

INTRODUCTION

The Coastal Plain from western Florida to the Rio Grande is an important agricultural area. Pumpage for irrigated agriculture, which is the largest use of ground water in the study area, was more than 4,500 Mgal/d in 1975. Withdrawal of ground water for municipal and industrial uses was estimated to be about 1,300 Mgal/d for the same year. An annual increase in ground-water withdrawal of 30 Mgal/d has occurred in the study area between 1950 and 1980 (Grubb, 1984).

The increasing annual withdrawals of ground water have resulted in lowering of the potentiometric surface, movement of the saline-fresh water interface into parts of the aquifer system that were previously fresh, and land-surface subsidence in areas of municipal and industrial wells that are pumping large volumes of ground water. Land-surface subsidence also has occurred in some rice-irrigated areas. These changes in the hydrologic system are affecting the availability of suitable quality water for irrigation and for municipal and industrial uses. The availability of suitable quality water will, in part, affect the economy of the area. The suitability of water for irrigation and for municipal and industrial uses as determined in part by the concentration of dissolved solids and by the water type.

In 1961, the U.S. Geological Survey began a study of the Tertiary and Quaternary aquifers, which supply most of the ground water used in the area, for the purpose of defining the geologic framework in which the aquifers exist, describing the chemistry of the ground water, and analyzing the regional ground-water flow patterns and flow systems. The Tertiary and Quaternary aquifers underlie about 230,000 mi² in parts of 10 States—Alabama, Arkansas, Florida, Illinois, Kentucky, Louisiana, Mississippi, Missouri, Tennessee, and Texas. The plan of the Gulf Coast Regional Aquifer System Analysis (RASAS) was described by Grubb (1984). The geologic framework of the Coastal Plain from western Florida to the Rio Grande was described by Horner (in press).

The aquifer and confining systems being studied consist mainly of sedimented Tertiary and Quaternary age. In ascending order, the Tertiary sediments consist of the Midway Group and lower part of the Wilcox Group of Pliocene age; the upper part of the Wilcox Group, the Claiborne Group and the Jackson Group of Eocene age; the Vicksburg Group of Oligocene age; sediments of Miocene age; and sediments of Pliocene age. The Tertiary sediments are composed predominantly of alternating beds of sand and clay, with some interbedded gravel, silt, lignite, and limestone. Thick clay layers restrict ground-water flow between the Tertiary sediments and to or from the underlying Cretaceous sediments. For the purpose of this study, the multiple aquifers in the Tertiary sediments were divided as follows: (1) Wilcox aquifer, (2) Claiborne aquifer, (3) Jackson aquifer, (4) Miocene aquifer, and (5) Pliocene aquifer. The Quaternary sediments are of Pleistocene and Holocene age and include alluvial, valley-fill, terrace, alluvial, and natural levee deposits of each age. For the purpose of this study, the multiple aquifers in the Quaternary sediments were not divided into smaller groups, as was done for the aquifers in the Tertiary sediments, because of the data in the files either were assigned as Quaternary or the water analyzed was from alluvial or terrace deposits both of which could be of Pleistocene or Holocene age. The multiple aquifers in the Quaternary sediments are referred to as the Quaternary aquifer in this report. A more detailed description of the geology and its areal distribution is presented by Horner (in press) and by Grubb (1984).

The purpose of this report is to describe the regional variation in dissolved-solids concentrations and primary water types in water from each of the designated aquifers and to relate the concentration of dissolved solids and the primary water type to the suitability of water for irrigation and for municipal and industrial uses. The concentration of dissolved solids, based on median concentration in 100-mi² areas, is described for each of the designated aquifers in ascending order on sheet 1. The statistical mode of the primary water type in individual 100-mi² areas is described similarly on sheet 2. The procedure for determining the median dissolved-solids concentrations and the statistical mode of the primary water type is described by Farniphan (written commun., 1986). Data used in this study were obtained from the water-quality files of the U.S. Geological Survey, files from the petroleum industry prepared by Taylor (1975), and files of the Texas Department of Water Resources.

