

**U.S DEPARTMENT OF THE INTERIOR
U.S. GEOLOGICAL SURVEY**

**PARTITIONING OF MINOR ELEMENTS AND MAJOR-
ELEMENT OXIDES BETWEEN ROCK COMPONENTS
AND CALCULATION OF THE MARINE DERIVED
FRACTION OF THE MINOR ELEMENTS IN ROCKS
OF THE PHOSPHORIA FORMATION, IDAHO AND
WYOMING**

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Abstract

Major-element-oxides from five sections of the Phosphoria Formation -- Mud Spring, Lakeridge, Wheat Creek, Fontenelle Creek, and Conant Creek -- are partitioned into normative-mineral components in order to determine the marine versus terrigenous fractions of the rocks. The components are apatite, dolomite, calcite, biogenic silica and organic matter, the marine-derived fraction of the samples, and a detrital component, the terrigenous fraction of the samples. The same proportionality constants are used in the calculation of each component for all the sections. The average sum of the components is 99.6 percent with an average standard deviation of 2.7 percent, demonstrating that the components have a relatively constant composition. The coherence of the normative scheme was also checked by comparing measured inorganic CO₂ to calculated CO₂; correlation coefficients were 0.99 or greater for all but one of the sections.

Assuming aluminum is entirely of terrigenous origin, X-Y plots of aluminum versus Ba, Li, Ga, and Sc, which show good correlations, demonstrate a relatively constant minor element content in the terrigenous fraction. The terrigenous fraction of the minor elements is determined primarily from a sample composed almost entirely of the terrigenous detrital component. There were exceptions where the terrigenous minor-element fraction was determined by the minor element content of a standard shale or by the minor-element/Al₂O₃ ratios extracted from a well defined minimum line formed on minor-element/Al₂O₃ graphs. The marine fraction of minor elements is the difference between bulk minor-element content of the samples and their terrigenous contribution of minor elements. Positive correlations exist between the marine fraction of several minor elements and the components apatite and organic matter, further validating the normative calculation.

INTRODUCTION

Over the past few decades oceanographers have examined the behavior of minor elements in seawater and sediment from different geochemical environments of the modern ocean (Elderfield, 1970; Bertine and Turekian, 1973; Bruland, 1983; Landing and Bruland, 1987; Jacobs and others, 1987; Piper, 1988; Anderson and others, 1989; Shaw and others, 1990; Emerson and Huested, 1991; Bruland and others, 1991). From this research the marine minor elements have been used to interpret the depositional environments of ancient sedimentary deposits (Piper, 1991; Piper and Medrano, 1994; Piper and Isaacs, 1995). The Phosphoria Formation is enriched in a number of the marine derived minor elements because of the low terrigenous sedimentation rate ($0.2-0.5 \text{ mg/yr/cm}^2$) and a large marine fraction, composed mainly of chert, apatite, calcite or/and dolomite, and organic matter. In this bulletin the marine fraction of minor elements and their *host* phases are calculated from major- and minor-element analyses of rock samples from the Phosphoria Formation using a normative procedure (described in detail in Medrano and Piper, 1992). In the next bulletin, the second part of this study, the *source* of the marine fraction of minor elements and their depositional environment will be discussed.

The normative procedure involves calculating the percentage of each mineral component in each sample. The components of the Phosphoria Formation are apatite, calcite, dolomite, organic matter, biogenic silica, and terrigenous detritus. Unlike the typical normative calculation these components, or minerals, are not fictitious. All have been identified in these rocks by X-ray diffraction and in thin sections using a petrographic microscope (Gulbrandsen, 1960a,b; McClellan and Lehr, 1969; Murata and others, 1972). All the components are composed of a single mineral, except the terrigenous detrital component and organic matter. The terrigenous detrital component is a multi-mineral component, composed of detrital quartz, clay minerals, feldspars, mica, and accessory minerals like zircon and rutile (Gulbrandsen, 1960a). After the

components are quantitatively determined, they are partitioned into either the marine and terrigenous fractions.

The minor-element content in the terrigenous fraction is determined from minor element/ Al_2O_3 graphs. The marine minor element content of the rocks is calculated as the difference between the terrigenous minor-element content and the bulk analyses. Key to calculating the marine minor elements is: 1) evaluating the quality of the initial data; 2) evaluating the constancy of composition for components at each section; 3) determining compositional variations of components between sections; 4) determining "proportionality constants" for each rock component; 5) evaluating the terrigenous fraction for constant minor-element composition; and 6) determining minor element/terrigenous fraction proportionality constants.

GEOLOGIC SETTING AND SAMPLING

Geologic Setting

The Phosphoria Formation, deposited in an interior sag basin (Wardlaw and Collinson, 1986) located along the western margin of the North American craton, consists of carbonaceous and phosphatic mudstone, phosphorite with minor amounts of carbonates, and dark gray chert (McKelvey, 1959). It was deposited over approximately 10 m.y., from the middle of the Leonardian to the middle of the Guadalupian stages in Permian time (Wardlaw and Collinson, 1984; Murchey and Jones, 1992). The basin is located in the eastern half of Idaho, western Wyoming, northern Utah, northeastern Nevada and southwest Montana (fig. 1). Paleomagnetic data indicate that at the time of deposition the basin lay 3° to 9° N of the equator (Sheldon, 1964).

Sample selection.

Rock samples from Conant Creek, Wheat Creek, Fontenelle Creek, Lakeridge, and Mud Spring (fig. 1) are described in U.S. Geological Survey publications (Sheldon and others, 1953; Sheldon and others, 1954; Smart and others, 1954; Sheldon, 1963).

Stratigraphic sections, showing sample positions, were constructed from these reports (fig. 2). These sites are located on an east-west transect of the area of deposition. Conant Creek, the easternmost site, is mainly a carbonate facies, possibly of shallow-water origin. Mud Spring, more than 500 km to the west, is a mudstone facies, possibly originating in deeper water (McKelvey 1959). The sections in between consist of these plus major phosphate and chert facies.

The Conant Creek, Fontenelle Creek, Wheat Creek, and Mud Spring samples are from surface outcrops or trenches and include mainly the Meade Peak Phosphatic Shale Member, with only a few samples from the Retort Phosphatic Shale Member and the Rex Chert Member, all of the Phosphoria Formation. The analyzed powders for these sections are aliquots of original ground powders (to 200 mesh). The samples from the Lakeridge section are from a core collected from a well deeper than 4200 m (Murata and others, 1972). Samples of Lakeridge core include all the members of the Phosphoria Formation, as well as interbedded units of the Park City Formation and Shedhorn Sandstone. (fig. 2C).

Figure 1. Locality map, showing sections sampled in this study, marked by small solid squares. Map is adapted from Yochelson (1968).

Figure 2. Stratigraphic columns of the Phosphoria Formation showing samples analyzed: Conant Creek (A), Wheat Creek (B), Fontenelle Creek (B), Mud Spring (C) and Lakeridge (C). Compiled from Smart and others, 1954, Sheldon 1963, Sheldon and others, 1953, Sheldon and others, 1954. Legend is given in figure 2A.

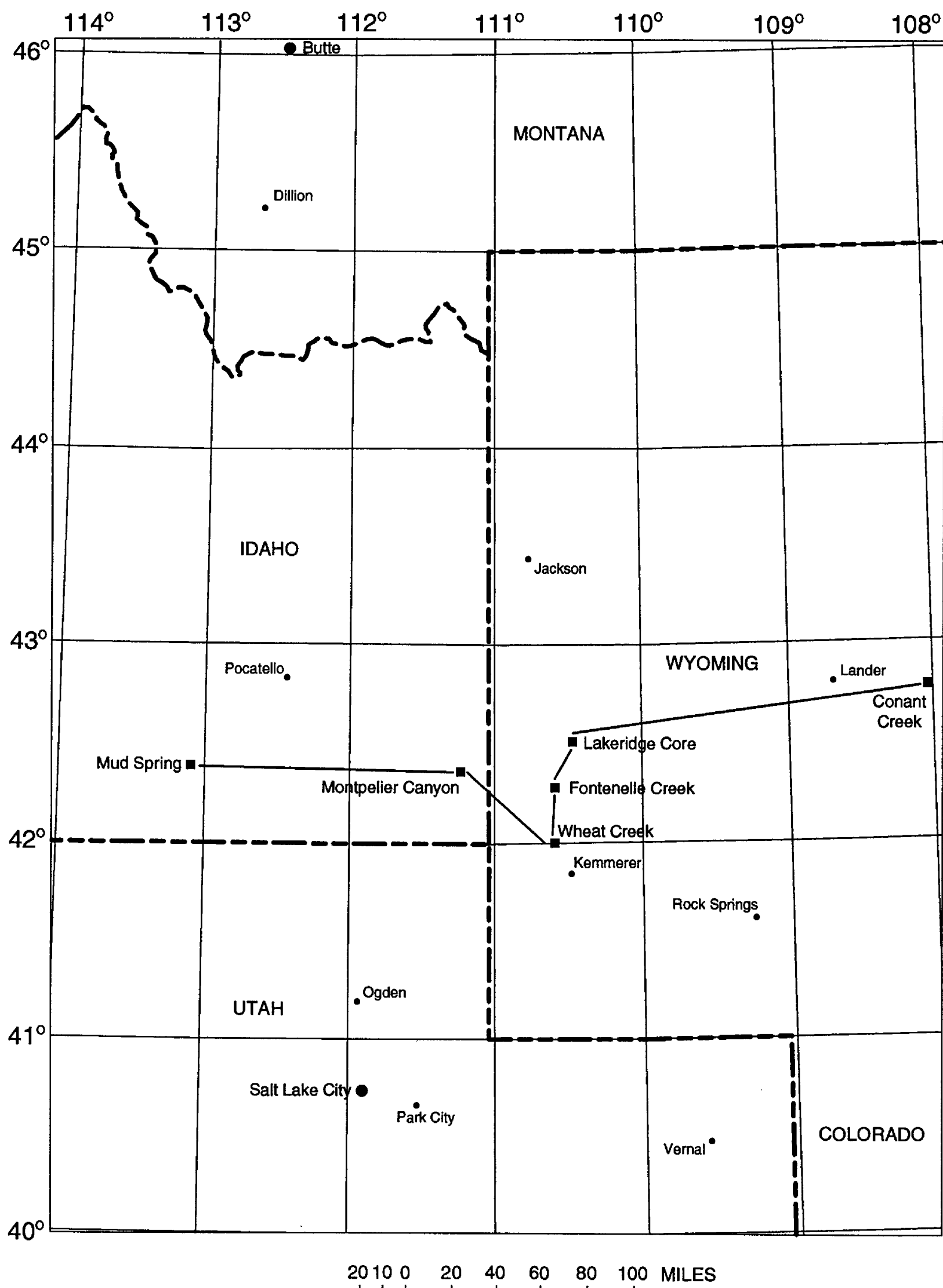


Figure 1

Conant Creek

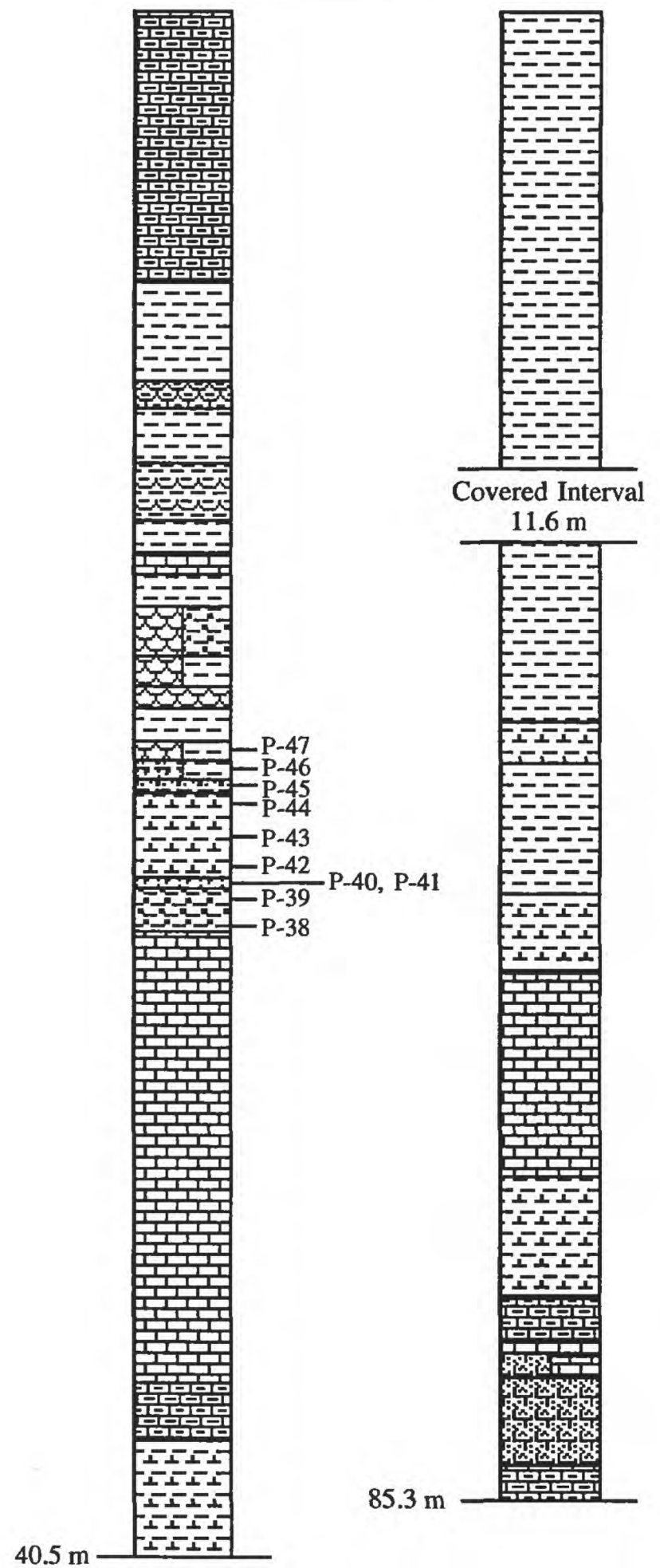
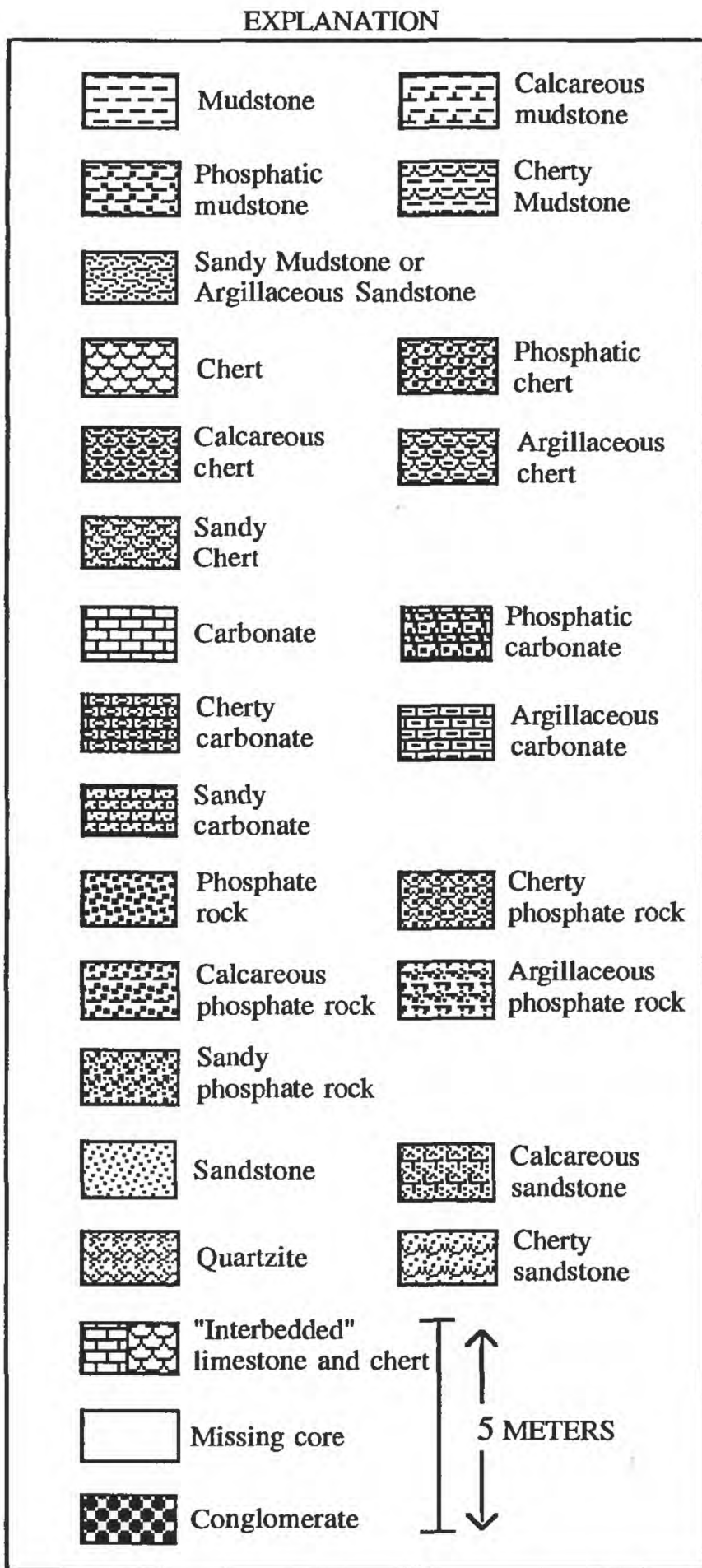
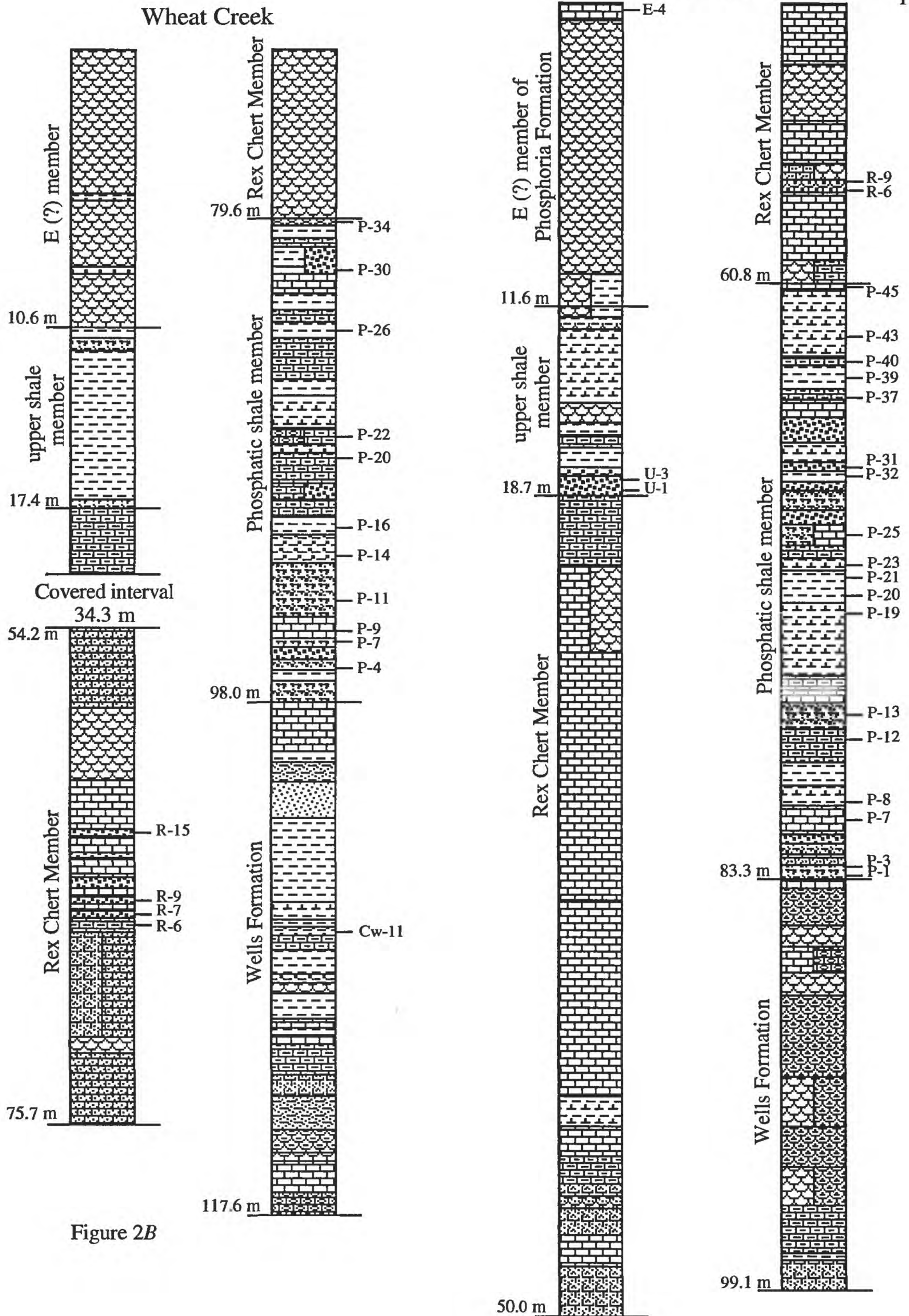


Figure 2A



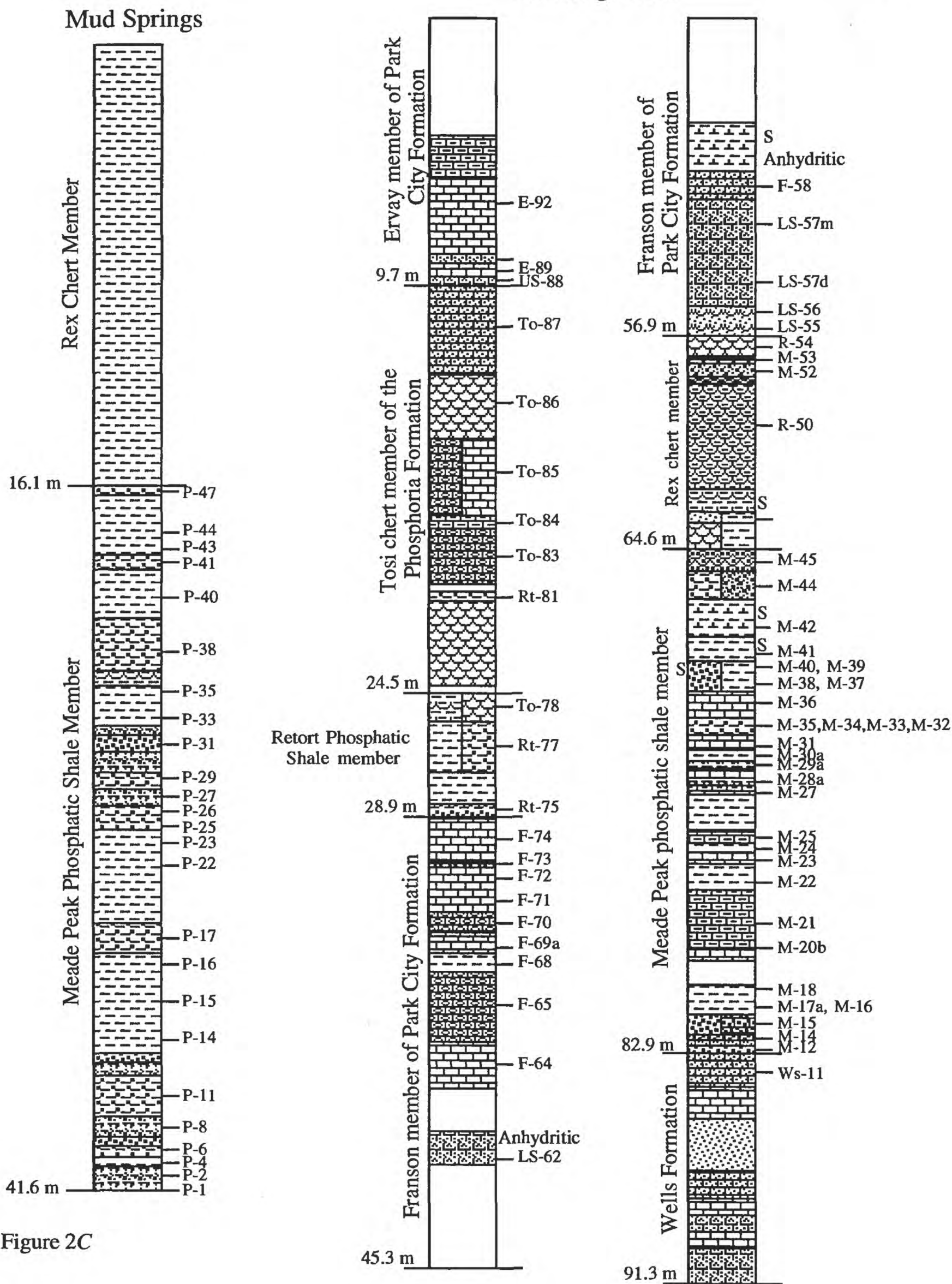


Figure 2C

ANALYTICAL TECHNIQUES AND EVALUATION OF THE DATA

Major-element-oxide analyses

Major-element oxides, except SiO_2 , were measured by inductively coupled plasma-atomic emission spectroscopy (ICP-AES) after acid digestion (Lichte and others, 1987a). Selected samples were analyzed by wavelength-dispersive X-ray fluorescence spectroscopy (XRF; Taggart and others, 1987). Loss on ignition (LOI) at 925°C was measured on the samples which were analyzed by XRF. The limit of detection and the precision and accuracy of each analytical technique were determined by repeated analysis of rock standards by the Branch of Geochemistry and are given in table 1. Wet chemical analyses (for P_2O_5 only), originally reported in U.S. Geological Survey Circulars 307, 325, 327 (Sheldon and others, 1953; Sheldon and others, 1954; Smart and others, 1954), are listed with the ICP-AES and XRF data in table 2.

Evaluating the quality of the major-element data is important because the major element data are used to determine the normative components and with each calculation any error in the original data is compounded. Accuracy of an analysis is its deviation from its true value. Unfortunately the true value cannot be known. What is known is the "accepted value," which represents the mean of all values from several different analytical techniques. The U.S. Geological Survey Laboratory has determined accuracy for its analytical techniques by comparing its analyses with "accepted values" (table 1).

We have further evaluated the quality of these data by comparing analyses of individual elements made by more than one technique. Ideally, the analytical results of multiple techniques should agree closely. For our data, X-Y plots of individual elements, measured by two techniques, show agreement within 5 percent of the one-to-one line in 75 percent of the cases (fig. 3). This procedure does not identify which technique is in error; rather it establishes that a finite systematic error of some percent exists in one or the other, or possibly both analyses. Analytical results for P_2O_5 (fig. 3A), measured by

wet chemistry and XRF (Mud Spring, Conant Creek, and Lakeridge sections) are in excellent agreement. In contrast, the ICP-AES data for the Wheat Creek and Fontenelle Creek sections are higher by approximately 13 percent and 9 percent, respectively, than the XRF data. These deviations suggest a systematic error in the ICP-AES data for a few elements.

Precision of an analytical technique is defined by the deviation of a set of determinations from their mean (table 1). The X-Y plots also provide a measure of precision (fig 3). Whereas accuracy is assumed to be represented by the approach of the correlation line to a slope of one, precision is evaluated by the degree of correlation to the line of best fit, i.e., the correlation coefficient. This is a measure of the combined precision of the two techniques. For major-element oxides used in the normative calculation (Al_2O_3 , MgO , CaO , P_2O_5) the correlation coefficients are $R \geq 0.99$ (fig. 3). The precision for the ICP-AES and the XRF data is lower than the maximum reported by the U.S. Geological Survey Laboratory (table 1).

We use the ICP data for the calculation of the components because it is the technique used to measure the minor elements and we want to be consistent by using the same data set within each section and between sections. The exception to this is the following: SiO_2 , not analyzed by ICP, is measured by XRF. In the case of Lakeridge, SiO_2 data is not available for all the samples, so they are calculated from the difference between the sum of all the major elements, plus a LOI term, and 100 percent.

Table 1. Detection limits, accuracy and precision for elements analyzed by the different analytical techniques were determined by repeated analysis of rock standards by the Branch of Geochemistry (Arbogast and others, 1990; Baedeker and others, 1987; Lichte and others, 1987).

Analytical Technique	Detection Limit	Accuracy	Precision
ICP-AES	Limits of detection, in weight percent, in parentheses: Al ₂ O ₃ (0.09), CaO (0.07), Fe ₂ O ₃ (0.07), K ₂ O (0.1), MgO (0.08), Na ₂ O (0.1), P ₂ O ₅ (0.02), and TiO ₂ (0.02). Limit of detection in parts per million in parentheses: Ag (2), As (10), Au (8), Ba (1), Be (1), Bi (10), Cd (2), Ce (4), Co (1), Cr (1), Cu (1), Eu (2), Ga (4), Ho (4), La (2), Li (2), Mn (10), Mo (2), Nb (4), Nd (4), Ni (2), Pb (4), Sc (2), Sn (10), Sr (2), Ta (40), Th (4), U (100), V (2), Y (2), Yb (1), and Zn (2).	Standards analyzed varied 0.5 to 10 % from the proposed value	±2 to 10 percent for concentrations greater than 10 times the lower limit of detection
XRF	Concentration Range SiO ₂ 0.10% to 99.0% Al ₂ O ₃ 0.10% to 28.0% Fe ₂ O ₃ 0.04% to 28.0% MgO 0.10% to 60.0% CaO 0.02% to 60.0% Na ₂ O 0.15% to 30.0% K ₂ O 0.02% to 30.0% TiO ₂ 0.02% to 10.0% P ₂ O ₅ 0.05% to 50.0% MnO 0.01% to 15.0% LOI 0.01% to 100.0%	Standards analyzed varied 0.5 to 3 % from the proposed value	Precision is better than ±5 percent and, depending on the element, as low as ±0.2 percent
ICP-MS	Limits of detection in parts per million: La (0.002), Ce (0.002), Pr (0.002), Nd (0.009), Sm (0.006), Eu (0.003), Gd (0.011), Tb (0.002), Dy (0.007), Ho (0.002), Er (0.007), Tm (0.002), Yb (0.006), Lu (0.002).	Accuracy similar to ICP-AES and neutron activation analysis	The average RSD for the REE is 2.5%. They range from 1.7% to 5.1%.
Total C by combustion	Concentration Range C 0.05% to 30%	Standard: MAG-1, \bar{X} = 2.28%, n = 12, s = 0.009%, proposed value: 2.15±0.40%	RSD is ≤ 5% for samples with > 0.01% C or an absolute standard deviation of 0.05% C, whichever is greater.
Total S by combustion	Concentration range S 0.05% to 30%	Standard: SDO-1, \bar{X} = 5.44%, n = 10, proposed value: 5.35±0.44%	RSD for standard is 0.2%
Carbonate carbon by coulometric titration	Concentration range CO ₂ 0.01% to 50.0%	Standard: GSP-1, \bar{X} = 0.10±0.003%, n = 12, proposed value: 0.11%	RSD is < 5% for concentration range of 0.01% to 36%.
F by ion-selective electrode	Concentration range F 100ppm to 2.7%	Standard: GSP-1, \bar{X} = 3240 ppm, n = 23, proposed value: 3630 ppm	RSD for standard is 11%.

Table 2. Major-element-oxide contents in samples of the Phosphoria Formation from Conant Creek, Wheat Creek, Fortenelle Creek, Lakeridge, Wyoming, and Mud Springs, Idaho.

