

Mineral Resources Program

Critical Minerals in Orogenic (Gold) and Coeur d'Alene-Type Mineral Systems of the United States



Data Report 1198

Cover. High-grade gold ore from the Sixteen to One mine, Alleghany district, California. (Photograph by Erin Marsh, U.S. Geological Survey)

Critical Minerals in Orogenic (Gold) and Coeur d'Alene-Type Mineral Systems of the United States

By Ryan D. Taylor and Albert H. Hofstra

Mineral Resources Program

Data Report 1198

**U.S. Department of the Interior
U.S. Geological Survey**

U.S. Geological Survey, Reston, Virginia: 2025

For more information on the USGS—the Federal source for science about the Earth, its natural and living resources, natural hazards, and the environment—visit <https://www.usgs.gov> or call 1–888–392–8545.

For an overview of USGS information products, including maps, imagery, and publications, visit <https://store.usgs.gov/> or contact the store at 1–888–275–8747.

Any use of trade, firm, or product names is for descriptive purposes only and does not imply endorsement by the U.S. Government.

Although this information product, for the most part, is in the public domain, it also may contain copyrighted materials as noted in the text. Permission to reproduce [copyrighted items](#) must be secured from the copyright owner.

Suggested citation:

Taylor, R.D., and Hofstra, A.H., 2025, Critical minerals in orogenic (gold) and Coeur d’Alene-type mineral systems of the United States: U.S. Geological Survey Data Report 1198, 47 p., <https://doi.org/10.3133/dr1198>.

Associated data for this publication:

Granitto, M., Hofstra, A.H., and Taylor, R.D., 2025, National Geochemical Database on Ore Deposits—New data featuring fusion digestion analytical methods: U.S. Geological Survey data release, <https://doi.org/10.5066/P1M9HNMW>.

ISSN 2771-9448 (online)

Contents

Abstract.....	1
Introduction.....	1
Purpose and Scope	1
Background—Mineral Systems and Ore Deposits.....	2
Occurrence of Orogenic and Coeur d’Alene-Type Mineral Systems.....	2
Formation of Orogenic Mineral Systems	8
Formation of Coeur d’Alene-type Mineral Systems.....	12
Mining and Beneficiation of Ore Deposits and Materials	14
Methods.....	16
Results	16
Geochemistry of Orogenic Gold Deposits	17
Geochemistry of Coeur d’Alene-type Deposits	21
Discussion—Potential Recovery of Critical Minerals.....	25
Conclusions.....	34
References Cited.....	35
Appendix 1. Descriptions for Coeur d’Alene-Type and Orogenic Gold Ore Samples Used in this Study.....	40

Figures

1. Map showing the locations of orogenic and Coeur d’Alene-type mineral systems in the continental United States	3
2. Simplified schematic model for a mineral system, including metamorphic shear zone hydrothermal systems	11
3. Schematic model of an orogenic system	11
4. Photomicrographs and scans of orogenic gold ore showing the residence of critical minerals	13
5. Graph showing the median concentrations of elements in orogenic gold ore with gold concentrations greater than or equal to 0.5 parts per million	17
6. Log-log plots of gold plotted against select critical minerals	20
7. Graph showing the median concentrations of elements in silver-rich Coeur d’Alene-type ore	22
8. Graph showing the median concentrations of elements in zinc-rich Coeur d’Alene-type ore	25
9. Log-log plots of silver plotted against select critical minerals in ore from Coeur d’Alene-type deposits	28
10. Log-log plots of lead plotted against select critical minerals in ore from Coeur d’Alene-type deposits	30
11. Log-log plots of zinc plotted against select critical minerals in ore from Coeur d’Alene-type deposits	32
12. Log-log plots of the major commodities in ore from Coeur d’Alene-type deposits	33
13. Log-log plots displaying positive correlations between silver and copper and arsenic and antimony in silver-rich veins from Coeur d’Alene-type deposits	33
14. Log-log plots of silver plotted against copper and arsenic plotted against antimony for tetrahedrite from Coeur d’Alene-type deposits	34

Tables

1. Examples of orogenic gold deposits in the United States	4
2. Examples of Coeur d'Alene-type deposits in the United States	9
3. Ore minerals containing critical mineral metals	14
4. Correlation coefficients of select metals commonly enriched in Coeur d'Alene-type deposits using all the samples compiled in the study	15
5. Basic statistics on gold-rich ore from orogenic gold deposits.....	18
6. Correlation coefficients of select metals commonly enriched in orogenic gold deposits using all samples compiled in this study	21
7. Basic statistics on silver-rich ore from Coeur d'Alene-type deposits.....	23
8. Basic statistics on zinc-rich ore from Coeur d'Alene-type deposits	26
9. Estimated tonnes of contained gold, arsenic, cobalt, antimony, tellurium, and tungsten from the Idaho-Maryland mine in Grass Valley, California	29
10. Estimated tonnes of manganese, antimony, and arsenic released in mine waste from silver-rich ore from the Coeur d'Alene area	31

Conversion Factors

U.S. customary units to International System of Units

Multiply	By	To obtain
Length		
foot (ft)	0.3048	meter (m)
mile (mi)	1.609	kilometer (km)
inch (in.)	25,400	micrometer (μm)
Mass		
ounce, avoirdupois (oz)	28.35	gram (g)
pound, avoirdupois (lb)	0.4536	kilogram (kg)
ton, short (2,000 lb)	0.9072	metric ton, tonne (t)

International System of Units to U.S. customary units

Multiply	By	To obtain
Length		
meter (m)	3.2808	foot (ft)
kilometer (km)	0.6214	mile (mi)
micrometer (μm)	0.0000394	inch (in.)
Mass		
gram (g)	0.0353	ounce, avoirdupois (oz)
kilogram (kg)	2.2046	pound, avoirdupois (lb)
metric ton, tonne (t)	1.1023	ton, short (2,000 lb)

Temperature in degrees Fahrenheit ($^{\circ}\text{F}$) may be converted to degrees Celsius ($^{\circ}\text{C}$) as follows:

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) / 1.8.$$

Temperature in degrees Celsius ($^{\circ}\text{C}$) may be converted to degrees Fahrenheit ($^{\circ}\text{F}$) as follows:

$$^{\circ}\text{F} = (1.8 \times ^{\circ}\text{C}) + 32.$$

Supplemental Information

Concentrations of chemical constituents are in parts per million (ppm), grams per tonne (g/t), or weight percent (wt. pct.).

Abbreviations

CM	critical mineral
ppm	parts per million
USGS	U.S. Geological Survey

Critical Minerals in Orogenic (Gold) and Coeur d'Alene-Type Mineral Systems of the United States

By Ryan D. Taylor and Albert H. Hofstra

Abstract

Orogenic and Coeur d'Alene-type mineral systems are produced by metamorphic devolatilization of thick volcanic or siliciclastic sedimentary rock sequences and the focused flow of hydrothermal fluids upwards along crustal-scale faults. Most orogenic systems are found along the Cordilleran orogen, stretching from California northwards into Alaska, whereas most Coeur d'Alene-type systems occur in the Proterozoic Belt Basin in Idaho and Montana. Although the deposit types in these systems are exploited for precious and base metals, potential exists for the production of critical minerals necessary for current (2025) societal needs in the United States. Publicly available geochemical data compiled for these mineral systems, coupled with mineralogical characteristics, indicate that several critical minerals could potentially be recovered from unmined resources and processed mine waste: arsenic, antimony, tellurium, cobalt, and tungsten from orogenic gold deposits and zinc, antimony, arsenic, and manganese from Coeur d'Alene-type systems. These critical minerals reside primarily in arsenopyrite (arsenic), scheelite (tungsten), siderite (manganese), sphalerite (zinc), tetrahedrite (antimony and arsenic), stibnite (antimony), and telluride (tellurium) minerals.

Introduction

Meeting the growing demand for commodities used in today's (2025) advanced technologies is a challenge for the global economy. Critical minerals are commodities (typically elements, but also compounds) deemed vital to the economic or national security of individual countries, and many are vulnerable to supply disruption (U.S. Department of the Interior, 2017; U.S. Department of Energy, 2021; U.S. Geological Survey [USGS], 2022a).

Most critical minerals are not the principal commodities produced from ore deposits. Rather, they are typically extracted as byproducts of principal commodity production (Nassar and others, 2015). The endowment of critical mineral byproducts in ore deposits can be quantified if the tonnage, grade, and concentration of the principal commodities and

critical minerals have been determined in a representative set of ore samples (for example, Werner and others, 2017). Unfortunately, geochemical data on the concentrations of critical minerals in ore are commonly lacking, are too few to be representative, or have not been compiled from previous studies. Therefore, the USGS Systems Approach to Critical Minerals Inventory, Research, and Assessment Project set out to fill this data gap by submitting ore samples from dozens of ore deposits for analysis of more than 60 elements, gathering legacy multielement geochemical data on ore samples from previous studies, and publishing the data (Granitto and others, 2020, 2021, 2025).

Analyses from individual deposits in the resulting databases (Granitto and others, 2020, 2021, 2025) were used to calculate critical mineral–principal commodity mass ratios, which can be multiplied by the tonnes of the principal commodity to calculate the approximate tonnes of a critical mineral in a deposit. Critical mineral grades can also be calculated from principal commodity grades using a similar method. The same calculation can also be extrapolated from individual deposits and applied to models for mineral deposit types. However, the quantity of geochemical data (Granitto and others, 2020, 2021, 2025) upon which these critical mineral tonnage and grade calculations are based is generally insufficient to be considered representative, so the endowments of critical minerals obtained are simply rough estimates with unquantified uncertainties. Nevertheless, these estimates provide an indication of which deposits and deposit types are likely to harbor significant quantities of critical minerals that could potentially be recovered, which may prompt economic assessments and development of metallurgical methods for their recovery.

Purpose and Scope

Throughout the last several years, the U.S. Government has taken several important steps to identify and ensure the supply of critical minerals. In 2017, the President issued Executive Order 13817 titled “A Federal Strategy to Ensure Secure and Reliable Supplies of Critical Minerals.” The Executive Order directed the USGS to develop a plan to improve the Nation's understanding of domestic critical

mineral resources. To implement Executive Order 13817, the Secretary of the Interior issued Secretarial Order 3359 titled “Critical Mineral Independence and Security.” In response, the USGS National Minerals Information Center identified the following list of 35 critical minerals with a high risk for supply disruption: aluminum (bauxite), antimony, arsenic, barite (BaSO_4), beryllium, bismuth, cesium, chromium, cobalt, fluor spar (CaF_2), gallium, germanium, graphite (natural), hafnium, helium, indium, lithium, magnesium, manganese, niobium, platinum group metals, potash (potassium-bearing compounds), rare earth elements group, rhenium, rubidium, scandium, strontium, tantalum, tellurium, tin, titanium, tungsten, uranium, vanadium, and zirconium (Fortier and others, 2018).

In 2018, the U.S. Congress allocated funds to the USGS Mineral Resources Program for the Earth Mapping Resources Initiative to generate topographic, geologic, geochemical, and geophysical maps and data that are needed to increase the inventory of domestic critical minerals (Day, 2019). The USGS Mineral Resources Program project titled “Systems Approach to Critical Minerals Inventory, Research, and Assessment Project” began on October 1, 2018, and is one of several projects that were designed to advance knowledge of critical minerals, including their abundance in the Nation’s ore deposits and mine waste, their sources, the processes that concentrate them in mineral systems, and how to assess their potential in known mining districts and frontier areas.

In 2020, the President determined that the undue reliance of the United States on critical mineral imports, in processed or unprocessed form, constitutes an unusual and extraordinary threat. As a result, the President declared a national emergency and issued Executive Order 13953 titled “Addressing the Threat to the Domestic Supply Chain From Reliance on Critical Minerals From Foreign Adversaries and Supporting the Domestic Mining and Processing Industries.” The 2020 Executive Order directed the U.S. Department of the Interior (which includes the USGS) to investigate the Nation’s undue reliance on critical mineral imports and to provide recommendations for executive action. On January 15, 2021, the U.S. Department of Energy announced the establishment of a Division of Minerals Sustainability to enable the ongoing transformation of the U.S. energy system and help secure a U.S. critical minerals supply chain (U.S. Department of Energy, 2021).

On May 7, 2021, as required by Section 7002 (“Mineral Security”) of Title VII (“Critical Minerals”) of the Energy Act of 2020 (Public Law 116–260, 134 Stat. 2419), the USGS National Minerals Information Center published an updated evaluation on the methodology used to evaluate mineral criticality and a revised draft list of critical minerals (Nassar and Fortier, 2021). Most recently, on February 22, 2022, a final list of 50 critical minerals was released after an extensive multiagency assessment and public comments (aluminum, antimony, arsenic, barite, beryllium, bismuth, cerium, cesium, chromium, cobalt, dysprosium, erbium, europium, fluor spar, gadolinium, gallium, germanium, graphite, hafnium, holmium,

indium, iridium, lanthanum, lithium, lutetium, magnesium, manganese, neodymium, nickel, niobium, palladium, platinum, praseodymium, rhodium, rubidium, ruthenium, samarium, scandium, tantalum, tellurium, terbium, thulium, tin, titanium, tungsten, vanadium, ytterbium, yttrium, zinc, and zirconium; USGS, 2022a). This report addresses the President’s and Secretary of the Interior’s orders by filling the knowledge gap on the abundance of critical minerals in orogenic and Coeur d’Alene-type mineral systems, particularly for orogenic gold and Coeur d’Alene-type polymetallic sulfide and antimony deposits. Geochemical data on ore and altered wall rock are used with information on production from and resources in each deposit type to estimate the amount of critical minerals present in processed mine waste resulting from previous mining activities and unmined resources. This report also discusses resources that could potentially be recovered in the future.

Background—Mineral Systems and Ore Deposits

Ore deposits are generated by mineral systems (for example, orogenic and Coeur d’Alene-type). Mineral systems operate in response to geodynamics, and sometimes climatic conditions, and generate one or more deposit types that are genetically related to one another (for example, Hofstra and Kreiner, 2020, and references therein). Metamorphic shear zone hydrothermal processes can produce orogenic and Coeur d’Alene-type mineral systems. Orogenic mineral systems can generate orogenic gold, orogenic antimony, orogenic mercury, and orogenic graphite deposits, of which orogenic gold is the most economically significant and likely to produce a mine. Coeur d’Alene-type mineral systems produce Coeur d’Alene-type polymetallic sulfide, antimony, and uranium deposits. By comparing the abundance of critical minerals in each deposit type in a system, we can learn where specific critical minerals are concentrated. This systems approach can, therefore, aid exploration and assessments of critical minerals in the United States and may eventually help mitigate the reliance on imports from foreign sources.

Occurrence of Orogenic and Coeur d’Alene-Type Mineral Systems

Both orogenic and Coeur d’Alene-type mineral systems are typically generated along dilatant structures during exhumation and dewatering of crustal metamorphic zones (Hofstra and Kreiner, 2020, and references therein). However, orogenic mineral systems occur in more reduced and sulfidic metamorphic zones that locally contain volcanogenic massive sulfide deposits and generate gold, antimony, mercury, and graphite deposits. Coeur d’Alene-type mineral systems occur in more oxidized metamorphic zones that locally contain

basin brine path zinc-lead and copper deposits and generate polymetallic sulfide, antimony, and uranium deposits (Hofstra and Kreiner, 2020, and references therein).

Orogenic mineral systems have also been referred to as “orogenic gold systems,” “deformation and metamorphism mineral systems,” and “metamorphic-hydrothermal mineral systems.” Orogenic gold deposits represent one of the most important lode sources for gold in the world (Goldfarb and others, 2001; Frimmel, 2008). In the United States, most orogenic gold deposits are located within accreted terranes of the western cordillera of North America (fig. 1). The highest concentration of orogenic gold deposits is in Alaska, California, and Oregon (table 1), but deposits are also in Alabama, Arizona, Georgia, Idaho, Michigan, and South Dakota (in the Homestake mine). Low-sulfide gold-quartz veins in North Carolina, Virginia, and Wyoming may also be orogenic gold deposits, but further studies are needed to confirm their classification.

The abundant placer gold that fueled the California gold rush from 1848 to 1855 was derived from large, eroded orogenic gold deposits. Orogenic gold deposits were commonly identified as mesothermal or hypothermal gold

deposits in the first half of the 20th century, emphasizing the inferred depth and temperature of deposit formation (Lindgren, 1933). More recently, orogenic gold deposits have been called low-sulfide gold-quartz veins for Phanerozoic examples and Homestake gold deposits for Phanerozoic to Archean types (Cox and Singer, 1986). Some have been called turbidite-hosted gold deposits (Keppie and others, 1986) or characterized by their location, such as Mother Lode gold (not shown in fig. 1; Goldfarb and others, 2005).

Coeur d’Alene-type polymetallic sulfide deposits are important sources of silver, lead, and zinc in the United States. The Coeur d’Alene-type mineral systems are named after the Coeur d’Alene district in Idaho and Montana, which is one of the largest silver producers in the world (Long, 1998b). These systems are also referred to as “mesothermal base and precious metal systems” (Harrison and others, 1986), “silver-lead-zinc veins in clastic metasedimentary terranes” (Beaudoin and Sangster, 1992), and “orogenic base-metal systems” (Skirrow and others, 2013). Although Coeur d’Alene-type mineral systems concentrate different commodities, they share broad geometric similarities with large kilometer-scale ore shoots.

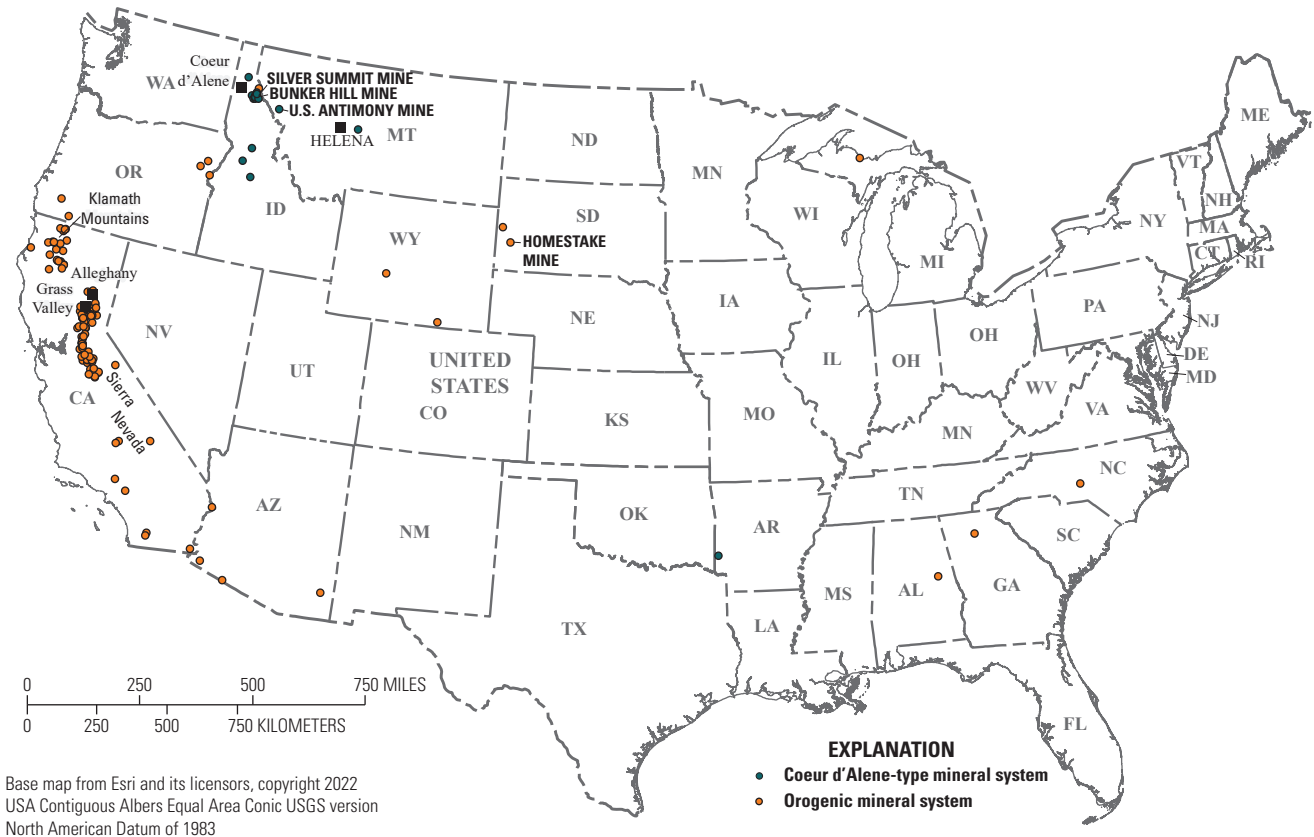


Figure 1. Map showing the locations of orogenic and Coeur d’Alene-type mineral systems in the continental United States.

Table 1. Examples of orogenic gold (Au) deposits in the United States.

[Modified from Long and others (1998), and references therein. —, no data; Unk, unknown; W.Y.O.D., Work Your Own Diggings; >, greater than; ≈, almost equal to. Abbreviations of U.S. States as follows: Ala., Alabama; Ariz., Arizona; Calif., California; Ga., Georgia; Mich., Michigan; N.C., North Carolina; Oreg., Oregon; S. Dak., South Dakota; Wyo., Wyoming]

Deposit name	State	Longitude	Latitude	Past production ore tonnage (short tons)	Troy ounces of gold produced	Other production
Alaska-Juneau	Alaska	−134.341	58.301	100,600,000	3,413,453	—
Banjo	Alaska	−150.733	63.567	13,653	6,260	Arsenic, tungsten
Big Hurrah	Alaska	−164.24	64.651	50,000	27,000	—
Bluff	Alaska	−163.736	64.579	Unk	Unk	—
Chichagoff-Hirst-Chichagoff	Alaska	−136.097	57.664	740,000	791,000	—
Cleary Hill	Alaska	−147.439	65.067	220,000	281,000	—
Dawson	Alaska	−132.705	55.471	Unk	15,814	—
Dolomi Area	Alaska	−132.07	55.144	Unk	9,645	—
Donlin Creek	Alaska	−158.184	62.054	Unk	Unk	—
Grant	Alaska	−147.957	64.882	118,684	18,393	—
Herbert Glacier	Alaska	−134.686	58.831	Unk	Unk	—
Hi Yu	Alaska	−147.282	65.075	70,000	110,000	—
Ivanhoe	Alaska	−135.103	58.882	3,500	340	—
Kensington-Jualin	Alaska	−135.087	58.866	3,500,000	792,803	—
Pogo	Alaska	−144.914	64.453	Unk	Unk	—
Rock Creek	Alaska	−165.41	64.61	811,810	Unk	—
Ryan Lode	Alaska	−147.986	64.868	328,600	19,521	—
Saddle	Alaska	−165.39	64.591	Unk	Unk	—
TMC1 (Rainbow Hill)	Alaska	−147.272	62.188	Unk	Unk	—
Treadwell	Alaska	−134.382	58.267	29,000,000	3,200,000	—
True North	Alaska	−147.56	65.053	11,740,000	Unk	—
Willow Creek-Independence- Lucky Shot-War Baby	Alaska	−149.5	61.75	550,000	606,400	—
Hog Mountain	Ala.	−85.851	33.076	187,000	24,300	Bismuth, manga- nese, arsenic
Fortuna	Ariz.	−114.332	32.553	153,000	131,000	—
Gold Prince	Ariz.	−109.583	32.194	66,749	22,280	—
Rasmussen	Ariz.	−113.35	32.074			—
Alabama-California	Calif.	−121.157	38.844	397,000	76,985	—
Allison Ranch	Calif.	−121.06	39.18	>94,000	165,000	—
Alto	Calif.	−120.605	37.894	820,000	57,000	—
American Girl	Calif.	−114.788	32.856	Unk	Unk	—
Angels	Calif.	−120.546	38.076	>517,000	>157,000	—
Argonaut	Calif.	−120.785	38.363	2,715,000	940,159	—
Bandarita	Calif.	−120.032	37.691	Unk	74,000	—
Beebe (Woodside-Eureka)	Calif.	−120.835	38.911	Unk	97,000	—
Big Canyon	Calif.	−120.904	38.611	> 438,000	107,000	—
Big Horn	Calif.	−117.716	34.225	31,000	4,675	—
Black Bear	Calif.	−123.152	41.259	Unk	>145,000	—
Black Oak	Calif.	−120.276	37.978	Unk	169,000	—

Table 1. Examples of orogenic gold (Au) deposits in the United States.—Continued

[Modified from Long and others (1998), and references therein. —, no data; Unk, unknown; W.Y.O.D., Work Your Own Diggings; >, greater than; ≈, almost equal to. Abbreviations of U.S. States as follows: Ala., Alabama; Ariz., Arizona; Calif., California; Ga., Georgia; Mich., Michigan; N.C., North Carolina; Oreg., Oregon; S. Dak., South Dakota; Wyo., Wyoming]

