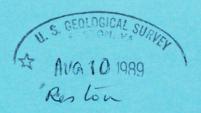
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# GROUND-WATER QUALITY IN DOUGLAS COUNTY, WESTERN NEVADA

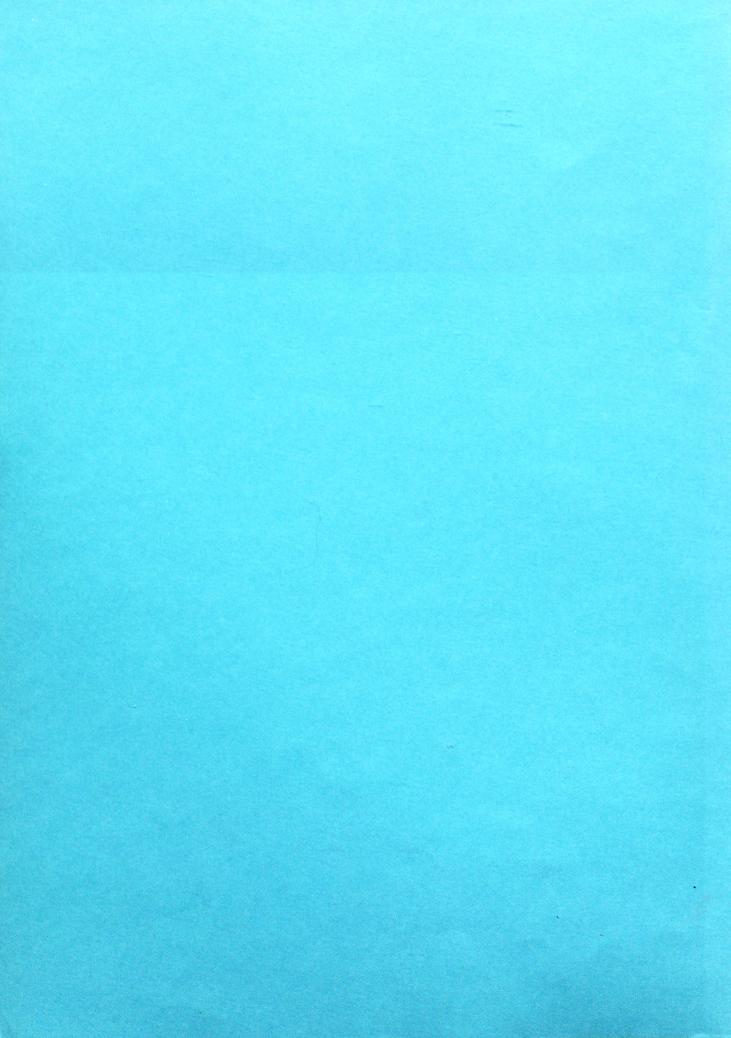
U.S. GEOLOGICAL SURVEY

Water-Resources Investigations Report 87-4269

Prepared in cooperation with the

DOUGLAS COUNTY DEPARTMENT OF PUBLIC WORKS





# GROUND-WATER QUALITY IN DOUGLAS COUNTY, WESTERN NEVADA

By Kerry T. Garcia

# U.S. GEOLOGICAL SURVEY

Water-Resources Investigations Report 87-4269

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Carson City, Nevada 1989

# DEPARTMENT OF THE INTERIOR

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U.S. GEOLOGICAL SURVEY

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#### CONVERSION FACTORS AND ABBREVIATIONS

"Inch-pound" units of measure used in this report may be converted to metric (International System) units by using the following factors:

Multiply	<u>By</u>	To obtain
Feet (ft) Miles (mi)	0.3048	Meters (m) Kilometers (km)
Square miles (mi <sup>2</sup> )	2.590	Square kilometers (km <sup>2</sup> )

For temperature, degrees Celsius (°C) may be converted to degrees Fahrenheit (°F) by using the formula °F = [(1.8)(°C)] + 32.

For temperature, degrees Fahrenheit (°F) may be converted to degrees Celsius (°C) by using the formula  $^{\circ}C = 0.5556$  (°F - 32).

#### ALTITUDE DATUM

In this report "sea level" refers to the National Geodetic Vertical Datum of 1929 (NGVD of 1929), which is derived from a general adjustment of the first-order leveling networks of both the United States and Canada.

# GROUND-WATER QUALITY IN DOUGLAS COUNTY, WESTERN NEVADA

By Kerry T. Garcia

#### ABSTRACT

A 182-percent increase in population within the last 10 years in Douglas County, Nevada, has raised concerns by county officials as to the possible effects land development may have on ground-water quality. Ground water is the principal source of drinking water in Douglas County. This study was developed in cooperation with the county to define ground-water quality conditions, examine any areal or temporal trends in the ground-water quality, and examine the need for continued ground-water monitoring program. Water-quality data were compiled and selected wells were sampled to define ground-water quality conditions.

Most ground water in Douglas County meets the State of Nevada drinking-water standards. Of the 333 sampling sites used in this analysis, water from 6 of them equaled or were greater than the standards for sulfate, 44 for fluoride, 4 for dissolved solids, 5 for nitrate as nitrate, 12 for arsenic, 33 for iron, and 18 for manganese. Ground water in the west-central, northern, and northeastern part of Carson Valley is influenced by geothermal water. Septic-tank effluent may contaminate the ground water in some areas of the county.

The predominant cations in Douglas County's ground water are calcium and sodium and the predominant anion is bicarbonate. Statistical distributions for sodium, arsenic, iron, manganese, and zinc showed one single population whereas distributions for dissolved solids, calcium, magnesium, bicarbonate, sulfate, chloride, fluoride, and nitrate as nitrate showed two discrete populations.

Analyses of water from selected wells in the county were examined for areal or temporal trends in dissolved-solids and nitrate concentrations. Dissolved-solids and nitrate concentrations for Gardnerville, Gardnerville Ranchos, and Minden municipal wells showed no overall trend for analyses spanning the years 1969-83. Dissolved-solids concentrations for the Topaz Lodge Subdivision well shows no consistent trend from 1961 to 1984. Nitrate concentrations in this well have shown a general increase from about 2 milligrams per liter (as nitrate) in 1961 to about 12 milligrams per liter in 1984.

On the basis of ground-water data in Douglas County, a continuing ground-water-quality monitoring program would include sample sites from heavily pumped municipal wells or wells in potentially high-risk pollution areas, and wells that may provide information about temporal changes in water quality.

#### INTRODUCTION

Ground water is the principal source of drinking water for Douglas County, Nev. (figure 1), except for the Lake Tahoe basin, where the lake is the principal source. Ground water also is used for irrigation in Carson and Antelope Valleys. In recent years, the quantity of ground water pumped for irrigation has decreased, whereas the quantity pumped for domestic supply has increased. This change in water use is due to the conversion of agricultural lands to residential developments. Rapid population growth has concerned county officials about the present and future effects of development on ground-water quality. In response to this concern, Douglas County and the U.S. Geological Survey entered into a cooperative agreement, and a general ground-water quality study began in October 1982.

# Purpose and Scope

The purposes of the study and this report were to (1) define ground-water quality conditions, (2) examine any areal or temporal trends in the ground-water quality, and (3) examine components for a continuing ground-water monitoring program in Douglas County. The scope of the study reported herein involved compiling data on wells and springs from local and State agency files and beginning a supplemental sampling program of 35 wells in August and September 1983 to extend areal coverage of ground-water quality data in Douglas County.

# Physical Setting

Douglas County comprises about 751 mi² of western Nevada (figure 1). The county is bounded on the west by Lake Tahoe and the Carson Range of the Sierra Nevada and on the east by part of the Pine Nut Mountains. North-south trending mountain ranges divide the county into three major valleys: the Lake Tahoe basin along the western border (about 56 mi²), Carson Valley in the west-central part (about 422 mi²), and the Nevada part of Antelope Valley in the southeastern apex of the county (about 115 mi²). The two principal rivers in the county are the Carson River in Carson Valley and the West Walker River in Antelope Valley.

Altitudes of the higher peaks in the Carson Range and Pine Nut Mountains range from about 8,500 to 10,000 feet above sea level. Carson and Antelope Valley altitudes range from about 4,500 to 5,500 feet. Because the Sierra Nevada forms the western boundary of Douglas County, the study area lies mainly in a zone of diminished precipitation (a "rain shadow") with respect to eastward-moving storms (Glancy and Katzer, 1976, page 17).

Most winter precipitation in the mountains is snow, but winter rains on the snowpack may contribute significantly to the precipitation total. Summer thunderstorms may produce heavy, localized amounts of precipitation but do little to supplement ground water, owing to fairly rapid evapotranspiration rates.

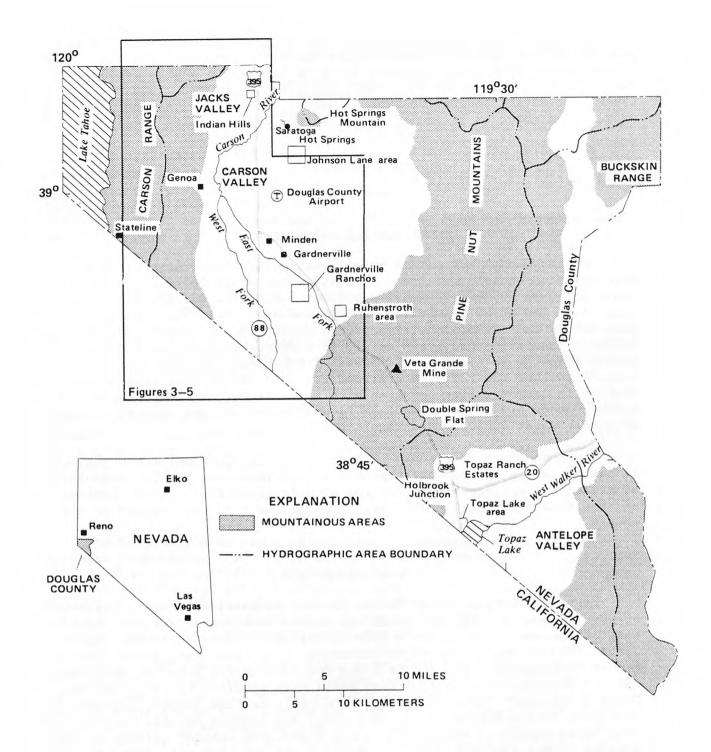


FIGURE 1.--Geographic features of Douglas County and areal extent of figures 3-5.

The mean annual temperature for 1940-81 (National Climatic Center, 1941-82) at the Douglas County Airport was 49 °F, and mean monthly temperatures ranged from a low of about 31 °F in January to a high of 69 °F in July. Mean annual temperature (National Climatic Center, 1959-82) at Topaz Lake during the period 1958-81 was 50 °F; mean monthly temperatures ranged from about 33 °F in January to about 70 °F in July.

The Indian Hills area, which local residents also refer to as Jacks Valley, is called the Indian Hills-Jacks Valley area in this report to avoid confusion with Jacks Valley proper.

### Population and Economy

Douglas County has one of the fastest population growth rates in Nevada. Between 1970 and 1980, population increased 182 percent, from about 6,900 to 19,400 (Governor's Office of Planning Coordination, 1984, page 37). The major population centers are Stateline (about 3,000) at Lake Tahoe, and the Minden-Gardnerville area (about 9,200) in Carson Valley. Population has grown mainly in response to increases in the gaming industry in northwestern Nevada. Other contributing growth factors include the relative availability and lower cost of residential housing in Carson Valley compared with that in the Lake Tahoe basin. In nearby Carson City, development in recent years has been severely limited by public water supply and sewage-disposal capacity.

The economy of the county is based primarily on tourism and gaming at casinos along the southeast shore of Lake Tahoe and in the communities of Minden and Gardnerville. Hay and alfalfa production, dairy farming, and the raising of beef cattle are also important to the economy of the county.

#### Acknowledgments

Darrel Rasner, Consumer Health Protection Services, Nevada Department of Human Resources (NCHPS), permitted access to his agency files where most of the water-quality data used in this study were obtained. Marvin Tebeau, Nevada Department of Environmental Protection, and Robert Downer, Downer Engineering, provided well-log information for selected municipal wells. Physical locations of some wells were provided by Gault Schenk, Genoa Postmaster; Maureen Crane, Gardnerville Ranchos General Improvement District; Shiela Robison, Town of Minden; and Jeannie Cordes, Gardnerville Town Water Company. Cathie Kite, Jack Moore, and Deborah Hildman of the Douglas County Assessor's Office provided the use of county computers to help locate well owners. Dave LaBarbara of the Minden-Gardnerville Sanitation District permitted water sampling of two monitoring wells. Numerous residents permitted access to and water sampling of their wells. The help of these individuals and agencies, and others who provided assistance is greatly appreciated.

#### HYDROGEOLOGY

# Generalized Geology

The Carson Range on the western side of Douglas County comprises mostly granitic igneous rocks (figure 2), including quartz monzonite and granodiorite (Moore, 1969, plate 1). The Pine Nut Mountains on the east-tern side of the county comprise partly consolidated basin-fill deposits of Tertiary age, including well-bedded, fine-grained, tuffaceous mudstone, siltstone, and shale (Moore, page 12); volcanic rocks, including andesite, rhyolite, and basalt; metavolcanic rocks; metasedimentary rocks, including shale, slate, tuffaceous siltstone, sandstone, and graywacke; and granitic rocks (Moore, plate 1). Erosional remnants of the Tertiary Hartford Hill Rhyolite Tuff of former usage are exposed in the Pine Nut Mountains and Buckskin Range, in the northeast part of the county (Moore, page 18).

Antelope and Carson Valleys are underlain by sedimentary basin-fill deposits of Quaternary age, derived from the surrounding mountainous areas. According to Donald H. Schaefer (U.S. Geological Survey, oral communication, 1986) qualitative inspection of the gravity map of Plouff (1984) shows the depth to bedrock in the Topaz Lake area of Antelope Valley to be less than 500 feet. Maurer (1985, plate 2) found that depth to bedrock in most of Carson Valley is greater than 1,000 feet, with a maximum depth greater than 5,000 feet on the west side of the valley.

A study of the topographic positions of the hot springs in Douglas County by Moore (1969, page 16) has shown that the hot springs are situated at topographic lows on the trace of a main fault line and that the logical point for water traveling along this fault plane to reach the surface would be at the lowest point.

#### Ground Water

The following discussion of ground-water movement in Carson Valley is taken from Maurer (1986). In Carson Valley, ground water occurs under both confined and unconfined conditions (Walters and others, 1970, page 16; Spane, 1977, page 139); however, no single confining layer exists over the entire valley (Dillingham, 1980). Ground-water movement in both confined and unconfined aquifers is in the direction of the hydraulic gradient, generally from areas of recharge to areas of discharge.

Water-level contours drawn using water-level measurements made during 1982 are shown in figure 3 for the unconfined aquifer and in figure 4 for the confined aquifer. Direction of water movement is perpendicular to the contour lines and is indicated in the figures by arrows. In the Jacks Valley area, water movement is southeast toward Carson Valley. In the eastern part of Carson Valley, ground-water movement is generally toward the center of the valley (toward the Carson River), then north.

Depth to water in wells completed in the unconfined aquifer ranges from about 5 feet on the valley floor to about 100 feet near the margins of the valley (figure 5). Wells completed in confined aquifers generally flow, and confined heads ranged from 5 to 20 feet above land surface.

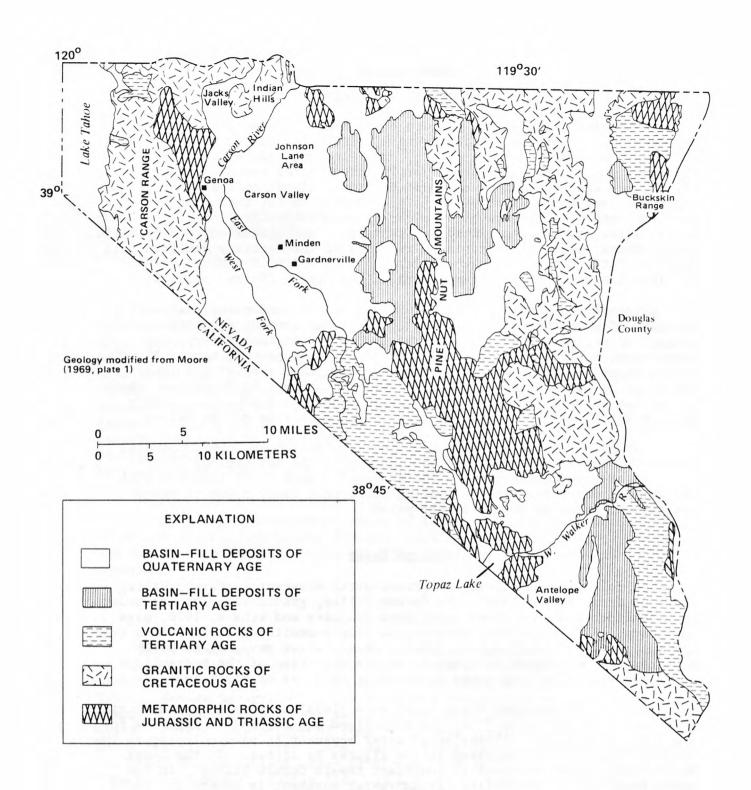


FIGURE 2,--Generalized geology.

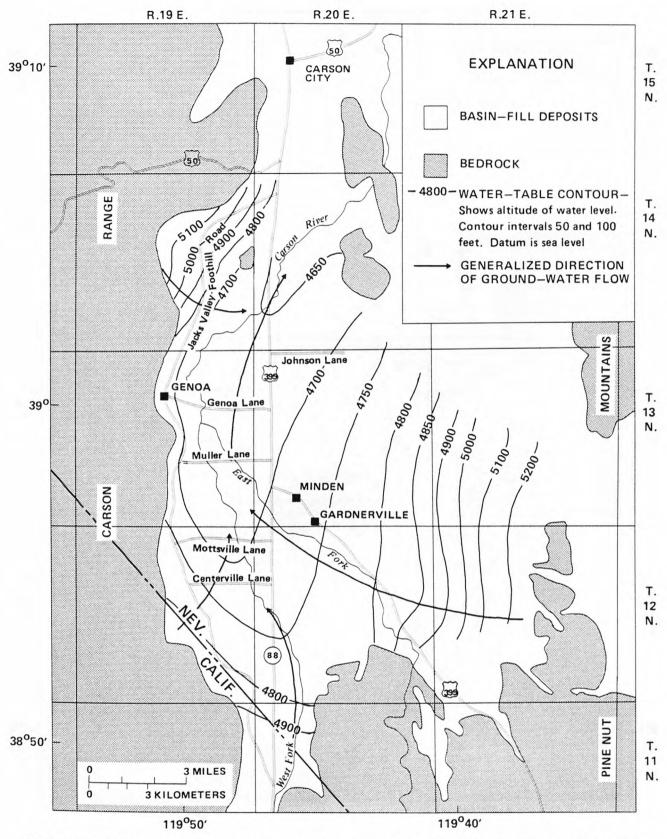


FIGURE 3.--Altitude of unconfined ground-water level in Carson Valley, May 1982 (from Maurer, 1986, fig. 3).

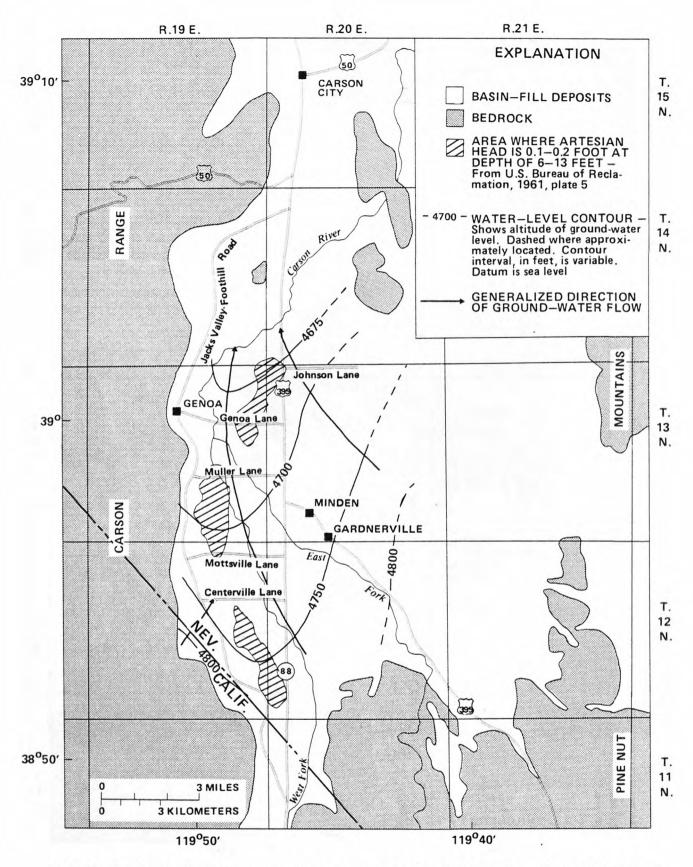


FIGURE 4.--Altitude of confined ground-water level in Carson Valley, May 1982 (from Maurer, 1986, fig. 3).

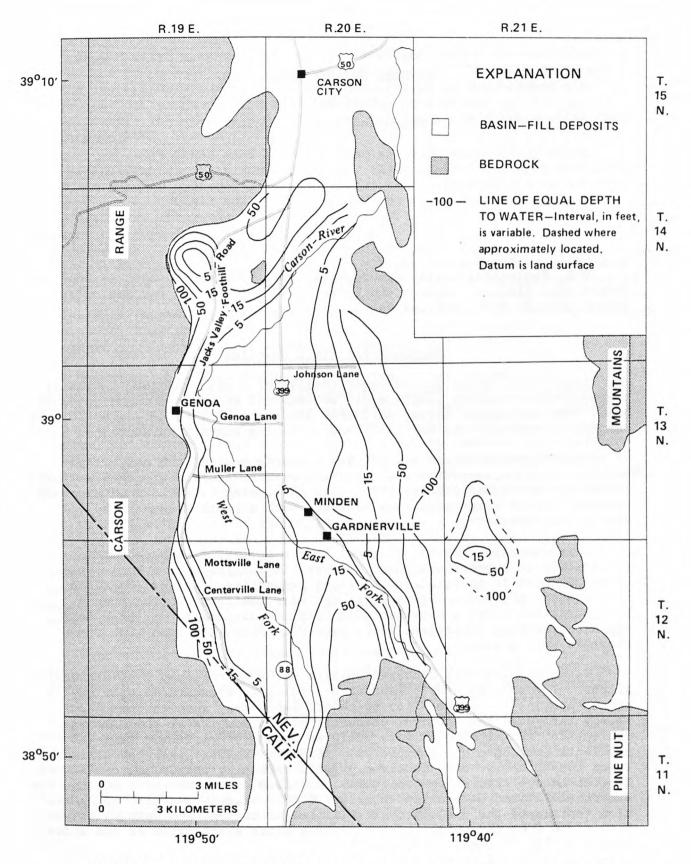


FIGURE 5.--Approximate depth to ground water in Carson Valley, 1982 (from Maurer, 1986, fig. 4).

#### METHODS OF DATA COLLECTION

#### Historical Data

# State and Local Agencies

About 1,380 water analyses were compiled from NCHPS files or published reports (1959-84 data). About 800 analyses were eliminated from the data compilation because specific well location was not provided or the given location could not be verified in the field.

# U.S. Geological Survey Data

U.S. Geological Survey ground-water quality historic data in Douglas County were limited. Glancy and Katzer (1976) list only six Geological Survey ground-water analyses for Douglas County.

# Supplemental Field Sampling

Areas that lacked ground-water quality data were examined to see if a ground-water quality sample could be obtained to broaden the areal data base. The Geological Survey collected and analyzed 2 ground-water quality samples in February and May 1983, and 33 during August and September 1983.

Specific conductance and pH were measured onsite with calibrated meters. Field bicarbonate and carbonate were determined by fixed-endpoint titration (Brown and others, 1974, page 43), using 0.1600 N sulfuric acid titrant. Water temperature was measured with a calibrated hand-held mercury thermometer.

Water samples from existing wells were collected using either a peristaltic or submersible pump; however, in one instance a bailer was used. Changes in physical and chemical properties of water (specific conductance, pH, and water temperature) were monitored during pumping to insure that samples represented aquifer conditions rather than local non-homogeneities (Claassen, 1982, page 7). Sampling began when these readings were stable.

Water samples were filtered on-site using a prerinsed 0.45-micrometer filter. The samples were preserved according to standard methods (Feltz and Anthony, 1984, pages 3-16) to obtain laboratory results that accurately reflected the chemical character of the samples at the time of collection. Water samples were analyzed for major dissolved ions and nutrients (nitrogen and phosphorus), dissolved solids, selected dissolved trace constituents, and dissolved organic carbon by the Geological Survey Central Laboratory, using procedures described in reports by Skougstad and others (1979) and Wershaw and others (1983). Dissolved-solids concentration represents the dissolved material (major cations and anions) in water determined from the weight of dry residue after evaporation of the water sample at 180 °C (Hem, 1985, page 156).

Ground-water samples analyzed for dissolved-solids concentrations by the Nevada State Health Laboratory were dried at 105 °C. The different drying temperatures, used by the two laboratories do not produce significantly different results for most of the more dilute natural waters compared with the other factors that may influence the determination (Hem, 1985, page 156).

# Data-Base Compilation

The data, after the rejection of the 800 analyses, were further reduced by comparing the balance of anions to cations for each analysis. Theoretically, the chemical balance between total milliequivalents per liter of anions and cations should be equal. Milliequivalents of a constituent are milligrams per liter divided by atomic or molecular weight, multiplied by ionic charge. Anion and cation milliequivalents seldom are equal but an analysis is generally acceptable if ions balance within a certain percentage (Hem, 1985, page 163). For this study, a 5 percent criterion was arbitrarily selected and 32 additional analysis were rejected.

The final data base (table 7) consisted of 546 analyses from 333 ground-water sampling sites (figure 6). Seventy-two of the sites (71 wells, 1 spring) had more than one chemical analysis. For purposes of interpretation within this report, mean concentrations of physical and chemical constituents from multisampled sites were computed to obtain one value per site (333 data points). Therefore, some individual analyses in the complete data base in table 7 may be higher or lower than statistics for the summary data base cited in selected tables and figures in this report.

