The Chemical Analysis of Argonne Premium Coal Samples

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The Chemical Analysis of Argonne Premium Coal Samples

Edited by Curtis A. Palmer

U.S. GEOLOGICAL SURVEY BULLETIN 2144

This manual presents analytical data from currently recommended procedures as well as procedures used in the 1980's by the geochemical laboratories of the U.S. Geological Survey for the chemical characterization of coal and a comparison of the results of these procedures for the Argonne Premium Coal samples.



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CONVERSION FACTORS

Both metric and inch-pound units are used by the different authors. Conversion factors are given below.

Multiply	By	To obtain
	Length	
nanometer (nm)	0.03937×10^{-6}	inch
micrometer (µm)	0.03937×10^{-3}	inch
millimeter (mm)	0.03937	inch
meter (m)	3.281	foot
inch	25.4	millimeter
	Mass	
picogram (pg)	1.543×10^{-11}	grain
nanogram (ng)	1.543×10^{-8}	grain
microgram (µg)	1.543×10^{-5}	grain
milligram (mg)	1.543×10^{-2}	grain
gram (g)	0.03527	ounce avoirdupois
	Volume	
microliter (µL)	0.016	minims
milliliter (mL)	0.0338	fluid ounce
cubic centimeter (cm ³)	0.06102	cubic inch
	Flow	
liter per minute (L/min)	0.2642	gallon/minute
	Force per unit area	
pound per square inch (lb/in ²)	6,895	pascal

For temperature conversions from degrees Celsius (°C) to degrees Fahrenheit (°F), use the following:

$(1.8 \times {}^{\circ}C) + 32 = {}^{\circ}F$

Standard reference materials.—The National Bureau of Standards (NBS) was renamed the National Institute of Standards and Technology (NIST) in 1988. In this report, standard reference materials from this agency are called NIST 1632a and so on, and the certificates of analysis of the reference materials are cited as National Bureau of Standards publications.

THE CHEMICAL ANALYSIS OF ARGONNE PREMIUM COAL SAMPLES

Edited by Curtis A. Palmer

The Chemical Analysis of Argonne Premium Coal Samples: An Introduction

By Curtis A. Palmer

ABSTRACT

Methods used to determine the concentrations of 67 elements in 8 Argonne Premium Coal samples are summarized in this bulletin. Seventeen different pro-cedures involving 11 different techniques were used. Sample identification, sample protocol, and sample descriptions are presented.

INTRODUCTION

This bulletin contains papers discussing the U.S. Geological Survey's chemical analyses of eight Argonne Premium Coal samples. The papers present analytical results obtained from all the methods used in the analyses and give a comparison of the results for elements where more than one technique was used. This paper describes the samples, explains the numbering protocol used throughout this bulletin, and lists the elements whose concentrations were determined by each technique used.

A detailed description of the samples, the reasons for their collection, and other background information have been reported by Vorres (1990, 1993). Table 1 contains the sample identification for the eight Argonne Premium Coal samples used in this study, with the seam, location, rank, and ash yield for each. The samples come from seven different States and range in rank from lignite to low volatile bituminous. The sample identification protocol is similar to that used by Vorres (1990). It consists of a two- to four-letter identifier that represents either the State or the seam from which the sample was obtained, followed by one or more spaces, the letters "PC" for Premium Coal, and a dash followed by sample numbers 1 through 8. In the papers that follow, splits are identified by the addition of a dash and another number after the sample number. For example, split 2 of Premium Coal sample 1 is identified as UF PC-1-2, with UF designating the Upper Freeport coal seam from which Premium Coal sample 1 was obtained.

Seventeen different procedures involving 11 different techniques were used to determine concentrations of 67 different elements. Ten of these procedures required coal ash, and the other seven used the whole coal. Table 2 shows the

 Table 1.
 Sample identification, location, rank, and ash yield of the eight Argonne Premium Coal samples analyzed in this work.

 [Ash yields are taken from table 1 of Doughten's paper this volume]

[Asn y	ields	are ta	акеп	Irom	lable	01	Dougnien	s paper,	unis	volume	

Sample ID	Seam	State	Rank	Ash yield (%)
UF PC-1	Upper	Pa.	Medium	13.5
	Freeport.		volatile bituminous.	
WY PC-2	Wyodak- Anderson.	Wyo.	Subbituminous	8.5
IL PC-3		I11.	High volatile bituminous.	16.2
PITT PC-4	Pittsburgh (No. 8).	Pa.	High volatile bituminous.	9.2
POC PC-5	Pocahontas	Va.	Low volatile	5.3
UT PC-6	-	Utah	bituminous. High volatile bituminous.	4.6
WV PC-7	Canyon. Lewiston- Stockton.	W. Va.	High volatile bituminous.	19.4
ND PC-8		N. Dak	. Lignite	9.5

Table 2. Tabulation of elements determined by different techniques.

[[]DCAES, direct-current arc atomic emission spectrography; XRF, X-ray fluorescence spectrometry; INAA, instrumental neutron activation analysis; ICAP-AES, inductively coupled argon plasma-atomic emission spectrometry, (R), Reston, (D), Denver; ICAP-MS, inductively coupled argon plasma-mass spectrometry; AAS, atomic absorption spectrometry; CVAAS, cold-vapor AAS; HGAAS, hydride-generation AAS; CHN, carbon-hydrogen-nitrogen elemental analysis; WC, whole coal]

_						Fechnique a	nd material						Total
Element –	DC. WC	AES Ash	WC X	RF Ash	INAA WC	ICAP- AES(R) Ash	ICAP- AES(D) Ash	ICAP-MS Ash	AAS Ash	CVAAS WC	HGAAS WC	CHN WC	number of procedures
H Li Be B C		X X				X	X X X ²		X			X X	1 H 2 Li 3 Be 2 B 1 C
N Na Mg Al Si	X X X	X X X X		$\begin{array}{c} \mathbf{X}^1\\ \mathbf{X}^1\\ \mathbf{X}^1\\ \mathbf{X}^1\end{array}$	Х	X X X	$egin{array}{c} X \ X^2 \ X^2 \ X^2 \ X^2 \end{array}$					Х	1 N 5 Na 5 Mg 5 Al 4 Si
P S Cl K Ca	X	X X X	\mathbf{X}^1 \mathbf{X}^1	$egin{array}{c} \mathbf{X}^1 \ \mathbf{X}^1 \ \mathbf{X}^1 \ \mathbf{X}^1 \ \mathbf{X}^1 \end{array}$	Х	X X X	$egin{array}{c} X^2 \ X^2 \ X^2 \end{array}$						5 P 1 S 1 Cl 5 K 5 Ca
Sc Ti V Cr Mn	X X X X	X X X X X X	Х	$egin{array}{c} \mathbf{X}^1 \ \mathbf{X} \ \mathbf{X}^1 \ \mathbf{X}^1 \end{array}$	X X	X X X X	X X ² X X X X						3 Sc 5 Ti 4 V 7 Cr 5 Mn
Fe Co Ni Cu Zn	X X X X X X	X X X X	X X X	X ¹ X X X	X X X X	X X X X	X ² X X X X X		Х				6 Fe 5 Co 7 Ni 6 Cu 6 Zn
Ga Ge As Se Br	X X	X X X			X X X			X X X			Х		3 Ga 3 Ge 3 As 2 Se 1 Br
Rb Sr Y Zr Nb	X	X X X X	X X X X X	X X X X X X	X X	X X	X X X ²	X X					4 Rb 6 Sr 5 Y 5 Zr 4 Nb
Mo Ag Cd Sn Sb		X X X			Х			X X X X X	X^4				2 Mo 2 Ag 2 Cd 2 Sn 2 Sb
Te Cs Ba La Ce	X	X X X	X X X	X X X	X X X X	Х	X^2	X X X ³ X ³					1 Te 2 Cs 7 Ba 5 La 5 Ce
Pr Nd Sm Eu Gd		X X X X			X X X			X ³ X ³ X ³ X ³ X ³					2 Pr 3 Nd 3 Sm 3 Eu 1 Gd
Tb Dy Ho Er Tm					Х			X ³ X ³ X ³ X ³ X ³					2 Tb 1 Dy 1 Ho 1 Er 1 Tm

 Table 2.
 Tabulation of elements determined by different techniques—Continued.

					-	Fechnique a	nd material						Total
Element	DC.	AES	X	RF	INAA	ICAP- AES(R)	ICAP- AES(D)	ICAP-MS	AAS	CVAAS	HGAAS	CHN	number of
-	WC	Ash	WC	Ash	WC	WC Ash	Ash	Ash	Ash	WC	WC	WC	procedures
Yb Lu Hf Ta W		Х			X X X X X X			X ³ X ³ X ³ X ³					3 Yb 1 Lu 2 Hf 2 Ta 2 W
Au Hg Tl Pb Bi	х	X X						X X X X	X^4	Х			1 Au 1 Hg 1 Tl 4 Pb 2 Bi
Th U					X X		Х	Х					2 Th 2 U
TOTALS	18	38	14	23	29	18	25	33	4	1	1	3	207

¹Wavelength-dispersive procedure; all other XRF procedures are energy-dispersive XRF.

²Sinter digestion procedure; all other ICAP-AES procedures are acid digestion procedures.

³Sinter digestion procedure; all other ICAP-MS procedures are acid digestion procedures.

⁴Graphite furnace AAS; all other AAS procedures are flame AAS.

elements, listed by atomic number, whose concentrations were determined by each procedure. Procedures include two direct-current arc atomic emission spectrography (DCAES) procedures, one on the ash and one on the whole coal; four X-ray fluorescence spectrometry (XRF) procedures, including both wavelength- and energy-dispersive procedures for the analysis of both the whole coal and the ash; and four atomic absorption spectrometry (AAS) procedures, including flame and graphite furnace AAS on the ash, cold-vapor atomic absorption spectrometry (CVAAS) procedures for mercury on the whole coal, and a hydride-generation AAS (HGAAS) procedure for selenium. Other techniques include instrumental neutron activation analysis (INAA) and combustion/gas chromatographic analyses for carbon, hydrogen, and nitrogen (CHN), which were determined on the whole coal only. Five inductively coupled argon plasma procedures were made on the ash only. Three of these procedures used an acid digest: the two inductively coupled argon plasma-atomic emission spectrometry (ICAP-AES) procedures (one in Reston and one in Denver) and the inductively coupled argon plasma-mass spectrometry (ICAP-MS) on the acid digest prepared in Denver. Two additional procedures used a sinter digest: the ICAP-AES and the ICAP-MS procedures in Denver. Ash yield (see paper by Doughten, this volume) and moisture content (see paper by Krasnow and Finkelman, this volume) are also reported. A detailed description of each of these procedures is given in the following papers.

In addition to the concentrations of 38 elements determined by DCAES on the ash, lower limits of detection were reported for another 24 elements, making a total of 62 elements reported by DCAES in the paper by Skeen, Libby, and Crandell. Of the 67 different elements with reported concentrations, 51 were determined by more than one technique. Results obtained by different techniques are compared by Palmer and Klizas in this volume.

This bulletin is a greatly expanded version of a preliminary report edited by Palmer and Walthall (1991) that contained information on 58 elements, 33 of which had been determined by more than one technique. All of the preliminary data were collected at the analytical facilities of the U.S. Geological Survey (USGS) in Reston, Va., and represented the procedures used to analyze coal through 1990. In 1990, work was begun to develop a procedure to determine trace-element concentrations by only high-precision techniques. The determination of several of the elements by DCAES was replaced by ICAP-MS, a technique available within the USGS only in laboratories in Denver, Colo. Because the dissolution procedures for ICAP-MS and ICAP-AES are the same, all routine ICAP-AES determinations on coal were moved to the Denver facilities. These two techniques were appropriate for most of the elements previously determined by the Reston high-precision techniques: INAA, XRF, and AAS. Those techniques, therefore, have been dropped for routine analysis, although INAA is still used for many samples when determinations on the whole coal are necessary or when sample size is critical. One element determined by INAA and not determined with the new procedures was selenium. A hydride-generation AAS method was developed in Denver to determine selenium. A new procedure to determine mercury by CVAAS was also added. CHN information is routinely obtained by ultimate and proximate analysis of coal by an outside laboratory, although CHN is still determined in USGS laboratories on some research samples.

The following chapters include papers modified from the open-file report by Palmer and Walthall (1991), as well as new papers describing the new Denver techniques. Results in each paper are reported in the manner in which results are normally reported by the laboratory performing the analysis. In all the papers except the last, results from methods used to analyze ash are reported on an ash basis, whereas results from methods used to analyze whole coal are reported on a whole-coal basis. Similarly, concentrations of major rock-forming elements are reported on an oxide basis or an element basis depending on the standard practice of the laboratory performing the analysis. The units parts per million (ppm) and micrograms per gram (μ g/g) are equivalent as are weight percent and percent (%); all results in this report are on a weight basis.

The comparison chapter by Palmer (1991) in the openfile report has been completely rewritten by Palmer and Klizas as the last paper in this report to include all the new techniques and a statistical analysis of results for all 51 elements that were determined by more than one procedure. All results in the last paper are converted to an element basis and a whole-coal basis and are expressed in weight percent and parts per million. Although only eight samples are included in this bulletin, comparison of the data for old and new techniques is useful in determining the accuracy of the data in the National Coal Resource Data System (NCRDS), which contains data for over 13,000 coal and rock samples, 7,400 of which are available on CD-ROM (Bragg and others, 1994).

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Rehydration of Desiccated Argonne Premium Coal Samples

By Marta R. Krasnow and Robert B. Finkelman

ABSTRACT

Eight Argonne Premium Coal samples stored in polyethylene bottles for several years exhibited substantial moisture losses. The samples retained an average of 23 percent of their original moisture content, with a range of 5 to 59 percent retention. Resaturated samples averaged 53 percent of the original moisture, with a range of 14 to 95 percent. If desiccated coal samples are to be analyzed, we recommend that moisture contents be determined prior to analysis. Drying coal samples prior to analysis would be an acceptable alternative if the original moisture content is known and the sample does not pick up moisture between drying and weighing for analysis.

INTRODUCTION

The Argonne National Laboratory has sealed its eight Premium Coal samples in glass ampoules filled with an inert gas (argon) to ensure that they retain their original properties during storage (Vorres, 1990). Once the ampoules are opened in a laboratory, however, sample alteration, such as dehydration and oxidation, can occur.

EXPERIMENTAL

We experienced a situation in which Argonne Premium Coal samples were stored in our laboratory for 2 to 4 years in polyethylene bottles, which allowed the samples to dehydrate (see columns 1 and 2 of table 1). A substantial amount of moisture had been lost from each sample during storage. This paper is intended to illustrate the degree of dehydration and to discuss the possibility of rehydrating the samples.

From 41 to 95 percent of the original moisture was lost during the unregulated storage of the "aged" samples. Sample WY PC-2 lost almost 20 weight percent moisture, and ND PC-8 lost more than 30 weight percent. Sample POC PC-5 lost the least absolute amount of moisture, slightly more than 0.5 weight percent.

To resaturate the samples, weighed splits of -100 mesh coal were placed in uncovered petri dishes in a vacuum des-

Table 1. Moisture of Argonne Premium Coal samples under different conditions.

[Values are in weight percent; all moisture values were determined by using the ASTM recommended procedure (ASTM, 1996a); Bit., bituminous; Sub., subbituminous; Lig., lignite]

Sample ident	ification	(1)	(2)	(3)
Sample	Rank	"Fresh" samples ¹	"Aged" samples ²	After resaturation ²
UF PC-1	Sub.	1.13	0.17 (15)	0.16 (14)
WY PC-2		28.09	8.79 (31)	9.76 (35)
IL PC-3		7.97	1.41 (18)	6.77 (85)
PITT PC-4	Bit.	1.65	.97 (59)	1.57 (95)
POC PC-5	Bit.	.65	.10 (15)	.23 (35)
UT PC-6		4.63	.21 (05)	3.29 (71)
WV PC-7		2.42	.85 (35)	1.29 (53)
ND PC-8		32.24	1.94 (06)	11.90 (37)

¹Samples in ampoules; data from Vorres (1993).

² Number in parentheses is the percentage of original moisture content.

iccator containing a barometer and a thermometer. Approximately 800 mL of deionized water was added to the base of the desiccator. The samples were kept under atmospheric pressure in the desiccator at 90 percent relative humidity and 22°C for 24 hours, after which they were removed from the desiccator and weighed. Several samples had increases far in excess of the original moisture loss. For example, sample UF PC-1, which had lost about 1 weight percent moisture, increased over 5 weight percent. This "excess" moisture may be due to condensation on the coal particles. We therefore allowed the resaturated samples to equilibrate with the ambient atmosphere for 24 hours (22°C-23°C, 60-64 percent relative humidity) prior to determining the moisture (referred to as resaturated moisture) by the American Society for Testing and Materials (ASTM) (1996a) procedure (table 1, column 3).

After the 24-hour residency in the desiccator at the 90 percent relative humidity and subsequent equilibration, moisture increased substantially in most samples. Sample IL PC-3 regained 67 percent of the original moisture (table 1, column 1), and UT PC-6 regained 66 percent. Sample IL PC-3 regained 5.36 weight percent moisture, and ND PC-8 gained almost 10 weight percent. Only sample UF PC-1

showed no gain in moisture. Despite the substantial increases in moisture for most samples, none of the samples, with the possible exception of PITT PC-4, recovered all the moisture lost on drying.

Experiments on water desorption and adsorption have been conducted by varying vapor pressure and holding the temperature constant, usually less than 40°C (Gauger, 1945; Allardice and Evans, 1978). Results of the experiments indicate that, once dry, a coal sample will not adsorb enough water to regain its original moisture content. There is no generally accepted mechanism to explain this phenomenon (Allardice and Evans, 1978); however, several theories have been offered. These include (1) the shrinking of coal on drying, which causes a collapse of some capillaries, so that the dried material can no longer hold or take up as much water as it held originally, and (2) the replacement of moisture on the walls of some capillaries by adsorbed gases, making it difficult to re-wet the capillaries (Gauger, 1945).

Vorres and Kolman (1988) and Vorres and others (1988) conducted drying and rehydration studies of Argonne Premium Coal samples. They concluded that coal rank, particle size, and degree of oxidation affected moisture removal and replacement.

The Argonne Premium Coal samples behave in a typical fashion with respect to rehydration. Improper storage can lead to substantial moisture loss. Resaturation generally will restore some of the lost moisture. These observations on moisture loss are important if measurements of physical properties are to be made on samples that have been stored under noncontrolled conditions for any length of time. These observations are also important for calculations involving chemical analysis of the raw coal (for example, instrumental neutron activation analysis or X-ray fluorescence analysis). Assuming that the coal samples have retained their original moisture contents can lead to errors of as much as 30 percent for low-rank coals (lignites and subbituminous: ND PC-8 and WY PC-2, respectively). Even for higher rank coal (for example, IL PC-3), the errors can be as high as 6.5 percent.

DISCUSSION

We recommend the following procedures to minimize errors caused by desiccation (especially for low-rank coal):

1. After the ampoules are opened, samples should be stored in a way to minimize dehydration.

- 2. Moisture content of each coal sample should be determined just prior to chemical analysis of whole coal.
- 3. If there is insufficient sample for moisture determination, rehydrate the sample to reduce the error.

The ASTM method for determining equilibrium moisture (ASTM, 1996b, D 1412–93) could be used to rehydrate the sample, but the method requires at least a 20-g sample, more time, and more equipment than the procedure described in this paper.

An alternative method would be to analyze a moisturefree sample by drying it (105°C for 24 hours) prior to analysis. Two assumptions are necessary: (1) that the published moisture value is applicable so that the chemical analysis can be recalculated to an as-received basis, and (2) that the sample does not pick up moisture between drying and weighing for analysis.

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Determination of 62 Elements in 8 Argonne Premium Coal Ash Samples by Automated Semiquantitative Direct-Current Arc Atomic Emission Spectrography

By Carol J. Skeen, B.J. Libby, and W.B. Crandell

ABSTRACT

The automated semiquantitative direct-current arc atomic emission spectrographic method was used to determine concentrations of 62 elements in 8 Argonne Premium Coal ash samples. Ashed samples of all eight coals were analyzed in triplicate to verify precision of the method. The precision for most elements was within ± 10 percent. The accuracy of this method is assumed to be limited to +50 percent or -33 percent because of the nature of the standard curves for each of the elements. Adjustments to the computer program were implemented to account for unique matrix interferences in these particular coal ash samples.

INTRODUCTION

Emission spectrographic analysis is based on the emission of light by atoms and ions returning to ground state after excitation in a direct-current (dc) arc. The light is passed through the slit of a spectrograph and diffracted by a grating. The slit image is focused on a photographic plate. Distinct wavelength positions for each element indicate the presence of that element, and the darkness of the resulting specific slit image is proportional to the concentration of the specific element present.

The automated dc atomic emission spectrographic analysis of a wide variety of geologic materials is a rapid, economical method for evaluating both the major- and trace-element composition. A total of 62 elements can be determined (Dorrzapf, 1973). This computerized procedure is semiquantitative, because it calculates concentrations by using prestored coefficients calculated from previously arced standards.

EXPERIMENTAL

In preparation of the samples for arcing, 15 mg of the ashed sample (-100 mesh) was mixed with 30 mg of graphite and transferred to a graphite crater electrode. The stan-

dard operating procedures for the spectrographic analysis are listed in table 1 (Dorrzapf and others, 1989). The Helz jet was used in preference to a Stallwood jet, because it simplifies the procedure for changing samples (Stallwood, 1954; Shaw and others, 1958; Helz, 1964). An iron bead

Table 1. Photoplate preparation conditions.

[Table modified from Dorrzapf and others (1989, table 4). A, ampere; ASTM, American Society for Testing and Materials; dc, direct current; L, liter; min, minute; s, second; V, volt]

Electrodes Cathode: ASTM type C-	
long. Anode: 6.3-mm d	
walled graphite (Ultra	Carbon no.
3170).	
Electrode charge 15 mg sample + 30 mg gr	aphite (type
UCP–2, 200 mesh).	
Spectrograph 3.4-m Ebert (Jarrell-Ash	Mark III).
Power source	
Excitation 15-A dc arc, set with emp	oty graphite
electrodes.	
Arc gap 4 mm, maintained throug	hout arcing.
Exposure 20 s at 5 A followed by 1	30 s at 15 A,
continuous arcing.	
Atmosphere 70% Ar + 30% O ₂ ; 6.6 L/	min flow rate,
with top of Helz jet not	zzle 2 mm
below top of electrode.	
Wavelength range 230.0-470.0 nm; first ord	er.
Grating 600 grooves/mm; 0.5 nm/	mm recipro-
cal linear dispersion.	
Slit 25 μ m wide and 2 mm hi	gh.
Filter 14% transmission neutral	-density filter
at slit.	
Illumination Arc image focused on col	limator by
450-mm focal-length c	ylindrical
quartz lens at slit.	
Mask at collimator	
Emulsion Eastman-Kodak III-0 (10)	2 ∞ 508 mm
plates).	
Processing Eastman-Kodak D-19 dev	veloper,
3.25 min at 20°C stirre	~
bursts; Eastman-Kodak	indicator stop
bath, 30 s; Eastman-Ko	
10 min; wash, 20–30 n	,
Eastman-Kodak photo-	flow, 1 min:
and dry with warm air	
and dry with walli all	ioi Jo mini.

Table 2. Element concentrations determined by direct-current arc spectrographic analyses of NIST standard reference material coal fly ash.

[Concentrations are in percentages (%) for the first nine elements (Si through Mn) and in micrograms per gram ($\mu g/g$) for the remaining elements (As through Zr). NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards; SD, standard deviation; USGS, U.S. Geological Survey]

	NIST	1633	NIST 1633a			
Element	NIST value ¹ (mean ± SD)	USGS average ²	NIST value ¹ (mean ± SD)	USGS average ³		
Si (%)		27±3.3	22.8±0.8	26±2.6		
Al		17±2.8	14	18±3.1		
Fe		8.0±0.93	9.40±0.10	11±1.7		
Ид		2.2±0.32	.455±0.010	.71±0.14		
Ca		5.4±1.2	1.11±0.01	1.3±0.31		
Na		.25±0.03	.17±0.01	.16±0.03		
.	1.72	1.3±0.15	1.88±0.06	1.5±0.19		
Ti		.80±0.15	.8	.85±0.18		
Mn	.0493±0.0007	.076±0.11	.0190	.026±0.08		
As (µg/g)	61±6	⁴ <100	145±15	190±49		
3	430	440±46		32±4.0		
3a		$1,600 \pm 490$	1,500	910±160		
3e	12	15±1.5	12	14±1.9		
Ce		170±29	180	160±53		
Co	38	38±4.9	46	38±6.2		
Cr	131±2	120±23	196±6	180±33		
Cu	128±5	100 ± 25	118±3	93±21		
Eu		3.1±0.5	4	3.4 ± 0.6		
Ja	49	39±5.8	58	54±11		
_a		96±13				
ло			29	28±6.2		
Nd		60 ±15		100 ± 29		
Ni	98±3	110 ± 15	127±4	140 ± 19		
р	70±4	74 ± 9.0	72.4 ± 0.4	76±12		
Sc		25 ± 4.0	40	29 ± 5.7		
Sr	1,380	$1,700 \pm 30$	830±30	900±140		
7	214±8	200 ± 24	300	240±36		
ζ		53 ± 7.8				
<i>(</i> Ъ		6.6 ± 0.9				
Zr		180 ± 29				

¹Values taken from NIST Certificates 1633 (National Bureau of Standards, 1975) and 1633a (National Bureau of Standards, 1979). Values with no error listed are NIST information values; all others are NIST certified values.

²Average of 35 determinations Feb. to July 1983.

³Average of 31 determinations June 1983 to Mar. 1984.

⁴Lower limit of determination.

was arced first and exposed on the plate for the iron calibration reference spectrum. Then each sample was arced at 5 amperes (A) for 20 seconds and then at 15 A for 130 seconds. For the iron and for each sample and standard, a cadmium lamp was exposed in two windows as a reference to be used on the scanning microphotometer.

The photoplate was developed and processed according to standard U.S. Geological Survey (USGS) procedures (Helz and others, 1969). The intensities of the spectra on the plate were recorded by a scanning microphotometer (Helz, 1965, 1973). The data were processed by a Hewlett-Packard 2100 computer, and a report was generated with the 62-element concentration information. This report was evaluated by the analyst, who verified the values reported (Golightly and others, 1977; Dorrzapf and others, 1989).

Programs for data collection and interpretation that were written (Walthall, 1974) for use on a mainframe computer were adapted for use on a minicomputer system (Thomas, 1979). The specific details about the procedure for calibration and the algorithm used for calculations were summarized by Dorrzapf and others (1989).

RESULTS AND DISCUSSION

For eight Argonne Premium Coal samples, an in-depth study was made for specific interference corrections in addi-

Element	UF PC-1-1	UF PC-1-2	UF PC-1-3	WY PC-2-1	WY PC-2-2	WY PC-2-3	IL PC-3-1	IL PC-3-2	IL PC-3-3
Si (%)	22	22	25	15	15	17	22	20	25
Al	15	19	19	5.0	6.4	7.2	12	12	13
Fe	12	9.9	13	3.0	3.0	3.6	14	14	16
Mg	.98	1.1	1.2	3.9	4.6	4.4	.69	.68	.77
Ca	5.1	4.4	4.7	12	15	17	9.6	9.4	11
Na	.31	.35	.34	1.4	1.9	1.4	.99	.89	.91
K	2.3	2.1	2.2	.36	.44	.36	1.5	1.6	1.7
Ti	.45	.29	.44	.35	.43	.48	.38	.40	.45
P	<.068	<.068	<.068	.24	.31	.39	<.068	<.068	<.068
Mn	.041	.038	.043	.049	.032	.036	.066	.064	.070
Ag (µg/g)	1.4	1.5	1.6	.65	.85	.88	2.2	2.6	2.4
As	230	280	160	<100	<100	<100	<100	<100	100
Au	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8
B	140	350	180	870	740	860	730	680	780
Ba	300	320	290	3,400	3,700	4,200	320	330	470
Be	11	8.9	11	1.6	1.6	1.8	5.5	5.9	6.6
Bi	<10	<10	<10	<10	<10	<10	<10	<10	<10
Cd	<32	<32	<32	<32	<32	<32	<32	<32	<32
Ce	<63	74	<63	74	120	110	<200	<200	170
Co	28	22	29	12	17	16	22	21	26
Cr	160	130	140	56	74	70	220	220	260
Cu	130	120	140	130	160	160	82	69	58
Dy	<22	<22	<22	<22	<22	<22	<22	<22	<22
Er	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6
Eu	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2
Ga	45	47	47	26	31	29	29	27	31
Gd	<32	<32	<32	<32	<32	<32	<32	<32	<32
Ge	26	25	29	<4.6	<4.6	<4.6	52	52	60
Hf	<15	<15	<15	<15	<15	<15	<15	<15	<15
Ho	<68	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8
In	<10	<10	<10	<10	<10	<10	<10	<10	<10
Ir	<15	<15	<15	<15	<15	<15	<15	<15	<15
La	64	39	50	56	77	84	54	55	63
Lu	<15	<15	<15	<15	<15	<15	<15	<15	<15
Mo	12	9.3	12	7.6	7.8	8.2	34	30	32
Nb	12	10	15	14	15	21	12	14	20
Nd	75	<32	37	<68	<68	<68	<68	<68	40
Ni	100	95	110	57	64	64	130	120	150
Os	<15	<15	<15	<15	<15	<15	<15	<15	<15
Pb	77	74	82	31	48	41	87	78	88
Pd	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Pr	<100	<100	<100	<100	<100	<100	<100	<100	<100
Pt	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2
Re	<10	<10	<10	<10	<10	<10	<10	<10	<10
Rh	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2
Ru	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2
Sb	<68	<68	<68	<68	<68	<68	<68	<68	<68
Sc	19	11	16	12	17	18	16	16	17
Sm	<10	<10	<10	<10	<10	<10	<10	<10	<10
Sn	13	12	12	6.4	10	14	10	15	20
Sr	490	420	490	2,000	2,800	3,100	250	230	270
Ta	<320	<320	<320	<320	<320	<320	<320	<320	<320
Tb	<32	<32	<32	<32	<32	<32	<32	<32	<32
Th	<46	<46	<46	<46	<46	<46	<46	<46	<46
Tl	<10	<10	<10	<10	<10	<10	<10	<10	<10
Tm	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6
U	<220	<220	<220	<220	<220	<220	<220	<200	<200
V	130	99	130	86	110	120	140	140	160
W	<15	<15	<15	<15	<15	<15	<15	<15	<15
Y	46	22	36	28	36	42	27	29	32

Yb.....

Zr

96

6.2

90

6.4

6.1

55

140

2.7

200

3.5

250

4.1

3.5 86

140

4.2

3.6 96

Table 3. Element concentrations determined by direct-current arc spectrographic analyses of the Argonne Premium Coal samples. [Concentrations are in percentages (%) for the first 10 elements and in micrograms per gram ($\mu g/g$) for the other 52 elements. H, interference]

Element	PITT PC-4-1	PITT PC-4-2	PITT PC-4-3	POC PC-5-1	POC PC-5-2	POC PC-5-3
Si (%)	26 24	29 21	32 24	18 12	17 14	19 14
Fe	19	20	22	9.8	8.3	10
Mg Ca	.71 5.0	.70 3.6	97 4.7	1.9 10	2.0 11	2.0 12
Na	.42	.36	.40	2.4	2.4	2.5
K	1.4	1.3	1.5	.65	.63	.73
Ti	.64	.57	.64	.53	.47	.58
P Mn	<.068 .033	<.068 .041	<.068 .042	<.068 .045	<.068 .043	<.068 .050
Ag (µg/g)	.95	.94	1.0	1.8	1.8	2.0
As	210	180	160	350	210	270
Au B	<6.8 530	<6.8 580	<6.8 570	<6.8 320	<6.8 310	<6.8 360
Ba	390	370	490	3,600	3,600	4,900
Be	12	13	13	14	11	13
Bi	22	17	<10	<10	<10	<10
Cd Ce	<32 <200	<32 <200	<32 <200	<32 220	<32 190	<32 270
Co	31	25	30	45	44	56
Cr	220	190	220	170	150	180
Cu	58 <22	89 <22	66 <22	420 <22	290 <22	380 <22
Dy Er	<22 <4.6	<22 <4.6	<22 <4.6	<22 <4.6	<22 <4.6	<22 <4.6
Eu	<2.2	<2.2	<2.2	<2.2	<4.2	<2.2
Ga	57	43	54	42	40	47
Gd Ge	<32 13	<32 13	<32 12	<32 8.7	<32 8.2	<32 9.2
Hf	<150	<150	<150	<15	<15	<15
Но	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8
In	<10	<10	<10	<10	<10	<10
lr La	<15 60	<15 66	<15 62	<15 130	<15 130	<15 150
La	<15	<15	<15	<15	<15	<15
Mo	9.4	7.4	9.0	37	41	46
Nb	18	22	18	24	14	22
Nd Ni	47 150	46 130	35 150	100 120	130 110	$\begin{array}{c} 110\\140 \end{array}$
Os	<15	<15	<15	<15	<15	<15
Pb	79	75	84	220	160	220
Pd	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Pr Pt	<100 <2.2	<100 <2.2	<100 <2.2	<100 <2.2	<100 <2.2	<100 <2.2
Re	<10	<10	<10	<10	<10	<10
Rh	<2.2	<2.2	<2.2	<2.2	<2.2	<2.2
Ru Sb	<2.2 <68	<2.2 <68	<2.2 <68	<2.2 <68	<2.2 <68	<2.2 <68
Sc	20	19	22	24	24	28
Sm	<10	<10	<10	14	13	16
Sn	18	11	13	2,800	550	720
Sr Ta	960 <320	900 <320	1,100 <320	2,000 <320	1,900 <320	2,300 <320
Tb	<32	<32	<32	<32	<32	<32
Th	<46	<46 <10	<46 <10	<46	<46	<46
Tl	<10	<10	<10	<10	<10	<10
Tm U	<4.6 <220	<4.6 <220	<4.6 <220	<4.6 <220	<4.6 <220	<4.6 <220
V	130	130	150	130	120	150
W Y	<15 31	<15 36	<15 33	<15 86	<15 81	<15 92
1	31	50	55	00	01	74
Yb	6.9	5.5	5.2	8.7	8.8	7.9

 Table 3.
 Element concentrations determined by direct-current arc spectrographic analyses of the Argonne Premium Coal samples— Continued.

Element	UT PC-6-1	UT PC-6-2	UT PC-6-3	WV PC-7-1	WV PC-7-2	WV PC-7-3	ND PC-8-1	ND PC-8-2	ND PC-8-3
Si (%)	16	20	20	22	30	28	5.7	6.6	6.6
Al	6.8	7.8	6.3	17	22	21	2.2	3.0	2.2
Fe	4.6	5.9	5.1	1.8	2.3	2.1	3.7	4.5	3.5
Mg	.93	.82	.89	.66	.43	.40	5.5	5.4	5.3
Ca	6.0	6.3	6.3	.40	.31	.49	13	14	15
Na	1.9	1.4	1.4	.19	.18	.19	H	2.6	H
K	.47	.31	.36	2.0	1.7	1.8	.40	.35	.41
Ti	.40	.43	.40	.95	.91	.90	.18	.18	.20
P	<.068	<.068	<.068	.086	<.068	<.068	<.13	.11	.098
Mn	.014	.010	.016	.0092	.012	.010	.11	.11	.11
Ag (µg/g)	750	8.9	9.6	2.6	2.2	2.2	<.10	<.10	<.10
As		<220	<220	<220	<220	<220	<220	<220	<220
Au		<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8
B		>1,000	>1,000	150	480	250	500	800	810
Ba		670	900	660	740	640	5,300	7,400	5,700
Be	3.3	5.2	4.4	15	16	16	2.8	3.5	2.4
Bi	<22	<22	<22	<22	<22	<22	<22	<22	<22
Cd	<22	<22	<22	<22	<22	<22	<22	<22	<22
Ce	160	160	130	130	140	130	<63	<63	<63
Co	15	13	14	3630	31	5.7	6.0	5.7	5.7
Cr	110	85	120	220	180	210	25	25	26
Cu	69	52	66	90	85	68	41	40	75
Dy	<10	<10	<10	<10	<10	<10	<10	<10	<10
Er	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6	<4.6
Eu	<.68	<.68	2.5	2.4	2.5	4.6	<.68	<.68	<.68
Ga	32	19	28	68	56	59	19	16	20
Gd	<22	<22	<22	<22	<22	<22	<22	<22	<22
Ge	<4.6	<4.6	<4.6	5.5	<4.6	6.2	<4.6	<4.6	<4.6
Hf	<10	<10	<10	<10	<10	<10	<10	<10	<10
Ho	<1.5	<1.5	<1.5	3.2	<1.5	1.7	<1.5	<1.5	<1.5
In	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8
Ir	<10	<10	<10	<10	<10	<10	<10	<10	<10
La	66	64	62	72	74	64	37	40	39
Lu	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8	<6.8
Mo	12	8.4	11	7.0	8.4	6.3	8.6	8.2	8.7
Nb	10	11	12	13	19	23	6.6	5.3	6.4
Nd	64	<150	<150	97	100	90	<32	<32	<150
Ni	84	89	87	120	100	110	17	17	22
Os	<10	<10	<10	<10	<10	<10	<10	<10	<10
Pb	45	36	43	93	81	69	28	22	32
Pd	<.68	<.68	<.68	<.68	<.68	<.68	<.68	<.68	<.68
Pr	9.5	9.3	8.4	10	9.3	<6.8	<6.8	<6.8	7.3
Pt	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Re	<10	<10	<10	<10	<10	<10	<10	<10	<10
Rh	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5	<1.5
Ru	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0
Sb	<32	<32	<32	<32	<32	<32	<32	<32	<32
Sc	16	15	14	36	31	32	9.2	9.1	10
Sm	8.0	8.2	7.8	8.9	8.7	<3.2	<3.2	<3.2	4.9
Sn	8.8	5.9	9.4	11	8.4	10	<4.6	<4.6	13
Sr	1,800	2,000	2,000	390	320	370	6,200	6300	7300
Ta	<320	<320	<320	<320	<320	<320	<320	<320	<320
Tb	<32	<32	<32	<32	<32	<32	<32	<32	<32
Th	<46	<46	<46	<46	<46	<46	<46	<46	<46
Tl	<10	<10	<10	<10	<10	<10	<1015	<10	<10
Tm	<3.2	<3.2	<3.2	<3.2	<3.2	<3.2	<3.2	<3.2	<3.2
U	<150	<150	<150	<150	<150	<150	<150	<150	<150
V	82	78	79	160	180	150	38	38	37
W	<10	<10	<10	<10	<10	<10	<10	<10	<10
Y	41	42	34	50	52	39	24	27	25
Yb	4.9	4.4	4.6	9.2	7.3	7.1	2.3	2.0	2.1
Zr	260	280	200	200	150	140	130	130	140

 Table 3.
 Element concentrations determined by direct-current arc spectrographic analyses of the Argonne Premium Coal samples— Continued.

tion to those that were automatically performed. Because of heterogeneity problems associated with barium, coupled with the suppression of the most sensitive analytical lines for barium due to the matrices of these coals, values from less sensitive lines were used. The values for strontium were improved after correcting for interferences from iron and nickel.

This semiquantitative approach achieves ranges and detectability comparable to those of the visual estimation procedure (Myers and others, 1961). Because the standards used do not closely match the approximate composition of the ash samples to be analyzed, the expected accuracy is limited to ± 1 step, which corresponds to roughly +50 percent or -33 percent of the reported value.

In the analyses of these Premium Coals, the National Institute of Standards and Technology (NIST) standard reference materials NIST 1633 and NIST 1633a (National Bureau of Standards, 1975, 1979) were included as control samples for evaluation of both precision and accuracy of this method. Table 2 lists the concentrations provided by NIST certificates, the mean concentrations determined over a 5-month period, and the associated relative standard deviations. Compositions determined for coal ashes are within the limits of precision and accuracy for which the method was designed. The high relative standard deviations for barium and zirconium indicate the heterogeneity documented for these reference materials (Filby and others, 1985). Heterogeneity possibly explains the large range of values for barium and zirconium in the Premium Coals as well as the interferences discussed above.

Table 3 lists data for all 62 elements in these eight Argonne Premium Coals. The "less than" symbol indicates that the concentration is less than the lower limit for a value that can be determined for that element, and the "greater than" symbol indicates that the concentration is greater than the highest value that can be reliably determined for that element. An "H" denotes the occurrence of an unresolved interference. Major elements are reported in percentages, and the trace elements in micrograms per gram.

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Determination of 18 Elements in 5 Whole Argonne Premium Coal Samples by Quantitative Direct-Current Arc Atomic Emission Spectrography

By Janet D. Fletcher and Carol J. Skeen

ABSTRACT

Quantitative multiple-element analysis of whole Argonne Premium Coal samples by direct-current arc atomic emission spectrography is possible with the use of a lithium carbonate buffer. Two spectrographic methods are described for the determination of 18 trace elements in 100mg samples of coal. Overall concentrations for calibration standards range from a low of 2 μ g/g to a high of 3 weight percent. For concentrations well above the lower determination limit, the typical accuracy is within ±20 percent, and the general precision of the method is ±10 percent.

INTRODUCTION

Most atomic-spectroscopic methods are designed for the analysis of ash from pulverized coals that have been oxidized at 500°C or 750°C (ASTM, 1984). The direct-current (dc) arc functions well for ash that is mixed with graphite powder, and many elements are effectively preconcentrated by the ashing process, thus providing improved detectability (Dorrzapf, 1973), but certain elements associated with organic phases, such as porphyrins, organometallics, or acid salts, may be volatilized and lost during the ashing process. The elements Ag, B, Ga, Ge, Mo, Ni, and Ti, which potentially can be determined by dc arc spectroscopy, are at least partially associated with organic phases in coals (Ruch and others, 1974; Gluskoter and others, 1977; Finkelman, 1980). Direct multiple-element analysis of whole coals circumvents the long intervals required for ashing, the losses due to volatilization, and the further exposure of samples to possible contamination. This paper describes the two directcurrent arc atomic emission spectrographic (DCAES) methods that have produced accurate determinations of 28 elements in the pulverized whole coal (Fletcher and Golightly, 1985). In this study, only 18 elements were determined in 5 Argonne Premium Coals. These methods, which have been applied principally to the analysis of coal microlithotypes, offer the basis for efficient, low-cost, multiple-element analysis of whole coals.

EXPERIMENTAL

Approach

A principal difficulty encountered in attempts to arc small quantities of pulverized coal directly is the rapid evolution of gases that occurs immediately following initiation of the arc and on the subsequent burning of the organic phases that remain in a cup-shaped electrode (anode). The rapidly evolved gases usually blow material from the anode cup, thus creating uncontrolled losses of the previously weighed sample, and the erratic flaming of the organic phase can produce unwanted spectral bands from carbonbased free radicals. These events constitute irreproducible processes that control the transport of material from the hot anode cup into the arc discharge. Such severe problems related to the arcing process have been solved by mixing powdered coal with a lithium carbonate buffer. This controls sample transport and excitation conditions in the arc column and greatly diminishes the possibility for flaming of the hot coal dissociation products. With these important aspects of arcing well controlled for coal samples, the methodology for dc arc spectrographic analysis becomes quite conventional.

Method

Preparation of the Samples

Splits of 100 mg of each whole Argonne Premium Coal sample, 100 mg of lithium carbonate, and 50 mg of pure graphite powder were thoroughly mixed and ground with an agate mortar and pestle to obtain a final homogeneous mixture. For samples that had especially high concentrations of analyte elements, a higher weight ratio of lithium carbonate to sample was necessary, but the ratio was no greater than 10:1. Twenty-five milligrams of the final homogeneous mixture was transferred into the appropriate graphite electrode and firmly tamped (Dorrzapf, 1973). These filled electrodes were dried in an oven at 110°C for 4 hours immediately before arcing. The drying step was necessary because it removes water and other readily volatilized com
 Table 1. Wavelengths, spectral lines, and determination limits

 for elements that can be determined by the direct-current arc

 atomic emission spectrography described in this paper.

