

WATER QUALITY OF STREAMS AND SPRINGS, GREEN RIVER BASIN, WYOMING

By

Lewis L. DeLong

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CONVERSION FACTORS

The following factors may be used to convert the inch-pound units
used in this report to the International System of Units (SI):

Multiply	By	To obtain
cubic foot per second	0.02832	cubic meter per second
ton per day	0.9072	megagram per day
mile	0.6214	kilometer

STATION NUMBERS

Routine surface-water sampling stations are identified by 8-digit station numbers. The complete station number, such as 09188500, includes the first two digits "09" that refer to the major drainage basin, and the remaining six digits "188500" that refer to individual station location. Progressively higher location numbers refer to locations progressively farther downstream.

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Abstract

Data on salinity, phosphorus, and trace elements in streams and springs within the Green River basin in Wyoming are summarized. Relative contributions of salinity are shown through estimates of annual loads and average concentrations at 11 water-quality measurement sites for the 1970-77 water years. A hypothetical diversion of 20 cubic feet per second from the Big Sandy River would decrease dissolved-solids concentrations in the Green River at Green River, Wyoming. This effect would be greatest during the winter months, decreasing dissolved-solids concentrations as much as 13 percent. Decreases in dissolved-solids concentrations during the remainder of the year generally would be less than 2 percent.

Unlike the dilution effect that overland runoff has on perennial streams, runoff in ephemeral and intermittent streams within the basin was found to be enriched by the flushing of salts from normally dry channels and basin surfaces.

Relative concentrations of sodium and sulfate in streams within the basin appear to be controlled by availability, whereas calcium concentrations appear to be controlled by solubility. A downstream trend

of increasing relative concentrations of sodium, sulfate, or both with increasing dissolved-solids concentrations was evident in all streams sampled.

Estimates of total phosphorus concentrations at water-quality measurement sites indicate that phosphorus is removed from water of the Green River as it passes through Fontenelle and Flaming Gorge Reservoirs. Total phosphorus concentrations at some stream sites are either directly or inversely related to streamflow, but at most sites a simple relation is not discernable.

Trace-element concentrations in many of the water samples from streams and springs were less than analytical-detection limits. A ranking procedure was used to calculate cumulative probabilities of concentrations in distributions affected by analytical detection limits. Thus, the number of samples with concentrations less than detection limits and the number and concentrations of samples with concentrations greater than detection limits were used to provide more realistic estimates of means and standard deviations.

1.0 INTRODUCTION

This report summarizes water-quality data for streams and springs in the Green River basin in Wyoming.

Demand for water in the Green River basin in Wyoming (fig. 1.0-1) is rapidly increasing due to development of extensive coal, oil, gas, and trona resources. The potential also exists for development of extensive oil-shale resources. The increasing demand for water and, consequently, an increasing demand for water-quality information has led to the initiation and expansion of data-collection programs by many State and Federal agencies. This report, a component of a complete basin study described by Lowham and others (1976), summarizes water-quality data and associated interpretations primarily for persons concerned with planning and managing water development and monitoring the resulting

effects on surface-water quality. Data analyzed are from samples collected between October 1965 and June 1979. Data are available from computer storage (U.S. Geological Survey, 1974) and may be found in publications such as Water Resources Data for Wyoming (U.S. Geological Survey, 1976; 1977).

Salinity, phosphorus, and trace elements are described in separate sections followed by a section describing methods used in this report to summarize water-quality data affected by analytical-detection limits.

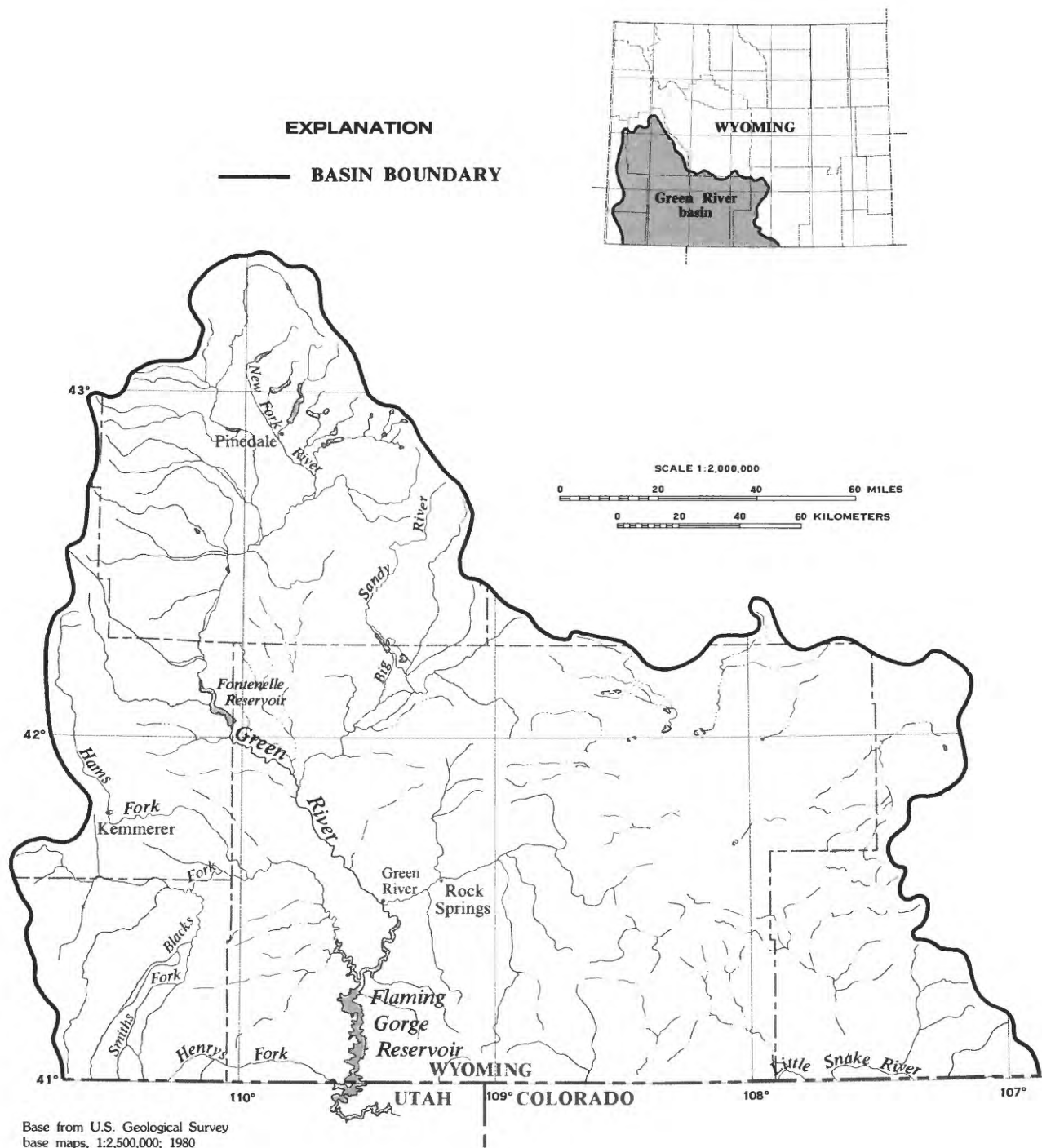


Figure 1.0-1 Location of the Green River basin in Wyoming.

1.0 INTRODUCTION

2.0 SALINITY

2.1 Dissolved-Solids Loads and Average Concentrations

Relative contributions of salinity may be seen in loads estimated at water-quality measurement sites on the Green River and its major tributaries.

Water quality, specifically salinity, is an important factor in determining water use and in assessing possible impacts of those uses with time. Development of extensive energy resources in the Green River basin and other parts of Wyoming could result in a significant increase in water consumption from the Green River (Wyoming Water Planning Program, 1970).

Quantitative description of salinity in the Green River and its major tributaries is useful in evaluating existing water quality as well as projecting impacts of proposed or existing surface-water developments. Dissolved-solids concentrations and loads in this report are estimated from periodic samples and

streamflow records using a regression model that relates dissolved-solids concentrations to daily streamflow and time of the year (DeLong, 1977). Considerable variability in annual loads and average concentrations at individual stations is apparent (figs. 2.1-1 and 2.1-2). Generally smaller annual loads and larger average concentrations during the 1977 water year were a result of less than normal runoff (table 2.1-1). The average discharge during the 1977 water year at station 09217000, Green River near Green River, Wyoming, was less than one-half the average discharge during the preceding 26 years (U.S. Geological Survey, 1977).

Table 2.1-1 Dissolved-solids loads and average concentrations, 1970-77 water years.

EXAMPLE: 74,020 Tons per year
166 Milligrams per liter

Station	1970	1971	1972	1973	1974	1975	1976	1977
09188500	74,020 166	95,080 140	102,400 155	81,230 183	94,260 158	74,910 146	88,120 165	55,870 213
09205000	52,890 97	68,730 80	79,230 74	67,670 94	68,900 86	66,120 84	65,990 85	39,660 129
09209400	227,200 188	378,500 169	442,400 175	291,800 199	335,400 192	294,500 166	342,900 184	144,600 220
09211200	299,500 232	443,000 214	543,100 213	347,500 240	436,400 226	379,600 217	439,300 226	180,900 226
09216000	68,240 2,252	75,870 1,727	92,490 1,057	97,690 1,537	102,300 1,383	92,440 1,420	93,760 1,563	75,200 2,186
09216050	----- -----	----- -----	----- -----	180,100 2,274	199,500 2,226	163,600 2,075	169,100 2,279	136,400 2,934
09217000	456,000 353	648,600 284	768,100 280	574,900 344	646,200 315	568,600 307	640,500 313	306,700 452
09222000	96,450 716	130,200 669	126,400 648	143,300 670	136,500 644	135,200 574	88,960 895	27,690 1,533
09224700	134,500 565	273,800 463	244,900 475	228,700 563	219,900 533	214,700 516	191,200 630	33,580 1,175
09229500	54,680 649	70,230 527	50,820 577	82,950 530	49,120 659	71,850 463	35,710 784	18,090 942
09234500	986,900 466	683,000 472	1,260,000 464	1,422,000 495	977,800 506	1,316,000 524	1,384,000 509	1,340,000 477

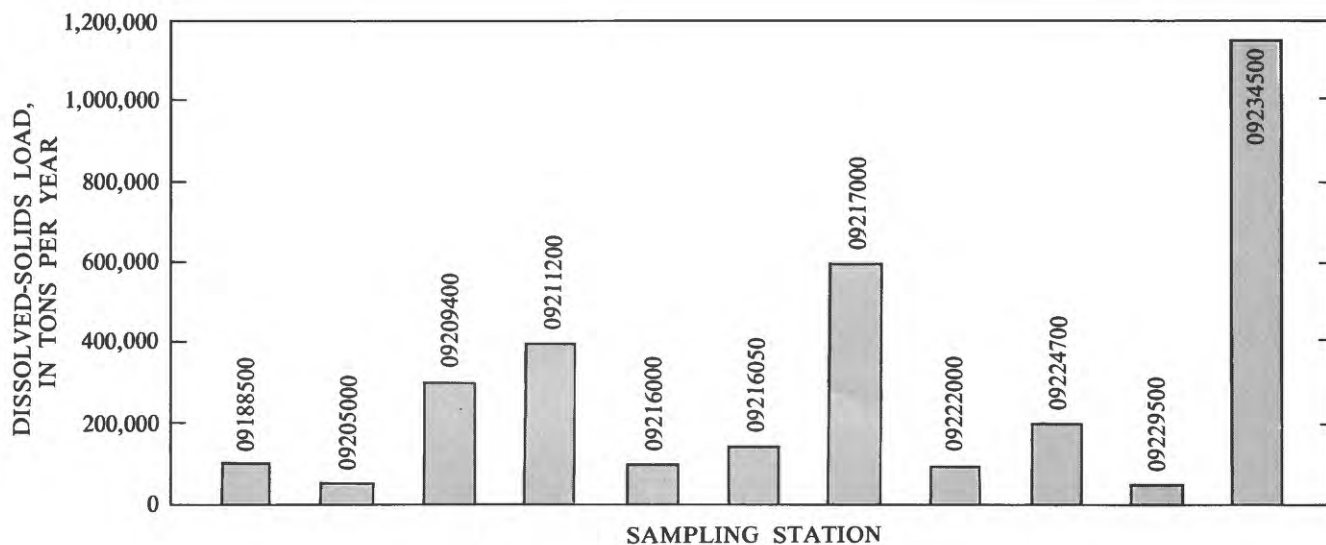


Figure 2.1-1 Average dissolved-solids loads, 1970-77 water years.

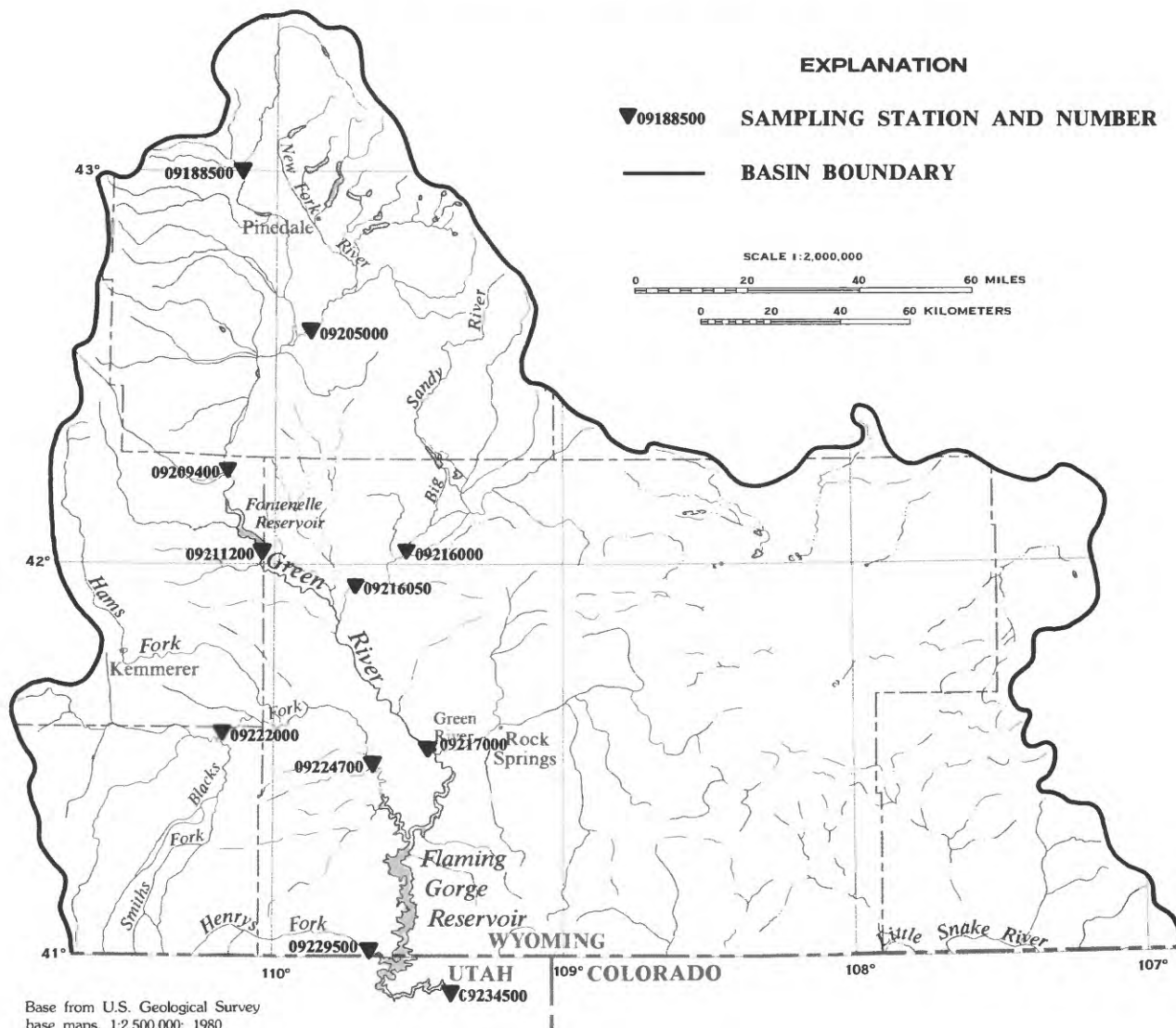


Figure 2.1-2 Location of sampling stations.

