

Microscopic Determination of the Nonopaque Minerals

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JOHN J. MATZKO

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A revision of Bulletin 848 by Larsen and Berman (1934),
including new optical data and new methods of determination

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PREFACE TO THE THIRD EDITION

The first edition of "The Microscopic Determination of the Nonopaque Minerals" by Esper S. Larsen, Jr., was published in 1921 as U.S. Geological Survey Bulletin 679 after more than 10 years of preparation. Preparation involved not only compilation of data from the literature, but also new determinations of optical properties on about 500 samples. These new descriptions accounted for 126 of the 294 pages of the 1921 edition.

The second edition, by Larsen and Harry Berman, both then at Harvard University, published in 1934 as U.S. Geological Survey Bulletin 848, increased the data in Bulletin 679 by more than 50 percent. The second edition was republished in Russia in 1965 (Petrov, 1965) with insertions of newer data from the literature.

Assembly of new data on optical properties of minerals was begun by Fleischer and Matzko in 1950, and compilation in the present tabular format was begun in 1965. Insofar as possible, data from the literature up to August 1, 1983 have been included in the tables. The introductory text on methods was written by Wilcox, who also critically reviewed and suggested minor modifications of the tables as prepared by Fleischer.

The present edition, in holding to the objective of mineral identifications, omits discussion of statistics of mineral optics, as covered in Chapter 3 of the 1934 edition. Modern treatments of the Gladstone-Dale relation and other aspects of the relations of refractive index to density and chemical composition can be found in Allen (1956), Jaffe (1956), Batsanov (1961), Anderson (1975), Pabst (1975), Mandarino (1976, 1978, 1979, 1981) and Bloss and others (1983). The Commission on New Minerals and Mineral Names, International Mineralogical Association (*American Mineralogist*, v. 67, p. 191-192), states for new minerals, "It is recommended that the relationship between chemical composition, density, and refractive indices, be checked by the Gladstone-Dale rule."

CONTENTS

	Page
Preface to the third edition -----	i
Chapter I. Introduction -----	1
Scope of the tables -----	1
Need for additional data -----	2
Acknowledgments -----	2
Chapter II. Determination of optical properties of transparent minerals by the immersion method -----	2
Refractive index -----	3
Immersion liquids -----	3
Central illumination ("Becke line") technique -----	4
Oblique illumination (Schroeder van der Kolk) technique -----	5
Dispersion coloration technique using central focal masking -----	5
Techniques of higher precision -----	7
Determination of refractive index of optically isotropic substances -----	7
Determination of refractive indices of anisotropic substances -----	9
Spindle stage -----	11
Birefringence -----	14
Optic angle and optic sign -----	15
Calculation of optic angle -----	15
Measurement of optic angle -----	18
Estimation of optic angle and determination of sign -----	18
Dispersion of the optic axes -----	19
Relation of optical to crystallographic directions -----	19
Extinction angle -----	20
Sign of elongation -----	20
Color and pleochroism -----	20
Chapter III. Identification tables -----	21
List of abbreviations -----	21
Arrangement of data -----	22
Use of the tables for identification -----	24
Examples -----	24
Chapter IV. Mineral groups -----	316
References cited -----	402
Index -----	408

FIGURES

Page

1.	Graph showing dispersion of the refractive index of a typical inorganic solid compared to dispersions of successive members of a typical set of organic immersion liquids -----	6
2-3.	Diagrams showing:	
2.	Formation of violet image of a fragment whose refractive index matches the liquid for the yellow band of wave-lengths of light -----	7
3.	Indicatrices of anisotropic crystals -----	10
4-6.	Sketches showing:	
4.	Spindle stage -----	12
5.	Technique for mounting fragment on spindle -----	13
6.	Withdrawal of liquid from immersion cell -----	13
7.	Diagrams showing examples of typical isogyre configurations -----	14
8.	Nomogram for determination of optic angle from ratio of partial birefringences -----	17
9.	Diagram showing curvatures of superimposed isogyres for centered optic axis figures -----	18

TABLES

Page

1.	Dispersion colors observed at edges of fragments on dark field with central focal masking technique -----	8
2.	Range of calculated optic angles, taking into account a possible ± 0.001 error of refractive index determinations -----	16
Tables 3-7	list properties for identification of the following minerals:	
3.	Isotropic minerals and mineraloids -----	30
4.	Uniaxial positive minerals -----	62
5.	Uniaxial negative minerals -----	84
6.	Biaxial positive minerals -----	124
7.	Biaxial negative minerals -----	208
Tables 8-38	list properties for identification of the following mineral groups:	
8.	Alunite group -----	316
9.	Amphibole group -----	318
10.	Apatite group -----	330
11.	Aragonite group -----	334
12.	Autunite group -----	335
13.	Barite group -----	337
14.	Beudantite group -----	337
15.	Calcite group -----	339
16.	Cancrinite group -----	341
17.	Chalcanthite group -----	342
18.	Copiapite group -----	343

TABLES—CONTINUED

19.	Crandallite group -----	344
20.	Dolomite group -----	346
21.	Epidote group -----	347
22.	Feldspar group -----	350
23.	Garnet group -----	355
24.	Humite group -----	358
25.	Melilite group -----	362
26.	Meta-autunite group -----	363
27.	Mica group -----	365
28.	Monticellite group -----	370
29.	Olivine group -----	371
30.	Osumilite group -----	373
31.	Pyrochlore group -----	375
32.	Pyroxene group -----	377
33.	Scapolite group -----	385
34.	Sodalite group -----	386
35.	Spinel group -----	387
36.	Stibiconite group -----	389
37.	Tourmaline group -----	391
38.	Zeolite group -----	394

Microscopic Determination of the Nonopaque Minerals

By Michael Fleischer, Ray E. Wilcox, and John J. Matzko

CHAPTER I. INTRODUCTION

Identification of an unknown mineral in former times was a haphazard operation, involving determination of a few optical, physical, and chemical properties for comparison with a woefully incomplete set of properties of known minerals. Today, not only can the diagnostic optical properties be determined more reliably and the chemical composition more rapidly, but several X-ray techniques are available for determination of diagnostic crystal structure.

Compilation of X-ray data in a form useful for mineral identification has proceeded apace and has far outstripped the compilation of new optical data, with the unfortunate result that the use of the polarizing microscope in mineralogic and petrologic investigations has decreased markedly; in addition, the time allotted in the curriculum of many colleges for training in optical crystallography has been reduced. The polarizing microscope nevertheless remains a powerful and sometimes indispensable tool in mineral identification, and to help restore its usefulness the authors have taken on the task of bringing up to date the tables compiled by Larsen and Berman (1934).

The tables, which appear in chapters 3 and 4 herein, are a compilation of properties of known minerals. They have been updated by adding optical data that have appeared in the literature since the 1934 edition and up until August 1, 1983. The range of optical properties of previously listed minerals and isomorphous groups has been corrected or extended, and the names and data for newly discovered minerals have been inserted. Methods for the measurement of optical properties are outlined in chapter 2.

Identification of an unknown by its optical properties proceeds most effectively in three stages: (1) measurement under the microscope of selected key

properties; (2) elimination from further consideration minerals not having these properties; and (3) determination of selected additional diagnostic properties to eliminate all but one or a very few candidates. For confirmation of that one mineral or a final choice between remaining candidates, one may then turn to X-ray or chemical analysis.

It is important that this process of elimination using the tables be given the same care and attention as the measurements of the basic optical properties, for to overlook a possible candidate at this stage can jeopardize the correctness of the eventual identification. For the final stage, one should consult published encyclopedic descriptive mineralogies (for example, Deer and others, 1962, 1963, 1978; Tröger, 1967, 1979; Winchell and Winchell, 1951) then return to the microscope to determine the needed additional diagnostic characteristics.

Scope of the Tables

For each known mineral, tables 3 to 7 list optical properties readily determinable by the polarizing microscope and its accessories, and they also note certain nonmicroscopic but useful identifying properties such as bulk color, fusibility, hardness, and chemical behavior. The minerals are grouped in separate tables according to basic optical character (isotropic, uniaxial, or biaxial) and according to optic sign (positive or negative). For convenience in entering the tables by means of optical data routinely obtained by the so-called immersion method (see chap. 2), the minerals in each table are arranged in order of increasing refractive index. The tables may also be entered with refractive indices estimated from thin sections, although a much greater number of listed minerals will usually have to be considered as possible candidates due to the uncertainty of the estimate.

Need for Additional Data

Although the tables in chapters 3 and 4 have been updated, they contain many imperfections and cannot be regarded either as complete or entirely correct. The reliability of the individual listings varies widely and for some minerals cannot be estimated, inasmuch as the source literature does not supply adequate descriptions of methods used. Indeed, in too many cases it is painfully evident that no set of minimum standards was adhered to in carrying out the original optical determination, much less in the preparation and purification of the minerals for chemical analysis. Still remaining to be filled are gaps in the knowledge of the ranges of optical properties of rare minerals, many of which, judging from their determined chemical compositions, are members of isomorphous series. Finally, there is the certainty that some naturally occurring minerals still wait to be discovered and described optically.

Optical methods of mineral identification are useful not only as an adjunct to the powerful methods of X-ray and electron microprobe but in many situations also as the most reliable and rapid means of identification. However, the gaps and flaws mentioned above in available optical data of minerals are a matter of concern particularly because identification rests on the process of elimination. One therefore must keep in mind the possibility, remote as it might be, that the unknown mineral under examination is not listed in the part of the table involved. The use of optical methods in the modern laboratory will not reach full effectiveness until these shortcomings are corrected. Fortunately, basic optical data can now be obtained with greater reliability and with considerably less tedium than formerly, thanks to the availability of more documented mineral specimens and more advanced development of instrumentation and methods. Also, many of the difficulties due to limited sample size that plagued the early workers in descriptive mineralogy are now avoidable because optical, X-ray, and chemical analyses can

be carried out on extremely small samples, using the spindle stage, the single-crystal X-ray goniometer, and the electron microprobe.

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CHAPTER II. DETERMINATION OF OPTICAL PROPERTIES OF TRANSPARENT MINERALS BY THE IMMERSION METHOD

Whereas the textures and interrelations of minerals in rocks, so important in petrologic interpretations, are well seen in thin section, measurements of principal refractive indices and other optical properties necessary for reliable identification are better accomplished on individual fragments by the so-called immersion method. Advantages of this method include the ability to manipulate crystal fragments individually without the handicap of interference from adjacent crystals.

For the immersion method, the sample is crushed to separate the indi-

vidual crystal units and expose fresh surfaces, so that fragments can be chosen and their refractive indices can be compared under the microscope with successive liquids of known refractive indices until a match is found. In the same mounts, birefringence, optic angle, optic sign, pleochroism, and the relationships of optical directions to crystallographic directions (extinction angle and sign of elongation, and so on) may be measured or estimated.

It is possible here to give only a brief review of established procedures and techniques; emphasis will be on several practical techniques not yet fully covered in most textbooks. For details of the conventional techniques, the reader is referred to textbooks such as Bloss (1961), Wahlstrom (1969), Hartshorne and Stuart (1969, 1970), Burri (1950), Tatarskii (1965), Stoiber and Morse (1972), Hutchison (1974, especially p. 78-93, 247-263), Schumann and Kornder (1973), McKie and McKie (1974), and el-Hinnawi (1966). Several useful works deal with problems encountered in special fields: industrial chemistry and product control (Hartshorne and Stuart, 1970; Chamot and Mason, 1958; McCrone and Delly, 1973; Lehr and others, 1967), ceramics (Insley and Frechette, 1955; Dietzel and Köppen, 1974), identification and characterization of synthetic inorganic substances (Winchell and Winchell, 1964; Kordes, 1960; Lehr and others, 1967), and organic compounds (Winchell, 1954).

Refractive Index

Three techniques are especially useful for determination of refractive index, the primary basis for entry into the tables: (1) central illumination ("Becke line"), (2) inclined illumination ("Schroeder van der Kolk"), and (3) dispersion coloration ("dispersion staining"). All three techniques depend on the same basic optic phenomenon: the smaller the difference of index between fragment and liquid, the less strongly diverted the light rays passing through an inclined interface. The techniques

differ only in the manner of illumination and in the resulting observed effects. Because the immersion liquids are so important for the correctness of the determination, their calibration and use will be dealt with first.

Immersion Liquids

Sets of liquids ranging in index from about 1.46 to 1.80 and separated by regular index intervals, say 0.005 or 0.002, are available commercially. In choosing the interval between liquids, it may be kept in mind that the greater initial cost of a set having the smaller interval may be more than compensated by the saving of time and the greater precision of results. The label on each bottle in the set should show not only the index of refraction for the D-line of the spectrum, but also its coefficients of temperature and dispersion. With some care and effort, sets of liquids can be made up in the laboratory from end members, as described by Harrington and Buerger (1931), Butler (1933), and Meyrowitz (1955, 1956). Useful general discussions of possible end-member liquids are given by Johannsen (1918, p. 259), Burri (1950, p. 221), Wilcox (1964), and Hartshorne and Stuart (1970, p. 265).

The refractive index of a typical organic immersion liquid decreases significantly with increasing temperature--on the order of 0.0004 to 0.0005 per degree Celsius; the corresponding change of index of the immersed crystal fragment is an order of magnitude smaller, thus negligible in the present application. A bottle label reading $n_D^{25} = 1.534$ indicates that the refractive index of its contained liquid is 1.534 at 25°C in light of the D-line (wavelength 589.3 nm). A further notation,

$$-\frac{dn}{dt} = 0.00039/^{\circ}\text{C},$$

indicates the magnitude of the inverse change of index with temperature. Thus, after reaching the match between fragment and liquid, it is necessary to make

an index correction for the difference between the temperature of the immersion mount itself and the temperature at which it was calibrated. If, for instance, the above liquid was found to match the index of a fragment at 28°C, the correction to be made would be 3×0.00039 , or 0.00117 , to be subtracted from 1.534 to give 1.53283 , rounded for the present application to 1.533 . This is the refractive index of the matching liquid under these conditions and thus of the index of the fragment.

The refractive index of each liquid in the set must be rechecked from time to time, depending on the stability of the liquid and on the accuracy required in use. Adequate for most mineralogical laboratories is the Abbe-type refractometer for indices up to about 1.70 with a precision about ± 0.0002 . Although a prism of special glass may be substituted to permit measurements to higher than 1.80 , the glass commonly used for this purpose is a high-lead glass that is vulnerable to corrosion from most liquids used in this higher range. Less expensive refractometers, such as that designed by E.E. Jelley (1934; see also Edwards and Otto, 1938), extend to 1.90 , are not damaged by liquids above 1.70 , and are precise enough (about ± 0.001) for much work. Whatever the refractometer, it too should be calibrated carefully according to the directions of the manufacturer (see also Fisher, 1958).

In some applications it is necessary to know the dispersion strength of the immersion liquids. This property may be expressed as the difference between refractive indices at two standard wavelengths, such as the F- and C-lines. A common expression of the dispersive power of the liquid is the coefficient:

$$v = \frac{n_D - 1}{n_F - n_C}$$

Dispersion can be determined by measurement of refractive indices of the liquid at the wavelengths of the F-, D-, and C-lines by minimum deviation on a goniometer using a prism-cell, or on an Abbe

refractometer making corrections for the dispersion of the glass of its reference prism. Considerably more convenient, however, is use of an Abbe refractometer equipped with Amici prisms, whereby a measure of the dispersion is obtained in the course of determination of n_D in white light.

Central Illumination ("Becke Line") Technique

The so-called Becke line test of the refractive index difference between fragment and immersion liquid is carried out in illumination consisting of a narrow cone of light symmetrically about the axis of the microscope. The fragment in liquid is brought into sharp focus. (If anisotropic, the fragment should be placed in a position of extinction, then the upper polar is withdrawn.) As the focus is raised slightly, that is, as distance between objective lens and object is increased, a bright-line (the Becke line) forms at the edge of the fragment and moves toward the substance of higher index. Thus, if refractive index of the fragment is higher, the Becke line moves inward; if lower, it moves outward. As the focus is raised further, the line fades and the image blurs. The basis for this fading and blurring has been variously ascribed to the crude lens effect of the solid fragment and to total reflection of rays by the very steep interfaces at the edge of the fragment.

Determination of the refractive index of the fragment becomes a matter of finding the matching liquid in a set of graduated index liquids. As this matching liquid is approached in successive liquids, the Becke line noticeably weakens but, up to a point, may be made more visible by stopping down the substage aperture iris diaphragm. In a liquid near the match and with white light for illumination, the edges of the fragment take on faint colors. Attention to the quality of these faint dispersion colors enables one to recognize a close approximation of the true match (Emmons and Gates, 1948). A more exact

match may be established using monochromatic light, such as that of the standard sodium line (589.3-nm wavelength); when the index of the solid and liquid are the same in monochromatic light, the Becke line disappears completely, and the fragment becomes invisible except for flaws and inclusions.

If the host fragment is crowded with inclusions, precise results are not attainable by the Becke line method, for as the matching liquid is approached in successive mounts, the still strong Becke lines of the crowded inclusions coalesce to form a single bright line that prevents recognition of the fading and disappearance of the Becke line of the host. A further difficulty, possibly related, arises from the so-called "spurious Becke line," a light line that moves opposite to the true Becke line. This situation is often encountered with fragments having appreciable light absorption (body color) or unfavorable fragment contour. Here at best one finds only a broad minimum in the intensity of the light lines through a series of adjacent immersion liquids, leaving in question the refractive index of the true match.

Oblique Illumination (Schroeder van der Kolk) Technique

The oblique illumination technique, details of which are given in most textbooks and in the excellent discussion by Wright (1911, p. 92-95), is capable of greater sensitivity than the Becke line method. Briefly, to obtain oblique illumination, the entering cone of light is blocked on one side at the proper level in the substage or in the microscope tube to produce a sidelighting effect that enhances the appearance of relief (mismatch) of the fragment in respect to the immersion liquid: one edge of the fragment is highlighted while the opposite edge is shaded. As a general rule, the index of the fragment is higher than the liquid if its brighter side is toward the more brightly illuminated portion of the field of view. It is advisable, however, to

check the behavior of a particular microscope and illuminating system in this respect by use of a fragment of known index in a liquid of known direction of mismatch.

In white light, as the indices of liquid and fragment become more nearly equal in successive mounts, the apparent relief of the fragment decreases. At match, one edge of the fragment is bluish and the opposite edge reddish. One learns to recognize the diagnostic shades and intensities of colors that are obtained on either edge when a match has been reached, keeping in mind that the quality of the colors is affected to some extent by the natural body color of the fragment as well as the spectrum of the light source. The match point is more precisely defined with monochromatic illumination, for at match the fragment completely disappears except for flaws and inclusions. In order to cancel disturbing effects due to differences in shape of opposite edges of the fragment, it is helpful to observe the fragment at the one position of the microscope stage, then rotate the fragment 180° on the stage and observe it again to confirm that the relief effect is in the same relation to the lighter side of the field.

Dispersion Coloration Technique Using Central Focal Masking

This method is rapid and in large part free of misleading effects such as the "spurious Becke line" encountered in the central illumination method. Although the precision of this method is somewhat less than that of oblique monochromatic illumination, it is quite adequate for purposes of crystal identification. In addition to its value for determination of refractive indices, the method enables rapid distinction between different mineral species in a fragmental mixture and determination of the kind and amount of undesired constituents in a product at successive stages

in a purification process. Details of the method and its variants may be found in descriptions by Cherkasov (1960), Wilcox (1962, 1983), Brown and McCrone (1963), Brown and others (1963), and Hartshorne and Stuart (1970, p. 437). Similar color effects are obtained in the "Grenzdarkfeld" technique described by Schmidt and Heidemanns (1958) and in the early "dispersion staining" technique of Crossmon (1948). In the two latter techniques, however, some light rays traverse the fragment at an appreciable angle to the axis of the microscope, leading to possible inaccuracies when working with strongly anisotropic substances.

The immersion mount is illuminated by a narrow beam of ordinary white light, produced by severely restricting the substage aperture diaphragm. With a small opaque dot mask installed in the objective near the focal point, the image of the fragment appears on the dark field in a color that indicates the degree of match or mismatch with the immersion liquid. These useful color effects result from the appreciably greater dispersion of the liquid over that of the solid of similar refractive index. Thus in figure 1, the refractive index of the glass matches that of the liquid for light of wavelength near that of the D-line (5893 Å or 589.3 nm) but is progressively less than that of the liquid at shorter wavelengths and greater at longer wavelengths.

The critical color effects begin to appear as the index of the immersion liquid approaches to within about 0.01 of that of the fragment, depending on the relative dispersive strengths of fragment and liquid. In liquids of index outside this range in either direction, the fragment edges are imaged brightly in the color of the source light. To determine whether to go to a liquid of higher or lower index, one simply observes the movement of the single strong Becke line as the focus is raised.

Under the microscope a fragment immersed in a matching liquid and illu-

minated by a thin beam of white light refracts the various wavelengths differently (fig. 2). Light beams of a wave-

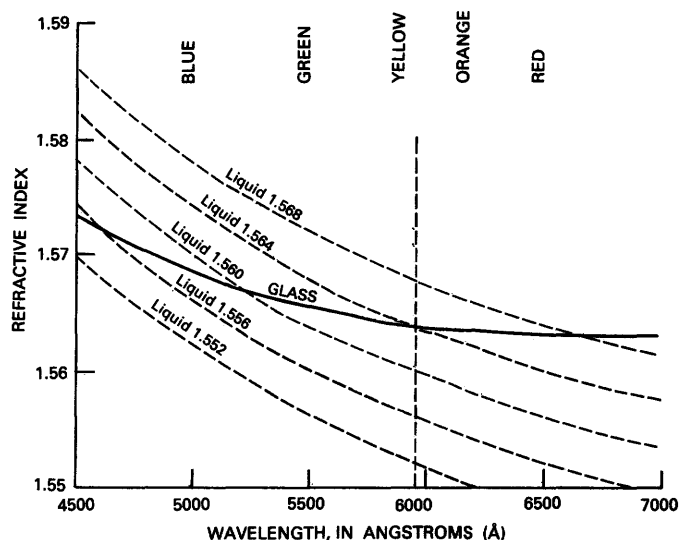


Figure 1. Dispersion of the refractive index of a typical inorganic solid compared to dispersions of successive members of a typical set of organic immersion liquids.

length close to the standard yellow wavelength, 589.3 nm, are blocked by the opaque dot at the focal point of the objective, while beams of shorter wavelengths (more blue) and of longer wavelengths (more red) bypass the opaque dot and combine again to form an image of the fragment edges in deep violet color. This violet image is the combined effect of the spectral colors remaining after the band near the yellows has been subtracted. The field of view is otherwise dark, because beams not passing through inclined surfaces of the fragment edge are also blocked by the opaque dot. By raising the focus slightly, a red outline of the image moves inward and a blue outline moves outward. (These, of course, are the Becke lines for light of the respective wavelengths.)

From figure 1 it can be further deduced that, in a liquid of slightly lower index, say by 0.002, the edges of the fragment would be seen as noticeably richer in reds; in this case a band in the blue-green would have been subtracted. Similarly the edges of the fragment in a liquid higher by 0.002 would be seen as richer in blue.

Table 1 gives word descriptors of the colors seen in different situations. Each person should learn the color of the match as seen by eye, however. To this end, one may make a mount of a solid in a liquid as closely matched as can be judged by inclined illumination (described above) using monochromatic sodium-D illumination, then change to central focal masking illumination to observe and fix in mind the quality of color seen at the match.

Techniques of Higher Precision

Very precise determination of refractive indices beyond the third decimal place is seldom required for identification of unknown minerals. Should greater precision be desired, for instance in the characterization of newly discovered minerals or in applications to crystal structures, a number of techniques are available, each of which requires special auxiliary equipment. Perhaps the most rapid of these is the wavelength-variation method introduced by Wherry (1918), Posnjak and Merwin (1922), and extended by Emmons (1929, 1943). A precision of ± 0.0001 refractive index is estimated for this method, as refined by Morse (1968) using a convenient substage variable interference filter and immersion liquids for which dispersion strengths have previously been determined. A further extension of the method enables determination of members of an isomorphous series, such as the plagioclase series, for which dispersion characteristics are also known.

Sueno (1933) and Micheelson (1957) use wavelength-variation techniques, adding glass powders of precisely known index to the immersion mount of the unknown substance to act in the role of refractometer, thus eliminating the necessity of temperature correction. In the double-variation method of Emmons (1929, 1943), both wavelength and temperature are changed at will to match solid and liquid at several wavelengths, enabling one to determine not only the refractive index of the solid at the D-

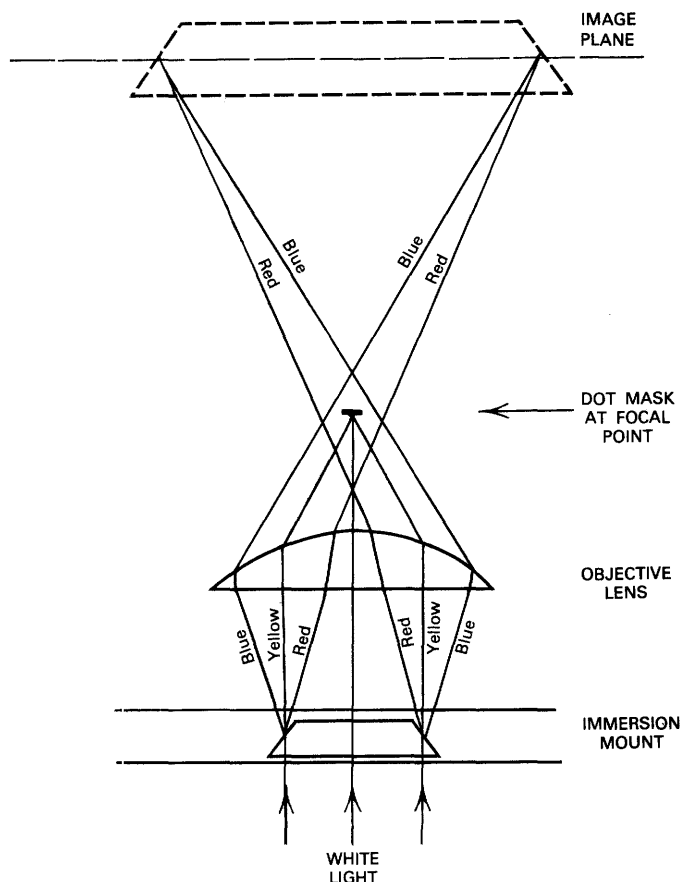


Figure 2. Formation of violet image (spectrum minus yellow band) of a fragment whose refractive index matches the liquid for the yellow band of wavelengths of light.

line, but also the dispersion curve. For greater sensitivity in determination of the index match between liquid and immersed solid, Saylor (1935) developed a double-diaphragm mask for the microscope objective and substage. Used with an ultrasensitive stage refractometer (Saylor, 1966), this mask enables precision to the fifth decimal place in determining the refractive index. Chao (1976) attains precision to the fourth place by interference microscopy.

Determination of Refractive Index of Optically Isotropic Substances

Light in an optically isotropic substance can vibrate in all directions in a plane perpendicular to its direction of transmission, and the refractive

Table 1. Dispersion colors observed at edges of fragments on dark field with central focal masking technique
[Modified from Cherkasov, 1960]

Matching wavelength (nm)	Observed color			Refractive index n_D of solid is
	At focus	Raising focus		
		Moving in	Moving out	
660	bright greenish blue	(none)	bright greenish blue	less than liquid
625	sky blue	(none)	sky blue	
600	blue	faint dark red	greenish blue	
<u>589</u>	DEEP VIOLET	WEAK RED	STRONG BLUE	SAME AS LIQUID
575	purple	red	blue	greater than liquid
540	reddish purple	orange-red	bluish-violet	
505	orange-red	orange	weak blue violet	
480	orange	yellow	weak violet	
465	bright gold	bright gold	(none)	

index is the same regardless of the directions of passage and vibration. Hence the refractive index of such a substance may be determined in a straightforward manner by making a series of immersion mounts of randomly oriented crushed fragments to find the matching liquid. Having found the liquid that matches the fragments, an appropriate correction for the temperature may be applied, as outlined in the section entitled "Immersion Liquids."

Determination of Refractive Indices of Anisotropic Substances

Light traveling in an anisotropic substance is polarized, that is, its vibrations are confined to two directions that are perpendicular both to each other and to the direction of transmission. Refractive indices depend on vibration directions and may vary between limits set by the extreme values of the principal refractive indices, determinations of which are necessarily more involved than for an isotropic substance because the fragment first must be properly oriented.

To aid in visualizing the behavior of light in anisotropic crystals, one may make use of the optical indicatrix, an ellipsoid in which the length of a given radius represents the refractive index of the light wave vibrating in the direction of the radius. Thus, the indicatrix of an isotropic substance is a sphere. Anisotropic crystals may be divided into two classes: uniaxial and biaxial. The indicatrix of a uniaxial crystal is an ellipsoid of revolution (fig. 3A), having its axis (called the optic axis) coincident with the crystallographic symmetry axis c , perpendicular to which is the circular section. Light traveling in the uniaxial crystal at an angle to the optic axis is polarized. One wave (called the ordinary or O wave) vibrates in the plane of the circular section and has a refractive index ω , which is represented by the radius of the circular section. The other wave (called the extraordinary or E wave) vibrates at right angles to that of ω

(fig. 3A) and has a refractive index ϵ' which is represented by the radius of the ellipsoid in vibration. The index ω is called the ordinary principal refractive index, and ϵ' is called the extraordinary refractive index. The extreme value of ϵ' , obtained when light travels perpendicular to the optic axis, is designated as ϵ , the extraordinary principal refractive index.¹ Note that, in this case, the other polarized wave again has ω as its refractive index. Light traveling parallel to the optic axis is not polarized and has the ordinary principal refractive index ω . Crystals in which ϵ is greater than ω are said to be optically positive (the indicatrix is a prolate ellipsoid, such as that of fig. 3A). Where ϵ is the lesser index (an oblate ellipsoid), the crystal is said to be optically negative.

The optical indicatrix of a biaxial crystal (fig. 3B) is an ellipsoid in which the three unequal symmetry axes represent the principal vibration directions X, Y, and Z of light for the three principal refractive indices: α , the lowest; β , intermediate; and γ , the highest.² Such an ellipsoid has two circular sections, the normals to which here are called the optic axes. These axes define the optic plane, and the angle between them is the optic angle $2V$. The optic plane includes the principal vibration directions X and Z as bisectrices of the optic angle. The crystal is said to be optically positive when $2V$ is acute about Z and optically negative when $2V$ is acute about X.

¹Other symbols for refractive indices used in some publications are,
for ω : n_ω , n_0 , n^0 , or merely 0,

for ϵ : n_ϵ , n_E , n^E , or E.

²Other symbols used in some publications are,

for α : n_α , n_X , n_p , or merely X,

for β : n_β , n_Y , n_m , or Y,

for γ : n_γ , n_Z , n_g , or Z.

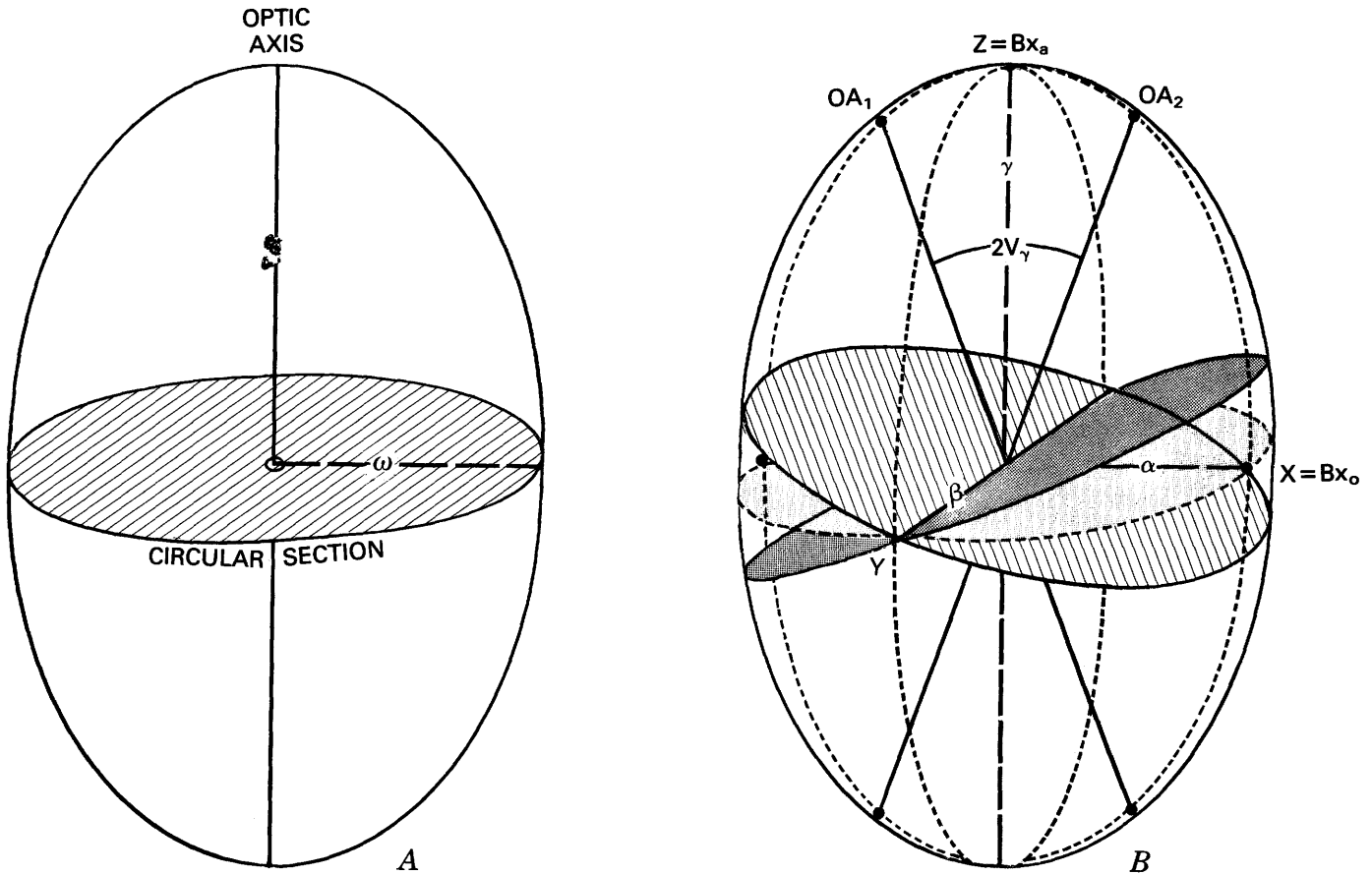


Figure 3. Indicatrices of anisotropic crystals. *A*, Uniaxial positive (a prolate ellipsoid of revolution, $\omega < \epsilon$). *B*, Biaxial positive ($2V_z < 90^\circ$).

Light traveling parallel to an optic axis vibrates in the circular section, is not polarized, and has β as its refractive index. (Although optic axes are customarily shown in the indicatrix, as in figure 3, it should be kept in mind that they really represent directions of travel and that the corresponding fundamental vibrational elements are the circular sections.) Light traveling at an angle to the optic axes is polarized into two waves vibrating at right angles to each other. One wave vibrates in the dihedral angle of the circular sections that includes the bisectrix X , and in the general case it has refractive index α' , a nonprincipal index between α and β . The other wave vibrates in the other dihedral angle that includes the bisectrix Z and has refractive index γ' between γ and β . Light traveling parallel to a symmetry plane of the indicatrix, XY , XZ , or YZ , will

have one wave with a principal refractive index, γ , β , or α , and generally all the other wave with a nonprincipal index. Light traveling parallel to a symmetry axis X , Y , or Z will have both waves with principal refractive indices.

As can be seen from the foregoing discussion, to determine a principal refractive index of an anisotropic crystal by the immersion method, the crystal fragment must be oriented so that the plane-polarized light from the polarizer of the microscope vibrates parallel to the corresponding principal vibration direction in the crystal. In practice, one has a choice of (1) the "method of random grain mounts," in which properly oriented fragments are sought in successive immersion mounts of multiple fragments, (2) the coated-slide technique, in which one seeks properly oriented fragments among the scattered fragments held on a thin film of adhe-

sive on a slide and then changes liquids as necessary to reach a match, or (3) a method wherein single fragments are mounted on a spindle for orientation so that each principal refractive index can be determined in its respective position.

The method of random grain mounts.
--This method, described in varying detail in textbooks, is carried out using loose-grain mounts in a chosen succession of immersion liquids. It needs no special auxiliary equipment but does require a thorough and logical procedure to ensure useful results in a reasonable time. The first step is to examine the behavior of the interference figures of a series of grains in a preliminary immersion mount to ascertain whether the unknown is uniaxial or biaxial. If uniaxial, every grain should show the ω -index at one or the other extinction position; one may, therefore, make a series of mounts in liquids leading to a match of ω . In another series of mounts, the other principal index, ϵ , is to be sought from grains showing highest interference colors, the correctness of the orientation being confirmed finally by the interference figure, which should be a symmetrically behaving flash figure.

For a biaxial substance, successive mounts may be made to match the index of grains that show very little or no interference color on the rotation between crossed nicols. This index is β , confirmed by a centered optic axis figure. Proceeding next through a series of mounts of progressively lower index and observing fragments showing relatively high interference colors, the extreme low index is sought; this will be the principal index, α , confirmed by an interference figure symmetrical about the crosshair that is perpendicular to the vibration direction of the polarizer. Helpful details in the use of such symmetrically behaving interference figures in this procedure are given by Slawson and Peck (1936) and by Stoiber and Morse (1972, p. 188-198, 211-214). (See also discussion in sec-

tion on "Spindle Stage," below). Similarly, the highest index, γ , is sought in grains showing high interference colors in mounts of index successively greater than β , again confirmed by the appropriately symmetrical interference figure.

If the unknown has marked cleavage oblique to the principal vibration directions, the chances are much reduced of finding fragments oriented suitably for determination of all principal refractive indices. Further, if the crushed sample contains two or more substances of overlapping optical properties, the unsophisticated or impatient investigator may settle on a set of totally misleading principal refractive indices and birefringence for the unknown "substance."

The coated slide technique.--In this technique, fragments of the unknown are scattered over a glass slide thinly coated with tacky material, such as gelatin (Olcott, 1960), thermal-setting epoxy (Langford, 1962), thick water glass (Lindberg, 1944), quick setting epoxy, or other colorless adhesive not affected by immersion liquids. Having once located a suitably oriented fragment by use of interference figures, the liquids can be changed to find the match for the respective principal refractive index without loss or disturbance of the fragment. In many cases, this is a considerable improvement over the method of random loose-grain mounts, in that new fragments need not be sought for each change of liquid. But this technique suffers also from the hazards and uncertainties arising when the unknown is composed of two or more substances of overlapping optical properties.

Spindle Stage

Most satisfactory for entry into the identification tables is a complete set of basic optical data from the unknown crystal fragment. For their determination, the spindle stage is well suited, enabling determination of all principal refractive indices as well as

optic angle and relation of optical symmetry axes to cleavage.³ A succession of fragments should of course be examined to establish the range of optical properties of the unknown mineral and whether the sample is composed of more than one mineral species. Procedures for the use of the spindle stage are given by Rosenfeld (1950), Wilcox (1959), Wilcox and Izett (1968), Hartshorne and Stuart (1970), Bloss and Light (1973), and Bloss (1981). The essentials will be outlined in the following.

To carry out the determination on the spindle stage, the crystal fragment is cemented on the tip of the spindle and placed on the base plate (fig. 4) attached to the stage of the microscope. With the fragment in the transparent immersion cell, orientation is accomplished by rotation about the horizontal (spindle) axis and vertical (microscope) axis. At the orientation for each principal refractive index, immersion liquids are changed successively to arrive at the match.

The adhesive for mounting the fragment must be insoluble in the immersion liquids, and ordinary water-soluble carpenter's glue works well for this if mixed with an appropriate amount of molasses, corn syrup, or similar substance to lengthen its period of tackiness. The mineral fragments, crushed to a convenient size (usually less than 0.2-mm diameter), are viewed under a stereomicroscope, and the desired fragment is isolated. The tip of the spindle is touched to a drop of the adhesive, then brought into the field of the stereomicroscope, holding the spindle as nearly vertical as practical and barely touching it to the fragment (fig. 5A).

³The universal stage (Emmons, 1943) also enables determination of these diagnostic properties to a high precision but requires an array of peripheral equipment as well as a much greater expenditure of time.

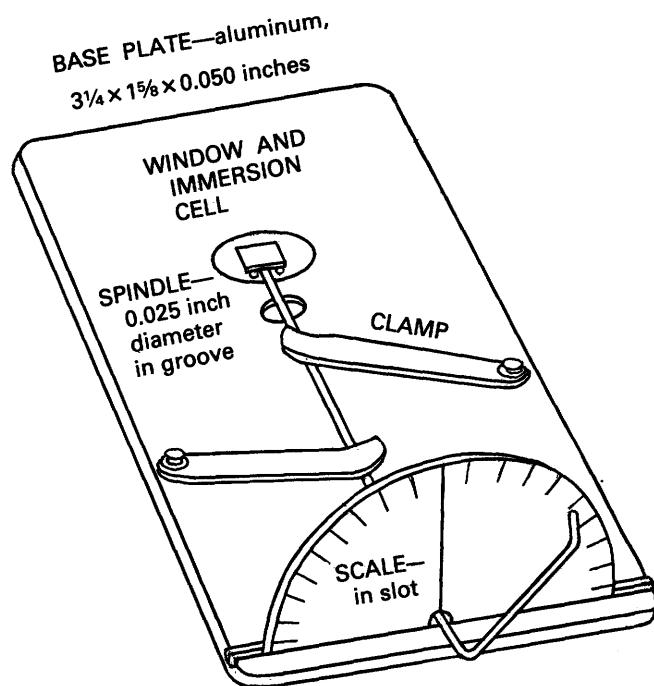


Figure 4. Spindal stage (modified from Wilcox, 1959).

With the fragment clinging to the spindle tip, the mount may be breathed on gently, whereupon the moisture-sensitive adhesive flows onto the fragment to hold it more securely. (Care must be taken, however, not to moisten it so much that the adhesive completely coats the fragment.) If the fragment is out of line, as in figure 5B, it may be gently nudged with a dissecting needle while viewed under the stereomicroscope to move it into the axis of rotation of the spindle (fig. 5C).

After allowing a minute or two for the adhesive to set, the spindle with fragment thus randomly mounted is clamped in the groove of the base plate, then pushed forward to insert the fragment into the immersion cell, and the small scale for the spindle arm is put in place. A trial immersion liquid is introduced at the front end of the cell, into which it moves by capillary action. Under low power, the image of the fragment is brought to the crosshair intersection by shifting the base plate. Liquids can be changed by holding a tab of blotting paper at the rear of the cell along the shank of the spindle (fig. 6). After the old liquid is

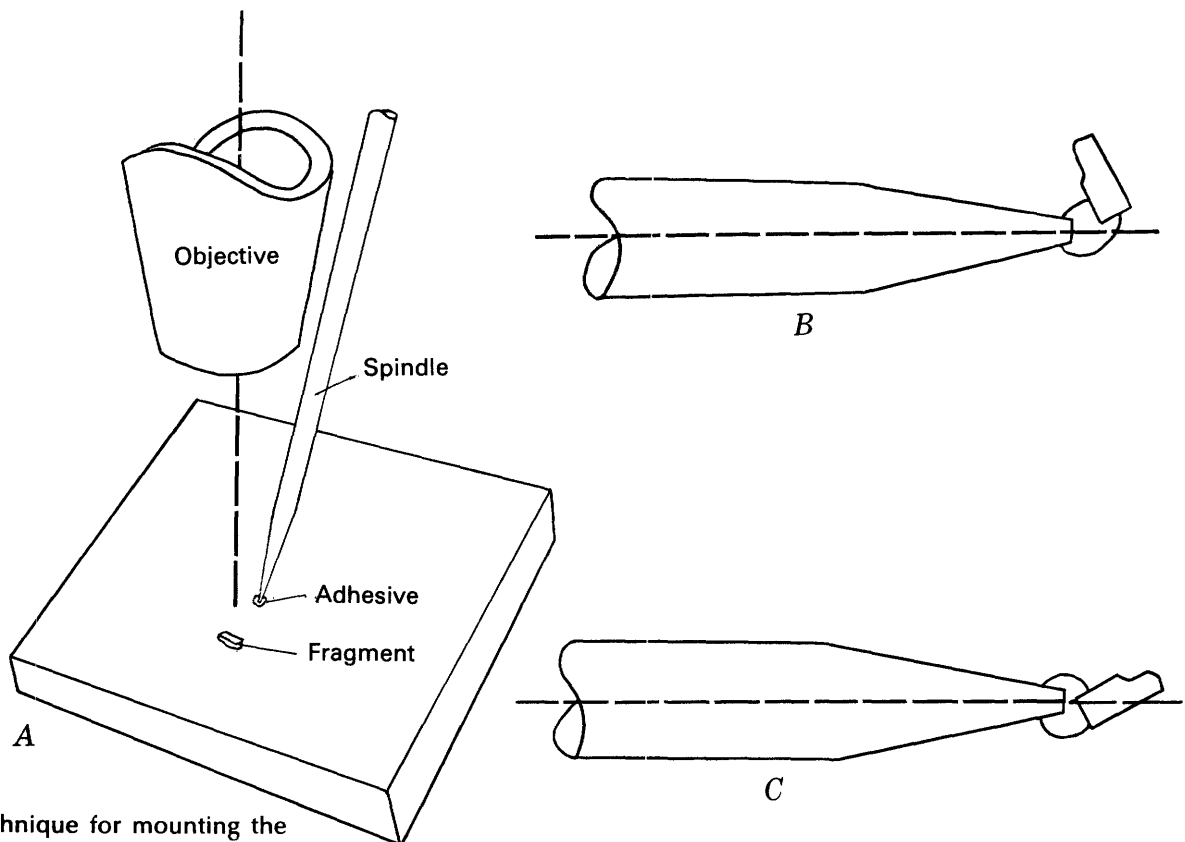


Figure 5. Technique for mounting the fragment on spindle: *A*, The spindle with adhesive is touched to the fragment. *B*, Fragment out of line. *C*, Fragment nudged into line with the spindle axis.

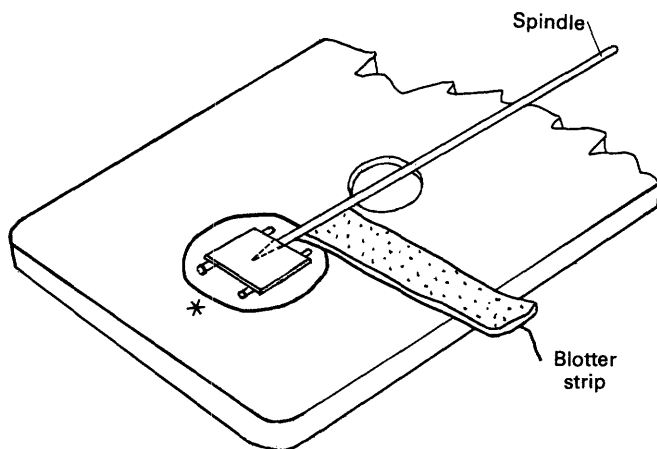


Figure 6. Withdrawal of liquid from immersion cell using absorbent paper.

removed, the new liquid is introduced at the front of the cell as before. This operation should be repeated once to flush out old liquid trapped in the cell and once again if the index change is large.

Orientation of the fragment for index determination may be carried out conoscopically (Wilcox, 1959, p. 1277), enabling at the same time an overview of other optical properties such as optic angle and optic sign. If the fragment is too small or if interference figures cannot be obtained, the somewhat more time-consuming orthoscopic procedure (Wilcox, 1959, p. 1281; Bloss and Riess, 1973; Bloss, 1981) can be used.

For a biaxial crystal, the correct orientation for index comparison is recognized conoscopically by an interference figure having bilateral symmetry along the crosshair perpendicular to the plane of the polarizer, as pointed out by Slawson and Peck (1936). Typical examples are shown in figure 7, in which the polarizer is NS. For any randomly mounted fragment, there are three orientations (within the upper hemisphere of rotation) that satisfy this criterion and at which the respective principal refractive indices are to be determined. It is advantageous to start with the orientation for β , that is, the orientation where the optic plane isogyre is symmetrically disposed along the EW crosshair. This isogyre is recognized, even before reaching the symmetrical setting, by its "counterrotating" behavior (Stoiber and Morse, 1972, p. 122, 172) or by the presence of the melatope itself, about which the isogyre pivots. With this isogyre symmetrically along the EW crosshair, the settings of spindle stage arm and microscope stage, at which β is to be determined, are recorded.

One then proceeds toward the setting for the next principal index by small rotations about the spindle axis followed by compensating rotations on the microscope stage to keep the one isogyre always well within the field of view. Eventually another symmetrical interference figure must be encountered, with an isogyre lying either along the EW crosshair, which is the desired setting, or along the NS crosshair, in which case the desired setting is made by rotating 90° on the microscope stage. The accessory plate may be used to establish whether it is the vibration direction of α or γ that is now NS, and the readings of spindle arm and microscope stage are recorded. One proceeds in the same manner to establish the setting for the third principal index. Converting now to orthoscopic illumination and a lower power objective, the index of the liquid is compared with each of the three principal indices at its respective setting by means of one of the index comparison techniques out-

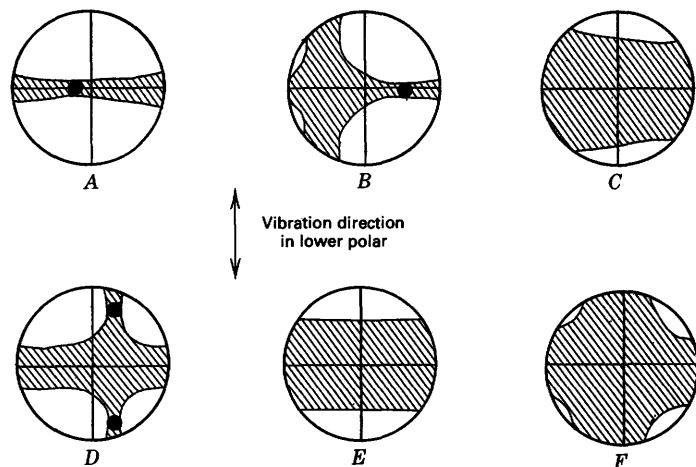


Figure 7. Examples of typical isogyre configurations to be seen when a principal vibration direction is horizontal EW; that is, when the crystal is oriented for comparison of the corresponding principal refractive index with the immersion liquid (polarizer NS).

lined in earlier sections. Liquids are changed accordingly to find the matching liquid for each index and, after correcting for temperature, the principal refractive indices are recorded.

The procedure for a uniaxial crystal is somewhat simpler, once having established from an optic axis figure that it is indeed uniaxial. Here one simply places an isogyre along the EW crosshair and, by incremental rotations of spindle and stage, "follows" this isogyre to the flash figure. The vibration direction of ω is now NS and, converting to orthoscopic illumination at this setting, ω may be compared with the liquid. Rotating 90° to the other extinction position similarly permits ϵ to be compared with the liquid.

Birefringence

The difference between the extreme principal refractive indices, α and γ , or ω and ϵ , ordinarily referred to as the birefringence of the mineral, is very helpful in identification. Birefringence is especially useful for identification of an unknown that is a member of an isomorphous series, for birefringence usually remains within relatively narrow limits while the refractive indices may have a wide range.

In routine immersion work, birefringence is usually taken as the difference between the already measured values of γ and α , or of ω and ϵ . Lacking these measurements, it can be estimated from the order of interference color observed between crossed polars, giving due consideration to the orientation and the probable thickness of the fragment. (This approach, of course, is the basis for rapid recognition of many minerals in thin section, where the thickness is known.) Maximum birefringence may also be determined on the spindle stage, following the suggestion of Fisher (1960, p. 87), by mounting the fragment on the spindle so that the optic plane can be made horizontal (flash figure), in which position the maximum retardation, R , is determined at the point where thicknesses is greatest using a compensator.⁴ That thickness, t , is measured with the ocular micrometer after rotating the fragment 90° about the spindle axis to show it in profile. Birefringence, B , is then $B = R/t$.

⁴If the optic plane of the randomly mounted fragment is oblique to the spindle axis, it cannot be made horizontal by simple rotation about the spindle axis. With practice, however, it is not difficult to correct the position of the fragment by (1) rotating the optic plane to verticality, (2) noting the angle of the optic plane with the spindle, (3) withdrawing the spindle, (4) dipping the spindle tip in solvent to remove the oil, (5) breathing gently on the mount to soften the adhesive, and finally (6) under the stereomicroscope with a dissecting needle, nudging the fragment the estimated amount necessary to place the optic plane in line with the spindle.

Optic Angle and Optic Sign

The optic angle, usually designated as $2V$, is the angle between the two optic axes (see fig. 3). A statement of the numerical value of the angle is not complete without an indication of optic sign. A positive optic angle may be shown, for example, as $(+)2V=47^\circ$, $2V_Z=47^\circ$, or $2V\gamma=47^\circ$. The size of the optic angle can be calculated from the relationships among the three principal indices of refraction or can be measured independently, thus offering a means of checking on the consistency of the measured values of principal refractive indices.

Calculation of Optic Angle

The relation of optic angle to the principal refractive indices may be expressed as

$$2V_Z = 2 \arccos \frac{\alpha}{\beta} \sqrt{\frac{\gamma^2 - \beta^2}{\gamma^2 - \alpha^2}} \quad (1)$$

where $2V_Z$ is the angle between the two optic axes across the bisectrix Z . A simpler approximation,

$$2V_Z = 2 \arccos \sqrt{\frac{\gamma - \beta}{\gamma - \alpha}} \quad (2)$$

is accurate within a degree or two if the optic angle or birefringence is not too high (see also Wright, 1951; Tröger, 1952).

Convenient nomograms based on the exact equation (Mertie, 1941) are reproduced by Bloss (1961, p. 158) and by Hartshorne and Stuart (1970, p. 393, and folding plate). Knowing any three terms of equation (1), the fourth term may quickly be determined from the nomogram. Thus, it is easy to determine not only the nominal value of $2V$, but also the range of uncertainty due to the estimated inaccuracies in the refractive index determination, as illustrated by the example in table 2 (see also

Table 2. Range of calculated optic angles, taking into account a possible ± 0.001 error of refractive index determinations

	α	β	γ	2V by exact formula	2V by approx. formula	2V by nomogram
Measured refractive indices (± 0.001)	1.534	1.545	1.560	(+)81° 46'	(+)81° 10'	(+)82°
Lowest worst case	1.535	1.544	1.561	(+)72° 40'	(+)72° 10'	(+)73°
Highest worst case	1.533	1.546	1.559	(+)89° 24'	90° 00'	(+)89°
Range, lowest to highest				16° 44'	17° 50'	16°

Willard, 1961). Obviously it is more realistic and informative to state the calculated value as $(+)2V_{\text{calc}}=82^\circ \pm 8^\circ$ than to imply an unjustifiable precision, such as $(+)2V_{\text{calc}}=81^\circ 46'$.

An approximate value of optic angle, quite adequate for much work, may be obtained from figure 8A. This nomogram is based on equation (2) using the form proposed by Tröger (1952). Like Tröger's nomogram, its error is considerably less than the error resulting from the usual uncertainties in the determination of the principal refractive indices themselves, especially for minerals of low or moderate birefringence. Should high birefringence justify refinement, a correction, to be added algebraically to the approximate value, may be obtained from figure 8B.

As an example, using the following measurements of a specimen of augite:

$$\begin{aligned}\alpha &= 1.680 \pm 0.001 & \gamma - \beta &= 0.019 \\ \beta &= 1.686 \pm 0.001 & \gamma - \alpha &= 0.025 \\ 2V_z (\text{measured}) &= 61^\circ \pm 2^\circ\end{aligned}$$

$$\gamma = 1.705 \pm 0.001 \quad \frac{\gamma - \beta}{\gamma - \alpha} = 0.760$$

From fig. 8A:
approximate $2V'_z = (+)58.7^\circ$

From fig. 8B:

correction $+ 0.6$
Corrected calculated
optic angle, $2V_z = 59.3^\circ$,
compared to value obtained from equation (1): $2V_z (\text{calc}) = 59.2^\circ \pm 11^\circ$.

Similarly, for a specimen of hastingsite:

$$\begin{aligned}\alpha &= 1.653 \pm 0.001 & \gamma - \beta &= 0.010 \\ \beta &= 1.665 \pm 0.001 & \gamma - \alpha &= 0.022 \\ 2V_x (\text{measured}) &= 80^\circ \pm 2^\circ \\ \gamma &= 1.675 \pm 0.001 & \frac{\gamma - \beta}{\gamma - \alpha} &= 0.455\end{aligned}$$

From fig. 8A:
approximate $2V'_x = (-)84.8^\circ$

From fig. 8B:
correction $+ 0.6$
Corrected calculated
optic angle, $2V_x = 84.2^\circ$,

compared to value obtained from equation (1): $2V_x (\text{calc}) = 84.2^\circ \pm 10.5^\circ$.

This relation between optic angle and the three principal refractive indices provides a useful test of the internal consistency of the measured results. If the value of optic angle calculated from measured α , β , and γ is not in reasonable agreement with the optic angle, there must be an error in one or more of the measured values, and a recheck is in order. For tables 6 and

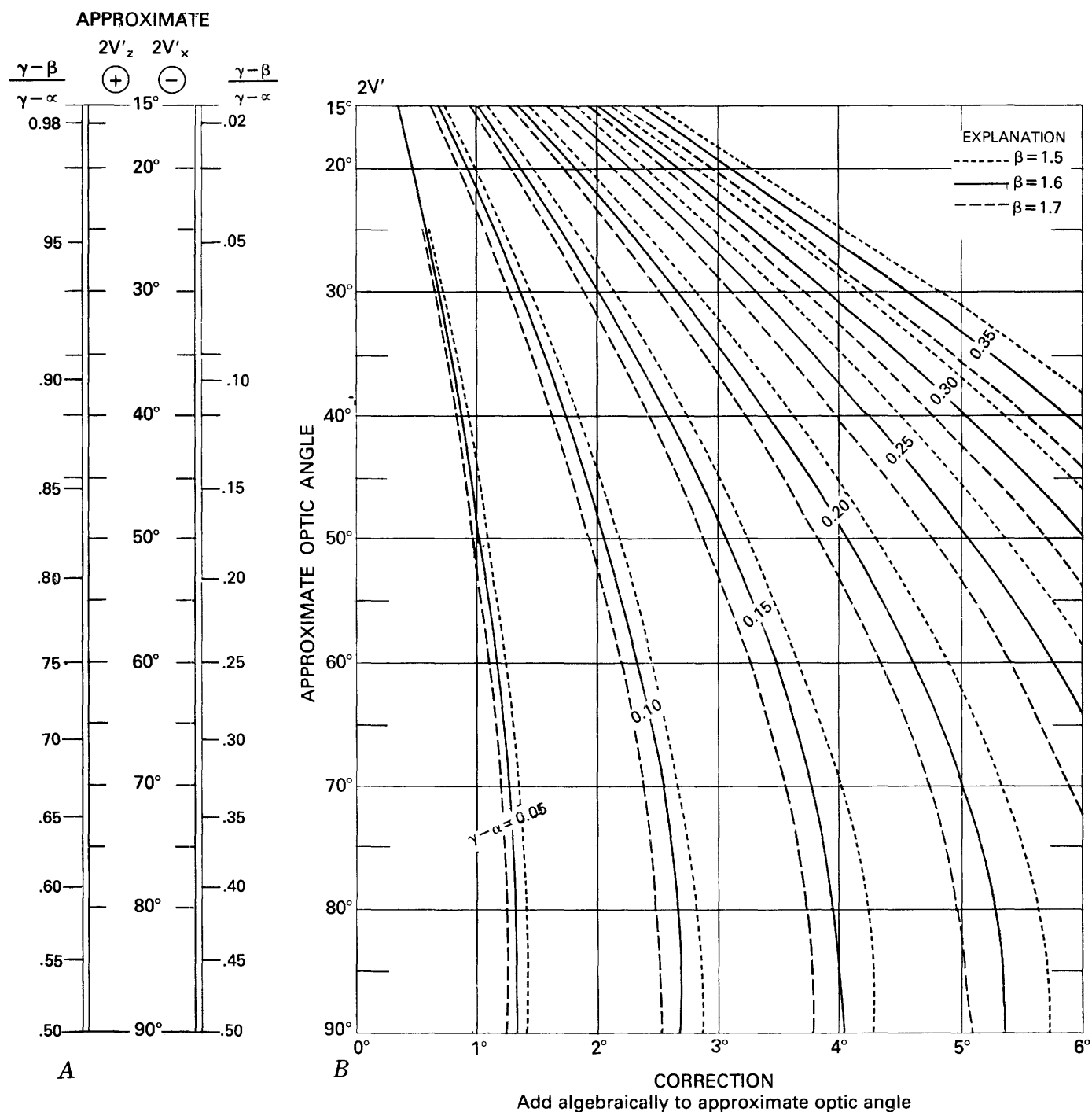


Figure 8. Nomogram for determination of optic angle from ratio of partial birefringences, $\frac{\gamma-\beta}{\gamma-\alpha}$: A, Approximate optic angle. B, Correction to be added algebraically.

7, this test has been run on the data as reported; where agreement is poor, the calculated value of optic angle has been inserted in parentheses below the reported optic angle as a cautionary signal. Also shown in the parentheses is a tolerance angle based on assumed "worst case" errors of ± 1 significant

figure in the values reported for principal refractive indices.⁵

⁵Unfortunately few data in the literature are accompanied by estimates of error. The "worst case" assumed here may err on the side of charity for many of the reported data and may be too pessimistic for others.

Measurement of Optic Angle

Among several means available to measure optic angles of crystal fragments in immersion mounts are the following:

(1) direct measurement conoscopically on the universal stage or spindle stage by placing the optic plane perpendicular to the rotation axis, then rotating from one melatope to the other across an identified bisectrix. Somewhat less reliably (Fairbairn and Podolsky, 1951), the optic angle may be determined by doubling the angle of rotation from one melatope to a bisectrix;

(2) measurement of the angle between stereographically plotted optic axis positions, as determined conoscopically by an extension of Mallard's method on the spindle stage (Wilcox and Izett, 1968; Noble, 1968);

(3) stereographic plotting of a series of orthoscopic extinction positions, determined on the spindle stage, from which the unique positions of the optic axes may be inferred (Wilcox, 1960; Tocher, 1962; Bloss and Reiss, 1973; Bloss, 1981). This procedure is analogous to the Berek-Dodge Method for universal stage (Emmons, 1943, p. 28), without requiring corrections of rotations.

Whatever the method of determining the optic angle, the identity of the acute bisectrix, X or Z, must also be determined to establish the optic sign. Direct measurement of optic angle by rotation from one optic axis to the other is of course to be preferred when the situation permits. On the universal stage, the optic plane may be brought directly into position for the rotational measurement, to which are applied corrections for index difference between mineral and hemispheres. On the spindle stage, the fragment may be nudged into position after the adhesive has been softened, and no correction to the rotational measurement is required. All three methods provide greater accuracy than can usually be obtained by calculation of optic angle from principal refractive indices, particularly for

substances of low or moderate birefringence.

Estimation of Optic Angle and Determination of Sign

For rough work, one may use the method developed by Wright (1911, p. 168) to obtain an estimate of optic angle. This requires an optic axis interference figure, that is, a melatope near the center of the conoscopic field, and judgment of the curvature of the isogyre when the optic plane is at 45° to the polars. The dark band of the isogyre is distinctly curved for small optic angles and straight diagonally across the field for optic angles approaching 90° . Estimates of optic angle to within about 10° can be made using figure 9, modified from Wright's

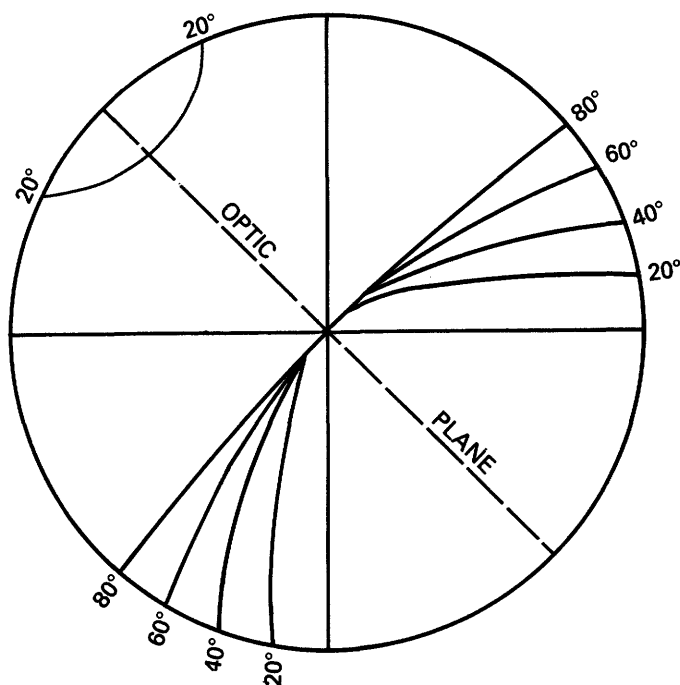


Figure 9. Curvatures of superimposed isogyres for centered optic axis figure of optic angles 20° , 40° , and 80° (after Wright, 1911).

(1911) composite of superposed isogyres, keeping in mind that objectives of higher numerical aperture will display a greater length of isogyre (see Winchell, 1965, p. 11). Determination of optic sign is made on the same interference figure by placing the optic plane in the diagonal position parallel to the vibration direction of the slow ray (high index) of the inserted accessory plate. For an optically positive crystal, insertion of the plate raises the retardation color on the concave side of the isogyre at the melatope.⁶

When both melatopes lie symmetrically disposed in the conoscopic field, one may use Mallard's method *sensu stricto*, on the basis of separation of the melatopes (Johannsen, 1918). When the melatopes are not symmetrically disposed, one may still estimate the size of the optic angle by comparison of their separation with the total diameter of the conoscopic field of known angular aperture. With one melatope and a bisectrix in the conoscopic field, the half-optic angle can be estimated in a similar manner but with reduced precision. When both melatopes lie outside the conoscopic field of a near-centered bisectrix figure or flash figure, the optic angle may be estimated roughly by the amount of rotation of the microscope stage that is required to move the isogyres from the crossed position to a certain radial distance representing a known angular aperture (Johannsen, 1918, p. 472; Kamb, 1958). Inasmuch as it is the darkest band of the diffuse and

varying isogyre that must be set at the reference radial distance, the latter is best chosen well within the field rather than at the extreme outer edge, as done in the theoretical treatment by Kamb (1958). A useful discussion of this and other conoscopic methods is given by el-Hinnawi (1966, p. 114-136).

Dispersion of the Optic Axes

If the directions of the optic axes change with wavelength, the mineral is said to have dispersion of the optic axes, a characteristic often helpful in distinguishing it from otherwise similar minerals. This property is included in the tables together with the nominal value of $2V$ for the standard yellow light. For example, for the biaxial positive minerals (table 6), wavellite at $\beta=1.524$ shows $2V_z$ as " 60° , $r > v$ wk," meaning that the optic angle for red is slightly greater and for blue slightly less than the 60° for yellow.

Dispersion of an optic axis may be recognized in white light illumination using a near-centered optic axis figure with the optic plane 45° to the polars. Because the melatope for each wavelength (color) is a point of no illumination and because these melatopes are now slightly spread out along the trace of the optic plane, the melatope for the red wavelengths is represented by a bluish color, while that for the blue wavelengths is represented by a reddish or, more commonly, brownish color. Thus, if the concave side of the isogyre is seen as bluish and the convex side reddish or brownish, it is concluded that the optic angle for red is greater than that for blue wavelengths, $r > v$.

Relation of Optical to Crystallographic Directions

The orientation of the indicatrix with respect to crystallographic elements, such as crystal axes, faces, cleavages, and twin planes, can be of much help in identification of an unknown, particularly after the number of

⁶To avoid misreading the direction of curvature, especially for a crystal of large optic angle, a setting is first made with the isogyre parallel to (not necessarily on) the NS crosshair. The stage is then rotated exactly 45° to read the curvature of the isogyre. The farther the melatope from the center of the field, of course, the smaller is the length of isogyre available on which to estimate curvature and, in general, the less precise is the estimate.

candidates has been reduced to a manageable few on the basis of the observed principal refractive indices, optic angle, and sign. Firm establishment of the relation of indicatrix axes to crystallographic axes would require coordinated optical and X-ray structural investigation of the same crystal fragment (see for instance Chao and Minkin, 1970). For many unknowns, however, much can be inferred from the behavior of interference figures of fragments lying on a prominent cleavage (see for instance Ehlers, 1980) and from such readily determinable properties as extinction angle and sign of elongation.

Extinction Angle

The extinction angle is the angle through which the microscope stage must be rotated to change it from a position in which a specified crystallographic direction is parallel to the crosshair to the extinction position in which the specified vibration direction is parallel to the crosshair. The vibration direction may be a principal symmetry axis of the indicatrix, such as X or Z, or, when the symmetry planes of the indicatrix are oblique to the section, merely an extinction position expressed then as X' or Z'. If the extinction angle is not given as such in tables 6 and 7, it often can be inferred from the basic information provided in the columns on "Optical orientation," "System," and "Cleavage." Helpful in visualizing these relationships are the admirable perspective sketches of typical crystals in Tröger (1979).

Extinction angles serve to distinguish between certain important minerals or mineral groups, for instance, between clinopyroxenes and clinoamphiboles using the angle Z:c. Extinction angles serve further to distinguish between members in an isomorphous series, such as the plagioclase series using X':a. On the spindle stage the crystallographic directions of a fragment may be inferred by orienting and plotting the bounding cleavage surfaces, from which the specified extinction angle can be read off

the stereogram. Alternatively the fragment can be mounted with a recognizable crystallographic zone axis parallel to the spindle axis so that the extinction angle can be read directly.

Sign of Elongation

The sign of elongation is positive if a characteristic direction of elongation of the fragment or crystal is less than 45° to the extinction position of the slow wave (high index). A mineral with two directions of good or excellent cleavage will generally break into elongate fragments. Thus, cleavage fragments of most amphiboles show positive elongation, since Z is at a low angle to c, defined by the intersection of the excellent prismatic cleavages. For most micas, both Y and Z lie in or very near the plane of the one perfect cleavage, so that cleavage flakes, viewed edge-on in thin section or on the spindle stage, show positive elongation. In random immersion mounts, of course, this relation is generally not seen because the flakes lie on their cleavage.

Color and Pleochroism

The color of the hand specimen, together with its luster, can be helpful in sight-identification of some minerals. Caution is required here, however, because the color can be affected appreciably by foreign inclusions, internal flaws, fractures, or by variations of certain minor chemical constituents. Under the stereo-microscope with standard conditions of illumination, small fragments may show characteristic colors and lusters, enabling approximate modal counts and removal of certain species by hand picking.

In plane-polarized transmitted light, the colors of small fragments are diagnostic of certain mineral species. Pleochroism, the change in color with the change in direction of vibration of the plane-polarized light, is especially useful in sight identification. Pleochroism is customarily described in

terms of colors observed for light vibrating parallel to the respective axes of the indicatrix: O and E for uniaxial crystals and X, Y, and Z for biaxial crystals. (Strictly speaking for monoclinic and triclinic crystals, however, the principal axes of the indicatrix do not necessarily coincide with the principal axes of the "absorption ovaloid." Clinopyroxenes provide an example, for which a more consistent convention was proposed by Hess (1949, p. 632).)

Depth of color for light transmitted through fragments in immersion mounts increases with the thickness of the fragment, so that the intensities of the observed colors may not correspond strictly to the pleochroic colors listed in tables 4-7. An added complication may arise in immersion mounts due to changes in dispersion colors or relief with rotation of the stage, for either may simulate pleochroism or mask a faint body color. These disturbing effects can largely be eliminated by swinging in the substage auxiliary condenser to flood the crystal with light. One must keep in mind, however, the differences in the spectral distribution of the source illumination from one microscope to another. In addition, one must be aware of the subjective nature of the descriptive terms used in the tables; these terms might be interpreted differently by different observers.

CHAPTER III. IDENTIFICATION TABLES (TABLES 3-7)

List of Abbreviations

a-----crystallographic axis a
Ab-----albite
abs-----absorption
acic-----acicular
Ak-----Akermanite
Al-----Almandine
alk-----alkaline
amor-----amorphous
An-----Anorthite, Andradite
anom-----anomalous
b-----crystallographic axis b
bb-----before the blowpipe

biax-----biaxial
biref-----birefringence, birefringent
c-----crystallographic axis c
Ca-----Calderite
calc-----calculated
char-----character
clv-----cleavage
Cn-----Celsian
cols-----colorless
comp-----composition
compd-----compound
concd-----concentrated
conch-----conchoidal
cryptocryst-----cryptocrystalline
cryst-----crystalline
crystn-----crystallization
cub-----cubic
dec-----decomposed
dehyd-----dehydrates
deliq-----deliquesces, deliquescent
dtd-----determined
diff-----difficult, difficulty
dil-----dilute
disp-----dispersion
diss-----dissolved
dist-----distinct
dod-----dodecahedral
E-----extraordinary wave
eff-----effervesces, effervescence
effl-----efflorescent
el-----elongation
En-----Enstatite
ext-----extinction
extr-----extreme
F-----fusibility
Fa-----Fayalite
fib-----fibers, fibrous
fluor-----fluoresces
Fo-----Forsterite
fr-----fracture
Fs-----Ferrosilite
fus-----fusible
G-----specific gravity
Ge-----Gehlenite
gel-----gelatinizes
Go-----Goldmanite
Gr-----Grossularite
grp-----group
H-----hardness
hex-----hexagonal
hyg-----hygroscopic
imperf-----imperfect
insol-----insoluble
isomor-----isomorphous
isot-----isotropic

Kn-----Knorringite
 lam-----lamellar, lamellae
 mass-----massive
 mcl-----monoclinic
 Me-----Meionite
 med-----medium
 mic-----micaceous
 mod-----moderate
 n-----refractive index
 neg-----negative
 O-----ordinary wave
 oct-----octahedral
 opt-----optical, optically
 Or-----Orthoclase
 org-----organic
 orth-----orthorhombic
 pct-----percent
 penet-----penetration
 perc-----perceptible
 perf-----perfect
 pl-----plane
 pleoc-----pleochroic, pleochroism
 poly-----polysynthetic
 pos-----positive
 pris-----prismatic
 ps-----pseudo
 Py-----Pyrope
 pyram-----pyramidal
 r-----optic angle in red light
 rect-----rectangular
 sepn-----separation
 ser-----mineral series
 sol-----soluble
 soln-----solution
 Sp-----Spessartite
 spher-----spherulites, spherulitic
 sq-----square
 str-----strong
 symm-----symmetrical
 synth-----synthetic
 tab-----tabular
 tcl-----triclinic
 tet-----tetragonal
 tetrah-----tetrahedral
 tr-----trace
 trig-----trigonal
 tw-----twinning
 u-----usually
 uniax-----uniaxial
 unk-----unknown
 Uv-----Uvarovite
 UV-----ultra-violet
 v-----optic angle in blue light
 var-----variety
 volat-----volatilizes

wk-----weak
 X, Y, Z-----vibration directions for
 α , β , γ , respectively
 $2V_x$ -----optic angle about X
 $2V_z$ -----optic angle about Z
 \perp -----perpendicular to

 ||-----parallel to
 >-----greater than
 <-----less than
 ~-----near, approximately
 \pm -----plus or minus
 001, 00 $\bar{1}$ -----Miller symbols for all
 faces of a form
 (conventional braces
 have been omitted)
 :-----angle with
 α , β , γ -----lowest, intermediate, and
 highest principal refractive
 indices of biaxial crystals,
 respectively
 ω , ϵ -----ordinary and extraordinary
 principal refractive indices
 of uniaxial crystals

Arrangement of Data

The minerals are divided among five tables according to their basic optical characteristics: isotropic minerals, including mineraloids and natural glasses, are in table 3; uniaxial positive minerals are in table 4; uniaxial negative minerals are in table 5; biaxial and positive minerals are in table 6; biaxial negative minerals are in table 7. The mineral names generally follow those of Fleischer (1983), and the chemical formulas attempt to convey the type of compounds and interchangeable elements without necessarily indicating lattice positions or other structural aspects.

In table 3 the isotropic substances are listed in order of increasing refractive index, n , shown in the second column. In the first column, titled "Other entries," arrows with values of refractive index indicate the next higher or lower listing of the particular mineral. Since the determined

value of refractive index of the unknown will generally fall between those of the representative entries of that mineral in the table, it is essential to give close attention to the indications of index variation shown in this column. Known variations of index too small to justify separate entries are indicated by square bracket symbols, with or without numerical values. The lack of a variability notation in this column opposite a mineral name, however, does not mean that variability does not exist, rather merely that none was found in published descriptions, a point to be kept in mind especially for the lesser known minerals of complex composition.

Next in table 3 are columns for crystal "System and habit" and for "Cleavage or fracture." Here note that the conventional enclosing braces denoting forms are omitted from the Miller symbols because of space limitations. The "Color" column indicates the color of fragments in reflected light. "Hardness, specific gravity, and fusibility" are shown in the next column, and finally in the "Remarks" column are given various additional helpful properties, such as optical anomalies, weight percentages of selected oxides of the chemical analysis, and behavior in chemical tests. Dilute acids are understood unless otherwise specified.

In tables 4 and 5 the uniaxial minerals are listed in order of increasing values of the ordinary principal refractive index, ω , with the extraordinary principal index, ϵ , given in the adjacent column. In addition to the properties already mentioned for table 3, the "Remarks" column of tables 4 and 5 gives information on twinning, sign of elongation, and pleochroism.

In tables 6 and 7, the biaxial minerals are listed in order of increasing intermediate principal refractive index, β , with the values of α and γ in adjacent columns and with birefringence in another column. (Parentheses around an index or birefringence indicate a value calculated from the reported value of optic angle). The column for "2V" gives the reported value of optic angle, and if this value does not appear to be

consistent with the reported values of α , β , and γ , the calculated value of optic angle is given in parentheses, along with a plus-or-minus tolerance based on possible errors of one in the last significant figure of the reported indices. (No calculated values are shown for minerals of low birefringence, since here normal errors in index would lead to wide variations in calculated optic angle). Also shown is the reported dispersion of the optic angle. The column for "Optic orientation" gives diagnostic relationships between indicatrix symmetry axes X, Y, and Z, and crystallographic axes, \underline{a} , \underline{b} , and \underline{c} , as well as extinction angle and sign of elongation. The statement "el clv pos," for instance, indicates positive elongation as seen with steeply dipping cleavage.

Because of the variations of optical character, some minerals are listed in more than one table. For example, a normally isotropic mineral that is weakly anisotropic in some occurrences may be listed in both table 3 (isotropic minerals) and in the appropriate table for anisotropic minerals. Furthermore, in some isomorphous series the optic sign may reverse due to progressively changing relations between the principal refractive indices. The melilite group, for instance, passes from optically uniaxial negative gehlenite through "isotropic" melilite to uniaxial positive akermanite and thus is listed in three tables. In the olivine group, the magnesium-rich members are optically positive, but as the Fe/Mg ratio increases the optic angle passes through 90° near Fa_{15} and the remainder of the series is optically negative. In table 6 (biaxial positive minerals), therefore, an olivine group mineral is entered at β -values of 1.651, 1.664, 1.674, and finally at 1.680, at which point a switch to table 7 (biaxial negative minerals) is indicated by the notation 1.680 neg. Due to thermal history or other factors, the cross-over in an isomorphous series may not take place precisely at the same value of β -index in every case; so that, with an unknown or observed optic angle near 90° , likely candidates should be sought

in both tables near the measured value of β .

Use of the Tables for Identification

In all cases, a systematic and thorough procedure must be carried out, and the following approach is recommended: For an isotropic unknown, one enters the table at the determined value of refractive index and then examines the other properties given for minerals of index within a reasonable range above and below the determined refractive index. Unfortunately for isotropic crystals there are no other optical-crystallographic parameters easily measured under the microscope, so one must depend heavily on cleavage, hardness, solubility, and so on. Those minerals that correspond to the unknown in these respects should be listed and consideration then given as to whether X-ray, chemical analysis, or another approach will be most effective in making the final assignment of identity.

For an anisotropic unknown, after having determined the values of principal indices, birefringence, optic angle, and optic sign, the following steps are recommended:

(1) enter the appropriate table at the determined value of ω or β .

(2) proceed both above and below in the table for a distance of at least 0.020 difference in index, listing the names of those minerals whose values of birefringence and optic angle are in reasonable agreement with those of the unknown.

(3) for each mineral so listed, check the remaining columns for distinguishing properties, returning to the microscope or the original sample as necessary to make diagnostic tests and crossing out each nonconforming mineral.

(4) for the remaining candidate or candidates, consult encyclopedic optical descriptions for further distinguishing properties.

(5) confirm or complete identification if necessary by X-ray and chemical analyses.

Examples

Example No. 1.--is crushed material from clusters of clear, colorless crystals in small lithophysae in rhyolite. Fragments mounted on the spindle stage show:

α 1.611 \pm .001	γ - α 0.011 \pm .002
β 1.614 \pm .001	$2V_Z$ 65 $^\circ$ \pm 2 $^\circ$ measured
γ 1.622 \pm .001	

Entering the table for biaxial positive minerals (table 6) at β =1.614 and proceeding toward lower and higher values of β , the following minerals seem reasonable possibilities (keeping in mind that, due to chemical complexity, some of them may be expected to vary in index even though it is not so indicated in the "Other entries" column).

Looking at the characteristics listed in other columns of table 6, tuhualite can be eliminated at once on the basis of its "dark blue to black" color; probably uranopilite ("bright yellow, straw yellow") can also be eliminated. In a simple hardness test, particles of the unknown leave scratches when rubbed between two glass slides, showing that it has a hardness of 5 or greater; this finding confirms the elimination of tuhualite and uranopilite, and also eliminates crandallite, uralborite, and buchwaldite. Under the microscope, the unknown appears to show one perfect cleavage, casting doubt on stokesite, which should show two cleavages. As a test of density, particles of the unknown in a drop of methylene iodide (sp gr 3.3) on a microscope slide are seen to rest at the bottom of the liquid; this eliminates stokesite (sp gr 3.19), afwillite (sp gr 2.62) and cuspidine (sp gr 2.86-3.05), all of which should float. The unknown is found to be insoluble eliminating hydrophilite, and is unaffected by acids, eliminating latiumite, afwillite, cuspidine, and uranopilite. This leaves only topaz which possesses all the observed characters of the unknown. According to the variation diagram given by Winchell and Winchell (1951, p. 509), the optical properties would imply a high fluorine variety of topaz.

	Other entries	β	Bire- fringence	$2V_z$
Cuspidine		1.595	.012	62°
Hydrophilite		1.605	.012	med
Latiumite		1.606	.014	83°-90°
Crandallite		1.607	.013	70°-75°
Uralborite		1.609	.011	85°
Topaz		1.610	.011	67°
Buchwaldite		1.610	.009	(?)
Tuhualite		1.612	.013	60°-70°
Stokesite		1.613	.010	70°
(UNKNOWN IN QUESTION)		1.614 ±.001	.011 ±.002	65° ± 2°
Topaz		1.618	.009	61°
Afwillite		1.619	.015	55°
Stokesite		1.621	.012	70°
Uranopilite		1.624	.011	small to large
Topaz		1.631	.099	48°

Example No. 2.--In a crushed sample of a coarse igneous rock as seen under the stereomicroscope, fragments of the unknown are clear and faintly honey-brown in color. Mounted on the spindle stage they show the following optical properties:

α 1.682 \pm .001
 β 1.691 \pm .001
 γ 1.696 \pm .001

$\gamma-\alpha$.014 \pm .002
 $2V_x$ 70 $^\circ$ \pm 2 $^\circ$ measured
 (73 $^\circ$ \pm 17 $^\circ$ calc)

Biaxial negative minerals of similar optical properties in table 7 within about 0.020 of this β -index form an extensive list as follows:

	Other entries	β	Birefringence	$2V_x$
Bustamite		1.675	.015	53 $^\circ$
Gerstmannite		1.675	.013	50 $^\circ$ -60 $^\circ$
Ferroaxinite		1.676	.011	69 $^\circ$
Bronzite, orthopyroxene		1.680	.011	79 $^\circ$
Tinzenite		1.681	.014	large
Manganaxinite		1.687	.014	75 $^\circ$
Lavenite		1.690	.016	68 $^\circ$
Chlorophoenicite		1.690	.015	83 $^\circ$
(UNKNOWN IN QUESTION)		1.691 \pm .001	.014 \pm .002	70 $^\circ \pm 2^\circ$
Tinzenite		1.693	.010	med
Bronzite, orthopyroxene		1.695	.015	69 $^\circ$
Bustamite		1.695	.015	43 $^\circ$
Arfvedsonite		1.695	.015	69 $^\circ$
Kempite		1.695	.014	med
Ferroaxinite		1.695	.011	70 $^\circ$
Ferro-richterite		1.699	.016	35 $^\circ$ (68 $^\circ \pm 15^\circ$)
Tinzenite		1.701	.011	63 $^\circ$
Barylite		1.702	.013	70 $^\circ$
Bustamite		1.708	.015	34 $^\circ$

At the outset it would appear doubtful that the unknown could be bustamite, due to the fairly large discrepancy in optic angle. This doubt is confirmed on fragments of the unknown on the spindle stage. There they are seen to have two good cleavages, whereas bustamite has three cleavages, two good and one perfect. Members of the axinite group (ferroaxinite, manganaxinite, and tinzenite) with lavenite and gerstmannite are similarly eliminated, inasmuch as they have only one cleavage. Cleavage fragments of the unknown show parallel extinction and positive elongation, eliminating arfvedsonite and ferrorichterite, which have inclined extinction. Although barylite fragments should also show two

good cleavages (001 and 100) and parallel extinction, the elongation would be negative, since $X=b$. A scratch test of fragments of the unknown between glass slides shows their hardness to be between 5 and 6, eliminating kempite and chlorophoenicite, having hardness 3.5, and confirming the elimination of gerstmannite, having hardness 4.5.

This leaves only orthopyroxene as a viable candidate; and, indeed, under the polarizing microscope the fragments show the pleochroism common in many orthopyroxenes: Z faint green, X faint reddish brown. An orthopyroxene with these refractive indices would have composition near En_{73} (Deer, Howie, and Zussman, v. 2A, 1978, p. 109).

TABLES 3–38

Table 3. Isotropic minerals and mineraloids

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	1.309	ICE H_2O	HEX	Conch	Cols	H 1.5 G 0.92	Melts at 0°C. Biref v low, pos.
	1.326	VILLIAUMITE NaF	CUB u mass	001 perf	Carmin-red to cols	H 2-3.5 G 2.79 F easy	Sol in H_2O . Also uniax neg, biref v low. Pleoc, 0 carmine-red, E golden-yellow.
	1.333	WATER H_2O	Liquid	---	Cols	G 1.00	---
	1.339	CRYOLITHIONITE $Na_3Li_3Al_2F_{12}$	CUB	011 dist	Cols	H 2.5-3 G 2.77 F easy	Diss by acids.
	1.340	HIERATITE K_2SiF_6	CUB cubo-oct	111 perf	Cols, gray	H ~ 2.5 G 2.67	Sol in hot water.
	1.362	CAROBBIITE KF	CUB	001	Cols	G 2.50 (synth) F easy	Sol in H_2O .
	1.364	NEIGHBORITE $NaMgF_3$	ORTH ps cub	Uneven	Cols to brownish	H 4.5 G 3.03	Slowly diss by hot acids. Biref 0.002.
	1.369	CRYPTOHALITE $(NH_4)_2SiF_6$	CUB cubo-oct	111 perf	Cols	H ~ 2.5 G 2.03 volat	Sol in H_2O .
	1.376	ELPASOLITE K_2NaAlF_6	CUB	Uneven	Cols	H 2.5 G 2.99 F 3	Slowly diss by concd H_2SO_4 .
1.427 v	1.38	Unnamed fluoride near Ralstonite $(Ca,Na)(Mg,Al)_2(F,OH)_6 \cdot H_2O$	CUB oct	---	Yellow	---	Diss by acids. Am. Mineral., 28, 283-284 (1943).
1.434 v	1.406	OPAL $SiO_2 \cdot xH_2O$	Amor	Conch	Cols, white, gray, green	H ~ 6 G 1.9-2.3	Insol in acids, diss by KOH soln.
1.467 v	1.423	MELANOPHLOGITE $SiO_2 + C, H, O, S$	TET ps cub	---	Cols, brownish	H 6.5-7 G 1.99 infus	Insol in acids. Turns black when heated (organic matter). Biref wk.
1.38 ^	1.427	RALSTONITE $Na_xMg_xAl_{2-x}(F,OH)_6 \cdot H_2O$	CUB	111 poor	Cols to yellowish	H 4.5 G 2.52-2.62 infus	Dec by H_2SO_4 . Opt anom, divided into oct biref sectors.
1.440 v	1.434	FLUORITE CaF_2	CUB	111 perf	Cols, green, violet, yellow	H 4 G 3.18 F 1.5	Diss by acids.

1.406 ^ 1.46	1.434	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols	H ~ 6 G 1.86	Insol in acids, diss by KOH soln. Contains 12.6% H_2O .
	1.439	SODIUM ALUM $\text{NaAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Cols	H ~ 3 G 1.67	Sol in H_2O .
1.443 v	1.440	CHUKHROVITE $\text{Ca}_3(\text{Y,Ce})\text{Al}_2(\text{SO}_4)\text{F}_{13} \cdot 10\text{H}_2\text{O}$	CUB oct "Porcelain-like"	111 indist uneven	Cols, white	H 3 G 2.27-2.40 F easy	Diss by acids. Anom biref.
1.434 ^ 1.457	1.440	FLUORITE, yttrian $(\text{Ca,Y})\text{F}_{2-3}$	CUB u mass	111	Cols, green, violet	H 4 G 3.41 F 2	Slowly diss by HCl. Contains 6.9% Y_2O_3 .
1.440 ^	1.443	CHUKHROVITE-(Ce) $\text{Ca}_3(\text{Ce,Y})\text{Al}_2(\text{SO}_4)\text{F}_{13} \cdot 10\text{H}_2\text{O}$	CUB	---	---	---	---
1.51 v	1.44	HISINGERITE $(\text{Fe}^{+3}, \text{Mg}, \text{Fe}^{+2})_2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3.5 G 2.5 infus	Dec by acids.
1.465 v	1.445	EVANSITE approx. $\text{Al}_3(\text{PO}_4, \text{SiO}_4)(\text{OH})_6 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols, yellowish	H 3-4 G 1.87-2.13	Diss by hot H_2SO_4 .
1.462 v	1.450	NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2\text{F} \cdot 19\text{H}_2\text{O}$	CUB	Conch	Cols to white	H 2.5 G 1.72	---
	1.455	SULPHOHALITE $\text{Na}_6(\text{SO}_4)_2\text{FCI}$	CUB dod	Conch	Cols	H 3.5 G 2.50 F 1	Slowly sol in H_2O .
	1.456	POTASSIUM ALUM $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Cols	H 2-2.5 G 1.76 F 1	Sol in H_2O .
1.440 ^	1.457	FLUORITE, yttrian $(\text{Ca,Y})\text{F}_{2-3}$	CUB	111 imperf	Yellowish	H 4.5 G 3.55 fus	Diss by acids.
	1.458	TSCHERMIGITE $(\text{NH}_4)\text{Al}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	White	H 1.5 G 1.65 F 1	Sol in H_2O . Opt anom.
1.434 ^	1.46	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	Cols, white, gray, green	H ~ 6 G ~ 2.2 infus	Insol in acids, diss by KOH soln.
1.48 v	1.46	Glass, "Lechatelierite" mainly SiO_2	Amor	Conch	Cols	H 6 G 2.19 infus	Natural fused silica from fulgurite. SiO_2 99%.
1.450 ^	1.462	NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2\text{F} \cdot 19\text{H}_2\text{O}$	CUB	Conch	Cols to white	H 2.5 G 1.72	---

1.406 ↓ 1.46	1.434	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	Col's	H ~ 6 G 1.86	Insol in acids, diss by KOH soln. Contains 12.6% H_2O .
	1.439	SODIUM ALUM $\text{NaAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Col's	H ~ 3 G 1.67	Sol in H_2O .
1.443 ↓ 1.443	1.440	CHUKHROVITE $\text{Ca}_3(\text{Y}, \text{Ce})\text{Al}_2(\text{SO}_4)_2 \cdot 10\text{H}_2\text{O}$	CUB oct "Porcelain-like"	111 indist uneven	Col's, white	H 3 G 2.27-2.40 F easy	Diss by acids. Anom biref.
1.434 ↓ 1.457	1.440	FLUORITE, yttrian $(\text{Ca}, \text{Y})\text{F}_{2-3}$	CUB u mass	111	Col's, green, violet	H 4 G 3.41 F 2	Slowly diss by HCl. Contains 6.9% Y_2O_3 .
1.440 ^	1.443	CHUKHROVITE-(Ce) $\text{Ca}_3(\text{Ce}, \text{Y})\text{Al}_2(\text{SO}_4)_2 \cdot 10\text{H}_2\text{O}$	CUB	---	---	---	---
1.51 ↓ 1.51	1.44	HISINGERITE $(\text{Fe}^{+3}, \text{Mg}, \text{Fe}^{+2})_2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3.5 G 2.5 infus	Dec by acids.
1.465 ↓ 1.462	1.445	EVANSITE approx. $\text{Al}_3(\text{PO}_4, \text{SiO}_4)(\text{OH})_6 \cdot x\text{H}_2\text{O}$	Amor	Conch	Col's, yellowish	H 3-4 G 1.87-2.13	Diss by hot H_2SO_4 .
	1.450	NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2 \cdot 19\text{H}_2\text{O}$	CUB	Conch	Col's to white	H 2.5 G 1.72	---
	1.455	SULPHOHALITE $\text{Na}_6(\text{SO}_4)_2\text{FCl}$	CUB dod	Conch	Col's	H 3.5 G 2.50 F 1	Slowly sol in H_2O .
	1.456	POTASSIUM ALUM $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	Col's	H 2-2.5 G 1.76 F 1	Sol in H_2O .
1.440 ^	1.457	FLUORITE, yttrian $(\text{Ca}, \text{Y})\text{F}_{2-3}$	CUB	111 imperf	Yellowish	H 4.5 G 3.55 fus	Diss by acids.
	1.458	TSCHERMIGITE $(\text{NH}_4)\text{Al}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$	CUB oct	Conch	White	H 1.5 G 1.65 F 1	Sol in H_2O . Opt anom.
1.434 ^	1.46	OPAL $\text{SiO}_2 \cdot x\text{H}_2\text{O}$	Amor	Conch	Col's, white, gray, green	H ~ 6 G ~ 2.2 infus	Insol in acids, diss by KOH soln.
1.48 ↓ 1.48	1.46	Glass, "Lechatelierite" mainly SiO_2	Amor	Conch	Col's	H 6 G 2.19 infus	Natural fused silica from fulgurite. SiO_2 99%.
1.450 ^	1.462	NATROPHOSPHATE $\text{Na}_7(\text{PO}_4)_2 \cdot 19\text{H}_2\text{O}$	CUB	Conch	Col's to white	H 2.5 G 1.72	---

1.471 △	1.48	FAUJASITE (Zeolite grp) $(\text{Na}_2, \text{Ca})(\text{Al}_2\text{Si}_4)_0_{12} \cdot 8\text{H}_2\text{O}$	CUB oct	111 dist	White	H 5 G 1.92 F 3	Dec by HCl. On losing H_2O , becomes uniax, pos, in 8 ² sectors.
	1.482	Unnamed calcium silicate	---	---	White, silky fibers	Soft	Am. Mineral., <u>26</u> , 375 (1941).
v 1.487	1.483	SODALITE (Sodalite grp) $\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}\text{Cl}_2$	CUB dod	110 poor	Gray, white, blue	H 6 G 2.30 F 3.5-4	Gel with acids. Cl 7.1, H_2O 2.5%.
	1.486	CRISTOBALITE SiO_2	TET ps cub, oct	---	White	H 6-7 G 2.3 infus	Insol in acids. Biref very low. Intricate tw.
1.483 △	1.487	SODALITE, var Hackmanite (Sodalite grp) $\text{Na}_8\text{Al}_6\text{Si}_6(\text{O}, \text{S})_{24}\text{Cl}_2$	CUB dod	110 poor	Reddish-violet, blue	H 5 G 2.2-2.3 F 4	Gel with acids. Color fades in sunlight.
	1.489	D'ANSITE $\text{Na}_{18}\text{Mg}(\text{SO}_4)_{10} \cdot 3\text{NaCl}$	CUB	---	Cols, yellow	G 2.66 F easy	Sol in H_2O .
[1.49	VISHNEVITE (Cancrinite grp) $(\text{Na}, \text{Ca})_{6-8}(\text{Al}_6\text{Si}_6\text{O}_{24})$ $(\text{SO}_4, \text{CO}_3, \text{Cl})_2 \cdot 1-2\text{H}_2\text{O}$	HEX	10T0 perf	Cols	H 5 G 2.32 F 2	Gel with acids. Also uniax pos and neg. Biref .002.
1.47 ^ 1.52	1.49	ALLOPHANE Hydrous aluminum silicate	Amor	---	White, blue, green	H 3 G 1.86 infus	Gel with acids. Comp variable, may contain $(\text{PO}_4)^{-3}$.
	1.490	SYLVITE KCl	CUB	Cub, perf	Cols, white, red, yellow	H 2 G 1.99 F 1.5	Sol in H_2O . Tastes bitter.
1.479 △ 1.505	1.493	ANALCIME (Zeolite grp) $\text{NaAlSi}_2\text{O}_6 \cdot \text{H}_2\text{O}$	CUB 211	100 poor	Cols, white	H 5 G 2.25 F 3.5	Gel with HCl.
v 1.509	1.494	HAUYNE (Sodalite grp) $(\text{Na}, \text{Ca})_{4-8}\text{Al}_6\text{Si}_6(\text{O}, \text{S})_{24}$ $(\text{SO}_4, \text{Cl})_2 \cdot 1-2$	CUB dod	110 poor	Blue, white	H 6 G 2.40 fus	Gel with acids. SO_3 9.8, Cl 1.3, Na_2O 19.9, CaO 4.9%.
1.470 ^	1.495	NOSEAN (Sodalite grp) $\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}(\text{SO}_4)$	CUB dod	110 poor	Blue, white	H 6 G 2.3-2.4 F 4.5	Gel with acids.
	1.498	ZAHERITE $\text{Al}_{12}(\text{SO}_4)_5(\text{OH})_{26} \cdot 20\text{H}_2\text{O}$	Mass, fine-grained	One clv	White	H 3.5 G 2.01 infus	Biref < .001.
	1.50	ROSIERESITE Hydrous phosphate of Pb, Cu, Al	Amor compact	---	Yellow, brown	G 2.2 infus	Diss by HNO_3 . Blackens bb.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.48 ^ 1.53	1.50	Glass rhyolitic to dacitic comp	Amor	Conch	Cols, gray	H 5-6 G ~ 2.4	Insol in acids.
□	1.50	STEVENSITE (Smectite grp) $Mg_3Si_4O_{10}(OH)_2 \cdot xH_2O$	MCL mass	---	Buff, pink	H 2.5 G 2.15-2.57 infus	Dec by HCl. Biref 0 to .02.
1.465 ^	1.50	EVANSITE $Al_3PO_4(OH)_6 \cdot 6H_2O$	Amor	Conch	Cols, brown, yellow	H 3-4 G 1.9-2.1 infus	Diss by hot H_2SO_4 .
1.522 v	1.500	LAZURITE (Sodalite grp) $(Na,Ca)_{7-8}(Al,Si)_{12}(O,S)_{24}$ $[SO_4, Cl_2, (OH)_2]$	CUB dod	110 poor	Azure blue	H 6 G 2.4 F 3.5	Gel with acids, gives off H_2S .
1.51]	1.50	BOLIVARITE $Al_2(PO_4)(OH)_3 \cdot 4-5H_2O$	Amor	Splintery to conch	Greenish-yellow to green	H 2.5 G 2.05	---
	1.502	NASTROPHITE $Na(Sn,Ba)PO_4 \cdot 9H_2O$	CUB	Conch	Cols	H 2 G 2.05	Dec by H_2O , diss by HCl.
	1.504	NABAPHITE $NaBaPO_4 \cdot 9H_2O$	CUB	100 dist	Cols	H ~ 2 G 2.3	Dec by H_2O , diss by acids.
1.48 ^	1.505	VASHEGYITE $Al_4(PO_4)_3(OH)_3 \cdot 13H_2O$ (?)	ORTH (?) cryptocryst	---	White, green, yellowish	H 3.5 G 1.98 infus	Diss by acids.
1.493 ^ 1.507	1.505	ANALCIME, cesian (Zeolite grp) $(Na,Cs)AlSi_2O_6 \cdot H_2O$	CUB	100 poor	Cols, white	H 5 G 2.60 F 4.5	Diff dec by acids. Cs_2O 14.9%.
1.505 ^ 1.525	1.507	POLLUCITE (Zeolite grp) $(Cs,Na)AlSi_2O_6 \cdot H_2O$	CUB u mass	Uneven	Cols, white	H 6.5 G 2.85 F diff	Dec by acids.
1.550 v	1.508	TYCHITE $Na_6Mg_2(CO_3)_4(SO_4)$	CUB oct	Conch	Cols	H 3.5-4 G 2.46-2.59 F 1	Slightly sol in H_2O , diss by acids with eff. Compare Northupite.
	1.509	LEUCITE $KAlSi_2O_6$	TET ps cub, 211	110 poor	Cols	H 5.5 G 2.47 infus	Dec by acids. In part cryptocryst.

1.494 ^	1.509	HAUYNE (Sodalite grp) (Na,Ca) ₄₋₈ Al ₆ Si ₆ (O,S) ₂₄ (SO ₄ ,Cl ₂) ₂	CUB dod	110 poor	Blue, white	H 6 G 2.51 fus	Gel with acids. Na ₂ O 13.2, K ₂ O 2.8, CaO 10.1%.
1.44 ◇ 1.57	1.51	HISINGERITE Fe ⁺³ ₂ Si ₂ O ₅ (OH) ₄ ·2H ₂ O	MCL cryptocryst	Conch	Brownish-black	H 3 G 2.5-3.0 infus	Dec by acids. In part cryptocryst.
[]	1.51	MOTUKOREAITE Na ₂ Mg ₃₈ Al ₂₄ (CO ₃) ₁₃ (SO ₄) ₈ (OH) ₁₀₈ ·56H ₂ O	HEX cryptocryst	---	White to pale yellowish-green	H 1-1.5 G 1.48-1.53	---
	1.513	CADWALADERITE Al(OH) ₂ Cl·4H ₂ O	Amor (?)	Conch	Lemon-yellow	G 1.6	---
1.550 v	1.514	NORTHUPITE Na ₆ Mg ₂ (CO ₃) ₄ Cl ₂	CUB oct	Conch	Cols	H 3.5-4 G 2.38 F 1	Slightly sol in H ₂ O, diss by acids with eff. Tw common. Compare Tychite.
[]	1.517	PLANERITE (Ca,Cu)Al ₆ (PO ₄) ₄ (OH) ₈ ·4H ₂ O (?)	Cryptocryst	---	Green	H 5 G 2.65	Slightly sol in acids. In part biref.
[]	1.517	SEPIOLITE, var Meerschaum Mg ₄ Si ₆ O ₁₅ (OH) ₂ ·6H ₂ O	ORTH cryptocryst	---	White	H 1-2 G 2.0	Biref 0 to 0.02.
1.49 ^	1.52	ALLOPHANE Hydrous aluminum silicate	Amor	---	White, brown	H 3 G 1.9 infus	Gel with acids. Fe ₂ O ₃ 5-6%.
1.500 ^	1.522	LAZURITE (Sodalite grp) (Na,Ca) ₇₋₈ (Al,Si) ₁₂ (O,S) ₂₄ [SO ₄ ,Cl ₂ ·(OH) ₂]	CUB dod	110 poor	Azure blue	H 6 G 2.4 F 3.5	Gel with acids, gives off H ₂ S.
1.507 ^	1.525	POLLUCITE (Zeolite grp) (Cs,Na)AlSi ₂ O ₆ ·H ₂ O	CUB u mass	Uneven	Cols, white	H 6.5 G 2.90 F diff	Dec by acids.
1.50 ◇ 1.57	1.53	Glass andesitic comp	Amor	Conch	Cols, gray	H 5-6 G ~ 2.5	Insol in acids.
[]	1.53	KEHOEITE (Zn,Ca)Al ₂ (PO ₄) ₂ (OH) ₂ ·5H ₂ O	Amor (?) mass	---	White	G 2.34 infus	Diss by acids. Status of mineral in doubt.
	1.530	WISEITE NaCa ₅ Al ₁₀ (SiO ₄) ₃ (PO ₄) ₅ (OH) ₁₄ ·10H ₂ O (?)	CUB mass	---	White, bluish, yellowish	H 3-4 G 2.2 F easy	---
1.572 v	1.534	LANGBEINITE K ₂ Mg ₂ (SO ₄) ₃	CUB tetrah	Conch	Cols, yellow	H 3.5-4 G 2.83 F 2	Slowly sol in H ₂ O.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
□	1.535	SUCCINITE (Amber) a hydrocarbon	Amor	Conch	Amber-yellow	Soft G 1.07 F easy	---
1.530 □ 1.54	1.537	TACHARANITE $\text{Ca}_{12}(\text{Mg}, \text{Al})_2\text{Si}_{18}\text{O}_{51} \cdot 18\text{H}_2\text{O}$ (?)	MCL mass	001 good	White	G 2.36	Biref low.
1.47 △	1.54	NEOTOCITE $\text{MnO}, \text{SiO}_2, \text{H}_2\text{O}$	Amor	Conch	Brown to black	H 4 G 2.7 F diff	Dec by acids.
1.47 △	1.54	Hydrated basaltic glass ("Chlorophaeite")	Amor	Conch	Yellow, brown, orange-brown	H 1.5-3 G 1.8-2.2	Color may darken on exposure.
	1.541	AJKAITE a resin	Amor	---	Pale yellow to red-brown	H 2.5 G 1.0 F easy and burns	Gives off H_2S when heated. C 80, H 10, O 9, S 1%.
	1.542	TELEGDITE a resin	Amor	---	---	H 2.5 G 1.09	Partly diss by alcohol. C 76.9, H 10.2, O 11.2, S 1.7%.
▽ 1.555	1.542	HALLOYSITE $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	MCL u crypto-cryst	---	White, blue	H 2 G 2.6 infus	Insol in acids. When dried at 60°C, <u>n</u> increases to 1.555.
	1.544	HALITE NaCl	CUB	Cub, perf	Cols, yellow, red	H 2 G 2.17 F 1.5	Sol in H_2O , salty taste.
1.514 ^	1.550	NORTHUPITE $\text{Na}_6(\text{Mg}, \text{Fe})_2(\text{CO}_3)_4\text{Cl}_2$	CUB oct	Conch	Bright green	H 3.5-4 G 2.52 F 1	Slightly sol in H_2O , diss by acids with eff. FeO up to 7.6%.
1.508 ^	1.550	FERROTCHITE $\text{Na}_6(\text{Fe}^{+2}, \text{Mg}, \text{Mn})(\text{SO}_4)(\text{CO}_3)$	CUB	Conch	Cols to light yellow	H 4 G 2.79	---
1.542 ^	1.555	HALLOYSITE $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	MCL u crypto-cryst	---	White, blue	H 2 G 2.6 infus	Insol in acids.
▽ 1.570	1.558	SERPENTINE $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	MCL or ORTH cryptocryst	---	Green	H 4 G 2.5 F 6	Dec by acids. Biref low. Contains NiO 2.6%. Species not detd.

1.59 ✓	1.56	ZARATITE $\text{Ni}_3(\text{CO}_3)(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	CUB	Conch	Emerald-green	H 3-3.5 G 2.57-2.69 infus	Diss by acids with eff.
1.606 ✓	1.560	GUTSEVICHITE $(\text{Al}, \text{Fe})_3(\text{PO}_4, \text{VO}_4)_2(\text{OH})_3 \cdot 8\text{H}_2\text{O} (?)$	Crusts, concretions	---	Yellow-olive to dark green	H 2.5 G 1.9-2.0	Diss by acids.
1.59 ✓	1.563	Ni analogue of Serpentine ("Garnierite") $(\text{Ni}, \text{Mg})_3\text{Si}_2\text{O}_5(\text{OH})_4$	MCL (?) cryptocryst	---	Green	G 2.58 infus	Dec by acids. (= Pecoraite or Nepouite?)
1.592 ✓	1.563	ZUNYITE $\text{Al}_{13}\text{Si}_5\text{O}_{20}(\text{OH}, \text{F})_{18}\text{Cl}$	CUB tetrah	111	Cols	H 7 G 2.87 infus	Insol in acids. F 12.5, Cl 2.2, H_2O 6.5%.
1.676 ✓	1.565	ALUMOPHARMACOSIDERITE $\text{KAl}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6.5\text{H}_2\text{O}$	CUB	---	White	G 2.68	---
1.59 ✓	1.569	CARBONATE-FLUORAPATITE ("Collophane")(Apatite grp) $\text{Ca}_5(\text{PO}_4, \text{CO}_3)_3(\text{F}, \text{OH})$	HEX cryptocryst	---	White, brown	H 3.5 G ~ 2.6 F diff	Diss by acids.
1.51 ✓ 1.59	1.57	HISINGERITE $\text{Fe}^{+3}_2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL microcryst	Conch	Brownish-black	H 3.5 G 2.5-3.0 infus	Dec by acids.
1.53 ✓ 1.61	1.57	Glass andesitic to latitic comp	Amor	Conch	Pale brown, yellowish	H 5 G ~ 2.06	Insol in acids.
1.558 ^	1.570	SERPENTINE, zincian $(\text{Mg}, \text{Zn})_3\text{Si}_2\text{O}_5(\text{OH})_4$	MCL or ORTH cryptocryst	---	Green	H 4 G 2.5 F 6	Dec by acids. ZnO 7.25%.
1.60 ✓	1.57	BORICKITE Hydrous Ca ferric phosphate	Amor	---	Reddish-brown	H 3.5 G ~ 2.7 F easy	Diss by acids.
	1.571	NITROBARITE $\text{Ba}(\text{NO}_3)_2$	CUB oct	---	Cols	G 3.25 F 1	Sol in H_2O .
1.534 ^	1.572	MANGANOLANGBEINITE $\text{K}_2\text{Mn}_2(\text{SO}_4)_3$	CUB tetrah	---	Rose-red	G 3.02	Sol in H_2O .
	1.573	MILLOSEVICHITE $(\text{Al}, \text{Fe})_2(\text{SO}_4)_3$	---	---	Cherry-red	G 1.72	Diss by acids. Fe_2O_3 8.8%.
	1.577	NAMUWITE $(\text{Zn}, \text{Cu})_4(\text{SO}_4)(\text{OH})_6 \cdot 4\text{H}_2\text{O}$	HEX	0001 perf	Pale sea-green	H 2 G 2.77	ZnO 37.8, CuO 22%.
1.65 ✓	1.58	KONINCKITE $\text{FePO}_4 \cdot 3\text{H}_2\text{O}$	TET spherulitic	---	Yellow	H 3.5 G 2.3 F 3	Diss by hot acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
□	1.582	CHINGLUSUITE $\text{Na}_4\text{Mn}_5\text{Ti}_3\text{Si}_{14}\text{O}_{41} \cdot 9\text{H}_2\text{O}$	Amor	Uneven	Black	H 2-3 G 2.15 F easy	Gel with acids.
1.57 ◇ 1.66	1.59	HISINGERITE $\text{Fe}^{+3}_2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3 G 3.0 infus	Dec by acids.
1.56 ◇ 1.61	1.59	ZARATITE $\text{Ni}_3(\text{CO}_3)(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	CUB	Conch	Emerald-green	H 3-3.5 G 2.6 infus	Diss by acids with eff.
1.569 ◇ 1.63	1.59	CARBONATE-FLUORAPATITE ("Collophane") (Apatite grp) $\text{Ca}_5(\text{PO}_4, \text{CO}_3)_3(\text{F}, \text{OH})$	HEX cryptocryst	---	White, brown	H 3.5 G 2.6-2.7 F diff	Diss by acids.
1.563 △	1.59	Ni analogue of Serpentine ("Garnierite") $(\text{Ni}, \text{Mg})_3\text{Si}_2\text{O}_5(\text{OH})_4$	MCL (?) cryptocryst	---	Green	H 2 G ~ 2.6 infus	Dec by acids. (= Pecoraite or Nepouite?)
1.667 v	1.59	SCHOENFLIESITE $\text{MgSn}(\text{OH})_6$	CUB mass	---	Brown	G 3.48 infus	Diss by HCl, slowly by NaOH soln.
1.563 ◇ 1.600	1.592	ZUNYITE $\text{Al}_{13}\text{Si}_5\text{O}_{20}(\text{OH}, \text{F})_{18}\text{Cl}$	CUB tetrah	111	Cols	H 7 G 2.89 infus	Insol in acids. F 5.5, Cl 2.6, H_2O 10.0%.
1.608 v	1.593	VOLTAITE $\text{K}_2\text{Fe}^{+2}_5\text{Fe}^{+3}_4(\text{SO}_4)_{12} \cdot 18\text{H}_2\text{O}$	CUB	Conch	Dull oil-green, brown, black	H 3 G 2.6-2.8	Diss by acids. Oil-green in thin section.
	1.593	GEORGEITE $\text{Cu}_5(\text{CO}_3)_3(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	Amor	---	Light blue	Soft G 2.55	---
└	1.595	COESITE SiO_2	MCL ps hex	Subconch	Cols	G 2.92 infus	Insol in acids. Biref .003 - .005.
1.64 v	1.595	VUDYAVRITE Hydrous silicate of Ce, La, Ti	Amor	Conch	Brown, cream, yellow	H ~ 3 G 2.40-2.52	Dec by acids.
□	1.596	STURTITE $(\text{Mn}, \text{Mg})_6\text{Fe}^{+3}_2\text{Si}_8\text{O}_{25} \cdot 23\text{H}_2\text{O}$ (?)	Amor	Conch	Brownish-black	H 3 G 2.05 F diff	Dec by acids.
	1.598	COMBEITE $\text{Na}_2\text{Ca}_2\text{Si}_3\text{O}_9$	TRIG	---	Cols	G 2.84	Diss by hot dilute HCl. Biref very low.

1.57 ◇ 1.67	1.60	BORICKITE Hydrous Ca ferric phosphate	Amor	---	Reddish-brown	H 3.5 G ~ 2.7 F easy	Diss by acids.
√ 1.70	1.60	STIBICONITE (Stibiconite grp) $Sb^{+3}Sb^{+5}_2O_6(O,OH)$	CUB u mass	Conch	Gray, white, yellow	H 4-6 G 5.1-5.6 infus	Insol in acids. <u>n</u> highly variable.
1.592 △	1.600	ZUNYITE $Al_{13}Si_5O_{20}(OH,F)_{18}Cl$	CUB tetrah	111	Cols	H 7 G 2.88 infus	Insol in acids. F 0.5, Cl 3.5, H ₂ O 11.4%.
□	1.602	EUDIALYTE $Na_4(Ca,Ce)_2(Fe^{+2},Mn^{+2})ZrSi_8O_{22}$ (OH,Cl) ₂	TRIG	0001 dist	Yellow, pink, brown	H 5-5.5 G 2.9-3.1 F 3	Gel with acids. Uniax, pos or neg.
1.560 ^	1.606	GUTSEVICHITE $(Al,Fe)_3(PO_4,VO_4)_2(OH)_3 \cdot 8H_2O$ (?)	Crusts, concretions	---	Dark green to brown	H 2.5 G 1.9-2.0	Diss by acids.
□	1.608	CALCIUM FERRI-PHOSPHATE $CaFe^{+3}_2(PO_4)_2(OH)_2 \cdot 9H_2O$ (?)	Amor	---	Light brown	H 2-3	Diss by acids.
1.593 ^	1.608	VOLTAITE $K_2Fe^{+2}_5Fe^{+3}_4(SO_4)_{12} \cdot 18H_2O$	CUB	Conch	Dull oil- green, brown, black	H 3 G 2.6-2.8	Diss by acids. Oil-green in thin section.
1.57 ◇ 1.65	1.61	Glass basaltic comp	Amor	Conch	Brown	H 5 G 2.8	---
1.59 △	1.61	ZARATITE $Ni_3(CO_3)(OH)_4 \cdot 4H_2O$	CUB	Conch	Emerald-green	H 3 G ~ 2.6 infus	Diss by acids.
□	1.61	DIADOCHITE $Fe_2(PO_4)(SO_4)(OH) \cdot 5H_2O$	TCL u cryptocryst	Uneven to conch	Brown, yellow	H 3-4 G 2.0-2.4 F easy	Diss by HCl.
√ 1.68	1.61	ALLANITE (Epidote grp) $(Ca,Ce,Y)_2(Al,Fe)_3(SiO_4)_3(OH)$	MCL metamict	---	Brown, black	H 6 G 2.96 F 3	Gel with acids. In part biref.
	1.613	HSIANGHUALITE $Ca_3Li_2Be_3(SiO_4)_3F_2$	CUB dod	---	White	H 6.5 G 2.98	---
v 1.643	1.614	MAYENITE $Ca_{12}Al_{14}O_{33}$	CUB mass	---	White	G 2.85 infus	Insol in dil acids.
v 1.635	1.616	PITTICITE Hydrous ferric arsenate-sulfate	Amor	---	Brown, yellow	H 2-3 G 2.2-2.5 F easy	Diss by HCl. Comp widely variable.
	1.618	METABORITE HBO ₂	CUB tetrah	Conch	Cols to brownish	H 5 G 2.47 F 1	Slowly sol in H ₂ O.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[1.62	ZIRFESITE $Zr_3Fe_2Si_4O_{17} \cdot 15H_2O$ (?)	Amor	---	Pale yellow	G 2.70	Gel with acids.
	1.625	BICCHULITE $Ca_2Al_2SiO_6(OH)_2$	CUB	---	Cols	G 2.75 (synth)	---
	1.629	KAMAISHILITE $Ca_2Al_2SiO_6(OH)_2$	TET	---	Cols	---	---
1.59 ^	1.63	CARBONATE-FLUORAPATITE ("Collophane") (Apatite grp) $Ca_5(PO_4,CO_3)_3(F,OH)$	HEX cryptocryst	---	White, brown	H 3.5 G 2.7 F diff	Diss by acids.
1.66 v	1.63	GRIPHITE $Na_4Ca_6(Mn,Fe^{+2},Mg)_{19}Li_2Al_8$ $(PO_4)_{24}(F,OH)_8$	CUB u mass	Conch	Brown	H 5.5 G 3.4 F easy	Diss by HCl.
1.685 v	1.63	TRITOMITE-(Y) ("Spencite") $(Y,Ca,La,Fe^{+2})_5(Si,B,Al)_3$ $(O,OH,F)_{13}$ (?)	TRIG (?) metamict	---	Reddish-brown	H 3.5 G 3.1 F 3	---
	1.633	BURTITE $CaSn(OH)_6$	CUB	001 good	Cols	H ~ 3 G 3.28	Diss by HCl.
1.616 ^	1.635	PITTICITE Hydrous ferric arsenate-sulfate	Amor		Brown, yellow	H 2-3 G 2.2-2.5 F easy	Diss by HCl. Comp widely variable.
	1.639	SALAMMONIAC NH_4Cl	CUB	111 imperf	Cols	H 1.5-2 G 1.53 F 1 volat	Sol in H_2O .
1.653 v	1.639	HARKERITE $Ca_{24}Mg_8Al_2(SiO_4)_8(BO_3)_6(CO_3)_{10} \cdot 2H_2O$	CUB oct	---	Cols	H 4 G 2.95	Diss by HCl. Anom biref.
[1.639	TOMBARTHITE $Y_4(Si,H_4)_4O_{12-x}(OH)_{4+2x}$	MCL metamict	Conch	Brown, black	H 5-6 G 3.51-3.65	---
	1.63- 1.66	THOROSTEENSTRUPINE $(Ca,Th,Mn)_3Si_4O_{12} \cdot 6H_2O$	Amor metamict	Conch	Dark brown	H 4 G 3.02	ThO ₂ 35.7%. Weakly magnetic.
1.685 v	1.64	THORITE $ThSiO_4 (+ xH_2O)$	TET metamict	---	Brown to yellow	H 2-4 G 3.8-4.3	Gel with warm acids.

✓ 1.68	1.64	PICITE Hydrous ferric phosphate	Amor	---	Dark brown	H 3-4 G 2.8 F easy	Diss by acids.
□	1.64	HOMILITE (altered) $\text{Ca}_2(\text{Fe}^{+2}, \text{Mg})\text{B}_2\text{Si}_2\text{O}_{10}$	MCL mass	Conch	Black	H 3 G 3.35	Gel with acids.
	1.64	UMBOZERITE $\text{Na}_3\text{Sr}_4\text{ThSi}_8(\text{O}, \text{OH})_{24}$	Amor metamict	Conch	Bottle-green to greenish- brown	H ~ 5 G 3.60	---
✓ 1.74	1.64	COFFINITE $\text{U}(\text{SiO}_4)_{1-x}(\text{OH})_{4x}$	TET metamict	---	Yellow, brown	G 3.6-3.9	---
1.595 ^	1.64	VUDYAVRITE Hydrous silicate of Ce, La, Ti	Amor	Conch	Brown, cream, yellow	H ~ 3 G 2.40-2.52	Dec by acids.
□	1.642	SAKHAITE $\text{Ca}_3\text{Mg}(\text{BO}_3)_2\text{CO}_3 \cdot x\text{H}_2\text{O}$ ($x < 1$)	CUB	---	Gray, white	H 5 G 2.78-2.84	Diss by HCl with eff.
1.614 ^	1.643	MAYENITE $\text{Ca}_{12}\text{Al}_{14}\text{O}_{33}$	CUB mass	---	Cols	G 2.85 infus	Insol in dil acids.
1.58 ^	1.65	KONINCKITE $\text{FePO}_4 \cdot 3\text{H}_2\text{O}$	TET spherulitic	---	Yellow	H 3.5 G 2.3 F 3	Diss by hot acids.
□	1.65	EGUEIITE Hydrous Ca ferric phosphate	Amor	---	Yellow-brown	G 2.60 F easy	Diss by acids.
✓ 1.675	1.65	GREENALITE $(\text{Fe}^{+2}, \text{Fe}^{+3})_{2-3}\text{Si}_2\text{O}_5(\text{OH})_4$	MCL microcryst	---	Green to brown	G 2.8-3.0	Dec by HCl.
1.61 ^	1.65	Glass basaltic comp	Amor	Conch	Brown	H ~ 5 G ~ 3	---
✓ 1.77	1.65	MELANOCERITE $(\text{Ca}, \text{Ce})_5(\text{Si}, \text{B})_3\text{O}_{12}(\text{OH}, \text{F}) \cdot x\text{H}_2\text{O}$ (?)	HEX metamict	Conch	Dark brown to black	H 5-6 G 4.1 infus	Dec by HCl. Apatite structural type.
1.648 uniax pos ◇ uniax neg 1.653	1.65	MELILITE (Melilite grp) $(\text{Ca}, \text{Na})_2(\text{Mg}, \text{Al})(\text{Si}, \text{Al})_2\text{O}_7$	TET	001 dist	Cols to brown	H 5 G 3.0 F 3	Gel with acids. Comp close to Gehlenite 50, Akermanite 50. Biref .001.
1.639 ^	1.653	HARKERITE $\text{Ca}_{24}\text{Mg}_8\text{Al}_2(\text{SiO}_4)_8(\text{BO}_3)_6(\text{CO}_3)_{10} \cdot 2\text{H}_2\text{O}$	CUB oct	---	Cols	H 4 G 2.95	Diss by HCl. Anom biref.
✓ 1.71	1.655	ZIRCON $\text{ZrSiO}_4 (+ x\text{H}_2\text{O})$	TET metamict	---	Yellow, brown	G 2.9 infus	Partly diss by acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.63 △	1.66	GRIPHITE $\text{Na}_4\text{Ca}_6(\text{Mn,Fe,Mg})_{19}\text{Li}_2\text{Al}_8(\text{PO}_4)_{24}(\text{F,OH})_8$	CUB u mass	Conch	Brown	H 5.5 G 3.4 F easy	Diss by HCl.
1.59 △	1.66	HISINGERITE $\text{Fe}^{+3}\text{Si}_2\text{O}_5(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	MCL cryptocryst	Conch	Brownish-black	H 3 G 3.0 F diff to infus	Dec by acids.
[]	1.66	VESUVIANITE $\text{Ca}_{10}(\text{Mg,Fe})_2\text{Al}_4(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)_5(\text{OH,F})_4$	TET metamict	---	Brown	H 6 G 3.3 F 3	Insol in acids.
1.59 ◇ 1.705	1.667	SCHOENFLIESITE $(\text{Mg,Mn})\text{Sn}(\text{OH})_6$	CUB mass	---	Brown	G 3.49 infus	Diss by HCl.
1.60 ^	1.67	BORICKITE Hydrous Ca ferric phosphate	Amor	---	Reddish-brown	H 3.5 G ~ 2.7 F 3-4	Diss by acids.
v 1.702	1.670	HYDROGROSSULAR (Garnet grp) $\text{Ca}_3\text{Al}_2(\text{SiO}_4)_{3-x}(\text{OH})_{4x}$	CUB dod	---	Cols	H 6-6.5 G 3.1 infus	Gel with acids. Some samples divided into biref sectors. H_2O 8-9%.
1.65 ^	1.675	GREENALITE $(\text{Fe}^{+2},\text{Fe}^{+3})_{2-3}\text{Si}_2\text{O}_5(\text{OH})_4$	MCL microcryst	---	Green to brown	G 3.0	Dec by HCl.
1.565 ◇ 1.704	1.676	PHARMACOSIDERITE $\text{KFe}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	CUB cubes, tetrah	100 imperf	Brown, green	H 2.5-3 G 2.8-3.0 F 1	Diss by HCl. Biref 0 to 0.01.
1.64 △	1.68	PICITE Hydrous ferric phosphate	Amor	---	Dark brown	H 3-4 G 2.8 F easy	Diss by acids.
1.67 □ 1.71	1.680	HUTTONITE $\text{ThSiO}_4 (+ x\text{H}_2\text{O})$	MCL metamict	---	Cols to yellow	---	---
1.73 ▽	1.68	DELVAUXITE $\text{CaFe}^{+3}_3(\text{PO}_4,\text{SO}_4)_2(\text{OH})_8 \cdot 4-6\text{H}_2\text{O} (?)$	Amor (?)	---	Brown	H 2.5-4 G 2.0-2.8	Diss by acids.

1.61 ^ 1.72	1.68	ALLANITE (Epidote grp) $(Ca,Ce,Y)_2(Al,Fe)_3(SiO_4)_3(OH)$	MCL metamict	---	Brown, black	H 6 G 3.4-3.7 F 3	In part biref. May gel with acids.
	1.68	SATPAEVITE $Al_{12}V^{+4}_2V^{+5}_6O_{37} \cdot 30H_2O$	ORTH (?)	One pina- coidal, perf	Canary- to saffron-yellow	H 1.5 G 2.4	Diss by acids. In section greenish-yellow to olive. U biaxial, pos.
1.63 ^ 1.763	1.685	TRITOMITE-(Y) ("Spencite") $(Y,Ca,La,Fe^{+2})_5(Si,B,Al)_3$ $(O,OH,F)_{13}$	TRIG metamict	---	Reddish-brown	H 3.5 G 3.3 F 3	---
1.64 ^ 1.715	1.685	THORITE $ThSiO_4 (+ xH_2O)$	TET metamict	---	Brown, yellow	H 2-4 G 4.4	Gel with acids. H_2O 9%.
[]	1.694	RHODIZITE $(Cs,K)Al_4Be_4B_{11}O_{25}(OH)_4$	CUB dod	111, T11 diff	Cols	H 8 G 3.31-3.44 infus	Insol in acids. Opt anom.
[v] 1.78	1.70	GADOLINITE $(Y,Ca)_2FeBe_2Si_2O_{10}$	MCL metamict	Conch	Greenish-black	H 7 G 4.0 infus	Gel with acids. Pale green in section.
1.60 ^ 1.74	1.70	STIBICONITE (Stibiconite grp) $Sb^{+3}_3Sb^{+5}_2O_6(O,OH)$	CUB u mass	Conch	Gray, white, yellow	H 4-6 G 5.1-5.6 infus	Insol in acids. <u>n</u> highly variable.
1.670 ^ 1.734	1.702	HYDROGROSSULAR (Garnet grp) $Ca_3Al_2(SiO_4)_{3-x}(OH)_{4x}$	CUB dod	---	Cols	H 6-6.5 G 3.35 infus	Diff sol in acids. May gel.
1.676 ^ 1.712	1.704	PHARMACOSIDERITE $KFe_4(AsO_4)_3(OH)_4 \cdot 6H_2O$	CUB cubes, tetrah	100 imperf	Brown, green	H 2.5-3 G 2.8-3.0 F 1	Diss by acids. Biref 0 to 0.01.
v 1.726	1.704	ROWLANDITE $Y_3(SiO_4)_2(OH,F) (?)$	Amor metamict	Conch	Green, red	H 6-7 G 4.4-4.5 infus	Gel with acids.
1.667 ^	1.705	WICKMANITE $MnSn(OH)_6$	CUB oct	001 good	Brownish-to honey-yellow	H 3.5 G 3.89	---
v 1.748	1.707	BERZELIITE $(Ca,Na)_3(Mg,Mn)_2(AsO_4)_3$	CUB	Subconch to uneven	Yellow, orange	H 5 G 4.08 F 3	Diss by acids. MnO 1.3%. Garnet-type structure.
1.655 ^ 1.79	1.71	ZIRCON $ZrSiO_4 (+ xH_2O)$	TET metamict	---	Yellow, brown	G 3.0-3.5 infus	Nearly insol in acids.
1.704 ^	1.712	BARIUM-PHARMACOSIDERITE $Ba(Fe,Al)_4(AsO_4)_3(OH)_5 \cdot 5H_2O (?)$	TET (?)	100 good	Yellow-brown	H 2-3 G 3.00	Diss by acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
v 1.741	1.714	PYROPE (Garnet grp) $Mg_3Al_2(SiO_4)_3$	CUB 110, 211	---	Cols	H 7 G 3.58 F 4	Insol in acids. Data for synth end-member.
v 1.738	1.714	SPINEL (Spinel grp) $MgAl_2O_4$	CUB oct	111 imperf	Cols, red, green	H 8 G 3.58 infus	Insol in acids. Data for synth end-member.
1.685 v 1.804	1.715	THORITE (Th,U)SiO ₄ (+ xH ₂ O)	TET metamict	---	Brown, green, yellow	H 4 G 3.8-4.4	Gel with acids.
1.68 v 1.75	1.72	ALLANITE (Epidote grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	MCL metamict	---	Brown, black	H 6 G 3.4-3.7 F 3	In part biref. May gel with acids.
v 1.75	1.72	BRITHOLITE (Ce,Ca) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX metamict	---	Yellow, brown	H 5 G 3.85 infus	Gel with HCl. Yellow in section. Apatite structural type.
[v] 1.79	1.72	BINDHEIMITE (Stibiconite grp) Pb ₂ Sb ₂ O ₆ (O,OH)	CUB u mass	Conch	Yellow, brown, gray	H 4-4.5 G 4.6-5.6 F 3-4	Dec by HCl.
[]	1.72	CALCIOURANOITE (Ca,Ba,Pb)U ₂ O ₇ ·5H ₂ O	Amor	---	Brown to orange	H 4 G 4.62	Weakly anisotropic.
	1.725	DZHALINDITE In(OH) ₃	CUB	---	Yellow-brown	G 4.34 infus	Pale yellow in section.
1.704 ^	1.726	ROWLANDITE Y ₃ (SiO ₄) ₂ (F,OH) (?)	Amor metamict	Conch	Green, red	H 6-7 G 4.4-4.5 infus	Gel with acids.
1.68 ^	1.73	DELVAUXITE CaFe ⁺³ ₃ (PO ₄ ,SO ₄) ₂ (OH) ₈ ·4-6H ₂ O (?)	Amor (?)	---	Brown	H 2.5-4 G 2.0-2.8	Diss by acids.
v 1.738	1.732	HELVITE (Helvite grp) (Mn,Fe,Zn) ₄ Be ₃ (SiO ₄) ₃ S	CUB tetrah	111 imperf	Yellow	H 6 G 3.22 F 3	Gel with acids. MnO 45.5, FeO 2.2, ZnO 1.2%.
1.702 v 1.752	1.734	GROSSULAR (Garnet grp) Ca ₃ Al ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	White, green, pink, brown	H 6-7 G 3.59 F 3	Insol in acids. Opt anom. Data for synth end member.

1.756 v 1.756	1.735	PERICLASE MgO	CUB 100, 111	100 perf, parting 110	Cols	H 5.5 G 3.56 infus	Diss by acids.
	1.735	VISMIRNOVITE $\text{ZnSn}(\text{OH})_6$	CUB	---	Pale yellow	H 4 G 4.04	Diss by HCl.
	1.736	SÖHNGEITE $\text{Ga}(\text{OH})_3$	CUB	---	Light brown	H 4-4.5 G 3.84 infus	---
1.732 1.744 1.714 1.747 1.64 1.92	1.738	HELVITE (Helvite grp) $(\text{Mn, Fe, Zn})_4\text{Be}_3(\text{SiO}_4)_3\text{S}$	CUB tetrah	111 imperf	Brown	H 6 G 3.4 F 3	Gel with acids. MnO 29.9, FeO 10.1, ZnO 5.2%.
	1.738	SPINEL (Spinel grp) $(\text{Mg, Fe})\text{Al}_2\text{O}_4$	CUB oct	111 imperf	Green, black	H 8 G 3.62 infus	Insol in acids. FeO 5.6, Fe_2O_3 1.5, MnO 0.1%.
	1.74	COFFINITE $\text{U}(\text{SiO}_4)_{1-x}(\text{OH})_{4x}$	TET metamict	---	Yellow, brown	G 3.6-3.9	---
	1.74	CARYOCERITE Borosilicate of Y, Ce, Ca, Th	ps trig, metamict	Conch	Brown	H 5-6 G 4.30	Dec by hot HCl. Swells without fusing bb. Compare Melanocerite.
1.77	1.74	THOROGUMMITE $(\text{Th, U})(\text{Si, P})\text{O}_{4-x}(\text{OH})_{4x}$	TET metamict	---	Brown to black	H 3 G ~ 4.6 infus	Gel with acids. Gum-like. Compare Thorite.
1.70 1.80	1.74	STIBICONITE (Stibiconite grp) $\text{Sb}^{+3}\text{Sb}^{+5}_2\text{O}_6(\text{O, OH})$	CUB u mass	Conch	Gray, brown, yellow	H 4-5.5 G 5.2-5.6 infus	Insol in acids.
1.714 1.766	1.741	PYROPE-ALMANDINE (Garnet grp) $(\text{Mg, Fe})_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	---	Red, brown	H 7 G 3.72 F 3.5	Insol in acids. Py 66, Al 20, Sp 1, Gr 11, An 2.
1.738 1.745	1.744	GENTHELVITE (Helvite grp) $(\text{Zn, Fe})_4\text{Be}_3(\text{SiO}_4)_3\text{S}$	CUB tetrah	---	Brown, red	H 6 G 3.66 F 4	Gel with acids. ZnO 46.0, FeO 6.1, MnO 1.5%.
1.744 1.747	1.745	DANALITE (Helvite grp) $(\text{Fe, Zn, Mn})_4\text{Be}_3(\text{SiO}_4)_3\text{S}$	CUB tetrah	---	Pink, brown	H 6 G 3.44 F 3.5	Gel with acids. FeO 22.4, ZnO 20.0, MnO 8.15%.
1.738 1.776	1.747	SPINEL, zincian (Spinel grp) $(\text{Mg, Fe, Zn})\text{Al}_2\text{O}_4$	CUB oct	---	Greenish-black	H 8 G 3.97 infus	Insol in acids. ZnO 18.2, MgO 16.8, FeO 1.9%.
1.745 1.760	1.747	HELVITE (Helvite grp) $(\text{Mn, Fe, Zn})_4\text{Be}_3(\text{SiO}_4)_3\text{S}$	CUB tetrah	---	Brown	H 7 G 3.31 F 3.5	Gel with acids. MnO 28.0, FeO 15.8, ZnO 7.8%.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.707 ◇ 1.777	1.748	BERZELIITE (Ca,Na) ₃ (Mg,Mn) ₂ (AsO ₄) ₃	CUB	Subconch to uneven	Yellow, orange	H 5 G 4.27 F 3	Diss by acids. MnO 8.8%. Garnet-type structure.
□	1.75	YTTRIALITE (Y,Th) ₂ Si ₂ O ₇	HEX (?) metamict	Conch	Olive-green	H 6.5 G 4.3-4.7 infus	Gel with acids.
□	1.75	HELLANDITE (Ca,Y) ₆ (Al,Fe)Si ₄ B ₄ O ₂₀ (OH) ₄	MCL metamict	---	Brown	H 5.5 G 3.70 F 2-3	Gel with acids.
1.72 ◇ 1.81	1.75	BRITHOLITE (Ce,La,Ca) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX metamict	---	Yellow, brown	H 5 G 3.85 infus	Gel with HCl. Apatite-like structure.
1.72 △	1.75	ALLANITE (Epidote grp) (Ca,Ce,Y) ₂ (Al,Fe) ₃ (SiO ₄) ₃ (OH)	MCL metamict	---	Brown, black	H 6 G 3.4-3.7 F 3	In part biref. May gel with acids.
46 1.734 ◇ 1.773	1.752	GROSSULAR (Garnet grp) Ca ₃ (Al,Fe) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Brown, white, green	H 6-7 G 3.57 F 3	Insol in acids. Gr 85, An 13, Py 1.
	1.755	ARSENOLITE As ₂ O ₃	CUB oct	111	White	H 1.5 G 3.80 F 1	Sol in H ₂ O. Volat when heated.
	1.755	NATANITE Fe ⁺² Sn(OH) ₆	CUB	---	Greenish-brown	H 5 G 4.07	Diss by HCl.
1.735 ^	1.756	PERICLASE (Mg,Fe)O	CUB	100 perf parting 110	Cols	H 5.5 G 3.56 infus	Diss by acids. FeO 5.8%.
	1.758	AZOVSHITE Fe ₃ (PO ₄)(OH) ₆	Amor (?)	---	Dark brown	H 4 G 2.5	Diss by acids. Faintly biref.
1.747 ^	1.760	DANALITE (Helvite grp) (Fe,Mn,Zn) ₄ Be ₃ (SiO ₄) ₃ S	CUB tetrah	---	Dark red	H 6 G 3.41 F 3	Gel with acids. FeO 38.4, MnO 7.1, ZnO 6.8%.
1.685 ^	1.763	TRITOMITE (Ce,Y,La,Th) ₅ (Si,B) ₃ (O,OH,F) ₁₃	TRIG metamict	---	Reddish-brown	H 4-5.5 G 4.2 F 3	Gel with acids.

1.741 ⬇ 1.782 □	1.766	PYROPE-ALMANDINE (Garnet grp) $(\text{Mg,Fe,Ca})_3(\text{Al,Fe})_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Red, brown	H 7 G 3.88 F diff	Insol acids. Py 43, Al 41, Sp 2, Gr 8, An 6.
	1.768	RINGWOODITE $(\text{Mg,Fe})_2\text{SiO}_4$	CUB	---	Purple	G 3.90	Structure of spinel type.
1.65 ⬆	1.77	MELANOCERITE $(\text{Ca,Ce})_5(\text{Si,B})_3\text{O}_{12}(\text{OH,F}) \cdot x\text{H}_2\text{O} (?)$	HEX metamict	Conch	Dark brown to black	H 5-6 G 4.1 infus	Dec by HCl. Apatite structural type.
1.74 ⬆	1.77	THOROGUMMITE $\text{ThSiO}_{4-x}(\text{OH})_{4x}$	TET square prisms metamict	---	Black	H 5.5 G 5.4 infus	Diff sol in acids, gel. Clouded in section.
1.752 ⬇ 1.796	1.773	GROSSULAR-SPESSARTINE (Garnet grp) $(\text{Ca,Mn,Fe})_3\text{Al}_2(\text{SiO}_4)_3$	CUB	---	Peach-tan	H 7 G 3.80 F 4	Insol in acids. Gr 56, Sp 25, Al 11, An 7.
1.747 ⬇ 1.798	1.776	HERCYNITE, magnesian ferrian ("Pleonaste") (Spinel grp) $(\text{Fe,Mg})(\text{Al,Fe})_2\text{O}_4$	CUB oct	Uneven	Greenish-black	H 8 G 4.20 infus	Insol in acids. Deep green in section. FeO 23.3, MgO 10.8, Fe ₂ O ₃ 11.7%.
1.748 ⬆	1.777	MANGANBERZELIITE $\text{Ca}_3\text{Mn}_2(\text{AsO}_4)_3$	CUB	Subconch to uneven	Yellow, orange	H 5 G 4.45 F 3	Diss by acids. Data for synth end-member. Garnet-type structure.
	1.78	HAWLEYITE CdS	CUB	---	Bright yellow	G 4.87 infus	Dec by acids.
1.70 ⬇ 1.81	1.78	GADOLINITE $\text{Y}_2\text{FeBe}_2\text{Si}_2\text{O}_{10}$	MCL metamict	Conch	Black to greenish-black	H 6.5-7 G 3.9-4.3 infus	Gel with acids. Heated material is biref with $n = 1.82$.
1.766 ⬇ 1.797	1.782	ALMANDINE-PYROPE (Garnet grp) $(\text{Fe,Mg,Ca})_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Red to brown	H 7 G 4.02 F 4	Insol in acids. Al 55, Py 27, Gr 14, Sp 3.
1.776 ⬇ 1.805	1.782	GAHNITE (Spinel grp) $(\text{Zn,Fe,Mg})\text{Al}_2\text{O}_4$	CUB oct	111 imperf	Gray, green, black	H 8 G 4.38 infus	Insol in acids. ZnO 31.4, FeO 8.5, MgO 2.4%.
1.71 ⬇ 1.82	1.79	ZIRCON $\text{ZrSiO}_4 (+x \text{H}_2\text{O} (?))$	TET metamict	---	Yellow, brown	H 6 G 3.9-4.0 infus	Insol in acids.
1.72 ⬇ 1.84	1.79	BINDHEIMITE (Stibiconite grp) $\text{Pb}_2\text{Sb}_2\text{O}_6(0,\text{OH})$	CUB u mass	Conch	Yellow, brown, gray	H 4-4.5 G 4.6-5.6 F 3-4	Dec by HCl.
	1.790	PHILIPSBORNITE (Crandallite grp) $\text{PbAl}_3(\text{AsO}_4)_2(\text{OH})_3 \cdot \text{H}_2\text{O}$	TRIG	Conch	Grayish-green	H ~ 4.5 G (4.33)	---

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.773 ◇ 1.800	1.796	GROSSULAR-UVAROVITE (Garnet grp) $\text{Ca}_3(\text{Al}, \text{Cr}, \text{Fe})_2(\text{SiO}_4)_3$	CUB dod	Uneven	Green	H 7 G 3.68 F 4	Insol in acids. Gr 52, Uv 40, An 8.
1.782 ◇ 1.805	1.797	ALMANDINE (Garnet grp) $(\text{Fe}, \text{Mg}, \text{Mn}, \text{Ca})_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Reddish-brown	H 7 G 4.10 F 4	Insol in acids. Al 60, Py 18, Sp 15, Gr 7.
1.776 ◇ 1.833	1.798	SPINEL, chromian (Spinel grp) $(\text{Mg}, \text{Fe})(\text{Al}, \text{Cr}, \text{Fe})_2\text{O}_4$	CUB oct	Uneven	Black	H 8 G 3.94 infus	Insol in acids. MgO 19.5, FeO 9.8, Al_2O_3 48.3, Cr_2O_3 19.6%.
1.74 ◇ 1.90	1.80	STIBICONITE (Stibiconite grp) $\text{Sb}^{+3}\text{Sb}^{+5}_2\text{O}_6(\text{O}, \text{OH})$	CUB u mass	Conch	Gray, brown, yellow	H 4-5.5 G 5.2-5.6 infus	Insol in acids.
	>1.80	MANGANOSTEENSTRUPINE $(\text{Ce}, \text{La}, \text{Th}, \text{Ca})\text{MnSiO}_3(\text{OH})_2 \cdot 2\text{H}_2\text{O}$	Amor	Conch	Black	H 5.5-6 G 3.29	Brownish-red in section.
48 1.796 ◇ 1.801	1.800	SPESSARTINE (Garnet grp) $\text{Mn}_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Red to brown	H 7 G 4.18 F 3	Insol in acids. Synth end-member.
	1.800	YAFSOANITE $(\text{Zn}, \text{Ca}, \text{Pb})_3\text{TeO}_6$	CUB	---	Light to dark brown	---	Diss by acids.
1.800 ◇ 1.805	1.801	GROSSULAR-ANDRADITE (Garnet grp) $(\text{Ca}, \text{Fe}, \text{Mn})_3(\text{Al}, \text{Fe})_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Green, red, brown	H 7 G 3.71 F 4.5	Insol in acids. Opt anom. Gr 52, An 42, Al 4, Sp 2.
1.766 ◇ 1.855	1.803	KNORRINGITE (Garnet grp) $(\text{Mg}, \text{Ca}, \text{Fe})_3(\text{Cr}, \text{Al})_2(\text{SiO}_4)_3$	CUB u mass	Uneven	Blue-green	H 7 G 3.76	Insol in acids. Kn 53, G 19, Al 14, Py 10, An 3.
1.715 △	1.804	THORITE $(\text{Th}, \text{U})\text{SiO}_4 (+ x\text{H}_2\text{O})$	TET metamict	---	Brown, green, yellow	H 4 G 4.8-6.4	Gel with acids.
1.782 ◇ 1.818	1.805	GAHNITE (Spinel grp) ZnAl_2O_4	CUB oct	---	Black	H 8 G 4.61 infus	Insol in acids. Synth end-member.
1.800 ◇ 1.814	1.805	SPESSARTINE (Garnet grp) $(\text{Mn}, \text{Fe})_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Red, brown	H 7 G 4.20 F 3.5	Insol in acids. Sp 80, Al 17, Py 2.
1.797 ◇ 1.814	1.805	ALMANDINE (Garnet grp) $(\text{Fe}, \text{Mg}, \text{Ca})_3\text{Al}_2(\text{SiO}_4)_3$	CUB 110, 211	Uneven	Red, brown	H 7 G 4.15 F 4	Insol in acids. Al 74, Py 11, Gr 11, Sp 4.

1.78 △	1.81	GADOLINITE $Y_2FeBe_2Si_2O_{10}$	MCL metamict	Conch	Black to greenish-black	H 6.5-7 G 3.9-4.3 infus	Gel with acids.
1.75 ^	1.81	BRITHOLITE $(Ce,La,Ca)_5(SiO_4,PO_4)_3(OH,F)$	HEX metamict	---	Yellow, brown	H 5 G 4.1 infus	Gel with HCl. Apatite-like structure.
1.82 □	1.81	ANTHOINITE $Al_2W_2O_9 \cdot 3H_2O$	MCL (?) mass	---	White, chalky	H 1 G 4.6	Slowly diss by acids, diss by strong KOH soln.
1.805 ◇ 1.818	1.814	SPESSARTINE-ALMANDINE (Garnet grp) $(Mn,Fe)_3Al_2(SiO_4)_3$	CUB 110, 211	Uneven	Red, brown	H 7 G 4.23 F 3.5	Insol in acids. Sp 53, Al 45, Py 1, Gr 1.
1.87 v	1.817	ROMEITE (Stibiconite grp) $(Ca,Na)_2Sb_2O_6(O,OH)$	CUB oct	Conch	Yellow	H 5.5-6.5 G 4.7-5.4 F diff	Insol in acids. Opt anom with low biref.
1.805 ^	1.818	GAHNITE (Spinel grp) $(Zn,Fe)(Al,Fe)_2O_4$	CUB oct	---	Black	H 8 G 4.60 infus	Insol in acids. FeO 1.7, Fe ₂ O ₃ 2.55%.
1.805 ◇ 1.830	1.818	ALMANDINE (Garnet grp) $(Fe,Mn)_3Al_2(SiO_4)_3$	CUB 110, 211	Uneven	Red, brown	H 7 G 4.25 F 4	Insol in acids. Al 74, Sp 20, Py 5, Gr 11.
1.79 △	1.82	ZIRCON, var Malacon (Naegite) $(Zr,Th,U)SiO_4 (+xH_2O)$	TET metamict	Conch	Brown, green	H 3-7 G 4.0-4.3 infus	Partly diss by acids.
1.801 ◇ 1.834	1.821	GOLDMANITE (Garnet grp) $Ca_3(V,Al,Fe)_2(SiO_4)_3$	CUB dod mass	---	Deep green to dull green	G 3.74	Weakly biref. V ₂ O ₃ 18.3, Fe ₂ O ₃ 5.4, Al ₂ O ₃ 4.9%.
1.801 ◇ 1.855	1.825	UVAROVITE-GROSSULAR (Garnet grp) $Ca_3(Cr,Al)_2(SiO_4)_3$	CUB dod	---	Green	H 7.5 G 3.81 infus	Insol in acids. n is for green light. Uv 72, Gr 26, An 1, Py 1.
1.801 ◇ 1.863	1.827	ANDRADITE-GROSSULAR (Garnet grp) $Ca_3(Fe,Al)_2(SiO_4)_3$	CUB 110, 211	Uneven	Red, brown, green	H 7 G 3.77 F 4	Insol in acids. An 57, Gr 32, Al 5, Py 3, Sp 3.
	1.83	LIANDRATITE $U^{+6}(Nb,Ta)_2O_8$	HEX u metamict	Conch	Yellow to yellow-brown	H 3.5 G 6.8	---
1.818 ^	1.830	ALMANDINE (Garnet grp) $Fe_3Al_2(SiO_4)_3$	CUB	Uneven	Red	H 7 G 4.32 F 3	Insol in acids. Synth end- member.
□	1.83	YTTROPYROCHLORE ("Obruchevite") (Pyrochlore grp) $(Y,Ca,Na,U)_{1-2}(Nb,Ta,Ti)_2(O,OH)_7$	CUB	Conch	Dark brown	H 4.5-5 G ~ 3.7	Insol in acids.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
1.798 ∇ 1.835	1.833	SPINEL, chromian (Spinel grp) (Mg,Fe)(Al,Cr,Fe) ₂ O ₄	CUB oct	Uneven	Black	H 7 G 3.93 infus	Insol in acids. MgO 19.3, FeO 11.7, Al ₂ O ₃ 45.4, Cr ₂ O ₃ 16.85, Fe ₂ O ₃ 5.8%.
1.821 ∇ 1.855	1.834	GOLDMANITE (Garnet grp) Ca ₃ V ₂ (SiO ₄) ₃	CUB	---	Green	G 3.765	Data for synth compd.
1.833 ∇ 1.90	1.835	HERCYNITE (Spinel grp) FeAl ₂ O ₄	CUB oct	Uneven	Greenish-black	H 8 G 4.40 infus	Insol in acids. Data calc for end-member.
	1.838	LIME CaO	CUB	100 perf	Cols	H 3-4 G 3.32 infus	Diss by acids, slightly sol in H ₂ O. Rapidly alters on exposure to moist air.
1.79 ∇ 1.94	1.84	BINDHEIMITE (Stibconite grp) Pb ₂ Sb ₂ O ₆ (OH)	CUB u mass	Conch	Yellow, gray, green	H 4-4.5 G 4.6-5.6 F 3-4	Dec by HCl.
1.825 ^	1.855	UVAROVITE (Garnet grp) Ca ₃ (Cr,Al) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Deep green	H 7.5 G 3.77 infus	Insol in acids. Opt anom. Uv 91, Gr 6, Py 2, An 1.
1.834 ^	1.855	GOLDMANITE (Garnet grp) (Ca,Mn) ₃ V ₂ (SiO ₄) ₃	CUB	---	Dark green	G 3.91	V ₂ O ₃ 24.9, Al ₂ O ₃ 2.0, Fe ₂ O ₃ 1.1, MnO 15.9%.
1.923 v	1.856	GALAXITE (Spinel grp) (Mn,Fe)(Al,Fe) ₂ O ₄	CUB oct	Conch	Black	H 7.5 infus	Insol in acids. MnO 33.1, FeO 2.3, Fe ₂ O ₃ 6.4%.
┌	>1.86	ARSENOBISMITE Bi ₂ (AsO ₄)(OH) ₃	CUB (?) ocherous masses	---	Yellow-brown to green	G ~ 5.7	---
1.827 ∇ 1.887	1.863	ANDRADITE (Garnet grp) (Ca,Fe,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃	CUB 110, 211	Uneven	Red, brown, green	H 7 G 3.72 F 3.5	Gel imperfectly with HCl. Opt anom. An 78.5, Gr 12, Al 5.5, Sp 4.
1.817 ∇ 2.09	1.87	ROMEITE (Stibiconite grp) (Ca,Mn) ₂ Sb ₂ O ₆ (OH)	CUB oct	Conch	Yellow	H 5.5-6.5 G 4.7-5.4 F 4.5	Insol in acids. Opt anom. MnO 6.3%.
1.915 v	1.87	PYROCHLORE (Pyrochlore grp) (Na,Ca) ₂ (Nb,Ta) ₂ O ₆ (OH,F)	CUB oct	Conch	Brown, yellow	H 5-6 G 3.77 infus	Dec by H ₂ SO ₄ .

1.88 2.00	1.88	CHEVKINITE (Ca,Ce,Th) ₄ (Fe,Mg) ₂ (Ti,Fe) ₃ Si ₄ O ₂₂	MCL metamict	Conch	Black to reddish-brown	H 5 G 4.3-4.6 F 4	Gel with acids. Red-brown in section. In part biref.
1.863 1.890	1.887	ANDRADITE (Garnet grp) Ca ₃ Fe ₂ (SiO ₄) ₃	CUB dod	---	Red	H 7 G 3.86 F 3.5	Gel imperfectly with acids. Data for synth end-member.
1.887 1.893	1.890	CALDERITE-ANDRADITE (Garnet grp) (Mn ⁺² ,Ca) ₃ Fe ⁺³ ₂ (SiO ₄) ₃	CUB	Conch	Red-brown	H 7 G 4.07	Fe ₂ O ₃ 29.0, MnO 19.3%. (Ca 47, An 45, Gr 8%). (Min. Abs. <u>32</u> , 310.)
1.890 1.94	1.893	ANDRADITE-SPESSARTINE (Garnet grp) (Ca,Mn) ₃ (Fe,Al) ₂ (SiO ₄) ₃	CUB 110, 211	---	Dark green	H 7 G 3.98 F 3.5	Nearly insol in acids. An 71, Sp 28, Al 1.
1.80 2.05	1.90	STIBICONITE (Stibiconite grp) Sb ⁺³ Sb ⁺⁵ ₂ O ₆ (OH)	CUB u mass	Conch	Yellow, gray, brown	H 4.5-5.5 G 5.2-5.6 infus	Insol in acids.
1.835 1.97	1.90	MAGNESIOCHROMITE (Spinel grp) MgCrO ₄	CUB oct	Uneven	Gray-green	H 7 G 4.39 infus	Insol in acids. Data for synth end-member.
1.87 1.925	1.915	BETAFITE (Pyrochlore grp) (Ca,Na,U) ₂ (Ti,Nb,Ta) ₂ O ₆ (OH)	CUB oct	Conch	Greenish-black	H 5 G 4.0 infus	Dec by H ₂ SO ₄ .
1.74 1.856	1.92	COFFINITE U(SiO ₄) _{1-x} (OH) _{4x}	TET metamict	---	Yellow, brown	G 4.2-4.8	---
1.915 1.94	1.923	GALAXITE (Spinel grp) (Mn,Fe)Al ₂ O ₄	CUB oct	Conch	Black	H 7.5 G 4.23 infus	Insol in acids. Mahogany-red in section. MnO 34.0, FeO 16.4, MgO 1.5%.
	1.925	MICROLITE (Pyrochlore grp) (Ca,Na) ₂ (Ta,Nb) ₂ O ₆ (OH,F)	CUB oct	Conch	Yellow, brown	H 5.5 G ~ 5.5 infus	Dec by H ₂ SO ₄ .
	1.93	NANTOKITE CuCl	CUB	011	Cols	H 2.5 G 4.1 fus	Sol in H ₂ O. Oxidizes rapidly. Data on Synth compound.
1.985	1.93	KALIPYROCHLORE (Pyrochlore grp) (K,Sr) ₂₋₄ Nb ₂ O ₆ (OH) ₂ ·xH ₂ O	CUB	---	Yellow, greenish	H 4-4.5 G 3.42	---
1.84 1.98	1.94	BINDHEIMITE (Stibiconite grp) Pb ₂ Sb ₂ O ₆ (OH)	CUB u mass	Conch	Gray, brown, green	H 4-4.5 G 5.0-5.6 F 3-4	Dec by HCl.
	1.94	ANDRADITE (Garnet grp) Ca ₃ (Fe,Ti) ₂ (SiO ₄) ₃	CUB 110, 211	Conch	Black	H 7 G 3.79 F 4	Gel with acids. TiO ₂ 9.4%.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	1.94	MELANITE $\text{Ca}_3(\text{Fe}^{+3}, \text{Ti})_2(\text{SiO}_4)_3$	---	---	---	G 3.79	TiO_2 9.4%.
[1.94	KIMZEYITE (Garnet grp) $\text{Ca}_3(\text{Zr}, \text{Ti})_2(\text{Al}_2\text{Si})\text{O}_{12}$	CUB	---	Brown	H 7 G 3.94	ZrO_2 29.9, TiO_2 5.6%.
1.925 ◇ 1.97	1.94	BETAFITE (Pyrochlore grp) $(\text{Ca}, \text{Na}, \text{U})_2(\text{Ti}, \text{Nb}, \text{Ta})_2\text{O}_6(\text{OH}, \text{F})$	CUB oct	Conch	Black, brown, yellow	H 5 G 5.2 infus	Dec by H_2SO_4 .
v 2.11	1.95	BRANNERITE $(\text{U}, \text{Th}, \text{Ce})\text{Ti}_2\text{O}_6$	MCL metamict	Conch	Black, brown, yellow	H 5-6 G 4.0-4.3 infus	Diss by acids.
	1.95	STETEFELDTITE (Stibiconite grp) $\text{Ag}_2\text{Sb}_2(\text{O}, \text{OH})_7 (?)$	CUB mass	---	Black, brown	H 3.5-4.5 G 4.1-4.6	---
1.90 ◇ 2.10	1.97	MAGNESIOCHROMITE (Spinel grp) $(\text{Mg}, \text{Fe})(\text{Cr}, \text{Al}, \text{Fe})_2\text{O}_4$	CUB oct	Uneven	Black	H 6 G 4.67 infus	Insol in acids. MgO 16.0, FeO 11.1, Cr_2O_3 43.6, Al_2O_3 24.4, Fe_2O_3 4%.
1.94 ◇ 2.00	1.97	URANMICROLITE ("Djalmaite") (Pyrochlore grp) $(\text{Ca}, \text{Na}, \text{U})_2(\text{Ta}, \text{Nb}, \text{Ti})_2\text{O}_6(\text{OH})$	CUB oct	Uneven	Brown, black	H 5.5 G 5.8	Dec by H_2SO_4 .
] 1.98	1.97	MOLURANITE $\text{H}_4\text{U}^{+4}(\text{UO}_2)_3(\text{MoO}_4)_7 \cdot 18\text{H}_2\text{O}$	Amor	Conch	Black	H 3-4 G ~ 4	Diss by acids.
[1.98	URANPYROCHLORE ("Hatchettolite") (Pyrochlore grp) $(\text{U}, \text{Ca}, \text{Ce})_2(\text{Nb}, \text{Ta})_2\text{O}_6(\text{OH}, \text{F})$	CUB oct	Conch	Brown	H 5 G ~ 4.8 infus	Insol in acids.
1.94 ^	1.98	SCHORLOMITE (Garnet grp) $\text{Ca}_3(\text{Fe}, \text{Ti})_2(\text{Si}, \text{Ti})_3\text{O}_{12}$	CUB 110, 211	Conch	Black	H 7 G 3.85 F 4	Gel with acids. Opt anom. TiO_2 16.9%.
	1.983	MAGNUSSONITE $\text{Mn}_5(\text{AsO}_3)_3(\text{OH}, \text{Cl})$	CUB and TET	---	Grass- to emerald-green	H 3.5-4.0 G 4.30	---
1.93 ^	1.985	KALIPYROCHLORE (Pyrochlore grp) $(\text{K}, \text{Sr})\text{Nb}_2\text{O}_6(\text{O}, \text{OH}) \cdot x\text{H}_2\text{O}$	CUB	---	Greenish	H 4-4.5 G 3.48	---
1.88 ^	2.00	CHEVKINITE $(\text{Ca}, \text{Ce}, \text{Th})_4(\text{Fe}, \text{Mg})_2(\text{Ti}, \text{Fe})_3\text{Si}_4\text{O}_{22}$	MCL metamict	Conch	Black	H 5 G 4.3-4.7 F 4	Gel with acids. Red-brown in section. In part biref.

1.97 ^ 2.02 []	2.00	PYROCHLORE (Pyrochlore grp) (Ca,Na) ₂ (Nb,Ta) ₂	CUB oct	111 fr conch	Brown, red	H 5 G 4.1-4.3 infus	Dec by H ₂ SO ₄ . Heated material had <u>n</u> 2.23.
	>2.00	BISMUTOMICROLITE ("Westgrenite") (Pyrochlore grp) (Bi,Ca)(Ta,Nb) ₂ O ₆ (OH)	CUB	Uneven	Yellow, pink, brown	H 5 G 6.5-7.2 infus	Insol in acids.
	>2.0	CAFARSITE Ca ₈ (Ti,Fe ⁺² ,Fe ⁺³ ,Mn) ₆₋₇ (As ⁺³ O ₃) ₁₂ ·4H ₂ O	CUB	Conch	Dark brown	H 5.5-6 G 3.90	---
2.00 ^ 2.023	2.02	BETAFITE (Pyrochlore grp) (Ca,Na,U) ₂ (Ti,Nb,Ta) ₂ O ₆ (OH,F)	CUB oct	Conch	Black	H 4.5 G 4.76 infus	Dec by H ₂ SO ₄ .
2.02 ^ 2.07	2.023	MICROLITE (Pyrochlore grp) (Ca,Na) ₂ Ta ₂ O ₆ (O,OH,F)	CUB oct	Conch	Brown, yellow	H 6 G 6.42 infus	---
	2.04	CHOLALITE PbCu(TeO ₃) ₂ ·H ₂ O	CUB	Uneven	Deep green	H 3 G 6.4	---
1.90 ^ []	2.05	STIBICONITE (Stibiconite grp) Sb ⁺³ Sb ⁺⁵ O ₆ (O,OH)	CUB u mass	Conch	Yellow, brown	H 4-4.5 G 5.2-5.6 infus	Insol in acids.
	2.05	GOETHITE ("Limonite") FeO(OH)	ORTH cryptocryst	---	Ocher-yellow	H 4 G ~ 3.8 infus	Diss by HCl.
	2.05	PERCYLITE PbCuCl ₂ (OH) ₂	CUB	100	Sky-blue	H 2.5 G 5.25 F 1	Diss by HNO ₃ . Sky-blue in section. A doubtful mineral.
	2.05	EULYTITE Bi ₄ Si ₃ O ₁₂	CUB tetrah	110 imperf	Brown, yellow, gray	H 4.5 G 6.11 F 2	Gel with acids. Opt anom.
2.19 ^	2.05	FERGUSONITE (Y,U)(Nb,Ta) ₄ O ₄	TET metamict	Conch	Dark brown	H 5 G 4.2-5.8 infus	Diss by hot concd H ₂ SO ₄ . Reddish-brown in section.
2.18 ^	2.06	ZIRKELITE (Ca,Th,Ce)Zr(Ti,Nb) ₂ O ₇	MCL ps cub	Uneven to conch	Brown to black	H 6 G 4.02 infus	Dec by hot HCl.
	2.06	LORANSKITE (Y,Ce,Ca)ZrTaO ₆ (?)	ORTH (?) metamict	Conch	Black, brown yellow	H 5 G 4.2-4.6	Status doubtful, perhaps = Euxenite.
2.24 ^ []	2.06	EUXENITE (Y,Ca,Ce,U,Th)(Nb,Ta,Ti) ₂ O ₆	ORTH metamict	Conch	Brownish-black	H 6.5 G 4.6-5.5 infus	Insol in acids. Reddish-brown in section. After ignition, <u>n</u> = 2.22.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[]	2.065	MOSESITE $\text{Hg}_2\text{N}(\text{Cl}, \text{SO}_4, \text{MoO}_4) \cdot \text{H}_2\text{O}$	CUB oct	Uneven	Lemon- to canary-yellow	H 3 G 7.72 volat	In part anom biref.
2.023 ∧ 2.18	2.07	MICROLITE (Pyrochlore grp) $(\text{Ca}, \text{Na})_2\text{Ta}_2\text{O}_6(0, \text{OH}, \text{F})$	CUB oct	Conch	Brown, yellow	H 6 G 5.95 infus	Insol in acids.
2.15 ∨	2.071	CHLORARGYRITE $\text{Ag}(\text{Cl}, \text{Br})$	CUB u mass	Sectile	Gray, yellowish	H 2.5 G 5.56 F 1	Diss by NH_4OH .
2.07 [] 2.11	2.08	BARIOPYROCHLORE ("Pandaite") (Pyrochlore grp) $(\text{Ba}, \text{Sr})_2(\text{Nb}, \text{Ti})_2(0, \text{OH})_7$	CUB	111 poor fr conch	Yellow, gray	H 4-4.5 G 3.7-4.0 infus	Insol in HCl.
[]	2.08	PLUMBOPYROCHLORE (Pyrochlore grp) $(\text{Pb}, \text{Y}, \text{U}, \text{Ca})_2\text{Nb}_2\text{O}_6(\text{OH})$	CUB	---	Greenish-yellow to brown	G 6.34	---
	2.085	GIANELLAITE $\text{Hg}_4\text{N}_2(\text{SO}_4)$	CUB	---	Straw-yellow	H ~ 3 G 7.19 volat	Dec by concd HCl.
[]	2.087	SENARMONTITE Sb_2O_3	CUB oct	111 poor	Cols	H 2 G 5.50 F 1.5 volat	Diss by HCl. Anom biref.
	2.09	BISMUTOSTIBICONITE (Stibiconite grp) $(\text{Bi}, \text{Fe})\text{Sb}_2\text{O}_7$	CUB	---	Yellow to yellow-brown	G 7.38	---
1.87 ∧ 2.20	2.09	ROMEITE, var Schneebergite (Stibiconite grp) $(\text{Ca}, \text{Fe}, \text{Mn}, \text{Na})_2(\text{Sb}, \text{Ti})_2\text{O}_6(0, \text{OH}, \text{F})$	CUB oct	111 dist	Honey-yellow	H 6.5 G 5.41 F diff	Insol in acids. Opt anom, low biref.
2.05 [] 2.20	~2.1	FERRITUNGSTITE $(\text{W}, \text{Fe}^{+3})(0, \text{OH})_3 (?)$	CUB	---	Yellow	G 4.4-5.2	Dec by HCl.
1.97 ∧ 2.12	2.10	CHROMITE (Spinel grp) $(\text{Mn}, \text{Fe}, \text{Mg})(\text{Cr}, \text{Al})_2\text{O}_4$	CUB oct	Uneven	Brownish-black	H 6 G 4.65 infus	Insol in acids. Cr_2O_3 49.7, Al_2O_3 17.6, Fe_2O_3 8.1, MnO 9.3, FeO 8.7, MgO 2.4, ZnO 4.1%.
2.20 ∨	2.10	THORIANITE $(\text{Th}, \text{U})\text{O}_2$	CUB	001 poor fr uneven	Black to brown	H 6 G 9.2 infus	Slowly diss by HNO_3 or H_2SO_4 .

□	2.10	THORUTITE (Th,U)Ti ₂ O ₆	MCL metamict	Conch	Brownish-black	H 5-6 G 5.82 infus	Compare Brannerite.
√ 2.38	2.105	LOPARITE (Perovskite grp) (Ce,Na,Ca) ₂ (Ti,Nb) ₂ O ₆	ORTH metamict	---	Reddish-brown	H 5-5.5 G 4.48 infus	Insol in acids.
1.95 ∧ 2.49	2.11	BRANNERITE (U,Th,Ce)Ti ₂ O ₆	MCL metamict	Conch	Black	H 4.5 G 4.2-4.6 infus	Diss with diff by hot HNO ₃ .
√ 2.3	2.11	DAVIDITE (La,Ce)(Y,U,Fe ⁺²)(Ti,Fe ⁺³) ₂₀ (O,OH) ₃₈	TRIG metamict	---	Brownish-black	H 6-7 G 4.4-4.6 infus	Dec by hot H ₂ SO ₄ .
	>2.11	CLIFFORDITE UTe ₃ O ₉	CUB oct	---	Bright yellow	H 4 G 6.6 (synth)	Diss by concd HCl.
2.10 ∧	2.12	CHROMITE (Spinel grp) FeCr ₂ O ₄	CUB oct	Uneven	Black	H 6 G 5.22 infus	Insol in acids. Data for synth compd.
√ 2.26	2.12	BISMUTITE Bi ₂ (CO ₃)O ₂	TET u earthy mass	001 dist	Yellow, brown	H 2.5-4 G 6.7-7.4 F 1.5	Diss by acids with eff.
	2.13 Na	BORNITE Cu ₅ FeS ₄	CUB u mass	Uneven	Red-brown, tarnishes purple	H 3 G 5.07 F 2	Dec by HNO ₃ .
2.11 □	2.13	MURATAITE (Na,Y) ₄ (Zn,Fe) ₃ (Ti,Nb) ₆ O ₁₈ (F,OH) ₄	CUB	Conch	Black	G 4.50-4.69	Insol in acids except HF.
√ 2.25	2.13	SAMARSKITE (Y,U,Ca,Fe)(Nb,Ta,Ti) ₂ O ₆	MCL metamict	Conch	Black	H 4 G 4.0-4.3 F 4.5	Diss by hot concd acids. Reddish-brown in section.
2.12 □ 2.15	2.13	YTTROCRASITE (Y,Th,U,Ca)Ti ₂ (O,OH) ₆	ORTH metamict	Conch	Black	H 4-6 G 4.80 infus	Diss by hot H ₂ SO ₄ . Amber in section. In part birefringent.
	2.14	OLDHAMITE (Ca,Mn)S	CUB	100	Pale brown	H 4 G 2.61 infus	Diss by acids. Rapidly dec in air. Meteorite mineral.
	2.14	FORMANITE (Y,U)(Ta,Nb) ₂ O ₄	TET metamict	Conch	Black	H 5.5-6.5 G 6.17 infus	Diss by hot concd H ₂ SO ₄ .
2.12 □ 2.18	2.14	AESCHYNITE-(Y), var Priorite (Y,Ce,Ca,Th)(Ti,Nb) ₂ O ₆	ORTH metamict	Conch	Brown to black	H 5-6 G 4.7-5.0 infus	Diss by hot concd acids.

Table 3. Isotropic minerals and mineraloids (continued)

56

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[]	2.15	YTTROTANTALITE (Y,U,Fe)(Ta,Nb)O ₄	ORTH metamict	010 indist fr conch	Black, brown, yellow	H 5 G 5.7-6.1 infus	Insol in acids. Reddish-brown in section.
2.071 ∧ 2.20	2.15	BROMARGYRITE, var Embolite Ag(Br,Cl)	CUB u mass	Sectile	Cols, yellow	H 1-1.5 G 5.6-6.0 F 1	Diss by NH ₄ OH.
[]	2.155	FERGUSONITE-BETA-(Ce) (Ce,La,Nd)NbO ₄	MCL metamict	---	Reddish-brown	H 6 G 5.34-5.44 infus	Dec by H ₂ SO ₄ .
	2.16 Li	BLAKEITE Ferric tellurite	Mass	---	Reddish-brown	H > 2 G > 3.1	Diss by acids.
	2.17	MANGANOSITE MnO	CUB oct, u mass	001 fair	Emerald-green, turns black	H 5-6 G 5.36 infus	Diss by acids. Emerald-green in powder and section.
2.07 ∧ 2.2	2.18	PYROCHLORE (Pyrochlore grp) (Ca,Na,Ce) ₂ Nb ₂ O ₆ (OH,F)	CUB oct	111 poor fr conch	Brownish-red	H 5 G 4.1-4.3 infus	Dec by H ₂ SO ₄ .
2.06 ∧ 2.28	2.18	ZIRKELITE (Ca,Th,Ce)Zr(Ti,Nb) ₂ O ₇	MCL ps cub	Conch	Black to dark brown	H 5-6 G 4.3-4.7 infus	Insol in acids. Reddish-brown in section.
2.05 ^	2.19	FERGUSONITE (Y,U,Ce)(Nb,Ta)O ₄	TET metamict	Conch	Black	H 6 G ~ 5.8 infus	Dec by H ₂ SO ₄ . After ignition, <u>n</u> = 2.14, biref.
	2.192	BIDEAUXITE Pb ₂ AgCl ₃ (F,OH) ₂	CUB	Conch	Cols to pale lavender	H 3 G 6.26 F 1	Diss by HNO ₃ , dec by HCl or by warm NH ₄ OH.
2.10 ^	2.20	THORIANITE (Th,U)O ₂	CUB	001 poor fr uneven	Black	H 6 G 9.32 infus	Slowly diss by HNO ₃ or H ₂ SO ₄ . Nearly opaque.
2.15 ∧ 2.253	2.20	BROMARGYRITE, var Iodobromite Ag(Br,Cl,I)	CUB	Sectile	Cols to yellow	H 2 G 6.0-6.2 F 1	Diss by NH ₄ OH.
	2.20	MIERSITE (Ag,Cu)I	CUB	011 perf sectile	Yellow	H 2.5 G 5-6.4 F 1	Diss by NH ₄ OH. Tw pl 111.

2.09 ^	2.20	ROMEITE, var Lewisite (Stibiconite grp) (Ca,Fe) ₂ (Sb,Ti) ₂ O ₆ (0,OH)	CUB oct	111 good	Honey-yellow to brown	H 5.5 G 4.95 fus	Insol in acids. Opt anom.
	2.205	KOBEITE (Y,U)(Ti,Nb) ₂ (0,OH) ₆ (?)	Amor	---	Dark brown	---	---
2.18 ^	2.2 calc	STIBIOBETAFITE (Pyrochlore grp) (Ca,Sb ⁺³) ₂ (Ti,Nb,Ta) ₂ (0,OH) ₇	CUB	---	Brownish-black	H ~ 5 G 5.30	Nb ₂ O ₅ 21.6, Ta ₂ O ₅ 19.3, TiO ₂ 16.5, Sb ₂ O ₃ 23.2, CaO 14.5%.
[]	2.22	POLYMIGNITE (Ca,Ce,Th,Fe)(Nb,Ti,Ta,Zr) ₄ O ₄	ORTH metamict	Conch	Black	H 6.5 G 4.8 infus	Reddish-brown in section.
2.19 [] 2.26	2.23	AESCHYNITE (Ce,Ca,Fe,Th)(Ti,Nb) ₂ (0,OH) ₆	ORTH metamict	Conch	Brownish-black	H 5.5 G 4.9-5.2 infus	Insol acids. Reddish-brown in section. Biref after ignition.
2.06 ^ 2.28	2.24	EUXENITE (Y,Ca,Ce,U,Th)(Nb,Ta,Ti) ₂ O ₆	ORTH metamict	Conch	Brownish-black	H 6.5 G 4.6-5.5 infus	Insol in acids. Reddish-brown in section.
2.13 ^	2.25	SAMARSKITE (Y,U,Ca,Fe)(Nb,Ta,Ti) ₂ O ₆	MCL metamict	Conch	Black	H 4-6 G 5.6-5.7 F 5	Diff sol hot acids. In section deep brown to opaque.
[]	2.25	POLYCRASE (Y,Ca,U,Th)(Ti,Nb) ₂ O ₆	ORTH metamict	Conch	Black	H 5-6 G 4.7-4.9 infus	Insol in acids. Reddish-brown in section. Ti analogue of Euxenite.
[]	2.25	TAZHERANITE (Zr,Ca,Ti) ₂ O ₂	CUB	Uneven	Orange to red	H 7.5 G 5.01 infus	---
2.20 ^	2.253	BROMARGYRITE Ag(Br,Cl,I)	CUB	Sectile	Yellow, green	H 2.5 G 6.3 F 1	Diss by NH ₄ OH.
2.12 ^	2.26	BISMUTITE Bi ₂ (CO ₃) ₂ O ₂	TET u mass earthy	001 dist	Yellow, brown	H 2.5-4 G 6.7-7.4 F 1.5	Diss by acids with eff.
[] 2.315	2.262	JIXIANITE Pb(W,Fe ⁺³) ₂ (0,OH) ₇	CUB	---	Brownish-black	H ~ 5 G 5.30	---
2.40 ^	2.265	PEROVSKITE, var Knopite and Dysanallyte (Perovskite grp) (Ca,Ce) ₂ (Ti,Nb) ₂ O ₆	ORTH ps cub	100	Black	H 5 G 4.1-4.2 infus	Dec by hot H ₂ SO ₄ . In part biref.
2.18 ^	2.28	ZIRKELITE (Ca,Th,Ce)Zr(Ti,Nb) ₂ O ₇	MCL ps cub	Conch	Black to dark brown	H 5-6 G 4.3-4.7 infus	Insol in acids. Reddish-brown in section.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
2.11 △	2.3	DAVIDITE (La,Ce)(Y,U,Fe ⁺²)(Ti,Fe ⁺³) ₂₀ (O,OH) ₃₈	TRIG metamict	---	Brownish-black	H 6-7 G 4.4-4.6 infus	Dec by hot H ₂ SO ₄ .
	2.3 (?)	TREVORITE (Spinel grp) NiFe ₂ O ₄	CUB oct	Uneven	Black	H 5 G 5.2 infus	Nearly opaque. Strongly magnetic. \bar{n} = 2.39 calc.
v 2.585	2.30	JACOBSITE (Spinel grp) (Mn,Fe ⁺² ,Mg)Fe ⁺³ ₂ O ₄	CUB oct	Uneven	Brownish-black	H 5.5-6.5 G 4.75-4.93 infus	Diss by HCl. Nearly opaque. Weakly magnetic.
	2.34	WINSTANLEYITE TiTe ₃ O ₈	CUB	---	Yellow, tan	H 4 G 5.57	Diss slowly in warm HNO ₃ .
	2.346 Na 2.313 Li	MARSHITE CuI	CUB tetrah	011 perf	Yellow, brown, red	H 2.5 G 5.68 F 1.5	Diss by acids. Disp exceeds that of diamond.
82 v 2.45	2.36 Li	FRANKLINITE (Spinel grp) (Zn,Mn,Fe ⁺²)(Mn,Fe ⁺³) ₂ O ₄	CUB oct	Uneven	Iron-black	H 6 G 5.2 infus	Diss by HCl. Reddish-brown in section. Weakly magnetic.
	2.37 Li	BUNSENITE NiO	CUB oct	---	Dark green	H 5.5 G 6.90 infus	Diff sol in acids.
v 2.40	2.37 Na 2.34 Li	SPHALERITE ZnS	CUB tetrah	110 perf	Cols to pale yellow	H 3.5 G 4.1 F 5	Diss by HCl. Luster resinous. Data for pure ZnS.
v 2.42	2.38 Na 2.34 Li	MAGNESIOFERRITE (Spinel grp) (Mg,Fe)Fe ₂ O ₄	CUB oct	Uneven	Black	H 6-6.5 G 4.5-4.7 infus	Diss by HCl. Nearly opaque to dark red in section. Strongly magnetic.
2.105 ^	2.38	LOPARITE (Perovskite grp) (Ce,Na,Ca) ₂ (Ti,Nb) ₂ O ₆	ORTH metamict	---	Black to brown	H 5-5.5 G 4.7-4.8 infus	Insol in acids.
2.265 ^	2.40	PEROVSKITE (Perovskite grp) Ca ₂ Ti ₂ O ₆	ORTH ps cub	100	Brown to black	H 5-5.5 G 4.1-4.48 infus	Dec by hot H ₂ SO ₄ . In part biref.
	2.40	ROMANECHITE BaMn ⁺² Mn ⁺⁴ ₈ O ₁₆ (OH) ₄	MCL ps orth	---	Black	H 5-6 G 4.4-4.7 infus	Diss by HCl with evolution of Cl.

