiO ₂43
Al ₂ O ₃	8.74
Fe ₂ O ₃	2.36
FeO.....	1.97
MnO.....	29.13
CaO.....	6.72
MgO.....	trace
Li ₂ O.....	.13
Na ₂ O.....	5.25
K ₂ O.....	trace
H ₂ O.....	3.67
CO ₂26
F.....	2.35
Cl.....	.25
	100.94
Excess O.....	1.05
	99.89

TURQUOISE.

From Los Cerillos, N. Mex. Described by Clarke and Diller in Bulletin 42. Analyses by F. W. Clarke.

A. Bright blue, faintly translucent in thin splinters.

B. Pale blue, with a slight greenish cast. Opaque and earthy in texture. Specific gravity, 2.805.

C. Dark green, opaque.

	A.	B.	C.
P ₂ O ₅	31.96	32.86	28.63
SiO ₂	1.15	.16	4.20
Al ₂ O ₃	39.53	36.88	37.88
Fe ₂ O ₃		2.40	4.07
CuO.....	6.30	7.51	6.56
CaO.....	.13	.38	undet.
H ₂ O.....	19.80	19.60	18.49
	98.87	99.79	99.83

XIII. VANADATES.

DESCLOIZITE.

A. From the Mayflower mine, Bald Mountain mining district, Beaverhead County, Mont. Yellow, friable, not crystallized.

B. From the Commercial mine, Georgetown, N. Mex. Brilliantly crystallized.

C. Cuprodescloizite from the Lucky Cuss mine, Tombstone, Ariz. Specific gravity, 5.88 at 19°.

Analyses by W. F. Hillebrand, with description in Bulletin 64.

	A.	B.	C.
V ₂ O ₅	20.80	20.44	19.79
As ₂ O ₅32	.94	1.10
P ₂ O ₅27	.26	.19
SiO ₂18	1.01	.80
CO ₂82
PbO	55.93	56.01	57.00
ZnO	15.94	17.73	4.19
CuO	1.15	1.05	11.21
FeO70	.07	trace
CaO10	.04	1.01
MgO06	.03	.04
K ₂ O10
Na ₂ O17
H ₂ O	4.37	2.45	2.50
Cl04	.07
	99.82	100.07	98.99 °

CARNOTITE.

From Montrose County, Colo. See memoir by Hillebrand and Ransome, Am. Jour. Sci., 4th ser., vol. 10, p. 120.

A, B, C. From Copper Prince claim, Roc Creek.

D, E. From Yellow Boy claim, La Sal Creek.

Analyses by W. F. Hillebrand.

	A.	B.	C.	D.	E.
V ₂ O ₅ -----	18.35	18.49	15.76	17.80	18.05
As ₂ O ₅ -----	.25	trace	none	none	none
P ₂ O ₅ -----	.33	.80	.40	trace	.05
SiO ₂ -----		.15	.13	5.05	.20
TiO ₂ -----	.10	.03	(?)	-----	(?)
CO ₂ -----	.33	.56	none	-----	none
SO ₃ -----	.12	-----	.18	-----	none
UO ₃ -----	52.25	54.89	47.42	52.28	54.00
MoO ₃ -----	.23	.18	.18	undet.	.05
CrO ₃ -----	trace	-----	-----	-----	-----
Fe ₂ O ₃ -----	1.77	.21	.72	3.36	.42
Al ₂ O ₃ -----	1.08	.09	.08	undet.	.29
PbO-----	.25	.13	.18	undet.	.07
CuO-----	.20	.15	.22	-----	trace
CaO-----	2.85	3.34	2.57	1.85	1.86
SrO-----	-----	.02	(?)	-----	trace
BaO-----	.72	.90	.65	3.21	2.83
MgO-----	.20	.22	.24	.17	.14
Li ₂ O-----	none	trace	(?)	trace	trace
Na ₂ O-----	.09	.14	.07	.07	.13
K ₂ O-----	6.73	6.52	6.57	5.32	5.46
H ₂ O at 105°-----	2.59	2.43	1.85	4.52	3.16
H ₂ O above 105°-----	3.06	2.11	2.79	3.87	2.21
Insoluble-----	8.34	7.10	19.00	-----	10.33
	99.84	98.46	99.01	97.50	99.25

XIV. ARSENATES AND ANTIMONATES.

OLIVENITE.

