

UNITED STATES DEPARTMENT OF THE INTERIOR
GEOLOGICAL SURVEY

Age and Petrology of the Tertiary As Sarat Volcanic Field,
Southwestern Saudi Arabia

By

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Open-File Report 91- *148*

Report prepared by the U.S. Geological Survey in cooperation with the
Deputy Ministry for Mineral Resources, Saudi Arabia

This report is preliminary and has not been reviewed for
conformity with U.S. Geological Survey editorial standards
and stratigraphic nomenclature.

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ABSTRACT

Harrat As Sarat forms the second smallest and southernmost of the basalt fields of western Saudi Arabia and is part of a voluminous Red Sea rift-related continental alkali basalt province. The rocks of the As Sarat were emplaced during the first stage of Red Sea rifting and represent the northernmost extension of the Tertiary Trap Series volcanics that occur mainly in the Yemen Arab Republic and Ethiopia. The field consists of up to 580 m of basalt flows, that are intruded by basaltic plugs, necks, minor dikes, and highly evolved peralkaline trachyte intrusions. K-Ar ages indicate that the As Sarat field formed between 31 and 22 Ma and contains an eruption hiatus of one million years that began about 25 Ma. Pre-hiatus flows are primarily hypersthene normative intersertal subalkaline basalt, whereas the majority of post-hiatus flows are nepheline normative alkali basalt and hawaiite with trachytic textures. Normative compositions of the basalts are consistent with their genesis by partial melting at varying depths. Trace element abundances in the basalt indicate that varying degrees of partial melting and fractional crystallization (or crystal accumulation) had major and minor roles, respectively, in development of compositional variation in these rocks. Modeling indicates that the pre-hiatus subalkaline basalts represent 8.10 percent mantle melting at depths of about 70 km and the post-hiatus alkali basalts represent 4-9 percent mantle melting at depths greater than 70 km.

INTRODUCTION

Cenozoic alkali basalt covers about 90,000 km² in western Saudi Arabia and is part of one of the world's largest alkali basalt provinces (Coleman and others, 1983). These basalts are similar to alkali basalts that occur in the Yemen Arab Republic and across the Red Sea in Ethiopia (Chiesa and others, 1989; and Mohr and Zanettin, 1989); all of these rocks have origins related to Red Sea rifting. Within western Saudi Arabia thirteen alkali basalt fields (harrats) have been delineated (Coleman and others, 1977).

The general characteristics of the Arabian alkali basalts have been described in reconnaissance studies of the region (e.g. Coleman and others, 1983). Several systematic studies of individual Red Sea rift-related basalt fields have been recently completed (Pallister, 1987; Camp and Roobol, 1989). The timing of uplift related to Red Sea rifting is discussed by Bohannon and others (1989), whereas Coleman and

McGuire (1988) and McGuire and Bohannon (1989) describe magma systems and mantle dynamics related to rifting. Hegner and Pallister (1989) present Pb, Sr, and Nd isotopic data for rift volcanics of the central Saudi Arabian coastal plain and infer from these data that magmas represented by these rocks represent mixing of asthenospheric and lithospheric components. Various aspects of the geology of the As Sarat field have been described by Coleman and others (1977), Overstreet and others (1977), Anderson (1979), and Stoesser (1984). Kellogg and Reynolds (1983) conducted a paleomagnetic study of the As Sarat field and identified a reversal within the basalt flow sequence. Their work helps constrain the timing of Red Sea rifting.

Basalt of the As Sarat forms the second smallest (about 750 km²), and most southerly of the Saudi Arabian alkali basalt fields (fig. 1). Exposures extend about 105 km northwestward from a point near the border with the Yemen Arab Republic and are an extension of the Tertiary plateau volcanic fields or Trap Series of Yemen (Capaldi and others, 1983; Chiesa and others, 1989). The field forms a contiguous series of deeply eroded buttes and mesas. The deeply dissected character of the field is a consequence of its lofty position directly east of the Red Sea escarpment. Peaks within the field are as much as 2850 m above sea level and are some of the highest in Saudi Arabia. The field is located about 300 km east of the currently active Red Sea spreading axis and 100 km north of the Tihama Asir complex considered by Coleman and others (1983) to be incipient oceanic crust formed by magmatic activity that accompanied spreading along the Red Sea rift. K-Ar ages presented by Coleman and others (1975, 1983) indicate that the basal and top flows of the As Sarat field were erupted at 30.0 ± 1.0 Ma and 25.3 ± 0.5 Ma, respectively, that two feeder necks were emplaced 24.7 ± 2.0 Ma and 25.4 ± 2.7 Ma, and a trachyte dome at 22.4 ± 0.2 Ma.

The As Sarat field is predominantly underlain by alkali basalt flows that unconformably overlie lateritized rocks of the Late Proterozoic Arabian Shield or Cambro-Ordovician Wajid Sandstone (Overstreet and others, 1977). The field also includes a small number of basaltic necks and basaltic dikes, and five late intrusive trachyte domes that were emplaced at the base of the flows. In the more deeply eroded area north of the As Sarat, numerous Tertiary basalt plugs, as much as 30 m in diameter, occur in an area that extends northwestwards for 65 km in a direction parallel to the Red Sea (fig. 1). These plugs are correlated with the basalts of the As Sarat field and indicate that the areal extent of the field was formerly much greater than at present.

The majority of faults that cut the As Sarat basalts are normal faults that parallel the long axis of the field and the Red Sea margin. These faults are concentrated along the margins of the field and most are down towards the center of the field (Anderson, 1979; Stoesser, 1984). These observations suggest that either the As Sarat basalts were erupted into an active graben or preserved in a graben that post-dates the basalts.

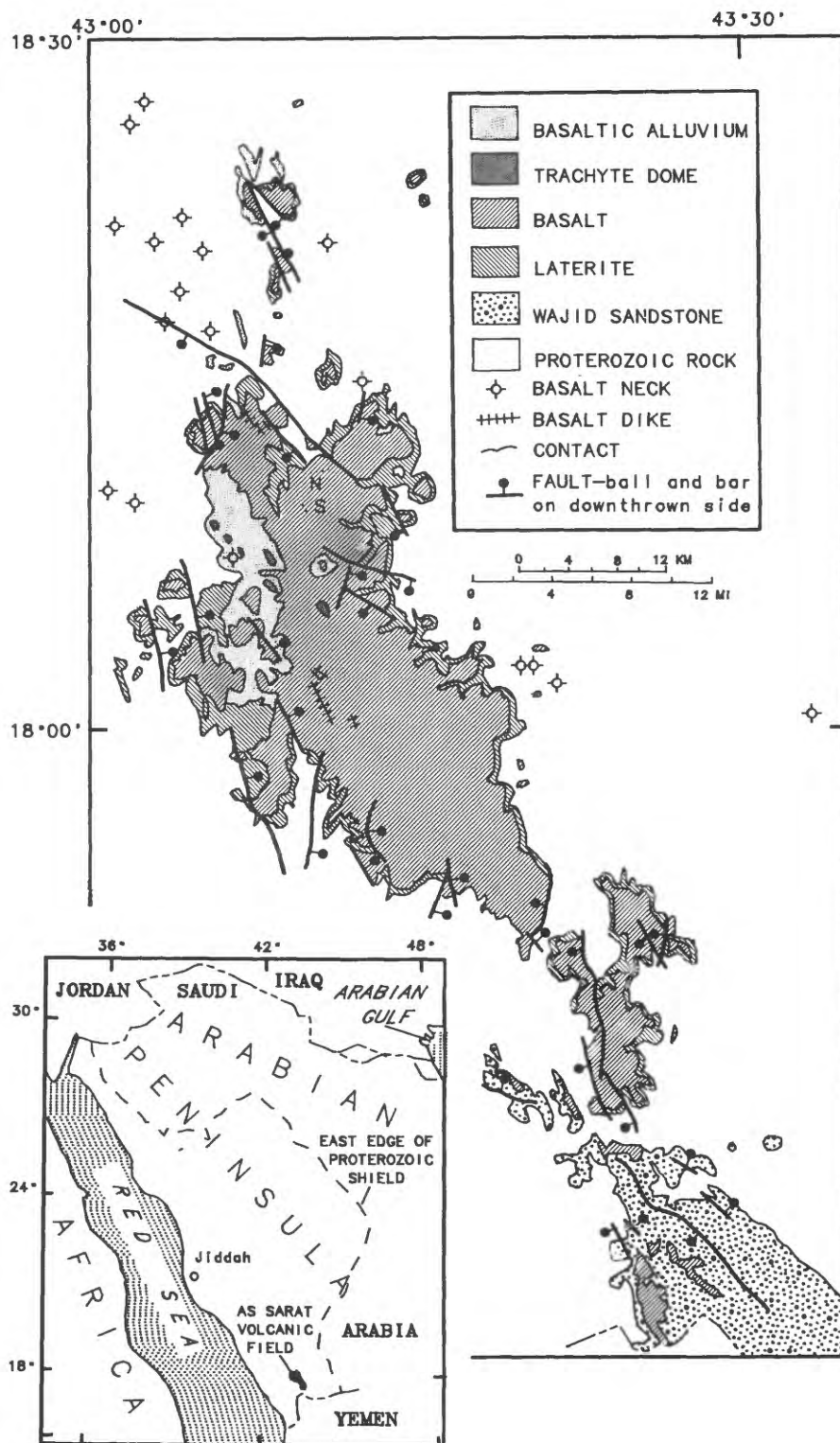


Figure 1.—Index and geologic sketch maps (after Anderson, 1979 and Stoesser, 1984) of the As Sarat basalt field, Kingdom of Saudi Arabia. Dotted lines denoted N and S show approximate locations of the northern and southern parts of the Composite Reference Section, respectively.

In order to study the flows of the As Sarat field, a two-part, composite reference section (CRS) in the northern part of the field was chosen for study (figs. 1, 2). The CRS was also included in the study of Kellogg and Reynolds (1983). The lower three-quarters of the CRS forms a ridge located 2.8 km north of a second ridge, where the upper one-quarter of the CRS is exposed. A distinctive thick flow at the top of the lower section and at the base of the upper section permits correlation between the two sections. The maximum total thickness of the CRS is about 580 m. Individual flows are from 3 to 20 meters thick and are at least 35 in number. Flows are mainly composed of massive, fine-grained, poorly to non-vesiculated basalt. Flows display columnar jointing from about their midline to their bases and some are characterized by subhorizontal jointing at their tops. The lower section includes several rather thick intervals composed of friable, scoriaceous basalt. No stratigraphic indicator is present to indicate the top of the basalt sequence and the uppermost flows of the As Sarat field may have been removed by erosion. The absence of well-developed soil horizons and of unconformities indicate that the CRS is an intact sequence, i.e. that no flows have been removed by between-eruption erosional events.

ACKNOWLEDGMENTS

The field work for this study was performed in accordance with an agreement between the Saudi Arabian Ministry of Petroleum and Mineral Resources and the U.S. Geological Survey (USGS). Reviews by J.S. Pallister, J.C. Ratte, and P. Mohr are much appreciated. Analysts from the Branch of Geochemistry (USGS) are thanked for their efforts.

ANALYTICAL METHODS

Major oxide and trace element compositions for 28 flows, 4 necks, 4 dikes, and 5 trachyte domes were determined in the Analytical Laboratories of the U.S. Geological Survey (USGS) in Denver, Colorado. Major oxide analyses were performed (analysts, J.E. Taggart, A.I. Bartel, K.C. Stewart, E.C. Robb, and D.F. Siems) using X-ray fluorescence techniques (Taggart and others, 1987), except FeO which was determined (analyst, E. Brandt) by wet chemistry (Jackson and others, 1987). $\text{Fe}^{2+}:\text{Fe}^{3+}$ ratios were adjusted to 9:1 (Ringwood, 1975) and major oxide compositions were recalculated to 100 percent on an anhydrous basis prior to norm calculation and plotting. Co, Cr, Cu, Ni, Sc, V, and Zn abundances were determined (analysts, P.H. Briggs, D.L. Fey, and L.A. Bradley) by inductively coupled plasma-atomic emission spectroscopy (Lichte and others, 1987). Rb, Sr, Y, Zr, Nb, and Ba abundances were determined (by E.A. du Bray) by energy-dispersive X-ray fluorescence spectroscopy (Elsass and du Bray, 1982) using ^{109}Cd and ^{241}Am radio-isotope excitation sources; these data are considered accurate within 5 percent of the reported abundances. X-ray Assay Laboratories, Ltd. (XRAL), Ontario,

AGE SCALE (M.Y.)