[All values in weight percent, as determined by inductively coupled plasma spectroscopy (I), X-ray-fluorescence spectroscopy (X), and wet chemical analysis (W). LOI, loss on ignition]

Conant Creek											Mud Springs							
Sample-----	P-47	P-46	P-45	P-44	P-43	P-42	P-41	P-40	P-39	P-38	P-47	P-44	P-43	P-41	P-40	P-38	P-35	
SiO2-----	62.9 (X)	37.4 (X)	26.7 (X)	40.8 (X)	38 (X)	38.7 (X)	39 (X)	55.5 (X)	40 (X)	27.1 (X)	47 (X)	69.8 (X)	62.3 (X)	50.4 (X)	61.1 (X)	47.7 (X)	62.9 (X)	
Al2O3-----	2.94 (I)	3.98 (I)	2.15 (I)	2.91 (I)	5.57 (I)	7.14 (I)	2.88 (I)	3.62 (I)	4.44 (I)	4.71 (I)	10.77 (I)	8.79 (I)	11.00 (I)	7.98 (I)	11.09 (I)	10.85 (I)	7.01 (I)	
Fe2O3-----	2.58 (X)	3.6 (X)	1.8 (X)	2.48 (X)	4.87 (X)	8.38 (X)	2.28 (X)	3.35 (X)	3.93 (X)	4.14 (X)	10.3 (X)	8.45 (X)	10.8 (X)	7.44 (X)	10.8 (X)	10.3 (X)	6.59 (X)	
	2.89 (I)	1.97 (I)	1.26 (I)	2.06 (I)	2.25 (I)	2.73 (I)	2.23 (I)	3.13 (I)	2.35 (I)	2.35 (I)	5.03 (I)	2.93 (I)	3.58 (I)	2.93 (I)	4.58 (I)	3.69 (I)	2.32 (I)	
MgO-----	2.81 (X)	2.04 (X)	1.22 (X)	2 (X)	2.12 (X)	2.61 (X)	2.22 (X)	2.92 (X)	2.27 (X)	2.26 (X)	4.99 (X)	2.87 (X)	3.55 (X)	2.92 (X)	4.8 (X)	3.8 (X)	2.27 (X)	
	2.59 (I)	3.50 (I)	6.01 (I)	9.15 (I)	9.13 (I)	7.92 (I)	1.54 (I)	6.54 (I)	5.98 (I)	7.34 (I)	0.71 (I)	0.71 (I)	0.85 (I)	0.40 (I)	0.90 (I)	0.80 (I)	0.60 (I)	
CaO-----	2.41 (X)	3.44 (X)	5.83 (X)	8.44 (X)	8.48 (X)	7.71 (X)	1.47 (X)	6.12 (X)	5.75 (X)	7.21 (X)	0.73 (X)	0.72 (X)	0.86 (X)	0.45 (X)	0.92 (X)	0.83 (X)	0.63 (X)	
	13.2 (I)	23.2 (I)	30.2 (I)	20.0 (I)	18.3 (I)	18.9 (I)	26.7 (I)	12.7 (I)	20.7 (I)	25.6 (I)	11.81 (I)	4.05 (I)	4.28 (I)	16.24 (I)	6.59 (I)	14.84 (I)	10.82 (I)	
Na2O-----	12.8 (X)	24 (X)	30.6 (X)	19.3 (X)	17.8 (X)	18.5 (X)	26.7 (X)	12 (X)	20.5 (X)	25.6 (X)	11.8 (X)	3.89 (X)	4.23 (X)	16.4 (X)	6.56 (X)	14.9 (X)	10.7 (X)	
	0.27 (I)	0.61 (I)	0.89 (I)	0.20 (I)	0.30 (I)	0.26 (I)	0.77 (I)	0.14 (I)	0.50 (I)	0.57 (I)	0.36 (I)	0.45 (I)	0.41 (I)	1.20 (I)	0.57 (I)	0.54 (I)	0.31 (I)	
K2O-----	0.28 (X)	0.61 (X)	0.65 (X)	0.26 (X)	0.35 (X)	0.33 (X)	0.75 (X)	0.19 (X)	0.51 (X)	0.8 (X)	0.43 (X)	0.48 (X)	0.46 (X)	1.21 (X)	0.82 (X)	0.57 (X)	0.35 (X)	
	1.33 (I)	1.98 (I)	0.95 (I)	1.39 (I)	2.80 (I)	3.73 (I)	1.31 (I)	2.02 (I)	2.36 (I)	2.60 (I)	1.67 (I)	2.03 (I)	2.38 (I)	1.21 (I)	2.70 (I)	2.81 (I)	1.86 (I)	
TiO2-----	1.21 (X)	1.88 (X)	0.84 (X)	1.22 (X)	2.49 (X)	3.36 (X)	1.2 (X)	1.79 (X)	2.15 (X)	2.34 (X)	1.62 (X)	1.99 (X)	2.38 (X)	1.17 (X)	2.7 (X)	2.73 (X)	1.81 (X)	
	0.06 (I)	0.07 (I)	0.02 (I)	0.07 (I)	0.13 (I)	0.18 (I)	0.02 (I)	0.10 (I)	0.12 (I)	0.13 (I)	0.13 (I)	0.22 (I)	0.25 (I)	0.08 (I)	0.25 (I)	0.13 (I)	0.08 (I)	
P2O5-----	0.12 (X)	0.16 (X)	0.08 (X)	0.14 (X)	0.25 (X)	0.33 (X)	0.12 (X)	0.21 (X)	0.23 (X)	0.21 (X)	0.38 (X)	0.51 (X)	0.54 (X)	0.32 (X)	0.57 (X)	0.48 (X)	0.31 (X)	
	4.51 (I)	11.5 (I)	15.5 (I)	2.62 (I)	3.94 (I)	3.27 (I)	17.04 (I)	1.08 (I)	8.52 (I)	9.71 (I)	10.49 (I)	3.18 (I)	4.53 (I)	11.89 (I)	4.81 (I)	10.99 (I)	8.02 (I)	
MnO-----	4.28 (X)	11.4 (X)	14.8 (X)	2.6 (X)	3.71 (X)	3.12 (X)	16.3 (X)	0.96 (X)	8.16 (X)	9.21 (X)	10.5 (X)	3.12 (X)	4.58 (X)	11.9 (X)	4.84 (X)	11 (X)	8 (X)	
	4.3 (W)	11.6 (W)	14.9 (W)	2.7 (W)	3.9 (W)	3.2 (W)	16.4 (W)	0.8 (W)	8.0 (W)	9.2 (W)	10.2 (W)	2.65 (W)	4.50 (W)	11.8 (W)	4.85 (W)	10.4 (W)	7.50 (W)	
LOI-----	0.02 (I)	0.02 (I)	0.02 (I)	0.03 (I)	0.02 (I)	0.03 (I)	0.01 (I)	0.03 (I)	0.02 (I)	0.03 (I)	0.01 (I)	0.01 (I)	0 (I)	0.01 (I)	0 (I)	0 (I)	0 (I)	
	8.54	11.4	15.3	21.9	20.7	19.9	7.16	15.2	14.5	19.2	10.3	8.5	8.64	6.17	6.16	6.6	5.11	
S-----	0.62	1.52	0.59	0.39	0.29	1.02	0.94	0.4	0.48	0.82	0.14	0.09	0.12	0.12	0.05	0.14	0.13	
C (Inorganic)	1.77	2.09	3.67	5.54	4.86	4.16	1.05	3.75	3.40	4.56	0.12	0.06	0.05	0.15	0.05	0.10	0.08	
C (Organic)	0.41	0.54	0.50	0.45	1.02	1.48	0.91	0.54	0.61	0.83	1.19	1.03	1.11	0.96	0.86	0.97	1.08	

Mud Springs																		
Sample-----	P-33	P-31	P-29	P-27	P-26	P-25	P-23	P-22	P-17	P-16	P-15	P-14	P-11	P-8	P-6	P-4	P-2	P-1
SiO2-----	58.2 (X)	18.8 (X)	38.5 (X)	38.1 (X)	41.4 (X)	58.3 (X)	64 (X)	78.9 (X)	52.6 (X)	83.5 (X)	85 (X)	87.9 (X)	59.5 (X)	32.1 (X)	24.8 (X)	91.2 (X)	51.9 (X)	25.9 (X)
Al2O3-----	11.23 (I)	3.33 (I)	4.95 (I)	5.50 (I)	7.39 (I)	7.41 (I)	6.41 (I)	6.39 (I)	10.62 (I)	8.66 (I)	3.78 (I)	3.42 (I)	6.33 (I)	4.12 (I)	3.10 (I)	1.83 (I)	6.58 (I)	2.02 (I)
Fe2O3-----	11 (X)	3.39 (X)	4.47 (X)	4.41 (X)	6.99 (X)	7.05 (X)	5.99 (X)	6.09 (X)	10.2 (X)	8.41 (X)	3.5 (X)	3.18 (X)	5.95 (X)	3.81 (X)	2.78 (X)	1.63 (X)	6.21 (X)	1.75 (X)
	4.39 (I)	1.13 (I)	1.53 (I)	1.30 (I)	2.15 (I)	2.29 (I)	1.77 (I)	2.42 (I)	3.85 (I)	2.57 (I)	2.43 (I)	2.15 (I)	1.89 (I)	1.30 (I)	1.07 (I)	1.62 (I)	1.90 (I)	1.20 (I)
MgO-----	4.44 (X)	1.4 (X)	1.46 (X)	1.37 (X)	2.17 (X)	2.29 (X)	1.73 (X)	2.31 (X)	3.87 (X)	2.54 (X)	2.32 (X)	2.01 (X)	1.87 (X)	1.35 (X)	1.05 (X)	1.5 (X)	1.88 (X)	1.22 (X)
	0.90 (I)	0.51 (I)	0.56 (I)	0.56 (I)	0.73 (I)	0.58 (I)	0.37 (I)	0.56 (I)	0.88 (I)	0.70 (I)	0.23 (I)	0.30 (I)	0.63 (I)	0.53 (I)	0.71 (I)	0.18 (I)	0.96 (I)	0.23 (I)
CaO-----	0.92 (X)	0.81 (X)	0.82 (X)	0.82 (X)	0.76 (X)	0.63 (X)	0.4 (X)	0.59 (X)	0.9	0.74 (X)	0.26 (X)	0.33 (X)	0.67 (X)	0.82 (X)	0.83 (X)	0.22 (X)	1 (X)	0.35 (X)
	6.96 (I)	37.86 (I)	27.30 (I)	35.42 (I)	22.26 (I)	13.55 (I)	11.54 (I)	3.28 (I)	12.63 (I)	9.18 (I)	2.55 (I)	1.47 (I)	13.97 (I)	30.38 (I)	35.42 (I)	1.50 (I)	17.92 (I)	36.68 (I)
Na2O-----	6.99 (X)	38.9 (X)	27.4 (X)	26.2 (X)	23 (X)	13.6 (X)	11.3 (X)	3.1 (X)	12.8 (X)	9.16 (X)	2.41 (X)	1.39 (X)	13.9 (X)	31.7 (X)	37.4 (X)	1.26 (X)	18.3 (X)	38.5 (X)
	0.36 (I)	0.55 (I)	0.70 (I)	0.92 (I)	0.82 (I)	1.82 (I)	1.67 (I)	0.47 (I)	0.41 (I)	0.34 (I)	0.14 (I)	0.09 (I)	1.01 (I)	0.80 (I)	0.59 (I)	0.07 (I)	0.49 (I)	0.50 (I)
K2O-----	0.42 (X)	0.8 (X)	0.71 (X)	0.74 (X)	0.88 (X)	1.79 (X)	1.62 (X)	0.5 (X)	0.43 (X)	0.39 (X)	0.18 (X)	0.15 (X)	1.01 (X)	0.82 (X)	0.63 (X)	0.07 (X)	0.53 (X)	0.51 (X)
	2.76 (I)	0.82 (I)	1.31 (I)	1.72 (I)	1.64 (I)	1.34 (I)	0.96 (I)	1.72 (I)	3.34 (I)	2.88 (I)	1.07 (I)	1.08 (I)	1.43 (I)	1.18 (I)	0.90 (I)	0.34 (I)	1.98 (I)	0.43 (I)
TiO2-----	2.74 (X)	0.76 (X)	1.23 (X)	1.27 (X)	1.6 (X)	1.29 (X)	0.91 (X)	1.67 (X)	3.26 (X)	2.63 (X)	1.03 (X)	1.05 (X)	1.36 (X)	1.13 (X)	0.87 (X)	0.31 (X)	1.92 (X)	0.4 (X)
	0.20 (I)	0.05 (I)	0.05 (I)	0.03 (I)	0.07 (I)	0.07 (I)	0.05 (I)	0.15 (I)	0.20 (I)	0.13 (I)	0.07 (I)	0.08 (I)	0.13 (I)	0.05 (I)	0.05 (I)	0.05 (I)	0.03 (I)	0.03 (I)
P2O5-----	0.46 (X)	0.12 (X)	0.23 (X)	0.24 (X)	0.36 (X)	0.47 (X)	0.41 (X)	0.33 (X)	0.45 (X)	0.38 (X)	0.11 (X)	0.11 (X)	0.43 (X)	0.27 (X)	0.14 (X)	0.08 (X)	0.32 (X)	0.06 (X)
	5.91 (I)	28.85 (I)	20.20 (I)	21.73 (I)	16.26 (I)	9.60 (I)	7.95 (I)	2.29 (I)	9.32 (I)	6.85 (I)	1.88 (I)	0.98 (I)	9.69 (I)	22.19 (I)	26.56 (I)	0.98 (I)	12.57 (I)	28.63 (I)
MnO-----	5.96 (X)	27.9 (X)	19.5 (X)	20.1 (X)	16.3 (X)	9.61 (X)	7.88 (X)	2.18 (X)	9.33 (X)	6.83 (X)	1.81 (X)	0.97 (X)	9.67 (X)	22.2 (X)	26.3 (X)	0.85 (X)	12.6 (X)	27.8 (X)
	5.70 (W)	28.4 (W)	19.2 (W)	19.5 (W)	16.4 (W)	9.35 (W)	7.75 (W)	2.20 (W)	9.15 (W)	6.75 (W)	1.80 (W)	0.95 (W)	9.40 (W)	21.5 (W)	27.1 (W)	0.90 (W)	12.8 (W)	28.0 (W)
LOI-----	0 (I)	0.01 (I)	0.02 (I)	0 (I)	0.05 (I)	0.06 (I)	0.52 (I)	0.01 (I)	0 (I)	0 (I)	0.02 (I)	0.01 (I)	0.06 (I)	0 (I)	0 (I)	0.02 (I)	0.02 (I)	0.01 (I)
	7.48	4.27	3.87	3.26	5.05	3.54	3.5	3.05	5	4.34	1.75	1.78	4.27	3.86	3.59	0.96	4.1	2.04
S-----	0.06	0.39	0.24	0.27	0.2	0.09	0.03	0.02	0.09	0.07	0.02	0.01	0.18	0.46	0.46	0.03	0.16	0.41
	C (Inorganic)	0.06	0.32	0.21	0.21	0.12	0.16	0.03	0.09	0.06	0.04	0.02	0.14	0.29	0.31	0.03	0.23	0.18
C (Organic)	1.05	0.77	0.59	0.48	0.45	0.36	0.33	0.32	0.39	0.41	0.36	0.34	0.41	0.54	0.59	0.20	0.49	0.31

Table 2. continued

Fontenelle Creek																		
Sample	E-4	U-3	U-1	R-9	R-6	P-45	P-43	P-40	P-39	P-37	P-32	P-31	P-25	P-23	P-21	P-20	P-19	P-13
SiO ₂	16.3 (X)	14.3 (X)	11.7 (X)	40.3 (X)	39.4 (X)	15.5 (X)	61.9 (X)	22.5 (X)	64.4 (X)	61.8 (X)	5.04 (X)	63 (X)	32.6 (X)	52.8 (X)	55.8 (X)	63.4 (X)	46.5 (X)	18.2 (X)
Al ₂ O ₃	2.65 (I)	2.84 (I)	1.34 (I)	1.15 (I)	3.4 (I)	0.04 (I)	3.02 (I)	0.09 (I)	3.4 (I)	3.97 (I)	0.72 (I)	8.51 (I)	5.29 (I)	10.21 (I)	10.02 (I)	9.45 (I)	6.05 (I)	4.73 (I)
Fe ₂ O ₃	2.63 (X)	2.61 (X)	1.29 (X)	1.24 (X)	3.09 (X)	0.73 (X)	2.87 (X)	0.65 (X)	3.4 (X)	3.9 (X)	0.64 (X)	8.26 (X)	5.13 (X)	10.2 (X)	10.4 (X)	9.22 (X)	5.86 (X)	4.55 (X)
	3.00 (I)	2.29 (I)	2.29 (I)	2.15 (I)	3.58 (I)	0.46 (I)	2.15 (I)	1.00 (I)	1.86 (I)	1.72 (I)	0.80 (I)	2.43 (I)	2.43 (I)	4.72 (I)	3.72 (I)	2.86 (I)	1.72 (I)	2.00 (I)
MgO	3.4 (X)	2.33 (X)	2.25 (X)	2.13 (X)	3.75 (X)	0.4 (X)	2.08 (X)	0.91 (X)	1.78 (X)	1.71 (X)	0.55 (X)	2.45 (X)	2.49 (X)	4.55 (X)	3.96 (X)	2.87 (X)	1.64 (X)	1.93 (X)
	0.45 (I)	0.45 (I)	3.32 (I)	6.47 (I)	0.51 (I)	18.26 (I)	5.15 (I)	14.11 (I)	3.98 (I)	4.32 (I)	0.33 (I)	0.61 (I)	4.32 (I)	1.29 (I)	0.95 (I)	0.55 (I)	7.80 (I)	2.99 (I)
CaO	0.57 (X)	0.56 (X)	3.3 (X)	6.36 (X)	0.58 (X)	16.9 (X)	4.75 (X)	13.7 (X)	3.83 (X)	4.18 (X)	0.38 (X)	0.64 (X)	4.34 (X)	1.34 (X)	1.05 (X)	0.8 (X)	7.58 (X)	3.02 (X)
	39.2 (I)	40.6 (I)	40.6 (I)	23.8 (I)	26.8 (I)	28 (I)	11.2 (I)	26.6 (I)	9.66 (I)	10.64 (I)	47.6 (I)	8.54 (I)	23.8 (I)	1.96 (I)	4.76 (I)	5.88 (I)	13.44 (I)	23.8 (I)
Na ₂ O	40.4 (X)	42 (X)	42.7 (X)	23.5 (X)	27.2 (X)	26.7 (X)	10.7 (X)	25.8 (X)	9.5 (X)	10.5 (X)	49.6 (X)	8.28 (X)	23.8 (X)	1.64 (X)	4.92 (X)	5.82 (X)	13.1 (X)	24.5 (X)
	0.95 (I)	0.63 (I)	0.54 (I)	0.28 (I)	0.58 (I)	0.05 (I)	0.14 (I)	0.14 (I)	0.14 (I)	0.14 (I)	1.49 (I)	1.62 (I)	0.47 (I)	0.12 (I)	0.42 (I)	1.19 (I)	1.30 (I)	0.43 (I)
K ₂ O	0.91 (X)	0.63 (X)	0.55 (X)	0.3 (X)	0.59 (X)	0.075 (X)	0.18 (X)	0.19 (X)	0.2 (X)	0.19 (X)	1.33 (X)	1.49 (X)	0.54 (X)	0.22 (X)	0.46 (X)	1.19 (X)	1.19 (X)	0.49 (X)
	0.82 (I)	0.70 (I)	0.44 (I)	0.31 (I)	0.82 (I)	0.12 (I)	0.80 (I)	0.12 (I)	0.92 (I)	1.13 (I)	0.24 (I)	2.04 (I)	1.44 (I)	2.88 (I)	2.76 (I)	2.40 (I)	1.44 (I)	1.32 (I)
TiO ₂	0.8 (X)	0.86 (X)	0.4 (X)	0.28 (X)	0.8 (X)	0.11 (X)	0.74 (X)	0.09 (X)	0.89 (X)	1.09 (X)	0.22 (X)	2.11 (X)	1.44 (X)	3.16 (X)	2.98 (X)	2.48 (X)	1.37 (X)	1.35 (X)
	0.07 (I)	0.07 (I)	0.07 (I)	0.05 (I)	0.13 (I)	0.01 (I)	0.08 (I)	0.01 (I)	0.12 (I)	0.12 (I)	0.03 (I)	0.33 (I)	0.17 (I)	0.51 (I)	0.45 (I)	0.42 (I)	0.17 (I)	0.15 (I)
P ₂ O ₅	0.12 (X)	0.12 (X)	0.05 (X)	0.06 (X)	0.23 (X)	0.03 (X)	0.18 (X)	0.02 (X)	0.21 (X)	0.24 (X)	0.02 (X)	0.65 (X)	0.34 (X)	0.66 (X)	0.69 (X)	0.7 (X)	0.43 (X)	0.23 (X)
	29.77 (I)	32.06 (I)	29.77 (I)	8.70 (I)	20.38 (I)	0.82 (I)	1.15 (I)	2.29 (I)	1.65 (I)	2.52 (I)	36.84 (I)	5.95 (I)	13.28 (I)	0.55 (I)	3.21 (I)	4.35 (I)	0.53 (I)	14.20 (I)
MnO	27.3 (X)	29.4 (X)	27.1 (X)	7.77 (X)	18.5 (X)	0.47 (X)	1.02 (X)	2.02 (X)	1.52 (X)	2.23 (X)	32.1 (X)	5.39 (X)	12.0 (X)	0.45 (X)	3.18 (X)	4.12 (X)	0.45 (X)	12.7 (X)
	7.4 (W)	29.2 (W)	26.7 (W)	7.8 (W)	18.4 (W)	0.5 (W)	1.1 (W)	1.9 (W)	1.5 (W)	2.2 (W)	31.8 (W)	5.3 (W)	18.8 (W)	0.2 (W)	3.2 (W)	3.9 (W)	0.7 (W)	14.4 (W)
LOI	0.02 (I)	0.01 (I)	0.01 (I)	0.02 (I)	0.01 (I)	0.02 (I)	0.02 (I)	0.02 (I)	0.02 (I)	0.02 (I)	0 (I)	0.02 (I)	0.02 (I)	0.01 (I)	0.02 (I)	0.03 (I)	0.04 (I)	0.01 (I)
	4.36	4.22	8.19	17	3.81	39	14.1	33.4	12.7	12.8	5.94	5.52	14.3	21.1	15.1	8.29	20.3	28.1
S	1.05	0.94	0.6	0.27	0.76	0.1	0.26	0.16	0.36	0.34	1.5	0.46	0.68	1.9	0.96	0.49	1.48	3.9
C(Inorganic)	0.81	0.49	2.09	4.48	0.45	10.34	3.21	9.32	2.54	2.64	1.06	0.20	2.62	0.15	0.08	0.06	4.75	2.09
C (Organic)	0.57	0.84	0.51	0.54	0.50	1.06	1.45	0.78	1.81	1.55	1.15	1.39	1.99	8.65	5.25	2.64	1.18	15.51

Fontenelle Creek					
Sample	P-12	P-8	P-7	P-3	P-1
SiO ₂	16.8 (X)	36.8 (X)	12.3 (X)	6.62 (X)	26.2 (X)
Al ₂ O ₃	2.65 (I)	8.69 (I)	2.64 (I)	1.53 (I)	2.08 (I)
	2.79 (X)	8.76 (X)	3.03 (X)	1.29 (X)	1.89 (X)
Fe ₂ O ₃	1.22 (I)	3.15 (I)	1.14 (I)	0.97 (I)	2.29 (I)
	1.1 (X)	3.26 (X)	1.08 (X)	0.94 (X)	2.45 (X)
MgO	14.28 (I)	1.06 (I)	10.46 (I)	0.56 (I)	0.53 (I)
	14.4 (X)	1.04 (X)	10.7 (X)	0.89 (X)	0.64 (X)
CaO	26.6 (I)	10.92 (I)	28 (I)	44.8 (I)	33.6 (I)
	25.9 (X)	11.6 (X)	28.7 (X)	47.7 (X)	36.1 (X)
Na ₂ O	0.08 (I)	0.28 (I)	0.28 (I)	1.08 (I)	0.57 (I)
	0.19 (X)	0.41 (X)	0.37 (X)	1.06 (X)	0.81 (X)
K ₂ O	1.09 (I)	2.88 (I)	1.06 (I)	0.52 (I)	0.62 (I)
	0.94 (X)	3.03 (X)	0.98 (X)	0.49 (X)	0.63 (X)
TiO ₂	0.07 (I)	0.33 (I)	0.10 (I)	0.07 (I)	0.07 (I)
	0.18 (X)	0.45 (X)	0.14 (X)	0.04 (X)	0.1 (X)
P ₂ O ₅	1.69 (I)	8.02 (I)	8.02 (I)	34.35 (I)	27.48 (I)
	1.36 (X)	7.82 (X)	7.03 (X)	32.1 (X)	25.4 (X)
	1.7 (W)	7.5 (W)	6.8 (W)	25.0 (W)	25.2 (W)
MnO	0.02 (I)	0.02 (I)	0.03 (I)	0 (I)	0.01 (I)
LOI	35.6	24.6	32.9	5.96	4.07
S	0.21	3.3	0.89	1.4	0.81
C (Inorganic)	9.07	0.21	6.88	0.81	0.39
C (Organic)	1.43	12.59	4.34	1.91	1.33

Table 2. continued

Wheat Creek															
Sample	R-15	R-9	R-7	R-8	P-34	P-30	P-26	P-22	P-20	P-16	P-14	P-11	P-9	P-7	P-4
SiO ₂	3.7 (X)	13.8 (X)	16.2 (X)	25.7 (X)	37.3 (X)	35.8 (X)	60.7 (X)	21.5 (X)	22.9 (X)	42.6 (X)	37.7 (X)	27.8 (X)	10.3 (X)	51.8 (X)	32.6 (X)
Al ₂ O ₃	0.86 (I)	2.65 (I)	1.13 (I)	0.01 (I)	2.08 (I)	7.18 (I)	10.02 (I)	3.78 (I)	3.59 (I)	8.69 (I)	8.69 (I)	6.24 (I)	1.64 (I)	10.02 (I)	4.73 (I)
Fe ₂ O ₃	0.46 (X)	2.49 (X)	0.9 (X)	0.12 (X)	1.98 (X)	7.07 (X)	10 (X)	3.93 (X)	3.68 (X)	8.46 (X)	8.38 (X)	6.01 (X)	2.04 (X)	9.78 (X)	4.4 (X)
	1.09 (I)	1.72 (I)	0.82 (I)	1.40 (I)	2.00 (I)	2.72 (I)	1.86 (I)	1.57 (I)	1.04 (I)	2.57 (I)	2.29 (I)	2.43 (I)	0.64 (I)	3.72 (I)	1.72 (I)
MgO	1.06 (X)	1.78 (X)	0.77 (X)	1.36 (X)	2.08 (X)	2.82 (X)	1.91 (X)	1.45 (X)	1.01 (X)	2.6 (X)	2.28 (X)	2.51 (X)	0.6 (X)	3.85 (X)	1.82 (X)
	0.35 (I)	0.51 (I)	0.35 (I)	13.81 (I)	1.06 (I)	3.82 (I)	1.18 (I)	13.45 (I)	14.11 (I)	4.48 (I)	4.81 (I)	2.99 (I)	15.80 (I)	1.18 (I)	0.51 (I)
	0.48 (X)	0.62 (X)	0.46 (X)	13.5 (X)	1.05 (X)	3.82 (X)	1.21 (X)	13.3 (X)	14 (X)	4.35 (X)	4.8 (X)	3.11 (X)	15.7 (X)	1.15 (X)	0.6 (X)
CaO	47.6 (I)	39.2 (I)	40.8 (I)	25.2 (I)	29.4 (I)	21.0 (I)	9.24 (I)	23.8 (I)	22.4 (I)	16.8 (I)	18.2 (I)	26.6 (I)	29.4 (I)	13.02 (I)	29.4 (I)
	50.2 (X)	42.2 (X)	43.1 (X)	25.1 (X)	29.8 (X)	21.8 (X)	9.2 (X)	23.7 (X)	22.8 (X)	16.8 (X)	18.8 (X)	27.9 (X)	29.8 (X)	13.0 (X)	30.6 (X)
Na ₂ O	1.49 (I)	1.32 (I)	1.20 (I)	0.09 (I)	0.77 (I)	0.82 (I)	0.78 (I)	0.46 (I)	0.47 (I)	0.55 (I)	0.50 (I)	0.74 (I)	0.27 (I)	0.51 (I)	0.96 (I)
	1.34 (X)	1.26 (X)	1.13 (X)	0.16 (X)	0.73 (X)	0.65 (X)	0.81 (X)	0.49 (X)	0.46 (X)	0.58 (X)	0.52 (X)	0.75 (X)	0.3 (X)	0.55 (X)	0.83 (X)
K ₂ O	0.12 (I)	0.86 (I)	0.28 (I)	0.05 (I)	0.52 (I)	1.92 (I)	2.84 (I)	1.08 (I)	1.06 (I)	2.64 (I)	2.64 (I)	1.92 (I)	0.94 (I)	3.12 (I)	1.80 (I)
	0.15 (X)	0.65 (X)	0.25 (X)	0.03 (X)	0.49 (X)	1.94 (X)	2.63 (X)	0.94 (X)	0.9 (X)	2.59 (X)	2.55 (X)	1.88 (X)	0.8 (X)	3.16 (X)	1.68 (X)
TiO ₂	0.05 (I)	0.12 (I)	0.05 (I)	0.01 (I)	0.07 (I)	0.17 (I)	0.45 (I)	0.13 (I)	0.15 (I)	0.33 (I)	0.33 (I)	0.17 (I)	0.07 (I)	0.33 (I)	0.15 (I)
	<0.02 (X)	0.18 (X)	0.04 (X)	<0.02 (X)	0.1 (X)	0.4 (X)	0.69 (X)	0.25 (X)	0.27 (X)	0.52 (X)	0.48 (X)	0.3 (X)	0.08 (X)	0.61 (X)	0.24 (X)
P ₂ O ₅	38.84 (I)	29.77 (I)	32.06 (I)	2.52 (I)	19.24 (I)	12.37 (I)	6.41 (I)	3.21 (I)	1.42 (I)	8.02 (I)	9.16 (I)	17.63 (I)	5.04 (I)	8.70 (I)	18.78 (I)
	32.8 (X)	27.2 (X)	27.8 (X)	1.98 (X)	17.5 (X)	11.5 (X)	5.67 (X)	2.77 (X)	1.28 (X)	7.5 (X)	8.39 (X)	16.2 (X)	4.35 (X)	8.06 (X)	17.1 (X)
	32.5 (W)	26.9 (W)	27.7 (W)	1.9 (W)	17.4 (W)	11.4 (W)	5.8 (W)	2.8 (W)	1.4 (W)	7.6 (W)	8.2 (W)	16.2 (W)	4.5 (W)	7.9 (W)	17.3 (W)
MnO	0.01 (I)	0.01 (I)	0.01 (I)	0.03 (I)	0.02 (I)	0.01 (I)	0.01 (I)	0.03 (I)	0.03 (I)	0.01 (I)	0.01 (I)	0.01 (I)	0.02 (I)	0.01 (I)	0.01 (I)
LOI	5.34	6.04	5.77	31.8	6.55	11.9	5.28	30.9	32.1	12.4	14	10.8	35.8	6.05	7.48
S	1.4	1.4	1.1	0.04	0.53	0.38	0.12	0.13	0.06	0.26	0.21	0.56	0.17	0.32	0.77
C(Inorganic)	1.01	0.92	0.97	8.56	1.43	2.11	0.44	8.02	8.51	2.45	2.71	1.90	9.50	0.49	1.47
C (Organic)	0.39	0.37	0.34	0.44	0.44	0.63	0.30	0.54	0.46	0.58	0.76	0.83	0.60	0.41	0.58

Lakeridge															
Sample	E-92C	US-91B	E-90A	E-89A	US-88C	TO-87B	TO-86D	TO-85A	TO-84B	RT-81A	TO-78C	RT-77	RT-75A	F-74	F-73
SiO ₂	0.81 (I)	0.87 (I)	0.74 (I)	0.98 (I)	0.45 (I)	1.3 (I)	2.1 (I)	5.1 (I)	2.8 (I)	2.1 (I)	36.6 (X)	50.9 (X)	3.4 (I)	8.85 (X)	1.8 (I)
Al ₂ O ₃	1.6 (I)	0.96 (X)	2.0 (I)	2.3 (I)	0.54 (X)	5.8 (I)	3.3 (I)	4.8 (X)	5.2 (I)	3.0 (I)	1.99 (X)	7.39 (X)	2.5 (I)	0.63 (X)	1.6 (I)
Fe ₂ O ₃	23.2 (I)	2.2 (I)	10.5 (I)	18.3 (I)	3.13 (X)	6.1 (I)	2.2 (I)	4.4 (X)	8.5 (I)	2.2 (I)	2.97 (X)	7.01 (X)	2.5 (I)	0.93 (X)	17 (I)
	29 (I)	1.91 (X)	22 (I)	31 (I)	6.80 (X)	13 (I)	4.9 (I)	14 (I)	20 (I)	24 (I)	10.2 (X)	2.77 (X)	39 (I)	16.3 (X)	17.1 (X)
CaO	0.32 (I)	0.36 (I)	0.34 (I)	0.38 (I)	0.27 (I)	0.36 (I)	0.38 (I)	0.82 (I)	0.57 (I)	0.84 (I)	19.3 (X)	10.8 (X)	1.5 (I)	29.8 (X)	29.1 (X)
	0.25 (I)	0.26 (I)	0.22 (I)	0.35 (I)	<0.15 (X)	0.49 (I)	0.70 (I)	1.7 (I)	0.97 (I)	0.68 (I)	0.33 (X)	1.00 (X)	0.97 (I)	0.34 (X)	0.93 (I)
K ₂ O	0.01 (I)	0.02 (I)	0 (I)	0.02 (I)	0.01 (I)	0.02 (I)	0.03 (I)	0.10 (I)	0.05 (I)	0.05 (I)	0.64 (X)	2.73 (X)	0.07 (I)	0.18 (X)	0.47 (I)
	0.85 (I)	0.08 (X)	2.0 (I)	4.1 (I)	0.04 (X)	1.7 (I)	1.3 (I)	0.29 (X)	3.4 (I)	15 (I)	0.11 (X)	0.31 (X)	23 (I)	0.07 (X)	5.0 (I)
MgO	0.115 (I)	0.028 (I)	0.076 (I)	0.101 (I)	0.062 (I)	0.058 (I)	0.026 (I)	0.052 (I)	0.062 (I)	0.022 (I)	1.68 (X)	5.03 (X)	0.014 (I)	3.60 (X)	0.032 (I)
	0.03 (X)	0.03 (X)	0.03 (X)	0.03 (X)	0.06 (X)	0.06 (X)	0.06 (X)	0.05 (X)	0.05 (X)	0.05 (X)	0.07 (X)	0.04 (X)	0.04 (X)	0.03 (X)	0.03 (X)
LOI	0.13	0.28	0.09	0.36	0.05	2.0	0.56	1.4	1.1	1.2	22.8	7.92		36.4	
S	11.0	1.12	0	9.93	5.76	3.85	1.10	4.07	5.51	1.84	0.84	3.9	2.2	0.52	0.52
C(Inorganic)	0.67	0.33	8.15	0.57	0.55	0.86	0.39	1.61	0.83	1.20	0.96	3.36	1.73	0.87	1.38
C (Organic)	0.10	0.34	0.21	0.39	0.46	0.19	0.17	0.37	0.42	1.7	0.25	0.67	2.8	0.54	0.5
F															

