

Prepared in cooperation with the Triangle Area Water Supply Monitoring Project
Steering Committee

Occurrence and Distribution of Mercury in Streams and Reservoirs in the Triangle Area of North Carolina, July 2007–June 2009



Scientific Investigations Report 2021–5027

Cover. Lake Michie at dam near Bahama, North Carolina (U.S. Geological Survey streamgaging station 02086490). Photograph by Ryan B. Rasmussen, U.S. Geological Survey.

Back cover. Little River at SR1461 near Orange Factory, North Carolina (U.S. Geological Survey streamgaging station 0208521324). Photograph by Ryan B. Rasmussen, U.S. Geological Survey.

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By Anna M. McKee, Sharon Fitzgerald, and Mary Giorgino

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Conversion Factors

U.S. customary units to International System of Units

Multiply	By	To obtain
Length		
inch (in.)	25.4	millimeter (mm)
foot (ft)	0.3048	meter (m)
mile (mi)	1.609	kilometer (km)
Area		
square mile (mi ²)	2.590	square kilometer (km ²)
Flow rate		
cubic foot per second (ft ³ /s)	0.02832	cubic meter per second (m ³ /s)

International System of Units to U.S. customary units

Multiply	By	To obtain
Length		
millimeter (mm)	0.03937	inch (in.)
meter (m)	3.281	foot (ft)
kilometer (km)	0.6214	mile (mi)
Area		
square kilometer (km ²)	0.3861	square mile (mi ²)
Flow rate		
cubic meter per second (m ³ /s)	35.31	cubic foot per second (ft ³ /s)

Temperature in degrees Celsius (°C) may be converted to degrees Fahrenheit (°F) as follows:
 $^{\circ}\text{F} = (1.8 \times ^{\circ}\text{C}) + 32$.

Datum

Vertical and horizontal coordinate information is referenced to the North American Datum of 1983 (NAD83).

Altitude, as used in this report, refers to distance above the vertical datum.

Supplemental Information

Specific conductance is given in microsiemens per centimeter at 25 degrees Celsius ($\mu\text{S}\cdot\text{cm}^{-1}$ at 25 °C).

Concentrations of chemical constituents in water are given in either milligrams per liter ($\text{mg}\cdot\text{L}^{-1}$) or micrograms per liter ($\mu\text{g}\cdot\text{L}^{-1}$).

Abbreviations

Chloro <i>a</i>	chlorophyll <i>a</i>
CV-AAS	cold vapor-atomic absorption spectrometry
CV-AFS	cold vapor-atomic fluorescence
DO	dissolved oxygen
DOC	dissolved organic carbon
FDA	U.S. Food and Drug Administration
FHg	filtered mercury in water
Hg	mercury
IQR	interquartile range
irl	interim reporting level
ISQG	interim sediment-quality guideline
<	less than
lrl	laboratory reporting level
MDL	method detection level
MeHg	methylmercury
MRL	method reporting level
N	number of samples
NADP	National Atmospheric Deposition Program
NCWQC	North Carolina water-quality criterion
NIST	National Institute of Standards and Technology
NRC	Nuclear Regulatory Commission
NWQL	U.S. Geological Survey National Water Quality Laboratory
Pheo <i>a</i>	pheophytin <i>a</i>
RPD	relative percent difference
RSD	relative standard deviation
SC	specific conductance
SD	standard deviation
SO ₄ ²⁻	sulfate
SSC	suspended-sediment concentration
TAWSMP	Triangle Area Water Supply Monitoring Project
THg	total mercury in bed sediment

TOC	total organic carbon
TOC _{sed}	total organic carbon in bed sediment
TOC _{water}	total organic carbon in water
TRHg	total recoverable mercury (in water)
USEPA	U.S. Environmental Protection Agency
USGS	U.S. Geological Survey

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Abstract

During the time period 2001–2006, the U.S. Geological Survey reported mercury-concentration measurements that exceeded the North Carolina water-quality criterion (NCWQC) of 0.012 microgram per liter for total recoverable mercury in streams and reservoirs across the Triangle Area of North Carolina. Mercury data were sparse, however, generally consisting of only one or two water samples per year. Additional monitoring and data analysis were needed to better determine the occurrence and distribution of mercury in the Triangle Area for all seasons and waterbody types as well as associations between mercury concentrations and water-quality and land-use parameters. Water at fifteen reservoir and 14 stream sites across the Triangle Area was sampled at various times between August 2007 and June 2009, with water samples collected from both the surfaces and bottoms of the water columns in reservoirs and from the surfaces of streams. A bed sediment sample was also collected at all reservoir sites and at all but one stream site. A total of 301 water samples was collected at reservoir sites. Filtered and total recoverable mercury were detected in at least one water sample collected from each reservoir site. A total of 77 water samples was collected from stream sites with filtered mercury detected in samples from one-half of these sites, and total recoverable mercury detected in at least one water sample from all but two sites. Total recoverable and filtered mercury concentrations exceeded the NCWQC for mercury more frequently in reservoir than in stream samples. Differences in sampling frequencies among seasons and between streams and reservoirs, however, may have negatively biased overall estimates of mercury concentrations in streams relative to reservoirs. Filtered mercury concentrations in surface-water samples from reservoirs and total recoverable mercury concentrations in bottom samples from reservoirs were highest in the fall, whereas no seasonal trends in filtered or total recoverable mercury were detected from stream samples. Total mercury concentrations were calculated for the bulk sample on the basis of the percentage of the grains in the bulk sample whose diameters that were smaller than 0.0625 millimeters. Total mercury concentrations in bed sediment were generally higher for samples

from reservoir sites compared to streams sites, although the highest total mercury concentration in bed sediment was from a stream site. Concentrations of total recoverable mercury in water samples from stream sites all fell within the general range for streams and lakes without on-site significant anthropogenic sources (for example, mercury mines or industrial pollution), whereas samples collected from eight reservoir sites had total mercury concentrations in a range characteristic of sites affected by mercury mines or industrial pollution. Results suggested that litterfall may be a source of mercury in streams, whereas atmospheric deposition is likely a dominant source for reservoirs; however, high concentrations of filtered and total recoverable mercury concentrations in the fall season in some reservoir-water samples may warrant further analysis of potential hydrologic factors. Mercury concentrations in all water and bed sediment samples were below levels expected to cause adverse effects to humans and aquatic biota, indicating that mercury levels at the study sites in the Triangle Area were unlikely to cause an immediate health risk to humans or aquatic organisms. The high variability among several sample replicates for total recoverable mercury, however, indicated that inferences from total recoverable mercury concentrations can be tenuous.

Introduction

The human-health risks from dietary exposure to methylmercury (MeHg) are a large driver of public concern over mercury contamination in the environment and aquatic food webs. Methylmercury bioaccumulates in fish and can cause neurotoxic effects if consumed by humans, particularly in developing fetuses and young children. Whereas mercury in the environment can come from natural causes like volcanic eruptions, an estimated two-thirds of the mercury mass in the current global flux originated in anthropogenic emissions from industrial processes and gold mining (Mason and others, 1994). In the 1970s, the United States and other industrialized countries made concerted efforts to reduce intentional mercury discharges into surface waters (Turner and Southworth, 1999), which generally resulted in decreased mercury levels in fish

and sediments at sites affected by industrial-mercury discharges (Wiener and others, 2003). The dispersion of mercury as a result of mercury cycling and atmospheric deposition, however, has resulted in mercury contamination even in areas in which no major anthropogenic source has ever been nearby.

Based on fish-tissue data routinely collected by the North Carolina (N.C.) Department of Environment and Natural Resources, a statewide advisory to limit or avoid consumption of fishes that could contain high levels of mercury was developed by the N.C. Department of Health and Human Services. Because of the statewide fish-consumption advisory, the State designated all named water bodies as impaired for mercury (North Carolina Department of Environment and Natural Resources, 2012). The U.S. Geological Survey (USGS), in partnership with local governments in the Triangle Area of North Carolina, has monitored mercury concentrations in streams and reservoirs since 1988 as part of a larger water-quality and streamflow-monitoring effort known as the Triangle Area Water Supply Monitoring Project (TAWSMP; Garrett and others, 1994).

From the inception of the mercury-monitoring program through 2001, mercury detections were rare in the Triangle Area. A methodological advancement in mercury detection and quantification implemented by the USGS National Water Quality Laboratory in 2001, however, improved the reporting level of by an order of magnitude from 0.1–0.2 microgram per liter ($\mu\text{g}\cdot\text{L}^{-1}$) to $0.01\ \mu\text{g}\cdot\text{L}^{-1}$ (Garbarino and Damrau, 2001). From 2001 to 2006, after the methods modification, mercury was frequently detected in the study streams and reservoirs, and in many instances total recoverable mercury (TRHg) was equal to or exceeded $0.012\ \mu\text{g}\cdot\text{L}^{-1}$, the North Carolina water-quality criterion (NCWQC) for mercury in freshwater (North Carolina Department of Environment and Natural Resources, 2007). Additionally, the new-method detection level¹ (MDL) of $0.005\ \mu\text{g}\cdot\text{L}^{-1}$ (Garbarino and Damrau, 2001) was well below the NCWQC for mercury. The NCWQC for mercury was derived from the 1984 U.S. Food and Drug Administration (FDA) fish-tissue action level for protecting human health from the consumption of methylmercury in fish and divided by a mercury-bioconcentration factor (North Carolina Department of Environment and Natural Resources, 2012). Whereas the FDA fish-tissue action level was developed for methylmercury, the NCWQC for mercury applies to TRHg, which includes MeHg, based on the logic that applying the criteria to TRHg would be protective of human health regardless of the proportion of TRHg was that was MeHg.

Whereas North Carolina does not regulate mercury concentrations in bed sediment, criteria (concentration limits) have been set by other States and countries. The Canadian Government recognizes 0.17 milligram per kilogram ($\text{mg}\cdot\text{kg}^{-1}$) as the interim sediment-quality guideline (ISQG) for mercury and $0.486\ \text{mg}\cdot\text{kg}^{-1}$ as the probable-effect level for aquatic biota

in freshwater (Smith and others, 1996; Canadian Council of Ministers of the Environment, 1999). Ranges below the ISQG are defined as rarely associated with adverse biological effects in freshwater biota; values between the ISQG and probable-effect level are occasionally associated with adverse biological effects; and values above the probable-effect level are frequently associated with adverse biological effects (Environment Canada, 1997). MacDonald and others (2000) developed another set of guidelines for sediment-dwelling biota to account for uncertainties in the application of the existing published sediment-quality guidelines. In the revised criteria, MacDonald and others (2000) suggested that bed sediment mercury concentrations below the threshold-effect concentration of $0.18\ \text{mg}\cdot\text{kg}^{-1}$ are not expected to cause adverse effects on sediment-dwelling organisms, but mercury concentrations in bed sediment above the probable-effect concentration of $1.06\ \text{mg}\cdot\text{kg}^{-1}$ were expected to cause adverse health effects in sediment-dwelling organisms. The threshold-effect concentration ($0.18\ \text{mg}\cdot\text{kg}^{-1}$) is commonly used as the benchmark by States (in the United States) in which mercury concentrations in bed sediment are monitored (for example, U.S. Environmental Protection Agency [USEPA], 2010).

Until 2007, mercury data throughout the study area were sparse—in general, measured in only one or two water samples per year. Additional monitoring and data analysis were needed to better determine the occurrence and distribution of mercury in the Triangle Area and factors associated with differences in mercury concentrations. To assess the status of mercury in streams and reservoirs across the Triangle Area, an intensive synoptic assessment of the occurrence and distribution of mercury at 14 streams and 15 reservoirs was conducted from 2007 through 2009. Associations between mercury concentrations in water and bed sediment samples were investigated, and data related to potential driving factors including season, ancillary constituents, precipitation, and land use within the drainage areas were collected and analyzed. Information from this study could be used to identify locations with mercury concentrations of potential concern and to assess the spatial and temporal variability of mercury levels in the Triangle Area.

Purpose and Scope

The purpose of this study was to describe the occurrence and distribution of mercury concentrations and the frequency of exceedances of the NCWQC for mercury at selected reservoir and stream sites around the Triangle Study Area. A secondary objective was to investigate potential seasonal, ancillary-constituent, precipitation, and land-use drivers associated with mercury concentrations in water and bed sediment. Nonpoint sources, including atmospheric deposition and surface runoff, were believed to be the primary sources of mercury in the study system (Sorensen and others, 1990; Fitzgerald and others, 1998; North Carolina Department of Environment and Natural Resources, 2012). Although North Carolina regulates mercury loads from permitted wastewater sites (National Pollutant Discharge Elimination System), these

¹Method detection level is the lowest concentration that can be measured and reported with 99% confidence that the analyte concentration is greater than zero (USEPA, 1997a).

facilities generally contribute lower quantities of mercury to surface waters in North Carolina than does atmospheric deposition (North Carolina Department of Environment and Natural Resources, 2012). Geologic sources of mercury were not measured.²

Methods

Study Area and Sampling Sites

This study was conducted in the Cape Fear River basin upstream from the confluence of the Haw and Deep Rivers and in the Neuse River basin upstream from the dam at Falls of the Neuse Reservoir (hereafter referred to as Falls Lake; [fig. 1](#)). The study area encompassed five counties (Orange, Durham, Chatham, Granville, and Wake) that included the towns and cities of Apex, Butner, Carrboro, Cary, Chapel Hill, Creedmoor, Durham, Garner, Hillsborough, Morrisville, Raleigh, and the Research Triangle Park. Twenty-nine sampling sites were distributed among nine water-supply reservoirs, selected tributaries to those reservoirs (stream sites), and near a run-of-river public-water supply (categorized as a stream site; [fig. 1](#), [table 1](#)). One reservoir site was located in each of seven small upland reservoirs, including Little River Reservoir, Lake Michie, Lake Butner, Lake Wheeler, Lake Benson, Cane Creek Reservoir, and University Lake. Two large, multipurpose reservoirs (Falls Lake and B. Everett Jordan Reservoir, hereafter identified as Jordan Lake; [fig. 1](#)) were each sampled at four sites.

Sample Collection and Processing

Since 1988, the USGS and TAWSMP have collected and analyzed water-quality samples from streams and reservoirs and collected continuous streamflow measurements from streams throughout the Triangle Area (Oblinger and Treece, 1996). Samples collected from sites for analysis were selected in 1988 as representative of water from major drinking-water supplies in the area and major tributaries to the drinking-water supplies (Oblinger, 2004). In 2007, supplementary mercury sampling was added to the scheduled sampling activities at USGS stations, and the number of samples collected at each site during the study period of July 2007 to June 2009 varied from 1 to 24. Fifteen reservoir sites were sampled either 8 or 12 times ([table 2](#)), at two depths (“surface”: 1 meter (m) below the surface; “bottom”: 1 m from the bottom of the reservoir). Fourteen stream sites were sampled 1 to 12 times each over the duration of the study ([table 2](#)) with one sample collected at each site from the surface of the water column. One bed sediment sample was collected at each site during the season of spring 2009 for all sites with one exception (USGS 02098198; [table 2](#)) at which

sample collection was not possible because of the high stream velocity. A total of 301 water samples were collected from reservoir sites (151 from the surface and 150 from the bottom, [table 2](#)), with the total number of samples collected at each site ranging from 16 (8 samples each from the surface and the bottom, [table 2](#)) to 24 (12 samples each from the surface and the bottom, [table 2](#)). The number of water samples collected at each stream site ranged from 1 (sites 24 and 29, [table 2](#)) to 12 (site 1, [table 2](#)), with a total of 77 water samples collected from stream sites.

Two mercury parameters in water were investigated: filtered mercury (FHg), which is an operational measure of mercury in a dissolved form or associated with particles and colloids smaller than 0.45 micron (μm); and TRHg, which includes all forms of mercury (inorganic and organic), including FHg and mercury associated with particles larger than 0.45 μm . Methylmercury, which is the predominant species of mercury that accumulates in fish tissue (Krabbenhoft and Rickert, 1995), was not specifically measured, but was included in FHg and TRHg measurements. Total mercury in bed sediment (THg) was measured for comparison with national background and baseline levels as well as for comparison with concentrations associated with adverse health effects to aquatic biota. Ancillary factors measured at the time of sample collection and investigated for potential relationships with mercury concentrations included concentrations of dissolved oxygen (DO), pH, and specific conductance (SC). Streamflow data (USGS, 2016b) were collected and analyzed in accordance with the Quality-Assurance Plan for Surface-Water Activities of the North Carolina Water Science Center (Rantz and others, 1982). Antecedent precipitation for a site was determined on the basis of total precipitation during three timeframes (1, 2, and 7 days) prior to sample collection from the most proximal USGS precipitation gage ([table 2](#)).

Water and bed sediment sample collection and processing followed established protocols specified in the USGS “National Field Manual for the Collection of Water-Quality Data” (National Field Manual; USGS, variously dated). Water samples were collected and processed according to established “Clean Hands/Dirty Hands” protocols (USGS, variously dated). In addition, equipment-cleaning procedures, sample containers, and shipping procedures used in the study were those specified in the USGS National Field Manual (USGS, variously dated). These protocols, which were developed to prevent contamination of samples containing low-level concentrations of trace elements, have been used by the USGS in the TAWSMP since 1994 (Oblinger, 2004). Bed sediment samples were collected with an Ekman dredge at the reservoir sites. Five samples were collected within a small radius in order to capture small-scale variability at each site. The overlying water was removed, and approximately the upper 5 centimeters (cm) of sediment were transferred in a Teflon beaker or Teflon spoon to a clean glass bowl. The bowl contents were then homogenized prior to removal of a sample aliquot for analysis. At the wadeable stream sites, samples of the upper 5 cm of sediment were collected in a short corer. Teflon sheets were used for in situ sectioning of the sediments at the top and bottom of the ring.

²The measurement of geologic mercury in the underlying bedrock was beyond the scope of this study. Horowitz and Stephens (2008) suggested that, relative to anthropogenic activities, the underlying rock type is unlikely to substantially influence sediment-associated mercury concentrations.

Table 1. Study sites and land-use characteristics.