2.37 ^ 2.43	2.40	SPHALERITE (Zn,Fe)S	CUB tetrah	110 perf	Brown	H 3.5-4 G 4.0-4.1 F 5	Diss by HCl. Luster resinous. Fe 5.5%.
	2.418	DIAMOND C	CUB tetrah	111 highly perf	Cols, yellow, blue	H 10 G 3.51 infus	Insol in acids. Tw pl 111. Disp extr.
2.38 ^	2.42	MAGNETITE (Spinel grp) FeFe ₂ O ₄	CUB oct	Parting 111 fr uneven	Black	H 6 G 5.17 infus	Diss by HCl. Nearly opaque. Strongly magnetic.
	>2.42	SILLENITE Bi ₁₂ SiO ₂₀	CUB	---	Green	H soft G 8.8 calc	---
2.40 ^ 2.47	2.43	SPHALERITE (Zn,Fe)S	CUB tetrah	110 perf	Brown to black	H 3.5-4 G 4.0 F 5	Diss by acids. Luster resinous. Fe 10.4%.
[]	2.45	BIXBYITE (Mn,Fe) ₂ O ₃	CUB	111 in traces	Black	H 6-6.5 G 4.95-5.05 F 4 to infus	Slowly diss by HCl.
2.36 ^	2.45 Na 2.41 Li	FRANKLINITE (Spinel grp) (Zn,Mn,Fe ⁺²)(Mn,Fe ⁺³) ₂ O ₄	CUB oct	Uneven	Iron-black	H 6 G 5.2 infus	Diss by HCl. Reddish-brown in section. Weakly magnetic.
2.43 ^	2.47	SPHALERITE (Zn,Fe)S	CUB tetrah	110 perf	Brown to black	H 3.5-4 G 3.9 F 4.5-5	Diss by acids. Luster resinous. Fe 17.1%.
2.11 ^	2.49	BRANNERITE (U,Ca,Ce)Ti ₂ O ₆	MCL metamict	Conch	Black	H 4.5 G 4.4-5.0 infus	Diff sol in acids. Light reddish-brown in section.
	2.49 Li	CADMIUM OXIDE CdO	CUB	111	Black	H 3 G 8.1-8.2 infus	Diss by HCl.
	2.5	STILLEITE ZnSe	CUB	---	---	H 4-5 G 5.29 calc F 5	---
	>2.5	CORDEROITE Hg ₃ S ₂ Cl ₂	CUB	---	Orange-pink	G (6.89)	---
	2.58 Na 2.49 Li	EGLESTONITE Hg ₆ Cl ₃ O ₂ H	CUB	---	Yellow	H 2-3 G 8.33-8.55	Dec by acids. Volat when heated. Anom biref. Darkens on exposure.
2.30 ^	2.585 Na 2.535 Li	JACOBSITE (Spinel grp) (Mn,Mg,Fe ⁺²)Fe ₂ O ₄	CUB oct	Uneven	Black to brownish-black	H 5.5-6.5 G 4.76 infus	Diss by HCl. Weakly to moderately magnetic.

Table 3. Isotropic minerals and mineraloids (continued)

Other entries	Refractive index	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
[]	2.6	MAGHEMITE $\gamma\text{-Fe}_2\text{O}_3$	CUB	---	Brownish-black	H 5 G ~ 4.8 infus	Diss by HCl. Strongly magnetic.
[]	2.60	CRYPTOMELANE $\text{K}(\text{Mn}^{+4}, \text{Mn}^{+2})_8\text{O}_{16}$	MCL ps tet u mass botryoidal	---	Steel-gray to black	H 6 G 4.2-4.4 infus	Diss by HCl. Streak brownish-black.
	2.68 Na 2.62 Li	HAUERITE MnS_2	CUB pyrito- hedral	001 perf	Brownish-black	H 4 G 3.46 F 3	Diss by HCl. Streak reddish.
	2.69 Na 2.68 Li	METACINNABAR HgS	CUB tetrah	Subconch to uneven	Grayish-black	H 3 G 7.65 volat	---
	2.78 Na 2.73 Li	ALABANDITE MnS	CUB 100, 110	100 perf	Iron-black	H 3.5-4 G 4.0 F 3	Diss by HCl. Streak green.
	2.85 Li	CUPRITE Cu_2O	CUB gyroidal	111 inter- rupted	Red	H 3.5-4 G 6.14 F 3	Diss by HCl. Streak crimson.
	2.93 Na 3.09 Li	DOMEYKITE Cu_3As	CUB u mass	Uneven	Tin-white to steel-gray	H 3-3.5 G 7.2-7.9 F 2	Dec by HNO_3 .
[] 3.54	3.02 Na 2.97 Li	TENNANTITE $(\text{Cu}, \text{Fe})_{12}(\text{As}, \text{Sb})_4\text{S}_{13}$	CUB u mass tetrah	---	Gray, iron-black	H 3-4.5 G 4.6-5.1 F 1.5	Dec by HNO_3 . In section bright red to opaque. Tw pl 111.
	3.24 Na 3.22 Li	TIEMANNITE HgSe	CUB tetrah	Uneven to conch	Steel- to blackish-gray	H 2.5 G 8.30-8.47 volat	Dec by HNO_3 .
	3.30 Na 4.16 Li	ALTAITE PbTe	CUB u mass	001 perf	Tin-white	H 3 G 8.15 F 1.5	Dec by HNO_3 .
	3.45 Na 3.37 Li	PETZITE Ag_3AuTe_2	CUB u mass	001, fr subconch	Steel-gray	H 2.5 G 9.13 F 1.5	Dec by HNO_3 .
	3.50 Na 3.79 Li	SPERRYLITE PtAs_2	CUB	001 indist fr conch	Tin-white	H 6-7 G 10.58 fus	---

3.80 Na 4.27 Li	LINNAEITE (Co,Ni) ₃ S ₄	CUB oct	001 imperf fr uneven	Steel-gray	H 4.5-5.5 G 4.85 F 2	Dec by HNO ₃ .
3.89 Na 3.87 Li	GALENA PbS	CUB	001 perf	Lead-gray	H 2.5-3 G 7.58 F 2	Dec by HNO ₃ .
4.50 Na 4.18 Li	PYRITE FeS ₂	CUB pyrito- hedral	Conch to uneven	Pale brass- yellow	H 6-6.5 G 5.02 F 2.5-3	Dec by HNO ₃ .

Table 4. Uniaxial positive minerals

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.46 √	1.309	1.312	.003	ICE H ₂ O	HEX mass	Conch	Col's	H 1.5, G 0.92	Melts at 0°C.
	1.378	1.390	.012	SELLAITE MgF ₂	TET prism	010, 110, perf	Col's	H 5, G 3.16, F 5	Diss by concd H ₂ SO ₄ . F with intumescence. El pos.
	1.40	---	mod	"CHRYSOCOLLA" Cu ₂ H ₂ Si ₂ O ₅ (OH) ₄ ·xH ₂ O	Fib c opal-like	---	Green	H 2+, G 2, infus	Dec by acids. Pleoc faint, 0 nearly col's, E pale bluish-green. Perhaps stained opal.
	1.439	1.442	.0029	KOGARKOITE Na ₃ (SO ₄)F	MCL ps trig	---	Col's	H 3.5, G 2.67, F 1.5	Sol in H ₂ O. Biref determined on synth crystal.
	1.443	1.445	.0012	SCHAIRERITE Na ₂₁ (SO ₄) ₇ F ₆ Cl	TRIG steep rhombs	Conch	Col's	H 3.5, G 2.62, F 1.5	Sol in H ₂ O.
	1.447	1.449	.002	GALEITE Na ₁₅ (SO ₄) ₅ F ₄ Cl	TRIG	---	White	G 2.61, F 1.5	Sol in H ₂ O.
1.40 △	1.0 red 1.45 Na 1.80 Tl	---	---	COVELLITE CuS	HEX plates	0001 perf	Indigo-blue	H 1.5-2, G 4.6- 4.76, F 2	Diss by HNO ₃ . Trans- lucent only in thinnest plates. In transmitted light, green and pleoc. Disp extr.
	1.46	1.57	.11	"CHRYSOCOLLA" (Cu,Al) ₂ H ₂ Si ₂ O ₅ (OH) ₄ ·xH ₂ O	MCL fib	---	Bluish-green	H 3, G 2.4, infus	Dec by acids. Pleoc faint, 0 nearly col's, E pale bluish-green. index increases with absorption of immersion liquid.
	1.460	1.478	.018	LANNONITE HCa ₄ Mg ₂ Al ₄ (SO ₄) ₈ F ₉ ·3H ₂ O	TET	---	White	G 2.22	Diss by acids.
	1.461	1.474	.013	TINCALCONITE Na ₂ B ₄ O ₇ ·5H ₂ O	TRIG	Hackly	Col's, white	H 2.5, G 1.88, F 2	Sol in H ₂ O. Data for synth compd.
	1.461	1.465	.004	CHABAZITE (Zeolite grp) (Ca,Na ₂)Al ₂ Si ₄ O ₁₂ ·6H ₂ O	TRIG rhombs	10Tl dist	White, red	H 4.5, G 2.05, F 3	Dec by HCl with sepn of slimy silica. 2V small to 0°.
	1.468	1.473	.005	ERIONITE (Zeolite grp) (K ₂ ,Ca,Na ₂) ₂ Al ₄ Si ₁₄ O ₃₆ · 15H ₂ O	HEX fib	---	White	G 2.02, F easy	Diff dec by HCl with sepn of silica. El pos.

	1.47	1.50	.03	HATCHETTITE $C_{40}H_{82}$	ORTH (?)	001 good	White	H 1, G. 0.9, F 1	Sol in org liquids. Melts at 65° and burns.
┌ 1.475	1.470	1.474	.004	GMELINITE (Zeolite grp) $(Na_2, Ca)Al_2Si_4O_{12} \cdot 6H_2O$	HEX	10T0 dist	White	H 4.5, G 2.0-2.1, F 3	Dec by HCl. 2V small to 0°. Tw axis <u>c</u> .
┌ 1.476	1.471	1.474	.003	HERSCHELITE (Zeolite grp) $(Na_2, Ca)Al_2Si_4O_{12} \cdot 6H_2O$	TRIG	---	White	H 4.5, G 2.1, F 3	Dec by HCl.
┌ 1.475	1.472	1.492	.020	GAGARINITE $NaCaY(F, Cl)_6$	HEX	Prism, fair	Creamy, yellow	H 3-4.5, G 4.1-4.2	Dec by acids. Anom biax, 2V up to 20°.
1.468 ^	1.477	1.480	.003	ERIONITE (Zeolite grp) $(K_2, Ca, Na_2)_2Al_4Si_{14}O_{36} \cdot 15H_2O$	HEX fib	---	White	G 2.08, F easy	Diff dec by HCl with sepn of silica. El pos.
	1.478	1.484	.006	BENTORITE $Ca_6(Cr, Al)_2(SO_4)_3(OH)_2 \cdot 26H_2O$	HEX	10T0 perf 0001 dist	Bright violet	H 2 2.03	Cr_2O_3 7.5, Al_2O_3 1.0%. Pleoc, E pale violet, O nearly cols.
	1.48	---	wk	FAUJASITE (Zeolite grp) $(Na_2, Ca)Al_2Si_4O_{12} \cdot 8H_2O$	CUB oct	111 dist	White	H 5, C 1.92, F 3	Dec by HCl. Uniax in 8 segments.
	1.480	1.493	.013	TETRANATROLITE (Zeolite grp) $Na_2Al_2Si_3O_{10} \cdot 2H_2O$	TET	---	White	H 5, G 2.25, F 2	Gel with HCl.
	1.481	1.483	.002	LEVYNE (Zeolite grp) $CaAl_2Si_4O_{12} \cdot 6H_2O$	TRIG	02Z1 dist	White	H 4, G 2.14, F 2-2.5	Gel with acids.
	1.484	1.603	.119	UREA $CO(NH_2)_2$	TET pyram	---	Yellow, brown	G 1.33	Diss by H_2O .
v 1.498	1.487	1.492	.005	APHTHALITE $(K, Na)_3Na(SO_4)_2$	TRIG tab, rhombs	10T0 fair 0001 poor	White	H 3, G 2.68, F 1.5	Slowly sol in H_2O .
1.461 ◇ 1.503	1.487	1.489	.002	CHABAZITE (Zeolite grp) $(Ca, Na_2)Al_2Si_4O_{12} \cdot 6H_2O$	TRIG rhombs	10T1 dist	White, red	H 4.5, G 2.08, F 3	Dec by HCl with sepn of silica. 2V small to 0°.
	1.488	1.500	.012	DOUGLASITE $K_2FeCl_4 \cdot 2H_2O$	MCL u mass	---	Light green, alters to reddish	G 2.16	Nearly uniax. Sol in H_2O . Tends to lie on face normal to opt axes.
	1.490	1.502	.012	NATROLITE (Zeolite grp) $(Na_2, Ca)Al_2Si_3O_{10} \cdot 2H_2O$	ORTH fib	110 perf 010 imperf	White	H 5, G 2.26, F 2-3	Gel with acids. El pos. Submicroscopic inter- growths of fibers give 2V near 0°.
	1.491	1.507	.016	GIUSEPPETTITE (Cancrinite grp) $(Na, K, Ca)_{7-8}(Si, Al)_{12}O_{24}(SO_4, Cl)_{1-2}$	HEX	---	Pale violet- blue	H 6-7, G 2.35	---

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.504 v	1.496	1.542	.046	PARAFFIN (mixt. of hydrocarbons)	HEX and ORTH plates and fib	---	White, cols	H 1, G 0.9, F very easy	Insol in acids.
1.492 ┌	1.496	1.502	.006	TUGTUPITE $\text{Na}_4\text{BeAlSi}_4\text{O}_{12}\text{Cl}$	TET	Pyramid, conch	Rose, gray, bluish	H 4, G 2.34	---
1.487 ^	1.498	1.503	.005	APHTHALITE $(\text{K}, \text{Na}, \text{Cu})_3\text{Na}(\text{SO}_4)_2$	TRIG	10T0 fair 0001 poor	White, blue, yellowish	H 3, G 2.68, F 1.5	Slowly sol in H_2O .
1.505 ┌	1.500	1.502	.002	GARRONITE (Zeolite grp) $\text{Na}_2\text{Ca}_5\text{Al}_{12}\text{Si}_{20}\text{O}_{64} \cdot 27\text{H}_2\text{O}$	ORTH ps tet, radiating	2 at 90°, poor	Cols	G 2.15, H 4	---
1.487 ^	1.503	1.507	.004	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na}_2)\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG rhombs	10T1 dist	White, red	H 4.5, G 2.09, F 3	Dec by HCl with sepn of silica. 2V small to 0°.
1.496 ^	1.504	1.550	.046	PARAFFIN (mixt. of hydrocarbons)	HEX (?) and ORTH, plates and fib	---	White, cols	H 1, G 0.9, F very easy	Insol in acids. Lies on base when crushed below cover glass.
	1.508	1.509	.001	LEUCITE KAlSi_2O_6	TET ps cub, 211	110 poor	Cols	H 6, G 2.5, infus	Dec by acids. Inclusions are characteristic. 2V small to 0°. Poly tw.
1.513 ┌	1.510	1.515	.005	FRANCOANELITE $\text{H}_6(\text{K}, \text{NH}_4)_3\text{Al}_5(\text{PO}_4)_8 \cdot 18\text{H}_2\text{O}$	TRIG nodular, pulverulent	---	Yellow, white	Very soft, G 2.27	Diss by acids.
	1.510	1.512	.002	FRANZINITE (Cancrinite grp) $(\text{Na}, \text{Ca})_7(\text{Si}, \text{Al})_{12}\text{O}_{24} (\text{SO}_4, \text{CO}_3, \text{OH}, \text{Cl})_3 \cdot \text{H}_2\text{O}$	HEX pris	---	White	H 5, G 2.49	---
1.518 ┌	1.511	1.518	.007	LEIFITE $\text{Na}_2(\text{Si}, \text{Al}, \text{Be})_7(\text{O}, \text{OH}, \text{F})_{14}$	TRIG pris	Pris dist	Cols	H 6, G 2.5, F easy	Insol in HCl. El pos.
	1.514	1.522	.008	NITROCALCITE $\text{Ca}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$	Fib, efflor	One perf	Cols	G 1.90	Readily sol in H_2O , hygroscopic. Data on natural material (synth material biax neg, $\beta = 1.498$).
1.515 ┌	1.515	1.54	.025	OZOCERITE (hydrocarbon)	ORTH (?) fib	---	Pale yellow	H 2.5, G 1.9-2.1, F very easy	Insol in acids. Burns. El neg. Lies on base.

1.515 ^ 1.522	neg	1.518	1.521	.003	DAVYNE (Cancrinite grp) (Na,Ca,K) ₈ Al ₆ Si ₆ O ₂₄ (Cl,SO ₄ ,CO ₃) ₂₋₃	HEX	10T0, 0001 perf	Cols	H 5.5, G 2.4+, F 4	Gel with acids. F with intumescence. Also iso- tropic and uniax neg.
1.537 v		1.518	1.561	.043	FIBROFERRITE Fe ⁺³ SO ₄ (OH)·5H ₂ O	MCL fib, ps trig	001 perf	Pale yellow	H 2.5, G 1.9-2.1 F 4.5-5	Dec by H ₂ O, diss by HCl. El pos. ² Pleoc wk, 0 nearly cols. E pale yellow.
1.555 v		1.52	1.55	.03	KOENENITE Na ₄ (Mg,Ca) ₉ Al ₄ (OH) ₂₂ Cl ₁₂	TRIG foliated	0001 mic	Yellow to red	H 1.5, G 1.98	Dec by H ₂ O, diss by acids. Pleoc, 0 red- brown, E cols. El neg.
		1.520	1.540	.020	ANDERSONITE Na ₂ Ca(UO ₂)(CO ₃) ₃ ·6H ₂ O	TRIG ps cub	---	Yellow green	G 2.8	Sol in H ₂ O. Pleoc, 0 cols, E pale yellow. Fluor bright green in UV.
		1.521	1.529	.008	MICROSOMMITE (Cancrinite grp) (Na ₂ ,Ca,K ₂) ₇₋₈ (Al,Si) ₁₂ O ₂₄ (Cl ₂ ,CO ₃ ,SO ₄) ₂₋₃	HEX	10T0 perf 0001 less so	Cols	H 6, G 2.4, F 4	Gel with acids. Poly- type of Davyne.
1.529 v		1.523	1.544	.021	WEDDELLITE CaC ₂ O ₄ ·2H ₂ O (oxalate)	TET	---	White	H 4, G 1.94, infus	Diss by acids.
1.528 v		1.523	1.529	.006	AFGHANITE (Cancrinite grp) (Na,Ca,K) ₈ (Al,Si) ₁₂ O ₂₄ (Cl,SO ₄ ,CO ₃) ₃ ·H ₂ O	HEX	10T0 perf	Bluish-white	H 5.5-6, G 2.55, fus	Gel with acids.
1.522 v		1.524	1.532	.008	BERLINITE AlPO ₄	TRIG	Conch	Cols	H 6.5, G 2.64, infus	Insol in acids. Iso- structural with Quartz.
1.552 v		1.525	1.525	<.001	KALBORSITE K ₆ Al ₄ BSi ₆ O ₂₀ [B(OH) ₄]Cl	TET	110 perf	Cols to rose	H 6, G 2.5	---
		1.536	1.572	.036	COQUIMBITE Fe ₂ (SO ₄) ₃ ·9H ₂ O	TRIG	10T1 imperf	Cols to violet	H 2.5, G 2.07, F 4.5-5	Sol in H ₂ O. Al ₂ O ₃ 6.9%.
		1.536	1.545	.009	ASHCROFTINE KNaCaY ₂ Si ₆ O ₁₂ (OH) ₁₀ ·4H ₂ O	TET acic	100 perf 001 good	White, pink	H 5, G 2.61	---
1.530 v		1.536	1.544	.008	YAGIITE (Na,K) ₃ Mg ₄ (Al,Mg) ₆ (Si,Al) ₂₄ O ₆₀	HEX	---	Light blue to cols	G 2.70	Pleoc wk, E cols, 0 very bright blue.
1.542 v		1.536	1.538	.002	FLUORAPOPHYLLITE KCa ₄ Si ₈ O ₂₀ (F,OH)·8H ₂ O	TET	001 perf 110 poor	Cols	H 5, G 2.44, F 2	Dec by HCl with sepn of slimy silica. Opt anom.
1.518 ^		1.537	1.572	.035	FIBROFERRITE Fe ⁺³ SO ₄ (OH)·5H ₂ O	MCL ps trig, fib	001 perf	Pale yellow	H 2.5, G 1.9-2.1, F 4.5-5	Dec by H ₂ O, diss by HCl. El pos. ² Pleoc wk, 0 nearly cols, E pale yellow.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ε							
1.546 ^ 1.536 ^ 1.540 ^ 1.558 ^ 1.536 ^ 1.52 ^	1.537	1.542	.005	ROEDDERITE (Osumilite grp) (Na,K) ₂ Mg ₅ Si ₁₂ O ₃₀	HEX	---	Cols	G 2.6	---
	1.540	1.560	.020	CHLORMAGALUMINITE (Mg,Fe ⁺²) ₄ Al ₂ (OH) ₁₂ (Cl,.5CO ₃)·2H ₂ O	HEX	0001 perf	Cols to greenish-brown	Soft, G 2.0-2.1, infus	Diss by acids.
	1.540	1.546	.006	OSUMILITE-(Mg) (Osumilite grp) (K,Na)(Mg,Fe) ₂ (Al,Fe) ₃ (Si,Al) ₁₂ O ₃₀ ·H ₂ O	HEX	---	Light blue	G 2.63	Pleoc wk, 0 cols, E pale blue.
	1.542	1.543	.001	HYDROXYAPOPHYLLITE KCa ₄ Si ₈ O ₂₀ (OH,F)·8H ₂ O	TET	001 perf	Cols	H 4.5-5, G 2.37, F 2	Dec by HCl with sepn of slimy silica. Opt anom.
	1.543	1.5455	.0025	EIFELITE (Osumilite grp) K ₂ Na ₄ Mg ₉ Si ₂₄ O ₆₀	HEX	---	---	---	Stated to be negative.
	1.544	1.553	.009	QUARTZ SiO ₂	TRIG hex prisms and pyramids	Conch	Cols	H 7, G 2.66, infus	Insol in acids, dec by HF.
	1.546	1.550	.004	OSUMILITE (K,Na)(Mg,Fe) ₂ (Al,Fe) ₃ (Si,Al) ₁₂ O ₃₀ ·H ₂ O	HEX	---	Cols, blue, red	G 2.62	Pleoc wk, 0 light blue, E cols.
	1.547	1.575	.028	BASSANITE 2CaSO ₄ ·H ₂ O	HEX ps hex, fib	---	White	G 2.7	Diss slowly by acids. El pos.
	1.550	1.650	.100	VATERITE CaCO ₃	HEX fib	---	Cols	G 2.64, infus	Diss by acids with eff. Spher aggregates give lower ns down to 1.538.
	1.552	1.558	.006	COQUIMBITE Fe ₂ (SO ₄) ₃ ·9H ₂ O	TRIG	10Tl imperf	Cols to violet	H 2.5, G 2.1, F 4.5-5	Sol in H ₂ O.
	1.555	1.585	.030	KOENENITE Na ₄ (Mg,Ca) ₉ Al ₄ (OH) ₂₂ Cl ₁₂	TRIG foliated	0001 mic	Yellow to red	H 1.5, G 2.15	Dec by H ₂ O, diss by acids. Pleoc, 0 red-brown, E cols. El neg.
	1.556	1.645	.089	JULIENITE Na ₂ CO(SCN) ₄ ·8H ₂ O	TET pris	001	Blue	G 1.65	Sol in H ₂ O. An artifact?
	1.558	1.613	.055	FERRINATRITE Na ₃ Fe(SO ₄) ₃ ·3H ₂ O	TRIG acic	10Tl perf 1120 less so	White, greenish	H 2.5, G 2.57, F 1.5	Sol in H ₂ O.

1.547 ^	1.558	1.586	.028	BASSANITE $2\text{CaSO}_4 \cdot \text{H}_2\text{O}$	HEX ps hex, fib	---	White	G 2.7	Diss slowly by acids. El pos.
v 1.592	1.559	---	Low to med	MERRIHUEITE (Osumilite grp) $(\text{K}, \text{Na})_2(\text{Fe}^{+2}, \text{Mg})_5\text{Si}_{12}\text{O}_{30}$	HEX	---	Greenish- blue	G 2.87 (calc)	Opt char unk. FeO 10.5%. Pleoc cols to greenish- blue.
v 1.59	1.559	1.580	.021	BRUCITE $\text{Mg}(\text{OH})_2$	TRIG tab to fib	0001 mic	White	H 2.5, G 2.39, infus	Diss by acids. Flexible. Luster pearly on base. Opt anom. El neg.
v 1.576	1.562	1.576	.014	CLINOCHLORE (Chlorite grp) $(\text{Mg}, \text{Fe})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	MCL ps hex, plates	0001 mic	White	H 2.5-3, G 2.5, F diff	Diff dec by acids. El neg. Fe_2O_3 1.7%.
	1.564	1.577	.013	PERHAMITE $\text{Ca}_3\text{Al}_7(\text{SiO}_4)_3(\text{PO}_4)_4(\text{OH})_3 \cdot 16\text{H}_2\text{O}$	HEX	0001 perf	Brown, white	H near 5, G 2.64	---
	1.565	1.575	.010	PINNOITE $\text{MgB}_2\text{O}_4 \cdot 3\text{H}_2\text{O}$	TET	Uneven	Yellowish	H 3.5, G 2.27, F 3	Diss by acids.
	1.569	1.581	.012	SCHAURTEITE $\text{Ca}_3\text{Ge}(\text{SO}_4)_2(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	HEX acic	---	White	G 2.65	---
	1.571	1.590	.019	NEFEDOVITE $\text{Na}_5\text{Ca}_4(\text{PO}_4)_4\text{F}$	TCL ps tet	Conch	Cols	H 4.5, G 3.01	Diss by acids.
v 1.583	1.572	1.592	.020	ALUNITE (Alunite grp) $\text{KAl}_3(\text{SO}_4)_2(\text{OH})_6$	TRIG tab 0001, u mass	0001 dist	White, yellow, reddish	H 3.5-4, G 2.6, infus	Insol in acids, but sol in acids after gentle heating. El neg.
	1.572	1.586	.014	EUCRYPTITE LiAlSiO_4	TRIG	Conch	Cols	H 5.5-6.5, G 2.66	Gel with acids. Fluor pink in UV.
	1.573	1.599	.026	TUNISITE $\text{NaCa}_2\text{Al}_4(\text{CO}_3)_4(\text{OH})_8\text{Cl}$	TET	001 perf pris good	Cols	H 4.5, G 2.51, infus	Insol in H_2O .
v 1.603	1.574	1.590	.016	NATROALUNITE (Alunite grp) $\text{NaAl}_3(\text{SO}_4)_2(\text{OH})_6$	TRIG tab 0001, u mass	0001 dist	White, yellow	H 4, G 2.78, infus	Insol in acids, but diss after gentle heating. El neg. Na_2O 7.6, K_2O 0.1%.
1.562 ^ 1.587	1.576	1.579	.003	CLINOCHLORE (Chlorite grp) $(\text{Mg}, \text{Fe})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	MCL ps hex, plates	001 perf	Green, white	H 2, G 2.7, F 5-5.5	Dec by H_2SO_4 . El neg. Abnormal blue interf color. Pleoc, X, Y green, Z cols.
	1.577	---	very low	NAMUWITE $(\text{Zn}, \text{Ca})_4(\text{SO}_4)(\text{OH})_6 \cdot 4\text{H}_2\text{O}$	HEX	0001 perf	Pale sea- green	G 2.77	Opt sign unk.
[]	1.580	1.588	.008	COERULEOLACTITE $(\text{Ca}, \text{Cu})\text{Al}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 4-5\text{H}_2\text{O}$	TCL (?) fib crusts	---	Milk-white, light blue	H 5, G 2.55- 2.70, infus	Diss by acids. Related to turquoise.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.572 ∇ 1.620	1.583	1.595	.012	ALUNITE (Alunite grp) $(K,Na)(Al,Fe)_3(SO_4)_2(OH)_6$	TRIG tab 0001 u mass	0001 dist	White, red, yellow	H 4, G 2.7, infus	Insol in acids, but diss by acids after gentle heating. El neg.
1.575 √ 1.604	1.585	1.656	.071	CACOXENITE $Fe^{+3}_9(PO_4)_4(OH)_{15} \cdot 18H_2O$	HEX needles \underline{c}	---	Yellow to brownish	H 3-4, G 2.3, F 2.5-3	Diss by acids. Pleoc, 0 cols to pale yellow, E orange- to canary-yellow.
1.594 v	1.586	1.595	.009	WARDITE $NaAl_3(PO_4)_2(OH)_4 \cdot 2H_2O$	TET	001 perf	Cols to pale green	H 5, G 2.87, F 3	Slowly diss by acids. Opt anom, biax in sectors. El neg.
1.576 ∇ 1.598	1.587	1.594	.007	CLINOCHLORE (Chlorite grp) $(Mg,Fe)_5Al(Si_3Al)O_{10}(OH)_8$	MCL ps hex, plates	001 perf	Greenish- gray	H 2.5, G 2.67, F 4.5	Slowly dec by acids. 2V 0-5°. El neg.
1.604 v	1.588	1.593	.005	EUDIALYTE $Na_4(Ca,Ce)_2FeZrSi_8O_{22}$ $(OH,Cl)_2$	TRIG	0001 dist 1020 poor	Yellow, pink, brown	H 5-5.5, G 2.85, F 3.0	Gel with acids. Opt anom, biax. Pleoc wk in pink and yellow.
89	1.588	1.589	.001	RINNEITE $K_3NaFeCl_6$	TRIG	1120 good	Cols, rose, yellow	H 3, G 2.35, F easy	Sol in H_2O . Anom biax. Abnormal interf colors in ultrablue.
1.559 ∇ 1.707	1.59	1.60	.01	BRUCITE $(Mg,Mn)(OH)_2$	TRIG	0001 perf	Light brown	H 2.5, G 2.5, infus	Diss by HCl. El neg. MnO 18.1, ZnO 3.7%.
	1.59	---	very wk	CHLORMANGANOKALITE K_4MnCl_6	TRIG	Conch	Yellow	H 2.5, G 2.31, F easy	Sol in H_2O . Deliq.
1.559 ^	1.592	---	low to med	MERRIHUEITE (Osumilite grp) $(K,Na)_2(Fe^{+2},Mg)_5Si_{12}O_{30}$	HEX	---	Greenish-blue	---	Opt char unk. MgO 4.4 FeO 24.9%.
1.586 ^	1.594	1.604	.010	WARDITE $NaAl_3(PO_4)_2(OH)_4 \cdot 2H_2O$	TET	001 perf	Cols to green	H 5, G 2.81, F 3	Slowly diss by acids. Opt anom, biax in sectors. El neg.
	1.597	1.612	.015	AMESITE (Kaolinite- Serpentine grp) $(Mg,Fe)_4Al_2(Al_2Si_2)O_{10}$ $(OH)_8$	MCL ps hex, plates	001 mic	Pale bluish-green	H 2-3, G 2.77, infus	Slowly dec by HCl. El neg.
1.587 ∇ 1.618	1.598	1.606	.008	CLINOCHLORE (Chlorite grp) $(Mg,Fe)_5Al(Si_3Al)O_{10}(OH)_8$	MCL ps hex, plates	001 mic	Green	H 2, G 2.80, F 5	Diff dec by HCl. 2V 0- 20°. El neg. FeO 9.1, Fe ₂ O ₃ 1.9%.

	1.598	1.598	very wk	COMBEITE $\text{Na}_2\text{Ca}_2\text{Si}_3\text{O}_9$	TRIG pris	---	Cols	G 2.84	Gel with hot HCl.
\vee 1.612	1.601	1.632	.031	NARSARSUKITE $\text{Na}_2(\text{Ti},\text{Fe})\text{Si}_4(\text{O},\text{F})_{11}$	TET tab	110 good 010 dist	Honey-yellow to reddish	H 6-7, G 2.75, F easy	Insol in acids. Pleoc wk, 0 yellow, E cols.
\wedge 1.574	1.603	1.611	.008	NATROALUNITE (Alunite grp) $(\text{Na},\text{K})\text{Al}_3(\text{SO}_4,\text{PO}_4)_2(\text{OH})_6$	TRIG tab 0001, u mass	0001 dist	White, yellow	H 4, G 2.78, infus	Insol in acids but diss in acids after being heated gently. Na_2O 3.9, K_2O 2.3, P_2O_5 5.1%.
	1.603	1.609	.006	CALCIUM CATAPLEIITE $\text{CaZrSi}_3\text{O}_9 \cdot 2\text{H}_2\text{O}$	HEX	One clv	Pale yellow, cream	H 4.5-5, G 2.77	
\wedge 1.585	1.604	1.676	.072	CACOXENITE $\text{Fe}^{+3}_9(\text{PO}_4)_4(\text{OH})_{15} \cdot 18\text{H}_2\text{O}$	HEX needles \underline{c}	---	Yellow to brownish	H 3-4, G 2.3, F 2.5-3	Diss by acids. Pleoc, 0 cols, E canary yellow.
	1.604	1.615	.011	SARCOLITE $(\text{Ca},\text{Na})_{7-8}\text{Al}_4\text{Si}_6\text{O}_{24}$ $(\text{OH})_2$ (?)	TET	---	Light rose	H 6, G 2.93, F 3 (?)	Gel with acids. Anom biax.
\diamond 1.588 1.624	1.604	1.607	.003	EUDIALYTE $\text{Na}_4(\text{Ca},\text{Ce})_2\text{FeZrSi}_8\text{O}_{22}$ $(\text{OH},\text{Cl})_2$	TRIG	0001 dist 1020 poor	Yellow, pink, brown	H 5-5.5, G 2.84, F 4-5	Gel with acids. Opt anom, biax. Pleoc wk in pink and yellow.
	1.606	1.620	.014	SARYARKITE $\text{Ca}(\text{Y},\text{Th})\text{Al}_5(\text{SiO}_4)_2$ $(\text{PO}_4,\text{SO}_4)_2(\text{OH})_7 \cdot 6\text{H}_2\text{O}$	HEX microcryst	---	White	H 3.5-4, G 3.1	Dec by acids. Related to Rhabdophane.
	1.606	1.608	.002	SOGDIANITE (Osumilite grp) $(\text{K},\text{Na})_2\text{Li}_2(\text{Li},\text{Fe}^{+3},\text{Al})_2$ $\text{ZrSi}_{12}\text{O}_{30}$	HEX	---	Violet	H 7, G 2.90	ZrO_2 9.8, TiO_2 2.9, FeO 1.2, Fe_2O_3 4.6, Na_2O 2.8, K_2O 4.8, Li_2O 3.7%. Reported as uniax neg.
\vee 1.625	1.610	1.620	.010	GORCEIXITE (Crandallite grp) $(\text{Ba},\text{Ca},\text{Ce})\text{Al}_3(\text{PO}_4)_2(\text{OH})_5 \cdot$ H_2O	TRIG u mass	---	Brown, white	H 6, G 3.1-3.3	---
\wedge 1.601	1.612	1.660	.048	NARSARSUKITE $\text{Na}_2(\text{Ti},\text{Fe})\text{Si}_4(\text{O},\text{F})_{11}$	TET tab	110 good 010 dist	Honey-yellow to reddish	H 6-7, G 2.78, F easy	Insol in acids. Pleoc wk, 0 yellow, E cols.
\vee 1.623	1.613	1.622	.009	CRANDALLITE (Crandallite grp) $\text{CaAl}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG u fib	0001 perf	Cols, yellow	H 5, G 2.8, F 3	Diff sol in acids. Fuses with intumescence. El neg.
	1.615	1.617	.002	FLUOCERITE $(\text{Ce},\text{La})\text{F}_3$	HEX	0001 perf	Reddish- yellow	H 4, G 5.8, infus	Insol in HCl or HNO_3 , diss by H_2SO_4 .
	1.618	1.630	.012	GAINESITE $\text{Na}_2\text{Zr}_2\text{Be}(\text{PO}_4)_4$	TET	Conch	Pale bluish- lavender	H 4, G 2.94	Insol in HCl.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.598 ^	1.618	1.621	.003	CLINOCHLORE (Chlorite grp) $(\text{Mg,Fe})_5\text{Al}(\text{Si}_3\text{Al})_0\text{}_{10}(\text{OH})_8$	MCL ps hex	001 perf	Green	H 2, G 2.8	Pleoc wk in green. FeO 13.7, Fe ₂ O ₃ 1.9%.
1.583 ^	1.620	1.641	.021	ALUNITE (Alunite grp) $\text{K}(\text{Al,Fe})(\text{SO}_4)_2(\text{OH})_6$	TRIG u mass	0001 dist	White to reddish	H 4, G 2.81, infus	Insol in acids, but after gentle heating diss by acids. El neg. Fe ₂ O ₃ 5.0%.
v 1.640	1.620	1.630	.010	GOYAZITE (Crandallite grp) $\text{SrAl}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG tab 0001	0001 perf	Cols to brown	H 4.5-5, G 3.15-3.26, F 4	Slowly diss by acids. Zoned. Anom biax on basal sections. Pleoc, 0 red-brown, E yellow.
1.613 ^ 1.627	1.623	1.634	.011	CRANDALLITE (Crandallite grp) $\text{CaAl}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG u fib	0001 perf	Cols to yellow	H 5, G 2.8, F 3	Diff sol in acids. Fuses with intumescence. El neg.
v 1.624	1.624	1.673	.049	WADEITE $\text{K}_2\text{CaZrSi}_4\text{O}_{12}$	HEX	2 dist fr conch	Cols, rose, lilac	H 6, G 3.11, F diff	Insol in acids. El pos.
1.624 ^	1.624	1.653	.029	WADEITE $\text{K}_2\text{CaZrSi}_4\text{O}_{12}$	HEX	2 dist fr conch	Cols, rose, lilac	H 6, G 3.10, F diff	Insol in acids. Anom biax. El pos.
1.604 ^	1.624	1.629	.005	EUDIALYTE $\text{Na}_4(\text{Ca,Ce})_2\text{FeZrSi}_8\text{O}_{22}(\text{OH,Cl})_2$	TRIG	0001 dist IT20 poor	Yellow, pink, brown	H 5-5.5, G 2.94, F 4-5	Gel with acids. Opt anom, biax. Pleoc wk in pink and yellow.
1.610 ^	1.625	---	wk	GORCEIXITE (Crandallite grp) $(\text{Ba,Ca,Ce})\text{Al}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG u mass	---	Brown, white	H 6, G 3.09-3.32	---
v 1.639	1.626	1.640	.014	SVANBERGITE (Beudantite grp) $\text{SrAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG ps cub	0001 dist	Cols, yellow, brown	H 4.5, G 2.98-3.22, F diff	Diss with diff by acids. Basal section may show 6 biax sectors.
1.62 □ 1.63	1.626	1.627	.001	METATORBERNITE (Meta-autunite grp) $\text{Cu}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$	TET tab 001	001 mic	Green	H 2.5, G 3.6-3.8, F 3	Diss by acids. Pleoc wk, 0 bright green, E cols. Not fluor in UV.
1.623 ^	1.627	1.632	.005	CRANDALLITE (Crandallite grp) $(\text{Ca,Sr,Ce})\text{Al}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG u fib	0001 perf	Cols to yellow	H 5, G 3.2, F 4	Diff sol in acids. SrO 5.1, RE ₂ O ₃ 10.5%.

	1.629	---	very low	KAMAISHILITE $\text{Ca}_2\text{Al}_2\text{SiO}_6(\text{OH})_2$	TET	---	Cols	---	Opt sign unk.
	1.629	1.632	.003	BABEPHITE $\text{BaBe}(\text{PO}_4)(\text{O},\text{F})$	TET tab	---	White	G 4.31	Insol in acids, diss by aqua regia or HF.
	1.630	1.638	.008	HAUCKITE $\text{Zn}_{18}(\text{Mg},\text{Mn})_{24}\text{Fe}^{+3}(\text{SO}_4)_4(\text{CO}_3)_2(\text{OH})_{81} (?)$	HEX	001 perf	Orange to yellow	H 2-3, G 3.10	Pleoc, 0 golden-brown, E pale yellow. ZnO 36.0, MnO 17.1%.
1.632	1.630	1.637	.007	MELILITE (Melilite grp) $(\text{Ca},\text{Na})_2(\text{Mg},\text{Al})(\text{Si},\text{Al})_2\text{O}_7$	TET	001, 110	Cols	H 5, G 2.95-3.05, F 3	Gel with acids. Fe_2O_3 1.9, FeO 3.1, Na_2O 3.75%.
	1.631	1.652	.021	FAHEYITE (Mn,Mg) $\text{Fe}^{+3}_2\text{Be}_2(\text{PO}_4)_4 \cdot 6\text{H}_2\text{O}$	HEX fib	Pris perf	White to brownish	G 2.66	Slowly diss by hot acids. El pos.
1.630 ^ 1.648	1.632	1.639	.007	AKERMANITE (Melilite grp) $\text{Ca}_2\text{MgSi}_2\text{O}_7$	TET	001, 110	Cols	H 5, G 2.95	Gel with acids. Data for synth compd.
	1.633	---	low	BURTITE $\text{CaSn}(\text{OH})_6$	TRIG	001 good	Cols	H ~ 3, G 3.28	Anom biax.
	1.636	1.647	.011	WOODHOUSEITE (Beudantite grp) $\text{CaAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001 good	Cols	H 4.5, G 3.01	Diff diss by acids. Basal section may show biax sectors.
1.626 ^	1.639	1.646	.007	SVANBERGITE (Beudantite grp) $\text{SrAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG ps cub	0001 dist	Cols, yellow, brown	H 4.5, G 2.98-3.22, F diff	Diff diss by acids. Basal section may show 6 biax sectors. SrO 21.2, CaO 1.0%.
1.620 ^	1.640	1.651	.011	GOYAZITE (Crandallite grp) $\text{SrAl}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG tab 0001	0001 perf	Cols to brown	H 4.5-5, G 3.15, F 4	Diff diss by acids. Anom biax. 2V 0-20°.
1.670	1.643	1.755	.112	PARISITE-(Y) ("Yttroparisite") $\text{Ca}(\text{Y},\text{Ce})_2(\text{CO}_3)_3\text{F}_2$	HEX	0001	Yellow, brown	---	No analysis, needs confirmation.
1.650	1.643	1.73	.087	SYNCHYSITE-(Y) ("Doverite") $\text{CaY}(\text{CO}_3)_2\text{F}$	ORTH ps hex, u mass	001	Red-brown	G 3.6-3.7 (?)	Diff diss by acids.
1.674	1.644	1.744	.100	SYNCHYSITE $\text{CaCe}(\text{CO}_3)_2\text{F}$	ps hex	---	Red-brown	H 4.5, G 3.9	Diff diss by acids.
	1.644	1.664	.020	TRISTRAMITE $(\text{Ca},\text{U}^{+4},\text{Fe}^{+3})(\text{PO}_4,\text{SO}_4) \cdot 2\text{H}_2\text{O}$	HEX fib	---	Pale- to greenish-yellow	G 3.8-4.2	Not fluor in UV light.
	1.646	1.647	.001	BRAITSCHITE $(\text{Ca},\text{Na}_2)_7(\text{Ce},\text{La})_2\text{B}_{22}\text{O}_{43} \cdot 7\text{H}_2\text{O}$	HEX mass	---	White to pink	G 2.90	---

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.632 ∇ 1.653 neg	1.648	1.649	.001	MELILITE (Melilite grp) $\text{Ca}_2(\text{Mg},\text{Al})(\text{Si},\text{Al})_2\text{O}_7$	TET	001, 110	Cols	H 5, G 2.9-3.1, F 3	Gel with acids. Data for synth Ak 60, Ge 40.
1.680	1.653	1.675	.022	PLUMBOGUMMITE (Crandallite grp) $\text{PbAl}_3(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG gum-like	---	Yellow, brown	H 4-5, G 4.0, F 4-5 (?)	Diss with diff by HNO_3 .
1.678	1.653	1.661	.008	FLORENCITE (Crandallite grp) $\text{CeAl}_3(\text{PO}_4)_2(\text{OH})_6 \cdot \text{H}_2\text{O}$	TRIG	0001	Pink	H 5, G 3.46, infus	Diss with diff by acids.
1.675	1.654	1.703	.049	RHABDOPHANE (Ce,La) $\text{PO}_4 \cdot \text{H}_2\text{O}$	HEX fib, mass	Uneven	Brown, white, reddish	H 3-4, G 3.5- 4.0, infus	Diss by acids.
	1.654	1.670	.016	PHENAKITE Be_2SiO_4	TRIG pris	1120 dist 1011 imperf fr conch	Cols, yellow	H 7.5, G 2.97, infus	Insol in acids. Tw pl 1010 penet.
1.662	1.655	1.656	.001	CAHNITE $\text{Ca}_2\text{BaSO}_4(\text{OH})_4$	TET	110 perf	Cols	H 3, G 3.06, F 3	Diss by acids. Tw. Abnormal interf colors in brown.
1.652 □ 1.658	1.656	1.708	.052	DIOPTASE $\text{CuSiO}_2(\text{OH})_2$	TRIG	1011 perf	Emerald- green	H 5, G 3.28, infus	Gel with acids. Anom biax. Abs in thick sections $O > E$.
	1.656	1.682	.026	DESPUJOLSITE $\text{Ca}_3\text{Mn}^{+4}(\text{SO}_4)_2(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	HEX	Conch	Lemon-yellow	H 2.5, G 2.46	Diss by HCl with evolution of chlorine. Pleoc wk, O pale yellow, E yellow.
	1.660	1.756	.096	RÖNTGENITE $\text{Ca}_2\text{Ce}_3(\text{CO}_3)_5\text{F}_3$	TRIG	---	Cols to light yellow	G 4.2	Diff diss by acids.
1.636 ^	1.662	1.669	.007	WOODHOUSEITE (Beudantite grp) $\text{CaAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001 good	Cols, gray, lilac	H 4-4.5, G 2.9- 3.0	Diff diss by acids. Basal section may show biax sectors.
	1.664	1.672	.008	MELIPHANITE var Gugiaite $(\text{Ca},\text{Na})_2\text{Be}(\text{Si},\text{Al})_2$ $(\text{O},\text{OH},\text{F})_7$	TET thin tab	010 perf 001 dist	Cols	H 5, G 3.03	Gel with acids.
1.692	1.665	1.685	.020	THOROGUMMITE $(\text{Th},\text{Ce})(\text{SiO}_4)_{1-x}(\text{OH})_{4x}$	TET u mass earthy	Conch	Yellow, brown	H 3, G 4.0-4.7, infus	Gel with acids.

	1.666	1.747	.081	PETERSITE (Y,Ce,La,Ca)Cu ₆ (PO ₄) ₃ (OH) ₆ ·3H ₂ O	HEX	---	Yellow-green	G 3.41	Dec by HCl. Pleoc, 0 light yellow-green, E green.
1.643 ^ 1.678	1.670	1.768	.098	PARISITE Ca(Ce,La) ₂ (CO ₃) ₂ F	TRIG	0001	Yellow	H 4.5, G 4.0-4.4, infus	Diss by acids. Pleoc, 0 pale yellow, E yellow- brown. El neg.
1.688 ^	1.67	1.70	.03	HINSDALITE (Beudantite grp) PbAl ₃ (PO ₄)(SO ₄)(OH) ₆	TRIG	0001 perf	Cols	H 4.5, G 3.65, infus	Anom biax in 6 basal sectors. El neg.
	1.673	1.678	.005	FILLOWITE Na ₂ Ca(Mn,Fe) ₇ (PO ₄) ₆	MCL ps hex	001 perf	Yellow, brown, cols	H 4.5, G 3.43, F 1.5	Diss by acids.
1.644 ^	1.674	1.770	.096	SYNCHYSITE CaCe(CO ₃) ₂ F	ps hex	---	Red-brown	H 4.5, G 3.9, infus	Diff diss by acids.
1.654 ^ 1.692	1.675	1.705	.030	RHABDOPHANE (Ce,La)PO ₄ ·H ₂ O	HEX fib, mass	Uneven	Brown, gray-green	H 3-4, G 3.3- 4.0, infus	Diss by acids.
1.681 ^	1.675	1.685	.010	BAZIRITE BaZrSi ₃ O ₉	HEX pris	0001	Cols	G 3.82 calc	Fluor whitish-blue in short-wave UV.
1.670 ^	1.678	1.767	.089	PARISITE Ca(Ce,La) ₂ (CO ₃) ₃ F ₂	TRIG	0001	Yellow, brown	H 4-5, G 4.0- 4.4, infus	Diss by acids. Pleoc, 0 pale yellow, E yellow-brown. El neg.
1.653 ^ 1.695	1.678	1.684	.006	FLORENCITE (Crandallite grp) CeAl ₃ (PO ₄) ₂ (OH) ₆	TRIG	0001	Brown	H 5-6, G 3.46, infus	Diff diss by acids.
1.653 ^	1.680	1.698	.018	PLUMBOGUMMITE (Crandallite grp) PbAl ₃ (PO ₄) ₂ (OH) ₅ ·H ₂ O	TRIG gum-like		Yellow, brown, gray	H 4.5, G 4.0- 4.9, F 4-5 (?)	Diss with diff by HNO ₃ .
	1.680	1.695	.015	BROCKITE (Ca,Th,La)(PO ₄)	HEX mass, earthy		Reddish- brown to yellow	G 3.9	Related to Rhabdophane.
1.675 ^	1.681	1.691	.010	BAZIRITE BaZrSi ₃ O ₉	HEX pris	0001	Cols	G 3.82 calc	Fluor whitish-blue in short-wave UV.
	1.688	1.698	.010	WEILERITE (Beudantite grp) BaAl ₃ (AsO ₄)(SO ₄)(OH) ₆ (?)	TRIG crusts	---	White	---	Diss by acids. Opt char unk.
1.67 ^	1.688	1.697	.009	HINSDALITE (Beudantite grp) PbAl ₃ (PO ₄)(SO ₄)(OH) ₆	TRIG	0001 perf	Cols	H 4.5, G 3.65, infus	Anom biax in 6 basal sectors. El neg.
	1.689	1.695	.006	ILIMAUSITE Ba ₂ Na ₄ CeFe ⁺³ Nb ₂ Si ₈ O ₂₈ · 5H ₂ O	HEX	Conch	Brownish- yellow	H ~ 4, G 3.6	---

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
$\sqrt{1.720}$	1.690	1.760	.070	XENOTIME YPO_4	TET	100 perf	Yellow, brown	H 4-5, G 4.7-5.1, infus	Insol in acids. Pleoc wk, O pale yellow, E greenish-yellow.
$\sqrt{1.714}$	1.691	1.719	.028	WILLEMITE Zn_2SiO_4	TRIG	0001, 1120	Green, white, red, brown	H 5.5, G 4.05-4.20, infus	Gel with acids. Fluor bright green in UV. MnO 0.12%.
$\sqrt{1.675}$ $\sqrt{1.721}$	1.692	1.732	.040	RHABDOPHANE-(La) $(\text{La,Ce})\text{PO}_4 \cdot \text{H}_2\text{O}$	HEX mass	Uneven	Brown, gray	H 3-4, G 4.4, infus	Diss by acids.
$\sqrt{1.665}$ $\sqrt{1.78(?)}$	1.692	1.710	.018	THOROGUMMITE $(\text{Th,Ce})(\text{Si,P})(\text{O,OH})_4$	TET u mass, earthy	Conch	Yellow, brown	H 3, G 4.0-4.7, infus	Gel with acids. Forms series with Thorite?
	1.694	1.701	.007	FLORENCITE-(La) (Crandallite grp) $(\text{La,Ce})\text{Al}_3(\text{PO}_4)_2(\text{OH})_6$	TRIG	Splintery	Cols to pale yellow	H 5, G 3.52	Diff diss by acids.
$\sqrt{1.717}$	1.695	1.785	.090	BASTNAESITE $(\text{Ce,Lu})\text{CO}_3\text{F}$	HEX pris	Parting 0001	Yellow, brown, reddish	H 4, G 4.9-5.1, infus	Diff diss by acids. Pleoc wk, abs E > 0.
$\sqrt{1.678}$ $\sqrt{1.713}$	1.695	1.705	.010	FLORENCITE (Crandallite grp) $\text{CeAl}_3(\text{PO}_4)_2(\text{OH})_6$	TRIG	0001	Cols, brown	H 6, G 3.69, infus	Diff diss by acids.
	1.697	1.704	.007	MUIRITE $\text{Ba}_{10}\text{Ca}_2\text{MnTiSi}_{10}\text{O}_{30}$ $(\text{OH,Cl,F})_{10}$	TET	001, 100 indist	Orange	H 2.5, G 3.86, F 3	Dec by acids. Anom purplish-brown interf colors.
	1.701	1.707	.006	KEMMLITZITE (Beudantite grp) $\text{SrAl}_3(\text{AsO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001	Grayish-brown	H 5.5, G 3.63	---
$\sqrt{1.722}$	1.702	1.706	.004	VESUVIANITE $\text{Ca}_{10}\text{Mg}_2\text{Al}_4(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)_5$ $(\text{OH,F})_4$	TET pris	110 poor	Cols, green, brown, yellow	H 6-6.5, G 3.37, F 3	Insol in acids. Pleoc wk. Anom biax. Abnormal interf colors in blue and brown.
	1.704	1.765	.061	GOUDEYITE $\text{Cu}_6(\text{Al,Y})(\text{AsO}_4)_3(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	HEX pris <u>c</u>	---	Yellow-green	H 3-4, G 3.50	Pleoc, O pale yellow-green, E green. Compare Mixite.
$\sqrt{1.722}$	1.705	1.777	.072	AGARDITE $(\text{Y,Ca})\text{Cu}_6(\text{AsO}_4)_3(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	HEX	---	Blue-green	G 3.72	Diss by acids.

1.59 ^	1.707	1.722	.015	AMAKINITE (Brucite ser) (Fe,Mg)(OH) ₂	TRIG	0001 fr uneven	Greenish, turns brown	H 3.5-4, G 2.98, infus	Diss by acids. Fe ₂ O ₃ 31.6, FeO 30.4, MgO 10.1, MnO 3.6%. Compare Brucite at $\omega = 1.59$.
1.730 v	1.71	(1.715)	.005	HIDALGOITE (Beudantite grp) PbAl ₃ (AsO ₄)(SO ₄)(OH) ₆	TRIG mass	---	White	H 4.5, G 3.96	Insol in acids.
1.695 ^	1.713	1.719	.006	FLORENCITE (Crandallite grp) CeAl ₃ (PO ₄) ₂ (OH) ₆	TRIG	0001	Cols, brown	H 6, G 3.4-3.6, infus	Diff diss by acids.
1.691 ^	1.714	1.732	.018	WILLEMITE (Zn,Mn) ₂ SiO ₄	TRIG	0001, 1120	Reddish- brown	H 5, G 4.05- 4.20, infus	Gel with acids. MnO 6.8, FeO 0.2%.
1.75 v	1.714	1.731	.017	OSARIZAWAITE PbCuAl ₂ (SO ₄) ₂ (OH) ₆	TRIG mass	---	Green	G 4.04	Diss by hot HNO ₃ . Anom biax.
	1.715	1.728	.013	JASMUNDITE Ca ₁₁ (SiO ₄) ₄ O ₂ S	TET	---	Dark brown	G 3.03	---
1.695 v	1.717	1.818	.101	BASTNAESITE (Ce,Ln)CO ₃ F	HEX pris	Parting 0001	Yellow, brown, reddish	H 4, G 4.9-5.1, infus	Diff diss by acids. Pleoc wk, abs E > 0.
1.730 v	1.719	1.733	.014	BROMELLITE BeO	HEX	1010 good 1120, 0001 poor	Cols	H 8.5-9, G 3.02, infus	Insol in acids.
1.81 [1.72	---	very wk	ZIRCON "Malacon" ZrSiO ₄	TET	Uneven	Brown	H 6-7, G ~ 4	Metamict. Insol in acids.
1.690 v	1.720	1.820	.100	XENOTIME YPO ₄	TET	100 perf	Yellow to reddish- brown	H 4-5, G 4.7-5.1, infus	Insol in acids. Pleoc wk, 0 pink to pale yellow, E pale greenish-yellow.
1.753 v									
1.692 ^	1.721	1.748	.027	RHABDOPHANE (Ce,Ln)PO ₄ ·H ₂ O	HEX u mass	Uneven	Greenish- yellow, brown	H 3-4, G 3.3-4.0, infus	Diss by acids.
1.705 ^	1.722	1.815	.093	AGARDITE (Y,Ca)Cu ₆ (AsO ₄) ₃ (OH) ₆ · 3H ₂ O	HEX	---	Blue-green	G 3.72	Diss by acids. Compare Mixite at $\omega = 1.730$.
1.702 ^	1.722	1.725	.003	VESUVIANITE Ca ₁₀ Mg ₂ Al ₄ (Si ₂ O ₇) ₂ (SiO ₄) ₅ (OH,F) ₄	TET pris	110 poor	Cols, green, brown, yellow	H 6-6.5, G 3.4, F 3	Insol in acids. Pleoc wk. Anom biax. Abnormal interf colors in blue and brown.
1.736 v	1.724	1.746	.022	CONNELLITE Cu ₁₉ (SO ₄)Cl ₄ (OH) ₃₂ ·3H ₂ O	HEX acic	---	Fine blue, greenish- blue	H 3, G 3.36, F 2.5	Diss by acids.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.780 ^v	1.728	1.730	.002	BRITHOLITE-(Y) ("Abukumalite") (Y,Ca,Ce) ₅ (SiO ₄ ,PO ₄) ₃ (OH,F)	HEX	0001, 10T0, imperf	Reddish-brown	H 6-6.5, G 4.25	Dec by acids. Apatite structural type.
1.717 [^] 1.757	1.730	1.830	.100	BASTNAESITE (Ce,La)CO ₃ F	HEX pris	parting 0001	Yellow, brown, reddish	H 4, G 4.9-5.1, infus	Diff diss by acids.
1.749 ^y	1.730	1.810	.080	MIXITE BiCu ₆ (AsO ₄) ₃ (OH) ₆ ·3H ₂ O	HEX acic	One perf	Emerald- to bluish-green	H 3-4, G 3.8, F 2	Diss by acids. Pleoc very wk in shades of green.
1.71 [^]	1.730	1.735	.005	HIDALGOITE (Beudantite grp) PbAl ₃ (AsO ₄)(SO ₄)(OH) ₆	TRIG	---	Pale green to emerald- green	H 4.5, G 3.95	Insol in acids. MoO ₃ 3.5%.
1.724 [^]	1.736	1.756	.020	CONNELLITE Cu ₁₉ (SO ₄)Cl ₄ (OH) ₃₂ ·3H ₂ O	HEX acic	---	Fine blue, greenish- blue	H 3, G 3.36, F 2.5	Diss by acids.
	1.738	1.752	.014	BUTTGENBACHITE Cu ₁₉ (NO ₃) ₂ Cl ₄ (OH) ₃₂ ·2H ₂ O	HEX acic	---	Azure-blue	H 3, G 3.33, F 2	Diss by acids or by NH ₄ OH.
	1.747	1.776	.029	FLEISCHERITE Pb ₃ Ge(SO ₄) ₂ (OH) ₆ ·3H ₂ O	HEX fib	---	White to pale rose	H soft, G 4.2-4.4	---
1.730 [^]	1.749	1.83	.081	MIXITE BiCu ₆ (AsO ₄) ₃ (OH) ₆ ·3H ₂ O	HEX acic	One perf	Blue-green	H 3-4, G 3.8, F 2	Diss by acids.
1.714 [^]	1.75	~1.76	~.01	OSARIZAWAITE Pb(Cu,Al) ₃ (SO ₄) ₂ (OH) ₆	TRIG	---	Green	---	Diss by hot HNO ₃ . Anom biax.
1.720 [^] 1.783	1.753	~1.84	~.09	CHERNOVITE (Xenotime ser) Y(As,P)O ₄	TET pris	100 perf	Brown, yellow	H 4-5, G 4.8, infus	Insol in acids. Forms a series with Xenotime.
	1.756	1.802	.046	BENITOITE BaTiSi ₃ O ₉	HEX pyram or tab	10T1 imperf fr conch	Blue	H 6, G 3.65, F 3	Insol in acids. Pleoc, 0 cols, E purplish-blue.
1.730 [^]	1.757	1.864	.107	HYDROXYLBASTNAESITE (Ce,La)CO ₃ (OH,F)	HEX pris	11T0 imperf	Reddish- brown, yellow	H 4, G 4.7, infus	Slowly diss by acids.
1.754 [□] 1.761	1.757	---	very wk	MCGOVERNITE (Mn,Mg,Zn) ₂₂ (AsO ₃)(AsO ₄) ₃ (SiO ₄) ₃ (OH) ₂	TRIG	0001 mic	Reddish- bronze	H 3, G 3.72	El neg.

	1.759	1.760	.001	THEOPHRASITE $\text{Ni}(\text{OH})_2$	TRIG	0001 perf	Emerald-green	H 3.5, G 4.00	Pleoc wk.
	1.765	1.800	.035	HENRITERMIERITE (Garnet grp) $\text{Ce}_3(\text{Mn,Al})_2(\text{SiO}_4)_2(\text{OH})_4$	TET	Conch	Clove-brown	G 3.34, fus	Dec by HCl. Anom biax. Pleoc wk, 0 pale yellow, E lemon-yellow.
1.76 □ 1.78 v 1.855	1.765	1.782	.017	STILLWELLITE $(\text{Ce,La,Ca})\text{BSiO}_5$	TRIG	Conch	Pinkish-gray, brown	H 5-6, G 4.6	Anom biax, 2V 0-6°.
	1.77	---	.03-.04	LUSUNGITE (Crandallite grp) $(\text{Sr,Pb})\text{Fe}^{+3}(\text{PO}_4)_2(\text{OH})_5 \cdot \text{H}_2\text{O}$	TRIG	---	Dark brown	---	Pleoc wk.
□ 1.795	1.778	1.801	.023	CONICALCITE $\text{CaCu}(\text{AsO}_4)(\text{OH})$	ORTH fib	Uneven	Pistachio-to emerald-green	H 4-5, G 4.13, F 3	In section green, pleoc wk. U biax.
1.692(?) v 1.837	1.78	1.79	.01	THORITE ThSiO_4	TET	110 dist	Brown, yellow	H 5, G 4.4-5.7, infus	Gel with acids.
1.728 ^	1.780	1.783	.003	BRITHOLITE-(Y) ("Abukumalite") $(\text{Y,Ca,Ce})_5(\text{SiO}_4)_3(\text{PO}_4)_3(\text{OH,F})$	HEX	0001, 10T0, imperf	Reddish-brown	H 6, G 4.35	Dec by acids. Apatite structural type.
1.753 ^	1.783	1.879	.096	CHERNOVITE YAsO_4	TET pris	100 perf	Yellow	H 4-5, G 4.87, infus	Insol in acids. Pleoc wk, 0 cols, E pale pinkish-yellow.
	1.794	1.803	.009	CARYINITE $(\text{Na,Ca,Pb})_3(\text{Mn,Mg,Fe}^{+3})_4(\text{AsO}_4)_4(?)$	MCL u mass	110, 010	Brown	H 3-4, G 4.25, F 2-3	Diss by acids. Orange to red in section, not pleoc.
	1.800	1.845	.045	STISHOVITE SiO_2	TET	---	Cols	G 4.35, infus	Insol in acids, very slowly diss by HF.
1.785 ┌	1.80	1.81	.01	NIGERITE $(\text{Zn,Mg,Fe})(\text{Sn,Zn})_2(\text{Al,Fe})_{12}\text{O}_{22}(\text{OH})_2$	TRIG	---	Dark brown to yellow	H 8-9, G 4.08-4.51	Insol in acids.
	1.805	---	wk	KRAISSLITE $(\text{Mn,Mg})_{24}\text{Zn}_3\text{Fe}^{+3}(\text{AsO}_3)_2(\text{AsO}_4)_3(\text{SiO}_4)_6(\text{OH})_{18}$	HEX	0001 perf	Deep coppery brown	H 3-4, G 3.88	Brittle.
└ 1.82	1.806	1.808	.002	CERITE $(\text{Ce,Ca})_9(\text{Mg,Fe})\text{Si}_7(\text{O,OH,F})_{28}$	TRIG	Uneven	Brown, reddish	H 5.5, G 4.75, infus	Gel with acids. Commonly biax, 2V 0-25°.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
	1.81	2.02	.21	GAUDEFROYITE $\text{Ca}_4\text{Mn}^{+3}_{3-x}(\text{BO}_3)_3(\text{CO}_3)(\text{O,OH})_3$	HEX pris	Pris	Black	H 6, G 3.35	Diss by HCl. Pleoc, 0 reddish-brown, E orange-red.
1.72 ^ 1.878	1.81	1.86	.05	ZIRCON ZrSiO_4	TET pris	110 fr uneven	Brown, yellow	H 7.5, G 4.56, infus	Insol in acids. El pos. Partly metamict.
v 1.85	1.81	1.84	.03	CHROMATITE CaCrO_4	TET	Conch	Lemon-yellow	---	Diss by acids.
	---	---	mod	DOWNEYITE SeO_2	TET	---	Cols to yellow	G calc 4.15	Sol in H_2O , very hygroscopic. Mean \bar{n} calc = 1.84.
1.78 ^	1.837	1.898	.061	THORITE $(\text{Th,U})\text{SiO}_4$	TET	110	Green, brown	H 5, G 6.63, infus	Gel with acids.
	1.842	1.848	.006	PARATACAMITE $\text{Cu}_2\text{Cl}(\text{OH})_3$	TRIG	10Tl good	Green	H 3, G 3.74, F easy	Diss by acids or NH_4OH . Tw 10Tl, poly.
	1.85	1.93	.08	ZEMANNITE $(\text{Zn,Fe})_2(\text{TeO}_3)_3\text{Na}_x\text{H}_{2-x}\cdot y\text{H}_2\text{O}$	HEX pris	---	Light to dark brown	Soft, G 4.36	Pleoc, E yellow-brown, 0 reddish brown, abs 0 > E.
1.81 ^	1.85	1.88	.03	CHROMATITE CaCrO_4	TET	Pris poor, fr conch	Lemon-yellow	---	Diss by acids.
1.77 ^	1.855	---	.03-.04	LUSUNGITE (Crandallite grp) $(\text{Sr,Pb})_3\text{Fe}_3(\text{PO}_4)_2(\text{OH})_5\cdot \text{H}_2\text{O}$	TRIG	---	Dark brown	---	Pleoc wk.
v 1.908	1.855	1.860	.005	BYSTROMITE MgSb_2O_6	TET u mass	---	Blue-gray	H 7, G 5.5 calc	Pleoc, 0 straw yellow, E brownish-yellow. Opt char unk.
	1.864	1.88	.016	WELINITE $(\text{Mn}^{+4},\text{W})_{1-x}(\text{Mn}^{+2},\text{W,Mg})_{3-y}\text{Si}(\text{O,OH})_7$	HEX	0001 poor to dist	Red-brown to reddish black	H 4, G 4.47	Dec by HCl. Blood-red in section. Anom biax with small 2V.
1.81 ^ 1.925	1.878	1.895	.017	ZIRCON ZrSiO_4	TET pris	110, fr uneven	Brown, yellow	H 7, G 4.62, infus	Insol in acids. El pos.

	1.880	1.884	.004	MORELANDITE (Apatite grp) $\text{Ba}_5(\text{AsO}_4)_3\text{Cl}$	HEX	0001 poor	Light yellow to gray	H 4.5, G 5.33	Diss by cold HNO_3 .
	1.89	2.00	.11	DENNINGITE (Mn,Zn) Te_2O_5	TET platy	001 perf fr conch	Cols to pale green	H 4, G 5.05, F easy	Diss by cold HCl, insol in HNO_3 . Anom biax, 2V 0- 15°.
	1.898	1.915	.017	HUEGELITE $\text{Pb}_2(\text{UO}_2)_3(\text{AsO}_4)_2(\text{OH})_4 \cdot$ $3\text{H}_2\text{O}$	MCL tab	100 good	Brown to orange- yellow	G 5.0	Diss by HNO_3 . Pleoc, 0 yellow, E cols to pale yellow. Anom interf color.
	1.90	2.12	.22	TRIPPKEITE CuAs_2O_4	TET	100 perf 110 good	Bluish- green	Soft, G 4.8, F easy	Diss by acids. Breaks into flexible pieces. Bluish-green in section, not pleoc.
1.855 ^	1.908	1.915	.007	BYSTROMITE MgSb_2O_6	TET u mass	---	Blue-gray	H 7, G 5.5-5.7	Diss by concd HCl + KI. Opt char unk.
	1.910	1.945	.035	GANOMALITE $\text{Pb}_6(\text{Ca,Mn})_4(\text{Si}_2\text{O}_7)_3(\text{OH})_2$	HEX tab 0001	0001, 1010 perf	Gray	H 3, G 5.74, F 3 (?)	Gel with acids. Anom biax. Biref variable.
v 1.930	1.912	1.925	.013	SCHEELITE CaWO_4	TET oct or tab	101 dist	White, yellow, gray, brown	H 4.5-5, G 6.1, F 5	Dec by HCl with sepn of yellow WO_3 . Fluor bright blue (purple) to yellow in UV (purple when Mo is present).
v 1.945	1.913	1.923	.010	NASONITE $\text{Pb}_6\text{Ca}_4(\text{Si}_2\text{O}_7)_3\text{Cl}_2$	HEX pris	0001, 1010 imperf	White	H 4, G 5.5, F easy	---
1.878 ^ 1.960	1.925	1.983	.058	ZIRCON ZrSiO_4	TET pris	110, fr uneven	Brown, cols, yellow, green, purple	H 7.5, G 4.70, infus	Insol in acids. El pos.
1.912 ^ 1.951	1.930	1.947	.017	SCHEELITE $\text{Ca(W,Mo)}\text{O}_4$	TET oct or tab	101 dist	White, brown	H 4.5-5, G 6.1, F 5	Dec by HCl with sepn of yellow WO_3 . Fluor yellow in UV.
	1.94	2.16	.22	BAOTITE $\text{Ba}_4(\text{Ti,Nb})_8\text{Si}_4\text{O}_{28}\text{Cl}$	TET	2 clvs	Light brown to black	H 6, G 4.42-4.71	Pleoc, 0 cols to brown, E greenish-yellow to dark red-brown.
1.913 ^	1.945	1.970	.025	NASONITE $\text{Pb}_6\text{Ca}_4(\text{Si}_2\text{O}_7)_3\text{Cl}_2$	HEX pris	0001, 1010 imperf	White	H 4, G 5.5, F easy	---
	1.948	1.958	.010	HEDYPHANE (Apatite grp) $(\text{Pb,Ca})_5(\text{AsO}_4)_3\text{Cl}$	HEX mass, fib	2 clvs	White	H 4, G 5.7, F 1	Diss by HNO_3 .
1.930 ^ 1.974	1.951	1.967	.016	POWELLITE (Scheelite ser) $\text{Ca(Mo,W)}\text{O}_4$	TET pyram	---	White, yellow	H 4, G 5.3, F 5	Dec by HCl with sepn of yellow WO_3 . Data for synth. CaMoO_4 60, CaWO_4 40%.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.925 ^	1.960	2.015	.055	ZIRCON ZrSiO_4	TET pris	110 fr uneven	Brown, yellow, cols	H 7.5, G 4.65 infus	Insol in acids. El pos.
	1.96	---	wk	DIXENITE $\text{Cu}^{+1}\text{Mn}_{14}\text{Fe}^{+3}(\text{AsO}_3)_5(\text{AsO}_4)(\text{SiO}_4)_2(\text{OH})_6$	TRIG plates	0001 mic	Dark brown	H 3-4, G 4.20-4.36	Dec by HCl. Glowing red in transmitted light.
	1.973Na 1.956Li	2.656Na 2.601Li	.683 .645	CALOMEL Hg_2Cl_2	TET	110 good 011 imperf sectile	Cols, gray	H 1.5, G 6.8-7.15, volat	---
1.951 ^	1.974	1.984	.010	POWELLITE (Scheelite ser) CaMoO_4	TET pyram	Uneven to conch	White	H 4, G 4.23, F 5	Diss by HCl. Data for synth. CaMoO_4 . Fluor yellow in UV.
1.993 □ 2.008	2.001	2.098	.097	CASSITERITE SnO_2	TET pyram	100 imperf fr subconch to uneven	Brown, black, white	H 6-7, G 6.98-7.02, infus	Insol in acids. Tw 011 poly. Sometimes dichroic in brown.
	2.013Na 1.990Li	2.029Na 2.005Li	.016 .015	ZINCITE ZnO	HEX	1010 good 1120, 0001 poor	Deep red	H 4, G 5.66, infus	Diss by acids. In section deep red, not pleoc. MnO 0.27%.
2.01 □ 2.05	2.03	---	high	VOLTZITE $\text{ZnS} + \text{Zn-salt of organic acid}$	HEX spherical globules	---	Yellow, brown, black	H 4-4.5, G 3.1-3.7, infus	Dec by acids, leaving an organic residue.
	>2.0	---	high	BEHIERITE $(\text{Ta}, \text{Nb})\text{BO}_4$	TET pyram	110, 010 dist	Grayish- pink	H 7-7.5, G 7.86, infus	Insol in acids. Mean $n_{\text{calc}} = 2.12$.
	>>1.74	---	~.1	SCHAFARZIKITE FeSb_2O_4	TET pyram	110 perf 100 good	Red to reddish- brown	H 3.5, G 5.5 calc	Pleoc, 0 straw yellow, E brownish yellow. Mean $n_{\text{calc}} = 2.10$.
2.27	2.10	2.26	.16	CALZIRTITE $\text{CaZr}_3\text{TiO}_9$	TET pris	Conch	Yellow- to dark-brown	H 6-7, G 5.07, infus	Partly diss by concd acids. Not pleoc.
	2.116	2.143	.027	PHOSGENITE $\text{Pb}_2\text{CO}_3\text{Cl}_2$	TET pris tab	001, 110 dist	White, gray, yellow	H 2-3, G 6.13, F 1	Diss with eff by HNO_3 . Pleoc wk, 0 reddish, E greenish.
	2.13	2.21	.08	PENFIELDITE $\text{Pb}_2\text{Cl}_3(\text{OH})$	HEX pris	001 dist	White	G 5.9-6.6, F 1	Diss by HNO_3 . Tw.
	>2.1	---	---	PRIDERITE $(\text{K}, \text{Ba})(\text{Ti}, \text{Fe})_8\text{O}_{16}$	TET pris	001 perf pris fair	Reddish- black	G 3.86, infus	Pleoc, 0 deep reddish-brown, E deep reddish-brown to black.

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2.15	2.275	.125	RANCIEITE (Ca,Mn ⁺²)Mn ⁺⁴ ₄ O ₉ ·3H ₂ O	HEX (?) mass	---	Dark brown, streak purplish	Soft. G 3.3	Diss by HCl. Opt char unk.
2.18	---	---	RUSSELLITE Bi ₂ WO ₆	TET mass	---	Yellow	H 3.5, G 7.37	Opt char unk.
2.18	2.19	.01	IODARGYRITE AgI	HEX thin plates	0001 perf sectile	Yellow, green, brown	H 1-1.5, G 5.6- 5.7, F 1	Becomes isotropic at 146°C. Abnormal green interf colors. Anom biax, 2V small. El neg.
2.19	2.21	.02	KLEINITE Hg ₂ N(Cl,SO ₄)·xH ₂ O	HEX short prisms	0001 easy uneven, 10T0 imperf	Yellow, orange	H 3.5-4, G ~ 8, volat	Diss by acids. Uniax above 130°C, biax neg at room temp.
2.19Li	2.21Li	.02	MACKAYITE FeTe ₂ O ₅ (OH)	TET	Subconch	Green to yellow-green	H 4.5, G 4.86, fus	Pleoc, 0 green, E yellow.
>2.0	---	---	ORDONEZITE ZnSb ₂ O ₆	TET tw	Conch	Pale to dark brown	H 6.5, G 6.64	Insol in acids. Tw pl (013). Mean <u>n</u> calc = 2.28.
2.21	---	wk	ZAVARITSKITE BiOF	TET	---	Gray	G (9.2)	Opt sign unk.
2.27	2.43	.16	PARATELLURITE TeO ₂	TET pyram	102 perf 110 dist	Grayish- yellow	H 3, G 5.60, fus	---
2.27Li	2.42Li	.15	TAPIOLITE Fe(Ta,Nb) ₂ O ₆	TET	---	Black	H 6, G 7.3-8.2, infus	Red-brown in section. Pleoc str, 0 yellow to reddish-brown, E nearly opaque.
2.27	2.36	.09	CALZIRITE CaZr ₃ TiO ₉	TET pris	Conch	Yellow- to dark-brown	H 6-7, G 5.0, infus	Partly diss by concd acids.
2.35	2.47	.12	TAPIOLITE Fe(Ta,Nb) ₂ O ₆	TET	---	Black	H 6-7, G 7.49, infus	---
2.36Na 2.33Li	2.38Na 2.35Li	.02 .02	WURTZITE (Zn,Fe)S	HEX pris	11T0 good 0001, 10T0 poor	Brownish- black	H 3.5-4, G 3.98, F diff	Diss by acids. Pleoc wk.
2.37Li	2.42Li	.05	FREUDENBERGITE Na ₂ (Ti,Fe) ₈ O ₁₆	MCL ps hex	2 good	Brownish- black	G 4.3	Insol in acids. Pleoc, 0 dark brown, E yellow- brown.
2.374Li	2.395Li	.021	WURTZITE (Zn,Cd)S	HEX pris	11T0 good 0001 poor	Green	H 3.5-4, G 4.44, F diss	Diss by acids. Data for 41.5 mol % CdS. Fluor yellow in UV.
2.39Li	2.41Li	.02	GREENOCKITE (Cd,Zn)S	HEX pris	10T0 good 11T0, 0001 poor	Orange	H 3-3.5, G 4.53, infus	Diss by acids. Data for 57.6 mol % CdS. Fluor orange-red in UV.

Table 4. Uniaxial positive minerals (continued)

Other entries	Refractive index		Birefringence	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
2.39 [^]	2.50Na 2.43Li	2.53Na 2.45Li	.024 .024	GREENOCKITE CdS	HEX pris	10T0 good 1120, 0001 poor	Yellow, green	H 3-3.5, G 4.9- 5.0, infus	Diss by acids. Data for synth CdS.
	2.45	2.51Li	.06	DERBYLITE Fe ₄ Ti ₃ SbO ₁₃ (OH)	MCL pris	Conch	Black	H 5, G 4.53, infus	Insol in acids. Not pleoc.
	2.476	2.485	.009	CHANGBAIITE PbNb ₂ O ₆	TRIG	0001 perf	Brown, yellow- brown, creamy-white	H 5-5.5, G 6.48	---
2.65 ^v	2.61Na 2.56Li	2.90Na 2.84Li	.287 .28	RUTILE TiO ₂	TET pris	110 dist 100 less so	Brown, red, yellow	H 6-6.5, G 4.22, infus	Insol in acids. Pleoc wk, abs 0 < E. El pos.
2.61 [^]	2.65	2.80	.15	RUTILE, var Ilmenorutile (niobian) (Ti,Nb,Fe)O ₂	TET pris	110 dist 100 less so	Black	H 6-6.5, G 4.36, infus	Insol in acids. Pleoc wk in brown, abs 0 < E.
	2.65Na 2.63Li	2.69Na 2.67Li	.043 .040	MOISSANITE SiC	HEX plates 0001	Conch	Green, blue- black, black	H 9.5, G 3.22, infus	Insol in acids. Pleoc, 0 light blue, E deep indigo blue. Data for synth compd.
	2.85Na 2.82Li	3.20Na 3.15Li	.347Na .327Li	CINNABAR HgS	TRIG	10T0 perf	Cochineal- red	H 2-2.5, G 8.09, volat	Streak scarlet. Disp very high. Circular polarization.

Table 5. Uniaxial negative minerals

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
	1.312	1.309	.003	MALLADRITE Na_2SiF_6	TRIG	---	Cols, rose	G 2.71 F 1	Sol in hot H_2O .
	1.327	---	very wk	VILLIAUMITE NaF	CUB ps tet, u mass	001 perf	Carmine-red	H 3.5 G 2.79 F 1.5	Sol in H_2O . Pleoc, 0 carmine-red to pink, E yellow.
	1.349	1.342	.007	CHIOLITE $\text{Na}_5\text{Al}_3\text{F}_{14}$	TET sq prisms	001 perf 011 dist	White	H 3.5-4 G 3.00 F 1.5	Diss by acids.
	1.388	1.385	.003	COLQUIRIITE CaLiAlF_6	TRIG	Conch	Cols, white	H 4 G 2.94	---
	1.406	1.309	.097	BARARITE $(\text{NH}_4)_2\text{SiF}_6$	HEX	0001 perf	White	H 2.5 G 2.15 volat	Sol in H_2O .
	1.458	1.432	.026	MENDOZITE $\text{NaAl}(\text{SO}_4)_2 \cdot 11\text{H}_2\text{O}$	MCL fib	100 good	Cols	H 3 G 1.75 F 1	Sol in H_2O .
1.489	1.462	1.460	.002	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG rhombs	10 $\bar{1}$ 1 dist	White, red	H 4.5 G 1.97 F 3	Dec by HCl with sepn of slimy silica. Anom biax.
1.491	1.464	1.458	.006	ETTRINGITE $\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12} \cdot 26\text{H}_2\text{O}$	HEX acic	10 $\bar{1}$ 0 perf	Cols, white	H 2-2.5 G 1.77 F 3	Diss by acids. Indices increase on dehydration.
	1.474	1.436	.038	HUMBERSTONITE $\text{K}_3\text{Na}_7\text{Mg}_2(\text{SO}_4)_6(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$	TRIG platy 0001	0001 perf	Cols	G 2.25	Diss by H_2O . El pos.
1.485	1.474	1.470	.004	HERSCHELITE (Zeolite grp) $(\text{Na}, \text{Ca}, \text{K})\text{AlSi}_2\text{O}_6 \cdot 3\text{H}_2\text{O}$	TRIG plates	---	White	H 4 G 2.06 F 3	Dec by acids.
1.465 1.484	1.477	1.475	.002	GMEINITE (Zeolite grp) $(\text{Na}_2, \text{Ca})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	HEX rhombs	10 $\bar{1}$ 0 good 0001	Cols, rose	H 4.5 G 2.04 F 3	Dec by acids. Anom biax with small 2V. Tw axis <u>c</u> .
	1.479	1.408	.071	ZHEMCHUZHNIKOVITE $\text{NaMg}(\text{Al}, \text{Fe})(\text{C}_2\text{O}_4)_3 \cdot 8\text{H}_2\text{O}$	TRIG acic	0001 fair	Smoky green	H 2 G 1.69 infus	An oxalate. Sol in H_2O . Pleoc, 0 greenish-yellow, E reddish-violet.
	1.481	1.461	.020	HANKSITE $\text{KNa}_{22}(\text{SO}_4)_9(\text{CO}_3)_2\text{Cl}$	HEX short prisms	0001 good	White	H 3-3.5 G 2.56 F 1.5	Easily sol in H_2O .

	1.487	1.484	.003	CRISTOBALITE SiO_2	TET ps cub, oct	---	Cols	H 6-7 G 2.33 infus	Insol in acids. Some samples are biax.
1.479 □ 1.493	1.487	1.486	.001	ANALCIME (Zeolite grp) $\text{NaAlSi}_2\text{O}_6 \cdot \text{H}_2\text{O}$	CUB, 211	001 poor	Cols, white	H 5 G 2.25 F 3.5	Dec by acids. Commonly biax. Lam tw on 001 and 110.
1.495 ┘	1.489	1.486	.003	OFFRETITE (Zeolite grp) $(\text{K}_2, \text{Ca})_5\text{Al}_{10}\text{Si}_{26}\text{O}_{72} \cdot 30\text{H}_2\text{O}$	HEX pris	0001	Cols, white	G 2.13 F 3	Dec by acids. El neg.
1.462 ^ 1.518	1.489	1.486	.003	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG rhombs	10T1 dist	White, red	H 4.5 G 2.06 F 3	Dec by HCl with sepn of slimy silica. Anom biax.
1.510 v	1.489	1.487	.002	LEVYNE (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG	0221 dist	White	H 4 G 2.14 F 2-2.5	Gel with acids.
	1.490	1.471	.019	LOEWEITE $\text{Na}_{12}\text{Mg}_7(\text{SO}_4)_{13} \cdot 15\text{H}_2\text{O}$	TRIG	Conch	White, yellow	H 2.5-3 G 2.40 F 1.5	Sol in H_2O . Opt anom.
	1.490	1.476	.014	WARDSMITHITE $\text{Ca}_5\text{MgB}_{24}\text{O}_{42} \cdot 30\text{H}_2\text{O}$	HEX platy	10T0 perf 0001 (?) good	White	H 2.5 G 1.88 F easy	Diss by acids.
8 1.464 ^	1.491	1.470	.021	ETTRINGITE $\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12} \cdot 26\text{H}_2\text{O}$	HEX acic	---	Cols, white	H 2-2.5 G 1.77 F 3	Diss by acids. Indices increase on dehydration.
	1.492	1.475	.017	CHARLESITE $\text{Ca}_6(\text{Al}, \text{Si})_2(\text{SO}_4)_2\text{B}(\text{OH}, \text{O})_{16} \cdot 26\text{H}_2\text{O}$	HEX	10T0 perf	Cols	H 2.5 G 1.77	---
	1.493	1.482	.011	WERMLANDITE $\text{Ca}_2\text{Mg}_{14}(\text{Fe}^{+3}, \text{Al})_4(\text{CO}_3)(\text{OH})_{42} \cdot 14-15\text{H}_2\text{O}$	HEX	0001 perf	Greenish- gray	H 1.5 G 1.93	Diss by HCl. In part biax, 2V 3-5°.
1.507 v	1.494	1.467	.027	THAUMASITE $\text{Ca}_3\text{Si}(\text{OH})_6(\text{CO}_3)(\text{SO}_4) \cdot 12\text{H}_2\text{O}$	HEX fib columnar	---	White	H 3.5 G 1.87 infus	Dec by acids with eff. El neg.
	1.494	1.489	.005	GOBBINSITE (Zeolite grp) $\text{Na}_4(\text{Ca}, \text{K}_2)\text{Al}_6\text{Si}_{10}\text{O}_{32} \cdot 12\text{H}_2\text{O}$	TET fib	---	White	G 2.19	El neg.
	1.495	1.460	.035	ACETAMIDE CH_3CONH_2	TRIG	Conch	Cols to grey	H 1-1.5 G 1.17 F easy	Diss by H_2O .

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.490 √ 1.515	1.499	1.493	.006	VISHNEVITE (Cancrinite grp) $\text{Na}_6\text{Ca}_2\text{Al}_6\text{Si}_6\text{O}_{24}$ $(\text{SO}_4, \text{CO}_3, \text{Cl}_2)_{2-4}$	HEX	10T0 perf	Cols	H 5 G 2.40 F 2	Gel with acids. SO_3 4.7, CO_2 2.2%. Some samples are isotropic, some uniax pos.
	1.502	1.449	.053	UNGEMACHITE $\text{K}_3\text{Na}_8\text{Fe}^{+3}(\text{SO}_4)_6(\text{OH})_2 \cdot 10\text{H}_2\text{O}$	TRIG	0001 perf	Cols, pale yellow	H 2.5 G 2.29	Diss by acids. El clv pos.
	1.504	1.49	.01	NYEREREITE $(\text{Na}, \text{K})_2\text{Ca}(\text{CO}_3)_2$	ORTH ps hex, tab	---	Cols	G 2.41-2.54 fus	Diss by acids with eff. Tw poly.
	1.505	1.486	.019	SACROFANITE (Cancrinite grp) $(\text{Na}, \text{Ca}, \text{K})_9(\text{Si}, \text{Al})_{12}\text{O}_{24}$ $[(\text{OH})_2, \text{SO}_4, \text{CO}_3, \text{Cl}_2]_3 \cdot x\text{H}_2\text{O}$	HEX	0001, 0T10 perf	Cols	H 5.5-6 G 2.42	---
	1.506	1.499	.007	MAZZITE (Zeolite grp) $\text{K}_2\text{CaMg}_2(\text{Al}, \text{Si})_{36}\text{O}_{72} \cdot 28\text{H}_2\text{O}$	HEX acic	---	Cols	G 2.11	---
	1.507	1.464	.043	MCALLISTERITE $\text{Mg}_2\text{B}_{12}\text{O}_{14}(\text{OH})_{12} \cdot 9\text{H}_2\text{O}$	TRIG	0001, 01T2 good	Cols	H 2.5 G 1.87	Diss by acids.
1.494 ^	1.507	1.468	.039	THAUMASITE $\text{Ca}_3\text{Si}(\text{OH})_6(\text{CO}_3)(\text{SO}_4) \cdot 12\text{H}_2\text{O}$	HEX fib columnar	---	White	H 3.5 G 1.87 infus	Dec by acids with eff. El neg.
	1.507	1.502	.005	TARANAKITE $\text{KAl}_3(\text{PO}_4)_3(\text{OH}) \cdot 9\text{H}_2\text{O}$	TRIG u mass	---	White	Soft G 2.09-2.15 F easy	Diss by acids.
√ 1.515	1.507	1.505	.002	GARRONITE (Zeolite grp) $\text{Na}_2\text{Ca}_5\text{Al}_{12}\text{Si}_{20}\text{O}_{64} \cdot 27\text{H}_2\text{O}$	ORTH ps tet radiating	2 at 90°	Cols	G 2.15	---
	1.510	1.486	.024	RETGERSITE $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$	TET tab	001 perf	Emerald-green	H 2.5 G 2.05	Sol in H_2O .
√ 1.530	1.510	1.486	.024	FLUOBORITE $\text{Mg}_3\text{BO}_3(\text{F}, \text{OH})_3$	HEX fib	0001 indist	White	H 3.5 G 2.96	Diff diss by acids. Comp near F end member.
1.489 ^	1.510	1.502	.008	LEVYNE (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG	02Z1 dist	White	H 4 G 2.14 F 2-2.5	Gel with acids.

1.52	1.511	1.495	.016	HYDROTALCITE $\text{Mg}_6\text{Al}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	TRIG plates	0001 perf	White, luster pearly	H 2 G 2.06 infus	Diss by acids. El pos. Anom biax.
	1.513	1.470	.043	ARCHERITE $(\text{K}, \text{NH}_4)\text{H}_2\text{PO}_4$	TET	---	Cols	Soft G 2.23	Sol in H_2O .
	1.515	1.417	.098	STEPANOVITE $\text{NaMgFe}^{+3}(\text{C}_2\text{O}_4)_3 \cdot 9\text{H}_2\text{O}$	TRIG	---	Yellow-green	H 3 G 1.61	An oxalate. Diss by H_2O . Pleoc, E cols, 0 yellow-green.
1.499 ^ 1.518 pos v 1.528	1.515	1.496	.019	CANCRINITE (Cancrinite grp) $\text{Na}_6\text{Ca}_2\text{Al}_6\text{Si}_6\text{O}_{24}(\text{CO}_3)_2$	HEX	10T0 perf	Cols	H 5 G 2.42 F 2	Gel with acids. CO_2 7.0, SO_3 1.4, Cl 0.4%.
1.507 ^	1.515	1.512	.003	GARRONITE (Zeolite grp) $\text{Na}_2\text{Ca}_5\text{Al}_{12}\text{Si}_{20}\text{O}_{64} \cdot 27\text{H}_2\text{O}$	ORTH ps tet radiating	2 at 90°	Cols	G 2.15	---
1.533 v	1.516	1.470	.046	BEIDELLITE (Smectite grp) $(\text{Na}, \text{Ca})_{0.33}\text{Al}_2(\text{Si}, \text{Al})_4\text{O}_{10}$ $(\text{OH})_2 \cdot x\text{H}_2\text{O}$	MCL u mass	001 mic	White, brown, yellow	H 1.5 G 2.6 infus	2V 0-16°. El clv pos. Indices increase on stand- ing in certain oils.
	1.517	~1.501	~.016	MEIXNERITE $\text{Mg}_6\text{Al}_2(\text{OH})_{18} \cdot 4\text{H}_2\text{O}$	TRIG	0001 perf	Cols	G 1.9 infus	Diss by acids.
1.489 ^	1.518	1.515	.003	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	TRIG rhombs	10T1 dist	White, red	H 4.5 G 2.18 F 3	Dec by HCl with sepn of slimy silica. Anom biax.
	1.519	1.503	.016	TEEPLEITE $\text{Na}_2\text{B}(\text{OH})_4\text{Cl}$	TET tab	Uneven	Cols	H 3-3.5 G 2.08 F 1	Diss by H_2O .
	1.52	1.51	.01	MOUNTKEITHITE $(\text{Mg}, \text{Ni})_{11}(\text{Fe}^{+3}, \text{Cr}, \text{Al})_3(\text{OH})_{24}$ $(\text{SO}_4, \text{CO}_3)_{3.5} \cdot 11\text{H}_2\text{O}$	HEX	0001 perf	Pale pink to white	H soft G 2.12	Diss by acids. Pleoc pale pink to cols.
	1.520	1.512	.008	TACHYHYDRITE $\text{CaMg}_2\text{Cl}_6 \cdot 12\text{H}_2\text{O}$	TRIG	10T1 perf	Wax- to honey- yellow	H 2 G 1.67 F 1	Easily sol in H_2O . Deliq.
	1.521	1.517	.004	CARLETONITE $\text{KNa}_4\text{Ca}_4\text{Si}_8\text{O}_{18}(\text{CO}_3)_4(\text{F}, \text{OH}) \cdot$ H_2O	TET	001 perf 110 good	Pink to blue	H 4-4.5 G 2.45	Blue var pleoc, 0 pale blue, E pale pinkish-brown.
	1.524	1.510	.014	MANASSEITE $\text{Mg}_6\text{Al}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	HEX foliated	0001 perf	White	H 2 G 2.05 infus	Diss by acids. El clv pos.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
	1.520 (mean \bar{n})		.005-.006	VIRGILITE $\text{Li}_x\text{Al}_x\text{Si}_{3-x}\text{O}_6$ ($x = 0.5$ to 1.0)	HEX	---	Cols	G 2.46 calc	---
	1.525	1.459	.066	NATROFAIRCHILDITE $\text{Na}_2\text{Ca}(\text{CO}_3)_2$	ORTH (?) ps hex	One perf 001 (?)	White	H 2.5	Poly tw. In part biax. = Nyerereite (?), $\omega = 1.504$.
	1.525	1.480	.045	BIPHOSPHAMMITE $(\text{NH}_4, \text{K})\text{H}_2\text{PO}_4$	TET	---	White to dark brown	G 2.04 volat	Diss in H_2O .
1.537 v	1.526	1.495	.031	SLAVIKITE $\text{NaMg}_2\text{Fe}_5(\text{SO}_4)_7(\text{OH})_6 \cdot 33\text{H}_2\text{O}$	TRIG tab	---	Yellow-green	G 1.95	Diss by acids. Pleoc, 0 lemon-yellow, E pale yellow to cols.
1.515 ^	1.528	1.503	.025	CANCRINITE (Cancrinite grp) $\text{Na}_6\text{Ca}_2\text{Al}_6\text{Si}_6\text{O}_{24}(\text{CO}_3)_2$	HEX	1010 perf	Cols	H 5 G 2.48 F 2	Gel with acids. El clv neg. CO_2 6.2, SO_3 0.2, Cl 0.1%.
88 1.510 ^ 1.547	1.530	1.507	.023	FLUOBORITE $\text{Mg}_3\text{BO}_3(\text{F}, \text{OH})_3$	HEX fib	0001 indist	White	H 3.5 G 2.95	Diff diss by acids. F 20.9, H_2O 3.2%.
1.528 ^	1.530	1.528	.002	LIOTTITE (Cancrinite grp) $(\text{Ca}, \text{Na}, \text{K})_8(\text{Si}, \text{Al})_{12}\text{O}_{24}$ $(\text{SO}_4, \text{CO}_3, \text{Cl}, \text{OH})_4 \cdot \text{H}_2\text{O}$	HEX flat prisms	---	Cols	H 5 G 2.56	SO_3 8.7, Cl 2.6, CO_2 2.1%.
1.539 v	1.531	1.527	.004	NEPHELINE $(\text{Na}, \text{K})\text{AlSiO}_4$	HEX	1010 dist 0001 imperf	Cols	H 6 G 2.58 F 3.5	Gel with acids. Luster greasy.
1.546 v	1.532	1.522	.010	MARIALITE (Scapolite grp) $3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf 110 less so	Cols	H 6 G 2.55 F 3	Insol in acids. Na_2O 7.1, CaO 6.3, K_2O 4.3, Cl 1.0, CO_2 2.0%.
1.530 □ 1.536	1.532	1.527	.005	KALIOPHILITE KAlSiO_4	HEX pris	1010 dist	Cols	H 6 G 2.60 F 3.5	Gel with acids.
1.553 v	1.532	1.529	.003	MILARITE (Osumilite grp) $\text{K}_2\text{Ca}_4\text{Al}_2\text{Be}_4\text{Si}_{24}\text{O}_{60} \cdot \text{H}_2\text{O}$	HEX	0001 indist	Cols, pale green, pink	H 5-5.5 G 2.57 F 3	Insol in acids. Basal section shows 6 biax segments.
	1.533	1.498	.035	FAIRCHILDITE $\text{K}_2\text{Ca}(\text{CO}_3)_2$	HEX plates	0001 good	Cols	G 2.45 F 2	Diss by acids with eff. El pos.

1.516 ^ 1.549	1.533	1.502	.031	BEIDELLITE (Smectite grp) $\text{Na}_{0.33}(\text{Al}, \text{Fe}^{+3})_2(\text{Al}, \text{Si})_4\text{O}_{10}$ (OH) ₂ ·xH ₂ O	MCL u mass	001 perf	White, brown, yellow	H 1.5 G 2.6 infus	2V small. El clv pos. Fe ₂ O ₃ 8.7%.
	1.534	---	---	STRAETLINGITE $\text{Ca}_2\text{Al}_2\text{SiO}_7 \cdot 8\text{H}_2\text{O}$	TRIG	001 perf	Cols to light green	G 1.9	---
	1.534	1.514	.020	ZINCALUMINITE $\text{Zn}_6\text{Al}_6(\text{SO}_4)_2(\text{OH})_{26} \cdot 5\text{H}_2\text{O}$	HEX or ps hex, plates	---	White	H 2.5-3 G 2.26 infus	Diss by acids. In part biax with small 2V.
1.554 v	1.535	1.531	.004	KALSILITE (K, Na)(Al, Fe)SiO ₄	HEX	---	Cols	G 2.65 F 3.5	Gel with acids.
1.545 v	1.537	0.490	.047	SCHROECKINGERITE $\text{NaCa}_3(\text{UO}_2)(\text{CO}_3)_3(\text{SO}_4)\text{F} \cdot$ 10H ₂ O	TCL ps hex	001	Greenish- yellow	H 2.5 G 2.54	Diss by H ₂ O. 2V 0-18°. Fluor yellow-green in UV.
1.526 ^	1.537	1.498	.039	SLAVIKITE $\text{NaMg}_2\text{Fe}_5(\text{SO}_4)_7(\text{OH})_6 \cdot 33\text{H}_2\text{O}$	TRIG tab	---	Yellow-green	G 1.45	Diss by acids. Pleoc, 0 lemon-yellow, E pale yellow to cols.
	1.538	1.535	.003	INDIALITE $\text{Mg}_2\text{Al}_4\text{Si}_5\text{O}_{18}$	HEX	---	Cols to blue	H 6-7 G 2.6 infus	Insol in acids. Dimorph with Cordierite.
1.531 ^ 1.546	1.539	1.536	.003	NEPHELINE (Na, K)AlSiO ₄	HEX	10TO dist 0001 imperf	Cols	H 6 G 2.60 F 3.5	Gel with acids. Luster greasy. K ₂ O 5.7, CaO 1.45%.
1.562 v	1.540	1.510	.030	BRUGNATELLITE $\text{Mg}_6\text{FeCO}_3(\text{OH})_{13} \cdot 4\text{H}_2\text{O}$	HEX tab 0001	0001 mic	White, brownish	H 2 G 2.14 infus	Diss by acids. El pos. Pleoc, 0 yellow-red, E cols.
	1.540	1.512	.028	MELLITE (aluminum mellitate) $\text{Al}_2[\text{C}_6(\text{COO})_6] \cdot 16\text{H}_2\text{O}$	TET	011 indist	Honey- yellow, white	H 2-2.5 G 1.64	Diss by HNO ₃ , dec by hot H ₂ O. In part biax with small 2V. Pleoc wk, E cols, 0 pale yellow.
	1.540	1.535	.005	TETRAKALSILITE (K, Na)AlSiO ₄	HEX	---	Cols	G 2.59 F 3.5	Gel with acids.
1.560 v	1.540	1.517	.023	VERMICULITE (Mg, Fe, Al) ₃ (Al, Si) ₄ O ₁₀ (OH) ₂ · 4H ₂ O	MCL ps hex	001 mic	Green, white	H 1.5 G 2.3	Dec by acids. Exfoliates when heated. El clv pos.
	1.542	1.533	.009	IOWAITE $\text{Mg}_4\text{Fe}^{+3}(\text{OH})_8\text{OCl} \cdot 2-4\text{H}_2\text{O}$	HEX	0001 perf	Bluish- green	H 1.5 G 2.11	Diss by acids. Anom biax. El clv pos.
1.537 ^	1.545	1.496	.049	SCHROECKINGERITE $\text{NaCa}_3(\text{UO}_2)(\text{CO}_3)_3(\text{SO}_4)\text{F} \cdot 10\text{H}_2\text{O}$	TCL ps hex	001	Greenish- yellow	H 2.5 G 2.55	Diss by H ₂ O. 2V 0-18°. Fluor yellow-green in UV.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.542 □ 1.553	1.545	1.518	.027	STICHTITE $\text{Mg}_6\text{Cr}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	TRIG plates 0001	0001 perf	Lilac	H 1.5-2 G 2.16 infus	Diss by acids. Anom biax. El pos. Pleoc wk, abs 0 > E.
1.532 ∇ 1.561	1.546	1.542	.004	MARIALITE (Scapolite grp) $3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf 110 less so	Cols	H 6 G 2.62 F 3	Insol in acids. Na_2O 10.5, CaO 4.8, K_2O 1.2, Cl 3.0, CO_2 1.1% (Me_{19}).
1.539 ^	1.546	1.542	.004	NEPHELINE $(\text{Na}, \text{K})\text{AlSiO}_4$	HEX	1010 dist 0001 imperf	Cols	H 6 G 2.65 F 3.5	Gel with acids. Luster greasy. K_2O 8.7, CaO 3.7%.
1.530 ∇ 1.570	1.547	1.522	.025	FLUOBORITE $\text{Mg}_3\text{BO}_3(\text{F}, \text{OH})_3$	HEX fib	0001 indist	White	H 3.5 G 2.92	Diff diss by acids. F 17.6%.
1.545 □ 1.549	1.547	1.536	.011	GYROLITE $\text{Ca}_2\text{Si}_3\text{O}_7(\text{OH})_2 \cdot \text{H}_2\text{O}$	HEX fib	0001 mic	White	H 3-4 G 2.39 F diff	Dec by acids. El clv pos.
1.533 ∇ 1.585	1.549	1.517	.032	BEIDELLITE (Smectite grp) $\text{Na}_{0.33}(\text{Al}, \text{Fe})_2(\text{Al}, \text{Si})_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	MCL u mass	001 mic	Brown, yellow	H 1.5 G 2.6 infus	2V small. El clv pos.
	1.549	1.522	.027	TRUSCOTTITE $(\text{Ca}, \text{Mn})_{14}\text{Si}_{24}\text{O}_{58}(\text{OH})_8 \cdot 2\text{H}_2\text{O}$	HEX	One mic	White	G 2.47	Dec by HCl. Commonly biax, 2V small.
	1.550	1.495	.055	ANTARCTICITE $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$	TRIG	0001 and pris, perf	Cols	G 1.7	Readily sol in H_2O .
∇ 1.570	1.550	1.522	.028	TAENIOLITE (Mica grp) $\text{KLiMg}_2\text{Si}_4\text{O}_{10}\text{F}_2$	MCL ps hex	001 perf	Cols, yellowish	H 3 G 2.82 F 3	Nearly insol in acids. Biax, 2V small. El clv pos.
1.532 ^	1.553	1.549	.004	MILARITE (Osumilite grp) $\text{K}_2\text{Ca}_2\text{Al}_2\text{Be}_4\text{Si}_{24}\text{O}_{60} \cdot \text{H}_2\text{O}$	HEX	0001 indist	Cols, pale green, pink	H 5.5-6 G 2.57 F 3	Insol in acids. Basal section may show 6 biax segments.
1.535 ^	1.554	1.550	.004	KALSILITE $(\text{K}, \text{Na})(\text{Al}, \text{Fe})\text{SiO}_4$	HEX	---	Cols	G 2.66 F 3.5	Gel with acids.
	1.556	1.540	.016	JOURAVSKITE $\text{Ca}_3\text{Mn}^{+4}(\text{SO}_4, \text{CO}_3)_2(\text{OH})_6 \cdot 12\text{H}_2\text{O}$	HEX	Pris, good	Greenish-yellow, orange	H 2.5 G 1.95	In section bright yellow, pleoc wk. Abnormal interf colors.

	1.557	1.529	.028	BARBERTONITE $\text{Mg}_6\text{Cr}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	HEX u mass	0001 perf	Lilac, rose-pink	H 1.5-2 G 2.10 infus	Diss by acids.
1.592 v	1.559	---	wk to med	MERRIHUEITE (Osumilite grp) $(\text{K}, \text{Na})_2(\text{Fe}^{+2}, \text{Mg})_5\text{Si}_{12}\text{O}_{30}$	HEX	---	---	G 2.87 calc	Opt char unk. FeO 10.5%.
	1.559	1.557	.002	HOLTEDAHLLITE $\text{Mg}_2\text{PO}_4(\text{OH})$	HEX	---	Cols	H 4.5-5 G 2.94	---
	1.560	1.507	.053	CHLORALUMINITE $\text{AlCl}_3 \cdot 6\text{H}_2\text{O}$	TRIG	---	Cols to yellowish	---	Sol in H_2O . Deliq.
1.593 v	1.560	1.540	.020	SAPONITE (Smectite grp) $(\text{Na}, \text{Ca})_{0.33}(\text{Mg}, \text{Fe})_3(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	MCL ps hex	001 perf	White, green	H 1-2 G 2.26 F diff	U biax, 2V small. El clv pos. Al_2O_3 7.2, Fe_2O_3 1.0%.
1.540 v 1.583	1.560	1.540	.020	VERMICULITE $(\text{Mg}, \text{Fe}, \text{Al})_3(\text{Al}, \text{Si})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	MCL ps hex	001 mic	Green	H 1.5 G 2.3	Dec by HCl. Exfoliates when heated. El clv pos. Fe_2O_3 4.2, FeO 1.8%.
1.572 v	1.560	1.545	.015	PYROAURITE $\text{Mg}_6\text{Fe}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	TRIG tab 0001 fib	0001 mic	White, yellowish	H 2.5 G 2.12 infus	Diss by acids. El pos. Pleoc, 0 yellow-red, E cols. Dimorph with Sjögrenite.
1.546 v 1.574	1.561	1.546	.015	SCAPOLITE (Scapolite grp) $3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl} -$ $3\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot \text{CaCO}_3$	TET	100 perf 110 less so	Cols, brown	H 6 G 2.63 F 3	Insol in acids. Na_2O 8.6, CaO 8.3, K_2O 1.1, Cl 2.2, CO_2 1.7, SO_3 0.4% (Me_{33}).
1.578 v	1.561	1.549	.012	LOVOZERITE $\text{Na}_3\text{ZrSi}_6(\text{O}, \text{OH})_{18}$	TRIG	Uneven to conch	Dark brown to black	H 5 G 2.64 F 2	Insol in acids. Tw.
1.540 v 1.584	1.562	1.543	.019	BRUGNATELLITE $\text{Mg}_6(\text{Fe}, \text{Cr})\text{CO}_3(\text{OH})_{13} \cdot 4\text{H}_2\text{O}$	HEX tab 0001	0001 mic	Brown	H 2 G 2.1 infus	Diss by acids. Cr_2O_3 4.9-6.2%.
	1.564	1.535	.029	PHLOGOPITE (Mica grp) $\text{K}(\text{Mg}, \text{Fe})_3\text{AlSi}_3\text{O}_{10}(\text{F}, \text{OH})_2$	MCL ps hex, platy	001 perf	Yellow, brown	H 2-2.5 G 2.75 F diff	Insol in acids. 2V 0-10°. El pos.
	1.565	1.535	.030	ZAKHAROVITE $\text{Na}_4(\text{Mn}, \text{Fe})_5\text{Si}_{10}\text{O}_{24}(\text{OH})_6 \cdot 5\text{H}_2\text{O}$	TRIG	0001 perf conch	Yellow	H 2 G 2.61	Strongly electromagnetic.
	1.565	1.560	.005	ZEOPHYLLITE $\text{Ca}_4\text{Si}_3\text{O}_8(\text{OH}, \text{F})_4 \cdot 2\text{H}_2\text{O}$	TCL ps trig	001 perf	Cols, white	H 3 G 2.76 F 2	Gel with acids. Biax, 2V 0-27°. El clv pos.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.576 v	1.567	1.566	.001	BRANNOCKITE (Osumilite grp) $\text{KLi}_3\text{Sn}_2\text{Si}_{12}\text{O}_{30}$	HEX	---	---	G 2.98	Fluor bluish white in UV. K_2O 3.7-4.5, Li_2O 3.75, SnO_2 28.2%.
	1.568	1.563	.005	REYERITE $(\text{Na},\text{K})_4\text{Ca}_{14}(\text{Si},\text{Al})_{24}\text{O}_{60}$ $(\text{OH})_5 \cdot 5\text{H}_2\text{O}$	trig radiating	0001 perf	Col's	G 2.55	Dec by HCl. El clv pos.
	1.568	1.564	.004	BERYL $\text{Be}_3\text{Al}_2\text{Si}_6\text{O}_{18}$	HEX pris	0001 imperf	Green, blue, col's	H 8 G 2.66 infus	Insol in acids. Pleoc variable. El neg. Data for nearly alkali-free mineral.
	1.569	1.547	.022	DESAUTELSITE $\text{Mg}_6\text{Mn}^{+3}_2(\text{CO}_3)(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	TRIG tab	0001 perf	Orange	G 2.13	Diss by acids.
	1.569	1.549	.020	KALISTRONTITE $\text{K}_2\text{Sr}(\text{SO}_4)_2$	TRIG	0001 perf	Col's	H 2 G 3.20	Dec by dil acids.
1.547 ^	1.57(?)	---	wk	LAWRENCITE $(\text{Fe},\text{Ni})\text{Cl}_2$	HEX tab	0001 imperf	Green, brown	G 3.16	A meteorite mineral. Diss in H_2O . Alters readily. El pos.
	1.570	1.534	.036	FLUOBORITE $\text{Mg}_3\text{BO}_3(\text{F},\text{OH})_2$	HEX fib	0001 indist	White	H 3.5 G 2.8	Diff diss by acids. F 14.8%.
	1.550 ^	1.570	1.540	TAENIOLITE (Mica grp) $\text{KLiMg}_2\text{Si}_4\text{O}_{10}\text{F}_2$	MCL ps hex	001 perf	Brown	H 3 G 2.82 F 3	Nearly insol in acids. Biax, $2V \sim 5^\circ$. El clv pos. TiO_2 2.0, FeO 1.9%.
	1.570	1.564	.006	CESANITE $\text{Ca}_2\text{Na}_3(\text{SO}_4)_3(\text{OH})$	HEX	---	Col's	H 2-3 G 2.79	---
	1.572	1.549	.023	PYROAURITE $\text{Mg}_6\text{Fe}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	TRIG tab 0001	0001 mic	White, yellow	H 2.5 G 2.12 infus	Diss by acids. El pos. Pleoc, 0 yellow-red, E col's. Dimorph with Sjögrenite.
1.560 ^	1.573	1.572	.001	STEACYITE $\text{Th}(\text{Ca},\text{Na})_2\text{K}_{1-x}\text{Si}_8\text{O}_{20}$ ($x = 0.2-0.4$)	TET	---	Dark brown	H 5 G (3.02)	---
	1.573	1.550	.023	SJÖGRENITE $\text{Mg}_6\text{Fe}_2\text{CO}_3(\text{OH})_{16} \cdot 4\text{H}_2\text{O}$	HEX plates	0001 perf	Yellowish	H 2.5 G 2.11 infus	Diss by acids. El pos. Dimorph with Pyroaurite.

1.609	1.573	1.572	.001	EKANITE (Th,U)(Ca,Fe,Pb) ₂ Si ₈ O ₂₀	TET	---	Cols, green, brown	G 3.3	---
	1.574	1.547	.027	PORTLANDITE Ca(OH) ₂	HEX	0001 perf	Cols	H 2 G 2.23 infus	Diss by acids. El clv pos. Luster pearly.
1.561 ^ 1.587	1.574	1.549	.025	SCAPOLITE (Scapolite grp) 3NaAlSi ₃ O ₈ ·NaCl - 3CaAl ₂ Si ₂ O ₈ ·CaCO ₃	TET	100 perf 110 less so	Cols, brown	H 5-6 G 2.68 F 4	Insol in acids. Na ₂ O 5.4, CaO 12.5, K ₂ O 1.0, CO ₂ 2.9, Cl 1.4%.
	1.574	1.552	.022	Unnamed P analogue of Troegerite (Autunite grp) (UO ₂) ₃ (PO ₄) ₂ ·8H ₂ O	TET	001 perf	Yellow to greenish	H 3 G 3.76	Diss by acids. El clv pos.
1.571 □	1.574	1.559	.015	SALEEITE (Autunite grp) Mg(UO ₂) ₂ (PO ₄) ₂ ·10H ₂ O	MCL ps tet tab	001 perf	Yellow	H 2-3 G 3.27	Diss by acids. El pos. Pleoc, 0 pale yellow, E cols. Fluor yellow in UV. Biax in part.
	1.575	1.57	.005	CALCIOFERRITE Ca ₂ Fe ₂ (PO ₄) ₃ (OH)·7H ₂ O	MCL scales, nodules	0001 perf	Yellow-green	H 2.5 G 2.53 F easy	Diss by HCl. El clv pos.
1.586 ^	1.575	1.572	.003	AUTUNITE (Autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ ·10-12H ₂ O	TET tab	001 perf 100 indist	Yellow to yellow-green	H 2-2.5 G 3.2 F 3	Diss by acids. Fluor yellow in UV. U biax, 2V 0-30°. El pos.
	1.576	1.546	.030	PARSETTENSITE (K,Na,Ca)(Mn,Al) ₇ Si ₈ O ₂₀ (OH) ₈ ·2H ₂ O (?)	MCL ps hex	001	Copper-red	G 2.59-2.68 fus	Dec by hot HCl. Pleoc, 0 greenish-yellow, E cols.
1.568 ^ 1.584	1.576	1.573	.003	BERYL Be ₃ Al ₂ Si ₆ O ₁₈	HEX pris	0001 imperf	Pale green	H 8 G 2.71 infus	Insol in acids. El neg. Pleoc variable. Fe ₂ O ₃ 0.7, total alkalies 0.7%.
1.561 ^	1.578	1.538	.040	LOVOZERITE Na ₃ ZrSi ₆ (O,OH) ₁₈	TRIG	Uneven to conch	Dark brown to black	H 5 G 2.63 F 2	Insol in acids. Tw.
	1.578	1.548	.030	NOVACEKITE (Autunite grp) Mg(UO ₂) ₂ (AsO ₄) ₂ ·12H ₂ O	TET tab	001 perf 100 dist	Yellow	H 2-2.5 G 3.23	Diss by acids. Fluor yellow in UV. Commonly biax, 2V 0-40°.
	1.578	1.559	.019	SODIUM AUTUNITE (Autunite grp) Na ₂ (UO ₂) ₂ (PO ₄) ₂ ·8H ₂ O	TET plates	001 perf 100 less so	Lemon-yellow	H 2-2.5 G 3.58	Diss by acids. El clv pos. Pleoc, 0 light yellow, E pale yellow. Fluor yellow- green in UV.
	1.578	1.577	.001	ARCTITE Na ₂ Ca ₄ (PO ₄) ₃ F	TRIG	001 perf	Cols	H 5 G 3.13	Diss by acids.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
	1.579	1.577	.002	CLINOCHLORE, var Penninite (Chlorite grp) (Mg,Fe) ₅ Al(Si,Al) ₄ O ₁₀ (OH) ₈	MCL ps trig, plates	001 perf	Green, brown, cols	H 2 G 2.7 F diff	Dec by H ₂ SO ₄ . Biax, 2V small. El clv pos. Pleoc, 0 green, E nearly cols. Abnormal blue interf colors.
	---	1.577	very low	NAMUWITE (Zn,Cu) ₄ (SO ₄)(OH) ₆ ·4H ₂ O	HEX	001 perf	Pale sea-green	G 2.77	Opt sign unk.
	1.580	1.575	.005	DARAPIOSITE (Osumilite grp) KNa ₂ Li(Mn,Zn) ₂ ZrSi ₁₂ O ₃₀	HEX	---	---	H 5 G 2.92	Na ₂ O 3.0, K ₂ O 5.1, ZrO ₂ 5.0, MnO 8.2, ZnO 7.8%.
	1.580	1.485	.095	BERBORITE Be ₂ (BO ₃)(OH,F)·H ₂ O	TRIG	0001 perf	Cols	H 3 G 2.20 infus	Insol in acids. El clv pos.
	1.580	1.550	.030	ZINNWALDITE (Mica grp) KLiFe ⁺² AlSi ₃ AlO ₁₀ (F,OH) ₂	MCL ps trig	001 perf	Brown	H 2.5-4 G 3.01 F 2.5	Insol in acids. 2V 0-30°.
	1.581	---	---	META-ANKOLEITE (Meta-autunite grp) (K ₂ ,Ba,Ca)(UO ₂) ₂ (PO ₄) ₂ ·6H ₂ O	TET thin plates	001 perf 100 dist	Yellow	G 3.54	Diss by acids. El pos. Fluor yellow-green in UV.
	1.582	1.564	.018	SABUGALITE (Autunite grp) HA1(UO ₂) ₄ (PO ₄) ₄ ·16H ₂ O	TET plates	001 perf	Yellow	H 2.5 G 3.20	Diss by acids. El clv pos. Fluor bright yellow in UV. Commonly biax, 2V 0°-med.
1.576 1.603 v	1.583	1.554	.029	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₁₂	MCL and TCL ps hex	001 mic	Brown	H 3.5 G 2.74 F 3.5	Dec by acids. El clv pos. Pleoc, E pale yellow, 0 green. FeO 26.7, Fe ₂ O ₃ 2.1%.
1.560 1.607 v	1.583	1.564	.019	VERMICULITE (Mg,Fe) ₃ (Al,Si) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	MCL ps hex	001 mic	Green	H 1.5 G 2.67	Dec by HCl. Exfoliates when heated. El clv pos. Fe ₂ O ₃ 6.2, FeO 2.6, MgO 27.3%.
1.599 v	1.584	1.545	.039	TALC Mg ₃ Si ₄ O ₁₀ (OH,F) ₂	TCL ps hex	001 perf	White to green	H 1 G 2.7 infus	Tric analogue of monoclinic Talc. 2V 0-10°. Pearly luster, greasy feel. El clv pos.
1.564 1.598 v	1.584	1.555	.029	PHLOGOPITE (Biotite ser, Mica grp) K(Mg,Fe) ₃ Si ₃ AlO ₁₀ (F,OH) ₂	MCL ps hex, plates	001 perf	Brown	H 2-2.5 G 2.86 F diff	Insol in acids. 2V 0-30°. El pos. FeO 4.2, Fe ₂ O ₃ 3.7%.

1.576 ^ 1.595	1.584	1.577	.007	BERYL $\text{Be}_3\text{Al}_2\text{Si}_6\text{O}_{18}$	HEX pris	0001 imperf	Azure-blue, green	H 8 G 2.72 infus	Insol in acids. El neg. Pleoc variable. Data for mineral with 1-2% alkalis.
1.549 ^	1.585	1.559	.026	BEIDELLITE (Smectite grp) $\text{Na}_{0.33}(\text{Al},\text{Fe})_2(\text{Al},\text{Si})_4\text{O}_{10}$ $(\text{OH})_2 \cdot x\text{H}_2\text{O}$	MCL u mass	001 mic	Brown	H 1.5 G 2.7 infus	Biax, 2V 0-30°. El clv pos.
1.600 v	1.586	1.559	.027	GLAUCONITE, var Skolite (Mica grp) $(\text{K},\text{Na})(\text{Al},\text{Fe},\text{Mg})_2(\text{Si},\text{Al})_4\text{O}_{10}$ $(\text{OH})_2$	MCL ps hex	001 perf	Cols, blue, green	H 2 G 2.6 fus	Dec by acids. 2V small. El clv pos. Fe_2O_3 6.4, Al_2O_3 18.2, FeO 2.6%.
[]	1.586	1.560	.026	URANOSPINITE (Autunite grp) $\text{Ca}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 10\text{H}_2\text{O}$	TET	001 perf	Yellow	H 2-3 G 3.45 fus	Diss by acids. Commonly biax. El clv pos. Pleoc, O pale yellow, E cols. Fluor bright yellow in UV.
1.575 ^	1.586	1.578	.008	AUTUNITE (Autunite grp) $\text{Ca}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 10\text{H}_2\text{O}$	TET tab	001 perf	Yellow to yellow-green	H 2-2.5 G 3.2 F 3	Diss by acids. Commonly biax, 2V 0-30°. El clv pos. Fluor yellow in UV.
	1.587	1.336	.251	NITRATITE (Soda niter) NaNO_3	TRIG	10T1 perf	White	H 1.5-2 G 2.27 F 1	Sol in H_2O . Taste cooling. Deflagrates when heated.
1.574 ^ 1.607	1.587	1.555	.032	MEIONITE (Scapolite grp) $3\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot \text{CaCO}_3 -$ $3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf 110 less so	Cols to brown	H 5-6 G 2.70 F 4-5	Insol in acids. CaO 15.8, Na_2O 2.4, K_2O 2.4, SO_3 0.9, CO_2 2.9% (Me_{70}).
	1.587	1.570	.017	SHAFRANOVSKITE $(\text{Na},\text{K})_6(\text{Mn},\text{Fe})_3\text{Si}_9\text{O}_{24} \cdot 6\text{H}_2\text{O}$	TRIG	0001 perf	Dark green	H 2-3 G 2.77	Magnetic.
1.606 v	1.588	1.560	.028	COALINGITE $\text{Mg}_{10}\text{Fe}_2\text{CO}_3(\text{OH})_{24} \cdot 2\text{H}_2\text{O}$	TRIG plates	0001	Reddish- brown	G 2.33 infus	Diss by acids. Pleoc, O golden-brown, E cols. Anom biax, 2V 5-20°. El clv pos.
1.619 v	1.589	1.583	.006	KAEMMERERITE (Chlorite grp) $(\text{Mg},\text{Al},\text{Cr})_6(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	MCL ps hex	001 perf	Rose, lilac	H 2.5 G 2.72 F diff	Insol in HCl. El clv pos. FeO 2.0, Cr_2O_3 7.9%. 2V 0-8°.
1.594 L	1.590	1.543	.047	MONOHYDROCALCITE $\text{CaCO}_3 \cdot \text{H}_2\text{O}$	HEX	---	White	Infus	Diss by acids with eff.
	1.59	1.56	.03	"CONNARITE" $(\text{Mg},\text{Ni})_3\text{Si}_2\text{O}_5(\text{OH})_4$ (?)	MCL ps hex (?)	001 perf	Yellow-green	H 2.5-3 G 2.5 infus	Pleoc wk. El clv pos (interlayered vermiculite- chlorite).
	1.590	1.585	.005	IRAQUITE $(\text{K},\text{Ce},\text{La},\text{Th})(\text{Ca},\text{La},\text{Na})_2\text{Si}_8\text{O}_{20}$	TET	2 good fr uneven	Pale green- ish yellow	H 4.5 G 3.27	In part biax, 2V up to 7°. Compare Ekanite.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.588 □ 1.596	1.591	1.573	.018	METAVOLTINE $K_2Na_6Fe^{+2}_2Fe^{+3}_6(SO_4)_{12}O_{2 \cdot 18H_2O}$	HEX scales	0001 perf	Yellow-brown, greenish-brown	H 2.5 G 2.53 F 5	Dec by hot H_2O , diss by acids. El clv pos. Pleoc, O yellow, E pale yellow.
1.559 ^	1.592	---	wk to med	MERRIHUEITE (Osumilite grp) $(K,Na)_2(Fe^{+2},Mg)_5Si_{12}O_{80}$	HEX	---	---	---	Opt char unk. MgO 4.4, FeO 29.7%.
	1.592	1.582	.010	TORBERNITE (Autunite grp) $Cu(UO_2)_2(PO_4)_2 \cdot 10-12H_2O$	TET tab	001 perf 100 indist	Green	H 2-2.5 G 3.22 F 3	Diss by acids. El clv pos. Pleoc, E pale green, O dark green to blue. Not fluor in UV.
1.560 ^	1.593	1.573	.020	SAPONITE (Smectite grp) $Na_{0.33}(Mg,Fe,Fe^{+3})_3(Si,Al)_4O_{10}(OH)_2 \cdot 4H_2O$	MCL ps hex	001 perf	White, green	H 1-2 G 2.39 F 4	U biax, 2V small. Fe_2O_3 19.9, FeO 5.3, Al_2O_3 5.7%.
	1.593	1.585	.008	MACHATSCHKIITE $(Ca,Na)_6(AsO_4)(AsO_3OH)_3(PO_4,SO_4) \cdot 15H_2O$	TRIG	Conch	Cols	H 2-3 G 2.5-2.6	Diss by acids.
1.605 v	1.595	1.455	.140	BUETSCHLIITE $K_2Ca(CO_3)_2$	TRIG	---	Cols	G 2.60	Dec by acids with eff.
1.584 ^ 1.608	1.595	1.587	.008	BERYL $Cs_x[Al_{2-x}Li_x]Be_3Si_6O_{18}$	HEX pris	0001 imperf	Pink	H 8 G 2.79 infus	Insol in acids. El neg. Na_2O 1.6, Li_2O 1.7, $(Cs,Rb,K)_2O$ 1.7%.
	1.595	1.589	.006	WENKITE $Ba_4Ca_6(Si,Al)_{20}O_{39}(OH)_2(SO_4)_3 \cdot xH_2O$ (?)	HEX pris	Pris, poor	White, gray	H 6 G 3.19	In part biax, 2V 0-10°. Cancrinite grp?
	1.597	1.570	.027	ABERNATHYITE (Meta-autunite grp) $K_2(UO_2)_2(AsO_4)_2 \cdot 4H_2O$	TET tab	001 perf	Yellow	H 2-3 G 3.74	Diss by acids. El pos. Pleoc, O yellow, E pale yellow. Fluor yellow-green in UV. In part biax.
1.584 ^ 1.640	1.598	1.558	.040	BIOTITE (Mica grp) $K(Mg,Fe)_3(Al,Fe)Si_3O_{10}(OH,F)_2$	MCL ps hex, plates	001 mic	Dark brown to green	H 2-2.5 G 2.95 F 3-4	2V 0-13°. El pos. FeO 13.85, Fe_2O_3 2.1, MnO 0.4%.
	1.598	1.598	<.001	COMBEITE $Na_2Ca_2Si_3O_9$	TRIG pris	---	Cols	G 2.84	Gel with hot HCl.

1.584 ^	1.599	1.554	.045	TALC (Mg,Fe) ₃ Si ₄ O ₁₀ (OH) ₂	TCL ps hex	001 perf	Green	H 1 G 2.87 infus	Insol in acids. 2V 0-28°. El clv pos. Luster pearly. FeO 13.4, Fe ₂ O ₃ 1.7%.
1.586 ∧ 1.618	1.600	1.585	.015	GLAUCONITE (Mica grp) (K,Na)(Al,Fe,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	MCL ps hex	001 perf	Green, blue, cols	H 2 G 2.59 fus	Dec by acids. 2V 0-10°. El clv pos. Fe ₂ O ₃ 14.1, Al ₂ O ₃ 8.9, FeO 1.6%.
1.611 v	1.600	1.598	.002	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ ·6H ₂ O	TET tab	001 perf	Yellow to greenish-yellow	H 2-2.5 G 3.5 F 3	Diss by acids. El clv pos. Commonly biax, 2V small. Fluor yellow-green in UV.
	1.601	1.480	.121	GRIMSELITE K ₂ Na(UO ₂)(CO ₃) ₃ ·H ₂ O	HEX prism	Conch	Yellow	H 2-2.5 G 3.30	Sol in H ₂ O. Pleoc, 0 yellow, E nearly cols.
	1.601	1.591	.010	DAVISONITE Ca ₃ Al(PO ₄) ₂ (OH) ₃ ·H ₂ O	HEX fib, botryoidal crusts	0001 perf	Cols, white	H 4.5 G 2.85	Diss by acids. El clv pos.
1.598 └	1.602	1.594	.008	TAKOVITE Ni ₆ Al ₂ (OH) ₁₆ (CO ₃ ,OH)·4H ₂ O	TRIG	---	Yellow-green to blue-green	G 2.80 infus	Diss by acids. Compare Hydrotalcite.
1.583 ∧ 1.634	1.603	1.555	.048	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₁₂	MCL and TCL ps hex	001 mic	Brown	H 3.5 G 2.80 F 3.5	Dec by acids. Pleoc, E pale yellow, 0 deep olive-brown. El pos. FeO 21.6, MnO 1.4, Fe ₂ O ₃ 6.4%.
1.622 v	1.603	1.598	.005	CARBONATE-HYDROXYLAPATITE (Apatite grp) Ca ₅ (PO ₄ ,CO ₃) ₃ (OH,F)	HEX u mass	---	Cols, gray, brown	H 5 G 2.9 F diff	Diss by HCl with slight eff. CO ₂ 3.4, SO ₃ 0.4, H ₂ O 3.9%.
	1.605	1.450	.155	EITELITE Na ₂ Mg(CO ₃) ₂	TRIG	0001	Cols	---	Dec by HCl with eff.
1.595 ^	1.605	1.453	.152	BUETSCHLIITE K ₂ Ca(CO ₃) ₂	TRIG	---	Cols	G 2.60	Dec by HCl with eff.
	1.605	1.574	.031	HEINRICHITE (Autunite grp) Ba(UO ₂) ₂ (AsO ₄) ₂ ·10-12H ₂ O	TET	001 perf 100 dist	Yellow to green	H 2.5	Diss by acids. Pleoc, 0 pale yellow, E cols. In part biax, 2V 0-20°. Fluor green to greenish-yellow in UV.
	1.606	1.608	.002	SOGDIANITE (Osumilite grp) (K,Na) ₂ Li ₂ (Li,Fe ⁺³ ,Al) ₂ ZrSi ₁₂ O ₃₀	HEX	---	Violet	H 7 G 2.90	ZrO ₂ 9.8, TiO ₂ 2.9, FeO 1.2, Fe ₂ O ₃ 4.6, Na ₂ O 2.8, K ₂ O 4.8, Li ₂ O 3.7%. Opt sign pos by indices.
1.588 ∧ 1.634	1.606	1.575	.031	COALINGITE Mg ₁₀ Fe ₂ CO ₃ (OH) ₂₄ ·2H ₂ O	TRIG platy	0001	Reddish-brown	G 2.32 infus	Diss by acids. El pos. Pleoc, 0 golden-brown, E cols. Anom biax, 2V 5-20°.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
$\sqrt{1.630}$	1.606	1.602	.004	EUDIALYTE $\text{Na}_4(\text{Ca,Ce})_2(\text{Fe}^{+2}, \text{Mn}^{+2})\text{ZrSi}_8\text{O}_{22}(\text{OH}, \text{Cl})_2$	TRIG	0001 dist 1020 poor	Yellow, pink, brown	H 5-5.5 G 2.85 F 3	Gel with acids. Pleoc wk in pink and yellow. Anom biax. Also uniax pos.
$\sqrt[4]{1.583}$ $\sqrt[4]{1.648}$	1.607	1.560	.047	VERMICULITE $(\text{Mg}, \text{Fe}, \text{Al})_3(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	MCL ps hex	001 perf	Green, brown	H 1.5 G 2.77 F 5	Dec by HCl. El clv pos. Exfoliates when heated. Fe_2O_3 4.5%.
$\sqrt[4]{1.587}$	1.607	1.571	.036	MEIONITE (Scapolite grp) $3\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot \text{CaCO}_3 - 3\text{NaAlSi}_3\text{O}_8 \cdot \text{NaCl}$	TET	100 perf 110 less so	Cols to brown	H 5-6 G 2.7 F 4-5	Insol in acids (Me_{84}).
[]	1.607	1.602	.005	JAMBORITE $(\text{Ni}^{+2}, \text{Ni}^{+3}, \text{Co}, \text{Fe})(\text{OH})_2(\text{OH}, \text{S}, \text{H}_2\text{O}) (?)$	HEX fib, platy	---	Green	G 2.67 infus	Slowly diss by cold HCl.
$\sqrt[4]{1.623}$	1.607	1.604	.003	WHITLOCKITE, carbonatian, var Martinite $\text{Ca}_9(\text{Mg}, \text{Fe})\text{H}(\text{PO}_4, \text{CO}_3)_7$	TRIG u mass	---	Cols	H 5 G 3.12 infus	Diss by acids.
$\sqrt[4]{1.595}$	1.608	1.599	.009	BERYL $\text{Cs}_x[\text{Al}_{2-x}(\text{Li}, \text{Fe})_x](\text{Be}_{3-x}\text{Fe}_x)\text{Si}_6\text{O}_{18}$	HEX pris	0001 imperf	Blue-green	H 8 G 2.92 infus	Insol in acids. El neg. FeO 2.2, Fe_2O_3 2.0. Total alkalies 8.2%.
	1.608	1.606	.002	SOGDIANITE (Osumilite grp) $(\text{K}, \text{Na})_2\text{Li}_2(\text{Li}, \text{Fe}^{+3}, \text{Al})_2\text{ZrSi}_{12}\text{O}_{30}$	HEX	0001 perf	Violet	H 7 G 2.90	---
	1.609	1.596	.013	HILAIRITE $\text{Na}_2\text{ZrSi}_3\text{O}_9 \cdot 3\text{H}_2\text{O}$	TRIG	Conch	Pale brown	H > 4 G 2.72	Slightly attacked by HCl at room temp.
$\sqrt[4]{1.573}$	1.609	1.604	.005	EKANITE $(\text{Th}, \text{U})(\text{Ca}, \text{Fe}, \text{Pb})_2\text{Si}_8\text{O}_{20}$	TET	---	Green, brown, cols	G 3.3	---
	1.610	1.605	.005	ZIRSINALITE $\text{Na}_6(\text{Ca}, \text{Mn}, \text{Fe})\text{ZrSi}_6\text{O}_{18}$	TRIG	Uneven to conch	Cols	H 5 G 2.90 F 3	Dec by warm HCl.
	1.610	1.607	.003	SUGILITE (Osumilite grp) $(\text{K}, \text{Na})(\text{Na}, \text{Fe})_2(\text{Li}, \text{Fe})\text{Si}_{12}\text{O}_{30}$	HEX	0001 poor	Light brownish-yellow	H 6-6.5 G 2.74	---

	1.611	1.592	.019	MELIPHANITE (Ca,Na) ₂ Be(Si,Al) ₂ (O,F) ₇	TET pyram	001 perf	Yellow	H 5-6 G 3.01 fus	Insol in acids. Anom biax. El clv pos.
1.600	1.611	1.594	.017	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ ·6H ₂ O	TET tab	001 perf	Yellow to greenish	H 2-2.5 G 3.5 F 3	Diss by acids. Fluor yellow-green in UV. Com- monly biax neg, 2V small.
	1.612	1.585	.027	SODIUM URANOSPINITE (Meta-autunite grp) (Na ₂ ,Ca)(UO ₂) ₂ (AsO ₄) ₂ ·5H ₂ O	TET tab, fib	001 perf 010, 100 dist	Yellow, yellow- green	H 2.5 G 3.85	Diss by acids. Fluor yellow-green in UV. Anom biax. Pleoc, E cols, 0 yellowish.
1.610 □ 1.616	1.613	1.585	.028	ZEUNERITE (Autunite grp) Cu(UO ₂) ₂ (AsO ₄) ₂ ·10-16H ₂ O	TET tab	001 perf 010 dist	Green	H 2-2.5 G 3.4 fus	Diss by acids. El clv pos. Does not fluor in UV.
	1.614	1.595	.019	SEMENOVITE (Ca,Ce,La,Na) ₁₀₋₁₂ (Fe,Mn) (Si,Be) ₂₀ (O,OH,F) ₄₈	ORTH ps tet	Uneven	Cols	H 3.5-4 G 3.14	Tw 110.
1.609 □ 1.618	1.614	1.608	.006	FLUOCERITE (Ce,La)F ₃	HEX	0001 dist	Wax-yellow to brown	H 4-5 G 5.8-6.0 infus	Insol in HCl or HNO ₃ , diss by H ₂ SO ₄ . Anom biax.
1.598 ◇ 1.640	1.617	1.575	.042	BIOTITE, var Manganophyllite (Mica grp) K(Mg,Fe,Mn) ₃ (Al,Fe)Si ₃ O ₁₀ (OH,F) ₂	MCL ps hex	001 perf	Reddish- brown	H 2.5-3 G 2.98 F 3-4	2V 0-30°. El clv pos. Pleoc, 0 dark brown, E brown. Fe ₂ O ₃ 5.8, FeO 1.2, MnO 6.2%.
	1.617	1.595	.022	KARNASURTITE (La,Ce,Th)(Ti,Nb)(Al,Fe) (Si,P) ₂ O ₇ (OH) ₄ ·3H ₂ O (?)	HEX (?)	One good	Yellow	H 2 G 2.92	Anom biax.
1.632 √	1.618	1.552	.066	CHALCOPHYLLITE Cu ₁₈ Al ₂ (AsO ₄) ₃ (SO ₄) ₃ (OH) ₂₇ · 33H ₂ O	TRIG tab	0001 perf	Deep emerald- green	H 2 G 2.69 F 2-2.5	Diss by acids or NH ₄ OH. El clv pos. Pleoc, 0 blue- green, E nearly cols. Indices increase with loss of H ₂ O.
1.600 ◇ 1.634	1.618	1.597	.021	GLAUCONITE (Mica grp) (K,Na)(Al,Fe,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	MCL ps hex	001 perf	Green	H 2 G 2.7 fus	Dec by acids. El clv pos. Pleoc, E yellow, 0 bluish- green. Fe ₂ O ₃ 18.8, Al ₂ O ₃ 8.5, FeO 4.0, MgO 3.6%.
1.632 √	1.618	1.606	.012	BURBANKITE (Na,Ca) ₃ (Sr,Ba,Ce) ₃ (CO ₃) ₅	HEX u mass	Pris, 0001	Yellow, pink	H 4.5-5 G 3.39-3.60 infus	Diss by acids with eff.
1.589 ^	1.619	1.605	.014	CLINOCHLORE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	MCL ps hex	001 perf	Olive- to dark-green	H 2 G 2.8 F 5	Dec by acids. El clv pos. Pleoc, E pale green, 0 green. 2V 0-5°.
1.611 □ 1.622	1.619	1.611	.008	CYMRITE BaAl ₂ Si ₂ (O,OH) ₈ ·H ₂ O	MCL ps hex	001 perf pris imperf	Cols	H 2-3 G 3.45	Slightly sol in acids. El clv pos.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ε							
1.636 ∇ 1.603 ∧ 1.640 1.607 ∧ 1.630 1.624 □ 1.630 1.653 ∇	1.619	1.618	.001	GILLESPIITE BaFeSi ₄ O ₁₀	TET platy	001 mic	Rose-red	H 3 G 3.33 F easy	Dec by HCl. El pos. Pleoc str, O pale pink, E deep rose-red.
	1.621	1.609	.012	KHANNESHITE (Na,Ca) ₃ (Ba,Sr,Ce,Ca) ₃ (CO ₃) ₅	HEX	Prism indist	Pale yellow	H soft G 3.8-3.9	Diss by HCl.
	1.622	1.606	.016	BAZZITE Be ₃ (Sc,Al,Fe) ₂ Si ₆ O ₁₈	HEX barrel-shaped	---	Bright blue	H 7 G 2.78 infus	Insol in acids. Pleoc, E azure-blue, O pale yellow to cols.
	1.622	1.614	.008	CARBONATE-FLUORAPATITE (Apatite grp) Ca ₅ (PO ₄ ,CO ₃) ₃ (F,OH)	HEX pris, fib	0001 perf	Gray to greenish	H 5 G 3.05 F 3	Diss by acids. El clv pos. Basal section may show biax segments.
	1.622	1.615	.007	HUNTITE CaMg ₃ (CO ₃) ₄	TRIG u mass	---	White	H 2-2.5 G 2.70 infus	Diss by HCl with eff.
	1.623	1.620	.003	WHITLOCKITE ("Merrillite") (Ca,Na) ₉ (Mg,Fe)H(PO ₄) ₇	TRIG	---	Cols	H 5 G 3.1 infus	Diss by acids. Meteorite mineral.
	1.623	1.619	.004	OLGITE Na(Sr,Ba)PO ₄	HEX	---	Bright blue, greenish-blue	H 4.5 G 3.94	---
	1.624	1.591	.033	TISINALITE (Lovozerite grp) Na ₃ H ₃ (Mn,Ca,Fe)TiSi ₆ (O,OH) ₁₈ ·2H ₂ O	TRIG	Uneven to conch	Yellow to orange	H 5 G 2.67	---
	1.627	1.582	.045	TROEGERITE (Autunite grp) (UO ₂) ₃ (AsO ₄) ₂ ·12H ₂ O (?)	TET tab 001	001 perf 100 good	Lemon-yellow	H 2-3 G 3.3 F 2.5	Diss by acids. Fluor Fluor yellow in UV.
	1.629	---	very low	KAMAISHILITE Ca ₂ Al ₂ SiO ₆ (OH) ₂	TET	---	Cols	---	Opt sign unk.
1.629	1.624	.005	MELILITE (Melilite grp) (Ca,Na) ₂ (Mg,Al)(Si,Al) ₂ O ₇	TET	001 dist	Cols	H 5 G 2.98 F 3	Gel with acids. MgO 8.2, FeO 2.1, Fe ₂ O ₃ 1.2, Al ₂ O ₃ 6.9, Na ₂ O 3.2%.	

	1.63	1.59	.04	HYDROHONESSITE $\text{Ni}_6\text{Fe}^{+3}_2(\text{SO}_4)(\text{OH})_{16} \cdot 7\text{H}_2\text{O}$	HEX	---	Bright yellow	---	Diss by acids.
1.606 ^ 1.652	1.630	1.620	.010	EUODIALYTE $\text{Na}_4(\text{Ca}, \text{Ce})_2(\text{Fe}^{+2}, \text{Mn})\text{ZrSi}_8\text{O}_{22}(\text{OH}, \text{Cl})_2$	TRIG	0001 dist 1010 poor	Yellow, brown, pink	H 5-5.5 G 3.03 F 3	Gel with acids. Pleoc wk in pink and yellow. Anom biax.
1.623 ^	1.630	1.627	.003	WHITLOCKITE $\text{Ca}_9(\text{Mg}, \text{Fe})\text{H}(\text{PO}_4)_7$	TRIG u mass	---	Cols	H 5 G 3.12 infus	Diss by acids.
1.625 ^	1.631	1.628	.003	METATORBERNITE (Meta-autunite grp) $\text{Cu}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 4-8\text{H}_2\text{O}$	TET tab	001 perf 100 indist	Green	H 2.5 G 3.7-3.8 F 3	Diss by acids. Pleoc, 0 dark green, E pale green to blue. Does not fluor in UV.
1.618 ^	1.632	1.575	.057	CHALCOPHYLLITE $\text{Cu}_{18}\text{Al}_2(\text{AsO}_4)_3(\text{SO}_4)_3(\text{OH})_{27} \cdot 33\text{H}_2\text{O}$	TRIG tab	0001 perf	Deep emerald-green	H 2 G 2.69 F 2-2.5	Diss by acids or NH_4OH . El clv pos. Pleoc, 0 blue-green, E nearly cols.
1.650 v	1.632	1.602	.030	BEMENTITE $\text{Mn}_8\text{Si}_6\text{O}_{15}(\text{OH})_{10}$	MCL fib	001 perf	Brown	H 4-6 G 2.7-3.1 F 3	Pleoc wk, 0 yellow, E nearly cols.
1.618 ^	1.632	1.620	.012	BURBANKITE $(\text{Na}, \text{Ca})_3(\text{Sr}, \text{Ba}, \text{Ce})_3(\text{CO}_3)_5$	HEX u mass	Pris, 0001	Yellow, pink	H 4.5-5 G 3.39-3.60 infus	Diss by acids with eff.
	1.633	1.590	.043	CUPRORIVAITE $\text{CaCuSi}_4\text{O}_{10}$	TET	001 perf	Blue	H 5 G 3.08	Insol in acids. El clv pos. Pleoc, 0 blue, E pale rose.
1.622 ^ 1.638	1.633	1.630	.003	FLUORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3\text{F}$	HEX pris	0001 imperf	Cols, green, brown	H 5 G 3.2 F 5	Diss by acids. El neg. Data for synth compd.
1.603 ^ 1.661	1.634	1.575	.059	STILPNOMELANE $\text{K}(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3}, \text{Al})_{10}\text{Si}_{12}\text{O}_{30}(\text{O}, \text{OH})_2$	MCL and TCL ps hex	001 mic	Brown	H 3.5 G 2.84 F 3.5	Dec by acids. Pleoc, 0 dark brown, E pale brown. El clv pos. FeO 22.7, Fe_2O_3 12.2%.
1.606 ^	1.634	1.590	.044	COALINGITE $\text{Mg}_{10}\text{Fe}_2\text{CO}_3(\text{OH})_{24} \cdot 2\text{H}_2\text{O}$	TRIG platy	0001	Reddish-brown	G 2.32 infus	Diss by acids. Pleoc, 0 deep golden brown, E cols. Anom biax.
1.618 ^ 1.643(?)	1.634	1.610	.024	GLAUCONITE (Mica grp) $(\text{K}, \text{Na})(\text{Fe}^{+3}, \text{Al}, \text{Mg}, \text{Fe}^{+2})_2(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2$	MCL ps hex	001 perf	Green	H 2 G 2.74 fus	Dec by acids. El clv pos. Pleoc, E yellow-green, 0 blue-green. Fe_2O_3 24.5, Al_2O_3 7.3, FeO 2.9%. 2V 0-15°.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
∇ 1.638	1.634	1.612	.022	DRAVITE (Tourmaline grp) $\text{NaMg}_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	$11\bar{2}0, 10\bar{1}1$ very poor	Brown	H 7 G 3.03 infus	Insol in acids. El neg. Pleoc, abs 0 > E. MgO 8.4, FeO 0.2, Na_2O 1.9%.
∇ 1.637	1.635	1.618	.017	ELBAITE (Tourmaline grp) $\text{Na}(\text{Al}, \text{Li})_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	$11\bar{2}0, 10\bar{1}1$ very poor	Red	H 7 G 3.02 infus	Insol in acids. El neg. Pleoc, abs 0 > E. Fe_2O_3 0.3, Li_2O 1.9, CaO 0.4, Na_2O 2.4%.
\square	1.635	---	wk	EYLETTERSITE (Crandallite grp) $(\text{Tn}, \text{Pb}, \text{Ba})\text{Al}_3(\text{PO}_4)_2(\text{OH})_6$ (?)	TRIG	---	White to cream	G 3.38-3.44	Fluor wk greenish-yellow in short-wave UV.
1.622 \wedge	1.636	1.618	.018	BAZZITE $\text{Be}_3(\text{Sc}, \text{Fe})_2\text{Si}_6\text{O}_{18}$	HEX barrel-shaped	---	Blue	H 7 G 2.78 infus	Insol in acids. Pleoc, E azure blue, 0 pale yellow to cols.
	1.637	1.615	.022	MITSCHERLICHITE $\text{K}_2\text{CuCl}_4 \cdot 2\text{H}_2\text{O}$	TET tw on 001	---	Green to blue	H 2.5 G 2.40 fus	Sol in H_2O . Pleoc, E sky-blue; 0 grass-green. Anom blax. Data for synth compd.
∇ 1.649	1.637	1.615	.022	CHAMOSITE (Chlorite grp) $(\text{Mg}, \text{Fe}^{+2}, \text{Fe}^{+3})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{O}, \text{OH})_8$	MCL ps hex	001 mic	Green	H 2.5 G 3.0 F diff	Dec by acids. Pleoc, E olive-green, 0 brownish-yellow. FeO 22.8, Fe_2O_3 9.1, Al_2O_3 22.1%.
\square	1.637	1.620	.017	METANOVAEKITE (Meta-autunite grp) $\text{Mg}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	TET tab	001 perf 100 dist	Yellow	H 2-2.5 G 3.66	Diss by acids. Fluor yellow in UV. Commonly biax, 2V 0-20°.
1.635 \diamond 1.646	1.637	1.621	.016	LIDDICOATITE (Tourmaline grp) $\text{Ca}(\text{Li}, \text{Al})_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{O}, \text{OH}, \text{F})_4$	TRIG pris	$11\bar{2}0, 10\bar{1}1$ very poor	Brown, pink, green	H 7.5 G 3.02 infus	Insol in acids. Pleoc, 0 dark brown, E light brown. El neg. CaO 4.2, Na_2O 0.9, Li_2O 2.5, F 1.7%.
1.634 \diamond 1.653	1.638	1.619	.019	UVITE (Tourmaline grp) $(\text{Ca}, \text{Na})(\text{Mg}, \text{Fe})_3\text{Al}_5\text{Mg}(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH}, \text{F})_4$	TRIG pris	$11\bar{2}0, 10\bar{1}1$ very poor	Brown	H 7 G 3.08 F 5	Insol in acids. El neg. Pleoc, abs 0 > E. MgO 15.4, FeO 0.5, CaO 5.5, Na_2O 0.2%.
1.633 \diamond 1.644	1.638	1.634	.004	FLUORAPATITE (Apatite grp) $(\text{Ca}, \text{Sr})_5(\text{PO}_4)_3\text{F}$	HEX pris	0001 imperf	Cols	H 5 G 3.32 F 5	Diss by acids. El neg. SrO 11.6%.

1.617 ^ 1.652	1.640	1.590	.050	SIDEROPHYLLITE (Biotite ser, Mica grp) $\text{KFe}_2\text{Al}(\text{Al}_2\text{Si}_2)_0_{10}(\text{F},\text{OH})_2$	MCL ps hex	001 mic	Black, dark brown	H 3 G 3.12 F 4	El clv pos. Pleoc, X brown, Y and Z dark brown. FeO 30.2, MnO 1.0%.
1.622 ^ 1.651	1.640	1.633	.007	CARBONATE-FLUORAPATITE ("Dehrnite") (Apatite grp) $(\text{Ca},\text{Na},\text{K})_5(\text{PO}_4)_3(\text{F},\text{OH})$	HEX pris, fib	0001 good	Gray to greenish	H 5 G 3.05 F 3	Diss by acids. El clv pos, el crystal neg.
	1.641	1.609	.032	METAHEINRICHITE (Meta-autunite grp) $\text{Ba}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	TET	001 perf 100 dist	Yellow to green	H 2.5 G 4.04	Diss by acids. Pleoc, 0 pale yellow, E cols. Fluor yellow-green in UV.
	1.642	1.608	.034	METAKAHLERITE (Meta-autunite grp) $\text{Fe}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	TET	001 perf 100 good	Sulfur-yellow	---	Diss by acids. Does not fluor in UV. Anom biax.
1.634(?) ^ 1.663	1.643	1.612	.031	CELADONITE (Mica grp) $(\text{K},\text{Na})(\text{Fe}^{+3},\text{Al},\text{Fe}^{+2},\text{Mg})_2(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2$	MCL ps hex	001 perf	Bright green	H 2 G 2.8 F 3	Dec by HCl. Pleoc, E yellow-green, 0 green. Fe_2O_3 24.3, Al_2O_3 1.0, FeO 9.8, MgO 3.7%.
	1.643	1.623	.020	ZUSSMANITE $\text{K}(\text{Fe},\text{Mg},\text{Mn})_{13}(\text{Al},\text{Si})_{18}\text{O}_{42}(\text{OH})_{14}$	TRIG tab	0001 perf	Pale green	G 3.15	El clv pos. Pleoc wk, 0 pale green, E cols.
	1.644	1.542	.102	EWALDITE $\text{Ba}(\text{Ca},\text{Y},\text{Na},\text{K})(\text{CO}_3)_2$	HEX	---	Blue-green, brick red	H 3 G 3.42	---
	1.644	1.617	.027	METAKIRCHHEIMERITE (Meta-autinite grp) $\text{Co}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	ORTH ps tet (?) plates	001 perf	Rose	H 2-2.5 G > 3.33	Diss by acids. El clv pos. Not fluor in UV.
1.638 ^ 1.651	1.644	1.638	.006	FLUORAPATITE (Apatite grp) $(\text{Ca},\text{Mn})_5(\text{PO}_4)_3\text{F}$	HEX pris	0001 imperf	Blue	H 5 G 3.26 F 5	Diss in acids. El neg. Fluor orange in UV. MnO 4.7%.
1.637 ^	1.646	1.625	.021	ELBAITE (Tourmaline grp) $\text{Na}(\text{Al},\text{Li},\text{Fe})_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	11 $\bar{2}$ 0, 10 $\bar{1}$ 1 very poor	Blue-green	H 7 G 3.09 infus	Insol in acids. El neg. Na_2O 2.5, Li_2O 1.2, FeO 4.9, MnO 0.7%.
	1.646	1.635	.011	ZAPATALITE $\text{Cu}_3\text{Al}_4(\text{PO}_4)_3(\text{OH})_9 \cdot 4\text{H}_2\text{O}$	TET	001 good	Pale blue	H 1.5 G 3.02	Pleoc, wk in green, abs E > 0. Diss by acids.
	1.646	1.639	.007	KELLYITE (Kaolinite-Serpentine grp) $(\text{Mn},\text{Al},\text{Mg})_3(\text{Si},\text{Al})_2\text{O}_5(\text{OH})_4$	TRIG and HEX granular	0001 perf	Yellow	G 3.07	Anom biax. Pleoc wk, E cols, 0 pale yellow.
	1.647	1.637	.010	AMINOFFITE $\text{Ca}_2(\text{Be},\text{Al})(\text{Si},\text{Al})_2\text{O}_7(\text{OH}) \cdot \text{H}_2\text{O}$	TET pyram	001 poor fr conch	Cols	H 5.5 G 2.94	Anom biax, 2V 0-15°.
1.607 ^	1.648	1.622	.026	VERMICULITE $(\text{Mg},\text{Ni},\text{Fe})_3(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	MCL ps hex	001 perf	Brownish-black	H 1.5 G 2.8 F 5	Dec by HCl. El clv pos. Exfoliates when heated. Fe_2O_3 19.2, FeO 5.0, MgO 13.9, NiO 8.6%.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.643 □ 1.651	1.648	1.624	.024	METAZEUNERITE (Meta-autunite grp) $\text{Cu}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	TET tab	001 perf 010 dist	Green	H 2.5 G 3.67	Diss by acids. Does not fluor in UV.
	1.648	1.625	.023	KAZAKOVITE $\text{Na}_6\text{H}_2\text{TiSi}_6\text{O}_{18}$	TRIG	Uneven to subconch	Pale yellow	H 4 G 2.84 F 1.5	Tw poly.
1.637 ∧ 1.685	1.649	1.643	.006	CHAMOSITE (Chlorite grp) $(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH}, \text{O})_8$	MCL ps hex	001 perf	Dark green	H 2 G 3.20	Dec by hot HCl. El clv pos. Pleoc, 0 olive-green, E pale yellow.
	1.649	1.641	.008	JEREMEJEVITE $\text{Al}_6\text{B}_5\text{O}_{15}(\text{OH})_3$	HEX pris	Conch	Cols to yellow-brown	H 6.5 G 3.28 infus	After ignition, diss by concd H_2SO_4 or KOH. El neg. Pleoc, 0 yellowish, E light blue. Anom biax.
104 1.724	1.65	1.60	.05	PYROCHROITE $\text{Mn}(\text{OH})_2$	TRIG	0001 perf	Cols to brown	H 2-2.5 G 3.1-3.2 infus	Diss by acids. Rapidly darkens in air. El clv pos.
1.632 ^	1.650	1.624	.026	BEMENTITE $\text{Mn}_8\text{Si}_6\text{O}_{15}(\text{OH})_{10}$	MCL fib	001 perf	Brown	H 4-6 G 2.7-3.1 F 3	Dec by acids. Pleoc wk, 0 yellow, E nearly cols.
1.644 ∧ 1.660	1.651	1.637	.014	STRONTIUM-APATITE (Apatite grp) $(\text{Sr}, \text{Ca}, \text{Ba})_5(\text{PO}_4)_3(\text{F}, \text{OH})$	HEX	---	Cols to green	H 5 G 3.84	Diss by acids. El neg. SrO 4.6, CaO 10.8, BaO 2.7%.
1.640 ^	1.651	1.644	.007	HYDROXYLAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3(\text{OH}, \text{F})$	HEX	---	Cols	H 5 G 3.21 infus	Diss by acids. El neg. F 0.16, H_2O 1.86%.
1.640 ∧ 1.672	1.652	1.595	.057	BIOTITE (Mica grp) $\text{K}(\text{Fe}, \text{Mg})_3(\text{Al}, \text{Fe})\text{Si}_3\text{O}_{10}(\text{OH}, \text{F})_2$	MCL ps hex	001 perf	Dark brown	H 2.5-3 G 3.05 F 4	El clv pos. Pleoc, 0 dark brown, E brown. FeO 19.9, MnO 0.4, Fe_2O_3 3.2%.
1.630 ^	1.652	1.640	.012	EUDIALYTE $\text{Na}_4(\text{Ca}, \text{Ce})_2(\text{Fe}^{+2}, \text{Mn})\text{ZrSi}_8\text{O}_{22}(\text{OH}, \text{Cl})_2$	TRIG	0001 dist 1010 poor	Pink	H 5-5.5 G 2.87 F 3	Gel with acids. Pleoc wk in pink. MnO 10.8, Fe_2O_3 1.9%.
1.647 □ 1.655	1.652	1.649	.003	"WILKEITE" (Apatite grp) $\text{Ca}_5(\text{SiO}_4, \text{PO}_4, \text{SO}_4)_3(\text{O}, \text{OH}, \text{F})$	HEX u mass	0001 imperf	Yellow, reddish	H 5 G 3.1 F diff	Dec by acids. P_2O_5 15.9, SO_3 13.5, SiO_2 12.0%. = Silician Sulfatian Apatite or phosphatian Fluorellestadite.

1.638 ∧ 1.660	1.653	1.637	.016	DRAVITE (Tourmaline grp) (Na,Ca)(Mg,Fe) ₃ Al ₆ (BO ₃) ₃ Si ₆ O ₁₈ (OH) ₄	TRIG pris	11 $\bar{2}$ 0, 10 $\bar{1}$ very poor	Dark brown	H 7 G 3.25 F 5	Insol in acids. El neg. Pleoc, 0 deep olive-green, E pale brown, abs 0 > E. MgO 9.3, FeO 3.6, Fe ₂ O ₃ 2.2, CaO 1.6, Na ₂ O 1.7%.
	1.653	1.642	.011	CHANTALITE CaAl ₂ SiO ₄ (OH) ₄	TET	---	Cols, white	G 2.8-2.9	---
1.629 ∧ 1.672	1.653	1.652	.001	MELILITE (Melilite grp) (Ca,Na) ₂ (Mg,Al)(SiAl)O ₇	TET	001 dist	Cols	H 5 G 3.0 F 4	Gel with acids. Data for synth Ge 50, Ak 50.
∨ 1.681	1.654	1.629	.029	FRIEDELITE (Mn,Fe) ₈ Si ₆ O ₁₅ (OH,Cl) ₁₀	MCL ps trig tab 0001	0001 perf 10 $\bar{1}$ 0 imperf	Pink, brown	H 4.5 G 3.17 fus	Dec by acids. Fus to black glass. Anom biax, 2V small. El pos.
┌ 1.660	1.655	1.645	.010	KONINCKITE FePO ₄ ·3H ₂ O	TET fib	---	Yellow-green	H 3 G 2.4 F 2.5-3	Diss by concd acids.
1.654 └ 1.703	1.655	1.650	.005	ELLESTADITE (Apatite grp) Ca ₅ (SiO ₄ ,PO ₄ ,SO ₄) ₃ (OH,Cl,F)	HEX	0001 imperf	Cols, pale rose, purple	H 5 G 3.1-3.2 F diff	Dec by acids. El neg. SiO ₂ 17.3, SO ₃ 26.7, P ₂ O ₅ 3.1%.
∨ 1.672	1.658	1.486	.172	CALCITE (Calcite grp) CaCO ₃	TRIG	10 $\bar{1}$ 1 perf	Cols, pink, brown	H 3 G 2.71 infus	Diss by acids with eff. Data for pure CaCO ₃ .
∨ 1.666	1.658	1.653	.005	CHLORAPATITE (Apatite grp) Ca ₅ (PO ₄) ₃ (Cl,F,OH)	MCL ps hex	001 imperf	Greenish- yellow	H 5 G 3.18 F 5	Diss by acids. El neg. Cl 3.7, F 1.15%.
	1.66	1.57	.09	MCKELVEYITE Ba ₃ Na(Ca,U)Y(CO ₃) ₆ ·3H ₂ O	TCL ps trig tab	---	Dark green	G 3.5-3.6	Diss by acids. Pleoc in green, abs 0 > E.
1.651 ^	1.660	1.640	.020	BELOVITE (Apatite grp) (Sr,Ce,Na,Ca) ₅ (PO ₄) ₃ (OH)	HEX pris	Pris, pinacoidal, imperf	Honey- yellow	H 5 G 4.19 F diff	Diss by acids. El neg.
	1.66	---	wk	FERMORITE (Apatite grp) (Ca,Sr) ₅ (PO ₄ ,AsO ₄) ₃ (OH,F)	HEX pyram	---	Pink to white	H 5 G 3.52 F diff	Diss by acids.
1.653 ∧ 1.661	1.660	1.639	.021	UVITE (Tourmaline grp) Ca(Mg,Fe) ₃ Al ₅ Mg(BO ₃) ₃ Si ₆ O ₁₈ (OH,F) ₄	TRIG pris	11 $\bar{2}$ 0, 10 $\bar{1}$ very poor	Dark brown	H 7 G 3.16 F 5	Insol in acids. El neg. Pleoc, abs 0 > E. MgO 11.0, FeO 5.8, Fe ₂ O ₃ 2.7, CaO 7.5%.
1.634 ∧ 1.685	1.661	1.584	.077	STILPNOMELANE K(Fe ⁺³ ,Fe ⁺² ,Mg) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₁₂	MCL and TCL ps hex	001 perf	Brown	H 3.5 G 2.80 F 3.5	Dec by acids. El clv pos. Pleoc, 0 deep brown, E golden-yellow. FeO 8.9, MnO 2.9, Fe ₂ O ₃ 19.4%.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.660 ^ 1.675	1.661	1.632	.029	SCHORL (Tourmaline grp) $\text{NaFe}_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	11 $\bar{2}$ 0, 10 $\bar{1}$ very poor	Black	H 7 G 3.17 F diff	Insol in acids. El neg. Pleoc, abs 0 > E. FeO 8.7, MgO 5.6, Fe ₂ O ₃ 0.8, Na ₂ O 2.3%.
1.667 v	1.661	1.646	.015	PENNANTITE (Chlorite grp) $\text{Mn}_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	MCL ps nex, plates	001 perf	Orange- brown	H 2.5 G 3.06	Pleoc in orange. El clv pos. MnO 38.9, Al ₂ O ₃ 18.6, Fe ₂ O ₃ 4.4%.
1.643 ^	1.663	1.644	.019	CELADONITE (Mica grp) $(\text{K}, \text{Na})(\text{Al}, \text{Fe}, \text{Mg})_2(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2$	MCL ps hex	001 perf	Bright green	H 2 G 2.8 F 3	Dec by HCl. El clv pos. Pleoc, E yellow-green, 0 emerald-green. Fe ₂ O ₃ 26.9, Al ₂ O ₃ 2.1, FeO 9.3%.
1.669 v	1.664	1.634	.030	PYROSMALITE $(\text{Fe}, \text{Mn})_8\text{Si}_6\text{O}_{15}(\text{OH}, \text{Cl})_{10}$	TRIG	0001 perf	Brown	H 4.5-5 G 3.11 fus	Dec by HCl. El clv pos. Pleoc, 0 light brown, E dark brown.
	1.664	1.646	.018	GONYERITE (Chlorite grp) $(\text{Mn}, \text{Mg})_5\text{Fe}(\text{Si}_3\text{Fe})\text{O}_{10}(\text{OH})_8$	ORTH (?) ps hex	001 perf	Deep brown	H 2.5 G 3.01	Dec by HCl. El clv pos. Pleoc, 0 light brown, E dark brown.
	1.666	1.653	.013	TIENSHANITE $\text{BaNa}_2\text{MnTiB}_2\text{Si}_6\text{O}_{20}$	HEX	0001 dist	Pistachio- green	H 6-6.5 G 3.29 F 2	---
1.658 ^	1.666	1.660	.006	CHLORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3\text{Cl}$	MCL ps hex	001 imperf	Cols	H 4.5 G 3.2 F 5	Diss by acids. El neg. Data for synth compd.
1.661 ^ 1.673	1.667	1.658	.009	PENNANTITE ("Grovesite") (Chlorite grp) $(\text{Mn}, \text{Fe})\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	MCL ps hex	001	Blackish- brown	G 3.15	Pleoc, 0 very dark brown, E reddish-brown.
1.664 □ 1.671	1.667	1.662	.005	STEENSTRUPINE $(\text{Ce}, \text{La}, \text{Na}, \text{Mn})_6(\text{Si}, \text{P})_6\text{O}_{18}(\text{OH})$	HEX	---	Brown, red, yellow	H 5 G 3.5 infus	Pleoc wk in brown and yellow.
	1.668	1.641	.027	MCGILLITE $(\text{Mn}, \text{Fe}^{+2}, \text{Mg})_8\text{Si}_6\text{O}_{15}(\text{OH})_8\text{Cl}_2$	MCL ps trig	0001 good	Pink	G 2.98	Dec by HCl.
1.664 ^ 1.682	1.669	1.631	.038	MANGANPYROSMALITE $(\text{Mn}, \text{Fe})_8\text{Si}_6\text{O}_{15}(\text{OH}, \text{Cl})_{10}$	HEX	0001 perf	Brown	H 4.5-5 G 3.13 fus	Dec by HCl. El clv pos. Pleoc, 0 greenish-yellow, E cols.

	1.669	1.657	.012	HARDYSTONITE (Melilite grp) $\text{Ca}_2\text{ZnSi}_2\text{O}_7$	TET granular	001 good 100, 110 rare	White	H 3 G 3.4 F diff	Gel with acids.
1.653 ^ 1.672	1.669	1.658	.011	GEHLENITE (Melilite grp) $\text{Ca}_2\text{Al}(\text{AlSi})\text{O}_7$	TET	001 imperf	Cols	H 6 G 3.04 F 6	Gel with acids. Data for synth compd.
1.658 ^ 1.710	1.672	1.501	.171	CALCITE (Calcite grp) $(\text{Ca}, \text{Mn}, \text{Mg})\text{CO}_3$	TRIG	10T1 perf	White, pink	H 3 G 2.82 infus	Diss by acids with eff. MnO 4.2, FeO 2.1, MgO 1.3%.
1.652 ^	1.672	1.624	.048	ANNITE (Biotite ser, Mica grp) $\text{KFe}_3\text{AlSi}_3\text{O}_{10}(\text{OH}, \text{F})_2$	MCL ps hex	001 perf	Dark brown	H 2.5-3 G 3.0 F 4	Pleoc, 0 dark brown, E brown. FeO 32.1, Fe_2O_3 3.1, TiO_2 3.6%.
1.669 ^	1.672	1.661	.011	MELILITE (Melilite grp) $(\text{Na}, \text{Ca})_2(\text{Mg}, \text{Fe}, \text{Al})(\text{Si}, \text{Al})_2\text{O}_7$	TET	001 dist	Brown	H 5 G 3.1 F 4	Gel with acids. Fe_2O_3 8.4, Al_2O_3 8.4, FeO 1.1, MgO 4.7, Na_2O 3.8%.
	1.672	1.527	.145	PARALSTONITE $(\text{Ba}, \text{Sr})\text{Ca}(\text{CO}_3)_2$	TRIG	---	White, cols	H 4-4.5 G 3.60 infus	Fluor pale to bright orange in long-wave UV.
1.667 ^	1.673	1.664	.009	PENNANTITE (Chlorite grp) $(\text{Mn}, \text{Fe})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	MCL ps hex	001 perf	Brown	H 2.5 G 3.04	Pleoc in browns. El clv pos.
	1.675	1.59	.085	CHLOROMAGNESITE MgCl_2	HEX plates	---	Cols	H soft G 2.33 F 2	Sol in H_2O , deliq.
1.661 ^ 1.693	1.675	1.643	.032	DRAVITE (Tourmaline grp) $\text{NaMg}_3(\text{Al}, \text{V})_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	11T0, 10T1 very poor	Green	H 7 G 3.13 F 5	Insol in acids. El neg. Pleoc, E pale green, 0 red- brown. MgO 6.3, FeO 1.3, V_2O_5 8.0%.
1.686 ^	1.678	1.630	.048	SPANGOLITE $\text{Cu}_6\text{Al}(\text{SO}_4)(\text{OH})_{12}\text{Cl} \cdot 3\text{H}_2\text{O}$	TRIG tab	0001 perf	Dark- to bluish- green	H 2-3 G 3.14 F 3	Diss by acids. El pos. Pleoc, 0 green, E bluish- green.
1.694 ^	1.680	1.501	.179	DOLOMITE (Dolomite grp) $\text{CaMg}(\text{CO}_3)_2$	TRIG	10T1 perf	White, buff	H 3.5-4 G 2.865 infus	Diss with eff in warm HCl. FeO 0.5%.
1.654 ^	1.680	1.655	.025	SINCOSITE $\text{CaV}^{+4}_2(\text{PO}_4)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	MCL ps tet, tab	001 good 100, 110 poor	Leek-green	H soft G 2.84 fus	Diss by acids to a blue soln. Pleoc, 0 gray- green, E nearly cols.
	1.681	1.643	.038	FRIEDELITE $(\text{Mn}, \text{Fe})_3\text{Si}_6\text{O}_{15}(\text{OH}, \text{Cl})_{10}$	MCL ps trig tab 0001	0001 perf 10T0 imperf	Brown	H 4.5 G 3.3 fus	Dec by HCl. El pos. Pleoc in browns.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.669 ^	1.682	1.650	.032	PYROSMALITE (Fe,Mn) ₈ Si ₆ O ₁₅ (OH,Cl) ₁₀	HEX	0001 perf	Brown	H 4.5-5 G 3.1 fus	Dec by HCl. Anom biax. El clv pos.
	1.683	1.672	.011	VERPLANCKITE Ba ₂ (Mn,Fe ⁺² ,Ti)Si ₂ O ₆ (O,OH,Cl,F) ₂ ·3H ₂ O	HEX	11 $\bar{2}$ 0 good	Orange to yellow	H 2.5-3 G 3.52	Pleoc, 0 orange-yellow, E cols.
v 1.716	1.684	1.672	.012	SVABITE (Apatite grp) Ca ₅ (AsO ₄ ,PO ₄) ₃ (F,OH)	HEX	---	Green	H 4-4.5 G 3.54 F 5	Diss by acids. P ₂ O ₅ 12.5, F 1.4%.
1.661 ^ 1.705	1.685	1.595	.090	STILPNOMELANE K(Fe ⁺³ ,Fe ⁺² ,Mg) ₁₀ Si ₁₂ O ₃₀ (O,OH) ₂	MCL and TCL ps hex	001 mic	Brownish-black	H 3.5 G 2.85 F 3	Dec by acids. El clv pos. Pleoc, 0 dark olive-brown, E yellow. FeO 13.7, Fe ₂ O ₃ 22.0%.
1.649 ^	1.685	1.670	.015	CHAMOSITE (Chlorite grp) (Fe ⁺² ,Fe ⁺³ ,Mg) ₅ Al(Si ₃ Al) ₁₀ (OH,O) ₈	MCL ps hex	001 perf	Brown	H 6 G 2.96	Dec by hot HCl. El clv pos. Pleoc, 0 dark green, E yellow.
	1.685	1.674	.011	PABSTITE Ba(Sn,Ti)Si ₃ O ₉	HEX	---	Cols, white	H 6 G 4.03	Anom blue-violet and yellow interf colors. Fluor bluish white in short-wave UV.
1.678 ^	1.686	1.635	.051	SPANGOLITE Cu ₆ Al(SO ₄)Cl(OH) ₁₂ ·3H ₂ O	TRIG tab	0001 perf	Dark- to bluish-green	H 2-3 G 3.14 F 3	Diss by acids. El clv pos. Pleoc, 0 green, E bluish-green.
	1.687	1.684	.003	JOHNBAUMITE (Apatite grp) Ca ₅ (AsO ₄) ₃ (OH)	HEX	100 dist	Cols, white	H 4.5 G 3.68	Fluor med pink-orange in short-wave UV.
	1.690	1.684	.006	COMBLAINITE (Ni ⁺² ,Co ⁺³)(OH) ₂ (CO ₃)·xH ₂ O	TRIG	---	Turquoise	G 3.05	---
v 1.705	1.691	1.527	.164	BENSTONITE (Ba,Sr) ₆ (Ca,Mn) ₆ Mg(CO ₃) ₁₃	TRIG	10 $\bar{1}$ 1 very good	White	H 3-4 G 3.60 infus	Diss by acids with eff. Fluor red in UV.
	1.691	1.641	.050	BANDYLITE CuB(OH) ₄ Cl	TET tab	001 perf	Deep blue	H 2.5 G 2.81 F 2	Diss by acids or by NH ₄ OH. El clv pos. Pleoc, 0 deep blue, E pale greenish-yellow.

	1.692	1.648	.044	WICKENBURGITE $\text{Pb}_3\text{CaAl}_2\text{Si}_{10}\text{O}_{24}(\text{OH})_6$	HEX tab	0001 indist	White, pink	H 5 G 3.85 F 3	Insol in acids. Fluor dull orange in UV.
1.675 ◇ 1.710	1.693	1.659	.034	SCHORL (Tourmaline grp) $\text{NaFe}_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	$11\bar{2}0$, $10\bar{1}1$ very poor	Black	H 7 G 3.2 F 5	Insol in acids. El neg. Pleoc, abs 0 > E. MgO 7.3, FeO 0.6, Fe_2O_3 9.2, CaO 1.5, Na_2O 2.6, TiO_2 2.2%.
1.680 ◇ 1.700	1.694	1.510	.184	DOLOMITE (Dolomite grp) $\text{Ca}(\text{Mg,Fe})(\text{CO}_3)_2$	TRIG	$10\bar{1}1$ perf	White, brown	H 3.5-4 G 2.92 infus	Diss by warm HCl. FeO 8.4, MnO 1.1, SrO 0.4%.
1.687 ▽ 1.712	1.694	1.519	.175	NORSETHITE (Dolomite grp) $\text{BaMg}(\text{CO}_3)_2$	TRIG	$10\bar{1}1$ perf	White to yellow-green	H 3.5 G 3.74-3.84 infus	Diss by acids.
	1.696	1.694	.002	ALFORSITE (Apatite grp) $\text{Ba}_5(\text{PO}_4)_3\text{Cl}$	HEX	---	Cols	G 4.80	Data for synth.
1.724 ▽	1.699	1.658	.041	PHOSPHURANYLITE $\text{Ca}(\text{UO}_2)_4(\text{PO}_4)_2(\text{OH})_4 \cdot 7\text{H}_2\text{O}$	ORTH plates	100 good 010	Lemon-yellow	H 2.5-3 G 4.1 fus	Diss by acids. Pleoc, E pale yellow, 0 golden-yellow. U biax, 2V 0-20°. Not fluor in UV.
1.694 ◇ 1.710	1.700	1.21	.179	DOLOMITE, zincian (Dolomite grp) $\text{Ca}(\text{Mg,Zn})(\text{CO}_3)_2$	TRIG	$10\bar{1}1$ perf	White, brown	H 3.5-4 G 2.95 infus	Diss by warm HCl. Fluor red in UV. FeO 2.2, ZnO 8.3, PbO 1.1%.
1.711 ▽	1.700	1.509	.191	MAGNESITE (Calcite grp) MgCO_3	TRIG u mass	$10\bar{1}1$ perf	Cols, white	H 3-4 G 2.96 infus	Diss by warm acids. Data for pure MgCO_3 .
1.714 ▽	1.702	1.700	.002	VESUVIANITE $\text{Ca}_{10}(\text{Mg,Fe})_2\text{Al}_4(\text{Si}_{20}\text{O}_{70})_2$ $(\text{SiO}_4)_5(\text{OH,F})_4$	TET	110 poor	Brown, green, blue	H 5.5-6 G 3.40 F 4	Nearly insol in HCl. FeO 2.0, Fe_2O_3 2.3, TiO_2 1.7%.
1.655 ◇ 1.752	1.703	1.699	.004	APATITE-BRITHOLITE (Apatite grp) $(\text{Ca,Ce})_5(\text{PO}_4)_3\text{SiO}_4\text{F}$	HEX mass	---	Reddish-brown	H 4.5 G 3.83 infus	Dec by HCl. P_2O_5 17.3, SiO_2 12.4, Ce_2O_3 12.9%.
	1.704	1.679	.025	SCHALLERITE $(\text{Mn,Fe})_{16}\text{Si}_{12}\text{As}^{+3}_3\text{O}_{36}(\text{OH})_{17}$	TRIG	0001 perf	Reddish-brown	H 4.5 G 3.37 F diff	Slowly dec by HCl. El clv pos. MnO 44.7, FeO 2.1%.
1.691 ^	1.705	---	very str	BENSTONITE $(\text{Ba,Sr})_6(\text{Ca,Mn})_6\text{Mg}(\text{CO}_3)_{13}$	TRIG	$10\bar{1}1$ very good	White	H 3-4 G 3.58 infus	Diss by HCl with eff. BaO 44.2, CaO 12.3, PbO 2.8, MgO 4.7, MnO 1.6%.
1.685 ◇ 1.735	1.705	1.596	.109	STILPNOMELANE $\text{K}(\text{Fe}^{+3}, \text{Fe}^{+2}, \text{Mg})_{10}\text{Si}_{12}\text{O}_{30}$ $(\text{O,OH})_2$	MCL and TCL ps hex	001 mic	Brownish-black	H 3.5 G 2.85 F 3	Dec by acids. El clv pos. Pleoc, E golden-yellow, 0 dark red-brown. Fe_2O_3 23.9, FeO 13.2%.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.700 ∧ 1.711	1.710	1.519	.191	KUTNOHORITE (Dolomite grp) $\text{Ca}(\text{Mn},\text{Mg})(\text{CO}_3)_2$	TRIG	10 $\bar{1}$ 1 perf	Pink, brown	H 3-4 G 3.00 infus	Diss by hot acids. MnO 17.6, MgO 10.0, FeO 1.3%.
1.672 ∧ 1.713	1.710	1.523	.189	CALCITE, cadmian (Calcite grp) $(\text{Ca},\text{Cd})\text{CO}_3$	TRIG	10 $\bar{1}$ 1 perf	Cols to yellow	G 2.9 infus	CaCO_3 73, CdCO_3 27%.
1.693 ∧ 1.735	1.710	1.664	.046	DRAVITE, chromian (Tourmaline grp) $\text{NaMg}_3(\text{Al},\text{Cr})_6(\text{BO}_3)_3\text{Si}_6\text{O}_{18}(\text{OH})_4$	TRIG pris	11 $\bar{2}$ 0, 10 $\bar{1}$ 1 poor	Green	H 7 G 3.13 infus	Insol in acids. El neg. Pleoc, abs 0 > E. Cr_2O_3 17.8, Fe_2O_3 1.0, FeO 1.8%.
1.700 ∧ 1.726	1.711	1.519	.192	MAGNESITE, nickeloan (Calcite grp) $(\text{Mg},\text{Ni})\text{CO}_3$	TRIG	10 $\bar{1}$ 1 perf	Green	H 3-4 G 3.15 infus	Diss by hot acids. NiO 13.55%.
1.710 ∧ 1.740	1.711	1.520	.191	DOLOMITE, ferroan (Dolomite grp) $\text{Ca}(\text{Mg},\text{Fe})(\text{CO}_3)_2$	TRIG	10 $\bar{1}$ 1 perf	Brown	H 3.5-4 G 3.01 infus	Diss by warm HCl. FeO 12.6, MnO 1.2%.
1.694 ^	1.712	1.512	.200	NORSETHITE (Dolomite grp) $\text{Ba}(\text{Mg},\text{Fe})(\text{CO}_3)_2$	TRIG	10 $\bar{1}$ 1 perf	White to yellow-green	H 3.5 G 3.85 infus	Diss by acids. FeO 3.6, MnO 0.8%.
	1.712	---	str	PALMIERITE $(\text{K},\text{Na})_2\text{Pb}(\text{SO}_4)_2$	TRIG plates	---	Cols	G 4.5 F easy	Dec by boiling H_2O , diss by HNO_3 . Luster pearly.
1.710 ∧ 1.731	1.713	1.519	.194	CALCITE, manganoan (Calcite grp) $(\text{Ca},\text{Mn})\text{CO}_3$	TRIG	10 $\bar{1}$ 1 perf	Pink	H 3 G 3.02 infus	Diss by acids with eff. MnO 20.1, ZnO 0.4%.
	1.714	1.702	.012	TRASKITE $\text{Ba}_9\text{Fe}_2\text{Ti}_2\text{Si}_{12}\text{O}_{36}(\text{OH},\text{Cl},\text{F})_6 \cdot 6\text{H}_2\text{O}$	HEX	Conch	Brownish-red	H 5 G 3.71 F 3.5	Insol in acids. Pleoc, 0 brown-red, E cols to straw-yellow.
1.702 ∧ 1.742	1.714	1.709	.005	VESUVIANITE $\text{Ca}_{10}\text{Mg}_2\text{Al}_4(\text{Si}_{20}\text{O}_{70})_2(\text{SiO}_4)_5(\text{OH},\text{F})_4$	TET	110 poor	Brown	H 5 G 3.38 F 4	Nearly insol in HCl. Fe_2O_3 3.3, MnO 3.3, BeO 1.6%.
1.684 ^	1.716	1.698	.018	SVABITE (Apatite grp) $\text{Ca}_5(\text{AsO}_4)_3(\text{Cl},\text{F})$	HEX	---	Green, brown	H 4-4.5 G 3.5 F 5	Diss by acids. Cl 1.9, F 0.45%.
	1.721	1.719	.002	SATTERLYITE $(\text{Fe}^{+2},\text{Mg},\text{Fe}^{+3})_2(\text{PO}_4)(\text{OH})$	HEX	---	Pale brown, pale yellow	H 4.5-5 G 3.68 fus	Diss by acids. Pleoc, 0 pale yellow, E brownish-yellow, abs E > 0.