Table 2. continued

Lakeridge															
Sample----	F-69A	F-68A	F-65B	F-64A	LS-62A	F-58B	LS-57M	LS-57D	LS-56A	LS-55B	R-54C	M-53B	M-52	R-50F	M-47
SiO ₂ -----		49.0 (X)			48.2 (X)		7.2 (X)		0.08 (I)	0.76 (I)	52.5 (X)	48.1 (X)		53.1 (X)	23.3 (X)
Al ₂ O ₃ -----	0.55 (I)	16.8 (I)	0.76 (I)	4.9 (I)	7.2 (I)	0.01 (I)	0.15 (I)	0.09 (I)			0.96 (I)	1.5 (I)	0.83 (I)	0.57 (I)	1.1 (I)
Fe ₂ O ₃ -----	0.45 (I)	17.2 (X)	2.0 (I)	7.3 (I)	6.75 (X)	0.54 (I)	3.8 (I)	0.12 (X)			1.05 (X)	1.48 (X)		0.70 (X)	0.66 (I)
		6.8 (I)			4.4 (I)			3.3 (I)	3.5 (I)	2.3 (I)	4.4 (I)	4.6 (I)	3.0 (I)	2.8 (I)	0.86 (X)
		6.92 (X)			4.15 (X)			3.00 (X)			4.07 (X)	4.46 (X)		2.45 (X)	3.2 (I)
MgO-----	18.3 (I)	3.8 (I)	20 (I)	6.1 (I)	4.8 (I)	16 (I)	13 (I)	12.8 (I)	0.58 (I)	16 (I)	6.0 (I)	7.3 (I)	0.32 (I)	5.2 (I)	2.99 (X)
		3.89 (X)			4.85 (X)			14.4 (X)			5.81 (X)	7.35 (X)		4.79 (X)	0.17 (I)
CaO-----	39 (I)	3.8 (I)	27 (I)	9.8 (I)	13 (I)	41 (I)	25 (I)	34 (I)	11 (I)	27 (I)	18 (I)	18 (I)	43 (I)	18 (I)	0.14 (X)
		3.53 (X)			12.8 (X)			32.4 (X)			16.8 (X)	17.5 (X)		17.5 (X)	39 (I)
Na ₂ O-----	0.18 (I)	1.6 (I)	0.43 (I)	1.4 (I)	1.8 (I)	0.07 (I)	0.22 (I)	0.24 (I)	0.36 (I)	0.63 (I)	0.65 (I)	0.82 (I)	1.5 (I)	0.46 (I)	38.2 (X)
		1.53 (X)			1.54 (X)			<0.15 (X)			0.48 (X)	0.66 (X)		0.38 (X)	1.1 (I)
K ₂ O-----	0.19 (I)	5.6 (I)	0.19 (I)	1.8 (I)	2.0 (I)	0.01 (I)	0.07 (I)	0.04 (I)	0.04 (I)	0.06 (I)	0.14 (I)	0.24 (I)	0.48 (I)	0.12 (I)	1.1 (I)
		5.84 (X)			1.90 (X)			0.03 (X)			0.12 (X)	0.21 (X)		0.11 (X)	0.84 (X)
TiO ₂ -----	0.02 (I)	0.48 (I)	0.01 (I)	0.08 (I)	0.17 (I)	0 (I)	0 (I)	0 (I)	0 (I)	0 (I)	0 (I)	0.01 (I)	0.02 (I)	0.01 (I)	0.15 (X)
		1.00 (X)			0.41 (X)			<0.02 (X)			0.05 (X)	0.08 (X)		0.05 (X)	0.02 (I)
P ₂ O ₅ -----	0.02 (I)	0.23 (I)	0.39 (I)	0.23 (I)	0.16 (I)	0.39 (I)	6.0 (I)	2.1 (I)	1.5 (I)	3.4 (I)	2.8 (I)	5.7 (I)	34 (I)	8.0 (I)	0.02 (I)
		0.24 (X)			0.14 (X)			1.72 (X)			2.31 (X)	4.60 (X)		6.73 (X)	0.05 (X)
MnO-----	0.030 (I)	0.022 (I)	0.037 (I)	0.053 (I)	0.032 (I)	0.036 (I)	0.034 (I)	0.040 (I)	0.022 (I)	0.028 (I)	0.035 (I)	0.039 (I)	0.017 (I)	0.026 (I)	25 (I)
		0.03 (X)			0.04 (X)			0.04 (X)			0.04 (X)	0.04 (X)		0.03 (X)	23.9 (X)
LOI-----		8.94			8.52			36.2			15.5	14.7		11.1	0.019 (I)
S-----	0.13	4.7	0.16	0.41	5.2	0.82	0.43	1.1	1.8	0.74	0.17	0.45	1.3	0.72	0.03 (X)
C(Inorganic)	12.0	1.28	10.3	3.57	2.78	11.7	6.74	10.5	1.94	6.05	4.46	4.48	0.78	3.00	3.65
C (Organic)	1.02	1.92	0.76	0.58	2.34	0.87	0.78	0.77	2.74	1.02	0.36	0.34	0.69	1.29	0.91
F-----	0.15	0.62	0.06	0.07	0.08	0.04	0.49	0.49	0.26	0.38	0.23	0.51	3.4	0.74	2.6
															0.81
															0.26
															1.30
															0.74
															2.5
															3

Lakeridge															
Sample----	M-42A	M-41	M-40	M-39B	M-38A	M-37A	M-36B	M-35A	M-34A	M-33A	M-32A	M-31A	M-30A	M-29A	M-24
SiO ₂ -----		62.6 (X)			35.0 (X)		63.9 (X)		8.1 (I)	12 (I)	53.3 (X)	6.4 (I)	12 (I)	23.8 (X)	49.1 (X)
Al ₂ O ₃ -----	3.8 (I)	7.4 (I)	1.3 (I)	4.2 (I)	8.7 (I)	6.5 (I)	3.8 (I)	11 (I)			12 (I)			2.1 (I)	12 (I)
Fe ₂ O ₃ -----	2.6 (I)	7.10 (X)	1.1 (I)	5.8 (I)	8.51 (X)	3.9 (I)	1.9 (I)	11.6 (X)	3.3 (I)	5.1 (I)	12.5 (X)	2.3 (I)	4.4 (I)	1.93 (X)	11.7 (X)
		4.9 (I)			10 (I)			5.1 (I)			4.5 (I)			2.8 (I)	4.5 (I)
		4.85 (X)			10.1 (X)			5.01 (X)			4.50 (X)			2.81 (X)	4.4 (X)
MgO-----	0.25 (I)	1.5 (I)	0.22 (I)	0.55 (I)	1.1 (I)	1.6 (I)	18 (I)	1.3 (I)	1.1 (I)	1.8 (I)	1.8 (I)	12 (I)	2.0 (I)	0.37 (I)	2.0 (I)
		1.43 (X)			1.00 (X)			1.24 (X)			1.83 (X)			0.31 (X)	1.97 (X)
CaO-----	38 (I)	9.9 (I)	48 (I)	38 (I)	12 (I)	21 (I)	22 (I)	2.8 (I)	25 (I)	11 (I)	7.4 (I)	18 (I)	7.1 (I)	36 (I)	9.7 (I)
		9.56 (X)			12.7 (X)			2.71 (X)			7.28 (X)			35.9 (X)	9.46 (X)
Na ₂ O-----	1.9 (I)	3.0 (I)	1.5 (I)	1.5 (I)	1.1 (I)	1.6 (I)	1.5 (I)	1.9 (I)	1.4 (I)	1.4 (I)	0.94 (I)	1.0 (I)	1.3 (I)	1.2 (I)	1.5 (I)
		2.69 (X)			0.96 (X)			1.88 (X)			0.92 (X)			1.06 (X)	1.43 (X)
K ₂ O-----	0.98 (I)	1.2 (I)	0.42 (I)	1.3 (I)	2.9 (I)	2.8 (I)	0.73 (I)	3.2 (I)	2.5 (I)	3.7 (I)	3.8 (I)	1.9 (I)	3.5 (I)	0.64 (I)	3.5 (I)
		1.20 (X)			2.82 (X)			3.42 (X)			3.77 (X)			0.58 (X)	3.51 (X)
TiO ₂ -----	0.08 (I)	0.10 (I)	0.03 (I)	0.10 (I)	0.20 (I)	0.17 (I)	0.05 (I)	0.25 (I)	0.10 (I)	0.28 (I)	0.33 (I)	0.17 (I)	0.27 (I)	0.05 (I)	0.28 (I)
		0.53 (X)			0.42 (X)			0.80 (X)			0.56 (X)			0.08 (X)	0.56 (X)
P ₂ O ₅ -----	25 (I)	0.57 (I)	34 (I)	25 (I)	9.4 (I)	14 (I)	0.30 (I)	1.2 (I)	19 (I)	6.9 (I)	4.8 (I)	1.2 (I)	4.1 (I)	23 (I)	6.0 (I)
		0.51 (X)			8.37 (X)			1.14 (X)			4.23 (X)			22.9 (X)	5.41 (X)
MnO-----	0.011 (I)	0.031 (I)	0.006 (I)	0.007 (I)	0.014 (I)	0.019 (I)	0.040 (I)	0.027 (I)	0.013 (I)	0.022 (I)	0.015 (I)	0.030 (I)	0.018 (I)	0.013 (I)	0.018 (I)
		0.04 (X)			0.02 (X)			0.03 (X)			0.02 (X)			0.02 (X)	0.02 (X)
LOI-----		6.46			18			7.47			9.65			5.40	9.76
S-----	1.4	1.0	1.2	5.4	9.8	3.5	0.78	3.2	3.1	3.7	3.7	1.8	3.7	2.0	3.7
C(Inorganic)	0.40	2.22	0.60	0.63	0.23	0.77	6.43	0.32	0.53	0.59	0.49	6.14	0.64	1.12	0.67
C (Organic)	0.60	1.08	1.22	3.89	10.4	3.84	1.49	3.36	3.40	3.21	3.96	2.82	5.38	2.40	9.72
F-----	3.2	0.08	3.7	3.2	1.3	1.7	0.07	0.18	0.26	0.82	0.60	0.16	0.50	2.4	1.38
															4.48
															0.61
															1.1

Table 2. continued

Sample	Lakeridge										
	M-23B	M-22B	M-21A	M-20B	M-18	M-17	M-16	M-15A	M-14	M-12B	WS-11D
SiO ₂		20.2 (X)			20.6 (X)			17.9 (X)	12.6 (X)		22.8 (X)
Al ₂ O ₃	1.6 (I)	4.9 (I)	1.9 (I)	10 (I)	5.3 (I)	5.1 (I)	5.3 (I)	4.9 (I)	3.6 (I)	0.43 (I)	6.1 (I)
Fe ₂ O ₃		4.63 (X)			5.12 (X)			4.73 (X)	3.54 (X)		5.58 (X)
	0.62 (I)	1.7 (I)	1.3 (I)	4.2 (I)	2.0 (I)	2.2 (I)	2.3 (I)	2.0 (I)	3.0 (I)	2.8 (I)	3.9 (I)
		1.68 (X)			1.94 (X)			2.01 (X)	2.94 (X)		3.70 (X)
MgO	20 (I)	9.6 (I)	0.61 (I)	0.68 (I)	7.0 (I)	3.0 (I)	12 (I)	2.2 (I)	1.1 (I)	0.46 (I)	12 (I)
CaO		10.6 (X)			7.37 (X)			2.05 (X)	0.99 (X)		14.1 (X)
	29 (I)	21 (I)	38 (I)	14 (I)	18 (I)	18 (I)	20 (I)	25 (I)	28 (I)	48 (I)	21 (I)
		19.9 (X)			18.6 (X)			25.6 (X)	27.7 (X)		19.7 (X)
Na ₂ O	0.61 (I)	0.84 (I)	1.1 (I)	1.8 (I)	0.81 (I)	0.96 (I)	1.2 (I)	1.3 (I)	0.89 (I)	1.1 (I)	0.49 (I)
K ₂ O		0.77 (X)			0.79 (X)			1.28 (X)	0.94 (X)		0.35 (X)
	0.37 (I)	1.4 (I)	0.65 (I)	3.1 (I)	1.7 (I)	1.7 (I)	1.6 (I)	1.6 (I)	1.3 (I)	0.14 (I)	2.4 (I)
		1.34 (X)			1.62 (X)			1.52 (X)	1.28 (X)		2.04 (X)
TiO ₂	0.03 (I)	0.12 (I)	0.03 (I)	0.28 (I)	0.13 (I)	0.13 (I)	0.13 (I)	0.07 (I)	0.07 (I)	0.02 (I)	0.13 (I)
P ₂ O ₅		0.21 (X)			0.27 (X)			0.25 (X)	0.18 (X)		0.30 (X)
	0.69 (I)	2.3 (I)	23 (I)	11 (I)	5.0 (I)	9.9 (I)	3.0 (I)	15 (I)	16 (I)	30 (I)	0.16 (I)
		1.95 (X)			4.39 (X)			15.1 (X)	16.9 (X)		0.10 (X)
MnO	0.015 (I)	0.011 (I)	0.006 (I)	0.012 (I)	0.017 (I)	0.013 (I)	0.015 (I)	0.008 (I)	0.005 (I)	0.017 (I)	0.044 (I)
LOI		0.01 (X)			0.02 (X)			0.01 (X)	<0.01 (X)		0.05 (X)
		30.4			27.9			22.9	25		27.6
	1.0	3.1	4.3	4.7	4.4	7.1	3.4	5.0	6.3	1.0	1.4
C(Inorganic)	10.8	6.62	1.26	0.13	4.69	1.74	6.03	1.49	0.93	1.66	8.08
C (Organic)	4.27	10.1	12.6	7.21	15.1	24.9	8.27	15.5	18.1	1.24	0.72
F	0.07	0.25	2.5	1.2	0.61	1.2	0.38	2.1	2.6	3.4	0.14

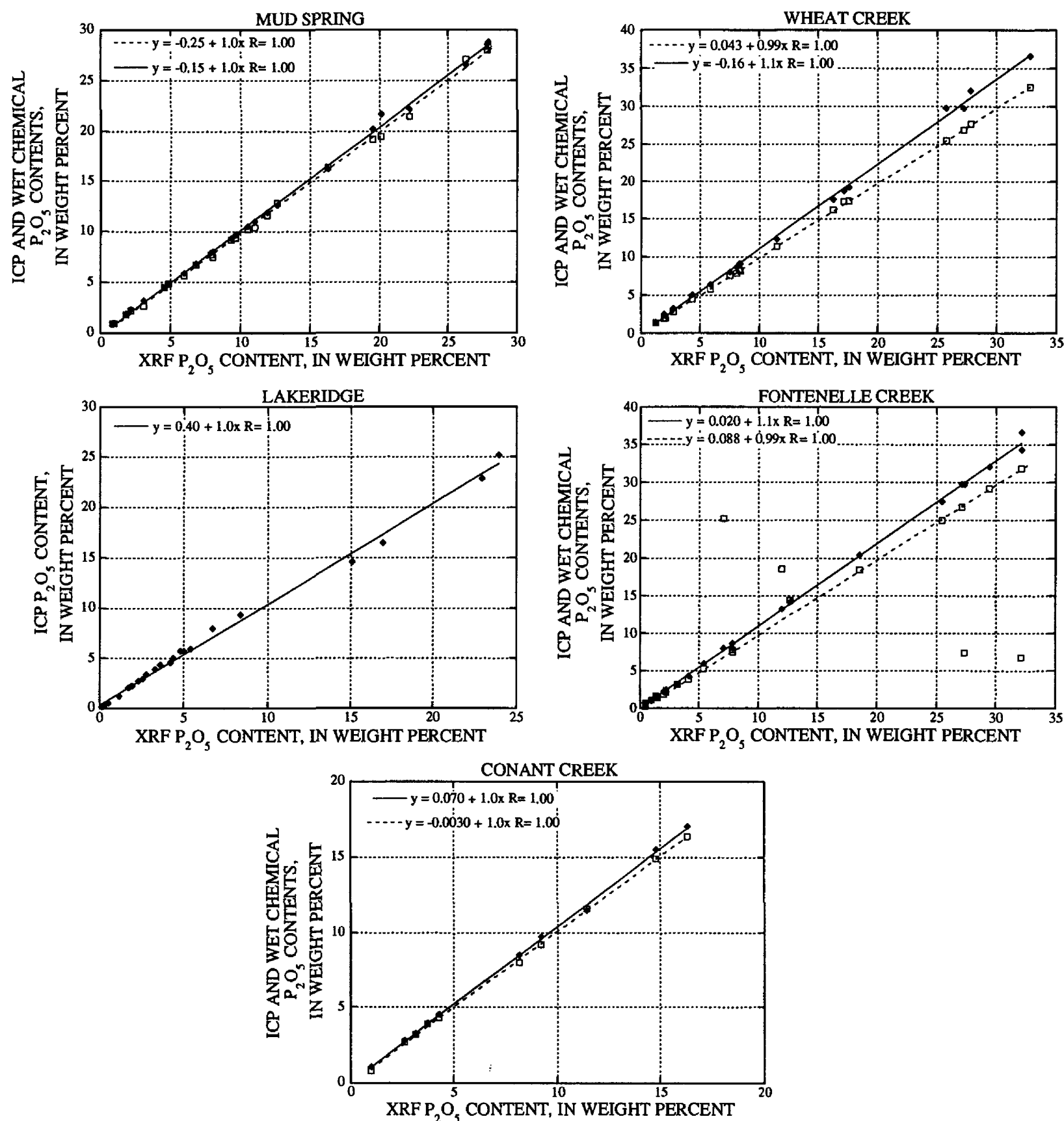


Figure 3. Comparison of measured P_2O_5 (A), Al_2O_3 (B), CaO (C), and MgO (D) contents by different analytical techniques: inductively coupled plasma-atomic emission spectroscopy (ICP), X-ray-fluorescence spectroscopy (XRF), and wet chemical analysis (WC). The open squares are the WC/XRF data and the solid diamonds are the ICP-AES/XRF data. Graphs include curves of best fit, equation for curve of best fit, correlation coefficients (R values) and the one-to-one line (the solid line). In figure 3A, the Fontenelle graph, the line of best fit does not include the four outlier data points.

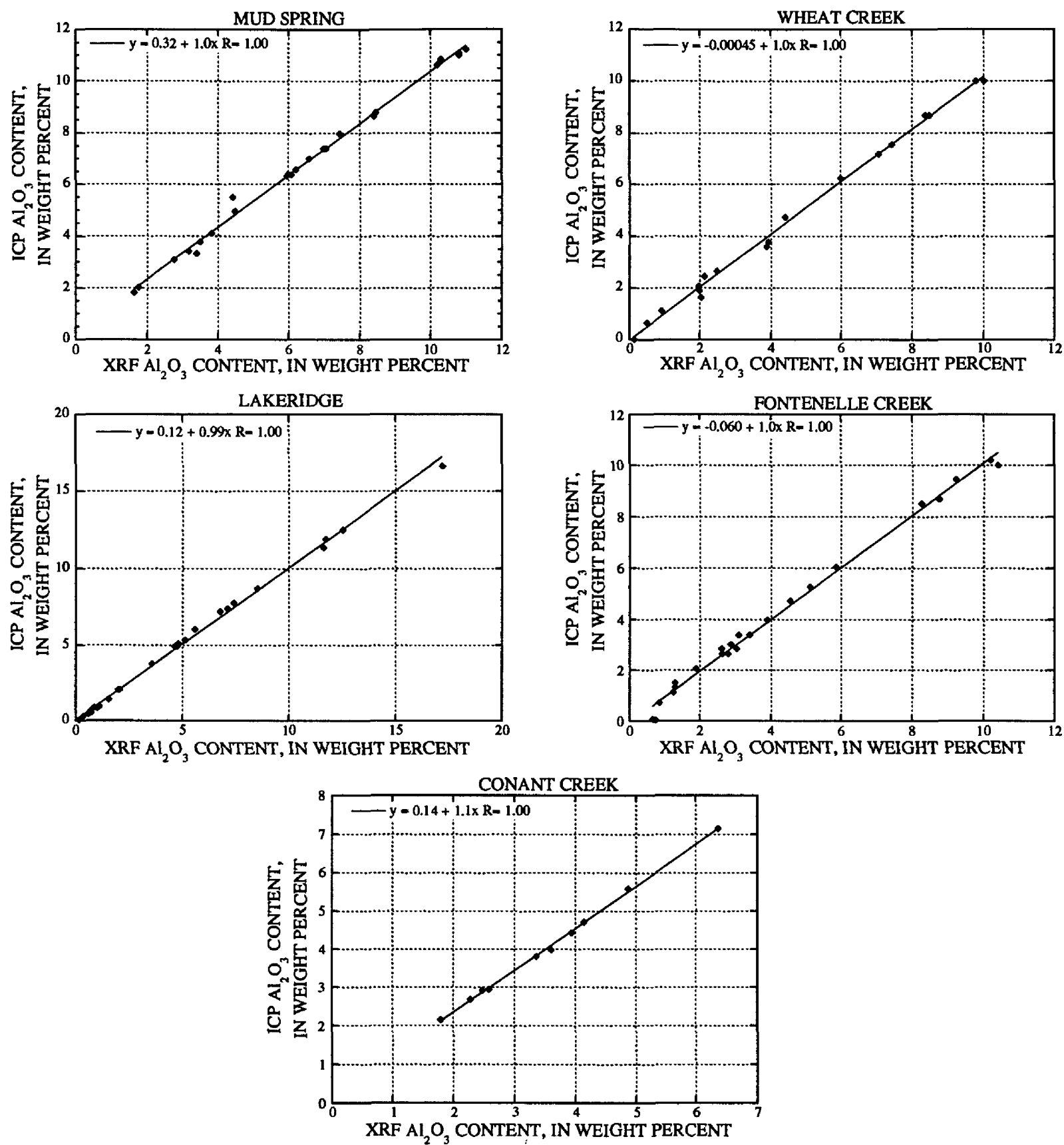


Figure 3. continued

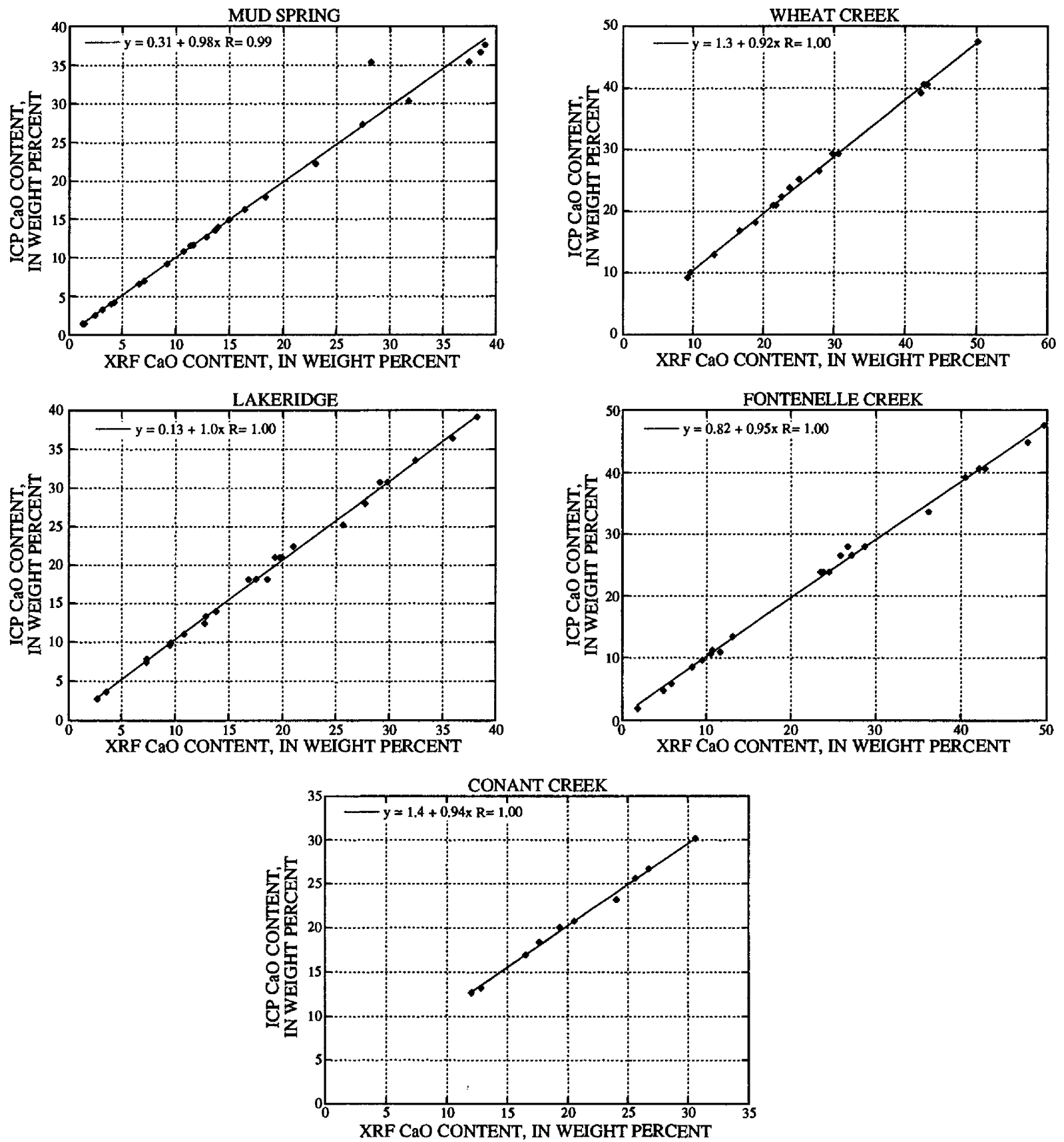


Figure 3. continued

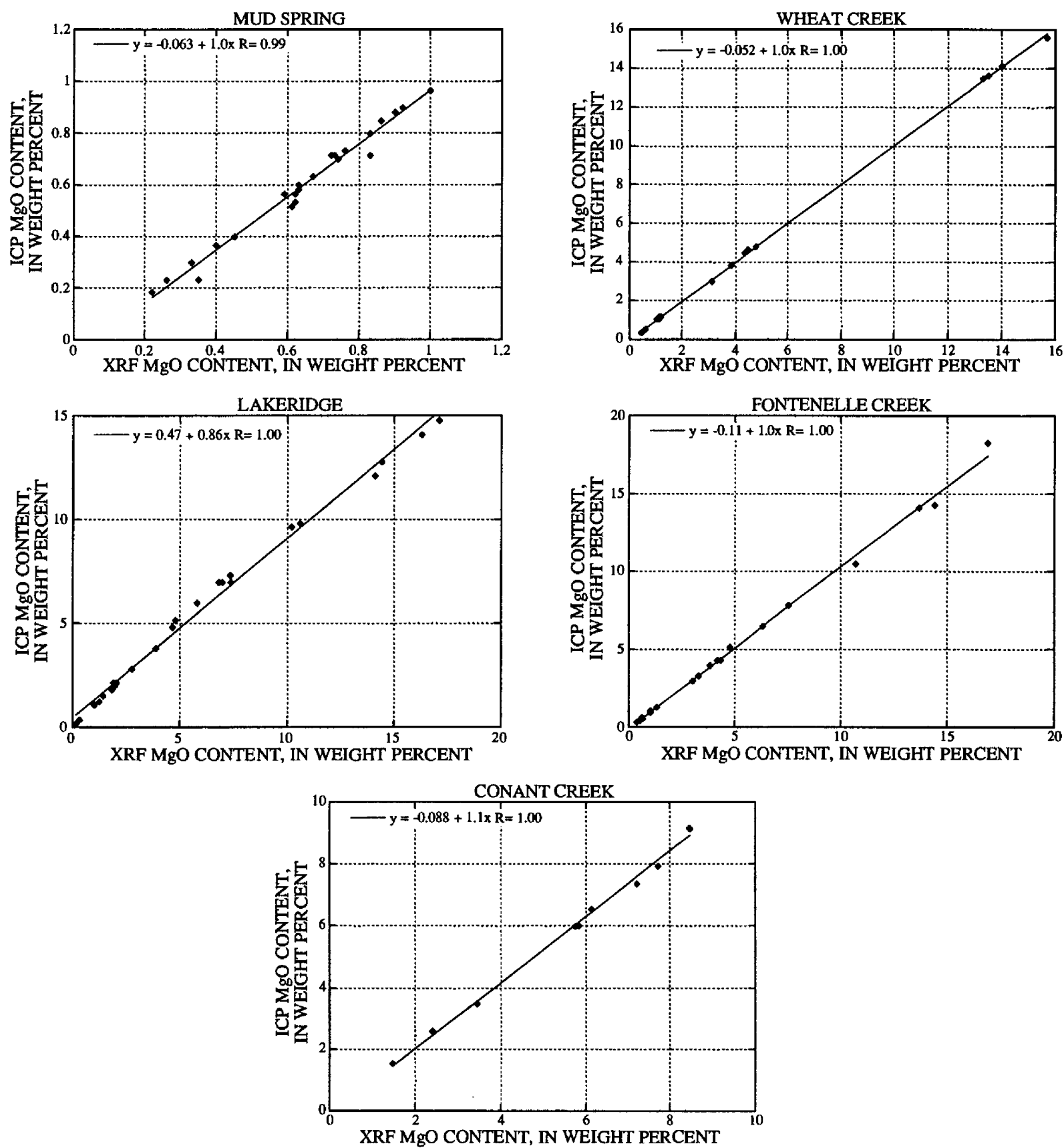


Figure 3. continued

Minor-element analyses

Minor elements were determined by inductively coupled plasma-emission spectroscopy (ICP-AES; table 3), and the rare earth elements (REE) by inductively coupled plasma-mass spectroscopy (ICP-MS; table 4; Lichte and others, 1987b). The range of detection, precision, and accuracy of each analytical technique is given in table 1. Minor elements that were above their detection level in only a few samples -- Bi, Nb, Sn, Ta and U -- are not reported in table 3.

La was the only minor element used to gauge the quality of minor-element data because it was analyzed by two techniques, ICP-MS and ICP-AES, and was significantly above the detection limit. X-Y plots of La (fig. 4) gave results similar to the X-Y plots for the major elements which were analyzed by both ICP-AES and XRF, $R \geq 0.98$ and deviation from the 1:1 curve of ≤ 14 percent.

-Miscellaneous analyses

Total sulfur and total carbon, evolved as oxides during combustion, were measured by absorption of infrared radiation (table 2; Jackson and others, 1987). Carbonate carbon was measured as CO_2 by coulometric titration (Jackson and others, 1987). For the Lakeridge samples, fluorine was determined by ion-selective electrode following LiBO_2 fusion and HNO_3 dissolution (Bodkin, 1977; Cremer and others, 1984). The ranges of detection, precision, and accuracy are given in table 1.

Table 3. Minor-element contents in samples of the Phosphoria Formation at Conant Creek, Wheat Creek, Fontenelle Creek, and Lakeridge, Wyoming and Mud Springs, Idaho

[All values in parts per million.]