2.0 SALINITY

2.1 Dissolved-Solids Loads and Average Concentrations

2.0 SALINITY--Continued

2.2 Estimating Effects of Water Development

Probable effects of alternative development plans on surface-water salinity may be evaluated by comparing estimated changes in dissolved-solids concentrations downstream from proposed development.

Consumptive use of surface water for energy-related development in southwestern Wyoming (such as diversions for powerplants, coal-slurry pipelines, and municipalities) is likely to have a greater effect on surface-water salinity than the actual mining of energy minerals (DeLong, 1978). Existing and planned mines in the Green River basin lie mainly in plains areas characterized by intermittent and ephemeral streams. Mining plans commonly assume total consumption of water within the area of the mine during both mining and reclamation. Even without total consumption, mines in the basin would not be persistent sources of salinity owing to the minimal precipitation and subsequent runoff from the mining areas. Unusually intense storms could cause saline runoff, but such storms are localized; and the runoff would not have a significant effect on major drainages.

Water would most likely be diverted from major perennial streams of the basin. The probable effect of proposed diversions made from different points within the stream system may be evaluated by es-

timating dissolved-solids concentrations at each point of diversion, computing the dissolved-solids load removed by diversion, and superimposing the changes in streamflow and dissolved-solids loads on estimates made for common downstream locations.

For example, a constant diversion of 20 cubic feet per second from the Big Sandy River at Gasson Bridge without return flow could potentially decrease dissolved-solids concentrations downstream in the Green River at Green River, Wyoming by as much as 13 percent during the winter months (fig. 2.2-1). Potential decreases in concentrations during the remainder of the year generally would be less than 2 percent. Concentrations would decrease downstream in the Green River because the concentrations in the Big Sandy River at the point of diversion are greater than the concentrations in the Green River. Conversely, an equivalent diversion from the mainstem of the Green River upstream from the Big Sandy River could result in a slight increase in dissolved solids downstream at the town of Green River.

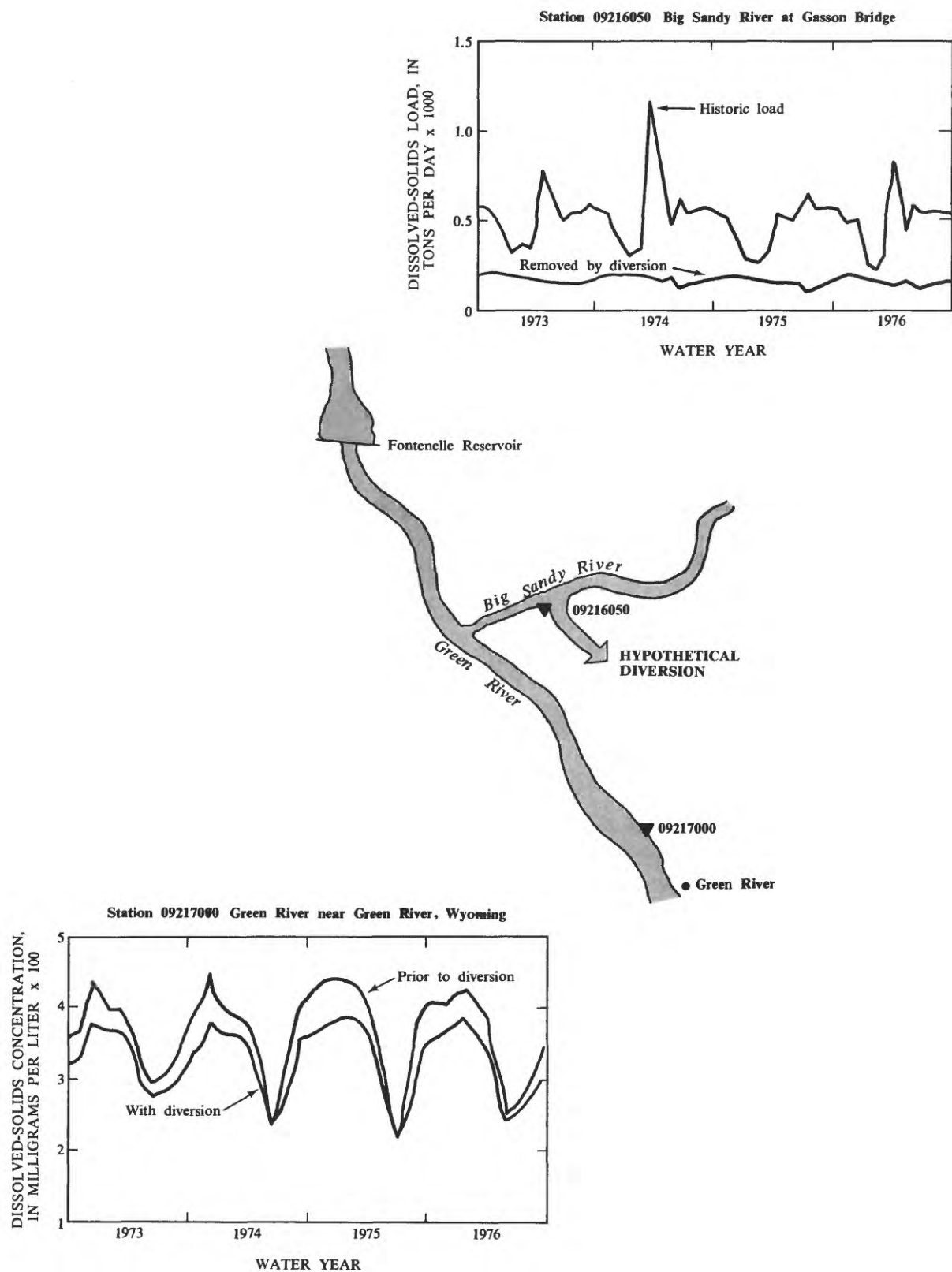


Figure 2.2-1 Estimation of downstream effect on dissolved-solids concentrations of a hypothetical diversion from the Big Sandy River.

2.0 SALINITY--Continued

2.2 Estimating Effects of Water Development

2.0 SALINITY--Continued

2.3 Dissolved-Solids Concentrations in Ephemeral and Intermittent Streams

Unlike the dilution effect that overland runoff has on the baseflow of most perennial streams, runoff in ephemeral and intermittent streams is enriched by the "flushing" of salts from normally dry channels and other basin surfaces.

The variable nature of flow in ephemeral and intermittent streams significantly affects corresponding water quality. During periods of infrequent precipitation and runoff, salts accumulate in channels and on other basin surfaces. Concentration of dissolved solids in the downstream edge of runoff reflects the availability of readily soluble salts. Dissolved-solids concentration continues to increase during runoff until salts are sufficiently flushed from inundated surfaces allowing the dilution effect of continued runoff to prevail (fig. 2.3-1).

The composite effect of overland runoff on perennial streams in the Green River basin, as shown by DeLong (1977), is dilution. Headwaters of most perennial streams within the basin are in mountainous areas characterized by greater relative precipitation and runoff. Dissolved-solids concentrations in overland runoff from these areas typically are less than the base-flow concentrations of recipient peren-

nial streams. Dissolved-solids concentrations typically are greater in snowmelt runoff from lower elevations characterized by ephemeral and intermittent streams than in snowmelt runoff from the mountainous areas. The fact that snowmelt runoff at lower elevations generally precedes that from the mountainous areas contributes to a seasonal trend in the relation between streamflow and dissolved-solids concentration of perennial streams (DeLong, 1977).

Data to illustrate the enrichment effect and flushing action were collected along Lost Creek (fig. 2.3-2) on April 1-2, 1976 (fig. 2.3-3). A plot of dissolved-solids concentration versus downstream distance shows the flushing action typical of the first flow of snowmelt runoff (fig. 2.3-4). The greatest concentration on both days was observed near the downstream edge of the water as it moved down the previously dry channel.

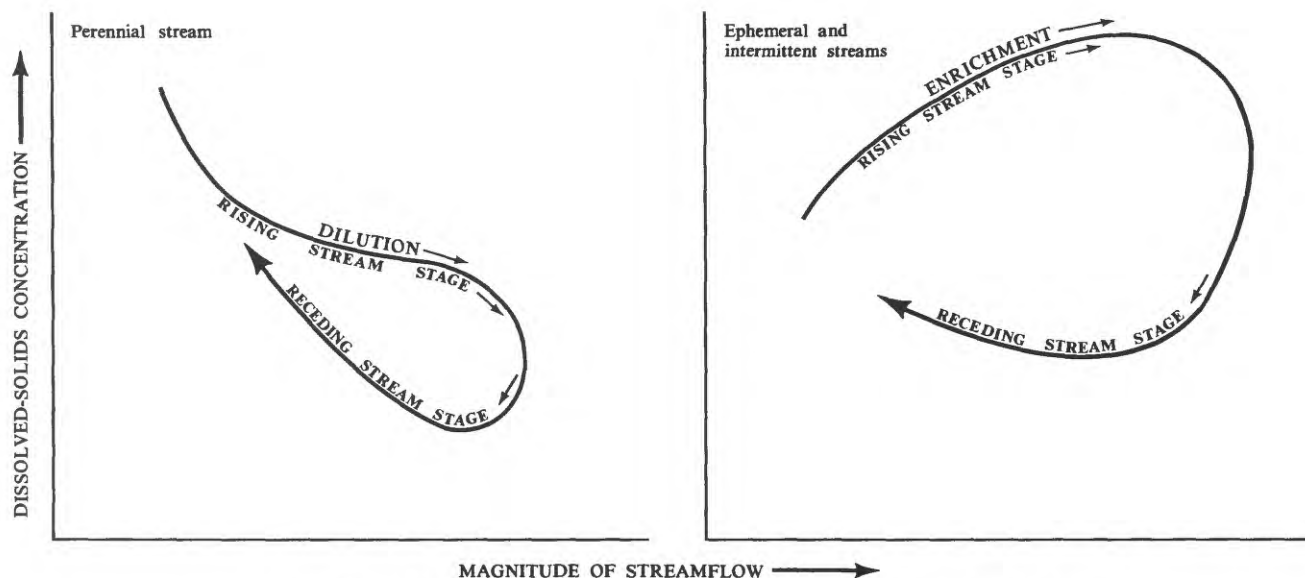


Figure 2.3-1 Comparison between dilution and enrichment effects.



Figure 2.3-3 Sampling leading edge of snowmelt runoff in Lost Creek, April 1, 1976.

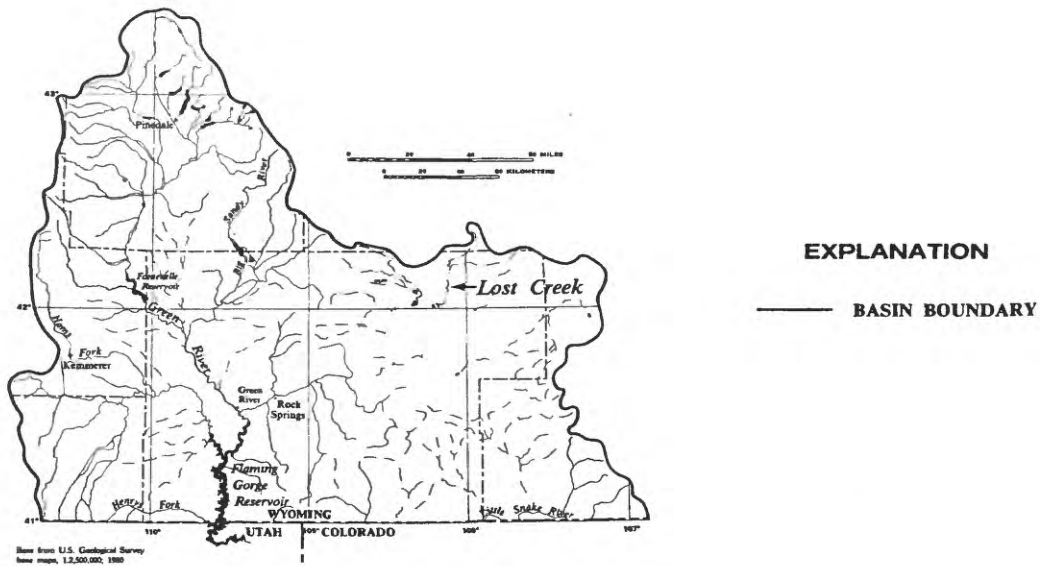


Figure 2.3-2 Location of Lost Creek.

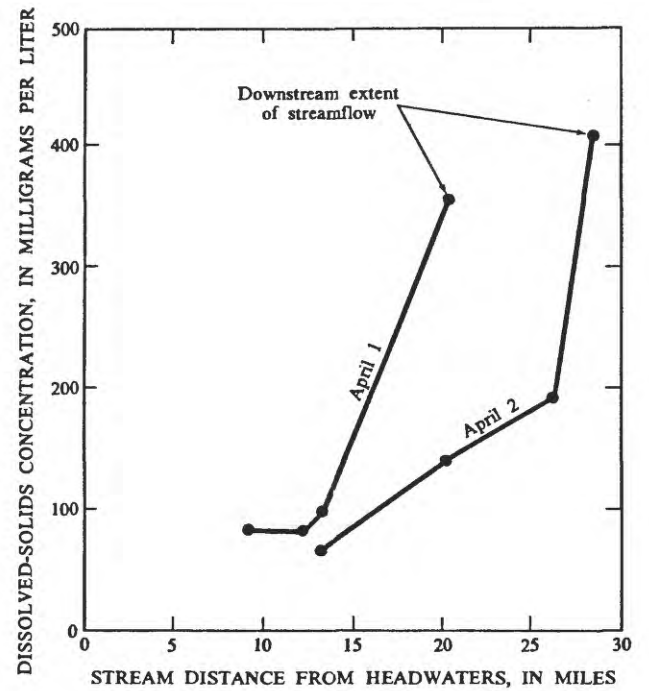


Figure 2.3-4 Flushing of Lost Creek Channel by snowmelt runoff, 1976.