	1.722	1.718	.004	TAAFFEITE $Mg_3Al_8BeO_{16}$	HEX	---	Mauve	G 3.61	---
	1.723	1.716	.007	HATRURITE Ca_3SiO_5	ps hex	---	Cols	---	Dec by H_2O .
1.699 ^	1.724	1.674	.050	PHOSPHURANYLITE $Ca(UO_2)_4(PO_4)_2(OH)_4 \cdot 7H_2O$	ORTH plates	100 good 010	Lemon- yellow	H 2.5-3 G 4.1 fus	Diss by acids. Pleoc, E pale yellow, 0 golden- yellow. U biax, 2V 0-20°.
1.65 ^	1.724	1.682	.042	PYROCHROITE $Mn(OH)_2$	TRIG	0001 perf	White to brown	H 1.5-2 G 3.1-3.25 infus	Diss by acids. El clv pos. Darkens in sunlight. Abs 0 > E.
1.711 v 1.788	1.726	1.528	.198	MAGNESITE (Calcite grp) $(Mg,Fe)CO_3$	TRIG	10Tl perf	White to brown	H 3-4 G 3.10 infus	Diss by hot acids. FeO 9.5, CaO 0.4%.
	1.727	---	---	KITTATINNYITE $Ca_4Mn^{+3}_4Mn^{+2}_2Si_4O_{16}(OH)_8 \cdot 18H_2O$	HEX	0001 perf	Bright yellow	H ~ 4 G 2.61	Pleoc wk, abs E > 0.
	1.728	---	---	WALLKILLDELLITE $Ca_4Mn^{+2}_6As^{+5}_4O_{16}(OH)_8 \cdot 18H_2O$	HEX	0001 perf	Dark red	G 2.85	Pleoc, 0 reddish-orange, E orange-pink, abs 0 > E.
	1.73	1.69	.04	BIRNESSITE $Na_4Mn_{14}O_{27} \cdot 9H_2O$	ORTH mass	---	Brownish- black	H 1.5 G 3.0 infus	Diss by HCl. Nearly opaque.
	~1.73	1.72	.01	MELANOCERITE $(Ca,Ce,Y)_5(Si,B)_3O_{12}(OH,F) \cdot xH_2O$	HEX tab	Conch	Deep brown to black	H 5-6 G 4.13 infus	Dec by acids. Pale yellow in section.
1.713 v 1.753	1.731	(1.55)	.18	RHODOCHROSITE (Calcite grp) $(Mn,Ca)CO_3$	TRIG	10Tl perf	Pink	H 4 G 3.05 infus	Diss by hot acids. MnO 33.4, CaO 21.0, FeO 2.7, MgO 2.2%.
□	1.733	1.714	.019	HEMATOLITE $(Mn,Mg,Al)_{15}(AsO_4)_2AsO_3(OH)_{23}$	TRIG	0001 perf	Brown to red	H 3.5 G 3.49 infus	Diss by acids. El clv pos. In section yellow to brown. Anom biax, small 2V.
v 1.750	1.734	1.592	.192	MINRECORDITE (Dolomite grp) $Ca(Zn,Mg)(CO_3)_2$	TRIG	10T4 perf	Cols	H 3.5 G 3.32	Slowly diss by cold HCl.
1.705 ^	1.735	1.625	.110	STILPNOMELANE $K(Fe,Mg,Al)_{10}Si_{12}O_{30}(OH)_{12}$	MCL and TCL ps hex	001 mic	Brownish- black	H 3.5 G 2.83 F 3	Dec by acids. El clv pos. Pleoc, E golden yellow, 0 deep red-brown. Fe_2O_3 31.7, FeO 1.3, MnO 2.6%.
1.72	1.735	(~1.650)	~.085	REEVESITE $Ni_6Fe_2(OH)_{16}CO_3 \cdot 4H_2O$	TRIG	---	Yellow to greenish- yellow	G 2.80-2.88 infus	Diss by acids.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.710 ^	1.735	1.655	.080	BUERGERITE (Tourmaline grp) $\text{NaFe}_3\text{Al}_6(\text{BO}_3)_3\text{Si}_6\text{O}_{21}\text{F}$	TRIG pris	Prism, dist	Dark brown	H 7 G 3.31 F diff	Insol in acids. El neg. Pleoc, 0 yellow-brown, E pale yellow.
	1.738	1.728	.010	STOTTITE $\text{FeGe}(\text{OH})_6$	TET ps oct	100, 010 good, 001	Brown	H 4-5 G 3.60	Luster greasy. Anom biax.
1.711(?) ^ 1.741	1.740	1.547	.193	KUTNOHORITE (Dolomite grp) $\text{CaMn}(\text{CO}_3)_2$	TRIG	10Tl perf	Pink, brown	H 3-4 G 3.14 infus	Diss by hot acids. MnO 31.8, ZnO 1.4, FeO 0.8, MgO 0.5%.
1.740 ^	1.741	1.536	.205	ANKERITE (Dolomite grp) $\text{Ca}(\text{Fe,Mg})(\text{CO}_3)_2$	TRIG	10Tl perf	Brown	H 3.5-4 G 3.12 infus	Diss by hot HCl. FeO 24.0, MnO 1.6%.
1.714 ^ 1.762	1.742	1.736	.006	VESUVIANITE $\text{Ca}_{10}(\text{Mg,Fe,Ti})_2\text{Al}_4(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)_5(\text{OH})_4$	TET	110 poor	Brown	H 5 G 3.41 F 4	Insol or partly dec by HCl. Fe_2O_3 4.4, FeO 1.6, TiO_2 4.7%.
	1.745	1.565	.180	ZHONGHUACERITE $\text{Ba}_2\text{Ce}(\text{CO}_3)_3\text{F}$	TRIG	---	Yellow	H 4.5-5 G 4.3	Diss by acids.
	1.747	1.741	.006	AKDALAITE $4\text{Al}_2\text{O}_3 \cdot \text{H}_2\text{O}$	HEX	Fr irreg	White	H 7 G 3.68	---
	1.748	1.645	.103	LAVENDULAN $\text{NaCaCu}_5(\text{AsO}_4)_4\text{Cl} \cdot 5\text{H}_2\text{O}$	ORTH ps hex	010 good 100, 001 dist	Greenish-blue	H 2.5 G 3.54 F easy	Diss by acids. Pleoc, E blue to greenish-blue, 0 pale blue to pale greenish-blue, abs E > 0.
1.734 ^	1.750	1.550	.200	MINRECORDITE (Dolomite grp) $\text{CaZn}(\text{CO}_3)_2$	TRIG	10T4 perf	Milk-white	H 3.5 G 3.45	Slowly diss by cold HCl.
	1.75	1.728	.02	JASMUNDITE $\text{Ca}_{11}(\text{SiO}_4)_2\text{O}_2\text{S}$	TET	---	---	G 3.03	---
	1.751	1.645	.106	CLARAITE $(\text{Cu,Zn})_3(\text{CO}_3)(\text{OH}) \cdot 4\text{H}_2\text{O}$	TCL (?) ps hex spherulitic	10T0 perf	Bluish-green	H ~ 2 G 3.33	r > v.
[]	1.751	1.748	.003	CAPPELENITE $(\text{Ba,Ca,Na})(\text{Y,La})_6\text{B}_6\text{Si}_3(\text{O,OH})_{27}$	HEX	---	Green to brown	H 6-6.5 G 4.41 F diff	Diss by HCl, gel.
1.703 ^ 1.823	1.752	1.748	.004	BRITHOLITE $(\text{Ce,Ca,Y})_5(\text{SiO}_4)_3(\text{PO}_4)_3(\text{OH,F})$	HEX	---	Brown	H 5 G 4.1-4.4 infus	Gel with acids. Apatite structural type.

	1.753	1.740	.004	OKANOGANITE (Na,Ca) ₃ (Y,Ce,Nd,La) ₁₂ Si ₆ B ₂ O ₂₇ F ₁₄	TRIG	---	Pale pink	H 4 G 4.35	Tw.
1.731 ∧ 1.803	1.753	1.560	.193	RHODOCHROSITE (Calcite grp) (Mn,Ca)CO ₃	TRIG	10I1 perf	Pink	H 4 G 3.3 infus	Diss by hot acids. MnO 44.3, CaO 13.4, MgO 2.3, FeO 0.8%.
	1.753	1.717	.036	WISERITE Mn ₄ B ₂ O ₅ (OH,Cl) ₄	TET fib	Clv across fibers, perf	Dark brown	G 3.42	Diss by HCl.
	1.755	1.731	.024	CONGOLITE (Fe,Mn,Mg) ₃ B ₇ O ₁₃ Cl	TRIG	---	Pale red	G 3.58	Diss by acids.
	1.756	1.680	.076	NATISITE Na ₂ TiSiO ₅	TET	001 perf 100 less so	Yellow- green to greenish- gray	H 3-4 G 3.15 calc F easy	Insol in dilute acids.
1.765 └	1.760	1.577	.183	CORDYLITE Ba(Ce,La) ₂ (CO ₃) ₃ F ₂	HEX pyram	0001 dist	Wax-yellow	H 4.5 G 4.10 infus	Diss by acids. Pleoc, 0 green-yellow, E brownish- yellow.
1.742 ∧ 1.795	1.762	1.750	.012	VESUVIANITE (Ca,Ce) ₁₀ (Mg,Fe) ₂ Al ₄ (Si ₂ O ₇) ₂ (SiO ₄) ₅ (OH) ₄	TET	110 poor	Dark brown	H 5 G 3.60 F 4	Insol in acids. (Ce,La) ₂ O ₃ 16.7, Fe ₂ O ₃ 5.4, FeO 5.5%.
	1.765	1.735	.030	SHERWOODITE Ca ₉ Al ₂ V ⁺⁵ ₂₄ V ⁺⁴ ₄ O ₈₀ ·56H ₂ O	TET	Conch to uneven	Blue-black	H 2 G 2.8	Diss by acids. Pleoc, 0 green, E blue.
	1.766	1.585	.178	HUANGHOITE Ba(Ce,La)(CO ₃) ₂ F	HEX	0001	Honey- yellow to yellow- green	H 4.5-5 G 4.58 infus	Diss by acids. Pleoc wk in yellow-green.
1.794 ∨	1.766	1.758	.008	CORUNDUM Al ₂ O ₃	TRIG	Parting 0001, 00I1 sometimes perf	Cols, red, blue, brown	H 9 G 4.00 infus	Insol in acids. Colored var pleoc. Anom biax, 2V 0-58°
	1.772	1.770	.002	SWEDENBORGITE NaBe ₄ SbO ₇	HEX pris	0001 dist	Cols to wine- yellow	H 8 G 4.29	Insol in acids. El crystal neg.
	1.775	1.765	.010	FRESNOITE Ba ₂ TiSi ₂ O ₈	TET	001 fair	Lemon- to canary- yellow	H 3-4 G 4.43 F 4	Insol in cold, dec by hot HCl. Pleoc, 0 cols, E yellow.
	1.778	1.660	.118	NORDENSKIÖLDINE CaSnB ₂ O ₆	TRIG tab	0001 perf fr conch	Cols to yellow	H 5.5-6 G 4.20 infus	Diff diss by acids. El neg.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
114	v 1.846	1.778 1.772	.006	CHROMDRAVITE (Tourmaline grp) $\text{Na}(\text{Mg}, \text{V}, \text{Al})_3(\text{Cr}, \text{Fe}^{+3})_6\text{B}_3\text{Si}_6\text{O}_{27}(\text{OH})_4$	TRIG	---	Dark green	G 3.40	Pleoc, 0 dark green, E yellow-green, abs 0 > E.
		1.78 1.75	.03	HIBONITE (Ca,Ce)(Al,Ti,Fe,Mg) $_{12}\text{O}_{19}$	HEX	0001 easy, parting 1010	Brownish-black	H 7.5-8 G 3.84 infus	Insol in acids. Pleoc, 0 cols, E blue.
		1.782 1.780	.002	CLARINGBULLITE $\text{Cu}_8\text{Cl}_2(\text{OH})_{14}\cdot\text{H}_2\text{O}$	HEX plates	0001 perf 1010, 1120 dist	Blue	G 3.9	Abnormal interf colors dark blue to yellow-green.
	1.726 v 1.816	1.788 1.570	.218	MAGNESITE-SIDERITE (Calcite grp) (Mg,Fe) CO_3	TRIG	1011 perf	White to brown	H 4 G 3.43 infus	Diss by hot acids. Calc for Mg:Fe = 1:1, FeO 35.9, MgO 20.1%.
	v 1.820	1.791 1.705	.086	JAROSITE (Alunite grp) $\text{KFe}_3(\text{SO}_4)_2(\text{OH})_6$	TRIG tab, rhombs	0001 dist	Yellow to brown	H 2.5-3.5 G 2.9-3.2 F 4.5	Diss by HCl. Pleoc, 0 yellow to brown, E paler to cols. Basal section may be divided into biax segments.
	1.766 ^	1.794 1.785	.009	CORUNDUM (Al,Fe) $_{203}$	TRIG	Parting 0001, 0011 sometimes perf	Brown	H 9 G 4.1 infus	Insol in acids. Fe_2O_3 9.2, TiO_2 0.4%.
	1.762 ^	1.795 1.775	.020	VESUVIANITE (Ca,Sb) $_{10}(\text{Mg}, \text{Fe})_2\text{Al}_4(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)_5(\text{OH})_4$	TET pris	110 poor	Pale green	H 5 F 4	Insol in acids. El neg. E cols, 0 pale yellow-green. Sb_2O_3 15.7, Fe_2O_3 3.3, FeO 4.1%.
		1.80 ---	str	CRONSTEDTITE (Kaolinite-Serpentine grp) $(\text{Fe}^{+2}, \text{Mg})\text{Fe}^{+3}(\text{SiFe})\text{O}(\text{OH})$	HEX, TRIG, and MCL polytypes	0001 perf	Black	H 3.5 G 3.34 F 4	Gel with acids. Pleoc, E dark reddish-brown, 0 nearly opaque.
		1.800 1.743	.057	FERRIDRAVITE (Tourmaline grp) $(\text{Na}, \text{K})(\text{Mg}, \text{Fe}^{+2})_3\text{Fe}^{+3}_6\text{B}_3\text{Si}_6\text{O}_{27}(\text{O}, \text{OH})_4$	TRIG	Uneven	Black	G 3.26	Streak brown. Insol in acids.
		1.800 1.750	.050	AMMONIOJAROSITE (Alunite grp) $(\text{NH}_4)\text{Fe}_3(\text{SO}_4)_2(\text{OH})_6$	TRIG tab	0001 dist	Pale yellow	G 3.11	Diss by HCl.

	1.80	1.795	.005	NIGERITE (Zn,Mg,Fe)(Sn,Zn) ₂ (Al,Fe) ₁₂ O ₂₂ (OH) ₂	TRIG	---	Dark brown	H 8-9 G 4.25	Insol in acids. Pleoc, 0 yellow-brown, E yellow. U opt pos.
	1.802	1.740	.062	QUETZALCOATLITE Zn ₈ Cu ₄ (TeO ₃) ₃ (OH) ₁₈	HEX	10T0 fair	Blue	H 3 G 6.05	Diss by acids. Pleoc, 0 blue-green, E nearly cols.
1.753 ∧ 1.816	1.803	1.584	.219	RHODOCHROSITE (Calcite grp) (Mn,Ca)CO ₃	TRIG	10T1 perf	Pink	H 4 G 3.65 infus	Diss by hot acids. FeO 3.8, MgO 0.6, CaO 1.3%.
	1.804	1.773	.031	CYRILLOVITE NaFe ₃ (PO ₄) ₂ (OH) ₄ ·2H ₂ O	TET	---	Yellow, brown, yellow- green	G 3.08 F 2-3	Diss by hot acids. Pleoc faint, abs 0 > E.
∨ 1.851	1.805	1.783	.022	HÖGBOMITE (Mg,Fe) ₂ (Al,Ti) ₅ O ₁₀	HEX and TRIG polytypes	10T0 good parting 0001	Brown, yellow	H 6.5 G 3.70 infus	Insol in acids. Pleoc, 0 deep yellow, E yellow. TiO ₂ 9.1, FeO 7.8%. Anom biax.
1.711 ∧ 1.84	1.81	1.58	.23	GASPEITE (Calcite grp) (Ni,Mg)CO ₃	TRIG	10T1 perf	Green	H 3-4 G 3.6 infus	Diss by hot acids. NiO 36.8, MgO 14.8, FeO 4.3%.
∨ 1.832	1.815	1.740	.075	NATROJAROSITE (Alunite grp) NaFe ₃ (SO ₄) ₂ (OH) ₆	TRIG plates	0001 perf	Yellow, brown	H 3 G 3.18 F 4.5	Diss by HCl. El pos. Pleoc in yellows, abs 0 > E. Anom biax. Basal sec- tions divided into seg- ments.
∨ 1.850	1.815	1.601	.214	SMITHSONITE (Calcite grp) (Zn,Mg)CO ₃	TRIG	10T1 perf	Milky white	H 4.5 infus	Diss by hot acids. ZnO 53.2, MgO 7.6, CaO 1.1%.
	1.815	1.761	.054	MOLYBDOPHYLLITE Pb ₂ Mg ₂ Si ₂ O ₇ (OH) ₂	TRIG lam	0001 perf	Cols to pale green	H 3-4 G 4.72 F diff	Cols in section. El pos.
1.788 ∧ 1.836	1.816	1.592	.224	SIDERITE (Calcite grp) (Fe,Mg,Mn)CO ₃	TRIG	10T1 perf	Brown	H 4 G 3.59 infus	Diss by warm acids. MgO 11.8, MnO 5.9, CaO 1.4%.
1.803 ∧ 1.836	1.816	1.597	.219	RHODOCHROSITE (Calcite grp) MnCO ₃	TRIG	10T1 perf	Pink	H 4 G 3.70 infus	Diss by warm acids. Calc for pure end member.
	1.816	1.728	.088	HYDRONIUM JAROSITE (Alunite grp) (H ₃ O)Fe ³⁺ ₃ (SO ₄) ₂ (OH) ₆	TRIG	0001	Yellow, brown	H 4-4.5 G 2.5-2.9, 3.17 calc	Diss by HCl.
	1.816	1.788	.028	PAINITE CaZrBa ₉ O ₁₈	HEX	---	Garnet-red	H 8 G 4.00	Insol in acids.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.791 ^	1.820	1.715	.105	JAROSITE (Alunite grp) $\text{KFe}_3(\text{SO}_4)_2(\text{OH})_6$	TRIG tab, rhombs	0001 dist	Yellow to brown	H 2.5-3.5 G 3.2 F 4.5	Diss by HCl. Pleoc in brown and yellow, abs 0 > E. Anom biax, with basal segments.
	1.82	1.78	.04	NANLINGITE $\text{CaMg}_4(\text{AsO}_3)_2\text{F}_4$	TRIG	One imperf	Brownish- red	G 3.93	---
1.752 ^	1.823	1.817	.006	BRITHOLITE $(\text{Ce}, \text{Ca}, \text{Y})_5(\text{SiO}_4, \text{PO}_4)_3(\text{OH}, \text{F})$	HEX	---	Brown	H 5 G 4.77 infus	Gel with acids. Apatite structural type.
	1.82- 1.83	---	---	ZAIRITE (Crandallite grp) $\text{Bi}(\text{Fe}, \text{Al})_3(\text{PO}_4)_2(\text{OH})_6$	TRIG	---	Greenish	G 4.37	---
1.816 ^ 1.836	1.830	1.605	.225	OTAVITE (Calcite grp) $(\text{Cd}, \text{Ca})\text{CO}_3$	TRIG	10Tl perf	White, yellow	G ~ 4.9 infus	Diss by acids with eff. CdCO_3 91, CaCO_3 9%.
1.815 ^	1.832	1.750	.082	NATROJAROSITE (Alunite grp) $\text{NaFe}_3(\text{SO}_4)_2(\text{OH})_6$	TRIG plates	0001 perf	Yellow, brown	H 3 G 3.18 F 4.5	Diss by HCl. El pos. Pleoc in yellow, abs 0 > E. Anom biax with basal seg- ments.
1.816 ^ 1.840	1.836	(1.610)	.226	RHODOCHROSITE (Calcite grp) $(\text{Mn}, \text{Zn}, \text{Fe})\text{CO}_3$	TRIG	10Tl perf	Pink	H 4 G 3.76 infus	Diss by warm acids. MnO 29.8, FeO 13.9, ZnO 14.9, CaO 3.1%.
	1.837	1.833	.004	JEANBANDYITE $(\text{Fe}^{+3}, \text{Mn}^{+2})\text{Sn}(\text{OH})_6$	TET ps cub	001 and 100 fair	Brown-orange	H 3.5 G 3.81	Diss slowly by cold HCl.
1.836 ^ 1.849	1.840	1.615	.225	RHODOCHROSITE, ferroan (Calcite grp) $(\text{Mn}, \text{Fe})\text{CO}_3$	TRIG	10Tl perf	Brown	H 4 G 3.72 infus	Diss by warm acids. MnO 35.4, FeO 26.1%.
1.81 ^	1.84	(1.62)	.22	GASPEITE (Calcite grp) NiCO_3	TRIG	10Tl perf	Emerald- green	H 3-4 G 3.7 infus	Diss by hot acids. NiO 55.5, MgO 3.3, FeO 0.1%.
1.78 ^	1.846	1.804	.042	HIBONITE $(\text{Ca}, \text{Ce})(\text{Al}, \text{Ti}, \text{Fe})_{12}\text{O}_{19}$	HEX	0001 easy, parting 10T0	Brownish- black	H 7.5-8 G 3.88 infus	Insol in acids. Pleoc, abs E > 0.
1.840 ^ 1.860	1.849	1.615	.234	SIDERITE, manganoan (Calcite grp) $(\text{Fe}, \text{Mn})\text{CO}_3$	TRIG	10Tl perf	Brown	H 3.5 G 3.81 infus	Diss by warm acids. FeO 47.9, MnO 9.5, MgO 3.2, CaO 0.2%.

1.815 △	1.850	1.623	.227	SMITHSONITE (Calcite grp) $ZnCO_3$	TRIG	10T1 perf	Cols, green	H 4.5 G 4.42 infus	Diss by acids. FeO 0.4, MnO 0.5%.
1.83 □ 1.87	1.85	(~1.81)	~.04	BEAVERITE (Alunite grp) $Pb(Cu,Fe,Al)_3(SO_4)_2(OH)_6$	TRIG plates, u earthy	---	Canary- yellow	G 4.08-4.36	Diss by acids.
1.805 ^	1.851	1.802	.049	HÖGBOMITE (Mg,Fe) $_2(Al,Ti)_5O_{10}$	HEX and TRIG (polytypes)	10T0 good parting 0001	Brown, yellow	H 6.5 G 3.81 infus	Insol in acids. Pleoc, 0 dark brown, E med-brown. Anom biax.
	1.855	1.60	.255	SPHAEROCOBAULTITE (Calcite grp) $CoCO_3$	TRIG	10T1 perf	Rose-red	H 3-4 G 4.1 infus	Diss by acids. Pleoc, 0 violet-red, E rose red.
1.849 ^ 1.875	1.860	1.625	.235	SIDERITE, zincian (Calcite grp) $(Fe,Mn,Zn)CO_3$	TRIG	10T1 perf	Brown	H 4 G 3.92 infus	Diss by hot acids. MnO 14.65, ZnO 11.55, MgO 0.3, FeO 36.4%.
	1.86	1.83	.03	ASBECASITE $Ca_3(Ti,Sn)As_6Be_2Si_2O_{20}$	TRIG	10T1 perf	Lemon- yellow	H 6.5-7 G 3.70	Anom biax, 2V 0-17°.
v 1.915	1.860	1.855	.005	BYSTROMITE $MgSb_2O_6$	TET u mass	---	Blue-gray	H 4 G 5.7	Insol in dil acids, diss by concd HCl + KI. Opt char unk.
v 1.898	1.870	1.792	.078	ARSENIOSIDERITE $Ca_3Fe_4(AsO_4)_4(OH)_6 \cdot 3H_2O$	MCL ps tet, fib	001 perf	Yellow, brown	H 4.5 G 3.5-3.6 F 3	Diss by acids. Pleoc, 0 brownish-red, E pale yellow.
	1.870	1.845	.025	DUSSERTITE (Crandallite grp) $BaFe_3(AsO_4)_2(OH)_5 \cdot H_2O$	TRIG	0001	Yellow, green	H 3.5 G 3.75	Diss by acids. Anom biax, 2V 0-20°.
1.860 ^	1.875	1.633	.242	SIDERITE (Calcite grp) $FeCO_3$	TRIG	10T1 perf	Brown	H 4 G 3.89 infus	Diss by hot acids. Calc for pure end-member.
	1.875	1.785	.090	PLUMBOJAROSITE (Alunite grp) $PbFe_6(SO_4)_4(OH)_{12}$	TRIG plates	10T4 fair	Brown	H soft G 3.63	Diss by HCl. Pleoc, 0 brown-red, E pale golden- yellow.
	1.882	1.785	.097	ARGENTOJAROSITE (Alunite grp) $AgFe_3(SO_4)_2(OH)_6$	TRIG plates	0001	Yellow, brown	G 3.66	Diss by HNO_3 . El pos. Pleoc in yellow, abs 0 > E.
1.870 ^	1.898	1.815	.083	ARSENIOSIDERITE $Ca_3Fe_4(AsO_4)_4(OH)_6 \cdot 3H_2O$	MCL ps tet, fib	001 perf	Yellow, brown	H 4 G 3.5-3.6 F 3	Diss by acids. Pleoc, 0 brownish-red, E pale yellow.
	1.91	(1.90)	.01	DAUBREEITE $BiO(OH,Cl)$	TET	001 perf	Yellow, brown	H 2-2.5 G 6.4-7.5 F easy	Diss by HCl. El clv pos.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
1.860 \wedge	1.915	1.908	.007	BYSTROMITE MgSb_2O_6	TET u mass	---	Blue-gray	H 7 G 5.7	Insol in dil acids, diss by concd HCl + KI. Opt char unk.
1.957 \vee	1.916	1.909	.007	BEUDANTITE (Beudantite grp) $\text{PbFe}_3(\text{AsO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001 easy	Olive, brown, yellow	H 4 G 4.0-4.3 F 3.5	Diss by acids. Pleoc, 0 yellow, E pale yellow. Anom biax. As:S = 4:1.
	1.93	---	wk	CORKITE (Beudantite grp) $\text{PbFe}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001 easy	Yellow, brown, greenish	H 4 G 4.2-4.3 F 4-5	Diss by HNO_3 . Abnormal green interf colors. Anom biax.
1.916 \wedge 1.98	1.957	1.943	.014	BEUDANTITE (Beudantite grp) $\text{PbFe}_3(\text{AsO}_4)(\text{SO}_4)(\text{OH})_6$	TRIG	0001 easy	Yellow, brown, black	H 4 G 4.0-4.3 F 3.5	Diss by HNO_3 . Pleoc, 0 yellow to red-brown, E cols to yellow. Anom biax. As:S = 1:1.
	1.97	---	wk	DIXENITE $\text{Cu}^{+1}\text{Mn}_{14}\text{Fe}^{+3}(\text{As}^{+3}\text{O}_3)_5$ $(\text{As}^{+5}\text{O}_4)(\text{SiO}_4)_2(\text{OH})_6$	TRIG plates	0001 mic	Black	H 3-4 G 4.20	Dec by HCl. Glowing red in transmitted light. El pos.
	1.977	1.967	.010	DUGGANITE $\text{Pb}_3(\text{Zn,Cu})_3(\text{Te}^{+6}\text{O}_6)\text{AsO}_4(\text{OH})_3$	HEX	---	Cols to green	H 3 G 6.33	---
	1.98	1.85	.13	DIABOLEITE $\text{Pb}_2\text{CuCl}_2(\text{OH})_4$	TET tab	001 perf	Sky-blue	H 2.5 G 5.42 F easy	Diss by HNO_3 . Pleoc, 0 deep blue, E nearly cols. El pos.
	1.99	---	mod	Unnamed (Crandallite grp) $\text{PbFe}_3(\text{AsO}_4)_2(\text{OH})_5\cdot\text{H}_2\text{O}$	TRIG	0001	Brown	---	Diss by HNO_3 . Anom biax, 2V 0-50°.
	2.0 red	---	extr	GRAPHITE C	HEX and TRIG	0001 perf	Black	H 1-2 G 2.2 infus	Insol in acids. Thinnest flakes are deep blue in transmitted light. Strongly pleoc. El pos.
\square	2.0	(1.8)	.025	JAGOITE $\text{Pb}_3\text{FeSi}_3\text{O}_{10}(\text{OH,Cl})$	TRIG	0001 perf	Yellow-green	H 3 G 5.43	El clv pos.
	2.00	2.14	.14	WAKEFIELDITE YVO_4	TET	---	Pale tan	H 5 G (4.26)	---
	2.01	1.99	.02	ARMANGITE $\text{Mn}_{26}\text{As}_{18}\text{O}_{50}(\text{OH})_4\text{CO}_3$	HEX	0001 fair	Black	H 4 G 4.43 F easy	Diss by HCl. Streak brown.

2.041 ^v	2.026	1.965	.061	CUMENGITE $\text{Pb}_4\text{Cu}_4\text{Cl}_8(\text{OH})_8 \cdot \text{H}_2\text{O}$	TET	101 good 110 dist	Indigo-blue	H 2.5 G 4.67 F 1	Diss by warm HNO_3 .
	>2.0	>2.0	str	KUSUITE $(\text{Ce}^{+3}, \text{Pb}^{+2}, \text{Pb}^{+4})\text{VO}_4$	TET	---	Black	H 4.5 G 5.30 calc	Streak rust-brown. In section honey-yellow.
	>2.0	>2.0	---	DREYERITE BiVO_4	TET plates	---	Orange- to brownish-yellow	H 2-3 G (6.25)	Pleoc, 0 bright yellow, E brownish-yellow.
	2.026	2.010	.016	HEDYPHANE (Apatite grp) $(\text{Ca}, \text{Pb})_5(\text{AsO}_4)_3\text{Cl}$	HEX (mcl, ps hex?)	10Tl	White	H 4.5 G 5.82 F 2	Diss by HNO_3 .
	2.03	2.00	.03	PSEUDOBOLEITE $\text{Pb}_5\text{Cu}_4\text{Cl}_{10}(\text{OH})_8 \cdot 2\text{H}_2\text{O}$	TET ps cub	001, 101 perf	Indigo-blue	H 2.5 G 4.85 F 1	Diss by warm HNO_3 . Pearly luster on clv.
	2.033	2.015	.018	BARYSILITE $\text{Pb}_8\text{Mn}(\text{Si}_2\text{O}_7)_3$	TRIG tab	0001 dist	Gray, white	H 3 G 6.72 F 2.5	Gel with acids. MnO 3.3, MgO 0.8, ZnO 0.3%.
	2.04	2.00	.04	SIMPSONITE $\text{Al}_4(\text{Ta}, \text{Nb})_3(\text{O}, \text{OH}, \text{F})_{14}$	TRIG	0001	White, yellow	H 7 G 6.6-7.3 infus	Fluor blue (pure Ta) to yellow (Nb present) in UV.
	2.041	1.926	.115	CUMENGITE $\text{Pb}_4\text{Cu}_4\text{Cl}_8(\text{OH})_8 \cdot \text{H}_2\text{O}$	TET	101 good 110 dist	Indigo-blue	H 2.5 G 4.67 F 1	Diss by warm HNO_3 .
	2.05	2.03	.02	BOLEITE $\text{Pb}_{26}\text{Ag}_{10}\text{Cu}_{24}\text{Cl}_{62}(\text{OH})_{48} \cdot 3\text{H}_2\text{O}$	TET ps cub	001 perf 101 good	Indigo-blue	H 3-3.5 G 5.05 F 1	Diss by warm HNO_3 . Pearly luster on clvs. Trillings on 001.
	2.05	---	very wk	EULYTITE $\text{Bi}_4\text{Si}_3\text{O}_{12}$	CUB tetrah	110 imperf	Gray, brown	H 4-4.5 G 6.62 F 2	Gel with acids.
	2.055	1.975	.080	GRIMALDIITE $\text{CrO}(\text{OH})$	TRIG tab	---	Deep red	G 4.12	Data for synth compd.
	2.057	2.046	.011	PYROMORPHITE (Apatite grp) $\text{Pb}_5(\text{PO}_4)_3\text{Cl}$	HEX pris	10Tl poor	Green, yellow, brown	H 3.5-4 G 7.0-7.1 F 1.5	Diss by HNO_3 . Luster resinous. Pleoc, 0 green, E greenish-yellow. Anom biax.
2.05 2.135 ^v	>2.05	---	---	KETTNERITE $\text{CaBiO}(\text{CO}_3)\text{F}$	TET plates	---	Brown to yellow	H "not too high" G 5.80	---
	2.06	2.05	.01	CUZTICITE $\text{Fe}^{+3}_2\text{Te}^{+6}\text{O}_6 \cdot 3\text{H}_2\text{O}$	HEX	---	Dark yellow	H 3 G 3.9	---

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ϵ							
2.033 ^	2.07	2.05	.02	BARYSILITE $\text{Pb}_8\text{Mn}(\text{Si}_2\text{O}_7)_3$	TRIG tab	0001 dist	Gray, white	H 3 G 6.72 F 2.5	Gel with acids.
	2.09	1.94	.15	HYDROCERUSSITE $\text{Pb}_3(\text{CO}_3)_2(\text{OH})_2$	TRIG plates	0001 perf	Cols	H 3.5 G 6.80 F 1.5	Diss by HNO_3 . Data for synth compd. El pos.
	2.125	2.059	.074	YEDLINITE $\text{Pb}_6\text{Cr}(\text{O},\text{OH})_8\text{Cl}_6$	TRIG	11 $\bar{2}$ 0 dist	Red-violet	H 2.5 G 5.85	Pleoc, 0 pale cobalt blue, E lavender.
2.11 □ 2.15	2.13	1.99	.14	BEYERITE $(\text{Ca},\text{Pb})\text{Bi}_2(\text{CO}_3)_2\text{O}_2$	TET tab	Conch	Yellow, green	H 2-3 G 6.56 F 1.5	Diss by acids.
2.057 ^ 2.18	2.135	2.12	.015	MIMETITE (Pyromorphite ser, Apatite grp) $\text{Pb}_5(\text{AsO}_4)_3\text{Cl}$	MCL ps hex	10 $\bar{1}$ 1 imperf	Yellow, brown, gray	H 3.5-4.5 G 7.1-7.24 F 1	Diss by HNO_3 . Commonly biax.
	2.145	2.006	.14	MATLOCKITE PbFCI	TET tab	001 perf	Yellow, greenish	H 2.5-3 G 7.12 F 1	Diss by warm HNO_3 . Anom biax. El clv pos.
	2.15	---	str	BISMOCLITE BiOCl	TET	001 perf	Yellow, brown	H 2-2.5 G 7.4-7.6 F 1	Diss by acids. El clv pos.
	2.155	1.975	.180	GRIMALDIITE $\text{CrO}(\text{OH})$	TRIG	---	Deep red	G 4.12 synth infus	---
	2.155	2.120	.035	PARAKHINITE $\text{PbCu}_3\text{TeO}_4(\text{OH})_6$	HEX	---	Dark green	H 3.5 G 6.7 calc	---
	>2.10	>2.10	mod to str	SCHUETTEITE $\text{Hg}_3(\text{SO}_4)_2\text{O}_2$	HEX	---	Yellow	H 3 G 8.2 volat	Diss by acids. Pleoc wk, 0 greenish-yellow, E orange yellow.
	2.18	---	---	RUSSELLITE Bi_2WO_6	TET mass	---	Yellow	H 3.5 G 7.37	Opt char unk.
	2.21	---	wk	ZAVARITSKITE BiOF	TET	---	Gray	G (9.2)	Opt sign unk.
2.43 v	2.25	2.20	.05	VANADINITE, arsenatian (Apatite grp) $\text{Pb}_5(\text{VO}_4, \text{AsO}_4)_3\text{Cl}$	HEX pris	---	Yellow	H 3 G 7.0 F 1.5	Diss by HNO_3 . El neg.

2.30 v	2.26	2.10	.16	HYDROHETAEROLITE Zn ₂ Mn ₄ O ₈ ·H ₂ O	TET fib	001 perf	Brownish-black	H 5-6 G 4.6 infus	Diss by HCl. Pleoc wk, 0 red-brown, E nearly opaque.
	2.27	2.18	.09	STOLZITE PbWO ₄	TET pyram	001 imperf	Green, gray, brown	H 2.5-3 G 7.9-8.4 F 2	Dec by HCl. Pleoc, 0 pale greenish-yellow, E bright yellow.
	2.275	2.15	.125	RANCIEITE (Ca,Mn ⁺²)Mn ⁺⁴ ₄ O ₉ ·3H ₂ O	HEX (?) mass	---	Dark brown, streak purplish	Soft G 3.3	Diss by HCl. Opt char unk.
	2.295	2.285	.01	FINNEMANITE Pb ₅ (AsO ₃) ₃ Cl	HEX pris	10Tl dist	Olive-green, gray, black	H 2-3 G 7.27 F 2	Diss by HNO ₃ . El neg.
2.35 v	2.30	2.07	.23	HETAEROLITE ZnMn ₂ O ₄	TET	001 indist	Black	H 6 G 5.2 infus	Diss by HCl. Red-brown in section. Pleoc wk, abs 0 > E.
2.27 ^ 2.35	2.30 Li	2.23 Li	.07	WULFENITE, tungstenian Pb(Mo,W)O ₄	TET	011 dist	Yellow	H 3 G 6.6 F 2	Dec by acids. MoO ₃ 21%, W/Mo = 0.7.
	2.30 Li	---	---	PLATTNERITE PbO ₂	TET	110	Black	H 5-5.5 G 8.5-9.5 F 2	Diss by HCl with evolution Cl ₂ . Nearly opaque. Basal section shows biax segments.
	2.32	2.12	.20	SANTANAITE 9PbO·2PbO ₂ ·CrO ₃	HEX	0001 perf 1210	Straw-yellow	H ~ 4	---
	2.32 Li	2.25 Li	.07	ECDEMITTE Pb ₆ As ₂ O ₇ Cl ₄	TET tab	001 dist	Yellow, green	H 2.5-3 G 7.14 F 1.5	Diss by HNO ₃ . In part biax.
2.31 □ 2.35	2.33	1.96	.37	GEIKIELITE (Mg,Fe)TiO ₃	TRIG rhombs	10Tl perf	Brownish-black	H 6 G 3.8-4.1 infus	Diff diss by HCl. Pleoc wk in red-brown, red, or purple.
2.30 ^	2.35	2.12	.23	HETAEROLITE ZnMn ₂ O ₄	TET	001 indist	Black	H 6 G 5.18 infus	Diss by HCl. Red-brown in section. Pleoc wk, abs 0 > E.
2.30 ^ 2.47	2.35	2.29	.06	WULFENITE, uranian Pb(Mo,U)O ₄	TET tab	011 dist	Yellow	H 3 G 6.6 F 2	Diss by HNO ₃ . UO ₃ 11.6%.
2.40 ┘	2.35 Li	2.33 Li	.02	"LORETTOITE" Pb ₇ O ₆ Cl ₂	TET (?) mass	001 perf	Orange-yellow	H 3 G 7.95 F 1	Diss by HNO ₃ . Probably synth material (?).
	2.36 Li	2.25 Li	.11	SCHWARTZEMBERGITE Pb ₅ (IO ₃)Cl ₃ O ₃	TET or ps tet	001 dist	Honey- to straw-yellow, reddish	H 2.5 G 7.39 F 1	Diss by HNO ₃ . Shows biax sectors.

Table 5. Uniaxial negative minerals (continued)

Other entries	Refractive index		Biref	MINERAL NAME and formula	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	ω	ε							
□	2.36 Li	2.31 Li	.05	LANGBANITE (Mn ⁺² , Sb, Ca) ₄ (Mn ⁺⁴ , Fe) ₃ SiO ₁₂	IRIG	---	Iron-black	H 6.5 G 4.6-4.8 infus	Diff diss by HCl. Pleoc wk in reddish-brown, abs 0 > E.
	2.41 Li	---	wk	MINIUM Pb ₃ O ₄	TET u mass	---	Brownish- red	H 2.5 G 9.1 F 2	Diss by acids. Abn green interf colors. Pleoc, deep red-brown to nearly cols.
2.25 ^	2.43	2.37	.06	VANADINITE (Apatite grp) Pb ₅ (VO ₄) ₃ Cl	HEX pris	---	Red, yellow, brown	H 3 G 6.65-6.98 F 1.5	Diss by HNO ₃ . El neg.
2.35 ^	2.45	2.33	.12	TAPIOLITE Fe(Ta, Nb) ₂ O ₆	TET	---	Black	H 6-7 G 7.11 infus	Pleoc str, dark red to black. U opt pos.
	2.46 Li 2.40	2.15 Li 2.15	.31	HAUSMANNITE Mn ⁺² Mn ⁺³ ₂ O ₄	TET oct	001 good	Brownish- black	H 5.5 G 4.84 infus	Diss by HCl. In section, brownish-red, not pleoc.
	2.47	2.26	.21	WULFENITE PbMoO ₄	TET tab	011 dist 001, 013 less so	Yellow, orange	H 3 G 6.7 F 2	Diss by HNO ₃ .
2.55 □ 2.60	2.48 Na 2.46	2.21 Na 2.07	.27 .39	PYROPHANITE MnTiO ₃	TRIG	0221 perf 1012 less so	Deep blood-red	H 5 G 4.54 infus	Nearly insol in HCl. In section yellow-red.
	2.50 Li	---	wk to mod	SENAITE Pb(Ti, Fe, Mn) ₂₁ O ₃₈	TRIG	---	Black	H 6 G 5.30	Nearly opaque. Not pleoc. Tw pl 1120 common.
	2.56	2.49	.07	ANATASE TiO ₂	TET oct	001, 011 perf	Brown to black	H 5.5-6 G 3.90 infus	Insol in acids. Pleoc in thick sections, 0 pale blue to yellow, E dark blue to orange.
	2.6 Li	---	extr	TRECHMANNITE AgAsS ₂	TRIG	1011 good 0001 dist	Scarlet- vermilion	H 1.5-2 F 1	Streak scarlet. Pleoc, 0 pale red, E cols.
	2.60	2.50	.10	BRAUNITE 3Mn ₂ O ₃ •MnSiO ₃	TET pyram	112 perf	Brownish- black	H 6-6.5 G 4.72-4.83 infus	Dec by HCl.
	2.665Li	2.535Li	.130	LITHARGE PbO	TET tab	110 perf	Red	H 2 G 9.13 F 1.5	Diss by HNO ₃ . Fuses to a yellow bead. El clv neg.

2.70	1.75	.95	CHALCOPHANITE (Zn,Mn)Mn ₃ O ₇ ·3H ₂ O	TCL	001 perf	Brownish-black	H 2.5 G 4.0 infus	Diss by HCl. Pleoc str, 0 nearly opaque, E deep red.
2.98 Li 3.09 Na	2.71 Li 2.79 Na	.27 .30	PROUSTITE Ag ₃ AsS ₃	TRIG	10T1 dist	Scarlet	H 2-2.5 G 5.57 F 1	Dec by HNO ₃ . Streak scarlet. Bright red in section, pleoc wk, 0 blood-red, E cochineal-red.
3.01 Li 3.22 Na	2.78 Li 2.94 Na	.23 .28	HEMATITE Fe ₂ O ₃	TRIG	Parting 0001	Red to black	H 5-6 G 5.26 infus	Diss by HCl. Streak red. Abs 0 > E.
3.08 Li	2.88 Li	.20	PYRRARGYRITE Ag ₃ SbS ₃	TRIG	10T1 dist	Dark red	H 2.5 G 5.85 F 1	Dec by HNO ₃ . Streak purplish-red. Red in section. Tw pl 1120.
3.25	2.40	.85	CREDNERITE CuMnO ₂	MCL ps hex	001, 100 perf T11 good	Iron-black	H 4 G 5.0 F diff	Diss by HCl. Poly tw.
3.30	2.90	.40	HOLLANDITE Ba(Mn ⁺⁴ , Mn ⁺²) ₈ O ₁₆	MCL ps tet	---	Black, gray	H 6 G 2.95 infus	Diss by HCl.
3.78	3.75	.03	GRATONITE Pb ₉ As ₄ S ₁₅	TRIG	Conch	Lead-gray	H 2.5 G 6.22 F 1	---
4.34 Li	2.03 Li	2.31	MOLYBDENITE MoS ₂	HEX	0001 perf	Lead-gray	H 1-1.5 G 4.7 infus	Dec by HNO ₃ . El clv pos.

Table 6. Biaxial positive minerals

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.301	<u>1.301</u>	1.307	.006	FERRUCCITE NaBF_4	11°	$X = \frac{b}{c}$ $Z = \frac{c}{a}$	ORTH tab	100, 010 001	Cols	H 3 G 2.50 F 1	Sl sol in H_2O , diss by acids.
	1.338	<u>1.338</u>	1.339	.001	CRYOLITE Na_3AlF_6	43° $r < v$	$X = \frac{b}{c}$ $Z:c = -44^\circ$	MCL	Parting 110, 001	White, reddish, brownish	H 2.5 G 2.97 F 2	Insol in H_2O , tw 110 lam.
	1.346	<u>1.348</u>	1.350	.004	WEBERITE $\text{Na}_2\text{MgAlF}_7$	80°	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	101 poor 010	Gray	H 3.5 G 2.96	---
	1.409	<u>1.412</u>	1.419	.010	PACHNOLITE $\text{NaCaAlF}_6 \cdot \text{H}_2\text{O}$	76° $r < v$ wk	$X = \frac{b}{c}$ $Z:c = 53-69^\circ$ disp str	MCL	001 fair	White	H 3 G 2.98, 2.97 calc	Tw pl 100.
	1.411	<u>1.416</u>	1.422	.011	CARLHINTZEITE $\text{Ca}_2\text{AlF}_7 \cdot \text{H}_2\text{O}$	77°	---	TCL pris	---	White, cols	G 2.36	Tw [010].
	1.413	(<u>1.417</u>)	1.423	.010	YAROSLAVITE $\text{Ca}_3\text{Al}_2\text{F}_{10}(\text{OH})_2 \cdot \text{H}_2\text{O}$	74°	---	ORTH fib	Pina- coidal	White	G 3.09	---
	1.425	<u>1.428</u>	1.438	.013	CALCJARLITE $(\text{Na}, \text{K})(\text{Ca}, \text{Sr})_3\text{Al}_3(\text{F}, \text{OH})_{16}$	72°	$Z:c = 10-15^\circ$	MCL	Uneven	Cols	H 4 G 3.51	---
	1.429	<u>1.433</u>	1.435	.006	JARLITE $\text{NaSr}_3\text{Al}_3(\text{F}, \text{OH})_{16}$	$\sim 90^\circ$	$Y = \frac{b}{c}$ $X:c = -6^\circ$	MCL tab	---	Cols	H 4-4.5 G 3.78-3.93 F easy	Formula in doubt.
	1.422	<u>1.435</u>	1.480	.058	SANTITE $\text{KB}_5\text{O}_6(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	70° ($58 \pm 5^\circ$)	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH	010 perf 100 dist	Cols	H 2.5 G 1.74	Diss by acids. Tw 01T.
	1.431	<u>1.438</u>	1.488	.057	SBORGITE $\text{NaB}_5\text{O}_6(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	35° ($42 \pm 6^\circ$) $r > v$ wk	---	MCL	---	Cols	H 3.5 G 1.71	Sol in H_2O .
	1.439	<u>1.442</u>	1.469	.030	STERCORITE $\text{HNa}(\text{NH}_4)\text{PO}_4 \cdot 4\text{H}_2\text{O}$	36° $r > v$ rather str	$Z = \frac{b}{c}$ $Y:c = 30^\circ$ disp str	TCL	---	Cols	H 2 G 1.6 F 1	Sol in H_2O . Section 010 ² shows 2 sets of tw lam at about 90° .
	1.441	<u>1.442</u>	1.444	.003	USOVITE $\text{Ba}_2\text{CaMgAl}_2\text{F}_{14}$	70°	---	MCL	One, perf	Brown	H 3.5 F easy	Pleoc, X and Y brownish-yellow, Z pale yellow.

1.478

1.466

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1.484

1.445	<u>1.460</u>	1.491	.046	MERCALLITE KHSO_4	56° (71+5°) $r < v$ wk	$X = \frac{b}{Z} = \frac{a}{a}$	ORTH	---	Cols, sky-blue	G 2.31 F 1	Sol in H_2O , giving an acid solution.
1.456	<u>1.460</u>	1.480	.024	CHESSEXITE $\text{Na}_4\text{Ca}_2(\text{Mg}, \text{Zn})_3\text{Al}_8$ $(\text{SO}_4)_{10}(\text{SiO}_4)_2(\text{OH})_{10} \cdot 40\text{H}_2\text{O}$	47°	$X = \frac{a}{Z} = \frac{b}{b}$	ORTH	010	White	G 2.04	Diss by HCl.
1.460	<u>1.461</u>	1.470	.010	ALUNOGEN $\text{Al}_2(\text{SO}_4)_3 \cdot 17\text{H}_2\text{O}$	31°	$X \sim \frac{b}{Z:c} = 42^\circ$	TCL tab or fib <u>c</u>	010 perf 100, 3I3	White	H 1.5-2 G 1.77 infus	Sol in H_2O . In closed tube, melts in its water of crystn.
1.454	<u>1.461</u>	1.471	.017	DORFMANITE $\text{Na}_2\text{HPO}_4 \cdot 2\text{H}_2\text{O}$	65° (80+14°) $r > v$ wk	---	ORTH	---	White	H 1-1.5 G 2.00	Sol in H_2O , gives an alk reaction.
---	<u>1.461</u>	---	low	SILHYDRITE $3\text{SiO}_2 \cdot \text{H}_2\text{O}$	---	---	ORTH	Uneven to subconch	White	G 2.14	Dec by HCl. Opt sign unk.
1.461	<u>1.463</u>	1.476	.015	PICROMERITE $\text{K}_2\text{Mg}(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$	48° $r > v$	$Y = \frac{b}{X:a} = -1^\circ$	MCL	20I perf	White	H 2.5 G 2.10 F 2	Sol in H_2O .
1.459	<u>1.464</u>	1.470	.011	ALUMINITE $\text{Al}_2\text{SO}_4(\text{OH})_4 \cdot 7\text{H}_2\text{O}$	Med to large (85+21°)	$X = e1$	MCL fib	---	White	H 1-2 G 1.66-1.8 infus	Diss by acids.
1.462	<u>1.465</u>	1.474	.012	MALLARDITE (Melanterite grp) $\text{MnSO}_4 \cdot 7\text{H}_2\text{O}$	Large	$Z:c = 44^\circ$	MCL	001 dist	Rose	G 1.84	Sol in H_2O .
1.462	<u>1.466</u>	1.469	.007	BOGGILDITE $\text{Na}_2\text{Sr}_2\text{Al}_2(\text{PO}_4)_9$	79°	---	MCL	---	Flesh-red	H 4-5 G 3.66 F 1-2	---
1.464	<u>1.468</u>	1.474	.010	KONYAITE $\text{Na}_2\text{Mg}(\text{SO}_4)_2 \cdot 5\text{H}_2\text{O}$	106° (79+24°)	---	MCL	---	Cols	H 2.5 G 2.097	Sol in H_2O . Stated to be biāx neg.
1.456	<u>1.469</u>	1.508	.052	LANSFORDITE $\text{MgCO}_3 \cdot 5\text{H}_2\text{O}$	60°	$X = \frac{b}{Z} \sim \frac{c}{c}$	MCL tab	001 perf 100 less so	Cols	H 2.5 G 1.71 infus	Diss by acids. Alters in air to Nesquehonite.
1.468	<u>1.470</u>	1.473	.005	ERIONITE (Zeolite grp) $(\text{K}_2, \text{Ca}, \text{Na}_2)_2\text{Al}_4\text{Si}_{14}$ $\text{O}_{36} \cdot 15\text{H}_2\text{O}$	---	e1 pos	HEX fib	---	---	G 2.05 F easy	Diff dec by HCl with sepn of silica.
1.468	<u>1.470</u>	1.474	.006	TRIDYMITE SiO_2	40°	$X = \frac{b}{Z} = \frac{c}{c}$ e1 neg	MCL ps hex, tab	---	Cols	H 6.5 G 2.25 infus	Insol in acids, diss by hot Na_2CO_3 soln. Tw, trillings, fourlings.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.478 ^v	1.467	1.471	1.476	.009	MELANTERITE (Fe,Mg)SO ₄ ·7H ₂ O	Large r > v	Y = $\frac{b}{c}$ Z:c = 77°	MCL el c	T01 perf 111 poor	Greenish	H 2 G 1.76 F easy	Sol in H ₂ O. MgO 7.45, CuO 0.3, ZnO 0.5%.
1.491(?) ^v	1.470	1.471	1.479	.009	BOUSSINGAULTITE (NH ₄) ₂ Mg(SO ₄) ₂ ·6H ₂ O	51° r > v perc	Y = $\frac{b}{c}$ Z:c = 12°	MCL	20T perf	Cols	H 2 G 1.72 F 1	Sol in H ₂ O. Continuous series to Mohrite?
	1.469	1.473	1.491	.022	META-ALUNOGEN Al ₄ (SO ₄) ₆ ·27H ₂ O	Large (51±12°)	---	---	---	White	---	---
	1.458	1.473	1.501	.043	BARRINGTONITE MgCO ₃ ·2H ₂ O(?)	(74±6°)	el pos	TCL fib	---	Cols	G 2.82 (calc)	Optics are discordant for compn given.
	1.466	1.474	1.495	.029	CARNALLITE KMgCl ₃ ·6H ₂ O	70° r < v	Z = $\frac{a}{c}$ X = $\frac{c}{b}$	ORTH	Conch	White, reddish	H 2.5 G 1.62 F 1-1.5	Sol in H ₂ O. Deliq.
1.473 □ 1.477	1.469	1.475	1.484	.015	THENARDITE Na ₂ SO ₄	83° r > v wk	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH	010 perf 101 fair	White	H 2.5-3 G 2.66 F 2	Sol in H ₂ O.
	---	1.47- 1.48	---	wk	HYDROBASALUMINITE Al ₄ SO ₄ (OH) ₁₀ ·12-36H ₂ O	---	---	---	---	Cols to yellowish	G 1.86	Diss by acids. Opt char unk.
	1.472	---	1.487	.015	MAKATITE Na ₂ Si ₄ O ₉ ·5H ₂ O	---	---	ORTH or MCL spher	---	White	G 2.07	Dec by acids. Opt char unk.
1.482 ^v	1.472	1.475	1.477	.005	MORDENITE (Zeolite grp) (Ca,K ₂ ,Na ₂)Al ₂ Si ₁₀ O ₂₄ ·7H ₂ O	80°	X = $\frac{c}{b}$ Y = $\frac{a}{b}$	ORTH	100 good 010 fair	Cols, pink, yellow	H 3-4 G 2.12 F 4-5	Partly dec by HCl.
1.471 ◇ 1.483	1.471	1.478	1.486	.015	MELANTERITE FeSO ₄ ·7H ₂ O	85° r > v wk	Y = $\frac{b}{c}$ Z:c = 77°	MCL el c	T01 perf 111 poor	Green	H 2 G 1.90 F easy	Sol in H ₂ O.
1.461 [^]	1.475	1.478	1.482	.010	ALUNOGEN Al ₂ (SO ₄) ₃ ·17H ₂ O	75°	X ~ $\frac{b}{c}$ Z:c = 42°	TCL tab or fib c	010 perf 100, 3T3	White	H 1.5-2 G 1.79 infus	Sol in H ₂ O. In closed tube, melts in its water of crystn.
1.489 ^v	1.478	1.479	1.482	.004	FERRIERITE (Zeolite grp) (Na,K) ₂ MgAl ₃ Si ₁₅ O ₃₆ (OH)·9H ₂ O	50°	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el pos	ORTH blades 100	100 perf	White	H 3 G 2.15 F 3-4	Insol in HCl.

1.493	1.478	1.479	1.481	.003	CLINOPTILOLITE (Zeolite grp) $(\text{Na}, \text{K}, \text{Ca})_{2-3}\text{Al}_3(\text{Al}, \text{Si})_2\text{Si}_{13}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	40° r > v	$X = \frac{b}{c}$ el cTv pos	MCL	010 perf	Cols	H 4 G 2.11 F 2	Dec by HCl. SiO_2 68.3, CaO 1.0, Na_2O 2.5%.
	1.47	1.48	1.49	.02	BOOTHITE $\text{CuSO}_4 \cdot 7\text{H}_2\text{O}$	Large	$Y = \frac{b}{c}$ $X \sim \frac{c}{c}$	MCL fib c	001 imperf	Blue	H 2-2.5 G 2.1 F 3	Sol in H_2O . Dehydrates readily. Cols in thin section.
	1.475	1.480	1.488	.013	DIETRICHITE $(\text{Zn}, \text{Fe}, \text{Mn})\text{Al}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Large	$X = \frac{b}{c}$ $Z:c = 29^\circ$	MCL fib c		White	H 2 infus	Sol in H_2O .
1.486	1.477	1.480	1.490	.013	NATROLITE (Zeolite grp) $\text{Na}_2\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 2\text{H}_2\text{O}$	63° r < v wk	$X = \frac{a}{c}$ $Z = \frac{c}{c}$ el pos	ORTH acic c	110 perf 010 imperf	White	H 5 G 2.22 F 2	Gel with acids. Tw pl 110, 100, rarely 301.
	1.475	1.480	1.490	.015	MAKATITE $\text{Na}_2\text{Si}_4\text{O}_9 \cdot 5\text{H}_2\text{O}$	(71+16°)		ORTH spher		White	H 4.5 G 2.07	---
	1.475	1.480	1.487	.012	MISENITE $\text{K}_2\text{SO}_4 \cdot 6\text{KHSO}_4 (?)$	Large	$Z = \frac{b}{c}$ $X:c = 29^\circ$	MCL fib c	010 dist	White	G 2.32 F 1.5	Sol in H_2O to an acid soln.
	---	1.480	---	wk	FAUJASITE (Zeolite grp) $(\text{Na}_2, \text{Ca})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 8\text{H}_2\text{O}$	~ 0°	---	CUB oct	111 dist	White	H 5 G 1.92 F 3	Dec by HCl. Shows 8 biax segments.
1.485	1.480	(1.480)	1.485	.005	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na}_2)\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	Small	---	TRIG rhombs	10T1 dist	White, red	H 4.5 G 2.06 F 3	Dec by HCl with sepn of slimy silica.
1.496	1.480	1.481	1.485	.005	DACHIARDITE (Zeolite grp) $(\text{K}_2, \text{Na}_2, \text{Ca})_5\text{Al}_{10}\text{Si}_{38}\text{O}_{96} \cdot 25\text{H}_2\text{O}$	55-60°	$X = \frac{b}{c}$ $Z:c = 35^\circ$	MCL	100, 001 perf	White	H 4-4.5 G 2.16 fus	Dec by acids. Mimetic tw perf.
	1.470	1.481	1.497	.027	RIVADAVITE $\text{Na}_6\text{MgB}_{24}\text{O}_{40} \cdot 22\text{H}_2\text{O}$	80° r > v	$Y = \frac{b}{a}$ $Z:a = -32^\circ$	MCL el b	100, T01 perf 010 poor	Cols, white	H 3.5 G 1.91 F 1.5	Diss by hot H_2O .
	1.470	---	1.493	.023	NICKEL-HEXAHYDRITE $(\text{Ni}, \text{Mg}, \text{Fe})\text{SO}_4 \cdot 6\text{H}_2\text{O}$	---	$Z:c = 45^\circ$	MCL crusts, fib	010 perf 100	Blue- green	---	Diss by H_2O . NiO 22.6, MgO 2.8, FeO 2.6%. Opt char unk.
	---	1.482	---	.002- .008	GMELINITE (Zeolite grp) $(\text{Na}_2, \text{Ca})\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	~ 0°		HEX	10T0 dist	White	H 4.5 G 2.0-2.1 F 3	Dec by HCl. Tw axis c.
1.475 ^	1.481	1.482	1.486	.005	MORDENITE var. Ashtonite (Zeolite grp) $(\text{Ca}, \text{Na}_2)\text{Al}_2\text{Si}_{10}\text{O}_{24} \cdot 7\text{H}_2\text{O}$	Large	$X = \frac{c}{a}$ $Y = \frac{a}{a}$	ORTH radi- ating	100 good 010 fair	White to red	H 3-4 G 2.12 F 4-5	Partly dec by HCl. SrO 1.3%.
	1.462	(1.483)	1.531	.069	BAYLISSITE $\text{K}_2\text{Mg}(\text{CO}_3)_2 \cdot 4\text{H}_2\text{O}$	64°	---	MCL	Conch	Cols	H 2-3 G 2.01	Diss by acids with eff. Dec by H_2O .

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.478 ^	1.479	1.483	1.488	.009	ZINC-COPPER MELANTERITE (Zn,Cu)SO ₄ ·7H ₂ O	Large disp wk	---	MCL fib	---	Light blue-green	H 2 G 2.02 F easy	Sol in H ₂ O. Dehydrates in air.
	1.480	1.483	1.486	.006	LEONITE K ₂ Mg(SO ₄) ₂ ·4H ₂ O	~ 90°	Y = $\frac{b}{a}$ Z: $\frac{a}{c}$ small	MCL pris	Conch	Cols, yellow	H 2.5-3 G 2.20 F easy	Sol in H ₂ O. Poly tw.
	1.468	1.484	1.515	.083	CALCLACITE CaCl ₂ ·Ca(C ₂ H ₃ O ₂) ₂ ·10H ₂ O	80° (72±5°)	---	MCL or TCL	---	Cols	G 1.5	Calcium chloride-acetate (an artifact?). Sol in H ₂ O. The synth salt has <u>ns</u> 1.468, 1.502, 1.523, 2V -58°, (81±1°).
1.470 ^	1.482	1.484	1.486	.004	TRIDYMITITE SiO ₂	86°	X = $\frac{b}{c}$ Z = $\frac{c}{a}$	MCL ps hex, tab	---	Cols	H 6.5 G 2.25 infus	Insol in acids. Diss by hot Na ₂ CO ₃ soln. Tw trillings, fourlings.
1.503 v	1.483	1.485	1.487	.004	PHILLIPSITE (Zeolite grp) (K ₂ ,Ca,Na ₂) ₁₋₂ (Al,Si) ₈ O ₁₆ ·6H ₂ O	63° r < v mod	X = $\frac{b}{c}$ Z:c = 19° el pos	MCL fib a	100, 010 dist	White	H 4 G 2.19 F 3	Gel with acids. Tw pl 001 and 011 penet giving ps orth forms.
(1.480) ^	1.485	1.485	1.488	.003	CHABAZITE (Zeolite grp) (Ca,Na ₂) ₂ Al ₂ Si ₄ O ₁₂ ·6H ₂ O	Med to large	---	TRIG	10I dist	Cols, white, red	H 4.5-5 G 2.1 F 3	Dec by HCl. Tw pl 10I1.
1.492 ^	---	1.485	---	low	MAGADIITE NaSi ₇ O ₁₃ (OH) ₃ ·3H ₂ O	---	el pos	MCL	---	White	---	Dec by HCl. Opt sign unk.
	1.472	---	1.502	.030	INDIGIRITE Mg ₂ Al ₂ (CO ₃) ₄ (OH) ₂ ·15H ₂ O	---	el pos	fib	---	Snow-white	H 2 G 1.6	Diss by acids. Opt char unk.
	1.484	1.486	1.502	.018	CYANOCHROITE K ₂ Cu(SO ₄) ₂ ·6H ₂ O	47° r < v str	Y = $\frac{b}{c}$ X:c = 19°	MCL crusts	20I perf	Greenish-blue	G 2.22 F 2	Sol in H ₂ O.
1.480 ^	1.483	1.486	1.495	.012	NATROLITE (Zeolite grp) Na ₂ Al ₂ Si ₃ O ₁₀ ·2H ₂ O	Med (60±20°) r < v wk	X = $\frac{a}{c}$ Z = $\frac{c}{b}$ el pos	ORTH acic c	110 perf 010 poor	White	H 5 G 2.22 F 2	Gel with acids. Tw pl 110, 100, rarely 301.
	1.470	1.487	1.540	.070	AMMONIOBORITE (NH ₄) ₂ B ₁₀ O ₁₆ ·5H ₂ O	59° r < v wk	Y = $\frac{b}{c}$ Z:c = 7°	MCL	---	White	Soft G 1.77 volat	Slowly sol in cold H ₂ O.

	1.476	---	1.500	.024	UKLONSKOVITE $\text{Na}_2\text{Mg}_2(\text{SO}_4)_2(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	---	---	MCL	T01, T11 good	Cols	G 2.42-2.5	Diss by acids.
	1.484	<u>1.487</u>	1.496	.012	TAMARUGITE $\text{NaAl}(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$	60° disp wk	$Y = \frac{b}{c}$ $X:c = 5^\circ$	MCL fib <u>c</u>	010 perf	Cols	H 3 G 2.07 fus	Sol in H_2O . Poly tw.
v 1.498	1.488	<u>1.488</u>	1.489	.001	HEULANDITE (Zeolite grp) $(\text{Ca}, \text{Na}_2)_2\text{Al}_3(\text{Al}, \text{Si})_2\text{Si}_{13}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	70° r > v	$Z = \frac{b}{c}$ el cTv neg	MCL	010 perf	White	H 4 G 2.14 F 2	Dec by HCl. SiO_2 61.8, CaO 5.5, Na_2O 1.1, K_2O 0.3%.
1.479 ^	1.489	<u>1.489</u>	1.49	.001	FERRIERITE (Zeolite grp) $(\text{Na}, \text{K})_2\text{MgAl}_3\text{Si}_{15}\text{O}_{36}(\text{OH}) \cdot 9\text{H}_2\text{O}$	Small	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el pos	ORTH blades 100	100 perf	White	H 3 G 2.21 F 3-4	Insol in HCl.
1.489 □ 1.496	1.480	<u>1.491</u>	1.509	.029	FLUELLITE $\text{Al}_2(\text{PO}_4)\text{F}_2(\text{OH}) \cdot 7\text{H}_2\text{O}$	85° r < v	$X = \frac{b}{c}$ $Y = \frac{c}{b}$	ORTH	010, 111 indist	White	H 3 G 2.16 infus	Insol in acids.
	1.488	<u>1.491</u>	1.505	.017	INDERITE $\text{MgB}_3\text{O}_3(\text{OH})_5 \cdot 5\text{H}_2\text{O}$	37°	$Z:c = 9^\circ$	MCL	110 good	Cols	H 2.5 G 1.79 F 1	Diss by acids.
1.471(?) ^	1.487	<u>1.491</u>	1.499	.012	MOHRITE $(\text{NH}_4)_2\text{Fe}(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$	~ 75°	$Y = \frac{b}{c}$	MCL	20I good 010 dist	Pale bluish- green	G 1.86	Sol in H_2O . Con- tinuous series to Boussingaultite?
1.485 ^	1.491	<u>1.492</u>	1.494	.003	CHABAZITE (Zeolite grp) $(\text{Ca}, \text{Na}_2)\text{Al}_2\text{Si}_4\text{O}_{12} \cdot 6\text{H}_2\text{O}$	58°	---	TRIG	10I1 dist	Cols, red, white	H 4-5 G 2.1 F 3	Dec by HCl. Tw pl 10I1.
1.479 ^	1.491	<u>1.493</u>	1.497	.006	CLINOPTILOLITE (Zeolite grp) $(\text{Na}_2, \text{Ca}, \text{K}_2)_2\text{Al}_3(\text{Al}, \text{Si})_2\text{Si}_{13}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	72° r > v	$X = \frac{b}{c}$ el cTv pos	MCL	010 perf	Cols	H 4 G 2.1 F 2	Dec by HCl. SiO_2 59.7, CaO 3.4, Na_2O 1.3, K_2O 4.2%.
	1.485	<u>1.494</u>	1.505	.020	CANAVESITE $\text{Mg}_2(\text{CO}_3)(\text{HBO}_3) \cdot 5\text{H}_2\text{O}$	Very large disp wk	$Z = \frac{b}{c}$	MCL fib	---	Milky white	G 1.8	---
	1.494	<u>1.495</u>	1.497	.003	ARCANITE K_2SO_4	67° r > v mod	$X = \frac{b}{c}$ $Z = \frac{c}{b}$	ORTH	010, 001 good	White	G 2.66 F 3	Sol in H_2O . Tw common.
	1.495	<u>1.496</u>	1.504	.009	STRUVITE $(\text{NH}_4)\text{MgPO}_4 \cdot 6\text{H}_2\text{O}$	37° r < v str	$X = \frac{b}{c}$ $Z = \frac{a}{b}$	ORTH	001 good 100 poor	Cols to yellow	H 2 G 1.71	Diss by acids.
1.481 ^	1.493	<u>1.496</u>	1.500	.007	DACHIARDITE (Zeolite grp) $(\text{K}_2, \text{Na}_2, \text{Ca})_5\text{Al}_{10}\text{Si}_{38}\text{O}_{96} \cdot 25\text{H}_2\text{O}$	69°	$X = \frac{b}{c}$ $Z:c = 35^\circ$	MCL	100, 001 perf	White	H 4-4.5 G 2.17 fus	Dec by acids. Mimetic tw 001.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.504	1.495	1.497	1.504	.009	YUGAWARALITE (Zeolite grp) $\text{CaAl}_2\text{Si}_6\text{O}_{16} \cdot 4\text{H}_2\text{O}$	78° $r > v$ wk	$Z = b$ $Y:c = 7^\circ$	MCL	Parting 010	Cols	H 4.5 G 2.23	Insol in acids. U opt neg.
	1.484	1.498	1.523	.039	ARISTARAINITE $\text{Na}_2\text{MgB}_{12}\text{O}_{20} \cdot 8\text{H}_2\text{O}$	70° $r > v$ wk	$X = b$ $Y:c = -38^\circ$	MCL	001, 100 perf 110	Cols	H 3.5 G 2.03 fus	Diss by HCl.
1.510	1.497	1.498	1.499	.002	GONNARDITE (Zeolite grp) $\text{Na}_2\text{CaAl}_4\text{Si}_6\text{O}_{20} \cdot 7\text{H}_2\text{O}$	Med to large	---	ORTH fib	---	White	G 2.26	---
1.488 1.507	1.496	1.498	1.504	.008	HEULANDITE (Zeolite grp) $(\text{Ca}, \text{K}_2)_{2-3}\text{Al}_3(\text{Al}, \text{Si})_2\text{Si}_{13}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	35°	$Z = b$ el cTv neg	MCL	010 perf	White	H 4 G 2.20 F 2	Dec by HCl. SiO_2 57.4, CaO 7.0, SrO 1.55, Na_2O 0.1, K_2O 1.4%.
	1.499	1.500	1.502	.003	WAIRAKITE (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 2\text{H}_2\text{O}$	75° $r > v$ wk	$X = b$ $Z \sim c$	MCL	---	White	H 5-5.6 G 2.26 F 2.5	Gel with acids.
	1.484	1.501	1.550	.066	GOWERITE $\text{CaB}_6\text{O}_{10} \cdot 5\text{H}_2\text{O}$	(63+4°) $r > v$ wk	$Y = b$ $Z:c = 27^\circ$	MCL	001 perf	White	H 3 G 2.00 fus	Diss by acids.
1.498 1.503	1.500	1.501	1.504	.004	WELLSITE (Zeolite grp) $(\text{K}_2, \text{Ca}, \text{Ba})\text{Al}_2\text{Si}_6\text{O}_{16} \cdot 6\text{H}_2\text{O}$	39°	$Z = b$ $X:c = -52^\circ$	MCL	---	Cols	H 4-5 G 2.3 F 3	Dec by acids. Complex tw. Perhaps a barian Phillipsite.
	1.497	1.502	1.539	.042	LIEBIGITE $\text{Ca}_2(\text{UO}_2)(\text{CO}_3)_3 \cdot 10\text{H}_2\text{O}$	40° $r > v$ med	$X = a$ $Z = c$	ORTH crusts	100	Green to yellow-green	H 2.5-3 G 2.41 infus	Diss by acids. Pleoc, X nearly cols, Y and Z pale yellow-green. Fluor green in UV.
	1.498	1.503	1.510	.012	HEXAHYDROBORITE $\text{Ca}[\text{B}(\text{OH})_4]_2 \cdot 2\text{H}_2\text{O}$	83° $r > v$ very str	$X:c = 14^\circ$	MCL pris	---	Cols, white	H 2.5 G 1.87	Diss by acids.
1.485	1.500	1.503	1.510	.010	PHILLIPSITE (Zeolite grp) $(\text{K}_2, \text{Ca}, \text{Na}_2)_{1-2}(\text{Al}, \text{Si})_8\text{O}_{16} \cdot 6\text{H}_2\text{O}$	Med $r < v$ mod	$X = b$ $Z:c = 19^\circ$ el pos	MCL fib a	100, 010 dist	White	H 4 G 2.15 F 3	Gel with acids. Tw pl 001 and 011, penet, giving orth forms.

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1.501	<u>1.503</u>	1.510	.009	PROSOPITE $\text{CaAl}_2(\text{F},\text{OH})_8$	63° r > v str	$Y = \frac{b}{c} = -35^\circ$	MCL tab 010	111 perf	Cols	H 4.5 G 2.38 infus	---
1.503	(1.504)	1.535	.032	SVEITE $\text{KAl}_7(\text{NO}_3)_4\text{Cl}_2(\text{OH})_{16} \cdot 8\text{H}_2\text{O}$	Small	---	MCL	001 perf	White	Very soft G 2.0	Dec by H_2O . Sol in acids.
1.504	<u>1.504</u>	1.553	.049	EVENKITE (n-tetracosene) $\text{C}_{21}\text{H}_{44}$	Small	---	MCL	One good	Cols	H 1 G 0.92	M.p. 50°C. Hydro- carbon. Poly tw.
1.501	<u>1.504</u>	1.520	.019	HYDROCHLORBORITE $\text{Ca}_2\text{B}_4\text{O}_4\text{Cl}(\text{OH})_7 \cdot 7\text{H}_2\text{O}$	46° r < v	$Y = \frac{b}{c} = 25^\circ$	MCL mass	001 good conch	Cols	H 2.5 G 1.85	Diss by hot H_2O .
1.501	<u>1.504</u>	1.509	.008	HARMOTOME (Zeolite grp) $(\text{Ba},\text{K}_2)_{1-2}(\text{Al},\text{Si})_8\text{O}_{16} \cdot 6\text{H}_2\text{O}$	78° disp crossed	$Z = \frac{b}{a} = 65^\circ$	MCL pris	010 good 001 less so	White	H 4.5 G 2.38 F 3.5	Dec by HCl. Cruciform tw, tw pl 001.
1.493	<u>1.505</u>	1.529	.036	ULEXITE $\text{NaCaB}_5\text{O}_9 \cdot 8\text{H}_2\text{O}$	70-73° (84+9°)	$X = \frac{b}{c} \sim 20^\circ$	TRIC fib	010 perf 120 good	White	H 2.5 G 1.96 F 1	Diss by acids. Partly dec by hot H_2O .
1.500	<u>1.505</u>	1.525	.025	LOUGHLINITE $\text{Na}_2\text{Mg}_3\text{Si}_6\text{O}_{16} \cdot 8\text{H}_2\text{O}$	60°	---	ORTH fib	---	White	G 2.165	Dec by HCl, slowly dec by H_2O .
1.502	<u>1.505</u>	1.514	.012	RHODESITE $(\text{Ca},\text{Na}_2,\text{K}_2)_8\text{Si}_{16}\text{O}_{40} \cdot 11\text{H}_2\text{O}$	Small (60+20°)	$X = \frac{b}{a}$	ORTH fib	100 good	White	G 2.36 fus	---
1.504	<u>1.505</u>	1.507	.003	MESOLITE (Zeolite grp) $\text{Na}_2\text{Ca}_2\text{Al}_6\text{Si}_9\text{O}_{30} \cdot 8\text{H}_2\text{O}$	Large r > v str	$Y = \frac{b}{c}$ $X \sim \frac{c}{a}$	MCL el = $\frac{b}{c}$	101, 101 perf	White, gray	H 2.5 G 2.26 F easy	Gel with acids. Tw pl 100, 2V may vary with temp.
1.494	<u>1.507</u>	1.528	.034	BISCHOFITE $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$	79° r > v wk	$X = \frac{b}{c}$ $Y : \frac{c}{a} = 9.5^\circ$	MCL fib	---	Cols	H 1.5 G 1.60 fus	Sol in H_2O . Deliq.
1.504	<u>1.507</u>	1.515	.011	HEULANDITE (Zeolite grp) $(\text{Ca},\text{Na},\text{Sr})_{2-3}\text{Al}_3$ $(\text{Al},\text{Si})_2\text{Si}_{13}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	10-40° (63+22°) r > v	$Z = \frac{b}{c}$ el cTv neg	MCL	010 perf	White	H 4 G 2.20 F 2	Dec by HCl. SiO_2 55.8, CaO 3.0, SrO 4.4, BaO 2.4, Na_2O 1.5, K_2O 1.3%.
1.504	<u>1.508</u>	1.543	.039	USSINGITE $\text{Na}_2\text{AlSi}_3\text{O}_8(\text{OH})$	32-39°	$Z : \frac{b}{c} = 3^\circ$	TCL	001 perf 110 fair	Rose- violet	H 6 G 2.48 F easy	Gel with acids.
1.502	<u>1.508</u>	1.525	.023	RABBITTITE $\text{Ca}_3\text{Mg}_3(\text{UO}_2)_2(\text{CO}_3)_6$ $(\text{OH})_4 \cdot 18\text{H}_2\text{O}$	(62+11°)	$Y = \frac{b}{c}$ $Z : \frac{c}{a} = 15^\circ$	MCL fib	001 and 2 pris	Pale green	H 2.5 G 2.6	Diss by acids, slowly dec by H_2O . Weakly fluor in UV.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.504 [^]	---	<u>1.508</u>	---	.001	LEUCITE KAlSi ₃ O ₆	Small	Z = $\frac{a}{b}$	TET ps cub	110 poor	Cols	H 6 G 2.5 infus	Dec by acids. Poly tw.
	1.493	<u>1.509</u>	1.561	.068	LARDERELLITE (NH ₄)B ₅ O ₆ (OH) ₄	60° r < v	X = $\frac{b}{c}$ Z:c = 30°	MCL tab	001, 010 perf	Cols	Soft G 1.91 volat	Diss by acids. Rhombs with angle 66°.
	1.505	<u>1.509</u>	1.513	.008	HARMOTOME (Zeolite grp) (Ba,K ₂) ₁₋₂ (Al,Si) ₈ O ₁₆ ·6H ₂ O	~ 90° disp crossed	Z = $\frac{b}{a}$ X:a = 68°	MCL pris	010 good 001 less so	White	H 4.5 G 2.44 F 3.5	Dec by HCl. Tw 001 cruciform.
	1.510	<u>1.510</u>	1.512	.002	OLYMPITE Na ₃ PO ₄	46°	---	ORTH	Conch	Cols	H 4 G 2.8	Sol in H ₂ O, gives an alk reaction.
	1.470	<u>1.510</u>	1.579	.109	STRONTIOBORITE SrB ₈ O ₁₁ (OH) ₄	85°	---	MCL plates	---	Cols	G 2.81	Diss by acids.
1.523 ^v	1.504	<u>1.510</u>	1.573	.069	PIRSSONITE Na ₂ Ca(CO ₃) ₂ ·2H ₂ O	30° r < v wk	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH pris or tab	Conch	Cols	H 3-3.5 G 2.36 F 2-2.5	Diss by acids with eff. Dec by hot H ₂ O.
	1.508	<u>1.510</u>	1.569	.061	HATCHETTITE C ₄₀ H ₈₂ (?)	26° r < v	---	ORTH(?)	001 good	Yellow	H 1 G 0.95 F 80°C	A paraffin hydro- carbon.
	1.510	<u>1.510</u>	1.545	.035	SOLONGOITE Ca ₂ B ₃ O ₄ Cl(OH) ₄	Very small r > v wk	Z:c = 2.5° el pos	MCL	---	Cols	G 2.51	Diss by HCl.
1.513 []]	1.504	<u>1.510</u>	1.519	.015	MOUNTAINITE (Ca,Na,K) ₂ Si ₄ O ₁₀ ·3H ₂ O	Med to large	Y = $\frac{b}{c}$ Z:c = 18°	MCL fib	---	White	G 2.36	---
	1.508	<u>1.510</u>	1.516	.008	DYPINGITE Mg ₅ (CO ₃) ₄ (OH) ₂ ·5H ₂ O	---	---	Globu- lar aggre- gates	---	White	G 2.15 calc	Diss by HCl with eff. Fluor light blue in UV.
1.498 [^]	1.508	<u>1.510</u>	1.513	.005	GONNARDITE (Zeolite grp) Na ₂ CaAl ₄ Si ₆ O ₂₀ ·7H ₂ O	Med to large	---	ORTH fib	---	White	G 2.27	---
	1.508	<u>1.511</u>	1.523	.015	BREWSTERITE (Zeolite grp) (Sr,Ba,Ca)Al ₂ Si ₆ O ₁₆ ·5H ₂ O	55° r > v wk	Z = $\frac{b}{c}$ X:c = 22° el clv neg	MCL el c	010 perf	White	H 5 G 2.32 F 3	Dec by acids.

133	1.505	<u>1.511</u>	1.516	.011	PETALITE $\text{LiAlSi}_4\text{O}_{10}$	83° $r > v$ wk	$Z = \frac{b}{a}$ $X:\underline{a} = -5^\circ$	MCL	001 perf 201 good	Cols, pink, white	H 6 G 2.39 F 5	Insol in acids.
	1.505	<u>1.512</u>	1.524	.019	FLAGSTAFFITE $\text{C}_{10}\text{H}_{22}\text{O}_3$	77° $r > v$	$Z = \frac{a}{c}$ $X = \underline{c}$	ORTH pris	110 imperf	Cols	Soft G 1.09 F 119°C	Sol in warm alcohol (= cis-terpin hydrate).
	1.553 v	1.501	<u>1.513</u>	1.536	.035	BOBIERRITE $\text{Mg}_3(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$	73° $r < v$ wk	$Y = \frac{b}{c}$ $Z:\underline{c} = 27^\circ$	MCL acic	010 perf Cols	H 2-2.5 G 2.20 fus	Diss by acids. Data for synth compound.
	1.525 v	1.511	<u>1.513</u>	1.518	.007	THOMSONITE (Zeolite grp) $\text{NaCa}_2\text{Al}_5\text{Si}_5\text{O}_{20} \cdot 6\text{H}_2\text{O}$	75° $r > v$	$X = \frac{a}{b}$ $Z = \underline{b}$	ORTH pris fib <u>c</u>	010 perf 100 good White	H 5 G 2.33 F 2	Gel with acids.
		1.510	<u>1.513</u>	1.517	.007	MONTEREGIANITE $(\text{Na},\text{K})_6(\text{Y},\text{Ca})_2\text{Si}_{16}\text{O}_{38} \cdot 10\text{H}_2\text{O}$	87°	$X = \frac{c}{a}$ $Y = \underline{a}$	ORTH acic, tab <u>a</u>	010 perf 001 good 100 fair White, cols	H 3.5 G 2.42	Dec by acids.
		---	<u>1.514</u>	---	---	KINGITE $\text{Al}_3(\text{PO}_4)_2(\text{OH},\text{F})_3 \cdot 9\text{H}_2\text{O}$	---	---	TCL mass	---	White	G 2.2-2.3 Opt char unk.
		1.508	<u>1.515</u>	1.523	.015	SCHERTELITE $(\text{NH}_4)_2\text{H}_2\text{Mg}(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	86°	$Z = \underline{a}$	ORTH	---	Cols	G 1.83 Dec by hot H_2O , diss by acids.
		1.513	(<u>1.515</u>)	1.518	.005	LOVDARITE $(\text{Na},\text{K})_2(\text{Be},\text{Al})\text{Si}_3\text{O}_8 \cdot 2\text{H}_2\text{O}$	70°	---	ORTH pris	100, 010, 001 dist Cols to yellowish	H 5-6 G 2.33 F 3	Insol in acids.
	1.534 v	1.518	<u>1.518</u>	1.561	.043	FIBROFERRITE $\text{Fe}(\text{SO}_4)(\text{OH}) \cdot 5\text{H}_2\text{O}$	Small	$Z = \frac{c}{a}$ el clv neg	MCL ps trig	001 perf Pale yellow, pale green	H 2.5 G 1.97 F 5	Dec by hot H_2O , diss by acids. Pleoc, X and Y cols, Z pale yellow.
		1.514	<u>1.518</u>	1.533	.019	NEWBERYITE $\text{MgHPO}_4 \cdot 3\text{H}_2\text{O}$	45° $r < v$ perc	$X = \frac{a}{b}$ $Y = \underline{b}$	ORTH tab 100	010 perf 001 imperf White	H 3-3.5 G 2.11 infus	Diss by acids.
		1.516	<u>1.518</u>	1.533	.017	FELSÖBANYAITE $\text{Al}_4(\text{SO}_4)(\text{OH})_{10} \cdot 5\text{H}_2\text{O}$	48° $r > v$	$X = \frac{a}{b}$ $Y = \underline{b}$	ORTH(?)	001 perf 100, 010 Cols	H 1.5 G 2.35 infus	Diss by acids. Clv fragments lath- shaped.
		1.501	<u>1.519</u>	1.755	.254	KLADNOITE $\text{C}_6\text{H}_4(\text{CO})_2\text{NH}$	25° (34+2°)	$Z = \frac{b}{c}$ $Y:\underline{c} = 16^\circ$	MCL	---	---	G 1.47 F 234°C = Phthalimide.
		1.505	<u>1.519</u>	1.533	.028	STANLEYITE $\text{VOSO}_4 \cdot 6\text{H}_2\text{O}$	Large	---	ORTH	---	Deep blue	H 1-1.5 G 1.95 Sol in cold H_2O . Pleoc, X and Y blue, Z pale blue.
	1.510	<u>1.520</u>	1.554	.044	NOBLEITE $\text{CaB}_6\text{O}_9(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	(76+4°) $r > v$ wk	$Y = \frac{b}{c}$ $Z:\underline{c} = -7^\circ$	MCL	100 perf	Cols	H 3 G 2.09	Diss by acids.
1.51 □ 1.53	---	<u>1.52</u>	---	.005- .010	BASALUMINITE $\text{Al}_4(\text{SO}_4)(\text{OH})_{10} \cdot 5\text{H}_2\text{O}$	---	el neg	Mass, compact	Conch	White	G 2.1	Diss by HCl. Opt char unk.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.510 ^	1.518	1.523	1.588	.070	HATCHETTITE C ₄₀ H ₈₂ (?)	33° r < v	Z = \underline{c}	ORTH(?)	001 good	White	H 1 G 0.95 F 80°C	A paraffin hydrocarbon.
	1.520	1.523	1.533	.013	GYPSUM CaSO ₄ ·2H ₂ O	58° r > v perc	Y = \underline{b} X:c = 37° disp str	MCL tab 010	010 perf 011, 100 poor	White	H 1.5-2 G 2.32 F 2.5-3,	Diss by HCl. Tw pl 100.
	1.521	1.523	1.533	.012	MASCAGNITE (NH ₄) ₂ SO ₄	52° r < v wk	X = \underline{c} Y = \underline{b}	ORTH el \underline{c}	001 dist	Cols, yellowish	H 2-2.5 G 177 F 1	Sol in H ₂ O. Volat when heated. Tw pl 110.
	1.518	---	1.530	.012	TACHARANITE Ca ₁₂ Al ₂ Si ₁₈ O ₅₁ ·18H ₂ O	---	X = fib	MCL fib	001 good	Cols	G 2.35	Opt char unk.
1.534 v	1.518	1.524	1.544	.026	WAVELLITE Al ₃ (PO ₄) ₂ (OH,F) ₃ ·5H ₂ O	60° r > v wk	X = \underline{b} Y = \underline{a} el pos	ORTH fib, radiating	110 perf 101 good 010 dist	White, yellow, green	H 3.5-4 G 2.36 infus	Diss by acids. Colored vars pleoc, X greenish, Z yellowish, abs X > Z.
	1.518	1.524	1.530	.012	MACDONALDITE BaCa ₄ Si ₁₆ O ₃₆ (OH) ₂ ·10H ₂ O	90°	X = \underline{c} Y = \underline{b}	ORTH acic	010 perf 001 good	Cols, white	H 3.5-4 G 2.27 F 5.5	Insol in cold acids, dec by boiling HCl (1:1).
	1.508	1.525	1.586	.078	SIDERONATRITE Na ₂ Fe(SO ₄) ₂ (OH)·3H ₂ O	58° r > v str	X = \underline{a} Y = \underline{b}	ORTH fib \underline{c}	001 perf	Orange to straw- yellow	H 2 G 2.28 F 2	Dec by hot H ₂ O, diss by HCl. Pleoc, X cols, Y pale yellow, Z pale amber-yellow.
	1.513	1.525	1.577	.064	STRONTIOGINORITE (Sr,Ca) ₂ B ₁₄ O ₂₃ ·8H ₂ O	50°	Y = \underline{b} Z:c = 40°	MCL	010 perf	White	H 3 G 2.25 fus	Diss by acids. Sr:Ca = 2:1 to 1:1.
	1.517	1.525	1.579	.062	GINORITE Ca ₂ B ₁₄ O ₂₃ ·8H ₂ O	42° inclined disp	Y = \underline{b} Z:c = 39°	MCL flat rhombs	010 perf	White	H 3.5 G 2.08 fus	Diss by acids.
	1.515	1.525	1.544	.029	PROBERTITE NaCaB ₅ O ₇ (OH) ₄ ·3H ₂ O	73° r > v	Y = \underline{b} Z:c = 12° el pos	MCL pris	110 perf	White	H 3.5 G 2.14 F 2	Diss by acids.
1.547 v	1.513	1.525	1.542	.029	VOGLITE Ca ₂ Cu(UO ₂)(CO ₃) ₄ ·6H ₂ O (?)	90° (81±8°)	X = \underline{b} Y = \underline{c} el cTv pos	MCL	010 perf	Emerald- to grass- green	Soft infus	Diss by acids. Pleoc, X deep blue, Y bluish-green, Z yellow. Poly tw.

	1.521	<u>1.525</u>	1.545	.024	SPADAITE $\text{MgSiO}_2(\text{OH})_2 \cdot \text{H}_2\text{O} (?)$	Small to med (49±12°)	ext ~ , el pos	ORTH(?) felted, columnar	---	Cream to pink	G 2.2 F 5	Dec by HCl.
	1.523	<u>1.525</u>	1.532	.009	CHALCOALUMITE $\text{CuAl}_4(\text{SO}_4)(\text{OH})_{12} \cdot 3\text{H}_2\text{O}$	Small r > v str	Ext 40° on flat face	MCL fib crusts	Several perf	Turquoise green to blue	H 2.5 G 2.29 F 5	Diss by hot acids. Tw pl el.
1.513 ↓ 1.533	1.523	<u>1.525</u>	1.532	.009	THOMSONITE (Zeolite grp) $\text{NaCa}_2\text{Al}_5\text{Si}_5\text{O}_{20} \cdot 6\text{H}_2\text{O}$	48° r > v	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH pris, fib c	010 perf 100 good	White	H 5 G 2.32 F 2	Gel with acids.
	1.508	<u>1.527</u>	1.550	.042	KALIBORITE $\text{KMg}_2\text{B}_{12}\text{O}_{16}(\text{OH})_{10} \cdot 4\text{H}_2\text{O}$	81°	Y = $\frac{b}{c}$ Z:c = 65°	MCL pris	001, T01 perf 100 good	White	H 4-4.5 G 2.13 F 1	Diss by acids.
1.537 ↓	1.523	<u>1.527</u>	1.545	.022	HYDROMAGNESITE $\text{Mg}_5(\text{CO}_3)_4(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	50°	Z = $\frac{b}{c}$ X:c = 47° el clv neg	MCL ps orth fib c	010 perf parting 100	White	H 4-4.5 G 2.22 infus	Diss by hot acids. Tw pl 100, poly common.
	1.526	<u>1.528</u>	1.551	.025	TERUGGITE $\text{Ca}_4\text{MgAs}_2\text{B}_{12}\text{O}_{22}(\text{OH})_{12} \cdot 12\text{H}_2\text{O}$	33° r > v wk	Z = $\frac{b}{c}$ X:c = -26°	MCL pris	001 good 110 fair	Cols, white	H 2.5 G 2.15 fus	Diss by acids.
	1.525	<u>1.528</u>	1.544	.019	SAZHINITE $\text{Na}_2\text{CeSi}_6\text{O}_{14}(\text{OH}) \cdot 5\text{H}_2\text{O}$	47°	Z = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH ps tet tab	100, 010, 001 perf	Gray, white, cream	H 2-3 G 2.61	---
	1.527	---	1.531	.004	BIRUNITE $\text{Ca}_{36}(\text{SiO}_3)_{17}(\text{CO}_3)_{17}(\text{SO}_4)_2 \cdot 30\text{H}_2\text{O} (?)$	---	---	ORTH(?) fib	One perf	White	H 2 G 2.36 infus	Gel with acids.
1.535 ↓	1.507	<u>1.529</u>	1.573	.067	MAGNESIOCOPIAPITE $(\text{Mg}, \text{Fe})\text{Fe}^{+3}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20\text{H}_2\text{O}$	73° r > v str	X ~ $\frac{b}{c}$ Y:c ~ 38°	TCL tab	010 perf I01 imperf	Sulfur- to golden- yellow	H 2.5-3 G 2.13 F 4-5	Diss by acids. Pleoc, X yellow-green, Y pale yellow, Z yellow to yellow-green.
1.548 ↓	1.522	<u>1.529</u>	1.577	.055	BOTRYOGEN $\text{MgFe}^{+3}(\text{SO}_4)_2(\text{OH}) \cdot 7\text{H}_2\text{O}$	41°	X = $\frac{b}{c}$ Z:c = 12°	MCL reni- form	010 perf 110 good	Red to orange	H 2-2.5 G 2.05-2.14 F 4.5-5	Dec by hot H ₂ O, diss by acids. Pleoc, X cols to brown, Y pale brown, Z golden- to orange-red.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.483	<u>1.530</u>	1.576	.093	CHALCONATRONITE $\text{Na}_2\text{Cu}(\text{CO}_3)_2 \cdot 3\text{H}_2\text{O}$	Large (93±2°)	Y = e1 Z:c small	MCL ps hex	---	Greenish-blue	Soft G 2.27	Dec by H ₂ O, diss by acids. Pleoc, X nearly cols, Y pale blue, Z blue.
	1.515	<u>1.530</u>	1.580	.065	EARLANDITE $\text{Ca}_3(\text{C}_6\text{H}_5\text{O}_7)_2 \cdot 4\text{H}_2\text{O}$	60°	---	MCL	---	White to pale yellow	G 1.95	Calcium citrate. Diss by acids.
	---	---	1.542	---	GLAUCOKERINITE $(\text{Zn,Cu})_{10}\text{Al}_4(\text{SO}_4)(\text{OH})_{30} \cdot 2\text{H}_2\text{O}$ (?)	---	Z = e1	Radiating, fib	---	Sky-blue	H 1 G 2.15	Concentric color banding. Opt char unk.
	1.525	<u>1.53</u>	1.550	.025	ROESSLERITE $\text{MgHAsO}_4 \cdot 7\text{H}_2\text{O}$	Small	Z = b X:a = 14°	MCL	110 imperf	Cols	H 2-3 G 1.94	Diss by acids.
	1.496	<u>1.531</u>	1.579	.083	CHALLANTITE $6\text{Fe}_2(\text{SO}_4)_3 \cdot \text{Fe}_2\text{O}_3 \cdot 6\text{H}_2\text{O}$ 63H ₂ O	80°	---	tab	---	Yellow	---	Slowly sol in H ₂ O, diss by HCl.
	1.530	<u>1.531</u>	1.534	.004	BUDDINGTONITE (Feldspar grp) $(\text{NH}_4)_2\text{Al}_2\text{Si}_6\text{O}_{16} \cdot \text{H}_2\text{O}$	---	Z = b X:a = 4°	MCL	001 good 010 dist	Cols	H 5.5 G 2.32 F 5	Insol in acids.
	1.5285	<u>1.531</u>	1.532	.004	Unnamed silicate $(\text{K,Na})_4\text{Ca}_2\text{AlSi}_6(\text{O,OH,F})_{18} \cdot \text{NaCl}$	29°	e1 clv neg	ORTH	One perf	Cols	H 4 G 2.58 F easy	Gel with acids (Am. Min. 44, 90).
v 1.541	1.528	<u>1.532</u>	1.539	.011	ALBITE plagioclase, plutic (Feldspar grp) $\text{NaAlSi}_3\text{O}_8$	74° r < v wk	X':a on 001 = 3° 010 = 20°	TCL	001 perf 010 good	Cols, white	H 6 G 2.61 F 4	End member, An ₀ . Insol in acids. Poly tw pl 010 almost universal. Other tw laws common.
v 1.542	1.527	<u>1.532</u>	1.538	.011	CORDIERITE $\text{Mg}_2\text{Al}_4\text{Si}_5\text{O}_{18}$	88°	X = c Z = b	ORTH	010 poor	Blue, cols	H 7 G 2.59 F 5.5	Nearly insol in acids. Fe ₂ O ₃ 1.5, FeO 2.2, Na ₂ O 0.2%. Tw 110, 120.
v 1.590	1.520	<u>1.533</u>	1.584	.064	KIESERITE $\text{MgSO}_4 \cdot \text{H}_2\text{O}$	55° r > v mod	Y = b Z:c = -77° disp dist	MCL	111, 110 perf 111, 101, 011 imperf	White	H 3.5 G 2.57 F 2-3	Slowly sol in H ₂ O, diss by acids. Poly tw.

1.525 ^	1.53	<u>1.533</u>	1.542	.012	THOMSONITE (Zeolite grp) $\text{Na}(\text{Ca}, \text{Sr})_2\text{Al}_5\text{Si}_5\text{O}_{20} \cdot 6\text{H}_2\text{O}$	52° $r > v$, dist	$X = \frac{a}{b}$ $Z = \frac{c}{b}$	ORTH pris, fib <u>c</u>	010 perf 100 good	White	H 5 G 2.37 F 2	Gel with acids.
	1.519	<u>1.534</u>	1.569	.050	TUNELLITE $\text{SrB}_6\text{O}_{10} \cdot 4\text{H}_2\text{O}$	(68+5°) $r > v$ wk	$Y = \frac{b}{c}$ $Z:\underline{c} = -5^\circ$	MCL	100 perf 001 dist	Cols	H 2.5 G 2.40 fus	Diss by acids.
	1.522	<u>1.534</u>	1.569	.047	HYDROBORACITE $\text{CaMgB}_6\text{O}_8(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	63° $r < v$ perc	$Y = \frac{b}{c}$ $X:\underline{c} = 33^\circ$ el neg	MCL tab 010	010 perf 100 less so	White	H 2-3 G 2.17 F 2	Diss by acids.
1.518 ^	1.533	<u>1.534</u>	1.575	.042	FIBROFERRITE $\text{Fe}(\text{SO}_4)(\text{OH}) \cdot 5\text{H}_2\text{O}$	Small	$Z = \underline{c}$	MCL ps trig	001 perf	Pale yellow	H 2.5 G 1.9-2.1 F 5	Diss by acids. Pleoc, X and Y nearly cols, Z pale yellow.
	1.520	---	1.558	.038	CHELKARITE $\text{CaMgB}_2\text{O}_4\text{Cl}_2 \cdot 7\text{H}_2\text{O}$ (?)	---	---	ORTH pris	Perf el	Cols to pale rose	G 2.21	Opt char unk.
1.524 ◇ 1.543	1.525	<u>1.534</u>	1.553	.028	WAVELLITE $\text{Al}_3(\text{PO}_4)_2(\text{OH}, \text{F})_3 \cdot 5\text{H}_2\text{O}$	72° $r > v$ wk	$X = \frac{b}{a}$ $Y = \frac{a}{b}$ el pōs	ORTH fib, radi- ating	110 perf 101 good 010 dist	White, yellow, green	H 3.5-4 G 2.36 infus	Diss by acids. Colored vars are pleoc, X greenish, Z yellowish, abs X > Z.
1.527 □ 1.537	1.531	<u>1.534</u>	1.538	.007	MINYULITE $\text{KAl}_2(\text{PO}_4)_2(\text{OH}, \text{F}) \cdot 4\text{H}_2\text{O}$	Large	$X = \frac{c}{a}$ $Y = \frac{a}{c}$	ORTH acic <u>c</u>	---	Cols	H 3.5 G 2.46 F 4-5	Diss by warm acids.
1.529 ◇ 1.540	1.52	<u>1.535</u>	1.578	.06	COIAPITE $\text{Fe}^{+2}\text{Fe}^{+3}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20\text{H}_2\text{O}$	70° (63+4°) $r > v$ str	$X \sim \frac{b}{c}$ $Y:\underline{c} \sim 38^\circ$	TCL tab	010 perf T01	Sulfur- to golden- yellow	H 2.5-3 G 2.1 F 4-5	Diss by acids. Pleoc, X and Y pale yellow, Z yellow.
	---	<u>1.535</u>	---	very low	NEKOITE $\text{Ca}_3\text{Si}_6\text{O}_{12}(\text{OH})_6 \cdot 5\text{H}_2\text{O}$	70°	X:needles = 26° $Z \sim \perp 100$	TCL acic	100 good	Cols	G 2.23	Tw.
	1.531	<u>1.536</u>	1.544	.013	PENTAHYDROBORITE $\text{CaB}_2\text{O}(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	73°	---	TCL u mass	---	Cols	H 2.5 G 2.01	Diss by acids.
	1.536	<u>1.536</u>	1.543	.007	ROEDDERITE (Osumilite grp) $(\text{Na}, \text{K})_2(\text{Mg}, \text{Fe})_5\text{Si}_{12}\text{O}_{30}$	5-8°	---	HEX	---	Cols	G 2.6	---
	1.535	<u>1.536</u>	1.538	.003	FLUORAPOPHYLLITE $\text{KCa}_4\text{Si}_8\text{O}_{20}\text{F} \cdot \text{H}_2\text{O}$	18-42°	$Z = \underline{c}$	TET	001 perf 110 poor 110 poor	Cols to yellow	H 5 G 2.3-2.4 F 2	Dec by HCl with sepn of slimy silica.
1.527 ^	1.527	<u>1.537</u>	1.550	.023	HYDROMAGNESITE $\text{Mg}_5(\text{CO}_3)_4(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	83°	$Z = \frac{b}{c}$ $X:\underline{c} = 47^\circ$	MCL ps orth, fib <u>c</u>	010 perf parting 100	White	H 4-4.5 G 2.17 infus	Diss by hot acids. Tw pl 100, poly common.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.535 ^ 1.545	1.535	<u>1.537</u>	1.545	.010	ASHCROFTINE KNaCaY ₂ Si ₆ O ₁₂ (OH) ₁₀ ·4H ₂ O	Small	Z = <u>c</u>	TET acic	100 perf 001 good	White, pink	H 5 G 2.61	---
	1.536	<u>1.538</u>	1.544	.008	NATROAPOPHYLLITE NaCa ₄ Si ₈ O ₂₀ F·8H ₂ O	32°	---	ORTH	001 perf	Cols to yellow	G 2.50	Slightly dec by HCl.
	1.536	<u>1.539</u>	1.603	.067	VOLKOVSKITE (Ca,Sr)B ₆ O ₁₆ ·3H ₂ O	24°	Y = <u>b</u> Z:a = 31°	MCL	010 perf 001 good	Cols	G 2.32	---
	1.525	<u>1.540</u>	1.590	.065	ALUMINOCOPIAPITE AlFe ⁺³ ₄ (SO ₄) ₆ O(OH)·20H ₂ O	Mod	---	TCL	010 perf	Yellow	---	Pleoc in yellow.
	1.541	<u>1.541</u>	1.547	.006	OSUMILITE-(Mg) (Osumilite grp) (K,Na)(Mg,Fe) ₂ (Al,Fe) ₃ (Si,Al) ₁₂ O ₃₀ ·H ₂ O	5-15°	---	HEX	---	Cols	G 2.63	---
	1.532 ^ 1.541 neg	<u>1.541</u>	1.546	.010	OLIGOCLASE plagioclase, plutonic (Feldspar grp) (Na,Ca)Al(Si,Al)Si ₂ O ₈	~ 90°	X':a on 001 = 1° 010 = 7°	TCL	001 perf 010 good	Cols	H 6 G 2.64	Data for An ₁₈ . Insol in acids. Tw poly on 010.
	1.532	<u>1.542</u>	1.556	.024	PHAUNOUXITE Ca ₃ (AsO ₄) ₂ ·11H ₂ O	80°	---	TCL	100 perf 011 good	Cols	G 2.28	Diss by acids.
	1.540	<u>1.542</u>	1.550	.010	RAITE Na ₄ Mn ₄ Si ₈ (OH) ₂₄ ·8-10H ₂ O	(~ 50°)	Z:c = 0- 30°	ORTH acic	100, 010, 001 perf	Gold to brown	H 3 G 2.39 F easy	Dec by acids. Pleoc, X nearly cols, Y yellowish, Z gold or brown. El pos.
	1.532 ^ 1.550	<u>1.542</u>	1.547	.009	CORDIERITE (Mg,Fe) ₂ Al ₄ Si ₅ O ₁₈	84°	X = <u>c</u> Z = <u>b</u>	ORTH	010 dist	Blue	H 7 G 2.64 F 5	Partly dec by acids. FeO 8.0, Fe ₂ O ₃ 0.6, Na ₂ O 1.0%.
	1.534 ^	<u>1.543</u>	1.561	.026	WAVELLITE Al ₃ (PO ₄) ₂ (OH,F) ₃ ·5H ₂ O	65° r > v wk	X = <u>b</u> Y = <u>a</u> el pos	ORTH radi- ating, fib	110 perf 101 good 010 dist	White, yellow, green	H 3.5-4 G 2.30 infus	Diss by acids. Colored var pleoc, X greenish, Z yellow- ish, abs X > Z.
	1.534	<u>1.543</u>	1.558	.024	GORDONITE MgAl ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	73° r > v	X ⊥ 010 Z:c = 30°	TCL laths	010 perf	Cols	H 3.5 G 2.23 F 3	Diss by acids.

v 1.550	1.536	---	1.550	.014	LOUDOUNITE NaCa ₅ Zr ₄ Si ₁₆ O ₄₀ (OH) ₁₁ · 8H ₂ O	---	---	fib	---	Green to white	H 5 G 2.48	Opt sign unk. Ext wavy.
	1.540	<u>1.543</u>	1.549	.009	CHKALOVITE Na ₂ BeSi ₂ O ₆	78° r > v	Y = <u>c</u>	ORTH	001 good 111 poor	White	H 6 G 2.66 F easy	Dec by acids with sepn of silica.
	1.540	<u>1.544</u>	1.559	.019	FARRINGTONITE (Mg,Fe) ₃ (PO ₄) ₂	55°	Z:c = 16°	MCL	100, 010 fair to good	Cols to yellowish	G ~ 2.80	FeO 3.7-5.4%. Meteorite mineral.
	1.542	<u>1.544</u>	1.551	.009	CAVANSITE Ca(VO)Si ₄ O ₁₀ ·4H ₂ O	52° r < v extr	X = <u>b</u> Y = <u>a</u>	ORTH pris	010 good	Greenish- blue	H 3-4 G 2.21-2.31	Diff diss by acids. Pleoc, X and Z cols, Y blue.
1.540 v 1.550	1.543	<u>1.544</u>	1.546	.003	EPIDIDYMITE NaBeSi ₃ O ₇ (OH)	22-30° r > v	X = <u>a</u> Z = <u>b</u>	ORTH plates	010, 001 perf	Cols	H 5.5 G 2.58 F 3	Insol in acids. Tw.
	1.525	<u>1.545</u>	1.595	.070	MAGNESIOCOPIAPITE (Mg,Fe)Fe ⁺³ (SO ₄) ₆ (OH) ₂ ·20H ₂ O	Med r > v str	X ~ b el clv pos	TCL tab	010 perf	Yellow	H 2.5-3 G 2.1 F 4.5-5	Diss by acids. Pleoc, X and Y nearly cols, Z deep yellow.
	1.532	<u>1.545</u>	1.572	.040	HALURGITE Mg ₂ B ₈ O ₁₀ (OH) ₈ ·H ₂ O	(71+6°)	X = <u>b</u> Z ~ <u>c</u>	MCL	---	Cols	H 2.5-3 G 2.19	Diss by boiling H ₂ O.
	1.539	<u>1.545</u>	1.551	.012	BRUSHITE CaHPO ₄ ·2H ₂ O	87° r > v crossed	Z = <u>b</u> X:c = -30°	MCL el <u>c</u>	010, 001 perf	White	H 2.5 G 2.30 F 3	Diss by acids.
v (1.569)	1.545	(<u>1.546</u>)	1.553	.008	CHRYSTILE (Serpentine grp) Mg ₃ Si ₂ O ₅ (OH) ₄	50°	---	MCL fib	---	Pale green	G 2.55 infus	Insol in acids. Fe ₂ O ₃ 2.3, Al ₂ O ₃ 1.2, FeO 1.6%.
	1.531	<u>1.546</u>	1.562	.031	BRASSITE MgHAsO ₄ ·4H ₂ O	Very large	Z = <u>c</u> Y = <u>a</u>	ORTH crusts	001 perf	White	G 2.28 infus	Diss by acids.
	1.545	<u>1.546</u>	1.549	.004	EUDIDYMITE NaBeSi ₃ O ₇ (OH)	23-30° r > v	Y = <u>b</u> Z:c = -55°	MCL tab	001 perf	White	H 6 G 2.58 F 3	Insol in acids. Tw pl 001, lam, always present.
1.525 v	1.541	<u>1.547</u>	1.564	.023	VOGLITE Ca ₂ Cu(UO ₂)(CO ₃) ₄ · 6H ₂ O (?)	60°	X = <u>b</u> Y = <u>c</u>	MCL	010 perf	Emerald- green	Soft infus	Poly tw lam. Pleoc str, X and Y deep bluish-green, Z yellow.
1.529 v 1.551	1.544	<u>1.548</u>	1.572	.028	BOTRYOGEN MgFe ⁺³ (SO ₄) ₂ (OH)·7H ₂ O	40° r > v str	X = <u>b</u> Z:c = 12° el clv pos	MCL reni- form	010 perf 110 good	Red to orange	H 2-2.5 G 2.08 F 4.5-5	Diss by acids. Pleoc, X yellow, Y red, Z orange-red.
	1.542	<u>1.548</u>	1.566	.024	SCHODERITE Al ₂ (VO ₄)(PO ₄)·8H ₂ O	(61+11°)	Y = <u>b</u> Z:c = 29°	MCL tab 010	---	Yellow- orange	H 2 G 1.88	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
140	1.547	<u>1.549</u>	1.559	.012	PARTHEITE $\text{CaAl}_2\text{Si}_2\text{O}_8 \cdot 2\text{H}_2\text{O}$	48° $r > v$ mod	X:el = 23-30°	MCL	clv el	White	G 2.39	---
	1.549	<u>1.549</u>	1.680	.131	STUDTITE $\text{UO}_4 \cdot 4\text{H}_2\text{O}$	Small	Z = el	MCL fib	---	Yellow	---	Slowly sol in cold dil HCl. Hydrous U-peroxide.
	1.545 $\hat{\Delta}$ 1.554	<u>1.550</u>	1.592	.062	FERRICOPIAPITE $\text{Fe}^{+3}\text{Fe}^{+3}_4(\text{SO}_4)_6\text{O}(\text{OH}) \cdot 20\text{H}_2\text{O}$	69° $r > v$	X ~ \underline{b}	TCL	010 perf	Yellow	H 2 G 2.2	Diss by acids. Pleoc in yellow.
	1.542 \wedge	<u>1.550</u>	1.556	.012	CORDIERITE $(\text{Mg}, \text{Fe})_2\text{Al}_4\text{Si}_5\text{O}_{18}$	85°	X = \underline{c} Z = \underline{b}	ORTH	010 dist	Blue	H 7 G 2.63 F 5	Partly dec by acids. FeO 8.4, Fe ₂ O ₃ 0.6%.
	1.550 neg $\hat{\Delta}$ 1.558	<u>1.550</u>	1.554	.008	ANDESINE plagioclase, plutonic (Feldspar grp) $(\text{Na}, \text{Ca})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	90°	X': \underline{a} on 001 = 0° 010 = -5°	TCL	001 perf 010 good	Cols	H 6 G 2.66 F 5.5	Insol in acids. Poly tw 010. Data for An ₃₃ .
	1.543 \wedge	<u>1.550</u>	1.553	.004	CHKALOVITE $\text{Na}_2\text{BeSi}_2\text{O}_6$	81° $r > v$	Y = \underline{c}	ORTH	001 good 111 poor	White	H 6 G 2.66 F easy	Dec by acids with sepn of silica.
		<u>1.551</u>	1.621	.072	VEATCHITE-A $\text{Sr}_2\text{B}_{11}\text{O}_{16}(\text{OH})_5 \cdot \text{H}_2\text{O}$	25°	X = \underline{c} Z = \underline{b}	TCL	100 perf 011, 011 good	Cols	G 2.73	Diss by acids.
	1.548 \wedge	<u>1.551</u>	1.587	.045	ZINCOBOTRYOGEN $(\text{Zn}, \text{Mg})\text{Fe}^{+3}(\text{SO}_4)_2(\text{OH}) \cdot 7\text{H}_2\text{O}$	(54+6°) $r > v$	X = \underline{b}	MCL pris	010 good	Orange-red	H 2.5 G 2.20	Diss by acids. Pleoc str, X yellow, Y and Z red.
		<u>1.552</u>	1.570	.030	RAUENTHALITE $\text{Ca}_3(\text{AsO}_4)_2 \cdot 10\text{H}_2\text{O}$	85°	Y:el = 5-10°	TCL spher	---	Snow-white	G 2.36	---
	1.570 \vee	<u>1.552</u>	1.570	.022	NORBERGITE (Humite grp) $\text{Mg}_3\text{SiO}_4(\text{F}, \text{OH})_2$	33° (51+13°)	X = \underline{a} Z = \underline{b}	ORTH	---	Cols	H 5 G 3.2 infus	Gel with acids. Data for synth F compd.
	1.552 neg $\hat{\Delta}$ 1.562	<u>1.552</u>	1.556	.007	ANDESINE plagioclase, volcanic (Feldspar grp) $(\text{Na}, \text{Ca})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	~ 90°	X': \underline{a} on 001 = 2° 010 = -4°	TCL	001 perf 010 good	Cols	H 6 G 2.67 F 5.5	Insol in acids. Poly tw 010. Data for An ₃₆ .
		<u>1.553</u>	1.638	.104	RHOMBOCLASE $\text{HFe}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	27° (53+3°)	X = \underline{c} Y = \underline{a}	ORTH tab 001	001 perf 110 good	Cols to yellowish	H 2 G 2.23 fus	Slowly sol in H ₂ O, diss by acids.

1.513 ^	1.543	<u>1.553</u>	1.620	.077	VEATCHITE $\text{Sr}_2\text{B}_{11}\text{O}_{16}(\text{OH})_5 \cdot \text{H}_2\text{O}$	44° r > v wk	Z = $\frac{b}{c}$ X:c = -30°	MCL	100, 011 perf	White	H 2 G 2.69 F 1.5	Diss by acids.
	1.551	<u>1.553</u>	1.621	.070	p-VEATCHITE $\text{Sr}_2\text{B}_{11}\text{O}_{16}(\text{OH})_5 \cdot \text{H}_2\text{O}$	37° (20+10°)	Y = $\frac{b}{c}$	MCL	---	Cols	H 2 G 2.60-2.69 F 1.5	Diss by acids.
	1.547	<u>1.553</u>	1.582	.035	BOBIERRITE $(\text{Mg}, \text{Fe})_3(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$	57° r < v	Y = $\frac{b}{c}$ Z:c = 27°	MCL acic	010 perf	Cols	H 2-2.5 G 2.3 fus	Diss by acids. FeO 15.1, MnO 2.95%.
	1.550	<u>1.553</u>	1.559	.009	CHAROITE $\text{K}(\text{Ca}, \text{Na})_2\text{Si}_4\text{O}_{10}$ (OH, F) · H ₂ O	29°	X = $\frac{b}{c}$ Z:c = 5° el pos	MCL	3 clv	Lilac to violet	G 2.54	In thick fragments, pleoc, X rose, Z cols.
	1.540	<u>1.553</u>	1.570	.030	LEMOYNITE $(\text{Na}, \text{K})_2\text{CaZr}_2\text{Si}_{10}\text{O}_{26} \cdot$ 5-6H ₂ O	80° r < v wk	Y = $\frac{b}{c}$ Z:a = 5° el pos	MCL	100	Cols	H 4 G 2.29	---
	1.550 ^ 1.575	<u>1.554</u>	1.586	.052	ZINCOCOPIAPITE $\text{ZnFe}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 18\text{H}_2\text{O}$	(78+5°) r < v str	X ~ $\frac{b}{c}$ el cTv pos	TCL tab	010 perf	Yellow- green	H 2 G 2.18 F 5	Diss by acids. Pleoc str in shades of yellow.
	1.547	---	1.571	.024	GLUCINE $\text{Ca}_2\text{Be}_8(\text{PO}_4)_4(\text{OH})_8 \cdot \text{H}_2\text{O}$	---	el pos, ext	Fine needles	---	Cols	H ~ 5 G 2.23-2.40	Diss by acids. Opt char unk.
	---	<u>1.554</u>	---	.016	GROTHINE Silicate of Ca, Al, Fe (?)	Med r < v	Y = $\frac{c}{a}$ Z = $\frac{a}{b}$	ORTH tab 010	---	Cols	G 3.09 infus	Dec by H ₂ SO ₄ .
	1.550 v 1.571	<u>1.554</u>	1.558	.005	CLINOCHLORE var. Corundophilite (Chlorite grp) $(\text{Mg}, \text{Fe})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}$ (OH) ₈	50° r < v	Z ~ $\frac{c}{a}$	MCL	001 mic	Green	H 3 G 2.6 F diff	Insol in acids.
	1.491	<u>1.555</u>	1.654	.163	WHEWELLITE (oxalate) $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$	83° r < v wk	X = $\frac{b}{c}$ Z:c = 30°	MCL	101 good 001, 010 dist	Cols	H 2.5-3 G 2.21 infus	Diss by acids. Tw pl 101.
1.549 ^	1.545	<u>1.555</u>	1.680	.135	STUDTITE $\text{UO}_4 \cdot 4\text{H}_2\text{O}$	Small	Z = el	MCL fib	---	Yellow	---	Slowly sol in cold dil HCl. Hydrous U-peroxide.
	1.532	<u>1.555</u>	1.591	.059	AMARILLITE $\text{NaFe}(\text{SO}_4)_2 \cdot 6\text{H}_2\text{O}$	Large r < v	Y = $\frac{b}{c}$ Z:c = 51°	MCL thick tab	110 good	Pale greenish- yellow	H 2.5-3 G 2.19	Sol in H ₂ O.
	1.546	<u>1.555</u>	1.587	.041	HARTITE $\text{C}_{19}\text{H}_{32}$ (?)	Med (57+7°) r < v perc	---	TCL	---	White	H < 1 G 1.06 F 1	A hydrocarbon. Diss by alcohol or ether.
	1.552	<u>1.555</u>	1.565	.013	WOODWARDITE $\text{Cu}_4\text{Al}_2(\text{SO}_4)(\text{OH})_{12} \cdot$ 3H ₂ O (?)	(58+19°)	---	ORTH(?) spher	---	Greenish- blue	G 2.38	---
	~1.571	<u>1.555</u>	1.565	.013	WOODWARDITE $\text{Cu}_4\text{Al}_2(\text{SO}_4)(\text{OH})_{12} \cdot$ 3H ₂ O (?)	(58+19°)	---	ORTH(?) spher	---	Greenish- blue	G 2.38	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.561 ┌ 1.551 └ 1.550 ┆ 1.568	1.551	<u>1.555</u>	1.562	.011	VAUXITE $\text{FeAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	32° (74+21°) $r > v$ str	$Z \sim \perp$ 001	TCL tab 010	---	Sky blue to vene- tian blue	H 3.5 G 2.39	Pleoc str, X and Z cols, Y blue.
	1.550	<u>1.557</u>	1.567	.017	LITHIOPHOSPHATE Li_3PO_4	69-88°	$X = a$ $Y = \underline{b}$	ORTH u mass	100 perf 110 dist	Cols to rose	H 4 G 2.46	Slowly sol in H_2O , diss by acids.
	1.551	<u>1.558</u>	1.582	.031	METAVARISCITE $\text{AlPO}_4 \cdot 2\text{H}_2\text{O}$	55° $r < v$ wk	$Y = \underline{b}$	MCL	010	Green	H 3.5 G 2.54 infus	Diss by HCl after being heated gently (color changes to lavender). Pleoc wk, X cols, Y and Z pale green.
	1.545	<u>1.554</u>	1.565	.020	Unnamed phosphate	~ 90°	ext large	MCL(?)	---	Cols	---	Occurs with Lacroix- ite (Am. Mineral., 57, 1914 (1972)).
	1.554	<u>1.558</u>	1.573	.019	PARAUAUXITE $\text{FeAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Med	---	TCL pris	010 perf	Cols	H 3 G 2.36	---
	1.555	<u>1.558</u>	1.562	.007	LABRADORITE plagioclase, plutonic (Feldspar grp) $(\text{Ca}, \text{Na})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	78°	$X': a$ on $001 = -6^\circ$ $010 = -17^\circ$	TCL	001 perf 010 good	Cols	H 6 G 2.69	Insol in acids. Poly tw 010. Data for An_{50} .
	1.536	<u>1.559</u>	1.697	.161	ZELLERITE $\text{Ca}(\text{UO}_2)(\text{CO}_3)_2 \cdot 5\text{H}_2\text{O}$	30-40° (47+2°) $r < v$ wk	$Z = \underline{c}$	ORTH fib	---	Light lemon- yellow	Soft G 3.25	Pleoc, X and Y cols, Z light green. Fluor green in UV.
	1.559	<u>1.560</u>	1.584	.025	BASSANITE $2\text{CaSO}_4 \cdot \text{H}_2\text{O}$	10-15°	$Z \parallel$ fib $X \perp$ fib el pos	HEX	---	White	G 2.7 F 3	Diss by HCl. Tw pl 100.
	1.557	<u>1.561</u>	1.569	.012	CLINOPHOSINAITE $\text{Na}_3\text{CaPSiO}_7$	80°		MCL	Conch	Pale lilac	H 4 G 2.85	Dec by HCl.
	1.494	<u>1.561</u>	1.692	.198	HUMBOLDTINE (oxalate) $\text{FeC}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$	Large	---	MCL ps orth	---	Yellow	H 1.5-2 G 2.28 infus	Diss by acids. Pleoc, X pale yellow- green, Y pale greenish-yellow, Z intense yellow.
	1.550	<u>1.561</u>	1.582	.032	METAVAUXTITE $\text{FeAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Large	$X = \underline{b}$ $Z: \underline{c} = -17^\circ$	MCL acic	---	Cols	H 3 G 2.34	---

	1.559	<u>1.562</u>	(1.572)	.013	MCNEARITE $\text{NaCa}_5\text{H}_4(\text{AsO}_4)_5 \cdot 4\text{H}_2\text{O}$	66° disp str	$\text{Y}:\underline{\text{c}} = 0-22^\circ$ el pos	TCL fib	One perf	White	G 2.60	Diss by acids.
1.552 ∧ 1.569	1.558	<u>1.562</u>	1.566	.008	LABRADORITE plagioclase, volcanic (Feldspar grp) $(\text{Ca}, \text{Na})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	76°	$\text{X}':\underline{\text{a}}$ on 001 = -15° 010 = -26°	TCL	001 perf 010 good	Cols	H 6 G 2.69 infus	Insol in acids. Poly tw 010. Data for An_{58} .
	1.561	<u>1.563</u>	1.567	.006	DICKITE $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	55-80° $r > v$ str	$\text{Z} = \underline{\text{b}}$ $\text{Y}:\underline{\text{a}} = 14-20^\circ$	MCL ps hex plates	001 perf	White	H 2 G 2.62 infus	Insol in acids.
	1.554	<u>1.564</u>	1.595	.041	BARICITE $(\text{Mg}, \text{Fe})_3(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$	59° $r < v$ wk	$\text{X} = \underline{\text{b}}$ $\text{Z}:\underline{\text{c}} = 32^\circ$	MCL plates 010	010 perf	Cols to pale blue	H 1.5-2 G 2.42	Compare Vivianite. FeO 19.9, Fe_2O_3 2.8%.
	1.547	<u>1.566</u>	1.594	.047	QUENSTEDTITE $\text{Fe}_2(\text{SO}_4)_3 \cdot 10\text{H}_2\text{O}$	70° (80+5°) $r < v$ str horizontal	---	TCL	010 perf 100 good	Violet	H 2.5 G 2.15	Sol in H_2O .
	1.562	<u>1.566</u>	1.587	.025	SENEGALITE $\text{Al}_2(\text{PO}_4)(\text{OH})_3 \cdot \text{H}_2\text{O}$	53° $r > v$	$\text{Z} = \underline{\text{a}}$ $\text{Y} = \underline{\text{c}}$	ORTH	---	Cols	G 2.55	---
1.571	1.561	<u>1.568</u>	1.585	.024	PICROPHARMACOLITE $\text{H}_2\text{Ca}_4\text{Mg}(\text{AsO}_4)_4 \cdot 11\text{H}_2\text{O}$	Med large	$\text{X} = \underline{\text{b}}$ $\text{Z}:\underline{\text{c}} = 8-15^\circ$ el pos	TCL fib	One good	White	G 2.64	Diss by acids.
1.579	1.568	<u>1.568</u>	1.587	.019	GIBBSITE $\text{Al}(\text{OH})_3$	0-5° $r < v$ or $r > v$	$\text{X} = \underline{\text{b}}$ $\text{Y}:\underline{\text{c}} = 69^\circ$	MCL ps hex, tab 001	001 perf	White, brownish	H 3 G 2.3-2.4 infus	Diss by hot concd H_2SO_4 or by hot alkalies.
1.565 □ 1.570	1.562	<u>1.568</u>	1.575	.013	ELPIDITE $\text{Na}_2\text{ZrSi}_6\text{O}_{15} \cdot 3\text{H}_2\text{O}$	84° $r < v$ dist	$\text{X} = \underline{\text{c}}$ $\text{Z} = \underline{\text{a}}$ el neg	ORTH el $\underline{\text{c}}$	110 dist	White, red, brown	H 5-7 G 2.63	Insol in acids. Pleoc wk in yellow, abs $\text{Y} > \text{X}$.
1.558 ∧ 1.568 neg	1.563	<u>1.568</u>	1.573	.010	BYTOWNITE plagioclase, plutonic (Feldspar grp) $(\text{Ca}, \text{Na})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	~ 90°	$\text{X}':\underline{\text{a}}$ on 001 = -16° 010 = -29°	TCL	001 perf 010 good	Cols	H 6 G 2.71 infus	Insol in acids. Poly tw on 010. Data for An_{70} .
	1.560	<u>1.569</u>	1.584	.024	GÖRGEYITE $\text{K}_2\text{Ca}_5(\text{SO}_4)_6 \cdot \text{H}_2\text{O}$	80°	---	MCL tab	100 dist	Cols, yellowish	H 3.5 G 2.8-2.9	Dec by H_2O , diss by acids.
1.582	1.563	<u>1.569</u>	1.581	.018	SUDOITE (Chlorite grp) $\text{Mg}_2(\text{Al}, \text{Fe})_3\text{Si}_3\text{AlO}_{10}$ (OH) ₈	54° (71+13°)	$\text{Z}:\underline{\text{c}} = 7^\circ$ el clv neg	MCL	001 perf	White	G 2.63	Dioctahedral chlor- ite.
□	1.568	<u>1.569</u>	1.581	.013	LIZARDITE (Serpentine grp) $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	Small	---	TRIG and HEX fib	001 perf	Green	G 2.58 infus	Nearly insol in acids. Fe_2O_3 3.6, FeO 1.1%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.562 ^ 1.569 neg	1.565	<u>1.569</u>	1.574	.009	BYTOWNITE plagioclase, volcanic (Feldspar grp) (Ca,Na)Al(Si,Al)Si ₂ O ₈	~ 90°	X':a on 001 = -23° 010 = -33°	TCL	001 perf 010 good	Cols	H 6 G 2.72 infus	Insol in acids. Poly tw on 010. Data for An ₇₅ .
(1.546) ^	1.569	(<u>1.569</u>)	1.570	.001	CHRYSTILE (Serpentine grp) Mg ₃ Si ₂ O ₅ (OH) ₄	Med	---	MCL fib	---	Dark green	G 2.53 infus	Insol in acids. Fe ₂ O ₃ 1.4, Al ₂ O ₃ 0.7, FeO 1.3%.
1.628 ^	1.555	<u>1.57</u>	1.585	.030	TENGERITE CaY ₃ (CO ₃) ₄ (OH) ₃ ·3H ₂ O	Large	X = e1	ORTH fib	---	White	Infus G 2.8	Diss by acids with eff.
1.552 ^	1.565	<u>1.570</u>	1.591	.026	NORBERGITE (Humite grp) Mg ₃ SiO ₄ (F,OH) ₂	49°	X = a Z = $\frac{b}{c}$	ORTH	---	Yellow to brown	H 5 G 3.20 infus	Gel with acids.
1.582 ^	1.569	<u>1.570</u>	1.582	.013	WAGNERITE (Mg,Fe) ₂ (PO ₄)F	27° r > v perc	Y = $\frac{b}{c}$ Z: $\frac{c}{a}$ = -22°	MCL	100, 120, 010 imperf	Cols, flesh, yellow	H 5-5.5 G 3.15 F 4	Slowly diss by acids.
1.582 ^	---	<u>1.570</u>	1.626	>.05	BUKOVSKYITE Fe ₂ (AsO ₄)(SO ₄)(OH)· 7H ₂ O	---	ext at 18°	MCL(?) needles	---	Yellow to gray- green	G 2.34	Diss by HCl. Opt char unk.
	1.543	<u>1.571</u>	1.577	.043	SHABYNITE Mg ₅ (BO ₃)(Cl,OH) ₂ (OH) ₅ · 4H ₂ O	---	X e1	MCL fib	---	Snow-white	H 3 G 2.32	Diss by acids.
1.589 ^	1.563	<u>1.571</u>	1.596	.033	HOERNESITE Mg ₃ (AsO ₄) ₂ ·8H ₂ O	60°	X = b Z:c = $\frac{b}{c}$ 31° e1 clv pos	MCL tab 010	010 perf	White	G 2.73 F 2-3	Diss by acids.
	1.565	<u>1.571</u>	1.584	.019 disp str	BRUCITE Mg(OH) ₂	30-70°	X = e1	TRIG fib	0001 perf	Greenish- white	H 2.5 G 2.4 infus	Diss by acids. Anom biax. Anom interf colors in some specimens.
	1.571	<u>1.571</u>	1.590	.019	NEFEDOVITE Na ₅ Ca ₄ (PO ₄) ₄ F	~ 0°	---	TCL ps tet	Conch	Cols	H 4.5 G 3.01	Diss by acids.
	1.570	<u>1.571</u>	1.578	.008	BANALSITE BaNa ₂ Al ₄ Si ₄ O ₁₆	41°	X = $\frac{c}{a}$ Y = $\frac{a}{b}$	ORTH	110, 001 good	White	H 6 G 3.06	---

1.554 ^ 1.585	1.571	<u>1.571</u>	1.576	.005	CLINOCHLORE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	Small r < v	Z ~ $\frac{c}{el}$ neg	MCL	001 perf	Green	H 3 G 2.66 F diff	Insol in acids.
1.555 ^	1.571	---	1.576	.005	WOODWARDITE Cu ₄ Al ₂ (SO ₄)(OH) ₁₂ · 3H ₂ O (?)	---	---	ORTH(?) spher	---	Greenish- blue	G 2.38	---
	1.562	<u>1.572</u>	1.585	.023	FERRARISITE Ca ₅ H ₂ (AsO ₄) ₄ ·9H ₂ O	83°	---	TCL	001 perf	Cols	G 2.63	Diss by HCl.
1.570 □ 1.575	1.572	<u>1.572</u>	1.594	.022	PREOBRAZHENSKITE Mg ₃ B ₁₁ O ₁₅ (OH) ₉	0-20°	X = $\frac{a}{Y}$ = $\frac{c}{c}$	ORTH	---	Lemon- yellow, cols	H 4.5-5 G 2.44	---
	1.562	(<u>1.572</u>)	1.583	.021	Unnamed hydrous calcium arsenate	~ 90°	Z = $\frac{b}{Y:c}$ = 8°	MCL	---	Cols	---	Opt char unk (Am. Mineral., 58, 561 (1973)).
	1.570	<u>1.572</u>	1.582	.012	FROLOVITE CaB ₂ (OH) ₈	75°	---	TCL	---	White	H 3.5 G 2.14	Diss by acids. Ext may be undulatory.
	1.572	(<u>1.572</u>)	1.582	.010	BAYERITE Al(OH) ₃	Small	---	MCL	---	White	---	Diss by hot alkalies.
	1.543	<u>1.575</u>	1.634	.091	METASIDERONATRITE Na ₂ Fe ⁺³ (SO ₄) ₂ (OH)·H ₂ O	60° (75+3°) r > v str	X = $\frac{a}{Y}$ = $\frac{b}{b}$	ORTH	001 mic 100, 010 perf	Golden- to straw- yellow	H 2.5 G 2.68	Diss by acids. Pleoc, X cols, Y light yellow, Z brownish-yellow.
1.554 ^	1.558	<u>1.575</u>	1.620	.062	CUPROCOPIAPITE CuFe ₄ (SO ₄) ₆ (OH) ₂ ·20H ₂ O	63° r > v str	X ~ $\frac{b}{el}$ cTv pos	TCL tab	010 perf	Greenish- yellow	H 2.5 G 2.1 F 5	Diss by acids. Pleoc in green, abs X and Z > Y.
	1.573	<u>1.575</u>	1.588	.015	AUGELITE Al ₂ (PO ₄)(OH) ₃	47°	X = $\frac{b}{Z:c}$ = 35°	MCL tab 001	110 perf 201 good	Cols	H 4.5-5 G 2.69 infus	Nearly insol in acids.
	1.570	<u>1.576</u>	1.614	.044	ANHYDRITE CaSO ₄	44° r < v	X = $\frac{a}{Z}$ = $\frac{c}{c}$	ORTH	010 perf 100 less so 001 good	Cols, bluish	H 3.5 G 2.98 F 3	Diss by HCl.
	1.573	<u>1.576</u>	1.579	.006	ILMAJOKITE (Na,Ce,Ba) ₂ TiSi ₃ O ₅ (OH) ₁₀ ·xH ₂ O	90°	---	MCL	2 at 72° perf	Bright yellow	H 1 G 2.20 fus	Dec by acids.
	1.554	<u>1.577</u>	1.618	.064	BONATTITE CuSO ₄ ·3H ₂ O	(75+4°)	---	MCL	010 dist	Pale blue	G 2.66 F 3	Sol in H ₂ O. Hydrates to chalcāthite. Tw pl 100 common.
	1.54	<u>1.578</u>	~1.60	.06	FICHELITE C ₁₉ H ₃₄	87°	opt pl 010 Z:c = 13°	MCL tab	001 perf 101 dist	White	H 1 G 1.03 F 46°C	Sol in ether. Tw pl 001. Dimethyl- isopropyl-perhydro- phenanthrene.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.588 ^v	1.575	<u>1.578</u>	1.584	.009	NIFONTOVITE $\text{Ca}_3\text{B}_6\text{O}_{12}(\text{OH})_{12} \cdot 2\text{H}_2\text{O}$	76° r > v str	el pos	MCL	Poor, el	Cols	H 3.5 G 2.36	Diss by acids. Anom interf colors.
	1.576	<u>1.579</u>	1.597	.031	COOKEITE (Chlorite grp) $\text{LiAl}_4(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	0-80° most ~ 50° (45+14°)	Z ~ c X edge el neg	MCL tab 001	001 mic	White, pink	H 2.5 G 2.67	Insol in acids. Ex-foliates when heated. Base shows 6 biax segments.
1.568 [^]	1.578	<u>1.579</u>	1.590	.012	GIBBSITE $\text{Al}(\text{OH})_3$	0-5° r < v or r > v	Y = b Z:c = -30°	MCL ps hex	001 perf	White, brownish	H 3 G 2.3-2.4 infus	Diss by hot concd H_2SO_4 or by hot alkalis.
1.588 ^v	1.578	<u>1.579</u>	1.583	.005	BAVENITE $\text{Ca}_4\text{Be}_2\text{Al}_2\text{Si}_9\text{O}_{26}(\text{OH})_2$	26-60°	X = c Z = b el neg	ORTH fib	100 good 001 fair	Cols, pink	H 5.5 G 2.71	Apparent tw pl 100.
1.591 ^v	1.543	<u>1.580</u>	1.617	.074	HAMBERGITE $\text{Be}_2\text{BO}_3(\text{OH},\text{F})$	90° r > v wk	X = a Y = b	ORTH pris c	010 perf 100 good	White	H 7.5 G 2.37 infus	Insol in acids except HF.
1.583 ^v	1.580	<u>1.580</u>	1.596	.016	NORDSTRANDITE $\text{Al}(\text{OH})_3$	Small	Z:c = 33° el neg	TCL	---	White	G 2.43	Poly tw.
	1.567	<u>1.581</u>	1.638	.071	KORNELITE $\text{Fe}_2(\text{SO}_4)_3 \cdot 7\text{H}_2\text{O}$	49-62° r > v perc	Z = b X:c = 17°	MCL pris	010 good	Pale rose, violet	G 2.3	Sol in H_2O . Poly tw pl 100.
	1.573	<u>1.581</u>	1.585	.012	SLAWSONITE (Feldspar grp) $(\text{Sr},\text{Ca})\text{Al}_2\text{Si}_2\text{O}_8$	82° r < v	Z = b X:a = 11°	MCL pris	001 good 100 fair	Light gray	H 5.5 G 3.12	---
1.570 [^]	---	<u>1.582</u>	1.631	>.05	BUKOVSKYITE $\text{Fe}_2(\text{AsO}_4)(\text{SO}_4)(\text{OH}) \cdot 7\text{H}_2\text{O}$	---	ext 18°	MCL(?) needles needles	---	Yellow to gray- green	G 2.34	Diss by HCl. Opt char unk.
1.570 [^]	1.577	<u>1.582</u>	1.595	.018	WAGNERITE $(\text{Mg},\text{Fe})_2(\text{PO}_4)\text{F}$	32° (64+14°) r > v perc	Y = b Z:c = -22°	MCL	100, 120, 010 imperf	Cols, flesh, yellow	H 5-5.5 G 3.23 F 4	Diss by acids. FeO 9.9, MnO 2.0%.
1.569 [^]	1.578	<u>1.582</u>	1.596	.018	SUDOITE (Chlorite grp) $\text{Mg}_2(\text{Al},\text{Fe})_3\text{Si}_3\text{AlO}_{10}(\text{OH})_8$	55°	Z:c = 7° el neg	MCL	001 perf	White	G 2.64	Dioctahedral chlorite.
	1.582	<u>1.582</u>	1.592	.010	XONOTLITE $\text{Ca}_6\text{Si}_6\text{O}_{17}(\text{OH})_2$	0-5°	X = b Z = c el pos	MCL and TCL fib	010 perf	Cols, pink	H 6 G 2.69 F 2.5	Dec by HCl with sepn of silica.

	1.550	<u>1.583</u>	1.641	.091	AJOITE (K,Na)Cu ₇ AlSi ₉ O ₂₄ (OH) ₆ · 3H ₂ O	80° (76±3°)	Z':c on 010 = 15°	TCL	---	Bluish- green	G 2.96	Dec by acids. Pleoc, X light bluish-green, Y and Z brilliant bluish-green.
1.580 ^	1.580	<u>1.583</u>	1.602	.022	NORDSTRANDITE Al(OH) ₃	24° (44±16°) r < v perc	el neg	TCL	110 perf	Cols, white	H ~ 3 G 2.42	---
	1.580	<u>1.583</u>	1.590	.010	WHITEITE Ca(Fe ⁺² ,Mn)Mg ₂ (PO ₄) ₄ Al ₂ (OH) ₂ ·8H ₂ O	40-50°	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	MCL	---	Tan	G 2.58	---
v 1.593	1.580	(<u>1.583</u>)	1.590	.010	MISERITE K(Ca,Ce) ₄ Si ₅ O ₁₃ (OH) ₃	60-78°	X = $\frac{c}{a}$ Z = $\frac{a}{b}$ el neg	TCL fib	100 good 010 imperf	Pink to violet	H 2.5 G 2.84	Insol in acids.
1.571 v 1.601	1.585	<u>1.585</u>	1.593	.008	CLINOCHLORE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	0-15° r < v	Z ~ $\frac{c}{a}$ el neg	MCL	001 perf	Cols to green	H 3 G 2.72 F 5	Insol in acids. FeO 3.3, Fe ₂ O ₃ 2.1%.
	1.563	<u>1.586</u>	1.619	.056	SIGLOITE (Fe ⁺² ,Fe ⁺³)Al ₂ (PO ₄) ₂ (O,OH) ₂ ·8H ₂ O	76° (81±4°) r < v str	---	TCL	010 perf 001 good	Straw- yellow	H 3 G 2.35	Diss by acids. Oxidation product of childrenite.
	1.584	<u>1.586</u>	1.600	.016	EAKERITE Ca ₂ SnAl ₂ Si ₆ O ₁₈ (OH) ₂ · 2H ₂ O	35° r > v str	X = $\frac{b}{a}$ Y:c = $\frac{a}{b}$ 23.5°	MCL	Conch	Cols to white	H 5.5 G 2.93	Insol in acids.
v 1.593	1.581	<u>1.586</u>	1.596	.015	SCHOLZITE CaZn ₂ (PO ₄) ₂ ·2H ₂ O	70° r < v	X = $\frac{a}{b}$ Y = $\frac{b}{a}$	ORTH	100 fair	Cols	H 3-3.5 G 3.11	Diss by acids.
1.579 ^	1.584	<u>1.588</u>	1.598	.014	COOKEITE (Chlorite grp) LiAl ₄ (Si ₃ Al) ₁₀ (OH) ₈	60°	Z ~ $\frac{c}{a}$ X edge el neg	MCL tab 001	001 mic	White, pink	H 2.5 G 2.61	Insol in acids. Ex- foliates when heated. Base shows 6 biax segments.
1.579 ^	1.586	<u>1.588</u>	1.593	.007	BAVENITE Ca ₄ Be ₂ Al ₂ Si ₉ O ₂₆ (OH) ₂	60°	X = $\frac{c}{a}$ Z = $\frac{b}{a}$ el neg	ORTH fib	100 good 001 fair	Cols, pink	H 4.5-5 G 2.73	Apparent tw pl 100.
	1.587	<u>1.588</u>	1.603	.016	PARASCHOLZITE CaZn ₂ (PO ₄) ₂ ·2H ₂ O	25° r > v	X = $\frac{b}{a}$ Z:c = $\frac{a}{b}$ = 13°	MCL	Parting 100	Cols to white	H 4 G 3.12	Tw on 100.
1.571 ^	1.579	<u>1.589</u>	1.609	.030	MANGANESE-HOERNESITE (Mn,Mg) ₃ (AsO ₄) ₂ ·8H ₂ O	65-70° (71±8°)	X = $\frac{b}{a}$ Z:c = $\frac{a}{b}$ = 31° el pos	MCL acic	010 perf	White	G 2.76	Diss by acids.
	1.580	<u>1.589</u>	1.605	.025	HEIDORNITE Na ₂ Ca ₃ B ₅ O ₈ (OH) ₂ (SO ₄) ₂ Cl	63-77° r < v	Y = $\frac{b}{a}$ Z:a = $\frac{a}{b}$ = 23°	MCL spear- like	001 perf	Cols	H 4-5 G 2.75	Diss by acids.
1.589 neg v 1.599	1.584	<u>1.589</u>	1.594	.010	CELSIAN (Feldspar grp) BaAl ₂ Si ₂ O ₈	86°	Y = $\frac{b}{a}$ Z:a = $\frac{a}{b}$ = 28°	MCL	001 perf 010 good	Cols	H 6 G 3.37 infus	Gel with HCl.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
[]	1.583	<u>1.589</u>	(1.601)	(.018)	SUDOITE (Chlorite grp) $Mg_2(Al, Fe^{+3})_3Si_3AlO_{10}(OH)_8$	70°	---	MCL	001 good	Pale green	G 2.68	MgO 14.1, Al ₂ O ₃ 35.7, Fe ₂ O ₃ 2.7, FeO 0.2, MnO 0.3%.
1.533 ^ 1.623	1.558	<u>1.590</u>	1.629	.071	SZOMOLNOKITE (Kieserite grp) (Fe,Mg)SO ₄ ·H ₂ O	Large r > v	Y = <u>b</u>	MCL	---	Yellow	H 2.5 G 2.8 fus	Slowly sol in H ₂ O. Fe:Mg = 3:2.
	1.587	<u>1.590</u>	1.597	.010	BULTFONTEINITE Ca ₂ SiO ₂ (OH,F) ₄	70° r > v	Z:c on 010 = 28° 100 = 47°	TCL spher	010, 001 fairly good	Pink	H 4.5 G 2.73	Gel with HCl. Poly tw.
1.580 ^	1.560	<u>1.591</u>	1.631	.071	HAMBERGITE Be ₂ BO ₃ (OH,F)	87° r > v wk	X = <u>a</u> Y = <u>b</u>	ORTH pris <u>c</u>	010 perf 100 good	White	H 7.5 G 2.35 infus	Insol in acids except HF.
[]	1.590	<u>1.592</u>	1.627	.037	CATAPLEIITE (Na ₂ ,Ca)ZrSi ₃ O ₉ ·2H ₂ O	Small r < v wk	---	HEX	1010 perf	Yellow, brown	H 5.6 G 2.75 F 3	Gel with acids. Complex tw. Uniax when heated to 200°C.
	1.586	<u>1.592</u>	1.614	.028	COLEMANITE Ca ₂ B ₆ O ₁₁ ·5H ₂ O	56° r > v wk	X = <u>b</u> Y:c = -6°	MCL	010 perf 001 dist	Cols	H 4.5 G 2.42 F 1.5	Diss by hot HCl.
	1.588	<u>1.592</u>	1.598	.010	ALTHAUSITE Mg ₂ (PO ₄)(OH,F,O)	~ 70°	X = <u>c</u> Y = <u>b</u> el neg	ORTH mass	010 perf 100 dist	Gray	H 3.5-4 G 2.97	---
1.586 ^	1.588	<u>1.593</u>	1.602	.014	SCHOLZITE CaZn ₂ (PO ₄) ₂ ·2H ₂ O	65° r < v	X = <u>a</u> Y = <u>b</u>	ORTH	100 fair	Cols	H 3-3.5 G 3.11	Diss by acids.
(1.583) ^	1.586	<u>1.593</u>	1.600	.014	MISERITE K(Ca,Ce) ₄ Si ₅ O ₁₃ (OH) ₃	52-78°	X = <u>c</u> Z = <u>a</u> el neg	TCL fib	100 good 010 imperf	Pink to violet	H 2.5 G 2.84	Insol in acids.
	1.562	<u>1.595</u>	1.632	.070	SZIMIKITE MnSO ₄ ·H ₂ O	~ 90°	Z = <u>b</u>	MCL fib or platy	One perf	White to rose	Soft G 3.15 infus	Sol in H ₂ O.
	1.590	<u>1.595</u>	1.602	.012	CUSPIDINE Ca ₄ Si ₂ O ₇ (F,OH) ₂	62° (81+19°) r > v	Y = <u>b</u> Z:c = 7° disp perc	MCL spear- shaped	001 dist	Pale rose, cols	H 5.5-6 G 2.86-3.05 F diff	Gel with acids. Poly tw.
1.600 []	1.594	<u>1.595</u>	1.599	.005	COESITE SiO ₂	54-64° r < v wk	X = <u>b</u> Z:c = 5°	MCL ps hex	Subconch	Cols	G 2.92 infus	Insol in acids.

1.592

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1.607

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1.589
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1.594	<u>1.595</u>	1.598	.004	FOSHAGITE $\text{Ca}_4\text{Si}_3\text{O}_9(\text{OH})_2$	---	el pos	TCL fib	---	White	H 3 G 2.36 infus	Gel with acids.
1.590	<u>1.596</u>	1.616	.026	PANASQUEIRAITE $\text{CaMg}(\text{PO}_4)(\text{OH},\text{F})$	51°	$Z = \frac{b}{c} = 22^\circ$	MCL	010 poor	Pink	H 5 G 3.27	---
1.577	<u>1.597</u>	1.616	.039	JOHANNITE $\text{Cu}(\text{UO}_2)_2(\text{SO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	~ 90° $r < v$ str	$X \sim \frac{b}{c}$ disp str	TCL pris tab 001	100 good	Emerald- to apple- green	H 2-2.5 G 3.32 infus	Diss by acids. Poly tw. Pleoc str, X cols, Y pale yellow, Z greenish- or canary-yellow.
1.593	<u>1.597</u>	1.619	.026	ISOKITE $\text{CaMg}(\text{PO}_4)\text{F}$	39-51° $r > v$	$Z = \frac{b}{c} = 32^\circ$ el neg	MCL fib	010 very good	Cols to buff	H 5 G 3.27	Diss by acids.
1.596	<u>1.598</u>	1.632	.036	PETARASITE $\text{Na}_5\text{Zr}_2\text{Si}_6\text{O}_{18}(\text{Cl},\text{OH}) \cdot 2\text{H}_2\text{O}$	29° $r < v$ wk	$X = \frac{b}{c}$ $Z:c = 41.5^\circ$	MCL	110 perf 001 dist	Greenish- yellow	H 5-5.5 G 2.88	Pleoc, X cols, Y and Z pale greenish- yellow.
1.580	<u>1.598</u>	1.627	.047	VIVIANITE $\text{Fe}_3(\text{PO}_4)_2 \cdot 8\text{H}_2\text{O}$	Large $r < v$ wk	$X = \frac{b}{c}$ $Z:c = 29^\circ$ disp str el clv pos	MCL	010 perf	Pale blue to bluish- green	H 1.5-2 G 2.66 F 1.5	Diss by HCl. Nearly cols fresh, oxidizes to deep blue, then pleoc, X deep blue, Y pale blue, Z olive to brown.
1.595	<u>1.598</u>	1.610	.015	KOSTYLEVITE $\text{K}_2\text{ZrSi}_3\text{O}_9 \cdot \text{H}_2\text{O}$	48°	$X = \frac{b}{c}$ $Z:c = 45^\circ$	MCL	110 perf	Cols	G 2.74	Dec by HCl.
1.598	<u>1.598</u>	1.605	.007	TAKOVITE $\text{Ni}_6\text{Al}_2(\text{OH})_{16}(\text{CO}_3,\text{OH}) \cdot 4\text{H}_2\text{O}$	Small	el pos	TRIG spher	---	Blue- green	---	Diss by HCl.
1.597	<u>1.599</u>	1.615	.018	AMESITE (Kaolinite-serpentine grp) $\text{Mg}_2\text{Al}(\text{SiAl})\text{O}_5(\text{OH})_4$	18° $r < v$	$Z:c = 0-20^\circ$ el neg	MCL or TCL ps hex plates	001 mic	Pale green	H 2-3 G 2.8 infus	Slowly dec by HCl. FeO ~ 8%.
1.593	<u>1.599</u>	1.608	.015	CELSIAN (Feldspar grp) $\text{BaAl}_2\text{Si}_2\text{O}_8$	85°	$Y = \frac{b}{a}$ $X:a \text{ on } 010 = 65^\circ$	MCL	001 perf 010 good	Cols	H 6 G 3.35 infus	Gel with HCl. Fe_2O_3 2.2, BaO 38.6, K_2O 0.5%.
1.593	---	1.608	.015	Unnamed silicate of K, Ca, Zr	---	---	MCL ps hex	---	Brown	---	(Am. Mineral. 50, 533 (1965)).
1.590	<u>1.600</u>	1.634	.044	HAIDINGERITE $\text{CaHAsO}_4 \cdot \text{H}_2\text{O}$	58° $r > v$ wk	$X = \frac{b}{c}$ $Z = \frac{c}{a}$ el pos	ORTH tab 010	010 perf	Cols	H 2-2.5 G 2.85 F 2.5	Diss by acids.
1.595	<u>1.600</u>	1.631	.036	PETARASITE $\text{Na}_5\text{Zr}_2\text{Si}_6\text{O}_{18}(\text{Cl},\text{OH}) \cdot 2\text{H}_2\text{O}$	43° $r < v$	$X = \frac{b}{c}$ $Z:c = 41.5^\circ$	MCL	110 perf 010 very good	Greenish- yellow	H 5-5.5 G 2.88	Pleoc, X cols, Y and Z pale greenish- yellow.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.597 □ 1.61	1.595	<u>1.60</u>	1.628	.033	CEBOLLITE (Ca,Na ₂) ₄ Al ₂ Si ₃ O ₁₂ (OH) ₂ (?)	58°	---	ORTH(?) fib	---	White	H 5 G 2.96 F 5	Gel with acids. Alteration product of Melilite.
	1.596	<u>1.600</u>	1.618	.022	CHIAVENNITE CaMnBe ₂ Si ₅ O ₁₃ (OH) ₂ · 2H ₂ O	50° r > v	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el neg	ORTH	001 perf 100, 010 good	Orange	H ~ 3 G 2.24	---
	---	<u>~1.6</u>	---	.014	MANANDOITE (Chlorite grp) LiAl ₄ BSi ₃ O ₁₀ (OH) ₈	17°	Z ~ c el cTv neg	MCL ps hex	001 mic	Col's	G 2.89 F easy	Dec by H ₂ SO ₄ . Basal section divided into 6 biax segments.
1.585 ∧ 1.620	1.599	<u>1.601</u>	1.609	.010	CLINOCHLORE (Chlorite grp) (Mg,Fe) ₅ Al(Si ₃ Al) ₁₀ (OH) ₈	12-22° r < v	Z ~ c el cTv neg	MCL	001 perf	Col's to green	H 3 G 2.78 F 4	Nearly insol in acids. FeO 11.4, Fe ₂ O ₃ 4.3%.
	1.582	<u>1.602</u>	1.629	.047	GOLDICHITE KFe(SO ₄) ₂ ·4H ₂ O	82°	X = $\frac{b}{c}$ Z:c = 11°	MCL	---	Pale green	G 2.43	Sol in H ₂ O.
	1.579	<u>1.603</u>	1.629	.050	METAVIVIANITE Fe ⁺² _{3-x} Fe ⁺³ _x (PO ₄) ₂ (OH) _x ·(8-x)H ₂ O	85°	X ⊥ 110	TCL pris el c	110 perf	Leek- green	G 2.69 calc F 1.5	Diss by HCl. Pleoc, X light blue to blue- green, Y yellow-green to light green, Z yellow, abs X > Z > Y.
1.617 ∨	1.593	<u>1.603</u>	1.623	.030	CHONDRODITE (Humite grp) (Mg,Fe) ₅ (SiO ₄) ₂ (F,OH) ₂	71° r > v wk	Z = $\frac{b}{c}$ X:c = 27°	MCL	100 poor	Yellow to red	H 6 G 3.18 infus	Gel with acids. Poly tw 001. Pleoc, X yellow, Y and Z nearly cols. FeO 2.8, F 7.9%.
1.608 ∨	1.594	<u>1.603</u>	1.615	.021	NATROMONTEBRASITE (Na,Li)AlPO ₄ (OH,F)	Very large	Z:tw lam = 29°	TCL	2 dist	White	H 5.5 G 3.09 F easy	Diff diss by acids. Poly tw in 2 directions.
	1.577	---	1.630	.053	GUNNINGITE (Zn,Mn)SO ₄ ·H ₂ O	---	---	MCL fine- grained	---	White	H 2.5 G 3.2 infus	Sol in H ₂ O.
1.636 ∨	1.599	<u>1.604</u>	1.638	.039	PECTOLITE NaCa ₂ Si ₃ O ₈ (OH)	59° (43+8°) r > v wk	---	TCL acic <u>b</u>	100, 001 perf	Col's	H 4.5-5 G 2.90 F 2	Partly dec by HCl.
	1.598	<u>1.604</u>	1.626	.028	METASCHODERITE Al ₂ (PO ₄)(VO ₄)·6H ₂ O	(56+10°)	Z = $\frac{b}{c}$ Y:c = -20°	MCL	---	Yellow- orange	---	---

1.623 v	1.580	<u>1.605</u>	1.644	.064	EUCHLORINE (K,Na) ₈ Cu ₉ (SO ₄) ₁₀ (OH) ₆ (?)	Med large (79+4°)	---	ORTH tab	2 clv	Emerald- green	---	Partly diss by H ₂ O. Pleoc, X pale green, Y green, Z yellow- green.
	1.593	<u>1.605</u>	1.627	.034	FUKALITE Ca ₄ Si ₂ O ₆ (OH) ₂ (CO ₃)	(74+7°)	---	ORTH	---	---	H ~ 4 G 2.77	Eff with acids.
	1.600	<u>1.605</u>	1.613	.013	HYDROPHILITE CaCl ₂ (?)	Med	---	ORTH ps tet	Pris, perf	White	G 2.2 F 2	Sol in H ₂ O, deliq. Lam tw pf 110. Data on synth compd.
	1.598	<u>1.606</u>	1.630	.032	HUMITE (Humite grp) Mg ₇ (SiO ₄) ₃ (F,OH) ₂	59° r > v wk	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH tab 010	001 perf	Cols	H 6 G 3.20 infus	Gel with acids. Data for synth Mg-F compd.
	1.596	<u>1.606</u>	1.621	.025	SCAWTITE Ca ₇ Si ₆ (CO ₃) ₀ 18•2H ₂ O	74°	Y = $\frac{b}{a}$ Z:a = 30°	MCL	001 perf 010 fair	Cols	H 4.5-5 G 2.74 infus	Eff and gel with acids.
	1.600	<u>1.606</u>	1.614	.014	LATIUMITE (Ca,K) ₈ (Al,Mg,Fe) (Si,Al) ₁₀ O ₂₅ (SO ₄)	83-90° r > v dist	Z = $\frac{b}{c}$ X:c = 16- 28°	MCL tab	100 perf	White	G 2.93	---
	1.598 1.624 v	<u>1.607</u>	1.631	.044	VIVIANITE Fe ₃ (PO ₄) ₂ •8H ₂ O	Large r < v wk	X = $\frac{b}{c}$ Z:c = 27° disp str el clv pos	MCL	010 perf	Blue	H 1.5-2 G 2.66 F 1.5	Diss by HCl. Pleoc, X dark blue, Y light blue, Z nearly cols. FeO 29.3, Fe ₂ O ₃ 14.0%.
	1.602	<u>1.607</u>	1.615	.013	CRANDALLITE (?) (Crandallite grp) CaAl ₃ (PO ₄) ₂ (OH) ₅ •H ₂ O	70-75°	---	TCL(?) ps trig	Basal, perf	White	H 3 G 2.50 F 2.5	Diss with diff in HCl (Am. Mineral., 48, 1144 (1963)).
	1.600	---	1.620	.020	ALDZHANITE CaMgB ₂ O ₄ Cl•7H ₂ O (?)	---	---	ORTH dipyra- midal	---	Cols to rose	G 2.21	Opt char unk.
	1.600	<u>1.608</u>	1.645	.045	CHURCHITE YPO ₄ •2H ₂ O	Med disp dist	X = $\frac{b}{c}$ Z:c = 35°	MCL laths el c	010, 100, 001	White, yellow	H 3 G 3.26 infus	Diss by hot acids.
1.603 1.624 v	1.605	<u>1.608</u>	1.612	.007	IMANDRITE Na ₁₂ Ca ₃ Fe ⁺³ ₂ Si ₁₂ O ₃₆	75°	---	ORTH	---	Honey- yellow	H 4 G 2.93	---
	1.597	<u>1.608</u>	1.632	.035	MONTEBRASITE (Li,Na)AlPO ₄ (OH,F)	Large r < v	---	TCL	100 perf 100 good 011 dist	White	H 6 G 3.04 F 2	Diss by H ₂ SO ₄ . Poly tw in 2 directions at 90°. Li ₂ O 7.5, Na ₂ O 3.5, F 1.4%.
	1.602	<u>1.609</u>	1.621	.019	BRAZILIANITE NaAl ₃ (PO ₄) ₂ (OH) ₄	75° r < v wk	Y = $\frac{b}{c}$ X:c = -20°	MCL	010 good fr conch	Yellow	H 5.5 G 2.98 F 4.5	Insol in HCl.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_Z$ ($2V_Z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.618	1.604	<u>1.609</u>	1.615	.011	URALBORITE $\text{CaB}_2\text{O}_2(\text{OH})_4$	85° $r > v$ str	---	MCL pris	Parallel to el dist	Cols	H 4 G 2.60	Diss by HCl. Anom interf colors in blue and brown.
	1.607	<u>1.610</u>	1.618	.011	TOPAZ $\text{Al}_2\text{SiO}_4(\text{F},\text{OH})_2$	67° $r > v$ dist	$X = \frac{a}{c}$ $Z = \frac{c}{a}$ el pos el clv neg	ORTH el \underline{c}	001 perf	Cols, yellow	H 8 G 3.57 infus	Insol in acids. F 20.4%.
	1.607	<u>1.610</u>	1.616	.009	BUCHWALDITE NaCaPO_4	---	el pos	ORTH	One clv	Cols	H < 3 G 3.21	Reported as opt neg.
	---	<u>~1.61</u>	---	wk	HARBORTITE $\text{Al}_3(\text{PO}_4)_2(\text{OH})_3 \cdot 3\text{H}_2\text{O}$ (?)	---	---	Spher	---	White to brown	H 5-5.5 G 2.80	---
	1.610	<u>1.610</u>	1.611	.001	FOGGITE $\text{CaAl}(\text{PO}_4)(\text{OH})_2 \cdot \text{H}_2\text{O}$	40-45°	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH plates	010 perf 100 good	Cols, white	H 4 G 2.78	---
	1.610	<u>1.611</u>	1.654	.044	CYCLOWOLLASTONITE CaSiO_3	Very small	---	TCL	---	Cols	H 5 G 2.91 F 4	Dec by acids. Also called Pseudowollastonite.
	1.608	<u>1.612</u>	1.621	.013	TUHUALITE $(\text{Na},\text{K})\text{Fe}^{+2}\text{Fe}^{+3}\text{Si}_6\text{O}_{15}$	60-70° $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{a}$ el pos	ORTH	100, 010, 001 good	Dark blue to black	H 3-4 G 2.89	Insol in acids. Pleoc str, X pale pink, Y violet or lavender, Z deep purplish-blue.
1.621	1.602	<u>1.613</u>	1.649	.047	ANAPAITE $\text{Ca}_2\text{Fe}(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	54° $r > v$ perc	on 100, ext:c = 15°	TCL radiating	001 perf 010 dist	Greenish-white	H 3.5 G 2.81	Diss by acids.
	1.609	<u>1.613</u>	1.619	.010	STOKESITE $\text{CaSnSi}_3\text{O}_9 \cdot 2\text{H}_2\text{O}$	70° $r < v$	$Y = \frac{b}{c}$ $X = \frac{c}{b}$	ORTH pyram	110 perf 010 imperf	Cols	H 6 G 3.19	---
	1.587	<u>1.615</u>	1.640	.053	MONETITE CaHPO_4	Large $r > v$ wk	---	TCL rhombs	3 pinacoidal, indist	Yellowish-white	H 3.5 G 2.93 infus	Diss by acids.
1.649	1.608	<u>1.615</u>	1.630	.022	MAGNIOTRIPLITE $(\text{Mg},\text{Fe},\text{Mn})_2\text{PO}_4(\text{F},\text{OH})$	66° $r > v$ perc	$Y = \frac{b}{c}$ $Z:\underline{c} = 53^\circ$	MCL	100 perf	Yellow	H 5.5 G 3.47 F 3	Diss by acids. Pleoc, X pale yellow, Y and Z deep yellow. MgO 27.9, FeO 13.4, MnO 7.7%.

	1.588	<u>1.617</u>	1.655	.067	CYANOTRICHITE $\text{Cu}_4\text{Al}_2(\text{SO}_4)(\text{OH})_{12}\cdot 2\text{H}_2\text{O}$	82° $r < v$ str	$Z = c$ el pos	ORTH acid <u>c</u>	---	Bright blue	G 2.8 F 3	Diss by acids. Pleoc str, X nearly cols, Y pale blue, Z bright blue.
1.603 ∇ 1.632	1.607	<u>1.617</u>	1.639	.032	CHONDRODITE (Humite grp) $(\text{Mg}, \text{Fe})_5(\text{SiO}_4)_2(\text{F}, \text{OH})_2$	80° (69+8°) $r > v$ wk	$Z = b$ $X:c = 28^\circ$	MCL	100 poor	Yellow to red	H 6 G 3.26 infus	Gel with acids. Poly tw 001. Pleoc, X yellow, Y and Z pale yellow. FeO 5.0, Fe ₂ O ₃ 0.8%.
	1.614	<u>1.617</u>	1.636	.022	HEMIMORPHITE $\text{Zn}_4\text{Si}_2\text{O}_7(\text{OH})_2\cdot \text{H}_2\text{O}$	46° $r > v$ str	$X = b$ $Z = c$	ORTH el <u>c</u>	110 perf 101 less so	Cols	H 5 G 3.45 F 6	Gel with acids.
	1.616	<u>1.617</u>	1.622	.006	KURUMSAKITE $(\text{Zn}, \text{Ni}, \text{Cu})_8\text{Al}_8\text{V}_2\text{Si}_5\text{O}_{35}\cdot 27\text{H}_2\text{O} (?)$	35°	---	fib	---	Greenish- to bright yellow	G 4.03	---
1.627 ∇	1.618	<u>1.618</u>	1.670	.052	DIADOCHITE $\text{Fe}_2(\text{PO}_4)(\text{SO}_4)(\text{OH})\cdot 5\text{H}_2\text{O}$	Small $r > v$ str	---	TCL u mass	Uneven to conch	Yellow to brown	H 3-4 G 2.0-2.4 fus	Diss by acids.
1.683 ∇	1.606	(<u>1.618</u>)	1.638	.032	KOETTIGITE $(\text{Zn}, \text{Mg})_3(\text{AsO}_4)_2\cdot 8\text{H}_2\text{O}$	Large $r < v$	$X = b$ el pos	MCL fib <u>c</u>	010 perf	White	H 3 G 2.9 F 3	Diss by acids. ZnO 27.7, MgO 7.2%.
1.643 ∇	1.608	<u>1.618</u>	1.636	.028	CLINOHUMITE (Humite grp) $\text{Mg}_9(\text{SiO}_4)_4(\text{F}, \text{OH})_2$	59° (74+8°) $r < v$	$Z = b$ $X:c = 9^\circ$	MCL	---	Cols	H 6 G 3.1 infus	Gel with acid. Data for synth F end- member.
1.629 ∇	1.610	<u>1.618</u>	1.635	.025	PREHNITE $\text{Ca}_2\text{Al}_2\text{Si}_3\text{O}_{10}(\text{OH})_2$	69° $r > v$	$X = a$ $Y = b$	ORTH	001 good	White, green	H 6 G 2.87 F 2	Slowly dec by HCl with sepn of silica. Fe ₂ O ₃ 1.5%.
1.634 ∇	1.613	<u>1.618</u>	1.635	.022	PARGASITE (Amphibole grp) $\text{NaCa}_2\text{Mg}_4\text{Al}(\text{Si}_6\text{Al}_2)\text{O}_{22}\text{O}_{22}(\text{OH}, \text{F})_2$	61°	$Y = b$ $Z:c = 26^\circ$	MCL pris <u>c</u>	110 perf at 124°	Cols to brown	H 6 G 3.07	Insol in acids. Al ₂ O ₃ 11.1, Fe ₂ O ₃ 0.7, FeO 1.7, MgO 20.6, CaO 12.5, Na ₂ O 2.5, K ₂ O 1.2%.
1.610 ∇ 1.631	1.616	<u>1.618</u>	1.625	.009	TOPAZ $\text{Al}_2\text{SiO}_4(\text{F}, \text{OH})_2$	61° $r > v$ dist	$X = a$ $Z = c$ el pos el clv neg	ORTH el <u>c</u>	001 perf	Cols, blue, yellow	H 8 G 3.55 infus	Insol in acids. F 17.2, H ₂ O 1.6%.
	1.616	<u>1.619</u>	1.631	.015	AFWILLITE $\text{Ca}_3\text{Si}_2\text{O}_4(\text{OH})_6$	55° $r < v$ perc	$Y = b$ $X:c = 30^\circ$ disp str	MCL pris <u>b</u>	001 perf 100 good	Cols	H 4 G 2.62 F diff	Diss by HCl, gel.
	1.61	<u>1.62</u>	1.71	.10	BISBEEITE $\text{CuSiO}_3\cdot \text{H}_2\text{O} (?)$	Small	el pos	ORTH thin laths	---	Bluish- white	Soft	Abs Z > (X,Y). Status of species in doubt.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
154	1.590	<u>1.620</u>	1.648	.058	NICKELBISCHOFITE $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$	87°	$Y = b$ $X:c = 8^\circ$	MCL	001 perf	Emerald-green	Soft	Sol H_2O , deliq. Pleoc; X pale green, Z green.
	1.60	<u>1.62</u>	1.65	.05	CHAVESITE Hydrous phosphate of Ca, Mn	Large	ext:tw pl ~ 30°	TCL	2 good ~ 90°	Cols	H ~ 3	Poly tw (Am. Mineral., 43, 1148 (1985)).
	1.605	(<u>1.620</u>)	1.632	.027	CUMINGTONITE (Amphibole grp) $(\text{Mg,Fe})_7\text{Si}_8\text{O}_{22}(\text{OH,F})_2$	89°	$Z:c = 14^\circ$	MCL	110 perf at 125°	---	H 4-4.5 G 3.04	FeO 6.5, MnO 0.2, MgO 28.1, CaO 1.3, Na ₂ O 2.4%.
	1.612	<u>1.620</u>	1.648	.036	TURQUOISE $\text{CuAl}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 5\text{H}_2\text{O}$	40° (57+7°) $r < v$ str	ext on 110 = 12° 001 = 5° 010 = 34° disp dist	TCL u mass	001 perf 010 good	Sky-blue, bluish-green	H 5 G 2.84 infus	Diss by acids only after gentle heating. Pleoc, X cols, Z pale bluish.
	1.611	<u>1.620</u>	1.645	.034	METAHAIWEEITE $\text{Ca}(\text{UO}_2)_2\text{Si}_6\text{O}_{15} \cdot x\text{H}_2\text{O}$	(63+7°)	---	---	---	Yellow to greenish-yellow	---	---
	1.601 ◇ 1.630	<u>1.620</u>	1.625	.005	CHAMOSITE (Chlorite grp) $(\text{Fe,Mg,Al})_6(\text{Si,Al})_4\text{O}_{10}(\text{OH})_8$	5° $r < v$	$Z \sim c$	MCL	001 perf	Green	H 3 G 2.90 F 3.5	Nearly insol in acids. FeO 19.1, Fe ₂ O ₃ 6.6%.
	1.627	<u>1.621</u>	1.666	.073	TINAKSITE $\text{K}_2\text{Na}(\text{Ca,Mn})_2\text{TiSi}_7\text{O}_{19}(\text{OH})$	76° disp str	---	TCL pris	010 perf 110 imperf	Pale yellow to pink	H 6 G 2.82-2.90	Insol in acids. Pleoc, X and Y cols, Z pale orange-yellow.
	1.613 ^	<u>1.621</u>	1.628	.012	STOKESITE $\text{CaSnSi}_3\text{O}_9 \cdot 2\text{H}_2\text{O}$	70° $r < v$	$X = c$ $Y = b$	ORTH pyram	110 perf 010 imperf	Cols	H 6 G 3.19	---
	1.647	<u>1.622</u>	1.658	.040	VESELYITE $(\text{Cu,Zn})_3(\text{PO}_4)(\text{OH})_3 \cdot 2\text{H}_2\text{O}$	39° $r < v$ st	$Y = b$ $Z:c = 36^\circ$	MCL pris	001, 110	Dark bluish-green	H 3.5 G 3.09 fus	Diss by acids. Pleoc.
	1.590 ^	<u>1.623</u>	1.663	.072	SZOMOLNOKITE (Kieserite grp) $\text{FeSO}_4 \cdot \text{H}_2\text{O}$	80° $r > v$ str	$Y = b$ $X:c = -26^\circ$	MCL pyram	Conch to uneven	Yellow to brown	H 2.5 G 3.05 fus	Slowly sol in H_2O .

1.606 ^ 1.638	1.607	<u>1.623</u>	1.643	.036	HUMITE (Humite grp) $(\text{Mg,Fe})_7(\text{SiO}_4)_3(\text{OH,F})_2$	81° r > v wk	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH tab 010	001 perf	Yellow, brown, cols	H 6 G 3.24 infus	Gel with acids. Pleoc, X and Z pale yellow, Y nearly cols. FeO 5.4, Fe ₂ O ₃ 0.5, F 5.4%.
1.608 ^	1.614	<u>1.624</u>	1.644	.030	MONTEBRASITE LiAlPO ₄ (OH,F)	70° r > v str	---	TCL	100 perf 110 good	White	H 6 G 3.01 F 2	Diss by H ₂ SO ₄ . Poly tw in 2 directions. Li ₂ O 10.1, Na ₂ O 0.14%.
1.607 ^	1.600	<u>1.624</u>	1.650	.050	VIVIANITE Fe ₃ (PO ₄) ₂ ·8H ₂ O	Large r < v wk	X = $\frac{b}{c}$ Z:c = 27° disp str	MCL	010 perf	Blue	H 2 G 2.66 F 1.5	Diss by HCl. Pleoc, X dark blue, Y light blue, Z nearly cols. FeO 33.0, Fe ₂ O ₃ 4.1%.
1.744 ^	1.622	<u>1.624</u>	1.642	.020	MANSFIELDITE (Variscite grp) Al(AsO ₄)·2H ₂ O	30° r > v str	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH	201 imperf	White to green	H 3.5-4 G 3.03	Diss by acids.
	1.622	<u>1.624</u>	1.633	.011	URANOPILITE (UO ₂) ₆ (SO ₄)(OH) ₁₀ ·12H ₂ O	Small to large r > v str	Z = $\frac{b}{c}$ Y:c = 20°	MCL	010 perf	Bright yellow, straw yellow	Soft G 3.96	Diss by acids. Fluor bright yellow in UV. Pleoc, X cols, Y and Z yellow.
1.637 ^	1.622	<u>1.624</u>	1.631	.009	CELESTITE (Barite grp) SrSO ₄	51° r < v	Y = $\frac{b}{c}$ Z = $\frac{c}{a}$	ORTH Tab 001	001 perf 210 good	Cols to pale blue	H 3-3.5 G 3.96 F 3	Slightly sol in acids.
	1.614	<u>1.625</u>	1.637	.023	PARAHOPEITE Zn ₃ (PO ₄) ₂ ·4H ₂ O	~ 90° r < v perc	X ~ $\frac{a}{b}$ Y:c = 30° on 100	TCL	010 perf	Cols	H 3.5-4 G 3.31 F easy	Diss by acids. Poly tw pl common 100.
	---	<u>1.625</u>	---	.007	SMOLIANINOVITE (Co,Ni,Mg,Ca) ₃ (Fe ⁺³ ,Al) ₂ (AsO ₄) ₄ ·11H ₂ O	---	el pos	ORTH fib	---	Yellow	H 2 G 2.46	Opt sign unk.
	1.623	<u>1.626</u>	1.631	.008	UDUMINELITE Ca ₃ Al ₈ (PO ₄) ₆ (OH) ₄ O ₄	77°	---	ORTH acic	110	White	---	---
	1.618	(~1.627)	1.642	.024	FLUCKITE CaMnH ₂ (AsO ₄) ₂ ·2H ₂ O	Large	---	TCL	010 perf 100 easy	Deep to pale rose	H 3.5-4 G 3.05	Diss by acids.
1.618 ^	1.615	<u>1.627</u>	1.670	.055	DIADOCHITE Fe ₂ (PO ₄)(SO ₄)(OH)·5H ₂ O	Small (57+5°) r < v str	---	TCL u mass	Uneven to conch	Yellow to brown	H 3-4 G 2.30 fus	Diss by acids.
1.649 ^	1.616	<u>1.628</u>	1.641	.025	ANTHOPHYLLITE (Mg,Fe) ₇ Si ₈ O ₂₂ (OH) ₂	79° (88+9°)	Y = $\frac{b}{c}$ Z = $\frac{c}{a}$ el pos	ORTH pris c	210 perf at 126°	Brown	H 6 G 3.10	Insol in acids. FeO 11.1, Al ₂ O ₃ 1.9%.
1.57 ^	1.622	---	1.640	.020	TENGERITE CaY ₃ (CO ₃) ₄ (OH) ₃ ·3H ₂ O	Large	---	ORTH fib	---	White	Infus G 2.8	Diss by acids with eff.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.618 v 1.647	1.625	<u>1.629</u>	1.655	.030	PREHNITE $\text{Ca}_2\text{Al}(\text{AlSi}_3)_0\text{}_{10}(\text{OH})_2$	63° (43+10°) $r > v$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	001 good	Green	H 6 G 2.93 F 2	Dec by HCl with sepn of silica. Fe_2O_3 3.5, FeO 0.5%.
	1.616	(<u>1.63</u>)	1.677	.061	CARBONATE-CYANOTRICHITE $(\text{Cu,Zn})_4\text{Al}_2(\text{CO}_3,\text{SO}_4)(\text{OH})_{12} \cdot 2\text{H}_2\text{O}$	55-60° $r > v$ str	el neg	ORTH fib	---	Pale blue to azure-blue	H ~ 2 G 2.66 fus	Diss by acids. Pleoc str, X cols, Z bright blue.
	1.623	<u>1.630</u>	1.684	.061	GUILDITE $\text{CuFe}(\text{SO}_4)_2(\text{OH}) \cdot 4\text{H}_2\text{O}$	Small to med (40+6°)	$Y = \frac{c}{a}$	MCL	100, 001 perf	Chestnut-brown	H 2.5 G 2.70	Diss by acids. Pleoc, X and Y pale yellow, Z greenish-yellow.
1.65 v	---	<u>1.63</u>	---	.02	HOMILITE (altered) $\text{Ca}_2(\text{Fe,Mg})\text{B}_2\text{Si}_2\text{O}_{10} \cdot \text{H}_2\text{O}$	Med large, $r < v$ very str	---	MCL	One imperf	Yellow	H 5 G 3 F 2	Gel with acids.
1.620 ^	1.630	<u>1.630</u>	1.635	.005	CHAMOSITE (Chlorite grp) $(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3})_5\text{Al}(\text{Si}_3\text{Al})_0\text{}_{10}(\text{OH},\text{O})_8$	Small	$Z \sim \frac{c}{a}$ el clv neg	MCL	001 perf	Green	H 3 G 2.9	Nearly insol in acids. Pleoc, X and Y deep green, Z greenish brown. FeO 28.7, Al_2O_3 22.9%.
1.608 ^	1.623	<u>1.631</u>	1.657	.034	CHURCHITE $\text{YPO}_4 \cdot 2\text{H}_2\text{O}$	Small (59+8°) disp dist	$X = \frac{b}{c}$ $Z:c = 31^\circ$ 31°	MCL el $\frac{c}{a}$	010, 100, 001	White, yellow	H 3 G 3.2 infus	Diss by hot acids.
1.618 ^	1.629	<u>1.631</u>	1.638	.009	TOPAZ $\text{Al}_2\text{SiO}_4(\text{F},\text{OH})_2$	48° $r > v$	$X = \frac{a}{b}$ $Z = \frac{c}{a}$ el pos el clv neg	ORTH el $\frac{c}{a}$	001 perf	Gray, brown, yellow	H 8 G 3.50 infus	Insol in acids. F 13.2%.
1.617 ^	1.619	<u>1.632</u>	1.653	.034	CHONDRODITE (Humite grp) $(\text{Mg,Fe})_5(\text{SiO}_4)_2(\text{F},\text{OH})_2$	80° $r > v$ wk	$Z = \frac{b}{c}$ $X:c = 27^\circ$	MCL	100 poor	Yellow, red	H 6 G 3.28 infus	Gel with acids. Poly tw 001. Pleoc, X yellow, Y and Z pale yellow. FeO 10.5, MnO 1.2, F 5.4%.
	1.605	<u>1.633</u>	1.703	.098	ROKŮHNITE $\text{FeCl}_2 \cdot 2\text{H}_2\text{O}$	64° $r < v$	$X = \frac{b}{c}$ $Z:c = -49^\circ$	MCL	110 very good	Light green	---	Sol in H_2O .
1.618 v 1.651	1.632	<u>1.634</u>	1.651	.019	PARGASITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{Al}(\text{Si}_6\text{Al}_2)_0\text{}_{22}(\text{OH},\text{F})_2$	64° (38+20°)	$Y = \frac{b}{c}$ $Z:c = 17^\circ$ el pos	MCL pris $\frac{c}{a}$	110 perf at 124°	Cols, brown	H 6 G 3.1	Insol in acids. FeO 4.0, Al_2O_3 13.3, CaO 10.2, Na_2O 2.6%.