Historically, most of the NCHPS analyses were for samples collected and submitted to the Nevada State Health Laboratory by homeowners seeking information about their domestic well water. In many such instances, the exact point of collection, the collection and preservation methods, and the type of sample container were unknown. The period of time between sample collection and laboratory analysis may have been lengthy, and under less-than-desirable storage conditions.

Many of the major dissolved constituents generally undergo little change during the period between collection and analysis. Calcium and bicarbonate, however, may decrease in concentration with time, because of precipitation of calcium carbonate. As a result, concentrations of calcium and bicarbonate, hence the concentration of dissolved solids, may be lower at the time of analysis than at the time of sampling. Denitrification (biological conversion of nitrate to nitrogen gas under anaerobic conditions) also would result in a lower concentration of nitrate in the sample at the time of analysis, but a reported nitrate concentration exceeding the Nevada public drinking-water standard would still indicate a nitrate problem in the well water.

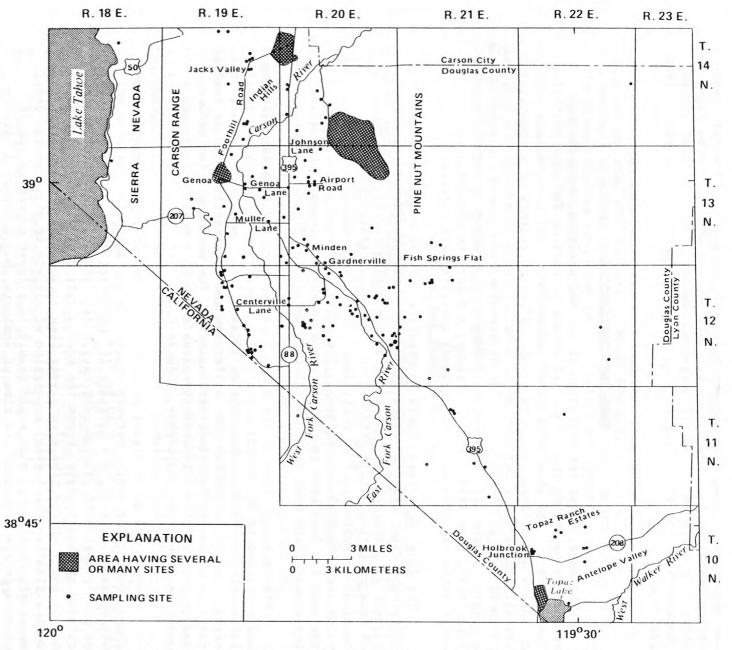


FIGURE 6.-Ground-water sampling sites for which water-quality data are available.

Concentrations of iron and manganese are determined on an aliquot that is acidified after being drawn off the unfiltered sample immediately following vigorous homogenization in the Nevada State Health Laboratory. If particulate matter containing acid-extractable iron or manganese were present in the sample at the time of collection, the measured concentrations could far exceed the amounts of iron and manganese that were in solution at the time of sampling. However, most domestic well water probably contains little, if any, particulate matter after the first few minutes of pumping. Also, if the water sample is not collected from a point between the well head and pressure tank, or if the well is not pumped adequately prior to sampling, the values obtained for iron and manganese (and for some other constituents) may not reflect the natural ground-water concentrations. Both NCHPS and U.S. Geological Survey data are considered adequate for use in evaluating general baseline waterquality conditions. However, because of the possible discrepancies, the data probably should not be used for detailed geochemical analysis.

# GROUND-WATER QUALITY

The quality of ground water in Douglas County generally is suitable for most uses. Ground water in Carson Valley differs from west to east, with generally lower dissolved solids on the west side. However, ground-water quality in the valley is suitable for most uses, except in areas near geothermal springs. Ground water associated with the geothermal springs characteristically has higher concentrations of dissolved solids, sodium, and sulfate than does the nonthermal water.

# Chemical Character of Ground Water

Trilinear diagrams can be used to show the chemical character of ground water. In a trilinear diagram, selected cations (positively charged ions--calcium, magnesium, and sodium plus potassium) and anions (negatively charged ions--bicarbonate plus carbonate, sulfate, and chloride) for each ground-water analysis are shown as a percentage of the total cations and anions, in milliequivalents per liter. The cations are plotted as single points on the left side triangle and anions on the right. Cation and anion plots for each sample then are projected into the central diamond field. A water type can be described depending on the location of the projected point in the central diamond. The trilinear type of diagram represented by figure 7 can be used to determine whether a particular water (1) is chemically similar to some other water, or (2) is a simple mixture of two chemically different water types (Hem, 1985, pages 177-179).

A water type in which one cation and one anion dominate (each amounts to 50 percent or more of the cations or anions, respectively) is designated by the names of the dominate cation and anion. A water type in which no one cation or anion dominate is designated a mixed-cation or mixed-anion type (Piper and others, 1953, page 26).

The trilinear diagrams for major dissolved constituents indicate that several types of ground water are present in Douglas County. However, no distinct boundaries, or chemical facies, delineating changes in chemical character were evident. Calcium-bicarbonate, sodium-bicarbonate, and calcium-sodium- or sodium-calcium-bicarbonate were the most common types (from about 73 percent of the sites analyzed). Calcium-magnesium-bicarbonate or magnesium-calcium-bicarbonate types (from about 11 percent of the sites sampled) were found throughout the county. Sodium-sulfate type water (from about 8 percent of the sites sampled) was most common near Saratoga Hot Springs in the Johnson Lane area. Waters from the remaining sites (about 8 percent) were of various chemical types.

Dashed lines on the trilinear diagram (figure 7) delineate high densities of data points and illustrate the type of ground water commonly found in the county. The points outside the dashed lines depict water samples that appear to differ in chemical character from the more common type of ground water for the area.

Trilinear diagrams for specific areas in the county provide a more detailed depiction of chemical character of ground water than the trilinear diagram for all sites in the county shown in figure 7. Figure 8A is a graphical representation of ground-water quality in the Topaz Lake, Holbrook Junction, and Topaz Ranch Estates areas near Antelope Valley and figure 8B represents the Double Spring Flat area and the area near Veta Grande Mine (figure 1).

Figures 8C-H show chemical character of ground water in most areas of Carson Valley. The chemical character of ground water on the west side of Carson Valley (generally west of Highway 395, figure 1) is somewhat similar between townships (figures 8C, E, G). However, the southwest part of the valley along Green Acres Drive has ground water with higher percentages of sodium plus potassium and sulfate (figure 8C) than other ground water found along the west side.

The chemical character of ground water on the east side of Carson Valley varies, generally among five water types (figures 8D, F, H). The ground water in the Minden-Gardnerville area and the southeast part of the valley are similar to one another (figures 8D and F).

Ground water near the Douglas County Airport has two water types (figure 8F); those associated with shallow wells (less than 150 feet total well depth) and deep wells (greater than 150 feet total well depth). The wells are drilled in alluvium and the ground water in the shallow wells appears to be less dilute compared with most other ground water in Douglas County. Calcium and magnesium comprise between 60 to 80 percent of the cations and sulfate and chloride about the same amount of the anions in these shallow wells near the airport. The shallow wells appear to be above an impervious clay layer of variable thickness, about 90 to 140 feet below land surface based on examination of drillers' logs in the area.

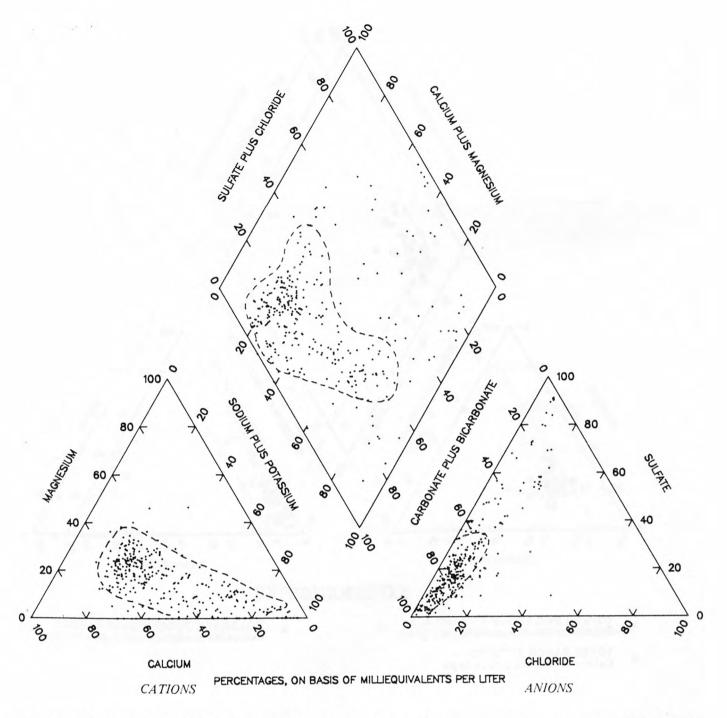
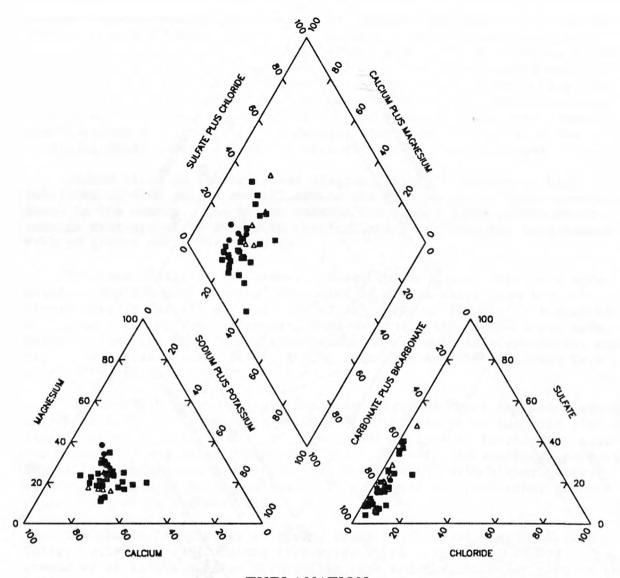


FIGURE 7.--Proportions of major dissolved constituents in ground water. Dashed lines enclose areas of most typical chemical character for sampling sites in study area. By convention, potassium, normally a minor constituent, is combined with sodium in calculating cation percentages. Similarly, carbonate, when present, is combined with bicarbonate. See text for discussion of this type of diagram.



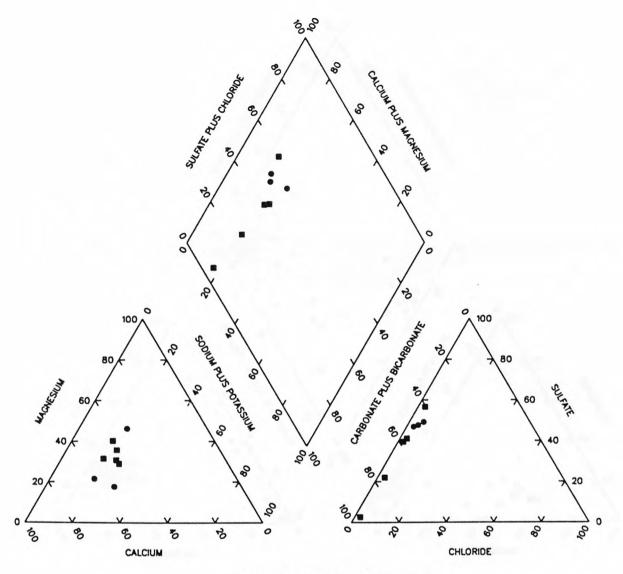
**EXPLANATION** 

- TOPAZ LAKE AREA--Calcium-bicarbonate type
- Δ TOPAZ RANCH ESTATES -- Calcium-bicarbonate type

HOLBROOK JUNCTION -- Calcium-bicarbonate type

A. Township 10 N., Range 22 E. (southern part of Douglas County); total, 42 samples.

FIGURE 8.-Proportions of major dissolved constituents in ground water. See figure 7 and text for discussion of this type of diagram.



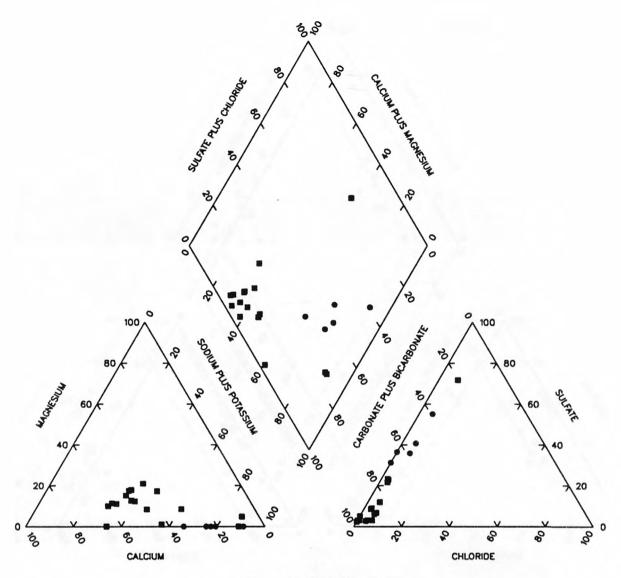
**EXPLANATION** 

DOUBLE SPRINGS FLAT -- Calcium-bicarbonate type

VETA GRANDE MINE AREA --Calcium-sulfate-bicarbonate type

B. Township 11 N., Range 21 E. (south-central part of Douglas County); total, eight samples.

FIGURE 8.-Continued.

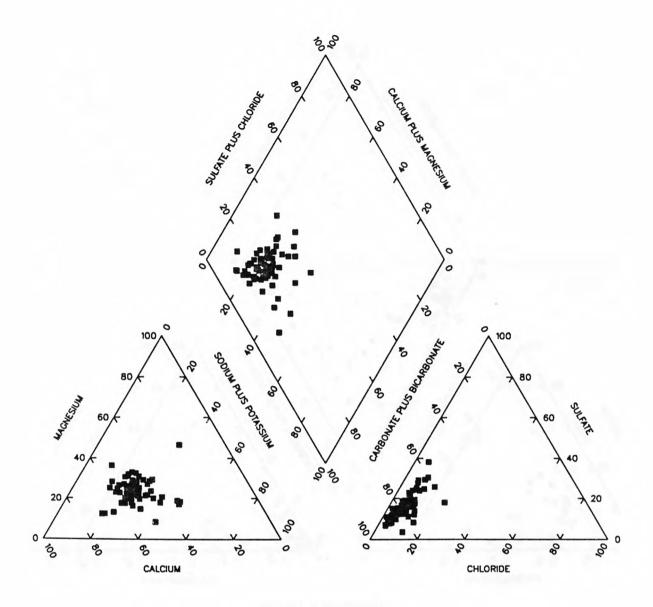


**EXPLANATION** 

 KINGSBURY AND FOOTHILL AREA --Calcium-sodium-bicarbonate type GREEN ACRES DRIVE AREA --Sodium-bicarbonate type

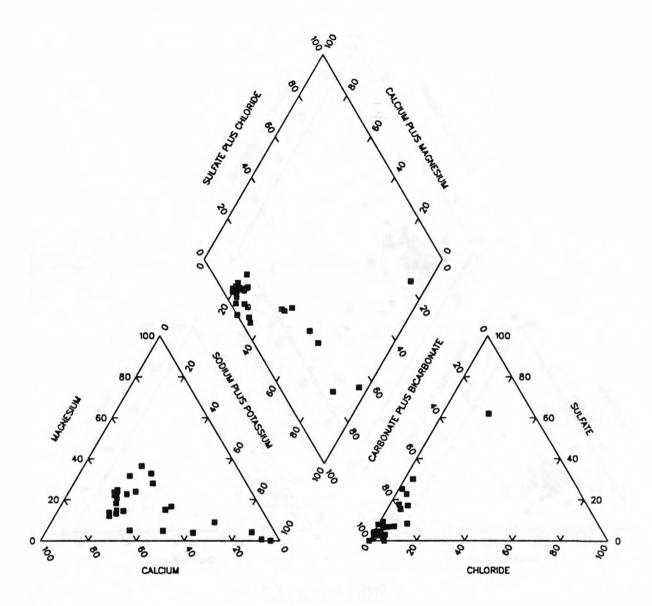
C. Township 12 N., Range 19 E. (southwest part of Carson Valley); total, 21 samples.

FIGURE 8.-Continued.



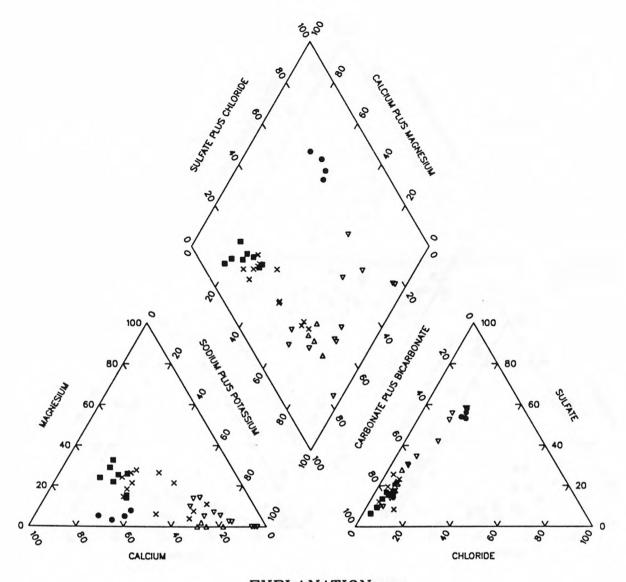
D. Township 12 N., Range 20 E. (southeast part of Carson Valley). Total, 68 samples; dominant chemical type, calcium—bicarbonate.

FIGURE 8 .- Continued.



E. Township 13 N., Range 19 E. (west-central part of Carson Valley). Total, 28 samples; dominant chemical type, calcium-sodiumbicarbonate.

FIGURE 8.-Continued.

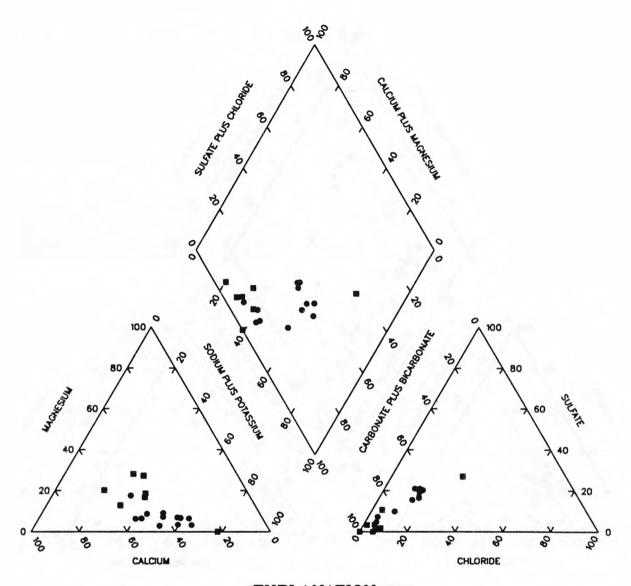


**EXPLANATION** 

- MINDEN-GARDNERVILLE AREA -- Calcium-bicarbonate type
- Δ DEEP WELLS, AIRPORT AREA -- Sodium-bicarbonate type

- SHALLOW WELLS, AIRPORT AREA --Calcium-sulfate type
- ∇ EAST JOHNSON LANE -Sodium-bicarbonate type
- OTHER SITES Sodium-calcium-bicarbonate type
- F. Township 13 N., Range 20 E. (east-central part of Carson Valley); total, 40 samples.

FIGURE 8 .- Continued.

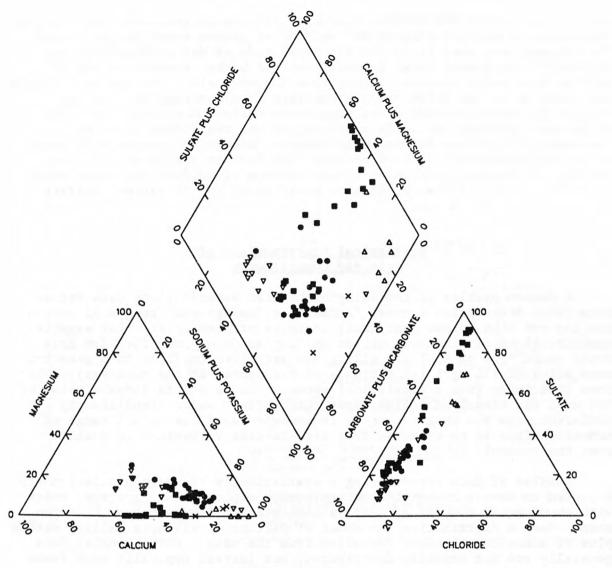


**EXPLANATION** 

JACKS VALLEY --Calcium-sodium-bicarbonate type JACKS VALLEY-INDIAN HILLS --Sodium-bicarbonate type

C. Township 14 N., Range 19 E. (northwest part of Carson Valley and Jacks Valley); total, 19 samples.

FIGURE 8.-Continued.



**EXPLANATION** 

- NORTH JOHNSON LANE --Sodium-sulfate type
- Δ EAST JOHNSON LANE --Sodium-sulfate type

- CENTRAL JOHNSON LANE --Sodium-bicarbonate type
- JACKS VALLEY-INDIAN HILLS --Sodium-calcium-bicarbonate type
- OTHER SITES --Sodium-bicarbonate type

H. Township 14 N., Range 20 E. (northeast part of Carson Valley); total, 70 samples.

FIGURE 8 .- Continued.

Some ground water north of Stephanie Way and east of East Valley Road (figure 9) in the Johnson Lane area (figure 1) generally has a higher percentage of sulfate (figure 8H) than other ground water in the county. The Johnson Lane area is on the southwest side of Hot Springs Mountain (figure 1) and ground water in wells contain higher concentrations of sulfate than wells located farther from the mountain. The analyses from two wells on Porter Drive and the analysis from Saratoga Hot Springs (figure 9) indicated that water from these wells and springs had over 90 percent sulfate; two wells southwest of the two Porter Drive wells are downgradient from Hot Springs Mountain, and contain water with about 49 and 53 percent sulfate. Generally, the farther west a well is from Hot Springs Mountain the lower the percentage of sulfate in ground water. Figure 9 also shows the approximate position of the 50-percent sulfate boundary by a dashed line.

# <u>Statistical Distributions of</u> <u>Selected Constituents</u>

A common problem in reviewing historical water-quality data for an area is to distinguish between "normal" or "background" ranges of concentration and high values that truly indicate problem areas. For example, concentrations of dissolved solids in Douglas County compiled for this study range from 50 to 1,170 milligrams per liter (mg/L) with a geometric mean value of 221 mg/L (an estimate of the median of the data set). What does this imply from a statistical sense? How is one to judge a value of 600 mg/L for dissolved solids; does that reflect water significantly different from the average, or is it an example of the normal range of concentrations to be expected from site-to-site variations in quality over the county?

Samples of data representing a statistically "normal" population are expected to have a symmetrical distribution about a mean (average) value with about equal numbers of samples both greater than and less than the mean. Such a distribution has about 67 percent of all data falling within plus or minus one standard deviation from the mean. Environmental data generally are not normally distributed, but instead typically have fewer values higher than the mean and more values lower than the mean. Statistics such as the standard deviation have less meaning for such data; however, the logarithms of the actual data may be normally distributed, providing what is termed a log-normal distribution.

One tool available to examine the statistical distribution of water-quality data is a plot of the distribution of observed concentrations on special probability graph paper (Velz, 1970, page 522-542; Sinclair, 1974). If the concentrations of dissolved solids in the ground water of an area have a statistically normal distribution, most points will plot on a straight line on such paper, and the distribution can be described by a mean value and by the standard deviation of that mean.

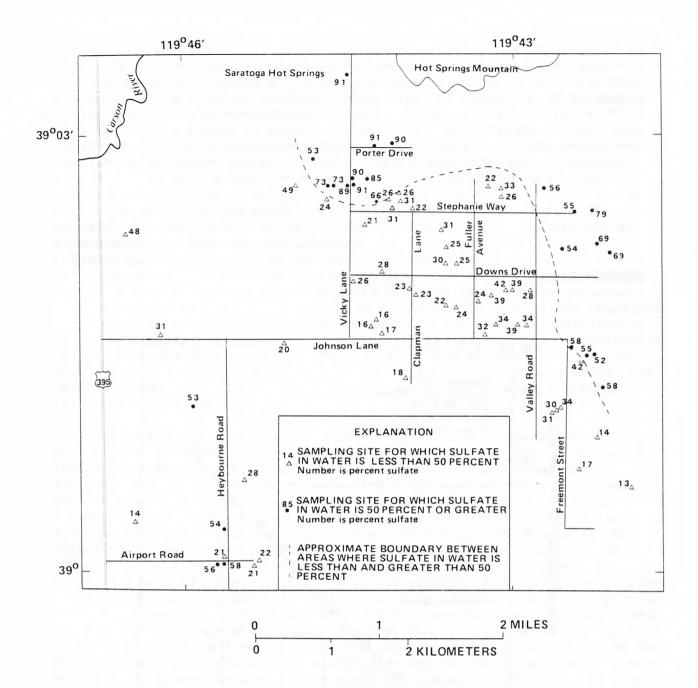


FIGURE 9.-Proportion of sulfate in water from sampling sites in the Johnson Lane area. Number indicates sulfate concentration as percentage of combined concentration of major anions (that is, sulfate plus chloride, carbonate, bicarbonate, and nitrate), with concentrations expressed in milliequivalents per liter.

The same technique can be used for data that are log-normally distributed by using graph paper with a logarithmic y axis. If data for a given water-quality constituent represent one log-normal population, they will form a straight-line plot on a log-probability graph. For example, the frequency distribution of sodium concentrations in Douglas County ground water is plotted in figure 10. The concentration is plotted on a vertical axis that is logarithmic (tic marks are spaced by powers of 10) and the horizontal axis is a special scale showing the probability of exceeding a particular concentration of sodium. Points shown in the illustration represent probability in levels of about 5 percent.

This graph imparts several types of information regarding the distribution of sodium in ground water. First, because the data plot is a fairly straight line on the log-probability graph, one can assume that the sodium analyses represent one population of data with a log-normal distribution.