[Drainage area (square miles) and land-use variables (shown as percent coverage within drainage area) were all calculated in StreamStats (U.S. Geological Survey [USGS], 2016a). NA, catchment unable to be defined]

Map no.	USGS station ID	Latitude	Longitude	USGS station name	Drainage area	Impervious surface	Urban	Wetland	Forest
Stream sites									
1	02085000	36.0711	−79.0956	ENO RIVER AT HILLSBOROUGH, NC	66	2.2	12.5	0.7	56.3
2	02085070	36.0722	−78.9078	ENO RIVER NEAR DURHAM, NC	142	2.8	17.8	0.5	59.2
3	0208521324	36.1417	−78.9192	LITTLE RIVER AT SR1461 NEAR ORANGE FACTORY, NC	78	0.6	5.9	1.0	59.7
5	02085500	36.1828	−78.8789	FLAT RIVER AT BAHAMA, NC	149	1.3	6.9	0.7	56.6
8	02086624	36.1279	−78.7983	KNAP OF REEDS CREEK NEAR BUTNER, NC	42	1.9	8.4	1.6	66.0
9	02086849	36.0593	−78.8325	ELLERBE CREEK NEAR GORMAN, NC	22	21.0	73.6	3.8	16.4
11	0208700780	35.9865	−78.7992	LITTLE LICK CR AB SR1814 NR OAK GROVE, NC	10	11.8	59.6	1.0	29.5
17	02096846	35.9872	−79.2061	CANE CREEK NEAR ORANGE GROVE, NC	8	0.3	4.4	0.1	72.8
20	02097314	35.8850	−78.9653	NEW HOPE CREEK NEAR BLANDS, NC	76	9.1	40.7	5.3	45.1
21	0209741955	35.8722	−78.9131	NORTHEAST CREEK AT SR1100 NR GENLEE, NC	21	14.9	56.9	3.7	33.7
22	02097464	35.9236	−79.1150	MORGAN CREEK NEAR WHITE CROSS, NC	8	0.4	5.3	0.2	69.9
24	02097521	35.8611	−79.0100	MORGAN CREEK NR FARRINGTON, NC	46	4.3	19.6	1.6	63.3
26	0209782609	35.7603	−78.9203	WHITE OAK CR AT MOUTH NEAR GREEN LEVEL, NC	12	8.1	33.2	5.8	43.3
29	02098198	35.6494	−79.0656	HAW R BELOW B. EVERETT JORDAN DAM NR MONCURE, NC	1,690	5.1	20.2	1.6	47.8
Reservoir sites									
4	0208524845	36.1150	−78.8692	LITTLE RIVER RESERVOIR AT DAM NEAR BAHAMA, NC	97	0.7	6.9	0.9	58.5
6	02086490	36.1508	−78.8300	LAKE MICHIE AT DAM NR BAHAMA, NC	168	1.2	6.7	0.9	57.1
7	02086569	36.1669	−78.7722	LAKE BUTNER AT DAM NEAR BUTNER, NC	29	0.2	2.7	1.0	69.8
10	02086920	36.0697	−78.7792	FALLS LAKE AT INTERSTATE 85 NEAR REDWOOD	513	2.7	13.9	1.9	56.1
12	0208703650	36.0153	−78.6911	FALLS LAKE AT STATE HWY 50 NR SANDY PLAIN, NC	620	2.7	14.2	2.1	55.5
13	0208708905	35.9786	−78.6328	FALLS LAKE AT STATE HWY 98 NR BAYLEAF, NC	705	2.5	13.3	2.2	56.6
14	0208718195	35.9414	−78.5836	FALLS LAKE ABOVE DAM AT FALLS, NC	772	2.5	14.2	2.1	56.9
15	02087588	35.6944	−78.6939	LAKE WHEELER ON SWIFT CREEK NEAR RALEIGH, NC	36	10.8	55.7	1.5	31.4
16	02087701	35.6625	−78.6114	LAKE BENSON AT DAM NEAR GARNER, NC	66	9.5	47.9	2.6	33.9
18	0209684980	35.9500	−79.2411	CANE CR RESERVOIR AT DAM NR WHITE CROSS, NC	32	0.5	5.4	0.2	61.1
19	0209699999	35.7053	−79.0844	B. EVERETT JORDAN LAKE, HAW R ARM NR HANKSCHAPEL	NA	NA	NA	NA	NA
23	0209749990	35.8969	−79.0922	UNIVERSITY LAKE AT INTAKES NR CHAPEL HILL, NC	30	0.8	7.4	0.2	72.4
25	0209768310	35.7989	−79.0058	JORDAN LAKE AT BUOY 12 AT FARRINGTON, NC	231	49.0	33.5	4.8	3.1
27	0209799150	35.7417	−79.0192	B.E. JORDAN LAKE AB US HWY 64 AT WILSONVILLE, NC	285	6.7	29.7	4.6	50.8
28	0209801100	35.7275	−79.0428	B.E. JORDAN LK AT BELLS LANDNG NR GRIFFINS XRDS, NC	NA	NA	NA	NA	NA

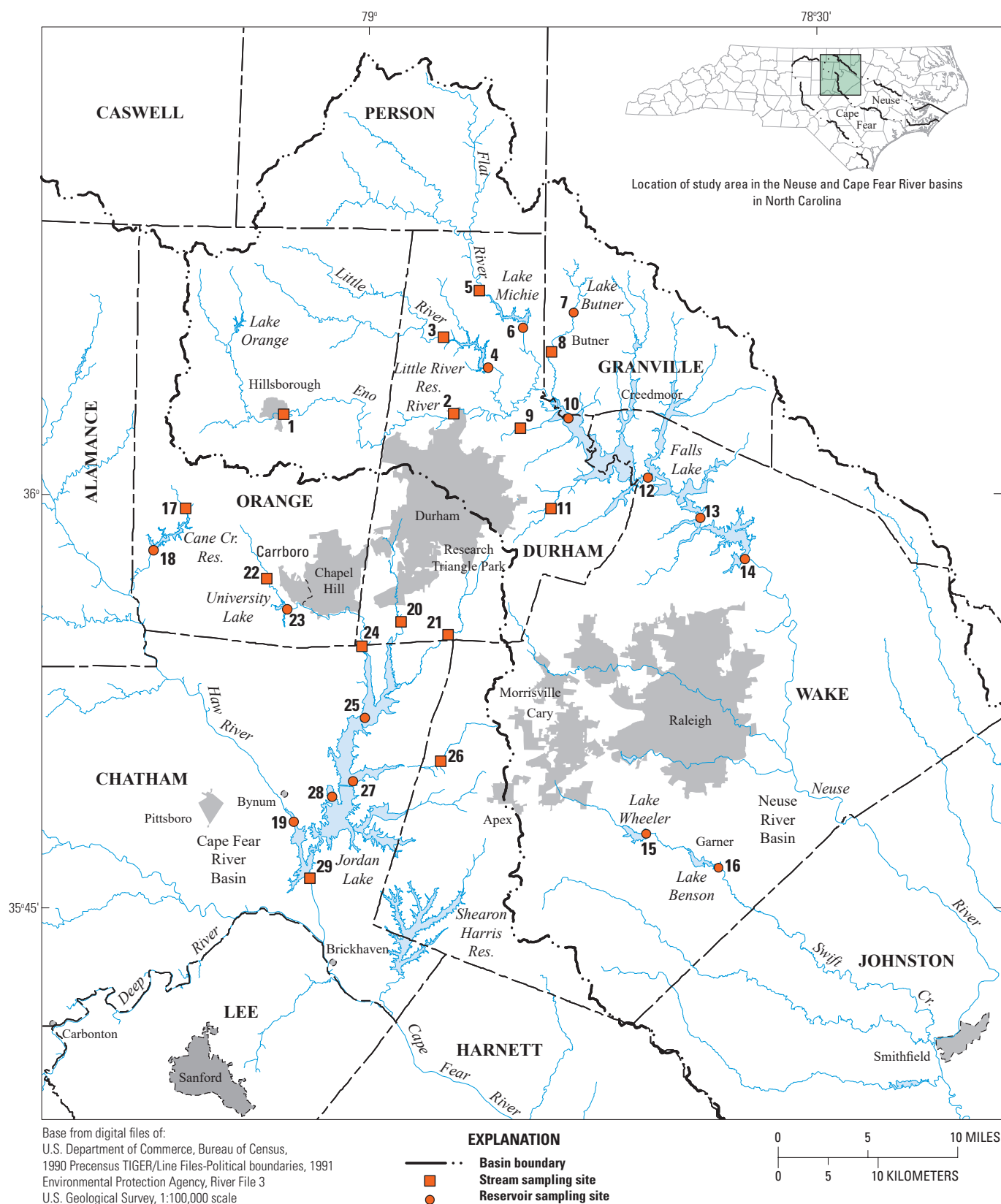


Figure 1. Map showing location of Triangle Area Water Supply Monitoring Project data-collection sites.

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Table 2. Precipitation gage and number of water samples collected by season and the number of bed sediment samples collected at each site.

[Site number refers to numbers on [figure 1](#). Spring samples were collected in March–May; summer, June–August; fall, September–November; and winter, December–February. Abbreviations: USGS, U.S. Geological Survey; N, number of samples collected; NA, not applicable; surface, 1 meter below the surface of the reservoir water column; bottom, 1 meter from the bottom of the reservoir]

Site number	USGS station ID (USGS precipitation gage ID)	Sampling location	Total N	Spring N	Summer N	Fall N	Winter N	Bed sediment N
Stream sites								
1	02085000 (360334078584145)	Surface of water column	12	2	4	2	4	1
2	02085070 (360334078584145)	Surface of water column	3	2	0	0	1	1
3	0208521324 (360334078584145)	Surface of water column	7	4	2	1	0	1
5	02085500 (360334078584145)	Surface of water column	2	1	0	1	0	1
8	02086624 (360419078543145)	Surface of water column	3	2	0	0	1	1
9	02086849 (360143078540945)	Surface of water column	4	1	0	2	1	1
11	0208700780 (355856078492945)	Surface of water column	3	1	0	1	1	1
17	02096846 (02096500)	Surface of water column	11	2	4	2	3	1
20	02097314 (355511078570745)	Surface of water column	4	3	0	0	1	1
21	0209741955 (355511078570745)	Surface of water column	5	4	0	0	1	1
22	02097464 (355511078570745)	Surface of water column	10	2	4	1	3	1
24	02097521 (355511078570745)	Surface of water column	1	1	0	0	0	1
26	0209782609 (355020078465645)	Surface of water column	11	2	4	2	3	1
29	02098198 (02098197)	Surface of water column	1	1	0	0	0	0
Reservoir sites								
4	0208524845 (360334078584145)	Bottom	8	2	4	2	0	1
		Surface	8	2	4	2	0	NA
6	02086490 (360334078584145)	Bottom	8	2	4	2	0	1
		Surface	8	2	4	2	0	NA
7	02086569 (0208706575)	Bottom	8	2	4	2	0	1
		Surface	8	2	4	2	0	NA
10	02086920 (355856078492945)	Bottom	10	2	4	2	2	1
		Surface	11	2	4	2	3	NA
12	0208703650 (0208706575)	Bottom	12	2	4	2	4	1
		Surface	12	2	4	2	4	NA
13	0208708905 (02087182)	Bottom	12	2	4	2	4	1
		Surface	12	2	4	2	4	NA

Table 2. Precipitation gage and number of water samples collected by season and the number of bed sediment samples collected at each site.—Continued

[Site number refers to numbers on [figure 1](#). Spring samples were collected in March–May; summer, June–August; fall, September–November; and winter, December–February. Abbreviations: USGS, U.S. Geological Survey; N, number of samples collected; NA, not applicable; surface, 1 meter below the surface of the reservoir water column; bottom, 1 meter from the bottom of the reservoir]

Site number	USGS station ID (USGS precipitation gage ID)	Sampling location	Total N	Spring N	Summer N	Fall N	Winter N	Bed sediment N
Reservoir sites—Continued								
14	0208718195	Bottom	12	2	4	2	4	1
	(02087182)	Surface	12	2	4	2	4	NA
15	02087588	Bottom	8	2	4	2	0	1
	(02087580)	Surface	8	2	4	2	0	NA
16	02087701	Bottom	8	2	4	2	0	1
	(354528078372645)	Surface	8	2	4	2	0	NA
18	0209684980	Bottom	8	2	4	2	0	1
	(02096500)	Surface	8	2	4	2	0	NA
19	0209699999	Bottom	12	2	4	2	4	1
	(02098197)	Surface	12	2	4	2	4	NA
23	0209749990	Bottom	8	2	4	2	0	1
	(355511078570745)	Surface	8	2	4	2	0	NA
25	0209768310	Bottom	12	2	4	2	4	1
	(355511078570745)	Surface	12	2	4	2	4	NA
27	0209799150	Bottom	12	2	4	2	4	1
	(02098197)	Surface	12	2	4	2	4	NA
28	0209801100	Bottom	12	2	4	2	4	1
	(02098197)	Surface	12	2	4	2	4	NA

Laboratory Analyses

In addition to FHg and TRHg, ancillary constituents for water samples measured by laboratory analysis included organic carbon (dissolved, DOC; and total in water, $\text{TOC}_{\text{water}}$), sulfate (SO_4^{2-}), chlorophyll *a* (Chloro*a*), and pheophytin *a* (Pheo*a*); these constituents were analyzed at the USGS National Water Quality Laboratory (NWQL) in Denver, Colorado (table 3). Methylmercury was not measured separately from FHg and TRHg but, if present, would be included in the FHg

and TRHg measurements. Suspended-sediment concentrations (SSCs) in stream samples were analyzed at the USGS Sediment Laboratory in Louisville, Kentucky (table 3). The clay and silt fraction of “fines” (particle diameters less than 63 μm) in bed sediment samples were analyzed for THg and total organic carbon in bed sediment (TOC_{sed}) (table 3) by total digestion at the USGS Georgia Sediment Partitioning Laboratory in Atlanta, Georgia.

Table 3. Laboratory-measured constituents in water and bed sediment samples. Minimum reporting level is defined as the smallest measured concentration of a substance that can be reliably measured.

[CV–AFS, cold vapor-atomic fluorescence spectrometry; CS analyzer, Carbon/Sulfur analyzer; CV–AAS, cold vapor-atomic absorption spectrometry; CAS, Chemical Abstracts Service; NA, not applicable; $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter; $\text{mg}\cdot\text{L}^{-1}$, milligram per liter; wt. %, weight percent; $\text{mg}\cdot\text{kg}^{-1}$, milligram per kilogram; mrl, minimum reporting level; lrl, laboratory reporting level; irl, interim reporting level]

Constituent	Method and instrumentation	CAS number	Reporting level and type	Reference
Water samples				
Mercury, filtered (FHg)	CV–AFS	7439-97-6	0.01 $\mu\text{g}\cdot\text{L}^{-1}$ lrl	(Garbarino and Damrau, 2001)
Mercury, total recoverable (TRHg)	CV–AFS	7439-97-6	0.01 $\mu\text{g}\cdot\text{L}^{-1}$ lrl	(Garbarino and Damrau, 2001)
Sulfate (SO_4^{2-})	Ion chromatography	14808-79-8	0.18 $\text{mg}\cdot\text{L}^{-1}$ lrl	(Fishman and Friedman, 1989)
Chlorophyll <i>a</i> ¹ (Chloro <i>a</i>)	Fluorometry	479-61-8	0.1 $\mu\text{g}\cdot\text{L}^{-1}$ mrl	(Arar and Collins, 1997)
Pheophytin <i>a</i> (Pheo <i>a</i>)	Fluorometry	603-17-8	0.1 $\mu\text{g}\cdot\text{L}^{-1}$ mrl	(Arar and Collins, 1997)
Organic carbon, filterable (dissolved, DOC)	Infrared spectrometry	NA	0.4 $\text{mg}\cdot\text{L}^{-1}$ lrl	(Brenton and Arnett, 1993)
Organic carbon, unfilterable (total, $\text{TOC}_{\text{water}}$)	Infrared spectrometry	NA	0.6 $\text{mg}\cdot\text{L}^{-1}$ irl	(Clesceri and others, 1998)
Suspended-sediment concentration (SSC) ²	Gravimetry	NA	1 $\text{mg}\cdot\text{L}^{-1}$ mrl	(Fishman and Friedman, 1989)
Bed sediment samples				
Organic carbon, total (TOC_{sed})	CS analyzer/infrared detection	NA	0.1 wt. % lrl	(Horowitz and others, 2001 and references therein)
Mercury, total digestion (THg)	CV–AAS	7439-97-6	0.01 $\text{mg}\cdot\text{kg}^{-1}$ lrl	(Horowitz and others, 2001 and references therein)
Percentage of bed sediment smaller than 0.0625 millimeters	Wet sieving, gravimetry	NA	wt. %	(Horowitz and others, 2001)

¹Reservoir samples only.

²Stream samples only.

Statistical Analysis

Non-detections of FHg or TRHg in environmental water samples were substituted with randomly generated concentration values between 0 and the MDL ($0.005 \mu\text{g}\cdot\text{L}^{-1}$; appendix 1) (Antweiler and Taylor, 2008). For comparison with benchmark concentrations for bed sediment, measured THg concentrations were calculated for the bulk sample on the basis of the percentage of the grains in the bulk sample whose diameters were smaller than the sieve diameter of 0.0625 millimeters (herein referred to as bulk-calculated THg concentrations; data available from USGS, 2016b). Non-parametric analyses were used to investigate associations between potential driver variables and FHg and TRHg concentrations for water samples and bulk-calculated THg concentrations in bed sediment samples. Stream, reservoir-surface, and reservoir-bottom samples were investigated independently in all statistical analyses, except for correlation analyses between reservoir-surface and reservoir-bottom mercury concentrations.

Mercury concentrations may vary by season owing to factors such as seasonal differences in reservoir stratification and atmospheric deposition. To investigate potential seasonal stratification at the reservoir sites, correlations between reservoir-surface and reservoir-bottom mercury concentrations were investigated with Spearman rank correlations (statistical significance when $p < 0.05$) for both FHg and TRHg. Owing to the myriad of factors that can affect mercury fluxes and rates of mercury reactions, significant correlations between surface and bottom mercury concentrations would be consistent with mixing between surface and bottom samples, whereas a lack of correlation could indicate stratification. Seasonal differences in FHg and TRHg mercury concentrations were tested by using non-parametric Wilcoxon Tests with seasonal median values for each site. If a seasonal effect was detected (statistical significance when Chi-square $p < 0.05$) in FHg or TRHg concentrations, non-parametric Spearman rank correlations (described below) between potential driver variables and mercury concentrations in water were determined by using site-specific seasonal median values for mercury concentrations, antecedent precipitation, and ancillary constituents (all sites: DO, pH, SC, DOC, $\text{TOC}_{\text{water}}$, SO_4^{2-} , Chloro a , Pheo a ; stream only: SSC, streamflow). If no seasonal effect was detected in FHg or TRHg concentrations, Spearman rank correlations between potential drivers and mercury concentrations were calculated by using the overall median values by site. For bed sediment bulk-calculated THg analyses, Spearman rank correlations were determined with ancillary constituents measured in water samples collected concurrently with the bed sediment samples. (All bed sediment samples were collected during spring 2009.)

Drainage-area and land-use variables investigated for associations with mercury concentrations were calculated in StreamStats (USGS, 2016a) and based on 2006 National Landcover Data (Fry and others, 2011); these variables included urban area, forested area, impervious surface area, and wetland area. For analysis, land-use variables were

calculated as a percentage of the drainage area (table 1). Directional associations between median mercury concentrations and ancillary constituents, antecedent precipitation, and land-use variables were investigated only if at least 50 percent of the median values for analysis were greater than the MDL. The interpretation of the value of the Spearman parameter ρ was similar to that of Pearson's r , whose values are between -1 and 1 . The closer the value is to -1 or 1 , the stronger the correlation; a value of $\rho = 0$ would indicate no correlation. Negative values indicate a negative directional association, and positive values for ρ indicate a positive directional association.

Increasing the number of statistical comparisons on a dataset also increases the probability of a statistically significant result caused by chance (for example, rejecting the null hypothesis when it is true; referred to as a type I error). One way to reduce a type I error is to use a smaller value for α , the maximum p -value for which a null hypothesis is rejected (for example, using $p < 0.01$ instead of $p < 0.05$ to indicate statistical significance). Error in the data caused by factors such as low precision in measured concentrations or high variability within the sample matrix may reduce the sensitivity of an analysis to detect a relationship if one is present (a type II statistical error—accepting the null hypothesis when it should be rejected). In this study, errors in water-sample mercury-concentration data were likely, particularly for mercury concentrations below the method reporting level (MRL) and samples potentially affected by high variability in measured TRHg concentrations (described below). Statistical significance was therefore determined by $p < 0.01$, and potentially relevant associations were discussed if $p < 0.05$ for Spearman rank analyses. All statistics were analyzed by using the program JMP (v. 14.2.0, SAS Institute Inc.).

Quality Assurance and Quality Control

The contamination, bias, and precision of measured constituent concentrations in environmental samples were assessed by using equipment blanks, vehicle (ambient–open bottle) blanks, field blanks, reference standards, split replicates, and field replicates (table 4). Additionally, method performance at the NWQL was assessed during the study period from data collected by the USGS Quality Systems Branch (described in Mueller and others, 2015; data available from USGS, undated a). The various blanks reveal contamination from the sampling environment, sampling equipment, or processing and analysis. Reference standards have a known concentration and provide an evaluation of the analytical method. Field replicates were used to quantify the overall variability caused by sampling, processing, and analysis in measured concentrations.

