

(200)

2290

no. 76-550

UNITED STATES
(DEPARTMENT OF INTERIOR)
GEOLOGICAL SURVEY,

[Reports - Open file series]

Geochemical and Hydrologic Data For Wells and
Springs in Thermal-Spring Areas of the Appalachians

William
By W. A. Hobba, Jr., J. C. Chemerys, D. W. Fisher,
and F. J. Pearson, Jr.

Open-File Report No. 76-550

Reston, Virginia

July, 1976

CONTENTS

	Page
Abstract-----	1
Introduction-----	2
Hydrologic techniques-----	3
Analytical techniques-----	4
Field procedure-----	4
Laboratory procedure-----	6
Summary-----	10
References-----	11

ILLUSTRATIONS

Figure 1. Map showing locations of sampled thermal spring areas in the Appalachians-----	13
2. Map showing geology and spring and well locations in New York - Lebanon Springs area-----	18
3. Map showing geology and spring and well locations in Pennsylvania - Perry County Warm Springs area-----	20
4. Map showing geology and spring and well locations in West Virginia - Berkeley Springs area-----	22
5. Map showing geology and spring and well locations in West Virginia - Minnehaha Springs area-----	24
6. Map showing geology and spring and well locations in Virginia - Bolar Springs, Hot Springs-Warm Springs, Falling Springs areas-----	26
6A. Detailed maps showing warm springs and well locations in Virginia-----	29
7. Map showing geology and spring and well locations in North Carolina - Hot Springs area-----	30
8. Map showing geology and spring and well locations in Georgia - Warm Springs area-----	32

TABLES

	Page
Table 1. Results of chemical analyses of waters from Appalachian warm springs areas----	14
2. Results of dissolved gas and isotope analyses of waters from Appalachian warm springs areas-----	16
Records of springs and wells in the Appalachian warm springs areas	
3. New York - Lebanon Springs area-----	19
4. Pennsylvania - Perry County Warm Springs area-----	21
5. West Virginia - Berkeley Springs area--	23
6. West Virginia - Minnehaha Springs area	25
7. Virginia - Bolar Springs, Hot Springs- Warm Springs, and Falling Springs area	27
8. North Carolina - Hot Springs area-----	31
9. Georgia - Warm Springs area-----	33
10. Factors for converting English units to international system(SI) units-----	34

GEOCHEMICAL AND HYDROLOGIC DATA FOR WELLS AND SPRINGS IN
THERMAL-SPRING AREAS OF THE APPALACHIANS

By

W. A. Hobba, Jr., J. C. Chemerys, D. W. Fisher,
and F. J. Pearson, Jr.

ABSTRACT

Current interest in geothermal potential of thermal-spring areas in the Appalachians makes all data on thermal springs and wells in these areas valuable. Presented here without interpretive comment are maps showing selected springs and wells and tables of physical and chemical data pertaining to these wells and springs. The chemical tables show compositions of gases (oxygen, nitrogen, argon, methane, carbon dioxide, and helium), isotope contents (tritium, carbon (13), and oxygen (18)), trace and minor element chemical data, and the usual complete chemical data.

INTRODUCTION

The U. S. Geological Survey is presently studying the ground-water hydrology and geochemistry of thermal spring areas in the Appalachians. This report presents data available through June 1976 collected for this study. Field work is still in progress, and more data will be available for some of these same sites at a later date and will be published in a similar format. A formal publication with interpretation is anticipated in the future, but these data are being released in the present form because of current interest in geothermally related data.

The stratigraphic nomenclature used in this report was determined from several sources and may not necessarily follow the usage of the U. S. Geological Survey.

The authors wish to thank all of the land owners, well drillers, and the various state geological surveys who cooperated in collecting these data.

HYDROLOGIC TECHNIQUES

During the summer of 1975, a reconnaissance of ten thermal-spring sites in the Appalachian was carried out. Much of the spring and well data given here was obtained by canvassing homeowners. However, some of the data for Virginia and New York were obtained from well drillers' files, and some of the data for Pennsylvania were obtained from drillers' records on file at the Pennsylvania Geological Survey.

After the wells and springs were inventoried, some were selected for sampling. Generally, ground water was sampled at three or more sites in the vicinity of each warm spring: the warm spring itself, a well or spring tapping the rock units adjacent to the rock unit from which the thermal spring issues, and a well or spring in the recharge area for the thermal spring, as nearly as it could be determined. These sites were sampled in the fall when ground water levels were low, and some were sampled again in the spring when levels were higher.

ANALYTICAL TECHNIQUES

Field Procedures

Samples were field-filtered, using pressurized nitrogen, through 0.20 μ m (micrometre) membrane filters, for all laboratory determinations except dissolved gases and stable carbon isotope analyses. Samples for cation analyses were acidified with hydrochloric acid. Specific conductances were measured in the field with field model conductivity meters and electrodes. Temperatures were measured with a 12-inch mercury-in-glass thermometer having a range of 0 to 50 degrees Celsius with markings every 0.2 degrees.

Sampling devices for the dissolved-gas analyses consisted of 500 ml (millilitre) glass flasks with evacuated side arms. In use, water was pumped into the flask to fill it, and an additional 2 litres of sample was pumped through the flask in order to flush adsorbed gases from the glass wall. After the sampler was filled and flushed, the inlet and outlet stopcocks were closed to the supply and to the atmosphere, respectively, then opened to the previously evacuated side chamber. The samplers were then returned to the laboratory for analysis of the vapor phase.

Samples for stable carbon isotope ratio measurement were fixed in the field to prevent loss of dissolved carbonate as gaseous CO₂. (Gleason, Friedman, and Hanshaw, 1969). The samples were collected with minimum aeration in 1-quart (0.946 litres) glass bottles. To this was added, by pipet, 50 ml of a strontium hydroxide solution, prepared by dissolving 1 lb (453.6 g) of reagent grade SrCl₂·6H₂O in a 5-pint (2.37 l) bottle of reagent grade concentrated (30 percent) NH₄OH. This solution raised the

pH of the sample, so that all the dissolved carbonate was present as the carbonate (CO_3) ion, which then combined with the strontium in the solution to precipitate SrCO_3 . Samples for stable oxygen isotope ratio analyses and tritium were collected in glass bottles with minimum aeration and exposure to the atmosphere.

pH, Alkalinity

pH measurements were made in the field when the samples were collected. Commercial pH meters, electrodes, and liquid buffers were used. The combination electrode was immersed in a bath through which the sample water flowed. This bath also served to maintain the pH buffers at the sample temperature. The meter and electrode were standardized with pH 7 and pH 4 buffers at the observed sample temperature, and the sample pH was then determined.

Alkalinities of freshly drawn samples were determined by the electro-metric titration method described in Brown, Skougstad, and Fishman (1970, p. 42-43).

Dissolved Oxygen and Sulfide

A commercial kit (Hach Chemical Co., Md1. OX-2P^{*}) was used for determinations of high dissolved oxygen concentrations. Concentrations of dissolved oxygen less than 2 mg/l (milligrams per litre) were determined more precisely by the modified Winkler method (Brown, Skougstad, and Fishman, 1970, p. 126). A Hach Model HS-6 sulfide kit (based on the

*The use of brand names in this report is for identification purposes only and does not imply endorsement by the U.S. Geological Survey.

methylene blue method) was used for determinations of sulfide in those samples containing less than 1 mg/l dissolved oxygen. Lower detection limit for sulfide by this method is about 0.02 mg/l. Sulfide was found in only one sample.

Laboratory Procedures

Appropriate analytical techniques described in Brown, Skougstad, and Fishman (1970) were used for the analyses of dissolved chemical constituents, except for the following:

- 1) Ammonia was separated in a micro-Kjeldahl still. In this device, high pH for distillation (achieved by making samples 0.1 normal in sodium hydroxide) is necessary for quantitative removal of the ammonia.
- 2) Nitrate was determined by the phenoldisulfonic acid method as described by Rainwater and Thatcher (1960, method D:25b-1).
- 3) Arsenic and selenium were analyzed by a graphite furnace atomic adsorption technique.

Copper was determined in all the laboratory-analyzed samples. Concentrations were less than 0.01 mg/l, except as noted in the table of analytical results. Bromide analyses were performed by the Atlanta Central Laboratory of the U. S. Geological Survey.

Dissolved Gases

Dissolved gases were analyzed chromatographically, using columns packed with porous polymer beads to achieve the required separations, hot wire thermal conductivity bridge detectors, and electronic integration and printout of detector signals for quantitative determinations of the various gases. Standard gas mixtures were used to calibrate the chromatograph response.

Argon was used as the carrier gas for helium analyses; a helium carrier was employed for the other determinations. Nitrogen, oxygen, and argon were separated on a 45-foot Porapak^{*} QS column in a dry ice-isopropanol bath (Porapak^{*} Brochure). Methane and carbon dioxide were separated from each other and from the major atmospheric gases on a Porapak^{*} QS-Porapak^{*} R column maintained at 50°C. Use of this combined column was suggested by Wilhite and Hollis (1968). Helium separations were made on a Porapak^{*} QS column in an ice-water bath.

Sample flasks are attached to the chromatograph inlet with short lengths of plastic tubing. Tubing, inlet, and calibrated chromatograph sample loop are evacuated, then the flask stopcock is opened to admit gases to the loop. The analytical sequence is started by actuating a valve-switching arrangement, which diverts carrier gas flow through the loop to the column. Volumes of the sampler side tube, plastic tubing, inlet system, and sample loop are known, so that total amounts of individual gases in the vapor phase can be calculated. From these data, together with the known volumes of the larger water chambers and published gas solubility data, dissolved gas concentrations and partial pressures at sampling temperatures are determined.

*Registered trademark of porous polymer beads sold by Waters Associates, Framingham, Mass.

Stable Isotopes

Stable isotope ratios were measured by a commercial laboratory using a mass spectrometer. Because it is difficult to measure absolute values of isotope ratios with any precision, such ratios are measured and reported relative to arbitrary but widely used standards. The measurements are reported in the δ notation, where:

$$\delta \text{ (o/oo)} = \left(\frac{R_{\text{sample}}}{R_{\text{standard}}} - 1 \right) \times 1000$$

R_{sample} and R_{standard} are the isotope ratios ($^{13}\text{C}/^{12}\text{C}$, $^{18}\text{O}/^{16}\text{O}$) in the sample and standard, respectively. The standard to which the carbon isotope ratios reported here are referred is the PDB standard (Craig, 1957), whereas the oxygen isotope ratios are referred to the SMOW standard (Craig, 1961). The precision of the analyses reported here is considered to be ± 0.1 o/oo.

The 1-quart samples treated in the field to precipitate all the carbonate as SrCO_3 were filtered under nitrogen and the SrCO_3 precipitate thoroughly rinsed and dried. This carbonate was then reacted with 100 percent H_3PO_4 to produce the CO_2 gas required for the mass spectrometric analysis of $^{13}\text{C}/^{12}\text{C}$ ratios (McCrea, 1950).

$^{18}\text{O}/^{16}\text{O}$ ratios were measured on a second CO_2 gas sample. This gas was prepared in the laboratory by equilibrating CO_2 with the water samples collected for that purpose. The $^{18}\text{O}/^{16}\text{O}$ ratio of the water is calculated from the $^{18}\text{O}/^{16}\text{O}$ ratio of the gas equilibrated with it.