	1.628	<u>1.635</u>	1.698	.070	SARMIENTITE $\text{Fe}_2(\text{AsO}_4)(\text{SO}_4)(\text{OH}) \cdot 5\text{H}_2\text{O}$	38°	$Y = \frac{b}{c}$ $Z : \underline{c} = 12^\circ$	MCL	---	Yellow-orange	G 2.58	Diss by acids.
	1.615	<u>1.635</u>	1.656	.041	SKŁODOWSKITE $\text{Mg}(\text{UO}_2)_2\text{Si}_2\text{O}_6(\text{OH})_2 \cdot 5\text{H}_2\text{O}$	~ 90° $r < v$ str	$Y = \frac{b}{c}$	MCL el <u>b</u>	100 perf	Lemon-yellow	H 2-3 G 3.64	Gel with acids. U opt neg. Pleoc, X cols, Y pale yellow, Z yellow.
1.632	1.617	<u>1.635</u>	1.652	.035	TILLEYITE $\text{Ca}_5\text{Si}_2\text{O}_7(\text{CO}_3)_2$	88° $r < v$ perc	$Z : \underline{c} = 12^\circ$	MCL	100 perf another at 42°	White	G 2.84 infus	Eff and gel with acids.
	1.618	<u>1.635</u>	1.650	.032	SEGELERITE $\text{CaMgFe}^{+3}(\text{PO}_4)_2(\text{OH}) \cdot 4\text{H}_2\text{O}$	Large	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010 perf	Pale green	H 4 G 2.67	Pleoc, X and Y cols, Z yellow.
	1.630	<u>1.636</u>	1.664	.034	HILGARDITE $\text{Ca}_2\text{B}_5\text{ClO}_8(\text{OH})_2$	35° (50+8°) $r > v$	$Y = \frac{b}{c}$ $Z : \underline{c} = 1.5^\circ$	MCL tab	010 perf 100 less so	Cols	H 5 G 2.71 F 2	Diss by acids.
(1.639)	1.630	<u>1.636</u>	1.664	.034	PARAHILGARDITE $\text{Ca}_2\text{B}_5\text{ClO}_8(\text{OH})_2$	35° (50+8°) $r > v$	---	TCL	010, 100 perf	Cols	H 5 G 2.71 F 2	Diss by acids.
1.654	1.632	<u>1.636</u>	1.643	.011	ANDALUSITE Al_2SiO_5	Large $r < v$ str	$X = \frac{a}{c}$ $Z = \frac{c}{a}$ el pos	ORTH fib <u>c</u>	110 perf	Cols, pink	H 7 G 3.15 infus	Insol in acids. Fe_2O_3 1.5%.
1.604 ◇ 1.664	1.631	<u>1.636</u>	1.660	.029	PECTOLITE $\text{Na}(\text{Ca}, \text{Mn})_2\text{Si}_3\text{O}_8(\text{OH})$	47° $r < v$ str	$Z \sim \frac{b}{c}$ $Y : \underline{a} = 9^\circ$ el pos	TCL pris <u>b</u>	100, 001 perf	Light red	H 5 G 2.97-3.13 F 2	Partly dec by acids. MnO 19.5, FeO 2.5%.
1.624 ◇ (1.697)	1.636	<u>1.637</u>	1.648	.012	BARITE BaSO_4	37° $r < v$ wk	$X = \frac{c}{b}$ $Y = \frac{b}{c}$ el pos	ORTH tab 001	001 perf 210 less so	White, blue, yellow	H 3-3.5 G 4.50 F 3	Insol in acids.
	1.634	---	1.642	.008	IRHEMITE $\text{Ca}_4\text{MgH}_2(\text{AsO}_4)_4 \cdot 4\text{H}_2\text{O}$	---	ext angle 25° el neg	MCL spher	---	White to pale rose	G 3.09	Diss by acids.
1.623 ◇ 1.653	1.628	<u>1.638</u>	1.655	.027	HUMITE (Humite grp) $(\text{Mg}, \text{Fe})_7(\text{SiO}_4)_3(\text{OH}, \text{F})_2$	76° $r > v$ wk	$X = \frac{a}{b}$ $Z = \frac{b}{a}$	ORTH tab 010	001 perf	Yellow, brown	H 6 G 3.28 infus	Gel with acids. Pleoc, X and Z pale yellow, Y nearly cols. FeO 7.9, Fe_2O_3 1.0, MnO 1.7, F 5.0%.
	1.620	<u>1.639</u>	1.686	.066	KRAUTITE $\text{MnAs}^{+5}\text{O}_3(\text{OH}) \cdot \text{H}_2\text{O}$	65+5°	$X = \frac{b}{c}$ el cTv pos	MCL	010 perf 101 good	Pink	H < 4 G 3.30	---
1.636 ^	1.638	(1.639)	1.670	.032	STRONTIOHILGARDITE $(\text{Sr}, \text{Ca})_8\text{B}_{18}\text{O}_{33}\text{Cl}_4 \cdot 4\text{H}_2\text{O}$	19-46° $r > v$	---	TCL	010, 100 perf	Cols	H 5 G 2.9 F 2	Diss by acids.
1.670	1.625	<u>1.640</u>	1.696	.071	STRUNZITE $\text{MnFe}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	(56+4°)	$Z : \underline{c} = 10^\circ$	TCL ps mcl laths	---	Straw- to brownish-yellow	G 2.47-2.56	Diss by acids. Pleoc wk in yellow, abs Z > X, Y.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
(1.620) ↓ 1.651	1.637	<u>1.640</u>	1.662	.025	PENKVKILSITE $\text{Na}_4\text{Ti}_2\text{Si}_8\text{O}_{22} \cdot 5\text{H}_2\text{O}$	(41±13°)	---	MCL or ORTH	001 perf	White	H 5 G 2.58 fus	Gel with acids. Clotted mass of fibers.
	1.630	<u>1.640</u>	1.655	.025	CUMMINGTONITE (Amphibole grp) $(\text{Mg}, \text{Fe})_7\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	78° $r > v$	$Y = \frac{b}{c}$ $Z:c = 20^\circ$ el pos	MCL pris c	110 perf at 125°	Brown, green	H 6 G 3.15 fus	Insol in acids. Pleoc wk, X cols, Y and Z pale green, FeO 16.8, Fe_2O_3 1.9, Al_2O_3 2.8, MnO 0.8%.
	1.64	<u>1.64</u>	1.66	.02	ROEBLINGITE $\text{Pb}_2(\text{Ca}, \text{Sr})_6(\text{Mn}, \text{Ca})(\text{SO}_4)_2\text{Si}_6\text{O}_{14}(\text{OH})_{10}$	Small	---	MCL u mass	001 perf 100 imperf	White	H 3 G 3.43 F 3	Gel with acids.
	1.640	<u>1.640</u>	1.657	.017	SARCOLITE $(\text{Ca}, \text{Na})_{7-8}\text{Al}_4\text{Si}_6\text{O}_{24}(\text{OH})_2 (?)$	Small $r > v$ str	---	TET cubo-oct	---	Reddish-white, flesh	H 6 G 2.7 F 3 (?)	Gel with acids.
	---	<u>1.64</u>	---	wk	NINGYOITE $(\text{U}, \text{Ca}, \text{Ce})_2(\text{PO}_4)_2 \cdot 1-2\text{H}_2\text{O}$	---	el pos	ORTH ps hex	---	Brown to brownish-green	---	Slightly pleoc in brown. Opt char unk.
v 1.650	1.615	---	1.685	.070	LIKASITE $\text{Cu}_6(\text{PO}_4)(\text{NO}_3)_2(\text{OH})_7$	---	$X = \frac{a}{b}$ $Z = \frac{b}{c}$	ORTH ps hex	001 perf	Blue	G 2.97	Diss by acids. Pleoc, X greenish-blue, Z pale blue. Opt char unk.
	1.633	<u>1.641</u>	1.652	.019	FAIRFIELDITE $\text{Ca}_2\text{Mn}(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	Large $r > v$ med	ext:c on 100 = 40°, on 010 = 10° disp perc	TCL pris foliated	001 perf 010 good 110 dist	White, yellowish	H 3.5 G 3.05 F 4	Diss by acids. MnO 19.7, FeO 1.0%.
	1.636	<u>1.641</u>	1.651	.015	ROSCHERITE $\text{Ca}(\text{Al}, \text{Fe}, \text{Mn})_3\text{Be}_2(\text{PO}_4)_3(\text{OH})_3 \cdot 2\text{H}_2\text{O} (?)$	Med (71±16°) $r > v$ str	$X = \frac{b}{c}$ $Y:c = -15^\circ$	MCL and TCL pris	001 perf 010 good	Brown to olive-green	H 4.5 G 2.93 fus	Diss by acids
	1.637	(<u>1.642</u>)	1.670	.033	TYRETSKITE $\text{Ca}_2\text{B}_5\text{O}_8(\text{OH})_2(\text{OH}, \text{Cl})$	46°	---	TCL fib	---	White to brownish	G 2.19 (?)	---
	1.632	<u>1.642</u>	1.657	.025	COLLINSITE $\text{Ca}_2(\text{Mg}, \text{Fe})(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	80°	el pos	TCL bladed	001, 010 fair	Light brown	H 3.5 G 2.99 fus	Diss by acids.

1.657 v 1.618 1.679	1.638	1.642	1.653	.015	MULLITE $\text{Al}_6\text{Si}_2\text{O}_{13}$	20-60° (63±16°) $r > v$ perc	$Y = b$ $Z = \frac{c}{\underline{c}}$ el pos	ORTH acic \underline{c}	010 perf	Cols, gray	H 6-7 G 3.23 infus	Insol in acids. Data for synth compd.
	1.640	1.642	1.647	.007	JUANITE $\text{Ca}_{10}\text{Mg}_4\text{Al}_2\text{Si}_{11}\text{O}_{39} \cdot 4\text{H}_2\text{O}$ (?)	50°	el pos	ORTH(?) fib	---	White, brownish	G 3.01 F 3	Dec by acids. Alteration product of Melilite.
	1.631	1.643	1.695	.064	RANSOMITE $\text{CuFe}_2(\text{SO}_4)_4 \cdot 6\text{H}_2\text{O}$	Med	---	MCL radi- ating el \underline{c}	010 perf	Sky-blue	H 2.5 G 2.63	Diss by acids.
	1.632	1.643	1.664	.032	CLINOHUMITE (Humite grp) $(\text{Mg}, \text{Fe})_9(\text{SiO}_4)_4(\text{F}, \text{OH})_2$	74° $r < v$	$Z = b$ $X \sim \frac{c}{\underline{c}}$	MCL	---	Yellow to brown	H 6 G 3.28 infus	Gel with acids. FeO 6.6, MnO 1.7, F 3.2%.
	1.638	1.643	1.650	.012	MAGNESIO-ARFVEDSONITE (Amphibole grp) $(\text{Na}, \text{Ca})_3(\text{Mg}, \text{Fe}^{+2})_4\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	80°	$Y = b$ $Z:\underline{c} = 39^\circ$ el pos	MCL pris \underline{c}	110 perf at 124°	Blue- green	H 5-6 G 3.17	Insol in acids. FeO 8.3, MnO 6.0%.
	1.620	(1.644)	1.674	.054	ZIRCOSULFATE $\text{Zr}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	75°	---	ORTH	---	White	H 2.5-3 G 2.95	Sol in H_2O .
	1.640	1.644	1.650	.010	RANKINITE $\text{Ca}_3\text{Si}_2\text{O}_7$	64°	$Y = b$ $X:\underline{c} = 15^\circ$	MCL	---	Cols	G 2.96 infus	Gel with acids.
1.663 v	1.618	1.645	1.679	.061	ERYTHRITE $(\text{Co}, \text{Ni}, \text{Mg})_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	Large $r < v$ wk	$X = b$ $Z:\underline{c} = 39^\circ$ el pos	MCL laths 010	010 perf	Crimson, peach- color	H 2 G 3.1 F 2	Diss by acids. Pleoc, X pale pink, Y violet, Z red. CoO 14.2, NiO 9.2, MgO 3.2%.
	1.641	1.645	1.672	.031	KURGANTAITE $(\text{Sr}, \text{Ca})_2\text{B}_4\text{O}_8 \cdot \text{H}_2\text{O}$ (?)	Small (43±11°)	el pos	TCL(?) mass	---	White	H ~ 6 G 2.9-3.0	Diss by acids. Perhaps = strontio- hilgardite.
1.657 v	1.640	1.645	1.652	.012	JADEITE (Pyroxene grp) $\text{Na}(\text{Al}, \text{Fe}^{+3})\text{Si}_2\text{O}_6$	67° $r > v$ mod	$Y = b$ $Z:\underline{c} = 40^\circ$	MCL fib \underline{c}	110 good at 87°	Cols to green	H 6 G 3.25 F 2.5	Insol in acids. Fe ₂ O ₃ 0.3, FeO 0.2, CaO 1.4%.
1.667 v	1.643	1.645	1.651	.008	MOSANDRITE var. Rinkolite $(\text{Na}, \text{Ca}, \text{Ce})_3\text{Ti}(\text{SiO}_4)_2\text{F}$	45-80° $r > v$ str	$Y = b$ $X:\underline{c} = 3^\circ$	MCL el \underline{c}	100 dist 010 poor	Yellow to reddish- brown	H 4-4.5 G 3.0-3.4 F 3	Gel with acids. Pleoc wk, abs Z > Y > X.
	1.643	1.645	1.649	.006	PELLYITE $\text{Ba}_2\text{Ca}(\text{Fe}^{+2}, \text{Mg})_2\text{Si}_6\text{O}_{17}$	47° $r > v$ very str	---	ORTH	Conch prism poor	Cols to pale- yellow	H 6 G 3.51	Dec by HCl.
1.622 1.658 v	1.638	1.647	1.682	.044	VESZELYITE $(\text{Cu}, \text{Zn})_3(\text{PO}_4)(\text{OH})_3 \cdot 2\text{H}_2\text{O}$	56° $r < v$ str	$Y = b$ $Z:\underline{c} = 36^\circ$	MCL pris	001, 110	Dark bluish- green	H 3.5 G 3.43 fus	Diss by acids. Pleoc, X blue, Z yellow-green.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.629 ^	1.637	<u>1.647</u>	1.668	.031	PREHNITE $\text{Ca}_2\text{Al}_2\text{Si}_3\text{O}_{10}(\text{OH})_2$	65° $r > v$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	001 good	Dark green	H 6 G 2.97 F 2	Dec by HCl with sepn of silica. Fe_2O_3 7.1%.
	1.642	<u>1.647</u>	1.672	.030	KORZHINSKITE $\text{CaB}_2\text{O}_4 \cdot \text{H}_2\text{O}$	44°	$X:c/v = 90^\circ$ el pos	ORTH(?) pris	One clv, el	Cols	---	Diss by acids. Tw.
1.64 □ 1.66	1.642	(<u>1.647</u>)	1.653	.011	HIORTDAHLITE $(\text{Ca}, \text{Na})_3\text{ZrSi}_2\text{O}_7$ $(\text{F}, \text{OH}, \text{O})_2$	~ 90° $r > v$ perc	opt pl ~ 111 Ext on 100 = 65°	TCL tab 100	Pris, dist	Yellow to brown	H 5.5 G 3.25 fus	Gel with acids. Poly tw. Pleoc, X cols, Y yellowish, Z wine-yellow.
	1.637	<u>1.648</u>	1.676	.039	LOSEYITE $(\text{Mn}, \text{Zn})_7(\text{CO}_3)_2(\text{OH})_{10}$	64° $r > v$ wk	$Y = \frac{b}{c}$	MCL el <u>b</u>	---	Bluish-white	H ~ 3 G 3.27 infus	Diss by acids.
1.662 v	1.643	<u>1.648</u>	1.674	.031	REDDINGITE $(\text{Mn}, \text{Fe})_3(\text{PO}_4)_2 \cdot 3\text{H}_2\text{O}$	(48+9°) $r > v$ dist	$Y = \frac{b}{c}$ $X = \frac{a}{b}$	ORTH oct, tab	010 poor fr uneven	Pink	H 3-3.5 G 3.27 fus	Diss by acids. Pleoc, X cols, Y pink, brown, Z pale yellow. MnO 48.2, FeO 2.2, MgO 0.4%.
1.615 ◇ 1.652	1.641	<u>1.649</u>	1.661	.020	MAGNIOTRIPLITE $(\text{Mg}, \text{Mn}, \text{Fe})_2\text{PO}_4(\text{F}, \text{OH})$	60° (79+12°) $r > v$ perc	$Y = \frac{b}{c}$ $X:a = 18^\circ$	MCL	100 perf	Yellow to brown	H 5.5 G 3.57 F 3	Diss by acids. Pleoc, X and Y light yellow, Z wine-yellow. MgO 17.1, MnO 25.9, FeO 13.0%.
1.659 v	1.644	<u>1.649</u>	1.663	.019	MESSELITE $\text{Ca}_2(\text{Fe}, \text{Mn})(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	Small (62+13°)	ext on 100 23° to <u>c</u>	TCL tab 100	One good	Cols to brownish	H 3-3.5 G 3.1	Diss by acids. FeO 11.5, MnO 7.35%.
1.628 ◇ 1.655	1.645	<u>1.649</u>	1.661	.016	ANTHOPHYLLITE (Amphibole grp) $(\text{Mg}, \text{Fe})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	59°	$Y = \frac{b}{c}$ $Z = \frac{c}{a}$ el pos	ORTH pris <u>c</u>	210 perf at 126°	Brown	H 6 G 3.24	Insol in acids. FeO 20.5, Fe_2O_3 1.8, Al_2O_3 1.8%.
1.657 v	1.649	<u>1.649</u>	1.655	.006	BOEHMITE $\text{AlO}(\text{OH})$	Small	---	ORTH	010 perf 100 less so	Cols	G 2.98 infus	Diss by hot NaOH soln. Data for synth compd.
	1.588	<u>1.650</u>	1.722	.134	KRAUSITE $\text{KFe}(\text{SO}_4)_2 \cdot \text{H}_2\text{O}$	Large	$Z = \frac{b}{c}$ $Y:c = -35^\circ$	MCL pris or flat tab	001 perf 100 good	Yellowish-green	H 2.5 G 2.84	Diss by acids. Pleoc wk, X cols, Y and Z pale yellow.

	1.600	<u>1.650</u>	1.722	.122	BIJVOETITE (Y,Dy,Gd) ₂ (UO ₂) ₄ (CO ₃) ₄ (OH) ₆ ·11H ₂ O	84°	$X = \frac{c}{a}$ $Y \sim \frac{a}{b}$	ORTH tab	One good	Yellow	G 3.9	Pleoc, X cols, Y pale yellow, Z dark yellow.
	1.626	<u>1.650</u>	1.686	.060	PSEUDOLAUEITE MnFe ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	80°	$Z = \frac{b}{a} = 12^\circ$	MCL pris and thick tab	---	Orange- yellow	H 3 G 2.46	Diss by acids.
	1.64- 1.65	---	1.67- 1.68	.03	CASSIDYITE Ca ₂ (Ni,Mg)(PO ₄) ₂ ·2H ₂ O	---	---	TCL fib	---	Pale- to bright- green	G 3.1-3.2	Opt char unk.
	1.633	---	1.672	.039	SHARPITE (UO ₂)(CO ₃)·H ₂ O (?)	---	el pos	ORTH(?)	---	Yellow- green	H 2 G > 3.3 infus	Diss by acids. Pleoc faint. Not fluor in UV.
1.641 ^	1.640	<u>1.650</u>	1.660	.020	FAIRFIELDITE (Fairfieldite grp) Ca ₂ (Mn,Fe)(PO ₄) ₂ ·2H ₂ O	86° r > v med	ext:c on 100 = 40° 010 = 10° disp perc	TCL pris, foli- ated	001 perf 010 good 110 dist	White, yellowish	H 3.5 G 3.02 F 4	Diss by acids. MnO 14.8, FeO 4.75%.
1.63 ^ 1.725	---	<u>1.65</u>	---	.02	HOMILITE (altered) Ca ₂ FeB ₂ Si ₂ O ₁₀ ·xH ₂ O	Small r > v or r < v str	---	MCL mass	---	Yellow	H 5 G 3.4 F 2	Gel with acids.
	1.645	<u>1.650</u>	1.655	.010	SAMUELSONITE (Ca,Ba)Ca ₈ (Fe,Mn) ₄ Al ₂ (PO ₄) ₁₀ (OH) ₂	Large	X:c = 22°	MCL pris	001 fair	Cols	H 5 G 3.27-3.33	
1.648 ^ 1.660	1.622	<u>1.651</u>	1.687	.065	PARASYMPLESITE Fe ₃ (AsO ₄) ₂ ·8H ₂ O	83°	$X = \frac{b}{c}$ $Z:c = 31^\circ$ el pos	MCL tab 010	010 perf	Greenish- blue	H 2 G 3.07	Diss by acids. Pleoc, X bluish- green, Z yellowish.
	1.639	<u>1.651</u>	1.681	.042	VUONNEMITE Na ₄ TiNb ₂ Si ₄ O ₁₇ ·2Na ₃ PO ₄	53° (66±6°)	Y:c = 4°	TCL	001 and 2 others perf	Light yellow	H 2-3 G 3.13	Dec by H ₂ O.
v 1.664	1.636	<u>1.651</u>	1.669	.033	FORSTERITE (Olivine grp) Mg ₂ SiO ₄	85° r < v	$X = \frac{b}{c}$ $Y = \frac{c}{a}$	ORTH equant	010, 001 poor	Cols to green	H 7 G 3.22 infus	Gel with acids. Data for synth compd (Fo ₁₀₀).
1.640 ^ 1.664	1.638	<u>1.651</u>	1.665	.027	TIRODITE (Cummingtonite ser, Amphibole grp) Mn ⁺² ₂ (Mg,Fe ⁺²) ₅ Si ₈ O ₂₂ (OH) ₂	89° r > v	$Y = \frac{b}{c}$ $Z:c = 19^\circ$ el pos	MCL pris c	110 perf at 125°	Brown, green	H 6 G 3.10 fus	Insol in acids. Pleoc wk, X cols, Y and Z brown-green. FeO 11.1, MnO 13.2%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.634 ∧ 1.652 neg	1.641	<u>1.651</u>	1.664	.023	PARGASITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{Al}$ $(\text{Si}_6\text{Al}_2)\text{O}_{22}(\text{OH}, \text{F})_2$	82°	$Y = b$ $Z:c = 18^\circ$ el pos	MCL pris <u>c</u>	110 perf at 124°	Brown	H 6 G 3.18	Insol in acids. Pleoc wk, X cols to yellow, Y and Z brown to green. FeO 8.9, Fe ₂ O ₃ 1.1, Al ₂ O ₃ 15.3, CaO 12.2%.
	1.632	<u>1.652</u>	1.693	.061	KORITNIGITE $\text{Zn}(\text{AsO}_3)(\text{OH})\cdot\text{H}_2\text{O}$	70°	---	TCL	010 perf 001, 100	Cols to white	H 2 G 3.54	Diss by acids.
1.649 ∧ 1.673	1.648	<u>1.652</u>	1.672	.024	TRIPLITE $(\text{Mn}, \text{Fe}, \text{Mg}, \text{Ca})_2\text{PO}_4(\text{F}, \text{OH})$	40°	$Y = b$ $X:a = 18^\circ$	MCL	100 perf 010 poor	Brown	H 5.5-6 G 3.6 F 3	Diss by acids. Pleoc in yellow. MnO 31.5, FeO 11.4, MgO 9.9%.
v 1.681	1.652	(<u>1.652</u>)	1.661	.009	NAGELSCHMIDTITE $\text{Ca}_{3-4}(\text{Si}, \text{P})_2\text{O}_8$	0-20°	---	MCL(?)	001 good 110 fair	Cols	G 3.04	---
1.638 ^	1.643	<u>1.653</u>	1.675	.032	HUMITE (Humite grp) $(\text{Mg}, \text{Fe})_7(\text{SiO}_4)_3(\text{OH}, \text{F})_2$	68° $r > v$ wk	$X = a$ $Z = \underline{b}$	ORTH tab 010	001 perf	Brown	H 6 G 3.27 infus	Gel with acids. Pleoc, Y pale yellow, X and Z golden-yellow. FeO 7.8, Fe ₂ O ₃ 2.7, MnO 0.7%.
	1.652	<u>1.653</u>	1.674	.022	KOTOITE $\text{Mg}_3\text{B}_2\text{O}_6$	21° $r > v$	$X = a$ $Y = \underline{b}$	ORTH mass	110 perf	Cols	H 6.5 G 3.10 infus	Diss by warm acids.
v 1.665	1.651	<u>1.653</u>	1.659	.008	GÖTZENITE $(\text{Ca}, \text{Na})_3(\text{Ti}, \text{Al})\text{Si}_2\text{O}_7$ $(\text{F}, \text{OH})_2$	~ 60° $r > v$ str	$X \sim \underline{c}$	TCL	100 perf 001 good	Cols, pink	H 6 G 3.11 F 3	Gel with acids. Poly tw.
v 1.669	1.650	<u>1.653</u>	1.658	.008	ENSTATITE (Orthopyroxene ser, Pyroxene grp) $\text{Mg}_2\text{Si}_2\text{O}_6$	59° $r < v$	$X = b$ $Y = \underline{a}$ el pos	ORTH pris <u>c</u>	210 good at 88°	Cols	H 5-6 G 3.18 F 6	Insol in acids. Tw 100. Data for synth compd En ₁₀₀ .
1.652 □ 1.663	1.62	<u>1.654</u>	1.689	.069	ANNABERGITE $\text{Ni}_3(\text{AsO}_4)_2\cdot 8\text{H}_2\text{O}$	~ 90°	$X = b$ disp dist el clv pos	MCL	010 perf	Apple-green to white	H 2.5-3 G 3.0-3.1 F 4	Diss by acids. U neg.
1.636 ∧ 1.671	1.650	<u>1.654</u>	1.670	.020	ANDALUSITE, manganian $(\text{Al}, \text{Mn})_2\text{SiO}_5$	70° (54+13°) $r < v$ str	$X = a$ $Z = \underline{c}$ el pos	ORTH fib <u>c</u>	110 perf	Green	H 7 G 3.17 infus	Insol in acids. Pleoc, X yellow-green, Y emerald green, Z golden yellow. Mn ₂ O ₃ 9.5, Fe ₂ O ₃ 3.7%.

	1.650	<u>1.654</u>	1.661	.011	VLADIMIRITE $H_2Ca_5(AsO_4)_4 \cdot 5H_2O$	70° r > v str	Z:c = 36°	MCL	---	Cols	H 3.5 G 3.14 F diff	Diss by acids. Reported as biax neg.
1.678 v	1.651	<u>1.654</u>	1.660	.009	CLINOENSTATITE (Pyroxene grp) $Mg_2Si_2O_6$	54°	X = b Z:c = 22°	MCL el c	110 good at 87°	Cols	H 5-6 G 3.28 infus	Insol in acids. Poly tw 100 character- istic. Synth compd.
	1.652	---	1.659	.007	YUKSPORITE $(Na,K)_4(Ca,Sr,Ba)_4$ $(Ti,Al,Fe)_3Si_8O_{16}$ $(F,Cl)_2 \cdot 4H_2O$ (?)	---	el neg	MCL(?) fib, platy	---	Pink to red	---	---
	1.648	<u>1.655</u>	1.670	.022	Unnamed phosphate of Ca, Ba, Mn, Fe, Al	80°	---	fib	---	Greenish-	---	Al_2O_3 8.0, Fe_2O_3 7.6, MnO 8.6, CaO 24.2, BaO 3.5, H_2O 3.3%.
1.649 v 1.667	1.648	<u>1.655</u>	1.662	.014	GEDRITE (Amphibole grp) $(Mg,Fe)_5Al_2(Si_6Al_2)O_{22}$ (OH) ₂	87°	Y = b Z = c el pos	ORTH pris c	210 perf at 126°	Brown	H 6 G 3.26	Insol in acids. FeO 14.6, Fe_2O_3 1.3, Al_2O_3 13.3%. ²³
1.662 v	1.648	<u>1.655</u>	1.662	.014	DICKINSONITE $(K,Ba)(Na,Ca)_5$ $(Mn,Fe,Mg)_{14}Al(PO_4)_{12}$ (OH,F) ₂	~ 90° r > v str	X = b Y:c = 15° el neg	MCL tab, foli- ated 001	001 perf	Olive- green	H 3-3.5 G 3.40 F easy	Diss by acids. Pleoc wk in green.
	1.649	<u>1.656</u>	1.714	.065	NATROCHALCITE $NaCu_2(SO_4)_2(OH) \cdot H_2O$	37° r < v str inclined	Y = b X:c = -12° el clv pos	MCL el c	001 perf	Emerald- green	H 4.5 G 3.49 F easy	Diss by acids.
	---	<u>1.656</u>	---	.052	DIOPHASE $CuSiO_2(OH)_2$	43°	---	TRIG	10Tl perf	Emerald- green	H 5 G 3.28 infus	Gel with acids. Pleoc wk, abs X > Y and Z.
	1.652	<u>1.656</u>	1.672	.020	EUCLASE $BeAlSiO_4(OH)$	48° r > v	Y = b Z:c = 42°	MCL el c	010 perf 100, 001 poor	Cols to pale blue	H 7.5 G 3.06 F 5.5	Insol in acids.
1.664 v	1.643	<u>1.657</u>	1.681	.038	LEUCOSPHEENITE $BaNa_4Ti_2B_2Si_{10}O_{30}$	78° r > v	Z = b Y ~ c el cTv neg	MCL el a	010 perf 001 fair	Cols to brown	H 6-6.5 G 3.07 F diff	Insol in acids. Tw pl 001.
1.649 ^	1.648	<u>1.657</u>	1.668	.020	BOEHMITE $AlO(OH)$	79°	---	ORTH	010 perf 100 less so	Cols	G 2.98 infus	Diss by hot NaOH soln.
1.642 ^	1.653	<u>1.657</u>	1.671	.018	MULLITE $Al_6Si_2O_{13}$	20-60° (57+14°) r > v perc	Y = b Z = c el pos	ORTH acic c	010 perf	Gray, lilac	H 6-7 G 3.23 infus	Insol in acids. Fe_2O_3 0.9, TiO_2 2.3%.
1.645 v 1.670	1.654	<u>1.657</u>	1.666	.012	JADEITE (Pyroxene grp) $Na(Al,Fe)Si_2O_6$	70° r > v mod	Y = b Z:c = 34°	MCL fib c	110 good at 87°	Cols to green	H 6 G 3.34 F 2.5	Insol in acids. Fe_2O_3 0.45, CaO 0.1, Na_2O 13.4%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.640	<u>1.658</u>	1.760	.120	METASTUDTITE $\text{UO}_4 \cdot 2\text{H}_2\text{O}$	48°	---	ORTH fib	---	Yellow	G 4.67 calc	Sol in hot HCl. Does not fluor in UV.
1.647 ^	1.640	<u>1.658</u>	1.695	.055	VESZELYITE $(\text{Cu}, \text{Zn})_3(\text{PO}_4)_3(\text{OH})_3 \cdot 2\text{H}_2\text{O}$	71° $r < v$ str	$Y = b$ $Z:c = 36^\circ$	MCL pris	001, 110	Dark bluish-green	H 3.5 G 3.43 fus	Diss by acids. Pleoc, X blue, Z yellow-green.
1.671 v	1.653	<u>1.659</u>	1.677	.024	SPODUMENE (Pyroxene grp) $\text{LiAlSi}_2\text{O}_6$	66° $r < v$	$Y = b$ $Z:c = 25^\circ$ el pos	MCL pris c	110 perf at 87°	Pink, gray, green	H 6.5 G 3.14 F 3.5	Insol in acids. Abs X > Y > Z in thick sections.
1.649 ^	1.653	<u>1.659</u>	1.676	.023	MESSELITE $\text{Ca}_2(\text{Fe}, \text{Mn})(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	25° (62+11°)	ext on 100 20° to c	TCL tab 100	One good	Cols to brownish	H 3-3.5 G 3.1	Diss by acids. FeO 20.9, Fe ₂ O ₃ 0.8, MnO 3.9, MgO 1.4%.
1.651 ^	1.628	<u>1.660</u>	1.705	.077	PARASYMPLESITE $\text{Fe}_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	~ 90° (82+2°)	$X = b$ $Z:c = 31^\circ$ el pos	MCL tab 010	010 perf	Greenish-blue	H 2 G 3.07	Diss by acids. Pleoc, X bluish-green, Y yellow, Z brownish-yellow.
1.718 v	1.645	<u>1.660</u>	1.715	.070	PLANCHEITE $\text{Cu}_8\text{Si}_8\text{O}_{22}(\text{OH})_4 \cdot \text{H}_2\text{O}$	Large (57+4°)	el ~ Z	ORTH fib	---	Blue	H 5 G 3.4	Slightly attacked by acids. Pleoc, X pale blue, Y and Z blue.
1.592 neg ^ 1.710	1.650	<u>1.660</u>	1.680	.030	STRENGITE, aluminian $(\text{Fe}, \text{Al})\text{PO}_4 \cdot 2\text{H}_2\text{O}$	Med Large $r > v$ str	$X = a$ $Y = b$	ORTH fib c	---	Gray, brown, red	H 4.5 G 2.6 fus	Diss by acids. Al ₂ O ₃ 12.5%.
1.664 ^	1.659	<u>1.660</u>	1.680	.021	SILLIMANITE Al_2SiO_5	20° $r > v$ perc	$Y = b$ $Z = c$ el pos	ORTH acic c	010 perf	White, brown	H 7 G 3.23 infus	Insol in acids. Some vars pleoc, cols to blue.
	1.656	<u>1.660</u>	1.668	.012	MAGNESIOAXINITE $\text{Ca}_2\text{MgAl}_2\text{BSi}_4\text{O}_{15}(\text{OH})$	Large	---	TCL	---	Pale blue	H 6.5 G 3.18 fus	Insol in acids.
	1.641	<u>1.660</u>	1.682	.041	JONESITE $\text{Ba}_4(\text{K}, \text{Na})_2\text{Ti}_4\text{Al}_2\text{Si}_{10}\text{O}_{36} \cdot 6\text{H}_2\text{O}$	77°	$X = b$ $Z = c$	ORTH bladed	010	Cols	H 3-4 G 3.25	Insol in HCl. Fluor orange in short-wave UV.
1.676 v	1.660	(<u>1.661</u>)	1.663	.003	PHARMACOSIDERITE $\text{KFe}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6-7\text{H}_2\text{O}$	Large $r < v$ str	---	CUB	100 imperf	Olive-green to brown	H 2.5 G 2.8-2.9 F 2	Diss by acids. Poly tw.
	1.629	<u>1.662</u>	1.729	.098	LINDACKERITE $\text{H}_2\text{Cu}_5(\text{AsO}_4)_4 \cdot 8\text{H}_2\text{O}$	73° $r < v$ str	$Y = b$ $X:el = 26^\circ$	MCL fib	010 perf	Apple-green	H 2-2.5 G 2.0-2.5 F 2-3 (?)	Diss by acids.

1.648 ◇ 1.680	1.655	<u>1.662</u>	1.683	.028	REDDINGITE (Mn,Fe) ₃ (PO ₄) ₂ ·3H ₂ O	65° r > v perc	Y = $\frac{b}{a}$ X = $\frac{a}{b}$	ORTH oct, tab	010 poor fr uneven	Pink	H 3-3.5 G 3.14 fus	Diss by acids. Pleoc, X cols, Y pinkish-brown, Z yellow. MnO 38.4, FeO 12.7%.
1.655 ^	1.658	<u>1.662</u>	1.671	.013	DICKINSONITE (K,Ba)(Na,Ca) ₅ (Mn,Fe,Mg) ₁₄ Al(PO ₄) ₁₂ (OH,F) ₂	Med r > v str	X = $\frac{b}{c}$ Y:c = 15° el neg	MCL tab, foli- ated 001	001 perf	Olive- green	H 3.5-4 G 3.40 F easy	Diss by acids. Pleoc wk in green.
1.667 ┐	1.658	<u>1.662</u>	1.668	.010	BORACITE Mg ₃ B ₇ O ₁₃ Cl	Large	X = $\frac{c}{a}$ Y = $\frac{a}{c}$	ORTH ps cub	Conch to uneven	White to greenish	H 7-7.5 G 2.95 F 2	Slowly diss by HCl. Inverts to cub at 265°C.
	1.598	<u>1.663</u>	1.737	.139	PARABUTLERITE Fe(SO ₄)(OH)·2H ₂ O	44-87° (90+2°) r > v mod	X = $\frac{b}{c}$ Y = $\frac{c}{b}$	ORTH pris	120 poor fr conch	Orange to brown	H 2.5 G 2.55	Diss by acids.
1.645 ^	1.629	<u>1.663</u>	1.701	.072	ERYTHRITE Co ₃ (AsO ₄) ₂ ·8H ₂ O	~ 90° r < v wk	X = $\frac{b}{c}$ Z:c = 32° el clv pos	MCL laths	010 perf	Crimson	H 2 G 3.06 F 2	Diss by acids. Pleoc, X pink, Y violet, Z red. CoO 33.4, FeO 4.0%.
1.657 ^	1.649	<u>1.664</u>	1.691	.042	LEUCOSPENITE BaNa ₄ Ti ₂ B ₂ Si ₁₀ O ₃₀	76° r > v perc	Z = $\frac{b}{c}$ Y ~ $\frac{c}{b}$ el neg	MCL el a	010 perf 001 fair	Cols to brown	H 6-6.5 G 3.09 F diff	Insol in acids. Tw pl 001.
	1.637	<u>1.664</u>	1.692	.055	WILHELMVIERLINGITE CaMnFe ³⁺ (PO ₄) ₂ (OH)·2H ₂ O	(90+4°)	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH	---	Yellow	H 4 G 2.58	Reported to be neg with 2V, 45°. Pleoc, X and Y ^x light yellow, Z yellow.
1.651 ◇ 1.674	1.653	<u>1.664</u>	1.686	.033	FORSTERITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	87° (71+7°) r < v	X = $\frac{b}{c}$ Y = $\frac{c}{b}$	ORTH equant	010, 001 poor	Cols to green	H 7 G 3.35 infus	Gel with acids. FeO 8.6, Fe ₂ O ₃ 0.4, MnO 0.2% (F ₆ g ₃).
1.636 ^	1.660	<u>1.664</u>	1.688	.028	SERANDITE Na(Mn,Ca) ₂ Si ₃ O ₈ (OH)	36° (45+11°)	Z = $\frac{b}{c}$ X:c = 57°	TCL el b	100, 001	Rose-red	G 2.32	MnO 29.0, Fe ₂ O ₃ 1.3%.
1.651 ◇ 1.677	1.651	<u>1.664</u>	1.678	.027	CUMMINGTONITE (Amphibole grp) (Fe,Mg) ₇ Si ₈ O ₂₂ (OH) ₂	86° r > v	Y = $\frac{b}{c}$ Z:c = 17° el pos	MCL pris c	110 perf at 125°	Brown, green	H 7 G 3.27 fus	Insol in acids. Pleoc in green. FeO 28.0, MnO 1.0, Al ₂ O ₃ 2.4%.
	1.655	<u>1.664</u>	1.675	.020	ATTAKOLITE (Ca,Mn,Sr) ₃ Al ₆ (PO ₄ ,SiO ₄) ₇ ·3H ₂ O	~ 84° r > v	---	ORTH mass	---	Pink	G 3.24 F easy	Partly dec by acids.
	1.656	<u>1.664</u>	1.672	.016	JUNITOITE CaZn ₂ Si ₂ O ₇ ·H ₂ O	(90+14°) r < v, wk	---	ORTH hemi- morph	010 good 100, 101 poor	Cols to milky- white	H 4.5 G 3.5 fus	Gel with acids.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
\vee 1.687	1.657	(1.664)	1.675	.018	ROSENBUSCHITE (Ca,Na) ₃ (Zr,Ti)Si ₂ O ₈ F	80°	X:c = 9° el neg	TCL	010 perf	Yellow, brown	H 5-6 G 2.97 F easy	Gel with acids. Pleoc wk, X cols, Z yellow.
	1.663	1.664	1.675	.012	GIANNETITE Silicate of Na, Ca, Mn, Zr, Ti	30°	Y:c = 22°	TCL pris	100 perf	Cols to pale yellow	---	Validity dubious.
	(1.639)	1.665	1.704	.065	TRIANGULITE Al ₃ (UO ₂) ₄ (PO ₄) ₄ (OH) ₅ ·5H ₂ O	80°	---	TCL	---	Yellow	G 3.7	---
1.653 ^	1.662	1.665	1.672	.010	GÖTZENITE (Ca,Na) ₃ (Ti,Al)Si ₂ O ₇ (F,OH) ₂	62° r > v str	X ~ c	TCL	100 perf 001 good	Pink, yellow	H 6 G 3.17 F 3	Gel with acids. Poly tw.
\vee 1.679	1.663	1.666	1.673	.010	LITHIOPHILITE Li(Mn,Fe)PO ₄	60° r < v str	X = c Y = a el neg	ORTH el c	100 perf 010 less so	Salmon-pink	H 5 G 3.43 F 1.5	Diss by acids. Pleoc, X deep pink, Y pale yellow, Z pale pink. MnO 42.6, FeO 2.9%.
1.675 ┐	1.650	1.667	1.688	.038	LUDLAMITE (Fe,Mg,Mn) ₃ (PO ₄) ₂ ·4H ₂ O	83° r > v	Y = b Z:c = -67°	MCL tab 001	001 perf 100 dist	Bright green	H 3.5-4 G 3.2 F 2-2.5	Diss by acids.
1.655 ◇ 1.681	1.657	1.667	1.678	.021	FERRO-GEDRITE (Amphibole grp) (Fe,Mg) ₅ Al ₂ (Si ₆ Al ₂)O ₂₂ (OH) ₂	87°	Y = b Z = c el pos	ORTH pris c	210 perf at 126°	Gray, brown	H 6 G 3.28	Insol in acids. FeO 18.3, Fe ₂ O ₃ 1.0, Al ₂ O ₃ 17.8%.
1.645 ^	1.662	1.667	1.681	.019	MOSANDRITE (Na,Ca,Ce) ₃ TiSi ₂ O ₇ (O,OH,F)	45-80° (62+13°) r > v	Y = b X:c = 3°	MCL el c	100 dist 010 poor	Yellow to reddish-brown	H 4-4.5 G 3.3-3.4 F 3	Gel with acids. Pleoc wk, abs Z > Y > X.
	1.646	1.668	1.705	.059	COBALTKORITNIGITE (Co,Zn)(As ⁺⁵ O ₃)(OH)·H ₂ O	78°	X:b = 3° Y:a =	TCL	010 perf 100 good	Deep purple	---	Pleoc, X deep violet, Y reddish-violet, Z bluish-violet.
\vee 1.686	1.665	1.669	1.768	.103	NENADKEVICHITE (Na,Ca,K)(Nb,Ti)Si ₂ O ₆ (O,OH)·2H ₂ O	38° (24+5°)	X = a Z = b	ORTH	001 poor	Yellow, rose	H 5 G 2.76 F diff	Diss by H ₂ SO ₄ . Pleoc wk, X cols, Y pale yellow, Z pale rose.

1.653 ∧ 1.677	1.665	<u>1.669</u>	1.674	.009	ENSTATITE (Orthopyroxene ser, Pyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	80° r < v	Y = a Z = c el pos	ORTH pris c	210 good at 88°	Green	H 5.5 G 3.2 F 5.5	Insol in acids. FeO 7.0%. Commonly faintly pleoc.
v 1.698	1.660	<u>1.670</u>	1.770	.110	LABUNTSOVITE (K,Ba,Na)(Ti,Nb) (Si,Al) ₂ (OH) ₇ ·H ₂ O	35° r > v	Y = b Z:c = 63°	MCL	T02 perf	Dark brown	H 5 G 2.87	Pleoc wk, X cols, Z yellow.
1.640 ∧ 1.670 neg	1.619	<u>1.670</u>	1.720	.101	STRUNZITE MnFe ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	(93+2°)	Z:c = 10°	TCL ps mcl laths	---	Straw- to brownish- yellow	G 2.47-2.56	Diss by acids. Pleoc wk in yellow, abs Z > X, Y.
v 1.681	1.664	<u>1.670</u>	1.685	.021	OMPHACITE (Pyroxene grp) (Ca,Na)(Mg,Fe,Al)Si ₂ O ₆	66° r < v mod	Y = b Z:c = 35°	MCL pris c	110 good at 87°	Green	H 5-6 G 3.40 F 5	Insol in acids. Pleoc wk in green. FeO 2.8, Fe ₂ O ₃ 1.8, Na ₂ O 4.5, CaO ³ 14.7%.
1.657 ∧ 1.681	1.662	<u>1.670</u>	1.682	.020	JADEITE (Pyroxene grp) (Na,Ca)(Al,Mg,Fe)Si ₂ O ₆	70° r > v mod	Y = b Z:c = 35°	MCL pris c	110 good at 87°	Green	H 6 G 3.35 F 2	Insol in acids. FeO 0.5, Fe ₂ O ₃ 3.2, Na ₂ O 9.4, CaO ³ 5.1%.
v 1.688	1.665	<u>1.670</u>	1.683	.018	PUMPELLYITE Ca ₂ (Mg,Fe)Al ₂ (SiO ₄) (Si ₂ O ₇)(OH) ₂ ·H ₂ O	15° (64+14°) r < v str rarely r > v	Y = b X:c = 15°	MCL	001 good 100 imperf	Green, bluish- green, brown	H 5.5-7 G 3.18	Insol in acids. Pleoc, X and Z pale yellow-green, Y green, abs Y > X, Z. FeO 1.4, Fe ₂ O ₃ 0.7%.
1.654 ∧ 1.722	1.662	<u>1.671</u>	1.691	.029	ANDALUSITE, manganian, ferrian (Al,Mn,Fe) ₂ SiO ₅	71° r < v str	el pos	ORTH fib c	110 perf	Green	H 7 G 3.22 infus	Insol in acids. Fe ₂ O ₃ 4.2, Mn ₂ O ₃ 4.8, TiO ₂ 1.0%.
1.669 neg ∧ 1.693 neg	1.660	<u>1.671</u>	1.687	.027	FERROTSCHERMAKITE (Amphibole grp) Ca ₂ (Fe ⁺² ,Mg) ₃ (Al,Fe ⁺³) ₂ Si ₆ Al ₂ O ₂₂ (OH,F) ₂	78° r > v	Y = b Z:c = 18°	MCL pris	110 good at 124°	Green, brown	H 6 G 3.2	Insol in acids. FeO 15.8, Fe ₂ O ₃ 9.3, Al ₂ O ₃ 11.2, CaO 10.2, Na ₂ O 1.3%.
	1.670	<u>1.671</u>	1.689	.019	HINSDALITE (Beudantite grp) PbAl ₃ (PO ₄)(SO ₄)(OH) ₆	0-30°	Z = c	TRIG	0001 perf	White, gray	H 4.5 G 3.65 infus	Insol in acids. Base divided into 6 segments.
1.659 ^	1.668	<u>1.671</u>	1.682	.014	SPODUMENE (Pyroxene grp) LiAlSi ₂ O ₆	65° r < v	Y = b Z:c = 24°	MCL pris c	110 good at 87°	Pink, gray, green	H 6.5 G 3.19 F 3.5	Insol in acids. Abs X > Y > Z.
v 1.678	1.665	<u>1.672</u>	1.695	.030	DIOPSIDE (Pyroxene grp) CaMgSi ₂ O ₆	57° r > v wk	Y = b Z:c = 39°	MCL pris c	110 good at 87°	Cols	H 5-6 G 3.28 infus	Insol in acids. FeO 1.5, Fe ₂ O ₃ 0.6%.
v 1.690	1.669	<u>1.672</u>	1.667	.008	MAGNESIUM CHLOROPHOENICITE (Mg,Mn) ₃ Zn ₂ (AsO ₄) (OH,O) ₆	r < v str	Y el	MCL fib	Perf	White	H 3-3.5 G 3.57	Diss by acids.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.672	<u>1.672</u>	1.676	.004	FILLOWITE $\text{Na}_2\text{Ca}(\text{Mn},\text{Fe})_7(\text{PO}_4)_6$	Small $r < v$	$X \sim c$	MCL ps trig	001 very good	Yellow, brown, cols	H 4.5 G 3.43 F 1.5	Diss by acids.
	1.672	<u>1.672</u>	1.673	.001	BARATOVITE $\text{KCa}_8\text{Li}_2\text{Si}_{12}\text{O}_{37}\text{F}$	60° $r > v$ str	$X:c = 50^\circ$	MCL	001 perf	White	H 3.5 G 2.92	---
	1.669	<u>1.673</u>	1.692	.023	GOEDKENITE $(\text{Sr},\text{Ca})_2\text{Al}(\text{PO}_4)_2(\text{OH})$	$45-50^\circ$	$X = b$	MCL	100 fair	Cols to pale yellow	H 5 G 3.83 calc	---
1.652 ∧ 1.683	1.665	<u>1.673</u>	1.682	.017	TRIPLITE $(\text{Mn},\text{Fe},\text{Mg},\text{Ca})_2\text{PO}_4(\text{F},\text{OH})$	$\sim 90^\circ$ $r > v$ str	$Y = b$ $Z:a = 42^\circ$	MCL	100 perf 010 poor	Salmon-pink	H 3.5 G 3.7 F 2.5	Diss by acids. Pleoc, X pale red-brown, Y cols. MnO 52.4, FeO 5.0, MgO 0.6, CaO 3.2%.
1.664 ∧ 1.680	1.656	<u>1.674</u>	1.695	.039	CHRYSLITE (Olivine grp) $(\text{Mg},\text{Fe})_2\text{SiO}_4$	89° $r < v$	$X = b$ $Y = c$	ORTH equant	010, 001 poor	Green	H 7 G 3.33 infus	Gel with acids. FeO 10.8, Fe ₂ O ₃ 1.7, MnO 0.2% (Fo ₈₈).
	1.663	<u>1.674</u>	1.699	.036	SPODIOSITE $\text{Ca}_2(\text{PO}_4)\text{F} (?)$	69° $r > v$ str	---	ORTH(?)	010 dist 001 indist	Ash-gray to brown	H 5 G 2.94 F diff	Diss by acids.
	1.665	<u>1.674</u>	1.686	.021	LAWSONITE $\text{CaAl}_2\text{Si}_2\text{O}_7(\text{OH})_2 \cdot \text{H}_2\text{O}$	82° $r > v$	$X = a$ $Y = b$	ORTH tab 001	010, 001 perf 110 good	Cols, bluish	H 8 G 3.08 F 4	Insol in acids, but after ignition gel with acids. Pleoc wk.
	1.671	<u>1.674</u>	1.684	.013	NATROPHILITE NaMnPO_4	75° $r < v$ str	$X = c$ $Y = a$	ORTH	100 good 010 indist	Deep wine-yellow	H 4.5-5 G 3.41 F 2-2.5	Diss by acids.
	1.672	<u>1.676</u>	1.683	.011	AKROCHORDITE $\text{Mn}_4\text{Mg}(\text{AsO}_4)_2(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	Med $r < v$	$X = b$ $Y:c = 45^\circ$	MCL radial aggregates	Two	Yellow-brown	H 3.5 G 3.26	Diss by acids.
(1.661) ∧ (1.700)	---	<u>1.676</u>	---	wk	PHARMACOSIDERITE $\text{KFe}^{+3}_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6-7\text{H}_2\text{O}$	Large $r < v$ str	---	CUB	100 imperf	Olive-green to brown	H 2.5 G 2.8-2.9 F 2	Diss by acids. Poly tw divided into segments.
1.664 ∧ 1.677 neg	1.660	<u>1.677</u>	1.693	.033	CUMMINGTONITE (Amphibole group) $(\text{Fe},\text{Mg})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	Large $r > v$	$Y = b$	MCL pris <u>c</u>	110 perf at 125°	Green to brown	H 7 G 3.47 fus	Insol in acids. FeO 31.2, Fe ₂ O ₃ 3.4%.

1.669 ^ 1.677 neg	1.672	<u>1.677</u>	1.682	.010	BRONZITE (Orthopyroxene ser, Pyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	~ 90°	Y = $\frac{a}{c}$ Z = $\frac{c}{a}$ el pos	ORTH pris <u>c</u>	210 good at 88°	Green	H 5.5 G 3.3 F 5	Nearly insol in acids. Faint pleoc common.
	1.672	<u>1.678</u>	1.712	.040	MAPIMITE Zn ₂ Fe ⁺³ ₃ (AsO ₄) ₄ ·4H ₂ O	50° r < v str	Y = $\frac{b}{a}$ Z:a = 13°	MCL	---	Blue to green	H 3 G 2.95	Pleoc, X pale yellow, Y deep blue, Z green- ish-yellow.
1.672 ^ 1.683	1.673	<u>1.678</u>	1.705	.032	DIOPSIDE (Pyroxene grp) Ca(Mg,Fe)Si ₂ O ₆	48°	Y = $\frac{b}{c}$ Z:c = 42°	MCL	110 good at 87°	Green to brown	H 6 G 3.24	Insol in acids. FeO 8.2, Fe ₂ O ₃ 1.4, Al ₂ O ₃ 4.5%.
1.654 ^	1.675	<u>1.678</u>	1.688	.013	CLINOHYPERSTHENE (Pyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	57°	X = $\frac{b}{c}$ Z:c = 36°	MCL el <u>c</u>	110 good at 87°	Green	H 5-6 G 3.3-3.4 infus	Insol in acids. Poly tw 100. FeO 12.6, MnO 0.5, Al ₂ O ₃ 0.5%.
1.643 ^ 1.714	1.668	<u>1.679</u>	1.700	.032	CLINOHUMITE (Humite grp) (Mg,Fe) ₉ (SiO ₄) ₄ (F,OH) ₂	Med (73+7°) r < v	Z = $\frac{b}{c}$ X ~ $\frac{c}{a}$	MCL	---	Yellow to red	H 6 G 3.3 infus	Gel with acids. FeO 9.9, MnO 0.3, Fe ₂ O ₃ 1.0, F 1.8%.
1.666 ^ 1.689	1.675	<u>1.679</u>	1.688	.013	LITHIOPHILITE Li(Mn,Fe)PO ₄	60° r < v str	X = $\frac{c}{a}$ Y = $\frac{a}{b}$ el neg	ORTH el <u>c</u>	100 perf 010 less so	Light brown	H 5 G 3.48 F 2	Diss by acids. MnO 31.9, FeO 11.0%.
1.674 ^ 1.680 neg	1.661	<u>1.680</u>	1.697	.036	CHRYSLITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	~ 90°	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH	010, 001 poor	Green	H 7 G 3.45 infus	Gel with acids (Fe ₈₆).
1.662 ^	1.672	<u>1.680</u>	1.700	.028	PHOSPHOFERRITE (Fe,Mn) ₃ (PO ₄) ₂ ·3H ₂ O	68° r > v dist	X = $\frac{a}{b}$ Y = $\frac{b}{a}$	ORTH	010 poor fr uneven	Green	H 4 G 3.3 fus	Diss by acids.
	1.678	<u>1.68</u>	1.683	.003	HARSTIGITE Ca ₆ (Mg,Mn)Be ₄ Si ₆ (O,OH) ₂₄	52° r < v wk	Y = $\frac{b}{a}$ Z = $\frac{a}{b}$	ORTH pris <u>c</u>	---	Cols	H 5.5 G 3.16	Insol in acids, but gel after being heated.
1.670 ^ 1.699	1.673	<u>1.681</u>	1.695	.022	OMPHACITE (Pyroxene grp) (Ca,Na)(Mg,Fe,Al)Si ₂ O ₆	70° r < v mod	Y = $\frac{b}{c}$ Z:c = 41°	MCL pris <u>c</u>	110 good at 87°	Green	H 5-6 G 3.34 F 5	Insol in acids. Pleoc wk in green. FeO 2.7, Fe ₂ O ₃ 1.3, Na ₂ O 4.6, CaO 14.7%.
(1.652) ^	1.680	(<u>1.681</u>)	1.698	.018	NAGELSCHMIDTITE Ca ₃₋₄ (Si,P) ₂ O ₈	0-20°	---	MCL(?)	001 good 110 fair	Cols	G 3.04	---
1.667 ^ 1.688	1.674	<u>1.681</u>	1.691	.017	FERRO-GEDRITE (Amphibole grp) (Fe,Mg) ₅ Al ₂ (Si ₆ Al ₂) ₂₂ (OH) ₂	80°	Y = $\frac{b}{c}$ Z = $\frac{c}{a}$	ORTH pris <u>c</u>	210 perf at 126°	Brown	H 6 G 3.36 F 6	Insol in acids. Pleoc X, Y yellow- green, Z blue-green. FeO 23.0, MnO 1.7, Al ₂ O ₃ 19.2%.
	1.676	(<u>1.681</u>)	1.690	.014	SATPAEVITE 6Al ₂ O ₃ ·V ₂ O ₄ ·3V ₂ O ₅ ·30H ₂ O	~ 70°	---	ORTH(?) micro- cryst	Pina- coidal perf	Canary- to saffron- yellow	H 1.5 G 2.4	Diss by acids. Pleoc wk, abs Z > X.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.670 ^	1.679	<u>1.681</u>	1.685	.006	JADEITE (Pyroxene grp) (Na,Ca)(Al,Fe)Si ₂ O ₆	64° r > v mod	Y = $\frac{b}{c}$ Z:c = 38°	MCL pris \underline{c}	110 good at 87°	Green	H 6 G 3.4 F 2	Insol in HCl. FeO 1.1, Fe ₂ O ₃ 5.7, Na ₂ O 13.0, CaO 2.7%.
(1.618) ^	1.662	<u>1.683</u>	1.717	.055	KOETTIGITE Zn ₃ (AsO ₄) ₂ ·8H ₂ O	74° r < v str	X = $\frac{b}{c}$ Z:c = 37°	MCL fib \underline{c}	010 perf	Carminered	H 2.5-3 G 3.33 F 3	Diss by acids. Pale pink in section, not pleoc.
1.678 ^ 1.685	1.676	<u>1.683</u>	1.705	.029	DIOPSIDE, manganian (Pyroxene grp) Ca(Mg,Mn,Zn)Si ₂ O ₆	60° r > v	Y = $\frac{b}{c}$	MCL pris \underline{c}	110 good at 87°	Brown	H 5.5 G 3.39 F 6	Insol in acids. MnO 7.4, ZnO 3.3, Fe ₂ O ₃ 0.4%.
1.673 ^ 1.704	1.675	<u>1.683</u>	1.692	.017	TRIPLITE (Mn,Fe,Mg,Ca) ₂ PO ₄ (F,OH)	80° r > v	Y = $\frac{b}{a}$ X:a = 42°	MCL	100 perf 010 poor	Reddishbrown	H 4.5 G 3.87 F 2.5	Diss by acids. Pleoc in brown, abs X > Y > Z. MnO 34.8, FeO 23.5, CaO 3.5%.
170 v 1.698	1.684	<u>1.684</u>	1.707	.023	PIGEONITE (Pyroxene grp) (Mg,Fe,Ca)(Mg,Fe)Si ₂ O ₆	18-26°	X = $\frac{a}{c}$ Z:c = 40°	MCL	110 good at 87°	Cols to green	H 5-6 G 3.4 infus	Insol in acids. FeO 20.1, Fe ₂ O ₃ 3.5, CaO 5.5%.
	1.65	---	1.72	.07	OTWAYITE Ni ₂ (CO ₃)(OH) ₂ ·H ₂ O	---	X = fib	ORTH fib ros- ettes	---	Bright green	G 3.41 infus	Diss slowly by cold dil HCl. Pleoc wk, abs Z > X. Opt char unk.
	1.646	<u>1.685</u>	1.745	.099	ERIOCHALCITE CuCl ₂ ·2H ₂ O	75° (80±2°) r < v str	X = $\frac{b}{a}$ Z = $\frac{a}{c}$	ORTH	110 perf 001 good	Bluishgreen	H 2.5 G 2.47 F easy	Sol in H ₂ O.
1.683 ^ 1.692	1.677	<u>1.685</u>	1.708	.031	DIOPSIDE (Pyroxene grp) Ca(Mg,Fe)Si ₂ O ₆	59° r > v wk	Y = $\frac{b}{c}$ Z:c = 39°	MCL pris \underline{c}	110 good at 87°	Cols to green	H 5-6 G 3.28 infus	Insol in acids. FeO 5.6, MnO 0.5, Fe ₂ O ₃ 3.2, Al ₂ O ₃ 2.2%.
1.669 ^	1.659	<u>1.686</u>	1.785	.126	NENADKEVICHITE (Na,Ca,K)(Nb,Ti)Si ₂ O ₆ (O,OH)·2H ₂ O	46° (58±2°)	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH	001 poor	Rose to brown	H 5 G 2.86 F diff	Diss by H ₂ SO ₄ . Pleoc wk, X cols, Y pale yellow, Z pale rose.
v 1.695	1.683	<u>1.687</u>	1.718	.035	BORNEMANITE BaNa ₄ Ti ₂ NbSi ₄ O ₁₇ (F,OH)·Na ₃ PO ₄	40°	X = $\frac{c}{a}$ Z = $\frac{a}{b}$ el pos	ORTH plates	001 perf	Pale yellow	H 3.5-4.5 G 3.48	Dec by acids with sepn of silica. Pleoc wk, X and Y cols, Z brownish.

(1.664) ^	1.678	<u>1.687</u>	1.705	.027	ROSEBUSCHITE (Ca,Na) ₃ (Zr,Ti)Si ₂ O ₈ F	78°	X ~ $\frac{c}{e}$ el pos	TCL	010 perf	Yellow, brown	H 5-6 G 3.35 F easy	Gel with acids. Pleoc wk. Y cols, Z yellow, abs Z > Y > X.
	1.684	<u>1.688</u>	1.705	.021	PENIKISITE Ba(Mg,Fe) ₂ Al ₂ (PO ₄) ₃ (OH) ₃	56° r > v str	X ~ $\frac{b}{Z:c} = -6°$	TCL	010, 100 fair to good	Blue to green	H 4 G 3.79	Pleoc, X grass-green, Y blue-green, Z pale pink, abs X ~ Y > Z.
1.681 ^	1.680	<u>1.688</u>	1.700	.020	FERRO-GEDRITE (Amphibole grp) (Fe,Mg) ₅ Al ₂ (Si ₆ Al ₂) ₀ ₂₂ (OH) ₂	83°	Y = $\frac{b}{Z} = \frac{c}{e}$ el pos	ORTH pris $\frac{c}{e}$	210 perf at 126°	Brown	H 6 G 3.37 F 6	Insol in acids. Pleoc, X light brown, Z pale blue-green. FeO 26.2, Fe ₂ O ₃ 3.6, Al ₂ O ₃ 17.7%.
1.670 ^ 1.705	1.686	<u>1.688</u>	1.699	.013	PUMPELLYITE Ca ₂ (Mg,Fe)Al ₂ (SiO ₄) (Si ₂ O ₇)(OH) ₂ ·H ₂ O	50° r < v str	Y = $\frac{b}{Z:c} = 32°$	MCL	001 good 100 imperf	Bluish- green	H 5.5-7 G 3.15	Insol in acids. Pleoc, X and Z pale yellow-green, Y green. FeO 3.0, Fe ₂ O ₃ 1.7%.
1.679 ^	1.688	<u>1.689</u>	1.695	.007	TRIPHYLITE Li(Fe,Mn)PO ₄	Small r < v str	Z = $\frac{b}{c}$	ORTH	100 perf 010 less so	Grayish- green	H 5 G 3.33 F 2	Diss by acids. FeO 29.6, MnO 14.6, MgO 1.1%.
1.709 v	1.675	<u>1.690</u>	1.735	.060	LEGRANDITE Zn ₂ (AsO ₄)(OH)·H ₂ O	36° (61+4°) r < v dist	X = $\frac{b}{Z:c} = -38°$	MCL pris	100 fair	Yellow, cols	H 5 G 3.98	Diss by acids. Pleoc faint in yellow.
1.672 ^	1.682	<u>1.690</u>	1.698	.016	CHLOROPHOENICITE (Mn,Mg) ₃ Zn ₂ (AsO ₄) (OH,O) ₆	83° r > v str	Y = $\frac{b}{c}$	MCL el $\frac{b}{c}$	100 good	Gray- green	H 3.5 G 3.46 F diff	Diss by acids. Color pink to purplish-red in str artificial light.
1.722 v	1.663	<u>1.691</u>	1.716	.053	ASTROPHYLLITE (K,Na) ₃ (Fe,Mn) ₇ (Ti,Nb) ₂ Si ₈ O ₂₄ (O,OH) ₇	Large r > v str	X ~ $\frac{b}{Z:c} = 3-19°$	TCL	001 perf	Yellow, brown	H 3 G 3.2 F 3	Dec by acids. Pleoc in yellow, abs X > Y > Z.
	1.686	<u>1.691</u>	1.708	.022	MARSTURITE Na ₂ Ca ₂ Mn ₆ Si ₁₀ O ₂₈ (OH) ₂	60°	---	TCL pris	110, 001 imperf	White to pale pink	H 6 G 3.46	MnO 34%.
	1.688	<u>1.691</u>	1.696	.008	FERROWYLLIEITE (Na,Ca,Mn)(Fe ⁺² ,Mn) (Fe ⁺² ,Fe ⁺³ ,Mg)Al(PO ₄) ₃	50° r < v str	---	MCL	010 perf 101 dist	Bluish- green	H 4	Pleoc, X smoky blue- grey, Y bluish-green, Z green.
1.685 ^ 1.699	1.687	<u>1.692</u>	1.715	.028	DIOPSIDE (Pyroxene grp) Ca(Mg,Fe)(Si,Al) ₂ O ₆	51°	Y = $\frac{b}{Z:c} = 42°$	MCL	110 good at 87°	Green, brown	H 6 G 3.33	Insol in acids. FeO 9.5, MnO 0.2, Fe ₂ O ₃ 1.6, Al ₂ O ₃ 2.3%.
	1.691	<u>1.692</u>	1.713	.022	YODERITE (Mg,Al) ₈ Si ₄ (O,OH) ₂₀	25-30°	X:a = 9°	MCL	Parting 001	Purple, emerald green	H 6 G 3.39	Pleoc, X pale blue, Y indigo, Z light olive-green. Y and Z yellow, X green.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
ν 1.703	1.691	<u>1.692</u>	1.700	.009	ZOISITE (Epidote grp) $\text{Ca}_2\text{Al}_3\text{Si}_3\text{O}_{12}(\text{OH})$	54° $r > v$ str	$X = \frac{c}{b}$ $Y = \frac{b}{a}$	ORTH	010, 100 good but diff, fr conch	Violet-blue to cols	H 6 G 3.35 F 3.5	Insol in acids, but gel after being heated. Abn interf colors. Pleoc u wk, X violet, Y deep blue, Z yellow-green.
	1.687	(<u>1.692</u>)	1.698	.011	TINZENITE (Axinite grp) $(\text{Ca}, \text{Mn}, \text{Fe})_3\text{Al}_2\text{BSi}_4\text{O}_{15}(\text{OH})$	75-85° $r < v$	---	TCL	100 good	Yellow	H 7 G 3.3 F 2	Insol in acids. Anom disp, deep blue, red-brown. CaO 17.0, MnO 14.8, Fe_2O_3 2.8, MgO 0.7%.
	1.672	<u>1.693</u>	1.710	.038	JAGOWERITE $\text{BaAl}_2(\text{PO}_4)_2(\text{OH})_2$	80° (97+6°)	---	TCL	100, 011 good 021 fair	Light green	H 4.5 G 4.01	Insol in acids. Fluor greenish-white in short-wave UV.
ν 1.719	1.686	<u>1.693</u>	1.714	.028	FASSAITE (Pyroxene grp) $\text{Ca}(\text{Mg}, \text{Fe}^{+3}, \text{Al})(\text{Si}, \text{Al})_2\text{O}_6$	56° $r > v$ mod	$Y = \frac{b}{a}$ $Z:c = 43^\circ$	MCL pris	110 good at 87°	Green	H 6 G 3.33	Insol in acids. FeO 3.1, Fe_2O_3 2.1, Al_2O_3 13.4, TiO_2 0.5%.
ν 1.700	1.691	<u>1.693</u>	1.719	.028	MANGAN-NEPTUNITE $\text{KNa}_2\text{Li}(\text{Mn}, \text{Fe})_2\text{Ti}_2\text{Si}_8\text{O}_{24}$	36° $r < v$ str	$Y = \frac{b}{a}$ $Z:c = 16-20^\circ$ el pos	MCL pris c	110 perf at 80°	Black to deep red	H 5-6 G 3.20 F 2.5	Insol in acids. Pleoc str, X pale yellow, Y brownish, Z orange-red, abs $Z > Y > X$.
ν 1.727	1.692	<u>1.695</u>	1.710	.018	BJAREBYITE $(\text{Ba}, \text{Sr})(\text{Mn}, \text{Fe}, \text{Mg})_2\text{Al}_2(\text{PO}_4)_3(\text{OH})_3$	~ 35° $r > v$ str	---	MCL spears	010, 100 perf	Emerald-green	H 4+ G 3.95	Pleoc wk, gray-tan to pale yellow-green. MnO 9.6, FeO 8.7%.
1.687 \wedge	1.682	<u>1.695</u>	1.720	.038	BORNEMANITE $\text{BaNa}_4\text{Ti}_2\text{NbSi}_4\text{O}_{17}(\text{F}, \text{OH}) \cdot \text{Na}_3\text{PO}_4$	66°	$X = \frac{c}{b}$ $Z = \frac{a}{b}$	ORTH platy	001 perf	Pale yellow	H 3.5-4.5 G 3.48	Dec by acids with sepn of silica. Pleoc wk, X and Y cols, Z brownish.
ν 1.711	1.695	<u>1.696</u>	1.715	.020	BEUSITE $(\text{Mn}, \text{Fe}, \text{Ca}, \text{Mg})_3(\text{PO}_4)_2$	25° $r > v$ str	$X = \frac{b}{a}$ $Z:c = -36^\circ$	MCL u mass	010 good 100 fair	Salmon-pink to reddish-brown	H 5 G 3.7 F 2	Diss by acids. MnO 24.8, FeO 16.8, CaO 12.2, MgO 2.4%.
1.692 \square 1.702	1.691	<u>1.696</u>	1.703	.012	BARYLITE $\text{BaBe}_2\text{Si}_2\text{O}_7$	60-81° $r > v$ wk	$X = \frac{b}{a}$ $Y = \frac{a}{b}$	ORTH	100 perf 210, 010	Cols, pink	H 6-7 G 4.02 infus	Insol in acids.

	1.691	<u>1.696</u>	1.725	.034	BRABANTITE $\text{CaTh}(\text{PO}_4)_2$	44°	---	MCL	---	Yellow, brown, green	G 5.20	Diss by HCl. Compare Cheralite, Monazite.
1.637 ^ 1.883	1.693	(1.697)	1.718	.025	BARITE (plumboan) $(\text{Ba,Pb})\text{SO}_4$	~ 50° $r < v$	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH	001 perf 210 less so	White	H 3 G ~ 5 F 2.5	Insol in acids.
1.670 ^	1.687	<u>1.698</u>	1.825	.138	LABUNTSOVITE $(\text{K,Ba,Na})(\text{Ti,Nb})$ $(\text{Si,Al})_2(\text{O,OH})_7 \cdot \text{H}_2\text{O}$	20° (35+3°) $r > \bar{v}$	$Y = \frac{b}{c}$ $Z:\underline{c} = 63^\circ$	MCL	T02 perf	Dark brown	H 5 G 2.87	Pleoc wk, X and Y cols, Z pale yellow.
	1.695	<u>1.698</u>	1.733	.038	EUCHROITE $\text{Cu}_2(\text{AsO}_4)(\text{OH}) \cdot 3\text{H}_2\text{O}$	29° $r > v$ mod	$X = \frac{c}{b}$ $Z = \frac{b}{c}$	ORTH pris \underline{c}	110, 101	Emerald- to leek- green	H 3.5-4 G 3.44 F 2-2.5	Diss by acids. Bright bluish-green in section, pleoc faint or absent.
1.684 ^ 1.714	1.696	<u>1.698</u>	1.721	.025	PIGEONITE (Pyroxene grp) $(\text{Mg,Fe,Ca})(\text{Mg,Fe})\text{Si}_2\text{O}_6$	25°	$X = \frac{a}{b}$ $Z:\underline{c} = 41^\circ$	MCL	110 good at 87°	Cols to green	H 5-6 G 3.38 infus	Insol in acids. Pleoc wk. FeO 21.4, Fe ₂ O ₃ 0.8, MnO 0.4, CaO 8.5%.
1.692 ^ 1.707	1.690	<u>1.699</u>	1.721	.031	DIOPSIDE, manganian (Pyroxene grp) $\text{Ca}(\text{Mg,Mn})\text{Si}_2\text{O}_6$	60° $r > v$ wk	$Y = \frac{b}{c}$ $Z:\underline{c} = 43^\circ$	MCL pris \underline{c}	110 good at 87°	Brown	H 6 G 3.3 infus	Insol in acids. MnO 10.0, FeO 1.65, Fe ₂ O ₃ 1.5%.
1.681 ^	1.689	<u>1.699</u>	1.718	.029	OMPHACITE (Pyroxene grp) $(\text{Ca,Na})(\text{Mg,Fe,Al})\text{Si}_2\text{O}_6$	64° (73+8°) $r < v$ mod	$Y = \frac{b}{c}$ $Z:\underline{c} = 41^\circ$	MCL pris \underline{c}	110 good at 87°	Green	H 6 G 3.35 F 5	Insol in acids. Pleoc wk in green. FeO 5.9, Fe ₂ O ₃ 3.8, Al ₂ O ₃ 6.1, CaO 20.6, Na ₂ O 1.7%.
\bar{v} 1.723	1.695	<u>1.699</u>	1.719	.024	GRAFTONITE $(\text{Fe,Mn,Ca})_3(\text{PO}_4)_2$	50° $r > v$ str	$X = \frac{b}{c}$ $Z:\underline{c} = -36^\circ$	MCL u mass	010 good 100 fair	Salmon- pink to red-brown	H 5 G 3.7 F 2	Diss by acids. FeO 28.8, MnO 16.0, CaO 12.8%.
	---	<u>1.70</u>	---	.044	PENNAITE Silicate of Na, Ca, Fe, Ti, Zr	25°	$Y:\underline{c} = 13^\circ$	TCL(?) acic	---	Yellow to light brown	---	Pleoc, X dark yellow, Y grayish-yellow, Z bright-yellow.
1.693 ^ 1.721	1.697	<u>1.700</u>	1.725	.028	MANGAN-NEPTUNITE $\text{KNa}_2\text{Li}(\text{Mn,Fe})_2\text{Ti}_2\text{Si}_8\text{O}_{24}$	36° $r < v$ str	$Y = \frac{b}{c}$ $Z:\underline{c} = 16-20^\circ$ el pos	MCL pris \underline{c}	110 perf at 80°	Red to black	H 5.5-6 G 3.2 F 2.5	Insol in acids. Pleoc str, X yellow, Y orange-red, Z deep red.
\bar{v} (1.722)	1.695	<u>1.70</u>	1.722	.027	GADOLINITE $\text{Y}_2\text{FeBe}_2\text{Si}_2\text{O}_{10}$	(52+11°)	$X = \frac{b}{c}$ $Z:\underline{c} = 4-13^\circ$	MCL	Conch	Black to dark brown	H 7 G 4.2 infus	Gel in part. Pale green, not pleoc in section.
1.676 ^	1.697	(1.700)	1.704	.007	PHARMACOSIDERITE $\text{KFe}^+3_4(\text{AsO}_4)_3(\text{OH})_4 \cdot 6-7\text{H}_2\text{O}$	Large $r < v$ str	---	CUB	100 imperf	Olive- green to brown	H 2.5 G 2.8-2.9 F 2	Diss by acids. Poly tw. Abnorm interf colors.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.692 ^	1.700	1.703	1.708	.008	ZOISITE (Epidote grp) $\text{Ca}_2\text{Al}_3(\text{SiO}_4)_3(\text{OH})$	71° $r < v$ or $r > v$	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH	010 perf but diff 100	Cols, green	H 6-7 G 3.35 F 3.5	Insol in acids, but gel after ignition. Abnorm interf colors. Fe_2O_3 3.1, FeO 1.3, MnO 0.8%.
1.717 v	1.700	1.703	1.706	.006	CLINOZOISITE (Epidote grp) $\text{Ca}_2\text{Al}_3(\text{SiO}_4)_3(\text{OH})$	~ 90° $r < v$ str	$Y = \frac{b}{c}$ $X \sim \frac{c}{b}$	MCL	001 perf	Cols	H 6.5 G 3.28 F 3	Insol in acids. Abnorm interf colors. Fe_2O_3 3.2, FeO 0.6%.
1.741 v (1.741)	1.701	1.703	1.706	.005	SERENDIBITE $\text{Ca}_2(\text{Mg},\text{Al})_6(\text{Si},\text{Al},\text{B})_6\text{O}_{20}$	~ 90° $r < v$ str	ext:tw lam ~ 40°	TCL platy	---	Sky-blue to indigo- blue	H 6.5 G 3.42 infus	Insol in acids. Poly tw. Pleoc, X pale green, Y nearly cols, Z deep blue.
1.728 v	1.694	1.704	1.719	.025	ROSELITE $\text{Ca}_2(\text{Co},\text{Mg})(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	75° $r < v$ perc	$Y = \frac{b}{c}$ $X:\frac{c}{b} = 0-4^\circ$	MCL	010 perf	Pale rose	H 3.5 G 3.69 F 3	Diss by acids. Pleoc wk in rose, abs $X > Y > Z$.
	1.698	1.704	1.720	.022	AKATOREITE $\text{Mn}_9(\text{Si},\text{Al})_{10}\text{O}_{23}(\text{OH})_9$	65.5°	---	TCL pris	010 perf 012 imperf	Orange- brown	H 6 G 3.48	Pleoc, X cols, Y pale yellow, Z light canary-yellow.
1.683 ^	1.696	1.704	1.713	.017	ZWIESELITE (Triplite ser) $(\text{Fe},\text{Mn},\text{Mg},\text{Ca})_2\text{PO}_4(\text{OH},\text{F})$	87° $r > v$	$Y = \frac{b}{c}$ $X:\frac{a}{b} = 42^\circ$	MCL	100 perf 010 poor	Dark brown	H 4.5 G 3.97 F 2.5	Diss by acids. Pleoc in brown, abs $X > Y > Z$. FeO 42.0, MnO 18.4, MgO 0.8, CaO 1.7%.
1.727 v (1.727)	1.703	1.704	1.710	.007	SAPPHIRINE $(\text{Mg},\text{Al})_8(\text{Al},\text{Si})_6\text{O}_{20}$	Med	$Y = \frac{b}{c}$	MCL	010, 001, 100 poor	Pale blue	H 7.5 G 3.44 infus	Insol in acids. Pleoc in blue, abs $Z > Y > X$.
	1.703	1.705	1.723	.020	KULANITE $\text{Ba}(\text{Fe},\text{Mn},\text{Mg})_2\text{Al}_2(\text{PO}_4)_3(\text{OH})_3$	32° $r > v$ str	$Y = \frac{b}{c}$ $Z:\frac{c}{b} = -8^\circ$	TCL plates	010 100 fair to good	Blue to green	H 4 G 3.91	Pleoc, X brownish- green, Y green, Z pale brown.
1.688 v 1.717	1.698	1.705	1.715	.017	PUMPELLYITE $\text{Ca}_2(\text{Mg},\text{Fe})\text{Al}_2(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{OH})_2 \cdot \text{H}_2\text{O}$	68° $r < v$ str rarely $r > v$	$Y = \frac{b}{c}$ $Z:\frac{c}{b} = 5-22^\circ$	MCL	001 good 100 imperf	Bluish- green	H 5.5-7 G 3.2	Insol in acids. Pleoc, X and Z yellow-green, Y green. Fe_2O_3 3.6, FeO 2.9%.
1.699 v 1.714	1.700	1.707	1.724	.024	SALITE (Diopside ser, Pyroxene grp) $\text{Ca}(\text{Mg},\text{Fe})(\text{Si},\text{Al})_2\text{O}_6$	52° (66±10°)	$Y = \frac{b}{c}$ $Z:\frac{c}{b} = 44^\circ$	MCL pris c	110 good at 87°	Green, brown	H 6 G 3.42	Insol in acids. FeO 9.8, MnO 0.3, Fe_2O_3 1.8, Al_2O_3 2.8%.

1.707 neg ∧ 1.730	1.689	<u>1.707</u>	1.727	.038	GAITITE $\text{Ca}_2(\text{Zn,Mg})(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	85° (92±6°)	---	TCL	010, 001, 011 good	Cols, white	H ~ 5	Zn:Mg = 1:1.
	1.703	<u>1.708</u>	1.729	.026	OXYCHILDRENITE $(\text{Mn}^{+2}, \text{Mg})(\text{Fe}^{+3}, \text{Mn}^{+3})_4$ $\text{Al}_4(\text{PO}_4)_4(\text{OH})_{14}$	32° (53±11°)	---	ORTH(?)	001 perf 010 less so	Reddish- brown	H 3.5-4 G 3.22 fus	Diss by acids. Pleoc, pale yellow to brownish. Fe_2O_3 18.5, Mn_2O_3 8.7, MnO 4.5%.
1.711 neg ∧ 1.744	1.704	<u>1.708</u>	1.714	.010	ALLANITE (Epidote grp) $(\text{Ca,Ce})_2(\text{Al,Fe})_3(\text{SiO}_4)_3$ (OH)	70-80° $r > v$	$Y = \frac{b}{c}$ $X:\underline{c} = \frac{5-}{40^\circ}$	MCL	001 imperf	Brownish- black	H 5-6 G 3.53 F 3	Slowly attacked by HCl and gel. G and refr indices decrease with hydration. U opt neg.
1.690 ^	1.702	<u>1.709</u>	1.740	.038	LEGRANDITE $\text{Zn}_2(\text{AsO}_4)(\text{OH}) \cdot \text{H}_2\text{O}$	36-50° $r < v$ dist	$X = \frac{b}{c}$ $Z:\underline{c} = -38^\circ$	MCL pris	100 fair	Cols to yellow	H 4.5 G 4.0	Diss by acids. Pleoc faint in yellow.
1.660 ∧ 1.733	1.708	<u>1.710</u>	1.746	.038	STRENGITE, manganian $(\text{Fe}^{+3}, \text{Mn}^{+3})(\text{PO}_4) \cdot 2\text{H}_2\text{O}$	42° $r > v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	---	Reddish- brown	H 4-5 G 2.8 fus	Diss by acids.
1.699 ∧ 1.719	1.700	<u>1.710</u>	1.733	.033	JOHANNSENITE (Pyroxene grp) $\text{Ca}(\text{Mn,Mg,Fe})\text{Si}_2\text{O}_6$	60° (68±7°) $r > v$	$Y = \frac{b}{c}$ $Z:\underline{c} = 49^\circ$.	MCL pris c	110 good at 87°	Brown	H 6 G 3.38 F 5	Insol in acids. MnO 22.6, FeO 1.5, Fe_2O_3 0.8%.
	1.707	<u>1.710</u>	1.730	.023	NAMBULITE $\text{NaLi}(\text{Mn,Mg})_8\text{Si}_{10}\text{O}_{28}$ (OH) ₂	30° $r > v$ wk	$X':\underline{c} = 19^\circ$ on 010	TCL	001 perf 100, 010 dist	Reddish- brown	H 6.5 G 3.51	---
1.720 ∧	1.702	<u>1.710</u>	1.723	.021	MERWINITE $\text{Ca}_3\text{Mg}(\text{SiO}_4)_2$	70° $r > v$ wk	$Z = \frac{b}{c}$ $X:\underline{c} = 36^\circ$	MCL gran- ular	010 poor	Cols	H 6 G 3.32 F 6	Gel with acid. Poly tw very common, 2 sets with angle 43° between sets.
1.696 ^	1.708	<u>1.711</u>	1.723	.015	BEUSITE $(\text{Mn,Fe,Ca})_3(\text{PO}_4)_2$	45° $r > v$ str	$X = \frac{b}{c}$ $Z:\underline{c} = -36^\circ$	MCL u mass	010 good 100 fair	Reddish- brown	H 5 G 3.70 F 2	Diss by acids. MnO 33.7, FeO 17.7, CaO 3.3, MgO 2.3%.
	1.709	<u>1.711</u>	1.724	.015	BRANDTITE $\text{Ca}_2\text{Mn}(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	23° $r < v$ str	$X = \frac{b}{c}$ $Y:\underline{c} = 7^\circ$	MCL pris	010 perf 001 good	Cols	H 3.5 G 3.67 F 2.5-3	Diss by acids. Tw common 100.
1.772 ∧	1.707	<u>1.712</u>	1.732	.025	MANGANHUMITE (Humite grp) $(\text{Mn,Mg})_7(\text{SiO}_4)_3(\text{OH})_2$	37° (54±11°) $r > v$ perc	$X = \frac{a}{b}$ $Z = \frac{b}{c}$	ORTH gran- ular	010	Pale to deep orange- brown	H 4 G 3.83	Gel with acid. MnO 57.1, MgO 14.2, FeO 1.0%.
	1.709	<u>1.712</u>	1.716	.007	OTTRELITE $(\text{Mn,Fe,Mg})_2\text{Al}_4\text{Si}_2\text{O}_{10}$ (OH) ₄	60-70° $r > v$ str	$X = \frac{b}{c}$ $Z:\underline{c} = 10^\circ$ el neg	MCL and TCL	001 perf	Pistachio green	H 6 G 3.52	Pleoc, X olive- yellow, Y pale yellow, Z cols. MnO 12.6, FeO 8.0, MgO 4.4%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
v (1.731)	1.703	1.713	1.722	.019	GERHARDTITE Cu ₂ (NO ₃)(OH) ₃	Large r < v, very str	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH hori- zontal vicinal stria- tions	001 perf 110 good	Emerald- green	H 2 G 3.42 F 2	Diss by acids. Pleoc, X and Y green Z blue.
	1.708	(1.713)	1.722	.014	VARULITE (Na,Ca)Mn(Mn,Fe ⁺² , Fe ⁺³) ₂ (PO ₄) ₃	Large r > v	---	MCL	001, 010	Olive- green to brown	H 5 G 3.5-3.6 fus	Diss by acids. Pleoc, X yellow- green, Z grass-green.
1.707 ^ 1.732	1.708	1.714	1.736	.028	FERROSALITE (Diopside ser, Pyroxene grp) Ca(Fe,Mg)Si ₂ O ₆	70° (56+10°) r > v wk	Y = $\frac{b}{c}$ Z:c = 43°	MCL pris c	110 good at 87°	Dark green	H 6 G 3.41 F 4	Insol in acids. FeO 18.6, MnO 0.2, Fe ₂ O ₃ 0.5, Al ₂ O ₃ 1.1%.
1.698 ^	1.714	1.714	1.742	.028	PIGEONITE (Pyroxene grp) (Mg,Fe,Ca)(Mg,Fe)Si ₂ O ₆	0-12°	X = $\frac{a}{b}$ Z:c = 41°	MCL	110 good at 87°	Green	H 5-6 G 3.44 F 5-6	Insol in acids. FeO 27.8, MnO 1.0, CaO 3.8, Fe ₂ O ₃ 1.7%.
1.679 ^	1.707	1.714	1.734	.027	CLINOHUMITE (Humite grp) (Mg,Fe) ₉ (SiO ₄) ₄ (OH,F) ₂	56° r < v	Z = $\frac{b}{c}$ X ~ $\frac{c}{b}$	MCL	---	Yellow to red	H 6 G 3.36 infus	Gel with acids. FeO 10.2, Fe ₂ O ₃ 1.2, TiO ₂ 5.1, F 0.1%.
v 1.728	1.712	1.714	1.726	.014	CHLORITOID (Fe,Mg,Mn) ₂ Al ₄ Si ₂ O ₁₀ (OH) ₄	51° r > v str	X = $\frac{b}{c}$ Z:c = 11° el neg	MCL and TCL	001 perf	Greenish- gray	H 6.5 G 3.54 F diff	Dec by H ₂ SO ₄ . Pleoc X green, Y slate blue, Z yellowish. FeO 21.2, Fe ₂ O ₃ 1.6, MnO 0.4%.
v 1.722	1.700	1.715	1.740	.040	DIASPORE AlO(OH)	84° r < v wk	Y = $\frac{b}{c}$ Z = $\frac{a}{b}$	ORTH blades 010	010 perf 110 less so	Cols	H 6.5-7 G 3.44 infus	Insol in acids. Luster pearly. Fe ₂ O ₃ 0.1, TiO ₂ 0.2%.
	1.700	1.715	1.732	.032	EVEITE Mn ₂ (AsO ₄)(OH)	65° (87+7°) r < v mod	---	ORTH tab 100	011 fair	Apple- green	H 4 G 3.67	Diss by acids.
	1.707	1.715	1.730	.023	LARNITE Ca ₂ SiO ₄	Mod large	Z = $\frac{b}{c}$ X:c = 14°	MCL	100	White	---	Gel with acids. Poly tw on 100.
v 1.725	1.711	1.715	1.724	.013	RHODONITE (Mn,Ca,Mg,Fe)SiO ₃	69°	Z':c on 010 = 9° 110 = 30°	TCL	100, 001 perf 010 less so	Pink	H 6 G 3.56 F 3	Slightly attacked by acids. MnO 40.2, CaO 6.7, MgO 4.6%.

1.707 ^ 1.723	1.710	<u>1.716</u>	1.736	.026	FERROAUGITE (Pyroxene grp) (Ca,Na)(Fe,Mg,Al,Ti) (Si,Al) ₂ O ₆	52°	$Y = \frac{b}{c}$ $Z:c = 49^\circ$	MCL	110 good at 87°	Green, brown	H 6 G 3.49	Insol in acids. FeO 20.2, MnO 1.1, Fe ₂ O ₃ 0.9, Al ₂ O ₃ 3.5%.
	1.712	<u>1.716</u>	1.725	.013	BREDIGITE Ca ₇ Mg(SiO ₄) ₄	10-34° (68+18°)	---	ORTH	130	White	G 3.38	Gel with acids. Tw 110.
v 1.767	1.707	(<u>1.717</u>)	1.778	.071	JOAQUINITE Ba ₂ NaCe ₂ Fe(Ti,Nb) ₂ Si ₈ O ₂₆ (OH,F)·H ₂ O	45° r < v	$Z = \frac{c}{a}$ $X = \frac{a}{b}$	MCL ps orth	---	Brown	H 5.5 G 3.62 F 2.5	Insol in acids. Pleoc wk, X and Y cols, Z light yellow.
	1.709	<u>1.717</u>	1.729	.020	STRINGHAMITE CaCuSiO ₄ ·2H ₂ O	80°	$X = \frac{b}{c}$ $Y:c = 2.5^\circ$	MCL	---	Deep azure- blue	G 3.67 calc	Pleoc, X light gray- blue, Y light blue, Z dark blue.
1.705 ^	1.711	<u>1.717</u>	1.727	.016	PUMPELLYITE Ca ₂ (Mg,Fe)Al ₂ (SiO ₄) (Si ₂ O ₇)(OH) ₂ ·H ₂ O	Large r < v str	$Y = \frac{b}{c}$ $Z:c = 25^\circ$	MCL	001 good 100 imperf	Bluish- green	H 5.5-7 G 3.25	Insol in acids. Pleoc, X and Z yellow-green, Y green. Fe ₂ O ₃ 6.7, FeO 3.9, MnO 0.2%.
	1.715	<u>1.717</u>	1.728	.013	KANOITE (Pyroxene grp) (Mn,Mg) ₂ Si ₂ O ₆	41°	$Y = \frac{b}{c}$ $Z:c = 42^\circ$	MCL	110 perf	Light pinkish- brown	H 6 G 3.66	Poly tw 100. MnO 31.2, MgO 15.1, FeO 2.6, Fe ₂ O ₃ 0.4%.
	1.713	<u>1.717</u>	1.723	.010	JOHACHIDOLITE CaAlB ₃ O ₇	75-80° r > v str	$X = \frac{a}{c}$ $Y = \frac{c}{b}$	ORTH mass	---	White	H 7.5 G 3.44 fus	Insol in acids. Fluor blue in UV.
1.703 ^ 1.720 neg v 1.733	1.715	<u>1.717</u>	1.721	.006	CLINOZOISITE (Epidote grp) Ca ₂ Al ₃ Si ₃ O ₁₂ (OH)	~ 90° r < v str	$Y = \frac{b}{c}$ $X \sim \frac{c}{b}$ el cTv pos	MCL	001 perf	Cols to green	H 6.5 G 3.21 F 3	Insol in acids. Abnorm interf colors. Fe ₂ O ₃ 3.0, FeO 0.8%.
	1.710	<u>1.718</u>	1.780	.070	STRONTIOJOAQUINITE Sr ₂ Ba ₂ (Na,Fe ⁺²) ₂ Ti ₂ Si ₈ O ₂₄ (O,OH) ₂ ·H ₂ O	35-45° r > v str	---	MCL	001 good	Green to yellow- green	H 5.5 G 3.68	---
1.660 ^	1.697	<u>1.718</u>	1.741	.044	PLANCHEITE Cu ₈ Si ₈ O ₂₂ (OH) ₄ ·H ₂ O	~ 90°	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH fib	---	Blue	H 5 G 3.65-3.80	Slightly attacked by acids. Pleoc, X pale blue, Y and Z blue.
	1.709	<u>1.718</u>	1.734	.025	METAHOHMANNITE Fe ₂ (SO ₄) ₂ (OH) ₂ ·3H ₂ O	(94+9°)	---	Mass, powdery	---	Orange	---	Diss by HCl. Pleoc, X pale yellow, Y reddish-yellow, Z reddish-brown.
	1.713	<u>1.718</u>	1.728	.015	WICKSITE NaCa ₂ (Fe ⁺² ,Mn) ₄ MgFe ⁺³ (PO ₄) ₆ ·2H ₂ O	(71+16°)	---	ORTH	010 good	Dark blue	H 4.5-5 G 3.54	Pleoc, X blue, Y greenish-blue, Z pale yellow-brown, abs X = Y > Z.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.710 ^ 1.728	1.710	<u>1.719</u>	1.738	.028	JOHANNSENITE (Pyroxene grp) $\text{CaMnSi}_2\text{O}_6$	70°	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 48^\circ$	MCL pris c	110 good at 87°	Brown to green	H 6 G 3.45 F 5	Insol in acids. MnO 26.8, FeO 1.0, MgO 1.0, Fe ₂ O ₃ 0.3%.
1.693 ^ 1.736	1.712	<u>1.719</u>	1.736	.024	FASSAITE (Pyroxene grp) $\text{Ca}(\text{Mg}, \text{Fe}, \text{Al})(\text{Al}, \text{Si})_2\text{O}_6$	55° (66+10°) $r > v$	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 46^\circ$	MCL pris c	110 good at 87°	Green	H 6 G 3.34	Insol in acids. FeO 0.2, Fe ₂ O ₃ 6.1, Al ₂ O ₃ 15.75, TiO ₂ 0.8%.
1.710 ^	1.714	<u>1.720</u>	1.730	.016	MERWINITE $\text{Ca}_3\text{Mg}(\text{SiO}_4)_2$	68° $r > v$ wk	$Z = \frac{b}{c}$ $X:c = \frac{b}{c} = 36^\circ$	MCL gran- ular	010 poor	Cols	H 6 G 3.32 F 6	Gel with acids. Poly tw very common, 2 sets with angle 43° between the sets.
1.700 ^	1.711	<u>1.721</u>	1.744	.033	NEPTUNITE $\text{KNa}_2\text{Li}(\text{Fe}, \text{Mn})_2\text{Ti}_2\text{Si}_8\text{O}_{24}$	62° $r < v$ str	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 16-20^\circ$ el pos	MCL pris c	110 perf	Red	H 5.5-6 G 3.2 F 2.5	Insol in acids. Pleoc str, X gold-orange, Y orange-red, Z deep red.
	1.707	<u>1.721</u>	1.739	.032	LEUCOPHOSPHITE $\text{KFe}_2(\text{PO}_4)_2(\text{OH}) \cdot 2\text{H}_2\text{O}$	84° $r < v$ str	$X = \frac{b}{c}$ $Z:c = \frac{b}{c} = 26^\circ$	MCL	001 perf	Buff	G 2.95	Diss by acids. Pleoc, cols to greenish-yellow.
	1.713	<u>1.721</u>	1.734	.021	HARADAITE $\text{SrV}^{+4}\text{Si}_2\text{O}_7$	105° (77+12°) $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	010 perf 100, 001 dist	Bright green	H 4.5 G 3.80	Pleoc, X cols to pale green, Y cols to light yellow-green, Z bluish-green. (Reported as opt neg.)
	1.712	<u>1.721</u>	1.731	.019	ADELITE $\text{CaMg}(\text{AsO}_4)(\text{OH}, \text{F})$	~ 90° $r < v$ perc	$X = \frac{a}{b}$ $Z = \frac{b}{c}$	ORTH u mass	Uneven to conch	Cols, gray	H 5 G 3.73 F easy	Diss by acids.
1.691 ^ 1.746	(1.694)	<u>1.722</u>	1.757	(.063)	ASTROPHYLLITE $(\text{K}, \text{Na})_3(\text{Fe}, \text{Mn})_7\text{Ti}_2\text{Si}_8\text{O}_{24}(\text{O}, \text{OH})_7$	85° $r > v$ str	$X \sim \frac{b}{c}$ $Z:c = \frac{b}{c} = 3-19^\circ$ el clv neg	TCL	001 perf	Red-brown	H 3 G 3.3 F 3	Dec by acids. Pleoc, X red-orange, Y yellow, Z lemon-yellow, abs $X > Y > Z$.
1.715 ^	1.702	<u>1.722</u>	1.750	.048	DIASPORE $\text{AlO}(\text{OH})$	82° $r < v$ wk	$Y = \frac{b}{c}$ $Z = \frac{a}{b}$	ORTH blades 010, el c	010 perf 110 less so	Cols	H 6.5-7 G 3.44 infus	Insol in acids. Luster pearly.

1.671 ◇ 1.730	(1.71)	<u>1.722</u>	1.74	(.03)	ANDALUSITE, manganian (Al,Mn,Fe) ₂ SiO ₅	80° r < v	X = a Y = $\frac{b}{c}$ el pos	ORTH pris	110 good	Green	H 7 G 3.3 infus	Insol in acids. Pleoc, X yellow- green, Y deep green, Z golden. Mn ₂ O ₃ 21.4, Fe ₂ O ₃ 4.5%.
1.70 ◇ 1.774	1.715	(<u>1.722</u>)	1.740	.025	GADOLINITE Y ₂ FeBe ₂ Si ₂ O ₁₀	65°	X = b Z:c = 4- 13°	MCL	Conch	Greenish- black	H 6.5 G 4.2 infus	Gel in part with acids.
1.716 ◇ 1.730	1.714	<u>1.723</u>	1.774	.060	AUGITE, zincian (Pyroxene grp) (Ca,Na)(Mg,Mn,Zn,Fe) Si ₂ O ₆	74° (47+5°)	Y = b Z:c = 55°	MCL pris	110 good at 87°	Greenish- brown	H 6 G 3.55 F 5	Insol in acids. MgO 5.9, ZnO 7.7, MnO 7.2, Fe ₂ O ₃ 4.2%.
1.699 ^	1.718	<u>1.723</u>	1.745	.027	GRAFTONITE (Fe,Mn,Ca) ₃ (PO ₄) ₂	52° r > v str	X = b Z:c = -36°	MCL u mass	010 good 100 fair	Reddish- brown	H 5 G 3.75 F 2	Diss by acids. FeO 34.3, MnO 20.1, CaO 5.1%.
	1.712	<u>1.725</u>	1.760	.048	KRINOVITE NaMg ₂ CrSi ₃ O ₁₀	61°	X = b disp str	TCL	---	Deep emerald- green	H 5.5-7 G 3.38	Pleoc str, X yellow- green, Y blue-green, Z greenish-black to dark reddish-brown. Meteorite mineral.
1.65 ^	1.715	<u>1.725</u>	1.738	.023	HOMILITE Ca ₂ (Fe,Mg)B ₂ Si ₂ O ₁₀	80° r > v rather str	Z = b Y ~ $\frac{c}{a}$	MCL tab 001	One indist	Black to dark brown	H 5 G 3.36 F 2	Gel with acids. Pleoc, X bluish- green, Y brownish- red, Z deep smoky gray to brownish- yellow.
1.715 ◇ 1.728	1.722	<u>1.725</u>	1.734	.012	RHODONITE (Mn,Ca,Mg,Fe)SiO ₃	70°	---	TCL	100, 001 perf 010 less so	Pink	H 6 G 3.54 F 3	Nearly insol in acids. MnO 41.4, CaO 6.6, MgO 2.3, FeO 1.9%.
	(1.708)	<u>1.726</u>	1.758	(.050)	CESIUM KUPLETSKITE (Cs,K,Na) ₃ (Mn,Fe ⁺²) ₇ (Ti,Nb) ₂ Si ₈ O ₂₄ (OH,F) ₇	75°	Z = a Y:b = 10°	TCL	001 perf	Gold-brown	H 4 G 3.68	Pleoc, X yellow- green, Y yellow to brown, Z brown.
1.749 v	1.725	<u>1.726</u>	1.730	.005	TRIPLOIDITE (Mn,Fe) ₂ (PO ₄)(OH)	Med r > v extr	X = b Z:c = 4° disp dist	MCL pris	010 good 120 fair	Yellow- ish- to reddish- brown	H 4.5-5.5 G 3.70 F 1.5	Diss by acids. Cols in section. Mn:Fe about 3:1.
1.730 v	1.715	<u>1.727</u>	1.749	.034	BABINGTONITE Ca ₂ (Fe ⁺² ,Mn)Fe ⁺³ Si ₅ O ₁₄ (OH)	76° r > v str	ext:c on 100 = 44° 010 = 31°	TCL pris c	001 perf 010, 100 less so	Greenish- to brownish- black	H 5.5-6 G 3.34 F 3	Insol in acids. Pleoc str, X deep green, Y lilac-brown, Z pale to deep brown. Fe ₂ O ₃ 15.6, FeO 7.8, MnO 1.8%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.695 ^	1.724	<u>1.727</u>	1.749	.025	BJAREBYITE (Ba,Sr)(Mn,Fe,Mg) ₂ Al ₂ (PO ₄) ₃ (OH) ₃	38° r > v str	Z:c = 8° el clv pos	MCL spears	010, 100 perf	Emerald-green	H 4+ G 3.90	Pleoc wk, X apple-to olive-green, Y brownish-green, Z cols to pale yellow. MnO 16.0, FeO 5.9, Fe ₂ O ₃ 3.3%.
1.704 ^	1.725	(<u>1.727</u>)	1.732	.007	SAPPHIRINE (Mg,Al) ₈ (Al,Si) ₆ O ₂₀	66°	Y = \underline{b}	MCL	010, 001, 100 poor	Yellow	H 7.5 G 3.4 infus	Insol in acids. Pleoc, X pale yellow, Y green, Z pale orange.
1.704 ^	1.725	<u>1.728</u>	1.755	.030	ROSELITE Ca ₂ (Co,Mg)(AsO ₄) ₂ ·2H ₂ O	60° (37+11°) r < v	Y = \underline{b} X:c = 0-4°	MCL	010 perf	Dark rose	H 3.5 G 3.7 F 3	Diss by acids. Pleoc, X dark rose, Y pale rose, Z paler rose.
1.719 △	1.716	<u>1.728</u>	1.745	.029	JOHANNSENITE, ferroan (Pyroxene grp) Ca(Mn,Fe,Mg)Si ₂ O ₆	64° (81+8°)	Y = \underline{b} Z:c = 48°	MCL pris c	110 good at 87°	Pale green	H 6 G 3.55	Insol in acids. MnO 14.1, FeO 13.4, MgO 2.2%.
1.725 ◇ 1.736	1.725	<u>1.728</u>	1.737	.012	RHODONITE (Mn,Fe,Ca,Mg)SiO ₃	70°	---	TCL	100, 001 perf 010 less so	Brownish-pink	H 4 G 3.65 F 3	Nearly insol in acids. MnO 29.2, FeO 14.5, CaO 6.6, MgO 1.9%.
1.714 △ 1.734	1.727	<u>1.728</u>	1.733	.006	CHLORITOID (Fe,Mg,Mn) ₂ Al ₄ Si ₂ O ₁₀ (OH) ₄	53° r > v str	X = \underline{b} Z:c = 12° el neg	MCL and TCL	001 perf	Greenish-gray	H 6.5 G 3.5 F diff	Dec by H ₂ SO ₄ . Pleoc, X green, Y blue, Z yellowish. FeO 27.1, MnO 0.2, Fe ₂ O ₃ 1.2%.
	1.696	<u>1.730</u>	1.798	.102	OJUELAITE ZnFe ⁺³ ₂ (AsO ₄) ₂ (OH) ₂ ·4H ₂ O	(73+2°)	X = \underline{b} Z = \underline{c}	MCL fib	---	Chartreuse	H 3 G 3.39	Sol in acids.
	1.715	<u>1.730</u>	1.795	.080	RUTHERFORDINE (UO ₂)CO ₃	53°	X = \underline{b} Y = \underline{c}	ORTH	010 perf	Yellow	G 5.7 infus	Diss by acids with eff. Pleoc, X nearly cols, Y pale yellow, Z pale greenish-yellow.
1.727 ◇ 1.734	1.716	<u>1.730</u>	1.746	.030	MANGANBABINGTONITE Ca ₂ (Mn,Fe) ₂ Fe ⁺³ Si ₅ O ₁₄ (OH)	80° r > v str	Z = \underline{b} Y ~ \underline{c}	TCL pris c	001 perf 010, 100 less so	Reddish-brown	H 6 G 3.45 F 3	Insol in acids. Pleoc, X green, Y pale rose, Z rose-brown. MnO 7.9, FeO 4.5, Fe ₂ O ₃ 12.3%.

1.723 ^ 1.741	1.725	<u>1.730</u>	1.750	.025	AUGITE, titanian (Pyroxene grp) (Ca,Na)(Mg,Fe,Al,Ti) (Si,Al) ₂ O ₆	40° (54+11°)	$Y = \frac{b}{c}$ $Z:c = 45^\circ$	MCL	110 good at 87°	Brown	H 6 G 3.42	Insol in acids. FeO 5.0, MnO 0.2, Fe ₂ O ₃ 4.4, Al ₂ O ₃ 7.35, TiO ₂ 5.5%.
1.722 ^	1.702	<u>1.730</u>	1.823	.121	KANONAITE (Andalusite ser) (Mn ⁺³ ,Al)AlSiO ₅	53° (60+2°) disp str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	110 poor	Greenish- black	H 6.5 G 3.40 calc	Pleoc, X yellow- green, Y bluish- green, Z deep golden yellow. Mn ₂ O ₃ 32.2%.
1.707 ^	1.713	<u>1.730</u>	1.748	.035	GAITITE Ca ₂ Zn(AsO ₄) ₂ ·2H ₂ O	88°	---	TCL	010, 001, 011 good	White	H 5 G 3.81	ZnO 15.3, MgO 1.2%.
1.778 ^	1.730	(<u>1.730</u>)	1.746	.016	MONAZITE, sulfatian (Ce,La,Th)(PO ₄ ,SO ₄)	5°	$X = \frac{b}{c}$	MCL	100 dist	Reddish- brown	G 4.54	SO ₃ 3.1%.
1.735 ^	1.727	<u>1.730</u>	1.735	.008	CARBOIRITE FeAl ₂ GeO ₅ (OH) ₂	55-70°	$Z:c = 4-$ 14° $r > v$	TCL	001 perf	Pale to bright green	---	Poly tw. Pleoc, X blue-green, Y light blue, Z cols to yellowish.
(1.713) ^ (1.766)	1.727	(<u>1.731</u>)	1.738	.011	ALLUAUDITE (Varulite ser) (Na,Ca) ₄ Fe ⁺² ₄ (Mn,Fe ⁺² , Fe ⁺³ ,Mg) ₈ (PO ₄) ₁₂	Large $r > v$	---	MCL	001	Greenish- black	H 5 G 3.5 fus	Diss by acids. Pleoc in green. "Huehner- kobelite."
	1.714	<u>1.732</u>	1.805	.091	LOPEZITE K ₂ Cr ₂ O ₇	50° (55+3°) $r > v$ med	---	TCL	010 perf 100, 001 dist	Orange- red	H 2.5 G 2.69	Sol in H ₂ O. Pleoc, X reddish-yellow, Y yellow, Z greenish- yellow.
1.714 ^ 1.753	1.726	<u>1.732</u>	1.753	.027	HEDENBERGITE (Diopside ser, Pyroxene grp) Ca(Fe,Mg)Si ₂ O ₆	49° $r > v$ wk	$Y = \frac{b}{c}$ $Z:c = 44^\circ$	MCL	110 good at 87°	Green	H 6 G 3.48 F 4	Insol in acids. Pleoc, X pale green, Z dark green. FeO 24.7, MnO 0.7, Fe ₂ O ₃ 1.9%.
1.742 ^	1.728	<u>1.732</u>	1.746	.018	PYROXMANGITE (Mn,Fe,Mg)SiO ₃	42° (57+14°)	---	TCL	---	Pink to dark brown	H 6 G 3.61	Insol in acids. MnO 45.5, FeO 1.3, MgO 3.8, CaO 0.7, Fe ₂ O ₃ 0.3%.
1.79 ^	1.720	<u>1.733</u>	1.935	.215	FERRIMOLYBDITE Fe ₂ (MoO ₄) ₃ ·8H ₂ O (?)	28° $r < v$ str	$Y = \frac{a}{b}$ $Z = \frac{c}{b}$	ORTH	001 dist fib <u>c</u>	Sulfur- yellow	H 1-2 G 2.99 F easy	Diss by acids, dec by NH ₄ OH. Pleoc, X and Y pale yellow, Z dirty gray to yellow- green.
	1.728	<u>1.733</u>	1.790	.062	BATISITE BaNa ₂ Ti ₂ (Si ₂ O ₇) ₂	8° (34+6°) $r < v$ str	$Y = \frac{b}{c}$ $Z = \frac{c}{b}$	ORTH	100 fair	Dark brown	H 6 G 3.43 F easy	Insol in acids. Pleoc, X cols, Y yellow-brown, Z reddish-brown.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.710 ^	1.730	1.733	1.763	.033	STRENGITE $\text{FePO}_4 \cdot 2\text{H}_2\text{O}$	Med (36+11°) $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	---	Rose to violet	H 4-5 G 2.8 fus	Diss by acids. Pleoc wk, in rose, abs Z > X.
1.717 ^	1.723	1.733	1.755	.032	MUKHINITE (Epidote grp) $\text{Ca}_2\text{Al}_2\text{V}(\text{SiO}_4)_3(\text{OH})$	88° (69+8°)	$Y = \frac{b}{c}$ $Z : a = \frac{b}{c} = 32^\circ$	MCL	001 perf 100 less so	Brownish-black	H 8	Diss with diff by acids. Pleoc str, X pale olive-green, Y pale red-brown, Z reddish-brown.
	1.557	1.734	2.07	.513	IDRIALITE Hydrocarbon, near $\text{C}_{22}\text{H}_{14}$	84° $r > v$ wk	$X = \frac{b}{c}$ $Z = \frac{c}{a}$ el cTv neg	ORTH	001 perf 100 poor	Yellow, greenish	H 1-1.5 G 1.22 F 1	Dimethylbenzphenanthrene. Pleoc, X pale yellow, Y and Z yellow.
1.758 v	1.719	(1.734)	1.778	.059	SEIDOZERITE $(\text{Na}, \text{Ca})_2(\text{Zr}, \text{Ti}, \text{Mn})_2\text{Si}_2\text{O}_7(\text{O}, \text{F})_2$	62° $r > v$ str	$X = \frac{b}{c}$ $Y : c = \frac{b}{c} = 13^\circ$	MCL	001 perf	Dark yellow to reddish-brown	H 4-5 G 3.43 F easy	Diff diss by HCl. Pleoc, X dark red, Y red, Z yellow.
1.730 ^	1.719	1.734	1.750	.031	BABINGTONITE $\text{Ca}_2(\text{Fe}^{+2}, \text{Mn})\text{Fe}^{+3}\text{Si}_5\text{O}_{14}(\text{OH})$	88° $r > v$ str	---	TCL pris c	001 perf 010, 100 less so	Greenish-black	H 5.5-6 G 3.48 F 3	Insol in acids. Pleoc, X green, Z brown. FeO 7.9, MnO 3.0, Fe ₂ O ₃ 13.6%.
1.730 ^	1.731	1.735	1.740	.009	CARBOIRITE $\text{FeAl}_2\text{GeO}_5(\text{OH})_2$	60-75° $r > v$	$Z : c = 7^\circ$	TCL	001 perf	Pale to bright green	---	Poly tw. Pleoc, X blue-green, Y light blue, Z cols to yellowish.
1.719 ^	1.727	1.736	1.751	.024	FASSAITE (Pyroxene grp) $\text{Ca}(\text{Mg}, \text{Fe}, \text{Al})(\text{Si}, \text{Al})_2\text{O}_6$	49° (76+10°) $r > v$	$Y = \frac{b}{c}$ $Z : c = \frac{b}{c} = 48^\circ$	MCL pris	110 good at 87°	Green	H 6 G 3.37	Insol in acids. FeO 1.2, Fe ₂ O ₃ 6.2, Al ₂ O ₃ 12.9, TiO ₂ 3.0%.
1.728 ^ 1.748	1.733	1.736	1.744	.011	RHODONITE $(\text{Mn}, \text{Fe}, \text{Mg}, \text{Ca})\text{SiO}_3$	68°	---	TCL	100, 001 perf 010 less so	Brownish-pink	H 4 G 3.65 F 3	Insol in acids. MnO 49.2, FeO 0.6, MgO 0.8, CaO 2.6, Fe ₂ O ₃ 0.3%.
	1.726	1.737	1.766	.040	INNELITE $\text{Na}_2(\text{Ba}, \text{K})_4(\text{Ca}, \text{Mg}, \text{Fe})\text{Ti}_3\text{Si}_4\text{O}_{18}(\text{OH}, \text{F})_{1.5}(\text{SO}_4)$	82° (64+6°) $r > v$ str	$Z \sim a$, 001 :X = 77° :Y = 12° :Z = 88°	TCL plates	010, 110, 101 perf 001	Yellow-brown	H 5 G 3.96	Nearly insol in acids. Poly tw. Anom blue interf colors. Pleoc, X and Y light yellow, Z pale brownish-yellow.

	1.735	<u>1.737</u>	1.80	.065	BARIO-ORTHOJOAQUINITE $\text{Ba}_2(\text{Ba}, \text{Sr})_2\text{Fe}^{+2}_2\text{Ti}_2\text{Si}_8\text{O}_{24}\text{O}_2\cdot\text{H}_2\text{O}$	10-15°	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	ORTH	---	Yellow-brown	G 3.96	---
	1.732	<u>1.737</u>	1.744	.012	CHAMBERSITE $\text{Mn}_3\text{B}_7\text{O}_{13}\text{Cl}$	83°	---	ORTH ps cub	Conch to uneven	Cols to deep purple	H 7 G 3.49 F 3-4	---
	1.588	<u>1.739</u>	1.898	.310	URICITE $\text{C}_5\text{H}_4\text{N}_4\text{O}_3$	84° (96±1°)	---	MCL	---	White	G 1.84 F 1	= Uric acid (2-,6-,8-trihydroxypurine). Insol in H_2O , diss by alkalies.
	1.726	<u>1.739</u>	1.788	.062	ANTLERITE $\text{Cu}_3(\text{SO}_4)(\text{OH})_4$	55° $r < v$ very str	$X = \frac{b}{c}$ $Y = \frac{c}{c}$ el pos	ORTH tab 010	010 perf 100 poor	Emerald-to light-green	H 3.5 G 3.88 F 2.5	Diss by HCl. Pleoc, X yellow-green, Y blue-green, Z green.
	1.730	(1.739)	1.748	.018	SUZUKIITE $\text{Ba}_2\text{V}^{+4}_2\text{O}_2\text{Si}_4\text{O}_{12}$	~ 90° $r < v$ very str	$X = \frac{a}{b}$ $Y = \frac{b}{b}$	ORTH	010 perf 001, 100 dist	Bright green	H 4-4.5 G 4.03	Pleoc, X pale green, Y yellow-green, Z bluish-green.
1.741	1.733	<u>1.740</u>	1.769	.036	LAMPROPHYLLITE $(\text{Sr}, \text{Ba})_2\text{Na}_2\text{Ti}_3(\text{SiO}_4)_4(\text{OH}, \text{F})_2$	32° (58+8°) $r > v$ str	$X = \frac{b}{c}$ $Z:c = 0-8^\circ$	MCL	100 perf 011 good 010 dist	Yellow-brown	H 4-5 G 3.50	Poly tw 100. Pleoc wk, X and Y light yellow, Z brownish-yellow, Y brown-yellow, Z brown.
1.753	1.739	<u>1.740</u>	1.760	.021	ARDENNITE $\text{Mn}_4(\text{Al}, \text{Mg})_6[(\text{V}, \text{As})\text{O}_4]\text{Si}_5\text{O}_{18}(\text{OH})_6$	Small $r > v$ or $r < v$ str	$Z = \frac{a}{b}$ X or $Y = \frac{b}{c}$	ORTH pris <u>c</u>	010 perf 110 dist	Yellow to brown	H 6.5 G 3.65 F 2.5	Nearly insol in acids. Pleoc wk to str, X brownish yellow, Y golden yellow, Z pale yellow.
	1.740	<u>1.740</u>	1.760	.020	ANDREMEYERITE $\text{BaFe}^{+2}_2\text{Si}_2\text{O}_7$	0-80° disp extr	$Z = \frac{b}{c}$	MCL pris	100, 010 perf	Pale emerald-green	H 3-3.5 G 2.41	Tw on (100). 2V 0° at 490 nm, to 40° at 540, to 0° at 585, to 80° at 640 nm. X:c 2° at 670 nm to 61° at 470 nm.
1.746	1.736	<u>1.740</u>	1.745	.009	STAUROLITE $(\text{Fe}, \text{Mg}, \text{Zn})_2\text{Al}_9(\text{Si}, \text{Al})_4(\text{Si}, \text{Al})_4\text{O}_{22}(\text{OH})_2$	83° $r > v$ wk	$Z = \frac{c}{b}$ $X \sim \frac{b}{c}$ el pos	MCL ps orth	010 dist	Yellow to reddish brown	H 7 G 3.75 infus	Insol in acids. Cruciform tw. Pleoc wk, X and Y light yellow, Z orange. FeO 12.7, Fe ₂ O ₃ 2.2%.
1.740 1.754	1.735	<u>1.741</u>	1.767	.032	BARYTOLAMPROPHYLLITE $(\text{Na}, \text{K})_2(\text{Ba}, \text{Ca}, \text{Sr})_2(\text{Ti}, \text{Fe})_3(\text{SiO}_4)_4(0, \text{OH})_2$	55° $r > v$ str	$X = \frac{b}{c}$ $Z:c = -3^\circ$	MCL	100 perf 011 good	Dark brown	H 4-5 G 3.50	Poly tw 100. Pleoc str, X light yellow, Z brown, abs Z > Y > X.
1.730 1.751	1.735	<u>1.741</u>	1.761	.026	FERROAUGITE (Pyroxene grp) $\text{Ca}(\text{Fe}, \text{Mg})(\text{Si}, \text{Al})\text{Si}_2\text{O}_6$	51°	$Y = \frac{b}{c}$ $Z:c = 40^\circ$	MCL	110 good at 87°	Brown	H 6 G 3.42	Insol in acids. FeO 24.0, Fe ₂ O ₃ 1.2, Al ₂ O ₃ 4.1, MnO 0.4%.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.703 ^	1.738	<u>1.741</u>	1.743	.005	SERENDIBITE $\text{Ca}_2(\text{Mg,Al})_6(\text{Si,Al,B})_6\text{O}_{20}$	~ 90° r < v str	Ext:tw lam = 40°	TCL plates	---	Sky-blue to indigo-blue	H 6.5 G 3.52 infus	Insol in acids. Poly tw. Pleoc, X pale yellow-green, Y nearly cols, Z deep blue.
1.810 v	1.722	<u>1.742</u>	1.763	.041	ADAMITE $\text{Zn}_2(\text{AsO}_4)(\text{OH})$	88° r < v str	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	ORTH	101 good 010 poor	Yellow, green, rose	H 3.5 G 4.44	Diss by acids. U opt neg. May contain Cu or Co.
1.732 ^ 1.750	1.738	<u>1.742</u>	1.754	.016	PYROXMANGITE $(\text{Mn,Fe,Mg,Ca})\text{SiO}_3$	39° (60+16°)	Z':c = 45° on T00	TCL	---	Pink to brown	H 6 G 3.76	Insol in acids. MnO 29.3, FeO 19.1, MgO 2.0, CaO 2.9%. (Series with Pyroxferroite.)
	1.715	<u>1.743</u>	1.783	.068	JOHILLERITE $\text{Na}(\text{Mg,Zn})\text{Cu}(\text{AsO}_4)_3$	80°	Z = $\frac{b}{c}$ X:c = 16°	MCL	010 perf 100, 001 good	Violet	H 3 G 4.15	Pleoc, X violet red, Y bluish violet, Z greenish-blue.
1.624 ^ 1.796	1.741	<u>1.744</u>	1.768	.027	SCORODITE (Variscite grp) $\text{FeAsO}_4 \cdot 2\text{H}_2\text{O}$	40° r > v str	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	ORTH pyram	201 imperf	Green to brown	H 3.5-4 G 3.14 F easy	Diss by acids. Fe_2O_3 25.8, Al_2O_3 4.7%.
1.708 ^ 1.768	1.738	<u>1.744</u>	1.753	.015	ALLANITE (Epidote grp) $(\text{Ca,Ce,Y})_2(\text{Al,Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	70-80° r > v	Y = $\frac{b}{c}$ X:c = 5-40°	MCL	001 imperf	Brownish-black	H 5-6 G 3.69 F 3	Slowly attacked by HCl, gel. U pleoc in browns and reds. G and n's decrease with alteration. Fe_2O_3 + FeO = 9.2%.
	1.698	<u>1.745</u>	1.793	.095	LIBETHENITE $\text{Cu}_2(\text{PO}_4)(\text{OH})$	90° r > v str	X = $\frac{b}{c}$ Y = $\frac{c}{c}$	ORTH	100, 010 poor	Olive- to blackish-green	H 4 G 3.97 F 2	Diss by acids. Pleoc in yellow and green.
1.75 _	---	<u>1.74</u>	---	---	TINTICITE $\text{Fe}_6(\text{PO}_4)_4(\text{OH})_6 \cdot 7\text{H}_2\text{O}$	---	---	ORTH(?) earthy masses	---	Creamy white	H 2.5 G 2.8	---
1.722 ^	1.740	<u>1.746</u>	1.765	.025	ASTROPHYLLITE $(\text{K,Na})_3(\text{Fe,Mn})_7\text{Ti}_2\text{Si}_8\text{O}_{24}(\text{O,OH})_7$	80° (59+10°) r > v str	X ~ $\frac{b}{c}$ Z:c = 3-19° el clv neg	TCL	001 perf	Brown	H 3 G 3.24 F 3	Dec by acids. Pleoc, yellow to orange, abs X > Y > Z. FeO 20.7, MnO 1.4, Fe_2O_3 11.2%.

1.740 ↓ 1.751 neg 1.753	1.739	<u>1.746</u>	1.753	.014	STAUROLITE, cobaltoan (Fe,Mg,Co) ₂ Al ₉ (Si,Al) ₄ O ₂₂ (OH) ₂	~ 90° r > v wk	Z = $\frac{c}{b}$ X ~ $\frac{b}{a}$ el pos	MCL ps orth	010 dist	Blue	H 7 G 3.76 infus	Insol in acids. Pleoc, X blue, Y violet-blue, Z violet. CoO 6.8- 8.5%.
	1.737	<u>1.747</u>	1.768	.031	CREASEYITE Pb ₂ Cu ₂ Fe ₂ Si ₅ O ₁₇ ·6H ₂ O	69° r < v wk	X = $\frac{a}{b}$ Z = $\frac{c}{b}$	ORTH fib	---	Green	H 2.5 G 4.1 fus	Dec by hot HNO ₃ . Pleoc, X and Z yellow-green, Y clear green, abs Z = X > Y.
	1.746	<u>1.748</u>	1.768	.022	PROSPERITE CaZn ₂ (AsO ₄) ₂ ·H ₂ O	34° r > v str	Y = $\frac{b}{c}$ Z:c = 27°	MCL prism	---	Cols, white	H 4.5 G 4.31	---
1.736 ^	1.739	<u>1.748</u>	1.760	.021	RHODONITE, ferroan (Mn,Fe,Mg,Ca)SiO ₃	75°	---	TCL	100, 001 perf 010 less so	Brownish- pink	H 4 G 3.76 F 3	Insol in acids. MnO 29.3, FeO 19.1, MgO 2.0, CaO 2.4%.
	1.746	<u>1.748</u>	1.756	.010	CHRYSOBERYL BeAl ₂ O ₄	70° r > v	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH tab 100 ps hex tw	110 dist 010 imperf	Green, red, yellow	H 8.5 G 3.75 infus	Insol in acids. Tw pl 031. Pleoc, X red, Y orange to yellow, Z emerald- green.
1.726 ^	1.748	<u>1.749</u>	1.753	.005	WOLFEITE (Triplodite ser) (Fe,Mn) ₂ (PO ₄)(OH)	Med r > v extr	X = $\frac{b}{a}$ disp dist	MCL pris	010 good 120 fair	Reddish- brown	H 4.5-5 G 3.8 fus	Diss by acids.
1.763 v	1.715	<u>1.75</u>	1.80	.085	ERYTHROSIDERITE K ₂ FeCl ₅ ·H ₂ O	62° r > v str	X = $\frac{a}{b}$ Z = $\frac{b}{a}$	ORTH tab 100	210, 011 perf	Red	G 2.37	Sol in H ₂ O. Yellow in thin section.
1.772 v	1.732	<u>1.750</u>	1.778	.046	PIEMONTE (Epidote grp) Ca ₂ (Al,Mn,Fe) ₃ (SiO ₄) ₃ (OH)	Large	Y = $\frac{b}{c}$ X:c = -5° el clv pos	MCL pris	001 perf	Reddish- brown	H 6 G 3.45	Insol in acids. Pleoc str, X buff, Y deep lavender, Z pink. Mn ₂ O ₃ 11.8, Fe ₂ O ₃ 3.8%.
1.742 ↓ 1.758	1.748	<u>1.750</u>	1.764	.016	PYROXFERROITE (Fe,Mn,Ca)SiO ₃	30°	---	TCL	---	Brown, yellow	H 6 G 3.80	Insol in acids. FeO 28.3, MnO 20.6, CaO 1.9%. (Series with Pyroxmangite.)
1.740 neg 1.741 ^	1.741	<u>1.751</u>	1.774	.033	ACMITE (Pyroxene grp) Na(Fe,Zn,Mn)Si ₂ O ₆	Med (68+7°)	Y = $\frac{b}{c}$ Z:c = 60°	MCL	110 good at 87°	Reddish- brown	H 6 G 3.59 F 3	Insol in acids. Fe ₂ O ₃ 12.9, FeO 1.4, ZnO 8.8, MnO 7.2, Na ₂ O 5.0%.
	1.745	<u>1.751</u>	1.760	.015	TIRAGALLOITE Mn ₄ As ⁵⁺ Si ₃ O ₁₂ (OH)	38-46° (79+16°)	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ Z ~ $\frac{c}{b}$ el pos	MCL	100 good	Orange	G 3.84	Insol in HCl.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
\vee 1.864	1.750	<u>1.751</u>	1.761	.011	SYNADELPHITE (Mn,Mg,Ca,Pb) ₉ (AsO ₃) (AsO ₄) ₂ (OH) ₉ ·2H ₂ O	37° r > v	---	TCL ps orth	010 imperf	Cols	H 4.5 G 3.64 F 2	Diss by acids. Pleoc wk, X and Y cols, Z light brown.
	1.740	<u>1.752</u>	(1.80)	(.06)	Unnamed ferric phosphate Fe ₃ (PO ₄) ₂ (OH) ₃ ·3-6H ₂ O	Med r < v str	---	---	---	Brown	G 2.55	Pleoc, X pale brown, Z reddish brown.
1.732 ^	1.745	<u>1.753</u>	1.771	.026	HEDENBERGITE (Diopside ser, Pyroxene grp) CaFeSi ₂ O ₆	52° (68+9°) r > v wk	Y = $\frac{b}{c}$ Z:c = 41°	MCL pris c	110 good at 87°	Greenish- black	H 6 G 3.65 fus	Insol in acids. FeO 29.7, MnO 0.9, Fe ₂ O ₃ 1.4%.
	1.745	<u>1.753</u>	1.778	.033	O'DANIELITE Na(Zn,Mg) ₃ (AsO ₄) ₃	60°	Z = $\frac{b}{c}$ Y:c = 18°	MCL	010, 100 perf, 001	Pale violet	H 3 G (4.49)	---
	1.750	<u>1.753</u>	1.780	.030	NAGASHIMALITE Ba ₄ (V ³⁺ ,Ti) ₄ B ₂ Si ₈ O ₂₇ Cl (O,OH) ₂	30° r > v str	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	ORTH	---	Greenish- black	G 4.08	Pleoc, X greenish- yellow, Y green, Z greenish-brown, abs Z > Y > X.
1.740 ^	1.753	<u>1.753</u>	1.770	.017	ARDENNITE Mn ₄ Al ₆ [(V,As)O ₄]Si ₅ O ₁₈ (OH) ₆	Small r > v or r < v str	Z = $\frac{a}{b}$ X or Y = $\frac{b}{c}$	ORTH pris c	010 perf 110 dist	Yellow to brown	H 6.5 G 3.65 F 2.5	Nearly insol in acids. Pleoc in yellow, wk to str.
1.746 ^ 1.751 neg ^	1.747	<u>1.753</u>	1.761	.014	STAUROLITE (Fe,Mg,Zn) ₂ Al ₉ (Si,Al) ₄ O ₂₂ (OH) ₂	80° r > v wk	Z = $\frac{c}{b}$ X ~ $\frac{b}{c}$	MCL ps orth	010 dist	Reddish- brown	H 7 G 3.77 infus	Insol in acids. Cruciform tw. Pleoc wk, X and Y light yellow, Z orange- yellow.
1.741 ^ 1.760	1.743	<u>1.754</u>	1.777	.034	BARYTOLAMPROPHYLLITE (Na,K) ₂ (Ba,Ca,Sr) ₂ (Ti,Fe) ₃ (SiO ₄) ₄ (O,OH) ₂	30° (70+7°) r > v str	X = $\frac{b}{c}$ Z:c = -7°	MCL	100 perf 011 good 010 dist	Dark brown	H 4-5 G 3.50	Poly tw 100. Pleoc, X light yellow, Y brown-yellow, Z brown.
\vee 1.775	1.725	<u>1.755</u>	1.815	.090	OLMSTEADITE KFe ²⁺ ₂ (Nb,Ta)(PO ₄) ₂ O ₂ · 2H ₂ O	~ 60° (73+3°)	X = $\frac{c}{a}$ Y = $\frac{a}{a}$	ORTH	100, 001 good	Deep brown to black	H 4 G 3.31	Dec by warm HCl. Pleoc, X blue-green, Y yellow, Z brown, abs X > Z > Y.
\vee 1.774	1.752	<u>1.756</u>	1.779	.027	AUSTINITE CaZn(AsO ₄)(OH)	41°	X = $\frac{a}{c}$ Y = $\frac{c}{c}$	ORTH	011 good	Cols	H 4-4.5 G 4.13	Diss by acids.

	1.730	<u>1.758</u>	1.838	.108	AZURITE $\text{Cu}_3(\text{CO}_3)_2(\text{OH})_2$	67° (64+2°) $r > v$ str	$X = b$ $Z:c = -13^\circ$ disp dist	MCL	011 perf inter- rupted, 100 fair	Azure- blue	H 3.5-4 G 3.77 F 3	Diss by acids with eff. Pleoc in blue, $Z > Y > X$.
(1.734) ^	1.725	<u>1.758</u>	1.830	.105	SEIDOZERITE $(\text{Na,Ca})_2(\text{Zr,Ti,Mn})_2\text{Si}_2\text{O}_7(\text{O,F})_2$	68° (71+2°) $r > v$ str	$X = b$ $Y:c = 13^\circ$	MCL	001 perf	Brownish- red to yellow	H 4-5 G 3.47 F easy	Diff diss by HCl. Pleoc, X dark red, Y red, Z yellow.
1.750 ^	1.756	<u>1.758</u>	1.768	.012	PYROXFERROITE $(\text{Fe,Ca,Mn})\text{SiO}_3$	40°	---	TCL	---	Yellow, brown	H 6 G 3.80	Insol in acids.
1.754 ^	1.749	<u>1.760</u>	1.779	.030	LAMPROPHYLLITE $(\text{Sr,Ba})_2\text{Na}_2\text{Ti}_3(\text{SiO}_4)_4(\text{O,F})_2$	32° (75+8°) $r > v$ str	$X = b$ $Z:c = 1-8^\circ$	MCL	100 perf 011 good	Yellow- brown	H 4-5 G 3.5	Poly tw 100. Pleoc wk, X and Y light yellow, Z brownish- yellow.
1.75 ^	1.733	<u>1.763</u>	1.807	.016	KREMERSITE (Erythrosiderite ser) $(\text{NH}_4,\text{K})_2\text{FeCl}_5 \cdot \text{H}_2\text{O}$	70°	$X = a$ $Z = b$	ORTH ps cub	---	Ruby red	G 2.18	Sol in H_2O . Data calc from end- members.
1.762 neg ^ 1.772	1.755	<u>1.763</u>	1.773	.018	EULITE (Orthopyroxene ser, Pyroxene grp) $(\text{Fe,Mg,Mn})_2\text{Si}_2\text{O}_6$	84° $r > v$	$X = a$ $Z = c$	ORTH pris c	210 good at 87°	Dark green	H 5.5 G 3.88 F 4	Insol in acids. FeO 41.7, MnO 5.0, Fe_2O_3 0.3% (En_{12}).
	1.765	<u>1.765</u>	1.800	.035	HENRITERMIERITE (Garnet grp) $\text{Ca}_3(\text{Mn,Al})_2(\text{SiO}_4)_2(\text{OH})_4$	Small	---	TET	Conch	Clove- to apricot- brown	G 3.34 fus	Dec by warm HCl. Pleoc wk, O pale yellow, E lemon- yellow.
	1.747	<u>1.765</u>	1.78	.033	JOESMITHITE $(\text{Ca,Pb})_3(\text{Mg,Fe}^{+2},\text{Fe}^{+3})_5\text{Si}_6\text{Be}_2\text{O}_{22}(\text{OH})_2$	60-70°	---	MCL	110 perf	Lustrous black	H 5.5 G 3.83	Insol in cold acids. Pleoc, X and Z olive, Y olive-brown, abs $Y > X, Z$.
	1.750	<u>1.766</u>	1.85	.10	KOMAROVITE $(\text{H,Ca})_2\text{Nb}_2\text{Si}_2\text{O}_{10}(\text{OH,F})_2 \cdot \text{H}_2\text{O}$	48°	$X = a$ $Z = b$	ORTH	001 fair	Pale rose	H 1.5-2	---
	1.765	---	1.775	.010	NATRODUFRENITE $\text{Na}(\text{Fe}^{+3},\text{Fe}^{+2})(\text{Fe}^{+3},\text{Al})_5(\text{PO}_4)_4(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	---	el pos	MCL spheru- litic	---	Bronze- green	G 3.20	Pleoc, X yellow, Z green.
	1.763	---	1.769	.006	ROWLANDITE $\text{Y}_3(\text{SiO}_4)_2(\text{OH,F}) (?)$	---	---	TCL(?)	Conch	Green, brown	H 6-6.5 G 4.85	Gel with acids. Com- monly metamict.
(1.731) ^ 1.802	1.760	(<u>1.766</u>)	1.775	.015	ALLUAUDITE $(\text{Na,Ca})\text{Fe}^{+2}(\text{Mn,Fe}^{+2},\text{Fe}^{+3},\text{Mg})_2(\text{PO}_4)_3$	Large $r > v$ mod	$Z = b$ $Y:c = 18^\circ$	MCL u mass	T01 perf 2 others	Greenish- black	H 5-5.5 G 3.5 F 3	Diss by acids.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
(1.717) ^	1.753	<u>1.767</u>	1.822	.069	JOAQUINITE $\text{Ba}_2\text{NaCe}_2\text{Fe}(\text{Ti}, \text{Nb})_2\text{Si}_8\text{O}_{26}(\text{OH}, \text{F}) \cdot \text{H}_2\text{O}$	30-55° $r < v$	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	MCL ps orth	---	Brown to black	H 5 G 3.98 F 2.5	Insol in acids. Pleoc wk, X and Y cols, Z light yellow.
	1.736	<u>1.767</u>	1.796	.060	ARTHURITE $\text{CuFe}_2(\text{AsO}_4, \text{PO}_4, \text{SO}_4)_2(\text{O}, \text{OH})_2 \cdot 4\text{H}_2\text{O}$	~ 90°	$Y = \frac{b}{c}$ $Z:c = 10^\circ$	MCL	---	Apple-green	G 3.2	Pleoc, X cols to pale green, Y grass green, Z olive-green, abs $Z > Y > X$.
1.744 ^	1.763	<u>1.768</u>	1.788	.025	ALLANITE (Epidote gr) $(\text{Ca}, \text{Ce}, \text{Y})_2(\text{Al}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	57° $r > v$	$Y = \frac{b}{c}$ $X:c = 5-40^\circ$	MCL	001 imperf	Dark brown	H 5.5 G 3.8 fus	Slowly attacked by HCl, gel. U pleoc in browns and reds. FeO 10.4, Fe ₂ O ₃ 3.8%.
	1.710	<u>1.770</u>	1.840	.130	ROSSITE $\text{CaV}_2\text{O}_6 \cdot 4\text{H}_2\text{O}$	Large disp str	$Z \sim \frac{c}{b}$ $Y:b = 45^\circ$	TCL pris tab	010 good	Yellow	H 2-3 G 2.45 F easy	Slowly sol in H ₂ O. Tw on 100.
	1.769	<u>1.770</u>	1.785	.016	HOLDENITE $(\text{Mn}, \text{Mg})_6\text{Zn}_3(\text{AsO}_4)_2\text{SiO}_4(\text{OH})_8$	30° $r > v$ perc	$X = \frac{c}{b}$ $Y = \frac{b}{b}$	ORTH tab	010 poor subconch	Pink to red	H 4 G 4.11	Diss by acids.
1.750 ^ 1.788	1.754	<u>1.772</u>	1.795	.041	PIEMONTE (Epidote grp) $\text{Ca}_2(\text{Mn}, \text{Al}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	85° $r < v$	$Y = \frac{b}{c}$ $X:c = -6^\circ$ el clv pos	MCL pris	001 perf	Reddish-brown	H 6 G 3.47	Insol in acids. Pleoc str, X yellow, Y pink, Z crimson. Mn ₂ O ₃ 22.0, Fe ₂ O ₃ 6.4%.
1.763 ^	1.768	<u>1.772</u>	1.790	.022	ORTHO FERROSILITE (Orthopyroxene ser, Pyroxene grp) $(\text{Fe}, \text{Mg})_2\text{Si}_2\text{O}_6$	~ 55° $r < v$	$X = \frac{b}{c}$ $Z = \frac{c}{c}$	ORTH	210 good at 87°	Dark green	H 5.5 F 4	Insol in acids. Extrapolated value for En ₀ .
1.712 ^	(1.761)	<u>1.772</u>	1.781	.020	MANGANHUMITE (Humite grp) $\text{Mn}_7(\text{SiO}_4)_3(\text{OH})_2$	84°	$X = \frac{a}{b}$ $Z = \frac{b}{b}$	ORTH	010	Orange-brown	H 4 G (4.05)	Gel with acids, MnO ~ 69, FeO ~ 1.0, MgO ~ 0.8%.
	1.752	<u>1.773</u>	1.796	.044	LUETHEITE $\text{Cu}_2\text{Al}_2(\text{AsO}_4)_2(\text{OH})_4 \cdot \text{H}_2\text{O}$	(88+5°) $r < v$ mod	$X = \frac{b}{c}$ $Z:c = 10^\circ$	MCL tab	100 fair to good	Blue to greenish-blue	H 3 G 4.28 F easy	Diss by hot HCl. Pleoc wk in pale blue, abs $Z = Y > X$.
1.813 v	1.772	(<u>1.773</u>)	1.792	.020	WARWICKITE $(\text{Mg}, \text{Ti}, \text{Fe}, \text{Al})_2(\text{BO}_3)_0$	Small	$X = \frac{c}{b}$ $Y = \frac{b}{b}$ el clv neg	ORTH pris	100 perf	Dark brown to black	H 3.5-4 G 3.35 infus	Dec by H ₂ SO ₄ . Pleoc X brown, Y and Z light brown.

	1.770	<u>1.774</u>	1.83	.06	TARAMELLITE $\text{Ba}_4(\text{Fe,Ti,Mg,V})_4\text{B}_2\text{Si}_6\text{O}_{29}\text{Cl}$	40° $r > v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el pos	ORTH pris \underline{c}	100 perf	Brownish-red	H 5.5 G 3.92 F easy	Insol in acids.
(1.722) ◇ 1.812	1.765	<u>1.774</u>	1.787	.022	GADOLINITE $(\text{Y,Ca})_2\text{FeBe}_2\text{Si}_2\text{O}_{10}$	Med (80+6°) $r < \underline{v}$	$X = \frac{b}{c}$ $Z:\underline{c} = 4-13^\circ$	MCL	Conch	Greenish-to brownish-black	H 6.5 G 4.1-4.5 infus	Diff diss by acids, gel.
1.756 ^ 1.795	1.770	<u>1.774</u>	1.783	.013	AUSTINITE $\text{Ca}(\text{Zn,Cu})(\text{AsO}_4)(\text{OH})$	Med $r > v$ wk	$X = \frac{a}{c}$ $Y = \frac{c}{b}$	ORTH	011 good	Cols to green	H 4-4.5 G 4.1	Diss by acids.
	1.765	<u>1.775</u>	1.800	.035	OGDENSBURGITE $\text{Ca}_4\text{Fe}^{+3}_6(\text{AsO}_4)_5(\text{OH})_{11} \cdot 5\text{H}_2\text{O}$	25° (65+7°)	---	Thin plates	One perf	Dark reddish-brown	H ~ 2 G 2.92	Pleoc moderate, $X < Y = Z$.
1.755 ^	1.765	<u>1.775</u>	1.835	.070	OLMSTEADITE $\text{KFe}^{+2}_2(\text{Nb,Ta})(\text{PO}_4)_2\text{O}_2 \cdot 2\text{H}_2\text{O}$	~ 60° (46+5°)	$X = \frac{c}{a}$ $Y = \frac{a}{b}$	ORTH	100, 001 good	Deep brown to black	H 4 G 3.36	Dec by warm HCl. Pleoc, X dark blue, Y light brown, Z dark brown, abs $X > Z > Y$.
	1.763	<u>1.777</u>	1.785	.022	YOSHIMURAIT $(\text{Ba,Sr})_2\text{TiMn}_2\text{Si}_2\text{O}_8(\text{PO}_4,\text{SO}_4)(\text{OH,Cl})$	80-90° (106+11°) $r > \underline{v}$	---	TCL	010 perf 101, 101 dist	Brown	H 4.5 G 4.13	Pleoc, X bright yellow, Y orange-brown, Z brown, abs $Z > Y > X$. Poly tw on 010.
(1.730) ◇ 1.789	1.777	<u>1.778</u>	1.823	.046	MONAZITE $(\text{Ce,La,Nd,Th})\text{PO}_4$	14.5°	$X = \frac{b}{c}$ $Z:\underline{c} = 2^\circ$	MCL	100 dist	Green	H 5-5.5 G 5.3 infus	Diff sol in HCl. Pleoc wk. UO_2 1.5, ThO_2 11.3%.
	1.76	<u>1.78</u>	>>1.85	>>.07	KARPATITE $\text{C}_{24}\text{H}_{12}$	Small $r < v$ extr	$X = \frac{b}{c}$ $Z:\underline{c} = 21^\circ$	MCL	001, 100, 201 perf	Pale	H < 1 G 1.35 F 1	A hydrocarbon, Coronene. Sol in org solvents.
	1.774	<u>1.780</u>	1.813	.039	BERAUNITE $\text{Fe}^{+2}\text{Fe}^{+3}_5(\text{PO}_4)_4(\text{OH})_5 \cdot 4\text{H}_2\text{O}$	Med large (47+8°) $r > v$ str	$Z = \frac{b}{c}$ $X:\underline{c} = 6^\circ$	MCL tab	100 good	Reddish-brown	H 2 G 2.9-3.1 F 3	Diss by acids. Tw. Pleoc str, X and Y pale yellow to flesh, Z carnelian red.
	1.779	<u>1.780</u>	1.816	.037	CHERALITE $(\text{Ce,La,Th,Ca})(\text{PO}_4,\text{SiO}_4)$	18°	$X = \frac{b}{c}$ $Z:\underline{c} = 7^\circ$	MCL	010 dist	Dark to pale green	H 5 G 5.28	Pleoc wk, X and Y green, Z yellow-green. Compare Monazite.
	1.77	<u>1.78</u>	1.80	.03	VANDENBRANDEITE $\text{Cu}(\text{UO}_2)(\text{OH})_4$	Large	Opt ax ~ ⊥ (001)	TCL	110 perf 2 others	Dark green	H 4 G 5.0	Diss by hot HCl. Pleoc, yellow-green to blue-green.
	1.77	<u>1.78</u>	1.80	.03	MEDAITE $\text{Mn}_6\text{V}^{+5}\text{Si}_5\text{O}_{18}(\text{OH})$	---	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el pos	MCL	100 good	---	G 3.70	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.794 v	1.776	1.780	1.805	.029	CARYINITE (Na,Ca,Pb) ₃ (Mg,Mn,Fe ³⁺) ₄ (AsO ₄) ₄ (?)	41° r > v wk	X = $\frac{c}{a}$ Y = $\frac{a}{b}$	MCL u mass	110, 010	Nut brown	H 4 G 4.27 F 2.5	Diss by HNO ₃ . Orange to red in section, not pleoc.
	1.753	1.782	1.815	.062	SHATTUCKITE Cu ₅ (SiO ₃) ₄ (OH) ₂	~ 90°	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH	010, 100	Blue	H 3.5 G 3.9-4.1	Pleoc, pale blue to deep blue.
	1.774	1.782	1.798	.028	RETZIAN-(Nd) Mn ₂ (Nd,Ce,La)(AsO ₄) ₄ (OH) ₄	69°	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH	---	Brown	---	Pleoc, X yellow, Y reddish-brown, Z brown.
1.772 ^	1.756	1.788	1.829	.073	PIEMONTE (Epidote grp) Ca ₂ (Mn,Al,Fe) ₃ (SiO ₄) ₃ (OH)	86° r < v	Y = $\frac{b}{a}$ X:c = -9° el clv pos	MCL pris	001 perf	Reddish-brown	H 6 G 3.47	Insol in acids. Pleoc str, X yellow, Y violet, Z red. Mn ₂ O ₃ 22.0, Al ₂ O ₃ 12.4, Fe ₂ O ₃ 6.4%.
	1.777	1.788	1.800	.023	RETZIAN Mn ₂ Ce(AsO ₄) ₄ (OH) ₄	Large r < v wk	Y = $\frac{b}{a}$ X = $\frac{c}{a}$	ORTH	---	Chocolate to chestnut-brown	H 4 G 4.15 F diff	Diss by acids. Pleoc, X cols, Y pale yellow-brown, Z reddish-brown.
1.778 ^ 1.801	1.787	1.789	1.839	.052	MONAZITE (Ce,La,Nd)PO ₄	10-16° r < v wk	X = $\frac{b}{a}$ Z:c = -4°	MCL	110 dist 010 diff	Brown to red	H 5-5.5 G 5.26 infus	Diff sol in HCl. Tw pl 100. Pleoc wk, X light yellow, Y dark yellow, Z greenish-yellow.
1.733 ^ 1.827	1.78	1.79	2.04	.26	FERRIMOLYBDITE Fe ₂ (MoO ₄) ₃ ·7-10H ₂ O	Small r < v dist	Y = $\frac{a}{b}$ Z = $\frac{c}{b}$	ORTH fib c	001 dist	Sulfur-yellow	H 1-2 G 2.99 F easy	Diss by HCl, dec by NH ₄ OH. Pleoc, X and Y pale yellow, Z yellow-green.
	1.75	1.79	1.85	.10	Hydrous U sulfate (?)	Med r < v str	Y = $\frac{b}{a}$ Z = $\frac{c}{b}$ el pos	ORTH 100 laths el c	---	Lemon-yellow	Soft	Diss by acids. Comp unknown.
	1.792	1.794	1.821	.029	JIMBOITE Mn ₃ B ₂ O ₆	35° r > v	Y = $\frac{b}{a}$ X = $\frac{a}{b}$ el pos	ORTH	110 perf parting 101	Light purplish-brown	H 5.5 G 3.98	Diss by acids. Tw pl 101.
1.780 ^	1.794	1.794	1.803	.009	CARYINITE (Na,Ca,Pb) ₃ (Mg,Mn,Fe ³⁺) ₄ (AsO ₄) ₄ (?)	Small r > v wk	X = $\frac{c}{a}$ Y = $\frac{a}{b}$	MCL u mass	110, 010	Brownish-red	H 3-4 G 4.3 F 2.5	Diss by HNO ₃ . Orange to red in section, not pleoc.

1.774 ^	1.780	<u>1.795</u>	1.815	.035	CONICALCITE $\text{CaCu}(\text{AsO}_4)(\text{OH})$	$\sim 90^\circ$ $r < v \text{ mod}$	---	ORTH equant	011 uneven	Grass- green	H 3 G 4.1 fus	Diss by acids. In thin-section yellow- green, faintly pleoc.
1.744 ^	1.785	<u>1.796</u>	1.814	.029	SCORODITE $\text{FeAsO}_4 \cdot 2\text{H}_2\text{O}$	75° $r > v$	$X = \frac{a}{c}$ $Y = \frac{c}{a}$	ORTH pyram	201 imperf	Green to brown	H 3.5 G 3.29 F 2-2.5	Diss by acids. Pleoc wk, X and Y blue, Z red. Al_2O_3 1.8%.
	1.79	(<u>1.79</u>)	1.82	.03	KRYZHANOVSKITE $\text{MnFe}^{+3}_2(\text{PO}_4)_2(\text{OH})_2 \cdot \text{H}_2\text{O}$	$40-45^\circ$ $r < v \text{ str}$	---	ORTH pris	001 perf	Brown to greenish- brown	H 3.5-4 G 3.31 F 2.5	Diss by acids. Pleoc str, X wine-yellow, Y orange-brown, Z reddish-brown.
	1.743	<u>1.80</u>	1.88	.137	TUNDRITE $\text{Na}_3(\text{Ce}, \text{La})_4(\text{Ti}, \text{Nb})_2\text{Si}_2\text{O}_8(\text{CO}_3)_3\text{O}_4(\text{OH}) \cdot 2\text{H}_2\text{O}$	Large	$Z:c = 14^\circ$	TCL	010 perf	Brownish- to greenish- yellow	H 3 G 3.7-4.0	Pleoc wk, X pale yellow, Z greenish- yellow.
	1.72	<u>1.80</u>	1.98	.26	HELMUTWINKLERITE $\text{PbZn}_2(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	70° $r < v \text{ dist}$	$X:c = 50^\circ$ $Z:b = 30^\circ$	TCL	---	Cols to light blue	H 4.5 G 5.3	Diss by HCl.
	1.775	<u>1.800</u>	1.846	.071	BOTALLACKITE $\text{Cu}_2(\text{OH})_3\text{Cl}$	Med large $r > v \text{ str}$	$X \perp \text{clv}$	MCL	One direction	Bluish- green	G ~ 3.6 fus	Diss by acids. Slightly pleoc.
	1.792	<u>1.801</u>	1.852	.060	JINSHAJIANGITE $(\text{Na}, \text{K})_5(\text{Ba}, \text{Ca})_4$ $(\text{Fe}, \text{Mn})_{15}(\text{Ti}, \text{Fe}^{+3}, \text{Nb})_8$ $(\text{SiO}_4)_{15}(\text{F}, \text{OH})_{10}$	72° $(47+6^\circ)$ $r < v$	---	MCL tab	010, 100 perf	blackish- red to golden-red	G 3.61 (H 430 kg/sq mm)	Pleoc, X light golden-yellow, Y brownish-yellow, Z brownish-red.
	1.783	<u>1.801</u>	1.834	.051	FLINKITE $\text{Mn}^{+2}_2\text{Mn}^{+3}(\text{AsO}_4)(\text{OH})_4$	Large $(74+5^\circ)$ disp wk	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH tab 001	One clv	Greenish- brown	H 4-4.5 G 3.87 F easy	Diss by acids. Pleoc, X yellow- to brownish-green, Y yellow-green, Z orange-brown.
1.789 v (1.821)	1.800	<u>1.801</u>	1.849	.049	MONAZITE $(\text{Ce}, \text{La}, \text{Nd}, \text{Th})\text{PO}_4$	11° $r < v \text{ wk}$	$X = \frac{b}{c}$ $Z:c = -10^\circ$	MCL	100 dist 010 diff	Yellow, red, brown	H 5 G 5.27 infus	Diff sol in acids. Pleoc wk in yellow. ThO_2 11.2%. (Solid soln with Huttonite.)
1.766 ^	1.782	<u>1.802</u>	1.835	.053	ALLUAUDITE $(\text{Na}, \text{Ca})_4\text{Fe}^{+2}_4(\text{Mn}, \text{Fe}^{+2}, \text{Fe}^{+3}, \text{Mg})_8(\text{PO}_4)_{12}$	79° $r > v \text{ mod}$	$Z = \frac{b}{c}$ $Y:c = 18^\circ$	MCL u mass	T01 perf 2 others	Greenish- black	H 5-5.5 G 3.58 F 3	Diss by acids. Pleoc, X light brown, Z dark brown.
1.82 v	1.793	<u>1.804</u>	1.87	.08	AENIGMATITE $\text{Na}_2\text{Fe}_5\text{TiSi}_6\text{O}_{20}$	Small	$Z:c = 40^\circ$	TCL pris	010, 100	Black	H 5.5 G 3.8 F 3	Insol in acids. Tw 110. Pleoc str, X yellow-brown, Y dark brown, Z dark brown to black.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.832 ^v	1.805	<u>1.805</u>	1.915	.110	LUDWIGITE (Mg,Fe) ₂ Fe ⁺³ BO ₅	Small r < v extr	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	Uneven, parting 001	Greenish-black	H 5 G 3.8 F diff	Diss by acids. Pleoc, X and Y dark green, Z brown.
	~1.81	(~1.81?)	~1.83	~.02	WELSHITE Ca ₂ Mg ₄ Fe ⁺³ Sb ⁺⁵ Be ₂ Si ₄ O ₂₀	(~ 25°)	---	TCL pris	Conch	Reddish-black	H 6 G 3.77	Opt char uncertain. 2E ~ 45°.
1.742 [^]	1.772	<u>1.810</u>	1.863	.091	OLIVENITE Cu ₂ (AsO ₄)(OH)	~ 90° r < v str	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH acic	101 conch	Olive-green	H 3 G 4.46 F 3	Diss by acids. In thin-section pale green, not pleoc.
	1.795	<u>1.810</u>	1.835	.040	TAVORITE LiFe(PO ₄)(OH)	Large r > v str	---	TCL	---	Yellow-green	G 3.29 fus	Diss by acids. Pleoc, X and Y yellow-green, Z cols.
1.852 ^v	1.808	<u>1.812</u>	1.838	.030	TOERNEBOHMITE (Ce,La) ₃ Si ₂ O ₈ (OH)	20-30° (43+10°) r < v dist	---	MCL ps hex	---	Light green to olive	H 4.5 G 4.94	Slowly diss by hot concd acids. Pleoc, X rose to greenish-yellow, Y bluish-green, Z rose, abs Y > X = Z.
	1.80	---	1.87	.07	KEYITE (Cu,Zn,Cd) ₃ (AsO ₄) ₂	Disp str	Y = $\frac{b}{c}$ X:c = 10°	MCL	001 good	Deep sky-blue	---	Pleoc, X pale blue, Y greenish blue, Z deep blue. Opt char unk.
1.774 [^]	1.801	<u>1.812</u>	1.824	.023	GADOLINITE Y ₂ Fe ⁺² Be ₂ Si ₂ O ₁₀	85° r < v	X = $\frac{b}{c}$ Z:c = 4-13°	MCL	Conch	Greenish-to brownish-black	H 6.5 G 4.33 infus	Gel with acids. Pleoc, olive to green.
1.837 ^v	1.810	<u>1.813</u>	1.855	.045	DUFRENITE Fe ⁺² Fe ⁺³ ₄ (PO ₄) ₃ (OH) ₅ ·2H ₂ O	Small r > v extr	Z = $\frac{b}{c}$ el neg	MCL fib	010 perf	Dark green	H 3.5-4 G 3.2-3.4 F 2.5	Diss by acids. Pleoc str, X yellow-brown, Y deep blue, Z deep greenish-blue, abs Z > Y > X.
(1.773) [^]	1.808	<u>1.813</u>	1.827	.019	WARWICKITE (Mg,Fe ⁺³ ,Ti,Al) ₂ (BO ₃)O	55°	X = $\frac{c}{b}$ Y = $\frac{b}{c}$ el cTv neg	ORTH pris	100 perf	Dark brown to black	H 3.5-4 G 3.35 infus	Dec by H ₂ SO ₄ . Pleoc, X and Z dark brown, Y red-brown.
	1.81	<u>1.815</u>	~1.85	.04	CORNWALLITE Cu ₅ (AsO ₄) ₂ (OH) ₄ ·H ₂ O	Small	---	MCL fib	Conch	Blackish-green	H 4.5 G 4.5 F 2-2.5	Diss by acids.

1.806 □	1.817	<u>1.818</u>	1.821	.004	CERITE (Ce,Ca) ₉ (Mg,Fe)Si ₇ (O,OH,F) ₂₈	Small r < v very str	---	TRIG	Uneven	Clove- brown, reddish- gray	H 5.5 G 4.8-4.9 infus	Gel with acids.
	1.715	<u>1.820</u>	1.880	.165	DOLEROPHANITE Cu ₂ (SO ₄)O	85° (110±1°) r > v very str crossed	Y = $\frac{b}{c}$ Z:c = -10°	MCL	T01 perf	Chestnut- to dark brown	H 3 G 4.17 fus	Diss by acids. Pleoc, X deep brown, Y brownish-yellow, Z lemon-yellow. Opt char neg?
1.804 ^	1.81	<u>1.82</u>	1.90	.09	AENIGMATITE Na ₂ Fe ₅ TiSi ₆ O ₂₀	30-50°	Z:c = 40°	TCL pris	010, 100	Black	H 5.5 G 3.8 F 3	Insol acids. Tw pl 110. Pleoc, X yellow-brown, Y red- brown, Z dark brown to black.
1.85 v	1.81	---	1.84	.03	DEERITE (Fe ⁺² ,Mn) ₆ (Fe ⁺³ ,Al) ₃ Si ₆ O ₂₀ (OH) ₅	---	Z = $\frac{c}{el}$ pos	MCL acic	110 good	Black	G 3.84	Opt char not stated.
	1.813	<u>1.820</u>	1.830	.017	ANDREWSITE (Cu,Fe)Fe ⁺³ ₃ (PO ₄) ₃ (OH) ₂	Large r > v extr crossed	---	ORTH fib	2 clv	Dark- to bluish- green	H 4 G 3.3-3.4	Diss by acids.
	1.809	<u>1.821</u>	1.857	.048	BETPAKDALITE CaFe ₂ H ₈ (AsO ₄) ₂ (MoO ₄) ₅ · 10H ₂ O	60°	Y = $\frac{b}{c}$ X:c = 12°	MCL pris	---	Lemon- yellow	H 3 G 3.0 F 2	Diss by HCl. Pleoc, X pale yellow, Y greenish-yellow, Z bluish-green, abs Z > Y > X.
1.801 ^ 1.900	1.821	(<u>1.821</u>)	1.825	.004	HUTTONITE cerian phosphatian (Th,Ce)(SiO ₄ ,PO ₄) ₃	21° r < v wk	Y = $\frac{b}{c}$	MCL	---	Yellow to reddish- brown	H 5 G 5.06 infus	(Solid soln with Monazite.)
	1.799	<u>1.822</u>	1.855	.056	AZOPROITE (Mg,Fe ⁺²) ₂ (Fe ⁺³ ,Ti,Mg) BO ₅	(81±14°)	X = $\frac{b}{a}$ Y = $\frac{a}{c}$	ORTH pris	010 good 001 less so, conch	Black	H 5.5 G 3.63	Diss by HCl. Pleoc, X pale green, Y dark green, Z reddish- brown.
1.79 ^	1.806	<u>1.827</u>	2.005	.178	FERRIMOLYBDITE Fe ₂ (MoO ₄) ₃ ·7-10H ₂ O	Small (41±2°) r < v dist	Y = $\frac{a}{c}$ Z = $\frac{c}{b}$	ORTH fib c	001 dist	Sulfur- yellow	H 1-2 G 2.99 F easy	Diss by HCl. Pleoc, X and Y pale yellow, Z yellow-green.
1.880 v	1.82	<u>1.83</u>	1.88	.06	ROCKBRIDGEITE (Fe ⁺² ,Mn)Fe ⁺³ ₄ (PO ₄) ₃ (OH) ₅	Med r > v very str	X = $\frac{c}{b}$	ORTH fib botry- oidal	100 perf 010 fair	Dark green	H 4.5 G 3.51 F easy	Diss by HCl. Pleoc, X yellow-brown, Y bluish-green, Z deep blue, abs Z > X > Y.
1.805 ^ 1.865	1.825	<u>1.832</u>	1.940	.115	LUDWIGITE (Mg,Fe) ₂ Fe ⁺³ BO ₅	25-40° r < v str	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	Uneven, parting 001	Greenish- black	H 5 G 3.9 F diff	Diss by acids. Pleoc, X blue-green, Y dark green, Z dark brown, abs Z > X > Y.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
2.02	---	<u>1.833</u>	---	wk	RUSAKOVITE (Fe,Al) ₅ (VO ₄ ,PO ₄) ₂ (OH) ₉ •3H ₂ O	~ 50° r > v str	---	Cryptocryst	---	Yellow-orange	H 1.5-2 G 2.76	Diss by acids. Opt char unk.
	1.820	<u>1.835</u>	1.920	.100	VOLBORTHITE Cu ₃ (VO ₄) ₂ •3H ₂ O	63° (47+3°) r < v str	---	MCL	001 perf	Yellow-green	H 3-4 G 3.4-3.6	Diss by acids. Pleoc, X and Y yellow, Z yellow-green.
1.813 ∧ 1.855	1.832	<u>1.837</u>	1.890	.058	DUFRENITE Fe ⁺² Fe ⁺³ ₄ (PO ₄) ₃ (OH) ₅ •2H ₂ O	Small r < v extr	Z = b el cTv neg	MCL fib	010 perf	Dark green to red-brown	H 3.5-4 G 3.2-3.4 F 2.5	Diss by acids. Pleoc str, X deep blue-green, Y yellow-brown, Z deep olive-brown.
1.833 └	1.792	<u>1.840</u>	1.888	.096	LAUTARITE Ca(IO ₃) ₂	~ 90° r > v mod	Y = b X:c = 25°	MCL pris	011 good	Cols to light yellow	H 3.5-4 G 4.59 F 1.5	Slightly sol in H ₂ O, diss by HCl.
	1.802	<u>1.840</u>	1.888	.086	ORTHOERICSSONITE BaMn ₂ (Fe ⁺³ O)Si ₂ O ₇ (OH)	~ 50° (85+3°) r > v str	X = b Y = c el cTv neg	ORTH	100 perf	Brownish-black	H 5.5 G 4.22	Pleoc, X yellow-brown, Y reddish brown, Z dark brown, abs Z > Y > X.
	1.842	<u>1.842</u>	1.848	.006	PARATACAMITE Cu ₂ (OH) ₃ Cl	0-50° r > v	---	TRIG	1011 good	Dark green	H 3 G 3.74 F easy	Diss by acids. Poly tw 1011.
1.82 ^	---	~1.84	---	---	ITOITE Pb ₃ Ge(SO ₄) ₂ O ₂ (OH) ₂	---	---	ORTH acic	---	White	G 6.67 calc	Opt char unk.
	1.840	<u>1.847</u>	1.892	.052	LAUBMANNITE Fe ⁺² ₃ Fe ⁺³ ₆ (PO ₄) ₄ (OH) ₁₂	Med r < v extr	---	ORTH fib	Clv fibers	Gray-green, yellow-green	H 3.5 G 3.33	Diss by acids. Pleoc, X buff, Y greenish-brown to olive-green, Z reddish-brown, abs Z > Y > X.
	1.840	---	1.870	.030	DEERITE (Fe ⁺² ,Mn) ₆ (Fe ⁺³ ,Al) ₃ ⁹ Si ₆ O ₂₀ (OH) ₅	---	Z = c el pos	MCL acic	110 good	Black	G 3.84	Opt char unk.
	1.85	<u>1.85</u>	1.88	.03	PARWELITE (Mn,Mg) ₅ Sb(As,Si) ₂ O ₁₂	27° r > v str	---	MCL pris	010 fair	Yellowish brown	H 5.5 G 4.62	Insol in dil HCl.

$\sqrt{1.89}$	1.85	<u>1.85</u>	1.87	.02	DUMONTITE $Pb_2(UO_2)_3(PO_4)_2(OH)_4 \cdot 3H_2O$	Small $r < v$ str	$X = \frac{b}{c}$ $Y = \frac{c}{a}$	MCL	---	Yellow	G 5.7	Diss by acids.
1.812 ^	1.845	<u>1.852</u>	1.878	.033	TOERNEBOHMITE $(Ce,La)_2AlSi_2O_8(OH)$	20-40° (55+8°) $r < v$ dist	---	MCL ps hex	---	Light green to olive	H 4.5 G 4.94	Slowly diss by hot concd acids. Pleoc, X rose to greenish- yellow, Y bluish- green, Z rose.
1.854 neg ^	1.820	<u>1.854</u>	1.888	.068	LIEBENBERGITE (Olivine grp) $(Ni,Mg,Fe,Co)_2SiO_4$	92° $r < v$	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010 poor to fair 100 poor	Yellowish green	H 6-6.5 G 4.60	X, Y cols to pale green, Z greenish yellow. NiO 56.3, MgO 6.5, FeO 4.4 CoO 1.8%.
	1.852	(1.854)	1.867	.015	LUDDENITE $Pb_2Cu_2Si_5O_{14} \cdot 14H_2O$	40°	---	MCL	---	Green	H 4 G 4.45 F easy	Pleoc, X and Y yellow-green, Z emerald-green.
1.837 ^	1.845	<u>1.855</u>	1.890	.045	DUFRENITE $Fe^{+2}Fe^{+3}_4(PO_4)_3(OH)_5 \cdot 2H_2O$	Small (57+6°) $r < v$ str	$Z = \frac{b}{a}$ el cTv neg	MCL fib	010 perf	Reddish- brown	H 3.5-4 G 3.2-3.4 F 2.5	Diss by acids. Pleoc str, X pale yellow- brown, Y yellow- brown, Z red-brown.
	---	<u>1.855</u>	>1.88	---	ANANDITE (Mica grp) $(Ba,K)(Fe,Mg)_3(Si,Al,Fe)_4O_{10}(OH)_2$	Disp str	$Y = \frac{b}{a}$ $Z:a = 12^\circ$ el clv pos	MCL ps hex plates	001 perf	Black	H 3-4 G 3.94	Nearly opaque. Pleoc, Y green, Z brown.
$\sqrt{1.92}$	1.85	<u>1.86</u>	1.92	.07	PURPURITE $(Mn^{+3},Fe^{+3})PO_4$	Med	$X = \frac{a}{b}$	ORTH	100 good 010 less so	Deep red or purple	H 4-4.5 G 3.4 F easy	Diss by acids. Pleoc str, X dark brownish- gray, Y scarlet, Z purple.
	1.835	---	1.910	.075	DEMESMAEKERITE $Pb_2Cu_5(UO_2)_2(SeO_3)_6(OH)_6 \cdot 2H_2O$	---	---	TCL el \perp 100	---	Bottle- green	H 3-4 G 5.28	Diss by dilute HNO ₃ .
1.751 ^ 1.88	1.851	<u>1.864</u>	1.894	.043	SYNADELPHITE $(Mn,Mg,Ca,Pb)_9(AsO_3)(AsO_4)_2(OH)_9 \cdot 2H_2O$	40° (68+6°) $r > v$	---	TCL ps orth	010 imperf	Cols to red-brown	H 4.5 G 3.8 F 2	Diss by HNO ₃ . Pleoc, X light brown, Y brown, Z dark red- brown.
1.832 ^	1.850	<u>1.865</u>	1.985	.135	LUDWIGITE $(Mg,Fe)_2Fe^{+3}BO_5$	25-40° (41+3°) $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{a}$	ORTH	Uneven, parting 001	Greenish- black	H 5 G 4.0 F diff	Slowly diss by acids. Nearly opaque, X and Y dark green, Z brownish-black.
$\sqrt{1.907}$	1.843	<u>1.870</u>	1.943	.100	TITANITE $(Ca,Y)TiSiO_4(O,OH)$	38° (65+3°) $r > v$ str	$Y = \frac{b}{c}$ $Z:c = 40^\circ$	MCL wedges	110 good	Yellow to reddish- brown	H 4.5 G 3.56 F 3	Dec by hot H ₂ SO ₄ . Pleoc, X pale yellow, Y pale greenish, Z pink to reddish- brown.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _z (2V _z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
196	1.817	1.879	2.057	.240	UVANITE (U,Ca) ₂ V ₆ O ₂₁ ·15H ₂ O (?)	52° (66±1°)	---	ORTH(?)	2 clv pinacoidal	Brownish-yellow	---	Diss by NH ₄ OH. Pleoc, X light brown to yellow, Y dark brown, Z greenish yellow. Not fluor in UV.
	1.841	(~1.88)	1.935	.094	PERETAITE CaSb ₄ O ₄ (OH) ₂ (SO ₄) ₂ ·2H ₂ O	Large	---	---	100	Cols	G 4.06	---
	1.87	1.880	1.98	.11	LEITEITE ZnAs ₂ O ₄	26° r < v str	Y = $\frac{b}{c}$ Z:c = 10°	MCL	100 perf	Cols to brown	G 4.61	Lam flexible, inelastic.
	1.910 v	1.880	1.935	.058	KASOLITE Pb(UO ₂)SiO ₄ ·H ₂ O	43° (27±8°)	X = $\frac{b}{c}$ Z:c = 1° el clv neg	MCL pris	001 perf 100, 010 indist	Ocher-yellow to amber	H 4.5 G 5.8-6.5	Gel with acids. Pleoc wk or absent.
	1.96 v	1.88	---	high	CHENEVIXITE Cu ₂ Fe ₂ (AsO ₄) ₂ (OH) ₄ ·H ₂ O	---	X = $\frac{b}{c}$ Z ~ $\frac{c}{b}$	MCL cryptocryst	---	Dark green to greenish-yellow	H 3.5-4.5 G 3.93 F 2.5	Diss by acids.
	1.83 ^	1.880	1.897	.022	ROCKBRIDGEITE (Fe,Mn)Fe ⁺³ ₄ (PO ₄) ₃ (OH) ₅	Med r > v very str	X = $\frac{c}{b}$	ORTH fib	100 perf 010 fair	Dark green	H 4.5 G 3.5	Diss by HCl. Pleoc, X yellow-brown, Y bluish-green, Z dark bluish-green.
	>1.87	(1.88)	<1.89	---	STIBIVANITE Sb ⁺³ ₂ V ⁺⁴ O ₅	85° r > v str	X = $\frac{b}{c}$ Z = $\frac{c}{b}$	MCL fib	---	Yellow-green	---	Opt sign unk. Pleoc, X and Y emerald-green, Z olive-green.
	---	1.88	---	---	RAUVITE Ca(UO ₂) ₂ V ₁₀ O ₂₈ ·16H ₂ O	---	---	Mass	---	Purplish-black	---	Opt char unk.
	(1.697) ^	1.883	1.894	.017	ANGLESITE (Barite grp) PbSO ₄	68-75° r < v str	Y = $\frac{b}{c}$ X = $\frac{c}{b}$	ORTH	001 good 210 dist	Cols	H 2.5-3 G 6.38 F 1.5	Slowly diss by HNO ₃ .
	1.85 ^	1.89	1.90	.020	DUMONTITE Pb ₂ (UO ₂) ₃ (PO ₄) ₂ (OH) ₄ ·3H ₂ O	Large r < v str	X = $\frac{b}{c}$ Y = $\frac{c}{b}$	MCL	---	Yellow	G 5.7	Diss by acids.
	1.89	1.89	2.00	.11	DENNINGITE (Mn,Zn)Te ₂ O ₅	0-15°	---	TET mass	001 perf	Pale green	H 4 G 5.05 F easy	Diss by cold HCl, insol in HNO ₃ .

(1.821)
^1.870
↓
1.970

1.882	(1.895)	1.915	.033	HALLIMONDITE $\text{Pb}_2(\text{UO}_2)(\text{AsO}_4)_2$	80° r > v	Z':c = 11° on 100, X':b = 9° on 001	TCL tab	Conch	Yellow	H ~ 3 G 6.39	Diss by HNO_3 .
1.888	1.895	1.915	.027	Unknown	62° r > v str	---	---	---	Pale green	---	Labelled "scorodite," Kiura, Japan. Larsen (1921).
1.898	(1.898)	1.915	.017	HUEGELITE $\text{Pb}_2(\text{UO}_2)_3(\text{AsO}_4)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	Small	---	MCL tab 100	100 good	Orange- yellow to brown	---	Pleoc, X yellow, Y orange-yellow, Z cols to pale yellow. Anom interf colors.
1.898	(1.899)	1.915	.017	HUEGELITE $\text{Pb}_2(\text{UO}_2)_3(\text{AsO}_4)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	25°	---	MCL	100 very good, conch	Brown to orange	H 2.5	Pleoc, X yellow, Y yellow to orange, Z cols to pale yellow.
<1.89	1.90	1.95	>.06	LAMMERITE $\text{Cu}_3(\text{AsO}_4)_2$	54+5°	X = $\frac{b}{c}$ Z:c = 40°	MCL tab	010 perf 100 good	Dark green	H 3.5-4 G 5.18	Pleoc, X bright blue, Y sky blue, Z bluish- green.
1.810	1.900	>2.01	>.20	DUTTONITE $\text{VO}(\text{OH})_2$	60° r < v mod	X = $\frac{a}{c}$ Y = $\frac{c}{a}$	MCL	---	Light brown	H 2.5 G 3.24	Pleoc, X pale pinkish-brown, Z pale brown.
1.890	1.90	1.99	.10	BELLINGERITE $\text{Cu}_3(\text{IO}_3)_6 \cdot 2\text{H}_2\text{O}$	Med r > v str	---	TCL pina- coidal	Subconch	Light green	H 4 G 4.89	Diss by HCl. Gives iodine vapor when heated in closed tube.
1.898	1.900	1.922	.024	HUTTONITE ThSiO_4	25° r < v mod	Y = $\frac{b}{c}$ Z ~ $\frac{c}{b}$	MCL	---	Cols to pale cream	G 7.1 infus	---
1.899	(1.901)	1.903	.004	QUEITITE $\text{Pb}_4\text{Zn}_2(\text{SO}_4)(\text{SiO}_4)(\text{Si}_2\text{O}_7)$	~ 90° r < v str	X = $\frac{b}{a}$ Z:a varies	MCL tab	---	Cols to pale yellow	H 4 G (6.07)	Diff diss by HNO_3 . Z:a = 40° at 405 nm, 0° at 592 nm.
1.900	1.907	2.034	.134	TITANITE CaTiSiO_5	27° r > v str	Y = $\frac{b}{c}$ Z:c = 40- 50°	MCL	110 good	Yellow, brown, cols	H 5 G 3.55 F 4	Dec by hot H_2SO_4 . Pleoc, X pale yellow, Y greenish, Z red to brown.
1.85	1.91	(2.20)	.35	SPIROFFITE $(\text{Mn}, \text{Zn})_2\text{Te}_3\text{O}_8$	55°	---	MCL mass	2 clv, conch	Red to purple	H 3.5 G 5.01 F easy	Diss by Hcl.
1.890	1.91	1.977	.087	SCHULTENITE PbHAsO_4	58° disp str	X = $\frac{b}{c}$ Z:c = 66° el pos	MCL thin plates, 010	010 good	Cols	H 2.5 G 5.94	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.880 ^	1.895	1.910	1.950	.055	KASOLITE $\text{Pb}(\text{UO}_2)\text{SiO}_4 \cdot \text{H}_2\text{O}$	43° (64±5°)	X = $\frac{b}{c}$ Z:c ~ 1° el clv neg	MCL pris	001 perf 100, 010 indist	Ocher-yellow to amber-brown	H 4.5 G 5.8-6.5	Gel with acids. Pleoc wk or absent.
	---	>1.9 <2.0	---	.110	PAULMOOREITE $\text{Pb}_2\text{As}^{+3}_2\text{O}_5$	65° r > v very str	Y = $\frac{b}{c}$ Z:c = 10° el clv neg	MCL	001 perf	Cols to light orange	H ~ 3 G 7.0:Luster	Diss by 1:1 HNO ₃ . adamantine.
	1.91	1.91	1.945	.035	GANOMALITE $\text{Pb}_6\text{Ca}_4\text{Si}_6\text{O}_{21}(\text{OH})_2$	Small	---	HEX tab 0001	10T0 perf 0001 less so	Gray	H 3 G 5.74 F 3	Gel with HNO ₃ .
	1.871	1.92	2.01	.139	CLAUDETITE As_2O_3	58° r < v str	Y = $\frac{b}{c}$ Z:c = 5°	MCL thin tab 010	010 perf	Cols	H 2.5 G. 4.15 volat	Slightly sol in hot H ₂ O. Tw on 100, penet.
	1.89	1.92	1.95	.06	TSUMEBITE $\text{Pb}_2\text{Cu}(\text{PO}_4)(\text{SO}_4)(\text{OH})$	88° r < v str	---	MCL tab	Uneven	Emerald-green	H 3.5 G 6.1 fus	Diss by HNO ₃ . Pleoc, X and Y pale blue, Z robin's egg blue.
1.86 ^	---	1.92	---	.04	PURPURITE $(\text{Mn}^{+3}, \text{Fe}^{+3})\text{PO}_4$	38°	X = $\frac{a}{c}$	ORTH	100 good 010 less so	Deep red or purple	H 4-4.5 G 3.2-3.4 F easy	Diss by acids. Pleoc str, X dark brownish- gray, Y scarlet, Z purple.
	1.920	(1.921)	1.943	.023	OTJISUMEITE PbGe_4O_9	20°	X:c = 3-5°	TCL	001 in traces	Cols to white	H 3 G (5.77)	Diss by hot HNO ₃ .
	1.880	1.928	2.029	.149	CESBRONITE $\text{Cu}_5(\text{TeO}_3)_2(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	72° r > v mod	X = $\frac{a}{c}$ Y = $\frac{c}{b}$	ORTH el $\frac{a}{c}$	021 good 010 poor	Green	H 3 G 4.45 F easy	Diss by acids. Pleoc, X pale bluish- green, Y yellow- green, Z deep emerald-green.
[~1.93	~1.93	1.97	.04	ZIRCON ZrSiO_4	0-10°	---	TET pris	110	Brown, pink, cols	H 7.5 G 4.7 infus	Insol in acids. Abnorm biax.
	1.920	1.960	2.20	.28	GRAEMITE $\text{CuTeO}_3 \cdot \text{H}_2\text{O}$	48°	---	ORTH	010 good parting 100	Blue- green	H 3-3.5 G 4.13 F easy	Diss by acids. Pleoc, X yellow- green, Y and Z blue- green.
1.88 ^	1.92	1.96	2.04	.12	CHENEVIXITE $\text{Cu}_2\text{Fe}_2(\text{AsO}_4)_2(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	(73±10°)	X = $\frac{b}{c}$ Z ~ $\frac{c}{b}$	MCL	---	Dark green	H 4 G 3.5 F 2.5	Diss by acids.