Deposit name	State	Longitude	Latitude	Past production ore tonnage (short tons)	Troy ounces of gold produced	Other production
Bonanza	Calif.	–120.384	37.991	Unk	73,000	—
Bonanza King	Calif.	–122.626	41.074	Unk	Unk	—
Briggs-Suitcase-Mineral Hill	Calif.	–117.185	35.938	3,643,600	70,186	—
Brown Bear	Calif.	–122.729	40.724	Unk	400,000–500,000	—
Brown's Valley	Calif.	–121.235	39.378	570,000	200,000	—
Brush Creek	Calif.	–120.88	39.502	>116,000	307,000	—
BUnker Hill	Calif.	–120.824	38.425	>888,000	266,000	—
Campo Seco (Lauren)	Calif.	–120.853	38.228	Unk	Unk	—
Carson Hill	Calif.	–120.506	38.023	>10,657,000	1,355,000	—
Central Eureka	Calif.	–120.797	38.383	1,110,586	456,250	—
Champion-Providence	Calif.	–121.037	39.259	772,000	203,579	—
Confidence-Lamphere	Calif.	–120.208	38.044	>169,195	191,000	—
Copper Basin	Calif.	–114.254	34.293	Unk	Unk	—
Cove	Calif.	–118.407	35.711	1,500,000	338,000	—
Crescent Mills	Calif.	–120.937	40.121	>900,000	>232,000	—
Delhi-St. Gothard	Calif.	–120.991	39.41	>54,000	73,000	—
Dewey	Calif.	–122.581	41.429	Unk	Unk	—
Diamond-Bullion-Alaska	Calif.	–121.051	39.194	Unk	169,000	—
Dorleska	Calif.	–122.911	41.058	Unk	Unk	—
Eagle-Shawmut	Calif.	–120.4	37.868	≈2,500,000	290,000	—
Empire-North Star	Calif.	–121.045	39.206	10,346,000	5,855,381	—
Enterprise	Calif.	–123.124	40.846	Unk	Unk	—
Ferguson (Anderson)	Calif.	–120.185	37.666	Unk	66,000	—
Four Hills	Calif.	–120.71	39.699	Unk	97,000	—
Fremont-Gover	Calif.	–120.826	38.435	>1,078,000	>673,495	—
Gaston Ridge	Calif.	–120.751	39.396	295,000	72,000	—
Gilta	Calif.	–123.325	41.195	Unk	Unk	—
Gladstone (Hazel)	Calif.	–122.585	40.725	>346,000	165,000	—
Globe	Calif.	–123.989	40.89	>60,000	113,970	—
Gold Bluff	Calif.	–120.821	39.582	Unk	73,000	—
Gold Cliff	Calif.	–120.544	38.066	Unk	137,000	—
Gold Hill	Calif.	–121.038	39.261	Unk	190,000	—
Goldbank	Calif.	–121.269	39.526	>154,000	115,000	—
Golden Center	Calif.	–121.063	39.215	>373,000	188,000	—
Golden Eagle	Calif.	–122.848	41.681	Unk	Unk	—
Governor (New York)	Calif.	–118.205	34.507	Unk	>73,000	—
Grizzly	Calif.	–120.222	37.941	Unk	73,000	—
Gwin	Calif.	–120.76	38.276	1,400,000	168,000	—
Hasloe	Calif.	–120.048	37.698	Unk	145,000	—
Herman	Calif.	–120.637	39.135	Unk	Unk	—

Table 1. Examples of orogenic gold (Au) deposits in the United States.—Continued

[Modified from Long and others (1998), and references therein. —, no data; Unk, unknown; W.Y.O.D., Work Your Own Diggings; >, greater than; ≈, almost equal to. Abbreviations of U.S. States as follows: Ala., Alabama; Ariz., Arizona; Calif., California; Ga., Georgia; Mich., Michigan; N.C., North Carolina; Oreg., Oregon; S. Dak., South Dakota; Wyo., Wyoming]

Deposit name	State	Longitude	Latitude	Past production ore tonnage (short tons)	Troy ounces of gold produced	Other production
Hickey	Calif.	–123.09	41.257	Unk	Unk	—
Hite	Calif.	–119.864	37.646	Unk	145,000	—
May Lundy	Calif.	–119.26	38.017	100,000	145,000	—
Idaho-Maryland-Eureka- Brunswick	Calif.	–121.038	39.224	>5,546,000	2,458,000	Tungsten
Italian	Calif.	–120.83	38.444	18,474	2,574	—
Jamestown	Calif.	–120.437	37.946	>16,832,000	2,310,000	—
Jamison	Calif.	–120.702	39.735	>334,000	89,000	—
Julian-Banner	Calif.	–116.556	33.075	Unk	242,000	—
Kennedy	Calif.	–120.779	38.367	4,340,000	1,742,000	—
Keystone	Calif.	–120.822	38.418	>1,310,000	>277,000	—
Lava Cap (Banner, Central)	Calif.	–120.97	39.228	>1,002,265	520,000	—
Lightner	Calif.	–120.56	38.073	>570,000	>145,000	—
Lincoln	Calif.	–120.8	38.395	Unk	97,000	—
Mammoth	Calif.	–118.501	35.623	>8,000	29,000	—
Mariposa	Calif.	–119.958	37.48	>114,000	116,000	—
Mary Harrison-Malvina	Calif.	–120.216	37.695	>463,000	128,000	—
Massachusetts Hill	Calif.	–121.006	39.249	280,000	191,360	—
McKeen	Calif.	–122.823	41.271	Unk	Unk	—
McKinley	Calif.	–122.769	41.762	Unk	Unk	—
Meadow Lake (Excelsior)	Calif.	–120.521	39.397	>10,000	9,700	—
Midas	Calif.	–122.985	40.39	179,187	176,236	—
Milkmaid-Franklin	Calif.	–122.669	40.722	Unk	121,000	—
Mount Gaines	Calif.	–120.177	37.54	>206,000	174,000	—
Mountain Laurel	Calif.	–123.105	41.26	Unk	Unk	—
Mountaineer	Calif.	–121.026	39.257	Unk	97,000–145,000	—
Murchie	Calif.	–120.987	39.265	708,500	283,000	—
Nashville	Calif.	–120.841	38.581	Unk	97,000	—
New York Hill	Calif.	–121.075	39.199	100,000	73,000	—
Old Eureka	Calif.	–120.799	38.388	>902,000	880,000	—
Omaha-Lone Jack- Homeward Bound- Wisconsin	Calif.	–121.063	39.187	220,000	154,000	—
Oneida	Calif.	–120.787	38.374	>304,000	>121,000	—
Ophir	Calif.	–121.124	38.905	Unk	242,000	—
Oriental	Calif.	–120.858	39.46	Unk	206,000	—
Original	Calif.	–119.864	37.673	175,156	79,637	—
Original Amador	Calif.	–120.826	38.423	>477,000	100,400	—
Pacific Quartz	Calif.	–120.799	38.727	Unk	72,000	—
Pennsylvania-W.Y.O.D.	Calif.	–121.055	39.209	>161,000	135,000	—
Pine Tree-Josephine	Calif.	–120.119	37.588	>540,000	>194,000	—

Table 1. Examples of orogenic gold (Au) deposits in the United States.—Continued

[Modified from Long and others (1998), and references therein. —, no data; Unk, unknown; W.Y.O.D., Work Your Own Diggings; >, greater than; ≈, almost equal to. Abbreviations of U.S. States as follows: Ala., Alabama; Ariz., Arizona; Calif., California; Ga., Georgia; Mich., Michigan; N.C., North Carolina; Oreg., Oregon; S. Dak., South Dakota; Wyo., Wyoming]

Deposit name	State	Longitude	Latitude	Past production ore tonnage (short tons)	Troy ounces of gold produced	Other production
Plumas-Eureka	Calif.	–120.708	39.758	Unk	>387,000	—
Plumbago (Croesus)	Calif.	–120.813	39.452	>152,000	169,000	—
Plymouth	Calif.	–120.843	38.475	2,150,000	656,000	—
Pocket Belt	Calif.	–120.345	37.962	Unk	267,000	—
Princeton	Calif.	–119.97	37.502	>497,000	213,000–290,000	—
Quartz Hill	Calif.	–122.999	41.743	Unk	Unk	—
Rainbow	Calif.	–120.834	39.464	Unk	121,000	—
Rawhide	Calif.	–120.446	37.961	Unk	Unk	—
Red Cloud	Calif.	–120.085	37.739	Unk	73,000	—
Red Ledge	Calif.	–121.149	39.295	Unk	Unk	—
Reid	Calif.	–122.429	40.661	>106,000	121,000	—
Rich Gulch	Calif.	–121.143	40.046	Unk	77,000	—
Rising Sun	Calif.	–120.968	39.107	Unk	>97,000	—
Royal-Mountain King	Calif.	–120.689	38.002	>6,184,000	433,000	—
Schroeder	Calif.	–122.79	41.75	Unk	Unk	—
Sheep Ranch	Calif.	–120.469	38.212	640,000	294,000	—
Sierra Buttes	Calif.	–120.643	39.574	>250,000	854,000–1,287,000	—
Sixteen-to-one	Calif.	–120.843	39.466	>1,067,000	>1,100,000	—
Sliger	Calif.	–120.932	38.941	>320,000	138,000	—
Soulsby	Calif.	–120.261	37.988	Unk	266,000	—
South Eureka	Calif.	–120.795	38.38	>1,148,000	256,500	—
South United	Calif.	–120.386	38.071	Unk	82,000	—
Spanish	Calif.	–120.786	39.382	159,115	31,679	—
Stonewall	Calif.	–116.569	32.983	Unk	100,000	—
Sultana-Orleans	Calif.	–121.038	39.194	400,000	364,000	—
Union-Church	Calif.	–120.827	38.648	>211,000	131,000–242,000	—
Utica	Calif.	–120.588	38.087	>2,815,000	861,000	—
Venecia	Calif.	–122.756	40.74	Unk	Unk	—
Walker	Calif.	–122.423	40.652	Unk	Unk	—
Washington	Calif.	–122.681	40.718	Unk	Unk	—
Washington	Calif.	–120.224	37.526	Unk	115,000	—
Wildman-Mahoney	Calif.	–120.808	38.394	>600,000	>161,000	—
Yankee John	Calif.	–122.46	40.533	Unk	Unk	—
Young America	Calif.	–120.631	39.615	Unk	73,000	—
Yuba	Calif.	–120.704	39.353	Unk	97,000	—
Zeibright	Calif.	–120.754	39.331	866,000	80,000	—
Zeila	Calif.	–120.763	38.345	>823,000	>242,000	—
Creighton (Franklin)	Ga.	–84.267	34.295	150,000	48,500	—
Golden Chest	Idaho	–115.834	47.617	61,704	11,874	Tungsten
Ropes	Mich.	–87.425	46.53	2,955,000	203,000	—

Table 1. Examples of orogenic gold (Au) deposits in the United States.—Continued

[Modified from Long and others (1998), and references therein. —, no data; Unk, unknown; W.Y.O.D., Work Your Own Diggings; >, greater than; ≈, almost equal to. Abbreviations of U.S. States as follows: Ala., Alabama; Ariz., Arizona; Calif., California; Ga., Georgia; Mich., Michigan; N.C., North Carolina; Oreg., Oregon; S. Dak., South Dakota; Wyo., Wyoming]

Deposit name	State	Longitude	Latitude	Past production ore tonnage (short tons)	Troy ounces of gold produced	Other production
Gold Hill group	N.C.	–80.346	35.513	Unk	141,000	—
Howie	N.C.	–80.715	34.956	Unk	51,300	—
Whitney group	N.C.	–80.366	35.5	Unk	3,000	—
Ashland	Oreg.	–122.791	42.182	Unk	63,000	—
Greenback	Oreg.	–123.306	42.654	Unk	182,000	—
Homestake	S. Dak.	–103.765	44.356	Unk	Unk	—
Keystone Consolidated	S. Dak.	–103.378	43.899	227,933	83,215	Arsenic
Wharf	S. Dak.	–103.838	44.356	Unk	Unk	—
South Pass-Atlantic City	Wyo.	–108.7	42.5	>43,000	299,900	—

¹The acronym “TMC” is not defined in Long and others (1998), and references therein.

In the United States, Coeur d'Alene-type mineral systems are largely in deformed and faulted metasedimentary rocks of the Proterozoic Belt Basin of Idaho and western Montana (Leach and others, 1998), the Paleozoic Arkoma basin in southwest Arkansas (not shown in [fig. 1](#); Howard, 1979), and the Spruce Creek sequence in the Katishna Hills at Quigley Ridge in Alaska (not shown in [fig. 1](#); Bundtzen, 1981; [fig. 1](#); [table 2](#)).

Formation of Orogenic Mineral Systems

A temporal relation exists between the accretion of juvenile crustal terranes and the formation of orogenic mineral systems (Goldfarb and others, 2001). Orogenic mineral systems form late during the collisional orogenic process and after regional metamorphism of the accreted terranes (Goldfarb and others, 2005). Deformation and prograde metamorphism of midcrustal rocks during an accretionary orogeny can release large volumes of fluid (Elmer and others, 2006; Phillips and Powell, 2010), bisulfide (HS[–]) ligands (Tomkins, 2010), and gold and other metals (Pitcairn and others, 2006, 2010) as a result of metamorphic reactions near the greenschist to amphibolite facies transition. These gold-rich hydrothermal fluids are subsequently focused and flow upwards along dilatant structures in concert with the earthquake cycle. High fluid pressure can induce fault ruptures and swarm seismicity, during which enormous volumes of metamorphic hydrothermal fluid can channel and flow upwards along major structures that only display small displacement (for example, Sibson, 1987; Sibson and others, 1988; Cox, 2016).

As the metamorphic hydrothermal fluids are pumped towards shallower depths, ore deposits may form within favorable chemical and structural traps. Metals are introduced

by ascending hydrothermal fluids, but additional metals may also be derived from local aquifers and host rocks through fluid–fluid and fluid–rock reactions at the deposit site. The metals within orogenic systems are dependent upon the types of rocks undergoing metamorphism at depth, the immediate host rock of the shear zone and veins, and the depth of formation ([fig. 2](#); Goldfarb and others, 2005). A continuum of deposits within this system goes from deeper level gold through shallower mercury deposits. The major commodity within orogenic gold deposits is gold. Other critical minerals that may be enriched in these deposits include arsenic, tellurium, and tungsten. Additionally, antimony, mercury, and graphite deposits may be present in other parts of the system; antimony and graphite are critical minerals (Hofstra and Kreiner, 2020).

The energy source required for forming an orogenic gold deposit is orogenesis and its associated metamorphism. Fluids, ligands, and metals are all sourced from supracrustal rocks undergoing metamorphism in the midcrust. Reduced oceanic volcanic arc and associated sedimentary rock domains containing pyrite (FeS₂) are likely to be a favorable protolith (source rock) that release fluids (Elmer and others, 2006; Phillips and Powell, 2010), ligands (Tomkins, 2010), and metals (Pitcairn and others, 2006, 2010), particularly during the greenschist to amphibolite facies metamorphic transition.

Fluid inclusion studies demonstrate the predominance of low-salinity water (H₂O)–carbon dioxide (CO₂; ± methane [CH₄], dinitrogen [N₂]) fluids in orogenic gold deposits (Goldfarb and Groves, 2015); there are rare examples containing only miniscule amounts of CO₂ (Taylor and others, 2021). Generation of an H₂O–CO₂ fluid composition would be expected through metamorphic devolatilization reactions occurring in the midcrust (Fyfe and others, 1978; Powell and others, 1991), and the general consistency in fluid

Table 2. Examples of Coeur d'Alene-type deposits in the United States.

[Modified from Long and others (1998), and references therein, Hofstra and others (2013), and Howard (1979). Ark., Arkansas; lbs, pounds; Mont., Montana; Neg, little or no production; Unk, unknown]

Deposit Name	State	Longitude	Latitude	Past production tonnage (short tons)	Troy ounces of silver produced	Pounds of zinc produced	Other production	Reference
Quigley Ridge	Alaska	−150.947	63.543	725	117,305	Neg	—	Long and others, 1998
Antimony district	Ark.	−94.235	34.185	Unk	Neg	Neg	5,400 short tons antimony ore	Howard, 1979
Atlas	Idaho	−115.776	47.466	Unk	Unk	Unk	—	Long and others, 1998
Bunker Hill—Last Chance	Idaho	−116.138	47.536	42,739,696	140,632,566	2,344,064,888	Cadmium, antimony, barite, cobalt	Long and others, 1998
Caledonia	Idaho	−116.168	47.524	263,182	8,092,307	Neg	—	Long and others, 1998
Coeur	Idaho	−115.992	47.489	2,476,692	40,259,304	474,912	—	Long and others, 1998
Conjecture	Idaho	−116.429	47.916	Unk	Unk	Unk	—	Long and others, 1998
Crescent	Idaho	−116.073	47.506	1,202,866	26,628,367	657,477	Antimony	Long and others, 1998
Dayrock	Idaho	−115.9	47.512	1,276,488	6,526,992	7,762,003	—	Long and others, 1998
Galena	Idaho	−115.965	47.477	6,287,684	137,169,731	3,851,306	—	Long and others, 1998
Gold Hunter	Idaho	−115.785	47.472	3,260,750	9,955,082	1,144,318	—	Long and others, 1998
Hecla	Idaho	−115.814	47.52	7,686,967	40,788,931	73,824,278	—	Long and others, 1998
Helena-Frisco	Idaho	−115.85	47.51	2,676,379	6,131,065	174,966,741	—	Long and others, 1998
Hercules	Idaho	−115.808	47.543	3,519,592	29,952,537	9,616,058	—	Long and others, 1998
Idora	Idaho	−115.86	47.566	Unk	Unk	Unk	—	Long and others, 1998
Jack Waite	Idaho Mont.	−115.744	48	693,149	641,863	20,424,000	—	Long and others, 1998
Last Chance	Idaho	−116.1486	47.51694	Unk	Unk	Unk	—	Long and others, 1998
Little Pittsburgh	Idaho	−116.2	47.485	Unk	Unk	Unk	—	Long and others, 1998
Lucky Friday	Idaho	−115.78	47.471	6,653,699	104,266,200	169,088,000	—	Long and others, 1998
Mineral Point	Idaho	−116.006	47.489	440,779	5,859,581	Neg	—	Long and others, 1998
Monitor (Interstate-Callahan)	Idaho	−115.887	47.544	3,708,030	4,210,670	401,256,743	—	Long and others, 1998
Murray district	Idaho	−115.834	47.617	Unk	Unk	Unk	Gold	Long and others, 1998
Nabob	Idaho	−116.207	47.488	Unk	Unk	Unk	—	Long and others, 1998

Table 2. Examples of Coeur d'Alene-type deposits in the United States.—Continued

[Modified from Long and others (1998), and references therein, Hofstra and others (2013), and Howard (1979). Ark., Arkansas; lbs, pounds; Mont., Montana; Neg, little or no production; Unk, unknown]

Deposit Name	State	Longitude	Latitude	Past production tonnage (short tons)	Troy ounces of silver produced	Pounds of zinc produced	Other production	Reference
Page—Blackhawk	Idaho	−116.201	47.528	4,521,461	15,365,503	552,978,641	—	Long and others, 1998
Polaris	Idaho	−116.052	47.502	320,783	7,368,759	29,718	—	Long and others, 1998
Rex	Idaho	−115.873	47.535	Unk	Unk	Unk	—	Long and others, 1998
Senator Stewart	Idaho	−116.171	47.525	1,041,814	6,610,160	323,888	—	Long and others, 1998
Sherman	Idaho	−115.82	47.525	661,071	3,930,759	8,546,128	—	Long and others, 1998
Sidney	Idaho	−115.192	47.488	1,071,197	1,931,081	171,151,606	—	Long and others, 1998
Silver Summit	Idaho	−116.025	47.506	798,761	19,932,835	119,200	—	Long and others, 1998
Standard—Mammoth	Idaho	−115.836	47.519	3,763,893	25,542,538	949,832	—	Long and others, 1998
Star-Morning	Idaho	−115.812	47.468	26,439,368	73,078,060	3,324,146,000	—	Long and others, 1998
Success	Idaho	−115.872	47.527	Unk	Unk	Unk	—	Long and others, 1998
Sunshine	Idaho	−116.068	47.501	12,794,736	366,177,233	8,668,646	354,174,808 lbs antimony	Long and others, 1998
Tamarack and Custer	Idaho	−115.848	47.536	2,819,472	8,753,391	156,417,665	—	Long and others, 1998
Tiger—Poorman	Idaho	−115.813	47.523	1,128,793	2,656,234	127,846	—	Long and others, 1998
Unity-Rescue	Idaho	−115.7	45.258	Unk	58,000	Neg	—	Long and others, 1998
Florence	Mont.	−110.743	46.942	Unk	Unk	Unk	—	Long and others, 1998
U.S. Antimony	Mont.	−115.59	47.572	Unk	Unk	Unk	15,400 tonnes antimony	Hofstra and others, 2013

characteristics between different orogenic gold deposits is suggestive of a common fluid source. Within accretionary belts such as the California Mother Lode (not shown in [fig. 1](#)), accreted oceanic volcanic arc and associated sedimentary rocks are the likely source of metamorphic fluids associated with orogenic gold deposits through dehydration reactions such as the conversion of chlorite to amphibole (Elmer and others, 2006; Phillips and Powell, 2010).

Although numerous potential ligands can be found in the hydrothermal fluid, bisulfide (HS^-) is the most important ligand for gold complexing (Loucks and Mavrogenes, 1999). Desulfidation reactions of a reduced rock containing pyrite as the sulfur source are necessary to generate these gold-complexing ligands. The pyrite to pyrrhotite ($\text{Fe}_{[1-x]}\text{S}$) reaction that releases sulfur occurs at the greenschist to

amphibole facies transition, which is the same important fluid-generating transition detailed in the previous paragraph (Tomkins, 2010). Major ore metals—including gold, arsenic, bismuth, tellurium, and tungsten—are released through prograde metamorphic reactions in metasedimentary rocks, particularly at the greenschist to amphibolite facies transition (Pitcairn and others, 2006). These ore metals are largely stripped from pyrite (Pitcairn and others, 2010), and although sedimentary pyrite is an important source for gold, basalt may also provide a source for the ore metals (Pitcairn and others, 2015).

Orogenic gold deposits form in association with crustal-scale shear zones that act as fluid pathways that link the metamorphic area of fluid, ligand, and metal generation with the mineralization site and shallower outflow region

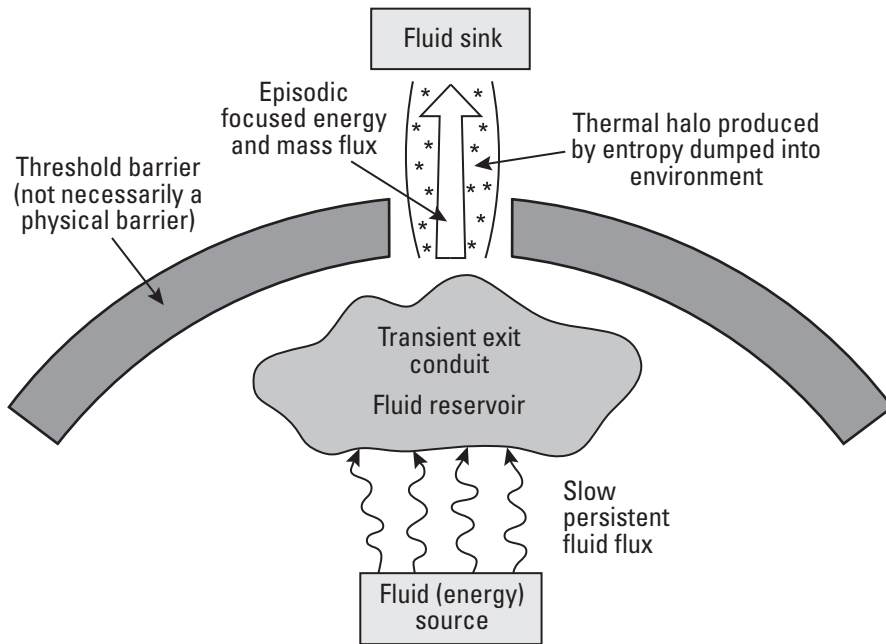


Figure 2. Simplified schematic model for a mineral system, including metamorphic shear zone hydrothermal systems. Modified from Wyman and others (2016).

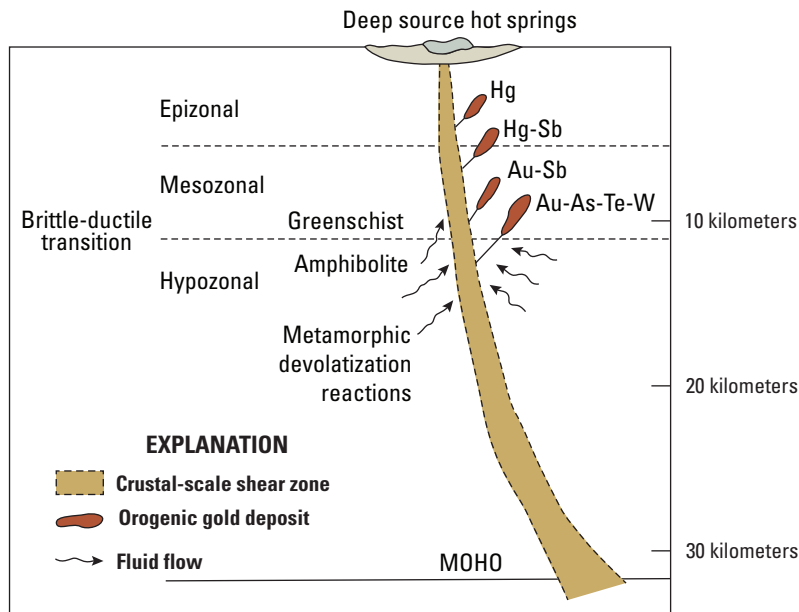


Figure 3. Schematic model of an orogenic system. Although different genetic models have been proposed, this model represents the generally accepted supracrustal metamorphic devolatilization model. Modified from Groves and others (2020). (As, arsenic; Au, gold; Hg, mercury; MOHO, Mohorovičić discontinuity; Sb, antimony; Te, tellurium; W, tungsten)

(fig. 3). The spatial association between major orogenic gold deposits and crustal-scale regional fault zones has been established (for example, Turneaure, 1955), although the sites of mineralization are typically in lower order faults where more efficient structural and chemical traps allow gold mineralization. Geopressurized fluids are channeled along the permeable first-order structures during seismic events (Sibson and others, 1988; Cox and others, 2001; Cox, 2016). In California, these major structures commonly mark boundaries between various amalgamated terranes.