From the American Eagle mine, Tintic mining district, Utah. Described by Hillebrand in Bulletin 20. Analysis by W. F. Hillebrand.

As ₂ O ₅	40.05
P ₂ O ₅06
CuO.....	55.40
Fe ₂ O ₃25
CaO.....	.16
ZnO.....	trace
H ₂ O.....	3.39
Quartz.....	.40
	<hr/> 99.71

ERINITE.

From the Mammoth mine, Tintic district, Utah. Described by Hillebrand and Washington in Bulletin 55. Analyses by W. F. Hillebrand.

	A.	B.
As ₂ O ₅	33.53	31.91
P ₂ O ₅10	-----
CuO.....	57.67	57.51
ZnO.....	1.06	.59
CaO.....	.32	.51
MgO.....	trace	trace
Fe ₂ O ₃14	.20
H ₂ O.....	7.22	9.15
	<hr/> 100.04	<hr/> 99.87

CLINOCLASITE.

From the Mammoth mine, Tintic district, Utah. Described by Hillebrand and Washington in Bulletin 55. Analyses by W. F. Hillebrand. Specific gravity, 4.38 at 19°.

	A.	B.
As ₂ O ₅	29.59	29.60
P ₂ O ₅05	.05
SiO ₂06	.06
CuO.....	62.34	62.54
ZnO.....	.96	.04
Fe ₂ O ₃12	.12
H ₂ O.....	7.73	7.72
	<hr/> 99.95	<hr/> 100.13

^a Assumed the same as in A.

CONICALCITE.

From the American Eagle mine, Tintic district, Utah. Analysis by W. F. Hillebrand, with description in Bulletin 20.

As ₂ O ₅	39.94
P ₂ O ₅14
CO ₂ (by difference)97
CuO.....	28.68
ZnO.....	2.86
CaO.....	19.79
MgO.....	.54
Fe ₂ O ₃36
Ag.....	.30
H ₂ O.....	5.52
Quartz.....	.90
	<hr/> 100.00

TYROLITE.

From the Mammoth mine, Tintic district, Utah. Described by Hillebrand and Washington in Bulletin 55. Analyses by W. F. Hillebrand. Specific gravity, 3.27 at 20°.5.

	A.	B.
As ₂ O ₅	28.78	26.22
P ₂ O ₅	trace	trace
CuO.....	45.22	46.38
ZnO.....	.04	trace
CaO.....	6.84	6.69
MgO.....	.05	.04
H ₂ O.....	17.26	17.57
SO ₃	(?)	2.27
	<hr/> 98.19	<hr/> 99.17

CHENEVIXITE.

From the American Eagle mine, Tintic district, Utah. Analyzed by W. F. Hillebrand and described in Bulletin 20.

As ₂ O ₅	35.14
CuO.....	26.31
CaO.....	.44
MgO.....	.16
Fe ₂ O ₃	27.37
Al ₂ O ₃66
H ₂ O.....	9.33
Quartz.....	.40
	<hr/> 99.81

MIXITE.

From the Mammoth mine, Tintic district, Utah. Described by Hillebrand and Washington in Bulletin 55. Analysis by W. F. Hillebrand.

As ₂ O ₅	28.79
P ₂ O ₅06
SiO ₂42
CuO.....	43.89
ZnO.....	2.70
CaO.....	.26
Bi ₂ O ₃	11.18
Fe ₂ O ₃97
H ₂ O.....	11.04
	<hr/>
	99.31

SCORODITE.

An incrustation on hot spring deposits, from Joseph's Coat Spring, Broad Creek, Yellowstone Park. Analyzed by J. E. Whitfield and described in Bulletin 55.

As ₂ O ₅	46.48
Fe ₂ O ₃	33.29
H ₂ O.....	15.50
SiO ₂	4.35
SO ₃84
	<hr/>
	100.46

BINDHELMITE.

From a claim near the Bertrand mine, Secret Canyon, Nevada. Analyzed by W. F. Hillebrand and described in Bulletin 20. Specific gravity, 5.01 at 19°, after correction for admixed quartz and cerussite.

Sb ₂ O ₅	35.20
PbO.....	49.50
CuO.....	.58
ZnO.....	.18
CaO.....	.66
MgO.....	.03
K ₂ O.....	.14
Na ₂ O.....	.21
Fe ₂ O ₃09
Ag.....	.29
CO ₂	3.35
H ₂ O.....	5.86
Quartz.....	4.59
	<hr/>
	100.68

XV. SULPHATES AND TELLURITES.

GYPSUM.