The tritium analyses were made (U.S.G.S., Reston) by liquid scintillation counting after electrolytic enrichment of the sample by a factor of about 17. Tritium concentrations are given in tritium units (TU). One TU corre-

sponds to a concentration of 1 tritium atom per 10^{18} hydrogen atoms and equals 3.2 picocuries per litre. The analytical errors reported are based on the statistics of counting random radioactive decay events and are 1 standard deviation (1σ) errors.

SUMMARY

Figure 1 shows the locations of the warm spring areas that were sampled in the Appalachian Mountains. Tables 1 and 2 list the springs and wells sampled and show the results of chemical, dissolved gas and isotope analyses. Arsenic and selenium results will be tabulated in a later data report. In general, arsenic and selenium concentrations in these waters is very low (less than 5 $\mu\text{g}/\text{l}$). Figures 2-8 show the locations of springs and wells in the warm springs areas. Tables 3-9 give the data for the springs and wells shown on the location maps.

REFERENCES

- Brown, Eugene, Skougstad, M. W., and Fishman, M. J., 1970, Methods for collection and analysis of water samples for dissolved minerals and gases: U.S. Geological Survey Techniques of Water-Resources Inv., chap. A1, 160 p.
- Butts, Charles, 1933, Geologic Map of the Appalachian Valley in Virginia: U.S. Geol. Survey, 1 p.
- Cardwell, D. E., Erwin, R. B., and Woodward, H. P., 1968, Geologic Map of West Virginia: West Virginia Geol. and Econ. Survey, 1 pl.
- Craig, H., 1957, Isotopic standard for carbon and oxygen and correction factors for mass spectrometric analysis of carbon dioxide: *Geochim. Cosmochim Acta*, v. 12, p. 133-149.
- Craig, Harmon, 1961, Standard for reporting concentrations of deuterium and oxygen-18 in natural waters: *Science*, v. 133, p. 1833-1834.
- Dyson, J. L., 1967, Geology and Mineral Resources of the Southern Half of the New Bloomfield Quadrangle, Pennsylvania: Pennsylvania Geol. Survey 4th ser., Atlas 137cd, 86 p.
- Gleason, J. D., Friedman, Irving, and Hanshaw, B. B., 1969, Extraction of dissolved carbonate species from natural waters for carbon isotope analysis: U.S. Geol. Survey Prof. Paper 650-D, p. D248-D250.
- Grimsley, G. P., 1916, Jefferson, Berkeley and Morgan Counties: West Virginia Geol. and Econ. Survey County Geologic Rept. Series, 644 p. 3 pl.
- Hewett, D. F., and Crickmay, G. W., 1937, The Warm Springs of Georgia Their Geologic Relations and Origin-A Summary Report: U.S. Geol. Survey Water Supply Paper 819

- Johnston, H. E., 1970, Ground-water Resources of the Loysville and Mifflintown
Quadrangles in South Central Pennsylvania: Pennsylvania Geol. Survey
4th Ser., Bull. W-27, 96 p.
- McCrea, J. M., 1950, On the Isotopic Chemistry of Carbonates and a
Paleotemperature Scale: Jour. Chem. Physics, v. 18, p. 849-857.
- Porapak (R) Brochure, Waters Associates, Framingham, Mass.
- Price, P. H., 1929, Pocahontas County: West Virginia Geol. and Econ.
Survey County Rept. Series, 531 p., 2pl.
- Rainwater, F. H., and Thatcher, L. L., 1960, Methods for Collection and
Analysis of Water Samples: U.S. Geological Survey Water-Supply
Paper 1454, 301 p.
- Reger, D. B., 1919, Webster County and Portions of Mingo District of
Randolph County, South of Valley Fork of Elk River: West Virginia
Geol. and Econ. Survey County Rept. Series, 682 p., 2 pl.
- Stose, G. W., and Stose, A. J., 1947, Origin of the Hot Springs at
Hot Springs, North Carolina: Am. Jour. Sci., v. 245, pt. 2, p. 624-
644.
- Wilhite, W. F., and Hollis, O. L., 1968, Jour. Gas Chromatography, v. 6,
p. 84-88.

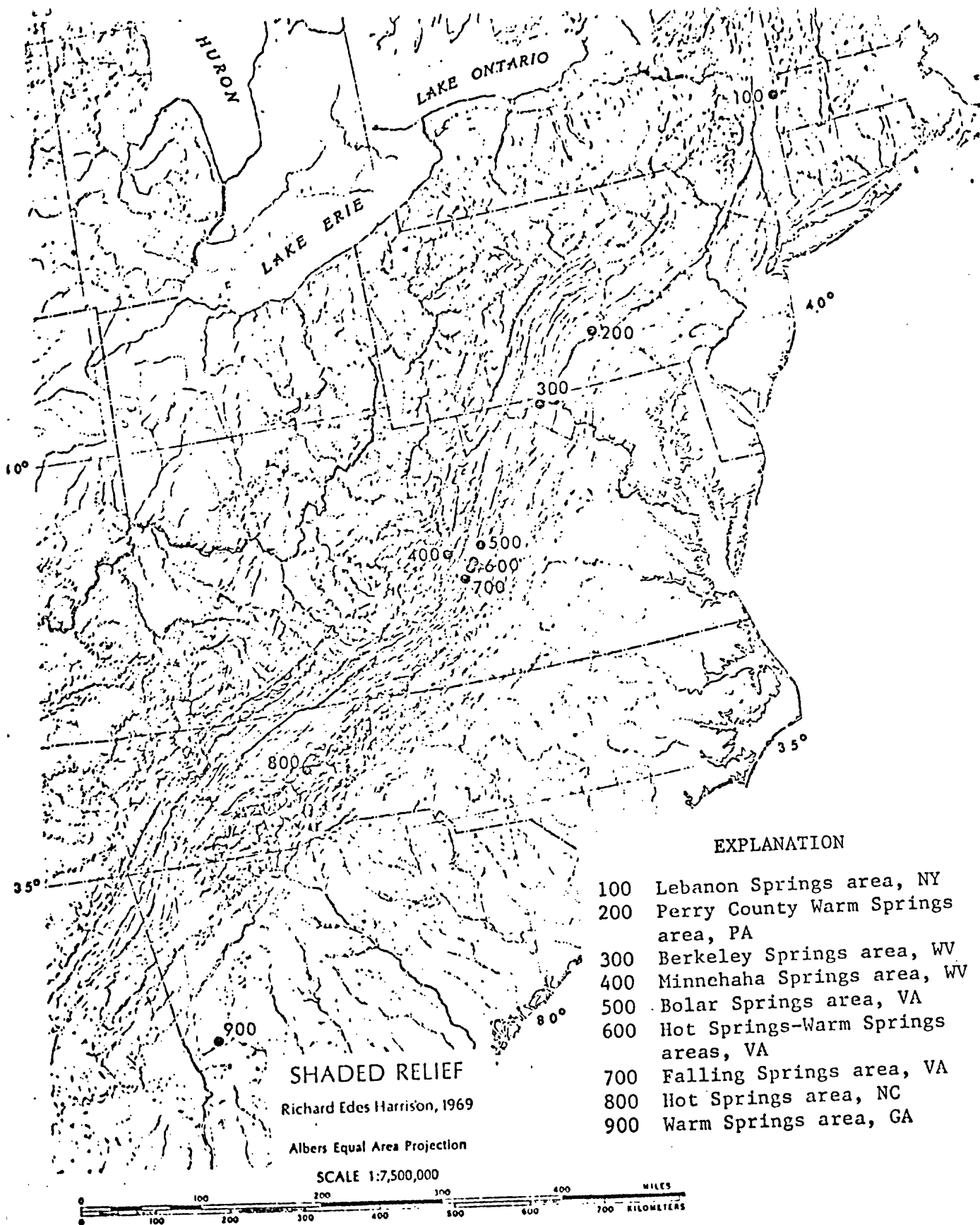


Figure 1.--Map showing locations of sampled thermal-spring areas in the Appalachians

Table 1.--Results of chemical analyses of waters from Appalachian warm springs areas
[Dissolved concentrations in milligrams per litre (mg/l)]