2.0 SALINITY--Continued

2.3 Dissolved-Solids Concentrations in Ephemeral and Intermittent Streams

2.0 SALINITY--Continued

2.4 Relative Calcium, Sulfate, and Sodium Concentrations

Relative concentrations of calcium in streams and springs in the Green River basin generally are controlled by solubility, whereas sulfate and sodium concentrations generally reflect sources.

Chemical equilibrium calculations (similar to those demonstrated by Hem, 1966, p. 64-77; 1970, p. 250-255) indicate that most stream water in the Green River basin is saturated or nearly saturated with respect to calcite (calcium carbonate). A similar condition exists in the major rivers of Russia (Alekin and Moricheva, 1957) as reported by Hem (1961, p. 14-15). Because calcite saturation typically persists in the study area from near the headwaters through downstream reaches, stream water dissolves very little calcium carbonate en route. In fact, calcium dissolved from other sources such as calcium sulfate would tend to precipitate as calcium carbonate. Dissolution of calcium sulfate and subsequent precipitation of calcium carbonate is hypothesized to account for a relative increase of sulfate concentration in Flaming Gorge Reservoir (Bolke, 1979, p. 3234). The streams in the study area, however, are not saturated with respect to other ions. Controlled

largely by availability and dissolution rates, concentrations of these ions continue to increase downstream (fig. 2.4-1). Initially large relative sulfate concentrations in the Green River are caused by sulfate concentrations of springs tributary to the head waters. Dilution by less concentrated tributaries causes a short-lived decrease in relative sulfate concentration that gradually is overcome by a continued increase in the absolute sulfate concentration downstream. The downstream trend of increasing relative concentrations of sulfate, sodium, or both with an overall increase in dissolved-solids concentration is typical of all streams sampled within the basin.

Analyses of water samples obtained from springs exhibit a similar trend (fig. 2.4-2); the largest concentrations of sulfate and sodium occur in samples with the largest dissolved-solids concentrations.

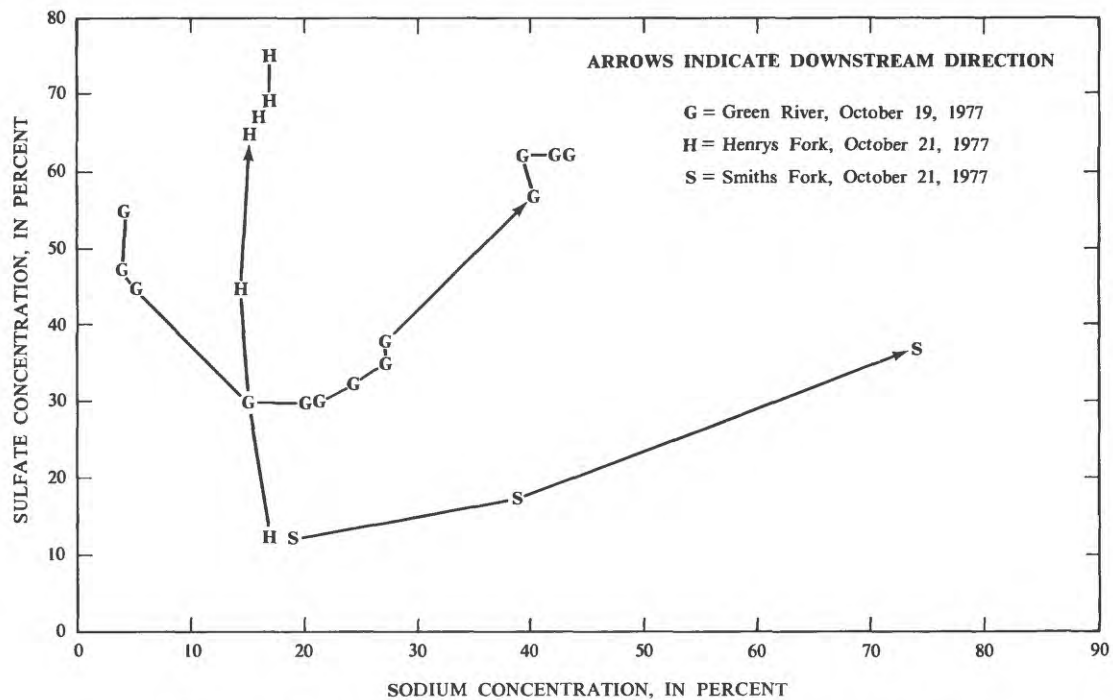


Figure 2.4-1 Downstream increase in sulfate and sodium concentrations.

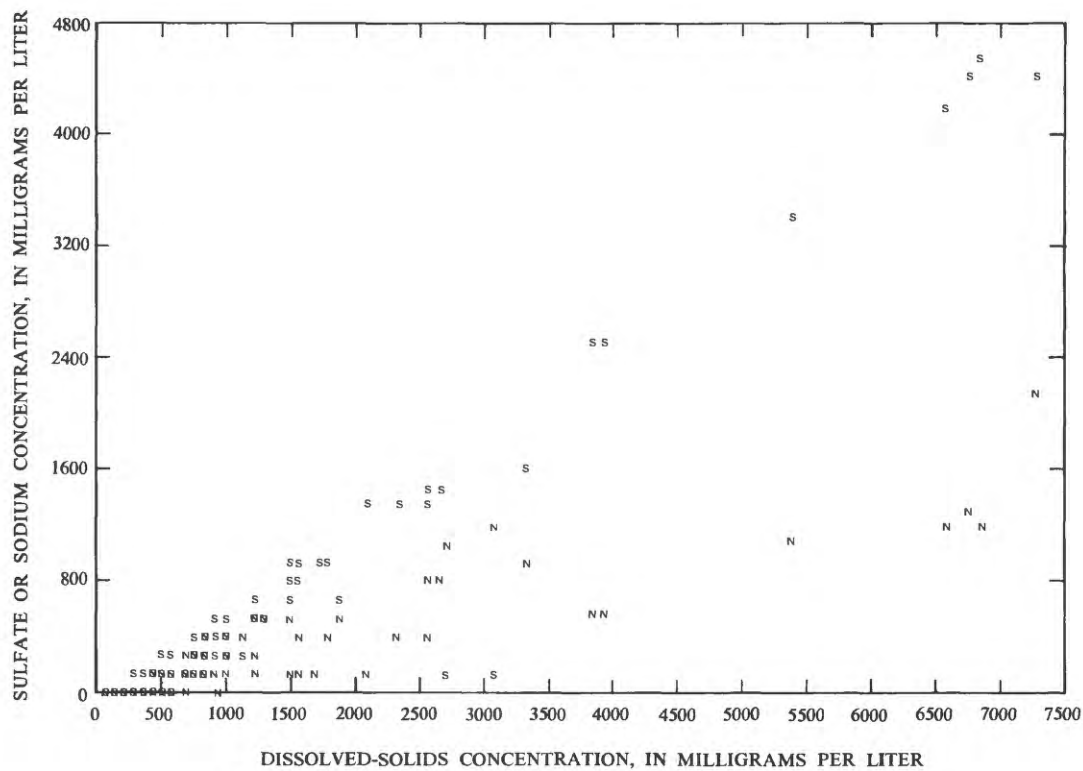


Figure 2.4-2 Relation of sulfate (S) and sodium (N) concentrations to dissolved-solids concentrations in samples from springs.

2.0 SALINITY--Continued

2.4 Relative Calcium, Sulfate, and Sodium Concentrations

2.0 SALINITY--Continued

2.5 Chemical Composition of Water in Samples from Springs

Variation of chemical composition and overall concentrations of water from springs is as great within individual geologic units as it is between different geologic units.

Water samples of springs flowing from the same geologic unit did not exhibit a characteristic chemical composition. The range of overall concentrations (fig. 2.5-1) and chemical composition (fig. 2.5-2) was as great in samples collected from springs scattered areally over a particular geologic unit as it was

between spring samples collected from different geologic units. It generally would not be possible to determine the geologic unit from which a spring sample was collected by chemical composition and overall concentration.

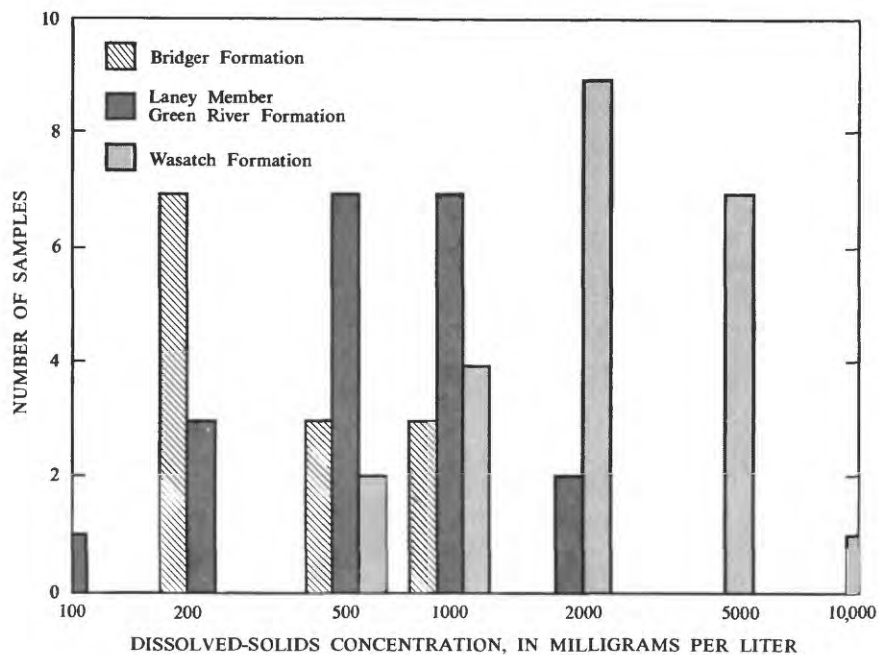


Figure 2.5-1 Dissolved-solids concentrations in samples collected from springs flowing from the Bridger and Wasatch Formations and the Laney Member of the Green River Formation.

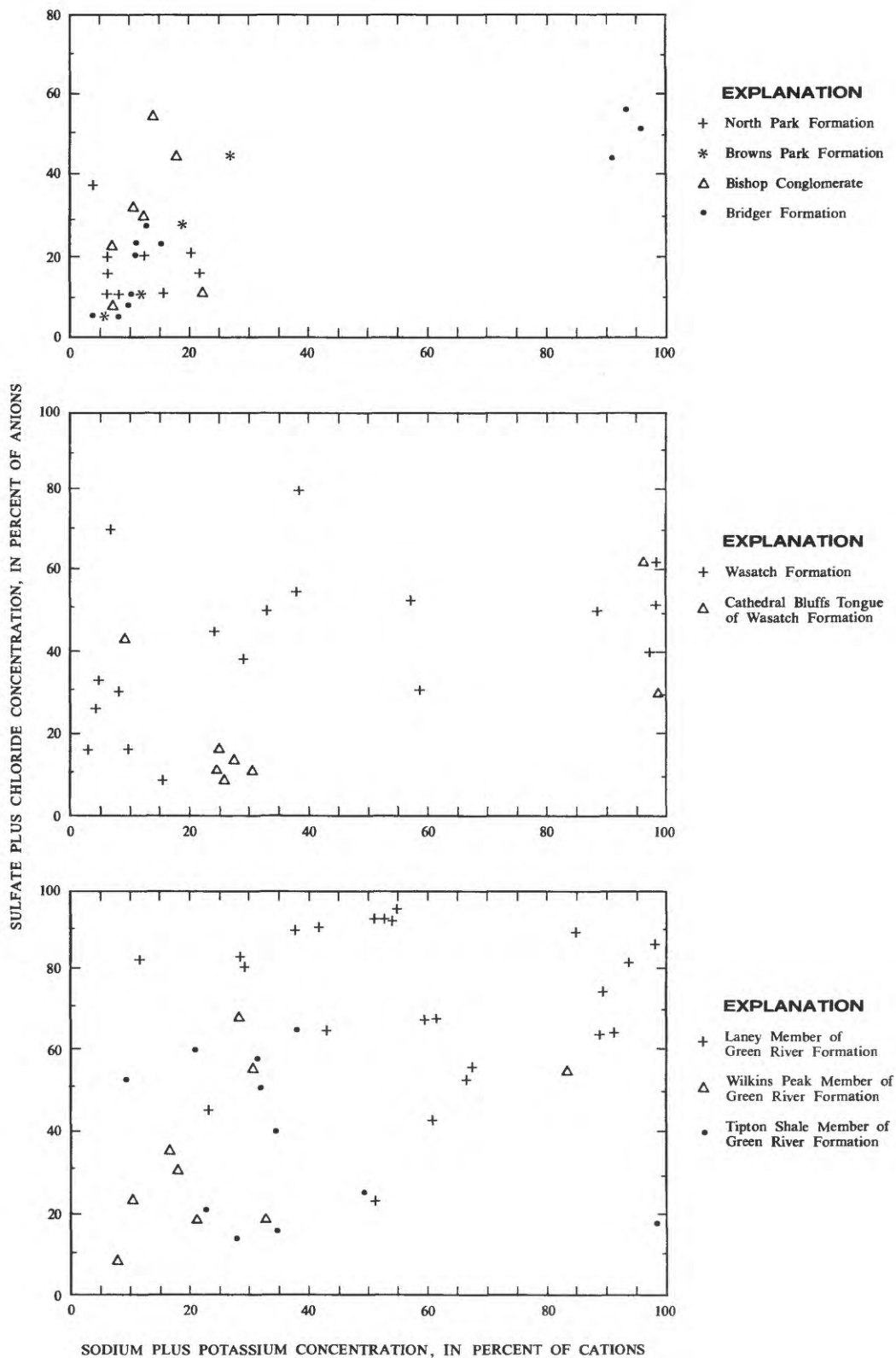


Figure 2.5-2 Relation of sulfate plus chloride concentrations to sodium plus potassium concentrations in samples from springs.

