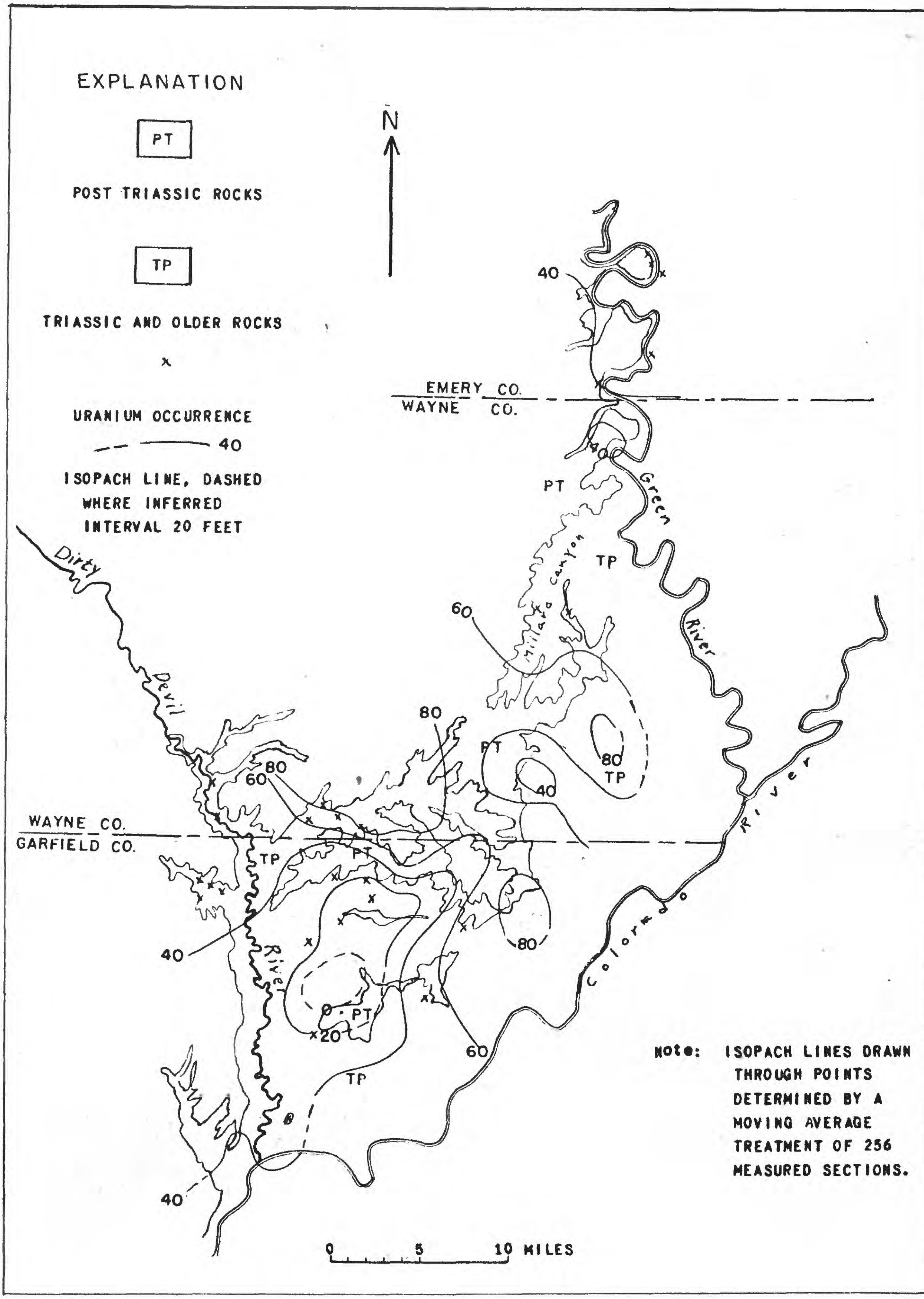



FIG. 21 .--ISOPACH MAP OF THE MOSS BACK MEMBER OF THE CHINLE FORMATION IN THE ORANGE CLIFFS AREA, EMERY, WAYNE AND GARFIELD COUNTIES, UTAH.