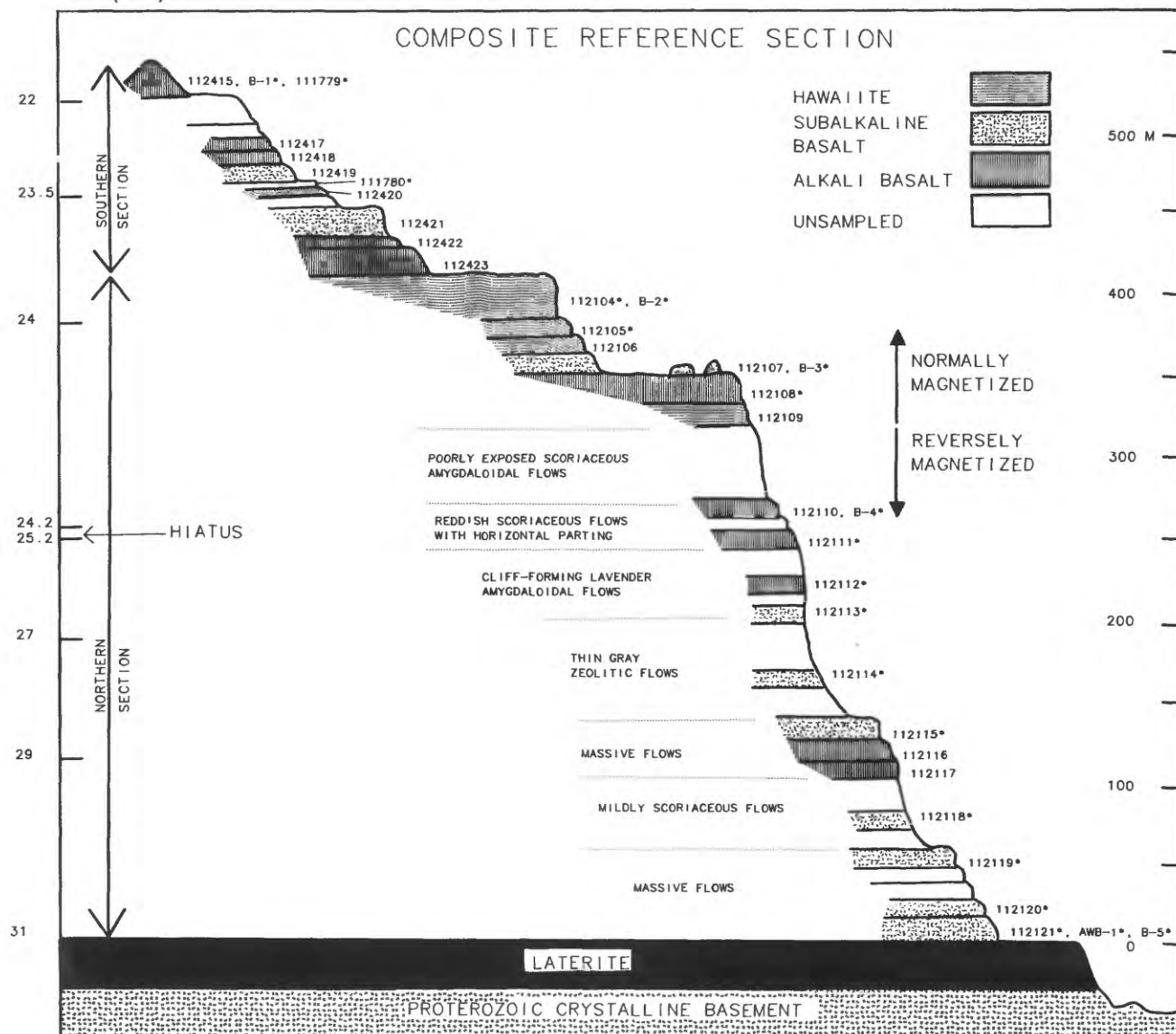


Figure 2.—Schematic cross section of the composite reference section (CRS). Asterisks following sample numbers indicate dated samples.

Canada determined rare-earth element (REE) and thorium abundances in samples from 7 flows, 1 neck, 1 dike, and 1 trachyte dome by neutron activation analysis and uranium by delayed neutron counting.

Compositions of feldspar, olivine, and pyroxene were determined using an ARL SEMQ electron microprobe equipped with 6 wavelength dispersive crystal spectrometers. A 5 micron (except feldspars, 25) beam and a 15 kV accelerating voltage with 10 nA sample current were used to make a minimum of three 20-s analyses on three or more grains per sample. Analyzed natural and synthetic minerals were used as standards; all data was reduced using MAGICIV (Colby, 1968).

Sample preparation (for age determinations) and argon and potassium analyses were conducted in USGS laboratories in Menlo Park, California, using techniques described in Dalrymple and Lanphere (1969). Whole rock samples were crushed, sieved to -60 + 100 mesh, washed, and treated for 30 min in 14 percent HNO₃ and 1 min in 5 percent HF solutions before loading into a high-vacuum gas-extraction system. Aliquots of the whole rock samples were analyzed for K₂O by a lithium metaborate flux fusion-flame photometry technique; lithium acts as an internal standard (Ingamells, 1970). Argon determinations were made in the USGS isotope geology labs using standard isotope-dilution techniques. Samples of clean argon gas were analyzed using a 600 sector, 15.2 cm-radius, Neir-type mass spectrometer, operated in the static mode. Ages were calculated using the constants of the subcommission on geochronology (Steiger and Jager, 1977).

K-AR GEOCHRONOLOGY

Twenty-two new age determinations on rocks of the As Sarat field are reported (table 1). The dated samples are from flows of the As Sarat volcanic field, with the exception of one age determination for a sample of a trachyte dome. Samples were selected at relatively evenly spaced intervals from the bottom of the CRS to its top in order to determine the duration and frequency of eruptive activity. Basalt flows of the As Sarat volcanic field range in age from about 31 m.y. to about 22 m.y. and the trachyte has an age of about 24 m.y.

Age determinations for two of the samples (nos. 112114 and 112119) have such large analytical uncertainty that they are of no value in age evaluation and three other samples (nos. 112105, 112118, and AWB-1) yield ages that are unreasonable when evaluated with respect to radiometric ages for stratigraphically adjacent flows. These samples appear to be weakly zeolitized, which has affected the isotopic systematics, and thus these data were excluded from our study.

Consideration of stratigraphic relations and age precisions calculated for individual samples allows re-evaluation of age accuracies. The K-Ar determinations

are arranged in stratigraphic succession and shown as bars, the lengths of which represent the limits of analytical uncertainty (precision) at one standard deviation (fig. 3). Because the relative ages of all flows are known from stratigraphy, each radiometric age can be compared with ages for overlying and underlying flows (McKee and Elston, 1980). The patterned areas in figure 3 show estimated accuracy age ranges (table 1) ascertained by this method for each of these samples. In addition to reducing age uncertainties for most flow units in the section, the data (fig. 3) indicate a one million year hiatus, between 25.2 and 24.2 Ma, and on that basis we refer herein to pre- and post-hiatus flows.

An approximate age scale (fig. 2) for the CRS was constructed using estimated age values. The As Sarat volcanics in the northern part of the field were erupted during a period of approximately 9 m.y. between 31 and 22 Ma. These data are comparable to K-Ar ages of 30 to 22 Ma for the equivalent Trap Series basalts directly to the south in the northern Yemen Arab Republic (Capaldi and others, 1983). In the CRS below the hiatus, the eruption rate decreases upward. Above the hiatus the eruption rate increases dramatically and most of the upper flows were erupted in a relatively short period of time between 24.2 and 23.5 Ma. The uppermost flow was erupted about 22.2 Ma.

Table 1.--Potassium-argon ages for volcanic rocks from the As Sarat volcanic field, Kingdom of Saudi Arabia.

Field Number	Latitude N.	Longitude E.	Rock type	K ₂ O wt. %	⁴⁰ Ar 10 ⁻¹¹ mol/g	⁴⁰ Ar ^{rad} %	Apparent age, m.y. ±1 sigma	Estimated Age Accuracy (m.y.)
111779	18°09'27"	43°09'58"	alkali basalt	1.321	4.1656	25.4	21.8 ±1.3	20.5-23.1
B-1	do.	do.	do.	1.315	4.30824	12.9	22.6 ±2.3	20.5-23.1
111780	18°09'34"	43°09'55"	do.	2.07	7.18045	46.7	23.9 ±0.7	23.2-24.2
B-2	18°10'56"	43°10'16"	hawaiite	1.504	5.37982	50.7	24.7 ±0.7	24.0-24.2
112104	do.	do.	do.	1.59	5.70614	70.6	24.8 ±0.4	-----
112105	do.	43°10'17"	do.	1.443	5.81538	69.8	23.0 ±0.4	-----
B-3	do.	43°10'20"	do.	1.576	5.04454	14.5	22.1 ±2.1	24.0-24.2
112108	do.	43°10'25"	alkali basalt	0.885	3.30741	20.2	25.8 ±1.8	24.0-24.2
B-4	18°10'55"	43°10'40"	do.	1.061	3.62366	51.8	23.6 ±0.6	24.0-24.2
112111	do.	43°10'41"	do.	0.55	2.08867	24.0	26.2 ±1.6	25.2-27.7
112112	18°10'56"	43°10'42"	do.	0.396	1.45544	11.5	25.4 ±2.9	25.2-27.7
112113	do.	43°10'45"	basalt	0.647	2.74638	27.8	29.3 ±1.6	27.7-30.1
112114	18°10'57"	43°10'50"	do.	0.39	0.882349	1.8	15.6 ±6.8	-----
112115	do.	43°10'51"	do.	0.28	1.11146	13.3	27.4 ±2.8	27.7-30.1
112118	18°10'58"	43°10'52"	do.	0.253	1.37174	27.9	37.3 ±2.0	-----
112119	do.	do.	do.	0.268	0.686926	3.9	17.7 ±4.5	-----
112120	do.	43°10'53"	do.	0.234	1.01168	58.1	29.8 ±0.7	29.1-30.1
112120	do.	do.	do.	0.234	0.968654	24.3	28.5 ±1.7	29.1-30.1
112121	do.	43°10'54"	do.	0.368	1.56358	51.3	29.3 ±0.8	29.1-30.1
B-5	18°10'43"	43°12'33"	do.	0.298	1.3466	21.2	31.1 ±2.1	29.1-33.2
AWB-1	18°09'25"	43°06'52"	do.	0.378	1.50521	31.0	27.5 ±1.3	-----
AW-2	18°09'25"	43°04'41"	trachyte	4.00	13.8664	73.5	23.9 ±0.4	23.5-24.3

Note: Uncertainties assigned to ages were determined using the curve of Tabor and others (1985). K-Ar ages were calculated using the constants for radioactive decay and abundance of ⁴⁰K recommended by the International Union of Geological Sciences Subcommittee on Geochronology (Steiger and Jager, 1977). These constants are: $\lambda_{\beta} = 0.581 \times 10^{-10}/\text{yr}$, $\lambda_{\text{beta}} = 4.962 \times 10^{-10}/\text{yr}$, and $^{40}\text{K}/\text{K}_{\text{total}} = 1.167 \times 10^{-4}$ mol/mol.

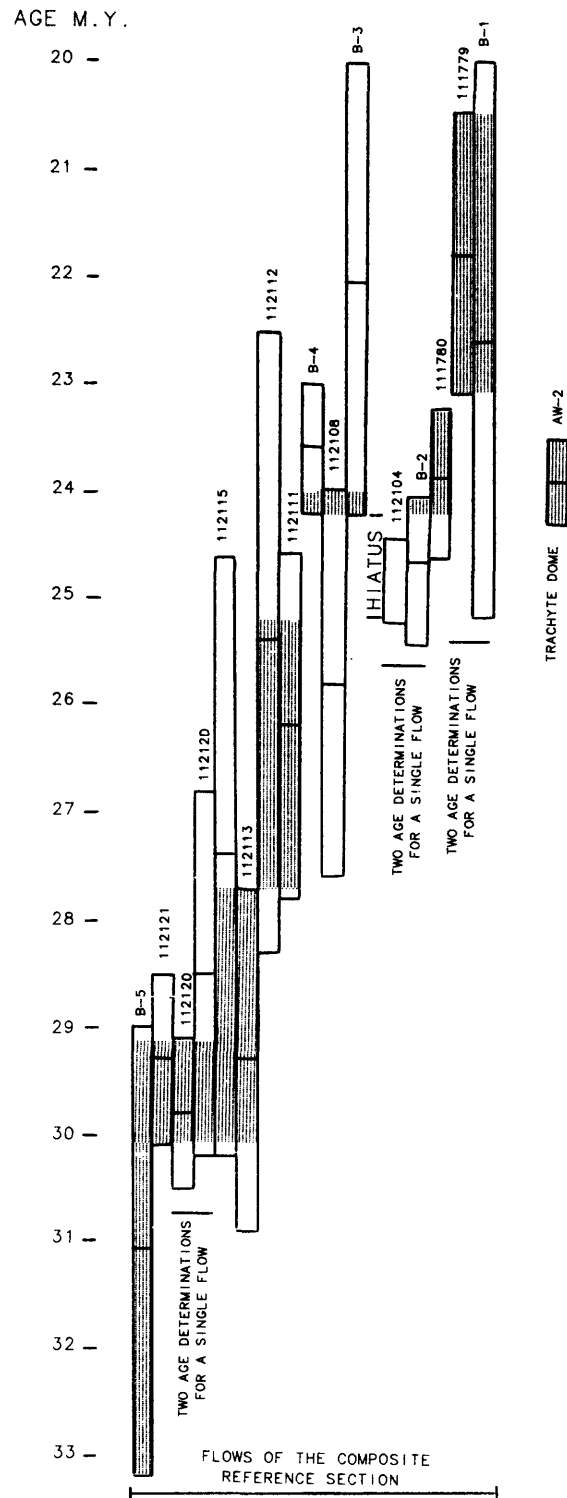


Figure 3.—Age evaluation diagram. Ages bars are plotted in order of decreasing stratigraphic age from left to right. Length of bar represents range of analytical uncertainty (one sigma); shaded area shows maximum uncertainty of the ages as indicated by stratigraphic relations.

PETROGRAPHIC CHARACTERISTICS

Basalts of the As Sarat include intersertal, trachytic, and pilotaxitic textural types. Pre-hiatus flows are dominated by massive intersertal basalt, whereas post-hiatus flows are texturally trachytic and pilotaxitic basalt. Stratigraphic separation of the textural types is not complete as the three types are intercalated to a limited extent. Finely devitrified green glass, variably altered to clay minerals, forms trace to 22 percent of the basalts. All three textural types include spherical amygdules filled with calcite and a very fine-grained (unknown) green mineral. Olivine is present in all basalt samples and is partially altered to iddingsite or serpentine in most places. Pyroxene in the basalts is reddish- to purplish-brown, unaltered titaniferous clinopyroxene. Plagioclase crystals are unaltered and occur principally in the groundmass of As Sarat basalt, though it forms phenocrysts in a few of the pilotaxitic and trachytic basalts. Apatite and Fe-Ti oxides are the characteristic accessory minerals.

The intersertal basalt is distinctly coarser grained than either of the other two textural types; groundmass crystals range from 0.1 to 0.5 mm. Olivine (5 to 27 percent, mean 14) forms subhedral phenocrysts (0.15 percent, mean 5) in the intersertal basalt and is most abundant in this textural type. Clinopyroxene (21 to 41 percent, mean 31) is subophitic to ophitic. Groundmass plagioclase (22 to 52 percent, mean 38) forms euhedral to subhedral grains. Opaque oxides (1 and 12 percent, mean 7) are mostly magnetite but include some ilmenite. Glass (0 to 18 percent, mean 5) is interstitial.

The trachytic basalts are platy due to plagioclase lath alignment and have an average groundmass grain size of 0.1 to 0.2 mm. The pilotaxitic basalts are very fine grained, with average groundmass grain sizes of generally less than 0.1 mm. The trachytic and pilotaxitic basalts contain phenocrysts of olivine (1 to 14 percent, mean 7) and/or plagioclase (37 to 60 percent, mean 45). Clinopyroxene (10 to 39 percent, mean 29) in the trachytic basalt is granular and very fine-grained. Opaque oxides (7 to 15 percent, mean 11) are mostly magnetite, although some of the trachytic basalts contain as much as or more ilmenite than magnetite. Glass (0 to 22 percent, mean 8) is interstitial.