2.0 SALINITY--Continued

2.5 Chemical Composition of Water in Samples from Springs

3.0 PHOSPHORUS

3.1 Effect of Reservoirs

Phosphorus is removed from the Green River as the water passes through Fontenelle and Flaming Gorge Reservoirs.

Control of phosphorus in surface waters of the Green River basin has become an important water-quality issue primarily as a result of eutrophication studies in the basin and increasing algal growth in Flaming Gorge Reservoir. The reservoir has been classified eutrophic at points within the Blacks Fork and Green River arms and at a point downstream from their confluence (U.S. Environmental Protection Agency, 1977, p.12). Algal blooms in the arms of the reservoir have grown increasingly larger during the last decade (CH2M Hill, 1977, p.13).

Estimating present phosphorus loading is a useful step in assessing probable effects of future development and controls on eutrophication. The effectiveness of Fontenelle and Flaming Gorge Reservoirs as phosphorus traps is readily apparent in estimates of loads and concentrations at adjoining stations (figs. 3.1-1 and 3.1-2). Estimates of phosphorus loads contributed to the reservoirs from direct runoff and other tributaries would be necessary to estimate phosphorus storage rates in the reservoirs. However, considering the quantities of phosphorus from the Green River alone that may have been stored in the reservoirs since their completions, it seems unlikely that algal growth would be a

simple function of phosphorus loading in any one season. Further study would be necessary to quantify dominant transport and growth mechanisms in the reservoirs. Results would be useful in examining feasibility of various control measures and effects of alternative development plans.

The increase in phosphorus load from station 09217000, Green River near Green River, Wyoming, to station 09217010, Green River below Green River, Wyoming is presumably a result of inflow between the two stations. Bitter Creek joins the Green River between the two stations, and sewage effluent is contributed both through Bitter Creek and directly to the Green River between the two stations.

Average loads (table 3.1-1) and geometric-mean concentrations (table 3.1-2) at individual stations were estimated by methods described in section 5. Logarithms of measured concentrations and calculated loads were used to transform data to the normal form. All phosphorus concentrations used in this report are total concentrations resulting from analysis of the total water-sediment mixture in the samples.

Table 3.1-1 Average total phosphorus loads, 1974-78 water years.

[L_Mean = Mean of logarithms (Base 10);
L_SD = Standard deviation of logarithms;
Avg_Load = Average load, in tons per day]

Station	L_Mean	L_SD	Avg_Load
09188500	-1.981486	0.651740	0.031
09205000	-1.838354	.707267	.053
09209400	-1.612998	.882251	.180
09211200	-1.603068	.551888	.055
09214500	-2.946421	1.156198	.027
09216000	-2.794096	.802085	.008
09216050	-2.270304	.821662	.030
09216300	-1.641843	.742699	.095
09217000	-1.013692	.627777	.269
09217010	-.694740	.572333	.473
09222300	-2.906847	.607950	.003
09222400	-2.360567	1.170525	.121
09224050	-1.612482	.413497	.038
09224450	-2.523287	1.188375	.107
09234500	-1.253458	.556600	.125

3.0 PHOSPHORUS--Continued

3.2 Relation to Streamflow

Total phosphorus concentration at some stream sites is related to streamflow.

Total phosphorus concentration in a stream may vary as a result of changes in streamflow (fig. 3.2-1). Dilution occurs if sources of additional water to the stream contain less phosphorus concentration relative to initial stream concentration. Conversely, enrichment of the stream occurs when additional water contains greater relative concentration. Increasing streamflows accompanied by rising stream stages also may increase total phosphorus concentration by capturing phosphorus previously stored in channels. For example, total phosphorus concentration in Salt Wells Creek, an intermittent plains stream at the southern end of the Green River basin,

is directly related to sediment concentration (Lowham and others, 1982).

Total phosphorus concentration in the Green River within Wyoming was not found to be related to magnitude of streamflow. In the Green River, as in most of its major tributaries, the lack of relation between total phosphorus and streamflow may be thought of as an averaging of dilution and enrichment effects. It should also be noted that phosphorus concentration is affected by its extensive use in biological cycles as a nutrient.

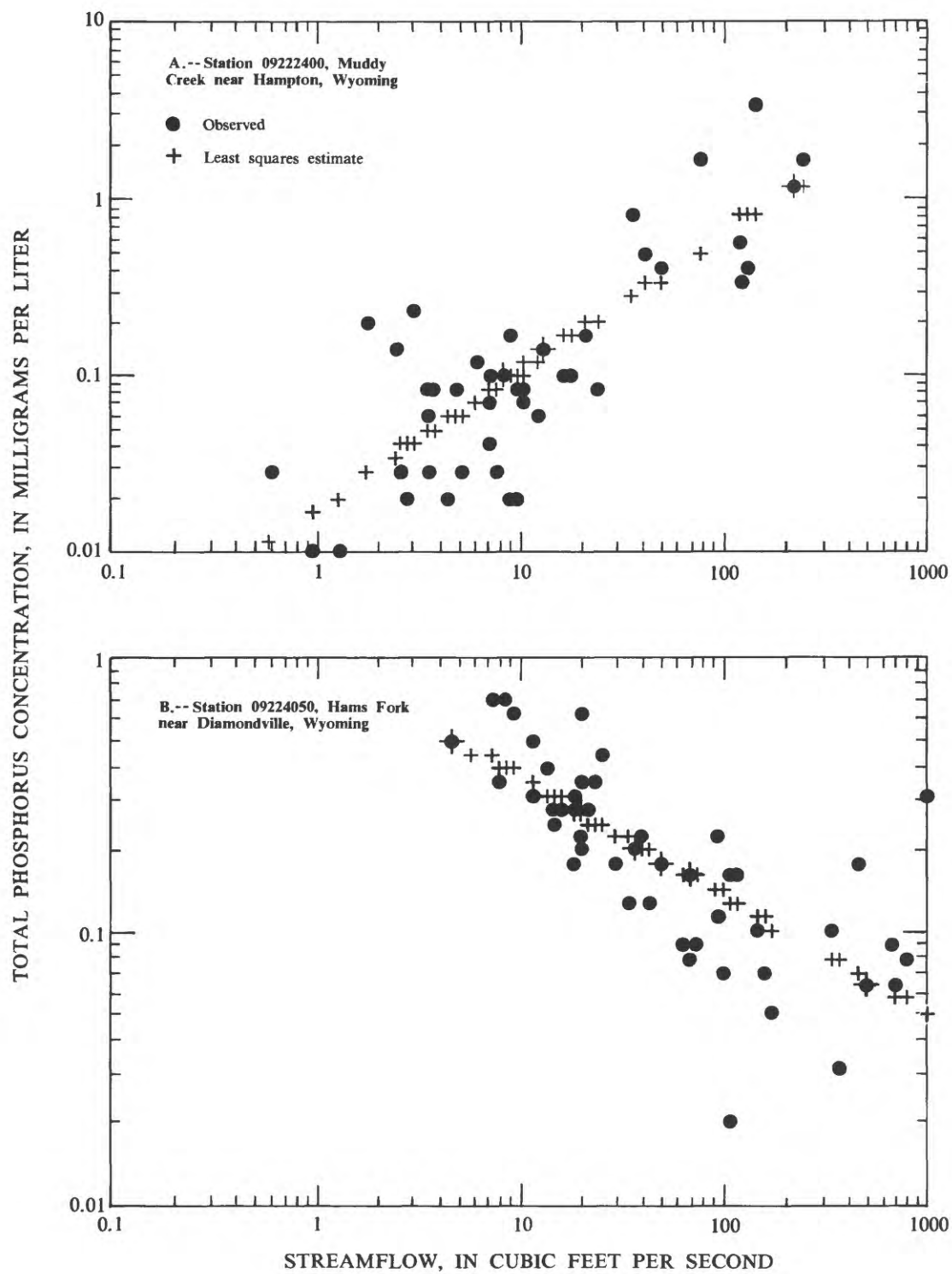


Figure 3.2-1 Relation of total phosphorus concentrations to streamflow.

4.0 TRACE ELEMENTS

Trace-element concentrations in many of the samples analyzed were less than detection limits.

Trace elements, as the term implies, normally are found at minute or "trace" concentrations in natural water. Consequently, trace-element concentrations are reported in micrograms per liter as opposed to the major constituents in water which are reported in milligrams per liter. As shown in the bar graph (fig. 4.0-1), a significant number of the samples contained concentrations less than prevalent analytical detection limits. Detection limits may vary among elements or analytical procedures or both. Trace-element concentrations summarized in this report were determined by atomic-absorption spectrophotometry (Skougstad and others, 1979). In general, detection limits range from 1 to 10 micrograms per liter for the dissolved analyses and from 1 to 100 for the total analyses. Although the exact concentration of a particular constituent in a sample may be less than detection limits, the sample can still be used in statistical analysis. The knowledge that the concentration is less than a certain value is useful information, obtained with considerable effort and expense. Section 5.0 describes the method used in this report to include data with values less than detection limits in the determination of means and standard deviations.

What is a typical value of total arsenic concentration? In table 4.0-1, it can be seen that geometric means of total arsenic concentrations at water-quality measurement sites differ by an order of magnitude. Numbers in columns represented by downstream-station numbers (09209400, for example; fig.

4.0-2) are geometric means of samples collected periodically at the same site. The columns headed "Misc-SW" (miscellaneous surface water) and "Springs" represent geometric means of samples collected one per site. Geometric means are used, rather than arithmetic means, because they provide a better measure of central tendency. Frequency distributions of logarithms of the data show more symmetry than do frequency distributions of the untransformed data. Summary statistics from which the table of geometric means was derived are tabulated in section 8.0 of this report.

Abbreviations

AS	Arsenic
B	Boron
BA	Barium
CD	Cadmium
CR	Chromium
CU	Copper
DI	Dissolved
FE	Iron
LI	Lithium
MN	Manganese
MO	Molybdenum
NI	Nickel
PB	Lead
SE	Selenium
TO	Total
V	Vanadium
ZN	Zinc

Table 4.0-1 Geometric means of trace-element concentrations, in micrograms per liter.

Element	Station number																	MISC SW	SPRINGS
	09209400	09211200	09216000	09216527	09216545	09216562	09216565	09216576	09216750	09216810	09216880	09217000	09217010	09222300	09222400	09224050	09235300		
Arsenic, dissolved	-	-	2	-	1	-	-	1	-	1	8	1	1	1	2	1	1	2	1
Arsenic, total	1	1	3	5	4	2	3	2	23	4	12	1	1	2	3	1	6	2	7
Barium, dissolved	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	87
Barium, total	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	10
Boron, dissolved	-	-	-	103	217	482	145	117	605	1,870	448	58	66	1,710	1,240	49	162	-	59
Cadmium, dissolved	1	1	2	-	-	2	1	1	-	2	1	1	1	1	1	1	2	-	0
Cadmium, total	6	9	7	-	6	7	7	-	5	-	6	6	5	6	5	5	6	-	3
Chromium, dissolved	-	-	-	-	-	-	-	-	-	10	6	-	-	-	-	-	-	-	6
Chromium, total	5	-	8	21	7	9	8	-	37	14	9	-	5	6	9	6	8	-	17
Copper, dissolved	2	2	3	2	3	3	1	1	-	3	3	2	1	4	2	1	4	1	1
Copper, total	14	11	13	19	13	14	13	-	14	17	16	8	12	12	8	14	-	-	18
Iron, dissolved	25	21	20	104	69	-	75	62	45	80	69	21	21	58	65	43	74	59	58
Lead, dissolved	1	2	2	-	-	3	1	2	-	4	3	1	3	-	3	-	2	1	1
Lead, total	75	-	63	-	47	35	34	-	93	62	55	-	-	39	46	-	66	-	34
Lithium, dissolved	6	5	35	-	63	213	105	50	-	381	100	9	10	78	130	10	72	-	43
Lithium, total	9	8	37	43	63	153	42	60	222	393	101	12	12	95	129	12	93	-	58
Manganese, dissolved	8	8	14	24	14	261	165	116	-	910	168	7	9	30	15	51	129	26	12
Molybdenum, dissolved	1	-	7	-	10	7	5	-	-	9	2	1	2	15	13	-	6	-	2
Molybdenum, total	2	1	7	2	10	5	6	4	5	9	4	1	2	14	10	1	6	-	3
Nickel, dissolved	2	2	1	-	3	7	-	-	-	11	5	2	2	8	6	2	4	-	3
Nickel, total	24	-	33	25	23	25	17	-	68	50	29	-	-	27	25	-	24	-	60
Selenium, dissolved	-	-	2	-	-	3	-	-	1	1	-	-	-	4	3	-	-	-	1
Selenium, total	-	-	2	-	1	3	-	1	4	1	1	-	-	5	2	-	1	-	2
Vanadium, dissolved	0	1	2	-	-	-	-	-	-	6	3	0	1	-	1	0	-	1	1
Zinc, dissolved	5	5	-	-	-	-	-	-	-	22	16	8	9	16	9	-	12	-	7
Zinc, total	19	16	29	63	34	40	28	22	256	53	47	15	13	36	31	17	48	10	60

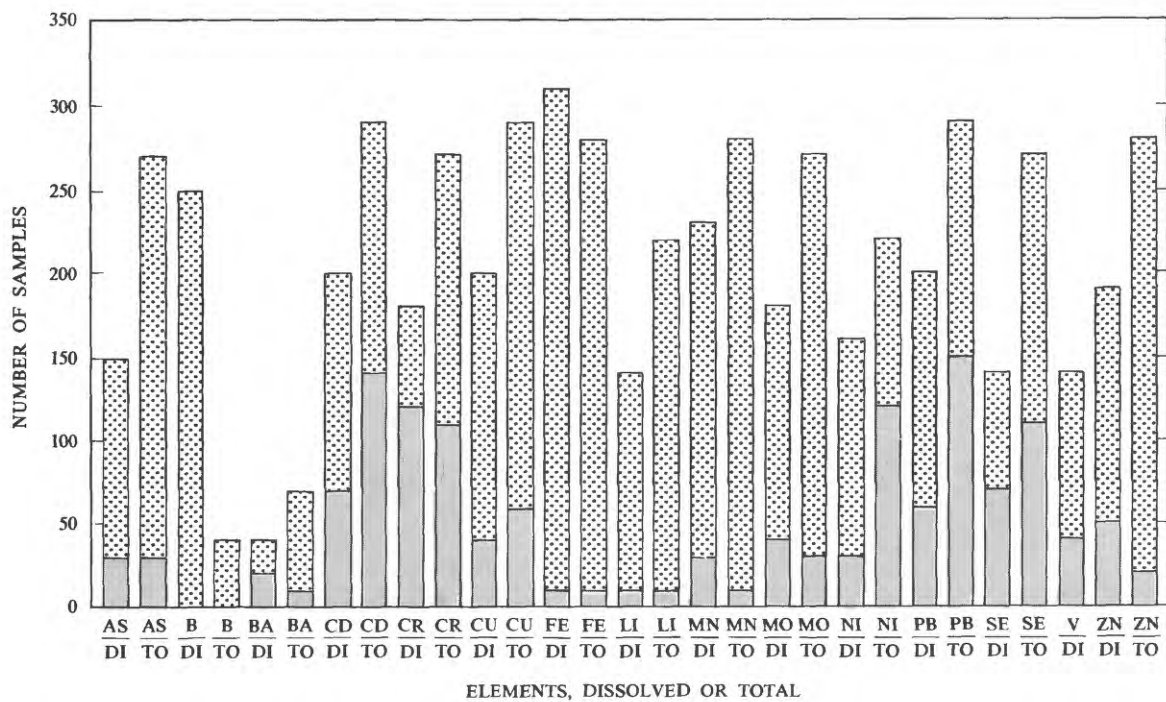


Figure 4.0-1 Relation of number of samples with concentrations greater than detection limits (pattern) to number of samples with concentrations less than detection limits (shaded).

