



EXPLANATION

Mining district—Approximate boundaries of mining district within which most mining activity occurred; name shown where known. Prospected mineral deposit type mixed and the resulting mine-drainage signatures identified by color.

Potential mine-drainage signatures of mineral deposit types—A brief summary of environmental weathering, gangue, and alteration minerals is given in parentheses. For each deposit type, estimated ranges of pH values and dissolved metal concentrations of drainage waters are listed for the different types or mineralogical assemblages. Composition ranges are defined by empirical drainage data compiled for similar mineral-deposit types (Smith and others, 1995; Plumlee and others, 1994). Inferred compositions of waters draining mine dumps are listed in decreasing order of likely acidity and dissolved metal content. ppm, parts per billion; pb, parts per billion; conc, dissolved concentrations; ±, may or may not be present.

Deposit types likely to generate predominantly acidic, metal-rich waters

- Quartz-sulfide epithermal deposits**—(Pyrite-earrinite-covellite-chalcocite veins, disseminated pyrite) have pH values that range from 1.5 to 3; Fe, Al, Mn conc, that range from hundreds to several thousands of ppm; Cu, Zn conc, that range from hundreds of pb to several ppm; As, Co, Ni, Pb, Co, U, and Th conc, that range from hundreds of pb to several ppm.
- Chlorite-type porphyry molybdenum deposits associated with U-rich granitic intrusions**—(Cov zone with pyrite-molybdenite, fluorite, topaz; pyrite-rich intermediate zone; sphalerite, carbonates on fringes) Waters draining pyrite-molybdenite core have pH values that range from 1 to 2; Fe, Al, Mn conc, that range from hundreds to thousands of ppm; Cu conc, in hundreds of pb; Zn conc, that range from 2 to 5; Fe, Al, Mn conc, in hundreds of pb; Zn, Cu conc, that range from <1 to 10 ppm. Waters draining heavy metal fringes have pH values generally greater than 5.5; Zn, Mn conc, that range from hundreds of pb to several tens of ppm; Fe conc, that range from hundreds of pb to several ppm; U conc, that range from 1 to 10 ppm. Waters draining intermediate zone have pH values generally greater than 5.5; Zn, Mn conc, that range from hundreds of pb to several tens of ppm; Fe conc, that range from hundreds of pb to several ppm; U conc, that range from 1 to 10 ppm.
- Hydrothermal vein dissemination in carbonate-poor rocks**—(Pyrite, chalcocite, sphalerite, galena, ± molybdenite) Waters draining underground workings and mine dumps have pH values that range from 2.5 to 5.5; Fe, Al, Mn conc, that range from tens to low hundreds of pb to several ppm; Cu conc, that range from hundreds of pb to several ppm; Pb, As conc, in tens of pb; Waters draining underground workings in carbonate-rich rocks (see carbonate-poor rocks) have pH values that range from 2.5 to 5.5; Fe, Al, Mn conc, that range from tens to low hundreds of pb to several ppm; Cu conc, that range from hundreds of pb to several ppm; Pb, As conc, in tens of pb; Waters draining underground workings in carbonate-poor igneous intrusive rocks have pH values that range from 2.5 to 5.5; Fe, Al, Mn conc, that range from tens to low hundreds of pb to several ppm; Cu conc, that range from 1 to 10 ppm; Pb, As conc, in several tens of pb.
- Uranium-rich veins and replacements in carbonate-rich sedimentary rocks**—(Pyrite, marcasite, uraninite, galena, calcite, rhodochrosite) Waters draining many underground workings have pH values generally greater than 5; Zn conc, that range from 1 to 200 ppm; Fe conc, that range from one to tens of ppm; Cu conc, that range from tens to one ppm; As conc, in several tens of pb; U conc, that range from 1 to 10 ppm; Pb, As conc, that range from 1 to 10 ppm; Fe conc, that range from 1 to 10 ppm; U conc, that range from 1 to 10 ppm; Pb, As conc, that range from 1 to 10 ppm.
- Polymetallic veins with abundant carbonates or that occur in walls altered to contain carbonates**—(Pyrite, chalcocite, sphalerite, galena, calcite, rhodochrosite) Waters draining many underground workings have pH values generally greater than 5; Zn conc, that range from 1 to 200 ppm; Fe conc, that range from one to tens of ppm; Cu conc, that range from tens to one ppm; As conc, in several tens of pb; U conc, that range from 1 to 10 ppm; Pb, As conc, that range from 1 to 10 ppm.
- Polymetallic veins with abundant carbonates**—(Pyrite, marcasite, uraninite, galena, calcite, other carbonates, ± sphalerite) Inferred compositions of waters draining mine dumps have pH values that range from 2.5 to 5.5; Fe, Al, Mn conc, that range from tens to low hundreds of pb to several ppm; Cu conc, that range from hundreds of pb to several ppm; Pb, As conc, in tens of pb; U conc, that range from 1 to 10 ppm; Pb, As conc, in several tens of pb.
- Massive sulfide deposits**—(Lenses of massive pyrite, chalcocite, and galena, with lesser chalcocite) Gneissification deposits enclosed in carbonate-bearing rocks may have drainage waters with pH values near 7; Zn conc, that range from several hundreds of pb to one ppm; Fe conc, that range from 1 to 10 ppm; U conc, in hundreds of pb; As in tens of pb.