	1.963	<u>1.963</u>	1.966	.003	HYALOTEKITE $\text{Pb}_2\text{Ba}_2\text{Ca}_2\text{B}_2(\text{Si,Be})_2\text{Si}_{80}\text{F}_{28}$	Small $r < v$ str	Opt pl \perp clv	TCL	2 clv at 90°	Cols	H 5-5.5 G 3.80 F 3 (?)	Insol in acids.
	1.952	---	2.002	.052	KURANAKHITE $\text{PbMn}^{+4}\text{Te}^{+6}\text{O}_6$	---	---	ORTH ps hex	---	Brown to black	H 4-5	Diss by HCl with evolution of chlorine. Opt sign unk.
1.907 ^	1.950	<u>1.970</u>	2.092	.142	TITANITE CaTiSiO_5	18° (46+2°) $r > v$ str	$Y = \frac{b}{c} = 40-50^\circ$	MCL	110 good	Yellow, brown, cols	H 5 G 3.52 F 4	Dec by hot H_2SO_4 . Pleoc, X pale yellow, Y greenish, Z red to brown.
	1.95	<u>1.97</u>	1.99	.04	BAYLDONITE $\text{Pb}(\text{Cu,Zn})_3(\text{AsO}_4)_2(\text{OH})_2 \cdot \text{H}_2\text{O}$	Large $r < v$ str	$X = \frac{b}{c}$ $Y:el = 45^\circ$	MCL fib \underline{c}	---	Grass- to dark- green	H 4.5 G 5.6 F easy	Diss by HNO_3 . Tw.
	1.96	<u>1.98</u>	2.06	.10	URANOSPHERITE $\text{Bi}_2\text{U}_2\text{O}_9 \cdot 3\text{H}_2\text{O}$	56° $r < v$ str	$Y = \frac{b}{c}$ $Z = \underline{c}$	MCL el \underline{c}	100	Reddish- orange, yellow	H 2-3 G 6.9	Diss by acids.
2.05 v	2.00	<u>2.01</u>	2.02	.02	CALCIOVOLBORTHITE $\text{CaCuVO}_4(\text{OH})$	Large $r > v$ str	$X = \frac{a}{b}$ $Y = \underline{b}$	ORTH	010, 001	Greenish- yellow	H 3-4 G 3.75 F 2 (?)	Diss by HCl. Pleoc wk.
	1.84	<u>>1.84</u>	>>1.84	high	METAROSSITE $\text{CaV}_2\text{O}_6 \cdot 2\text{H}_2\text{O}$	Large	---	TCL	One perf	Dull yellow	Soft F easy	Diss by HCl.
1.835 ^ 2.05	2.01	<u>2.02</u>	2.06	.05	VOLBORTHITE $\text{Cu}_3(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	Large $r < v$ str	---	MCL	001 perf 010	Dark green to yellow- green	H 3-4 G 3.42-3.62	Diss by acids. Tw. Pleoc, X and Y yellow, Z yellow- green.
	1.958	<u>2.037</u>	2.245	.287	SULFUR S	69° $r < v$	$X = \frac{a}{b}$ $Y = \underline{b}$	ORTH pris	001, 110, 111 im- perf	Yellow	H 2 G 2.07 F 1	Insol in acids, diss by CS_2 . Burns. Tw pl 10f. Pleoc.
2.11 v	1.98	<u>2.040</u>	2.13	.15	FERSMITE $(\text{Ca,Ce,Na})(\text{Nb,Ta,Ti})_2(\text{O,OH,F})_6$	Med (82+15°)	---	ORTH	Uneven subconch	Dark yellow, brown, black	H 4-5 G 4.7-4.9 infus	Nearly insol in acids.
	2.016	<u>2.040</u>	2.130	.114	GAMAGARITE $\text{Ba}_4(\text{Fe,Mn})_2\text{V}_4\text{O}_{15}(\text{OH})_2$	47-62° $r < v$ str	$Y = \underline{b}$	MCL	2 clv	Dark brown	H 4.5-5 G 4.63	Pleoc, red-brown to yellow.
2.25 v	1.90	<u>2.05</u>	2.50	.60	MANGANITE $\text{MnO}(\text{OH})$	Small (71+5°) $r > v$ str	$X \sim \frac{a}{b}$ $Y \sim \underline{b}$	MCL pris	010 perf 110, 001 good	Steel- gray to iron- black	H 4 G 4.33 infus	Diss by HCl. Pleoc wk, abs Z > X and Y.
	---	<u>2.05</u>	---	high	FERNANDINITE $\text{CaV}^{+4}_2\text{V}^{+5}_2\text{O}_{30} \cdot 14\text{H}_2\text{O} (?)$	---	---	Crypt- ocryst	---	Dull green	---	Diss by acids. Not pleoc. Opt char unk.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
2.01 ^	2.01	<u>2.05</u>	2.09	.08	CALCIOVOLBORTHITE CaCuVO ₄ (OH)	83° r > v str	X = <u>a</u> Y = <u>b</u>	ORTH	010, 001	Greenish-yellow	H 3-4 G 3.75 F 2 (?)	Diss by HCl. Pleoc wk.
2.07 ┘	2.05	<u>2.05</u>	2.07	.02	CARMINITE PbFe ₂ (AsO ₄) ₂ (OH) ₂	Med r < v str	X = <u>c</u>	ORTH acic	Pris	Carmine to lilac	H 3.5 G 5.22 F easy	Diss by HNO ₃ . Pleoc, X and Z red-brown, Y greenish-brown.
	1.96	<u>2.055</u>	>2.11	>.15	LUDLOCKITE (Fe ⁺² , Pb)As ⁺⁵ ₂ O ₆	---	Z ~ <u>a</u> el pōs	TCL tw	011 perf	Red	H 1.5-2 G 4.40	Diss by concd acids. Sectile and flexible. Pleoc, X yellow, Y deep yellow, Z orange-yellow.
	>2.01	<u>>2.01</u>	>2.01	high	GERSTLEYITE Na ₂ (Sb, As) ₈ S ₁₃ ·2H ₂ O	Large (?)	---	MCL	010, 101 perf	Cinnabar to blackish-red	H 2.5 G 3.62 F 2	Diss by alkalis. Pleoc, X salmon-red, Y and Z deep blood-red. Opt char unk.
	>2.01	<u>>2.01</u>	>2.01	.07	SUNDIVSITE Pb ₁₀ (SO ₄)Cl ₂ O ₈	---	---	MCL	100 perf	Cols to white	H 3 G 7.0	---
	---	<u>2.087</u>	---	wk	SENARMONTITE Sb ₂ O ₃	---	---	CUB oct	111 in traces	Cols	H 2 G 5.50 F 1.5 volat	Diss by HCl. Anom biaxial.
	>2.05	<u>>2.05</u>	>2.05	~.14	RYNERSONITE Ca(Ta, Nb) ₂ O ₆	---	X = <u>c</u> Y = <u>b</u>	ORTH fib	Uneven	Creamy white to pink	H ~ 4.5 G 6.40	Pleoc, X and Y straw yellow, Z light straw yellow.
	2.05	(<u>2.11</u>)	2.20	.15	BLIXITE Pb ₂ Cl(O, OH) ₂	80°	---	ORTH	One dist	Pale yellow	H 3 G 7.35	Diss by HNO ₃ .
2.040 ^	2.095	<u>2.11</u>	2.23	.135	FERSMITE (Ca, Ce, Na)(Nb, Ta, Ti) ₂ (O, OH, F) ₆	Med	---	ORTH	Uneven, subconch	Brown to black	H 4-5 G 4.7-4.9 infus	Nearly insol in acids.
	2.08	<u>2.11</u>	---	---	DUHAMELITE Pb ₂ Cu ₄ Bi(VO ₄)(OH) ₃ ·8H ₂ O	---	Z:c = 3°	ORTH	---	Yellow-green	H 3 G 5.80	Opt sign unk. Pleoc wk in yellow. Diss by HNO ₃ .
	2.110	<u>2.112</u>	2.165	.055	KHINITE PbCu ₃ Te ⁺⁶ O ₄ (OH) ₆	20°	---	ORTH	001 fair	Dark green	H 3.5 G 6.7	---

2.10	---	2.30	.20	ROOSEVELTITE BiAsO_4	---	---	MCL	Conch	Gray to yellow	H 4-4.5 G 6.9-7.1 F 2-3	Diss by acids. Opt char unk.
>2.11	---	<2.13	<.02	SCHNEIDERHOEHNITE $\text{Fe}^{+2}_8\text{As}^{+3}_{10}\text{O}_{23}$	---	---	TCL	100 perf 2 others	Dark brown	H 3 G 4.3	Pleoc, reddish-brown to bright yellow, abs $X < Y < Z$.
2.115	<u>2.135</u>	2.26	.125	RAJITE $\text{CuTe}^{+4}_2\text{O}_5$	40° disp wk	---	MCL	010	Green	H 4 G 5.75 fus	Diss by cold acids. Pleoc wk in green, abs $Z > Y > X$.
2.01	---	---	---	SANTAFEITE $\text{Na}_2(\text{Mn}, \text{Ca}, \text{Sr})_6\text{Mn}^{+4}_3(\text{V}, \text{As})_6\text{O}_{28} \cdot 8\text{H}_2\text{O}$	---	$X = \underline{c}$	ORTH ros- ettes	010 perf 110 dist	Brownish-black	G 3.38 F easy	Diss by HCl. Pleoc yellow-brown to dark red-brown, abs $X > Y > Z$. Opt char unk.
2.14	---	2.15	.01	EZTLITE $\text{Pb}_2\text{Fe}^{+3}_6(\text{Te}^{+4}\text{O}_3)_3(\text{Te}^{+6}\text{O}_6)(\text{OH})_{10} \cdot 8\text{H}_2\text{O}$	---	$Z:\underline{c} = 3^\circ$	MCL	001	Deep orange	---	Sign unk. Not pleoc.
---	<u>2.15</u>	---	high	CUPROTUNGSTITE $\text{Cu}_2(\text{WO}_4)(\text{OH})_2$	---	---	Cryptocryst	---	Green	Fus	Dec by HCl. Opt char unk.
2.14	<u>2.15</u>	2.18	.04	ATELESTITE $\text{Bi}_8(\text{AsO}_4)_3\text{O}_5(\text{OH})_5$	44° $r < v$ str	---	MCL	010 indist	Sulfur-yellow	H 4.5-5 G 6.8-7.1 F 1.5	Diss by HCl.
>2.09	<u>>2.09</u>	>2.09	---	MOUNANAITE $\text{PbFe}_2(\text{VO}_4)_2(\text{OH})_2$	---	On 010, ext: $\underline{c} = 38^\circ$	TCL el \underline{c}	---	Brownish-red	G 4.85	Tw common. Pleoc, brownish-red to brownish-yellow. Opt char unk.
2.130	<u>2.16</u>	2.195	.065	PREISINGERITE $\text{Bi}_3(\text{AsO}_4)_2\text{O}(\text{OH})$	90°	---	TCL tab	---	White to gray	G (7.24)	---
2.12	<u>2.17</u>	2.31	.19	MELANOTEKITE $\text{Pb}_2\text{Fe}_2\text{Si}_2\text{O}_9$	67° $r > v$ str	---	ORTH pris	2 clv	Black	H 6.5 G 5.73 F 2-2.5	Dec by HNO_3 . Pleoc str, X nearly cols, Y pale red-brown, Z deep reddish-brown.
2.15	<u>2.17</u>	2.25	.10	MANGANOTANTALITE $\text{Mn}(\text{Ta}, \text{Nb})_2\text{O}_6$	$r < v$ disp extr	$X = \underline{a}$ $Y = \underline{b}$	ORTH	010 perf	Deep red to black	H 5-6 G 7.5 infus	Insol in acids. Pleoc, X and Y deep red, Z pink.
2.17	<u>2.17</u>	2.18	.01	GEORGIADESITE $\text{Pb}_8(\text{AsO}_4)_2\text{Cl}_7(\text{OH})_3$	Large $r < v$ str	$Y = \underline{b}$ $Z = \underline{c}$	MCL	---	White	H 3.5 G 7.1 F easy	Diss by HNO_3 .
2.11	<u>2.18</u>	2.22	.11	BALYAKINITE CuTeO_3	80°	$X = \underline{a}$ $Y = \underline{b}$	ORTH	---	Gray-green to bluish-green	G 5.6	---
2.14	---	2.24	.10	FORNACITE $(\text{Pb}, \text{Cu})_3[(\text{Cr}, \text{As})\text{O}_4]_2(\text{OH})$	---	---	MCL	---	Olive-green	G 6.2	Slightly pleoc. Opt char uncertain.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
v ~2.23	2.00 _{Li}	2.18 _{Li}	2.35 _{Li}	.35	TELLURITE TeO ₂	~ 90° r < v med	X = $\frac{b}{c}$ Z = $\frac{c}{a}$ el pos	ORTH acid <u>c</u>	010 perf	Colc	H 2 G 5.90 F 2	Diss by HCl.
	2.14	---	2.23	.09	WODGINITE (Ta,Nb,Sn,Mn,Fe) ₁₆ O ₃₂	---	Z:c = 26°	MCL	---	Red-brown	H 6 G 7.6 infus	Insol in acids. Pleoc, light yellow to reddish-brown.
	2.14	---	2.315	.18	VIGEZSITE (Ca,Ce)(Nb,Ta,Ti) ₂ O ₆	Large	X = $\frac{c}{a}$ Z = $\frac{a}{b}$	ORTH prism	100 dist	Orange-yellow	G (5.34)	Sign unk. Not pleoc.
	1.94	2.20	2.51	.57	LEPIDOCROCITE FeO(OH)	83° (96±4°)	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH blades 100	010 perf 100 less so 010 fair	Red	H 5 G 4.09 infus	Slowly diss by HCl. Pleoc str, X yellow, Y red, Z deeper red.
	2.13	2.20	2.40	.27	ANGELELLITE Fe ₄ (AsO ₄) ₂ O ₃	Med large	---	TCL	001	Blackish-brown	H 5.5 G 4.86	Pleoc str, yellow to deep red.
	2.10	2.20	2.31	.21	KENTROLITE Pb ₂ Mn ₂ Si ₂ O ₉	88° r < v str	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH pris	110 dist	Dark reddish-brown	H 5 G 6.19 F 2-2.5	Dec by HNO ₃ . Pleoc, X pink, Z brownish-red, abs Z > Y > X.
	2.19	2.20	2.33	.14	TRIPUHYITE FeSb ₂ O ₆	Small r < v very str	---	TET mass	---	Greenish-yellow, yellow-brown	H ~ 6 G 5.6-5.8 F 4-5 (?)	Insol in acids.
	2.18	(2.21)	2.26	.08	EUXENITE (Y,Ca,Ce,U,Th) (Nb,Ta,Ti) ₂ O ₆	70°	Y = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH	Conch	Brownish-black	H 5-5.5 G 5.0-5.4 infus	Dec by hot concd acids.
2.17 ∇ 2.32	2.17	2.21	2.25	.08	MANGANOTANTALITE (Mn,Fe)(Ta,Nb) ₂ O ₆	Large r < v str	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH pris	101 dist 100 poor	Red to reddish-brown	H 6 G 7.52 infus	Insol in acids. Pleoc, X and Y deep red, Z pink, abs X > Y >> Z.
v 2.30	2.19	2.21	2.25	.06	AESCHYNITE-(Y) (Y,Ca,Fe,Th)(Ti,Nb) ₂ (O,OH) ₆	74°	el neg	ORTH	100, 010, 001 perf uneven	Yellow	H 5 G 4.4 infus	Insol in HCl, diss by H ₂ SO ₄ .
	---	2.21	---	low	ROMEITE (Ca,Fe,Mn,Na) ₂ (Sb,Ti) ₂ O ₆ (O,OH,F)	Large	---	CUB oct	111 imperf fr uneven	Yellow	H 6.5 G 5.0 infus	Insol in acids. Abnorm violet interf colors.
	2.185	2.219	2.266	.081	HEYITE Pb ₅ Fe ⁺² ₂ (VO ₄) ₂ O ₄	89° (82±3°)	Y = $\frac{b}{c}$ X:c = 36°	MCL	Uneven	Yellow-orange	H 4 G 6.3	Tw 110.

2.36 v	2.17	<u>2.22</u>	2.32	.15	HUEBNERITE (Wolframite ser) (Mn,Fe)WO ₄	73°	$X = \frac{b}{c}$ $Z:c = 18^\circ$ el clv pos	MCL bladed, tab 100	010 perf	Brownish- red	H 5-5.5 G 7.35 F 4	Dec by HCl with sepn of yellow WO ₃ . Pleoc, X pale yellow, Y yellow-brown, Z green.
	2.20	<u>2.22</u>	2.26	.06	COTUNNITE PbCl ₂	67°	$X = \frac{b}{c}$ $Z = \frac{c}{a}$ el cTv pos	ORTH acic <u>a</u>	010 perf	White, yellow	H 2.5 G 5.8 F 1	Diss by hot water.
	2.19	<u>2.23</u>	2.35	.16	PLUMBOTELLURITE PbTeO ₃	~ 50°	---	ORTH	---	Gray to brown	Soft G 7.2	---
~2.18 ^	2.20	---	2.27	.07	WODGINITE (Ta,Nb,Sn,Mn,Fe) ₁₆ O ₃₂	---	---	MCL	---	Red-brown, dark brown	H 6 G 7.2-7.4 infus	Insol in acids.
2.05 ^	2.24 _{Li}	<u>2.25</u> _{Li}	2.53 _{Li}	.29	MANGANITE MnO(OH)	Small r > v str	$X \sim \frac{a}{b}$ $Y \sim \frac{b}{c}$	MCL pris	010 perf 110, 001 less so	Steel- gray to black, streak brownish	H 4 G 4.33 infus	Diss by HCl. Pleoc wk, abs Z > X and Y.
	2.24	---	2.26	.02	OBOYERITE Pb ₆ H ₆ (TeO ₃) ₃ (TeO ₆) ₂ · 2H ₂ O	---	---	TCL fib	---	White	H 1.5 G 6.46	Opt sign unk.
	2.18	<u>2.26</u>	2.35	.17	DESCLOIZITE Pb(Zn,Cu)VO ₄ (OH)	~ 90° r < v str	$X = \frac{c}{b}$ $Y = \frac{b}{a}$	ORTH	Conch	Brownish- red, brown, black	H 3.5 G 5.9-6.2 F 1.5	Diss by HNO ₃ . Pleoc, X and Y yellow, Z brownish.
	2.17	<u>2.26</u>	2.34	.17	MOTTRAMITE Pb(Cu,Zn)VO ₄ (OH)	~ 90° r < v str	$X = \frac{c}{b}$ $Y = \frac{b}{a}$	ORTH	Conch	Reddish- brown, olive- green	H 3.5 G 5.9-6.2 F 1.5	Diss by HNO ₃ . Pleoc, X and Y greenish- yellow, Z brownish- yellow.
	2.20	(~2.27)	2.42	.22	SCHUMACHERITE Bi ₃ (VO ₄) ₂ O(OH)	Large	$X:c = 22^\circ$	TCL tab <u>b</u>	Conch	Yellow	H 3 G 6.90	---
	2.24	<u>2.27</u>	2.31	.07	MENDIPITE Pb ₃ O ₂ Cl ₂	~ 90° r < v str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el pos	ORTH fib	110 perf 100, 010 less so	White	H 2.5 G 7.24 F 1	Diss by HNO ₃ .
	2.27	<u>2.27</u>	2.30	.03	RASPITE PbWO ₄	Very small	$Y = \frac{b}{c}$ $Z:c = 30^\circ$	MCL tab 001	100 perf	Brownish- yellow	H 2.5-3 G 8.46 F 2.5-3	Dec by HNO ₃ . Abs Z > X and Y.
2.21 ^ 2.32 neg	2.28	<u>2.30</u>	2.34	.06	AESCHYNITE (Ce,Ca,Fe,Th)(Ti,Nb) ₂ (O,OH) ₆	80°	---	ORTH	Conch	Brownish- black	H 5.5 G 5.2 infus	Partly dec by hot acids. Commonly metamict.
2.40 v	---	<u>2.30</u>	---	low	PEROVSKITE (Ca,Ce,Na) ₂ (Ti,Nb) ₂ O ₆	~ 90°	$X = \frac{a}{b}$ $Z = \frac{b}{c}$	ORTH ps cub	100, 110	Brownish- black,	H 5.5-6 G 4.0-4.2	Insol in acids. Pleoc wk, X light green, Z darker green.

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _Z (2V _Z calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
204 2.21 ^	2.26	<u>2.32</u>	2.43	.17	TANTALITE (Fe,Mn)(Ta,Nb) ₂ O ₆	Large r < v str	X = $\frac{a}{b}$ Y = $\frac{b}{a}$	ORTH pris	010 dist 100 less so	Black	H 6 G 6.6-8.0 infus	Insol in acids. Nearly opaque. Pleoc str, X pale yellow, Y red-brown, Z dark red-brown.
	2.31	<u>2.35</u>	2.40	.09	ASHANITE (Nb,Ta,U,Fe,Mn) ₄ O ₈	---	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH	Conch	Brown to black	H 6-6.5	Pleoc, X light brown- ish-red, Z dark brownish-red.
	2.30 _{Li}	<u>2.35</u> _{Li}	2.40 _{Li}	.10	NADORITE PbSbO ₂ Cl	Large r > v str	X = $\frac{b}{a}$ Z = $\frac{a}{b}$ el pos	ORTH tab 010	010 perf	Brown to yellow	H 3.5-4 G 7.02 F 1.5	Diss by HNO ₃ . Tw on 101.
	2.39	(<u>2.36</u>)	2.52	.13	THOREAULITE SnTa ₂ O ₆	30-35° r > v str	Y = $\frac{b}{a}$ Z:c = 58°	MCL	100 good 011 less so	Brown	H 5.5-6 G 7.5-7.9	---
	2.28 _{Li}	<u>2.36</u> _{Li}	2.48 _{Li}	.20	BRACKEBUSCHITE Pb ₂ (Mn,Fe)(VO ₄) ₂ ·H ₂ O	Large r > v rather str	---	MCL acic	---	Dark brown to black	G 6.05 F 2 (?)	Pleoc str, X nearly cols, Y dark reddish- brown, Z clear reddish-brown.
	2.28 _{Na}	<u>2.38</u> _{Na}	2.49 _{Na}	.21								
	2.31 _{Li}	<u>2.36</u> _{Li}	2.46 _{Li}	.15	WOLFRAMITE (Fe,Mn)WO ₄	60-70° r < v	X = $\frac{b}{a}$ Z:c = 26° el clv pos	MCL tab 100	010 perf	Brownish- black	H 5-5.5 G 7.4 F 3	Dec by hot HCl with sepn of yellow WO ₃ . Nearly opaque. Pleoc str, abs X > Y > Z.
	2.31 _{Li}	<u>2.37</u> _{Li}	2.66 _{Li}	.35	CROCOITE PbCrO ₄	55°(Na) r > v very str, in- clined	Y = $\frac{b}{a}$ Z:c = 5°	MCL	110 dist	Hyacinth- red, orange	H 2.5-3 G 6.0-6.1 F 1.5	---
	2.34 _{Li}	<u>2.38</u> _{Li}	2.65 _{Li}	.31	PHOENICOCHROITE 8Pb ₂ (CrO ₄) ₀	58°(Na) r > v mod	X = $\frac{b}{a}$ Y:c = -2°	MCL tab	20I good	Coch- ineal-, hyacinth- red	H 2.5-3 G 7.01 F easy	Diss by HNO ₃ .
	2.38 _{Na}	<u>2.44</u> _{Na}	2.65 _{Na}	.27								
2.22 ^	2.38 _{Li}	<u>2.39</u> _{Li}	2.42 _{Li}	.04	PSEUDOBROOKITE Fe ₂ TiO ₅	~ 50° r < v	X = $\frac{b}{a}$ Z = $\frac{a}{b}$	ORTH tab 100	010 dist	Dark brown to black	H 6 G 4.3-4.4 infus	Insol in acids. Pleoc wk in reddish- brown, abs Z > Y > X.
	2.37	<u>2.40</u>	2.46	.09	STIBIOTANTALITE Sb(Ta,Nb)O ₄	75° r < v str	X = $\frac{a}{b}$ Z = $\frac{b}{a}$	ORTH	110 good	Dark brown, yellow- brown	H 5.5 G 6.2-7.9 F 4	Insol in acids. Data for mineral with G 6.8.

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2.39 _{Li}	<u>2.40_{Li}</u>	2.43 _{Li}	.04	BISMUTOTANTALITE Bi(Ta,Nb)O ₄	80° r < v str	X = $\frac{a}{b}$ Z = $\frac{b}{a}$	ORTH	110 good	Dark- brown, yellow- brown	H 5 G 8.1-8.8	---
---	<u>2.40</u>	---	low	PEROVSKITE CaTiO ₃	~ 90°	X = $\frac{a}{b}$ Z = $\frac{b}{a}$	ORTH ps cub	100, 110	Brownish- black to yellow	H 5-6 G 4.0-4.2 infus	Insol in acids. Pleoc wk in green or brown.
2.40	<u>2.42</u>	2.46	.06	STIBIOCOLUMBITE Sb(Nb,Ta)O ₄	73° r > v str	X = $\frac{a}{b}$ Z = $\frac{b}{a}$	ORTH	110 good	Dark brown, yellow- brown	H 5.5 G 5.6-6.6 F 4	Insol in acids. Data for mineral with G 6.3.
>2.42	---	>2.42	---	BISMITE Bi ₂ O ₃	Disp str	---	MCL mass	---	Gray- green, yellow	H 4.5 G 8.6	Opt char unk.
2.37 _{Li}	<u>2.45_{Li}</u>	2.65 _{Li}	.28	MONTROYDITE HgO	Large	Z = $\frac{a}{b}$ Y = $\frac{b}{a}$ (?)	ORTH pris	010 perf	Deep red, orange, brown	H 2-3 G 11.2 volat	Diss by acids.
2.45 _{Li}	<u>2.45_{Li}</u>	2.51 _{Li}	.06	DERBYLITE Fe ₄ Ti ₃ SbO ₁₃ (OH)	Small	---	MCL pris	Conch	Black	H 5 G 4.53 infus	Insol in acids. In thin-section dark brown, not pleoc.
2.58	<u>2.59</u>	2.70	.12	BROOKITE TiO ₂	15-30° (Na), ~ 0° for green, disp very str	Y = $\frac{a}{b}$ X = $\frac{b}{a}$ (red) X = $\frac{c}{b}$ (blue)	ORTH	120 indist	Brown to black	H 5.5-6 G 4.14 infus	Insol in acids. Pleoc wk, orange- brown to dark brown, abs Z > Y > X.
2.57 _{Li}	<u>2.61_{Li}</u>	2.71 _{Li}	.14	MASSICOT PbO	~ 90° (67+17°) disp str	Y = $\frac{a}{b}$ (?)	ORTH tab 100	010 perf	Yellow	Soft G 9.6 F 1.5	Diss by HNO ₃ . Pleoc, Y light yellow, Z deep yellow. Opt neg in blue light.
---	<u>2.63</u>	---	---	CLINOBISSVANITE BiVO ₄	Disp str	---	MCL	010 perf	Orange to yellow	Soft G 6.95 calc	Opt char unk.
---	<u>2.63</u> red	---	high	TENORITE CuO	Large r < v	Y = $\frac{b}{c}$ Z = $\frac{c}{b}$	MCL laths	111, 101	Steel- gray to black	H 3.5 G 6.45	Diss by HCl. Tw common on 011. Pleoc in brown. Opt char unk.
---	<u>3.17</u> blue	---	high								
>2.72 _{Li}	<u>>2.72_{Li}</u>	---	extr	LORANDITE TlAsS ₂	Large r > v str	Z = $\frac{b}{a}$ X ~ $\frac{b}{a}$ 100	MCL	100 perf 201, 001 good	Coch- ineal- red	H 2-2.5 G 5.53 F 1	Diss by HNO ₃ . Deep red in thin-section. Pleoc wk, Y purple- red, Z orange-red.
---	<u>>2.72_{Li}</u>	---	very high	GETCHELLITE AsSbS ₃	80° r > v str crossed	Z = $\frac{b}{a}$ Y:a = 15°	MCL	001 perf sectile	Dark red	H 1.5-2 G 3.92 F 1	---

Table 6. Biaxial positive minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_z$ ($2V_z$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	---	$>2.72_{Li}$	---	very high	VRBAITE $Tl_4Hg_3Sb_2As_8S_{20}$	Large $r > v$ str	---	ORTH	010 easy	Brownish-gray to reddish	H 3.5 G 5.30 F easy	Diss by HCl.
	---	$>2.72_{Li}$	---	very high	DUFRENOYSITE $Pb_2As_2S_5$	---	---	MCL	010 perf	Lead-gray	H 3 G 5.53 fus	Opt char unk.
	---	$>2.72_{Li}$	---	very high	MIARGYRITE $AgSbS_2$	Med	---	MCL	010 traces	Iron-black	H 2-2.5 G 5.2 F 1	Dec by HNO_3 . In thin-section blood-red.
	---	---	---	extr	PYROSTILPNITE Ag_3SbS_3	---	$Y = \frac{b}{c}$ $X:c = 8^\circ$	MCL	010 perf	Hyacinth-red	H 2 G 5.94 F 1	Tw pl 100.
	2.82_{Na}	2.93_{Na}	3.14_{Na}	.32	ENARGITE Cu_3AsS_4	$(76 \pm 8^\circ)$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH tab 001	110 perf 100, 010 dist	Gray-black, iron-black	H 3 G 4.45 fus	Tw pl 320.
	2.96_{Na}	3.09_{Na}	3.42_{Na}	.46	KERMESITE Sb_2S_2O	Med $(70 \pm 6^\circ)$	$Z = \frac{b}{c}$	TCL pris	001 perf 100 less so	Cherry-red	H 1-1.5 G 4.68 volat	---
	3.14	$3.17_{(8520\text{ nm})}$	3.28	.14	BOURNONITE $PbCuSbS_3$	52°	---	ORTH pris tab	010 imperf	Steel-gray	H 2.5-3 G 5.83 fus	Dec by HNO_3 . Tw on 110 common.

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Table 7. Biaxial negative minerals

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
208 1.435	1.324	1.3245	1.325	.001	AVOGADRITE (K,Cs)BF ₄	90°	X = $\frac{c}{b}$ Y = $\frac{c}{b}$	ORTH tab	---	Yellow to reddish	G 2.62	Slightly sol in H ₂ O, diss by acids.
	1.393	1.395	1.397	.004	MIRABILITE Na ₂ SO ₄ ·10H ₂ O	76° r < v str crossed	X = $\frac{b}{c}$ Z:c = 30°	MCL pris	100 perf 001, 010, 011	White	H 1.5-2 G 1.49 F 1.5	Sol in H ₂ O, effloresces.
	1.407	1.414	1.415	.008	THOMSENOLITE NaCaAlF ₆ ·H ₂ O	50° r < v wk	Z = $\frac{b}{c}$ X:c = -52°	MCL cube-like	001 perf 110 dist	Cols	H 2 G 2.48 F 1.5	Diss by H ₂ SO ₄ .
	1.405	1.425	1.440	.035	NATRON Na ₂ CO ₃ ·10H ₂ O	71° (81+7°) r < v perc crossed	X = $\frac{b}{c}$ Z:c = 41°	MCL	001 good 010 imperf	White	H 1-1.5 G 1.48 F 1	Sol in H ₂ O, effloresces.
	1.429	1.433	1.436	.007	JARLITE NaSr ₃ Al ₃ (F,OH) ₁₆ (?)	81-90°	Y = $\frac{b}{c}$ X:c = -6°	MCL tab	---	Cols	H 4-4.5 G 3.78 - 3.93 F easy	---
	1.424	1.436	1.438	.012	WILCOXITE MgAl(SO ₄) ₂ ·18H ₂ O	48°	---	TCL	---	Cols, white	H 2 G 1.58	Sol in H ₂ O.
	1.430	1.452	1.458	.028	KALINITE KAl(SO ₄) ₂ ·11H ₂ O (?)	52° disp wk	Z = $\frac{b}{c}$ Y:c = 13°	MCL fib c	---	White	H 2-2.5 G 1.75 F 1	Sol in H ₂ O.
	1.440	1.453	1.454	.014	LECONTITE (K,NH ₄)Na(SO ₄)·2H ₂ O	30° r < v rather str	Y = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH pris	011	Cols	H 2-2.5 G 1.74 F 1	Sol in H ₂ O.
	1.448	1.454	1.456	.008	GEARKSUTITE CaAl(F,OH) ₅ ·H ₂ O	Small to med	X = $\frac{b}{c}$ Y:c Target el pos	MCL acic c	---	White, chalky	H 2 G 2.74 F 2	Diss by acids.
	1.433	1.455	1.461	.028	EPSOMITE MgSO ₄ ·7H ₂ O	52° r < v wk	X = $\frac{b}{c}$ Z = $\frac{a}{c}$	ORTH el c	010 good 101 dist	White	H 2 G 1.68 infus	Sol in H ₂ O, taste bitter.
	1.435	1.455	1.459	.024	WATTEVILLITE Na ₂ Ca(SO ₄) ₂ ·4H ₂ O	48°	---	MCL (?) acic	---	Cols	G 1.81 F diff	Dec by H ₂ O with sepn of gypsum, diss by acids.
1.448 1.461	1.340	1.456	1.459	.119	SASSOLITE B(OH) ₃	5-14°	X:c = 84-88° b ~ opt pl	TCL tab 001	001 perf	White, pearly	H 1 G 1.48 F 1	Sol in H ₂ O. Tw pl 001.

	1.452	<u>1.456</u>	1.458	.006	TIKHONENKOVITE $\text{SrAlF}_4(\text{OH})\cdot\text{H}_2\text{O}$	70°	---	MCL pris	001 perf fr conch	Cols to rose	H 3.5 G 3.26	---
	(1.405)	<u>1.460</u>	1.487	(.082)	ROSTITE $\text{AlSO}_4(\text{OH})\cdot 5\text{H}_2\text{O}$	68°	$X = \frac{c}{b}$ $Y = \frac{a}{b}$	ORTH	---	Cols	G 1.93	Diss by acids. Reported α given as "1.44 calc."
1.455 □ 1.466	1.449	<u>1.461</u>	1.463	.014	MENDOZITE $\text{NaAl}(\text{SO}_4)_2\cdot 11\text{H}_2\text{O}$	56°	$X = \frac{b}{c}$ $Y:\underline{c} = 30^\circ$	MCL laths <u>c</u>	100 good	White	H 3 G 1.77 F 1	Sol in H_2O . Data on synth compd.
┐ 1.466	---	<u>1.461</u>	---	low	SILHYDRITE $3\text{SiO}_2\cdot\text{H}_2\text{O}$	---	---	ORTH	fr uneven to subconch	White	G 2.14	Dec by HCl. Opt sign unk.
	1.438	<u>1.463</u>	1.465	.027	HEXAHYDRITE $\text{MgSO}_4\cdot 6\text{H}_2\text{O}$	29° $r > v$	$Y = \frac{b}{c}$ $X:\underline{c} = -25^\circ$	MCL	---	Cols	G 1.8 infus	Sol in H_2O . Data on synth compd.
	---	<u>1.465</u>	---	---	ZINCFAUSERITE $(\text{Mn},\text{Mg},\text{Zn})\text{SO}_4\cdot 7\text{H}_2\text{O} (?)$	Large	---	ORTH	010 good	Pale rose	H 2.5 G 2.00 infus	Sol in H_2O , efflo- resces.
1.455 ∧ 1.480	1.447	(<u>1.468</u>)	1.470	.023	GOSLARITE, magnesian (Epsomite grp) $(\text{Zn},\text{Mg})\text{SO}_4\cdot 7\text{H}_2\text{O}$	Small	$X = \frac{a}{b}$ $Z = \frac{b}{c}$ el cTv neg	ORTH	010 perf	White	H 2 G 1.8 infus	Sol in H_2O .
	1.447	<u>1.469</u>	1.472	.025	BORAX $\text{Na}_2\text{B}_4\text{O}_7\cdot 10\text{H}_2\text{O}$	40° $r > v$ str crossed	$X = \frac{b}{c}$ $Z:\underline{c} = -55^\circ$	MCL pris c tab 100	100 perf 110 good	White	H 2-2.5 G 1.71 F 1-1.5	Sol in H_2O . Tw pl 100 rare.
	1.451	<u>1.470</u>	1.478	.027	KANEMITE $\text{NaHSi}_2\text{O}_4(\text{OH})_2\cdot 2\text{H}_2\text{O}$	46° $r > v$	$X = \frac{b}{c}$ $Y = \frac{a}{c}$	ORTH	010 perf 100 good	Cols, white	H 4 G 1.93	Dec by acids.
1.455 ∧ 1.493	1.447	<u>1.470</u>	1.473	.026	EPSOMITE, nickeloan $(\text{Mg},\text{Ni})\text{SO}_4\cdot 7\text{H}_2\text{O}$	47°	$X = \frac{b}{c}$ $Z = \frac{a}{c}$ clv pos	ORTH	010 fair	Pale blue- green	H 2 G 1.8 infus	Sol in H_2O . NiO 12.2%.
	1.463	<u>1.470</u>	1.471	.008	PARALUMINITE $\text{Al}_4(\text{SO}_4)_5\cdot 15\text{H}_2\text{O}$	Small	$X = \text{el}$	Fib, mass	---	White, chalky	Soft	---
	1.460	<u>1.471</u>	1.474	.014	STARKEYITE $\text{MgSO}_4\cdot 4\text{H}_2\text{O}$	(55±18°)	---	MCL	---	White	G ~ 1.8 infus	Sol in H_2O . (Synth compd is α biax pos, β 1.491.)
	1.455	<u>1.472</u>	1.487	.032	KERNITE $\text{Na}_2\text{B}_4\text{O}_7\cdot 4\text{H}_2\text{O}$	80° $r > v$ dist	$Z = \frac{b}{c}$ $X:\underline{a} = 38^\circ$	MCL	001, 100 perf 201 fair	White, cols	H 2.5 G 1.90 F 1-1.5	Sol in H_2O .
	1.459	<u>1.473</u>	1.483	.024	JURBANITE $\text{AlSO}_4(\text{OH})\cdot 5\text{H}_2\text{O}$	80°	$Y = \frac{b}{c}$ $Z:\underline{a} = -5^\circ$	MCL	---	Cols	H 2.5 G 1.79	---
	1.465	<u>1.473</u>	1.477	.012	SASAIT $(\text{Al},\text{Fe})_{14}(\text{PO}_4)_{11}(\text{OH})_7$ $(\text{SO}_4)\cdot 83\text{H}_2\text{O} (?)$	---	$X = \frac{c}{b}$ $Y = \frac{a}{b}$ el cTv pos	ORTH (?)	001 perf	White, chalky	G 1.75 infus	Diss by acids. Fe_2O_3 1.1%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.465 □ 1.484	---	<u>1.474</u>	---	.002-.008	GMELINITE (Zeolite grp) $(Na,Ca)Al_2Si_4O_{12} \cdot 6H_2O$	Small	---	HEX	10T0 dist	White	H 4.5 G 2.0-2.1 F 3	Dec by HCl. Tw axis <u>c</u> .
1.580	---	<u>1.474</u>	---	0-.008	HISINGERITE $Fe^{+3}_2Si_2O_5(OH)_4 \cdot 2H_2O$	Small	---	MCL u mass	Conch	Dark brown	H 3.5 G ~ 2.5 infus	Dec by acids.
	---	<u>1.47-1.48</u>	---	wk	HYDROBASALUMINITE $Al_4SO_4(OH)_{10} \cdot 12-36H_2O$	---	---	---	---	Cols to yellowish	G 1.86	Diss by acids. Opt char unk.
1.482	1.471	<u>1.475</u>	1.476	.005	SODIUM DACHIARDITE (Zeolite grp) $(Na,Ca,K)_{4-5}Al_8Si_{40}O_{96} \cdot 26H_2O$	88-92°	$Z:c = 18^\circ$	MCL	---	Cols	G 2.17	---
1.480	1.472	<u>1.475</u>	1.477	.005	MORDENITE (Zeolite grp) $(Ca,Na,K)Al_2Si_{10}O_{24} \cdot 7H_2O$	80°	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	ORTH	100 perf 010 good	Cols	H 5 G 2.12 F 3-4	Some samples appear to be monoclinic.
	1.461	<u>1.478</u>	1.484	.023	CREEDITE $Ca_3Al_2(SO_4)(F,OH)_{10} \cdot 2H_2O$	64° $r > v$ wk	$Y = \frac{b}{c}$ $Z:c = 41^\circ$	MCL pris <u>c</u>	100 perf	Cols	H 4 G 2.72 F diff	Diss by acids.
	1.358	<u>1.479</u>	1.530	.172	BARENTSITE $Na_7Al(CO_3)_2(HCO_3)_2F_4$	62° $r < v$ wk	$X = \frac{c}{c}$	TCL ps hex	001, 100 perf	Cols	H ~ 3 G 2.56	Eff in acids, slowly dec by H_2O .
	1.476	<u>1.479</u>	1.481	.005	TVEITITE $Ca_{1-x}(Y,Ce)_x F_{2+x}$ ($x \sim 0.3$)	34°	---	MCL ps cub	---	White to pale yellow	---	Poly tw. Fluor faint yellow-orange in short-wave UV.
1.484	1.476	<u>1.479</u>	1.479	.003	CLINOPTILOLITE (Zeolite grp) $(Na,K,Ca)_{2-3}Al_3(Al,Si)_2Si_{13}O_{36} \cdot 12H_2O$	Small $r < v$ str	$X = \frac{b}{b} (?)$ $Z:a = 15^\circ$ el pos (?)	MCL tab 010	010 perf	Cols	H 3.5-4 G 2.2 F 2	Dec by HCl.
(1.468) ^	1.457	<u>1.480</u>	1.484	.027	GOSLARITE (Epsomite grp) $ZnSO_4 \cdot 7H_2O$	46° $r > v$ wk	$X = \frac{a}{b}$ $Z = \frac{b}{b}$ el cTv pos	ORTH acic <u>c</u>	010 perf	Cols, yellowish	H 2-2.5 G 1.98 infus	Sol in H_2O .
	1.47	<u>1.48</u>	1.49	.02	BOOTHITE $CuSO_4 \cdot 7H_2O$	Large	$Y = \frac{b}{c}$ $X \sim \frac{c}{c}$	MCL fib <u>c</u>	001 imperf	White, silky	H 2-2.5 G 2.1 fus	Sol in H_2O . Dehydrates to Chalcánthite.

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1.472	---	1.487	.015	MAKATITE $\text{Na}_2\text{Si}_4\text{O}_9 \cdot 5\text{H}_2\text{O}$	---	---	ORTH(?) spher- ulitic	---	White	G 2.07	Dec by HCl. Opt char unk.
1.391	<u>1.481</u>	1.486	.095	DARAPSKITE $\text{Na}_3(\text{SO}_4)(\text{NO}_3) \cdot \text{H}_2\text{O}$	27° $r > v$ rather str	$X = \frac{b}{c}$ $Z:c = 12^\circ$ el clv pos	MCL	010 perf 100	Cols	H 2-3 G 2.20 F 1-2	Sol in H_2O . Poly tw 100.
1.380	<u>1.482</u>	1.573	.193	KALICINITE KHCO_3	81° (83±1°)	$Y = \frac{b}{c}$ $X:c = 30^\circ$	MCL el <u>b</u>	100, 001, 101	White	G 2.16 F easy	Sol in H_2O .
1.462	<u>1.482</u>	1.495	.033	AUBERTITE $\text{CuAl}(\text{SO}_4)_2\text{Cl} \cdot 14\text{H}_2\text{O}$	71° $r > v$ mod	---	TCL	010 perf	Azure- blue	G 1.82	Sol in H_2O .
1.462	<u>1.482</u>	1.490	.028	MORAESITE $\text{Be}_2\text{PO}_4(\text{OH}) \cdot 4\text{H}_2\text{O}$	65°	$Z = \frac{b}{c}$ $Y:c = 11-$ 23°	MCL acic	010, 001 good	White	G 1.81	Diss by acids.
1.470	---	1.493	.023	NICKEL-HEXAHYDRITE $(\text{Ni}, \text{Mg}, \text{Fe})\text{SO}_4 \cdot 6\text{H}_2\text{O}$	---	$Z:c = 45^\circ$	MCL crusts	010 perf 100	Blue- green	---	Sol in H_2O . NiO 22.6, MgO 2.8, FeO 2.6%. Opt char unk.
1.479 1.488	<u>1.482</u>	1.484	.007	PICKERINGITE (Halotrichite grp) $\text{MgAl}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Med	$Y = \frac{b}{c}$ $Z:c = 30^\circ$	MCL fib	010 poor fr conch	Cols to yellow	H 2-2.5 G 1.8 infus	Sol in H_2O , taste astringent. In closed tube melts in H_2O of crystn.
1.469	<u>1.482</u>	1.490	.021	REVDITE $\text{Na}_2\text{Si}_2\text{O}_5 \cdot 5\text{H}_2\text{O}$	75°	$Y:c = 0-$ 10°	TCL	100, 010 perf	Cols	H 2 G 1.94	Sol in H_2O , gives an alk soln.
1.478	<u>1.482</u>	1.482	.004	APJOHNITE (Halotrichite grp) $(\text{Mn}, \text{Mg})\text{Al}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	Small	$Y = \frac{b}{c}$ $Z:c = 29^\circ$	MCL fib <u>c</u>	---	White, silky	H 1.5 G 1.78 infus	Sol in H_2O . In closed tube melts in H_2O of crystn.
1.477	<u>1.483</u>	1.489	.012	BIEBERITE $\text{CoSO}_4 \cdot 7\text{H}_2\text{O}$	88° disp wk	$Y = \frac{b}{c}$ $Z:c \sim 30^\circ$	MCL	001 perf 110 fair	Rose- to carmine- red	H 2 G 1.95 infus	Sol in H_2O . Loses H_2O on exposure.
1.479 ^	<u>1.484</u>	1.486	.005	CLINOPTILOLITE (Zeolite grp) $(\text{Na}, \text{K}, \text{Ca})_2\text{Al}_3$ $(\text{Al}, \text{Si})_2\text{Si}_{13}\text{O}_{36} \cdot 12\text{H}_2\text{O}$	Large $r < v$ str	$X = \frac{b}{a} (?)$ $Z:a = 15^\circ$ el pos (?)	MCL	010 perf	Cols	H 3.5-4 G 2.2 F 2	Dec by HCl.
1.442	<u>1.485</u>	1.490	.048	HUNGCHAOITE $\text{MgB}_4\text{O}_5(\text{OH})_4 \cdot 7\text{H}_2\text{O}$	36°	---	TCL ps hex	---	White	---	Diss by acids.
1.479	<u>1.485</u>	1.489	.010	BARRERITE (Zeolite grp) $(\text{Na}, \text{K}, \text{Ca})_2\text{Al}_2\text{Si}_{18}\text{O}_{60} \cdot 6-7\text{H}_2\text{O}$	---	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	---	Cols	G 2.13 fus	Sodium analogue of Stellerite.
1.477	<u>1.485</u>	1.486	.009	PHOSPHORROESSLERITE $\text{MgHPO}_4 \cdot 7\text{H}_2\text{O}$	38° $r > v$	$X = \frac{b}{c}$ $Z:c = 7^\circ$	MCL	Conch	Cols to yellow	H 2.5 G 1.73 infus	Diss by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orien- tation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.507	---	1.485	---	low	MAGADIITE NaSi ₇ O ₁₃ (OH) ₃ ·3H ₂ O	---	el pos	MCL	---	White	---	Dec by acids. Opt sign unk.
	1.483	1.486	1.487	.004	BLOEDITE Na ₂ Mg(SO ₄) ₂ ·4H ₂ O	71° r < v str	Y = $\frac{b}{c}$ X:c = 37°	MCL tab 001	Conch	Cols	H 2.5-3 G 2.25 F 1.5	Sol in H ₂ O.
1.509	1.484	1.486	1.487	.003	PHILLIPSITE (Zeolite grp) (K,Ca,Na) ₁₋₂ (Al,Si) ₈ O ₁₆ ·6H ₂ O	Med r > v	---	MCL	010, 100 good	Cols	H 4-4.5 G 2.2 F 3	Gel with acids. Penet tw.
1.480 □ 1.493	---	1.487	---	.002	ANALCIME (Zeolite grp) NaAlSi ₂ O ₆ ·H ₂ O	Very small	---	CUB trape- zohe- dral	100 imperf	Cols	H 4-4.5 G 2.25 F 3.5	Dec by HCl. Tw grating.
	1.482 ^	1.488	1.490	.010	HALOTRICHITE FeAl ₂ (SO ₄) ₄ ·22H ₂ O	Med	Y = $\frac{b}{c}$ Z:c = 38°	MCL fib c	010 poor fr conch	White, silky	H 2 G 1.89 F 4.5-5	Sol in H ₂ O. Taste astringent.
1.493	1.486	1.488	1.489	.003	VANTHOFFITE Na ₆ Mg(SO ₄) ₄	83° r < v wk	---	MCL	---	Cols	H 3.5 G 2.69 F 3	Sol in H ₂ O.
	1.484	(1.488)	1.490	.006	CHABAZITE (Zeolite grp) (Ca,Na)Al ₂ Si ₄ O ₁₂ ·6H ₂ O	60-70°	---	TRIG	10T1 poor	White, reddish	H 4 G 2.05- 2.10 F 3	Dec by HCl. Tw 10T1, 0001.
1.502	1.448	1.489	1.493	.045	BURKEITE Na ₆ (CO ₃)(SO ₄) ₂	34° r > v dist	X = $\frac{c}{b}$ Z = $\frac{b}{c}$	ORTH	Conch	White	H 3.5 G 2.57 F 1.5	Sol in H ₂ O.
	1.473	(1.489)	1.490	.017	Unnamed Na-Mg-U sulfate-borate	17-27°	ext 28°	TCL	---	Pale yellow	---	Tw. Fluor bright yellow-green in UV. Am. Mineral., 62, 1261 (1977).
1.502	1.482	1.489	1.496	.014	STILBITE (Zeolite grp) NaCa ₂ Al ₅ Si ₁₃ O ₃₆ ·14H ₂ O	43° (90+17°)	Y = $\frac{b}{a}$ X:a = 2-9°	MCL acic a	010 perf	White, pink, yellow	H 4 G 2.15 F 3	Dec by HCl. Tw 001 cruciform, penet. Var with high Na.

	1.485	<u>1.490</u>	1.494	.009	AMICITE (Zeolite grp) $K_2Na_2Al_4Si_4O_{16} \cdot 5H_2O$	82°	---	MCL radi- ating	---	Cols	H soft G 2.06- 2.23	---
	1.454	<u>1.491</u>	1.500	.046	BAYLEYITE $Mg_2(UO_2)_2(CO_3)_3 \cdot 18H_2O$	30° (51+6°)	$Z = b$ $X:c = 8-15^\circ$	MCL	---	Sulfur- yellow	G 2.05	Sol in H_2O . Pleoc wk, X pinkish, Y and Z pale greenish-yellow. Fluor yellow-green in UV.
v 1.512	1.482	<u>1.492</u>	1.493	.011	PENTAHYDRITE (Chalcanthite grp) $MgSO_4 \cdot 5H_2O$	45° r < v	$X \sim b$	TCL	---	Cols	G 1.77 infus	Sol in H_2O . Data for synth compd.
	1.472	---	1.502	.030	INDIGIRITE $MgAl(CO_3)_2(OH) \cdot 7.5H_2O$	---	el pos	fib	---	Snow- white	H 2 G 1.6	Diss by acids.
1.470 ^	1.470	<u>1.493</u>	1.500	.030	MORENOSITE (Epsomite grp) $NiSO_4 \cdot 7H_2O$	42° r > v wk	$X = b$ $Z = a$	ORTH acic <u>c</u>	010 dist	Apple- green	H 2-2.5 G 1.95 infus	Sol in H_2O .
(1.488) ^	1.487	<u>1.493</u>	1.495	.008	CHABAZITE (Zeolite grp) $(Ca,Na)Al_2Si_4O_{12} \cdot 6H_2O$	60°	---	TRIG	10 $\bar{1}$ 1 poor	White, reddish	H 4 G 2.08 F 3	Dec by HCl. Tw 10 $\bar{1}$ 1, 0001.
	1.416	<u>1.494</u>	1.542	.126	TRONA $Na_3(CO_3)(HCO_3) \cdot 2H_2O$	75° r < v str	$X = b$ $Z:c = 83^\circ$	MCL el <u>b</u>	100 perf 10 $\bar{1}$	White	H 2.5-3 G 2.15 F 3	Sol in H_2O , gives an alk soln.
	1.465	<u>1.494</u>	1.495	.030	BIANCHITE $(Zn,Fe,Mg)SO_4 \cdot 6H_2O$	10-20° r > v wk	$Y = b$ $X:c = -26^\circ$	MCL	---	White to pale green	H 2.5 G 2.03 infus	Sol in H_2O . Tw 001. ZnO 20.0, FeO 8.8%.
	1.476	(<u>1.494</u>)	1.496	.020	Unnamed Na-Mg-U sulfate-borate	~ 30°	---	---	---	Yellow	---	Fluor bright yellow- green in UV. Pleoc, X cols, Z yellowish. Am. Mineral., 62, 1261 (1977).
	1.486	<u>1.494</u>	1.496	.010	STELLERITE (Zeolite grp) $CaAl_2Si_7O_{18} \cdot 7H_2O$	38-48°	$X = a$ $Y = c$ el cTv neg	ORTH	010 perf	Cols to flesh-red	H 3.5-4 G 2.14 F easy	Dec by HCl. Not tw.
	1.468	---	1.498	.030	FERROHEXAHYDRITE $FeSO_4 \cdot 6H_2O$	---	---	MCL fib	---	Cols	---	Sol in H_2O . Opt sign unk, put here by analogy to related minerals.
	1.471	<u>1.496</u>	1.497	.026	MOORHOUSEITE $(Co,Ni,Mn)SO_4 \cdot 6H_2O$	10-30°	$Y = b$	MCL mass	Conch	Pink	H 2.5 G 1.97 infus	Sol in H_2O . Pleoc, abs X > Z. Data on synth $CoSO_4 \cdot 6H_2O$.
	1.465	<u>1.498</u>	1.504	.039	NITROCALCITE $Ca(NO_3)_2 \cdot 4H_2O$	50° disp wk	$X \perp$ clv el clv pos	MCL pris	One perf	Cols	Soft G 1.90 F easy	Sol in H_2O , hygro- scopic. Data on synth compd.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.495	1.498	1.502	.007	GOOSECREEKITE (Zeolite grp) $\text{CaAl}_2\text{Si}_6\text{O}_{16} \cdot 5\text{H}_2\text{O}$	82°	$Y = b$ $Z:c = 46^\circ$	MCL	010 perf	Cols	H 4.5 G 2.21	Insol in HCl.
	1.490	1.498	1.501	.011	HYDROGLAUBERITE $\text{Na}_4\text{Ca}(\text{SO}_4)_3 \cdot 2\text{H}_2\text{O}$	60°	$Z:c = 3.5^\circ$ el pos	MCL acic	010	White	G 2.40	Dec by H_2O , diss by HCl.
	1.493	1.499	1.503	.010	YUGAWARALITE (Zeolite grp) $\text{CaAl}_2\text{Si}_6\text{O}_{16} \cdot 4\text{H}_2\text{O}$	65-71° $r < v$	$Z = b$ $X:c = -9^\circ$	MCL	Parting 010	Cols	H 4.5 G 2.23	Insol in acids.
	1.499	1.500	1.501	.002	MERLINOITE (Zeolite grp) $(\text{K}, \text{Ca}, \text{Na})_7\text{Si}_{23}\text{Al}_9\text{O}_{64} \cdot 23\text{H}_2\text{O}$	56° $r > v$	$X = b$ $Z = a$ el pos	ORTH	---	White	G 2.14-2.27	---
	1.442	(1.500)	1.504	.062	ADMONTITE $\text{Mg}_2\text{B}_{12}\text{O}_{20} \cdot 15\text{H}_2\text{O}$	30° $r < v$	$X:c = 45^\circ$	MCL tab 100	Conch	Cols	H 2-3 G 1.82	---
	---	1.500	---	wk	BILINITE (Halotrichite grp) $\text{Fe}^{+2}\text{Fe}^{+3}_2(\text{SO}_4)_4 \cdot 22\text{H}_2\text{O}$	---	ext 35-39°	MCL radiating, fib	---	White to yellowish	H 2 G 1.88	Diss in H_2O .
	1.442	(1.500)	1.504	.062	Unnamed magnesium borate	~ 30° $r < v$	---	ORTH	---	Cols	---	Am. Mineral., 62, 1261 (1977).
	1.377	1.501	1.583	.206	NAHCOLITE NaHCO_3	75° (72+1°) $r > v$ wk	$Y = b$ $X:c = 27^\circ$	MCL pris	101 perf 111 good	White	H 2.5 G 2.21 F 1	Sol in H_2O , gives an alk soln.
	1.490	---	1.502	.012	BEARSITE $\text{Be}_2\text{AsO}_4(\text{OH}) \cdot 4\text{H}_2\text{O}$	---	$Z:c = 9^\circ$	MCL fib	---	White	G 2.2	---
	1.498	(1.501)	1.502	.004	WAIRAKITE (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 2\text{H}_2\text{O}$	75° $r > v$	$X \sim b$ $Z \sim c$	MCL	---	White	H 5.5-6 G 2.26 F 2.5	Gel with acids. U biax pos.
1.489 ^	1.494	1.502	1.507	.013	STILBITE (Zeolite grp) $\text{NaCa}_2\text{Al}_5\text{Si}_{13}\text{O}_{36} \cdot 14\text{H}_2\text{O}$	30-50° (76+18°)	$Y = b$ $X:a = 2-9^\circ$	MCL acic a	010 perf	White, pink, yellow	H 4 G 2.15 F 3	Dec by HCl. Tw 001, cruciform, penet.
	1.414	1.503	1.527	.113	NESQUEHONITE $\text{Mg}(\text{HCO}_3)(\text{OH}) \cdot 2\text{H}_2\text{O}$	53° $r < v$ wk	$X = a$ $Z = b$	ORTH el c	110 perf 001 poor	Cols	H 2.5 G 1.85 infus	Diss by cold acids with eff.

1.508 v	1.455	<u>1.503</u>	1.549	.094	HELLYERITE $\text{NiCO}_3 \cdot 6\text{H}_2\text{O}$	85°	---	---	One perf 2 good	Light blue	H 2.5 G 1.97 infus	Diss by acids with eff. Faintly pleoc.
	1.472	<u>1.503</u>	1.526	.054	AKSAITE $\text{MgB}_6\text{O}_{10} \cdot 5\text{H}_2\text{O}$	80°	$X = \frac{c}{b}$ $Z = \frac{b}{a}$	ORTH	100, 101	Cols, gray	H 2.5 G 2.0	Diss by acids.
	1.498	<u>1.503</u>	1.506	.008	HARMOTOME (Zeolite grp) $(\text{Ba}, \text{K})_{1-2}(\text{Si}, \text{Al})_8\text{O}_{16} \cdot 6\text{H}_2\text{O}$	75-82°	$Z = \frac{b}{a}$ $Y:c = \frac{b}{a} 28^\circ$	MCL	010 good 001 fair	Cols	H 4.5 G 2.44 F 3.5	Dec by HCl. Tw cruciform, penet on 001.
	1.426	(<u>1.504</u>)	1.508	.082	Unnamed carbonate of K, Ca	~ 25°	$X = e1$	ORTH	---	Pale rose	---	Am. Mineral., 58, 139 (1973).
	1.332	<u>1.504</u>	1.504	.172	NITER KNO_3	5-7° $r < v$ wk	$X = \frac{c}{b}$ $Z = \frac{b}{a}$	ORTH	011 perf 010 good 110 poor	Cols	H 2 G 2.11 F 1	Sol in H_2O .
	1.492	<u>1.505</u>	1.517	.025	INYOITE $\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 13\text{H}_2\text{O}$	80° $r < v$ wk	$Y = \frac{b}{a}$ $X:c = \frac{b}{a} 36^\circ$	MCL tab 100	001 good 010	Cols	H 2.5-3 G 2.13 F 2	Sol in hot H_2O .
	1.494	<u>1.505</u>	1.516	.022	KAINITE $\text{MgSO}_4 \cdot \text{KCl} \cdot 3\text{H}_2\text{O}$	85° $r > v$ wk	$Y = \frac{b}{a}$ $Z:c = \frac{b}{a} 13^\circ$	MCL tab 001	001 perf 110 dist	Cols	H 2.5-3 G 2.13 F 2	Sol in H_2O .
	1.344	<u>1.506</u>	1.506	.162	NITROMAGNESITE $\text{Mg}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$	5° $r < v$ perc	---	MCL pris	110 perf	Cols	G 1.58	Sol in H_2O , taste bitter.
	1.420	<u>1.506</u>	1.524	.104	THERMONATRITE $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$	48° $r < v$ wk	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH flat- tened 001 or 010	100 diff	White	H 1-1.5 G 2.26 F 1.5	Sol in H_2O , gives an alk soln.
	1.470	<u>1.506</u>	1.527	.057	EZCURRITE $\text{Na}_4\text{B}_{10}\text{O}_{17} \cdot 7\text{H}_2\text{O}$	73° $r > v$	---	TCL	110, 010 good 100, T26 fair	Cols	H 3-3.5 G 2.1 F easy	Sol in H_2O .
1.525 v	1.498	<u>1.506</u>	1.508	.010	SEPIOLITE $\text{Mg}_4\text{Si}_6\text{O}_{15}(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	50°	$Z = e1$	ORTH fib	---	White to yellow	H 2 G 2.3 infus	Dec by HCl with sepn of silica.
	1.486 v 1.518	<u>1.507</u>	1.509	.005	NICKELBLOEDITE $\text{Na}_2(\text{Ni}, \text{Mg})(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$	60-70°	$Y = \frac{b}{a}$	MCL	---	Light green	G 2.3	Sol in H_2O . Ni 8.8, Mg 3.0, Cu 0.6, Co 0.3%.
	1.503 v	<u>1.508</u>	1.528	.055	AKSAITE $\text{MgB}_6\text{O}_{10} \cdot 5\text{H}_2\text{O}$	(73+4°)	$X = \frac{c}{b}$ $Z = \frac{b}{a}$	ORTH	100, 101	Cols, gray	H 2.5 G 2.0	Diss by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.486 \triangle	1.505	<u>1.509</u>	1.511	.006	PHILLIPSITE (Zeolite grp) (K,Ca,Na) ₁₋₂ (Al,Si) ₈ O ₁₆ ·6H ₂ O	Med r > v	---	MCL	010, 100 good	Cols	H 4-4.5 G 2.2 F 3	Gel with acids. Penet tw.
1.515 \checkmark	1.505	<u>1.509</u>	1.509	.004	COWLESITE (Zeolite grp) CaAl ₂ Si ₃ O ₁₀ ·5-6H ₂ O	30°	X = a Y = $\frac{b}{c}$ el pos	ORTH bladed	010 perf	Cols, white	H 5-5.5 G 2.13	---
	1.470	<u>1.510</u>	1.579	.109	STRONTIOBORITE SrB ₈ O ₁₁ (OH) ₄	80-100° (102±2°)	---	MCL plates	---	Cols	G 2.81	Diss by acids.
	1.465	<u>1.51</u>	1.540	.075	SWARTZITE CaMg(UO ₂)(CO ₃) ₃ · 12H ₂ O	(40°)	---	MCL pris	---	Green	G 2.3	Diss by acids. Pleoc, X cols, Y and Z green. Fluor bright green in UV.
	1.490	<u>1.510</u>	1.524	.034	KURNAKOVITE MgB ₃ O ₃ (OH) ₅ ·5H ₂ O	80° r > v	Y ~ b Z:c = -22°	TCL	010 perf 110 good	Cols	H 3 G 1.86 F 2	Diss by acids.
1.531 \checkmark	1.479	<u>1.510</u>	1.511	.032	SAPONITE (Smectite grp) (Na,Ca) _{0.33} Mg ₃ (Al,Si) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	20-30°	X \perp 001 el pos	MCL scales	001 perf	White, waxy	H 1-2 G 2.2 F diff	Dec by acids.
1.538 \checkmark	1.492	<u>1.510</u>	1.522	.030	URANOSPATHITE HAl(UO ₂) ₄ (PO ₄) ₄ · 40H ₂ O	69-76° r > v	X = c Y = $\frac{b}{c}$	ORTH plates	001 perf 100, 010 good	Yellow to pale green	H 2-2.5 G 2.50	Diss by acids. Pleoc, X pale yellow, Y and Z yellow. Fluor wk yellow-green in UV.
	1.498	<u>1.510</u>	1.517	.019	JOKOKUITE MnSO ₄ ·5H ₂ O	70-80°	---	TCL	No clv	Pale pink	H 2.5 G 2.03	Sol in H ₂ O.
	1.496	---	1.512	.016	Unnamed Na-Mg-U- borate-sulfate A	Small	---	---	---	Yellow	---	Dec by H ₂ O. Pleoc, X cols, Z yellow. Fluor bright yellow-green in UV. Am. Mineral., 62, 1261 (1977).
	1.500	<u>1.511</u>	1.513	.013	EPISTILBITE (Zeolite grp) CaAl ₂ Si ₆ O ₁₆ ·5H ₂ O	40° r > v str	Y = b Z:c = 9°	MCL el c	010 perf	Cols	H 4 G 2.25 F 3	Dec by acids.

	1.508	<u>1.511</u>	1.512	.004	GARRONITE (Zeolite grp) $\text{Na}_2\text{Ca}_5\text{Al}_{12}\text{Si}_{20}\text{O}_{64} \cdot 27\text{H}_2\text{O}$	0-30°	---	ORTH ps tet	2 poor at 90°	Cols	H 4.5 G 2.15	---
1.521	1.483	<u>1.512</u>	1.530	.047	INDERBORITE $\text{CaMg}[\text{B}_3\text{O}_3(\text{OH})_5]_2 \cdot 6\text{H}_2\text{O}$	77°	$Z = \frac{b}{c}$ $X:c = 2^\circ$	MCL	100 good fr conch	Cols	H 3.5 G 2.00 fus	Diss by acids.
	1.494	<u>1.512</u>	1.524	.030	NASINITE $\text{Na}_2\text{B}_5\text{O}_8(\text{OH}) \cdot 2\text{H}_2\text{O}$	67° (78+7°)	$Y = \frac{b}{a}$ $Z:a = 7^\circ$	MCL	---	Cols	F 1.5	Sol in H_2O .
1.492 ^ (1.526)	1.495	<u>1.512</u>	1.518	.023	PENTAHYDRITE (Chalcanthite grp) $(\text{Mg,Cu})\text{SO}_4 \cdot 5\text{H}_2\text{O}$	55° $r < v$	$X \sim \frac{b}{c}$	TCL	---	Blue	G 2.01 infus	Sol in H_2O . CuO 9.0, ZnO 5.6, ^2FeO 1.4%.
1.534	1.494	<u>1.512</u>	1.512	.018	MONTMORILLONITE (Smectite grp) $(\text{Na,Ca})_{0.33}(\text{Al,Mg,Fe})_2$ $\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	Small	$Y = \frac{b}{c}$ $X \sim \frac{c}{a}$	MCL u mass	001 perf	Cols, gray, green	H 1-2 G 2.3	Dec by acids.
	1.497	<u>1.512</u>	1.513	.016	META-ALUMINITE $\text{Al}_2\text{SO}_4(\text{OH})_4 \cdot 5\text{H}_2\text{O}$	Small	$Z = \frac{b}{c} (?)$ $Y:el = 43^\circ$	MCL laths	---	White	G 1.85	---
1.524	1.504	<u>1.512</u>	1.516	.012	LAUMONTITE var Leonhardite (Zeolite grp) $\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 4\text{H}_2\text{O}$	44° (70+19°) $r < v$ str	$Y = \frac{b}{c}$ $Z:c = 30^\circ$	MCL	010, 110 good	White	H 3.5-4 G 2.3 F 2.5-3	Gel with acids. H_2O 12.8%.
1.535	1.512	<u>1.514</u>	1.515	.003	OKENITE $\text{CaSi}_2\text{O}_4(\text{OH})_2 \cdot \text{H}_2\text{O}$	Large	$Z \sim \frac{c}{a}$	TCL fib $\frac{c}{a}$	010 perf	White	H 5 G 2.3 F 2.5	Gel with acids.
	---	<u>1.514</u>	---	---	KINGITE $\text{Al}_3(\text{PO}_4)_2(\text{OH,F})_3 \cdot 9\text{H}_2\text{O}$	---	---	TCL (?) mass	---	White	G 2.2-2.3	Opt char unk.
	1.508	<u>1.515</u>	1.520	.012	NACAPHITE $\text{Na}_2\text{Ca}(\text{PO}_4)\text{F}$	80° $r > v$ wk	---	ORTH	Conch	Cols	H 3 G 2.85	Diss by HCl.
	1.445	<u>1.515</u>	1.523	.078	GAYLUSSITE $\text{Na}_2\text{Ca}(\text{CO}_3)_2 \cdot 5\text{H}_2\text{O}$	35° $r < v$ str crossed	$X = \frac{b}{c}$ $Z:c = -15^\circ$ disp str	MCL el $\frac{a}{c}$	110 perf 001 diff	Cols	H 2.5-3 G 1.99 F 1.5	Dec by H_2O , diss by acids with eff.
1.509 ^	1.512	<u>1.515</u>	1.517	.005	COWLESITE (Zeolite grp) $\text{CaAl}_2\text{Si}_3\text{O}_{10} \cdot 5-6\text{H}_2\text{O}$	44-53°	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el pos	ORTH bladed	010 perf	Cols, white	H 5-5.5 G 2.13	---
	1.501	<u>1.516</u>	1.525	.024	LETOVICITE $(\text{NH}_4)_3\text{H}(\text{SO}_4)_2$	75°	$Z = \frac{b}{c}$ $Y:c = -12^\circ$	MCL tab 001	001 dist	Cols	G 1.83 volat	Sol in H_2O . Tw lam.
1.528	1.512	<u>1.516</u>	1.518	.006	GISMONDINE (Zeolite grp) $(\text{Ca,Na})\text{Al}_2\text{Si}_2\text{O}_8 \cdot 4\text{H}_2\text{O}$	Large $r < v$ wk	$Y = \frac{b}{c}$ $Z:c = 42^\circ$	MCL ps tet	---	Cols	H 4.5 G 2.2 fus	Gel with acids. Tw.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
218 [] 1.507 ^ 1.51 □ 1.53 1.512 ^	1.513	<u>1.516</u>	1.518	.005	LOVDARITE (Na,K,Ca) ₂ (Be,Al)Si ₃ O ₈ ·2H ₂ O	~ 90°	---	ORTH pris	100, 010, 001	Cols to yellowish	H 6.5 G 2.33 F 3.5	Insol in acids.
	1.501	<u>1.517</u>	1.518	.017	SYNGENITE K ₂ Ca(SO ₄) ₂ ·H ₂ O	28° r < v str	Z = $\frac{b}{c}$ X:c = -2° el neg	MCL laths el c	110, 100 perf 010	White	H 2.5 G 2.60 F 1.5-2	Dec by H ₂ O, diss by acids. Tw pl 100 common.
	1.507	<u>1.517</u>	1.521	.014	NATROSILITE Na ₂ Si ₂ O ₅	55° r < v	Y ~ c	MCL ps hex thick tab	100 perf 001 dist 011 poor	Cols	G 2.48 F 2	Dec by acids. Anom blue interf colors.
	1.508	<u>1.518</u>	1.522	.014	ILESITE (Mn,Zn,Fe)SO ₄ ·4H ₂ O	Med	Y = $\frac{b}{c}$ Z:c = 5°	MCL tab 100	---	Clear green	G 2.25	Sol in H ₂ O. Data on synth compd.
	1.512	<u>1.518</u>	1.519	.007	SCOLECITE (Zeolite grp) CaAl ₂ Si ₃ O ₁₀ ·3H ₂ O	35° r < v str	Z = $\frac{b}{c}$ X:c = 18° el neg	MCL el c	110 perf	White	H 5 G 2.27 F 2	Gel with acids. Tw 100 common
	1.513	<u>1.518</u>	1.520	.007	NICKELBLOEDITE Na ₂ Ni(SO ₄) ₂ ·4H ₂ O	60-70°	Y = $\frac{b}{c}$	MCL	---	Light green	G 2.43	Sol in H ₂ O. Ni 13.3, MgO 1.0, FeO 0.9%.
	1.433	<u>1.519</u>	1.528	.095	WEGSCHEIDERITE Na ₅ (CO ₃)(HCO ₃) ₃	(34+4°)	---	TCL	---	Cols	G 2.34 F easy	Sol in H ₂ O, gives an alk soln.
	---	<u>1.52</u>	---	.005-.01	BASALUMINITE Al ₄ SO ₄ (OH) ₁₀ ·5H ₂ O	---	el neg	Mass compact	Conch	White	G 2.1	Diss by HCl. Opt char unk.
	---	<u>1.52</u>	---	wk	CHLOROCALCITE KCaCl ₃	---	---	ORTH (?) ps cub	Three clv	Cols	H 2.5-3	Sol in H ₂ O, deliq. Poly tw. Validity dubious.
	1.496	<u>1.521</u>	1.539-1.544	.043-.048	INDERBORITE CaMg[B ₃ O ₃ (OH) ₅] ₂ ·6H ₂ O	80-86°	Z = $\frac{b}{c}$ X:c = 2.5°	MCL	100 good fr conch	Cols	H 3.5 G 2.00 fus	Diss by acids.
	1.510	<u>1.521</u>	1.523	.013	BIKITAITE LiAlSi ₂ O ₆ ·H ₂ O	45° r < v	Z = $\frac{b}{c}$ X:c = 28°	MCL	001, 100 poor	Cols	H 6 G 2.30	---
	1.504	<u>1.522</u>	1.539	.035	HANNAYITE (NH ₄) ₂ H ₄ Mg ₃ (PO ₄) ₄ ·8H ₂ O	~ 90° r < v wk	---	TCL tab 100	111 complete 100, 010 poor	Cols to yellowish	G 2.03 fus	Diss by acids. Data on synth compd.

v 1.539	1.518	<u>1.522</u>	1.524	.006	MICROCLINE (Feldspar grp) KAlSi ₃ O ₈	60-70° r > v	X':a on 001 = 0-10° 010 = 5°	TCL	001 perf 010 good	White, pink, green	H 6 G 2.56 F 4	Insol in acids. Poly grating tw on 010 and 100.	
	1.519	<u>1.522</u>	1.524	.005	ANORTHOCLASE (Feldspar grp) (K,Na)AlSi ₃ O ₈	51° r > v wk	X':a on 001 = 0-3° 010 = 8-9°	TCL	001 perf 010 good	White, cols	H 6 G 2.57 F 4	Insol in acids. Tw pl 010, 100, giving very fine grating. K ₂ O 3.7, Na ₂ O 8.4, CaO 0.7%. (K ₂ O and Na ₂ O interchanged?)	
	1.504	<u>1.523</u>	1.531	.027	IVANOVITE Chloride-borate of Ca	58-72° r < v	X = <u>b</u> Z:c = 20-26° el clv pos	MCL ps hex	010 good	---	H 2-3 G 1.96	Sol in H ₂ O.	
	(1.492)	(<u>1.524</u>)	1.526	(.034)	SWINEFORDITE (Smectite grp) (Li,Ca,Na)(Al,Li,Mg) ₄ (Si,Al) ₈ O ₂₀ (OH,F) ₄	29-45°	X ~ <u>c</u> Y = <u>a</u>	MCL mass	---	Greenish- gray to olive	H ~ 1	Pleoc wk, Y nearly cols, Z pale yellow- brown.	
	1.518	<u>1.524</u>	1.530	.012	MACDONALDITE BaCa ₄ Si ₁₆ O ₃₆ (OH) ₂ · 10H ₂ O	90°	X = <u>c</u> Y = <u>b</u>	ORTH acic	010 perf 001 good	Cols, white	H 3.5-4 G 2.27 F 5.5	Dec by boiling HCl.	
	1.512 ^	1.513	<u>1.524</u>	1.525	.012	LAUMONTITE (Zeolite grp) CaAl ₂ Si ₄ O ₁₂ ·4H ₂ O	25° r < v str	Y = <u>b</u> Z:c = 20-30°	MCL	010, 110 good	White	H 3.5-4 G 2.3 F 2.5-3	Gel with acids.
	1.522 [v 1.544	1.519	<u>1.524</u>	1.525	.006	ORTHOCLASE (Feldspar grp) KAlSi ₃ O ₈	46° r > v wk	Y or Z = <u>b</u> X:a = 3-12° r < v	MCL	001 perf 010 good	Cols, flesh, pink, etc	H 6 G 2.57 F 4	Insol in acids. Tw axis <u>c</u> comp pl 010, others less common.
	[v 1.529	1.520	<u>1.525</u>	1.525	.005	SANIDINE (Feldspar grp) (K,Na)AlSi ₃ O ₈	24°	X':a on 010 = 6°	MCL	001 perf 010 good	White, cols	H 6 G 2.56	Insol in acids.
		1.510	<u>1.525</u>	1.536	.026	URALOLITE CaBe ₃ (PO ₄) ₂ (OH) ₂ ·4H ₂ O	(80+9°)	Z':el = 20°	MCL fib aggre- gates and needles	el indist	Cols, white	H 2.5 G 2.1	---
		1.518	---	1.530	.012	TACHARANITE Ca ₁₂ Al ₂ Si ₁₈ O ₅₁ ·18H ₂ O	---	X:fib = 0°	MCL fib	001 good	Cols	G 2.36	Opt char unk.
1.506 v 1.553	1.516	<u>1.525</u>	1.526	.010	SEPIOLITE Mg ₄ Si ₆ O ₁₅ (OH) ₂ ·6H ₂ O	Med	---	ORTH fib	---	White	H 2 G 2.3 infus	Dec by HCl with sepn of silica. Fe ₂ O ₃ 2.5%.	

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.512 ∧ 1.537	1.513	(1.526)	1.534	.021	SIDEROTIL (Chalcanthite grp) (Fe,Cu)SO ₄ ·5H ₂ O	60°	---	TCL	---	Pale green to blue	H 2-3 G 2.1-2.2	Sol in H ₂ O.
1.535	1.521	1.526	1.528	.007	CORDIERITE Mg ₂ Al ₄ Si ₅ O ₁₈	64°	X = $\frac{c}{b}$ Y = $\frac{b}{c}$	ORTH ps hex	010 fair	Cols	H 7 G 2.51 F 5.5	Partly dec by acids. Data for synth compd.
	1.452	1.527	1.538	.086	STENONITE (Sr,Ba,Na) ₂ Al(CO ₃)F ₅	43°	X = $\frac{b}{c}$ Z:c = -32°	MCL	001, 120	Cols, white	H 3.5 G 3.86	Diss by acids.
	1.429	1.528	1.538	.109	AMEGHINITE NaB ₃ O ₃ (OH) ₄	33° r < v	Z = $\frac{b}{c}$ X:c = 9°	MCL	100 good 010, 001 poor	Cols	H 2.5 G 2.03 F easy	Sol in H ₂ O. Fluor pale blue in UV.
1.516 ∧ 1.543	1.522	1.528	1.530	.008	GISMONDINE (Zeolite grp) (Ca,Na)Al ₂ Si ₂ O ₈ ·4H ₂ O	15-90° r < v wk	Y = $\frac{b}{c}$ Z:c = 42°	MCL ps tet	---	Cols	H 4.5 G 2.22 F 3	Gel with acids. Tw pl 110, 001.
1.525 ∧ 1.534	1.523	1.529	1.530	.007	SANIDINE (Feldspar grp) (Na,K)AlSi ₃ O ₈	33°	---	MCL	001 perf 010 good	Cols	H 6 G 2.56	Insol in acids.
	1.365	1.530	1.595	.265	GLUSHINSKITE (an oxalate) MgC ₂ O ₄ ·2H ₂ O	(58±1°)	---	MCL	---	---	H 2 G 1.85 infus	Diss by HCl.
1.536	1.518	1.530	1.542	.024	MINASRAGRITE VO(SO ₄)·5H ₂ O	Med large (89±9°)	X = $\frac{b}{c}$ Z ~ $\frac{c}{b}$	MCL pris	---	Blue	G 2.03 F easy	Sol in H ₂ O. Pleoc str, X deep blue, Y blue, Z nearly cols, abs X > Y > Z.
1.546	1.522	1.530	1.533	.011	PALYGORSKITE MgAlSi ₄ O ₁₀ (OH)·4H ₂ O	61°	---	ORTH fib	---	White	G 2.13 - 2.27 infus	Insol in acids.
	1.522	1.530	1.531	.009	FEDORITE (Na,K)Ca(Si,Al) ₄ (O,OH) ₁₀ ·1.5H ₂ O	32° r < v	X ~ $\frac{c}{b}$ Y ~ $\frac{b}{c}$	TCL	001 perf	Cols to pale red	G 2.58 F easy	Insol in acids.
1.510 ∧ 1.565	1.490	1.531	1.534	.044	SAPONITE (Smectite grp) (Ca,Na) _{0.33} (Mg,Fe) ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	Med (30±8°)	X ⊥ 001 Z = el	MCL plates	001 perf	White, pink, gray, green	H 1.2 G 2.3 F diff	Dec by acids.
	1.522	(1.531)	1.536	.014	BOYLEITE (Zn,Mg)SO ₄ ·4H ₂ O	70°	---	MCL	Uneven	White	G 2.41 infus	Sol in H ₂ O.

	1.524	<u>1.532</u>	1.536	.012	KOKTAITE (NH ₄) ₂ Ca(SO ₄) ₂ ·H ₂ O	72°	Y = 2° b Z:c = 2° el pos	MCL acic	---	Cols	G 2.09	Dec by H ₂ O, diss by acids. Tw 100.
	1.529	<u>1.532</u>	1.533	.004	DELHAYELITE (Na,K) ₁₀ Ca ₅ Al ₆ Si ₃₂ O ₈₀ (Cl,F,SO ₄) ₃ ·18H ₂ O	~ 90°	X = a Y = c	ORTH platy	010 dist	Cols	H ~ 4 G 2.60 F 3	Diss by acids. Wavy ext, anomal blue-gray b.
	1.511	<u>1.533</u>	1.535	.024	NYEREREITE (Na,K) ₂ Ca(CO ₃) ₂	29°	X = c Y = a	ORTH tab	---	Cols	G 2.54 fus	Diss by acids with eff. Tw.
	1.515	<u>1.533</u>	1.535	.020	SEARLESITE NaBSi ₂ O ₅ (OH) ₂	Med	Z = b X:c = 30°	MCL pris	100 perf	White	Soft G 2.45 F easy	Diss by acids.
1.529 □ 1.535	1.488	<u>1.534</u>	1.556	.068	ARTINITE Mg ₂ CO ₃ (OH) ₂ ·3H ₂ O	70°	Y = b Z:c = 30°	MCL acic, radi- ating, botry- oidal	100 perf 001 good	White, silky	H 2.5 G 2.02	Diss by acids with eff.
1.512 △	1.503	<u>1.534</u>	1.534	.031	MONTMORILLONITE (Smectite grp) (Na,Ca) _{0.33} (Al,Mg,Fe) ₂ Si ₄ O ₁₀ (OH) ₂ ·xH ₂ O	Small	Y = b X ~ 1 001	MCL ps hex earthy	001 perf	White, gray, pink	H 1-2 G 2.06- 2.23	Dec by acids.
1.529 △ 1.543	1.527	<u>1.534</u>	1.535	.008	ALBITE plagioclase, volcanic (Feldspar grp) (Na,Ca,K)AlSi ₃ O ₈	~ 50°	X':a on 001 = 3° 010 = 23°	TCL	001 perf 010 good	Cols	H 6 G 2.60 F 4	Insol in acids. Forms series with Anorthite and Sanidine. Data for end member Ab ₁₀₀ An ₀ Or ₀ .
v (1.543)	1.528	(1.534)	1.536	.008	APLOWITE (Co,Mn,Ni)SO ₄ ·4H ₂ O	Med	---	MCL	---	Pink	H 3 G 2.33 infus	Sol in H ₂ O.
	1.410	<u>1.535</u>	1.543	.133	NATRITE Na ₂ CO ₃	28°	Y ~ b	MCL	001 perf 010, 100 less so	---	H 3.5 G 2.54	Sol in H ₂ O. Poly tw.
	1.500	<u>1.535</u>	1.560	.060	MEYERHOFFERITE Ca ₂ B ₆ O ₁₁ ·7H ₂ O	78° r > v perc	Z':c on 100 = 25° 010 = 60°	TCL tab 100	010 perf	Cols, white	H 2 G 2.12 F easy	Diss by acids.
	1.515	<u>1.535</u>	1.536	.021	GLAUBERITE Na ₂ Ca(SO ₄) ₂	7° r > v str	Z = b Y:a = 12° el clv pos	MCL tab 001	001 perf 110 in- dist brittle	Gray, cols, yellow- red	H 2.5-3 G 2.75- 2.85 F 1.5	Dec by hot H ₂ O, diss by acids.
1.514 ^	1.530	<u>1.535</u>	1.540	.010	OKENITE CaSi ₂ O ₄ (OH) ₂ ·H ₂ O	Large	Z ~ c	TCL fib c	010 perf	White	H 5 G 2.3 F 2.5	Gel with acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.531	<u>1.535</u>	1.541	.010	VERTUMNITE $\text{Ca}_4\text{Al}_4\text{Si}_4\text{O}_6(\text{OH})_{24} \cdot 3\text{H}_2\text{O}$	62°	$X = \frac{c}{b}$ $Z:b = \frac{c}{b} = 16^\circ$	MCL	Conch brittle	Col's	H 5 G 2.15	Gel with acids.
1.526 ^ 1.546	1.530	<u>1.535</u>	1.538	.008	CORDIERITE $(\text{Mg}, \text{Fe})_2\text{Al}_4\text{Si}_5\text{O}_8$	84°	$X = \frac{c}{b}$ $Y = \frac{c}{b}$	ORTH ps hex	010 fair 001 poor	Blue, green, col's	H 7 G 2.59 F 5.5	Partly dec by acids. Tw sectors, also lam. Pleoc, X col's to pale yellow, Y pale to dark blue, Z pale blue. FeO 2.1%.
	1.423	<u>1.536</u>	1.554	.131	TESCHEMACHERITE $(\text{NH}_4)\text{HCO}_3$	49° (41+2°) $r < v$	$X = \frac{a}{b}$ $Y = \frac{a}{b}$	ORTH mass	110 perf	Col's to yellowish	H 1.5 G 1.57 volat	Sol in H_2O , eff with acids.
v 1.561	1.494	<u>1.536</u>	1.536	.042	BEIDELLITE (Montmorillonite grp) $(\text{Na}, \text{Ca})_{0.33}\text{Al}_2(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	9-16°	$X = \frac{c}{b}$ $Z = \frac{a}{b}$ el cTv pos	MCL earthy	001 perf	White, reddish, gray	G 2.0 infus	Refractive indices increase when immersed in certain oils.
1.530 ^	1.513	<u>1.536</u>	1.545	.032	MINASRAGRITE $\text{VO}(\text{SO}_4) \cdot 5\text{H}_2\text{O}$	Med large	$X = \frac{b}{c}$ $Z \sim \frac{c}{b}$	MCL pris	---	Blue	G 2.03 F easy	Sol in H_2O . Pleoc str, X deep blue, Y blue, Z nearly col's, abs X > Y > Z.
	1.527	<u>1.536</u>	1.541	.014	ROZENITE $\text{FeSO}_4 \cdot 4\text{H}_2\text{O}$	Large $r < v$ wk	---	MCL	---	Col's to greenish	G 2.20	Sol in H_2O .
	---	---	1.542	---	GLAUCOKERINITE $(\text{Zn}, \text{Cu})_{10}\text{Al}_4\text{SO}_4(\text{OH})_{30} \cdot 2\text{H}_2\text{O} (?)$	---	$Z = \text{el}$	Radiating, fib	---	Sky-blue	H 1 G 2.15	Concentric color banding. Opt char unk.
1.545	1.490	<u>1.537</u>	1.538	.048	SCHROECKINGERITE $\text{NaCa}_3(\text{UO}_2)(\text{CO}_3)_3(\text{SO}_4)\text{F} \cdot 10\text{H}_2\text{O}$	18°	$X = \frac{c}{b}$ $Y = \frac{c}{b}$	TCL ps hex	001	Greenish-yellow	H 5 G 2.54	Diss by H_2O . Fluor yellow-green in UV.
(1.526) ^	1.501	<u>1.537</u>	1.539	.038	CHALCANTHITE $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$	56° $r < v$ perc	---	TCL pris	110 im-perf fr conch	Dark blue	H 2.5 G 2.29	Sol in H_2O .
	1.455	<u>1.538</u>	1.545	.090	IKAITE $\text{CaCO}_3 \cdot 6\text{H}_2\text{O}$	~ 45° (31+5°) disp inclined	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 17^\circ$	MCL	---	Col's	G 1.77	Dec by acids.

	1.516	<u>1.538</u>	1.547	.031	KORSHUNOVSKITE $Mg_2Cl(OH)_3 \cdot 3.5H_2O$	62°	el neg	TCL pris	---	Cols	H ~ 2 G 1.80	Sol in dil acids.
1.510 ^	(1.522)	<u>1.538</u>	1.542	(.020)	ARSENURANOSPATHITE $HAl(UO_2)_4(AsO_4)_4 \cdot 40H_2O$	52° r > v	X = $\frac{c}{el}$ pos	TET or ORTH ps tet	001 perf 100, 010 good	Pale yellow	H 2 G 2.54	Sol in acids. Fluor wk in UV.
	1.534	<u>1.538</u>	1.543	.009	CANASITE $(Na,K)_6Ca_5Si_{12}O_{30}(OH,F)_4$	58°	Y = $\frac{b}{Z}$	MCL	2 perf at 118°	Greenish- yellow	G 2.71 F easy	Dec by acids. Poly tw.
	1.496	<u>1.539</u>	1.557	.061	BIRINGUCCITE $Na_4B_{10}O_{16}(OH)_2 \cdot 2H_2O$	63°	Y = $\frac{b}{Z}$ a = 5°	MCL acic	001, 100 good	Cols	---	Diss in H ₂ O.
1.522 ^	1.536	<u>1.539</u>	1.541	.005	ANORTHOCLASE (Feldspar grp) $(K,Na,Ca)AlSi_3O_8$	62° r > v wk	ext on 010 = 3°	TCL	010, 001 perf	White, pink, etc	H 6 G 2.62 F 4	Insol acids. Tw pl 010, 100, giving very fine grating. Na ₂ O 7.2, K ₂ O 3.3, CaO 3.7%.
	1.522	<u>1.540</u>	1.552	.030	SULFOBORITE $Mg_3B_2(SO_4)(OH)_{10}$	70-88°	Y = $\frac{b}{X}$ a =	ORTH pris	110 good 001 fair	Cols	H 4-4.5 G 2.42 fus	Diss by acids.
1.541 pos ^ 1.546	1.536	<u>1.541</u>	1.546	.010	OLIGOCLASE plagioclase, plutonic (Feldspar grp) $(Na,Ca)(Al,Si)_2Si_2O_8$	~ 90° r < v wk	X': a on 001 = 1° 010 = 7°	TCL	001 perf 010 good	White	H 6 G 2.64 F 4-5	Insol in acids. Tw pl 010 poly, almost universal. Data for An ₁₈ .
	1.520	<u>1.541</u>	1.545	.025	LUENEBOURGITE $Mg_3B_2(PO_4)_2(OH)_6 \cdot 5H_2O$	62° (47±15°)	---	MCL fib	Pris	Cols to brownish	H 2 G 2.05 fus	Diss by acids.
1.537 ┘	1.466	<u>1.542</u>	1.596	.130	DAWSONITE $NaAlCO_3(OH)_2$	77° r < v wk	X = $\frac{a}{Y}$ c =	ORTH bladed acic	110 perf	Cols, white	H 3 G 2.44 infus	Diss by acids with eff.
	1.528	<u>1.542</u>	1.549	.021	KOVDORSKITE $Mg_{10}(PO_4)_4(CO_3)_2(OH)_4 \cdot 9H_2O$	81° r > v wk	Z:c = 1-3° el pos	MCL	Conch to uneven	Pale rose	H 4 G 2.60	---
	1.535	<u>1.542</u>	1.549	.014	FOSHALLASITE $Ca_3Si_2O_7 \cdot 3H_2O$	12-18° (90±16°)	---	MCL(?) tab	100 perf	Snow- white	H 2.5-3 G 2.5	---
1.528 ^	1.538	<u>1.543</u>	1.548	.010	GISMONDINE (Zeolite grp) $(Ca,Na)Al_2Si_2O_8 \cdot 4H_2O$	15-90° r < v wk	Y = $\frac{b}{Z}$ c = 42°	MCL ps tet	---	Cols, white	H 4.5 G 2.22 F 3	Gel with acids. Tw common, tw pl 110, 001.
	1.536	---	1.550	.014	LOUDOUNITE $NaCa_5Zr_4Si_{16}O_{40}(OH)_{11} \cdot 8H_2O$	---	---	fib	---	Green to white	H 5 G 2.48	Sign unk. Ext wavy.
1.534 ^ 1.552	1.538	<u>1.543</u>	1.546	.008	OLIGOCLASE plagioclase, volcanic (Feldspar grp) $(Na,Ca)(Al,Si)_2Si_2O_8$	65°	X': a on 001 = 3° 010 = 4°	TCL	001 perf 010 good	White, cols	H 6 G 2.63 F 4-5	Insol in acids. Tw pl 010 poly. Data for An ₂₀ .

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
(1.534) ^	1.533	(1.543)	1.546	.013	APLOWITE (Co,Mn,Ni)SO ₄ ·4H ₂ O	Med	---	MCL	---	Pink	H 3 G 2.33 infus	Sol in H ₂ O.
	1.539	1.543	1.544	.005	EPIDIDYMIT NaBeSi ₃ O ₇ (OH)	2-3°	X = $\frac{a}{b}$ Z = $\frac{b}{a}$	ORTH	001 good 010 perf	Cols	H 5.5 G 2.61 F easy	Insol in acids. Multiple tw.
	1.522	1.544	1.552	.030	MUNDRABILLAITE (NH ₄) ₂ Ca(HPO ₄) ₂ ·H ₂ O	(61+9°)	X = $\frac{b}{c}$ Y: $\frac{c}{b}$ = 26°	---	---	Cols	Soft G 2.05	Sol in H ₂ O.
	1.533	1.544	1.548	.015	BEHOITE Be(OH) ₂	82° r < v str	Y = $\frac{a}{c}$ Z = $\frac{c}{a}$	ORTH	Conch	White	H 4 G 1.92	Diss by acids.
	1.533	1.544	1.547	.014	PENTAGONITE Ca(VO)Si ₄ O ₁₀ ·4H ₂ O	50° r > v str	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH	010 good	Greenish-blue	H 3-4 G 2.33	Pleoc str, X and Z cols, Y blue.
1.524 ^ 1.568	1.542	1.544	1.546	.004	HYALOPHANE (Feldspar grp) (K,Ba)Al(Si,Al) ₃ O ₈	75° r > v	Z = $\frac{b}{a}$ X: $\frac{a}{b}$ = -18°	MCL	010, 001 perf	Cols, white	H 6 G 2.79 F diff	Insol in acids. BaO 6.9% (Cn ₃₇).
1.560 v	1.525	1.545	1.545	.020	VERMICULITE (Mg,Fe,Al) ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	~ 0°	X ⊥ plates	MCL	001 perf	Greenish, brownish	H 1.5 G 2.28 F 5	Dec by HCl. Exfoliates when heated. FeO 0.7, Fe ₂ O ₃ 4.2%.
	1.533	1.545	1.547	.014	MOOREITE (Mg,Zn,Mn) ₈ SO ₄ (OH) ₁₄ · 3H ₂ O	50° r > v wk	X = $\frac{b}{c}$ Z: $\frac{c}{b}$ = 44° el clv pos	MCL tab to platy 010	010 perf	Cols, white	H 3 G 2.47 infus	Diss by acids. Mg:Zn:Mn = 4:2:1.
1.541 ^ 1.550	1.541	1.546	1.550	.009	OLIGOCLASE plagioclase, plutonic (Feldspar grp) (Na,Ca)(Al,Si) ₂ Si ₂ O ₈	~ 78°	X': $\frac{a}{a}$ on 001 = 1° 010 = 2°	TCL	001 perf 010 good	Cols, white	H 6 G 2.65 F 4-5	Insol in acids. Tw pl 010, poly. Data for An ₂₅ .
	1.487	1.546	1.560	.073	BRADLEYITE Na ₃ Mg(PO ₄)(CO ₃)	49° r < v	X = $\frac{b}{c}$ Y: $\frac{c}{b}$ = 7°	MCL flat- tened 001	Conch	White, gray	H 3.5 G 2.72	Dec by H ₂ O, diss by acids with eff.
	1.507	1.546	1.569	.062	CARBOBORITE Ca ₂ Mg(CO ₃)B ₂ (OH) ₈ ·4H ₂ O	75°	Y = $\frac{b}{c}$ Z: $\frac{c}{b}$ = 12° el pos	MCL pris	100 perf	Cols	H 2 G 2.12	Fluor white in UV, pale green phosphorescence.
1.530 ^ (~1.57)	1.528	1.546	1.548	.020	PALYGORSKITE (Mg,Al,Fe) ₂ Si ₄ O ₁₀ (OH)· 4H ₂ O	30-35°	---	ORTH fib	---	White, greenish	G 2.4-2.5 infus	Insol in acids. Fe ₂ O ₃ 6.6%.

1.535 ∧ 1.560	1.540	<u>1.546</u>	1.548	.008	CORDIERITE (Mg,Fe) ₂ Al ₄ Si ₅ O ₁₈	68°	X = $\frac{c}{b}$ Y = $\frac{a}{b}$	ORTH ps hex	010 fair 001 poor	Blue, green, cols	H 7 G 2.61 F 5.5	Partly dec by acids. Tw sectors, also lam. Pleoc, X pale yellow, Y dark blue, Z blue. FeO 5.1, Fe ₂ O ₃ 0.4%.
	1.438	<u>1.547</u>	1.595	.157	OXAMMITE (NH ₄) ₂ C ₂ O ₄ ·H ₂ O	62° r < v dist	X = $\frac{c}{a}$ Y = $\frac{a}{a}$ el cTlv pos	ORTH acic $\frac{c}{c}$	001 dist	Cols to yellowish	H 2.5 G 1.5 F volat	An oxalate. Sol in H ₂ O.
	1.537	<u>1.547</u>	1.549	.012	GYROLITE Ca ₂ Si ₃ O ₇ (OH) ₂ ·H ₂ O	Small	el clv pos	HEX	0001 perf	Cols	H 3-4 G 2.40 F diff	Dec by acids.
v 1.564	1.522	(<u>1.548</u>)	1.549	.027	PHLOGOPITE (Mica grp) KMg ₃ Si ₃ AlO ₁₀ (F,OH)	14° r < v	Y = b el cTlv pos	MCL ps hex	001 perf	Cols	H 2-2.5 G 2.88	Data for synth fluor- phlogopite. (Pleoc, X cols, Y = Z brown, abs X < Y < Z?)
v 1.555	1.529	<u>1.548</u>	1.553	.024	LEPIDOLITE (Mica grp) K(Li,Al) ₃ (Si,Al) ₄ O ₁₀ (F,OH) ₂	39° (54+12°) r > v	Z':clv = 0-7° el clv pos	MCL ps hex	001 perf	Cols, pink, gray	H 2.5-4 G 2.81 F 2.5	Insol in acids. FeO 0.1, Fe ₂ O ₃ 0.2, MnO 0.7, Li ₂ O 5.6%.
	1.522	<u>1.549</u>	1.549	.027	TRUSCOTTITE (Ca,Mn) ₁₄ Si ₂₄ O ₅₈ (OH) ₈ ·2H ₂ O	Small	---	HEX	One perf	White	G 2.47	Dec by HCl.
[]	1.519	<u>1.550</u>	1.559	.040	"RECTORITE" Na(Mg,Al) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂ ·2H ₂ O	0-39° (56+7°)	---	MCL	001	White	H 1-2 G 2.2 infus	Interstratified Pyrophyllite- Vermiculite.
	1.520	---	1.558	.038	CHELKARITE CaMgB ₂ O ₄ Cl ₂ ·7H ₂ O (?)	---	---	ORTH pris	Perf el	Cols to pale rose	G 2.21	Opt char unk.
[] 1.59	1.530	(<u>1.550</u>)	1.551	.021	SAUCONITE (Smectite grp) Na _{0.33} Zn ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	0-20°	X ⊥ clv Y = b el cTlv pos	MCL u mass	001 perf	White, yellow, brown	H 1.5-2 G 2.61	Dec by acids. ZnO 35.7, MgO 2.4, Al ₂ O ₃ 0.4%.
	1.537	<u>1.550</u>	1.556	.019	NAUJAKASITE Na ₆ (Fe,Mn)(Al,Fe) ₄ Si ₈ O ₂₆	52-71°	---	MCL ps hex plates	001 perf	Silvery, gray	H 2-3 G 2.62 F 3	Gel with HCl.
1.546 ∧ 1.550 pos	1.546	<u>1.550</u>	1.554	.008	ANDESINE plagioclase, plutonic (Feldspar grp) (Na,Ca)(Al,Si) ₂ Si ₂ O ₈	~ 90	X':a on 001 = 0° 010 = -5°	TCL	001 perf 010 good	White, cols	H 6 G 2.65 F 4.5	Insol in acids. Poly tw 010. Data for An ₃₃ .
	1.530	---	1.559	.029	YOFORTIERITE (Mn,Mg) ₅ Si ₈ O ₂₀ (OH) ₂ · 8-9H ₂ O	---	X:fib = 8°	ORTH or mon fib	---	Pink to violet	H 2.5 G 2.18	Opt char unk. Pleoc.
	1.519	<u>1.552</u>	1.561	.042	BALAVINSKITE Sr ₂ B ₆ O ₁₁ ·4H ₂ O	Large (54+6°)	---	---	---	---	---	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref.	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.535	<u>1.552</u>	1.553	.018	SATIMOLITE $\text{KNa}_2\text{Al}_4\text{B}_6\text{O}_{15}\text{Cl}_3 \cdot 13\text{H}_2\text{O}$	Small	---	ORTH	---	White	G 2.1	---
1.543 ^ 1.552 pos	1.549	<u>1.552</u>	1.556	.007	ANDESINE plagioclase, volcanic (Feldspar grp) $(\text{Na,Ca})(\text{Al,Si})_2\text{Si}_2\text{O}_8$	~ 90°	X':a on 001 = 2° 010 = -4°	TCL	001 perf 010 good	Cols	H 6 G 2.66 F 4.5	Insol in acids. Tw pl 010 poly. Data for An ₃₈ .
v 1.567	1.485	<u>1.553</u>	1.570	.085	ALUMOHYDROCALCITE $\text{CaAl}_2(\text{CO}_3)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	64° (51+3°)	X = b ext = 7- 10°	MCL	100 perf	Chalky to pale blue	H 2.5 G 2.23 infus	Dec by hot H ₂ O, diss by acids with eff.
	1.550	<u>1.553</u>	1.554	.004	CHAROITE $(\text{K,Ba,Sr})(\text{Ca,Na})_2\text{Si}_4\text{O}_{10}(\text{OH,F}) \cdot \text{H}_2\text{O}$	30°	ext:el = 5°	MCL	Three clv	Lilac to violet	G 2.54	In thick fragments pleoc, X rose, Z cols. U pos.
v 1.570	1.522	<u>1.553</u>	1.553	.031	TAENIOLITE (Mica grp) $\text{KLiMg}_2\text{Si}_4\text{O}_{10}\text{F}_2$	0-5° r > v	Y = b el cTv pos	MCL ps hex	001 perf	Cols to brown	H 2.5-4 G 2.83	Insol in acids.
1.525 ^	(1.522)	<u>1.553</u>	1.579	(.057)	SEPIOLITE (ferrian) $(\text{Mg,Fe})_4\text{Si}_6\text{O}_{15}(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	83°	el pos	ORTH fib	001, 010	Brown	H 2.5 G 2.3 infus	Dec by HCl. Fe ₂ O ₃ 10.0, FeO 1.2%.
	1.535	<u>1.553</u>	1.557	.022	HYDROCALUMITE $\text{Ca}_4\text{Al}(\text{OH})_7 \cdot 3\text{H}_2\text{O}$	25° (50+12°) r < v	Y = b el pos	MCL ps hex	001 perf	Cols to light green	H 3 G 2.15 infus	Diss by acids. Becomes uniax at about 90°C.
	1.541	<u>1.553</u>	1.560	.019	BERYLLITE $\text{Be}_3\text{Si}_4(\text{OH})_2 \cdot \text{H}_2\text{O}$	Small (74+12°)	el pos	Fib	---	White	Soft G 2.20	---
[1.549	1.541	<u>1.553</u>	1.557	.016	EDINGTONITE (Zeolite grp) $\text{BaAl}_2\text{Si}_3\text{O}_{10} \cdot 4\text{H}_2\text{O}$	54° r < v wk	X = c Z = a el neg	ORTH ps tet	110 perf 110	Cols	H 4 G 2.75 F 5	Gel with acids.
	1.498	<u>1.554</u>	1.594	.096	MINGUZZITE $\text{K}_3\text{Fe}(\text{C}_2\text{O}_4)_3 \cdot 3\text{H}_2\text{O}$	78°	---	MCL tab	010 perf	Green to yellow-green	G 2.03	An oxalate. Sol in H ₂ O. Pleoc, X yellow green, Z bright emerald-green.
	1.545	<u>1.554</u>	1.565	.020	Unnamed phosphate	~ 90° disp wk	---	MCL	---	Pale yellow, pale green	H 4.5 G 3.13	Diss by acids.

	1.531	<u>1.555</u>	1.570	.039	SHORTITE $\text{Na}_2\text{Ca}_2(\text{CO}_3)_3$	75° $r < v$ mod	$X = c$ $Z = \underline{b}$	ORTH hemi- morph	010 dist	Cols to yellowish	H 3 G 2.6 F 1	Dec by H_2O , diss by acids with eff.
1.548 ∧ 1.567	1.537	<u>1.555</u>	1.559	.022	LEPIDOLITE (Mica grp) $\text{K}(\text{Li}, \text{Al})_3(\text{Si}, \text{Al})_4\text{O}_{10}$ (F, OH) ₂	38° $r > v$	$Z' : \text{clv} =$ 0-7° el clv pos	MCL ps hex	001 perf	Cols, pink, lilac	H 2.5-4 G 2.86 F 2.5	Insol in acids. Fe_2O_3 1.1, FeO 0.7, MnO 0.9, Li_2O 5.5%.
v 1.568	1.545	<u>1.555</u>	1.555	.010	LIZARDITE (Serpentine grp) $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	Very small	$Y = \underline{b}$ $Z = \underline{a}$	MCL fib	001 perf	Green	G 2.55 infus	Slowly attacked by acids. Fe_2O_3 2.2%.
	1.552	(1.556)	1.559	.007	HALLOYSITE, chromian $(\text{Al}, \text{Cr})_2\text{Si}_2\text{O}_5(\text{OH})_4$	~ 90°	---	MCL mass	---	Pale greenish- blue	H 2.5 G 2.1 infus	Insol in acids.
┌ 1.57	1.545	---	1.56	.015	"MN-PALYGORSKITE" $\text{NaMgMn}(\text{Fe}, \text{Al})_3\text{Si}_7\text{O}_{20}$ (OH) ₂ ·10H ₂ O (?)	Large	$Z = \underline{c}$	Spher- ulitic	---	Rose to red	G 2.62	Pleoc, X yellow, Z red- brown, abs Z > X. Am. Mineral., 55, 2139 (1970).
	1.552	<u>1.558</u>	1.561	.009	BERYLLONITE NaBePO_4	68° $r < v$ wk	$X = \underline{b}$ $Z = \underline{c}$	MCL ps orth tab	010 perf 100 good	Cols to pale yellow	H 5.5-6 G 2.81 F 3	Slowly diss by acids.
	1.514	<u>1.559</u>	1.562	.048	ZINCSILITE $\text{Zn}_3\text{Si}_4\text{O}_{10}(\text{OH})_2$ 4H ₂ O (?)	0-22°	$Z : \underline{c} = 3^\circ$	MCL	001 perf	White to bluish	H 1.5-2 G 2.69	---
	1.551	<u>1.559</u>	1.562	.011	ARMENITE (Osumilite grp) $\text{BaCa}_2\text{Al}_6\text{Si}_9\text{O}_{30} \cdot 2\text{H}_2\text{O}$	60°	---	HEX	Three clv	Cols	H 7-8 G 2.77	---
	1.541	<u>1.560</u>	1.567	.026	FENAKSITE $(\text{K}, \text{Na}, \text{Ca})_4(\text{Fe}^{+2}, \text{Fe}^{+3},$ $\text{Mn})_2\text{Si}_8\text{O}_{20}(\text{OH}, \text{F})$	77° (62±10°)	$Z = \underline{b}$	TCL	001 perf	Pale rose	H 5-5.5 G 2.74 F easy	Dec by acids.
1.545 ∧ 1.581	1.540	<u>1.560</u>	1.560	.020	VERMICULITE $(\text{Mg}, \text{Fe}, \text{Al})_3(\text{Si}, \text{Al})_4$ $\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	6-8°	$X \perp 001$ el clv pos	MCL	001 perf	Green to brown	H 1.5 G 2.29 F 5	Exfoliates when heated. Dec by HCl. Fe_2O_3 4.2, FeO 1.8%.
	1.545	<u>1.560</u>	1.561	.016	LOVOZERITE $(\text{Na}, \text{Ca})_3(\text{Zr}, \text{Ti})\text{Si}_6$ (O, OH) ₁₈	Small	$X \sim \underline{c}$	TRIG	---	Dark brown to black	H 5 G 2.38 F easy	Insol in acids. Poly tw.
1.546 ∧ 1.569	1.551	<u>1.560</u>	1.562	.011	CORDIERITE $(\text{Mg}, \text{Fe})_2\text{Al}_4\text{Si}_5\text{O}_{18}$	42°	$X = \underline{c}$ $Y = \underline{b}$	ORTH ps hex	010 fair	Blue, green, cols	H 7 G 2.66 F 5	Partly dec by acids. Tw sectors, also lam. Pleoc, X cols, Y dark blue, Z blue. FeO 7.1, Fe_2O_3 0.2%.
	1.552	<u>1.5605</u>	1.561	.009	KULKEITE $\text{Mg}_8\text{Al}(\text{Si}_7\text{Al})\text{O}_{20}(\text{OH})_{10}$	24°	$Y = \underline{a}$ $Z = \underline{b}$	MCL	001 perf	Cols	H 2 G 2.70	Interlayered talc- chlorite.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.536 ∧ 1.588	1.526	<u>1.561</u>	1.561	.035	BEIDELLITE (Smectite grp) $\text{Na}_{0.33}\text{Al}_2(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	47° (small)	X = $\frac{c}{a}$ Z = $\frac{a}{b}$ el cTv pos	MCL earthy	001 perf	White, reddish, gray	G 2.0 infus	Refractive indices increase when immersed in certain oils.
~1.585 v	1.553	---	1.569	.016	KARPINSKITE (Mg, Ni) ₂ Si ₂ O ₅ (OH) ₂ (?) (?)	---	Z:c = 12°	MCL (?)	---	Cols to greenish-blue	H 2.5-3 G 2.6	---
	1.547	---	1.571	.024	GLUCINE $\text{Ca}_2\text{Be}_8(\text{PO}_4)_4(\text{OH})_8 \cdot \text{H}_2\text{O}$	---	el pos ext	Fine needles	---	Cols	H ~ 5 G 2.23-2.40	Diss by acids. Opt char unk.
	1.547	<u>1.562</u>	1.567	.020	POLYHALITE $\text{K}_2\text{Ca}_2\text{Mg}(\text{SO}_4)_4 \cdot 2\text{H}_2\text{O}$	62-70°	Z':tw on 10I = 28°	TCL fib b tab	10I perf parting 010	Red, yellow, cols	H 3.5 G 2.78 F 1.5	Dec by H ₂ O, diss by acids. Poly tw.
	1.557	<u>1.562</u>	1.563	.006	NACRITE $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	40-90° r > v wk	Z = $\frac{b}{a}$ X:c = 12°	MCL tab wedge-shaped	001 perf	White	G 2.60 infus	Insol in acids. Luster pearly.
	1.521	<u>1.563</u>	1.585	.064	SIDORENKITE $\text{Na}_3\text{Mn}(\text{PO}_4)(\text{CO}_3)$	63° (70+4°)	el $\frac{b}{a}$	MCL ps orth	100, 010 perf	Pale rose	H ~ 2 G 2.90	Diss by acids.
	1.541	<u>1.563</u>	1.564	.023	TANCOITE $\text{HNa}_2\text{LiAl}(\text{PO}_4)_2(\text{OH})$	23° r < v wk	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	MCL	010, 001 fair	Pale pink	H 4-4.5 G 2.75	Diss by acids.
	1.560	<u>1.563</u>	1.515	.005	SCHODERITE $\text{Al}_2(\text{PO}_4)(\text{VO}_4) \cdot 8\text{H}_2\text{O}$	42°	X = $\frac{b}{a}$ Y:c = 25°	MCL	---	Yellow-orange	H 2 G 1.92	Pleoc in yellow, abs Y > Z > X.
(1.548) ∧ 1.584	1.535	<u>1.564</u>	1.565	.030	PHLOGOPITE (Mica grp) $\text{K}(\text{Mg}, \text{Fe})_3\text{Si}_3\text{AlO}_{10}(\text{F}, \text{OH})_2$	10° r < v wk	Y = $\frac{b}{c}$ X ~ $\frac{c}{a}$ el cTv pos	MCL platy	001 perf	Yellow, brown	H 2-2.5 G 2.75 F diff	Insol in acids. Pleoc, X cols, Y = Z brown, abs X < Y < Z.
	1.552	<u>1.564</u>	1.571	.019	JENNITE $\text{Ca}_9\text{H}_2\text{Si}_6\text{O}_{18}(\text{OH})_8 \cdot 6\text{H}_2\text{O}$	74°	Y: el = 35-40°	TCL bladed fib	001 perf	White	G 2.32	---
1.531 ^	1.485	<u>1.565</u>	1.572	.087	SAPONITE (Smectite grp) $(\text{Ca}, \text{Na})_{0.33}(\text{Mg}, \text{Fe})_3(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	Small (32+5°)	X ⊥ 001	MCL plates	001 perf	Dark green	H 1.5 G 2.31 F 4	Gel with acids. Pleoc, X pale yellow, Y olive-green, Z brown-green. FeO 7.8, Fe ₂ O ₃ 7.3, Al ₂ O ₃ 9.1%.

	1.536	<u>1.565</u>	1.570	.034	FLUOBORITE $Mg_3BO_3(F,OH)_3$	Small	---	HEX fib	0001 indist	White	H 3.5 G 2.82	Diff diss by acids. F 15.6, H_2O 6.2%.
v 1.592	1.550	<u>1.565</u>	1.570	.020	VARISCITE $AlPO_4 \cdot 2H_2O$	Med r < v wk	X = a Z = b el cTv neg	ORTH	010 good 001 poor	Green to cols	H 4.5 G 2.57- 2.75 infus	Insol in acids, but diss after being heated gently.
	1.554	<u>1.565</u>	1.573	.019	REEDMERGNERITE (Feldspar grp) $NaBSi_3O_8$	80°	---	TCL	001 perf	Cols	H 6-6.5 G 2.70 F 2	Insol in acids.
	1.553	<u>1.565</u>	1.567	.014	MORINITE $NaCa_2Al_2(PO_4)_2$ (OH,F) $_5 \cdot 2H_2O$	40° r < v wk	Y = b	MCL	100 perf	Wine-red, cols	H 4.5 G 2.96 F easy	Insol in acids.
	1.557	<u>1.565</u>	1.571	.014	VIITANIEMIITE $Na(Ca,Mn)Al(PO_4)$ (F,OH) $_3$	81°	Y = b	MCL tab	10I good	Gray, white	H 5 G 3.25	---
1.559 □ 1.569	1.560	<u>1.565</u>	1.566	.006	KAOLINITE $Al_2Si_2O_5(OH)_4$	24-50° r > v	Z = b X:c = 1-4°	MCL u earthy masses	001 perf	White to brownish	H 2-2.5 G 2.60 infus	Insol in acids.
┐ 1.577	1.560	<u>1.565</u>	1.566	.006	ZEOPHYLLITE $Ca_4Si_3O_8(OH,F)_4 \cdot 2H_2O$	Small	Z = b el cTv pos	TRIG	0001 perf	Cols, white	G 2.75 F easy	Gel with acids. Uniax and biax segments. Exfoliates when heated.
	1.546	<u>1.566</u>	1.573	.027	FURONGITE $Al_2(UO_2)(PO_4)_2(OH)_2 \cdot$ $8H_2O$	65°	---	TCL	3 perf	Bright yellow to lemon- yellow	G 2.86	Fluor in UV. Str in yellow-green.
	1.554	<u>1.566</u>	1.567	.013	TOBERMORITE $Ca_5Si_6O_{16}(OH)_2 \cdot 4H_2O$	26-33°	---	ORTH	---	Cols	G 2.45	---
v 1.577	1.561	<u>1.566</u>	1.567	.006	ANTIGORITE (Serpentine grp) $(Mg,Fe)_3Si_2O_5(OH)_4$	48° r > v	X = c el cTv pos	MCL	001 perf	Green to white	H 3-4 G 2.61 infus	Dec by acids.
1.553 ◇ ~1.587	1.504	<u>1.567</u>	1.575	.071	ALUMOHYDROCALCITE $CaAl_2(CO_3)_2(OH)_4 \cdot 3H_2O$	46-70°	X = b ext 7-10°	MCL	100 perf	White	H 2.5 G 2.23 infus	Dec by hot H_2O , diss by acids with eff.
1.555 ◇ 1.577	1.542	<u>1.567</u>	1.570	.028	LEPIDOLITE (Mica grp) $K(Li,Fe,Al)_3(Si,Al)_4$ $O_{10}(F,OH)_2$	34° r > v	X ~ c el cTv pos	MCL ps hex	001 perf	Cols, pink, brown	H 2.5-4 G 2.95 F 2.5	Insol in acids. Fe_2O_3 0.8, FeO 2.0, MnO 2.0, Li_2O 5.3%.
y 1.592	1.55	<u>1.567</u>	1.572	.022	URANOSPINITE (Autunite grp) $Ca(UO_2)_2(AsO_4)_2 \cdot 10H_2O$	0-62° r > v mod	X = c	TET tab	001 perf 100 dist	Lemon- yellow to green	H 2-3 G 3.45 fus	Diss by acids. Pleoc, X cols, Y and Z pale yellow.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.557	<u>1.567</u>	1.570	.013	OLSHANSKYITE $\text{Ca}_3\text{B}_4(\text{OH})_{18}$	54° $r > v$ wk	ext at 30°	MCL (?) fib	---	Cols	H 4	Diss by acids. Poly tw.
	1.520	<u>1.568</u>	1.591	.071	BONSHEDTITE $\text{NaFe}^{+2}(\text{PO}_4)(\text{CO}_3)$	$(68+3^\circ)$	$X = \underline{b}$ $Y = \underline{c}$	MCL ps orth tab	010, 100 perf	Cols, rose	H 4 G 3.16	Dec by acids with eff.
	1.541	(<u>1.568</u>)	1.570	.029	PSEUDO-AUTUNITE $(\text{H}_3\text{O})_4\text{Ca}_2(\text{UO}_2)_2$ $(\text{PO}_4)_4 \cdot 5\text{H}_2\text{O}$ (?)	32° $r > v$ dist	$Y = \underline{c}$	ORTH ps hex plates	001 perf	Pale yellow	G 3.28	Diss by acids. Fluor bright greenish-yellow in UV.
	1.562	<u>1.568</u>	1.574	.012	ELPIDITE $\text{Na}_2\text{ZrSi}_6\text{O}_{15} \cdot 3\text{H}_2\text{O}$	89°	$Z = \underline{a}$	ORTH fib	110 dist	White to tan	H 6 G 2.63	Insol in acids. U biax pos.
1.568 pos 1.577	1.563	<u>1.568</u>	1.573	.010	BYTOWNITE plagioclase, plutonic (Feldspar grp) $(\text{Ca}, \text{Na})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	$\sim 90^\circ$	$X': \underline{a}$ on 001 = -16° 010 = -29°	TCL	001 perf 010 good	White, gray	H 6 G 2.71 infus	Insol in acids. Tw pl 010 poly. Data for An ₇₀ .
1.544 1.583	1.564	<u>1.568</u>	1.572	.008	CELSIAN var Kasoite (Feldspar grp) $(\text{Ba}, \text{K})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	80°	$X: \underline{c}$ on 001 = $10-13^\circ$ 010 = $2-3^\circ$	MCL	001 perf 010 dist	Cols	H 5.5 G 3.00	Not tw. BaO 25.0% (Cn ₆₅).
1.555 1.638	1.562	<u>1.568</u>	1.570	.008	LIZARDITE (Serpentine grp) $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$	45°	---	MCL fib	001 perf	Green	G 2.58 infus	Insol in acids. Pleoc wk, X yellow, Y and Z green.
1.560 1.572	1.559	<u>1.569</u>	1.573	.014	SEKANINAITE (Cordierite grp) $(\text{Fe}, \text{Mg})_2\text{Al}_4\text{Si}_5\text{O}_{18}$	70°	$X = \underline{c}$ $Y = \underline{b}$	ORTH ps hex	100 fair	Blue to violet	H 7 G 2.77 F 5	Partly dec by acids. Tw pl 110, 310. Pleoc, X cols, Y and Z light blue, abs $Z > Y > X$. FeO 17.8%.
	1.563	<u>1.569</u>	1.573	.010	ARMSTRONGITE $\text{CaZrSi}_6\text{O}_{18} \cdot 2.5\text{H}_2\text{O}$	$r < v$	$Z = \underline{b}$ $Y: \underline{c} = 6^\circ$	MCL	001 perf 100 fair	Brown	H 4.5 G 2.58	Insol in acids.
1.561 1.586	1.560	<u>1.569</u>	(<u>1.570</u>)	.010	NONTRONITE var Volchonskoite (Smectite grp) $\text{Na}_{0.33}(\text{Fe}, \text{Cr})_2$ $(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	Small	$Y = \underline{b}$	MCL earthy	001 perf	Bright green	H 2 G 2.36 infus	Dec by acids. Cr_2O_3 22.1%.

1.569 pos 1.577	1.565	1.569	1.574	.009	BYTOWNITE plagioclase, volcanic (Feldspar grp) (Ca,Na)Al(Si,Al)Si ₂ O ₈	~ 90°	X':a on 001 = -23° 010 = -33°	TCL	001 perf 010 good	Cols, white	H 6 G 2.72 infus	Insol in acids. Tw poly. Data for An ₇₅ .
1.572	1.567	1.569	1.570	.003	PHOSINAITE Na ₃ H ₂ (Ca,Ce)(SiO ₄) (PO ₄)	69°	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH columnar	100 perf 010, 110 imperf	Cols to rose	H 3.5 G 2.62, 3.00 F 3	Slowly diss by concd acids.
	1.536	1.570	1.571	.035	MASUTOMILITE (Mica grp) K(Li,Mn,Al) ₃ (Si,Al) ₄ O ₁₀ (F,OH) ₂	28° r > v wk	Y = $\frac{b}{c}$ Z:a = 3° el clv neg	MCL	001 perf	Pale purplish- pink	H 2.5 G 2.90	Insol in acids. Pleoc, X and Z cols to pale purple, Y purple. MnO 4.3, Li ₂ O 5.8%.
1.578	1.543	1.570	1.577	.034	NOVACEKITE (Autunite grp) Mg(UO ₂) ₂ (AsO ₄) ₂ ·12H ₂ O	0-40°	X = $\frac{c}{a}$	TET tab	001 perf 100 dist	Yellow	H 2-2.5 G 3.23	Diss by acids. Fluor yellow in UV.
1.553 ^	1.540	1.570	1.570	.030	TAENIOLITE (Mica grp) KLiMg ₂ Si ₄ O ₁₀ F ₂	5° r < v	Y = $\frac{b}{c}$ el clv pos	MCL ps hex	001 perf	Cols to brown	H 2.5-4 G 2.87	Insol in acids. Pleoc wk, X cols, Z yellow.
1.582 v	---	1.570	1.626	---	BUKOVSKYITE Fe ₂ (AsO ₄)(SO ₄)(OH)· 7H ₂ O	---	ext at 18°	MCL (?) acic	---	Yellow to gray- green	G 2.34	Diss by HCl. Opt char unk.
1.582 v	1.559	1.570	1.574	.015	SALEEITE (Autunite grp) Mg(UO ₂) ₂ (PO ₄) ₂ ·10H ₂ O	0-65° r < v str	X = $\frac{c}{a}$	TET tab	001 perf 010, 110 indist	Lemon- yellow	H 2-3 G 3.27	Diss by acids. Pleoc, X cols, Z greenish- yellow. Fluor lemon- yellow in UV.
	1.564	1.570	1.574	.010	TOSUDITE (interlayered- Montmorillonite- Chlorite)	59°	el pos	MCL	001 perf 010	Cols	H ~ 1 G 2.83	---
	1.543	1.571	1.577	.034	SHABYNITE Mg ₅ (BO ₃)(Cl,OH) ₂ (OH) ₅ ·4H ₂ O	(49+9°)	X clv	MCL fib	---	White	H 3 G 2.32	Diss by acids.
1.564 ┌	1.524	1.571	1.583	.059	ROEMERITE Fe ⁺² Fe ⁺³ ₂ (SO ₄) ₄ ·14H ₂ O	51° r > v str crossed	ext on clv = 33°	TCL tab 001	010 perf 001 good	Brown to yellow	H 3-3.5 G 2.17 F 4.5-5	Sol in H ₂ O. Pleoc, X reddish yellow, Y pale yellow, Z yellow-brown.
1.546 ^	1.56	(~1.57)	1.58	.02	MANGANOPALYGORSKITE (Mn,Mg,Al) ₂ Si ₄ O ₁₀ (OH)·4H ₂ O	Med large	Z = $\frac{c}{a}$	ORTH fib	---	Rose to red	G 2.62	Pleoc, X yellow, Z red- brown.
	1.546	1.572	1.576	.030	DEFERNITE Ca ₃ (CO ₃) ₂ (OH,Cl) ₈ ·H ₂ O	42° r < v mod	X = $\frac{c}{a}$ Y = $\frac{b}{c}$	ORTH	---	Cols	G 2.5	Diss by cold HCl with eff.
1.580 v	(1.533)	1.572	1.573	.040	HAIWEEITE Ca(UO ₂) ₂ Si ₆ O ₁₅ ·5H ₂ O	18° r > v str	Y = $\frac{b}{c}$	MCL spher- ulitic	100 good	Pale yellow	H 3.5 G 3.35	Pleoc wk, Y pale yellow, Z cols. Fluor pale green in UV.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.562	(1.572)	1.583	.021	Unnamed hydrous calcium arsenate	~ 90°	Z = $\frac{b}{c}$ Y:c = 8°	MCL (?)	---	Cols	---	Opt char unk. Am. Mineral., 53, 561 (1968).
1.566 □ 1.581	1.570	1.572	1.573	.003	ENGLISHITE $K_4Na_2Ca_9Al_{18}(PO_4)_6$ $(PO_3OH)_{12}(OH)_{36} \cdot 8H_2O$	Small	X = $\frac{c}{b}$ Z ~ $\frac{b}{c}$ el cTv pos	MCL (?) plates	001 perf	Cols, white	H 2 G 2.65	---
	1.568	1.574	1.580	.012	OVERITE $CaMgAl(PO_4)_2(OH) \cdot 4H_2O$	65-85° r > v wk	X = $\frac{c}{b}$ Y = $\frac{b}{c}$	ORTH	010 perf	Pale green, cols	H 4 G 2.53 F 2	Diss by hot acids.
v 1.610	1.57	1.574	1.580	.010	BASSETITE (Meta-autunite grp) $Fe(UO_2)_2(PO_4)_2 \cdot 8H_2O$	62° r > v str	X = $\frac{b}{c}$ Z:c = -4°	MCL ps tet	010 perf	Olive-green, olive-brown	H 2.5 G 3.4	Diss by acids. Pleoc, X pale yellow, Y and Z deep yellow. Does not fluor in UV.
	1.548	1.574	1.582	.034	LITIDIONITE $KNaCuSi_4O_{10}$	56°	---	TCL	---	Blue	G 2.75	---
	1.546	1.575	1.579	.033	BRAMMALLITE (Mica grp) $(Na, H_3O)(Al, Mg, Fe)$ $(Si, Al)_4O_{10}[(OH)_2 \cdot H_2O]$	Small variable	X ~ $\frac{c}{b}$ el cTv pos	MCL	001 perf	White	H 1-2 G 2.69	Dec by acids.
v 1.587	1.555	1.575	1.581	.026	TOBELITE (Mica grp) $(NH_4, K)Al_2(Si_3Al)O_{10}(OH)_2$	28° (57±10°)	el pos	MCL	001 perf	White	---	---
□ 1.586	1.555	1.575	1.577	.022	AUTUNITE (Autunite grp) $Ca(UO_2)_2(PO_4)_2 \cdot 10-12H_2O$	0-53° u 10-30° r > v str	X = $\frac{c}{b}$	TET tab	001 perf	Lemon- to sulfur-yellow to greenish	H 2-2.5 G 3.1-3.2 fus	Diss by acids. Pleoc, X cols to pale yellow, Y and Z deep yellow. Fluor yellow-green in UV.
	1.567	1.576	1.579	.012	PANETHITE $(Na, Ca, K)(Mg, Fe, Mn)PO_4$	51°	---	MCL	---	Pale amber	G 2.95	Tw. FeO 5.3, MnO 1.7%. Meteorite mineral.
	1.558	1.576	1.593	.035	XIANGJIANGITE $(Fe^{+3}, Al)(UO_2)_4(PO_4)_2$ $(SO_4)_2(OH) \cdot 22H_2O$	(87±7°)	---	ORTH ps tet	---	Yellow	H 1-2 G 2.9-3.1	Diss by acids. Not fluor in UV. Pleoc wk in yellows.

	(1.558)	<u>1.576</u>	1.582	(.024)	MATULAITE $\text{CaAl}_{18}(\text{PO}_4)_{12}(\text{OH})_{20} \cdot 20\text{H}_2\text{O}$	60° r < v very str	$Y = \frac{b}{c}$ $Z:c = 8^\circ$	MCL	100 perf	Cols, white	H 1 G 2.33	---
1.567 ^	1.548	<u>1.577</u>	1.579	.031	ZINNWALDITE (Mica grp) $\text{KLiFe}^{+2}\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{F},\text{OH})_2$	30° r > v	$X \sim c$ el cTv pos	MCL ps hex	001 perf	Brown	H 2.5-4 G 3.01 F 2.5	Insol in acids. FeO 10.4, MnO 1.7, Fe_2O_3 0.6, Li_2O 2.0%.
	1.574	(<u>1.577</u>)	1.577	.003	EKATERINITE $\text{Ca}_2\text{B}_4\text{O}_7(\text{Cl},\text{OH})_2 \cdot 2\text{H}_2\text{O}$	Very small	---	HEX (?)	---	White, rose	H ~ 1 G 2.44	Diss by acids. Indices decrease in oils.
1.568 ◇ 1.584	1.571	<u>1.577</u>	1.583	.012	BYTOWNITE plagioclase, plutonic (Feldspar grp) $(\text{Ca},\text{Na})\text{Al}(\text{Al},\text{Si})\text{Si}_2\text{O}_8$	79°	$X':a$ on 001 = -30° 010 = -37°	TCL	001 perf 010 good	White, gray, cols	H 6 G 2.74 infus	Insol in acids. Tw pl 010, 001 poly. Data for An_{88} .
1.569 ◇ 1.584	1.572	<u>1.577</u>	1.583	.011	BYTOWNITE plagioclase, volcanic (Feldspar grp) $(\text{Ca},\text{Na})\text{Al}(\text{Al},\text{Si})\text{Si}_2\text{O}_8$	81°	$X':a$ on 001 = -34° 010 = -36°	TCL	001 perf 010 good	White	H 6 G 2.74 infus	Insol in acids. Tw pl 010, 001 poly. Data for An_{90} .
1.566 ◇ 1.603	1.567	<u>1.577</u>	1.578	.011	ANTIGORITE (Serpentine grp) $(\text{Mg},\text{Fe},\text{Cr})_3\text{Si}_2\text{O}_5(\text{OH})_4$	58° r < v str	$X = \frac{c}{b}$	MCL laths	001 perf	Lavender	H 4 G 2.6 F diff	Insol in acids. Abnorm blue interf color. Cr_2O_3 4.4%.
	1.544	<u>1.578</u>	1.601	.057	KROEHNKITE $\text{Na}_2\text{Cu}(\text{SO}_4)_2 \cdot 2\text{H}_2\text{O}$	78° r < v inclined	$Y = \frac{b}{c}$ $X:c = 48^\circ$	MCL pris	010 perf fr conch	Blue to greenish-blue	H 2 G 2.90 F easy	Sol in H_2O . Tw 101 common.
1.576 □ 1.583	1.544	<u>1.578</u>	1.586	.042	TARASOVITE (Mica grp) $(\text{Ca},\text{Na})_{0.42}\text{KNa}(\text{H}_3\text{O})\text{Al}_8(\text{Si},\text{Al})_{16}\text{O}_{40}(\text{OH})_8 \cdot 2\text{H}_2\text{O}$	23° (51±6°)	---	MCL scales	---	White	G 2.36	Interlayered mica-clay.
	1.570	---	1.584	.014	URANOSILITE $\text{UO}_3 \cdot 7\text{SiO}_2$	---	el pos	ORTH acic	---	Yellow-white	G 3.25	---
	1.572	<u>1.578</u>	1.582	.010	MONTGOMERYITE $\text{Ca}_4\text{MgAl}_4(\text{PO}_4)_6(\text{OH})_4 \cdot 12\text{H}_2\text{O}$	75° r < v wk	$Z = \frac{b}{c}$ $X:c = 60^\circ$ el clv neg	MCL laths	010 perf	Deep green to cols	H 4 G 2.53	Pleoc, X cols to pale green, Y and Z cols.
v 1.588	1.576	<u>1.578</u>	1.578	.003	CLINOCHLORE (Chlorite grp) $(\text{Mg},\text{Fe})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_8$	~ 0° r > v wk	$X \sim c$ el cTv pos	MCL ps hex	001 perf	Green	H 2.5 G 2.7 F diff	Dec by H_2SO_4 . Abnormal blue interf colors. Pleoc, X nearly cols, Y and Z green.
	1.567	<u>1.579</u>	1.581	.014	AGRELLITE $\text{NaCa}_2\text{Si}_4\text{O}_{10}(\text{F},\text{OH})$	47°	---	TCL	1T0, 110 perf 010 poor	White	H 5.5 G 2.90	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
\square	1.569	<u>1.579</u>	1.583	.014	SUDOITE (Chlorite grp) $Mg_2(Al,Fe)_3Si_3AlO_{10}(OH)_8$	78° $r < v$ str	$Z \sim \perp c$ el pos (?)	MCL plates	001 perf	Cols to green	H 2-3 G 2.68	Partly dec by acids. Dioctahedral chlorite.
1.572 ^	1.56	<u>1.580</u>	1.581	.021	HAIWEEITE $Ca(UO_2)_2Si_6O_{15} \cdot 5H_2O$	$16-20^\circ$ $r > v$ str	$Y = \underline{b}$	MCL spherulitic	110 good	Yellow	H 3.5 G 3.35	Pleoc wk, Y pale yellow, Z cols. Fluor faint green in UV.
1.474 ^	1.562	<u>1.580</u>	1.582	.020	HISINGERITE $Fe^{+3}_2Si_2O_5(OH)_4 \cdot 2H_2O$	Small $r > v$ wk	$Z = \underline{c}$	MCL u mass	---	Brown	H 3.5 G 2.3-2.6	---
1.560 ^ 1.607	1.561	<u>1.581</u>	1.581	.020	VERMICULITE $(Mg,Fe,Al)_3(Si,Al)_4O_{10}(OH)_2 \cdot 4H_2O$	$\sim 0^\circ$	$X \perp$ plates el clv pos	MCL	001 perf	Green, brown	H 1.5 G 2.6 F 5	Dec by HCl. Exfoliates when heated. Fe_2O_3 7.4, FeO 1.1%.
	1.575	<u>1.581</u>	1.583	.008	KINGSMOUNTITE $(Ca,Mn^{+2})_4(Fe^{+2},Mn)Al_4(PO_4)_6(OH)_4 \cdot 12H_2O$	62°	$Z:el = 35^\circ$	MCL	---	Cols	H 2.5 G 2.51	MnO 8.1, FeO 4.1%.
1.585]	1.576	<u>1.581</u>	1.584	.008	SORENSENITE $Na_4SnBe_2Si_6O_{16}(OH)_4$	$0-75^\circ$ str, inclined	---	MCL	2 at 63°	Cols to pink	G 2.90	Anom yellow-brown and blue interference colors.
1.570 ^	---	<u>1.582</u>	1.631	---	BUKOVSKYITE $Fe_2(AsO_4)(SO_4)(OH) \cdot 7H_2O$	---	ext at 18°	MCL (?) acid	---	Yellow to gray-green	G 2.34	Diss by HCl. Opt char unk.
1.628 v	1.548	(<u>1.58</u>)	1.600	.052	LAUSENITE $(Fe,Al)_2(SO_4)_3 \cdot 6H_2O$	Large	---	MCL fib	---	Cols	---	Fe/Al = 1.03.
1.590 v	1.552	<u>1.582</u>	1.587	.035	MUSCOVITE (Mica grp) $KAl_2(Si_3Al)O_{10}(OH)_2$	$30-47^\circ$ $r > v$	$Z = \underline{b}$ $X:c \sim 0^\circ$ el clv pos	MCL ps hex	001 perf	Pale brown, pale green	H 2.5-3 G 2.77 F 5.5	Insol in acids. Tw pl 001.
1.570 ^	1.565	<u>1.582</u>	1.585	.020	SALEEITE (Autunite grp) $Mg(UO)_2(PO_4)_2 \cdot 10H_2O$	$0-65^\circ$ $r < v$ str	$X = \underline{c}$	TET tab	001 perf 010, 100 indist	Lemon-yellow	H 2-3 G 3.27	Diss by acids. Pleoc, X cols, Z greenish-yellow. Fluor lemon-yellow in UV.

v 1.598	1.564	<u>1.582</u>	1.584	.020	SABUGALITE (Autunite grp) $\text{HA1}(\text{UO}_2)_4(\text{PO}_4)_4 \cdot 16\text{H}_2\text{O}$	35°	el pos	TET plates	001 perf	Bright yellow	H 2.5 G 3.20	Diss by acids. Pleoc, X cols, Y and Z pale yellow. Fluor bright lemon-yellow in UV.
	1.576	<u>1.582</u>	1.584	.008	TERSKITE $\text{Na}_4\text{ZrSi}_6\text{O}_{16} \cdot 2\text{H}_2\text{O}$	53° r > v wk	---	ORTH ps tet	---	Pale lilac	H 5 G 2.71	Insol in acids. Bright green in UV.
	1.574	<u>1.582</u>	1.582	.008	HOPEITE $\text{Zn}_3(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	Small r < v wk	$X = \frac{a}{c}$ $Y = \frac{c}{c}$	ORTH tab pris	010 perf 100 good	Cols to yellow	H 3.5 G 3.05 F easy	Diss by acids.
	1.576	<u>1.582</u>	1.584	.008	GUERINITE $\text{Ca}_5\text{H}_2(\text{AsO}_4)_4 \cdot 9\text{H}_2\text{O}$	7-15° r > v str also r < v	Z = el	MCL acic	Three clv	Cols	H 1.5 G 2.68	Diss by acids.
v 1.603	1.514	<u>1.583</u>	1.595	.081	STRONTIODRESSERITE $(\text{Sr}, \text{Ca})\text{Al}_2(\text{CO}_3)_2(\text{OH})_4 \cdot \text{H}_2\text{O}$	42°	$Y = \frac{el}{el}$ $X \perp el$	ORTH laths	---	White	G 2.71	Diss by HCl with eff.
	1.554	<u>1.583</u>	1.583	.029	STILPNOMELANE $\text{K}(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3}, \text{Al})_{10}\text{Si}_{12}\text{O}_{30}(\text{OH})_{12}$	~ 0°	$X = \frac{b}{c}$ $Y \sim \frac{c}{c}$	MCL also TCL	001 perf	Golden- to reddish- brown	H 3-4 G 2.74 F 4.5	Dec by acids. Pleoc, X pale yellow-brown, Y and Z dark brown. FeO 26.7, Fe ₂ O ₃ 2.1, MnO 0.4%.
	(1.573)	<u>1.583</u>	1.583	.010	THREADGOLDITE $\text{Al}(\text{UO}_2)_2(\text{PO}_4)_2(\text{OH}) \cdot 8\text{H}_2\text{O}$	70°	$Y = \frac{b}{c}$ $Z : \frac{c}{c} = 4^\circ$ el neg	MCL tab	001, 100 010, 012	Greenish- yellow	G 3.4	Fluor green in long- wave UV.
	1.574	<u>1.583</u>	1.588	.014	URANOCIRCITE $\text{Ba}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 12\text{H}_2\text{O}$	60-70°	$X = \frac{c}{c}$	TET tab	001 perf	Yellow- green	H 2 G 3.8	Diss by acids. Pleoc wk, X nearly cols, Y and Z pale yellow. Fluor green in UV.
1.568 ◇ 1.589	1.580	<u>1.583</u>	1.586	.006	CELSIAN (Feldspar grp) $(\text{Ba}, \text{K})\text{Al}(\text{Si}, \text{Al})\text{Si}_2\text{O}_8$	74°	Z:c on 100 = 33°	MCL	001 perf 010 good	Cols	H 5.5 G 3.30	Not tw (Cn ₈₅).
v 1.599	1.545	<u>1.584</u>	1.584	.039	TALC $\text{Mg}_3\text{Si}_4\text{O}_{10}(\text{OH}, \text{F})_2$	0-10° r > v	$X \sim \frac{c}{c}$ $Z = \frac{b}{b}$ el pos	TCL plates	001 perf	White to green	H 1 G 2.7 infus	Tcl polymorph. Pearly luster, greasy feel.
1.564 ◇ 1.598	1.555	<u>1.584</u>	1.588	.033	PHLOGOPITE (Mica grp) $\text{K}(\text{Mg}, \text{Fe})_3\text{Si}_3\text{AlO}_{10}(\text{F}, \text{OH})_2$	23-30° r < v wk	$Y = \frac{b}{c}$ $X \sim \frac{c}{c}$ el pos	MCL plates	001 perf	Brown	H 2-2.5 G 2.86 F diff	Insol in acids. Pleoc, X cols to pale yellow, Y = Z brown, abs X < Y < Z. FeO 4.2, Fe ₂ O ₃ 3.7%.
	1.568	<u>1.584</u>	1.585	.017	LAPLANDITE $\text{Na}_4\text{CeTiPSi}_7\text{O}_{22} \cdot 5\text{H}_2\text{O}$	Small	---	ORTH fib radi- ating	Splintery	Gray to yellowish	H 2-3 G 2.83 F 2	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.577 ^	1.570	1.584	1.585	.015	TORREYITE (Mg,Mn) ₅ Zn ₂ (SO ₄) (OH) ₁₂ ·4H ₂ O	40°	X = \underline{b}	MCL	010 good	White	G 2.66	Diss by acids. Poly tw.
	1.575	1.584	1.589	.014	ANORTHITE plagioclase, plutonic, and volcanic (Feldspar grp) CaAl ₂ Si ₂ O ₈	77° r < v	X':a on 001 = -40° 010 = -39°	TCL	001 perf 010 good	Cols, white, pink	H 6-6.5 G 2.76	GeI with acids. Tw poly 010, 001 almost universal, other tw laws common. Data for An ₁₀₀ .
	1.582	1.584	1.584	.002	ZECTZERITE NaLiZrSi ₆ O ₁₅	~ 0° r > v wk	X = \underline{a} Y = \underline{b}	ORTH	100, 010 perf	Cols to pink	H 6 G 2.79	Fluor light yellow in short-wave UV.
	1.575	1.585	1.595	.020	WHITEITE-(MN) (Mn,Ca)Mg ₂ (Fe,Mn)Al ₂ (PO ₄) ₄ (OH) ₂ ·8H ₂ O	80-90°	X = \underline{b} Y = \underline{c}	MCL	001 good	Tan to brown	H 3-4 G 2.67	Diss by acids. Tw 001. FeO 7.9, MnO 7.6, MgO 10.1%.
	1.564	1.585	1.585	.021	URAMPHITE (Meta-autunite grp) (NH ₄)(UO ₂)PO ₄ ·3H ₂ O	0-3°	---	---	2 dist	Bottle-green to pale green	G 3.7	Diss by acids. Fluor med yellow-green in UV. Pleoc, X cols, Y and Z pale green.
~1.562 ^	1.570	---	1.594	.024	KARPINSKITE (Mg,Ni) ₂ Si ₂ O ₅ (OH) ₂	---	Z:c = 0-12° el pos	MCL(?)	---	Blue	H 2.5-3 G 2.6	Cryptocryst. NiO 2.1%.
1.611 ^	1.545	1.586	1.589	.044	BANNISTERITE (Na,K)(Mn,Fe,Al) ₅ (Si,Al) ₆ O ₁₅ (OH) ₅ ·2H ₂ O	Med (30+9°) r < v wk	Z ~ \underline{a}	MCL	---	Brown	G 2.84	Pleoc, X nearly cols, Y and Z brown.
1.569 ^ 1.600	1.545	1.586	1.589	.044	NONTRONITE (Smectite grp) Na _{0.33} Fe ⁺³ ₂ (Si,Al) ₄ O ₁₀ (OH) ₂ ·xH ₂ O	Small to med r < v	Y = \underline{b} Z ~ \underline{c}	MCL u mass	001 perf	Dark brown to olive-green	H 1-2 G 2.50	Dec by acids. Pleoc, X pale yellow, Y olive-green, Z yellow green.
(1.600) ^	1.559	1.586	1.586	.027	GLAUCONITE var Skolite (Mica grp) (K,Na)(Al,Fe,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	Small r < v	Y = \underline{b} X:c = 10°	MCL	001 perf	Cols, blue, green	H 2 G 2.6 fus	Dec by acids. Fe ₂ O ₃ 6.4, Al ₂ O ₃ 18.2, FeO 2.6%.

1.567 ^	1.540	---	1.592	.052	ALUMOHYDROCALCITE ("Knipovichite") $\text{Ca}(\text{Al}, \text{Cr})_2(\text{CO}_3)_2$ $(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	---	$X = b$ ext 7-10°	MCL fib	100 perf	Pink	H 2.5 G 2.23 infus	Diss by acids with eff. Cr_2O_3 8.8%.
1.590 v	1.52	1.587	1.613	.093	LANTHANITE $(\text{La}, \text{Ce}, \text{Nd})_2(\text{CO}_3)_3 \cdot 8\text{H}_2\text{O}$	62° r < v wk	$X = b$ $Y = \frac{c}{a}$ el cTv pos	ORTH platy	010 perf	Cols, pink	H 2.5-3 G 2.69 infus	Diss by acids with eff. Tw and comp pl 101.
1.575 ^	1.560	1.587	1.595	.035	TOBELITE (Mica grp) $(\text{NH}_4, \text{K})\text{Al}_2\text{Si}_3\text{AlO}_{10}$ $(\text{OH})_2$	30° (56+8°)	el pos	MCL	001 perf	Yellow- green	---	---
1.571 □	1.572	1.587	1.600	.028	ILLITE (Mica grp) $(\text{K}, \text{H}_3\text{O})(\text{Al}, \text{Mg}, \text{Fe})_2$ $(\text{Si}, \text{Al})_4\text{O}_{10}[(\text{OH})_2 \cdot \text{H}_2\text{O}]$	Small to med (85+9°)	$X \sim \frac{c}{a}$ el cTv pos	MCL	001 perf	Cols, brown, green	H 1-2 G 2.7	Slightly dec by acids.
1.590 v	1.574	1.587	1.599	.025	KRAUSKOPFITE $\text{BaSi}_2\text{O}_4(\text{OH})_2 \cdot 2\text{H}_2\text{O}$	88° r > v dist	$X = \frac{b}{a}$ $Y : a = 6^\circ$	MCL	---	Cols	G 3.14 F 5	Dec by acids.
	1.578	1.587	1.595	.017	LEIGHTONITE $\text{K}_2\text{Ca}_2\text{Cu}(\text{SO}_4)_4 \cdot 2\text{H}_2\text{O}$	60° (86+13°) r > v str	$X \sim \frac{b}{a}$ $Y \sim \frac{c}{a}$ $Z \sim \frac{a}{b}$	TCL ps orth pris	---	Pale blue	H 3 G 2.95	Diss by acids. Multiple tw on 100 and 010.
	1.548	1.588	1.600	.052	GORDONITE $\text{MgAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Med r > v	$X \sim \frac{b}{a}$	TCL tab	010 perf 100 fair	White to pale green	H 3.5 G 2.23 F 3	Diss by acids. Fe^{+3} present.
	1.550	1.588	1.590	.040	COCONINOITE $\text{Fe}_2\text{Al}_2(\text{UO}_2)_2(\text{PO}_4)_4$ $(\text{SO}_4)(\text{OH})_2 \cdot 20\text{H}_2\text{O}$	40° (19+8°)	---	MCL (?) aggre- gates	---	Creamy yellow	Soft G 2.70	Diss by acids. Pleoc, X cols, Y and Z pale yellow. Not fluor in UV.
1.578 ◇	1.583	1.588	1.589	.006	KAEMMERERITE (Chlorite grp) $(\text{Mg}, \text{Al}, \text{Cr})_6(\text{Si}, \text{Al})_4\text{O}_{10}$ $(\text{OH})_8$	8° r > v wk	$X \sim \frac{c}{a}$ el pos	MCL	001 perf	Rose, lilac	H 2.5 G 2.72 F diff	Insol in HCl. FeO 2.0, Cr_2O_3 7.9%.
1.561 ◇	1.559	1.588	1.588	.029	BEIDELLITE (Smectite grp) $\text{Na}_{0.33}(\text{Al}, \text{Fe})_2(\text{Si}, \text{Al})_4$ $\text{O}_{10}(\text{OH})_2 \cdot x\text{H}_2\text{O}$	Small to med	$X = \frac{c}{a}$ $Z = \frac{a}{b}$	MCL earthy	001 perf	White to red	G 2.1 infus	Dec by acids.
1.600 v	1.556	1.589	1.601	.045	PYROPHYLLITE $\text{Al}_2\text{Si}_4\text{O}_{10}(\text{OH})_2$	62° r > v wk	$Z = \frac{b}{a}$ $X : c = 10^\circ$ el clv pos	MCL	001 perf	White, brown, yellow	H 1-2 G 2.7-2.9 infus	Nearly insol in acids.
	1.583	1.589	1.594	.011	PHARMACOLITE $\text{CaHAsO}_4 \cdot 2\text{H}_2\text{O}$	79° r > v	$Z = \frac{b}{a}$ $X : c = -29^\circ$	MCL acic	010 perf	Cols	H 2-2.5 G 2.5-2.7 F 2.5	Diss by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.583 ◇ 1.589 pos	1.583	<u>1.589</u>	1.593	.010	CELSIAN (Feldspar grp) $BaAl_2Si_2O_8$	88°	Z:c on 100 = 26°	MCL	001 perf 010 good	Cols	H 5.5 G 3.35	Not tw (Cn_{94}).
1.582 ◇ 1.595	1.563	<u>1.590</u>	1.594	.031	MUSCOVITE var Phengite (Mica grp) $KAl_2(Si_3Al)O_{10}(OH)_2$	33° r > v	Z = $\frac{b}{c}$ X ~ $\frac{c}{a}$ el pos	MCL ps hex tab	001 perf	Pale brown, green	H 2.5-3 G 2.82 F 5.5	Insol in acids. Fe_2O_3 1.5, FeO 1.9, MgO 4.0%.
	1.525	<u>1.590</u>	1.593	.068	BRENKITE $Ca_2CO_3F_2$	27°	X = $\frac{c}{a}$ Y = $\frac{a}{b}$	ORTH laths	---	Cols	H 5 G 3.10	Diss by HCl with eff.
(1.550) ◇ (1.614)	1.57	<u>1.59</u>	1.60	.03	SAUCONITE (Smectite grp) $Na_{0.33}Zn_3(Si,Al)_4O_{10}(OH)_2 \cdot 4H_2O$	0-20°	Y = $\frac{b}{c}$ el cTv pos	MCL mass	001 perf	Cols	H 2 G 2.7	Dec by acids.
	1.572	<u>1.590</u>	1.601	.029	KOLBECKITE $ScPO_4 \cdot 2H_2O$	60° (75+8°) r > v wk	X = $\frac{c}{a}$ Z = $\frac{a}{b}$	MCL pris	010 good fr conch	Blue-gray	H 3.5-4 G 2.39 F easy	Diss by acids.
1.587 ^	1.532	<u>1.590</u>	1.614	.082	LANTHANITE-(Nd) (Nd,La) $_2(CO_3)_3 \cdot 8H_2O$	61°	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH	010 perf 101 good	Cols	G 2.8 infus	Diss by HCL with eff.
	1.575	<u>1.590</u>	1.601	.026	DALYITE $K_2ZrSi_6O_{15}$	72° r > v wk	X:c = 7°	TCL pris	101, 010 good	Cols	H 7.5 G 2.66 F 3	Gel with acids.
1.630 v	1.575	<u>1.590</u>	1.594	.019	CARPHOLITE (Mn,Fe)Al $_2Si_2O_6(OH)_4$	50° r > v str	X = $\frac{b}{c}$ Z = $\frac{c}{a}$ el pos	ORTH fib c	Pris at 68°	Pale yellow	H 5-6 G 3.04 F 3.5	Nearly insol in acids. Pleoc, X and Y pale yellow, Z cols. MnO 16.1, FeO 0.6, Fe_2O_3 2.4%.
	1.581	<u>1.590</u>	1.591	.010	TUSCANITE $K(Ca,Na)_6(Si,Al)_{10}O_{22}[SO_4,CO_3,(OH)_2] \cdot H_2O$	40°	Z = $\frac{b}{c}$ X:c = 40°	MCL platy	100 dist	Cols	H 5.5-6 G 2.83	Compare Latiumite.
1.609 v	1.582	<u>1.590</u>	1.591	.009	LATIUMITE (Ca,K) $_8(Al,Mg,Fe)(Si,Al)_{10}O_{25}(SO_4)$	40° r > v str	Z = $\frac{b}{c}$ X:c = 16-28°	MCL tab	100 perf	White	H 5.5-6 G 2.93 F easy	Gel with acids. Tw pl 100. Mottled extinction.
1.585 ┌	1.572	<u>1.591</u>	1.598	.026	PRICEITE $Ca_4B_{10}O_{19} \cdot 7H_2O$	32° (62+10°)	X:plates = 65°	TCL (?) rhombic plates	110, 001 perf	White, chalky	H 3-3.5 G 2.42 F 2-3	Diss by acids.

	1.569	<u>1.592</u>	1.620	.051	LOKKAITE (Y,Ca) ₂ (CO ₃) ₃ ·2H ₂ O	(94±4°)	Z = $\frac{c}{el}$ pos	ORTH fib	---	White	Infus	Eff in acids.
1.567 ^	1.566	<u>1.592</u>	1.596	.030	URANOSPINITE (Autunite grp) Ca(UO ₂) ₂ (AsO ₄) ₂ ·10H ₂ O	25-35° (42±10°) r > v mod	X = $\frac{c}{el}$ pos	TET plates	001 perf 100 dist	Lemon- yellow	H 2-3 G 3.45 fus	Pleoc, X cols, Y and Z yellow. Fluor bright lemon-yellow in UV.
	1.573	<u>1.592</u>	1.599	.026	GAIDONNAYITE Na ₂ ZrSi ₃ O ₉ ·2H ₂ O	59°	X = $\frac{a}{Y}$ = $\frac{b}{c}$	ORTH	Conch	Cols	H 6 G 2.67 F 3	Gel with acids.
1.565 ^ 1.660 pos	1.572	<u>1.592</u>	1.597	.025	VARISCITE, ferrian (Al,Fe)PO ₄ ·2H ₂ O	Med large (55±10°)	X = $\frac{a}{Z}$ = $\frac{b}{r}$ < $\frac{v}{w}$ wk	ORTH	010 good 001 poor	White, green	H 4.5 G 2.57- 2.75 infus	Insol in acids but diss after heating gently.
1.586 □ 1.595	1.581	<u>1.592</u>	1.592	.011	TORBERNITE (Autunite grp) Cu(UO ₂) ₂ (PO ₄) ₂ ·8-12H ₂ O	Small r > v	X = $\frac{c}{el}$ pos	TET tab	001 perf	Emerald- green	H 2-2.5 G 3.22 F easy	Diss by acids. Pleoc, X cols to pale green, Y and Z pale to dark green. Not fluor in UV.
v 1.611	1.563	<u>1.593</u>	1.593	.030	GANOPHYLLITE (Na,K)(Mn,Fe,Al) ₅ (Si,Al) ₆ O ₁₅ (OH) ₅ ·2H ₂ O	~ 0° r < v	X = $\frac{c}{Z}$ = $\frac{b}{el}$ pos	MCL	001 perf	Yellow, brown	H 4 G 2.8 F 3	Gel with acids. Pleoc wk, X yellow, Y and Z nearly cols.
	1.570	<u>1.593</u>	1.595	.025	LEUCOPHANITE (Ca,Na) ₂ BeSi ₂ (O,OH,F) ₇	47°	X = $\frac{c}{Z}$ = $\frac{b}{el}$ pos	TCL ps orth tab 001	001 perf 100 dist	White to pale yellow	H 3.5-4 G 2.96 F 3	Insol in acids. Poly tw 110. Luminesces violet-blue in long- wave UV.
v 1.615	1.581	<u>1.593</u>	1.602	.021	TREMOLITE (Amphibole grp) Ca ₂ Mg ₅ Si ₈ O ₂₂ (OH,F) ₂	86°	Y = $\frac{b}{Z:c}$ = 21° el pos	MCL	110 good	Cols	H 5.5 G 3.02	Insol in acids. Data for synth compd.
	1.520	<u>1.594</u>	1.595	.075	HYDRODRESSERITE BaAl ₂ (CO ₃) ₂ (OH) ₄ ·3H ₂ O	17°	el pos	TCL radi- ating spher- ulitic	010, 2T0 perf	White	H 3-4 G 2.80	---
	1.572	<u>1.595</u>	1.614	.042	JOHANNITE Cu(UO ₂) ₂ (SO ₄) ₂ (OH) ₂ · 6H ₂ O	90° (83±6°) r > v str	X ~ $\frac{b}{Y:c}$ = 5-8° disp str	TCL tab el c	100 good	Green to greenish- yellow	H 2.5 G 3.32 infus	Diss by acids. Poly tw 100. Pleoc, X cols, Y pale yellow, Z greenish-yellow. Not fluor in UV.
1.590 ^ 1.598	1.569	<u>1.595</u>	1.609	.040	MUSCOVITE var Phengite (Mica grp) K(Al,Mg,Fe) ₂ (Si ₃ Al) O ₁₀ (OH) ₂	40° (71±6°) r > v	Z = $\frac{b}{X}$ ~ $\frac{c}{el}$ pos	MCL ps hex tab	001 perf	Emerald- green	H 2.5-3 G 2.82 F 5.5	Insol in acid. Pleoc, X blue-green, Y pale yellow-green, Z bluish. F 3.4, Fe ₂ O ₃ 3.1, Cr ₂ O ₃ 3.7, MnO 3.1%.
	1.581	<u>1.596</u>	1.601	.020	BALIPHOLITE BaMg ₂ LiAl ₃ Si ₄ O ₁₂ (OH) ₈	70°	X = $\frac{b}{Y}$ = $\frac{c}{c}$	ORTH fib c	010 perf 100, 110 dist	Yellow- ish-white	G 3.4	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
240												
	1.560	<u>1.597</u>	1.600	.032	THADEUITE Ca(Fe ⁺² ,Mn) ₂ Mg(PO ₄) ₂ (OH,F) ₂	33°	X = $\frac{c}{b}$ Y = $\frac{a}{b}$	ORTH	010 perf	Yellow-orange	H 4 G 3.25	---
	1.575	<u>1.597</u>	1.598	.023	CHRYSOCOLLA (Cu,Al) ₂ H ₂ Si ₂ O ₅ (OH) ₄ ·xH ₂ O	Small	---	MCL fib, u mass	---	Bluish-green, sky-blue	H 2 G 2.4 infus	Dec by acids. Optics highly variable.
	1.596	<u>1.597</u>	1.597	.001	EMELEUSITE (Osumilite grp) Na ₄ Li ₂ Fe ⁺³ ₂ Si ₁₂ O ₃₀	0-30° r > v str	X = $\frac{b}{a}$ Y = $\frac{c}{a}$	ORTH tab $\frac{b}{a}$	---	Cols	H 5-6 G 2.76	---
	1.516	<u>1.598</u>	1.621	.105	AMARANTITE FeSO ₄ (OH)·3H ₂ O	30° (53+3°) r < v str	X ~ $\frac{1}{2}$ 100 Y:c = 47°	TCL acic columnar	010, 100 perf	Red to orange-red	H 2.5 G 2.19 F 4.5-5	Dec by H ₂ O, diss by acids. Pleoc, X cols, Y pale yellow Z reddish-brown.
	1.584 ∧ 1.617	<u>1.598</u>	1.599	.041	BIOTITE (Mica grp) K(Mg,Fe ⁺²) ₃ (Al,Fe ⁺³) ₃ Si ₃ O ₁₀ (OH,F) ₂	10-13° r < v	Y = $\frac{b}{c}$ X ~ $\frac{c}{a}$ el pos	MCL plates	001 perf	Dark brown to green	H 2-2.5 G 2.95 F 3-4	Pleoc, X yellow, Y brown, Z dark brown, abs X < Y < Z. FeO 13.8, Fe ₂ O ₃ 2.1, MnO 0.4%.
	1.595 ∧ 1.608	<u>1.598</u>	1.600	.035	MUSCOVITE (Mica grp) KA ₂ (Si ₃ Al)O ₁₀ (OH) ₂	38° r > v	Z = $\frac{b}{c}$ X ~ $\frac{c}{a}$ el pos	MCL ps hex tab	001 perf	Pale brown, greenish, red	H 3-4 G 2.87 F diff	Insol in acids. FeO 0.9, Fe ₂ O ₃ 2.8%.
	1.585 ∧ 1.607	<u>1.598</u>	1.606	.021	MONTEBRASITE LiAlPO ₄ (OH,F)	77°	---	TCL	100 perf 110 good	White	H 5.5-6 G 3.10 F easy	Diff diss by acids. Li ₂ O 9.6, Na ₂ O 1.2, F 3.2%.
	1.595 □ 1.600	<u>1.598</u>	1.605	.019	HOWLITE Ca ₂ B ₅ SiO ₉ (OH) ₅	Large (74+12°)	X = $\frac{b}{a}$ Z:a = 55°	MCL tab 100	---	White	H 3.5 G 2.58 F 2	Insol in acids.
		<u>1.598</u>	1.602	.018	MILLISITE (Na,K)CaAl ₆ (PO ₄) ₄ (OH) ₉ ·3H ₂ O	Med	X = $\frac{c}{a}$ el neg	MCL fib $\frac{c}{a}$	---	White, gray	H 5.5 G 2.83 F 3.5	Diss by acids.
	1.582 ^	<u>1.598</u>	1.599	.010	HOPEITE Zn ₃ (PO ₄) ₂ ·4H ₂ O	37° r < v wk	X = $\frac{a}{c}$ Y = $\frac{c}{a}$	ORTH tab pris	010 perf 100 good	Cols, White	H 3.5 G 3.05-3.12 F easy	Diss by acids.

1.584 ↓ ~1.613	1.554	<u>1.599</u>	1.602	.048	TALC (Mg,Fe) ₃ Si ₄ O ₁₀ (OH) ₂	Small r > v	X ~ $\frac{c}{b}$ Z = $\frac{b}{a}$ el pos	MCL	001 perf	Green	H 1 G 2.87 infus	Insol in acids. Luster pearly. FeO 13.4, Fe ₂ O ₃ 1.7%.
1.586 ↓ 1.65	1.589	<u>1.600</u>	1.610	.021	NONTRONITE (Smectite grp) Na _{0.33} Fe ⁺³ ₂ (Si,Al) ₄ O ₁₀ (OH) ₂ ·xH ₂ O	40° r < v	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el cTv pos	MCL u mass	001 perf	Dark brown to green	H 1-2 G 2.27	Dec by acids. Pleoc in brown or green.
1.586 ↓ 1.618	1.585	(<u>1.600</u>)	1.600	.015	GLAUCONITE (Mica grp) (K,Na)(Al,Fe,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	Small r < v	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el cTv pos	MCL	001 perf	Green, blue, cols	H 2 G 2.59 fus	Dec by acids. Fe ₂ O ₃ 14.1, Al ₂ O ₃ 8.9, FeO 1.6%.
	1.518	<u>1.601</u>	1.601	.083	DRESSERITE Ba ₂ Al ₄ (CO ₃) ₄ (OH) ₈ ·3H ₂ O	30-40°	X = $\frac{a}{c}$ Z = $\frac{c}{b}$	ORTH fib	---	White	H 2.5 G 2.96 infus	Diss in acids with eff.
	1.588	<u>1.601</u>	1.610	.022	PARAUMBITE K ₃ Zr ₂ HSi ₆ O ₁₈ ·7H ₂ O	82°	X = $\frac{c}{b}$ Y = $\frac{b}{c}$	ORTH	010 mic 100 perf	Cols	H ~ 4.5 G 2.59	---
	1.595	<u>1.601</u>	1.604	.009	HURLBUTITE CaBe ₂ (PO ₄) ₂	70° r > v wk	X = $\frac{b}{c}$ Y = $\frac{c}{b}$	ORTH pris	Conch	Cols to greenish- white	H 6 G 2.88 F diff	Slowly diss by acids. Striated on 110.
	1.526	<u>1.602</u>	1.602	.076	ANTHONYITE Cu(OH,Cl) ₂ ·3H ₂ O	3°	Y = $\frac{b}{c}$ Z:c = 13°	MCL pris	100 good	Lavender	H 2 sectile	Diss by acids. Pleoc, X lavender, Y and Z deep smoky blue, abs Z = Y > X.
	1.586	<u>1.602</u>	1.606	.020	SPENCERITE Zn ₄ (PO ₄) ₂ (OH) ₂ ·3H ₂ O	49° r > v mod	Z = $\frac{b}{a}$ X ~ $\frac{a}{b}$ el pos	MCL tab 100	100 perf 010 good 001 fair	White	H 3 G 3.14 F easy	Diss by acids. Tw and comp pl 100, poly.
v 1.622	1.596	<u>1.602</u>	1.603	.007	META-AUTUNITE (Meta-autunite grp) Ca(UO ₂) ₂ (PO ₄) ₂ ·4-6H ₂ O	0-15°	Z = $\frac{c}{b}$	TET tab	---	Yellow	H 2-2.5 G 3.5 fus	Diss by acids. Pleoc. X cols, Y and Z yellow. Fluor yellow-green in UV.
1.583 ↓ 1.634	1.555	<u>1.603</u>	1.555	.048	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) ₁₀ Si ₁₂ O ₃₀ (OH) ₁₂	~ 0°	X = $\frac{b}{c}$ Y ~ $\frac{c}{b}$	MCL also TCL	001 perf	Golden- to reddish- brown	H 3-4 G 2.78 F 4-5	Dec by acids. Pleoc, X pale yellow, Y and Z olive-brown. FeO 21.6, Fe ₂ O ₃ 6.4, MnO 1.4%.
1.598 □ 1.605	1.571	<u>1.603</u>	1.606	.035	PARAGONITE (Mica grp) NaAl ₂ (Si ₃ Al)O ₁₀ (OH) ₂	30° r > v	Z = $\frac{b}{c}$ el pos	MCL ps hex plates	001 perf	Cols to green	H 2.5 G 2.87 F diff	Insol in acids.
1.598 □ 1.607	1.587	<u>1.603</u>	1.612	.025	BERTRANDITE Be ₄ Si ₂ O ₇ (OH) ₂	70-75° r < v wk	X = $\frac{a}{c}$ Z = $\frac{c}{b}$	ORTH columnar	110 perf 100, 010, 001 good	Cols	H 6.5 G 2.60 infus	Insol in acids. Tw 011.
	1.585	<u>1.603</u>	1.604	.019	WIGHTMANITE Mg ₅ (BO ₃)O(OH) ₅ ·2H ₂ O	33° r < v	Z:b = 5°	MCL ps hex pris	101 perf 101 good	Cols	H 5.5 G 2.59 infus	Diff sol in cold dil acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
242	1.596	1.603	1.606	.010	WEEKSITE $K_2(UO_2)_2Si_6O_{15} \cdot 4H_2O$	60° r > v str	X = b Y = $\frac{c}{c}$ el cTv pos	ORTH ps tet	010 perf	Pale yellow-green	H 1.5 G 4.1	Pleoc, X cols, Y pale yellow-green, Z yellow-green. Not fluor in UV.
	1.598	1.603	1.607	.009	HILLEBRANDITE $Ca_2SiO_3(OH)_2$	Med r < v str	Z = $\frac{c}{c}$ el pos	MCL fib	110	White	H 5-6 G 2.69 F diff	Gel with acids. Ultra-blue interf colors.
	1.577 ^	1.595	1.603	.009	ANTIGORITE (Serpentine grp) $(Mg,Fe)_3Si_2O_5(OH)_4$	Small	X = c el cTv pos	MCL	001 perf	Brown	H 3 G 2.4-2.6 F diff	Dec by HCl. FeO 21.0, Fe ₂ O ₃ 0.5, MnO 2.5%.
	1.582	1.604	1.609	.027	NIAHITE $(NH_4)(Mn^{+2},Mg)PO_4 \cdot H_2O$	(50±11°)	---	ORTH	---	Pale orange	Soft G 2.39	---
	1.585	1.604	1.612	.027	NAKAURIITE $Cu_8(SO_4)_4(CO_3)(OH)_6 \cdot 48H_2O$	(65±9°)	el pos	ORTH fib	---	Sky-blue	G 2.39	Pleoc, X cols, Y light greenish blue, Z pale sky-blue to light blue.
	1.598 ^	1.51	1.605	.101	AMARANTITE $FeSO_4(OH) \cdot 3H_2O$	28° r < v str	Y:c = 47° X $\sim \perp$ 100	TCL acic columnar	010, 100 perf	Red, orange-red	H 2.5 G 2.11 F 4.5-5	Dec by H ₂ O, diss by acids. ² Pleoc, X cols, Y pale yellow, Z reddish-brown.
	1.574	1.605	1.605	.031	HEINRICHITE (Autunite grp) $Ba(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$	0-20°	---	TET	001 perf 100 dist	Yellow to green	H 2.5	Diss by acids. Pleoc, X cols, Y and Z pale yellow. Fluor green to yellow in UV.
	1.580	1.605	1.605	.025	MONTDORITE (Mica grp) $(K,Na)_2(Fe^{+2},Mn,Mg)_5Si_8O_{20}(F,OH)_4$	0-3°	---	MCL	001 perf	Green to brownish-green	G 3.15	---
	1.593	1.605	1.613	.020	MAGNESIO-ANTHOPHYLLITE (Amphibole grp) $Mg_7Si_8O_{22}(OH)_2$	65° (78±12°)	Y = b Z = $\frac{c}{c}$ el pos	ORTH	210 perf	Cols to brown	H 5.5-6 G 3.0	Insol in acids. Fe ₂ O ₃ 0.25, Al ₂ O ₃ 1.6%.
	1.598	1.605	1.608	.010	BRIANITE $Na_2CaMg(PO_4)_2$	64°	---	MCL	---	Cols	H 4-5 G 3.1	Poly tw, ext angle between lam 2-3°. Meteorite mineral.
1.581 ^ 1.648	1.560	1.607	1.607	.047	VERMICULITE $(Mg,Fe,Al)_3(Si,Al)_4O_{10}(OH)_2 \cdot 4H_2O$	Small	X \perp plates el pos	MCL	001 perf	Green, brown	H 1.5 G 2.77 F 5	Dec by HCl. Exfoliates when heated. Fe ₂ O ₃ 4.5%.

1.598 ∧ 1.612	1.593	<u>1.607</u>	1.614	.021	AMBLYGONITE (Li,Na)AlPO ₄ (F,OH)	61-70° r > v	X:c = 56° Y:b = 17°	TCL	100 perf 110 good	White	H 5.5-6 G 3.05 F easy	Diff diss by acids. Li ₂ O 8.3, Na ₂ O 2.4, F 6.2%.
	1.583	<u>1.608</u>	1.633	.050	CHUDOBAITE (Mg,Zn) ₅ H ₂ (AsO ₄) ₄ · 10H ₂ O	79° (89±5°)	---	TCL	010, 100 good	Pink	H 2.5-3 G 2.94	---
	1.544	<u>1.608</u>	1.608	.064	WONESITE (Mica grp) (Na,K)(Mg,Fe) ₆ (Si,Al) ₈ O ₂₀ (OH,F) ₄	0-5°	---	MCL	001 perf	Cols to brown	---	Pleoc, X pale brown, Y and Z brown.
1.598 ∧ 1.609	1.571	<u>1.608</u>	1.611	.040	MUSCOVITE (Mica grp) K(Al,V) ₂ (Si ₃ Al) ₀ 10 (OH) ₂	34° r > v	Z = b X:c = 0-5° el pos	MCL ps hex	001 mic	Green	H 3 G 2.93 F 5	Insol in acids. Pleoc, X cols, Z pale blue. V ₂ O ₅ 3.4, Cr ₂ O ₃ 0.2, FeO 0.5, MgO 1.1%.
	1.590	<u>1.608</u>	1.611	.021	LAWSONBAUERITE (Mn,Mg) ₉ Zn ₄ (SO ₄) ₂ (OH) ₂₂ ·8H ₂ O	42° r > v str	Y = b Z:c = 7°	MCL	---	Cols to white	H 4.5 G 2.87	---
	---	<u>1.608</u>	---	.011	MONSMEDITE K ₂ O·Ti ₂ O ₃ ·8SO ₃ ·15H ₂ O	~ 52°	---	ORTH ps cub	cub 2	Dark green to black	G 3.00 H 2	---
1.608 ∧ 1.620	1.573	<u>1.609</u>	1.611	.038	MUSCOVITE var Phengite (Mica grp) K(Al,Mg) ₂ (Si ₃ Al) ₀ 10 (OH) ₂	31° r > v	Z = b X ~ c el pos	MCL ps hex	001 perf	Pale brown	H 3 G 2.88 F 5	Insol in acids. FeO 2.8, MgO 4.1, Fe ₂ O ₃ 4.1, TiO ₂ 1.0%.
	1.597	<u>1.609</u>	1.615	.018	MAGBASITE KBa(Al,Sc)(Mg,Fe) ₆ Si ₆ O ₂₀ F ₂	70°	Z:c = 10°	Acic	---	Cols, rose- violet	H 5 G 3.41	Pleoc, X and Y cols, Z lilac.
1.590 ∧	1.603	<u>1.609</u>	1.615	.012	LATIUMITE (Ca,K) ₈ (Al,Mg,Fe) (Si,Al) ₁₀ O ₂₅ (SO ₄)	72-83° r > v str	Z = b X:c = 16- 28°	MCL tab	100 perf	White	H 5-6 G 2.93 F easy	Gel with acids. Tw pl 100. Mottled extinc- tion.
	1.596	(<u>1.610</u>)	1.619	.023	UMBITE K ₂ ZrSi ₃ O ₉ ·H ₂ O	80°	---	ORTH platy	010 mic 100 perf	Cols to yellowish	H 4.5 G 2.79	Dec by HCl. Weakly fluor yellow-white in UV.
v 1.626	1.600	<u>1.610</u>	1.616	.016	SAINFELDITE Ca ₅ H ₂ (AsO ₄) ₄ ·4H ₂ O	80°	X = b Y:c = 20°	MCL flat- tened on 100	---	Cols, light pink	G 3.04	Diss by acids.
1.574 ∧	1.603	<u>1.610</u>	1.617	.014	BASSETITE (Meta-autunite grp) Fe(UO ₂) ₂ (PO ₄) ₂ ·8H ₂ O	~ 90° r > v str	X = b Y:c = 18.5° el clv pos	MCL ps tet	010 perf	Olive- green to brownish	H 2.5 G 3.4	Diss by acids. Pleoc, X and Y yellow, Z dark olive-brown. Not fluor in UV.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.607	<u>1.610</u>	1.616	.009	BUCHWALDITE NaCaPO_4	---	el pos	ORTH	One clv	Cols	H < 3 G 3.21	Biax pos (?).
1.603 ^	1.605	<u>1.61</u>	1.612	.007	HILLEBRANDITE $\text{Ca}_2\text{SiO}_3(\text{OH})_2$	51° $r < v$ str	$Z = \frac{c}{b}$ el pos	MCL fib	110	White	H 5-6 G 2.69 F diff	Gel with acids. Shows ultra-blue interf colors.
1.593 ^	1.571	<u>1.611</u>	1.612	.041	GANOPHYLLITE (Na,K)(Mn,Fe,Al) ₅ (Si,Al) ₆ O ₁₅ (OH) ₅ ·2H ₂ O	~ 0° $r < v$	$X = \frac{c}{b}$ $Z = \frac{b}{a}$ el pos	MCL	001 perf	Cinnamon-brown	H 4 G 2.8 F 3	Gel with acids. Pleoc, X dark yellow-brown, Y and Z pale yellow-brown.
1.586 ^	1.574	<u>1.611</u>	1.612	.038	BANNISTERITE (Na,K)(Mn,Fe,Al) ₅ (Si,Al) ₆ O ₁₅ (OH) ₅ ·2H ₂ O	22° $r < v$ wk	$Z \sim \frac{a}{b}$	MCL	---	Brown	G 2.84	Pleoc wk, X cols, Y and Z pale yellow.
1.609 [1.620	1.592	<u>1.612</u>	1.621	.029	HERDERITE $\text{CaBePO}_4(\text{F},\text{OH})$	67-75° $r > v$ inclined	$Y = \frac{b}{a}$ $Z:\underline{c} = -3^\circ$	MCL pris	110 poor	Cols to yellowish	H 5-5.5 G 3.01-3.08 F diff	Diss by acids. Tw pl 001 penet.
	1.585	<u>1.612</u>	1.612	.027	SODIUM URANOSPINITE (Meta-autunite grp) (Na ₂ ,Ca)(UO ₂) ₂ (AsO ₄) ₂ ·5H ₂ O	Small	---	TET tab	001 perf 010, 100 dist	Yellow-green to yellow	H 2.5 G 3.85	Diss by acids. Pleoc, X cols, Y and Z yellow. Fluor yellow-green in UV.
1.624 v	1.586	<u>1.612</u>	1.613	.027	MARGARITE Sodium analogue (Mica grp) $\text{CaAl}_2(\text{Si}_2\text{Al}_2\text{O}_{10}(\text{OH})_2$	50° $r < v$	$Z = \frac{b}{a}$ $Y:\underline{a} = 7^\circ$ el pos	MCL	001 perf	Cols	H 4 G 3.06 fus	Insol in acids. Na ₂ O 5.6, CaO 3.3%, Li not detd.
1.607 ^	1.598	<u>1.612</u>	1.621	.023	MONTEBRASITE $\text{LiAlPO}_4(\text{OH},\text{F})$	82° $r > v$	---	TCL	100 perf 110 good	White	H 5.5-6 G 3.05 F easy	Diff diss by acids. Li ₂ O 9.2, Na ₂ O 0.6, F 4.4%.
1.606 [1.615	1.596	<u>1.612</u>	1.615	.019	PHOSPHOPHYLLITE $\text{Zn}_2(\text{Fe},\text{Mn})(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	40-50° $r > v$	$Z = \frac{b}{a}$ $Y:\underline{c} = 50^\circ$	MCL thick tab	001 perf 010, 102 dist	Cols to bluish-green	H 3-3.5 G 3.13 F easy	Diss in acids. Tw on 100 common, sometimes poly. Fluor violet in short-wave UV.
	1.571	<u>1.613</u>	1.623	.052	KTENASITE (Cu,Zn) ₅ (SO ₄) ₂ (OH) ₆ ·6H ₂ O	51°	$Z = \frac{b}{a}$ $X \sim \frac{c}{b}$	MCL	---	Blue-green	H 2-2.5 G 2.97	Diss by acids and NH ₄ OH. Pleoc, X cols, Y bluish-green, Z light green.