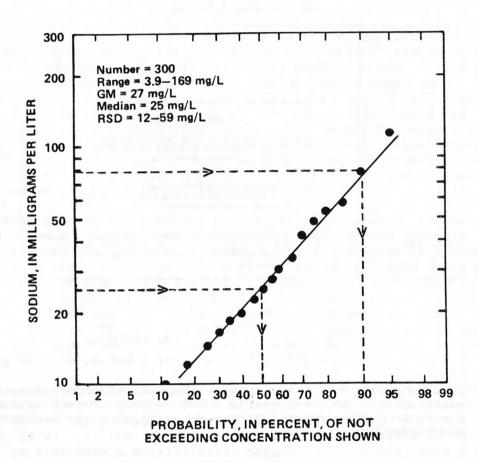


FIGURE 10,—Frequency distribution for sodium in ground water.

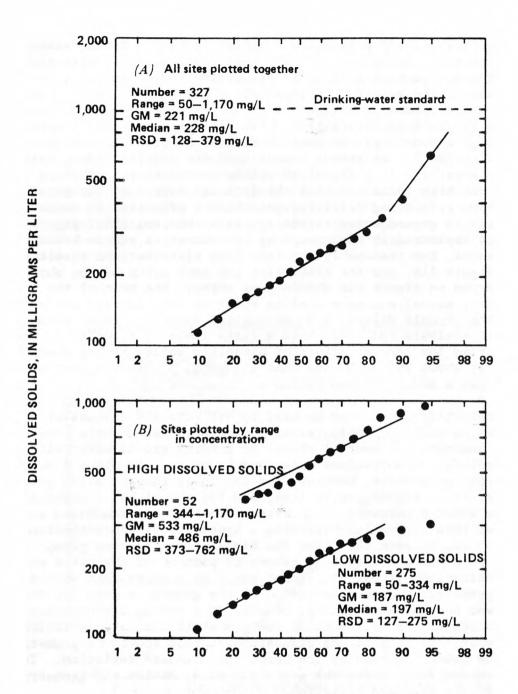
Dashed lines show the following examples: The probability that an individual sodium concentration will not exceed 25 milligrams per liter is only 50 percent; in contrast, the probability of not exceeding 78 milligrams per liter is about 90 percent. Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

Furthermore, useful statistics may be read directly off the graph. In the example for sodium (figure 10), the concentration corresponding to the 50-percent probability (25 mg/L) is the median concentration of sodium in the sampled ground water of Douglas County. Other probabilities of not exceeding a given concentration may be directly obtained; for example, if an industry wishing to locate in the county required water with a sodium concentration not exceeding 78 mg/L, figure 10 shows that about 90 percent of the sampled wells meet that criterion.

Figure 11A is a log-probability plot for dissolved-solids concentrations in Douglas County ground water. Unlike the plot for sodium, the data seem to follow two linear trends rather than one straight line, with the two segments intersecting at a dissolved-solids concentration of about 340 mg/L. Such a distribution indicates that the data may represent two populations of sites, perhaps reflecting differing geochemical processes or sources of dissolved solids in ground water (Sinclair, 1974; Nowlin, 1982, page 24). the data are replotted in two groups by concentration, those below 340 mg/L and those above, two reasonably straight-line distributions result as presented in figure 11B, and the statistics for each group may be obtained. The lines drawn on figure 11B should pass through the mean of the logarithm (the geometric means) and have a slope equal to the standard deviation of the logarithm (Dennis Helsel, U.S. Geological Survey, written communication, 1987). This analysis indicates that a large group of sites (275) exists with relatively low concentrations of dissolved solids having a median concentration of about 200 mg/L, and that a second, high-concentration group (52 sites) has a median concentration of about 490 mg/L.

The probability plots may be used to estimate the chances of finding ground-water exceeding drinking-water standards. For example, the data in figure 11A indicate the overall chance of finding ground water with dissolved-solids concentrations exceeding 1,000 mg/L is about 2 percent for the county as a whole; however, for the lower concentration group (figure 11B) it is significantly less, and for the higher concentration group it is about 6 percent. With respect to the question posed at the beginning of this discussion regarding a hypothetical concentration of 600 mg/L, it can be seen that, for the higher concentration group, this would only be a bit above average (about 65 percent of the sites would have lower concentrations, about 35 percent would be higher), and within plus or minus one geometric standard deviation of the geometric mean for the group. For the lower concentration group, however, the 600 mg/L concentration would be quite abnormal, and well outside one geometric standard deviation of the geometric mean. To compute one standard deviation above the geometric mean, multiply the geometric mean by the geometric standard deviation. one standard deviation below the geometric mean, divide the geometric mean by the geometric standard deviation.

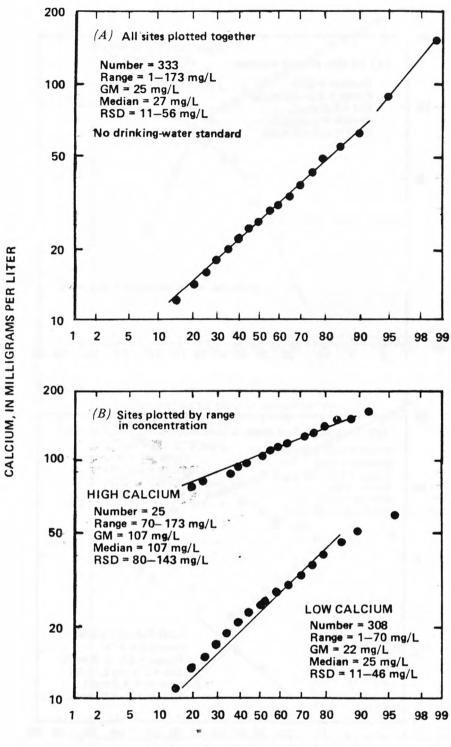
Log-probability plots for some of the common ions contributing to dissolved solids (calcium, magnesium, sodium, bicarbonate, sulfate, chloride, fluoride, and nitrate) and selected minor elements (iron, manganese, zinc, and arsenic) are shown in figures 10-20. Of the common ions, only sodium (figure 10) appears to be adequately represented by a single lognormal population. Arsenic, iron, manganese, and zinc also plot as one lognormal distribution (figures 19 and 20). All the others may be divided into two discrete lognormal populations—a lower concentration group generally representing two-thirds or more of the sampled sites, and a higher concentration group generally representing less than one-third of the sites.



# PROBABILITY, IN PERCENT, OF NOT EXCEEDING CONCENTRATION SHOWN

FIGURE 11.--Frequency distribution for dissolved solids in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.



PROBABILITY, IN PERCENT, OF NOT EXCEEDING CONCENTRATION SHOWN

FIGURE 12.—Frequency distribution for calcium in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

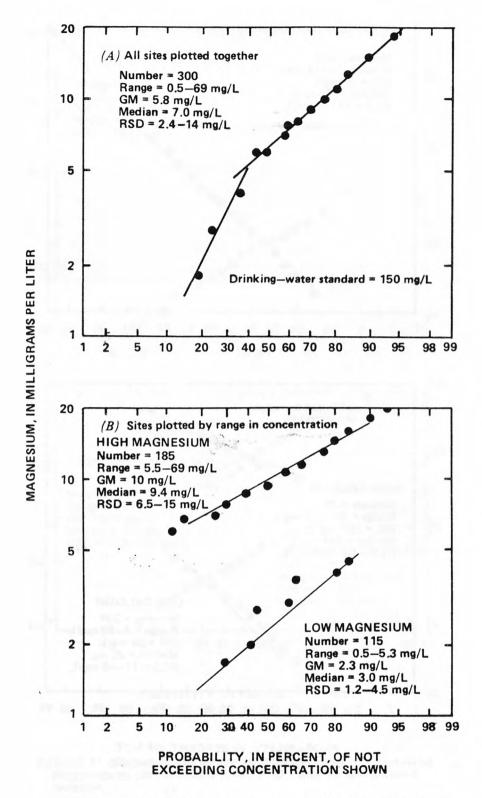


FIGURE 13.—Frequency distribution for magnesium in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

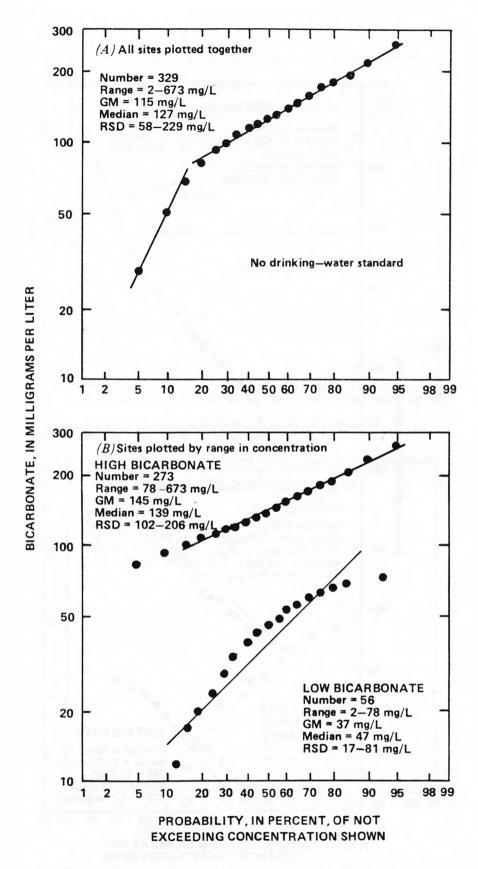


FIGURE 14.—Frequency distribution for bicarbonate in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

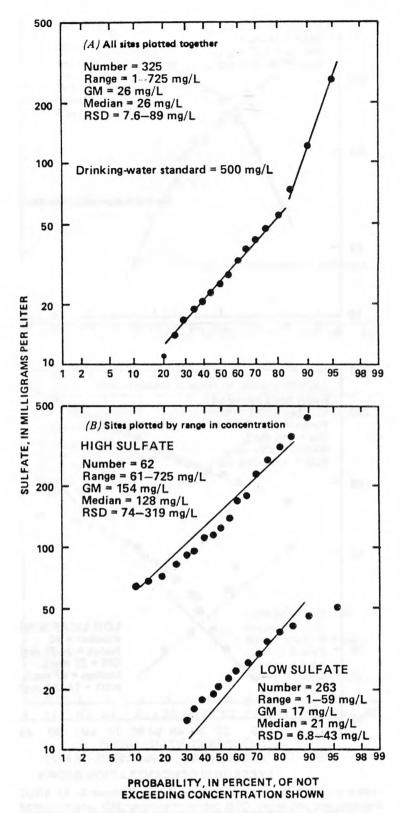


FIGURE 15.—Frequency distribution for sulfate in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

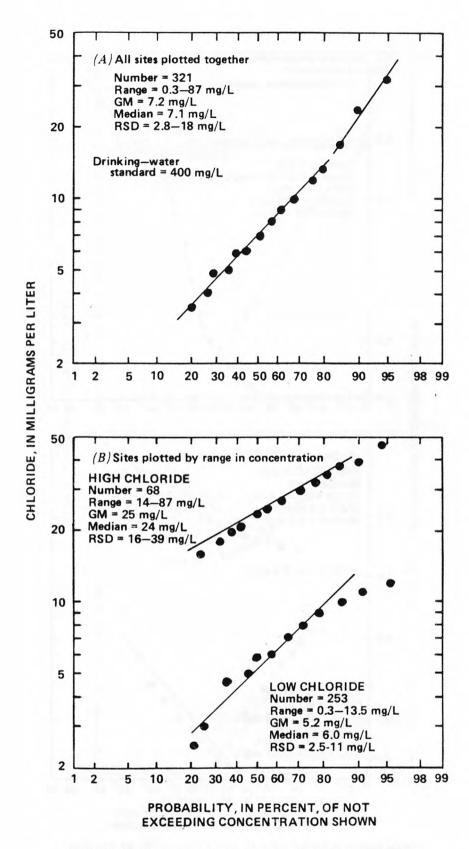


FIGURE 16.—Frequency distribution for chloride in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

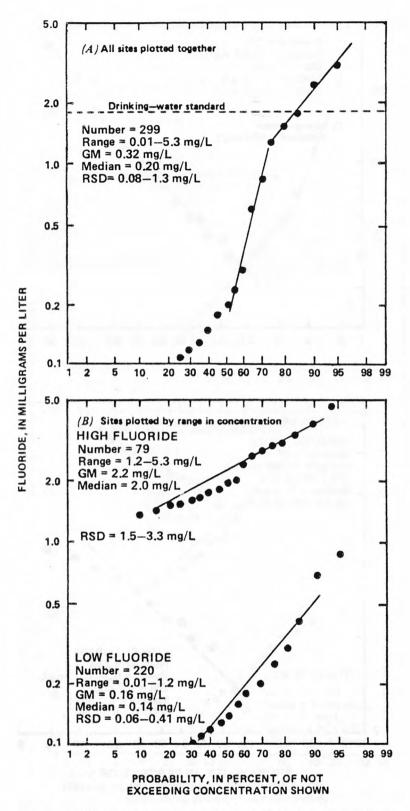


FIGURE 17.—Frequency distribution for fluoride in ground water.
Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

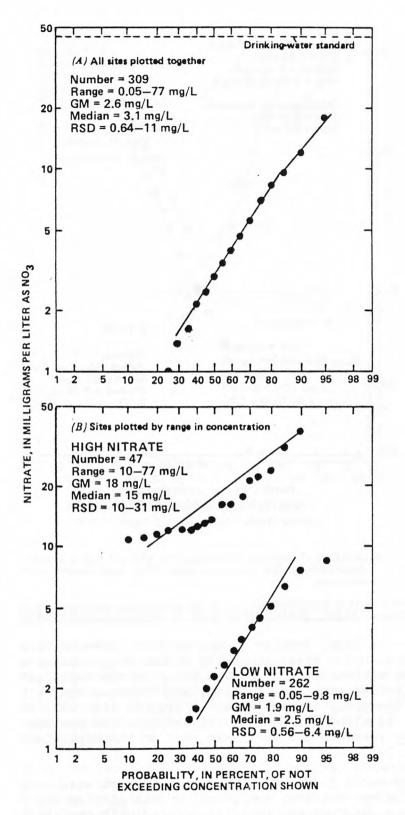


FIGURE 18.—Frequency distribution for nitrate in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

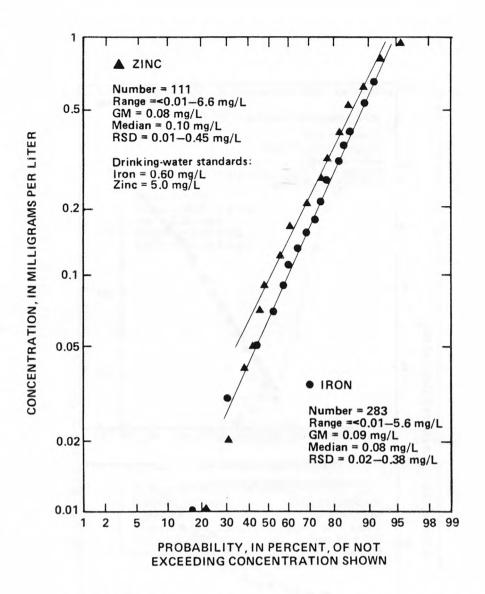


FIGURE 19.--Frequency distributions for iron and zinc in ground water.

Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

# Areal Distribution of Selected Constituents

The Topaz Lake, Douglas County Airport, Johnson Lane, and Jacks Valley-Indian Hills areas generally differ in ground-water chemistry from other parts of the county. Ground water in the Topaz Lake area contains dissolved solids, calcium, sulfate, and nitrate, that plot in the high concentration group (upper graphs on figures 11B, 12B, 15B, and 18B). This water is along the contact between alluvial-fan deposits and metasedimentary rocks in the northwest part of the Topaz Lake area (Nowlin, 1982, page 24).

Ground water in the Douglas County Airport area contains dissolved solids, calcium, sulfate, and chloride that plot in the high concentratio group. This water generally is associated with the shallow wells (less than 150 feet deep) in the area.

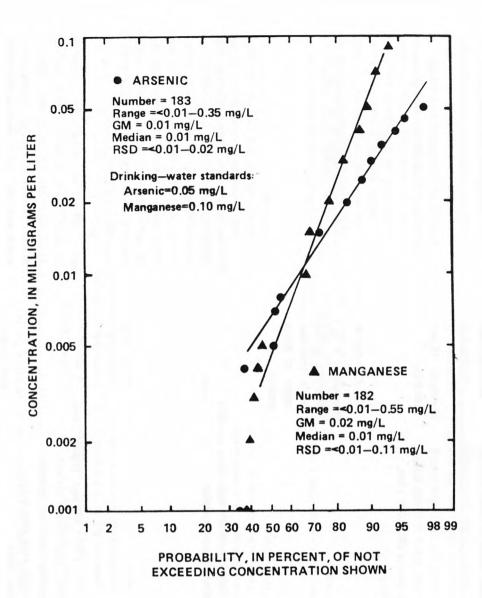


FIGURE 20.--Frequency distributions for arsenic and manganese in ground water. Abbreviations: GM, geometric mean; RSD, range for one standard deviation.

Ground water in the Johnson Lane area contains dissolved solids, calcium, sulfate, chloride, and fluoride that are in the high concentration group. Ground water in the northeast part of this area generally contains high concentrations of sulfate that reflect mixing of sulfate-rich geothermal water.

Ground water in the Jacks Valley-Indian Hills area contains fluoride and chloride that are in the high concentration group. The water in this area also is influenced by geothermal water but does not contain sulfate concentrations as high as those in the northeastern part of the Johnson Lane area.

# TABLE 1.--Source, significance, and standards or criteria for dissolved constituents and properties of water

[Modified slightly from Nowlin (1982, table 2). Abbreviations: NCHPS, Nevada Consumer Health Protection Services, 1977; EPA, U.S. Environmental Protection Agency, 1976; mg/L, milligrams per liter]

Constituent or characteristic	Source or cause of occurrence	Normal range in concentration	Standards or 1 criteria for use	Significance for use
Color	Organic compounds derived from decaying organic matter, peat, and other natural organic deposits, industrial wastes, and sewage.	Generally low in ground water; surface water draining swamps may exceed 200 units (Hem, 1985).	Standard: 15 units (NCHPS).	Esthetically objectionable in drinking water and in food and beverage processing.
Turbidity	Suspended sediments, precipitates, and colloids. May be derived from soil erosion, industrial wastes, sewage, or chemical reactions, such as the oxidation of dissolved iron.		Standards: 1 to 5 standard turbidity units depending upon number of hookups (NCHPS).	Esthetically objectionable in drinking water.
Hardness (as CaCO <sub>3</sub> )	Derived principally from dissolved calcium and magnesium. With the exception of alkalies such as sodium and potassium, all metals in water contribute to hardness.	Commonly 200 to 300 mg/L in carbonate ground water.	None	Consumes soap and detergents before a lather will form, resulting in soap curds depositing on sinks and bathtubs. Hard water form scales in pipes, boilers, and water heaters. Water of hardness up to 60 mg/L considered soft; 61 to 120 mg/L, moderately hard; 121 to 180 mg/L, hard; more than 180 mg/L, very hard.
Calcium (Ca) and magnesium (Mg)	Dissolved from rocks and soils, especially those containing limestone, dolomite, and gypsum.	Calcium: 1 to 1,000 mg/L (Todd, 1970). Magnesium: Normally much less than calcium and usually less than sodium. (Hem, 1985).	None	Impart hardness and scale-forming properties to water (see hardness). High concentrations unsuitable for laundries, steam plants, textile processing, dyeing, and electroplating. Small amounts desirable to prevent corrosion.

streets and highways.

TABLE 1.--Source, significance, and standards or criteria for dissolved constituents and properties of water--Continued

Constituent or characteristic	Source or cause of occurrence	Normal range in concentration	Standards or 1 criteria for use	Significance for use
Sodium (Na) and potassium (K)	Dissolved from most rocks and soils. High concentrations may be found in natural brines, industrial waste, and sewage.	Sodium: Generally 1 to 1,000 mg/L (Hem, 1985). Potassium: Commonly 0.1 to 0.5 times sodium; generally less than 10 mg/L.	Criteria: Sodium, 10 to 200 mg/L; Potassium, 1,000 to 2,000 mg/L (McKee and Wolfe, 1963).	Concentrations greater than 50 mg/L may cause foaming in boilers. Combine with chloride to impart salty taste. Sodium may contribute to hypertension and cardiovascular diseases. Sodium may be objectionable in irrigation water in concentrations that vary with the type of crops and soils.
Bicarbonate (HCO <sub>3</sub> ) and carbonate (CO <sub>3</sub> )	Dissolved from most rocks and soils by carbon dioxide reacting with carbonate minerals such as limestone and dolomite. The carbonate species (CO <sub>3</sub> ) can only exist if pH is 8.3 or more.	Bicarbonate: Generally less than 200 mg/L in surface water and 500 mg/L in ground water. Carbonate: Generally less than 10 mg/L.	None	Increase the alkalinity and usually the pH of water. In combination with calcium and magnesium, causes scales in pipes and, upon heating, may release corrosive carbon dioxide.
Sulfate (SO <sub>4</sub> )	Dissolved from rocks and soils containing gypsum and sulfide or sulfate minerals. Commonly associated with coal deposits, metallic ore deposits, and geothermal areas. May be derived from industrial wastes and atmospheric pollution.	Generally ranges from 1 to 1,000 mg/L.	Standards: 250 mg/L unless alternate supply unavailable, then 500 mg/L (NCHPS).	Forms boiler scale in combination with calcium. Causes bitter taste when combined in high concentrations with other ions and may have laxative effects when ingested in higher concentrations than an individual is accustomed to. Combines with hydrogen ions in low pH water to form sulfuric acid.
Chloride (Cl)	Dissolved in varying amounts from all rocks and soils. High concentrations may be derived from marine and desert evaporites containing brines. Commonly present in sewage and industrial wastes. May be derived from salts used for control of ice on	Commonly less than 100 mg/L in potable water (Hem, 1985).	Standards: 250 mg/L unless alternate supply unavailable, then 400 mg/L (NCHPS).	May contribute to corrosiveness of water.  Imparts salty taste in concentrations as low as 100 mg/L. The chloride ion is very stable in ground water and is often used as a tracer of the movement of wastes in aquifers.

Constituent or characteristic	Source or cause of occurrence	Normal range in concentration	Standards or 1 criteria for use	Significance for use
Fluoride (F)	Dissolved in small amounts from most rocks and soils. Added to many public water supplies to inhibit tooth decay.	Commonly less than 1.0 mg/L in potable natural water (Hem, 1985).	Standards: 1.4 to 2.4 mg/L depending upon average air temperatures (NCHPS).	Concentrations between 0.6 and 1.7 mg/L may have beneficial effects on the structure and resistance to decay of children's teeth. Concentrations in excess of 6.0 mg/L may cause mottling and disfiguration of teeth.
Dissolved solids	Sum of all minerals dissolved from rocks and soils. Concentrations may be increased by industrial wastes or sewage.	Ground water generally in the range of 25 to 1,000 mg/L (Hem, 1985).	Standards: 500 mg/L unless alternate supply not available, then 1,000 mg/L (NCHPS).	General indicator of the mineralization or inorganic chemical content of water.  Specific effects upon water uses depend upon the individual constituents present. Water containing more than 1,000 mg/L are unsuitable for most uses.
Nitrate (NO <sub>3</sub> )	Derived from the atmosphere or may be leached from decaying organic matter, fertilizers, sewage, industrial wastes, or animal wastes.	0 to 10 mg/L.	Standard: 45 mg/L as NO (10 mg/L as N) (NCHPS).	Concentrations in excess of 45 mg/L (as 0 3 may cause methemoglobinemia (infant cyanosis or "blue-baby" syndrome) in infants. May be internally reduced to form nitrite compounds suspected to be carcinogenic. Encourages growth of algae and other organisms that may produce objectionable tastes and odors.
Phosphorus (P), phosphate (PO <sub>4</sub> )	Derived from phosphate minerals (notably apatite) common in many rocks and soils. May be present in sewage from human or animal wastes and from additives to synthetic detergents.	•	Recommendation for fresh water aquatic life: 0.025 to 0.05 mg/L as P (EPA).	Encourages growth of nuisance algae in lakes and streams.
Arsenic (As)	Associated with volcanic minerals and metallic ore deposits.  Common in water of thermal springs.		Standard: 0.05 mg/L (NCHPS).	Toxic to humans, animals, and vegetation.

TABLE 1.--Source, significance, and standards or criteria for dissolved constituents and properties of water--Continued

Constituent or characteristic	Source or cause of occurrence	Normal range in concentration	Standards or criteria for use	Significance for use
Iron (Fe)	Dissolved from iron-bearing minerals present in most rocks and soils. Found in some industrial wastes and can be corroded from pipes, well casings, pumps, and other equipment. Also can be concentrated in wells and springs by certain bacteria.	Concentrations in ground water as high as 1,000 to 10,000 ug/L may be common in some aquifers.  Areal distribution is often erratic (Hem, 1985).	Standards: 0.3 mg/L unless alternate supply is unavailable, then 0.6 mg/L (NCHPS).	Oxidizes to a reddish-brown sediment.  Stains utensils, enamelware, clothing, and plumbing fixtures. May cause taste and odor problems objectionable for food and beverage processing.
Manganese (Mn)	Dissolved from some rocks, soils, and lake-bottom sediments. Generally associated with iron; often associated with acid drainage from mines.	Generally less than 20 ug/L; usually less than iron.	Standard: 0.10 mg/L (NCHPS).	Oxidizes to dark brown or black sediment.  Problems similar to those of iron.
Methylene-blue active substances (MBAS)	Synthetic detergents in domestic and industrial wastes.	Not found in natural water.	None	Objectionable tastes and odors, causes foam.  Presence in water serves as an indicator of recent contamination by wastes.

Standards are legal limits for concentrations or values of given constituents in water; criteria are recommended limits for specific water uses, based on current scientific knowledge. Standards and criteria quoted above are for drinking water unless otherwise stated.

The source, normal range, criteria for various types of water use, and significance of various water-quality constituents reported in this study are given in table 1. A comparison of the analyses in table 7 with these standards and criteria indicate that ground water in most of Douglas County is suitable for most drinking-water uses. Concentrations of most constituents generally increase in northerly and easterly directions in Carson Valley.