Within orogenic gold deposits, gold may be in its native form, as gold telluride minerals, or as invisible gold found as nanoinclusions or structurally bound within sulfide minerals (generally, pyrite or arsenopyrite). Ore deposition is postulated to occur through various physicochemical processes that lower gold solubility within the ore fluid. Fluid–rock interaction is of primary importance, especially for nanoinclusions or structurally bound gold associated with sulfides. During fluid interaction of rocks with high iron (Fe) to iron plus magnesium (Mg; Fe+Mg) ratios, gold will become destabilized from its bisulfide ligand, and pyrite will form from the reaction of iron in the host rock with sulfur in the fluid along with concomitant deposition of gold (Phillips and Groves, 1983; Böhlke, 1988). In addition to iron-rich rocks, carbon-rich rocks are also important reductants of the fluid that can lead to gold mineralization (Cox and others, 1991). Hydraulic fracturing, an important process in the formation of orogenic gold deposits, can trigger pressure fluctuations within the ore-forming environment from supralithostatic to hydrostatic pressure conditions. Pressure changes alone are an effective mechanism that alter gold solubility and cause gold precipitation (for example, Groves and others, 1987; Loucks and Mavrogenes, 1999). Agglomeration of colloidal gold in the hydrothermal fluid resulting from pressure changes may result in large

sponge- and dendrite-like masses of gold (Taylor and others, 2021). Additional mechanisms of lesser importance that may result in gold precipitation include pH decrease of the hydrothermal fluid, fluid mixing, changes in oxygen fugacity (fO_2), and temperature changes.

Hydrothermal alteration related to gold precipitation occurs at the deposit site due to fluid–rock interaction and may occur pre-, syn-, and postmineralization. Early alteration may increase the competency of the host rock, allowing brittle fractures to develop that host gold mineralization (for example, Poulsen and others, 1986). The composition and structure of the wall rock plays a significant role in what alteration products are present. Quartz-sericite (muscovite)-pyrite and carbonate (calcite, ankerite, ferroan dolomite, magnesite, and siderite) alteration are fundamental alteration types. Talc and fuchsite may substitute for sericite or muscovite within ultramafic wall rocks, and magnesite and siderite are the accompanying dominant carbonate minerals.

In addition to gold and silver, other ore minerals may include pyrite, arsenopyrite ($FeAsS$), galena (PbS), sphalerite (ZnS), chalcopyrite ($CuFeS_2$), stibnite (Sb_2S_3), scheelite ($CaWO_4$), graphite (C), rutile (TiO_2), tetrahedrite ($[Cu,Fe,Zn,Ag]_{12}[Sb,As]_4S_{13}$), and various telluride minerals (fig. 4); of these ore minerals, arsenopyrite (arsenic), stibnite (antimony), scheelite (tungsten), graphite, and telluride minerals (tellurium) are important host phases for critical minerals (table 3). The most common telluride minerals in orogenic gold deposits include calaverite ($AuTe_2$), sylvanite ($AgAuTe_4$), hessite (Ag_2Te), and petzite (Ag_3AuTe_2 ; table 3). Pyrite may contain critical metals such as arsenic, cobalt, and nickel, but the metals are likely found as minor to trace components within the crystals (Taylor and others, 2021). Gangue (the valueless rock or mineral aggregates in an ore) mineralogy is dominated by quartz and various carbonate minerals and lesser amounts of mica (such as muscovite, chlorite, and fuchsite).

Formation of Coeur d'Alene-type Mineral Systems

Coeur d'Alene-type mineral systems form in a manner similar to orogenic mineral systems, but there are some key differences: Coeur d'Alene-type mineral systems have different source rocks and more saline fluids, have chloride as the major ligand, and comprise a different suite of metals. Major commodities within Coeur d'Alene-type polymetallic sulfide deposits are silver, lead, zinc, and copper; minor commodities in these deposits include gold and byproduct cadmium, cobalt, antimony, and barite. Of the major and minor commodities, antimony, barite, cobalt, and zinc are considered critical minerals. Additional critical minerals that may be enriched in these polymetallic sulfide deposits include arsenic, bismuth, gallium, germanium, indium, and

manganese. Antimony and uranium deposits occur in other parts of the system; antimony is a critical mineral (Hofstra and Kreiner, 2020).

Coeur d'Alene-type systems have also been called “mesothermal base and precious metal systems” (Leach and others, 1988), “polymetallic sulfide and antimony deposits” (Long, 1998a), or “silver-lead-zinc veins in clastic metasedimentary rock terranes” (Beaudoin and Sangster, 1992). In the United States, these types of deposits are largely in deformed and metamorphosed rocks of the Proterozoic Belt Basin along the Lewis and Clark line (a fault zone that extends from near Wallace, Idaho [not shown in fig. 1], to east of Helena, Montana; Wallace and others, 1990) in Idaho and Montana (fig. 1). Zinc-, lead-, silver-, and gold-rich ores are in different districts and are suggested to have formed during the Proterozoic or Late Cretaceous–early Tertiary (Zartman and Stacey, 1971; Leach and others, 1998; Ramos and Rosenberg, 2012). Metamorphism accompanying orogenesis provides the energy required to form Coeur d'Alene-type deposits. Fluids, ligands, and metals are all sourced from siliciclastic sedimentary rocks undergoing metamorphism in a similar manner to what is outlined for orogenic mineral systems (refer to the “Formation of Orogenic Mineral Systems” section).

Oxygen isotope studies of vein quartz from the Coeur d'Alene region indicate a mix of meteoric and metamorphic fluids derived from rocks of the Belt Basin were involved in ore deposition (Constantopoulos and Larson, 1991; Constantopoulos, 1994). In addition to oxygen isotopes, lead isotopes of sulfides and strontium isotopes of vein carbonate minerals further suggest that strontium and lead were derived from rocks of the Belt Basin and pre-existing lead-zinc mineralization, either in the Proterozoic or Mesozoic (Rosenberg and Larson, 1996; Leach and others, 1998; Fleck and others, 2002). Another study of lead, strontium, and neodymium isotopes in carbonate minerals suggested that these elements were also derived from Archean basement rocks (Ramos and Rosenberg, 2012). Ore deposition occurred between 250 and 350 degrees Celsius ($^{\circ}C$), based upon fluid inclusion analysis (Leach and others, 1988). Fluid mixing of deeply sourced and meteoric fluids is an important process leading to mineralization (Beaudoin and Sangster, 1992). However, the correlation between productive veins and specific stratigraphy highlights the importance of local fluid–rock interactions and favorable rheology to mineralization; base metal veins are more commonly located within quartzite, whereas silver-rich veins are within silicified siltstone of the Proterozoic Revett Formation (Mauk and White, 2004), and quartz-stibnite veins are in the Proterozoic Prichard Formation for deposits in the Coeur d'Alene region (Hofstra and others, 2013).

Although bisulfide complexes are the most important ligand for complexing gold within orogenic gold deposits, fluid inclusion analyses by Hofstra and others (2013) indicate that antimony and silver-rich veins within the Coeur d'Alene region contain low concentrations of hydrogen sulfide (H_2S). In contrast, the moderate to high salinity indicates that

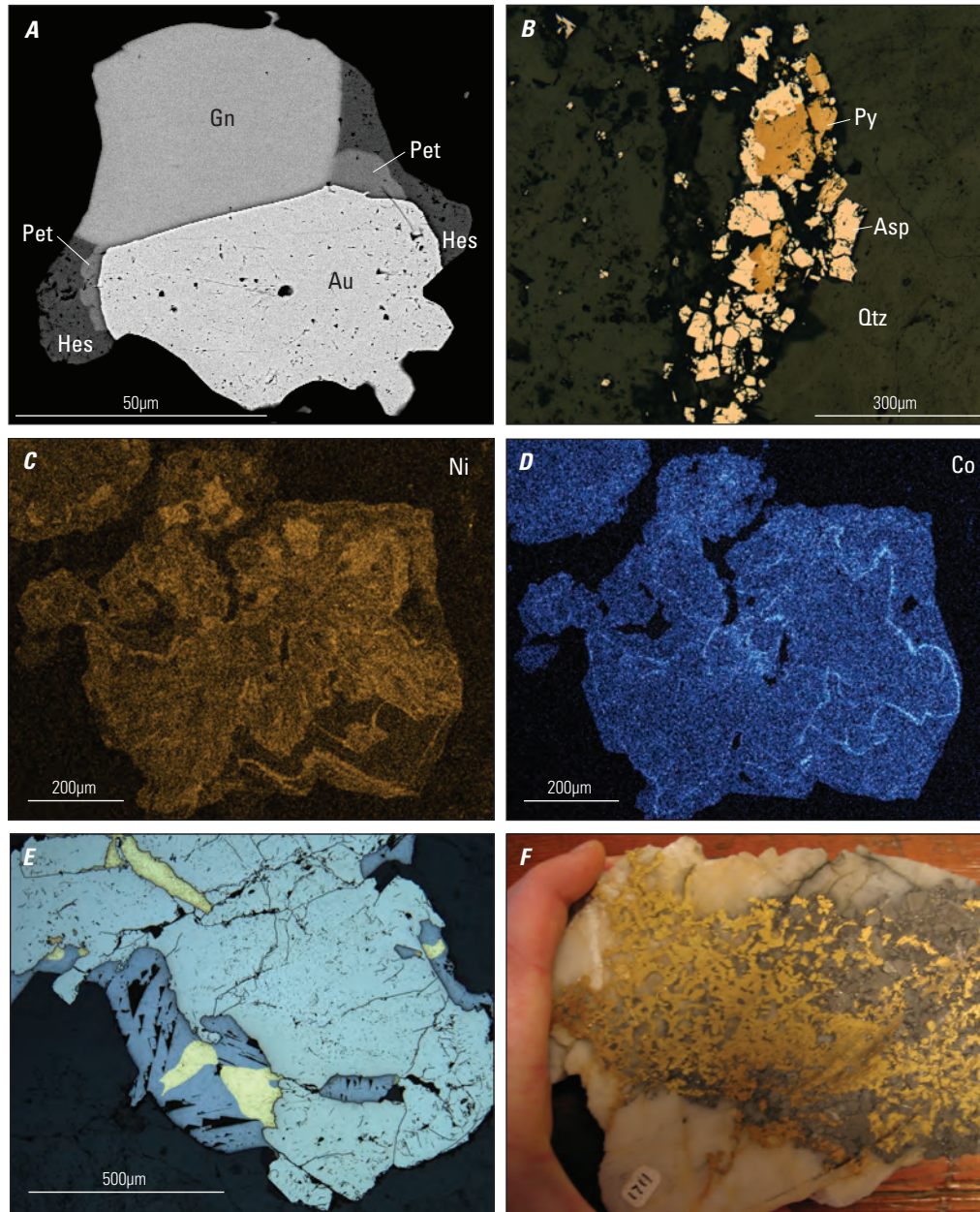


Figure 4. Photomicrographs and scans of orogenic gold ore showing the residence of critical minerals (μm , micrometer). *A*, A backscattered electron image of native gold (Au), galena (Gn), hessite (Hes), and petzite (Pet) from the Idaho-Maryland mine in Grass Valley, California. Modified from Taylor and others (2021). *B*, A reflected light photomicrograph of arsenopyrite (Asp) and pyrite (Py) within a quartz (Qtz) vein, Empire deposit in Grass Valley, California. (Photomicrograph by R. Taylor, U.S. Geological Survey) *C–D*, Qualitative energy-dispersive X-ray elemental maps illustrating nickel (Ni) and cobalt (Co) zoning within a pyrite crystal from the Idaho-Maryland mine in Grass Valley, California. Modified from Taylor and others (2021). *E*, Gold intergrown with galena and within fractures in pyrite. Sample from Grass Valley, California. (Photomicrograph by R. Taylor, U.S. Geological Survey) *F*, Photograph of high-grade gold ore with native gold and arsenopyrite from the Sixteen-to-One deposit in Alleghany, California. (Photograph by Erin Marsh, U.S. Geological Survey)

Table 3. Ore minerals containing critical mineral metals.

[CdA, Coeur d'Alene-type; OroAu, orogenic gold; X, 0 to 0.125; Abbreviations of chemical elements as follows: Ag, silver; As, arsenic; Au, gold; C, carbon; Ca, calcium; Cd, cadmium; Co, cobalt; Cu, copper; Fe, iron; Ga, gallium; Ge, germanium; In, indium; Mn, manganese; Ni, nickel; O, oxygen; Pb, lead; S, sulfur; Sb, antimony; Te, tellurium; W, tungsten; Zn, zinc]

Mineral	Stoichiometry	Critical mineral	System type
Arsenopyrite	FeAsS	As	CdA, OroAu
Graphite	C	Graphite	OroAu
Pyrite	FeS ₂	As, Co, Ni	CdA, OroAu
Scheelite	CaWO ₄	W	CdA, OroAu
Stibnite	Sb ₂ S ₃	Sb	CdA, OroAu
Calaverite	AuTe ₂	Te	OroAu
Hessite	Ag ₂ Te	Te	OroAu
Petzite	AuAg ₃ Te ₂	Te	OroAu
Sylvanite	AuAgTe ₄	Te	OroAu
Sphalerite	ZnS	Cd, In, Ge, Ga	CdA, OroAu
Tetrahedrite	(Cu,Fe,Zn,Ag) ₁₂ (Sb,As) ₄ S ₁₃	Sb	CdA
Boulangerite	Pb ₃ Sb ₄ S ₁₁	Sb	CdA
Gersdorffite	NiAsS	Co, Ni	CdA
Pyrrhotite	Fe _(1-x) S	Co, Ni	CdA, OroAu
Siderite	FeCO ₃	Mn	CdA

chloride complexes were important ligands transporting metals during formation of silver and antimony veins. Chloride complexes are also inferred to be important in the Coeur d'Alene base metal veins, which has been demonstrated for zinc and lead mobility during metamorphism (Hammerli and others, 2015).

The source of metals in Coeur d'Alene-type polymetallic veins is considered the Belt Basin. Sedimentary rocks, potentially the pyrite-bearing rocks of the lower part of the Prichard Formation, which contains sedex lead-zinc deposits in the Canadian part of the Belt Basin, likely released lead, zinc, silver, and antimony during metamorphism. This model is supported by lead isotope studies showing that Coeur d'Alene ores have Proterozoic lead signatures similar to those signatures of the Sullivan lead-zinc sedex deposit in Canada (Fleck and others, 2002). The silver-rich veins are hosted in the Revett Formation, which regionally contains sedimentary rock-hosted copper deposits (Boleneus and others, 2005), including the Snowstorm deposit at the east end of the district. The correlation between copper, silver, and antimony in these veins suggests that they formed as ascending fluids that contained antimony, silver, and variable lead and zinc mixed with copper-bearing fluids derived from the Revett Formation (table 4).

The shear zone veins of the Coeur d'Alene region are located within the western part of the Lewis and Clark line, which extends from Coeur d'Alene, Idaho, to Helena, Montana (Wallace and others, 1990). This fault zone contains high-angle faults of variable kinematics that have been intermittently active since the middle Proterozoic with

major strike-slip movement in the Late Cretaceous (Bennett and Venkatakrishnan, 1982; Wallace and others, 1990). Individual faults within this fault zone may stretch for as many as 250 kilometers along strike. Ore-bearing veins are preferentially located within fractures and faults that intersect large folds within brittle host rocks; the ore province itself is located within the most deformed part of the Lewis and Clark line.

For Coeur d'Alene-type mineral systems, phyllic alteration composed of sericite, quartz, and pyrite is common for the wall rocks adjacent to veins (Beaudoin and Sangster, 1992). In the Coeur d'Alene region, however, alteration may be broader and is typified by bleaching of the rock through destruction of feldspar and hematite and the addition of siderite and sulfides (Fryklund, 1964; Mauk and Strand, 2002).

Mining and Beneficiation of Ore Deposits and Materials

Underground mining techniques have historically been the most common method of mining orogenic gold deposits, but open-pit mining is not uncommon. The grade-tonnage model of Bliss (1986) indicates that 50 percent of deposits have gold grades of at least 16 grams per tonne (g/t) and ore tonnages of at least 30,000 tonnes, which are indicative of the high-grade low-tonnage nature of these deposits. Because of the continued increase in gold prices, open-pit bulk mining of lower grade, hydrothermal alteration zones that would have previously been considered waste material has become

Table 4. Correlation coefficients of select metals commonly enriched in Coeur d'Alene-type deposits using all the samples compiled in the study (n=75).

[Compositional data are from Granitto and others (2020, 2021, 2025). —, not applicable; Abbreviations of chemical elements as follows: Ag, silver; Au, gold; As, arsenic; Bi, bismuth; C, carbon; Cd, cadmium; Co, cobalt; Cu, copper; Fe, iron; Ga, gallium; Ge, germanium; In, indium; Mn, manganese; Pb, lead; Pd, palladium; S, sulfur; Sb, antimony; Se, selenium; Te, tellurium; W, tungsten; Zn, zinc]

Element	Au	Ag	As	Bi	C	Cd	Co	Cu	Fe	Ga	Ge	In	Mn	Pb	Pd	S	Sb	Se	Te	W	Zn
Au	—	0.80	0.46	0.10	0.05	0.13	0.62	0.74	0.00	-0.08	-0.15	0.14	0.04	-0.15	0.79	0.12	0.60	0.00	-0.04	-0.01	0.06
Ag	0.80	—	0.31	0.15	0.19	-0.07	0.29	0.63	-0.02	-0.10	-0.16	-0.01	0.34	-0.16	0.63	-0.10	0.49	0.27	0.19	-0.04	-0.15
As	0.46	0.31	—	0.43	0.22	-0.02	0.27	0.44	0.42	-0.12	-0.17	0.05	0.24	-0.31	0.38	-0.05	0.30	-0.02	0.14	-0.07	-0.09
Bi	0.10	0.15	0.43	—	0.21	-0.06	-0.1	0.48	0.24	-0.08	-0.19	0.15	0.12	-0.14	0.08	-0.06	0.44	0.05	0.32	-0.05	-0.11
C	0.05	0.19	0.22	0.21	—	-0.35	-0.12	0.29	0.77	0.12	-0.40	-0.13	0.80	-0.35	0.10	-0.44	0.06	-0.24	-0.13	-0.22	-0.40
Cd	0.13	-0.07	-0.02	-0.06	-0.35	—	0.15	-0.15	-0.3	0.27	0.28	0.34	-0.31	-0.04	0.13	0.77	-0.11	-0.02	0.06	0.00	0.77
Co	0.62	0.29	0.27	-0.1	-0.12	0.15	—	0.40	0.06	0.11	0.12	-0.02	-0.08	-0.22	0.49	0.14	0.16	-0.09	-0.11	-0.06	0.15
Cu	0.74	0.63	0.44	0.48	0.29	-0.15	0.40	—	0.21	-0.11	-0.24	0.14	0.15	-0.20	0.52	-0.05	0.75	-0.03	-0.04	-0.07	-0.15
Fe	0.00	-0.02	0.42	0.24	0.77	-0.30	0.06	0.21	—	0.02	-0.31	-0.07	0.67	-0.46	0.08	-0.40	-0.02	-0.15	-0.09	0.00	-0.31
Ga	-0.08	-0.10	-0.12	-0.08	0.12	0.27	0.11	-0.11	0.02	—	0.33	-0.08	-0.08	-0.25	0.07	0.24	-0.21	0.00	-0.06	-0.04	0.19
Ge	-0.15	-0.16	-0.17	-0.19	-0.40	0.28	0.12	-0.24	-0.31	0.33	—	-0.12	-0.27	-0.09	-0.10	0.11	0.01	-0.07	-0.09	-0.09	0.20
In	0.14	-0.01	0.05	0.15	-0.13	0.34	-0.02	0.14	-0.07	-0.08	-0.12	—	-0.10	-0.08	0.16	0.48	0.12	0.01	0.10	0.02	0.46
Mn	0.04	0.34	0.24	0.12	0.80	-0.31	-0.08	0.15	0.67	-0.08	-0.27	-0.10	—	-0.22	0.10	-0.36	0.02	-0.14	-0.14	0.06	-0.34
Pb	-0.15	-0.16	-0.31	-0.14	-0.35	-0.04	-0.22	-0.20	-0.46	-0.25	-0.09	-0.08	-0.22	—	-0.30	0.27	-0.06	-0.12	-0.13	0.20	-0.07
Pd	0.79	0.63	0.38	0.08	0.10	0.13	0.49	0.52	0.08	0.07	-0.10	0.16	0.10	-0.30	—	0.04	0.40	0.01	-0.02	-0.03	0.10
S	0.12	-0.10	-0.05	-0.06	-0.44	0.77	0.14	-0.05	-0.40	0.24	0.11	0.48	-0.36	0.27	0.04	—	-0.02	-0.11	-0.11	0.02	0.90
Sb	0.6	0.49	0.30	0.44	0.06	-0.11	0.16	0.75	-0.02	-0.21	0.01	0.12	0.02	-0.06	0.40	-0.02	—	-0.05	-0.05	-0.05	-0.09
Se	0.00	0.27	-0.02	0.05	-0.24	-0.02	-0.09	-0.03	-0.15	0.00	-0.07	0.01	-0.14	-0.12	0.01	-0.11	-0.05	—	0.83	-0.02	-0.03
Te	-0.04	0.19	0.14	0.32	-0.13	0.06	-0.11	-0.04	-0.09	-0.06	-0.09	0.10	-0.14	-0.13	-0.02	-0.11	-0.05	0.83	—	-0.03	-0.05
W	-0.01	-0.04	-0.07	-0.05	-0.22	0.00	-0.06	-0.07	0.00	-0.04	-0.09	0.02	0.06	0.20	-0.03	0.02	-0.05	-0.02	-0.03	—	-0.03
Zn	0.06	-0.15	-0.09	-0.11	-0.40	0.77	0.15	-0.15	-0.31	0.19	0.20	0.46	-0.34	-0.07	0.10	0.90	-0.09	-0.03	-0.05	-0.03	—

more common since the Bliss (1986) grade-tonnage model was created. Inclusion of lower grade alteration zones has led to changes when applying the grade-tonnage model of Bliss (1986) to the current (2025) economic environment (Davies and others, 2020).

Mining techniques for Coeur d'Alene-type systems are similar to techniques for orogenic mineral systems because both are vein deposits. Underground mining methods predominate, but open-pit mining is occasionally used. Underground mining by stoping is common, but the room-and-pillar mining method is also considered a viable method. Using tailings as backfill is a common practice.

Beneficiation of ore materials is done through initial crushing and milling, gravity separation, flotation, and cyanide extraction with carbon leaching (Adams, 2016). Coarse-grained native gold from orogenic gold deposits has historically been concentrated through gravity separation techniques because of its high specific gravity that permits easy separation from other associated gangue and ore minerals in veins. The gravity separation method allows the remaining material, including critical minerals, to be collected in piles that are free of any processing chemicals. Another standard practice that is especially efficient for recovery of fine gold particles, low grade ore, or gold that is inadequately separated from the rock is cyanide extraction and carbon leaching. Cyanide extraction and carbon leaching can be done by itself after crushing and milling or after an additional step of concentrating ore minerals through froth flotation. Refractory gold, such as gold tellurides, requires oxidation, flotation, and leaching to free the gold at high recovery rates (Zhang and others, 2010). Roasting followed by cyanide leaching has been the preferred method of gold recovery from telluride ores (Ellis and Deschênes, 2016). Preleaching of tellurium from gold telluride ore prior to removal of gold has been investigated as a way to collect tellurium and to avoid any detrimental effects that this critical mineral has on the recovery of gold (Yang and others, 2021). Mercury amalgamation was an early method used for gold recovery but is rarely used today because of health and environmental hazards related to mercury.

Beneficiation of Coeur d'Alene-type ores begins with milling (crushing and grinding) and is followed by flotation or gravity separation that produces the metal concentrates (McMahon and others, 1974). Flotation reagents include sodium cyanide, zinc sulfate, lime, copper sulfate, xanthate, and methyl isobutyl carbonyl, which are common for processing polymetallic mineralization. Sintering of concentrates may be done depending on the processing methodology. Concentrates are then shipped to a smelter.

Methods

Geochemical data for altered and mineralized rocks from orogenic gold deposits discussed in this report were compiled and released by Granitto and others (2020, 2021, 2025). The data from Granitto and others (2020) include geochemical data for 44 archived orogenic gold mine samples, and 33 of those 44 samples come from the United States (13 different deposits). The other 11 samples come from Australia and South Africa and are not considered in this report. Additionally, five of the United States deposit analyses reported in Granitto and others (2020) are also excluded from this report because of controversial or unlikely classification as orogenic gold-type deposits. Granitto and others (2020) also include data for numerous samples from the Proterozoic Homestake mine in South Dakota.

Additional legacy data for orogenic gold deposits published in Granitto and others (2021) comprise 75 samples of analyzed drill core from the Idaho-Maryland mine in Grass Valley, Calif., and 12 samples previously analyzed by the Mineral Deposit Research Unit, University of British Columbia, which include 6 samples from Grass Valley, Calif.; 5 samples from the Mother Lode belt in California (not shown in [fig. 1](#)); and 1 sample from the Alleghany district, Calif. Newly produced data from 35 samples of Mesozoic–Cenozoic gold deposits compiled by Granitto and others (2025) comprise 6 samples from other mines in the Sierra Nevada foothills, Calif.; 21 samples from mines in the Klamath Mountains, Calif.; 7 samples from deposits in Alaska (not shown in [fig. 1](#)); and 1 sample from Idaho. Additionally, Granitto and others (2020, 2021, 2025) contain information regarding the analytical methods used and data sources. Orogenic gold ore samples included in this study are briefly described and tabulated in [appendix 1 \(table 1.1\)](#). Data on tailings or mine waste for orogenic gold deposits were not available.

Geochemical data on mineralized rock samples from Coeur d'Alene-type deposits discussed in this report were compiled and released by Granitto and others (2020, 2025). The data for Coeur d'Alene-type deposits in Granitto and others (2020, 2025) include geochemical data for 75 archived ore samples from 26 different mines and locations in the Coeur d'Alene region in Idaho and Montana. Coeur d'Alene-type ore samples included in this study are briefly described and tabulated in [appendix 1 \(table 1.2\)](#). Data on tailings or mine waste from Coeur d'Alene-type deposits were not available because tailings or mine waste samples were not collected.