Sam- ple No.	Date	Tem- pera- ture (°C)	pH	Silica (SiO ₂)	Alu- min- um (Al)	Iron (Fe)	Man- gane- se (Mn)	Zinc (Zn)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Stron- tium (Sr)	Sod- ium (Na)	Po- tas- sium (K)	Lith- ium (Li)	Alka- lin- ity (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Fluo- ride (F)	Bro- mide (Br)	Ni- trate (NO ₃)	Ammo- nium (NH ₄)	Phos- pho- rus (PO ₄)	Bo- ron (B)
NEW YORK - LEBANON SPRINGS AREA																							
101	10-75	21.9	7.60	12	0.014	0.00	0.00	0.00	35	16	0.4	6.9	1.2	0.00	166	25	6.5	0.3	0.056	1.1	0.05	0.03	0.02
109	10-75	14	7.60	11	0.014	0.01	0.00	0.00	31	17	0.4	7.0	1.3	0.00	168	23	6.4	0.3	0.043	0.4	0.04	0.02	0.02
114	10-75	10.9	7.20	6.0	0.000	0.01	0.11	0.00	133	12	2.4	28	0.8	0.01	186	--	0.8	0.5	0.012	0.5	0.09	0.01	0.03
116	10-75	10.4	7.55	14	0.031	0.01	0.00	0.11	49	22	0.1	2.8	0.4	0.00	241	21	1.1	0.3	0.015	0.2	0.03	0.03	0.00
PENNSYLVANIA - PERRY COUNTY - WARM SPRINGS AREA																							
201	10-75	17.8	7.2	9.0	0.000	0.01	0.00	0.00	38	3.3	0.3	1.6	0.5	0.00	135	8.3	0.8	0.1	0.012	0.1	0.08	0.10	0.00
202	10-75	12	6.85	4.2	0.000	0.02	0.00	0.00	25	1.4	0.1	0.6	0.6	0.00	72	0.4	1.4	0.1	0.016	0.1	0.02	0.62	0.00
216	10-75	12.8	6.50	7.3	0.014	0.00	0.00	0.00	17	2.1	0.1	1.4	0.6	0.00	58	6.3	1.1	0.1	0.030	0.5	0.09	1.2	0.00
218	10-75	12.0	7.20	5.9	0.000	0.00	0.00	0.00	65	5.9	0.4	3.1	0.9	0.00	176	29	14	0.2	0.043	6.9	0.06	0.15	--
WEST VIRGINIA - BERKELEY SPRINGS AREA																							
301	10-75	22.2	6.99	9.5	0.024	0.00	0.00	0.00	45	5.6	0.5	4.1	1.0	0.00	168	17	6.4	0.1	0.023	0.5	0.03	0.13	0.02
302	10-75	13.0	6.8	20	0.034	0.96	0.10	0.11	80	11	0.4	6.2	0.6	0.03	254	55	1.1	0.2	0.018	0.2	0.22	0.02	0.03
a303	9-75	13.9	7.30	18	0.000	1.4	0.08	0.00	55	20	1.0	17	0.4	0.04	226	42	23	0.2	0.086	0.1	0.38	0.07	0.03
b307	10-75	16	6.97	7.8	0.007	0.00	0.00	0.00	106	24	1.7	3.8	0.7	0.00	410	25	10	0.2	0.063	6.5	0.05	0.02	0.02
WEST VIRGINIA - MINNEHAHA SPRINGS AREA																							
401	11-75	20.1	7.88	8.4	0.000	0.00	0.00	0.00	39	6.0	0.4	2.2	0.8	0.00	116	38	0.8	0.2	0.006	0.1	0.04	0.24	0.01
402	11-75	11.6	7.30	14	0.000	0.25	0.19	2.8	61	15	0.4	4.2	0.4	0.02	230	37	0.8	0.2	0.016	0.1	0.18	0.14	0.01
403	11-75	13.3	7.20	6.8	0.006	0.00	0.00	0.04	36	0.7	0.1	1.2	0.8	0.00	84	2.8	3.4	0.1	0.020	23	0.01	0.65	--
406	11-75	10.3	7.63	5.0	0.000	0.00	0.00	0.00	36	1.3	0.0	0.4	0.8	0.00	113	5.9	0.8	0.1	0.010	0.1	0.11	0.14	0.00
c407	11-75	10.8	5.50	6.8	0.000	0.03	0.00	0.05	14	1.2	0.0	2.6	1.5	0.00	20	1.6	5.3	0.1	0.018	27	0.10	0.11	--
VIRGINIA - BOLAR SPRINGS AREA																							
d501	12-75	22.2	7.48	11	0.000	0.00	0.00	0.00	58	14	0.3	1.6	2.3	0.01	205	29	0.8	0.3	0.011	0.9	0.12	0.03	0.03
VIRGINIA - HOT SPRINGS - WARM SPRINGS AREA																							
602	11-75	39.9	6.65	21	0.000	0.17	0.00	0.05	132	40	1.3	7.0	13	0.05	454	130	2.6	1.1	0.029	0.1	0.09	0.02	0.03
605	11-75	15.2	7.20	8.2	0.000	0.00	0.00	0.03	60	9.7	0.3	1.7	2.9	0.01	174	35	9.5	0.3	0.031	0.8	0.06	0.04	0.00
e607A	11-75	19.6	7.15	15	0.000	0.16	0.08	0.39	113	34	1.1	2.9	5.8	0.03	--	114	2.4	0.6	0.024	0.1	0.08	0.13	0.02
f607B	2-76	17.9	7.6	7.2	--	--	0.14	0.07	53	18	0.4	2.6	3.0	0.01	--	37	2.8	0.4	0.024	0.4	--	0.13	0.02
g607C	2-76	19.5	6.9	14	0.018	0.34	0.27	1.0	119	35	1.1	4.9	11	0.04	--	116	2.3	0.9	0.028	0.1	--	0.11	0.04
h607D	2-76	32.2	--	19	0.029	0.04	0.04	0.18	123	37	1.4	5.6	13	0.04	--	132	2.4	1.0	0.025	0.1	--	0.13	0.02
619	11-75	15.8	7.32	8.9	0.000	0.00	0.00	0.04	80	14	0.5	2.1	2.6	0.02	196	85	2.2	0.5	0.018	2.4	0.05	0.04	0.02
625	11-75	12.0	7.30	6.0	0.000	0.16	0.02	0.95	60	16	0.5	1.5	1.2	0.03	248	14	0.7	0.4	0.010	0.1	0.06	0.00	0.01
627	11-75	10.5	7.48	5.4	0.000	0.00	0.00	0.10	96	7.6	0.1	5.7	0.6	0.00	300	10	10	0.1	0.050	10	0.10	0.02	--
1631	11-75	35.4	7.32	21	0.000	0.00	0.00	0.05	112	28	1.9	3.7	7.4	0.04	197	232	1.5	1.4	0.016	0.1	0.08	0.01	0.02
633	12-75	9.1	4.50	3.0	0.048	0.01	0.00	0.00	5	0.2	0.0	0.4	0.2	0.00	--	0.4	0.6	0.0	0.018	0.5	0.05	0.00	0.01

See footnotes at end of table.

Table 1.--Results of chemical analyses of waters from Appalachian warm springs areas--Continued

Sam- ple No.	Date	Tem- pera- ture (°C)	pH	Silica (SiO ₂)	Alu- min- um (Al)	Iron (Fe)	Man- ga- nese (Mn)	Zinc (Zn)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Stron- tium (Sr)	Sod- ium (Na)	Po- tas- sium (K)	Lith- ium (Li)	Alka- lin- ity (HCO ₃)	Sul- fate (SO ₄)	Chlo- ride (Cl)	Fluo- ride (F)	Bro- mide (Br)	Ni- trate (NO ₃)	Ammo- nium (NH ₄)	Phos- pho- rus (PO ₄)	Bo- ron (B)
VIRGINIA - FALLING SPRINGS AREA																							
701	12-75	25.0	7.32	18	0.000	0.00	0.00	0.00	164	35	2.4	3.8	16	0.17	315	289	2.5	1.2	0.028	1.3	0.10	0.07	0.02
702	11-75	27.2	7.15	24	.000	.16	.02	.30	234	55	3.5	6.6	2.5	.30	423	507	4.3	2.4	.044	.1	.08	.00	.04
NORTH CAROLINA - HOT SPRINGS AREA																							
801	12-75	39.3	7.50	31	.023	.00	.00	.05	135	29	3.8	10	10	.07	126	369	3.8	1.4	.049	.1	.07	.03	.02
803	12-75	15.2	6.75	25	.000	.12	.62	.25	13	10	.1	6.6	4.5	.02	114	9.3	5.9	.2	.047	.1	.10	.38	.01
804	12-75	12.5	7.10	7.4	.000	.02	.02	.04	4.9	3.4	.0	.9	1.5	.00	29	4.8	.8	.1	.010	1.4	.04	.12	.00
805	12-75	13.2	7.75	11	.000	.01	.00	.03	19	11	.0	.3	2.2	.00	108	1.8	.5	.2	.017	.4	.08	.12	.01
GEORGIA - WARM SPRINGS AREA																							
901	12-75	30.9	7.40	20	.000	.00	.00	.00	22	13	.1	1.2	3.8	.00	122	7.1	1.9	.1	.004	.3	.04	.08	.02
905	12-75	17.7	4.00	7.5	.015	.00	.00	.00	.3	.2	.0	.9	.6	.00	--	.4	1.2	.0	.011	.2	.09	.00	.00
907	12-75	33.8	6.42	21	.000	.00	.00	.00	12	9.7	.0	1.0	4.0	.00	--	5.7	1.2	.1	.013	.1	.05	.04	.00
909	12-75	17.0	7.02	31	.000	.10	.07	.03	19	3.4	.1	10.0	3.5	.00	94	6.3	1.8	.3	.017	.1	.08	.13	.02

a Contains 0.03 mg/l of copper (Cu).

b Contains 0.05 mg/l of copper (Cu).

c Contains 0.07 mg/l of copper (Cu).

d Contains 0.02 mg/l of copper (Cu).

e Sample pumped from 215 foot depth.

f Sample collected at 335 foot depth with sampling tube on logger.

g Sample collected at 400 foot depth with sampling tube on logger.

h Sample collected at 745 foot depth with sampling tube on logger.

i Contains 0.75 mg/l hydrogen sulfide (H₂S).

Table 2.--Results of dissolved gas and isotope analyses of waters from Appalachian warm springs areas

Sample number	Date	Temperature (°C)	Gas (concentrations in mg/l)							Isotope results				
			Field dissolved oxygen (O ₂)	Lab dissolved oxygen (O ₂)	Nitrogen (N ₂)	Argon (Ar)	Methane (CH ₄)	Carbon dioxide (CO ₂)	Helium (He)	Tritium		δ ¹³ C o/oo	δ ¹⁸ O o/oo	
										T.U.	±1σ			
NEW YORK - LEBANON SPRINGS AREA														
101	10-75	21.9	4.5	4.3	24	0.96	<.01	6.4	Tr, <.002	11.6	±1.0	-11.9	-9.9	
109	10-75	14	3.0	2.4	24	.90	<.01	3.7	<.002	1.0	±1.4	-13.1	-10.0	
114	10-75	10.9	7.2	7.3	20	.78	<.01	6.6	<.002	--	--	--	--	
116	10-75	10.4	.8	(a)	(a)	(a)	<.01	9.8	<.002	.6	±1.0	-11.5	-9.4	
PENNSYLVANIA - PERRY COUNTY WARM SPRINGS AREA														
201	10-75	17.8	6.0	6.5	20	.82	<.01	8.2	<.002	6.0	±1.0	-13.4	-7.7	
202	10-75	12	6.5	(a)	(a)	(a)	<.01	9.6	<.002	--	--	--	--	
216	10-75	12.8	8	8.7	19	.82	<.01	22	<.002	15.5	±1.4	-12.9	--	
218	10-75	12.0	10.2	8.7	18	.68	<.01	8.5	<.002	78.1	±4.1	-12.3	-7.5	
WEST VIRGINIA - BERKELEY SPRINGS AREA														
301	10-75	22.2	4.5	3.6	22	--	<.01	25	<.002	2.7	±.8	-15.2	-8.7	
302	10-75	13.0	10	8.8	19	--	<.01	20.7	--	36.2	±2.4	-15.5	-7.9	
303	9-75	13.9	0	1.4	26	.89	.020	15	<.002	--	--	--	--	
307	10-75	16	6.7	(a)	(a)	(a)	<.01	55	<.002	71.8	±3.7	-15.2	-7.8	
WEST VIRGINIA - MINNEHAHA SPRINGS AREA														
401	11-75	20.1	3.5	3.2	19	.80	<.01	2.7	Tr, <.0001	2.5	±.9	-11.4	-8.5	
402	11-75	11.6	<.1	.3	28	1.06	.006	16.6	<.0001	41.0	±2.4	-14.4	-8.1	
403	11-75	13.3	9.0	--	--	--	--	--	--	--	--	--	--	
406	11-75	10.3	11.2	10.3	19	.73	<.01	3.3	<.0001	62.9	±3.6	-13.1	-7.2	
407	11-75	10.8	10.5	8.9	22	.78	<.01	67	<.0001	87.6	±4.5	-22.9	-8.0	
VIRGINIA - BOLAR SPRINGS AREA														
501	12-75	22.2	5.0	4.9	22	.82	<.005	16	Tr, <.0002	24.4	±2.0	-9.2	-8.0	
VIRGINIA - HOT SPRINGS - WARM SPRINGS AREA														
602	11-75	39.9	2.0	.6	17	.72	Tr, <.006	136	.0007	-3.1	±1.8	-5.6	-7.4	
605	11-75	15.2	9.2	8.5	19	.72	<.01	14.4	<.0001	41.0	±2.4	-8.8	-7.9	
b607A	11-75	19.6	.5	.4	25	.96	<.01	115	.0010	2.4	±2.3	-7.5	-7.8	
619	11-75	15.8	8.0	7.1	19	.74	<.007	13.4	Tr, <.0002	36.0	±2.8	-10.4	-7.7	
625	11-75	12.0	1.7	.2	30	1.13	Tr, <.02	15.5	<.0001	4.9	±1.8	-7.7	-7.9	
627	11-75	10.5	7.0	7.1	26	.89	<.007	31	<.0001	64.6	±3.5	--	--	
631	11-75	35.4	.1	<.01	16	.79	Tr, <.006	21	.0001	.9	±.9	-8.2	-7.9	
633	12-75	9.1	8.5	8.0	19	.78	<.005	50	<.0001	--	--	--	--	

a Air contaminated sample.