Element	Wavelength, (nm) ¹	Spectrum ¹	Determination limits (%) ²
G	roup I, invola	atile element	s
Al	265.248	Ι	0.01-2.0
	266.039	Ι	.01-2.0
В	249.773	Ι	.0051
Ba	455.403	II	.00205
Ca	315.887	II	.01–3
~	422.673	I	.00501
Co	345.350	Ι	.000205
Cr	302.156	Ι	.002–.2
	425.435	Ι	.000202
Cu	327.396	Ι	.000202
Fe	259.837	II	.05–3
	302.107	Ι	.002-1.0
Mg	277.983	Ι	.01-0.2
	285.213	Ι	.000202
Mn	279.482	Ι	.00021
	279.827	Ι	.0002–.2
Mo	317.035	Ι	.0005–.2
Nb	316.340	II	.002–.1
Ni	305.082	Ι	.0021
	341.476	Ι	.000205
	349.296	Ι	.01–.2
Si	251.920	Ι	.2–4
Ti	308.940	II	.0022
	316.257	II	.005–.2
V	318.341	Ι	.00051
Zr	327.926	II	.001–.2
	Group II, vol	atile elemen	ts
Ag	338.289	Ι	.0002–.02
As	278.020	Ī	.022
Bi	306.772	Ĩ	.0002005
Cd	326.106	Ĩ	.00021
Ga	294.364	Ĩ	.000205
Ge	265.118	Ĩ	.000505
	303.906	Ĩ	.0002005
Hg	253.652	Ī	.002–.1
DL	202.206	Ť	0002 01

¹Wavelength and spectrum are from the NIST Wavelength Tables (Meggers and others, 1975). I line emitted by normal atom; II line emitted by singly ionized atom.

I

I

I

I

.0002–.01

.0002-.02

.0002 - .02

.001-.05

283.306

317.505

276.787

334.502

Pb

Sn

T1.....

Zn

²Lower and upper limits for each element are in units of weight percent.

ponents that could cause the loss of sample material from the anode just after initiation of the arc discharge.

Preparation of Standards

Calibration standards consisted of homogeneous mixtures of oxides and carbonates of the analyte elements in a lithium carbonate matrix. Dilutions of commercially available standards, which contain 43 elements in lithium carbonate (Spex Industries, Metuchen, New Jersey), provided calibration standards for the concentration range from 1 to 1,000 μ g/g for each element of interest. Individual standards were diluted on a weight-weight basis with high-purity lithium carbonate (<10 μ g/g total impurities).

Reference standards were prepared from the National Institute of Standards and Technology (NIST), formerly the National Bureau of Standards (NBS), coal reference materials NIST 1632, NIST 1632a, and NIST 1635 (NBS, 1974, 1978a,b), which were diluted with lithium carbonate in the same fashion as the samples. Drying and handling of NIST standards followed the procedure used for samples.

Arcing of Samples and Standards

All samples and standards were arced in an argon-oxygen, or argon, laminar stream that is concentric to the anode and is introduced through an alumina nozzle arrangement known as a Helz jet (Helz, 1964). Both the arcing conditions and the atmosphere were chosen to give complete volatilization of analyte elements from the anode cup into the arc column and to effectively excite those atomic energy levels giving the spectral lines listed in table 1, without causing high spectral background. For the volatile elements (group II, table 1), the objective was to vaporize and to excite these elements over a relatively long interval while distilling insignificant amounts of matrix elements into the arc column. The present method was one adapted from that of Annell (1967) for volatile elements in silicate and carbonate rocks. For elements in chemical forms that exhibit low volatility (group I, table 1), total vaporization of each sample into the arc column was necessary for an accurate determination.

Complete details on the spectrographic equipment and the conditions for arcing samples and for making the necessary measurements are given in table 2. Maintaining a 4mm gap between the tips of the electrodes was essential to the achievement of the accuracy and precision that this approach is capable of producing.

ACCURACY AND PRECISION

The accuracy of analysis by DCAES is dependent on the successful element-by-element calibrations of an instrument with standard materials that closely resemble the materials to be analyzed. For coals, the effective matrix of the "arced sample" was modified through the use of a lithium carbonate buffer. This modification of the sample matrix made the arced sample resemble the lithium carbonate matrix of the Spex calibration standards. The quantity of lithium carbonate relative to that of the sample was sufficient to control the fusion, vaporization, transport, and excitation processes. The concentration ranges for the elements

Table 2.	Equipment and operating conditions for direct-current arc atomic emission spectrography.
[A, ampere	; ASTM, American Society for Testing and Materials; dc, direct current; min, minute; s, second; T, transmission; V, volt]

Subject	Group I, involatile elements	Group II, volatile elements
Sample	100 mg pulverized whole coal mixed with 100 mg Li ₂ CO ₃ powder (American Potash) and 50 mg graphite powder (-200 mesh, Ultra Carbon); 25 mg of mixture tamped into anode cup.	100 mg pulverized whole coal mixed with 100 mg Li ₂ CO ₃ powder (American Potash) and 100 mg graphite powder (-200 mesh, Ultra Carbon); 50 mg of mixture tamped into anode cup.
Spectrograph	Ebert mounting, 3.4-m focal length, 0.5-nm/mm reciprocal linear dispersion in first order. Grating: 600 grooves/mm, blazed for 300 nm.	Eagle mounting, 3-m focal length, 0.55-nm/mm reciprocal linear dispersion in first order. Grating: 590 grooves/mm, blazed for 300 nm.
Slit	$25 \ \mu m \propto 2 \ mm$ for spectra from standards and samples; $25 \ \mu m \propto 4 \ mm$ for spectra from iron arc.	25 μ m ∞ 2.5 mm for spectra from standards and samples; 25 μ m ∞ 5 mm for spectra from iron arc.
Wavelength range	240–360 nm, first order	250-340 nm, second order.
Illumination	Arc image focused on collimator mirror by a 450- mm focal length cylindrical quartz lens located at the entrance slit.	Arc image focused on grating by a 450-mm focal length cylindrical quartz lens located near the entrance slit.
Filters	Neutral density, 35% T plus 75% T, for exposures of samples and standards. Two-step neutral- density filter, 40% T: 100% T, for iron arc expo- sures used in calibration of the photographic emulsion.	None for exposures of samples and standards. Two- step neutral-density filter, 50% T: 100% T, for iron arc exposure used in calibration of the pho- tographic emulsion.
Electrodes	Cathode: Graphite rod (Ultra Carbon no. 5001) 3.2 mm in diameter and 3.8 cm long . Anode: Thin-walled graphite electrode (Ultra Car- bon no. 1590) 3.6 mm in diameter.	Cathode: Same as for group I. Anode: Graphite electrode (Ultra Carbon no. 3170) 5.7 mm in diameter .
Excitation	 Arc current: Stepped arc current, 5 A dc for 10 s, 15 A dc for 95 s, across constant 4-mm arc gap. Voltage source of 300 V, open circuit. Electrode supporting the sample is the anode. Atmosphere: 80% Ar, 20% O₂; 6.6 L/min through Helz jet (Helz, 1964). 	 Arc current: Stepped arc current, 8 A dc for 10 s, 25 A for 110 s, across constant 4-mm arc gap. Voltage source of 300 V, open circuit. Electrode supporting the sample is the anode. Atmosphere: Ar, 6.6 L/min through Helz jet (Helz, 1964).
Photography	Eastman-Kodak III-0 emulsion on 101-∞ 254-mm glass substrate. Emulsion was processed in Eastman-Kodak D-19 developer for 3 min (20°C), short stop solution for 30 s, and fixer for 10 min. Then, the plate was washed in tap water for 10 to 20 min, allowed to drain, and dried with warm air for 5 min.	Same as for group I elements.
Microphotometry	All microphotometry was done by conventional methods, such as those described by ASTM (ASTM, 1971, p. 79–80).	Same as for group I elements.

determined by the dc arc spectrographic methods described in this work are summarized in table 1. Elements exhibiting the largest deviations are aluminum, calcium, manganese, and silicon. Experience in the analyses of other coals, vitrinites, exinites, and inertinites indicates that the deviations for the elements observed here are random, rather than systematic. Measurement errors for the spectrographic method for concentrations well above (>5 times) the determination limits are typically ± 20 percent, and the precision of the method is ± 10 percent.

RESULTS AND DISCUSSION

Only five of the eight Argonne Premium Coals were available for analysis when this method was developed. Because this method is labor intensive and the accuracy and precision for this method at the detection limits for a majority of the elements are no better than the other methods implemented in the analysis of these coals, the analyses of the other three coals were not carried out.

Only 18 elements were determined in the 5 whole Argonne Premium Coals. The results of these analyses are

Sample	Mg	Al	Ca	Ti	Mn	Si	Fe
UF PC-1-1	0.097	2	0.57	0.10	0.0053	3	2
UF PC-1-2	.10	2	.54	.10	.0059	4	3
UF PC-1-3	.10	2	.63	.10	.0048	3	2
WY PC-2-1	.20	.47	1	.048	.0010	.80	.34
WY PC-2-2	.20	.54	1	.046	.0016	1.0	.33
WY PC-2-3	.20	.44	1	.038	.0014	.52	.20
IL PC-3-1	.096	2	2	.089	.015	3	2
IL PC-3-2	.099	2	2	.10	.015	4	2
IL PC-3-3	.10	2	1	.086	.017	3	3
PITT PC-4-1	.045	1	.24	.086	.0016	2	1
PITT PC-4-2	.048	2	.34	.090	.0028	3	1
PITT PC-4-3	.052	$\overline{2}$.28	.097	.0018	2	2
POC PC-5-1	.079	.80	.69	.084	.0016	.83	.60
POC PC-5-2	.074	.92	.93	.084	.0024	1.0	.80
POC PC-5-3	.079	.84	.79	.078	.0017	.80	.68

Table 3. Concentrations (in weight percent) of 7 elements in Argonne Premium Coals determined by directcurrent arc atomic emission spectrography.

 Table 4.
 Concentrations (in micrograms per gram) of 11 elements in Argonne Premium Coals determined by direct-current arc atomic emission spectrography.

Sample	V	Cr	Ni	Cu	Zn	Ga	Co	Ge	Zr	Ba	Pb
UF PC-1-1	38	32	22	20	30	11	5	4	36	88	8
UF PC-1-2	45	38	29	18	30	11	5	4	29	72	10
UF PC-1-3	41	26	24	16	40	8	5	3	34	84	6
WY PC-2-1	10	8	<10	5	<20	4	<5	<2	<20	200	<2
WY PC-2-2	14	6	<10	10	<20	2	<5	<2	<20	260	<2
WY PC-2-3	17	8	<10	6	<20	3	<5	<2	<20	300	<2
IL PC-3-1	52	54	28	9	250	8	<5	8	22	140	9
IL PC-3-2	59	57	35	10	260	5	<5	7	25	120	8
IL PC-3-3	44	47	37	9	320	7	<5	12	16	92	6
PITT PC-4-1	20	23	13	5	20	6	<5	3	23	66	4
PITT PC-4-2	25	27	14	6	30	5	<5	<2	21	72	3
PITT PC-4-3	27	21	11	7	30	5	<5	<2	28	60	5
POC PC-5-1	16	17	11	18	<20	4	5	<2	25	400	5
POC PC-5-2	21	17	11	13	<20	3	5	<2	34	460	4
POC PC-5-3	18	14	<10	27	<20	4	5	<2	32	500	5

shown in tables 3 and 4. The determination of silver, boron, and molybdenum required special treatment and preparation time, and so they were eliminated from the routine for analysis. As, Bi, Cd, Hg, Nb, Sn, and Tl were not determined because of the nature of the matrix of these particular coals.

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Determination of Major and Trace Elements in Eight Argonne Premium Coal Samples (Ash and Whole Coal) by X-Ray Fluorescence Spectrometry

By John R. Evans, George A. Sellers, Robert G. Johnson, Davison V. Vivit, and Judy Kent

ABSTRACT

X-ray fluorescence (XRF) spectrometric methods were used in the analysis of eight Argonne Premium Coal samples. Trace elements (Cr, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Ba, La, and Ce) in both coal ash and whole coal were determined by energy-dispersive X-ray fluorescence spectrometry. Major elements (Na, Mg, Al, Si, P, S, K, Ca, Ti, Mn, and Fe) in coal ash and trace elements (Cl and P) in whole coal were determined by wavelength-dispersive X-ray fluorescence spectrometry. The experimental XRF methods and procedures used to determine these major and trace elements are described.

INTRODUCTION

Energy-dispersive X-ray fluorescence (EDXRF) spectrometry and wavelength-dispersive X-ray fluorescence (WDXRF) spectrometry are used routinely in the determination of major and trace elements in silicate rocks (Norrish and Hutton, 1969; Johnson, 1984); however, the analysis of whole coals by XRF spectrometric techniques is more difficult because of the problem of the very light coal matrix and the scarcity of reliable coal standards. Because coal ash is more similar to silicate matrix rocks, EDXRF and WDXRF techniques developed for silicates can be used for the determination of major and trace elements in coal ash samples.

The rapidity, sensitivity, accuracy, and precision of Xray fluorescence spectrometric methods are well documented for a wide range of geologic materials (Rose and others, 1963; Norrish and Hutton, 1969; Johnson, 1984; Johnson and others, 1986; Johnson and Fleming, 1987; Evans and Jackson, 1989). Analysis of whole coal by XRF spectrometric techniques has also proven to be successful in many studies (Kuhn and others, 1975; Johnson and others, 1989). Therefore, the determinations of major and minor elements in eight Argonne Premium Coal samples by XRF spectrometric techniques contributed an important part of the geochemical data base compiled for these materials. This study was not intended to include interpretations of the differences of behavior between various coal ranks of the samples studied.

EXPERIMENTAL

Coal Ash-EDXRF

All the Argonne Premium Coal samples were first ashed at 525°C. This is a lower temperature than prescribed by the American Society for Testing and Materials (ASTM, 1996) method (750°C); however, our method eliminated all combustible material while retaining the same or higher concentrations of volatile material. Sample preparation of coal ash samples followed procedures described in other publications (Johnson, 1984; Johnson and others, 1986; Evans and Jackson, 1989). A Kevex 700 EDXRF spectrometer with a Kevex 8000 analyzer was used to fluoresce coal ash samples powdered to approximately 100 mesh. These powders were pressed into cups made of Mylar film (6.35 μ m) pulled tightly over an aluminum ring with a Teflon collar. The resultant surface appears to be planar.

Appropriate secondary targets were used (table 1). Each sample was fluoresced, and intensity measurements were determined after making background and spectral overlap corrections. The ratio of the analyte line intensity to the secondary target Compton scatter intensity was used in determining elemental concentrations. The Compton ratio method corrects for matrix effects, particle size variations, packing density variations, heterogeneity effects, instru-

Table 1.Secondary targets used for EDXRFanalysis.

Element	Secondary target
Cr	Fe
Ni, Cu, Zn	Ge
Rb, Sr, Y, Zr, Nb	Ag
Ba, La, Ce	Gd

mental fluctuations, and other sources of error inherent in EDXRF determinations.

Trace-element concentrations for coal ash samples were determined from calibration graphs that were constructed by plotting intensity ratio versus the known concentrations for a selected set of standard reference materials (Abbey, 1983).

Whole Coal-EDXRF

Whole-coal samples were prepared by using procedures similar to those described for EDXRF analyses of the coal ash (Johnson, 1984; Johnson and others, 1986; Evans and Jackson, 1989). All intensity measurements were made on a Kevex 700 spectrometer with a Kevex 7000 analyzer. Each whole-coal sample was fluoresced using a secondary target (table 1). Corrections for background interferences and spectral line overlaps were made before integration of the analyte line intensity. Trace elements in whole coal samples were determined by EDXRF by using interelement influence coefficients calculated from fundamental parameters (Johnson and Fleming, 1987). Characterizations of the coal samples by other analytical techniques must be made before trace-element determinations can be obtained with this method. Even though carbon, hydrogen, nitrogen, and oxygen constitute the largest percentage of the whole coal, these elements have very little bearing on absorption and enhancement effects. Major-element concentrations, as determined from the coal ash (see the next section, "Coal Ash-WDXRF"), identify the most important influences on absorption and enhancement effects necessary to generate accurate interelement influence coefficients from the fundamental parameters algorithm.

The complexities of the fundamental parameters algorithms used in this study are beyond the scope of this paper. Detailed explanations of all equations and variables inherent in the matrix correction procedures were given by Sherman (1959), Rousseau (1984a,b), and Johnson and Fleming (1987).

The lack of a sufficient number of whole-coal standards and the ultimate degradation over time of these standards are major difficulties involved in the characterization of coals. For these reasons, it is not possible to construct routine calibration graphs of standard reference materials for elements of interest; therefore, we must use the fundamental parameters algorithm. Using this algorithm allows the investigator to make accurate trace-element determinations in whole coal with as few as one well-characterized standard. In this study, the National Institute of Standards and Technology (NIST), formerly the National Bureau of Standards (NBS), whole-coal reference materials, NIST 1632a and NIST 1632b (NBS, 1978a, 1985) were used to calculate pure element intensities.

Coal Ash-WDXRF

The fusion method was used to produce glass disks of coal ash samples (Johnson and others, 1989). This method eliminates the need for matrix correction routines, since the significant dilution of the sample by the flux corrects for heterogeneity effects, particle size variations, and other sources of error from instrumental fluctuations. A sample/flux ratio is chosen to yield linear calibration curves over the range of concentrations found in both samples and standards, without the use of a heavy absorber, such as La_2O_3 . This sample/flux ratio is needed because of the low final concentrations of sample components in the sample/flux mix.

A 1:9 dilution of sample to flux is obtained by mixing 0.600 g of the coal ash with 5.400 g of a 2:1 mixture of lithium tetraborate to lithium metaborate. This mixture is carefully transferred to a platinum-gold crucible, and three drops of a 15 percent hydrobromic acid solution are added as a wetting agent. An automatic Claisse fluxer is used to heat/mix the sample to temperatures reaching 1,200°C for approximately 20 minutes. After the sample cools to room temperature, a thin glass disk with a planar analytical surface is produced, which is adequate for WDXRF analysis. Elemental intensity measurements are made on a Diano XRD-8300 wavelength-dispersive X-ray fluorescence spectrometer.

Standards used in the construction of calibration graphs were silicate matrix materials selected from those tabulated by Abbey (1983). Because the coal ash matrix closely resembles silicate materials in composition, calibration graphs obtained from silicate standard reference materials are reliable for major-element determinations in coal ash. Standards are prepared for WDXRF analysis in a manner identical to that described above. Calibration graphs were constructed by plotting the analyte intensity with the known concentration for a selected set of standard reference materials for each element of interest. The intensities for the major elements in the coal ash samples were then used in the individual calibration graphs.

A set of synthetic silicate standards was spiked with sulfur before fusion because the chemical matrix of typical silicate rock standards does not have sulfur concentrations similar to those in the coal ash matrix. Because some sulfur is volatilized during fusion, a portion of the fused standard was analyzed by a LECO sulfur analyzer to determine the actual sulfur concentration in the standard. The sulfur determinations of the standards were used to prepare calibration graphs like those described above for the silicate matrix materials tabulated by Abbey (1983).

Whole Coal-WDXRF (Determination of Cl and P)

Briquettes of the whole-coal samples were produced by mixing 0.500 g of the coal with 0.500 g of microgranular

Table 2. Major-oxide concentrations (in weight percent) in coal ash determined by WDXRF (reported on an ash basis).

 [nd, not determined]

Coal sample	Na ₂ O	MgO	Al_2O_3	SiO_2	P_2O_5	SO_3	K ₂ O	CaO	TiO ₂	MnO	Fe ₂ O ₃
UF PC-1-1	0.3	1.0	21.6	41.9	0.1	1.0	2.4	4.1	1.0	<0.1	21.0
UF PC-1-2	.2	1.0	21.6	42.7	.1	nd	2.5	4.2	1.0	<.1	19.9
UF PC-1-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
WY PC-2-1	1.6	4.9	15.0	31.0	.7	6.4	.4	18.1	1.0	<.1	5.6
WY PC-2-2	1.3	5.2	14.9	31.5	.7	9.9	.4	18.9	1.0	<.1	5.4
WY PC-2-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
IL PC-3-1	1.1	.9	14.1	40.2	.1	4.0	1.5	7.8	.7	<.1	23.9
IL PC-3-2	.8	.9	14.0	40.2	.1	3.6	1.5	7.7	.7	<.1	23.8
IL PC-3-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
PITT PC-4-1	.6	.6	20.3	45.4	.2	2.0	1.5	3.1	1.1	<.1	22.6
PITT PC-4-2	.4	.6	20.0	45.8	.2	1.9	1.5	3.0	1.1	<.1	22.2
PITT PC-4-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
POC PC-5-1	1.9	2.1	19.6	32.7	.1	6.6	.7	11.7	1.2	<.1	14.0
POC PC-5-2	1.8	2.1	19.5	32.2	.1	6.0	.7	11.7	1.2	<.1	14.1
POC PC-5-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
UT PC-6-1	3.8	1.4	15.2	41.5	.1	8.3	.6	12.2	.8	<.1	9.0
UT PC-6-2	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
UT PC-6-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
WV PC-7-1	.3	.7	30.8	54.0	.1	.7	2.9	.4	2.1	<.1	2.8
WV PC-7-2	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
WV PC-7-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
ND PC-8-1	7.1	7.3	9.0	17.3	.4	20.5	.4	22.6	.3	.1	6.7
ND PC-8-2	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
ND PC-8-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd

cellulose for 10 minutes on a shaker mill and subsequently pressing the mixture against a fibrous cellulose backing at 276 MPa for approximately 30 seconds (Johnson and others, 1989).

The difficulties experienced in the analysis of whole coals by EDXRF also apply for WDXRF. Reliable whole-coal standard reference materials are scarce. Because these standards are not commercially available, synthetic standards as well as coal samples characterized by other laboratories were used. Only three NIST coal standards were used in this study: NIST 1633, 1633a, and 1635 (NBS, 1975, 1979, and 1978b). Spiked graphite samples with varying concentrations of chlorine and phosphorus served as the synthetic whole-coal standards. All standards were prepared identically to those for the whole-coal samples. Intensity measurements for chlorine and phosphorus were made on a Diano XRD-8300 wavelength-dispersive X-ray fluorescence spectrometer.

Calibration graphs were constructed by plotting the analyte intensity versus the known concentration for a set of standards. The intensities for chlorine and phosphorus in the whole-coal samples were then used to calculate chlorine and phosphorus concentrations from the regression curves.

RESULTS AND DISCUSSION

In this study, eight Argonne Premium Coal samples were analyzed by EDXRF and WDXRF spectrometry. Determinations of major oxides in coal ash are detailed in table 2; trace elements in coal ash in table 3; chlorine and phosphorus oxide in whole coal in table 4; and trace elements in whole coal in table 5. The precision and accuracy for the analysis of coal ash samples by EDXRF and WDXRF closely approximate the precision and accuracy for the analysis of silicates. A study by Johnson and others (1989) estimated an average relative difference of ± 2 to ± 5 percent for WDXRF determinations of major elements (Na, Mg, Al, Si, P, S, K, Ca, Ti, Mn, and Fe) in coal ash samples. Trace-element determinations (Cr, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Ba, La, and Ce) by EDXRF for silicate rocks were estimated to have an accuracy of $<\pm 5$ percent for the ratiocalibration graph method (Johnson, 1984). This level of accuracy is also expected for EDXRF trace-element determinations on coal ash.

The precision and accuracy of the EDXRF and WDXRF analyses of whole-coal samples were more difficult to estimate, since a wide range of acceptable standards

Table 3. Trace-element concentrations (in micrograms per gram, $\mu g/g$) in coal ash determined by EDXRF (reported on an ash basis). [nd, not determined]

Coal sample	Cr	Ni	Cu	Zn	Rb	Sr	Y	Zr	Nb	Ba	La	Ce
UF PC-1-1	186	104	146	162	160	440	74	205	20	380	52	112
UF PC-1-2	196	118	152	142	154	450	72	196	16	405	60	136
UF PC-1-3	196	84	158	150	158	450	84	205	20	370	70	130
WY PC-2-1	86	50	142	114	40	3,100	40	260	14	3,200	<30	72
WY PC-2-2	96	46	158	130	40	3,000	42	275	12	3,300	<30	50
WY PC-2-3	108	44	154	130	44	3,100	44	265	10	3,100	<30	<30
IL PC-3-1	265	88	66	984	90	188	30	146	12	465	<30	42
IL PC-3-2	265	122	58	737	90	178	20	130	<10	460	<30	52
IL PC-3-3	270	120	54	750	92	188	26	136	14	455	<30	40
PITT PC-4-1	200	98	70	92	102	760	56	225	22	370	<30	80
PITT PC-4-2	190	96	66	92	94	740	56	230	20	410	50	108
PITT PC-4-3	195	112	66	88	100	710	48	215	20	385	46	104
POC PC-5-1	235	132	240	82	50	2,100	134	320	16	3,200	3	100
POC PC-5-2	225	146	245	100	40	2,100	126	315	16	3,600	50	162
POC PC-5-3	235	150	225	88	42	2,100	130	320	16	3,500	44	168
UT PC-6-1	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
UT PC-6-2	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
UT PC-6-3	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd	nd
WV PC-7-1	238	91	130	64	218	393	110	412	40	567	75	145
WV PC-7-2	234	83	115	57	232	402	99	411	43	577	91	163
WV PC-7-3	237	87	120	59	224	411	118	421	46	552	96	154
ND PC-8-1	26	25	36	61	nd	6,700	nd	68	<10	5,000	nd	nd
ND PC-8-2	28	17	35	54	nd	6,700	nd	74	<10	4,900	nd	nd
ND PC-8-3	29	19	34	55	nd	6,700	nd	47	<10	4,700	nd	nd

Table 4. Chlorine and phosphorus oxide concentrations (in weight percent) in whole coal determined by WDXRF.

Coal sample	Cl	P_2O_5	Coal sample	Cl	P ₂
UF PC-1-1	0.15	<0.01	POC PC-5-1	0.16	<0.
UF PC-1-2	.14	<.01	POC PC-5-2	.16	<.
UF PC-1-3	.15	<.01	POC PC-5-3	.16	<.
WY PC-2-1	<.01	.06	UT PC-6-1	<.01	<.
WY PC-2-2	<.01	.05	UT PC-6-2	<.01	<.
WY PC-2-3	<.01	.06	UT PC-6-3	<.01	<.
IL PC-3-1	.05	<.01	WV PC-7-1	.04	<.
IL PC-3-2	.05	<.01	WV PC-7-2	.04	<.
IL PC-3-3	.06	<.01	WV PC-7-3	.05	<.
PITT PC-4-1	.06	.02	ND PC-8-1	<.01	
PITT PC-4-2	.06	.02	ND PC-8-2	<.01	
PITT PC-4-3	.07	.02	ND PC-8-3	<.01	

Table 5. Trace-element concentrations (in micrograms per gram, $\mu g/g$) in whole coal determined by EDXRF.

[nd, not determined]

Coal sample	Cr	Ni	Cu	Zn	Rb	Sr	Y	Zr	Nb	Ba	La	Ce
UF PC-1-1	15	27	20	35	23	61	nd	24	nd	54	31	27
UF PC-1-2	23	22	18	32	21	63	nd	22	nd	68	9	46
UF PC-1-3	19	21	17	31	20	61	nd	24	nd	59	9	48
WY PC-2-1	3	9	18	26	14	292	nd	19	nd	404	4	18
WY PC-2-2	5	7	17	29	14	300	nd	21	nd	407	4	21
WY PC-2-3	nd	8	17	21	11	291	nd	17	nd	415	11	35
IL PC-3-1	29	33	15	137	22	33	nd	24	nd	112	16	23
IL PC-3-2	36	30	13	105	17	30	nd	22	nd	122	15	39
IL PC-3-3	41	33	15	186	21	35	nd	24	nd	129	15	64
PITT PC-4-1	6	17	10	17	8	59	nd	15	nd	46	23	19
PITT PC-4-2	10	11	9	14	8	61	nd	17	nd	22	1	25
PITT PC-4-3	9	15	10	15	8	69	nd	18	nd	19	2	6
POC PC-5-1	5	11	16	14	5	86	nd	10	nd	134	1	4
POC PC-5-2	7	9	16	14	4	77	nd	8	nd	124	4	19
POC PC-5-3	4	12	19	15	5	93	nd	11	nd	164	10	8
UT PC-6-1	<10	5	8	<2	3	58	2	20	<10	28	9	20
UT PC-6-2	<10	5	8	<2	2	65	<2	17	<10	32	9	13
UT PC-6-3	<10	5	8	<2	3	66	<2	16	<10	28	11	12
WV PC-7-1	50	17	30	9	43	87	19	106	13	230	15	64
WV PC-7-2	48	18	30	10	43	87	20	102	<10	202	18	54
WV PC-7-3	48	19	32	10	40	83	17	101	13	200	14	48
ND PC-8-1	<10	4	8	5	4	800	<2	20	<10	1000	8	21
ND PC-8-2	<10	10	10	4	5	802	<2	19	<10	1040	6	19
ND PC-8-3	<10	4	10	6	3	753	<2	17	<10	1060	10	17

was not available. However, Johnson and others (1989) estimated the average relative difference for chlorine and phosphorus oxide determinations on whole coals to be ± 10 percent. Trace-element (Cr, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Ba, La, and Ce) determinations on the whole coal by EDXRF generally show close agreement (± 10 percent) between replicate samples. A wide variance was noted, however, when the whole-coal trace-element results were compared with the results obtained on the coal ash. Further investigation is needed to evaluate more clearly the accuracy of the matrix correction method for whole coals.

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Determination of 29 Elements in 8 Argonne Premium Coal Samples by Instrumental Neutron Activation Analysis

By Curtis A. Palmer

ABSTRACT

Twenty-nine elements have been determined in triplicate splits of the eight Argonne Premium Coal samples by instrumental neutron activation analysis. Data for control samples NIST (National Institute of Standards and Technology) 1633 (fly ash) and NIST 1632b (bituminous coal) are also reported. The factors that could lead to errors in analysis of these samples, such as spectral overlaps, low sensitivity, and multiple sources of interfering nuclear reactions, are discussed.

INTRODUCTION

The U.S. Geological Survey (for example, Zubovic and others, 1979, 1980; Oman and others, 1981; Currens and others, 1986, 1987) and other laboratories (for example, Gluskoter and others, 1977) have used instrumental neutron activation analysis (INAA) for the determination of major, minor, and trace elements in thousands of coal samples. The application of INAA for the analysis of coal has been described in several papers (for example, Block and Dams, 1973; Ondov and others, 1975; Rowe and Steinnes, 1977a,b; Swaine, 1985; Palmer and Baedecker, 1989). The analysis of coal by INAA is especially useful because determinations are made on the whole coal in contrast to other techniques in which the ash is used as the sample matrix. Therefore, INAA can be used to measure elements that might be volatilized during ashing, such as bromine. All elements are determined on the same sample split so that elein understanding ment ratios used geochemical environments are not affected by inhomogeneities in a coal sample. In addition, INAA has very low detection limits for many elements, can be easily automated, and provides precise data for many major, minor, and trace elements.

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EXPERIMENTAL

Three splits of approximately 500 mg of each of the eight Argonne Premium Coal samples were weighed and heat sealed in 1.5-cm³ polyethylene vials. These samples were irradiated for 8 hours in the TRIGA research reactor facility of the U.S. Geological Survey in Denver, Colo., at a neutron flux of $3\infty 10^{12}$ neutrons/cm² sec. After a delay of 3 days to eliminate or reduce short-lived activity, the samples were shipped by overnight delivery to laboratories in Reston, Va., for gamma-ray counting.

The samples were counted at three different times on high-resolution coaxial germanium and germanium (lithium) detectors for gamma-ray spectroscopy. The first count was started approximately 4 days after irradiation. A second count was started at 17 days after irradiation after allowing the short-lived activities (especially ²⁴Na, half-life=15 hours) to decay, and then a third count was begun approximately 2 months after irradiation to obtain higher precision on the measurement of the long-lived radionuclides. The gamma-ray detectors were coupled to multichannel pulseheight analyzers, which are capable of dividing the spectrum into 4,096 energy increments or channels. An automatic sample changer similar to that described by Massoni and others (1973) was used to change the samples. All spectra were processed by using the computer program SPECTRA (Baedecker and Grossman, 1989, 1994).

SAMPLES AND STANDARDS

The eight Argonne National Laboratory Premium Coal samples used in this study have been described previously (Vorres, 1990). The convention for sample identification is the same as described by Palmer in the Introduction of this volume. Three multiple-element standards, NIST (National Institute of Standards and Technology) 1632a, NIST 1633a, **Table 1.** Comparison of concentrations determined (in micrograms per gram, $\mu g/g$) from this study with literature values.

[Concentrations in this study were determined by instrumental neutron activation analysis (INAA). Blank, no values available from listed source]

F1 (NIST 16	33, fly ash	NIST 1632b, bitt	uminous coal
Element	This study	Literature ¹	This study	Literature ²
Na	3,060±64	3,200±400	513±5	515±11
K	16,000±640	$16,100\pm1,500$	740±37	748±28
Sc	27.3±0.27	27±1	2.060 ± 0.02	1.9
Cr	113.4±8.6	127±6	10.4±0.3	11
Fe	62,600±630	62,000±3,000	7,780±160	$7,590 \pm 450$
Со	40.8±2.0	41.5±1.2	2.33±0.04	2.29±0.17
Ni	92±10	98±9	8.1±2	6.1±0.27
Zn	183±9.5	216±25	11.7±1.4	11.89±0.78
As	56.0±1.6	58.0±4	3.80±0.11	3.73±0.09
Se	9.0±0.54	10.2 ± 1.4	1.24±0.10	1.29±0.11
Br	6.5±0.32	12±4	21.3±1.1	17
Rb	10±6.3	125±10	4.2±0.76	5.05±0.11
Sr	1,340±67	$1,700\pm300$	97±5.8	102
Sb	6.61±0.2	6.9±0.6	.259±0.01	.24
Cs	7.76±0.23	8.6±1.1	.414±0.012	.44
Ва	$2,450\pm74$	$2,700\pm200$	69±2.8	67.5±2.1
La	80.4±1.6	82±2	4.80±0.01	5.1
Ce	139 ± 2.7	146±15	9.19±0.18	9
Nd	55±7.2	57.8±1.6	<12	
Sm	14.2±0.3	12.4±0.9	.899±0.009	.87
Eu	2.68±0.08	2.5±0.4	.176±0.004	.17
Tb	1.79±0.054	1.9±0.3	.104±0.003	
Yb	5.99±0.18	7±3	.366±0.01	
Lu	1.12±0.046	1.0 ± 0.1	.099±0.004	
Hf	7.12±0.21	7.9±0.4	.410±0.002	.43
Та	2.03±0.06	1.8±0.3	.194±0.03	
W	4.93±0.25	4.6±1.6	.52±0.05	.48
Th	23.1±0.46	24.8±2.2	1.321±0.026	1.342±0.036
U	10.7±0.9	12.0±0.5	.42±0.05	.436±0.012

¹Values taken from Ondov and others (1975).

²Values taken from NIST (National Institute of Standards and Technology) 1632b Certificate (National Bureau of Standards, 1985). Values with no error listed are NIST information values; all others are NIST certified values.

and Eastman-Kodak TEG–50–B, and two control samples, NIST 1633 (fly ash; different from 1633a) and NIST 1632b (bituminous coal), were included with each irradiation. The element concentration values for the NIST standards used for analysis have been reported previously (Palmer and Baedecker, 1989) and are largely based on the results of Ondov and others (1975).

A comparison of the results of this study with literature values for the control samples is given in table 1. The analytical errors reported for the control NIST 1632b in this study are based on counting statistics at the one-sigma level. NIST certified and information values are shown for NIST 1632b. Our determinations of concentrations in control 1632b agree with all certified values within the stated errors and generally agree, within 10 percent, with the NIST information values that have no reported errors.

RESULTS AND DISCUSSION

The concentrations and their associated errors based on counting statistics for 29 elements for each of the Premium Coal samples are shown in table 2. Iron is the only major element (concentrations >1 percent) determined, and sodium and potassium are the only minor elements (concentrations <1 percent, >0.1 percent) determined. All other elements determined are trace elements. For many elements, the concentration values ranged over a factor of 5 among the eight Argonne Premium Coal samples.

The errors reported in table 2 are based on counting statistics only. Generally, the precision of the data based on the replicate analyses is within the counting errors for elements where the reported error is greater than 5 percent. For some elements with small counting errors, the analytical

Element	UF PC-1-1	UF PC-1-2	UF PC-1-3	WY PC-2-1	WY PC-2-2	WY PC-2-3	IL PC-3-1	IL PC-3-2	IL PC-3-3
Na	336±2%	343±2%	$342\pm3\%$	1,140±2%	1,170±2%	1,130±2%	1,030±2%	1,010±2%	1,030±2%
K	2,640±4%	2,620±4%	2,800 $\pm5\%$	268±10%	340±11%	270±17%	1,990±4%	1,860±5%	1,990±6%
Sc	4.04±1%	4.01±1%	4.14 $\pm1\%$	1.67±2%	1.69±2%	1.67±1%	2.55±1%	2.57±1%	2.64±1%
Cr	20.4±3%	20.3±3%	20.4 $\pm5\%$	5.7±6%	6.22±4%	6.23±4%	33.7±2%	32.5±3%	33.2±2%
Fe	17,600±1%	18,100±1%	17,900±1%	3,810±2%	3,790±2%	3,390±2%	26,600±1%	26,700±1%	26,800±2%
Co	5.27±2%	5.41±2%	5.31±2%	1.66±3%	1.73±2%	1.65±2%	4.21±6%	$\begin{array}{c} 4.21{\pm}6\%\\ 17.8{\pm}13\%\\ 243{\pm}3\%\\ 4.65{\pm}4\%\\ 4.03{\pm}5\%\end{array}$	4.53±2%
Ni	15.9±16%	13.2±18%	14.3±19%	4.4±29%	5.4±24%	5.0±24%	24.2±11%		21.0±2%
Zn	18.2±8%	17.0±14%	23.9±5%	11.0±4%	11.6±5%	11.2±9%	218±2%		200±2%
As	16.6±3%	17.6±3%	17.1±3%	3.67±3%	3.23±3%	3.75±3%	4.46±4%		4.90±3%
Se	2.09±8%	1.88±7%	1.8±22%	1.57±11%	1.49±8%	1.63±6%	3.78±4%		5.07±4%
Br Rb Sr Sb Cs	66±8% 19.4±8% 56±16% .558±5% 1.48±3%	$65\pm8\%$ 19.0 $\pm8\%$ 58 $\pm19\%$.50 $\pm8\%$ 1.52 $\pm3\%$	66±8% 20.0±8% 33±22% .535±5% 1.56±3%	$2.96\pm8\%$ <5 $263\pm5\%$.180 $\pm6\%$.163 $\pm7\%$	2.88±8% <5 248±5% .203±5% .173±12%	2.79±8% <5 245±5% .191±5% .160±11%	$\begin{array}{c} 6.6{\pm}8\%\\ 15.5{\pm}14\%\\ 42{\pm}29\%\\ .848{\pm}3\%\\ 1.14{\pm}4\% \end{array}$	$\begin{array}{c} 6.5{\pm}8\%\\ 17.2{\pm}11\%\\ 36{\pm}22\%\\ .913{\pm}3\%\\ 1.12{\pm}3\%\end{array}$	6.8±8% 14.6±11% 40±18% .870±3% 1.15±3%
Ba	52±14%	6827±11%	63±15%	324±3%	318±3%	297±3%	94±7%	$91\pm7\%$	96±7%
La	10.0±2%	10.0±2%	10.3±3%	5.32±3%	5.34±3%	5.38±3%	6.11± 3%	6.10 $\pm3\%$	6.09±3%
Ce	18.1±3%	18.2±3%	18.7±2%	9.07±3%	9.5±5%	9.6±5%	12.5±3%	12.5 $\pm3\%$	12.8±3%
Nd	<20	<18	8.0±29%	<13	<11	<9	<10	<8	<11
Sm	1.98±2%	1.94±2%	2.00±2%	.968±2%	.988±2%	.977±2%	1.19±4%	1.183 $\pm2\%$	1.22±3%
Eu	.389±3%	.399±3%	$.406\pm6\%$.201±4%	.199±4%	$.195\pm6\%$.229±3%	.222±5	.227±4%
Tb	.275±5%	.252±4%	$.260\pm4\%$.122±7%	.122±4%	$.115\pm5\%$.147±5%	.131±5%	.136±5%
Yb	.91±4%	.84±5%	$.90\pm4\%$.42±8%	.443±5%	$.40\pm8\%$.51±7%	.511±5%	.551±5%
Lu	.227±5%	.213±5%	$.220\pm5\%$.101±5%	.103±5%	$.098\pm5\%$.140±12%	.131±8%	.136±10%
Hf	.647±4%	.660±4%	$.67\pm9\%$.621±3%	.622±3%	$.560\pm5\%$.54±8%	.532±4%	.578±4%
Ta	.207±5%	.224±5%	.227±5%	.141±5%	.155±5%	.142±7%	.188±5%	.197±5%	.189±6%
W	1.00±6%	1.09±7%	1.09±8%	.38±10%	.36±12%	.40±12%	1.53±5%	1.52±6%	1.50±6%
Th	2.59±2%	2.64±2%	1.547±2%	1.56±3%	1.58±3%	2.01±2%	1.95±2%	1.95±2%	1.99±3%
U	.81±8%	.81±8%	.86±9%	.55±10%	.54±9%	.52±8%	4.46±4%	4.06±4%	4.28±4%

Table 2. Concentrations (in micrograms per gram, $\mu g/g$) of 29 elements in Argonne Premium Coal samples.

[Concentrations were determined by instrumental neutron activation analysis (INAA). Errors in percent are based on one-sigma counting statistics only]

Element	PITT PC-4-1	PITT PC-4-2	PITT PC-4-3	POC PC-5-1	POC PC-5-2	POC PC-5-3	UT PC-6-1	UT PC-6-2	UT PC-6-3
Na	343±2%	343±3%	344±3%	797±2%	773±2%	778±2%	1,485±2%	1,472±2%	1,425±2%
K	$1,090\pm5\%$	1,120±6%	$1,090 \pm 7\%$	284±11%	260±14%	330±15%	180±25%	<250	252±12%
Sc	2.61±2%	2.56±1%	2.55±1%	$1.80 \pm 2\%$	$1.79 \pm 2\%$	$1.77 \pm 1\%$.832±2%	.801±1%	.805±2%
Cr	15.0±3%	15.0±3%	$14.5 \pm 3\%$	9.3±4%	9.06±3%	8.85±3%	$5.25 \pm 4\%$	5.03±4%	$5.62 \pm 3\%$
Fe	$13,600 \pm 2\%$	13,400±2%	13,300±2%	5,040±2%	5,210±2%	$5,020 \pm 2\%$	$3,270\pm 2\%$	$3,100\pm 2\%$	$3,150\pm 2\%$
Со	$2.65 \pm 3\%$	2.61±4%	2.60±6%	4.09±2%	4.13±2%	$3.98 \pm 2\%$	$1.01 \pm 2\%$.992±3%	1.00±2%
Ni	9.4±21%	11.3±17%	<12	9.2±17%	8.5±19%	8.2±19%	<3	3.6±32%	3.7±26%
Zn	$8.9 \pm 4\%$	7.8±7%	10.5±12%	5.74±5%	6.15±5%	7.3±5%	$4.80 \pm 5\%$	7.43±4%	6.82±4%
As	8.44±3%	8.30±3%	$8.5 \pm 4\%$	10.43±3%	10.6±3%	9.89±3%	.51±7%	.50±9%	.44±8%
Se	$1.49 \pm 7\%$	$1.60 \pm 7\%$	1.34±7%	$2.3 \pm 28\%$	2.5±18%	2.601±11%	$1.03 \pm 10\%$	1.16±10%	1.03±8%
Br	15.7±8%	15.5±8%	15.5±8%	50±8%	50±8%	49±8%	1.14±5	1.06±5%	1.05±5%
Rb	9.0±16%	7.2±16%	7.1±16%	<6	<6	<3	<2	$1.00 \pm 22\%$.95±25%
Sr	58±12%	61±22%	65±19%	97±23%	116±7%	$101 \pm 11\%$	71±7%	68±8%	70±8%
Sb	.239±6%	.240±5%	.221±6%	.625±4%	.458±4%	$.535 \pm 4\%$	118±7%	.098±9%	.106±8%
Cs	.737±4%	.69±9%	.67±8%	.256±5%	.283±17%	.247±11%	.123±6%	.148±6%	.119±11%
Ва	44±13%	51±11%	44±12%	197±5%	203±5%	193±4%	35.7±8%	38.1±8%	47±7%
La	6.19±3%	6.15±3%	6.11±3%	6.78±3%	6.77±3%	6.73±3%	3.41±3%	$3.27 \pm 3\%$	3.24±3%
Се	$11.5 \pm 3\%$	11.5±5%	11.1±3%	11.6±3%	$11.8 \pm 3\%$	$11.5 \pm 3\%$	4.88±3%	4.67±3%	4.84±3%
Nd	<12	<8	<10	<18	<16	<14	<3	$2.5 \pm 6\%$	1.6 ± 26
Sm	$1.100 \pm 2\%$	$1.09 \pm 2\%$	$1.09 \pm 2\%$	$1.23 \pm 2\%$	$1.21\pm 2\%$	$1.21\pm 2\%$.517±3%	$.500 \pm 3\%$.507±3%
Eu	.221±4%	.214±4%	.208±6%	.243±4%	.241±4%	.218±10%	.100±6%	.098±5%	.098±4%
Tb	.133±5%	.131±10%	.132±8%	.158±6%	.173±4%	.169±4%	$.065 \pm 5\%$.054±5%	.054±5%
Yb	.478±5%	.463±5%	.468±5%	.555±5%	.56±10%	.545±5%	.204±7%	.198±15%	.211±18%
Lu	.121±6%	.117±10%	.120±5%	.132±5%	.142±5%	.132±5%	.036±9%	.026±11%	.029±7%
Hf	.518±4%	.49±8%	.500±4%	.441±6%	.429±4%	.444±4%	.478±3%	.499±5%	.458±3%
Га	.168±5%	.163±5%	.165± 5%	.117±12%	.108±8%	.123±6%	.060±7%	.065±7%	.061±6%
W	.79±6%	.80±8%	.76±8%	.84±7%	.91±8%	.77±8%	.42±10%	.40±11%	.38±9%
Γh	$1.53 \pm 3\%$	1.46±3%	1.44±3%	1.19±3%	1.12±8%	1.17±3%	.609±3%	.614±3%	.622±3%
U	.36±13%	.37±10%	.45±11%	.47±16%	.51±13%	.55±13%	.76±8%	.70±8%	.76±9%

Table 2. Concentrations (in micrograms per gram, $\mu g/g$) of 29 elements in Argonne Premium Coal samples—Continued.