Table 2 shows the mean concentrations of selected constituents at sites where the depth is less than 200 feet and also where the depth is 200 feet or more. Generally, the ground water from depths of 200 feet or more is of slightly better quality. However, the ratio of the number of sites less than 200 feet deep to the number of sites equal to or greater than 200 feet deep is about 3.5:1, and may reflect some bias.

# Comparison of Water-Quality Conditions with Water-Quality Standards

Nevada public drinking-water standards for selected chemical constituents are listed in table 3 (Nevada Legislative Counsel Bureau, 1986). Both primary and secondary standards have been established for community-supply wells. Only primary standards for nitrate and secondary standards for other chemical constituents are applicable for noncommunity supply wells (restaurant, church). State drinking-water standards have not been established for calcium, sodium, potassium, alkalinity, bicarbonate, and carbonate. Nevada public-supply standards are not enforced for individual private water supplies; however, they are used as recommended maximum concentrations (Darrel Rasner, NCHPS, oral communication, 1986).

The following discussions describe the major ions, nitrate, and trace constituents that equaled or exceeded the recommended drinking water standards. Concentrations of copper and zinc exceeded the criteria in one well and two wells, respectively. Concentrations of chloride, magnesium, and barium were within recommended levels in all wells sampled.

#### Sulfate

Concentrations above 750 mg/L may have a laxative effect on consumers (National Academy of Science and National Academy of Engineering, 1973, page 89). Sulfate concentrations in water from six sites (eight analyses) exceeded the drinking-water standard of 500 mg/L and ranged from 514 to 725 mg/L (table 7). Five sites are in the Johnson Lane area (figures 1 and 21) near Saratoga Hot Springs, where geothermal water probably contributes to the sulfate concentration in the ground water. The sixth site is Saratoga Hot Springs.

TABLE 2.--Statistical data for selected constituents in water from sites less than 200 feet deep and sites greater than or equal to 200 feet deep

[Concentrations, in milligrams per liter, constituents were averaged for sites with more than one analysis; values are rounded.]

Constit- uent	Number of sites		25th percentile			50th percentile		75th percentile		Mean	
	Less than 200 feet	Greater than or equal to 200 feet	Less than 200 feet	Greater than or equal to 200 feet	Less than 200 feet	Greater than or equal to 200 feet	Less than 200 feet	Greater than or equal to 200 feet	Less than 200 feet	Greater than or equal to 200 feet	
Calcium Magnesium Sodium Potassium Bicarbonate	258 76 17 13 252 76 3.0 1.2 231 69 14 15 228 69 2.0 1.0 255 76 98 81		27 6.1 25 3.0 130	6.1 25 2.5	44 10 49 4.0 170	39 10 50 3.0 180	36 7.7 37 2.9 140	28 6.5 37 2.5			
Carbonate Sulfate Chloride Fluoride	251 254 256 230	76 76 76 74	0.0 13 4.0 .1	0.0 16 4.6 .1	0.0 27 7.0 .2	0.0 24 7.0 .2	0.0 48 12 1.4	1.8 51 10 1.1	60 11 .8	2.0 47 9.1 .9	
Dissolved solids	254	76	160 .	157	230	215	284	294	267	233	
Nitrate as NO3 Arsenic Barium Boron Copper	250 214 53 41 66	73 68 20 13 30	1.2 0.00 .02 .02 0.00	0.6 0.00 .03 .01 0.00	3.1 < .01 .04 .10 .02	< .01 .06 .07 0.00	7.6 .02 .08 .20	4.6 .02 .11 .14	6.2 .01 .05 .20	4.1 .01 .07 .09 < .01	
Iron Manganese Zinc	237 219 83	74 65 32	0.00 .03	.02 0.00 .01	.06 .01 .14	< .01 < .02	.17 .02 .34	.30 .02 .10	.20 .03 .30	.23 .02 .30	

### Fluoride

Fluoride in low concentrations has proven to be helpful in preventing dental caries. Significantly high concentrations of fluoride in drinking water (6 mg/L), however, can produce dental fluorosis (mottling of teeth), which primarily affects children (National Academy of Science and National Academy of Engineering, 1973, page 66).

Fluoride concentrations in water from 44 sites (48 analyses) exceeded the drinking-water standard of 1.8 mg/L and ranged from 1.8 to 5.3 mg/L (table 7). Most excessive fluoride concentrations were in the Johnson Lane and Jacks Valley-Indian Hills areas (figures 1 and 21), where geothermal water may influence the ground-water chemistry. Geothermal water generally contains 1-10 mg/L fluoride (Ellis and Mahon, 1977, page 66).

# TABLE 3.--Nevada drinking-water standards for selected chemical constituents

# [Concentrations in milligrams per liter]

Primary	y standard	Secondary standard			
Constituent	Concentration	Constituent	150 500 400 1,000		
Fluoride Nitrate as NO3 Arsenic Barium	<sup>a</sup> 1.8 45 .05 1.0	Magnesium Sulfate as SO <sub>4</sub> Chloride Dissolved solids			
		Copper Iron Manganese Zinc	1.0 .60 .10 5.0		

 $<sup>^{\</sup>rm a}$  Recommended maximum concentrations for fluoride are temperature dependent. The 1.8 mg/L is based on the maximum annual average daily air temperature at Minden, Nev., of 66.8  $^{\circ}\text{F}.$ 

#### Dissolved Solids

High concentrations of dissolved solids in drinking water can affect its taste and, like sulfate, can have laxative effects. High concentrations of dissolved solids in drinking water also can corrode or encrust water-supply pipes.

Dissolved-solids concentrations in water from three wells in the Johnson Lane area exceeded the drinking-water standard of 1,000 mg/L and ranged from 1,030 to 1,170 mg/L (table 7). These high concentrations are associated with the high sulfate concentrations near Saratoga Hot Springs (figures 1 and 21). The dissolved-solids concentrations of Saratoga Hot Springs is estimated to be about 1,100 mg/L.

Dissolved-solids concentrations in water from four wells (five analyses) in the Douglas County Airport area (figure 1) were greater than 660 mg/L and averaged 841 mg/L (table 7). Three of these wells have an average depth of about 120 feet; the depth of the fourth well is not known. Average dissolved-solids concentrations from five other wells (eight analyses) in the same area was about 166 mg/L. Four of these wells have an average depth of about 365 feet; the depth of the fifth well is not known. The clay layer mentioned earlier appears to separate the deep wells from the shallow. Dissolved-solids concentrations in water from wells drilled through the clay layer are generally less than 200 mg/L. Dissolved-solids concentrations in water from wells completed above the clay are generally greater than 600 mg/L. Thus, the area of high dissolved-solids concentrations near the airport probably is restricted to the shallower wells (figure 22). For a well having more than one analysis, dissolved-solids concentrations were averaged to provide one point for plotting in figure 22.

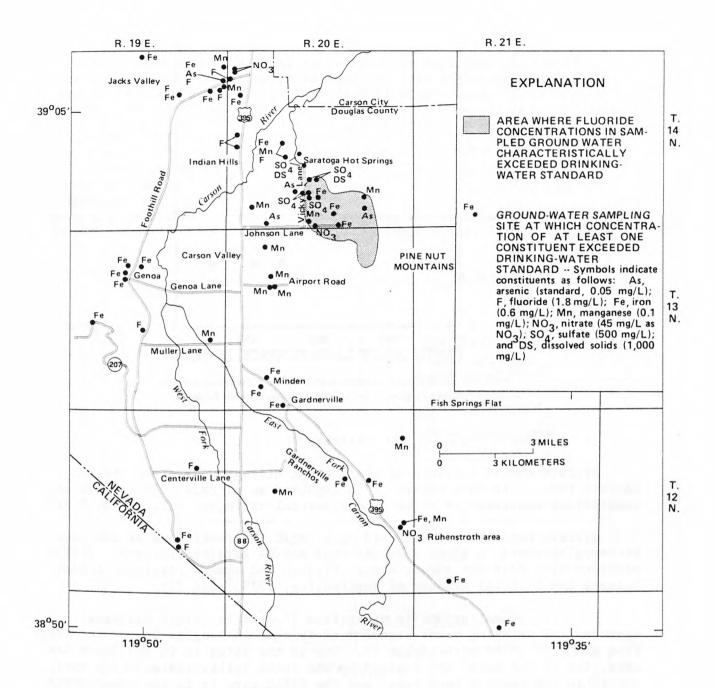


FIGURE 21.--Ground-water sampling sites in northern Douglas County where Nevada drinking-water standards were exceeded.

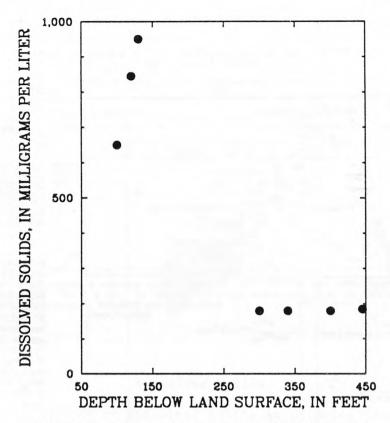


FIGURE 22,--Relation between well depth and dissolved-solids concentration in ground water near Douglas County Airport.

# Nitrate

Nitrate can be reported as the element nitrogen (N) or as the ion nitrate ( $NO_3$ ). In this report it is reported as nitrate. To convert concentrations expressed as nitrate to elemental nitrogen, multiply by 0.2259.

Nitrate concentrations exceeding 45 mg/L in drinking water can cause methemoglobinemia, a blood disorder that may be fatal to infants. This disorder usually does not affect older children and adults (National Academy of Science and National Academy of Engineering, 1973, page 73).

Nitrate concentrations in water from five wells (eight analyses) exceeded the drinking water standard of 45 mg/L in Douglas County and ranged from about 48 to 84 mg/L (table 7). One of the sites is in the Topaz Lake area, two of the sites are located in the Jacks Valley-Indian Hills area, one is in the Johnson Lane area, and the fifth site is in the Ruhenstroth area (figures 21 and 23). Sources of high nitrate in ground water may be from dissolution of mineral deposits, contamination from septic tank effluent, leaching of fertilizers, and buried decomposed natural organic material (Feth, 1966).

Nowlin (1982, page 46) states that elevated nitrate concentrations in the Topaz Lake area may indicate contamination by septic-tank effluent. Although the concentrations did not exceed the drinking-water criteria, they were within 50 percent of applicable State standards for public-water supplies.

#### Arsenic

Concentrations of arsenic exceeding 0.05 mg/L can cause mild chronic poisoning resulting in fatigue and loss of energy (National Academy of Science and National Academy of Engineering, 1973, page 56). Arsenic concentrations equaled or exceeded the drinking-water standards of 0.05 mg/L in water from 12 wells (21 analyses) in Douglas County and ranged from 0.05 to 0.38 mg/L. Two of the wells are in the Johnson Lane area, one is along Johnson Lane, and another is in the Jacks Valley-Indian Hills area (figure 21). The other eight wells (17 analyses) are in the Topaz Lake area (figure 23), and averaged about 0.15 mg/L of arsenic with a maximum of 0.38 mg/L. The occurrence of arsenic in the Topaz Lake area probably is related to mineralization along the alluvium-bedrock contact at the northern boundary of the alluvial fan (Nowlin, 1982, page 46).

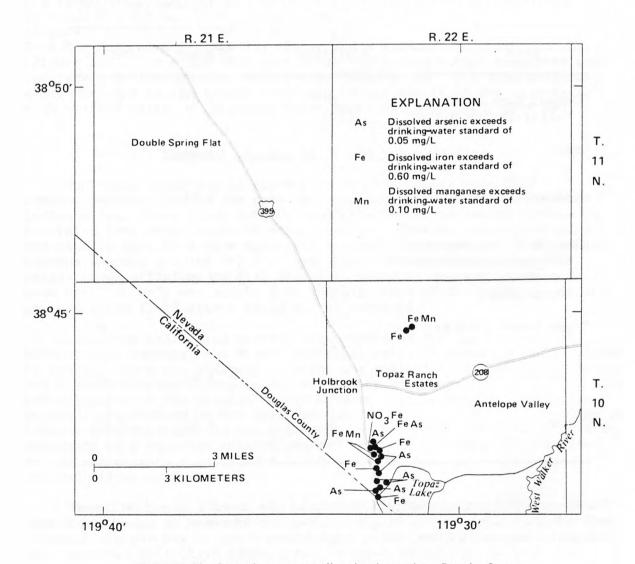


FIGURE 23.--Ground - water sampling sites in southern Douglas County where Nevada drinking - water standards were exceeded.

Concentrations of iron exceeding 0.30 mg/L in drinking water may make water esthetically unsuitable for use. Excessive iron may cause spotting of laundered clothes, iron deposits in drinking-water supply pipes, and metallic taste (National Academy of Science and National Academy of Engineering, 1973, page 69).

Iron concentrations exceeded the drinking-water standard of 0.60~mg/L in water from 33 wells (39 analyses) in Douglas County (figures 21 and 23) and ranged from 0.63~to~5.6~mg/L.

	Number of sites		Iron concentration (milligrams per liter)			
Area	(number of analyses)	Highest	Average			
Johnson Lane, Jacks Valley-Indian Hills	6 (7)	2.01	1.05			
Genoa	4 (4)	1.13	.93			
Minden-Gardnerville	3 (3)	1.76	1.27			
Along U.S. Highway 395 between southeastern						
Carson Valley and Topaz Lake	5 (6)	2.50	1.36			
Topaz Ranch Estates	2 (2)	. 99	.96			
Topaz Lake	7 (11)	3.42	1.46			
Other	6 (6)	5.60	2.01			

The excessive iron concentrations may be the result of oxidation of reduced iron minerals; or biogeochemical reactions with organic and inorganic materials (Hem, 1985, page 77-84).

### Manganese

Concentrations of manganese exceeding 0.05 mg/L affect drinking water in a manner much like iron (National Academy of Science and National Academy of Engineering, 1973, page 71). The spotting of clothes, deposits in pipes and plumbing fixtures, and detectable taste are common problems.

Manganese concentrations equaled or exceeded the drinking-water standard of 0.10 mg/L in water from 18 wells (21 analyses) and ranged from 0.10 to 0.55 mg/L. Four of these are the same shallow wells in the Douglas County Airport area (figures 1 and 21) where the water contained elevated dissolved-solids concentrations. The highest manganese concentration in water from these wells was 0.55 mg/L; the average was 0.26 mg/L. Four wells with water containing high manganese concentrations are in the Jacks Valley-Indian Hills and Johnson Lane areas, two are northwest of Saratoga Hot Springs, and one is between the Carson River and Johnson Lane (figure 21). Six other wells are in the southeastern part of Carson Valley, Pine Nut Mountains (figure 21), Topaz Ranch Estates, and the Topaz Lake area (figure 23). The other well with water containing a high manganese concentration is located near Muller Lane (figure 21). The high manganese concentrations in the ground water may be the result of the association with thermal water or manganese-oxide deposits (Hem, 1985, page 89).

# Temporal Changes in Ground-Water Quality

Municipal water-supply systems in Douglas County have been sampled periodically in the past, but few data were available for the Indian Hills, Topaz Ranch Estates, and other small community systems to determine long-term trends in water quality. However, dissolved-solids and nitrate data from four municipal systems--Gardnerville, Gardnerville Ranchos, Minden (figure 24) and Topaz Lodge Subdivision (figure 25) wells--were sufficient to draw graphs of temporal changes. Data points were based on only one sample from varying times of the year, so seasonal concentration fluctuations could not be assessed.

Dissolved-solids and nitrate concentrations for the Gardnerville, Gardnerville Ranchos, and Minden municipal wells (figures 26 and 27) showed no overall trend for analyses spanning the years 1969-83. Dissolved-solids and nitrate concentrations in the wells were statistically correlated with precipitation at the Douglas County Airport to determine if a relation existed. Streamflow in the East Fork Carson River and dissolved-solids and nitrate concentrations in the wells were also statistically correlated to determine if a relation existed between wet and dry years and dissolved solids and nitrate fluctuations. Correlations were generally poor and are not discussed further.

Though no overall trend was observed, the temporal changes do not appear to be due to seasonal changes, precipitation, or streamflow. The temporal changes may be the result of a combination of changes in land use, improved collection techniques, improved analytical methods, or inaccuracies in the data.

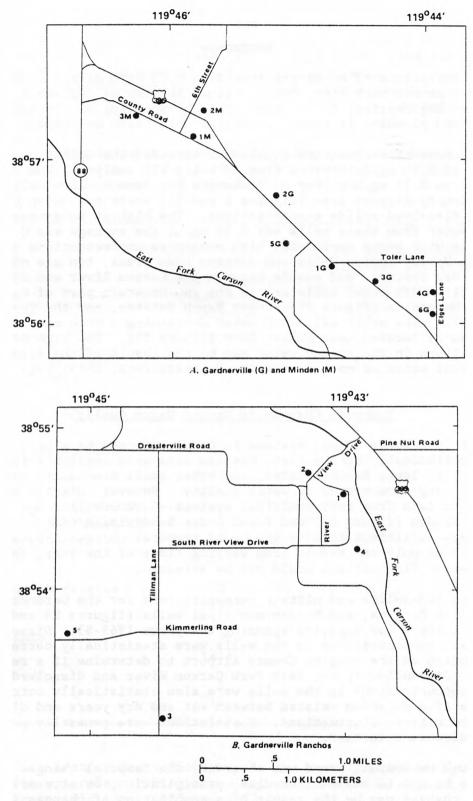


FIGURE 24.-Municipal wells in Minden, Gardnerville, and Gardnerville Ranchos. Well numbers are indicated and are referenced to figures 26 and 27.

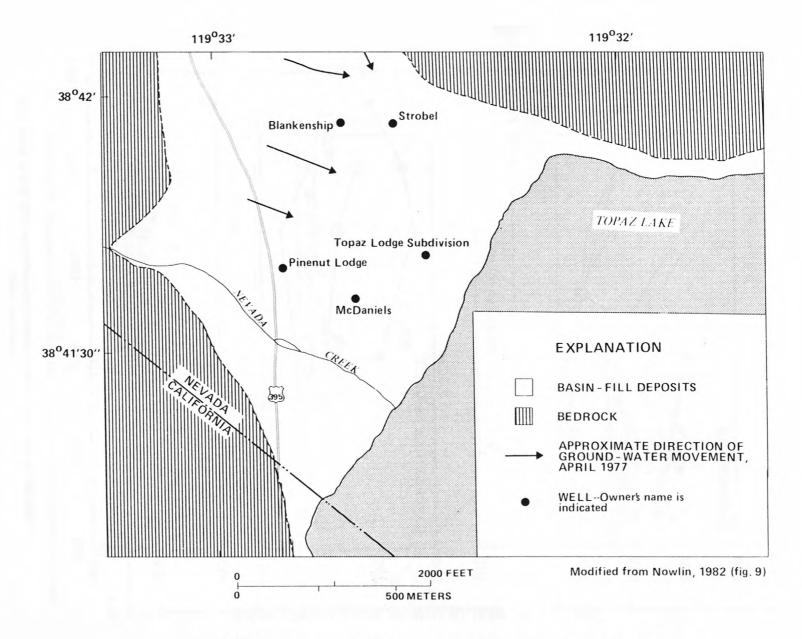


FIGURE 25.-Topaz Lodge Subdivision well and four other wells near Topaz Lake.

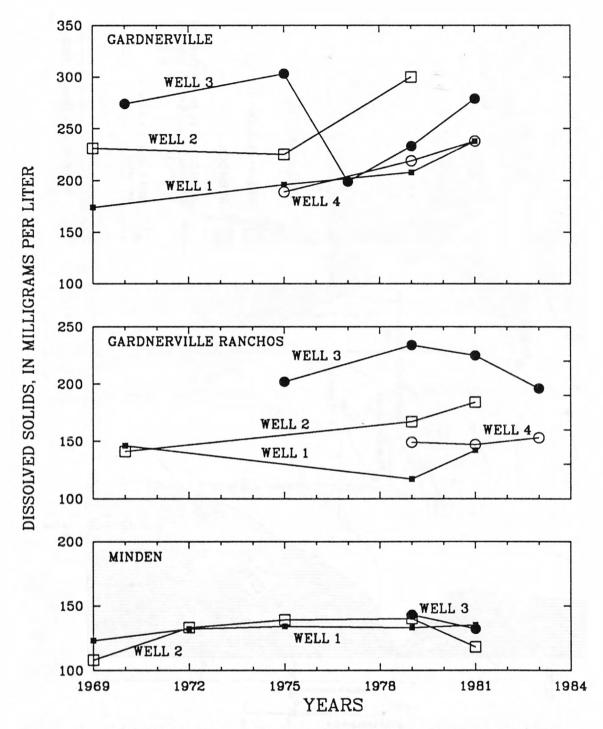


FIGURE 26.—Changes in dissolved-solids concentration with time in water from municipal wells.

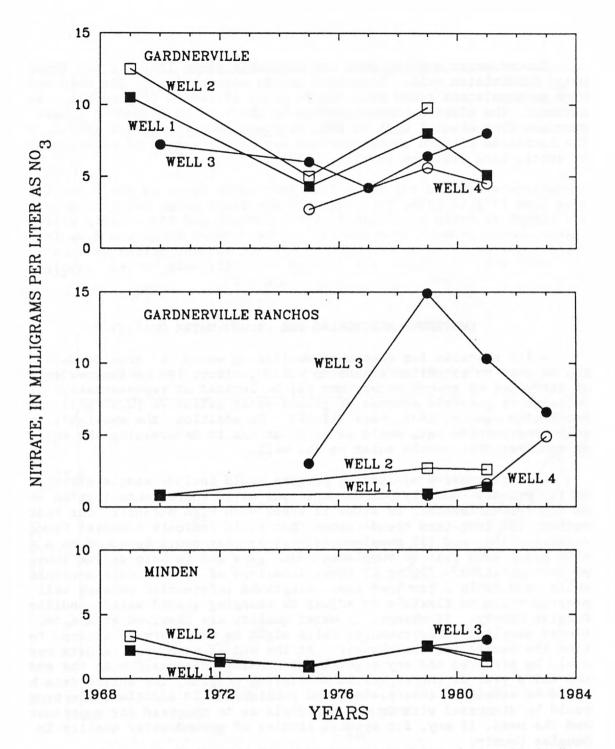


FIGURE 27.-Changes in nitrate concentration with time in water from municipal wells.

Ground-water quality data are available since 1961 for the Topaz Lodge Subdivision well. Dissolved-solids concentrations for this well show no consistent trend over the 24 years of record (figure 28). In contrast, the nitrate concentrations in the well have shown a general increase from about 2 mg/L in 1961 to about 12 mg/L in 1984 (figure 28). The increased nitrate concentrations may be the result of contamination by septic-tank effluent (Nowlin, 1982).

Nitrate samples collected from four other wells in the Topaz Lake area from 1972 to 1984, together with the Topaz Lodge Subdivision samples, are listed in table 4. Water from the Strobel and Blankenship wells show little change between 1977 and 1983. The Pinenut Lodge and McDaniels wells, however, show an increase in nitrate concentrations between 1977 and 1983. This might have been due to the difference between sample-preservation procedures used in 1977 and 1983.

# CONTINUED MONITORING FOR GROUND-WATER QUALITY

Wells suitable for continued monitoring would (1) provide data that can be used to establish trends in water quality, (2) be located in areas of heavy use of ground water, and (3) be located at representative sites relative to possible sources of ground-water pollution (U.S. Environmental Protection Agency, 1975, page III-16). In addition, the availability of well-construction data would be of great use in determining the aquifer or aquifers that supply water to the well.

A comprehensive monitoring program would include sample sites defined as (1) primary--heavily pumped municipal wells where contamination would be highly detrimental, or sites in areas with high potential for contamination; (2) long-term trend--sites that would indicate temporal trends in water-quality; and (3) supplemental -- sites that could be added on a onetime basis each year to supplement data gaps and to help define sources of contamination. Figure 29 shows locations of 31 candidate monitoring wells, and table 5 provides some background information on each well. The program would be flexible to adjust to changing ground-water conditions in Douglas County. If changes in water quality are observed at a site, additional sampling of surrounding wells might be necessary to attempt to define the causes for the changes. At the end of each year, a data report would be prepared and any significant findings discussed. At the end of the third year of the suggested monitoring program, the entire data base would be compiled, interpreted, and published. In addition, the program would be discussed with County officials as to the need for continuation and the need, if any, for special studies of ground-water quality in Douglas County.

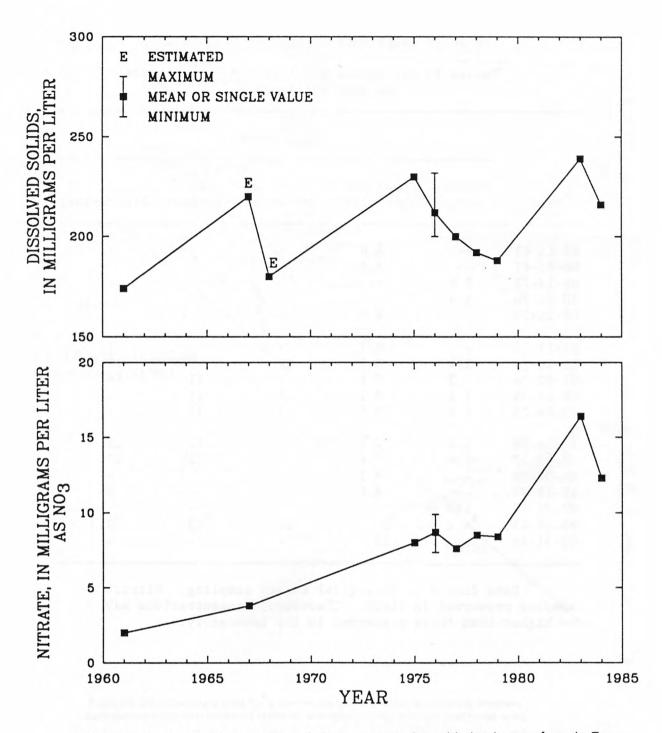


FIGURE 28.—Changes in dissolved-solids and nitrate concentrations with time in water from the Topaz Lodge Subdivision well near Topaz Lake.

TABLE 4.--Concentrations of nitrate (as NO<sub>3</sub>) in water from selected wells near Topaz Lake (figure 25)

[Values in milligrams per liter; "--" indicates no data available.]

