



Deposit Model for Volcanogenic Uranium Deposits

By George N. Breit and Susan M. Hall

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Introduction

Volcanism is a major contributor to the formation of important uranium deposits both close to centers of eruption and more distal as a result of deposition of ash with leachable uranium (Nash, 2010). Hydrothermal fluids that are driven by magmatic heat proximal to some volcanic centers directly form some deposits. These fluids leach uranium from U-bearing silicic volcanic rocks and concentrate it at sites of deposition within veins, stockworks, breccias, volcanoclastic rocks, and lacustrine caldera sediments (Goodell, 1985). The volcanogenic uranium deposit model presented here summarizes attributes of those deposits and follows the focus of the International Atomic Energy Agency (IAEA) caldera-hosted uranium deposit model. Although inferred by some to have a volcanic component to their origin, iron oxide-copper-gold (IOCG) deposits with economically recoverable uranium contents are not considered in this model.

The IAEA's tabulation of volcanogenic uranium deposits lists 100 deposits in 20 countries, with major deposits in Russia, Mongolia, and China. Collectively these deposits are estimated to contain uranium resources of approximately 500,000 tons (t) of uranium (U), which amounts to 6 percent of the known global resources (IAEA, 2009). Prior to the 1990s, these deposits were considered to be small (less than 10,000 t U) with relatively low to moderate grades [0.05 to 0.2 weight percent (wt.%) U]. Recent availability of information on volcanogenic uranium deposits in Asia highlighted the large resource potential of this deposit type. For example, the Streltsovskoye district in eastern Russia produced more than 100,000 t U as of 2005; with equivalent resources remaining (IAEA, 2009; Dahlkamp, 2010a). Known volcanogenic uranium deposits within the United States are located in Idaho, Nevada, Oregon, and Utah. These deposits produced an estimated total of 800 t U during mining from the 1950s through the 1970s and have known resources of 30,000 t U (IAEA, 2009). The most recent estimate of speculative resources proposed an endowment of 200,000 t U (Department of Energy, 1980).

Geochemical and Geologic Setting

Formation of volcanogenic uranium deposits is dependent on extraction of uranium from felsic rocks, transport by hydrothermal solutions, and deposition induced by chemical or physical changes. Uranium concentrations in volcanic rocks increase with differentiation to more felsic compositions. Midocean ridge basalts typically contain 0.2 parts per million (ppm) U (Kelemen and others, 2004), whereas the median content of U in rhyolite in the United States is 5 ppm (Cenozoic rhyolites, n=1,400; U.S. Geological Survey, 2011). Felsic volcanic rocks proximal to uranium deposits typically contain uranium concentrations in excess of 10 ppm. Enriched uranium contents are characteristic of rocks with aluminous and alkaline affinities (Cuney and Kyser, 2009). This favorability is partly attributed to the low degree to which silica is polymerized in aluminous and alkaline melts. The depolymerization favors retention of uranium and other high-field strength elements in the magma, rather than

sequestering these elements in early formed refractory phases such as monazite or zircon. High fluorine contents that are typical of uranium-enriched volcanic rocks, similarly limits linking of silica in magmas, which favors the retention of uranium in the melt.

Volcanogenic uranium deposits formed in diverse tectonic settings including: rifts (Osamu Utsumi, Brazil; Lapido Loureiro and others, 1988), hot spots (McDermitt caldera, Nevada; Castor and others, 1996), back arcs (Macusani, Peru; Cheilletz and others, 1992), and subduction zones where the stress field has shifted to be extensional (Marysvale, Utah; Cunningham and others, 1998). The age of rocks hosting these deposits ranges from Proterozoic through the Tertiary; no change in deposit characteristics is apparent with age. Although the origin of felsic magmas in all favorable settings is complex and controversial, enhanced heat flow because of thinning of the crust and rise of mafic magmas likely induce partial melting of relatively uranium-rich upper crustal rocks to create melt compositions that favor enrichment of uranium and other incompatible elements (estimated average U content of upper continental crustal rocks is 2.8 ppm U; Rudnick and Gao, 2004).