Basalt dikes and necks of the As Sarat field are petrographically similar to the post-hiatus basalts. Some of the dikes and necks contain abundant clinopyroxene phenocrysts that may indicate crystal settling and accumulation in the magma chamber. Two of the basalt necks contained sparse small (<2 cm) peridotite (wehrlite) xenoliths and plagioclase xenocrysts.

The trachyte domes are composed of a leucocratic porphyry that contains about 30 percent carlsbad-twinned phenocrysts of anorthoclase 0.8 to 2.5 mm long in a trachytic groundmass. The groundmass is composed of feldspar, and minor opaque oxides and aegirine-augite, and has an average grain size of 0.1 mm. Small

quantities of feldspathoids are also present as phenocrysts and in the groundmass. Accessory minerals are biotite, arfvedsonite, opaque oxides, zircon, and apatite.

CHEMICAL DATA

Whole Rock

The alkaline/subalkaline classification of Irvine and Baragar (1971) was used to establish the alkalinity of As Sarat rocks (fig. 4), whereas, the classification system of the International Union of Geological Sciences (Le Bas and others, 1986) was used to determine rock names based on chemical compositions (fig. 4). Pre-hiatus flows in the CRS are composed, with two exceptions, of subalkaline basalt (average 3.02 percent $\text{Na}_2\text{O} + \text{K}_2\text{O}$ and 46.7 percent SiO_2 , anhydrous). The post-hiatus flows include alkali basalt, hawaiite, and minor weakly alkaline basalt (average 4.52 percent $\text{Na}_2\text{O} + \text{K}_2\text{O}$ and 47.9 percent SiO_2 , anhydrous). The intrusive trachytes are alkaline (average 11.36 percent $\text{Na}_2\text{O} + \text{K}_2\text{O}$ and 63.0 percent SiO_2 , anhydrous) and the presence of alkali pyroboles indicates that they are peralkaline.

The As Sarat basalts are silica undersaturated. The majority of pre-hiatus basalts are hypersthene normative, whereas most of the post-hiatus basalts and all of the neck and dike samples are nepheline normative (as much as 9.8 percent). Compositions of dike and neck samples are similar to those of post-hiatus flows (figs. 4, 5; table 2). A distinctive feature of the As Sarat basalts is their unusually high TiO_2 contents. Average TiO_2 contents of the pre- and post-hiatus flow rocks, necks, and dikes are 1.92, 2.62, 2.63, and 2.94 anhydrous weight percent, respectively. The As Sarat rocks follow an AFM trend that is similar to that for Hawaiian alkaline basalts (fig. 6). The former have compositions very similar to those of the Termaber Formation of the Ethiopian Plateau (Mohr and Zanettin, 1989) but are somewhat more alkaline than the main Yemen Trap Series (Chiesa and others, 1989).

The abundances of the major oxides vary considerably between flows of the CRS (fig. 5); variation is most erratic for the pre-hiatus basalts. In post-hiatus basalts the average contents of Al_2O_3 , Na_2O , K_2O , TiO_2 , and P_2O_5 , are distinctly greater, whereas total iron, MgO , CaO , and $\text{Na}_2\text{O}/\text{K}_2\text{O}$ are distinctly lower than in pre-hiatus subalkaline basalts of the CRS (table 2). A significant compositional discontinuity (fig. 5) is coincident with the eruption hiatus (between samples 112111 and 112110). Sample 112110 is compositionally transitional relative to pre- and post-hiatus basalt compositions. Its transitional character may indicate that initial (alkaline) lava erupted following the hiatus was contaminated by solidified, unerupted (subalkaline) lava remaining in eruption conduits at the end of pre-hiatus volcanism. In rocks of the As Sarat field, abundances of Al_2O_3 , Na_2O , K_2O , P_2O_5 , Ba, and Zr decrease with increasing MgO , whereas abundances of CaO , increase with increasing MgO ; these variations are in accord with observed variation in mineral modes. Abundances of total iron, TiO_2 , and MnO are uncorrelated with MgO variation.

Compositional data for each of the As Sarat field components (pre- and post-hiatus lava flows, necks, dikes, and trachyte domes) are characterized by groups of oxides and elements whose abundances display mutual covariation (table 3); correlation coefficients for abundances of oxides and elements within these groupings exceed 0.9. These covariance relations suggest that compositional variation within each component was controlled by a different mineral assemblage. Minerals inferred to have controlling influence on compositional variation occur as phenocryst phases. Fractional crystallization and partial melting, which can lead to removal, accumulation, and retention of crystals, result in predictable compositional variation because compositions of crystallizing/melting minerals are known or inferred. An additional consequence of fractional crystallization and partial melting is variable incompatible trace element enrichment of the liquid; an elemental covariation grouping, involving combinations of K_2O , Na_2O , Rb, Sr, Ba, Nb, Y, Zr, and Zn, denotes this process. Covariance relations among the rock groups indicates mineral component crystallization/crystal-settling control as follows:

pre-hiatus basalts: olivine, Fe-Ti oxides, apatite

dikes: clinopyroxene, Fe-Ti oxides

post-hiatus basalts: clinopyroxene

trachytes: plagioclase, aegirine, Fe-Ti oxides

necks: clinopyroxene, Fe-Ti oxides, apatite

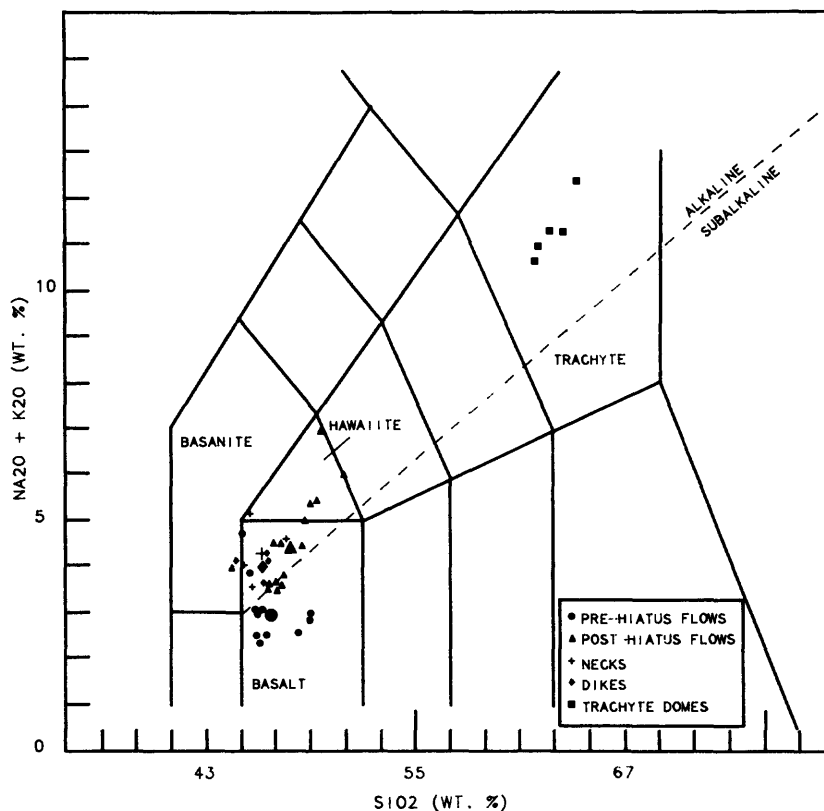


Figure 4.—Total alkalis-silica volcanic rock classification diagram after IUGS (Le Bas and others, 1986) showing compositions of samples from the As Sarat field (analyses recalculated to 100 percent anhydrous). Dashed line is the alkaline-subalkaline dividing line of Irvine and Baragar (1971). Enlarged plot symbols represent average compositions for rock type denoted by that symbol type.

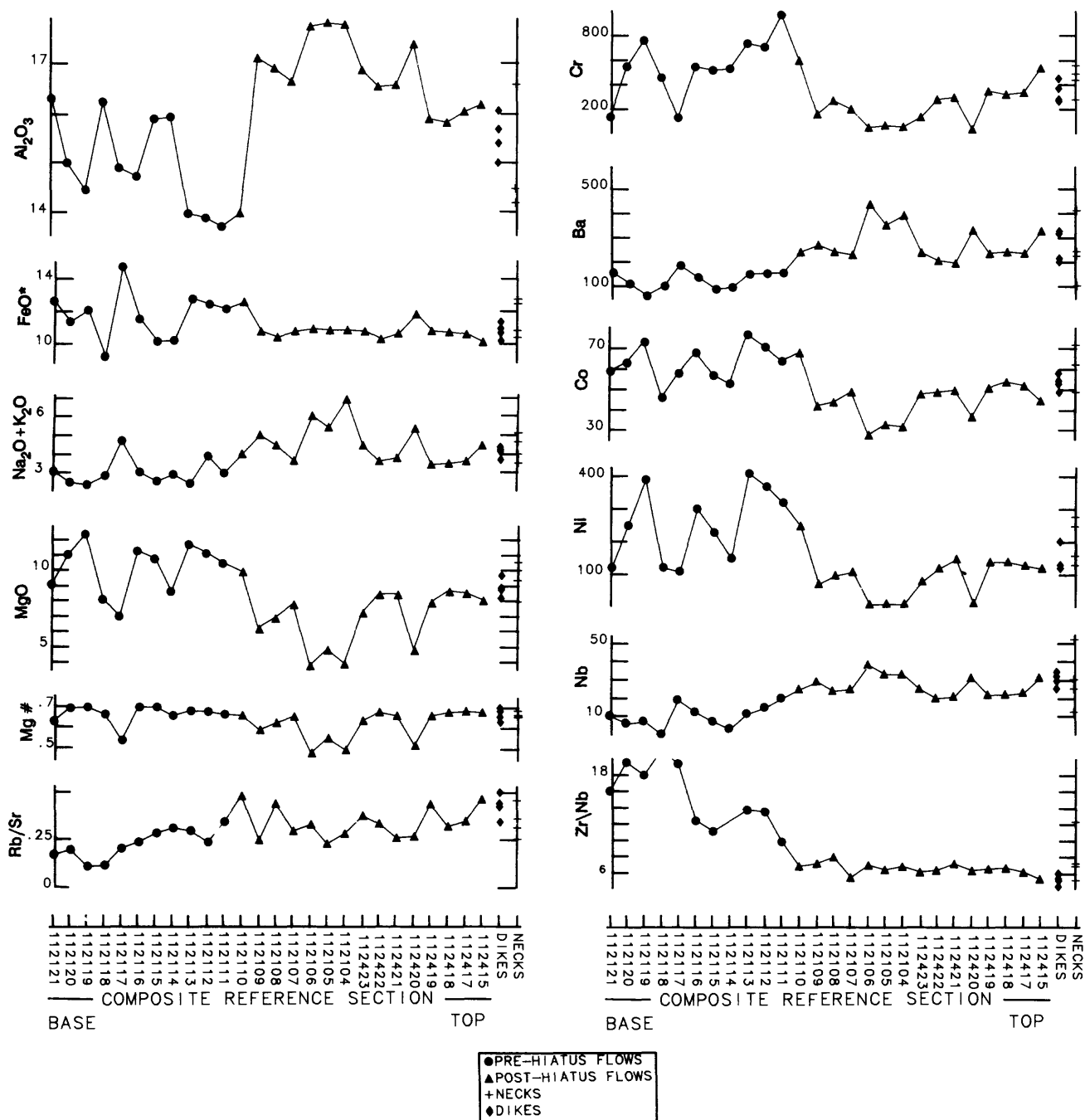


Figure 5.--Diagrams showing variations in major and trace element abundances versus stratigraphic position (analyses recalculated to 100 percent anhydrous). Data for major oxides expressed as weight percent; trace element data expressed in parts per million.

Table 2.--Major element, CIPW normative (calculated for analyses normalized to 100 percent, anhydrous) and trace element compositions for samples of the As Sarat field, Kingdom of Saudi Arabia.