A. From Hillsboro, New Brunswick. Analysis by George Steiger.

B. From the Western Plaster Works, Alabaster, Mich. Analysis by George Steiger.

C. From east of Cascade, Black Hills, South Dakota. Analysis by George Steiger.

D. From Rico-Aspen mine, Rico district, Colorado. Analysis by W. F. Hillebrand.

E, F. From Nephi, Utah. Analyses by E. T. Allen. Some anhydrite must be present.

	A.	B.	C.	D.	E.	F.
SO ₃	46.18	46.18	45.45	45.07	48.14	39.53
CO ₂85	1.54	.65	7.73
Cl	trace	.03			trace	.04
SiO ₂10	.51		
TiO ₂				trace		
Al ₂ O ₃	} .10	} .08	.12	.03		} .14
Fe ₂ O ₃09		
CaO	32.37	32.33	32.44	32.49	35.29	38.46
SrO10		
MgO	trace	.05	.33	.92	trace	.24
Na ₂ O	} .10	} .14		trace		.07
K ₂ O19
H ₂ O	20.94	20.96	20.80	19.67	15.88	12.69
Insoluble10	.05				.45
Organic matter				present		
	99.79	99.82	100.09	100.42	99.96	99.54

BROCHANTITE.

From the United Verde mine, Jerome, Ariz. Analysis by W. F. Hillebrand.

SO ₃	16.75
CuO	69.45
PbO04
Fe ₂ O ₃ (from gangue)28
CaO	trace
H ₂ O	12.05
P ₂ O ₅	trace
CO ₂71
Insoluble70
	99.98

ANTLERITE.

From the Antler mine, Yucca Station, Mohave County, Ariz. Described by Hillebrand as a new species in Bulletin 55. Specific gravity, 3.93 at 16°, corrected for gangue. Analyses by W. F. Hillebrand.

A, B. First lot received.

C. Later sample.

	A.	B.	C.
SO ₃	18.78	18.48	20.11
CuO	62.48	62.69	63.26
ZnO27	.27	.04
CaO06	.05	.04
MgO03	.03	-----
Al ₂ O ₃ , Fe ₂ O ₃14	.12	.15
SiO ₂06	undet	.05
H ₂ O	10.21	10.18	10.05
Insoluble gangue	8.01	8.02	6.27
	100.04	99.84	99.97

PISANITE.

From Bingham, Utah. Analysis by W. F. Hillebrand.

SO ₃	28.52
CuO	12.60
FeO	14.13
ZnO10
H ₂ O	44.92
	<hr/> 100.27

PICRALLUMOGENE.

From near Las Vegas, N. Mex. Analysis by W. F. Hillebrand.

SO ₃	37.51
Al ₂ O ₃	9.71
MnO06
MgO	6.75
Na ₂ O	2.09
K ₂ O28
H ₂ O	43.75
	<hr/> 100.15

HALOTRICHITE.

From the headwaters of Gila River, Grant County, N. Mex., about 40 miles north of Silver City. Analyzed by F. W. Clarke, and described in Bulletin 9.

SO ₃	37.19
Al ₂ O ₃	7.27
FeO.....	13.59
H ₂ O.....	40.62
Insoluble.....	.50
	99.17

ALUNOGEN.

A. From headwaters of Gila River, Grant County, N. Mex., about 40 miles north of Silver City. Analyzed by F. W. Clarke, and described in Bulletin 9. Color, pinkish.

B. From the calcite spring, Yellowstone Park. Fine, white, silky fibers. Analysis by J. E. Whitfield.

C. From the Grand Canyon, Yellowstone Park. Analysis by J. E. Whitfield.

	A.	B.	C.
SO ₃	34.43	40.65	38.22
Al ₂ O ₃	15.52	15.72	16.80
Fe ₂ O ₃		trace	trace
CaO.....		trace	none
MgO.....		1.53	.18
K ₂ O.....			.01
Na ₂ O.....			.63
H ₂ O.....	42.56	41.81	43.64
SiO ₂09	1.38
Insoluble.....	7.62		
	100.13	99.80	100.86

COPIAPITE.