Conant Creek											Mud Springs				
Sample	P-47	P-46	P-45	P-44	P-43	P-42	P-41	P-40	P-39	P-38	P-47	P-44	P-43	P-41	P-40
Mn---	149	119	168	199	186	226	88	249	177	210	85	86	30	46	21
Ag----	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	4	4	5	5	6
As----	5	20	10	10	20	30	40	10	20	20	30	10	10	20	30
Ba----	68	84	59	84	132	161	93	115	120	132	4460	1830	4860	1040	601
Be----	1	2	1	0.5	1	2	1	0.5	1	2	3	1	2	2	2
Cd----	1	1	1	1	5	6	1	1	2	6	9	6	4	8	3
Co----	5	4	3	4	5	6	5	5	5	5	21	14	5	6	7
Cr----	374	466	315	262	516	735	371	281	435	544	1970	1110	1680	1060	1420
Cu----	37	19	8	17	18	11	11	20	22	10	225	221	255	169	126
Ga----	2	5	2	2	8	9	2	4	5	6	14	12	13	6	11
Li----	50	52	27	31	54	64	32	39	42	38	25	18	21	14	22
Mo----	38	8	8	13	6	6	18	32	12	12	10	6	4	12	11
Ni----	63	31	15	25	31	37	29	49	33	35	376	304	126	298	286
Pb----	2	16	31	21	37	31	144	18	65	67	11	11	12	10	10
Sc----	4	5	4	3	6	8	3	4	5	6	11	10	12	9	12
Sr----	175	429	499	137	196	168	579	83	358	448	1170	254	393	470	338
Th----	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	4	5	9	8	< 4	8
V----	54	69	40	41	69	91	44	48	58	72	145	114	149	118	170
Y----	46	102	109	26	45	50	92	20	59	65	158	86	157	173	76
Zn----	23	32	30	14	779	903	314	28	551	1080	1240	883	512	1570	1230

Mud Springs															
Sample	P-38	P-35	P-33	P-31	P-29	P-27	P-26	P-25	P-23	P-22	P-17	P-16	P-15	P-14	P-11
Mn---	21	13	20	98	176	21	361	442	4040	112	18	16	124	87	473
Ag----	7	6	5	11	7	6	5	3	3	3	4	4	< 2	< 2	3
As----	30	20	30	10	< 10	10	10	10	< 10	10	20	10	< 10	< 10	10
Be----	569	336	534	292	230	219	276	219	821	253	867	1010	373	202	286
Be----	2	2	3	1	1	2	2	1	1	1	2	2	< 1	< 1	1
Cd----	3	2	2	13	18	9	29	11	55	3	7	5	3	< 2	12
Co----	5	3	6	13	9	4	9	10	108	5	3	5	7	4	20
Cr----	1660	1940	2200	1750	904	1010	1040	442	234	910	1910	1740	570	570	542
Cu----	200	143	258	179	94	80	78	61	96	49	197	195	96	74	113
Ga----	12	11	11	5	5	8	8	6	6	7	13	11	< 4	< 4	7
Li----	20	14	31	22	20	21	26	20	15	20	31	25	9	13	23
Mo----	16	9	18	7	9	9	9	13	18	6	17	8	10	7	11
Ni----	240	153	340	177	381	144	418	584	986	271	182	187	150	129	319
Pb----	14	11	12	11	5	9	12	6	11	6	17	14	5	4	9
Sc----	12	9	12	6	6	4	8	7	6	7	13	11	3	3	6
Sr----	522	462	792	1010	527	669	407	246	136	71	545	867	266	59	307
Th----	9	5	6	< 4	< 4	< 4	6	5	8	7	8	4	< 4	< 4	4
V----	155	141	222	830	203	296	229	143	109	110	223	173	62	60	159
Y----	109	173	133	463	183	171	122	86	59	52	253	208	41	34	142
Zn----	883	517	1320	874	1400	742	1430	1540	2110	903	534	613	506	364	1220

Mud Springs					Fontenelle Creek										
Sample	P-8	P-6	P-4	P-2	P-1	E-4	U-3	U-1	R-9	R-6	P-45	P-43	P-40	P-39	P-37
Mn---	25	23	170	121	104	160	43	95	150	110	170	150	160	120	120
Ag----	3	5	3	< 2	< 2	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4
As----	10	10	< 10	20	< 10	20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20
Ba----	189	154	520	191	92	140	74	28	51	91	20	110	33	120	110
Be----	2	2	< 1	2	< 1	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2
Cd----	31	29	< 2	13	5	< 4	6	5	< 4	< 4	< 4	20	10	30	32
Co----	3	3	4	4	5	4	5	5	4	5	2	5	4	5	5
Cr----	1250	1610	65	803	289	470	560	620	430	1100	130	370	130	450	480
Cu----	96	74	26	84	34	43	20	20	25	26	8	47	20	48	43
Ga----	5	5	< 4	9	< 4	< 8	< 8	< 8	< 8	< 6	< 8	10	< 8	10	10
Li----	18	25	14	69	21	23	20	20	10	23	6	20	6	20	20
Mo----	10	7	27	21	13	10	8	7	< 4	4	< 4	31	10	36	23
Ni----	139	129	122	245	107	34	39	53	20	31	10	66	22	86	79
Pb----	15	12	< 4	9	10	140	20	10	< 8	6	< 8	< 8	< 8	< 6	< 8
Sc----	6	5	< 2	8	3	4	5	4	< 4	5	< 4	< 4	< 4	< 4	4
Sr----	741	946	44	453	951	660	660	330	240	510	82	85	180	100	120
Th----	< 4	< 4	< 4	4	< 4	< 8	< 8	< 8	< 6	< 8	< 8	< 6	< 8	< 8	< 8
V----	515	515	30	123	44	110	110	260	44	86	25	56	38	110	120
Y----	429	501	22	195	347	170	570	260	81	250	10	26	10	26	31
Zn----	552	493	214	823	298	160	640	540	200	290	89	420	120	610	770

Sample	Fontenelle Creek													Wheat Creek	
	P-32	P-31	P-25	P-23	P-21	P-20	P-19	P-13	P-12	P-8	P-7	P-3	P-1	R-15	R-9
Mn---	38	120	160	72	120	210	330	74	170	120	210	35	84	81	72
Ag----	< 4	< 4	< 4	9	5	< 4	< 4	10	< 4	10	5	< 4	< 4	< 4	< 4
As----	< 20	< 20	< 20	30	20	< 20	< 20	< 20	< 20	30	< 20	< 20	< 20	< 20	20
Ba----	65	230	160	290	290	280	190	130	79	250	88	80	85	210	210
Be----	2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	2
Cd----	20	10	44	220	190	150	42	150	57	280	140	10	27	10	6
Co----	4	8	8	4	9	10	10	5	6	7	7	4	4	4	4
Cr----	1200	630	800	1600	2700	500	120	2500	270	2800	1100	550	530	460	620
Cu----	24	42	56	170	140	49	36	160	31	210	70	43	66	20	20
Ga----	< 8	10	10	41	20	10	< 8	10	< 8	27	10	9	10	< 8	< 8
Li----	8	22	22	40	44	20	10	30	7	41	10	9	10	6	9
Mo----	10	20	56	580	240	58	10	200	46	300	69	20	32	8	31
Ni----	24	76	120	440	220	130	88	360	120	500	290	46	59	28	41
Pb----	< 8	10	10	10	20	20	9	10	< 8	20	< 8	< 8	10	10	29
Sc----	5	8	6	10	10	9	5	7	< 4	10	5	6	< 4	< 4	6
Sr----	990	190	330	58	110	140	94	540	190	290	370	1000	460	1200	1100
Th----	< 8	< 8	< 8	10	9	< 8	< 8	< 8	< 8	< 8	< 8	< 8	< 8	< 8	< 8
V----	180	160	340	11000	1700	740	140	2200	450	2200	610	150	200	110	71
Y----	750	120	200	40	190	79	20	150	36	280	140	340	320	210	200
Zn----	330	610	960	1700	600	1000	510	1600	890	2200	1100	350	170	650	530

Sample	Wheat Creek														
	R-7	R-6	P-34	P-30	P-26	P-22	P-20	P-16	P-14	P-11	P-9	P-7	P-4	P-3	P-1
Mn---	76	260	140	110	58	210	200	110	96	65	120	84	84	80	120
Ag----	< 4	< 4	< 4	6	< 4	5	< 4	7	10	9	< 4	9	9	7	< 4
As----	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20	< 20
Ba----	650	48	150	240	320	120	120	270	270	200	66	320	230	260	220
Be----	2	< 2	< 2	2	2	< 2	< 2	< 2	< 2	< 2	< 2	2	< 2	< 2	< 2
Cd----	37	85	20	44	20	23	20	33	70	41	7	4	63	10	7
Co----	4	6	5	3	3	2	3	3	5	4	3	2	5	2	4
Cr----	480	70	550	1900	1200	600	310	1900	1600	2500	370	1300	1100	170	660
Cu----	21	22	52	80	27	51	36	69	84	88	20	86	46	27	28
Ga----	< 8	< 8	< 8	10	10	< 8	< 8	10	20	10	< 8	10	9	< 8	< 8
Li----	7	5	10	40	30	20	10	31	34	30	9	45	26	10	8
Mo----	< 4	5	10	27	10	20	10	8	< 4	57	5	36	24	20	8
Ni----	20	21	40	200	97	96	100	170	200	180	52	260	81	87	24
Pb----	63	20	10	10	10	< 8	< 8	10	10	10	< 8	24	20	10	9
Sc----	< 4	< 4	6	10	10	5	4	10	10	9	< 4	10	7	6	5
Sr----	1100	110	620	430	230	210	130	360	360	610	220	300	600	110	810
Th----	< 8	< 8	< 8	< 8	9	< 8	< 8	< 8	< 8	< 8	< 8	< 8	< 8	< 8	< 8
V----	160	42	110	350	350	150	140	440	1300	820	70	410	290	100	110
Y----	350	40	280	270	170	58	22	190	220	250	43	180	300	38	190
Zn----	540	230	330	1000	480	410	410	770	830	710	220	1200	630	750	200

Sample	Lakeridge														
	Cw-11	E-92C	US-91B	E-90A	E-89A	US-88C	TO-87B	TO-86D	TO-85A	TO-84B	RT-81A	TO-78C	RT-77	RT-75A	F-74
Mn---	120	890	220	590	780	480	450	200	400	480	170	500	240	110	200
Ag----	< 4	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2
As----	< 20	< 10	< 10	< 10	< 10	< 10	10	< 10	< 10	< 10	< 10	< 10	< 10	< 10	< 10
Ba----	180	16	40	20	27	48	62	66	120	62	92	82	110	150	34
Be----	< 2	< 1	< 1	< 1	< 1	< 1	< 1	< 1	1	< 1	1	< 1	1	1	< 1
Cd----	20	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	8	8	< 2	< 2
Co----	4	3	3	3	4	4	4	4	6	5	4	4	6	5	2
Cr----	140	120	210	130	240	170	180	180	400	370	410	260	570	670	150
Cu----	34	22	34	36	29	57	61	52	60	78	57	42	63	42	7
Ga----	< 8	< 4	< 4	< 4	< 4	< 4	< 4	< 4	6	4	< 4	< 4	8	8	< 4
Li----	7	8	20	14	16	12	18	36	43	32	31	26	50	31	7
Mo----	5	5	7	8	8	14	17	16	13	19	11	10	16	12	15
Ni----	22	15	24	21	24	39	41	38	53	59	45	35	110	90	16
Pb----	< 8	< 4	20	9	11	9	28	21	32	32	40	21	25	19	6
Sc----	< 4	< 2	< 2	2	3	2	2	2	6	4	3	3	7	7	2
Sr----	460	74	99	96	140	130	100	73	150	190	700	160	340	1000	130
Th----	< 8	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	14	< 4
V----	36	18	23	21	29	19	22	26	59	57	39	43	66	160	41
Y----	120	20	84	29	62	59	25	16	40	31	78	26	110	460	50
Zn----	180	3	8	< 2	2	3	4	64	22	19	13	880	500	55	3

Lakeridge

Sample	F-73	F-72B	F-71B	F-70E	F-69A	F-68A	F-65B	F-64A	LB-62A	F-58B	LS-57M	LS-57D	LB-56A	LS-55B	R-54C
Mn---	250	280	220	370	230	170	290	410	250	280	260	310	170	220	270
Ag----	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2
As----	< 10	< 10	< 10	< 10	< 10	32	< 10	< 10	10	< 10	< 10	< 10	< 10	< 10	< 10
Ba----	42	65	37	52	19	27	110	170	36	11	15	10	47	19	23
Be----	< 1	< 1	< 1	< 1	< 1	3	< 1	< 1	< 1	< 1	< 1	< 1	< 1	< 1	< 1
Cd----	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	2	< 2
Co----	3	2	2	6	3	20	4	9	10	2	4	4	3	3	5
Cr----	170	130	130	380	10	140	74	210	120	20	200	130	110	150	250
Cu----	21	15	25	120	8	55	36	130	51	12	79	60	69	42	79
Ga----	< 4	< 4	< 4	< 4	< 4	17	< 4	4	6	< 4	< 4	< 4	< 4	< 4	< 4
Li----	6	3	6	17	6	110	8	15	23	< 2	4	2	11	7	9
Mo----	10	9	15	37	2	27	11	27	8	2	15	13	13	10	16
Ni----	18	14	29	75	5	59	23	84	36	5	38	35	36	36	49
Pb----	8	< 4	< 4	6	< 4	10	< 4	9	14	< 4	< 4	< 4	< 4	< 4	< 4
Sc----	4	2	< 2	4	< 2	15	< 2	4	6	< 2	2	< 2	< 2	< 2	3
Sr----	160	99	140	110	180	70	89	70	300	630	400	270	180	230	220
Th----	< 4	< 4	< 4	< 4	< 4	7	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4	< 4
V----	38	29	39	66	14	180	11	34	36	6	14	7	6	14	24
Y----	60	20	40	58	5	10	6	10	11	9	170	62	40	130	52
Zn----	7	2	6	< 2	6	27	5	7	25	11	25	23	11	83	30

Lakeridge

Sample	M-53B	M-52	R-50F	M-48B	M-47	M-45	M-44A	M-42A	M-41	M-40	M-39B	M-38A	M-37A	M-36B	M-35A
Mn---	300	130	200	95	140	150	72	82	240	44	56	110	150	310	210
Ag----	< 2	< 2	< 2	2	< 2	< 2	< 2	< 2	3	< 2	7	65	6	< 2	3
As----	< 10	< 10	< 10	10	< 10	< 10	< 10	< 10	< 10	< 10	< 10	76	20	< 10	20
Ba----	94	150	45	200	140	270	270	270	180	290	60	86	110	100	63
Be----	< 1	2	< 1	2	< 1	< 1	4	2	< 1	3	4	2	2	< 1	2
Cd----	< 2	2	15	< 2	4	< 2	< 2	< 2	< 2	< 2	4	8	110	< 2	5
Co----	5	4	5	7	4	4	4	4	7	4	8	8	8	4	9
Cr----	450	900	320	1000	420	720	2100	1200	240	630	1900	1800	1800	110	730
Cu----	84	87	56	84	86	85	110	94	73	50	74	110	58	21	43
Ga----	< 4	4	< 4	12	< 4	< 4	10	6	6	5	11	22	16	< 4	12
Li----	13	14	15	61	12	19	46	11	8	9	25	48	30	5	25
Mo----	23	17	120	22	39	31	33	25	25	16	59	220	24	5	14
Ni----	65	47	73	190	69	52	150	41	90	92	190	400	100	20	70
Pb----	7	< 4	10	20	< 4	< 4	6	6	6	< 4	7	32	11	< 4	13
Sc----	3	5	< 2	13	5	9	10	12	5	12	14	14	13	3	10
Sr----	260	2000	350	110	760	800	1000	980	130	1500	1300	400	670	98	120
Th----	< 4	5	< 4	8	< 4	< 4	< 4	7	< 4	< 4	4	4	6	< 4	5
V----	50	100	56	140	55	100	170	220	56	62	170	230	110	26	120
Y----	130	900	230	34	240	640	340	530	31	500	300	290	89	6	33
Zn----	58	310	600	41	170	21	22	13	39	88	190	280	6000	28	160

Lakeridge

Sample	M-34A	M-33A	M-32A	M-31A	M-30A	M-29A	M-28A	M-27B	M-25	M-24	M-23B	M-22B	M-21A	M-20B	M-18
Mn---	99	170	120	230	140	100	89	210	140	94	120	87	45	91	130
Ag----	5	5	6	3	9	< 2	4	< 2	6	9	2	7	14	10	9
As----	10	20	21	10	22	10	21	10	36	33	< 10	29	40	41	42
Ba----	97	68	87	170	69	180	210	74	120	130	43	120	140	67	160
Be----	2	2	2	1	2	< 1	2	< 1	2	2	< 1	2	3	5	3
Cd----	24	15	20	11	66	52	30	5	63	260	20	210	540	360	160
Co----	7	9	8	5	8	4	6	3	9	6	2	4	< 1	10	3
Cr----	1000	1400	1400	890	2100	740	1200	110	1500	1900	280	1200	950	950	2800
Cu----	40	46	41	23	89	57	61	16	54	94	16	71	330	140	160
Ga----	11	16	15	8	16	< 4	10	< 4	14	9	< 4	< 4	< 4	15	< 4
Li----	24	36	41	25	44	13	23	4	29	25	4	20	17	39	43
Mo----	13	9	17	11	59	16	19	5	47	170	19	210	200	240	330
Ni----	73	91	100	65	220	67	100	17	130	290	43	270	1400	360	710
Pb----	10	12	15	6	15	5	10	< 4	10	17	5	12	23	16	14
Sc----	8	14	13	8	14	8	10	3	13	10	2	6	6	11	8
Sr----	800	350	230	140	200	940	860	120	290	500	140	220	1400	740	360
Th----	< 4	6	6	< 4	6	< 4	< 4	< 4	6	< 4	< 4	< 4	< 4	7	< 4
V----	150	140	180	95	180	76	140	32	220	770	250	1500	2600	1300	1500
Y----	62	72	37	29	83	270	120	8	54	84	8	59	260	160	210
Zn----	900	580	600	320	1500	1200	950	130	1300	4300	440	2800	9400	8900	3200

Lakeridge

Sample	M-17	M-16	M-15A	M-14	M-12B	WS-11D
Mn---	100	120	60	39	130	340
Ag---	15	8	13	13	< 2	< 2
As---	46	21	36	59	< 10	10
Ba---	100	160	240	150	290	210
Be---	5	2	3	3	< 1	1
Cd---	590	360	190	110	29	< 2
Co---	2	5	3	2	4	8
Cr---	3300	580	3300	5500	330	82
Cu---	320	99	210	480	130	46
Ga---	< 4	< 4	< 4	29	4	6
Li---	42	26	55	79	12	32
Mo---	670	410	290	360	30	10
Ni---	1100	660	880	1200	110	30
Pb---	22	8	20	25	7	10
Sc---	8	6	9	8	3	6
Sr---	550	270	830	890	1900	100
Th---	4	< 4	< 4	< 4	< 4	< 4
V---	3900	2400	1500	1100	120	42
Y---	280	58	450	270	450	6
Zn---	7000	5200	3700	4800	790	43

Table 4. Rare earth elements in samples from the Phosphoria Formation from Conant Creek, Fontenelle Creek, Wheat Creek, and Lakeridge, Wyoming, and Mud Spring, Idaho. [All values in parts per million.]

Conant Creek											Mud Spring				
Sample	P-47	P-46	P-45	P-44	P-43	P-42	P-41	P-40	P-39	P-38	P-47	P-44	P-43	P-41	P-40
La---	33	71	74	19	34	36	62	14	37	41	91	71	110	120	68
Ce---	21	30	27	15	26	27	35	15	23	22	46	50	57	44	52
Pr---	4.9	8.9	8.3	3	5.3	5.6	7.8	2.6	5.1	5.2	19	14	19	17	12
Nd---	22	41	37	13	22	25	33	11	22	23	83	59	79	71	53
Sm---	3.5	6.4	5.6	2.2	3.8	4.2	5.2	2	3.6	3.7	13	9.9	13	11	9.0
Eu---	0.81	1.3	1.2	0.43	0.84	0.88	1.3	0.43	0.83	0.82	2.9	2	2.5	2.5	1.9
Gd---	4.2	7.2	6.8	2.7	3.8	4.3	6.3	2	4.4	4.1	14	9.2	14	13	9.3
Tb---	0.6	1.1	1	0.4	0.87	0.67	0.96	0.33	0.63	0.69	2.1	1.6	2.1	1.8	1.4
Dy---	3.9	7.7	7.4	2.2	4	4.5	6	2.3	4.1	4.7	13	9.1	13	12	8.5
Ho---	0.86	1.7	1.6	0.46	0.87	0.93	1.4	0.43	0.93	1	2.9	1.9	2.9	2.7	1.7
Er---	2.4	5	4.7	1.4	2.7	3	3.9	1.5	2.8	2.8	9.2	5.7	9.3	7.9	5.1
Tm---	0.35	0.67	0.63	0.2	0.35	0.37	0.52	0.19	0.41	0.41	1.2	0.72	1.2	1.0	0.8
Yb---	2.2	5	4.7	1.5	2.5	2.8	3.7	1.6	2.7	3	8	5.5	8.2	7.4	4.4

Mud Spring															
Sample	P-38	P-35	P-33	P-31	P-29	P-27	P-26	P-25	P-23	P-22	P-17	P-16	P-15	P-14	P-11
La---	77	120	88	290	100	120	77	66	47	43	150	120	23	21	87
Ce---	50	42	43	57	36	42	46	51	44	30	57	42	13	14	42
Pr---	16	21	15	42	15	17	12	10	7.6	8	29	22	4.1	3.9	13
Nd---	68	91	63	180	63	72	51	45	33	34	130	96	18	16	56
Sm---	11	14	9.8	28	10	11	8.4	7.5	5.4	5.7	22	16	3	2.9	8.8
Eu---	2.4	3.2	2.2	6.3	2.1	2.4	1.7	1.6	1.1	1.1	4.7	3.6	0.68	0.56	2
Gd---	11	16	10	31	12	13	9.2	7.5	5.3	5.3	24	18	3.4	3.2	10
Tb---	1.8	2.3	1.5	4.7	1.7	1.9	1.4	1.2	0.87	0.81	3.6	2.7	0.52	0.42	1.6
Dy---	11	14	9.5	30	11	12	8.9	7.5	5.4	5.3	22	17	3.3	2.8	10
Ho---	2.3	3.1	2.2	7.1	2.6	2.8	2.1	1.7	1.3	1.1	4.9	3.7	0.7	0.65	2.3
Er---	6.9	8.8	6.8	22	7.5	8.6	6	5.1	3.9	3.2	14	11	2	2	6.8
Tm---	0.92	1.1	0.93	2.6	0.99	1	0.73	0.65	0.55	0.43	2	1.5	0.25	0.23	0.95
Yb---	6.1	8.3	6.4	19	7	7.6	5.6	4.4	3.7	3	13	9.7	2.1	1.7	6.3

Mud Spring						Fontenelle Creek									
Sample	P-8	P-8	P-4	P-2	P-1	E-4	U-3	U-1	R-9	R-6	P-45	P-43	P-40	P-39	P-37
La---	220	240	15	110	210	110	360	150	45	160	6.2	17	8.5	19	22
Ce---	65	52	16	46	61	40	110	49	22	79	3.7	16	5.6	19	21
Pr---	30	29	2.6	16	29	14	57	20	5.7	20	0.3	2.7	0.6	3	3.2
Nd---	140	130	12	68	130	60	240	88	27	84	3.6	13	5.2	15	16
Sm---	22	21	1.8	11	21	9.4	41	15	4.3	13	0.62	2.2	0.99	2.7	2.7
Eu---	5.1	4.6	0.39	2.5	4.8	2.5	10	3.5	0.93	3.4	0.1	0.47	0.16	0.56	0.84
Gd---	30	26	2	13	26	11	47	19	6	17	1	2.7	1	2.7	3
Tb---	4.2	3.9	0.32	2	3.6	1.7	7.1	2.6	0.77	2.5	0.11	0.34	0.15	0.4	0.46
Dy---	29	29	2.1	13	24	11	42	18	4.6	16	0.96	2.4	1.1	2.5	3.2
Ho---	7.1	6.6	0.43	3.1	5.3	2.5	9.1	3.8	1.2	4.2	0.16	0.58	0.3	0.65	0.68
Er---	23	22	1.3	9.5	16	7.1	26	11	3.4	11	0.65	1.8	0.67	1.9	2.2
Tm---	3	3	0.15	1.3	2.1	0.91	3.5	1.4	0.44	1.5	0.06	0.23	0.12	0.24	0.26
Yb---	21	20	1.4	8.3	13	6	19	8.1	2.8	7.7	0.53	1.5	0.57	1.7	1.7

Fontenelle Creek														Wheat Creek	
Sample	P-32	P-31	P-25	P-23	P-21	P-20	P-19	P-13	P-12	P-6	P-7	P-3	P-1	R-15	R-9
La---	330	70	150	37	140	72	26	85	27	200	95	210	220	160	120
Ce---	83	55	54	49	100	74	37	28	22	62	25	42	52	40	54
Pr---	42	11	18	8.8	23	11	4.2	9.6	3.9	23	10	23	29	15	14
Nd---	190	46	76	41	100	49	17	44	17	98	43	99	130	66	62
Sm---	31	8.7	11	8	18	8.7	2.9	6.4	2.6	14	5.8	13	18	9.5	9.3
Eu---	7.4	1.9	2.4	1.6	4.5	1.8	0.59	1.5	0.63	3.2	1.3	3.5	4.4	2.3	1.8
Gd---	44	9.7	13	7.3	20	7.9	2.5	8.1	2.9	16	7.5	18	22	13	12
Tb---	6	1.4	1.9	1.3	3.1	1.4	0.45	1.2	0.44	2.5	1	2.7	3.1	1.7	1.7
Dy---	39	8.7	12	7	19	8.2	2.6	8.1	2.8	17	6.9	18	21	12	11
Ho---	10	2.1	2.9	1.5	4.5	1.8	0.59	1.8	0.84	4	1.8	4.6	4.6	2.9	2.7
Er---	30	6.7	8.3	4.3	13	5.5	2.1	5.9	1.8	12	5.6	14	14	8.8	8.9
Tm---	3.8	0.93	1.1	0.63	1.7	0.63	0.25	0.72	0.25	1.6	0.69	1.9	1.9	1.2	1.2
Yb---	23	5.9	6.1	3.6	9.6	4	2	4.7	1.7	11	5.2	12	11	6.7	7.6

Wheat Creek

Sample	R-7	R-6	P-34	P-30	P-26	P-22	P-20	P-16	P-14	P-11	P-9	P-7	P-4	P-3	P-1
La---	230	22	210	220	150	54	19	140	160	160	27	110	210	35	130
Ce---	59	7.6	71	68	85	31	19	57	57	43	9.6	54	53	46	32
Pr---	26	3.1	28	24	17	6.4	2.5	16	18	19	2.9	14	21	6.2	13
Nd---	120	16	120	100	73	30	13	63	77	79	13	59	90	27	57
Sm---	17	2.8	17	14	11	4.3	2.2	9.7	12	12	2.3	10	14	5.1	7.8
Eu---	3.9	0.56	3.7	3.2	2.4	1.1	0.4	2.6	2.7	3.2	0.48	2.4	3.4	1	2.1
Gd---	23	3.6	22	18	12	5.3	1.9	13	14	15	3.1	12	19	5	12
Tb---	3.2	0.47	3	2.5	1.9	0.73	0.36	1.8	2.1	2.3	0.4	1.8	2.6	0.63	1.6
Dy---	22	3.1	20	16	13	4.7	1.8	12	14	15	2.6	11	18	4.8	11
Ho---	5.2	0.58	4.3	3.9	2.8	1	0.38	2.6	3.2	3.5	0.55	2.6	4.1	1	2.4
Er---	16	1.9	12	12	8.2	3	1.3	7.8	9.8	11	1.8	7.8	13	3	8.3
Tm---	2.3	0.2	1.6	1.8	1.2	0.46	0.22	1.1	1.3	1.6	0.25	1.2	1.9	0.47	1.2
Yb---	13	1.4	9.2	10	6.7	2.3	1.5	6.7	8.2	10	1.6	6.6	12	3.1	7.8

Wheat Creek

Lakeridge

Sample	Cw-11	E-92C	US-91B	E-90A	E-89A	US-88C	TO-87B	TO-86D	TO-85A	TO-84B	RT-81A	TO-78C	RT-77	RT-75A	F-74
La---	88	11	55	20	44	39	16	11	33	19	50	20	87	340	32
Ce---	41	5.2	25	8.0	18	16	7.7	9.0	27	10	20	12	48	110	17
Pr---	8.7	2.1	10	2.8	6.4	6.5	2.4	1.9	5.6	3.0	5.9	2.9	13	50	4.5
Nd---	36	8.4	41	10	25	25	9.5	8.1	21	11	22	11	47	200	17
Sm---	6	1.6	7.5	1.8	4.4	4.6	1.8	1.4	3.9	2.0	3.5	1.9	7.8	33	3.0
Eu---	1.6	0.34	1.4	0.47	1.2	0.86	0.37	0.29	0.77	0.42	0.98	0.35	1.9	6.8	0.60
Gd---	7.9	1.8	8.9	2.2	5.3	5.3	2.1	1.6	4.3	2.2	4.5	2.2	9.0	38	4.1
Tb---	1.1	0.24	1.3	0.32	0.75	0.76	0.29	0.21	0.62	0.34	0.66	0.31	1.4	5.6	0.60
Dy---	7.3	1.7	7.7	2.1	4.8	5.0	2.0	1.5	4.1	2.5	4.8	2.0	8.5	35	3.7
Ho---	1.6	0.39	1.7	0.50	1.1	1.2	0.48	0.32	0.91	0.57	1.1	0.49	2.0	7.7	0.87
Er---	5.1	1.2	4.5	1.5	3.2	3.3	1.4	1.0	2.6	1.7	3.6	1.5	5.8	22	2.6
Tm---	0.6	0.16	0.56	0.23	0.40	0.45	0.20	0.13	0.36	0.23	0.48	0.21	0.80	2.8	0.36
Yb---	3.1	0.96	3.1	1.4	2.2	2.6	1.2	0.78	2.2	1.5	2.7	1.3	4.4	16	2.2

Lakeridge

Sample	F-73	F-72B	F-71B	F-70E	F-69A	F-88A	F-65B	F-64A	LB-62A	F-58B	LS-57M	LB-57D	LS-56A	LS-55B	R-54C
La---	39	14	26	32	11	52	6.5	13	21	5.3	100	34	22	78	33
Ce---	21	6.1	12	19	7.7	69	7.3	23	39	5.6	28	10	6.1	29	11
Pr---	5.2	1.8	5.2	6.8	1.2	7.5	1.4	3.1	4.7	1.2	16	5.6	3.6	13	5.1
Nd---	20	7.1	21	28	3.8	25	5.6	12	18	5.0	67	22	15	54	20
Sm---	4.0	1.3	3.7	5.3	0.80	4.6	1.1	2.7	3.3	1.2	12	3.7	2.7	10	3.3
Eu---	0.97	0.31	0.70	1.1	0.23	0.75	0.22	0.51	0.59	0.17	2.7	0.77	0.61	2.1	0.79
Gd---	4.3	1.6	4.7	5.5	0.7	4.1	0.9	2.5	3.1	1.3	15	4.6	3.5	12	4.1
Tb---	0.70	0.22	0.58	0.86	0.06	0.72	0.13	0.36	0.52	0.19	2.3	0.71	0.47	1.8	0.56
Dy---	4.7	1.6	3.8	5.4	0.60	4.4	0.83	2.4	3.2	1.2	14	4.7	3.0	11	3.6
Ho---	1.0	0.38	0.79	1.2	0.13	0.95	0.20	0.50	0.68	0.26	3.2	1.2	0.69	2.4	0.89
Er---	3.1	1.1	2.1	3.1	0.36	3.1	0.50	1.5	1.9	0.71	9.0	3.6	1.9	6.7	2.5
Tm---	0.42	0.18	0.29	0.40	0.05	0.49	0.07	0.21	0.31	0.09	1.2	0.58	0.24	0.82	0.35
Yb---	2.7	1.1	1.7	2.2	0.27	3.2	0.38	1.3	1.8	0.55	6.1	4.0	1.3	4.4	2.1