Results

The nugget effect of gold within orogenic gold deposits causes difficulties in interpreting quantitative geochemical associations because large and valuable particles of gold may be heterogeneously distributed in the ore and can affect the

results of a smaller sample analysis (Dominy, 2014). However, there should be a general correlation between higher grade samples. Additionally, the methodology for measuring metal content differs among many of the analyses and therefore has differing detection limits (Granitto and others, 2020, 2021, 2025). Median geochemical values are preferred more than mean values because numerous analyses are below detection limits and therefore do not offer a quantitative concentration.

Geochemistry of Orogenic Gold Deposits

Samples with gold values above 0.5 parts per million (ppm; 0.5 g/t or 0.16 ounces per short ton) cutoff (n=67) are plotted relative to average upper crustal abundance in figure 5. The median, minimum, and maximum concentrations of each element are listed in table 5.

Analyzed samples contain high concentrations of antimony, arsenic, cadmium, cobalt, copper, iron, molybdenum, silver, tellurium, tungsten, and zinc. Gold concentrations are plotted against the concentrations of some select critical minerals (arsenic, bismuth, cobalt,

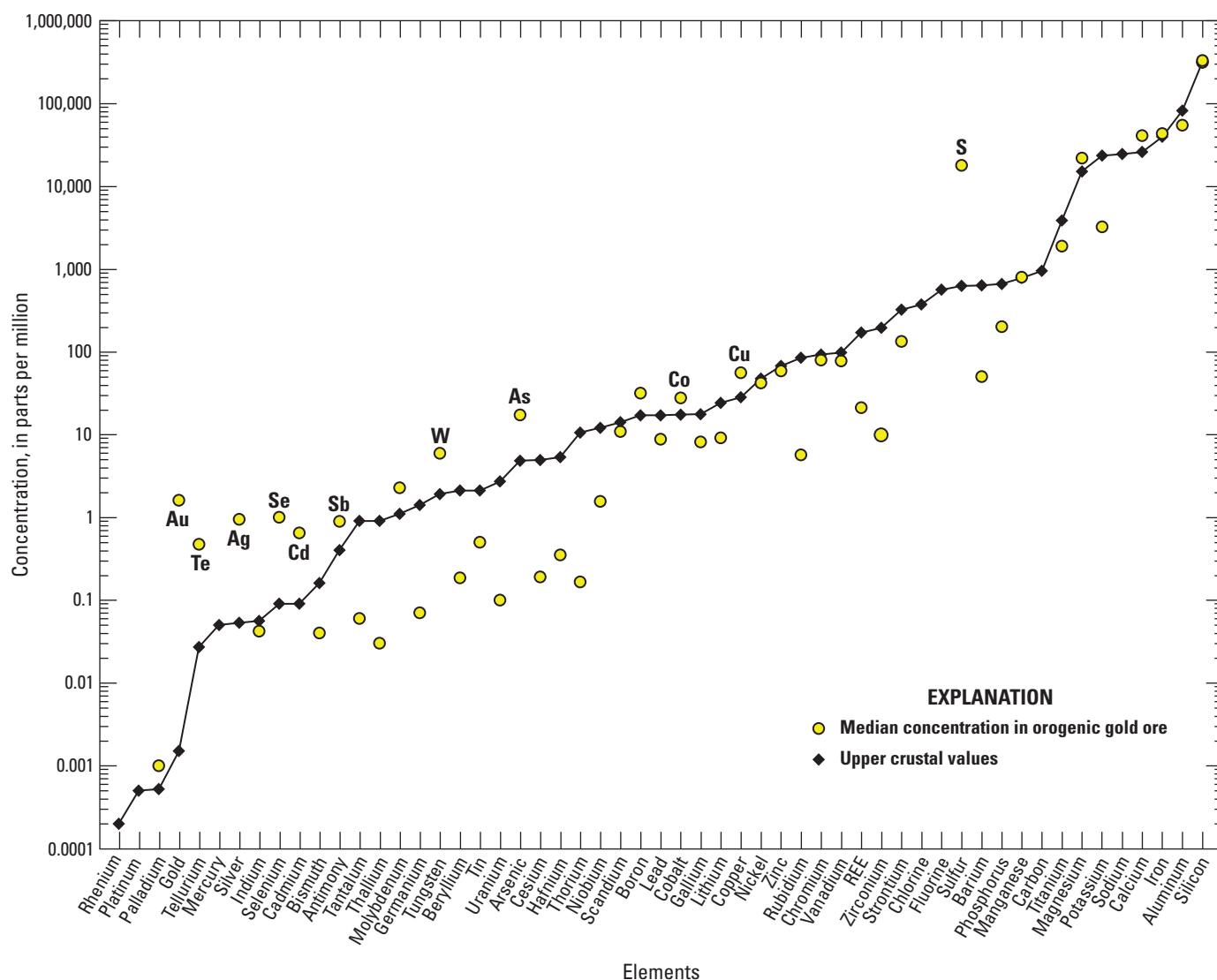


Figure 5. Graph showing the median concentrations of elements in orogenic gold ore with gold concentrations greater than or equal to 0.5 parts per million (equivalent to 0.5 grams per tonne; n=67) relative to the upper crust. Important potential byproduct commodities are individually labeled. Based on geochemical data from Granitto and others (2020, 2021, 2025). Crustal abundance values are from Hu and Gao (2008) for tellurium, Israel Science and Technology (2007) for carbon, and Rudnick and Gao (2014) for all other elements. (Au, gold; Te, tellurium; Ag, silver; Se, selenium; Cd, cadmium; Sb, antimony; W, tungsten; As, arsenic; Co, cobalt; Cu, copper; S, sulfur; REE, rare earth elements)

antimony, tellurium, and tungsten) for all samples in figure 6. Correlation coefficients for critical minerals with gold are reported in table 6. A much stronger correlation between gold and tellurium may be found for individual deposits, especially those deposits hosted by local intrusions within metasedimentary rock-dominated orogens (Goldfarb and others, 2005, 2017). For example, gold mined at

the Kensington deposit, Alaska (not shown in fig. 1), is predominantly hosted in telluride minerals and would therefore have a very strong correlation between the two metals; no ore geochemistry for the Kensington deposit is available for this study, but analysis by Heinchon (2019) indicated a significant correlation between gold and tellurium recording a coefficient of determination (R^2) value of 0.82.

Table 5. Basic statistics on gold-rich ore (gold concentration greater than 0.5 parts per million) from orogenic gold deposits.

[Compositional data are from Granitto and others (2020, 2021, 2025); BDL, below detection limit; ppm, parts per million; wt. pct., weight percent; >, greater than. Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; As, arsenic; Au, gold; B, boron; Ba, barium; Be, beryllium; Bi, bismuth; Ca, calcium; Cd, cadmium; Ce, cerium; Co, cobalt; Cr, chromium; Cs, cesium; Cu, copper; Dy, dysprosium; Er, erbium; Eu, europium; Fe, iron; Ga, gallium; Gd, gadolinium; Ge, germanium; Hf, hafnium; Ho, holmium; In, indium; K, potassium; La, lanthanum; Li, lithium; Lu, lutetium; Mg, magnesium; Mn, manganese; Mo, molybdenum; Nb, niobium; Nd, neodymium; Ni, nickel; P, phosphorous; Pb, lead; Pd, palladium; Pr, praseodymium; Pt, platinum; Rb, rubidium; Re, rhenium; S, sulfur; Sb, antimony; Sc, scandium; Se, selenium; Si, silica; Sm, samarium; Sn, tin; Sr, strontium; Ta, tantalum; Tb, terbium; Te, tellurium; Th, thorium; Ti, titanium; Tl, thallium; Tm, thulium; U, uranium; V, vanadium; W, tungsten; Y, yttrium; Yb, ytterbium; Zn, zinc; Zr, zirconium]

Element	Number of analyses	Median	Minimum	Maximum
Ag ppm	67	0.90	BDL	331
Al wt. pct.	67	5.48	0.08	9.7
As ppm	67	16.50	BDL	131,000
Au ppm	67	1.60	0.50	1,320
B ppm	23	31	BDL	182
Ba ppm	67	50	5	979
Be ppm	67	0.19	BDL	0.51
Bi ppm	67	0.04	BDL	335
Ca wt. pct.	67	3.9	BDL	19.1
Cd ppm	67	0.64	BDL	802
Ce ppm	67	6.01	0.2	186
Co ppm	67	28.5	BDL	598.5
Cr ppm	67	76	BDL	749
Cs ppm	67	0.2	BDL	7
Cu ppm	67	58	BDL	4,200
Dy ppm	28	0.785	BDL	5.58
Er ppm	28	0.4	BDL	3.22
Eu ppm	28	0.24	BDL	3.94
Fe wt. pct.	67	4.44	0.13	27.1
Ga ppm	67	8.61	BDL	22.6
Gd ppm	28	0.795	BDL	8.37
Ge ppm	62	0.075	BDL	6
Hf ppm	67	0.3	BDL	8
Ho ppm	28	0.14	BDL	1.13
In ppm	62	0.044	BDL	1.4
K wt. pct.	67	0.33	BDL	4.09
La ppm	67	2.2	BDL	94.2
Li ppm	67	9.7	BDL	60
Lu ppm	28	BDL	BDL	0.55

Table 5. Basic statistics on gold-rich ore (gold concentration greater than 0.5 parts per million) from orogenic gold deposits.—Continued

[Compositional data are from Granitto and others (2020, 2021, 2025); BDL, below detection limit; ppm, parts per million; wt. pct., weight percent; >, greater than. Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; As, arsenic; Au, gold; B, boron; Ba, barium; Be, beryllium; Bi, bismuth; Ca, calcium; Cd, cadmium; Ce, cerium; Co, cobalt; Cr, chromium; Cs, cesium; Cu, copper; Dy, dysprosium; Er, erbium; Eu, europium; Fe, iron; Ga, gallium; Gd, gadolinium; Ge, germanium; Hf, hafnium; Ho, holmium; In, indium; K, potassium; La, lanthanum; Li, lithium; Lu, lutetium; Mg, magnesium; Mn, manganese; Mo, molybdenum; Nb, niobium; Nd, neodymium; Ni, nickel; P, phosphorous; Pb, lead; Pd, palladium; Pr, praseodymium; Pt, platinum; Rb, rubidium; Re, rhenium; S, sulfur; Sb, antimony; Sc, scandium; Se, selenium; Si, silica; Sm, samarium; Sn, tin; Sr, strontium; Ta, tantalum; Tb, terbium; Te, tellurium; Th, thorium; Ti, titanium; Tl, thallium; Tm, thulium; U, uranium; V, vanadium; W, tungsten; Y, yttrium; Yb, ytterbium; Zn, zinc; Zr, zirconium]

Element	Number of analyses	Median	Minimum	Maximum
Mg wt. pct.	67	2.49	BDL	12
Mn wt. pct.	67	0.0819	BDL	0.784
Mo ppm	67	2.07	BDL	1,255
Nb ppm	67	1.6	BDL	42
Nd ppm	28	1.7	0.1	72.9
Ni ppm	67	42	BDL	711
P wt. pct.	67	0.02	BDL	0.19
Pb ppm	67	8.7	BDL	61,700
Pd ppm	22	0.001	BDL	0.004
Pr ppm	28	0.395	BDL	21
Pt ppm	22	BDL	BDL	BDL
Rb ppm	67	6	0.2	210
Re ppm	62	BDL	BDL	0.03
S wt. pct.	67	1.77	BDL	29.5
Sb ppm	67	0.88	0.3	621,000
Sc ppm	67	10.9	BDL	38.3
Se ppm	62	1	BDL	270
Si wt. pct.	23	31.2	1.6	46.7
Sm ppm	28	0.65	BDL	12
Sn ppm	67	0.5	BDL	4
Sr ppm	67	131	2.1	2,920
Ta ppm	67	0.06	BDL	1.5
Tb ppm	28	0.12	BDL	1.06
Te ppm	62	0.445	BDL	40.9
Th ppm	67	0.17	BDL	29.2
Ti wt. pct.	67	0.199	BDL	2.24
Tl ppm	62	0.03	BDL	12
Tm ppm	28	BDL	BDL	0.52
U ppm	67	0.1	BDL	11
V ppm	67	78	BDL	284
W ppm	67	5.5	BDL	>200
Y ppm	67	7.8	BDL	27.3
Yb ppm	28	0.3	BDL	3.7
Zn ppm	67	60	BDL	31,400
Zr ppm	67	10.1	BDL	278

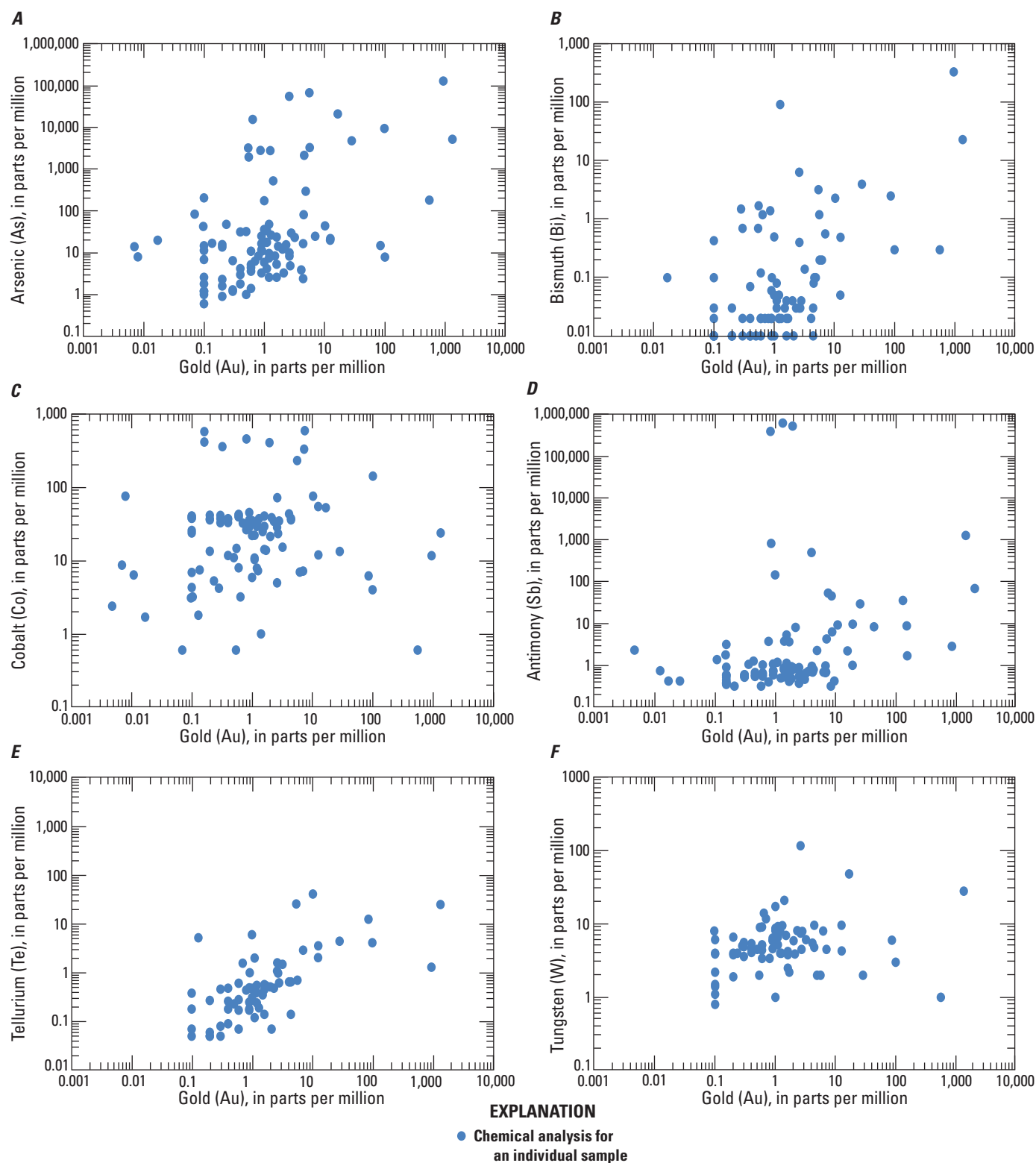


Figure 6. Log-log plots of gold plotted against select critical minerals *A*, arsenic, *B*, bismuth, *C*, cobalt, *D*, antimony, *E*, tellurium, and *F*, tungsten in ore from orogenic gold deposits.

Table 6. Correlation coefficients of select metals commonly enriched in orogenic gold deposits using all samples compiled in this study (n=150).

[Compositional data are from Granitto and others (2020, 2021, 2025); —, not applicable; Abbreviations of chemical elements as follows: Au, gold; As, arsenic; Bi, bismuth; Cd, cadmium; Co, cobalt; Cu, copper; Pb, lead; Sb, antimony; Te, tellurium; W, tungsten; Zn, zinc]

Element	Au	As	Bi	Cd	Co	Cu	Pb	Sb	Te	W	Zn
Au	—	0.46	0.56	0.72	−0.06	−0.01	0.66	−0.03	0.27	0.10	0.56
As	0.46	—	0.37	0.01	−0.01	−0.02	−0.01	−0.02	−0.01	−0.07	0.15
Bi	0.56	0.37	—	0.05	−0.03	0.15	0.10	0.14	0.17	−0.04	0.03
Cd	0.72	0.01	0.05	—	−0.03	0.03	0.92	0.00	−0.01	0.14	0.82
Co	−0.06	−0.01	−0.03	−0.03	—	0.11	−0.04	−0.07	−0.04	0.05	0.03
Cu	−0.01	−0.02	0.15	0.03	0.11	—	0.03	−0.01	0.11	−0.03	0.01
Pb	0.66	−0.01	0.10	0.92	−0.04	0.03	—	0.01	0.24	0.10	0.70
Sb	−0.03	−0.02	0.14	0.00	−0.07	−0.01	0.01	—	−0.02	−0.04	0.02
Te	0.27	−0.01	0.17	−0.01	−0.04	0.11	0.24	−0.02	—	−0.04	−0.01
W	0.10	−0.07	−0.04	0.14	0.05	−0.03	0.10	−0.04	−0.04	—	0.09
Zn	0.56	0.15	0.03	0.82	0.03	0.01	0.70	0.02	−0.01	0.09	—

Geochemistry of Coeur d'Alene-type Deposits

The median, minimum, and maximum concentrations of each element in silver-rich Coeur d'Alene-type ore samples with a cut off value of 250 ppm silver (n=44) are listed in [table 7](#) and are plotted relative to average upper crustal abundance in [figure 7](#). The median, minimum, and maximum concentrations of each element in zinc-rich Coeur d'Alene-type ore samples with a cut off value of 3 weight percent zinc (n=36) are listed in [table 8](#) and plotted relative to average upper crustal abundance in [figure 8](#).

Major ore minerals in Coeur d'Alene-type deposits include sphalerite, galena, tetrahedrite, stibnite, chalcopyrite, pyrite, and arsenopyrite ([table 3](#)). Other important minerals include boulangerite ($\text{Pb}_5\text{Sb}_4\text{S}_{11}$) and scheelite. Gersdorffite (NiAsS) is uncommon except for at the Silver Summit mine where it was a large enough component to be considered cobalt and nickel ore (Fryklund, 1964). Pyrrhotite ($\text{Fe}_{[1-X]}\text{S}$) from the Coeur d'Alene region contains per mille levels of cobalt and nickel (Fryklund and Harner, 1955). Common gangue mineralogy includes quartz, siderite, dolomite, and calcite. Antimony has been produced in the Coeur d'Alene region, and antimony contents are typically higher in silver-rich ores and in quartz-stibnite veins (Leach and others, 1998).

Notable enrichments of critical minerals compared to upper crustal abundance for silver-rich ore (silver concentration greater than 250 ppm) include antimony, arsenic, bismuth, indium, and manganese ([table 7](#)). Critical mineral enrichments relative to upper crustal abundance for zinc-rich ores (zinc concentration greater than 3 weight percent) include antimony, arsenic, bismuth, cobalt, germanium, indium, manganese, selenium, and zinc ([table 8](#)). Zinc is a primary commodity mined in Coeur d'Alene-type ores (Long, 1998a, b) but is still considered a critical mineral

(USGS, 2022a). Tungsten is elevated in a few samples that can be either silver- or zinc-rich but is more commonly found at levels near or below the detection limit.

Of the enriched critical minerals in Coeur d'Alene ore, only a few have median concentrations in the analyzed ore greater than 50 ppm. In silver-rich ore ([table 7](#)), the highest median values for the enriched critical minerals include antimony (median value of 1,635 ppm), arsenic (median value of 154.5 ppm), manganese (median value of 1.02 weight percent), and zinc (median value of 1.62 weight percent). In zinc-rich ore ([table 8](#)), the highest median values for the enriched critical minerals include antimony (median value of 323.5 ppm), manganese (median value of 0.41 weight percent), and zinc (median value of 18.5 weight percent). The median value for cobalt in zinc-rich ores analyzed in this study is 27.95 ppm; however, cobalt was stockpiled as a byproduct at the Bunker Hill zinc smelter (Davis and Buck, 1958). The concentration of the major commodities silver, lead, and zinc are plotted against the concentrations of select critical minerals (antimony, arsenic, bismuth, cobalt, indium, manganese, selenium, tungsten) in [figures 9–11](#), respectively, and the major commodities are plotted against each other in [figure 12](#). Chemical data important for silver-rich ores are plotted in [figure 13](#) for comparison with mineral chemistry data for tetrahedrite in [figure 14](#).

Using a limited dataset, Long (1998a) showed that there is neither a significant correlation among silver, lead, or zinc grades for lead-zinc veins, nor among silver, copper, and lead grades for silver veins. The additional dataset analyzed in this report supports the assertions made by Long (1998a). The strongest positive correlation coefficients with silver in Coeur d'Alene-type deposits ([table 4](#)) are for gold (0.80), copper (0.63), and palladium (0.63); the strongest positive correlation coefficients with zinc are sulfur (0.90), cadmium (0.77), and indium (0.46); the strongest positive correlation coefficient

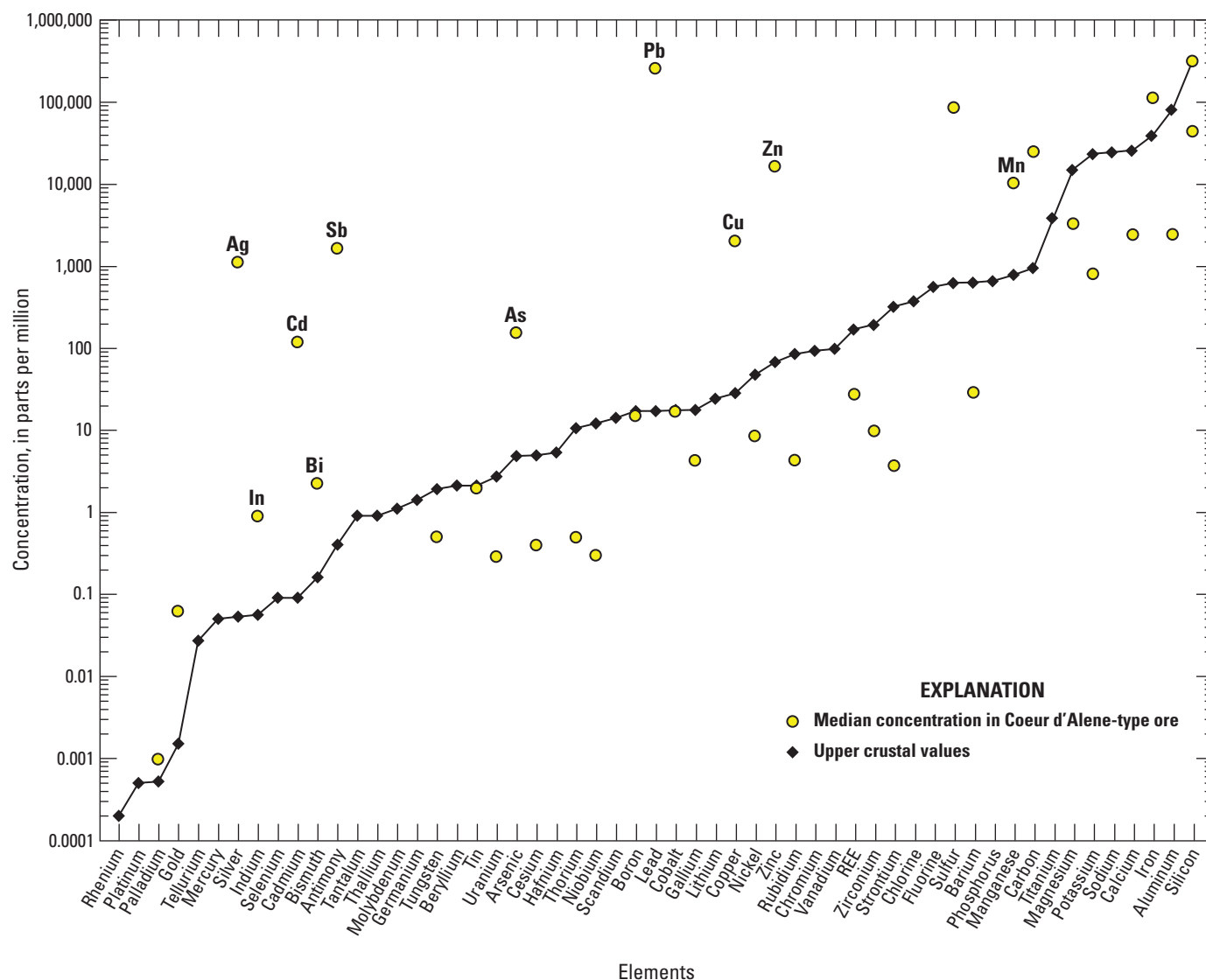


Figure 7. Graph showing the median concentrations of elements in silver-rich Coeur d'Alene-type ore with silver concentrations greater than 250 parts per million silver ($n=44$) relative to upper crust. Based on geochemical data from Granitto and others (2020, 2021, 2025). Crustal abundance values are from Hu and Gao (2008) for tellurium, Israel Science and Technology (2007) for carbon, and Rudnick and Gao (2014) for all other elements. (Ag, silver; In, indium; Cd, cadmium; Sb, antimony; Bi, bismuth; As, arsenic; Pb, lead; Cu, copper; Zn, zinc; Mn, manganese; REE, rare earth elements)

with lead is sulfur (0.27); and the strongest positive correlation coefficients with antimony are copper (0.75), gold (0.60), and silver (0.49).