A. From the Redington mine, Knoxville, Cal.

B. From Sulphur Bank, Lake County, Cal.

Analyses by W. H. Melville. Description by Melville and Lindgren in Bulletin 61.

	A.	B.
SO ₃	39.97	38.82
Al ₂ O ₃37
Fe ₂ O ₃	26.54	26.79
FeO46	3.28
MnO21	trace
CaO25
MgO	3.06	.16
H ₂ O	30.43	29.58
Residue75
	100.67	100.00

KNOXVILLITE.

From the Redington mine, Knoxville, Cal. Described as a new species by Melville and Lindgren in Bulletin 61. Analysis by W. H. Melville.

SO ₃	35.91
Cr ₂ O ₃	7.41
Al ₂ O ₃	4.83
Fe ₂ O ₃	15.86
FeO	3.81
NiO83
MgO	3.22
H ₂ O at 100°	9.30
H ₂ O above 100°	17.60
Residue	1.73

100.00

REDINGTONITE.

From the Redington mine, Knoxville, Cal. Described as a new species by Melville and Lindgren in Bulletin 61. Analysis by W. H. Melville. Specific gravity, 1.761.

SO ₃	35.35
Cr ₂ O ₃	7.51
Al ₂ O ₃	5.14
Fe ₂ O ₃19
FeO	4.58
NiO	1.00
MnO	trace
MgO	1.85
H ₂ O at 100°	27.09
H ₂ O above 100°	14.34
Residue	3.46

100.51

ALUNITE.

A. From Knickerbocker Hill, Custer County, Colo. Analysis by L. G. Eakins.

B. From Tres Cerritos Buttes, southwest of Indian Gulch, Mariposa County, Cal. Analysis by William Valentine.

	A.	B.
SO ₃	35.91	38.50
Al ₂ O ₃	38.91	38.05
Fe ₂ O ₃23
CaO.....	.35	.55
MgO.....	trace	trace
K ₂ O.....	4.03	4.48
Na ₂ O.....	4.32	2.78
H ₂ O.....	13.03	11.92
SiO ₂	2.82	2.64
TiO ₂40
P ₂ O ₅		trace
	99.37	99.55

JAROSITE.

A. From Black Iron mine, Eagle County, Colo.

B. From Pigeon mine, Rico district, Colorado.

Analyses by W. F. Hillebrand.

	A.	B.
SO ₃	30.72	28.20
Fe ₂ O ₃	46.37	43.81
PbO.....	1.80	
MgO.....	trace	trace
CaO.....		.06
K ₂ O.....	8.14	7.44
Na ₂ O.....	.27	.08
H ₂ O.....	11.40	11.64
Al ₂ O ₃		1.00
P ₂ O ₅33
As ₂ O ₅51	
SiO ₂72	7.35
	99.93	99.91

NATROJAROSITE.

From Soda Springs Valley, on road from Sodaville to Vulcan copper mine, Nevada. Described by Hillebrand and Penfield in Am. Jour. Sci., 4th ser., vol. 14, p. 211. Analysis by W. F. Hillebrand. Specific gravity, 3.18 at 30.5°.

SO ₃	30.96
Fe ₂ O ₃	50.98
CaO.....	.04
K ₂ O.....	.35
Na ₂ O.....	6.03
H ₂ O at 105°.....	.12
H ₂ O above 105°.....	11.03
SiO ₂23
As ₂ O ₅20
	<hr/>
	99.94

PLUMBOJAROSITE.

From Cooks Peak, New Mexico. Described as a new species by Hillebrand and Penfield in Am. Jour. Sci., 4th ser., vol. 14, p. 211. Analysis by W. F. Hillebrand. Specific gravity, 3.668 at 30.5°.

SO ₃	27.06
Fe ₂ O ₃	42.37
Al ₂ O ₃ ?.....	.10
PbO.....	19.84
CuO.....	.27
CaO.....	.05
MgO.....	.01
K ₂ O.....	.17
Na ₂ O.....	.21
H ₂ O at 105°.....	.02
H ₂ O above 105°.....	9.54
SiO ₂51
	<hr/>
	100.15

EMMONSITE.