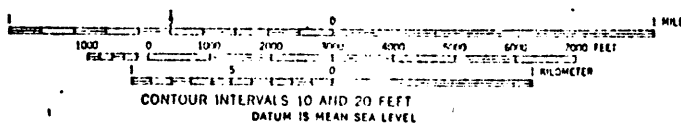
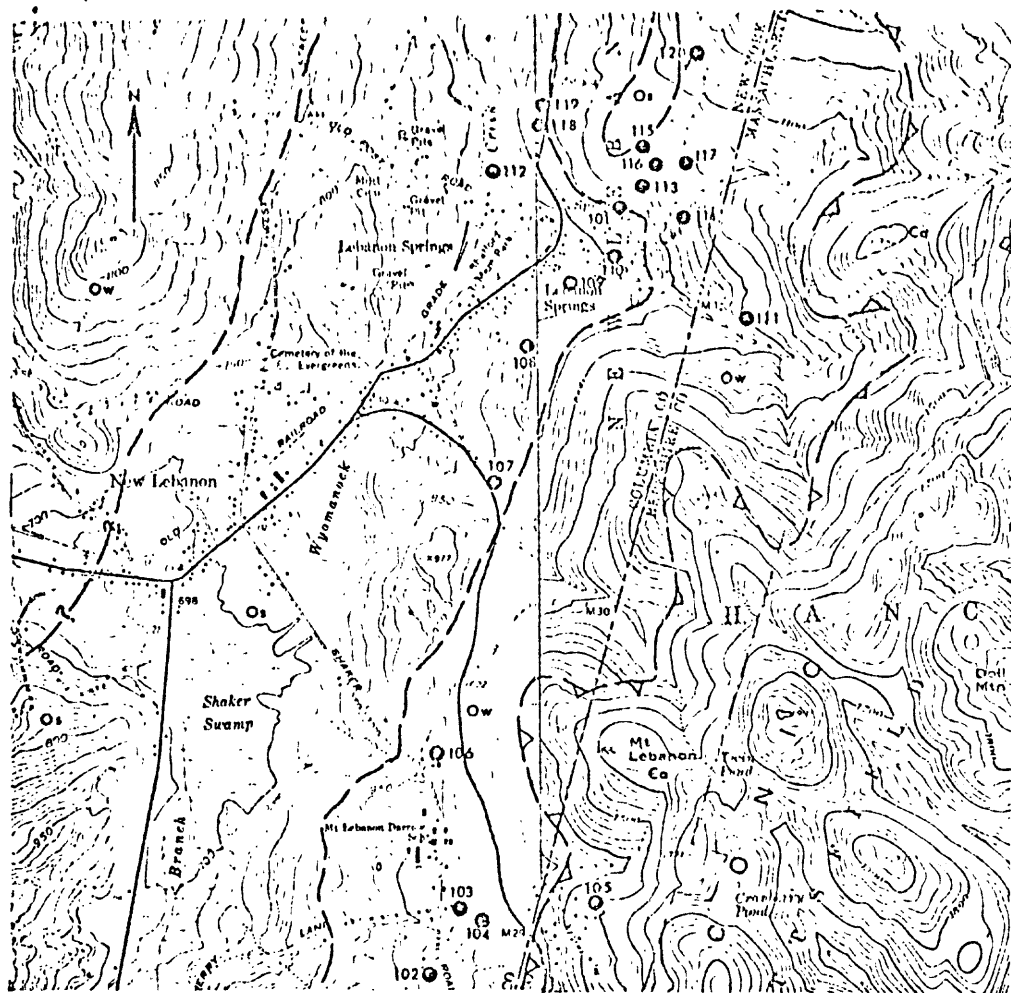
b Sample pumped from 215 foot depth.







Tr Trace amount present.

Table 2.--Results of dissolved gas and isotope analyses of waters from Appalachian warm springs areas--Continued

Sample number	Date	Temper- ature (°C)	Gas (concentrations in mg/l)						Isotope results				
			Field dissolved oxygen (O ₂)	Lab dissolved oxygen (O ₂)	Nitrogen (N ₂)	Argon (Ar)	Methane (CH ₄)	Carbon dioxide (CO ₂)	Helium (He)	Tritium			
										T.U.	±1σ		
VIRGINIA - FALLING SPRINGS AREA													
701	12-75	25.0	7.0	7.2	15	0.66	<0.005	33	<0.0003	24.4	±2.2	-6.8	-7.7
702	11-75	27.2	.1	(a)	(a)	(s)	Tr, <.03	140	.0020	10.9	±1.0	-8.0	-7.9
NORTH CAROLINA - HOT SPRINGS AREA													
801	12-75	39.3	3.2	3.5	15	.61	<.005	7.9	.0007	-.2	±2.1	-9.5	-6.3
803	12-75	15.2	1.4	<.02	25	.84	.06	38	<.0001	40.6	±2.4	-19.3	-6.2
804	12-75	12.5	10	10.5	22	.82	<.005	3.0	<.0001	52.2	±3.0	-12.0	-5.9
805	12-75	13.2	10	9.3	21	.79	<.005	2.7	<.0001	6.0	±1.0	-13.4	-6.2
GEORGIA - WARM SPRINGS AREA													
901	12-75	30.9	7.0	6.6	20	.81	<.005	11	<.0001	.8	±.9	-13.7	-4.5
905	12-75	17.7	8.2	7.9	14	.58	<.005	55	<.0002	34.6	±2.4	-30.1	-4.4
907	12-75	33.8	7.0	5.6	23	.58	<.01	14	<.0001	-1.0	±2.2	-16.4	-4.3
909	12-75	17.0	.4	<.02	25	.84	<.01	17	<.0002	15.1	±1.4	-20.0	-4.2

a Air contaminated sample.
Tr Trace amount present.



EXPLANATION	
	CAMERIAN ORDOVICIAN
Walloomsac Slate	
	
Stockbridge Limestone	
	
Austerlitz Phyllite	
	Fault
	Spring
	Well

(Geology modified after unpublished maps by D. W. Fisher, 1960, 1970. Modifications based on well driller's logs.)

Figure 2.--Map showing geology and spring and well locations in New York - Lebanon Springs area

Table 3.--Records of springs and wells in the Appalachian warm springs areas

Latitude-longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Casing depth (ft)	Well diameter (in)	Date drilled	Altitude (feet above mean sea level)	Water level (land surface, ft)	Date measured	Yield (gpm)	Specific conductance (microhmhos at 25°C/cm)	Temperature (°C)	Date sampled	pH	Topographic situation	Geology
NEW YORK-LEBANON SPRINGS AREA																	
4228500732211.1	*101	Lebanon Spring	P	--	--	--	--	910	--	--	105	260	21.9	10- 4-75	7.6	S	Stockbridge Limestone
4226460732254.1	102	D. McDonald		332	82	6	1946	r50	--	--	145	290	18.7	5-20-76	7.72	S	Walloomsac Slate
4226570732247.1	103	Darrow School	T	132	26	6	1962	1075	r16	10-	-62	15	13.5	9-18-75	7.3	S	Do.
4226550732243.1	104do.....	U	258	32	6	1965	1125	e15	--	--	10	--	--	--	S	Do.
4226570732218.1	105	Shaker Hts. Inc. Assoc.	P	400	--	6	--	1470	--	--	--	320	--	9-16-75	8.0	S	Austerlitz Phyllite ^b
4227220732252.1	106	Darrow School	T	338	20	6	1965	970	r15	8-25-65	6	--	--	--	--	S	Walloomsac Slate
4228050732239.1	107	Ed Blomgren	H	96	--	6	--	825	--	--	e<10	--	--	--	--	T	Stockbridge Limestone
4228270732232.1	108	Donald Kline	H	--	--	6	--	840	--	--	--	360	14.6	9-16-75	8.0	S	Do.
4228380732223.1	*109	John Koepf	H	111	70	6	1970	785	a+1.8	--	--	10+	260	9-16-75	7.6	V	Do.
4228420732212.1	110	E. Arcouet	H	137	38	6	1967	805	a+1.2	--	--	8	340	9-18-75	8.0	T	Do.
4228330732144.1	111	M. Ahlert	H	20	20	48	1700's	1000	e15	10- 3-75	--	--	--	--	--	S	Walloomsac Slate
4228560732238.1	112	F. Cummings	H	144	7	6	1967	775	0.0	9-16-75	16	235	11.1	9-16-75	8.0	T	Stockbridge Limestone (?)
4228530732206.1	113	S. Stouter	H	123	120	6	--	960	30.1	10- 4-75	--	--	--	--	--	S	Do.
4228480732157.1	*114	B. Schell	H	155	155	6	--	950	30.07	5-20-76	20+	--	--	9-18-75	7.7	Do.	Do.
4228480732157.1			H	155	155	6	--	950	73.7	9-17-75	6	825	14.2	9-17-75	7.2	S	Stockbridge Limestone-Limestone
4228590732206.1	115	K. Duffy	H	140	138	6	1961	1025	--	--	6	370	12.1	9-17-75	7.8	S	Walloomsac Slate
4228570732203.1	*116	F. Amlaw	H	184	151	6	1968	1005	78.0	9-17-75	10	380	13.2	9-17-75	7.5	S	Limestone
4228570732156.1	117	W. Weigan	U	--	--	36	--	1000	e5	9-18-75	--	--	--	--	--	S	Walloomsac Slate
4229030732229.1	118	G. W. Winquist	H	180?	--	6	--	880	<20	9-18-75	--	260	10.6	9-18-75	8.0	S	Stockbridge Limestone
4229070732228.1	119	R. Francher	H	272	--	6	--	885	0.0	9-18-75	good	260	--	9-18-75	8.0	S	Do.
4229150732154.1	120	F. Amlaw	S	--	--	--	--	1150	--	--	e5	110	12.2	9-17-75	5.8	S	Till (?)