[Concentrations were determined by instrumental neutron activation analysis (INAA). Errors in percent are based on one-sigma counting statistics only]

Element	WV PC-7-1	WV PC-7-2	WV PC-7-3	ND PC-8-1	ND PC-8-2	ND PC-8-3
Na	460±10%	351±3%	352±2%	5,400±2%	5,330±3%	5,150±2%
К	$5,170 \pm 3\%$	4,970±6%	$5,000 \pm 4\%$	$280 \pm 28\%$	<400	$300 \pm 16\%$
Sc	$7.69 \pm 1\%$	7.54±1%	7.61±1%	.846±2%	.828±2%	.865±3%
Cr	36.2±1%	35.6±2%	35.7±2%	$2.23\pm5\%$	$2.16 \pm 5\%$	2.33±5%
Fe	4,040±3%	$3,840 \pm 2\%$	4,080±2%	$5,530 \pm 2\%$	$5,290 \pm 2\%$	5,630±4%
Со	7.65 ±2%	7.91±2%	7.72±3%	.771±3%	.761±3%	.802±3%
Ni	$13.0 \pm 20\%$	$18 \pm 20\%$	13.1±19%	<6	<3	<3
Zn	$14.0 \pm 4\%$	12.5±4%	$14.0 \pm 4\%$	5.71±5%	$5.5 \pm 7\%$	5.86±5%
As	5.8±12%	$5.29 \pm 3\%$	$7.40 \pm 3\%$	2.67±4%	2.61±4%	2.61±4%
Se	5.3±7%	$5.54 \pm 5\%$	$5.5 \pm 7\%$.51±18%	.61±15%	.63±149
Br	16.9±2%	16.9±3%	16.9±3%	1.49±6%	1.34±6%	1.43±6%
Rb	$29.8 \pm 3\%$	$28.5 \pm 3\%$	31.2±4%	.36±30%	.99±26%	<1
Sr	60±11%	73±19%	58±17%	628±4%	580±9%	583±5%
Sb	.521 ±2%	.540±3%	.502±4%	.148±7%	.155±8%	.158±9%
Cs	$1.88 \pm 2\%$	$1.87 \pm 2\%$	$1.84 \pm 3\%$.072±10%	$.097 \pm 23\%$.089±9%
Ва	166±11%	116±5%	116±6%	699±4%	660±4%	671±3%
La	21.7 ±2%	21.6±2%	21.3±2%	$2.79 \pm 3\%$	2.84±3%	2.84±3%
Ce	$36.0 \pm 2\%$	35.2±3%	36.1±2%	4.34±3%	$4.33 \pm 3\%$	4.68±5%
Nd	12.4 ±8%	$11.0 \pm 7\%$	12.4±9%	<3	2.1±25%	2.5 ± 229
Sm	$3.56 \pm 2\%$	$3.50 \pm 2\%$	3.51±2%	.420±3%	.419±3%	.388±3%
Eu	$.674 \pm 2\%$.651±2%	.697±4%	.079±4%	.077±4%	.087±4%
Tb	.408 ±2%	.390±2%	.402±3%	.052±9%	.052±15%	.063±7%
Yb	$1.69 \pm 3\%$	$1.61 \pm 7\%$	$1.52 \pm 5\%$.317±9%	.298±10%	.245±6%
Lu	$.202 \pm 3\%$.194±5%	.191±6%	.039±13%	.035±9%	.035±8%
Hf	$1.88 \pm 2\%$	$1.80 \pm 2\%$	$1.80 \pm 2\%$.342±4%	.341±3%	.340±4%
Та	.675±2%	.617±3%	.630±5%	.095±6%	.088±5%	.095±5%
W	1.33 ±7%	1.19±8%	2.53±4%	.38±17%	.25±31%	.43±139
Th	6.49 ±2%	6.26±3%	6.49±3%	1.057±2%	1.027±2%	1.113±2%
U	$1.67 \pm 5\%$	1.60±7%	$1.64 \pm 8\%$.52±16%	.46±1%	<.2

Table 2. Concentrations (in micrograms per gram, $\mu g/g$) of 29 elements in Argonne Premium Coal samples—Continued. [Concentrations were determined by instrumental neutron activation analysis (INAA). Errors in percent are based on one-sigma counting statistics only]

Table 3. Long-lived (>10 hours) radionuclides.

[Table modified from Palmer and Baedecker (1989). d, day; h, hour; yr, year; µg/g, micrograms per gram]

	Indicator		Preferred gamma	Limit of determination	Potential spectral interferences				
Element	radionuclide	Half-life	(γ) energy	(μ g/g except % as indicated)	Radionuclide	Energy	Radionuclide	Energy	
Na	²⁴ Na	15.0 h	1,368.9 2,753.9	10					
K	^{42}K	12.4 h	1,524.7	.01%					
Sc		84 d	1,120.5	.01					
Cr		27.8 d	320.1	.5	¹⁷⁷ Lu	321.3	¹⁴⁷ Nd	319.4	
Fe		45.6 d	1,099.3	50	¹⁸² Ta	1,289.1	ING	517.4	
1.6	10	45.0 u		75	14	1,209.1			
Со	⁶⁰ Co	5.3 yr	1,291.5 1,173.2 1,332.5	.2					
Ni	⁵⁸ Co	71.3 d	810.8	.55	¹⁵² Eu	810.8			
Zn		245 d	1,115.4	1	¹⁶⁰ Tb	1,115.1			
As		245 d 26.4 h	559.0	.1	10	1,115.1			
		120.4 fl 120.0 d	264.6	.1	¹⁸² Ta	264.1			
Se					112	204.1			
Br	⁸² Br	35.4 h	554.3	.5 .5					
	0(776.5	5					
Rb		18.7 d	1,076.8	5					
Sr	⁸⁵ Sr	64.0 d	514.0	50					
Sb	¹²² Sb	67.2 h	564.0	.05					
	¹²⁴ Sb	60.0 d	1,691.0	.1					
Cs		2.1 yr	795.8	.1					
Ba		12.0 d	476.3	50					
La	¹⁴⁰ La	40.2 h	1,596.6	.02					
La	La	40.2 II	487.0	.02					
C.	¹⁴¹ Ce	2254			¹⁵⁴ Eu	1456			
Ce		32.5 d	145.4	.5	Eu	145.6			
Nd	¹⁴⁷ Nd	11.1 d	531.0	2	2205 7	102 7			
<u>S</u> m		46.8 h	103.2	.5	²³⁹ Np	103.7			
Eu	¹⁵² Eu	12.7 yr	779.1	.04					
			1,408.1	.01					
Tb	¹⁶⁰ Tb	72.1 d	298.6	.05	²³³ Pa	299.9			
			1,178.1	.1					
Yb	¹⁷⁵ Yb	101.0 h	396.1	.1	¹⁴⁷ Nd	398.2	²³³ Pa	398.2	
			282.6	.2					
Lu	¹⁷⁷ Lu	6.7 d	208.4	.01	²³⁹ Np	209.7			
Hf		42.5 d	482.2	.1	түр	207.1			
111	111	42.5 u	133.1	.05					
Та	¹⁸² Ta	11514							
Та	102 Ia	115.1 d	1,221.3	.02					
	197557	24.01	1,189.2	.03					
W	^{187}W	24.0 h	479.5	.1	1475 5 4				
			685.7	.1	¹⁴⁷ Nd	685.9			
Th		27.0 d	311.9	.1					
U	²³⁹ Np	2.3 d	277.6	.5					

precision is poorer because of the other sources of error such as sample homogeneity or positioning during counting.

Errors reported in table 2 were generally less than 10 percent except for nickel, rubidium, and neodymium, in which the concentration was near or below the detection limit for all samples. Errors were also greater than 10 percent for barium in UF PC-1; uranium in PITT PC-4, POC PC-5, and ND PC-8; and ytterbium in UT PC-6 (table 2). Errors reported for potassium are variable even at the same concentration because it has the shortest half-life of the elements determined in this study, and the detection limit varies by nearly an order of magnitude during the 2-day counting cycle for the entire sample set.

In table 2, the concentrations reported for nickel in WY PC-2 and UT PC-6 and for rubidium in UT PC-6 and ND

PC-8 are actually below the expected detection limits given in table 3 because the values in table 3 are determined for a "typical" coal matrix. The detection limits for individual coal samples may change because of variations in the concentrations of the most sensitive elements that dominate the gamma-ray spectrum and because of variations in the intensities of spectral interferences. The percent correction of each spectral interference for all premium coals is given in table 4. Generally, only a small correction is needed for most elements. Some elements, such as nickel, selenium, and samarium in some samples, require changes larger than 10 percent.

In addition to corrections made because of spectral interferences, barium and the light rare earth elements lanthanum, cerium, neodymium, and samarium were corrected

 Table 4. Average percent corrections made for spectral interferences on counts with the lowest errors during INAA (instrumental neutron activation analysis) of eight Argonne Premium Coal samples.

[- indicates no correction was made; <0.1 indicates an extremely small correction was made. No corrections were required for spectral interference of the
¹⁶⁰ Tb line by the ²³³ Pa line in any of these samples. Data for individual splits have been reported by Palmer (1991)]

Element	Cr	Fe	Ni	Zn	Se	Ce	Sm	Yb	Lu	W
Interfering	177-	192-	152-	160	182-	154-	220	1475 7.4	220	1475
radionuclide	¹⁷⁷ Lu	¹⁸² Ta	¹⁵² Eu	¹⁶⁰ Tb	¹⁸² Ta	¹⁵⁴ Eu	²³⁹ Np	¹⁴⁷ Nd	²³⁹ Np	¹⁴⁷ Nd
	and (or) ¹⁴⁷ Nd							and (or) ²³³ Pa		
UF PC-1	2.3	<0.1	10.5	3.3	14.5	0.2	4.5	0.3	3.0	0.4
WY PC-2	4.8	.2	142	2.7	13.1	.5	5.4	.4	5.3	_
IL PC-3	1.2	<.1	4.5	.2	7.5	_	26.5	1.5	_	.6
PITT PC-4	1.7	<.1	7.8	4.0	14.2	.4	3.9	.4	3.8	.4
POC PC-5	2.5	<.1	10.0	6.2	7.8	.3	3.5	.6	3.1	_
UT PC-6	1.7	.1	12.3	2.1	10.9	_	9.7	.4	1.3	.7
WV PC-7	1.5	.5	14.8	6.3	13.5	.3	3.0	.6	_	1.7
ND PC-8	4.4	<.1	9.6	2.3	22.2	_	8.9	.5	_	_

 Table 5.
 Average percent correction made for fission product interferences during INAA (instrumental neutron activation analysis) of eight Argonne Premium Coal samples.

[Apparent concentration of each fission product is given in micrograms per gram $(\mu g/g)$ per $\mu g/g$ of U. <0.1 indicates a correction of less than 0.1 percent was made. Data for individual splits have been reported by Palmer (1991)]

Fission product Apparent concentration	¹³¹ Ba (¹)	¹⁴⁰ La (¹)	¹⁴¹ Ce 0.27	¹⁴⁷ Nd 0.17	¹⁵³ Sm 0.00008
UF PC-1	4.9	0.1	1.2	² 0.6	<0.1
WY PC-2	.6	.2	1.5	² .8	<.1
IL PC-3	16.1	1.0	8.2	² 6.8	<.1
PITT PC-4	3.4	.1	.9	² .7	<.1
POC PC-5	.9	.1	1.2	² .5	<.1
UT PC-6	7.1	.4	4.0	² 5.0	<.1
WV PC-7	3.8	.1	1.3	2.2	<.1
ND PC-8	.1	.3	2.7	² 2.9	<.1

¹Apparent concentration of element in $\mu g/g$ per $\mu g/g$ of U is time dependent.

²Correction made on upper limit value.

for interference because of neutron-induced fission of ²³⁵U; table 5 shows the percent correction for these elements. However, corrections for barium and lanthanum are time dependent and therefore vary during the counting of the samples. The concentration of barium, Ba_{corr} (corrected for the time-dependent fission correction factor), was calculated by using the formula:

$$Ba_{corr} = Ba_{meas} - 2.9Ue^{0.0402t}$$

and the concentration of lanthanum, La_{corr} (corrected for the time-dependent fission correction factor), was calculated by using:

$La_{corr} = La_{meas} - 0.002723 Ue^{0.3592t}$

where t = time after bombardment in days, Ba_{meas} and La_{meas} are the uncorrected barium and lanthanum concentrations, and U is the concentration of uranium. The constant 0.002723 in the La equation is calculated by assuming a ²³⁵U cross section of 580 barns, which agrees with the experimental data within ±1 percent. The half-lives for ¹³¹Ba and ²³⁹Np (U) and ¹⁴⁰La were taken from table 3. The half-life of ¹⁴⁰Ba, which decays to the measured ¹⁴⁰La, was assumed to be 12.8 days. The fission correction factors are generally quite small except for barium and cerium in IL PC-3, which are about 16 and 8 percent, respectively, and for barium and

neodymium in UT PC-6, which have correction factors as high as 7 and 5 percent, respectively.

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Determination of Selected Elements in Coal Ash from Eight Argonne Premium Coal Samples by Atomic Absorption Spectrometry and Atomic Emission Spectrometry

By Michael W. Doughten

ABSTRACT

Methods for the determination of 22 elements in coal ash from Argonne Premium Coals by inductively coupled argon plasma-atomic emission spectrometry and flame and graphite furnace atomic absorption spectrometry are described. Coal ashes were analyzed in triplicate to determine the precision of the methods. Results of the analyses of National Institute of Standards and Technology (NIST) standard reference materials 1633 and 1633a are reported. Accuracy of the methods was determined by comparing element concentrations in standard reference materials determined in this study with their certified values and literature values.

INTRODUCTION

Procedures are described and results are presented for the determination of 22 elements in the coal ashes from 8 Argonne Premium Coal samples by inductively coupled argon plasma-atomic emission spectrometry (ICAP-AES) and atomic absorption spectrometry (AAS). Results of the ICAP-AES and AAS analyses of two standard reference materials from the National Institute of Standards and Technology (formerly the National Bureau of Standards, NBS), NIST 1633 and 1633a (coal fly ashes), are included and are compared with their certified values as well as with other values reported in the literature. Cadmium and lead were determined by graphite furnace atomic absorption spectrometry (GFAAS); cobalt and lithium were determined by flame atomic absorption spectrometry (FAAS). All other elements were determined by ICAP-AES. All the analyses described in this paper were performed in the U.S. Geological Survey laboratory in Reston, Va.

SAMPLE PREPARATION

Raw coal samples were ashed by weighing 70 g of coal into a previously weighed porcelain crucible. The cru-

cible was placed in an electric furnace, which was slowly heated to 200°C. After 1.5 hours at 200°C, the temperature was increased to 350°C and was held at that temperature for 2 hours. The temperature was then increased to 525°C and maintained for about 36 hours. After the sample cooled for 1 to 2 hours, the weight of the ash was determined by subtracting the weight of the crucible from the weight of the crucible plus ash. Ash yield was reported as percent ash and was calculated by:

Percent ash =
$$\frac{(\text{weight ash})}{(\text{weight coal})} \times 100$$

Percent ash data are listed in table 1.

Sample solutions for analysis by ICAP-AES and AAS were prepared by weighing 100 mg of the coal ash and placing it in a 75-mL Teflon screwcap bomb, then adding 7 mL of concentrated nitric acid. The bomb was capped and heated on a hotplate overnight at 200°C. After cooling, the bomb was uncapped, and 2 mL of concentrated nitric acid, 2 mL of concentrated perchloric acid, and 10 mL of hydro-fluoric acid were added. All acids used were reagent grade. The bomb was then recapped and again heated on a hotplate at 200°C for 4 hours. The bomb was uncapped, and the solution evaporated to dryness. The sample was allowed to cool,

Table 1. Coal ash yields for ArgonnePremium Coal samples.

[Coal	ash	yields	determined	by	Larry	Win-
ters, U	J.S. (Geolog	ical Survey]			

Coal sample	Percent ash
UF PC-1-1 WY PC-2-1 IL PC-3-1 PITT PC-4-1	8.5
POC PC-5-1 UT PC-6-1 WV PC-7-1 ND PC-8-1	4.6 19.4

and 10 mL of 2*N* hydrochloric acid was added. The bomb was recapped once again and gently heated until the solution was clear, indicating complete dissolution. The solution was then transferred to a 15-mL polyethylene tube. This digestion procedure was used to determine that the concentration of the coal ash in solution was 1 percent.

SAMPLE ANALYSIS

The determinations of Ba, Be, Cr, Cu, Mn, Ni, Sr, V, Y, Zn, K, P, and Ti were made directly on this solution by ICAP-AES using a Jarrell-Ash model 1160 Atomcomp ICP system. Cobalt and lithium were determined on this solution by FAAS using a Perkin-Elmer model 5000 atomic absorption spectrometer. A deuterium arc lamp background corrector was used for Co. Lithium requires no background correction. Concentrations for Co and Li were calculated from a calibration curve established by analyzing a set of cobalt and lithium standard solutions.

The sample solution was diluted 1 to 10 with 2N hydrochloric acid and analyzed for sodium and magnesium by ICAP-AES. Cadmium was determined on this solution by using a Perkin-Elmer atomic absorption spectrometer with a graphite furnace assembly (model HGA 500) and a Zeeman background correction system. Lead was determined on this solution by using a Perkin-Elmer model 603 atomic absorption spectrometer with a graphite furnace

Table 2.	GFAAS (graphite furnace atomic absorption spectrom-
etry) operation	ating conditions.

	Cd	Pb
Wavelength Bandpass Source lamp Graphite tube Sample size	228.8 nm 0.7 nm Hollow cathode Uncoated 20 μL	283.3 nm 0.7 nm Hollow cathode Uncoated 10 μL
$\begin{array}{c} Matrix modifier, 2 \ percent \\ solution of \ NH_4H_2PO_4 \\ Drying temperature \\ Drying time: \\ Ramp \\ Hold \\ Charring temperature \\ Charring time: \\ Ramp \\ Hold \\ Hold \end{array}$	10 μL 110°C 10 sec 30 sec 250°C 5 sec 25 sec	10 μL 110°C 0 sec 30 sec 950°C 0 sec 20 sec
Atomizing temperature Atomizing time: Ramp Hold Background correction Calibration standard con- centration range Lower limit	2,300°C 0 sec 5 sec Zeeman 0–4 ng/mL 0.2 ng/mL	2,700°C 0 sec 5 sec Deuterium 0–0.2 μg/mL 0.01 μg/mL

 Table 3.
 FAAS (flame atomic absorption spectrometry) operating conditions.

[Flame is fuel lean air-acetylene]

Со	Li
240.7 nm	670.8 nm
0.7 nm	0.4 nm
Hollow cathode	Hollow cathode
Deuterium	None
0–5 μg/mL 0.1 μg/mL	0–2 μg/mL 0.1 μg/mL
	240.7 nm 0.7 nm Hollow cathode Deuterium 0–5 μg/mL

 Table 4. ICAP-AES (inductively coupled argon plasma-atomic emission spectrometry) wavelengths, calibration ranges, and detection limits.

Element	Wavelength (nm)	Calibration range ¹ (µg/mL)	Limit of detection (in solution) (µg/mL)
Al	308.2	0-2	0.5
Ba	455.4	0-20	.5
Be	313.0	0-1	.1
Ca	317.9	0-1	.5
Fe	259.9	0-20	.5
K	766.5	10-200	10
Mg	280.2	0-10	.5
Mn	257.6	0-10	.5
Na	589.0	0-10	.5
Ni	231.6	0-1	.1
P Sr Ti V Zn	214.9 407.7 334.9 292.4 213.8	0-50 0-5 0-100 0-1 0-1	1 .1 .1 .1

¹Calibration standard concentration range.

assembly (model HGA 2100) and a deuterium arc lamp background correction system. A 2 percent solution of ammonium phosphate (NH₄H₂PO₄) (see table 2) was used as a matrix modifier for both Cd and Pb. Concentrations for Cd and Pb were calculated from a calibration curve (absorbance versus concentration (μ g/g)) established from analyzing a set of Cd and Pb standard solutions. This diluted solution was further diluted to 1 to 100 with 2*N* hydrochloric acid and analyzed for aluminum, calcium, and iron by ICAP-AES.

All calibration solutions for AAS and AES were prepared in 2N hydrochloric acid. Instrumental operating parameters for GFAAS and FAAS are listed in tables 2 and 3. ICAP-AES wavelengths and concentration ranges are listed in table 4. Trace- and major-element concentrations are listed in tables 5 and 6.

	concentrations (in m			

[Cd and Pb were determined by graphite furnace atomic absorption spectrometry, Co and Li were determined by flame atomic absorption spectrometry, and all other element concentrations were determined by inductively coupled argon plasma-atomic emission spectrometry. –, no data]

Coal ash	Ba	Be	Cd	Со	Cr	Cu	Li	Mn	Ni	Pb	Sr	V	Y	Zn
UF PC-1-1	400	11	0.53	38	150	140	110	300	110	58	440	190	69	140
UF PC-1-2	410	11	.51	38	140	140	110	310	100	55	450	190	69	140
UF PC-1-3	420	11	.54	35	150	140	110	300	110	51	430	190	68	150
WY PC-2-1	180	3.1	1.1	19	73	140	48	240	58	37	1,800	160	46	120
WY PC-2-2	180	2.8	1.1	19	75	140	49	250	55	35	1,700	160	44	120
WY PC-2-3	170	2.9	1.1	19	71	150	48	220	61	32	1,800	160	44	120
IL PC-3-1	420	4.8	3.7	26	190	65	48	470	110	42	180	200	26	1,000
IL PC-3-2	540	4.6	4.0	28	200	64	49	480	120	39	180	200	26	1,100
IL PC-3-3	540	4.6	3.5	26	190	64	48	470	110	39	180	190	26	1,200
PITT PC-4-1	460	8.3	.64	26	160	60	95	200	99	32	700	160	47	89
PITT PC-4-2	450	8.3	.63	28	150	58	93	190	90	32	700	160	46	81
PITT PC-4-3	430	8.5	.68	26	140	60	94	190	88	32	680	160	46	84
POC PC-5-1	310	15	1.6	71	180	220	110	300	130	48	1.600	200	120	73
POC PC-5-2	310	15	1.4	71	160	220	110	300	120	45	1,600	200	130	77
POC PC-5-3	310	15	1.4	71	180	220	110	300	130	45	1,600	200	110	74
UT PC-6-1	690	2.8	1.3	18	110	84	110	88	73	34	1,300	86	45	140
UT PC-6-2	660	2.7	1.3	18	99	82	110	88	71	34	1,300	87	46	130
WV PC-7-1	610	9.7	.40	40	210	100	150	80	81	63	250	220	55	65
WV PC-7-2	630	9.8	.37	43	210	100	150	78	82	61	240	230	54	66
WV PC-7-3	680	10.	.44	42	200	100	150	77	82	63	260	220	57	63
ND PC-8-1	4,500	1.9	.50	<10	27	59	28	840	15	16	5.400	37	20	54
	4.100	1.8	.46	<10	26	37	28	830	15	16		36	19	49
ND PC-8-3	4,700	_	.48	<10	24	52	30	850	12	16	5,300	38	_	47
ND PC-8-2	4,100	1.8	.46	<10	26	37	28	830	15	16	5,300	36	19	

 Table 6.
 Major-element concentrations (in weight percent) in coal ash from eight Argonne

 Premium Coal samples.

Coal ash	Al	Ca	Fe	K	Mg	Na	Р	Ti
UF PC-1-1	11.9	3.31	13.8	2.10	0.61	0.24	0.056	0.58
UF PC-1-2	11.9	3.30	14.0	2.10	.61	.24	.059	.59
UF PC-1-3	11.8	3.28	13.6	2.10	.61	.24	.061	.58
WY PC-2-1	8.18	14.2	4.21	.33	2.83	1.40	.31	.66
WY PC-2-2	8.21	13.6	4.46	.33	2.78	1.37	.29	.66
WY PC-2-3	8.26	14.3	4.28	.31	2.81	1.40	.30	.65
IL PC-3-1	7.75	5.82	16.8	1.20	.47	.60	.036	.43
IL PC-3-2	7.70	5.96	16.4	1.20	.48	.61	.037	.43
IL PC-3-3	7.72	6.00	17.2	1.20	.47	.58	.037	.42
PITT PC-4-1	10.8	2.29	14.6	1.20	.41	.35	.12	.62
PITT PC-4-2	10.9	2.23	14.3	1.20	.41	.35	.12	.61
PITT PC-4-3	10.8	2.21	14.2	1.20	.40	.34	.12	.60
POC PC-5-1	10.4	8.75	9.67	.51	1.06	1.57	.042	.72
POC PC-5-2	10.4	8.53	9.52	.53	1.04	1.57	.041	.72
POC PC-5-3	10.3	8.53	9.50	.53	1.04	1.54	.040	.72
UT PC-6-1	7.56	9.07	6.16	.30	.64	2.75	.020	.50
UT PC-6-2	7.50	8.70	6.09	.30	.65	2.70	.021	.49
WV PC-7-1	16.1	.29	2.04	2.52	.48	.17	.046	1.34
WV PC-7-2	16.4	.30	1.96	2.58	.49	.18	.045	1.34
WV PC-7-3	16.3	.30	1.95	2.55	.49	.18	.038	1.34
ND PC-8-1	4.17	16.0	4.91	.13	4.16	4.84	.13	.21
ND PC-8-2	4.22	16.1	4.95	.16	4.16	4.89	.13	.20
ND PC-8-3	4.25	16.5	5.05	.15	4.26	5.00	.13	.21

[Element concentrations were determined by inductively coupled argon plasma-atomic emission spectrometry]

[Concentrations for the first 14 elements are in micrograms per gram ($\mu g/g$); others are in weight percent. Methods used in this study: ICAP-AES, inductively coupled argon plasma-atomic emission spectrometry; GFAAS, graphite furnace atomic absorption spectrometry; FAAS, flame atomic absorption spectrometry. NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards (NBS). Certified and information values for NIST 1633 are from NBS (1975). —, no data]

Element	This study				NBS	Gladney and others	Kane	Ondov and others	Rowe and Steinnes
Element	Method	Conc.	(1975) ¹	(1987)	(1989)	(1975)	(1977)		
Ba $(\mu g/g)$	ICAP-AES	2,700	_	$2,665 \pm 160$	_	$2,700 \pm 200$	2,540		
Be	ICAP-AES	11	—	12.1 ± 1.0	—	—	—		
Cd	GFAAS	1.5	1.45 ± 0.06	1.47 ± 0.15	1.56 ± 0.26	—	—		
Со	FAAS	41	38	40 ± 3	_	41.5 ± 1.2	40.3 ± 0.4		
Cr	ICAP-AES	130	131 ± 2	127 ± 10	—	127 ± 6	129.2 ± 2.7		
Cu	ICAP-AES	130	128 ± 5	129 ± 7	130 ± 6	_	115 ± 8		
Li	FAAS	160	—	170 ± 80	170 ± 13	_	_		
Mn	ICAP-AES	510	493 ± 7	494 ± 20	496 ± 34	496 ± 19	488 ± 14		
Ni	ICAP-AES	94	98 ± 7	98 ± 6	_	98 ± 9	69 ± 7		
Pb	GFAAS	70	70 ± 4	72 ± 6	67 ± 4	75 ± 5	—		
Sr	ICAP-AES	1,400	$1,430 \pm 60$	$1,380 \pm 100$	_	$1,700 \pm 300$	$1,430 \pm 60$		
V	ICAP-AES	220	214 ± 8	224 ± 24	_	235 ± 13	237 ± 20		
Y	ICAP-AES	66	—	64 ± 4	_	62 ± 10	—		
Zn	ICAP-AES	210	210 ± 20	211 ± 11	210 ± 10	216 ± 25	201 ± 6		
Al (%)	ICAP-AES	11.9	_	12.6 ± 0.6	_	12.7 ± 0.5	12.35 ± 0.25		
Ca	ICAP-AES	4.58	—	4.65 ± 0.34	_	4.7 ± 0.6	4.69 ± 0.14		
Fe	ICAP-AES	6.05	—	6.16 ± 0.27	_	6.2 ± 0.3	6.2 ± 0.05		
K	ICAP-AES	1.80	1.72	1.69 ± 0.09	_	1.61 ± 0.15	1.80 ± 0.13		
Mg	ICAP-AES	1.33	1.98	1.5 ± 0.3	1.17 ± 0.049	1.8 ± 0.4	1.78 ± 0.2		
Na	ICAP-AES	.31	.307	$.313 \pm 0.02$	$.295 \pm 0.023$	$.32 \pm 0.04$	$.283 \pm 0.014$		
Р	ICAP-AES	.12	—	$.101 \pm 0.018$	—	—	—		
Ti	ICAP-AES	.77	—	$.71 \pm 0.05$	—	$.74 \pm 0.03$	$.70 \pm 0.03$		

¹Values with no error listed are NIST information values; all others are NIST certified values.

DISCUSSION

NIST standard reference materials 1633 and 1633a were used as control standards for each determination. Each Argonne Premium Coal sample was run in triplicate (labeled 1, 2, and 3) with the exception of UT PC-6, which was run in duplicate due to a lack of available sample. Data for these analyses are shown in tables 5 and 6. The replicate analyses show the precision of the methods used. The relative percent standard deviation was generally about ± 5 percent. Comparison of the analyses of these control standards with NIST certified values (NBS, 1975, 1979), with AAS values (Kane, 1989), and with values determined by instrumental neutron activation analysis (Ondov and others, 1975; Rowe and Steinnes, 1977) shows the accuracy of the methods (tables 7 and 8).

ICAP-AES in Reston of samples WY PC-2 and POC PC-5 determined lower concentrations of barium and stron-

tium than other methods of analysis (see table 2 in paper by Palmer and Klizas, this volume). They were probably caused by incomplete dissolution of barium sulfate present in the sample. Low Sr results may have been caused by Sr co-precipitating with Ba.

Sample IL PC-3 showed a wide concentration range for Ba (420–540 μ g/g). Sample ND PC-8 showed low concentrations of potassium. These values were close to the detection limit, and that may account for the error. This also could be due to sampling error or incomplete digestion of barite that may be present in the sample.

Beryllium values determined by ICAP-AES were corrected for vanadium and titanium interferences, and zinc values determined by ICAP-AES were corrected for interferences by manganese, iron, and vanadium. Interference corrections vary depending on the instrument operating conditions used and should be determined before the start of the analysis. **Table 8.** Comparison of element concentrations in NIST standard reference material 1633a determined in this study with certified values and with concentrations determined in other studies.

[Concentrations for the first 14 elements are in micrograms per gram ($\mu g/g$); others are in weight percent. Methods used in this study: ICAP-AES, inductively coupled argon plasma-atomic emission spectrometry; GFAAS, graphite furnace atomic absorption spectrometry; FAAS, flame atomic absorption spectrometry. NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards (NBS). Certified and information values for NIST 1633a are from NBS (1979). —, no data]

Element	This st	udy	NBS (1979) ¹	Gladney and others (1987)	Kane (1989)
Element	Method	Conc.	NBS (1979)	Gradicy and others (1987)	Kane (1969)
Ba (µg/g)	ICAP-AES	1,400	1,500	$1,420 \pm 100$	
Be	ICAP-AES	12	12	12.8 ± 0.6	_
Cd	GFAAS	.94	1.0 ± 0.15	1.12 ± 0.17	0.98 ± 0.08
Со	FAAS	46	46	43 ± 3	_
Cr	ICAP-AES	190	196 ± 6	194 ± 7	_
Cu	ICAP-AES	110	118 ± 3	120 ± 4	114 ± 7
Li	FAAS	170	_	165 ± 50	184 ± 14
Mn	ICAP-AES	180	179 ± 8	188 ± 15	160 ± 12
Ni	ICAP-AES	130	127 ± 4	124 ± 13	_
Pb	GFAAS	71	72.4 ± 0.4	72 ± 4	62 ± 4
Sr	ICAP-AES	850	830 ± 30	810 ± 40	_
V	ICAP-AES	290	297 ± 6	294 ± 18	_
Y	ICAP-AES	87	_	82 ± 6	_
Zn	ICAP-AES	210	220 ± 10	226 ± 22	211 ± 11
Al (%)	ICAP-AES	14.0	14.3 ± 1.0	14.4 ± 0.6	_
Ca	ICAP-AES	1.14	1.11 ± 0.01	1.14 ± 0.06	_
Fe	ICAP-AES	9.18	9.40 ± 0.1	9.37 ± 0.23	_
K	ICAP-AES	1.95	1.88 ± 0.06	1.88 ± 0.05	_
Mg	ICAP-AES	.44	$.455 \pm 0.01$	$.457 \pm 0.045$	$.436 \pm 0.005$
Na	ICAP-AES	.17	$.17 \pm 0.01$	$.173 \pm 0.011$	$.158 \pm 0.014$
Р	ICAP-AES	.18	_	$.169 \pm 0.024$	_
Ti	ICAP-AES	.92	0.8	$.823 \pm 0.034$	—

¹Values with no error listed are NIST information values; all others are NIST certified values.

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Determination of 25 Elements in Coal Ash from 8 Argonne Premium Coal Samples by Inductively Coupled Argon Plasma-Atomic Emission Spectrometry

By Paul H. Briggs

ABSTRACT

Twenty-five major and trace elements were determined in coal ash material by inductively coupled argon plasmaatomic emission spectrometry (ICAP-AES). Two decomposition techniques were used. Coal ashes were analyzed in triplicate to determine the precision of the method. The National Institute of Standards and Technology (NIST), formerly the National Bureau of Standards (NBS), standard reference material 1632a and U.S. Geological Survey (USGS) standard reference material CLB-1 were used to assess the accuracy of the method.

INTRODUCTION

Inductively coupled argon plasma-atomic emission spectrometry (ICAP-AES) is rapidly becoming a common method to determine many major and trace elements in geologic materials. An overview of ICAP-AES analysis was given by Lichte and others (1987). Recently, the ICAP-AES method was expanded to include the analysis of coal by using an acid dissolution procedure (Doughten and Gillison, 1990). This work included new methods combining results from two different decomposition procedures that were used to determine the concentrations of 25 elements in coal ashes from 8 Argonne Premium Coal samples by ICAP-AES. Fourteen elements were determined by an acid decomposition using a mixture of hydrochloric, nitric, perchloric, and hydrofluoric acids at a low temperature in a method described by Crock and others (1983). Eleven additional elements were determined by a sodium peroxide sinter decomposition technique modified from one described by Borsier and Garcia (1983). The digested sample was aspirated into the ICAP discharge where the elemental emission signal was measured simultaneously for the elements of interest.

EXPERIMENTAL

All ICAP-AES measurements described in this paper were performed on a Thermo Jarrell-Ash model 1160 Plasma Atomcomp simultaneous instrument in the U.S. Geological Survey (USGS) laboratory in Denver, Colo. ICAP-AES calibration was performed using USGS reference material BHVO-1, Canadian Certified Reference Materials Project SY-3, and four multielement solutions. The wavelengths, operating ranges, and decomposition methods are given in table 1. The ICAP-AES operating conditions are given in table 2.

Two decomposition procedures were used to determine the 25 elements. The acid decomposition technique was used for the determination of the trace elements Be, Co, Cr, Cu, Li, Mn, Ni, Sc, Sr, Th, V, Y, and Zn and the major element Na. The trace-element suite was chosen in order to give the best reporting limits for the 100-fold dilution and the ease of solubility by the acid decomposition. Sodium is reported with the trace suite because sodium peroxide was the sintering flux used for the sample decomposition for major-element determinations.

Coal ash sample solutions from the acid decomposition were prepared in the following manner: a 0.200-g sample, to which a solution containing 100 μ g lutetium had been added as an internal standard, was digested and evaporated to dryness in a 30-mL Teflon vessel with 3 mL HCl, 2 mL HNO₃, 1 mL HClO₄, and 2 mL HF at 110°C. An additional 1 mL HClO₄ was added to the residue and taken to dryness again at 160°C. One milliliter HNO₃ and one drop 30 percent H₂O₂ were added to the residue, and 20 mL of 1 percent HNO₃ was added to the solution. The solution was transferred to a 13∞100-mm polypropylene test tube and capped until ready for ICAP-AES analysis. All reagents used in the procedures were reagent grade or better. It should be noted that this solution was used for both ICAP-AES and inductively coupled argon plasma-mass spectrometry (ICAP-MS) to minimize duplication of digestion and maximize the efforts of the laboratory staff. (See ICAP-MS analysis in Meier's paper in this volume.)

Table 1. Wavelengths, operating ranges, and decomposition methods used for determining concentrations of 25 elements in coal ash from 8 Argonne Premium Coal samples by ICAP-AES (inductively coupled argon plasma-atomic emission spectrometry).

 $[\lambda, wavelength; nm, nanometer; ppm, parts per million; \%, weight percent]$

Element	$\lambda(\text{nm})$	Range	Decomposition
Al	309.2	0.04-100 %	Sinter
B	249.7	40–10,000 ppm	Sinter
Ba	455.4	8–10,000 ppm	Sinter
Be	313.0	4–10,000 ppm	Acid
Ca	317.9	0.04–100 %	Sinter
Со	228.6	4–10,000 ppm	Acid
Cr	267.7	4–10,000 ppm	Acid
Cu	324.7	4–10,000 ppm	Acid
Fe	271.4	0.04–100 %	Sinter
K	766.4	0.04–100 %	Sinter
Li	670.7	8–10,000 ppm	Acid
Mg	285.2	0.04–100 %	Sinter
Mn	257.6	16–10,000 ppm	Acid
Na	588.9	0.01–100 %	Acid
Ni	231.6	8–10,000 ppm	Acid
Р	213.6	0.04-100 %	Sinter
Sc	424.6	8–10,000 ppm	Acid
Si	251.6	0.04–100 %	Sinter
Sr	421.5	8–10,000 ppm	Acid
Th	401.9	16–10,000 ppm	Acid
Ti	334.9	0.04-100 %	Sinter
V	292.4	8–10,000 ppm	Acid
Y	321.6	8–10,000 ppm	Acid
Zn	213.8	8–10,000 ppm	Acid
Zr	339.1	16–10,000 ppm	Sinter

The 11 elements reported from the Na_2O_2 sinter decomposition are the major elements Al, Ca, Fe, K, Mg, P, Si, and Ti, and the trace elements B, Ba, and Zr. The Na_2O_2 sinter technique was used to decompose resistant mineral phases like barite and zircon (for the elements Ba and Zr) and to make soluble boron and silicon, which are volatilized in the acid decomposition. The large dilution factor (1:400) does not degrade the reportability for the major elements because of their high concentrations in the ashed coals.

Coal ash sample solutions from the sinter decomposition were prepared in the following manner: a 0.100-g sample and 0.5 g of finely ground Na_2O_2 were mixed in a 5-mL graphite crucible with a Teflon stirring rod and sintered for 35 minutes at 445°C. The crucible was allowed to cool to

 Table 2.
 Operating conditions for ICAP-AES (inductively coupled argon plasma-atomic emission spectrometry).

250 W .5 L/min 8 L/min .7 mL/min 4.5 mm above load coil Iddified Babington
Iodified Babington

room temperature and was placed in a 50-mL Teflon beaker. Twenty milliliters of deionized H₂O, 20 mL of 20 percent HNO₃, and 200 μ g of lutetium in solution (an internal standard) were added in that order. The solution was transferred to a 13 ∞ 100-mm test tube and analyzed by ICAP-AES. As discussed earlier, this solution was used for both ICAP-AES and ICAP-MS. (See ICAP-MS analysis in Meier's paper in this volume.)

RESULTS AND DISCUSSION

Table 3 gives the results of triplicate analyses of the eight ashed coal samples digested by the acid decomposition method. Table 4 presents the results of triplicate analyses of the eight ashed coal samples digested by the sinter decomposition technique. Generally, the precision for both decomposition techniques is within $\pm 5-10$ percent relative standard deviation (RSD).

The copper content of sample ND PC-8-3 is dissimilar to the contents of the other replicates and is attributed to a contaminated acid digestion. A high value occurs for zinc in sample PITT PC-4-1. Again, the probable explanation is contamination from the acid digestion. Boron and barium have erroneous values for samples UF PC-1-3 and show large disagreement for all splits of WV PC-7. Low values of MgO, CaO, and TiO₂ for sample IL PC-3-3 are attributed to the sinter preparation rather than ICAP-AES analysis.

Confirmation of accuracy was evaluated by data for NIST standard reference material 1632a and USGS standard reference material CLB-1 that have undergone the two decomposition techniques. Tables 5 and 6 compare data for this study from two ashed coal standards with values from other studies (Gladney and others, 1984; J.S. Kane, unpub. data, 1990). The data from this study show good agreement with results from other studies. **Table 3.** Concentrations of Na_2O (in weight percent) and trace elements (in parts per million) in eight Argonne Premium Coal samples ashed and then digested by acid decomposition.

[Three analyses were de	one for each sample by ICAP-AES	inductively coupled argon p	lasma-atomic emission spectrometry)]

Coal ash	Na ₂ O	Be	Co	Cr	Cu	Li	Mn	Ni	Sc	Sr	Th	V	Y	Zn
UF PC-1-1		11	39	160	140	110	330	110	29	430	21	200	63	140
UF PC-1-2		11	39	160	140	110	330	110	29	430	21	200	60	140
UF PC-1-3	.35	11	40	160	130	110	320	110	29	430	19	200	56	140
WY PC-2-1		3	18	74	150	46	240	55	19	3,000	16	170	42	150
WY PC-2-2	1.9	3	21	76	160	46	240	57	19	3,000	19	170	42	130
WY PC-2-3	1.9	3	20	75	160	46	220	55	20	3,100	15	170	43	140
IL PC-3-1		4	29	300	67	46	570	180	17	190	15	240	26	1,100
IL PC-3-2		4	27	210	69	47	520	120	17	200	13	230	26	1,300
IL PC-3-3	.86	4	26	210	64	46	520	120	16	190	12	220	23	1,200
PITT PC-4-1	.50	8	28	170	59	89	230	97	27	690	19	178	43	150
PITT PC-4-2		8	29	170	68	90	230	100	27	700	15	180	42	88
PITT PC-4-3	.49	8	28	170	62	89	220	100	27	680	17	180	43	88
POC PC-5-1	2.1	15	75	190	310	100	340	140	33	2,100	23	230	110	100
POC PC-5-2		15	73	180	310	100	340	139	34	2,000	20	230	110	100
POC PC-5-3	2.1	15	75	180	310	110	350	140	33	2,100	22	230	110	100
UT PC-6-1	3.9	3	21	150	94	120	93	79	17	1.300	13	100	45	100
UT PC-6-2	3.8	3	20	110	92	120	92	74	16	1,300	13	96	43	110
UT PC-6-3	3.8	3	20	110	89	120	89	75	16	1,300	12	95	43	100
WV PC-7-1	.24	11	38	200	110	140	71	81	38	320	32	220	63	55
WV PC-7-2		11	39	200	110	144	70	84	38	320	33	220	67	54
WV PC-7-3		11	38	200	110	140	72	81	38	320	34	220	62	58
ND PC-8-1	7.0	2	8	23	48	28	850	14	8	5,200	6	41	23	54
ND PC-8-2		$\overline{2}$	10	23	51	28	870	16	8	5,200	8	40	24	49
ND PC-8-3	7.0	2	9	26	130	28	870	15	8	5,200	7	41	23	50

Table 4. Concentrations of major oxides (in weight percent) and trace elements (in parts per million) in eight

 Argonne Premium Coal samples ashed and then digested by sinter decomposition.