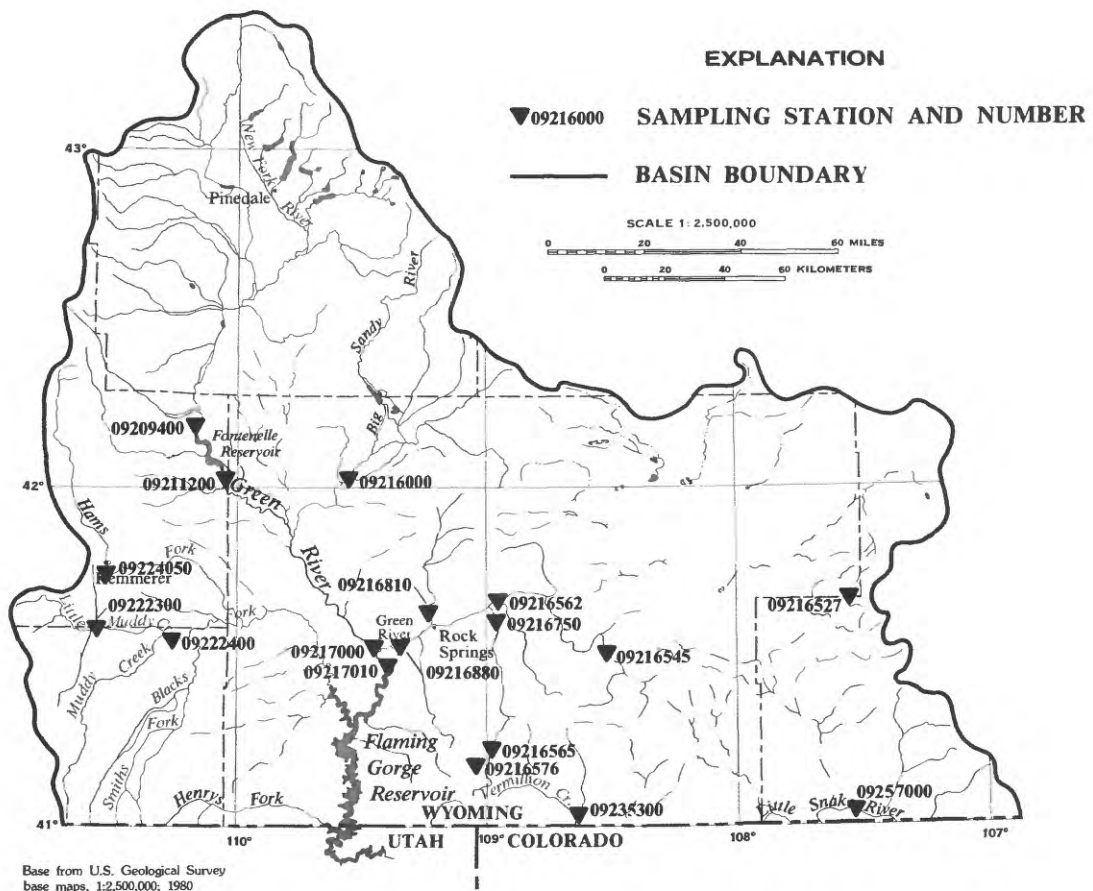


Figure 4.0-2 Location of sampling stations.

4.0 TRACE ELEMENTS

5.0 ESTIMATION OF MEANS AND STANDARD DEVIATIONS OF DISTRIBUTIONS AFFECTED BY ANALYTICAL DETECTION LIMITS

5.1 Censored Sample Distributions

Water-quality data are termed censored data when the constituent concentrations are less than analytical detection limits and cannot be quantified.

Limited analytical sensitivity often presents a problem in estimating means and standard deviations. Many constituents exist in natural waters at concentrations less than the detection limits of prevalent analytical procedures (fig. 5.1-1). For example, if the detection limit for a particular constituent and attendant analytical procedure is 100 micrograms per liter, concentrations of less than 100 micrograms per liter would be qualified as "F100," "not detected," or simply "0." Data containing such qualified values are referred to as censored. The number of data that are less than the detection limit is known, but their individual magnitudes are unknown. Because of this, averages of the logarithmically transformed or untransformed data cannot be computed directly.

Because censored sample distributions commonly result from the measurement of processes and abundance in nature, many methods have been used to statistically analyze censored distributions. If the qualified values compose less than 20 percent of the

data, they may be replaced by arbitrary values directly outside the range of detection of the analytical method used. Seven-tenths of the lower limit of detection was used by Miesch (1976). Methods given by Cohen (1959) and summarized by Miesch (1967) allow statistical analysis of censored-data distributions composed of greater than 20-percent qualified values. Jennings and Benson (1969) used a theorem of conditional probability to estimate probability of occurrence of annual floods from censored data. A ranking procedure, described in section 5.2, was used in this report to calculate cumulative probabilities of concentrations in distributions affected by analytical detection limits. Thus, the number of samples with concentrations less than detection limits and the number and concentrations of samples with concentrations greater than detection limits were used to provide more realistic estimates of means and standard deviations.

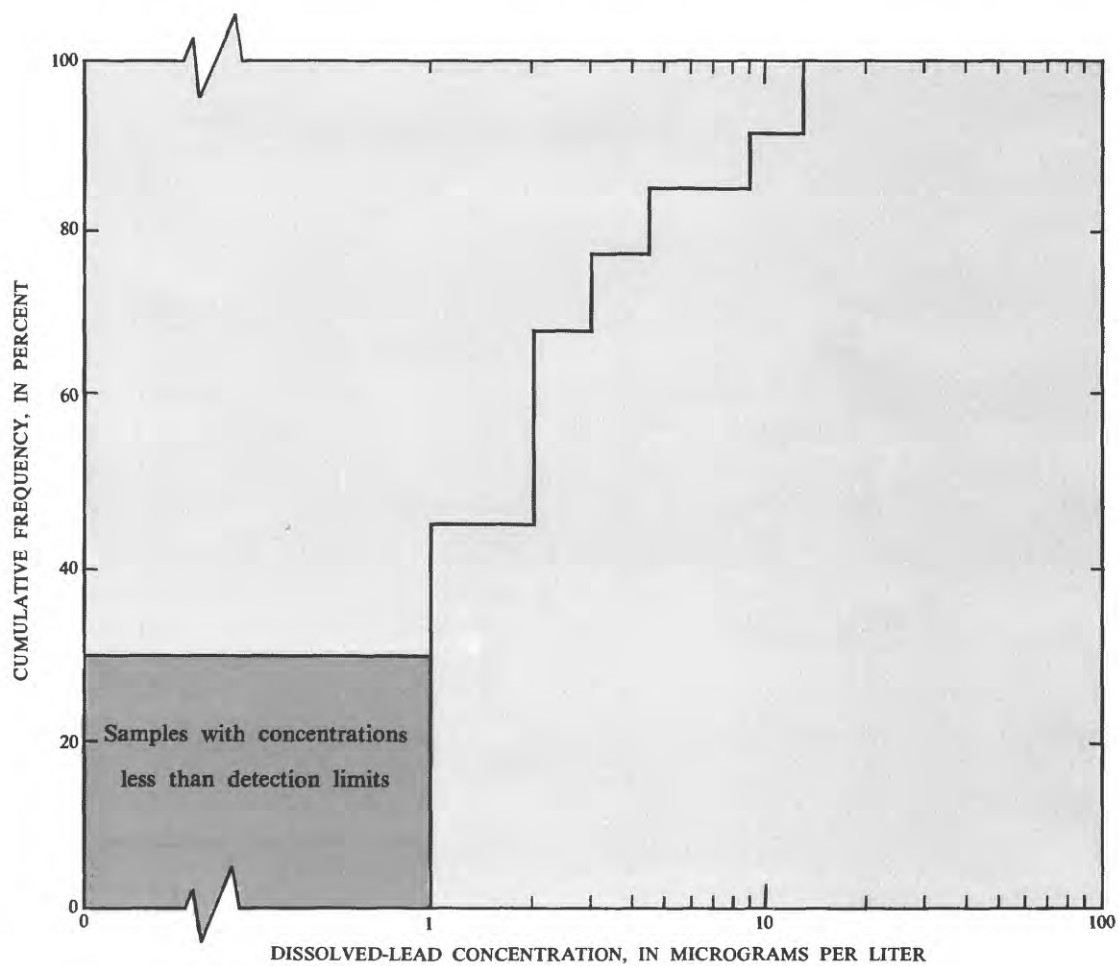


Figure 5.1-1 Dissolved-lead concentration in water samples at station 09217000, Green River near Green River, Wyoming.

5.0 ESTIMATION OF MEANS AND STANDARD DEVIATIONS OF DISTRIBUTIONS AFFECTED BY ANALYTICAL DETECTION LIMITS

5.1 Censored Sample Distributions

5.0 ESTIMATION OF MEANS AND STANDARD DEVIATIONS OF DISTRIBUTIONS AFFECTED BY ANALYTICAL DETECTION LIMITS--Continued

5.2 Geometric Means

Population mean and standard deviation are estimated in this report from censored data distributions by a method that uses the number of data less than detection limits as well as the number and individual magnitudes of data greater than detection limits.

Means and standard deviations in this report are estimated from a least-squares fit of data plotted on normal probability paper. When the underlying distribution is normal, data plot as a straight line. Data in this report were transformed to the normal form by using logarithms of the data, and the underlying distributions are referred to as lognormal. The antilog of the mean of the logarithms is an estimate of the geometric mean. The definition of the geometric mean used in this report is the strict mathematical definition: the n^{th} root of the product of n values.

Cumulative probabilities were estimated from the following equation (Blom, 1958; Barr and others, 1976):

$$\text{Cumulative probability} = (R_i - \frac{3}{8}) \times 100 \div [(n + \frac{1}{4})]$$

where n is the total number of samples including qualified values, and R_i is rank in ascending order (1,2,3,...,n) of the i^{th} sample. Qualified values receive ranks lower than the smallest unqualified value. For

example, if 20 of 100 samples were below the detection limit for a particular constituent, the 20 qualified values would receive ranks 1-20, and the unqualified value of smallest known magnitude would receive rank 21.

When logarithms of the known magnitudes are plotted on the vertical coordinate and corresponding cumulative probabilities are plotted on the horizontal coordinate, an estimate of the mean is the value of the vertical coordinate where the fitted line intersects a cumulative probability of 50 percent (Bowker and Lieberman, 1972) as shown in figure 5.2-1. The estimate of standard deviation is the difference on the vertical scale between the points where the fitted line passes through cumulative probabilities of 50 and 84.13 percent. Because of the large quantity of data analyzed in this report, the entire process was computerized using programs written for the Statistical Analysis Systems¹ (Barr and others, 1976).

¹ The use of trade names in this report is for identification only and does not constitute endorsement by the U.S. Geological Survey.

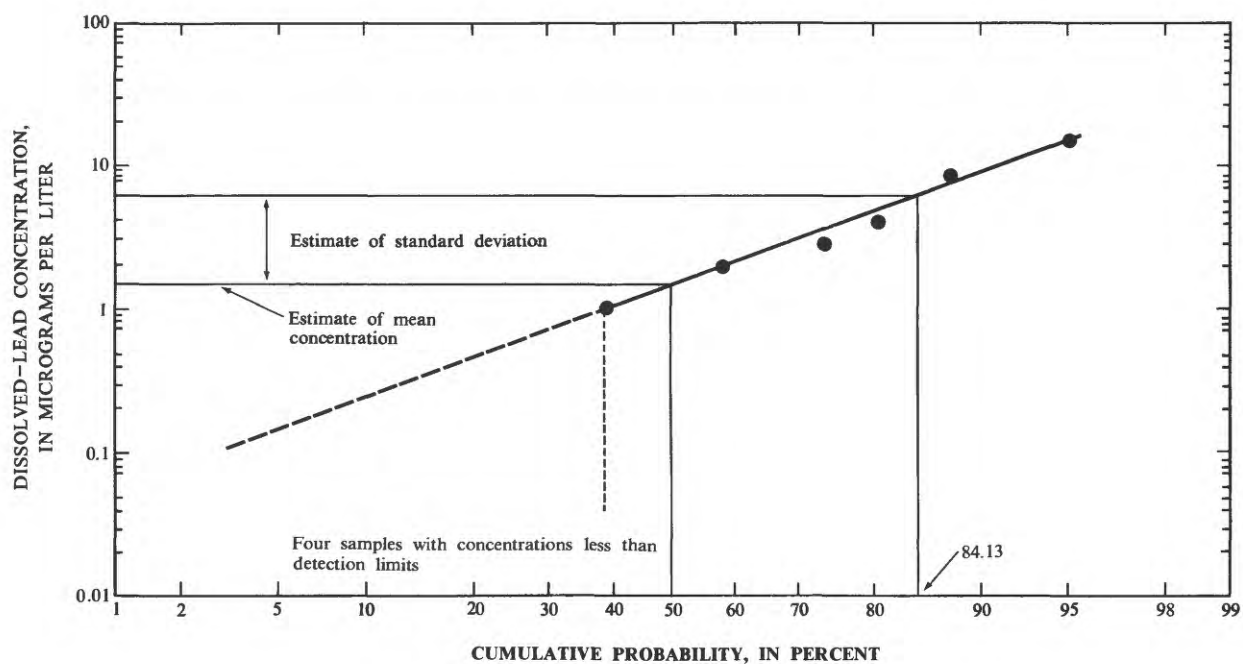


Figure 5.2-1 Determination of mean concentration and standard deviation of a data set containing censored data, station 09217000, Green River near Green River, Wyoming.

5.0 ESTIMATION OF MEANS AND STANDARD DEVIATIONS OF DISTRIBUTIONS AFFECTED BY ANALYTICAL DETECTION LIMITS--Continued

5.2 Geometric Means

5.0 ESTIMATION OF MEANS AND STANDARD DEVIATIONS OF DISTRIBUTIONS AFFECTED BY ANALYTICAL DETECTION LIMITS--Continued

5.3 Arithmetic Means

In this report, arithmetic means of censored lognormal distributions are estimated from means and standard deviations of logarithms of the data.