Lakeridge

Sample	M-53B	M-52	R-50F	M-48B	M-47	M-45	M-44A	M-42A	M-41	M-40	M-39B	M-36A	M-37A	M-36B	M-35A
La---	70	600	140	46	220	420	270	430	35	410	230	220	80	12	47
Ce---	22	150	49	44	59	85	100	260	44	140	90	70	41	18	62
Pr---	12	86	24	7.4	25	54	35	64	6.1	44	25	27	9.3	2.3	9.3
Nd---	48	340	95	27	93	200	130	250	22	160	91	100	35	9.2	34
Sm---	8.3	59	16	5.3	15	30	22	43	4.2	24	13	16	5.5	1.7	6.3
Eu---	1.7	13	3.2	0.85	3.7	5.4	5.1	9.3	0.64	5.7	3.2	3.3	1.2	0.30	1.1
Gd---	10	76	19	4.9	18	37	26	48	4.0	32	17	20	6.4	1.4	5.6
Tb---	1.5	11	2.8	0.76	2.6	5.5	3.9	7.1	0.67	4.6	2.5	3.0	0.91	0.20	0.89
Dy---	9.5	68	17	4.7	16	35	25	42	4.3	29	17	19	6.2	1.5	5.6
Ho---	2.3	16	3.8	1.0	3.7	8.5	5.7	9.0	0.95	7.0	4.0	4.7	1.5	0.32	1.2
Er---	6.7	45	10	3.0	11	24	16	25	2.9	20	12	13	4.6	0.98	3.5
Tm---	0.89	5.7	1.2	0.48	1.4	3.1	2.1	3.0	0.44	2.6	1.8	1.8	0.65	0.14	0.51
Yb---	4.9	31	6.5	3.0	7.5	17	12	16	2.7	14	9.1	11	3.9	0.94	3.2

Lakeridge

Sample	M-34A	M-33A	M-32A	M-31A	M-30A	M-29A	M-28A	M-27B	M-25	M-24	M-23B	M-22B	M-21A	M-20B	M-18
La----	73	78	54	28	71	190	110	10	60	74	8.7	47	160	130	160
Ce----	40	72	44	22	43	41	42	13	47	33	5.3	20	30	110	46
Pr---	7.6	14	8.0	4.5	10	19	11	1.9	9.4	8.9	1.2	6.2	16	20	19
Nd----	27	50	28	17	40	73	44	7.2	35	34	4.6	23	62	73	72
Sm---	4.3	8.8	4.8	3.0	6.9	11	7.0	1.5	6.0	5.5	0.88	3.8	9.5	13	11
Eu----	0.88	1.9	0.86	0.65	1.4	2.4	1.5	0.26	1.1	1.2	0.17	0.74	2.1	3.0	2.1
Gd----	4.6	8.7	4.4	3.1	7.4	14	7.9	1.3	5.9	6.0	0.8	4.2	12	14	13
Tb----	0.70	1.4	0.63	0.46	1.1	2.1	1.2	0.18	0.86	0.92	0.11	0.61	2.0	2.2	1.9
Dy---	4.6	8.4	3.8	2.9	6.9	14	7.7	1.2	5.4	5.7	0.80	4.1	13	14	13
Ho----	1.1	1.8	0.88	0.68	1.6	3.5	1.9	0.26	1.2	1.4	0.17	0.99	3.4	3.1	3.1
Er----	3.3	4.8	2.7	1.9	4.6	10	5.7	0.82	3.5	4.4	0.57	2.8	11	8.8	9.3
Tm---	0.46	0.63	0.37	0.26	0.62	1.4	0.79	0.11	0.50	0.60	0.07	0.40	1.4	1.1	1.2
Yb----	2.7	3.7	2.4	1.6	3.9	8.1	4.7	0.73	3.1	3.7	0.44	2.5	8.6	6.5	7.6

Lakeridge

Sample	M-17	M-16	M-15A	M-14	M-12B	WS-11D
La----	190	42	330	140	270	20
Ce----	42	31	64	35	60	34
Pr---	21	7.4	37	22	48	3.9
Nd----	79	29	140	85	190	14
Sm---	12	5.2	20	15	33	2.5
Eu----	2.6	1.0	4.2	3.0	7.0	0.40
Gd----	16	5.3	25	18	38	2.1
Tb----	2.4	0.84	3.7	2.6	5.5	0.31
Dy---	15	5.3	25	17	33	2.1
Ho----	3.9	1.2	6.3	4.2	7.5	0.44
Er----	12	3.6	19	13	21	1.3
Tm---	1.6	0.49	2.6	1.7	2.6	0.21
Yb----	9.9	3.0	16	11	14	1.3

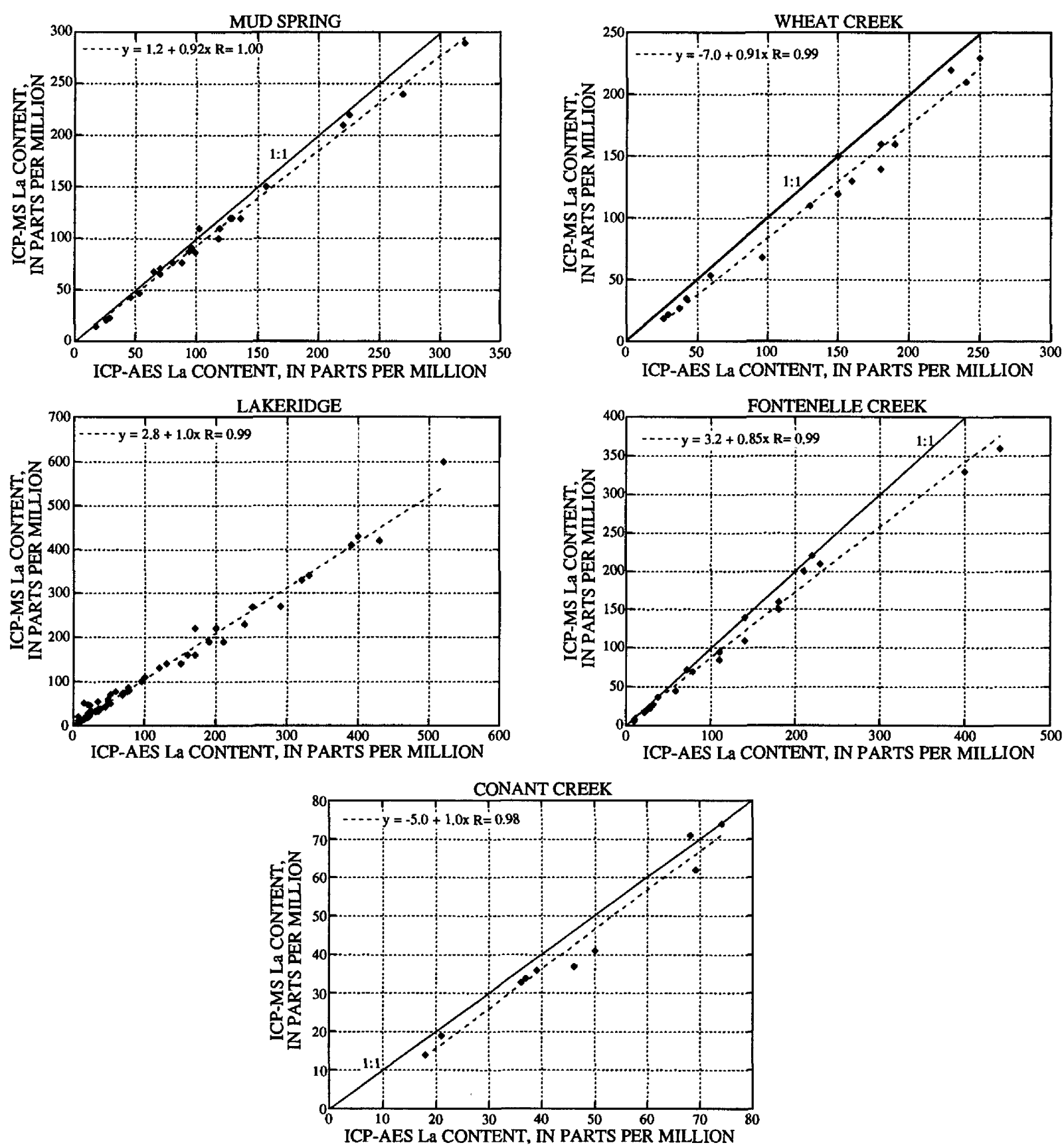


Figure 4. Graphs comparing La content analyzed by inductively coupled plasma-atomic emission spectroscopy (ICP-AES) and by inductively coupled plasma-mass spectroscopy (ICP-MS). Graphs include line of best fit, equation for line of best fit, correlation coefficient (R values) and the one-to-one line (the solid line).

CONSTANCY OF COMPOSITION FOR ROCK COMPONENTS

The components are assumed to have constant composition within a section. Simple x-y plots can demonstrate their constancy of composition. Our interpretation, if two elements of a graph strongly correlate and extrapolate to the origin, is 1) those two elements are almost entirely in that one component, 2) their element to element ratio in that component is relatively constant, and, therefore, the general composition of the component is also constant.

X-y plots, in which the data scatter above a well-defined minimum line that extends to the origin, indicate that one element is entirely in one component and the other element is primarily in the same component, but is also in one or more other components. We interpret the minimum line to represent a constant element to element ratio in the component they have in common, a good indication that the component itself has constant composition. We assume that the data which make up the minimum line represent samples composed of only one component. Given the ubiquitous distribution of all the components, this assumption is not likely valid. Another problem with this interpretation is that it doesn't account for the error in the analyses, which is 5 to 10 percent. Despite these problems the points must at least approach this condition (of both being solely in one component) in order to form a line.

These graphs are used to check variations in component composition between sections.

The detrital component is defined as the sum of all minerals of terrigenous origin, and is uniquely defined by Al_2O_3 (Isaacs, 1980). Graphs of K_2O , TiO_2 and Fe_2O_3 versus Al_2O_3 are used to evaluate the composition of the detrital component (fig. 5). A strong correlation exists within each section between Al_2O_3 and both K_2O and TiO_2 . The correlation coefficients for the K_2O - Al_2O_3 graph range from $R = 0.96$ to 0.99 , except Mud Spring, for which $R = 0.87$. The correlation coefficients for TiO_2 and Al_2O_3 range from $R = 0.93$ to 0.97 , except Mud Spring, for which $R = 0.79$. The Fe_2O_3 - Al_2O_3 graph

shows a well defined minimum line. All the graphs support the idea that 1) these elements are almost entirely in the detrital component and 2) relatively constant composition exists within each section. Fe_2O_3 data, which scatters above a minimum line, must be in another component such as pyrite or iron oxide.

Small differences in composition in the detrital component exist between sections. The most pronounced difference is represented by the $\text{K}_2\text{O}/\text{Al}_2\text{O}_3$ ratio (0.62), for Conant Creek, the easternmost section (fig. 5). It is significantly higher than at the other locations, whose $\text{K}_2\text{O}/\text{Al}_2\text{O}_3$ ratios range from 0.23 to 0.31. Small differences in the TiO_2 content exist between sections, represented by the $\text{TiO}_2/\text{Al}_2\text{O}_3$ ratio, which ranges from 0.018 to 0.044 (fig. 5). A couple of factors possibly cause these differences. (1) The proportions of the minerals which compose the detrital component might vary as a function of distance from source. (2) The composition of the minerals that comprise the detrital component could vary as a result of different sources (Peterson, 1980). Or (3) it could be analytical error. It is probably a result of a combination of these factors. However the variations are small.

Geologic evidence supports a constant terrigenous composition. Paleogeographic reconstructions of the area in which the Phosphoria Formation was deposited remained relatively unchanged for at least 10 m.y. years and continued unchanged into the Triassic with the deposition of the Dinwoody Formation (McKelvey, 1959). The only difference between the two formations was the lack of a significant biogenic input during the deposition of the Dinwoody (Peterson, 1980). The Phosphoria Formation, as was mentioned earlier, had a very low sedimentation rate. This was the result of being surrounded by other basins. The only sources of terrigenous detritus was the Antler Orogenic belt to the west, which had been eroded to low relief by Permian time (Cook, 1988), and the Lemhi uplift to the north, the source of the Shedhorn Sandstone (Peterson, 1980).

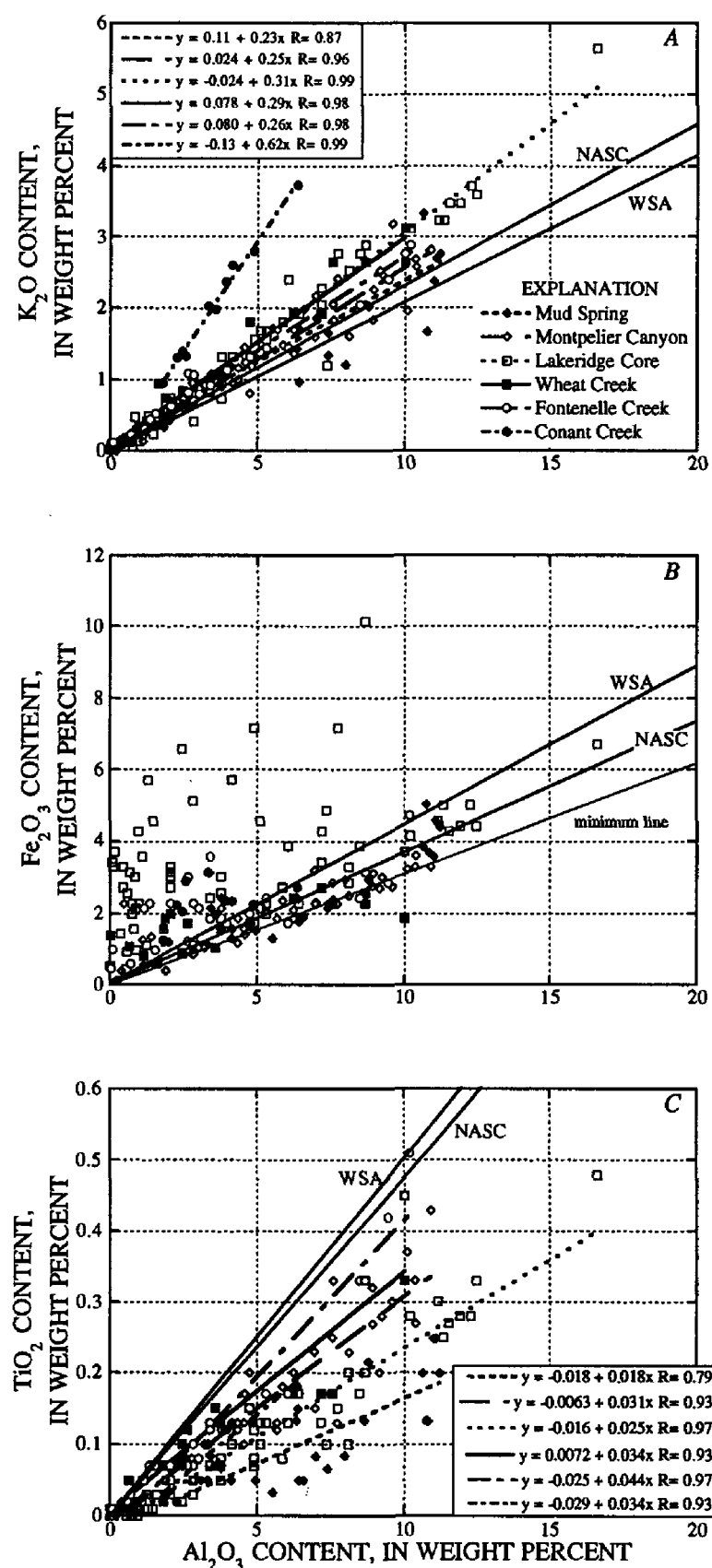


Figure 5. TiO₂ (A), Fe₂O₃ (B), and K₂O (C) versus Al₂O₃ content, as determined by inductively coupled plasma-atomic emission spectroscopy. The line for the world shale average (WSA) and North American shale composite (NASC) are shown for comparison purposes. Legend shown in figure 5A. Note the K₂O content at Conant Creek is higher than at all other sections and the Lakeridge samples have excess iron which scatters above the minima.

The graph of CaO versus P_2O_5 , used to evaluate constancy of apatite composition, shows a well-defined minimum, which intersects the origin (fig. 6). P_2O_5 in the apatite fraction dominates the total P_2O_5 inventory in all samples. The scatter above the minimum is attributed to CaO in dolomite and calcite, and less so to CaO in the detrital component. The minimum line is essentially the same as the line representing the CaO/ P_2O_5 ratio in stoichiometric apatite (McClellan and Lehr, 1969), an indication that the apatite component is clearly the dominant CaO-containing component in some of the rocks and that the composition of apatite is constant.

The MgO versus CO_2 graph shows most of the data along the line of stoichiometric dolomite ($MgCa(CO_3)_2$), demonstrating that the dolomite composition at all locations is relatively constant (fig. 7A). A few samples from Montpelier Canyon are exceptions. Their CaO/ CO_2 and MgO/ CO_2 ratios indicate they contain calcite, rather than dolomite.

Scatter of the Lakeridge data is interpreted as a result of a systematic analytical error. Data above the dolomite line were one analytical batch and those below the line another batch. The difference in MgO analyses between batches is confirmed by 35 samples that were analyzed by both XRF and ICP. The graph of ICP-AES versus XRF MgO data shows that the ICP-AES data is 13 percent lower than the XRF data (fig. 3D).

The graph of CaO and CO_2 sample data was considered as a possible gauge for the composition of calcite (fig. 7B). The samples with significant amounts of CO_2 plot either above the line for stoichiometric dolomite or below the line for stoichiometric calcite, except for the few samples from Montpelier Canyon. This observation in conjunction with the MgO/ CO_2 graph, indicates that the bulk of the carbonates at these five sections is dolomite. As a result, calcite could not be evaluated for constancy of composition. It was assumed to be stoichiometric ($CaCO_3$), with insignificant MgO and minor-element substitution (Gulbrandsen, 1960b).

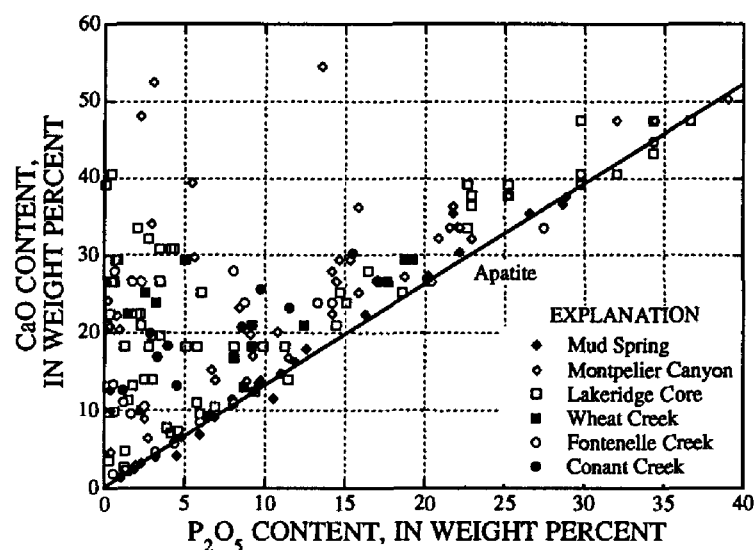


Figure 6. CaO versus P_2O_5 content, as determined by inductively coupled plasma-atomic emission spectroscopy.. The line represents the ratio of CaO: P_2O_5 in stoichiometric apatite.

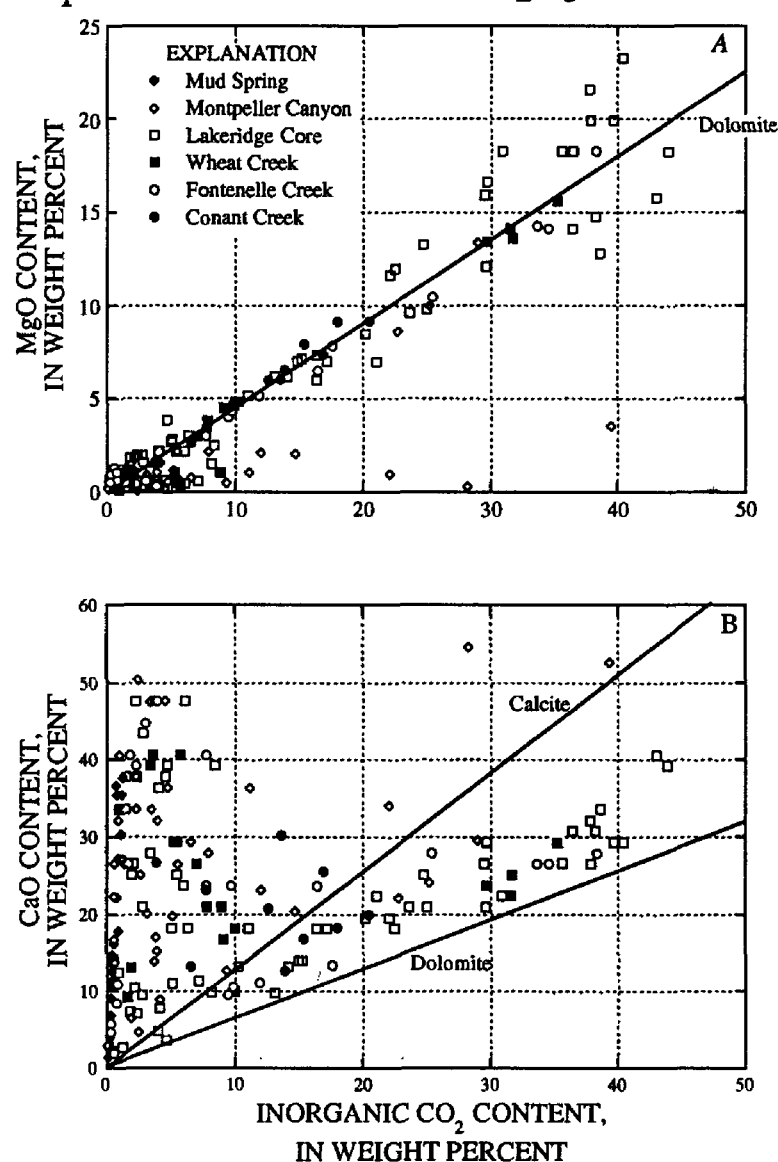


Figure 7. MgO (A,) and CaO (B) versus inorganic- CO_2 content. Legend shown in figure 7A. In figure 7A, the line represents the MgO: CO_2 ratio in dolomite with a composition of $MgCa(CO_3)_2$ and in figure 7B, the lines represent the CaO: CO_2 ratio in dolomite and calcite with a composition of: $MgCa(CO_3)_2$ and $CaCO_3$, respectively.

The composition of organic matter could not be checked, but published accounts suggest a relatively constant composition. The average C content of kerogen was 68.5 weight percent with a $\sigma = 6.0$ (Powell and others, 1975).

By graph or by published accounts, the components of the Phosphoria Formation are relatively constant in composition. Even the detrital component which is composed of several minerals shows little variation, possibly a reflection of relatively constant depositional conditions for 10 million years.

PROPORTIONALITY CONSTANTS FOR ROCK COMPONENTS

Rock components were calculated using the formulas in table 5. The formulas differ slightly from those previously used for the Montpelier Canyon section (Medrano and Piper, 1992). These changes reflect the larger data base. The new proportionality constants were determined through an iterative process, whereby each calculated component was plotted against the sum of the components with the aim of obtaining closure for the sum of components to 100 percent and a random distribution of each component about 100 percent. The changes from the previous calculations include the following:

Table 5. Formulas used to calculate the normative components.

Component	Formula
Detrital component	$6.0 \times \text{Al}_2\text{O}_3$
Dolomite	$(\text{MgO} - 0.07 \times \text{Al}_2\text{O}_3) / 0.22$
Calcite	$(\text{CaO} - 0.15 \times \text{Al}_2\text{O}_3 - 0.304 \times \text{Dolomite} - 0.555 \times \text{Apatite}) / 0.56$
Biogenic silica	$\text{SiO}_2 - 3.5 \times \text{Al}_2\text{O}_3$
Apatite	$(\text{P}_2\text{O}_5 - 0.01 \times \text{Al}_2\text{O}_3) / 0.41$
Organic matter	$\text{Organic C} \times 1.4$
Calculated CO_2	$0.48 \times \text{Dolomite} + 0.01 \times \text{Apatite} + 0.44 \times \text{Calcite}$

1) The MgO and CaO content in the detrital component is 1.2 and 1.4 weight percent lower, respectively.

2) The Al_2O_3 /detrital component proportionality constant is reduced from 6.4 to 6.0.

3) The formula for dolomite from the Lakeridge core was adjusted to account for a systematic analytical error in the MgO ICP data, as mentioned earlier. They are: $(\text{MgO} - 0.1 \times \text{Al}_2\text{O}_3) / 0.27$, if the MgO/CO_2 ratio > 0.48 ; $(\text{MgO} - 0.1 \times \text{Al}_2\text{O}_3) / 0.25$, if $0.48 \geq \text{MgO}/\text{CO}_2$ ratio > 0.41 ; and $(\text{MgO} - 0.1 \times \text{Al}_2\text{O}_3) / 0.23$, if the MgO/CO_2 ratio ≤ 0.41 . At Mud Spring there was no calcite and ≤ 2.3 weight percent dolomite.

4) The $\text{SiO}_2:\text{Al}_2\text{O}_3$ ratio in the detrital component, the proportionality constant in the formula for biogenic silica (table 5), was reduced from 3.75 to 3.5. The effect is to increase the biogenic silica.

5) The organic carbon/organic matter proportionality constant is reduced from 1.7 to 1.4. This change is based on chemical analyses of organic matter (kerogen) in Phosphoria Formation from published literature (Powell and others, 1975).

The results are listed in table 6 and are presented as bar graphs (fig. 8) to allow for comparison of the five locations.

Figure 8. Bar graphs showing calculated abundance of the terrigenous component, dolomite, calcite, biogenic/diagenetic silica, apatite and organic matter in each sample analyzed for all the sections; Mud Spring (A), Conant Creek (B), Wheat Creek (C), Fontenelle Creek (D), Lakeridge (E). Legend is given in figure 8A.

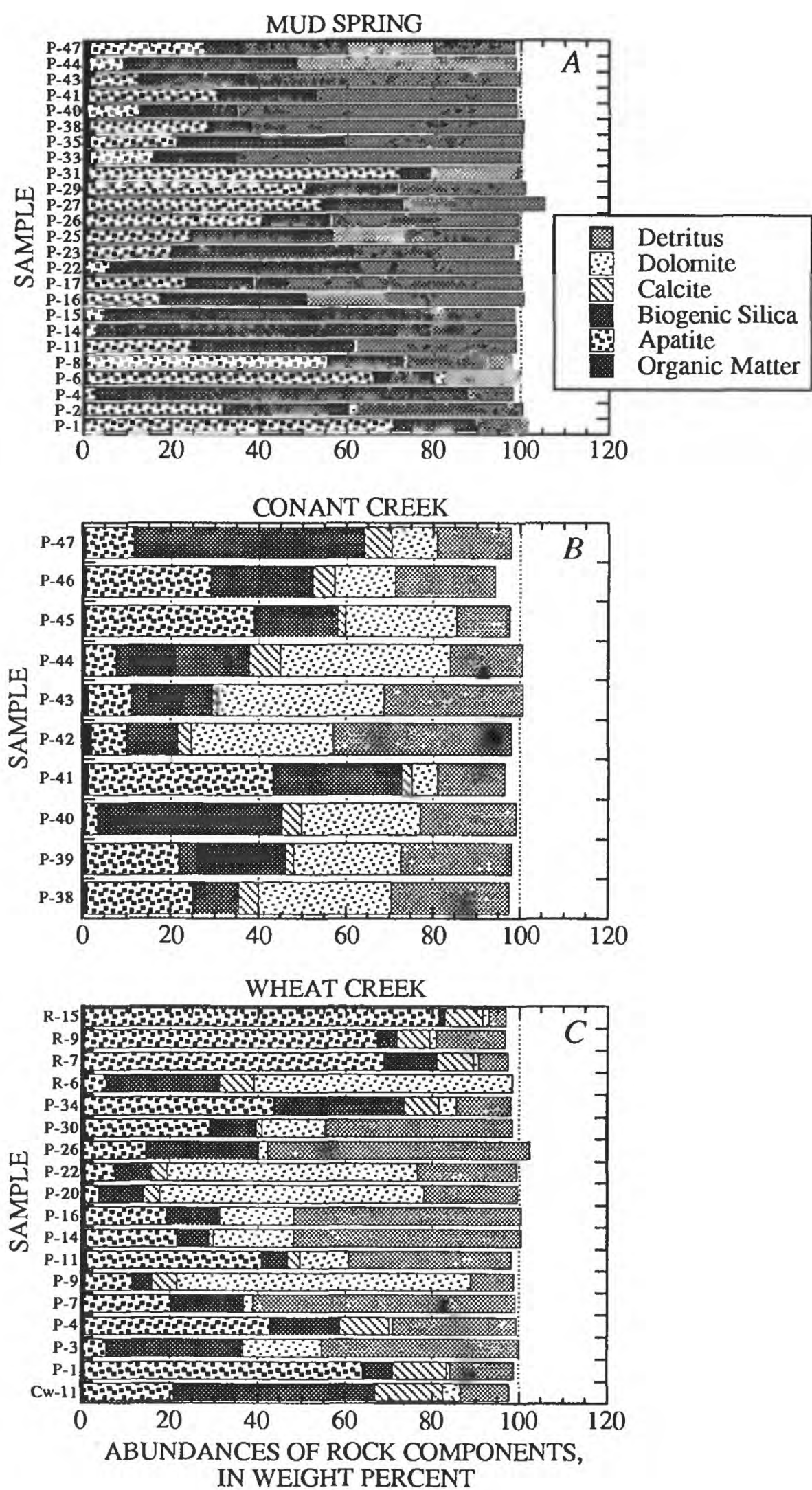


Figure 8

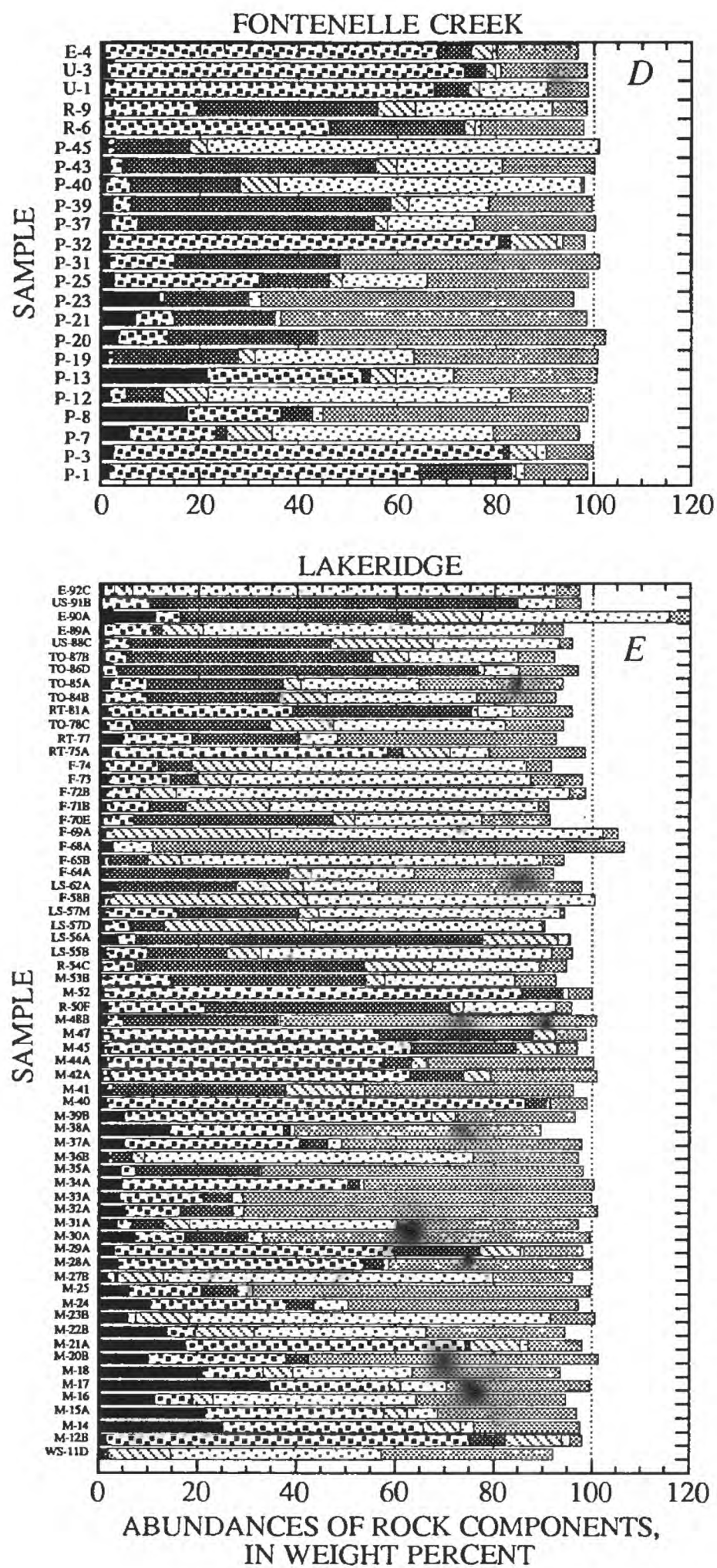


Figure 8. continued

Table 6. Component abundances in samples of the Phosphoria Formation from five sections: Conant Creek, Mud Spring, Fontenelle Creek, Wheat Creek and Lakeridge