Low accuracy or precision of constituent measurements in environmental samples may reduce the sensitivity of statistical analyses to detect associations between mercury concentrations and ancillary constituents and potentially determine

whether or not a sample exceeds the NCWQC for mercury. The variability of sampling and analysis was assessed by using paired or triplicate sets of water samples (Fitzgerald, 2020). The precision levels of analyses of water collected in field replicates was evaluated in terms of relative percent difference (RPD, the absolute difference between replicate-pair concentrations divided by the average concentration multiplied by 100); relative standard deviation (RSD, the standard deviation of triplicate concentrations divided by the average concentration multiplied by 100); and absolute concentration differences among replicates. Replicate sets for which concentrations in all replicates were below the MDL (table 4) were assigned an RPD or RSD of 0 percent and an absolute difference of 0 concentration units. Mercury concentrations in replicate samples with measured concentrations below the method reporting level (MRL) (mercury was detected, but the measured concentration was estimated at a value below the MRL)

were used in the same way as concentrations above the MRL in calculations. RPDs and RSDs less than 25 percent were considered to have acceptable reproducibility. The potential bias of THg and TOC_{sed} analyses in bed sediment samples was assessed through concurrent analysis of several reference standards with known concentrations. The precision levels of these analyses were determined from field and laboratory analytical replicates.

Results

Water-quality data presented in this report, including sample mercury concentrations, are available through the U.S. Geological Survey National Water Information System (USGS, 2016b) and are summarized for the entire study period in table 5.

Table 4. Numbers (and relative percentages) of quality-assurance and quality-control samples for each measured constituent relative to the number of environmental samples screened.

[Environmental samples, water and bed sediment samples analyzed for the study herein; blank, sample that is prepared with water intended to be free of measurable concentrations of the analytes of interest; equipment blank, blank used to demonstrate that the sample-collection and sample-processing equipment is not introducing contamination; vehicle blank, used to demonstrate that sample-handling and transport have not introduced contamination; field blank, used to demonstrate that contamination was not introduced during collection and processing; replicate, two or more samples collected, prepared, and analyzed such that the samples are considered to be essentially identical in composition; field replicate set, concurrently collected samples; split replicate, a single large sample split into two samples (U.S. Geological Survey, variously dated); reference materials, materials with a known composition or concentration that are used for testing analytical accuracy, bias, and precision. Abbreviations: %, percent; FHg, filtered mercury; TRHg, total recoverable mercury; SO₄²⁻, sulfate; DOC, dissolved organic carbon; TOC_{water}, total organic carbon in water; Chloro_a, chlorophyll *a*; Pheo_a, pheophytin *a*; SSC, suspended-sediment concentration; THg, total mercury in bed sediment; TOC_{sed}, total organic carbon in bed sediment]

Sample type	Constituent	Number of environmental samples	Equipment blanks	Vehicle blanks	Field blanks	Field replicate sets	Split replicates	Reference materials
Water	Total	378	2 (<1%)	11 (3%)	11 (3%)	39 (10%)	0 (0%)	0 (0%)
	FHg	378	1 (<1%)	10 (3%)	10 (3%)	13 (3%)	0 (0%)	0 (0%)
	TRHg	378	2 (<1%)	11 (3%)	11 (3%)	13 (3%)	0 (0%)	0 (0%)
	SO ₄ ²⁻	378	1 (<1%)	8 (2%)	9 (2%)	13 (3%)	0 (0%)	0 (0%)
	DOC	376	0 (0%)	11 (3%)	10 (3%)	13 (3%)	0 (0%)	0 (0%)
	TOC _{water}	219	0 (0%)	0 (0%)	0 (0%)	13 (6%)	0 (0%)	0 (0%)
	Chloro _a	148	0 (0%)	0 (0%)	0 (0%)	28 (19%)	0 (0%)	0 (0%)
	Pheo _a	148	0 (0%)	0 (0%)	0 (0%)	28 (19%)	0 (0%)	0 (0%)
	SSC	66	0 (0%)	0 (0%)	0 (0%)	3 (5%)	0 (0%)	0 (0%)
Bed sediment	Total	28	0 (0%)	0 (0%)	0 (0%)	2 (7%)	3 (11%)	7 (25%)
	THg	28	0 (0%)	0 (0%)	0 (0%)	2 (7%)	3 (11%)	7 (25%)
	TOC _{sed}	28	0 (0%)	0 (0%)	0 (0%)	2 (7%)	3 (11%)	7 (25%)

Table 5. Summary statistics for concentrations of mercury and ancillary constituents in water samples.

[Randomly generated concentration values between 0 and the method detection level (MDL, 0.005 microgram per liter [$\mu\text{g}\cdot\text{L}^{-1}$]) were substituted for non-detections in computations. Abbreviations: N, number of samples collected; SD, standard deviation; ND, not detected; NC, data for constituent was not collected; NA, not applicable; <, less than]

Constituent	Unit	Statistic	Reservoir		Stream
			Surface	Bottom	
Water					
Mercury, filtered (FHg)	Microgram per liter (μg·L ⁻¹)	N	151	150	77
		Mean	0.010	0.005	0.003
		Median	0.004	0.003	0.003
		SD	0.015	0.008	0.002
		Minimum	ND	ND	ND
		Maximum	0.099	0.071	0.011
Mercury, total recoverable (TRHg)	μg·L ⁻¹	N	151	150	77
		Mean	0.032	0.009	0.006
		Median	0.012	0.008	0.004
		SD	0.067	0.010	0.006
		Minimum	ND	ND	ND
		Maximum	0.497	0.103	0.032
Dissolved oxygen (DO)	Milligram per liter (mg·L ⁻¹)	N	141	139	72
		Mean	8.3	4.1	8.1
		Median	8.5	2.6	8.2
		SD	2.4	4.1	2.2
		Minimum	0.3	0.1	0.1
		Maximum	13.3	12.6	12.9
pH	Standard units	N	141	140	73
		Mean	7.4	6.9	6.9
		Median	7.3	6.9	6.9
		SD	0.6	0.4	0.3
		Minimum	6.4	5.7	6.2
		Maximum	9.1	8.1	7.8
Specific conductance (SC)	Microsiemens per centimeter at 25 degrees Celsius (μS·cm ⁻¹ at 25 °C)	N	151	150	77
		Mean	125	147	129
		Median	105	138	102
		SD	64	74	90
		Minimum	43	48	31
		Maximum	389	506	500
Sulfate (SO ₄ ²⁻)	mg·L ⁻¹	N	151	150	77
		Mean	10.9	9.9	9.6
		Median	7.9	7.6	6.1
		SD	7.0	8.1	11.3
		Minimum	2.7	0.1	0.8
		Maximum	34.3	50.1	67.4

Table 5. Summary statistics for concentrations of mercury and ancillary constituents in water samples.—Continued

[Randomly generated concentration values between 0 and the method detection level (MDL, 0.005 microgram per liter [$\mu\text{g}\cdot\text{L}^{-1}$]) were substituted for non-detections in computations. Abbreviations: N, number of samples collected; SD, standard deviation; ND, not detected; NC, data for constituent was not collected; NA, not applicable; <, less than]

Constituent	Unit	Statistic	Reservoir		Stream
			Surface	Bottom	
Water—Continued					
Chlorophyll <i>a</i> (Chloro <i>a</i>)	μg·L ⁻¹	N	147	NC	NC
		Mean	25.1	NC	NC
		Median	22.7	NC	NC
		SD	17.4	NC	NC
		Minimum	1.9	NC	NC
		Maximum	96.0	NC	NC
Pheophytin <i>a</i> (Pheo <i>a</i>)	μg·L ⁻¹	N	147	NC	NC
		Mean	11.4	NC	NC
		Median	9.3	NC	NC
		SD	8.4	NC	NC
		Minimum	1	NC	NC
		Maximum	49.4	NC	NC
Organic carbon, filterable (dissolved, DOC)	mg·L ⁻¹	N	150	149	77
		Mean	7.16	7.77	7.63
		Median	7.00	7.57	6.94
		SD	1.20	1.73	3.20
		Minimum	4.84	4.71	2.49
		Maximum	11.00	16.80	18.40
Organic carbon, filterable (total, TOC _{water})	mg·L ⁻¹	N	151	NC	68
		Mean	9.6	NC	9.5
		Median	9.5	NC	8.3
		SD	1.6	NC	4.7
		Minimum	6.2	NC	3.1
		Maximum	15.3	NC	26.8
Suspended-sediment concentration (SSC)	mg·L ⁻¹	N	NC	NC	66
		Mean	NC	NC	52
		Median	NC	NC	12
		SD	NC	NC	93
		Minimum	NC	NC	1
		Maximum	NC	NC	419
Bed sediment					
Mercury, total digestion (THg) [bulk-calculated]	Milligram per kilogram (mg·kg ⁻¹)	N	NA	15	13
		Mean	NA	0.07 (0.06)	0.11 (0.04)
		Median	NA	0.07 (0.07)	0.05 (<0.01)
		SD	NA	0.03 (0.03)	0.21 (0.11)
		Minimum	NA	0.02 (0.02)	0.02 (<0.01)
		Maximum	NA	0.13 (0.13)	0.82 (0.39)

Table 5. Summary statistics for concentrations of mercury and ancillary constituents in water samples.—Continued

[Randomly generated concentration values between 0 and the method detection level (MDL, 0.005 microgram per liter [$\mu\text{g}\cdot\text{L}^{-1}$]) were substituted for non-detections in computations. Abbreviations: N, number of samples collected; SD, standard deviation; ND, not detected; NC, data for constituent was not collected; NA, not applicable; <, less than]

Constituent	Unit	Statistic	Reservoir		Stream
			Surface	Bottom	
Bed sediment—Continued					
Organic carbon, total (TOC _{sed})	Percent (%)	N	NA	15	13
		Mean	NA	2.79	1.47
		Median	NA	2.70	1.30
		SD	NA	0.89	0.63
		Minimum	NA	1.60	0.20
		Maximum	NA	4.10	2.20

Quality-Assurance and Quality-Control Results

Contamination in Blanks

FHg was detected in one field blank and one vehicle blank (water intended to be free of analytes of interest that is used to demonstrate that sample-handling and transport have not introduced contamination) at concentrations below the MRL of $0.010 \mu\text{g}\cdot\text{L}^{-1}$ (Fitzgerald, 2020). FHg was either not detected or detected at concentrations below the MRL in water samples collected on the date associated with the vehicle blank. The FHg concentration in the one water sample that was collected from stream site 22 in association with the field blank from February 8, 2008 was above the MRL but did not exceed the NCWQC for mercury. Because FHg was not detected in any of the other nine samples from site 22, any contamination present in the sample collected on February 8, 2008, would not affect the overall median value for stream site 22.

TRHg was also detected in a vehicle blank at a level below the MRL. TRHg concentrations in two of the six environmental water samples collected on the date when the vehicle blank was contaminated were greater than the NCWQC and thus raised the possibility that contamination may have caused these environmental samples to be mis-categorized as exceedances. Intermittent high variability in sample replicates for TRHg concentrations (described below) also indicated that error in TRHg concentration estimates for environmental samples was likely not uncommon over the course of the study. For a conservative estimate of the number of samples that tested above the NCWQC for mercury, the frequencies of FHg concentrations (which did not appear to have precision issues similar to those for TRHg) exceeding the NCWQC were used for comparison with the frequencies of TRHg exceedances. Frequent DOC contamination was detected in 9 of 11 field blanks (Fitzgerald, 2020). Levels of DOC detected in these blanks ($1.1 \text{ mg}\cdot\text{L}^{-1}$ maximum) were all

lower than the minimum DOC concentration in environmental samples ($2.5 \text{ mg}\cdot\text{L}^{-1}$ and $4.7 \text{ mg}\cdot\text{L}^{-1}$ in stream and reservoir samples, respectively).

Bias

The USGS Quality Systems Branch Inorganic Blind Sample Project noted a moderate negative bias in FHg concentrations of about 12 to 13 percent and about 6.2 to 12 percent for TRHg from July 2007 through approximately February 2008 (USGS, undated b). For environmental samples, the mean and median of FHg concentrations above the MRL were lower between July 2007 and February 2008 than between March 2008 and June 2009; however, the proportion of FHg values above the MRL was higher during the earlier period. Under the assumption that the negative bias affected samples collected from all sites during this period, the potential consequences included lower seasonal median concentrations for summer, fall, and winter surface samples from reservoir sites (for which there were significant differences in seasonal median values, described below), and lower overall median concentrations for stream sites and bottom samples from reservoir sites (for which there were no significant differences in seasonal median values, described below). In surface samples from reservoir sites, differences in mercury detections, NCWQC exceedances, or concentrations between spring and other seasons would likely be more pronounced than indicated by the data. Additionally, the frequency of detections and NCWQC exceedances may have been greater for samples collected during this period than indicated by the data with negative bias. If the negative bias resulted in more non-detections for FHg or TRHg, less precision in the concentration estimates for the data could reduce the probability of detecting associations between mercury and predictor variables. No systematic differences were observed between known and measured concentrations of THg or TOC_{sed} in bed sediment (appendix 2).

Precision

All RPDs for FHg, SO_4^{2-} , $\text{TOC}_{\text{water}}$, and DOC were below 25 percent (appendix 3). *Chloro*a had two replicate sets with an RSD or RPD greater than 25 percent (32 and 34 percent) (Fitzgerald, 2020). One of these sets had an absolute difference of 1.0, which is low relative to all of the *Chloro*a concentrations measured in the environmental samples (median: $22.3 \mu\text{g}\cdot\text{L}^{-1}$, max: $96 \mu\text{g}\cdot\text{L}^{-1}$) (Fitzgerald, 2020). The other had an absolute difference of $12.1 \mu\text{g}\cdot\text{L}^{-1}$ (Fitzgerald, 2020). Errors in the measurements of physical or chemical water-quality parameters in environmental samples could obscure any existing correlative relationship with mercury concentrations. One replicate SSC set had an RPD of 40 percent; however, this set's low absolute difference ($2 \text{ mg}\cdot\text{L}^{-1}$) relative to concentrations measured in other environmental samples (median: $12 \text{ mg}\cdot\text{L}^{-1}$, max: $419 \text{ mg}\cdot\text{L}^{-1}$) indicated that these differences were unlikely to affect results of statistical analyses (Fitzgerald, 2020).

Nearly one-third of TRHg replicate sets (4 of 13) had RPDs greater than 25 percent (Fitzgerald, 2020); 3 of these had absolute differences greater than the NCWQC for mercury. All replicate sets with RPDs greater than 25 percent were from reservoir sites; however, only 3 of the 13 replicate sets were collected from stream sites. Of the replicate sets with TRHg RPDs greater than 25 percent, absolute concentration differences ranged from $0.008 \mu\text{g}\cdot\text{L}^{-1}$ to $0.032 \mu\text{g}\cdot\text{L}^{-1}$, and RPDs ranged from 67 to 145 percent (appendix 3). More than 90 percent of the environmental water samples had TRHg concentrations below the maximum absolute difference of $0.032 \mu\text{g}\cdot\text{L}^{-1}$ in replicate samples. Although they were not measured in reservoir-water samples in this study, turbidity and suspended sediment can cause high variability in TRHg measurements; for this reason, occasional high variability in TRHg concentrations in this study was not unexpected. The frequency of absolute differences greater than the NCWQC of $0.012 \mu\text{g}\cdot\text{L}^{-1}$ among the replicate sets, however, indicates that caution is needed when TRHg results are interpreted. Furthermore, of the environmental water samples in which FHg was above the MRL, five exhibited higher concentrations of FHg than TRHg (USGS, 2016b), indicating a known low frequency of error in estimates of mercury concentrations in environmental water samples. One consequence of error in measured mercury concentrations is the potential for obscuring associations with ancillary constituents, antecedent precipitation, or land-use variables. Therefore, the lack of a statistically significant association between FHg or TRHg and a given predictor variables does not preclude an ecologically relevant association.

With regard to TRHg exceedances of the NCWQC in environmental samples, differences in TRHg concentration measurements greater than the NCWQC itself for nearly one in three replicate sets indicates that the frequency of exceedances reported herein should be regarded as a general (not absolute) estimate. Measurement precision in replicate samples was better for FHg, which is a component of TRHg. The

frequency of FHg concentrations greater than $0.012 \mu\text{g}\cdot\text{L}^{-1}$ was used as a conservative estimate of TRHg exceedances of the NCWQC for mercury, because samples with FHg concentrations less than $0.012 \mu\text{g}\cdot\text{L}^{-1}$ may have had TRHg concentrations greater than or equal to $0.012 \mu\text{g}\cdot\text{L}^{-1}$.

One of two field-replicate sets showed a high degree of variation in bed sediment THg concentration ($0.03 \text{ mg}\cdot\text{kg}^{-1}$ and $0.10 \text{ mg}\cdot\text{kg}^{-1}$) (Fitzgerald, 2020). This variation could be caused by analytical variation, sample heterogeneity, or small-scale variability in bed sediment THg concentrations at that site. Field and laboratory replicates for TOC_{sed} were in acceptable agreement (Fitzgerald, 2020).

Water-Sample Mercury Detections, Concentrations, and Exceedances

FHg was detected in 109 reservoir-water samples (36 percent; table 6, fig. 2), and TRHg was detected in 183 reservoir-water samples (61 percent; table 6, fig. 2). Both FHg and TRHg were detected in at least two samples from each reservoir site (fig. 2). Mercury was detected more often in samples collected from the surface of the water column in reservoirs versus the bottom, with FHg and TRHg detections in at least one surface sample from all sites, and at least one bottom sample from all but two reservoir sites (table 6, fig. 2). For stream sites, FHg was detected in 9 and TRHg in 24 of 77 water samples (12 percent and 31 percent, respectively; table 6, fig. 3), with at least 1 FHg detection at 7 sites and at least 1 TRHg detection at 12 of 14 stream sites (table 6, fig. 3). The maximum mercury concentrations detected in water samples were $0.099 \mu\text{g}\cdot\text{L}^{-1}$ FHg and $0.497 \mu\text{g}\cdot\text{L}^{-1}$ TRHg, both in the same reservoir-surface sample from site 27 (table 5, figs. 4, 5). FHg concentrations in streams were comparatively low, with only one sample at a concentration greater than the MRL ($0.011 \mu\text{g}\cdot\text{L}^{-1}$ at site 22; table 5, fig. 6). The maximum TRHg concentration detected in stream-water samples was $0.032 \mu\text{g}\cdot\text{L}^{-1}$ from site 9 (table 5, fig. 6).

FHg and TRHg exceeded the NCWQC for mercury more often in reservoir-surface samples (25 percent and 50 percent, respectively; table 6, fig. 2) compared to reservoir-bottom samples (7 percent and 26 percent, respectively; table 6, fig. 2). The two reservoir sites with the highest frequency of FHg concentrations in surface samples exceeding the NCWQC for mercury were sites in Falls Lake and Jordan Lake (reservoir sites 12 and 28, respectively; fig. 7). No stream samples had concentrations of FHg that exceeded the NCWQC for mercury (table 6, fig. 7). In contrast, TRHg concentrations exceeded the NCWQC for mercury in eight stream-water samples (10 percent; table 6, fig. 3); the excess was spread among several sites without a clear spatial pattern (fig. 8). TRHg concentrations exceeded the NCWQC in 114 reservoir-water samples (38 percent; table 6, fig. 2), with the highest frequencies of exceedances in reservoir-surface samples generally concentrated around Jordan Lake and Falls Lake (fig. 8).

Table 6. Number of water samples and sites with mercury-concentration detections and exceedances.