Tetrahedrite ($[\text{Cu,Fe,Zn,Ag}]_{12}[\text{Sb,As}]_4\text{S}_{13}$) is the main ore mineral for silver (table 3). Although the chemical formula for the tetrahedrite series of minerals contains copper, iron, zinc, antimony, and arsenic, there is no demonstrated correlation with silver and iron or zinc in ore. There is a demonstrated positive correlation between silver and copper (0.63), and arsenic and antimony in ore (fig. 13) even though tetrahedrite crystals display negative correlations between silver and copper (-0.987) and arsenic and antimony (-0.958 ; fig. 14) because of elemental substitutions in the tetrahedrite crystal

structure (for example, Biagioni and others, 2020). Fryklund (1964) hypothesized that significant bismuth and mercury may be found within tetrahedrite ore, although the correlation coefficient between bismuth and silver is low and mercury was not analyzed.

Abundant carbonate phases, including siderite, dolomite, and calcite compose much of the gangue in Coeur d'Alene-type deposits. Calculated correlation coefficients between carbon and iron (0.77), carbon and manganese (0.80), and iron and manganese (0.67) indicate that siderite (FeCO_3) is the main residence of iron and manganese (table 4). Critical minerals that may be minor constituents in sphalerite include gallium, germanium, indium, and manganese (Cook and

Table 7. Basic statistics on silver-rich ore (silver concentration greater than 250 parts per million) from Coeur d'Alene-type deposits.

[Compositional data are from Granitto and others (2020, 2021, 2025). BDL, below detection limit; ppm, parts per million; wt. pct., weight percent; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; As, arsenic; Au, gold; B, boron; Ba, barium; Be, beryllium; Bi, bismuth; Ca, calcium; Cd, cadmium; Ce, cerium; Co, cobalt; Cr, chromium; Cs, cesium; Cu, copper; Dy, dysprosium; Er, erbium; Eu, europium; Fe, iron; Ga, gallium; Gd, gadolinium; Ge, germanium; Hf, hafnium; Ho, holmium; In, indium; K, potassium; La, lanthanum; Li, lithium; Lu, lutetium; Mg, magnesium; Mn, manganese; Mo, molybdenum; Nb, niobium; Nd, neodymium; Ni, nickel; P, phosphorous; Pb, lead; Pd, palladium; Pr, praseodymium; Pt, platinum; Rb, rubidium; Re, rhenium; S, sulfur; Sb, antimony; Sc, scandium; Se, selenium; Si, silica; Sm, samarium; Sn, tin; Sr, strontium; Ta, tantalum; Tb, terbium; Te, tellurium; Th, thorium; Ti, titanium; Tl, thallium; Tm, thulium; U, uranium; V, vanadium; W, tungsten; Y, yttrium; Yb, ytterbium; Zn, zinc; Zr, zirconium]

Element	Number of analyses	Median	Minimum	Maximum
Ag ppm	44	1,115	252	45,500
Al wt. pct.	44	0.24	BDL	6.25
As ppm	44	154.5	BDL	5,880
Au ppm	33	0.063	BDL	4.41
B ppm	43	15	BDL	1,730
Ba ppm	44	28.65	2.1	45,100
Be ppm	44	BDL	BDL	6
Bi ppm	44	2.25	0.1	2,390
C wt. pct.	29	2.46	0.04	9.22
Ca wt. pct.	44	0.24	BDL	17.9
Cd ppm	44	118.5	0.2	2,580
Ce ppm	44	5.3	BDL	78.2
Co ppm	44	17.25	BDL	240
Cr ppm	44	BDL	BDL	18
Cs ppm	44	0.4	BDL	22.4
Cu ppm	44	2,000	14	180,000
Dy ppm	44	1.635	BDL	29.2
Er ppm	44	0.83	BDL	12.8
Eu ppm	44	0.6	BDL	16.6
Fe wt. pct.	44	11.05	0.34	39
Ga ppm	44	4.315	0.23	20.6
Gd ppm	44	1.665	BDL	42.8
Ge ppm	44	BDL	BDL	12
Hf ppm	44	BDL	BDL	3
Ho ppm	44	0.305	BDL	4.97
In ppm	44	0.9	BDL	48.5
K wt. pct.	44	0.08	BDL	7.67
La ppm	44	3.3	BDL	35.1
Li ppm	44	BDL	BDL	59
Lu ppm	44	0.12	BDL	1.28
Mg wt. pct.	44	0.325	BDL	1.09
Mn wt. pct.	44	1.02	BDL	7.62
Mo ppm	44	BDL	BDL	21
Nb ppm	44	0.3	BDL	18.3
Nd ppm	44	3.55	BDL	65
Ni ppm	44	8.5	BDL	87
P wt. pct.	44	BDL	BDL	0.91
Pb ppm	44	253,000	35	769,000
Pd ppm	29	0.001	BDL	0.009

Table 7. Basic statistics on silver-rich ore (silver concentration greater than 250 parts per million) from Coeur d'Alene-type deposits.—Continued

[Compositional data are from Granitto and others (2020, 2021, 2025). BDL, below detection limit; ppm, parts per million; wt. pct., weight percent; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; As, arsenic; Au, gold; B, boron; Ba, barium; Be, beryllium; Bi, bismuth; Ca, calcium; Cd, cadmium; Ce, cerium; Co, cobalt; Cr, chromium; Cs, cesium; Cu, copper; Dy, dysprosium; Er, erbium; Eu, europium; Fe, iron; Ga, gallium; Gd, gadolinium; Ge, germanium; Hf, hafnium; Ho, holmium; In, indium; K, potassium; La, lanthanum; Li, lithium; Lu, lutetium; Mg, magnesium; Mn, manganese; Mo, molybdenum; Nb, niobium; Nd, neodymium; Ni, nickel; P, phosphorous; Pb, lead; Pd, palladium; Pr, praseodymium; Pt, platinum; Rb, rubidium; Re, rhenium; S, sulfur; Sb, antimony; Sc, scandium; Se, selenium; Si, silica; Sm, samarium; Sn, tin; Sr, strontium; Ta, tantalum; Tb, terbium; Te, tellurium; Th, thorium; Ti, titanium; Tl, thallium; Tm, thulium; U, uranium; V, vanadium; W, tungsten; Y, yttrium; Yb, ytterbium; Zn, zinc; Zr, zirconium]

Element	Number of analyses	Median	Minimum	Maximum
Pr ppm	44	0.68	BDL	12.1
Pt ppm	29	BDL	BDL	0.009
Rb ppm	44	4.3	0.2	410
Re ppm	35	BDL	BDL	0.003
S wt. pct.	43	8.4	0.2	31.5
Sb ppm	44	1,635	12.8	228,000
Sc ppm	44	BDL	BDL	BDL
Se ppm	43	BDL	BDL	1,440
Si wt. pct.	44	4.33	BDL	34.5
Sm ppm	44	1.6	BDL	29.3
Sn ppm	44	2	BDL	1,090
Sr ppm	44	3.7	BDL	2,060
Ta pp	44	BDL	BDL	BDL
Tb ppm	44	0.25	BDL	5.46
Te ppm	43	BDL	BDL	32.7
Th ppm	44	0.5	BDL	4.1
Ti wt. pct.	44	BDL	BDL	0.1
Tl ppm	44	BDL	BDL	6.8
Tm ppm	44	0.13	BDL	1.64
U ppm	44	0.29	BDL	6.34
V ppm	44	BDL	BDL	23
W ppm	44	0.5	BDL	1,530
Y ppm	44	6.7	BDL	141
Yb ppm	44	0.9	BDL	9.1
Zn ppm	44	16,200	24	523,000
Zr ppm	44	9.8	BDL	94.4

others, 2009). The weak negative correlation between zinc and manganese (-0.34) in ore indicates that manganese does not reside in sphalerite. The low correlation coefficients between zinc and gallium (0.19) and zinc and germanium (0.20), together with the low median concentrations of gallium (6 ppm) and germanium (2 ppm) in zinc ore, suggest that these critical minerals would unlikely be economic to recover.

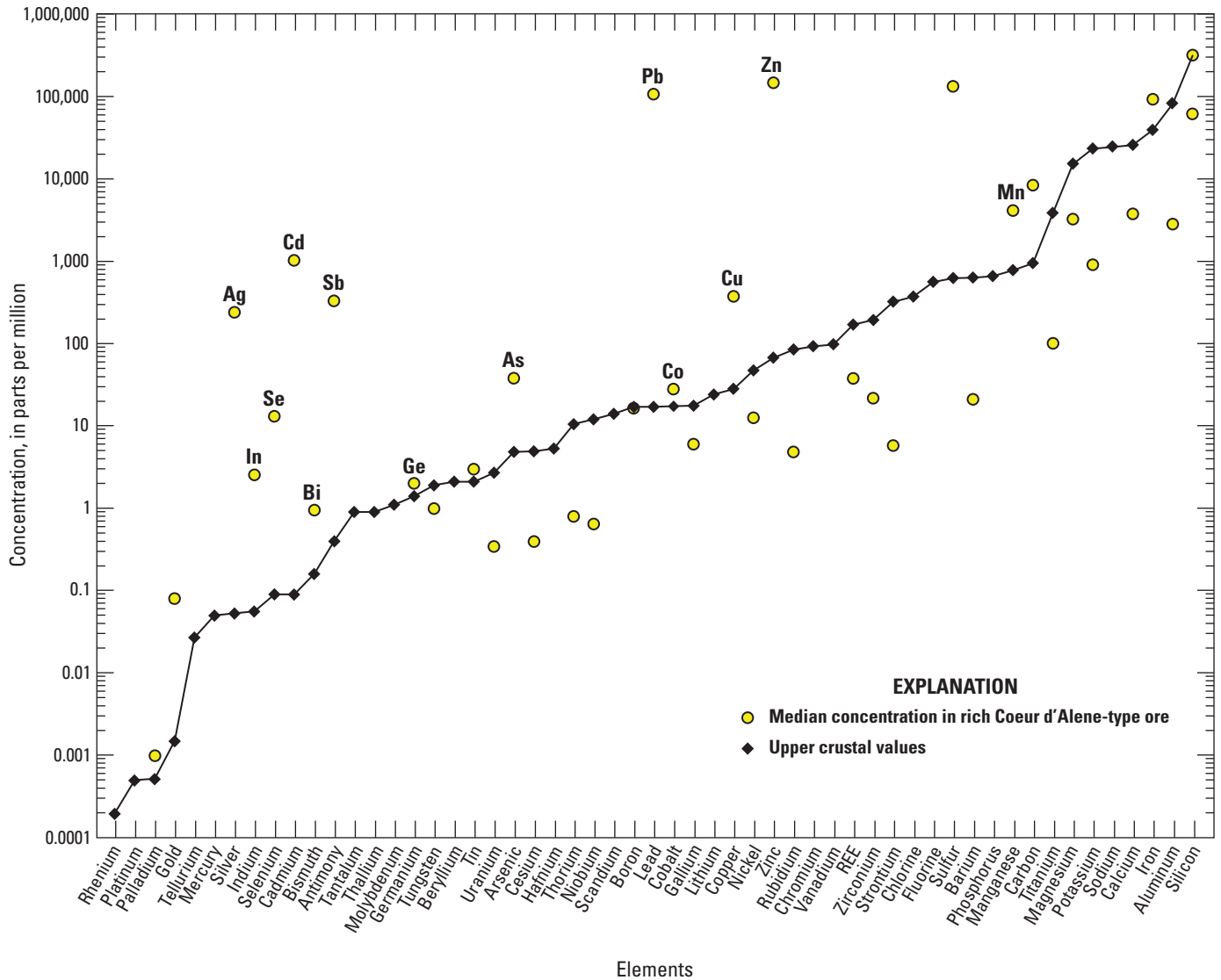


Figure 8. Graph showing the median concentrations of elements in zinc-rich Coeur d'Alene-type ore with zinc concentrations greater than 3 weight percent ($n=36$) relative to upper crust. Based on geochemical data from Granitto and others (2020, 2021, 2025). Crustal abundance values are from Hu and Gao (2008) for tellurium, Israel Science and Technology (2007) for carbon, and Rudnick and Gao (2014) for all other elements. (Ag, silver; In, indium; Se, selenium; Cd, cadmium; Bi, bismuth; Sb, antimony; Ge, germanium; As, arsenic; Pb, lead; Co, cobalt; Cu, copper; Zn, zinc; Mn, manganese)

Discussion—Potential Recovery of Critical Minerals

Pitcairn and others (2006) showed that many elements, including antimony, arsenic, cadmium, gold, silver, and tungsten are released from a rock package undergoing prograde metamorphism and are then concentrated within an ascending package of metamorphic-hydrothermal fluid. Because metals are introduced into mineralized veins from hydrothermal fluids, there may be compositional correlation among metals in orogenic gold deposits. However, some of these metals are liberated from different source rocks during

reactions occurring at different temperatures and, therefore, also precipitate out of solution at differing temperatures and depths compared to gold. For example, the lack of correlation between gold and antimony results from different depths and temperatures of their precipitation from the hydrothermal fluid (Groves and others, 1998).

Alternatively, some metals that are enriched and show a positive correlation with gold content are derived from the surrounding host rock. For example, Goldfarb and others (1997) and Haeberlin and others (2003, 2004) used lead isotopes of ore minerals in Phanerozoic deposits to show correlations with host rock and heterogeneities within individual districts that have diverse host rock types.

Table 8. Basic statistics on zinc-rich ore (with zinc concentration greater than 3 weight percent) from Coeur d'Alene-type deposits.

[Compositional data are from Granitto and others (2020, 2021, 2025). BDL, below detection limit; ppm, parts per million; wt. pct., weight percent; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; As, arsenic; Au, gold; B, boron; Ba, barium; Be, beryllium; Bi, bismuth; Ca, calcium; Cd, cadmium; Ce, cerium; Co, cobalt; Cr, chromium; Cs, cesium; Cu, copper; Dy, dysprosium; Er, erbium; Eu, europium; Fe, iron; Ga, gallium; Gd, gadolinium; Ge, germanium; Hf, hafnium; Ho, holmium; In, indium; K, potassium; La, lanthanum; Li, lithium; Lu, lutetium; Mg, magnesium; Mn, manganese; Mo, molybdenum; Nb, niobium; Nd, neodymium; Ni, nickel; P, phosphorous; Pb, lead; Pd, palladium; Pr, praseodymium; Pt, platinum; Rb, rubidium; Re, rhenium; S, sulfur; Sb, antimony; Sc, scandium; Se, selenium; Si, silica; Sm, samarium; Sn, tin; Sr, strontium; Ta, tantalum; Tb, terbium; Te, tellurium; Th, thorium; Ti, titanium; Tl, thallium; Tm, thulium; U, uranium; V, vanadium; W, tungsten; Y, yttrium; Yb, ytterbium; Zn, zinc; Zr, zirconium]

Element	Number of analyses	Median	Minimum	Maximum
Ag ppm	36	239.5	BDL	7,830
Al wt. pct.	35	0.28	BDL	3.53
As ppm	36	38	BDL	9,720
Au ppm	26	0.0805	0.002	1.32
B ppm	34	16.5	BDL	1,730
Ba ppm	36	20.9	BDL	1,070
Be ppm	36	BDL	BDL	BDL
Bi ppm	36	0.95	BDL	1,750
C wt. pct.	23	0.83	0.04	10
Ca wt. pct.	35	0.37	BDL	3.06
Cd ppm	36	1,014.5	0.07	2,580
Ce ppm	36	8.25	BDL	78.2
Co ppm	36	27.95	0.6	105
Cr ppm	36	BDL	BDL	18
Cs ppm	36	0.4	BDL	11.5
Cu ppm	36	372.5	35	103,000
Dy ppm	36	1.75	BDL	29.2
Er ppm	36	0.86	BDL	12.8
Eu ppm	36	0.685	BDL	16.6
Fe wt. pct.	35	9.14	2.49	36.8
Ga ppm	36	6.02	0.8	20.6
Gd ppm	36	1.89	BDL	42.8
Ge ppm	36	2	BDL	12
Hf ppm	36	BDL	BDL	5
Ho ppm	36	0.325	BDL	4.97
In ppm	36	2.55	BDL	48.5
K wt. pct.	35	0.09	BDL	1.67
La ppm	36	3.95	BDL	35.1
Li ppm	36	BDL	BDL	37
Lu ppm	36	0.115	BDL	1.28
Mg wt. pct.	35	0.32	BDL	1.42
Mn wt. pct.	35	0.41	0.0072	3.75
Mo ppm	36	BDL	BDL	2
Nb ppm	36	0.65	BDL	18.3
Nd ppm	36	4.7	BDL	65
Ni ppm	36	12.5	BDL	66
P wt. pct.	35	BDL	BDL	0.91
Pb ppm	36	105,500	BDL	617,000
Pd ppm	23	0.001	BDL	0.004

Table 8. Basic statistics on zinc-rich ore (with zinc concentration greater than 3 weight percent) from Coeur d'Alene-type deposits.—Continued

[Compositional data are from Granitto and others (2020, 2021, 2025). BDL, below detection limit; ppm, parts per million; wt. pct., weight percent; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; As, arsenic; Au, gold; B, boron; Ba, barium; Be, beryllium; Bi, bismuth; Ca, calcium; Cd, cadmium; Ce, cerium; Co, cobalt; Cr, chromium; Cs, cesium; Cu, copper; Dy, dysprosium; Er, erbium; Eu, europium; Fe, iron; Ga, gallium; Gd, gadolinium; Ge, germanium; Hf, hafnium; Ho, holmium; In, indium; K, potassium; La, lanthanum; Li, lithium; Lu, lutetium; Mg, magnesium; Mn, manganese; Mo, molybdenum; Nb, niobium; Nd, neodymium; Ni, nickel; P, phosphorous; Pb, lead; Pd, palladium; Pr, praseodymium; Pt, platinum; Rb, rubidium; Re, rhenium; S, sulfur; Sb, antimony; Sc, scandium; Se, selenium; Si, silica; Sm, samarium; Sn, tin; Sr, strontium; Ta, tantalum; Tb, terbium; Te, tellurium; Th, thorium; Ti, titanium; Tl, thallium; Tm, thulium; U, uranium; V, vanadium; W, tungsten; Y, yttrium; Yb, ytterbium; Zn, zinc; Zr, zirconium]

Element	Number of analyses	Median	Minimum	Maximum
Pr ppm	36	1.075	BDL	12.1
Pt ppm	23	BDL	BDL	0.006
Rb ppm	36	4.85	0.2	90.4
Re ppm	26	BDL	BDL	0.003
S wt. pct.	33	13.1	3.5	31.5
Sb ppm	36	323.5	2.7	228,000
Sc ppm	36	BDL	BDL	BDL
Se ppm	34	13	BDL	251
Si wt. pct.	33	8.83	0.13	26.6
Sm ppm	36	1.6	BDL	29.3
Sn ppm	36	3	BDL	1,090
Sr ppm	36	5.75	BDL	243
Ta pp	36	BDL	BDL	BDL
Tb ppm	36	0.295	BDL	5.46
Te ppm	34	BDL	BDL	30
Th ppm	36	0.8	BDL	6.1
Ti wt. pct.	35	0.01	BDL	0.14
Tl ppm	36	BDL	BDL	6.8
Tm ppm	36	0.125	BDL	1.64
U ppm	36	0.345	BDL	6.34
V ppm	36	BDL	BDL	87
W ppm	36	1	BDL	1,530
Y ppm	36	7.8	BDL	141
Yb ppm	36	0.8	BDL	9.1
Zn ppm	36	185,000	49,700	541,000
Zr ppm	36	21.5	BDL	160

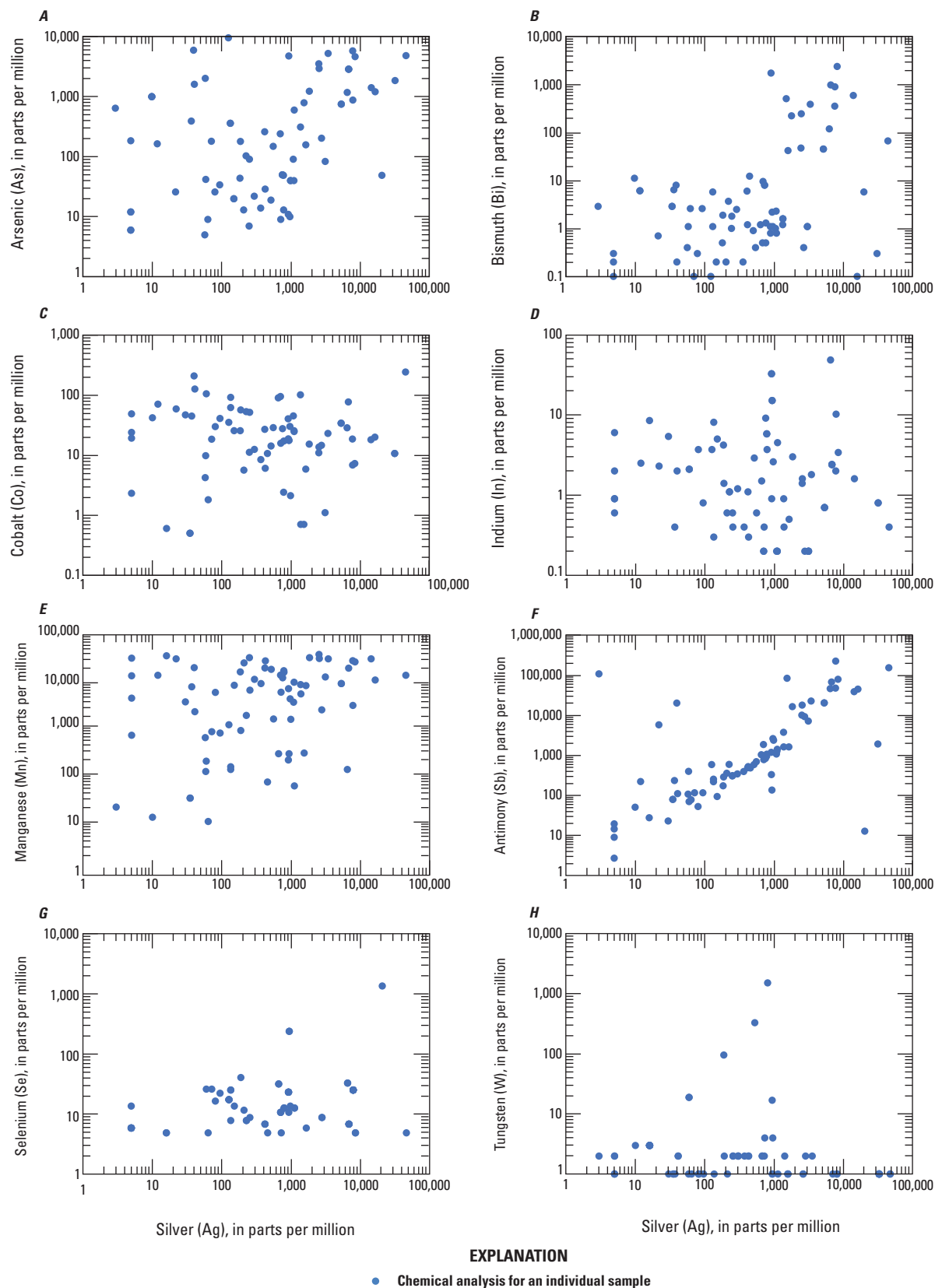


Figure 9. Log-log plots of silver plotted against select critical minerals *A*, arsenic, *B*, bismuth, *C*, cobalt, *D*, indium, *E*, manganese, *F*, antimony, *G*, selenium, and *H*, tungsten in ore from Coeur d'Alene-type deposits. Based on geochemical data from Granitto and others (2020, 2021, 2025).

Table 9. Estimated tonnes of contained gold, arsenic, cobalt, antimony, tellurium, and tungsten from the Idaho-Maryland mine in Grass Valley, California.

