

Prepared in cooperation with the SCHENECTADY COUNTY PLANNING DEPARTMENT



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CONSIDERATIONS FOR MONITORING WATER QUALITY OF THE SCHENECTADY AQUIFER, SCHENECTADY COUNTY, NEW YORK

Ronald V. Allen and Roger M. Waller

U.S. GEOLOGICAL SURVEY

Water Resources Investigations 80-103

Prepared in cooperation with the SCHENECTADY COUNTY PLANNING DEPARTMENT



Albany, New York

1981

UNITED STATES DEPARTMENT OF THE INTERIOR

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CONVERSION FACTORS

The following factors may be used to convert inch-pound units of measurement in this report to the International System (SI) units.

Multiply	By	<u>To obtain</u>
foot (ft)	0.3048	meter (m)
mile (mi)	1.609	kilometer (km)

CONSIDERATIONS FOR MONITORING WATER QUALITY OF THE SCHENECTADY AQUIFER, SCHENECTADY COUNTY, NEW YORK

By

Ronald V. Allen and Roger M. Waller

ABSTRACT

Public water-supply systems in eastern Schenectady County obtain water from sand and gravel units that form a virtually continuous aquifer system contiguous to the Mohawk River. Water in the aquifer is principally under water-table conditions and in hydraulic contact with the river, so that pumping of most wells induces recharge from the river. Direct recharge to the aquifer from precipitation and runoff occurs throughout the valley floor.

No water-quality deterioration from toxic substances, including pesticides, has been detected from chemical analyses of water since 1972. Geohydrologic conditions at six well fields were evaluated to determine the feasibility of a monitoring system to provide warning of contamination before it reaches a pumping center. Potential contamination sources in the area are landfills, gravel pits, industrial sites, and transportation corridors. Wells that could be used for monitoring were present in two of the well fields. At the other sites, two or more wells would need to be installed to provide minimal means of detecting contaminants migrating toward a pumping center.

Water-quality monitoring, as required by the New York State Department of Health since the early 1970's, includes an annual evaluation of public water supplies to determine concentrations of inorganic chemicals and some heavy metals. The Mohawk River is sampled 9 miles east of the Niskayuna water supply (the well field farthest downstream), and 9 miles west of Rotterdam Junction (the site farthest upstream). Phenol concentrations have been noted in the river analyses, and chloride increases occur in the ground-water analyses. Chemical analyses of three water samples from privately owned wells near well fields showed minor concentrations of arsenic, lead, and zinc; however, all three metals could be derived from domestic plumbing.

Monitoring sites near each well field are indicated. Monitoring would be most effective by constructing pairs of wells to sample both shallow and deep zones of the aquifer. Frequency of water sampling for chemical analysis would be determined after an initial sampling period. An annual sample probably would be sufficient at most sites under ordinary circumstances.

INTRODUCTION

Ground water used for public supply in Schenectady County has a potential for contamination. Six public water-supply systems in eastern Schenectady County have well fields in sand and gravel deposits that underlie the Mohawk River valley; these deposits form an aquifer system that is locally referred to as the Schenectady aquifer.

A recent hydrologic study for Schenectady County (C. T. Male Associates, P. C., 1978) designated nine parts of the aquifer system to be protected from land-use practices that may contaminate the aquifer and result in shutdown of a well field for an indefinite period. The report also documented potential sources of contamination and recommended that the county develop a "... centrally coordinated water quality monitoring and surveillance program" where "... land use activity could affect the quality of the supply" (C. T. Male Associates, P. C., 1978, p. vii).

The U.S. Geological Survey, in cooperation with the Schenectady County Planning Department, evaluated the local hydrology to determine areas of recharge and possible sources of contamination. Monitoring and surveillance of the quality of ground water near the well fields could prevent contaminated water from being pumped into a distribution system that serves the public.

Methods and Scope

The six well fields--in the villages of Rotterdam Junction and Scotia, the towns of Glenville, Rotterdam, and Niskayuna, and the city of Schenectady-were evaluated for probable source and direction of recharge, potential sources of contamination, availability of wells for monitoring, and results of present or former monitoring programs. Information on the hydrology and geology, in addition to much well data, were available from a study by Winslow and others (1965). Possible sources of contamination were designated by C. T. Male Associates, P. C. (1978) and public water-supply authorities of the six well fields. Additional well inventories and water samplings were done by the USGS. The New York State Department of Health (NYSDH) and New York State Department of Environmental Conservation (NYSDEC) provided information on current and former water-quality programs. Additional water-quality and well data were obtained from files of the USGS in Albany.

Appreciation is extended to the city, county, town, and village officials who made their time available to aid this study.

HYDROLOGY OF THE SCHENECTADY AQUIFER

The six public water-supply well fields (fig. 1) tap the Schenectady aquifer, which consists of deposits of coarse sand and gravel that were deposited in a shale bedrock valley by glacial meltwaters and subsequently reworked, in part, by the Mohawk River. The deposits form an extensive linear water-table aquifer system ranging from 30 to 100 feet thick in the city of Schenectady. The system is in hydraulic contact with the Mohawk River. The downstream, southernmost field (Niskayuna) is separated from the upstream sand and gravel units by bedrock in the vicinity of Lock 7.

The Mohawk River is the dominant hydrologic factor in the aquifer system. Under nonflood conditions, the river receives discharge from the aquifer except near well fields in which pumping from the aquifer creates radial

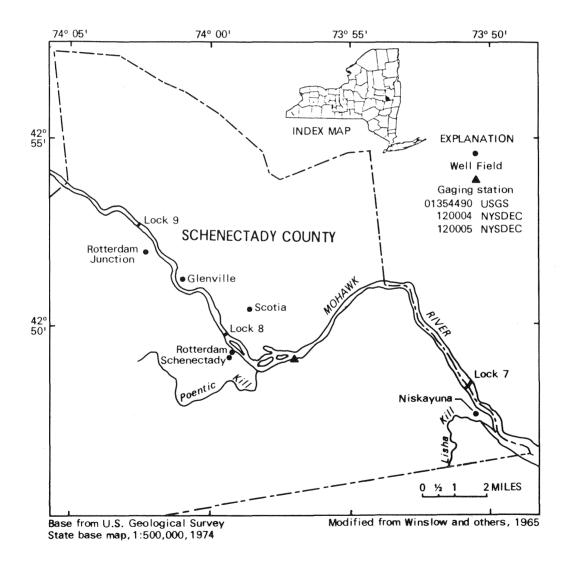


Figure 1.--Major geographic features and location of well fields.

flow toward the pumping centers. Under flood conditions the river recharges the aquifer. Recharge also occurs from precipitation on the valley floor, by underflow from valley deposits upgradient, and from runoff and seepage from adjacent highlands. Consequently, normal ground-water flow is toward the stream in a downvalley direction. Because of the interplay between the river and the aquifer, water moving down the valley may leave and reenter the aquifer several times on its way to the Hudson River. Therefore, the extent and distribution of pumping can greatly influence the movement of contaminants near or below the water table.

WELL FIELDS STUDIED

Village of Rotterdam Junction

The Rotterdam Junction well field, about 300 feet southwest of the Mohawk River (fig. 2), has two wells screened 51 to 83 feet below land surface. Pumping in the well field causes ground water to flow mainly southwestward from the river. Some ground water probably flows eastward and northwestward

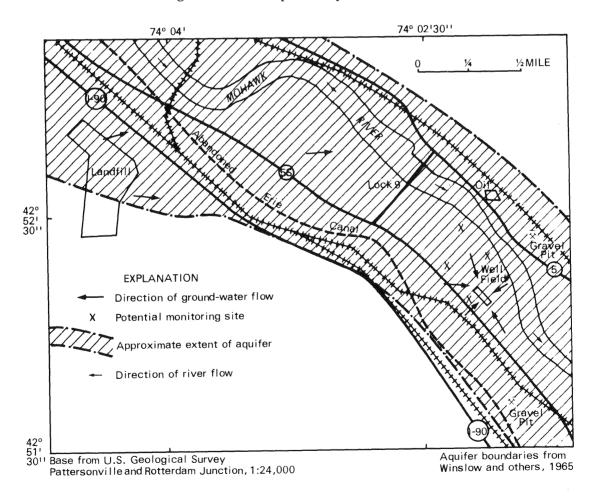


Figure 2.--Village of Rotterdam Junction area showing relation of well field to sources of potential contamination.

toward the field. Winslow and others (1965, p. 38) reported excellent hydraulic contact between the river and the Schenectady aquifer.

Potential sources of contamination of the aquifer are road salt on Route 5S, an abandoned oil-tank storage lot north of the river, a junk-car storage lot southwest of the well field, and a landfill more than a mile to the west. Although these sites are improbable sources of contamination, they are possible sources to monitor.

If a gravel pit to the south were to receive wastes, it could introduce contaminants into the aquifer because it is open to the water table. During periods of low river level and heavy pumpage in the village well field, ground-water conceivably could move upvalley toward the well field.