Coal ash	Al_2O_3	CaO	Fe ₂ O ₃	K ₂ O	MgO	P_2O_5	SiO_2	TiO ₂	В	Ba	Zr
UF PC-1-1	21	4.1	17	2.5	0.92	0.10	40	0.89	260	390	180
UF PC-1-2	22	4.2	18	2.7	1.0	.10	41	.92	230	380	180
UF PC-1-3	23	4.2	18	2.6	.98	.10	42	.96	360	470	190
WY PC-2-1	14	17	5.2	.38	3.8	.57	25	.95	1,000	3,300	210
WY PC-2-2	14	16	5.9	.50	3.8	.56	26	.97	1,300	3,400	220
WY PC-2-3	16	19	5.8	.54	4.4	.67	28	1.1	1,100	3,600	230
IL PC-3-1	16	8.3	21	1.7	.80	.06	41	.74	1,000	510	170
IL PC-3-2	15	8.1	22	1.6	.77	.05	40	.73	1,000	480	140
IL PC-3-3	14	7.5	20	1.6	.70	.04	36	.66	970	530	120
PITT PC-4-1	20	2.8	19	1.5	.60	.23	42	.93	520	420	200
PITT PC-4-2	20	2.8	19	1.5	.60	.22	42	.92	530	420	200
PITT PC-4-3	19	2.9	19	1.5	.59	.22	41	.89	510	410	190
POC PC-5-1	21	12	14	.83	1.8	.10	31	1.3	260	4,100	320
POC PC-5-2	20	12	13	.82	1.7	.10	30	1.2	240	3,800	300
POC PC-5-3	20	12	13	.72	1.7	.09	29	1.2	240	3,800	290
UT PC-6	15	12	9.0	.70	1.0	.04	41	.85	2,700	710	360
UT PC-6-2	16	13	9.1	.73	1.1	.04	43	.85	2,700	730	390
UT PC-6-3	16	13	9.6	.76	1.1	.03	44	.89	2,900	800	370
WV PC-7-1	30	.40	3.0	2.9	.70	.09	53	1.9	200	570	340
WV PC-7-2	33	.44	2.9	3.3	.77	.09	59	2.1	400	740	380
WV PC-7-3	31	.42	2.8	3.0	.72	.09	55	2.0	360	620	350
ND PC-8-1	7.9	21	6.7	.56	6.1	.30	15	.32	830	5,500	120
ND PC-8-2	7.9	21	7.0	.41	6.2	.30	15	.32	850	5,600	140
ND PC-8-3	7.8	21	6.9	.45	6.1	.29	15	.31	830	5,600	120

[Three analyses were done for each sample by ICAP-AES (inductively coupled argon plasma-atomic emission spectrometry)]

Table 5. Concentrations of Na_2O (in weight percent) and trace elements (in parts per million) in two coal reference materials ashed and then digested by acid decomposition.

[USGS, U.S. Geological Survey; ICAP-AES, inductively coupled argon plasma-atomic emission spectrometry; NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards]

	USG	S CLB-1	NIST	1632a
	This study, ICAP-AES	J.S. Kane (USGS, unpub. data, 1990) ^{1,2}	This study, ICAP-AES	Gladney and others (1984) ^{1,2}
Na ₂ O (%)	0.28	0.31	0.48	0.48
Be (ppm)	14	16	5	6
Co (ppm)	91	87	29	27
Cr (ppm)	130	130	150	150
Cu (ppm)	140	140	68	70
Li (ppm)	85	110	130	150
Mn (ppm)	110	120	120	130
Ni (ppm)	250	250	79	85
Sc (ppm)	27	25	27	27
Sr (ppm)	880	930	370	380
Th (ppm)	20	18	20	19
V (ppm)	160	160	180	190
Y (ppm)	64	58	34	34
Zn (ppm)	660	660	120	120
% ash	7.8		23.5	

¹Reported on a whole-coal basis and converted to an ash basis for comparison.

²Compilation of data obtained by various methods.

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Table 6. Concentrations of major oxides (in weight percent) and trace elements (in parts per million) in two coal reference materials ashed and then digested by sinter decomposition.

[USGS, U.S. Geological Survey; ICAP-AES, inductively coupled argon plasma-atomic emission spectrometry; NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards]

	USG	S CLB-1	NIST	1632a
	This study, ICAP-AES	J.S. Kane (USGS, unpub. data, 1990) ^{1,2}	This study, ICAP-AES	Gladney and others (1984) ^{1,2}
Al ₂ O ₃ (%)	18	19	23	24
CaO (%)	2.6	2.9	1.4	1.4
$Fe_2O_3(\%)$	14	16	6.6	6.8
K ₂ O (%)	.87	.96	2.1	2.1
MgO (%)	.53	.62	.70	.85
P ₂ O ₅ (%)	.89	.95	.22	.25
SiO ₂ (%)	30	33	53	55
$TiO_{2}(\%)$.91	.98	1.1	1.2
B (ppm)	82	46	220	230
Ba (ppm)	430	470	440	530
Zr (ppm)	170	160	230	230
% ash	7.8		23.5	

¹Reported on a whole-coal basis and converted to an ash basis for comparison.

²Compilation of data obtained by various methods.

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Determination of 33 Elements in Coal Ash from 8 Argonne Premium Coal Samples by Inductively Coupled Argon Plasma-Mass Spectrometry

By Allen L. Meier

ABSTRACT

Thirty-three elements were determined in the ash of eight Argonne Premium Coal samples by inductively coupled argon plasma-mass spectrometry (ICAP-MS). Two sample digestion procedures were used, a sodium peroxide sinter to dissolve resistant minerals and an acid digestion technique for acid-soluble minerals in the coal ash. Hf, Ta, W, and the rare earth elements La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, and Yb were determined by ICAP-MS in the solution from the sodium peroxide sinter. Ga, Ge, As, Rb, Nb, Mo, Ag, Cd, Sn, Sb, Te, Cs, Au, Tl, Pb, Bi, and U were determined by ICAP-MS in the acid solution. These solutions were also used for inductively coupled argon plasma-atomic emission spectrometry (ICAP-AES) determination of other elements to give nearly total elemental coverage except for the volatile elements, halogens, and elements not retained because of combustion in the ashing process. The technique to determine the value for each element was selected to provide the best possible precision and determination limit while minimizing interferences.

INTRODUCTION

Inductively coupled argon plasma-mass spectrometry (ICAP-MS) is one of the newest instrumental analytical techniques to be used for elemental determination in geologic materials. The U.S. Geological Survey (USGS) began use of the technique in 1985 and developed methods using ICAP-MS for the determination of rare earth elements (REE's) and platinum-group elements and for the analysis of coal. ICAP-MS is attractive for these applications because it has multielement measurement capabilities with very low detection limits. Most elements are detected directly in solutions in the range of 1 to 100 pg/mL. These detection limits are often 100 to 1,000 times lower than those routinely achieved by inductively coupled argon plasma-atomic emission spectrometry (ICAP-AES). The response is linear with concentration over about 6 to 8

orders of magnitude, making calibration quite uncomplicated. Another advantage of ICAP-MS is that the mass spectra of elements are relatively simple. Problems with the techniques arise from spectral interferences from molecular species and effects from the sample matrix. These problems are minimized by the use of corrections for spectral overlaps and internal standards for matrix effects. For the analysis of coal, ICAP-MS and ICAP-AES are used as complementary techniques. ICAP-AES is used to determine the elements that normally have higher concentrations in the coal ash. These are primarily the lower mass elements where the ICAP-MS technique has more interferences. The ICAP-AES technique is also used for other elements that are normally found in coal ash above the detection limits for the technique and where the precision by this technique is better than the precision of the ICAP-MS technique. ICAP-MS is used to determine the elements where a lower limit of detection is necessary to determine normal concentrations found in coal and for elements where the ICAP-AES technique suffers from interferences.

EXPERIMENTAL

SINTER METHOD

Hf, Ta, W, and the rare earth elements La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, and Yb are made soluble in a 0.1-g coal ash sample by sintering with sodium peroxide, leaching with water, and acidifying with nitric acid in a preparation technique modified from one described by Borsier and Garcia (1983). Details of the procedure are described by Briggs in this volume. The elements were then determined by ICAP-MS at lower reporting limits, in parts per million, of 2.0 La, 3.0 Ce, 0.5 Pr, 2.0 Nd, 0.5 Sm, 0.2 Eu, 1.0 Gd, 0.5 Tb, 0.2 Dy, 0.5 Ho, 0.2 Er, 0.5 Tm, 0.5 Yb, 1.0 Hf, 1.0 Ta, and 1.0 W. Lutetium was added as an internal standard to correct for instrument instability and oxide interferences. Two-point calibration for each element was made by using the average intensity of five blanks taken through the entire procedure, and the intensities were acquired on a solution of a glass reference standard containing a known concentration of each element. The standard solution was run at 15 sample intervals, drift was calculated, and correction was applied between standards. All new determinations reported in this paper result from work done in the USGS laboratory in Denver, Colo.

INTERFERENCES FOR THE SINTER METHOD

Isobaric interferences of some metal oxide ions are quite high for selected REE's. Therefore, conditions that minimize the oxide ions were used. The delivery line from the nebulizer spray chamber to the plasma torch was cooled to 10°C to reduce the amount of water vapor (the main source of oxygen) that enters the plasma. Compromise conditions of power and the sheath gas flow rate were selected to achieve a balance between sensitivity and oxide. A method modified from Lichte and others (1987) was used to minimize and correct for these oxide isobaric overlaps. Oxide interference was subtracted by using the ratio of oxide ions to element ions in single-element standards and the oxide/ion ratio of the internal standard. In a 3-hour period of running samples, the oxide ratios can drift by as much as 100 percent. This is probably due to a gradual closing of the sampler cone, although several factors are involved. The oxide correction of PrO on gadolinium-157 must be very accurate. Even after the oxide abundance is minimized, a 10 percent error in the oxide ratio correction can result in a 20 percent error in the gadolinium result. The metal oxide/metal ion ratios of the REE's all responded similarly to plasma conditions (Lichte and others, 1987). Lutetium, used as an internal standard, was also used to track the drift in the oxide/metal ratio through a sample run. The oxide ratios of overlapping elements were measured in standard solutions and compared to the LuO+/Lu+ response. These ratios were used to mathematically subtract the oxide interference and to correct for changes in the other metal oxide ratios due to matrix or drift in the sample run.

ACID DIGESTION METHOD

Ga, Ge, As, Rb, Nb, Mo, Ag, Cd, Sn, Sb, Te, Cs, Au, Tl, Pb, Bi, and U were made soluble in a 0.2-g coal ash sample by heating with a mixture of hydrochloric, nitric, perchloric, and hydrofluoric acids (Crock and others, 1983). Details of the procedure are described by Briggs in this volume. The elements were then determined by ICAP-MS at lower reporting limits, in parts per million, of 0.1 Ga, 0.5 Ge, 1.0 As, 0.5 Rb, 2.0 Nb, 0.5 Mo, 0.5 Ag, 0.2 Cd, 1.0 Sn, 0.5 Sb, 0.5 Te, 0.1 Cs, 0.1 Au, 0.5 Tl, 2.0 Pb, 0.1 Bi, and 0.2 U. Lutetium and indium were added as internal standards to correct for instrument instability and oxide interferences. Two-point calibration for each element was made by using the average intensity of five blanks taken through the entire procedure, and the intensities were acquired on a solution of a glass reference standard containing a known concentration of each element. The standard solution was run at 15 sample intervals, drift was calculated, and correction was applied between standards. Oxide interference was subtracted by using the ratio of oxide ions to element ions in single-element standards and the oxide/ion ratio of the internal standard.

INTERFERENCES FOR THE ACID DIGESTION METHOD

Interferences in ICAP-MS come from matrix effects, instrumental drift, and isobaric overlap of some elemental isotopes and molecular ions formed in the plasma, resulting in suppression or enhancement of measured ion intensity. An internal standard was added to minimize matrix effects and instrumental drift. The isotopes measured were selected to minimize isobaric overlap from other elements and molecular species that might be present. Oxide overlaps were subtracted by measuring the ratio of oxide to element for single-element standards in each run and applying this ratio to each sample.

RESULTS AND DISCUSSION

The values obtained for triplicate analyses of the eight Argonne Premium Coal reference samples digested using the sinter method are given in table 1. The values obtained for triplicate analyses of the eight Argonne Premium Coal reference samples prepared by using the acid digestion method are given in table 2.

The wide elemental coverage and the low limits of determination of the ICAP-MS technique make it a worthwhile tool for the analysis of coal ash. The accuracy and precision of the methods are adequate for the determination of trace elements in coal ash. Tables 3–6 show values determined by ICAP-MS on solutions obtained by the two dissolution methods of the ash of National Institute of Standards and Technology (NIST) reference standard materials 1632b (coal) and 1633a (coal fly ash). The tables compare values obtained in this study with the reference values. These comparisons show that reasonable accuracy is achieved by these methods. Unfortunately, many elements determined have not been reported for these reference materials, so accuracy

Table 1. (Concentrations of rare earth and other elements determined by triplicate ICAP-MS analyses of eight Argonne Premium Coal samples that were ashed and
	y the sinter method.

Sample no. La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Но	Er	Tm	Yb	Hf	Та	W
UF PC-1-1 55 UF PC-1-2 60 UF PC-1-3 63	111	13.5	53	10.9	2.4	10	1.6	10.2	1.9	5.4	1.0	5.5	6.0	1.6	8.3
	128	14.5	55	11.2	2.6	12	1.8	10.9	2.3	6.2	1.1	5.8	5.9	1.4	9.3
	136	14.9	58	12.2	2.5	11	1.8	12.2	2.4	6.2	1.1	6.8	6.8	1.3	10
WY PC-2-1 47	94	10.8	43	8.3	1.8	7.6	1.2	7.4	1.4	4.1	.78	3.8	7.6	1.5	5.7
WY PC-2-2 44	90	10.2	40	7.6	1.9	7.2	1.2	8.3	1.6	3.7	.70	4.2	8.0	1.4	4.8
WY PC-2-3 51	100	11.7	44	8.9	2.2	8.8	1.3	8.8	1.6	4.7	.79	5.2	8.4	1.2	5.8
IL PC-3-1	80	8.8	34	6.6	1.4	6.2	.84	6.3	1.2	3.4	.54	3.2	5.7	1.2	12
	81	9.3	35	6.7	1.2	5.4	.94	5.6	1.0	2.9	.60	3.2	4.4	1.1	13
	67	7.3	28	5.3	1.1	5.0	.72	4.9	.91	2.6	.48	3.2	4.2	.99	11
PITT PC-4-1 52 PITT PC-4-2 50 PITT PC-4-3 50	116	12.8	51	9.9	1.8	8.4	1.2	8.4	1.7	4.1	.79	4.8	6.9	1.4	9.7
	112	12.1	46	8.2	2.0	7.6	1.3	8.4	1.6	4.0	.78	4.5	6.6	1.5	9.4
	106	11.6	44	8.8	1.8	7.4	1.2	8.7	1.5	4.3	.75	4.7	6.5	1.4	8.9
POC PC-5-1 113	227	26.1	104	19.9	4.1	20	3.2	22.2	4.3	11.3	1.8	10.9	11	2.5	22
POC PC-5-2 106	212	24.5	93	18.2	4.0	20	3.1	20.1	3.9	10.4	1.8	10.8	11	2.1	21
POC PC-5-3 102	206	22.7	90	18.2	3.9	18	2.9	19.4	3.9	10.1	1.8	10.7	10	1.9	19
UT PC-6-1 52	97	11.4	46	7.9	1.7	7.6	1.1	7.1	1.3	3.7	.63	4.1	11	1.2	11
UT PC-6-2 56	105	12.1	46	8.2	1.9	8.3	1.3	7.5	1.4	4.2	.72	4.3	13	1.3	11
UT PC-6-3 54	102	12.1	45	8.5	1.9	8.4	1.2	7.5	1.4	4.2	.68	4.6	12	1.2	11
WV PC-7-1 86	184	19.9	75	14.0	3.1	12	1.9	12.6	2.5	6.9	1.2	7.4	12	3.2	8.5
WV PC-7-2 92	198	22.1	82	15.2	3.2	14	2.2	14.3	2.8	6.8	1.4	8.6	12	3.5	91
WV PC-7-3 90	191	20.8	74	14.1	3.2	11	2.0	13.7	2.5	6.9	1.3	7.9	12	2.9	8.6
ND PC-8-122ND PC-8-224ND PC-8-321	43	4.3	16	2.8	.56	3.4	.50	3.8	.73	2.0	< .5	2.2	4.3	<1	3.6
	44	4.9	17	2.7	.55	2.8	.50	3.7	.73	2.4	< .5	2.6	4.7	<1	4.0
	42	4.4	15	2.5	.61	3.0	.55	3.4	.71	2.3	< .5	2.4	4.3	<1	3.8

[All concentrations are in parts per million. ICAP-MS, inductively coupled argon plasma-mass spectrometry]

Table 2. Concentrations of trace elements determined by triplicate ICAP-MS analyses of eight Argonne Premium Coal samples that were ashed and dissolved by the the acid digestion method.

Sample no. A	g	As	Au	Bi	Cd	Cs	Ga	Ge	Мо	Nb	Pb	Rb	Sb	Sn	Te	Tl	U
UF PC-1-1 2.7) 1	22	< 3	0.89	0.97	13.7	40.4	32.5	22.4	19	57	166	3.8	7.1	<0.5	9.2	8.7
UF PC-1-2 3.0		18	< 3	.89	1.6	13.5	38.6	31.8	16.6	17	57	160	3.4	7.2	<.5	8.9	8.6
UF PC-1-3 4.0		19	< 3	.83	.98	13.8	39.6	32.4	16.5	18	55	163	3.7	7.0	<.5	8.7	8.4
WY PC-2-1 2.0	1	34	<3	.65	1.2	2.4	21.9	4.2	6.1	16	19	21.2	1.8	3.8	<.5	.90	7.6
WY PC-2-2 2.0		31	<3	.65	1.3	2.4	21.4	4.2	5.8	15	19	21.2	1.8	3.8	<.5	.85	7.3
WY PC-2-3 3.9		22	<3	.58	1.7	2.2	24.0	4.5	6.3	18	18	21.2	2.2	3.8	<.5	.79	7.4
IL PC-3-1 2.7	3	22	<3	.55	4.1	8.8	21.3	50.0	44.7	12	53	96.4	4.5	4.3	<.5	10.5	42.2
IL PC-3-2 3.8		24	<3	.54	4.7	8.8	22.5	53.2	34.4	14	54	99.3	5.0	4.6	<.5	10.4	40.6
IL PC-3-3 4.9		25	<3	<.5	5.8	8.3	22.4	52.4	37.4	15	50	95.9	5.2	4.3	<.5	9.1	36.7
PITT PC-4-1 2.0	2	78	<3	1.3	1.1	8.7	34.5	12.3	7.7	17	41	87.0	2.0	6.2	<.5	6.3	6.5
PITT PC-4-2 3.2		83	<3	1.3	.96	8.8	35.2	13.3	7.9	18	43	87.7	2.2	6.2	<.5	6.2	6.1
PITT PC-4-3 4.9		89	<3	1.2	1.4	8.7	37.4	15.1	8.4	20	40	90.2	2.3	6.3	<.5	5.9	6.0
POC PC-5-1 2.0	2 1	76	<3	.92	.64	4.3	32.1	4.4	49.3	19	34	38.5	5.4	6.9	<.5	4.4	12.2
POC PC-5-2 4.2		85	<3	.98	1.0	4.3	33.9	5.1	50.0	20	32	39.8	5.5	6.9	<.5	4.5	11.6
POC PC-5-3 6.0		98	<3	1.0	1.7	4.1	35.7	6.8	53.8	25	31	41.2	6.1	7.2	<.5	4.4	11.9
UT PC-6-1 2.	5.	7.2	<3	.77	1.4	3.1	20.6	4.1	8.9	11	31	21.9	1.9	3.6	<.5	.62	18.5
UT PC-6-2 3.		<1	<3	.78	1.7	3.1	21.5	5.1	9.0	12	32	22.8	1.9	3.7	<.5	.64	17.9
UT PC-6-3 4.2		<1	<3	.75	1.9	3.0	22.8	5.5	9.8	14	32	23.6	2.0	3.5	<.5	.66	17.8
WV PC-7-1 3.0	1	32	<3	1.2	.26	12.3	45.8	8.2	6.3	29	61	178	2.7	9.0	<.5	2.7	11.2
WV PC-7-2 6.4		34	<3	1.2	.23	12.2	49.6	8.8	6.6	31	65	189	2.8	9.0	<.5	2.8	11.4
WV PC-7-3 9.		37	<3	1.2	.37	12.0	49.8	8.8	6.8	34	64	188	3.2	9.1	<.5	2.7	10.8
ND PC-8-1 9 ND PC-8-2 1.0 ND PC-8-3 2.1	5	17 7.4 <1	<3 <3 <3	< 0.5 < 0.5 < 0.5	.31 .45 .52	$1.1 \\ 1.0 \\ 1.1$	12.7 12.0 12.1	3.4 3.9 4.3	4.1 4.4 4.6	7.8 8.4 8.8	20 8.8 16	14.4 14.8 15.4	1.4 1.5 2.0	2.9 3.0 5.9	<.5 <.5 <.5	<.5 <.5 <.5	4.4 4.9 4.8

[All concentrations are in parts per million. ICAP-MS, inductively coupled argon plasma-mass spectrometry]

Table 3. Comparison of reference values with mean concentrations of trace elements determined by 23 replicate ICAP-MS analyses of NIST standard reference material 1632b (coal) that was ashed and dissolved by the acid digestion method.

[All concentrations are in parts per million (ppm). The reference values for NIST standard reference material 1632b (coal) are from National Bureau of Standards (1985). ICAP-MS, inductively coupled argon plasma-mass spectrometry; NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards; RSD, relative standard deviation (in percent); STD, standard deviation (in parts per million); —, no data]

	Ref.		This study	
Element	value (ppm)	Mean (ppm)	STD (ppm)	RSD (%)
Ag As Au Bi Cd	54.7 	1.7 63 <3.0 1.7 1.1	$ \begin{array}{r} 1.3 \\ 4.2 \\ - \\ .2 \\ .3 \\ \end{array} $	75 7
Cs Ga Ge Mo Nb	6.5 13.2 	6.0 44 33 13 24	1.4 3.1 4.5 .5 1.3	24 7 13 4 5
Pb Rb Sb Sn Te Tl	54.0 74.3 	63 75 3.6 9.5 <.5 2.3	8.2 21.2 .3 .5 - .3	$ \begin{array}{r} 13 \\ 28 \\ 8 \\ 5 \\ - \\ 12 \\ 2 $
U	6.4	6.6	.6	8

Table 4.Comparison of reference values with meanconcentrations of trace elements determined by 23 replicateICAP-MS analyses of NIST standard reference material1633a (coal fly ash) that was dissolved by the acid digestionmethod.

[All concentrations are in parts per million (ppm). The reference values for NIST standard reference material 1633a (coal fly ash) are from National Bureau of Standards (1979). ICAP-MS, inductively coupled argon plasma-mass spectrometry; NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards; RSD, relative standard deviation (in percent); STD, standard deviation (in parts per million); -, no data]

	Ref.		This study	
Element	value (ppm)	Mean (ppm)	STD (ppm)	RSD (%)
Ag As Au Bi Cd	145 1	2.3 172 <3.0 1.3 1.2	1.7 9.1 - .2 .2	76 5 -13 15
Cs Ga Ge Mo Nb	11 58 33.9 29	10 65 39 35 30	1.9 5.3 2.5 1.0 2.5	19 8 6 3 8
Pb Rb Sb Sn Te	72.4 131 6.8 10	83 148 7.3 9.2 <.5	7.4 30.7 .5 .7	9 21 7 7
Tl U	5.7 10.2	6.5 12	.6 .9	9 8

Table 5. Comparison of reference values with mean concentrations of rare earth and other elements determined by 22 replicate ICAP-MS analyses of NIST standard reference material 1632b (coal) that was ashed and dissolved by the sinter method.

[All concentrations are in parts per million (ppm). The reference values for NIST standard reference material 1632b (coal) are from National Bureau of Standards (1985). ICAP-MS, inductively coupled argon plasma-mass spectrometry; NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards; RSD, relative standard deviation (in percent); STD, standard deviation (in parts per million); —, no data]

	Ref.		This study	
Element	value (ppm)	Mean (ppm)	STD (ppm)	RSD (%)
La Ce Pr Nd Sm	75.0 132.4 — 12.8	64.3 124.6 13.6 53.2 10.9	3.11 7.18 .73 2.98 .65	4.8 5.8 5.4 5.6 6.0
Eu Gd Tb Dy Ho	2.5 	2.5 10.2 1.5 9.4 1.7	.19 .84 .11 .57 .11	7.5 8.3 7.0 6.1 6.4
Er Tm Yb Hf Ta W	 6.3 7.1	5.3 .8 5.0 6.8 2.0 6.8	.36 .07 .37 .55 .24 .51	6.8 8.6 7.5 8.1 12.1 7.5

cannot be estimated using these materials. Precision is given as standard deviation and relative standard deviation for each element determined in the reference materials. For most elements, precision is better than 10 percent relative standard deviation. Precision for some elements is poorer, especially as detection limits are approached. Silver concentrations determined on solutions obtained by the acid digestion method have the most variation. The lack of precision can be attributed to sampling variation as well as instrumental variation.

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Borsier, M., and Garcia, M., 1983, Analyse automatique d'echantillons geologiques par plasma ICP: Spectrochimica Acta, v. 38B, nos. 1/2, p. 123–127. **Table 6.** Comparison of reference values with mean concentrations of rare earth and other elements determined by 22 replicate ICAP-MS analyses of NIST standard reference material 1633a (coal fly ash) that was dissolved by the sinter method.

[All concentrations are in parts per million (ppm). The reference values for NIST standard reference material 1633a (coal fly ash) are from National Bureau of Standards (1979). ICAP-MS, inductively coupled argon plasma-mass spectrometry; NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards; RSD, relative standard deviation (in percent); STD, standard deviation (in parts per million); —, no data]

	Ref.	f. This study						
Element	value (ppm)	Mean (ppm)	STD (ppm)	RSD (%)				
La Ce Pr Nd Sm	180.0 — — —	87.30 168.12 19.16 75.71 16.34	5.28 12.36 1.32 4.91 .82	6.0 7.4 6.9 6.5 5.0				
Eu Gd Tb Dy Ho	4.0 	3.97 16.16 2.51 15.48 2.88	.22 1.52 .18 1.04 .16	5.7 9.4 7.1 6.7 5.5				
Er Tm Yb Hf Ta W	 7.6 	8.51 1.22 7.65 8.09 2.14 6.11	.44 .08 .46 .82 .19 .52	5.1 6.7 6.0 10.2 9.0 8.5				

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Determination of Mercury and Selenium in Eight Argonne Premium Coal Samples by Cold-Vapor and Hydride-Generation Atomic Absorption Spectrometry

By Richard M. O'Leary

ABSTRACT

The methods for the determination of mercury and selenium in whole coal by cold-vapor atomic absorption spectrometry (CVAAS) and hydride-generation atomic absorption spectrometry (HGAAS) are described. The Argonne Premium Coal samples were analyzed in triplicate to determine the precision of the method. The averaged values ranged from 0.01 to 0.39 μ g/g for mercury and 0.60 to 6.2 μ g/g for selenium. Mercury and selenium were also determined in standard reference materials from the National Institute of Standards and Technology (NIST), the Community Bureau of Reference (BCR) of the Commission of the European Communities, the National Research Council of Canada (NRCC), and the U.S Geological Survey (USGS). Results obtained by these methods were compared with the published values to determine the accuracy of the methods.

INTRODUCTION

Analytical techniques for the determination of mercury and selenium in coal are increasing in importance. With the passage of the 1990 Clean Air Act Amendments (CAAA), the Environmental Protection Agency (EPA) has been given the authority to set emission standards for a number of potentially hazardous air pollutants (HAP's) generated by a number of specific combustion sources. Mercury, selenium, and nine other elements present in coal are among the 189 pollutants identified as air toxins in the CAAA legislation. Although their concentrations in coal are minor, they represent a potentially significant release of mercury and selenium to the environment because of the large tonnage of coal burned in powerplants.

The most common instrumental technique for determining mercury is cold-vapor atomic absorption spectrometry (CVAAS). Some CVAAS techniques preconcentrate the mercury using a gold amalgam, then thermally release the mercury by inductively heating the gold; however, alternative techniques analyze the mercury vapor directly. Other instrumental techniques include graphite furnace atomic absorption spectroscopy (GFAAS), cold-vapor atomic fluorescence spectroscopy (CVAFS), and neutron activation analysis (NAA). Various sample preparation procedures include oxygen bomb combustion, direct coal combustion, microwave digestion, and hotplate acid digestion.

In the procedure described here, a 0.150-g coal sample was decomposed by a heated mixture of nitric and sulfuric acids and vanadium pentoxide. The sample solution was introduced to a continuous-flow CVAAS system, where it was complexed and reduced with a solution of hydroxylamine hydrochloride and sodium chloride prior to further reduction with stannous chloride. The mercury vapor was then separated from the liquid in the phase separator before entering the quartz cell for the determination of the mercury concentration.

Like mercury, selenium can be determined by a number of different instrumental techniques such as GFAAS, NAA, and hydride-generation atomic absorption spectrometry (HGAAS). Of the AAS techniques, hydride generation is the technique of choice because it is relatively interference free as compared to GFAAS.

In the method used in this study, a 0.100-g sample of pulverized coal was digested at 150–200°C with a mixture of concentrated sulfuric, nitric, and perchloric acids until a clear to yellow solution was reached. After the addition of 6 M hydrochloric acid, the solution was allowed to set to permit the selenium to reduce the Se⁺³ state. The diluted solution was then introduced by way of an autosampler and peristaltic pump to a Varian VGA-76 hydride generator coupled to an atomic absorption spectrophotometer. The sample solution was then mixed with concentrated hydrochloric acid and 0.35 percent sodium borohydride, and the resultant selenium hydride was then transported with argon gas to an air-acetylene flame-heated quartz cell for atomization and estimation of the selenium concentration.

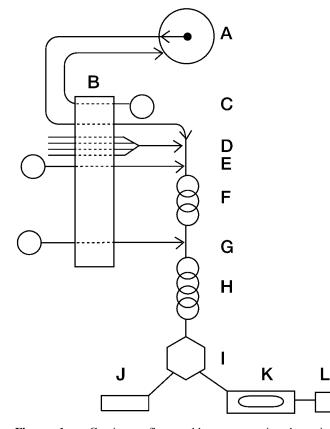


Figure 1. Continuous-flow cold-vapor atomic absorption spectrometry (CVAAS) manifold for determining mercury. A, Autosampler, 8 mL/min; B, 8-channel peristaltic pump; C, 1 percent nitric acid wash, 7 mL/min; D, air, 48 mL/min; E, complexing-reducing solution, 3.5 mL/min; F, 20-turn mixing coil; G, stannous chloride solution, 3.5 mL/min; H, 80-turn mixing coil; I, liquid-gas separator; J, liquid to waste; K, flow-through cell in atomic absorption spectrophotometer; L, strip chart recorder.

PROCEDURE FOR MERCURY

Approximately (scooped) 0.1 g vanadium pentoxide,^{1,2} 1.5 mL concentrated sulfuric acid, and 3.5 mL concentrated nitric acid were added to 0.150 g of whole coal in a $16\infty150$ -mm disposable test tube and mixed. The test tube was placed in an aluminum heating block and covered with a watch glass. The temperature was ramped gradually to 150° C over a 2-hour period. The tube was heated overnight at this temperature and then removed and allowed to cool. The sample was diluted to 15 mL with water, capped, and shaken for 5 minutes. It was then centrifuged at 1,000 rpm (revolutions per minute) for 5 minutes, and approximately
 Table 1. Operating conditions for determination of mercury and selenium by cold-vapor atomic absorption spectrometry (CVAAS) and hydride-generation AAS (HGAAS), respectively.

	Hg	Se
Source	Hg hollow cathode lamp.	Electrodeless discharge
	cathode famp.	lamp (EDL).
Slit	0.7 nm	0.7 nm
Wavelength	253.7 nm	196.0 nm
Quartz cell temperature	100°C	2,000°C
Mode	Absorbance	Absorbance
Recorder	2 mV, 5 mm/ min	10 mV, 5 mm/ min
AA recorder mode	TC3	TC3
Sensitivity (peak height)	$1 \ \mu g/L = 19 \ mm$	$10 \ \mu g/L = 72 \ mm$

12 mL of solution was transferred to a $16\infty100$ -mm disposable test tube.

The mercury was determined by using a continuousflow-through CVAAS system (fig. 1) as described by Kennedy and Crock (1987). The test tube containing the sample solution was placed in the autosampler. The sample solution was fed from the autosampler by a peristaltic pump into a continuous-flow system, where it was mixed with a reducing-complexing solution of 3 percent hydroxylamine hydrochloride (reagent grade) and 3 percent sodium chloride (reagent grade) in 10 percent sulfuric acid. Next, the sample was further reduced with a 10 percent stannous chloride in 10 percent hydrochloric acid. The sample then entered a phase separator where the mercury gas passed through the flow-through cell of the AAS for measurement and the liquid was discharged to waste. The absorbance indicating the mercury concentration was recorded on a strip chart, and peak heights were measured. The operating conditions for the AAS are shown in table 1.

The samples were compared against a calibration curve generated by analyzing standards in the 1- to $15-\mu g/L$ range. The calibration standards were made by dilution of a 1.47- $\mu g/mL$ mercury solution (standard reference material (SRM) 1641c) obtained from NIST (National Institute of Standards and Technology). The calibration standards contained a final concentration of 3.7 M nitric acid, 1.8 M sulfuric acid, and 0.5 percent (w/v, weight per volume) sodium dichromate (reagent grade).

PROCEDURE FOR SELENIUM

The method used for determining selenium is a modification of that described by Aruscavage (1977). Twenty milliliters of concentrated hydrochloric acid³ and 2 mL concentrated sulfuric acid were added to 0.100 g of whole coal

¹Unless otherwise noted, all chemicals used in the mercury determinations are of Baker "Instra Analyzed" quality or are labeled "Suitable for mercury determinations," and the water is deionized.

 $^{^{2}}$ Some brands of vanadium pentoxide (reagent grade) contain trace amounts of mercury and need to be roasted at 500°C prior to use.

³Unless otherwise noted, all chemicals used in the selenium determinations are Baker "Instra Analyzed" or of equal purity, and the water is deionized.

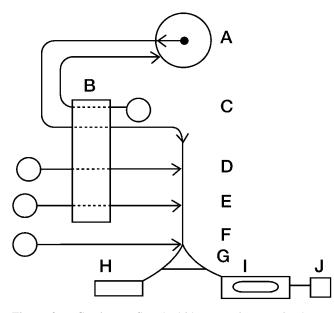


Figure 2. Continuous-flow hydride-generation atomic absorption spectrometry (HGAAS) manifold for determining selenium. A, Autosampler, 8 mL/min; B, 4-channel peristaltic pump; C, deionized water wash, 8 mL/min; D, concentrated hydrochloric acid, 1 mL/min; E, sodium borohydride solution, 1 mL/min; F, nitrogen purge gas, 90 mL/min; G, liquid-gas phase separator; H, liquid to waste; I, heated quartz furnace in atomic absorption spectrophotometer; J, strip chart recorder.

in a 250-mL Erlenmeyer flask. After the mixture was allowed to stand overnight, 3 mL of perchloric acid (redistilled) was added. A claw refluxer was added to the flask, and the solution was heated on a hotplate at $150-200^{\circ}$ C for 30 minutes. The refluxers were removed, and heating of the solution was continued until the development of dense white fumes and a clear to yellow solution. The solution then was removed from the hotplate. When the solution was cool, 25 mL of 6 M hydrochloric acid was added, and the solution was allowed to stand for 30 minutes to permit the selenium to reduce to the Se⁺³ state. The contents of the flask were transferred to a 60-mL polyethylene bottle and diluted to 55 g with water.

The selenium was determined by the HGAAS system using a Varian VGA-76 hydride generator coupled to a Perkin-Elmer atomic absorption spectrophotometer (fig. 2). The solution was transferred to $13\infty100$ -mm test tubes and placed in an autosampler. The sample solution was then fed from the autosampler to the hydride-generation system by a peristaltic pump, where it was mixed with concentrated hydrochloric acid and 0.35 percent sodium borohydride. The selenium hydride was then transported with argon gas to the air-acetylene flame-heated quartz furnace of the AAS for atomization. The absorbance indicating the selenium concentration was registered on a strip chart recorder, and peak heights were measured. The operating conditions for the AAS are shown in table 1.

Table 2. Mercury (Hg) and selenium (Se) contents in eight Argonne Premium Coals determined by cold-vapor and hydridegeneration atomic absorption spectrometry (CVAAS and HGAAS), respectively.

[All values in micrograms per gram $(\mu g/g)$]

Sample ID		Hg, CVAAS		Se,
	1st day	2d day	Average	HGAAS
UF PC-1-1 UF PC-1-2 UF PC-1-3 Average	0.38 .37 .44	0.39 .40 .37 —	 0.39	2.3 2.0 1.5 1.9
WY PC-2-1 WY PC-2-2 WY PC-2-3 Average	.13 .12 .11 —	.11 .13 .14	 .12	1.8 2.8 2.5 2.4
IL PC-3-1 IL PC-3-2 IL PC-3-3 Average	.09 .10 .09	.10 .08 .09	 09	4.1 4.0 4.3 4.1
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 Average	.15 .17 .15	.16 .17 .16	 16	1.8 1.6 1.6 1.7
POC PC-5-1 POC PC-5-2 POC PC-5-3 Average	.06 .06 .06	.06 .06 .05	 06	2.9 2.9 3.1 3.0
UT PC-6-1 UT PC-6-2 UT PC-6-3 Average	.02 .01 .01	.01 .01 .01	 01	1.2 1.3 1.3 1.3
WV PC-7-1 WV PC-7-2 WV PC-7-3 Average	.08 .07 .10	.08 .08 .10	 08	6.1 5.6 6.9 6.2
ND PC-8-1 ND PC-8-2 ND PC-8-3 Average	.08 .08 .07	.08 .08 .08 	 08	.60 .60 .60

The samples were compared against a calibration curve generated by analyzing standards in the 5- to $20-\mu g/L$ range. The calibration standards were made by dilution of a commercially prepared $10-\mu g/g$ selenium standard in 10 percent HCl. The calibration standards contained a final concentration of 3 M hydrochloric acid and 0.72 M sulfuric acid.

DISCUSSION

The eight Argonne Premium Coal samples were analyzed for selenium in triplicate on one day and for mercury in triplicate on two nonconsecutive days. All analyses were performed in the U.S. Geological Survey laboratory in Denver, Colo. The averaged values for mercury range from 0.01 to 0.39 μ g/g, and the averaged values for selenium range from 0.60 to 6.2 μ g/g as shown in table 2. As a measure of quality control, several standard reference materials were **Table 3.** Comparison of recommended values of mercury concentrations in standard reference materials with mean concentrations determined in this study by cold-vapor atomic absorption spectrometry (CVAAS).

[*Column headings*: RV, recommended value of mercury concentration from the references A–E; SDEV, standard deviation; %RSD, relative standard deviation, in percent; %R, percent recovery, which compares the values obtained by CVAAS in this study with the recommended values; µg/g, micrograms per gram. *Sources*: NIST, U.S. National Institute of Standards and Technology, formerly the National Bureau of Standards (NBS); NRCC, National Research Council of Canada; USGS, U.S. Geological Survey; BCR, Community Bureau of Reference of the Commission of the European Communities. *References*: A, Lengyel and others, 1994; B, NBS, 1974; C, NBS, 1978a; D, Govindaraju, 1989; E, Griepink and others, 1986]

Standard So				DV	Thi			
	Source Reference Description		RV (µg/g)	Mean (µg/g)	SDEV (µg/g)	%RSD	%R	
SRM 1632b	NIST	А	Coal	0.077 ± 0.017	0.068	0.005	7.4	88
SRM 1632	NIST	В	Coal	$.12 \pm 0.02$.091	.011	12	76
SRM 1632a	NIST	С	Coal	$.13 \pm 0.03$.119	.007	5.9	91
MESS-1	NRCC	D	Marine sediment	.171	.175	.02	11	102
SDO-1	USGS	D	Shale	.19	.182	.011	6	96
CRM 181	BCR	E	Coal	$.138 \pm 0.011$.143	.005	3.5	104

Table 4. Comparison of recommended values of selenium concentrations in standard reference materials with mean concentrations determined in this study by hydride-generation atomic absorption spectrometry (HGAAS).

[*Column headings*: RV, recommended value of selenium concentration from the references A–C; SDEV, standard deviation; %RSD, relative standard deviation, in percent; %R, percent recovery, which compares the values obtained by HGAAS in this study with the recommended values; µg/g, micrograms per gram. *Sources*: NIST, U.S. National Institute of Standards and Technology, formerly the National Bureau of Standards (NBS); USGS, U.S. Geological Survey. *References*: A, NBS, 1978b; B, Stephen A. Wilson, USGS, oral commun., 1994; C, Eric P. Welsch, USGS, written commun., 1991]

Standard Source Reference				RV	Thi	s study, HGA	AAS	
	Reference	Description	κν (μg/g)	Mean (µg/g)	SDEV (µg/g)	%RSD	%R	
SRM 1635 CLB-1 ALF	NIST USGS USGS	A B C	Coal Coal Alfalfa	0.9 2.1 .19	0.95 2.5 .18	0.10 .2 .02	11 8 11	108 119 95

also analyzed for mercury and selenium by CVAAS and HGAAS, and the results are reported in tables 3 and 4, respectively. The standards chosen were coal, shale, a plant, and marine sediment. The tables compare the published values of the reference materials with the mean and standard deviation obtained from these methods.

The lower limits of determination for mercury and selenium are 0.01 and 0.1 μ g/g, respectively, which are based on three times the standard deviation of the blank. The precision of the mercury values is in the range of 3.5 to 12 percent relative standard deviation (%RSD), and the accuracy, based on the percent recovery (%R), which compares this method's values with the recommended values (RV), ranges from 76 to 104 percent (table 3). The precision of the values for selenium is in the range of 8 to 11 %RSD, and the accuracy or %R ranges from 95 to 119 percent of the recommended values (table 4). Both methods offer a

technique that is simple and rapid, and both are applicable to a wide range of organically based samples.

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Determination of Carbon, Hydrogen, and Nitrogen in Eight Argonne Premium Coal Samples by Using a Gas Chromatographic Analyzer with a Thermal Conductivity Detector

By Carol J. Skeen and Zoe A. Brown

ABSTRACT

The carbon, hydrogen, and nitrogen contents of eight Argonne Premium Coals were determined by using the Perkin-Elmer 240B elemental gas chromatographic (GC) analyzer with a thermal conductivity detector (TCD). Precision for the analysis of these samples is within the accepted 0.1 percent relative standard deviation. The carbon content ranged from 56 to 86 percent; the hydrogen content ranged from 3.7 to 5.6 percent; and the nitrogen content ranged from 0.93 to 2.2 percent. Because these ranges are typical for coals, the NIST (National Institute of Standards and Technology) 1635 coal standard reference material was chosen as the control standard to evaluate the accuracy of the method.

INTRODUCTION

Analysis of a substance for carbon, hydrogen, and nitrogen generally requires drastic treatment of the material in order to convert the elements into a form readily determined by routine analytical techniques. A common way to convert the carbon, hydrogen, and nitrogen to gaseous products is to carry out an oxidation in a quartz combustion tube through which is forced a stream of carrier gas. The stream transports the volatile products to the part of the apparatus where they can be separated for measurement.

The combustion train is packed with silver compounds to remove any halogen and sulfur compounds generated, because these compounds interfere with the determination of carbon dioxide and water. Before reaching the combustion train, the helium and oxygen flow through scrubbers packed with colorcarb and anhydron to remove extraneous contaminants.

The Perkin-Elmer 240B elemental gas chromatographic analyzer with a thermal conductivity detector gives excellent results for finely ground, dry materials, especially **Table 1.** Comparison of published concentrations of carbon, hydrogen, and nitrogen in NIST standard reference material 1635 (coal) with concentrations determined in this study by using a gas chromatographic analyzer with a thermal conductivity detector (GC/TCD).

[All concentrations are in weight percent. NIST, National Institute of Standards and Technology, formerly the National Bureau of Standards (NBS)]

Element	This study, GC/TCD	NBS (1978)	Gladney and others (1987)
C	63.0	62.2 ± 1.8	62.6
H	3.98	3.96 ± 0.03	4.07
N	1.5	1.0 ± 0.1	1.26

materials high in organic matter, with the following concentration ranges: 0.1–100 percent for carbon, 0.01–12 percent for hydrogen, and 0.10–18 percent for nitrogen.

ANALYTICAL PROCEDURE

Standard operating procedures for use of the Perkin-Elmer 240B and a revised statistical computer program (Abramowitz, 1964) were implemented for the analysis of the eight Argonne Premium Coals. Oxygen was the combustion gas, and helium was the carrier gas. The instrument was calibrated by oxidizing three samplings of standard acetanilide, all of approximately the same weight (1.0 to 1.3 mg). The furnace temperatures were 950°C for the combustion tube and 650°C for the reduction tube. The sample weights used were between 1.0 mg and 1.3 mg.