It often is desirable to estimate the arithmetic mean of a data population that is lognormally distributed. For example, it is useful to estimate annual load of a particular constituent at selected points along a stream in order to quantitatively determine sources or sinks of the particular constituent. If all the water passing a point on the stream were collected for a full year, the overall concentration of that constituent in the water multiplied by its total volume would give the desired load. If the total water volume were collected in small-sample volumes, the average load could be estimated from the arithmetic mean of the loads computed from each small-sample volume. In practice, only a few small-sample volumes are collected during the year, and the ratio of total water in the samples collected to total water passing a particular point along the stream may be less than 1 to 1 trillion. Because of this ratio and the underlying lognormal distribution, the arithmetic mean of the few samples is very sensitive to occasional large loads, and it can be shown (Sichel, 1952) that the arithmetic mean estimated from the few samples generally is greater than the true arithmetic mean of the total water volume.

A better estimate of the arithmetic mean of a censored lognormal distribution can be made using Sichel's "t" estimator (Sichel, 1952) in conjunction

with estimates of geometric mean and standard deviation described earlier in this section. The "t" estimator based on Fisher's "Method of Maximum Likelihood" (Fisher, 1921) is an estimate of the arithmetic mean of the total water volume and may be computed from the following power series (Sichel, 1952):

$$t = e^{\bar{X}} \left[1 + \frac{1}{2} V + \frac{n-1}{2 \cdot 2! (n+1)} V^2 + \frac{(n-1)^2}{2^3 3! (n+1) (n+3)} V^3 + \dots \right],$$

where \bar{X} = mean of the logarithms (base e),

V = variance of the logarithms or square of the standard deviation of the logarithms (base e), and

n = the total number of samples including those with concentrations less than detection limits.

NOTE: \bar{X} (base e) = 2.3026 \bar{X} (base 10), and

V (base e) = $(2.3026)^2 V$ (base 10).

The equations above were used to estimate average phosphorus loads presented in section 3.1, which are again presented in table 5.3-1 for the reader's convenience.

Table 5.3-1 Average total phosphorus loads, 1974-78 water years.

[L_Mean=Mean of logarithms (Base 10);
L_SD=Standard deviation of logarithms;
Avg_Load=Average load, in tons per day]

Station	L_Mean	L_SD	Avg_Load
09188500	-1.981486	0.651740	0.031
09205000	-1.838354	.707267	.053
09209400	-1.612998	.882251	.180
09211200	-1.603068	.551888	.055
09214500	-2.946421	1.156198	.027
09216000	-2.794096	.802085	.008
09216050	-2.270304	.821662	.030
09216300	-1.641843	.742699	.095
09217000	-1.013692	.627777	.269
09217010	-.694740	.572333	.473
09222300	-2.906847	.607950	.003
09222400	-2.360567	1.170525	.121
09224050	-1.612482	.413497	.038
09224450	-2.523287	1.188375	.107
09234500	-1.253458	.556600	.125

**5.0 ESTIMATION OF MEANS AND STANDARD DEVIATIONS OF DISTRIBUTIONS
AFFECTED BY ANALYTICAL DETECTION LIMITS--Continued**
5.3 Arithmetic Means

6.0 LOCATION OF WATER-QUALITY MEASUREMENT SITES

Water-quality measurement sites summarized in this report are referred to in three groups: routine surface-water sampling stations, miscellaneous surface-water sampling sites, and springs.

Routine surface-water sampling stations are part of a nationwide water-quality monitoring network operated by the U.S. Geological Survey in cooperation with other Federal and State agencies. Each routine surface-water sampling station in this report has a unique number. The complete 8-digit number, such as 09188500, includes the first two digits "09" that refer to the major drainage basin, and the remaining six digits "188500" that refer to individual station location. Increasing location numbers refer to locations progressively farther downstream.

Miscellaneous surface-water sites and springs represent a general reconnaissance and are not part

of a routine measurement network. Data analyzed in this report are available from computer storage (U.S. Geological Survey, 1974) and may be found in publications such as Water Resources Data for Wyoming (U.S. Geological Survey, 1976; 1977). The location of each routine surface-water sampling station is shown in figure 6.0-1; the stations are described in table 6.0-1. The location of each of the more than 500 spring and miscellaneous surface-water sites for which data were analyzed in this report but not referred to individually is available through the data sources mentioned above.

Table 6.0-1 Description of routine surface-water sampling stations.

Station number	Latitude	Longitude	Location
09188500	430108	1100703	Green River at Warren Bridge, near Daniel, Wyoming
09205000	423402	1095546	New Fork River near Big Piney, Wyoming
09209400	421134	1100945	Green River near LaBarge, Wyoming
09211200	420116	1100257	Green River below Fontenelle Reservoir, Wyoming
09214500	421412	1091844	Little Sandy Creek above Eden, Wyoming
09216000	420037	1093457	Big Sandy River below Eden, Wyoming
09216050	415643	1094104	Big Sandy River at Casson Bridge, near Eden, Wyoming
09216300	414552	1094405	Green River at Big Island, near Green River, Wyoming
09216527	413938	1073328	Separation Creek, near Riner, Wyoming
09216545	412935	1083047	Bitter Creek near Bitter Creek, Wyoming
09216562	413852	1085950	Bitter Creek above Salt Wells Creek, near Salt Wells, Wyoming
09216565	411156	1085952	Salt Wells Creek near South Baxter, Wyoming
09216576	411225	1090309	Gap Creek below Bean Springs Creek, near South Baxter, Wyoming
09216750	413750	1085918	Salt Wells Creek near Salt Wells, Wyoming
09216810	413556	1091354	Killpecker Creek at Rock Springs, Wyoming
09216880	413300	1091815	Bitter Creek below Little Bitter Creek, near Kanda, Wyoming
09217000	413059	1092654	Green River near Green River, Wyoming
09217010	412946	1092617	Green River below Green River, Wyoming
09222000	412708	1101020	Blacks Fork near Lyman, Wyoming
09222300	413454	1103342	Little Muddy Creek near Glencoe, Wyoming
09222400	413217	1101343	Muddy Creek near Hampton, Wyoming
09224050	414506	1103157	Hams Fork near Diamondville, Wyoming
09224450	413556	1095928	Hams Fork near Granger, Wyoming
09224700	413246	1094134	Blacks Fork near Little America, Wyoming
09229500	410045	1094020	Henrys Fork near Manila, Utah
09234500	405430	1092520	Green River near Greendale, Utah
09235300	410054	1083839	Vermillion Creek near Hiawatha, Colorado
09257000	410142	1073255	Little Snake River near Dixon, Wyoming

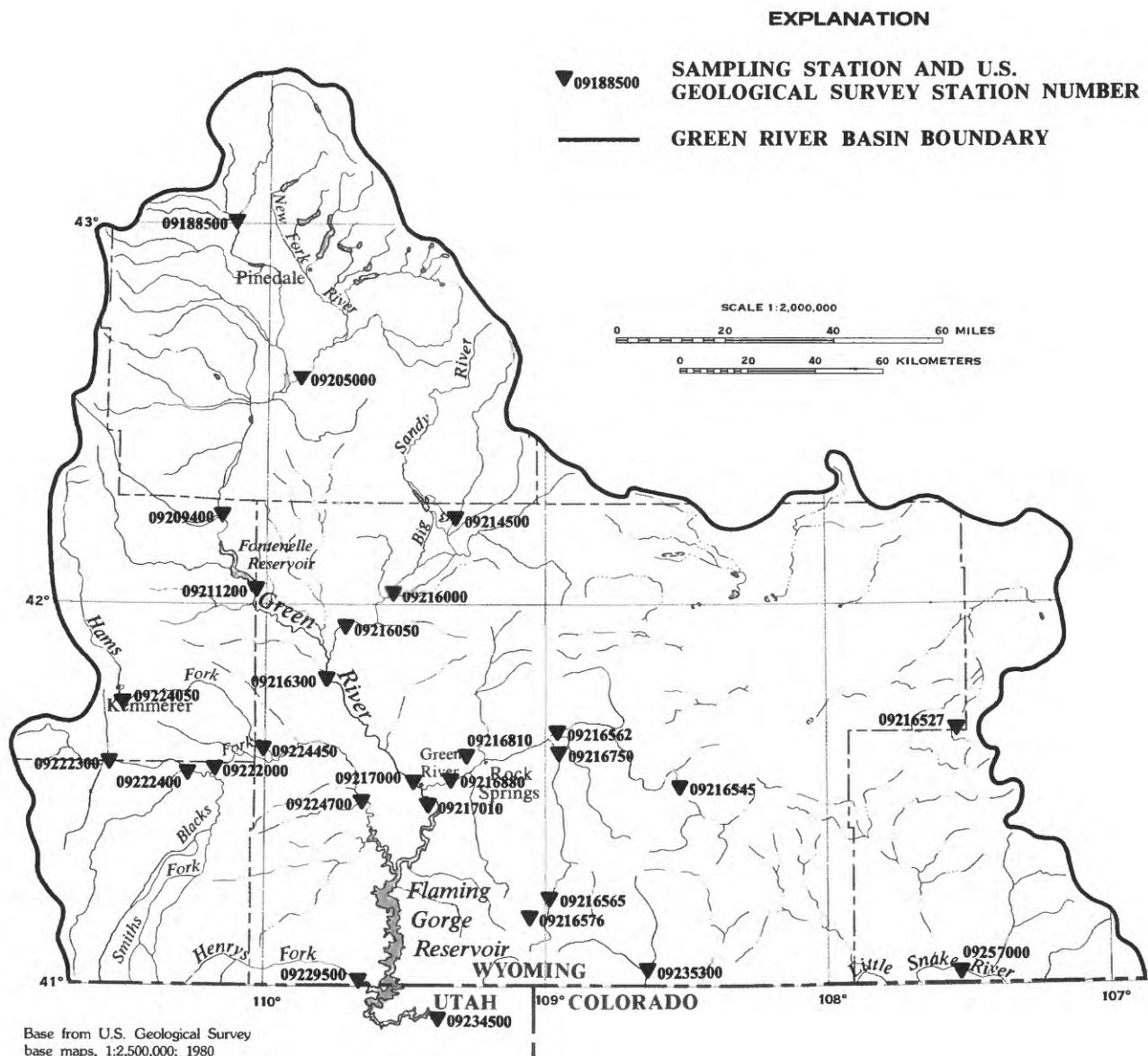


Figure 6.0-1 Location of routine surface-water sampling stations.

6.0 LOCATION OF WATER-QUALITY MEASUREMENT SITES

7.0 SELECTED REFERENCES

- Alekin, O. A., and Moricheva, N. P., 1957, The problem of the stability of the carbonate system in natural water: *Akademiya Nauk SSSR Doklady [Geochem. Sec.]*, v. 117, p. 1030.
- Barr, A. J., Goodnight, J. H., Sall, J. P., and Helwig, J. T., 1976, *A user's guide to SAS 76*: Raleigh, N.C., SAS Institute, Inc., 329 p.
- Blom, G. 1958, *Statistical estimates and transformed beta variables*: New York, John Wiley, 462 p.
- Bolke, E. L., 1979, Dissolved-oxygen depletion and other effects of storing water in Flaming Gorge Reservoir, Wyoming and Utah: U.S. Geological Survey Water-Supply Paper 2058, 41 p.
- Bowker, A. H., and Lieberman, G. J., 1972, *Engineering Statistics* (2d ed.): Englewood Cliffs, N.J., Prentice-Hall, p. 453-454.
- CH2M Hill, Inc., 1977, Clean water report for southwestern Wyoming: Technical report prepared for the Southwestern Wyoming Water Quality Planning Association by CH2M Hill, Inc., Denver, Colo., 224 p.
- Cohen, A. C., Jr., 1959, Simplified estimators for the normal distribution when samples are singly censored or truncated: *Technometrics*, v. 1, no. 3, p. 217-237.
- DeLong, L. L., 1977, An analysis of salinity in streams of the Green River Basin, Wyoming: U.S. Geological Survey Water-Resources Investigations 77-103, 32 p.
- _____, 1978, Predicting effects of coal development on surface-water salinity, Green River Basin, Wyoming [abstract]: Protection of the Hydrologic Environment of Surface Mined Lands (Symposium), American Geophysical Union meeting, San Francisco, Calif., Dec, 1978, p. 1067.
- Fisher, R. A., 1921, On the mathematical foundation of theoretical statistics, *in* Royal Society of London, Philosophical Transactions: London, ser. A, v. 222, p. 309-368.
- Hem, J. D., 1961, Calculation and use of ion activity: U.S. Geological Survey Water-Supply Paper 1535C, 17 p.
- _____, 1966, Chemical controls of irrigation drainage water composition: American Water Resources Conference, 2d, Chicago, Illinois, 1966, Proceedings, p. 64-77.
- _____, 1970, Study and interpretation of the chemical characteristics of natural water: U.S. Geological Survey Water-Supply Paper 1473, 363 p.
- Jennings, M. E., and Benson, M. A., 1969, Frequency curves for annual flood series with some zero events or incomplete data: *Water Resources Research*, v. 5, no. 1, p. 276-280.
- Landwehr, J. M., 1978, Some properties of the geometric mean and its use in water quality standards: *Water Resources Research*, v. 14, no. 2, p. 467-473.
- Lowham, H. W., DeLong, L. L., Collier, K. R., and Zimmerman, E. A., 1982, Hydrology of Salt Wells Creek--A plains stream in southwestern Wyoming: U.S. Geological Survey Water-Resources Investigations 81-62, 52 p.
- Lowham, H. W., DeLong, L. L., Peter, K. D., Wangsness, D. J., Head, W. J., and Ringen, B. H., 1976, A plan for the study of water and its relation to economic development in the Green River and Great Divide Basins in Wyoming: U.S. Geological Survey Open-File Report 76-349, 92 p.
- Miesch, A. T., 1967, Methods of computation for estimating geochemical abundance: U.S. Geological Survey Professional Paper 574-B, 15 p.
- _____, 1976, Geochemical survey of Missouri, methods of sampling, laboratory analysis, and statistical reduction of data: U.S. Geological Survey Professional Paper 954-A, 39 p.
- Riffenburg, H. B., 1925, Chemical character of ground water of the Northern Great Plains: U.S. Geological Survey Water-Supply Paper 560-B, p. 31-52.
- Sichel, H. S., 1952, New methods in the statistical evaluation of mine sampling data: London, Institute of Mining and Metallurgy Transactions, v. 61, p. 261-288.
- Skougstad, M. W., Fishman, M. J., Friedman, L. C., Erdman, D. E., and Duncan, S. S., 1979, Methods for determination of inorganic substances in water and fluvial sediments: U.S. Geological Survey Techniques of Water-Resources Investigations, Book 5, Chapter A1, 626 p.
- U.S. Environmental Protection Agency, 1977, Report on Flaming Gorge Reservoir, Sweetwater County, Wyoming, and Daggett County, Utah, EPA Region VIII: U.S. Environmental Protection Agency, National Eutrophication Survey Working Paper No. 885, 63 p.
- U.S. Geological Survey, 1974, WATSTORE - U.S. Geological Survey National Water Data Storage and Retrieval System: U.S. Geological Survey INF-74-23, 15 p.
- _____, 1976, Water resources data for Wyoming, Water Year 1976, Volume 2, Green River, Bear River, and Snake River Basins: U.S. Geological Survey Water-Data Report WY 76-2, 436 p.