[All values in weight percent]

Conant Creek							
Sample	Detrital component	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Total
P-47	18	11	6.1	53	11	0.6	99
P-46	24	15	4.8	23	28	0.8	96
P-45	13	27	1.5	19	38	0.7	99
P-44	17	41	6.2	30	6.8	0.6	102
P-43	33	40	0.3	18	9.5	1.4	103
P-42	43	34	2.3	12	7.8	2.1	100
P-41	16	6.2	2.6	30	42	1.3	97
P-40	23	28	3.6	42	2.5	0.8	100
P-39	27	26	1.3	24	21	0.9	100
P-38	28	32	3.8	11	24	1.2	99

Mud Spring							
Sample	Detrital component	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Total
P-47	65	0	0	9.3	25	1.7	101
P-44	53	0.4	0	39	7.5	1.4	101
P-43	66	0.3	0	24	11	1.6	102
P-41	48	0	0	22	29	1.3	100
P-40	67	0.5	0	22	11	0.9	102
P-38	65	0.2	0	9.7	27	1.4	103
P-35	42	0.5	0	38	19	1.5	102
P-33	67	0.5	0	19	14	1.5	102
P-31	20	1.3	0	7.0	70	1.1	100
P-29	30	1.0	0	21	49	0.8	102
P-27	33	0.8	0	19	53	0.7	106
P-26	44	1.0	0	16	39	0.6	101
P-25	44	0.3	0	32	23	0.5	101
P-23	38	0	0	42	19	0.5	100
P-22	38	0.5	0	57	5.4	0.4	101
P-17	64	0.6	0	15	22	0.5	103
P-16	52	0.4	0	33	16	0.6	103
P-15	23	0	0	72	4.5	0.5	99
P-14	21	0.3	0	76	2.3	0.5	100
P-11	38	0.9	0	37	23	0.6	100
P-8	25	1.1	0	18	54	0.7	98
P-6	19	2.3	0	14	65	0.8	100
P-4	11	0.2	0	85	2.4	0.3	99
P-2	39	2.3	0	29	31	0.7	102
P-1	12	0.4	0	19	70	0.4	102

Table 6. Continued

Fontenelle Creek							
Sample	Detrital component	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Total
E-4	16	1.2	0	7.0	72	0.8	97
U-3	17	1.1	0	4.4	78	1.2	102
U-1	8.0	15	0	7.0	73	0.7	103
R-9	6.9	29	5.4	36	21	0.7	100
R-6	20	1.3	0	28	50	0.7	99
P-45	0.2	83	3.4	15	1.5	1.5	105
P-43	18	22	4.3	51	2.7	2.0	101
P-40	0.5	64	7.1	22	5.6	1.1	101
P-39	20	17	3.2	52	3.9	2.5	100
P-37	24	18	2.0	48	6.0	2.2	100
P-32	4.3	1.3	0	2.5	89	1.6	99
P-31	51	0.1	0	33	14	1.9	101
P-25	32	18	0	14	32	2.8	99
P-23	61	2.6	0	17	1.1	12	94
P-21	60	1.1	0	20	7.6	7.3	97
P-20	57	0.0	0	30	10	3.7	101
P-19	36	34	3.0	25	1.1	1.6	101
P-13	28	12	0.5	1.6	34	22	99
P-12	16	64	8.0	7.5	4.1	2.0	102
P-8	52	2.1	0	6.4	19	18	98
P-7	17	47	4.6	2.4	20	6.1	96
P-3	9.2	2.1	0	1.3	84	2.7	99
P-1	12	1.8	0	19	67	1.9	102
Wheat Creek							
Sample	Detrital component	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Total
R-15	4.0	1.4	0	1.4	89	0.6	97
R-9	16	1.5	0	4.5	73	0.5	95
R-7	6.8	1.2	0	12	78	0.5	99
R-6	0.1	62	5.3	26	6.1	0.6	100
P-34	12	4.2	3.2	30	47	0.6	97
P-30	43	15	0	11	30	0.9	100
P-26	60	2.2	0	26	15	0.4	104
P-22	23	60	1.3	8.3	7.7	0.8	101
P-20	22	63	1.5	10	3.4	0.7	100
P-16	52	18	0	12	19	0.8	102
P-14	52	19	0	7.3	22	1.1	102
P-11	37	12	0	6.0	43	1.2	99
P-9	10	70	1.7	4.5	12	0.8	100
P-7	60	2.2	0	17	21	0.6	101
P-4	28	0.8	5.5	16	46	0.8	97
P-3	45	19	0.5	31	5.4	0.6	101
P-1	15	0.9	0	7.0	73	0.9	96
Cw-11	11	4.2	13	46	22	0.6	97

Table 6. Continued

Lakeridge							
Sample	Detrital component	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Total
E-92C	4.9	86	3.7	0	2.0	0.9	97
US-91B	5.2	7.7	0.2	75	9.5	0.5	98
E-90A	4.4	38	14	47	4.7	11	120
E-89A	5.9	67	8.3	1.8	10	0.8	94
US-88C	2.7	26	21	41	5.2	0.8	96
TO-87B	7.7	22	7.2	50	4.2	1.2	92
TO-86D	12	7.2	1.2	73	3.1	0.6	97
TO-85A	31	24	3.6	28	7.1	2.3	95
TO-84B	17	30	9.6	27	8.3	1.2	93
RT-81A	12	7.2	1.5	36	37	1.7	96
TO-78C	12	35	13	28	5.4	1.3	95
RT-77	46	7.6	0	22	14	4.7	94
RT-75A	20	8.0	10	3.1	55	2.4	99
F-74	5.3	52	16	6.8	11	1.2	92
F-73	11	61	6.9	5.6	12	1.9	98
F-72B	3.4	80	7.4	0	6.7	1.4	99
F-71B	2.2	55	17	7.5	8.4	1.5	91
F-70E	15	26	4.5	40	6.1	0.9	92
F-69A	3.3	67	33	0	0	1.4	105
F-68A	100	8.0	0	0	0.2	2.7	111
F-65B	4.5	73	6.5	7.9	0.9	1.1	94
F-64A	29	21	4.4	37	0.4	0.8	93
LS-62A	43	15	13	24	0.2	3.3	99
F-58B	0.1	58	40	0	0.9	1.2	100
LS-57M	0.9	49	3.9	25	15	1.1	94
LS-57D	0.6	47	29	7.1	5.1	1.1	90
LS-56A	0.5	2.1	16	70	3.6	3.8	96
LS-55B	4.5	59	7.1	16	8.4	1.4	96
R-54C	5.8	22	14	47	6.7	0.5	95
M-53B	8.7	27	3.9	39	14	0.5	93
M-52	5.0	1.0	0	8.5	84	1.0	99
R-50F	3.4	19	2.8	50	20	1.8	96
M-48B	67	0.4	0	32	3.1	1.6	104
M-47	6.5	0.1	4.9	31	55	0.8	99
M-45	4.0	0.4	8.7	22	61	1.0	97
M-44A	35	0.4	3.5	5.8	55	1.7	102
M-42A	23	0	6.0	11	61	0.8	102
M-41	44	2.9	13	35	1.2	1.5	98
M-40	7.5	0.4	1.4	4.4	84	1.7	99
M-39B	25	0.6	5.3	0	61	5.4	98
M-38A	52	0.8	0	1.4	23	15	92
M-37A	51	2.7	0	5.6	35	5.4	100
M-36B	23	66	2.4	4.3	0.6	2.1	98

Table 6. Lakeridge continued

Sample	Detrital component	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Total
M-35A	68	0.4	0	25	2.7	4.7	101
M-34A	49	0.9	0	2.5	45	4.8	102
M-33A	74	2.2	0	6.0	16	4.5	103
M-32A	75	2.1	0	11	11	5.5	104
M-31A	39	42	5.2	6.5	2.9	3.9	99
M-30A	69	3.1	0	13	9.8	7.5	102
M-29A	12	0.7	8.8	17	56	3.4	99
M-28A	43	1.1	0	4.0	49	4.1	101
M-27B	17	67	9.2	0.7	1.4	1.9	97
M-25	71	3.0	0	7.5	14	6.3	102
M-24	49	6.8	0	5.7	27	11	99
M-23B	10	73	11	0	1.6	6.0	101
M-22B	29	34	12	0	5.4	14	96
M-21A	11	1.6	11	1.1	56	18	98
M-20B	61	0	0	4.5	28	10	103
M-18	32	24	6.1	0	12	21	95
M-17	31	9.2	2.5	0	24	35	101
M-16	32	41	4.2	0.3	7.1	12	96
M-15A	29	6.2	5.0	0	36	22	98
M-14	23	2.7	7.7	0	40	25	99
M-12B	2.6	1.7	12	7.6	73	1.7	98
WS-11D	36	43	12	1.0	0.2	1.0	94

A critical step in the process of calculating the components is the calculation of the detrital component using a proportionality constant extracted from the residual versus Al_2O_3 graph (fig. 9). The residual is the difference between 100 percent and the sum of the apatite, dolomite, calcite and organic matter. Therefore, the detrital component or terrigenous fraction depends on the correct calculation of the residual components. The components, apatite, dolomite, and calcite, three of the four components of the residual, were independently checked by graphing measured CO_2 versus calculated CO_2 , where the calculated CO_2 = sum of CO_2 in calcite, dolomite, and apatite (fig 10; Gulbrandsen, 1960a, b, 1970). The correlation coefficient for data from all the sections is $R \geq 0.99$, except for Mud Spring, which was 0.89.

The overall calculations of the components are also checked by the approach of the sum of the components to 100 percent. The average sum for all the sections is 99.6 percent and the average standard deviation is 2.7 percent. The sum of each section is in table 7.

Table 7. The average of the sum of the rock components for each section and its standard deviation.

Section name	Average	Standard deviation
Conant Creek	99.5	2.2
Fontenelle Creek	99.7	2.4
Wheat Creek	99.3	2.3
Lakeridge	98.1	4.7
Mud Spring	101.2	1.7

The sum of the Lakeridge core components does not approach 100 percent as closely as the other sections. One possible reason is the high S (up to 10 weight percent) and Fe, which can not be completely accounted for by organic matter (Powell and others, 1975) or apatite (Compton and others, 1993), as it is at the other sections. Pyrite was identified in many of the samples, but there was no clear correlation between excess sulfur ($\text{Total S} - 0.27 \times \text{Organic C} - 0.077 \times \text{P}$) and excess Fe ($\text{Total Fe}_2\text{O}_3 - 0.35 \times$

Al_2O_3) (fig. 11). Thus, pyrite could not be calculated as a component in the Lakeridge core. Iron and sulfur in these rocks clearly needs further study.

The small amount of scatter around 100 percent for the sum of the components and the one to one line in the graph of calculated CO_2 versus measured CO_2 indicates the validity of calculating marine and detrital components based on major-element-oxide analyses. The small amount of scatter further suggests little variation in the composition of the individual components between samples in this data set.

Figure 9. Residual versus Al_2O_3 content. The residual is the difference between 100 and the sum of the components: dolomite, calcite, apatite, and organic matter. The slope of the line is the proportionality constant used to determine the detrital component from Al_2O_3

Figure 10. Carbonate content, estimated from calculated abundance of dolomite, calcite, and apatite, versus carbonate content measured by induction-furnace carbon analysis. Shown are the best fit curves and their corresponding correlation coefficients.

Figure 11. Fe_2O_3 content minus that in the terrigenous component versus S content adjusted for that in organic matter and apatite for Lakeridge samples. Lines represent the Fe_2O_3 :S ratio in amorphous FeS and pyrite.

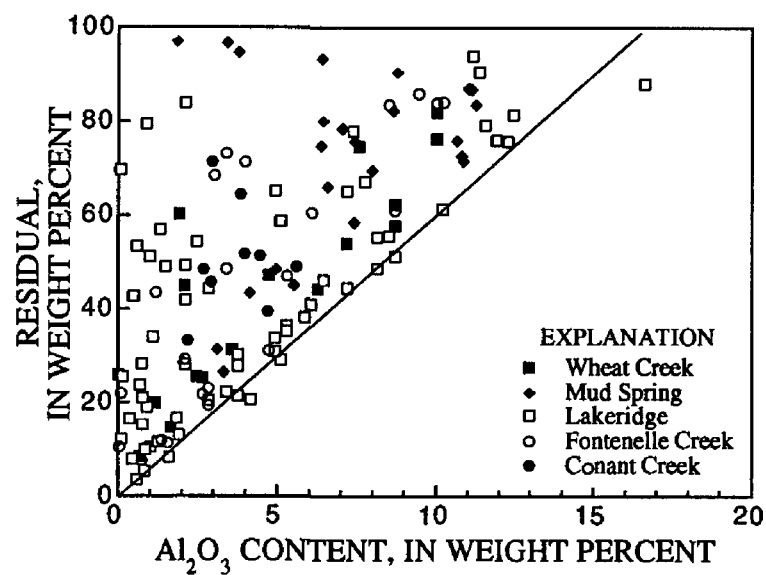


Figure 9

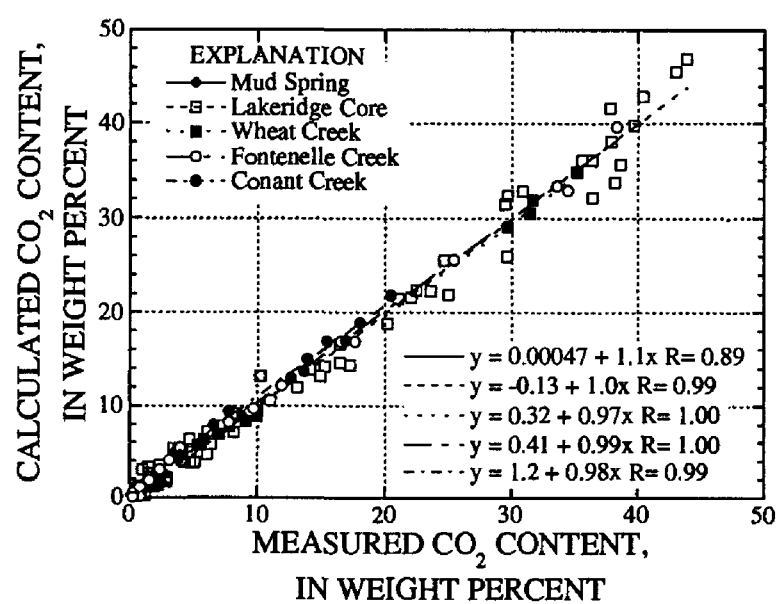


Figure 10

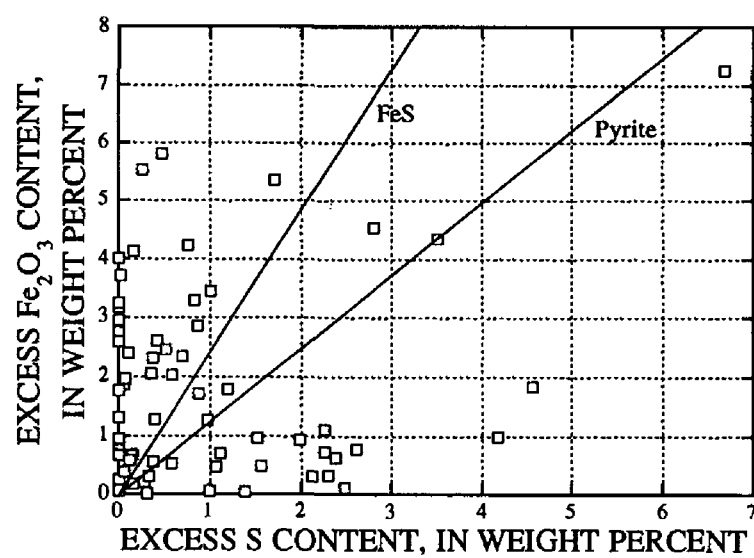


Figure 11

DETERMINATION OF THE TERRIGENOUS MINOR ELEMENT CONTENT

Assuming all the Al_2O_3 in the samples is contained in the terrigenous fraction and the Al_2O_3 content is constant, we conclude that minor elements that correlate strongly with Al_2O_3 are also contained solely in the terrigenous fraction and their content constant. Samples from Conant Creek and Mud Spring show Sc correlates strongly with Al_2O_3 , where $R = 0.93$ and 0.95 respectively, both having near origin intercepts (fig. 12O, 13D), despite the fact that Sc content only ranges from 2 ppm (the detection limit) to 13 ppm. The other sections have a slight scatter above the minimum line, but also correlate with Al_2O_3 .

Ga, again, at Conant Creek and Mud Spring correlates strongly with Al_2O_3 , where $R = 0.96$ and 0.91 respectively, and have near origin intercepts (fig. 12I, 13B), despite the Ga content in these samples being very close to the detection limit of 4 ppm. The other sections have a small amount of scatter above the minimum line, but also correlate with Al_2O_3 .

Li correlates with Al_2O_3 at all the sections, where R ranges from 0.68 to 0.86 , with the exception of Mud Spring, where $R = 0.22$ (fig. 12J, 13C), owing mostly to the line of best fit approaching a horizontal line. These trends closely follow the trends in the K_2O versus Al_2O_3 graph (fig. 5A), where Conant Creek has the highest trend and Mud Spring the lowest. This results possibly from the substitution of Li for K in clays. The Li plot shows that, whereas the Li content is constant in the terrigenous fraction within each section, there appears to be some small variation between sections, similar to that found for K_2O . However, the variation is small considering the extent of the area of deposition and the time span over which the Phosphoria Formation was deposited.

The Li content of the Lakeridge samples has the greatest amount of scatter; it ranges between the Mud Spring and Conant Creek data. This greater range of values in comparison to other sections is similar for other minor elements from the Lakeridge core. It possibly results from these samples containing a wider range of lithologies and

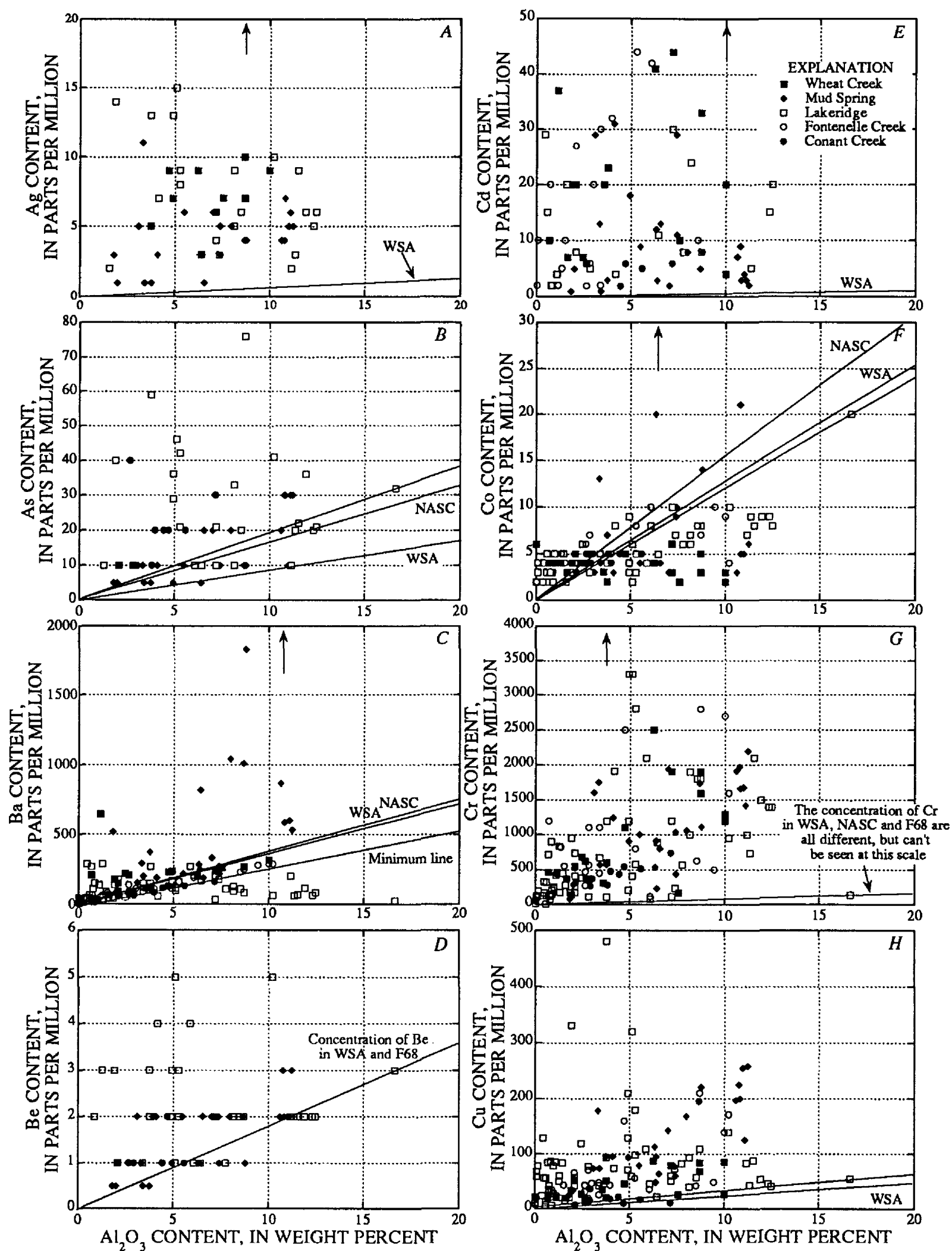


Figure 12

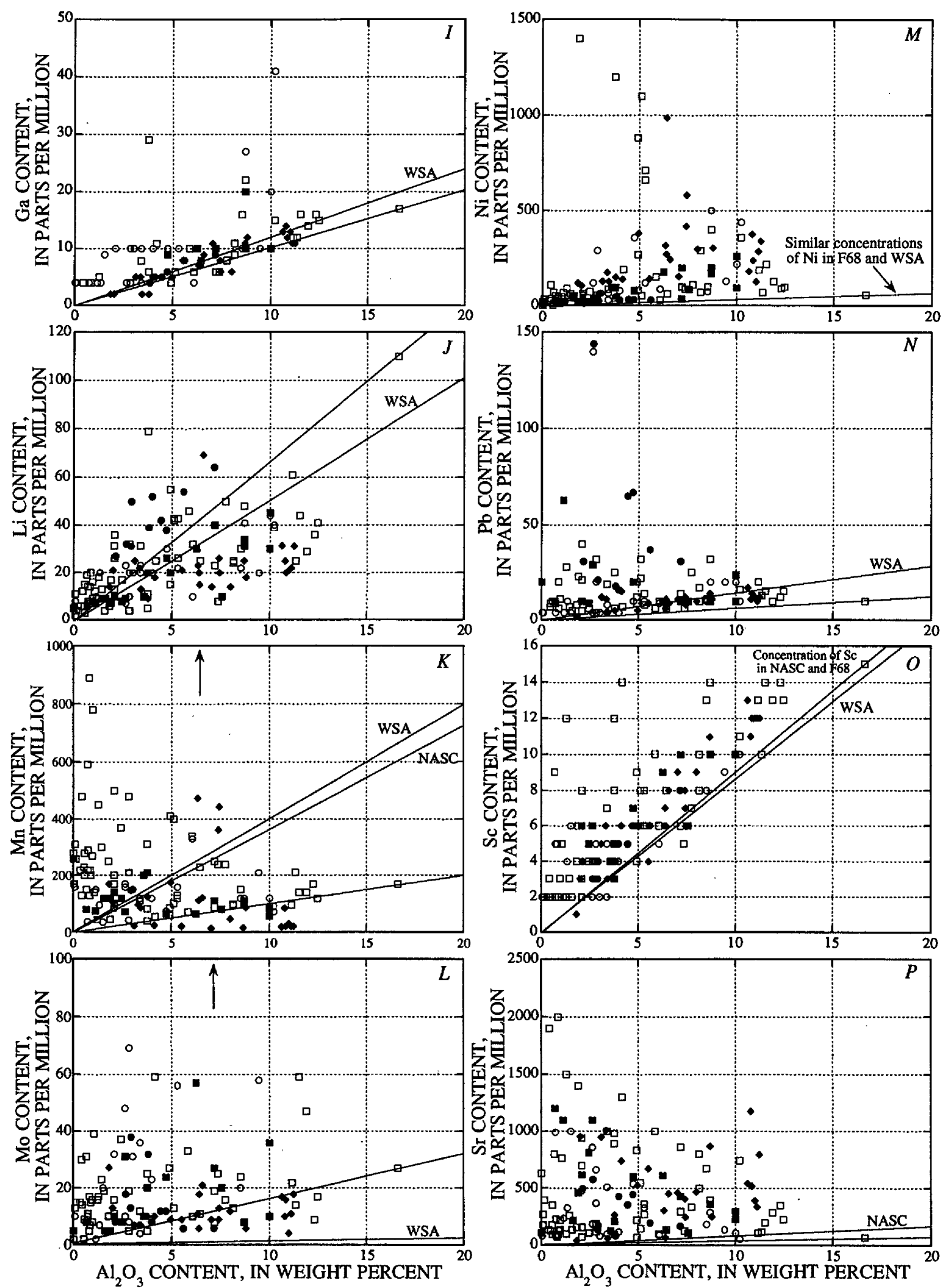


Figure 12. continued

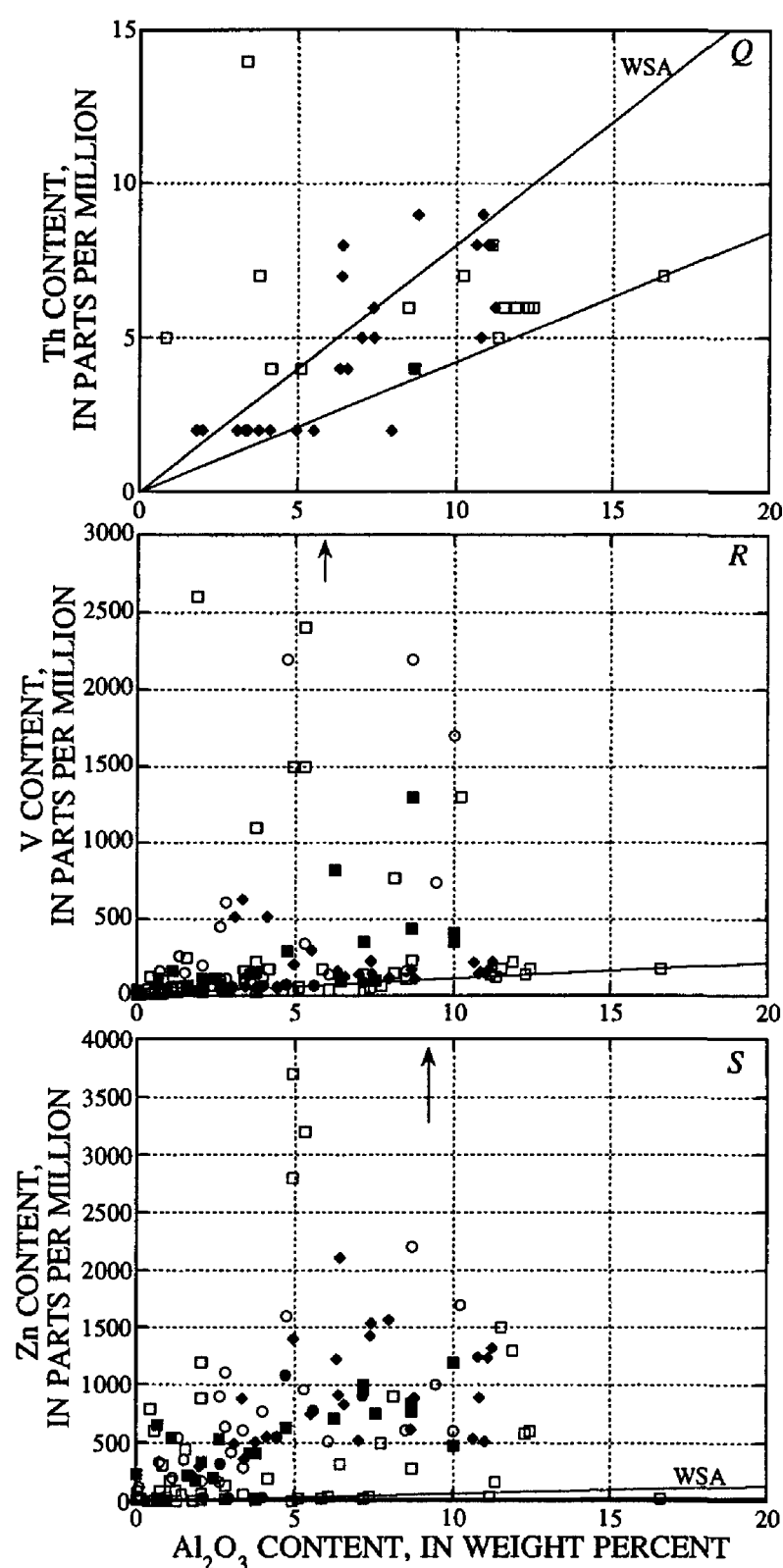


Figure 12. Minor-element versus Al_2O_3 content, as determined by inductively coupled plasma-atomic emission spectroscopy: A) Ag; B) As; C) Ba; D) Be; E) Cd; F) Co; G) Cr; H) Cu; I) Ga; J) Li; K) Mo; L) Ni; M) Pb; N) Sc; O) Sr; P) Th; Q) V; R) Y; S) Zn.

Lines drawn represent the minor element: Al_2O_3 ratios in the world shale average (WSA) and the North American shale composite (NASC). The unmarked line is either the minimum line for this graph relative to the Al_2O_3 content or is a line which intersects the F68a data point. The legend is given in figure 9E. Samples with minor-element content below the detection limit are not included. The arrows pointing up indicate data for some samples that are outside the limits of the graph. Limits of the graphs were chosen so as to include as much data as possible, yet show the minimum relative to the Al_2O_3 content.

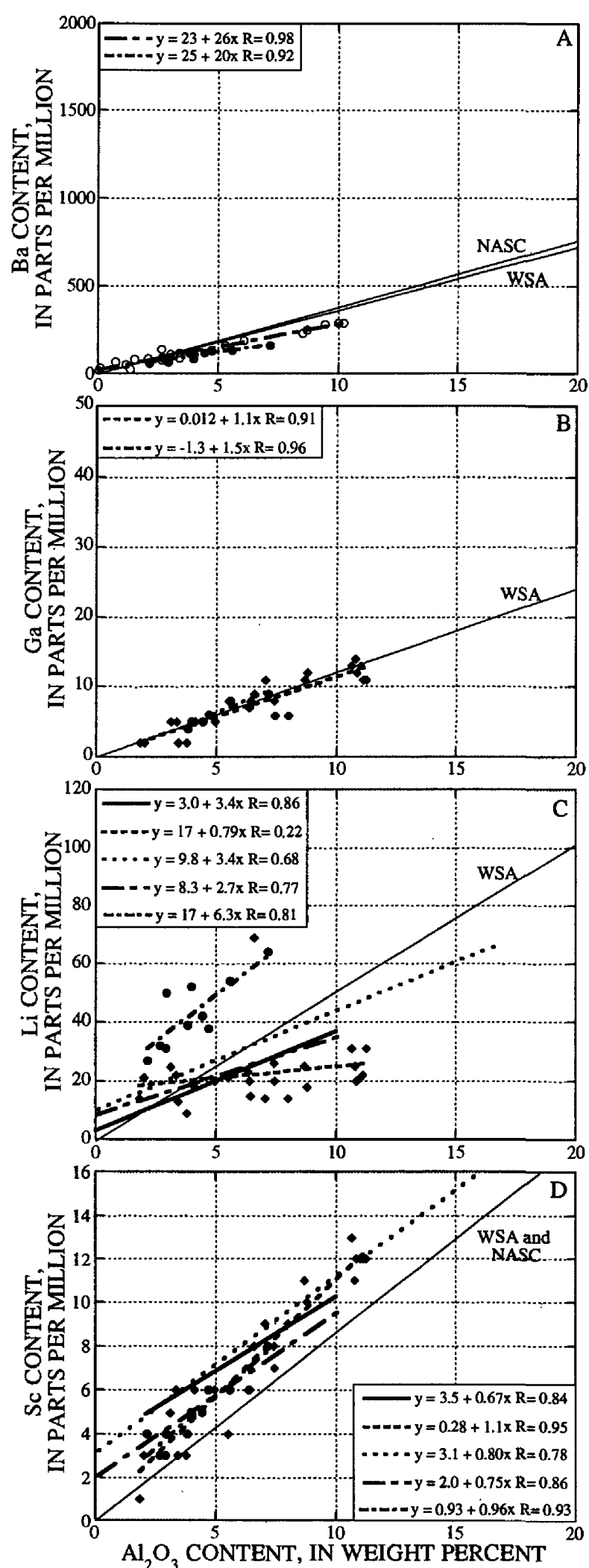


Figure 13. Selected sections of Ba, Ga, Li and Sc content versus Al_2O_3 content, which exhibit a strong correlation: (A) Ba content at Fontenelle Creek and Conant Creek, with their lines of best fit, and lines representing Ba content in WSA and NASC; (B) Ga content at Mud Spring and Conant Creek, with lines of best fit and line representing Ga content in WSA. C) Li content at Conant Creek and Mud Spring, with lines of best fit and equations for all sections and line representing Li content in WSA. D) Sc content at Mud Spring and Conant Creek, with lines of best fit and equations for all sections, and lines representing Sc content in WSA and NASC. Legend in figure 9E.

representing all the members of the Phosphoria Formation, unlike the other sections, which are primarily from the lower Meade Peak Phosphatic Shale member. Also the Lakeridge core is unweathered, which may have effected the minor element content relative to the other sections, which are from surface outcrops or trenches.