Flows														
Field No.	112121	112120	112119	112118	112117	112116	112115	112114	112113	112112	112111	112110	112109	112108
Latitude: (18°N.)	10°58"	10°58"	10°58"	10°58"	10°58"	10°57"	10°57"	10°57"	10°56"	10°56"	10°55"	10°55"	10°56"	10°56"
Longitude: (43°E.)	10°54"	10°53"	10°53"	10°52"	10°52"	10°51"	10°51"	10°48"	10°45"	10°42"	10°41"	10°40"	10°32"	10°25"
Chemical analyses, weight percent														
SiO ₂	43.9	43.7	45.0	48.0	43.9	44.5	46.6	46.5	43.1	45.1	44.0	43.3	47.6	47.4
Al ₂ O ₃	15.6	14.1	14.1	15.9	14.5	14.2	15.3	15.1	13.1	13.7	13.1	13.6	16.7	16.5
Fe ₂ O ₃	3.56	3.54	3.64	4.49	6.36	6.90	4.18	7.31	8.67	5.78	7.54	8.03	4.40	4.97
FeO	9.30	7.97	8.96	5.50	9.12	5.40	6.50	3.59	4.62	7.58	5.27	5.46	7.02	6.15
MgO	8.70	10.4	12.1	7.94	6.79	10.9	10.4	8.17	11.0	11.0	10.0	9.66	6.09	6.73
CaO	8.89	9.81	9.48	12.1	7.98	9.68	9.79	10.3	9.09	9.53	11.0	10.6	7.97	8.81
Na ₂ O	2.40	1.92	2.05	2.54	3.82	2.48	2.21	2.47	1.70	3.22	2.29	2.83	3.87	3.40
K ₂ O	0.53	0.43	0.25	0.25	0.78	0.43	0.26	0.29	0.57	0.58	0.55	1.03	0.99	0.95
TiO ₂	2.26	1.74	1.67	1.13	3.33	1.89	1.03	1.27	2.01	2.11	1.88	2.59	2.30	2.18
P ₂ O ₅	.37	.27	.20	.15	.71	.30	.16	.18	.26	.39	.33	0.41	.61	.50
MnO	.18	.17	.18	.16	.19	.17	.18	.18	.20	.19	.19	.19	.18	.17
H ₂ O ¹	3.47	5.66	2.31	1.66	3.10	3.42	3.09	4.39	6.53	1.48	4.16	2.36	1.84	2.25
Total	99.16	99.71	99.94	99.82	100.58	100.27	99.70	99.75	100.85	100.66	100.13	100.06	99.57	100.01
CIPW norms, weight percent														
Q	-	-	-	-	-	-	-	-	-	-	-	-	-	-
or	3.3	2.7	1.5	1.5	4.8	2.6	1.6	1.8	3.6	3.5	3.4	6.3	6.0	5.8
ab	21.3	17.3	17.8	22.0	24.4	19.7	19.4	22.1	15.4	17.3	15.5	10.3	32.3	29.4
an	31.7	30.5	29.3	31.9	20.7	27.4	32.3	30.9	28.2	21.5	25.0	22.0	25.9	27.7
ne	-	-	-	-	4.8	1.2	-	-	-	5.6	2.6	7.8	.7	.1
di	9.8	15.8	14.3	23.2	12.7	16.6	13.9	17.8	14.8	19.4	24.4	24.0	8.7	11.4
hy	3.9	5.8	5.3	3.1	-	-	11.1	9.4	8.6	-	-	-	-	-
ol	22.6	21.8	26.0	14.1	21.9	26.2	17.6	13.4	22.6	25.7	22.6	21.5	18.6	18.5
mt	2.1	1.9	2.0	1.6	2.5	1.9	1.7	1.7	2.1	2.1	2.0	2.1	1.8	1.8
il	4.5	3.5	3.3	2.2	6.5	3.7	2.0	2.6	4.1	4.1	3.7	5.1	4.5	4.3
ap	0.9	0.7	0.5	0.4	1.7	.7	0.4	0.5	0.7	0.9	0.8	1.0	1.5	1.2
Trace elements, parts per million														
Rb	12	15	4	4	21	12	8	10	14	14	20	30	15	24
Sr	713	763	362	338	1042	512	279	321	470	595	575	621	604	543
Y	22	20	20	21	20	20	23	22	19	24	21	21	27	25
Zr	160	117	126	71	369	149	78	91	151	202	196	172	208	190
Nb	10	6	7	<3	19	12	7	3	11	15	20	25	29	24
Ba	156	111	62	102	187	138	90	94	151	152	156	242	269	242
Co	59	63	73	46	58	68	57	53	77	71	64	68	42	44
Cr	140	550	760	460	130	550	520	530	740	710	970	600	160	270
Cu	68	58	89	60	61	100	86	54	79	79	69	81	49	45
Ni	120	250	390	120	110	300	230	150	410	370	320	250	72	96
Sc	27	30	30	40	10	27	33	34	24	24	27	23	20	23
V	210	220	240	250	200	230	220	250	240	240	220	260	150	170
Zn	88	79	83	61	120	75	67	73	88	97	96	97	70	72

Table 2.—Major element, CIPW normative (continued)

Flows														
Field No.	112107	112106	112105	112104	112423	112422	112421	112420	112419	112418	112417	112415	112101	112102
Latitude: (18°N.)	10°56"	10°56"	10°56"	10°56"	09°40"	09°39"	09°37"	09°35"	09°33"	09°31"	09°29"	09°27"	55°44" ⁿ²	55°46" ⁿ²
Longitude: (43°E.)	10°20"	10°19"	10°17"	10°16"	09°53"	09°53"	09°54"	09°54"	09°55"	09°56"	09°57"	09°58"	15°26"	15°51"
Chemical analyses, weight percent														
SiO ₂	45.8	49.0	48.3	49.1	46.2	46.5	46.4	47.7	45.7	45.2	44.7	46.7	44.7	50.3
Al ₂ O ₃	16.2	17.1	17.4	17.6	16.6	16.2	16.2	16.9	15.4	15.3	15.3	15.9	14.1	17.4
Fe ₂ O ₃	5.11	6.14	5.04	5.18	4.32	3.88	3.54	5.71	6.65	5.53	4.79	4.07	2.98	4.97
FeO	6.38	5.45	6.53	6.50	7.18	7.04	7.70	6.83	4.90	5.82	6.22	6.78	9.74	6.24
MgO	7.58	3.67	4.70	3.92	7.19	8.31	8.29	4.71	7.72	8.39	8.17	7.99	9.48	3.73
CaO	9.77	5.90	6.47	6.12	9.38	10.00	9.07	6.54	10.3	10.2	9.98	9.49	9.96	6.93
Na ₂ O	2.63	4.22	4.03	5.35	3.38	2.70	2.91	4.03	2.40	2.49	2.52	3.16	3.15	4.26
K ₂ O	0.92	1.59	1.28	1.52	1.03	0.83	0.82	1.19	0.95	0.92	0.91	1.26	1.25	1.84
TiO ₂	2.56	2.58	3.20	2.87	2.62	2.17	2.26	2.92	2.59	2.52	2.47	2.44	2.84	2.20
P ₂ O ₅	.33	0.67	0.55	0.66	0.37	.31	.37	.66	.36	.36	.35	0.56	0.56	0.93
MnO	.17	.17	.16	.17	.17	.17	.18	.19	.17	.18	.16	.17	.21	.22
H ₂ O ¹	2.80	2.92	2.75	.66	1.10	1.71	1.47	2.31	3.32	3.25	2.26	1.03	.71	2.07
Total	100.25	99.41	100.41	99.57	99.54	99.82	99.21	99.69	100.46	100.16	97.83	99.55	99.68	101.09
CIPW norms, weight percent														
Q	-	-	-	-	-	-	-	-	-	-	-	-	-	-
or	5.6	9.8	7.8	9.1	6.2	5.0	5.0	7.3	5.8	5.6	5.7	7.6	7.5	11.0
ab	22.9	37.2	35.1	36.5	22.5	22.9	25.3	35.2	21.0	20.4	20.8	22.5	13.8	36.5
an	30.6	24.0	26.3	19.8	27.6	30.3	29.5	25.3	29.4	28.9	29.1	25.9	20.9	23.2
ne	-	-	-	5.1	3.6	.2	-	-	-	.8	.9	2.6	7.1	-
di	14.0	1.6	2.5	5.5	14.2	15.0	11.6	3.1	17.2	17.2	16.7	14.9	20.7	4.6
hy	.5	8.4	6.5	-	-	-	.7	4.8	2.0	-	-	-	-	3.9
ol	18.8	10.4	12.5	15.1	18.1	19.9	21.0	15.1	16.8	19.5	19.3	18.8	21.2	12.6
mt	1.8	1.8	1.8	1.8	1.8	1.7	1.8	2.0	1.8	1.8	1.8	1.7	2.0	1.8
il	5.0	5.1	6.3	5.5	5.1	4.2	4.4	5.7	5.1	5.0	4.9	4.7	5.5	4.2
ap	0.8	1.7	1.3	1.6	0.9	0.8	0.9	1.6	0.9	0.9	0.9	1.4	1.3	2.2
Trace elements, parts per million														
Rb	15	28	20	25	19	16	16	20	26	22	21	29	30	40
Sr	503	840	876	884	497	471	602	736	592	670	598	617	734	1222
Y	22	25	25	28	27	22	22	26	20	19	23	23	25	25
Zr	138	265	213	226	154	128	151	198	145	146	142	166	201	253
Nb	25	38	33	33	25	20	21	31	22	22	23	31	39	52
Ba	229	435	352	390	237	205	195	328	236	240	235	326	297	581
Co	49	28	33	32	48	49	50	37	51	54	52	45	56	21
Cr	200	49	67	59	140	280	300	41	350	320	340	540	640	<1
Cu	59	26	30	24	61	71	59	26	67	69	63	49	64	12
Ni	110	10	11	11	81	120	150	14	140	140	130	120	190	<2
Sc	26	7	10	10	25	28	22	14	25	26	26	23	25	5
V	200	67	98	74	190	200	190	140	230	230	220	190	210	85
Zn	71	70	56	66	71	68	68	66	77	70	74	69	85	87

Table 2.—Major element, CIPW normative (continued)

	Necks				Dikes				Trachyte domes				
Field No.	112052	112054	112087	112103	112092	112094	112097	112098	108557	108894	108896	108897	108899
Latitude: (18°N.)	02°14"	09°28"	00°11"	00°06"	00°27"	59°43" ²	02°34"	02°01"	07°40"	07°23"	07°34"	09°22"	08°24"
Longitude: (43°E.)	21°40"	19°50"	33°01"	17°21"	11°23"	10°31"	10°29"	10°26"	07°17"	08°40"	08°23"	04°43"	06°30"
Chemical analyses, weight percent													
SiO ₂	45.1	45.0	45.2	46.9	45.6	45.8	45.3	44.1	63.2	59.5	58.8	61.3	60.8
Al ₂ O ₃	14.3	14.1	14.3	16.3	15.1	15.7	15.2	14.7	17.6	17.6	17.3	17.5	17.9
Fe ₂ O ₃	3.33	3.6	4.06	4.99	3.86	3.56	3.43	2.49	2.48	1.99	3.52	2.72	2.10
FeO	7.71	9.58	9.39	6.58	7.68	7.68	7.26	9.37	1.50	3.27	2.08	1.84	2.75
MgO	9.24	10.0	10.4	7.86	9.54	8.06	8.69	8.68	0.17	0.51	0.64	0.39	0.46
CaO	10.4	9.86	9.32	8.35	9.77	9.73	9.60	10.8	.85	1.74	1.88	1.32	1.37
Na ₂ O	3.40	2.84	2.85	3.54	2.65	2.80	2.78	2.72	7.44	6.56	6.19	6.61	6.62
K ₂ O	1.69	1.15	0.63	1.00	0.96	1.28	1.43	1.38	4.73	3.97	3.92	4.31	4.37
TiO ₂	2.99	2.74	2.27	2.39	2.55	2.71	2.90	3.37	.05	.22	.21	.14	.17
P ₂ O ₅	.64	0.49	.35	0.46	.37	0.56	0.36	0.47	.05	.18	.19	.11	.11
MnO	.20	.20	.18	.17	.17	.17	.16	.17	.21	.20	.20	.20	.17
H ₂ O ¹	.86	.50	.29	1.02	1.88	1.88	1.95	.70	1.05	3.88	4.44	3.15	2.57
Total	99.86	100.11	99.24	99.56	100.13	99.93	99.06	98.95	99.33	99.62	99.37	99.59	99.39
CIPW norms, weight percent													
Q	-	-	-	-	-	-	-	-	-	-	-	.3	-
or	10.1	6.8	3.8	6.0	5.8	7.7	8.7	8.3	28.5	24.5	24.5	26.5	26.7
ab	11.1	14.9	19.4	26.8	19.7	20.6	18.0	11.6	61.9	58.1	55.3	58.1	58.0
an	19.0	22.5	24.7	26.1	27.0	27.1	25.6	24.3	.7	7.2	8.3	5.6	6.3
ne	9.8	5.0	2.7	2.0	1.8	2.0	3.4	6.4	1.2	-	-	-	-
di	23.5	19.2	16.2	10.7	16.2	15.1	17.2	22.1	2.9	.6	0.2	0.4	-
hy	-	-	-	-	-	-	-	-	-	3.6	8.8	7.9	4.1
ol	17.5	23.1	25.9	20.9	21.9	19.1	18.9	17.7	4.0	4.4	1.1	-	3.5
mt	1.8	2.1	2.1	1.8	1.8	1.8	1.7	1.9	.6	.9	.9	.7	.8
il	5.8	5.2	4.4	4.6	4.9	5.3	5.7	6.5	.1	.4	.4	.3	.3
ap	1.5	1.2	0.8	1.1	.9	1.4	0.9	1.1	.1	.5	.5	.3	.3
Trace elements, parts per million													
Rb	32	31	13	22	21	33	25	25	165	103	111	118	106
Sr	882	675	502	692	490	646	560	717	91	284	301	128	189
Y	24	26	24	24	26	22	24	24	41	29	28	29	26
Zr	272	217	148	173	151	185	151	140	937	656	609	758	645
Nb	52	30	12	25	25	34	29	32	146	103	111	107	93
Ba	413	226	103	244	216	321	316	327	196	885	868	723	623
Co	49	62	72	49	58	49	55	53	<1	2	3	<1	2
Cr	490	560	440	280	450	280	270	370	<1	<1	<1	<1	<1
Cu	59	77	87	37	77	69	58	61	7	7	6	7	6
Ni	170	260	290	140	20	120	120	130	<2	<2	<2	<2	<2
Sc	24	26	23	20	27	22	25	25	<2	<2	<2	<2	<2
V	240	220	250	160	220	220	240	250	<2	<2	2	<2	<2
Zn	92	99	110	83	85	84	72	80	140	95	110	110	100

Table 2.—Major element, CIPW normative (continued)

	AS SARAT AVERAGES						MORB ³	Lith ⁴	Rahat ⁵
	CRS flows	Subalka- line flows	Alkaline flows	Necks	Dikes	Trachyte Domes			
Chemical analyses, weight percent									
SiO ₂	45.92	44.94	46.64	45.55	45.20	60.72	49.34	46.68	47.12
Al ₂ O ₃	15.45	14.43	16.19	14.75	15.18	17.58	17.04	15.41	16.08
Fe ₂ O ₃	3.74	3.33	4.04	3.66	3.34	1.66	1.99	4.30	-
FeO	7.99	8.78	7.41	8.25	8.00	3.10	6.82	6.18	11.85
MgO	8.10	9.76	6.87	9.38	8.74	0.43	7.19	7.28	7.67
CaO	9.16	9.79	8.71	9.48	9.98	1.43	11.72	9.68	9.78
Na ₂ O	2.96	2.46	3.33	3.16	2.74	6.68	2.73	3.17	3.59
K ₂ O	0.81	0.45	1.08	1.12	1.26	4.26	0.16	0.86	0.80
TiO ₂	2.25	1.85	2.55	2.60	2.88	0.16	1.49	1.86	2.48
P ₂ O ₅	0.40	0.30	0.47	0.48	0.44	0.13	.16	.35	.50
MnO	0.18	0.18	0.17	0.19	0.17	0.20	.17	.17	.19
H ₂ O ¹	2.74	3.57	2.14	0.67	1.60	3.02	1.17	2.37	.45
Total	99.70	99.84	99.60	99.66	99.52	99.37	99.98	98.31	100.06
CIPW norms, weight percent									
or	5.0	2.8	6.6	6.7	7.6	26.2	1.0	5.3	4.7
ab	24.9	21.7	27.2	18.4	17.4	58.8	23.4	25.9	23.2
an	27.4	28.1	26.8	23.2	26.0	5.6	34.2	26.4	25.4
ne	.5	-	1.0	4.8	3.4	-	-	1.2	3.9
di	13.8	16.6	11.9	17.5	17.7	.8	14.5	17.5	16.3
hy	-	.2	-	-	-	4.2	7.1	-	-
ol	21.1	24.3	18.7	21.5	19.4	3.1	2.5	17.5	18.9
mt	1.9	2.0	1.8	1.9	1.8	.8	2.9	1.7	1.9
il	4.4	3.7	5.0	5.0	5.6	.3	2.9	3.4	4.7
ap	1.0	.7	1.2	1.2	1.1	.3	0.4	0.9	1.2
Trace elements, parts per million									
Rb	18	12	22	25	26	128	1.16	22	7
Sr	601	543	644	688	603	173	40	424	550
Y	23	21	24	25	24	27	43	-	25
Zr	167	156	176	203	157	675	95	151	197
Nb	20	10	27	30	30	110	<30	-	23
Ba	214	127	277	247	295	716	7.8	-	108
Co	52	63	46	58	54	2	32	-	-
Cr	376	551	248	443	343	1	297	180	215
Cu	61	73	52	65	66	7	77	-	72
Ni	163	252	97	215	143	2	97	-	122
Sc	24	28	21	23	25	2	61	-	23
V	197	229	174	218	233	2	292	-	228
Zn	76	84	71	96	80	111	-	-	86