Deposit types likely to generate both acidic and near-neutral, metal-rich waters

- Polymetallic veins and replacements in carbonate-rich sedimentary rocks with low carbonate content**—(Pyrite, chalcocite, sphalerite, galena, calcite, rhodochrosite) Waters draining many underground workings have pH values generally greater than 5; Zn conc, that range from 1 to 200 ppm; Fe conc, that range from one to tens of ppm; Cu conc, that range from tens to one ppm; As conc, in several tens of pb; U conc, that range from 1 to 10 ppm; Pb, As conc, that range from 1 to 10 ppm.
- Uranium-rich veins with abundant carbonates**—(Pyrite, marcasite, uraninite, galena, calcite, other carbonates, ± sphalerite) Inferred compositions of waters draining mine dumps have pH values that range from 2.5 to 5.5; Fe, Al, Mn conc, that range from tens to low hundreds of pb to several ppm; Cu conc, that range from hundreds of pb to several ppm; Pb, As conc, in tens of pb; U conc, that range from 1 to 10 ppm; Pb, As conc, in several tens of pb.
- Massive sulfide deposits**—(Lenses of massive pyrite, chalcocite, and galena, with lesser chalcocite) Gneissification deposits enclosed in carbonate-bearing rocks may have drainage waters with pH values near 7; Zn conc, that range from several hundreds of pb to one ppm; Fe conc, that range from 1 to 10 ppm; U conc, in hundreds of pb; As in tens of pb.

Scale: 1 : 750,000
1 inch equals approximately 12 miles

Map of OF-95-26 Digital Production Note: Digital geologic and geologic information prepared with GSI/MAP running under DOS 5.0 on an IBM compatible PC. The map was prepared with ArcView 3.5 and Canvas 3.5 running under System 7.5 on an Apple Macintosh personal computer.

This report is preliminary and has not been reviewed for conformity with U.S. Geological Survey editorial standards and stratigraphic nomenclature. Any use of trade names is for descriptive purposes only and does not imply endorsement by the USGS.

Introduction

Land management agencies are currently faced with the daunting task of identifying, assessing, and prioritizing abandoned mine sites on public lands for remediation. In Colorado, the number of abandoned mine sites on public and private lands is in excess of 20,000 (Herron, oral comm., 1995). Nationwide, the total number of abandoned mine sites is in excess of 500,000. Because of the large number of sites to be evaluated, procedures must be developed that can streamline and facilitate the site assessment process.

Geologic characteristics of mineral deposits are a fundamental and predictable control on the environmental effects of mining and mineral processing, and on the natural environmental conditions that exist in mineralized areas prior to mining (Plumlee and others, 1994; Smith and others, 1994; Plumlee and others, 1993; Plumlee and others, 1992; Ficklin and others, 1992; Kwong, 1993). Other important controls, such as geochemical and hydrogeological processes, climate, and mining and mineral processing methods, generally serve to modify the environmental effects dictated by geologic characteristics.