Village well water has been analyzed annually by NYSDH for several years. Comparison of a 1960 analysis (see well RWD3, table 5) with a 1973 analysis (see well F, table 6) indicates that chloride concentration in the aquifer has increased from 27 to 68 mg/L in the village well field.

None of the observed wells could be used for monitoring. Sites for observation wells should be near the river; northwest, north-northeast, and north-northwest of the well field; and near Route 55 (fig. 2).

Town of Glenville

The Glenville well field, 800 feet northeast of the river (fig. 3), has three screened wells, which tap the aquifer at 55 feet below land surface. A fourth well is planned. An abandoned gravel pit just northeast of the well field probably overlies the principal route of ground-water travel toward the river. Pumping withdrawals cause radial flow of ground water toward the well field and induces eastward and northeastward flow from the river. Stream infiltration is greater during summer than winter because river levels are higher during the summer navigation season, and river water is warmer and thus has a higher viscosity.

Potential sources of ground-water contamination include accidental spills along transportation routes to the north (a railroad 0.2 mile from the well field and Route 5, 0.4 mile from the well field) and the abandoned gravel pit immediately to the northeast. In addition, the river receives industrial effluent from a chemical plant 0.5 mile upstream.

Water samples have been collected annually and analyzed by NYSDH and a consultant chemist. A 1958 analysis (see well 251-401-11, table 5) showed a chloride concentration of 6 mg/L, whereas in 1971 the chloride was 23 mg/L (well B, table 6). Two wells were located for sampling in this study. Water from the Pucci well (well 1, fig. 3), 0.5 mile east of the field, was sampled August 29, 1979. Concentrations of minor elements, nutrients, and organic compounds were within recommended limits, and no pesticides were detected (table 1). The Widmer well (well 2, fig. 3) was sampled on August 30, 1979. This water had 1 mg/L of arsenic; zinc and iron were unusually high (320 and 450 ug/L, table 2). All three elements could be derived from plumbing metals.

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Analyses of five samples by NYSDH for nutrients and major ions (both raw and distribution water) during 1971-78 indicate concentrations below recommended limits for these constituents. However, raw water from wells require routine analysis for organic compounds and minor elements to establish a baseline for future comparisons.

No wells are available for monitoring in the immediate vicinity of the well field. Potential locations for observation wells are 900 feet northwest and 800 feet northeast of the well field.

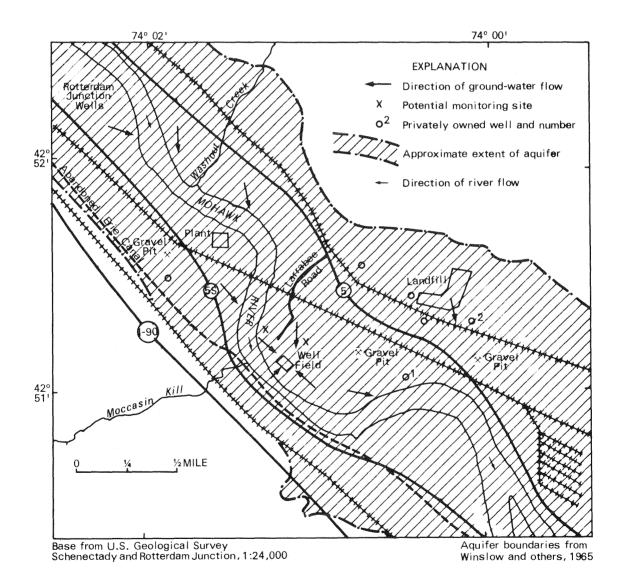


Figure 3.--Town of Glenville area showing relation of well field to sources of potential contamination.

Village of Scotia

The Scotia well field, 0.9 mile northeast of the river (fig. 4), has three wells screened between 60 and 98 feet below land surface. The wells probably receive little infiltration from the river because of their relatively long distance from it and because the pumping level is generally 10 to 20 feet higher than the stream elevation above Lock 8 (fig. 1). Recharge is derived principally from the north and from an unnamed stream that drains the higher elevations and crosses the aquifer 0.2 mile northeast of the well field.

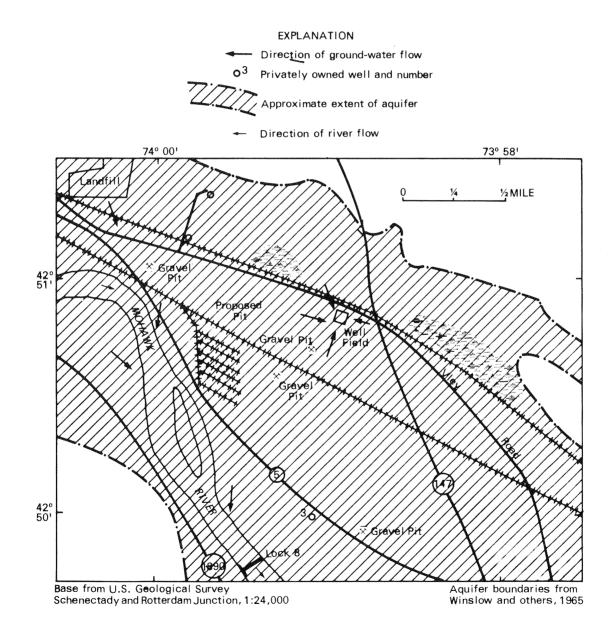


Figure 4.--Village of Scotia area showing relation of well field to sources of potential contamination.

The area surrounding the well field is prone to contamination from accidental spills. Transportation lanes immediately to the north (Route 147, Vley Road, and Amtrak) overlie the probable area of pumping influence. The town of Glenville landfill is about 1 mile northwest of the well field, and a large gravel pit 0.1 mile south of the well field has been excavated to the upper limits of the water table, which was visible in June 1979. The planned expansion of the pit westward will eventually increase this area of vulnerability. Contaminants introduced to the lower elevations of the pit would be in contact with the upper zone of the source of water withdrawn at the well field.

Analyses of raw water that was routinely sampled by NYSDH from 1972-78 showed no concentrations of the constituents analyzed to be above established or recommended limits. Analyses of a combined sample from two of the wells in 1943 (table 5) and one in 1971 (table 6) show that chloride has increased from 3 to 43 mg/L. One well was located in this study to obtain a water sample. The Lewis well (well 3, fig. 4), along Route 5 and south of the field, was sampled August 30, 1979 (table 3) and showed arsenic to be present. Sodium and chloride were somewhat higher than in most analyses. This is not surprising because the site is subject to road-salt contamination.

The well field may require four or five observation wells. The Village of Scotia has abandoned two wells along Vley Road just north of their present wells; if accessible, these could be used for monitoring. Observation wells tapping the upper part of the aquifer could provide early evidence of contaminants moving toward production wells, and an observation well screened in a lower part could identify contaminant migration or mixing characteristics within the cone of depression.

Town of Rotterdam

The Rotterdam well field is southwest of Isle of the Oneidas Island, about 300 feet from the river (fig. 5). The field consists of three wells 80 feet deep. Ground water moves principally from the river to the well field. The quality of water infiltrating from the river can be determined by analysis of water pumped from wells 61 and 54 (Winslow and others, 1965, fig. 26).

Route 5S, 400 feet southwest, and Route I-890, 900 feet southwest, may contribute chloride from winter road salt. The aquifer is also subject to spills from vehicles on the highways.

Chemical analyses of water from the Rotterdam well field have been made routinely by the NYSDH for several years. A 1971 analysis, presented in table 6 (site G), shows chloride at 19 mg/L. One of the wells was sampled for analysis of organic compounds in 1978 by the USGS; no contaminants were detected.

The Mantika well (well no. 4, fig. 5), 0.4 mile northwest of the well field, reportedly taps the aquifer at 60 feet below land surface. Water sampled from this well on August 29, 1979 was free of pesticides and contained concentrations below recommended limits for minor elements (table 4). However, the sum of iron plus manganese (0.54 mg/L) and the phenol concentration (0.0001 mg/L) equaled NYSDEC quality standards. The chloride concentration, 16 mg/L, was slightly above that from most other wells in the area.

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Two of the well field s original observation wells--wells 54 and 61, to the north and northeast, were found in July 1979. A portable pump would be required for sampling. An additional well west of the field should also be considered for monitoring.

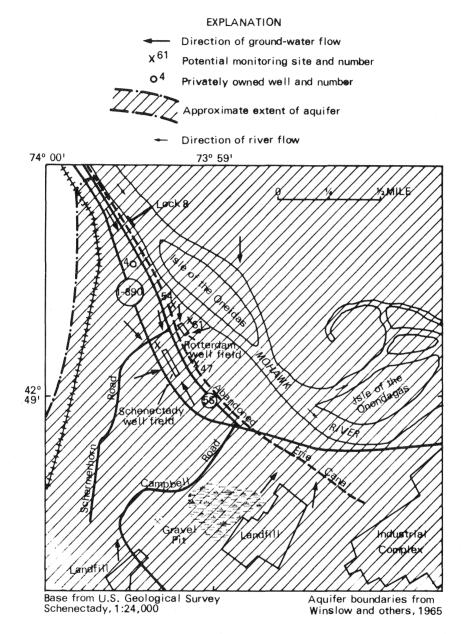


Figure 5.--Town of Rotterdam and city of Schenectady area showing relation of well field to sources of potential contamination.