DISSOLVED-SOLIDS

Because of the solvent property of water, the water in an aquifer can contain many substances in ionic form, that is, electrically charged parts of a molecule. These substances, which usually are mineral in solution, are collectively called dissolved solids. The concentration of dissolved solids in freshwater is, on the average, so small that it is expressed in milligrams per liter (mg/L). However, the concentration of dissolved solids in natural water is not fixed or unvariable. Consequently, the concentration can vary by many orders of magnitude as a result of geochemical and biochemical interaction between water and the geologic material through which it moves. A simple method for categorizing ground water based on dissolved-solids concentration is presented in the following table (Winlow and Kiser, 1956):

| Water category | Dissolved-solids concentration (milligrams per liter) |
|-------------------|---|
| Fresh | Less than 1,000 |
| Slightly saline | 1,000 to 3,000 |
| Moderately saline | 3,000 to 10,000 |
| Very saline | 10,000 to 35,000 |
| Brine | 35,000 or more |

Dissolved-solids concentration or salinity of water in the aquifer systems in the study area generally increases with depth in the hydrogeopressed (normally pressurized) zones. Maximum salinity at most locations occurs near the bottom of this zone or the top of the geopressure zone. The geopressure zone is described by Jones (1968) as that part of the aquifer or aquifer system that has a fluid pressure which exceeds the hydrostatic pressure of a column of water containing 80,000 mg/L dissolved solids. Geopressure may be expressed as a ratio of the measured fluid pressure in an aquifer to the pressure due to the weight of overlying deposits, computed for the depth at which the aquifer occurs.

Geopressure conditions occur in aquifers when water that is released from the connecting aquifer materials is blocked from discharging from the aquifer, thus forming geopressured zones. Three types of blocks or seals that cause geopressured zones to form are: (1) small reservoirs sealed by pyroclastic; (2) large reservoirs sealed by down faulting against a thick shale, upthrown, and sealed downward by regional tectonic changes; and (3) relative position of fault seals in upthrown and downthrown blocks (Dolan, 1963, p. 402).

Within geopressured zones, the pressure forces saline formation water toward normally pressured aquifers. The intervening clay beds function like membrane filters to inhibit the water, which subsequently dilutes the water in the receiving aquifer. Geopressure and trapped radiogenic heat cause diagenesis of montmorillonite to illite in the clay beds. The released freshwater resulting from the diagenesis of montmorillonite either remains in the transferred aquifer or flows into and dilutes the water in normally pressured aquifers. Diagenetic processes which are described by the diagenesis of clay and which may be responsible for halting or even reversing those flows, are considered the dominant factor in the preservation of geopressure (Flett, 1976, p. 16).

Geopressure zones are reported in the Wilcox Group, Claiborne Group, Jackson Group, Miocene, and Pliocene deposits and are most prevalent at depths between 5,000 and 15,000 ft. The presence of geopressure zones is not treated here as it is applied to fields. However, the salinity of the applied water is critical for optimum growth of crops. In contrast, modern water management techniques and a variety of available water-treatment processes make possible the use of raw water of almost any quality to produce an acceptable municipal and industrial water supply. However, the cost of producing water of acceptable quality is the volume needed needs to be considered. Following are some general guidelines and recommended limits for dissolved-solids concentration in water that is to be used for irrigation or municipal and industrial purposes (B.S. Environmental Protection Agency, 1973, p. 323-328).

Acceptable limits of dissolved solids in irrigation water may vary from less than 100 to more than 4,000 mg/L. For example, a spring irrigation of water with a dissolved-solids concentration of about 100 mg/L could ultimately damage sensitive crops such as citrus fruit, whereas water with a dissolved-solids concentration of 4,000 mg/L may be used to successfully produce salt-tolerant field crops on land with good drainage. This phenomenon occurs because healthy plant growth may be enhanced by one or more of the following humidity of the plant environment, drainage and mineral characteristics of the soil, water requirements of the plant, duration and frequency of irrigation, the plant's tolerance to salinity, and the presence or absence of sodium ions in the dissolved solids. The relationship of these various factors to each other and to the specific plant needs to be determined before acceptable limits of dissolved solids can be estimated for growing specific crops (B.S. Environmental Protection Agency, 1973, p. 323-328).