The NIST (National Institute of Standards and Technology, formerly the National Bureau of Standards) 1635 coal standard reference material (SRM) was analyzed as a control standard at the same time as the last three premium coals. Table 1 shows these results along with the NIST SRM

Table 2. Comparison of published concentrations of carbon, hydrogen, and nitrogen in eight Argonne Premium Coals with concentrations determined in this study by using a gas chromatographic analyzer with a thermal conductivity detector (GC/TCD).

[All concentrations are in weight percent]

		С		Н	Ν		
Coal samples	This study	Vorres (1990)	This study	Vorres (1990)	This study	Vorres (1990)	
UF PC-1-1 UF PC-1-2	77 76		4.6 4.7		2.2 1.8		
UF PC-1-3 UF Average		74.23	4.7	4.08	2.0	1.35	
WY PC-2-1 WY PC-2-2 WY PC-2-3	60 60		4.4 4.6		1.2 1.2		
WY Average WY Repeat	60 61	68.43	4.5	4.88	1.2	1.02	
IL PC-3-1 IL PC-3-2 IL PC-3-3	64 65		4.5 4.5		1.6 1.8		
IL PC-3-3	65	65.65	4.5	4.23	1.7	1.16	
PITT PC-4-1 PITT PC-4-2	74 75		5.0 5.0		2.2 2.0		
PITT PC-4-3 PITT Average .		75.50	5.0	4.83	2.1	1.49	
POC PC-5-1 POC PC-5-2 POC PC-5-3	86 86		4.4 4.3		1.9 2.0		
POC Average	86	86.71	4.4	4.23	2.0	1.27	
UT PC-6-1 UT PC-6-2 UT PC-6-3	72 74 74		5.5 5.5 5.6		1.6 1.8 1.4		
UT Average	73	76.89	5.5	5.49	1.6	1.50	
WV PC-7-1 WV PC-7-2 WV PC-7-3 WV Average	64 65 66 65	66.20	4.3 4.3 4.2 4.3	4.21	1.3 1.4 1.4 1.4	1.25	
ND PC-8-1	56	00.20	3.8		1.2	1.25	
ND PC-8-2 ND PC-8-3	57 57	(5.95	3.7 3.7	4.26	1.0 .93	1.0.4	
ND Average ND Repeat	57 57	65.85	3.7	4.36	1.0	1.04	

values (National Bureau of Standards, 1978) and the analytical values of Gladney and others (1987).

RESULTS AND DISCUSSION

A study was made to ascertain the detection limits of this method (Filby and others, 1985). By diluting pure acetanilide with ultra-pure silica to prepare three analytical standards— (1) 7.1 percent C, 0.67 percent H, and 1.04 percent N, (2) 0.71 percent C, 0.067 percent H, and 0.104 percent N, and (3) 0.071 percent C, 0.0067 percent H, and 0.010 percent N—and by using acetanilide undiluted (71.07 per-

cent C, 6.71 percent H, and 10.36 percent N), the lowest detection limits were calculated to be 0.1 percent C, 0.01 percent H, and 0.10 percent N. The experiment using various sampling weights also validated that the analytical curves were linear from the detection limit to the highest standard.

The results of these coal analyses were compared with the published data for the Argonne Premium Coal Sample Program (Vorres, 1990) for carbon, hydrogen, and nitrogen determined on dried whole coals. The values were in good agreement with all the coals except for the carbon values for the subbituminous coal (WY) and the lignite (ND). To verify the accuracy of this paper's results, analyses of these two samples were repeated with special attention given to proper drying of the samples before analysis. The carbon values obtained by the repeat analyses, 61 percent for WY and 57 percent for ND, are in agreement with values obtained by the initial analyses. Table 2 shows the results for the replicate analyses of these coals, the repeats, and Vorres' published data.

The differences between the results from Argonne National Laboratories (Vorres, 1990) and the data reported in this paper could be due to oxidation of these two coals. Argonne went to great lengths to seal these coals in an oxygen-free environment. Because analysis in the U.S. Geological Survey laboratories was not done immediately after the ampoules were opened, it is likely that the subbituminous coal and the lignite oxidized. Bituminous coals are characteristically more stable.

The precision of this method is within the 0.1 percent relative standard deviation, which is well within the accepted deviation for this type of analysis. The analysis of NIST 1635 indicates that the accuracy is also excellent.

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Compilation of Multitechnique Determinations of 51 Elements in 8 Argonne Premium Coal Samples

By Curtis A. Palmer and Sarah A. Klizas

ABSTRACT

Eight Argonne Premium Coal samples were analyzed by the U.S. Geological Survey. The concentrations of 51 elements were determined by two or more techniques on each sample. The analyses were performed by energy- and wavelength-dispersive X-ray fluorescence spectrometry, instrumental neutron activation analysis, inductively coupled argon plasma-atomic emission spectroscopy, atomic absorption spectrometry, inductively coupled argon plasmamass spectrometry, and direct-current arc spectrographic analysis. All data are compiled on a whole-coal basis for ease of comparison. The ash values are also included so that data can be converted to an ash basis if desired.

INTRODUCTION

Although the eight Argonne Premium Coal samples analyzed in this study are not defined as "reference standards" by Argonne National Laboratories, they are extremely important because of the care that has been taken in collection, preparation, and storage. A detailed description of the background information for these samples has been reported by Vorres (1990, 1993). However, these samples have not been widely analyzed for trace elements. The analytical laboratories of the U.S. Geological Survey analyzed these samples to further characterize them and to provide a foundation for a trace-element data base.

Most quantitative techniques used for elemental analyses of geologic samples offer high levels of precision and accuracy for selected elements in certain types of samples over specific ranges of concentrations, but all analytical techniques have certain characteristic limitations. For example, matrix-induced spectral interferences can result in incorrect determinations of trace elements. Even if properly corrected, these interferences may lead to reduced sensitivity or precision for a given element. Generally, the concentrations of elements determined by another technique on the same matrix will not be affected by the same interferences. A multitechnique approach for major- and trace-element analysis was taken to provide the high degree of reliability desired to characterize these materials. In addition, this information may be useful in evaluating data from a single technique for coal analysis for laboratories that do not have all techniques available. Semiquantitative analytical techniques, although not offering the precision or accuracy of the quantitative techniques, rapidly provide a large volume of data. Some of the data obtained by these low-precision techniques are not easily obtained by quantitative methods, but can be useful in the overall characterization of these materials.

This paper (1) summarizes the results of the multitechnique analyses of the Argonne Premium Coals, (2) discusses some discrepancies in the data, and (3) determines "recommended values" or "best averages" depending on the precision of the data. Each of the eight Premium Coal samples has been analyzed in triplicate for 68 elements. Fifty-one elements were determined by more than one technique. Although up to seven different techniques were used for some elements, there are not enough high-precision data to recommend values for all elements in all coals using common criteria for establishing such values (Kane and others, 1990). Therefore, modified criteria were designed for this data set. They allowed definition of "recommended values" on slightly less than half of the elements included in the data set.

SAMPLES AND TECHNIQUES

Three splits of each of the Argonne Premium Coal samples were analyzed by multiple techniques. The samples and the sample identification protocol are described in this volume by Palmer (see p. 1).

Ideally, solid samples of the whole coal would be analyzed by instrumental techniques because this type of analysis avoids problems caused by volatilization of elements during ashing and problems caused by incomplete sample dissolution. The procedures used for determining element concentrations instrumentally on the whole coal are discussed in this volume in the following papers:

Author	Procedure
Fletcher and Skeen	Quantitative DCAES, direct-current arc atomic emission spectrography
Evans and others	WDXRF, wavelength-dispersive X- ray fluorescence spectrometry
	EDXRF, energy-dispersive XRF
Palmer	INAA, instrumental neutron activa- tion analysis
O'Leary	CVAAS and HGAAS, cold-vapor and hydride-generation atomic absorption spectrometry
Skeen and Brown	Gas chromatographic analysis with a thermal conductivity detector

Although the sensitivity of INAA was acceptable for most of the 29 elements determined, the sensitivities of the other whole-coal procedures were marginal for many elements. Therefore, coal ash procedures were also used for WDXRF, EDXRF, and DCAES (see list below) to concentrate the trace elements and thereby increase sensitivities.

Techniques that require analysis of coal ash were used as described in the following papers in this volume:

Author	Procedure
Skeen and others	Semiquantitative DCAES
Evans and others	WDXRF EDXRF
Doughten	ICAP-AES, inductively coupled argon plasma-atomic emission spectrometry
	FAAS, flame atomic absorption spec- trometry
	GFAAS, graphite furnace atomic absorption spectrometry
Briggs	ICAP-AES
Meier	ICAP-MS, inductively coupled argon plasma-mass spectrometry

All samples were ashed at 525°C to limit volatilization of lead, cadmium, and other moderately volatile trace elements. Ash yields were determined on the same splits used for the analyses and were used to calculate data as if determined on a whole-coal basis. The 525°C ash yields, which are not directly comparable to those determined by ASTM (American Society for Testing and Materials, 1996) ash procedures (750°C) but are generally similar, can be used to recalculate back to an ash basis if desired.

DISCUSSION OF RESULTS

Direct comparison of results presented in previous papers in this volume is difficult because data are presented in three different forms depending on the analytical technique used and the material analyzed. Concentrations are reported on an ash basis for some procedures, on a wholecoal basis for some procedures, and on an oxide basis of the ash for major elements determined by WDXRF and DCAES. To facilitate a direct comparison of the data, the ash data have been recalculated to whole-coal values and converted to an element basis for those elements reported on an oxide basis. The entire recalculated data set for all splits can be found in appendix 1. The number of significant figures given in the original papers has been maintained in the converted values.

A careful examination of appendix 1 shows that analytical procedures can be classified into two categories: highly precise (HP) procedures shown in bold, which generally have a relative standard deviation of less than 5 percent, and procedures that are less precise (LP). The precision was calculated by determining the percent of difference between the three individual data points and their mean for each sample-element pair. Using the accuracy guidelines discussed later in this paper, each test for each element was given a rating of good, usable, or poor precision. Comparisons of the different ratings for all elements determined by each technique were made. Finally, the techniques were divided into the two precision groups (LP and HP) based on which rating they received most frequently. In this study, the two DCAES procedures (ash and whole coal) and the X-ray whole-coal procedures were classified as LP procedures; INAA, ICAP-AES, ICAP-MS, CVAAS, HGAAS, FAAS, GFAAS, and the other X-ray procedures were classified as HP procedures. It should be noted that no procedures had the same precision for all elements in all samples. For the designated HP techniques, most determinations were of high precision, but as expected, determinations near the detection limit for some samples had poorer precision. LP procedures generally had lower precision for all samples and elements.

Statistical approaches are useful for large data sets; however, often they do not provide the detail that is useful in evaluating individual problems in the data. Even though the individual samples were analyzed only in triplicate, the complete data set requires 18 pages (appendix 1). A summary of the data is given in tables 1 and 2. Table 1 presents the method averages of the major rock-forming-element data determined on each of the three splits of the eight Argonne Premium Coal samples. Table 2 is a similar table for the trace-element data.

Statistical analysis of the data in appendix 1 is given in appendix 2. These data include the number of samples for which values were determined, the arithmetic mean (mean), the standard deviation, the relative standard deviation, the

COMPILATION OF MULTITECHNIQUE DETERMINATIONS

Table 1. Average concentrations of major elements in weight percent based on triplicate analyses.

[Values in brackets are averages of two analyses, and numbers in parentheses are based on only one analysis. Data from high-precision techniques are in bold except outliers. An * indicates that the analysis was done in Denver. Complete data set is given in appendix 1. Material analyzed: (A) = ash, (C) = whole coal. -, no data]

	Argonne Premium Coal sample									
Element and technique –	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-8		
Si ICAP-AES(A)* WDXRF(A) DCAES(A) DCAES(C)	2.6 [2.67] 3.1 3	1.0 [1.24] 1.4 .77	2.9 [3.04] 3.6 3	1.8 [1.96] 2.7 2	0.75 [.804] .95 .88	0.91 (.893) .86 —	5.0 (4.90) 5.2	0.67 (.768) .60		
Al WDXRF(A) ICAP-AES(A) ICAP-AES(A)* DCAES(A) DCAES(C)	[1.54] 1.60 1.6 2.4 2	[.672] .699 .66 .53 .48	[1.21] 1.25 1.3 2.0 2	[.981] .997 .96 2.1 2	[.548] .549 .57 .71 .85	(.370) [.347] .38 .32 —	(3.17) 3.15 3.2 3.9 —	(.45) .400 .40 .23		
Fe INAA(C) WDXRF(A) ICAP-AES(A) DCAES(A) DCAES(A) DCAES(C)	1.78 [1.93] 1.86 1.7 1.6 2	.366 [.33] .367 .34 .27 .29	2.67 [2.70] 2.72 2.4 2.4 2	1.35 [1.44] 1.32 1.2 1.9 1	.509 [.521] .507 .50 .50 .69	.317 (.29) [.282] .30 .24	.397 (.37) .385 .39 .40	.547 (.45) .472 .46 .37		
Mg WDXRF(A) ICAP-AES(A) ICAP-AES(A)* DCAES(A) DCAES(C)	[.079] .082 .078 .15 .099	[.26] .239 .21 .37 .20	[.09] .077 .073 .12 .098	[.03] .038 .033 .073 .048	[.067] .0548 .0554 .11 .077	(.048) [.030] .030 .041 —	(.08) .094 .085 .097 —	(.42) .398 .351 .51		
Ca WDXRF(A) ICAP-AES(A) ICAP-AES(A)* DCAES(A) DCAES(C)	[.40] .445 .40 .64 .58	[1.12] 1.19 1.0 1.2 1	[.90] .960 .92 1.6 2	[.20] .206 .19 .41 .29	[.443] .456 .45 .58 .80	(.401) [.409] [.43] .29 —	(.06) .058 .058 .078 —	(1.54) 1.54 1.4 1.3 —		
Na INAA(C) ICAP-AES(A) ICAP-AES(A)* WDXRF(A) DCAES(A)	.0341 .032 .034 [.03] .045	.115 .119 .12 [.092] .13	.102 .097 .11 (.13) .15	.0343 .032 .034 [.03] .036	.0782 .0826 .0826 [.071] .13	.146 [.126] .13 (.13) .072	.0388 .034 .036 (.05) .036	.529 .466 .493 (.50) (.25)		
K INAA(C) WDXRF(A) ICAP-AES(A) DCAES(A) DCAES(A)	.269 [.27] .283 .29 .30	.0292 [.03] .028 .033 .033	.195 [.20] .194 .22 .26	.110 [.11] .110 .11 .13	.029 [.03] .028 .035 .036	[.022] (.02) [.014] .028 .017	.505 (.46) .495 .49 .36	[.029] (.03) .014 .037 .037		
Ti WDXRF(A) ICAP-AES(A) ICAP-AES(A)* DCAES(A) DCAES(C)	0.080 .079 .075 .053 .10	[0.051] .056 .051 .036 .044	[0.07] .070 .069 .066 .092	[0.059] .056 .050 .057 .091	[0.040] .038 .040 .028 .082	(0.02) [.023] .023 .019	(0.24) .260 .23 .18	(0.02) .020 .019 .018		
P WDXRF(A) ICAP-AES(A) ICAP-AES(A)* WDXRF(C) DCAES(A)	[.006] .0079 .0059 —	[.03] .025 .022 .03 .026	[.004] .0059 .004 	[.01] .011 .0080 .009	[.002] .0022 [.0023] 	(.001) [.00094] .0007 (.00074)	(.007) .0084 .008 (.017)	(.02) .012 .012 .02 .011		
Mn WDXRF(A) ICAP-AES(A) ICAP-AES(A)* DCAES(A) DCAES(C)	.0041 .0044 .0055 .0053			.0018 .0021 .0036 .0020			.0015 .0014 .0020	(.007) .0080 .0082 .010		

Table 2. Average concentrations of trace elements in parts per million based on triplicate analyses.

[Bracketed values are averages of two analyses, and numbers in parentheses are based on only one analysis. Data from high-precision techniques are in bold except outliers. An * indicates that the analysis was done in Denver. Complete data set is given in appendix 1. Material analyzed: (A) = ash, (C) = whole coal. -, no data]

Element and technique _	Argonne Premium Coal sample									
Element and teeninque =	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-		
i										
FAAS(A) ICAP-AES(A)*	15 15	4.1 3.9	7.8 7.7	8.6 8.9	5.8 5.5	5.1 5.5	29 27	2.7 2.7		
e										
ICAP-AES(A)	1.5	.25	.76	.77	.80	[.13]	1.9	[.18]		
ICAP-AES(A)* DCAES(A)	1.5 1.4	.3 .14	.6 .97	.7 1.2	.80 .67	.1 .20	2.1 3.0	.2 .28		
ICAP-AES(A)* DCAES(A)	38 30	96 70	160 120	48 51	13 17	130	56 57	79 67		
	100	1 (0			4 =0	010		0.44		
INAA(C)	4.06 3.9	1.68	2.59 2.7	2.57 2.5	1.79 1.8	.813 .75	7.62 7.4	.846 .8		
ICAP-AES(A)* DCAES(A)	3.9 2.1	1.6 1.3	2.6	2.5 1.9	1.8 1.3	.69	6.4	.8 .90		
ICAP-AES(A)	26	14	32	15	11	[4.0]	44	3.5		
ICAP-AES(A)*	27	14	37	17	12	4.5	43	3.9		
DCAES(A) DCAES(C)	16 41	9.0 14	24 52	13 24	7.1 18	3.7	32	3.6		
r										
INAA(C)	20.3	6.1	33.1	14.8	9.1	5.30	35.8	2.24		
EDXRF(A)	26.0	[8.2]	43.2	17.9	12.3	—	45.9	2.6		
ICAP-AES(A)	20	6.2	31	14	9.2	[4.8]	40	2.4		
ICAP-AES(A)*	22 19	6.4	36 35	16 8	9.7 5	5.2	39 49	2.3		
EDXRF(C) DCAES(A)	19	[4] 5.7	33	19	8.8	4.8	49 39	2.4		
DCAES(C)	32	7	52	24	16	_	_			
0										
INAA(C)	5.33	1.68	4.31	2.62	4.07	1.00	7.74	.778		
FAAS(A) ICAP-AES(A)*	5.0 5.3	1.6 1.7	4.3 4.4	2.4 2.6	3.8 4.0	[.83] .94	8.1 7.5	.9		
DCAES(A)	3.5	1.3	4.4 3.7	2.6	4.0 2.6	.64	6.3	.9		
DCAES(C)	5	_	_		5		_			
i										
INAA(C)	14.5	4.9	21.0	[10.3]	8.6	[3.4]	15.4	_		
EDXRF(A)	14 14	3.9 4.9	18 18	9.3 8.5	7.58 6.7	[2 2]	17 16	1.9 1.3		
ICAP-AES(A) ICAP-AES(A)*	14	4.9	[19]	8.5 9.1	0.7 7.4	[3.3] 3.5	16	1.3		
EDXRF(C)	23	8	32	14	11	5	18	6		
DCAES(A)	14	5.2	22	13	6.5	4.0	21	1.8		
DCAES(C)	25	—	33	12	[11]	—	—	_		
	20.4	12.9	9.6	6.2	12.5		23.6	3.3		
EDXRF(A) ICAP-AES(A)	20.4 19	12.9	9.0 10	0.2 5.5	12.5	[3.8]	23.0 19	3.3 4.7		
ICAP-AES(A)*	19	12	10	5.8	16	4.2	21	[4.6]		
EDXRF(C)	18	17	14	10	17	8	31	9		
DCAES(A)	17	13	11	6.5	19	2.9	16	4.9		
DCAES(C)	18	7	9	6	19	_	_	_		
$\mathbf{INAA}(\mathbf{C})$	19.7	11.3	220	9.1	6.4	6.35	13.5	5.69		
INAA(C) EDXRF(A)	20.5	11.5	107	9.1 8.4	0.4 4.8	-	13.5	5.09		
ICAP-AES(A)	20.3 19	10.0	1 80	7.8	4.0	[6.2]	12	3. 4 4.6		
ICAP-AES(A)*	19	12	190	[8.1]	5.3	4.8	11	4.8		
EDXRF(C)	33	25	140	15	14	_	10	5		
DCAES(C)	33	_	280	27	_	_	_			

COMPILATION OF MULTITECHNIQUE DETERMINATIONS

	Argonne Premium Coal sample										
Element and technique –	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-8			
a											
ICAP-AES(A)*	5.34	1.91	3.57	3.28	1.80	1.00	9.39	1.16			
DCAES(A)	6.2	2.4	4.7	4.7	2.3	1.2	12	1.7			
DCAES(C)	10	3	7	5	4	_	_	_			
e ICAP-MS(A)*	4.35	0.36	8.40	1.25	0.29	0.23	1.7	0.37			
DCAES(A)	4.55 3.6	0.50	8.9	1.23	.46		[1.1]				
DCAES(C)	4	_	9	(3)	.+0	_	—	_			
S											
INAA(C)	17.1	3.6	4.7	8.42	10.3	.48	6.2	2.63			
ICAP-MS(A)*	16.1	2.4	3.8	7.6	9.9	(.33)	6.7	(1.6)			
DCAES(A)	30	—	_	17	15	_	_	_			
	1.02	1.50	4.20	1 -	25	1.00		-0			
INAA(C) HGAAS(C)	1.92 1.9	1.56 2.4	4.29 4.1	1.5 1.7	2.5 3.0	1.08 1.3	5.4 5.8	.58 .60			
	10	2.1		1.7	0.0	1.0	0.0	.00			
b INAA(C)	19.5	_	16	7.7	_	.98	29.7	[.93]			
ICAP-MS(A)*	22.0	1.80	15.7	8.12	2.11	1.06	36.3	1.41			
EDXRF(A)	21.2	3.5	15	9.1	2.3	_	43.6	_			
EDXRF(C)	21	13	20	8	5	3	42	4			
ſ											
INAA(C)	49	252	39	61	105	70	60	597			
EDXRF(A)	60 50	260	29.9 20	68	110		78.0	640 500			
ICAP-AES(A) ICAP-AES(A)*	59 58	150 260	29 31	64 64	85 110	[60] 60	49 62	500 490			
EDXRF(C)	62	294	33	63	85	63	86	490 780			
DCAES(A)	63	220	40	91	110	89	70	630			
EDXRF(A)	10	3.6	4.1	4.9	6.89	-	21.2	_			
ICAP-AES(A)	9.3	3.8	4.2	4.3	6.2	2.0	11	[1.8]			
ICAP-AES(A)*	8.1	3.6	4.0	4.0	5.8	2.0	12	2.2			
EDXRF(C)	4.7	3.0	4.7	3.1	4.6	(2) 1.8	19 9.1	2.4			
DCAES(A)	4./	5.0	4./	5.1	4.0	1.0	9.1	2.4			
r ICAP-AES(A)*	24	19	23	18	16	17	69	12			
EDXRF(A)	27.3	22.7	22.3	20.5	16.9	_	80.5	6.5			
EDXRF(C)	23	19	23	17	10	18	103	18			
DCAES(A)	11	17	17	12	12	11	34	13			
DCAES(C)	33	—	21	24	30	_	_	_			
	2.4	1.4	2.2	1 7	1 1		(1	70			
ICAP-MS(A)*	2.4 2.5	1.4 1.0	2.2 [2.1]	1.7 1.9	1.1 .85	.57	6.1 8.3	.79			
EDXRF(A) EDXRF(C)	2.5	1.0	[2.1]	1.9	.05	_	0.3 [13]	_			
DCAES(A)	1.7	1.4	2.5	1.8	1.1	.51	3.6	.58			
lo											
ICAP-MS(A)*	2.50	.52	6.23	.74	2.67	.42	1.3	.42			
DCAES(A)	1.5	.67	5.2	.80	2.2	.48	1.4	.81			
g											
ICAP-MS(A)*	.46	.23	.62	.31	.23	.15	1.2	.16			
DCAES(A)	.21	.067	.367	.088	.099	.47	.45	_			
d TCAD MS(A)*	.17	.12	.79	.11	.072	.077	055	040			
ICAP-MS(A)* GFAAS(A)	.17 .071	.12 .094	.79 .61	.061	.072	.077 [.060]	.055 .078	.040 .046			
GFAAS(A)	· · · -					[]					
n ICAP-MS(A)* DCAES(A)	.96	.32	.72 2.4	.57 1.3	.37	.16 .37	1.7 1.9	.38			

 Table 2.
 Average concentrations of trace elements in parts per million based on triplicate analyses—Continued.

Element en déceloisere	Argonne Premium Coal sample								
Element and technique –	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-8	
Sb									
INAA(C) ICAP-MS(A)*	.531 .49	.191 .16	.877 .79	.233 .20	.539 .27	.107 .089	.521 .56	.153 .16	
Cs									
INAA(C) ICAP-MS(A)*	1.52 1.89	.191 .19	.877 1.4	.70 .80	.262 .22	.130 .14	1.86 2.33	.086 .10	
Ba									
INAA(C)	61	310	94	47	198	36	133	680	
EDXRF(A)	52	270	74.5	35.7	180		110	460	
ICAP-AES(A) ICAP-AES(A)*	55 56	15 290	81 82	41 39	16 210	31 35	120 120	420 530	
EDXRF(C)	50 60	410	121	39	140	33 29	210	1030	
DCAES(A)	41	320	60	38	210	36	130	580	
DCAES(C)	81	250	120	66	450	_			
La	10.1		C 10	< 1 -		2 21	<u> </u>	2.02	
INAA(C)	10.1	5.35	6.10 5 0	6.15	6.76	3.31	21.5	2.82	
ICAP-MS(A)*	7.7 8.2	4.0	5.9	4.6	5.67	2.5	17	1.9	
EDXRF(A) EDXRF(C)	0.2 16	6	15	[4.4] 9	2.2 5	10	17 16	8	
DCAES(A)	6.9	6.1	9.3	5.8	7.2	2.9	10	3.7	
Ce									
INAA(C)	18.3	9.4	12.6	11.3	11.6	4.81	35.8	4.45	
ICAP-MS(A)*	17.6	8.0	12	10.3	11.3	4.7	37.1	4.1	
EDXRF(A)	17.0	[5.2]	7.2	9.0	7.6	_	29.9	_	
EDXRF(C) DCAES(A)	40 (10)	25 9	40 (28)	17	10 12	15 6.9	55 26	19	
Pr									
ICAP-MS(A)	1.93	.926	1.3	1.12	1.29	.546	4.06	.43	
DCAES(A)	—	—	—	_	—	.42	[1.87]	—	
Nd INAA(C)	(8.0)					2.0	11.9	2.3	
ICAP-MS(A)*	7.5		5.2	4.3		2.0	11.9	2.5 1.5	
DCAES(A)	8	_	(6.5)	3.9	6.0	(2.9)	19	-	
Sm									
INAA(C)	1.97	.978	1.20	1.10	1.22	.508	3.52	.409	
ICAP-MS(A)* DCAES(A)	1.55	.71	1.0	.83	.995 .76	.37 .37	2.80 [1.7]	.26 (.47)	
Eu		—		_	.70	.57	[1./]	(.47)	
INAA(C)	.398	.198	.226	.214	.234	.099	.674	.081	
ICAP-MS(A)*	.34	.17	.20	.17	.21	.084	.61	.054	
DCAES(A)	_	_	_	_	_	(.11)	.61	_	
Tb	2(2	100	120	122	1/8	0.50	400	0.50	
INAA(C)	.262 .23	.120 .10	.138	.132	.167	.058	.400	.056	
ICAP-MS(A)*	.25	.10	.14	.11	.16	.055	.40	.055	
Yb INAA(C)	.88	.42	.52	.470	.555	.204	1.61	.287	
ICAP-MS(A)*	.00 .81	.42	.52	.470	.555	.204	1.01	.207	
DCAES(A)	.84	.29	.61	.54	.45	.20	1.5	.20	
Hf		<0.4		-0	100		1.02		
INAA(C) ICAP-MS(A)*	.66 .85	.601 .68	.55 .76	.50 .61	.438 .56	.478 .56	1.83 2.3	.341 .42	
Ta									
INAA(C)	.219	.146	.191	.166	.116	.048	.641	.092	
ICAP-MS(A)*	.20	.12	.18	.13	.11	.046	.583		

 Table 2.
 Average concentrations of trace elements in parts per million based on triplicate analyses—Continued.

	Argonne Premium Coal sample								
Element and technique -	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-8	
W INAA(C) ICAP-MS(A)*	1.06 1.2	.38 .46	1.52 1.62	.78 .86	.84 1.1	.42 .51	1.68 [1.6]	.35 .36	
Pb GFAAS(A) ICAP-MS(A)* DCAES(A) DCAES(C)	7.4 7.6 10 8	2.9 1.6 3.4	6.5 8.5 14 8	2.9 3.8 7.3 4	2.4 1.7 11 5	[1.6] 1.5 1.9 —	12 12 16	1.5 1.6 2.6	
Bi ICAP-MS(A*) DCAES(A)	.12	.053	[.088] —	.12 [1.8]	.052	035	.23	_	
Th INAA(C) ICAP-AES(A)*	2.26 2.7	1.71 1.4	1.96 2.2	1.48 1.6	1.16 1.2	.615 .58	6.41 6.4	1.07 .7	
U INAA(C) ICAP-MS(A)*	.83 1.2	.54 .63	4.27 6.46	.39 .57	.51 .631	.74 .852	1.64 2.16	[.49] .45	

Table 2. Average concentrations of trace elements in parts per million based on triplicate analyses-Continued.

geometric mean, the deviation of the arithmetic and geometric means, and an analysis of the kurtosis and skewness for the HP techniques excluding outliers (values with only one significant figure and HP values excluded because of the 40 percent rule discussed in the next paragraph). A similar analysis for all values, including LP values, outlier values, and other excluded values, is also given in appendix 2.

Another approach to analyzing the data is to define the agreement between techniques in a useful, nonstatistical manner and then discuss individual cases of disagreement. In a practical sense, for major elements (elements with concentrations generally greater than 0.1 percent; table 1), procedures are said to have "good accuracy" if the standard deviation of the individual determinations for a given sample determined by a given technique is ± 5 percent of the mean of all of the HP procedures and does not disagree by more than ± 0.5 pecent absolute. For trace elements (table 2) "good agreement" is defined as ± 10 percent of the mean. "Usable agreement" is four times the uncertainty of "good agreement" or ± 20 percent for the majors and ± 40 percent for the traces. Excluded from the agreement analysis were values of only one significant figure. If more than two tests were used, a mean value for a given technique differing by more than 40 percent from the mean of the remaining values was reason for excluding a given technique. This is the "40 percent rule." The excluded technique was said to have poor agreement for elements in those samples. In addition, the Grubbs test (Taylor, 1987) was made for all suspected HP outliers, using the mean and standard deviation for all HP techniques. Outliers are reported in table 3 under exceptions. They were excluded from the determination of agreement except for cases where the outlier was the only value for an element determined by a given technique. Figure 1 summarizes the decisions required to determine the agreement.

Trace-element criteria were applied to those samples containing elements that are traditionally considered major or minor elements (table 1), but whose mean concentration for HP techniques was less than 0.1 percent. This included phosphorus and manganese for all samples and magnesium, sodium, potassium, and titanium (see table 3) for four or more samples. It should be noted that the ± 5 percent criterion for "good agreement" is better than expected for some HP techniques for some samples as concentrations approach the detection limit. For example, counting errors of as high as 28 percent are reported for potassium by INAA (see paper by Palmer, this volume). The criterion of 20 percent required for "usable agreement" is much smaller than the inherent precision for many of the LP techniques. Skeen and others, in this volume, report possible errors of +50 percent or -33 percent because of the nature of the standards for each of the elements. It is not surprising, therefore, that not all HP techniques have "good agreement" for all samples and that most LP techniques have "poor agreement" for most samples. Most of these disagreements were within ±50 percent of the HP mean concentrations, and scatter in intertechnique comparison plots simply demonstrates the poorer precision of the LP techniques. A complete discussion of the precision of all values in this study is beyond the scope of this paper, but an indication of the precision can be obtained by examining appendix 1 and by the relative standard deviation given in appendix 2.

Table 3. Agreement between techniques for major elements using data from table 1 and criteria described in text.

[The following abbreviations were used under Exceptions: Samples: 1 = UF PC-1; 2 = WY PC-2; 3 = IL PC-3; 4 = PITT PC-4; 5 = POC PC-5; 6 = UT PC-6; 7 = WV PC-7; 8 = ND PC-8. Those that differ from the overall agreement are followed by g = good agreement; u = usable agreement; p = poor agreement; or x = no values or only upper limits found. High-precision (HP) techniques are in bold for consistency with other tables. HP values not used to calculate the mean are identified with an s for values with only one significant figure, f for values rejected because of the 40 percent rule (see text), or an o for single values rejected using the Grubbs outlier test (Taylor, 1987). All values including outliers were used to determine overall agreement. The mean of these values was used for comparison in evaluating low-precision values. Sample numbers in parentheses are samples where only single determinations are available. Agreement calculated using trace-element criteria because of low concentration in sample was designated with a t. Material analyzed: (A) = ash; (C) = whole coal; * = done in Denver]

Element	Exceptions to major elements	HP techniques	Overall agreement	Exceptions	LP techniques	Overall agreement	Exceptions
Si		ICAP-AES(A)* WDXRF(A)	Good Good	2u,3o,8= 2,4,5u,(8)ou	DCAES(A) DCAES(C)	Usable Poor	3,6,7g;2,4,5p 3sg;1s,4s,5u;6–8x
A1		WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Good	(8)ou (6)u 30,40,50	DCAES(A) DCAES(C)	Poor Poor	6u 1s,3s,4s;6–8x
Fe		INAA(C) WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Good Good	6,8u 1,2,4,(8)u 6u 1,3,4,8u	DCAES(A) DCAES(C)	Usable Usable	5g;2,4,8p 3s,4s,5p;6–8x;1s
Mg	. 1,3–7t	WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Usable Good Good	1g;(6)p;3s,4s,(7)s 2,8u	DCAES(A) DCAES(C)	Poor Usable	7g;6u 6–8x
Ca	. 7t	WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Good	(7)s 1,2u 2,4u;6o	DCAES(A) DCAES(C)	Poor Poor	2,7,8u 2su;6-8x;3s
Na	. 1,4,5,7t	INAA(C) ICAP-AES(A) ICAP-AES(A)* WDXRF(A)	Good Good Good Usable	6–8u 3,5,6,8u 2u (8)g;(7)sp;1s,4s	DCAES(A)	Usable	4,7g;3,5,6,(8)p
К	. 2,5,6,8t	INAA(C) WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Usable Good	6,8u (6)su;2s,5s,(8)s 1,3,4,7g;8fp 2,3,5,8u	DCAES(A)	Usable	3,7p
Ti	. 1–6,8t	WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Good	(6)su:3s,(8)s	DCAES(A) DCAES(C)	Usable Usable	3,4,8g;7p 4,5p;6–8x
Р	. 1–8t	WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Usable Good Good	4sg;(8)sp;1s,2s, 3s,5s;(6)s,(7)s 1,4,6u;3,7= 1,4;3s,6su;7s	WDXRF(C) DCAES(A)	Poor Poor	4sg;2su;8sp;1,3,5–7x 8g;7p;1,3–5x
Mn	. 1–8t	WDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Usable Good Good	1–7x,8sg	DCAES(A) DCAES(C)	Usable Usable	2,3,4,6p 4g;3p;6–8x

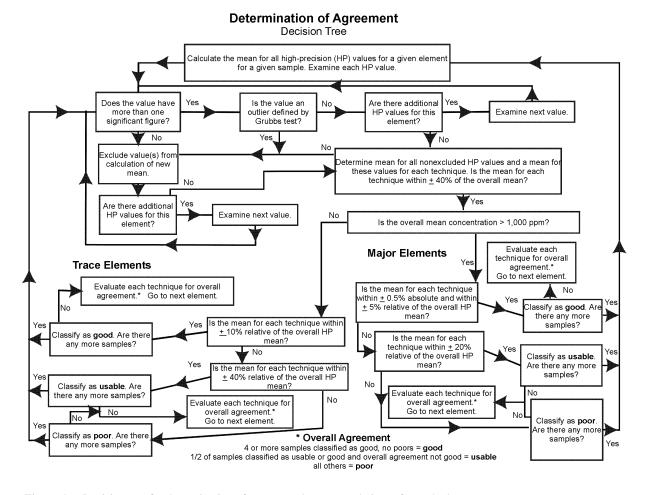


Figure 1. Decision tree for determination of agreement between techniques for each element.

A summary of the agreement for major elements is given in table 3. Data for elements by specific techniques were classified as in "overall good agreement" with the mean of the HP procedures if at least half of the individual samples were in "good agreement" using the previously mentioned criteria, and no samples had "poor agreement." If half or more of the samples determined were in "good" or "usable agreement" and the technique was not classified as in "good agreement," the element had "overall usable agreement." All others had "overall poor agreement" except where the technique was the only HP technique. In this case, agreement could not be determined. The mean of the samples for this sole HP technique was used to assess the agreement of the LP techniques, and no accuracy designation was given. All HP techniques listed in table 3 were in "overall good agreement" except for sodium, magnesium, manganese, and phosphorus determined by WDXRF on the ash, and potassium determined by ICAP-AES in Reston. These were classified as having "overall usable agreement."

In contrast, all LP techniques had "overall poor" or "overall usable agreement."

Table 4 summarizes the agreement of trace elements. All agreements were evaluated using trace-element criteria discussed previously and summary classifications similar to those in table 3. HP procedures generally had "good agreement" and none had "poor agreement," whereas LP procedures generally had "usable" or "poor agreement," with only an occasional "good agreement."

Most of the data fall within expected precision limits, but barium is an example of a case where determining an element by more than one technique can make a significant difference. Agreement between techniques can be graphically represented by plotting the concentrations of all elements determined by one technique versus all corresponding concentrations by a second technique and comparing these points to a theoretical line with zero intercept and a slope of 1. Figure 2, for example, shows the comparison of INAA and ICAP-AES data from table 2 for all elements that the two techniques have in common. There is

Table 4. Agreement between techniques for trace elements using data from table 2 and criteria described in text.

[The following abbreviations were used under Exceptions: Samples: 1 = UF PC-1; 2 = WY PC-2; 3 = IL PC-3; 4 = PITT PC-4; 5 = POC PC-5; 6 = UT PC-6; 7 = WV PC-7; 8 = ND PC-8. Those that differ from the overall agreement are followed by g = good agreement; u = usable agreement; p = poor agreement; or x = no values or only upper limits found. High-precision (HP) techniques are in bold for consistency with other tables. HP values not used to calculate the mean are identified with an s for values with only one significant figure, an f for values rejected because of the 40 percent rule (see text), or an o for single values rejected using the Grubbs outlier test (Taylor, 1987). All values including outliers were used to determine overall agreement. Equal sign "=" indicates there are no other high-precision values; italicized "good" agreements indicate that the only high-precision techniques used in determining the accuracy were modifications of the same tests, ICAP-AES(A) done in Reston and Denver. The mean of these values was used for comparison in evaluating low-precision values. Sample numbers in parentheses are samples where only single determinations are available. Material analyzed: (A) = ash; (C) = whole coal; * = done in Denver. —, no agreement could be calculated]

Element	HP techniques	Overall agreement	Exceptions	LP techniques	Overall agreement	Exceptions
Li	FAAS(A) ICAP-AES(A)*	Good Good				
Be	ICAP-AES(A)	Good	2–4,6,8=	DCAES(A)	Poor	1g;3,5u
	ICAP-AES(A)*	Good	2s,3s,6su;4s,8s			
В	ICAP-AES(A)*	—	1-8=	DCAES(A)	Usable	4,7g;6x
Sc	INAA(C) ICAP-AES(A)*	Good Good	8= 8s	DCAES(A)	Usable	3,8g;1p
V	ICAP-AES(A) ICAP-AES(A)*	Good Good		DCAES(A) DCAES(C)	Usable Poor	8g 2g;6–8x
Cr	INAA(C)	Good	7u	EDXRF(C)	Usable	3g;4s,5sp;6,8x; 2s
	EDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Usable Good Good	8g;6x;2o 3,4u	DCAES(A) DCAES(C)	Good Poor	1,2,4,5u 2sg;6–8x
Со	INAA(C) FAAS(A)	Good Good	8= 6u;8x	DCAES(A) DCAES(C)	Usable Usable	4g 1sg;2–4,6–8x; 5s
	ICAP-AES(A)*	Good	8su			
Ni	INAA(C) EDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Good Good	4,5u;8x 2,8u;6x 5,8u 1,3ou	EDXRF(C) DCAES(A) DCAES(C)	Poor Usable Poor	7u;2s,6s,8s 1g;4p 4u;2,6–8x
Cu	EDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Usable	7,8u;6x 5,7,8u 1–4,6,7g;8op	EDXRF(C) DCAES(A) DCAES(C)	Usable Usable Usable	1g;4,6s,7,8sp 2,3g;5p 1,4g;2,5p;6–8x
Zn	INAA(C) EDXRF(A)	Good Usable	3,5,8u 1,2,4,5,7,8g;3fp,6x	EDXRF(C)	Poor	8sg;3,7u;6x
	ICAP-AES(A) ICAP-AES(A)*	Good Good	5,8u 4o,6u	DCAES(C)	Poor	2,5–8x
Ga	ICAP-AES(A)*	_	1-8=	DCAES(A) DCAES(C)	Usable Poor	4,8p 6–8x;2s,3s,4s, 5s
Ge	ICAP-MS(A)*	_	1-8=	DCAES(A) DCAES(C)	Usable Usable	3,4g;5p;2,6,8x 1s,3sg;(4)sp; 2,5–8x
As	INAA(C) ICAP-MS(A)*	Good Usable	2,3,6,8u 1,3-5,7g;8o	DCAES(A)	Poor	2,3,6–8x
Se	INAA(C) HGAAS(C)*	Good Good	2u 2u,7o			
Rb	INAA(C)	Good	7,8u;2,5x	EDXRF(C)	Usable	1,4sg;2,5s,6s,
	ICAP-MS(A)* EDXRF(A)	Good Good	2,8u 2,7u;6,8x			8sp

Element	HP techniques	Overall agreement	Exceptions	LP techniques	Overall agreement	Exceptions
Sr	INAA(A) EDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Usable Good	1,3,6u 7,8u;6x 1,4,6g;2fp 8u	EDXRF(C) DCAES(A)	Good Usable	2,5,7,8u 1,5g;4,6p
Y	EDXRF(A)	Usable	1–3,5g;6,8x;7fp	EDXRF(C)	Usable	(6)sg,7p;1–5, 8x
	ICAP-AES(A) ICAP-AES(A)*	Good Good	8u 1u	DCAES(A)	Usable	1p
Zr	ICAP-AES(A)* EDXRF(A)	Good Good	8u,6= 8u;6x	EDXRF(C) DCAES(A) DCAES(C)	Usable Usable Usable	2,3,6g;8p 1,7,8p 3g;5p;2,6–8x
Nb	ICAP-MS(A)* EDXRF(A)	Good Usable	2,7u;6,8= 1,3,4g;6,8x	EDXRF(C) DCAES(A)	Poor Usable	1–6,8x 4,6g;7p
Мо	ICAP-MS(A)*	_	1-8=	DCAES(A)	Usable	4g;1,8p
Ag	ICAP-MS(A)*	_	1-8=	DCAES(A)	Poor	8x
Cd	ICAP-MS(A)* GFAAS(A)	Usable Usable	5,6,8g;1p 5,8g			
Sn	ICAP-MS(A)*	_	1-8=	DCAES(A)	Poor	7g
Sb	INAA(C) ICAP-MS(A)*	Good Good	5u 5u			
Cs	INAA(C) ICAP-MS(A)*	Good Good	1,3,7u 1,3,7u			
Ba	INAA(C) EDXRF(A) ICAP-AES(A) ICAP-AES(A)*	Good Good Usable Good	3,4,8u 4,7,8u;6x 1,3,4,6,7g;2f,5fp	EDXRF(C) DCAES(A) DCAES(C)	Usable Good Poor	1g;2,3,7,8p 1,3,8u 2u;6–8x
La	INAA(C)	Usable	3,5g	EDXRF(C)	Poor	2s,5s,7u;4s,6s, 8s
	ICAP-MS(A)* EDXRF(A)	Usable Usable	5,3,7g 1,7g;5fp;2,3,6,8x	DCAES(A)	Usable	6g;3,8p
Ce	INAA(C) ICAP-MS(A)* EDXRF(A)	Good Good Usable	2–5u 3,5u 1g;6,8x	EDXRF(C) DCAES(A)	Poor Poor	5g 2,5,7u;4,8x
Pr	ICAP-MS(A)*	_	1-8=	DCAES(A)	Poor	6u;1–5,8x
Nd	INAA(C) ICAP-MS(A)*	Usable Good	(1),6g;2–5x 8u;2–5=	DCAES(A)	Usable	1s,4g;7p;2,8x
Sm	INAA(C) ICAP-MS(A)*	Usable Usable	3g 3g	DCAES(A)	Poor	5,6u;1–4x
Eu	INAA(C) ICAP-MS(A)*	Good Good	4,8u 4,8u	DCAES(A)	Usable	6u,7g;1–5,8x
Tb	INAA(C) ICAP-MS(A)*	Good Good				
Yb	INAA(C) ICAP-MS(A)*	Good Good	8u 8u	DCAES(A)	Usable	1,6,7g
Hf	INAA(C) ICAP-MS(A)*	Good Good	1,3,5,7u 1,3,5,7u			

 Table 4.
 Agreement between techniques for trace elements using data from table 2 and criteria described in text— Continued.