City of Schenectady

The Schenectady well field, which has 12 wells, is 1,200 feet southwest of the Mohawk River and 900 feet south of the Rotterdam well field (fig. 5). The water originates primarily from induced river infiltration from the north at Lock 8, but water flows from all other directions (Winslow and others, 1965).

The river is a potential source of contamination to the well field. Route I-890, which traverses the length of the well field 300 feet to the southwest, may contribute chloride leachate from road-salt applications and is also subject to accidental spills. Potential areas of contamination to the south include a truck terminal and gravel pits near Campbell Road. An industrial complex, 1 mile to the southeast and downgradient from the well field, may be too far away to threaten the well field.

A review of analyses made by NYSDH during 1971-78 shows the water to be of excellent quality. Several determinations for pesticides revealed zero or near-zero concentrations (see table 7). Fluoride and manganese concentrations were above recommended limits in a few analyses. Selected analyses by the USGS are given in tables 5 and 6. Two of the city wells were sampled in 1978 by the USGS for analyses of organic compounds, but none were noted.

An excess of chloride, which is a common problem in shallow ground waters in much of the Northeast, is also common in this area. Table 5 shows that chloride concentration in the Schenectady City water averaged 6 mg/L in the 1940's and 18 mg/L (table 6) in the 1970's (the State and Federal recommended limit is 250 mg/L).

Observation well 47 (Winslow and others, 1965, fig. 26) is available to sample water moving toward the well field from the river. The observation wells to the northwest, nos. 57, 36, 46, and 24 (Winslow and others, 1965, fig. 26), were not found during field reconaissance and are assumed to have been destroyed during construction of Route I-890. A well drilled in this area north of Schermerhorn Road would give early notice of contaminants moving toward the well field from the northwest. Observation wells could be drilled 0.1 mile northwest of Campbell Road to monitor the southern part of the area influenced by pumping.

Town of Niskayuna (Water District 5)

Niskayuna's Water District 5 well field is along the south shore of the Mohawk River 0.6 mile downstream from Lock 7 (fig. 6). Its four wells are about 60 feet deep and are located 150 to 200 feet southwest of the river, which is the principal source of water to the well field. This reach of the river is subject to flooding as a result of ice jams; flood water in February 1979 reached within a few feet of the pumping station.

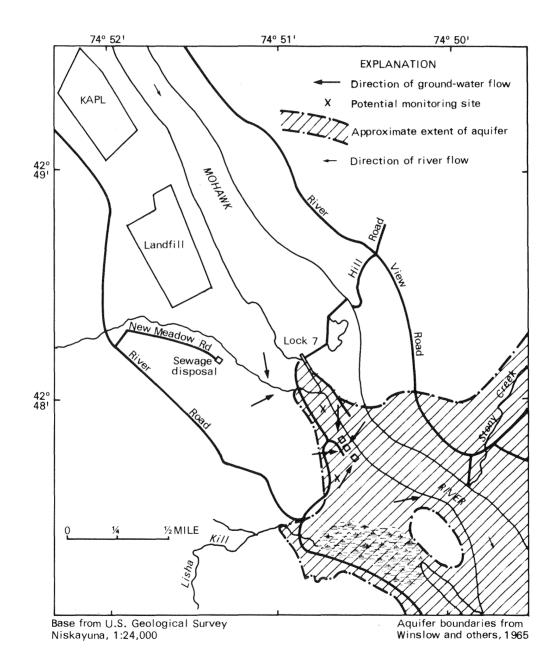


Figure 6.--Town of Niskayuna area showing relation of well field to sources of potential contamination.

A potential source of contamination to the Schenectady aquifer is runoff in the unnamed stream that drains the area occupied by the sewage-disposal plant on New Meadow Road. The stream discharges into the river below Lock 7, 0.4 mile upstream from the well field. Other areas of potential contamination along the south side of the river are the town of Niskayuna landfill, 1.5 miles northwest of the well field, the Knolls Atomic Power Laboratory (KAPL), and General Electric (GE) plant, about 2 miles north of the well field (not shown in fig. 6).

Chemical analyses of water from the well field have been made routinely by the NYSDH for several years. Water from the well field contains large amounts of iron and manganese, which are removed by chemical treatment (see table 6). Observation wells could be located in a number of places. Two potential sites are between the bedrock outcrop and the river, 0.3 mile upstream from the well field, and adjacent to Lock 7 Road, about 1,800 feet west of the well field.

MOHAWK RIVER MONITORING

River water has been sampled monthly since 1972 at Route 5 on the north side of the Mohawk River (USGS station 01354490, fig. 1). In addition, water from the river at Lock 10, 7 miles upstream from Rotterdam Junction, and Cresent Dam, 9 miles downstream from Niskayuna's well field, have been sampled by NYSDEC. Crescent Dam is the only monitoring site at present. The period of record and types of analyses are as follows:

Sites	Record	Type of analysis
USGS 01354000, at Tribes Hill	Monthly, April 1973 - March 1979	Chemical, nutrients, heavy metals
USGS 01354160, Lock 10 at Cranesville	Monthly April 1969 - April 1979	Chemical, nutrients, heavy metals
NYSDEC 12-0300, Lock 10	Biweekly November 1967. to present	Chemical, biological, heavy metals
USGS 01354490, at Schenectady, at Rte. 5	Monthly April 1969 - May 1979	Chemical, nutrients, heavy metals
NYSDEC 12-0004.5; at Schenectady, at Rte. 5 (Washington Avenue Branch)	Biweekly October 1964 - October 1976	Chemical, biological, heavy metals
NYSDEC 12-0002, at Cresent Dam	Biweekly Sept. 1965 - Sept. 1976	Chemical, nutrients, biological, heavy metals

The analyses indicates that only phenols have been present occasionally. Chloride does not seem to have been increasing since 1972, although Peters and Turk (in press) indicate a 75-percent increase in mean chloride concentration in the Mohawk River basin since the 1950's. They also conclude that 41 percent of the current chloride load (maximum 28 mg/L) is due to road-salt operations in the basin. The river-monitoring system can be a useful indicator for aquifer protection, but only in a broad sense and on a long-term basis.

MONITORING DISCUSSION

An adequate monitoring system would include observation wells for each well field. The appraisals given in the previous section suggest observation well sites at each field. However, if only one or two are installed in areas where ground-water flow toward the well field is indicated, local contamination might not be detected. The first priority in site selection would be downgradient from areas having a known source of contamination.

Pairs of observation wells would enable drawing water from both the top of the aquifer and from greater depth because some contaminants move near the top of an aquifer whereas others disperse and move downward. For economy, chemical analyses could be limited to suspected constituents or those most critical in human consumption. Because ground-water movement is relatively slow, monthly or annual sampling may be adequate.

The Scotia well field, in particular, seems to have the greatest potential for contamination and, in addition, has the least, if any, directly induced river water. Potential sources of contamination are on all sides of the field.

Of critical importance is curtailment of pumping to allow spilled contaminants to move away from a well field. Once a contaminant is known to have entered a system, a program of well installation and sampling can be set up to map the movement of the contaminanted water or, if necessary, to pump the contaminated water from the aquifer.

SELECTED REFERENCES

- C. T. Male Associates, P. C., 1978, Schenectady County aquifer preservation study and report: C. T. Male Associates, P. C., 3000 Troy Rd., Schenectady, N.Y., 89 p.
- Peters, N. E., and Turk, J. T., Increase in sodium and chloride in Mohawk River, New York, attributed to road salt: American Water Resources Association Bulletin (in press).
- Simpson, E. S., 1952, The ground-water resources of Schenectady County, New York: New York State Water Power and Control Commission, Bulletin GW-30, 110 p.
- Winslow, J. D., Stewart, H. G., Jr., Johnston, R. H., and Crain, L. J., 1965, Ground-water resources of eastern Schenectady County, New York, with emphasis on infiltration from the Mohawk River: New York State Conservation Department, Bulletin 57, 148 p.