Vein, stockwork, breccia-hosted, and stratabound deposits proximal to volcanic centers have generally similar grades, ore mineralogy, and share common clay mineral alteration assemblages. Vein and stockwork deposits fill fractures and faults that cut volcanic, subvolcanic, and basement rocks. They extend for tens to several hundreds of meters along strike and tens to hundreds of meters in dip. Subvolcanic eruption breccias form pipes that host uranium deposits tens of meters in diameter; the dimensions of these breccias typically decrease with depth. Eruption breccia, flow breccia, and volcanoclastic rock host deposits that are conformable to the bounds of these units, although the feeder for ore-forming fluids is commonly a fracture. These stratabound deposits can extend for hundreds of meters laterally and range from less than one to tens of meters in thickness. Stratiform deposits hosted by lacustrine sediments deposited within a caldera share characteristics of more distal sedimentary-type uranium deposits in which air-fall volcanic ash has a prominent role, but are included in this model because of the inferred role of magmatic heat in fluid circulation. These intracaldera lacustrine deposits can extend laterally for several hundreds of meters and be meters to tens of meters thick. Generally these deposits are of lower grade than other types of volcanogenic deposits.

Ore Mineralogy and Alteration

Primary ore phases in volcanogenic uranium deposits are typically a mixture of uraninite, pitchblende, and coffinite, with brannerite and uranothorite common in deposits formed in more alkaline rocks. Significant contents of fluorite and molybdenite (jordesite) are common in most volcanogenic uranium deposits; in some deposits they are sufficiently abundant to be recovered as byproduct commodities. Deposits in peralkaline rocks have associated enrichments of rare-earth elements, thorium, and zirconium. A few Hercynian deposits in Europe and Asia are polymetallic as a result of a complex, multistaged hydrothermal history and are noted for their enrichments of copper, lead, silver, and zinc (Dahlkamp, 2010a).

Pre-ore alteration at many deposits includes alkali metasomatism that is followed by varying amounts of quartz, sericite, pyrite, and carbonate mineral veining. Minerals produced during wall-rock alteration include: albite, potassium feldspar, smectite, illite (sericite), kaolinite, dickite, chlorite, calcite, hematite, and pyrite. Growth of uranium minerals is commonly associated with argillic alteration and fluorite. Post-ore fluids redistribute uranium, and produce late-stage carbonate minerals, sulfates, and additional argillic alteration. Alteration of wall rock along vein margins is limited to a few meters, whereas the alteration of the eruption breccias and volcanoclastic deposits is more complete likely reflecting the greater permeability of these host rocks. Paragenesis of the ore and gangue phases is

difficult to resolve, and is considered as evidence for multiple fluids with distinct compositions during the history of the uranium mineralizing system.

Mineralizing Conditions

Evidence for the composition of fluids responsible for volcanogenic uranium mineralization is based on stable isotopic and fluid inclusion studies of gangue minerals. Results of these analyses support the mixing of meteoric and magmatic water during formation of most deposits. Volatiles, notably fluorine, chlorine, and CO₂ are considered to have a magmatic origin. Fluid inclusion studies indicate mineralization temperatures between 150 and 300 °C and fluid compositions with low to moderate salinities (less than 8 wt.% KCl eq.) (Dahlkamp, 2010a; 2010b). Smaller, low-temperature deposits (less than 100 °C) within volcanoclastic rocks are speculated to have formed by the release of uranium during weathering of the volcanic rocks.