1.599 ^ ~1.630	1.580	---	1.615	.035	MINNESOTAITE (Fe ⁺² ,Mg) ₃ Si ₄ O ₁₀ (OH) ₂	Small	el pos	MCL	001 perf	Cols to yellow- green	G 3.01	Pleoc, X pale yellow, Y and Z pale blue- green.
	1.560	<u>1.614</u>	1.615	.055	PREISWERKITE (Mica grp) NaMg ₂ Al(Al ₂ Si ₂) ₁₀ (OH) ₂	5-7°	---	MCL	001 perf	Pale green	H 2.5 G 2.96	---
	1.585	<u>1.614</u>	1.614	.029	VIMSITE CaB ₂ O ₂ (OH) ₄	28°	el pos	MCL	One clv	Cols	H 4 G 2.54	Diss by acids.
	1.586	<u>1.614</u>	1.621	.035	CAYSICHITE (Y,Ca) ₄ Si ₄ O ₁₀ (CO ₃) ₃ · 4H ₂ O	53-73°	X = $\frac{b}{a}$ Y = $\frac{a}{b}$ el pos	ORTH radi- ating	---	Cols to yellow	H 4.5 G 3.03	Diss by cold HCl with eff.
1.59 ^	1.575	(<u>1.614</u>)	1.615	.040	SAUCONITE (Smectite grp) Na _{0.33} Zn ₃ (Si,Al) ₄ O ₁₀ (OH) ₂ ·4H ₂ O	0-20°	Y = $\frac{b}{a}$	MCL mass	001 perf	Cols, white, yellow	H 2 G 2.7	Dec by acids. ZnO 23.1, Al ₂ O ₃ 17.0%.
	1.595	<u>1.614</u>	1.614	.019	SEMENTOVITE (Ca,Ce,La,Na) ₁₀₋₁₂ (Fe,Mn)(Si,Be) ₂₀ (O,OH,F) ₄₈	0-40°	---	ORTH ps tet	Uneven	Cols	H 3.5-4 G 3.14	Tw 110.
	1.600	<u>1.615</u>	1.629	.029	LEHIITE (Na,K) ₂ Ca ₅ Al ₈ (PO ₄) ₈ (OH) ₁₂ ·6H ₂ O	Large (67+8°)	el neg	MCL fib	---	White	H 5.5 G 2.89	Diss by acids.
1.593 ^ 1.632	1.600	<u>1.615</u>	1.627	.027	TREMOLITE (Actinolite ser, Amphibole grp) Ca ₂ (Mg,Fe) ₅ Si ₈ O ₂₂ (OH) ₂	79° (63+8°) r < v wk	Y = $\frac{b}{a}$ Z:c = 15° el pos	MCL	110 perf at 124°	Cols to green	H 5.5-6 G 2.98	Insol in acids. FeO 0.6%.
v (1.627)	1.605	<u>1.615</u>	1.622	.017	RICHTERITE (Amphibole grp) Na ₂ Ca(Mg,Ti,Al) ₅ Si ₈ O ₂₂ (OH) ₂	Med large r < v wk	Y = $\frac{b}{a}$ Z:c = 25° el pos	MCL	110 perf at 129°	White	H 5.5-6 G 3.07	Insol in acids.
	(1.559)	<u>1.616</u>	1.624	(.065)	PHARALUMITE Al ₂ (UO ₂) ₃ (PO ₄) ₂ (OH) ₆ · 10H ₂ O	40°	X = $\frac{b}{c}$ Y ~ $\frac{c}{b}$	MCL pris	---	Lemon- yellow	H 3 G 3.5	Pleoc, X cols, Y and Z pale yellow. Not fluor in UV.
1.624	1.597	<u>1.616</u>	1.624	.027	SANBORNITE BaSi ₂ O ₅	66°	Z ~ $\frac{c}{b}$	TCL platy	001 perf 010 imperf	White	H 5 G 4.19	Dec by acids. Poly tw on 010.
1.598 ^ 1.640	1.575	<u>1.617</u>	1.621	.046	BIOTITE, var Manganophyllite (Mica grp) K(Mg,Fe ⁺² ,Mn) ₃ (Al,Fe ⁺³)Si ₃ O ₁₀ (OH,F) ₂	30° r < v	Y = $\frac{b}{a}$ el pos	MCL plates	001 perf	Reddish- brown	H 2.5-3 G 2.98 F 3-4	Pleoc, X brown, Y and Z dark brown. Fe ₂ O ₃ 5.8, FeO 1.2, MnO 6.2%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
246	1.600	---	1.620	.020	ALDZHANITE $\text{CaMgB}_2\text{O}_4\text{Cl} \cdot 7\text{H}_2\text{O}$ (?)	---	---	ORTH dipyram	---	Cols to rose	G 2.21	Opt char unk.
	1.606	<u>1.617</u>	1.620	.014	REINHARDBRAUNSITE $\text{Ca}_5(\text{SiO}_4)_2(\text{OH},\text{F})_2$	44-50° $r > v$ dist	$Z = \frac{b}{c}$ $X:c = 18^\circ$	MCL	001 dist	Light pink	H 5-6 G 2.84	---
	(1.600) ^ 1.634	<u>1.618</u>	1.619	.020	GLAUCONITE (Mica grp) $(\text{K},\text{Na})(\text{Al},\text{Fe},\text{Mg})_2$ $(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2$	20° $r < v$	$Y = \frac{b}{c}$ $X:c = 10^\circ$	MCL	001 perf	Green	H 2 G 2.7 fus	Dec by acids. Pleoc, X yellow, Y and Z bluish-green. Fe_2O_3 18.8, Al_2O_3 8.5, FeO 4.0, MgO 3.0%.
	1.636	<u>1.619</u>	1.620	.040	GRANDIDIERITE $(\text{Mg},\text{Fe})\text{Al}_3(\text{BO}_4)(\text{SiO}_4)_2$	24° $r < v$	$X = \frac{a}{c}$ $Y = \frac{c}{b}$ el cTv pos	ORTH	100 perf 010 poor	Blue	G 2.91 infus	Insol in acids. Pleoc, X pale blue, Y yellow- brown, Z deep blue, abs $Z > X > Y$.
	1.588 ^ 1.637	<u>1.619</u>	1.619	.014	CLINOCHLORE (Chlorite grp) $(\text{Mg},\text{Fe})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}$ $(\text{OH})_8$	~ 0° $r > v$	$X \sim \frac{c}{b}$ el cTv pos	MCL	001 perf	Olive- to dark- green	H 2 G 2.8 F 5	Dec by acids. Pleoc, X pale green, Y and Z green.
		<u>1.619</u>	1.621	.010	CYMRITE $\text{BaAl}_2\text{Si}_2(\text{O},\text{OH})_8 \cdot \text{H}_2\text{O}$	0-5°	---	MCL ps hex	---	Cols, dark green, brown	H 2-3 G 3.41- 3.45	Slight sol in HCl. Exfoliates when heated.
		<u>1.62</u>	1.633	.050	Unnamed Cu-Zn arsenate	---	---	TCL	2 pina- coidal	Light rose	---	Diss by HCl. Am. Mineral., 44, 1323 (1959).
	1.609 ^	<u>1.620</u>	1.623	.043	MUSCOVITE, var Phengite (Mica grp) $\text{K}(\text{Al},\text{Fe},\text{Mg})(\text{Si}_3\text{Al})\text{O}_{10}$ $(\text{OH})_2$	35° $r > v$	$Z = \frac{b}{c}$ $X:c = 0-5^\circ$ el pos	MCL ps hex plates	001 perf	Cols to pale green	H 5 G 2.88	Insol in acids. Fe_2O_3 8.0, FeO 2.5, MgO 1.9%.
		<u>1.620</u>	1.621	.037	NISSONITE $\text{Cu}_2\text{Mg}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 5\text{H}_2\text{O}$	19° $r > v$ str	$Z = \frac{b}{c}$ $Y:c = 6^\circ$	MCL tab	100 fair	Bluish- green	H 2.5 G 2.73	Diss by acids. Pleoc, X cols, Y and Z turquoise-blue.
	1.612 ^	<u>1.620</u>	1.629	.025	HYDROXYLHERDERITE $\text{CaBePO}_4(\text{OH},\text{F})$	74° $r < v$ inclined	$Y = \frac{b}{c}$	MCL pris	110 poor	Cols	H 5-5.5 G 3.08	Diss by acids. Tw pl 001 penet.
	1.610	<u>1.620</u>	1.623	.013	SUOLUNITE $\text{Ca}_2\text{Si}_2\text{O}_5(\text{OH})_2 \cdot \text{H}_2\text{O}$	30-35° (57+19°)	$X = \frac{a}{c}$ $Y = \frac{c}{b}$	ORTH	---	White	G 2.68	Abnormal dark gray to yellowish interf colors.

1.627 v	1.612	1.621	1.622	.010	CARBONATE-FLUORAPATITE ("Dehrnite") (Apatite grp) $(Ca,Na,K)_5(PO_4,CO_3)_3F$	Small	$X = \frac{c}{el}$ $c \overline{TV}$ pos	HEX	0001 perf	Cols, yellow	H 5 G 3.04 fus	Diss by acids. Base divided into biax segments.
1.632 ┐	1.618	1.621	1.622	.004	METATORBERNITE (Meta-autunite grp) $Cu(UO_2)_2(PO_4)_2 \cdot 8H_2O$	~ 0°	$X = \frac{c}{c}$	TET square tab	001 perf	Emerald- green	H 2.5 G 3.6 F easy	Diss by acids. Anom interf colors. Not fluor in UV.
1.602 ^	1.604	1.622	1.630	.026	META-AUTUNITE (Meta-autunite grp) $Ca(UO_2)_2(PO_4)_2 \cdot 4-6H_2O$	0-15°	$Z = \frac{c}{c}$	TET tab	---	Yellow	H 2-2.5 G 3.5 fus	Diss by acids. Pleoc, X cols, Y and Z yellow. Fluor yellow-green in UV.
	1.607	1.623	1.628	.021	VLASOVITE $Na_2ZrSi_4O_{11}$	40-56° r > v dist	Opt pl 010	MCL	010 good	Cols, brownish	H 6 G 2.97	---
1.612 ┐	1.610	1.623	1.623	.013	META-URANOCIRCITE (Meta-autunite grp) $Ba(UO_2)_2(PO_4)_2 \cdot 8H_2O$	0-20° r > v	$X = \frac{c}{c}$	ORTH ps tet	001 perf 100 dist	Yellow- green	H 2-2.5 G 4.08 F 3	Diss by acids. Pleoc, X cols, Y and Z pale yellow. Poly tw. Fluor green in UV.
1.612 ^ 1.643	1.592	1.624	1.625	.033	EPHESITE (Margarite ser, Mica grp) $NaLiAl_2(Si_2Al_2)O_{10}$ (OH,F) ₂	18° r < v	$Z = \frac{b}{Y:a}$ $a = 7^\circ$ el pos	MCL plates	001 perf	Pink to brown	H 5-7 G 2.98	Insol in acids. Na_2O 7.4, Li_2O 3.5%.
	---	1.624	1.624	---	FRAIPONTITE (Serpentine grp) $(Zn,Al)_3(Si,Al)_2O_5$ (OH) ₄	15-20°	el pos	MCL fib	---	Yellow- ish-white	Soft fus	Gel with acids.
	---	1.625	1.645	---	Unnamed hydrous copper aluminum sulfate	Rather large	---	Radi- ating	---	Sky-blue	---	Pleoc str. Am. Mineral., 57, 1004 (1972).
	1.614	1.625	1.637	.023	PARAHOPEITE $Zn_3(PO_4)_2 \cdot 4H_2O$	~ 90° r < v perc	$X \sim \frac{a}{a}$	TCL	010 perf	Cols	H 3.5-4 G 3.31 F easy	Diss by acids. Poly tw common on 100.
1.605 ^ 1.630	1.615	1.625	1.634	.019	GEDRITE (Amphibole grp) $(Mg,Fe)_5Al_2(Si_6Al_2)O_{22}$ (OH) ₂	71° (86±12°)	$Y = \frac{b}{Z}$ $Z = \frac{c}{c}$ el pos	ORTH	210 perf	Cols to brown	H 5.5-6 G 2.87	Insol in acids. Al_2O_3 16.1, Fe_2O_3 0.3, FeO 1.5%.
	---	1.625	---	.007	SMOLIANINOVITE $(Co,Ni,Mg,Ca)_3(Fe^{+3},$ $Al)_2(AsO_4)_4 \cdot 11H_2O$	---	el pos	ORTH fib	---	Yellow	H 2 G 2.46	Opt sign unk.
	1.605	1.626	1.633	.028	JIMTHOMPSONITE $(Mg,Fe)_5Si_6O_{16}(OH)_2$	62° r > v wk	$X = \frac{a}{Y}$ $Y = \frac{b}{b}$	ORTH	210 perf at 38°	Cols to light pinkish- brown	---	FeO 12.2, MnO 0.7%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.610 [^]	1.616	<u>1.626</u>	1.634	.018	SAINFELDITE $\text{Ca}_5\text{H}_2(\text{AsO}_4)_4 \cdot 4\text{H}_2\text{O}$	83°	$X = \frac{b}{c}$ $Y:c = 20^\circ$	MCL flat-tened 100	---	Cols, pink	G 3.04	Diss by acids.
	1.589	<u>1.627</u>	1.628	.039	CUPRORIVAITE $\text{CaCuSi}_4\text{O}_{10}$	14°	---	TET tab	001 perf	Azure-blue	H 5 G 3.08	Insol in acids. Pleoc, X pale rose, Z blue.
	1.610	<u>1.627</u>	1.633	.023	CLINOHOLMQUISTITE (Amphibole grp) $\text{Li}_2(\text{Mg}, \text{Fe}^{+2})_3\text{Al}_2\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	58°	$Y = \frac{b}{c}$ $Z = \frac{c}{a}$ $X:a = 16^\circ$ el pos	MCL pris	110 perf	Blue	G 3.00	---
1.615 [^] 1.631	1.616	(<u>1.627</u>)	1.632	.016	RICHTERITE, var Magnophorite (Amphibole grp) $\text{Na}_2\text{Ca}(\text{Mg}, \text{Fe}^{+2}, \text{Ti})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	70°	$Z:c = 26^\circ$ el pos	MCL pris	110 perf	Cols to brown	H 5-6 G 3.12	Insol in acids. Fe_2O_3 0.6, Al_2O_3 1.7, FeO 0.6, TiO_2 3.5%.
1.640 ^v	1.619	<u>1.627</u>	1.629	.012	FEDOROVSKITE (Roweite ser) $\text{Ca}_2(\text{Mg}, \text{Mn})_2\text{B}_4\text{O}_7(\text{OH})_6$	Small $r < v$ str	$Z = \frac{b}{c}$ $Y = \frac{c}{a}$ el cTv pos	ORTH fib	100 perf	Cols to pink	H 4.5 G 2.65	Diss by acids. Pleoc, X cols, Z yellow. MnO 5%.
1.621 [^]	1.622	<u>1.627</u>	1.627	.005	CARBONATE-FLUORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4, \text{CO}_3)_3(\text{F}, \text{OH})$	0-25°	---	HEX u mass	0001 poor	Cols to brown	H 5 G 3.1-3.2	Diss by acids with slight eff.
(1.58) [^]	1.598	<u>1.628</u>	1.654	.056	LAUSENITE $\text{Fe}_2(\text{SO}_4)_3 \cdot 6\text{H}_2\text{O}$	Large (64+4°)	$X:c = 27^\circ$	MCL fib <u>c</u>	---	Cols	---	Sol in H_2O .
	1.602	<u>1.628</u>	1.632	.030	SWITZERITE $(\text{Mn}, \text{Fe})_3(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	42°	$Y = \frac{b}{c}$ $Z:c = 10^\circ$	MCL bladed	100 perf 010 fair	Pink, brown	H 2.5 G 2.98	---
1.644 ^v	1.618	<u>1.628</u>	1.631	.013	WOLLASTONITE CaSiO_3	38° $r > v$	$Y:b = 3-5^\circ$	TCL	100 perf 001, 102 good	White	H 5 G 2.92 F 4	Dec by HCl. Fe_2O_3 0.2, FeO 0.1, MnO 0.1%.
	1.585	<u>1.630</u>	1.630	.045	TROEGERITE (Autunite grp) $(\text{UO}_2)_3(\text{AsO}_4)_2 \cdot 12\text{H}_2\text{O}$	~ 0° $r > v$ mod	$X = \frac{c}{a}$ $Z:a = 13^\circ$	TET tab	001 perf 100 good	Lemon-yellow	H 2-3 G 3.3 F 2.5	Diss by acids. Not pleoc. Fluor yellow in UV.
~1.613 [^] 1.652	1.592	---	1.632	.040	MINNESOTAITE (Talc ser) $(\text{Fe}^{+2}, \text{Mg})_3\text{Si}_4\text{O}_{10}(\text{OH})_2$	Small	el pos	MCL	001 perf	Gray-green	G 3.03	Pleoc, X pale green, Y and Z bluish-green.

	1.610	(1.63)	1.639	.029	GATUMBAITE $\text{CaAl}_2(\text{PO}_4)_2(\text{OH})_2 \cdot \text{H}_2\text{O}$	65°	$Z = \underline{b}$	MCL fib	One lon- gitudinal	White	H < 5	Diss by acids. Luster pearly.
1.643 v	1.606	(1.63)	1.632	.026	CELADONITE (Mica grp) $\text{K}(\text{Mg}, \text{Fe}^{+2})(\text{Fe}^{+3}, \text{Al})\text{Si}_4\text{O}_{10}(\text{OH})_2$	Small r > v	$Y = \underline{b}$ $X \sim \underline{c}$ el cTv pos	MCL	001 perf	Green	H 2 G 2.7 F 3	Dec by HCl. Pleoc, X yellow-green, Y and Z green. Fe_2O_3 12.0, Al_2O_3 6.7, FeO 2.1, MgO 5.8%.
1.625 v 1.655	1.616	1.630	1.641	.025	ANTHOPHYLLITE (Amphibole grp) $(\text{Mg}, \text{Fe})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	88°	$Y = \underline{b}$ $Z = \underline{c}$ el pos	ORTH	210 perf	Brown	H 5.5-6 G 3.09	Insol in acids. Al_2O_3 0.2, Fe_2O_3 1.0, FeO ² 0.3 11.7%.
1.590 v 1.636	1.613	1.630	1.632	.019	CARPHOLITE $(\text{Mn}, \text{Fe})\text{Al}_2\text{Si}_2\text{O}_6(\text{OH})_4$	50° r > v str	$X = \underline{b}$ $Z = \underline{c}$ el pos	ORTH fib c	Pris at 68°	Yellow	H 5.5-6 G 2.94 F 3.5	Insol in acids. Pleoc, X and Y pale yellow, Z cols. MnO 19.9, FeO 2.3, Fe_2O_3 2.0%.
1.644 v	1.620	1.630	1.635	.015	MAGNESIO-CUMMINGTONITE (Amphibole grp) $(\text{Mg}, \text{Mn}, \text{Fe})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	74°	$Z:c = 16^\circ$ $Y = \underline{b}$ el pos	MCL	110 perf	Pink	H 6.5 G 3.07	Insol in acids. Pleoc, X and Z cols, Y yellow. MgO 25.9, MnO 8.7%.
(~1.66) v	1.613	(1.63)	1.645	.032	SODIUM BOLTWOODITE $(\text{H}_3\text{O})(\text{Na}, \text{K})(\text{UO}_2)(\text{SiO}_4) \cdot \text{H}_2\text{O}$	Large	---	ORTH	010 perf 001 imperf	Pale yellow	G 4.1	Pleoc, X cols, Z pale yellow.
1.646 v	1.608	1.631	1.642	.034	LAZULITE $(\text{Mg}, \text{Fe})\text{Al}_2(\text{PO}_4)_2(\text{OH})_2$	70° r < v perc	$Y = \underline{b}$ $X \sim \underline{c}$ el cTv neg	MCL	110 good 101	Blue	H 5-6 G 3.08 infus	Insol in HCl. Pleoc, X cols, Y and Z blue. FeO 2.6%.
(1.627) v 1.699	1.617	1.631	1.637	.020	RICHTERITE (Amphibole grp) $\text{Na}_2\text{Ca}(\text{Mg}, \text{Mn}, \text{Fe})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	68°	$Y = \underline{b}$ $Z:c = 18^\circ$ el pos	MCL	110 perf at 124°	Brown	H 5-6 G 3.08	Insol in acids. Pleoc, X and Y pale yellow, Z straw-yellow. Al_2O_3 3.0, Fe_2O_3 3.7, FeO 0.6, MnO 5.4%.
1.644 v	1.624	1.631	1.637	.013	ECKERMANNITE (Amphibole grp) $\text{Na}_3(\text{Mg}, \text{Fe})_4\text{AlSi}_8\text{O}_{22}(\text{OH})_2$	72°	$X:c = 49^\circ$	MCL	110 perf	Green	H 5-6 G 3.16	Insol in acids. Pleoc, X rose, Y lilac, Z light blue. Fe_2O_3 8.2, MnO 1.2%.
	1.620	1.631	1.633	.013	WOLLASTONITE-2M, Parawollastonite CaSiO_3	44° r > v	$Y = \underline{b}$ $X:c = 34^\circ$	MCL	---	White	---	Dec by acids.
	1.589	1.632	1.634	.045	KARLITE $\text{Mg}_7(\text{BO}_3)_3(\text{OH}, \text{Cl})_5$	24°	$X = \underline{c}$ $Y = \underline{b}$	MCL fib	001 perf	White to light green	H 5.5 G 2.82	Insol in cold dil HCl.
	1.603	1.632	1.632	.029	CARYOPILITE (Serpentine grp) $(\text{Mn}, \text{Mg})_3\text{Si}_2\text{O}_5(\text{OH})_4$	~ 0°	$X = \underline{c}$	MCL fib	001 perf	Brown	H 5.5 G 2.8-2.9 F easy	Dec by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.617	<u>1.632</u>	1.640	.023	CHESTERITE (Mg,Fe ⁺²) ₁₇ Si ₂₀ O ₅₄ (OH) ₆	71° r > v wk	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	110 perf at 45°	Cols to light pinkish-brown	---	FeO 14.1, MnO 1.0%.
1.615 ∧ 1.643	1.622	<u>1.632</u>	1.642	.020	ACTINOLITE (Amphibole grp) Ca ₂ (Mg,Fe) ₅ Si ₈ O ₂₂ (OH) ₂	80° r < v	Y = $\frac{b}{c}$ Z:c = 26° el pos	MCL	110 perf at 124°	Green	H 5.5-6 G 3.18	Insol in acids. Pleoc. FeO 7.2, Fe ₂ O ₃ 2.8, Al ₂ O ₃ 2.2%.
	---	<u>1.632</u>	1.634	---	KAHLERITE (Autunite grp) Fe(UO ₂) ₂ (AsO ₄) ₂ ·xH ₂ O	9-33°	---	TET tab	001 perf	Yellow-green	---	Diss by acids. Not fluor in UV.
	1.620	<u>1.633</u>	1.640	.020	GARRELSITE Ba ₃ NaSi ₂ B ₇ O ₁₆ (OH) ₄	72°	Z = $\frac{b}{c}$	MCL bipyram	---	Cols	G 3.8	---
1.630 └	1.619	<u>1.633</u>	1.635	.016	KINOSHITALITE (Mica grp) (Ba,K,Na)(Mg,Mn,Al) ₃ Si ₂ Al ₂ O ₁₀ (OH,F) ₂	23°	el pos	MCL	001 perf	Yellow-brown	H 2.5-3 G 3.23-3.30	Pleoc wk, X very light yellow, Y and Z yellow. BaO 17.9, K ₂ O 3.3, MnO 7.4, Mn ₂ O ₃ 3.2%.
	1.622	<u>1.633</u>	1.638	.016	LIBERITE Li ₂ BeSiO ₄	66°	Z:c = 41°	MCL	010 perf 100, 001 dist	Yellow to brown	H 7 G 2.69	---
[1.630	<u>1.633</u>	1.636	.006	DANBURITE CaB ₂ Si ₂ O ₈	88° r < v str	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH el c	001 poor	Cols to yellowish	H 7 G 2.94 F diff	Insol in acids.
∨ 1.685	1.59	<u>~1.63</u>	~1.64	~.05	ROSCOELITE (Mica grp) K(V,Al,Mg) ₂ AlSi ₃ O ₁₀ (OH) ₂	Med r < v	Z = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el pos	MCL	001 perf	Brownish-green	H 3 G 2.8 F 3	Insol in acids. V ₂ O ₃ 17.4, Al ₂ O ₃ 21.9, FeO 1.7%.
1.603 ∧ 1.661	1.575	<u>1.634</u>	1.634	.059	STILPNOMELANE K(Fe ⁺² ,Mg,Fe ⁺³ ,Al) ₁₀ Si ₁₂ O ₃₀ (OH) ₁₂	~ 0°	X = $\frac{b}{c}$ Y ~ $\frac{c}{b}$	MCL also tric	001 perf	Brown	H 3-4 G 2.84 F 4-5	Dec by acids. Pleoc, X pale brown, Y and Z dark brown. FeO 22.7, Fe ₂ O ₃ 12.2%.
∨ 1.645	1.616	<u>1.634</u>	1.646	.030	HOLMQUISTITE (Amphibole grp) Li ₂ (Mg,Fe ⁺²) ₃ Al ₂ Si ₈ O ₂₂ (OH) ₂	56° (78+8°) r > v wk	Z = $\frac{c}{b}$ el pos	ORTH pris	210 perf	Violet	H 5.5-6 G 3.04	Pleoc, X and Y cols, Z lilac. FeO 7.3, Fe ₂ O ₃ 1.0, Al ₂ O ₃ 14.3, Li ₂ O 3.2%.

1.618 ^	1.610	1.634	1.634	.024	GLAUCONITE (Mica grp) (K,Na)(Fe ⁺³ ,Al,Mg, Fe ⁺²) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	10° r < v	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$	MCL	001 perf	Green	H 2 G 2.74 fus	Dec by acids. Pleoc, X yellow-green, Y and Z blue-green. Fe ₂ O ₃ 24.9, Al ₂ O ₃ 7.3, FeO 2.9%.
	1.560	1.635	1.635	.075	GILALITE Cu ₅ Si ₆ O ₁₇ ·7H ₂ O	Small	---	MCL fib	---	Green	H 2 G 2.82	---
	1.615	1.635	1.656	.041	SKLODOWSKITE Mg(UO ₂) ₂ Si ₂ O ₆ (OH) ₂ · 5H ₂ O	~ 90° r > v str	Y = $\frac{b}{c}$ = el	MCL pris $\frac{b}{c}$ tab spher- ulitic	100 perf	Yellow	H 2-3 G 3.64	Pleoc, X cols, Y yellow, Z pale yellow. Not fluor in UV.
1.632 └	1.617	1.635	1.652	.035	TILLEYITE Ca ₅ Si ₂ O ₇ (CO ₃) ₂	92° r > v perc	Y = $\frac{b}{c}$ X: $\frac{a}{b}$ = 18°	MCL	100 perf	White	G 2.84	Eff and gel with acids.
	1.611	1.635	1.643	.032	BURANGAITE (Na,Ca) ₂ (Fe,Mg) ₂ Al ₁₀ (PO ₄) ₈ (OH,O) ₁₂	58° r > v	Z = $\frac{b}{c}$ X: $\frac{c}{b}$ = 11°	MCL pris	100 perf	Bluish to bluish- green	H 5 G 3.05	Pleoc, X light blue, Y dark blue, Z cols. Fe ₂ O ₃ 1.1, FeO 6.3%.
1.651 v	1.624	1.635	1.654	.030	BAKERITE Ca ₄ B ₄ (BO ₄)(SiO ₄) ₃ (OH) ₃ ·H ₂ O	85° (105+8°)	Y = $\frac{b}{c}$ X: $\frac{c}{b}$ = 44°	MCL pris tab spher- ulitic	---	White	H 4.5 G 2.88	---
	1.615	1.635	1.638	.023	METALODEVITE (Meta-autunite grp) Zn(UO ₂) ₂ (AsO ₄) ₂ ·10H ₂ O	27-37°	---	ORTH tab	---	Pale yellow to olive	G 4.00	Weakly fluor yellow- green in UV.
1.630 □ 1.640	1.619	1.635	1.642	.023	NORDITE (La,Ce)(Sn,Ca)(Na,Mn) ₃ (Zn,Mg)Si ₆ O ₁₇	31° (66+10°)	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	100 good	Light brown	H 5-6 G 3.43 F easy	Dec by acids.
1.644 v	1.627	1.635	1.639	.012	ANDALUSITE Al ₂ SiO ₅	82° r < v	X = $\frac{c}{b}$ Y = $\frac{b}{c}$ el neg	ORTH	110 good 100 poor	Pink, white	H 6-6.5 G 3.14 infus	Insol in acids. Pleoc wk, X rose, Y and Z greenish-yellow. Fe ₂ O ₃ 0.8%.
	1.560	1.635	1.635	.075	GILALITE Cu ₅ Si ₆ O ₁₇ ·7H ₂ O	Small	---	MCL	---	Green	H 2 G 2.72	---
~1.650 v	1.596	(1.636)	1.637	.041	JOLIOTITE (UO ₂)(CO ₃)·xH ₂ O (x < 2)	Small	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH crusts	100	Lemon- yellow	H 1-2 G 4.55 calc	Pleoc, X cols, Z yellow. Fluor wk in UV.
1.619 ^	1.602	1.636	1.639	.037	GRANDIDIERITE (Mg,Fe)Al ₃ (BO ₄)(SiO ₄) O ₂	30° r < v str	X = $\frac{a}{c}$ Y = $\frac{c}{b}$ el cTv pos	ORTH	100 perf 010 poor	Blue- green	G 2.99 infus	Insol in acids. Pleoc, X greenish-blue, Y cols, Z pale bluish- green.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.630 \wedge 1.644	1.621	1.636	1.640	.019	FERROCARPHOLITE (Fe,Mg)Al ₂ Si ₂ O ₆ (OH) ₄	56° r > v str	X = $\frac{b}{c}$ Z = $\frac{c}{a}$ el pos	ORTH fib \underline{c}	Pris at 68°	Yellow to green	H 5.5-6 G 3.0 F 3	Insol in acids. Pleoc, X and Y yellow-green, Z blue-green. FeO 15.5, MnO 0.1, Fe ₂ O ₃ 3.9%.
	1.624	1.636	1.642	.018	BERTOSSAITE (Li,Na) ₂ CaAl ₄ (PO ₄) ₄ (OH,F) ₄	Med r < v str	X = $\frac{a}{c}$ Y = $\frac{c}{b}$	ORTH	100 good	Pink	H 6 G 3.10	Slowly diss by HNO ₃ .
	1.584	1.637	1.670	.086	USHKOVITE MgFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	50°	ext at 26°	TCL	010 good	Light yellow	G 3.38	Diss by acids.
	1.610	1.637	1.652	.042	FABIANITE CaB ₃ O ₅ (OH)	66° r < v wk	Y = $\frac{b}{a}$ X:a = 22°	MCL pris	110	Cols	H 6 G 2.77	Diss by acids. Fluor brownish-yellow in UV.
252 1.619 \wedge 1.649	1.615	1.637	1.638	.023	CHAMOSITE (Chlorite grp) (Mg,Fe ⁺² ,Fe ⁺³) ₅ Al(Si ₃ Al) ₁₀ (OH,0) ₈	15°	X \perp \underline{c}	MCL	001 perf	Green	H 2.5 G 3.0 F diff	Dec by acids. Pleoc, X olive-green, Y and Z brownish-yellow. FeO 22.8, Fe ₂ O ₃ 9.1, Al ₂ O ₃ 22.1%.
	1.620	1.637	1.642	.022	GLAUCOPHANE (Amphibole grp) Na ₂ (Mg,Fe) ₃ Al ₂ Si ₈ O ₂₂ (OH) ₂	51° r < v str	Y = $\frac{b}{c}$ Z:c = 5° el pos	MCL pris	110 perf at 121°	Blue	H 6 G 3.13 F 3.5	Insol in acids. Pleoc, X cols, Y lavender, Z blue. FeO 10.6, Fe ₂ O ₃ 2.0, Al ₂ O ₃ 10.1%.
1.568 \wedge	1.623	---	1.640	.017	NEPOUITE (Serpentine grp) (Ni,Mg) ₃ Si ₂ O ₅ (OH) ₄	---	---	MCL fib	001 perf	Dark green	H 6 G 3.2 infus	Insol in acids. Pleoc wk, X dark green, Z yellow-green. NiO 43.7, MgO 8.2%.
	1.596	1.639	1.670	.074	SUANITE Mg ₂ B ₂ O ₅	70° (79+3°) r > v wk	X = $\frac{b}{c}$ Y:c = 23°	MCL fib	010	White	H 5.5 G 2.91 F diff	Slowly diss by HCl.
1.636 \square 1.642	1.618	1.639	1.653	.035	INESITE Ca ₂ Mn ₇ Si ₁₀ O ₂₈ (OH) ₂ ·5H ₂ O	75° r > v wk	Ext from clv on 010 = 29° 001 = 14°	TCL pris	010 perf 100 good	Rose-red to flesh	H 5.5 G 3.02 F 3	Dec by HCl with sepn of silica.
	1.619	1.639	1.643	.024	TROLLEITE Al ₄ (PO ₄) ₃ (OH) ₃	49° r > v wk	---	MCL mass	---	Pale green	H 5.5 G 3.09 infus	Nearly insol in acids. Fe 2%.

1.644	1.624	1.639	1.643	.019	ROSCHERITE $\text{Ca(Al,Fe,Mn)}_3\text{Be}_2(\text{PO}_4)_3$ $(\text{OH})_3 \cdot 2\text{H}_2\text{O}$ (?)	Large (54+14°) $r > v$ str crossed	$X = \frac{b}{c} = -15^\circ$ $Y: \underline{c} =$	MCL pris	001 good 010 fair	Dark brown	H 4.5 G 2.93 fus	Diss by acids. Pleoc, X yellow to olive, Y yellow-brown, Z chest- nut brown. Abnormal interf colors.
1.617 ↕ 1.652	1.590	1.640	1.640	.050	SIDEROPHYLLITE (Biotite ser, Mica grp) $\text{KFe}_2\text{Al}(\text{Al}_2\text{Si}_2)_0^{10}$ $(\text{F,OH})_2$	Small $r < v$	$Y = \frac{b}{el}$ $el \text{ pos}$	MCL plates	001 perf	Dark brown	H 2.5-3 G 3.12 F 4	Pleoc, X brown, Y and Z dark brown. FeO 30.2, MnO 1.0%.
	1.624	1.640	1.650	.026	OURSINITE $(\text{Co,Mg})(\text{UO}_2)_2\text{Si}_2\text{O}_7 \cdot$ $6\text{H}_2\text{O}$	(76+9°)	$Y = \underline{c}$	ORTH	---	Pale yellow	G (3.67)	Does not fluor UV.
	1.632	---	1.646	.014	ZYKAITE $\text{Fe}^{+3}_4(\text{AsO}_4)_3\text{SO}_4(\text{OH}) \cdot$ $15\text{H}_2\text{O}$	Large	$el \text{ pos}$	ORTH	Uneven	Gray- white, yellow- green	G 2.50	---
1.627 ↕ 1.660	1.631	1.640	1.641	.010	ROWEITE $\text{Ca}_2(\text{Mn,Mg})_2\text{B}_4\text{O}_7(\text{OH})_6$	20° $r < v$ str	$Z = \frac{b}{c}$ $Y = \underline{c}$	ORTH tab	101 poor	Light brown	H 5 G 2.73 F 2	Diss by acids. MnO 17.9, MgO 7.6, FeO 1.2%.
	1.633	1.640	1.645	.012	Unnamed Ca_2SiO_4	57°	---	ORTH fib	$clv \perp el$	White	G 2.97 calc infus	Gel with acids. Am. Mineral., 51, 1771 (1966).
	---	1.64	---	wk	NINGYOITE $(\text{U,Ca,Ce})_2(\text{PO}_4)_2 \cdot$ $1-2\text{H}_2\text{O}$	---	$el \text{ pos}$	ORTH ps hex	---	Brown to brownish- green	---	Pleoc wk in brown. Opt char unk.
	1.583	1.641	1.648	.065	SERPIERITE $\text{Ca}(\text{Cu,Zn})_4(\text{SO}_4)_2(\text{OH})_6 \cdot$ $3\text{H}_2\text{O}$	(37+6) $r > v$ str	$Y = \frac{b}{a}$ $X: \underline{a} = 24^\circ$	MCL laths	100 perf	Sky-blue	G 3.07	Diss by acids. Pleoc, X pale green, Y and Z bluish-green.
	1.607	1.641	1.672	.065	PAPAGOITE $\text{CaCuAlSi}_2\text{O}_6(\text{OH})_3$	78° (86+4°) $r > v$ wk	$Z = \frac{b}{c}$ $X: \underline{c} = 44^\circ$	MCL tab	100 good	Blue	H 5-5.5 G 3.25	Pleoc, X cols to pale greenish-blue, Y blue, Z deep greenish-blue.
	1.609	1.641	1.641	.032	METAHEINRICHITE (Meta-autunite grp) $\text{Ba}(\text{UO}_2)_2(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	0-20°	---	TET tab	001 perf 100 dist	Yellow to green	H 2.5 G 4.04	Diss by acids. Pleoc, X cols, Y and Z pale yellow. Fluor yellow- green in UV.
	1.624	1.641	1.647	.023	ROSENHAHNITE $\text{Ca}_3\text{Si}_3\text{O}_8(\text{OH})_2$	60-68°	On 100 $Z': \underline{c} = 36^\circ$	TCL laths	001 perf 100, 010 good	Cols	H 4.5-5 G 2.89 F 3	Slightly attacked by acids.
1.646	1.635	(1.642)	1.642	.007	KILLALAITITE $2\text{Ca}_3\text{Si}_2\text{O}_7 \cdot \text{H}_2\text{O}$	26°	$Y = \frac{b}{c}$ $Z: \underline{c} = 16^\circ$	MCL	100 perf 010, 001 good	Cols	G 2.88 calc	Complex penet tw.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
$\sqrt{1.653}$	1.618	<u>1.642</u>	1.652	.034	SOUZALITE (Mg,Fe) ₃ (Al,Fe) ₄ (PO ₄) ₄ (OH) ₆ ·2H ₂ O	68° r > v str	X = $\frac{a}{b}$ el pos	MCL (?) fib	100 perf	Green	H 5.5-6 G 3.09 infus	Diff diss by HCl. Poly tw. Pleoc, X green, Y blue, Z yellow.
$\sqrt{1.645}$	1.633	<u>1.642</u>	1.652	.019	MAGNESIO-HORNBLLENDE (Amphibole grp) (Ca,Na) ₂ (Mg,Fe) ₄ Al (Si ₇ Al) ₀ 22(OH,F) ₂	88°	Y = $\frac{b}{c}$ Z:c = 16° el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.10	Insol in acids. Pleoc, X pale yellow, Y and Z green. FeO 8.7, Fe ₂ O ₃ 2.7, Al ₂ O ₃ 5.0%.
	1.627	<u>1.642</u>	1.644	.017	PALERMOITE (Li,Na) ₂ (Sr,Ca)Al ₄ (PO ₄) ₄ (OH) ₄	~ 20° r < v mod	Y = $\frac{a}{c}$ X = $\frac{c}{b}$ el neg	ORTH pris, el c	100 perf 001 fair	Cols, white	H 5.5 G 3.22	---
	1.555	<u>1.643</u>	1.658	.103	SIBIRSKITE CaHBO ₃	43°	---	MCL	---	Cols	---	Diss by acids.
$\sqrt{1.650}$	1.559	<u>1.643</u>	1.655	.096	HOHMANNITE Fe ₂ (SO ₄) ₂ (OH) ₂ ·7H ₂ O	40° r > v str	Y:c = 23°	TCL pris	010 perf	Amaranth-red to brown	H 3 G 2.2	Diss by acids. Pleoc, X pale yellow, Y pale greenish-yellow, Z dark greenish-brown.
(1.63) ↓ (1.662)	1.612	<u>1.643</u>	1.643	.031	CELADONITE (Mica grp) K(Mg,Fe ⁺²)(Fe ⁺³ ,Al)Si ₄ O ₁₀ (OH) ₂	Small r > v	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el cTv pos	MCL	001 perf	Bright green	H 2 G 2.8 F 3	Dec by HCl. Pleoc, X yellow-green, Y and Z green. Fe ₂ O ₃ 29.3, Al ₂ O ₃ 1.0, FeO 9.8, MgO 3.7%.
$\sqrt{1.632}$ ↓ 1.663	1.629	<u>1.643</u>	1.650	.021	ACTINOLITE (Amphibole grp) Ca ₂ (Mg,Fe) ₅ Si ₈ O ₂₂ (OH) ₂	73° r < v	Y = $\frac{b}{c}$ Z:c ~ 20° el pos	MCL pris	110 perf at 124°	Green	H 5.5-6 G 3.17	Insol in acids. Pleoc. FeO 8.3, Fe ₂ O ₃ 2.4%.
$\sqrt{1.624}$ ↓ 1.652	1.632	<u>1.643</u>	1.646	.014	MARGARITE (Mica grp) CaAl ₂ (Al ₂ Si ₂)O ₁₀ (OH) ₂	63° r < v	Z = $\frac{b}{c}$ Y:c = 7°	MCL	001 perf	Cols	H 4 G 3.06 F diff	BeO 1.8 to 3.3%.
	1.637	<u>1.643</u>	1.648	.011	KOASHVITE Na ₆ (Ca,Mn)(Ti,Fe)Si ₆ O ₁₈ ·H ₂ O	83° r > v wk	---	ORTH	Conch	Pale yellow	H 6 G 3.00 F 3	---
$\sqrt{1.658}$	1.639	<u>1.643</u>	1.646	.007	HIORTDAHLITE (Ca,Na) ₃ ZrSi ₂ O ₇ (O,OH,F) ₂	83° r < v	Ext on 100 = 65°	TCL tab	Pris at about 90°	Yellow, brown	H 5.5 G 3.26 F 3	Gel with acids. Poly tw, comp pl 100. Pleoc wk, X cols, Y and Z yellow.

1.630 ↓ 1.665	1.630	<u>1.644</u>	1.652	.022	TIRODITE (Cummingtonite ser, Amphibole grp) $Mn_2(Mg,Fe^{+2})_5Si_8O_{22}(OH)_2$	73°	$Z:c = 20^\circ$ $Y = \frac{b}{a}$ el pos	MCL pris	110 perf	Pink to greenish	H 5.5-6 G 3.10	Insol in acids. MnO 16.6, FeO 4.5%.
1.628 ↓ (1.652)	1.631	<u>1.644</u>	1.646	.015	WOLLASTONITE (Ca,Mn)SiO ₃	45° r > v	$Y \sim \frac{b}{a}$ $X:c \sim 40^\circ$	TCL	100 perf 001, 102 good	White	H 5 G 3.03 F 4	Dec by HCl. MnO 6.3, FeO 0.7%.
1.635 ^	1.637	<u>1.644</u>	1.650	.013	ANDALUSITE Al ₂ SiO ₅	73° r < v	$X = \frac{c}{a}$ $Y = \frac{b}{a}$ el neg	ORTH	110 good 100 poor	White, pink	H 6-6.5 G 3.16 infus	Insol in acids. Pleoc wk, X rose, Y and Z yellow.
1.631 ↓ 1.656	1.636	<u>1.644</u>	1.649	.013	ECKERMANNITE (Arfvedsonite ser, Amphibole grp) $Na_3(Mg,Fe)_4AlSi_8O_{22}(OH)_2$	74°	$X:c = 25^\circ$ el neg	MCL pris	110 good	Bluish- green	H 6 G 3.16	Insol in acids. Pleoc, X bluish-green, Y bright blue-green, Z yellow-green. Na ₂ O 9.8, FeO 2.7, Fe ₂ O ₃ 8.0, MgO 9.1%.
1.653 v	1.637	<u>1.644</u>	1.645	.008	JEREMEJEVITE Al ₆ B ₅ O ₁₅ (F,OH) ₃	18° r > v dist	$X = \frac{c}{a}$	HEX pris	Conch	Cols to brown	H 6.5 G 3.28 infus	Insol in acids.
	1.589	<u>1.645</u>	1.659)	.056	CAMPIGLIAITE Cu ₄ Mn(SO ₄) ₂ (OH) ₆ ·4H ₂ O	52° r < v wk	$Z = \frac{b}{a}$ $X \sim \frac{a}{b}$	MCL el b	---	Light blue.	---	Tw 100.
1.634 ^	1.624	<u>1.645</u>	1.651	.027	HOLMQUISTITE (Amphibole grp) $Li_2(Mg,Fe^{+2})_3Al_2Si_8O_{22}(OH)_2$	50° r > v wk	$Z = \frac{c}{a}$ el pos	ORTH pris	210 perf	Blue	H 5.5-6 G 3.06	Pleoc, X grayish- yellow, Y pink-violet, Z dark bluish-violet. FeO 8.9, Fe ₂ O ₃ 2.2, Al ₂ O ₃ 13.0, Li ₂ O 2.4%.
1.642 ↓ 1.664	1.634	<u>1.645</u>	1.658	.024	EDENITE (Hornblende ser, Amphibole grp) $NaCa_2(Mg,Fe)_5Si_7AlO_{22}(OH)_2$	74° (94+10°)	$Y = \frac{b}{a}$ el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.10	Insol in acids. Pleoc. FeO 9.0, Fe ₂ O ₃ 3.4, MnO 0.3%.
1.655 v	1.637	<u>1.645</u>	1.649	.012	HUREAULITE $Mn_5(PO_4)[PO_3(OH)]_2 \cdot 4H_2O$	75° r < v very str	$X = \frac{b}{a}$ $Z:c = 65^\circ$	MCL tab	100 good	Pink to brown	H 3.5 G 3.19 F 2	Diss by acids. Pleoc, X cols, Y pale pink to yellow, Z pink to brown.
	1.556	<u>1.646</u>	1.652	.096	DONNAYITE $Sr_3CaNaY(CO_3)_6 \cdot 3H_2O$	0-20° u 5-10°	---	TCL	001 indist to fair	Yellow to cols	H 3 G 3.35	Diss by acids with eff.
	1.558	<u>1.646</u>	1.648	.090	WELOGANITE $Sr_3Na_2Zr(CO_3)_6 \cdot 3H_2O$	0-15°	---	TCL ps hex el c	001 perf fr conch	Lemon- yellow to amber	H 3.5 G 3.20	Diss by acids with eff.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
v (1.699)	1.575	<u>1.646</u>	1.650	.075	SZAIBELYITE (Sussexite ser) $\text{MgBO}_2(\text{OH})$	$\sim 25^\circ$ $r > v$	---	MCL fib	110 perf	White	H 3 G 2.65 F 3	Diss by acids. Poly tw.
1.631 $\hat{\vee}$ 1.674	1.620	<u>1.646</u>	1.656	.036	LAZULITE $(\text{Mg,Fe})\text{Al}_2(\text{PO}_4)_2(\text{OH})_2$	67° $r < v$ perc	$Y = \frac{b}{c}$ $X \sim \frac{c}{c}$	MCL	110 good 101	Blue	H 5-6 G 3.19 infus	Insol in acids. Pleoc, X cols, Y and Z blue. FeO 8.9, Fe_2O_3 0.9%.
	1.640	(<u>1.646</u>)	1.651	.011	RUSTUMITE $\text{Ca}_{10}(\text{Si}_2\text{O}_7)_2\text{SiO}_4\text{Cl}_2$ $(\text{OH})_2$	80°	$Y = \frac{b}{c}$ $X:c = 5^\circ$	MCL tab	100 poor	Cols	G 2.9	Lam tw on 100.
	1.639	<u>1.646</u>	1.646	.007	KELLYITE (Kaolinite-Serpentine grp) $(\text{Mn,Mg,Al})_3(\text{Si,Al})_2\text{O}_5$ $(\text{OH})_4$	$5-30^\circ$ $r > v$ mod	$X = \frac{c}{c}$ el cTv pos	TRIG and HEX	0001 perf	Yellow	G 3.07	Pleoc, X cols to greenish-yellow, Y and Z pale yellow to reddish-brown.
	1.637	<u>1.647</u>	1.647	.010	NIMITE (Chlorite grp) $(\text{Ni,Mg,Fe})_5\text{Al}(\text{Si}_3\text{Al})$ $\text{O}_{10}(\text{OH})_8$	15°	---	MCL	001 good	Yellow-green	H 3 G 3.16	Pleoc wk, X greenish-yellow, Z apple green. NiO 29.5, FeO 2.8, Fe_2O_3 4.4%.
1.607 \wedge	1.622	<u>1.648</u>	1.651	.029	VERMICULITE $(\text{Mg,Ni,Fe,Al})_3(\text{Si,Al})_4$ $\text{O}_{10}(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	Small	$X \perp$ plates el pos	MCL	001 perf	Brownish-black	H 1.5 G 2.8 F 5	Dec by HCl. Exfoliates when heated. Fe_2O_3 19.2, FeO 5.0, MgO 13.9, NiO 8.6%.
1.637 $\hat{\vee}$ 1.659	1.634	<u>1.648</u>	1.653	.019	CROSSITE (Glaucophane ser, Amphibole grp) $\text{Na}_2(\text{Mg,Fe}^{+2})_3$ $(\text{Al,Fe}^{+3})_2\text{Si}_8\text{O}_{22}(\text{OH})_2$	Small ($61 \pm 13^\circ$)	$X:c = 8^\circ$ $Y = \frac{b}{c}$ el neg	MCL pris	110 perf	Blue	H 5 G 3.14	Insol in acids. Pleoc, X yellowish, Y lavender, Z blue. FeO 7.5, Fe_2O_3 6.4, Al_2O_3 8.4%.
	1.646	<u>1.648</u>	1.650	.004	KILCHOANITE $\text{Ca}_3\text{Si}_2\text{O}_7$	$46-54^\circ$ $r > v$ str	---	ORTH	---	Cols	G 2.99	Gel with acids. Abnormal ultrablue and brown interf colors.
	1.585	<u>1.649</u>	1.660	.075	DEVILLINE $\text{CaCu}_4(\text{SO}_4)_2(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	39° ($44 \pm 4^\circ$)	$Z = \frac{b}{c}$ $Y \sim \frac{c}{c}$	MCL tab	001 perf 110, 101 dist	Emerald-to bluish-green	H 2.5 G 3.13 F 3.5	Diss by acids. Tw 010. Pleoc, X pale green, Y Venice green, Z turquoise.

1.667 v	1.627	1.649	1.660	.033	URANOPHANE $\text{Ca}(\text{UO}_2)_2\text{Si}_2\text{O}_7 \cdot 6\text{H}_2\text{O}$	60° $r > v$ (U r < v)	$Z = \frac{b}{a}$	MCL acid	100 perf	Lemon- to pale- yellow	H 2.5 G 3.3 infus	Gel with acids. Pleoc, wk in yellows. Ab- normal blue interf colors.
1.664 v	1.641	1.649	1.655	.014	MONTICELLITE $\text{Ca}(\text{Mg,Fe})\text{SiO}_4$	80° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010 poor	White, gray	H 5.5 G 3.06 F 6	Gel with acids. FeO 1.4%.
1.667 v	1.642	1.649	1.649	.007	CHLORAPATITE (Apatite grp) $\text{Ca}_5(\text{PO}_4)_3(\text{Cl,F,OH})$	Small 5-10°	---	MCL ps hex	001 imperf	Cols	H 5 G 3.18 F 5	Diss by acids. Cl 1.31, F 0.45, H ₂ O 0.7%.
1.637 v 1.666 ^	1.643	1.649	1.649	.006	CHAMOSITE (Chlorite grp) $(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3})_5\text{Al}$ $(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH},\text{O})_8$	~ 0°	$X = \frac{c}{a}$ el pos	MCL	001 perf	Dark green	H 3 G 3.08	Dec by HCl. Pleoc, X pale yellow, Y and Z olive-green.
1.646 v	1.602	1.650	1.654	.052	VITUSITE $\text{Na}_3(\text{Ce,Ln})(\text{PO}_4)_2$	30°	---	ORTH	---	Pale pink, gray	H 4.5 G 3.60- 3.70	Diss by HCl. Tw.
	1.610	1.650	1.682	.072	EPISTOLITE $\text{Na}_2(\text{Nb,Ti})_2\text{Si}_2\text{O}_9 \cdot x\text{H}_2\text{O}$	80° $r < v$ perc	$Y = \frac{b}{a}$ $Z:c = \frac{b}{a} 7^\circ$	TCL plates	001 perf 110 dist	White, gray, yellow	H 1-1.5 G 2.89	---
1.675 v	1.640	(1.650)	1.653	.013	FERROBUSTAMITE $\text{Ca}(\text{Fe}^{+2}, \text{Ca}, \text{Mn})\text{Si}_2\text{O}_6$	60°	$X:c = 44^\circ$	TCL	100 perf 110, 110 good	Pink to brown	H 6 G 3.09	FeO 15.4, MnO 1.6%.
(1.636) v	1.604	---	1.651	.047	JOLIOTITE $(\text{UO}_2)(\text{CO}_3) \cdot x\text{H}_2\text{O}$ (x = 2?)	Small	$X = \frac{a}{b}$ $Y = \frac{b}{a}$	ORTH crusts	100	Lemon- yellow	H 1-2 G 4.55 calc	Pleoc, X cols, Z yellow. Fluor wk in UV.
1.600 v	1.625	1.65	1.655	.030	NONTRONITE (Smectite grp) $\text{Na}_{0.33}\text{Fe}^{+3}_2(\text{Si,Al})_4\text{O}_{10}$ $(\text{OH})_2 \cdot x\text{H}_2\text{O}$	33° $r < v$	$Y = \frac{b}{a}$ $X \sim \frac{c}{a}$	MCL u mass	001 perf	Green to yellow	H 1-2 G 2.27	Dec by acids. Pleoc, X nearly cols, Y and Z yellow to greenish- yellow.
1.664 v	1.629	1.650	1.658	.029	EOSPHORITE (Childrenite ser) $(\text{Mn,Fe})\text{AlPO}_4(\text{OH})_2 \cdot \text{H}_2\text{O}$	45° (63+8°) $r < v$	$X = \frac{b}{a}$ $Z:c = 5^\circ$	MCL ps orth	001 poor	Pink to brown	H 5 G 3.10	Diss by acids. MnO 29.9, FeO 1.4%.
	1.623	1.650	1.651	.028	FRIEDELITE $(\text{Mn,Fe})_8\text{Si}_6\text{O}_{15}$ $(\text{OH,Cl})_{10}$	Small	el clv pos	HEX	0001 perf	Rose-red	H 4.5-5 G 3.07 F 4	Dec by HCl. Pleoc, X cols, Y and Z greenish- yellow.
	1.610	1.650	1.650	.040	APACHITE $\text{Cu}_9\text{Si}_{10}\text{O}_{28} \cdot 11\text{H}_2\text{O}$	Small	---	MCL	---	Blue	H 2 G 2.80	---
	1.624	1.650	1.650	.026	BEMENTITE $\text{Mn}_8\text{Si}_6\text{O}_{15}(\text{OH})_{10}$	~ 0°	$X = \frac{c}{a}$ el cTv pos	MCL	001 perf	Brown	H 5 G 2.98 F 3	Dec by acids. Pleoc, X nearly cols, Y and Z yellow.

258

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.590	<u>1.651</u>	1.657	.067	BORCARITE $\text{Ca}_4\text{MgB}_4\text{O}_{16}(\text{OH})_6(\text{CO}_3)_2$	30° r < v	Z = $\frac{b}{c}$ = 28°	MCL	100 perf	Cols, greenish-blue	H 4 G 2.77	Diss by acids.
1.635 ^	1.625	<u>1.651</u>	1.664	.039	BAKERITE $\text{Ca}_4\text{B}_4(\text{BO}_4)(\text{SiO}_4)_3(\text{OH})_3 \cdot \text{H}_2\text{O}$	67°	Y = $\frac{b}{c}$ X:c = 44°	MCL pris tab spherulitic	---	White	H 4-5 G 2.88	---
	1.618	<u>1.652</u>	1.682	.064	SCHOONERITE $\text{Fe}^{+2}_2\text{MnZnFe}^{+3}(\text{PO}_4)_3(\text{OH})_2 \cdot 9\text{H}_2\text{O}$	70-80° (85±3°)	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH tab 010 el 100	010 perf 001 good	Brown to red-brown	H 4 G 2.89	Diss by cold acids. Pleoc, X yellow, Y pale brown, Z brown.
	1.612	<u>1.652</u>	1.675	.063	LIROCONITE $\text{Cu}_2\text{AlAsO}_4(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	72° r < v mod	Y = $\frac{b}{a}$ Z:a = 25°	MCL pris	110, 011 poor fr conch	Blue, green	H 2.5 G 2.96 F easy	Diss by acids.
1.640 ^ 1.672	1.595	<u>1.652</u>	1.656	.061	BIOTITE (Mica grp) $\text{K}(\text{Fe}, \text{Mg})_3(\text{Al}, \text{Fe}^{+3})\text{Si}_3\text{O}_{10}(\text{OH}, \text{F})_2$	Small r < v	Y = $\frac{b}{a}$ el pos	MCL plates	001 perf	Dark brown	H 2.5-3 G 3.05 F 4	Pleoc, X brown, Y and Z dark brown. FeO 19.9, MnO 0.4, Fe ₂ O ₃ 3.2%.
1.599 ^	1.600	<u>1.652</u>	(1.655)	(.055)	WILLEMSEITE (Talc ser) $(\text{Ni}, \text{Mg})_3\text{Si}_4\text{O}_{10}(\text{OH})_2$	27°	---	MCL	001 perf	Light to apple-green	H 2 G 3.28	Pleoc, X yellowish-green, Z bluish-green. NiO 34.5, MgO 7.1%.
1.651 pos ^ 1.664	1.642	<u>1.652</u>	1.660	.018	PARGASITE (Hastingsite ser, Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{Al}(\text{Si}_6\text{Al}_2)\text{O}_{22}(\text{OH})_2$	88°	Y = $\frac{b}{c}$ Z:c = 28° el pos	MCL pris	110 perf at 124°	Brown	H 5-6 G 3.14	Insol in acids. Pleoc. Al ₂ O ₃ 19.3, Fe ₂ O ₃ 0.9, FeO 6.1, MgO 14.3%.
1.644 ^	1.639	(<u>1.652</u>)	1.656	.017	WOLLASTONITE (Ca, Fe)SiO ₃	59°	Y ~ $\frac{b}{c}$ X:c = 3°	TCL	100 perf 001, 102 good	Brown	H 5 G 3.07 F 4	Dec by HCl. FeO 8.7, MnO 2.7, Fe ₂ O ₃ 0.6%.
1.690 ^	1.645	<u>1.652</u>	1.656	.011	LAVENTITE (Na, Ca) ₃ ZrSi ₂ O ₇ (O, OH, F) ₂	86° r > v wk	el pos	ORTH(?) tab 100	100	Light yellow	H 6 G 3.25	Poly tw 100. Dimorph of Lavenite at β = 1.690 (?).

1.643 △ 1.655	1.643	<u>1.652</u>	1.654	.011	BITYITE (Margarite ser, Mica grp) $\text{CaLiAl}_2(\text{AlBeSi}_2)_0_{10}(\text{OH})_2$	Med	$X = c$ $el \text{ c}\bar{T} \text{ pos}$	MCL	001 perf	White, yellow	H 3.5 G 3.05 F easy	Insol in HCl. Basal section divided into 6 segments.
	1.559	<u>1.653</u>	1.680	.121	CALLAGHANITE $\text{Cu}_2\text{Mg}_2\text{CO}_3(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	55° $r > v \text{ str}$	$Z:c = 18^\circ$	MCL pyramid	111, $\bar{T}11$ perf	Azure- blue	H 3-3.5 G 2.71	Diss by acids with eff. Pleoc in blue, abs $Z > Y > X$.
1.649 □ 1.658	1.625	<u>1.653</u>	1.670	.045	DATOLITE $\text{CaBSiO}_4(\text{OH})$	73° $r > v \text{ wk}$	$Y = b$ $Z:a = 1-3^\circ$	MCL $el \text{ c}$	---	Cols	H 5.5 G 3.00 F 2.5	Gel with acids.
1.642 ^	1.619	<u>1.653</u>	1.660	.041	GORMANITE $(\text{Fe}^{+2}, \text{Mg})_3\text{Al}_4(\text{PO}_4)_4(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	53° $r > v$	---	TCL bladed	001 parting	Blue- green	---	Pleoc, X and Z cols, Y blue. FeO 14.7%.
1.644 ^	1.640	<u>1.653</u>	1.653	.013	JEREMEJEVITE $\text{Al}_6\text{B}_5\text{O}_{15}(\text{OH})_3$	Variable 0-33°	$X = c$	HEX pris	Conch	Cols to brown	H 6.5 G 3.28 infus	Insol in acids. Basal section divided into 6 sections.
	1.567	<u>1.654</u>	1.722	.155	TATARSKITE $\text{Ca}_6\text{Mg}_2(\text{SO}_4)_2(\text{CO}_3)_2\text{Cl}_4(\text{OH})_4 \cdot 7\text{H}_2\text{O}$	(79+1°)	$el \text{ pos}$	ORTH	---	Cols, yellow	H 2.5 G 2.34	---
1.705 v	1.618	<u>1.654</u>	1.655	.037	KIVUITE $(\text{Th}, \text{Ca}, \text{Pb})\text{H}_2(\text{UO}_2)_4(\text{PO}_4)_2(\text{OH})_8 \cdot 7\text{H}_2\text{O} (?)$	0-5° $r > v$	---	ORTH(?)	---	Yellow	---	Dec by HNO_3 . Pleoc, X cols, Y and Z greenish-yellow.
	1.595	<u>1.654</u>	1.670	.075	CALCIBORITE CaB_2O_4	54°	$Ext:el = 22^\circ$	MCL radi- ating	Conch	White	H 3.5 G 2.88	Diss by acids.
	1.650	<u>1.654</u>	1.661	.011	VLADIMIRITE $\text{H}_2\text{Ca}_5(\text{AsO}_4)_4 \cdot 5\text{H}_2\text{O}$	70° (106+21°) $r > v \text{ str}$	$Z:c = 36^\circ$	MCL	---	Cols	H 3.5 G 3.14 F diff	Diss by acids.
1.630 ◇ 1.667	1.642	<u>1.655</u>	1.661	.019	GEDRITE (Amphibole grp) $(\text{Mg}, \text{Fe}^{+2})_5\text{Al}_2(\text{Si}_6\text{Al}_2)\text{O}_{22}(\text{OH})_2$	Large $r > v$	$Y = b$ $Z = \bar{c}$ $el \text{ pos}$	ORTH pris \bar{c}	210 perf	Brown, gray	H 6 G 3.18 F 4	Insol in acids. FeO 9.2, Fe_2O_3 0.2, Al_2O_3 23.8%.
1.645 ◇ 1.667	1.649	<u>1.655</u>	1.659	.010	HUREAULITE $(\text{Mn}, \text{Fe})_5(\text{PO}_4)_2[\text{PO}_3(\text{OH})]_2 \cdot 4\text{H}_2\text{O}$	75° $r < v \text{ str}$	$X = b$ $Z:c = 55^\circ$	MCL tab	100 good	Pink to brown	H 3.5 G 3.18 F 2	Diss by acids. Pleoc, X cols, Y pale pink or yellow, Z pink or brown.
1.644 ◇ 1.683	1.645	<u>1.656</u>	1.661	.016	MAGNESIO-ARFVEDSONITE (Amphibole grp) $(\text{Na}, \text{Ca})_3(\text{Mg}, \text{Fe}^{+2})_4\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	57°	$Y = b$ $Z:c = 56^\circ$	MCL	110 perf at 124°	Bluish- green	H 6 G 3.23	Insol in acids. Pleoc, X blue-green, Y green, Z pale green. Fe_2O_3 8.4, FeO 13.4, MnO 1.5, MgO 7.8, F 3.3%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.569	<u>1.657</u>	1.686	.117	CALKINSITE (Ce,La) ₂ (CO ₃) ₃ ·4H ₂ O	54° r < v	Y = $\frac{c}{a}$ Z = $\frac{a}{b}$	ORTH platy	010 perf 101 fair	Pale yellow	H 2.5 G 3.28	Diss by acids with eff. Commonly tw 101. Pleoc wk.
	1.650	(<u>1.657</u>)	1.660	.010	DELLAITE Ca ₆ Si ₃ O ₁₁ (OH) ₂	65°	Z':el = 20°	---	---	Cols	---	---
	1.612	<u>1.658</u>	1.682	.070	LAUEITE MnFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	50° (70+3°) r < v str	Z:c ~ 30°	TCL	010 perf brittle	Brown	H 3 G 2.45	Diss by acids. Pleoc wk.
	1.615	---	1.685	.070	LIKASITE Cu ₆ (PO ₄)(NO ₃) ₂ (OH) ₇	---	X = $\frac{a}{b}$ Z = $\frac{b}{c}$	ORTH ps hex	001 perf	Blue	G 2.97	Diss by acids. Pleoc, X green-blue, Z pale blue. Opt char unk.
	(1.640)	<u>1.658</u>	1.664	(.024)	JUNGITE Ca ₂ Zn ₄ Fe ⁺³ ₈ (PO ₄) ₉ (OH) ₉ ·16H ₂ O	~ 60° r < v str	Y = $\frac{a}{b}$ Z = $\frac{c}{a}$	ORTH tab	010 perf	Yellow to yellow-green	H 1 G 2.84	---
v 1.660	1.622	<u>1.658</u>	1.687	.065	ANNABERGITE (Erythrite ser) Ni ₃ (AsO ₄) ₂ ·8H ₂ O	84° r > v str crossed	X = $\frac{b}{c}$ Z:c = 36° el clv pos	MCL pris acic	010 perf	Apple-green	H 2.5 G 3.07 F 4	Diss by acid. With increase of Co, becomes gray, then pink.
v 1.686	1.598	<u>1.658</u>	1.660	.062	HENDRICKSITE (Mica grp) K(Zn,Mn) ₃ Si ₃ AlO ₁₀ (OH) ₂	8°	---	MCL	001 perf	Dark reddish-brown	H 3 G 3.4	Dec by acids. ZnO 19.8, MnO 12.5, Fe ₂ O ₃ 4.85%.
\sqrt{v} 1.695	1.639	<u>1.658</u>	1.670	.031	JAHNSITE CaMn(Mg,Fe ⁺²) ₂ Fe ⁺³ ₂ (PO ₄) ₄ (OH) ₂ ·8H ₂ O	Large	Z = $\frac{b}{c}$ X:c = 18° el pos	MCL pris tab	001 good	Brown to yellow	H 4 G 2.72	Diss by acids. Tw 001. Pleoc, X pale purple to brown, Y red-brown, Z cols to yellow, abs Y > Z > X.
	1.63	<u>1.658</u>	1.66	.03	STEWARTITE MnFe ⁺³ ₂ (PO ₄) ₂ (OH) ₂ ·8H ₂ O	Large r < v str	X:c = 56°	TCL	010 good	Brownish-yellow	G 2.94	Pleoc wk.
	(~1.640)	<u>1.658</u>	1.664	(~.024)	JUNGITE Ca ₂ Zn ₄ Fe ⁺³ ₈ (PO ₄) ₉ (OH) ₉ ·16H ₂ O	~ 60° r < v str	Y = $\frac{a}{b}$ Z = $\frac{c}{a}$	ORTH tab	010 perf	Yellow to yellow-green	H 1 G 2.84	---
1.643 ^	1.652	<u>1.658</u>	1.665	.013	HIORTDAHLITE (Ca,Na) ₃ ZrSi ₂ O ₇ (O,OH,F) ₂	~ 90° r < v	Ext on 100 = 65°	TCL tab	Pris at ~ 90°	Yellow, brown	H 5.5 G 3.25 F 3	Gel with acids. Poly tw pl 100. Pleoc wk, X cols, Y and Z pale brownish-yellow.

1.648 ^	1.641	1.659	1.662	.021	FERRO-GLAUCOPHANE (Amphibole grp) $\text{Na}_2(\text{Fe}^{+2}, \text{Mg})_3\text{Al}_2\text{Si}_8\text{O}_{22}(\text{OH})_2$	42° $r < v$ str	$Y = \frac{b}{c}$ $Z:c = 6^\circ$ el pos	MCL pris	110 perf at 121°	Blue	H 6 G 3.23 F 3	Insol in acids. Pleoc, X cols, Y violet, Z blue. FeO 17.9, Fe ₂ O ₃ 4.0, Al ₂ O ₃ 10.1%.
	1.648	1.659	1.660	.012	CLINTONITE (Mica grp) $\text{Ca}(\text{Mg}, \text{Al})_3(\text{Al}_3\text{Si})\text{O}_{10}(\text{OH})_2$	5-33°	$Z = \frac{b}{c}$ el pos	MCL ps hex	001 perf	Reddish- brown	H 5 G 3.07 infus	Insol in acids. "Brittle mica." Pleoc, X cols to pale yellow, Z brown to green.
1.658 ^	1.623	1.660	1.694	.071	ERYTHRITE $\text{Co}_3(\text{AsO}_4)_2 \cdot 8\text{H}_2\text{O}$	87° $r > v$ wk	$X = \frac{b}{c}$ $Z:c = 31^\circ$ el pos	MCL pris	010 perf	Pink to red	H 3 G 3.06 F 2	Diss by acids. Pleoc, X pink, Y pale violet, Z red.
	1.640	1.660	1.675	.035	TILASITE $\text{CaMg}(\text{AsO}_4)\text{F}$	83° $r < v$ wk	$Z = \frac{b}{c}$ $X:c \sim 30^\circ$	MCL	10I good	Gray, green	H 5 G 3.77 F easy	Diss by acids. Tw 001 common.
1.640 ^	1.648	1.660	1.663	.015	ROWEITE $\text{Ca}_2(\text{Mn}, \text{Zn})_2\text{B}_4\text{O}_7(\text{OH})_6$	15° (53+18°) $r < v$ str	$Z = \frac{b}{c}$ $Y = \frac{c}{a}$	ORTH tab	101 poor	Light brown	H 5 G 2.92 F 1	Diss by acids. MnO 28.3, ZnO 3.1, MgO 1.7%.
(1.63) ^	1.645	(~1.66)	1.672	.027	SODIUM BOLTWOODITE $(\text{H}_3\text{O})(\text{Na}, \text{K})(\text{UO}_2)\text{SiO}_4 \cdot \text{H}_2\text{O}$	Large	---	ORTH	010 perf 001 imperf	Pale yellow	G 4.1	Pleoc, X cols, Z pale yellow.
1.634 ^	1.584	1.661	1.661	.077	STILPNOMELANE $\text{K}(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3}, \text{Al})_{10}\text{Si}_{12}\text{O}_{30}(\text{OH})_{12}$	~ 0°	$Y = \frac{b}{c}$ $X \sim \frac{c}{a}$ el cTv pos	TCL also MCL	001 perf	Dark brown	H 3-4 G 2.80 F 4.5	Dec by acids. Pleoc, X golden yellow, Y and Z dark brown. FeO 8.9, Fe ₂ O ₃ 19.4, MnO 2.9%.
1.685	1.640	1.661	1.663	.023	SWAMBOITE $\text{U}_3\text{H}_6(\text{UO}_2)_6(\text{SiO}_4)_6 \cdot 30\text{H}_2\text{O}$	34°	---	MCL acic	---	Pale yellow	G 4.0	Pleoc, X cols, Y and Z pale yellow. Does not fluor in UV.
v 1.667	1.646	1.661	1.661	.015	PENNANTITE (Chlorite grp) $\text{Mn}_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{CH})_8$	~ 0° $r < v$ str	---	MCL	001 perf	Orange- brown	H 2-3 G 3.06	MnO 38.9, Al ₂ O ₃ 18.6, Fe ₂ O ₃ 4.4%.
	1.640	1.662	1.667	.027	VÄYRYNENITE $\text{MnBePO}_4(\text{OH}, \text{F})$	46° $r > v$ mod	$Y = \frac{b}{c}$ $Z:c = -31^\circ$	MCL pris	001 perf 100 good	Rose	H 5 G 3.18	---
1.643 ^	1.644	(1.662)	1.663	.019	CELADONITE (Mica grp) $\text{K}(\text{Mg}, \text{Fe}^{+2})(\text{Fe}^{+3}, \text{Al})\text{Si}_4\text{O}_{10}(\text{OH})_2$	Small $r > v$	$Y = \frac{b}{c}$ $X \sim \frac{c}{a}$ el cTv pos	MCL	001 perf	Bright green	H 2 G 2.8 F 3	Dec by HCl. Pleoc, X and Y yellow-green, Z emerald-green. FeO 9.2, Fe ₂ O ₃ 26.9, Al ₂ O ₃ 2.1%.
v 1.697	1.654	1.662	1.668	.014	MAGNESIO-RIEBECKITE (Amphibole grp) $(\text{Na}, \text{Ca})_2(\text{Mg}, \text{Fe}^{+2})_3\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH})_2$	Med large	$Y = \frac{b}{c}$ $X:c = 9-20^\circ$ el neg	MCL pris	110 perf at 124°	Dark blue	H 5 G 3.2	Pleoc, X dark blue, Y indigo blue, Z yellow- green.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.643 △ 1.67	1.656	<u>1.662</u>	1.668	.012	HELLANDITE (Ca,Y) ₆ (Al,Fe ⁺³)Si ₄ B ₄ O ₂₀ (OH) ₄	87°	X = <u>b</u>	MCL	100 dist	Brown	H 4.5 G 3.3 F 2-3	Gel with acids. Tw 100.
	1.640	<u>1.663</u>	1.665	.025	SEAMANITE Mn ₃ BPO ₄ (OH) ₆	40° r < v	Y = <u>b</u> Z = <u>c</u>	ORTH acic	001 dist	Pale yellow	H 4 G 3.08 F easy	Diss by acids.
	1.650	<u>1.663</u>	1.670	.020	ACTINOLITE (Amphibole grp) Ca ₂ (Mg,Fe) ₅ Si ₈ O ₂₂ (OH) ₂	65° r < v	Y = <u>b</u> Z:c = 16° el pos	MCL pris	110 perf at 124°	Green	H 5.5-6 G 3.17	Insol in acids. Pleoc. FeO 14.9, MnO 0.3, Fe ₂ O ₃ 3.9%.
	1.637	<u>1.664</u>	1.692	.055	WILHELMVIERLINGITE CaMnFe ⁺³ (PO ₄) ₂ (OH)·2H ₂ O	(90+4°)	X = <u>b</u> Y = <u>a</u>	ORTH	---	Yellow	H 4 G 2.58	Pleoc, X and Y light yellow, Z yellow. Reported to have 2V = 45° calc.
	1.640	<u>1.664</u>	1.675	.035	CYANOPHILLITE Cu ₁₀ Al ₄ Sb ⁺³ ₆ O ₂₅ ·25H ₂ O	(67+7°)	X = <u>c</u>	ORTH spherulitic	001 perf	Greenish-blue	H 2 G 3.10	---
	1.646	<u>1.664</u>	1.676	.030	TANEYAMALITE Na(Mn ⁺² ,Mg,Fe ⁺³) ₁₂ Si ₁₂ (O,OH) ₄₄	70°	el pos	TCL	010 perf	Greenish gray-yellow	H 5 G 3.30 calc	Pleoc, X and Y nearly cols, Z pale yellow.
	1.643	<u>1.664</u>	1.670	.027	RANUNCULITE HAl(UO ₂)(PO ₄)(OH) ₃ ·4H ₂ O	(56+10°)	---	MCL ps orth	---	Gold-yellow	G 3.4	Pleoc, X pale greenish yellow, Y pale yellow.
	1.650 ◇ 1.681	<u>1.664</u>	1.670	.030	CHILDRENITE (Fe,Mn)AlPO ₄ (OH) ₂ ·H ₂ O	45° r < v	X = <u>b</u> Z:c = 4°	MCL ps orth	001 poor	Pink to brown	H 5 G 3.14 F 4	Diss by acids. FeO 14.6, MnO 13.9%.
	1.652 ◇ 1.669	<u>1.664</u>	1.671	.021	TSCHERMAKITE (Amphibole grp) Ca ₂ (Mg,Fe ⁺²) ₃ Al ₂ (Si ₆ Al ₂)O ₂₂ (OH) ₂	81°	Y = <u>b</u> ext Tn-clined el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.15	Insol in acids. Pleoc. FeO 9.5, Fe ₂ O ₃ 2.4, TiO ₂ 2.4%.
1.652 ◇ 1.687	1.652	<u>1.664</u>	1.672	.020	MAGNESIO-HASTINGSITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ Fe ⁺³ Si ₆ Al ₂ O ₂₂ (OH) ₂	80°	Y = <u>b</u> Z:c = 19° el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.18	Insol in acids. Pleoc, X pale brown, Y dark brown, Z green-brown. FeO 8.7, Fe ₂ O ₃ 2.6, Al ₂ O ₃ 14.3, TiO ₂ 1.3%.

1.649 ^ 1.694	1.654	<u>1.664</u>	1.674	.020	MONTICELLITE $\text{Ca}(\text{Mg}, \text{Fe})\text{SiO}_4$	82° r > v	$X = \frac{b}{Z} = \frac{a}{a}$	ORTH	010 poor	White, gray	H 5.5 G 3.2 F 6	Gel with acids. FeO 7.6, MnO 0.2%.
[]	1.646	<u>1.664</u>	1.664	.018	GONYERITE (Chlorite grp?) $(\text{Mn}, \text{Mg})_5\text{Fe}^{+3}(\text{Si}_3\text{Fe}^{+3})_8\text{O}_{10}(\text{OH})_8$	~ 0° r < v str	el clv pos	ORTH(?)	001 perf	Dark brown	H 2.5 G 3.01	Dec by HCl. Pleoc, X dark brown, Z light brown.
	1.654	<u>1.664</u>	1.666	.012	CUPROSKLODOWSKITE $\text{Cu}(\text{UO}_2)_2(\text{SiO}_3)_2(\text{OH})_2 \cdot 5\text{H}_2\text{O}$	Med r > v	$X \sim \frac{a}{Y} \sim \frac{b}{b}$	TCL fib	100	Yellow- green	G 3.85	Pleoc wk, X cols, Z greenish-yellow. Not fluor in UV.
	1.638	<u>1.665</u>	1.676	.038	KINOITE $\text{Ca}_2\text{Cu}_2\text{Si}_3\text{O}_8(\text{OH})_4$	68° r < v	$X = \frac{b}{Z} \sim \frac{c}{c}$	MCL tab	010 perf 100, 001 dist	Azure- blue	H 5 G 3.16	Dec by HCl. Pleoc, X pale green, Y blue, Z dark blue.
1.644 ^ 1.677	1.650	<u>1.665</u>	1.679	.029	GRUNERITE (Cummingtonite ser, Amphibole grp) $(\text{Fe}^{+2}, \text{Mg}, \text{Mn})_7\text{Si}_8\text{O}_{22}(\text{OH})_2$	87° r < v	$Y = \frac{b}{Z:c} = 16^\circ$ el pos	MCL pris	110 perf at 124°	Gray, brown	H 6 G 3.31 fus	Insol in acids. Pleoc, X and Y yellow, Z brownish-yellow. FeO 21.4, MnO 8.0, Fe ₂ O ₃ 0.6%.
v 1.676	1.654	<u>1.665</u>	1.668	.009	MAGNESIOAXINITE $\text{Ca}_2(\text{Mg}, \text{Fe}, \text{Mn})\text{Al}_2\text{BSi}_4\text{O}_{15}(\text{OH})$	76° r < v	$X \sim \perp \text{111}$	TCL	100 good	Brown	H 6.5 G 3.19 F 2	Insol in acids. CaO 20.3, MgO 4.2, FeO 5.2, MnO 1.4, Fe ₂ O ₃ 1.7%.
	1.638	<u>1.666</u>	1.682	.044	LEPERSONNITE $\text{CaO} \cdot (\text{Gd}, \text{Dy}, \text{Y})_2\text{O}_3 \cdot 24\text{UO}_3 \cdot 8\text{CO}_2 \cdot 4\text{SiO}_2 \cdot 60\text{H}_2\text{O}$	73°	---	ORTH acic	---	Yellow	G 3.97	Pleoc, X pale yellow, Y and Z bright yellow.
	1.649	<u>1.666</u>	1.676	.027	UPALITE $\text{Al}(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_3$	(74+9°)	$X = \frac{b}{Y} = \frac{a}{a}$	ORTH acic	010 dist	Amber- yellow	G 3.5	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
1.649 ^ 1.685	1.656	<u>1.666</u>	1.666	.010	CHAMOSITE (Chlorite grp) $(\text{Fe}^{+2}, \text{Fe}^{+3}, \text{Mg})_5\text{Al}(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH}, \text{F})_8$	~ 6°	$X \sim \frac{c}{\text{el pos}}$	MCL plates	001 perf	Dark green	H 2.5 G 2.96 fus	Dec by acids. FeO 26.2, Fe ₂ O ₃ 18.8%.
v 1.686	1.520	<u>1.667</u>	1.668	.148	STRONTIANITE (Aragonite grp) SrCO ₃	7° r < v wk	$X = \frac{c}{Z} = \frac{a}{a}$ el neg	ORTH pris	110 good	White, gray	H 3.5 G 3.68 infus	Diss by HCl with eff.
1.649 ^	1.642	<u>1.667</u>	1.669	.027	URANOPHANE $\text{Ca}(\text{UO}_2)_2\text{Si}_2\text{O}_7 \cdot 6\text{H}_2\text{O}$	32° r < v str	$Z = \frac{b}{b}$	MCL acic	100 perf	Lemon- to pale- yellow	H 2.5 G 3.8 infus	Gel with acids. Pleoc wk, X cols, Y pale yellow, Z canary yellow.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.655 ^ 1.681	1.656	1.667	1.672	.016	GEDRITE (Amphibole grp) $(Mg, Fe^{+2})_5Al_2(Si_6Al_2)O_{22}(OH)_2$	57° $r > v$	$Y = \underline{b}$ $Z = \underline{c}$ el pos	ORTH pris \underline{c}	210 perf	Brown, gray	H 6 G 3.22 F 4	Insol in acids. FeO 17.1, Fe ₂ O ₃ 4.3, Al ₂ O ₃ 10.35%.
1.661 ^ 1.673	1.658	1.667	1.667	.009	PENNANTITE (Chlorite grp) $Mn_5Al(Si_3Al)O_{10}(OH)_8$	Small	---	MCL ps hex ro-settes	---	Dark brown	G 3.15	"Grovesite." Pleoc in browns and reds.
	1.662	1.667	1.669	.007	CLINOHEDRITE $CaZnSiO_3(OH)_2$	Med $r > v$	$Z = \underline{b}$ $Y:\underline{c} = -28^\circ$ el neg	MCL	010 perf	Cols	H 5.5 G 3.33 F 4	Gel with acids.
1.649 ^	1.665	1.667	1.667	.002	CHLORAPATITE (Apatite grp) $Ca_5(PO_4)_3(Cl, F, OH)$	10°	---	MCL ps hex	---	Cols	H 5 G 3.18 F 5	Diss by acids. Cl 6.2, F 0.1, H ₂ O 0.9%.
1.655 ^	1.657	1.667	1.671	.014	HUREAULITE $(Mn, Fe)_5(PO_4)_2$ $[PO_3(OH)]_2 \cdot 4H_2O$	61°	$X = \underline{b}$ $Z:\underline{c} = 61^\circ$	MCL tab	100 good	Pink to brown	H 3.5 F 2	Diss by acids. MnO 22.4, FeO 20.7%.
	1.635	1.668	1.702	.067	SYMPLESITE $Fe_3(AsO_4)_2 \cdot 8H_2O$	87° $r > v$ str	$X \perp 1\bar{1}0$ $Z:\underline{c} = 32^\circ$ el clv pos	TCL pris	110 perf	Green to blue	H 2.5 G 3.01 infus	Diss by acids. Pleoc, X deep blue, Y cols, Z yellow-green.
	1.662	1.668	1.672	.010	ARROJADITE $(K, Ba)(Na, Ca)_5(Fe, Mn, Mg)_{14}Al(PO_4)_{12}(OH, F)$	80° $r < v$ str	$X = \underline{b}$ $Y:\underline{c} = 21^\circ$	MCL u mass	001 good 201 dist	Dark green	H 5 G 3.55	Diss by acids. Pleoc, X cols, Y pale green, Z pale yellow-green.
	1.658	1.669	1.670	.012	ZINKOSITE $ZnSO_4$	Small $r < v$ str	$X = \underline{b}$ $Z = \underline{c}$	ORTH plates	---	White	G 3.7	Diss by H ₂ O. Alters on exposure. Natural occurrence doubtful.
1.684 v	1.665	1.669	1.672	.007	LITHIOPHILITE (Triphylite grp) $Li(Mn, Fe)PO_4$	80° $r < v$ str	$X = \underline{c}$ $Y = \underline{a}$	ORTH pris	100 perf 010 less so	Yellow- ish-brown	H 4-5 G 3.45	Diss by acids. MnO 22.6, FeO 12.8, MgO 6.3%.
1.664 ^ 1.693	1.658	1.669	1.676	.018	FERROTSCHERMAKITE (Amphibole grp) $(Ca, Na)_2(Fe^{+2}, Mg)_3Al_2(Si_6Al_2)O_{22}(OH)_2$	(77+13°)	$Y = \underline{b}$ $Z:\underline{c} = 7^\circ$ el pos	MCL pris	110 perf at 124°	Green	H 5-6	Pleoc, X tan, Y green, Z bluish green.

1.670 pos ^	1.619	1.670	1.720	.101	STRUNZITE $\text{MnFe}^{+3}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Med (87±2°)	Z:c = 10°	MCL	---	Straw-yellow	G 2.52	Pleoc wk in yellows, abs Z > X > Y.
1.664 ^ 1.678	1.657	1.670	1.679	.022	MAGNESIO-HORNBLENDE (Amphibole grp) $\text{Ca}_2(\text{Mg,Fe}^{+2})_4\text{Al}(\text{Si}_7\text{Al})\text{O}_{22}(\text{OH})_2$	78°	Y = $\frac{b}{a}$ el pos	MCL pris	110 perf at 124°	Brown	H 5-6 G 3.23	Insol in acids. Pleoc. FeO 13.2, Fe ₂ O ₃ 4.7, MnO 0.3, MgO 10.6%.
	1.526	1.671	1.672	.146	ALSTONITE $\text{BaCa}(\text{CO}_3)_2$	6° r > v wk	X = $\frac{c}{b}$ Z = $\frac{b}{a}$	ORTH el c	110 imperf	Cols	H 4.5 G 3.71 F diff	Diss by acids with eff.
1.652 ^	1.624	1.672	1.672	.048	ANNITE (Biotite ser, Mica grp) $\text{KFe}^{+2}_3\text{AlSi}_3\text{O}_{10}(\text{OH,F})_2$	5° r < v	Y = $\frac{b}{a}$ el pos	MCL plates	001 perf	Dark brown	H 2.5-3 G 3.0 F 4	Pleoc, X brown, Y and Z dark brown. FeO 32.1, Fe ₂ O ₃ 3.1, TiO ₂ 3.2%.
	1.632	1.672	1.672	.040	Serpentine grp mineral (Mg,Ni,Fe,Al) ₃ (Si,Al) ₂ O ₅ (OH) ₄	0-10°	---	MCL	001 good	Dark green	---	Species not identified. Pleoc, X yellow-green, Y and Z dark green.
	1.650	1.672	1.677	.027	PARASPURRITE $\text{Ca}_5(\text{SiO}_4)_2(\text{CO}_3)$	77° (50±11°)	X = $\frac{b}{a}$ Z:c ~ 30°	MCL	001 poor	Cols	G 3.03 infus	Dec by HCl with eff. Poly tw.
	1.634	1.673	1.685	.051	DURANGITE $\text{NaAl}(\text{AsO}_4)\text{F}$	45° (57±5°) r > v wk	Y = $\frac{b}{a}$ X:c = -25°	MCL pris	110 dist	Orange-red	H 5 G 4.0 F 2	Pleoc, X orange-yellow, Y pale orange, Z cols.
v 1.682	1.662	1.673	1.684	.022	TRIPLITE (Mn,Fe,Mg,Ca) ₂ PO ₄ (F,OH)	88°	Y = $\frac{b}{a}$ Z:a = 42°	MCL u mass	001 good 010 fair	Dark brown	H 5-5.5 G 3.58 F easy	Diss by acids. MnO 53.8, FeO 6.7, CaO 2.2%.
1.665 [1.604	1.674	1.731	.127	BUTLERITE $\text{Fe}^{+3}\text{SO}_4(\text{OH}) \cdot 2\text{H}_2\text{O}$	Large	---	MCL pris	100 perf	Deep orange	H 2.5 G 2.55	Pleoc, X nearly cols, Y and Z pale yellow.
	1.643	1.674	1.701	.058	CLINOKURCHATOVITE $\text{Ca}(\text{Mg,Fe,Mn})\text{BO}_5$	85°	---	MCL	010 perf	Cols	H 4.5 G 3.07	Poly tw. Diss by acids.
v 1.681	1.642	1.674	1.699	.057	KURCHATOVITE $\text{Ca}(\text{Mg,Mn,Fe})\text{BO}_5$	82° r > v	X = $\frac{b}{a}$ Z = $\frac{c}{a}$	ORTH mass	One perf 2 imperf	White	H 4.5 G 3.07	Diss by warm HCl.
1.670 □ 1.676	1.640	1.674	1.679	.039	SPURRITE $\text{Ca}_5(\text{SiO}_4)_2(\text{CO}_3)$	40° r > v wk	X = $\frac{b}{a}$ Z ~ $\frac{a}{b}$	MCL pris	001 perf 100 good	Cols	H 5 G 3.01 infus	Dec by HCl with eff. Poly tw 001.
1.646 ^	1.645	1.674	1.680	.035	SCORZALITE (Lazulite ser) $(\text{Fe,Mg})\text{Al}_2(\text{PO}_4)_2(\text{OH})_2$	64° (48±8°) r < v	Y = $\frac{b}{a}$ X ~ $\frac{c}{a}$	MCL	110 good	Dark blue	H 6 G 3.36 infus	Insol in acids. Pleoc, X cols, Y and Z dark blue. FeO 17.7, Fe ₂ O ₃ 3.0, MnO 0.5%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.696 ^v	1.660	1.674	1.675	.015	KORNERUPINE Mg ₃ Al ₆ (Si,Al,B) ₅ O ₂₁ (OH)	20° r > v wk	X = $\frac{c}{a}$ Z = $\frac{b}{a}$ el neg	ORTH pris	110 good	Green to brownish-green	H 6.5 G 3.28 F 6	Insol in acids. Pleoc variable, X yellow to green, Y cols to pale brown, Z pale to dark green.
1.691 ^v	1.655	1.675	1.685	.030	DUMORTIERITE Al ₇ (BO ₃)(SiO ₄) ₃ O ₃	Med r < v	X = $\frac{c}{a}$ Z = $\frac{a}{b}$ el neg	ORTH acic \underline{c}	100 dist	Blue	H 7 G 3.25 infus	Insol in acids. Pleoc str, X deep blue, Y yellow, Z cols.
(1.650) ∧ 1.695	1.664	1.675	1.679	.015	BUSTAMITE (Ca,Mn) ₃ Si ₃ O ₉	50-55°	X:a ~ 15° Y:b ~ 35° Z:c ~ 35°	TCL	100 perf 110, 110 good	Pink	H 6 G 3.30	MnO 27.7, CaO 24.9%.
	1.665	1.675	1.678	.013	GERSTMANNITE (Mg,Mn) ₂ ZnSiO ₄ (OH) ₂	50-60°	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH pris	010 good	White to pale pink	H 4.5 G 3.68	---
	1.529	1.676	1.677	.148	WITHERITE (Aragonite grp) BaCO ₃	16° r > v wk	X = $\frac{c}{a}$ Z = $\frac{a}{b}$ el neg	ORTH el \underline{c}	010 dist 110 imperf	Cols, white	H 3-3.5 G 4.29 F 2.5-3	Diss by HCl. Tw pl 110 universal.
1.665 ∧ 1.681	1.670	1.676	1.681	.011	FERROAXINITE Ca ₂ (Fe,Mg,Mn)Al ₂ BSi ₄ O ₁₅ (OH)	69°	X ~ \perp T11	TCL	100 good	Violet	H 6.5 G 3.26 F 2	Insol in acids. CaO 19.6, FeO 8.1, MgO 2.1, MnO 1.8, Fe ₂ O ₃ 0.9%.
	1.660	1.676	1.686	.026	FERRIPYROPHYLLITE Fe ⁺³ ₂ Si ₄ O ₁₀ (OH) ₂	(76+9°)	---	MCL	---	Brownish-yellow	H 1.5-2 G 2.99	---
	1.629	1.677	1.679	.050	SAMPLEITE NaCaCu ₅ (PO ₄) ₄ Cl·5H ₂ O	23° r > v	X = $\frac{b}{a}$ Y = $\frac{a}{c}$	ORTH	010 perf 110, 001 good	Blue	H 4 G 3.20 F 2	Diss by acids. Pleoc, X blue-green, Y and Z turquoise-blue, abs Z = Y > X.
1.665 ∧ 1.677 pos ∧ 1.682	1.660	1.677	1.693	.033	GRUNERITE (Cumingtonite ser, Amphibole grp) (Fe ⁺² ,Mg) ₇ Si ₈ O ₂₂ (OH) ₂	87°	Y = $\frac{b}{c}$ ext Tn-clined el pos	MCL pris \underline{c}	110 perf at 110°	Green to brown	H 7 G 3.47	Insol in acids. Pleoc, X pale yellow, Z brown. FeO 31.2, Fe ₂ O ₃ 3.4%.
1.677 pos ∧ 1.680	1.672	1.677	1.682	.010	BRONZITE (Orthopyroxene ser, Pyroxene grp) (Mg,Fe) ₂ Si ₂ O ₆	~ 90°	Y = $\frac{a}{c}$ Z = $\frac{c}{b}$ el pos	ORTH pris	210 good at 87°	Green, brown	H 5.5 G 3.3 F 5	Nearly insol in acids. Faint pleoc common.

1.670 ↓ 1.685	1.665	<u>1.678</u>	1.684	.019	EDENITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_5\text{Si}_7\text{Al}$ $\text{O}_{22}(\text{OH})_2$	65°	$Y = \frac{b}{c}$ $Z:c = 19^\circ$ el pos	MCL	110 perf at 124°	Light green	H 5.5 G 3.15	Insol in acids. Pleoc, X yellow, Y pale green, Z dark green. FeO 13.4, Fe_2O_3 6.2, Al_2O_3 7.8%.
	1.569	<u>1.679</u>	1.708	.139	CARBOCERNAITE (Ca,Ce,Na,Sr) CO_3	56° (51+2°) $r > v$	$X = \frac{b}{a}$ $Y = \frac{a}{c}$	ORTH	---	Cols to yellow	H 3 G 3.66	Diss by acids.
	1.624	<u>1.679</u>	1.705	.081	POSNJAKITE $\text{Cu}_4\text{SO}_4(\text{OH})_6 \cdot \text{H}_2\text{O}$	57° (67+3°)	---	MCL	---	Dark blue	H 2-3 G 3.32 F easy	Diss by HCl. Pleoc, X bluish, Y darker blue, Z greenish blue, abs Y > Z > X.
v 1.695	1.530	<u>1.680</u>	1.685	.155	ARAGONITE (Aragonite grp) CaCO_3	18° $r < v$ wk	$X = \frac{c}{b}$ $Z = \frac{b}{c}$ el neg	ORTH acic <u>c</u>	010 dist	Cols	H 3-4 G 2.94 infus	Diss in acids with eff. Poly tw on 110 common.
	1.648	<u>1.680</u>	1.716	.068	METAKOETTIGITE (Zn, Fe^{+3} , Fe^{+2}) ₃ (AsO_4) ₂ · 8(H_2O , OH)	87°	---	TCL tab	110 perf	Bluish- gray	G (3.03)	Pleoc, X deep blue, Y yellow, Z light yellow.
1.680 pos ↓ 1.704	1.661	<u>1.680</u>	1.697	.036	CHRYSLITE (Olivine grp) (Mg, Fe) ₂ SiO_4	90°	$X = \frac{b}{a}$ $Z = \frac{a}{c}$	ORTH	010, 001 poor	Pale green	H 7 G 3.15 infus	Gel with acids. Data for Fo_{86} .
v 1.780	1.67	<u>1.680</u>	1.703	.03	ALLEGHANYITE (Mn, Zn) ₅ (SiO_4) ₂ (OH) ₂	66-86° $r > v$	$Z = \frac{b}{c}$ $X:c \sim 30^\circ$	MCL	fr conch	Brownish to red- dish-pink	H 5 G 3.70 F 3.5	Gel with acids.
1.652 ^	1.667	<u>1.680</u>	1.687	.020	PARGASITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe})_4\text{Al}$ (Si_6Al_2) $\text{O}_{22}(\text{OH})_2$	74°	$Y = \frac{b}{c}$ el pos	MCL	110 perf at 124°	Brown, green	H 5.5 G 3.22	Insol in acids. Pleoc. FeO 9.9, Fe_2O_3 4.9, TiO_2 1.9, MgO 11.6%.
	1.67	<u>1.68</u>	1.69	.02	METAVANMEERSSCHEITE $\text{U}(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_6 \cdot$ $2\text{H}_2\text{O}$	83° (92+3°)	---	ORTH tab	010 good 100 less so	Yellow	G (4.49)	Pleoc in yellow shades.
	(1.668)	<u>1.680</u>	1.686	(.018)	DAVREUXITE $\text{Mn}_2\text{Al}_{12}(\text{SiO}_4)_7\text{O}_3(\text{OH})_6$	70°	el pos	MCL fib	---	Creamy white to pale rose	G 3.15	---
1.677 ↓ 1.695	1.674	<u>1.680</u>	1.685	.011	BRONZITE (Orthopyroxene ser, Pyroxene grp) (Mg, Fe) ₂ Si_2O_6	79° $r > v$	$Y = \frac{a}{c}$ $Z = \frac{c}{b}$ el pos	ORTH pris <u>c</u>	210 good at 87°	Greenish	H 5.5 G 3.3 F 5	Nearly insol in acids. Faint pleoc common. FeO 11.1, MnO 0.5%.
	---	<u>1.680</u>	---	weak	BALANGEROITE (Mg, Fe^{+2} , Fe^{+3} , Mn) ₄ $\text{Si}_{15}(\text{O}, \text{OH})_{90}$	---	---	ORTH fib	1 or more good	Brown	G 2.98	Pleoc, yellow brown 001, dark brown ⊥ 001. Am. Mineral., 68, 214-219 (1983).

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.674 ^	1.635	<u>1.681</u>	1.698	.063	KURCHATOVITE $\text{Ca}(\text{Mg}, \text{Mn}, \text{Fe})\text{BO}_5$	66° (61+4°) $r > v$ wk	$X = \frac{b}{c}$ $Z = \frac{c}{a}$	ORTH tab mass	One perf 2 imperf	Gray to cols	H 4.5 G 3.02	Diss by acids. Fluor bright violet in long- wave UV.
1.664 ^	1.646	<u>1.681</u>	1.687	.041	CHILDRENITE $(\text{Fe}, \text{Mn})\text{AlPO}_4(\text{OH})_2 \cdot \text{H}_2\text{O}$	50° $r > v$	$X = \frac{b}{c}$ $Y:c = 8^\circ$	MCL ps orth	001 poor	Pink to brown	H 5 G 3.19 F 4	Diss by acids. FeO 28.6, MnO 3.1%.
1.667 ^ 1.710	1.671	<u>1.681</u>	1.690	.019	FERRO-GEDRITE (Amphibole grp) $(\text{Fe}^{+2}, \text{Mg})_5\text{Al}_2(\text{Si}_6\text{Al}_2)\text{O}_{22}(\text{OH})_2$	75° (86+12°) $r > v$	$Y = \frac{b}{c}$ $Z = \frac{c}{a}$ el pos	ORTH pris $\frac{c}{a}$	210 perf	Brown, gray	H 6 G 3.33 F 4	Insol in acids. Pleoc. FeO 24.4, Fe_2O_3 0.9, Al_2O_3 13.7%.
1.676 ^ 1.687	1.674	<u>1.681</u>	1.688	.014	TINZENITE (Axinite grp) $(\text{Ca}, \text{Mn}, \text{Fe})_3\text{Al}_2\text{BSi}_4\text{O}_{15}(\text{OH})$	Large $r < v$	$X \perp T_{11}$	TCL	100 good	Yellow- green	H 6.5 G 3.28 F 2	Insol in acids. CaO 15.0, MnO 15.8, MgO 1.3, FeO 0.4%.
	(1.62)	<u>1.682</u>	1.688	(.070)	MUNDITE $\text{Al}(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_3 \cdot 5.5\text{H}_2\text{O}$	33°	el neg	ORTH	001, 100, 010 good	Pale yellow	G (4.30)	Pleoc wk.
	1.637	<u>1.682</u>	1.694	.057	WROEWOLFEITE $\text{Cu}_4(\text{SO}_4)(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	53°	---	MCL	010, 100, 001 perf	Deep greenish- blue	H 2.5 G 3.27	Pleoc, X light blue, Y deep greenish-blue, Z med greenish-blue, abs $Y > Z \gg X$.
1.677 ^ 1.700	1.666	<u>1.682</u>	1.698	.032	DANNEMORITE (Cumingtonite ser, Amphibole grp) $\text{Mn}_2(\text{Fe}^{+2}, \text{Mg})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$	88° $r < v$	$Y = \frac{b}{c}$ $Z:c = 15^\circ$ el pos	MCL	110 perf at 125°	Gray	H 5.5-6 G 3.50	Insol in acids. Pleoc, X red-violet, Y pale violet, Z blue. FeO 22.6, MnO 15.6%.
1.673 ^	1.671	<u>1.682</u>	1.690	.019	TRIPLITE $(\text{Mn}, \text{Fe}, \text{Mg}, \text{Ca})_2\text{PO}_4(\text{F}, \text{OH})$	Large $r > v$	$Y = \frac{b}{c}$ $Z:a = 42^\circ$	MCL	001 good 010 less so	Brown	H 5-5.5 G 3.8	Diss by acids.
1.656 ^ 1.695	1.667	<u>1.683</u>	1.692	.025	MAGNESIO-ARFVEDSONITE (manganian) (Amphibole grp) $(\text{Na}, \text{Ca})_3(\text{Mg}, \text{Fe}^{+2})_4\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	62° (73+9°)	$Y = \frac{b}{c}$ $Z:c = 3^\circ$	MCL	110 perf at 124°	Dark bluish- green	H 5.5-6 G 3.27	Insol in acids. Pleoc, X and Y blue-green, Z yellow-green. FeO 1.0, Fe_2O_3 9.1, MgO 11.3, MnO 7.7%.

	1.525	<u>1.684</u>	1.686	.161	BARYTOCALCITE $\text{BaCa}(\text{CO}_3)_2$	15° $r > v$ wk	$Z = \frac{b}{c}$ $X:c = 64^\circ$	MCL el \underline{c}	110 perf 001 poor	Cols, white	H 4 G 3.65 F diff	Diss by HCl with eff.
	1.593	<u>1.684</u>	1.698	.105	YAVAPAIITE $\text{KFe}(\text{SO}_4)_2$	30° (41+3°) $r > v$ str	$Z = \frac{b}{c}$ $X:c = 6^\circ$	MCL	001, 100 perf 110 dist	Pink to cols	H 2.5-3 G 2.88	Dec by hot H_2O , diss by HCl.
	1.668	<u>1.684</u>	1.685	.017	WALSTROMITE $\text{BaCa}_2\text{Si}_3\text{O}_9$	30°	---	TCL	---	Cols	G 3.67	---
1.669 ^ 1.700	1.675	<u>1.684</u>	1.685	.010	TRIPHYLITE $\text{Li}(\text{Fe,Mg,Mn})\text{PO}_4$	25° $r < v$ str	$X = \frac{c}{a}$ $Y = \frac{a}{c}$	ORTH pris	100 perf 010 less so	Green	H 4-5 G 3.44	Diss by acids. FeO 32.9, MnO 3.1, MgO 7.4%.
-1.63 ^	1.610	<u>1.685</u>	1.704	.094	ROSCOELITE (Mica grp) $\text{K}(\text{V,Al,Mg})_2\text{AlSi}_3\text{O}_{10}$ (OH) ₂	10-15° (51+3°) $r < v$ str	$Z = \frac{b}{c}$ $X \sim \frac{c}{a}$ el pos	MCL plates	001 perf	Green	H 3 G 2.97 F 3	Slowly dec by HCl. Pleoc, X olive-green, Z greenish-brown. Char- acteristic green interf color.
1.661 ^ 1.705	1.595	<u>1.685</u>	1.685	.090	STILPNOMELANE $\text{K}(\text{Fe}^{+2}, \text{Mg}, \text{Fe}^{+3}, \text{Al})_{10}$ $\text{Si}_{12}\text{O}_{30}(\text{OH})_{12}$	~ 0°	$Y = \frac{b}{c}$ $X \sim \frac{c}{a}$ el cTv pos	TCL and MCL	001 perf	Dark brown	H 3-4 G 2.84 F 4	Dec by acids. Pleoc, X pale yellow, Y and Z dark olive-brown. FeO 13.7, Fe_2O_3 22.0%.
1.674 ^	1.650	<u>1.685</u>	1.712	.062	SODDYITE $(\text{UO}_2)_5\text{Si}_2\text{O}_9 \cdot 6\text{H}_2\text{O}$	84°	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH platy	001 perf 111 good	Greenish- yellow	H 3-4 G 4.70 infus	Gel with acids. Not fluor in UV.
1.705 ^	1.659	<u>1.685</u>	1.702	.043	TARBUTTITE $\text{Zn}_2(\text{PO}_4)(\text{OH})$	50° (77+5°)	Disp str	TCL pris	001 perf	Pale green	H 3.5-4 G 4.15 F easy	Diss by acids.
1.680 ^ 1.694	1.664	<u>1.685</u>	1.694	.030	URANOPHANE-BETA $\text{Ca}(\text{UO}_2)_2\text{Si}_2\text{O}_7 \cdot 6\text{H}_2\text{O}$	63° $r > v$ str crossed	$X = \frac{b}{c}$ $Z:c = 26^\circ$	MCL tab	010 good 100 poor	Yellow to yellow- green	H 2-3 G 3.8-3.9	Gel with acids. Pleoc wk, X cols, Y and Z light yellow.
1.710 ^	1.672	<u>1.685</u>	1.698	.026	TALMESSITE $\text{Ca}_2\text{Mg}(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	~ 90°	---	TCL	---	Cols	H 5 G 3.42	"Arsenate-belovite." Diss by acids.
1.678 ^ 1.693	1.670	<u>1.685</u>	1.693	.023	FERRO-HORNBLende (Amphibole grp) $\text{Ca}_2(\text{Fe}^{+2}, \text{Mg})_4\text{Al}$ (Si_7Al) ₀₂₂ (OH,F) ₂	61° (72+10°)	$Y = \frac{b}{c}$ $Z:c = 18^\circ$ el pos	MCL pris	110 perf at 124°	Greenish- brown	H 5-6 G 3.2	Insol in acids. Pleoc, X greenish-yellow, Y olive-green, Z bluish- green. FeO 18.1, Fe_2O_3 5.4, Al_2O_3 13.9%.
1.666 ^	1.670	<u>1.685</u>	1.685	.015	CHAMOSITE (Chlorite grp) ($\text{Fe}^{+2}, \text{Fe}^{+3}, \text{Mg}$) ₅ Al (Si_3Al) ₀₁₀ (O,OH) ₈	~ 0°	$X \sim \frac{c}{a}$ el pos	MCL plates	001 perf	Dark brown	H 2.5 G 2.96 fus	Dec by acids.
	1.627	<u>1.686</u>	1.690	.063	SOBOLEVITE $\text{Na}_2\text{Ca}_2\text{MnTi}_3\text{Si}_4\text{O}_{18} \cdot$ $4\text{Na}_3\text{PO}_4$	29°	$Y = \frac{b}{c}$ $X:c = 32^\circ$	MCL	110 perf	Brown	H 4.5-5 G 3.03	Dec by acids. Partly dec by H_2O , alk reaction.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.666	<u>1.686</u>	1.694	.028	CLINOTYROLITE $\text{Ca}_2\text{Cu}_9[(\text{As},\text{S})\text{O}_4]_4$ (O,OH) $_{10}\cdot 10\text{H}_2\text{O}$	66°	$Y = b$ $X:c = \frac{b}{a} = 7-8^\circ$	MCL silky	---	Emerald-green	G 3.22	Diss by acids.
1.667 ^	1.525	<u>1.686</u>	1.690	.165	STRONTIANITE (Aragonite grp) SrCO_3	8° (17+4°) $r < v$ wk	$X = c$ $Z = \frac{c}{a}$ el neg	ORTH pris	110 good	White, gray	H 3.5 G 3.81 infus	Diss by acids with eff. CaO 1.0, BaO 3.3%.
1.658 ^	1.624	<u>1.686</u>	1.686	.062	HENDRICKSITE (Mica grp) $\text{K}(\text{Zn},\text{Mn})_3\text{Si}_3\text{AlO}_{10}(\text{OH})_2$	2-5°	---	MCL	001 perf	Dark red-brown	H 3 G 3.41	Dec by acids. ZnO 21.4, MnO 13.9, FeO 0.7, Fe_2O_3 4.85%.
1.691]	1.640	<u>1.686</u>	1.702	.062	CHERNYKHITE (Mica grp) $(\text{Ba},\text{Na})(\text{V}^{+3},\text{Al})_2$ $(\text{Si},\text{Al})_4\text{O}_{10}(\text{OH})_2$	11-12° (60+4°)	---	MCL	001 perf	Olive-to dark-green	H 2.5-4 G 3.15	V_2O_3 18.6, V_2O_4 5.3, BaO 9.5%.
1.682 [1.689	1.662	<u>1.686</u>	1.692	.030	KAINOSITE $\text{Ca}_2(\text{Y},\text{Ce})_2(\text{SiO}_4)_3$ (CO_3) $\cdot \text{H}_2\text{O}$	40° (53+9°) $r < v$ dist	$X = c$ $Y = \frac{c}{b}$	ORTH	2 clv at 90°	Cols, rose, brown	H 5.5 G 3.51- 3.65	Diss by acids with slight eff.
1.664 ^ 1.709	1.676	<u>1.687</u>	1.695	.019	MAGNESIO-HASTINGSITE (Amphibole grp) $\text{NaCa}_2(\text{Mg},\text{Fe}^{+2})_4(\text{Fe}^{+3},\text{Ti})\text{Si}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	80°	$Y = b$ $Z:c = \frac{b}{a} = 15^\circ$ el pos	MCL pris	110 perf at 124°	Green	H 5-6 G 3.22	Insol in acids. Pleoc. FeO 6.9, Fe_2O_3 4.8, Al_2O_3 13.9, TiO_2 4.4%.
1.681 ^ 1.693	1.678	<u>1.687</u>	1.692	.014	MANGANAXINITE $\text{Ca}_2(\text{Mn},\text{Fe})\text{Al}_2\text{BSi}_4\text{O}_{15}$ (OH)	75°	$X \perp T11$	TCL	100 good	Yellow-brown	H 6.5 G 3.31 F 2	Insol in acids. MnO 11.7, FeO 3.3, MgO 0.3, CaO 18.2%.
	1.658	(<u>1.687</u>)	1.710	.052	MAGNESIUM ASTROPHYLLITE $(\text{K},\text{Na})_4\text{Mg}_2(\text{Fe},\text{Mn})_5\text{Ti}_2$ $\text{Si}_8\text{O}_{24}(\text{O},\text{OH},\text{F})_7$	82°	$Y = b$ $Z:c = \frac{b}{a} = -6^\circ$	TCL	100, 010 perf	Straw yellow	---	Pleoc, X bright yellow, Z gray, $Z > Y > X$.
	1.632	<u>1.688</u>	1.738	.106	SODIUM-ZIPPEITE $\text{Na}_4(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10}\cdot 4\text{H}_2\text{O}$	80° (84+2°)	$Z = c$	ORTH plates	010 perf	Yellow to orange	H 2	Diss by acids. Tw common. Pleoc, X cols, Y pale yellow, Z yellow. Fluor bright yellow in UV.

v 1.713	1.658	(1.688)	1.714	.056	ALVANITE $\text{Al}_6(\text{VO}_4)_2(\text{OH})_{12} \cdot 5\text{H}_2\text{O}$	80-85° r < v str	X:b = 14° el pos and neg	MCL	010 perf	Bluish- green to bluish- black	H 3-3.5 G 2.41	Diss by hot acids. Poly tw.
	1.664	1.688	(1.707)	(.043)	WEILITE CaHAsO_4	82°	Z:c = 20° Y:c = 27°	TCL	---	White	G 3.48	Diss by acids.
	1.641	1.690	1.705	.064	LANGITE $\text{Cu}_4\text{SO}_4(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	56-70°	X = c Y = b	ORTH	001 good 010 dist	Sky-blue to blue- green	H 3 G 3.31	Diss by acids. Poly tw 110 common. Pleoc wk, abs Y > Z > X.
	1.640	1.690	1.690	.050	SHUBNIKOVITE $\text{Ca}_2\text{Cu}_8(\text{AsO}_4)_6\text{Cl}(\text{OH}) \cdot 7\text{H}_2\text{O}$ (?)	Small r < v	---	ORTH(?) platy	---	Light blue	H 2 F easy	---
	1.660	1.690	1.698	.038	BERGENITE $(\text{Ba}, \text{Ca})_2(\text{UO}_2)_3(\text{PO}_4)_2$ $(\text{OH})_4 \cdot 5.5\text{H}_2\text{O}$	> 45° (54+7°)	---	MCL tab	---	Yellow	G (4.69)	Fluor pale green in short-wave UV.
v 1.726	1.673	1.690	1.705	.032	TYROLITE $\text{CaCu}_5(\text{AsO}_4)_2(\text{CO}_3)$ $(\text{OH})_4 \cdot 6\text{H}_2\text{O}$	70° (86+7°) r > v str	X = b Z = a	ORTH laths	001 perf	Green to blue	H 1.5 G 3.1-3.2 F 2.5	Diss by HNO_3 or NH_4OH . Pleoc, X cols, Y and Z green.
	1.666	1.690	1.690	.024	CALUMETITE $\text{Cu}(\text{Cl}, \text{OH})_2 \cdot 2\text{H}_2\text{O}$	2°	X = c Y = a el cTv pos	ORTH	001 good	Azure- blue	H 2	Diss by acids. Pleoc wk, in blue, abs Z > Y > X.
1.652 ∇ 1.723	1.675	1.690	1.693	.018	SINCOSITE $\text{CaV}^{+4}_2(\text{PO}_4)_2(\text{OH})_4 \cdot 3\text{H}_2\text{O}$	Med r > v str	---	MCL ps tet tab	001 good 100, 110 poor	Emerald- green to golden	H 3 G 2.84 fus	Diss by acids. Pleoc, X nearly cols, Z gray- green.
	1.679	1.690	1.695	.016	LAVENTITE $(\text{Na}, \text{Ca})_3\text{ZrSi}_2\text{O}_7$ $(\text{O}, \text{OH}, \text{F})_2$	68° r < v wk	Y = b X:c = -20°	MCL tab on 100	100 good	Yellow, brown	H 6 G 3.41 fus	Diff diss by acids. Tw pl 100, lam.
	1.682	1.690	1.697	.015	CHLOROPHOENICITE $(\text{Mn}, \text{Zn})_5\text{AsO}_4(\text{OH})_7$	83° r > v str	Y = b	MCL pris	100 perf	Gray- green	H 3-3.5 G 3.55 F diff	Diss by acids. Color pink, purple-red in str artificial light.
v (1.720)	1.653	(1.691)	1.691	.038	FERRI-ANNITE (Mica grp) $\text{K}(\text{Fe}^{+2}, \text{Mg})_3(\text{Fe}^{+3}, \text{Al})$ $\text{Si}_3\text{O}_{10}(\text{OH})_2$	0-10°	---	MCL	001 perf	Reddish- brown	---	Pleoc, X reddish- brown, Y and Z pale yellow-green. Tw common.
	1.675 ∇ 1.722	1.670	1.691	1.692	.022	DUMORTIERITE $\text{Al}_7(\text{BO}_3)(\text{SiO}_4)_3\text{O}_3$	Small r < v	Z = a X = c	ORTH acic c	100 dist	Blue	H 7 G 3.3 infus
	1.670	1.692	1.713	.043	PARAKELDYSHITE $\text{Na}_2\text{ZrSi}_2\text{O}_7$	84°	X ~ b Y ~ c	TCL	001 perf 110, 110 good	Cols	H 5.5-6 G 3.39	Fluor cream in short- wave UV.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.710 ∨	1.680	1.692	1.720	.040	VYUNTSPAKHKITE Y ₄ Al ₂ AlSi ₅ O ₁₈ (OH) ₅	68°	---	MCL	---	Cols	H 6-7 G 4.02	Insol in HCl.
	1.670	1.692	1.701	.031	KAERSUTITE (Amphibole grp) NaCa ₂ (Mg,Fe ⁺²) ₄ TiSi ₆ Al ₂ O ₂₂ (OH) ₂	81° (64+8°)	Y = $\frac{b}{c}$ Z:c = 9° el pos	MCL pris	110 perf at 124°	Dark brown	H 6 G 3.3	Insol in acids. Pleoc, X yellow-brown, Y red-brown, Z dark red-brown. FeO 5.7, Fe ₂ O ₃ 3.3, Al ₂ O ₃ 14.2, TiO ₂ 5.7%.
	---	1.692	1.699	---	KECKITE Ca(Mn,Zn) ₂ Fe ⁺³ ₃ (PO ₄) ₄ (OH) ₃ ·2H ₂ O	---	X:c = 15-22° Z = $\frac{b}{c}$	MCL	001, 100	Brown	H 4.5 G 2.6	Pleoc, X red-brown, Y yellow, Z bright yellow, abs X > Y > Z.
1.685 ^	1.672	1.693	1.710	.038	JAGOWERITE BaAl ₂ (PO ₄) ₂ (OH) ₂	100° (83+6°)	---	TCL	100, 0T1 good 02T fair	Light green	H 4.5 G 4.01	Insol in acids. Fluor green-white in UV. (Reported as opt pos.)
	1.674	1.693	1.700	.026	FERRO-FERRI-TSCHERMAKITE (Hornblende ser, Amphibole grp) Ca ₂ (Fe ⁺² ,Mg) ₃ Fe ⁺³ ₂ Si ₆ Al ₂ O ₂₂ (OH) ₂	62°	Y = $\frac{b}{c}$ el pos	MCL pris	110 perf at 124°	Green	H 5.5 G 3.2	Insol in acids. Pleoc. FeO 20.4, Fe ₂ O ₃ 5.5, TiO ₂ 1.8%.
	1.687 ∧ 1.695	1.687	1.693	1.697	.010	TINZENITE (Axinite grp) (Ca,Fe,Mn) ₃ Al ₂ BSi ₄ O ₁₅ (OH)	Med	X ⊥ T11	TCL	100 good	Yellow	H 6.5 G 3.39 F 2
1.664 ∧ 1.720	---	1.693	---	.005	PHARMACOSIDERITE KFe ₄ (AsO ₄) ₃ (OH) ₄ ·6-7H ₂ O	Large r > v	---	MCL ps cub	100 poor	Yellow to brown	H 2.5 G 3.0	Diss by HCl. Abnormal interf colors in blue and brown. Crystals zoned.
	1.674	1.694	1.706	.032	KIRSCHSTEINITE (Monticellite ser) Ca(Fe,Mg)SiO ₄	65° (75+7°) r > v	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH	010 poor	Cols	H 5-5.5 G 3.31 F 6	Gel with acids. FeO 19.3, MnO 1.1, Fe ₂ O ₃ 1.4%.
	1.680 ^	1.540	1.695	1.703	.163	ARAGONITE (Aragonite grp) (Ca,Pb)CO ₃	23° r < v wk	X = $\frac{c}{a}$ Z = $\frac{b}{c}$	ORTH acic	010 dist	Cols	H 3-4 G 3.05 infus
1.718 ∨	1.620	1.695	1.730	.110	ANCYLITE SrCe(CO ₃) ₂ (OH)·H ₂ O	70° (66+2°) r < v	X = $\frac{a}{b}$ Z = $\frac{c}{a}$	ORTH ps oct	---	Rose, pale yellow	H 4.5 G 3.82 infus	Diss by acids with eff.

	1.65	---	1.72	.07	OTWAYITE $\text{Ni}_2\text{CO}_3(\text{OH})_2 \cdot \text{H}_2\text{O}$	---	X fib el neg	ORTH fib ro- settes	---	Bright green	G 3.41 infus	Diss slowly in dil HCl. Pleoc wk, abs Z > X. Opt char unk.
	1.668	<u>1.695</u>	1.698	.030	BOLTWOODITE $(\text{H}_3\text{O})\text{K}(\text{UO}_2)\text{SiO}_4 \cdot \text{H}_2\text{O}$	Large ($36 \pm 11^\circ$)	---	MCL ps orth fib <u>b</u>	010 perf 001 poor	Pale yellow	H 3.5-4 G 3.6	Abnormal blue interf colors. Fluor dull green in UV.
1.658 ^	1.682	<u>1.695</u>	1.707	.025	JAHNSITE $\text{CaMn}(\text{Mg}, \text{Fe}^{+2})\text{Fe}^{+3}_2$ $(\text{PO}_4)_4(\text{OH})_2 \cdot 8\text{H}_2\text{O}$	Large	Z = <u>b</u> el pos	MCL pris tab	001 good	Brown to yellow	H 4 G 2.72	Diss by acids. Tw 001. Pleoc, X brown, Y red- brown, Z yellow, abs Y > Z > X. Fe_2O_3 22.8, MnO 10.2, CaO 2.6%.
	1.67	<u>1.695</u>	1.698	.02	MARICITE NaFePO_4	43° r > v wk	X = <u>a</u> Y = <u>b</u>	ORTH	---	Cols to gray to pale brown	H 4-4.5 G 3.66	---
1.680 ^ 1.713	1.685	<u>1.695</u>	1.700	.015	BRONZITE (Orthopyroxene ser, Pyroxene grp) $(\text{Mg}, \text{Fe})_2\text{Si}_2\text{O}_6$	69° r > v	Y = <u>a</u> Z = <u>c</u> el pos	ORTH pris <u>c</u>	210 good at 87°	Green, brown	H 5.5 G 3.43 F 5	Nearly insol in acids. Faint pleoc common. FeO 17.1, Fe_2O_3 0.95, MnO 0.4% (En_{73}).
1.675 ^ 1.708	1.682	<u>1.695</u>	1.697	.015	BUSTAMITE $(\text{Mn}, \text{Ca})_3\text{Si}_3\text{O}_9$	43°	X:a ~ 15° Y:b ~ 35° Z:c ~ 35°	TCL	100 perf 110, 110 good	Pink	H 6 G 3.39	MnO 31.7, CaO 18.2, FeO 0.5, MgO 1.2%.
1.683 ^	1.685	<u>1.695</u>	1.698	.013	ARFVEDSONITE (Amphibole grp) $(\text{Na}, \text{Ca})_3(\text{Fe}^{+2}, \text{Mg})_4$ $\text{Fe}^{+3}\text{Si}_8\text{O}_{22}(\text{OH}, \text{F})_2$	69°	X:c = 29° el neg	MCL	110 perf at 124°	Dark green	H 5-6 G 3.38	Insol in acids. Pleoc, X and Z dark green, Y brown. FeO 20.6, Fe_2O_3 9.1, MnO 2.3, Na_2O 6.8, K_2O 3.6%.
	1.684	<u>1.695</u>	1.698	.014	KEMPITE $\text{Mn}_2\text{Cl}(\text{OH})_3$	Med	X = <u>c</u> Y = <u>b</u>	ORTH pris	---	Emerald- green	H 3.5 G 2.94	Diss by HCl.
1.693 ^ 1.701	1.687	<u>1.695</u>	1.698	.011	FERROAXINITE $\text{Ca}_2(\text{Fe}, \text{Mn}, \text{Mg})\text{Al}_2\text{BSi}_4$ $\text{O}_{15}(\text{OH})$	70°	X ~ \perp T11	TCL	100 good	Brownish- blue	H 6.5 G 3.28 F 2	Insol in acids. CaO 19.4, FeO 7.7, MnO 2.3, MgO 1.7, Fe_2O_3 2.2%.
1.674 ^	1.682	<u>1.696</u>	1.699	.017	KORNERUPINE $\text{Mg}_3\text{Al}_6(\text{Si}, \text{Al}, \text{B})_5\text{O}_{21}$ (OH)	48° r > v wk	X = <u>c</u> Z = <u>b</u> el neg	ORTH pris	110 good	Greenish- black	H 6.5-7 G 3.45 F 6	Insol in acids. Pleoc, X cols to green, Y yellow, Z dark green.
1.662 ^ 1.700	1.694	<u>1.697</u>	1.698	.004	RIEBECKITE (Amphibole grp) $\text{Na}_2(\text{Fe}^{+2}, \text{Mg})_3\text{Fe}^{+3}_2\text{Si}_8$ $\text{O}_{22}(\text{OH})_2$	81°	Y = <u>b</u> X ~ <u>c</u> el neg	MCL pris	110 perf at 124°	Dark blue	H 5 G 2.94	Insol in acids. Pleoc, X and Y dark blue, Z yellow-green. FeO 9.5, Fe_2O_3 18.3, MnO 6.0%.
1.692 ^ 1.704	1.669	<u>1.698</u>	1.705	.036	SINHALITE MgAlBO_4	56° disp str	---	ORTH	---	Brown to yellow	H 7 G 3.50	Insol in acids. Pleoc, X dark brown, Y green, Z pale brown.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.656	<u>1.699</u>	1.731	.075	KUPLETSKITE (K,Na) ₃ (Mn,Fe) ₇ (Ti,Nb) ₂ Si ₈ O ₂₄ (OH) ₇	79° r > v str	el pos	TCL	001 perf	Dark brown	H 3 G 3.22 F easy	Gel with acids. Pleoc, X orange-yellow, Z brown, abs Z > X. MnO 23.6, FeO 5.4, Fe ₂ O ₃ 7.8%.
1.646 ^ 1.713	1.639	(<u>1.699</u>)	1.704	.065	SUSSEXITE (Mn,Mg)BO ₂ (OH)	Small r > v	el neg	ORTH fib <u>c</u>	---	Brown	H 3 G 3.12 F 3	Slowly diss by acids. MnO 37.6, ZnO 3.9%.
1.631 ^	1.688	<u>1.699</u>	1.704	.016	FERRO-RICHTERITE (Amphibole grp) Na ₂ Ca(Fe ⁺² ,Mg) ₅ Si ₈ O ₂₂ (OH) ₂	35° (68+15°) r < v	Y = b ext <u>Tn</u> - clined el pos	MCL	110 perf at 124°	Brown to dark green	H 5-6 G 3.4	Insol in acids. Pleoc. FeO 27.3, Fe ₂ O ₃ 6.2, MnO 1.1, CaO 6.1%.
1.724 v	1.660	<u>1.700</u>	1.701	.041	PHOSPHURANYLITE Ca(UO ₂) ₃ (PO ₄) ₂ (OH) ₂ · 6H ₂ O	5-20° r > v str	Z = c X = <u>a</u>	ORTH laths	100 perf 010 good	Golden-yellow	H 2.5 G 4.1	Pleoc, X cols to pale yellow, Y and Z golden-yellow. Not fluor in UV.
1.682 ^	1.679	<u>1.700</u>	1.719	.040	GRUNERITE (Cummingtonite ser, Amphibole grp) (Fe,Mg) ₇ Si ₈ O ₂₂ (OH) ₂	86° r < v	Y = b Z:c = 14° el pos	MCL pris	110 perf at 124°	Greenish-black	H 6 G 3.4	Pleoc, X and Y cols, Z yellow to brown. FeO 45.0, MnO 0.4%.
1.713 v	1.684	<u>1.700</u>	1.703	.019	TARAMITE, var Mboziite (Amphibole grp) Na ₂ Ca(Fe ⁺² ,Mg) ₃ Al ₂ (Si ₆ Al ₂)O ₂₂ (OH) ₂	(46+15°)	Y = b	MCL	110 perf at 124°	Blue-green	---	Pleoc, X light yellow, Y violet gray, Z blue. FeO 13.0, Fe ₂ O ₃ 9.6, Al ₂ O ₃ 15.8%.
1.684 ^	1.696	<u>1.700</u>	1.702	.006	LITHIOPHILITE (Triphylite grp) Li(Mn,Fe)PO ₄	~ 70° r < v	X = c Y = <u>a</u>	ORTH	100 perf	Brown, salmon	H 4-5 G 3.55 F 1.5	Diss by acids. MnO 37.3, FeO 8.6, MgO 0.7%.
1.693 ^	1.693	<u>1.701</u>	1.704	.011	TINZENITE (Axinite grp) (Ca,Mn,Fe) ₃ Al ₂ BSi ₄ O ₁₅ (OH)	63° r < v	X \perp T11	TCL	100 good	Yellow	H 6.5-7 G 3.29	Insol in acids. MnO 21.2, CaO 12.5, Fe ₂ O ₃ 1.6%.
1.693 ^	1.695	<u>1.702</u>	1.708	.013	BARYLITE BaBe ₂ Si ₂ O ₇	70°	X = b Y = <u>c</u>	ORTH	001, 100 good	White	H 7 G 4.07 infus	Insol in acids. Fluor bright blue in UV.

1.680 ∧ 1.720	1.683	<u>1.704</u>	1.722	.039	CHRYSLITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	85°	X = $\frac{b}{a}$ Z = $\frac{a}{b}$	ORTH	010 fair	Green	H 7 G 3.53 infus	Gel with acids. FeO 21.6, MnO 0.3, Fe ₂ O ₃ 1.6% (Fe ₇₅).
1.720	1.680	<u>1.704</u>	1.712	.032	PARNAUITE Cu ₉ (AsO ₄) ₂ (OH) ₁₀ ·7H ₂ O	60°	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH blades on 010	---	Green, blue, blue- green	H 2 G 3.09	Pleoc, X pale green, Y yellow-green, Z blue- green, abs Z > Y > X.
1.685 ∧ 1.735	1.596	<u>1.705</u>	1.705	.109	STILPNOMELANE K(Fe ⁺² , Fe ⁺³ , Mg, Al) ₁₀ Si ₁₂ O ₃₀ (OH) ₁₂	~ 0°	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el cTv pos	MCL or TCL	001 perf	Dark brown	H 3-4 G 2.82 F 4	Dec by acids. Pleoc, X yellow, Y and Z dark brown. FeO 13.2, Fe ₂ O ₃ 23.9%.
1.685 ∧	1.660	<u>1.705</u>	1.713	.053	TARBUTITE Zn ₂ (PO ₄)(OH)	50°	Disp str	TCL pris	001 perf	Cols, yellow	H 4 G 4.12 F easy	Diss by acids.
1.654 ∧	1.675	<u>1.705</u>	1.705	.030	KIVUITE (Th, Ca, Pb)H ₂ (UO ₂) ₄ (PO ₄) ₂ (OH) ₈ ·7H ₂ O	0-5° r > v	---	ORTH(?)	---	Yellow	---	Dec by HNO ₃ . Pleoc, X cols, Y and Z greenish-yellow.
	1.690	<u>1.705</u>	1.711	.021	YFTISITE (Y, Dy, Er) ₄ (Ti, Sn)O (SiO ₄) ₂ (F, OH) ₆	Large r > v dist	---	ORTH	---	Yellowish	H 3.5-4 G 3.96	Abnormal blue interf color.
	1.678	<u>1.706</u>	1.721	.043	ERNSTITE (Mn ⁺² _{1-x} Fe ⁺³ _x)Al(PO ₄) (OH) _{2-x} O _x	74° r > v	Z = $\frac{b}{c}$ Y:c = -4°	MCL radi- ating	010, 100 good	Yellow- brown	H 3-3.5 G 3.07	Oxidation product of Eosphorite. Untw. Pleoc, X yellow-brown, Y red-brown, Z pale yellow.
	1.688	<u>1.707</u>	1.725	.037	LERMONTOVITE U ⁺⁴ (PO ₄)(OH)·H ₂ O	(88+6°)	Z = $\frac{c}{b}$	ORTH fib	---	Gray- green	---	Opt sign not given.
1.723	1.704	<u>1.707</u>	1.710	.006	SAPPHIRINE (Mg, Al) ₈ (Si, Al) ₆ O ₂₀	85° r < v str	Y = $\frac{b}{c}$ Z:c = 12°	MCL tab	010, 100 poor	Pale blue to green	H 7.5 G 3.50 infus	Insol in acids. Pleoc, X pale greenish-blue, Y blue to dark green, Z blue to yellow-green. FeO 1.5, Fe ₂ O ₃ 0.6%.
1.707 pos	1.689	<u>1.707</u>	1.727	.038	GAITITE Ca ₂ (Zn, Mg)(AsO ₄) ₂ · 2H ₂ O	85° (92+6°)	---	TCL	010, 001, 011 good	Cols, white	H ~ 5	Zn:Mg = 1:1.
1.695 ∧	1.695	<u>1.708</u>	1.710	.015	BUSTAMITE (Mn, Ca, Fe) ₃ Si ₃ O ₉	34° r < v	X:a ~ 15° Y:b ~ 35° Z:c ~ 35°	TCL	100 perf 110, 110 good	Pink	H 6 G 3.46	MnO 30.4, CaO 12.8, FeO 6.9, MgO 1.2%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.687 ^ 1.731	1.688	<u>1.709</u>	1.714	.026	HASTINGSITE (Amphibole grp) $\text{NaCa}_2(\text{Fe}^{+2}, \text{Mg})_4\text{Fe}^{+3}\text{Si}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	51°	$Y = b$ $Z:c = 12^\circ$ el pos	MCL pris	110 perf at 124°	Dark green	H 5-6 G 3.45	Insol in acids. Pleoc, X greenish-yellow, Y dark olive, Z dark green. FeO 22.9, Fe ₂ O ₃ 5.9, Al ₂ O ₃ 11.5, TiO ₂ 1.7%.
1.692 ^ 1.730	1.687	<u>1.710</u>	1.725	.038	KAERSUTITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{TiSi}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	77°	$Y = b$ $Z:c = 4^\circ$ el pos	MCL pris	110 perf at 124°	Brownish-black	H 6 G 3.3	Insol in acids. Pleoc, X pale yellow-brown, Y red-brown, Z dark red-brown. FeO 4.9, Fe ₂ O ₃ 9.3, Al ₂ O ₃ 15.5, TiO ₂ 5.9%.
1.685 ^	1.695	<u>1.710</u>	1.725	.030	TALMESSITE $\text{Ca}_2(\text{Mg}, \text{Co}, \text{Ni})(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	Large	---	TCL	---	Pink, green	H 5 G 3.57	"Arsenate-belovite." Diss in acids. Pink var is pleoc: cols to rose. Up to 8.5% FeO or NiO.
1.681 ^	1.694	<u>1.710</u>	1.722	.028	FERRO-GEDRITE (Amphibole grp) $(\text{Fe}^{+2}, \text{Mg})_5\text{Al}_2(\text{Si}_6\text{Al}_2)\text{O}_{22}(\text{OH})_2$	82° r > v	$Y = b$ $Z = \bar{c}$ el pos	ORTH pris <u>c</u>	210 perf	Green	H 6 G 3.57 F 4	Insol in acids. Pleoc, X pale green, Y brownish-green, Z bluish-green. Fe ₂ O ₃ 2.3, Al ₂ O ₃ 19.7, FeO 35.5, MnO 2.3%.
1.729 v	1.705	<u>1.711</u>	1.715	.010	CHLORITOID $(\text{Fe}, \text{Mg}, \text{Mn})_2\text{Al}_4\text{Si}_2\text{O}_{10}(\text{OH})_4$	60°	$X = b$ $Z:c = 12^\circ$ el clv pos	MCL	001 perf	Greenish-gray	H 6.5 G 3.5 F diff	Dec by H ₂ SO ₄ . Pleoc, X green, Y blue, Z yellow. FeO 19.9, Fe ₂ O ₃ 4.6%.
1.708 pos ^ 1.738	1.706	<u>1.711</u>	1.714	.008	ALLANITE (Epidote grp) $(\text{Ce}, \text{Ca}, \text{Y})_2(\text{Al}, \text{Fe})_3(\text{SiO}_4)_3(\text{OH})$	61°	$Y = b$	MCL	001, 100, 110 poor	Brown to black	H 6 G 3.8 F 3	Slowly attacked by HCl, gel. Fe ₂ O ₃ 4.9, FeO 9.1, ThO ₂ 1.7%.
1.725 v	1.670	<u>1.712</u>	1.750	.080	WHITMOREITE $(\text{Fe}^{+2}, \text{Mn})\text{Fe}^{+3}_2(\text{PO}_4)_2(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	Rather large	$Y = b$ $Z \sim \bar{c}$	MCL	100 fair	Brown to greenish-brown	H 3 G 2.88	Tw 100. Pleoc wk, X cols, Y pale yellow, Z yellow-green.
(1.699) ^ 1.728	1.642	<u>1.713</u>	1.721	.079	SUSSEXITE $(\text{Mn}, \text{Mg})\text{BO}_2(\text{OH})$	37°	el neg	ORTH fib <u>c</u>	---	Brown	H 5 G 3.05 F 3	Slowly diss by HCl. MnO 40.4%.

1.690 ^	1.654	<u>1.713</u>	1.722	.068	LANGITE $\text{Cu}_4\text{SO}_4(\text{OH})_6 \cdot 2\text{H}_2\text{O}$	$\sim 70^\circ$ (42+5°)	$X = c$ $Y = \underline{b}$	ORTH	001 good 010 dist	Blue to blue- green	H 3 G 3.31	Diss by acids. Poly tw 110. Pleoc wk, abs Y > Z > X.
	1.703	<u>1.713</u>	1.722	.019	GERHARDTITE $\text{Cu}_2\text{NO}_3(\text{OH})_3$	Large r > v str	$X = a$ $Z = \underline{c}$	ORTH hori- zontal stria- tions	001 perf 100 less so	Emerald- green	H 2 G 3.40 F 2	Diss by acids. Pleoc, X and Y green, Z blue.
1.695 ^ 1.728	1.703	<u>1.713</u>	1.717	.014	HYPERSTHENE (Orthopyroxene ser, Pyroxene grp) $(\text{Mg}, \text{Fe})_2\text{Si}_2\text{O}_6$	53°	$Y = a$ $Z = \underline{c}$ el pōs	ORTH	210 good	Green	H 5.5 G 3.53 F 5	Insol in acids. Faint pleoc common. FeO 25.7, Fe_2O_3 1.0% (En ₅₇).
1.700 ^	1.705	<u>1.713</u>	1.715	.010	TARAMITE, var Mboziite (Amphibole grp) $\text{Na}_2\text{Ca}(\text{Fe}^{+2}, \text{Mg})_3\text{Al}_2$ $(\text{Si}_6\text{Al}_2)_0\text{O}_{22}(\text{OH})_2$	54°	$Y = b$ $Z:c = 11^\circ$ el pos	MCL	110 perf at 124°	Blue- green	---	Pleoc, X yellow, Y blue-green, Z deep blue. FeO 18.7, Fe_2O_3 13.1, Al_2O_3 10.3, MnO 1.9%.
1.740 v	1.690	<u>1.714</u>	1.735	.045	SCHOEPITE $\text{UO}_3 \cdot 2\text{H}_2\text{O}$	85° r > v str	$X = c$ $Y = \underline{b}$ el pōs	ORTH tab	001 perf	Lemon- yellow to brownish	H 2.5 G 4.8-5.0	Diss by acids. Pleoc, X nearly cols, Y and Z lemon-yellow to bright yellow. Fluor pale green in UV.
	1.701	<u>1.714</u>	1.720	.019	NIOCALITE $\text{Ca}_4\text{NbSi}_2\text{O}_{10}(\text{O}, \text{F})$	56°	$X = b$ $Z:c = 12^\circ$	MCL	Conch	Lemon- yellow	H 6 G 3.32	Tw.
	1.663	<u>1.715</u>	1.734	.071	RUIZITE $\text{CaMn}^{+3}\text{Si}_2\text{O}_6(\text{OH}) \cdot 2\text{H}_2\text{O}$	60° r > v str inclined	$Y = b$ $Z:c = 44^\circ$	MCL el <u>b</u>	---	Orange to brown	H 5 G 2.9	Diss by warm acids. Tw common 100. Pleoc, X pale orange, Y orange, Z yellow.
	1.667	(<u>1.715</u>)	1.737	.070	Unnamed arsenate- sulfate of Cu	Med large	---	ORTH(?)	---	Blue to blue- green	---	Pleoc, X cols, Y and Z pale green to bluish- green.
	1.665	<u>1.715</u>	1.715	.050	KHIBINSKITE $\text{K}_2\text{ZrSi}_2\text{O}_7$	6-16°	$Z = b$ $X:c = 34^\circ$	MCL mass	6 clv	Cols	H 4.5-5.5 G 3.35	---
┌	1.704	<u>1.715</u>	1.724	.020	XANTHOXENITE $\text{Ca}_4\text{Fe}_2(\text{PO}_4)_4(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	Large r < v str	---	Plates	---	Yellow to brown	H 2.5 G 2.97	Pleoc, X cols, Z yellow. Abnormal blue interf colors. Fe_2O_3 21.1, MnO 4.5%.
	1.704	<u>1.715</u>	1.718	.014	VANMEERSSCHEITE $\text{U}(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_6 \cdot$ $4\text{H}_2\text{O}$	56°	---	ORTH tab	010 good 100 less so	Yellow	G (4.67)	Pleoc in yellow.
	1.603	<u>1.716</u>	1.742	.139	DUNDASITE $\text{PbAl}_2(\text{CO}_3)_2(\text{OH})_4 \cdot 2\text{H}_2\text{O}$	Large (48+2°)	$X = a$ $Y = \underline{b}$	ORTH aggre- gates	010 perf	White	H 2 G 3.55 infus	Diss by acids with eff.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
$\begin{smallmatrix} \perp \\ 1.722 \end{smallmatrix}$	1.682	<u>1.716</u>	1.729	.047	GLAUCOCHROITE CaMnSiO_4	61° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{a}$	ORTH pris	---	Pale pink to violet	H 6 G 3.41 F 3	Gel with acids. MnO 32.7, FeO 4.2, MgO 2.1%.
$\begin{smallmatrix} \vee \\ 1.779 \end{smallmatrix}$	1.695	<u>1.716</u>	1.725	.030	SONOLITE, zincian $(\text{Mn,Zn})_9(\text{SiO}_4)_4(\text{OH,F})_2$	Large	---	MCL	---	Dark brown to brownish-black	H 5.5 G 3.8	Gel with acids. Tw common 001.
	1.700	<u>1.716</u>	1.726	.026	WOHLERITE $\text{NaCa}_2(\text{Zr,Nb})\text{Si}_2\text{O}_8$ (O,OH,F)	71-79° $r < v$ dist	$Z = b$ el cTv neg	MCL tab	010 dist	Yellow-brown	H 6 G 3.42 F 3.5	Gel with HCl. Pleoc wk, X and Y cols, Z yellow.
$\begin{smallmatrix} \vee \\ 1.748 \end{smallmatrix}$	1.703	<u>1.716</u>	1.721	.018	JULGOLDITE (Pumpellyite grp) $\text{Ca}_2(\text{Fe}^{+2},\text{Mg})(\text{Fe}^{+3},\text{Al})_2$ $(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{OH})_2 \cdot \text{H}_2\text{O}$	Med	---	MCL	100, 001 good	Green	H 4.5 G 3.3	Gel with hot HCl. Pleoc, X and Z pale green, Y blue-green. Tw 001.
$\begin{smallmatrix} \vee \\ 1.739 \end{smallmatrix}$	1.709	<u>1.716</u>	1.723	.014	THALENITE $\text{Y}_3\text{Si}_3\text{O}_{10}(\text{OH}) (?)$	70° (90+17°) $r < v$	$Y:c$ small	MCL pris	110 perf	Reddish-black to black	H 6 G 4.16	---
$\begin{smallmatrix} 1.710 \\ \vee \\ 1.731 \end{smallmatrix}$	1.657	<u>1.717</u>	1.767	.110	ZIPPEITE $\text{K}_4(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 4\text{H}_2\text{O}$	Large	$Z = c$ $X = \frac{b}{a}$ el cTv pos	ORTH plates	010 perf	Yellow to orange	H 2 G 3.66	Diss by acids. Tw common. Pleoc, X cols, Y pale yellow, Z yellow. Fluor bright yellow in UV.
$\begin{smallmatrix} \square \\ \end{smallmatrix}$	1.685	<u>1.717</u>	1.720	.035	KOZULITE (Amphibole grp) $(\text{Na,K})_3(\text{Mn,Mg})_4$ $(\text{Fe}^{+3},\text{Al})\text{Si}_8\text{O}_{22}(\text{OH})_2$	35° $r > v$ wk	$Y = b$ $X:c = 25^\circ$ el neg	MCL pris	110 perf	Reddish-black to black	H 5 G 3.30	Pleoc, X yellow-brown, Y red-brown, Z dark brown, abs $Z > Y > X$. MnO 2.8, Fe ₂ O ₃ 2.9%.
$\begin{smallmatrix} 1.695 \\ \downarrow \\ 1.734 \end{smallmatrix}$	1.621	<u>1.718</u>	1.745	.124	ANCYLITE $\text{SrCe}(\text{CO}_3)_2(\text{OH}) \cdot \text{H}_2\text{O}$	70° (53+2°) $r < v$ wk	$X = \frac{a}{a}$ $Z = \frac{c}{c}$	ORTH	---	Yellow, pink, brown	H 4.5 G 4.00 infus	Diss by acids with eff.
	1.709	<u>1.718</u>	1.734	.025	METAHOHMANNITE $\text{Fe}_2(\text{SO}_4)_2(\text{OH})_2 \cdot 3\text{H}_2\text{O}$	(86+9°)	---	Mass, powdery	---	Orange	---	Diss by HCl. Pleoc, X pale yellow, Y reddish-yellow, Z reddish-brown. Reported as biax pos.

1.705 □ 1.73	1.715	<u>1.719</u>	1.720	.005	VESUVIANITE $\text{Ca}_{10}\text{Mg}_2\text{Al}_4(\text{SiO}_4)_5$ $(\text{Si}_2\text{O}_7)_2(\text{OH})_4$	30-60°	---	TET	110 poor	Yellow, brown, green	H 6.5 G 3.4 F 3	Partly dec by HCl. Anom biax, may show ab- normal interf colors in blue or brown.
	1.660	<u>1.720</u>	1.728	.068	HYDROASTROPHYLLITE $(\text{H}_3\text{O}, \text{K}, \text{Ca})(\text{Fe}, \text{Mn})_{5-6}$ $\text{Ti}_2\text{Si}_6(\text{O}, \text{OH})_{31}$	40°	---	TCL	2 sets	Dark brown	G 3.15	Pleoc, X bright yellow, Y orange- yellow, Z dull yellow, abs $Z > Y > X$.
(1.691) ^	1.677	(<u>1.720</u>)	1.721	.044	FERRI-ANNITE (Mica grp) $\text{K}(\text{Fe}^{+2}, \text{Mg})_3(\text{Fe}^{+3}, \text{Al})$ $\text{Si}_3\text{O}_{10}(\text{OH})_2$	Small	---	MCL	001 perf	Reddish- brown	---	Pleoc, X brownish- red, Y and Z pale greenish-brown. Tw common.
1.694 △ 1.734	1.689	<u>1.720</u>	1.728	.039	KIRSCHSTEINITE (Monticellite ser) $\text{Ca}(\text{Fe}, \text{Mg})\text{SiO}_4$	51° $r > v$	$X = \frac{b}{c}$ $Y = \frac{c}{a}$	ORTH	010 poor	Gray	H 5.5 G 3.43 F 6	Gel with acids. FeO 29.3, Fe_2O_3 0.7, MgO 4.9, MnO 1.65%.
1.704 ◇ 1.733	1.698	<u>1.720</u>	1.736	.038	HYALOSIDERITE (Olivine grp) $(\text{Mg}, \text{Fe})_2\text{SiO}_4$	80°	$X = \frac{b}{c}$ $Z = \frac{a}{b}$	ORTH	010 fair	Green	H 7 G 3.60 infus	Gel with acids. FeO 27.8, Fe_2O_3 2.1, MnO 0.4% (Fo ₆₆).
	1.701	<u>1.720</u>	1.734	.033	HOWIEITE $\text{Na}(\text{Fe}, \text{Mn})_{10}(\text{Fe}^{+3}, \text{Al})_2$ $\text{Si}_{12}\text{O}_{31}(\text{OH})_{13}$	65° (80+7°) $r < v$ str	---	TCL	010 good 100 fair	Dark green	G 3.38	Pleoc, X pale golden yellow, Y dark lilac gray, Z dull green.
1.717 pos ◇ 1.736	1.713	<u>1.720</u>	1.727	.014	EPIDOTE (Epidote grp) $\text{Ca}_2(\text{Al}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	89° $r > v$ str	$X \sim \frac{c}{b}$ $Y = \frac{b}{c}$ el cTv pos	MCL el <u>b</u>	001 perf	Light green	H 7 G 3.40 F 4	Insol in acids. Fe_2O_3 6.6, FeO 0.4%.
1.704 ^	1.667	<u>1.720</u>	1.737	.070	PARNAUTE $\text{Cu}_9(\text{AsO}_4)_2(\text{SO}_4)(\text{OH})_{10} \cdot$ $7\text{H}_2\text{O}$	60° $r < v$	$X = \frac{b}{c}$ $Y = \frac{a}{b}$	ORTH blades	---	Green, blue	H 2 G 3.09	Pleoc, X cols, Z greenish-blue.
	1.715	<u>1.720</u>	1.725	.010	TRIMERITE $(\text{Ca}, \text{Mn})\text{BeSiO}_4$	83°	Opt pl and $X \sim \perp$ 001	TCL ps hex tab	001 dist	Cols	H 6 G 3.47 F diff	Gel with HCl. Basal section shows 3 radial segments.
	1.713	<u>1.721</u>	1.734	.021	HARADAITE SrVSi_2O_7	75° (103+12°) $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	010 perf 100, 001 dist	Bright green	H 4.5 G 3.80	Pleoc, X cols to pale green, Y cols to light yellow-green, Z bluish- green.
1.717 □ 1.724	1.713	<u>1.721</u>	1.728	.015	KYANITE Al_2SiO_5	83° $r > v$ wk	$Z:c = -30^\circ$ on 100 el pos	TCL bladed	100 perf 010 good	Bluish, greenish, cols	H 5.5-7 G 3.59 infus	Insol in acids. Pleoc wk, X cols, Y and Z pale blue. Tw pl comp face 100.
	1.687	<u>1.722</u>	1.731	.044	TARAPACAITE K_2CrO_4	52° $r > v$ wk	$X = \frac{b}{c}$ $Y = \frac{a}{b}$	ORTH ps hex	010 perf 001 good	Bright yellow	G 2.74	Sol in H_2O .

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.691 ^	1.686	<u>1.722</u>	1.723	.037	DUMORTIERITE $Al_7(BO_3)(SiO_4)_3O_3$	13° $r > v$ wk	$Z = \frac{a}{c}$ $X = \frac{c}{c}$	ORTH acic <u>c</u>	100 dist	Brown	H 7 G 3.41 infus	Insol in acids. Tw pl 001. Pleoc, X black, Y deep red-brown, Z brown.
	1.716	(<u>1.723</u>)	1.723	.007	HATRURITE Ca_3SiO_5	Small	---	ps hex	---	Cols	---	Dec by H_2O .
	1.675	<u>1.723</u>	1.765	.090	CLINOCHALCOMENITE $CuSiO_3 \cdot 2H_2O$	78° ($84+3^\circ$) $r < v$	$Y = \frac{b}{c}$ $Z:c = 10^\circ$ el pos	MCL	110 perf	Bluish-green	H 2 G 3.35	Pleoc, X and Y cols, Z bluish-green.
1.690 ^ 1.746	1.698	<u>1.723</u>	1.745	.047	LAVENTITE $(Na,Ca)_3ZrSi_2O_7$ (O,OH,F) ₂	80° $r < v$	$Y = \frac{b}{c}$ $X:c = -20^\circ$	MCL tab on 100	100 good	Yellow, brown	H 6 G 3.5 fus	Diff diss by acids. Tw pl 100 lam. Pleoc, X cols to pale yellow, Y light yellow, Z golden yellow.
280 1.707 ^ 1.741	1.720	<u>1.723</u>	1.725	.005	SAPPHIRINE $(Mg,Al)_8(Si,Al)_6O_{20}$	52° $r < v$ str	$Y = \frac{b}{c}$ $Z:c = 7^\circ$	MCL tab	010, 100 poor	Pale blue to green	H 7.5 G 3.58 infus	Insol in acids. Pleoc, X nearly cols, Y and Z blue, abs $Z > Y > X$.
1.700 ^	1.674	<u>1.724</u>	1.724	.050	PHOSPHURANYLITE $Ca(UO_2)_3(PO_4)_2(OH)_2 \cdot 6H_2O$	$5-10^\circ$ $r > v$ str	$X = \frac{a}{c}$ $Z = \frac{c}{c}$	ORTH laths	100 perf 010 good	Golden-yellow	H 2.5 G 4.1	Diss by acids. Pleoc, X cols to pale yellow, Y and Z golden-yellow. Not fluor in UV.
1.712 ^	1.676	<u>1.725</u>	1.745	.069	WHITMOREITE $(Fe^{+2},Mn)Fe^{+3}_2(PO_4)_2(OH)_2 \cdot 4H_2O$	$60-65^\circ$	$Y = \frac{b}{c}$ $Z \sim \frac{c}{c}$	MCL	100 fair	Brown to greenish-brown	H 3 G 2.87	Tw 100. Pleoc, X and Y light greenish-brown, Z dark greenish-brown.
1.723 □ 1.728	1.692	<u>1.725</u>	1.738	.046	PHOSPHOSIDERITE $FePO_4 \cdot 2H_2O$	62° $r > v$ str	$Y = \frac{b}{c}$ $X:c = 4^\circ$	MCL tab 010	010 good	Reddish-violet	H 3.5-4 G 2.76 F easy	Diss by acids. Tw 101. Pleoc, X pale rose, Y carmine-red, Z cols.
	1.700	<u>1.725</u>	1.730	.030	VUAGNATITE $CaAlSiO_4(OH)$	48° $r < v$ str	$X = \frac{c}{b}$ $Y = \frac{b}{c}$ el neg	ORTH	---	White	G 3.22-3.42	---
1.690 ^	1.694	<u>1.726</u>	1.730	.036	TYROLITE $CaCu_5(AsO_4)_2(CO_3)(OH)_4 \cdot 6H_2O$	36° $r > v$ str	$X = \frac{b}{a}$ $Z = \frac{a}{a}$	ORTH laths	001 perf	Pale green to sky-blue	H 2 G 3.27 F easy	Diss by acids or NH_4OH . Pleoc, X and Z grass-green, Y yellow green.
v 1.766	1.711	<u>1.727</u>	1.740	.029	TEPHROITE (Olivine grp) $(Mn,Mg)_2SiO_4$	85° $r > v$	$X = \frac{b}{c}$ $Y = \frac{c}{c}$	ORTH	001 good 010 fair	Reddish-brown to gray	H 5 G 3.72 F 3.5	Gel with acids. MnO 47.1, MgO 18.1, Fe_2O_3 0.3%.

	1.718	(~1.727)	1.728	.010	BARIUM- PHARMACOSIDERITE $\text{Ba}(\text{Fe}, \text{Al})_4(\text{AsO}_4)_3$ $(\text{OH})_5 \cdot 5\text{H}_2\text{O} (?)$	0-39°	---	TET (?)	100 good	Yellow- brown	H 2-3 G 3.00	Diss by acids.
	1.681	1.728	1.769	.088	COBALTOMENITE $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$	83° $r < v$	$Z:c = 13^\circ$	MCL	Conch	Pink	H 2.5 G 3.42	Data for synth compd. Pleoc, X and Y pink, Z red.
1.713 ^	1.670	1.728	1.732	.062	SUSSEXITE $\text{MnBO}_2(\text{OH})$	30°	el neg	ORTH fib <u>c</u>	---	Brown	H 5 G 3.30 F 3	Calc for pure end members.
	1.670	1.728	1.732	.062	SARCOPSIDE $(\text{Fe}^{+2}, \text{Mn}, \text{Mg})_3(\text{PO}_4)_2$	28° $r > v$	---	MCL	001, 100 good 010 poor	Gray to brown	H 4 G 3.79	Diss by acids. Poly tw 001.
v 1.758	1.705	1.728	1.730	.025	HOLTITE $\text{Al}_6(\text{Ta}, \text{Sb}, \text{Li})[(\text{Si}, \text{As})$ $\text{O}_4]_3(\text{BO}_3)(\text{O}, \text{OH})_3$	27° $r < v$ str	$X = \frac{c}{b}$ $Y = \frac{b}{c}$	ORTH pris	100 good	Buff to greenish	H 8.5 G 3.90 infus	Insol in acids. Tw 110. Not pleoc.
1.713 v 1.747	1.715	1.728	1.731	.016	FERROHYPERSTHENE (Orthopyroxene ser, Pyroxene grp) $(\text{Mg}, \text{Fe})_2\text{Si}_2\text{O}_6$	51°	$Y = \frac{a}{c}$ $Z = \frac{c}{b}$ el pos	ORTH	210 good	Green	H 5-6 G 3.60 F 5	Insol in acids. Faint pleoc common. FeO 29.4, MnO 0.2, Fe_2O_3 2.1% (En_{45}).
	1.720	1.728	1.735	.015	LANDESITE $\text{Mn}^{+2}\text{Fe}^{+3}(\text{PO}_4)_2(\text{OH})_3 \cdot$ $3\text{H}_2\text{O} (?)$	Large	el pos	ORTH	010 perf	Reddish- brown	H 3-3.5 G 3.03	Diss by acids. Pleoc, X dark brown, Y light brown, Z yellow.
1.725 └	1.690	1.729	1.750	.060	BERMANITE $\text{Mn}^{+2}(\text{Mn}^{+3}, \text{Fe}^{+3})_2(\text{PO}_4)_2$ $(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	72° $r < v$ str	$X = \frac{b}{c}$ $Y:c = -36^\circ$	MCL ps orth	001 perf 010	Reddish- brown	H 3.5 G 2.85 F easy	Diss by acids. Poly tw. Pleoc, X light red, Y pale yellow, Z deep red. Mn_2O_3 30.6, Fe_2O_3 3.2, MnO 12.8%.
1.711 ^	1.725	1.729	1.730	.005	CHLORITOID $(\text{Fe}, \text{Mg}, \text{Mn})\text{Al}_4\text{Si}_2\text{O}_{10}$ $(\text{OH})_4$	56°	$X = \frac{b}{c}$ $Z:c = 20^\circ$ el clv neg	MCL	001 perf	Greenish- gray	H 6.5 G 3.5 F diff	Dec by H_2SO_4 . Pleoc, X green, Y blue, Z yellow. FeO 22.2, Fe_2O_3 4.9, TiO_2 2.7%.
	1.605	(1.73)	1.738	.133	Unnamed barium uranyl arsenate	Small	---	ORTH(?) acic	---	Yellow	---	Pleoc, X cols, Z yellow. Not fluor in UV. Am. Mineral., 58, 561 (1973).
	1.602	1.730	1.732	.130	MCGUINNESSITE $(\text{Mg}, \text{Cu})_2(\text{CO}_3)(\text{OH})_2$	(very small)	$X:c = 11^\circ$	MCL fib	---	Blue- green to blue	H 2.5 G 3.02- 3.22	Diss by cold acids. Pleoc, X pale green, Y and Z light bluish- green.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.710 ^	1.694	<u>1.730</u>	1.757	.063	KAERSUTITE (Amphibole grp) $\text{NaCa}_2(\text{Mg}, \text{Fe}^{+2})_4\text{TiSi}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	80°	$Y = \frac{b}{c}$ $Z \sim \frac{c}{a}$ el pos	MCL pris	110 perf at 124°	Brownish-black	H 6 G 3.3	Insol in acids. Pleoc, X yellow, Y red-brown, Z dark brown. FeO 7.7, Fe ₂ O ₃ 12.4, TiO ₂ 6.0%.
	(1.690)	<u>1.730</u>	1.749	(.059)	PHURCALITE $\text{Ca}_2(\text{UO}_2)_3(\text{PO}_4)_2(\text{OH})_4 \cdot 4\text{H}_2\text{O}$	68°	el pos	ORTH	001, 010 perf 100	Yellow	H 3 G 4.14	Diss by acids. Pleoc, X bright yellow, Y pale yellow. Not fluor in UV.
1.717 ^	1.674	<u>1.731</u>	1.754	.080	ZIPPEITE $\text{K}_4(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 4\text{H}_2\text{O}$	45° $r > v$ str	$Z = \frac{c}{a}$ $X = \frac{b}{a}$ el cTv pos	ORTH plates	010 perf	Yellow to orange	H 2 G 3.61	Diss by acids. Pleoc, X cols, Y and Z yellow. Fluor bright yellow in UV.
1.709 ^	1.705	<u>1.731</u>	1.732	.027	HASTINGSITE (Amphibole grp) $\text{NaCa}_2(\text{Fe}^{+2}, \text{Mg})_4\text{Fe}^{+3}\text{Si}_6\text{Al}_2\text{O}_{22}(\text{OH})_2$	25°	$Y = \frac{b}{c}$ $Z:c = 13^\circ$ el pos	MCL pris	110 perf at 124°	Dark green	H 5-6 G 3.43	Insol in acids. Pleoc. FeO 22.0, Fe ₂ O ₃ 11.6, Al ₂ O ₃ 11.5, MnO 0.6%.
	1.711	<u>1.731</u>	1.732	.021	CHALCOMENITE $\text{CuSeO}_3 \cdot 2\text{H}_2\text{O}$	Small $r < v$ or $r > v$ str	$X = \frac{a}{b}$ $Y = \frac{c}{a}$	ORTH	None	Clear blue	H 2.5 G 3.36 fus	Diss by acids. Pleoc, X light blue, Y and Z darker blue.
	1.680	<u>1.732</u>	1.775	.095	Unnamed uranyl phosphate	80°	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el cTv neg	ORTH	001 good	Yellow	---	Pleoc, X cols, Y yellow, Z yellow-gold. Fluor green in UV. Am. Mineral., 59, 212 (1974).
1.720 ^ 1.750	1.710	<u>1.733</u>	1.750	.040	HYALOSIDERITE (Olivine grp) $(\text{Mg}, \text{Fe})_2\text{SiO}_4$	79°	$X = \frac{b}{c}$ $Z = \frac{a}{b}$	ORTH	010 fair	Green	H 5 G 3.69 F 6	Gel with acids. FeO 32.0, Fe ₂ O ₃ 2.0, MnO 0.5% (Fe ₆₀).
	1.724	<u>1.733</u>	1.739	.015	CHALCOCYANITE CuSO_4	Large $r > v$	$X = \frac{b}{c}$ $Y = \frac{a}{b}$	ORTH tab	None	Pale green to blue	H 3.5 G 3.65	Diss by H ₂ O.
1.718 ^	1.656	<u>1.734</u>	1.776	.120	ANCYLITE $\text{SrCe}(\text{CO}_3)_2(\text{OH}) \cdot \text{H}_2\text{O}$	70° $r < v$ wk	$X = \frac{a}{b}$ $Z = \frac{c}{a}$	ORTH	---	Yellow, brown, pink	H 4.5 G 4.0 infus	Diss by acids with eff.
	1.679	<u>1.734</u>	1.742	.063	HUEMULITE $\text{Na}_4\text{MgV}_{10}\text{O}_{28} \cdot 24\text{H}_2\text{O}$	28° (41+5°) $r > v$ str	---	TCL	001 perf 010 less so	Yellow-orange	H 2.5 G 2.39	Diss by cold H ₂ O. Not fluor in UV. Pleoc, X light yellow, Y gold, Z yellow-orange.

1.768 v 1.768	1.708	1.734	1.758	.050	ADAMITE (Olivinite grp) $\text{Zn}_2(\text{AsO}_4)(\text{OH})$	87° r > v str	X = a Z = $\frac{b}{c}$	ORTH	101 good	Cols, green, pink	H 3.5 G 4.43	Diss by acids. CuO 0.6%.
	1.723	1.734	1.736	.013	GAGEITE (Mn,Mg,Zn) $_8\text{Si}_3\text{O}_{10}(\text{OH})_8$	Med r < v str	Z = c el pos	ORTH acic $\frac{c}{b}$	---	Pink, brown	H 3 G 3.58 infus	Dec by HCl.
1.705 ^ 1.745	1.625	1.735	1.735	.110	STILPNOMELANE $\text{K}(\text{Fe}^{+2}, \text{Fe}^{+3}, \text{Al})_{10}\text{Si}_{12}\text{O}_{30}(\text{OH})_{12}$	~ 0°	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el pos	MCL or TCL	001 perf	Dark brown	H 3-4 G 2.83 F 3	Dec by acids. Pleoc, X golden yellow, Y and Z deep reddish-brown. FeO 1.3, Fe ₂ O ₃ 31.7, MnO 2.6%.
	1.707	1.735	1.738	.031	BERAUNITE $\text{Fe}^{+2}\text{Fe}^{+3}_5(\text{PO}_4)_4(\text{OH})_5 \cdot 4\text{H}_2\text{O}$	25° r < v	X:c = 4-8°	MCL	100 perf	Green	H 3.5-4 G 2.9-3.1 F easy	Diss by acids. Pleoc, X blue-green, Y pale olive-green, Z olive- green.
1.771 v	1.715	1.735	1.745	.030	SICKLERITE $\text{Li}(\text{Mn}^{+2}, \text{Fe}^{+3})\text{PO}_4$	Med large r > v very str	X = a el cTv pos	ORTH	100 good	Yellow- ish- to dark brown	H 4 G 3.3 F easy	Diss by acids. Pleoc, X deep red, Y pale red, Z very pale red. MnO 33.6, Mn ₂ O ₃ 2.1, Fe ₂ O ₃ 11.3%.
	1.720	1.736	1.738	.018	GITTINSITE $\text{CaZrSi}_2\text{O}_7$	30°	X:c = 5- 10°	MCL fib el $\frac{c}{b}$	---	White	H 3.5-4 G 3.6	---
1.728 └	1.640	1.736	1.750	.110	HYDROZINCITE $\text{Zn}_5(\text{CO}_3)_2(\text{OH})_6$	40° r < v str	X = $\frac{b}{c}$ Z:c = 40°	MCL laths 100	100 perf	White	H 2.5 G 4.0 infus	Diss by acids with eff. Fluor pale blue in UV.
1.750 v	1.654	1.736	1.764	.110	LOMONOSOVITE $\text{Na}_2\text{Ti}_2\text{Si}_2\text{O}_9 \cdot \text{Na}_3\text{PO}_4$	60°	el pos	TCL	100 perf	Yellow to dark brown	H 3-4 G 3.15	Dec by H ₂ O. Pleoc, X yellow, Y orange- yellow, Z dark brown.
1.721 └	1.715	1.736	1.739	.024	RENARDITE $\text{Pb}(\text{UO}_2)_4(\text{PO}_4)_2(\text{OH})_4 \cdot 7\text{H}_2\text{O}$	~ 40° r > v	X = a Y = $\frac{c}{b}$ el cTv pos	ORTH pris	100 perf	Yellow	H 3.5 G 4.35 F easy	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
1.720 ^ 1.754	1.721	1.736	1.740	.019	EPIDOTE (Epidote grp) $\text{Ca}_2(\text{Al}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	78° (54+14°) r > v str	Y = $\frac{b}{c}$ X ~ $\frac{c}{b}$ el cTv pos	MCL el $\frac{b}{c}$	001 perf	Green	H 7 G 3.44 F 4	Insol in acids. Fe ₂ O ₃ 9.5, FeO 0.8%.
	1.723	1.737	1.756	.043	ROSELITE-BETA $\text{Ca}_2\text{Co}(\text{AsO}_4)_2 \cdot 2\text{H}_2\text{O}$	80-90° r < v	---	TCL	010 perf	Dark rose-red	H 3.5-4 G 3.71	Pleoc, X pink, Y pale pink, Z cols.
	1.708	1.738	1.747	.039	ZIRCOPHYLLITE (K, Na, Ca) $_3(\text{Mn}, \text{Fe})_7$ (Zr, Nb) $_2\text{Si}_8\text{O}_{27}(\text{OH})_4$	62° r > v str	Y = a X:b = 10° el neg	TCL	001 perf	Dark brown	H 4-5 G 3.34	Pleoc, X and Y dark yellow, Z brown.
1.711 ^ 1.755	1.721	1.738	1.742	.021	ALLANITE (Epidote grp) $(\text{Ce}, \text{Ca}, \text{Y})_2(\text{Al}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	Large (51+13°) r > v str	Y = $\frac{b}{c}$	MCL	001, 100, 110 poor	Brown, black	H 6 G 4.0 F 3	Gel with acids. Pleoc. Fe ₂ O ₃ 17.0, MnO 0.6, ThO 0.9%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.716 ^	1.719	1.739	1.748	.029	THALENITE $Y_3Si_3O_{10}(OH)$ (?)	67° $r < v$	$Y \sim c$	MCL tab 100	110 perf	Reddish to black	H 6-6.5 G 4.40	---
	1.730	(1.739)	1.748	.018	SUZUKIITE $Ba_2V^{+4}_2O_2Si_4O_{12}$	~ 90° $r < v$ str	$X = a$ $Y = \underline{b}$	ORTH	010 perf 100, 001 dist	Bright green	H 4-4.5 G 4.03	Pleoc, X pale green, Y yellowish-green, Z bluish-green, abs $X < Y < Z$.
	1.722	(1.739- 1.740)	1.745	.023	STEIGERITE $AlVO_4 \cdot 3H_2O$	55-60°	---	MCL	010 good	Canary yellow	---	Diss by acids.
1.714 ^	1.683	1.740	1.755	.072	SCHOEPITE $UO_3 \cdot 2H_2O$	Med $r > v$ str	$X = c$ $Y = \underline{b}$ el pos	ORTH tab	001 perf	Yellow	H 2.5 G 4.8-5.0	Diss by acids. Pleoc, X cols, Y yellow, Z lemon-yellow. Fluor pale green in UV.
1.751 pos 1.756	1.720	1.740	1.757	.037	ACMITE-AUGITE (Pyroxene grp) (Na,Ca)(Fe ⁺³ ,Al,Mg, Fe ⁺²)Si ₂ O ₆	82-88° $r < v$	$Y = b$ $X:c = 15^\circ$ el neg	MCL	110 good at 87°	Dark green	H 6 G 3.40	Insol in acids. Pleoc, X bright green, Y and Z greenish-yellow. FeO 7.5, Fe ₂ O ₃ 14.0, Na ₂ O 5.3%.
	1.726	(1.740)	1.747	.021	STRASHIMIRITE $Cu_8(AsO_4)_4(OH)_4 \cdot 5H_2O$	70°	$Z:el = 5^\circ$	MCL spherulites	---	Green	G 3.8 calc	Pleoc wk, Y very pale yellow-green, Z yellow-green.
1.723 ^	1.731	1.741	1.743	.012	SAPPHIRINE (Mg,Al,Fe ⁺³) ₈ (Si,Al) ₆ O ₂₀	40° $r < v$ str	$Y = b$ $Z \sim \underline{c}$	MCL tab	010, 100 poor	Blue	H 7.5 G 3.60 infus	Insol in acids. Pleoc, X pale brown, Y and Z deep blue. FeO 3.0, Fe ₂ O ₃ 10.4%.
	1.724	1.742	1.746	.022	HODGKINSONITE $MnZn_2SiO_5 \cdot H_2O$	52° $r > v$ str	$Y = b$ $Z:c = 38^\circ$	MCL pyram	001 good	Pale reddish-brown	H 5 G 4.01 F easy	Gel with acids. Pleoc, X and Z lavender, Y pale green.
	1.735	1.742	1.745	.010	HAGENDORFITE (Na,Ca)Mn (Fe ⁺² ,Fe ⁺³ ,Mg) ₂ (PO ₄) ₃	68°	$Z = b$ $Y:c = -22^\circ$	MCL	---	Green	H 3.5 G 3.71	Diss by acids. Pleoc, X yellow brown, Y green, Z blue-green.
	1.720	1.743	1.748	.028	SANEROITE $Na_2(Mn^{+2},Mn^{+3})_{10}Si_{11}VO_{34}(OH)_4$	44°	el pos	TCL	2 clv perf	Deep orange	G 3.47	Pleoc, X deep orange, Y lemon-yellow, Z yellow-orange.
	1.738	1.743	1.746	.008	SURINAMITE (Al,Mg,Fe) ₃ (Si,Al) ₂ (O,OH) ₈	68° disp str	$Y = b$ $Z:clv = 44^\circ$	MCL plates	010 perf	Blue	G > 3.3	---

1.770 V	1.670	1.744	1.782	.112	MURMANITE $\text{Na}_2(\text{Ti}, \text{Nb})_2\text{Si}_2\text{O}_9 \cdot x\text{H}_2\text{O}$	60° (68+2°) $r > \bar{v}$	X: a small Z = \bar{b} el cTv pos	TCL	100 perf	Pink to yellow- brown	H 2 G 2.8	Pleoc, X light brown, Y brown to pink, Z dark brown, abs Z > Y > X.
1.752 U	1.703	1.744	1.786	.083	AHLFELDITE $\text{NiSeO}_3 \cdot 2\text{H}_2\text{O}$	87°	Z: c = 12°	MCL	Conch	Light green	H 2.5 G 3.51	Data for synth compd. Pleoc, X and Y pale green, Z green.
	---	1.74- 1.75	---	---	TINTICITE $\text{Fe}_6(\text{PO}_4)_4(\text{OH})_6 \cdot 7\text{H}_2\text{O}$	---	---	ORTH(?) earthy masses	---	Creamy white	H 5 G 2.8	---
	1.707	1.745	1.776	.069	SHCHERBAKOVITE $(\text{K}, \text{Na}, \text{Ba})_3(\text{Ti}, \text{Nb})_2\text{Si}_4\text{O}_{14}$	82° $r < \bar{v}$ dist	X = $\frac{b}{a}$ Y = $\frac{\bar{a}}{\bar{b}}$ el pos	ORTH	110 perf	Brown to dark brown	H 6.5 G 3.21	Pleoc, X pale yellow, Y yellow, Z brownish- yellow.
1.738 □ 1.752	1.658	1.746	1.751	.093	AURICALCITE $(\text{Zn}, \text{Cu})_5(\text{CO}_3)_2(\text{OH})_6$	Small $r < \bar{v}$ str	X = $\frac{a}{b}$ Y = $\frac{\bar{a}}{\bar{b}}$	ORTH fib c	010 perf	Pale green or blue	H 1-2 G 3.64 infus	Diss by acids or NH_4OH . Pleoc wk, X cols, Y^4 and Z faint blue-green.
	1.703	1.746	1.789	.086	LIBETHENITE $\text{Cu}_2\text{PO}_4(\text{OH})$	85-90° $r > \bar{v}$ str	X = $\frac{b}{c}$ Y = $\frac{\bar{c}}{\bar{a}}$	ORTH short pris	100, 010 poor	Dark emerald- green	H 4-4.5 G 3.97 F 2.5	Diss by acids. Pleoc, X pale green to yellow, Y bright green to green-yellow, Z yellow to yellow-green.
1.723 ^	1.720	1.746	1.760	.040	LAVENTITE $(\text{Na}, \text{Ca})_3\text{ZrSi}_2\text{O}_7$ (O, OH, F) ₂	73° $r < \bar{v}$	Y = $\frac{b}{c}$ X: c = -20°	MCL tab	100 good	Yellow to brown	H 6 G 3.55 fus	Diss with diff by acids. Pleoc, X pale yellow, Y greenish- yellow, Z orange- yellow. TiO_2 11.3%.
1.728 ◇ 1.762	1.736	1.747	1.752	.016	EULITE (Orthopyroxene ser, Pyroxene grp) $(\text{Fe}, \text{Mg})_2\text{Si}_2\text{O}_6$	63°	Y = $\frac{a}{c}$ Z = $\frac{\bar{c}}{\bar{a}}$ el pos	ORTH	210 good	Green	H 5-6 G 3.78 F 5	Nearly insol in acids. FeO 39.4, MnO 0.9% (En_{30}).
1.716 ◇ 1.79	1.728	1.748	1.754	.026	JULGOLDITE (Pumpellyite grp) $\text{Ca}_2(\text{Fe}^{+2}, \text{Mg})(\text{Fe}^{+3}, \text{Al})_2$ $(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{OH})_2 \cdot \text{H}_2\text{O}$	30-60°	---	MCL	100, 001 good	Greenish- black	H 4.5 G 3.31	Gel with hot HCl. Tw 001. Pleoc, X pale yellow, Y blue-green, Z yellowish-red.
	1.739	1.749	1.752	.013	PRZHEVALSKITE $\text{Pb}(\text{UO}_2)_2(\text{PO}_4)_2 \cdot 4\text{H}_2\text{O}$	30° (57+19°)	el neg	ORTH tab	001 good	Bright yellow	---	Pleoc, X cols, Y pale yellow, Z deep yellow.
1.736 ^	1.670	1.750	1.778	.108	LOMONOSOVITE $\text{Na}_2\text{Ti}_2\text{Si}_2\text{O}_9 \cdot \text{Na}_3\text{PO}_4$	~ 60°	el pos	TCL	100 perf	Dark brownish- yellow	H 3-4 G 3.15	Dec by H_2O . Pleoc, X yellow, Y orange- yellow, Z dark brown.
	1.72	1.75	1.82	.10	MAGNESIUM-ZIPPEITE $\text{Mg}_2(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 16\text{H}_2\text{O}$	Large	Z = $\frac{c}{a}$	ORTH plates	010 perf	Yellow to orange	H 2	Diss by acids. Tw com- mon. Pleoc, X pale yellow, Y yellow, Z dark yellow. Fluor bright yellow in UV.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.733 ↓ 1.770	1.721	<u>1.750</u>	1.765	.044	HYALOSIDERITE (Olivine grp) (Mg,Fe) ₂ SiO ₄	85° (70+5°)	X = $\frac{b}{a}$ Z = $\frac{a}{b}$	ORTH	010 fair	Green	H 5 G 3.7 F 5.5	Gel with acids. FeO 36.8, MnO 0.5% (Fe ₅₂).
v 1.776	1.733	<u>1.750</u>	1.762	.029	PIEMONTE (Epidote grp) Ca ₂ (Al,Mn,Fe) ₃ Si ₃ O ₁₂ (OH)	78° r > v str	Y = $\frac{b}{a}$	MCL el $\frac{b}{a}$	001 perf	Reddish-brown	H 6.5 G 3.40 F 3	Pleoc, X yellow, Y pale amethyst, Z deep purplish-red. Fe ₂ O ₃ 11.3, Mn ₂ O ₃ 1.0%.
1.746 pos ↓ 1.753 pos	1.743	<u>1.751</u>	1.758	.015	STAUROLITE (Fe,Mg,Zn) ₂ Al ₉ (Si,Al) ₄ O ₂₂ (OH) ₂	87° r > v wk	Z = $\frac{c}{b}$ X ~ $\frac{b}{a}$ el cTv pos	MCL ps orth	010 good	Reddish-brown	H 7 G 3.74 infus	Insol in acids. Pleoc, X cols, Y pale yellow, Z golden yellow. FeO 11.2, Fe ₂ O ₃ 2.9%. U opt pos.
	1.735	<u>1.753</u>	1.767	.032	SURSASSITE Mn ₅ Al ₄ Si ₅ O ₂₁ ·3H ₂ O (?)	65° (82+7°) r < v str	Y = $\frac{b}{a}$ X:c = 55°	MCL el $\frac{b}{a}$	001	Brown to reddish-brown	G 3.26	Pleoc, X and Z cols to pale yellow, Y golden to reddish-brown. May show abnormal interf colors.
1.736 ↓ 1.768	1.729	<u>1.754</u>	1.776	.047	EPIDOTE (Epidote grp) Ca ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	73° (85+5°) r > v str	Y = $\frac{b}{a}$ X:c = 8° el clv pos	MCL	001 perf	Green	H 7 G 3.45 F 4	Insol in acids. Pleoc, X pale yellow, Y pale greenish-yellow, Z greenish-yellow. Fe ₂ O ₃ 14.0, FeO 1.2%.
	1.743	<u>1.754</u>	1.764	.021	CARACOLITE Na ₃ Pb ₂ (SO ₄) ₃ Cl	Very large r > v str	---	MCL ps hex	---	Cols	H 4.5 F 2	Insol in H ₂ O, dec by HCl. Complex tw.
	(1.710)	<u>1.755</u>	1.775	(.065)	SCHUILINGITE PbCu(Nd,Gd)(CO ₃) ₃ (OH)·1.5H ₂ O	66°	Ext 0-10°	ORTH	el	Turquoise-to azure-blue	H 3-4 G 5.2 F diff	Diss by HNO ₃ with eff.
	1.731	<u>1.755</u>	1.755	.024	ERICAITE (Fe,Mn,Mg) ₃ B ₇ O ₁₃ Cl	Small	---	ORTH ps trig	---	Pale red	G 3.5	Diss by acids.
1.738 ↓ 1.784	1.740	<u>1.755</u>	1.760	.020	ALLANITE (Epidote grp) (Ce,Ca,Y) ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	Large r > v str	Y = $\frac{b}{a}$	MCL	001, 100, 110 poor	Brownish-black	H 6 G 3.90 F 3	Gel with acids. Pleoc. Fe ₂ O ₃ 3.2, FeO 10.7, MnO 1.9, TiO ₂ 1.7, ThO ₂ 0.9%.