TABLES 1-5

CHEMICAL ANALYSES OF WELL WATER

Table	1Town of Glenville, Pucci well	16
	2Town of Glenville, Widmer well	17
	3Village of Scotia, Lewis well	18
	4Town of Rotterdam, Mantika well	19
	5Selected Schenectady County wells	20

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Abbreviations Used in Tables 1-5

ALK	Alkalinity	MIN	Minimum
COND	Conductance	NC	Noncarbonate
DEG	Degree	NITR	Nitrogen
DIS, DISS	Dissolved	NONCARB	Noncarbonate
Е	Estimated (reported	PCI/L	Picocuries per liter
	well depths)	PHOS	Phosphate
FLD	Field	SAR	Sodium absorption ratio
KJD	Kjeldahl	SP, SPEC	Specific
LSD	Land surface datum	TEMP	Temperature
MBAS	Methlene blue active	TON/AFT	Tons per acre foot
	substance	TOT	Total
MG/L	Milligrams per liter	UGL	Micrograms per liter
MEQ/L	Millequivalents per liter		

Table 1.--Chemical analysis of water from Pucci well (well 1) Town of Glenville

[Analysis by U.S. Geological Survey]

SAMPLE LOCATION: RALPH PUCCI STATION ID: 425102074003601 LAT.LONG.SEQ.: 425102 0840037 01 DATE OF COLLECTION: BEGIN--790829 END--TIME--1445 STATE CODE: 36 COUNTY CODE: 093 PROJECT IDENTIFICATION: 443607200 DATA TYPE: 2 SOURCE: GROUND WATER GEOLOGIC UNIT: 1125DGV COMMENTS: COLL BY RV ALLEN FILTERED FOR METALS BUT ACID ADDED BEFORE FILTERING AND NO VISIBLE SEDIMENT ON FILTER ALDRIN TOTAL (WATER) UG/L 0.00 NITR. NH4 AS NH4 DIS MG/L 0.01 NITR. NO2 AS NO2 DIS MG/L 0.00 ALK, TOT(CACO₃) MG/L 160 ANALYZING AGENCY 80010 NITR. NO_3 AS NO_3 DIS MG/L 3.9 ARSENIC TOTAL UG/L NITROGEN DIS ORG ASN MG/L 1 0.00 CADMIUM TOTAL UG/L 0 NITROGEN DISS AS N MG/L 0.87 CALCIUM DISS MG/L 54 NITROGEN DISS KJD 0.00 MG/L 16 NITROGEN NH4 ASN DIS MG/L CARBON DIOXIDE MG/L 0.01 NITROGEN NO2 ASN DIS MG/L MG/L 0.00 CARBON TOT ORGANIC 1.2 CHLORDANE TOT(WATER) UG/L 0.0 NITROGEN NO3 ASN DIS MG/L 0.87 CHLORIDE DISS MG/L 11 NO2+NO3 AS N DISS 0.87 MG/L CHROMIUM TOTAL UG/L 10 OIL AND GREASE MG/L 1 0.0 COPPER TOTAL UG/L 40 PCB TOTAL (WATER) UG/L DDD TOTAL (WATER) UG/L 0.00 PCN TOTAL (WATER) UG/L 0.0 (WATER) UG/L 0.00 PERTHANE TOTAL UG/L 0.00 DDE TOTAL 0.00 DDT TOTAL (WATER) UG/L PH FIELD 7.3 DEPTH BELOW LSD (FT) PH LAB E 68.0 7.7 DEPTH OF WELL IN FT. Ε 75.0 PHENOLS UG/L 0 DIAZINON TOT (WATER) UG/L 0.00 0.00 PHOS ORTHO DIS AS P MG/L DIELDRIN TOT (WATER) UG/L 0.00 PHOSPHATE DIS ORTHO MG/L 0.00 ENDOSULFAN I TOTAL UG/L 0.00 PHOSPHORUS DIS AS P MG/L 0.00 ENDRIN TOTAL (WATER) UG/L 0.00 POTASSIUM DISS 0.9 MG/L ETH PARTH TOT(WATER) UG/L 0.00 0.7 POTASSIUM 40.D.PCI/L ETH TRITH TOT(WATER) UG/L 0.00 PUMP PERIOD (MIN) 20.0 0.00 ETHION TOTAL (WATER) UG/L RESIDUE DIS CALC SUM MG/L 228 FLUORIDE DISS MG/L 0.1 RESIDUE DIS TON/AFT 0.36 33 HARDNESS NONCARB MG/L **RESIDUE DIS 180C** MG/L 266 HARDNESS TOTAL MG/L 190 SAMPLE SOURCE CODE 30 HEPT EPOX TOT(WATER) UG/L 0.00 SAR 0.2 HEPTACHLOR T. (WATER) UG/L 0.00 SELENIUM TOTAL UG/L 0 70 IRON TOTAL UG/L SILICA DISSOLVED MG/L 9.4 LEAD TOTAL UG/L 0 SODIUM + POTASSIUM MG/L 8.2 LINDANE TOTAL (WATER) UG/L 0.00 7.3 SODIUM DISS MG/L MAGNESIUM DISS MG/L 14 SODIUM PERCENT 8 MALATHION TOT(WATER) UG/L 0.00 SP. CONDUCTANCE FLD 423 MANGANESE TOTAL UG/L 20 SP. CONDUCTANCE LAB 405 MERCURY TOTAL UG/L < 0.5 SULFATE DISS MG/L 31 MET PARTH TOT(WATER) UG/L 0.00 TOXAPHENE TOT (WATER) UG/L 0.0 MET TRITH TOT(WATER) UG/L 0.00 WATER TEMP (DEG C) 13.5 ZINC TOTAL UG/L 60 CATIONS ANIONS (MG/L) (MEQ/L) (MG/L) (MEQ/L) CALCIUM DISS CHLORIDE DISS 54 2.695 11 0.311 MAGNESIUM DISS 14 1.152 0.1 0.006 FLUORIDE DISS 0.9 0.024 31 POTASSIUM DISS SULFATE DISS 0.646 SODIUM DISS 7.3 0.318 ALK.TOT(CACO₃) 160 3.197 NO_2+NO_3 AS N D 0.87 0.063 TOTAL 4.187 TOTAL 4.220

Table 2.--Chemical analysis of water from Widmer well (well 2) Town of Glenville

[Analysis by U.S. Geological Survey]

SAMPLE LOCATION: FRED WIDMER STATION ID: 425118074000801 LAT.LONG.SEQ.: 425118 0740008 01 DATE OF COLLECTION: BEGIN--790830 END--TIME--1045 STATE CODE: 36 COUNTY CODE: 093 PROJECT IDENTIFICATION: 443607200 DATA TYPE: 2 SOURCE: GROUND WATER GEOLOGIC UNIT: COMMENTS: COLL BY RV ALLEN ALDRIN TOTAL (WATER) UG/L 0.00 NITR. NH4 AS NH4 DIS MG/L 0.01 NITR. NO2 AS NO2 DIS MG/L 0.00 87 ALK, TOT(CACO₃) MG/L ANALYZING AGENCY 80010 NITR. NO_3 AS NO_3 DIS MG/L 1.2 NITROGEN DIS ORG ASN MG/L ARSENIC TOTAL UG/L 1 0.00 CADMIUM TOTAL UG/L 0 NITROGEN DISS AS N MG/L 0.27 NITROGEN DISS KJD CALCIUM DISS 0.00 MG/L 48 MG/L CARBON DIOXIDE MG/L 2.1 NITROGEN NH4 ASN DIS MG/L 0.01 0.9 CARBON TOT ORGANIC MG/L NITROGEN NO2 ASN DIS MG/L 0.00 NITROGEN NO3 ASN DIS MG/L CHLORDANE TOT (WATER) UG/L 0.0 0.27 NO2+NO3 AS N DISS MG/L 0.27 CHLORIDE DISS MG/L 6.4 CHROMIUM TOTAL UG/L 10 OIL AND GREASE MG/L 1 COPPER TOTAL 20 PCB TOTAL (WATER) UG/L 0.0 UG/L DDD TOTAL (WATER) UG/L 0.00 PCN TOTAL (WATER) UG/L 0.0 (WATER) UG/L DDE TOTAL 0.00 PERTHANE TOTAL UG/L 0.00 (WATER) UG/L 0.00 7.9 DDT TOTAL PH FIELD DEPTH BELOW LSD (FT) E 60.0 PH LAB 7.8 DEPTH OF WELL IN FT. E 92.0 PHENOLS UG/L 0 MG/L 0.00 DIAZINON TOT (WATER) UG/L 0.00 PHOS ORTHO DIS AS P DIELDRIN TOT (WATER) UG/L 0.00 PHOSPHATE DIS ORTHO MG/L 0.00 ENDOSULFAN I TOTAL UG/L 0.00 PHOSPHORUS DIS AS P MG/L 0.00 ENDRIN TOTAL (WATER) UG/L 0.00 POTASSIUM DISS MG/L 1.4 ETH PARTH TOT(WATER) UG/L 0.00 POTASSIUM 40.D.PCI/L 1.0 PUMP PERIOD (MIN) ETH TRITH TOT(WATER) UG/L 0.00 30.0 ETHION TOTAL (WATER) UG/L 0.00 RESIDUE DIS CALC SUM MG/L 210 FLUORIDE DISS 0.1 RESIDUE DIS TON/AFT 0.34 MG/L **RESIDUE DIS 180C** HARDNESS NONCARB MG/L 82 MG/L 251 HARDNESS TOTAL MG/L 170 SAMPLE SOURCE CODE 30 HEPT EPOX TOT(WATER) UG/L 0.00 SAR 0.2 HEPTACHLOR T. (WATER) UG/L 0.00 SELENIUM TOTAL UG/L 0 IRON TOTAL UG/L 450 SILICA DISSOLVED MG/L 11 LEAD TOTAL 0 SODIUM + POTASSIUM MG/L 7.5 UG/L 0.00 LINDANE TOTAL (WATER) UG/L SODIUM DISS MG/L 6.1 MAGNESIUM DISS MG/L 12 SODIUM PERCENT 7 MALATHION TOT (WATER) UG/L 0.00 SP. CONDUCTANCE FLD 378 MANGANESE TOTAL UG/L 20 SP. CONDUCTANCE LAB 347 MG/L MERCURY TOTAL UG/L <0.5 SULFATE DISS 71 MET PARTH TOT(WATER) UG/L 0.00 TOXAPHENE TOT(WATER) UG/L 0.0 MET TRITH TOT(WATER) UG/L 0.00 WATER TEMP (DEG C) 8.0 ZINC TOTAL UG/L 320 CATIONS ANIONS (MEQ/L) (MG/L) (MEQ/L) (MG/L) CALCIUM DISS 48 2.396 CHLORIDE DISS 6.4 0.181 0.988 FLUORIDE DISS 0.1 0.006 MAGNESIUM DISS 12 POTASSIUM DISS 1.4 0.036 SULFATE DISS 71 1.479 87 1.739 SODIUM DISS 6.1 0.266 ALK. TOT(CACO3) NO2+NO3 AS N D 0.27 0.020 TOTAL 3.683 3.422 TOTAL PERCENT DIFFERENCE = 3.69