[Number of reservoir and stream-water samples and number of study sites with filtered mercury (FHg) and total recoverable mercury (TRHg) detections (method detection level = 0.005 microgram per liter [$\mu\text{g}\cdot\text{L}^{-1}$]) and concentrations that exceeded the North Carolina water-quality criterion for mercury of 0.012 $\mu\text{g}\cdot\text{L}^{-1}$. The total numbers of samples or sites for the category are noted in parentheses]

Mercury constituent	Unit	Number of detections			Number of exceedances		
		Reservoir surface	Reservoir bottom	Stream	Reservoir surface	Reservoir bottom	Stream
FHg	Samples	69 (151)	40 (150)	9 (77)	39 (151)	11 (150)	0 (77)
	Sites	15 (15)	13 (15)	7 (14)	13 (15)	8 (15)	0 (14)
TRHg	Samples	95 (151)	88 (150)	24 (77)	76 (151)	38 (150)	8 (77)
	Sites	15 (15)	15 (15)	12 (14)	15 (15)	15 (15)	5 (14)

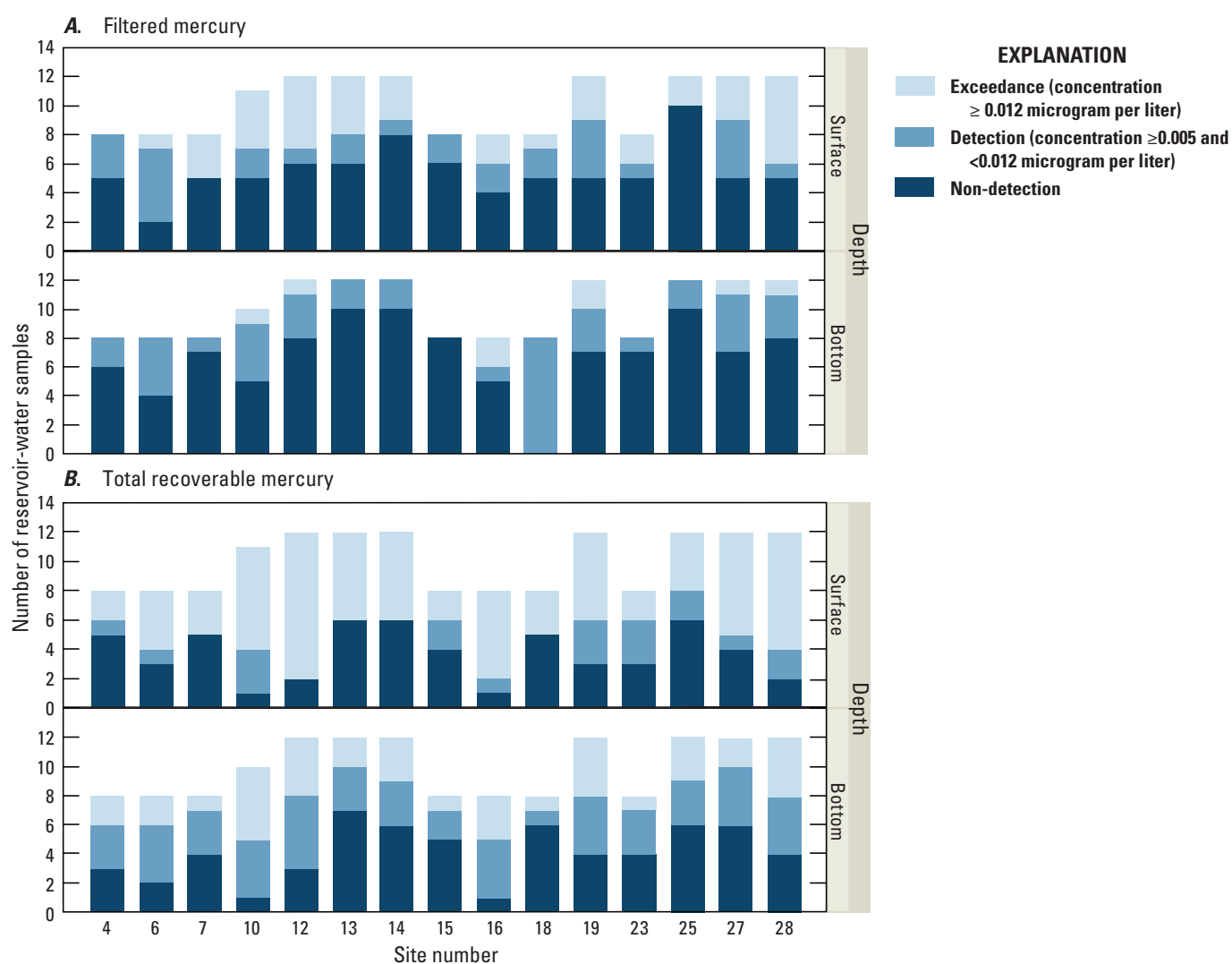


Figure 2. Graphs showing number of reservoir-water samples by sample site and depth with A, filtered mercury and B, total recoverable mercury detections and concentrations that exceeded the North Carolina water-quality criterion for total mercury.

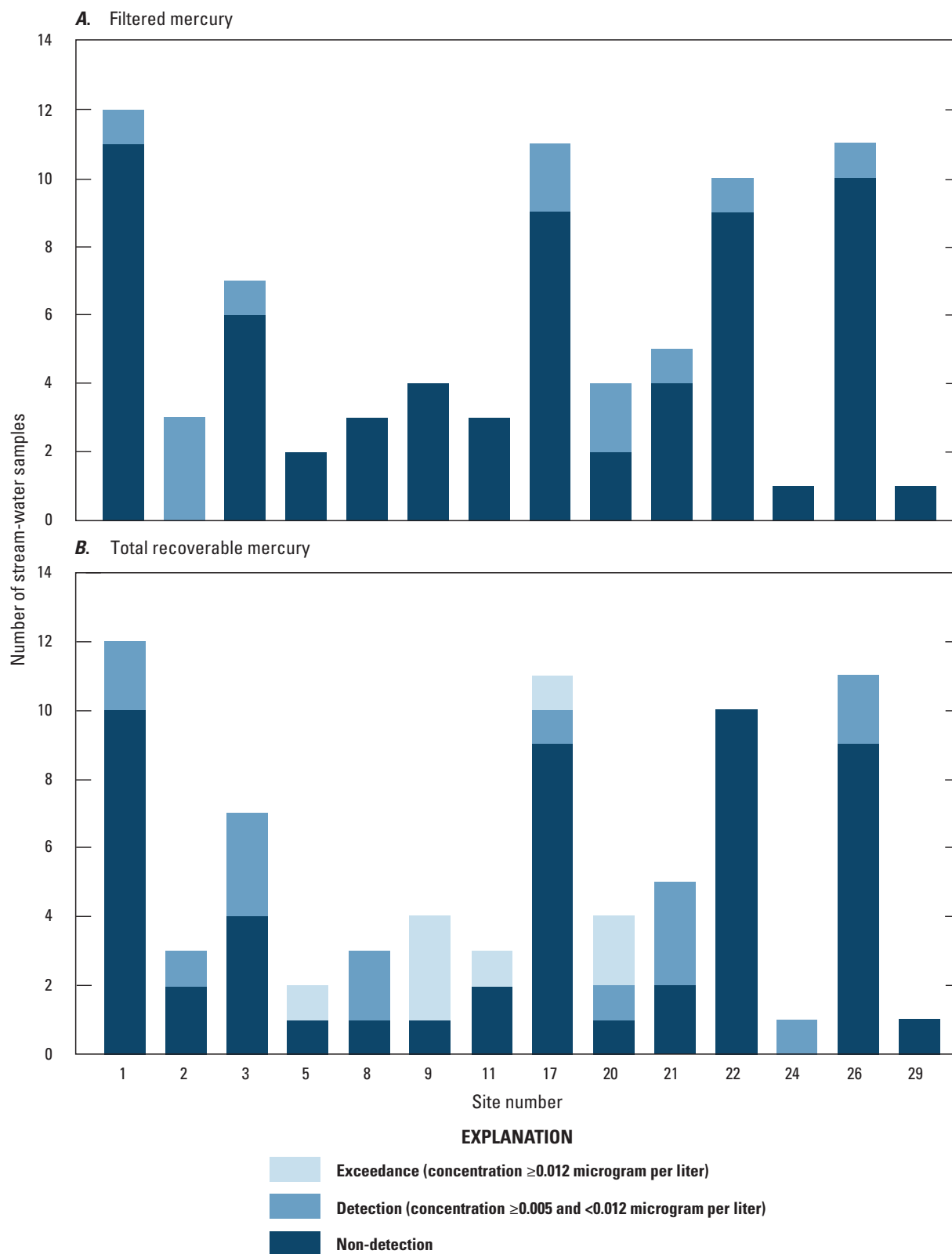


Figure 3. Graphs showing number of stream-water samples by sample site with *A*, filtered mercury and *B*, total recoverable mercury detections and concentrations that exceeded the North Carolina water-quality criterion for total mercury.

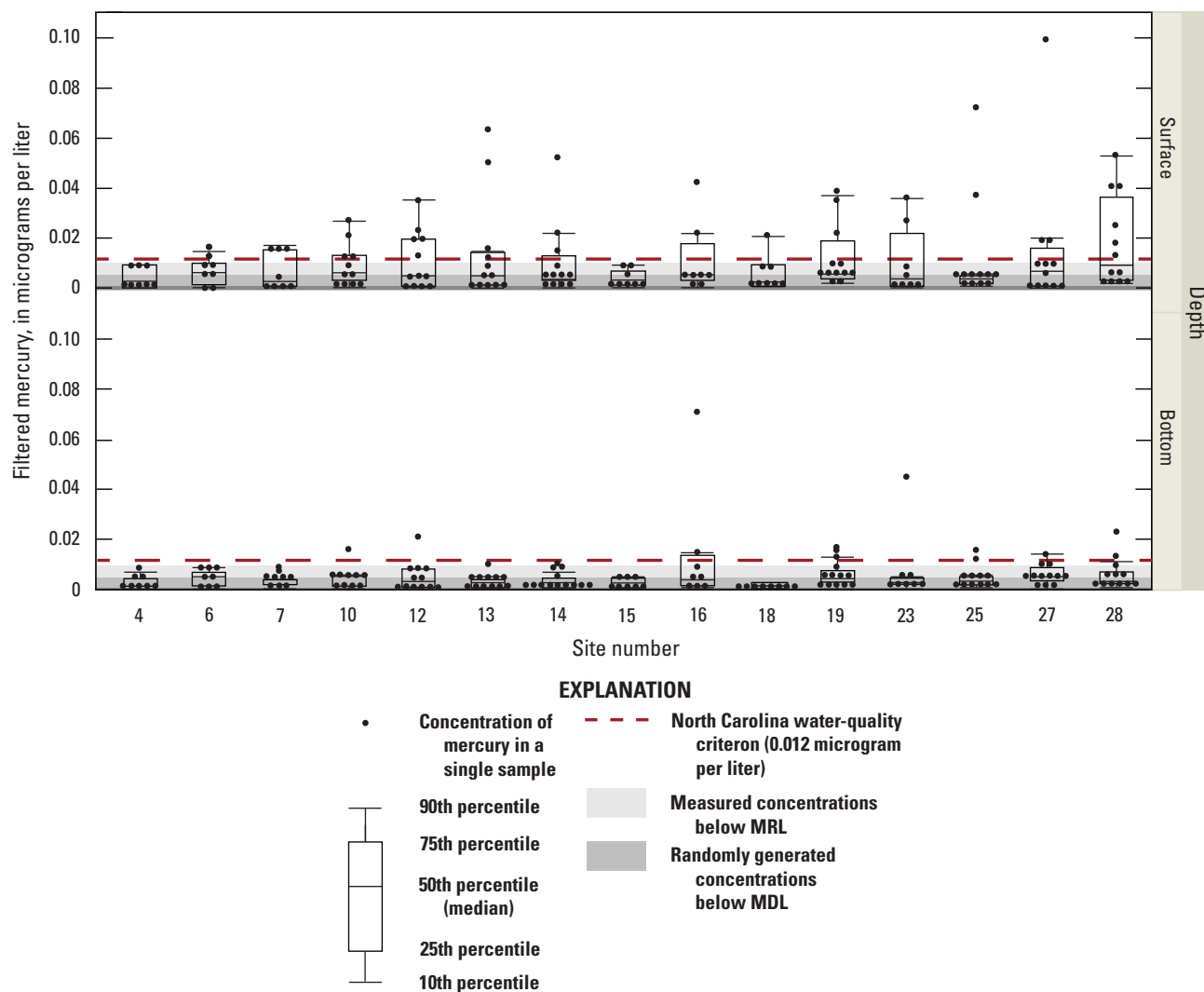


Figure 4. Boxplot showing filtered mercury concentrations in reservoir-water samples, by sample site and depth. Randomly generated values between 0 and 0.005 $\mu\text{g}\cdot\text{L}^{-1}$ were substituted for non-detections. Boxplots show the median and interquartile range (IQR), with whiskers extending to $1.5 \times \text{IQR}$ from the first and third quartiles. $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter; MRL, method reporting level; MDL, method detection level.

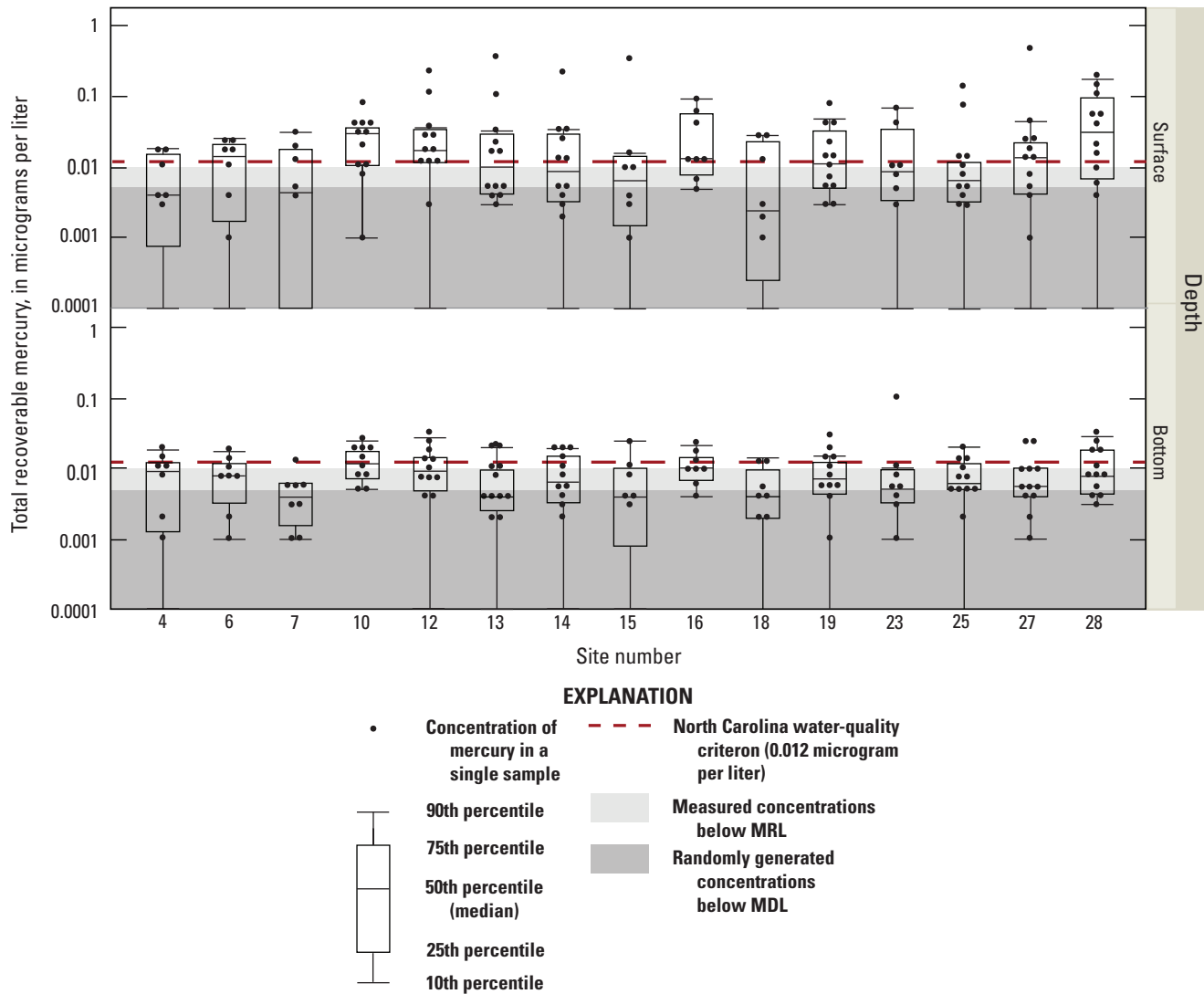


Figure 5. Boxplot showing total recovered mercury concentrations on log scale in reservoir-water samples, by sample site and depth. Randomly generated values between 0 and 0.005 $\mu\text{g}\cdot\text{L}^{-1}$ were substituted for non-detections. Generated values of 0 are not shown on log scale. Boxplots show the mean and interquartile range (IQR), with whiskers extending to $1.5 \times \text{IQR}$ from the first and third quantiles. $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter; MRL, method reporting level; MDL, method detection level.

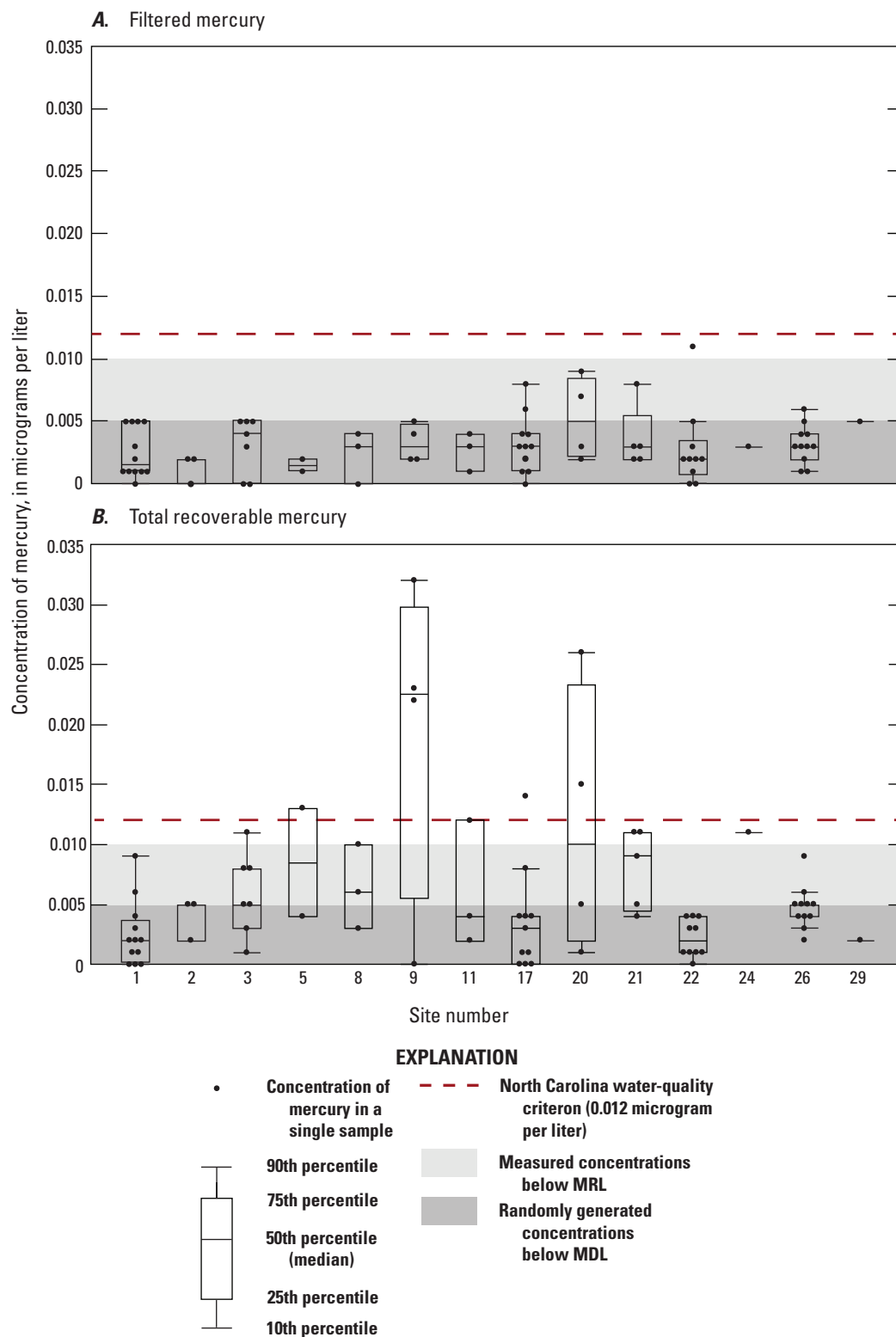


Figure 6. Boxplots showing *A*, filtered mercury, and *B*, total recoverable mercury concentrations in stream-water samples, by sample site. Randomly generated values between 0 and 0.005 $\mu\text{g}\cdot\text{L}^{-1}$ were substituted for non-detections. Boxplots show the mean and interquartile range (IQR), with whiskers extending to $1.5 \times \text{IQR}$ from the first and third quantiles. $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter; MRL, method reporting level; MDL, method detection level.