[Element ratios are median ratios. Total commodity value based on calculated total content and 2021 annual average price for the commodity (U.S. Geological Survey, 2022b). Compositional data are from Granitto and others (2020, 2021, 2025), production data are from Kulla (2017), and resource data are from Pease (2009). t, tonnes; USD, U.S. dollars; /, per; Abbreviations of chemical elements as follows: As, arsenic; Au, gold; Co, cobalt; Sb, antimony; Te, tellurium; W, tungsten]

Metric	Estimation
Gold	
Total t	96.84
Total value	\$5,604,373,000 USD
Production t	68.44
Resources t	28.41
Median ratios	
As/Au	6.15
Co/Au	18.43
Sb/Au	0.59
Te/Au	0.28
W/Au	4.67
Arsenic	
Total t	595.57
Total value	\$1,240,000 USD
Tailings t	420.91
Unmined t	174.72
Cobalt	
Total t	1784.62
Total value	\$4,697,000 USD
Tailings t	1261.25
Unmined t	523.56
Antimony	
Total t	56.89
Total value	\$1,110,000 USD
Tailings t	40.21
Unmined t	16.69
Tellurium	
Total t	27.44
Total value	\$14,518,000 USD
Tailings t	19.39
Unmined t	8.05
Tungsten	
Total t	451.91
Total value	\$26,000 USD
Tailings t	319.38
Unmined t	132.58

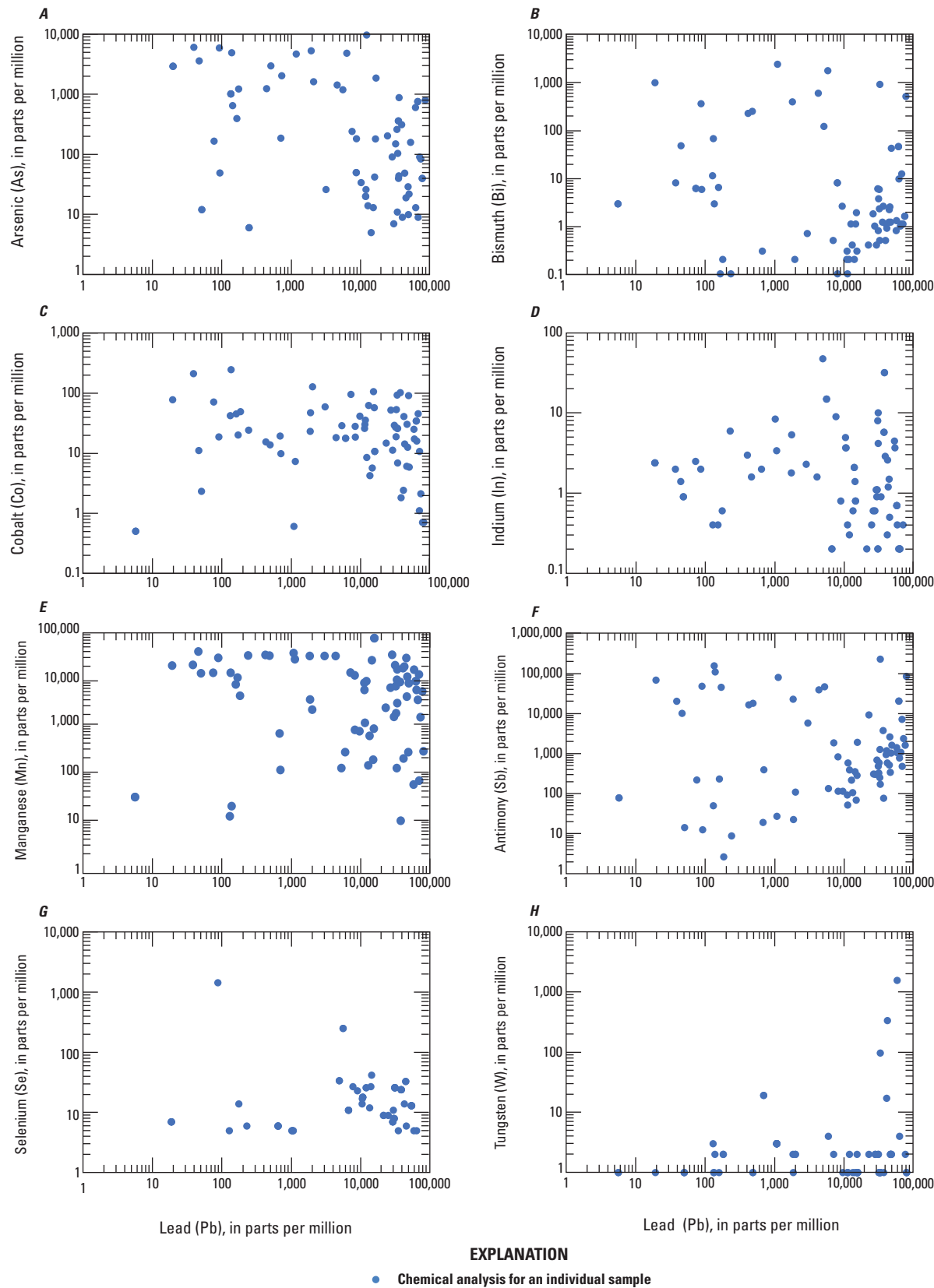


Figure 10. Log-log plots of lead plotted against select critical minerals *A*, arsenic, *B*, bismuth, *C*, cobalt, *D*, indium, *E*, manganese, *F*, antimony, *G*, selenium, and *H*, tungsten in ore from Coeur d'Alene-type deposits. Based on geochemical data from Granitto and others (2020, 2021, 2025).

Table 10. Estimated tonnes of manganese, antimony, and arsenic released in mine waste from silver-rich ore from the Coeur d'Alene area.

[Element ratios are median ratios. Total commodity value based on calculated total content and 2021 annual average price for the commodity (U.S. Geological Survey, 2022b). Compositional data are from Granitto and others (2020, 2021, 2025) and production data are from Long (1998b); t, tonnes; USD, U.S. dollars; /, per; Abbreviations of chemical elements as follows: Ag, silver; As, arsenic; Mn, manganese; Sb, antimony; Zn, zinc]

Element	Estimation
Production Ag t	34,300
Total Ag value	\$27,569,225,250 USD
Mn/Ag	4.31
Sb/Ag	1.33
As/Ag	0.08
Tailings Mn t	147,886.23
Total Mn value	\$769,000 USD
Tailings Sb t	45,588.12
Total Sb value	\$522,623,000 USD
Tailings As t	2,707.58
Total As value	\$3,466,000 USD

Serpentinite host rocks would therefore offer ores that may be more enriched in metals like nickel and cobalt than ores hosted in granitoids (Taylor and others, 2021).

The most enriched critical minerals within orogenic mineral systems include antimony, arsenic, cobalt, manganese, tellurium, and tungsten. Arsenic occurs in common arsenopyrite and arsenic-rich pyrite; antimony is in stibnite; tellurium is in gold-, silver-, lead-, and nickel-bearing telluride minerals; and tungsten is in scheelite (table 3). Elemental correlations between gold and metals sourced in the host rock, such as lead, nickel, and cobalt, are dependent upon the type of host rock and relate to fluid–rock interaction that leads to gold precipitation along with other ore minerals such as galena and nickel- and cobalt-bearing pyrite. Cobalt and nickel are in growth zones in pyrite that are interpreted to indicate periods of significant fluid–rock interaction in veins hosted by mafic to ultramafic rocks, as exemplified at Grass Valley, Calif. (Taylor and others, 2021). Local fluid–rock interaction has also been interpreted as the cause of cobalt mineralization in orogenic gold deposits in northern Finland (Witt and others, 2020). Additionally, cobalt may be in pyrrhotite, pentlandite, or as cobaltite. The data in this study indicate that antimony, arsenic, cobalt, tellurium, and tungsten could potentially be recovered from orogenic gold ores.

Tungsten (for example, in Grass Valley, Calif.) and antimony (for example, in Woxi, China) have been produced from orogenic gold mines and continue to be a potential critical mineral byproduct of future resource extraction. Although gold tellurides may comprise a significant gold resource in many orogenic gold deposits, tellurium has not been produced (Goldfarb and others, 2017).

Using the ratio between median concentrations of critical minerals (CM) to gold (CM/Au), and multiplying this ratio by gold production and resources, an estimate of critical minerals

that could potentially be recovered from orogenic gold deposits can be reached. Because byproducts are produced from ore, only data for samples with more than 0.5 ppm gold are used. We focus on the Idaho-Maryland mine because 41 of the 66 samples with gold values greater than the cutoff are from the Idaho-Maryland mine, and its production and resources are well constrained unlike historical gold mines in California.

From 1866 to 1955, the Idaho-Maryland mine produced 2,414,000 ounces of gold (Kulla, 2017). Total measured plus indicated mineral resources as of 2007 for the Idaho-Maryland mine indicate an additional 1,002,000 ounces of gold (Pease, 2009). In total, production plus measured plus indicated mineral resources equals 3,416,000 ounces of gold. The potential dollar value based upon the annual average commodity price from 2021 (USGS, 2022b) for antimony, arsenic, cobalt, tellurium, and tungsten from tailings (previous gold production) and remaining resources is more than \$21.5 million U.S. dollars (table 9).

This study indicates that the zinc-rich veins in Coeur d'Alene-type mineral systems contain insignificant amounts of bismuth (median value of 0.95 ppm), gallium (median value of 6.02 ppm), germanium (median value of 2 ppm), and indium (median value of 2.55 ppm), although zinc is considered a critical mineral and is found at elevated concentrations (table 8). However, manganese and antimony and minor amounts of arsenic could be recovered from silver-rich veins if mineral processing procedures are optimized for their recovery. Antimony resources also reside in stibnite veins.

Using the ratio of median concentrations of critical minerals to those of silver and zinc (CM/Ag and CM/Zn) and multiplying these ratios by silver and zinc production and resources, respectively, allows for the calculation of an estimate of critical minerals that could potentially be

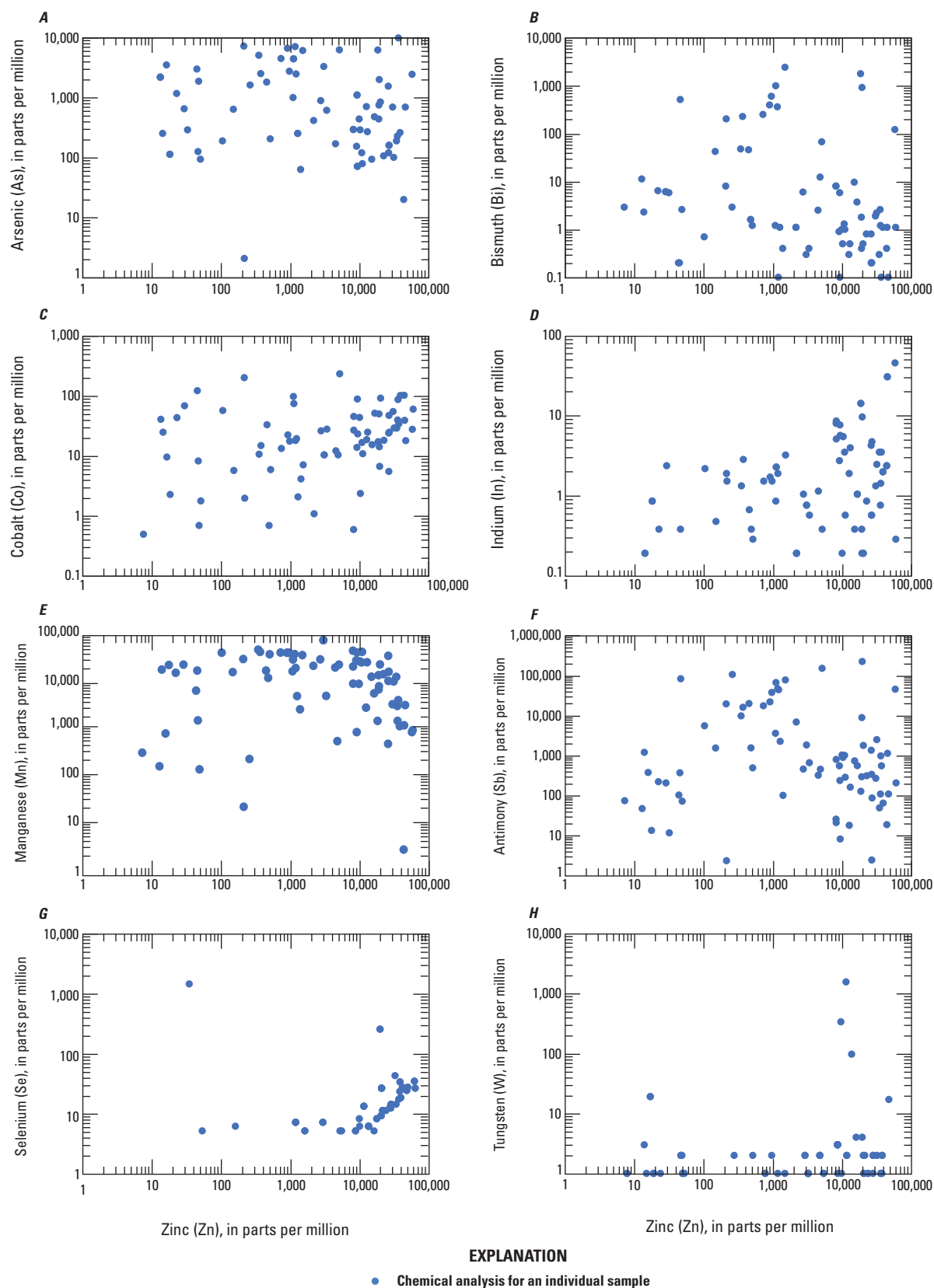


Figure 11. Log-log plots of zinc plotted against select critical minerals *A*, arsenic, *B*, bismuth, *C*, cobalt, *D*, indium, *E*, manganese, *F*, antimony, *G*, selenium, and *H*, tungsten in ore from Coeur d'Alene-type deposits. Based on geochemical data from Granitto and others (2020, 2021, 2025).

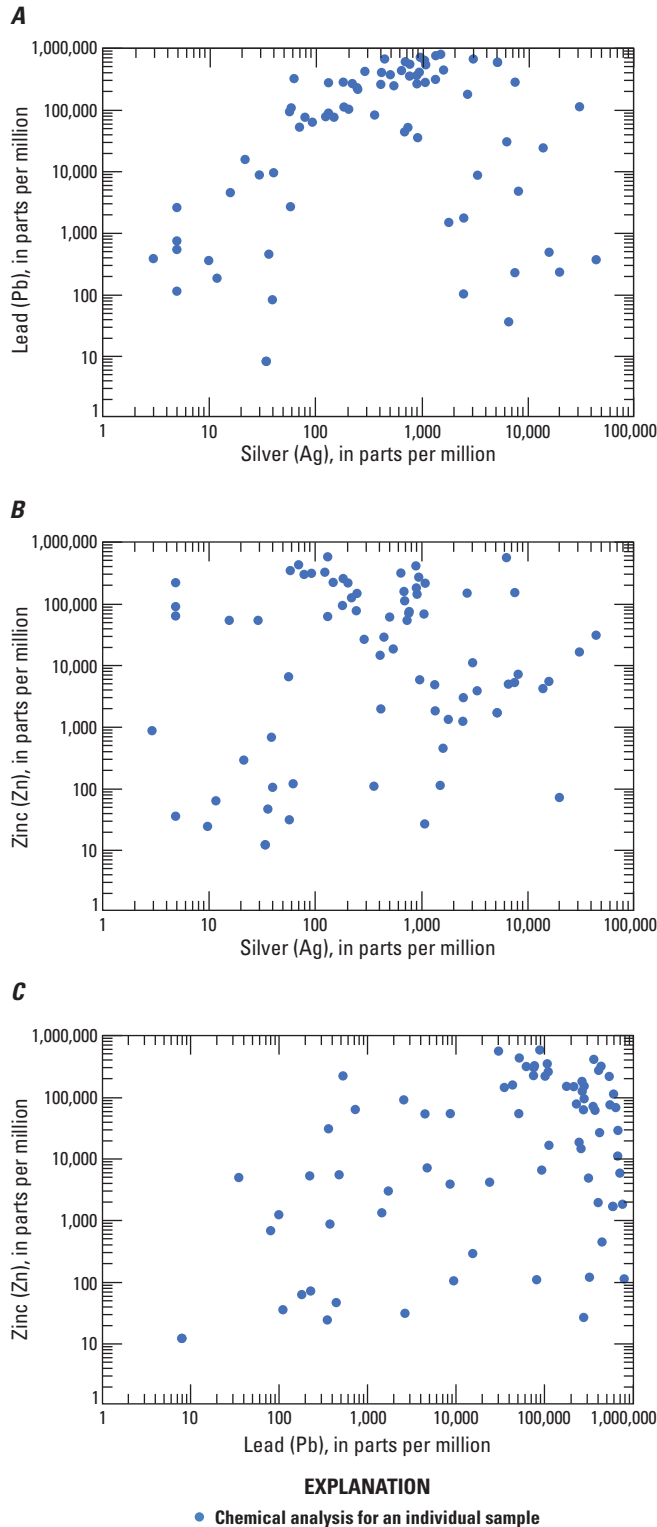


Figure 12. Log-log plots of the major commodities *A*, lead and silver, *B*, zinc and silver, and *C*, zinc and lead in ore from Coeur d’Alene-type deposits. Based on geochemical data from Granitto and others (2020, 2021, 2025).

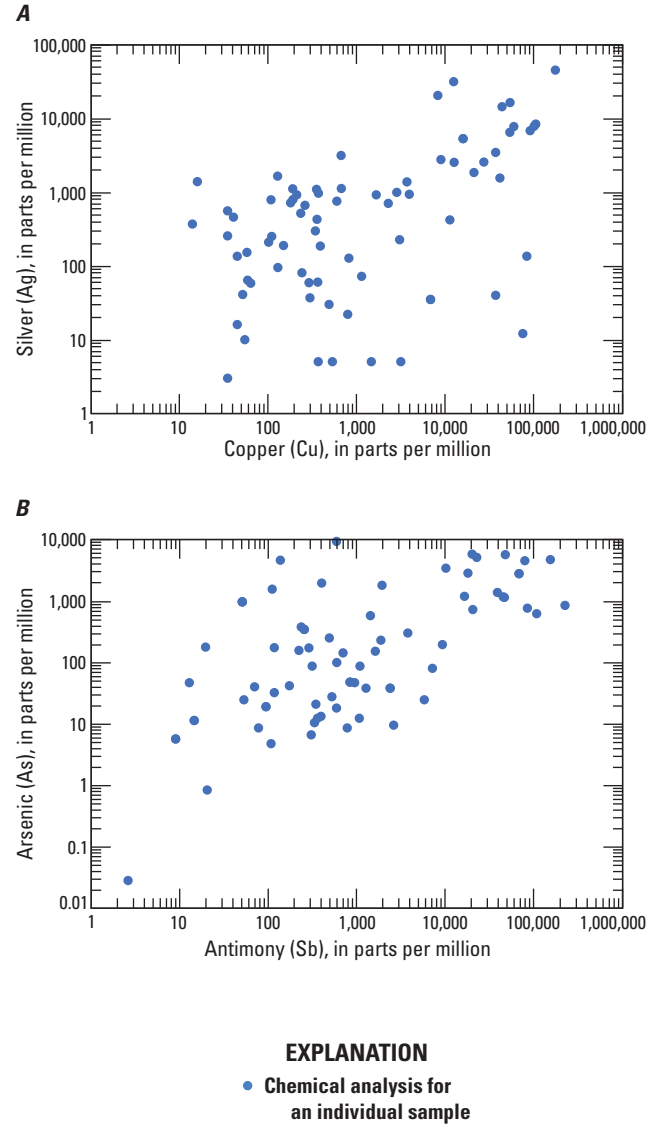


Figure 13. Log-log plots displaying positive correlations between *A*, silver and copper and *B*, arsenic and antimony in silver-rich veins from Coeur d’Alene-type deposits. Based on geochemical data from Granitto and others (2020, 2021, 2025).

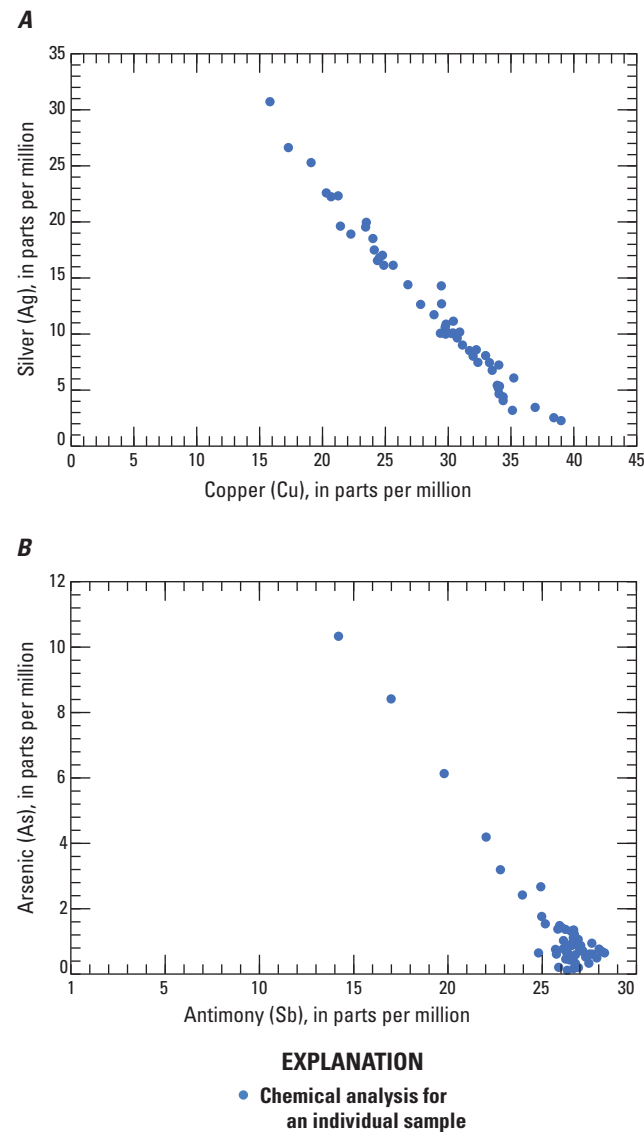


Figure 14. Log-log plots of *A*, silver plotted against copper and *B*, arsenic plotted against antimony for tetrahedrite from Coeur d’Alene-type deposits. Data from O’Leary and Sack (1987), Sack and others (2002, 2005), and references therein.

recovered as byproducts from Coeur d’Alene-type mineral systems. Because byproducts are produced from ore, only data for samples greater than 250 parts per million silver are used. Our dataset includes 44 silver-rich samples from the Coeur d’Alene district that are used for these calculations (table 10).

From the discovery of the Coeur d’Alene ores in 1883 through 1997, 34,300 tonnes of silver and 2,870,000 tonnes of zinc were produced (Long, 1998b). Production has continued since 1997 and is still ongoing as of 2025, so these values are minimums, and additional resources still exist. For our calculations, however, the 1883–1997 production values from Long (1998b) will be used for an estimate of antimony, arsenic, and manganese contents. Although a few mines

produced antimony as a primary commodity, our calculated antimony values are only as a byproduct of silver-rich ore. The U.S. Antimony mine, which is considered the second largest quartz-stibnite vein deposit in the United States, has production and reserves of 15,400 tonnes of antimony (Hofstra and others, 2013). For comparison, our calculations estimate more than 45,500 tonnes of antimony as a potential byproduct of silver mining in Coeur d’Alene-type polymetallic sulfide ore. The potential dollar value based upon the annual average commodity price from 2021 (USGS, 2022b) for antimony, arsenic, and manganese for these critical mineral quantities that may remain in the tailings and waste is more than \$525 million U.S. dollars. In addition, a minimum of 650,000 tonnes of zinc with a potential dollar value of more than \$2 billion U.S. dollars (based on the annual average commodity price from 2021 [\$1.45 per pound]) were disposed of downstream of mining activity (Long, 1998b).

Tailings found in the Coeur d’Alene region may be isolated in tailings impoundments, but a large total mass of the historically produced tailings were released and reside in the downstream channels, floodplains, and interconnected lakes of the Coeur d’Alene River Basin. Of the approximately 110 million tonnes of tailings material that was produced into the 1990s, it was estimated that 50–60 percent was discharged into the Coeur d’Alene River or its associated tributaries, and the remainder remains in dumps or impoundments or was backfilled into underground mine workings (Long, 1998b). The deleterious effects toward the environment of the downstream disposal of tailings necessitates proper mitigation, and superfund cleanup has already begun to consolidate tailings and waste rock from the Coeur d’Alene River Basin into centralized impoundment areas (U.S. Environmental Protection Agency, 2017). Reprocessing this material for primary commodities, along with the expected quantities of critical minerals, could potentially contribute to funding this cleanup.

Conclusions

Deposit types in orogenic gold and Coeur d’Alene-type systems are exploited for precious and base metals, but potential exists to also produce critical minerals to meet current (2025) societal needs in the United States. Publicly available geochemical data compiled for these mineral systems coupled with mineralogical characteristics, indicate that the following minerals could potentially be recovered from unmined resources and processed mine waste: arsenic, antimony, tellurium, cobalt, and tungsten from orogenic gold systems and zinc, antimony, arsenic, and manganese from Coeur d’Alene-type systems. These critical minerals reside primarily in arsenopyrite (arsenic), scheelite (tungsten), siderite (manganese), sphalerite (zinc), tetrahedrite (antimony and arsenic), stibnite (antimony), and telluride (tellurium) minerals.