This map, compiled by the U.S. Geological Survey (USGS) in cooperation with the Colorado Geological Survey (CGS) and the U.S. Bureau of Land Management (BLM), is the first of a series of maps designed to demonstrate how geologic and mineralogical information can be used on a regional scale to help assess the potential for mineral-related and natural environmental problems in mining districts, unmineralized areas, and surrounding watersheds. A GIS (Geographic Information System) format was used to integrate geologic, geochemical, water-quality, climate, land-use, and ecological data from diverse sources. Colorado was chosen as the prototype because of the extensive geologic and geochemical data available digitally from the USGS, the State of Colorado, and other sources. Similar maps can be developed for other states or areas at a variety of scales.

The map shows the potential for mineral-related problems that may exist in Colorado metal-mine districts, as indicated by the geologic characteristics of the mineral deposits that occur in the respective districts. Likely mine-drainage signatures are defined for each mining district based on: (1) a review of the geologic characteristics of the mining district, including mineralogy, trace-element content, host-rock lithology, and

wall-rock alteration, and (2) results of site-specific studies on the geologic controls on mine-drainage composition (Smith and others, 1994; Plumlee and others, 1993; Ficklin and others, 1992).

Uses of the Map

This map was designed primarily for use in abandoned mine lands assessments, but can also be used for other applications such as watershed and ecosystem management, and environmental prediction and mitigation.

In abandoned mine lands assessments, land management agencies can use the map to help screen and identify mining districts with the greatest potential for acid-mine drainage, so that these districts can receive higher priority for detailed on-site characterization studies and remediation. The map is not intended to replace on-site field studies of abandoned mine sites. The map can be used, however, to supplement information gathered during field studies. In some on-site assessments, time and monetary constraints allow collection of only minimal water-quality data such as the pH and conductivity of mine-drainage waters. For these sites, the map can be used to place constraints on the types and concentration ranges of metals that are likely to be present in drainage waters. The map can also be used to estimate the composition of waters draining mine sites during storm events, waters that on-site sampling would measure only if fortuitously timed to coincide with the storm event.

Wildlife managers can use information on the map to help screen mining districts or unmineralized areas that have the highest potential for contributing metals of particular concern for aquatic life or wildlife health. The map also can be used to guide regional water-quality studies by identifying watersheds potentially affected by natural or mining-related metal contamination.

The map can be used by both the mining industry and land management or regulatory agencies to better estimate and locate the potential for mineral-related problems that may result from future mineral resource development. Combined with standard regulatory procedures used to predict acid-drainage potential (Ferguson and Moran, 1991), the geologic information can be incorporated into the earliest phases of the geologic characterization of the mining district, including mineralogy, trace-element content, host-rock lithology, and

mineral deposit types in defined geographic areas and climates. Similarly, the most beneficial mitigative procedures can be determined based on the geologic information developed on-site, thereby avoiding more costly remedial procedures after development and potential environmental problems have started.

Acknowledgments

Numerous digital data files were generated in the process of making the Colorado assessment map and many derivative digital data files were made by combining files or portions of files. The mine-drainage hazards digital data file was compiled by Geoffrey S. Plumlee, Kathleen S. Smith, and Walter H. Ficklin from the USGS. The mineralogical digital data file was compiled by Randall K. Streufert, Mark W. Davis, and James A. Carpenter from the USGS and Alan R. Wallace, Margo I. Toth, J. Thomas Nash, Geoffrey S. Plumlee, and Steven M. Smith from the USGS. Digital data compilation and graphical manipulation were done by Steven M. Smith, Gregory S. Lee, Geoffrey S. Plumlee, and Richard Tripp from the USGS and David Taylor from the BLM. Reviewers Byron R. Berger and Arthur Bookstemon, and editors Sherman Marsh and Susan Kropfshofer provided many helpful suggestions that greatly improved the map's quality.