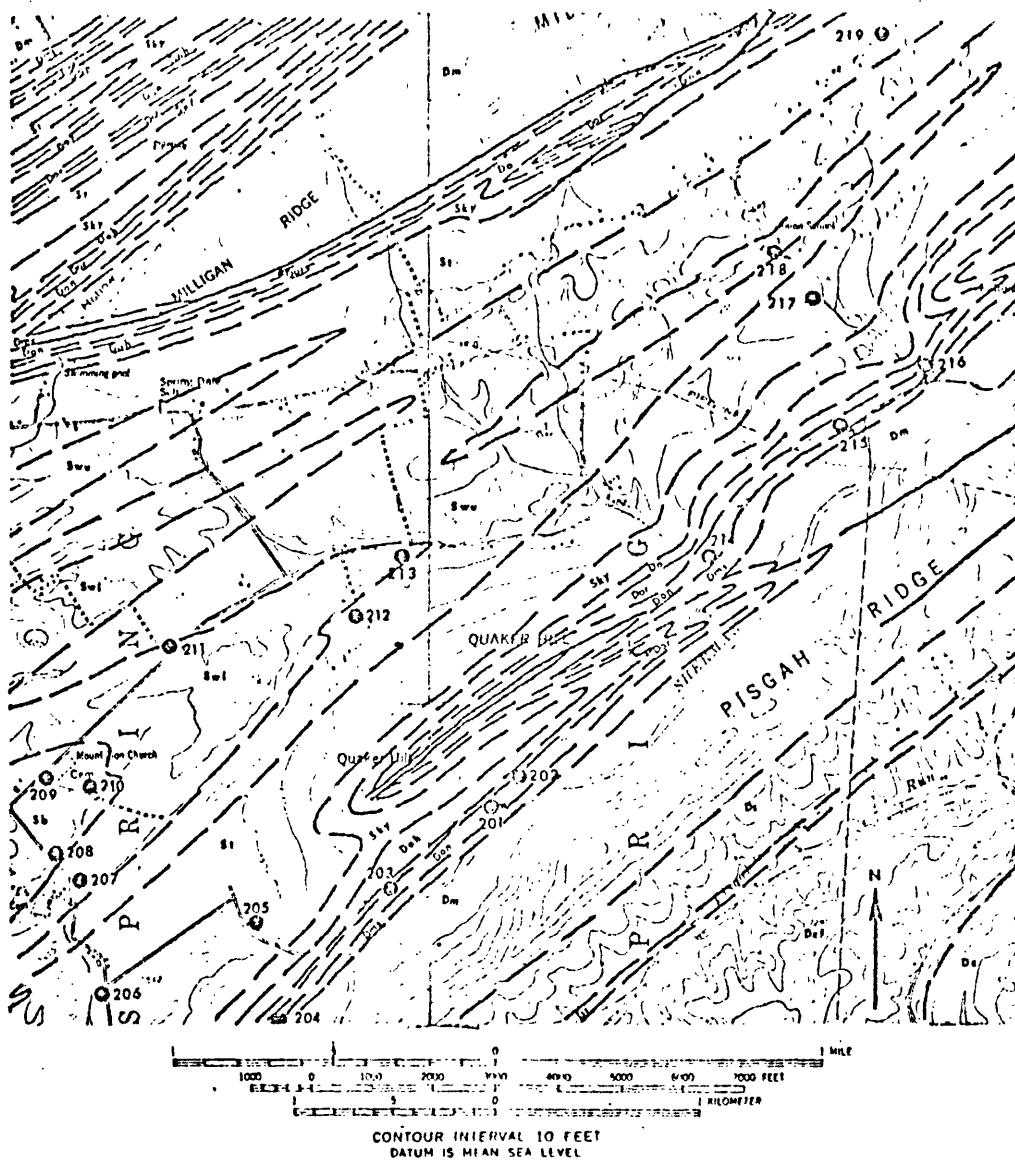
* Sampled for chemical analysis.

e Estimated.

r Reported.

a Above land surface.

b Nomenclature not adopted by U.S. Geological Survey.



EXPLANATION		
DEVONIAN	DEVONIAN	SILURIAN
Dc Catskill Formation	Dm Shamokin Member of Montebello Formation	Sky Keyser Formation
Def Catskill and Fort Littleton Formations, undivided	Don Onondaga Formation	St Tonoloway Formation
Dr Rush Formation	Dor Ridgeley Member of Oriskany Formation	Swu Upper Member of Wills Creek Formation
Ds Sherman Ridge Formation	Doh Oriskany and Helderberg Formations, undivided	Swl Lower Member of Wills Creek Formation
Dm Montebello Formation		Sb Bloomsburg Formation
Fault	Spring	Well

(Geology after Dyson, 1967, and J. T. Miller, 1961, in Johnston, 1970.)

Figure 3.--Map showing geology and spring and well locations in Pennsylvania - Perry County Warm Springs area

Table 4.--Records of springs and wells in the Appalachian warm springs areas

Latitude-longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Casing depth (ft)	Well diameter (in)	Date drilled	Altitude (feet above mean sea level)	Water level (land surface, ft)	Date measured	Yield (gpm)	Specific conductance (microhmhos at 25°C/cm)	Temperature (°C)	Date sampled	pH	Topographic situation	Geology
PENNSYLVANIA - PERRY COUNTY - WARM SPRINGS AREA																	
4019430771447.1	*201	Perry County warm springs	C	--	--	--	--	460	--	10- 6-75	140	190	17.8	10- 6-75	7.2	V	Oriskany Formation
4019480771441.1	*202	Lloyd Hetrick	H	123	123	6	1973	510	13.3	5-19-76	165	200	17.7	5-19-76	7.6	S	Do.
4019300771508.1	203	Max Kretzing	U	--	--	--	--	460	12.3	5-19-76	>10	125	12.0	10- 6-75	6.3	V	Do.
4019090771532.1	204do.....	U	83	0	6	1970	510	3.0	9-11-75	--	170	15.1	9-11-75	6.2	S	Do.
4019250771536.1	205do.....	H	63	?	5	1969	510	16.6	10- 7-75	--	535	13.4	9-10-75	7.5	S	Tonoloway Limestone
4019130771609.1	206	Harry Crozer	H	43	20	6	1962	515	16	4- -62	25	630	--	4- -62	--	V	Do.
4019320771614.1	207	Paul Crozer	H	75	29	6	1963	545	15	5- -63	25	--	--	--	--	S	Wills Creek Formation
4019360771620.1	208	Banks Scheibly	U	400	40	15	--	550	a+	3- -65	--	--	--	--	--	S	Bloomsbury Formation
4019490771621.1	209	Mt. Zion Church	T	80	31	6	1956	590	37	10- -56	15	180	--	10- -56	--	H	Do.
4019470771612.1	210	Banks Scheibly	H	61	25	6	1957	560	12	9- -57	8	--	--	--	--	S	Do.
4020100771555.1	211	C. A. Shope	H	70	53	6	1957	595	40	7- -57	10	--	--	--	--	S	Wills Creek Formation
4020140771515.1	212	Luther Snyder	H	96	74	6	1956	575	50	5- -56	10	370	--	5- -56	--	S	Do.
4020240771505.1	213	Bertram Shuey	H	104	?	6	1960	540	e25	9-10-75	--	425	12.6	9-10-75	8.0	S	Do.
4020240771401.1	214	W. A. Miller	H	93	50	6	1973	435	e12	9- -75	20	<50	13.2	9-11-75	5.7	T	Oriskany Formation
4020450771334.1	215	Spring	U	--	--	--	--	440	--	--	e25	--	13.4	9-10-75	--	V	Do.
4020540771316.1	*216	H. Stambaugh Spring	H	--	--	--	--	460	--	--	?	120	13.3	9-10-75	6.1	V	Do.
4021050771339.1	217	Morris Loy	H	63	30+	6	1963	505	r10	--	>40	440	14.2	9-10-75	7.3	S	Tonoloway Limestone
4021130771347.1	*218do.....	U	--	--	--	--	525	--	10- 7-75	7.5	300	12.0	10- 7-75	--	S	Do.
4021490771324.1	219	Robert Miller	H	235	120	6	1975	645	61.3	9-11-75	12	--	--	--	--	S	Do.

*Sampled for chemical analysis.

e Estimated.

r Reported.

a Above land surface.

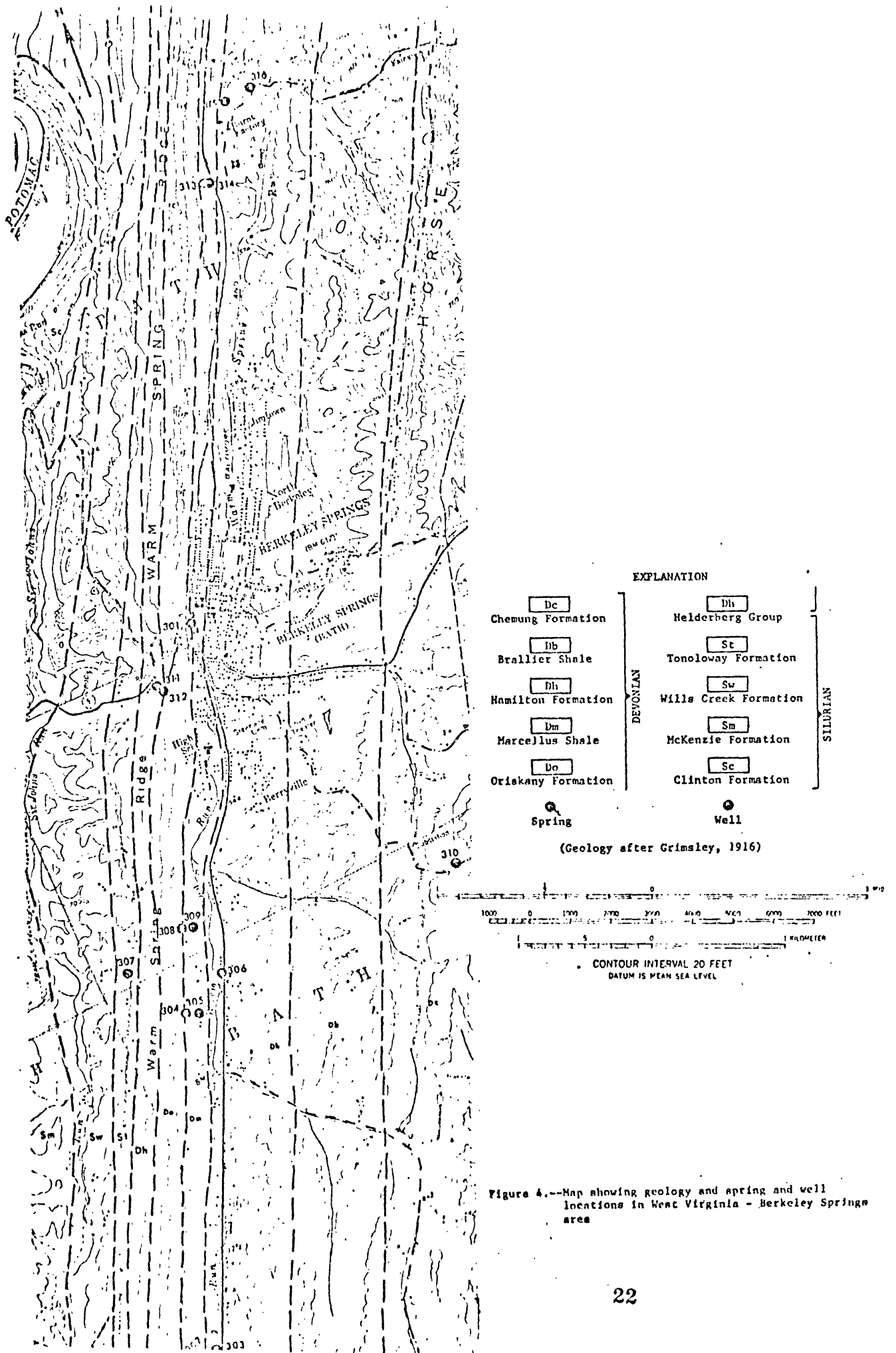


Figure 4.--Map showing geology and spring and well locations in West Virginia - Berkeley Springs area