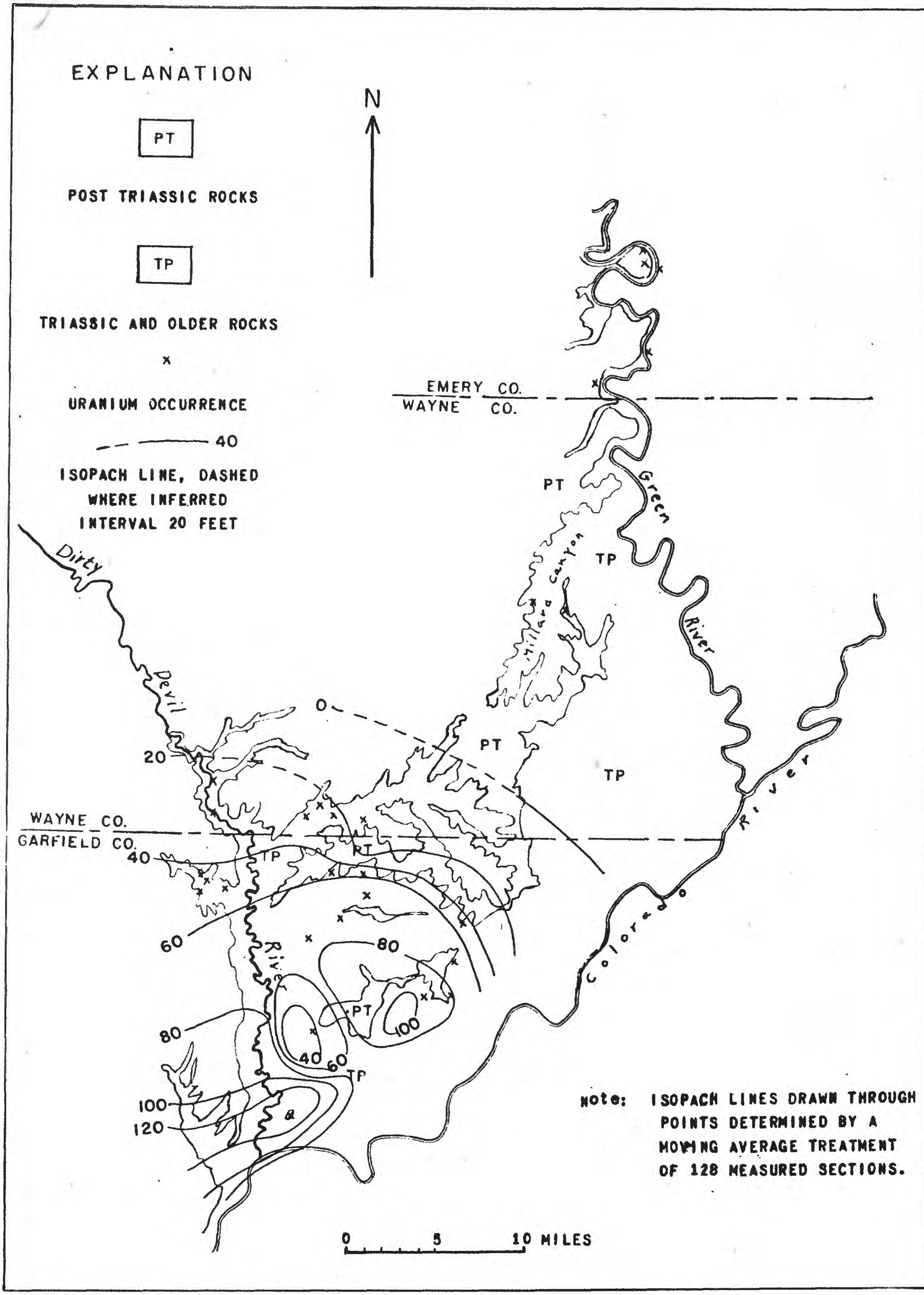



FIG. 22.--ISOPACH MAP OF THE MONITOR BUTTE MEMBER OF THE CHINLE FORMATION IN THE ORANGE CLIFFS AREA, EMERY, WAYNE AND GARFIELD CO NTIES, TAH.