¹ H₂O is loss on ignition

² Latitude is prefixed by 17°

³ Mid ocean ridge basalt (Engel and others, 1965; Hart, 1971)

⁴ Average of 8 Red Sea basalts (sample numbers 165006, 165582, 165567, 165661, 165881, 175834, 165767, and 175771) from the central Arabian coastal plain (Pallister, 1987)

⁵ Average of 64 alkali olivine basalts from Harrat Rahat (Camp and Roobol, 1989)

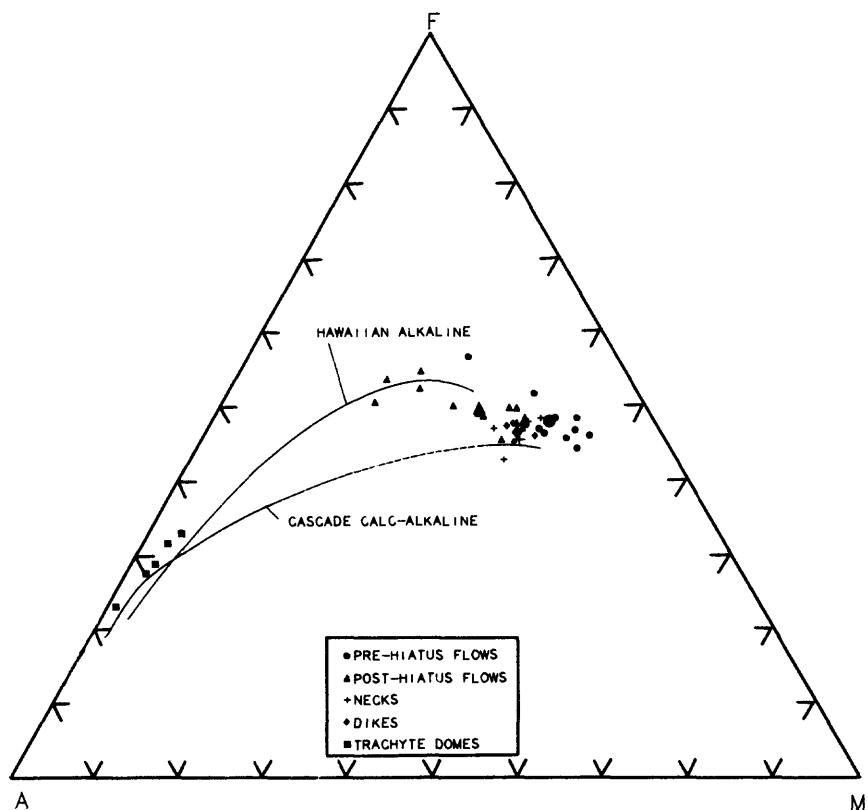


Figure 6.--Ternary AFM diagram showing compositions of samples from the As Sarat field. Enlarged plot symbols represent average compositions for rock type denoted by that symbol type. Trend lines from Irvine and Baragar (1971).

Table 3.--Groupings, by lithology, whose constituent oxide and elemental pairs display abundance covariation in components of the As Sarat field. Most oxide/element pairs within a grouping have correlation coefficients ≥ 0.9 .

Lithology	Oxide/element groupings	Inferred controlling phases
Pre-hiatus basalt	FeO+TiO ₂ +P ₂ O ₅ +Zr+Sr+Zn+Co MgO+Ni+Cr K ₂ O+Rb+Ba+Zn	iron-titanium oxide and apatite olivine melt incompatible trace element enrichment
Post-hiatus basalt	CaO+MgO+Co+Cu+Ni+Sc+V K ₂ O+Ba+Nb+Zr+P ₂ O ₅	Ca/Mg-clinopyroxene melt incompatible trace element enrichment and apatite
Basalt necks	MgO+FeO+Co+Cu+Ni+Zn+Nb+Sc CaO+MnO+Cr+Sc K ₂ O+TiO ₂ +P ₂ O ₅ +Sr+Ba+Zr+Rb+Nb	Mg/Fe-clinopyroxene unknown melt incompatible trace element enrichment, apatite, and iron-titanium oxide
Basalt dikes	MgO+Co+Sc+Y+Cu+Ni+V K ₂ O+Ba Ti ₂ O+V	Mg-clinopyroxene melt incompatible trace element enrichment iron-titanium oxide
Trachyte domes	SiO ₂ +K ₂ O+Na ₂ O+Zr+Y+Zn+Nb FeO+MgO+CaO+TiO ₂ +P ₂ O ₅ +Sr+Ba+Ca	melt incompatible trace element enrichment aegirine and apatite

The abundances of the trace elements also vary considerably within the flows of the CRS (fig. 5); variation is about equally erratic, over similar ranges, in pre- and post-hiatus basalts. In the post-hiatus basalts the average contents of Rb, Sr, Ba, Zr, and Nb are distinctly greater, whereas those of Co, Cr, Cu, Ni, Sc, V, and Zn are distinctly lower than in the pre-hiatus flows of the CRS (table 2). The previously described compositional discontinuity coincident with the hiatus is also denoted by trace element abundances; sample 112110 is similarly compositionally transitional. In rocks of the As Sarat field abundances of Co, Cu, Ni, Sc, V, Zn, and Cr, increase with increasing MgO, whereas abundances of Ba, Y, Nb, Rb, Zr, and Sr decrease with increasing MgO. Ternary proportions of Ti, Zr, and Y (fig. 7) in As Sarat basalt are similar to those characteristic of within-plate basalts; three of the pre-hiatus basalts have compositions similar to that of ocean floor basalt.

Total REE contents and chondrite-normalized REE patterns for selected samples of the CRS are strikingly similar (table 4, fig. 8). Individual patterns show slight light-REE (LREE) enrichment and very small positive Eu anomalies. Post-hiatus basalts are more LREE enriched ($\text{La/Lu}_{\text{CN}} = 6.36$ versus 3.50) than average pre-hiatus basalt. Chondrite normalized patterns and REE contents for one basalt neck and one basalt dike are essentially indistinguishable from those of the pre- and post-hiatus flow rocks, respectively. However, the trachyte contains a much greater total-REE content and its chondrite normalized REE pattern shows strong LREE enrichment and moderate heavy REE (HREE) enrichment centered on a moderate negative europium anomaly. REE contents of the As Sarat basalts are similar to those reported for alkali basalts by Camp and Roobol (1989) for Miocene to Recent flows of Harrat Rahat (600 km to the north-northwest) and by Pallister (1987) for Tertiary flows and dikes of the Red Sea coastal plain (400 km to the northwest).

Spidergrams (fig. 9) facilitate comparison of incompatible trace element abundances in the As Sarat basalts with those in other continental alkali basalts. With the exception of weakly positive Sr and P anomalies in the pre- and post-hiatus basalts, respectively, and moderate negative Ba and Th anomalies in all components of the As Sarat field, spidergrams for the As Sarat basalts are relatively smooth and gently negatively sloping. Abundances of the incompatible trace elements are greater in the post-hiatus basalts than in the pre-hiatus basalts; in particular the post-hiatus basalts are more notably enriched in the most incompatible trace elements. Basalt necks and dikes of the As Sarat field are characterized by spidergrams that are most similar to those of the pre- and post-hiatus basalts, respectively. Positive Nb anomalies characteristic of many continental alkali basalts (Norry and Fitton, 1983) are nearly absent in the As Sarat basalts. The overall abundances of the incompatible trace elements in the As Sarat basalts are low relative to values summarized by Camp and Roobol (1989) for continental alkali basalts, including other Red Sea rift-related basalts. In addition the As Sarat basalts are much less compatible-trace-element depleted (table 2; fig.9); average Ni and Cr values for flows of the CRS and for alkali olivine basalt of Harrat Rahat (Camp and Roobol, 1989) are 163 versus 122 and 376 versus 215, respectively.

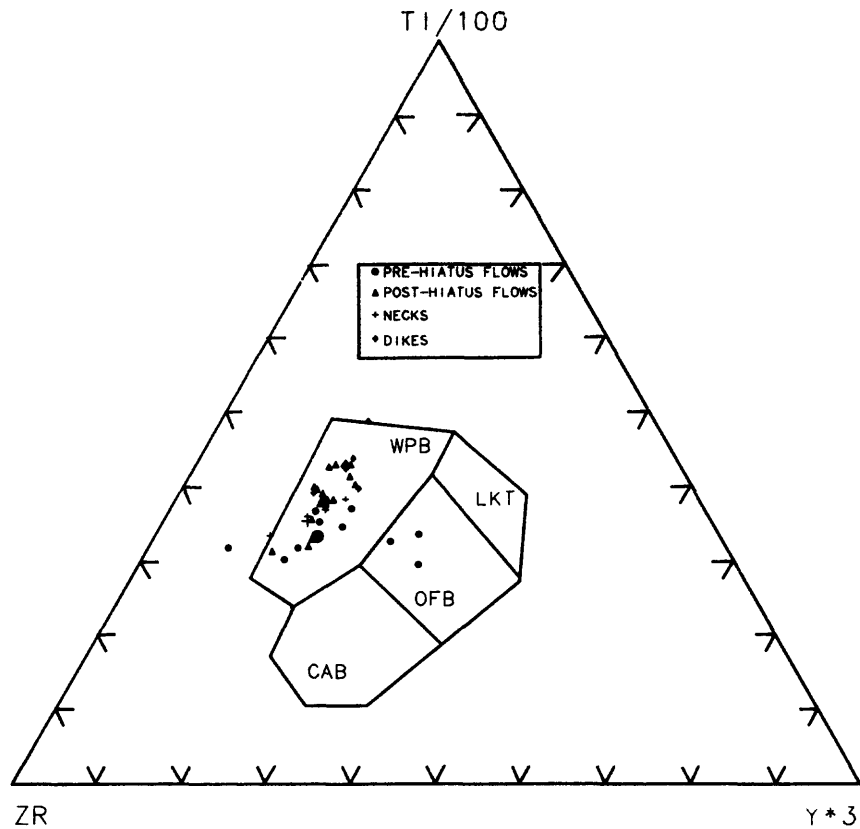


Figure 7.—Ternary diagram showing the proportions of Zr, Y, and Ti in samples of the As Sarat field. Enlarged plot symbols represent average compositions for rock type denoted by that symbol type. Field boundaries after Pearce and Cann (1973); WPB is within-plate basalt, LKT is low-potassium tholeiite, OFB is ocean-floor basalt, CAB is calc-alkali basalt.

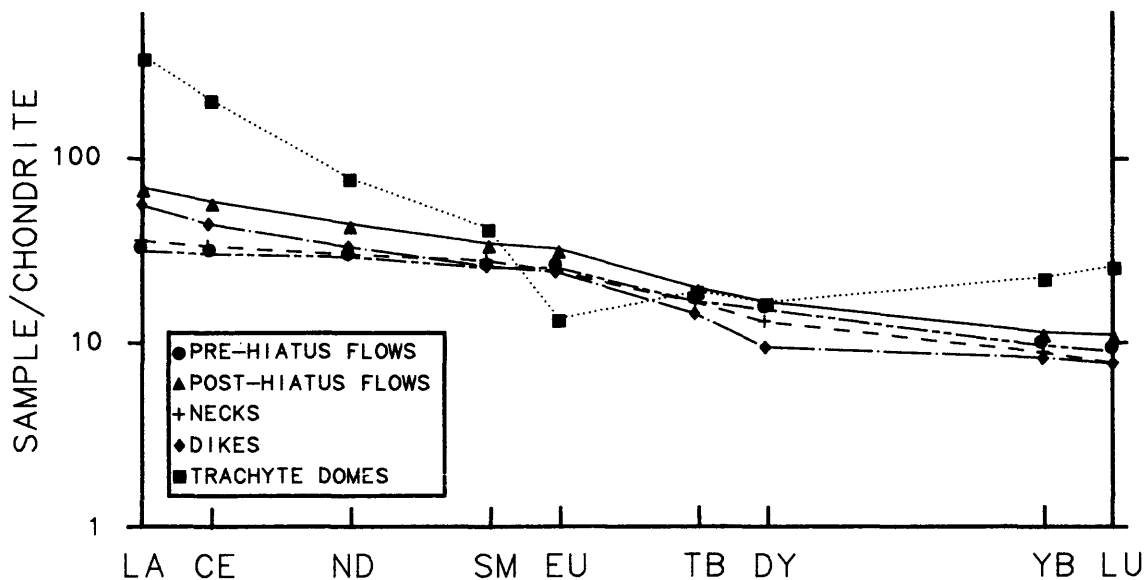


Figure 8.—Chondrite normalized REE plot showing average compositions of As Sarat field components.

Table 4.--Rare-element, uranium, and thorium analyses for selected samples as the As Sarat basalt field, Kingdom of Saudi Arabia. Hyphen indicates no data available.