Table 3.--Chemical analysis of water from Lewis well (well 3) Village of Scotia

[Analysis by U.S. Geological Survey]

SAMPLE LOCATION: H R LEWIS STATION ID: 4249570735900 LAT.LONG.SEQ.: 424957 0735900 01 DATE OF COLLECTION: BEGIN--790830 END--TIME--0905 STATE CODE: 36 COUNTY CODE: 093 PROJECT IDENTIFICATION: 443607200 SOURCE: GROUND WATER GEOLOGIC UNIT: DATA TYPE: 2 COMMENTS: COLL BY RV ALLEN PUMPED DIR FROM WELL NITR. NH4 AS NH4 DIS MG/L 0.01 ALDRIN TOTAL (WATER) UG/L 0.00 270 NITR. NO2 AS NO2 DIS MG/L 0.03 ALK, TOT(CACO3) MG/L NITR. NO_3 AS NO_3 DIS MG/L ANALYZING AGENCY 80010 12 ARSENIC TOTAL UG/L 1 NITROGEN DIS ORG ASN MG/L 0.00 0 2.7 UG/L NITROGEN DISS AS N MG/L CADMIUM TOTAL CALCIUM DISS 84 NITROGEN DISS KJD 0.00 MG/L MG/L 105 CARBON DIOXIDE MG/L NITROGEN NH4 ASN DIS MG/L 0.01 CARBON TOT ORGANIC MG/L 2.4 NITROGEN NO2 ASN DIS MG/L 0.01 NITROGEN NO3 ASN DIS MG/L 2.7 CHLORDANE TOT(WATER) UG/L 0.0 NO2+NO3 AS N DISS 37 2.7 CHLORIDE DISS. MG/T. MG/L CHROMIUM TOTAL UG/L 20 OIL AND GREASE MG/L 1 COPPER TOTAL UG/L 10 PCB TOTAL (WATER) UG/L 0.0 DDD TOTAL (WATER) UG/L 0.00 PCN TOTAL (WATER) UG/L 0.0 (WATER) UG/L 0.00 PERTHANE TOTAL 0.00 DDE TOTAL HG/LDDT TOTAL (WATER) UG/L 0.00 PH FIELD 6.7 DEPTH BELOW LSD (FT) E 29.1 PH LAB 7.6 DEPTH OF WELL IN FT. E 32.0 PHENOLS UG/L 0 DIAZINON TOT (WATER) UG/L 0.00 PHOS ORTHO DIS AS P 0.00 MG/L DIELDRIN TOT (WATER) UG/L 0.00 PHOSPHATE DIS ORTHO MG/L 0.00 0.00 ENDOSULFAN I TOTAL UG/L PHOSPHORUS DIS AS P MG/L 0.00 ENDRIN TOTAL (WATER) UG/L 0.00 POTASSIUM DISS MG/L 1.0 ETH PARTH TOT (WATER) UG/L 0.00 POTASSIUM 40.D.PCI/L 0.7 ETH TRITH TOT(WATER) UG/L 0.00 PUMP PERIOD (MIN) 20.0 ETHION TOTAL (WATER) UG/L 0.00 RESIDUE DIS CALC SUM MG/L 421 RESIDUE DIS TON/AFT 0.54 FLUORIDE DISS MG/L 0.1 HARDNESS NONCARB MG/L 22 **RESIDUE DIS 180C** MG/L 397 HARDNESS TOTAL MG/L 290 SAMPLE SOURCE CODE 47 HEPT EPOX TOT(WATER) UG/L 0.00 SAR 1.1 HEPTACHLOR T. (WATER) UG/L SELENIUM TOTAL 0.00 0 UG/L IRON TOTAL UG/L 70 SILICA DISSOLVED MG/L 12 LEAD TOTAL UG/L 0 SODIUM + POTASSIUM MG/L 45 LINDANE TOTAL (WATER) UG/L 0.00 SODIUM DISS MG/L 44 25 MAGNESIUM DISS MG/L 20 SODIUM PERCENT MALATHION TOT(WATER) UG/L 0.00 SP. CONDUCTANCE FLD 750 MANGANESE TOTAL 10 707 UG/L SP. CONDUCTANCE LAB MERCURY TOTAL UG/L < 0.5 SULFATE DISS MG/L 49 MET PARTH TOT(WATER) UG/L 0.0 0.00 TOXAPHENE TOT (WATER) UG/L MET TRITH TOT(WATER) UG/L 0.00 WATER TEMP (DEG C) 11.5 ZINC TOTAL UG/L 30 CATIONS ANIONS (MG/L) (MEQ/L) (MG/L) (MEQ/L) CALCIUM DISS 84 4.192 CHLORIDE DISS 37 1.044 MAGNESTUM DISS 20 1.646 FLUORIDE DISS 0.1 0.006 POTASSIUM DISS 1.0 0.026 SULFATE DISS 49 1.021 SODIUM DISS 44 1.914 ALK.TOT(CACO₃) 270 5.395 2.7 0.193 NO2+NO3 AS N D TOTAL 7.776 TOTAL 7.657 PERCENT DIFFERENCE = 0.78

Table 4.---Chemical analysis of water from Mantika well (well 4) Town of Rotterdam

[Analysis by U.S. Geological Survey]