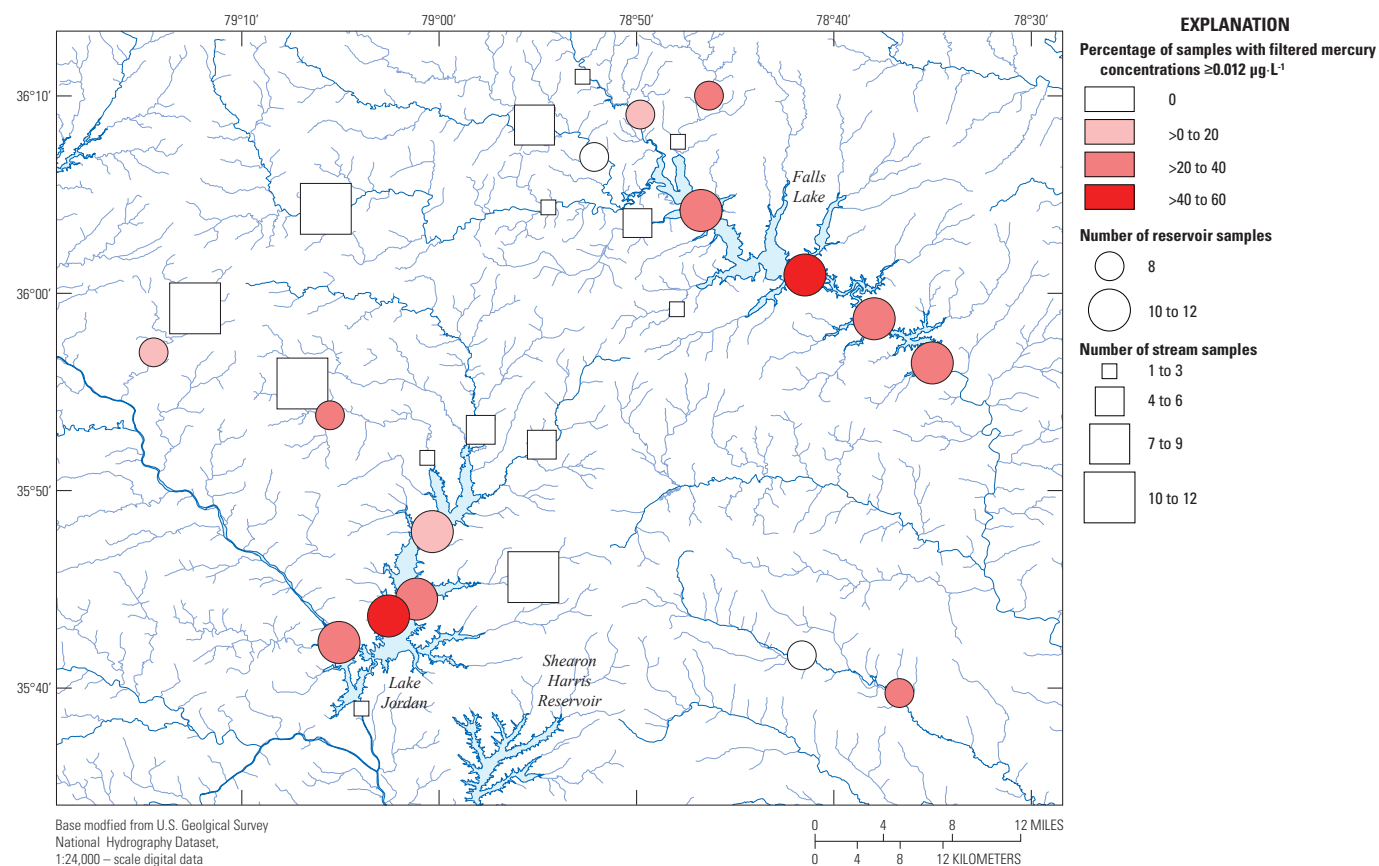


Figure 7. Map showing percentages of reservoir-surface and stream-water samples with filtered mercury concentrations exceeding the North Carolina water-quality criterion for total mercury ($\geq 0.012 \mu\text{g}\cdot\text{L}^{-1}$). $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter.

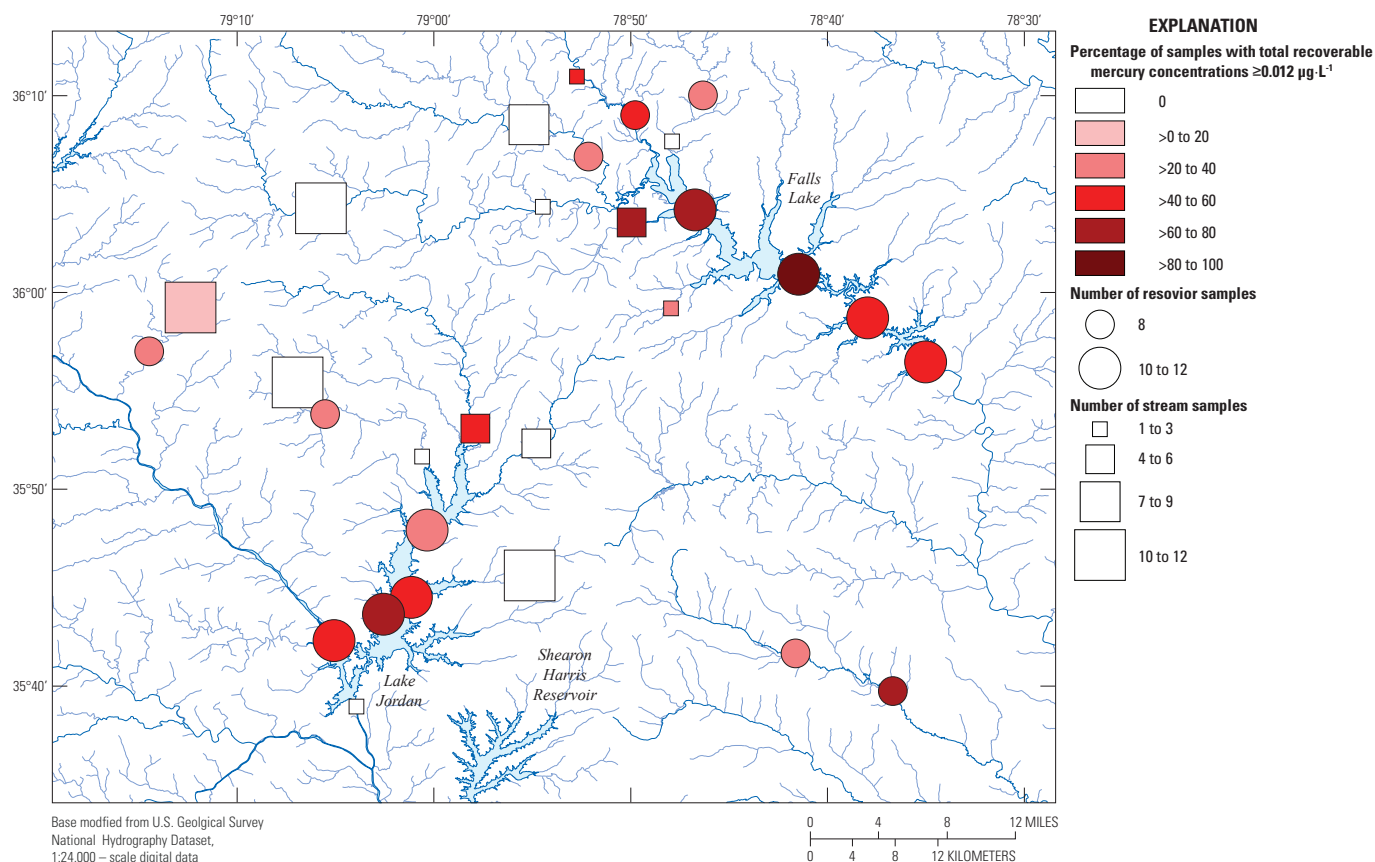


Figure 8. Map showing percentages of reservoir-surface and stream-water samples with total recoverable mercury concentrations exceeding the North Carolina water-quality criterion for total mercury ($\geq 0.012 \mu\text{g}\cdot\text{L}^{-1}$). $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter.

Seasonal Variation of Mercury Concentrations in Water Samples

Correlations between reservoir-surface and reservoir-bottom mercury concentrations were significant for every season except for winter TRHg concentrations (table 7), suggesting that seasonal lake stratification was not common in the reservoir sites. The correlations between reservoir-surface and reservoir-bottom mercury concentrations were not particularly strong in some cases (for example, $\rho = 0.502$ between surface and bottom TRHg concentrations in the spring season) and thus suggested that the degree of mixing might differ across seasons.

Seasonal patterns of median FHg and TRHg concentrations were different between reservoir surface- and bottom-water samples. Seasonal differences in median FHg concentrations were detected in reservoir surface-water samples ($X^2 = 9.108$, $p = 0.028$; fig. 9) and in median TRHg concentrations for reservoir bottom-water samples ($X^2 = 12.208$, $p = 0.007$; fig. 10) suggesting that the degree of mixing might have varied among seasons. Median reservoir-surface FHg concentrations and median reservoir-bottom TRHg concentrations were greatest in the summer and fall (fig. 9 and fig. 10, respectively). No significant differences were detected in seasonal median

reservoir-bottom FHg ($X^2 = 7.206$, $p = 0.066$) or reservoir-surface TRHg concentrations ($X^2 = 7.12$, $p = 0.068$). Stream samples also did not exhibit seasonal patterns in median FHg or TRHg concentrations ($X^2 = 2.778$, $p = 0.427$; $X^2 = 3.654$, $p = 0.301$, respectively). Concentrations of FHg in reservoir-surface samples most frequently exceeded the NCWQC for mercury in the fall followed by the summer (fig. 9, table 8). Seasonal patterns of median TRHg concentrations in reservoir bottom-water samples were similar to median FHg concentrations in reservoir surface-water samples, with concentrations most frequently exceeding the NCWQC for mercury in the fall followed by the summer (fig. 10, table 8).

Table 7. Spearman correlations (ρ with p-value in parentheses) by season between reservoir-surface and reservoir-bottom mercury concentrations for concurrently collected samples.

[FHg, filtered mercury; TRHg, total recoverable mercury; <, less than]

Season	FHg	TRHg
Spring	0.869 (<0.001)	0.502 (0.015)
Summer	0.523 (0.001)	0.660 (<0.001)
Fall	0.622 (0.013)	0.668 (0.007)
Winter	0.591 (0.010)	0.140 (0.578)

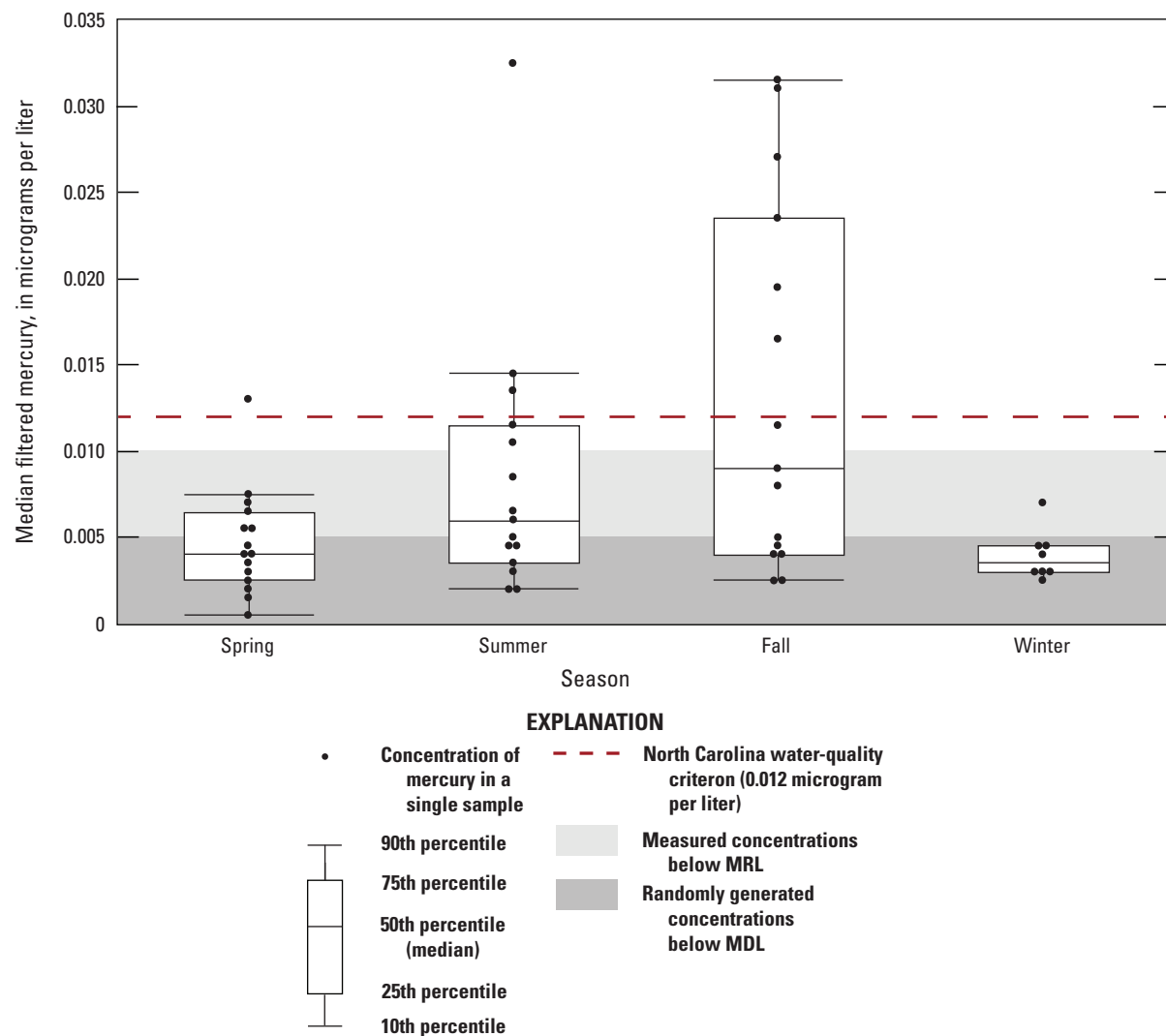


Figure 9. Boxplot showing seasonal median (by site) filtered mercury concentrations for reservoir surface-water samples. Randomly generated values between 0 and $0.005 \mu\text{g}\cdot\text{L}^{-1}$ were substituted for non-detections. Samples were only collected from 8 of the 15 sites during winter. Boxplots show the mean and interquartile range (IQR) with whiskers extending to $1.5 \times \text{IQR}$ from the first and third quantiles. MRL, method reporting level; MDL, method detection level; $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter.

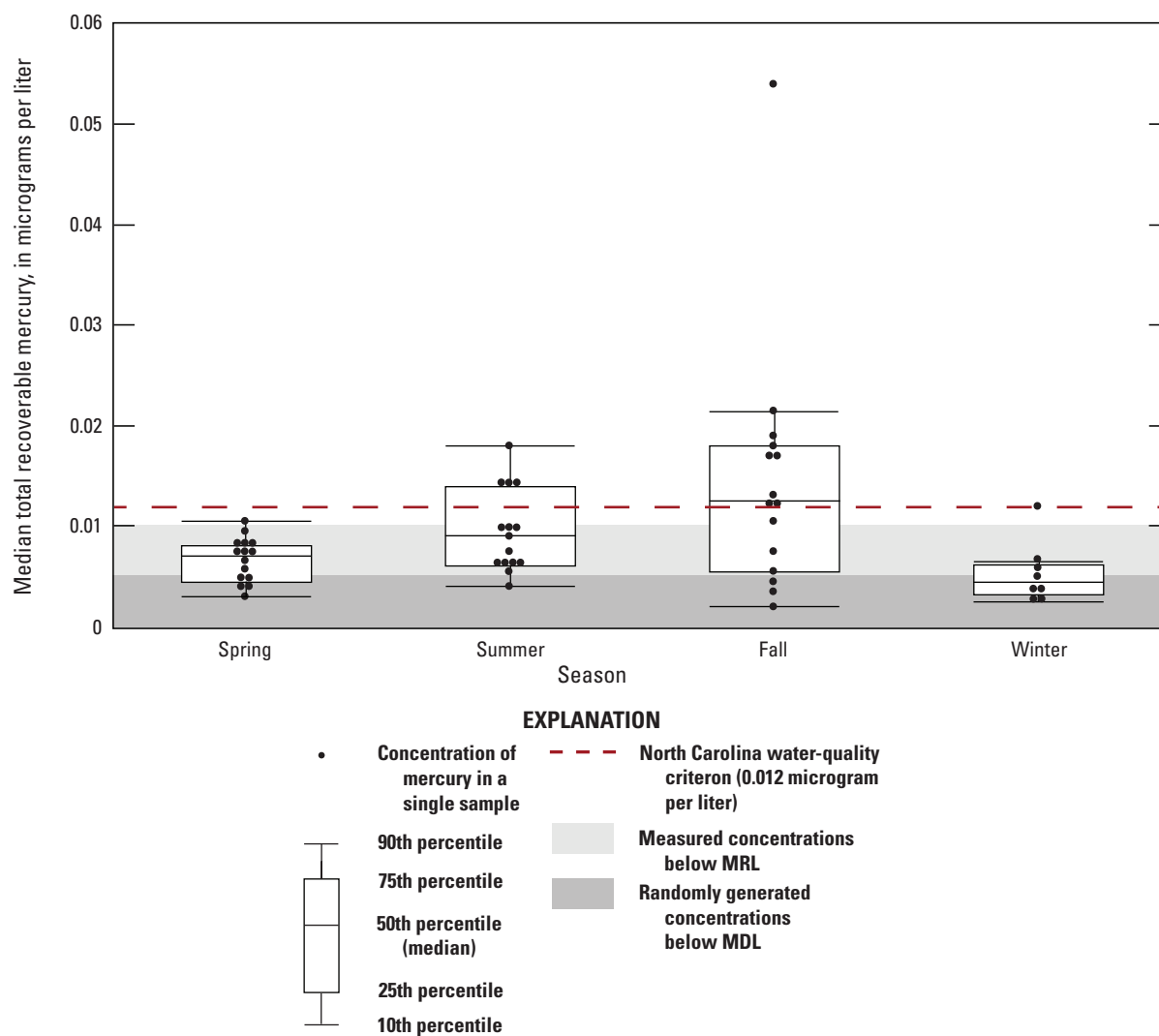


Figure 10. Boxplot showing seasonal median (by site) total recoverable mercury concentrations for reservoir bottom-water samples. Randomly generated values between 0 and $0.005 \mu\text{g}\cdot\text{L}^{-1}$ were substituted for non-detections. Samples were only collected from 8 of the 15 sites during winter. Boxplots show the mean and interquartile range (IQR) with whiskers extending to $1.5 \times \text{IQR}$ from the first and third quantiles. MRL, method reporting level; MDL, method detection level; $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter.