	Pre-hiatus basalt			Post-hiatus basalt				Basalt neck	Basalt dike	Trachyte dome	MORB ¹	Lith ²	Rahat ³
Field No.	112121	112120	112116	112109	112104	112420	112418	112087	112097	108557			
Chemical analyses, parts per million													
La	12	10	15	23	28	26	18	12	19	120	-	22	15.96
Ce	31	24	36	51	57	56	41	30	39	191	9.99	49	38.76
Nd	21	19	21	28	31	30	23	20	22	52	8.79	29	21.06
Sm	6.2	4.9	5.8	7.1	7.7	7.7	5.8	6.0	5.7	9.0	3.09	7.3	4.84
Eu	2.35	1.85	2.12	2.35	2.57	2.84	2.04	1.99	1.95	1.13	1.15	2.5	1.81
Tb	0.96	0.81	0.86	1.0	1.03	0.96	0.80	0.85	0.75	1.00	-	1.2	-
Dy	6.0	4.9	5.0	6.0	5.7	6.0	4.7	4.6	3.4	6.0	-	6.9	4.38
Yb	2.5	2.0	1.9	2.6	2.7	3.0	2.0	2.0	1.9	5.4	2.94	3.4	1.97
Lu	.38	.27	.28	0.40	0.39	.44	.28	.27	.27	0.94	-	0.44	0.31
Th	1.0	.7	1.2	2.1	2.3	2.3	1.8	1.1	2.1	30.0	-	2.6	-
U	.3	.3	.5	.7	.7	.6	.6	.3	.9	8.0	-	.7	-

1 Mid-ocean ridge basalt (Schilling, 1971)

2 Average of 9 analyses (sample numbers 165006, 165571, 165881, 175780, 165882, 165661, 175764, 175776, and 165767) presented by Pallister (1987) for Red Sea basalts from the central Arabian coastal plain

3 Alkali olivine basalt from Harrat Rahat (Camp and Roobol, 1989)

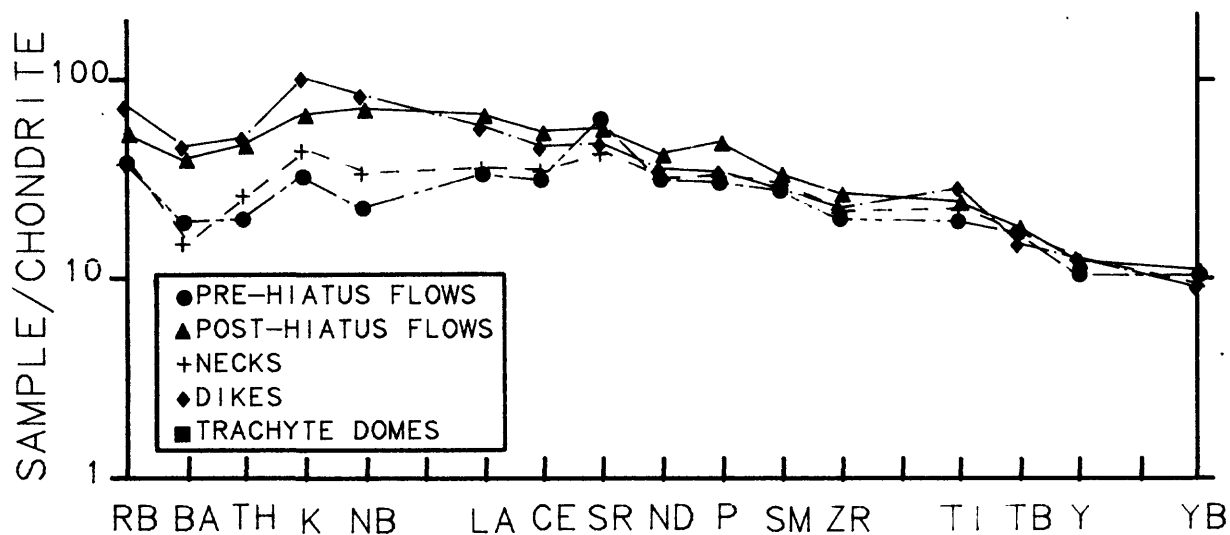


Figure 9.--Spidergrams showing average incompatible trace element abundances in As Sarat field components; normalized to chondrite values of Thompson and others (1983). Trace elements are in order of increasing incompatibility with mantle rocks; the most incompatible elements are on the left.

Mineral Chemistry

Plagioclase phenocryst compositions were determined for 10 samples from the As Sarat field (table 5, fig. 10). Compositional zoning in plagioclase grains is minor, as indicated by core to rim analytical traverses and by small standard deviations computed from nine analyses per sample, except in samples 112108, 112105, and 112422, which have distinctly more calcic cores. Plagioclase compositions in flows of the CRS range from calcic andesine to sodic bytownite; the average composition is labradorite. Plagioclase compositions do not vary systematically with position in the CRS though they may vary systematically within single flows. Feldspars in the necks and dike are compositionally indistinguishable from those in flows of the CRS. Feldspars in the trachytes are anorthoclase.

Phenocryst and groundmass olivines were analyzed from nine As Sarat basalt samples (table 6). Phenocryst core compositions vary from Fo 86 to 77, and are zoned outward to lower Fo values at the rims that are similar to groundmass olivines. Groundmass olivines are not strongly zoned, but display considerable between sample variation (Fo 81-61). The calcium content of phenocryst cores is highly variable, but shows a systematic increase outward in the zoned rims.

Most of the As Sarat basalts contain olivine and plagioclase phenocrysts, but a few flows, dikes, and necks also contain clinopyroxene phenocrysts. Table 7 presents average analyses for both groundmass and phenocrystal clinopyroxene from nine samples. Clinopyroxenes in the As Sarat basalts display little between sample compositional variation or zoning within individual grains, and with one exception, are all titaniferous magnesian salites (fig. 11). Relic cores of a magnesian augite that occurs in dike sample 112094 (table 7) as well as the forsteritic olivine phenocryst cores (i.e. Fo 86-83) may represent xenocrystic phases or phenocrysts whose compositions represent equilibrium with melt in the source region.

Table 5.--Representative compositions of feldspar phenocrysts (except 112422) in rocks of the As Sarat volcanic field. Mean compositions and associated standard deviations (\pm) were calculated from nine (three analyses on each of three grains) analyses/sample.

Field No.	Flow rocks						Neck	Dike	Trachyte
	112119	112111	112108	112105	112422	112415	112103	112094	108897
SiO ₂	52.97 \pm 1.76	50.54 \pm 0.85	55.18 \pm 5.19	56.45 \pm 2.88	54.23 \pm 3.84	50.57 \pm 0.41	52.94 \pm 1.69	51.06 \pm 0.60	66.39 \pm 0.48
Al ₂ O ₃	29.35 \pm 1.15	31.05 \pm 0.77	27.99 \pm 3.81	26.85 \pm 2.02	28.25 \pm 2.59	30.81 \pm 0.31	29.26 \pm 1.27	30.27 \pm 0.50	19.65 \pm 0.54
FeO	0.55 \pm 0.11	0.78 \pm 0.15	0.60 \pm 0.32	0.67 \pm 0.02	0.70 \pm 0.24	0.62 \pm 0.06	0.32 \pm 0.06	0.56 \pm 0.09	0.33 \pm 0.17
CaO	12.29 \pm 1.37	14.27 \pm 0.74	10.17 \pm 4.47	8.95 \pm 2.51	10.88 \pm 3.18	13.72 \pm 0.34	11.80 \pm 1.30	13.88 \pm 0.52	.61 \pm 0.43
Na ₂ O	4.44 \pm 0.70	3.27 \pm 0.35	5.28 \pm 1.81	5.95 \pm 1.10	4.83 \pm 1.58	3.48 \pm .21	4.72 \pm 0.76	3.45 \pm 0.31	8.15 \pm 0.56
K ₂ O	0.28 \pm 0.12	0.10 \pm 0.03	0.76 \pm 1.07	0.65 \pm 0.58	0.51 \pm 0.52	0.20 \pm 0.02	0.34 \pm 0.17	0.20 \pm 0.03	4.62 \pm 1.10
Ab	38.9	29.1	46.3	52.6	43.2	31.1	41.2	30.7	70.7
An	59.5	70.3	49.3	43.7	53.8	67.8	56.9	68.2	2.9
Or	1.6	0.6	4.4	3.8	3.0	1.2	2.0	1.2	26.4
	labradorite	bytownite	andesine	andesine	labradorite	labradorite	labradorite	labradorite	anorthoclase

Table 6.—Composition of olivines in rocks of the As Sarat volcanic field. Abbreviations used: Fo - forsterite component, Ca - Calcium in weight percent, W - total number of analyses per sample, N.O. - not observed in thin section.

	112119 Flow	112117 Flow	112111 Flow	112108 Flow	112105 Flow	112422 Flow	81 Flow	112094 Dike	112103 Neck
PHENOCRYSTS									
Fo range	85.9-69.8	76.8-73.5	78.8-63.3	76.9-58.3	N.O.	83.4-67.6	86.3-73.2	84.8-80.7	83.9-68.7
Fo ave.	82.3	73.4	73.8	72.2		74.6	81.4	83.4	79.6
Ca range	0.24-0.51	0.20-0.45	0.14-0.44	0.21-0.49		0.24-0.47	0.20-0.32	0.19-0.28	0.13-0.30
Ca ave.	0.32	0.25	0.26	0.30		0.33	0.25	0.23	0.21
N	31	9	12	10		12	11	14	8
GROUNDMASS									
Fo range	71.6-61.1	74.6-62.5	59.4-57.6	N.O.	62.5-57.2	67.7-64.5	70.1-69.5	80.8-76.8	71.2-65.9
Fo ave.	66.6	67.7	58.6		60.3	65.9	69.8	79.0	68.5
Ca range	0.50-0.59	0.28-0.57	0.40-0.52		0.34-0.59	0.47-0.54	0.40-0.44	0.22-0.30	0.28-0.41
Ca ave.	0.55	0.43	0.47		0.46	0.50	0.42	0.27	0.36
N	4	4	6		6	3	2	4	9

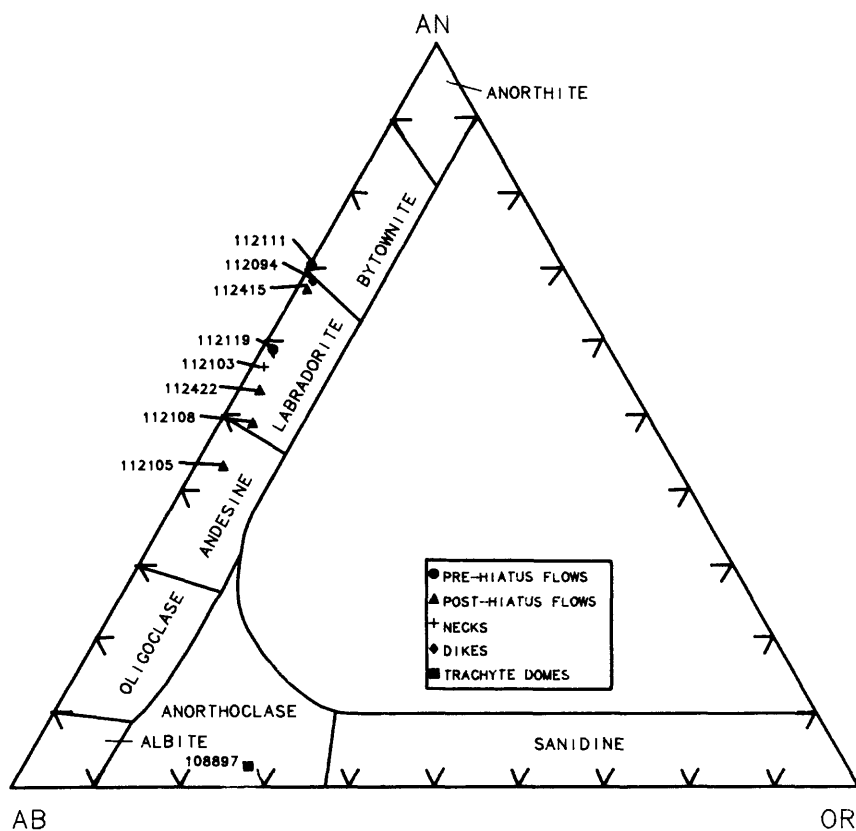


Figure 10.—Ternary diagram showing molecular proportions of feldspar end members in selected feldspars from rocks of the As Sarat field.

Table 7.—Compositions of clinopyroxenes in rocks of the As Sarat volcanic field. Mean oxide weight percent (m) and absolute standard deviation (sd) are presented. Total iron as FeO. Abbreviations: N, Number of analyses used for the average; WO, EN, and FS are the wollastonite, enstatite, and ferrosillite components, respectively.