# Well owner

	***************************************					
Date	Pinenut Lodge	Topaz Lodge Subdivision	J. McDaniels	F. Strobel	J. Blankenship	
09-11-61		2.0				
06-02-67		3.8				
06-13-72	1.8					
10-02-74	1.6					
09-26-75		8.0				
02-17-76		8.7	9.6		11	
05-20-76	1.4	9.8	11	9.5	12	
07-02-76	. 3	9.6	14	11	16	
08-16-76	1.6	8.2	16	11	14	
10-26-76	1.6	8.0	15	11	11	
12-16-76	1.8	7.7	11	11	11	
03-02-77	1.4	7.6	9.2	11	10	
06-08-78		8.5				
10-18-79		8.4				
07-26-79	1.3					
08-17-83	a4.2	<sup>a</sup> 16	<sup>a</sup> 20	<sup>a</sup> 12	<sup>a</sup> 9.7	
01-31-84		12				

a Data from U.S. Geological Survey sampling. Nitrate samples preserved in field. Therefore, concentrations may be higher than those preserved in the laboratory.

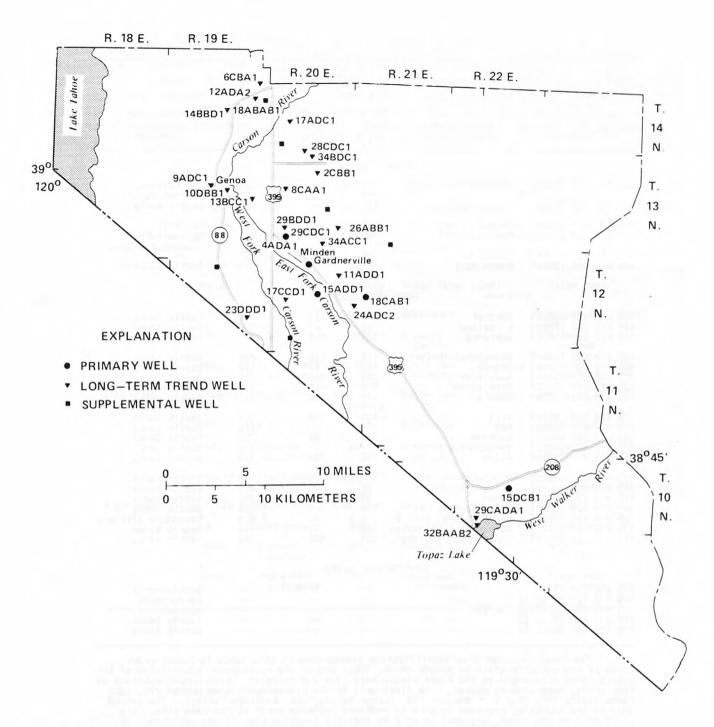


FIGURE 29.--Candidate sites for a continuing ground-water monitoring program. Abbreviated local well numbers (table 5) are indicated for primary and trend wells.

TABLE 5.--Data on candidate sites for a continuing ground-water monitoring program

["--" indicates no data available]

Site identification <sup>1</sup> (see fig. 29 for location)					land-	below surface itum	Dissolved		
		. 29		e of owner rmer owner)	Total depth	First opening	solids (milligrams per liter)	Potential contaminant sources	
					PRIM	ARY WEL	LS		
105	N12	E20	4ADA1	Gardne	rville Well 6	300	100	266	Agriculture,
105	N12	E20	15ADD1	Rancho	s Well 4	370		150	urban runoff Septic tanks
105	N12	E21	18CAB1	Dougla	s Landfill			219	Landfill
105	N13	E20	29CDC1	Minden	Well 1	398		131	Agriculture, urban runoff
			18ABAB1	Impala		425	151	166	
106	N10	E22	15DCB1	Ranch	Well			316	Septic tanks
					LONG-	TERM WE	LLS		
105	N12	E19	230001	Castee	l	141	121	74	Septic tanks
			17CCD1	Kingsl		91	67	150	Septic tanks
105	NIZ	E20	11ADD1	Curren	ce	125	105	191	Septic tanks
105	N12	E20	24ADC2		thal (McCarthy)	145	122	290	Septic tanks
	N13		10DBB1 9ADC1	Rogney		115 180	80 156	132 182	Septic tanks Septic tanks
			13BCC1	Settle		518	150	134	Secondary effluen
105	N13	E20	2CBB1	Hastie		176	156	261	Septic tanks
105	N13	E20	8CAA1	Witt		135	110	951	Septic tanks
			26ABB1	May		150	90	226	Septic tanks
			29BDD1 34ACC1	Rathbu		118	93	286 216	Septic tanks Septic tanks
			12ADA2	Plume	ditter	155	120	159	Septic tanks
105	N14	E19	14BBD1	Southw	ick	100	76	82	Septic tanks
105	N14	E20	6CBA1	Steven	S	94	70	409	Septic tanks
			28CDC1	Erven		88	68	290	Septic tanks
			34BDC1 17ADC1		Bernasconi)	100	:::	233	Septic tanks
			32BAAB2		ds area well 5 iak (McDaniels)	160		2,650 264	Secondary effluen Septic tanks
			29CADA1		enson (Strobel)	183	140	348	Septic tanks
					SUPPLEM	ENTAL S	SITES		
105	N12	F19	3 or 4						Septic tanks
105	N12	E20	29 or 32					••	Agriculture
105	N13	E20	14,15,22	, or 23	••		**	••	Agriculture
			32 or 33 29 or 32		::			::	Septic tanks
103	M14	220	27 01 32						Septic tanks

<sup>1</sup> The local (Nevada) site-identification system used in this table is based on an index of hydrographic areas in Nevada (Rush, 1968) and on the rectangular subdivision of the public lands referenced to the Mount Diablo base line and meridian. Each number consists of four units, separated by spaces. The first unit is the hydrographic-area number (90, Lake Tahoe basin; 102, Churchill Valley; 105, Carson Valley; 106, Antelope Valley). The second unit is the township, preceded by an N to indicate location north of the base line. The third unit is the range, preceded by an E to indicate location east of the meridian. The fourth unit consists of the section number and letters designating the quarter section, quarter-quarter section, and so on (A, B, C, and D indicate the northeast, northwest, southwest, and southeast quarters, respectively), followed by a number indicating the sequence in which the well was recorded. For example, well 105 N12 E20 4ADA1 is in Carson Valley. It is the first well recorded in the NE 1/4 of the SE 1/4 of the NE 1/4 of section 4, Township 12 North, Range 20 East, Mount Diablo base line and meridian.

Chemical analyses would include major dissolved constituents (calcium, magnesium, sodium, potassium, bicarbonate, carbonate, sulfate, chloride, fluoride, and silica), dissolved nutrients (organic nitrogen, ammonia, nitrite, nitrite plus nitrate, phosphorus, and orthophosphate), trace constituents (arsenic, barium, boron, cadmium, chromium, copper, iron, lead, manganese, mercury, selenium, silver, and zinc), and dissolved organic carbon. The following table shows the sites, sampling frequency, and constituents to be determined during the program.

	Firs	t year	Secon	d year	Third year		
Site	Sampling frequency	Constituents	Sampling frequency		Sampling frequency	Constituents	
Primary	Every other month	Major ions	Quarterly	Major ions	Semi- annually	Major ions	
	do.	Nutrients	do.	Nutrients	do.	Nutrients	
	do.	Organic carbon	do.	Organic carbon	do.	Organic carbon	
	do.	Trace con- stituents	Annually	Trace con- stituents	Annually	Trace con- stituents	
Long-term	Quarterly	Major ions	Semi- annually	Major ions	Annually	Major ions	
	do.	Nutrients	do.	Nutrients	do.	Nutrients	
	do.	Organic carbon	do.	Organic carbon	do.	Organic carbon	
	Annually	Trace con- stituents	Annually	Trace con- stituents	do.	Trace con- stituents	
Supplemental	Annually	Major ions	Annually	Major ions	Annually	Major ions	
	do.	Nutrients	do.	Nutrients	do.	Nutrients	
	do.	Organic carbon	do.	Organic carbon	do.	Organic carbon	
	do.	Trace con- stituents	do.	Trace con- stituents	do.	Trace con- stituents	

# SUMMARY AND CONCLUSIONS

During the last decade, Douglas County, Nev., experienced a 182-percent increase in population. Ground water is the principal source of drinking water for much of the county, especially in Carson Valley (figure 1). Carson Valley is noted for its irrigated agriculture, light industry, and housing development. In recent years, withdrawal of ground water for agricultural purposes has decreased, whereas the use of ground water for domestic purposes has increased. Rapid population growth in Douglas County has concerned local decisionmakers about present and future effects of development on ground-water quality (and quantity).

This report defines the general quality of sampled ground water, by mainly using data from 1959 to 1984; and examines water-quality trends in time and space in Douglas County. A summary of water-quality information for sampled ground water, by major populated areas, is given in table 6. The amount of data in the Lake Tahoe area and in sparsely populated outlying areas of the county is limited; therefore, these data are not included in table 6.

In general, the quality of ground water in much of the county meets drinking-water standards and criteria and is, therefore, suitable for most purposes. In Carson Valley, concentrations of most constituents generally increase in a northerly and easterly direction, corresponding to the direction of ground-water flow. Countywide, calcium and sodium are the predominant cations, and bicarbonate is the predominant anion in the sampled ground water. The ionic type of ground water most common throughout the study area are calcium-bicarbonate, sodium-bicarbonate, and calcium-sodium or sodium-calcium bicarbonate.

A statistical evaluation of constituents in ground water indicates that sodium, arsenic, iron, manganese, and zinc each have a single statistical "population," whereas dissolved solids, calcium, magnesium, bicarbonate, sulfate, chloride, fluoride, and nitrate each have two discrete "populations." Because of known hydrogeologic conditions in certain areas of Douglas County, the two data distributions may be caused by: Water flowing from metasedimentary rocks into alluvial-fan deposits in the Topaz Lake area; water above the clay layers underlying the Douglas County Airport; and geothermal water in the Jacks Valley-Indian Hills and Saratoga Hot Springs areas.

The concentrations of most chemical constituents in ground water in Douglas County meet drinking-water standards. Water samples that exceeded drinking-water standards for fluoride, nitrate, arsenic, iron, and manganese were found in the Jacks Valley-Indian Hills area; for sulfate, fluoride, dissolved solids, nitrate, arsenic, iron, and manganese in the Johnson Lane area; for iron in wells in the Genoa, Minden-Gardnerville, and Gardnerville Ranchos areas; for manganese in shallow wells near the Douglas County Airport; for nitrate, iron, and manganese in wells in the Ruhenstroth area; and for nitrate, arsenic, iron, and manganese in the Topaz Lake area (figures 21, 23, and table 7).

TABLE 6.--Summary of water-quality information for sampled ground water, by major populated area [N, number of sites; Max., maximum; Min., minimum; E, estimated]

		Site depth <sup>1</sup>			0	issolved	solids	2		
Area <sup>3</sup>	N	N Median		Min.	N	Median	Max.	Min.	Constituents exceeding drinking-water criteria 4	Potential sources of contamination
Entire county	264	145	580	0	327	228	1,170	50	See below	See below
Jacks Valley- Indian Hills	22	134	350	94	23	182	409	97	Arsenic 1/23 (0.05) Fluoride 3/23 (2.0) Iron 3/23 (1.03) Manganese 2/21 (0.20) Nitrate 2/22 (69.5)	Septic tanks
Johnson Lane	45	135	420	60	64	272	1,170	204	Arsenic 2/64 (0.065) Dissolved solids 3/64 (1,170) Fluoride 33/65 (5.3) Iron 3/64 (2.01) Manganese 2/62 (0.39) Nitrate 1/65 (70) Sulfate 5/65 (725)	Septic tanks
Saratoga Hot Springs	1	0			1	1,100E			Dissolved solids Sulfate (678)	Geothermal water
Genoa	9	159	225	83	14	145	182	91	Iron 4/14 (1.13)	Septic tanks
Douglas County Airport	8	298	440	100	9	186	951	160	Manganese 4/9 (0.55)	Agriculture
Minden- Gardnerville	13	300	398	90	14	234	344	130	Iron 3/14 (1.76)	Agriculture, urban runoff
Ruhenstroth	13	169	480	100	15	285	519	238	Iron 1/15 (2.50) Manganese 1/15 (0.80) Nitrate 1/15 (84)	Septic tanks
Gardnerville Ranchos	11	200	450	77	11	162	382	115	Iron 1/11 (2.45)	Urban runoff
Topaz Lake	24	165	318	0	30	234	536	154	Arsenic 8/30 (0.38) Iron 7/30 (3.42) Manganese 2/29 (0.56) Nitrate 1/30 (48)	Septic tanks, urban runoff

Depth, in feet below land surface.
Concentration, in milligrams per liter.
See figure 1 for location.

Ground water in the west-central, northern, and northeastern part of Carson Valley is influenced by mixing with geothermal water.

Dissolved-solids and nitrate concentrations in water from the Gardnerville, Gardnerville Ranchos, and Minden municipal wells indicated no overall trend for the years from 1969 to 1983. Dissolved-solids concentrations in water from the Topaz Lodge Subdivision well indicated no consistent trend from 1961 to 1984. In contrast, nitrate concentrations in water from the Topaz Lodge Subdivision well indicated a general increase from 1961 to 1984.

<sup>4</sup> For each constituent: number to left of slash indicates number of sites exceeding criteria; number to right of slash indicates total number of sites; and number in parentheses is maximum determined concentration, in milligrams per liter.

These conclusions are based primarily on water-quality data from the historical data base, which consists largely of analyses for major dissolved constituents and a limited number of trace elements. Because of the paucity of available data concerning manmade organic compounds and indicator bacteria, an evaluation of these variables in ground water of Douglas County was beyond the scope of this study.

A comprehensive program for future monitoring of ground-water quality in Douglas County would need to include periodic sampling of primary (heavily pumped or high potential for contamination) wells, long-term trend (possible changes with time) sites, and supplemental (to supplement data gaps) sites. The frequency of sample collection could be quarterly or more frequent for the heavily used wells and sites used for determining long-term water-quality trends during the first year, and probably annually for the supplemental wells. Frequency of sample collection after the first year might be changed for the heavily pumped wells and sites with possible changes in time to meet changing needs for problem identification. Supplemental wells would be selected and sampled as considered necessary.

## BASIC DATA ON GROUND-WATER QUALITY

The following tabulation includes analytical results for 546 samples from 333 ground-water sites (wells and springs). The sites are listed in order of increasing hydrographic-area, township, range, and section numbers.

Each sample has been assigned a sequential reference number (from 1 through 546). The site information and analytical results for each sample are listed herein, by reference number, in four groups of data (group 1, pages 65-74; group 2, pages 75-84; group 3, pages 85-94; and group 4, pages 95-104). For example, data for sample number 35 appear in four places: on pages 65, 75, 85, and 95.

<u>Site Identification</u>: Indicates section number, location within the section, and sequence number. <u>See table 5, footnote 1 (p. 58), for description of complete site-numbering system.</u>

Total depth of well: Sources of information: L, driller's log (on file at Nevada Division of Water Resources, Carson City); M, measured in field; R, reported. Reported depths should be considered approximate due to inconsistencies between using land surface or top of casing as datum.

Analyst: D, Douglas County Sewer Improvement District No. 1; E, Environmental Science Laboratory;
N, Nevada State Health Laboratory; U, U.S. Geological Survey.

Solids, residue: "c" indicates computed dissolved-solids sum with bicarbonate multiplied by 0.492 to make result comparable with residue value.

Abbreviations: E, estimated; MG/L, milligrams per liter; NTU, nephelometric turbidity units.

	SITE			TOTAL DEPTH OF	DATE	SAMP	LED	
REF.	IDENTI- FICATION		LOCATION	WELL (FEET)	MONTH	DAY	YEAR	ANA- LYST
		Lake Ta	ahoe basin (90), T. 13 N., R. 18	<u>E.</u>				
1 2 3	3CDD1 3CDD2	DOUGLAS COUNTY FIRE DOUGLAS COUNTY SCHOOL	ZEPHYR COVE GEORGE WHITTELL HIGH SCHOOL	95L 264L	7 7 1	11 11 2	1963 1963 1973	N N N
		Lake Ta	ahoe basin (90), T. 14 N., R. 18	E				
4	3D	GLENBROOK GOLF COURSE	GLENBROOK	OM	6	20	1961	N
		Churchill	Valley basin (102), T. 14 N., R.	22 E.				
5	13D1 13D2	SUNRISE CAMP SPRING SUNRISE CAMP SPRING	N14E2213D N14E2213D	OM OM	9	17 17	1979 1979	N N
		Carson Va	alley basin (105), T. 11 N., R. 2	20 E.				
7	7ADC1	NEDDENRIEP	0.9MI S STATELINE .3MI E 395	••	8	24	1983	U
		Carson Va	alley basin (105), T. 11 N., R. 2	21 E.				
8 9 10 11 12	9ABD1 9A1 9BAA1 20CD1 22DDB1	B. ZERBY VETA GRANDE MINE ELDON HANNEMAN DOUD SPRING ALFRED KNOOP	CARTERS STATION N11E219A CARTERS STATION DOUBLE SPRING FLAT AREA DOUBLE FLAT	140R 175R 0R 180R	6 5 7 5 8	19 22 16 7 20	1981 1975 1978 1970 1973	N N N N
13 14 15 16	26BA1 26BCA1 35ADD1		DOUBLE SPRING FLAT AREA DOUBLE SPRING FLAT AREA DEAD HORSE FLAT AREA	0R 145L 175L	5 11 3	6 24 4	1977 1970 1980 1981	N U N N
		Carson V	alley basin (105), T. 11 N., R. 2	22 E.				
17 18	9B1 9B2	MILL CANYON SPRING N MILL CANYON SPRING N	N11E229B N11E229B	OR OR	9	17 17	1979 1979	N N
		Carson V	alley basin (105), T. 12 N., R.	19 E.				
19 20 21 22 23	2BDD1 3BBA1 3BBA3 3BBC1 3BBC2	FEIL ARTESIAN PAUL BRITTENDAHL GENE WALTERS JOE VARGA M.BRODIE	ON MOTTSVILLE LN 2MI W 395 228 BEVERLY 232 BEVERLY 274 BEVERLY FOOTHILL	262R   94R	8 10 3 9 2	18 28 22 8 6	1983 1980 1982 1980 1980	U N N
24 25 26 27 28	3BCB1 3CDC1 11CDC1 12CCC1 13BAA1	H.M. EVANS CRAIG LODATO BROCKLISS ARTESIAN SAYED BIEGELOW ARTESIAN	MOTT COURT 1207 FOOTHILL ROAD ON CNTRVILLE LN 2.2MI W 395 1008 MARSHA LANE ON CNTRVILLE LN 1.1MI W 395	115R 90R 60R 117R 222R	10 9 8 5 8	19 10 18  18	1978 1978 1983 1977 1983	N N U N U
29 30 31 32 33	14CCA1 15AAB1 15BBB1 23BAA1 23BDA2	TED WHITE KENNETH PINNON TOMERLIN INDIAN ROAD RANCH NORMAN MCMURRAY	797 FOOTHILL 986 BOLLEN CIRCLE FOOTHILL FOOTHILL ROAD 741 INDIAN TRAIL	300R 176R  82R	10 11 9 10 4	28  3 25	1980 1977 1976 1974 1978	N N N N
34 35 36 37 38 39	23DDD1 25CDD1 26AAB1 26ABB2 26ACC1 26BAD1	GARY CASTEEL HERITAGE RANCH AL WASKOVIAK PEDRO VILLALOBOS DAVID HELLWINKEL CASINELLA	581 GREEN ACRES 0.1MI N FAIRVIEW LN 1MI W 395 567 GREEN ACRES 507 GREEN ACRES 500 FOOTHILL SE CORNER GREEN ACRES FOOTHILL	141L 100R 122L	11 8 4 10 12 9	3 24  9 16 8	1980 1983 1976 1979 1981 1980	N U N N N

TABLE 7.--Water-quality data--Continued

				DEPTH OF WELL,	SAMP	LE DA	TE	ANA- LYST
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	
		Carson Va	lley basin (105), T. 12 N., R.	20 E.				
40 41 42 43 44	1DCC1 1DCD1 4AAD1 4AAD2	NANCIE COZZA SKINNER A. CECIL STODIEK GARDNERVILLE TOWN 4	1228 FISH SPRINGS ROAD SHEEP CAMP FISH SPRGS ROAD ELGES LANE 0.2MI S OF TOLLER ON ELGES	192R 96R 305L	3 6 6 6	17 13 24 14	1982 1976 1972 1975 1979	2 2 2
45 46 47 48 49	4ABC1	GARDNERVILLE TOWN 3	NEAR HIGHWAY MAINT STATION	335L	4 5 6 4 6	14 11 24 26 14	1981 1970 1975 1977 1979	N N N N
50 51 52 53 54	4ACB1 4ADA1 4BAB1	DOUGLAS COUNTY GARDNERVILLE TOWN 6 GARNDERVILLE TOWN 1	LAMPE PARK 0.2MI N 395 ON ELGES AT WATER CO YARD	270L 300L	46226	14 29 5 3 24	1981 1979 1984 1969 1975	N N N N
55 56 57 58 59	58801 7CAA1 7DCC1	STODIECK KEITH CORNFORTH CENTERVILLE BAR	0.3MI N WATERLOO 0.7MI E 395 HIGHWAY 88 CENTERVILLE ROAD & 395	79L	6 4 8 11 7	14 14 30 12 6	1979 1981 1983 1967 1978	X U X X
60 61 62 63 64	8CDC1 8DAD1 8DDC1 9BAC1	ELMER HELLWINKEL JAMES LAWRENCE GARY STONE GORDON FRICKE	CENTERVILLE ROAD ROCK CREEK RANCH CENTERVILLE ROAD CENTERVILLE ROAD	70L 130R	7 4 5 1 4	8 10 19 28	1980 1975 1969 1972 1975	N N N N
65 66 67 68 69	9BBA1 9CAD1 10CCB1 10CCC1 10DAD1	LEONARD WEISSMAN NINA FOSTER ROBINSON GARY GARCIA 7-11 MARKET	1186 CENTERVILLE 1188 DRESSLERVILLE 1029 DRESSLERVILLE ROAD 1025 DRESSLERVILLE ROAD SOUTH 395	110R 140R 200L	8 12 3 5 9	18 29 14 3 9	1982 1980 1978 1977 1982	N N N N
70 71 72 73 74	10DCC1 11ADC1 11ADD1	GARD RANCHOS NO 2  ROBERT LANGE TOM CURRENCE	OFF RIVERVIEW NEAR FAIRWAY  1122 LINDA ANNE 1119 LINDA ANNE	445L 121R 125L	46595	29 25 12 16 1	1970 1979 1981 1975 1981	N N N N N N N N N N N N N N N N N N N
75 76 77 78 79	11CAA1 12CAA1 12CAC1 12CAC1 12CAD1	PAUL FAULSTICH D. KENT BYERS DAVE DURYEE ROBERT OSBORNE ROSEMARY NELSON	S.E. OF GARDNERVILLE HELLMAN DRIVE 1069 SEGO CIRCLE 1056 JEWEL CIRCLE HELLMAN DRIVE	343L 180R 158L 200L	8 2 8 7 3	14 23 28 12	1962 1971 1967 1974 1979	N N N N N N N N N N N N N N N N N N N
80 81 82 83	12CBA1 14ADDC1 14BDA1	ROBERT FERRIS U.S. BUREAU SPORT FISH WASHOE TRIBAL OFFICE		150R 400L 61R	6 8 1 9	11 13 4 4	1975 1962 1965 1979	N N U N
84	15AAB1	GARD RANCHOS NO 1	ON FAIRWAY DRIVE	450L	4	28	1970	N
85 86 87 88 89	15ABB1 15ABD2 15ABD1	JOHN WINTERS DON WOODWARD HOOVER	965 DRESSLERVILLE ROAD 961 RIVERVIEW 961 RIVERVIEW	230R 77R 130R	6 5 6 10 2	25 12 25 20	1979 1981 1979 1976 1973	N N N N
90 91 92 93 94	15ADD1 16CCD1 16DCA1	GARD RANCHOS NO 4 BING GRAVEL PIT DAVID ANDERSON	BTWN GREEN CT/PUTTER DRIVE  PIT ROAD 1325 MUIR DRIVE	370L 385R	9 5 2 10 9	5 12 16 17 11	1979 1981 1983 1972 1979	N N N N