Element	HP techniques	Overall agreement	Exceptions	LP techniques	Overall agreement	Exceptions
Та	INAA(C) ICAP-MS(A)*	Good Good	4u;8= 4u;8x			
W	INAA(C) ICAP-MS(A)*	Good Good	5u 5u;7o			
Pb	GFAAS(A) ICAP-MS(A)*	Good Good	2–5u 2–5u	DCAES(A) DCAES(C)	Poor Usable	1,6,7u 1s,3sg;5sp;2, 6–8x;4s
Bi	ICAP-MS(A)*	—	1–7=,8x	DCAES(A)	Poor	1–3,5–8x
Th	INAA(C) ICAP-AES(A)*	Good Good	8= 8su			
U	INAA(C) ICAP-MS(A)*	Usable Usable	2,6,8g 2,6,8g			

 Table 4.
 Agreement between techniques for trace elements using data from table 2 and criteria described in text—

 Continued.

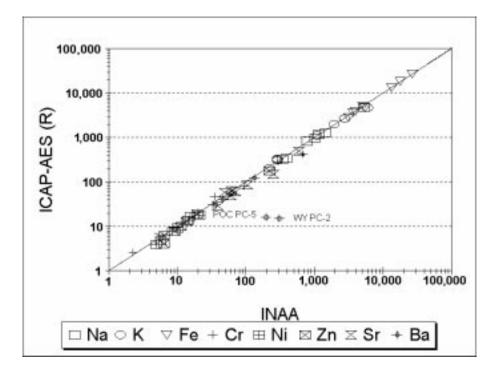


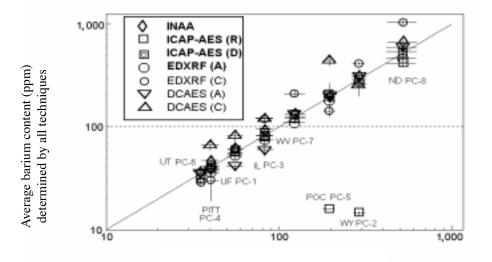
Figure 2. A comparison of average concentrations (in parts per million) of all elements determined by both inductively coupled argon plasma-atomic emission spectrometry in Reston, Va. (ICAP-AES (R)), on the ash and instrumental neutron activation analysis (INAA) on the whole coal. All data determined on the ash are converted to a whole-coal basis. The diagonal line is the line of perfect agreement.

relatively little scatter (excellent agreement) in most of the data; therefore, the few problems with the data are easily recognizable. The most obvious discrepancy in the data is that the barium concentration determined by ICAP-AES is more than an order of magnitude smaller in WY PC-2 and POC PC-5 than the concentration determined by INAA.

The data for barium determined by the seven different techniques are shown in figure 3, plotting barium determined by all techniques versus the mean barium concentration determined by high-precision techniques. Although there is scatter among data from different techniques, ICAP-AES (R) data for WY PC-2 and POC PC-5 are clearly off the correlation line. The disagreement of ICAP-AES (R) data with data from all other techniques suggests that barium is present in a species, probably BaSO₄, which is not dissolved by the Reston ICAP-AES acid dissolution procedures (see paper by Doughten, this volume). ICAP-AES procedures done in Denver use a sinter dissolution procedure and yield Ba concentrations that agree with the INAA data (see paper by Briggs, this volume). Both of these coals contain enough sulfate sulfur (Vorres, 1990) to account for all barium being $BaSO_4$ in the original coal. Solubility studies of these coals by Finkelman and others

(1990), however, show that barium in these two samples is soluble in ammonium acetate and is therefore readily exchangeable. This suggests that $BaSO_4$ is not in the original samples of WY PC-2 and POC PC-5 but that it is formed in the ashing process. Clearly, the stronger dissolution procedures (see paper by Briggs, this volume) should be used.

Some interesting consequences result from the "agreement" rules. Although zinc determined by EDXRF on the ash had "good agreement" in six samples, the mean of the zinc values for IL PC-3 was only slightly greater than onehalf of the zinc values determined by INAA and ICAP-AES for that sample. This difference led to a classification of "poor agreement" for zinc in IL PC-3 and an overall "usable agreement." Although these results are unusual and can be treated as outliers, as will be discussed later, they relate to the overall reliability of a technique. Outliers also led to usable ratings for copper determined by ICAP-AES (Denver), arsenic determined by ICAP-AES (Denver), yttrium determined by EDXRF, and barium determined by ICAP-AES (Reston). Barium had only "overall usable agreement" because of the "poor agreement" in WY PC-2 and POC PC-5, probably caused by incomplete dissolution. Strontium,



Mean barium content (ppm) determined by HP techniques

Figure 3. The comparison of mean concentration (in parts per million, ppm) in eight Argonne Premium Coal samples of barium determined by all high-precision techniques excluding outliers with the average concentrations of barium determined by all techniques. All plotted data are determined on, or converted to, a whole-coal basis. The diagonal line is the line of perfect agreement. The high-precision techniques in the legend are in bold. All techniques are defined in appendix 1; letters in parentheses have the following meanings: C, determined on whole coal; A, determined on ash (but converted to a whole-coal basis for this figure); R, analysis performed on ash by the U.S. Geological Survey (USGS) in Reston, Va.; D, analysis performed on ash by the USGS in Denver, Colo. Error bars given are the range in the y direction and the standard deviation of the mean in the x direction.

 Table 5.
 Recommended or average values for major elements in percentages, determined only by high-precision (HP) techniques using the criteria described in text.

[Parentheses indicate that the value is only an average of concentrations determined by the HP techniques and is not a recommended value. Complete data set is given in appendix 1. Statistical analysis of all values, including errors and criteria for not recommending values, is given in appendix 2. "Excluded values" column indicates that the specified HP techniques were not used in determining the given value. The bracketed numbers with the letters indicate the samples for which the techniques were excluded and the reason the values were rejected. The following abbreviations were used under "Excluded values": 1 = UF PC-1; 2 = WY PC-2; 3 = IL PC-3; 4 = PITT PC-4; 5 = POC PC-5; 6 = UT PC-6; 7 = WV PC-7; 8 = ND PC-8; o = determined value was an outlier; s = only one significant figure in all determined values; f = values rejected because of forty percent rule (see text); and n = no values were determined or all reported values were upper limits. All techniques referred to under "Excluded values" were determined on the ash]

	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-8	No. HP techniques	Excluded s values
Si	2.63	(1.1)	3.05	1.86	0.76	0.91	5.0	(0.67)	2	WDXRF [80]
Al	1.58	.68	1.24	.99	.552	(.37)	3.16	.400	3	WDXRF [80]
Fe	(1.8)	(.35)	(2.6)	(1.33)	.51	(.30)	.39	(.49)	4	
Mg	.080	(.23)	.075	.036	.058	(.033)	.090	(.38)	3	WDXRF [3s,4s,7s]
Ca	(.42)	(1.12)	.93	(.197)	.452	.41	.058	1.49	3	WDXRF [7s]
Na	.033	(.11)	(.10)	.033	.079	(.135)	.0348	(.50)	4	WDXRF [1s,4s,7s]
K	.28	(.030)	(.20)	.111	(.031)	(.025)	.49	(.034)	4	WDXRF [2s,5s,6s,8s] ICAP-AES [6f,8f]
Ti	.078	.053	.070	.055	.039	.023	(.25)	.0192	3	WDXRF [3s,6s,8s]
Р	(.007)	.024	(.0059)	(.010)	.00222	(.00095)	(.0083)	.012	3	WDXRF [1–3s,5–8s] ICAP-AES* [3s,6s,7s]
Mn	.0043	.0020	.0081	.0019	.0017	.000416	.00147	.0081	3	WDXRF [1–8n]

determined by ICAP-AES in Reston, showed trends similar to those of barium except to a lesser extent. Strontium had one outlier ("poor"), three "good," and four "usable" values.

All techniques for cadmium, lanthanum, samarium, and uranium showed only overall "usable agreement" even though some samples showed "good agreement" for each of these elements. In these cases it is difficult to determine which technique may be in error. For cadmium, samarium, and uranium, there are only two HP techniques. For years INAA has been considered an excellent technique for the rare earth elements and may provide the best data for lanthanum and samarium. EDXRF is generally not considered to be the best technique for rare earth elements such as lanthanum because of spectral overlaps and values generally near the detection limits. ICAP-MS should produce good results for the rare earth elements but does not always agree well with INAA. This disagreement may be due to incomplete dissolution of some rare-earth-bearing species such as zircon.

Not surprisingly, EDXRF values for cerium yielded only "usable agreement" because cerium concentrations were very near the detection limits for this technique. Other techniques for cerium showed "good agreement." However, chromium determined by EDXRF, well above the detection limit, also showed only "usable agreement," whereas all other techniques showed "good agreement."

The overall agreement of niobium determined by EDXRF and neodymium determined by INAA was usable only because several samples had values at or below the detection limits. Ge, Ga, Mo, Ag, Sn, Pr, and Bi all had only one HP technique, so no rating could be determined. Beryllium and vanadium were determined only by ICAP-AES in both Reston and Denver. The "good agreement" that was expected for these techniques is printed in italics in table 4 because these techniques are modifications of the same technique.

The causes of all of the discrepancies are not known. Overall, however, the data are generally useful and provide an excellent base for further study.

DETERMINATION OF RECOMMENDED VALUES

Because some elements were determined by only one high-precision technique, because some element concentrations approached their detection limits, and because some samples contained interfering elements, recommended values cannot be reliably calculated for all elements in all coals. Tables 5 and 6 present recommended and average values for concentrations determined by high-precision techniques.

Recommended values were determined by using procedures similar to those used in determining agreement ratings. The mean value was considered a recommended value if the relative standard deviation of all individual determinations of HP techniques excluding outliers was less than 5 percent for major elements or 10 percent for trace elements using the criteria discussed earlier, and there were at least four individual determinations. If a recommended value could not be determined, the value for the statistical parameter responsible for rejection was boxed in appendix 2, and an average of all HP techniques was reported in parentheses **Table 6.** Recommended or average values for trace elements in parts per million, determined only by high-precision (HP) techniques using the criteria described in text.

[Parentheses indicate that the value is only an average of concentrations determined by the HP techniques and is not a recommended value. Complete data set is given in appendix 1. Statistical analysis of all values, including errors and criteria for not recommending values, is given in appendix 2. "Excluded values" column indicates that the specified HP techniques were not used in determining the given value. The bracketed numbers with the letters indicate the samples for which the techniques were excluded and the reason the values were rejected. The following abbreviations were used under "Excluded values": 1 = UF PC-1; 2 = WY PC-2; 3 = IL PC-3; 4 = PITT PC-4; 5 = POC PC-5; 6 = UT PC-6; 7 = WV PC-7; 8 = ND PC-8; o = determined value was an <u>o</u>utlier; s = only one <u>significant figure in all determined values;</u> f = values rejected because of <u>forty percent rule</u> (see text); and $n = \underline{n}o$ values were determined or all reported values were upper limits. All techniques referred to under "Excluded values" were determined on the ash except INAA, which was determined on the whole coal (* indicates technique was done in Denver). —, no data]

	UF PC-1	WY PC-2	IL PC-3	PITT PC-4	POC PC-5	UT PC-6	WV PC-7	ND PC-8	No. HP techniques	Excluded values
Li Be	14.8 1.5	4.0 (.25)	7.8 (.76)	8.6 (.77)	5.6 .80	5.3 (.125)	28 2.0	2.69 (.175)	$\frac{2}{2}$	ICAP-AES* [2–4,6,
B Sc V	(38) 3.99 26.5	(96) 1.66 14	(160) 2.65 35	(47.8) 2.53 16	(13.1) 1.78 11.5	(127) .78 4.3	(56) 7.5 43.3	(79) (.80) 3.7	$\begin{array}{c}1\\2\\2\end{array}$	8s] ICAP-AES* [8s]
Cr Co	(22) 5.2	(6.5) 1.65	(36) 4.4	(16) 2.6	(10) 3.9	5.2 .93	40 7.8	2.4 (.78)	4 3	EDXRF [6n] FAAS [8n], ICAP-AES* [8s}
Ni	(15)	(4.7)	(19)	9.2	(7.6)	3.4	16	(1.5)	4	EDXRF [6n]; INAA [8n]
Cu Zn	18.9 20	13.0 11.0	10.1 (200)	5.8 8.3	(14) (5)	4.1 (6)	(21) 12	(4.2) (5.2)	3 4	EDXRF [6n] EDXRF [3f,6n]
Ga Ge As Se Rb	(5.34) (4.35) 16.7 (1.9) 21	(1.9) (.36) (3.0) (2.0) (2.7)	(3.57) (8.4) (4.2) (4.2) 15.5	(3.3) (1.2) 8.0 1.6 8.3	(1.8) (.29) 10.1 (2.7) 2.2	(1.00) (.23) (.45) (1.2) 1.02	(9.4) (1.67) (6.4) 5.6 (36)	(1.16) (.37) (2.4) .59 (1.2)	1 1 2 2 3	INAA [2n,5n]; EDXRF [6n,8n]
Sr	(57)	257	(32)	64	(100)	64	(60)	(560)	4	ICAP-AES [2f]; EDXRF [6n]
Y Zr Nb Mo	(9) 26 2.5 (2.5)	3.7 (21) (1.2) (.52)	(4.1) (23) (2.2) (6.3)	(4.4) 19 1.8 (.74)	6.3 16.5 (1.0) (2.7)	2.0 (17.3) (.57) (.42)	11.7 75 (7) (1.27)	(2.1) (9) (.79) (.41)	3 2 2 1	EDXRF [6n,8n,7f] EDXRF [6n] EDXRF [6n,8n]
Ag Cd Sn Sb Cs	(.5) (.12) (.95) .51 (1.7)	(.24) (.11) (.32) (.18) .195	(.6) (.7) (.71) .84 (1.1)	(.3) (.08) (.552) (.22) .75	(.23) (.08) (.371) (.4) .24	(.15) (.07) (.17) (.10) .14	(1.2) (.07) (1.75) $.54 (2.1)$	(.15) (.043) (.4) .15 (.09)	1 2 1 2 2	
Ba	(56)	300	(83)	(40)	200	35	(120)	(500)	4	ICAP-AES [2f,5f]; EDXRF [6n]
La	(9)	(4.7)	6.0	(5.1)	(6.2)	(2.9)	(19)	(2.5)	3	EDXRF [2n,3n,5f, 6n,8n]
Ce Pr Nd	17 (1.9) 7.7	(8) (.93) (3.4)	$(11) \\ (1.4) \\ (4.9)$	$(10) \\ (1.12) \\ (4.3)$	(11) (1.31) (5.0)	4.7 (.55) (2.1)	(34) (4.1) (14)	4.3 (.43) (1.8)	3 1 2	EDXRF [6n,8n] INAA [2–5n]
Sm Eu Tb Yb Hf	(1.8) .37 .248 .85 (.8)	(.8) (.18) .11 (.40) .64	(1.1) .21 .14 .52 (.7)	$(1.0) \\ (.19) \\ .12 \\ .450 \\ (.56)$	(1.1) .22 .165 .56 (.50)	(.44) .092 .056 .20 .52	(3.2) .64 .398 1.6 (2.1)	(.33) (.07) (.055) (.26) (.38)	2 2 2 2 2 2	
Ta W Pb Bi	.21 (1.2) 7.5 (.117)	$(.12) \\ (.42) \\ (2.3) \\ (.053) \\ (1.6)$.18 (1.73) (7) (.088)		.12 (1.0) (2.1) (.051)	(.052) .45 1.52 (.035)	.63 (1.7) 12.1 (.23)	(.093) (.36) (1.5) (1.1)	2 2 2 1	ICAP-MS* [8n]
Th U	(2.5) (1.0)	(1.6) .58	2.1 (5)	1.5 (.5)	1.16 (.57)	.60 .79	6.4 (1.9)	(1.1) .46	2 2	ICAP-AES [8s]

in tables 5 and 6. Values excluded from the determination of recommended values and the reason for exclusion are given in the final column of tables 5 and 6. After the analyses were completed, 43 percent of the values reported in tables 5 and 6 were recommended values.

CONCLUSIONS

A multitechnique approach is the best method to differentiate "good" values from "poor" values. Differences are caused by spectral interferences, volatilization due to ashing, or incomplete sample dissolution. Interferences for a given element usually differ for each technique. Losses caused by volatilization can be determined by comparing data from whole-coal procedures and ash procedures. Insolubility problems can be identified by comparing data from techniques not requiring dissolution with data from techniques requiring dissolution.

This paper does not recommend values for all elements, but it does provide reliable data for many trace elements. It provides manipulations of the data that will allow readers to make their own interpretations and judgments. It also demonstrates that some techniques are more reliable than others for individual elements, and they depend on the concentration of an element. This paper shows that the more high-precision tests that can be run on a sample to measure certain elements, the greater the reliability the data and the greater the likelihood of determining a recommended value. More important, it points out the uncertainties in attempting to obtain reliable data from a single technique for coals of widely differing types, it provides a basis for determining some uncertainties of the techniques, and it should aid in the evaluation of data determined by different techniques.

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Techniques.—The techniques include inductively coupled argon plasma-atomic emission spectrometry (ICAP-AES) used in both Denver (D) and Reston (R) labs, inductively coupled argon plasma-mass spectrometry (ICAP-MS), flame (FAAS) and graphite furnace (GFAAS) atomic absorption spectrometry, hydride-generation atomic absorption spectrometry (HGAAS), direct-current arc atomic emission spectrography (DCAES), wavelength-dispersive (WDXRF) and energy-dispersive (EDXRF) X-ray fluorescence spectrometry, and instrumental neutron activation analysis (INAA).

Material analyzed.—Whole coal (WC) and ash were analyzed. Concentrations originally reported on an ash basis were converted to a whole-coal basis, and concentrations originally reported as an oxide were converted to an elemental basis for ease of comparison. The ash yield was also reported.

Location.—Materials were analyzed in U.S. Geological Survey laboratories in Denver (D), Colo., and Reston (R), Va.

Concentrations.—The concentrations are reported as weight percent and parts per million (ppm), which is equivalent to micrograms per gram.

Type styles and parentheses.—Average values in parentheses indicate that only one value was reported. Techniques in **bold** type are high-precision techniques; concentrations (except outliers) determined by these techniques are also in **bold** type. Techniques in regular type are less precise than those in bold; concentrations determined by the low-precision techniques are also in regular type.

Outliers.—Concentrations in braces are considered outliers for the statistical analysis used to determine the recommended values or the best average (see table 6 and appendix 2). Single outliers were determined by the Grubbs test (J.K. Taylor, 1987, Quality assurance of chemical measurements, Chelsea, Mich., Lewis Publishers, Inc., 328 p.). Multiple outliers for a given sample technique were determined by the 40 percent rule (see paper by Palmer and Klizas, this volume). Values with one significant figure were treated as outliers.

[Appendix 2 begins on p. 96]

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

	Ash (%)		Si (weight	t percent)			Al	(weight perce	nt)	
Technique Material analyzed	550°C WC	ICAP-AES Ash	WDXRF Ash	DCAES Ash	DCAES WC	Ash	ICAP-AES Ash	Ash	DCAES Ash	DCAES WC
Location	R	D	R	R	R	R	R	D	R	R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	13.5 13.5 13.5 13.5	2.5 2.6 2.7 2.6	2.64 2.69 2.67	3.0 3.0 3.4 3.1	3 4 3 3	1.54 1.54 1.54	1.61 1.61 1.59 1.60	1.5 1.5 1.6 1.6	2.0 2.6 2.6 2.4	2 2 2 2
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	8.5 8.5 8.5 8.5	.99 1.0 1.1 1.0	1.23 1.25 1.24	1.3 1.3 1.5 1.4	.80 1.00 .52 .77	.675 .670 .672	.695 .698 .705 .699	.63 .63 .72 .66	.42 .54 .61 .53	.47 .54 .44 .48
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	16.2 16.2 16.2 16.2	3.1 3.0 {2.7} 2.9	3.04 3.04 3.04	3.6 3.2 4.0 3.6	3 4 3 3	1.21 1.20 1.21	1.26 1.25 1.25 1.25	{1.4} 1.3 1.2 1.3	1.9 1.9 2.1 2.0	2 2 2 2
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	9.2 9.2 9.2 9.2	1.8 1.8 1.8 1.8	1.95 1.97 - 1.96	2.4 2.7 2.9 2.7	2 3 2 2	.988 .974 .981	.994 1.00 .994 .997	.97 .97 {.92} .96	2.2 1.9 2.2 2.1	1 2 2 2
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	5.3 5.3 5.3 5.3	0.77 0.75 0.72 0.75	.810 .798 — .804	.95 .90 1.00 .95	.83 1.00 .80 .88	.550 .547 548	.551 .551 .546 .549	{.59} .56 .56 .57	.64 .74 .74 .71	.80 .92 .84 .85
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	4.6 (4.6)	.88 .92 .94 .91	.893 - (.893)	.74 .92 .92 .86	 	.370 	.348 .345 .347	.36 .39 .39 .38	.31 .32 .29 .32	- - -
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	19.4 (19.4)	4.8 5.3 5.0 5.0	4.90 — (4.90)	4.3 5.8 5.4 5.2	 	3.17 (3.17)	3.12 3.18 3.16 3.15	3.1 3.2 3.2 3.2	3.3 4.3 4.1 3.9	- - -
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	9.5 (9.5)	.67 .67 .67 .67	{.768} (0.768)	.54 .63 .63 .60		{.45} (.45)	.396 .400 .404 .400	.40 .40 .40 .40	.21 .28 .21 .23	

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

			Fe (weight	t percent)				Mg (weight percent	t)	
Technique Material analyzed Location	INAA WC R	WDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	DCAES Ash R	DCAES WC R	WDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	DCAES Ash R	DCAES WC R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	1.76 1.81 1.79 1.78	1.98 1.88 1.93	1.86 1.89 1.84 1.86	1.6 1.7 1.7 1.7	1.6 1.3 1.7 1.6	2 3 2 2	0.079 .079 .079	0.082 .082 .081 .082	0.075 .078 .080 .078	0.13 .15 .16 .15	0.097 .100 .100 .099
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	.381 .379 .339 .366	.33 .32 	.358 .379 .363 .367	.31 .35 .35 .34	.26 .26 .31 .27	.34 .33 .20 .29	.25 .27 26	.241 .236 .239	.20 .20 .23 .21	.33 .39 .37 .37	.20 .20 .20 .20
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	2.66 2.67 2.68 2.67	2.71 2.70 2.70	2.72 2.66 2.79 2.72	2.4 2.5 2.3 2.4	2.3 2.3 2.6 2.4	2 2 3 2	{.09} {.09} 	.076 .078 .076 .077	.078 .075 .068 .073	.11 .11 .12 .12	.096 .099 .100 .098
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	1.36 1.34 1.33 1.35	1.45 1.43 1.44	1.34 1.32 1.31 1.32	1.2 1.2 1.3 1.2	1.7 1.8 2.0 1.9	1 1 2 1	{.04} {.03} .03	.038 .038 .038 .038	.033 .033 .033 .033	.065 .064 .089 .073	.045 .048 .052 .048
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	.504 .521 .502 .509	.519 .522 .521	.513 .505 .504 .507	.52 .48 .48 .50	.52 .44 .53 .50	.60 .80 .68 .69	.067 .067 — .067	.0562 .0551 .551 .0548	.0575 .0543 .0543 .0554	.10 .11 .11 .11	.079 .074 .079 .077
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	.327 .310 .315 .317	.29 (.29)	.284 .281 282	.29 .29 .31 .30	.21 .27 .23 .24	 	.048 (.048)	.030 .030 030	.028 .031 .031 .030	.043 .038 .041 .041	
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	.404 .384 .408 .399	.37 	.396 .381 .379 .385	.41 .39 .38 .39	.35 .45 .41 .40	 	{.08} (.08)	.092 .095 .096 .094	.082 .090 .084 .085	.128 .084 .078 .097	
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	.553 .529 .560 .547	.45 (.45)	.466 .470 .480 .472	.45 .47 .46 .46	.35 .43 .33 .37	 	.42 - (.42)	.395 .395 .405 .398	.34 .355 .349 .351	.52 .51 .51 .51	

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

		Ca	(weight perce	ent)			Na	(weight percer	nt)	
Technique Material analyzed Location	WDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	DCAES Ash R	DCAES WC R	INAA WC R	ICAP-AES Ash R	ICAP-AES Ash D	WDXRF Ash R	DCAES Ash R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	0.40 .40 .40	0.447 .445 .443 .445	0.40 .40 .40 .40	0.69 .59 .63 .64	0.57 .54 .63 .58	0.0336 .0343 .0342 .0341	0.032 .032 .032 .032	0.034 .034 .034 .034	{0.03} {.03} 	0.042 .047 .046 .045
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	1.10 1.15 1.12	1.21 1.16 1.22 1.19	1.0 .97 1.1 1.0	1.0 1.3 1.4 1.2	1 1 1 1	.114 .117 .113 .115	.119 .119 .119 .119	.12 .12 .12 .12	.10 .083 092	.12 .16 .12 .13
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	.90 .89 — .90	.943 .965 .972 .960	.96 .94 .87 .92	1.6 1.5 1.8 1.6	2 2 1 2	.103 .101 .103 .102	.097 .099 .094 .097	.11 .11 .10 .11	.13 {.09} 	.16 .14 .15 .15
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	.20 .20 .20 .20	.211 .205 .203 .206	.18 .18 .19 .19	.46 .33 .43 .41	.24 .34 .28 .29	.0343 .0342 .0344 .0343	.032 .032 .031 .032	.034 .034 .033 .034	{.04} {.03} - .03	.039 .033 .037 .036
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	.443 .443 .443	.464 .452 .452 .456	.45 .45 .45 .45	.53 .58 .64 .58	.69 .93 .79 .80	.0797 .0773 .0778 .0782	.0832 .0832 .0816 .0826	.0826 .0826 .0826 .0826	.073 .069 .071	.13 .13 .13 .13
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	.401 	.418 .401 .409	{.35} .43 .43 .43	.28 .29 .29 .29	 	.148 .147 .143 .146	.127 .124 126	.13 .13 .13 .13	.13 (.13)	.087 .064 .064 .072
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	{.06} 	.057 .058 .058 .058	.056 .061 .058 .058	.078 .060 .095 .078	 	{.0460} .0351 .0352 .0388	.033 .035 .036 .034	.0346 .0346 .0346 .0346	{.05} 	.037 .035 .037 .036
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	1.54 	1.52 1.53 1.57 1.54	1.4 1.4 1.4 1.4	1.2 1.3 1.4 1.3	 	.540 .533 .515 .529	.460 .464 .475 .466	.493 .493 .493 .493	.50 (.50)	.25 .25 (.25)

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

		K	(weight perce	ent)			Ti	(weight percen	t)	
- Technique Material analyzed Location	INAA WC R	WDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	DCAES Ash R	WDXRF Ash R	ICAP-AES Ash R	G ICAP-AES Ash D	DCAES Ash R	DCAES WC R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	0.264 .262 .280 .269	0.26 .28 27	0.283 .283 .283 .283	0.28 .30 .29 .29	0.31 .28 .30 .30	.078 .082 080	0.078 .080 .078 .079	0.072 .074 .078 .075	0.061 .039 .059 .053	0.10 .10 .10 .10
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	.0268 .0335 .0272 .0292	{.03} {.03} .03	.028 .028 .026 .028	.027 .035 .038 .033	.031 .037 .031 .033	.051 .052 051	.056 .056 .055 .056	.048 .049 .056 .051	.030 .037 .041 .036	.050 .050 .038 .044
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	.199 .186 .199 .195	.21 .20 20	.194 .194 .194 .194	.23 .22 .22 .22	.24 .26 .28 .26	{.07} {.07} 	.070 .070 .070 .070	.072 .071 .064 .069	.062 .065 .073 .066	.089 .100 .086 .092
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	.109 .112 .109 .110	.11 .11 - .11	.110 .110 .110 .110 .110	.11 .11 .11 .11	.13 .12 .14 .13	.059 .060 .059	.057 .056 .055 .056	.051 .051 .049 .050	.059 .052 .059 .057	.086 .090 .097 .091
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	.0284 .026 .033 .029	{.03} {.03} .03	.027 .028 .028 .028	.036 .036 .032 .035	.034 .033 .039 .036	.040 .039 040	.038 .038 .037 .038	.044 .038 .038 .040	.028 .025 .031 .028	.084 .084 .078 .082
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	.018 <.025 .0252 .022	{.02} 	{.014} {.014} 	.027 .027 .029 .028	.022 .014 .017 .017	{.02} 	.023 0.023 .023	.023 .023 .023 .023	.018 .020 .018 .019	
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	.517 .497 .500 .505	.46 (.46)	.489 .501 .495 .495	.46 .53 .48 .49	.39 .33 .35 .36	.24 	.260 .260 .260 .260	.22 .24 .23 .23	.18 .18 .17 .18	
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	.028 <.040 .030 .029	{.03} 	{.012} {.016} {.015} {.014}	.044 .032 .036 .037	.038 .033 .039 .037	{.02} 	.020 .019 .020 .020	.018 .018 .020 .019	.017 .017 .019 .018	

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

		Р (weight percer	nt)			Mn	(weight percer	nt)	
Technique Material analyzed Location	WDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	WDXRF WC R	DCAES Ash R	WDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	DCAES Ash R	DCAES WC R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	{0.005} {.006} .006	0.0076 .0080 .0082 .0079	0.0059 .0059 .0059 .0059	<0.005 <.005 <.005 —	<0.0092 <.0092 <.0092 —	<0.01 <.01 	0.0041 .0042 .0041 .0041	0.0044 .0044 .0043 .0044	0.0055 .0051 .0058 .0055	0.0053 .0059 .0048 .0053
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	{.03} {.03} .03	.026 .024 .026 .025	.021 .021 .025 .022	.03 .02 .03 .03	.020 .026 .033 .026	<.007 <.007 	.0020 .0021 .0019 .0020	.0020 .0020 .0019 .0020	.0042 .0027 .0031 .0033	.0010 .0016 .0014 .0013
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	{.004} {.004} 004	.0058 .0060 .0060 .0059	{.004} {.004} {.003} .004	<.005 <.005 <.005 —	<.011 <.011 <.011	<.01 <.01 	.0076 .0076 .0076 .0076	.0092 .0084 .0084 .0087	.011 .010 .011 .011	.015 .015 .017 .016
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	{.01} {.01} 01	.011 .011 .011 .011	.0092 .0088 .0060 .0080	.009 .009 .009 .009	<.0063 <.0063 <.0063 —	<.007 <.007 	.0018 .0017 .0017 .0018	.0021 .0021 .0020 .0021	.0030 .0038 .0039 .0036	.0016 .0028 .0018 .0020
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	{.002} {.002} .002	.0022 .0022 .0021 .0022	.0023 .0023 {.002} .0023	<.005 <.005 <.005	<.0036 <.0036 <.0036	<.004 <.004 	.0016 .0016 .0016 .0016	.0018 .0018 .0019 .0018	.0024 .0023 .0026 .0024	.0016 .0024 .0017 .0019
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	{.0001} (.001)	.00092 .00097 00094	{.0008} {.0008} {.0006} .0007	<.005 <.005 <.005	<.0031 <.0031 .00074 (.00074)	<.004 	.00041 .00041 .00041	.00043 .00042 .00041 .00042	.00064 .00046 .00074 .00061	
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	{.007} (.007)	.0089 .0087 .0074 .0084	{.008} {.008} {.008} {.008}	<.005 <.005 <.005	.017 <.013 <.013 (.017)	<.015 — — —	.0016 .0015 .0015 .0015	.0014 .0014 .0014 .0014	.0018 .0023 .0019 .0020	
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	{.02} (.02)	.012 .012 .012 .012	.012 .012 .012 .012	.02 .02 .02 .02	.012 .010 .0093 .011	{.007} 	.0080 .0079 .0081 .0080	.0081 .0083 .0083 .0082	.010 .010 .010 .010	

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

	Li	(ppm)		Be (ppm)		B (pp	m)
- Technique Material analyzed	FAAS Ash	ICAP-AES Ash	ICAP-AES Ash	ICAP-AES Ash	DCAES Ash	ICAP-AES Ash	DCAES Ash
Location	R	D	R	D	R	D	R
UF PC-1-1	15	15	1.5	1.5	1.5	35	19
UF PC-1-2	15	15	1.5	1.5	1.2	31	47
UF PC-1-3	15	15	1.5	1.5	1.5	49	24
AVG PC-1	15	15	1.5	1.5	1.4	38	30
WY PC-2-1	4.1	3.9	.26	{.3}	.14	85	74
WY PC-2-2	4.2	3.9	.24	{.3}	.14	110	63
WY PC-2-3	4.1	3.9	.25	. 3}	.15	94	73
AVG PC-2	4.1	3.9	.25	.3	.14	96	70
IL PC-3-1	7.8	7.9	.78	{.6}	.89	160	120
IL PC-3-2	7.9	7.6	.75	<u>{</u> .6}	.96	160	110
IL PC-3-3	7.8	7.5	.75	{.6}	1.1	160	130
AVG PC-3	7.8	7.7	.76	.6	.97	160	120
PITT PC-4-1	8.7	9.2	.76	{.7}	1.1	48	49
PITT PC-4-2	8.6	8.3	.76	{.7}	1.2	49	53
PITT PC-4-3	8.6	8.2	.78	{.7}	1.2	47	52
AVG PC-4	8.6	8.9	.77	.7	1.2	48	51
POC PC-5-1	5.8	5.3	.80	.80	.74	14	17
POC PC-5-2	5.8	5.3	.80	.80	.58	13	16
POC PC-5-3	5.8	5.8	.80	.80	.69	13	19
AVG PC-5	5.8	5.5	.80	.80	.67	13	17
UT PC-6-1	5.1	5.5	.13	{.1}	.15	120	<46
UT PC-6-2	5.1	5.5	.12	{.1}	.24	120	<46
UT PC-6-3	—	5.5	_	{.1}	.20	130	<46
AVG PC-6	5.1	5.5	.13	.1	.20	130	_
WV PC-7-1	29	27	1.9	2.1	2.9	39	29
WV PC-7-2	29	27.9	1.9	2.1	3.1	78	93
WV PC-7-3	29	27	1.9	2.1	3.1	51	49
AVG PC-7	29	27	1.9	2.1	3.0	56	57
ND PC-8-1	2.7	2.7	.18	{.2}	.27	79	48
ND PC-8-2	2.7	2.7	.17	{.2}	.33	81	76
ND PC-8-3	2.8	2.7	_	{.2}	.23	79	77
AVG PC-8	2.7	2.7	.18	.2	.28	79	67

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

		Sc (ppm)			V (p	pm)	
- Technique	INAA	ICAP-AES	DCAES	ICAP-AES	ICAP-AES	DCAES	DCAES
Material analyzed	WC	Ash	Ash	Ash	Ash	Ash	WC
Location	R	D	R	R	D	R	R
UF PC-1-1	4.04	3.9	2.6	26	27	18	38
UF PC-1-2	4.01	3.9	1.5	26	27	13	45
UF PC-1-3	4.14	3.9	2.2	26	27	17	41
AVG PC-1	4.06	3.9	2.1	26	27	16	41
WY PC-2-1	1.67	1.6	1.0	14	14	7.3	10
WY PC-2-2	1.69	1.6	1.4	14	14	9.3	14
WY PC-2-3	1.67	1.7	1.5	14	14	10.0	17
AVG PC-2	1.68	1.6	1.3	14	14	9.0	14
IL PC-3-1	2.55	2.8	2.5	32	39	23	52
IL PC-3-2	2.57	2.8	2.6	32	37	23	59
IL PC-3-3	2.64	2.6	2.8	31	36	26	44
AVG PC-3	2.59	2.7	2.6	32	37	24	52
PITT PC-4-1	2.61	2.5	1.8	15	17	12	20
PITT PC-4-2	2.56	2.5	1.7	15	17	12	25
PITT PC-4-3	2.55	2.5	2.0	15	17	14	27
AVG PC-4	2.57	2.5	1.9	15	17	13	24
POC PC-5-1	1.80	1.8	1.3	11	12	6.9	16
POC PC-5-2	1.79	1.8	1.3	11	12	6.4	21
POC PC-5-3	1.77	1.8	1.5	11	12	7.9	18
AVG PC-5	1.79	1.8	1.3	11	12	7.1	18
UT PC-6-1	.832	.78	.74	4.0	4.6	3.8	—
UT PC-6-2	.801	.74	.69	4.0	4.4	3.6	
UT PC-6-3	.805	.74	.64	_	4.4	3.6	
AVG PC-6	.813	.75	.69	4.0	4.5	3.7	—
WV PC-7-1	7.69	7.4	7.0	43	43	31	_
WV PC-7-2	7.54	7.4	6.0	45	43	35	_
WV PC-7-3	7.61	7.4	6.2	43	43	29	—
AVG PC-7	7.62	7.4	6.4	44	43	32	—
ND PC-8-1	.846	{.8}	.87	3.5	3.9	3.6	—
ND PC-8-2	.828	{.8}	.86	3.4	3.8	3.6	_
ND PC-8-3	.865	{.8}	.95	3.6	3.9	3.5	_
AVG PC-8	.846	.8	.90	3.5	3.9	3.6	

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

				Cr (ppm)						Co (ppm)		
Technique Material analyzed Location	WC	EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES Ash R	DCAES WC R	INAA WC R	FAAS Ash R	ICAP-AES Ash D	DCAES Ash R	DCAES WC R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	20.3 20.4	25.1 26.5 26.5 26.0	20 19 20 20	22 22 22 22 22	15 23 19 19	22 17 19 19	32 38 26 32	5.27 5.41 5.31 5.33	5.1 5.1 4.7 5.0	5.3 5.3 5.4 5.3	3.8 3.0 3.9 3.5	5 5 5 5
WY PC-2-1 WY PC-2-2. WY PC-2-3 AVG PC-2	6.22 6.23	7.3 8.2 {9.18} 8.2	6.2 6.4 6.0 6.2	6.3 6.5 6.4 6.4	3 5 - 4	4.8 6.3 5.9 5.7	8 6 8 7	1.66 1.73 1.65 1.68	1.6 1.6 1.6 1.6	1.5 1.8 1.7 1.7	1.0 1.4 1.4 1.3	<5 <5 <5
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	32.5 33.2	42.9 42.9 43.7 43.2	29 32 31 31	41 34 34 36	29 36 41 35	36 36 42 38	54 57 47 52	4.21 4.21 4.53 4.31	4.2 4.5 4.2 4.3	4.7 4.4 4.2 4.4	3.6 3.4 4.2 3.7	<5 <5 <5
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	15.0 14.5	18.4 17.5 17.9 17.9	15 14 13 14	16 16 16 16	6 10 9 8	20 17 20 19	23 27 21 24	2.65 2.61 2.60 2.62	2.4 2.6 2.4 2.4	2.6 2.7 2.6 2.6	2.8 2.3 2.8 2.6	<5 <5 <5
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	9.06 8.85	12.5 11.9 12.5 12.3	9.5 8.5 9.5 9.2	10.1 9.5 9.5 9.7	5 7 4 5	9.0 8.0 9.5 8.8	17 17 14 16	4.09 4.13 3.98 4.07	3.8 3.8 3.8 3.8	4.0 3.9 4.0 4.0	2.4 2.3 3.0 2.6	5 5 5 5
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	5.03 5.62	 	5.1 4.6 - 4.8	5.5 5.1 5.1 5.2	<10 <10 <10	5.1 3.9 5.5 4.8		1.01 .992 1.00 1.00	.83 .83 83	.97 .92 .92 .94	.69 .60 .64 .64	
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	35.6 35.7	46.2 45.4 46.0 45.9	41 41 39 40	39 39 39 39 39	50 48 48 49	43. 35 41 39		7.65 7.86 7.72 7.74	7.8 8.4 8.2 8.1	7.4 7.6 7.4 7.5	7.0 5.8 6.0 6.3	
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	2.16 2.33	2.5 2.7 2.8 2.6	2.6 2.5 2.3 2.4	2.2 2.2 2.5 2.3	<10 <10 <10	2.4 2.4 2.5 2.4		.771 .761 .802 .778	<1 <1 <1	$\{.8\}$ $\{1\}$ $\{.8\}$.9	.54 .57 .54 .55	

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

				Ni (ppm)						Cu (ppm	n))		
Technique Material analyzed Location	WC	EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES Ash R	DCAES WC R	EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES Ash R	DCAES WC R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	13.2 14.3	14.0 15.9 11 14	15 14 15 14	15 15 15 17	27 22 21 23	14 13 14 14	22 29 24 25	19.7 20.5 21.3 20.4	19 19 19 19	19 19 18 19	20 18 17 18	17 16 19 17	20 18 16 18
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	5.4 5.0	4.3 3.9 3.7 3.9	4.9 4.7 5.2 4.9	4.7 4.9 4.7 4.8	9 7 8 8	4.8 5.4 5.4 5.2	<10 <10 <10	12.1 13.4 13.1 12.9	12 12 13 12	13 14 14 14	18 17 17 17	11 14 14 13	5.0 10 6.0 7
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	17.8 21.0	14 20 19 18	18 19 18 18	{29} 19 19 19 19	33 30 33 32	21 19 24 22	28 35 37 33	11 9.4 8.7 9.6	10 10 10 10	11 11 10 11	15 13 15 14	13 11 9.4 11	9.0 10 9.0 9
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	11.3 <12	9.0 8.8 10.3 9.3	9.1 8.3 8.1 8.5	8.9 9.2 9.2 9.1	17 11 15 14	14 12 13 13	13 14 11 12	6.4 6.1 6.2	5.5 5.3 5.5 5.5	5.4 6.2 5.7 5.8	10 9 10 10	5.3 8.2 6.1 6.5	5.0 6.0 7.0 6
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	8.5 8.2	7.00 7.74 8.0 7.58	6.9 6.4 6.9 6.7	7.4 7.4 7.4 7.4	11 9 12 11	6.4 5.8 7.4 6.5	11 11 <10 11	12.7 13.0 11.9 12.5	12 12 12 12	16 16 16 16	16 16 19 17	22 15 20 19	18 13 27 19
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	3.2 3.7	 	3.4 3.3 - 3.3	3.6 3.4 3.5 3.5	5 5 5 5	3.9 4.1 4.0 4.0	 	 	3.9 3.8 - 3.8	4.3 4.2 4.1 4.2	8 8 8 8	3.2 2.4 3.0 2.9	
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	18 14.3	18 16 17 17	16 16 16 16	16 16 16 16	17 18 19 18	23 19 21 21	 	25.2 22.3 23.3 23.6	19 19 19 19	21 21 21 21	30 30 32 31	17 16 13 16	
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	<3 <3	2.4 1.6 1.8 1.9	1.4 1.4 1.1 1.3	1.3 1.5 1.4 1.4	$\begin{array}{c} 4\\10\\4\\6\end{array}$	1.6 1.6 2.1 1.8	 	3.4 3.3 3.2 3.3	5.6 3.5 4.9 4.7	4.5 4.8 {12} 4.6	8 10 10 9	3.9 3.8 7.1 4.9	