Ba and Al_2O_3 have correlation coefficients at Conant Creek and Fontenelle Creek of $R = 0.92$ and 0.98 respectively (fig. 12C, 13A). Wheat Creek scatters slightly above but has a minimum line which coincides with the line of best fit for Conant Creek and Fontenelle Creek. Some of the data from Mud Spring scatters well above the minimum line and Lakeridge has a small group of samples that lie below the minimum line.

Studies of modern sediment show strong correlations between the terrigenous fraction of the sediment and Ba, Li, Ga, and Sc, and less so Th and Co, and (Piper and Isaacs, 1995, Piper and others, in press). Here in the Phosphoria Formation, Li, Ga and Sc also show strong correlations despite some scatter from a couple of sections and despite the Ga and Sc data ranging to near the detection limit. Ba also shows a strong correlation, with the exception of Mud Spring and Lakeridge sections. Th and Co were below their detection limit in too many cases to be used as indicators of constant composition. Clearly though, for the elements, Ba, Li, Ga, and Sc, the graphs indicate they are primarily in a terrigenous fraction of relatively constant composition. We conclude from these relations that all minor elements have a constant concentration in the terrigenous fraction.

The minor-element contents in the terrigenous fraction are determined by one of three possible methods:

- 1) Minimum line method: the minimum line from the minor element, Al_2O_3 graphs, a line formed by a group of points which have the least amount of a given minor element relative to Al_2O_3 , and which intersects the origin, can be used to determine the minor element content of the terrigenous fraction. This method is best used when the minor element has only a slight marine enrichment or forms a well defined minimum line

such is seen in the graphs of Pb and V (fig. 12*N, R*). However, for minor elements, like Cr, Mo, and Ni, the data forms no well defined minimum line (fig. 12*G, L, M*). One problem with this method, which was discussed earlier, is the assumption that the samples which form the minimum line have solely a terrigenous minor-element fraction. However, the fact that a group of points form a line means they at least approach this condition.

2) Single analysis method: the minor element content of a sample, in which the terrigenous fraction comprises > 95 weight percent of the rock, like sample F68A of the Lakeridge core, is taken to be the minor element content of the terrigenous fraction for all the samples. This method works well when a large portion of data shows significant marine minor element enrichment. As the Al_2O_3 content increases, i.e., the marine component decreases, the minor element content should approach the value in the sample with >95 detritus. A good example is Ni or Sr (fig. 12*M, P*). For elements that are slightly above the detection limit, yet are primarily in the terrigenous fraction, it represents a sample with the greatest amount of the minor element with the smallest amount of analytical error. An example of this is Be and Sc (fig. 12*D, O*). The minor element content of the 5 percent of the sample which is not terrigenous is a potential problem with this method. There could be contributions of minor elements from the marine environment, such as from the retention of organic matter in the sediment, or from the precipitation of authigenic minerals, like Mn precipitation under bottom water oxidizing conditions, or Mo precipitation under bottom water sulfate reducing conditions. Relying on one analysis, which is bound to have certain amount of analytical error, is another problem with this technique.

3) Standard shale method: a standard shale analysis, such as the World Shale Average (WSA; Turekian and Wedepohl, 1961; Wedepohl, 1969-1978), or North American Shale Composite (NASC; Gromet and others, 1984) for the minor elements, and, for the rare earths', another WSA (Piper, 1974), or another NASC (Haskin and

Haskin, 1966), (table 8) is used for the minor element content of the terrigenous fraction. Obviously one problem with this method is the possibility of the Phosphoria's terrigenous fraction having a composition significantly different from the average shales. Major and minor elements graphed against Al_2O_3 , that show a significant correlation, have contents similar to the values in WSA and/or NASC (fig. 5 and 12C, I, J, and N, table 8). The minor element contents of Ba, Ga, Li, and Sc in the terrigenous fraction of our samples are 429 ppm, 17 ppm, 110 ppm, and 15 ppm, respectively, as compared to 546, 18, 76, and 13 in WSA (Turekian and Wedepohl, 1961; Wedepohl, 1969-1978).

We have chosen to primarily use the single analysis method (2) for determining the minor element content of the terrigenous fraction. We took the minor element content of sample F68A to be the content of terrigenous fraction (which includes As, Be, Co, Ga, Mn, Ni, Pb, Sc, Sr, Th, V), except in the following cases.

1. If the F68A element content is below the detection limit. The Ag and Cd contents of the terrigenous fraction are determined using their contents in the standard shale (WSA).
2. If the minor-element content of sample F68A is significantly higher than the WSA and the graph of the minor element and Al_2O_3 forms no minimum line. Cr, Cu, Li, and Mo contents of the terrigenous fraction are determined by the standard shale method (WSA). It is suspected that the F68A values for Cr, Cu, and Mo contain a marine fraction.
3. If the minor-element content of sample F68A is significantly different from a well defined minimum line. Zn and Ba contents in the terrigenous fraction are determined by the well defined minimum line.

The REE contents of the terrigenous fraction was not determined in the same way, i.e., using the values from sample F68A. In every case sample F68A was the minimum point in rare-earth-element, Al_2O_3 graphs (fig. 14). However, there was for most of the graphs a well defined minimum line, which wasn't always the same value as sample

F68A. It was also noted that both the minimum line and sample F68A were unusual in that they had high La:Ce ratios compared to WSA (Piper, 1974) and NASC (Haskin and Haskin, 1966). We concluded that there must be a marine fraction of rare earth elements in these samples.

The rare earth element contents of the terrigenous fraction, in particular, the Ce and La contents of the terrigenous fraction, were calculated by forcing the data to a shale pattern. They were determined by normalizing the REE contents, attained using the minimum line from REE/Al₂O₃ plots, and the REE contents of sample F68A values with the WSA and NASC values. The pattern most resembling a normalized shale pattern was the minimum line values normalized to the NASC values. The fraction of the NASC values which most closely produced the minimum line values, was then used to calculate the rare earth element contents in the terrigenous fraction (table 8).

Figure 14. Rare-earth-elements versus Al₂O₃ content, determined by inductively coupled plasma-mass spectroscopy: A) La; B) Ce; C) Pr; D) Nd; E) Sm; F) Eu; G) Gd; H) Tb; I) Dy; J) Ho; K) Er; L) Tm; M) Yb.. The unmarked line represents the calculated terrigenous fraction of minor elements relative to the Al₂O₃ content. Legend is given in figure 12A.

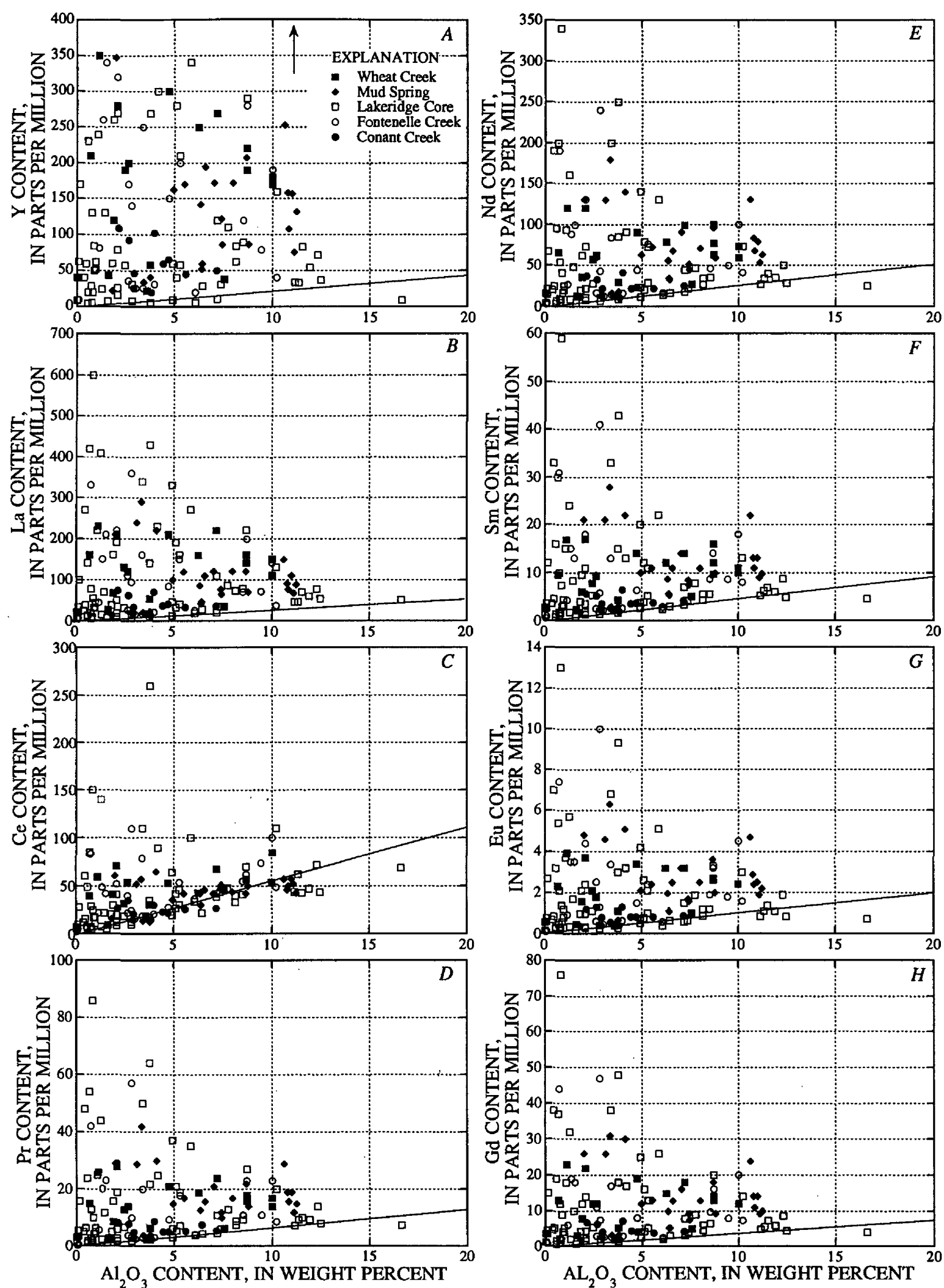


Figure 14

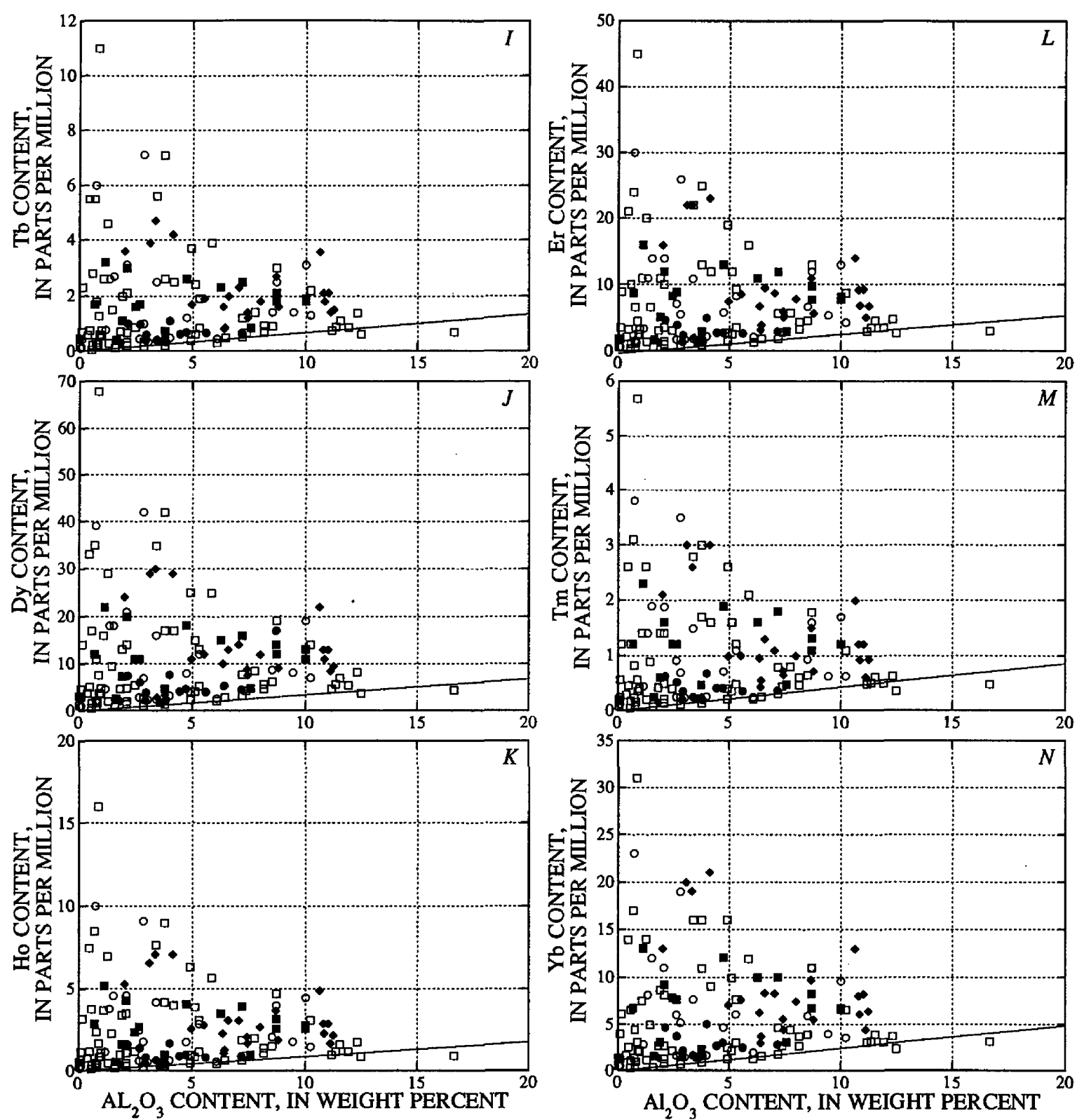


Figure 14. continued

Table 8. Major and minor-element composition of terrigenous fraction. All values are in parts per million.

	World Shale Average ¹	North American Shale Composite ²	Value determined from sample F68	Value from minimum line	Value used in calculations ⁴
Major-oxides					
SiO ₂	58.4	64.8	49.0		58.3
Al ₂ O ₃	15.1	16.9	16.6		16.7
Fe ₂ O ₃	6.84	6.29	6.8	5.14	6.2
MgO	2.49	2.85	3.8	1.14	1.2
CaO	3.09	3.58	3.6	3.58	2.5
Na ₂ O	1.3	1.13	1.6		
K ₂ O	3.19	3.94	5.6	4.17	4.2
TiO ₂	0.77	0.79	0.48	0.48	0.48
P ₂ O ₅	0.16	0.14	0.23	0.23	0.17
CO ₂	2.63 ³	—	—		
Minor elements					
Ag	0.05-0.90	—	—		0.48*
As	13	28.4	32		32
Ba	546	636	27	429	429
Be	2-3	—	3	3	3
Cd	0.8	—	—		0.8*
Co	19	25.7	20		20
Cr	83	124.5	140		83*
Cu	35	—	55	55	35*
Ga	18	—	17	17	17
Li	76		110		76*
Mn	600	600	168	168	168
Mo	0.7-2.0	—	27		2*
Ni	42	58	59		59
Pb	21.6	—	10	14	10
Sc	13	14.9	15	15	15
Sr	24-359	142	70	70	70
Th	12	12.3	7		7
V	98-2600	—	180	180	180
Zn	100	—	27	60	60
Rare Earth Elements					
Y	27	27	10		36
La	41	32	52	52	43
Ce	83	70	69	60	93
Pr	10.1	7.9	7.5	10.6	10.5
Nd	38	31	25	38	41
Sm	7.5	5.7	4.6	7.8	7.6
Eu	1.61	1.24	0.75	1.25	1.65
Gd	6.35	5.21	4.1	7.35	6.9
Tb	1.23	0.85	0.72	1.19	1.13
Dy	5.5	—	4.4	7.2	5.5*
Ho	1.34	1.04	0.95	1.62	1.38
Er	3.75	3.4	3.1	4.5	4.5
Tm	0.63	0.5	0.49	0.70	0.70
Yb	3.53	3.1	3.2	4.3	4.1

¹ World Shale Average (major-element oxides, Turekian and Wedepohl, 1961; minor elements, Wedepohl, 1969-1972; rare-earth elements, Piper, 1974)

² North American Shale Composite (major and minor elements on a volatile free basis: Gromet and others, 1984; rare-earth elements, Haskin and Haskin, 1966)

³ CO₂, Wedepohl, 1969-1972

⁴ Numbers with * in this column were taken from the World Shale Average (the first column)

MARINE MINOR ELEMENTS CALCULATED

The marine minor elements and REEs were calculated by subtracting the fraction calculated to be contributed by the terrigenous material from the bulk content:

$$C = A - B \times (\text{terrigenous fractional content of sample})$$

Where A = total minor-element content of sample, B = minor-element content of terrigenous material, C = marine derived minor-element content in each sample (tables 9 and 10). The error in the calculation of C increases as the minor element content of the sample approaches that of the terrigenous fraction, or the error in C increases as C goes to zero. For those elements that are primarily marine in origin the error is small, for example Cr, Ni, and Sr. For those elements which are primarily terrigenous in origin the error is large, for example Ga, Sc, and Co.

A correlation table for each stratigraphic section was created for components in the marine fraction (dolomite, calcite, biogenic silica, apatite and organic matter) and the marine fraction of minor elements (table 11). These correlations provide a measure of the validity to the calculations of the marine fraction of minor elements. For example, the marine fraction of the heavy metals correlates strongly with organic matter and the marine fraction of REEs correlate with apatite. These associations have been reported in earlier studies, which examined merely the bulk composition of samples (Nissenbaum and others, 1976; Altschuler and others, 1967).

SUMMARY

The goal of this study is to calculate the fraction of the minor elements in these rock samples that is marine derived. This is done by first calculating the components -- dolomite, calcite, apatite, organic matter, biogenic silica and the terrigenous detritus component. These components were then partitioned into marine and terrigenous fractions. The minor element content of sample F68a, a sample composed of >95 weight percent terrigenous detritus, was taken to be the minor element content of the terrigenous

fraction of all the samples. There were a few exceptions, where either the minor element content of a standard shale was used or the minor element content was determined from extracting a minor element/ Al_2O_3 ratio from a well defined minimum line from a minor element/ Al_2O_3 graph. The rare earth element contents and, in particular, the Ce and La contents of the terrigenous fraction, were calculated by forcing the data to a shale pattern. The marine fraction of minor elements and REEs was then calculated from the difference between the bulk minor element content of each sample and the minor elements contributed by the terrigenous fraction. The interpretation, using these calculated marine derived minor elements, of the environment of deposition will be the topic of another bulletin.

Table 9. Calculated marine minor-element contents in samples of the Phosphoria Formation at Conant Creek, Wheat Creek, Fontenelle Creek, and Lakeridge, Wyoming and Mud Springs, Idaho

[All values in parts per million. Blank indicate element is below the detection limit.]

Sample	Conant Creek										Mud Springs				
	P-47	P-46	P-45	P-44	P-43	P-42	P-41	P-40	P-39	P-38	P-47	P-44	P-43	P-41	P-40
Ag-----											4	4	5	5	6
As-----	0	12	6	4	9	16	35	3	11	11	9	0	0	5	9
Ba-----	0	0	4	9	0	0	24	17	6	11	4183	1604	4577	835	315
Be-----	0	1	1	0	0	1	1	0	0	1	1	0	0	1	0
Cd-----	1	1	1	1	5	6	1	1	2	6	8	8	3	8	2
Co-----	1	0	0	1	0	0	2	0	0	0	8	3	0	0	0
Cr-----	359	446	304	248	488	699	358	262	413	521	1916	1066	1625	1020	1365
Cu-----	31	11	3	11	6	0	5	12	13	0	202	203	232	152	103
Ga-----	0	1	0	0	3	2	0	0	1	1	4	4	2	0	0
Li-----	37	34	17	18	29	31	20	22	22	17	0	0	0	0	0
Mn-----	119	79	146	170	130	154	61	211	132	163	0	0	0	0	0
Mo-----	38	8	8	13	5	5	18	32	11	11	9	5	3	11	10
Ni-----	53	17	7	15	11	12	19	35	17	18	338	273	87	270	247
Pb-----	0	14	30	19	34	27	142	16	62	64	5	6	5	5	3
Sc-----	1	1	2	0	1	2	1	1	1	2	1	2	2	2	2
Br-----	163	412	490	125	173	138	568	67	339	428	1125	217	347	437	291
V-----	22	26	17	10	9	14	15	5	10	21	29	19	30	32	50
Zn-----	12	18	22	4	759	877	304	14	535	1063	1201	851	472	1541	1190

Sample	Mud Springs														
	P-38	P-35	P-33	P-31	P-29	P-27	P-26	P-25	P-23	P-22	P-17	P-16	P-15	P-14	P-11
Ag-----	7	6	5	11	7	6	5	3	3	3	4	4	1	1	3
As-----	9	7	8	4	0	0	0	0	0	0	0	0	0	0	0
Ba-----	310	156	245	206	103	77	86	28	656	89	594	787	276	114	123
Be-----	0	1	1	0	0	1	1	0	0	0	0	0	0	0	0
Cd-----	2	2	1	13	18	9	29	11	55	3	6	5	3	1	12
Co-----	0	0	0	9	3	0	0	1	100	0	0	0	2	0	12
Cr-----	1606	1905	2144	1733	879	983	1003	405	202	878	1857	1697	551	553	510
Cu-----	177	128	234	172	84	68	62	45	83	36	175	177	88	67	100
Ga-----	2	4	0	2	0	3	1	0	2	1	3	3	0	0	1
Li-----	0	0	0	7	0	0	0	0	0	0	0	0	0	0	0
Mn-----	0	0	0	64	126	0	287	387	3975	48	0	0	86	53	409
Mo-----	15	8	17	7	8	8	8	12	17	5	16	7	10	7	10
Ni-----	202	128	300	165	363	125	392	558	963	248	144	156	137	117	297
Pb-----	7	7	5	9	2	6	8	4	7	2	11	9	3	2	5
Sc-----	2	3	2	3	2	0	1	0	0	1	3	3	0	0	0
Sr-----	476	433	745	996	506	646	376	215	109	44	500	631	250	45	280
V-----	38	65	101	594	150	237	149	63	40	41	108	80	21	23	91
Zn-----	844	492	1280	862	1382	722	1403	1513	2087	880	496	582	492	352	1197

Sample	Mud Springs					Fontenelle Creek									
	P-8	P-6	P-4	P-2	P-1	E-4	U-3	U-1	R-9	R-6	P-45	P-43	P-40	P-39	P-37
Ag-----	3	5	3	1	1										
As-----	2	4	1	7	1	15									
Ba-----	83	74	473	31	40	72	1	0	21	3	19	32	31	32	8
Be-----	1	1	0	1	0										
Cd-----	31	29	1	13	5	2	6	5	2	2	2	20	10	30	32
Co-----	0	0	2	0	3	1	2	3	3	1	2	1	4	1	0
Cr-----	1229	1595	76	770	279	457	546	813	424	1083	130	355	130	433	460
Cu-----	87	67	22	50	30	37	14	17	23	19	8	41	20	41	35
Ga-----	1	2	0	3	0	1	1	3	3	1	4	7	4	7	6
Li-----	0	11	6	39	12	11	7	14	5	7	6	6	8	4	2
Mn-----	0	0	152	55	84	133	14	81	138	76	170	120	159	86	80
Mo-----	10	7	27	20	13	10	8	7	2	4	2	31	10	36	23
Ni-----	124	118	116	222	100	25	29	48	16	19	10	57	22	56	65
Pb-----	13	10	1	5	9	138	18	9	3	6	4	2	4	2	2
Sc-----	2	2	0	2	1	2	2	3	1	2	2	0	2	0	0
Sr-----	724	933	36	425	943	849	648	324	235	496	82	72	180	86	103
V-----	471	482	10	52	22	81	79	246	32	49	25	23	37	73	77
Zn-----	537	482	207	799	291	150	630	535	196	278	89	409	120	598	756

Table 9. continued

Sample	Fontenelle Creek													Wheat Creek	
	P-32	P-31	P-25	P-23	P-21	P-20	P-19	P-13	P-12	P-8	P-7	P-3	P-1	R-15	R-9
Ag----				9	5			10		10	5			2	2
As----				10	1					13					15
Ba----	46	11	24	27	32	37	34	8	11	26	15	41	31	193	142
Be----	2														2
Cd----	20	10	44	220	190	150	42	150	57	280	140	10	27	10	6
Co----	3	0	2	0	0	0	3	0	3	0	4	2	2	3	1
Cr----	1196	588	774	1549	2650	453	90	2476	257	2757	1086	542	520	457	607
Cu----	22	24	45	149	119	29	23	150	25	192	64	40	62	19	14
Ga----	3	2	5	31	10	1	0	5	1	19	7	8	8	3	1
Li----	5	0	0	0	0	0	0	8	0	1	0	2	1	3	0
Mn----	31	34	107	0	19	115	269	26	143	32	181	20	63	74	45
Mo----	10	19	55	579	239	57	9	199	48	299	69	20	32	8	31
Ni----	21	46	101	404	185	97	67	343	111	469	280	41	52	26	32
Pb----	4	5	7	4	14	14	5	7	2	15	2	3	9	10	27
Sc----	4	0	1	1	1	0	0	3	0	2	2	5	0	1	4
Sr----	987	154	308	15	68	100	69	520	179	254	358	994	471	1197	1089
V----	152	68	283	10890	1592	638	75	2149	421	2106	579	133	178	103	42
Zn----	327	579	941	1663	564	966	488	1583	880	2169	1090	344	163	648	520

Sample	Wheat Creek														
	R-7	R-6	P-34	P-30	P-26	P-22	P-20	P-16	P-14	P-11	P-9	P-7	P-4	P-3	P-1
Ag----	2	2	2	6	2	5	2	7	10	9	2	9	9	7	2
As----															
Ba----	621	48	96	55	62	23	28	46	46	39	24	62	108	65	157
Be----	2			1	0							0			
Cd----	37	85	20	44	20	23	20	33	70	41	7	4	63	10	7
Co----	3	6	3	0	0	0	0	0	0	0	1	0	0	0	1
Cr----	454	70	540	1864	1150	581	292	1857	1557	2469	362	1250	1076	132	668
Cu----	19	22	46	65	6	43	28	51	66	75	17	65	38	11	23
Ga----	3	4	2	3	0	0	1	2	12	4	2	0	4	0	2
Li----	2	5	1	7	0	3	0	0	0	2	2	0	4	0	0
Mn----	65	260	119	38	0	172	164	22	8	2	103	0	16	4	95
Mo----	2	5	10	26	9	20	10	7	1	56	5	35	23	19	8
Ni----	18	21	33	175	62	83	87	139	169	158	46	225	64	60	15
Pb----	62	20	9	6	4	2	2	5	5	6	3	18	17	5	8
Sc----	1	2	4	4	1	2	1	2	2	3	1	1	3	0	3
Sr----	1095	110	611	400	188	194	115	323	323	584	213	258	580	78	800
V----	146	42	88	272	242	109	101	346	1208	753	52	302	239	18	83
Zn----	536	230	323	974	444	396	397	739	799	688	214	1164	613	723	191

Sample	Wheat Creek	Lakeridge													
	Cw-11	E-92C	US-91B	E-90A	E-89A	US-88C	TO-87B	TO-86D	TO-85A	TO-84B	RT-81A	TO-78C	RT-77	RT-75A	F-74
Ag----	2														
As----		3					8								
Ba----	131	0	18	1	2	36	29	12	0	0	38	28	0	62	11
Be----									0		1		0	0	
Cd----	20											8	8		
Co----	2	2	2	2	3	3	2	2	0	2	2	2	0	1	1
Cr----	131	116	206	128	235	168	174	170	375	356	400	250	531	653	146
Cu----	30	20	32	34	27	56	58	46	49	72	53	38	67	35	5
Ga----	2								1	1			1	5	
Li----	0	4	16	11	12	10	12	27	20	19	22	17	15	15	3
Mn----	101	882	211	583	770	475	437	179	349	451	149	479	162	76	191
Mo----	5	5	7	8	8	14	17	16	12	19	11	10	15	12	15
Ni----	15	12	21	18	21	37	36	31	35	49	38	28	63	78	13
Pb----	3		19	9	10	9	27	20	29	30	39	20	20	17	5
Sc----	0			1	2	2	1	0	1	1	1	1	0	4	1
Sr----	452	71	95	93	136	128	95	64	129	178	691	151	307	988	126
V----	16	9	14	13	18	14	8	4	4	26	17	21	0	123	31
Zn----	173	0	5		0	1	0	57	4	9	6	873	472	43	0

Table 9. continued

Lakeridge															
Sample	F-73	F-72B	F-71B	F-70E	F-69A	F-68A	F-65B	F-64A	LS-62A	F-58B	LS-57M	LS-57D	LS-56A	LB-55B	R-54C
Ag----						0			0						
As----						0									
Ba----	0	50	28	0	5	0	91	44	0	11	11	8	45	0	0
Be----						0									
Cd----														2	
Co----	1	1	2	3	2	0	3	3	1	2	4	4	3	2	4
Cr----	161	127	128	368	7	57	70	186	84	20	199	130	110	146	245
Cu----	17	14	24	115	7	20	34	120	36	12	79	60	69	40	77
Ga----						1		0	0						
Li----	0	0	4	6	4	34	5	0	0		3	2	11	4	5
Mn----	232	274	216	345	224	2	282	360	178	280	258	309	169	212	260
Mo----	10	9	15	37	2	25	11	26	7	2	15	13	13	10	16
Ni----	12	12	28	66	3	0	20	67	13	5	37	35	36	33	46
Pb----	7			5		0		6	10						
Sc----	2	1		2		0		0	0		2				2
Sr----	152	97	138	100	178	0	86	49	270	630	399	270	180	227	216
V----	18	23	35	39	8	0	3	0	0	6	12	6	5	6	14
Zn----	0	0	5		4	0	2	0	0	11	24	23	11	80	27

Lakeridge															
Sample	M-53B	M-52	R-50F	M-48B	M-47	M-45	M-44A	M-42A	M-41	M-40	M-39B	M-36A	M-37A	M-36B	M-35A
Ag----				2					3		7	65	6		3
As----				0								59	4		0
Ba----	57	129	30	0	112	253	119	173	0	258	0	0	0	3	0
Be----		2		0			3	1		3	3	0	0		0
Cd----		2	15		4						4	8	110		4
Co----	3	3	4	0	3	3	0	0	0	3	3	0	0	0	0
Cr----	443	896	317	944	415	717	2071	1181	203	824	1879	1757	1756	91	674
Cu----	81	85	55	61	84	84	98	86	58	47	65	92	40	13	19
Ga----		3		1			4	2	0	4	7	14	8		1
Li----	6	10	12	10	7	16	19	0	0	3	6	8	0	0	0
Mn----	285	122	194	0	129	143	13	44	166	31	14	22	64	272	96
Mo----	23	17	120	21	39	31	32	25	24	16	59	219	23	5	13
Ni----	60	44	71	151	65	50	129	28	64	88	175	369	70	7	30
Pb----	6		10	13			2	4	2		5	27	6		6
Sc----	2	4		3	4	8	5	9	0	11	10	6	5	0	0
Sr----	254	1997	348	63	755	797	975	964	99	1495	1283	363	634	82	72
V----	34	91	50	20	43	93	107	179	0	49	125	136	18	0	0
Zn----	53	307	598	1	166	19	1	0	12	84	175	249	5969	14	119

Lakeridge															
Sample	M-34A	M-33A	M-32A	M-31A	M-30A	M-29A	M-28A	M-27B	M-25	M-24	M-23B	M-22B	M-21A	M-20B	M-18
Ag----	5	5	6	3	9		4		6	9	2	7	14	10	9
As----	0	0	0	0	0	6	7	5	13	17		20	36	21	32
Ba----	0	0	0	5	0	126	25	1	0	0	2	0	91	0	24
Be----	1	0	0	0	0		1		0	1		1	3	3	2
Cd----	24	14	19	11	65	52	30	5	62	260	20	210	540	360	180
Co----	0	0	0	0	0	2	0	0	0	0	0	0		0	0
Cr----	960	1339	1338	858	2043	730	1164	96	1441	1860	272	1176	941	899	2774
Cu----	23	20	15	10	65	53	46	10	29	77	13	61	326	119	169
Ga----	3	4	3	2	5		3		3	1				5	
Li----	0	0	0	0	0	4	0	0	0	0	0	0	8	0	19
Mn----	17	46	0	165	24	79	17	181	20	12	104	37	26	0	77
Mo----	12	8	16	10	58	16	18	5	46	169	19	209	200	239	329
Ni----	44	48	56	42	179	60	75	7	88	261	37	253	1393	324	691
Pb----	5	5	8	2	8	4	6		3	12	4	9	22	10	11
Sc----	1	3	2	2	4	6	4	0	2	3	1	2	4	2	3
Sr----	766	298	178	113	152	931	830	108	240	466	133	199	1392	697	338
V----	62	7	45	26	55	54	62	1	91	682	233	1447	2580	1190	1443
Zn----	871	536	555	297	1458	1193	924	120	1257	4271	434	2782	9393	8863	3181

Table 9. continued

Sample	Lakeridge					
	M-17	M-16	M-15A	M-14	M-12B	WS-11D
Ag-----	15	8	13	13		
As-----	36	11	27	52		0
Ba-----	0	24	114	53	279	54
Be-----	4	1	2	2		0
Cd-----	590	360	190	110	29	
Co-----	0	0	0	0	3	1
Cr-----	3275	554	3276	5481	328	52
Cu-----	309	88	200	472	129	33
Ga-----				25	4	0
Li-----	19	2	33	62	10	4
Mn-----	49	67	10	1	126	279
Mo-----	669	409	289	360	30	9
Ni-----	1082	641	863	1187	108	9
Pb-----	19	5	17	23	7	6
Sc-----	3	1	5	5	3	1
Sr-----	529	248	809	874	1898	75
V-----	3845	2343	1447	1059	115	0
Zn-----	6982	5181	3682	4586	788	21

Table 10. Calculated marine rare earth elements in samples from the Phosphoria Formation from Conant Creek, Fontenelle Creek, Wheat Creek, and Lakeridge, Wyoming, and Mud Spring, Idaho. [All values in parts per million.]