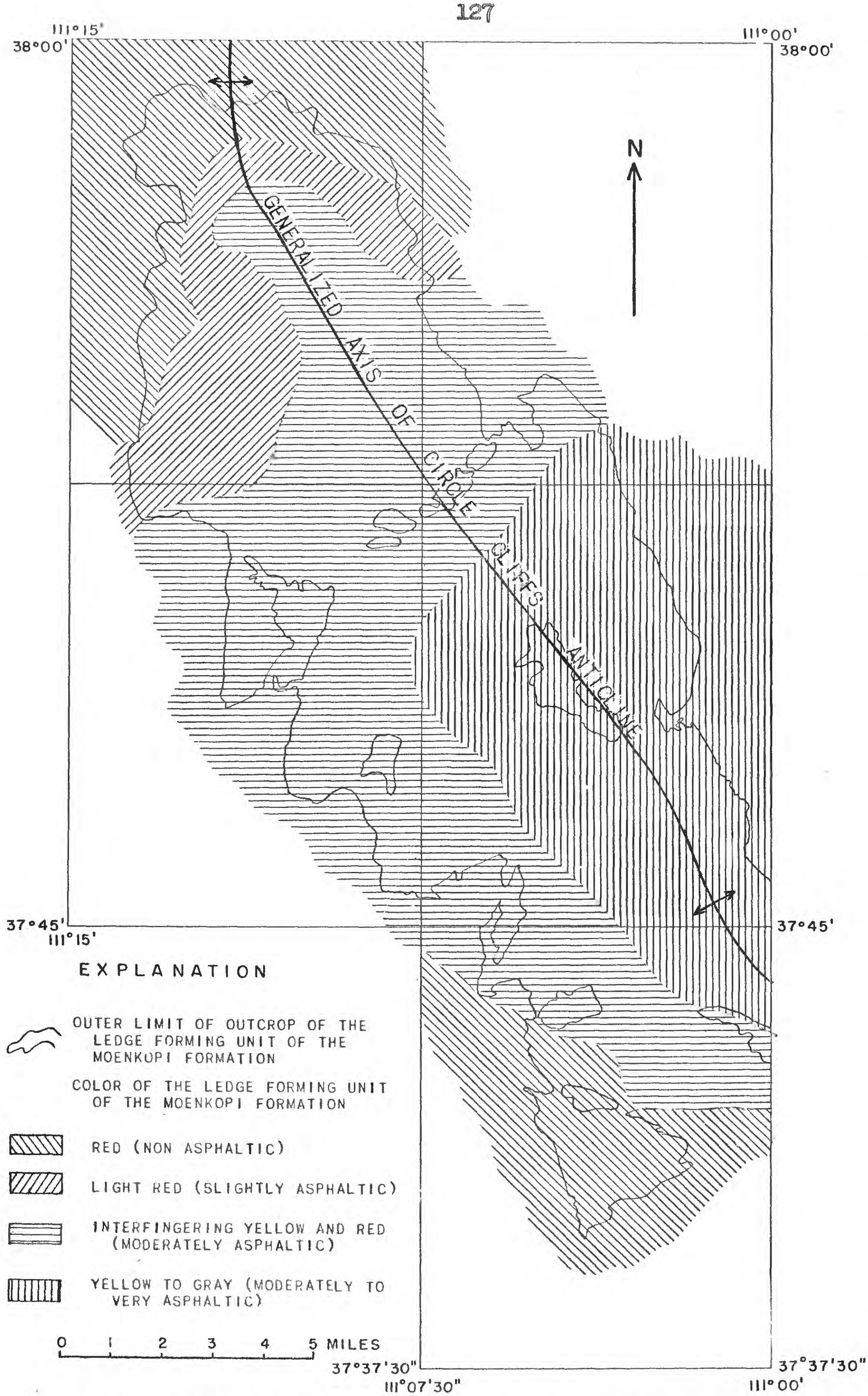



Fig. 25.--DISTRIBUTION OF ASPHALTIC ROCK IN THE LEDGE FORMING UNIT OF THE MOENKOPI FORMATION, CIRCLE CLIFFS AREA, GARFIELD COUNTY, UTAH.