1.740 ^ 1.767	1.732	1.756	1.770	.038	ACMITE, manganoan, var Blanfordite (Pyroxene grp) (Na,Ca)(Fe ³⁺ ,Mg,Mn) Si ₂ O ₆	80° r > v	Y = $\frac{b}{c}$ X:c = 9° el neg	MCL	110 good at 87°	Pink	H 6 G 3.28	Pleoc, X pink, Y pale blue, Z blue. Fe ₂ O ₃ 18.2, Al ₂ O ₃ 5.9, MnO 4.7, FeO 1.1%.
1.765 v	1.740	1.756	1.762	.022	UREYITE (Pyroxene grp) NaCrSi ₂ O ₆	65°	Y = $\frac{b}{c}$ X:c = 14°	MCL	110 good	Emerald- green	---	Insol in acids. Pleoc, X and Z dark green, Y yellow-green.
	1.692	1.757	1.800	.108	MUNIRITE NaV ⁵⁺ O ₃ ·2H ₂ O	75°	X = $\frac{c}{a}$ Y = $\frac{a}{b}$	ORTH	---	Greenish- white	G 2.43	Sol in H ₂ O.
1.728 ^	1.745	1.758	1.760	.015	HOLTITE Al ₆ (Ta,Sb,Li)[(Si,As) O ₄] ₃ (BO ₃)(O,OH) ₃	52° r < v	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH pris	100 good	Buff, greenish to brown	H 8.5 G 3.90 infus	Insol in acids. Poly tw 110. Pleoc, X yellow, Y and Z cols. Fluor orange to yellow in UV.
	1.705	1.760	1.770	.065	PARASCHOEPITE UO ₃ ·2H ₂ O (?)	46° r > v	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH	---	Yellow	H 2	Pleoc, X cols, Y and Z yellow.
	1.724	1.760	(1.772)	(.048)	NIOBOPHYLLITE (K,Na) ₃ (Fe,Mn) ₆ (Nb,Ti) ₂ Si ₈ (O,OH,F) ₃₁	60°	Y = $\frac{a}{b}$ X:b = 13°	TCL	001 perf	Brown	G 3.43	Pleoc, X and Y brownish-yellow, Z orange-red.
	1.72	1.76	1.76	.04	META-URANOPIILITE (UO ₂) ₆ (SO ₄)(OH) ₁₀ ·5H ₂ O	Small	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH(?) acic	---	Yellow	---	Fluor yellow-green in UV.
1.747 ^ 1.763 pos	1.752	1.762	1.772	.020	EULITE (Orthopyroxene ser, Pyroxene grp) (Fe,Mg) ₂ Si ₂ O ₆	87°	Y = $\frac{a}{b}$ Z = $\frac{c}{b}$ el pos	ORTH	210 good	Dark brown	H 5-6 G 3.84 F 4	Nearly insol in acids. FeO 44.5, Fe ₂ O ₃ 1.9%.
	1.698	1.765	1.799	.101	RICHELSDORFITE Ca ₂ Cu ₅ Sb(AsO ₄) ₄ Cl (OH) ₆ ·6H ₂ O	69° r > v	Y = $\frac{a}{b}$ Z = $\frac{b}{c}$	MCL	001 very good	Sky-blue to tur- quoise	G 3.20	Pleoc, X pale blue, Y greenish-blue, Z bright greenish-blue.
1.756 ^	1.742	1.765	1.767	.025	UREYITE (Pyroxene grp) NaCrSi ₂ O ₆	34°	Y = $\frac{b}{c}$ X:c = 22°	MCL	110 good	Emerald- green	---	Pleoc, X yellow, Y emerald-green, Z dark blue-green.
1.727 ^ 1.802	1.750	1.766	1.779	.029	TEPHROITE (Olivine grp) (Mn,Mg) ₂ SiO ₄	67° (83+8°) r > v	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH	001 good 010 fair	Brown, gray	H 6 G 3.87 F 3.5	Gel with acids. MnO 60.7, MgO 6.6, FeO 0.2%.
	1.705	1.767	1.769	.064	SIMPLITITE CaV ₄ O ₉ ·5H ₂ O	25° r > v wk	X = $\frac{b}{c}$ Z:c = -58°	MCL ps tet	010 perf	Dark green	H 1 G 2.64	Diss by acids. Pleoc, X yellow, Y and Z green.
	1.736	1.767	1.796	.060	ARTHURITE CuFe ₂ (AsO ₄ ,PO ₄ ,SO ₄) ₂ (O,OH) ₂ ·4H ₂ O	~ 90°	Y = $\frac{b}{c}$ Z:c = 10°	MCL	---	Apple green	G 3.2	Pleoc, X cols to pale green, Y gray-green, Z olive-green, abs Z > Y > X.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.756 $\hat{\Delta}$ 1.786	1.741	<u>1.767</u>	1.789	.048	ACMITE-AUGITE (Pyroxene grp) $(Na,Ca)(Fe^{+3},Mg)Si_2O_6$	85° $r > v$	$Y = b$ $X:c = 12^\circ$ el neg	MCL	110 good at 87°	Green	H 6 G 3.56	Insol in acids. Pleoc, X deep green, Y yellow- green, Z brown-green, abs $X > Y = Z$. Fe_2O_3 19.1, FeO 5.2, MnO_2 0.5%.
	1.729	<u>1.767</u>	1.772	.043	SHUIKITE (Pumpellyite grp) $Ca_2(Mg,Al)(Cr,Al)$ $(SiO_4)(Si_2O_7)(OH)_2 \cdot H_2O$	40-50° $r < v$ str	el pos	MCL fib	001 perf	Dark brown	H 6 G 3.24	Pleoc, X violet blue, Y yellow green, Z dark violet. Cr_2O_3 19.3, Fe_2O_3 1.7%.
	1.761	(<u>1.767</u>)	1.772	.011	TADZHIKITE $Ca_3(Ce,Y)_2(Ti,Al,Fe)B_4$ Si_4O_{22}	88°	---	MCL	100 good	Grayish- brown	H 6 G 3.73	---
288 1.754 $\hat{\Delta}$ 1.784	1.740	<u>1.768</u>	1.787	.047	EPIDOTE (Epidote grp) $Ca_2(Al,Fe)_3Si_3O_{12}(OH)$	74° $r > v$ str	$Y = b$ $X:c = 13^\circ$ el clv pos	MCL	001 perf	Green	H 7 G 3.48 F 4	Insol in acids. Pleoc wk. Fe_2O_3 17.9, FeO 0.4, MnO 0.5%.
1.734 $\hat{\Delta}$ 1.810	1.742	<u>1.768</u>	1.773	.031	ADAMITE (Olivenite grp) $(Zn,Cu)_2(AsO_4)(OH)$	47° $r > v$ str	$X = a$ $Z = \underline{b}$	ORTH	101 good	Sea-green	H 3.5 G 4.32	Diss by acids. Pleoc, X coils, Y blue-green, Z yellow-green. ZnO 31.9, CuO 23.4, CoO 0.5%.
	1.761	<u>1.768</u>	1.769	.008	CORUNDUM Al_2O_3	17°	$X = c$	HEX	Parting 0001	Blue, gray, brown	H 9 G 4.02 infus	Anom biax. Tw common 1011. Pleoc wk in blue.
\vee 1.831	1.730	(<u>1.769</u>)	1.771	.041	CONICALCITE $CaCuAsO_4(OH)$	25° $r > v$ very str	$Z = el$	ORTH	011 fr uneven	Pista- chio- to emerald- green	H 4.5 G 3.96 F 3	Diss by acids. Pleoc wk, X yellow, Y pale green, Z pale blue- green.
	1.760	<u>1.769</u>	1.769	.009	ESPERITE $(Ca,Pb)ZnSiO_4$	5-40°	---	MCL	010, 100 dist	White	H 5 G 4.28	Gel with acids. Fluor bright yellow in short- wave UV.
1.744 \wedge	1.682	<u>1.770</u>	1.81	.13	MURMANITE $Na_2(Ti,Nb)_2Si_2O_9 \cdot xH_2O$	57-64° $r > v$	$X:a$ small $Z = b$ el pos	TCL	100 perf	Pink to yellow- brown	H 2 G 2.8	Pleoc, X light brown, Y brown to pink, Z dark brown.

	1.72	<u>1.77</u>	1.81	.09	ZINC-ZIPPEITE $\text{Zn}_2(\text{UO}_2)_6(\text{SO}_4)_3(\text{OH})_{10} \cdot 16\text{H}_2\text{O}$	Large	$Z = \underline{c}$	ORTH plates	010 perf	Yellow	H 2	Diss by acids. Pleoc, X cols, Y pale yellow, Z yellow. Fluor yellow in UV. Data for synth compd.
	1.720	<u>1.770</u>	1.800	.080	MELONJOSEPHITE $\text{CaFe}^{+2}\text{Fe}^{+3}(\text{PO}_4)_2(\text{OH})$	80-85° disp str	$X = \underline{c}$ $Z = \underline{b}$	ORTH fib	el	Dark green	H < 5 G 3.65	Diss by HCl. Pleoc, X deep brown, Y greenish-brown, Z yellow-gold.
1.750 ∧ 1.786	1.742	<u>1.770</u>	1.786	.044	HORTONOLITE (Olivine grp) $(\text{Fe,Mg})_2\text{SiO}_4$	69°	$X = \underline{b}$ $Z = \underline{a}$	ORTH	010 fair	Green	H 5 G 3.9 F 5	Gel with acids (Fo_{43}).
	1.734	<u>1.770</u>	1.773	.039	VINOGRADOVITE $(\text{Na,Ca,K})_4\text{Ti}_4\text{AlSi}_6\text{O}_{23}(\text{OH}) \cdot 2\text{H}_2\text{O}$	41° $r > v$	$Z:\underline{c} = 7^\circ$	MCL	010 perf	Cols	H 4 G 2.88	Gel with acids.
	1.728	<u>1.771</u>	1.800	.072	BROCHANTITE $\text{Cu}_4\text{SO}_4(\text{OH})_6$	75° $r < v$ med	$Y = \underline{b}$ $X \sim \underline{a}$ el cTv pos	MCL pris	100 perf	Emerald-green	H 3.5-4 G 3.97 F 3.5	Diss by acids or NH_4OH . Tw 100 common. Pleoc in blue-greens.
	1.727	<u>1.771</u>	1.798	.071	MARGAROSANITE $\text{Pb}(\text{Ca,Mn})_2\text{Si}_3\text{O}_9$	78° (74+3°) $r < v$	$X':010 = 42^\circ$	TCL lam	010 perf 100 good 001 fair	Cols	H 2.5-3 G 4.33	Dec by HNO_3 with sepn of silica. Fluor pink or blue in UV. F 2 in reducing flame.
	1.726	<u>1.771</u>	1.780	.054	PARADAMITE $\text{Zn}_2\text{AsO}_4(\text{OH})$	50°	---	TCL	010 perf	Pale yellow	G 4.55 F 3	Diss by acids.
	1.751	<u>1.771</u>	1.782	.031	LEUCOPHOENICITE $\text{Mn}_7(\text{SiO}_4)_3(\text{OH})_2$	74° $r > v$ wk	$X \perp \text{clv}$	MCL ps orth	---	Light to deep pink	H 5.5 G 3.85 F 3	Gel with acids. Poly tw on 001.
1.735 ∧ 1.805	1.750	<u>1.771</u>	1.776	.026	FERRISICKLERITE $\text{Li}(\text{Fe}^{+3}, \text{Mn}^{+2})\text{PO}_4$	52° $r > v$	$X = \underline{a}$ el cTv pos	ORTH	100 good 010	Dark brown	H 4.5 G 3.30 F easy	Diss by acids.
┌ 1.779	1.755	<u>1.772</u>	1.774	.019	ALLACTITE $\text{Mn}_7(\text{AsO}_4)_2(\text{OH})_8$	Small $r > v$ str	$Y = \underline{b}$ $X:\underline{c} = 51^\circ$	MCL pris	001 dist	Purplish-red	H 6.5 G 3.83 F diff	Diss by HCl. Pleoc, X blood red, Y pale yellow, Z sea-green.
	1.772	<u>1.775</u>	1.777	.005	BRITHOLITE $(\text{Ce,Ca})_5(\text{SiO}_4)_3\text{PO}_4(\text{OH},\text{F})$	0-44°	$X = \underline{c}$ $Y = \underline{a}$	HEX pris	---	Brown	H 5-5.5 G 4.0-4.5	Apatite-type structure.
	1.679	<u>1.776</u>	1.807	.128	SAHAMALITE $(\text{Mg,Fe})\text{Ce}_2(\text{CO}_3)_4$	57° $r < v$	$Y = \underline{b}$ $Z:\underline{c} = 29^\circ$	MCL tab	010 poor	Cols	G 4.30 infus	Diss by concd HCl with eff.
1.750 ∧ 1.803	1.746	<u>1.776</u>	1.793	.047	PIEMONTE (Epidote grp) $\text{Ca}_2(\text{Al,Mn,Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	87° (73+5°) $r > v$ str	$Y = \underline{b}$	MCL el <u>b</u>	001 perf	Purplish-red	H 4.5 G 3.44 F 3	Pleoc, X yellow, Y pink, Z deep pink. Fe_2O_3 11.9, Mn_2O_3 6.8%.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.745	<u>1.777</u>	1.84	.095	NICKEL-ZIPPEITE $Ni_2(UO_2)_6(SO_4)_3(OH)_{10} \cdot 16H_2O$	Large	$Z = \underline{c}$	ORTH plates	010 perf	Yellow to orange	H 2	Diss by acids. Pleoc, X cols, Y pale yellow, Z yellow. Fluor yellow in UV. Data for synth compd.
	1.740	<u>1.777</u>	1.780	.040	BELYANKINITE $Ca(Ti,Zr,Nb)_5O_{12} \cdot 9H_2O (?)$	23°	$Y = \underline{b}$ $X = \underline{c}$	ORTH	One perf	Cols to brown	H 2.3 G 2.36 infus	Dec by warm acids. Pleoc, X dark brown, Y and Z light brown.
	1.756	<u>1.777</u>	1.794	.038	ORIENTITE $Ca_2Mn^{+2}Mn^{+3}_2Si_3O_{10}(OH)_4$	80° $r < v$ str	$X = \underline{b}$ $Y = \underline{c}$	ORTH	001, 120 imperf	Deep brown	G 3.05	Dec by HCl. Pleoc, X red-brown, Y yellow, Z yellow-brown.
	1.763	<u>1.777</u>	1.785	.022	YOSHIMURAITE $(Ba,Sr)_2TiMn_2(SiO_4)_2(PO_4,SO_4)(OH,Cl)$	90-95° (74+11°) $r < v$	---	TCL	010 perf 101, 101 dist	Brown	H 4.5 G 4.13	Poly tw on 010. Pleoc, X bright yellow, Y orange-brown, Z brown, abs $Z > Y > X$. Reported as biax pos.
	1.747	<u>1.779</u>	1.84	.09	COBALT-ZIPPEITE $Co_2(UO_2)_6(SO_4)_3(OH)_{10} \cdot 16H_2O$	Large	$Z = \underline{c}$	ORTH plates	010 perf	Yellow, orange	H 2	Diss by acids. Pleoc, X cols, Y pale yellow, Z yellow. Fluor yellow in UV. Data for synth compd.
1.716 ^	1.763	<u>1.779</u>	1.793	.030	SONOLITE $Mn_9(SiO_4)_4(OH,F)_2$	78° $r > v$	$X:\underline{c} = 9^\circ$	MCL pris	---	Reddish-brown	H 5.5 G 3.82	Gel with acids. Tw on 001 common.
	1.67	<u>1.78</u>	1.78	.11	NULLAGINITE $Ni_2(OH)_2CO_3$	Small	$X:\underline{c} = 6^\circ$	MCL fib	---	Bright green	G 3.56 infus	---
v 1.832	1.725	<u>1.780</u>	1.790	.065	BILLIETITE $BaU_6O_{19} \cdot 11H_2O$	35° (45+5°) $r > v$ str	$X = \underline{c}$ $Y = \underline{a}$ el cTv pos	ORTH ps hex	001 perf	Golden-yellow	G 5.32	Tw on 111 and 110 common. Pleoc, X pale yellow, Y and Z deep golden-yellow.
1.680 ^	1.756	<u>1.780</u>	1.792	.036	ALLEGHANYITE $Mn_5(SiO_4)_2(OH)_2$	72° $r > v$	$Z = \underline{b}$ Ext:tw pl = 22°	MCL	Conch	Pink to reddish-brown	H 5.5 G 4.02 F 2.5	Gel with acids. Poly tw on 001 common.
	1.767	<u>1.782</u>	1.791	.024	TEINEITE $CuTeO_3 \cdot 2H_2O$	36° (75+10°)	$X = \underline{a}$ $Y = \underline{b}$	ORTH	010 good 001, 100 poor	Blue	H 2.5 G 3.80 F 2	Pleoc, X greenish-blue, Y blue, Z indigo blue, abs $Z > Y > X$.

	1.720	<u>1.783</u>	1.842	.122	ROSSITE $\text{CaV}_2\text{O}_6 \cdot 4\text{H}_2\text{O}$	85°	$Z \sim c$ $Y:b \sim 45^\circ$	TCL pris	010 good	Yellow	H 2.5 G 2.45 F easy	Slowly diss by H_2O . Tw 100. Data for synth compd.
1.768 ^	1.751	<u>1.784</u>	1.797	.046	EPIDOTE (Epidote grp) $\text{Ca}_2(\text{Al},\text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	64° $r > v$ str	$Y = b$ $X \sim c$ el cTv pos	MCL	001 perf	Green	H 7 G 3.48 F 4	Insol in acids. Fe_2O_3 17.2, FeO 0.5%.
	1.765	<u>1.784</u>	1.799	.034	MALAYAITE CaSnSiO_5	85°	---	MCL	---	Pale yellow	H 4 G 4.55	Fluor yellow-green in short-wave UV.
1.755 ^ 1.815	1.769	<u>1.784</u>	1.791	.022	ALLANITE (Epidote grp) $(\text{Ce},\text{Ca},\text{Y})_2(\text{Al},\text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	60° $r > v$ str	$Y = b$	MCL	001, 100, 110 poor	Brownish- black	H 6 G 3.87 F 3	Slowly attacked by HCl, gel. Pleoc, X cols to red-brown, Y yellow- brown to pale green, Z green-brown to dark red-brown.
	1.77	<u>1.785</u>	1.815	.045	BARBOSALITE $\text{Fe}^{+2}\text{Fe}^{+3}_2(\text{PO}_4)_2(\text{OH})_2$	64° $r > v$ str	---	MCL ps tet tab	100	Dark blue- green	H 2 G 3.65	Slowly diss by hot HCl. Pleoc, X and Y dark blue-green, Z dark olive-green.
	1.755	<u>1.785</u>	1.785	.030	THEISITE $\text{Cu}_2\text{Zn}_5(\text{As},\text{Sb})_2\text{O}_8(\text{OH})_{14}$	~ 0°	---	ORTH ps hex	001	Pale blue- green	H 1.5 G 4.3	Not pleoc.
1.767 ^ 1.812	1.751	<u>1.786</u>	1.800	.049	ACMITE (Pyroxene grp) $(\text{Na},\text{Ca})(\text{Fe}^{+3},\text{Mg},\text{Fe}^{+2})\text{Si}_2\text{O}_6$	69° $r > v$	$Y = b$ $X:c = 1^\circ$ el neg	MCL	110 good at 87°	Green	H 6 G 3.55	Pleoc, X and Y deep green, Z yellow-brown. Fe_2O_3 25.4, FeO 3.7%.
1.770 ^ 1.790	1.758	<u>1.786</u>	1.804	.046	HORTONOLITE, manganian, zincian (Olivine grp) $(\text{Fe},\text{Mn},\text{Mg},\text{Zn})_2\text{SiO}_4$	77° $r > v$	$X = b$ $Z = a$	ORTH	010 fair	Reddish- brown	H 6 G 4.1 F 3.5	Gel with acids. Pleoc wk. FeO 35.5, MnO 16.9, ZnO 10.7, MgO 5.6%.
	1.779	<u>1.786</u>	1.790	.011	KOLICITE $\text{Mn}_7\text{Zn}_4(\text{AsO}_4)_2(\text{SiO}_4)_2(\text{OH})_8$	78° $r < v$ str	$Y = c$ $Z = a$	ORTH	---	Yellow- orange	H 4.5 G 4.17	Pleoc, X cols to pale yellow, Y yellow- orange, Z light yellow, abs $Y = Z > X$.
	1.749	<u>1.790</u>	1.821	.072	REINERITE $\text{Zn}_3(\text{AsO}_3)_2$	(80±3°)	---	ORTH	110 good 011, 111	Blue to greenish- yellow	H 5-5.5 G 4.27	Diss by cold HCl.
	1.70	<u>1.79</u>	1.793	.09	EPI-INTHINITE $\text{UO}_3 \cdot x\text{H}_2\text{O}$ (?)	Small	$X = c$ $Y = b$	ORTH	---	Yellow	---	Pleoc, X pale yellow, Y yellow, Z deep yellow. Stated to be Schoepite, optics are different.
1.748 ^ 1.814	1.776	<u>1.79</u>	1.820	.044	JULGOLDITE (Pumpellyite grp) $\text{Ca}_2(\text{Fe},\text{Mg})(\text{Fe},\text{Al})_2(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{OH})_2 \cdot \text{H}_2\text{O}$	80°	---	MCL	100, 001 good	Greenish- black	H 4.5 G 3.4	Gel with hot HCl. Pleoc, X and Y pale greenish-yellow, Z olive-green. Tw 001.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.786 ↓ 1.818 v 1.835	1.761	1.790	1.806	.045	HORTONOLITE (Olivine grp) (Fe,Mg) ₂ SiO ₄	65° (72+5°) r > v	X = b Z = $\frac{a}{b}$ el cTv pos	ORTH	010 fair	Green to brown	H 5 G 3.90 F 5	Gel with acids (Fo ₃₃).
	1.772	1.790	1.796	.024	PSEUDOMALACHITE Cu ₅ (PO ₄) ₂ (OH) ₄ ·H ₂ O	50°	X:c = 21°	MCL pris	010 good	Dark green	H 4.5-5 G 4.35 F 2	Diss by acids. Tw 100. Pleoc wk.
	1.765	1.792	(1.82)	(.055)	VANDENBRANDEITE Cu(UO ₂)(OH) ₄	~ 90°	---	TCL tab	110 perf 2 others	Dark green	H 4 G 5.0	Diss by acids. Not fluor in UV.
1.784 □ 1.798	1.756	1.793	1.809	.053	THORTVEITITE (Sc,Y) ₂ Si ₂ O ₇	50-70° r > v dist	Y = b	MCL pris	110 good	Gray-green	H 6.5 G 3.58 F diff	Insol in acids. Poly tw.
	1.752	1.795	1.800	.048	PUMPELLYITE-(MN) (Pumpellyite grp) Ca ₂ (Mn ⁺² ,Mg)(Al,Mn ⁺³) ₂ (SiO ₄)(Si ₂ O ₇)(OH) ₂ ·H ₂ O	(37+8°)	---	MCL	001 perf	Pink to brownish-pink	H ~ 5 G 3.34	Mn ₂ O ₃ 77, Fe ₂ O ₃ 24, MnO 13.4%. Pleoc, X pale pink, Y and Z brownish-pink.
	1.773	1.795	1.815	.042	MACFALLITE Ca ₂ (Mn ⁺³ ,Al) ₃ (SiO ₄) (Si ₂ O ₇)(OH) ₃	(86+6°)	Z = b	MCL	001 perf	Red-brown, maroon	H > 5 G 3.43	Tw 100. Pleoc, X yellow, Y light brown, Z dark brown.
	1.780	1.795	1.805	.025	JAROSEWICHITE Mn ⁺² ₃ Mn ⁺³ (AsO ₄)(OH) ₆	(78+9°)	X = a Y = $\frac{b}{a}$	ORTH pris	---	Bright red	G 3.66	Streak reddish-orange. Pleoc, X med brownish-red, Z dark brownish-red.
	1.758	1.796	1.810	.052	TLALOCITE (Cu,Zn) ₁₆ (TeO ₃)(TeO ₄) ₂ Cl(OH) ₂₅ ·27H ₂ O	64°	---	MCL (?) spherulitic	---	Blue	H 1 sectile	Diss by HNO ₃ . Pleoc, X yellow-green, Y and Z blue-green, abs Z > Y > X.
	1.773	1.797	1.814	.041	BRUEGGENITE Ca(IO ₃) ₂ ·2H ₂ O	88° (79+6°) r > v mod to str	Z = b X:a = 9° el neg	MCL	Conch	Cols to bright yellow	H 3.5 G 4.24	Slightly sol in H ₂ O.
	1.720	1.798	1.805	.085	GUILLEMINITE Ba(UO ₂) ₃ (SeO ₃) ₂ (OH) ₄ ·3H ₂ O	35° r > v str	X = c Y = $\frac{b}{c}$ el cTv neg	ORTH	100 perf 010 good	Canary-yellow	G 4.88	Pleoc str, X bright yellow, Y yellow, Z cols.

	1.775	<u>1.798</u>	1.800	.025	BOSTWICKITE $\text{CaMn}^{+3}_6\text{Si}_3\text{O}_{16}\cdot 7\text{H}_2\text{O}$	25°	$r < v$ str	---	---	Dark red	H 1 G 2.93	Diss by HCl. Pleoc, X and Y red-brown, Z yellow-brown, abs X = Y > Z.
	<1.79	<u>1.798</u>	1.802	>.01	COMPREGNACITE $\text{K}_2\text{U}_6\text{O}_{19}\cdot 11\text{H}_2\text{O}$	10-15°	$X = \frac{c}{a}$ $Y = \frac{a}{b}$ el cTv pos	ORTH	001 perf	Yellow	G 5.03	Diss by acids. Tw 110 common. Pleoc, X cols, Y and Z yellow.
	---	<u>1.80</u>	---	Large	CRONSTEDTITE $(\text{Fe}^{+2}, \text{Mg})_2\text{Fe}^{+3}(\text{SiFe}^{+3})_4\text{O}_5(\text{OH})_4$	~ 0°	---	MCL ps hex	001 perf	Blackish- brown	H 3.5 G 3.34 F 4	Gel with acids. Pleoc, X brown, Y and Z black.
	1.74	<u>1.80</u>	1.85	.11	SABINAITE $\text{Na}_9\text{Zr}_4\text{Ti}_{12}\text{O}_9(\text{CO}_3)_8$	85°	$X \sim \perp$ plates el pos	MCL ps hex plates	001 perf	White	G 3.31	Insol cold acids, dec by hot acids.
	1.780	---	1.810	.030	Unnamed manganese silicate $(\text{Mn}, \text{Fe}^{+2})_3\text{Si}_2\text{O}_7$	60°	---	---	2 clv dist	Rose-red	---	Am. Mineral., 43, 793 (1958).
	1.787	<u>1.800</u>	1.805	.018	KIDWELLITE $\text{NaFe}^{+3}_9(\text{PO}_4)_6(\text{OH})_{10}\cdot 5\text{H}_2\text{O}$	Large	$Y = \frac{b}{a}$	MCL acic	100 perf	Pale green to yellow	H 3 G > 3.3	---
	1.715	<u>1.801</u>	1.87	.155	MANDARINOITE $\text{Fe}^{+3}_2\text{Se}_3\text{O}_9\cdot 4\text{H}_2\text{O}$	85°	$X = \frac{b}{a}$ $Z:c = 2^\circ$ el pos	MCL pris $\frac{c}{a}$	---	Light green	H 2.5 G 2.93	Tw 100. Pleoc, X cols, Y and Z pale green.
1.766 ^	1.782	<u>1.802</u>	1.812	.030	TEPHROITE (Olivine grp) Mn_2SiO_4	65° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010, 001 good	Reddish- brown	H 6 G 4.03	Gel with acids. Pleoc, X brown, Y and Z black.
1.776 ^ 1.810	1.768	<u>1.803</u>	1.843	.075	PIEMONTE (Epidote grp) $\text{Ca}_2(\text{Al}, \text{Mn}, \text{Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	68° (92+3°) $r > v$ str	$Y = \frac{b}{a}$	MCL	001 perf	Purplish- red	H 6.5 G 3.52 F 3	Pleoc, X yellow, Y pink, Z deep pink. Fe_2O_3 11.0, Mn_2O_3 12.0, MnO 0.6%.
	1.775	<u>1.803</u>	1.812	.037	GINIITE $\text{Fe}^{+2}\text{Fe}^{+3}_4(\text{PO}_4)_4(\text{OH})_2\cdot 2\text{H}_2\text{O}$	55°	$X = \frac{c}{a}$ $Y = \frac{b}{a}$	ORTH	Conch	Blackish- green to blackish- brown	H 3-4 G 3.41	Pleoc, X light brown, Y dark brown, Z dark blue-green.
	1.793	<u>1.803</u>	1.808	.015	PERLOFFITE $\text{Ba}(\text{Mn}, \text{Fe}^{+2})_2\text{Fe}^{+3}_2(\text{PO}_4)_3(\text{OH})_3$	70-80° $r < v$ str	$X = \frac{b}{a}$ $Y:c = 42^\circ$	MCL spear- shaped	100 perf	Dark brown, greenish- brown	H 5 G 4.00	Diss by acids. Pleoc, X and Z greenish-brown, Y light greenish-brown.
v 1.835	(1.563)	<u>1.805</u>	1.851	(.288)	TYUYAMUNITE $\text{Ca}(\text{UO}_2)_2(\text{VO}_4)_2\cdot 5-8\text{H}_2\text{O}$	42°	$X = \frac{c}{a}$ $Y = \frac{a}{b}$	ORTH	001 perf 010, 100 good	Canary- yellow	H 2 G 3.6 F easy	Pleoc wk, X cols, Y and Z greenish-yellow. Not fluor in UV.
v 1.825	1.730	<u>1.805</u>	1.820	.090	BECQUERELITE $\text{CaU}_6\text{O}_{19}\cdot 11\text{H}_2\text{O}$	30° (47+3°) $r > v$	$X = \frac{c}{a}$ $Y = \frac{b}{a}$ el cTv pos	ORTH ps hex	001 perf	Golden- to lemon- yellow	H 2.5 G 5.3	Pleoc, X cols to pale yellow, Y and Z deep yellow.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.771 [^]	1.790	1.805	1.820	.025	FERRISICKLERITE $\text{Li}(\text{Fe}^{+3}, \text{Mn}^{+2})\text{PO}_4$	Very large	$X = \frac{c}{a}$ $Y = \frac{a}{b}$	ORTH	010 perf 100 good	Dark brown	H 4.5 G 3.41 F easy	Diss by acids. Pleoc wk, X orange-yellow, Z gold-yellow.
	---	1.80- 1.81	---	---	MUSKOXITE $\text{Mg}_7\text{Fe}^{+3}_4\text{O}_{13} \cdot 10\text{H}_2\text{O}$	10-40°	el pos	TRIG platy 0001	0001 perf	Dark reddish-brown	H 3 G 3.15	Not pleoc.
1.793 ^u	1.793	1.807	1.809	.016	SARKINITE $\text{Mn}_2(\text{AsO}_4)(\text{OH})$	83° (41+18°)	$Y = \frac{b}{a}$ $X:c = -54^\circ$	MCL tab	100 good	Flesh-red to yellow	H 3 G 3.15	---
1.768 [^]	1.772	1.810	1.863	.091	OLIVENITE $\text{Cu}_2(\text{AsO}_4)(\text{OH})$	76° (97+3°) $r > v$ str	$X = \frac{a}{b}$ $Z = \frac{b}{c}$	ORTH	101 fr conch	Olive-green	H 3 G 4.4 F easy	Diss by acids or NH_4OH . Pleoc, X pale green, Y bright green, Z pale green.
	1.74	1.81	1.81	.07	GERASIMOVSKITE $(\text{Mn}, \text{Ca})(\text{Nb}, \text{Ti})_5\text{O}_{12} \cdot 9\text{H}_2\text{O} (?)$	18°	el pos	MCL (?) platy	One clv	Brown, gray	H 2 G 2.55	---
v 1.850	1.785	1.810	1.820	.035	VANDENDRIESSCHEITE $\text{PbU}_7\text{O}_{22} \cdot 12\text{H}_2\text{O}$	Med $r > v$ str	$X = \frac{c}{a}$ $Z = \frac{a}{b}$ el cTv pos	ORTH tab	001 perf	Orange to yellow-brown	H 3 G 4.5-5.5	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
	1.787	1.810	1.816	.029	ARSENOCASITE $\text{Mn}_5(\text{AsO}_4)_2(\text{OH})_4$	53°	$X = \frac{b}{a}$ $Z = \frac{c}{b}$ el cTv pos	ORTH u mass	010 perf	Red	H 5.5 G 4.16	---
v ~1.86	1.795	---	1.815	.020	PARSONSITE $\text{Pb}_2(\text{UO}_2)(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	---	$Y = \frac{b}{a}$ $Z:c = 11-26^\circ$	TCL	010 poor fr conch	Pale lemon-yellow	H 2.5-3 G 5.73	Not fluor in UV. Opt char unk.
1.786 [^]	1.770	1.812	1.830	.060	ACMITE (Pyroxene grp) $(\text{Na}, \text{Ca})(\text{Fe}^{+3}, \text{Mg})\text{Si}_2\text{O}_6$	58° (65+4°) $r > v$	$Y = \frac{b}{a}$ $X:c = 6^\circ$ el neg	MCL	110 good at 87°	Green	H 6 G 3.56	Insol in HCl. Pleoc, X pale yellow-green, Y yellow-green, Z emerald-green. Fe_2O_3 31.4, Al_2O_3 1.9, FeO 0.8%.
v 1.835	(1.786)	1.813	1.852	(.066)	BAFERTISITE $\text{Ba}(\text{Fe}, \text{Mn})_2\text{TiSi}_2\text{O}_7$ (O, OH) ₂	80°	$Y = \frac{b}{a}$ $Z:c = 6^\circ$ el clv neg	MCL	001 perf	Orange to red	H 5 G 3.97-4.25	Pleoc, X yellowish-red, Z pale yellow. FeO 10.8, MnO 12.8, Fe_2O_3 3.7%.

1.79 ^	1.776	1.814	1.836	.060	JULGOLDITE (Pumpellyite grp) $\text{Ca}_2(\text{Fe,Mg})(\text{Fe,Al})_2$ $(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{OH})_2 \cdot \text{H}_2\text{O}$	50-70°	---	MCL	100, 001 good	Greenish- black	H 4.5 G 3.60	Gel with hot HCl. Tw 001. Pleoc, X pale brown, Y pale brownish- green, Z deep emerald- green. Fe_2O_3 29.6, FeO 8.7%.
	1.775	1.815	1.825	.050	PASCOITE $\text{Ca}_3\text{V}_{10}\text{O}_{28} \cdot 17\text{H}_2\text{O}$	50° $r > v$ very str	$X = \underline{b}$	MCL	010 poor	Dark red- orange	H 2.5 G 2.46 F easy	Diss in H_2O . Pleoc, X light yellow, Y yellow, Z orange.
1.784 ^ 1.857	1.791	1.815	1.822	.031	ALLANITE (Epidote grp) $(\text{Ce,Ca,Y})_2(\text{Al,Fe})_3\text{Si}_3$ $\text{O}_{12}(\text{OH})$	40° (56+8°) $r > v$ str	$Y = \underline{b}$ $X:\underline{c} = 20^\circ$	MCL	001, 100, 110 poor	Brownish- black	H 6 G 3.95 F 3	Gel with HCl. Pleoc, X cols to red-brown, Y yellow-brown to pale green, Z green-brown to dark red-brown. Fe_2O_3 10.3, FeO 7.3, MnO 6.7, TiO_2 2.0, ThO_2 1.0%.
	1.715	1.817	1.820	.105	JAROSITE (Alunite grp) $\text{KFe}_3(\text{SO}_4)_2(\text{OH})_6$	Small	---	TRIG	0001 good	Amber yellow	H 3 G 3.2	Diss by HCl. Pleoc, X pale yellow, Y and Z brownish-yellow.
1.790 ^ 1.828	1.777	1.818	1.828	.051	FERROHORTONOLITE, manganon (Olivine grp) $(\text{Fe,Mn})_2\text{SiO}_4$	52° $r > v$	$X = \underline{b}$ $Z = \underline{a}$	ORTH	010 fair	Reddish- brown	H 6 G 4.21 F 3.5	Gel with acids. FeO 41.5, MnO 21.4, MgO 4.5, Fe_2O_3 0.7%.
v 1.85	1.770	(1.82)	1.820	.050	MITRIDATITE $\text{Ca}_3\text{Fe}^{+3}_4(\text{PO}_4)_4(\text{OH})_6 \cdot$ $3\text{H}_2\text{O}$	Small	$X \sim \perp 100$	MCL	---	Brownish- green	G 3.24	Pleoc, X pale greenish- yellow, Y and Z deep greenish-brown.
v 1.889	1.730	1.820	1.93	.20	IRIGINITE $(\text{UO}_2)\text{Mo}_2\text{O}_7 \cdot 3\text{H}_2\text{O}$	Large	$Y = \underline{b}$ $Z = \underline{c}$	MCL	---	Canary yellow	H 4-5 G 3.94	Diss by warm HCl. Not fluor in UV.
	1.715	1.820	1.880	.165	DOLEROPHANITE $\text{Cu}_2(\text{SO}_4)\text{O}$	105° (?) (70+2°) $r > v$ very str, crossed	$Y = \underline{b}$ $Z:\underline{c} = -10^\circ$	MCL	101 perf	Chestnut- to dark brown	H 3 G 4.17 fus	Diss by acids. Pleoc, X deep brown, Y brown- ish-yellow, Z lemon- yellow. Reported as opt pos.
	1.762	1.820	1.825	.063	CORNETITE $\text{Cu}_3(\text{PO}_4)(\text{OH})_3$	35° $r < v$ str	$X = \underline{a}$ $Z = \underline{b}$	ORTH	---	Light to dark blue	H 5 G 4.10	Diss by acids. Not pleoc.
	1.770	1.821	1.860	.090	CALCURMOLITE $\text{Ca}(\text{UO}_2)_3(\text{MoO}_4)_3(\text{OH})_2 \cdot$ $11\text{H}_2\text{O}$	(80+3°)	---	ORTH pris	---	Honey- yellow	---	Pleoc, X cols, Y pale yellow, Z yellow. Fluor bright yellow- green in UV.
	1.753	1.824	1.830	.077	IIMORIITE $\text{Y}_2(\text{SiO}_4)(\text{CO}_3)$	31°	---	TCL	011 good	Gray to tan	H 5.5-6 G 4.47	---
1.805 ^	1.730	1.825	1.830	.100	BECQUERELITE $\text{CaU}_6\text{O}_{19} \cdot 11\text{H}_2\text{O}$	30° $r > v$	$X = \underline{c}$ $Y = \underline{b}$ el cTv pos	ORTH ps hex	001 perf	Golden- to lemon- yellow	H 2.5 G 5.1	Pleoc, X cols to pale yellow, Y and Z deep yellow.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.81 ┌	1.788	<u>1.825</u>	1.830	.042	HANCOCKITE (Epidote grp) (Pb,Ca,Sr) ₂ (Al,Fe) ₃ Si ₃ O ₁₂ (OH)	38° r > v	Y = \underline{b}	MCL	001	Brownish-red	H 6.5 G 4.03 F 3	Abs Z > X.
1.818 └ 1.843	1.788	<u>1.828</u>	1.840	.052	FERROHORTONOLITE (Olivine grp) (Fe,Mg) ₂ SiO ₄	58° r > v	X = \underline{b} Z = \underline{a}	ORTH	010 fair	Reddish-brown	H 6 G 4.15 F 3.5	Gel with acids. FeO 58.6, MnO 0.85% (Fe ₁₇).
	1.672	<u>1.83</u>	1.83	.16	ROSASITE (Cu,Zn) ₂ (CO ₃)(OH) ₂	Small r < v str	X = el	MCL	---	Bluish-green	G 4.09 fus	Diss by acids with eff. Pleoc, X cols, Y and Z blue.
┌ 1.85	1.69	<u>1.83</u>	1.83	.14	GLAUKOSPHERITE (Cu,Ni) ₂ (CO ₃)(OH) ₂	Small	X:c = 7°	MCL fib spherulitic	001 good	Green	H 3-4 G 3.78-3.96	Diss by acids with eff. Pleoc, X deep green, Y and Z apple-green.
	1.66	<u>1.831</u>	1.915	.255	UMOHOITE (UO ₂)MoO ₄ ·4H ₂ O	65° r > v str	Y = \underline{b} X ~ \underline{a}	MCL tab	---	Blue-black	H 2 G 4.6	Pleoc, X dark blue, Y light blue, Z olive-green.
(1.769) ^	1.800	<u>1.831</u>	1.846	.046	CONICALCITE CaCuAsO ₄ (OH)	Large r > v str	X = \underline{a} Y = \underline{b}	ORTH pris	011 fr uneven	Dark- to yellow-green	H 4-5 G 4.33 F 3	Diss by acids. Pleoc, X green, Y yellow-green, Z bluish-green.
1.780 ^	1.733	<u>1.832</u>	1.839	.106	BILLIETITE BaU ₆ O ₁₉ ·11H ₂ O	35° (28+4°) r > v str	X = \underline{c} Y = \underline{a} el cTv pos	ORTH ps hex	001 perf	Golden-yellow	G 5.32	Tw common on 111 and 110. Pleoc, X and Z pale green, Y bright green.
1.815 ┌	1.750	<u>1.832</u>	1.832	.082	NATROJAROSITE (Alunite grp) NaFe ₃ (SO ₄) ₂ (OH) ₆	Very small	X = \underline{c} Y = \underline{a} el cTv pos	TRIG ps orth	0001 perf	Yellow-brown	H 3 G 3.2 F 3	Diss by HCl. Pleoc, X nearly cols, Y and Z pale yellow.
	---	<u>1.833</u>	---	wk	RUSAKOVITE (Fe,Al) ₅ (VO ₄ ,PO ₄) ₂ (OH) ₉ ·3H ₂ O	---	---	Crypto-cryst	---	Yellow-orange	H 1.5-2 G 2.76	Diss by acids. Opt char unk.
	1.783	<u>1.834</u>	1.866	.083	DELRIOITE SrCaV ₂ O ₆ (OH) ₂ ·3H ₂ O	Med to large	---	MCL acic	---	Yellow-green	H 2 G 3.1	Diss by H ₂ O. Tw common 100. Pleoc, X cols, Y pale yellow, Z deeper yellow.

1.805 ^	(1.664)	<u>1.835</u>	1.865	(.201)	TYUYAMUNITE $\text{Ca}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 5-8\text{H}_2\text{O}$	45° r < v wk	$X = \frac{c}{r}$ $Y = \frac{a}{r}$	ORTH	001 perf 010, 100 good	Canary- to greenish- yellow	H 2 G 3.85 F easy	Pleoc wk, X cols, Y and Z greenish-yellow. Not fluor in UV.
1.813 ^	1.805	<u>1.835</u>	1.862	.057	BAFERTISITE $\text{Ba}(\text{Fe,Mn})_2\text{TiSi}_2\text{O}_7$ (O,OH) ₂	86°	$Y = \frac{b}{Z:c} = 6^\circ$ el clv neg	MCL	001 perf	Orange to red	H 5 G 3.97- 4.25	Poly tw 001. Pleoc, X reddish-brown, Y yellow, Z greenish- yellow.
1.790 ^ 1.86	1.789	<u>1.835</u>	1.845	.056	PSEUDOMALACHITE $\text{Cu}_5(\text{PO}_4)_2(\text{OH})_4 \cdot \text{H}_2\text{O}$	50° r < v str	$X:c = 21^\circ$	MCL pris	010 good	Dark green	H 4.5-5 G 4.35 F 2	Diss by acids. Tw 100. Pleoc wk.
	1.809	<u>1.838</u>	1.859	.050	LINARITE $\text{PbCu}(\text{SO}_4)(\text{OH})_2$	80° r < v str	$Z = \frac{b}{X:c} = -24^\circ$	MCL el 010	100 perf	Deep azure- blue	H 2.5 G 5.35 F easy	Diss by HNO_3 . Tw com- mon 100, less so 001. Pleoc, X pale blue, Y clear blue, Z Prussian blue.
	1.81	---	1.84	.03	DEERITE $(\text{Fe}^{+2}, \text{Mn})_6(\text{Fe}^{+3}, \text{Al})_3$ $\text{Si}_6\text{O}_{20}(\text{OH})_5$	---	$Z = \frac{c}{el}$ pos	MCL or ORTH acic	110 good	Black	G 3.84	Opt char unk.
	1.80	---	1.87	.07	KEYITE $(\text{Cu,Zn,Cd})_3(\text{AsO}_4)_2$	Disp str	$X:c = 10^\circ$ $Y = \frac{b}{r}$	MCL	001 good	Deep sky- blue	---	Diss by conc acids. Pleoc, X pale blue, Y greenish blue, Z deep blue. Opt char unk.
	1.760	<u>1.840</u>	1.870	.110	CALCIOURANOITE $(\text{Ca,Ba,Pb})\text{U}_2\text{O}_7 \cdot$ $5\text{H}_2\text{O}$	(61±2°)	$X = \frac{c}{el}$ neg	ORTH(?)	---	Brown to orange	H 4 G 4.02- 4.28	Pleoc, X and Y lemon- yellow, Z cols.
	1.775	<u>1.840</u>	1.844	.069	CHALCOSIDERITE $\text{CuFe}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 4\text{H}_2\text{O}$	22° r > v str crossed	---	TCL pris	001 perf 010 good	Siskin- green	H 4.5 G 3.22	Fe-analogue of Turquoise. Pleoc wk.
	1.795	<u>1.842</u>	1.874	.079	STRANSKIITE $\text{Zn}_2\text{Cu}(\text{AsO}_4)_2$	80°	---	TCL	010 perf 100 good	Blue	H 4 G 5.23	---
	1.825	<u>1.842</u>	1.857	.032	DIETZEITE $\text{Ca}_2(\text{IO}_3)_2(\text{CrO}_4)$	86° r < v very str	$Y = \frac{b}{Z:c} = 6^\circ$	MCL tab	100 poor fr conch	Golden- yellow	H 3.5 G 3.62	Diss by hot H_2O .
1.828 ^ 1.850	1.803	<u>1.843</u>	1.851	.048	FAYALITE, manganoean (Olivine grp) $(\text{Fe,Mn})_2\text{SiO}_4$	48° r > v wk	$X = \frac{b}{Z} = \frac{a}{r}$	ORTH	010 good	Reddish- brown	H 6 G 4.3 F 3.5	Gel with acids. FeO 56.9, MnO 9.7, MgO 1.1, Fe ₂ O ₃ 1.2%.
	---	<u>1.84-</u> <u>1.85</u>	---	---	ITOITE $\text{Pb}_3\text{Ge}(\text{SO}_4)_2\text{O}_2(\text{OH})_2$	---	---	ORTH acic	---	White	G 6.7 calc	Opt char unk.
	1.65	<u>1.85</u>	1.90	.25	VANURALITE $\text{Al}(\text{UO}_2)_2(\text{VO}_4)_2(\text{OH}) \cdot$ $11\text{H}_2\text{O}$	44° r < v wk	$Z = \frac{b}{X} = \frac{c}{r}$	MCL	001 easy	Lemon- yellow	H 2 G 3.62	Tw 001. Pleoc str, X cols, Y and Z yellow.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	1.77	<u>1.85</u>	1.89	.12	DERRIKSITE $\text{Cu}_4(\text{UO}_2)(\text{SeO}_3)_2(\text{OH})_6 \cdot \text{H}_2\text{O}$	(68+19°)	$X = \frac{a}{b}$ $Y = \frac{b}{a}$	ORTH	010 good	Green	G 4.72	----
	1.740	---	1.855	.115	SINOITE $\text{Si}_2\text{N}_2\text{O}$	Small	$Z = \frac{c}{a}$	ORTH	---	Gray	G 2.82	---
	1.79	<u>1.85</u>	1.89	.10	MROSEITE $\text{CaTe}^{+4}(\text{CO}_3)_2$	74° $r < v$ str	$X = \frac{a}{b}$ $Y = \frac{c}{a}$	ORTH	---	Cols to white	H 4 G 4.35	Diss by cold acids with eff.
1.810 ^	1.780	<u>1.850</u>	1.860	.080	VANDENDRIESSCHEITE $\text{PbU}_{70}\text{O}_{22} \cdot 12\text{H}_2\text{O}$	60° (40+4°) $r > v$ str	$X = \frac{c}{a}$ $Z = \frac{a}{b}$ el cTv pos	ORTH tab	001 perf	Orange-to yellow-brown	H 3 G 4.5-5.5	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
(1.82) ^	1.785	<u>1.85</u>	1.85	.065	MITRIDATITE $\text{Ca}_2\text{Fe}^{+3}_4(\text{PO}_4)_4(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	5-10°	$X \sim \perp 100$	MCL	---	Deep red to brownish-green	G 3.24	Pleoc, X pale greenish-yellow, Y and Z deep greenish-brown.
	1.82	<u>1.85</u>	1.85	.03	BURKHARDTITE $\text{Pb}_2(\text{Fe}^{+3}, \text{Mn}^{+3})\text{Te}^{+4}\text{AlSi}_3\text{O}_{12}(\text{OH})_2 \cdot \text{H}_2\text{O}$	Small	$X = \frac{c}{a}$	MCL ps hex	001 perf	Carmine-to violet-red	H 2 G 4.96	Insol in acids. Pleoc, X pale magenta, Y and Z carmine-red.
1.843 ^ 1.854	1.816	<u>1.850</u>	1.863	.047	FAYALITE (Olivine grp) $(\text{Fe}, \text{Mg})_2\text{SiO}_4$	60° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010 fair	Dark brown	H 6 G 4.36 F 3.5	Gel with acids (Fe_7).
	1.847	<u>1.850</u>	1.850	.003	ROMEITE (Stibiconite grp) $(\text{Ca}, \text{Fe}, \text{Mn}, \text{Na})_2(\text{Sb}, \text{Ti})_2\text{O}_6(\text{O}, \text{OH}, \text{F})$	Med	---	CUB oct	Conch	Brown	H 6 G 5.07 infus	Insol in acids. Opt anom. Segmented. Poly tw. Abnormal interf colors.
1.850 ^ 1.854 pos 1.869 v	1.820	<u>1.854</u>	1.888	.068	LIEBENBERGITE (Olivine grp) $(\text{Ni}, \text{Mg}, \text{Fe}, \text{Co})_2\text{SiO}_4$	88° $r > v$	$X = \frac{b}{a}$ $Z = \frac{a}{b}$	ORTH	010 poor to fair 100 poor	Yellowish-green	H 6-6.5 G 4.60 calc	Pleoc, X and Y cols to pale green, Z greenish yellow. NiO 56.3, MgO 6.5, FeO 9.4, CoO 1.8%.
1.907 v	1.674	<u>1.855</u>	1.880	.206	STRELKINITE $\text{Na}(\text{UO}_2)(\text{VO}_4) \cdot 3\text{H}_2\text{O}$	Med	$X = \frac{c}{a}$ $Y = \frac{b}{a}$ el cTv pos	ORTH plates	001 perf	Gold- to canary-yellow	H 2-2.5 G 3.70-3.91	Diss by acids. Pleoc wk, Y yellow, Z pale yellow. Fluor wk, dirty green in UV.

1.815 ^	1.813	1.857	1.891	.078	ALLANITE (Epidote grp) $(\text{Ce,Ca,Y})_2(\text{Al,Fe})_3\text{Si}_3\text{O}_{12}(\text{OH})$	80° r > v str	$Y = \frac{b}{c}$ $X:c = 34^\circ$	MCL	001, 100 poor	Brownish- black	H 6 G 4.13 F 3	Pleoc, X red-yellow, Y yellow to brown, Z dark red-brown. Fe_2O_3 15.2, FeO 8.9, MnO 1.9%.
~1.81 ^	1.85	---	1.862	.01	PARSONSITE $\text{Pb}_2(\text{UO}_2)(\text{PO}_4)_2 \cdot 2\text{H}_2\text{O}$	---	$Y = \frac{b}{c}$ $Z:c = 2^\circ$	TCL	010 poor fr conch	Pale yellow	H 2.5-3 G 5.73	Not fluor in UV. Opt char unk.
1.835 ^	1.80	1.86	1.88	.080	PSEUDOMALACHITE $\text{Cu}_5(\text{PO}_4)_2(\text{OH})_4 \cdot \text{H}_2\text{O}$	46° r < v str	$X:c = 23^\circ$	MCL pris	010 good	Dark green	H 4.5-5 G 4.35 F 2	Diss by acids. Tw 100. Pleoc wk.
	1.820	1.86	1.88	.06	CORNWALLITE $\text{Cu}_5(\text{AsO}_4)_2(\text{OH})_4 \cdot \text{H}_2\text{O}$	Small r < v med	$Y = \frac{b}{c}$ $Z \perp \text{clv}$	MCL pris	One perf	Emerald- green	H 4.5 G 4.52 F 2.5	Diss by acids.
	1.831	1.861	1.880	.049	ATACAMITE $\text{Cu}_2\text{Cl}(\text{OH})_3$	75° r < v str	$X = \frac{b}{c}$ $Y = \frac{a}{b}$	ORTH pris tab	010 perf 101 fair	Bright green	H 3.5 G 3.76 F 3.5	Diss by acids. Pleoc, X pale green, Y yellow- green, Z grass-green.
	1.818	1.866	1.909	.091	CALEDONITE $\text{Pb}_5\text{Cu}_2(\text{CO}_3)(\text{SO}_4)_3(\text{OH})_6$	~ 85° r < v wk	$X = \frac{c}{a}$ $Y = \frac{a}{b}$	ORTH	010 perf	Bluish- green	H 2.5-3 G 5.76 fus	Commonly striated.
1.854 ^	1.827	1.869	1.879	.052	FAYALITE (Olivine grp) Fe_2SiO_4	48° r > v	$X = \frac{b}{c}$ $Z = \frac{a}{b}$	ORTH	010 fair	Dark brown	H 6 G 4.22 F 3.5	Gel with acids. FeO 65.0, Fe_2O_3 0.4, MnO 1.0%.
1.93 v	1.675	1.870	1.885	.210	METATYUYAMUNITE $\text{Ca}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3-5\text{H}_2\text{O}$	30-45° r < v	$X = \frac{c}{a}$ $Y = \frac{a}{b}$	ORTH	001 perf 010, 100 good	Canary- to greenish- yellow	H 2 G 3.85 F easy	Pleoc wk, X cols, Y and Z pale yellow. Not fluor in UV.
(1.92) v	1.727	1.870	1.883	.156	ILVAITE $\text{Ca}(\text{Fe}^{+2}, \text{Mn})_2\text{Fe}^{+3}(\text{SiO}_4)_2(\text{OH})$	20-30° r < v str	$X = \frac{c}{a}$ $Y = \frac{b}{c}$	ORTH	010 good 001 dist	Brownish- black	H 5.5-6 G 3.98 F 2.5	Gel with acids. Pleoc, X dark green, Y yellow- brown to dark brown, Z dark brown, abs X > Y > Z.
	1.81	1.87	1.92	.11	BAHIANITE $\text{Al}_5\text{Sb}_3\text{O}_{14}(\text{OH})_2$	Large r > v	---	MCL ps orth	---	Tan, brown, cols	G 4.9-5.46 (5.26 calc)	---
1.898 v	1.792	1.870	1.870	.078	ARSENIOSIDERITE $\text{Ca}_3\text{Fe}^{+3}_4(\text{AsO}_4)_4(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	~ 0°	el pos	MCL ps tet fib	001 perf	Yellow to brown	H 4.5 G 3.60 F 3	Diss by acids. Pleoc, X cols, Y and Z dark reddish-brown.
	1.756	1.874	1.896	.140	CLINOCLASE $\text{Cu}_3\text{AsO}_4(\text{OH})_3$	50° (44+2°) r < v very str	$Y = \frac{b}{c}$ $Z \sim \frac{a}{b}$ el cTv pos	MCL pris	001 perf	Greenish- blue	H 2.5-3 G 4.38	Diss by acids. Pleoc, X pale blue-green, Y light blue-green, Z blue-green.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
300												
2.02	1.655	<u>1.875</u>	1.909	.254	MALACHITE $\text{Cu}_2\text{CO}_3(\text{OH})_2$	44° $r < v$ str	$Y = \frac{b}{c}$ $X:\underline{c} = 23^\circ$	MCL	$20\bar{1}$ perf 010 good	Bright green	H 3.5-4 G 4.05 F 2	Diss by acids with eff. Pleoc, X nearly cols, Y yellow-green, Z deep green.
	1.786	<u>1.875</u>	1.875	.089	PLUMBOJAROSITE (Alunite grp) $\text{PbFe}_6(\text{SO}_4)_4(\text{OH})_{12}$	Small	$X \perp$ plates	TRIG plates	$10\bar{1}4$ fair	Brown	G 3.63	Diss by HCl. Pleoc, X golden-yellow, Y and Z brownish-red. Basal plates divided into sectors.
	1.810	<u>1.880</u>	1.925	.115	Unknown, labeled "Scorodite"	$\sim 90^\circ$ ($75+2^\circ$)	---	---	---	Bright green	---	Data of Larsen, 1921, p. 132.
	---	<u>1.88</u>	---	.10	CHEVKINITE (Ca,Ce,Th) $_4$ (Fe,Mg) $_2$ (Ti,Fe $^{+3}$) $_3\text{Si}_4\text{O}_{22}$	Med	$Z = \frac{b}{c}$ $X:\underline{c} = 11-26^\circ$	MCL	Fr conch	Velvet-black	H 5 G 4.3-4.6 F 4	Gel with acids. Pleoc, X nearly cols, Y pale red-brown, Z dark red-brown. In part isotropic, metamict.
1.927	1.860	<u>1.880</u>	1.893	.033	FRONDELITE $\text{MnFe}^{+3}_4(\text{PO}_4)_3(\text{OH})_5$	Med $r > v$ str	$X = \underline{c}$	ORTH fib	100 perf 010 good 001 poor	Brown	H 4.5 G 3.48	Pleoc, X pale yellow-brown, Y and Z orange-brown, abs $Z > Y > X$. Fe_2O_3 48.8, MnO 7.7%.
	---	<u>1.88</u>	---	---	RAUVITE $\text{Ca}(\text{UO}_2)_2\text{V}^{+5}_{10}\text{O}_{28}$	---	---	Mass	---	Purplish-black	---	Opt char unk.
	>1.87	---	<1.89	---	STIBIVANITE $\text{Sb}^{+3}\text{V}^{+4}_4\text{O}_5$	85° $r > v$ str	$X = \frac{b}{c}$ $Z = \underline{c}$	MCL fib	---	Yellow-green	---	Pleoc str, X emerald-green with olive tint, Y emerald-green, Z olive-green. Sign unk.
1.92	1.863	<u>1.885</u>	1.890	.027	FOURMARIERITE $\text{PbU}_4\text{O}_{13} \cdot 4\text{H}_2\text{O}$	50° $r > v$	$X = \frac{c}{a}$ $Y = \underline{a}$	ORTH tab	001 perf 100 good	Reddish-orange	H 3.5 G 5.74	Pleoc, X cols, Y pale yellow, Z amber.
[]	1.86	<u>1.887</u>	1.91	.05	HETEROSITE (Fe $^{+3}$,Mn $^{+3}$) PO_4	37°	$X = \underline{a}$	ORTH	100 good 010 poor	Deep rose to purple	H 4 G 3.3	Diss by acids. Pleoc, X greenish-gray, Y and Z deep red to purple, abs $Z = Y > X$. Abnormal green interf colors near optic axis.
1.820 \wedge	1.764	<u>1.889</u>	1.936	.172	IRIGINITE (UO_2) $\text{Mo}_2\text{O}_7 \cdot 3\text{H}_2\text{O}$	60°	$Y = \frac{b}{c}$ $Z = \underline{c}$	MCL	---	Canary yellow	H 4-5 G 3.84	Diss by warm HCl. Not fluor in UV.

	(1.80)	<u>1.89</u>	1.91	.11	WYARTITE $\text{Ca}_3\text{U}^{+4}(\text{UO}_2)_6(\text{CO}_3)_2(\text{OH})_{18} \cdot 3\text{-}5\text{H}_2\text{O}$	48°	$X = c$ $Y = \underline{b}$	ORTH	001 perf and 010	Black to violet- black	H 3-4 G 4.69	Pleoc, X gray, Y violet, Z lavender- blue.
1.870 ^	1.815	<u>1.898</u>	1.898	.083	ARSENIOSIDERITE $\text{Ca}_3\text{Fe}^{+3}_4(\text{AsO}_4)_4(\text{OH})_6 \cdot 3\text{H}_2\text{O}$	~ 0°	el pos	MCL ps tet fib	001 perf	Yellow to brown	H 4.5 G 3.60 F 3	Diss by acids. Pleoc, X cols, Y and Z dark red-brown.
1.925 v	1.76	<u>1.90</u>	1.92	.16	CARNOTITE $\text{K}_2(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	50° $r < v$	$X = c$ $Y = \underline{b}$ $Z:a = 14^\circ$	MCL	001 perf 010, 110 poor	Lemon- to bright- yellow	H soft G 4.70 F diff	Pleoc, X cols to gray, Y and Z canary- to lemon-yellow. Not fluor in UV.
	1.775	<u>1.900</u>	1.920	.145	XOCOMEATLITE $\text{Cu}_3(\text{TeO}_4)(\text{OH})_4$	41±2°	---	ORTH(?) fib	---	Rich green	H 4 G 4.65	Diss by HNO_3 . Pleoc in bluish-green, abs Z > X = Y.
	1.899	<u>1.901</u>	1.903	.004	QUEITITE $\text{Pb}_4\text{Zn}_2(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{SO}_4)$	~ 90° $r < v$ str	---	MCL tab	010, 001 poor	Pale yellow to cols	H 4 G 6.07	Diff sol in HNO_3 .
	1.803	<u>1.905</u>	1.945	.142	WIDENMANNITE $\text{Pb}_2(\text{UO}_2)(\text{CO}_3)_3$	63°	$X = b$ $Y = \underline{a}$ el pos	ORTH tab 010	010 perf	Yellow	H 2 G 6.89 calc	Diss by HNO_3 with eff. Not pleoc. Not fluor in UV.
1.895 ┐	1.873	<u>1.905</u>	1.910	.037	YEATMANITE $\text{Mn}_7\text{Zn}_8\text{Sb}_2\text{Si}_4\text{O}_{28}$	49° $r < v$ med	---	TCL	100 perf	Deep brown	H 4 G 5.02 F 4	Dec by HCl. Tw on 010.
	1.785	<u>1.906</u>	1.917	.132	MASUYITE $\text{UO}_2 \cdot 2\text{H}_2\text{O} (?)$	50° (32±3°) $r > v$	$X = c$ $Y = \underline{b}$	ORTH tab 001	001 perf 010 good	Reddish- to brownish- orange	G 5.08	Tw 110 and 130 common. Pleoc, X pale yellow, Y and Z deep golden. Not fluor in UV.
1.855 ^	1.770	<u>1.907</u>	1.915	.145	STRELKINITE $\text{Na}_2(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 6\text{H}_2\text{O}$	Med (26±3°)	$X = c$ $Y = \underline{b}$ el pos	ORTH plates	001 perf	Gold- to canary- yellow	H 2-2.5 G 3.7-3.91	Diss by acids. Pleoc wk, Y yellow, Z pale yellow. Fluor wk, dirty green in UV.
	1.685	<u>1.91</u>	1.93	.245	IANTHINITE $\text{UO}_2 \cdot 5\text{UO}_3 \cdot 10\text{H}_2\text{O}$	Large (30±15°)	$X = c$ $Y = \underline{b}$	ORTH acic	001 perf 100 good	Black to violet	H 2.5	Pleoc, X cols, Y purple, Z violet.
1.952 v	1.750	<u>1.910</u>	1.945	.195	FRANCEVILLITE $\text{Ba}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 5\text{H}_2\text{O}$	46°	$X = c$ $Y = \underline{b}$ el pos	ORTH	001 perf	Yellow	H 3 G 4.0	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
	1.885	<u>1.910</u>	1.913	.028	BARTELKEITE $\text{PbFe}^{+2}\text{Ge}_3\text{O}_8$	35° $r < v$	$Z = \underline{b}$	MCL	101 dist	Cols to pale green	H ~ 4 G (4.97)	Diss by hot HCl.
	1.820	<u>1.920</u>	1.955	.135	LAZARENKOITE $(\text{Ca}, \text{Fe}^{+2})\text{Fe}^{+3}\text{As}^{+3}_7 \cdot 3\text{H}_2\text{O}$	~ 30°	el pos	ORTH fib	---	Orange	H 1 G 3.45	Diss by HCl. Pleoc, X pale yellow, Y pale brown, Z rose-brown.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.870	1.830	(1.92)	1.925	.095	ILVAITE $\text{Ca}(\text{Fe}, \text{Mn})_2\text{Fe}^{+3}(\text{SiO}_4)_2(\text{OH})$	20-30° $r < v$ str	$X = \frac{c}{b}$ $Y = \frac{a}{b}$	ORTH	010 good 001 dist	Brownish-black	H 5.5-6 G 3.98 F 2.5	Gel with acids. Pleoc, X dark green, Y and Z dark brown, abs X > Y > Z.
1.885	1.85	1.92	1.94	.09	FOURMARIERITE $\text{PbU}_{40}\text{O}_{13} \cdot 4\text{H}_2\text{O}$	Large $r > v$	$X = \frac{c}{a}$ $Y = \frac{a}{a}$	ORTH tab	001 perf 100 good	Reddish-orange	H 3.5 G 5.74	Pleoc, X cols, Y pale yellow, Z deeper yellow.
	1.810	1.923	1.933	.123	KOLFANITE $\text{Ca}_2\text{Fe}^{+3}(\text{AsO}_4)_3\text{O}_2 \cdot 2\text{H}_2\text{O}$	5-7° (24+4°)	---	MCL	One	Red to yellow	G 3.3	Pleoc, X pale yellow, Y and Z dark orange.
1.90	1.750	1.925	1.950	.200	CARNOTITE $\text{K}_2(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3\text{H}_2\text{O}$	43° (38+2°) $r < v$	$X = \frac{c}{b}$ $Y = \frac{b}{b}$ $Z:a = 14^\circ$	MCL	001 perf 010, 110 poor	Lemon- to bright yellow	H soft G 4.70 F diff	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
	---	1.926	---	.04	Unnamed lead-zinc arsenate	~ 90° $r > v$	---	MCL tw	---	Yellow-brown to cols	---	Am. Mineral., 47, 418 (1962).
1.880	1.915	1.927	1.939	.024	ROCKBRIDGEITE, oxidized (Frondelite ser) $(\text{Fe}^{+2}, \text{Mn})\text{Fe}^{+3}_4(\text{PO}_4)_3(\text{OH})_5$	Large $r < v$	---	ORTH	100 perf 010 good	Green to brown	H 4.5 G 3.48	Diss by acids. Pleoc, X pale yellow-brown, Y and Z orange-brown. Fe_2O_3 55.2, MnO 4.1%.
1.870	1.77	1.93	1.97	.20	METATYUYAMUNITE $\text{Ca}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 3-5\text{H}_2\text{O}$	40-55° $r < v$	$X = \frac{c}{a}$ $Y = \frac{a}{a}$	ORTH	001 perf 010, 100 good	Canary- to greenish-yellow	H 2 G 3.85 F easy	Pleoc, X nearly cols, Y canary yellow, Z darker yellow. Not fluor in UV.
	1.886	1.930	1.939	.053	FERSMANITE $(\text{Ca}, \text{Ce}, \text{Na})(\text{Nb}, \text{Ta}, \text{Ti})_2(\text{O}, \text{OH}, \text{F})$	0-7° (48+6°)	$Y = \frac{b}{c}$ $X \sim \frac{c}{c}$	MCL	---	Yellow to brown	H 5.5 G 3.44	---
1.87	---	1.93	---	wk	CORKITE $\text{PbFe}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$	Med $r < v$ very str	$X = \frac{c}{c}$	TRIG	0001	Olive-green to brown	H 4 G 4.20 F 4-5	Basal section divided into biax segments. Abnormal interf colors green, blue, brown.
	1.922	1.933	1.938	.016	PLUMBOTSUMITE $\text{Pb}_5\text{Si}_4\text{O}_8(\text{OH})_{10}$	32° (68+15°) $r < v$	---	ORTH ps hex	001 perf	Cols	H 2 G 5.6	Diff sol hot HNO_3 .

1.77	<u>1.935</u>	1.960	.190	SENGIERITE $\text{Cu}_2(\text{UO}_2)_2(\text{VO}_4)_2(\text{OH})_2 \cdot 6\text{H}_2\text{O}$	38° r < v str	X = \underline{c}	MCL plates	001 perf	Greenish-yellow	H 2.5 G 4.41	Diss by acids. Pleoc, X pale greenish-yellow, Y and Z greenish-yellow.
---	<u>1.94</u>	1.95	---	SAYRITE $\text{Pb}_2(\text{UO}_2)_5\text{O}_6(\text{OH})_2 \cdot 4\text{H}_2\text{O}$	Large	Y = \underline{b} X ~ \underline{c} Z ~ \underline{a} el neg	MCL	102 dist	Yellowish to reddish-orange	G (6.76)	---
1.897	<u>1.940</u>	1.942	.045	SCHIEFFELINITE $\text{Pb}(\text{Te,S})\text{O}_4 \cdot \text{H}_2\text{O}$	24° r < v wk	X = \underline{b} Y = \underline{c}	ORTH	010 easy	Cols, white	H 2 G 4.98 F easy	---
(1.79)	<u>1.94</u>	1.96	(.17)	Unnamed Ca-Sr-U oxide	38°	---	ORTH ps hex	001 good	Red-orange	G 5.29	Am. Mineral., <u>45</u> , 254 (1960).
1.70	<u>1.95</u>	2.04	.34	HYDROTUNGSTITE $\text{H}_2\text{WO}_4 \cdot \text{H}_2\text{O}$	52° r < v	Y = \underline{b}	MCL plates	010 poor	Dark- to yellow-green	H 2 G 4.60	Insol in acids, diss by NH_4OH . Pleoc, X cols, Y and Z yellow-green to dark green, abs Z > Y > X.
---	<u>1.95</u>	1.985	---	KLEBELSBERGITE $\text{Sb}_4\text{O}_4(\text{OH})_2\text{SO}_4$	~ 70°	X = \underline{b} Y = \underline{c}	ORTH	---	Yellow	G 4.62	Not pleoc.
(1.75)	<u>1.95</u>	1.97	.22	RAMEAUTE $\text{K}_2\text{CaU}_6\text{O}_{20} \cdot 9\text{H}_2\text{O}$	32°	X = \underline{b} Z:c = 5° el clv pos	MCL ps hex	010 good	Orange	G 5.60	Tw 100.
1.89	<u>1.95</u>	2.02	.13	CEROTUNGSTITE $\text{Ce}_2\text{W}_2\text{O}_6(\text{OH})_3$	(Large)	Z = \underline{b} X ~ \underline{a}	MCL	001 perf	Orange-yellow	H 1	Tw 001.
1.85	<u>1.95</u>	1.96	.11	CHAPMANITE $\text{SbFe}_2(\text{SiO}_4)_2(\text{OH})$	Small	Z = el X \perp flat face	ORTH laths	Conch	Olive-green, deep yellow	H 2.5 G 3.75	Insol in acids.
1.92	<u>1.95</u>	1.96	.04	LARSENITE PbZnSiO_4	80° (?)	X = \underline{a}	ORTH pris	120 good	White	H 3 G 5.90	Luster greasy.
1.92	<u>1.95</u>	1.96	.04	MANGANOSTIBITE $(\text{Mn,Fe})_7\text{SbAsO}_{12}$	Small	---	ORTH fib	---	Black to reddish-brown	H 5 G 4.95 infus	Dec by HCl. Pleoc, X reddish-brown, Z nearly opaque.
1.92	<u>1.95</u>	1.95	.03	KATOPTRITE $(\text{Mn,Mg,Fe})_{13}(\text{Al,Fe})_4\text{Sb}_2\text{Si}_2\text{O}_{28}$	Small r > v str inclined	Y = \underline{b} Z:a = -3°	MCL pris	001 perf	Jet black, red in thin splinters	H 5.5 G 4.56	Insol in acids. Pleoc str, reddish-yellow to reddish-brown.
---	<u>1.95</u>	~1.985	---	KLEBELSBERGITE $\text{Sb}_4\text{O}_4(\text{OH})_2(\text{SO}_4)$	70°	X = \underline{b} Y = \underline{a}	ORTH powdery	---	Yellow to orange-yellow	G 4.62	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.910 $\hat{\vee}$ >2	1.785	<u>1.952</u>	2.002	.217	FRANCEVILLITE (Ba,Pb)(UO ₂) ₂ (VO ₄) ₂ ·5H ₂ O	52°	X = $\frac{c}{b}$ Y = $\frac{b}{a}$ el cTv pos	ORTH	001 perf	Yellow	H 3 G 4.55	Pleoc, X cols, Y and Z yellow. Not fluor in UV.
1.94 $\hat{\vee}$ 1.985	1.702	<u>1.955</u>	1.965	.263	POUGHITE Fe ₂ (TeO ₃) ₂ (SO ₄)·3H ₂ O	22° r > v very str	X = $\frac{b}{a}$ Y = $\frac{a}{c}$	ORTH	010 perf 101 good	Greenish-yellow	H 2.5 G 3.76 fus	Diss by acids. Pleoc, X nearly cols, Y pale greenish-yellow, Z pale sulfur-yellow.
	1.946	<u>1.958</u>	1.964	.018	ALAMOSITE PbSiO ₃	65° r < v very str	Y = $\frac{b}{a}$ Z:c = 8°	MCL Fib	010 perf	Cols	H 4-5 G 6.49 fus	Gel with acids.
\vee 2.115	1.815	<u>1.960</u>	1.960	.145	TLAPALLITE H ₆ (Ca,Pb) ₂ (Cu,Zn) ₃ (SO ₄)(Te ⁺⁴ O ₃)(Te ⁺⁶ O ₆)	Small	---	MCL	---	Green	H 3 G 4.7 (5.05 calc)	CuO 15.6, CaO 8.3, ZnO 0.8%.
	1.73	<u>1.96</u>	1.98	.25	MELANOVANADITE Ca ₂ V ⁺⁴ ₄ V ⁺⁵ ₆ O ₂₅ ·xH ₂ O	Med	Z = b Y:c = 15°	TCL pris	One perf	Black, brown	H 2.5 G 3.48 F easy	Diss by acids. Pleoc, X yellow-brown, Y dark red-brown, Z dark red-brown to opaque.
1.92 \square 1.97	1.94	<u>1.96</u>	1.96	.02	BEUDANTITE PbFe ₃ (AsO ₄)(SO ₄)(OH) ₆	0-50° r > v	X = $\frac{c}{b}$ el cTv pos	TRIG	0001 good	Yellow, brown, green	H 4 G 4.1 F 3.5	Diss by hot HNO ₃ . Complex tw. Base divided into biax segments. Abnormal interf colors.
	1.945	<u>1.966</u>	1.983	.038	OLSACHERITE Pb ₂ (SO ₄)(SeO ₄)	80°	X = $\frac{a}{c}$ Y = $\frac{c}{b}$	ORTH	101 good 010 poor	Cols	H 3-3.5 G 6.55 calc	---
	1.93	<u>1.97</u>	2.01	.08	BISMUTOFERRITE BiFe ₂ (SiO ₄) ₂ (OH)	---	---	ORTH mass	---	Yellow-green	G 4.47	---
	1.871	<u>1.975</u>	2.005	.134	WALPURGITE (BiO) ₄ (UO ₂)(AsO ₄) ₂ O ₄ ·3H ₂ O	52°	Y:c = 8°	TCL tab	010 perf	Yellow to yellow-green	H 5 G 5.95 F 1.5	Tw 010. Pleoc, X pale yellow, Y and Z green.
\vee ~2.09	1.956	<u>1.978</u>	1.980	.024	WOELSENDORFITE (Pb,Ca)U ₂ O ₇ ·2H ₂ O	Large	---	ORTH	001 good	Orange-red	H 5 G 5.37	PbO 15.5, CaO 3.2%.
	1.78	<u>1.98</u>	2.1	.32	KARAPTITE (= Coronene) C ₂₄ H ₁₂	(69±7°)	---	MCL needles	One clv	Yellow	H 1 G 1.40 F 1	Sol in benzene.

	1.89	<u>1.98</u>	2.02	.13	YTTROTUNGSTITE $\text{Yb}_2\text{O}_6(\text{OH})_3$	68°	$Z = b$ $X:c = 26^\circ$	MCL ps orth laths	010 perf 101 poor	Yellow	G 5.96	Tw 100.
\vee >2.0	(1.97)	<u>1.98</u>	1.99	(.02)	RICHETITE Oxide of Pb and U	Large	---	MCL ps hex tab	010 perf	Black	G 7.25	Tw common. Pleoc, X pale brown, Y and Z dull brown.
1.955 ^	1.72	<u>1.985</u>	1.990	.27	POUGHITE $\text{Fe}_2(\text{TeO}_3)_2(\text{SO}_4) \cdot 3\text{H}_2\text{O}$	15-20° r > v str	$X = b$ $Y = \underline{a}$	ORTH	010 perf 101 good	Greenish- yellow	H 2.5 G 3.76 fus	Diss by acids. Pleoc, X nearly cols, Y pale greenish-yellow, Z pale yellow.
	1.952	---	2.002	.05	KURANAKHITE $\text{PbMn}^{+4}\text{Te}^{+6}\text{O}_6$	---	---	ORTH ps hex	---	Reddish- brown to black	H 4.5	Diss by hot HCl. Opt char unk.
	1.970	<u>1.992</u>	2.011	.041	Unnamed arsenate analogue of Tsumebite $\text{Pb}_2\text{Cu}(\text{AsO}_4)(\text{SO}_4)(\text{OH})$	88°	---	MCL	---	Green	---	Pleoc, X pale green, Y and Z bottle-green. Am. Mineral., <u>51</u> , 258 (1966).
	1.990	<u>1.993</u>	1.994	.004	ELYITE $\text{Pb}_4\text{Cu}(\text{SO}_4)(\text{OH})_8$	66°	$Y = b$ $X:c = 45^\circ$	MCL pris tab	---	Lavender	H 2 G 6.3	Pleoc str in shades of lavender, abs Z > Y > X.
	1.960	<u>1.995</u>	2.020	.060	JAMESITE $\text{Pb}_2\text{Zn}_2\text{Fe}^{+3}_5\text{O}_4(\text{AsO}_4)_5$	75° r > v	$Y:a$ on (001) = 5°	TCL tab	---	Reddish- brown	H ~ 3	Pleoc, X and Y pale brown, Z deep reddish- brown.
1.952 ^	---	>2	---	---	CURIENITE $\text{Pb}(\text{UO}_2)_2(\text{VO}_4)_2 \cdot 5\text{H}_2\text{O}$	66°	$X = c$ $Y = \underline{b}$	ORTH	---	Canary yellow	G 4.88	---
	1.82	>2.0	>2.0	>.18	GRANTSITE $\text{Na}_4\text{Ca}_x\text{V}^{+4}_{2x}\text{V}^{+5}_{12-2x}\text{O}_{32} \cdot 8\text{H}_2\text{O}$	---	$Z = \underline{b}$	MCL	---	Olive- green	H 1 G 2.94	Pleoc, X green, Y greenish-brown, Z brown, abs Z > Y > X. Mean \bar{n} calc 1.965.
	1.87	<u>2.00</u>	2.01	.14	LEADHILLITE $\text{Pb}_4(\text{SO}_4)(\text{CO}_3)_2(\text{OH})_2$	10° r < v str	$Z = b$ $X:c = 5^\circ$ el-clv pos	MCL ps hex	001 perf fr conch	Cols, gray, blue, green	H 2.5-3 G 6.55 F 1.5	Diss by HNO_3 with eff.
1.98 ^	~1.8	>2.0	<2.07	---	RICHETITE Hydrous oxide of Pb, U	Large	---	MCL (?)	010 perf	Black	G 7.25	Pleoc, X pale brown, Y and Z dull brown.
	1.797	>2.0	>2.0	>.2	BARNESITE $\text{Na}_2\text{V}_6\text{O}_{16} \cdot 3\text{H}_2\text{O}$	---	$Z = b$ $X:c = 5^\circ$	MCL bladed	---	Dark red	G 3.15 F 2	Diss by HCl. Pleoc, X yellow, Y orange- yellow, Z red. Mean refractive index 2.037.
1.986 ^	1.930	<u>2.002</u>	2.020	.090	LINDGRENITE $\text{Cu}_3(\text{MoO}_4)_2(\text{OH})_2$	71° (52+3°) r > v	$Z = b$ $X:c = -7^\circ$ el-clv neg	MCL tab	010 perf 001, 101 poor	Apple- green	H 4.5 G 4.26 F easy	Diss by acids.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
1.99 v 2.04	1.928	<u>2.007</u>	2.036	.108	LANARKITE $\text{Pb}_2(\text{SO}_4)_0$	60° $r > v$ str inclined	$Y = \frac{b}{c}$ $Z:c = 30^\circ$	MCL el <u>b</u>	201 perf	Gray, greenish	H 2.5 G 6.92	Diss by KOH or warm HNO_3 . Poly tw 010 rare. Fluor yellow in UV.
	(1.85)	<u>2.01</u>	2.06	(.21)	AGRINIERITE $(\text{K}_2, \text{Ca}, \text{Sr})\text{U}_3\text{O}_{10} \cdot 4\text{H}_2\text{O}$	55°	$X = \frac{a}{b}$ $Y = \frac{b}{c}$ el cTv neg	ORTH tab	001 good	Orange	G 5.7	Sector tw.
	1.92	<u>2.01</u>	2.02	.10	LORENZENITE $\text{Na}_2(\text{Ti}, \text{Zr})_2\text{Si}_2\text{O}_9$	25-49° $r > v$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	100 perf 110 good	Brown to black	H 6 G 3.37 F easy	Insol in acids. Pleoc, X and Y reddish-yellow, Z pale yellow.
	1.942	<u>2.010</u>	2.024	.082	WHERRYITE $\text{Pb}_4\text{Cu}(\text{CO}_3)(\text{SO}_4)_2$ $(\text{Cl}, \text{OH})_2\text{O}$	50°	Wavy ext	MCL	---	Light green	G 6.45 (7.2 calc) fus	Diss by HNO_3 .
	1.96	<u>2.01</u>	2.04	.08	BROWNMILLERITE $\text{Ca}_2(\text{Al}, \text{Fe})_2\text{O}_5$	75°	---	ORTH	---	Reddish-brown	G 3.76 infus	Data for synth compd. Pleoc, X and Y yellow-brown, Z dark brown.
	2.00	<u>2.01</u>	2.02	.02	TURANITE $\text{Cu}_5(\text{VO}_4)_2(\text{OH})_4 (?)$	Med $r > v$ str	---	Reniform, radiating, fib	---	Olive-green	H 5	Pleoc, X and Y brown, Z green.
	2.00	<u>2.01</u>	2.02	.02	CALCIOVOLBORTHITE $\text{CaCu}(\text{VO}_4)(\text{OH})$	~ 90° $r < v$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH	010 perf 001 good	Yellow, green, brown	H 3.5 G 3.75	Diss by HCl. Pleoc wk.
	<2.01	> <u>2.01</u>	>2.01	---	HENDERSONITE $\text{Ca}_2\text{V}^{+4}\text{V}^{+5}_8\text{O}_{24} \cdot 8\text{H}_2\text{O}$	Med $r > v$	$X = \frac{a}{b}$ $Y = \frac{b}{c}$	ORTH fib	---	Dark greenish-black	H 2.5 G 2.79	Diss by HNO_3 . Pleoc, X yellow-green, Y green, Z brown. Mean n calc 2.10.
	>2.01	> <u>2.01</u>	>2.01	high	GERSTLEYITE $\text{Na}_2(\text{Sb}, \text{As})_8\text{S}_{13} \cdot 2\text{H}_2\text{O}$	Large (?)	---	MCL (?)	010, 101 perf	Cinnabar-red to reddish-black	H 2.5 G 3.62 F 2	Diss by alkalis. Pleoc, X salmon-red, Y and Z deep blue-red. Opt char unk.
	1.905	<u>2.02</u>	>2.02	>.115	NAVAJOITE $\text{V}_2\text{O}_5 \cdot 3\text{H}_2\text{O}$	---	$Z = \frac{b}{c}$	MCL fib <u>b</u>	---	Dark brown	H 1 G 2.56	Pleoc, X and Y yellow-brown, Z dark brown.

2.01 □ 2.04	1.96	<u>2.02</u>	2.06	.10	PERRIERITE (Ca,Ce,Th) ₄ (Mg,Fe) ₂ (Ti,Fe) ₃ Si ₄ O ₂₂	63° r > v str	Z = $\frac{b}{a}$ X:a = 24°	MCL	---	Brownish-black	H 6 G 4.3-4.45 F 4	Pleoc, X yellow-brown, Y deep reddish-brown, Z dark brown to opaque, abs Z > Y > X.
1.88 ^	1.97	<u>2.02</u>	2.05	.08	CHEVKINITE (Ca,Ce,Th) ₄ (Fe,Mg) ₂ (Ti,Fe) ₃ Si ₄ O ₂₂	Large	Z = $\frac{b}{c}$ X:c = 11- 26°	MCL	Conch	Brown to black	H 5.5 G 4.63 F 4	Tw 001. Pleoc, X dark brown, Y and Z nearly opaque.
	2.018	<u>2.023</u>	2.025	.007	SONORAITE FeTeO ₃ (OH)·H ₂ O	23°	---	MCL plates	---	Yellowish green	H 3 G 3.95	Diss by acids.
2.24 v	1.82	(<u>2.03</u>)	2.04	.22	TUNGSTITE WO ₃ ·H ₂ O	27° r < v	X = $\frac{c}{b}$ Y = $\frac{b}{a}$ el cTv pos	ORTH	001 perf	Dark to light yellow	H 1.5 G 5.52	Diss by NH ₄ OH, insol in HCl.
	1.990	<u>2.030</u>	2.035	.045	ARSENDESCLOIZITE PbZn(AsO ₄)(OH)	~ 30° r > v	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH tab 001	---	Pale yellow	H ~ 4 G (6.57)	---
	1.98	<u>2.04</u>	2.10	.12	FIEDLERITE Pb ₃ Cl ₄ (OH) ₂	Large r < v	Z = $\frac{b}{c}$ Y:c = -34°	MCL tab 100	100 good	Cols	H 3.5 G 5.88	Diss by HNO ₃ . Tw on 100 common.
2.01 ^	1.95	<u>2.04</u>	2.06	.11	LORENZENITE Na ₂ Ti ₂ Si ₂ O ₉	37-41° r > v	X = $\frac{a}{b}$ Y = $\frac{b}{a}$	ORTH	100 perf 110 good	Brown to black	H 6 G 3.43 F easy	Insol in acids. Pleoc, X and Y pale reddish-yellow, Z pale yellow.
2.129 v	2.01	<u>2.04</u>	2.04	.07	VESIGNIEITE BaCu ₃ (VO ₄) ₂ (OH) ₂	60°	Ext:tw pl = 10°	MCL ps hex	001 good	Yellow-to olive-green	H 3.5 G 4.05	Poly tw.
2.01 □ 2.05	2.01	<u>2.04</u>	2.07	.06	VOLBORTHITE Cu ₃ (VO ₄) ₂ ·3H ₂ O	Large r < v str	---	MCL	001 perf	Yellow to dark green	H 3.8 G 3.42	Tw. Diss by acids. Pleoc, X and Y yellow, Z yellow-green.
	(1.91)	<u>2.05</u>	2.15	(.24)	SCHMITTERITE (UO ₂)TeO ₃	75°	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	ORTH	100	Light yellow	H 1 G 6.9	---
	1.908	<u>2.05</u>	2.065	.157	PINAKIOLITE (Mg,Mn ⁺²) ₂ Mn ⁺³ BO ₅	32°	X = $\frac{b}{a}$ Z = $\frac{a}{b}$ el cTv pos	MCL	010 perf	Black	H 6 G 3.88 F 5	Diss by HCl. Pleoc in red-browns.
	---	<u>2.05</u>	---	str	FERNANDINITE CaV ⁺⁴ ₂ V ⁺⁵ ₁₀ O ₃₀ ·14H ₂ O	---	---	Crypto-cryst	---	Dull green	---	Diss by acids. Not pleoc. Opt char unk.
2.061 ┐	2.042	<u>2.050</u>	2.050	.008	PYROMORPHITE (Apatite grp) Pb ₅ (PO ₄) ₃ Cl	~ 0°	X = $\frac{c}{a}$	HEX pris	1010	Green, brown, yellow	H 4 G 7.05 F 1.5	Diss by HNO ₃ . Pleoc, X greenish-yellow, Z green.
	1.96	---	>2.10	>.14	KARIBIBITE Fe ⁺³ ₂ As ⁺³ ₄ (O,OH) ₉	Large	---	ORTH fib	---	Brownish-yellow	Soft G 4.07	Pleoc wk, straw-yellow to brownish-yellow.
2.08 ┐	2.03	<u>2.06</u>	2.08	.05	DUFTITE PbCuAsO ₄ (OH)	Large r > v	---	ORTH	X = $\frac{c}{b}$ Y = $\frac{b}{a}$	Canary yellow	G 4.88	---

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
808 1.978 ^	1.786	<u>2.070</u>	2.075	.289	SALESITE $\text{Cu}(\text{IO}_3)(\text{OH})$	0-5° (12+2°) $r > v$ extr	$X = a$ $Y = \frac{b}{5}$ el pos	ORTH	110 perf	Green	H 3 G 4.77	Diss by HNO_3 . Pleoc, X cols, Y light bluish-green, Z bluish-green.
	1.804	<u>2.074</u>	2.076	.272	CERUSSITE (Aragonite grp) PbCO_3	9° $r > v$ str	$X = c$ $Y = \frac{b}{5}$	ORTH	110, 021 good	Cols, white	H 3.5 G 6.55 F easy	Diss in dil HNO_3 with eff. Tw on 110 common.
	1.95	<u>2.08</u>	2.11	.16	CAFETITE $\text{Ca}(\text{Fe,Al})_2\text{Ti}_4\text{O}_{12} \cdot 4\text{H}_2\text{O}$	Med	$Z:c = 3^\circ$ el pos	MCL fib \underline{c}	Pris	Pale yellow to cols	H 4.5 G 3.28 F easy	Insol in acids.
	1.96	<u>2.09</u>	2.10	.14	EMMONSITE $\text{Fe}_2\text{Te}_3\text{O}_9 \cdot 2\text{H}_2\text{O}$	20° $r > v$ str	$Y = \frac{b}{5}$	TCL fib	010 perf	Yellow-green	H 5 G 4.59	Tw. Pleoc, X light green, Z darker green.
	2.05	---	2.09	.04	WOELSENDORFITE $\text{PbU}_2\text{O}_7 \cdot 2\text{H}_2\text{O}$	---	---	ORTH	001 good	Orange-red	G 6.08	PbO 23.1, CaO 1.2%.
	---	<u>~2.09</u>	---	.01	MONTANITE $\text{Bi}_2\text{TeO}_6 \cdot 2\text{H}_2\text{O}$	Small $r < v$ extr	---	MCL (?) fib	---	Yellowish white to grayish-yellow	Soft G 3.79 F 1.5	Diss by acids. Abnormal green interf colors.
	1.982	<u>2.095</u>	2.19	.208	CARLFRIESITE $\text{CaTe}^{+4}_2\text{Te}^{+6}\text{O}_8$	80°	$Y = \frac{b}{5}$ $X:\underline{c} = 28^\circ$	MCL	010 fair	Bright yellow	H 3.5 G 6.3	Diss by hot dil HCl. Pleoc wk in yellows, abs $Z > X = Y$.
	2.01	---	---	---	SANTAFEITE $\text{Na}_2(\text{Mn,Ca,Sr})_6\text{Mn}^{+4}_3$ $(\text{V,As})_6\text{O}_{28} \cdot 8\text{H}_2\text{O}$	---	$X = \underline{c}$	ORTH ro-settes	010 perf 110 dist	Brownish-black	G 3.38 F easy	Diss by HCl. Pleoc, yellow-brown to dark red-brown, abs $X > Y > Z$. Opt char unk.
	1.70 _{Li}	<u>2.10_{Li}</u>	2.23 _{Li}	.53	METAHEWETTITE $\text{CaV}_6\text{O}_{16} \cdot 9\text{H}_2\text{O}$	52°	$Z = \underline{b}$	MCL laths	---	Deep red	Soft G 2.51-2.94	Diss by acids. Pleoc, X light orange-yellow, Y and Z deep red.
	>2.10	<u>>2.10</u>	>2.10	---	NAMIBITE CuBi_2VO_6	Mod for red, small for blue light	$Z = \frac{b}{5}$ $X:a = 12^\circ$	MCL tw	100 good	Dark green	H 4.5-5 G 6.86	Diss by acids. Pleoc, X yellow-green, Y pistachio green, Z dark green, abs $Z > Y > X$.
	>2.09	<u>>2.09</u>	>2.09	---	MOUNANAITE $\text{PbFe}_2(\text{VO}_4)_2(\text{OH})_2$	---	Ext: $c = 38^\circ$ on (010)	TCL el \underline{c}	---	Brownish-red	G 4.85	Tw common. Pleoc, brownish-yellow to brownish-red. Opt char unk.

2.07 ┌	2.00	<u>2.10</u>	2.11	.11	CLARKEITE (Na,Ca,Pb) ₂ U ₂ (O,OH) ₇	40° r < v	---	ORTH(?)	Conch	Dark brown	H 4.5 G 6.3	Pleoc wk in deep orange.
	2.07	<u>2.10</u>	2.12	.05	TRIGONITE Pb ₃ Mn(AsO ₃) ₂ (AsO ₂ OH)	Large r < v	Y = $\frac{b}{c}$ ext: $\frac{c}{a} = 45^\circ$	MCL domatic	010 perf 101 good	Bright yellow to brownish	H 2.5 G 6.5 F 2	Diss by acids.
	2.06	<u>2.11</u>	2.15	.09	CURITE Pb ₂ U ₅ O ₁₇ ·4H ₂ O	Large r > v str	X = $\frac{b}{a}$ Y = $\frac{a}{b}$	ORTH	100 perf	Deep orange- red	H 4.5 G 7.26	Diss by acids. Pleoc, X pale yellow, Y orange, Z red- brown. Not fluor in UV.
	2.08	<u>2.11</u>	---	---	DUHAMELITE Pb ₂ Cu ₄ Bi(VO ₄)(OH) ₃ · 8H ₂ O	---	---	ORTH	---	Yellow- green	H 3 G 5.80	Sign unk. Pleoc wk in yellow. Diss by acids.
1.960 ^	1.915	<u>2.115</u>	2.115	.200	TLAPALLITE H ₆ (Ca,Pb) ₂ (Cu,Zn) ₃ SO ₄ (Te ⁺⁴ O ₃)(Te ⁺⁶ O ₆)	---	---	MCL	---	Green	H 3 G 5.38	PbO 13.6, CaO 4.3, CuO 15.8, ZnO 0.7%.
2.04 ^	2.08	<u>2.12</u>	2.16	.08	LAURIONITE PbCl(OH)	Large	X = $\frac{c}{a}$ Y = $\frac{a}{c}$	ORTH tab	101 perf	Cols, white	H 3.5 G 6.14 F 1	Diss by HNO ₃ .
	2.053	<u>2.129</u>	2.133	.08	VESIGNIEITE BaCu ₃ (VO ₄) ₂ (OH) ₂	---	Ext:tw pl = 10°	MCL	001 good	Yellow, olive	H 3.5 G 4.3-4.6	Poly tw.
└ 2.18	1.95	<u>2.13</u>	2.21	.26	KASSITE CaTi ₂ O ₄ (OH) ₂	58° r > v very str	X = $\frac{c}{b}$ Z = $\frac{b}{c}$	ORTH	010 perf 101 dist	Pale yellow	G 3.42	Pleoc wk, cols to pale yellow.
	2.12	<u>2.13</u>	2.135	.015	MIMETITE (Apatite grp) Pb ₅ (AsO ₄) ₃ Cl	29°	X = $\frac{c}{a}$	MCL ps hex	---	Yellow, green	H 3.5 G 7.1 F 1	Diss by HNO ₃ . Basal section in 6 segments. Tw.
	2.13	<u>2.137</u>	2.142	.012	YTTROCRASITE (Y,Th,Ca,U)(Ti,Fe) ₂ (O,OH) ₆	60-70°	---	ORTH pris	---	Greenish brown	H 5.5-6 G 5.32	Pleoc, X cols to pale brown, Y brown, Z brown to greenish- brown, abs Z > Y > X.
	2.06	<u>2.14</u>	2.17	.11	Unnamed lead oxychloride	Large (?)	Ext:tw = 30° el pos	MCL (?)	010 dist	White	H 3	Poly tw. Am. Mineral., 59, 211 (1974).
	2.12	<u>2.14</u>	~2.15	~.03	MOLYBDOMENITE PbSeO ₃	~ 80°	el pos	MCL	001 perf	Cols to yellowish	H 3.5 G 7.07	Diss by HNO ₃ .
	2.14	---	2.15	.01	EZTLITE Pb ₂ Fe ⁺³ ₆ (Te ⁺⁴ O ₃) ₃ (Te ⁺⁶ O ₆)(OH) ₁₀ ·8H ₂ O	---	Z:c = 3°	---	001	Deep orange	---	Sign unk. Not pleoc.
	---	<u>2.15</u>	---	str	CUPROTUNGSTITE Cu ₂ WO ₄ (OH) ₂	---	---	Crypto- cryst	---	Green	fus	Dec by HCl. Opt char unk.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
v 2.36	2.05	<u>2.15</u>	2.20	.15	PARALAURIONITE PbCl(OH)	Med large r < v str	Y = $\frac{b}{c}$ Z:c = 25°	MCL tab	001 perf	Cols, white	Soft G 6.12 F 1	Diss by HNO ₃ . Tw 100 very common.
	2.01	<u>2.15</u>	2.15	.14	MATLOCKITE PbFCl	~ 0°	X = $\frac{c}{el}$ cTv pos	TET	001 perf	Yellow	H 3 G 7.12 F 1	Diss by HNO ₃ . Anom biax.
	2.13	<u>2.16</u>	2.195	.065	PREISINGERITE Bi ₃ (AsO ₄) ₂ O(OH)	90°	---	TCL tab	---	White to gray	G (7.24)	---
	---	<u>2.16</u>	---	.05	BISMUTITE Bi ₂ (CO ₃)O ₂ +xH ₂ O	Small to med	X = $\frac{c}{el}$ pos	ORTH platy	---	Yellow, dark gray	H 3.5 G 7.0 F 1.5	Diss by acids with eff. Striated 100.
	>2.1	---	<2.2	<0.1	RODALQUILARITE H ₃ Fe ⁺³ ₂ (TeO ₃) ₄ Cl	38°	---	TCL	One good	Emerald- green	H 2-3 G 5.1	Not pleoc.
	2.11	<u>2.18</u>	2.22	.11	BALYAKINITE CuTeO ₃	100° (72+22°)	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	---	Gray- green, bluish- green	G 5.6	Reported as opt pos.
	2.04 1.77 _{Li}	---	2.18 <u>2.18</u> _{Li}	.14(?) .58 _{Li}	HEWETTITE CaV ₆ O ₁₆ ·9H ₂ O	Med (55+4) _{Li}	Z = $\frac{b}{c}$	MCL	---	Deep red	---	Diss by acids. Pleoc, X and Y light orange- yellow, Z dark red.
	2.00 _{Li}	<u>2.18</u> _{Li}	2.35 _{Li}	.35 _{Li}	TELLURITE TeO ₂	~ 90° r > v mod	X = $\frac{b}{c}$ Z = $\frac{c}{el}$ cTv pos	ORTH	010 perf	White	H 2 G 5.90 fus	Diss by HCl. Flexible.
	2.16	<u>2.18</u>	2.18	.02	KLEINITE Hg ₂ N(Cl,SO ₄)·xH ₂ O	Small r < v very str	---	HEX pris	0001 good	Yellow to orange	H 3.5-4 G 8.0 volat	Diss by acids. Uniax pos above 130°C. Poly tw.
	2.13	<u>2.19</u>	2.20	.07	BADDELEYITE ZrO ₂	30° r > v str	Y = $\frac{b}{c}$ X:c = 13°	MCL pris	001 perf 010 good	Cols, brown, black	H 6.5 G 5.74 infus	Insol in acids. Poly tw on 100, 110. Pleoc wk, abs X > Y > Z.
v 2.38	1.94	<u>2.20</u>	2.51	.57	LEPIDOCROCITE FeO(OH)	83°	X = $\frac{b}{c}$ Y = $\frac{c}{a}$	ORTH tab	010 perf 100 good 001 fair	Ruby red	H 5 G 4.09 infus	Diss by HCl. Pleoc, X yellow, Y reddish- orange, Z dark red orange, abs Z > Y > X.
	2.14	---	2.315	.175	VIGEZSITE (Ca,Ce)(Nb,Ta,Ti) ₂ O ₆	Large	X = $\frac{c}{a}$ Z = $\frac{a}{b}$	ORTH prism	100 dist	Orange- yellow	G (5.54)	Sign unk. Not pleoc.

	2.12	<u>2.20</u>	2.23	.11	NATRONIOBITE NaNbO_3	10-30° (61+22°) $r < v$	$X':c = 10-15^\circ$	MCL	---	Yellow to black	H 5.5-6 G 4.40	---
2.27	2.11	<u>2.22</u>	2.22	.11	VAUQUELINITE $\text{Pb}_2\text{Cu}(\text{CrO}_4)(\text{PO}_4)(\text{OH})$	~ 0°	$X = e1$	MCL wedge-shaped	Uneven	Green, brown	H 2.5-3 G 6.06 F 2	Diss by HNO_3 . Pleoc, X pale green, Y and Z pale brown.
2.41	2.15	<u>2.22</u>	2.23	.08	GOETHITE $\text{FeO}(\text{OH})$	Small $r > v$ str	$X = \frac{b}{c}$ $Y = \frac{c}{a}$	ORTH	010 perf 100 good	Yellow-brown, reddish-brown	H 5 G 3.8-4.2 infus	Diss by HCl. Pleoc, X clear yellow, Y brownish-yellow, Z orange-yellow, abs Z > Y > X.
	2.186	<u>2.222</u>	2.224	.038	FERVANITE $\text{Fe}^{+3}_4(\text{VO}_4)_4 \cdot 5\text{H}_2\text{O}$	~ 0°	---	MCL (?) fib <u>c</u>	---	Golden-brown	---	Diss by acids.
	2.10	---	2.30	.20	ROOSEVELTITE BiAsO_4	---	---	MCL	Conch	Gray to yellow	H 4-4.5 G 6.9-7.1 F 2-3	Diss by acids. Opt char unk.
	---	<2.23 >2.18	<2.26 >2.23	---	ONORATOITE $\text{Sb}_8\text{O}_{11}\text{Cl}_2$	---	$X:c = 12^\circ$ $Y:a = 8^\circ$	TCL ps mcl acic	---	White	G 5.3	Diss by HCl.
	>2.11	>2.11	>2.11	---	MOCTEZUMITE $\text{Pb}(\text{UO}_2)(\text{TeO}_3)_2$	5-10°	---	MCL	100 perf	Bright orange	H 3 G 5.73	Diss by HCl or NaOH.
(2.03) ^	2.09	<u>2.24</u>	2.26	.17	TUNGSTITE $\text{WO}_3 \cdot \text{H}_2\text{O}$	27° $r < v$	$X = \frac{c}{b}$ $Y = \frac{b}{a}$	ORTH	001 perf 010 good	Light- to greenish-yellow	H 1.5 G 5.52	Insol in acids, diss by NH_4OH .
	2.16	<u>2.24</u>	2.25	.09	CHLOROXIPHITE $\text{Pb}_3\text{CuCl}_2(\text{OH})_2\text{O}_2$	70° $r > v$ str	$X = \frac{c}{b}$ $Z = \frac{b}{a}$	MCL bladed	101 perf 100 good	Olive-green	H 2.5 G 6.93 F easy	Diss by HNO_3 . Pleoc, X pale yellow-brown, Y brown, Z green.
	2.24	---	2.26	.02	OBOYERITE $\text{Pb}_6\text{H}_6(\text{TeO}_3)_3(\text{TeO}_6)_2 \cdot 2\text{H}_2\text{O}$	---	---	TCL fib	---	White	H 1.5 G 6.46	Opt sign unk.
2.19 □ 2.28	2.24	---	2.26	.02	FERGUSONITE-BETA YNbO_4	Small to 34°	---	MCL ps tet	Conch	Black	H 5.6 G 5.2-5.8 infus	Dec by H_2SO_4 .
	2.17	<u>2.26</u>	2.32	.15	MOTTRAMITE $\text{Pb}(\text{Cu}, \text{Zn})(\text{VO}_4)(\text{OH})$	73° $r > v$ str	$X = \frac{c}{b}$ $Y = \frac{b}{a}$	ORTH	Conch	Brownish-red to black, green	H 3.5 G 5.9 F 1.5	Diss by HNO_3 . Pleoc, X pale yellow, Y and Z deep reddish-brown.
	2.21 _{Li}	<u>2.31</u> _{Li}	2.33 _{Li}	.12 _{Li}								
	2.18	<u>2.27</u>	2.35	.17	DESCLOIZITE $\text{Pb}(\text{Zn}, \text{Cu})(\text{VO}_4)(\text{OH})$	~ 90° $r > v$ str	$X = \frac{c}{b}$ $Y = \frac{b}{a}$	ORTH	Conch	Red, brown, black	H 3.5 G 6.2 F 1.5	Diss by HNO_3 . Pleoc, X yellow, Y and Z deep reddish-brown.
2.30 □	---	<u>2.25</u>	2.40	---	IRANITE $\text{Pb}_{10}\text{Cu}(\text{CrO}_4)_6(\text{SiO}_4)_2(\text{F}, \text{OH})_2$	---	Ext:el = 5°	TCL	---	Saffron-yellow	---	Pleoc, brownish-orange to yellow-orange.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	2V _x (2V _x calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
(2.37) $\overset{v}{\square}$ 2.27 2.35	---	<u>2.29</u>	---	str	FERROCOLUMBITE (Fe,Mn)(Nb,Ta) ₂ O ₆	82°	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	010 perf 100 good	Black	H 6 G 5.3 infus	Insol in acids. Tw pl 201.
	---	<u>2.30</u>	---	up to .025	LUESHITE NaNbO ₃	46-80°	---	ORTH ps cub	---	Reddish-brown	H 5.5 G 4.35-4.52	Insol in acids.
	2.28	<u>2.31</u>	2.34	.06	MACQUARTITE Pb ₃ Cu(CrO ₄)SiO ₃ (OH) ₄ ·2H ₂ O	85°	Y = $\frac{b}{c}$ X:c = +35°	MCL	100 good	Orange	H 3.5 G 5.49	Dec by concd HCl.
	2.10	<u>2.32</u>	2.65	.55	HEMIHEDRITE Pb ₁₀ Zn(CrO ₄) ₆ (SiO ₄) ₂ F ₂	88° horizontal disp	X:110 = 30°	TCL tab	110 poor	Bright orange to brown	H 3-3.5 G 6.42	Diss by HNO ₃ . Tw common 223 by reflection, less common 010.
	2.12	<u>2.32</u>	2.32	.20	SEELIGERITE Pb ₃ (IO ₃)Cl ₃ O	4°	---	ORTH ps tet plates	001 perf 110 good	Bright yellow	G 6.8	---
2.30 pos \wedge	2.27	<u>2.32</u>	2.34	.07	NIOBO-AESCHYNITE (Ce,Ca,Th)(Nb,Ti) ₂ (O,OH) ₆	85°	X = $\frac{a}{b}$ Y = $\frac{b}{c}$	ORTH	Conch	Black	H 5-5.5 G 5.04	---
	2.30 _{Li}	<u>2.34_{Li}</u>	2.36 _{Li}	.06	NADORITE PbSbO ₂ Cl	~ 90°	X = $\frac{b}{c}$ Y = $\frac{c}{a}$ el cTv pos	ORTH tab	010 perf	Brown to yellow	H 3.5-4 G 7.02 F 1.5	Diss by HNO ₃ . Tw pl 101.
	2.31	<u>2.35</u>	2.4	.1	ASHANITE (Nb,Ta,U,Fe,Mn) ₄ O ₈	70-75°	X = $\frac{b}{c}$ Y = $\frac{a}{b}$	ORTH	Conch	Brownish-black	G 6.61	Pleoc, brownish-yellow to reddish-brown, abs Z > X.
	2.18	<u>2.35</u>	2.35	.17	VALENTINITE Sb ₂ O ₃	Very small r < v	X = $\frac{a}{b}$ Y = $\frac{b}{c}$ el pos	ORTH pris	110 perf 010 poor	Cols to yellow	H 2.5 G 5.76 F 1.5	Diss by HCl. Prism faces commonly striated 001.
	2.25 _{Li}	<u>2.35_{Li}</u>	2.36 _{Li}	.11 _{Li}	SCHWARTZEMBERGITE Pb ₆ (IO ₃) ₂ Cl ₄ O ₂ (OH) ₂	Small disp wk	---	ORTH ps tet	001 good	Honey-yellow to reddish-brown	H 2 G 7.39 F 1	Diss by acids. Crystals divided into biax sectors.
	2.20	<u>2.36</u>	2.36	.16	EMBREYITE Pb ₅ (CrO ₄) ₂ (PO ₄) ₂ ·H ₂ O	0-11°	Y = $\frac{b}{c}$	MCL tab	Conch	Orange	H 3.5 G 6.45	Pleoc, X honey-yellow, Y and Z amber.
2.16 \wedge	2.30	<u>2.36</u>	2.38	.08	BISMUTITE Bi ₂ (CO ₃)O ₂	45°	X = $\frac{c}{a}$ el pos	ORTH platy	---	Yellow to dark gray	H 3.5 G 8.11 F 1.5	Diss by acids. Striated 100.

	2.32	<u>2.36</u>	2.37	.05	PYROBELONITE $\text{PbMn}(\text{VO}_4)(\text{OH})$	29° $r > v$	$X = \frac{a}{c}$ $Y = \frac{c}{a}$	ORTH acic	Conch	Bright red	H 3.5 G 6.45	Pleoc in reddish-brown, abs $Y > X$ and Z .
2.29 ^ (2.41)	2.34	(<u>2.37</u>)	2.40	.06	MAGNOCOLUMBITE $(\text{Mg}, \text{Fe}, \text{Mn})(\text{Nb}, \text{Ti})_2\text{O}_6$	80°	$X = \frac{a}{b}$ $Y = \frac{b}{a}$	ORTH	010, 100	Black to brownish- black	H 6 G 5.2 infus	Insol in acids. Pleoc, X brownish-yellow, Z brownish-red.
2.19 ^	2.21	<u>2.38</u>	2.41	.20	BADDELEYITE ZrO_2	31° $r > v$ str	$Y = \frac{b}{c}$ $X:c = 13^\circ$	MCL pris	001 perf 010 good	Cols, brown, black	H 6.5 G 5.62 infus	Insol in acids. Poly tw on 100 and 110. Pleoc wk, abs $X > Y > Z$.
	2.37	(<u>2.38</u>)	2.39	.02	LANDAUITTE $\text{NaMnZn}_2(\text{Ti}, \text{Fe})_6\text{Ti}_{12}\text{O}_{38}$	60° $r > v$ wk	---	MCL	Conch	Black to greenish- brown	H 7.5 G 4.42	Pleoc, X and Z bottle green, Y green, abs $Z > Y > X$.
2.27 □ 2.40	---	<u>2.38</u>	---	wk	PEROVSKITE CaTiO_3	$\sim 90^\circ$ $r > v$	$X = \frac{a}{c}$ $Y = \frac{c}{a}$	ORTH ps cub	100 poor fr sub- conch	Brown to black	H 5.5 G 4.0-4.3 infus	Dec by hot H_2SO_4 . Penet or poly tw on 001.
	2.25	<u>2.382</u>	2.41	.16	CHERVETITE $\text{Pb}_2\text{V}_2\text{O}_7$	50° disp wk inclined	---	MCL	100, 010 (?)	Cols, gray, brown	H < 3 G 6.31	Tw 100.
2.22 ^	2.275	<u>2.41</u>	2.415	.140	GOETHITE $\text{FeO}(\text{OH})$	Small $r > v$ str	$X = \frac{b}{c}$ $Y = \frac{c}{b}$	ORTH pris	010 perf 100 good	Yellow- brown, reddish- brown	H 5.5 G 4.28 infus	Diss by HCl . Pleoc, X clear yellow, Y brown- ish-yellow, Z orange- yellow, abs $Z > Y > X$.
(2.37) ^	2.36	(<u>2.41</u>)	2.44	.08	MANGANOCOLUMBITE $(\text{Mn}, \text{Mg}, \text{Fe})(\text{Nb}, \text{Ta})_2\text{O}_6$	80°	$X = \frac{a}{b}$ $Y = \frac{b}{a}$	ORTH	010 perf 100 good	Reddish- brown to black	H 6 G 5.4 infus	Insol in acids. Tw pl 201. Pleoc, X yellow- brown, Z red-brown. MnO 11.8, FeO 7.9%.
	>2.42	---	>2.42	---	BISMITE Bi_2O_3	Disp str	---	MCL mass	---	Gray, green, yellow	H 4.5 G 8.6	Opt char unk.
	2.44	<u>2.47</u>	2.48	.04	GIRDITE $\text{Pb}_3\text{H}_2(\text{TeO}_3)\text{TeO}_6$	70° $r > v$ str	$Z = \frac{b}{c}$ $X:c = 34^\circ$	MCL spher	---	Cols	H 2 G 5.5 fus	Diss by HNO_3 .
	<1.83	<u>2.49</u>	>2.70	>.87	MARGARITASITE $(\text{Cs}, \text{K}, \text{H}_3\text{O})_2(\text{UO}_2)_2$ $(\text{VO}_4)_2 \cdot \text{H}_2\text{O}$	45.5°	---	MCL	---	Yellow	G 5.41	Data for synth Cs compd.
	2.41 _{Li}	<u>2.50_{Li}</u>	2.51 _{Li}	.10 _{Li}	PUCHERITE BiVO_4	19° $r < v$	$X = \frac{c}{a}$ $Y = \frac{a}{c}$ el cTv pos	ORTH tab	001 perf	Reddish- brown	H 4 G 6.57 F 2	Diss by acids.
	2.45 _{Li}	<u>2.55_{Li}</u>	2.55 _{Li}	.10 _{Li}	HEMATITE, var Turgite (impure) $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$	$\sim 0^\circ$	---	TRIG mass compact	Subconch	Red to brownish- black	H 5.5 G 5 infus	Diff diss by HCl . Pleoc wk in reds.

Table 7. Biaxial negative minerals (continued)

Other entries	Refractive index			Biref	MINERAL NAME and formula	$2V_x$ ($2V_x$ calc) disp	Optical orientation	SYSTEM and habit	CLEAVAGE or fracture	Color	Hardness, specific gravity, and fusibility	Remarks
	α	β	γ									
	2.52 _{Li}	2.61 _{Li}	2.67 _{Li}	.15 _{Li}	KOECHLINITE Bi_2MoO_6	Large $r < v$ str	$X = \frac{c}{a}$ $Y = \frac{a}{b}$ el cTv neg	ORTH plates	010 perf	Greenish-yellow	G 8.3 fus	Diss by HCl. Contact and penet tw 101. Striated 010.
	---	2.63 calc	---	---	CLINOISVANITE BiVO_4	Disp str	---	MCL	010 perf	Orange to yellow	G 6.95 calc	Opt char unk.
	---	2.63 red 3.17 bTue	---	str	TENORITE CuO	---	$Y = \frac{b}{c}$ $Z = \frac{c}{a}$	MCL	In zones perp 011, 011	Steel-gray to black	H 3.5 G 6.45	Diss by HCl. Tw common on 011. Pleoc in browns. Opt char unk.
	2.35 _{Li}	2.64 _{Li}	2.66 _{Li}	.31 _{Li}	TERLINGUAITE Hg_2ClO	20° (Li) $r < v$	---	MCL pris	T01 perf	Sulfur-to greenish-yellow	H 2.5 G 8.73	Volat when heated.
	2.538 2.46 _{Li}	2.684 2.59 _{Li}	2.704 2.61 _{Li}	.166 .15 _{Li}	REALGAR AsS	41° $r < v$	$Y = \frac{b}{c}$ $X:c = -11^\circ$	MCL pris	010 good T01, 100 120 poor	Red to orange-yellow	H 2 G 3.51 F 1	Dec by HNO_3 . Contact tw 100. Striated 001.
	---	>2.72 _{Li}	---	very str	DUFRENOYSITE $\text{Pb}_2\text{As}_2\text{S}_5$	---	---	MCL	010 perf	Lead gray	H 3 G 5.53 fus	Opt char unk.
	---	>2.72	---	very str	POLYBASITE $(\text{Ag,Cu})_{16}\text{Sb}_2\text{S}_{11}$	22°	$X = \frac{c}{a}$ $Y = \frac{a}{b}$	MCL ps hex tab	001 poor fr uneven	Black	H 2.5 G 6.31 F 1	Dec by HNO_3 . Tw pl 110.
	---	>2.72	---	very str	MIARGYRITE AgSbS_2	Med	---	MCL	010 poor	Black	H 2.5 G 5.25 fus	Red in thin splinters.
	2.4 _{Li}	2.81 _{Li}	3.02 _{Li}	.62 _{Li}	ORPIMENT As_2S_3	76° $r > v$ str	$X = \frac{b}{c}$ $Z:c = 2^\circ$ el cTv pos	MCL ps orth orth pris	010 perf sectile	Lemon- to brownish-yellow	H 2 G 3.49 volat	Tw pl 100. Pleoc, Y yellow, Z greenish-yellow.
	---	~3	---	extr	XANTHOCONITE Ag_3AsS_3	34° $r < v$	$X = \frac{a}{b}$ $Z = \frac{b}{c}$ el cTv pos	MCL ps orth tab	100 perf	Red, orange, yellow	H 2.5 G 5.5 F 1	Dec by HNO_3 . Tw 100.
	---	~3	---	extr	LIVINGSTONITE HgSb_4S_8	---	---	MCL acic	100 perf	Lead gray	H 2 G 4.81 F 1	Volat when heated. Pleoc wk in reds.

2.78 _{Li}	3.06 _{Li}	3.07 _{Li}	.29 _{Li}	HUTCHINSONITE
3.08 _{Na}	3.18 _{Na}	3.19 _{Na}	.11 _{Na}	(Pb,Tl) ₂ As ₅ S ₉
---	3.27	---	very str	SMITHITE
				AgAsS ₂
3.19 _{Li}	4.05 _{Li}	4.30 _{Li}	1.11 _{Li}	STIBNITE
				Sb ₂ S ₃

20° Li	X = a	ORTH	010 good	Deep	H 2	Pleoc wk in reds.
37° Na	Y = $\frac{b}{c}$	pris	fr conch	cherry-	G 4.6	
r < v extr				red	F 1	
65°	Y = $\frac{b}{c}$	MCL	100 perf	Pale red	H 2	Dec by HNO ₃ .
	Z:c = 6°				G 4.88	
	el clv pos				F 1	
26° Li	X = c	ORTH	010 perf	Lead gray	H 2	---
(46±2°)	Y = $\frac{a}{b}$	pris	100, 110		G 4.63	
	el cTv neg		poor		F 1	

CHAPTER IV. MINERAL GROUPS (TABLES 8-38)

For these tables, the minerals are arranged in groups according to the chemical and physical characteristics they hold in common, following the terminology of Fleischer (1983). For each group, a brief statement of these common characteristics is given, followed by representative examples in tabular form.

These are arranged in order of increasing refractive index, similar to the arrangement in the identification tables above but omitting in most cases those properties that are the same or nearly the same for all members of the group.

Not included are several groups, notably the chlorite, kaolinite-serpentine, and montmorillonite groups, mainly because of difficulties of nomenclature and lack of reliable criteria for distinguishing between the members under the microscope.

Table 8. Alunite group

The minerals of this group have the general formula $AB_6(SO_4)_4(OH)_{12}$, in which A may be K_2 , Na_2 , $(NH_4)_2$, Ag_2 , $(H_3O)_2$, Ca (the hydronium ion), and Pb; B may be Al and Fe^{+3} . In beaverite and osarizawaite, A = Pb, B = Cu_2Al_4 or $Cu_2(Fe^{+3},Al)_4$.

Alunite	$KAl_3(SO_4)_2(OH)_6$
Alunite	$(K,Na)Al_3(SO_4)_2(OH)_6$
Ammoniojarosite	$(NH_4)Fe_3(SO_4)_2(OH)_6$
Argentojarosite	$AgFe_3(SO_4)_2(OH)_6$
Beaverite	$PbCu(Fe,Al)_2(SO_4)_2(OH)_6$
Hydronium Jarosite	$(H_3O)Fe_3(SO_4)_2(OH)_6$
Jarosite	$KFe_3(SO_4)_2(OH)_6$
Natroalunite	$NaAl_3(SO_4)_2(OH)_6$
Natroalunite	$(Na,K)Al_3(SO_4)_2(OH)_6$
Natrojarosite	$NaFe_3(SO_4)_2(OH)_6$
Osarizawaite	$PbCuAl_2(SO_4)_2(OH)_6$
Plumbojarosite	$PbFe_6(SO_4)_4(OH)_{12}$

These minerals are trigonal (pseudo-cubic), space group $R\bar{3}m$ (except plumbojarosite, $R\bar{3}m$). Distinct to perfect basal cleavage is characteristic, with hardness between 2 and 4. Optically, they are uniaxial, although some samples of jarosite are anomalously biaxial. The minerals of this group are closely related to those of the beudantite and plumbogummite groups.

Table 8. Alunite group (continued)

Refractive index		Biref	NAME	G
ω	ϵ			
1.572	1.592	(+).020	ALUNITE	2.6
1.574	1.590	(+).016	NATROALUNITE	2.78
1.583	1.595	(+).012	ALUNITE	2.7
1.603	1.611	(+).008	NATROALUNITE	2.8
1.714	1.731	(+).017	OSARIZAWAITE	4.04
1.791	1.705	(-).086	JAROSITE	2.9-3.2
1.800	1.750	(-).050	AMMONIOJAROSITE	3.11
1.815	1.740	(-).075	NATROJAROSITE	3.18
1.816	1.728	(-).088	HYDRONIUM JAROSITE	3.1-3.2 calc.
1.820	1.715	(-).105	JAROSITE	3.2
1.832	1.750	(-).082	NATROJAROSITE	3.18
1.85	(1.81)	(-).04	BEAVERITE	4.08-4.36
1.875	1.785	(-).090	PLUMBOJAROSITE	3.63
1.882	1.785	(-).097	ARGENTOJAROSITE	3.66

Table 9. Amphibole group

The amphiboles form a complex group of double-chain silicates having the general formula $A_{0-1}B_2C_5T_8O_{22}(OH,F,Cl)_2$, in which A may be Ca, Na, and K; B may be Fe²⁺, Mg, Mn, and Li; C may be Al, Cr, Fe²⁺, Fe³⁺, Mg, and Mn; T may be mainly Si and Al but may have Cr, Fe³⁺, and Ti. The nomenclature used here generally follows that recommended by the International Mineralogical Association (Leake, 1978).

A. Fe-Mg-Mn AMPHIBOLES		Mg
1. Orthorhombic forms		Mg + Fe ²⁺
Anthophyllite	} (Mg, Fe ²⁺) ₇ Si ₈ O ₂₂ (OH) ₂	0.1-0.9
Magnesio-anthophyllite		0.9-1.0
Gedrite		0.1-0.9
Magnesio-gedrite	} (Mg, Fe ²⁺) ₅ Al ₂ Si ₆ Al ₂ O ₂₂ (OH) ₂	{
Ferro-gedrite		0.9-1.0
Holmquistite	Li ₂ (Mg, Fe ²⁺) ₃ Al ₂ Si ₈ O ₂₂ (OH, F) ₂	0.1-0.9
2. Monoclinic forms		
Cummingtonite	} (Mg, Fe ²⁺) ₇ Si ₈ O ₂₂ (OH) ₂	0.3-0.7
Grunerite		0.0-0.3
Tiroadite	} Mn ²⁺ (Mg, Fe ²⁺) ₅ Si ₈ O ₂₂ (OH) ₂	0.5-1.0
Dannemorite		0.0-0.5
Clinoholmquistite	Li ₂ (Mg, Fe ²⁺) ₃ Al ₂ Si ₈ O ₂₂ (OH) ₂	0.1-0.9
B. CALCIC AMPHIBOLES		Mg + Fe ²⁺
Tremolite		0.9-1.0
Actinolite	} Ca ₂ (Mg, Fe ²⁺) ₅ Si ₈ O ₂₂ (OH) ₂	{ 0.5-0.9
Ferro-actinolite		0.0-0.5
Edenite	NaCa ₂ (Mg, Fe ²⁺) ₅ Si ₇ AlO ₂₂ (OH) ₂	0.5-1.0

Table 9. Amphibole group (continued)

			Mg
B. <u>CALCIC AMPHIBOLES</u> (continued)			
Hastingsite	} NaCa ₂ (Mg, Fe ⁺²) ₄ Fe ⁺³ Si ₆ Al ₂ O ₂₂ (OH) ₂	{	0.0-0.3
Magnesio-hastingsite			0.7-1.0
(Hornblende)			
Magnesio-hornblende	} Ca ₂ (Mg, Fe ⁺²) ₄ AlSi ₇ AlO ₂₂ (OH, F) ₂	{	0.5-1.0
Ferro-hornblende			0.0-0.5
Kaersutite	NaCa ₂ (Mg, Fe ⁺²) ₄ TiSi ₆ Al ₂ O ₂₂ (OH) ₂		0.5-1.0
Pargasite	NaCa ₂ (Mg, Fe ⁺²) ₄ AlSi ₆ Al ₂ O ₂₂ (OH) ₂		0.3-1.0
Tschermakite	} Ca ₂ (Mg, Fe ⁺²) ₃ Al ₂ Si ₆ O ₂₂ (OH) ₂	{	0.5-1.0
Ferrotschermakite			0.0-0.5
Ferro-ferri-tschermakite	Ca ₂ (Fe ⁺² , Mg) ₅ Fe ₂ ⁺³ Si ₆ Al ₂ O ₂₂ (OH) ₂		
C. <u>SODIC-CALCIC AMPHIBOLES</u>			
Richterite	} Na ₂ Ca(Mg, Fe ⁺²) ₅ Si ₈ O ₂₂ (OH) ₂	{	0.5-1.0
Ferro-richterite			0.0-0.5
Taramite	Na ₂ Ca(Fe ⁺² , Mg) ₃ Al ₂ Si ₆ Al ₂ O ₂₂ (OH) ₂		0.0-0.5
D. <u>ALKALI AMPHIBOLES</u>			Mg + Fe ⁺²
Arfvedsonite	} Na ₃ (Mg, Fe ⁺²) ₄ Fe ⁺³ Si ₈ O ₂₂ (OH, F) ₂	{	0.0-0.5
Magnesio-arfvedsonite			0.5-1.0

Table 9. Amphibole group (continued)

			Mg
D. <u>ALKALI AMPHIBOLES</u> (continued)			
Eckermannite	}	$\text{Na}_3(\text{Mg}, \text{Li}, \text{Fe}^{+2})_4(\text{Al}, \text{Fe}^{+3})\text{Si}_8\text{O}_{22}(\text{OH})_2$	0.5-1.0
Kozulite			
<hr/>			
Glaucophane	}	$\text{Na}_2(\text{Mg}, \text{Fe}^{+2})_3\text{Al}_2\text{Si}_8\text{O}_{22}(\text{OH})_2$	0.5-1.0
Ferro-glaucophane			0.0-0.5
<hr/>			
Crossite	$\text{Na}_2(\text{Mg}, \text{Fe}^{+2})_3(\text{Al}, \text{Fe}^{+3})_2\text{Si}_8\text{O}_{22}(\text{OH})_2$		
<hr/>			
Riebeckite	}	$\text{Na}_2(\text{Mg}, \text{Fe}^{+2})_3\text{Fe}_2^{+3}\text{Si}_8\text{O}_{22}(\text{OH})_2$	0.0-0.5
Magnesio-riebeckite			0.5-0.1

Members of this group may be orthorhombic, space group Pnma, or monoclinic, space group C2/m (also P2₁/m). They are generally prismatic with perfect cleavage on {210} at 124-126°. The orthorhombic amphiboles have orientation $Y=b$ and $Z=c$; most monoclinic amphiboles have $Y=b$ and $Z:c=4-26^\circ$, thus with positive elongation. Some alkali amphiboles, such as riebeckite, eckermannite, and crossite, have $Z:c > 45^\circ$, therefore with negative elongation. All amphiboles have hardness from 4.5 to 6.5 and are insoluble in all acids except HF. In general, the refractive indices increase with increasing content of iron and titanium, but the substitutions are so complex that it is difficult to deduce chemical composition from refractive indices alone.

Table 9. Amphibole group (continued)

A. Fe-Mg-Mn AMPHIBOLES

1. Orthorhombic forms

321

Refractive index			NAME	2V	Wt. pct. selected oxides			Remarks
α	β	γ			Al ₂ O ₃	Fe ₂ O ₃	FeO	
1.593	<u>1.605</u>	1.613	MAGNESIO-ANTHOPHYLLITE	(-)65°	1.6	0.3	---	Li ₂ O ₃ 3.2%. Pleoc, X and Y cols, Z lilac. Li ₂ O 2.4%. Pleoc, X grayish-yellow, Y lavender, Z blue.
1.613	<u>1.625</u>	1.634	MAGNESIO-GEDRITE	(-)71° (86+12°)	16.1	0.3	1.5	
1.616	<u>1.628</u>	1.641	ANTHOPHYLLITE	(+)79° (88+9°)	1.9	---	11.1	
1.616	<u>1.630</u>	1.641	ANTHOPHYLLITE	(-)88°	0.2	1.0	11.7	
1.616	<u>1.634</u>	1.646	HOLMQUISTITE	(-)56° r > v wk	14.3	1.0	7.3	
1.624	<u>1.645</u>	1.651	HOLMQUISTITE	(-)50° r > v wk	13.0	2.2	8.9	
1.645	<u>1.649</u>	1.661	ANTHOPHYLLITE	(+)59°	1.8	1.8	20.5	
1.648	<u>1.655</u>	1.662	GEDRITE	(+)87°	13.3	1.3	14.6	
1.654	<u>1.660</u>	1.667	ANTHOPHYLLITE	(+)81°	8.1	2.2	18.4	
1.657	<u>1.667</u>	1.678	GEDRITE	(+)87°	17.8	1.0	18.3	

Table 9. Amphibole group (continued)

A. Fe-Mg-Mn AMPHIBOLES (continued)

1. Orthorhombic forms (continued)

Refractive index			NAME	2V	Wt. pct. selected oxides			Remarks
α	β	γ			Al ₂ O ₃	Fe ₂ O ₃	FeO	
1.671	<u>1.681</u>	1.690	FERRO-GEDRITE	(-)75°	18.7	0.9	24.4	MnO 2.3%, Pleoc, X pale green, Y brown-green, Z bluish green.
1.680	<u>1.688</u>	1.700	FERRO-GEDRITE	(+)83°	17.7	3.6	26.2	
1.694	<u>1.710</u>	1.722	FERRO-GEDRITE	(-)82°	19.7	2.3	33.5	

2. Monoclinic forms

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides				Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	
1.610	<u>1.627</u>	1.633	CLINOHOLMQUISTITE	(-)58°	(X:a=16°)	13.5	0.4	5.9	0.5	Pleoc wk, X cols, Y and Z pale green.
1.630	<u>1.640</u>	1.655	CUMMINGTONITE	(+)78° r > v	20°	2.8	1.9	16.8	0.8	
1.630	<u>1.644</u>	1.652	TIRODITE	(+)73°	20°	---	---	5.0	16.6	

Table 9. Amphibole group (continued)

A. Fe-Mg-Mn AMPHIBOLES (continued)

2. Monoclinic forms (continued)

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides				Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	
1.638	<u>1.651</u>	1.665	TIRODITE	(+)89° r > v	19°	---	---	11.1	13.2	Pleoc wk in green.
1.651	<u>1.664</u>	1.678	CUMMINGTONITE	(+)86° r > v	17°	2.4	---	28.0	1.0	Pleoc in green.
1.650	<u>1.665</u>	1.679	GRUNERITE	(-)87° r < v	16°	---	0.6	21.9	8.0	Pleoc, X and Y yellow, Z brownish yellow.
1.660	<u>1.677</u>	1.693	GRUNERITE	(-)large r < v	---	---	3.4	31.2	---	Pleoc, X pale yellow, Z brown.
1.666	<u>1.682</u>	1.698	DANNEMORITE	(-)88° r < v	15°	---	---	22.6	15.6	Pleoc, X red- violet, Y pale violet, Z blue.
1.679	<u>1.700</u>	1.719	GRUNERITE	(-)86° r < v	14°	---	---	45.0	0.4	Pleoc, X and Y cols, Z yellow to brown.

Table 9. Amphibole group (continued)

B. CALCIC AMPHIBOLES

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides						Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	Na ₂ O	
1.581	<u>1.593</u>	1.602	TREMOLITE	(-)86°	21°	---	---	---	---	---	---	Synthetic fluor-tremorite.
1.600	<u>1.615</u>	1.627	TREMOLITE	(-)79° (63+8°) r < v wk	15°	---	---	0.6	---	---	---	
1.613	<u>1.618</u>	1.635	PARGASITE	(+)61°	26°	11.1	0.7	1.7	---	12.5	2.5	K ₂ O 1.2%.
1.622	<u>1.632</u>	1.642	ACTINOLITE	(-)80°	26°	2.2	2.8	7.2	---	---	---	
1.632	<u>1.634</u>	1.651	PARGASITE	(+)64°	17°	13.3	---	4.0	---	10.2	2.6	
1.633	<u>1.642</u>	1.652	MAGNESIO-HORNBLLENDE	(-)88°	16°	5.0	2.7	8.7	---	---	---	Pleoc, X yellow, Y and Z green.
1.629	<u>1.643</u>	1.650	ACTINOLITE	(-)73° r < v	---	---	2.4	0.8	---	---	---	
1.634	<u>1.645</u>	1.658	EDENITE	(-)74° (94+10°)	---	---	3.4	9.0	0.3	---	---	
1.641	<u>1.651</u>	1.664	PARGASITE	(+)82°	18°	15.3	1.1	8.9	---	12.2	---	

Table 9. Amphibole group (continued)

B. CALCIC AMPHIBOLES (continued)

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides						Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	Na ₂ O	
1.642	<u>1.652</u>	1.660	PARGASITE	(-)88°	---	19.3	0.9	6.1	---	---	---	MgO 14.3%.
1.650	<u>1.663</u>	1.670	ACTINOLITE	(-)65°	---	---	3.9	14.9	0.3	---	---	
1.650	<u>1.664</u>	1.671	TSCHERMAKITE	(-)81°	---	---	2.4	9.5	---	---	---	TiO ₂ 2.4%.
1.652	<u>1.664</u>	1.672	MAGNESIO-HASTINGSITE	(-)80°	19°	14.3	2.6	8.7	---	---	---	TiO ₂ 1.3%. Pleoc, X pale brown, Y dark brown, Z greenish-brown.
1.657	<u>1.670</u>	1.679	MAGNESIO-HORNBLENDE	(-)78°	---	---	4.7	13.2	0.3	---	---	MgO 10.6%.
1.660	<u>1.671</u>	1.687	FERROTSCHERMAKITE	(+)78°	18°	11.2	9.3	15.8	---	10.2	1.3	
1.665	<u>1.678</u>	1.684	EDENITE	(-)65°	19°	7.8	6.2	13.4	---	---	---	Pleoc, X yellow, Y pale green, Z dark green.
1.667	<u>1.680</u>	1.687	PARGASITE	(-)74°	---	---	4.9	9.9	---	---	---	MgO 11.6, TiO ₂ 1.9%.
1.670	<u>1.685</u>	1.693	FERRO-HORNBLENDE	(-)61° (72+10°)	18°	13.9	5.4	18.1	---	---	---	Pleoc, X greenish-yellow, Y olive-green, Z bluish-green.

Table 9. Amphibole group (continued)

B. CALCIC AMPHIBOLES (continued)

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides						Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	Na ₂ O	
1.676	<u>1.687</u>	1.695	MAGNESIO-HASTINGSITE	(-)80°	15°	13.9	4.8	6.9	---	---	---	TiO ₂ 4.4%.
1.670	<u>1.692</u>	1.701	KAERSUTITE	(-)81° (64+8°)	9°	14.2	3.3	5.7	---	---	---	TiO ₂ 5.7%. Pleoc, X yellow-brown, Y red-brown, Z dark red- brown.
1.674	<u>1.693</u>	1.700	FERRO-FERRI- TSCHERMAKITE	(-)62°	---	---	5.5	20.4	---	---	---	Ti ₂ 1.8%.
1.688	<u>1.709</u>	1.714	HASTINGSITE	(-)51°	12°	11.5	5.9	22.9	---	---	---	TiO ₂ 1.7%. Pleoc, X greenish- yellow, Y dark olive, Z dark green.
1.687	<u>1.710</u>	1.725	KAERSUTITE	(-)77°	4°	15.5	9.3	4.9	---	---	---	TiO ₂ 5.9%.
1.694	<u>1.730</u>	1.757	KAERSUTITE	(-)80°	small	---	12.4	7.7	---	---	---	TiO ₂ 6.0%.
1.705	<u>1.731</u>	1.732	HASTINGSITE	(-)25°	13°	11.5	11.6	22.0	0.6	---	---	

Table 9. Amphibole group (continued)

C. SODIC-CALCIC AMPHIBOLES

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides					Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	
1.605	<u>1.615</u>	1.622	RICHTERITE	med large r < v wk	25°	---	---	---	---	---	TiO ₂ 3.5%.
1.616	(<u>1.627</u>)	1.632	RICHTERITE	(-)70°	26°	1.7	0.6	0.6	---	---	
1.617	<u>1.631</u>	1.637	RICHTERITE	(-)68°	18°	3.0	3.7	0.6	5.4	---	
1.688	<u>1.699</u>	1.704	FERRO- RICHTERITE	(-)35° (68+15°) r < v	---	---	6.2	27.3	1.1	6.1	Pleoc, X greenish-yellow, Y dark olive, Z dark green.
1.684	<u>1.700</u>	1.703	TARAMITE	(-) (46+15°)	---	15.8	9.6	13.0	---	---	
1.705	<u>1.713</u>	1.715	TARAMITE	(-)54°	11°	10.3	13.1	18.7	1.9	---	

Table 9. Amphibole group (continued)

D. ALKALI AMPHIBOLES

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides						Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	Na ₂ O	
1.624	<u>1.631</u>	1.637	ECKERMANNITE	(-)72°	41°	1.2	8.2	---	1.2	---	---	TiO ₂ 0.3%. Pleoc, X rose, Y lilac, Z pale blue.
1.620	<u>1.637</u>	1.642	GLAUCOPHANE	(-)51° r < v str	5°	10.1	2.0	10.6	---	---	---	Pleoc, X cols, Y lavender, Z blue.
1.638	<u>1.643</u>	1.650	MAGNESIO- ARFEDSONITE	(+)80°	39°	---	---	8.3	6.0	---	---	
1.636	<u>1.644</u>	1.649	ECKERMANNITE	(-)74°	65°	---	9.5	1.9	---	---	---	Pleoc, X and Y bluish-green, yellow-green.
1.634	<u>1.648</u>	1.653	CROSSITE	(-)small (61+13°)	82°	8.4	6.4	7.5	---	---	---	Pleoc, X yellow, Y lavender, Z blue.
1.645	<u>1.656</u>	1.661	MAGNESIO- ARFEDSONITE	(-)57°	56°	---	8.4	13.4	1.5	---	---	MgO 7.8, F 3.3%. Pleoc, X blue-green, Y green, Z pale green.
1.641	<u>1.659</u>	1.662	FERRO- GLAUCOPHANE	(-)42° r < v str	6°	10.1	4.0	17.9	---	---	---	Pleoc, X cols, Y violet, Z blue.
1.654	<u>1.662</u>	1.668	MAGNESIO- RIEBECKITE	(-)med large	81°	0.4	15.9	3.6	0.1	---	4.5	Pleoc, X dark blue, Y indigo blue, Z yellow-green.

Table 9. Amphibole group (continued)

D. ALKALI AMPHIBOLES (continued)

Refractive index			NAME	2V	Z:c	Wt. pct. selected oxides						Remarks
α	β	γ				Al ₂ O ₃	Fe ₂ O ₃	FeO	MnO	CaO	Na ₂ O	
1.667	<u>1.683</u>	1.692	MANGANOAN MAGNESIO- ARFEDSONITE	(-)6° (72+9°)	3°	---	9.1	1.0	7.7	---	9.0	Pleoc, X and Y blue-green, Z yellow-green.
1.694	<u>1.697</u>	1.698	RIEBECKITE	(-)81°	X~c	---	18.3	9.5	6.0	---	---	Pleoc, X and Y dark blue, Z yellow-green.
1.685	<u>1.717</u>	1.720	KOZULITE	(-)35° r > v wk	65°	1.7	2.9	0.0	28.0	1.1	8.4	Pleoc, X yellow- brown, Y reddish brown, Z dark brown.

Table 10. Apatite group

This group includes phosphates, arsenates, and vanadates of general formula $A_5(XO_4)_3(F,Cl,OH)$, in which A may be Ba, Ca, Ce, K, Na, Pb, Sr, and Y; X may be As, C, P, S, Si, and V. Whereas the silicates britholite, britholite-(Y), ellestadite, and hydroxyl-ellestadite are not, strictly speaking, members of the apatite group, they are included here because they intergrade with the phosphates of the apatite group.

Alforsite	$Ba_5(PO_4)_3Cl$
Apatite-britholite	$(Ca,Ce)_5(PO_4,SiO_4)_3F$
Belovite	$(Sr,Ce,Na,Ca)_5(PO_4)_3(OH)$
Britholite	$(Ce,Ca,Y)_5(SiO_4)_3(OH,F)$
Britholite-(Y)	$(Y,Ca,Ce)_5(SiO_4,PO_4)_3(OH,F)$
Carbonate-fluorapatite	$Ca_5(PO_4,CO_3)(F,OH)$
Carbonate-hydroxylapatite	$Ca_5(PO_4,CO_3)_3(OH,F)$
Chlorapatite	$Ca_5(PO_4)_3(Cl,F,OH)$
Ellestadite	$Ca_5[(Si,S,P)O_4]_3(F,OH,Cl)$
Fermorite	$(Ca,Sr)_5(PO_4,AsO_4)(OH,F)$
Fluorapatite	$Ca_5(PO_4)_3F$
Hedyphane	$(Ca,Pb)_5(AsO_4)_3Cl$
Hydroxylapatite	$Ca_5(PO_4)_3(OH,F)$
Hydroxyl-ellestadite	$Ca_5[(Si,S)O_4]_5(OH,Cl,F)$
Johnbaumite	$Ca_5(AsO_4)_3(OH)$
Mimetite	$Pb_5(AsO_4)_3Cl$
Morelandite	$(Ba,Ca,Pb)_5(AsO_4)_3Cl$
Pyromorphite	$Pb_5(PO_4)_5Cl$
Strontium-apatite	$(Sr,Ca,Ba)_5(PO_4)_3(F,OH)$
Svabite	$Ca_5(AsO_4,PO_4)_3(F,OH)$
Vanadinite	$Pb_5(VO_4)_3Cl$

Table 10. Apatite group (continued)

These minerals are hexagonal or monoclinic (pseudo-hexagonal). They are infusible to difficultly fusible, except for the lead members, which fuse easily. All are readily attacked by HNO_3 ; britholite gels and britholite-(Y) is decomposed. Optically they are uniaxial or biaxial with small optic angle. All except vanadinite have low birefringence.

Refractive index			Biref	NAME	2V	G	Remarks
α or ϵ	β or ω	γ or ϵ					
	<u>1.569</u>			CARBONATE- FLUORAPATITE		2.6	Isotropic.
	<u>1.59</u>			CARBONATE- FLUORAPATITE		2.6-2.7	Isotropic.
1.598	<u>1.603</u>		.005	CARBONATE- HYDROXYL- APATITE	(-)0°	2.9	
1.612	<u>1.621</u>	1.622	.010	CARBONATE- FLUORAPATITE	(-)small	3.04	
1.614	<u>1.622</u>		.008	CARBONATE- FLUORAPATITE	(-)0°	3.05	
1.622	<u>1.627</u>	1.627	.005	CARBONATE- FLUORAPATITE	(-)0-25°	3.1-3.2	
	<u>1.63</u>			CARBONATE- FLUORAPATITE		2.7	Isotropic.
1.630	<u>1.633</u>		.003	FLUORAPATITE	(-)0°	3.2	
1.634	<u>1.638</u>		.004	FLUORAPATITE	(-)0°	3.32	
1.633	<u>1.640</u>		.007	CARBONATE- FLUORAPATITE	(-)0°	3.05	
1.638	<u>1.644</u>		.006	FLUORAPATITE	(-)0°	3.26	
1.642	<u>1.649</u>	1.649	.007	CHLORAPATITE	(-)5-10°	3.18	
1.637	<u>1.651</u>		.014	STRONTIUM- APATITE	(-)0°	3.84	
1.644	<u>1.651</u>		.007	HYDROXYL- APATITE	(-)0°	3.21	F 0.16, H ₂ O 1.86%.

Table 10. Apatite group (continued)

Refractive index			Biref	NAME	2V	G	Remarks
α or ϵ	β or ω	γ or ϵ					
1.650	<u>1.654</u>		.004	HYDROXYL-ELLESTADITE	(-)0°	3.02	
1.650	<u>1.655</u>		.005	ELLESTADITE	(-)0°	3.1-3.2	SiO ₂ 17.3, SO ₃ 20.7, P ₂ O ₅ 3.1%.
1.653	<u>1.658</u>		.005	CHLORAPATITE	(-)0°	3.18	
1.640	<u>1.660</u>		.020	BELOVITE	(-)0°	4.18	
	<u>1.66</u>		wk	FERMORITE	(-)0°	3.52	
1.660	<u>1.666</u>		.006	CHLORAPATITE	(-)0°	3.2	
1.665	<u>1.667</u>	1.667	.002	CHLORAPATITE	(-)10°	3.18	
1.672	<u>1.684</u>		.012	SVABITE	(-)0°	3.54	P ₂ O ₅ 12.5, F 1.4%.
1.684	<u>1.687</u>		.003	JOHNBAUMITE	(-)0°	3.68	
1.694	<u>1.696</u>		.002	ALFORSITE	(-)0°	4.80	
1.699	<u>1.703</u>		.004	APATITE-BRITHOLITE	(-)0°	3.83	P ₂ O ₅ 17.3, SiO ₂ 12.9, Ce ₂ O ₃ 12.9%.
	<u>1.72</u>		.000	BRITHOLITE	---	3.85	Isotropic.
	<u>1.728</u>	1.730	.002	BRITHOLITE-(Y)	(+)0°	4.25	
	<u>1.75</u>		.000	BRITHOLITE	---	3.85	Isotropic.
1.748	<u>1.752</u>		.004	BRITHOLITE	(-)0°	4.1-4.4	
1.772	<u>1.775</u>	1.777	.005	BRITHOLITE	(-)0-44°	4.0-4.5	
	<u>1.780</u>	1.783	.003	BRITHOLITE-(Y)	(+)0°	4.35	
	<u>1.81</u>		.000	BRITHOLITE	---	4.1	Isotropic.

Table 10. Apatite group (continued)

Refractive index			Biref	NAME	2V	G	Remarks
α or ϵ	β or ω	γ or ϵ					
	<u>1.880</u>	1.884	.004	MORELANDITE	(+)0°	5.33	Anom biax. Pleoc, 0 green, E greenish yellow.
	<u>1.948</u>	1.958	.010	HEDYPHANE	(+)0°	5.7	
2.010	<u>2.026</u>		.016	HEDYPHANE	(-)0°	5.82	
2.042	<u>2.050</u>	2.050	.008	PYROMORPHITE	(-)small	7.05	
2.046	<u>2.057</u>		.011	PYROMORPHITE	(-)0°	7.0-7.1	
2.12	<u>2.13</u>	2.135	.015	MIMETITE	(-)29°	7.1	Arsenatian var.
2.12	<u>2.135</u>		.01	MIMETITE	(-)0°	7.1-7.24	
2.20	<u>2.25</u>		.05	VANADINITE	(-)0°	7.0	
2.37	<u>2.43</u>		.06	VANADINITE	(-)0°	6.65- 6.98	

Table 11. Aragonite group

Minerals of this group have the general formula RCO_3 , in which R may be Ca, Sr, Ba, and Pb. They are orthorhombic, space group Pmcn , with twinning on {110} very common, giving pseudo-hexagonal aggregates. Hardness is 3 to 4. The minerals are soluble in HCl with effervescence. All are biaxial negative, with very high birefringence and small optic angle.

Refractive index			Biref	NAME and formula	(-)2V	G	Remarks
α	β	γ					
1.520	<u>1.667</u>	1.668	.148	STRONTIANITE SrCO_3	7°	3.68	
1.529	<u>1.676</u>	1.677	.148	WITHERITE BaCO_3	16°	4.29	
1.530	<u>1.680</u>	1.685	.155	ARAGONITE CaCO_3	18°	3.32	
1.525	<u>1.686</u>	1.690	.165	STRONTIANITE $(\text{Sr}, \text{Ba}, \text{Ca})\text{CO}_3$	8°	3.81	BaO 3.3, CaO 1.0%.
1.540	<u>1.695</u>	1.703	.163	ARAGONITE $(\text{Ca}, \text{Pb})\text{CO}_3$	23°	3.05	Pb 5.2%.
1.804	<u>2.074</u>	2.076	.272	CERUSSITE PbCO_3	9°	6.55	

Table 12. Autunite group

The minerals of this group have the general formula $A(UO_2)_2(XO_4)_2 \cdot 8-12H_2O$, in which A may be Ba, Ca, Cu, Fe^{+2} , $1/2 (HA1)$, Mg, Mn, Na_2 , and (UO_2) , and X may be As, P, and V. The water is of zeolitic character, and the minerals are readily transformed reversibly into the respective lower hydrates of the meta-autunite group (which see).

Autunite	$Ca(UO_2)_2(PO_4)_2 \cdot 10-12H_2O$
Heinrichite	$Ba(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$
Kahlerit	$Fe(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$
Novacekite	$Mg(UO_2)_2(AsO_4)_2 \cdot 10-12H_2O$
Sabugalite	$HA1(UO_2)_4(PO_4)_4 \cdot 16H_2O$
Saleeite	$Mg(UO_2)_2(PO_4)_2 \cdot 10H_2O$
Sodium autunite	$Na_2(UO_2)_2(PO_4)_2 \cdot 8H_2O$
Torbernite	$Cu(UO_2)_2(PO_4)_2 \cdot 10-12H_2O$
Troegerite	$(UO_2)(UO_2)_2(AsO_4)_2 \cdot 12H_2O$
Unnamed analogue of Troegerite	
Uranocircite	$Ba(UO_2)(PO_4)_2 \cdot 12H_2O$
Uranospinite	$Ba(UO_2)(AsO_4)_2 \cdot 8H_2O$
Zeunerite	$Cu(UO_2)(AsO_4)_2 \cdot 10-12H_2O$

These minerals are tetragonal, space group $I4/mmm$. Crystals are typically square tabular in form with yellow to green color. Optically the members have negative sign, mostly uniaxial, but often showing anomalous biaxial character. Most show moderate to weak birefringence with pleochroism in shades of yellow or green. All except those containing Fe or Cu fluoresce under ultraviolet light.

Refractive index			Biref	NAME	(-)2V	G
α or ϵ	β or ω	γ				
1.55	<u>1.567</u>	1.572	.022	URANOSPINITE	62°	3.45
1.543	<u>1.570</u>	1.577	.034	NOVACEKITE	40°	3.23
1.559	<u>1.570</u>	1.574	.015	SALEEITE	65°	3.27

Table 12. Alunite group (continued)

Refractive index			Biref	NAME	(-)2V	G
α or ϵ	β or ω	γ				
1.552	<u>1.574</u>	1.577	.022	Unnamed analogue of TROEGERITE	0°	3.76
1.559	<u>1.574</u>		.015	SALEEITE	0°	3.27
1.572	<u>1.575</u>		.003	AUTUNITE	0°	3.2
1.555	<u>1.575</u>		.022	AUTUNITE	53°	3.1-3.2
1.548	<u>1.578</u>	1.585	.030	NOVACEKITE	0°	3.23
1.559	<u>1.578</u>		.019	SODIUM AUTUNITE	0°	3.58
1.564	<u>1.582</u>		.018	SABUGALITE	0°	3.20
1.565	<u>1.582</u>		.020	SALEEITE	65°	3.27
1.564	<u>1.582</u>	1.584	.020	SABUGALITE	35°	3.20
1.574	<u>1.583</u>	1.588	.014	URANOCIRCITE	60-70°	---
1.560 _e	<u>1.586</u>	1.596	.026	URANOSPINITE	0°	3.45
1.578	<u>1.586</u>		.008	AUTUNITE	0°	3.2
1.582	<u>1.592</u>		.010	TORBERNITE	0°	3.22
1.566	<u>1.592</u>		.030	URANOSPINITE	25-35°	3.45
1.581	<u>1.592</u>	1.592	.011	TORBERNITE	small	3.22
1.574	<u>1.605</u>	1.634	.031	HEINRICHITE	0°	---
1.585	<u>1.613</u>		.028	ZEUNERITE	0°	3.4
1.582	<u>1.627</u>		.045	TROEGERITE	0°	3.3
---	<u>1.632</u>		---	KAHLERITE	9-33°	---

Table 13. Barite group

Members of this group are sulfates with the general formula $XS\text{O}_4$, in which X may be Ba, Sr, and Pb. They are orthorhombic, space group Pnma , with perfect cleavage on {001}. Optically they are biaxial positive, with $\underline{X}=\underline{c}$, $\underline{Y}=\underline{b}$, dispersion $\underline{r} < \underline{v}$.

Refractive index			Biref	NAME and formula	(+)2V	G
α	$\underline{\beta}$	γ				
1.622	<u>1.624</u>	1.631	.009	CELESTITE SrSO_4	51°	3.96
1.636	<u>1.637</u>	1.648	.012	BARITE BaSO_4	37°	4.50
1.693	(<u>1.697</u>)	1.718	.025	BARITE (Ba,Pb) SO_4	$\sim 50^\circ$	~ 5
1.877	<u>1.883</u>	1.894	.017	ANGLESITE PbSO_4	$68-75^\circ$	6.38

Table 14. Beudantite group

The minerals of this group have the general formula $\text{AB}_3(\text{SO}_4)(\text{XO}_4)(\text{OH})_6$, in which A may be Ba, Ca, Pb, and Sr; B may be Al and Fe^{+3} ; X may be P and As, with the ratio $\text{SO}_4:\text{XO}_4$ varying somewhat from 1. The minerals of this group are closely related to the sulfates of the alunite group and the phosphates of the plumbo-gummite group.

Beudantite	$\text{PbFe}_3(\text{AsO}_4, \text{SO}_4)_2(\text{OH})_6$
Corkite	$\text{PbFe}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$
Hidalgoite	$\text{PbAl}_3(\text{AsO}_4)(\text{SO}_4)(\text{OH})_6$
Hinsdalite	$\text{PbAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$
Kemmlitzite	$(\text{Sr}, \text{Ce})\text{Al}_3(\text{AsO}_4)(\text{SO}_4, \text{PO}_4)(\text{OH})_6$
Svanbergite	$\text{SrAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$
Weilerite	$\text{BaAl}_3(\text{AsO}_4)(\text{SO}_4)(\text{OH})_6$
Woodhouseite	$\text{CaAl}_3(\text{PO}_4)(\text{SO}_4)(\text{OH})_6$

Table 14. Beudantite group (continued)

They are trigonal, pseudo-cubic, space group $\bar{R}3m$. Distinct to perfect basal cleavage and low birefringence are characteristic, with anomalous biaxial samples not uncommon.

Refractive index			Biref	NAME	2V	G
α or ϵ	β or ω	γ or ϵ				
	<u>1.626</u>	1.640	.014	SVANBERGITE	(+)0°	2.98-3.22
	<u>1.636</u>	1.647	.011	WOODHOUSEITE	(+)0°	3.01
	<u>1.639</u>	1.646	.007	SVANBERGITE	(+)0°	2.98-3.22
	<u>1.662</u>	1.669	.007	WOODHOUSEITE	(+)0°	2.9-3.0
	<u>1.67</u>	1.70	.03	HINS DALITE	(+)0°	3.65
	<u>1.671</u>	1.689	.019	HINS DALITE	(+)0-30°	3.65
	<u>1.688</u>	1.698	.010	WEILERITE	(+)0°	---
	<u>1.688</u>	1.697	.009	HINS DALITE	(+)0°	3.65
	<u>1.701</u>	1.707	.006	KEMMLITZITE	(+)0°	3.63
	<u>1.71</u>	---	.005	HIDALGOITE	(+)0°	3.96
	<u>1.730</u>	1.735	.005	HIDALGOITE	(+)0°	3.95
1.909	<u>1.916</u>		.007	BEUDANTITE	(-)0°	4.0-4.3
---	<u>1.93</u>		wk	CORKITE	(-)0°	4.2-4.3
---	<u>1.93</u>	---	wk	CORKITE	(-)med	4.20
1.943	<u>1.957</u>		.014	BEUDANTITE	(-)0°	4.0-4.3
1.94	<u>1.96</u>	1.96	.02	BEUDANTITE	(-)6-50°	4.1

Table 15. Calcite group

The calcite group minerals have the general formula RCO_3 , in which R may be Ca, Cd, Co, Fe^{+2} , Mg, Mn, Ni, and Zn. Complete solid solution exists between Ca and Cd, Ca and Mn, Mn and Fe, and Mg and Fe, probably also between $(\text{Mg}, \text{Fe}^{+2})$ and (Zn, Co) , although not many representatives are known.

Calcite	CaCO_3
Gaspeite	$(\text{Ni}, \text{Mg})\text{CO}_3$
Magnesite	$(\text{Mg}, \text{Fe})\text{CO}_3$
Otavite	$(\text{Cd}, \text{Ca})\text{CO}_3$
Rhodochrosite	$(\text{Mn}, \text{Ca})\text{CO}_3$
Siderite	$(\text{Fe}, \text{Mg})\text{CO}_3$
Smithsonite	$(\text{Zn}, \text{Mg})\text{CO}_3$
Sphaerocobaltite	CoCO_3

These minerals are trigonal, space group $\bar{R}3c$, with perfect rhombohedral cleavage, hardness 3 to 5. Optically they are uniaxial negative with very high birefringence (compare the related dolomite group).

Refractive index		Biref (-)	NAME	G	Weight %			
ω	ϵ				MnO	FeO	MgO	Other
1.658	1.486	.172	CALCITE	2.71	---	---	---	end member
1.672	1.501	.171	CALCITE	2.82	4.2	2.1	1.3	
1.700	1.509	.191	MAGNESITE	2.96	---	---	47.8	end member
1.710	1.523	.189	CALCITE	---	---	---	---	CdO 20.1
1.711	1.519	.192	MAGNESITE	3.15	---	---	---	NiO 13.55
1.713	1.519	.194	CALCITE	3.02	20.1	---	---	ZnO 0.4
1.726	1.528	.198	MAGNESITE	3.10	---	9.5	---	
1.731	(1.55)	.18	RHODOCHROSITE	3.05	33.4	2.7	2.2	CaO 21.0
1.753	1.560	.193	RHODOCHROSITE	3.3	44.3	0.8	2.3	CaO 13.4

Table 15. Calcite group (continued)

Refractive index		Biref (-)	NAME	G	Weight %			
ω	ϵ				MnO	FeO	MgO	Other
1.788	1.570	.218	MAGNESITE	3.43	---	(35.9)	(20.1)	Mg/Fe = 1
1.803	1.584	.219	RHODOCHROSITE	3.65	---	3.8	0.6	CaO 1.3
1.81	1.58	.23	GASPEITE	3.6	---	4.3	14.8	NiO 36.8
1.815	1.601	.214	SMITHSONITE	---	---	---	7.6	ZnO 53.2 CaO 1.1
1.816	1.592	.224	SIDERITE	3.59	5.9	---	11.8	CaO 1.4
1.816	1.597	.219	RHODOCHROSITE	3.70	61.7	---	---	end member
1.830	1.605	.225	OTAVITE	~4.8	---	---	---	CdO 67.8
1.836	(1.61)	.226	RHODOCHROSITE	3.76	29.8	13.9	---	{ ZnO 14.9 CaO 3.1
1.840	1.615	.225	RHODOCHROSITE	3.72	35.4	26.1	---	
1.84	(1.62)	.22	GASPEITE	3.7	---	0.1	3.3	NiO 55.5
1.849	1.615	.234	SIDERITE	3.81	9.5	47.9	3.2	
1.850	1.623	.227	SMITHSONITE	4.42	0.5	0.4	---	
1.855	1.60	.225	SPHAERO-COBALTITE	4.1	---	---	---	
1.860	1.625	.235	SIDERITE	3.92	14.7	36.4	0.3	ZnO 11.55
1.875	1.633	.242	SIDERITE	3.89	---	62.0	---	end member

Table 16. Cancrinite group

The cancrinite group consists of hexagonal silicates of Ca, Na, and K, also containing carbonate (cancrinite), sulfate (vishnevite), or chloride. Davyne, microsommite, and afghanite are polytypes with nearly the same composition but with different unit cells. They are characterized by low to moderate birefringence and perfect prismatic cleavage.

Afghanite	$(\text{Na,Ca,K})_8(\text{Si,Al})_{12}\text{O}_{24}(\text{Cl},\text{SO}_4,\text{CO}_3)_3\cdot\text{H}_2\text{O}$
Cancrinite	$(\text{Na,Ca})_{7-8}(\text{Si}_6\text{Al}_6)\text{O}_{24}(\text{CO}_3,\text{SO}_4)_{1-2}\cdot 2\text{H}_2\text{O}$
Davyne	$(\text{Na,Ca,K})_{7-8}(\text{Si,Al})_{12}\text{O}_{24}(\text{Cl},\text{SO}_4,\text{CO}_3)_{2-3}$
Franzinite	$(\text{Na,Ca})_7(\text{Si,Al})_{12}\text{O}_{24}(\text{SO}_4,\text{CO}_3)_3\cdot\text{H}_2\text{O}$
Giuseppettite	$(\text{Na,K,Ca})_{7-8}(\text{Si,Al})_{12}\text{O}_{24}(\text{SO}_4,\text{Cl})_{1-2}$
Liottite	$(\text{Ca,Na,K})_8(\text{Si,Al})_{12}\text{O}_{24}(\text{SO}_4,\text{CO}_3,\text{OH},\text{Cl})_4\cdot\text{H}_2\text{O}$
Microsommite	$(\text{Na,Ca,K})_{7-8}(\text{Si,Al})_{12}\text{O}_{24}(\text{Cl},\text{SO}_4,\text{CO}_3)_{2-3}$
Sacrofanite	$(\text{Na,Ca,K})_9(\text{Si,Al})_{12}\text{O}_{24}[(\text{OH})_2,\text{SO}_4,\text{CO}_3,\text{Cl}_2]_3\cdot x\text{H}_2\text{O}$
Vishnevite	$(\text{Na,Ca,K})_{6-7}\text{Si}_6\text{Al}_6\text{O}_{24}(\text{SO}_4,\text{CO}_3)_{1-2}\cdot\text{H}_2\text{O}$

Refractive index		Biref	NAME	Weight %					
ω	ϵ			SO ₃	CO ₂	Cl	Na ₂ O	K ₂ O	CaO
1.491	1.507	(+).016	GIUSEPPELTITE	---	---	---	---	---	---
1.499	1.493	(-).006	VISHNEVITE	4.7	2.2	0.4	20.8	1.0	1.1
1.505	1.486	(-).019	SACROFANITE	---	---	---	---	---	---
1.510	1.512	(+).002	FRANZINITE	10.7	1.5	0.4	11.5	4.2	12.1
1.515	1.496	(-).019	CANCRINITE	1.4	7.0	0.4	18.7	0.6	4.8
1.518	1.521	(+).003	DAVYNE	---	---	---	---	---	---
1.521	1.529	(+).008	MICROSOMMITE	---	---	---	---	---	---
1.523	1.529	(+).006	AFGHANITE	8.5	0.4	4.6	12.6	2.7	16.5
1.528	1.533	(+).005	AFGHANITE	9.7	0.8	4.5	12.4	0.9	12.3

Table 16. Cancrinite group (continued)

Refractive index		Biref	NAME	Weight %					
ω	ϵ			SO ₃	CO ₂	Cl	Na ₂ O	K ₂ O	CaO
1.528	1.503	(-).025	CANCRINITE	0.2	6.2	0.1	9.8	1.2	11.7
1.530	1.528	(-).002	LIOTTITE	8.7	2.1	2.6	8.0	5.0	16.7

Table 17. Chalcanthite group

The minerals of this group are hydrous sulfates of the general formula $ASO_4 \cdot 5H_2O$, in which A may be Cu, Fe^{+2} , Mg, Mn, and Zn. They are triclinic, optically biaxial negative, with moderate $2V$, $r < v$. Hardness and specific gravity are low. All are soluble in water and on heating lose water but do not fuse.

Refractive index			Biref	NAME and formula	(-)2V	Color	Remarks
α	β	γ					
1.482	<u>1.492</u>	1.493	.011	PENTAHYDRITE $MgSO_4 \cdot 5H_2O$	45°	Col's	CuO 9.0, ZnO 5.6, FeO 1.4%.
1.498	<u>1.510</u>	1.517	.019	JOKOKUITE $MnSO_4 \cdot 5H_2O$	70-80°	Pale pink	
1.495	<u>1.512</u>	1.518	.023	PENTAHYDRITE $(Mg, Cu, Zn)SO_4 \cdot 5H_2O$	55°	Blue	
1.513	(<u>1.526</u>)	1.534	.021	SIDEROTIL $(Fe, Cu)SO_4 \cdot 5H_2O$	60°	Pale green to blue	
1.501	<u>1.537</u>	1.539	.038	CHALCANTHITE $CuSO_4 \cdot 5H_2O$	56°	Dark blue	

Table 18. Copiapite group

Minerals of this group are sulfates having the general formula $AFe_4^{+3}(SO_4)_6 \cdot 20H_2O$ or $BFe_4^{+3}(SO_4)_6O(OH) \cdot 20H_2O$, in which A may be Ca, Cu, Fe^{+2} , Mg, and Zn; B may be Al and Fe^{+3} . All are triclinic with perfect {010} cleavage, low hardness and density. They are optically biaxial positive with moderate 2V, $r > v$ strong, X near b, and pleochroic in yellows or greens. Optical data are not available for calcio-copiapite. All are dissolved by acids.

Refractive index			Biref	NAME and formula	(+)2V	Pleochroism
α	β	γ				
1.507	<u>1.529</u>	1.573	.067	MAGNESIOCOPIAPITE $MgFe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	73°	X yellow-green, Y yellow, Z yellow to yellow-green.
1.520	<u>1.535</u>	1.578	.058	COPIAPITE $Fe^{+2}Fe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	70°	X, Y pale yellow, Z yellow.
1.525	<u>1.540</u>	1.590	.065	ALUMINOCOPIAPITE $AlFe_4^{+3}(SO_4)_6O(OH) \cdot 20H_2O$	med	Yellows.
1.525	<u>1.545</u>	1.595	.070	MAGNESIOCOPIAPITE $(Mg, Fe^{+2})Fe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	66°	X, Y very pale yellow, Z deep yellow.
1.530	<u>1.550</u>	1.592	.062	FERRICOPIAPITE $Fe^{+3}Fe_4^{+3}(SO_4)_6O(OH) \cdot 20H_2O$	69°	Yellows.
1.534	<u>1.554</u>	1.586	.052	ZINCOCOPIAPITE $ZnFe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	(78+5°)	Strong in shades of yellow.
1.558	<u>1.575</u>	1.620	.062	CUPROCOPIAPITE $CuFe_4^{+3}(SO_4)_6(OH)_2 \cdot 20H_2O$	63°	Greens, abs X and Z > Y.

Table 19. Crandallite (Plumbogummite) group

The minerals of this group have the general formula $AB_3(XO_4)_2(OH)_5 \cdot H_2O$ or $AB_3(XO_4)_2(OH)_6$, in which A may be Ba, Ca, Ce, Nd, Pb, Sr, and Th; B may be Al and Fe^{+3} ; X may be P and As. Minerals of this group are closely related to those of the alunite and beudantite groups.

Arsenocrandallite	$(Ca,Sr)Al_3[(As,P)O_4]_2(OH)_5 \cdot H_2O$
Crandallite	$CaAl_3(PO_4)_2(OH)_5 \cdot H_2O$
Dussertite	$BaFe_3^{+3}(AsO_4)_2(OH)_5 \cdot H_2O$
Eylattersite	$(Th,Pb)Al_3(PO_4)_2(OH)_6$
Florencite	$CeAl_3(PO_4)_2(OH)_6$
Gorceixite	$(Ba,Ca,Ce)Al_3(PO_4)_2(OH)_5 \cdot H_2O$
Goyazite	$SrAl_3(PO_4)_2(OH)_5 \cdot H_2O$
Lusungite	$(Sr,Pb)Fe_3^{+3}(PO_4)_2(OH)_5 \cdot H_2O$ (?)
Philipsbornite	$PbAl_3(AsO_4)_2(OH)_5 \cdot H_2O$
Plumbogummite	$PbAl_3(PO_4)_2(OH)_5 \cdot H_2O$
Unnamed	$PbFe_3^{+3}(AsO_4)_2(OH)_5 \cdot H_2O$
Zairite	$Bi(Fe^{+3},Al)_3(PO_4)_2(OH)_6$

These minerals are trigonal, pseudocubic, space group $R\bar{3}m$ or $R\bar{3}m$. Distinct to perfect basal cleavage and low birefringence (except for Lusungite) are characteristic. An unnamed triclinic (?) dimorph of crandallite with $\beta=1.607$ (see table 7) is also included in this group. Optical data on waylandite have not been recorded, other than that it is uniaxial negative.

Refractive index		Biref	NAME	H	ϵ	Remarks
ω	ϵ					
1.610	1.620	(+).010	GORCEIXITE	6	3.1-3.3	
1.613	1.622	(+).009	CRANDALLITE	5	2.8	

Table 19. Crandallite (Plumbogummite) group (continued)

Refractive index		Biref	NAME	H	G	Remarks
ω	ϵ					
1.620	1.630	(+).010	GOYAZITE	4.5-5	3.15-3.26	Anom biax. Pleoc, 0 red-brown, E yellow.
1.623	1.634	(+).011	CRANDALLITE	5	2.8	
1.625	---	low	ARSENO-CRANDALLITE	5.5	3.25	
1.625	---	(+)low	GORCEIXITE	6	3.09-3.32	
1.635	---	(-)low	EYLETTERSITE	---	3.4	Fluor wk greenish-yellow in UV.
1.640	1.651	(+).011	GOYAZITE	4.5-5	3.15	Anom biax, 2V 0-20°.
1.653	1.675	(+).022	PLUMBOGUMMITE	4-5	4.0	
1.653	1.661	(+).008	FLORENCITE	5	3.46	
1.678	1.684	(+).006	FLORENCITE	5-6	3.46	
1.680	1.698	(+).018	PLUMBOGUMMITE	4.5	4.0-4.9	
1.695	1.705	(+).010	FLORENCITE	6	3.69	
1.713	1.719	(+).006	FLORENCITE	6	3.4-3.6	
1.77	---	(+).03-.04	LUSUNGITE	---	---	
1.790	---	Isotropic	PHILIPSBORNITE	4.5	4.33	
1.82-1.83	---	---	ZAIRITE	---	4.37	
1.855	---	(+).03-.04	LUSUNGITE	---	---	
1.870	1.845	(-).025	DUSSERTITE	3.5	3.75	
1.99	---	(-)mod	Unnamed	---	---	Anom biax, 2V 0-50°.

Table 20. Dolomite group

The minerals of this group have the general formula $AB(CO_3)_2$, in which A may be Ca and Ba, and B may be Mg, Fe^{+2} , Mn^{+2} , and Zn. Dolomite, Ankerite, and Kutnohorite probably form a complete series of solid solutions.

Ankerite	$CaFe(CO_3)_2$
Dolomite	$CaMg(CO_3)_2$
Kutnohorite	$CaMn(CO_3)_2$
Minrecordite	$CaZn(CO_3)_2$
Norsethite	$BaMg(CO_3)_2$

The minerals are trigonal, space group $R\bar{3}$. All are optically uniaxial negative and, like the members of the calcite group, are characterized by very high birefringence and perfect rhombohedral cleavage. All are infusible and difficultly soluble in dilute acids (compare the related calcite group).

Refractive index		Biref	NAME	G	Weight %		
ω	ϵ				FeO	MnO	Others
1.680	1.501	.179	DOLOMITE	2.865	0.5	---	
1.694	1.510	.184	DOLOMITE	2.92	8.4	1.1	SrO 0.4
1.694	1.519	.175	NORSETHITE	3.74- 3.84	---	---	
1.700	1.521	.179	DOLOMITE	2.95	2.2	---	ZnO 8.3, PbO 1.1
1.710	1.519	.191	KUTNOHORITE	3.00	1.3	17.6	MgO 10.0
1.711	1.520	.191	DOLOMITE	3.01	12.6	1.2	
1.712	1.512	.200	NORSETHITE	3.88	3.6	0.8	
1.734	1.542	.192	MINRECORDITE	3.32			
1.740	1.547	.193	KUTNOHORITE	3.14	0.8	31.8	ZnO 1.4, MgO 0.5
1.741	1.536	.205	ANKERITE	3.12	24.0	1.6	
1.750	1.550	.200	MINRECORDITE	3.45			

Table 21. Epidote group

The minerals of the epidote group are silicates having AlO_6 - and $AlO_4(OH)_2$ -chains linked by independent SiO_4 - and Si_2O_7 -groups. The general formula is $A_2B_3(Si_2O_7)(SiO_4)O(OH)$ or $A_2B_3(Si_3O_{12})(OH)$, in which A may be Ca, Ce, La, Y, Pb, and Sr, and B may be Al, Fe^{+3} , and V.

Allanite	$(Ca,Ce,La,Y)_2(Al,Fe^{+3},Fe^{+2})_3Si_3O_{12}(OH)$
Clinozoisite	$Ca_2Al_3(Si_3O_{12})(OH)$
Epidote	$Ca_2(Al,Fe)_3(Si_3O_{12})(OH)$
Hancockite	$(Pb,Ca,Sr)_2(Al,Fe)_3(Si_3O_{12})(OH)$
Mukhinitite	$Ca_2(Al,V)_3(Si_3O_{12})(OH)$
Piemontite	$Ca_2(Al,Mn^{+3},Fe^{+3})_3(Si_3O_{12})(OH)$
Zoisite	$Ca_2Al_3(Si_3O_{12})(OH)$

All are monoclinic, except orthorhombic zoisite, and all have perfect {001} cleavage, except allanite and zoisite. They are optically biaxial with moderate to large optic angle, most with strong dispersion and $X = \text{or} \sim c$, $Y = b$. Refractive indices and birefringence increase with increase in iron or manganese content. Pleochroism is marked in piemontite, mukhinitite, and in the higher-iron epidotes.

Many allanites are isotropic metamict because of the presence of U and Th. In this state, refractive index and density are highly variable: n 1.61 to 1.75; G 2.96 to 3.74. Data for such metamict varieties are not listed here but are included in the table for isotropic minerals (table 3).

Refractive index			Biref	NAME	2V	Orienta- tion	Remarks
α	β	γ					
1.691	<u>1.692</u>	1.700	.009	ZOISITE	(+)54° $r > v$ str	$X = c$	Abn interf colors. Pleoc u wk, X violet, Y deep blue, Z yellow-green.
1.700	<u>1.703</u>	1.708	.008	ZOISITE	(+)71° $r < v$ or $r > v$	$X = c$	Abn interf colors. Fe_2O_3 3.1, FeO 1.3, MnO 0.3%.
1.700	<u>1.703</u>	1.706	.006	CLINO- ZOISITE	$\sim 90^\circ$ $r < v$ str	$X \sim c$	Abn interf colors. Fe_2O_3 3.2, FeO 0.6%.

Table 21. Epidote group (continued)

Refractive index			Biref	NAME	2V	Orienta- tion	Remarks
α	β	γ					
1.704	<u>1.708</u>	1.714	.006	ALLANITE	(+)70-80° r > v	X:c = 5-40°	U neg. Slowly gel in HCl.
1.706	<u>1.711</u>	1.714	.008	ALLANITE	(-)61°		Slowly gel in HCl. Fe ₂ O ₃ 4.9, FeO 9.1, ThO ₂ 1.7%.
1.715	<u>1.717</u>	1.721	.006	CLINO-ZOISITE	~ 90° r < v str	X ~ c	Abn interf colors. Fe ₂ O ₃ 3.0, FeO 0.8%.
1.713	<u>1.720</u>	1.727	.014	EPIDOTE	(-)89° r > v str		Fe ₂ O ₃ 6.6, FeO 0.4%.
1.723	<u>1.733</u>	1.755	.032	MUKHINITE	(+)88° (69+8°)	Z:a = 32°	Pleoc str, X
1.721	<u>1.736</u>	1.740	.019	EPIDOTE	(-)78° (54+14°) r > v str	X ~ c	Fe ₂ O ₃ 9.5, FeO 0.8%.
1.721	<u>1.738</u>	1.742	.021	ALLANITE	(-)Large (51+13°) r > v str		Gel with acids. Fe ₂ O ₃ 17.0, MnO 0.6, ThO ₂ 0.9%.
1.738	<u>1.744</u>	1.753	.015	ALLANITE	(+)70-80° r > v	X ~ c = 5-40°	Pleoc in brown and red.
1.732	<u>1.750</u>	1.778	.046	PIEMONTITE	(+)Large	X:c = -5°	Pleoc str, X buff, Y deep lavender, Z pink. Fe ₂ O ₃ 3.8, Mn ₂ O ₃ 11.8%.

Table 21. Epidote group (continued)

Refractive index			Biref	NAME	2V	Orientation	Remarks
α	β	γ					
1.733	<u>1.750</u>	1.762	.029	PIEMONTITE	(-)78° r > v str		Pleoc, X yellow, Y pale amethyst, Z deep purplish red. Fe ₂ O ₃ 11.3, Mn ₂ O ₃ 1.0%.
1.729	<u>1.754</u>	1.776	.047	EPIDOTE	(-)73° (85°+5°) r > v str	X:c = 8°	Pleoc, X pale yellow, Y pale greenish yellow, Z greenish yellow. Fe ₂ O ₃ 14.0, FeO 1.2%.
1.740	<u>1.755</u>	1.760	.020	ALLANITE	(-)Large r > v str		Gel with acids. Fe ₂ O ₃ 3.2, FeO 10.7, MnO 1.9, TiO ₂ 1.7, ThO ₂ 0.9%.
1.763	<u>1.768</u>	1.788	.025	ALLANITE	(+)57° r > v	X:c = 5-40°	Slowly gel with HCl. Fe ₂ O ₃ 3.8, FeO 10.4%.
1.740	<u>1.768</u>	1.787	.047	EPIDOTE	(-)74° r > v str	X:c = 13°	Fe ₂ O ₃ 17.9, FeO 0.4, MnO 0.5%.
1.754	<u>1.772</u>	1.795	.041	PIEMONTITE	(+)85° r < v	X:c = -6°	Pleoc str, X yellow, Y violet, Z red.
1.746	<u>1.776</u>	1.793	.047	PIEMONTITE	(-)87° r > v str		Pleoc, X yellow, Y pink, Z deep pink. Fe ₂ O ₃ 11.9, Mn ₂ O ₃ 6.8%.
1.751	<u>1.784</u>	1.797	.046	EPIDOTE	(-)64° r > v str		Fe ₂ O ₃ 17.2, FeO 0.5%.