Table 5.--Records of springs and wells in the Appalachian warm springs areas

Latitude-longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Casing depth (ft)	Well diameter (in)	Date drilled	Altitude (feet above mean sea level)	Water level (land surface, ft)	Date measured	Yield (gpm)	Specific conductance (micro-mhos at 25°C/cm)	Temperature (°C)	Date sampled	pH	Topographic situation	Geology
WEST VIRGINIA -- BERKELEY SPRINGS AREA																	
3937370781347.1	*301	Berkeley Springs	P	--	--	--	--	620	--	10-2-75	1635	235?	21.8	10-2-75	6.3	V	Oriskany Formation
3934520781518.1	*302	Ford Wachter	U	398	62	6	1974	785	165	5-18-76	1715	250	22.2	5-18-76	7.0	S	Marcellus Shale and Oriskany Formation
3934510781513.1	*303do.....	H	198	136	7	1969	760	21.4	9-30-75	10	480	13.9	9-30-75	6.5	T	Marcellus Shale and Oriskany Formation
3936090781438.1	304	WV Dept. of Highways	U	?	--	--	1974	705	dry	--	--	--	--	--	--	--	Oriskany Formation (?)
3936070781434.1	305do.....	H	550	43	6	1974	695	25.1	8-20-75	8	--	--	--	--	--	Marcellus Shale and Hamilton Formation
3936140781423.1	306	Bohler & McKoy	H	148	41	7	1969	670	9	8-22-69	15	340	--	8-22-69	7.0	T	Marcellus Shale and Hamilton Formation
3936230781449.1	*307	H. Hovermale	H	173	36	6	1967	840	48.4	10-1-75	4	500?	16.6	10-1-75	6.2	S	Tonoloway Formation
3936280781428.1	308	Kirkpatrick & Brown	U	150	--	6	1968	705	dry	--	--	--	--	--	--	S	Oriskany Formation
3636270781425.1	309do.....	I	148	--	6	1968	680	.7	8-20-75	60	--	--	--	--	S	Marcellus Shale
3936150781302.1	310	Fred Unger	H	335	41	7	1969	950	e60	8-22-69	--	--	--	--	--	S	Chemung Formation
3937240781404.1	311	R. Glass	H	147	20	6	1960	825	0	10-23-70	--	655	--	10-23-70	7+	H	Holderberg Group
3937220781404.1	312do.....	U	180	20	10	1954	835	r140	-54	>30	--	--	--	--	H	Do.
3939110781244.1	313	Pa. Glass & Sand Co.	U	450	30	8	1969	660	115	3- -69	50	400	13.7	8-20-75	7.2	S	Marcellus Shale and Oriskany Formation
3939110781245.1	314do.....	N	465	30	8	1969	660	115	5- -69	e50	--	--	--	--	S	Do.
3939280781220.1	315do.....	N	360	--	6	1945	670	60-70	--	100	--	--	--	--	S	Oriskany Formation (?)
3939270781230.1	316do.....	H	170	19	9	1957	600	--	--	12	--	--	--	--	S	Marcellus Shale (?)

*Sampled for chemical analysis.

e Estimated.

r Reported.

b Nomenclature not adopted by U.S. Geological Survey.

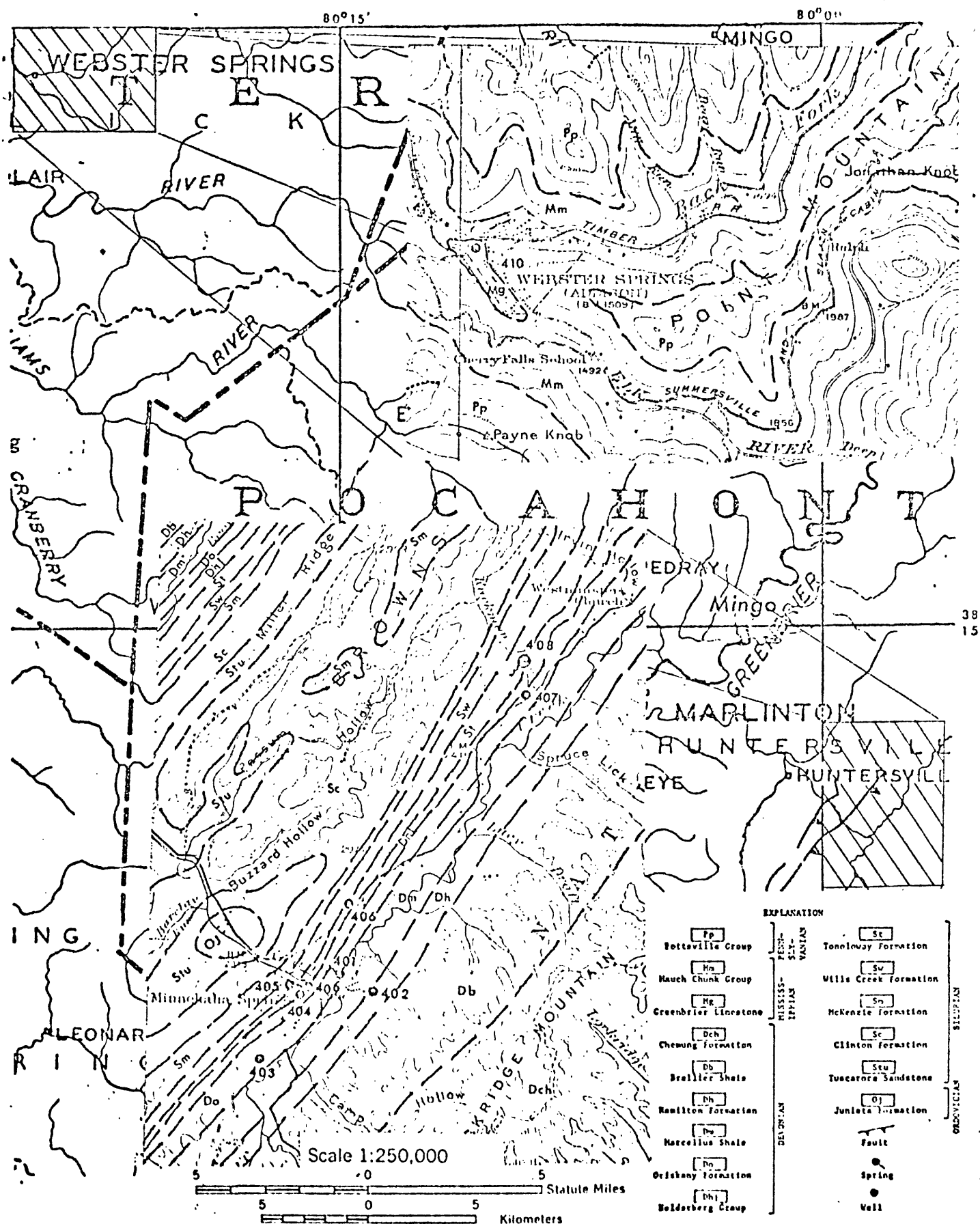


Figure 5.--Map showing geology and spring and well locations in West Virginia - Minnehaha Springs area (Geology after Price, 1929, and Reger, 1919)

Table 6.--Records of springs and wells in the Appalachian warm springs areas

Latitude-longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Casing depth (ft)	Well diameter (in)	Date drilled	Altitude (feet above mean sea level)	Water level (land surface, ft)	Date measured	Yield (gpm)	Specific conductance (micro-mhos at 25°C/cm)	Temperature (°C)	Date sampled	pH	Topographic situation	Geology
WEST VIRGINIA - MINNEHAHA SPRINGS AREA																	
3809500795829.1	*401	Minnehaha spring	R	--	--	--	--	2310	--	11-18-75	220	240	20.4	11-18-75	7.8	V	Oriskany Formation
3809450795814.1	*402	Camp Minnehaha	H	300	60	6	1971	2440	43.7	11-19-75	3	285	20.3	6-9-76	7.9	S	Hamilton (?) Formation
3809180795907.1	*403	A. T. White	H	76	76	6	1953	2335	p52.9	6-11-76	--	355	13.2	6-11-76	7.3	S	Oriskany Formation
3809430795848.1	404	J. Buzzard	H	46	27	6	1950	2355	r38	--	4.5	140	13.8	10-23-75	5.8	S	Do.
3809460795854.1	405	M. Buzzard	H	84	84	6	1958	2335	--	--	>20	80+	12.5	10-23-75	4.9	S	Do. (?)
3810170795826.1	*406	Floyd Davis	H	--	--	--	--	2320	--	--	e20	180	10.4	10-23-75	7.6	T	Tonoloway Formation
3811370795704.1	*407	R. Shinaberry	H	82	78	6	--	2365	37.9	10-22-75	9	110	13.5	10-22-75	5.5	T	Oriskany Formation
3811480795704.1	408	M. Workman	H	--	--	--	--	2435	--	--	e500	225	10.4	10-23-75	7.4	S	Tonoloway Formation
3809460795845.1	409	J. B. Worth	H	78	35	6	1976	2320	r2	--	6	--	--	--	--	--	Oriskany Formation
3828460802450.1	410	T. Rose	U	71	20?	6	1912	1460	2.5	6-8-76	--	6100	13.8	6-8-76	7.7	V	Greenbrier Limestone

*Sampled for chemical analysis.

e Estimated.

r Reported.

p Well pumped recently.

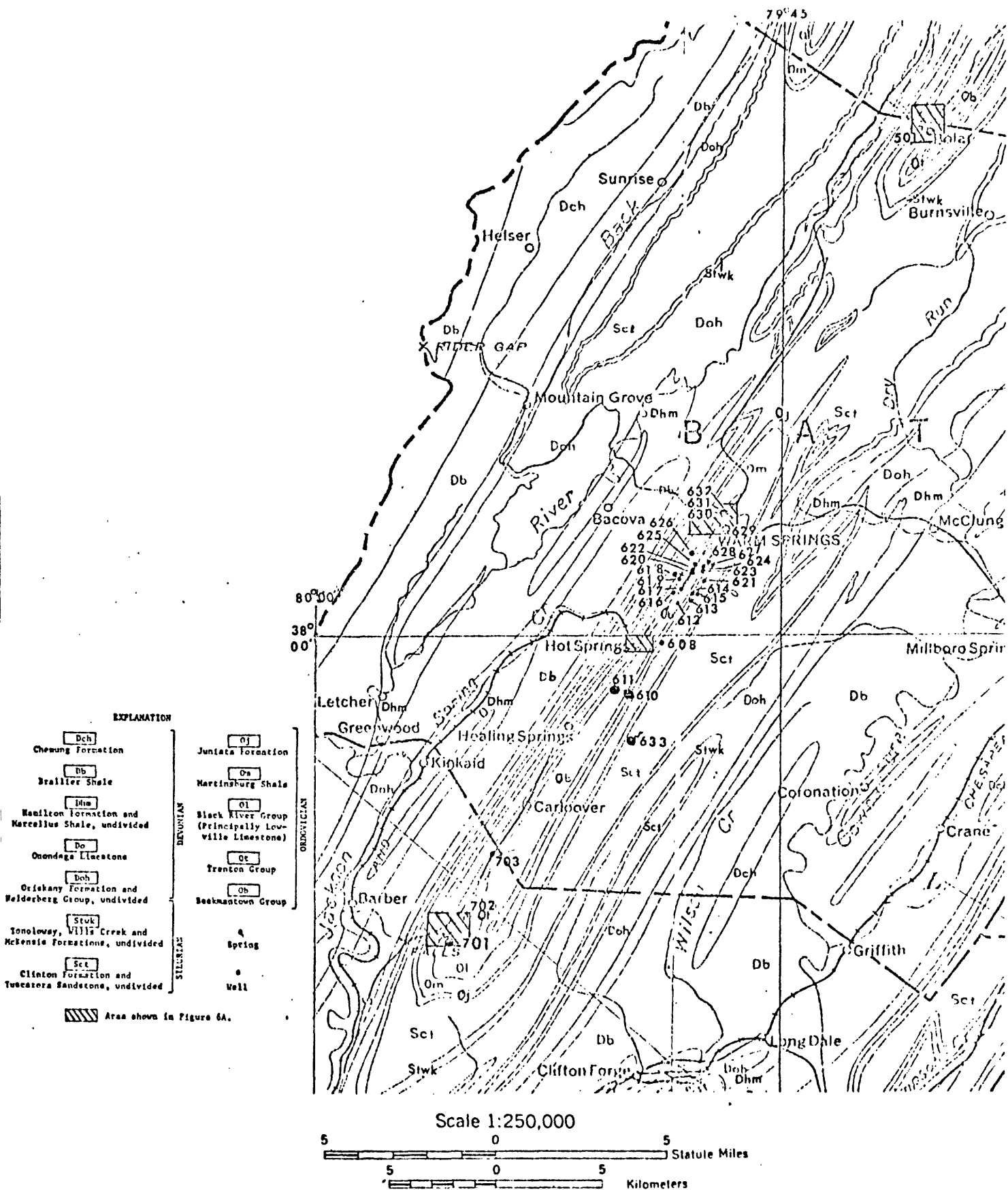


Figure 6.--Map showing geology and spring and well locations in Virginia - Bolar Springs, Hot Springs-Warm Springs, Falling Springs areas (Geology after Butts, 1933)