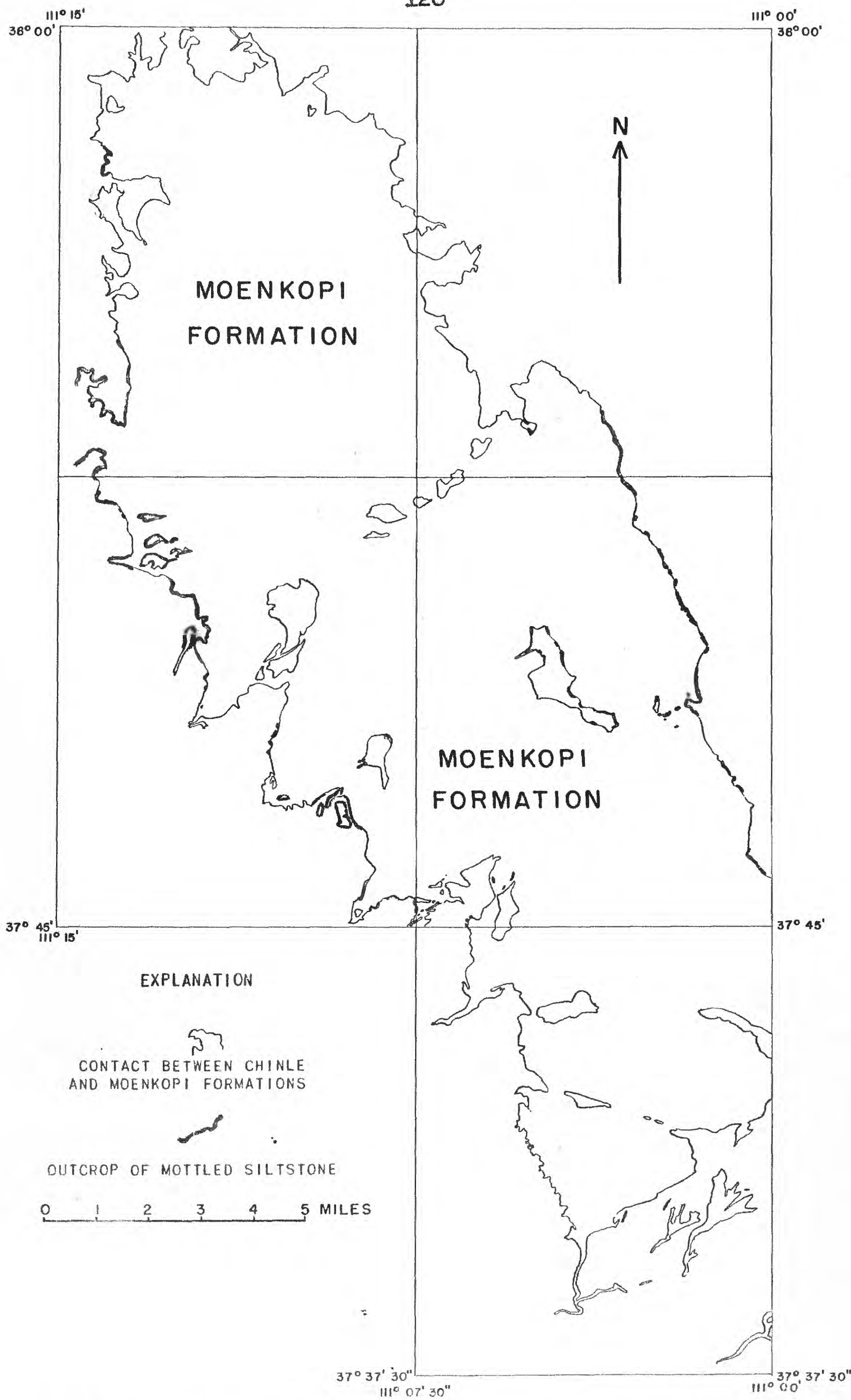



Fig. 26.--DISTRIBUTION OF MOTTLED SILTSTONE UNIT AT THE BASE OF THE CHINLE FORMATION, CIRCLE CLIFFS AREA, GARFIELD COUNTY, UTAH.