A hydrous ferric tellurite from near Tombstone, Ariz. Analyzed partially by W. F. Hillebrand, and described by him as a new species in Proc. Colorado Sci. Soc., vol. 2, pt. 1, p. 21. The data relative to the composition of emmonsite are as follows:

	I.	II.	III.	IV.
Te(Se)	59.77	59.15	59.05	59.14
Fe	14.00	14.06	14.90	14.20
ZnO				1.94
CaO56
H ₂ O	3.28			

XVI. MOLYBDATES, TUNGSTATES, AND URANATES.

POWELLITE.

From the Seven Devils mining district, Idaho. Analyzed by W. H. Melville, and described by him as a new mineral species in Bulletin 90. Specific gravity, 4.526.

MoO ₃	58.58
WO ₃	10.28
CaO	25.55
MgO16
Fe ₂ O ₃	1.65
Al ₂ O ₃	trace
CuO	trace
SiO ₂	3.25
S	undet.
	<hr/> 99.47

HÜBNERITE.

From the Royal Albert vein, Uncompahgre district, Ouray County, Colo. Analyzed by W. F. Hillebrand, and described by him in Bulletin 20. Specific gravity, 7.177 at 24°.

WO ₃	75.58
Cb ₂ O ₅ (?)05
SiO ₂62
MnO	23.40
FeO24
CaO13
	<hr/> 100.02

URANINITE.

Analyses by W. F. Hillebrand. Discussed in Bulletins 78 and 90. In the original publications the gaseous constituent of uraninite was supposed to be nitrogen, as indeed it was in part. The discovery of helium, however, has shed new light upon the subject, and the analyses have been corrected accordingly,

A, B, C, D, E. Crystallized uraninite from Hale's quarry, Glastonbury, Conn. Nitrogen was certainly present in these specimens.

	A.	B.	C.	D.	E.	
UO ₃ -----	22.08	23.35	22.22	26.48	23.03	
UO ₂ -----	59.13	58.01	59.31	57.43	59.93	
ThO ₂ -----	9.57	9.78	10.31	9.79	11.10	
CeO ₂ -----				.25		
ZrO ₂ ?-----		.46		.13		
(LaDi) ₂ O ₃ -----				.20		
(YEr) ₂ O ₃ -----						
PbO-----	3.14	3.24	3.07	3.26	3.08	
CaO-----	.08		undet.	.08	.11	
MgO-----			undet.			
Alkalies-----			undet.	trace		
H ₂ O-----	.97	undet.	undet.	.61	.43	
He, etc-----	undet.	undet.	undet.	undet.	.34	
Fe ₂ O ₃ -----	1.21	.33	.67	.40	.29	
MnO-----				trace		
SiO ₂ -----	1.06		.25	.16	.16	
P ₂ O ₅ -----					.02	
F-----					.04	
Insoluble-----	.85	1.74	.42	.70	.89	
Cb ₂ O ₅ -----	.96					
	99.05	96.91	96.25	99.49	99.42	
Specific gravity-----	9.139	9.051		9.587	9.622	

F, G, H. Brilliantly crystallized, from Branchville, Conn.

I, J. Altered uraninite, from the Flat Rock mine, Mitchell County, N. C.

K. Massive, from near Blackhawk, Colo.

L. From Marietta, S. C.

	F.	G.	H.	I.	J.	K.	L.
UO ₃	13.27	21.54	14.00	50.83	44.11	25.26	} 83.95
UO ₂	72.25	64.72	70.99	39.31	46.56	58.51	
TiO ₂						trace	
ZrO ₂	} 7.20	.33	} 6.52			7.59	.20
ThO ₂		6.93		2.78	} 3.04		1.65
CeO ₂26		.22	.19
(LaDi) ₂ O ₃50			2.05
(YEr) ₂ O ₃20			6.16
PbO	4.35	4.34	4.35	4.20	4.53	.70	3.58
ZnO44	
FeO				trace		.32	
Fe ₂ O ₃11	.28	.27				trace
MnO10	.07	(?)			.16	
CaO18	.22	.30	.85	.23	.84	.41
MgO			} .15	} .30	} .25	trace?	trace
Alkalies						trace?	trace
H ₂ O68	.67	.68	1.21	undet.	1.96	undet.
He, etc	undet.	undet.	.38	.05	undet.	.02	undet.
SiO ₂03	.13	.20	.08	.13	2.79	
P ₂ O ₅				(?)		.22	
As ₂ O ₅43	
CuFeS ₂12	
FeS ₂24	
Insoluble04	.14	1.40	.10	.06		.20
	98.21	99.37	99.24	100.67	98.91	99.82	98.39
Specific gravity	9.733	9.560	9.348	8.086	9.492	8.068	