Table 7.--Records of springs and wells in the Appalachian warm springs areas

Latitude- longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Cas- ing diam- eter (in)	Date drill- ed	Altitude (feet above mean sea level)	Water level (land sur- face, ft)	Date measured	Yield (gpm)	Specific conduct- ance (micro- mhos at 25°C/cm)	Tem- per- ature (°C)	Date sampled	pH	Topo- graph- ic situa- tion	Geology
VIRGINIA - BOLAR SPRINGS AREA																
3812580794026.1	*501	Bolar Spring	R	--	--	--	2235	--	12-16-75 6-9-76	2065 2600	325 335	22.2 20.2	12-16-75 6-9-76	7.5 7.4	C	Lowville Limestone
VIRGINIA - HOT SPRINGS - WARM SPRINGS AREA																
3759540794950.1	601	Octagon spring	R	--	--	--	--	--	--	--	--	36.5	11-21-75	--	V	Do.
3759520794948.1	*602	Boiler spring	R	--	--	--	--	--	11-21-75 6-10-76	60 60	700 645	39.9 40.0	11-21-75 6-10-76	6.6 6.5	V	Do.
3759520794947.1	603	Sulphur spring	--	--	--	--	--	--	--	--	--	36.5	11-21-75	--	V	Do.
3759500794941.1	604	Magnesia spring	--	--	--	--	--	--	--	--	--	35.2	6-10-76	--	V	Do.
3759490794940.1	*605	Cold Magnesia spring	--	--	--	--	--	--	--	--	--	28.4	11-21-75	--	V	Do.
3759470794954.1	606	Soda spring	--	--	--	--	--	--	11-26-75 6-10-76	135 320	330 265	15.2 14.0	11-21-75 6-10-76	7.2 7.2	V	Do.
3759560794921.1	*607	Homestead hot well	U	756	150	6	1974	2440	9-4-75	220	420	19.6	11-20-75	7.2	V	Do. b Beekmantown Group
3759490794904.1	608	Homestead well	U	341	101	6	1964	2650	-64	15	--	--	--	--	S	Trenton and Black River Groups
3759380794922.1	609do.....	--	200?	--	--	2560	dry	--	--	--	--	--	--	S	Do.
3758430794952.1	610do.....	--	808	3-7	5	1963	2840	8-75	--	--	--	--	--	S	Do. b
3758470795022.1	611	Erwin Solomon	U	300?	--	6	--	2620	12-17-75	--	--	--	--	--	S	Beekmantown Group
3800460794841.1	612	Western Auto Store	C	235	205	6	--	2450	100+?	9	640	17.2	9-4-75	7.0	S	Do.
3800490794818.1	613	R. Robertson	H	412	62	4	--	2610	r312	3	--	--	--	--	S	Trenton and Black River Groups
3800540794819.1	614	Herbert Kriser	U	350	70	6	1970	2580	-70	--	--	--	--	--	S	Do.
3800530794800.1	615	Unknown	--	212	116	--	--	2820	10-11-73	20	--	--	--	--	S	Do.
3801020794848.1	616	J. Woodzell	H	100+	--	6	1957	2360	9-4-75	5	550	--	9-5-75	--	S	Do. b
3801020794837.1	617	R. Pampillonla	H	330	--	4	--	2370	--	12	--	--	--	--	S	Beekmantown Group
3801290794850.1	618	T. K. Ellis	--	228	43	5	1962	2225	r15	42	--	--	--	--	S	Martinsburg Shale, Trenton, and Black River Groups
3801150794843.1	*619	M. Dunn	H	--	--	--	2290	--	11-24-75	1025	410	15.8	11-24-75	7.3	C	Trenton and Black River Groups
3801350794815.1	620	T. K. Ellis	--	333	85	5	--	2400	r120	.5	--	--	6-10-76	7.4	S	Beekmantown Group
3801260794751.1	621	Ramsay ?	--	?	--	--	2640	r dry	--	--	--	--	--	--	S	Trenton and Black River Groups
3801410794816.1	622	T. K. Ellis	--	400	50	6	--	2340	r42	12	--	--	--	--	S	Beekmantown Group

* Sampled for chemical analysis.

e Estimated.

r Reported.

b Nomenclature not adopted by U.S. Geological Survey.

Table 7.--Records of springs and wells in the Appalachian warm springs areas--Continued

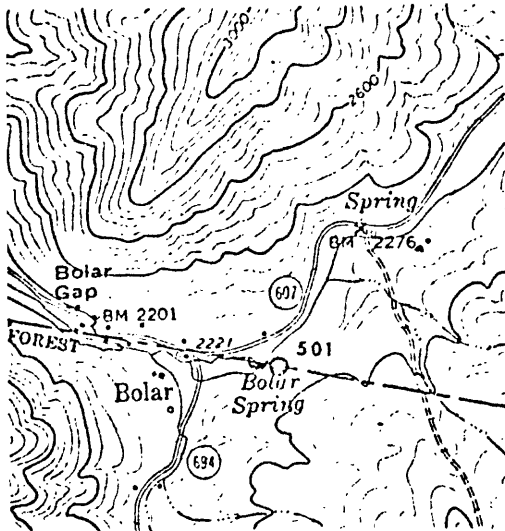
Latitude- longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Cas- ing diam- eter (ft)	Well diam- eter (in)	Date drill- ed	Altitude (feet above mean sea level)	Water level (land sur- face, ft)	Date measured	Yield (gpm)	Specific conduct- ance (micro- mhos at 25°C/cm)	Tem- per- ature (°C)	Date sampled	pH	Topo- graph- ic situa- tion	Geology
VIRGINIA - HOT SPRINGS - WARM SPRINGS AREA--Continued																	
3801510794746.1	623	J. Oliver	-	350	--	--	--	2590	r dry	--	--	--	--	--	--	S	Trenton and Black Riverb Group
3801560794742.1	624do.....	-	500	--	--	--	2580	r dry	--	--	--	--	--	--	S	Beekmantown ^b Group
3802010794807.1	*625	Valley View Cottages	P	761	64	5	1954	2500	r65	--	8	350	14.8	9- 4-75	--	S	Martinsburg Shale, Trenton and Black Riverb Groups
3802090794806.1	626	Unknown	-	125	--	--	--	2630	--	--	--	--	--	--	--	S	Martinsburg Shale, Trenton and Black Riverb Groups
3801590794738.1	*627	T. H. Bonds	H	242	130+	6	--	2580	163.2	9- 5-75	5	455	13.0	9- 5-75	7.0	S	Martinsburg Shale Trenton and Black Riverb Groups
3802070794743.1	628	Bath County High School	T	240	--	8	--	2540	r180	-55	12	360	11.9	9- 4-75	7.3	S	Do.
3802420794659.1	629	G. Gardner	H	200+	--	5	--	2495	--	--	13	275	16.4	9- 4-75	7.4	S	Do.
3803130794652.1	630	Warm Spring (ladies' pool)	M	--	--	--	--	2325	--	11-25-75	185	600	34.9	9- 3-75	7.1	V	Lowville Limestone
3803130794652.2	*631	Warm Spring (children's pool)	M	--	--	--	--	2325	--	6-10-76	410	--	34.8	11-25-75	--	--	--
3803130794651.1	632	Warm Spring (mens' pool)	M	--	--	--	--	2325	--	11-25-75	180	640	35.3	9- 3-75	7.0	V	Do.
3757550794952.1	*633	Ingalls Airport "Spring"	C	--	--	--	--	3690	e7	12-16-75	15	50	35.4	11-20-75	7.3	--	--
										11-25-75	450	--	35.0	9- 3-75	7.2	V	Do.
										6-10-76	410	--	35.4	6-10-76	--	--	--
										12-16-75	15	50	9.1	12-16-75	4.5	H	Clinton Formation
VIRGINIA - FALLING SPRINGS AREA																	
3752090795600.1	*701	Falling Spring	U	--	--	--	--	2195	--	12-15-75	3240	800	25.0	12-15-75	7.3	C	Lowville Limestone ^b
3752580795539.1	*702	H. D. Webb	S	167	7	6	1973	2235	?	6-11-76	4640	530	20.8	6-11-76	7.3	V	Beekmantown Group
3753580795416.1	703	Va. Polytech.	U	1000	1000	2	1975	2451	125	1-15-76	--	--	--	--	--	S	Do.

* Sampled for chemical analysis.

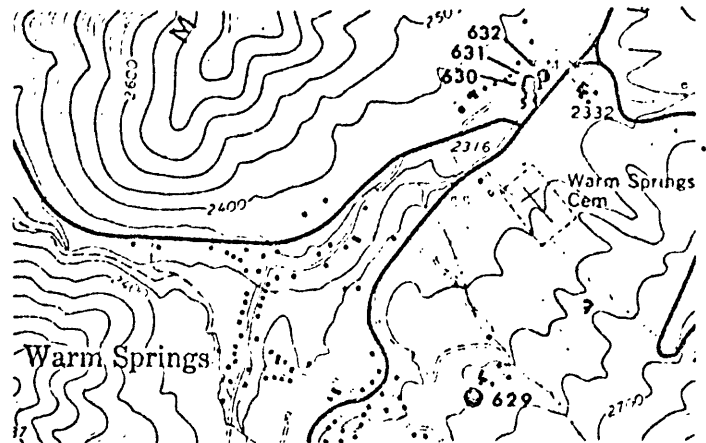
e Estimated.

r Reported.

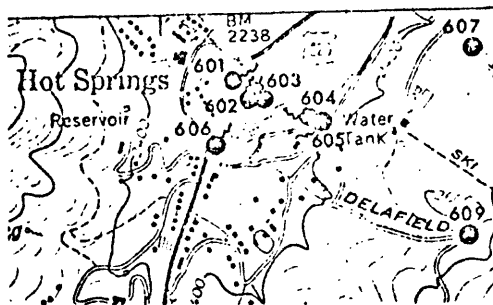
b Nomenclature not adopted by U.S. Geological Survey.