Map Preparation

Numerous digital data files exist for Colorado, including streams, cities, drainage divides, geospatial data, land use data, geology, and many more. Data files showing locations of streams, cities, drainage divides, and roads are available on CD-ROM in digital file format (DLG format) (U.S. Geological Survey, 1990, 1992). The DLG files were downloaded and translated into files compatible with the program GSI/MAP for use on IBM-compatible computers (Seiner and Taylor, 1992). Boundaries of national forests, national parks, and monuments, and Indian reservations were digitized from BLM land classification maps. Additional digital files including geologic information, mining districts, and other pertinent data were digitized using GSI/MAP. All digital files were then converted to ASCII coordinate files and transferred to an Apple Macintosh computer. On this map, the Macintosh, the data files were plotted using the program

MacGrid, by RockWare, Inc. For final map preparation, the graphics files were brought as layers into the object-oriented graphics program Canvas, by Deneba Software, for final map preparation. All data layers used in this compilation are currently being translated into ARC/INFO format for future use in a GIS environment.

Mining Districts

Locations and boundaries of major mining districts as well as locations of mining projects were compiled from several sources. Maps prepared by the CGS were used to identify the locations and boundaries of major gold-producing districts, all metal-mining districts, and smaller metal-mining projects (Streufert and Davis, 1990; Streufert and Capps, 1994). Locations and boundaries of uranium and vanadium districts (Streufert and Davis, 1990; Streufert and Capps, 1994) and others (1985). Fluorspar districts were compiled from Brady (1975). A limited number of other mining districts were added to the map.

Unmineralized Areas

There are a number of areas outside known Colorado mining districts that are mineralized but that have not had appreciable mining activity. These mineralized areas, such as the headwaters of the Alamosa River south of Summitville (Bove and others, 1995), typically contain abundant sulfide minerals such as pyrite (iron sulfide) that, when weathered, can be a significant source of natural acid drainage and metal contamination. Unmineralized areas that occur outside the known mining districts are identified on the map to the best of our knowledge. A number of Colorado mountains (areas near or within the 20-in. precipitation contour), there is a much greater potential for perennial or long-term epithermal surface drainage from rain workings or springs. Due to the larger volumes of water and lower rates of evaporation, metals and acid in these waters may persist well downstream from mines. There is, at the same time, a relatively high likelihood that the contaminated waters could be diluted from unmineralized waters draining unmineralized areas. Evaporation and the formation of soluble salts remain an important process in cooler, winter climates if rainfall or snowfall occurs sporadically. Plumlee, Gray, and others (1995) and Plumlee, Smith, and others (1992) documented the important role of soluble salts in the generation of acidic, metal-rich pulses from the Summitville mine during snowmelt or summer thunderstorms that follow extended dry periods; the mine is located at 11,500 feet elevation in an area receiving more than 50 in./yr. of precipitation.

Rivers and Streams Affected by Metals

The Colorado Water Quality Control Division (WQCD) periodically reviews Colorado rivers and streams for impact from a variety of contaminants such as metals, fertilizers, and bacteria, and prepares maps showing the results of the reviews. The layer on this map showing streams and rivers affected by metals was modified from a map prepared by the WQCD (1989). Affected reaches of streams and rivers were identified by WQCD as those for which aquatic life and/or agricultural uses frequently were exceeded for one or more of various metals such as copper, zinc, and cadmium.

Mean Annual Precipitation

Contours of mean annual precipitation amounts were digitized from a map developed by the Colorado Climate Center (1984), on the basis of data contained in Doerken and others (1984). Only areas receiving more than 20 in. of precipitation annually are shown on the map. The 20 in. mean annual precipitation contour was chosen somewhat arbitrarily, but it represents what we believe to be the approximate boundary between predominantly dry (arid) and moist (semi-arid) parts of the state.