				DEPTH OF WELL,	SAMP	LE DA	TE	
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	LYS
		Carson Va	alley basin (105), T. 12 N., R. 2	20 E.				
95 96 97 98 99	17BAA1 17CBD1 17CCD1 17CDB1 17CDC1	GUS HERZOG J. HELLWINKEL BRUCE KINGSLAND L.E. BANKUS J. BAILEY	1089 CENTERVILLE VERDE WAY KINGSLAND COURT 816 ROJO 807 ROJO	81L 82R 91L 84L 105R	8 10 3 8 10	29 18 3 18 17	1979 1972 1978 1969 1972	N N N N
100 101 102 103 104	17CDC2 17CDD1 17DBB1 19BBB1 21DBC1	C. WASS JERRY CLARK ALEX MALAVAZOS SCHWAKE GARD RANCHOS NO 3	1071 ORO 1091 ORO WAY 1125 AMARILLO 1.1MI S CNTRVILLE .5MI W 395 ON TILLMAN 0.6MI S KIMMERLING	105R 90L 104R  460L	10 11 8 8 6	17 2 16 22 20	1972 1977 1982 1983 1975	N N N U N
105 106 107 108 109	23DCA1 23ACA1	FISH HATCHERY SETTLEMEYER	SOUTH 395 SOUTH 395	240R	6 5 2 6 5	25 12 16 5 9	1979 1981 1983 1972 1974	2 2 2 2 2 2 2
110 111 112 113	23DAD1 23DAD1 24AAC1	BETTY ADAMS GARY WILLIAMS RICHARD SANDBORN	660 STONESTHROW ABOVE FISH HATCHERY 754 MUSTANG	172L 182L	3 11 4	7 2 20 14	1978 1976 1979 1982	N N N
114	24AAD1	WALT LEE	1976 SORREL	200R	7		1978	N
115 116 117 118	24ACA1 24ADC1 24ADC2	LANE HOLDEMAN PHIL SULLIVAN ANDREW MCCARTHY	1933 SORREL 716 MUSTANG LANE ARABIAN	150R 190L 145L	10 11 1 8	9 20 29	1979 1976 1977 1978	N N N
119	24ADC2	HAWKS	MUSTANG LANE	180R	11		1976	N
120 121 122 123 124	24BCA1 24DAA1 24DAD1 24DBD1 24DCA1	BECKER F.J. MCCANN STEVEN BOHLER JOHN GRANT NORMAN SCOTT	1830 COLT LANE 1965 ARABIAN 1986 PALOMINO 1940 PALOMINO 629 APPALOOSA	100R 170R 480R 220R 168R	1 6 10 11 4	20 7 27 19	1977 1979 1977 1974 1976	N N N N
125 126 127 128 129	24DCA2 24DCB1	HILL RAY WILLIAMS	1902 PALOMINO 628 APPALOOSA	150R 150R	1 12 12 12 5	20 27 16 27 12	1977 1976 1976 1976 1977	N N N
130 131 132 133	24DCB2 24DDD1 25BDB1	MORRIS RANSTROM DAN SHIPMAN WASHOE TRIBE CAMPGRO	SW CORNER APPALOOSA PALOMINO 619 STAGECOACH SOUTH OF GARDNERVILLE	140R 60L	8 8 8 9	8 23 8 2	1978 1978 1978 1975	N N N
		Carson V	alley basin (105), T. 12 N., R.	21 E.				
134 135	4DBA1 5CAC1	MARSHALL NORRIS	1263 MARJ LANE 1250 RON LANE	71L	1 2 9	18 10 15	1979 1976 1980	N
136 137 138	5CAC2 5CAC3	ELDON MILLER R. HINTZMAN	1261 RON LANE 1260 RON LANE	74L	8 2	12	1980 1976	N
139 140 141	5DBA1 5DBB1 5DBC1	BOBEK VEHLING CAROL KRULL KARL AMDAL	1261 MYERS 1277 MYERS 1256 MYERS	80L 169R	2 6 5	10 12 8 1	1976 1976 1978 1979	N N N
142 143	5DBD3	RON FINCH	2331 JAN COURT OR 1250 MYERS	62L	2	10	1976	N
144 145 146 147	17CCA1 18CAB1	FRED SMITH DOUGLAS COUNTY ANIMAL	O.5 MI S PINENUT ROAD LANDFILL	435L	8 2 4 2 3	31 27 13	1982 1980 1981 1983	N
147 148	180001	DOUGLAS COUNTY PARKS	E END PINTO CIRCLE	380L		19	1979	N
149 150 151	32CBA1 35BAA1	WASHOE INDIAN TRIBAL RICHARD MORGAN	BODIE FLAT CALLE PEQUENO	580L 100R	1 2 5	10 15	1981 1981 1979	N N

TABLE 7.--Water-quality data--Continued

				DEPTH OF WELL,	SAMP	LE DA	TE	
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	ANA- LYST
		Carson V	alley basin (105), T. 12 N., R.	22 E.				
152 153 154 155	14CC1 14CC2 26 26	SLATERS MINE SPRING SLATERS MINE SPRING WINTERS MINE SPRING WINTERS MINE SPRING	UNKNOWN UNKNOWN UNKNOWN UNKNOWN	OM OM OM	6 9 9	17 17 17	1979 1979 1979 1979	N N N
		Carson V	alley basin (105), T. 13 N., R.	19 E.				
156 157 158 159 160	3BDA1 3CCB1 3CCC1 4DDA1 9AAA1	HOMER ANGELO S. HOLTEN CHARLES CROWL D.J. HENRY ANTHONY STEPHENSON	JACKS VALLEY ROAD HOLTEN CIRCLE TRAIL COURT CENTENNIAL DRIVE 2398 CENTENNIAL DRIVE	128L 210R 180R 231L	5 6 2 5 9	9 11 9	1974 1979 1979 1974 1981	N N N N
161 162	9AAC1	ALDUS BAGNE	2330 MARJORY LANE	225L	4	23 13	1974 1980	N
163 164 165	9ACD1 9ACD2 9ADC1	BOB BRUSH M. MILUCK STANLEY HOLLISTER	TOBOGGAN ROAD END OF 5TH GENOA NW CORNER GENOA FIFTH	208R 225L 180L	5 7 7 3	5 28 12	1976 1976 1969	N N N
166 167	9DAA1	W. MERRELL	PINK HOUSE		7	6	1978 1979	N
168 169 170	9DAA2 9DAA3	A. JUCHTZER NEVADA STATE PARK	GENOA MORMON STATION	83L 159L	7 6 6	8 2 14	1980 1976 1979	N N
171 172 173 174 175	9DAB1 9DABD1 10ACC1 10BBA1 10BBD1	ROBERT CARVER R. BOMMERITO ROBERT STEEN L.E. GOODNIGHT BRENDAN RILEY	GENOA SALOON GENOA 2279 MEADOWLARK LANE 2399 JACKS VALLEY ROAD 2290 PIONEER TRAILS	100R	7 6 8 6 8	6 2 21 28 8	1978 1976 1979 1982 1975	N N N N
176 177 178 179 180	10BDD1 11ACC1	JOHN LERETTE SETTLEMEYER	264 SCHOOLHOUSE DRIVE STP EFFLUENT PROJECT	90R 	3 1 3 4	29 24 10 22	1977 1982 1982 1982 1982	N D D
181 182 183 184 185	11CCA1 12BBA1 13BCC1 14AAA1	STONE ELEMORE SETTLEMEYER ARTESIAN SETTLEMEYER ARTESIAN SETTLEMEYER	WILLOW BEND 1MI N GENOA LN 1.3MI W 395 0.5MI S GENOA LN 1.4MI W 395 STP EFFLUENT PROJECT	110R  518L	5 9 8 8	5 17 18 25 29	1982 1976 1983 1983 1982	מכנגם
186 187 188 189 190	17	FOREST SERVICE	SPRING IN SECTION 17	0м	34459	24 10 22 5 26	1982 1982 1982 1982 1973	0000
191 192 193 194 195	20ABA1 21 22ABC1 24CAD1 27BAC1	SPRING FOREST SERVICE WALLEYS HOT SPRINGS DANGBERG ARTESIAN MIKE SARGENT	SPRING IN SECTION 20 SPRING IN SECTION 21 FOOTHILL ROAD 1 MI W 395 0.1 MI N MULLER FOOTHILL ROAD	0M 0M 0M 401R 80L	7 8 11 8 5	7 30 10 18 17	1971 1973 1959 1983 1973	N N U U
196 197 198	34BCA1 35CBB1	GEOFF BLAKESLEE KNOX JOHNSON	1483 FOOTHILL 0.3MI N WATERLOO 2.3MI W 395	::	11 9 9	27	1973 1982 1983	N N U

				DEPTH OF WELL,	SAMPLE DATE		TE	
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	ANA-
		Carson Va	lley basin (105), T. 13 N., R.	20 E.				
199 200 201 202	2ACB1 2BAA1 2BAB1	W. BOLT AUGUST BELLAS JOHN SOUTHERN	2555 LENA 2578 MCKAY 2589 FREMONT	263R 310L 223R	6 9 9 8	1 18 18 28	1982 1979 1979 1978	N N N
203	201101				9	22	1978	N
204 205 206 207 208	2BAC1 2BAD1 2CBA1 2CBA2 2CBB1	SAUNDERS	2554 NYE DRIVE 2571 NYE DRIVE 1639 NANSUE LANE 1635 NANSUE LANE 1631 NANSUE LANE	230L 210R 172L 176R	7 7 9 6 10	16 28 7 8 3	1979 1980 1977 1978 1978	N N N N
209 210 211 212 213	2CDA1 3BCB1 4ADA1 5AAA1 5CBA1	FRED STANFIELD UNKNOWN EARL P. IFLAND L. A. ENGINEERING JOHN INDIANO	1708 DAVIS LANE 2533 CLAPHAM JOHNSON AND CLAPHAM 75 FT S JOHNSON LANE OFF HEYBOURNE	200R 132L 300L 100L	3 6 9 2 11	10 4 27 26	1980 1979 1971 1981 1978	N N N N N N N N N N N N N N N N N N N
214 215	7ACD1	ALDAX	ALDAX RANCH	400L	12	15 15	1969 1970	N
216 217	8ABB1	VALMAN CORP	150 FT E HEYBOURNE	297L	1	6	1981 1981	N N
218	8CAA1	HERB WITT	MILKY WAY FARMS	135L	12	10	1975	N
219 220 221	8CAD1 8CDA1 8CDA2	FXO SYSTEMS	MILKY WAY FARMS 1089 AIRPORT ROAD 1051 AIRPORT ROAD	440L 115L	6 3 1	25 8 31 30	1976 1982 1973 1976	N N N
222 223	8DCA1	DOUGLAS COUNTY AIRPORT	AIRPORT OLD WELL		5	18	1973	N
224 225 226 227 228	8DCA2 11ABD1 11BAB1 12BCA1	DOUGLAS COUNTY AIRPORT CECIL REED HARRY POWERS COULTER	AIRPORT NEW WELL REED CIRCLE 2433 FREMONT 0.6 MI S SUNRISE ROAD	330L 202L  300L	6 6 12 8	6 6 1 29 19	1980 1980 1974 1981 1983	N N N N
229 230 231 232 233	18BAA1 19AAA1 19CBA1 19CCB1 26ABB1	SETTLEMEYER ARTESIAN DANGBERG ARTESIAN DOUGLAS STP NORTH WE DOUGLAS STP WEST WEL EARL MAY	GENOA LANE AND 395 0.8MI N MULLER 0.45MI E 395 0.3 MI W 395 0.35 MI N MULLER 0.5 MI W 395 0.1 MI N MULLER EAST VALLEY ROAD	318R 10M 9M 150L	8 8 8 8	22 18 31 31 19	1983 1983 1983 1983 1983	UUUU
234 235 236 237 238	29BCD1 29CDC1	DON CALLAHAN	FRIEDA LANE WELL 1 NEAR BENTLEY	908	2 2 12 6 1	27 3 22 24 21	1970 1969 1972 1975 1981	N N N N
239 240 241 242 243	30DCA1 30DDC1 31BCC1 31CAA1 32BAC1	ALDO BIAGGI TOWN OF MINDEN DANGBERG ORIG HOME DANGBERG TOWN OF MINDEN	COUNTY ROAD WELL 3 1648 COUNTY ROAD 0.5MI W 395 1MI N WATERLOO N13E2031CAA WELL 2 BETWEEN 4TH AND 5TH	108R 310L  413L 300L	1 6 8 12 2	2 25 22 	1980 1979 1983 1969 1969	Z Z C Z Z
244 245 246 247 248	32DAB1	GARDNERVILLE TOWN 2	OXOBY TRAILER COURT	290L	12 6 6 2 6	22 24 25 3 24	1972 1975 1979 1969 1975	N N N N
249 250 251 252	32DDA1	GARDNERVILLE TOWN 5	ON HUSSMAN AVE	310L	6 8 2 4	14 31 19 14	1979 1978 1980 1981	N N N
252 253	33CBA1	CARSON VALLEY MEAT	GARDNERVILLE	137L	5	23	1968	N
254 255 256	34ACC1	SPOONHUNTER	ORCHARD ROAD		6 8 8	2 18 22	1969 1969 1983	N N U

				DEPTH OF WELL,	SAMP	SAMPLE DATE		
REF.	SECTION	NAME		TOTAL (FEET)	MONTH	DAY	YEAR	ANA
			alley basin (105), T. 13 N., R. 21					
257 258 259	28CCB1 29DBC1	FISH SPRING FLAT WELL ROBERT CARLSON	FISH SPRING FLAT 1651 CARLSON 1650 CARLSON FISH SPRING FLAT	95R 102L	5 3 7	14 20 24	1970 1979 1980	U N N
260	290BC2 33CDD1	JOHN SHERWOOD FINGAR	1650 CARLSON FISH SPRING FLAT	140L 132L	9	23	1982 1983	N U
		Carson Va	alley basin (105), T. 14 N., R. 19	E.				
262 263 264	1ADC1 1DAB1 1DAD1	AMES ROBINSON JERRY GAGE DONALD SWEZEY	3677 CINDY LANE 3669 BUCKS WAY 3666 CHEROKEE	1231	6	8	1976 1976 1976	N N
265 266	1DDD1	L.G. DUFFRIN	3615 CINDYS TRAIL	118L	7	13	1978 1977	N
267	3BBD1	JIM HALEN	JUUU ULD CLEAR CREEK KUAD	94L	0	29	1973	N
268 269 270 271	4ABA1 11CAA1 11CAA2	VICTORIA LINDERHOLM RHONDA KILTY CHARLENE ROSS	3000 OLD CLEAR CREEK ROAD OLD CLEAR CREEK ROAD 3488 ZURICH 3489 ZURICH COURT	165L 175R	5 3 9	27 28 2	1975 1974 1978 1982	N N N
272 273 274 275 276	11CAD1 11CBD1 12AAA1 12ADA1 12ADA2	H.L.ROSSE J.R. STEVENS K. PORTH ROBERT LARSON ROBERT FOSTER	478 BAVARIAN 3459 ALPINE VIEW 3577 CHEROKEE 3526 CHEROKEE 3530 CHEROKEE	165L 80R 120R 146L 100R	6 2 6 4 9	8 22 14 2 17	1975 1979 1972 1976 1976	N N N N
277 278 279 280 281	12ADA3 12ADB1 12ADC1 12ADD1 12BDA1	ROBERT PENNINGTON BERNING HARVEY MILLARD RAY BACON DOUGLAS COUNTY SCHOOL	478 BAVARIAN 3459 ALPINE VIEW 3577 CHEROKEE 3526 CHEROKEE 3530 CHEROKEE 3523 CHEROKEE 3543 ARCADIA COURT 3522 ARCADIA 780 PAWNEE JACKS VALLEY ELEM SCHOOL	142L 101R 103R 109R 350L	7 3 8 5	14 17 6 22 20	1981 1980 1973 1977 1980	N N N N
282 283 284 285 286	14BAA1 14BAB1 15CDB1 26CAD1 26CCA1	ANTHONY E. BOWERS KIRBY CLENDENON ASCUAGA HARVEY GROSS MEADOW GOLD	500 ALPINE VIEW COURT 3396 BERNICE COURT 0.5 MI SE OF RANCH HARVEYS GUN CLUB JAMES CANYON RANCH	210R 85L  385R	11 12 8 10 6	28 23 18 16	1977 1979 1983 1979 1975	N N U N
287 288 289	26CCA1 26CCB1 35CBB1	MEADOW GOLD HARVEY GROSS TYSON ARTESIAN	JAMES CANYON RANCH JAMES CANYON RANCH 2.2MI N GENOA 0.4MI E FOOTHILL	••	11 8 8	7 23 18	1961 1983 1983	N U U
		Carson V	alley basin (105), T. 14 N., R. 20	0 E.				
290 291 292	6CAA1 6CAB1	MORRIS ROBERT HOOD	3695 SUMMERHILL 3686 SILVERADO	175R 120R	12 1 8	13 21 26	1979 1981 1982	N N
293 294	6CBA1 6CBA2	STUART STEVENS ELAINE CUNNINGHAM	3687 SHAWNEE 3689 SHAWNEE	94L 107R	3	5	1979 1979	N
295 296 297 298 299	6CDC1 7BBB1 7BCB1 7BCD1 7BDD1	TAYLOR GOATS KEITH NEUMAN NEIL LEONARD RICH MCGILL OPALITE WELL	3607 SILVERADO 3592 GREEN ACRES 3519 GREEN ACRES 3514 SHAWNEE INDIAN HILLS	154L 150R 165L 252L	1 2 5 5 12	18 11 14 15 13	1979 1980 1982 1976 1977	N N N N
300 301 302 303 304	7CBC1 7DBA1 18CDA1 19BAD1	LITTLE CANYON WELL R. COX HOBO HOT SPRINGS WELL LUTTRELL	INDIAN HILLS 3480 VISTA GRANDE INDIAN HILLS 1.5MI N CARSONR 0.05MI W 395	345L 340L 160R	12 6 4 11 8	13 22 25 26 24	1977 1980 1979 1977 1983	N N N N
305	20AAB1	WETLANDS WELL	1.25MI E395, 3.5MI N STEPH. WAY	25M	8	29	1979 1983	E
306 307 308	20DAA1	WETLANDS WARM WELL	1.5MI E395, 3MI N STEPH. WAY	27M	8	26	1979	UEU
309	21CDD1	SARATOGA HOT SPRINGS	N END OF VICKY LANE	OM	5	14	1970	Ü

				DEPTH OF WELL,	SAMPLE DATE			
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	ANA- LYST
		Carson Va	Hay begin (10E) T 1/ N D	20 E.				
310 311 312 313 314	26CCB1 26CDC1 26CDD1 27DCB1 27DCD1	GLORIA LOHNER WILLIAM BARDO SANDY MCFALL BOB LAWRENCE	1650 STEPHANIE 1680 STEPHANIE WAY 2844 WADE 2832 SQUIRES	245R 262R  165R	9 6 6 3 7	21 16 28 24 8	1980 1980 1978 1980 1978	N N N N
315 316 317 318	27DCD1 28ACD1 28ADC1	JOHN SCHWEITZER TERRY BROWN JIM LIEBHEN	2820 SQUIRES 101 PORTER 1360 PORTER	190R 125R 90L	12 11 9	8	1981 1981 1978 1981	N N N
319	28CBA1	UNRUHS TURF FARM	JOHNSON LANE AREA	420L	5	19	1980	N
320 321 322 323	28CCB1 28CDA1 28CDB1	JANSSEN VIRGIL STIDHAM E.M. ROBERTS		97R 	11 3 10 10	20 19 8 10	1979 1979 1971 1978	N N N
324	28CDB2		1268 ESTHER		7	20	1981	N
325 326	28CDC1	STEVE ERVEN	1269 MELBOURNE	88L	4	18 11	1979 1980	N
327 328 329	28DBC1 28DBD1 28DCB1	LOU GLASGLOW OZNOWICZ CUSHMAN	2855 VICKY 1322 SARATOGA 2847 VICKY	100R 60R	12 2	11	1980 1977 1979	N
330 331 332 333 334	28DCC1 28DCD1 28DDA1 28DDC1	H. KUNKEL VERNON PACK LARRY HICKMAN B. FORD	KIM PLACE 1347 KIM PLACE 1378 KIM PLACE 1353 KIM PLACE	100R 110R 115R	1 10 10 8 8	14 26 22 24	1982 1972 1981 1979 1978	N N N N
335	28DDC2	PERRY WINGERT	1350 STEPHANIE WAY	108R	2	22	1979	N
336 337 338 339	28DDD1 28DDD2 30BDB1	PEG QUIRK SHARON MURRAY STOKES	1375 KIM PLACE 1396 STEPHANIE WAY CRADLEBAUGH RANCH	92L 115L	8 8 12 8	30  3 23	1982 1979 1979 1983	N N U
340 341 342 343 344	31AAC1 32CDC1 33ABB1 33ADC1 33DAA1	DANGREDG ARTESTAN UELL	0.5 MI S CARSON R 0.2 MI E 395 50 FEET N OF JOHNSON LANE 1314 RAELINE 1330 DOWNS 2664 CLAPHAM LANE		3	30 3 14 25 27	1983 1970 1981 1981 1979	U N N N
345 346 347 348 349	33DAD1 33DBB1 33DCA1 33DCD1 33DDC1	CHARLES PORTERFIELD PETE FERLISI ROBERT QUILICI ROBERT GOODPASTURE W.E. JOHNSTON	2653 CLAPHAM LANE 2679 VICKY 1338 JUDY 1319 JUDY 1354 JOHNSON LANE	119R  146R 135R	7 3 9 9	18 17 10 18	1978 1982 1980 1980 1979	2 2 2 2 2
350 351 352 353 354	34BAB1 34BCA1 34BCC1 34BDB1 34BDC1	GARY SALTSGIVER UNKNOWN PARDEW HARLEY KULKIN ANGELO BERNASCONI	2618 GORDON KAYNE DRIVE 2717 CLAPHAM 2739 GORDON	112R 85R 120R 100R		5 3 8	1978 1978 1983 1980 1978	N N U N
355 356 357 358 359	34BDC2 34CAC1 34CDB1 34DAA1 34DAB1	DON COOK JIM WORDEN MIERAU LOUZETTA RICHARD VADENAIS SCOTT MILLER	2721 GORDON 2651 GORDON 2642 STEWART 1588 JONES 1552 JONES	115R 105R 147L 120R	7 12 12 1 5	26  14 10	1979 1975 1975 1979 1978	N N N N
360 361 362 363 364	34DAB2 34DBC1 34DBD1 34DCC1 34DCC2	KUPER TRANSAMERICA MORTGAGE LARRY REED LEONARD LARSON RENE OJEDA	1558 JONES 2653 FULLER 2661 WADE 1510 JOHNSON LANE 2626 WADE	140R 100R 150R	7 7 11 2 5	16 16 21 3	1979 1979 1979 1976 1981	N N N N N N
365 366 367 368 369	34DDC1 34DDD1 35BAD1 35BCA1 35BDA1	JOE ALOTTA TREADWAY G.P. LUNGSTROM JOHN DESMOND ROLAND SAMUELSON	1569 CHIQUITA 1575 CHIQUITA 1668 CHOWBUCK 1649 CHOWBUCK 1690 CHOWBUCK	213L 210R 275L	7 3 6 9 4	17 2 13	1978 1980 1978 1976 1981	N

				DEPTH OF WELL,	SAME	LE DA	TE	ANA-
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	ANA LYS
		Antelope V	alley basin (106), T. 10 N., R	. 22 E.				
370 371 372 373 374	9BBC1 9BDB1 9CBB1 10ABB1 10ABB2	LLOYD GATHRIGHT JAMES DELIMA MERLE WILLIAMS WELL 6 RAY LAGUE	1238 SANDSTONE 3666 TOPAZ RANCH DRIVE 1290 SANDSTONE SHALE DR TOPAZ RANCH ESTATE 3986 SHALE DRIVE	320L 202R	9 10 7 5 12	13 19 18	1978 1978 1978 1980 1980	N N N N
375 376 377 378 379	15ABC1 15ABC2 18ABD1 18ADB1 18ADC1	DON BENGES LEO HAWKINS HOLBROOK JUNCTION WELL JUNCTION BAR WELL JUANITAS	3950 WALKER VIEW ROAD 3962 WALKER VIEW ROAD HOLBROOK JUNCTION HOLBROOK JUNCTION HOLBROOK JUNCTION	267L 196R 100R	4 7 12 5 7	17 12 6	1977 1978 1980 1981 1978	N N N N
380 381 382 383 384	18ADC2 28DDD1 29BBCD1	SIERRA SAVINGS TOPAZ STATE PARK FAIR R	HOLBROOK JUNCTION TOPAZ LAKE TOPAZ LAKE AREA	196R	5 7 3 5 7	12 26 5 20 2	1981 1979 1976 1976 1976	N N N N N
385 386 387 388 389	29BBDB1	MOORE P	TOPAZ LAKE AREA	80R	8 10 3 5 7	16 26 2 20 2	1976 1976 1977 1976 1976	2 2 2 2 2 2
390 391 392 393 394	29BCCB1	GLENCO SPRING	TOPAZ LAKE AREA	ОМ	8 10 12 3 3	16 26 16 2 5	1976 1976 1976 1977 1976	N N N N
395 396 397 398 399					5 7 8 10 12	20 16 26 16	1976 1976 1976 1976 1976	N N N N
400 401 402 403 404	29BDCA2	BRECKENRIDGE J	TOPAZ LAKE AREA	150L	5 7 8 10 3	20 2 16 26 2	1976 1976 1976 1976 1977	N N N N
405 406 407 408 409	29BDDA1	HALLERT J	TOPAZ LAKE AREA	158M 230R	8 10 12 3 8	16 26 16 2 16	1976 1976 1976 1977 1976	N N N
410 411 412 413 414	29BDDD1	BROWN D	TOPAZ LAKE AREA	140R	3 5 7 8 10	2 20 2 16 26	1977 1976 1976 1976 1976	N N N N
415 416 417 418 419	29CACA1	BLANKENSHIP J	TOPAZ LAKE AREA	170L	12 3 2 5 7	16 2 17 20 2	1976 1977 1976 1976 1976	N N N N
420 421 422 423 424			The Sales of the S		8 10 12 3 8	16 26 16 2 17	1976 1976 1976 1977 1983	N N N N
425 426 427 428 429	29CACC1	BARNICLE L	TOPAZ LAKE AREA	245R	2 5 7 8 10	17 20 2 16 26	1976 1976 1976 1976 1976	N N N N

				DEPTH OF WELL,	SAMP	LE DA	TE	
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	ANA-
		Antelope	Valley basin (106), T. 10	N., R. 22 E.				
430 431 432 433 434	29CACC2 29CADA1	CULLISON R D STROBEL F	TOPAZ LAKE AREA TOPAZ LAKE AREA	183L	12 3 10 5 7	16 2 26 20 2	1976 1977 1976 1976 1976	N N N
435 436 437 438 439					8 10 12 3 8	16 26 16 2 17	1976 1976 1976 1977 1983	× × × U
440 441	29CADD1	BINGHAM L	TOPAZ LAKE AREA	140L	10 12 3	26 16	1976 1976	N
442 443 444	29CBDB1	CLUB HORIZONS	TOPAZ LAKE AREA	285L	3 6 8	16 2 13 5	1977 1972 1974	N N
445 446 447 448 449					5 7 8 10 12	20 2 16 26 16	1976 1976 1976 1976 1976	N N N
450 451 452 453 454	29CCAA1 29CCDA1	GREG MAR SITE TOPAZ LODGE	TOPAZ LAKE AREA TOPAZ LAKE AREA	200M 318L	3 10 7 10 5	27 7 2 20	1977 1976 1972 1974 1976	N N N N
455 456 457 458 459					7 8 10 12 3	2 16 26 16 2	1976 1976 1976 1976 1977	N N N N
460 461 462 463 464	29CCDD1	PINENUT LODGE	TOPAZ LAKE AREA	240R	1 6 10 5 7	21 13 2 20 2	1981 1972 1974 1976 1976	N N N N
465 466 467 468 469					8 10 12 3 7	16 26 16 2 26	1976 1976 1976 1977 1979	N N N N
470 471 472 473 474	29CDAB1	WESTON M	TOPAZ LAKE AREA		8 2 5 7 8	17 17 20 2 16	1983 1976 1976 1976 1976	UNNN
475 476 477 478 479	29CDBA1 29CDCC1	PAOLOZZI J TOPAZ SUNRISE SUBD	TOPAZ LAKE AREA TOPAZ LAKE AREA	175L 203L	10 12 3 10 10	26 16 2 26 1	1976 1976 1977 1976 1967	N N N N
480 481 482 483 484					10 4 5 12 12	26 12 30 16 27	1967 1971 1972 1976 1976	N N N
485 486 487 488 489	29DCCC1	TOPAZ LODGE SUBD LOT	TOPAZ LAKE AREA	83L	3 9 6 10 9	2 11 2 29 26	1977 1961 1967 1968 1975	N N N N