The recommended limit for dissolved-solids concentrations in municipal water supplies is 500 mg/L (U.S. Environmental Protection Agency, 1973). In some areas, water of higher concentration are being used, however, drinking water containing high concentrations of dissolved solids are likely to contain substances that are highly objectionable due to mineral taste, physiological effects, or economic consequences.

The maps on this sheet show the areal distribution of the median dissolved-solids concentrations in water from each aquifer group in the study area. The median concentration of dissolved solids for each 100-mi² area was computed and plotted at the center of each 100-mi² area. Areas of equal dissolved-solids concentration were connected to define ranges of dissolved-solids concentrations. In those areas where no ranges appear or are discontinued, the available aquifer group is most likely not present or is not being used as a source of water. The maps constructed to show the distribution of dissolved solids for each of the aquifer groups are based on chemical analyses of water from about 21,000 sites. The median dissolved-solids concentrations in water from aquifers in each group will be discussed in ascending order (oldest to youngest):

Wilcox Aquifers

Aquifers in the Wilcox Group are separated from the underlying Cretaceous aquifers by the thick clay layer of the Midway Group. Aquifers in this group are the oldest and lowest aquifers in the Gulf Coast aquifer systems except for about 20,000 mi² in the northern part of the Mississippi embayment which includes some Upper Cretaceous sediments (Grubb, 1984).

The Wilcox aquifer underlies most of the study area. However, water-chemistry data are few in the following parts of the study area—all of Alabama, south one-half and northeast one-fourth of Louisiana, south one-half of both Mississippi and Alabama, and southwest Texas. There are no data from the coastline to 45 to 145 mi inland because of the depth of aquifers and the excessive salinity of the water. Dissolved-solids concentrations decrease with distance inland from the coastline and nearest to the upbay limit, which is approximately at the study area boundary on the north side.

Median dissolved-solids concentrations in water from the Wilcox aquifer increase markedly in a southerly direction from a line extending from northern Mobile County in southern Texas to southern Cherokee County, Texas, to northern Vernon Parish, Louisiana, to western West Carroll Parish, Louisiana, to northern Lawrence County, Mississippi, to northern Wayne County, Mississippi, to middle Clark County, Alabama. The increase is from about 3,000 mg/L (upper limit of slightly saline water) at the line to as much as 200,000 mg/L (concentrated brine) in some areas along the coastline and nearest to the southernmost Alabama. North of this line, the median dissolved-solids concentration decreases from about 3,000 mg/L to less than 500 mg/L. There are, however, local areas north of the line where the concentrations may be as much as 10,000 mg/L.

Geopressure zones occur along the downbay extent of the mappable data for the Wilcox aquifers (Jones, 1968). The effect of these zones on the concentrations of dissolved solids, however, does not appear to be significant at a regional scale.

Claiborne Aquifers

Aquifers in the Claiborne Group have an areal distribution of dissolved-solids concentrations similar to that of aquifers in the Wilcox Group. The main difference is that the line of markedly increasing concentrations of dissolved solids across the study area is located farther to the south or downbay. This has resulted in a slightly larger area of freshwater in the Claiborne Group than in the Wilcox Group even though the upbay limit of the Claiborne Group also is located farther to the south.

The area of low or no data between the downbay extent of the mappable data for the Claiborne Group and the coastline ranges from 25 to 130 mi, which is a smaller area than that for the Wilcox Group. There also are few data in Missouri and no data in a circular area located in northeast Texas and northwest Louisiana called the Sabine uplift where the Claiborne Group is not present. Other areas of few data occur in the following parts of the study area—Alabama, northern Arkansas, northeastern and southern Mississippi, and southern Louisiana.

Median dissolved-solids concentrations in water from the Claiborne aquifer increase markedly in a southerly direction from a line extending from middle Webb County in southwest Texas to southern Trinity County, Texas, to middle Vernon Parish, Louisiana, to southern Madison Parish, Louisiana, to northern Lawrence County, Mississippi, to northern Wayne County, Mississippi, to middle Washington County, Alabama. The concentrations range from the upper limit of slightly saline water (3,000 mg/L) at the line to concentrated brine (200,000 mg/L) in some areas along the coastline and nearest to the upbay area in Texas. North of this line, the water in the Claiborne aquifer is fresh having less than 1,000 mg/L of dissolved solids and in most of this area having less than 500 mg/L dissolved solids.