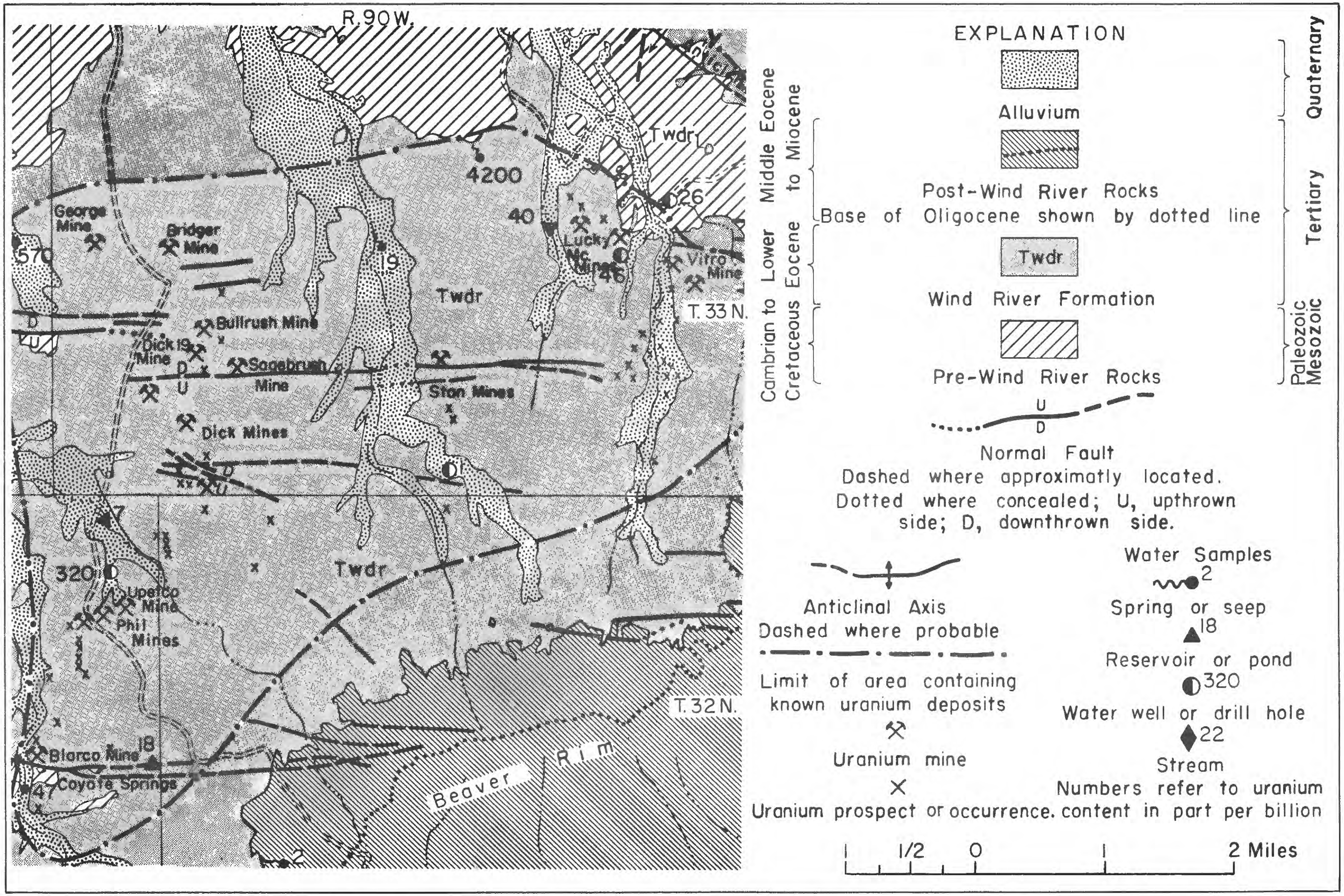
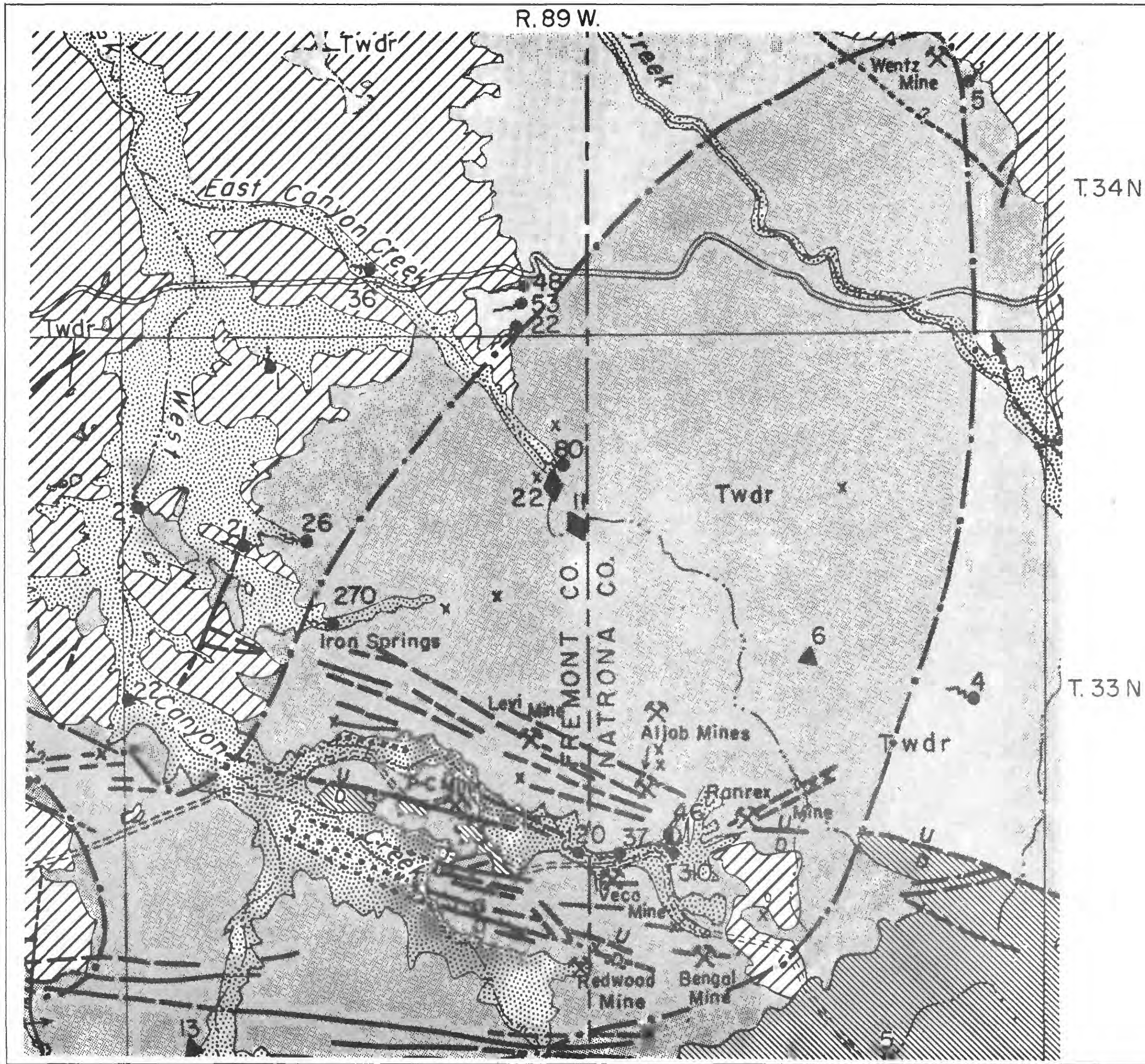



FIG. 66-GENERALIZED GEOLOGIC MAP, WESTERN GAS HILLS AREA




FOR EXPLANATION
SEE FIG. 66

FIG. 67–GENERALIZED GEOLOGIC MAP, EASTERN GAS HILLS AREA

1953, p. 91).