Climate Factors

Climate factors such as mean annual precipitation must be considered because of their effects on ground- and surface-water flow, water chemistry, and evaporation. In arid climates, ground-water tables are generally deep, and are intersected by open pits, underground mine workings generally intersect the water tables only in areas of steep

topography. Open pits or waters that do flow from springs or mines have a high likelihood of evaporating or flowing underground rather than contaminating the surface environment for long distances downstream. Due to the lack of weathering, chemical weathering tends to be much more limited in arid climates than in wet climates. When acid waters do form as a result of occasional storm events, they eventually evaporate and leave behind their metals and acid in the form of highly soluble salts (Plumlee, Gray, and others, 1995; Plumlee, Smith, and others, 1995). These salts can be transported by wind for significant distances and are readily dissolved during subsequent storm events. Thus, in arid and semiarid climates, there are some rivers, such as the Yampa River and some of its tributaries west of Steamboat Springs, that are affected by metals released by coal mining activities.

Mine-Drainage Models of Mineral Deposits and Likely Mine Drainage Signatures of Mining Districts

A compilation of mine-drainage compositions from diverse mineral deposit types shows that mine waters draining mineral deposits with different geologic characteristics have metal concentrations and pH values that cluster within distinctive ranges (Smith and others, 1994; Plumlee and others, 1993; Plumlee and others, 1992; Ficklin and others, 1992). In addition, the elemental suites of metals present in mine-drainage waters typically reflect the same characteristic suites of metals contained in the mineral deposits. For example, deposits containing pyrite (an iron sulfide), enargite (a copper-arsenic sulfide), and other copper sulfides in highly altered host rocks at Summitville, Colorado predominantly generate highly acidic waters with thousands of parts per million (ppm) dissolved iron and aluminum, hundreds of ppm dissolved copper and zinc, and several to tens of ppm dissolved cadmium, cobalt, arsenic, nickel, and uranium. These water compositions result from the lack of acid buffering capacity in the highly altered wall-rocks, coupled with the wall-rock nature of the ores. In addition, the copper-rich nature of the ores at Summitville is reflected by similar enrichments of copper over zinc in the mine-drainage waters compared to those draining most other mineral deposit types. In contrast, deposits of pyrite, sphalerite (a zinc sulfide), and galena (a lead sulfide) that occur in carbonate-rich host rocks at Leadville, Colorado tend to generate

drainage waters that can contain high levels of dissolved zinc, but are generally of near neutral pH. Mine-drainage waters may also be affected by the mining technique used to extract the ore. Generally, waters from open-pit mines and water draining mine dumps tend to have somewhat more acidic, metal-rich compositions than waters draining underground workings; this is related to the increased surface areas exposed to weathering in open-pit mines and mine dumps, increased availability of oxygenated waters, and increased opportunities for evaporative concentration (Plumlee and others, 1993).

These empirical studies show that, given a good knowledge of mineral deposit geology in a given district, it is possible to estimate or predict likely ranges of pH and metals present in mine-drainage waters. The mine-drainage model of mineral deposits (Smith and others, 1994; Plumlee and others, 1993), for example, shows that mine waters from a given district grouped according to their geologic characteristics and their resulting potential for mine-drainage hazards. The deposit types are listed in the map explanation in decreasing order of acidity and metal content (that is, increasing water quality of their potential mine-drainage compositions). This map and other best knowledge of the geology of each district.

Because mineral-deposit geology is highly complex, with significant spatial variations in ore mineralogy, host rock lithology, and host rock alteration possible in the same mining district, the explanation shows a range in mine-drainage compositions that might occur in different mineralogical zones, wall-rock alteration zones, or host rocks for various deposit types within a given mining district.

Empirical data on mine drainage compositions have not been collected for all deposit types and all geologic characteristics. In districts containing deposit types for which empirical data are lacking, some limits on drainage water compositions can be inferred by analogy with deposit types of similar geology.

References

Beach, R.A., Gray, A.W., Peterson, E.K., and Roberts, C.A., 1985. Availability of Federal land for mine development and development in western states—Colorado, 1984. U.S. Bureau of Mines Special Publication 40, p. 2 pl.