Sample No.	-----112119-----				112111		112108		112105		112422	
	Flow				Flow		Flow		Flow		Flow	
	Phenocryst		Groundmass		Groundmass		Phenocryst		Groundmass		Groundmass	
	m	sd	m	sd	m	sd	m	sd	m	sd	m	sd
SiO ₂	49.19	1.48	50.30	1.71	50.05	0.90	50.22	1.10	50.58	0.30	51.00	0.41
TiO ₂	1.86	0.58	1.94	0.45	2.75	.48	2.76	0.44	2.10	.33	2.38	.25
Al ₂ O ₃	4.94	1.45	4.94	1.32	4.16	.64	4.41	.63	2.69	.36	4.42	.40
CaO	22.09	.49	21.44	.92	20.98	.34	21.21	.25	20.80	.52	20.94	2.0
FeO	7.89	.92	7.88	.79	8.92	.21	8.30	.32	8.68	.21	7.51	3.0
MgO	13.33	.88	13.09	.69	12.79	.52	12.92	.48	11.87	.15	13.29	2.6
sum	99.32		99.58		99.65		99.83		96.72		99.53	
N	11		14		5		9		4		8	
WO	47.2		46.8		45.9		46.5		47.2		46.3	
EN	39.6		39.7		38.9		39.4		37.4		40.8	
FS	13.2		13.4		15.2		14.2		15.4		12.9	

Sample No.	81		112103		----- 112094 -----					
	Flow		Neck		Dike					
	Groundmass		Groundmass		Relic Core		Phenocryst Rim		Groundmass	
	m	sd	m	sd	m	sd	m	sd	m	sd
SiO ₂	49.20	1.48	50.27	0.60	50.82	0.61	46.26	1.76	46.60	2.10
TiO ₂	2.78	0.70	2.09	.27	1.04	.17	2.94	0.58	3.31	0.73
Al ₂ O ₃	5.09	1.65	3.94	.77	6.69	.52	8.20	2.30	7.24	2.32
CaO	20.93	.40	22.52	.34	17.73	.50	21.73	.76	22.37	.60
FeO	7.04	.67	7.21	.13	6.92	.23	7.48	.33	8.22	.51
MgO	13.39	.86	13.80	.22	16.11	.52	12.19	.90	12.25	.91
sum	98.42		99.82		99.31		98.80		100.00	
N	2		8		4		13		5	
WO	46.5		47.6		38.9		49.4		48.8	
EN	41.3		40.5		49.2		37.1		37.2	
FS	12.2		11.9		11.9		13.5		14.0	

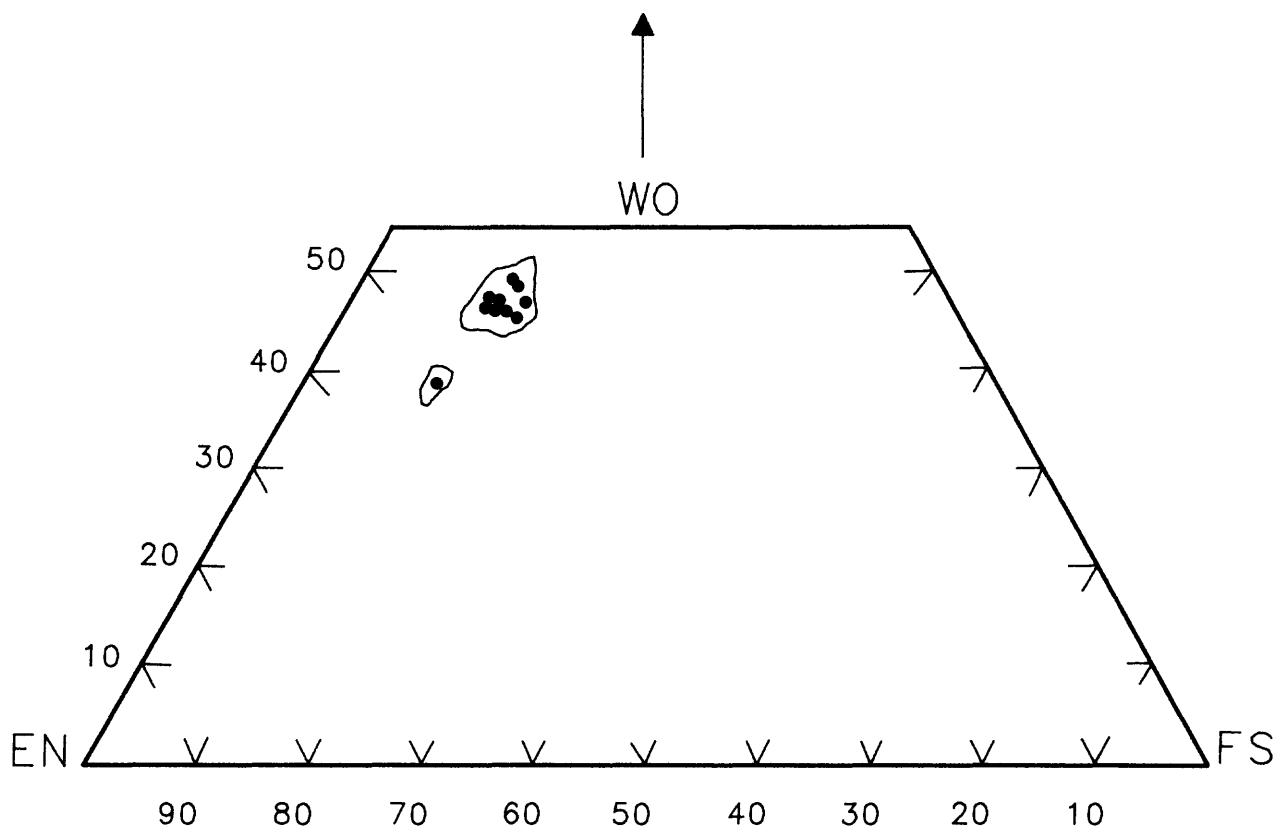


Figure 11.—Ternary diagram showing composition of clinopyroxene from As Sarat basalts. Dots show average compositions from table 7. The field boundaries show the limit of all data.

PETROGENESIS AND TECTONIC SETTING

The change from subalkaline to alkaline basalt in the As Sarat field is significant and places certain constraints on the petrogenesis of these rocks. The invariant point in the basalt experimental system crosses the plane of silica saturation (fig. 12) as the pressure at which partial melting occurs increases (O'Hara, 1968). This relationship has been used to explain the coexistence of alkaline and subalkaline basaltic rocks in a single tectonic setting. As suggested by Pallister (1987) the occurrence of both alkaline and subalkaline basalt in association with Red Sea rift may result from partial melting of the mantle at various depths. In particular, the subalkaline pre-hiatus As Sarat basalts may represent partial mantle melting at approximately 2 GPa (20 kb, 70 km), whereas the alkaline, post-hiatus As Sarat basalts may represent melting at depths greater than 70 km (fig. 12). As Sarat basalts display some compositional deviation from the olivine-clinopyroxene cotectic (fig. 12); these deviations probably represent varying amounts of olivine \pm clinopyroxene fractionation or accumulation (table 3).

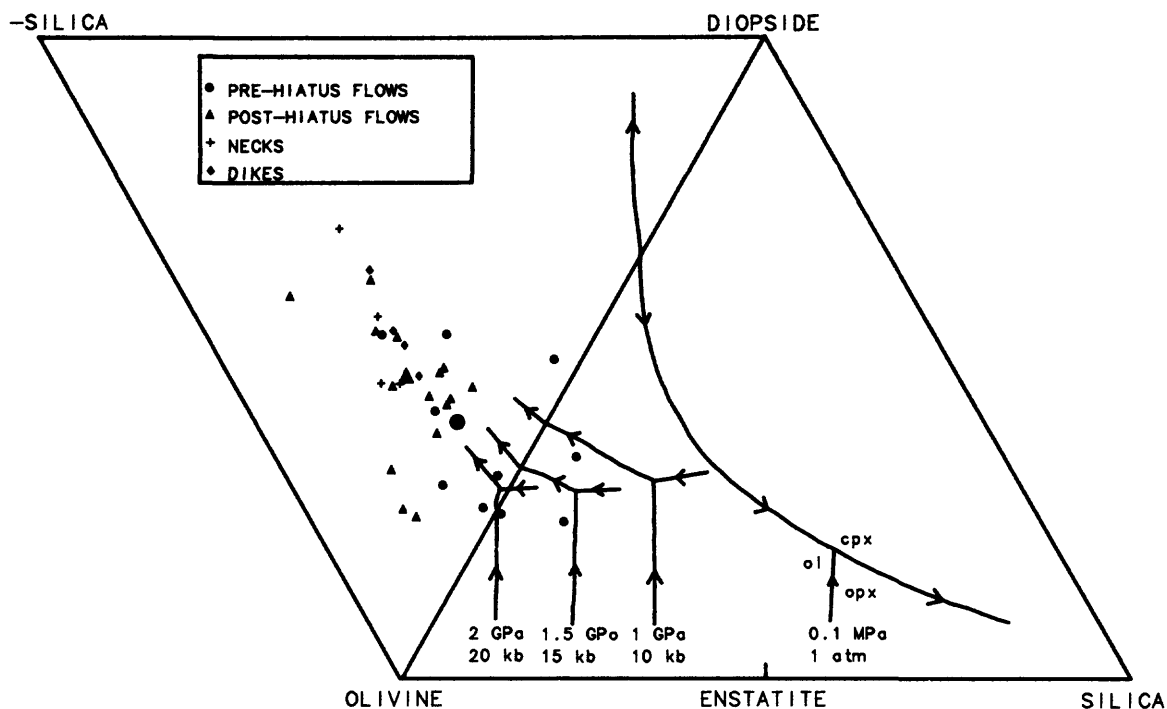


Figure 12.—Compositions of As Sarat basalt and liquidus phase relations for basaltic melts from Stolper (1980) and Walker and others (1979) projected from plagioclase on the diopside-olivine-silica saturation surface of the simplified basalt phase system using the methods of Walker and others (1979) and Pallister (1987). Enlarged plot symbols represent average compositions for rock type denoted by that symbol type.

REE abundances in pre-and post-hiatus basalt are similar, though LREE abundances in the post-hiatus basalts are almost double those in the pre-hiatus basalts, which suggests that all As Sarat basalt was produced by partial melting of chemically similar mantle material. $^{87}\text{Sr}/^{86}\text{Sr}$ ratios for three samples, from the base, middle, and top of the CRS, are 0.7034, 0.7033, and 0.7036, respectively (Coleman and others, 1977) and are consistent with oceanic mantle derivation (Faure, 1986). The absence of Nb anomalies, considered indicators of asthenospheric mantle partial melting (Thompson, 1986), in As Sarat basalts (fig. 9) suggests that they were generated from lithospheric mantle. Mantle-like initial strontium ratios and the absence of continental crustal xenoliths suggest little or no contamination of As Sarat basalt by evolved upper crustal rocks of the Precambrian basement.

Process identification diagrams use the geochemical behavior of incompatible and extremely incompatible elements to help determine whether partial melting, fractional crystallization, or a combination of these processes had dominant roles in basalt compositional evolution (Allegre and Minster, 1978; Minster and Allegre, 1978). Trace element abundance interrelationships in samples of As Sarat basalt (pre-and post-hiatus, necks and dikes) define steep linear regression lines (fig. 13

A-E) that trend toward the origin on these diagrams. Allegre and Minster (1978) and Minster and Allegre (1978) suggest that these characteristics indicate batch partial melting. Linear data arrays, (fig. 13 B, C in particular) on these plots indicate that As Sarat magmas experienced minimal subsequent crystal fractionation or accumulation. Increases in the abundances of incompatible elements, such as Ba and Zr, that accompany relatively minor MgO decreases indicate that varying degrees of partial melting played a dominant role in development of As Sarat compositional variation (table 2).

The major role of partial melting in the genesis of the As Sarat magmas is in contrast to processes identified as being responsible for compositional variation in volcanic rocks of Harrat Rahat (Camp and Roobol, 1989), the only other well studied Saudi Arabian Tertiary volcanic field. Composition data for rocks of Harrat Rahat indicate that partial melting was an essential but not the sole factor responsible for their compositional variation. Camp and Roobol (1989) suggest that Harrat Rahat magmas also experienced open-system fractionation. The age difference between Harrat Rahat rocks (10 Ma-present; Camp and Roobol, 1989) and As Sarat basalt (31-22 Ma), and the fact that different petrologic processes are inferred to have operated in the two systems suggest that the tectonic processes (rifting) responsible for extrusion of these rocks may also have been significantly different in the two areas.

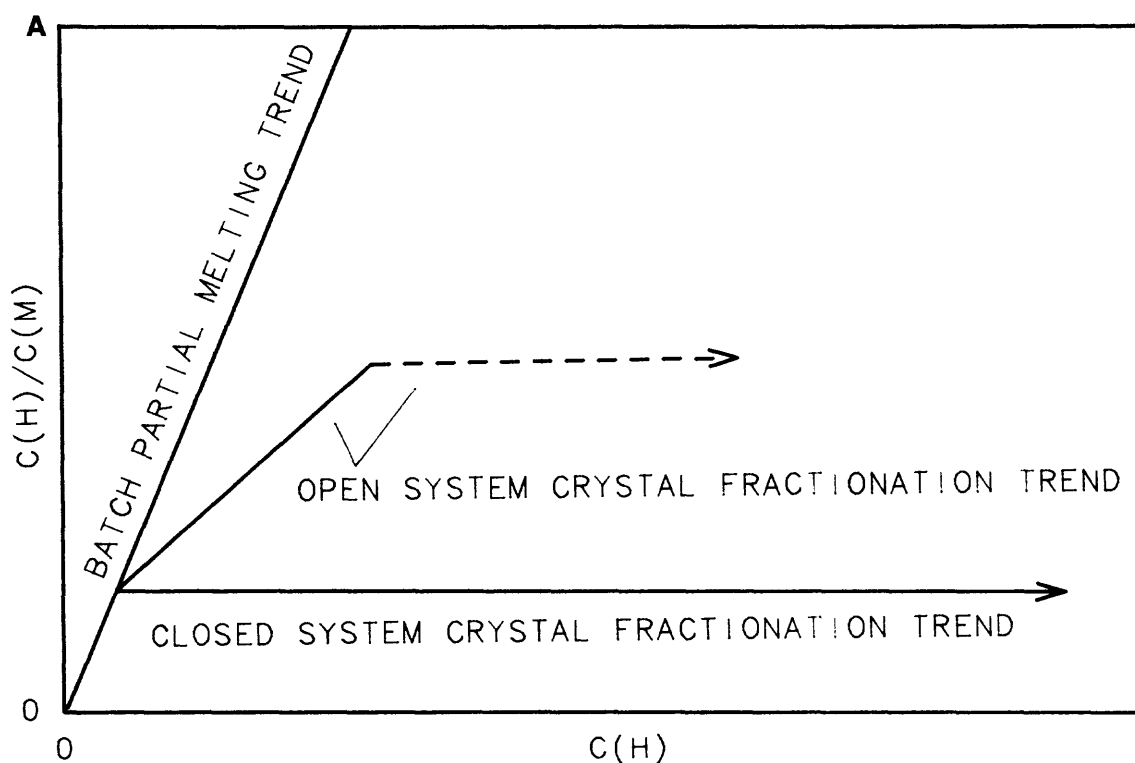


Figure 13.—Process identification trace element variation diagrams. **A**, theoretical, variation trends for different magmatic process (trend lines after Allegre and Minster, 1978); $C(M)$ and $C(H)$ are abundances of magmatophile and hypermagmatophile elements, respectively. **B-E**, variation diagrams for As Sarat basalt; solid lines are linear regression lines. Dashed fractionation trend lines, in **D** and **E**, from Camp and Roobol (1989) and Feigenson and others (1983), respectively.