Bolar Spring Area



Warm Springs Area



Hot Springs Area



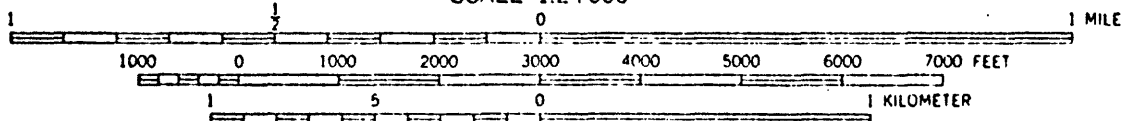
Falling Springs Area

EXPLANATION


 Spring

 Well

SCALE 1:24 000



CONTOUR INTERVAL 20 FEET

CONTOUR INTERVAL 40 FEET

DATUM IS MEAN SEA LEVEL

Figure 6A.--Detailed maps showing warm springs and well locations in Virginia.

Table 8.--Records of springs and wells in the Appalachian warm springs areas

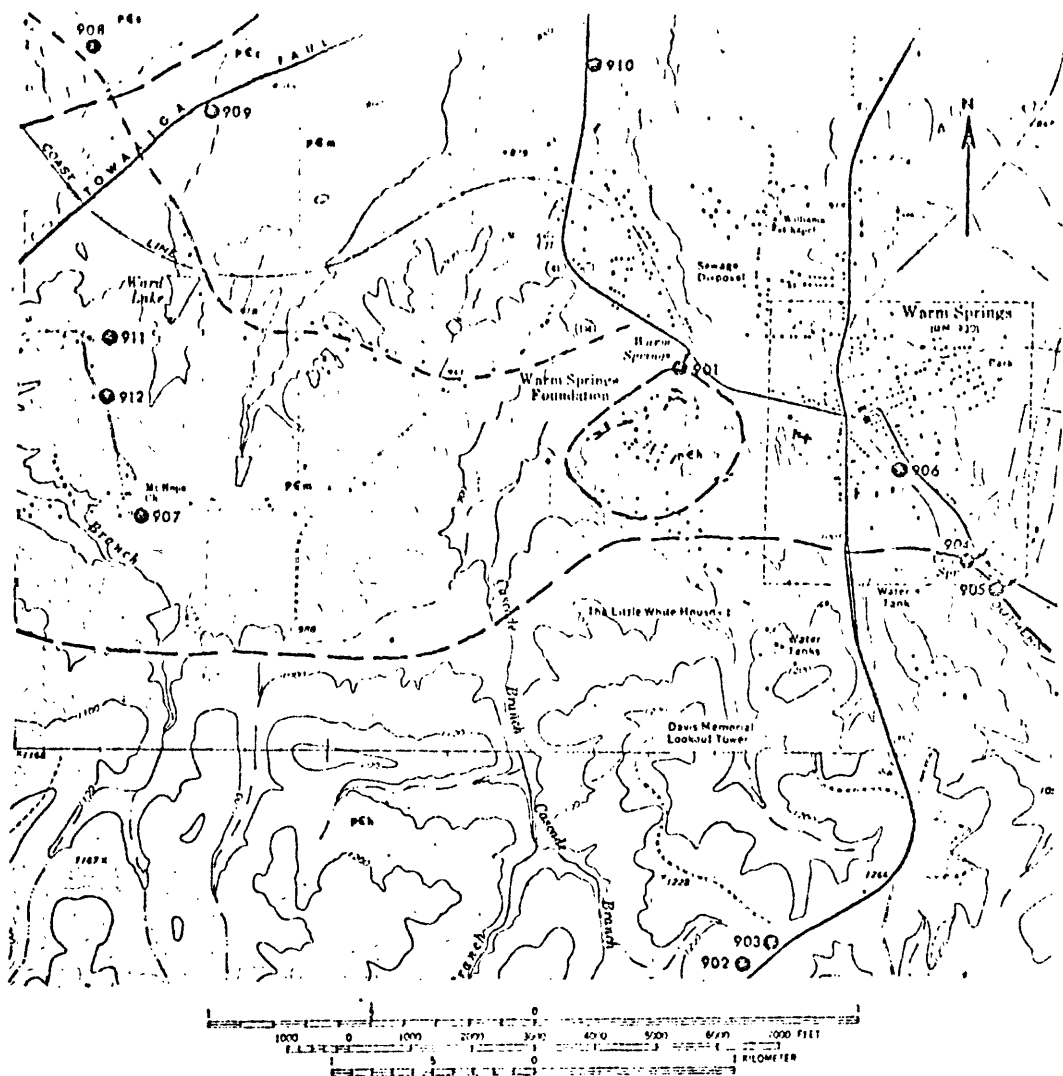
Latitude-longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Casing depth (ft)	Well diameter (in)	Date drilled	Altitude (feet above mean sea level)	Water level (land surface, ft)	Date measured	Yield (gpm)	Specific conductance (microhmhos at 25°C/cm)	Temperature (°C)	Date sampled	pH	Topographic situation	Geology
NORTH CAROLINA - HOT SPRINGS AREA																	
3553500824933.1	*801	Hot Spring	U	--	--	--	--	1315	8.85	12- 6-75	8.5	710	39.3	12- 6-75	7.5	V	Shady Dolomite
3553330824941.1	802	J. R. Henderson	U	--	--	--	--	--	8.86	5- 3-76	6.8	830	41.4	5- 3-76	7.5	T	Alluvium
3553170824948.1	*803	City Well	P	385	22	6	1966	1380	dry	11- 1-75	--	--	--	--	--	T	Murray Shale
3553370825143.1	*804	Fairview Water Assoc.	P	320	178	5	1972	1730	100	10-31-75	50	200	15.2	12- 6-75	6.7	T	Hesse Quartzzite
3554150825446.1	*805	Bubbling Springs	P	--	--	--	--	1315	--	12- 5-75	480	165	13.2	12- 5-75	7.75	V	Rome Formation
3553360825143.1	806	Kail Graham	H	140	--	6	--	1310	--	--	--	--	--	--	--	T	Great Smoky ^b Quartzite
3554300825031.1	807	W. Gorneflo	H	210	--	6	--	1350	--	--	--	--	--	--	--	T	Shady Dolomite
3554210825032.1	808	R. Williamson	H	160	30	6	1970	1320	r10	--	6	285	13.6	10-31-75	8.0	T	Rome Formation
3553590824933.1	809	W. Ramsey	U	200+?	--	6	--	1520	--	--	--	--	--	--	--	T	Shady Dolomite
3553570825146.1	*810	Paul Lovin	I	245	200	4	1966	1630	r200	--	12	245	15.0	5- 3-76	8.1	S	Do.
3554000825103.1	811	Burlin Shetley	U	238	196	6	1956	1610	r138	--	60	--	--	--	--	S	Do.

* Sampled for chemical analysis.

e Estimated.

r Reported.

b Nomenclature not adopted by U.S. Geological Survey.



CONTOUR INTERVAL 20 FEET
DATUM IS MEAN SEA LEVEL

EXPLANATION	
<p>pGm Manchester Schist</p>	<p>— Fault</p>
<p>pGh Hollis Quartzite</p>	<p>Q Spring</p>
<p>pGs Snelson Granite</p>	<p>O Well</p>
<p>pGc Carolina Gneiss</p>	

(Geology after Hewett and Crickmay, 1937)

Figure 8.--Map showing geology and spring and well locations
in Georgia - Warm Springs area

Table 9.--Records of springs and wells in the Appalachian warm springs areas

Latitude-longitude number	Spring or well number	Owner or name	Use	Well depth (ft)	Casing depth (ft)	Well diameter (in)	Date drilled	Altitude (feet above mean sea level)	Water level (land surface, ft)	Date measured	Yield (gpm)	Specific conductance (micro-mhos at 25°C/cm)	Temperature (°C)	Date sampled	pH	Topographic situation	Geology
GEORGIA - WARM SPRINGS AREA																	
3253330844126.1	*901	Warm Springs	M	--	--	--	--	845	--	12-11-75	885	182	30.9	12-10-75	7.40	V	Hollis Quartzite ^b
3251550844111.1	902	E. Neely	U	184	--	5	--	1225	65.7	5-5-76	h1085	210	30.5	5-5-76	6.95	H	Do.
3251590844106.1	903	Butts	U	66	--	48	1924	1225	69.5	5-5-76	--	--	--	--	--	H	Do.
3253010844029.1	904	Cold Spring	P	--	--	--	--	885	61.7	5-5-76	--	--	--	--	--	S	Do.
3252560844023.1	*905	South Spring	N	--	--	--	--	880	65.1	12-8-75	e800	<50	17.7	12-8-75	4.0	S	Do.
3253160844042.1	906	Ellerson	U	<20	--	--	--	920	13.9	5-5-76	245	<50	17.8	5-5-76	4.9	S	Manchester (?) Schist
3253080844308.1	*907	James Phillips	H	240	240	6	1965	945	81.2	12-13-34	--	100	33.8	12-10-75	6.42	S	Do.
3254240844316.1	908	B. Jewett Barnes	H	98	80	6	1955	890	r20	12-10-75	35	<50	17.6	12-10-75	--	H	Snelson Granite
3254130844258.1	*909	Andy Baxley	H	106	2	6	1975	840	6	--	--	158	17.0	12-11-75	7.02	S	Carolina (?) Gneiss
3254220844141.1	910	Chic Sparte Wear	N	<100	--	--	--	855	--	--	--	50	--	5-6-76	--	H	Manchester Schist
3253370844312.1	911	B. J. Harrington	H	45	--	--	--	920	--	--	--	80	--	5-6-76	--	S	Do.
3253270844313.1	912	W. L. Pritt	H	40	--	--	--	940	--	--	--	100	--	5-6-76	--	S	Do.

* Sampled for chemical analysis.

e Estimated.

r Reported.

h This measurement erroneously high because of stored water draining from reservoir.

b Nomenclature not adopted by U.S. Geological Survey.

Table 10.-- FACTORS FOR CONVERTING ENGLISH UNITS TO
INTERNATIONAL SYSTEM (SI) UNITS

The following factors may be used to convert the English units published herein to the International System of Units (SI).

Multiply English units	By	To obtain SI units
Length		
inches (in)	25.4	millimetres (mm)
	.0254	metres (m)
feet (ft)	.3048	metres (m)
miles (mi)	1.609	kilometres (km)
Area		
square miles (mi ²)	2.590	square kilometres (km ²)
Volume		
gallons (gal)	3.785	litres (l)
	3.785	cubic decimetres (dm ³)
	3.785x10 ⁻³	cubic metres (m ³)
million gallons (10 ⁶ gal)	3785	cubic metres (m ³)
	3.785x10 ⁻³	cubic hectometres (hm ³)
cubic feet (ft ³)	28.32	cubic decimetres (dm ³)
	.02832	cubic metres (m ³)
Flow		
cubic feet per second (ft ³ /s)	28.32	** litres per second (l/s)
	28.32	cubic decimetres per second (dm ³ /s)
	.02832	cubic metres per second (m ³ /s)
gallons per minute (gpm)	.06309	litres per second (l/s)
	.06309	cubic decimetres per second (dm ³ /s)
	6.309x10 ⁻⁵	cubic metres per second (m ³ /s)
million gallons per day (mgd)	43.81	cubic decimetres per second (dm ³ /s)
	.04381	cubic metres per second (m ³ /s)

** The unit litre is accepted for use with the International System (SI). See NBS Special Bulletin 330, p. 13, 1972 edition.