- _____. 1977, Water resources data for Wyoming, Water Year 1977, Volume 2, Green River, Bear River, and Snake River Basins: U.S. Geological Survey Water-Data Report WY 77-2, 484 p.
- Welder, G. E., 1968, Ground-water reconnaissance of the Green River Basin, southwestern Wyoming: U.S. Geological Survey Hydrologic Investigations Atlas HA-290.
- Welder, G. E., and McGreevy, L. J., 1966, Ground-water reconnaissance of the Great Divide and Washakie Basins and some adjacent areas, southwestern Wyoming: U.S. Geological Survey Hydrologic Investigations Atlas HA-219.
- Wyoming Water Planning Program, 1970, Water and related land resources of the Green River Basin, Wyoming: Wyoming State Engineer's Office, Report No. 3 and addendum, 167 p.

**8.0 STATISTICAL SUMMARY OF TRACE-ELEMENT CONCENTRATIONS
AT ROUTINE SURFACE-WATER SAMPLING STATIONS**

8.0 STATISTICAL SUMMARY OF TRACE-ELEMENTS CONCENTRATIONS

N1 = Number of samples with concentrations greater
than detection limits;
N2 = Total number of samples;
Log_Mean = Mean of logarithm (Base 10);
Log_SD = Standard deviation of logarithms

ARSENIC

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	20	28	0.129831	0.298508	20	22	0.820537	0.718648
Springs	9	18	.011747	.539464	31	43	-.018871	1.052088
09209400	-	-	-----	-----	10	13	-.047066	.399289
09211200	-	-	-----	-----	11	12	.024868	.200920
09214500	-	-	-----	-----	1	3	-----	-----
09216000	12	12	.280207	.319742	13	13	.418590	.490640
09216527	-	-	-----	-----	5	5	.654461	.627264
09216545	4	4	.150515	.193351	15	15	.552681	.378719
09216562	-	-	-----	-----	9	10	.309929	.547071
09216565	-	-	-----	-----	11	11	.455642	.483767
09216576	4	4	.119280	.341083	5	5	.388897	.441358
09216750	-	-	-----	-----	8	8	1.357502	.505310
09216810	10	12	.081779	.359547	12	12	.569967	.227697
09216880	12	12	.917538	.466349	12	12	1.068463	.502949
09217000	9	13	-.045118	.235257	11	13	.023935	.225697
09217010	10	13	.036004	.276979	10	12	.162171	.240669
09218300	-	-	-----	-----	1	2	-----	-----
09222300	4	5	.068841	.276916	12	14	.176440	.296950
09222400	5	5	.311261	.436568	16	16	.449214	.353226
09224050	11	12	.092761	.220079	11	12	.171454	.274278
09235300	5	5	.060206	.204129	13	13	.743762	.889410
09257000	5	5	.180618	.215411	4	4	.225772	.215199

BARIUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	12	20	1.938395	0.262092	16	17	2.436076	0.761356
Springs	5	13	.983534	1.200210	39	43	2.290691	.750487

BORON

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	32	32	1.770796	0.624186				
Springs	55	56	2.014201	.700003	6	7	1.311696	0.340256
09188500	3	3	1.359727	.135025	12	12	1.890589	.278190
09211200	-	-	-----	-----	11	11	1.803293	.196321
09216000	-	-	-----	-----	12	12	2.538700	.132279
09216527	5	5	2.012591	.259358	-	-	-----	-----
09216545	15	15	2.337002	.095846	-	-	-----	-----
09216562	10	10	2.682833	.242573	-	-	-----	-----
09216565	12	12	2.161539	.132544	-	-	-----	-----
09216576	6	6	2.067107	.147130	-	-	-----	-----
09216750	8	8	2.781742	.118190	-	-	-----	-----
09216810	12	12	3.272463	.288056	-	-	-----	-----
09216880	12	12	2.651740	.542541	-	-	-----	-----
09217000	13	14	1.763570	.173855	-	-	-----	-----
09217010	12	13	1.817728	.176067	-	-	-----	-----
09222300	14	14	3.231787	.368344	-	-	-----	-----
09222400	16	16	3.091807	.375553	-	-	-----	-----
09224050	12	12	1.689086	.161156	-	-	-----	-----
09235300	14	14	2.208319	.093588	-	-	-----	-----

8.0 STATISTICAL SUMMARY OF TRACE ELEMENTS--CONTINUED

CADMIUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	10	30	-0.795112	1.060640	15	29	0.491781	0.452993
Springs	4	21	-.500116	.423996	21	54	.093764	.703853
09209400	10	12	.066096	.342347	6	13	.772779	.625052
09211200	8	12	.123962	.507693	4	12	.975346	.199190
09213810	-	-	-----	-----	10	11	.993019	.442685
09214500	3	3	.318081	.365852	2	3		
09216000	11	12	.263827	.291957	10	13	.863976	.387338
09216527	1	3	-----	-----	2	5	-----	-----
09216545	-	-	-----	-----	8	15	.749743	.480722
09216562	4	4	.313818	.263536	9	12	.846733	.427020
09216565	3	6	-.234757	.555482	7	11	.830627	.320409
09216576	3	4	.092551	.309216	1	5	-----	-----
09216750	-	-	-----	-----	7	8	.676736	.603059
09216810	10	12	.268225	.346760	-	-	-----	-----
09216880	12	12	.125429	.169609	9	12	.777981	.485836
09217000	10	13	.070180	.413639	6	13	.772779	.625052
09217010	10	13	.085375	.505554	5	13	.722970	.581831
09218500	1	2	-----	-----	-	-	-----	-----
09222000	1	2	-----	-----	-	-	-----	-----
09222300	5	5	.120412	.215411	9	14	.783303	.304374
09222400	3	5	-.160737	.646569	7	16	.735464	.572777
09224050	9	12	-.035486	.309981	4	12	.739703	.543437
09224700	1	2	-----	-----	-	-	-----	-----
09235300	4	5	.357354	.213456	9	13	.775982	.668251
09257000	2	7	-----	-----	2	7	-----	-----

CHROMIUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	17	28	0.746585	0.354923	17	23	1.238870	0.723039
Springs	5	20	.999999	.000001	36	49	.508931	.431486
09188500	1	2	-----	-----	-	-	-----	-----
09209400	1	12	-----	-----	5	13	.664901	.632350
09211200	3	12	-----	-----	3	12	-----	-----
09214500	1	3	-----	-----	-	-	-----	-----
09216000	2	12	-----	-----	5	13	.881846	.561556
09216527	-	-	-----	-----	4	5	1.324008	.326512
09216545	1	4	-----	-----	9	15	.874015	.639644
09216562	2	4	-----	-----	7	10	.965461	.263489
09216565	-	-	-----	-----	5	11	.891405	.448951
09216576	1	4	-----	-----	1	5	-----	-----
09216750	1	2	-----	-----	8	8	1.563576	.338769
09216810	7	12	1.000000	.342960	11	12	1.133528	.206338
09216880	5	12	.781080	.318005	7	12	.961010	.485396
09217000	2	12	-----	-----	5	12	-----	-----
09217010	1	13	-----	-----	6	13	.688890	.445714
09222000	1	2	-----	-----	6	14	.792121	.298034
09222400	1	5	-----	-----	9	16	.940825	.435381
09224050	3	12	-----	-----	5	12	.781080	.318005
09224450	1	2	-----	-----	-	-	-----	-----
09224700	1	2	-----	-----	1	2	-----	-----
09235300	2	5	-----	-----	8	13	.928160	.827881

8.0 STATISTICAL SUMMARY OF TRACE-ELEMENT CONCENTRATIONS

8.0 STATISTICAL SUMMARY OF TRACE ELEMENTS--CONTINUED

COPPER

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	14	30	-0.196227	0.724915	27	29	1.265916	0.604515
Springs	12	12	.006371	.511108	46	55	.392030	.555859
09188500	1	2	-----	-----	-	-	-----	-----
09205000	1	2	-----	-----	-	-	-----	-----
09209400	11	12	.256785	.328950	6	13	1.153448	.612506
09211200	8	12	.256399	.321647	5	12	1.030615	.559687
09214500	3	3	.518767	.274389	2	3	-----	-----
09216000	12	12	.486634	.334626	11	13	1.128196	.331904
09216527	3	3	.359727	.346241	5	5	1.288181	.413175
09216545	4	4	.526802	.303186	13	15	1.128270	.231736
09216562	4	4	.495568	.168245	10	10	1.132222	.328266
09216565	4	6	.011131	.384125	8	11	1.111833	.281811
09216576	3	4	.000000	.454777	4	5	-----	-----
09216750	2	2	-----	-----	8	8	-----	-----
09216810	12	12	.526394	.355560	11	12	1.160326	.338300
09216880	12	12	.465810	.333080	10	12	1.229205	.279016
09217000	9	12	.187153	.501330	9	13	1.195193	.645858
09217010	11	13	.145221	.389794	8	13	.914087	.201174
09218500	1	2	-----	-----	2	2	-----	-----
09222000	1	2	-----	-----	1	1	-----	-----
09222300	5	5	.571466	.281104	11	14	1.063983	.322018
09222400	4	5	.313309	.594195	14	16	1.089621	.295527
09224050	6	12	-.009318	.507931	7	12	.922058	.183940
09224450	2	2	-----	-----	-	-	-----	-----
09224700	1	2	-----	-----	2	2	-----	-----
09235300	4	5	.586468	.178366	10	13	1.135870	.326827
09257000	5	7	.003865	.357468	3	7	-----	-----

IRON

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	45	46	1.765056	0.339468	-	-	-----	-----
Springs	61	64	1.780503	.937305	-	-	-----	-----
09209400	11	12	1.400799	.294776	-	-	-----	-----
09211200	11	12	1.320100	.330483	-	-	-----	-----
09216000	11	12	1.297491	.398695	-	-	-----	-----
09216527	5	5	2.016298	.533520	-	-	-----	-----
09216545	14	15	1.837800	.478823	-	-	-----	-----
09216565	12	12	1.872401	.434491	-	-	-----	-----
09216576	6	6	1.790166	.759865	-	-	-----	-----
09216750	8	8	1.655812	.440449	-	-	-----	-----
09216810	11	12	1.901847	.485277	-	-	-----	-----
09216880	12	12	1.837550	.333920	-	-	-----	-----
09217000	11	12	1.330325	.493496	-	-	-----	-----
09217010	10	13	1.314672	.292042	-	-	-----	-----
09222300	13	14	1.763910	.403631	-	-	-----	-----
09222400	16	16	1.815513	.486836	-	-	-----	-----
09224050	12	12	1.635262	.362127	-	-	-----	-----
09235300	14	14	1.869206	.595897	-	-	-----	-----
09257000	7	7	1.773376	.259552	-	-	-----	-----
09276562	10	10	1.676582	.372749	-	-	-----	-----

8.0 STATISTICAL SUMMARY OF TRACE ELEMENTS--CONTINUED

LEAD

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	18	30	0.000112	0.920091	18	29	1.535608	0.511762
Springs	11	21	-.222502	.696572	32	55	.760753	.910449
09188500	1	2	-----	-----	-	-	-----	-----
09209400	7	12	.165097	.474368	6	13	1.873261	.348653
09211200	9	12	.179545	.597228	1	12	-----	-----
09216000	9	12	.333600	.556084	7	13	1.799527	.566849
09216527	2	3	-----	-----	2	5	-----	-----
09216545	3	4	-----	-----	7	15	1.669501	.435848
09216562	4	4	.508356	.227389	8	10	1.549301	.557620
09216565	3	6	.125100	.659609	6	11	1.527325	.625957
09216576	3	4	.271389	.658625	1	5	-----	-----
09216750	2	2	-----	-----	7	8	1.970058	.295581
09216810	12	12	.610015	.371272	9	12	1.792447	.517318
09216880	11	12	.467953	.298461	4	12	1.741503	.628700
09217000	9	13	.168593	.612174	1	12	-----	-----
09217010	12	13	.477245	.605951	2	13	-----	-----
09218500	1	2	-----	-----	-	-	-----	-----
09222000	1	2	-----	-----	-	-	-----	-----
09222300	3	5	-----	-----	9	14	1.589685	.571406
09222400	4	5	.413372	.746369	7	16	1.665209	.539228
09224050	-	-	-----	-----	5	12	-----	-----
09224450	8	13	.357533	.275219	-	-	-----	-----
09235300	4	5	.260963	.491114	6	13	1.820204	.729260
09257000	4	7	.044845	.535510	2	7	-----	-----

MANGANESE

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	28	31	1.076541	0.314760	-	-	-----	-----
Springs	47	54	1.381461	.965381	-	-	-----	-----
09209400	8	12	.886961	.315082	-	-	-----	-----
09211200	8	12	.890813	.431432	-	-	-----	-----
09216000	11	12	1.148731	.260186	-	-	-----	-----
09216527	3	3	1.382042	.486011	-	-	-----	-----
09216545	4	4	1.150515	.286933	-	-	-----	-----
09216562	4	4	2.416439	.105102	-	-	-----	-----
09216565	6	6	2.218251	.408684	-	-	-----	-----
09216576	4	4	2.064683	.152227	-	-	-----	-----
09216810	12	12	2.959255	.599444	-	-	-----	-----
09216880	12	12	2.225580	.232974	-	-	-----	-----
09217000	9	12	.822687	.437793	-	-	-----	-----
09217010	10	13	.964853	.409923	-	-	-----	-----
09222300	5	5	1.476042	.405551	-	-	-----	-----
09222400	5	5	1.180618	.215411	-	-	-----	-----
09224050	12	12	1.708687	.386140	-	-	-----	-----
09235300	5	5	2.109506	.241465	-	-	-----	-----
09257000	6	6	1.421057	.320326	-	-	-----	-----