TABLE 7.--Water-quality data--Continued

				DEPTH OF WELL,	SAMP	LE DA	TE	
REF.	SECTION	NAME	LOCATION	TOTAL (FEET)	MONTH	DAY	YEAR	LYS
		Antel	ope Valley basin (106), T. 10 N.,	R. 22 E.				
490 491 492 493 494	2900001	TOPAZ LODGE SUBD L	OT (Continued)		2 5 7 8 10	17 20 2 16 26	1976 1976 1976 1976 1976	2 2 2 2
495 496 497 498 499					12 3 6 10 8	16 2 8 18 17	1976 1977 1978 1979 1983	N N N N N U
500 501 502 503 504	32ABBB1	DEVENPECK G	TOPAZ LAKE AREA		1 4 10 12 3	31 12 26 16 2	1984 1972 1976 1976 1977	N N N N N N N N N N N N N N N N N N N
505 506 507 508 509	32BAAA1	BOYINGER	TOPAZ LAKE AREA		3 2 5 7 8	23 17 20 2 16	1972 1976 1976 1976 1976	N N N N N
510 511 512 513 514	32BAAB1	KAY B	TOPAZ LAKE AREA	100L	10 12 3 4 10	26 16 2 12 26	1976 1976 1977 1971 1976	N N N N N N N N N N N N N N N N N N N
515 516 517 518 519	32BAAB2	MCDANIELS J	TOPAZ LAKE AREA	160M	2 5 7 8 10	17 20 2 16 26	1976 1976 1976 1976 1976	2 2 2 2
520 521 522 523 524	32BADA1	SAVAGE B	TOPAZ LAKE AREA	80L	12 3 8 5 7	16 2 17 20 2	1976 1977 1983 1976 1976	2 Z U Z Z
525 526 527 528 529	32BADB1	BOXSTALLER	TOPAZ LAKE AREA		8 10 12 3	16 26 16 2 17	1976 1976 1976 1977 1976	N N N
530 531 532 533 534	32BCDA1	KAHN M	TOPAZ LAKE AREA	200M	5 8 10 3	20 16 26 2	1976 1976 1976 1977 1970	2 2 2 2 2 2 2 2
535 536 537 538 539	32BCDD1 32BDBA1	KOCH H GIOMI LOT 11	TOPAZ LAKE AREA TOPAZ LAKE AREA	95L	3 5 7 8 12	2 20 2 16 16	1977 1976 1976 1976 1976	2 2 2 2 2
540 541 542 543 544	32BDBB1	RIORDAN	TOPAZ LAKE AREA	146R	9 2 7 8 10	17 17 2 16 26	1974 1976 1976 1976 1976	N N N N N
545 546					12	16 2	1976 1977	N

TABLE 7.--Water-quality data--Continued

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		La	ke Taho	e basin	(90), T.	13 N., R	18 E.		
1 2 3	6.6 6.6 7.4	0 3 3	0.0 .0 .8	120 80 84	34 13 22	12 7	::	37 36 6.0	::
		La	ke Taho	e basin	(90), T.	14 N., R	18 E.		
4	7.6	0	0.0	132	46	4.4		46	
		Church	ill Val	ley basi	in (102),	T. 14 N.	, R. 22 E	<u>.</u>	
5	7.0 6.9	::	28 4.0	::	21 21	::	12 11	::	1.5
		Cars	on Vall	ey basir	(105),	T. 11 N.,			
7	6.7		0.5	90	24	7.4	9.3		3.3
		Cars			(105),				
8 9 10 11 12	7.5 8.2 7.8 8.2 7.5	7 7 3  7	3.1 .8 .5  2.2	404 399 491 145 286	119 120 83 32 70	26 24 69 16 27	42 71 55  28	18	1 6 4 1
13 14 15 16	7.4 8.1 7.3 7.4	7  3 3	3.5 1.3 .4	275 212 154 102	69 52 37 21	25 20 15 12	26 21 8	33	1 1 3
		Cars	son Val	ley basir	1 (105),	T. 11 N.,	R. 22 E.		
17 18	7.5 7.8	::	16 15	::	25 24	::	8.7 8.1	::	5.2
		Cars		ley basir	1 (105),				
19 20 21 22 23	7.4 7.3 7.4 7.2 7.5	3 5 7 7	0.3 .2 2.9 1.4 .3	33 51 51 45 41	8.6 17 17 13 13	2.7 2 3 2	9 10 10 11	::	0.7
24 25 26 27 28	7.6 7.2 6.9 7.7 8.4	3 3 7	.5 .8 .3	19 36 28 184 7	6 11 9.0 72 1.7	1 1.4 1.7	13 8 7.4 107 23	::	1 1 .8 3
29 30 31 32 33	7.9 7.4 7.5 7.4 7.1	3 5 3 3	.2 2.8 .8 1.7	29 30 27 5 26	10 12 9.0 2.0 7.0	1 0 1 0 2	5 7 10 18 9	::	1 0 1 1 2
34 35 36 37 38 39	7.4 6.6 8.5 7.8 9.3 8.1	7 3 3 15 7	2.8 .5 6.8 .3 3.2 4.4	13 22 10 13 10 18	5.0 6.4 5 4	0 1.4 0 0 0	17 3.9 17 19 45 15	::	1 2.3 1 1 2 1

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Cars	on Vall	ey basir	(105), 1	. 12 N.,	R. 20 E.		
40 41 42 43 44	7.9 8.1 7.6 8.0 7.9	3 3 3 3	1.2 .2 .7 .7	96 111 200 115 159	27 38 59 33 39	7 4 13 8 15	15 37 18 19	30	3 3 2 3
45 46 47 48 49	7.8 7.6 7.9 7.9 7.9	5 5 3 3	.2 .1 1.0 .1	141 208 231 144 177	40 58 66 38 43	10 16 16 12 17	18  21 17 18	21	3 4 3 4
50 51 52 53 54	7.4 7.6 7.8 7.2 7.4	3 3 2 3	.2 .1 .6 .8	208 110 189 144 131	57 26 51 40 36	16 11 15 11	19 15 23 	20	3 4
55 56 57 58 59	7.5 7.2 7.4 7.6 7.2	3 3 3 3	.2 .4 .5 .0 .2	166 175 88 108 85	40 47 23 13 19	16 14 7.5 18 9	18 19 14  16	25	3 2.9
60 61 62 63 64	7.5 7.8 7.6 7.1 7.3	7 0 6 1 0	.7 .3 2.4 .3 .5	90 130 84 100 91	21 34 22 24 25	9 11 7 10 7	17 21  19	14 6	3 2
65 66 67 68 69	7.5 7.3 7.9 7.4 8.1	5 3 5 7 5	2.0 .6 5.0 .3	90 79 92 83 193	26 20 22 25 54	6 7 9 5 14	13 10 15 14 58	••	2 2 2 3
70 71 72 73 74	7.9 7.5 7.4 8.0 7.8	4 8 3 3 3	.0 .1 .2 .7	76 120 105 142 105	22 30 29 42 32	5 11 8 9 6	17 16 24 15	15	3 2 2 3
75 76 77 78 79	7.8 8.0 8.1 7.8 8.0	0 3 0 0 3	.0 .4 .0 .4	132 124 108 120 110	40 35 30 33 31	11 9 8 9 8	21 17	21 24 16	3
80 81 82 83 84	8.0 7.3 7.3 7.1 7.8	3 4  35 5	1.0 3.0 14	107 220 204 86 76	33 62 58 23 24	6 16 14 7 4	19  24 14	30  13	3 3 3
85 86 87 88 89	7.7 7.8 7.3 7.4 7.0	3 17 2 3 10	9.9 .4 10 .8	81 76 96 81 84	21 22 27 21 22	7 5 7 7	15 14 16 32	13	3 2 2 2 2
90 91 92 93 94	7.2 7.5 7.7 8.0 7.6	3 3 3	.5 .2 .5 .6	86 80 77 112 89	23 22 21 30 24	7 6 6 9 7	14 14 14  19	  47	2 2 2 2 2 2 2

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Cars		ey basir	(105), 1	. 12 N.	R. 20 E.		
95 96 97 98 99	7.0 7.1 7.5 7.2 7.6	3 3 3 3	0.3 .9 .3 1.1 .7	120 108 108 96 132	30 27 25 24 35	11 10 11 9	15  13 	12  15 16	3
100 101 102 103 104	7.0 7.8 8.0 7.2 8.2	5 3 5 7	.9 .7 2.2 .3 2.3	88 128 93 90 61	24 30 24 24 18	7 13 8 7.2 4.0	14 39 12 44	13	3 2 2.7 2.7
105 106 107 108 109	7.8 7.6 8.0 8.1 8.1	3 3 5 0	.2 .9 .3 .4	179 131 73 92 221	42 36 21 29 62	18 10 5 5	19 25 40  17	22	3 2 3
110 111 112 113 114	7.9 7.6 7.6 8.3 7.8	3 7 3 3	2.7 2.5 .3 .4 .3	106 139 304 294 193	26 36 102 101 64	10 12 12 10 8	19 19 32 28 25	::	4 2 3 3 3
115 116 117 118 119	7.6 8.2 8.0 8.0	3 7 5 3	.8 3.4 .2 .3 1.8	304 195 215 170 193	87 60 68 50 54	21 11 11 11 14	31 27 31 23 27	::	4 3 2 3 4
120 121 122 123 124	8.0 7.6 8.3 7.9 7.9	7 3 7 5 7	.2 .4 6.1 1.1 1.6	213 236 140 139 174	72 63 33 36 50	8 19 14 12 12	23 18 26 30 30	::	2 2 4 4 4
125 126 127 128 129	7.9 8.1 8.0 8.0	25 3 7 7 5	14 2.8 1.7 1.4 .3	183 286 346 326 326	50 70 102 96 96	14 27 22 21 21	35 31 37 35 33	::	4 4 5 5 5 5
130 131 132 133	7.7 8.0 8.2 7.7	5 3 3 3	.6 .2 .5	348 175 154 131	85 42 45 36	33 17 10 10	33 19 23 31	::	6 2 3 6
13/	7.8			ley basi		. 15 14.	, R. 21 E.		1
134 135 136 137 138	7.8 8.0 7.9 8.0 8.1	3 7 7 3	0.2 .7 1.2 .2 .8	178 96 101 120 106	50 32 32 38 36	13 4 5 6 4	24 53 53 43 51	::	1 2 3 2 2
139 140 141 142 143	7.8 8.0 8.1 7.9 8.0	3 3 3 3	2.9 .5 1.9	191 164 157 163 150	55 49 48 47 47	13 10 9 11 8	25 24 28 28 31	::	1 2 2 2 2 2
144 145 146 147 148	8.9 7.8 7.5  7.9	5 3 7 	2.2 .4 .5 4.0 2.1	23 79 84 81 111	9.0 25 27 26 36	0 4 4 3.8 5	85 45 42 45 37	::	1 2 2 2.3 2
149 150 151	7.8 7.9 7.7	7 7 3	3.6 3.0 .6	229 238 144	57 59 38	21 22 12	30 25 29	::	6 6 3

TABLE 7.--Water-quality data--Continued

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Cars	on Vall	ey basin	(105), T	. 12 N.,	R. 22 E.		
152 153 154 155	6.9 7.2 7.5 6.9	::	11 3.8 20 27	••	10 38 5.4 7.5	::	8.4 16 5.0 6.4	••	0.6 2.1 1.5 1.9
		Cars	on Vall	ey basin	(105), T	. 13 N.,	R. 19 E.		
156 157 158 159 160	7.7 7.7 7.6 7.9 7.2	0 3 5 3	0.3 1.4 .3	54 113 113 75 118	20 27 32 20 39	1 11 8 6 5	12 13 11 12 14	::	2 2 2 2 2 2
161 162 163 164 165	8.9 9.6 7.7 8.1 7.4	5 7 5 7 8	3.3 1.2 .4 8.0 5.8	0 5 82 42 124	0 2 23 15 40	0 6 1 6	27 29 10 29	18	1 1 4 3
166 167 168 169 170	7.2 6.7 6.8 6.8 7.4	3 5 5 5	.2 .1 .6 .7	104 101 121 94 88	30 29 37 31 27	7 7 7 4 5	10 11 13 12 10	::	4 4 4 3
171 172 173 174 175	6.9 7.2 7.6 7.5 8.0	7 3 7 7 5	3.3 .7 4.0 3.6 3.2	76 80 96 90 90	22 27 27 26 26	5 3 7 6	8 9 10 9	::	2 3 2 2 2 2
176 177 178 179 180	7.6 8.5  7.4	3	.4	94	26 40 43 55	7 2 18 27	10 128 75 82	••	1
181 182 183 184 185	8.2 8.6 8.2 8.2	5	2.0	44 8 47	16 2.3 13 30	1 3.5 24	18 22 16 75	::	2.6
186 187 188 189 190	7.3	5	   .5	  86	27 46  18	14 17  10	32 39  9	••	6
191 192 193 194 195	7.8 8.1 9.1 8.3 8.2	9 3  2	.4	72 80 26 26 52	26 17 9.6 7.6 16	2 9 .5 1.8 3.0	12 137 24 28	16	6 2.9 3.4 3.0
196 197 198	7.2 7.6 7.2	10	.8 .6 .4	85 62 38	24 20 8.8	6 3 3.8	18 9 7.9	::	5 3 1

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Cars	on Vall	ey basir	(105), 1	. 13 N.,	R. 20 E.		
199 200 201 202 203	8.9 8.2 7.7 7.9 8.5	3 7 3 10 7	5.5 4.8 .6 3.8 4.8	13 38 62 10 10	5 15 15 4 4	0 6 0	87 101 57 86 88	::	1 2 4 1 1
204 205 206 207 208	7.7 7.6 7.7 8.0 8.1	3 25 2 3 5	.3 13.0 .5 1.1 .3	70 92 27 27 27	23 27 9 9	3 6 1 1	78 71 57 57 63	::	3 2 4 4 4 4
209 210 211 212 213	8.4 7.8 7.9 8.3 7.8	17 3 2 3 3	5.7 .2 .3 .4 2.0	8 122 120 37 350	3 29 30 13 130	0 12 11 1 6	86 23  31 55	23	2 5 2 7
214 215 216 217 218	8.4 8.1 8.3 8.3	5 3 3 3	.3 .1 .4 .6	52 64 25 34 418	19 21 10 12 146	1 2 0 1 13	35 30 120	26 19 	3 3 7
219 220 221 222 223	8.5 8.0 8.1 8.1 7.9	3 5 3 2	.4 .8 .6 .4	18 431 336 484 20	7 164 130 172 8	0 5 3 13	35 97  112 32	70 	6 3
224 225 226 227 228	8.3 8.5 8.0 7.9 8.0	3 5 3 3	.5 .6 .3 .4	28 28 67 45 63	11 11 17 13 15	0 0 6 3 6.2	29 29 49 58 51	::	3 3 4 3 4.5
229 230 231 232 233	8.3 8.1 7.5 7.4 7.5	= ::	.3 .4 .5 .5	32 70 240 440 120	10 22 54 97 32	1.7 3.7 25 49 9.1	26 14 110 140 25	::	3.6 3.0 2.7 10 2.6
234 235 236 237 238	7.4 8.1 8.2 8.0 8.1	12 2 5 3 7	1.8 1.8 .2 .8 1.1	224 72 72 57 66	64 22 24 18 20	16 4 3 3 4	17 12	35 12 13	2 3
239 240 241 242 243	7.4 7.7 8.3 7.7 8.3	3 3 7 4	.4 .1 .5 .4 2.7	193 90 24 84 68	46 23 6.2 22 21	19 8 2.1 7	19 15 24	14 13	3 2.1 
244 245 246 247 248	8.2 8.0 7.8 7.4 7.9	3 4 4 3	.2 .7 1.3 2.0 1.7	76 53 87 168 162	24 18 25 43 45	4 2 6 15 12	17 14  20	11  16	2 2 3
249 250 251 252 253	7.8 7.2 7.6 7.7 7.3	3 3 35 3 5	.2 .7 12 .2 1.7	249 100 125 115 220	60 27 32 31 62	24 8 11 9 16	20 16 16 16	   24	3 3 2
254 255 256	8.0 7.9 7.9	0 2	.3	216 220 120	59 64 36	17 15 8.1	 26	21 22	1.9

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Cars	on Vall	ey basir	(105), 1	. 13 N.,	R. 21 E.		
257 258 259 260 261	7.8 7.9 7.2 7.9 7.9	5 7 10	2.5 .3 3.8 .4	109 163 157 123 160	29 39 38 31 43	9 16 15 11 12	35 32 37 25	35  	7 5 4 2.2
		Cars	son Vall	ey basir	(105), 1	. 14 N.	R. 19 E.		
262 263 264 265 266	8.0 7.6 7.6 8.1 8.0	7 7 3 3 8	8.5 5.3 2.7 1.9 1.0	37 39 37 43 47	13 14 13 14 14	1 1 2 3	12 11 12 15 10	::	1 2 2 1 1
267 268 269 270 271	7.8 7.4 7.7 7.3 7.9	7 3 3 20 3	4.8 1.6 .6 12	101 52 59 51 38	24 19 17 12 15	10 1 4 5 0	20 14 13 11 60	::	1 1 1 1
272 273 274 275 276	7.3 8.1 7.8 7.9 8.0	3 45 3 3	2.4 .7 18 6.0 1.8	82 45 68 49 44	23 13 26 18 16	6 3 1 1	23 12  32 37	32	2 2 2
277 278 279 280 281	7.4 7.4 7.8 8.1 8.7	5 7 15 5 17	2.1 2.0 4.0 1.9 4.0	51 67 56 48 51	17 22 19 16 17	2 3 2 2 2 2	33 31 36 31	27 	1 2
282 283 284 285 286	7.2 7.6 7.1 7.3 7.2	3 7  3 3	1.2 5.0 .8	115 89 60 14 118	28 29 18 4.0 39	11 4 3.7 1 5	20 17 5.6 71 13	••	2 2 2.5 2
287 288 289	6.8 7.0 7.9	0	1.5	88 86 77	27 27 26	5 4.5 2.8	7.6 38	20	3.1 2.5
200	7.7			ley basir		r. 14 N.	R. 20 E.		
290 291 292 293 294	7.6 7.6 7.6 7.1	7 3 3 3 3 3	1.7 .3 .3 .3	71 119 111 187 170	22 36 33 60 55	4 7 7 9 8	27 27 27 44 40	::	1 2 1 2 2
295 296 297 298 299	7.8 7.8 7.3 7.8 7.9	3 17 3 7	.7 5.6 1.2 8.0 2.8	96 60 110 125 96	27 19 31 37 27	7 3 8 8 7	27 33 32 45 50	::	1 2 1 2
300 301 302 303 304	7.9 7.6 8.1 8.0 8.3	7 7 3 10	2.2 1.3 1.3 1.5 .6	96 120 119 12 13	27 35 31 3.0 3.5	7 8 10 1	50 46 30 62 44	::	1 2 2 1 1
305 306 307 308 309	7.6 7.4 7.0 7.3 9.0	::	9.7	139 140 186 170 429	54 53 74 66 172	1 .7 .1 0.7	122 110 162 170	160	4.7 4

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Cars	on Vall	ey basin	(105), T	. 14 N.,	R. 20 E.		
310 311 312 313 314	8.2 8.7 7.7  8.0	7 7 3 3 3	0.5 .3 .7 .4 .5	43 18 160 77 6	14 7 59 21 18	2 0 3 6 4	84 93 150 43 57	::	2 0 2 5 2
315 316 317 318 319	7.9 8.9 8.7 8.0 8.2	7 5 3 5 7	1.1 .6 .5 1.8 1.2	77 433 445 308 107	21 173 178 123 38	6 0 0 0 3	41 169 177 151 54	::	3 4 4 7
320 321 322 323 324	7.4 7.9 8.2 8.1 8.2	3 3 2 3 5	1.0 .1 .2 .3	136 297 172 139 184	43 117 69 54 72	7 1 0 1	53 152  134 121	120	5 2 4
325 326 327 328 329	7.2 7.9 8.1 7.6 8.6	3 7 7 7 3	2.2 1.0 1.6	108 62 330 203 263	35 15 132 78 105	5 6 0 2 0	65 50 156 136 142	=======================================	5 4 4 3
330 331 332 333 334	8.9 8.2 8.2 7.8 8.1	3 3 15 7 3	.4 .5 5.3 2.4 .6	273 76 127 64 77	109 30 46 19 21	0 0 3 4 6	138  77 69 41	77 	3 3 5 4
335 336 337 338 339	8.1 8.0 7.8 7.8 8.4	3 3 3 25	2.6 .7 .3 13	68 77 61 58 14	19 21 18 15 4.4	5 6 4 5 .8	41 46 53 55 33	::	4 4 4 2.2
340 341 342 343 344	8.2 8.1 7.9 7.8 7.8	18 3 3 5	1.1 2.4 .3 1.7	17 28 56 46 53	4.7 11 16 12 13	1.3 0 4 4 5	42 48 57 59	36	4.3 5 3 4
345 346 347 348 349	7.2 7.8 7.5 7.6 7.6	3 5 7 5 3	.3 .6 .6 .5	65 61 80 71 182	16 16 19 17 45	6 5 8 7 17	52 57 45 52 42	::	4 4 4 6
350 351 352 353 354	8.0 7.7 7.8 7.9	5 3  3 7	1.5 .6 .4 4.4	41 76 110 37 56	10 19 27 10 14	4 7 9.5 3 5	63 45 87 55 54	::	3 4  4 3
355 356 357 358 359	8.0 8.2 8.2 8.2 7.9	3 3 3 7	.1 .6 .5 3.3 9.1	61 51 47 46 61	16 14 14 15 18	5 4 3 2 4	53 50 50 60 56	::	4 4 4 5
360 361 362 363 364	7.6 7.8 7.3 8.2 7.8	3 3 3 5	.3 .2 .4 .4	66 56 47 40 56	20 16 14 11 14	4 4 3 3 5	52 52 61 57 59	::	65544
365 366 367 368 369	7.5 7.8 8.8 9.2 9.1	3 7 3 7 3	2.2 2.1 1.3 4.8	95 50 35 8 20	25 15 14 3 8	8 3 0 0	38 54 132 96 113	::	5 1 0 0

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Antel	ope Val	ley basi	in (106),	T. 10 N.	, R. 22 E		
370 371 372 373 374	7.8 8.0 7.8 8.2 8.1	5 7 7 3	2 1.3 3.5 .4 .4	268 168 197 186 143	84 46 59 58 39	14 13 12 10 11	27 20 24 28 21	::	1 1 1 1 1 1 1
375 376 377 378 379	8.0 8.1 7.4 7.2 7.1	7 5 3 3	1.5 .6 .1 .5	169 189 200 220 269	52 59 52 55 60	9 10 17 20 29	30 25 25 22 18	::	1 1 3 3 3 3
380 381 382 383 384	7.3 7.6 7.5 7.6 7.6	3 8 7 70	10 7	202 181 127 126 117	48 51 36 34 32	20 13 9 10 9	17 46 20 20	::	3 2 1 1
385 386 387 388 389	7.4 7.3 7.4 7.5 7.7	7 35 25 3 3	11 10 1	129 129 126 95 91	35 35 34 28 28	10 10 10 6 5	21 18 18	::	1 1 1
390 391 392 393 394	7.3 7.2 7.5 7.3 8.0	3 7 7 7 3	1 1 8 1	97 97 101 102 373	29 29 29 31 110	6 7 6 24	22 20 17 23	::	1 1 1 1
395 396 397 398 399	8.0 8.0 7.7 7.8 7.8	3 5 3 3 3	1 2 2 2	397 373 379 375 387	116 108 112 112 112	26 25 24 23 26	23  23 24		1  2 1
400 401 402 403 404	8.1 8.0 8.0 7.9 8.0	7 3 17 25 7	10  13 6	157 155 157 157 157	48 47 48 48 49	9 9 9 9	21  24 19	::	1
405 406 407 408 409	7.9 7.6 7.8 7.6 7.5	3 7 3 5	2 2 1 0	145 145 149 144 150	48 48 48 48	6 7 6 5	26 27 24 24	::	1 .0
410 411 412 413 414	7.5 7.8 7.6 7.7 7.5	25 3 3 3 5	10	116 118 116 123 126	40 39 40 41 42	4 5 4 5 5	21 17  23		.0
415 416 417 418 419	7.8 7.5 7.9 7.9 7.9	7 3 3 3	4 2	132 125 151 160 154	43 42 44 46 45	6 5 10 11 10	22 19 20 18		1.0 3 3
420 421 422 423 424	7.8 7.7 7.8 7.6 7.2	3 7 3	2 4 1 .8	156 159 160 157 154	46 47 46 48 45	10 10 11 9	18 18 16 17	::	3 3 2 2.8
425 426 427 428 429	7.7 7.8 8.2 7.7 7.5	3 3 3 5	2 3	129 140 131 136 140	40 41 41 43 43	7 9 7 7 8	19 14  17	::	3 3