Uranium is generally thought to be transported in the ore-forming fluids as the more soluble U(VI) species (uranyl ion), rather than as U(IV), which is the oxidation state in the primary ore minerals. Accordingly, formation of ore is attributed to chemical reduction. Proposed reductants include reduced sulfur species (H₂S, thiosulfate, and other polythionates) (Nash, 2010) and ferrous iron in wall-rock minerals. Carbonaceous matter in lacustrine sediments and some volcanoclastic rocks likely contributed to reduced sulfur forms in those rocks by local low-temperature bacterial sulfate reduction. Reduction of uranium by dissolved sulfur species would require mixing of oxidized and reduced fluids, which is problematic in some vein systems. Reaction with ferrous iron in the wall rock is offered to account for the frequent association of hematite with reduced uranium oxides in some deposits. Although silicic volcanic rocks have relatively low iron content, only 0.5 weight percent Fe²⁺ is necessary to reduce sufficient uranium to produce a volumetrically equivalent rock containing 1 wt.% U. The median iron content of Cenozoic rhyolites in the United States is 1.7 wt.% (U.S. Geological Survey, 2011). In some systems boiling of the ore-forming fluid, whereby soluble uranium complexes are destabilized by loss of CO₂ and fluoride, may have been more significant than uranium reduction in ore formation (Cunningham and others, 1998).

Weathering Effects

Weathering of volcanogenic uranium deposits typically forms supergene deposits of potential ore grade. Minerals in weathered deposits include silicates (uranophane and beta-uranotile), phosphates (autunite), hydroxides (curite), and some vanadates (carnotite). Gangue minerals associated with the supergene accumulations are secondary kaolinite and smectite with ferric oxyhydroxides.

Exploration

Areas favorable for volcanogenic uranium deposits typically have uraniferous basement rocks. Partial melting and fluid interaction with the basement are thought to contribute to the uranium content of the felsic volcanic rocks that host the deposits. Late Mesozoic deposits in Russia and China are known to overlie felsic Proterozoic basement rocks with uranium contents in excess of 10 ppm. The area of the giant Strel'tsovskoye district experienced two episodes of silicic magmatism marked by uranium enrichment, prior to the late Mesozoic volcanism that is linked to ore formation (Golubev and others, 2010). The identification of regions with progressive concentration of uranium by successive magmatic episodes may aid in identification of favorable terrain.

Detection of volcanogenic uranium deposits relies on geophysical and geochemical surveys, as well as subsequent target drilling. Radiometric measurements are the most effective geophysical

method for detection of potential areas for volcanogenic uranium deposits. This approach would detect uranium enrichment in deposits, as well as favorable rocks to host these deposits. Elevated concentrations of uranium, fluoride, and molybdenum in rock, water, and stream sediments are favorable indicators. At Streltsovskoye drill testing of fluorite veins that have no apparent uranium enrichment at the surface was the best technique to locate ore bodies. Volcanogenic uranium deposits that have no surface expression are suited for exploration methods applied to other uranium deposits, including geochemical sampling of soil gas (radon) and groundwater for uranium and indicator elements.

Environmental Concerns

Environmental hazards associated with volcanogenic uranium deposits are attributed to the release of uranium, radioactive daughter products, and associated elements. Exposure to uranium in dusts and water is regulated because of its chemical toxicity. Oxidation of the primary uranium minerals during weathering or exposure during mining operations produces the soluble uranyl ion. This form of uranium is readily transported in surface and groundwater, although reactions along the transport path of solutions with high contents of uranium near a uranium deposit may cause low solubility uranium silicates, phosphates, hydroxides, or vanadates to form. Alternatively, even small concentrations of dissolved uranyl ion can accumulate to economic concentration in reducing environments such as wetlands. The U.S. Environmental Protection Agency has set the drinking-water standard for uranium to be 30 µg/L, which can be greatly exceeded in uranium mines (Caine and others, 2011). The products of radioactive decay, notably radium and radon, are considered radiological hazards and are also regulated by the U.S. Environmental Protection Agency. Radium forms a soluble ion that is readily leached from uranium ore. Dissolved sulfate can limit radium transport by forming low-solubility minerals such as barite which will incorporate radium. Radon is a short-lived radioactive gas that is a likely hazard proximal to the uranium ore bodies and waste piles. In contrast to many base-metal deposits, uranium deposits have generally low contents of sulfide minerals, and typically have gangue mineral and wall-rock composition with moderate to high acid-buffering capacities. The circum-neutral pH of drainage from volcanogenic uranium deposits limits the mobility of many elements associated with these deposits. Nonetheless, concentrations of dissolved fluoride and molybdenum may exceed regulatory standards in areas of undeveloped and mined volcanogenic uranium deposits.

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