Reference

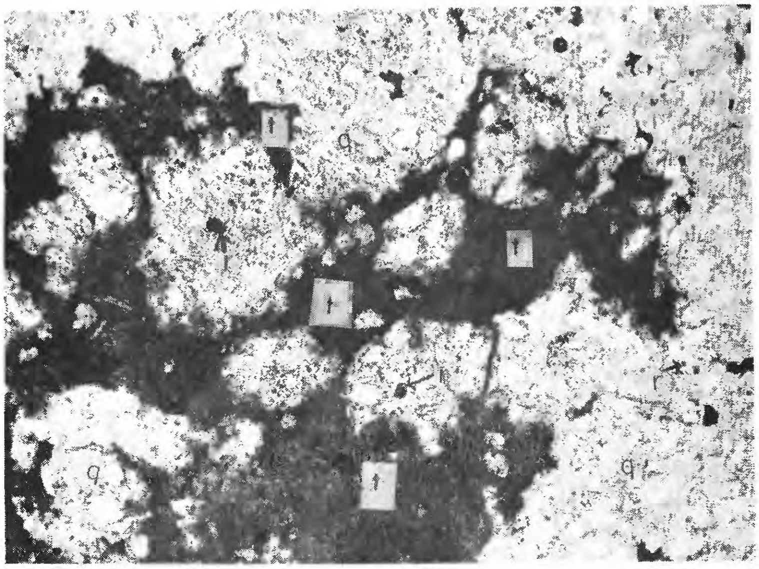
Everhart, D. L. and Wright, R. J., 1953, The geologic character of typical pitchblende veins: Econ. Geology, v. 48, p. 77-96.

Uranium minerals in a silicified log from Monument No. 2 mine, Apache County, Ariz., by T. L. Finnell

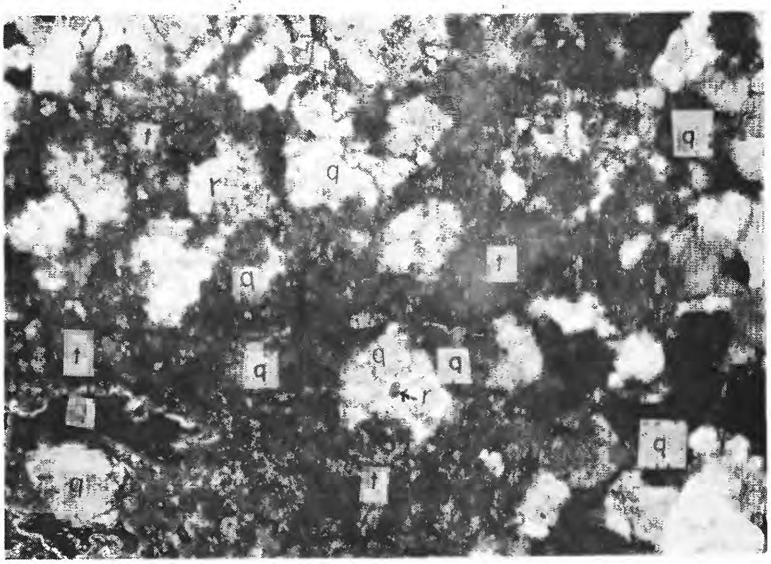
An occurrence of secondary uranium minerals along the rim of a 2-foot diameter silicified log from the north workings of the Monument No. 2 mine was sectioned and studied because it seemed to be an impregnation of fine- to medium-grained sandstone that penetrates the log as much as one inch along cracks. A thin section normal to the cell structure of the log (fig. 105, A) shows preservation of the wood cell structure even though the log is almost completely replaced by silica. Under cross nicols, the relict cell structure is partly obscured (fig. 105, B) because the anhedral to subhedral quartz grains replaced the log without apparent relation to the size and shape of the wood cells.

Secondary uranium minerals penetrate the edges of the log along the quartz crystal boundaries, and completely surround some quartz crystals. If the secondary uranium minerals had penetrated all of the silicified log in the same manner, the resultant effects could easily be misinterpreted as an impregnation of sandstone by the uranium minerals. Ore textures of uranium deposits in sandstone involving impregnation of quartz grains by uranium minerals should be examined with the thought that the quartz is not necessarily detrital and that it may represent silicified wood.

Secondary uranium minerals occupy two separate and distinct positions in the sections. Rauvite(?) (deep orange, high index of refraction) occupies



A. Wood cell structures outlined by minute inclusions in quartz (q). Rauvite(?) (r) occupies some cell pores and forms relicts in tyuyamunite (t) that penetrates the quartz. Plain light, X104.



B. Same as A. Polarized light. Crystals of quartz are not obviously related to the wood cell structure. Tyuyamunite and rauvite(?) penetrate the log along quartz crystal boundaries. X104.