SAMPLE LOCATION: PAUL MANTIKA STATION ID: 424938073593101 LAT.LONG.SEO.: 424938 0735931 01 DATE OF COLLECTION: BEGIN--790829 END--TIME--1725 STATE CODE: 36 COUNTY CODE: 093 PROJECT IDENTIFICATION: 443607200 SOURCE: GROUND WATER GEOLOGIC UNIT: DATA TYPE: 2 COMMENTS: COLL BY RV ALLEN ALDRIN TOTAL (WATER) UG/L 0.00 NITR. NH4 AS NH4 DIS MG/L 0.08 ALK, TOT (CACO3) MG/L 110 NITR. NO2 AS NO2 DIS MG/L 0.03 NITR. NO_3 AS NO_3 DIS MG/L ANALYZING AGENCY 80010 0.04 NITROGEN DIS ORG ASN MG/L ARSENIC TOTAL UG/L 1 0.02 CADMIUM TOTAL 0 UG/L NITROGEN DISS AS N MG/L 0.10 CALCIUM DISS MG/L 44 NITROGEN DISS KJD MG/L 0.08 8.5 CARBON DIOXIDE MG/L NITROGEN NH4 ASN DIS MG/L 0.06 CARBON TOT ORGANIC 3.6 NITROGEN NO2 ASN DIS MG/L MG/L 0.01 CHLORDANE TOT(WATER) UG/L 0.0 NITROGEN NO3 ASN DIS MG/L 0.01 CHLORIDE DISS MG/L 16 NO2+NO3 AS N DISS MG/L 0.02 CHROMIUM TOTAL UG/L 10 OIL AND GREASE MG/L 2 COPPER TOTAL UG/L 10 PCB TOTAL (WATER) UG/L 0.0 0.00 PCN TOTAL DDD TOTAL (WATER) UG/L (WATER) UG/L 0.0 0.00 DDE TOTAL (WATER) UG/L PERTHANE TOTAL UG/L 0.00 0.00 DDT TOTAL (WATER) UG/L PH FIELD 7.4 20.0 DEPTH BELOW LSD (FT) E PH LAB 7.8 DEPTH OF WELL IN FT. E 60.0 PHENOLS UG/L 1 DIAZINON TOT (WATER) UG/L 0.00 PHOS ORTHO DIS AS P MG/L 0.01 DIELDRIN TOT (WATER) UG/L 0.00 PHOSPHATE DIS ORTHO MG/L 0.03 ENDOSULFAN I TOTAL 0.00 UG/L PHOSPHORUS DIS AS P MG/L 0.00 ENDRIN TOTAL (WATER) UG/L 0.00 POTASSIUM DISS MG/L 1.3 ETH PARTH TOT(WATER) UG/L 0.00 POTASSIUM 40.D.PCI/L 1.0 ETH TRITH TOT(WATER) UG/L 0.00 PUMP PERIOD (MIN) 25.0 0.00 ETHION TOTAL (WATER) UG/L RESIDUE DIS CALC SUM MG/L 178 FLUORIDE DISS 0.1 RESIDUE DIS TON/AFT 0.25 MG/L HARDNESS NONCARB MG/L 28 **RESIDUE DIS 180C** MG/L 187 HARDNESS TOTAL MG/L 140 SAMPLE SOURCE CODE 30 HEPT EPOX TOT(WATER) UG/L 0.00 SAR 0.4 HEPTACHLOR T. (WATER) UG/L 0.00 SELENIUM TOTAL IIG/I. 0 IRON TOTAL UG/L 120 SILICA DISSOLVED MG/L 6.1 LEAD TOTAL UG/L 0 SODIUM + POTASSIUM MG/L 12 0.00 SODIUM DISS LINDANE TOTAL (WATER) UG/L MG/L 11 MAGNESIUM DISS MG/L 6.8 SODIUM PERCENT 15 MALATHION TOT(WATER) UG/L 0.00 SP. CONDUCTANCE FLD 370 MANGANESE TOTAL UG/L 420 SP. CONDUCTANCE LAB 314 MERCURY TOTAL UG/L 40.5 SULFATE DISS MG/L 26 MET PARTH TOT(WATER) UG/L 0.00 TOXAPHENE TOT(WATER) UG/L 0.0 MET TRITH TOT(WATER) UG/L 0.00 WATER TEMP (DEG C) 16.5 UG/L ZINC TOTAL 0 CATIONS ANIONS (MG/L) (MG/L) (MEQ/L)(MEQ/L)CALCIUM DISS 44 2.196 CHLORIDE DISS 0.452 16 MAGNESIUM DISS 6.8 0.560 FLUORIDE DISS 0.1 0.006 POTASSIUM DISS 1.3 0.034 SULFATE DISS 26 0.542 SODIUM DISS 11 0.479 110 ALK.TOT(CACO3) 2.198 NO2+NO3 AS N D 0.02 0.002 TOTAL 3.267 TOTAL 3.197 PERCENT DIFFERENCE = 1.08

Table 5.--Chemical analyses of water from selected wells in Schenectady County, N.Y.

Number	Well number or owner	Previous well number ^a	Depth (feet)	Water-bearing formation	Date of collec- tion	Dis- solved solids	Silica (SiO ₂)	Iron (Fe)	Manga- nese (Mn)	Cal- cium (Ca)	Mag- nesi- um (Mg)
1	Scotia	Sn 4&5	70 & 85	Pleistocene sand	5-26-43			.03			
2	Schen	Sn 127	44	Pleistocene gravel	6- 1-28						
3	Schen	Sn 130 ^b	62	Pleistocene gravel	11- 9-49			.23			
4	Schen	Sn 130 ^b	62	Pleistocene gravel	9-23-48	187	6.5	.05		49	8.6
5	Schen	Sn 133 ^b	62	Pleistocene gravel	8-22-47	173	6.0	.09	.13	46	7.5
6	Schen	Sn 134	57	Pleistocene gravel	10- 5-46	195		.03	.05		
7	Schen	Sn 126-128	c	Pleistocene gravel	10-13-40	212		.15			
8	Schen	Sn 126-128	c	Pleistocene gravel	1-27-42	216		.2			
9	Schen	Sn 129-138	c	Pleistocene gravel	9-14-44	218		.1			
10	Glen	251-401-11	51	Sand and gravel	1- 9-58			.08			
11	RWD3	252-402-16 (SN 229)	r63	Sand and gravel	12- 7-60			.08			

[Analyses by New York State Department of Health unless otherwise indicated. Dissolved constituents given in parts per million.]

- <u></u>	Sodium and	Bicar-	Sul-	Chlo-	Fluo-	Ni-	Hardness	s (calculated	as CaCO ₃)	Total alka-	
	potassium	bonate	fate	ride	ride	trate		Car-	Noncar-	linity	
Number	(Na + K)	(HCO3)	(SO ₄)	(C1)	(F)	(NO3)	Total	bonate	bonate	(as CaCO ₃)	pН
1		137		3.0			128	112	16	112	7.5
2		122		5.5			119	100	19	100	
3		150	32	7.6	.0	.6	154	123	31	123	7.8
4	9.2	157	30	7.2	.1	.3	158	128	30	128	7.7
5	4.4	152	23	4.4	.1	.3	146	125	21	125	7.7
6		159	10	8.6	.05		148	130	18	130	7.7
7		154		7.0			174	126	48	126	7.5
8		161		5.5			165	132	33	132	7.5
9		146		6.0			164	120	44	120	7.6
10		192		6		1.5	220	220			7.7
11				27		6.5	310	310			7.5

a Well numbers in Winslow and others, 1965.

b Analysis by the Quality of Water Branch, U.S. Geological Survey.

c Analysis by Schenectady Sewage Disposal Laboratory.

TABLES 6-7

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Table	6Chemical	analyses of	water from	community	systems	
	in Schene	ectady Count	y, New York			22

Table 7.--Chemical analyses (pesticides and related constituents) of water from City of Schenectady, New York, wells..... 28 Table 6.--Chemical analyses of water from community systems in Schenectady County, New York

	COLUMN(S) ON THIS PAGE		USGS-ASSIC LATITUDE-LONC NUMBER	GITUDE	TEM (QW SITE) AND RAW SOUR(OF WATER SAMPI	CE			
		A	424934073575	5900 s	COTIA(V)-WELL				
		B	42495007359		LENVILLE-WELL				
		Č	42495007359		LENVILLE-WELL				
		D	424745073503		ISKAYUNA-WELLS	5			
		Е	424745073503		ISKAYUNA-WELLS				
		F	424718073585		OTTERDAM JUNCI				
		G	425725073585		OTTERDAM JUNCT				
SYSTEM(S) ON THIS PAGE	A	В	С	D	D	Е	Е	F	G
TYPE OF WATER SAMPLED.		RAW	TREATED	RAW	TREATED	RAW	TREATED	RAW	DISTRBN
DATE	12/01/71	12/01/71	12/01/71	12/01/71	04/11/75	12/01/71	04/11/75	02/28/73	12/01/71
ALUMINUM UG/L	5.0	89	7.0	67	13	63	20	7.0	46
ARSENIC UG/L	1	2	1	2	1	1	0	0	3
BARIUM UG/L	35	24	< 12	28	32	23	25	37	22
BERYLLIUM UG/L	< 2.0	< 2.0	< 2.0	< .80	< .70	< .90	< .80	< 4.0	< 1.0
BICARBONATE MG/L	210	190	189	115	145	148	178	330	185
BISMUTH UG/L	< 8.0	< 6.0	< 6.0	< 4.0	< 2.0	< 5.0	< 2.0	< 10	< 5.0
BORON UG/L	33	26	24	31	25	25	25	64	30
CADMIUM UG/L	0	0	0	0	0	0	0	0	0
CALCIUM MG/L	70	58	19	37	50	37	45	98	57
CARBONATE MG/L	0	0	0	0	0	0	0	0	0
CHLORIDE MG/L	43	23	22	17	18	21	23	68	19
CHROMIUM UG/L	< 8	< 6	< 6	< 4	< 2	< 5	< 2	< 10	< 5
COBALT UG/L	< 16	< 12	< 12	< 8.0	< 1.0	< 10	< 2.0	< 10	< 10
COLIFORM COL/100 ML									
COPPER UG/L	22	< 2.0	14	1.0	.60	37	35	3.0	8.0
CYANIDE MG/L	0	0	0	0	0.1	0	0	0	0
DISS SOLIDS SUM MG/L	320	237	255	173	206	208	240	469	226
FLUORIDE MG/L	.10	.10	.10	.20		.20	.10	.10	.20
GALLIUM UG/L	< 2.0	< 2.0	< 2.0	< .80		< .90	< .80	< 5.0	< 1.0
GERMANIUM UG/L	< 8.0	< 6.0	< 6.0	< 4.0	< 2.0	< 5.0	< 2.0	< 10	< 5.0