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

			Zn	(ppm)				Ga (ppm)			Ge (ppm)	
Technique Material analyzed Location	WC	EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES WC R	ICAP-MS Ash D	DCAES Ash R	DCAES WC R	ICAP-MS Ash D	DCAES Ash R	DCAES WC R
UF PC-1-1		21.9	19	19	35	30	5.45	6.1	11	4.39	3.5	4
UF PC-1-2		19.2	19	19	32	30	5.21	6.3	11	4.29	3.4	4
UF PC-1-3		20.3	20	19	31	40	5.35	6.3	8	4.37	3.9	3
AVG PC-1	19.7	20.5	19	19	33	33	5.34	6.2	10	4.35	3.6	4
WY PC-2-1	11.0	9.7	10	13	26	<20	1.86	2.2	4	.36	<.4	<2
WY PC-2-2	11.6	11.1	10	11	29	<20	1.82	2.6	2	.36	<.4	<2
WY PC-2-3	11.2	11.1	10	12	21	<20	2.04	2.5	3	.36	<.4	<2
AVG PC-2	11.3	10.6	10	12	25	_	1.91	2.4	3	.36	_	_
IL PC-3-1	218	{94}	160	180	137	250	3.45	4.7	8	8.10	8.4	8
IL PC-3-2		$\{112\}$	180	210	105	260	3.64	4.4	5	8.62	8.4	7
IL PC-3-3		$\{115\}$	190	190	186	320	3.63	5.0	7	8.49	9.7	12
AVG PC-3		107	180	190	140	280	3.57	4.7	, 7	8.40	8.9	9
PITT PC-4-1	8.9	8.5	8.2	{14}	17	20	3.17	5.2	6	1.13	1.2	3
PITT PC-4-2		8.5	7.4	8.1	14	30	3.24	4.0	5	1.22	1.2	<2
PITT PC-4-3		8.1	7.7	8.1	15	30	3.44	5.0	5	1.39	1.1	<2
AVG PC-4		8.4	7.8	8.1	15	27	3.28	4.7	5	1.25	1.2	(3)
POC PC-5-1	5.74	4.3	3.9	5.3	14	<20	1.70	2.2	4	.23	.46	<2
POC PC-5-2			4.1	5.3	14	<20	1.80	2.1	3	.27	.43	<2
POC PC-5-3		4.7	3.9	5.3	15	<20	1.89	2.5	4	.36	.49	<2
AVG PC-5		4.8	4.0	5.3	14	-	1.80	2.3	4	.29	.46	
UT PC-6-1	4.80) _	6.4	4.6	<2	_	.948	1.5	_	.19	<.2	_
UT PC-6-2			6.0	5.1	<2	_	.989	.9	_	.24	<.2	_
UT PC-6-3			-	4.6	<2	_	1.05	1.3	_	.25	<.2	_
AVG PC-6			6.2	4.8		_	1.00	1.2	_	.23	<u>2</u>	_
						_			_			_
WV PC-7-1		12	13	11	9	_	8.88	13	—	1.6	1.1	_
WV PC-7-2		11	13	11	10	—	9.62	11	—	1.7	<.89	—
WV PC-7-3		12	12	11	10	—	9.66	11	—	1.7	1.2	—
AVG PC-7	13.5	12	13	11	10	_	9.39	12	_	1.7	1.1	—
ND PC-8-1			5.1	5.1	5	_	1.21	1.8	_	.32	<.44	_
ND PC-8-2		5.1	4.7	4.7	4	—	1.14	1.5	—	.37	<.44	_
ND PC-8-3			4.5	4.7	6	_	1.14	1.9	_	.41	<.44	_
AVG PC-8	5.69	5.4	4.6	4.8	5	_	1.16	1.7	_	.37	_	_

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

		As (ppm)		Se (ppm)		Rb (pr	om)	
Technique Material analyzed Location	WC	ICAP-MS Ash D	DCAES Ash R	INAA WC R	HGAAS WC D	INAA WC R	ICAP-MS Ash D	EDXRF Ash R	EDXRF WC R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	17.6 17.1	16.4 15.9 16.1 16.1	31 38 22 30	2.09 1.88 1.8 1.92	2.3 2.0 1.5 1.9	19.4 19.0 20.0 19.5	22.9 21.6 21.6 22.0	21.6 20.8 21.3 21.2	23 21 20 21
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	3.23 3.75	2.8 2.6 1.9 2.4	<9 <9 <9 —	1.57 1.49 1.63 1.56	1.8 2.8 2.5 2.4	<5 <5 <5	1.80 1.80 1.80 1.80	3.4 3.4 3.7 3.5	14 14 11 13
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	4.65 4.90	3.6 3.7 4.0 3.8	<16 <16 <16 —	3.78 4.03 5.07 4.29	4.1 4.0 4.3 4.1	15.5 17.2 14.6 16	15.6 16.0 15.6 15.7	15 15 15 15	22 17 21 20
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	8.30 8.5	7.1 7.6 8.2 7.6	19 17 15 17	1.49 1.60 1.34 1.5	1.8 1.6 1.6 1.7	9.0 7.2 7.1 7.7	8.00 8.06 8.30 8.12	9.38 8.6 9.20 9.1	8 8 8
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	10.6 9.89	9.33 9.81 10.5 9.9	18 11 14 15	2.3 2.5 2.60 2.5	2.9 2.9 3.1 3.0	<6 <6 <3	2.04 2.11 2.18 2.11	2.6 2.1 2.2 2.3	5 4 5 5
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	.50 .44	.33 <.05 <.05 (.33)	<10 <10 <10	1.03 1.16 1.03 1.08	1.2 1.3 1.3 1.3	<2 1.00 .95 .98	1.01 1.06 1.10 1.06	 	3 2 3 3
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	5.29 7.40	6.4 6.6 7.2 6.7	<43 <43 <43 —	5.3 5.54 5.5 5.4	6.1 5.6 {6.9} 5.8	29.8 28.5 31.2 29.7	35.0 36.9 36.9 36.3	42.3 45.0 43.4 43.6	43 43 40 42
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	2.61 2.61	1.6 {.70} <.1 (1.6)	<21 <21 <21 	.51 .61 .63 .58	.60 .60 .60 .60	.86 .99 <1 .93	1.37 1.41 1.46 1.41	 	4 5 3 4

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

			St	r (ppm)					Y (ppm)		
Technique Material analyzed Location		EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES Ash R	EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES Ash R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	56 58 33 49	59 61 61 60	59 61 58 59	58 58 58 58	61 63 61 62	66 57 66 63	10 9.7 11 10	9.3 9.3 9.2 9.3	8.5 8.1 7.6 8.1		6.2 3.0 4.9 4.7
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	263 248 245 252	260 260 260 260	{153} {145} {153} 150	260 260 260 260	292 300 291 294	170 240 260 220	3.4 3.6 3.7 3.6	3.9 3.7 3.7 3.8	3.6 3.6 3.7 3.6	 	2.4 3.1 3.6 3.0
L PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	42 36 40 39	30.5 28.8 30.5 29.9	29 29 29 29	31 32 31 31	33 30 35 33	40 37 44 40	4.9 3.2 4.2 4.1	4.2 4.2 4.2 4.2	4.2 4.2 3.7 4.0	 	4.4 4.7 5.2 4.7
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	58 61 65 61	70 68 65 68	64 64 63 64	64 64 63 64	59 61 69 63	88 83 100 91	5.1 5.1 4.4 4.9	4.3 4.2 4.2 4.3	4.0 3.9 4.0 4.0	 	2.8 3.3 3.0 3.1
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	97 116 101 105	110 110 110 110	85 85 85 85	110 110 110 110	86 77 93 85	110 100 120 110	7.10 6.68 6.89 6.89	6.4 6.4 5.8 6.2	5.8 5.8 5.8 5.8	 	4.6 4.3 4.9 4.6
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	71 68 70 70	 	60 59 60	60 60 60 60	58 65 66 63	83 92 92 89	 	1.8 2.1 - 2.0	2.1 2.0 2.0 2.0	2 <2 <2 (2)	1.9 1.9 1.6 1.8
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	64 59 58 60	76.2 78.0 79.7 78.0	49 47 51 49	62 62 62 62	87 87 83 86	76 62 72 70	{21.4} {19} {22.9} 21.2	11 11 11 11	12 13 12 12	19 20 17 19	9.7 10 7.6 9.1
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	628 580 583 597	640 640 630 640	510 500 500 500	490 490 490 490	800 800 750 780	590 600 690 630	 	1.9 1.8 1.8	2.2 2.3 2.2 2.2	<2 <2 <2 -	2.3 2.6 2.4 2.4

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

			Zr (ppm)				Nb (ppm)		Mo (ppm)	
Technique	ICAP-AES	EDXRF	EDXRF	DCAES	DCAES	ICAP-MS	EDXRF	EDXRF	DCAES	ICAP-MS	DCAES
Material analyzed		Ash	WC	Ash	WC	Ash	Ash	WC	Ash	Ash	Ash
Location	. D	R	R	R	R	D	R	R	R	D	R
UF PC-1-1		27.7	24	13	36	2.6	2.7	_	1.6	3.02	1.6
UF PC-1-2	. 24	26.5	22	7.4	29	2.3	2.2	_	1.3	2.24	1.3
UF PC-1-3	. 25	27.7	24	12	34	2.4	2.7	—	2.0	2.23	1.6
AVG PC-1	. 24	27.3	23	11	33	2.4	2.5	_	1.7	2.50	1.5
WY PC-2-1		22.1	19	12	<20	1.4	1.2	_	1.2	.52	.65
WY PC-2-2		23.4	21	17	<20	1.3	1.0	_	1.3	.49	.66
WY PC-2-3		22.5	17	21	<20	1.5	.85	—	1.8	.54	.70
AVG PC-2	. 19	22.7	19	17	-	1.4	1.0	_	1.4	.52	.67
IL PC-3-1		23.7	24	14	22	1.9	1.9	_	19	7.24	5.5
IL PC-3-2		21.1	22	16	25	2.3	<1.6	—	2.3	5.57	4.9
IL PC-3-3		22.0	24	23	16	2.4	2.3	—	3.2	6.06	5.2
AVG PC-3	. 23	22.3	23	17	21	2.2	2.1	_	2.5	6.23	5.2
PITT PC-4-1	. 18	20.7	15	10	23	1.6	2.0	_	1.7	.71	.86
PITT PC-4-2		21.2	17	15	21	1.7	1.8	_	2.0	.73	.70
PITT PC-4-3		19.8	18	11	28	1.8	1.8	—	1.7	.77	.83
AVG PC-4	. 18	20.5	17	12	24	1.7	1.9	—	1.8	.74	.80
POC PC-5-1	. 17	17.0	10	13	25	1.0	.85	_	1.3	2.61	2.0
POC PC-5-2		16.7	8	11	34	1.1	.85	_	.74	2.65	2.2
POC PC-5-3		17.0	11	12	32	1.3	.85	—	1.2	2.85	2.4
AVG PC-5	. 16	16.9	10	12	30	1.1	.85	—	1.1	2.67	2.2
UT PC-6-1	. 17	_	20	12	_	.51	_	<10	.46	.41	.55
UT PC-6-2		_	17	13	_	.55	_	<10	.51	.41	.39
UT PC-6-3		_	16	9.2	_	.64	_	<10	.55	.45	.51
AVG PC-6	. 17	_	18	11	-	.57	-	_	.51	.42	.48
WV PC-7-1	. 66	80.0	106	47	_	5.6	7.8	13	2.5	1.2	1.3
WV PC-7-2	. 74	79.8	102	29	_	6.0	8.3	<10	3.7	1.3	1.6
WV PC-7-3		81.6	101	27	-	6.6	8.9	13	4.5	1.3	1.2
AVG PC-7	. 69	80.5	103	34	-	6.1	8.3	13	3.6	1.3	1.4
ND PC-8-1	. 11	6.5	20	12	_	.74	<1	<10	.63	.39	.82
ND PC-8-2		7.0	19	12	_	.80	<1	<10	.51	.42	.78
ND PC-8-3		4.5	17	13	_	.84	<1	<10	.61	.44	.83
AVG PC-8	. 12	6.5	18	13	_	.79	_	_	.58	.42	.81

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

	Ag (j	ppm)	Cd (p	pm)	Sn (p	opm)	Sb	(ppm)	Cs	(ppm)
Technique Material analyzed Location	Ash	DCAES Ash R	ICAP-MS Ash D	GFAAS Ash R	ICAP-MS Ash D	DCAES Ash R	INAA WC R	ICAP-MS Ash D	INAA WC R	ICAP-MS Ash D
	D	R	D	ĸ	D			D	ĸ	2
UF PC-1-1	0.36	0.19	0.14	0.072	0.96	1.76	0.558	0.51	1.48	1.89
UF PC-1-2	.40	.20	.22	.069	.97	1.62	.50	.46	1.52	1.89
UF PC-1-3	.62	.22	.14	.073	.95	1.62	.535	.50	1.56	1.89
AVG PC-1	.46	.21	.17	.071	.96	1.67	.531	.49	1.52	1.89
WY PC-2-1	.17	.055	.10	.094	.32	.544	.180	.15	.180	.19
WY PC-2-2		.072	.11	.094	.32	.85	.203	.15	.203	.20
WY PC-2-3		.075	.15	.094	.32	1.2	.191	.18	.191	.19
AVG PC-2		.067	.12	.094	.32	.86	.191	.16	.191	.19
IL PC-3-1	.44	.356	.67	.60	.70	1.6	.848	.73	.848	1.4
IL PC-3-2		.421	.76	.65	.75	2.4	.913	.81	.913	1.4
IL PC-3-3		.324	.94	.57	.70	3.2	.870	.84	.870	1.4
AVG PC-3		.367	.79	.61	.70	2.4	.877	.79	.877	1.4
PITT PC-4-1	.18	.087	.10	.059	.57	1.7	.239	.18	.737	.80
PITT PC-4-2		.086	.092	.058	.57	1.0	.240	.20	.69	.81
PITT PC-4-3		.092	.13	.063	.58	1.2	.221	.21	.67	.80
AVG PC-4	.31	.088	.11	.061	.57	1.3	.233	.20	.70	.80
POC PC-5-1	.14	.095	{.034}	.085	.37	148	.625	.29	.256	.23
POC PC-5-2	.22	.095	.053	.074	.37	29	.458	.29	.283	.23
POC PC-5-3		.106	.090	.074	.38	38	.535	.32	.247	.22
AVG PC-5		.099	.072	.078	.37	72	.539	.27	.262	.22
UT PC-6-1	.097	.55	.064	.060	.17	.40	.118	.087	.123	.14
UT PC-6-2		.41	.078	.060	.17	.28	.098	.087	.148	.14
UT PC-6-3		.44	.087		.16	.43	.106	.092	.119	.14
AVG PC-6		.47	.077	.060	.16	.37	.100	.089	.130	.14
WV PC-7-1	.70	.50	.050	.078	1.7	2.13	.521	.52	1.88	2.33
WV PC-7-2		.43	.044	.072	1.7	1.61	.540	.54	1.87	2.33
WV PC-7-3		.43	.072	.085	1.7	1.94	.502	.62	1.84	2.33
AVG PC-7	1.2	.45	.055	.078	1.7	1.9	.521	.56	1.86	2.33
ND PC-8-1	.095	<.0095	.029	.048	.28	<.5	.148	.13	.072	.10
ND PC-8-2		<.0095	.043	.044	.28	<.5	.155	.14	.097	.095
ND PC-8-3		<.0095	.049	.046	.56	1.23	.158	.19	.089	.10
AVG PC-8			.040	.046	.38	(1.23)	.153	.16	.086	.10

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

				Ba (ppm)					La (ppm)			
Technique Material analyzed Location	WC	EDXRF Ash R	ICAP-AES Ash R	ICAP-AES Ash D	EDXRF WC R	DCAES Ash R	DCAES WC R	INAA WC R	ICAP-MS Ash D	EDXRF Ash R	EDXRF WC R	DCAES Ash R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	68 63	51 54.7 50 52	54 55 57 55	53 51 63 56	54 68 59 60	40 43 39 41	88 72 84 81	10.0 10.0 10.3 10.1	6.8 8.1 8.1 7.7	7.0 8.1 9.5 8.2	31 9 9 16	8.6 5.3 6.7 6.9
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	318 297	270 280 260 270	{15} {15} {14} 15	280 290 310 290	404 407 415 410	290 310 360 320	200 260 300 250	5.32 5.34 5.38 5.35	4.2 3.4 4.3 4.0	<3 <3 <3 —	4 4 11 6	4.8 6.5 7.1 6.1
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	91 96	75.3 74.5 73.7 74.5	68 87 87 81	83 77 86 82	112 122 129 121	51 53 76 60	140 120 92 120	6.11 6.10 6.09 6.10	6.5 6.5 4.8 5.9	<5 <5 <5	16 15 15 15	8.7 8.9 10 9.3
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	51 44	34.0 37.7 35.4 35.7	42 41 39 41	39 39 38 39	46 22 19 29	36 34 45 38	66 72 60 66	6.19 6.15 6.11 6.15	4.6 4.6 4.6 4.6	<3 4.6 4.2 4.4	23 1 2 9	5.5 6.1 5.7 5.8
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	203 193	170 190 180 180	$\{16\}\ \{16\}\ \{16\}\ 16\}\ 16$	220 200 200 210	134 124 164 140	190 190 260 210	400 460 500 450	6.78 6.77 6.73 6.76	5.98 5.61 5.41 5.67	$ \{ 1.6 \} \\ \{ 2.6 \} \\ \{ 2.3 \} \\ 2.2 $	1 4 10 5	6.9 6.9 7.9 7.2
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	37 36	 	$ \begin{array}{r} 32\\ 30\\ -\\ 31 \end{array} $	33 34 37 35	28 32 28 29	34 31 41 36	 	3.41 3.27 3.24 3.31	2.3 2.8 2.3 2.5	 	9 9 11 10	3.0 2.9 2.9 2.9
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	116 116	110 112 107 110	120 120 130 120	110 140 120 120	230 202 200 210	130 140 120 130	 	21.7 21.6 21.3 21.5	17 18 17 17	15 18 19 17	15 18 14 16	14 14 12 14
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	660 671	480 470 450 460	430 390 450 420	520 530 530 530	1,000 1,040 1,060 1,030	500 700 540 580		2.79 2.84 2.84 2.82	1.9 1.9 1.9 1.9	 	8 6 10 8	3.5 3.8 3.7 3.7

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

		Ce (ppm)					opm)		Nd (ppm)			Sm (ppm)	
Technique Mat. analyzed Location	INAA WC R	ICAP-MS Ash D	EDXRF Ash R	EDXRF WC R	DCAES Ash R	ICAP-MS Ash D	DCAES Ash R	INAA WC R	ICAP-MS Ash D	DCAES Ash R	INAA WC R	ICAP-MS Ash D	DCAES Ash R
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	18.1 18.2 18.7 18.3	15.0 17.3 18.4 17.6	15.1 18.4 17.5 17.0	27 46 48 40	<9 10 <9 (10)	1.82 1.96 2.01 1.93	<14 <14 <14 	<20 <18 8.0 (8.0)	7.2 7.4 7.8 7.5	10 <4 5.0 8	1.98 1.94 2.00 1.97	1.47 1.51 1.65 1.55	<1.4 <1.4 <1.4
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	9.07 9.5 9.6 9.4	8.0 7.6 8.5 8.0	6.1 4.2 <3 5.2	18 21 35 25	6.3 10 9.3 9	.918 .867 .994 .926	<9 <9 <9 —	<13 <11 <9	3.7 3.4 3.7 3.6	<6 <6 <6	.968 .988 .977 .978	.71 .65 .76 .71	<.8 <.8 <.8 —
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	12.5 12.5 12.8 12.6	13 13 11 12	6.8 8.4 6.5 7.2	23 39 64 40	<32 <32 28 (28)	1.4 1.5 1.2 1.3	<17 <17 <17 	<10 <8 <11	5.5 5.7 4.5 5.2	<11 <11 6.5 (6.5)	1.19 1.18 1.22 1.20	1.1 1.1 .86 1.0	<1.6 <1.6 <1.6
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	11.5 11.5 11.1 11.3	10.7 10.3 9.75 10.3	7.4 9.94 9.57 9.0	19 25 6 17	<18 <18 <18 	1.18 1.11 1.07 1.12	<10 <10 <10 	<12 <8 <10	4.7 4.3 4.0 4.3	4.3 4.2 3.2 3.9	1.10 1.09 1.09 1.10	.91 .75 .81 .83	<.9 <.9 <.9 —
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	11.6 11.8 11.5 11.6	12.0 11.2 10.9 11.3	{5.30} 8.59 8.90 7.6	4 19 8 10	12 10 14 12	1.38 1.29 1.20 1.29	<6 <6 <6	<18 <16 <14	5.51 4.9 4.7 5.1	5.3 6.9 5.8 6.0	1.23 1.21 1.21 1.22	1.05 .965 .965 .995	.74 .69 .85 .76
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	4.88 4.67 4.84 4.81	4.5 4.83 4.69 4.7	 	20 13 12 15	7.4 7.4 6.0 6.9	.524 .557 .557 .546	.44 .42 .39 .42	<3 2.5 1.6 2.0	2.1 2.1 2.1 2.1	2.9 <7 <7 (2.9)	.517 .500 .507 .508	.36 .37 .39 .37	.37 .38 .36 .37
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	36.0 35.2 36.1 35.8	35.7 38.4 37.1 37.1	28.2 31.7 29.9 29.9	64 54 48 55	25 27 25 26	3.86 4.28 4.04 4.06	1.94 1.79 <2 1.87	12.4 11.0 12.4 11.9	15 16 14 15	19 19 18 19	3.56 3.50 3.51 3.52	2.72 2.95 2.74 2.80	1.7 1.7 <.6 1.7
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	4.34 4.33 4.68 4.45	4.2 4.1 4.0 4.1	 	21 19 17 19	<6 <6 <6	.41 .47 .42 .43	<.7 <.7 <.7	<3 2.1 2.5 2.3	1.5 1.6 1.4 1.5	<3 <3 <14	.420 .419 .388 .409	.27 .26 .24 .26	<.3 <.3 .47 (.47)

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples-Continued.

[Analyses of splits for each sample are followed by an average. -, no data. See definitions on p. 77]

	Eu (ppm) Tb (ppm		b (ppm)) Yb (ppm)			Н	f (ppm)	Ta	a (ppm)	W (ppm)			
Technique Mat. analyzed Location	WC	ICAP-MS Ash D	DCAES Ash R	INAA WC R	ICAP-MS Ash D	INAA WC R	ICAP-MS Ash D	DCAES Ash R	INAA WC R	ICAP-MS Ash D	INAA WC R	ICAP-MS Ash D	INAA WC R	ICAP-MS Ash D
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	.399 .406	0.32 .35 .34 .34	<0.3 <.3 <.3	0.275 .252 .260 .262	0.22 .24 .24 .23	0.91 .84 .90 .88	0.74 .78 .92 .81	0.84 .82 .86 .84	0.647 .660 .67 .66	0.81 .80 .92 .85	0.207 .224 .227 .219	0.22 .19 .18 .20	1.00 1.09 1.09 1.06	1.1 1.3 1.3 1.2
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	.199 .195	.15 .16 .19 .17	<.2 <.2 <.2	.122 .122 .115 .120	.10 .10 .11 .10	.42 .443 .40 .42	.32 .36 .44 .37	.23 .30 .35 .29	.621 .622 .560 .601	.65 .68 .71 .68	.141 .155 .142 .146	.13 .12 .10 .12	.38 .36 .40 .38	.48 .41 .49 .46
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	.222	.23 .19 .18 .20	<.4 <.4 <.4	.147 .131 .136 .138	.14 .15 .12 .14	.51 .511 .551 .52	.52 .52 .52 .52	.63 .58 .68 .61	.54 .532 .578 .55	.92 .71 .68 .76	.188 .197 .189 .191	.19 .18 .16 .18	1.53 1.52 1.50 1.52	1.9 2.1 1.7 1.62
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	.214	.17 .18 .17 .17	<.3 <.3 <.3 —	.133 .131 .132 .132	.11 .12 .11 .11	.478 .463 .468 .470	.44 .42 .43 .43	.63 .51 .48 .54	.518 .49 .500 .50	.63 .61 .60 .61	.168 .163 .168 .166	.13 .14 .13 .13	.79 .80 .76 .78	.89 .86 .82 .86
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	.241 .218	.22 .21 .21 .21	<.3 <.2 <.2	.158 .173 .169 .167	.17 .16 .15 .16	.555 .56 .545 .555	.578 .572 .567 .572	.46 .47 .42 .45	.441 .429 .444 .438	.58 .58 .53 .56	.117 .108 .123 .116	.13 .11 .10 .11	.84 .91 .77 .84	1.2 1.1 1.0 1.1
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	.098 .098	.078 .087 .087 .084	<.2 <.1 .11 (.11)	.065 .054 .054 .058	.051 .060 .055 .055	.204 .198 .211 .204	.19 .20 .21 .20	.22 .20 .21 .21	.478 .499 .458 .478	.51 .60 .56 .56	.060 .046 .038 .048	.055 .060 .056 .046	.42 .40 .38 .42	.51 .51 .51 .51
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	.651 .697	.60 .62 .62 .61	.47 .48 .89 .61	.408 .390 .402 .400	.37 .43 .39 .40	1.69 1.61 1.52 1.61	1.4 1.7 1.5 1.5	1.8 1.4 1.4 1.5	1.88 1.80 1.80 1.83	2.3 2.3 2.3 2.3	.675 .617 .630 .641	.62 .68 .56 .583	1.33 1.19 2.53 1.68	1.6 {18} 1.7 1.6
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	.077 .087	.053 .052 .058 .054	<.2 <.2 <.2	.052 .052 .063 .056	.053 .053 .058 .055	.317 .298 .245 .287	.21 .25 .23 .23	.22 .19 .20 .20	.342 .341 .340 .341	.41 .45 .41 .42	.095 .088 .095 .092	<0.1 <0.1 <0.1	.38 .25 .43 .35	.34 .38 .36 .36

Appendix 1. Comparison of the concentrations of 51 elements determined by multiple analytical techniques in 8 Argonne Premium Coal samples—Continued.

		Pb (p	pm)		Bi (p	pm)	Th	(ppm)	U	(ppm)
Technique Material analyzed Location	GFAAS Ash R	ICAP-MS Ash D	DCAES Ash R	DCAES WC R	ICAP-MS Ash D	DCAES Ash R	INAA WC R	ICAP-AES Ash D	INAA WC R	ICAP-MS Ash D
UF PC-1-1 UF PC-1-2 UF PC-1-3 AVG PC-1	7.8 7.4 6.9 7.4	7.7 7.7 7.4 7.6	10 10 11 10	8.0 10 6.0 8	0.12 .12 .11 .12	<1.4 <1.4 <1.4	2.59 2.64 1.55 2.26	2.8 2.8 2.6 2.7	0.81 .81 .86 .83	1.2 1.2 1.1 1.2
WY PC-2-1 WY PC-2-2 WY PC-2-3 AVG PC-2	3.1 3.0 2.7 2.9	1.6 1.6 1.5 1.6	2.6 4.1 3.5 3.4	<2 <2 <2 <2	.055 .055 .049 .053	<.9 <.9 <.9 —	1.56 1.58 2.01 1.71	1.4 1.6 1.3 1.4	.55 .54 .52 .54	.65 .62 .63 .63
IL PC-3-1 IL PC-3-2 IL PC-3-3 AVG PC-3	6.8 6.3 6.3 6.5	8.6 8.7 8.1 8.5	14 13 14 14	9 8 6 8	.089 .087 <.08 .088	<1.7 <1.7 <1.7	1.95 1.95 1.99 1.96	2.4 2.1 1.9 2.2	4.46 4.06 4.28 4.27	6.84 6.58 5.95 6.46
PITT PC-4-1 PITT PC-4-2 PITT PC-4-3 AVG PC-4	2.9 2.9 2.9 2.9 2.9	3.8 4.0 3.7 3.8	7.3 6.9 7.7 7.3	4 3 5 4	.12 .12 .11 .12	2.0 1.5 <.9 1.8	1.53 1.46 1.44 1.48	1.8 1.4 1.6 1.6	.36 .37 .45 .39	.60 .56 .55 .57
POC PC-5-1 POC PC-5-2 POC PC-5-3 AVG PC-5	2.5 2.4 2.4 2.4	1.8 1.7 1.6 1.7	12 8.5 12 11	5 4 5 5	.049 .052 .053 .052	<.5 <.5 <.5 —	1.19 1.12 1.17 1.16	1.2 1.1 1.2 1.2	.47 .51 .55 .51	.646 .615 .631 .631
UT PC-6-1 UT PC-6-2 UT PC-6-3 AVG PC-6	1.6 1.6 — 1.6	1.4 1.5 1.5 1.5	2.1 1.7 2.0 1.9	 	.035 .036 .035 .035	<1 <1 <1	.609 .614 .622 .615	.60 .60 .55 .58	.76 .70 .76 .74	.851 .823 .819 .852
WV PC-7-1 WV PC-7-2 WV PC-7-3 AVG PC-7	12 12 12 12	12 12 12 12	18 16 13 16	 	.23 .23 .23 .23	<5 <5 <5	6.49 6.26 6.49 6.41	6.2 6.4 6.6 6.4	1.67 1.60 1.64 1.64	2.17 2.21 2.10 2.16
ND PC-8-1 ND PC-8-2 ND PC-8-3 AVG PC-8	1.5 1.5 1.5 1.5	1.9 .84 1.5 1.6	2.7 2.1 3.0 2.6	 	<.05 <.05 <.05 —	<2 <2 <2 -	1.06 1.03 1.11 1.07	{.6} {.8} {.7} .7	.52 .46 <.2 .49	.42 .47 .46 .45

Appendix 2. Statistical parameters for the data in appendix 1.

Concentrations of 51 elements in 8 Argonne Premium Coals were determined by multiple analytical techniques (appendix 1). Statistical parameters for non-outlier concentrations determined by high-precision (HP) techniques and for concentrations (including outliers) determined by all techniques (HP and less precise techniques) are listed in this appendix in separate columns for each element; only one column is needed for elements for which all values were determined by HP techniques and no concentrations were outliers (Li, Cs, Tb, Hf, Ta, U). Concentrations of major elements (Si, Al, Fe, Mg, Ca, Na, K, Ti, P, Mn) are in weight percent (wt. %); concentrations of trace elements are in parts per million (ppm).

The mean values in HP columns are recommended values if there is no box for that sample in that column. They are concentrations determined from the arithmetic mean of the HP values, except values that were excluded in tables 5 and 6 of the paper (this volume) by Palmer and Klizas. The mean concentration is a recommended value if all the following conditions are met:

- (1) the number of determinations (*n*) is greater than 3,
- (2) the relative standard deviation (Rel Std Dev) is less than 5 percent for major elements present in concentrations greater than 0.1 percent or is less than 10 percent for trace elements or for major elements present in concentrations less than 0.1 percent (see paper by Palmer and Klizas), and
- (3) the deviation of the arithmetic mean and the geometric mean (Dev of Means) is less than 1 percent.

If any of these conditions is not met (as indicated by a box around the value), the mean is reported as an average value in tables 5 and 6 of the paper by Palmer and Klizas.

The standard deviation represents the error in the recommended values and was used to determine significant figures. The kurtosis and skewness are also included. ERR indicates that there were insufficient data to calculate the statistical parameter.

Appendix 2. Statistical parameters for the data in appendix 1–Continued.

	Si HP	Si All	Al HP	Al All	Fe HP	Fe All	Mg HP	Mg All	Ca HP	Ca All
UF PC-1										
n Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Geometric Mean (wt. %) . Dev of Means (%) Kurtosis Skewness WY PC-2	5 2.63 .081 3.09 2.62 .04 .641 -1.05	11 3.0 .43 14.6 2.9 .89 2.68 1.53	8 1.58 .046 2.94 1.57 .04 753 277	14 1.8 .38 20.4 [1.8 1.76 .508 1.22	11 1.8 .11 5.93 1.8 .16 .0521 304	17 1.8 .34 18.6 1.8 1.39 8.44 2.32	8 .080 .0023 2.93 .079 .04 .900 904	14 .10 .028 28.8 .09 3.33 .797 1.42	8 .42 .023 5.48 .42 .13 -2.17 .590	14 .5 .10 21.0 .5 1.98 -1.30 .573
n Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Geometric Mean (wt. %) . Dev of Means (%) Kurtosis Skewness	1.1 .49	11 .27 24.9 1.0 3.51 .766 720	8 .034 4.94 .68 .11 973 567	14 .6 .10 17.1 [.6 1.50 959 680	11 .35 .025 6.99 .35 .23 -1.00 255	17 .33 .048 14.8 .32 1.22 1.70 -1.3	8 .028 .028 .028 .028 .028 .028 .028 .02	14 .25 .066 26.5 .24 2.91 .321 1.22	8 1.12 .087 7.70 1.12 .27 124 840	14 1.1 .13 11.6 1.1 .61 147 .735
IL PC-3 <i>n</i>	4 3.05 .041 1.35 3.04 .01 1.79 .713	11 3.2 .43 13.3 3.2 .76 0299 1.05	7 1.24 .035 2.79 1.24 .03 -1.29 .235	14 1.6 .38 24.3 [1.5 2.72 -2.08 .337	11 2.6 .16 2.6 .18 .782 -1.32	17 2.5 .28 11.0 2.5 .60 195 588	6 .075 .0037 4.94 .075 .11 3.97 -1.89	14 .09 .016 17.6 .09 1.43 966 .393	8 .93 .039 4.16 .93 .08 -1.40 532	14 1.2 .44 35.4 1.2 5.37 951 .920
PITT PC-4 <i>n</i>	5 1.86 .088 4.72 1.86 .09 -3.16 .638	11 2.2 .46 20.7 2.2 1.83 951 .872	7 .99 .011 1.14 .99 .01 -2.26 0177	14 1.4 .54 39.7 [1.3 6.96 -1.61 .732	11 1.33 .071 5.33 1.33 .13 .191 0179	17 1.4 .29 20.7 1.4 1.94 .218 .827	6 .036 .0027 7.71 .035 .25 -3.33 -5.7×10-15	14 .05 .017 35.9 .04 5.13 2.30 1.55	8 .197 .0099 5.02 .197 .11 -1.21 31 3	14 .093 35.7 .25 5.23 .330 1.24
POC PC-5 <i>n</i>	4 .76 .038 4.95 .76 .09 .257 .358	10 .9 .10 12.1 .8 .65 -1.39 .397	7 .552 .0062 1.12 .552 .01 -1.05 .812	14 .7 .13 19.9 .6 1.71 511 .944	11 .51 .015 2.98 .51 .04 33 892	17 .54 .085 15.8 .53 1.01 5.51 2.26	8 .058 .0055 9.37 .058 .36 225 1.29	14 .07 .021 28.2 .07 3.43 416 .956	8 .452 .0069 1.52 .452 .01 .264 0133	14 6 15 27.7 .5 3.08 1.41 1.47
UT PC-6 <i>n</i>	4 .027 2.97 .91 .03 -2.57 .257	7 .89 .068 7.66 .89 .28 5.17 -2.20	6 .37 .020 5.31 .37 .12 -1.89 .045	9 .35 .035 10.1 [.35 .47 -1.01 351	9 .30 .016 5.37 .30 .13 -1.19 .509	12 .28 .034 12.0 .28 .74 .994 -1.14	$ \begin{array}{r} 6 \\ .033 \\ .0074 \\ \hline 22.5 \\ .032 \\ \hline 1.77 \\ 5.57 \\ 2.33 \\ \end{array} $	9 .036 .0071 20.0 .035 1.72 -1.04 820	5 .41 .013 3.16 .41 .04 -3.07 293	9 .36 .063 17.3 .36 1.43 -1.98 435
n Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Geometric Mean (wt. %) . Dev of Means (%) Kurtosis Skewness.	4 5.0 .22 4.32 5.0 .07 1.50 1.19	7 5.1 48 9.51 5.1 .39 .274 105	7 3.16 .024 .75 3.16 0 861 802	10 3.4 12.9 3.4 .68 1.64 1.78	10 .39 .014 3.52 .39 .06 -1.30 .242	13 .39 .024 6.21 .39 .18 1.69 .647	6 .090 .0057 6.39 .090 .17 -1.67 470	10 .09 .014 15.9 .09 .99 5.40 2.14	6 .058 .0017 2.89 .058 .03 2.50 1.15	$\begin{array}{c} 10\\ .06\\ .013\\ 19.6\\ .06\\ 1.46\\ 4.10\\ 2.14 \end{array}$
ND PC-8 <i>n</i>	3 .67 0 .67 0 ERR ERR	7 .65 .068 10.4 .65 .47 1.90 0112	6 .400 .0036 .90 .400 0 -1.94 202	10 .35 .088 24.7 .34 3.40 534 -1.06	10 .49 .042 8.61 .49 .32 793 .975	13 .46 .067 14.6 .46 1.06 .404 533	7 .38 .029 7.73 .38 .26 -2.14 0612	10 .42 .068 16.2 .42 1.15 -1.37 -1.07	7 1.49 .061 4.07 1.49 .07 -2.33 114	10 1.4 .12 8.06 1.4 .31 .515 935

Appendix 2.	Statistical	l parameters	for the	data in ap	ppendix 1	-Continued.
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	Na HP	Na All	K HP	K All	Ti HP	Ti All	P HP	P All	Mn HP	Mn All
UF PC-1										
n Mean (wt. %) Standard Dev (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Dev (%) Geometric Mean (wt. %) Dev of Means (%) Dev of Means (%) Standard Dev (%) Skewness Skewness		14 .035 .0055 15.6 .035 1.03 .810 1.41	11 .28 .013 4.53 .28 .09 111 038	14 .28 .015 5.20 .28 .13 276 .169	8 .0032 4.08 .077 .07 .293 614	14 .08 .017 21.8 .08 2.62 .893 547	6 .007 .0011 16.3 .007 1.12 -3.03 .115	8 .007 .0012 18.1 .006 1.42 -1.60 .409	6 .0043 .00014 3.24 .0042 .04 -2.30 -7.44×10 ⁻¹⁶	12 .0048 .00067 13.9 .0048 .87 -1.42 .471
<i>n</i>		14	9	14	8	14	6	14	6	12
Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%)[Geometric Mean (wt. %) Dev of Means (%) Kurtosis Skewness	.11 .012 10.2 .11 .55 4.60 -2.19	.12 .016 13.9 [.12 .75 4.02 -1.45	.030 .0044 14.6 .030 .89 489 1.05	.031 .0039 12.8 .030 .73 598 .753	.053 .0033 6.27 .053 .17 -1.75 420	.047 .0081 17.2 .047 1.55 132 815	.024 .0023 9.72 .024 .41 -2.00 568	.026 .0043 16.6 .026 1.32 -1.23 .0122	$\begin{array}{r} .0020\\ 7.52 \times 10^{-5}\\ 3.80\\ .0020\\ .06\\104\\ .313\end{array}$.0022 .00084 39.0 .0020 6.54 2.40 1.34
IL PC-3	10	14						0		10
n	10 .10 .010 9.67 .10 .39 4.46 1.92	14 .022 19.6 [.11 1.64 .00612 1.14	11 .20 .013 6.20 .20 .17 .170 .811	14 .22 .028 12.9 .21 .71 1.00 1.30	6 .070 .0028 4.04 .069 .07 4.51 -1.99	14 .07 .012 16.9 .07 1.28 1.23 .807	3 .0059 .00012 1.95 .0059 .01 ERR -1.73	8 .005 .0012 25.1 .004 2.83 -1.67 .278	6 .0081 .00065 8.03 .0081 .26 3 .857	12 .011 .0033 30.9 .010 4.07 453 .949
<i>n</i>	9	14	11	14	8	14	6	11	6	12
Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Geometric Mean (wt. %) Dev of Means (%) Kurtosis Skewness POC PC-5		.034 .0028 8.33 .034 .31 .366 .845	.111 .0025 2.21 .111 .02 -1.14 .865	.115 .0091 7.90 .115 .27 3.67 2.00	.055 .0040 7.35 .055 .24 -1.51 177	.06 .016 25.0 .06 2.53 .671 .145	.010 .0020 20.8 .009 2.16 1.30 -1.28	.009 .0014 15.3 .009 1.24 2.61 -1.24	.0019 .00019 9.99 .0019 .42 -2.69 -2.3×10 ⁻¹⁵	.0024 .00082 34.8 .0022 5.02 156 1.10
n	11	14	9	14	8	14	5	8	6	12
Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Geometric Mean (wt. %) Dev of Means (%) Kurtosis Skewness	.079 .0047 5.92 .079 .17 1.04	.09 .022 24.4 .09 2.40 .317 1.41	.031 .0039 12.9 .030 .72 -1.30 .592	.031 .0039 12.5 .031 .72 789 .438	.039 .0022 5.65 .039 .13 4.49 2.03	.05 .0202 44.1 .04 7.84 .287 1.33	$\begin{array}{r} .00222\\ 8.4 \times 10^{-5}\\ 3.77\\ .00220\\ .06\\612\\512\end{array}$.0021 .00013 6.09 .0021 .16 -1.92 .105	.0017 .00013 7.74 .0017 .25 -2.25 .326	.0019 .00038 19.3 .00191 1.64 -1.24 .711
UT PC-6 <i>n</i>	9	12	5	10	5	9	2	7	5	8
Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%)[Geometric Mean (wt. %) Dev of Means (%)	.135 .0089 6.64 .134 .19 -1.33 .640	12 .12 .030 25.3 [.11 3.82 [.0315 -1.19	.025 .0043 17.0 .025 1.34 3.15 -1.68	.021 .0056 26.1 .021 3.29 -1.59 0162	023 0 0 .023 0 ERR ERR	.021 .0022 10.5 .021 .51 -1.61 621	.00095 3.5×10 ⁻⁵ 3.74 .00095 .04 ERR ERR	.0008 .00014 16.9 .0008 1.32 436 465	.000416 8.94×10 ⁻⁶ 2.15 .000416 .02 .312 1.26	.00049 .000127 26.0 .00048 2.57 1.05 1.56
WV PC-7										
n Mean (wt. %) Standard Dev (wt. %) Rel Std Dev (%) Geometric Mean (wt. %) Dev of Means (%) Kurtosis Skewness ND PC-8	8 .0348 .00085 2.45 .0348 .03 2.95 -1.07	13 .037 .00499 13.4 .037 .73 3.58 2.10	10 .49 .022 4.50 .49 .09 .0548 0434	13 .46 .064 13.9 .46 1.01 .345 -1.24	7 .25 .015 <u>6.27</u> .25 .17 -1.10 473	10 .22 .036 15.8 .22 1.22 -1.37 623	3 .0083 .00081 9.77 .0083 .33 ERR -1.62	8 .0091 .0032 35.5 .0088 4.08 7.15 2.63	6 .00147 8.16×10 ⁻⁵ 5.57 .0015 .13 3 .857	9 .0016 .00030 18.5 .0016 1.38 1.69 1.43
n	10 .50 .027 5.40 .50 .13 720 .305	11 .079 16.6 [.47 1.75 [8.14 -2.69	5 .034 .0063 18.6 .034 1.31 1.05 1.19	12 .03 .010 34.5 .03 7.32 .646 597	6 .0192 .00083 4.35 .0192 .08 -2.27 297	10 .019 .0012 6.20 .019 .18 -1.05 559	6 .012 0 .012 0 ERR ERR ERR	14 .014 32.7 .013 4.96 961 .625	6 .0081 .00016 1.97 .0081 .02 -1.31 .0405	10 .009 .0011 12.3 .009 .67 948 .497

Appendix 2. Statistical parameters for the data in appendix 1–Continued.

	Li HP	Be HP	Be All	B HP	B All	Sc HP	Sc All	V HP	V All
UF PC-1									
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 14.8 0 14.8 0 -3.33 1.37	6 1.5 0 1.5 0 ERR ERR	9 1.5 .1 6.82 1.5 .23 9.00 -3.00	3 38 9.4 24.5 38 1.92 ERR 1.37	6 34 12 35.3 32 5.71 -1.67 .127	6 3.99 .094 2.37 3.99 .02 384 .863	9 3.4 .99 29.4 3.2 5.19 302 -1.142	6 26.5 .55 2.07 26.5 .02 -3.33 0	12 28 9.6 34.9 26 5.99 301 .448
WY PC-2 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 4.0 .11 2.85 4.0 .03 -2.14 .357	3 .25 .01 4.0 .250 .000534 ERR 1.46×10 ⁻¹⁷	9 .22 .056 25.9 .21 3.53 -1.68 817	3 96 13 13.7 96 .62 ERR .966	6 83 17 21.5 82 1.74 .0757 .724	6 1.66 .034 2.05 1.66 .02 -1.78 479	9 1.5 .23 14.6 1.5 1.17 4.54 -2.09	6 14 .00 0 14 0 ERR ERR	12 13 2.8 22.1 12 2.59 344 624
IL PC-3 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 7.8 .19 2.41 7.8 .02 446 668	3 .76 .017 2.3 .7599 .000172 ERR 1.73	9 .8 .16 19.6 .8 1.62 .109 .949	3 160 2.9 1.80 160 .01 ERR -1.73	6 140 23 16.5 140 1.17 -2.43 272	6 2.65 .094 3.55 2.64 .05 -2.01 .577	9 2.6 .11 4.04 2.6 .07 -1.39 .429	6 35 3.3 9.48 34 .37 -2.08 .309	12 36 11 30.7 35 4.19 .239 .837
PITT PC-4 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 8.6 .36 4.17 8.6 .07 .521 .673	3 .77 .012 1.51 .77 .01 ERR 1.73	9 .9 .21 23.5 .9 2.28 -1.43 .911	3 47.8 .95 1.99 47.8 .01 ERR .158	6 50 2.4 4.84 50 .10 -1.35 .613	6 2.53 .055 2.18 2.53 .02 -1.36 .583	9 2.3 .36 15.6 2.3 1.20 -1.03 962	$ \begin{array}{r} 6 \\ 16 \\ 1.1 \\ 6.85 \\ 16 \\ .20 \\ -3.33 \\ 0 \end{array} $	12 17 4.7 27.42 17 3.15 .710 1.17
POC PC-5 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 5.6 .27 4.78 5.6 .10 -1.88 960	6 .80 0 .80 0 -3.33 1.37	9 .76 .077 10.2 .75 .52 3.24 -1.89	3 13.1 .64 4.86 13.1 .08 ERR 1.73	6 15 2.6 16.8 15 1.17 -1.34 .487	6 1.78 .023 1.32 1.78 .01 -2.41 245	9 1.6 .21 13.0 1.6 .83 787 -1.07	6 11.5 .55 4.76 11.5 .09 -3.33 0	12 12 4.4 36.1 11 6.08 .210 .740
UT PC-6 <i>n</i>	5 5.3 .25 4.72 5.3 .09 -3.33 609	2 .125 .0071 5.66 .125 .08 ERR ERR	10 .16 .039 24.5 .16 2.45 .312 1.19	3 127 5.2 4.09 127 .05 ERR 1.73	3 127 5.2 4.09 127 .05 ERR 1.73	6 .78 .037 4.75 .78 .09 -1.52 145	9 .75 .060 8.04 .75 .30 00338 618	5 4.3 .27 6.27 4.27 .16 -2.41 166	8 4.05 .38 9.43 4.03 .39 -1.55 .185
WV PC-7 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness ND PC-8	6 28 1.0 3.70 28 .06 -3.33 -7.7×10 ⁻¹²	6 2.0 .11 5.48 2.0 .13 5 4.76×10 ⁻¹⁸	12 2.3 .51 22.0 2.3 2.10 -1.53 .746	3 56 20 35.9 50 4.25 ERR 1.09	6 56 24 43.1 50 8.02 851 .686	6 7.5 .14 1.84 7.5 .01 -1.95 .474	9 7.13 .62 8.69 7.11 .36 .154 -1.26	6 43.3 .82 1.88 43.3 .01 6.00 2.449	9 39 6.1 15.4 39 1.17 784 -1.03
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 2.69 .078 2.88 2.69 .03 6.00 2.45	2 .175 .0071 4.04 .175 .04 ERR ERR	10 .22 .054 24.2 .22 2.42 .0493 1.09	3 79 1.1 1.38 79 .01 ERR 1.73	6 73 12 17.0 72 1.52 5.65 -2.36	3 .80 .049 6.07 .20 .20 267 .231	9 .83 .064 7.71 .83 .26 230 .314	6 3.7 .21 5.80 3.7 .14 -2.15 232	9 3.6 .18 4.97 3.6 .11 -1.19 .440

Appendix 2. Statistical parameters for the data in appendix 1–Continued.