References Cited

- Adams, M., ed., 2016, Gold ore processing—Project development and operations (2d ed.): Amsterdam, Netherlands, Elsevier, 1,040 p.
- Beaudoin, G., and Sangster, D.F., 1992, A descriptive model for silver-lead-zinc veins in clastic metasedimentary terranes: *Economic Geology*, v. 87, no. 4, p. 1005–1021, accessed November 2023 at <https://doi.org/10.2113/gsecongeo.87.4.1005>.
- Bennett, E.H., and Venkatakrishnan, R., 1982, A palinspastic reconstruction of the Coeur d'Alene mining district based on ore deposits and structural data: *Economic Geology*, v. 77, no. 8, p. 1851–1866, accessed November 2023 at <https://doi.org/10.2113/gsecongeo.77.8.1851>.
- Biagioni, C., George, L.L., Cook, N.J., Makovicky, E., Mořlo, Y., Pasero, M., Sejkora, J., Stanley, C.J., Welch, M.D., and Bosi, F., 2020, The tetrahedrite group—Nomenclature and classification: *American Mineralogist*, v. 105, no. 1, p. 109–122, accessed November 2023 at <https://doi.org/10.2138/am-2020-7128>.
- Bliss, J.D., 1986, Grade and tonnage model of low-sulphide Au-quartz veins, in Cox, D.P. and Singer, D.A., eds., *Mineral deposit models: U.S. Geological Survey Bulletin 1693*, p. 239–243, accessed September 2023 at <https://pubs.usgs.gov/bul/b1693/>.
- Břohlke, J.K., 1988, Carbonate-sulfide equilibria and “stratabound” disseminated epigenetic gold mineralization—A proposal based on examples from Alleghany, California, U.S.A.: *Applied Geochemistry*, v. 3, no. 5, p. 499–516, accessed September 2023 at [https://doi.org/10.1016/0883-2927\(88\)90022-4](https://doi.org/10.1016/0883-2927(88)90022-4).
- Boleneus, D.E., Applegate, L.M., Stewart, J.H., and Zientek, M.L., 2005, Stratabound copper-silver deposits of the Mesoproterozoic Revett Formation, Montana and Idaho: U.S. Geological Survey Scientific Investigations Report 2005–5231, 60 p., 3 pls., accessed November 2023 at <https://doi.org/10.3133/sir20055231>.
- Bundtzen, T.K., 1981, Geology and mineral deposits of the Kantishna Hills, Mt. McKinley quadrangle, Alaska: College, Alaska, University of Alaska, M.S. thesis, 238 p.
- Constantopoulos, J., 1994, Oxygen isotope geochemistry of the Coeur d'Alene mining district, Idaho: *Economic Geology*, v. 89, no. 4, p. 944–951, accessed November 2023 at <https://doi.org/10.2113/gsecongeo.89.4.944>.
- Constantopoulos, J., and Larson, P.B., 1991, Oxygen and hydrogen isotope geochemistry of the Star-Morning mine, Coeur d'Alene district, Idaho: *Geology*, v. 19, no. 2, p. 131–134, accessed November 2023 at [https://doi.org/10.1130/0091-7613\(1991\)019<0131:OAHIGO>2.3.CO;2](https://doi.org/10.1130/0091-7613(1991)019<0131:OAHIGO>2.3.CO;2).
- Cook, N.J., Ciobanu, C.L., Pring, A., Skinner, W., Shimizu, M., Danyushevsky, L., Saini-Eidukat, B., and Melcher, F., 2009, Trace and minor elements in sphalerite—A LA-ICPMS study: *Geochimica et Cosmochimica Acta*, v. 73, no. 16, p. 4761–4791, accessed December 2023 at <https://doi.org/10.1016/j.gca.2009.05.045>.
- Cox, D.P., and Singer, D.A., eds., 1986, *Mineral deposit models: U.S. Geological Survey Bulletin 1693*, [variously pagged], accessed September 2023 at <https://pubs.usgs.gov/bul/b1693/>.
- Cox, S.F., 2016, Injection-driven swarm seismicity and permeability enhancement—Implications for the dynamics of hydrothermal ore systems in high fluid-flux, overpressured faulting regimes—An invited paper: *Economic Geology*, v. 111, no. 3, p. 559–587, accessed September 2023 at <https://doi.org/10.2113/econgeo.111.3.559>.
- Cox, S.F., Knackstedt, M.A., and Baun, J., 2001, Principles of structural control on permeability and fluid flow in hydrothermal systems: *Reviews in Economic Geology*, v. 14, p. 1–24, accessed September 2023 at <https://doi.org/10.5382/Rev.14.01>.
- Cox, S.F., Wall, V.J., Etheridge, M.A., and Potter, T.F., 1991, Deformational and metamorphic processes in the formation of mesothermal vein-hosted gold deposits—Examples from the Lachlan fold belt in central Victoria, Australia: *Ore Geology Reviews*, v. 6, no. 5, p. 391–423, accessed September 2023 at [https://doi.org/10.1016/0169-1368\(91\)90038-9](https://doi.org/10.1016/0169-1368(91)90038-9).
- Davies, R.S., Groves, D.I., Trench, A., Dentith, M., and Sykes, J.P., 2020, Appraisal of the USGS Three-Part Mineral Resource Assessment through estimation of the orogenic gold endowment of the Sandstone Greenstone Belt, Yilgarn Craton, Western Australia: *Mineralium Deposita*, v. 55, p. 1009–1028, accessed December 2023 at <https://doi.org/10.1007/s00126-019-00916-1>.
- Davis, H.W., and Buck, C.R., 1958, Cobalt, *in* *Metals and minerals (except fuels): U.S. Bureau of Mines Minerals Yearbook 1956*, v. 1, p. 379–392.
- Day, W.C., 2019, The Earth Mapping Resources Initiative (Earth MRI)—Mapping the Nation's critical mineral resources (ver. 1.2, September 2019): U.S. Geological Survey Fact Sheet 2019–3007, 2 p., accessed September 11, 2023, at <https://doi.org/10.3133/fs20193007>.
- Dominy, S.C., 2014, Predicting the unpredictable—Evaluating high-nugget effect gold deposits, in *Monitoring and exploiting the reserve*, chap. 8 of *Mineral resource and ore reserve estimation: Melbourne, Australia, Australasian Institute of Mining and Metallurgy*, p. 659–678, accessed February 2025 at http://cdn.ceo.ca.s3-us-west-2.amazonaws.com/lesojp7-2014_D_NuggetevaluationMONO30%20%281%29.pdf.

- Ellis, S., and Deschênes, G., 2016, Chapter 51—Treatment of gold-telluride ores, chap. 51 of Adams, M.D. ed., *Gold ore processing—Project development and operations* (2d ed.): Elsevier, p. 919–926, accessed September 2023 at <https://doi.org/10.1016/B978-0-444-63658-4.00051-7>.
- Elmer, F.L., White, R.W., and Powell, R., 2006, Devolatilization of metabasic rocks during greenschist-amphibolite facies metamorphism: *Journal of Metamorphic Geology*, v. 24, no. 6, p. 497–513, accessed September 2023 at <https://doi.org/10.1111/j.1525-1314.2006.00650.x>.
- Fleck, R.J., Criss, R.E., Eaton, G.F., Cleland, R.W., Wavra, C.S., and Bond, W.D., 2002, Age and origin of base and precious metal veins of the Coeur d'Alene mining district, Idaho: *Economic Geology*, v. 97, no. 1, p. 23–42, accessed November 2023 at <https://doi.org/10.2113/gsecongeo.97.1.23>.
- Fortier, S.M., Nassar, N.T., Lederer, G.W., Brainard, J., Gambogi, J., and McCullough, E.A., 2018, Draft critical mineral list—Summary of methodology and background information—U.S. Geological Survey technical input document in response to Secretarial Order No. 3359: U.S. Geological Survey Open-File Report 2018–1021, 15 p., accessed April 18, 2020, at <https://doi.org/10.3133/ofr20181021>.
- Frimmel, H.E., 2008, Earth's continental crustal gold endowment: *Earth and Planetary Science Letters*, v. 267, no. 1–2, p. 45–55, accessed September 2023 at <https://doi.org/10.1016/j.epsl.2007.11.022>.
- Fryklund, V.C., 1964, Ore deposits of the Coeur d'Alene district, Shoshone County, Idaho: U.S. Geological Survey Professional Paper 445, 103 p., accessed November 2023 at <https://doi.org/10.3133/pp445>.
- Fryklund, V.C., and Harner, R.S., 1955, Comments on minor elements in pyrrhotite: *Economic Geology*, v. 50, no. 3, p. 339–344, accessed December 2023 at <https://doi.org/10.2113/gsecongeo.50.3.339>.
- Fyfe, W.S., Price, N.J., and Thompson, A.B., 1978, *Fluids in the Earth's crust*: Amsterdam, Netherlands, Elsevier, 383 p.
- Goldfarb, R.J., Baker, T., Dubé, B., Groves, D.I., Hart, C.J.R., and Gosselin, P., 2005, Distribution, character, and genesis of gold deposits in metamorphic terranes, chap. 14 of Hedenquist, J.W., Thompson, J.F.H., Goldfarb, R.J., and Richards, J.P., eds., *Economic Geology 100th anniversary volume*: Society of Economic Geologists, p. 407–450.
- Goldfarb, R.J., Berger, B.R., George, M.W., and Seal, R.R., II, 2017, Tellurium, chap. R of Schulz, K.J., DeYoung, J.H., Jr., Seal, R.R., II, and Bradley, D.C., eds., *Critical mineral resources of the United States—Economic and environmental geology and prospects for future supply*: U.S. Geological Survey Professional Paper 1802, p. R1–R27, accessed December 13, 2023, at <https://doi.org/10.3133/pp1802>.
- Goldfarb, R.J., and Groves, D.I., 2015, Orogenic gold—Common or evolving fluid and metal sources through time: *Lithos*, v. 233, p. 2–26, accessed September 2023 at <https://doi.org/10.1016/j.lithos.2015.07.011>.
- Goldfarb, R.J., Groves, D.I., and Gardoll, S., 2001, Orogenic gold and geologic time—A global synthesis: *Ore Geology Reviews*, v. 18, no. 1–25, p. 1–75, accessed September 2023 at [https://doi.org/10.1016/S0169-1368\(01\)00016-6](https://doi.org/10.1016/S0169-1368(01)00016-6).
- Goldfarb, R.J., Miller, L.D., Leach, D.L., and Snee, L.W., 1997, Gold deposits in metamorphic rocks of Alaska, chap. 6 of Goldfarb, R.J., and Miller, L.D., eds., *Mineral deposits of Alaska: Economic Geology Monograph Series*, v. 9, p. 151–190, accessed September 2023 at <https://doi.org/10.5382/Mono.09.07>.
- Granitto, M., Emsbo, P., Hofstra, A.H., Orkild-Norton, A.R., Bennett, M.M., Azain, J.S., Koenig, A.E., and Karl, N.A., 2020, Global geochemical database for critical minerals in archived mine samples: U.S. Geological Survey data release, accessed December 13, 2023, at <https://doi.org/10.5066/P9Z3XL6D>.
- Granitto, M., Hofstra, A.H., and Taylor, R.D., 2025, National Geochemical Database on Ore Deposits—New data featuring fusion digestion analytical methods: U.S. Geological Survey data release, at <https://doi.org/10.5066/P1M9HNMW>.
- Granitto, M., Schmidt, D.E., Karl, N.A., and Khoury, R.M., 2021, National Geochemical Database on Ore Deposits—Legacy data: U.S. Geological Survey data release, accessed December 13, 2023, at <https://doi.org/10.5066/P944U7S5>.
- Groves, D.I., Goldfarb, R.J., Gebre-Mariam, M., Hagemann, S.G., and Robert, F., 1998, Orogenic gold deposits—A proposed classification in the context of their crustal distribution and relationship to other gold deposit types: *Ore Geology Reviews*, v. 13, no. 1–5, p. 7–27, accessed September 2023 at [https://doi.org/10.1016/S0169-1368\(97\)00012-7](https://doi.org/10.1016/S0169-1368(97)00012-7).
- Groves, D.I., Phillips, G.N., Ho, S.E., Houston, S.M., and Standing, C.A., 1987, Craton-scale distribution of Archaean greenstone gold deposits—Predictive capacity of the metamorphic model: *Economic Geology*, v. 82, no. 8, p. 2045–2058, accessed September 2023 at <https://doi.org/10.2113/gsecongeo.82.8.2045>.

- Groves, D.I., Santosh, M., Deng, J., Wang, Q., Yang, L., and Zhang, L., 2020, A holistic model for the origin of orogenic gold deposits and its implications for exploration: *Mineralium Deposita*, v. 55, p. 275–292, accessed November 2023 at <https://doi.org/10.1007/s00126-019-00877-5>.
- Haeberlin, Y., Moritz, R., and Fontboté, L., 2003, Paleozoic orogenic gold deposits in the eastern Central Andes and its foreland, South America: *Ore Geology Reviews*, v. 22, no. 1–2, p. 41–59, accessed October 2023 at [https://doi.org/10.1016/S0169-1368\(02\)00108-7](https://doi.org/10.1016/S0169-1368(02)00108-7).
- Haeberlin, Y., Moritz, R., Fontboté, L., and Cosca, M., 2004, Carboniferous orogenic gold deposits at Patata, eastern Andean Cordillera, Peru—Geological and structural framework, paragenesis, alteration, and $^{40}\text{Ar}/^{39}\text{Ar}$ geochronology: *Economic Geology*, v. 99, no. 1, p. 73–112, accessed September 2023 at <https://doi.org/10.2113/gsecongeo.99.1.73>.
- Hammerli, J., Spandler, C., Oliver, N.H.S., Sossi, P., and Dipple, G.M., 2015, Zn and Pb mobility during metamorphism of sedimentary rocks and potential implications for some base metal deposits: *Mineralium Deposita*, v. 50, p. 657–664, accessed December 2023 at <https://doi.org/10.1007/s00126-015-0600-5>.
- Harrison, J.E., Leach, D.L., and Kleinkopf, M.D., 1986, Resource appraisal maps for mesothermal base- and precious-metal veins in the Wallace 1° x 2° quadrangle, Montana and Idaho: U.S. Geological Survey Miscellaneous Investigations Series IMAP 1509–I, scale 1:250,000, accessed November 2023 at <https://pubs.usgs.gov/publication/i1509I>.
- Heinchon, S.H., 2019, Metal and mineral zoning and ore paragenesis at the Kensington Au-Te deposit, SE Alaska: Fairbanks, Alaska, University of Alaska Fairbanks, M.S. thesis, 208 p., accessed September 2023 at <http://hdl.handle.net/11122/10504>.
- Hofstra, A.H., and Kreiner, D.C., 2020, Systems-deposit s-commodities-critical minerals table for the Earth Mapping Resources Initiative: U.S. Geological Survey Open-File Report 2020–1042, 24 p., accessed September 11, 2023, at <https://doi.org/10.3133/ofr20201042>.
- Hofstra, A.H., Marsh, E.E., Todorov, T.I., and Emsbo, P., 2013, Fluid inclusion evidence for a genetic link between simple antimony veins and giant silver veins in the Coeur d’Alene mining district, ID and MT, USA: *Geofluids*, v. 13, no. 4, p. 475–493, accessed November 2023 at <https://doi.org/10.1111/gfl.12036>.
- Howard, J.M., 1979, Antimony district of southwest Arkansas: Arkansas Geological Commission Information Circular IC–24, 29 p., 1 pl.
- Hu, Z., and Gao, S., 2008, Upper crustal abundances of trace elements—A revision and update: *Chemical Geology*, v. 253, no. 3–4, p. 205–221, accessed March 2022 at <https://doi.org/10.1016/j.chemgeo.2008.05.010>.
- Israel Science and Technology, 2007, List of periodic table elements sorted by abundance in the Earth’s crust: Israel Science and Technology, accessed March 2, 2022, at <https://www.science.co.il/elements/>.
- Keppie, J.D., Boyle, R.W., and Haynes, S.J., eds., 1986, Turbidite-hosted gold deposits: Geological Association of Canada Special Paper no. 32, 186 p.
- Kulla, G., 2017, Technical report on the Idaho-Maryland project—Grass Valley California, USA: Rise Gold Corp., prepared by Amec Foster Wheeler, 179 p, accessed June 2022 at https://www.risegoldcorp.com/uploads/content/I-M_Tech_Report.pdf.
- Leach, D.L., Hofstra, A.H., Church, S.E., Snee, L.W., Vaughn, R.B., and Zartman, R.E., 1998, Evidence for Proterozoic and Late Cretaceous-early Tertiary ore-forming events in the Coeur d’Alene district, Idaho and Montana: *Economic Geology*, v. 93, no. 3, p. 347–359, accessed November 2023 at <https://doi.org/10.2113/gsecongeo.93.3.347>.
- Leach, D.L., Landis, G.P., and Hofstra, A.H., 1988, Metamorphic origin of the Coeur d’Alene base- and precious-metal veins in the Belt Basin, Idaho and Montana: *Geology*, v. 16, no. 2, p. 122–125, accessed October 2023 at <https://pubs.geoscienceworld.org/gsa/geology/article-abstract/16/2/122/204629/Metamorphic-origin-of-the-Coeur-d-Alene-base-and>.
- Lindgren, W., 1933, Mineral deposits (4th ed.): New York, McGraw-Hill, 930 p.
- Long, K.R., DeYoung, J.H., Jr., and Ludington, S., 1998, Database of significant deposits of gold, silver, copper, lead, and zinc in the United States: U.S. Geological Survey Open-File Report 98–206–A,B, 60 p., accessed August 2022 at <https://doi.org/10.3133/ofr98206AB>.
- Long, K.R., 1998a, Grade and tonnage models for Coeur d’Alene-type polymetallic veins: U.S. Geological Survey Open-File Report 98–583, 28 p.
- Long, K.R., 1998b, Production and disposal of mill tailings in the Coeur d’Alene mining region, Shoshone County, Idaho—Preliminary estimates: U.S. Geological Survey Open-File Report 98–595, 14 p.
- Loucks, R.R., and Mavrogenes, J.A., 1999, Gold solubility in supercritical hydrothermal brines measured in synthetic fluid inclusions: *Science*, v. 284, no. 5423, p. 2159–2163, accessed September 2023 at <https://doi.org/10.1126/science.284.5423.2159>.

- McMahon, A.D., Cotterill, C.H., Dunham, J.T., and Rice, W.L., 1974, The U.S. zinc industry—A historical perspective: U.S. Bureau of Mines Information Circular 8629, 76 p.
- Mauk, J.L., and Strand, A.D., 2002, Ore bodies of the Coeur d'Alene district, Idaho—Stratigraphic controls and alteration halos, in 2002 AusIMM Annual Conference, Auckland, New Zealand, 2002, Proceedings: AusIMM, p. 261–266.
- Mauk, J.L., and White, B.G., 2004, Stratigraphy of the Proterozoic Revett Formation and its control on Ag-Pb-Zn vein mineralization in the Coeur d'Alene district, Idaho: *Economic Geology*, v. 99, no. 2, p. 295–312, accessed October 2023 at <https://doi.org/10.2113/gsecongeo.99.2.295>.
- Nassar, N.T., and Fortier, S.M., 2021, Methodology and technical input for the 2021 review and revision of the U.S. Critical Minerals List: U.S. Geological Survey Open-File Report 2021–1045, 31 p., accessed December 13, 2023, at <https://doi.org/10.3133/ofr20211045>.
- Nassar, N.T., Graedel, T.E., and Harper, E.M., 2015, By-product metals are technologically essential but have problematic supply: *Science Advances*, v. 1, no. 3, 10 p.
- O'Leary, M.J., and Sack, R.O., 1987, Fe-Zn exchange reaction between tetrahedrite and sphalerite in natural environments: *Contributions to Mineralogy and Petrology*, v. 96, p. 415–425, accessed December 2023 at <https://doi.org/10.1007/BF01166687>.
- Pease, R.C., 2009, Idaho-Maryland mine project, Grass Valley CA—Technical report: Emgold Mining Corporation, 110 p.
- Phillips, G.N., and Groves, D.I., 1983, The nature of Archaean gold-bearing fluids as deduced from gold deposits of Western Australia: *Journal of the Geological Society of Australia*, v. 30, no. 1–2, p. 25–39, accessed August 2023 at <https://doi.org/10.1080/00167618308729234>.
- Phillips, G.N., and Powell, R., 2010, Formation of gold deposits—A metamorphic devolatilization model: *Journal of Metamorphic Geology*, v. 28, no. 6, p. 689–718, accessed August 2023 at <https://doi.org/10.1111/j.1525-1314.2010.00887.x>.
- Pitcairn, I.K., Craw, D., and Teagle, D.A.H., 2015, Metabasalts as sources of metals in orogenic gold deposits: *Mineralium Deposita*, v. 50, p. 373–390, accessed September 2023 at <https://doi.org/10.1007/s00126-014-0547-y>.
- Pitcairn, I.K., Olivo, G.R., Teagle, D.A.H., and Craw, D., 2010, Sulfide evolution during prograde metamorphism of the Otago and Alpine schists, New Zealand: *The Canadian Mineralogist*, v. 48, no. 5, p. 1267–1295, accessed September 2023 at <https://doi.org/10.3749/canmin.48.5.1267>.
- Pitcairn, I.K., Teagle, D.A.H., Craw, D., Olivo, G.R., Kerrich, R., and Brewer, T.S., 2006, Sources of metals and fluids in orogenic gold deposits—Insights from the Otago and Alpine Schists, New Zealand: *Economic Geology*, v. 101, no. 8, p. 1525–1546, accessed September 2023 at <https://doi.org/10.2113/gsecongeo.101.8.1525>.
- Poulsen, K.H., Ames, D.E., Lau, S., and Brisbin, W.C., 1986, Preliminary report on the structural setting of gold in the Rice Lake area, Uchi subprovince, southeastern Manitoba: Geological Survey of Canada Paper 86–1B, p. 213–221.
- Powell, R., Will, T.M., and Phillips, G.N., 1991, Metamorphism in Archaean greenstone belts—Calculated fluid compositions and implications for gold mineralization: *Journal of Metamorphic Geology*, v. 9, no. 2, p. 141–150, accessed September 2023 at <https://doi.org/10.1111/j.1525-1314.1991.tb00510.x>.
- Ramos, F.C., and Rosenberg, P.E., 2012, Age and origin of quartz-carbonate veins associated with the Coeur d'Alene mining district, Idaho and western Montana—Insights from isotopes and rare earth elements: *Economic Geology*, v. 107, no. 6, p. 1321–1339, accessed December 2023 at <https://doi.org/10.2113/econgeo.107.6.1321>.
- Rosenberg, P.E., and Larson, P.B., 1996, The age of quartz-carbonate veins associated with the Coeur d'Alene mineralization [abs.]: Geological Society of America Program with Abstracts, v. 28, no. 7, p. A475.
- Rudnick, R.L., and Gao, S., 2014, 4.1—Composition of the continental crust, chap. 4 of Holland, H.D., and Turekian, K.K., eds., *Treatise on geochemistry* (2d ed.): Oxford, Pergamon Press, v. 4, 51 p., accessed March 2022 at <https://doi.org/10.1016/B978-0-08-095975-7.00301-6>.
- Sack, R.O., Fredericks, R., Hardy, L.S., and Ebel, D.S., 2005, Origin of high-Ag fahlores from the Galena mine, Wallace, Idaho, U.S.A.: *American Mineralogist*, v. 90, no. 5–6, p. 1000–1007, accessed December 2023 at <https://doi.org/10.2138/am.2005.1651>.
- Sack, R.O., Kuehner, S.M., and Hardy, L.S., 2002, Retrograde Ag-enrichment in fahlores from the Coeur d'Alene mining district, Idaho, USA: *Mineralogical Magazine*, v. 66, no. 1, p. 215–229, accessed December 2023 at <https://doi.org/10.1180/0026461026610024>.
- Sibson, R.H., 1987, Earthquake rupturing as a mineralizing agent in hydrothermal systems: *Geology*, v. 15, no. 8, p. 701–704, accessed July 2023 at <https://pubs.geoscienceworld.org/gsa/geology/article-abstract/15/8/701/204481/Earthquake-rupturing-as-a-mineralizing-agent-in?redirectedFrom=fulltext>.

- Sibson, R.H., Robert, F., and Poulsen, K.H., 1988, High-angle reverse faults, fluid-pressure cycling, and mesothermal gold-quartz deposits: *Geology*, v. 16, no. 6, p. 551–555, accessed July 2023 at <https://pubs.geoscienceworld.org/gsa/geology/article-abstract/16/6/551/190624/High-angle-reverse-faults-fluid-pressure-cycling?redirectedFrom=fulltext>.
- Skirrow, R.G., Huston, D.L., Mernagh, T.P., Thorne, J.P., Dulfer, H., and Senior, A.B., 2013, Critical commodities for a high-tech world—Australia's potential to supply global demand: Canberra, Geoscience Australia, [variously paged].
- Taylor, R.D., Monecke, T., Reynolds, T.J., and Monecke, J., 2021, Paragenesis of an orogenic gold deposit—New insights on mineralizing processes at the Grass Valley district, California: *Economic Geology*, v. 116, no. 2, p. 323–356. [Also available at <https://doi.org/10.5382/econgeo.4794>.]
- Tomkins, A.G., 2010, Windows of metamorphic sulfur liberation in the crust—Implications for gold deposit genesis: *Geochimica et Cosmochimica Acta*, v. 74, no. 11, p. 3246–3259, accessed August 2023 at <https://doi.org/10.1016/j.gca.2010.03.003>.
- Turneure, F.S., 1955, Metallogenic provinces and epochs, chap. 2 of Bateman, A.M., Ferguson, H.G., Chace, F.M., Burbank, W., McKinstry, H.E., Leighton, M.M., Noble, J., Broderick, T.M., Davidson, D.H., DeFord, R.K., Harder, E.C., Jahns, R.H., Legget, R.F., Sayer, A.N., and Schwartz, G.M., *Economic Geology fiftieth anniversary volume—1905–1955*: Lancaster, Pa., Lancaster Press, Inc., p. 38–98.
- U.S. Department of Energy, 2021, Department of Energy launches Minerals Sustainability Division to enable the ongoing transformation of the U.S. energy system and help secure a U.S. critical minerals supply chain: U.S. Department of Energy, January 15, 2023, accessed September 11, 2023, at <https://www.energy.gov/articles/departement-energy-launches-minerals-sustainability-division-enable-ongoing-transformation>.
- U.S. Department of the Interior, 2017, Critical mineral independence and security: U.S. Department of the Interior, Secretarial Order no. 3359, 2 p., accessed December 13, 2023, at https://www.doi.gov/sites/doi.gov/files/uploads/so_criticalminerals.pdf.
- U.S. Environmental Protection Agency, 2017, Superfund cleanup implementation plan, 2016–2025—Bunker Hill Mining and Metallurgical Complex superfund site: U.S. Environmental Protection Agency, 124 p., accessed September 11, 2023, at <https://semspub.epa.gov/work/10/100038724.pdf>.
- U.S. Geological Survey [USGS], 2022a, U.S. Geological Survey releases 2022 list of critical minerals: U.S. Geological Survey, February 22, 2022, accessed September 11, 2023, at <https://www.usgs.gov/news/national-news-release/us-geological-survey-releases-2022-list-critical-minerals>.
- U.S. Geological Survey [USGS], 2022b, Mineral commodity summaries 2022: U.S. Geological Survey, 202 p., accessed September 11, 2023, at <https://doi.org/10.3133/mcs2022>.
- Wallace, C.A., Lidke, D.J., and Schmidt, R.G., 1990, Faults of the central part of the Lewis and Clark line and fragmentation of the Late Cretaceous foreland basin in west-central Montana: *GSA Bulletin*, v. 102, no. 8, p. 1021–1037, accessed October 2023 at <https://pubs.geoscienceworld.org/gsa/gsabulletin/article-abstract/102/8/1021/182450/Faults-of-the-central-part-of-the-Lewis-and-Clark?redirectedFrom=fulltext>.
- Werner, T.T., Mudd, G.M., and Jowitt, S.M., 2017, The world's by-product and critical metal resources—Part II—A method for quantifying the resources of rarely reported metals: *Ore Geology Reviews*, v. 80, p. 658–675, accessed July 2023 at <https://doi.org/10.1016/j.oregeorev.2016.08.008>.
- Witt, W.K., Hagemann, S.G., Roberts, M., and Davies, A., 2020, Cobalt enrichment at the Juomasuo and Hangaslampi polymetallic deposits, Kuusamo Schist Belt, Finland—A role for an orogenic gold fluid?: *Mineralium Deposita*, v. 55, p. 381–388, accessed September 2023 at <https://doi.org/10.1007/s00126-019-00943-y>.
- Wyman, D.A., Cassidy, K.F., and Hollings, P., 2016, Orogenic gold and the mineral systems approach—Resolving fact, fiction and fantasy: *Ore Geology Reviews*, v. 78, p. 322–335, accessed August 2023 at <https://doi.org/10.1016/j.oregeorev.2016.04.006>.
- Yang, W., Lan, X., Wang, Q., Dong, P., and Wang, G., 2021, Selective pre-leaching of tellurium from telluride-type gold concentrate: *Frontiers in Chemistry*, v. 9, 10 p., accessed September 2023 at <https://doi.org/10.3389/fchem.2021.593888>.
- Zartman, R.E., and Stacey, J.S., 1971, Lead isotopes and mineralization ages in Belt supergroup rocks, northwestern Montana and northern Idaho: *Economic Geology*, v. 66, no. 6, p. 849–860, accessed October 2023 at <https://doi.org/10.2113/gsecongeo.66.6.849>.
- Zhang, J., Zhang, Y., Richmond, W., and Wang, H.-p., 2010, Processing technologies for gold-telluride ores: *International Journal of Minerals, Metallurgy, and Materials*, v. 17, p. 1–10, accessed September 2023 at <https://doi.org/10.1007/s12613-010-0101-6>.