SYSTEM(S) ON THIS PAGE	DISTRBN	B	C	D	D	E	E	F	G
TYPE OF WATER SAMPLED		RAW	TREATED	RAW	TREATED	RAW	TREATED	RAW	DISTRBN
DATE		12/01/71	12/01/71	12/01/71	04/11/75	12/01/71	04/11/75	02/28/73	12/01/71
HARDNESS TOTAL MG/L	249	202	66	131	166	130	153	352	188
HARDNESS NONCARB MG/L	77	47	0	36	47	9	7	81	36
IRON UG/L	220	16	4.0	670	650	10	50	< 10	58
LEAD UG/L	< 8.0	< 6.0	< 6.0	< 4.0	< 2.0	< 5.0	< 2.0	< 10	< 5.0
LITHIUM UG/L	< 10	< 10	< 10	< 10	6.0	< 10	6.0	< 10	< 10
MAGNESIUM MG/L	18	14	4.5	9.3	10	9.2	9.8	26	11
MANGANESE UG/L	< 8.0	< 6.0	< 6.0	280	1100	< 5.0	1.0	< 8.0	130
MBAS MG/L	.03	.01	.02	.02	0	.02	0	.02	.01
MERCURY UG/L	< .50	< .50	< .50	< .50	< .50	< .50	< .50	< .50	< .50
MOLYBDENUM UG/L	< 2.0	< 2.0	< 2.0	< .80	< .70	< .90	< .80	< 5.0	< 1.0
NICKEL UG/L NITRATE AS N MG/L NITRITE AS N MG/L NITROGEN NH4 AS N MG/L NITROGEN NH4+ORG-N MG/L	< 8.0 1.1 08	< 6.0 .30 .03	< 6.0 .90 02	6.0 _40 _40	2.0 .23 0 .04	< 5.0 .30 .12	< 2.0 .22 0 .20	< 10 4.8 .01	< 5.0 .10
PH UNITS PHENOLS UG/L PHOSPHORUS AS P MG/L POTASSIUM MG/L RUBIDIUM UG/L	7.7 1.0 .01 1.5	7.7 5.0 0 1.2	7.6 0 .80	7.1 1.0 .01 1.4 	7.1 01 1.1 	7.9 0 .00 1.4 	7.6 .02 1.2 	7.8 .00 2.3 	7.6 1.0 .80 1.4
SELENIUM UG/L	0	0	2	0	0	0	0	0	0
SILICA MG/L	7.0	7.3	6.9	7.2	6.8	7.3	6.7	9.1	6.5
SILVER UG/L	< .50	<.30	<.30	<.20	< .20	<.30	< .20	< 1.0	< .30
SODIUM MG/L	20	10	72	10	10	27	27	34	11
SPECIFIC COND UMHOS	582	422	436	305	369	366	418	818	396
STRONTIUM UG/L	200	140	62	130	220	140	210	240	240
SULFATE MG/L	56	30	36	34	39	32	39	64	28
TIN UG/L	< 8.0	< 6.0	< 6.0	< 4.0	< 2.0	< 5.0	< 2.0	< 10	< 5.0
TITANIUM UG/L	< 4.0	< 3.0	< 3.0	< 2.0	< 2.0	< 3.0	< 2.0	< 10	< 3.0
VANADIUM UG/L	< 4.0	< 3.0	< 3.0	< 2.0	< 2.0	< 3.0	< 2.0	< 10	< 3.0
ZINC UG/L	400	< 250	< 250	< 170	0 < 2.0	< 210	0	50	< 210
ZIRCONIUM UG/L	< 16	< 12	< 12	< 8.0		< 10	< 3.0	< 22	< 10

		COLUMN ON THIS	I(S) LATIT	GS-ASSIGNED UDE-LONGITUDE NUMBER	AND RA	W SITE) NAME AW SOURCE ER SAMPLED			
		A	4249	10073591700	SCHENECT	TADY(C)-WELLS			
SYSTEM(S) ON THIS PAGE.	• TREATED	A	A	A	A	A	A	A	A
TYPE OF WATER SAMPLED		TREATED	TREATED	TREATED	TREATED	TREATED	TREATED	TREATED	TREATED
DATE		07/12/71	10/14/71	01/13/72	04/06/72	07/13/72	10/18/72	01/10/73	04/17/73
ALUMINUM UG/L	9.0	3.0	7.0	5.0	73	12	47	13	6.0
ARSENIC UG/L	0	0	0	2	0	1	0	0	0
BARIUM UG/L	34	41	42	33	28	26	36	26	29
BERYLLIUM UG/L	<.50	< .60	< 2.0	< 1.0	< 1.0	< 1.0	< 2.0	< 2.0	< 2.0
BICARBONATE MG/L	162	192	200	170	152	160	201	173	194
BISMUTH UG/L	< 5.0	< 3.0	< 6.0	< 5.0	< 5.0	< 5.0	< 6.0	< 5.0	< 5.0
BORON UG/L	45	18	39	41	20	16	22	27	17
CADMIUM UG/L	0	0	0	0	0	0	0	0	0
CALCIUM MG/L	52	60	61	54	52	52	62	58	63
CARBONATE MG/L	0	0	0	0	0	0	0	0	0
CHLORIDE MG/L CHROMIUM UG/L COBALT UG/L COLIFORM COL/100 ML COPPER UG/L	16 < 5 < 5.0 21	18 < 6 < 2.0 160	18 < 6 < 3.0 	16 < 5 < 5.0 14	16 < 3 < 5.0 330	14 < 5 < 5.0 500	18 < 6 < 6.0 3.0	18 < 5 < 6.0 140	18 < 6 < 6.0 390
CYANIDE MG/L	0	0	0	0	0	0	0	0.1	.01
DISS SOLIDS SUM MG/L	201	232	238	212	197	198	239	225	239
FLUORIDE MG/L	.20	.40	.90	1.1	< 20	1.2	.10	.90	.90
GALLIUM UG/L	ND	< 2.0	< 3.0	< 1.0	< 3.0	< 5.0	< 3.0	< 3.0	< 3.0
GERMANIUM UG/L	< 5.0	< 6.0	< 6.0	< 5.0	< 5.0	< 10	< 6.0	< 5.0	< 6.0

Table 6.--Chemical analyses of water from community systems in Schenectady County, New York (Continued)

SYSTEM(S) ON THIS PAGE. TYPE OF WATER SAMPLED DATE	. RAW	A RAW 07/12/71	A RAW 10/14/71	A RAW 01/13/72	A RAW 04/06/72	A RAW 07/13/72	A RAW 10/18/72	A RAW 01/10/73	A RAW 04/17/73
MAGNESIUM MG/L	5.9 190	12 93	11 180	11 180	10 140	9.8	10	10 95	10
MANGANESE UG/L MBAS MG/L	.01	.01	.02	.02	.01	95 .02	150 .02	.07	190 .01
MERCURY UG/L	< .50	< .50	< .50	< .50	< .50	< .50	< .50	< .50	< .50
MOLYBDENUM UG/L	.70	< 2.0	< 2.0	< 2.0	< 2.0	< 1.0	< 3.0	< 3.0	< 0
NICKEL UG/L	< 2.0	< 6.0	< 6.0	< 3.0	< 6.0	< 5.0	< 6.0	< 5.0	< 6.0
NITRATE AS N MG/L	.10	.44	.20	.30	.30	.20	.30	.30	.50
NITRITE AS N MG/L	0	0							
NITROGEN NH4 AS N MG/L	0	0							
NITROGEN NH4+ORG-N MG/I		0	.02	.11	0	.17	.08	•06	.10
PH UNITS	7.8	7.8	7.6	7.8	7.4	7.8	8.0	7.8	8.3
PHENOLS UG/L	0	0	0	0	1.0				
PHOSPHORUS AS P MG/L	.02	.01	.01	.01	.01	.02	.02	•02	.01
POTASSIUM MG/L	1.3	2.1	1.4	1.5	1.4	1.1	1.4	1.1	1.2
RUBIDIUM UG/L	< 2.0	< .40							
SELENIUM UG/L	1	2	6	1	1	0	7	2	0
SILICA MG/L	5.1	6.5	7.1	8.6	8.0	6.8	7.7	7.0	7.0
SILVER UG/L	< .50	< .60	< .60	< 2.0	< 2.0	< 1.0	< .60	< .50	< .60
SODIUM MG/L	11	11	12	12	10	8.6	11	9.4	10
SPECIFIC COND UMHOS	336	433	422	427	413	382	422	395	404
STRONTIUM UG/L	300	240	320	290	280	230	300	290	260
SULFATE MG/L	25	34	30	33	32	28	30	31	33
TIN UG/L	< 5.0	< 6.0	< 6.0	< 6.0	< 6.0	< 11	< 6.0	< 5.0	< 6.0
TITANIUM UG/L	< 3.0	< 6.0	< 3.0	< 3.0	< 6.0	< 5.0	3.0	< 5.0	< 4.0
VANADIUM UG/L	< 5.0	< 3.0	< 3.0	< 3.0	< 3.0	< 3.0	< 3.0	< 5.0	< 6.0
ZINC UG/L	< 300	< 270	< 270	600	< 250	< 480	< 370	0	0
ZIRCONIUM UG/L	ND	< 3.0	< 13	< 13	< 12	< 11	< 8.0	< 5.0	< 8.0

Table 6.--Chemical analyses of water from community systems in Schenectady County, New York (Continued)