Table 8. Number and percentage of samples with mercury-concentration detections and exceedances by waterbody type, sample depth, and season.

[Number of reservoir and stream-water samples with filtered mercury (FHg) and total recoverable mercury (TRHg) detections (method detection level = 0.005 microgram per liter [$\mu\text{g}\cdot\text{L}^{-1}$]) and concentrations that exceeded the North Carolina water-quality criterion for mercury of $0.012\ \mu\text{g}\cdot\text{L}^{-1}$. Abbreviations: N, number of samples; %, percent]

Waterbody type	Depth	Season	N	FHg detections	FHg exceedances	TRHg detections	TRHg exceedances
Reservoir	Surface	Spring	30	12 (40%)	2 (7%)	21 (70%)	15 (50%)
		Summer	60	31 (52%)	20 (33%)	36 (60%)	31 (52%)
		Fall	30	17 (57%)	12 (40%)	21 (70%)	19 (63%)
		Winter	31	9 (29%)	5 (16%)	17 (56%)	11 (36%)
	Bottom	Spring	30	6 (20%)	0 (0%)	18 (60%)	2 (7%)
		Summer	60	14 (23%)	4 (7%)	39 (65%)	22 (37%)
		Fall	30	12 (40%)	6 (20%)	20 (67%)	12 (40%)
		Winter	30	8 (31%)	1 (3%)	11 (41%)	2 (9%)
Stream	Surface of the water column	Spring	28	6 (15%)	0 (0%)	11 (34%)	2 (5%)
		Summer	18	1 (5%)	0 (0%)	2 (15%)	0 (0%)
		Fall	12	1 (6%)	0 (0%)	6 (56%)	5 (44%)
		Winter	19	1 (3%)	0 (0%)	5 (43%)	1 (10%)

Bed Sediment

Mercury was detected in all 28 bed sediment samples, with measured THg concentrations ranging from 0.02 to $0.82\ \text{mg}\cdot\text{kg}^{-1}$ and bulk-calculated THg concentrations ranging from <0.01 to $0.39\ \text{mg}\cdot\text{kg}^{-1}$ in stream sediment samples (mean = $0.04\ \text{mg}\cdot\text{kg}^{-1}$, median $<0.01\ \text{mg}\cdot\text{kg}^{-1}$; table 5) and from 0.02 to $0.13\ \text{mg}\cdot\text{kg}^{-1}$ in reservoir samples (mean = $0.06\ \text{mg}\cdot\text{kg}^{-1}$, median = $0.07\ \text{mg}\cdot\text{kg}^{-1}$; table 5, fig. 11). Only one site (site 24, a stream site) had a bulk-calculated THg concentration of $0.39\ \text{mg}\cdot\text{kg}^{-1}$ (table 5) that exceeded the level below which adverse effects to sediment-dwelling organisms were unlikely ($0.18\ \text{mg}\cdot\text{kg}^{-1}$); however, $0.39\ \text{mg}\cdot\text{kg}^{-1}$ is below the $1.06\ \text{mg}\cdot\text{kg}^{-1}$ concentration of total mercury in bed sediment that is expected to cause adverse effects to sediment-dwelling organisms (MacDonald and others, 2000). All FHg concentrations in water samples collected simultaneously with the bed sediment samples were below the MRL, and no significant associations were detected between bulk-calculated bed sediment THg and FHg for stream-water samples ($\rho = -0.1864$, $p = 0.5421$), reservoir surface-water samples ($\rho = 0.1425$, $p = 0.6125$), or bottom-water samples ($\rho = 0.0705$, $p = 0.8029$). No significant correlations were detected between bed sediment bulk-calculated THg concentrations and TRHg in stream-water samples ($\rho = 0.4308$, $p = 0.1416$), reservoir surface-water samples ($\rho = -0.1978$, $p = 0.4797$), or reservoir bottom-water samples ($\rho = 0.1764$, $p = 0.5294$).

Ancillary Constituent, Precipitation, and Land-Use Associations with Median Mercury Concentrations in Water and Bed Sediment

Land-use variables (percentage of drainage area composed of urban development, forest, impervious surface, and wetland; table 1) were all highly correlated among each other for both stream and reservoir sites. The weakest correlation (smallest absolute value of the Spearman ρ) between land-use types in stream catchments was between forest and wetland areas ($|\rho| = 0.670$; $p = 0.0087$). For the 13 reservoir catchments for which drainage-area and land-use percentages could be calculated, the weakest correlation in land-use types was between urban and wetland area ($|\rho| = 0.725$; $p = 0.005$). If significant correlations were found between mercury concentrations and multiple land-use variables, this could be an artifact of the significant correlations of the percentages of areas with different land-use types within catchments. The interpretation of land-use associations with mercury concentrations might thus be confounded by the correlations among land-use variables.

The only water-sample mercury constituents that exhibited seasonal patterns in median concentrations and thus were investigated for seasonal associations with potential driver variables were reservoir-surface FHg concentrations and reservoir-bottom TRHg concentrations. Numerous non-detections of FHg prevented the analysis of reservoir-bottom median FHg and stream median FHg associations, as well as reservoir-surface spring and winter median FHg concentrations, with potential driver variables. No correlations were detected between summer or fall reservoir-surface FHg

concentrations and any of the ancillary constituents, antecedent precipitation, or land-use variables (tables 9 and 10). The only potential association between reservoir-water sample mercury concentrations and ancillary constituents was between reservoir-bottom winter TRHg and DO ($\rho = 0.810$, $p = 0.015$; table 9); however, reservoir-bottom TRHg concentrations exhibited significant and potential associations with several other potential driver variables. Reservoir-bottom summer TRHg concentrations were significantly associated with drainage area ($\rho = 0.777$, $p = 0.002$; table 10) and were potentially associated with antecedent precipitation and all of the land-use variables (table 10). Reservoir-bottom winter TRHg concentrations were also potentially associated with antecedent precipitation; however, potential associations between reservoir-bottom winter TRHg concentrations and antecedent precipitation were positive in contrast with the potential associations between reservoir-bottom summer TRHg concentrations and antecedent precipitation, which were negative (table 10). TRHg in streams was positively associated with $\text{TOC}_{\text{water}}$ ($\rho = 0.883$, $p = <0.001$; table 9) and SSC ($\rho = 0.728$, $p = 0.005$; table 9) and potentially associated with streamflow

($\rho = 0.674$, $p = 0.016$; table 9). Stream TRHg concentrations also exhibited a potential association with wetlands ($\rho = 0.619$, $p = 0.018$; table 10).

Bed sediment THg was significantly associated with one potential driver variable and potentially associated with several potential driver variables (tables 11 and 12). For reservoir sites, bed sediment THg was significantly associated with the amount of precipitation within 7 days prior to sample collection ($\rho = 0.708$, $p = 0.003$; table 12) and potentially associated with SO_4^{2-} ($\rho = -0.577$, $p = 0.024$; table 11), drainage area ($\rho = -0.581$, $p = 0.038$; table 12), and TOC_{sed} ($\rho = 0.593$, $p = 0.020$; table 12). THg concentrations in bed sediment from streams were also potentially correlated with TOC_{sed} , although the direction of the association differed from the direction of association between bed sediment THg and TOC_{sed} in reservoirs ($\rho = -0.610$, $p = 0.027$; table 12).

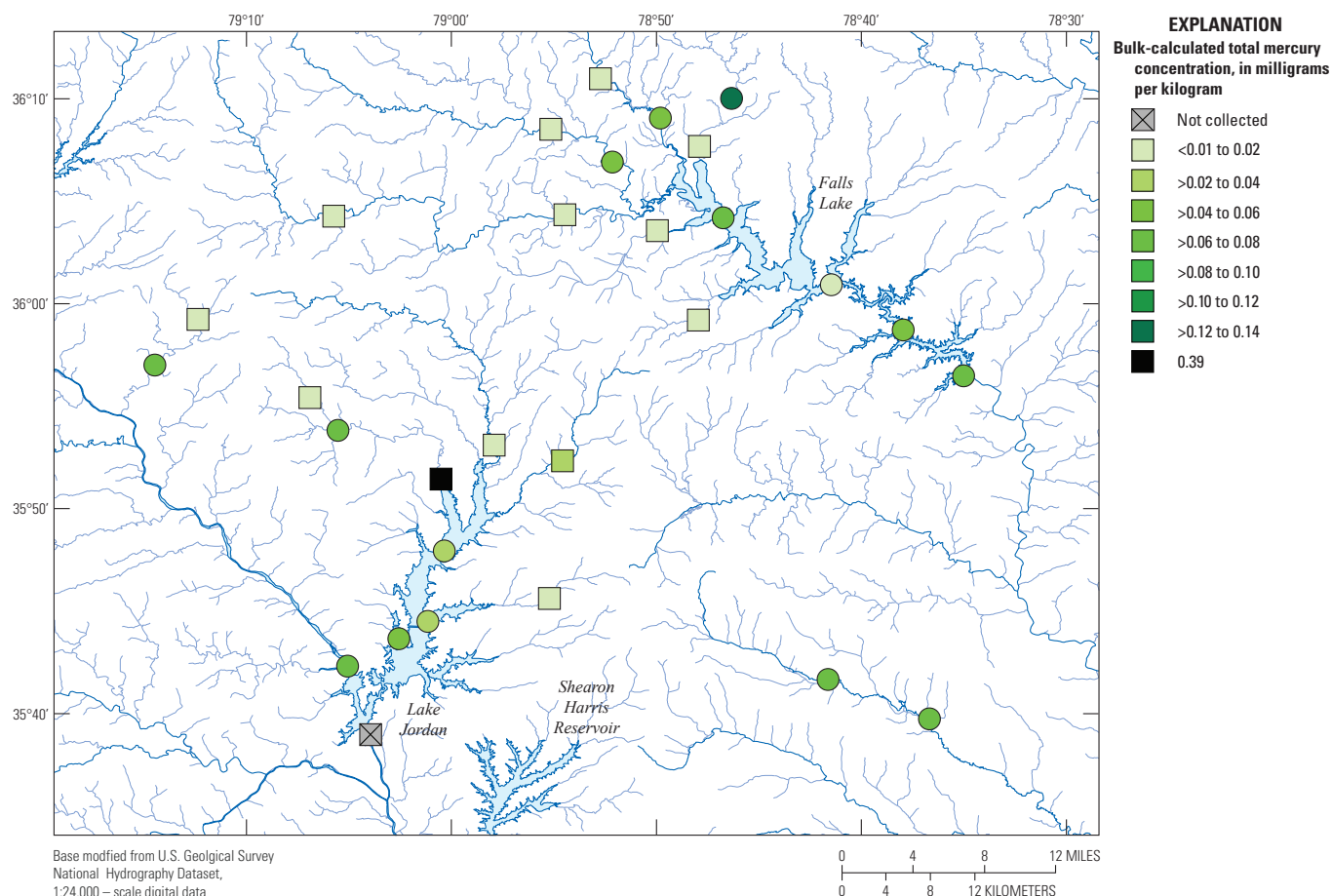


Figure 11. Map showing bed sediment bulk-calculated total mercury concentrations. Circles indicate reservoir sites; squares indicate stream sites.

Table 9. Spearman rank correlations (ρ , with p-value in parentheses) between median mercury concentrations and ancillary constituents for water samples.

[Mercury concentrations (FHg or TRHg) in samples in which FHg or TRHg were not detected were substituted with randomly generated values below the MDL for median computations. Abbreviations: FHg, filtered mercury; TRHg, total recoverable mercury; MDL, method detection level; DO, dissolved oxygen; SC, specific conductance; SO_4^{2-} , sulfate; *Chloro**a*, chlorophyll *a*; *Pheo**a*, pheophytin *a*; DOC, dissolved organic carbon; $\text{TOC}_{\text{water}}$, total organic carbon in water; SSC, suspended-sediment concentration; NA, no association was calculated because of either no variability in potential driver variable values, or more than 50 percent of median values for correlative analysis were below the MDL; NM, not measured]

Waterbody type	Mercury constituent	Depth	Season	DO	pH	SC	SO ₄ ²⁻	Chloro _a	Pheo _a	DOC	TOC _{water}	SSC	Streamflow		
Reservoir	FHg	Surface	Spring	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA		
			Summer	0.453 (0.090)	0.482 (0.069)	0.2068 (0.460)	0.501 (0.057)	−0.476 (0.073)	−0.391 (0.149)	0.023 (0.934)	−0.357 (0.191)	NM	NM		
			Fall	0.238 (0.393)	0.217 (0.438)	0.308 (0.264)	0.360 (0.188)	0.386 (0.155)	0.295 (0.286)	−0.267 (0.337)	0.047 (0.869)	NM	NM		
			Winter	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA		
	TRHg	Bottom	Overall	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA	
		Surface	Overall	0.100 (0.724)	0.032 (0.909)	0.470 (0.077)	0.386 (0.155)	0.190 (0.499)	0.322 (0.242)	−0.106 (0.708)	0.224 (0.423)	NM	NM	NM	
			Bottom	Spring	0.312 (0.257)	0.372 (0.173)	0.224 (0.422)	0.280 (0.312)	NM	NM	0.047 (0.869)	NM	NM	NM	
		Summer		−0.242 (0.386)	0.136 (0.628)	0.170 (0.545)	0.377 (0.166)	NM	NM	−0.399 (0.141)	NM	NM	NM		
		Fall		0.188 (0.501)	0.192 (0.492)	0.449 (0.094)	0.195 (0.487)	NM	NM	0.181 (0.519)	NM	NM	NM		
		Winter		0.810 (0.015) ¹	−0.122 (0.774)	0.262 (0.531)	0.262 (0.531)	NM	NM	0.262 (0.531)	NM	NM	NM		
		Stream	FHg	Surface of the water column	Overall	NA	NA	NA	NA	NA	NA	NA	NA	NA	NA
			TRHg		Overall	−0.315 (0.272)	−0.043 (0.884)	−0.055 (0.851)	0.308 (0.284)	NM	NM	0.463 (0.091)	0.883 (<0.001) ^{1, 2}	0.728 (0.005) ^{1, 2}	0.674 (0.016) ¹

¹ $p < 0.05$.² $p < 0.01$.

Table 10. Spearman rank correlations (Spearman's ρ , with p-value in parentheses) for associations between seasonal median and overall median mercury concentrations and land use for reservoir and stream sites.

[Mercury concentrations (FHg or TRHg) for samples in which FHg or TRHg were not detected were substituted with randomly generated values below the MDL for median computations. Abbreviations: FHg, filtered mercury; TRHg, total recoverable mercury; MDL, method detection level; Precip, total precipitation antecedent to sample collection for the number of days listed; NA, no variability in potential driver variable values or more than 50 percent of median values for correlative analysis below the MDL]

Waterbody type	Mercury constituent	Depth	Season	Precip _{1day}	Precip _{2day}	Precip _{7day}	Drainage area	Urban	Forest	Impervious surface	Wetland
Reservoir	FHg	Surface	Spring	NA	NA	NA	NA	NA	NA	NA	NA
			Summer	-0.364 (0.183)	-0.468 (0.079)	-0.230 (0.410)	0.551 (0.051)	0.113 (0.713)	-0.154 (0.615)	0.094 (0.761)	0.501 (0.081)
			Fall	-0.403 (0.137)	-0.265 (0.340)	-0.381 (0.161)	0.380 (0.200)	0.163 (0.596)	0.033 (0.915)	0.110 (0.720)	0.135 (0.660)
			Winter	NA	NA	NA	NA	NA	NA	NA	NA
	TRHg	Bottom	Overall	NA	NA	NA	NA	NA	NA	NA	NA
		Surface	Overall	NA	-0.618 (0.014) ¹	-0.387 (0.154)	0.534 (0.060)	0.270 (0.373)	-0.322 (0.284)	0.393 (0.184)	0.344 (0.250)
			Spring	-0.034 (0.903)	0.100 (0.722)	-0.278 (0.317)	0.388 (0.190)	0.388 (0.190)	-0.363 (0.223)	0.465 (0.109)	0.452 (0.121)
		Bottom	Summer	-0.571 (0.026) ¹	-0.631 (0.012) ¹	-0.168 (0.549)	0.777 (0.002) ^{1,2}	0.596 (0.031) ¹	-0.580 (0.038) ¹	0.555 (0.049) ¹	0.671 (0.012) ¹
			Fall	-0.433 (0.107)	-0.233 (0.403)	-0.460 (0.085)	0.357 (0.231)	0.071 (0.817)	0.055 (0.859)	0.093 (0.762)	0.028 (0.929)
			Winter	0.203 (0.630)	0.792 (0.019) ¹	0.807 (0.015) ¹	-0.600 (0.208)	-0.029 (0.957)	-0.486 (0.329)	0.429 (0.397)	-0.314 (0.544)
			Overall	NA	NA	NA	NA	NA	NA	NA	NA
Stream	FHg	Surface of the water column	Overall	NA	NA	NA	NA	NA	NA	NA	NA
	TRHg		Overall	0.227 (0.435)	0.368 (0.195)	0.276 (0.340)	0.118 (0.689)	0.463 (0.095)	-0.313 (0.276)	0.490 (0.075)	0.619 (0.018) ¹

¹p<0.05.

²p<0.01.

Table 11. Spearman rank correlations (ρ , with p-value in parentheses) between bed sediment bulk-calculated total mercury concentrations and ancillary constituents for water samples collected simultaneously with bed sediment samples for reservoir and stream sites.

[Abbreviations: DO, dissolved oxygen; SC, specific conductance; SO_4^{2-} , sulfate; Chloro a , chlorophyll a ; Pheo a , pheophytin a ; DOC, dissolved organic carbon; $\text{TOC}_{\text{water}}$, total organic carbon in water; SSC, suspended-sediment concentration]

Waterbody type	Sampling depth	pH	DO	SC	SO_4^{2-}	Chloro a	Pheo a	DOC	$\text{TOC}_{\text{water}}$	SSC	Streamflow
Reservoir	Surface	−0.380 (0.163)	0.145 (0.606)	−0.433 (0.107)	−0.492 (0.063)	−0.145 (0.607)	−0.464 (0.081)	−0.300 (0.277)	−0.281 (0.310)	NM	NM
	Bottom	−0.300 (0.278)	0.2004 (0.474)	−0.441 (0.100)	−0.577 (0.024) ¹	NM	NM	−0.411 (0.128)	NM	NM	NM
Stream	Surface of water column	0.370 (0.214)	−0.291 (0.335)	0.415 (0.151)	0.503 (0.080)	NM	NM	−0.223 (0.464)	0.000 (1.000) ²	−0.600 (0.400)	−0.169 (0.620)

¹ $p < 0.05$.

² $\text{TOC}_{\text{water}}$ measured in only 4 of 13 stream-water samples that were collected simultaneously with bed sediment samples.

Table 12. Spearman rank correlations (ρ , with p-value in parentheses) for associations between bed sediment bulk-calculated total mercury concentrations and potential driver variables such as percentages of organic carbon in bed sediment, antecedent precipitation, drainage area, and land use.

[Abbreviations: TOC_{sed} , percentage of organic carbon in bed sediment; Precip, antecedent precipitation for the number of days preceding sample collection]

Waterbody type	TOC_{sed}	Precip _{1day}	Precip _{2day}	Precip _{7day}	Drainage area	Urban	Forest	Impervious surface	Wetland
Reservoir	0.593 (0.020) ¹	0.009 (0.974)	0.346 (0.206)	0.708 (0.003) ^{1,2}	−0.581 (0.038) ¹	−0.228 (0.453)	0.415 (0.158)	−0.404 (0.171)	−0.490 (0.089)
Stream	−0.610 (0.027) ¹	−0.434 (0.138)	0.406 (0.169)	−0.140 (0.647)	−0.066 (0.830)	−0.052 (0.865)	0.325 (0.279)	0.003 (0.993)	0.250 (0.409)

¹ $p < 0.05$.

² $p < 0.01$.