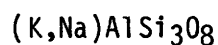
Table 21. Epidote group (continued)

Refractive index			Biref	NAME	2V	Orienta- tion	Remarks
α	β	γ					
1.769	<u>1.784</u>	1.791	.022	ALLANITE	(-)60° r > v str		Slowly gel with HCl.
1.756	<u>1.788</u>	1.829	.073	PIEMONTE	(+)86° r < v	X:c = -9°	Pleoc str, X yellow, Y violet, Z red.
1.788	<u>1.825</u>	1.830	.042	HANCOCKITE	38° r > v	Y = <u>b</u>	Abs Z > X.

Table 22. Feldspar group

Minerals of the feldspar group are framework silicates of the general formula $XAlSi_3O_8$ or $Y(Z,Si)Si_2O_8$, in which X may be K and Na, less commonly NH_4 ; Y may be Ca, less commonly Ba and Sr; Z may be Al, less commonly B and Fe^{+3} . Two subgroups of wide occurrence and importance in the classification of igneous rocks are the alkali feldspars and the plagioclases. The barium, strontium, boron, and NH_4 feldspars occur only infrequently or rarely.

Alkali feldspars



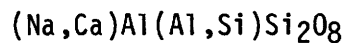
Microcline

Orthoclase

Anorthoclase

Sanidine-High Albite

Plagioclase



Albite

An 0-10

Oligoclase

An 10-30

Andesine

An 30-50

Labradorite

An 50-70

Table 22. Feldspar group (continued)

Bytownite	An 70-90
Anorthite	An 90-100
Barium feldspars	
Banalsite	$\text{BaNa}_2\text{Al}_4\text{Si}_4\text{O}_{16}$
Celsian	$(\text{Ba}, \text{K})\text{Al}(\text{Al}, \text{Si})\text{Si}_2\text{O}_8$
Hyalophane	$(\text{K}, \text{Ba})\text{Al}(\text{Al}, \text{Si})\text{Si}_2\text{O}_8$
Buddingtonite	$(\text{NH}_4)\text{AlSi}_3\text{O}_8$
Reedmergnerite	NaBSi_3O_8
Slawsonite	$(\text{Sr}, \text{Ca})\text{Al}_2\text{Si}_2\text{O}_8$

Minerals of the feldspar group may also be divided on the basis of thermal history. Alkali feldspars that have been rapidly quenched from magmatic temperatures form the sanidine-high albite series. Those that have cooled slowly (annealed) under plutonic conditions or have formed at low temperatures include microcline and orthoclase. Anorthoclases may be an intermediate type. Similarly, the plagioclase series may be divided into quenched types (labeled "volcanic" in tables 6 and 7), annealed types ("plutonic"), and intermediate types. If the aim is to infer thermal history, advantage may be taken of the differences in lattice structure seen in optical orientation or in X-ray. If, on the other hand, the aim is only to infer anorthite content, a convenient approach is to determine the α -refractive index on cleavage flakes (Tsuboi, 1923; Morse, 1968). Useful mineralogical and petrographic discussions of the feldspars are given by Deer, Howie, and Zussman (1963, v. 4), Tröger (1967, pt. 2), and Smith (1974).

The feldspars are monoclinic or triclinic. The space groups are microcline and anorthoclase, C_1 ; orthoclase and sanidine-high albite, C_2/m ; low albite, C_1 ; anorthite, P_1 ; and celsian, I_2/c . They have good to perfect {010} and {001} cleavages, and hardness 5.5 to 6. Simple twins are common, with polysynthetic twinning especially prominent in the plutonic feldspars, less so in the volcanic plagioclases. Except for those high in barium and calcium, the feldspars are essentially unaffected by HCl. All are dissolved by HF.

Table 22. Feldspar group (continued)

Refractive index			Biref	NAME and composition	2V	X':a on		Poly tw	System
α	β	γ				{001}	{010}		
1.519	<u>1.522</u>	1.524	.005	ANORTHOCLASE Or 21, Ab 73, An 6 (?)	(-)51° r > v wk	2°	8°	fine grating	TCL
1.519	<u>1.523</u>	1.525	.006	MICROCLINE Or 85, Ab 14, An 1	(-)76°	10-15°	7.5°	coarse grating	TCL
1.519	<u>1.524</u>	1.525	.006	ORTHOCLASE Or 84, Ab 14, An 2	(-)46°	0°	6°	---	MCL
1.520	<u>1.525</u>	1.525	.005	SANIDINE Or 78, Ab 22	(-)24°	0°	5-8°	---	MCL
1.523	<u>1.529</u>	1.530	.007	SANIDINE Or 41, Ab 58, An 1	(-)33°	0°	---	---	MCL
1.530	<u>1.531</u>	1.534	.004	BUDDINGTONITE (NH ₄)AlSi ₃ O ₈	(+)	0°	4°	---	MCL
1.525	<u>1.531</u>	1.532	.007	ANORTHOCLASE Or 22, Ab 76, An 2	(-)47°	---	---	fine grating	TCL
1.528	<u>1.532</u>	1.539	.011	ALBITE, plutonic plagioclase, An 0	(+)74° r < v wk	3°	20°	{010}	TCL
1.527	<u>1.534</u>	1.535	.008	ALBITE, volcanic plagioclase, An 0	(-)50°	3°	23°	less common	TCL
1.536	<u>1.539</u>	1.541	.005	ANORTHOCLASE Or 19, Ab 63, An 18	(-)62°	---	3°	fine grating	TCL
1.536	<u>1.541</u>	1.546	.010	OLIGOCLASE, plutonic plagioclase, An 18	~ 90°	1°	7°	{010}	TCL

Table 22. Feldspar group (continued)

Refractive index			Biref	NAME and composition	2V	X' :a on		Poly tw	System
α	β	γ				{001}	{010}		
1.538	<u>1.543</u>	1.546	.008	OLIGOCLASE, volcanic plagioclase, An 20	(-)65°	3°	4°	less common	TCL
1.542	<u>1.544</u>	1.546	.004	HYALOPHANE Cn 37	(-)75°	0°	-18°	---	MCL
1.541	<u>1.546</u>	1.550	.009	OLIGOCLASE, plutonic plagioclase, An 25	(-)78°	1°	2°	{010}	TCL
1.546	<u>1.550</u>	1.554	.008	ANDESINE, plutonic plagioclase, An 33	~ 90°	0°	-5°	{010}	TCL
1.549	<u>1.552</u>	1.556	.00	ANDESINE, volcanic plagioclase, An 36	~ 90°	2°	-4°	less common	TCL
1.555	<u>1.558</u>	1.562	.007	LABRADORITE, plutonic plagioclase, An 50	(+)78°	-6°	-17°	{010}	TCL
1.558	<u>1.562</u>	1.566	.008	LABRADORITE, volcanic plagioclase, An 58	(+)76°	-15°	-26°	less common	TCL
1.570	<u>1.571</u>	1.578	.008	BANALSITE	41°				ORTH
1.554	<u>1.565</u>	1.573	.019	REEDMERGNERITE	(-)80°	---	---	---	TCL
1.564	<u>1.568</u>	1.572	.008	CELSIAN, var Kasoite, Cn 35	(-)80°	10-13°	2-3°	untw	MCL

Table 22. Feldspar group (continued)

Refractive index			Biref	NAME and composition	2V	X':a on		Poly tw	System
α	β	γ				{001}	{010}		
1.563	<u>1.568</u>	1.573	.010	BYTOWNITE, plutonic plagioclase, An 70	$\sim 90^\circ$	-16°	-29°	{010}, peri- cline	TCL
1.565	<u>1.569</u>	1.574	.009	BYTOWNITE, volcanic plagioclase, An 75	$\sim 90^\circ$	-23°	-33°	less common	TCL
1.571	<u>1.577</u>	1.583	.012	BYTOWNITE, plutonic plagioclase, An 88	$(-)79^\circ$	-30°	-37°	{010}, peri- cline	TCL
1.572	<u>1.577</u>	1.583	.011	BYTOWNITE, volcanic plagioclase, An 90	$(-)81^\circ$	-34°	-36°	less common	TCL
1.573	<u>1.581</u>	1.585	.012	SLAWSONITE (Sr,Ca)Al ₂ Si ₂ O ₈	$(+)82^\circ$ $r < v$	$Z = \frac{b}{c}$ $X:c = 11^\circ$		---	MCL
1.580	<u>1.583</u>	1.586	.006	CELSIAN, Cn 85	$(-)74^\circ$	$Z:c$ on {100} = 33°		untw	MCL
1.575	<u>1.584</u>	1.589	.014	ANORTHITE, plagioclase, An 100	$(-)77^\circ$	-40°	-39°	{010}, peri- cline	TCL
1.583	<u>1.589</u>	1.593	.010	CELSIAN, Cn 94	$(-)88^\circ$	$Z:c$ on {100} = 26°		untw	MCL
1.593	<u>1.599</u>	1.608	.015	CELSIAN, Cn 94	$(+)85^\circ$	$Y = \frac{b}{c}$		---	MCL

Table 23. Garnet group

The minerals of this group are orthosilicates having the general formula $X_3Y_2(SiO_4)_3$, in which X may be Fe^{+2} , Mn^{+2} , and Mg; Y may be Al, Fe^{+3} , Cr, less commonly V, Ti, Mn^{+3} , and Zr. Substitution is common among the end members listed below, with complete solid solution between some.

Almandine	(Al)	$Fe_3^{+2}Al_2(SiO_4)_3$
Andradite	(An)	$Ca_3Fe_2^{+3}(SiO_4)_3$
Melanite		$Ca_3(Fe^{+3},Ti)_2(SiO_4)_3$
Schorlomite		$Ca_3(Fe^{+3},Ti)_2[(Si,Ti)O_4]_3$
Calderite	(Ca)	$Mn_3Fe_2^{+3}(SiO_4)_3$
Goldmanite	(Go)	$Ca_3V_2(SiO_4)_3$
Grossular	(Gr)	$Ca_3Al_2(SiO_4)_3$
Henritermierite		$Ca_3(Mn^{+3},Al)_2(SiO_4)_2(OH)_4$
Hydrogrossular		$Ca_3Al_2(SiO_4)_{3-x}(OH)_{4x}$
Kimzeyite		$Ca_3(Zr,Ti)_2Al_2SiO_{12}$
Knorringite	(Kn)	$Mg_3Cr_2(SiO_4)_3$
Pyrope	(Py)	$Mg_3Al_2(SiO_4)_3$
Spessartine	(Sp)	$Mn_3Al_2(SiO_4)_3$
Uvarovite	(Uv)	$Ca_3Cr_2(SiO_4)_3$

The garnets are cubic, space group $Ia\bar{3}d$, except for henritermierite, which is tetragonal. Their fracture is subconchoidal, hardness 6 to 7.5, specific gravity 3.35 to 4.32; and, except for hydrogrossular and henritermierite, they are nearly insoluble in HCl and only slowly attacked by cold HF.

Optically garnets have very high refractive index and are usually isotropic, but some, especially the andradites, grossulars, and uvarovites, may show low birefringence with complex grating twinning. Because of the many possible substitutions in the solid solution series, refractive index alone is not a sufficient basis on which to infer chemical composition but is helpful together with density and the length of a_0 of the unit cell. The overlaps in the following tabulation dictate caution in deducing composition from physical properties. If a garnet is known to be in the series Gr-An, however, the composition may be inferred from refractive index, specific gravity, or the length of the cell edge. In the Al-Sp-Py series, refractive index and Mn-content will usually give an approximate determination of the composition.

Relations between composition and physical properties were discussed by Ford (1915), Fleischer (1937), and Winchell (1958), and properties of synthesized

Table 23. Garnet group (continued)

garnets are described by Skinner (1956), whose data are included in the tabulation below. Graphs showing these relations are given by Sriramadas (1957). Types of occurrence are discussed by Wright (1938), and Deer, Howie, and Zussman (1962, v. 1, p. 77-112).

Refractive index	NAME	G	Cell edge a_0 (Å)	Composition					Remarks
				Al	An	Gr	Py	Sp	
1.670	HYDROGROSSULAR	---	---	---	---	---	---	---	H ₂ O 9.3%.
1.702	HYDROGROSSULAR	3.35	---	---	---	---	---	---	H ₂ O 4.6%.
1.714	PYROPE	3.582	11.459	---	---	---	100	---	Synth.
1.734	GROSSULAR	3.594	11.851	---	---	100	---	---	Synth.
1.737	GROSSULAR	---	---	27	---	53	20	---	
1.741	PYROPE- ALMANDINE	3.72	---	20	2	11	66	0.6	
1.752	GROSSULAR- ANDRADITE	3.57	---	---	13	85	1	---	
1.766	PYROPE- ALMANDINE	3.88	11.580	41	6	8	43	1.5	
1.773	GROSSULAR- SPESSARTINE	3.80	---	11	7	56	---	25	
1.782	ALMANDINE- PYROPE	4.02	---	55	---	14	27	3	
1.796	GROSSULAR- UVAROVITE	3.68	---	---	8	52	1	---	Uv 39.
1.797	ALMANDINE	4.10	---	60	---	7	18	15	
1.800	SPESSARTINE	4.190	11.621	---	---	---	---	100	Synth.
1.801	GROSSULAR- ANDRADITE	3.71	11.94	4	42	52	---	2	
1.803	KNORRINGITE	3.85	---	14	3	19	10	1	Kn 53.
1.805	SPESSARTINE	4.20	---	17	---	---	2	80	
1.808	ALMANDINE	4.15	---	74	---	11	11	5	

Table 23. Garnet group (continued)

Refractive index	NAME	G	Cell edge a_0 (Å)	Composition					Remarks
				Al	An	Gr	Py	Sp	
1.814	SPESSARTINE-ALMANDINE	4.23	---	45	---	1	1	53	
1.818	ALMANDINE-SPESSARTINE	4.25	---	74	---	1	5	20	
1.821	GOLDMANITE	3.74	12.011	---	17	20	3	---	Go 60.
1.825	UVAROVITE-GROSSULAR	3.81	---	---	1	26	1	---	Uv 72.
1.827	ANDRADITE-GROSSULAR	3.77	---	57	5	32	3	3	
1.830	ALMANDINE	4.318	11.526	100	---	---	---	---	Synth.
1.834	GOLDMANITE	3.765	---	---	---	---	---	---	Go 100, synth.
1.855	UVAROVITE	3.77	---	1	---	6	2	---	Uv 91.
1.855	GOLDMANITE	3.91	11.974	---	---	---	---	---	V ₂ O ₃ 24.9%.
1.863	ANDRADITE	3.72	11.996	5	79	12	---	4	
1.875	CALDERITE-GROSSULAR	4.05	---	---	---	---	---	---	
1.887	ANDRADITE	3.859	12.048	---	100	---	---	---	Synth.
1.890	CALDERITE-ANDRADITE	4.07	---	---	---	---	---	---	
1.893	ANDRADITE-SPESSARTINE	3.98	---	1	71	---	---	28	
1.94	MELANITE	3.79	---	---	---	---	---	---	TiO ₂ 9.4%.
1.94	KIMZEYITE	3.94	12.46	---	---	---	---	---	TiO ₂ 5.6, ZrO ₂ 29.9%.
1.98	SCHORLOMITE	3.85	---	---	---	---	---	---	TiO ₂ 16.9%.

Table 24. Humite group

Minerals of this group are morphotropic orthosilicates with the general formula $nXSiO_4 \cdot Y(OH,F)_2$ in which n may be 1, 2, 3, or 4; X may be Mg, Fe, and Mn; Y may be Mg and Mn. The site of the Ti, present in some clinohumites, has not been clearly established.

Alleghanyite	$2Mn_2SiO_4 \cdot Mn(OH)_2$
Chondrodite	$2(Mg,Fe)_2SiO_4 \cdot Mg(OH,F)_2$
Clinohumite	$4(Mg,Fe)_2SiO_4 \cdot Mg(OH,F)_2$
Humite	$3(Mg,Fe)_2SiO_4 \cdot Mg(OH,F)_2$
Leucophoenicite	$3Mn_2SiO_4 \cdot Mn(OH)_2$
Manganhumite	$3Mn_2SiO_4 \cdot Mn(OH)_2$
Norbergite	$(Mg,Fe)_2SiO_4 \cdot Mg(OH,F)_2$
Sonolite	$4Mn_2SiO_4 \cdot Mn(OH,F)_2$

The orthorhombic members have space group $Pbnm$; the monoclinic members, $P2_1/c$, commonly with polysynthetic twinning on $\{001\}$. Cleavage of the orthorhombic members is perfect $\{001\}$; that of the monoclinic members is poor or lacking. Hardness is 5 to 6; specific gravity, 3.2 to 4.0. All gelatinize in HCl.

Optically the members of this group are biaxial with moderately strong birefringence and Z parallel to the b crystallographic axis. The Mg-Fe members are optically positive, colorless to yellow; the Mn members (except Manganhumite) are optically negative, pink to reddish brown.

Table 24. Humite group (continued)

Refractive index			Biref	NAME	2V	System	Orientation	Remarks
α	β	γ						
1.548	<u>1.552</u>	1.570	.022	NORBERGITE	(+)33° (51+13°)	ORTH	X = <u>a</u>	Synth Mg-F compd.
1.565	<u>1.570</u>	1.591	.026	NORBERGITE	(+)49°	ORTH	X = <u>a</u>	
1.593	<u>1.603</u>	1.623	.030	CHONDRODITE	(+)71° r > v wk	MCL	X:c = 27°	Poly tw {001}. Pleoc, X yellow, Y and Z nearly cols. FeO 2.8, F 7.9%.
1.598	<u>1.606</u>	1.630	.032	HUMITE	(+)59° r > v wk	ORTH	X = <u>a</u>	Synth Mg-F compd.
1.607	<u>1.617</u>	1.639	.032	CHONDRODITE	(+)80° (69+8°) r > v wk	MCL	X:c = 28°	Poly tw {001}. Pleoc, X yellow, Y and Z pale yellow. FeO 5.0, Fe ₂ O ₃ 0.8%.
1.608	<u>1.618</u>	1.636	.028	CLINOHUMITE	(+)59° (79+8°) r < v	MCL	X:c = 9°	Synth Mg-F compd.
1.607	<u>1.623</u>	1.643	.036	HUMITE	(+)81° r > v wk	ORTH	X = <u>a</u>	Pleoc, X and Z pale yellow, Y nearly cols. FeO 5.4, Fe ₂ O ₃ 0.5, F 5.4%.
1.619	<u>1.632</u>	1.653	.034	CHONDRODITE	(+)80° r > v wk	MCL	X:c = 27°	Pleoc, X yellow, Y and Z pale yellow. FeO 10.5, MnO 1.2, F 5.4%.

Table 24. Humite group (continued)

Refractive index			Biref	NAME	2V	System	Orientation	Remarks
α	β	γ						
1.628	<u>1.638</u>	1.655	.027	HUMITE	(+)76° r > v wk	ORTH	X = <u>a</u>	Pleoc, X and Z pale yellow, Y nearly cols. FeO 7.9, Fe ₂ O ₃ 1.0, MnO 1.7, F 5.0%.
1.632	<u>1.643</u>	1.664	.032	CLINOHUMITE	(+)74° r < v	MCL	X ~ <u>c</u>	FeO 6.6, MnO 1.7, F 3.2%.
1.643	<u>1.653</u>	1.675	.032	HUMITE	(+)68° r > v wk	ORTH	X = <u>a</u>	Pleoc, X and Z golden yellow, Y pale yellow. FeO 7.8, MnO 0.7, Fe ₂ O ₃ 2.7%.
1.668	<u>1.679</u>	1.700	.032	CLINOHUMITE	(+)Med r < v	MCL	X ~ <u>c</u>	FeO 9.9, MnO 0.3, Fe ₂ O ₃ 1.0, F 1.8%.
1.67	<u>1.680</u>	1.703	.033	ALLEGHANYITE	(-)66-86° r > v	MCL	X:a = ~ 30°	
1.707	<u>1.712</u>	1.732	.025	MANGANHUMITE	(+)37° r > v	ORTH	X = <u>a</u>	MnO 57.1, MgO 14.2, FeO 1.0%.
1.707	<u>1.714</u>	1.734	.027	CLINOHUMITE	(+)56° r < v	MCL	X ~ <u>c</u>	FeO 10.2, Fe ₂ O ₃ 1.2, TiO ₂ 5.1, F 0.1%.
1.695	<u>1.716</u>	1.725	.030	SONOLITE	(-)Large	MCL	---	Tw {001} common zincian var.
1.751	<u>1.771</u>	1.782	.031	LEUCOPHOENICITE	(-)74° r > v wk	MCL, ps orth	X \perp clv	Poly tw on {001}.

Table 24. Humite group (continued)

Refractive index			Biref	NAME	2V	System	Orientation	Remarks
α	$\underline{\beta}$	γ						
(1.761)	<u>1.772</u>	1.781	.020	MANGANHUMITE	(+)84°	ORTH	X = <u>a</u>	MnO ~ 69, Fe ~ 1.0, MgO ~ 0.8%.
1.763	<u>1.779</u>	1.793	.030	SONOLITE	(-)78° r > v	MCL, pris	X: <u>c</u> = 9°	Tw {001} common.
1.756	<u>1.780</u>	1.792	.036	ALLEGHANYITE	(-)72° r > v	MCL	Ext to tw plane = 22°	Poly tw on {001} common.

Table 25. Melilite group

This group consists of silicates with the general formula $A_2YZ_2(O,OH,F)_7$, in which A may be Ca and Na; Y may be Mg, Al, Fe, Be, and Zn; and Z may be Si and Al. The mineral melilite is commonly formulated in terms of its end members akermanite (Ak) and gehlenite (Ge), neglecting the presence of Na.

Akermanite	(Ak)	$Ca_2MgSi_2O_7$
Gehlenite	(Ge)	$Ca_2Al(SiAl)O_7$
Hardystonite		$Ca_2ZnSi_2O_7$
Melilite		$(Ca,Na)(Mg,Al,Fe)(Si,Al)_2O_7$
Meliphanite		$(Ca,Na)Be(Si,Al)_2(O,OH,F)_7$

These minerals are tetragonal, space group $P4_2/m$, with distinct basal cleavage and hardness 5 to 6. They are optically uniaxial and, except for meliphanite, have low birefringence, ranging from optically positive for akermanite to negative for gehlenite.

A related mineral, leucophanite, $(Ca,Na)_2BeSi_2(O,OH,F)_7$, is triclinic, pseudo-orthorhombic. It is biaxial negative with $\beta=1.593$ (table 7).

Refractive index		Biref	NAME	G	Remarks
ω	ϵ				
1.611	1.592	(-).019	MELIPHANITE	3.01	
1.629	1.624	(-).005	MELILITE	2.98	Al_2O_3 6.9, Fe_2O_3 1.2, FeO 2.1, MgO 8.2, Na_2O 3.2%.
1.630	1.637	(+).007	MELILITE	2.96-3.05	Al_2O_3 6.9, Fe_2O_3 1.9, FeO 3.1, MgO 8.2, Na_2O 3.2%.
1.632	1.639	(+).007	AKERMANITE	2.961	Synth, Ak 100.
1.648	1.649	(+).001	MELILITE	2.9-3.1	Synth, Ak 60, Ge 40.
1.65	1.65	.00	MELILITE	3.0	Ak 50, Ge 50.
1.653	1.652	(-).001	MELILITE	3.0	Synth, Ak 50, Ge 50.
1.664	1.672	(+).019	MELIPHANITE	3.03	

Table 25. Melilite group (continued)

Refractive index		Biref	NAME	G	Remarks
ω	ϵ				
1.669	1.657	(-).012	HARDYSTONITE	3.4	
1.669	1.658	(-).011	GEHLENITE	3.04	
1.672	1.661	(-).011	MELILITE	3.1	Al ₂ O ₃ 8.4, Fe ₂ O ₃ 8.4, FeO 1.1, MgO 4.7, Na ₂ O 3.8%.

Table 26. Meta-autunite group

The minerals of this group, many of which are lower hydrates of respective members of the autunite group, have the general formula $A(UO_2)_2(XO_4)_2 \cdot 4-8H_2O$, in which A may be K₂, (NH₄)₂, Ca, Ba, Mg, Fe⁺², Co, Cu, and Zn; X may be P and As.

Abernathyite	$K_2(UO_2)_2(AsO_4)_2 \cdot 8H_2O$
Bassetite	$Fe(UO_2)_2(PO_4)_2 \cdot 8H_2O$
Meta-ankoleite	$(K,Ba,Ca)(UO_2)_2(PO_4)_2 \cdot 6H_2O$
Meta-autunite	$Ca(UO_2)_2(PO_4)_2 \cdot 6H_2O$
Metaheinrichite	$Ba(UO_2)_2(AsO_4)_2 \cdot 8H_2O$
Metakahlerite	$Fe(UO_2)_2(AsO_4)_2 \cdot 8H_2O$
Metakirchheimerite	$Co(UO_2)_2(AsO_4)_2 \cdot 8H_2O$
Metalodevite	$Zn(UO_2)_2(AsO_4)_2 \cdot 10H_2O$
Metanovacekite	$Mg(UO_2)_2(AsO_4)_2 \cdot 8H_2O$
Metatorbernite	$Cu(UO_2)_2(PO_4)_2 \cdot 4-8H_2O$
Meta-uranocircite	$Ba(UO_2)_2(PO_4)_2 \cdot 6H_2O$
Metazeunerite	$Cu(UO_2)_2(AsO_4)_2 \cdot 8H_2O$
Sodium Uranospinite	$(Na_2,Ca)(UO_2)_2(AsO_4)_2 \cdot 5H_2O$
Uramphite	$(NH_4)_2(UO_2)_2(PO_4)_2 \cdot 6H_2O$

Table 26. Meta-autunite group (continued)

These minerals are tetragonal, orthorhombic, or monoclinic, with square tabular form and perfect basal cleavage. Optically they are uniaxial or biaxial, mostly negative, with low to medium birefringence, weakly to moderately pleochroic in yellow or green. All except those containing Fe, Cu, or Co fluoresce under UV light.

Refractive index			Biref	NAME	2V	System	G
α or ϵ	β or ω	γ or ϵ					
1.57	<u>1.574</u>	1.580	.006	BASSETITE	(-)62°	MCL	3.4
---	<u>1.581</u>		---	META-ANKOLEITE	(-)0°	TET	3.54
1.564	<u>1.585</u>	1.585	.021	URAMPHITE	(-)3°	ORTH(?)	3.7
1.570	<u>1.597</u>		.027	ABERNATHYITE	(-)0°	TET	3.74
1.598	<u>1.600</u>		.002	META-AUTUNITE	(-)0°	TET	3.5
1.596	<u>1.602</u>	1.603	.007	META-AUTUNITE	(-)15°	TET	3.5
1.603	<u>1.610</u>	1.617	.014	BASSETITE	~ 90°	MCL	3.4
1.594	<u>1.611</u>		.017	META-AUTUNITE	(-)0°	TET	3.5
1.585	<u>1.612</u>		.027	SODIUM URANOSPINITE	(-)0°	TET	3.85
1.618	<u>1.621</u>	1.622	.004	METATORBERNITE	(-)0°	TET	3.6
1.604	<u>1.622</u>	1.630	.026	META-AUTUNITE	(-)15°	TET	3.5
1.603	<u>1.622</u>		.019	META- URANOCIRCITE	(-)0°	TET	4.0
1.610	<u>1.623</u>	1.623	.013	META- URANOCIRCITE	(-)20°	ORTH	4.08
	<u>1.626</u>	1.627	.001	METATORBERNITE	(+)0°	TET	3.6- 3.8
1.628	<u>1.631</u>		.003	METATORBERNITE	(-)0°	TET	3.7- 3.8
1.619	<u>1.635</u>	1.638	.019	METALODEVITE	(-)27-37°	ORTH	(4.0)
1.620	<u>1.637</u>		.017	METANOVACEKITE	(-)0°	TET	3.66

Table 26. Meta-autunite group (continued)

Refractive index			Biref	NAME	2V	System	G
α or ϵ	β or ω	γ or ϵ					
1.609	<u>1.641</u>		.032	METAHEINRICHITE	(-)0°	TET	4.04
1.608	<u>1.642</u>		.034	METAKAHLERITE	(-)0°	TET	---
1.617	<u>1.644</u>		.027	METAKIRCH- HEIMERITE	(-)0°	TET	---
1.624	<u>1.648</u>		.024	METAZEUNERITE	(-)0°	TET	3.67

Table 27. Mica group

The minerals of this group are silicates having the general formula $XY_{2-3}Z_4O_{10}(OH,F)_2$, in which X may be K, Na, Ca, Ba, and H_3O ; Y may be Al, Fe^{+3} , Cr, V^{+3} , Mg, Fe^{+2} , Mn, Zn, and Li; Z may be Si, Al, and Be. The so-called brittle micas and the hydromicas are included here.

Anandite	$(Ba,K)(Fe,Mg)_3(Si,Al,Fe)_4O_{10}(OH)_2$
Annite	$K(Fe^{+2},Fe^{+3},Ti)_3(Si,Al)_4O_{10}(OH,F)_2$
Biotite	$K(Mg,Fe)_3(Si_3Al)O_{10}(OH,F)_2$
Bityite	$CaLiAl_2(AlBeSi_2)O_{10}(OH)_2$
Brammallite	$(Na,H_3O)(Al,Mg)_2(Si,Al)_4O_{10}(OH)_2 \cdot nH_2O$
Celadonite	$(K,Na)(Al,Fe,Mg)_2(Si,Al)_4O_{10}(OH)_2$
Chernykhite	$(Ba,Na)(V,Al)_2(Si,Al)_4O_{10}(OH)_2$
Clintonite	$Ca(Mg,Al)_3(Si,Al)_4O_{10}(OH)_2$
Ephesite	$Na(LiAl_2)(Si_2Al_2)O_{10}(OH,F)_2$
Ferri-annite	$K(Fe^{+2},Mg)_3(Fe^{+3},Al)Si_3O_{10}(OH)_2$
Glaucosite	$(K,Na)(Al,Fe,Mg)_2(Si,Al)_4O_{10}(OH)_2$
Hendricksite	$K(Zn,Mn)_3(Si_3Al)O_{10}(OH,F)_2$
Illite	$(K,H_3O)(Al,Mg,Fe)_2(Si,Al)_4O_{10}(OH)_2 \cdot nH_2O$

Table 27. Mica group (continued)

Kinoshitalite	$(\text{Ba}, \text{K}, \text{Na})(\text{Mg}, \text{Mn}, \text{Al})_3(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH}, \text{F})_2$
Lepidolite	$\text{K}(\text{Li}, \text{Fe}, \text{Al})_3(\text{Si}, \text{Al})_4\text{O}_{10}(\text{F}, \text{OH})_2$
Margarite	$\text{CaAl}_2(\text{Al}_2\text{Si}_2)\text{O}_{10}(\text{OH})_2$
Masutomilite	$\text{K}(\text{Li}, \text{Mn}, \text{Al}, \text{Fe})_3(\text{Si}, \text{Al})_4\text{O}_{10}(\text{F}, \text{OH})_2$
Montdorite	$(\text{K}, \text{Na})_2(\text{Fe}^{+2}, \text{Mn}, \text{Mg})_5(\text{Si}, \text{Al})_8\text{O}_{20}(\text{F}, \text{OH})_4$
Muscovite	$\text{KAl}_2(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH}, \text{F})_2$
Paragonite	$\text{NaAl}_2(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH}, \text{F})_2$
Phlogopite	$\text{K}(\text{Mg}, \text{Fe})_3(\text{Si}_3\text{Al})\text{O}_{10}(\text{F}, \text{OH})_2$
Preiswerkite	$\text{NaMg}_2\text{Al}(\text{Al}_2\text{Si}_2)\text{O}_{10}(\text{OH})_2$
Roscoelite	$\text{K}(\text{V}, \text{Al})_2(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH}, \text{F})_2$
Siderophyllite	$\text{K}(\text{Fe}, \text{Mg})_3(\text{Si}_3\text{Al})\text{O}_{10}(\text{F}, \text{OH})_2$
Sodium phlogopite	$\text{NaMg}_3(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_2$
Taeniolite	$\text{KLiMg}_2\text{Si}_4\text{O}_{10}\text{F}_2$
Tarasovite	$(\text{K}, \text{Na}, \text{H}_3\text{O})\text{Al}_2(\text{Si}, \text{Al})_4\text{O}_{10}(\text{OH})_2 \cdot n\text{H}_2\text{O}$
Tobelite	$(\text{NH}_4, \text{K})\text{Al}_2(\text{Si}_3\text{Al})\text{O}_{10}(\text{OH})_2$
Wonesite	$(\text{Na}, \text{K})(\text{Mg}, \text{Fe})_6(\text{Si}, \text{Al})_8\text{O}_{20}(\text{OH}, \text{F})_4$
Zinnwaldite	$\text{K}(\text{Li}, \text{Fe}^{+2}, \text{Al})_3(\text{Si}_3\text{Al})\text{O}_{10}(\text{F}, \text{OH})_2$

These minerals are monoclinic but show complex polytypic relations. Nearly all are biaxial negative, but optic angle may be near 0° , so that some have been listed also in table 5 (uniaxial negative) as well as table 7. The acute bisectric (X) is nearly normal to the plane of the basal cleavage, which is perfect ("micaeous"). $Y=b$ in most of the group, but $Z=b$ in muscovite, paragonite, and in the brittle micas margarite, ephesite, and clintonite. The birefringence is mostly 0.03-0.04, less for glauconite, celadonite, and the brittle micas, higher for those of high index, the latter also showing greater pleochroism.

Table 27. Mica group (continued)

Refractive index			Biref	NAME	(-)2V	Weight %			
α	β	γ				FeO	MnO	Fe ₂ O ₃	Other
1.522	(1.548)	1.549	.027	PHLOGOPITE, synth F	14°	---	---	---	
1.529	1.548	1.553	.024	LEPIDOLITE	39°	0.1	0.7	0.2	
1.522	1.553	1.553	.031	TAENIOLITE	0-5°	---	---	---	
1.537	1.555	1.559	.022	LEPIDOLITE	38°	.07	.09	1.1	
1.535	1.564	1.565	.030	PHLOGOPITE	10°	---	---	---	
1.542	1.567	1.570	.028	LEPIDOLITE	34°	2.0	2.0	0.8	
1.534	1.568	1.570	.036	MASUTOMILITE	30°	1.5	8.1	0.4	Li ₂ O 4.5
1.540	1.570	1.570	.030	TAENIOLITE	5°	---	---	---	
1.546	1.575	1.579	.033	BRAMMALITE	small	---	---	---	
1.555	1.575	1.581	.026	TOBELITE	28°			0.6	(NH ₄) ₂ O 3.5, K ₂ O 2.3
1.548	1.577	1.579	.031	ZINNWALDITE	30°	10.4	1.7	0.6	
1.544	1.578	1.586	.042	TARASOVITE	23°	---	---	---	
1.552	1.582	1.587	.035	MUSCOVITE	30-47°	---	---	---	
1.555	1.584	1.588	.033	PHLOGOPITE	23-30°	4.2	---	3.7	TiO ₂ 0.4
1.559	1.586	1.586	.027	GLAUCONITE, var Skolite	small	2.6	---	6.4	Al ₂ O ₃ 18.2
1.560	1.587	1.595	.035	TOBELITE	30°			1.0	(NH ₄) ₂ O 3.85, K ₂ O 3.25
1.572	1.587	1.600	.028	ILLITE	small to med	---	---	---	
1.555	1.590	1.590	.035	MUSCOVITE	0-15°	---	---	---	
1.563	1.590	1.594	.031	MUSCOVITE, var Phengite	33°	1.9	---	1.5	MgO 4.0

Table 27. Mica group (continued)

Refractive index			Biref	NAME	(-)2V	Weight %			
α	β	γ				FeO	MnO	Fe ₂ O ₃	Other
1.569	<u>1.595</u>	1.609	.040	MUSCOVITE, var Phengite	40°	3.4	3.9	3.1	MgO 3.1, Cr ₂ O ₃ 3.6
1.558	<u>1.598</u>	1.599	.041	BIOTITE	10-13°	13.9	0.4	2.1	
1.565	<u>1.598</u>	1.600	.035	MUSCOVITE	38°	0.9	---	2.8	
1.585	(~1.60)	1.600	.015	GLAUCONITE	small	1.6	---	14.1	Al ₂ O ₃ 8.9
1.571	<u>1.603</u>	1.606	.035	PARAGONITE	30°	---	---	---	
1.580	<u>1.605</u>	1.605	.025	MONTDORITE	3°	17.1	9.9	---	MgO 4.3, Al ₂ O ₃ 5.0
1.544	<u>1.608</u>	1.611	.040	WONESITE	0-5°	---	---	---	
1.571	<u>1.608</u>	1.611	.040	MUSCOVITE	34°	0.5	---	---	MgO 1.1, Cr ₂ O ₃ 0.2
1.573	<u>1.609</u>	1.615	.042	MUSCOVITE, var Phengite	31°	2.8	---	4.1	MgO 4.1, TiO ₂ 1.0
1.586	<u>1.612</u>	1.613	.027	MARGARITE (Sodium analogue)	50°	---	---	---	
1.560	<u>1.614</u>	1.615	.055	PREISWERKITE	5-7°	---	---	---	
1.575	<u>1.617</u>	1.621	.046	BIOTITE, var Mangano- phyllite	30°	1.2	6.2	5.8	
1.597	<u>1.618</u>	1.619	.020	GLAUCONITE	20°	4.0	---	18.8	Al ₂ O ₃ 8.5, MgO 3.6
1.580	<u>1.620</u>	1.623	.043	MUSCOVITE, var Phengite	35°	2.5	---	8.0	MgO 1.9
1.592	<u>1.624</u>	1.625	.033	EPHESITE	18°	---	---	---	Na ₂ O 7.4, Li ₂ O 3.5
1.606	(<u>1.630</u>)	1.632	.026	CELADONITE	small	2.1	---	12.4	MgO 5.8, Al ₂ O ₃ 6.7
1.60	<u>1.63</u>	1.63	.03	MUSCOVITE, var Mariposite	~ 0°	---	---	---	

Table 27. Mica group (continued)

Refractive index			Biref	NAME	(-)2V	Weight %			
α	β	γ				FeO	MnO	Fe ₂ O ₃	Other
1.619	<u>1.633</u>	1.635	.016	KINOSHITALITE	23°	---	7.4	0.7	BaO 17.8, Mn ₂ O ₃ 3.2
1.59	> <u>1.63</u>	>1.64	>.05	ROSCOELITE	med	1.7	---	---	Al ₂ O ₃ 21.9, V ₂ O ₃ 17.4
1.610	<u>1.634</u>	1.634	.024	GLAUCONITE	10°	2.9	---	24.9	Al ₂ O ₃ 7.3
1.590	<u>1.640</u>	1.640	.050	SIDERO- PHYLLITE	small	30.2	1.0	---	
1.612	<u>1.643</u>	1.643	.031	CELADONITE	small	9.8	---	24.3	MgO 3.7
1.632	<u>1.643</u>	1.646	.014	MARGARITE	63°	---	---	---	
1.595	<u>1.652</u>	1.656	.061	BIOTITE	small	19.9	0.4	3.2	
1.643	<u>1.652</u>	1.654	.011	BITYITE	med	---	---	---	
1.598	<u>1.658</u>	1.660	.062	HENDRICKSITE	8°	---	12.5	4.8	ZnO 19.8
1.648	<u>1.659</u>	1.660	.012	CLINTONITE	5-33°	---	---	---	
1.644	---	1.663	.019	CELADONITE	small	9.2	---	26.9	Al ₂ O ₃ 2.1
1.624	<u>1.672</u>	1.672	.048	ANNITE	5°	32.1	---	3.1	TiO ₂ 3.6
1.610	<u>1.676</u>	1.676	.066	PHLOGOPITE (titanian)	19°	4.7	0.2	6.7	TiO ₂ 4.4
1.610	<u>1.685</u>	1.704	.094	ROSCOELITE	10-15°	---	---	---	
1.624	<u>1.686</u>	1.686	.062	HENDRICKSITE	2-5°	0.7	13.9	4.8	ZnO 21.4
1.640	<u>1.686</u>	1.702	.062	CHERNYKHITE	11-12°	---	---	---	
1.653	(1.691)	1.691	.038	FERRI-ANNITE	0-10°	24.5	---	8	Al ₂ O ₃ 5.5
1.660	(1.720)	1.728	.068	FERRI-ANNITE	40°	29.9	---	13.2	Al ₂ O ₃ 1.4
---	<u>1.855</u>	>1.88	>.025	ANANDITE	(+)?	33.1	0.7	7.0	BaO 20.4

Table 28. Monticellite group

This group, sometimes considered to be a subgroup of the olivine group, includes minerals having the general formula CaXSiO_4 , in which X may be Mg, Fe^{+2} , and Mn. All members are orthorhombic, space group Pmcn . Optically they are biaxial negative, with medium to large optic angle, $r > v$, $X=b$, $Z=a$. All gelatinize with HCl.

Refractive index			Biref	NAME and formula	(-)2V	Weight %		
α	$\underline{\beta}$	γ				FeO	Fe_2O_3	MnO
1.641	<u>1.649</u>	1.655	.014	MONTICELLITE CaMgSiO_4	80°	1.4	---	---
1.654	<u>1.664</u>	1.674	.020	MONTICELLITE $\text{Ca}(\text{Mg},\text{Fe})\text{SiO}_4$	82°	7.6	---	0.2
1.674	<u>1.694</u>	1.706	.032	KIRSCHSTEINITE $\text{Ca}(\text{Fe},\text{Mg})\text{SiO}_4$	65°	19.3	1.4	1.1
1.695	<u>1.716</u>	1.725	.030	GLAUCOCHROITE $\text{Ca}(\text{Mn},\text{Fe})\text{SiO}_4$	61°	4.2	---	32.7
1.689	<u>1.720</u>	1.728	.039	KIRSCHSTEINITE $\text{Ca}(\text{Fe},\text{Mg})\text{SiO}_4$	51°	29.3	0.7	1.6

Table 29. Olivine group

Minerals of this group have the general formula X_2SiO_4 , in which X may be Mg, Fe, Mn, Ni, and rarely Zn. There is a complete solid solution series from Fe to Mg, referred to as the olivine series.

Liebenbergite	$(Ni,Mg,Fe)_2SiO_4$
Olivine series	$(Mg,Fe)_2SiO_4$
Forsterite	Fo 90-100
Chrysolite	Fo 70-90
Hyalosiderite	Fo 50-70
Hortonolite	Fo 30-50
Ferrohortonolite	Fo 10-30
Fayalite	Fo 0-10
Tephroite	$(Mn,Mg)_2SiO_4$

The members are orthorhombic, space group $Pmcn$ (except liebenbergite, possibly $Pbnm$), with distinct {010} cleavage, hardness 6.5 to 7, high specific gravity. Optically they are biaxial, with $X=b$, $Z=a$, moderate to high birefringence, refractive indices increasing with Fe-content (compare the monticellite group).

Refractive index			Biref	NAME	2V	Weight %			
α	β	γ				FeO	Fe ₂ O ₃	MnO	MgO
1.636	<u>1.651</u>	1.669	.033	FORSTERITE (synthetic) Fo 100	(+)85°	---	---	---	57.3
1.653	<u>1.664</u>	1.686	.033	FORSTERITE Fo 93	(+)87°	8.6	0.4	0.2	---
1.656	<u>1.674</u>	1.695	.039	CHRYSLITE Fo 88	(+)89°	10.8	1.7	0.2	46.5
1.661	<u>1.680</u>	1.697	.036	CHRYSLITE Fo 86	~ 90°	---	---	---	---
1.683	<u>1.704</u>	1.722	.039	CHRYSLITE Fo 76	(-)85°	21.6	1.6	0.3	---

Table 29. Olivine group (continued)

Refractive index			Biref	NAME	2V	Weight %			
α	β	γ				FeO	Fe ₂ O ₃	MnO	MgO
1.698	<u>1.720</u>	1.736	.038	HYALOSIDERITE Fo 65	(-)80°	27.8	2.1	0.4	---
1.711	<u>1.727</u>	1.740	.029	TEPHROITE	(-)85°	---	0.3	47.1	18.1
1.710	<u>1.733</u>	1.750	.040	HYALOSIDERITE Fo 63	(-)79°	32.0	2.0	0.5	31.0
1.721	<u>1.750</u>	1.765	.044	HYALOSIDERITE Fo 60	(-)85°	36.8	---	0.5	27.0
1.750	<u>1.766</u>	1.779	.029	TEPHROITE	(-)67°	0.2	---	60.7	6.6
1.742	<u>1.770</u>	1.786	.044	HORTONOLITE Fo 43	(-)69°	---	---	---	---
1.758	<u>1.786</u>	1.804	.046	HORTONOLITE, zincian	(-)77°	35.5	(ZnO 10.7)	16.9	5.6
1.761	<u>1.790</u>	1.806	.045	HORTONOLITE Fo 33	(-)65°	---	---	---	---
1.782	<u>1.802</u>	1.812	.030	TEPHROITE	(-)65°	0.9	---	65.1	2.5
1.777	<u>1.818</u>	1.828	.051	FERRO- HORTONOLITE, manganoan	(-)52°	41.5	0.7	21.4	4.5
1.788	<u>1.828</u>	1.840	.052	FERRO- HORTONOLITE Fo 20	(-)58°	58.6	---	0.8	---
1.803	<u>1.843</u>	1.851	.048	FERRO- HORTONOLITE, manganoan	(-)48°	56.9	1.2	9.7	1.1
1.816	<u>1.850</u>	1.863	.047	FAYALITE Fo 7	(-)60°	---	---	---	---
1.820	<u>1.854</u>	1.888	.068	LIEBENBERGITE	(-)88°	4.4	(NiO 56.3)	(CaO 1.8)	6.5
1.827	<u>1.869</u>	1.879	.052	FAYALITE	(-)48°	65.0	0.4	1.0	---

Table 30. Osumilite group

This group consists of silicates with double rings of $(\text{Si,Al})_{12}\text{O}_{30}$. The general formula is $\text{A}_{1-2}\text{B}_{2-3}\text{C}_3\text{Z}_{12}\text{O}_{30} \cdot n\text{H}_2\text{O}$, in which A may be Ba, Ca, K, and Na; B may be Fe^{+2} , Li, Mg, Mn, and Na; C may be Al, Be, Fe^{+2} , Fe^{+3} , Li, Mg, Sn, Ti, Zn, and Zr; Z may be Al and Si. The minerals are hexagonal, space group P6/mcc , or pseudohexagonal. Optically they are uniaxial or nearly so, with low birefringence.

Brannockite	$\text{KLi}_2\text{Sn}_3\text{Si}_{12}\text{O}_{30}$
Darapiosite	$\text{KNa}_2\text{Li}(\text{Mn,Zn})_2\text{ZrSi}_{12}\text{O}_{30}$
Eifelite	$\text{KNa}_4\text{Mg}_9\text{Si}_{24}\text{O}_{60}$
Emeleusite	$\text{Na}_4\text{Li}_2\text{Fe}_2^{+3}\text{Si}_{12}\text{O}_{30}$
Merrihueite	$(\text{K,Na})_2(\text{Fe}^{+2},\text{Mg})_5\text{Si}_{12}\text{O}_{30}$
Milarite	$\text{KCa}_2\text{AlBe}_2\text{Si}_{12}\text{O}_{30} \cdot 0.5\text{H}_2\text{O}$
Osumilite	$(\text{K,Na})(\text{Fe}^{+2},\text{Mg})_2(\text{Al,Fe})_3(\text{Si,Al})_{12}\text{O}_{30}$
Osumilite-(Mg)	$(\text{K,Na})(\text{Mg,Fe}^{+2})_2(\text{Al,Fe})_3(\text{Si,Al})_{12}\text{O}_{30} \cdot \text{H}_2\text{O}$
Roedderite	$(\text{Na,K})_2(\text{Mg,Fe}^{+2})_5\text{Si}_{12}\text{O}_{30}$
Sogdianite	$(\text{K,Na})\text{Li}_2(\text{Li,Fe}^{+2})_2\text{ZrSi}_{12}\text{O}_{30}$
Sugilite	$(\text{K,Na})(\text{Na,Fe}^{+3})_2(\text{Li}_2\text{Fe}^{+3})\text{Si}_{12}\text{O}_{30}$
Yagiite	$(\text{Na,K})_{1-2}\text{Mg}_2(\text{Al,Mg})_3(\text{Si,Al})_{12}\text{O}_{30}$
(Armenite, $\text{BaCa}_2\text{Al}_6\text{Si}_9\text{O}_{30} \cdot \text{H}_2\text{O}$, may belong to this group.)	

Refractive index		Biref	NAME	Weight %				Remarks
ω	ϵ			MgO	FeO	Na_2O	K_2O	
1.532	1.529	(-).003	MILARITE	---	---	---	---	$2V_Z = 5-8^\circ$.
1.536	1.543	(+).006	ROEDDERITE	19.0	0.4	5.3	3.8	
1.536	1.544	(+).008	YAGIITE	10.5	2.4	3.7	1.4	
1.537	1.542	(+).005	ROEDDERITE	19.5	2.0	4.0	3.3	
1.540	1.546	(+).006	OSUMILITE-Mg	9.5	2.2	0.3	4.0	

Table 30. Osumilite group (continued)

Refractive index		Biref	NAME	Weight %				Remarks
ω	ϵ			MgO	FeO	Na ₂ O	K ₂ O	
1.541	1.547	(+).006	OSUMILITE-Mg	7.5	4.0	---	4.1	2V _z = 5-15°.
1.543	1.545	(+).002	EIFELITE	---	---	---	---	
1.546	1.550	(+).004	OSUMILITE	6.7	10.0	1.0	1.4	Fe ₂ O ₃ 2.0%.
1.553	1.549	(-).004	MILARITE	---	---	---	---	
1.559	---	---	MERRIHUEITE	---	10.5	---	---	
1.567	1.566	(-).001	BRANNOCKITE	---	---	0.7	3.5-4.5	SnO ₂ 28.2, Li ₂ O ₃ 3.7%. Fluor in UV.
1.580	1.575	(-).005	DARAPIOSITE	---	---	3.0	5.1	2V = 5°, ZrO ₂ 5.0, MnO 8.2, ZnO 7.8%.
1.592	---	---	MERRIHUEITE	---	24.9	---	---	
1.597	1.596	(-).001	EMELEUSITE	0.1	---	12.0	---	2V = 0-30°, r > v str. Fe ₂ O ₃ 12.1, Li ₂ O 2.8%.
1.606	1.608	(+).002	SOGDIANITE	---	1.2	2.8	4.8	TiO ₂ 2.9, Fe ₂ O ₃ 4.6, ZrO ₂ 9.8, Li ₂ O 3.7%. Violet.
1.610	1.607	(-).003	SUGILITE	---	0.2	4.4	3.8	Fe ₂ O ₃ 12.8, Li ₂ O 3.1%.

Table 31. Pyrochlore group

The minerals of this group are complex oxides having the general formula $A_{1-2}B_2O_6(O,OH,F)$, in which A may be Ba, Bi, Ca, Ce, K, Na, Pb, Sb^{+3} , Sn, Sr, Th, U, Y, and Zr; B may be Nb, Ta, and Ti. Jixianite is a related mineral with W and Fe^{+3} .

In the Pyrochlore subgroup, $Nb > Ta$, and $Nb + Ta > 2Ti$.

In the Microlite subgroup, $Ta > Nb$, and $Nb + Ta > 2Ti$.

In the Betafite subgroup, $Nb + Ta < 2Ti$.

Bariopyrochlore	$(Ba,Sr)_2(Nb,Ti)_2(O,OH)_7$
Betafite	$(Ca,Na,U)_2(Ti,Nb,Ta)_2O_6(OH)$
Bismutomicrolite	$(Bi,Ca)(Ta,Nb)_2O_6(OH)$
Jixianite	$Pb(W,Fe^{+3})_2(O,OH)_7$
Microlite	$(Ca,Na)_2(Ta,Nb)_2O_6(OH)$
Plumbopyrochlore	$(Pb,Y)(Nb,Ta)_2(O,OH)_7$
Pyrochlore	$(Na,Ca)_2(Nb,Ta)_2O_6(OH,F)$
Stibiobetafite	$(Ca,Sb^{+3})(Ti,Nb,Ta)_2(O,OH)_7$
Uranmicrolite	$(Ca,Na,U)_2(Ta,Nb,Ti)_2O_6(OH)$
Uranpyrochlore	$(Ca,Na,U)_2(Nb,Ta,Ti)_2(O,OH)_7$
Ytropyrochlore	$(Y,Na,Ca,U)_{1-2}(Nb,Ta,Ti)_2(O,OH)_7$
Zirkelite	$(Ca,Zr,Th)_2(Ti,Nb)_2O_7$

These minerals are cubic, space group $Fd3m$, commonly occurring as octahedra, with conchoidal fracture. Nearly all are infusible. Optically they are isotropic with high refractive index.

Refractive index	NAME	H	G	Color	Action in acid
1.83	YTROPYROCHLORE	4.5-5	3.7	Dark brown	Insol.
1.87	PYROCHLORE	5-6	3.77	Brown, yellow	Dec by H_2SO_4 .
1.915	BETAFITE	5	4.0	Greenish black	Dec by H_2SO_4 .
1.925	MICROLITE	5.5	5.5	Yellow, brown	Dec by H_2SO_4 .

Table 31. Pyrochlore group (continued)

Refractive index	NAME	H	G	Color	Action in acid
1.94	BETAFITE	5	5.2	Black, brown, yellow	Dec by H ₂ SO ₄ .
1.97	URANMICROLITE	5.5	5.8	Brown, black	Dec by H ₂ SO ₄ .
1.98	URANPYROCHLORE	5	4.8	Brown	Insol.
2.0	PYROCHLORE	5	4.1-4.3	Brown, red	Dec by H ₂ SO ₄ .
>2.00	BISMUTOMICROLITE	5	6.5-7.2	Yellow, brown	Insol.
2.02	BETAFITE	4.5	4.76	Black	Dec by H ₂ SO ₄ .
2.023	MICROLITE	6	6.42	Brown, yellow	
2.06	ZIRKELITE	6	4.02	Brown, black	Dec by hot HCl.
2.07	MICROLITE	6	5.95	Brown, yellow	Insol.
2.08	BARIOPYROCHLORE	4-4.5	3.7-4.0	Yellow, gray	Insol.
2.08	PLUMBOPYROCHLORE	---	6.34	Green-yellow to brown	---
2.18	PYROCHLORE	5	4.1-4.3	Brownish red	Dec by H ₂ SO ₄ .
2.18	ZIRKELITE	5-6	4.3-4.7	Black to dark brown	Insol.
2.2	STIBIOBETAFITE	~ 5	5.30	Brownish black	---
2.26	JIXIANITE	5	5.30	Brownish black	---
2.28	ZIRKELITE	5-6	4.3-4.7	Black to dark brown	Insol.
2.315	JIXIANITE	5	5.30	Brownish black	---

Table 32. Pyroxene group

The minerals of the pyroxene group are single-chain silicates with the general formula $(W)_{1-p}(X,Y)_{1+p}Z_2O_6$. One subgroup is orthorhombic, in which $p=1$ and X may be mainly Mg and Fe^{+2} . Another subgroup includes the monoclinic members, in which p ranges from 0 to 1, and W may be Ca and Na ; X may be Mg , Fe^{+2} , Mn , Ni , and Li ; Y may be Al , Fe^{+3} , Cr , and Ti ; Z may be Si and Al . A discussion of nomenclature is given by Deer, Howie, and Zussman (1978, v. 2A).

<u>Orthopyroxenes</u>	$(Mg,Fe)_2Si_2O_6$
Enstatite	En 100-87
Bronzite	En 87-70
Hypersthene	En 70-50
Ferrohypersthene	En 50-30
Eulite	En 30-10
(Orthoferrosilite	En 10-0)
<u>Clinopyroxenes</u>	
Acmite	$NaFe^{+3}Si_2O_6$
Aegirine-augite	$(Na,Ca)(Fe^{+3},Mg,Al)(Si,Al)_2O_6$
Augite	$(Ca,Na)(Mg,Fe,Al,Ti)(Si,Al)_2O_6$
Clinoenstatite	$Mg_2Si_2O_6$
Clinohypersthene	$(Mg,Fe)_2Si_2O_6$
Diopside	$CaMgSi_2O_6$
Fassaite	$Ca(Mg,Fe^{+3},Al)(Si,Al)_2O_6$
Ferroaugite	$(Ca,Na)(Fe,Mg,Al,Ti)(Si,Al)_2O_6$
Ferrosalite	$Ca(Fe,Mg)Si_2O_6$
Hedenbergite	$CaFe^{+2}Si_2O_6$
Jadeite	$NaAlSi_2O_6$
Johannsenite	$CaMnSi_2O_6$
Kanoite	$(Mn,Mg)_2Si_2O_6$
Omphacite	$(Ca,Na)(Mg,Fe^{+2},Fe^{+3},Al)Si_2O_6$

Table 32. Pyroxene group (continued)

Clinopyroxenes (continued)

Pigeonite	$(\text{Mg}, \text{Fe}^{+2}, \text{Ca})(\text{Mg}, \text{Fe}^{+2})\text{Si}_2\text{O}_6$
Salite	$\text{Ca}(\text{Mg}, \text{Fe})\text{Si}_2\text{O}_6$
Spodumene	$\text{LiAlSi}_2\text{O}_6$
Ureyite	$\text{NaCrSi}_2\text{O}_6$

The orthorhombic pyroxenes, space group Pbca, form a solid solution series from enstatite through eulite, with no natural occurrences of the theoretical iron end member, orthoferrosilite. Crystals are commonly prismatic, cleavage is good {210} at 87-88°, hardness 5 to 6. All members are insoluble in or only slightly attacked by acids. Birefringence is low, refractive indices increase with increasing iron content, with optic sign changing from positive to negative at about β 1.68 and back to positive at about β 1.76. All members of the series have $X=\underline{b}$, $Z=\underline{c}$, positive elongation. Although the enstatite end member is not pleochroic, many intermediate members show weak pleochroism, X pink, Y yellow, Z pale green.

The monoclinic pyroxenes have space group P2/c (except pigeonite, P2₁/c), good cleavage on {110} at about 87°, hardness 5 to 6. All are insoluble in or only very slightly attacked by acids. Optically, they are biaxial, with refractive indices increasing with increasing iron and titanium content; $Y=\underline{b}$ (except clinoenstatite, clinohypersthene, and some pigeonites, where $X=\underline{b}$). Minerals of the series diopside-hedenbergite-johannsenite along with the augites are optically positive, with moderate 2V and birefringence, $Z:\underline{c}=40-50^\circ$. Pigeonite, a Ca-poor member, is characterized by low 2V. The series jadeite to aegirine to aegirine-augite has high 2V (60-90°), with indices and birefringence increasing with increase in Fe-content. For aegirines, $Z:\underline{c}=75-89^\circ$.

Table 32. Pyroxene group (continued)

	Refractive index			Biref	NAME	2V	Opt orien- tation	System	Weight %				Remarks
	α	$\underline{\beta}$	γ						FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
379	1.640	<u>1.645</u>	1.652	.012	JADEITE	(+)67°	$\gamma = \frac{b}{Z:c} = 22^\circ$	MCL	---	---	---	---	Synth.
	1.650	<u>1.653</u>	1.658	.008	ENSTATITE En 100	(+)59° $r < v$	$\chi = \frac{b}{Z} = \frac{b}{c}$	ORTH	---	---	---	---	
	1.651	<u>1.654</u>	1.660	.009	CLINO- ENSTATITE En 100	(+)54°	$\chi = \frac{b}{Z:c} = 22^\circ$	MCL	---	---	---	---	
	1.654	<u>1.657</u>	1.666	.012	JADEITE	(+)70° $r > v$ med	$\gamma = \frac{b}{Z:c} = 34^\circ$	MCL	---	---	0.4	---	Pleoc, wk greens.
	1.653	<u>1.659</u>	1.677	.024	SPODUMENE	(+)66° $r < v$	$\gamma = \frac{b}{Z:c} = 25^\circ$	MCL	---	---	---	---	
	1.665	<u>1.669</u>	1.674	.009	ENSTATITE En 91	(+)80° $r < v$ wk	$\chi = \frac{b}{Z} = \frac{b}{c}$	ORTH	7.0	---	---	---	
	1.664	<u>1.670</u>	1.685	.021	OMPHACITE	(+)66° $r < v$ med	$\gamma = \frac{b}{Z:c} = 35^\circ$	MCL	2.8	---	1.8	---	
	1.662	<u>1.670</u>	1.682	.020	JADEITE	(+)70° $r > v$ med	$\gamma = \frac{b}{Z:c} = 35^\circ$	MCL	0.5	---	3.2	---	
	1.668	<u>1.671</u>	1.682	.014	SPODUMENE	(+)65° $r < v$	$\gamma = \frac{b}{Z:c} = 24^\circ$	MCL	---	---	0.7	---	

Table 32. Pyroxene group (continued)

Refractive index			Biref	NAME	2V	Opt orien- tation	System	Weight %				Remarks
α	$\underline{\beta}$	γ						FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.665	<u>1.672</u>	1.695	.030	DIOPSIDE	(+)57° r > v wk	$Y = \frac{b}{c}$ $Z:\underline{c} = 39^\circ$	MCL	1.5	0.1	0.6	---	TiO ₂ 1.1%.
1.673	<u>1.678</u>	1.705	.032	DIOPSIDE	(+)48° r > v	$Y = \frac{b}{c}$ $Z:\underline{c} = 42^\circ$	MCL	8.2	0.3	1.4	4.5	
1.675	<u>1.678</u>	1.688	.013	CLINO- HYPERSTHENE	(+)57°	$X = \frac{b}{c}$ $Z:\underline{c} = 36^\circ$	MCL	12.6	0.5	---	0.5	
1.674	<u>1.680</u>	1.685	.011	BRONZITE	(-)79°	$X = \frac{b}{c}$ $Z = \underline{c}$	ORTH	11.1	0.5	---	---	Pleoc, wk greens.
1.673	<u>1.681</u>	1.695	.022	OMPHACITE	(+)70° r < v med	$Y = \frac{b}{c}$ $Z:\underline{c} = 41^\circ$	MCL	2.7	---	1.3	8.3	
1.679	<u>1.681</u>	1.685	.006	JADEITE	(+)64° r > v med	$Y = \frac{b}{c}$ $Z:\underline{c} = 38^\circ$	MCL	1.1	---	5.7	---	
1.676	<u>1.683</u>	1.705	.029	DIOPSIDE, manganoan	(+)1.0° r > v	$Y = \frac{b}{c}$ $Z:\underline{c} = 5^\circ$	MCL	---	7.4	0.4	0.4	ZnO 3.3%.
1.684	<u>1.684</u>	1.707	.023	PIGEONITE	(+)18- 26°	$Y = \frac{b}{c}$ $Z:\underline{c} = 40^\circ$	MCL	20.1	0.4	3.5	---	CaO 5.5%.
1.677	<u>1.685</u>	1.708	.031	DIOPSIDE	(+)59° r > v wk	$Y = \frac{b}{c}$ $Z:\underline{c} = 39^\circ$	MCL	5.6	0.5	3.2	2.2	

Table 32. Pyroxene group (continued)

Refractive index			Biref	NAME	2V	Opt orien- tation	System	Weight %				Remarks
α	$\underline{\beta}$	γ						FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.687	<u>1.692</u>	1.715	.028	DIOPSIDE	(+)51°	$Y = \frac{b}{Z:c} = 42^\circ$	MCL	9.5	0.2	1.6	2.3	TiO ₂ 0.5%.
1.686	<u>1.693</u>	1.719	.028	FASSAITE	(+)56° r > v med	$Y = \frac{b}{Z:c} = 43^\circ$	MCL	3.1	---	2.1	13.4	
1.685	<u>1.695</u>	1.700	.015	BRONZITE En 73	(-)69° r > v	$X = \frac{b}{Z} = \frac{b}{c}$	ORTH	17.1	0.4	1.0	---	CaO 5.5%.
1.696	<u>1.698</u>	1.721	.025	PIGEONITE	(+)25°	$Y = \frac{b}{Z:c} = 41^\circ$	MCL	21.4	0.4	0.8	---	
1.690	<u>1.699</u>	1.721	.031	DIOPSIDE, manganoan	(+)60° r > v wk	$Y = \frac{b}{Z:c} = 43^\circ$	MCL	1.7	10.0	1.5	0.3	MgO 10.5%.
1.689	<u>1.699</u>	1.718	.029	OMPHACITE	(+)64° r < v med	$Y = \frac{b}{Z:c} = 40^\circ$	MCL	5.9	---	3.8	6.1	
1.695	<u>1.700</u>	1.720	.025	AUGITE	(+)50°	$Y = \frac{b}{Z:c} = 43^\circ$	MCL	14.1	0.2	1.3	2.9	TiO ₂ 1.3%.
1.700	<u>1.707</u>	1.724	.024	SALITE	(+)52°	$Y = \frac{b}{Z:c} = 44^\circ$	MCL	9.8	0.3	1.8	2.8	
1.700	<u>1.710</u>	1.733	.033	JOHANNSENITE	(+)60° r > v	$Y = \frac{b}{Z:c} = 49^\circ$	MCL	1.5	22.6	0.8	---	
1.703	<u>1.713</u>	1.717	.014	HYPERSTHENE En 57	(-)53°	$X = \frac{b}{Z} = \frac{b}{c}$	ORTH	25.7	---	1.0	---	

Table 32. Pyroxene group (continued)

Refractive index			Biref	NAME	2V	Opt orien- tation	System	Weight %				Remarks
α	β	γ						FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.708	<u>1.714</u>	1.736	.028	FERROSALITE	(+)70° r > v wk	$Y = \frac{b}{c}$ $Z = \frac{c}{b}$	MCL	18.6	0.2	0.5	1.1	
1.714	<u>1.714</u>	1.742	.028	PIGEONITE	(+)0- 12°	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 41^\circ$	MCL	27.8	1.0	1.7	0.9	CaO 3.8%.
1.710	<u>1.716</u>	1.736	.028	FERROAUGITE	(+)52°	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 49^\circ$	MCL	20.2	1.1	3.5	0.9	TiO ₂ 1.2%.
1.715	<u>1.717</u>	1.728	.013	KANOITE	(+)41°	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 48^\circ$	MCL	3.1	31.2	---	---	MgO 15.1%.
1.710	<u>1.719</u>	1.738	.028	JOHANNSENITE	(+)70°	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 48^\circ$	MCL	1.0	26.8	0.3	---	
1.712	<u>1.719</u>	1.736	.024	FASSAITE	(+)55° r > v	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 46^\circ$	MCL	0.2	---	6.1	15.8	CaO 25.3%.
1.714	<u>1.723</u>	1.774	.060	AUGITE, zincian	(+)74° (47+5°)	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 55^\circ$	MCL	3.7	7.2	4.2	1.2	ZnO 7.7%.
1.715	<u>1.728</u>	1.731	.016	FERROHYPER- STHENE En 45	(-)51°	$X = \frac{b}{c}$ $Z = \frac{c}{b}$	ORTH	29.4	0.2	2.1	---	
1.725	<u>1.730</u>	1.750	.025	AUGITE, titanian	(+)40°	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 45^\circ$	MCL	5.0	0.2	4.4	7.4	TiO ₂ 5.5%.
1.726	<u>1.732</u>	1.753	.027	HEDENBERGITE	(+)49° r > v wk	$Y = \frac{b}{c}$ $Z:c = \frac{b}{c} = 44^\circ$	MCL	24.7	0.7	1.9	---	Pleoc, X pale green, Z dark green.

Table 32. Pyroxene group (continued)

Refractive index			Biref	NAME	2V	Opt orien- tation	System	Weight %				Remarks
α	$\underline{\beta}$	γ						FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.727	<u>1.736</u>	1.751	.024	FASSAITE	(+)49° r > v	$Y = \underline{b}$ $Z:\underline{c} = 48^\circ$	MCL	1.2	---	6.2	12.9	TiO ₂ 3.0%.
1.720	<u>1.740</u>	1.757	.037	AEGIRINE- AUGITE	(-)85° r < v	$Y = \underline{b}$ $Z:\underline{c} = 75^\circ$	MCL	7.5	0.6	14.0	2.2	Pleoc, X bright green, Y and Z greenish yellow. Na ₂ O 5.3%.
1.735	<u>1.741</u>	1.761	.026	FERROAUGITE	(+)51°	$Y = \underline{b}$ $Z:\underline{c} = 40^\circ$	MCL	24.0	0.4	1.2	4.1	TiO ₂ 1.6%.
1.736	<u>1.747</u>	1.752	.016	EULITE En 30	(-)63°	$X = \underline{b}$ $Z = \underline{c}$	ORTH	39.4	0.9	---	---	
1.741	<u>1.751</u>	1.774	.033	ACMITE	(+)mod	$Y = \underline{b}$ $Z:\underline{c} = 60^\circ$	MCL	1.4	7.2	12.9	---	Zn 8.8%.
1.745	<u>1.753</u>	1.771	.026	HEDENBERGITE	(+)52° r > v wk	$Y = \underline{b}$ $Z:\underline{c} = 41^\circ$	MCL	29.7	0.9	1.4	---	
1.732	<u>1.756</u>	1.770	.038	ACMITE	(-)80° r > v	$Y = \underline{b}$ $Z:\underline{c} = 81^\circ$	MCL	1.1	4.7	18.2	5.9	Pleoc, X pink, Y pale blue, Z blue. TiO ₂ 0.6%.
1.740	<u>1.756</u>	1.762	.022	UREYITE	(-)65°	$Y = \underline{b}$ $Z:\underline{c} = 76^\circ$	MCL	---	---	---	---	
1.752	<u>1.762</u>	1.772	.020	EULITE En ~ 11	(-)87° r < v str	$X = \underline{b}$ $Z = \underline{c}$	ORTH	44.5	---	1.9	---	Dark brown.

Table 32. Pyroxene group (continued)

Refractive index			Biref	NAME	2V	Opt orien- tation	System	Weight %				Remarks
α	$\underline{\beta}$	γ						FeO	MnO	Fe ₂ O ₃	Al ₂ O ₃	
1.741	<u>1.767</u>	1.789	.048	AEIRINE- AUGITE	(-)85° r > v	$Y = \underline{b}$ $Z:\underline{c} = 78^\circ$	MCL	5.2	0.5	19.1	1.3	Pleoc, X deep green, Y yellow-green, Z brown-green, abs X > Y = Z. Na ₂ O 7.2, TiO ₂ 1.6%.
1.751	1.786	1.800	.049	ACMITE	(-)69° r > v	$Y = \underline{b}$ $Z:\underline{c} = 89^\circ$	MCL	3.7	0.2	25.4	2.9	Pleoc, X and Y deep green, Z yellow-brown. Na ₂ O 11.7, TiO ₂ 0.7%.
1.770	1.812	1.830	.060	ACMITE	(-)58° r > v	$Y = \underline{b}$ $Z:\underline{c} = 84^\circ$	MCL	0.8	---	31.4	1.9	Pleoc, X pale yellow-green, Y yellow-green, Z emerald green, TiO ₂ 0.8%.

Table 33. Scapolite group

The minerals of this group are solid solutions of the end-members marialite, $\text{Na}_8(\text{AlSi}_3\text{O}_8)_6(\text{Cl}_2, \text{SO}_4, \text{CO}_3)$, and meionite, $\text{Ca}_8(\text{Al}_2\text{Si}_2\text{O}_8)_6(\text{CO}_3, \text{Cl}_2, \text{SO}_4)$, which themselves are not found as such in nature. The composition and paragenetic relations of the scapolites have been reviewed in detail by Shaw (1960).

Minerals of this group are tetragonal, space group $P4/m$, with cleavages $\{100\}$ perfect, $\{110\}$ less so, hardness 5 to 6. Optically they are uniaxial negative, refractive indices and birefringence increasing with Ca-content.

Refractive index		Biref	NAME	Weight %					
ω	ϵ			Na_2O	K_2O	CaO	Cl	CO_2	SO_3
1.532	1.522	(-).010	MARIALITE	7.1	4.3	6.3	1.0	2.0	---
1.546	1.542	(-).004	MARIALITE Me 19	10.5	1.2	4.8	3.0	1.1	0.03
1.561	1.546	(-).015	SCAPOLITE Me 33	8.6	1.1	8.3	2.2	1.7	0.4
1.574	1.549	(-).025	SCAPOLITE Me 54	5.4	1.0	12.5	1.4	2.9	0.1
1.587	1.555	(-).032	MEIONITE Me 70	2.4	2.4	15.8	0.1	2.9	0.9
1.607	1.571	(-).036	MEIONITE Me 84	2.0	0.3	20.4	0.2	4.1	0.3

Table 34. Sodalite group

The minerals of this group are sodium- and calcium-aluminosilicates having the general formula $(\text{Na,Ca})_{6-8}\text{Al}_6\text{Si}_6(\text{O,S})_{24}[\text{SO}_4,\text{Cl}_2,(\text{OH})_2]_{1-2}$.

Hauyne	$(\text{Na,Ca})_{6-8}\text{Al}_6\text{Si}_6(\text{O,S})_{24}(\text{SO}_4,\text{Cl}_2)_{1-2}$
Lazurite	$(\text{Na,Ca})_{7-8}(\text{Al,Si})_{12}(\text{O,S})_{24}[\text{SO}_4,\text{Cl}_2(\text{OH})_2]$
Nosean	$\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}(\text{SO}_4)$
Sodalite	$\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}\text{Cl}_2$

These minerals are cubic, space group $P\bar{4}3m$ or $P\bar{4}3n$, habit dodecahedral, color blue or white, cleavage {110} poor, hardness 5 to 6. They fuse at 3.5-4.5, gelatinize with HCl. Optically they are isotropic, with refractive index increasing with Ca-content. The group is discussed by Taylor (1967) and by Hogarth and Griffin (1976).

Refractive index	NAME	G	Weight %				
			Na ₂ O	CaO	SO ₃	Cl	Other
1.470	NOSEAN	2.32	20.7	1.7	8.8	0.7	H ₂ O 2.5%.
1.483	SODALITE	2.30	---	---	---	7.1	
1.487	SODALITE var. Hackmanite	2.2-2.3	---	---	---	---	
1.494	HAUYNE	2.40	19.9	4.9	9.8	1.3	
1.495	NOSEAN	2.3-2.4	---	---	---	---	K ₂ O 2.8%.
1.500 ✓	LAZURITE	2.4	---	---	---	---	
1.509	HAUYNE	2.51	13.2	10.1	---	---	
1.522	LAZURITE	2.4	---	---	---	---	

Table 35. Spinel group

Minerals of this group have the general formula AB_2O_4 , in which A may be Mg, Fe^{+2} , Zn, Mn^{+2} , rarely Ni, Cu, Co, Ti, and Ge; B may be Al, Fe^{+3} , Cr, Mn^{+3} , rarely Fe^{+2} and V^{+3} . Common end members are spinel (Mg-Al), hercynite (Fe-Al), gahnite (Zn-Al), magnesiochromite (Mg-Cr), chromite (Fe-Cr), magnesioferrite (Mg-Fe), magnetite (Fe-Fe), and jacobsite (Mn-Fe).

Chromite	$(Fe^{+2}, Mg)(Cr, Al, Fe^{+3})_2O_4$
Franklinite	$(Zn, Mn^{+2}, Fe^{+2})(Mn^{+3}, Fe^{+3})_2O_4$
Gahnite	$Zn(Al, Fe^{+3})_2O_4$
Galaxite	$(Mn, Fe^{+2})(Al, Fe^{+3})_2O_4$
Hercynite	$(Fe^{+2}, Mg)(Al, Fe^{+3})_2O_4$
Jacobsite	$(Mn, Fe^{+2}, Mg)Fe^{+3}_2O_4$
Magnesiochromite	$(Mg, Fe^{+2})(Cr, Al, Fe^{+3})O_4$
Magnesioferrite	$(Mg, Fe^{+2})Fe^{+3}_2O_4$
Magnetite	$Fe^{+2}Fe^{+3}_2O_4$
Spinel	$(Mg, Fe^{+2})Al_2O_4$
Trevorite	$NiFe_2O_4$

Rare minerals of the group include coulsonite (Fe-V), cuprospinel (Cu-Fe), ulvöspinel ($TiFe^{+2}_2O_4$), and its germanium analogue brunogeierite, $GeFe_2O_4$. Ringwoodite, $(Mg, Fe)_2SiO_4$, has spinel-type structure ($n=1.768$).

These minerals are cubic, commonly octahedral, space group $Fd\bar{3}m$. They are characterized by high refractive index and specific gravity, lack of cleavage, hardness 7.5 to 8, and infusibility. Those members high in Fe^{+3} or V^{+3} (magnetite, coulsonite, ulvöspinel, brunogeierite) are opaque or nearly so. The aluminum and chromium end-members are insoluble in acids; those with dominant Fe^{+3} are dissolved by HCl.

Table 35. Spinel group (continued)

Refractive index	NAME	G	Weight %					Remarks
			Cr ₂ O ₃	Fe ₂ O ₃	FeO	MnO	MgO	
1.714	SPINEL	3.58	---	---	---	---	---	Synth end member.
1.738	SPINEL	3.62	---	1.5	5.6	0.1	---	
1.747	SPINEL, zincian	3.97	---	---	1.9	---	16.8	ZnO 18.2.
1.776	HERCYNITE	4.20	---	11.7	23.3	---	10.8	"Pleonaste"
1.782	GAHNITE	4.38	---	---	8.5	---	2.4	ZnO 31.4.
1.798	SPINEL, chromian	3.94	19.6	---	9.8	---	19.5	Al ₂ O ₃ 48.3.
1.805	GAHNITE	4.61	---	---	---	---	---	Synth end member.
1.818	GAHNITE	4.60	---	2.9	1.7	---	---	
1.833	SPINEL, chromian	3.93	16.8	5.8	11.7	---	19.3	Al ₂ O ₃ 45.4.
1.835	HERCYNITE	4.40	---	---	---	---	---	Synth end member.
1.856	GALAXITE	---	---	6.4	2.3	33.1	---	
1.90	MAGNESIO-CHROMITE	4.39	---	---	---	---	---	Synth end member.
1.923	GALAXITE	4.23	---	---	16.4	34.0	1.5	
1.97	MAGNESIO-CHROMITE	4.67	43.6	4.0	11.1	---	16.0	Al ₂ O ₃ 24.4.
2.10	CHROMITE	4.65	49.7	8.1	8.7	9.3	2.4	Al ₂ O ₃ 17.6, ZnO 4.1.
2.12	CHROMITE	5.22	---	---	---	---	---	Synth end member.
2.30	JACOBSITE	4.84	---	---	---	---	---	Wk magn.

Table 35. Spinel group (continued)

Refractive index	NAME	G	Weight %					Remarks
			Cr ₂ O ₃	Fe ₂ O ₃	FeO	MnO	MgO	
2.3 (?)	TREVORITE	5.2	---	---	---	---	---	Str magn, n(calc) = 2.39.
2.38 Na 2.34 Li	MAGNESIO- FERRITE	4.5- 4.7	---	---	---	---	---	Str magn.
2.36 Li	FRANKLINITE	5.2	---	---	---	---	---	Wk magn.
2.42 Na	MAGNETITE	5.17	---	---	---	---	---	Str magn.
2.45 Na 2.41 Li	FRANKLINITE	5.2	---	---	---	---	---	Wk magn.
2.585 Na 2.535 Li	JACOBSITE	4	---	---	---	---	---	Wk-mod magn.

Table 36. Stibiconite group

Minerals of this group have the general formula $A_{1-2}B_2O_6(O,OH,F)$, in which A may be Bi⁺³, Ca, Na, Pb, Sb⁺³, Ag, K, Fe, and Mn; B may be Sb⁺⁵ and Ti.

Bindheimite	Pb ₂ Sb ₂ O ₆ (O,OH)
Bismutostibiconite	(Bi,Fe)Sb ₂ O ₇
Romeite	(Ca,Na,Fe,Mn) ₂ (Sb,Ti) ₂ O ₆ (O,OH)
Stetefeldite	Ag ₂ Sb ₂ (O,OH) ₇ (?)
Stibiconite	SbSb ₂ O ₆ (O,OH)

These minerals are cubic, space group $Fd\bar{3}m$, isostructural with those of the pyrochlore group. They are commonly massive, fine-grained, and highly variable in physical properties. Specific gravity is high; optically isotropic.

Table 36. Stibiconite group (continued)

Refractive index	NAME	G	Remarks
1.60	STIBICONITE	5.1-5.6	Insol in acids. Infus.
1.70	STIBICONITE	5.1-5.6	Insol in acids. Infus.
1.72	BINDHEIMITE	4.6-5.6	Dec by HCl. F 3-4.
1.74	STIBICONITE	5.2-5.6	Insol in acids. Infus.
1.79	BINDHEIMITE	4.6-5.6	Dec by HCl. F 3-4.
1.80	STIBICONITE	5.2-5.6	Insol in acids. Infus.
1.817	ROMEITE	4.7-5.4	Insol in acids. F diff.
1.84	BINDHEIMITE	4.6-5.4	Dec by HCl. F 3-4.
1.87	ROMEITE	4.7-5.4	Insol in acids. F 4.5.
1.90	STIBICONITE	5.2-5.6	Insol in acids. Infus.
1.94	BINDHEIMITE	5.0-5.6	Dec by HCl. F 3-4.
1.95	STETEFELDITE	4.1-4.6	
2.05	STIBICONITE	5.2-5.6	Insol in acids. Infus.
2.09	BISMUTOSTIBICONITE	7.38	
2.09	ROMEITE, var Schneebergite	5.41	Insol in acids. F diff. Anom biref.
2.20	ROMEITE, var Lewisite	4.95	Insol in acids. Fus. Anom biref.

Table 37. Tourmaline group

Minerals of this group are borosilicates having the general formula $AX_3Y_6(BO_3)_3Si_6O_{18}(OH,F)_4$, in which A may be Ca, Na, and K; X may be Al, Fe⁺², Fe⁺³, Li, Mg, Mn⁺², and V⁺³; Y is mainly Al, but may also be Cr⁺³, Fe⁺³, and V⁺³.

Buergerite	$NaFe^{+3}Al_6(BO_3)_3Si_6O_{18}(O,OH,F)_4$
Chromdravite	$Na(Mg,V,Al)_3(Cr,Fe^{+3})_6(BO_3)_3Si_6O_{18}(OH)_4$
Dravite	$(Na,Ca)(Mg,Fe)_3Al_6(BO_3)_3Si_6O_{18}(OH,F)_4$
Elbaite	$Na(Al,Li,Fe)_3Al_6(BO_3)_3Si_6O_{18}(OH,F)_4$
Ferridravite	$(Na,K)(Mg,Fe^{+2})_3(Fe^{+3},Al)_6(BO_3)_3Si_6O_{18}(O,OH)_3$
Liddicoatite	$Ca(Li,Al)_3Al_6(BO_3)_3Si_6O_{18}(O,OH,F)_4$
Schorl	$(Na,Ca)(Mg,Fe)_3(Al,Fe^{+3})_6(BO_3)_3Si_6O_{18}(OH)_4$
Uvite	$Ca(Mg,Fe)_3Al_5Mg(BO_3)_3Si_6O_{18}(OH)_4$

These minerals are trigonal, space group $R\bar{3}m$, with characteristic prismatic habit, lack of cleavage (except buergerite), specific gravite 3.0-3.3. All have hardness of 7 and are insoluble in acids. Optically they are uniaxial negative with moderate to strong birefringence and negative elongation. Absorption in the pleochroic members is $O > E$.

Table 37. Tourmaline group (continued)

Refractive index		Biref (-)	NAME	Pleochroism		Weight %					
ω	ϵ			O	E	Fe ₂ O ₃	FeO	MgO	CaO	Na ₂ O	Other
1.634	1.612	.022	DRAVITE	---	---	---	.02	8.4	---	1.9	
1.635	1.618	.017	ELBAITE	---	---	.03	---	---	0.4	2.4	Li ₂ O 1.9.
1.637	1.621	.016	LIDDICOATITE	Dark brown	Pale brown	---	---	---	4.2	0.9	Li ₂ O 2.5, F 1.7.
1.638	1.619	.019	UVITE	Dark	Light	---	0.5	15.4	5.5	0.2	
1.646	1.625	.021	ELBAITE	---	---	---	4.9	---	---	2.5	Li ₂ O 1.2, MnO 0.7.
1.653	1.637	.016	DRAVITE	Dark olive green	Pale brown	2.2	3.6	9.3	1.6	1.9	
1.660	1.639	.021	UVITE	Dark	Light	2.7	5.8	11.0	7.9	---	
1.661	1.632	.029	SCHORL	---	---	0.8	8.7	5.6	---	2.3	
1.675	1.643	.032	DRAVITE	Red-brown	Pale green	---	1.3	6.3	---	---	V ₂ O ₃ 8.0.
1.693	1.659	.034	SCHORL	---	---	9.2	0.6	7.3	1.5	2.6	TiO ₂ 2.2.
1.710	1.664	.046	DRAVITE, chromian	---	---	1.8	1.8	---	---	---	Cr ₂ O ₃ 17.8.
1.735	1.625	.110	BUERGERITE	Yellow-brown	Pale yellow	17.6	1.3	---	0.7	2.5	

Table 37. Tourmaline group (continued)

Refractive index		Biref (-)	NAME	Pleochroism		Weight %					
ω	ϵ			O	E	Fe ₂ O ₃	FeO	MgO	CaO	Na ₂ O	Other
1.778	1.772	.006	CHROMDRAVITE	Dark green	Yellow-green	7.65	---	9.05	0.16	2.66	Cr ₂ O ₃ 31.6, V ₂ O ₃ 1.46.
1.800	1.743	.057	FERRIDRAVITE	Very dark brown	Light brown	38.4	7.5	5.6	---	2.2	K ₂ O 1.0.

Table 38. Zeolite group

The zeolite minerals form a complex group of hydrated aluminosilicates with ratio $O:(Si+Al)=2$. They have the general formula $A(XO_2)_n \cdot mH_2O$, in which A may be Na, Ca, and K, and less commonly Ba, Sr, Mg, Li(7), and Cs; X may be Si and Al. They are characterized by ready loss of their water, which can be restored, at least in part, and by their ability to exchange basic cations.

Amicite	$K_2Na_2Al_4Si_4O_{16} \cdot 5H_2O$
Analcime	$(Na, K, Cs)AlSi_2O_6 \cdot H_2O$
Barrerite	$(Na, K, Ca)_2Al_2Si_7O_{18} \cdot 7H_2O$
Brewsterite	$(Sr, Ba, Ca)Al_2Si_6O_{16} \cdot 5H_2O$
Chabazite	$(Ca, Na_2)Al_2Si_4O_{12} \cdot 6H_2O$
Clinoptilolite	$(Na_2, K_2, Ca)_{2-3}Al_3(Al, Si)_2Si_{13}O_{36} \cdot 12H_2O$
Cowlesite	$CaAl_2Si_3O_{10} \cdot 6H_2O$
Dachiardite	$(Ca, Na_2, K_2)_5Al_{10}Si_{38}O_{96} \cdot 25H_2O$
Edingtonite	$BaAl_2Si_3O_{10} \cdot 4H_2O$
Epistilbite	$CaAl_2Si_6O_{16} \cdot 5H_2O$
Erionite	$(Ca, Na_2, K_2)Al_2Si_6O_{16} \cdot 6H_2O$
Faujasite	$Na_2CaAl_4Si_{10}O_{28} \cdot 20H_2O$
Ferrierite	$(Na, K)_2MgAl_3Si_{15}O_{36}(OH) \cdot 9H_2O$
Garronite	$Na_2Ca_5Al_{12}Si_{20}O_{64} \cdot 27H_2O$
Gismondine	$CaAl_2Si_2O_8 \cdot 4H_2O$
Gmelinite	$(Na_2, Ca)Al_2Si_4O_{12} \cdot 6H_2O$
Gobbinsite	$Na_4(Ca, K_2)Al_6Si_{10}O_{32} \cdot 12H_2O$
Gonnardite	$Na_2CaAl_4Si_6O_{20} \cdot 7H_2O$
Goosecreekite	$CeAl_2Si_6O_{16} \cdot 5H_2O$
Harmotome	$(Ba, K_2)_{1-2}(Al, Si)_iO_{16} \cdot 6H_2O$
Herschelite	$(Na_2, Ca, K_2)AlSi_2O_6 \cdot 3H_2O$
Heulandite	$(Na, K)_{2-3}Al_3(Al, Si)_2Si_{13}O_{36} \cdot 12H_2O$
Laumontite	$CaAl_2Si_4O_{12} \cdot 4H_2O$

Table 38. Zeolite group (continued)

Levyne	$\text{CaAl}_2\text{Si}_3\text{O}_{10} \cdot 5\text{H}_2\text{O}$
Mazzite	$(\text{K}_2, \text{Na}_2, \text{Ca})_6\text{Mg}_3\text{Al}_{13}\text{Si}_{27}\text{O}_{80} \cdot 16\text{H}_2\text{O} (?)$
Merlinoite	$(\text{K}, \text{Ca}, \text{Na})_7\text{Si}_{23}\text{Al}_9\text{O}_{64} \cdot 23\text{H}_2\text{O}$
Mesolite	$\text{Na}_2\text{CaAl}_6\text{Si}_9\text{O}_{30} \cdot 8\text{H}_2\text{O}$
Mordenite	$(\text{Ca}, \text{Na}_2, \text{K}_2)\text{Al}_2\text{Si}_{10}\text{O}_{24} \cdot 7\text{H}_2\text{O}$
Natrolite	$(\text{Na}_2, \text{Ca})\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 2\text{H}_2\text{O}$
Offretite	$(\text{K}_2, \text{Ca})_3\text{Al}_5\text{Si}_{13}\text{O}_{36} \cdot 14\text{H}_2\text{O}$
Paulingite	
Phillipsite	$(\text{K}_2, \text{Na}_2, \text{Ca})_{1-2}(\text{Al}, \text{Si})_8\text{O}_{16} \cdot 6\text{H}_2\text{O}$
Pollucite	$(\text{Cs}, \text{Na})\text{AlSi}_2\text{O}_6 \cdot m\text{H}_2\text{O}$
Scolecite	$\text{CaAl}_2\text{Si}_3\text{O}_{10} \cdot 3\text{H}_2\text{O}$
Sodium dachiardite	$(\text{Na}_2, \text{Ca}, \text{K})_{4-5}\text{Al}_8\text{Si}_{40}\text{O}_{96} \cdot 26\text{H}_2\text{O}$
Stellerite	$\text{CaAl}_2\text{Si}_9\text{O}_{18} \cdot 7\text{H}_2\text{O}$
Stilbite	$\text{NaCa}_2\text{Al}_5\text{Si}_{13}\text{O}_{36} \cdot 14\text{H}_2\text{O}$
Tetranatrolite	$\text{Na}_2\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 2\text{H}_2\text{O}$
Thomsonite	$\text{NaCa}_2\text{Al}_5\text{Si}_5\text{O}_{20} \cdot 6\text{H}_2\text{O}$
Wairakite	$\text{CaAl}_2\text{Si}_4\text{O}_{12} \cdot 2\text{H}_2\text{O}$
Wellsite	$(\text{Ba}, \text{Ca}, \text{K}_2)\text{Al}_2\text{Si}_3\text{O}_{10} \cdot 3\text{H}_2\text{O}$
Yugawaralite	$\text{CaAl}_2\text{Si}_6\text{O}_{16} \cdot 4\text{H}_2\text{O}$

The basic structure of zeolite minerals is an open framework of $(\text{Si}, \text{Al})\text{O}_4$ tetrahedra, which accounts for their cation exchange and absorption capacity and for the ability of water molecules to move out of and into the framework. In the zeolite group are representatives of every crystal system except triclinic. Hardness is 3.5 to 5, specific gravity is low (mostly 2.0-2.3), cleavage is good to perfect. Most are gelatinized or otherwise decomposed by acids.

These minerals vary in optical character but typically have low refractive index and low to very low birefringence. Depending on the degree of dehydration of a given species, the optic angle may vary widely, even to changing optic sign and orientation. This, together with the overlap of properties and the lack of adequate documentation of some of the examples, makes it difficult to distinguish between members of this group by optical means.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta, \omega,}{\text{or } \underline{n}}$	γ or ϵ					
---	<u>1.461</u>	1.465	.004	CHABAZITE	(+)0°	TRIG	
1.460	<u>1.462</u>	---	.001	CHABAZITE	(-)0°	TRIG	Anom biax.
---	<u>1.468</u>	1.473	.008	ERIONITE	(+)0°	HEX	Diff dec by acids. El pos.
1.468	<u>1.470</u>	1.473	.005	ERIONITE	(+)0°	HEX	Diff dec by acids.
---	<u>1.470</u>	1.474	.004	GMELINITE	(+)0°	TRIG	Tw. 2V 0° to small.
---	<u>1.471</u>	1.474	.003	HERSCHELITE	(+)0°	TRIG	
---	<u>1.473</u>	---	.000	PAULINGITE	---	CUB	
1.470	<u>1.474</u>	---	.004	HERSCHELITE	(-)0°	TRIG	
1.471	<u>1.475</u>	1.476	.005	SODIUM- DACHIARDITE	(-)88- 92°	MCL	Z:c = 18°.
1.472	<u>1.475</u>	1.477	.005	MORDENITE	(-)80°	ORTH	X = $\frac{a}{c}$, Z = $\frac{c}{a}$
---	<u>1.477</u>	1.480	.003	ERIONITE	(+)0°	HEX	Diff dec by HCl. El pos.
1.475	<u>1.477</u>	---	.002	GMELINITE	(-)0°	TRIG	Anom biax. Tw.
1.476	<u>1.479</u>	1.479	.003	CLINOPTILOLITE	(-)small	MCL	X = $\frac{b}{a}$?, Z:a = 15°.
1.478	<u>1.479</u>	1.481	.003	CLINOPTILOLITE	(+)40°	MCL	X = $\frac{b}{a}$, el clv pos.
1.478	<u>1.479</u>	1.482	.004	FERRIERITE	(+)50°	ORTH	Insol in HCl. X = $\frac{a}{b}$, Y = $\frac{b}{a}$.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta}{\text{or } \frac{\omega}{n}}$	γ or ϵ					
---	<u>1.479</u>	---	.000	ANALCIME	---	CUB	Opt anom.
---	<u>1.480</u>	1.493	.013	TETRANATROLITE	(+)0°	TET	
1.477	<u>1.480</u>	1.490	.013	NATROLITE	(+)63°	ORTH	X = $\frac{a}{c}$, Z = $\frac{e}{l}$ pos.
---	<u>1.48</u>	---	wk	FAUJASITE	---	CUB	On losing H ₂ O, becomes uniax pos.
1.480	(<u>1.480</u>)	1.485	.005	CHABAZITE	(+)small	TRIG	
---	<u>1.481</u>	1.483	.002	LEVYNE	(+)0°	TRIG	
---	<u>1.482</u>	(1.484-1.490)	.002-.008	GMELINITE	(+) ~ 0°	HEX	
1.481	<u>1.482</u>	1.486	.005	MORDENITE	(+)large	ORTH	Partly dec by HCl. X = $\frac{c}{a}$, Y = $\frac{a}{b}$.
1.481	<u>1.484</u>	1.486	.005	CLINOPTILOLITE	(-)large	MCL	X = $\frac{b}{c}$ (?), Z:c = 15°.
1.483	<u>1.485</u>	1.487	.004	PHILLIPSITE	(+)63°	MCL	X = $\frac{b}{c}$, Z:c = 19°
1.479	<u>1.485</u>	1.489	.010	BARRERITE	(-78°)	ORTH	X = $\frac{a}{b}$, Y = $\frac{b}{c}$.
1.485	<u>1.485</u>	1.488	.003	CHABAZITE	(+)med	TRIG	Tw.
1.483	<u>1.486</u>	1.495	.012	NATROLITE	(+)med	ORTH	X = $\frac{a}{c}$, Z = $\frac{e}{l}$, tw.
1.484	<u>1.486</u>	1.487	.003	PHILLIPSITE	(-)med	MCL	Penet tw.
---	<u>1.487</u>	1.489	.002	CHABAZITE	(+)0°	TRIG	Anom biax.
1.486	<u>1.487</u>	---	.001	ANALCIME	(-)small	CUB	Anom biax, lam tw.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta}{\text{or } \underline{n}}$, $\frac{\omega}{\text{or } \underline{n}}$	γ or ϵ					
1.488	<u>1.488</u>	1.489	.001	HEULANDITE	(+)70°	MCL	El clv neg.
1.484	(<u>1.488</u>)	1.490	.006	CHABAZITE	(-)60-70°	TRIG	Anom biax, tw.
1.482	<u>1.489</u>	1.496	.014	STILBITE	(-)43°	MCL	Y = b, X:a = 2-9°, e $\bar{1}$ neg. Cruciform penet tw.
1.489	<u>1.489</u>	1.492	.003	FERRIERITE	(+)small	ORTH	Insol in HCl. El pos.
1.486	<u>1.489</u>	---	.003	OFFRETITE	(-)0°	HEX	El pos.
1.486	<u>1.489</u>	---	.003	CHABAZITE	(-)0°	TRIG	Anom biax.
1.487	<u>1.489</u>	---	.002	LEVYNE	(-)0°	TRIG	
1.490	<u>1.490</u>	1.502	.012	NATROLITE	(+)0°	ORTH	El pos.
1.485	<u>1.490</u>	1.494	.009	AMICITE	(-)82°	MCL	
1.491	<u>1.492</u>	1.494	.003	CHABAZITE	(+)58°	TRIG	Tw.
1.491	<u>1.493</u>	1.497	.006	CLINOPTILOLITE	(+)72°	MCL	El clv pos.
---	<u>1.493</u>	---	.000	ANALCIME	---	CUB	
1.486	<u>1.494</u>	1.496	.010	STELLERITE	(-)38-48°	ORTH	X = a, Y = $\frac{c}{2}$, e $\bar{1}$ clv neg.
1.489	<u>1.494</u>	---	.005	GOBBINSITE	(-)0°	TET	El neg.
1.493	<u>1.496</u>	1.500	.007	DACHIARDITE	(+)69°	MCL	Mimetic tw.
1.495	<u>1.497</u>	1.504	.009	YUGAWARALITE	(+)78°	MCL	Insol in acids. U opt neg.
1.495	<u>1.498</u>	1.502	.007	GOOSECREEKITE	(-)82°	MCL	Y = b, Z:c = 46°.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta, \omega,}{\text{or } \underline{n}}$	γ or ϵ					
1.497	(1.498)	1.499	.002	GONNARDITE	(+)med to large	ORTH	
1.496	1.498	1.504	.008	HEULANDITE	(+)35°	MCL	El clv neg.
1.499	1.500	1.502	.003	WAIKITE	(+)med	MCL	
---	1.500	1.502	.002	GARRONITE	(+)0°	TET	
1.499	1.500	1.501	.002	MERLINOITE	(-)56°	ORTH	X = \underline{b} , Z = \underline{a} , e $\overline{1}$ pos.
1.498	(1.501)	1.502	.004	WAIKITE	(-)75°	MCL	X ~ \underline{b} , Z ~ \underline{c} , U biax pos.
1.500	1.501	1.504	.004	WELLSITE	(+)39°	MCL	Complex tw.
1.494	1.502	1.507	.013	STILBITE	(-)30-50°	MCL	Y = \underline{b} , X:a = 2-9°, e $\overline{1}$ neg. Cruciform penet tw.
1.498	1.503	1.506	.008	HARMOTOME	(-)75-82°	MCL	Z = \underline{b} , Y:c = 28°
---	1.503	1.507	.004	CHABAZITE	(+)0°	TRIG	2V small
1.500	1.503	1.510	.010	PHILLIPSITE	(+)med	MCL	Tw.
1.501	1.504	1.509	.008	HARMOTOME	(+)78°	MCL	Cruciform tw.
1.504	1.505	1.507	.003	MESOLITE	(+)large	MCL	
---	1.505	---	.000	ANALCIME	---	CUB	Diff dec by acids. Cs ₂ O 14.9%.
1.499	1.506	---	.007	MAZZITE	(-)0°	HEX	
1.505	1.507	1.507	.002	GARRONITE	(-)0°	ORTH, ps tet	

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta}{\text{or } \frac{\omega}{n}}$	γ or ϵ					
1.504	<u>1.507</u>	1.515	.011	HEULANDITE	(-)10-40°	MCL	El clv neg.
---	<u>1.507</u>	---	.000	POLLUCITE	---	CUB	F diff.
1.505	<u>1.509</u>	1.509	.004	COWLESITE	(-)30°	ORTH	X = a, Y = <u>b</u> , e $\bar{1}$ pos.
1.505	<u>1.509</u>	1.513	.008	HARMOTOME	~ 90°	MCL	Cruciform tw.
1.505	<u>1.509</u>	1.511	.006	PHILLIPSITE	(-)med	MCL	Penet tw.
1.508	<u>1.510</u>	1.513	.005	GONNARDITE	(+)large	ORTH	
1.502	<u>1.510</u>	---	.008	LEVYNE	(-)0°	TRIG	
1.508	<u>1.511</u>	1.523	.015	BREWSTERITE	(+)55°	MCL	El clv neg.
1.508	<u>1.511</u>	1.512	.004	GARRONITE	(-)0-30°	ORTH	
1.500	<u>1.511</u>	1.513	.013	EPISTILBITE	(-)40°	MCL	X = <u>b</u> , Z:c = 9°
1.504	<u>1.512</u>	1.516	.012	LAUMONTITE	(-)44°	MCL	Y = <u>b</u> , Z:c = 30°
1.511	<u>1.513</u>	1.518	.007	THOMSONITE	(+)75°	ORTH	X = a, Z = <u>b</u>
1.512	<u>1.515</u>	1.517	.005	COWLESITE	(-)44-53°	ORTH	X = a, Z = <u>c</u> , e $\bar{1}$ pos.
1.512	<u>1.515</u>	1.515	.003	GARRONITE	(-)0°	ORTH, ps tet	
1.512	<u>1.516</u>	1.518	.006	GISMONDINE	(-)large	MCL	X = <u>b</u> , Y ~ <u>c</u> , Tw.
1.512	<u>1.518</u>	1.519	.007	SCOLECITE	(-)35°	MCL	Z = <u>b</u> , X:c = 18°, e $\bar{1}$ neg.
1.515	<u>1.518</u>	---	.003	CHABAZITE	(-)0°	TRIG	Anom biax.

Table 38. Zeolite group (continued)

Refractive index			Biref	NAME	2V	System	Remarks
α or ϵ	$\frac{\beta}{\text{or } \underline{n}}$, $\frac{\omega}{\text{or } \underline{n}}$	γ or ϵ					
1.513	<u>1.524</u>	1.525	.012	LAUMONTITE	(-)25°	MCL	$Y = \underline{b}$, $Z:c$ = 20-30°.
1.523	<u>1.525</u>	1.532	.009	THOMSONITE	(+)48°	ORTH	$X = \underline{a}$, $Z = \underline{b}$
---	<u>1.525</u>	---	.000	POLLUCITE	---	CUB	
1.522	<u>1.528</u>	1.530	.008	GISMONDINE	(-)15- 90°	MCL	$X = \underline{b}$, $Y \sim$ \underline{c} . \overline{Tw} .
1.530	<u>1.533</u>	1.542	.012	THOMSONITE	(+)52°	ORTH	$X = \underline{a}$, $Z = \underline{b}$
1.538	<u>1.543</u>	1.548	.010	GISMONDINE	(-)15- 90°	MCL	$X = \underline{b}$, $Y \sim$ \underline{c} . \overline{Tw} .
1.541	<u>1.553</u>	1.557	.016	EDINGTONITE	(-)54°	ORTH	$X = \underline{c}$, $Z =$ \underline{a} , $e\overline{l}$ neg.

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INDEX

INDEX

Abbreviations, list of --- 21-22
Abernathyite --- 96, 363, 364
"Absorption ovaloid" --- 20
Acetamide --- 85
Acknowledgements --- 2
Acmite --- 185, 287, 291, 294, 377,
383, 384
Acmite-augite --- 284, 288
Actinolite --- 250, 254, 262, 318, 324,
325
Adamite --- 184, 283, 288
Adelite --- 178
Admontite --- 214
Aegirine --- 378
Aegirine-augite --- 377, 383, 384
Aenigmatite --- 191, 193
Aeschynite --- 57, 203
Aeschynite-(Y) --- 55, 202
Afghanite --- 65, 341
Afwillite --- 153
Agardite --- 74, 75
Agrellite --- 233
Agrinierite --- 306
Anlfeldite --- 285
Ajkaite --- 36
Ajoite --- 147
Akatoreite --- 174
Akdalaite --- 112
Akermanite --- 71, 362
Akrochordite --- 168
Aksaite --- 215
Alabandite --- 60
Alamosite --- 304
Albite --- 136, 221, 350, 351, 352
Aldzhanite --- 151, 246
Alforsite --- 109, 330, 332
Allactite --- 289
Allanite --- 39, 43, 44, 46, 175, 184,
188, 276, 283, 286, 291, 295,
299, 347, 348, 349, 350
Alleghanyite --- 267, 290, 358, 360,
361
Allophane --- 32, 33, 35
Alluaudite --- 181, 187, 191
Almandine --- 48, 49, 355, 356, 357
Almandine-pyrope --- 47, 356
Almandine-spessartine --- 357
Alstonite --- 265
Altaite --- 60
Althausite --- 148
Aluminite --- 125

Aluminocopiapite --- 138, 343
 Alumohydrocalcite --- 226, 229, 237
 Alumopharmacosiderite --- 37
 Alunite --- 67, 68, 70, 316, 317
 Alunogen --- 125, 126
 Alvanite --- 271
 Amakinite --- 75
 Amarantite --- 240, 242
 Amarillite --- 141
 Amblygonite --- 243
 Ameghinite --- 220
 Amesite --- 68, 149
 Amicite --- 213, 394, 398
 Aminoffite --- 103
 Ammonioborite --- 128
 Ammoniojarosite --- 114, 316, 317
 Alalcime --- 32, 33, 34, 85, 212, 394,
 397, 398, 399
 Anandite --- 195, 365, 369
 Anapaite --- 152
 Anatase --- 122
 Ancyrite --- 272, 278, 282
 Andalusite --- 157, 162, 167, 179, 251,
 255
 Andersonite --- 65
 Andesine --- 140, 225, 226, 350, 353
 Andradite --- 50, 51, 355, 357
 Andradite-rossular --- 49, 357
 Andradite-spessartine --- 51, 357
 Andremerite --- 183
 Andrews site --- 193
 Angelellite --- 202
 Anglesite --- 196, 337
 Anhydrite --- 145
 Ankerite --- 112, 346
 Annabergite --- 162, 260
 Annite --- 107, 265, 365, 369
 Anorthite --- 236, 351, 354
 Anorthoclase --- 219, 223, 350, 351,
 352
 Antarcticite --- 90
 Anthoinite --- 49
 Anthonyite --- 241
 Anthophyllite --- 155, 160, 249, 318,
 321
 Antigorite --- 229, 233, 242
 Antlerite --- 183
 Apachite --- 257
 Apatite-britholite --- 109, 330, 332
 Aphanthalite --- 63, 64
 Apjohnite --- 211
 Aplowite --- 221, 224
 Aragonite --- 267, 272, 334
 Arcanite --- 129

Archerite --- 87
 Arctite --- 93
 Ardennite --- 183, 186
 Arfvedsonite --- 273, 319
 Argentojarosite --- 117, 316, 317
 Aristarainite --- 130
 Armangite --- 118
 Armenite --- 227, 373
 Armstrongite --- 230
 Arrangement of data in tables --- 22-24
 Arrojadite --- 264
 Arsendescloizite --- 307
 Arseniosiderite --- 117, 299, 301
 Arsenobismite --- 50
 Arsenoclasite --- 294
 Arsenocrandallite --- 344, 345
 Arsenolite --- 46
 Arsenuranospathite --- 223
 Arthurite --- 188, 287
 Artinite --- 221
 Asbecasite --- 117
 Ashanite --- 204, 312
 Ashcroftine --- 65, 138
 Astrophyllite --- 171, 178, 184
 Atacamite --- 299
 Atelestite --- 201
 Attakolite --- 165
 Aubertite --- 211
 Augelite --- 145
 Augite --- 179, 181, 377, 378, 381, 382
 Aurichalcite --- 285
 Austinite --- 186, 189
 Autunite --- 93, 95, 232, 335, 336
 Avogadrite --- 208
 Axial angle, see Optic angle
 Azoprobeite --- 193
 Azovskite --- 46
 Azurite --- 187
 Babefphite --- 71
 Babingtonite --- 179, 182
 Baddeleyite --- 310, 313
 Bafertisite --- 294, 297
 Bahianite --- 299
 Bakerite --- 251, 258
 Balangeroite --- 267
 Balavinskite --- 225
 Balipholite --- 239
 Balyakinite --- 201, 310
 Banalsite --- 144, 351, 353
 Bandyte --- 108
 Bannisterite --- 236, 244
 Baotite --- 79
 Bararite --- 84
 Baratovite --- 168

Barbertonite --- 91
 Barbosalite --- 291
 Barentsite --- 210
 Baricite --- 143
 Bario-orthojoaquinite --- 183
 Bariopyrochlore ("Pandaite") --- 54,
 375, 376
 Barite --- 157, 173, 337
 Barium-pharmacosiderite --- 43, 281
 Barnesite --- 305
 Barrerite --- 211, 394, 397
 Barringtonite --- 126
 Bartelkeite --- 301
 Barylite --- 172, 274
 Barysilite --- 119, 120
 Barytocalcite --- 269
 Barytolamprophyllite --- 183, 186
 Basaluminite --- 133, 218
 Bassanite --- 66, 67, 142
 Bassetite --- 232, 243, 363, 364
 Bastnaesite --- 74, 75, 76
 Batisite --- 181
 Bavenite --- 146, 147
 Bayerite --- 145
 Bayldonite --- 199
 Bayleyite --- 213
 Baylissite --- 127
 Bazirite --- 73
 Bazzite --- 100, 102
 Bearsite --- 214
 Beaverite --- 117, 316, 317
 Becke line --- 4-5
 colored --- 4
 spurious --- 5
 technique --- 4-5
 Becquerelite --- 293, 295
 Behierite --- 80
 Benoitite --- 224
 Beidellite --- 87, 89, 90, 95, 222,
 228, 237
 Bellingerite --- 197
 Belovite --- 105, 330, 332
 Belyankinite --- 290
 Bementite --- 101, 104, 257
 Benitoite --- 76
 Benstonite --- 108, 109
 Bentorite --- 63
 Beraunite --- 189, 283
 Berborite --- 94
 Bergenite --- 271
 Berlinite --- 65
 Bermanite --- 281
 Bertossaite --- 252
 Bertrandite --- 241

Beryl --- 92, 93, 95, 96, 98
 Beryllite --- 226
 Beryllonite --- 227
 Berzeliite --- 43, 46
 Betafite --- 51, 52, 53, 375, 376
 Betpakdalite --- 193
 Beudantite --- 118, 304, 337, 338
 Beusite --- 172, 175
 Beyerite --- 120
 Bianchite --- 213
 Bicchulite --- 40
 Bideauxite --- 56
 Bieberite --- 211
 Bijvoetite --- 161
 Bikitaite --- 218
 Bilinite --- 214
 Billietite --- 290, 296
 Bindheimite --- 44, 47, 50, 51, 389,
 390
 Biotite --- 96, 99, 104, 240, 245, 258,
 365, 368, 369
 Biphosphammite --- 88
 Birefringence --- 14-15
 Biringuccite --- 223
 Birnessite --- 111
 Birunite --- 135
 Bisbeeite --- 153
 Bischofite --- 131
 Bismite --- 205, 313
 Bismoclite --- 120
 Bismutite --- 55, 57, 310, 312
 Bismutoferrite --- 304
 Bismutomicrolite ("Westgrenite") ---
 53, 375, 376
 Bismutostibiconite --- 54, 389, 390
 Bismutotantalite --- 205
 Bityite --- 259, 365, 369
 Bixbyite --- 59
 Bjarebyite --- 172, 180
 Blakeite --- 56
 Blixite --- 200
 Bloedite --- 212
 Bobierite --- 133, 141
 Boehmite --- 160, 163
 Boggildite --- 125
 Boleite --- 119
 Bolivarite --- 34
 Boltwoodite --- 273
 Bonattite --- 145
 Bonshtedtite --- 230
 Boothite --- 127, 210
 Boracite --- 165
 Borax --- 209
 Borcarite --- 258

Borickite --- 37, 39, 42
 Bornemanite --- 170, 172
 Bornite --- 55
 Bostwickite --- 293
 Botallackite --- 191
 Botryogen --- 135, 139
 Bournonite --- 206
 Boussingaultite --- 126
 Boyleite --- 220
 Brabantite --- 173
 Brackebuschite --- 204
 Bradleyite --- 224
 Braitschite --- 71
 Brammallite --- 232, 365, 367
 Brandtite --- 175
 Brannerite --- 52, 55, 59
 Brannockite --- 92, 373, 374
 Brassite --- 139
 Braunite --- 122
 Brazilianite --- 151
 Bredigite --- 177
 Brenkite --- 238
 Brewsterite --- 132, 394, 400
 Brianite --- 242
 Britholite --- 44, 46, 49, 112, 116,
 289, 330, 331, 332
 Britholite-(Y) --- 76, 77, 330, 331,
 332
 Brochantite --- 289
 Brockite --- 73
 Bromargyrite --- 56, 57
 Bromellite --- 75
 Bronzite --- 169, 266, 267, 273, 377,
 380, 381
 Brookite --- 205
 Brownmillerite --- 306
 Brucite --- 67, 68, 144
 Bruegggenite --- 292
 Brugnatellite --- 89, 91
 Brunogeierite --- 387
 Brushite --- 139
 Buchwaldite --- 152, 244
 Buddingtonite --- 136, 351, 352
 Buergerite --- 112, 391, 392
 Buetschliite --- 96, 97
 Bukovskyite --- 144, 146, 231, 234
 Bultfonteinite --- 148
 Bunsenite --- 58
 Burangaite --- 251
 Burbankite --- 99, 101
 Burkeite --- 212
 Burkhardtite --- 298
 Burtite --- 40, 71
 Bustamite --- 266, 273, 275

Butlerite --- 265
 Buttgenbachite --- 76
 Bystromite --- 78, 79, 117, 118
 Bytownite --- 143, 144, 230, 231, 233,
 351, 354
 Cacozenite --- 68, 69
 Cadmiumoxide --- 59
 Cadwaladerite --- 35
 Cafarsite --- 53
 Cafetite --- 308
 Cannite --- 72
 Calciborite --- 259
 Calcioferrite --- 93
 Calciouranoite --- 44, 297
 Calciovolborthite --- 199, 200, 306
 Calcite --- 105, 107, 110, 339
 Calcium catapleiite --- 69
 Calcium ferri-phosphate --- 39
 Calcjarlite --- 124
 Calclacite --- 128
 Calcurmolite --- 295
 Calderite --- 355
 Calderite-andradite --- 51, 357
 Calderite-grossular --- 357
 Caledonite --- 299
 Calkinsite --- 260
 Callaghanite --- 259
 Calomel --- 80
 Calumetite --- 271
 Calzirtite --- 80, 81
 Campigliaite --- 255
 Canasite --- 223
 Canavesite --- 129
 Cancrinite --- 87, 88, 341, 342
 Cappelenite --- 112
 Caracolite --- 286
 Carboborite --- 224
 Carbocernaite --- 267
 Carboirite --- 181, 182
 Carbonate-cyanotrichite --- 156
 Carbonate-fluorapatite --- 37, 38, 40,
 100, 103, 247, 248, 330, 331
 Carbonate-hydroxylapatite --- 97, 330,
 331
 Carletonite --- 87
 Carlfriesite --- 308
 Carlhintzeite --- 124
 Carminite --- 200
 Carnallite --- 126
 Carnotite --- 301, 302
 Carobbiite --- 30
 Carpholite --- 238, 249
 Caryinite --- 77, 190
 Caryocerite --- 45

Caryopilite --- 249
 Cassidyite --- 161
 Cassiterite --- 80
 Catapleiite --- 148
 Cavansite --- 139
 Caysichite --- 245
 Cebollite --- 150
 Celadonite --- 103, 106, 249, 254, 261,
 365, 366, 368, 369
 Celestite --- 155, 337
 Celsian --- 147, 149, 230, 235, 238,
 351, 353, 354
 Central focal masking --- 507
 Central illumination, technique --- 4-5
 Cerite --- 77, 193
 Cerotungstite --- 303
 Cerussite --- 308, 334
 Cesanite --- 92
 Cesbronite --- 198
 Cesium kupletskite --- 179
 Cnabazite --- 62, 63, 64, 84, 85, 87,
 127, 128, 129, 212, 213, 394,
 396, 397, 398, 399, 400
 Chalcanthite --- 222, 342
 Chalcoalumite --- 135
 Chalcocyanite --- 282
 Chalcomenite --- 282
 Chalconatronite --- 136
 Chalcophyllite --- 99
 Chalcophanite --- 123
 Chalcophyllite --- 101
 Chalcosiderite --- 297
 Challantite --- 136
 Chambersite --- 183
 Chamosite --- 102, 104, 108, 154, 156,
 252, 257, 263, 269
 Changbaiite --- 82
 Chantalite --- 105
 Chapmanite --- 303
 Charlesite --- 85
 Charoite --- 141, 226
 Chavesite --- 154
 Chelkarite --- 137, 225
 Chenevixite --- 196, 198
 Cheralite --- 189
 Chernovite --- 76, 77
 Chernykhite --- 270, 365, 369
 Chervetite --- 313
 Chessexite --- 125
 Chesterite --- 250
 Chevkinite --- 51, 52, 300, 307
 Chiavennite --- 150
 Childrenite --- 262, 268
 Chinglusuite --- 38

Cniolite --- 84
 Cnkalovite --- 139, 140
 Chloraluminite --- 91
 Chlorapatite --- 105, 106, 257, 264,
 330, 331, 332
 Chlorargyrite --- 54
 Chloritoid --- 176, 180, 276, 281
 Chlormagaluminite --- 66
 Chlormanganokalite --- 68
 Chlorocalcite --- 218
 Chloromagnesite --- 107
 Chlorophoenicite --- 171, 271
 Chloroxiphite --- 311
 Choloalite --- 53
 Chondrodite --- 150, 153, 156, 358, 359
 Chromatite --- 78
 Chromdravite --- 114, 391, 393
 Chromite --- 54, 55, 387, 388
 Chrysoberyl --- 185
 Chrysocolla --- 240
 "Chrysocolla" --- 62
 Chrysolite --- 168, 169, 267, 275, 371
 Chrysotile --- 139, 144
 Chudobaite --- 243
 Chukhrovite --- 31
 Chukhrovite-(Ce) --- 31
 Churchite --- 151, 156
 Cinnabar --- 82
 Claraite --- 112
 Claringbullite --- 114
 Clarkeite --- 309
 Claudetite --- 198
 Cliffordite --- 55
 Clinobisvanite --- 205, 314
 Clinochalcomenite --- 280
 Clinochlore --- 67, 68, 70, 94, 99,
 141, 145, 147, 150, 233, 246
 Clinoclase --- 299
 Clinoenstatite --- 163, 377, 378, 379
 Clinohedrite --- 264
 Clinoholmquistite --- 248, 318, 322
 Clinohumite --- 153, 159, 169, 176,
 358, 359, 360
 Clinohypersthene --- 169, 377, 378, 380
 Clinokurchatovite --- 265
 Clinophosinaite --- 142
 Clinoptilolite --- 129, 127, 210, 211,
 394, 396, 397, 398
 Clinotyrolyte --- 270
 Clinozoisite --- 174, 177, 347, 348
 Clintonite --- 261, 365, 366, 369
 Coalingite --- 95, 97, 101
 Cobaltkoritnigite --- 166
 Cobaltomenite --- 281

Cobalt-zippeite --- 290
 Coconinoite --- 237
 Coeruleolactite --- 67
 Coesite --- 38, 148
 Coffinite --- 41, 45, 51
 Colemanite --- 148
 Collinsite --- 158
 Color --- 20-21
 Colquiriite --- 84
 Combeite --- 38, 69, 96
 Comblainite --- 108
 Compreignacite --- 293
 Congolite --- 113
 Conichalcite --- 77, 191, 288, 296
 "Connarite" --- 95
 Connellite --- 75, 76
 Cookeite --- 146, 147
 Copiapite --- 137, 343
 Coquimbite --- 65, 66
 Corderoite --- 59
 Cordierite --- 136, 138, 140, 220, 222,
 225, 227
 Cordylite --- 113
 Corkite --- 118, 302, 337, 338
 Cornetite --- 295
 Cornwallite --- 192, 299
 Corundum --- 113, 114, 288
 Cotunnite --- 203
 Coulsonite --- 387
 Covellite --- 62
 Cowlesite --- 216, 217, 394, 400
 Crandallite --- 69, 70, 344, 345
 Crandallite(?) --- 151
 Creaseyite --- 185
 Crednerite --- 123
 Creedite --- 210
 Cristobalite --- 33, 85
 Crocoite --- 204
 Cronstedtite --- 114, 293
 Crossite --- 256, 320, 328
 Cryolite --- 124
 Cryolithionite --- 30
 Cryptohalite --- 30
 Cryptomelane --- 60
 Crystallographic directions, relations
 to optical directions --- 19-20
 Cumengite --- 119
 Cumingtonite --- 154, 158, 165, 168,
 318, 322, 323
 Cuprite --- 60
 Cuprocopiapite --- 145, 343
 Cuprorivaite --- 101, 248
 Cuprosklodowskite --- 263
 Cuprospinel --- 387

Cuprotungstite --- 201, 309
 Curienite --- 305
 Curite --- 309
 Cuspidine --- 148
 Cuzticite --- 119
 Cyanochroite --- 128
 Cyanophillite --- 262
 Cyanotrichite --- 153
 Cyclo wollastonite --- 152
 Cymrite --- 99, 246
 Cyrilovite --- 115
 Dachiardite --- 127, 129, 394, 398
 Dalyite --- 238
 Danalite --- 45, 46
 Danburite --- 250
 Dannemorite --- 268, 318, 323
 D'ansite --- 33
 Darapiosite --- 94, 373, 374
 Darapskite --- 211
 Datolite --- 259
 Daubreeite --- 117
 Davidite --- 55, 58
 Davisonite --- 97
 Davreuxite --- 267
 Davyne --- 65, 341
 Dawsonite --- 223
 Deerite --- 193, 194, 297
 Defernite --- 231
 Delhayelite --- 221
 Dellaite --- 260
 Delrioite --- 296
 Delvauxite --- 42, 44
 Demesmaeckerite --- 195
 Denningite --- 79, 196
 Derbylite --- 82, 205
 Derriksite --- 298
 Desautelsite --- 92
 Descloizite --- 203, 311
 Despujolsite --- 72
 Devilline --- 256
 Diaboleite --- 118
 Diadochite --- 39, 153, 155
 Diamond --- 59
 Diaspore --- 176, 178
 Dickinsonite --- 163, 165
 Dickite --- 143
 Dietrichite --- 127
 Dietzeite --- 297
 Diopside --- 167, 169, 170, 171, 173,
 377, 378, 380, 381
 Dioptase --- 72, 163
 Dispersion coloration technique --- 5-7
 Dispersion colors, central focal
 masking --- 8

Dispersion of refractive index --- 6
 Dispersion staining technique --- 5
 Dixenite --- 80, 118
 Dolerophanite --- 193, 295
 Dolomite --- 107, 109, 110, 346
 Domeykite --- 60
 Donnayite --- 255
 Dorfmanite --- 125
 Douglasite --- 63
 Downeyite --- 78
 Dravite --- 102, 105, 107, 110, 391,
 392
 Dresserite --- 241
 Dreyerite --- 119
 Dufrenite --- 192, 194, 195
 Dufrenoyite --- 206, 314
 Duftite --- 307
 Dugganite --- 118
 Dunamelite --- 200, 309
 Dumontite --- 195, 196
 Dumortierite --- 266, 271, 280
 Dundasite --- 277
 Durangite --- 265
 Dussertite --- 117, 344, 345
 Duttonite --- 197
 Dypingite --- 132
 Dzhaliindite --- 44
 Eakerite --- 147
 Earlandite --- 136
 Ecdemite --- 121
 Eckermannite --- 249, 255, 320, 328
 Edenite --- 255, 267, 318, 324, 325
 Edingtonite --- 226, 394, 401
 Eglestonite --- 59
 Egueite --- 41
 Eifelite --- 66, 373, 374
 Eitelite --- 97
 Ekanite --- 93, 98
 Ekaterinite --- 233
 Elbaite --- 102, 103, 391, 392
 Ellestadite --- 105, 330, 332
 Elongation, sign of --- 20
 Elpasolite --- 30
 Elpidite --- 143, 230
 Elyite --- 305
 Embreyite --- 312
 Emeleusite --- 240, 373, 374
 Emmonsite --- 308
 Enargite --- 206
 Englishite --- 232
 Enstatite --- 162, 167, 377, 378, 379
 Eosphorite --- 257
 Ephesite --- 247, 365, 366, 368
 Epididymite --- 139, 224

Epidote --- 279, 283, 286, 288, 291,
 347, 348, 349
 Epi-ianthinite --- 291
 Epistilbite --- 216, 394, 400
 Epistolite --- 257
 Epsomite --- 208, 209
 Ericaite --- 286
 Eriochalcite --- 170
 Erionite --- 62, 63, 125, 394, 396
 Ernstite --- 275
 Erythrite --- 159, 165, 261
 Erythrosiderite --- 185
 Esperite --- 288
 Ettringite --- 84, 85
 Euchlorine --- 151
 Euchroite --- 173
 Euclase --- 163
 Eucryptite --- 67
 Eudialyte --- 39, 68, 69, 70, 98, 101,
 104
 Eudidymite --- 139
 Eulite --- 187, 285, 287, 377, 378, 383
 Eulytite --- 53, 119
 Euxenite --- 53, 57, 202
 Evansite --- 31, 32, 34
 Eveite --- 176
 Evenkite --- 131
 Ewaldite --- 103
 Extinction angle --- 20
 Extinction position --- 18
 Extraordinary principle refractive
 index --- 9
 Extraordinary ray --- 9
 Eylettersite --- 102, 344, 345
 Ezcurrite --- 215
 Eztlite --- 201, 309
 Fabianite --- 252
 Faheyite --- 71
 Fairchildite --- 88
 Fairfieldite --- 158, 161
 Farringtonite --- 139
 Fassaite --- 172, 178, 182, 377, 381,
 382, 383
 Faujasite --- 32, 33, 63, 127, 394, 397
 Fayalite --- 297, 298, 299, 371, 372
 Fedorite --- 220
 Fedorovskite --- 248
 Felsöbanyaite --- 133
 Fenaksite --- 227
 Fergusonite --- 53, 56
 Fergusonite-beta --- 311
 Fergusonite-beta-(Ce) --- 56
 Fermorite --- 105, 330, 332
 Fernandinite --- 199, 307

Ferrarisite --- 145
 Ferri-annite --- 271, 279, 365, 369
 Ferricopiapite --- 140, 343
 Ferridravite --- 114, 391, 393
 Ferrierite --- 126, 129, 394, 396, 398
 Ferrimolybdite --- 181, 190, 193
 Ferrinatrite --- 66
 Ferripyrophyllite --- 266
 Ferrisicklerite --- 289, 294
 Ferritungstite --- 54
 Ferro-actinolite --- 318
 Ferroaugite --- 177, 183, 377, 382, 383
 Ferroaxinite --- 266, 273
 Ferrobustamite --- 257
 Ferrocapholite --- 252
 Ferrocolumbite --- 312
 Ferro-ferri-tschermakite --- 272, 319,
 326
 Ferro-gedrite --- 166, 169, 171, 268,
 276, 318, 322
 Ferro-glaucophane --- 261, 320, 328
 Ferronexahydrite --- 213
 Ferro-hornblende --- 269, 319, 325
 Ferrohortonolite --- 295, 296, 371, 372
 Ferrohypersthene --- 281, 377, 382
 Ferro-richterite --- 274, 319, 327
 Ferrosalite --- 176, 377, 382
 Ferrotschermakite --- 167, 264, 319,
 325
 Ferrotychite --- 36
 Ferrowyllieite --- 171
 Ferruccite --- 124
 Fersmanite --- 302
 Fersmite --- 199, 200
 Fervanite --- 311
 Fibroferrite --- 65, 133, 137
 Fichtelite --- 145
 Fiedlerite --- 307
 Fillowite --- 73, 168
 Finnemanite --- 121
 Flagstaffite --- 133
 Fleischerite --- 76
 Flinkite --- 191
 Florencite --- 72, 73, 74, 75, 344, 345
 Florencite-(La) --- 74
 Fluckite --- 155
 Fluellite --- 129
 Fluoborite --- 86, 88, 90, 92, 229
 Fluocerite --- 69, 99
 Fluorapatite --- 101, 102, 103, 330,
 331
 Fluorapophyllite --- 65, 137
 Fluorite --- 30, 31
 Foggite --- 152

Formanite --- 55
 Fornacite --- 201
 Forsterite --- 161, 165, 371
 Foshagite --- 149
 Foshallasite --- 223
 Fourmarierite --- 300, 302
 Fraipontite --- 247
 Francevillite --- 301, 304
 Francoanellite --- 64
 Frankdicksonite --- 32
 Franklinite --- 58, 59, 387, 389
 Franzinite --- 64, 341
 Fresnoite --- 113
 Freudenbergite --- 81
 Friedelite --- 105, 107, 257
 Frolovite --- 145
 Frondelite --- 300
 Fukalite --- 151
 Furongite --- 229
 Gadolinite --- 43, 47, 49, 173, 179,
 189, 192
 Gagarinite --- 63
 Gageite --- 283
 Gahnite --- 47, 48, 49, 387, 388
 Gaidonnayite --- 239
 Gainesite --- 69
 Gaitite --- 175, 181, 275
 Galaxite --- 50, 51, 387, 388
 Galeite --- 62
 Galena --- 61
 Gamagarite --- 199
 Ganomalite --- 79, 198
 Ganophyllite --- 239, 244
 Garrelsite --- 250
 Garronite --- 64, 86, 87, 217, 394,
 399, 400
 Gaspeite --- 115, 116, 339, 340
 Gatumbaite --- 249
 Gaudefroyite --- 78
 Gaylussite --- 217
 Gearksutite --- 208
 Gedrite --- 163, 247, 259, 264, 318,
 321
 Genlenite --- 107, 362, 363
 Geikielite --- 121
 Genthelvite --- 45
 Georgeite --- 38
 Georgiadesite --- 201
 Gerasimovskite --- 294
 Gerhardtite --- 176, 277
 Gerstleyite --- 200, 306
 Gerstmannite --- 266
 Getchellite --- 205
 Gianellaite --- 54

Giannetite --- 166
 Gibbsite --- 143, 146
 Gilalite --- 251
 Gillespite --- 100
 Giniite --- 293
 Ginorite --- 134
 Girdite --- 313
 Gismondine --- 217, 220, 223, 394, 400,
 401
 Gittinsite --- 283
 Giuseppettite --- 63, 341
 Glass --- 34, 35, 37, 39, 41
 Glass (obsidian) --- 32
 Glass, "Lechatelierite" --- 31
 Glauberite --- 221
 Glaucocroite --- 278, 370
 Glaucokeirinite --- 136, 222
 Glauconite --- 95, 97, 99, 101, 236,
 241, 246, 251, 365, 366,
 367, 368, 369
 Glaucophane --- 252, 320, 328
 Glaukosphaerite --- 296
 Glucine --- 141, 228
 Glushinskite --- 220
 Gmelinite --- 63, 84, 127, 210, 394,
 396, 397
 Gobbinsite --- 85, 394, 398
 Goedkenite --- 168
 Goethite --- 311, 313
 Goethite ("Limonite") --- 53
 Goldichite --- 150
 Goldmanite --- 49, 50, 355, 357
 Gonnardite --- 130, 132, 394, 399, 400
 Gonyerite --- 106, 263
 Goosecreekite --- 214, 394, 398
 Gorceixite --- 69, 70, 344, 345
 Gordonite --- 138, 237
 Gormanite --- 259
 Goslarite --- 209, 210
 Götzenite --- 143, 162, 166
 Goudeyite --- 74
 Gowerite --- 130
 Goyazite --- 70, 71, 344, 345
 Graemite --- 198
 Graftonite --- 173, 179
 Grandidierite --- 246, 251
 Grantsite --- 305
 Graphite --- 118
 Gratonite --- 123
 Greenalite --- 41, 42
 Greenockite --- 81, 82
 "Grenzdunkfeld" technique --- 6
 Grimaldiite --- 119, 120
 Grimselite --- 97

Griphite --- 40, 42
 Grossular --- 44, 46, 355, 356
 Grossular-andradite --- 48, 356
 Grossular-spessartine --- 47, 356
 Grossular-uvarovite --- 48, 356
 Grothine --- 141
 Grunerite --- 263, 266, 274, 318, 323
 Guerinite --- 235
 Guildite --- 156
 Guilleminite --- 292
 Gunningite --- 150
 Gutsevichite --- 37, 39
 Gypsum --- 134
 Gyrolite --- 90, 225
 Hagendorfite --- 284
 Haidingerite --- 149
 Haiweeite --- 231, 234
 Halite --- 36
 Hallimondite --- 197
 Halloysite --- 36, 227
 Halotrichite --- 212
 Halurgite --- 139
 Hambergite --- 146, 148
 Hancockite --- 296, 347, 350
 Hanksite --- 84
 Hannayite --- 218
 Haradaite --- 178, 279
 Harbortite --- 152
 Hardystonite --- 107, 362, 363
 Harkerite --- 40, 41
 Harmotome --- 131 132, 215, 394, 399,
 400
 Harstigite --- 169
 Hartite --- 141
 Hastingsite --- 276, 282, 319, 326
 Hatchettite --- 63, 132, 134
 Hatrurite --- 111, 280
 Hauckite --- 71
 Hauerite --- 60
 Hausmannite --- 122
 Hauyne --- 33, 35, 386
 Hawleyite --- 47
 Hedenbergite --- 181, 186, 377, 378,
 382, 383
 Hedyphane --- 79, 119, 330, 333
 Heidornite --- 147
 Heinrichite --- 97, 242, 335, 336
 Hellandite --- 46, 262
 Hellyerite --- 215
 Helmutwinklerite --- 191
 Helvite --- 44, 45
 Hematite --- 123, 313
 Hematolite --- 111
 Hemihedrite --- 312

Hemimorphite --- 153
 Hendersonite --- 306
 Hendricksite --- 260, 270, 365, 369
 Henritermierite --- 77, 187, 355
 Hercynite --- 47, 50, 387, 388
 Herderite --- 244
 Herschelite --- 63, 84, 394, 396
 Hetaerolite --- 121
 Heterosite --- 300
 Heulandite --- 129, 130, 131, 394, 398,
 399, 400
 Hewettite --- 310
 Hexahydrite --- 209
 Hexahydroborite --- 130
 Heyite --- 202
 Hibonite --- 114, 116
 Hidalgoite --- 75, 76, 337, 338
 Hieratite --- 30
 Hilairite --- 98
 Hilgardite --- 157
 Hillebrandite --- 242, 244
 Hinsdalite --- 73, 167, 337, 338
 Hiortdahlite --- 160, 254, 260
 Hisingerite --- 31, 35, 37, 38, 42,
 210, 234
 Hodgkinsonite --- 284
 Hoegbomite --- 115, 117
 Hoernesite --- 144
 Hohmannite --- 254
 Holdenite --- 188
 Hollandite --- 123
 Holmquistite --- 250, 255, 318, 321
 Holtedahlite --- 91
 Holtite --- 281, 287
 Homilite --- 179
 Homilite (altered) --- 41, 156, 161
 Hopeite --- 235, 240
 Hortonolite --- 289, 291, 292, 371, 372
 Howieite --- 279
 Howlite --- 240
 Hsianghualite --- 39
 Huanghoite --- 113
 Huebnerite --- 203
 Huegelite --- 79, 197
 Huemulite --- 282
 Humberstonite --- 84
 Humboldtine --- 142
 Humite --- 151, 155, 157, 162, 358,
 359, 360
 Hungchaoite --- 211
 Huntite --- 100
 Hureaulite --- 255, 259, 264
 Hurlbutite --- 241
 Hutchinsonite --- 315

Huttonite --- 42, 193, 197
 Hyalophane --- 224, 351, 353
 Hyalosiderite --- 279, 282, 286, 371,
 372
 Hyalotekite --- 199
 Hydrated basaltic glass --- 32, 36
 Hydroastrophyllite --- 279
 Hydrobasaluminite --- 126, 210
 Hydroboracite --- 137
 Hydrocalumite --- 226
 Hydrocerussite --- 120
 Hydrochlorborite --- 131
 Hydrodresserite --- 239
 Hydroglauberite --- 214
 Hydrogrossular --- 42, 43, 355, 356
 Hydrohetaerolite --- 121
 Hydrononessite --- 101
 Hydromagnesite --- 135, 137
 Hydronium jarosite --- 115, 316, 317
 Hydrophilite --- 151
 Hydrotalcite --- 87
 Hydrotungstite --- 303
 Hydrous U sulfate --- 190
 Hydroxyapophyllite --- 66
 Hydroxylapatite --- 104, 330, 331
 Hydroxylbastnaesite --- 76
 Hydroxyllelestadite --- 330, 332
 Hydroxylherderite --- 246
 Hydrozincite --- 283
 Hypersthene --- 277, 377, 381
 Ianthinite --- 301
 Ice --- 30, 62
 Identification of an unknown by its
 optical properties --- 1
 use of tables for --- 24-27
 Idrialite --- 182
 Iimoriite --- 295
 Ikaite --- 222
 Ilesite --- 218
 Ilimaussite --- 73
 Illite --- 237, 365, 367
 Ilmajokite --- 145
 Ilvaite --- 299, 302
 Imandrite --- 151
 Immersion liquids --- 3-4
 change of refractive index with
 temperature --- 3-4
 dispersion strength of --- 4
 Immersion methods --- 2-3
 Inderborite --- 217, 218
 Inderite --- 129
 Index of refraction, see Refractive
 index
 Indialite --- 89

Indicatrices --- 10
 Indigirite --- 128, 213
 Inesite --- 252
 Innelite --- 182
 Interference figures --- 11, 13
 Inyoite --- 215
 Iodargyrite --- 81
 Iowaite --- 89
 Iraquite --- 95
 Irhtemite --- 157
 Iriginite --- 295, 300
 Ironite --- 311
 Isogyre configuration --- 14
 Isogyres, curvature of --- 18
 Isokite --- 149
 Itoite --- 194, 297
 Ivanovite --- 219
 Jacobsite --- 58, 59, 387, 388, 389
 Jadeite --- 159, 163, 167, 170, 377,
 379, 380
 Jagoite --- 118
 Jagowerite --- 172, 272
 Jahnsite --- 260, 273
 Jamborite --- 98
 Jamesite --- 305
 Jarlite --- 124, 208
 Jarosewichite --- 292
 Jarosite --- 114, 116, 295, 316, 317
 Jasmundite --- 75, 112
 Jeanbandyite --- 116
 Jennite --- 228
 Jeremejevite --- 104, 255, 259
 Jimboite --- 190
 Jimthompsonite --- 247
 Jinshajiangite --- 191
 Jixianite --- 57, 375, 376
 Joaquinite --- 177, 188
 Joesmithite --- 187
 Jonachidolite --- 177
 Johannite --- 149, 239
 Johannsenite --- 175, 178, 180, 377,
 378, 381, 382
 Johillerite --- 184
 Johnbaumite --- 108, 330, 332
 Jokokuite --- 216, 342
 Joliotite --- 251, 257
 Jonesite --- 164
 Jouravskite --- 90
 Juanite --- 159
 Julgoldite --- 278, 285, 291, 295
 Julienite --- 66
 Jungite --- 260
 Junitoite --- 165
 Jurbanite --- 209

Kaemmererite --- 95, 237
 Kaersutite --- 272, 276, 282, 319, 326
 Kahlerite --- 250, 335, 336
 Kainite --- 215
 Kainosite --- 270
 Kalborsite --- 65
 Kaliborite --- 135
 Kalicinite --- 211
 Kalinite --- 208
 Kaliophilite --- 88
 Kalipyrochlore --- 51, 52
 Kalistrontite --- 92
 Kalsilite --- 89, 90
 Kamaishilite --- 40, 71, 100
 Kanemite --- 209
 Kanoite --- 177, 377, 382
 Kanonaite --- 181
 Kaolinite --- 229
 Karaptite (= coronene) --- 304
 Karibibite --- 307
 Karlite --- 249
 Karnasurtite --- 99
 Karpatite --- 189
 Karpinskite --- 228, 236
 Kasolite --- 196, 198
 Kassite --- 309
 Katoptrite --- 303
 Kazakovite --- 104
 Keckite --- 272
 Kehoeite --- 35
 Kellyite --- 103, 256
 Kemmlitzite --- 74, 337, 338
 Kempite --- 273
 Kentrolite --- 202
 Kermesite --- 206
 Kernite --- 209
 Kettnerite --- 119
 Keyite --- 192, 297
 Khanneshite --- 100
 Khibinskite --- 277
 Kninite --- 200
 Kidwellite --- 293
 Kieserite --- 136
 Kilchoanite --- 256
 Killalaite --- 253
 Kimzeyite --- 52, 355, 357
 Kingite --- 133, 217
 Kingsmountite --- 234
 Kinoite --- 263
 Kinoshitalite --- 250, 366, 369
 Kirschsteinite --- 272, 279, 370
 Kittatinnyite --- 111
 Kivuite --- 259, 275
 Kladnoite --- 133

Klebelsbergite --- 303
 Kleinite --- 81, 310
 Knorringite --- 48, 355, 356
 Koashvite --- 254
 Kobeite --- 57
 Koechlinite --- 314
 Koenenite --- 65, 66
 Koettigite --- 153, 170
 Kogarkoite --- 62
 Koktaite --- 221
 Kolbeckite --- 238
 Kolfanite --- 302
 Kolicite --- 291
 Komarovite --- 187
 Koninckite --- 37, 41, 105
 Konyaite --- 125
 Koritnigite --- 162
 Kornelite --- 146
 Kornerupine --- 266, 273
 Korshunovskite --- 223
 Korzhinskite --- 160
 Kostylevite --- 149
 Kotoite --- 162
 Kovdorskite --- 223
 Kozulite --- 278, 320, 329
 Kraisslite --- 77
 Krausite --- 160
 Krauskopfite --- 237
 Krautite --- 157
 Kremersite --- 187
 Krinovite --- 179
 Kroehnkite --- 233
 Kryzhanovskite --- 191
 Ktenasite --- 244
 Kulanite --- 174
 Kulkeite --- 227
 Kupletskite --- 274
 Kuranakhite --- 199, 305
 Kurchatovite --- 265, 268
 Kurgantaite --- 159
 Kurnakovite --- 216
 Kurumsakite --- 153
 Kusuite --- 119
 Kutnohorite --- 110, 112, 346
 Kyanite --- 279
 Labradorite --- 142, 143, 350, 353
 Labuntsovite --- 167, 173
 Lammerite --- 197
 Lamprophyllite --- 183, 187
 Lanarkite --- 306
 Landauite --- 313
 Landesite --- 281
 Langbanite --- 122
 Langbeinite --- 35

Langite --- 271, 277
 Lannonite --- 62
 Lansfordite --- 125
 Lanthanite --- 237
 Lanthanite-(Nd) --- 238
 Laplandite --- 235
 Larderellite --- 132
 Larnite --- 176
 Larsenite --- 303
 Latiumite --- 151, 238, 243
 Laubmannite --- 194
 Laueite --- 260
 Laumontite --- 217, 219, 394, 400, 401
 Laurionite --- 309
 Lausenite --- 234, 248
 Lautarite --- 194
 Lavendulan --- 112
 Lavenite --- 258, 271, 280, 285
 Lawrencite --- 92
 Lawsonbauerite --- 243
 Lawsonite --- 168
 Lazarenkoite --- 301
 Lazulite --- 249, 256
 Lazurite --- 34, 35, 386
 Leadhillite --- 305
 Lecontite --- 208
 Legrandite --- 171, 175
 Lehiite --- 245
 Leifite --- 64
 Leightonite --- 237
 Leiteite --- 196
 Lemoynite --- 141
 Leonite --- 128, 212
 Lepersonnite --- 263
 Lepidocrocite --- 202, 310
 Lepidolite --- 225, 227, 229, 366, 367
 Lermontovite --- 275
 Letovicite --- 217
 Leucite --- 34, 64, 132
 Leucophanite --- 239
 Leucophoenicite --- 289, 358, 360
 Leucophosphite --- 178
 Leucosphenite --- 163, 165
 Levyne --- 63, 85, 86, 395, 397, 398,
 400
 Liandratite --- 49
 Liberite --- 250
 Libethenite --- 184, 285
 Liddicoatite --- 102, 391, 392
 Liebenbergite --- 195, 298, 371, 372
 Liebigite --- 130
 Likasite --- 158, 260
 Lime --- 50
 Linarite --- 297

Lindackerite --- 164
 Lindgrenite --- 305
 Linnaeite --- 61
 Liottite --- 88, 341, 342
 Liroconite --- 258
 Litharge --- 122
 Lithiophilite --- 166, 169, 264, 274
 Lithiophosphate --- 142
 Litidionite --- 232
 Livingstonite --- 314
 Lizardite --- 143, 227, 230
 Loewite --- 85
 Lokkaite --- 239
 Lomonosovite --- 283, 285
 Loparite --- 55, 58
 Lopezite --- 181
 Lorandite --- 205
 Loranskite --- 53
 Lorenzenite --- 306, 307
 "Lorettoite" --- 121
 Loseyite --- 160
 Loudounite --- 139, 223
 Loughlinite --- 131
 Lovdarite --- 133, 218
 Lovozerite --- 91, 93, 227
 Luddenite --- 195
 Ludlamite --- 166
 Ludlockite --- 200
 Ludwigite --- 192, 193, 195
 Lueneburgite --- 223
 Lueshite --- 312
 Luetheite --- 188
 Lusungite --- 77, 78, 344, 345
 Macdonaldite --- 134, 219
 Macfallite --- 292
 Machatschkiite --- 96
 Mackayite --- 81
 Macquartite --- 312
 Magadiite --- 128, 212
 Magbasite --- 243
 Maghemite --- 60
 Magnesio-anthophyllite --- 242, 318,
 321
 Magnesio-arfvedsonite --- 159, 259,
 268, 319, 328, 329
 Magnesioaxinite --- 164, 263
 Magnesiochromite --- 51, 52, 387, 388
 Magnesiocopiapite --- 135, 139, 343
 Magnesio cummingtonite --- 249
 Magnesioferrite --- 58, 387, 389
 Magnesio-gedrite --- 318, 320, 321
 Magnesio-hastingsite --- 262, 270, 319,
 325, 326

Magnesio-hornblende --- 254, 265, 319,
 324, 325
 Magnesio-riebeckite --- 261, 320, 328
 Magnesite --- 109, 110, 111, 339, 340
 Magnesite-siderite --- 114
 Magnesium chlorophoenicite --- 167, 270
 Magnesium-zippeite --- 285
 Magnetite --- 59, 387, 389
 Magniotriplite --- 152, 160
 Magnocolumbite --- 313
 Magnussonite --- 52
 Makatite --- 126, 127, 211
 Malachite --- 300
 Malayaite --- 291
 Malladrite --- 84
 Mallardite --- 125
 Manandoite --- 150
 Manasseite --- 87
 Mandarinoite --- 293
 Manganaxinite --- 270
 Manganbabingtonite --- 180
 Manganberzeliite --- 47
 Manganese-hoernesite --- 147
 Manganhumite --- 175, 188, 358, 360,
 361
 Manganite --- 199, 203
 Mangan-neptunite --- 172, 173
 Manganocolumbite --- 313
 Manganolangbeinite --- 37
 Manganopalygorskite --- 231
 Manganosite --- 56
 Manganosteenstrupine --- 48
 Manganostibite --- 303
 Manganotantalite --- 201, 202
 Manganpyrosmalite --- 106
 Mansfieldite --- 155
 Mapimite --- 169
 Margaritasite --- 313
 Margarite --- 244, 254, 366, 368, 369
 Margarosanite --- 289
 Marialite --- 88, 90, 385
 Maricite --- 273
 Marshite --- 58
 Marsturite --- 171
 Mascagnite --- 134
 Massicot --- 205
 Masutomilite --- 231, 366, 367
 Masuyite --- 301
 Matlockite --- 120, 310
 Matulaite --- 233
 Mayenite --- 39, 41
 Mazzite --- 86, 395, 399
 Mcallisterite --- 86
 McGillite --- 106

MCGovernite --- 76
 Mcguinnessite --- 281
 Mckelveyite --- 105
 Mcnearite --- 143
 Medaite --- 189
 Meionite --- 95, 98, 385
 Meixnerite --- 87
 Melanite --- 52, 355, 357
 Melanocerite --- 41, 47, 111
 Melanophlogite --- 30, 32
 Melanotekite --- 201
 Melanovanadite --- 304
 Melanterite --- 126
 Melilite --- 41, 71, 72, 100, 105, 107,
 362, 363
 Meliphanite --- 72, 99, 362
 Mellite --- 89
 Melonjosephite --- 289
 Mendipite --- 203
 Mendozite --- 84, 209
 Mercallite --- 125
 Merlinoite --- 214, 395, 399
 Merrihueite --- 67, 68, 91, 96, 373,
 374
 Merwinite --- 175, 178
 Mesolite --- 131 395, 399
 Messelite --- 160, 164
 Meta-aluminite --- 217
 Meta-alunogen --- 126
 Meta-ankoleite --- 94, 363, 364
 Meta-autunite --- 97, 99, 241, 247,
 363, 364
 Metaborite --- 39
 Metacinnabar --- 60
 Metahaiweeite --- 154
 Metaheinrichite --- 103, 253, 363, 365
 Metahewettite --- 308
 Metahohmannite --- 177, 278
 Metakanlerite --- 103, 363, 365
 Metakirchheimerite --- 103, 363, 365
 Metakoettigite --- 267
 Metalodevite --- 251, 363, 364
 Metanovacekite --- 102, 363, 364
 Metarossite --- 199
 Metaschoderite --- 150
 Metasideronatrite --- 145
 Metastudtite --- 164
 Metatorbernite --- 70, 101, 247, 363,
 364
 Metatyuyamunite --- 299, 302
 Meta-uranocircite --- 247, 287, 363,
 364
 Metavanmeersscheite --- 267
 Metavariscite --- 142

Metavauxite --- 142
 Metavivianite --- 150
 Metavoltine --- 96
 Metazeunerite --- 104, 363, 365
 Meyerhofferite --- 221
 Miargyrite --- 206, 314
 Microcline --- 219, 350, 351, 352
 Microlite --- 51, 53, 54, 375, 376
 Microsomite --- 65, 341
 Miersite --- 56
 Milarite --- 88, 90, 373, 374
 Millisite --- 240
 Millosevichite --- 37
 Mimetite --- 120, 309, 330, 333
 Minasragrite --- 220, 222
 Minguzzite --- 226
 Minium --- 122
 Minnesotaite --- 245, 248
 Minrecordite --- 111, 112, 346
 Minyulite --- 137
 Mirabilite --- 208
 Misenite --- 127
 Miserite --- 147, 148
 Mitridatite --- 295, 298
 Mitscherlichite --- 102
 Mixite --- 76
 Mn-palygorskite --- 227
 Moctezumite --- 311
 Mohrite --- 129
 Moissanite --- 82
 Moluranite --- 52
 Molybdenite --- 123
 Molybdomenite --- 309
 Molybdophyllite --- 115
 Monazite --- 181, 189, 190, 191
 Monetite --- 152
 Monohydrocalcite --- 95
 Monsmedite --- 243
 Montanite --- 308
 Montdorite --- 242, 366, 368
 Montebrasite --- 151, 155, 240, 244
 Monteregeianite --- 133
 Montgomeryite --- 233
 Monticellite --- 257, 263, 370
 Montmorillonite --- 217, 221
 Montroydite --- 205
 Mooreite --- 224
 Moorhouseite --- 213
 Moraesite --- 211
 Mordenite --- 126, 127, 210, 395, 396,
 397
 Morelandite --- 79, 330, 333
 Morenosite --- 213
 Morinite --- 229

Mosandrite --- 159, 166
 Mosesite --- 54
 Mottramite --- 203, 311
 Motukoreaite --- 35
 Mounanaite --- 201, 308
 Mountainite --- 132
 Mountkeithnrite --- 87
 Mroseite --- 298
 Muirite --- 74
 Mukhinite --- 182, 347, 348
 Mullite --- 159, 163
 Mundite --- 268
 Mundrabillaite --- 224
 Munirite --- 287
 Murataite --- 55
 Murmanite --- 285, 288
 Muscovite --- 234, 238, 239, 240, 243,
 246, 366, 367, 368
 Muskoxite --- 294
 Nabaphite --- 34
 Nacaphite --- 217
 Nacrite --- 228
 Nadorite --- 204, 312
 Nagashimalite --- 186
 Nagelschmidtite --- 162, 169
 Nahcolite --- 214
 Nakauriite --- 242
 Nambulite --- 175
 Namibite --- 308
 Namuwite --- 37, 67, 94
 Nanlingite --- 116
 Nantokite --- 51
 Narsarsukite --- 69
 Nasinite --- 217
 Nasonite --- 79
 Nastrophite --- 34
 Natanite --- 46
 Natisite --- 113
 Natrite --- 221
 Natroalunite --- 67, 69, 316, 317
 Natroapophyllite --- 138
 Natrochalcite --- 163
 Natrodufrenite --- 187
 Natrofairchildite --- 88
 Natrojarosite --- 115, 116, 296, 316,
 317
 Natrolite --- 63, 127, 128, 395, 397,
 398
 Natromontebrasite --- 150
 Natron --- 208
 Natroniobite --- 311
 Natrophilite --- 168
 Natrophosphate --- 31
 Natrosilite --- 218

Naujakasite --- 225
 Navajoite --- 306
 Nefedovite --- 67, 144
 Neighborite --- 30
 Nekoite --- 137
 Nenadkevichite --- 166, 170
 Neotocite --- 32, 36
 Nepheline --- 88, 89, 90
 Nepouite --- 252
 Neptunite --- 178
 Nesquehonite --- 214
 Newberyite --- 133
 Ni analogue of serpentine --- 37, 38
 Nianite --- 242
 Nickelbischofite --- 154
 Nickelbloedite --- 215, 218
 Nickel-hexahydrite --- 127, 211
 Nickel-zippeite --- 290
 Nifontovite --- 146
 Nigerite --- 77, 115
 Nimate --- 256
 Ningyoite --- 158, 253
 Niobo-aeschynite --- 312
 Niobophyllite --- 287
 Niocalite --- 277
 Nissonite --- 246
 Niter --- 215
 Nitratite --- 95
 Nitrobarite --- 37
 Nitrocalcite --- 64, 213
 Nitromagnesite --- 215
 Nobleite --- 133
 Nontronite --- 230, 236, 241, 257
 Norbergite --- 140, 144, 358, 359
 Nordenskiöldine --- 113
 Nordite --- 251
 Nordstrandite --- 146, 147
 Norsethite --- 109, 110, 346
 Northupite --- 35, 36
 Nosean --- 32, 33, 386
 Novacekite --- 93, 231, 335, 336
 Nullaginite --- 290
 Nyerereite --- 86, 221
 Oblique illumination, technique --- 5
 Oboyerite --- 203, 311
 O'danielite --- 186
 Offretite --- 85, 395, 398
 Ogdensburgite --- 189
 Ojuelaite --- 180
 Okanoganite --- 113
 Okenite --- 217, 221
 Oldhamite --- 55
 Olgite --- 100

Oligoclase --- 138, 223, 224, 350, 352,
 353
 Olivenite --- 192, 294
 Olmsteadite --- 186, 189
 Olsacherite --- 304
 Olshanskyite --- 230
 Olympite --- 132
 Omphacite --- 167, 169, 173, 377, 379,
 380, 381
 Onoratoite --- 311
 Opal --- 30, 31
 Optic angle --- 15-19
 calculation of --- 15-17
 estimation of --- 18-19
 measurement of --- 18
 nomogram for --- 17
 relation to the principal
 refractive indices --- 16-17
 Optic axes, dispersion of --- 19
 Optic sign --- 15-19
 determination of --- 18-19
 Optical data, reliability of --- 2
 Optical directions, relation to
 crystallographic directions ---
 19-20
 Optical indicatrix --- 7
 Ordinary wave --- 9
 Ordonezite --- 81
 Orientite --- 290
 Orpiment --- 314
 Orthoclase --- 219, 350, 351, 352
 Orthoericssonite --- 194
 Orthoferrosilite --- 188, 377, 378
 Usarizawaite --- 75, 76, 316, 317
 Osumilite --- 66, 373, 374
 Osumilite-(Mg) --- 66, 138, 373, 374
 Otavite --- 116, 339, 340
 Otjisumeite --- 198
 Ottrelite --- 175
 Otwayite --- 170, 273
 Oursinite --- 253
 Overite --- 232
 Oxammite --- 225
 Oxychildrenite --- 175
 Ozocerite --- 64
 Pabstite --- 108
 Pachnolite --- 124
 Painite --- 115
 Palermoite --- 254
 Palmierite --- 110
 Palygorskite --- 220, 224
 Panasqueiraite --- 149
 Panethite --- 232
 Papagoite --- 253

Parabutlerite --- 165
 Paradamite --- 289
 Paraffin --- 64
 Paragonite --- 241, 366, 368
 Parahilgardite --- 157
 Parahopeite --- 155, 247
 Parakeldyshite --- 271
 Parakhinite --- 120
 Paralaurionite --- 310
 Paralstonite --- 107
 Paraluminite --- 209
 Paraschoepite --- 287
 Parascholzite --- 147
 Paraspurrite --- 265
 Parasymplesite --- 161, 164
 Paratacamite --- 78, 194
 Paratellurite --- 81
 Paraumbite --- 241
 Paravauxite --- 142
 Pargasite --- 153, 156, 162, 258, 267,
 319, 324, 325
 Parisite --- 73
 Parisite-(Y) --- 71
 Parnauite --- 275, 279
 Parsettensite --- 93
 Parsonsite --- 294, 299
 Partheite --- 140
 Parwelite --- 194
 Pascoite --- 295
 Paulingite --- 32, 395, 396
 Paulmooreite --- 198
 Pectolite --- 150, 157
 Pellyite --- 159
 Penfieldite --- 80
 Penikisite --- 171
 Penkvilksite --- 158
 Pennaite --- 173
 Pennantite --- 106, 107, 261, 264
 Pentagonite --- 224
 Pentahydrite --- 213, 217, 342
 Pentahydroborite --- 137
 Percylite --- 53
 Peretaite --- 196
 Perhamite --- 67
 Periclase --- 45, 46
 Perloffite --- 293
 Perovskite --- 57, 58, 203, 205, 313
 Perrierite --- 307
 Petalite --- 133
 Petarasite --- 149
 Petersite --- 73
 Petzite --- 60
 Pharalumite --- 245

Pharmacosiderite --- 42, 43, 164, 168,
 173, 272
 Pharmacolite --- 237
 Phaunouxite --- 138
 Phenakite --- 72
 Philipsbornite --- 47, 344, 345
 Phillipsite --- 128, 130, 212, 216,
 395, 397, 399, 400
 Phlogopite --- 91, 94, 225, 228, 235,
 366, 367, 369
 Phoenicochroite --- 204
 Phosgenite --- 80
 Phosinaite --- 231
 Phosphoferrite --- 169
 Phosphophyllite --- 244
 Phosphorroesslerite --- 211
 Phosphosiderite --- 280
 Phosphuranylite --- 109, 111, 274, 280
 Pnuralite --- 282
 Picite --- 41, 42
 Pickeringite --- 211
 Picromerite --- 125
 Picropharmacolite --- 143
 Piemontite --- 185, 188, 190, 286, 289,
 293, 347, 348, 349, 350
 Pigeonite --- 170, 173, 176, 378, 380,
 381, 382
 Pinakiolite --- 307
 Pinnoite --- 67
 Pirssonite --- 132
 Pitticite --- 39, 40
 Plancheite --- 164, 177
 Planerite --- 35
 Plattnerite --- 121
 Pleochroism --- 20-21
 Plumbogummite --- 72, 73, 344, 345
 Plumbojarosite --- 117, 300, 316, 317
 Plumbopyrochlore --- 54, 375, 376
 Plumbotellurite --- 203
 Plumbotsumite --- 302
 Pollucite --- 34, 35, 395, 400, 401
 Polybasite --- 314
 Polycrase --- 57
 Polyhalite --- 228
 Polymignite --- 57
 Portlandite --- 93
 Posnjakite --- 267
 Potassium alum --- 31
 Poughite --- 304, 305
 Powellite --- 79, 80
 Prehnite --- 153, 156, 160
 Preisingerite --- 201, 310
 Preiswerkite --- 245, 366, 368
 Preobrazhenskite --- 145

Priceite --- 238
 Priderite --- 80
 Principal refractive indices --- 9
 relation to optic angle --- 15
 Probertite --- 134
 Prosopite --- 131
 Prosperite --- 185
 Proustite --- 123
 Przhevalskite --- 285
 Pseudo-autunite --- 230
 Pseudoboleite --- 119
 Pseudobrookite --- 204
 Pseudolaueite --- 161
 Pseudomalachite --- 292, 297, 299
 Pucherite --- 313
 Pumpellyite --- 167, 171, 174, 177
 Pumpellyite-(Mn) --- 292
 Purpurite --- 195, 198
 p-Veatchite --- 141
 Pyrargyrite --- 123
 Pyrite --- 61
 Pyroaurite --- 91, 92
 Pyrobelonite --- 313
 Pyrochlore --- 50, 53, 56, 375, 376
 Pyrochroite --- 104, 111
 Pyromorphite --- 119, 307, 330, 333
 Pyrope --- 44, 355, 356
 Pyrope-almandine --- 45, 47, 356
 Pyrophanite --- 122
 Pyrophyllite --- 237
 Pyrosmalite --- 106, 108
 Pyrostilpnite --- 206
 Pyroxferroite --- 185, 187
 Pyroxmangite --- 181, 184
 Quartz --- 66
 Queitite --- 197, 301
 Quenstedtite --- 143
 Quetzalcoatlite --- 115
 Rabbittite --- 131
 Raite --- 138
 Rajite --- 201
 Ralstonite --- 30
 Rameauite --- 303
 Rancieite --- 81, 121
 Rankinite --- 159
 Ransomite --- 159
 Ranunculite --- 262
 Raspite --- 203
 Rauenthalite --- 140
 Ravvite --- 196, 300
 Realgar --- 314
 "Rectorite" --- 225
 Reddingite --- 160, 165
 Reedmergnerite --- 229, 351, 353

Reevesite --- 111
 Refractive index
 change with temperature --- 3-4
 determination --- 3-14
 by method of random grain mounts
 --- 11
 by spindle stage--- 11-14
 central illumination (Beke line)
 technique --- 4-5
 dispersion coloration technique
 --- 5-7
 for anisotropic substances ---
 9-11
 for isotropic substances ---
 7-9
 oblique illumination (Schroeder
 van der Kolk) technique --- 5
 wave-length variation method ---
 7
 dispersion of --- 6
 rechecking --- 4
 symbols --- 9
 Refractometer --- 4
 Reinerite --- 291
 Reinhardbraunsite --- 246
 Reliability of optical data --- 2
 Renardite --- 283
 Retgersite --- 86
 Retzian --- 190
 Retzian-(Nd) --- 190
 Revdite --- 211
 Reyerite --- 92
 Rhabdophane --- 72, 73, 75
 Rhabdophane-(La) --- 74
 Rhodesite --- 131
 Rhodizite --- 43
 Rhodochrosite --- 111, 113, 115, 116,
 339, 340
 Rhodonite --- 176, 179, 180, 182, 185
 Rhomboclase --- 140
 Richelsdorfite --- 287
 Richetite --- 305
 Richterite --- 245, 248, 249, 319, 327
 Riebeckite --- 273, 320, 329
 Ringwoodite --- 47, 387
 Rinneite --- 68
 Rivadavite --- 127
 Rockbridgeite --- 193, 196, 302
 Rodalquilarite --- 310
 Roeblingite --- 158
 Roedderite --- 66, 137, 373
 Roemerite --- 231
 Roesslerite --- 136
 Rokūhnite --- 156

Romanechite --- 58
 Romeite --- 49, 50, 54, 57, 202, 298,
 389, 390
 Röntgenite --- 72
 Rooseveltite --- 201, 311
 Rosasite --- 296
 Roscherite --- 158, 253
 Roscoelite --- 250, 269, 366, 369
 Roselite --- 174, 180
 Roselite-beta --- 283
 Rosenbuschite --- 166, 171
 Rosenhahnite --- 253
 Rosieresite --- 33
 Rossite --- 188, 291
 Rostite --- 209
 Roweite --- 253, 261
 Rowlandite --- 43, 44, 187
 Rozenite --- 222
 Ruizite --- 277
 Rusakovite --- 194, 296
 Russellite --- 81, 120
 Rustumite --- 256
 Rutherfordine --- 180
 Rutile --- 82
 Rutile, var Ilmenorutile --- 82
 Rynersonite --- 200
 Sabinaite --- 293
 Sabugalite --- 94, 235, 335, 336
 Sacrofanite --- 86, 341
 Sahamalite --- 289
 Sainfeldite --- 243, 248
 Sakhaite --- 41
 Salammoniac --- 40
 Saleeite --- 93, 231, 234, 335, 336
 Salesite --- 308
 Salite --- 174, 378, 381
 Samarskite --- 55, 57
 Sampleite --- 266
 Samuelsonite --- 161
 Sanbornite --- 245
 Saneroite --- 284
 Sanidine --- 219, 220, 350, 351, 352
 Santafeite --- 201, 308
 Santanaite --- 121
 Santite --- 124
 Saponite --- 91, 96, 216, 220, 228
 Sapphirine --- 174, 180, 275, 280, 284
 Sarcolite --- 69, 158
 Sarcopside --- 281
 Sarkinite --- 294
 Sarmientite --- 157
 Saryarkite --- 69
 Sasaite --- 209
 Sassolite --- 208

Satimolite --- 226
 Satpaevite --- 43, 169
 Satterlyite --- 110
 Sauconite --- 225, 238, 245
 Sayrite --- 303
 Sazhinite --- 135
 Sborgite --- 124
 Scapolite --- 91, 93, 385
 Scawtite --- 151
 Schafarzikite --- 80
 Schairerite --- 62
 Schallerite --- 109
 Schaurteite --- 67
 Scheelite --- 79
 Schertelite --- 133
 Schieffelinite --- 303
 Schmietterite --- 307
 Schneiderhoehnite --- 201
 Schoderite --- 139, 228
 Schoenfliesite --- 38, 42
 Schoepite --- 277, 284
 Scholzite --- 147, 148
 Schoonerite --- 258
 Schorl --- 106, 109, 391, 392
 Schorlomite --- 52, 355, 357
 Schroeckingerite --- 89, 222
 Schroeder van der kolk technique --- 5
 Schuetteite --- 120
 Schuilingite --- 286
 Schultenite --- 197
 Schumacherite --- 203
 Schwartzembergite --- 121, 312
 Scolecite --- 218, 395, 400
 Scorodite --- 184, 191
 Scorzalite --- 265
 Seamanite --- 262
 Searlesite --- 221
 Seeligerite --- 312
 Segelerite --- 157
 Seidozerite --- 182, 187
 Sekaninaite --- 230
 Sellaite --- 62
 Semenovite --- 99, 245
 Senaite --- 122
 Senarmontite --- 54, 200
 Senegalite --- 143
 Sengierite --- 303
 Sepiolite --- 35, 215, 219, 226
 Serandite --- 165
 Serendibite --- 174, 184
 Serpentine --- 36, 37
 Serpentine group mineral --- 265
 Serpierite --- 253
 Shabynite --- 144, 231

Shafranovskite --- 95
 Sharpite --- 161
 Shattuckite --- 190
 Shcherbakovite --- 285
 Sherwoodite --- 113
 Shortite --- 227
 Shubnikovite --- 271
 Shuiskite --- 288
 Sibirskite --- 254
 Sicklerite --- 283
 Siderite --- 115, 116, 117, 339, 340
 Sideronatrium --- 134
 Siderophyllite --- 103, 253, 366, 369
 Siderotil --- 220, 342
 Sidorenkite --- 228
 Sigloite --- 147
 Silhydrite --- 125, 209
 Sillenite --- 59
 Sillimanite --- 164
 Simplotite --- 287
 Simpsonite --- 119
 Sincosite --- 107, 271
 Sinhalite --- 273
 Sinoite --- 298
 Sjögrenite --- 92
 Sklodowskite --- 157, 251
 Slavikite --- 88, 89
 Slawsonite --- 146, 351, 354
 Smithite --- 315
 Smithsonite --- 115, 117, 339, 340
 Smolianinovite --- 155, 247
 Sobolevite --- 269
 Sodalite --- 33, 386
 Sodalite, var Hackmanite --- 33
 Soddyite --- 269
 Sodium alum --- 31
 Sodium autunite --- 93, 335, 336
 Sodium boltwoodite --- 249, 261
 Sodium dachiardite --- 210, 395, 396
 Sodium phlogopite --- 366
 Sodium uranospinite --- 99, 244, 363,
 364
 Sodium-zippeite --- 270
 Sogdianite --- 69, 97, 98, 373, 374
 Söhngeite --- 45
 Solongoite --- 132
 Sonolite --- 278, 290, 358, 360, 361
 Sonoraite --- 307
 Sorensenite --- 234
 Souzalite --- 254
 Spadaite --- 135
 Spangolite --- 107, 108
 Spencerite --- 241
 Sperryllite --- 60

Spessartine --- 48, 355, 356
 Spessartine-almandine --- 49, 357
 Sphaerocobaltite --- 117, 339, 340
 Sphalerite --- 58, 59
 Spindle stage, mounting fragment on
 --- 13
 Spindle stage, orthoscopic procedure
 --- 13
 Spindle stage, sketch --- 12
 Spindle stage technique --- 11-14
 Spinel --- 44, 45, 48, 50, 387, 388
 Spiroffite --- 197
 Spodiosite --- 168
 Spodumene --- 164, 167, 378, 379
 Spurrite --- 265
 Stanleyite --- 133
 Starkeyite --- 209
 Staurolite --- 183, 185, 186, 286
 Steacyite --- 92
 Steenstrupine --- 106
 Steigerite --- 284
 Stellerite --- 213, 395, 398
 Stenonite --- 220
 Stepanovite --- 87
 Stercorite --- 124
 Stereographic plotting of extinction
 positions --- 18
 Stetefeldtite --- 52, 389, 390
 Stevensite --- 34
 Stewartite --- 260
 Stibiconite --- 39, 43, 45, 48, 51, 53,
 389, 390
 Stibiobetafite --- 57, 375, 376
 Stibiocolumbite --- 205
 Stibiotantalite --- 204
 Stibivanite --- 196, 300
 Stibnite --- 315
 Stichtite --- 90
 Stilbite --- 212, 214, 395, 398, 399
 Stilleite --- 59
 Stillwellite --- 77
 Stilpnomelane --- 94, 97, 101, 105,
 108, 109, 111, 235, 241,
 250, 261, 269, 275, 283
 Stishovite --- 77
 Stokesite --- 152, 154
 Stolzite --- 121
 Stottite --- 112
 Straetlingite --- 89
 Stranskiite --- 297
 Strashimirite --- 284
 Strelkinitite --- 298, 301
 Strengite --- 164, 175, 182
 Stringhamite --- 177

Tamarugite --- 129
 Tancoite --- 228
 Taneyamalite --- 262
 Tantalite --- 204
 Tapiolite --- 81, 122
 Taramellite --- 189
 Taramite --- 274, 277, 319, 327
 Taranakite --- 86
 Tarapacaite --- 279
 Tarasovite --- 233, 366, 367
 Tarbuttite --- 269, 275
 Tatarskite --- 259
 Tavorite --- 192
 Tazheranite --- 57
 Teepleite --- 87
 Teineite --- 290
 Telegdite --- 36
 Tellurite --- 202, 310
 Tengerite --- 144, 155
 Tennantite --- 60
 Tenorite --- 205, 314
 Tephroite --- 280, 287, 293, 371, 372
 Terlinguaite --- 314
 Terskite --- 235
 Teruggite --- 135
 Teschemacherite --- 222
 Tetrakalsilite --- 89
 Tetranatrolite --- 63, 395, 397
 Thadevite --- 240
 Thalenite --- 278, 284
 Thaumaside --- 85, 86
 Theisite --- 291
 Thenardite --- 126
 Theophrastite --- 77
 Thermonatrite --- 215
 Thomsenolite --- 208
 Thomsonite --- 133, 135, 137, 395, 400,
 401
 Thoreaulite --- 204
 Thorianite --- 54, 56
 Thorite --- 40, 43, 44, 48, 77, 78
 Thorogummite --- 45, 47, 72, 74
 Thorosteenstrupine --- 40
 Thortveitite --- 292
 Thorutite --- 55
 Threadgoldite --- 235
 Tiemannite --- 60
 Tienshanite --- 106
 Tikhonkovite --- 209
 Tilasite --- 261
 Tilleyite --- 157, 251
 Tinaksite --- 154
 Tincalconite --- 62
 Tinticite --- 184, 285

Tinzenite --- 172, 268, 272, 274
 Tiragalloite --- 185
 Tirodite --- 161, 255, 318, 322, 323
 Tisinalite --- 100
 Titanite --- 195, 197, 199
 Tlalocite --- 292
 Tlapallite --- 304, 309
 Tobelite --- 232, 237, 366, 367
 Tobermorite --- 229
 Toernebohmite --- 192, 195
 Tombarthite --- 40
 Topaz --- 152, 153, 156
 Torbernite --- 96, 239, 335, 336
 Torreyite --- 236
 Tosudite --- 231
 Traskite --- 110
 Trechmannite --- 122
 Tremolite --- 239, 245, 318, 324
 Trevorite --- 58, 387, 389
 Triangulite --- 166
 Tridymite --- 125, 128
 Trigonite --- 309
 Trimerite --- 279
 Triphylite --- 171, 269
 Triplite --- 162, 168, 170, 265, 268
 Triploidite --- 179
 Trippkeite --- 79
 Tripuhyite --- 202
 Tristramite --- 71
 Tritomite --- 46
 Tritomite-(Y) ("Spencite") --- 40, 43
 Troegerite --- 100, 248, 335, 336
 Trolleite --- 252
 Trona --- 213
 Truscottite --- 90, 225
 Tschermakite --- 262, 319, 325
 Tschermigite --- 31
 Tsumebite --- 198
 Tugtupite --- 64
 Tuhualite --- 152
 Tundrite --- 191
 Tunellite --- 137
 Tungstite --- 307, 311
 Tunisite --- 67
 Turanite --- 306
 Turquoise --- 154
 Tuscanite --- 238
 Tveitite --- 210
 Tychite --- 34
 Tyretskite --- 158
 Tyrolite --- 271, 280
 Tyuyamunite --- 293, 297
 Uduminelite --- 155
 Uklonskovite --- 129

Uranospinite --- 95, 229, 239, 335, 336
 Uranpyrochlore ("Hatchettolite") ---
 52, 375, 376
 Urea --- 63
 Ureyite --- 287, 378, 383
 Uricite --- 183
 Ushkovite --- 252
 Usovite --- 124
 Ussingite --- 131
 Uvanite --- 196
 Uvarovite --- 50, 355, 357
 Uvarovite-grossular --- 49, 357
 Uvite --- 102, 105, 391, 392
 Valentinite --- 312
 Vanadinite --- 120, 122, 330, 331, 333
 Vandenbrandeite --- 189, 292
 Vandendriesscheite --- 294, 298
 Vanmeersscheite --- 277
 Vanthoffite --- 212
 Vanuralite --- 297
 Variscite --- 229, 239
 Varulite --- 176
 Vashegyite --- 32, 34
 Vaterite --- 66
 Vauquelinite --- 311
 Vauxite --- 142
 Väyrynenite --- 261
 Veatchite --- 141
 Veatchite-A --- 140
 Vermiculite --- 89, 91, 94, 98, 103,
 224, 227, 234, 242, 256
 Verplanckite --- 108
 Vertumnite --- 222
 Vesignieite --- 307, 309
 Vesuvianite --- 42, 74, 75, 109, 110,
 112, 113, 114, 279
 Veszelyite --- 154, 159, 164
 Vigezzite --- 202, 310
 Viitaniemiite --- 229
 Villiaumite --- 30, 84
 Vimsite --- 245
 Vinogradovite --- 289
 Virgilite --- 88
 Viseite --- 35
 Vishnevite --- 33, 86, 341
 Vismirnovite --- 45
 Vitusite --- 257
 Vivianite --- 149, 151, 155
 Vladimirite --- 163, 259
 Vlasovite --- 247
 Voglite --- 134, 139
 Volborthite --- 194, 199, 307
 Volkovskite --- 138
 Voltaite --- 38, 39

Voltzite --- 80
 Vrbaitite --- 206
 Vuagnatite --- 280
 Vudyavrite --- 38, 41
 Vuonnemite --- 161
 Vyuntspakhkite --- 272
 Wadeite --- 70
 Wagnerite --- 144, 146
 Wairakite --- 130, 214, 395, 399
 Wakefieldite --- 118
 Wallkilldellite --- 111
 Walpurgite --- 304
 Walstromite --- 269
 Wardite --- 68
 Wardsmithite --- 85
 Warwickite --- 188, 192
 Water --- 30
 Wattevilleite --- 208
 Wave-length variation method --- 7
 Wavellite --- 134, 137, 138
 Weberite --- 124
 Weddellite --- 65
 Weeksite --- 242
 Wegscheiderite --- 218
 Weilerite --- 73, 337, 338
 Weillite --- 271
 Welinite --- 78
 Wellsite --- 130, 395, 399
 Weloganite --- 255
 Welshite --- 192
 Wenkite --- 96
 Wermlandite --- 85
 Wherryite --- 306
 Whewellite --- 141
 Whiteite --- 147
 Whiteite-(Mn) --- 236
 Whitlockite --- 98, 100, 101
 Whitmoreite --- 276, 280
 Wickenburgite --- 109
 Wickmanite --- 43
 Wicksite --- 177
 Widenmannite --- 301
 Wightmanite --- 241
 Wilcoxite --- 208
 Wilhelmvierlingite --- 165, 262
 "Wilkeite" --- 104
 Willemite --- 74, 75
 Willemseite --- 258
 Winstanleyite --- 58
 Wiserite --- 113
 Witherite --- 266, 334
 Wodginite --- 202, 203
 Woehlerite --- 278
 Woelsendorfite --- 304, 308

Wolfeite --- 185
 Wolframite --- 204
 Wollastonite --- 248, 255, 258
 Wollastonite-2M --- 249
 Wonesite --- 243, 366, 368
 Woodhouseite --- 71, 72, 337, 338
 Woodwardite --- 141, 145
 Wroewolfeite --- 268
 Wulfenite --- 121, 122
 Wurtzite --- 81
 Wyartite --- 301
 Xanthoconite --- 314
 Xanthoxenite --- 277
 Xenotime --- 74, 75
 Xiangjiangite --- 232
 Xocomecatlite --- 301
 Xonotlite --- 146
 Yafsoanite --- 48
 Yagiite --- 65, 373
 Yaroslavite --- 124
 Yavapaiite --- 269
 Yeatmanite --- 301
 Yedlinite --- 120
 Yftisite --- 275
 Yoderite --- 171
 Yofortierite --- 225
 Yoshimuraite --- 189, 290
 Yttrialite --- 46
 Yttrocrasite --- 55, 309
 Yttrpyrochlore ("Obruchevite") --- 49,
 375
 Yttrotantalite --- 56
 Yttrotungstite --- 305
 Yugawaralite --- 130, 214, 395, 398
 Yuksporite --- 163
 Zaherite --- 33
 Zairite --- 116, 344, 345
 Zakharovite --- 91
 Zapatalite --- 103
 Zaratite --- 37, 38, 39
 Zavaritskite --- 81, 120
 Zektzerite --- 236
 Zellerite --- 142
 Zemannite --- 78
 Zeophyllite --- 91, 229
 Zeunerite --- 99, 335, 336
 Zhemchuzhnikovite --- 84
 Zhonghuacerite --- 112
 Zincaluminite --- 89
 Zinc-copper melanterite --- 128
 Zincfauserite --- 209
 Zincite --- 80
 Zincobotryogen --- 140
 Zincocopiapite --- 141, 343

Zincsilite --- 227
Zinc-zippeite --- 289
Zinkosite --- 264
Zinnwaldite --- 94, 233, 366, 367
Zippeite --- 278, 282
Zircon --- 41, 43, 47, 78, 79, 80, 198
Zircon "Malacon" --- 49, 75
Zircophyllite --- 283
Zircosulfate --- 159
Zirfesite --- 40
Zirkelite --- 53, 56, 57, 375, 376
Zirsinalite --- 98
Zoisite --- 172, 174, 347
Zunyite --- 37, 38, 39
Zussmanite --- 103
Zwieselite --- 174
Zykaite --- 253