	COLUM ON THIS	N(S) LATIT	GS-ASSIGNED UDE-LONGITUDE NUMBER	AND RA	N SITE) NAME AW SOURCE ER SAMPLED			
	А	4249	10073591701	SCHENECT	FADY(C)-WELLS			
SYSTEM(S) ON THIS PAGE A TYPE OF WATER SAMPLED RA DATE11/10	W RAW	A RAW 10/14/71	A RAW 01/13/72	A RAW 04/06/72	A RAW 07/13/72	A RAW 10/18/72	A RAW 01/10/73	A RAW 04/17/73
ALUMINUM UG/L14ARSENIC UG/L0BARIUM UG/L30BERYLLIUM UG/L<BICARBONATE MG/L150	0 38 .50 < .60	8.0 0 40 < 2.0 202	48 1 41 .< 2.0 206	41 0 34 < 2.0 182	7.0 2 30 < 1.0 180	11 1 36 < 2.0 201	48 0 28 < 2.0 180	4.0 0 31 < 2.0 192
BISMUTH UG/L< 5BORON UG/L36CADMIUM UG/L0CALCIUM MG/L48CARBONATE MG/L0	16 0 63	< 6.0 48 0 66 0	< 6.0 33 0 65 0	< 6.0 27 0 62 0	< 5.0 17 0 57 0	< 6.0 25 0 60 0	< 5.0 24 1 61 0	< 5.0 20 0 68 0
	< 6	19 < 6 < 3.0 25	20 < 6 < 6.0 3.0	20 < 3 < 6.0 5.0	14 < 5 < 5.0 4.0	17 < 6 < 6.0 5.0	25 < 5 6.0 1.0	16 < 6 < 6.0 40
	248 .10 0 D < 2.0	0 246 < 3.0 < 6.0	0 253 .10 < 2.0 < 6.0	0 233 .10 < 3.0 < 6.0	0 214 .10 < 5.0 < 11	0 236 .10 < 3.0 < 6.0	0.1 233 .10 < 3.0 < 5.0	.01 240 .10 < 3.0 < 6.0
LEAD UG/L < 5	41 .0 3.0	210 44 14 < 13 < 10	208 39 5.0 < 6.0 < 10	196 47 12 < 3.0 < 10	183 35 9.0 < 5.0 < 10	191 26 12 < 6.0 < 10	193 46 < 5.0 < 5.0 < 10	211 53 12 < 6.0

SYSTEM(S) ON THIS PAGE	TREATED	A	A	A	A	A	A	A	A
TYPE OF WATER SAMPLED.		TREATED	TREATED	TREATED	TREATED	TREATED	TREATED	TREATED	TREATED
DATE		07/12/71	10/14/71	01/13/72	04/06/72	07/13/72	10/18/72	01/10/73	04/17/73
HARDNESS TOTAL MG/L HARDNESS NONCARB MG/L IRON UG/L LEAD UG/L LITHIUM UG/L	158 25 130.0 < 5.0 3.0	191 34 9.0 4.0 2.0	198 34 15 < 13 < 10	176 37 8.0 8.0 < 10	164 39 35 3.0 < 10	166 35 11 37 < 10	196 31 13 < 6.0 < 10	185 43 < 5.0 < 5.0 < 10	198 39 11 < 6.0
MAGNESIUM MG/L	6.9	10	11	10	8.2	8.9	10	9.7	10
MANGANESE UG/L	270	290	460	260	150	170	50	160	210
MBAS MG/L	.01	.01	.02	.01	.01	.02	.02	.02	.02
MERCURY UG/L	< .50	< .50	< .50	< .50	< .50	< .50	< .50	< .50	< .50
MOLYBDENUM UG/L	.70	< 2.0	< 2.0	2.0	< 1.0	< 1.0	< 3.0	< 3.0	< 2.0
NICKEL UG/L NITRATE AS N MG/L NITRITE AS N MG/L NITROGEN NH4 AS N MG/L NITROGEN NH4+ORG-N MG/I	< 3.0 .10 0 	< 6.0 .26 0 .03 .33	15 .10 .15	< 3.0 0 .04	5.0 .10 .02	22 .50 .16	< 6.0 .30 .11	< 5.0 .50 .06	< 6.0 .50 .06
PH UNITS PHENOLS UG/L PHOSPHORUS AS P MG/L POTASSIUM MG/L RUBIDIUM UG/L	7.7 0 .04 1.3 < 2.0	7.7 0 1.5 < .40	7.4 0 .01 1.4	7.7 0 .18 1.5 	7.5 28 .08 1.1 	7.7 .26 1.0 	8.2 .01 1.4 	7.5 .28 1.1 	8.2 .01 1.1
SELENIUM UG/L	3	2	5	2	0	0	4	3	0
SILICA MG/L	5.6	7.4	8.0	6.9	5.2	6.1	7.7	6.3	7.9
SILVER UG/L	<.50	<.60	< .60	< 1.0	< 1.0	< 1.0	< .60	< .50	< .60
SODIUM MG/L	11	10	11	11	9.4	8.6	10	10	11
SPECIFIC COND UMHOS	370	406	420	364	351	358	420	392	387
STRONTIUM UG/L	320	280	300	340	320	260	310	290	300
SULFATE MG/L	28	30	28	28	30	27	31	35	31
TIN UG/L	< 5.0	< 6.0	< 6.0	< 5.0	< 5.0	< 10	< 6.0	< 5.0	< 6.0
TITANIUM UG/L	12	< 6.0	< 3.0	< 3.0	7.0	< 5.0	2.0	< 5.0	< 4.0
VANADIUM UG/L	< 5.0	< 3.0	< 3.0	< 3.0	< 3.0	< 3.0	< 3.0	< 5.0	< 6.0
ZINC UG/L	< 340	< 240	< 270	< 490	< 220	< 450	< 370	< 330	
ZIRCONIUM UG/L	ND	< 3.0	< 13	< 11	< 11	< 10	< 8.0	< 5.0	< 8.0

COLUMN(S) ON THIS PAGE	USGS-A LATITUDE-I NUM		SYSTEM (OR AND RAW OF WATER	SOURCE	
A B	4249100	73591701 73591700	SCHENECTADY SCHENECTADY	(C)-WELLS	
SYSTEM(S) ON THIS PAGE TYPE OF WATER SAMPLED DATE	RAW	A RAW 07/12/71	A RAW 10/14/71	A RAW 04/06/72	A RAW 07/13/72
TOT ORG CARBON MG/L PCB UG/L PCN UG/L	5.0	6.0 0	0	0	0
ALDRIN UG/L CHLORODANE UG/L	0	0 0	0 0	0 0	0 0
DDD UG/L DDE UG/L DDT UG/L	0 0 0	0 0 0	0 0 0	0 0 0	0 0 0
DIAZINON UG/L DIELDRIN UG/L	0	0 < .01	0 < .01	0 0	0 0
ENDRIN UG/L ETHION UG/L HEPTACHLOR UG/L HEPTACHLOR EPOXIDE UG/L	0 0 0	0 0 0 0	0 0 0 0	0 0 0 0	0 0 0 0
LINDANE UG/L	0	0	0	0	0
MALATHION UG/L METHYOXYCHLOR UG/L METHYL PARATHION UG/L METHYL TRITHION UG/L PARATHION UG/L	0 0 	0 0 0 0	0 0 0 0	0 0 0 0	0 0 0 0
TOXAPHENE UG/L RITHION UG/L 2,4-D UG/L 2,4,5-T UG/L SILVEX UG/L	0 0 0 0	0 0 0 0	0 0 0 0	0 0 0 0	0 0 0 0
SYSTEM(S) ON THIS PAGE FYPE OF WATER SAMPLED	B TREATED	B TREATED 07/12/71	B TREATED	B TREATED 04/06/72	B TREATED
COT ORG CARBON MG/L PCB UG/L PCN UG/L	7.0	4.0	0	0	1.0
ALDRIN UG/L CHLORODANE UG/L	0	0 0	0 0	0 0	0 0
DDD UG/L DDE UG/L DDT UG/L DIAZINON UG/L DIELDRIN UG/L	0 0 0	0 0 0 0	0 0 .02 < .01	0 0 0 0	0 0 0 0
NDRIN UG/L STHION UG/L HEPTACHLOR UG/L HEPTACHLOR EPOXIDE UG/L	0 0 0	0 0 0 0	0 0 0 0	0 0 0 0	0 0 0 0
LINDANE UG/L	0	0	0	0	0
MALATHION UG/L METHYOXYCHLOR UG/L METHYL PARATHION UG/L METHYL TRITHION UG/L MARATHION UG/L	0 0 0	0 0 0 0	0 0 0 0 0	0 0 0 0 0	0 0 0 0
COXAPHENE UG/L RITHION UG/L 2,4-D UG/L	 0 0	0 0 0	0 0 0	0 0 0	0 0 0
2,4,5-T UG/L SILVEX UG/L	0 0	0 0	0 0	0 0	0 0

Table 7.--Chemical analyses (pesticides and related constituents) of water from City of Schenectady, New York, wells