Discussion

In this study, mercury concentrations in water and bed sediment samples from streams and reservoirs were measured and assessed for frequency of occurrences, concentrations, and exceedances relative to State, national, and international benchmark concentrations. Mercury levels were generally greater at reservoir sites than stream sites for both water and bed sediment samples, and mercury concentrations more frequently exceeded the NCWQC for mercury in reservoirs than in streams. The experimental design of this study, however, confounds an ecological interpretation of this comparison for water samples. Stream sites were most frequently sampled in the spring, when FHg concentrations in reservoir surface-water samples and TRHg concentrations in reservoir bottom-water samples are relatively low. Stream sites were also least frequently sampled in the fall, when FHg concentrations in reservoir surface-water samples and TRHg concentrations in reservoir bottom-water samples are highest. If mercury

concentrations in stream- and reservoir-water samples follow similar seasonal patterns, the overall detection and exceedance frequencies and median mercury concentrations may be biased low for streams and high for reservoirs.

At least 20 percent of surface samples from each reservoir site had TRHg concentrations exceeding the NCWQC for mercury, suggesting that mercury exceedances were not isolated in a limited number of sites in the Triangle Area. Similar inferences could be made on the basis of estimates of NCWQC exceedances made by using FHg concentrations that exceeded the NCWQC in reservoir surface-water samples; at least one sample collected in all but two reservoir sites exceeded the NCWQC. FHg concentrations did not exceed the NCWQC for mercury in any stream sites; this result could indicate that mercury concentrations were generally low in streams. Alternatively, the lack of FHg concentrations exceeding the NCWQC for mercury in stream samples may have been a result of the limited number of samples collected

during the summer, or of samples collected in the five sites with TRHg exceedances of the NCWQC, where a high percentage of mercury in streams sites occurred in particulate form.

Trends in Mercury Concentrations in Water and Bed Sediment

Concentrations of mercury constituents in samples collected from reservoir depths that exhibited seasonal differences (FHg in reservoir-surface samples, TRHg in reservoir-bottom samples) were both highest in the fall. Atmospheric deposition peaks in the summer in the southeastern United States (Selin and Jacob, 2008), indicating that other factors besides atmospheric deposition are driving this trend. None of the investigated potential driver variables were associated with reservoir-surface FHg in the fall, nor were any associated with reservoir-bottom TRHg in the fall. Whereas the degree of reservoir mixing was assessed in this study by the strength and significance of correlations between surface and bottom mercury concentrations, other hydrologic factors (for example, inflows from stream and rivers) were not investigated. Wildman (2016) found that when inflow was high into a reservoir in Oklahoma, FHg concentrations were controlled by inflow hydrology, whereas biogeochemistry drove mercury flux when inflows were low through sequestration of mercury at the bottom of reservoirs under anoxic conditions. Similarly, Regnell (1994) found that under anoxic conditions and increased microbial activity, TRHg and MeHg concentrations increased in the water after release from the sediment. Low DO as a driver of TRHg in fall reservoir-bottom samples seems unlikely because TRHg concentrations were not associated with DO. TRHg in winter reservoir-bottom samples, however, were potentially positively associated with DO, which would not be expected if anoxic conditions had led to an increase in TRHg concentrations in bottom-water samples. Further analysis or additional studies to assess potential hydrologic influences on seasonal mercury trends may be warranted.

Summer TRHg concentrations in reservoir-bottom samples were significantly associated with drainage area. Potential associations between summer TRHg in reservoir-bottom samples and potential driver variables included (1) negative associations with precipitation one or two days prior to sample collection, (2) forest area, and (3) positive associations with urban, impervious-surface, and wetland areas. Atmospheric deposition is frequently the primary source of mercury in lake systems (Sorensen and others, 1990; USEPA, 1997b). The relationship between drainage-area and reservoir-bottom TRHg levels in the summer could be due to greater total masses of mercury deposited from the atmosphere in larger basins; at least some of this mercury would then be transported to reservoirs in runoff from impervious surfaces in the catchment (Lee and Iverfeldt, 1991; Brigham and others, 2009).

Potential associations between antecedent-precipitation and reservoir-bottom TRHg in the summer were negative, which would be contrary to expectation if wet deposition were a dominant source of mercury in reservoirs in the summer. Data collected between 2007 and 2009 from the National Atmospheric Deposition Program (NADP, 2020) site proximal to the Triangle Area (NC26, located in Candor, N.C.) indicated that TRHg concentrations in precipitation peaked in the summer with a median concentration of $0.014 \mu\text{g}\cdot\text{L}^{-1}$, (fig. 12) which was higher than the median summer TRHg concentrations in reservoir-bottom samples ($0.009 \mu\text{g}\cdot\text{L}^{-1}$). This association might be an instance of a type I error or indicate an inverse relation between precipitation and summer TRHg concentrations in reservoir-bottom samples.

Mercury concentrations in streams are often a result of terrestrial processes (Brigham and others, 2009; Bradley and others, 2011). Litterfall can be a source of mercury inputs to streams equal to, if not more so, wet atmospheric deposition in fluvial systems (Journey and others, 2012). That TRHg in streams in this study was significantly associated with $\text{TOC}_{\text{water}}$ and SSC suggests that inputs of particulate-bound mercury were dominant sources of mercury to stream systems. Mercury has a strong affinity for particulate organic matter (Haitzer and others, 2003; Brigham and others, 2009) and can bind to litterfall, which serves as an indirect input of mercury to streams from atmospherically derived mercury (Grigal, 2002; Journey and others, 2012; Bravo and others, 2018). Horowitz and Stephens (2008) also found that stream sites in basins with more than 50 percent forested cover had bed sediment mercury levels above levels measured at baseline sites (that is, population density below the 50th percentile and less than 5 percent urban) as a result of dry atmospheric deposition on leaf surfaces and possibly from higher levels of organic matter entering the streams. The mean percentage of forest cover in drainage basins in this study was 51 percent for reservoir sites (ranging from 3 to 72 percent) and 52 percent for stream sites (ranging from 16 to 73 percent). Whereas no association between mercury in stream water and forest cover was detected in this study, dry deposition on leaf cover from the abundant forested area at the study sites could be a possible source of mercury in bed sediment at the stream sites.

A potential positive association was also detected for stream TRHg (and reservoir-bottom summer TRHg) with wetland area. Numerous studies have noted associations between wetland abundance and MeHg (Wentz and others, 2014), which was not directly measured but would be included in the FHg and TRHg operational fractions if present. If wetlands are a source of MeHg in stream and reservoir systems in the Triangle Area, this relation might indicate that catchments with a higher percentage of wetland coverage have a higher ratio of MeHg to TRHg than catchments with less wetland coverage.

No association was detected between bulk-calculated bed sediment THg and land-use features except for a potential negative association with drainage area at reservoir sites. Sedimentation of particle-bound mercury is often the primary

process that removes mercury in the water column (Sorensen and others, 1990; USEPA, 1997a). Marvin-DiPasquale and others (2009) found that in stream systems, the percentage of fines in bed sediment had a strong correlation ($r^2 = 0.70$) with bed sediment THg. A later exploration of the association between drainage area and the percentage of fines in reservoir-bed sediment (defined as the percentage of sample grains less than $63\ \mu\text{m}$ in diameter) indicated that, with the exclusion of reservoir site 27, the percentage of fines in sediment was generally greater for samples collected from reservoir sites with smaller drainage areas than those with larger drainage areas (Pearson's $r = 0.588$, $p = 0.0442$).

Bulk-calculated bed sediment THg concentrations in reservoirs were potentially negatively associated with SO_4^{2-} in water samples from the bottom of the water column. Mercury flux at the sediment-water interface is influenced by numerous factors (Kudo and Hart, 1974; Dyrssen and Wedborg, 1991) that complicate an ecological interpretation of this result. Microbial methylation rates of mercury in bed sediment are generally highest in the top layer of bed sediment (Korthals and Winfrey, 1987). Sulfate-reducing bacteria in sulfate concentrations in bed sediment can affect rates of MeHg production (Langer and others, 2001). If the association between THg and SO_4^{2-} is ecologically significant, one potential explanation could be that SO_4^{2-} at the sediment-water interface is used by sulfate-reducing bacteria in the methylation and release of mercury from bed sediment to the water column.

The direction of association between TOC_{sed} and THg in bed sediment differed between reservoir and stream sites. THg in reservoir-bed sediments was positively associated with TOC_{sed} , whereas THg in stream-bed sediment was negatively associated with TOC_{sed} . Many factors influence

mercury flux in bed sediment (Kudo and Hart, 1974; Dyrssen and Wedborg, 1991), including organic carbon. For example, mercury is known to bind strongly with dissolved organic matter, which can affect the production and bioaccumulation of methylmercury (for review, see Ravichandran, 2004). Stream sites had relatively little variation in TOC_{sed} concentrations compared to reservoir sites. (The range of bulk-calculated THg in reservoir versus stream sites was $0.019\text{--}0.129$ [0.026 standard deviation] $\text{mg}\cdot\text{kg}^{-1}$ and $<0.001\text{--}0.035$ [0.010 standard deviation] $\text{mg}\cdot\text{kg}^{-1}$, respectively.) This lack of variation suggests that if the relationship between stream-bed sediment THg and TOC_{sed} were significant, any causative association had relatively little influence on THg concentrations in bed sediment.

Reservoir-bed sediment THg was also significantly correlated with antecedent precipitation during the seven days prior to sample collection. Mercury uptake in bed sediments can be influenced by factors including substrate type, presence of oxides, concentrations of mercury in water at the water-sediment interface, and movement of water at the water-sediment interface (Kudo and Hart, 1974; Jackson, 1989). Wet deposition or surface runoff and transport of litterfall may have resulted in higher concentrations of mercury in the water column. Kudo and Hart (1974) found that the rate of inorganic mercury uptake in bed sediment from the water-sediment interface was greatest when mercury was first introduced to the system. If precipitation resulted in direct or indirect inputs of mercury to the reservoirs, this input suggests that seven days may have been a long enough time window for mercury inputs to the reservoirs to be taken up by the bed sediment.

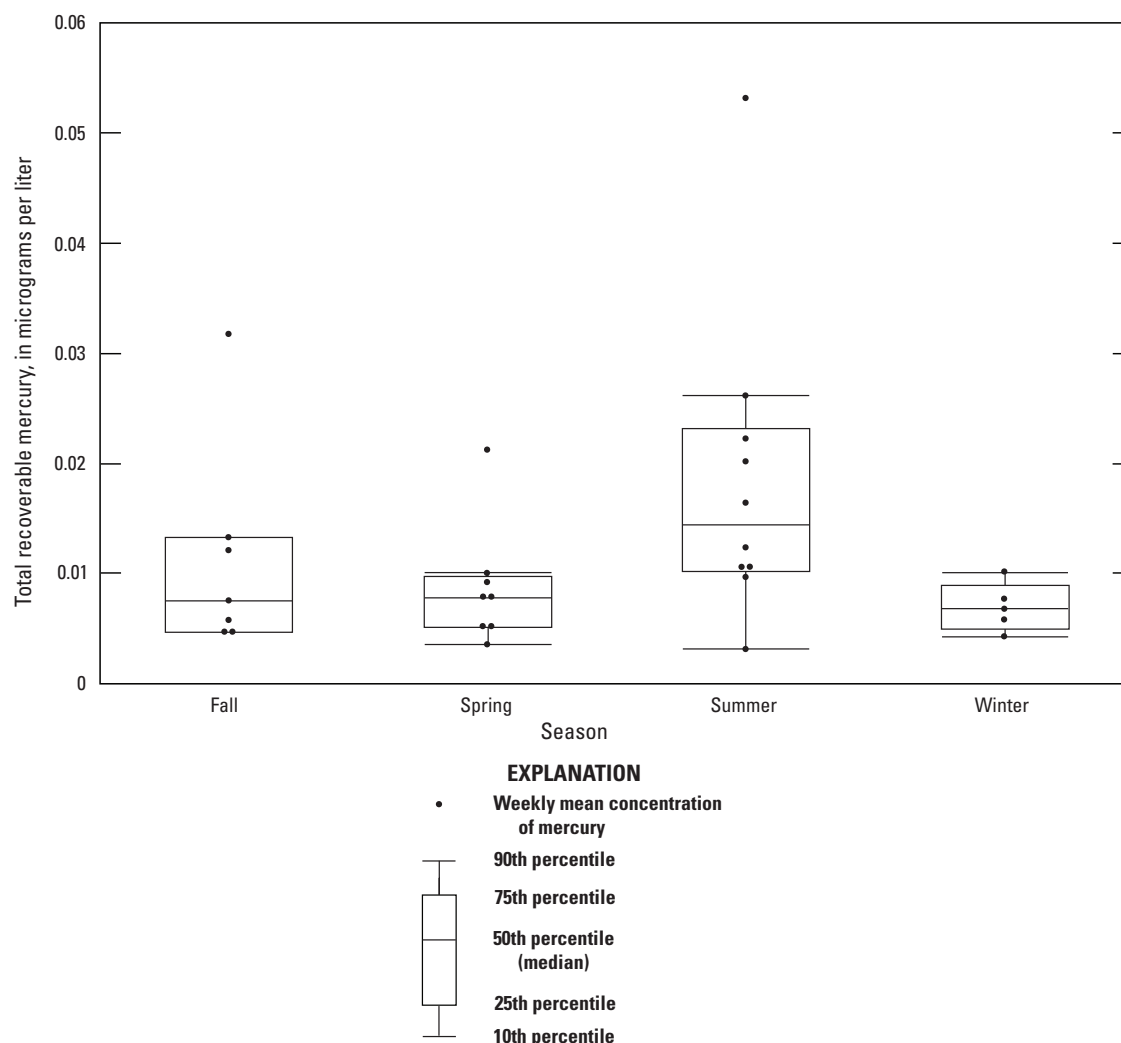


Figure 12. Boxplot showing weekly average concentrations of total recoverable mercury in precipitation samples collected between 2007 and 2009 from the National Atmospheric Deposition Program site in Candor, North Carolina.

Comparisons of Mercury Concentrations with National and Benchmark Levels

Mercury concentrations in water samples in this study were below levels associated with health risks to humans and aquatic biota. FHg and TRHg in all water samples were well below the USEPA national drinking-water regulatory criterion for mercury of $2 \mu\text{g}\cdot\text{L}^{-1}$ (USEPA, 2009). All FHg and TRHg concentrations in water samples were also below the TRHg aquatic-life criteria concentrations of $1.4 \mu\text{g}\cdot\text{L}^{-1}$ for acute exposure and $0.77 \mu\text{g}\cdot\text{L}^{-1}$ for chronic exposure (USEPA, 1996; Buchman, 2008). Mercury concentrations below the aquatic-life criteria are expected to prevent adverse effects associated with mercury exposure to fish, invertebrates, and other aquatic biota (USEPA, 1996).

At least one water sample from about one-half of the reservoir sites had a TRHg concentration similar to those measured in streams and lakes classified as having a significant on-site anthropogenic source of mercury.

TRHg concentrations in water collected during the last several decades of the 20th century from lakes and streams across the United States in areas with and without on-site significant anthropogenic sources (for example, mercury mines or industrial pollution) typically ranged from 0.1 to $0.4 \mu\text{g}\cdot\text{L}^{-1}$ and from 0.003 to $0.08 \mu\text{g}\cdot\text{L}^{-1}$, respectively (Wiener and others, 2003). TRHg mercury concentrations at the stream sites in this study all fell within the range for sites without significant anthropogenic sources; however, a total of 12 water samples from eight reservoir sites had TRHg concentrations in or above the range of those affected by anthropogenic sources. Six of these eight sites with relatively high TRHg concentrations in water samples were from Falls Lake and Jordan Lake. A study of mercury in the Sandy Creek catchment, which flows into Jordan Lake, suggested that the application of mercury-based fungicides to athletic fields at Duke University was a legacy source of mercury in the catchment (Deonarine and others, 2015). A concentration of $0.007 \mu\text{g}\cdot\text{L}^{-1}$ TRHg was measured in water samples from the most downstream

site on Sandy Creek, suggesting that this tributary was not a large source of mercury to Jordan Lake. They also suggested, however, that Sandy Creek may be a source of chronically low levels of MeHg that could bioaccumulate in fishes in Jordan Lake.

Background levels of mercury contamination refer to natural geochemical concentrations that are unaffected by human activity (Wentz and others, 2014). Background levels of mercury in freshwater sediment range between <0.01 and $0.05 \text{ mg}\cdot\text{kg}^{-1}$ (Buchman, 2008; Horowitz and Stephens, 2008). Measured and bulk-calculated concentrations of THg in reservoir-bed sediment in this study generally exceeded $0.05 \text{ mg}\cdot\text{kg}^{-1}$, a value that suggests that anthropogenic sources of mercury, such as atmospheric deposition, contribute to levels of mercury in bed sediment in Triangle-Area reservoirs. Total digestion was used to measure bed sediment THg concentrations in this study, indicating that mercury bound to the mineral matrix may have been included in the measurements as well as sorbed mercury. Horowitz and Stephens (2008) suggested that, relative to anthropogenic activities, the underlying rock type is unlikely to substantially influence sediment-associated mercury concentrations. The use of total digestion for measuring bed sediment mercury concentrations may have led to higher measurements than would functionally be available if geologic mercury bound to the matrix had been included in the measurement.

In contrast to background levels of a contaminant, baseline levels refer to concentrations of a contaminant for a specific area over a specific period of time and usually have some degree of anthropogenic influence (for example, atmospheric deposition; Park and others, 2008; Wentz and others, 2014). Samples collected by the USGS from 1991 to approximately 1999 indicated that baseline sediment THg concentrations ranged from 0.02 to $0.06 \text{ mg}\cdot\text{kg}^{-1}$ for stream sites across the United States in catchments with population densities below the 50th percentile and areas with less than 5 percent urban land-use (Horowitz and Stephens, 2008; Wentz and others, 2014). Four stream-bed sediment samples in this study had measured THg concentrations that exceeded the baseline range of 0.02 – $0.06 \text{ mg}\cdot\text{kg}^{-1}$ for stream sites across the United States (Horowitz and Stephens, 2008). Another national-scale USGS survey of THg in stream-bed sediment conducted between 1998 and 2005 found a mean of $0.09 \text{ mg}\cdot\text{kg}^{-1}$ and median of $0.03 \text{ mg}\cdot\text{kg}^{-1}$ for unmined basins (Scudder and others, 2009). The mean and median measured THg concentrations for streams in this study were $0.11 \text{ mg}\cdot\text{kg}^{-1}$ and $0.05 \text{ mg}\cdot\text{kg}^{-1}$, respectively. This indicates elevated levels of mercury in stream-bed sediment in the Triangle Area relative to levels in unmined basins during the time interval 1998–2005 (Scudder and others, 2009). Atmospheric-deposition rates of mercury are greater in the eastern half of the continental United States (east of the 98th meridian) compared to the western half (Shacklette and Boerngen, 1984; Gustavsson and others, 2001; Wentz and others, 2014). High rates of atmospheric deposition in the eastern United States could be one explanation for elevated THg concentrations in stream-bed sediment from the study area relative to concentrations at the national scale.

Adverse outcomes to sediment-dwelling organisms from exposure to elevated mercury levels in bed sediment were generally unlikely in the study area. Relative to benchmark standards indicative of health risks to aquatic biota, all bed sediment bulk-calculated mercury concentrations were below the probable-effect concentration of $1.06 \text{ mg}\cdot\text{kg}^{-1}$, the concentration at which adverse effects to sediment-dwelling organisms are expected to occur more often than not (MacDonald and others, 2000). All bed sediment bulk-calculated mercury concentrations were also below the probable effect level of $0.486 \text{ mg}\cdot\text{kg}^{-1}$, which is the concentration frequently associated with adverse effects to aquatic biota (Smith and others, 1996). Only one stream site had a bulk-calculated bed sediment THg above both the commonly used threshold for regulation in the United States of $0.18 \text{ mg}\cdot\text{kg}^{-1}$ (USEPA, 2010) and the lowest effect level concentration of $0.20 \text{ mg}\cdot\text{kg}^{-1}$ (the concentration at which ecotoxic effects become apparent; Canadian Ministry of Environment and Energy, 1993).