**8.0 STATISTICAL SUMMARY OF TRACE-ELEMENT CONCENTRATIONS
AT ROUTINE SURFACE-WATER SAMPLING STATIONS--Continued**

MOLYBDENUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	17	28	0.186752	0.463614	22	23	0.542350	0.368419
Springs	11	20	.116621	.856924	40	49	.658481	.528263
09209400	8	12	-.074683	.337426	9	13	.309024	.339851
09211200	10	12	-----	-----	8	12	.050823	.460341
09216000	11	12	.845351	.159766	13	13	.822063	.182663
09216527	1	3	.989760	.128127	5	5	.396454	.338472
09216545	-	-	-----	-----	15	15	.981920	.124783
09216562	4	4	.837562	.328619	10	10	.668950	.329222
09216565	6	6	.704218	.108664	11	11	.788009	.174924
09216576	4	4	-----	-----	5	5	.580618	.177187
09216750	1	1	-----	-----	7	8	.719668	.249777
09216810	12	12	.937133	.194318	12	12	.935540	.231719
09216880	11	12	.364290	.356283	12	12	.618091	.235470
09217000	10	12	.147227	.330284	11	13	.174865	.357341
09217010	11	13	.191475	.381034	13	13	.335575	.206833
09222300	5	5	1.168462	.201923	14	14	1.143121	.178839
09222400	5	5	1.120231	.090542	16	16	.983419	.274418
09224050	3	12	-----	-----	8	12	.095261	.416334
09235300	5	5	.781310	.123136	13	13	.791704	.118735

NICKEL

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	7	8	0.398370	0.222160	4	7	1.776302	0.297087
Springs	3	10	-.368910	1.481439	2	14	-----	-----
09209400	10	12	.292006	.334505	3	13	1.378886	.827378
09211200	11	11	.333047	.224974	3	12	-----	-----
09216000	8	12	.168307	.451319	7	13	1.512381	.598411
09216527	1	3	-----	-----	3	5	1.395153	.531671
09216545	4	4	.464333	.143466	9	15	1.366830	.391023
09216562	4	4	.838911	.097542	7	10	1.396986	.470240
09216565	4	4	-----	-----	7	10	1.238846	.558720
09216576	3	3	-----	-----	-	-	-----	-----
09216750	1	2	-----	-----	7	8	1.835021	.276580
09216810	12	12	1.030229	.202344	11	12	1.696427	.376603
09216880	12	12	.717118	.164120	3	12	1.468189	.490272
09217000	9	12	.244374	.397151	2	13	-----	-----
09217010	11	13	.361741	.225273	2	13	-----	-----
09222300	4	5	.911930	.089733	8	14	1.427589	.337161
09222400	4	4	.778151	.167845	8	15	1.398650	.438420
09224050	11	12	.296336	.278263	2	12	-----	-----
09235300	4	4	.595053	.105730	8	13	1.374996	.486366
09257000	3	5	-----	-----	2	5	-----	-----

8.0 STATISTICAL SUMMARY OF TRACE ELEMENTS--CONTINUED

VANADIUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	8	8	0.128514	0.398020	-	-	-----	-----
Springs	6	10	.126936	1.163908	-	-	-----	-----
09209400	7	11	-.473593	.542967	-	-	-----	-----
09211200	6	10	-.275350	.209085	-	-	-----	-----
09216000	8	9	.187322	.264914	-	-	-----	-----
09216562	4	4	-----	-----	-	-	-----	-----
09216565	1	4	-----	-----	-	-	-----	-----
09216576	3	3	-----	-----	-	-	-----	-----
09216750	1	2	-----	-----	-	-	-----	-----
09216810	10	12	.775107	.365733	-	-	-----	-----
09216880	11	12	.458768	.235883	-	-	-----	-----
09217000	9	12	-.334517	.292131	-	-	-----	-----
09217010	9	13	-.294024	.285565	-	-	-----	-----
09222300	2	5	-----	-----	-	-	-----	-----
09222400	3	4	.103193	.354257	-	-	-----	-----
09224050	7	12	-.606321	.557978	-	-	-----	-----
09235300	1	4	-----	-----	-	-	-----	-----
09257000	4	5	-.000422	.472970	-	-	-----	-----

ZINC

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	13	28	0.817970	0.256938	21	23	1.780899	0.832538
Springs	13	20	1.031358	1.169338	37	49	.946874	.721753
00921654	4	4	1.150515	.193351	-	-	-----	-----
09209400	8	12	.725504	.367199	12	13	1.269561	.330986
09211200	6	12	.666381	.402542	10	12	1.211982	.297092
09216000	10	12	-----	-----	12	12	1.455238	.368390
09216527	2	3	-----	-----	5	5	1.796454	.439200
09216545	-	-	-----	-----	15	15	1.528279	.280901
09216562	4	4	-----	-----	10	10	1.604372	.270300
09216565	3	6	-----	-----	11	11	1.452835	.427063
09216576	2	4	-----	-----	5	5	1.336248	.252922
09216750	2	2	-----	-----	12	12	2.407419	-----
09216810	12	12	1.342717	.216335	12	12	1.727641	.144387
09216880	12	12	1.215361	.197282	12	12	1.672141	.379441
09217000	11	13	.919952	.382234	12	12	1.190275	.253936
09217010	10	13	.954629	.210667	12	12	1.115017	.173497
09222300	5	5	1.215836	.243899	14	14	1.552561	.238731
09222400	4	5	.947439	.433372	16	16	1.491454	.364050
09224050	9	12	-----	-----	12	12	1.241008	.394740
09235300	4	5	1.093829	.341455	13	13	1.683755	.709803
09257000	3	6	-----	-----	5	5	.996866	.308909

**8.0 STATISTICAL SUMMARY OF TRACE-ELEMENT CONCENTRATIONS
AT ROUTINE SURFACE-WATER SAMPLING STATIONS--Continued**

8.0 STATISTICAL SUMMARY OF TRACE ELEMENTS--CONTINUED

LITHIUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	8	8	1.632546	0.519093	7	7	1.765445	0.386219
Springs	9	10	1.324586	.399124	11	14	1.280540	.360265
09209400	8	12	.786650	.221763	12	13	.962015	.374549
09211200	5	11	.725219	.252495	11	12	.920644	.216203
09216000	12	12	1.543404	.190029	12	12	1.572303	.211556
09216527	2	2	-----	-----	5	5	1.637278	.159310
09216545	4	4	1.798836	.143467	15	15	1.798008	.087710
09216562	4	4	3.328993	.179931	10	10	2.184411	.207562
09216565	4	4	2.019795	-----	10	10	1.623754	.179381
09216576	3	3	1.700124	-----	5	5	1.775704	.148912
09216750	2	2	2.303728	-----	8	8	2.347035	.119905
09216810	12	12	2.581393	.209001	12	12	2.003110	.287161
09216880	12	12	2.001080	.267131	-	-	-----	-----
09217000	12	12	.944538	.250176	13	13	1.080709	.231994
09217010	12	13	1.008945	.163110	13	13	1.094254	.185240
09222300	5	5	1.890124	.121724	14	14	1.975659	.142084
09222400	4	4	2.112620	.352620	15	15	2.110786	.296124
09224050	12	12	1.006598	.186333	12	12	1.088904	.245827
09235300	4	4	1.857357	.070703	13	13	1.967343	.230328

SELENIUM

Site or Station	Dissolved				Total			
	N1	N2	Log Mean	Log SD	N1	N2	Log Mean	Log SD
Misc-SW	11	28	-0.215167	0.330025	16	22	0.250972	0.589242
Springs	6	18	-.385419	.690792	15	43	-.334270	.726928
09209400	-	-	-----	-----	1	13	-----	-----
09211200	-	-	-----	-----	1	12	-----	-----
09216000	12	12	.296359	.325257	13	13	.367550	.331818
09216527	-	-	-----	-----	2	5	-----	-----
09216545	2	4	-----	-----	10	15	-.040697	.137239
09216562	4	4	.475772	.628479	10	10	.527397	.543573
09216565	5	6	-----	-----	10	11	-----	-----
09216576	3	4	-----	-----	5	5	.060206	-----
09216750	2	2	.150515	-----	8	8	.651189	.242497
09216810	10	12	.031903	.276520	11	12	.088521	.248603
09216880	5	12	-----	-----	7	12	-.061425	.343872
09217000	-	-	-----	-----	3	13	-----	-----
09217010	1	13	-----	-----	4	13	-----	-----
09222300	4	5	.555508	.499714	14	14	.667948	.365931
09222400	5	5	.406685	.318780	16	16	.286248	.358683
09224050	1	12	-----	-----	1	12	-----	-----
09235300	3	5	-----	-----	10	13	.000737	.338809

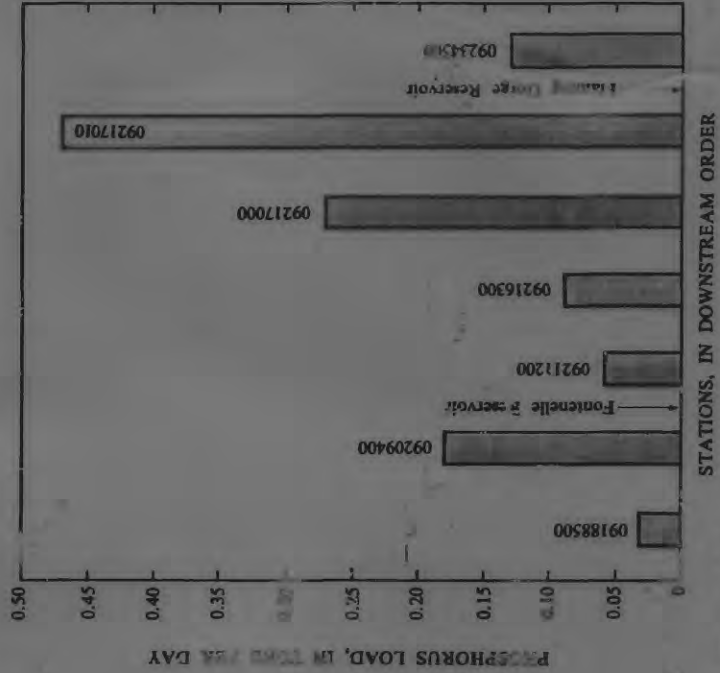


Figure 3.1-1 Effect of Fontenelle and Flaming Geologic Reservoirs on phosphorus loads in the Green River, 1974-78 water years.

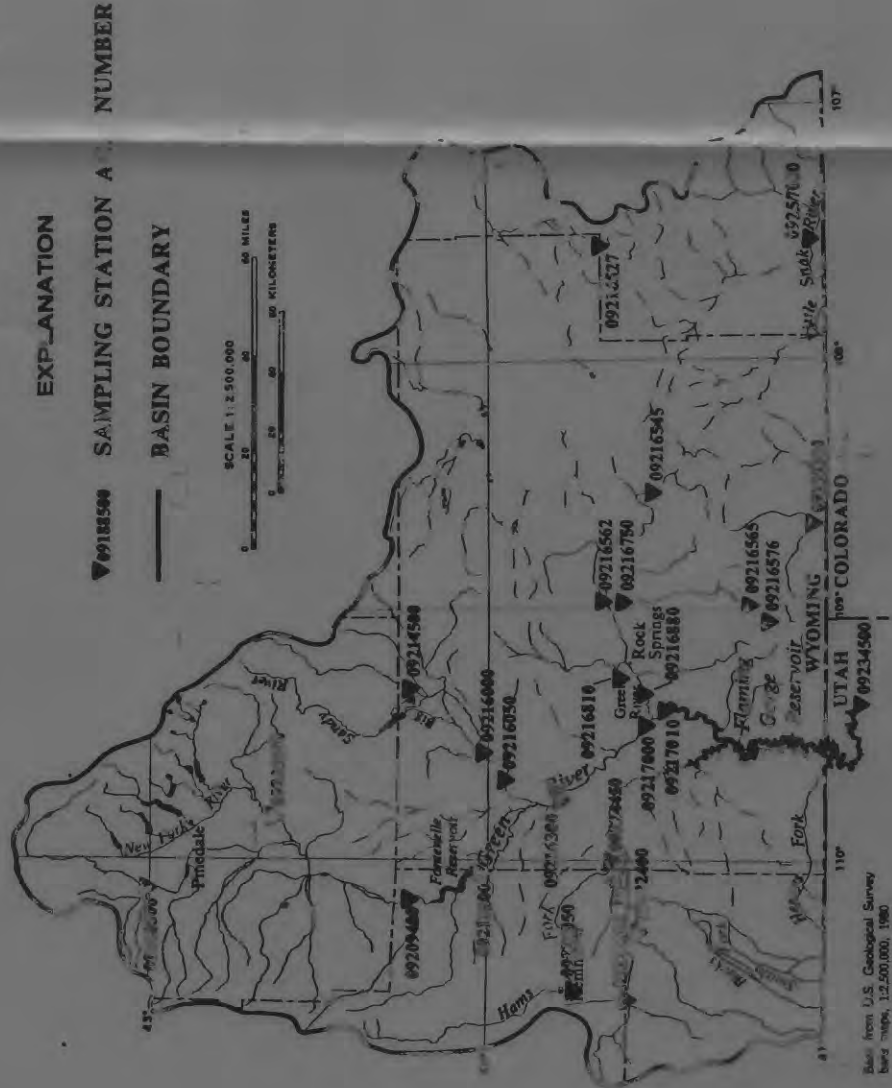


Figure 3.1-2 Location of sampling stations.

Table 3.1-2 Total phosphorus concentrations, 1974-78 water years.

[N1 = Number of samples with concentration greater than detection limits; N2 = Total number of samples; GM = Geometric mean, in milligrams per liter; Log_Mean = Mean of the logarithms (Base 10); Log_SD = Standard deviation of the logarithms]

Station	N1	N2	GM	Log_Mean	Log_SD
09188500	33	44	0.014	-1.865142	0.358878
09205000	56	67	.015	-1.834875	.434438
09209400	48	66	.013	-1.869689	.507258
09211200	50	68	.010	-1.985871	.388170
09214500	29	29	.052	-1.287484	.402490
09216000	48	62	.017	-1.767476	.707539
09216050	40	44	.038	-1.421177	.671681
09216300	48	67	.011	-1.976795	.549073
09216527	11	11	.212	-.672897	.572031
09216545	48	48	.142	-.846441	.449616
09216562	29	29	.099	-1.003964	.645013
09216565	33	33	.075	-1.125269	.578884
09216576	18	18	.125	-.902135	.407013
09216750	13	13	1.573	-.196720	.554926
09216810	41	41	.225	-.646987	.456018
09216880	47	47	5.102	-.707711	.372935
09217000	47	51	.030	-1.31853	.454085
09217010	46	47	.060	-1.219940	.411432
09222300	43	43	.073	-1.136858	.389523
09222400	40	40	.145	-.838748	.579339
09224050	46	46	.204	-.690592	.337454
09224450	55	63	.030	-1.520790	.603649
09234500	34	46	.011	-1.974433	.413104
09235300	40	41	.149	-.826137	.706462
09257000	8	8	.050	-1.304814	.534566