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	Cr HP	Cr All	Co HP	Co All	Ni HP	Ni All	Cu HP	Cu All	Zn HP	Zn All
UF PC-1										
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis	$\begin{array}{r} 22 \\ 2.6 \\ \hline 11.9 \\ 22 \\ .62 \\530 \end{array}$	21 23 5.2 22.8 22 2.20 3.11	9 5.2 4.25 5.2 .08 3.43	15 4.8 .71 14.7 4.8 1.22 2.20	$ \begin{array}{c} 12\\ 15\\ 2.3\\ \hline 15.5\\ \hline 1.04\\ 4.65\\ \end{array} $	21 17 5.0 28.8 17 3.62 .00992		18 18 1.4 7.37 18 .26 783	12 20 1.8 9.10 20 .36 2.36	18 24 7.0 29.0 23 3.63 238
Skewness WY PC-2	.930	1.53	-1.76	-1.70	1.36	1.03	553	401	1.30	.997
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	.69 10.6 6.5	20 6 1.3 21.0 6 2.42 1.55 209	9 1.65 .088 5.32 1.65 .13 .326 .112	12 1.6 .21 13.7 1.5 1.02 3.71 -1.71	$ \begin{array}{r} 12 \\ 4.7 \\ .50 \\ \hline 10.8 \\ 4.6 \\ .56 \\106 \\583 \\ \end{array} $	18 5.3 1.4 25.9 5.2 2.70 2.58 1.69	9 13.0 .79 6.08 12.9 .16 -1.35 001	18 12.7 3.3 26.2 12.2 4.39 1.22 836	12 11.0 .96 8.73 10.9 .34 .343 .593	15 14 6.2 44.8 13 7.21 2.08 1.83
IL PC-3 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness PITT PC-4	5.2 14.6 35 .96 -1.38	21 38 7.7 19.9 38 1.76 .582 .992	9 4.4 .18 4.35 4.4 .08 636 .867	12 4.2 .37 8.87 4.2 .39 1.25 -1.12	11 19 2.5 12.9 19 .78 2.74 .172	21 24 6.7 28.2 23 3.67 823 .690	9 10.1 .78 7.74 10.1 .27 211 411	18 11 1.9 17.7 11 1.36 .569 1.19	9 200 25 12.3 200 .68 .393 .486	18 190 60 32.3 180 5.37 0399 .320
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	$ \begin{array}{r} 1.6 \\ 10.3 \\ 16 \\ .49 \\529 \end{array} $	21 16 4.7 28.9 16 4.78 .905 .00908	9 2.6 .10 4.03 2.6 .07 .280 -1.12	12 2.6 .15 5.95 2.6 .17 222 499	11 9.2 .89 9.66 9.2 .40 2.16 1.30	20 11 2.5 22.8 11 2.34 170 .826	9 5.8 .40 6.95 5.8 .21 -1.77 .221	18 7 1.6 24.2 6 2.45 .548 1.33	11 8.3 .83 9.91 8.3 .41 4.78 1.92	18 13 7.3 56.4 11 12.3 1.87 1.63
POC PC-5 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	$ \begin{array}{c} 10 \\ 1.4 \\ \hline 14.1 \\ 10 \end{array} $	21 10 3.2 32.3 10 5.38 .782 .592	9 3.9 .13 3.20 3.9 .05 -1.45 .058	12 3.6 .65 18.1 3.5 1.83 .553 -1.43	12 7.6 .79 10.4 7.6 .48 .160 .611	20 8 1.8 21.5 8 2.07 251 .873	$9 \\ 14 \\ 1.9 \\ 13 \\ .85 \\ -1.75 \\ .725$	18 16 4.1 25.4 16 2.76 1.84 1.29	$ \begin{array}{r} 12 \\ 5 \\ 1.0 \\ \hline 19.8 \\ 5 \\ \hline 1.75 \\ .562 \\ .725 \\ \end{array} $	15 7 3.9 56.5 6 12.29 .597 1.48
UT PC-6 <i>n</i>	8 5.2 .31 6.04 5.1 .16 .874 293	11 5.1 9.44 5.0 .45 3.11 -1.54	8 .93 .072 7.74 .93 .27 -1.18 645	11 .9 .15 17.5 .8 1.55 -1.03 705	7 3.4 .17 4.99 3.4 .11 638 .169	13 3.9 .67 16.92 3.9 1.25 704 .872	5 4.1 .21 5.11 4.1 .11 -1.96 236	11 5 2.2 45.9 4 9.49 927 .829	$ \begin{array}{r} 8 \\ 6 \\ 1.1 \\ 18.8 \\ 6 \\ \hline 1.53 \\ -1.51 \\ .401 \end{array} $	8 6 1.1 19.2 6 1.53 -1.51 .401
WV PC-7 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness ND PC-8	40 3.8 9.51	15 40 3.7 9.34 40 915 .330	9 7.8 .34 4.33 7.8 .08 .0131 .805	12 7.4 .79 10.7 7.4 .57 .765 -1.15	12 16 1.2 7.43 16 .26 .325 0667	18 17 2.2 12.99 17 .75 1.46 1.08	9 21 2.1 10.0 21 .44 0445 .710	15 22 5.4 24.7 21 2.88 300 .518	12 12 1.1 9.17 12 .38 -1.01 .425	15 12 1.5 12.52 12 .75 5667 412
ND PC-8 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	8.79 2.4 .35	15 [2.4 .19 7.84 2.4 .28 482 415	3 .78 .021 2.75 .78 .02 ERR 1.32	9 .7 .15 20.0 .7 1.89 -1.10 181	$9 \\ 1.5 \\ .37 \\ 24.3 \\ 1.5 \\ 2.32 \\ 3.38 \\ 1.63$	15 2 2.3 91.4 2 23.46 9.67 2.98	$ \begin{array}{r} $	15 6 2.9 49.6 5 10.7 .0149 1.08	12 5.2 .47 9.06 5.1 .37 -1.28 .215	15 5.1 .57 11.0 5.1 .58 438 163

Appendix 2. Statistic	al paramete	rs for the da	ata in appendi	x I—Conti	nued.					
	Ga HP	Ga All	Ge HP	Ge All	As HP	As All	Se HP	Se All	Rb HP	Rb All
UF PC-1										
<i>n</i>	3	9	3	9	6	9	6	6	9	12
Mean (ppm) Standard Dev (ppm)	5.34 1.21	7 2.3	4.35 .053	3.9 .48	16.7 .59	21 8.0	1.9 .27	1.9 .27	21 1.2	21 1.2
Rel Std Dev (%)	2.26	32.2	1.22	12.4	3.52	37.5	14.1	14.1	5.93	5.98
Geometric Mean (ppm)	5.3	7	4.35	3.8	16.6	20	1.9	1.9	21	21
Dev of Means (%) Kurtosis	.02 ERR	4.15 260	0 ERR	.73 487	.05 428	5.23 1.57	.87 .556	.87 .556	.16 545	.16 629
Skewness	49	1.16	-1.46	698	1.0	1.62	368	368	192	031
WY PC-2										
<i>n</i>	3	9	3	3	6	6	6	6	6	9
Mean (ppm) Standard Dev (ppm)	1.9 .12	2.4 .70	.36 0	.36 0	3.0 .71	3.0 .71	2.0 .55	2.0 .55	2.7 .94	6 5.3
Rel Std Dev (%)	6.15	28.6	0	0	23.4	23.4	27.9	27.9	35.3	86.9
Geometric Mean (ppm) Dev of Means (%)	1.9 .12	2.4 3.18	.36 0	.36 0	2.9	2.9 2.64	1.9	1.9 3.06	2.5	4 40.76
Kurtosis	ERR	2.46	ERR	ERR	558	558	-1.17	-1.17	-3.16	-1.33
Skewness	1.508	1.55	-2.45	-2.45	523	523	.953	.953	.0580	.877
IL PC-3										
<i>n</i>	3.57	9 5	8.4	9 9	6 4.2	6 4.2	6 4.2	6 4.2	9 15.5	12 17
Mean (ppm) Standard Dev (ppm)	.107	1.6	.27	9 1.4	4.2	4.2 .53	4.2 .45	4.2 .45	.78	2.4
Rel Std Dev (%)	2.99	31.4	3.21	16.1	12.6	12.6	10.7	10.7	5.00	14.5
Geometric Mean (ppm) Dev of Means (%)	3.57 .03	5 4.03	8.4 .03	9 1.04	4.2 .67	4.2 .67	4.2 .45	4.2 .45	15.5 .11	16 .87
Kurtosis	ERR	.440	ERR	3.82	-2.06	-2.06	3.44	3.44	2.55	1.70
Skewness	-1.71	1.14	-1.293	1.69	051	051	1.72	1.72	1.40	1.65
PITT PC-4		0		7	(0	6	(0	10
<i>n</i> Mean (ppm)	3.3	9 4	3	7 1.5	6 8.0	9 11	6 1.6	6 1.6	9 8.3	12 8.2
Standard Dev (ppm)	.14	1.0	.13	.68	.56	4.6	.15	.15	.82	.71
Rel Std Dev (%) Geometric Mean (ppm)	4.27 3.3	22.7 4	10.5 1.2	46.7 1.4	6.92 8.0	41.8 10	9.64 1.6	9.64 1.6	9.83 8.3	8.64 8.2
Dev of Means (%)	0	2.43	.37	6.69	.21	7.15	.39	.39	.44	.34
Kurtosis	ERR	-1.44	ERR	6.59	.001	937	1.23	1.23	-1.02	484
Skewness	1.26	0535	.87	2.55	-1.15	1.00	087	087	312	.058
POC PC-5 <i>n</i>	3	9	3	6	6	9	6	6	6	9
Mean (ppm)	1.8	2.6	.29	.4	10.1	12	2.7	2.7	2.2	3
Standard Dev (ppm)	.10	.90	.065	.10	.52	2.8	.30	.30	.20	1.3 41.8
Rel Std Dev (%) Geometric Mean (ppm)	5.31 1.8	34.8 2.5	23.2	28.0 .4	5.16 10.1	24.2 11	2.7	11.0 2.7	8.76 2.2	41.0
Dev of Means (%)	.09	5.08	1.76	3.78	.11	2.18	.52	.52	.30	7.17
Kurtosis Skewness	ERR 158	660 .936	ERR 1.06	-1.81 415	-2.57 192	3.74 1.99	-1.34 173	-1.34 173	4.68 2.12	-1.03 .980
UT PC-6	150	.950	1.00	115	172	1.99	175	175	2.12	.900
n	3	6	3	3	4	4	6	6	5	8
Mean (ppm)	1.00	1.1	.23	.23	.45	.45	1.2	1.2	1.02	1.6
Standard Dev (ppm) Rel Std Dev (%)	.051 5.20	.24 21.1	.033 14.7	.033 14.7	.082	.082 19.7	.12	.12 10.4	.057 5.64	.90 55.2
Geometric Mean (ppm)	.99	1.1	.22	.22 .71	.44	.44	1.2	1.2	1.02	1.5
Dev of Means (%)	.09 EDD	1.74 213	.71 ERR	.71 ERR	1.65	1.65 1.2	.46 -1.97	.46 –1.97	.13	12.68 - 1.02
Kurtosis Skewness	ERR 1.73	1.08	-1.55	-1.55	-1.34	-1.34	178	-1.97	628 .138	1.02
WV PC-7										
<i>n</i>	3	6	3	5	6	6	5	6		12
Mean (ppm)	9.4	11	1.67	1.5	6.4	6.4	5.6	5.8	36	38
Standard Dev (ppm) Rel Std Dev (%)	.44 4.68	1.5 14.0	.058 3.46	.29 19.9	.81	.81 12.6	.30 5.30	.59 10.1	6.1	5.7 15.2
Geometric Mean (ppm)	9.4	10	1.67	1.4	6.4	6.4	5.6	5.8	36	37
Dev of Means (%) Kurtosis	.07 ERR	.79 .605	.04 ERR	1.68 -2.85	.67 –1.11	.67 –1.11	.11 2.86	.40 2.06	1.25 -1.48	1.13 - 1.26
Skewness	-1.72	.890	-1.73	590	306	306	1.40	1.54	-1.48	469
ND PC-8										
<i>n</i>	3	6	3	3	4	5	6	6	5	8
Mean (ppm) Standard Dev (ppm)	1.16 .04	1.4 .34	.37 .045	.37 .045	2.4 .51	2.0 .87	.59 .042	.59 .042	1.2 .27	2 1.5
Rel Std Dev (%)	.04 3.47	.34 23.4	12.3	.045 12.3	21.4	42.5	7.04	7.04	21.0	69.6
Geometric Mean (ppm)	1.16	1.4	.36	.36	2.3	1.8	.59	.59	1.2	2
Dev of Means (%) Kurtosis	.04 ERR	2.25 -2.14	.51 ERR	.51 ERR	2.15	11.43 154	.22 4.60	.22 4.60	-2.37	22.09 539
Skewness	1.73	.486	331	331	-1.98	-1.17	-1.99	-1.99	645	1.01

	1		11							
	Sr HP	Sr All	Y HP	Y All	Zr HP	Zr All	Nb HP	Nb All	Mo HP	Mo All
UF PC-1										
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness WY PC-2	57 7.6 13.4 56 1.13 10.8	18 59 7.0 11.9 58 .90 11.8 -3.08	991.011.29.56.0992.146	12 8 2.3 28.7 8 5.38 .761 -1.11	6 26 1.7 6.67 26 .19 -2.46 .0672	15 24 7.8 32.8 22 7.22 .395 708	6 2.5 .21 8.61 2.5 .30 -2.15 232	9 2.2 .49 22.2 2.1 2.61 202 878	3 2.5 .39 15.6 2.5 .77 ERR 1.73	6 2.0 .62 30.8 1.9 4.01 664 .558
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	257 6.6 2.55 257 .03 362	18 240 49 20.5 230 2.42 106 -1.07	9 3.7 .13 3.65 3.7 .06 2.11 152	12 3.5 .40 11.4 3.5 .71 5.55 -2.25	$ \begin{array}{r} 6 \\ 21 \\ 2.1 \\ \hline 10.3 \\ 21 \\ .45 \\ -1.97 \\ 179 \\ \end{array} $	12 19 3.1 16.1 19 1.38 1.65 -1.05	$ \begin{array}{r} 6 \\ 1.2 \\ .25 \\ \hline 20.5 \\ \hline 1.2 \\ \hline 1.87 \\ -1.1 \\453 \\ \end{array} $	9 1.3 .28 21.7 1.3 2.13 .745 .339	3 .52 .021 4.15 .52 .06 ERR 611	6 .59 .087 14.7 .59 .91 -2.58 .0571
IL PC-3 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness PITT PC-4	32 4.5 13.9 32 .81 .849	18 34 4.9 14.6 33 .96 592 .847	9 4.1 .46 11.1 4.1 .58 2.23 558	12 4.3 .52 12.1 4.2 .71 1.10 250	$ \begin{array}{r} 6 \\ 23 \\ 3.0 \\ \hline 13.3 \\ 23 \\ .71 \\ 1.65 \\ .875 \\ \end{array} $	15 22 3.8 17.6 21 1.61 .0515 634	5 2.2 23 11.2 2.1 .51 -3.09 47	8 2.3 .43 18.7 2.2 1.39 3.03 1.52	3 6.3 .92 14.7 6.3 .70 ERR 1.20	6 5.7 .83 14.43 5.7 .81 2.56 1.51
n	64 3.0 4.73 64 .10 1.44	18 70 11 16.1 68 1.07 3.43 1.95	9 4.4 .45 10.3 4.3 .45 0734 1.11	12 4.0 .72 17.9 4.0 1.56 295 214	6 19 1.5 7.65 19 .24 -2.45 .344	15 18 4.5 24.8 18 3.16 .802 .0791	6 1.8 .13 7.45 1.8 .23 1.3 .440	9 1.8 .14 7.63 1.8 .25 370 .631	3 .74 .033 4.51 .74 .07 ERR 1.12	6 .77 .066 8.69 .77 .31 -1.75 .540
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	$ \begin{array}{c} 100 \\ 12 \\ \hline 11.3 \\ 100 \\ .62 \\ -1.12 \\ \end{array} $	18 100 13 12.8 100 .80 -1.12 394	9 6.3 .52 8.24 6.3 .30 -1.59 .341	12 5.9 .89 15.2 5.8 1.14 703 479	6 16.5 .81 4.92 16.4 .10 1.48 -1.49	15 17 7.7 45.2 16 8.66 1.00 1.30	$ \begin{array}{r} 6 \\ 1.0 \\ .19 \\ \hline 19.0 \\ \hline 1.0 \\ \hline 1.39 \\ 1.37 \\ 1.31 \\ \end{array} $	9 1.0 .22 21.1 1.0 1.99 -1.53 .296	3 2.7 .13 5.1 2.7 .09 ERR 1.48	6 2.5 .31 12.8 2.4 .71 927 277
UT PC-6 <i>n</i>	5.2 8.16 63 .28 -1.87	14 70 12 17.2 70 1.26 .249 1.21	5 2.0 .12 6.12 2.0 .16 2.00 -1.36	8 1.9 .17 8.67 1.9 .35 .994 -1.01	3 17.3 .58 3.33 17.3 .04 ERR 1.73	9 15 3.4 21.9 15 2.48 0765 756	3 .57 .070 11.8 .56 .50 ERR 1.1	6 .54 .062 11.6 .53 .52 1.54 .837	3 .42 .023 5.34 .42 .10 ERR 1.65	6 .45 .063 13.9 .45 .79 -1.11 .760
w V PC-7 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness ND PC-8	60 11 17.5 61 1.41 805	18 70 13 18.2 70 1.74 -1.14 .042	6 11.7 .82 7.00 11.6 .20 300 .857	15 14 4.9 34.0 14 5.52 -1.33 .454	6 75 6.7 8.91 75 .34 -2.08 488	12 70 26 37.7 70 8.60 563 527	$ \begin{array}{r} 6 \\ 7 \\ 1.3 \\ 18.3 \\ \hline 7 \\ \hline 1.43 \\ -2.05 \\ .054 \\ \end{array} $	11 7 6.0 82.1 7 11.46 303 .611	3 1.27 .049 3.83 1.27 .05 ERR 586	6 .61 .21 35.6 .58 5.74 -3.2 .003
n	560 65 11.6 560 .61 -2.04	18 600 100 16.7 600 1.30 458 .641 .641	5 2.1 .22 10.4 2.1 .45 -2.37 559	8 2.2 .26 11.7 2.2 .62 130 380	6 9 <u>3.4</u> <u>38.3</u> 8 <u>7.14</u> -2.21 115	12 12 4.8 39.1 11 8.59 527 .0785	3 .79 .050 6.34 .79 .14 ERR 59	6 .7 .12 18.2 .7 1.44 -1.40 199	3 .41 .024 5.83 .41 .11 ERR 609	6 .6 .22 35.5 .6 5.73 -3.21 .00375

Appendix 2. Statistical parameters for the data in appendix 1–Continued.

Appendix 2.	Statistical	parameters for the data in appendix 1—Continued.	

	Ag HP	Ag All	Cd HP	Cd All	Sn HP	Sn All	Sb HP	Sb All	Cs HP
UF PC-1									
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness		6 .3 .17 50.2 .3 10.22 .687 1.11	$ \begin{array}{r} 6\\ .12\\ .057\\ \hline 49.7\\ .11\\ \hline 9.95\\ .664\\ 1.11\\ \end{array} $	6 .12 .057 49.7 .11 9.95 .664 1.11	3 .95 0 .95 0 ERR ERR ERR	6 1.3 .39 30.5 1.3 4.09 -3.10 .071	6 .51 .034 6.64 .51 .19 .364 156	6 .51 .034 6.64 .51 .19 .364 156	6 1.7 .20 12.0 1.7 .61 -3.16 063
WY PC-2 <i>n</i>		6 .2 .11 70.6 .1 23.26 .353 1.03	$ \begin{array}{r} 6 \\ .11 \\ .020 \\ \hline 18.5 \\ .11 \\ \hline 1.27 \\ 3.70 \\ 1.91 \\ \end{array} $	6 .11 .020 18.5 .11 1.27 3.70 1.91	3 .323 0 .323 0 ERR ERR ERR	6 .36 58.9 .5 13.95 .201 1.17	$ \begin{array}{r} 6 \\ .18 \\ .021 \\ \hline 11.6 \\ .18 \\ .58 \\ -1.62 \\397 \\ \end{array} $	6 .18 .021 11.6 .18 .00579 -1.62 397	6 .195 .010 4.86 .195 .10 -1.22 .198
IL PC-3 <i>n</i> Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	18 28.9 <u>.6</u> <u>2.96</u> ERR	6 .5 .18 36.5 .5 5.19 .333 1.14	$ \begin{array}{r} 6 \\ .7 \\ .14 \\ \hline 19.4 \\ .7 \\ \hline 1.45 \\ 1.66 \\ 1.36 \\ \end{array} $	6 .7 .14 19.4 .7 1.45 1.66 1.36	3 .71 .094 13.3 .70 .06 ERR 1.73	6 2 1.1 67.7 1 22.21 966 .809	6 .84 .062 7.45 .83 .24 1.49 882	6 .84 .062 7.45 .83 .24 1.49 882	6 1.1 .29 25.3 1.1 -3.17 .019
PITT PC-4 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	13 13 3 3 	6 .2 .14 73.3 .2 23.82 .213 1.14	6 .08 .028 34.3 .08 4.83 779 .738	6 .08 .028 34.3 .08 4.83 779 .738	3 .552 .0053 .93 .573 0 ERR 1.73	6 .9 .46 50.6 .8 10.7 .120 .954	6 .022 .022 .022 .02 .43 890 340	6 .022 10.0 .22 .43 890 340	6 .75 .061 8.16 .75 .28 -2.24 432
POC PC-5 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	23 093 40.6 <u>.22</u> <u>6.04</u> ERR	6 .16 .090 55.6 .15 12.2 .760 1.33	5 .08 .014 19.0 .074 1.61 1.119 968	6 .07 .02 30.9 .07 5.12 088 920	3 .371 0 0 .371 0 ERR ERR ERR	6 40 57 159 5 700 4.29 2.02	$ \begin{array}{r} 6 \\ .4 \\ .33.8 \\ .4 \\ .4 \\ .4.81 \\ -1.69 \\ .508 \\ \end{array} $	6 .4 .14 33.8 .4 4.81 -1.69 .508	6 .24 .024 9.91 .24 .40 .192 .876
UT PC-6 <i>n</i>	15 050 32.8 <u>.15</u> <u>4.21</u> ERR	6 .3 .18 58.7 .3 19.1 -2.11 .155	$5 \\ .07 \\ .012 \\ 17.5 \\ .07 \\ 1.19 \\ -1.52 \\ .824$	5 .07 .012 17.5 .07 1.19 -1.52 .824	3 .17 .03 15.8 .17 .89 ERR -1.73	6 .3 .12 45.8 .2 9.21 -1.99 .471	6 .10 .012 12.3 .10 .60 .00139 .967	6 .10 .012 12.3 .10 .60 .00139 .967	6 .14 .012 8.68 .14 .32 -1.55 706
WV PC-7 <i>n</i>	53 43.2 <u>1.2</u> <u>7.17</u> ERR	6 .8 .55 64.6 .7 16.9 .243 1.21	$ \begin{array}{r} 6 \\ .07 \\ .016 \\ \hline 24.9 \\ .07 \\ \hline 3.23 \\ .971 \\ -1.23 \\ \end{array} $	6 .07 .016 24.9 .07 3.23 971 -1.23	3 1.75 0 1.75 0 ERR ERR ERR	6 1.8 .18 10.1 1.8 .41 .666 .961	6 .54 .041 7.65 .54 .23 3.66 1.75	6 .54 .041 7.65 .54 .23 3.66 1.75	6 2.1 .27 12.9 2.1 .63 -3.30 011
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	 	3 .15 .062 39.7 .15 5.76 ERR .219	$ \begin{array}{r} 6\\ .043\\ .0072\\ \hline 18\\ \hline .041\\ \hline 1.58\\ 1.52\\ -1.45\\ \end{array} $	6 .043 .0072 18 .041 1.58 1.52 -1.45	3 .4 .16 43.3 .4 5.84 ERR 1.73	4 .6 .45 76.2 .5 21.9 1.96 1.53	5 .15 .010 6.80 .15 .19 906 516	6 .15 .020 12.7 .15 .64 2.45 1.33	6 .09 .012 13.0 .09 .77 1.86 -1.34

Appendix 2. Statistical parameters for the data in appendix 1–Continued.

	Ba HP	Ba All	La HP	La All	Ce HP	Ce All	Pr HP	Pr All	Nd HP	Nd All
UF PC-1	Da III	DaAn		La All	cem	CC All	11111	11 All	Num	Nu Ali
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	56 .46	21 60 13 21.1 60 2.18 .742 .855	991.215.491.08-1.64105	15 10 6.0 60.5 9 9.62 13.1 3.52	9 17 1.5 8.6 17 .35 .121 -1.23	13 20 12 52.4 20 10.3 2.09 1.75	3 1.9 .16 8.08 1.9 .23 ERR -1.73	3 1.9 .16 8.08 1.9 .23 ERR -1.73	4 7.7 .68 8.87 7.7 .31 3.90 -1.97	6 8 1.6 21.3 7 2.05 .983 402
WY PC-2 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	22 7.17	21 300 120 40.0 200 35.37 1.25 -1.30	$ \begin{array}{r} 6 \\ 4.7 \\ .82 \\ \hline 17.6 \\ 4.6 \\ \hline 1.39 \\ -1.18 \\623 \\ \end{array} $	12 5 2.0 37.3 5 5.29 4.76 2.00	$ \begin{array}{r} 8 \\ 8 \\ $	14 12 8.1 69.9 10 17.20 5.17 2.21	3 .93 .085 9.08 .93 .28 ERR 176	3 .085 9.08 .93 .28 ERR 176	3.4 0 3.4 0 8.7 0 8.7 0 8.7 0 8.7 0 8.7 0 8.7 0 8.7 0 8.7 0 8.7 0 0 8.7 0 0 8.7 0 0 8.7 0 0 8.7 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	3 3.4 0 3.4 0 ERR ERR
IL PC-3 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	82 .54	21 90 23 26.1 90 3.31 0928 .519	5 6.0 .22 3.50 6.0 .05 -3.33 604	12 9 4.1 45.1 8 8.97 829 .881	9 11 2.7 25.6 10 3.46 -1.32 822	13 20 16 81.6 20 27.17 4.30 2.05	3 1.4 .17 14.1 1.3 .71 ERR -1.73	3 1.4 .17 14.1 1.3 .71 ERR -1.73	3 4.9 0 4.9 0 ERR ERR	4 5.6 .81 14.5 5.5 .82 4.00 2.00
PITT PC-4 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	40 4.5	21 40 12 29.0 40 3.49 1.80 .888	$ \begin{array}{r} 8 \\ 5.1 \\ .86 \\ \hline 16.7 \\ 5.1 \\ \hline -2.12 \\ .541 \\ \end{array} $	14 6 5.1 84.8 5 24.5 11.0 3.12	$9 \\ 10 \\ 1.3 \\ 12.5 \\ 10 \\ .77 \\ 2.36 \\ -1.32$	12 12 5.2 43.9 11 7.32 3.48 1.84	3 1.12 .058 5.09 1.12 .08 ERR 1.73	3 1.12 .058 5.09 1.12 .08 ERR 1.73	3 4.3 .52 12 4.3 .51 ERR 1.73	6 4.1 .50 13.5 4.1 .81 127 956
POC PC-5 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	14 7.17 190 .24	21 200 130 65.0 100 35.8 1.25 .894	$ \begin{array}{r} 6 \\ 6.2 \\ \hline 10.1 \\ 6.2 \\ .43 \\ -2.17 \\345 \end{array} $	15 5 2.6 47.5 5 17.3 548 287	$ \begin{array}{r} 8 \\ 11 \\ 1.6 \\ 14.8 \\ 11 \\ \hline 1.10 \\ .967 \\ -1.45 \\ \end{array} $	15 11 3.5 33.4 10 6.19 1.74 .331	3 1.31 .090 6.19 1.31 .13 ERR 722	3 1.31 .090 6.19 1.31 .13 ERR 722	3 5.0 .29 5.81 5.0 .11 ERR 1.73	6 5.5 .77 13.9 5.5 .81 1.86 1.36
UT PC-6 <i>n</i> Standard Dev (ppm) Standard Dev (ppm) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness WV PC-7	2.0 5.62 35 .14	14 34 3.7 10.8 34 .55 164 0660	$ \begin{array}{r} 6 \\ 2.9 \\ .50 \\ \hline 17.3 \\ 2.9 \\ \hline 1.31 \\ -2.95 \\ .390 \\ \end{array} $	12 5 3.1 67.7 4 18.1 .200 1.38	6 4.7 .14 3.02 4.7 .04 065 846	12 8 4.8 61.2 7 13.9 2.89 1.79	3 .55 .023 4.30 .55 .06 ERR -1.73	6 .48 .069 14.5 .48 .90 -2.35 0511	$5 \\ 2.1 \\32 \\ 18.3 \\ 2.1 \\ 1.41 \\ -2.05 \\543 \\ \end{array}$	6 2.2 .44 21.1 2.2 1.96 620 056
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness ND PC-8	$ \begin{array}{r} 120\\ 17\\ 14.1\\ 120\\ .78\\ \end{array} $	18 140 37 26.1 130 2.82 1.28 1.52	9 19 2.4 12.6 19 .72 914 .018	15 17 3.0 17.8 17 1.51 928 .121	9 34 3.5 10.2 34 .48 614 678	15 40 11 30.0 40 3.74 1.59 1.34	3 4.1 .21 4.78 4.1 .09 ERR 077	5 3 1.2 38.1 3 7.08 -3.17 546	$ \begin{array}{r} 6 \\ 14 \\ 1.6 \\ \hline 12.0 \\ 13 \\ .56 \\ -1.535 \\ .358 \\ \end{array} $	9 15 3.1 20.4 15 1.86 -1.67 .355
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	100 20.0 500 1.66 729	1860021035.0600 $4.93.4341.25$	$ \begin{array}{r} 6 \\ 2.5 \\ .51 \\ \hline 21.4 \\ \hline 2.4 \\ \hline 1.97 \\ -3.31 \\ .005 \\ \end{array} $	12 4 2.5 63.2 4 16.3 1.37 1.47	6 4.3 .38 9.16 4.3 .34 -1.70 .482	9 9 7.4 82.7 7 32.8 -1.41 .924	3 .43 .058 14.0 .43 .62 ERR 1.73	3 .43 .058 14.0 .43 .62 ERR 1.73	$5 \\ 1.8 \\ .57 \\ 30.5 \\ 1.8 \\ 4.9 \\ 2.4 \\ -1.2 \\ $	5 1.8 .57 30.5 1.8 4.9 2.4 -1.2

Appendix 2. Statistical parameters for the data in appendix 1–Continued.

	Sm HP	Sm All	Eu HP	Eu All	Tb HP	Yb HP	Yb All	Hf HP	Ta HP
UF PC-1									
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness WY PC-2	6 1.8 .24 14.3 1.7 .87 -2.89 132	6 1.8 .24 14.3 1.7 .87 -2.89 132	6 .37 .035 9.63 .37 .39 -2.17 217	6 .37 .035 9.63 .37 .39 -2.17 217	6 .248 .0190 7.66 .247 .25 .088 026	6 .85 .073 8.6 .85 .32 -1.63 61	9 .85 .059 6.97 .84 .22 52 45	6 .8 .11 14.5 .7 .86 -1.12 .617	6 .21 .020] 9.30 .21 .37 -1.56 24
w Y PC-2 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness IL PC-3	6 8 .15 18.0 8 <u>1.39</u> -2.97 133	6 .8 .15 18.0 .8 1.39 -2.97 133	6 .18 .021 12.0 .18 .64 -1.37 934	6 .18 .021] 12.0 .18 .64 -1.37 934	6 .11 .009 8.96 .11 .34 -2.08 206	6 .40 .048 12.2 .39 .66 54 54 848	9 .36 .071 19.6 .36 1.92 23 633	6 .64 .053 8.17 .64 .28 .108 288	6 .12 .019 14.8 .13 .98 .270 694
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness PITT PC-4	6 1.1 .15 13.6 1.1 .91 4.90 -2.16	6 1.1 .15 13.6 1.1 .91 4.90 -2.16	6 .21 .021 10.4 .21 .47 -1.37 991	6 .21 .021 10.4 .21 .47 -1.37 991	6 .14 .011 8.0 .14 .27 173 586	6 .52 .015 2.85 .52 .03 4.15 1.91	9 .56 .061 10.9 .55 .50 .670 1.31	$ \begin{array}{r} 6 \\ .7 \\ .15 \\ \hline 22.4 \\ .6 \\ \hline 1.95 \\ 1.38 \\ 1.26 \\ \end{array} $	6 .18 .013] 7.04 .18] .23 2.79 -1.56
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 1.0 .15 16.0 .9 1.13 -1.83 518	6 1.0 .15 16.0 .9 1.13 -1.83 518	6 .19 .025 12.0 .19 .60 -2.74 .033	6 .19 .025] 12.0 .19 .60 -2.74 .033	6 .12 .011 8.87 .12 .33 -2.48 371	6 .450 .0243 5.14 .449 .11 2.04 11	9 .48 .063 13.08 .48 .70 4.57 1.95	$ \begin{array}{r} 6 \\ .56 \\ .062 \\ \hline 11.2 \\ .56 \\ .52 \\ -2.64 \\ .018 \\ \end{array} $	6 .15 .019 12.4 .15 .65 -2.86 108
POC PC-5 <i>n</i>	6 1.1 .13 12.0 1.1 .62 -2.59 309	9 1.0 .20 20.6 1.0 2.01 -1.45 147	6 .22 .015 6.61 .22 .18 -1.84 .660	6 .015 6.61 .22 .18 -1.84 .660	6 .165 .0074 5.39 .164 .12 -1.09 52	6 .56 .012 2.11 .56 .02 22 635	9 .53 .059 11.24 .52 .60 805 964	6 .073 14.5 .50 .87 -2.73 .217	6 .12 .011] 9.48 .11 .40 802 .15
UT PC-6 <i>n</i>	6 .44 .076 17.3 .43 1.26 -3.27 .0210	9 .069 16.7 .41 1.17 -1.65 .852	6 .092 .0086 9.54 .091 .38 -1.15 585	7 .094 .0105 11.31 .094 .56 37 07	6 .056 .0051 9.01 .056 .33 .395 1.04	6 .20 .008 3.91 .20 .06 509 429	9 .009 4.35 .20 .08 008 .067	6 .52 .053 10.3 .52 .43 57 .72	6 .052 .0087 16.7 .052 1.28 .02 -1.08
WV PC-7 n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness ND PC-8	6 3.2 .40 13.1 3.1 .70 -3.03 098	8 .76 27.1 2.7 3.82 -1.03 644	6 .64 .036 5.75 .64 .14 -1.31 .416	9 .6 .12 19.6 .6 1.67 1.74 .76	6 .398 .020 5.08 .397 .11 .689 .318	6 1.6 .12 7.49 1.6 .24 -1.27 256	9 1.6 .15 9.59 1.6 .403 -1.32 .362	6 2.1 .28 12.6 2.1 .68 -3.22 049	6 .63 .043] 6.99 .63 .20 .190 49
n	6 .33 .10 28.4 .34 4.11 869 559	7 .35 .10 28.4 .34 4.11 869 559	6 .07 .015 22.4 .07 2.10 -2.52 .138	6 .07 .015] 22.4 .07] 2.10 -2.52 .138	$ \begin{array}{r} 6 \\ .055 \\ .0044 \\ \hline 8.06 \\ .055 \\ .26 \\ 1.05 \\ 1.43 \\ \end{array} $	6 .26 .041 15.9 .26 1.02 -1.18 .548	9 .24 .043 18.0 .24 1.36 237 .854	$ \begin{array}{c} 6 \\ .38 \\ .046 \\ \hline 12.4 \\ .38 \\ .63 \\ -1.95 \\ .41 \\ \end{array} $	3 .093 .0040] 4.36 .093 .06 ERR -1.73

	W HP	W All	РЬ НР	Pb All	Bi HP	Bi All	Th HP	Th All	U HP
UF PC-1									
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Means (%) Dev of Means (%) Kurtosis Skewness WY PC-2	$ \begin{array}{r} 6 \\ 1.2 \\ .13 \\ \hline 1.1 \\ .48 \\ -1.56 \\ .56 \\ \end{array} $	6 1.2 .13 11.0 1.1 .48 -1.56 .56	6 7.5 .33 4.39 7.5 .08 1.34 -1.22	12 8 1.5 18.4 8 1.54 902 .469	3 .117 .0058 4.95 .117 .08 ERR -1.73	3 .117 .0058 4.95 .117 .08 ERR -1.73	$ \begin{array}{r} 6\\ 2.5\\ .48\\ \hline 19.2\\ 2.5\\ 1.96\\ 4.82\\ -2.12\\ \end{array} $	6 2.5 .48 19.2 2.5 1.96 4.82 -2.12	6 1.0 .18 18.4 1.0 1.55 -2.91 .102
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 .42 .055 12.7 .42 .66 -1.68 .535	6 .42 .055 12.7 .42 .66 -1.68 .535	$ \begin{array}{r} 6 \\ 2.3 \\ .75 \\ \hline 3.2 \\ 2.2 \\ \hline 4.72 \\ -3.01 \\ .122 \\ \end{array} $	9 2.6 .90 34.3 2.5 5.84 -1.03 .096	3 .053 .0034 6.45 .053 .14 ERR -1.73	3 .053 .0034 6.45 .053 .14 ERR -1.73	$ \begin{array}{r} 6 \\ 1.6 \\ .25 \\ \hline 16.2 \\ 1.6 \\ .95 \\ 2.21 \\ 1.17 \\ \end{array} $	6 1.6 .25 16.2 1.6 .95 2.21 1.17	6 .58 .054 9.35 .58 .35 -2.66 -4.6×10 ⁻¹⁵
IL PC-3 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness PITT PC-4	6 1.73 .256 14.4 1.69 .82 637 .929	6 1.73 .256 14.4 1.69 .82 637 .929	$ \begin{array}{r} 6 \\ 7 \\ 1.1 \\ \hline 15.6 \\ 7 \\ \hline 1.04 \\ -2.91 \\019 \\ \end{array} $	12 9 3.0 32.6 9 4.59 637 .885	2 .088 .0011 1.61 .088 .01 ERR ERR	2 .088 .0011 1.61 .088 .01 ERR ERR	6 2.1 .19 9.28 2.1 .33 3.31 1.81	6 2.1 .19 9.28 2.1 .33 3.31 1.81	$\begin{array}{r} 6 \\ 5 \\ 1.2 \\ \hline 23.1 \\ 5 \\ 2.27 \\ -2.70 \\ .164 \end{array}$
n Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 .82 .049 5.82 .82 .15 80 .43	6 .82 .049 5.82 .82 .15 80 .43	6 3.4 .50 15.0 3.3 .94 -3.14 .066	12 5 1.8 40.1 4 6.86 680 .962	3 .117 .0053 4.95 .116 .08 ERR -1.73	5 .8 .91 118 .3 125 -2.36 .776	6 1.5 .15 9.52 1.5 .36 1.66 1.34	6 1.5 .15 9.52 1.5 .36 1.66 1.34	6 .5 .10 21.4 .5 2.02 -2.37 228
POC PC-5 <i>n</i>	$ \begin{array}{r} 6 \\ 1.0 \\ .16 \\ \hline 96 \\ \hline -1.17 \\ .281 \\ \end{array} $	6 1.0 .16 16.7 .96 1.17 -1.17 .281	$\begin{array}{r} 6 \\ 2.1 \\ .40 \\ \hline 19.3 \\ 2.0 \\ \hline 1.68 \\ -2.96 \\064 \end{array}$	12 5 3.9 78.5 4 29.52 0035 1.17	3 .051 .0022 4.06 .051 .06 ERR -1.29	3 .051 .0022 4.06 .051 .06 ERR -1.29	6 1.16 .043 3.71 1.15 .06 -1.48 809	6 1.16 .043 3.71 1.15 .06 -1.48 809	6 .57 .071 12.5 .57 .68 -1.81 400
UT PC-6 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness	6 .45 .062 13.53 .450 .78 -2.85 182	6 .45 .062 13.53 .450 .78 -2.85 182	5 1.52 .084 5.50 1.52 .12 612 512	8 1.7 .25 15.0 1.7 .94 333 .968	3 .035 .0007 1.63 .035 .01 ERR 1.73	3 .035 .0007 1.63 .035 .01 ERR 1.73	6 .60 .025 4.13 .60 .07 3.37 -1.70	6 .60 .025 4.13 .60 .07 3.37 -1.70	6 .79 .056 7.08 .78 .21 674 502
WV PC-7 <i>n</i> Mean (ppm) Standard Dev (ppm) Rel Std Dev (%) Geometric Mean (ppm) Dev of Means (%) Kurtosis Skewness ND PC-8	$5 \\ 1.7 \\ .52 \\ \hline 31.3 \\ 1.62 \\ 3.56 \\ 2.29 \\ 1.42 \\ \hline$	6 4 6.5 151.0 2 80.2 5.89 2.42	$\begin{bmatrix} 6\\12.1\\0\\12.1\\0\\12.1\\0\\ERR\\ERR\\ERR \end{bmatrix}$	9 13 2.2 16.4 13 1.06 1.94 1.72	3 0 0 23 0 ERR ERR ERR	3 .23 0 .23 0 ERR ERR	6 6.4 .15 2.33 6.4 .02 -1.33 315	6 6.4 .15 2.33 6.4 .02 -1.33 315	6 1.9 .29 15.3 1.9 .98 -3.09 .0423
n	$ \begin{array}{r} 6 \\ .36 \\ .060 \\ \hline 16.8 \\ .35 \\ \hline 1.34 \\ 2.21 \\ -1.10 \\ \end{array} $	6 .060 16.8 .35 1.34 2.21 -1.10	6 1.5 <u>.34</u> 23.0 1.4 2.72 3.07 -1.20	9 1.8 .67 35.8 1.7 6.08 .034 .564	0 ERR ERR ERR ERR ERR ERR ERR	0 ERR ERR ERR ERR ERR ERR ERR	3 1.1 .04 3.79 1.1 .05 ERR .72	6 .88 .21 .18 .23 .86 -2.20 228	5 .46 .04 7.87 .46 .24 1.9 .544

Appendix 2. Statistical parameters for the data in appendix 1–Contin	nued.
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