M. Nivenite from Llano County, Tex.

N. Somewhat altered uraninite from Villeneuve, Province of Quebec, Canada.

O. Massive uraninite from Johanngeorgenstadt, Saxony.

	M.	N.	O.
UO ₃	20.89	34.67	22.33
UO ₂	44.17	41.06	59.30
ThO ₂	6.69	6.41	} none
ZrO ₂34	(?)	
CeO ₂34	.40	
(La, Di) ₂ O ₃	2.36	1.11	
(Y, Er) ₂ O ₃	9.46	2.57	
CaO32	.39	1.00
PbO	10.08	11.27	6.39
H ₂ O	1.48	1.47	3.17
He, etc08	.10	trace
SiO ₂46	.19	.50
Fe ₂ O ₃14	.10	.21
Al ₂ O ₃ ?20
MgO17
Na ₂ O31
Bi ₂ O ₃09	.75
CuO17
MnO09
P ₂ O ₅06
As ₂ O ₅			2.34
V ₂ O ₅ , MoO ₃ , WO ₃75
SO ₃19
Insoluble.....	1.47	.13	-----
	98.28	99.96	97.93
Specific gravity.....	8.29	-----	6.89

The following uraninites are all from Norway, and were analyzed for comparison with the American material.

P. Bröggerite from Anneröd, near Moss. Specific gravity, 8.893.

Q. From Elvestad. Specific gravity, 9.145.

R. From Elvestad. Specific gravity, 8.320.

S. From Skraatorp. Specific gravity, 8.966.

T. From Huggenäskilen. Specific gravity, 8.930.

U. Cleveite from Arendal. Specific gravity, 7.500.

V. From Arendal.

	P.	Q.	R.	S.	T.	U.	V.
UO ₃	30.63	25.36	22.04	32.00	35.54	41.71	26.80
UO ₂	46.13	50.74	43.03	43.88	43.38	24.18	44.18
ZrO ₂ ²06	.08					
ThO ₂	6.00	8.48		8.98	6.63	3.66	4.15
CeO ₂18	.21	8.43	.17	.20		none
(La, Di) ₂ O ₃27	.26		.36	.23		.67
(Y, Er) ₂ O ₃	1.11	1.10		.97	1.03		9.05
PbO.....	9.04	10.06	8.58	9.46	9.44	10.54	10.95
CaO.....	.37	.77	.37	.36	.41	1.06	.61
MgO.....	traces	traces	.13	traces	.13	.10	.04
Alkalies.....						.23	.15
H ₂ O.....	.74	.73	.74	.77	.79	1.23	undet.
He, etc.....	.17	.18	.15	.15	.15	undet.	.18
Fe ₂ O ₃25	.21	.30	.09	.32	.03	.24
MnO.....		.06					
SiO ₂22	.38	.29	.53	.49	.90	.50
P ₂ O ₅02	.04	trace	(?)	trace		trace
Insoluble.....	4.42	.45	15.45	1.54	.42	1.10	1.19
	99.61	99.11	99.51	99.26	99.16	94.50	98.71

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PUBLICATIONS OF UNITED STATES GEOLOGICAL SURVEY.

[Bulletin No. 220.]

The serial publications of the United States Geological Survey consist of (1) Annual Reports, (2) Monographs, (3) Professional Papers, (4) Bulletins, (5) Mineral Resources, (6) Water-Supply and Irrigation Papers, (7) Topographic Atlas of the United States—folios and separate sheets thereof, (8) Geologic Atlas of the United States—folios thereof. The classes numbered 2, 7, and 8 are sold at cost of publication; the others are distributed free. A circular giving complete lists may be had on application.

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- PP 18. Chemical composition of igneous rocks expressed by means of diagrams, with reference to rock classification on a quantitative chemico-mineralogical basis, by J. P. Iddings. 1903. 98 pp., 8 pls.
- B 220. Mineral analyses from the laboratories of the United States Geological Survey, 1880 to 1903, tabulated by F. W. Clarke, chief chemist. 1903. 119 pp.

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Correspondence should be addressed to—

THE DIRECTOR,

UNITED STATES GEOLOGICAL SURVEY,

WASHINGTON, D. C.

NOVEMBER, 1903.

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