Conant Creek											Mud Spring				
Sample	P-47	P-46	P-45	P-44	P-43	P-42	P-41	P-40	P-39	P-38	P-47	P-44	P-43	P-41	P-40
Y----	40	93	104	20	33	35	86	12	49	55	135	67	133	156	52
La----	25	61	68	11	20	18	55	4	26	29	63	48	82	99	39
Ce----	5	8	15	0	0	0	20	0	0	0	0	1	0	0	0
Pr----	3.0	6.4	6.9	1.2	1.8	1.1	6.1	0.2	2.3	2.2	12.2	8.5	12.1	12.0	5.0
Nd----	15	31	32	6	8	7	26	2	11	11	56	37	52	51	26
Sm----	2.2	4.6	4.6	0.9	1.3	0.9	4.0	0.3	1.6	1.6	8.1	5.9	8.0	7.4	3.9
Eu----	0.52	0.91	0.99	0.14	0.29	0.17	1.03	0.05	0.39	0.35	1.83	1.13	1.41	1.71	0.80
Gd----	2.98	5.55	5.91	1.50	1.49	1.34	5.19	0.42	2.56	2.15	9.54	5.56	9.45	9.70	4.71
Tb----	0.40	0.83	0.85	0.20	0.29	0.19	0.78	0.07	0.33	0.37	1.37	1.00	1.35	1.26	0.65
Dy----	2.9	6.4	6.7	1.2	2.2	2.1	5.1	1.0	2.6	3.1	9.4	6.2	9.4	9.4	4.8
Ho----	0.62	1.37	1.42	0.24	0.41	0.34	1.18	0.11	0.56	0.61	2.01	1.17	1.99	2.04	0.78
Er----	1.6	3.9	4.1	0.6	1.2	1.1	3.2	0.5	1.6	1.5	6.3	3.3	6.3	5.7	2.1
Tm----	0.23	0.50	0.54	0.08	0.12	0.07	0.41	0.03	0.22	0.21	0.75	0.35	0.74	0.67	0.13
Yb----	1.5	4.0	4.2	0.8	1.1	1.0	3.0	0.7	1.6	1.8	5.3	3.3	5.5	5.4	1.7

Mud Spring															
Sample	P-38	P-35	P-33	P-31	P-29	P-27	P-26	P-25	P-23	P-22	P-17	P-16	P-15	P-14	P-11
Y----	86	158	109	456	152	159	106	70	45	38	230	189	33	27	128
La----	49	102	59	281	87	106	58	47	30	27	123	98	13	12	71
Ce----	0	3	0	38	8	11	7	10	8	0	0	0	0	0	7
Pr----	9.2	16.6	7.9	39.9	11.9	13.5	7.3	5.3	3.6	4.0	22.3	16.5	1.7	1.7	9.0
Nd----	41	74	35	172	51	58	33	27	17	18	104	75	9	8	40
Sm----	6.1	10.8	4.7	26.5	7.7	8.5	5.0	4.1	2.5	2.8	17.2	12.1	1.3	1.3	5.9
Eu----	1.33	2.51	1.09	5.97	1.61	1.86	0.97	0.87	0.47	0.47	3.65	2.74	0.29	0.22	1.37
Gd----	6.51	13.10	5.35	29.62	9.95	10.72	6.14	4.43	2.65	2.66	19.60	14.42	1.84	1.78	7.38
Tb----	1.06	1.82	0.74	4.47	1.36	1.53	0.90	0.70	0.44	0.38	2.88	2.11	0.26	0.19	1.17
Dy----	7.4	11.7	5.8	28.9	9.4	10.2	6.5	5.1	3.3	3.2	18.5	14.1	2.1	1.7	7.9
Ho----	1.40	2.52	1.27	6.82	2.19	2.34	1.49	1.09	0.77	0.57	4.02	2.98	0.39	0.37	1.78
Er----	4.0	6.9	3.8	21.1	6.2	7.1	4.0	3.1	2.2	1.5	11.1	8.7	1.0	1.1	5.1
Tm----	0.46	0.81	0.46	2.46	0.78	0.77	0.42	0.34	0.28	0.16	1.55	1.14	0.09	0.09	0.68
Yb----	3.4	6.6	3.6	18.2	5.8	6.2	3.8	2.6	2.1	1.4	10.4	7.6	1.2	0.9	4.7

Mud Spring						Fontenelle Creek									
Sample	P-8	P-6	P-4	P-2	P-1	E-4	U-3	U-1	R-9	R-6	P-45	P-43	P-40	P-39	P-37
Y----	420	494	18	181	343	164	564	257	79	243	10	19	10	19	22
La----	209	232	10	93	205	103	353	147	42	151	6	9	8	10	12
Ce----	42	35	6	9	50	25	94	42	16	60	3	0	5	0	0
Pr----	27.4	27.0	1.4	11.9	27.7	12.3	55.2	19.2	5.0	17.9	0.3	0.8	0.5	0.9	0.7
Nd----	130	122	7	52	125	53	233	85	24	76	4	6	5	7	6
Sm----	20.1	19.6	1.0	8.0	20.1	8.2	39.7	14.4	3.8	11.4	0.6	0.8	0.9	1.1	0.9
Eu----	4.69	4.29	0.21	1.85	4.60	2.24	9.72	3.37	0.82	3.06	0.10	0.17	0.15	0.22	0.25
Gd----	28.29	24.72	1.24	10.28	25.16	9.90	45.82	18.45	5.52	15.59	0.98	1.45	0.96	1.29	1.38
Tb----	3.92	3.69	0.20	1.55	3.46	1.52	6.91	2.51	0.69	2.27	0.11	0.14	0.14	0.17	0.21
Dy----	27.6	28.0	1.5	10.8	23.3	10.1	41.1	17.6	4.2	14.9	0.9	1.4	1.1	1.4	1.9
Ho----	6.76	6.34	0.28	2.56	5.13	2.28	8.86	3.69	1.10	3.92	0.16	0.33	0.29	0.37	0.33
Er----	21.9	21.2	0.8	7.7	15.5	6.4	25.2	10.6	3.1	10.1	0.6	1.0	0.6	1.0	1.1
Tm----	2.83	2.87	0.07	1.02	2.02	0.80	3.38	1.34	0.39	1.36	0.06	0.10	0.12	0.10	0.09
Yb----	20.0	19.2	0.9	6.7	12.5	5.3	18.3	7.8	2.5	6.9	0.5	0.8	0.5	0.9	0.7

Fontenelle Creek														Wheat Creek	
Sample	P-32	P-31	P-25	P-23	P-21	P-20	P-19	P-13	P-12	P-6	P-7	P-3	P-1	R-15	R-9
Y----	746	102	189	18	168	59	7	140	30	261	134	337	316	209	194
La----	328	48	138	11	114	46	10	73	20	178	68	206	215	156	113
Ce----	79	8	24	0	44	21	3	2	7	14	9	33	40	36	39
Pr----	41.5	5.6	14.7	2.4	16.7	5.0	0.4	6.6	2.2	17.5	8.2	22.0	27.7	14.6	12.3
Nd----	188	25	63	16	75	26	2	32	10	77	38	95	125	84	55
Sm----	30.7	4.8	8.6	3.3	13.4	4.4	0.1	4.2	1.4	10.0	4.5	12.3	17.1	9.2	8.1
Eu----	7.33	1.06	1.68	0.59	3.51	0.86	0.00	1.03	0.37	2.34	1.02	3.35	4.19	2.23	1.54
Gd----	43.70	6.18	10.81	3.07	15.85	3.99	0.00	6.14	1.80	12.40	6.32	17.37	21.14	12.73	10.90
Tb----	5.95	0.82	1.54	0.61	2.42	0.76	0.04	0.68	0.26	1.91	0.81	2.60	2.96	1.66	1.52
Dy----	38.8	5.9	10.3	3.6	15.7	5.1	0.6	6.5	1.9	14.1	6.0	17.5	20.3	11.8	10.1
Ho----	9.94	1.40	2.46	0.65	3.67	1.02	0.09	1.41	0.42	3.28	1.56	4.47	4.43	2.85	2.48
Er----	29.8	4.4	6.9	1.5	10.3	2.9	0.5	4.6	1.1	9.7	4.8	13.6	13.4	8.6	8.2
Tm----	3.77	0.57	0.88	0.20	1.28	0.23	0.00	0.52	0.14	1.24	0.57	1.84	1.81	1.17	1.09
Yb----	22.8	3.8	4.8	1.1	7.1	1.7	0.5	3.5	1.0	8.9	4.5	11.6	10.5	6.5	6.9

Table 10. continued

Wheat Creek															
Sample	R-7	R-6	P-34	P-30	P-26	P-22	P-20	P-16	P-14	P-11	P-9	P-7	P-4	P-3	P-1
La----	348	40	276	254	148	50	14	171	201	237	39	158	290	22	185
	227	22	205	201	124	44	10	118	138	144	23	84	198	15	124
Ce----	53	8	59	28	29	10	0	8	8	8	0	0	27	4	18
Pr---	25.3	3.1	26.7	19.5	10.7	4.0	0.2	10.5	12.5	15.1	1.9	7.7	18.0	1.4	11.5
Nd----	117	16	115	82	48	21	4	42	56	84	9	34	78	8	51
Sm---	16.5	2.8	16.1	10.7	6.4	2.6	0.6	5.7	8.0	9.2	1.6	5.4	11.8	1.7	6.7
Eu----	3.79	0.56	3.49	2.49	1.41	0.73	0.04	1.74	1.84	2.58	0.32	1.41	2.93	0.25	1.86
Gd----	22.53	3.60	21.14	15.03	7.85	3.74	0.41	9.40	10.40	12.42	2.42	7.85	17.04	1.87	10.98
Tb----	3.12	0.47	2.86	2.01	1.22	0.47	0.12	1.21	1.51	1.86	0.29	1.12	2.28	0.32	1.43
Dy---	21.6	3.1	19.3	13.6	9.7	3.5	0.6	9.1	11.1	12.9	2.1	7.7	16.4	2.3	10.2
Ho----	5.11	0.58	4.13	3.31	1.97	0.69	0.08	1.88	2.48	2.98	0.41	1.77	3.71	0.37	2.20
Er----	15.7	1.9	11.4	10.1	5.5	2.0	0.3	5.5	7.5	9.3	1.4	5.1	11.7	1.0	7.6
Tm---	2.25	0.20	1.51	1.50	0.78	0.30	0.07	0.73	0.93	1.34	0.18	0.78	1.70	0.15	1.10
Yb----	12.7	1.4	8.7	8.2	4.2	1.4	0.6	4.6	6.1	8.5	1.2	4.1	10.8	1.2	7.2

Wheat Creek		Lakeridge													
Sample	Cw-11	E-92C	US-91B	E-90A	E-89A	US-88C	TO-87B	TO-86D	TO-85A	TO-84B	RT-81A	TO-78C	RT-77	RT-75A	F-74
Y----	116	18	82	27	60	58	22	12	29	25	74	22	93	453	48
La----	63	9	53	18	41	38	13	6	20	12	45	15	67	331	30
Ce----	30	1	20	4	13	13	1	0	0	0	8	0	5	91	12
Pr---	7.5	1.6	9.5	2.3	5.8	6.2	1.6	0.6	2.4	1.2	4.6	1.6	8.1	47.9	3.9
Nd----	31	6	39	8	23	24	6	3	8	4	17	6	28	192	15
Sm---	5.1	1.2	7.1	1.5	4.0	4.4	1.2	0.5	1.6	0.7	2.6	1.0	4.3	31.4	2.6
Eu----	1.41	0.26	1.31	0.40	1.10	0.82	0.24	0.08	0.26	0.14	0.77	0.14	1.13	6.46	0.51
Gd----	7.12	1.46	8.54	1.89	4.89	5.11	1.57	0.74	2.19	1.03	3.64	1.34	5.79	36.59	3.73
Tb----	0.97	0.18	1.24	0.27	0.68	0.73	0.20	0.07	0.27	0.15	0.52	0.17	0.87	5.37	0.54
Dy---	6.7	1.4	7.4	1.9	4.5	4.9	1.6	0.8	2.4	1.6	4.1	1.3	5.9	33.9	3.4
Ho----	1.44	0.32	1.63	0.44	1.02	1.16	0.37	0.15	0.49	0.34	0.93	0.32	1.36	7.42	0.80
Er----	4.6	1.0	4.3	1.3	2.9	3.2	1.1	0.4	1.2	0.9	3.0	0.9	3.7	21.1	2.4
Tm---	0.52	0.13	0.52	0.20	0.36	0.43	0.15	0.04	0.15	0.11	0.39	0.12	0.47	2.86	0.32
Yb----	2.6	0.8	2.9	1.2	2.0	2.5	0.9	0.3	0.9	0.8	2.2	0.8	2.5	15.2	2.0

Lakeridge															
Sample	F-73	F-72B	F-71B	F-70E	F-69A	F-68A	F-65B	F-64A	LS-62A	F-58B	LS-57M	LS-57D	LS-56A	LS-55B	R-54C
Y----	58	19	39	53	4	0	4	0	0	9	170	62	40	128	50
La----	34	13	25	26	10	9	5	0	2	5	100	34	22	76	31
Ce----	11	3	10	5	5	0	3	0	0	6	27	9	6	25	6
Pr---	4.0	1.4	5.0	5.3	0.9	0.0	0.9	0.0	0.2	1.2	15.9	5.5	3.6	12.5	4.5
Nd----	15	6	20	22	2	0	4	0	0	5	67	22	15	52	18
Sm---	3.2	1.0	3.5	4.2	0.6	0.0	0.8	0.5	0.0	1.2	11.9	3.7	2.7	9.7	2.9
Eu----	0.79	0.25	0.66	0.86	0.18	0.00	0.15	0.02	0.00	0.17	2.69	0.76	0.60	2.03	0.69
Gd----	3.54	1.37	4.55	4.48	0.47	0.00	0.59	0.47	0.13	1.30	14.94	4.56	3.47	11.69	3.70
Tb----	0.58	0.18	0.56	0.69	0.02	0.00	0.08	0.05	0.03	0.19	2.29	0.70	0.48	1.75	0.49
Dy---	4.1	1.4	3.7	4.6	0.4	0.0	0.6	0.8	0.8	1.2	14.0	4.7	3.0	10.8	3.3
Ho----	0.85	0.33	0.76	1.00	0.08	0.00	0.14	0.09	0.09	0.26	3.19	1.19	0.68	2.34	0.81
Er----	2.6	0.9	2.0	2.4	0.2	0.0	0.3	0.2	0.0	0.7	9.0	3.6	1.9	6.5	2.2
Tm---	0.34	0.16	0.27	0.30	0.03	0.00	0.04	0.00	0.01	0.09	1.19	0.56	0.24	0.79	0.31
Yb----	2.2	1.0	1.6	1.6	0.1	0.0	0.2	0.1	0.0	0.5	6.1	4.0	1.3	4.2	1.9

Lakeridge															
Sample	M-53B	M-52	R-50F	M-48B	M-47	M-45	M-44A	M-42A	M-41	M-40	M-39B	M-38A	M-37A	M-36B	M-35A
Y----	127	898	229	10	238	639	327	522	15	497	291	271	71	0	9
La----	66	598	139	17	217	418	255	420	16	407	219	198	58	2	18
Ce---	14	145	46	0	53	81	67	239	3	133	67	21	0	0	0
Pr---	11.1	85.5	23.6	0.4	24.3	53.6	31.3	61.6	1.5	43.2	22.4	21.5	3.9	0.0	2.2
Nd----	44	338	94	0	90	198	116	241	4	157	81	79	14	0	6
Sm---	7.6	58.6	15.7	0.2	14.5	29.7	19.3	41.3	0.8	23.4	11.1	12.0	1.6	0.0	1.1
Eu----	1.56	12.92	3.14	0.00	3.59	5.33	4.52	8.93	0.00	5.58	2.79	2.44	0.36	0.00	0.00
Gd----	9.40	75.66	18.77	0.28	17.55	36.73	23.57	46.44	0.95	31.48	15.28	16.40	2.88	0.00	0.91
Tb----	1.40	10.94	2.76	0.00	2.53	5.46	3.50	6.84	0.17	4.52	2.22	2.41	0.33	0.00	0.12
Dy---	9.0	67.7	16.8	1.0	15.6	34.8	23.1	40.8	1.9	28.6	15.6	16.1	3.4	0.3	1.9
Ho----	2.18	15.93	3.75	0.08	3.61	8.45	5.21	8.69	0.34	6.90	3.66	3.98	0.80	0.01	0.26
Er----	6.3	44.8	9.8	0.0	10.7	23.8	14.4	24.0	0.9	19.7	10.9	10.7	2.3	0.0	0.4
Tm---	0.83	5.67	1.18	0.01	1.35	3.07	1.85	2.84	0.13	2.55	1.43	1.43	0.29	0.00	0.03
Yb----	4.5	30.8	6.4	0.3	7.2	16.8	10.6	15.1	0.9	13.7	8.1	8.9	1.8	0.0	0.4

Table 10. continued

Lakeridge															
Sample	M-34A	M-33A	M-32A	M-31A	M-30A	M-29A	M-28A	M-27B	M-25	M-24	M-23B	M-22B	M-21A	M-20B	M-18
Y----	44	45	10	15	58	266	104	2	28	66	5	48	256	138	199
La----	52	46	22	11	41	185	91	3	29	53	5	34	155	104	146
Ce----	0	3	0	0	0	29	2	0	0	0	0	0	19	53	16
Pr----	2.5	6.3	0.1	0.5	2.7	17.7	6.5	0.1	1.9	3.8	0.2	3.1	14.8	13.6	15.7
Nd----	7	20	0	1	12	68	26	0	6	14	1	11	57	48	59
Sm----	0.6	3.2	0.0	0.1	1.6	10.1	3.7	0.2	0.6	1.8	0.2	1.6	8.6	8.3	8.6
Eu----	0.08	0.68	0.00	0.01	0.26	2.19	0.79	0.00	0.00	0.40	0.01	0.25	1.91	1.99	1.56
Gd----	1.24	3.61	0.00	0.44	2.63	13.14	4.93	0.13	0.97	2.64	0.14	2.17	11.22	9.77	10.81
Tb----	0.15	0.57	0.00	0.04	0.32	1.96	0.71	0.00	0.05	0.37	0.00	0.28	1.87	1.51	1.54
Dy----	1.9	4.3	0.0	0.8	3.1	13.3	5.3	0.3	1.5	3.0	0.3	2.5	12.4	10.6	11.3
Ho----	0.43	0.78	0.00	0.15	0.65	3.33	1.31	0.05	0.21	0.73	0.04	0.56	3.24	2.25	2.66
Er----	1.1	1.5	0.0	0.2	1.5	9.4	3.8	0.1	0.3	2.2	0.1	1.5	10.5	6.0	7.9
Tm----	0.12	0.11	0.00	0.00	0.14	1.31	0.49	0.00	0.00	0.26	0.00	0.19	1.32	0.67	0.98
Yb----	0.7	0.7	0.0	0.0	1.1	7.6	2.9	0.0	0.2	1.7	0.0	1.3	8.1	4.0	6.3

Lakeridge						
Sample	M-17	M-16	M-15A	M-14	M-12B	WS-11D
Y----	269	47	439	262	449	0
La----	177	28	317	130	269	4
Ce----	14	1	37	14	58	0
Pr----	17.8	4.1	33.9	19.6	47.7	0.1
Nd----	66	16	128	76	189	0
Sm----	9.7	2.8	17.8	13.3	32.8	0.0
Eu----	2.09	0.48	3.71	2.63	6.96	0.00
Gd----	13.89	3.11	22.97	16.44	37.82	0.00
Tb----	2.05	0.48	3.37	2.34	5.47	0.00
Dy----	13.3	3.6	23.4	15.8	32.9	0.1
Ho----	3.48	0.78	5.89	3.89	7.46	0.00
Er----	10.6	2.2	17.7	12.0	20.9	0.0
Tm----	1.39	0.27	2.39	1.54	2.56	0.00
Yb----	8.6	1.7	14.8	10.1	13.9	0.0

Table 11. Components and calculated marine minor element correlation table for A) Conant Creek ($R=0.71$ is significant at the 1 percent level, R -values with significance at 1 percent level or better have been shaded), and B) Mud Spring ($R=0.49$ is significant at the 1 percent level, R -values with significance at 1 percent level or better have been shaded)

A	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	As	Cr	Cu	Mn	Mo	Ni	Pb	Sr	V	Zn	La
Dolomite	1.00															
Calcite	-0.23	1.00														
Biogenic SiO ₂	-0.47	0.55	1.00													
Apatite	-0.57	-0.25	-0.28	1.00												
Organic matter	0.22	-0.50	-0.61	-0.08	1.00											
As	-0.42	-0.33	-0.34	0.60	0.53	1.00										
Cr	0.18	-0.34	-0.65	-0.09	0.86	0.29	1.00									
Cu	-0.42	0.55	0.88	-0.31	-0.61	-0.48	-0.43	1.00								
Mn	0.71	0.06	0.02	-0.66	-0.07	-0.64	-0.13	-0.10	1.00							
Mo	-0.46	0.53	0.92	-0.26	-0.47	-0.30	-0.51	0.77	0.16	1.00						
Ni	-0.47	0.57	0.87	-0.35	-0.39	-0.36	-0.30	0.85	0.07	0.95	1.00					
Pb	-0.35	-0.37	-0.25	0.64	0.31	0.89	0.05	-0.42	-0.48	-0.17	-0.28	1.00				
Sr	-0.52	-0.25	-0.36	0.98	-0.04	0.61	-0.01	-0.36	-0.64	-0.32	-0.37	0.68	1.00			
V	-0.54	0.39	-0.08	0.48	-0.12	0.08	0.25	0.14	-0.55	-0.04	0.14	-0.06	0.50	1.00		
Zn	0.40	-0.49	-0.71	-0.08	0.77	0.30	0.81	-0.57	0.08	-0.50	-0.37	0.31	0.05	-0.05	1.00	
La	-0.57	-0.16	-0.23	0.92	-0.15	0.41	-0.06	-0.20	-0.69	-0.30	-0.32	0.33	0.87	0.63	-0.24	1.00

B	Dolomite	Biogenic SiO ₂	Apatite	Organic matter	Ag	Ba	Cd	Cr	Cu	Mn	Mo	Ni	Sr	V	Zn	La
Dolomite	1.00															
Biogenic SiO ₂	-0.33	1.00														
Apatite	0.55	-0.66	1.00													
Organic matter	-0.13	-0.47	-0.04	1.00												
Ag	0.14	-0.56	0.37	0.49	1.00											
Ba	-0.32	-0.16	-0.25	0.58	0.04	1.00										
Cd	0.29	-0.22	0.39	-0.23	0.00	-0.14	1.00									
Cr	0.13	-0.60	0.09	0.71	0.58	0.33	-0.23	1.00								
Cu	-0.26	-0.41	-0.19	0.79	0.44	0.60	-0.26	0.76	1.00							
Mn	-0.21	0.13	-0.08	-0.26	-0.14	-0.04	0.73	-0.39	-0.16	1.00						
Mo	-0.03	0.21	-0.14	-0.30	-0.27	-0.27	0.04	-0.31	-0.24	0.25	1.00					
Ni	-0.23	-0.03	-0.12	-0.11	-0.05	-0.02	0.65	-0.32	-0.09	0.84	0.18	1.00				
Sr	0.31	-0.72	0.70	0.35	0.36	0.17	0.04	0.57	0.34	-0.29	-0.17	-0.27	1.00			
V	0.63	-0.44	0.73	-0.01	0.51	-0.25	0.40	0.30	-0.02	-0.12	-0.24	-0.20	0.54	1.00		
Zn	-0.17	-0.28	-0.02	0.13	0.20	-0.01	0.50	-0.10	0.07	0.57	0.03	0.87	-0.13	-0.12	1.00	
La	0.57	-0.59	0.86	0.08	0.38	-0.13	0.23	0.32	0.05	-0.20	-0.20	-0.34	0.74	0.82	-0.26	1.00

Table 11. (cont.) Components and calculated marine minor elements correlation table for C) Fontenelle Creek (R=0.51 is significant at the 1 percent level, R-values with significance at 1 percent level or better have been shaded) and D) Wheat Creek (R=0.56 is significant at the 1 percent level, R-values with significance at 1 percent level or better have been shaded)

C	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Cd	Cr	Cu	Mn	Mo	Ni	Pb	Sr	V	Zn	La
Dolomite	1.00															
Calcite	0.83	1.00														
Biogenic SiO ₂	0.01	0.24	1.00													
Apatite	-0.48	-0.53	-0.56	1.00												
Organic matter	-0.21	-0.24	-0.31	-0.18	1.00											
Cd	-0.22	-0.21	-0.22	-0.35	0.82	1.00										
Cr	-0.42	-0.44	-0.35	0.02	0.83	0.78	1.00									
Cu	-0.32	-0.30	-0.24	-0.19	0.92	0.89	0.87	1.00								
Mn	0.69	0.64	0.19	-0.40	-0.40	-0.29	-0.58	-0.46	1.00							
Mo	-0.26	-0.26	-0.17	-0.29	0.74	0.83	0.68	0.85	-0.46	1.00						
Ni	-0.16	-0.17	-0.31	-0.28	0.91	0.93	0.78	0.93	-0.31	0.85	1.00					
Pb	-0.24	-0.24	-0.22	0.33	-0.09	-0.09	-0.04	-0.01	0.07	-0.07	-0.10	1.00				
Sr	-0.38	-0.39	-0.60	0.91	-0.08	-0.29	0.05	-0.13	-0.32	-0.27	-0.21	0.39	1.00			
V	-0.20	-0.22	-0.13	-0.24	0.55	0.63	0.43	0.64	-0.38	0.93	0.68	-0.07	-0.24	1.00		
Zn	-0.22	-0.21	-0.24	-0.27	0.85	0.86	0.67	0.81	-0.33	0.74	0.93	-0.16	-0.22	0.59	1.00	
La	-0.51	-0.55	-0.56	0.85	-0.04	-0.08	0.24	0.00	-0.52	-0.12	-0.08	0.09	0.76	-0.17	-0.05	1.00

D	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Ag	Ba	Cd	Cr	Cu	Mn	Mo	Ni	Pb	Sr	V	Zn	La
Dolomite	1.00																	
Calcite	0.05	1.00																
Biogenic SiO ₂	-0.15	0.72	1.00															
Apatite	-0.64	-0.21	-0.38	1.00														
Organic matter	0.16	-0.13	-0.39	-0.12	1.00													
Ag	-0.17	-0.21	-0.15	-0.24	0.58	1.00												
Ba	-0.41	-0.06	-0.05	0.65	-0.39	-0.31	1.00											
Cd	0.17	0.21	0.05	-0.19	0.37	0.35	-0.03	1.00										
Cr	-0.30	-0.41	-0.36	0.01	0.62	0.69	-0.24	0.26	1.00									
Cu	-0.10	-0.12	-0.18	-0.14	0.67	0.72	-0.30	0.32	0.76	1.00								
Mn	0.68	0.38	0.11	-0.20	-0.16	-0.59	-0.06	0.18	-0.61	-0.27	1.00							
Mo	-0.21	-0.23	-0.22	0.06	0.31	0.50	-0.26	-0.10	0.55	0.48	-0.39	1.00						
Ni	-0.03	-0.39	-0.25	-0.35	0.48	0.79	-0.41	0.15	0.77	0.82	-0.49	0.50	1.00					
Pb	-0.29	-0.12	-0.10	0.53	-0.41	-0.17	0.88	0.14	-0.17	-0.22	-0.01	-0.07	-0.26	1.00				
Sr	-0.60	-0.14	-0.35	0.98	-0.16	-0.28	0.88	-0.20	-0.02	-0.15	-0.17	0.05	-0.37	0.56	1.00			
V	-0.14	-0.29	-0.30	-0.07	0.67	0.70	-0.18	0.47	0.72	0.68	-0.47	0.20	0.66	-0.14	-0.09	1.00		
Zn	-0.35	-0.49	-0.25	0.00	0.15	-0.74	-0.06	0.06	0.52	0.58	-0.67	0.48	0.82	0.07	-0.02	0.48	1.00	
La	-0.70	-0.19	-0.18	0.68	0.08	0.14	0.51	0.20	0.45	0.30	-0.43	0.10	0.07	0.43	0.64	0.28	0.30	1.00

Table 11. (cont.) Components and marine minor elements correlation table for E)Lakeridge (R=0.32 is significant at the 1 percent level, R-values with a significance of 1 percent or better are shaded)

E	Dolomite	Calcite	Biogenic SiO ₂	Apatite	Organic matter	Ba	Cr	Cu	Mn	Mo	Ni	Pb	Sr	V	Zn	La
Dolomite-----	1.00															
Calcite-----	0.44	1.00														
Biogenic SiO ₂ -----	-0.23	-0.04	1.00													
Apatite-----	-0.56	-0.25	-0.23	1.00												
Organic matter-----	-0.24	-0.16	-0.33	0.10	1.00											
Ba-----	-0.29	-0.01	-0.06	0.73	-0.10	1.00										
Cr-----	-0.45	-0.31	-0.35	0.35	0.78	0.07	1.00									
Cu-----	-0.35	-0.09	-0.13	0.33	0.76	0.22	0.73	1.00								
Mn-----	0.62	0.32	0.26	-0.43	-0.37	-0.17	-0.50	-0.28	1.00							
Mo-----	-0.20	-0.16	-0.29	0.10	0.31	-0.04	0.68	0.73	-0.34	1.00						
Ni-----	-0.26	-0.10	-0.29	0.23	0.88	0.07	0.72	0.90	-0.35	0.85	1.00					
Pb-----	-0.31	-0.23	0.25	0.01	0.33	-0.15	0.27	0.36	0.02	0.29	0.36	1.00				
Sr-----	-0.49	-0.09	-0.28	0.94	0.15	0.69	0.33	0.39	-0.40	0.14	0.30	0.00	1.00			
V-----	-0.14	-0.09	-0.30	0.12	0.85	0.00	0.52	0.69	-0.30	0.92	0.87	0.25	0.19	1.00		
Zn-----	-0.26	-0.18	-0.31	0.20	0.73	-0.04	0.52	0.65	-0.37	0.74	0.79	0.23	0.27	0.81	1.00	
La-----	-0.46	-0.19	-0.19	0.87	0.13	0.75	0.33	0.31	-0.35	0.17	0.21	-0.06	0.82	0.15	0.09	1.00

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