Appendix 1. Descriptions for Coeur d'Alene-Type and Orogenic Gold Ore Samples Used in this Study

This appendix contains two tables that briefly describe the ore samples included in this study. [Table 1.1](#) contains descriptions for orogenic gold ore samples and [table 1.2](#) contains descriptions for Coeur d'Alene-type ore samples. The geochemical data associated with these tables are found in Granitto and others (2020, 2021, 2025). Only samples from the United States are included.

Table 1.1. Information on orogenic gold ore samples included in this study.

[Geochemical results are in Granitto and others (2020, 2021, 2025). ID, identification; S, south; ft, foot; —, no data; No., number; approx., approximately; Mtns, mountains; WYOD, Work Your Own Diggings]

Collection ID	State	District	Deposit or mine	Sample description
OD14855	S. Dak.	Terraville-Homestake	Homestake mine	3,800 ft level, drift S of 49 crosscut
OD11950	S. Dak.	Terraville-Homestake	Homestake mine	2,300 ft level, 26 pillar main ledge ore body
OD11902	S. Dak.	Terraville-Homestake	Homestake mine	2,450 ft level
11052	S. Dak.	Terraville-Homestake	Homestake mine	—
11044	S. Dak.	Terraville-Homestake	Homestake mine	Below water line
JAR4	Mont.	Jardine-Crevasse	Jardine mine	—
OD22224	Alaska	Treadwell	Juneau	Dump outside mouth of tunnel
OD11904	S. Dak.	Terraville-Homestake	Homestake mine	1,700 ft level
AL4	Alaska	Juneau	Treadwell	—
11054	S. Dak.	Terraville-Homestake	Homestake mine	Homestake
21830	S. Dak.	Terraville-Homestake	Homestake mine	Ore face
11034	S. Dak.	Terraville-Homestake	Homestake mine	Homestake
23424	S. Dak.	Terraville-Homestake	Homestake mine	No. 11 ledge, 3,800 ft level
19294	S. Dak.	Terraville-Homestake	Homestake mine	7,850 level, 25A top stope, 19 ledge
OD22208	Alaska	Juneau	Perseverance	—
OD14851	S. Dak.	Terraville-Homestake	Homestake mine	3,350 ft level, 16C stope 9 ledge ore body
OD11940	S. Dak.	Terraville-Homestake	Homestake mine	1,700 ft level, 14 pillar main ledge ore body
OD11931	S. Dak.	Terraville-Homestake	Homestake mine	5,000 ft level, south drift
OD11943	S. Dak.	Terraville-Homestake	Homestake mine	4,100 ft level, W crosscut 9 ledge ore body
OD10459	Calif.	Grass Valley	Original Empire	Deepest level ore approx. 8,000 ft
OD11942	S. Dak.	Terraville-Homestake	Homestake mine	2,900 ft level, 25D stope
OD11903	S. Dak.	Terraville-Homestake	Homestake mine	2,450 ft level
13519	Calif.	Shingle Springs	Big Canyon	—
13518	Calif.	Shingle Springs	Big Canyon	—
13517	Calif.	Shingle Springs	Big Canyon	—
19286	S. Dak.	Terraville-Homestake	Homestake mine	7,400 level, 22 top stope, 19 ledge
17985	Idaho	Yellow Pine	Yellow Pine mine	—
OD07880	Calif.		Rymal mine	North Fork Feather River, Meadow View
EMTA-1-1	Calif.	Grass Valley	Empire	Collected from a vein exposed in the tourist adit
EMTA-2-2	Calif.	Grass Valley	Empire	Collected from a vein exposed in the tourist adit
EUR-D-1	Calif.	Mother Lode	Eureka	—
EUR-D-2	Calif.	Mother Lode	Eureka	—
HAR-1	Calif.	Mother Lode	Harvard	—
HAR-2	Calif.	Mother Lode	Harvard	—
IDH001-B	Calif.	Grass Valley	Idaho-Maryland	High-grade drill core intercept
OX-DOWN-2	Calif.	Downieville	Oxford	—
SO-1	Calif.	Allegheny	Sixteen-to-One	—
UHD-3A	Calif.	Grass Valley	Union Hill	Dump at the mine
WYOD-1B	Calif.	Grass Valley	WYOD	Dump at the mine
WYOD-2	Calif.	Grass Valley	WYOD	Dump at the mine
C-504783	Calif.	Klamath Mtns	Golden Eagle 1	Collected from a dump for Taylor and others (2022)
C-504784	Calif.	Klamath Mtns	Golden Eagle 2	Collected from a dump for Taylor and others (2022)

Table 1.1. Information on orogenic gold ore samples included in this study.—Continued

[Geochemical results are in Granitto and others (2020, 2021, 2025). ID, identification; S, south; ft, foot; —, no data; No., number; approx., approximately; Mtns, mountains; WYOD, Work Your Own Diggings]

Collection ID	State	District	Deposit or mine	Sample description
C-504785	Calif.	Klamath Mtns	Schroeder 1	Collected from a dump for Taylor and others (2022)
C-504786	Calif.	Klamath Mtns	Schroeder 2	Collected from a dump for Taylor and others (2022)
C-504787	Calif.	Klamath Mtns	Yankee John 1	Collected from a dump for Taylor and others (2022)
C-504789	Calif.	Klamath Mtns	Yankee John 2	Collected from a dump for Taylor and others (2022)
C-504790	Calif.	Klamath Mtns	McKinley 1	Collected from a dump for Taylor and others (2022)
C-504791	Calif.	Klamath Mtns	McKinley 2	Collected from a dump for Taylor and others (2022)
C-504792	Calif.	Klamath Mtns	Jillson	Collected from a dump for Taylor and others (2022)
C-504793	Calif.	Klamath Mtns	New York	Collected from a dump for Taylor and others (2022)
C-504794	Calif.	Klamath Mtns	McKeen	Collected from a dump for Taylor and others (2022)
C-504795	Calif.	Klamath Mtns	Cleaver	Collected from a dump for Taylor and others (2022)
C-504796	Calif.	Klamath Mtns	Maple	Collected from a dump for Taylor and others (2022)
C-504797	Calif.	Klamath Mtns	Ozark	Collected from a dump for Taylor and others (2022)
C-504798	Calif.	Klamath Mtns	Quartz Hill	Collected from a dump for Taylor and others (2022)
C-504799	Calif.	Klamath Mtns	Hickey	Collected from a dump for Taylor and others (2022)
C-504800	Calif.	Klamath Mtns	Washington	Collected from underground mine face for Taylor and others (2022)
C-504801	Calif.	Klamath Mtns	Mojo	Collected from underground mine face for Taylor and others (2022)
C-504802	Calif.	Klamath Mtns	Dean	Collected from underground mine face for Taylor and others (2022)
C-504803	Calif.	Klamath Mtns	Dean #p3	Collected from underground mine face for Taylor and others (2022)
C-504804	Calif.	Klamath Mtns	Dean #p3 N-S	Collected from underground mine face for Taylor and others (2022)
C-506231	Calif.	Allegheny	Oriental	—
C-506232	Calif.	Mother Lode	Carson Hill 3	—
C-506233	Calif.	Honey Lake	Honey Lake 1	—
C-506234	Calif.	Mother Lode	Empire Plymouth	—
C-506235	Calif.	Mother Lode	Eureka	—
C-506236	Alaska	Kantishna	Stampede mine	—
C-506237	Alaska	Kantishna	Stampede mine	—
C-506238	Alaska	Fairbanks	Ryan Lode	Surficial vein outcrop
C-506239	Alaska	Fairbanks	Ryan Lode	Surficial vein outcrop
C-506240	Alaska	Fairbanks	Ryan Lode	Surficial vein outcrop
C-506241	Alaska	Fairbanks	Ryan Lode	Surficial vein outcrop
C-506242	Alaska	Fairbanks	Ryan Lode	Surficial vein outcrop
C-506244	Calif.	Bagby district	Josephine	—
IM19_13_3272.95	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3278	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3280.4	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3283.2	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3286.6	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3289.4	Calif.	Grass Valley	Idaho-Maryland	Drill core

Table 1.1. Information on orogenic gold ore samples included in this study.—Continued

[Geochemical results are in Granitto and others (2020, 2021, 2025). ID, identification; S, south; ft, foot; —, no data; No., number; approx., approximately; Mtns, mountains; WYOD, Work Your Own Diggings]

Collection ID	State	District	Deposit or mine	Sample description
IM19_13_3292.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3296.65	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3300	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3303.7	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3307	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3310.65	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3313.7	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3317.4	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3320.2	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3323.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3326.35	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3328.45	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3330.5	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3334	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3337.7	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3340.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3344.05	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3347	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3350.65	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3354.3	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3356.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3360.2	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3364.05	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3368	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3371.3	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3374.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3377.85	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3380.55	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3384	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3387.05	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3389.6	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3391.2	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3393.65	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3395.35	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3397.2	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3399.6	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3402.45	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3405	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3407.6	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3410.3	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3413.15	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3417	Calif.	Grass Valley	Idaho-Maryland	Drill core

Table 1.1. Information on orogenic gold ore samples included in this study.—Continued

[Geochemical results are in Granitto and others (2020, 2021, 2025). ID, identification; S, south; ft, foot; —, no data; No., number; approx., approximately; Mtns, mountains; WYOD, Work Your Own Diggings]

Collection ID	State	District	Deposit or mine	Sample description
IM19_13_3420.35	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3424.45	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3428.4	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3432.35	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3436.45	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3440.45	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3445.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3451.1	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3457.1	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3463.1	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3468.65	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3472.65	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3476.3	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3479.5	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3482.9	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3486.55	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3490	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3493	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3495.5	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3498.2	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3502	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3505.35	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3508.8	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3512.35	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3515.9	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3519	Calif.	Grass Valley	Idaho-Maryland	Drill core
IM19_13_3522.1	Calif.	Grass Valley	Idaho-Maryland	Drill core
CDA-GC-200	Idaho	Summit	Golden Chest	Ore

Table 1.2. Information on Coeur d'Alene-type ore samples included in this study.

[Geochemical results are in Granitto and others (2020, 2021, 2025). Lab, laboratory; ID, identification; oz/ton, ounce per short ton; <, less than; Co., company; ±, plus or minus; %, percent; approx., approximately; ', feet; N., north; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; Au, gold; Ba, barium; C, carbon; Cu, copper; Fe, iron; Mn, manganese; Pb, lead; Sb, antimony; Zn, zinc]

Lab ID	Field ID	State	District	Deposit or mine	Sample description
C368513	RM0282	Idaho	Hayden Lake	Couer d'Alene pit	Au quartz; mixture of massive stibnite stringers, arsenopyrite, siderite, and quartz; oxidation of Fe-Cu sulfide minerals present in a widely distributed pattern, makes ID of mineral species difficult
C377832	RM1079	Idaho	Hunter	Lucky Friday mine	High-grade vein 25–30 oz/ton Ag; galena-siderite-sphalerite-pyrite-quartz; a massive but open distribution or cloud-like mixture of galena and sphalerite with inclusions of siderite and quartz; sulfide approx. 80%
C380213	RM0276	Idaho	Lelande	Star-Morning mine	Pb-Zn ore, stringer in Revett quartzite; massive, extremely fine grain sphalerite with tiny stringers of galena in quartz vein, sulfide amount approx. 50%
C447150	RM0261	Montana	Neihart	Florence mine	Slab vein of galena-pyrite-chalcopryite; mostly coarsely crystalline siderite with veinlets of galena, sphalerite, chalcopryite, and pyrite throughout; siderite appears to be Mn-rich and galena appears to be Ag-rich
C447153	RM0263	Idaho	Yreka	Last Chance mine	Tetrahedrite and galena with Ag, collected by T. Hoskins; splotches of fine grain silver-copper sulfide in quartz vein, sulfide content <1%
C447154	RM0264	Idaho	Evolution	Galena mine	Vein; Ag:Pb is 1:1, collected by R.H. Vaughn; massive, fine-to-coarse granular galena with pyrite blebs distributed throughout galena; sulfide content approx. 50%; matrix rock is white quartz with included tan siderite(?)
C447155	RM0265	Idaho	Yreka	Page mine	Vein; 14% across 7', collected by R.H. Vaughn; massive, extremely fine grain galena associated with wispy elongated "clouds" of fine grain sphalerite in quartz; sulfide content approx. 80%; small amount chalcopryite scattered throughout
C447157	RM0267	Idaho	Yreka	Bunker Hill mine	Ore; Truman vein; collected by Bunker Hill; solid massive galena with tiny sphalerite and arsenopyrite inclusions, small inclusions of shattered white quartz; sulfide content approx. 80%
C447166	RM0283	Idaho	Yreka	Sidney mine	Zn ore, collected by A.B. Griggs; nearly pure solid sphalerite with tiny galena stringers running subparallel throughout sphalerite, galena content approx. 5%, sphalerite approx. 95%
C447169	RM0292	Idaho	Yreka	Page mine	Collected by Fred Howell; solid massive galena with small inclusions of arsenopyrite, pyrite, and sphalerite, inclusions of white quartz and feldspars
C447170	RM0293	Idaho	Yreka	Sidney mine	Partially oxidized ore, collected by A.B. Griggs; medium to large round blebs of nearly pure, massive, granular sphalerite imbedded in nearly pure massive galena; sulfide content approx. 90%
C447178	RM0300	Idaho	Placer Center	Standard-Mammoth mine	Galena; massive fine to coarse grain galena with tiny stringers of sphalerite; rock matrix unknown; sulfide content approx. 30%
C447187	RM0308	Idaho	St. Regis	Interstate-Callahan	Pb-Zn ore; massive mixture of fine grain sphalerite, galena and pyrrhotite with quartz; sulfide content approx. 90%
C447203	RM0322	Idaho	Lelande	Star-Morning mine	Sulfide ore; sulfide replacement in brecciated quartzite, collected by A.B. Griggs; stringers and veinlets of galena and sphalerite in quartz; sulfide amount is approx. 30%
C447206	RM0325	Idaho	Yreka	Sidney mine	Brecciated ore, collected by A.B. Griggs; intimate intergrowth of major sphalerite and pyrrhotite with lesser amounts of galena and singular blebs of chalcopryite; sulfide content approx. 40%
C447396	RM0634	Idaho	Yreka	Bunker Hill mine	Galena and siderite ore; numerous chunks; solid massive galena with large siderite inclusions throughout galena; sulfide approx. 80%
C447397	RM0636	Idaho	Yreka	Bunker Hill mine	Bunker Hill ore(?), Pate stope(?); 15' from Dull fault; sample made up of many broken pieces and crumbs; a massive mixture of galena and sphalerite associated with thin quartz and feldspar veinlet; small, isolated areas of pyrite; sulfide approx. 80%; outer surfaces of galena coated with tannish-white Pb oxidation
C447447	RM1087	Idaho	Hunter	Lucky Friday mine	Revett Formation(?); coarsely crystalline galena with occasional scatterings of microscopic pyrite, rarely chalcopryite, mixed with siderite in dark grey quartz; sulfide approx. 30%
C447453	RM1098	Idaho	Evolution	Sunshine mine	Silver ore; Sunshine Mining Co.; two samples; widely scattered microscopic tetrahedrite blebs throughout massive, coarsely crystalline siderite; sulfide <1%

Table 1.2. Information on Coeur d'Alene-type ore samples included in this study.—Continued

[Geochemical results are in Granitto and others (2020, 2021, 2025). Lab, laboratory; ID, identification; oz/ton, ounce per short ton; <, less than; Co., company; ±, plus or minus; %, percent; approx., approximately; ', feet; N., north; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; Au, gold; Ba, barium; C, carbon; Cu, copper; Fe, iron; Mn, manganese; Pb, lead; Sb, antimony; Zn, zinc]

Lab ID	Field ID	State	District	Deposit or mine	Sample description
C447456	RM1107	Idaho	Hunter	Lucky Friday mine	Ore, six pieces; very friable banded sphalerite, siderite, and galena; galena massive coarsely crystalline; siderite in brown rhombs, sphalerite dark brown and dull lustered; sulfide approx. 70%
C447457	RM1108	Idaho	Hunter	Lucky Friday mine	Banded ore; massive siderite with numerous inclusions; quartz veinlet; sulfide approx. 40%
C447460	RM1113	Idaho	Hunter	Lucky Friday mine	Host Revett Formation; galena-sphalerite-siderite; high-grade silver ore; abundant tannish colored siderite in a galena sphalerite matrix, all in an Fe-, Mg-, Al-silicate rock; irregular, discontinuous quartz veins; sulfide approx. 40%
C-513302	VCF-200-53	Idaho	Evolution	Silver Summit	Ag-rich ore
C-513303	CDA-SS-7	Idaho	Evolution	Silver Summit	Disseminated, Cu-rich ore
C-513304	SS-8-96	Idaho	Evolution	Silver Summit	Ag+Pb-rich ore
C-513305	CDA-CR-1	Idaho	Evolution	Coeur	Pyritic
C-513306	CDA-CR-14	Idaho	Evolution	Coeur	Zn-rich ore
C-513308	CDA-81L-14	Idaho	Evolution	Coeur	Ag-rich ore
C-513309	CDA-110	Idaho	Evolution	Coeur	Ag-Sb ore
C-513310	SSH-3100L-104A	Idaho	Yreka	Sunshine	Ag-rich ore
C-513311	SSH-3100L-104B	Idaho	Yreka	Sunshine	Ag+Pb-rich ore
C-513312	SSH-3100L-WC	Idaho	Yreka	Sunshine	Pb+Ag+Sb-rich ore
C-513313	SSH-3100L-116	Idaho	Yreka	Sunshine	Pb-rich ore
C-513314	VF-351-53	Idaho	Yreka	Page	Zn-rich ore
C-513315	VF-373-53	Idaho	Yreka	Page	Zn-rich ore
C-513316	VF-204-53	Idaho	Placer Center	Dayrock	Zn+Pb-rich ore
C-513317	CDA-LF-075	Idaho	Hunter	Lucky Friday	Zn-rich ore
C-513319	CDA-LF-202	Idaho	Hunter	Lucky Friday	Zn-rich ore
C-513320	GH-4900L-11W	Idaho	Hunter	Gold Hunter	Pyritic
C-513321	GH-4900L-30A	Idaho	Hunter	Gold Hunter	Pb-rich ore
C-513322	GH-4900L-30B	Idaho	Hunter	Gold Hunter	Pyritic
C-513323	GH-4900L-30C	Idaho	Hunter	Gold Hunter	Ag+Sb-rich ore
C-513324	GH-4900L-52D	Idaho	Hunter	Gold Hunter	Zn+Pb-rich ore
C-513325	GH-4900L-52U	Idaho	Hunter	Gold Hunter	Ag+Pb+Zn-rich ore
C-513326	GH-4900L-54U	Idaho	Hunter	Gold Hunter	Ag+Pb±Ba-rich ore
C-513327	GH-4900L-56	Idaho	Hunter	Gold Hunter	Zn-rich ore
C-513328	GH-4900L-185A	Idaho	Hunter	Gold Hunter	Pyritic
C-513330	GH-4900L-185B	Idaho	Hunter	Gold Hunter	Pb-rich ore
C-513331	VCF-108-58A	Idaho	Yreka	Bunker Hill	Zn-rich ore
C-513332	VCF-1018-58B	Idaho	Yreka	Bunker Hill	Pb-rich ore
C-513333	CDA-BH-2	Idaho	Yreka	Bunker Hill	Zn-rich ore
C-513334	BH-1000L-Quill	Idaho	Yreka	Bunker Hill	Pyritic Zn-rich ore
C-513335	VF-159-53	Idaho	Placer Center	Galena	Zn+Pb-rich ore
C-513336	CDA-GA-201	Idaho	Placer Center	Galena	Ag-rich ore
C-513337	GA-37L-34L-125A	Idaho	Placer Center	Galena	Cu-rich ore
C-513338	GA-37L-34L-125B	Idaho	Placer Center	Galena	Ag-rich ore

Table 1.2. Information on Coeur d'Alene-type ore samples included in this study.—Continued

[Geochemical results are in Granitto and others (2020, 2021, 2025). Lab, laboratory; ID, identification; oz/ton, ounce per short ton; <, less than; Co., company; ±, plus or minus; %, percent; approx., approximately; ', feet; N., north; Abbreviations of chemical elements as follows: Ag, silver; Al, aluminum; Au, gold; Ba, barium; C, carbon; Cu, copper; Fe, iron; Mn, manganese; Pb, lead; Sb, antimony; Zn, zinc]

Lab ID	Field ID	State	District	Deposit or mine	Sample description
C-513339	GA-46L-105	Idaho	Placer Center	Galena	Pyritic wallrock
C-513341	CDA-IC-14	Idaho	St. Regis	Interstate-Callahan	Zn-rich ore
C-513342	CDA-IC-15	Idaho	St. Regis	Interstate-Callahan	Zn-rich ore
C-513343	CDA-STAR	Idaho	Lelande	Star-Morning	Zn+Pb-rich ore
C-513344	CDA-SM-100	Idaho	Lelande	Star-Morning	Zn+Pb-rich ore
C-513345	CDA-TA-5	Idaho	Placer Center	Tamarack	Zn-rich ore
C-513346	CDA-TA-16	Idaho	Placer Center	Tamarack	Zn+Pb-rich ore
C-513347	VF-433-5	Idaho	Hunter	Atlas	Cu-rich ore
C-513348	CDA-AT-204	Idaho	Hunter	Atlas	Pyritic (±C) ore?
C-513352	CDA-7	Idaho	Summit	N. of Golden Chest	Pb+Zn-rich ore
C-513354	CDA-81L-56	Idaho	Murray	Murray district	Zn-rich ore
C-513355	CDA-JW-200	Idaho	Eagle sub-district	Jack Waite	Pb-rich ore
C-513356	CDA-84-606	Idaho	Eagle sub-district	Jack Waite	Pyritic Pb+Zn-rich ore
C-513357	CCDA-SC-208	Idaho	Placer Center	Success	Zn-rich ore
C-513358	VCF-171-56	Idaho	Placer Center	Rex	Zn-rich ore
C-513359	VCF-176-53	Idaho	Placer Center	Idora	Zn+Pb-rich ore
C-513360	CDA-LP-300	Idaho	Yreka	Little Pittsburgh	Pyritic Ag?-rich ore
C-513361	CDA-NB-12	Idaho	Yreka	Nabob	Pyritic Zn-rich ore
C-513362	CDA-SB-1	Montana	Prospect Creek	US Antimony mine	Sb-rich ore

References Cited

- Granitto, M., Emsbo, P., Hofstra, A.H., Orkild-Norton, A.R., Bennett, M.M., Azain, J.S., Koenig, A.E., and Karl, N.A., 2020, Global geochemical database for critical minerals in archived mine samples: U.S. Geological Survey data release, accessed December 13, 2023, at <https://doi.org/10.5066/P9Z3XL6D>.
- Granitto, M., Hofstra, A.H., and Taylor, R.D., 2025, National Geochemical Database on Ore Deposits—New data featuring fusion digestion analytical methods: U.S. Geological Survey data release, at <https://doi.org/10.5066/P1M9HNMW>.
- Granitto, M., Schmidt, D.E., Karl, N.A., and Khoury, R.M., 2021, National Geochemical Database on Ore Deposits—Legacy data: U.S. Geological Survey data release, accessed December 13, 2023, at <https://doi.org/10.5066/P944U7S5>.
- Taylor, R.D., Morgan, L.E., Jourdan, F., Monecke, T., Marsh, E.E., and Goldfarb, R.J., 2022, Late Jurassic-Early Cretaceous orogenic gold mineralization in the Klamath Mountains, California—Constraints from ⁴⁰Ar/³⁹Ar dating of hydrothermal muscovite: *Ore Geology Reviews*, v. 141, 19 p. [Also available at <https://doi.org/10.1016/j.oregeorev.2021.104661>.]

Publishing support provided by the Science Publishing Network,
Denver and Reston Publishing Service Centers

For more information concerning the research in this report,
contact the

Center Director, USGS Geology, Geophysics, and Geochemistry
Science Center

Box 25046, Mail Stop 973

Denver, CO 80225

(303) 236-1800

Or visit the Geology, Geophysics, and Geochemistry Science Center
website at

<https://www.usgs.gov/centers/gggsc>