Conclusions

Seasonal trends and associations between potential driver variables and mercury concentrations suggest that atmospheric deposition is a primary source of mercury to reservoirs from both direct and indirect input from surface runoff, and that litterfall might be a primary source of mercury to streams in the Triangle Area. High concentrations of FHg and TRHg in the fall in some reservoir-water samples may warrant further analysis to determine potential hydrologic drivers of mercury concentrations in water. None of the analyses of water or bed sediment samples in this study indicated a severe or immediate risk to humans or aquatic biota; however, TRHg concentrations in surface-reservoir samples from the entire study area frequently exceeded the NCWQC for mercury. FHg concentrations, which served as conservative estimates of TRHg to control for the possibility of errors in TRHg measurements, exceeded the NCWQC for mercury in 25 percent of all reservoir-surface samples. Most of the water samples with TRHg concentrations in the general range for sites across the United States with on-site significant anthropogenic sources were from Jordan Lake and Falls Lake. These results might suggest that continued monitoring of these reservoirs is warranted. Mercury concentrations in stream-water and sediment samples were generally low relative to reservoir sites, and all stream TRHg concentrations were within the general range found across the United States for sites not affected by on-site mercury mines or industrial-pollution sources. The experimental design of the study, however, may have resulted in negative bias in overall mean and median FHg and TRHg concentrations of stream-water samples relative to overall mean and median FHg and TRHg concentrations in reservoir-water samples (or positive bias in reservoir-water samples relative to stream-water samples). Mercury concentrations in bed sediment from streams and reservoirs indicated that levels were

elevated relative to baseline national levels. Higher rates of atmospheric deposition in the eastern half of the United States could be the cause of elevated THg levels in bed sediment in the study area relative to national baseline levels. Only one bed sediment sample had a bulk-calculated THg concentration above the commonly used guideline of $0.18 \text{ mg} \cdot \text{kg}^{-1}$ in the United States. Future monitoring of bed sediment THg concentrations at this stream site with an anomalous THg concentrations may be warranted.

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Appendix 1. Randomly Generated Numbers Below the Method Detection Level for Use in Statistical Analysis

Table 1.1. Randomly generated numbers below the method detection level for use in statistical analysis.

[FHg, filtered mercury; TRHg, total recoverable mercury; >, greater than; MDL, method detection level]

Site	Collection date	FHg	TRHg	Site	Collection date	FHg	TRHg
1	Oct. 2, 2007	0	0.004	6	June 3, 2008	0	0.001
1	Dec. 3, 2007	0.001	>MDL	6	Aug. 26, 2008	0	0.004
1	Feb. 6, 2008	0.001	0.003	6	Aug. 26, 2008	0.001	0.002
1	Apr. 2, 2008	0.001	0.002	6	June 17, 2009	0.002	>MDL
1	June 2, 2008	0.005	0.002	7	Aug. 28, 2007	0	0.005
1	Aug. 4, 2008	0.003	0	7	Aug. 28, 2007	0.004	>MDL
1	Oct. 17, 2008	0.005	0.001	7	Oct. 18, 2007	0.004	0
1	Dec. 10, 2008	0.002	0	7	Oct. 18, 2007	0	0.001
1	Feb. 12, 2009	0.005	0.001	7	June 16, 2008	0.004	>MDL
1	Apr. 27, 2009	0.001	0	7	Aug. 12, 2008	0.002	>MDL
1	June 8, 2009	0.001	0.002	7	Oct. 3, 2008	0.001	0.004
2	Jan. 7, 2009	0	>MDL	7	Oct. 3, 2008	0.004	0.003
2	Mar. 1, 2009	0.002	0.002	7	Apr. 1, 2009	0.001	0
2	May 27, 2009	0.002	0.005	7	Apr. 1, 2009	0.002	0.003
3	Sept. 6, 2008	0.005	>MDL	7	June 26, 2009	0.001	0
3	Mar. 1, 2009	0	0.003	7	June 26, 2009	0.004	0.001
3	May 5, 2009	0	0.005	8	Apr. 1, 2008	0.004	>MDL
3	May 19, 2009	0.004	0.005	8	Dec. 12, 2008	0.003	>MDL
3	June 5, 2009	0.005	>MDL	8	May 27, 2009	0	0.003
3	June 23, 2009	0.003	0.001	9	Sept. 6, 2008	0.002	>MDL
4	Aug. 28, 2007	0	0	9	Sept. 26, 2008	0.004	>MDL
4	Oct. 30, 2007	0.002	>MDL	9	Jan. 6, 2009	0.002	>MDL
4	Apr. 24, 2008	0.002	>MDL	9	May 19, 2009	0.005	0
4	June 3, 2008	0.003	0.004	10	Feb. 13, 2008	0.003	>MDL
4	June 3, 2008	0	0	10	June 17, 2008	0.003	>MDL
4	Aug. 26, 2008	0.001	0	10	Aug. 20, 2008	0.001	>MDL
4	Aug. 26, 2008	0	0.002	10	Aug. 20, 2008	0.005	>MDL
4	Oct. 23, 2008	0.001	0.003	10	Oct. 14, 2008	0.001	>MDL
4	Apr. 30, 2009	0.002	0.004	10	Feb. 5, 2009	0	>MDL
4	Apr. 30, 2009	0.002	0.001	10	Apr. 15, 2009	0.004	>MDL
4	June 17, 2009	0.004	>MDL	10	Apr. 15, 2009	0	>MDL
5	Sept. 6, 2008	0.002	>MDL	10	June 23, 2009	0.003	0.001
5	May 12, 2009	0.001	0.004	10	June 23, 2009	0.002	0.005
6	Aug. 28, 2007	>MDL	0	11	Sept. 26, 2008	0.003	>MDL
6	Apr. 24, 2008	0.004	>MDL	11	Feb. 18, 2009	0.001	0.004
6	June 3, 2008	0	0.001	11	May 12, 2009	0.004	0.002

Table 1.1. Randomly generated numbers below the method detection level for use in statistical analysis.—Continued

[FHg, filtered mercury; TRHg, total recoverable mercury; >, greater than; MDL, method detection level]

Site	Collection date	FHg	TRHg	Site	Collection date	FHg	TRHg
12	Aug. 13, 2007	0.001	>MDL	14	Dec. 4, 2008	0.003	0.003
12	Feb. 11, 2008	0.001	>MDL	14	Feb. 6, 2009	0.004	0
12	Feb. 11, 2008	0	0	14	Feb. 6, 2009	0.003	0
12	June 17, 2008	0.004	0.003	14	Apr. 9, 2009	0.004	>MDL
12	June 17, 2008	0.001	0.004	14	June 23, 2009	0.003	>MDL
12	Aug. 20, 2008	0.001	>MDL	15	Aug. 29, 2007	0.003	>MDL
12	Oct. 14, 2008	0.001	>MDL	15	Aug. 29, 2007	0.004	>MDL
12	Oct. 14, 2008	0.004	>MDL	15	Nov. 1, 2007	0	0
12	Dec. 5, 2008	0	0	15	Nov. 1, 2007	0.003	0.003
12	Dec. 5, 2008	0	0.004	15	Apr. 28, 2008	0.001	0.001
12	Feb. 4, 2009	0.004	>MDL	15	Apr. 28, 2008	0.005	0
12	Feb. 4, 2009	0.003	>MDL	15	June 25, 2008	0.003	0.004
12	Apr. 14, 2009	0.001	>MDL	15	June 25, 2008	0.005	0.004
12	June 22, 2009	0.004	>MDL	15	Aug. 21, 2008	0.002	0.003
13	Aug. 14, 2007	0	0.004	15	Aug. 21, 2008	0.001	0
13	Aug. 14, 2007	0	0.002	15	Oct. 21, 2008	0	0.004
13	Oct. 4, 2007	0.004	0.004	15	Apr. 16, 2009	0.004	>MDL
13	Dec. 12, 2007	0.004	0.004	15	Apr. 16, 2009	0.002	>MDL
13	Dec. 12, 2007	0.004	0.002	15	June 19, 2009	0	>MDL
13	Feb. 11, 2008	0.001	0.005	16	Aug. 29, 2007	0.004	>MDL
13	Feb. 11, 2008	0.001	0.004	16	Aug. 29, 2007	0.001	>MDL
13	Apr. 9, 2008	0	0.005	16	Nov. 1, 2007	0.005	>MDL
13	Apr. 9, 2008	0.002	0.004	16	Apr. 28, 2008	0.004	>MDL
13	June 18, 2008	0.001	>MDL	16	Aug. 21, 2008	0.003	0.005
13	Aug. 19, 2008	0.001	>MDL	16	Aug. 21, 2008	0.004	0.004
13	Dec. 4, 2008	0.002	0.005	16	May 1, 2009	0	>MDL
13	Dec. 4, 2008	0.004	0	16	May 1, 2009	0.001	>MDL
13	Apr. 10, 2009	0.003	0.003	16	June 25, 2009	0.002	>MDL
13	Apr. 10, 2009	0.002	0.004	17	Aug. 20, 2007	0.004	0
13	June 23, 2009	0.004	>MDL	17	Feb. 6, 2008	0.003	0.003
14	Aug. 14, 2007	0	0.002	17	June 2, 2008	0.003	0
14	Aug. 14, 2007	0.005	0.004	17	Aug. 4, 2008	0.003	0.004
14	Oct. 4, 2007	0.003	>MDL	17	Oct. 17, 2008	0.004	0
14	Oct. 4, 2007	0.002	>MDL	17	Dec. 10, 2008	0.002	0.001
14	Dec. 12, 2007	0	0.005	17	Feb. 12, 2009	0.001	0.004
14	Dec. 12, 2007	0.001	0.002	17	Apr. 27, 2009	0.001	0.004
14	Feb. 11, 2008	0.004	0.005	17	June 15, 2009	0	0.001
14	Feb. 11, 2008	0.001	0.005	18	Aug. 31, 2007	0	>MDL
14	Apr. 9, 2008	0.003	0.004	18	Oct. 5, 2007	0.003	0.003
14	Apr. 9, 2008	0	0.005	18	Oct. 5, 2007	0.002	0.004
14	June 18, 2008	0.002	>MDL	18	Apr. 21, 2008	0.001	0
14	Oct. 15, 2008	0	>MDL	18	Apr. 21, 2008	0.003	0.002
14	Dec. 4, 2008	0.005	0.003	18	June 26, 2008	0	>MDL

Table 1.1. Randomly generated numbers below the method detection level for use in statistical analysis.—Continued

[FHg, filtered mercury; TRHg, total recoverable mercury; >, greater than; MDL, method detection level]

Site	Collection date	FHg	TRHg	Site	Collection date	FHg	TRHg
18	Aug. 4, 2008	0.001	0.001	23	Aug. 1, 2008	0.003	0.001
18	Aug. 4, 2008	0.003	0	23	Apr. 17, 2009	0.002	>MDL
18	Oct. 20, 2008	0.002	0.002	23	June 29, 2009	0	0.005
18	Oct. 20, 2008	0	0.005	23	June 29, 2009	0.002	0.005
18	Apr. 21, 2009	0	0.004	24	May 28, 2009	0.003	>MDL
18	June 16, 2009	0.002	0	25	Oct. 17, 2007	0.004	>MDL
18	June 16, 2009	0.001	0.002	25	Oct. 17, 2007	0.005	>MDL
19	Oct. 16, 2007	0.004	0.005	25	Dec. 6, 2007	0.004	0.004
19	Oct. 16, 2007	0.003	0.004	25	Dec. 6, 2007	0.001	0.005
19	Dec. 6, 2007	0.003	>MDL	25	Feb. 26, 2008	0.005	0.005
19	Apr. 22, 2008	0.004	>MDL	25	Feb. 26, 2008	0.003	0
19	June 19, 2008	0.004	0.003	25	Apr. 23, 2008	0.002	>MDL
19	June 19, 2008	0.003	0.005	25	Apr. 23, 2008	0.005	>MDL
19	Dec. 15, 2008	0.002	>MDL	25	June 19, 2008	0.002	0.003
19	Dec. 15, 2008	0	0.001	25	June 19, 2008	0.002	0.002
19	Feb. 10, 2009	0.003	0.003	25	Aug. 6, 2008	0.005	>MDL
19	Feb. 10, 2009	0	0	25	Aug. 6, 2008	0.001	>MDL
19	Apr. 22, 2009	0.005	>MDL	25	Oct. 22, 2008	0.004	0.003
19	June 30, 2009	0.005	>MDL	25	Oct. 22, 2008	0.005	0.005
20	Jan. 7, 2009	0.003	0.001	25	Feb. 10, 2009	0.001	0.005
20	May 28, 2009	0.002	>MDL	25	Feb. 10, 2009	0.004	0.005
21	Apr. 29, 2008	0.002	>MDL	25	Apr. 20, 2009	0.002	>MDL
21	Dec. 12, 2008	0.002	>MDL	25	Apr. 20, 2009	0.003	>MDL
21	Mar. 1, 2009	0.003	0.004	25	June 30, 2009	0.004	0
21	May 28, 2009	0.003	0.005	25	June 30, 2009	0.001	0.005
22	Aug. 31, 2007	0	0.004	26	Aug. 20, 2007	0.003	0.003
22	Feb. 8, 2008	>MDL	0.004	26	Oct. 2, 2007	0.003	0.004
22	Apr. 11, 2008	0.002	0.003	26	Feb. 6, 2008	0.003	0.005
22	June 27, 2008	0.002	0.001	26	Apr. 2, 2008	0.003	>MDL
22	Aug. 1, 2008	0.001	0.004	26	June 2, 2008	0.004	0.005
22	Oct. 20, 2008	0	0	26	Aug. 4, 2008	0.005	0.002
22	Dec. 10, 2008	0.005	0.003	26	Oct. 17, 2008	0.002	0.004
22	Feb. 12, 2009	0.002	0.001	26	Dec. 10, 2008	0.004	0.005
22	Apr. 28, 2009	0.003	0.001	26	Feb. 12, 2009	0.001	0.005
22	June 15, 2009	0.002	0.001	26	June 8, 2009	0.001	0.004
23	Aug. 31, 2007	0.003	>MDL	27	Dec. 5, 2007	0.004	>MDL
23	Aug. 31, 2007	0.005	>MDL	27	Apr. 23, 2008	0	0.001
23	Oct. 5, 2007	0.004	>MDL	27	Apr. 23, 2008	0.003	0.005
23	Apr. 11, 2008	0.001	>MDL	27	June 20, 2008	0.001	0.005
23	Apr. 11, 2008	0	0.004	27	June 20, 2008	0.005	0.004
23	June 27, 2008	0.005	0.003	27	Oct. 22, 2008	0.001	0.004
23	June 27, 2008	0.003	0.003	27	Oct. 22, 2008	0.005	0.001
23	Aug. 1, 2008	0.001	0	27	Dec. 17, 2008	0.002	0

Table 1.1. Randomly generated numbers below the method detection level for use in statistical analysis.—Continued

[FHg, filtered mercury; TRHg, total recoverable mercury; >, greater than; MDL, method detection level]

Site	Collection date	FHg	TRHg
27	Dec. 17, 2008	0.004	0.002
27	Feb. 11, 2009	0.002	0.004
27	Apr. 23, 2009	0.001	>MDL
27	Apr. 23, 2009	0	0.005
28	Oct. 17, 2007	0.001	>MDL
28	Dec. 5, 2007	0.003	0.004
28	Dec. 5, 2007	0.002	0.003
28	Feb. 27, 2008	0.002	>MDL
28	Apr. 23, 2008	0.003	>MDL
28	June 20, 2008	0.003	>MDL
28	Dec. 17, 2008	0.004	>MDL
28	Dec. 17, 2008	0.002	0.004
28	Feb. 11, 2009	0.003	>MDL
28	Feb. 11, 2009	0.004	0.004
28	Apr. 23, 2009	0.002	0
28	Apr. 23, 2009	0.003	0.005
28	June 29, 2009	0.004	>MDL
29	Apr. 29, 2008	0.005	0.002

Appendix 2. Accuracy of Bed Sediment Total Mercury and Total Organic-Carbon Analyses Determined for Reference Materials

Table 2.1. Accuracy of bed sediment total mercury and total organic carbon analyses determined for reference materials.

[Data are from Fitzgerald (2020). NIST, National Institute of Standards and Technology; NRC, Nuclear Regulatory Commission; mg·kg⁻¹, milligram per kilogram; ±, plus or minus; —, not determined; wt. %, weight percent; <, less than; RSD, relative standard deviation]

Reference material	Reported	Found	Absolute difference	RSD (percent)
Total mercury (mg·kg ⁻¹)				
NIST 2709	1.40±0.08	1.40	0	0
USGS SGR-1B	0.3	0.23	0.07	26
USGS SDO-1	0.19±0.08	0.16	0.03	17
NRC MESS-3	0.09±0.01	0.10	0.01	11
USGS SCO-1	0.05	0.07	0.02	33
USGS QLO-1	0.01	0.02	0.01	67
Minimum			0	0
Median			0.02	22
Maximum			0.07	67
Total organic carbon (wt. %)				
USGS SGR-1B	28	26	2	7
USGS SDO-1	9.95±0.44	9.7	0.15	2.5
NRC MESS-3	2	2.0	0	0.0
NIST 2709	1.2	1.0	0.2	18
USGS SCO-1	0.8±0.1	0.9	0.1	12
USGS QLO-1	<0.1	<0.1	—	—
USGS GSP-2	<0.1	<0.1	—	—
Minimum			0	0
Median			0.15	7.4
Maximum			2.0	18

Appendix 3. Precision of Constituent Analyses in Replicate Water Samples

Table 3.1. Precision of constituent analyses in replicate water samples.

[RPD, relative percent difference; RSD, relative standard deviation; $\mu\text{g}\cdot\text{L}^{-1}$, microgram per liter; $\text{mg}\cdot\text{L}^{-1}$, milligram per liter, NA, not applicable]

Constituent	Total number of sets	Number of sets with all concentrations below the detection level	RPD or RSD* (percent)			Unit	Absolute difference		
			Minimum	Median	Maximum		Minimum	Median	Maximum
Sulfate (SO_4^{2-})	13	NA	0	0.6	5.8	$\text{mg}\cdot\text{L}^{-1}$	0	0.04	0.30
Mercury, filterable (FHg)	13	12	0	0	4.4	$\mu\text{g}\cdot\text{L}^{-1}$	0	0	0.001
Mercury, total recoverable (TRHg)	13	8	0	0	145.5	$\mu\text{g}\cdot\text{L}^{-1}$	0	0	0.032
Chlorophyll <i>a</i> (Chloro <i>a</i>)	28	NA	0	7.0	33.6	$\mu\text{g}\cdot\text{L}^{-1}$	0	1.3	13.5
Pheophytin <i>a</i> (Pheo <i>a</i>)	28	NA	0	7.4	45.8	$\mu\text{g}\cdot\text{L}^{-1}$	0	0.9	8.5
Organic carbon, filterable (dissolved, DOC)	13	NA	0	1.3	6.3	$\text{mg}\cdot\text{L}^{-1}$	0	0.08	0.70
Organic carbon, unfilterable (total, $\text{TOC}_{\text{water}}$)	13	NA	0	1.3	10.3	$\text{mg}\cdot\text{L}^{-1}$	0	0.1	1.5
Suspended-sediment concentration (SCC)	3	NA	13	18	40.0	$\text{mg}\cdot\text{L}^{-1}$	1	1	2

*RPD was calculated for two replicates, and RSD was calculated for three replicates.

For more information about this publication, contact:
Director, South Atlantic Water Science Center
U.S. Geological Survey
1770 Corporate Drive, Suite 500
Norcross, GA 30093

For additional information visit <https://www.usgs.gov/centers/sa-water>

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