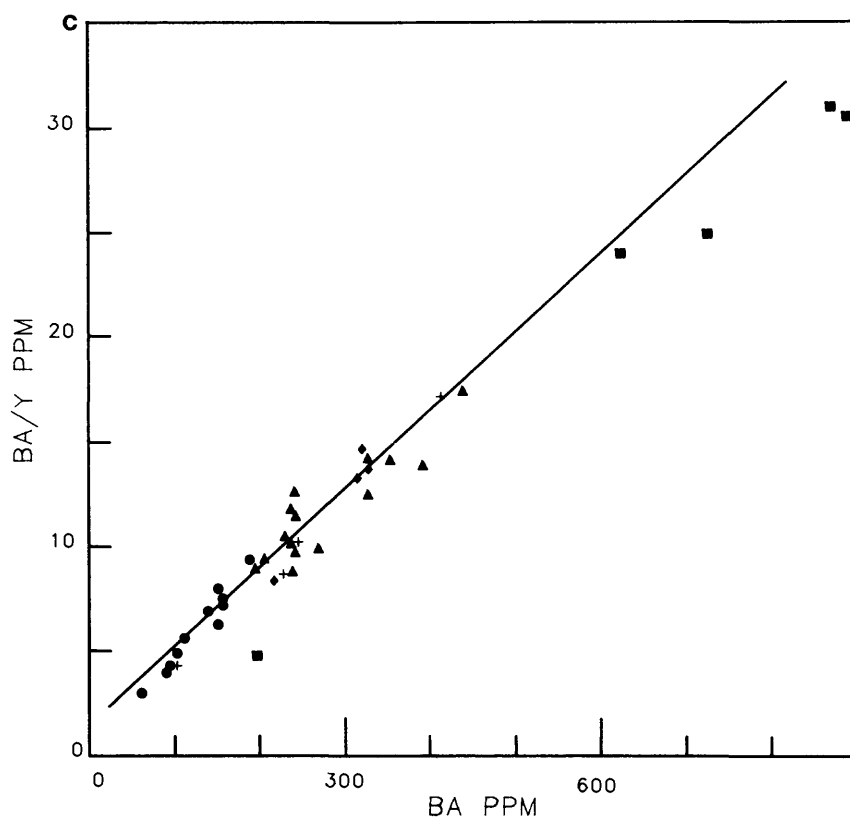
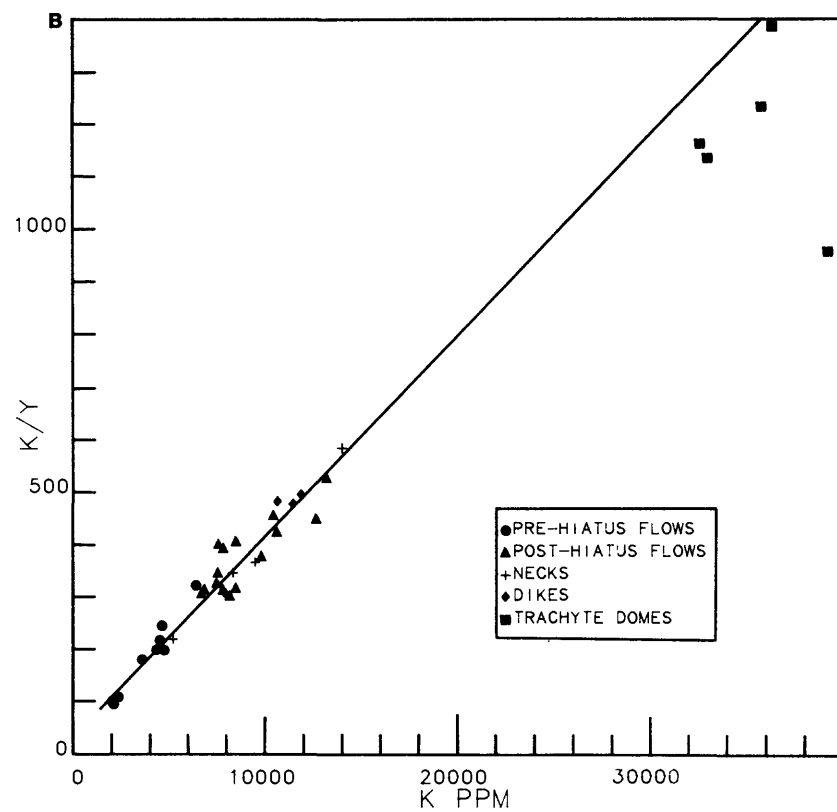


Figure 13.—Process identification trace element variation diagrams—Continued

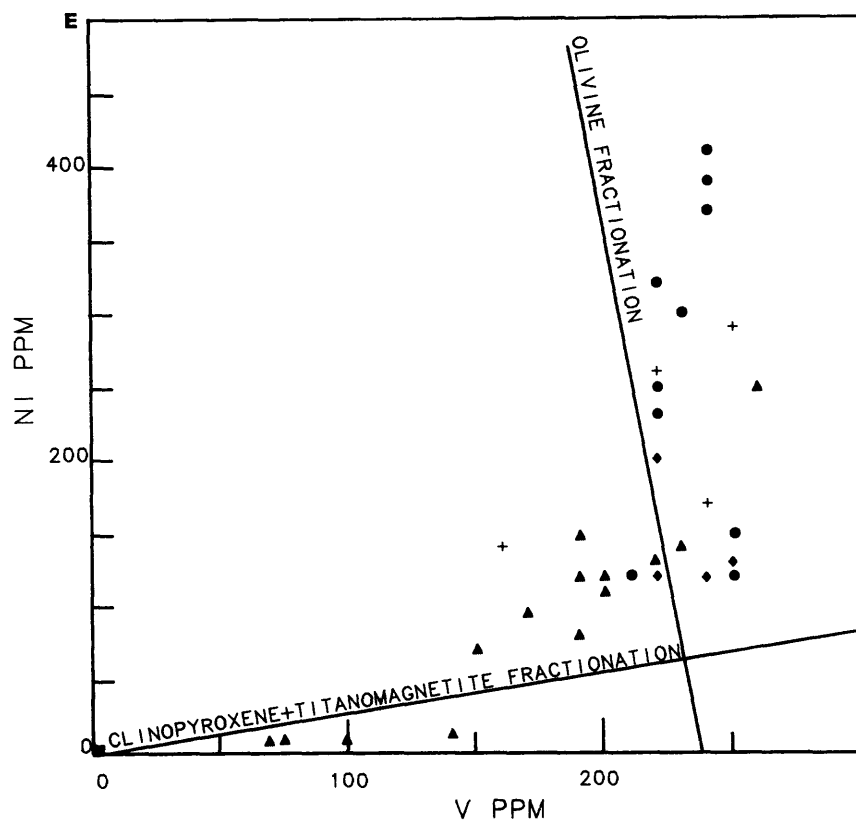
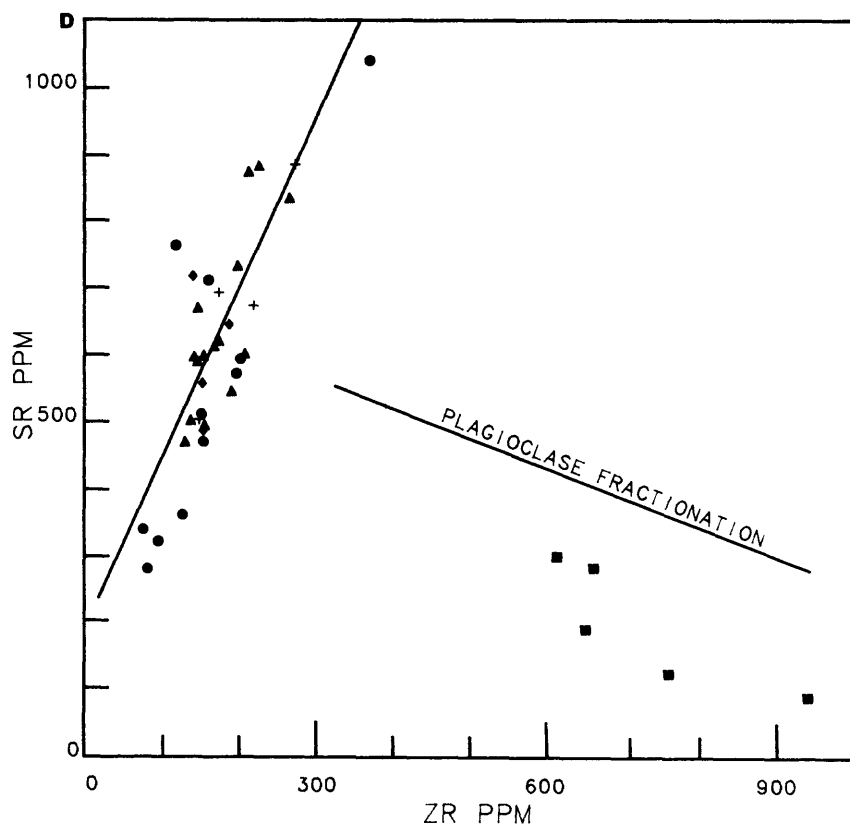


Figure 13.—Process identification trace element variation diagrams—Continued

Trace element abundances can be used to infer degrees of partial melting for a given igneous lithology. Given that bulk distribution coefficients for strongly incompatible trace elements are extremely low, the expression for Rayleigh melting reduces to $C_l/C_0 = 1/F$, where C_l is the abundance of a trace element in the liquid, C_0 is its abundance in the source, and F is the fraction of partial melting (Shaw, 1970). Using this relation and assuming a mantle source that contains 1 ppm Rb (Frey and others, 1978), calculated partial melt percentages for the pre- and post-hiatus basalts, necks, and dikes are 8.3, 4.6, 4.0, 3.9, respectively. Assuming a mantle source that contains 15.5 ppm Zr (Frey and others, 1978), the same calculated partial melt percentages are 9.9, 8.8, 7.6, and 9.9, respectively. Post-hiatus basalt appears to represent a smaller percentage of partial melting than pre-hiatus basalt. The basalt necks and, less certainly, the dikes were generated by quantities of partial melting most similar to those characteristic of post-hiatus basalt.

REE abundances characteristic of the pre- and post-hiatus basalts are a function REE distribution coefficients. As suggested above, the post-hiatus basalts may represent a smaller degree of partial melting, which maximizes REE liquid abundances because the REE have small distribution coefficients in the environment of basalt genesis (Arth, 1976). The LREE are particularly incompatible in this environment (Arth, 1976) so that relatively smaller amounts of partial melting, such as that responsible for the post-hiatus basalts, will yield relatively LREE enriched liquids. Very modest positive Eu anomalies characteristic of the As Sarat basalts may result from the addition of a minor amount of feldspar (Hanson, 1980).

Polybaric crystal fractionation and(or) addition of phases had a minor role in development of As Sarat compositional variation. Their roles are most apparent in comparisons of As Sarat basalt compositions to compositions along the clinopyroxene-olivine cotectic (fig. 12). As Sarat basalts exhibit considerable deviation from cotectic compositions; these deviations are inferred to result from small amounts of olivine and(or) clinopyroxene fractionation. Covariation of major oxide and trace element abundances (table 3) substantiate the inference that some of the compositional variation is attributable to olivine and(or) clinopyroxene fractionation. Variation of some trace element abundances such as Ni and V (fig. 13E) also indicate that crystallization of olivine, clinopyroxene, and iron-titanium oxides contributed to As Sarat compositional variation. These features must be placed in the perspective offered by petrographic observations, however. Specifically, cumulate textures, megacrysts, and reaction textures have not been observed in As Sarat basalt, which implies that neither crystal fractionation nor accumulation occurred on a major scale.

The trachytes of the As Sarat field appear to have evolved from basalt by extensive fractionation, probably within a basaltic magma chamber. Plagioclase fractionation is well demonstrated by the large negative Eu anomaly that characterizes the trachyte REE pattern (fig. 8). Likewise, abundances of Sr and Zr (fig. 13D) indicate that plagioclase fractionation played a major role in derivation of

the trachytes from the basalt. Middle REE depletion in the trachyte may be a consequence of apatite fractionation; apatite is characterized by relatively elevated partition coefficients for the middle REE (Hanson, 1980). Compositional variation among the trachytes may be controlled by additional feldspar fractionation in addition to clinopyroxene and iron-titanium oxide fractionation (fig. 13D). Note that a large composition gap, between basalt and trachyte, exists in the As Sarat magmatic series for which intermediate members are not presently known, although several unsampled small "andesitic" intrusions mapped by Anderson (1979) in the southern part of the field may represent such rocks.

A tabulation of ages (Coleman and others, 1983; fig. 15) for spatially associated rocks of the Red Sea and Dead Sea rifts indicates that these volcanic rocks are approximately coeval. These ages are bimodally distributed which suggests that spreading at the Red Sea rift has been a two-stage process as first proposed by Girdler and Styles (1974). The first stage of magmatism began about 32 Ma and the second about 5 Ma. Volcanism during the first period was widespread and voluminous and led to the formation of the Trap Series or plateau volcanics that occur mainly in Ethiopia and Yemen, with minor extensions into Somalia and Saudi Arabia. In Ethiopia the bulk of the basaltic rocks were erupted between 32-15 Ma and 4.5-0 Ma (Jones, 1976; Mohr, 1983). The main Trap Series of the southern Yemen Arab Republic was formed in two stages between 30 and 19 Ma, with a hiatus between 26 and 23 Ma (Civetta and others, 1978). In northern Yemen and southern Saudi Arabia, the early basalts formed between 31 and 22 Ma (Coleman and others, 1983; Capaldi and others, 1983; present paper); evidence presented here indicates a hiatus at about 25-24 Ma. The appearance of basalt over a widespread area at about the same time (i.e. 31-30 Ma) suggests that these rocks, including the As Sarat basalts, formed during crustal thinning along the flanks of a zone of crustal extension in the southern Red Sea region (Coleman and others, 1983; Kellogg and Reynolds, 1983).

First stage rifting is approximately coeval with pre-hiatus As Sarat basalt, which was erupted between 31.1 and 25.2 Ma. Hempton (1987) and Bohannon and others (1989) suggest that spreading along the Red Sea rift ceased in response to a collision involving the Arabian and Eurasian plates. However, after 24 Ma magmatism continued in the region for another 6 m.y. The period between spreading episodes along the rift represents a time of regional plate reorganization in the Middle East that began about 25 Ma; renewed spreading did not begin again along the axis of the Red Sea rift until this reorganization was complete (about 5 Ma) and regionally coherent mantle flow was reestablished (Hempton, 1987). The effects of reorganization may have been complete somewhat earlier further south in the Ethiopia-Yemen region, however, as indicated by 10 Ma-old basalts. The chemical discontinuity between pre- and post-hiatus As Sarat basalt implies a change in the nature of magma genesis that is coincident with the beginning of this period of postulated regional plate reorganization.

DATA STORAGE

All results obtained in this study are contained in this report therefore no Data File was established.

No updated information was added to the Mineral Occurrence Documentation System (MODS) data bank, and no new files were established.

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