There are geopressure zones along the downbay extent of the mappable data for the Claiborne aquifer but like those in the Wilcox aquifer, they do not appear to affect the dissolved-solids concentrations significantly at a regional scale.

Jackson Aquifers

The Jackson Group is predominantly marine sediments of clay and sand and is considered a confining layer rather than an aquifer in most of the study area. However, some water-chemistry data is available in a few fairly large areas where aquifers occur within the sediments. The areas are in south central Arkansas and in a 45-mile-wide area extending from southeast Texas to the Texas-Louisiana border.

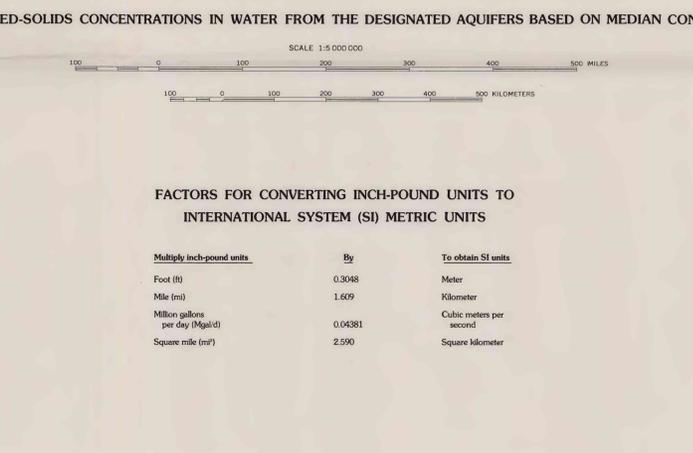
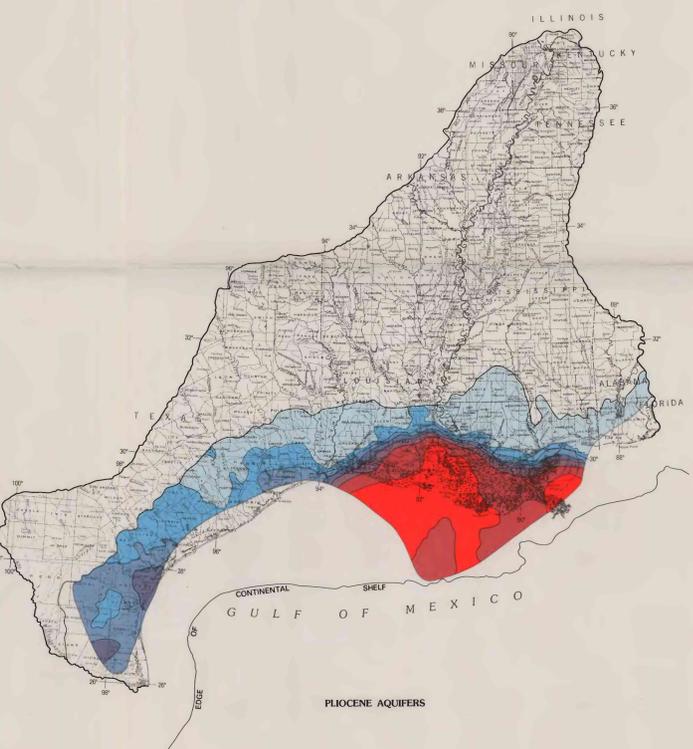
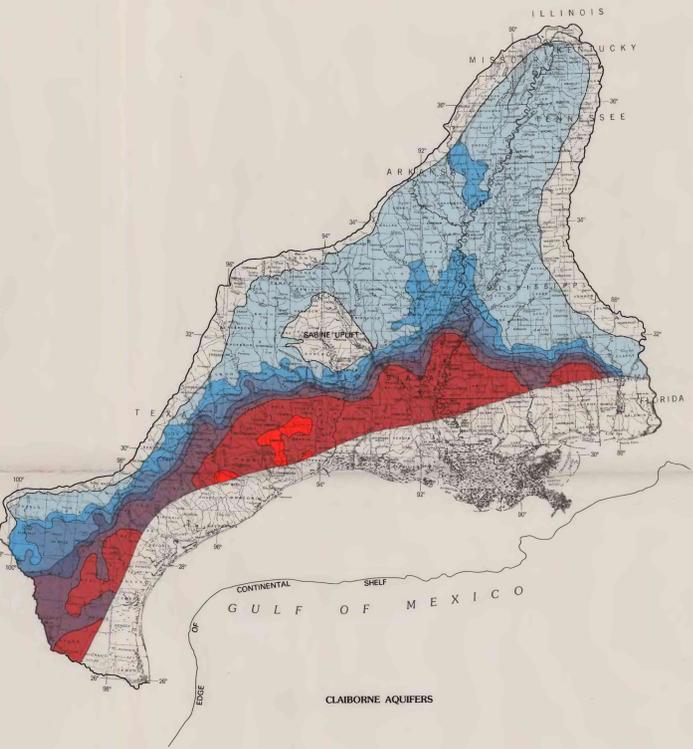
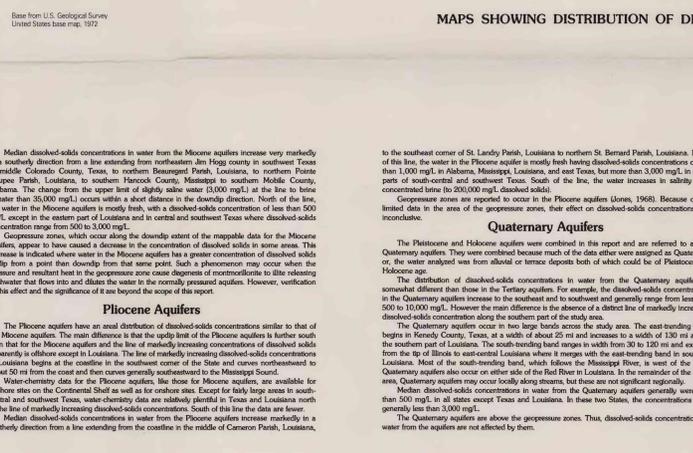
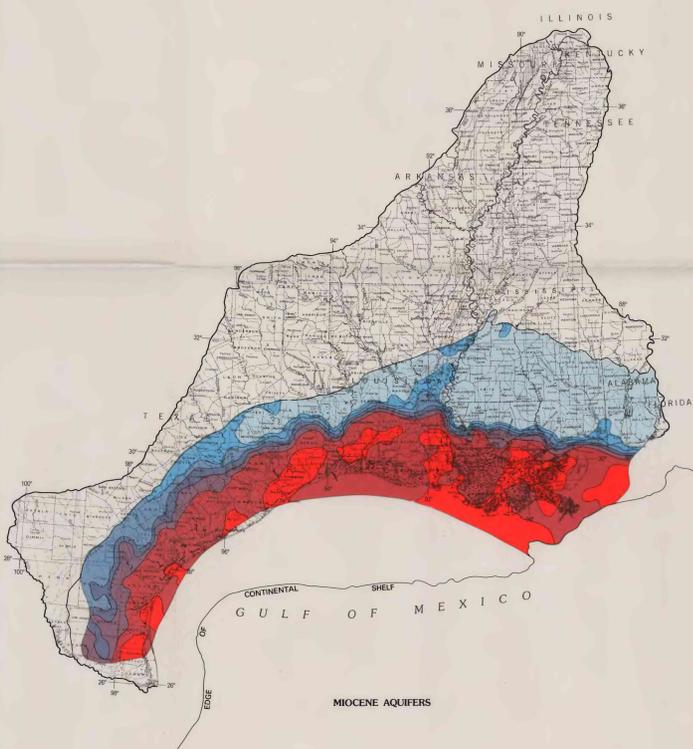
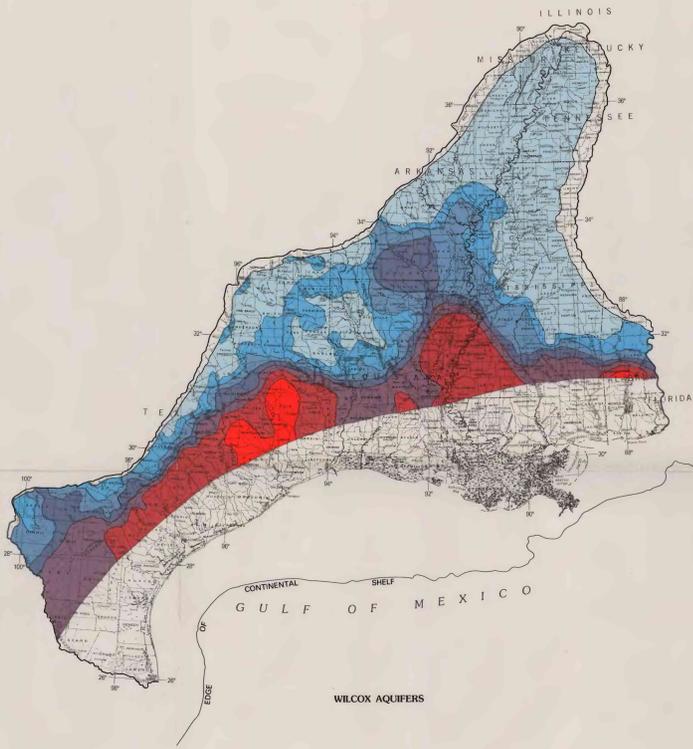
Median dissolved-solids concentrations in water from the Jackson aquifer in these areas in Arkansas generally are less than 3,000 mg/L. However, in Texas the median concentrations range from less than 1,000 mg/L (freshwater) to greater than 35,000 mg/L (brine).