FIG. 105 .--PHOTOMICROGRAPHS OF SILICIFIED LOG PARTLY REPLACED BY RAUVITE(?) AND TYUYAMUNITE.

(including the Moss Back member as presently defined, Stewart, 1957) and the Salt Wash member of the Morrison formation of Late Jurassic age. The estimated arithmetic mean compositions of sandstones from the two units are shown on table 30.

Sandstones of the Shinarump member are estimated to contain about twice as much titanium, zirconium, manganese, copper, chromium, and vanadium as sandstones of the Salt Wash member, about three times as much aluminum, iron, strontium, and boron, and about 10 times as much nickel and cobalt. The uranium contents of both units are similar, but sandstones of the Shinarump appear to contain more yttrium, ytterbium, arsenic, zinc, and lead. Part of the difference in the chemical compositions of these two rock units may be due to the somewhat tuffaceous and arkosic nature of sandstones of the Shinarump member, whereas sandstones of the Salt Wash member are well sorted or relatively clean quartzose sandstones commonly cemented by calcite.

Most of the uranium deposits of the Morrison formation occur in uppermost sandstone strata of the Salt Wash member; however, except for potassium and calcium no significant chemical differences between sandstones of the uppermost strata and intermediate and basal strata were detected in the samples studied. Sandstones of the uppermost strata contain on the average less potassium and calcium than sandstones lower in the section. Copper and vanadium occur in abnormally high amounts in unmineralized sandstone of the Salt Wash member in an area that coincides roughly with the Uravan mineral belt (TEI-490, p. 59-60). The distribution of elements commonly associated with igneous or crystalline rocks indicates three possible secondary source areas for detrital minerals in sandstone of the

Table 30. Estimated mean compositions of sandstone of the Shinarump member of the Chinle formation and sandstone of the Salt Wash member of the Morrison formation. (Composition shown in parts per million).

| Element | Sandstone of the Shinarump member1/ G.M.3/ | G.D.4/ | A.M.5/ | Sandstone of the Salt Wash member2/ G.M.3/ | G.D.4/ | A.M.5/ |
|---|---|---|---|---|---|---|
| Si | >100,000 | | | >100,000 | | |
| Al | 32,000 $\times\!\div$ 1.25 | 2.28 | 45,000 | 11,900 $\times\!\div$ 1.18 | 1.89 | 14,600 |
| Fe | 6,400 $\times\!\div$ 1.28 | 2.46 | 9,700 | 2,400 $\times\!\div$ 1.19 | 1.91 | 3,000 |
| Mg | 1,800 $\times\!\div$ 1.39 | 3.40 | 3,700 | 2,300 $\times\!\div$ 1.33 | 2.85 | 4,000 |
| Ca | 6,800 $\times\!\div$ 1.69 | 7.16 | 45,000 | 33,000 $\times\!\div$ 1.40 | 3.47 | 72,000 |
| Na | 790 $\times\!\div$ 1.46 | 4.04 | 2,100 | 890 $\times\!\div$ 1.42 | 3.73 | 2,100 |
| K | 3,200 $\times\!\div$ 1.39 | 3.44 | 6,900 | ~3,000 | | |
| Ti | 920 $\times\!\div$ 1.26 | 2.39 | 1,300 | 510 $\times\!\div$ 1.20 | 1.96 | 640 |
| Zr | 120 $\times\!\div$ 1.34 | 2.94 | 220 | 100 $\times\!\div$ 1.27 | 2.40 | 150 |
| Mn | 170 $\times\!\div$ 1.65 | 6.48 | 950 | 220 $\times\!\div$ 1.33 | 2.89 | 380 |
| Ba | 450 $\times\!\div$ 1.38 | 3.32 | 920 | 340 $\times\!\div$ 1.34 | 3.00 | 630 |
| Sr | 82 $\times\!\div$ 1.46 | 4.10 | 220 | 49 $\times\!\div$ 1.28 | 2.50 | 70 |
| Cu | 20 $\times\!\div$ 1.40 | 3.45 | 43 | 13 $\times\!\div$ 1.28 | 2.49 | 20 |
| Cr | 13 $\times\!\div$ 1.31 | 2.74 | 21 | 7 $\times\!\div$ 1.25 | 2.27 | 9 |
| V | 22 $\times\!\div$ 1.34 | 2.98 | 40 | 10 $\times\!\div$ 1.32 | 2.81 | 18 |
| Y | 5 $\times\!\div$ 1.22 | 2.06 | 7 | < 5 | | |
| As | 13 $\times\!\div$ 1.20 | 1.91 | 17 | <10 | | |
| Zn | <20 | | | <20 | | |
| Se | | | <0.5 to 15 | | | <0.5 to 8 |
| U | 2.4 $\times\!\div$ 1.21 | 2.00 | 3 | 1.8 $\times\!\div$ 1.21 | 1.99 | 2 |

1/ Estimate based on 97 samples.