Bove, D.J., Barry, T., Kirtz, J., Hon, K., Wilson, A.B., VanLoenen, R.L., and Kartham, R.K., 1995. Geology of hydrothermal altered areas within the upper Alamosa River basin, Colorado, and probable effects on water quality. In Posy, H.H., Pendleton, J.A., and Van Zyl, D., eds., Summitville Fluorspar Proceedings; Colorado Geological Survey Special Publication 28, p. 5-41.

Brady, B.T., 1975. Map showing fluorapatite deposits in Colorado. U.S. Geological Survey Mineral Investigations Series Map 70, scale 1:250,000.

Colorado Climate Center, 1984. Colorado Average Annual Precipitation Map, 1984. Colorado Water Quality Control Division, 1989. Colorado River Basin Water Quality Control Plan, 1984. Analysis of Colorado average annual precipitation for the 1951-1980 period. Fort Collins, Colorado: U.S. Geological Survey, Report 84-4, 53 p.

Doerken, N.J., McKee, T.B., and Morin, K.A., 1991. The prediction of acid rock drainage—lessons from the database. Proc. 2nd Int. Conference on the Abatement of Acidic Drainage, MEND Program, v. 3, p. 85-106.

Ficklin, W.H., Plumlee, G.S., Smith, K.S., and McHugh, J.B., 1992. Geochemical classification of mineral deposits and natural drainages in mineralized areas. Proc. 7th Int. Conference on the Abatement of Acidic Drainage, MEND Program, v. 3, p. 381-384.

Kwong, Y.T.J., 1993. Prediction and prevention of acid rock drainage from a geologic and mineralogical perspective. Ontario, Canada, MEND Program, Project 1.32.1.

Plumlee, G.S., Gray, J.E., Roebler, M.M., Jr., Coohough, M., and Van Zyl, D., eds., Proceedings, Summitville Fluorspar '95, Colo. Geol. Survey Special Publication 38, p. 13-22.

Plumlee, G.S., Smith, K.S., and Ficklin, W.H., 1994. Geochemical models of mineral deposits, and geologic characteristics of mineral deposits in Colorado. U.S. Geological Survey Open-File Report 94-203, 7 p.

Plumlee, G.S., Smith, K.S., Ficklin, W.H., and Briggs, P.H., 1992. Geologic and geochemical controls on the composition of mine drainages and natural drainages in mineralized areas. In Posy, H.H., Pendleton, J.A., and Van Zyl, D., eds., Proceedings, Summitville Fluorspar '95, Colo. Geol. Survey Special Publication 38, p. 23-34.

Seiner, G.L., and Taylor, R.K., 1992. System 9, GSI/MAP, other programs for the IBM PC and compatible microcomputers to assist workers in the earth sciences. U.S. Geological Survey Open-File Report 93-511, 363 p.

Smith, K.S., Plumlee, G.S., and Ficklin, W.H., 1994. Predicting water contamination from metal mines and mining waste: Notes, Workshop No. 2. Int. Conf. on the Abatement of Acidic Drainage. U.S. Geological Survey Open-File Report 94-264, 112 p.

Streufert, R.K., and Taylor, R.K., 1994. Location map and descriptions of metal occurrences in Colorado with notes on geochemical implications. Colorado Geological Survey Series 28, 35 p.

Streufert, R.K., and Davis, M.W., 1990. Gold districts and placers of Colorado. Part 1 of 2 in Davis and Streufert, Gold occurrences of Colorado. Colo. Geol. Survey Resource Series 28, 11 p.

U.S. Geological Survey, 1990. 1:250,000-scale Digital Line Graph (DLG) data. Hydrography and transportation. U.S. Geological Survey, Series DD2-4, CD-ROM.

U.S. Geological Survey, 1993. 1:100,000-scale Digital Line Graph (DLG) data. Hydrography and transportation. U.S. Geological Survey, U.S. Geodata, Area 9—Central Plains States, CD-ROM.

Vanderwall, J.W., 1947. Mineral resources of Colorado. Denver: Mineral Resources Board, 547 p.

MAP SHOWING POTENTIAL METAL-MINE DRAINAGE HAZARDS IN COLORADO, BASED ON MINERAL-DEPOSIT GEOLOGY

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