Vicksburg Aquifers

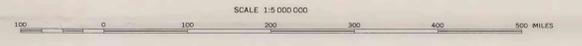
The Vicksburg Group of Oligocene age is also composed of predominantly marine sediments and is considered a confining layer in this study. However, water is obtained locally from the Vicksburg Group in small areas in Mississippi (Gard, 1979). Because these aquifers are not significant regionally, the dissolved-solids concentrations were not mapped in this study.

Miocene Aquifers

The distribution of median dissolved-solids concentrations in water from the Miocene aquifer is similar to those for the Wilcox and Claiborne aquifers in the southern part of the study area. Because the Miocene aquifer underlies the Wilcox and Claiborne aquifers stratigraphically, they generally are at shallower depths. Also the downbay extent of data for the Miocene aquifer is beyond the coastline and includes parts of the Continental Shelf. The upbay limit of the Miocene aquifer is further south than that for either the Wilcox or Claiborne aquifer. As a result there are data for only the western tip of Florida and the southern one-half of the study areas in each of the following States—Alabama, Louisiana, Mississippi, and Texas.



MAPS SHOWING DISTRIBUTION OF DISSOLVED-SOLIDS CONCENTRATIONS IN WATER FROM THE DESIGNATED AQUIFERS BASED ON MEDIAN CONCENTRATION PER 100-SQUARE-MILE AREA



FACTORS FOR CONVERTING INCH-POUND UNITS TO INTERNATIONAL SYSTEM (SI) METRIC UNITS

| Multiple inch-pound units | By | To obtain SI units |
|----------------------------------|---------|-------------------------|
| Foot (ft) | 0.3048 | Meter |
| Mill (mil) | 1.609 | Millimeter |
| Million gallons per day (Mgal/d) | 0.04381 | Cubic meters per second |
| Square mile (mi ²) | 2.590 | Square kilometer |



EXPLANATION
DISSOLVED-SOLIDS CONCENTRATION, IN MILLIGRAMS PER LITER

| |
|-------------------------|
| 0-500 |
| 500-1,000 |
| 1,000-3,000 |
| 3,000-10,000 |
| 10,000-35,000 |
| 35,000-100,000 |
| 100,000-200,000 |
| No data |
| Boundary of study area |
| Boundary of upbay limit |

DISSOLVED-SOLIDS CONCENTRATIONS AND PRIMARY WATER TYPES, GULF COAST AQUIFER SYSTEMS, SOUTH-CENTRAL UNITED STATES

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1988

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