2/ Estimate based on 96 samples.

3/ Geometric mean showing the 99 percent confidence interval for the population geometric mean: the limits of the confidence interval are determined from Student's t distribution (Fisher and Yates, 1953, p. 1 and 40) where t is the deviation in units of estimated standard error.

For a lognormal distribution: Confidence interval of geometric mean $\equiv \times\!\div \text{antilog} \left[ t \; \frac{\text{Log G.D.}}{\sqrt{n-1}} \right]$

4/ Geometric deviation or antilog of the log standard deviation.

5/ Arithmetic mean. The most efficient estimate of the arithmetic mean of a lognormal population may be obtained from the following equation if n is large:

$$\text{Log}_{10} \text{ estimated arithmetic mean} \equiv \text{Log}_{10} \text{ G.M.} - 1.1513 (\text{Log}_{10} \text{ G.D.})^2$$ (Sichel, 1952)

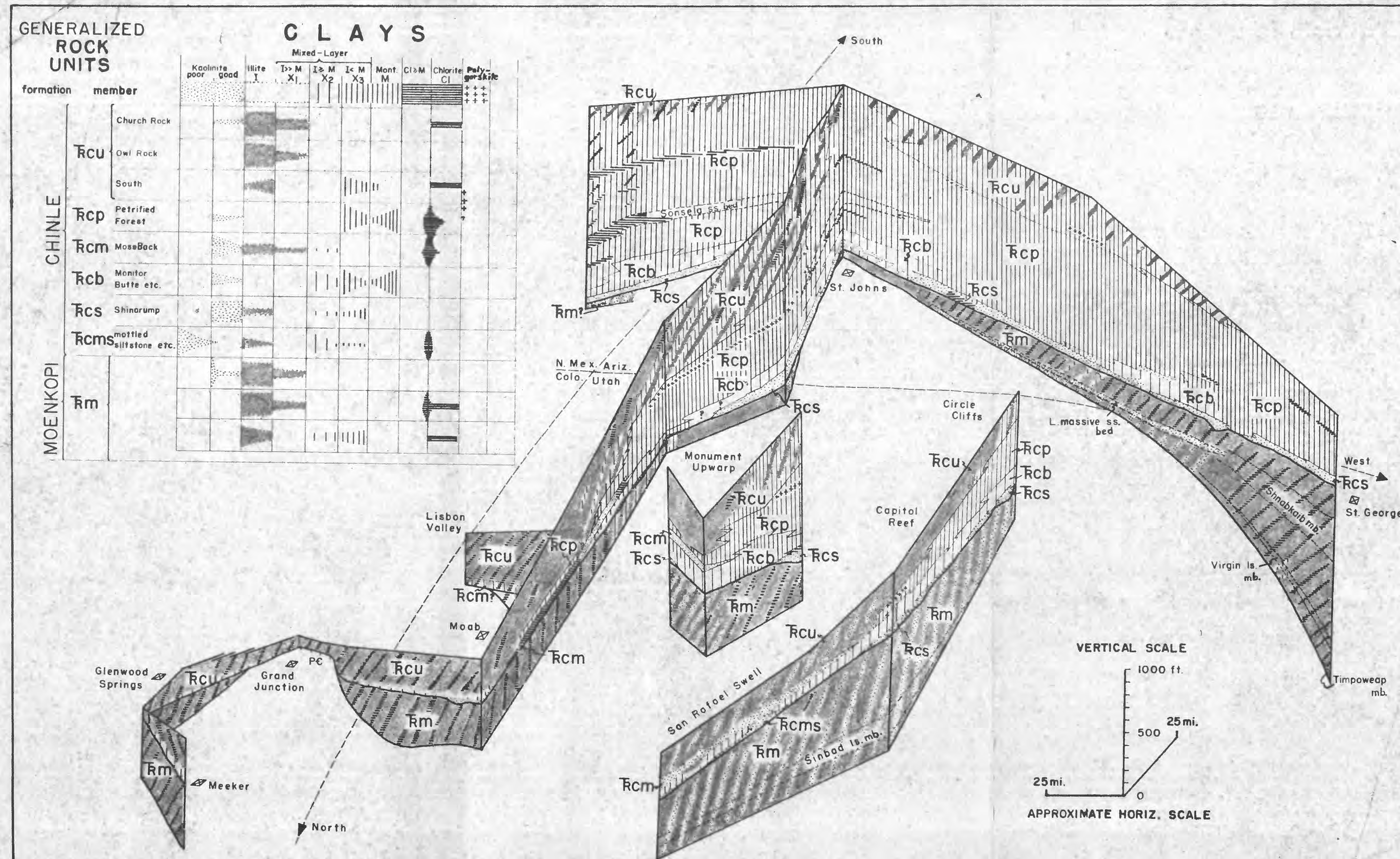



Fig. 112 CLAYS IN TRIASSIC ROCKS OF THE COLORADO PLATEAU