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Antel	ope Val	ley basi	n (106),	T. 10 N.	, R. 22 E		
430 431 432 433 434	7.7 7.6 7.7 7.6 7.7	7 5 3 3	3 1 2 3	135 133 152 232 221	41 40 46 65 62	8 8 9 17 16	13 12 20 23	::	3 3 2
435 436 437 438 439	7.6 7.4 7.8 7.4 7.2	3 7 7 5	8 8 3 .5	216 213 216 207 225	60 59 60 60 62	16 16 16 14 17	23 23 21 21	::	2 2 1 1.7
440 441 442 443 444	7.4 7.7 7.5 7.7 7.9	3 3 5 5	3 5 2 1 0	110 112 117 308 299	31 30 32 93 90	8 9 9 18 18	14 12 11  30	35	4 4 2
445 446 447 448 449	7.9 8.0 7.8 7.6 8.0	3 3 3 7	1  3 3	302 290 295 286 282	93 88 90 88 85	17 17 17 16 17	28  28 28	:	2 2 2
450 451 452 453 454	7.9 7.4 7.7 7.5 7.8	7 3 3 0 3	0 4 1 1	287 101 108 99 103	87 29 32 28 28	17 7 7 7 8	26 13  16 10	12	3 4
455 456 457 458 459	7.9 7.6 7.6 7.8 7.5	3 3 3 3	1 0 0	92 99 105 100 100	27 28 29 27 27	6 7 8 8 8	14 10 8		4 4 4
460 461 462 463 464	7.4 7.4 7.6 8.0	5 0 3 7	1 0 0 1	125 100 90 89 56	32 30 26 24 16	11 6 6 7 4	11 15 8	13	5 4 4
465 466 467 468 469	7.7 7.4 7.8 7.5 7.3	3 3 3 3 3	1 1 0 .3	95 94 94 94 90	25 26 26 26 23	8 7 7 7 8	11 10 8 9	:-	4 3 4 4
470 471 472 473 474	7.0 7.7 7.9 7.6 7.7	3 3 3 3	1 1	88 110 100 96 98	23 31 27 27 26	7.3 8 8 7 8	8.9 18 11	::	3.5
475 476 477 478 479	7.4 7.7 7.5 7.5 7.4	3 7 3 3 10	2 1 1 2 10	103 112 110 109 160	28 30 31 32 40	8 9 8 7 15	11 10 10 12	::	4 4 4
480 481 482 483 484	7.7 8.3 7.5 7.7 7.8	0 3 2 17 7	0 0 1 10 4	152 196 208 205 174	40 59 51 64 53	13 12 19 11 10	14 13	19 24 13	6
485 486 487 488 489	8.0 7.6 7.8 8.0 7.8	3 0 0 0 3	0	126 92 98 106 117	49 16 34 28 32	1 13 3 9	12   22	28 31 15	5 5

REF.	pH (STAND- ARD UNITS)	COLOR (PLAT- NUM- COBALT UNITS)	TUR- BID- ITY (NTU)	HARD- NESS (MG/L AS CACO3)	CALCIUM DIS- SOLVED (MG/L AS CA)	MAGNE- SIUM- DIS- SOLVED (MG/L AS MG)	SODIUM- DIS- SOLVED (MG/L AS NA)	SODIUM+ POTAS- SIUM, DIS- SOLVED (MG/L AS NA)	POTAS- SIUM, DIS- SOLVED (MG/L AS K)
		Antel	ope Val	ley bas	in (106),	T. 10 N.	, R. 22 E		
490 491 492 493 494	7.8 8.1 8.1 7.6 7.7	3 3 3 3	.0	114 116 107 110 110	29 30 28 29 29	10 10 9 9	22 22  17	::	5
495 496 497 498 499	7.9 7.7 7.8 7.6 7.4	3 3 3	1 .0 .8 .4	114 112 116 115 146	29 30 30 28 37	10 9 10 11 13	16 15 16 17 22	::	5 5 5 5 5 5 5
500 501 502 503 504	7.7 7.4 7.7 7.9 7.4	3 2 3 3 7	2 1 4 1	137 148 155 150 151	35 40 39 37 39	12 12 14 14 13	21  15 14 13	18	6 6 6
505 506 507 508 509	7.6 7.8 8.2 7.6	0 3 3 3 3	1 1 .0	124 132 130 139 139	35 33 34 34 36	9 12 11 13 12	18 11 	17  	6
510 511 512 513 514	7.5 7.7 7.5 7.7 7.6	3 3 5 3	1 2 1 .0	137 138 137 168 169	35 34 35 42 43	12 13 12 16 15	13 12 11 	13	5 6 5 6
515 516 517 518 519	7.5 7.9 7.6 7.3 7.4	3 3 3 3 3	1 1 2	147 159 167 181 181	39 42 42 46 46	12 13 15 16 16	18 11  15	••	5
520 521 522 523 524	7.7 7.5 7.0 8.0 8.0	7 3 3 3	.6 .0	169 160 204 94 91	43 41 52 26 25	15 14 18 7 7	13 11 14 10	::	6.4
525 526 527 528 529	7.7 7.6 8.0 7.8 7.8	3 7 3 3	1 1 .0	91 94 95 94 86	25 26 25 26 23	7 7 8 7 7	11 11 9 17	::	5 5 5 4
530 531 532 533 534	7.7 7.6 7.7 7.5 7.8	3 3 3 2	1.0	89 93 86 84 180	24 24 23 22 48	7 8 7 7 15	10 8	   59	4
535 536 537 538 539	8.2 7.9 7.8 7.6 7.9	7 3 3 3 3	7 .0  1	195 87 80 90 91	55 20 19 21 20	14 9 8 9	39 7  8	::	3 3
540 541 542 543 544	8.0 8.2 8.1 8.1	3 3 3 5	1	101 100 104 99 98	29 30 30 28 31	7 6 7 7 5	23 22  21	::	3 4
545 546	8.2 8.1	7	4	102 100	31 32	6 5	21 20	::	4 3

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
				Lake Taho	e basin	(90), T.	13 N., R	2. 18 E.			
1 2 3	137 120 98	0 0 0	112 98 80	48 53 4	13 4 5	0.06	::	::	210 119 131	25.0 8.3	::
				Lake Taho	e basin	(90), T.	14 N., F	2. 18 E.			
4	181	0	148	77	2				160		
			Ch	urchill Va	alley bas	sin (102)	. T. 14 M	I., R. 22 E	L.		
5	67 88	::	::	32 5.9	5.5	::	::	::	430 480	0.05	::
			c	arson Vall		1 (105),	T. 11 N.,	R. 20 E.			
7	60	0	51	9.6	38	0.10	34	179		3.6	
			2	arson Vall	ey basir	1 (105),	T. 11 N.	R. 21 E.			
8 9 10 11	293 261 351 169	0 16 0 0	240 246 288	231 254 259 39	16 23 12 4.0	0.20 .01 .10	::	::	622 754 691	0.8 .3 .5	::
12	161	0	132	180	6	.11	••		469	2.9	
13 14 15 16	161 194 129 134	0 0 0	132 106 110	183 112 72 3	8 6.0 3	.13	::	::	433 249 178	3.5 5.7 5.2	::
				Carson Va	lley bas	in (105)	T. 11 N.	, R. 22 E			
17 18	82 94	::	:-	7.8 7.4	7.7 7.0	Ξ	II.	::	400 480	6.0	
			2	Carson Val	ley basi	n (105),	T. 12 N.	R. 19 E.			
19 20 21 22 23	64 83 78 73 73	0 0 0 0	54 68 64 60	1.5 2 2 2 2	0.3 2 0 2 3	0.20 .16 .20 .20	31 21	74   	102 109 90 76	4.8 3.3 2.3 2.6	::
24 25 26 27 28	49 46 43 90 55	0 0 0 0	40 38 37 74 48	1 3 5.5 278 13	0 2 1.7 23 1.1	.56 .50 .20 2.48 .70	28  31	74  93	63 80  589	2.4 8.2 5.3 2.3	::
29 30 31 32 33	46 37 49 39 46	0 0 0 0	38 30 40 32 38	2 9 4 10 3	0 1 1 1 1 2	.18 .23 .09 .29	::	::	68 72 88 77 76	.8 1.8 4.0 1.1 3.3	::
34 35 36 37 38 39	29 32 20 37 5 41	0 0 6 0 18 0	24 30 28 30 40 34	19 2.8 17 18 53 15	.7 2 0 4 0	.77 .20 .84 .87 2.02 .26	19	50   	74 74 83 159 75	2.4	::

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
			<u>c</u>	arson Vall	ey basir	(105),	T. 12 N.	R. 20 E.			
40 41 42 43 44	105 159 205 129 168	0 0 0 0	86 130 168 106 138	21 46 63 27 30	9 10 20 16 12	0.11 .10 .03 .12 .10	32   	::	145 247 301 189 219	4.1 5.1 4.8 2.7 5.6	::
45 46 47 48 49	163 266 295 146 200	0 0 0 0	134 218 242 120 164	30 17 24 37 22	10 8 10 11 7	.15 .10 .11 .12	::	::	238 274 303 199 233	4.5 7.2 6.0 4.2 6.4	::
50 51 52 53 54	266 122 217 159 190	0 0 0 0	218 100 178 130 156	24 25 37 24 24	6 7 11 10 8	.17 .14 .06 .17	::	••	279 166 266 196 208	8.0 3.8 4.5 4.3 8.0	::
55 56 57 58 59	217 176 120 144 117	0 0 0 0	178 144 94 118 96	24 27 16 24 15	6 5 4.0 7 3	.21 .07 .20	34	149	238 174  185 161	5.1 10.5 4.0 13.6 7.9	::
60 61 62 63 64	124 154 76 112 100	0 0 0 0	102 126 62 92 82	14 23 20 12 23	3 9 4 3 8	.15 .05 .12 .13	::	::	178 207 172 178 158	6.2 6.7 31.0 5.6 7.7	::
65 66 67 68 69	112 95 112 107 315	0 0 0 0	92 78 92 88 258	23 18 20 17 39	4 7 5 9	.10 .11 .16 .31	32   36	••	180 133 155 115 382	7.7 12.1 4.0 2.9 10.4	
70 71 72 73 74	110 146 127 168 117	0 0 0 0 0 0	90 120 104 138 96	11 24 23 25 22	5 8 7 11 9	.08 .12 .14 .08	::	::	141 167 184 232 191	2.7 2.6 6.0 4.0	
75 76 77 78 79	183 154 134 142 115	0 0 0 0	150 126 110 116 94	14 27 5 19 32	6 13 12 8 6	.14	::		110 209 189 196 197	5.0 4.1 12.6 7.7 3.2	::
80 81 82 83 84	122 281 253 100 105	0 0 0 0	100 230 82 86	19 24 21 20 11	12 8 10 4 5	.07  .10 .15	33	::	124 296 307c 139 146	7.1 25.0 19 1.1	0.16
85 86 87 88 89	100 93 120 110 95	0 0 0 0	82 76 98 90 78	18 21 17 28 23	8 6 5 26 5	.11 .15 .07 .61	::	::	117 142 180 212 154	.9 1.4 5.4 6.0 3.3	::
90 91 92 93 94	98 98 95 195 112	0 0 0 0	80 78 160 92	21 21 24 35 19	6 5 4 8 8	.13 .13 .14 1.09 .14	::	::	149 147 153 253 162	.8 1.6 4.9 9.0 3.2	::

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
			2	arson Vall	ey basir	(105),	T. 12 N.,	R. 20 E.			
95 96 97 98 99	129 139 115 127 171	0 0 0	106 114 94 104 140	25 11 30 13 13	6 3 5 5 6	0.10 .23 .15 	::	::	176 170 150 168 191	9.0 5.4 4.8 5.0 5.5	::
100 101 102 103 104	115 163 176 120 105	0 0 0 0 12	94 134 144 99 110	11 14 18 14 32	5 2 2.8 15	.17 .11 .88 	28 35	154	134 186 202  202	4.8 3.0 5.3 8.0 3.0	::
105 106 107 108 109	207 176 149 110 268	0 0 0 0	170 144 122 90 220	16 20 24 39 17	7 7 7 6 5	.12 .16 .31 .00	::	::	234 225 196 186 317	14.9 10.3 6.6 1.0 3.7	::
110 111 112 113 114	127 200 246 222 232	0 0 0 10	104 164 202 202 190	36 14 94 86 36	10 6 36 39 7	.13 .08 .12 .09 < .10	::	::	186 236 430 456 309	1.0 3.6 23.4 19.8 7.7	::
115 116 117 118 119	351 220 207 156 242	0 10 0 0	288 200 170 128 198	52 34 64 64 37	5 9 15 7 7	.12 .04 .04 .10	::	::	394 284 324 257 279	11.3 6.9 11.9 10.4 7.0	::
120 121 122 123 124	239 283 119 149 227	0 0 6 0	196 232 110 122 186	29 17 71 61 24	10 6 8 14 18	.02 .09 .13 .08	::	::	285 285 253 261 290	9.6 10.1 1.3 4.6 8.3	::
125 126 127 128 129	240 290 320 312 307	0 0 0 0	196 238 262 256 252	30 41 30 32 27	20 32 40 40 41	.08 .01 .10 .00	::	::	294 444 544 509 518	8.6 37.6 84.0 81.3 75.0	0.24
130 131 132 133	293 183 154 146	0 0 0	240 150 126 120	40 24 48 55	43 11 10 12	< .10 < .10 < .10 .09	::	::	505 239 238 296	69.4 16.8 9.2 1.9	::
			400	Carson Va			T. 12 N		272		
134 135 136 137 138	232 188 183 195 188	0 0 0 0	190 154 150 160 154	22 43 55 40 45	4 6 5 7	0.28 .19 .39 .21 .14	=======================================	=	232 251 264 256 244	1.4 .7 1.0 .9 1.0	
139 140 141 142 143	249 215 220 212 220	0 0 0 0	204 176 180 174 180	26 21 19 23 24	6 5 5 4 6	.22 .12 .15 .20	::	::	261 231 221 238 248	2.4 1.5 1.0 1.3	::
144 145 146 147 148	56 142 149  139	7 0 0  0	58 116 122 117 114	131 27 28 27 49	15 17 16 8	.26 .18 .13 .20	28  30	217	284 196 243  212	.1 21.1 21.3 19.5 7.9	::
149 150 151	205 207 178	0 0 0	168 170 146	102 91 34	11 6 6	.05 .16 .18	::	::	426 384 220	4.4 4.1 1.3	::

									And the second s	The second secon	
REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
			<u>c</u>	arson Vall	ey basir	(105),	T. 12 N.	, R. 22 E.			
152 153 154 155	78 160 24 30	::	::	7.6 5.6 6.4	5.5 10 5.5 6.3	::		::	630 140 180	<0.02 .17 .17 .16	
			<u>c</u>	arson Val	ley basir	(105),	T. 13 N.	, R. 19 E.			
156 157 158 159 160	102 151 132 110 163	0 0 0 0	84 124 108 90 134	3 6 3 4 11	0 1 2 1 0	0.01 .06 .11 .00	   21	::	112 159 155 122 161	2.1 3.5 16.4 1.6	::
161 162 163 164 165	24 20 122 88 181	14 14 0 0	48 44 100 72 148	13 23 10 25 10	3 0 1 1 3	.11 .13 .16 1.38	::	. :	90 92 177 131 182	.0 1.5 .4 1.3	::
166 167 168 169 170	134 117 127 112 117	0 0 0 0	110 96 104 92 96	6 7 15 8 3	5 5 11 4 4	< .10 .12 .09 .09	::	••	151 144 201 178 140	12.2 23.8 28.5 22.8 10.9	0.09
171 172 173 174 175	107 115 122 132 112	0 0 0 0	88 94 100 108 92	4 4 3 3 3	2 0 1 1 4	< .10 .03 .08 .09	28	::	122 127 98 150 160	6.3 3.9 8.4 7.7 7.6	.06
176 177 178 179 180	124	0	102 220 246 300	2	4	.00		• • • • • • • • • • • • • • • • • • • •	131 748 543 502 616	9.5 1.7 .5 2.4	
181 182 183 184 185	83 57 83	0 3 0	68 52 69 140	15 10 13	5 1.6 3.5	.39 .70 .40	55 50	116 134	151	.7 .0  1.1 .8	
186 187 188 189 190	124		120 130 102	::	1	.01	::	••	365 295 308	.3 .3 .1	
191 192 193 194 195	132 126 12 81 110	0 0 24 0 0	108 104 66 90	0 200 18 13	1 5 46 2.6 3	.07 .13 5.00 .40 .56	61 48	137	121 113 492c  160	.1 1.2 .3 	.06
196 197 198	85 93 63	0 0	70 76 54	5 0 1.5	16 0 .6	.09 .09 .20	21 22	70	168 102	23.9 1.2 .5	::

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
			2	arson Vall	ey basir	(105),	T. 13 N.,	R. 20 E.			
199 200 201 202 203	20 27 120 49 20	12 6 0 0	40 34 98 40 36	113 171 49 112 114	25 27 15 27 25	4.68 3.72 2.26 4.20 4.20	::	::	307 362 275 298 304	0.7 .1 2.9 .0	::
204 205 206 207 208	105 93 108 110 132	0 0 0	86 76 88 90 108	93 144 54 44 54	23 25 10 8 9	2.94 2.12 1.50 1.82 1.60	::	::	308 348 265 293 261	9.8 2.6 .3 .9 3.5	::
209 210 211 212 213	146 146 151 81 149	10 0 0 6 0	140 120 124 78 122	26 13 29 21 237	12 14 10 5 66	3.12 .19 .18 .94 .75	::	::	296 233 235 186 663	4.5 11.5 3.9 .1	::
214 215 216 217 218	107 90 63 63 237	8 0 6 6	88 74 64 64 194	13 19 28 24 357	5 9 3 5 87	.45 .45 1.04 1.04 .25	::	::	163 163 160 175 951	.5 .4 .0 .1	::
219 220 221 222 223	51 185 120 244 76	14 0 0 0	70 152 98 200 62	19 356 271 382 20	5 82 77 84 4	1.15 .28 .53 .29 .80	56  	::	175 894 701 996 170	.0 .6 .0 6.4	::
224 225 226 227 228	68 63 168 146 180	6 8 0 0	68 68 138 120 156	19 19 24 28 17	4 4 10 9 8.9	.60 .59 .83 3.04 1.20	   58	238	160 159 269 243	.6 .4 2.4 2.2 .7	::
229 230 231 232 233	84 98 456 673 160	0 0 0 0	72 85 379 524 134	19 17 92 180 30	4.9 4.8 14 40 8.4	.40 .20 .50 .30 .20	52 45 59 63 51	152 152 569 873 226	::	1.1 1.5 12.0 2.9	:-
234 235 236 237 238	293 93 88 83 78	0 0 0 0	240 76 72 68 64	39 16 17 15 16	10 5 6 7 3	.12 .10 .04 .06 .08	::	::	344 123 132 134 135	6.2 2.2 1.3 1.0 1.7	::
239 240 241 242 243	200 107 69 117 90	0 0 0 0	164 88 61 96 74	33 17 20 11 15	10 7 1.9 5 4	.10 .13 .30 .12 .07	39	124	276 143  156 108	12.8 2.5 .4 .8 3.3	::
244 245 246 247 248	93 88 102 200 202	0 0 0 0	76 72 84 164 166	16 15 17 22 20	5 8 4 5 9	.13 .08 .13 .05	::	::	133 139 140 231 225	1.5 .9 2.5 12.5 5.0	::
249 250 251 252 253	300 129 146 142 281	0 0 0 0	246 106 120 116 230	20 21 29 22 19	8 5 5 7	.14 .21 .75 .20	::	::	300 182 188 170 293	9.8 .4 3.7 3.7 16.5	::
254 255 256	283 285 180	0	232 234 153	15 14 21	5 7 5	.20	36	216	288 297	8.0 10.7 3.1	::

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
			2	arson Vall	ey basir	(105),	T. 13 N.	R. 21 E.			
257 258 259 260 261	140 166 156 171 230	0 0 0	136 128 140 179	57 45 44 45 27	8 25 31 7 7.1	0.18 .24 .17 .10	56 22	239	339 320 271	20.9 14.7 .7 3.1	::
				Carson Vall		100000	T. 14 N.	R. 19 E.			
262 263 264 265 266	73 76 78 81 73	0 0 0 0	60 62 64 66 60	1 5 4 3 1	3 2 0 5 3	0.12 .00 .06 .35 .14	::	••	100 128 116 114 97	5.9 1.7 .3 7.6 8.5	::
267 268 269 270 271	124 98 100 66 81	0 0 0 0	102 80 82 54 66	19 0 3 7 44	7 4 2 2 2 36	.09 .00 .04 .14 2.70	   32	::	227 102 120 97 230	4.1 .5 .7 5.0	::
272 273 274 275 276	127 78 110 100 98	0 0 0 0	104 64 90 82 80	2 0 29 21 20	7 0 13 15	.36 < .10 2.00 1.53 1.66			187 98 191 182 170	10.0 4.5 .2 .0	::
277 278 279 280 281	105 98 90 95 90	0 0 0 0 8	86 80 74 78 90	19 25 23 25 11	13 15 12 15 8	1.32 1.74 1.81 1.40 .28	::		168 158 185 172 145	5.2  .0 6.5	
282 283 284 285 286	171 129 88 183 122	0 0 0 0	140 106 75 150 100	0 4 2.4 9	6 4 .8 7 9	.30 .01  .11 .05	22	90	190 143  203 199	3.9 9.2 .8 .1 14.6	
287 288 289	112 120 140	0 0 0	92 100 118	34 5 30	3 1.2 10	1.30	28 24	130 197	126	3.0 4.1 2.5	::
226				Carson Val			T. 14 N.	, R. 20 E.			
290 291 292 293 294	117 134 120 183 185	0 0 0	96 110 98 150 152	16 25 20 31 27	10 13 12 17 15	1.44 .72 .63 .69 .68	62	::	182 261 259 409 371	3.0 28.1 40.6 69.5 59.6	::
295 296 297 298 299	137 112 161 173 183	0 0 0 0	112 92 132 142 150	12 22 18 18 27	10 13 12 32 15	.99 1.93 1.45 1.28 .45	::	::	200 174 244 239 259	15.1 .7 6.6 9.8 3.9	::
300 301 302 303 304	178 224 166 61 66	0 0 0 0	146 184 136 50 54	26 18 14 68 42	15 17 12 20 10	.45 .31 .61 2.84 2.00	38	166	285 265 245 229	4.4 4.7 13.3 .5	::
305 306 307 308 309	47 50 17 39 4	0  0 7	41	300 300 418 470 678	21 23 37 38 39	2.90	39  35 20	548 798	603 842 1,100e	.4 .4 .3 .4	::

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUÉ AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
				Carson Val	ley basi	n (105)	T. 14 N.	R. 20 E.			
310 311 312 313 314	59 37 56 144 115	10 0 0	52 50 46 118 94	125 123 362 35 58	28 31 31 6 14	3.76 5.32 3.20 1.55 3.08	::	::	295 306 719 236 269	1.4 .0 .0 2.6 3.1	::
315 316 317 318 319	129 10 0 27 110	0 2 4 0	106 12 8 22 90	40 725 710 522 121	6 38 39 30 11	1.22 3.36 3.30 1.53 1.12	  28	::	256 1173 1170 946 344	1.9 .1 .3 2.5 1.5	::
320 321 322 323 324	124 24 98 81 85	0 0 0 0	102 20 80 66 70	117 553 308 288 319	14 28 23 23 25	1.12 3.60 3.30 3.13 2.98	::	::	347 926 618 607 663	2.3 .2 .6 7.2 12.3	::
325 326 327 328 329	183 142 12 41	0 0 0 0 4	150 116 10 34 10	63 34 600 400 533	19 8 34 24 26	1.51 1.57 3.58 2.80 3.96	::	::	327 253 1032 689 898	1.2 2.1 .2 .0	::
330 331 332 333 334	0 171 93 139 134	4 0 0 0	8 140 76 114 110	514 75 182 54 43	28 16 11 28 8	3.80 2.82 1.79 1.54 1.58	49 47 	::	859 331 402 284 245	2.9 1.1 1.7 2.4	::
335 336 337 338 339	144 146 146 144 81	0 0 0 0	118 120 120 118 70	36 34 38 37 16	8 8 9 9 4.8	1.50 1.52 1.75 1.65 1.00	73  51	   152	247 253 254 248	2.8 2.9 .7	::
340 341 342 343 344	54 78 154 122 137	0 0 0 0	45 64 126 100 112	47 32 37 47 40	6.1 6 7 11 12	.80 1.10 1.65 2.78 2.56	59  68 	193	142 250 258 266	.4 .0 2.6 1.9 2.2	::
345 346 347 348 349	137 132 156 161 173	0 0 0 0	112 108 128 132 142	38 46 27 28 44	7 12 11 10 21	1.78 2.66 1.15 1.94 .70	66	::	262 274 204 233 367	2.6 1.7 .4 5.9 70.0	::
350 351 352 353 354	120 159 129 122	0 0  0 0	98 130  106 100	54 35 56 38 39	13 8 27 6 10	3.00 1.40 1.73 1.70 1.62	61	::	271 235  240 233	.3 4.2 23.9 2.3 2.4	::
355 356 357 358 359	122 134 105 117 105	0 0 12 0	100 110 110 96 86	52 38 40 46 82	15 16 13 12 19	1.41 1.98 2.02 2.46 1.37	::	::	248 228 226 251 297	5.2 3.3 3.2 1.8 2.7	::
360 361 362 363 364	102 134 98 110 102	0 0 0 0	84 110 80 90 84	70 41 68 52 59	17 11 18 11 21	1.30 1.68 1.52 2.12 1.75	::	::	269 261 276 252 249	3.8 4.2 3.6 3.1 3.8	::
365 366 367 368 369	112 107 17 22 2	0 0 10 14 16	92 88 34 46 34	74 55 202 111 174	18 11 35 30 28	.82 1.63 4.72 4.96 4.96	::	::	290 269 461 322 386	2.7 .1 .1	::

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
			Ar	itelope Val	ley basi	n (106),	T. 10 N.	, R. 22 E.			
370 371 372 373 374	181 176 220 173 171	0 0 0 2	148 144 180 146 140	140 32 48 87 38	8.0 10 5.0 6	0.24 .15 .12 .20 .14	::	::	414 235 289 297 193	0.3 7.6 4.1 2.0 8.5	::
375 376 377 378 379	200 205 239 232 264	0 0 0 0	164 168 196 190 216	52 68 56 44 50	7 5 11 11 10	.01  .15 .21 .16	::		281 277 347 334 349	8.9 2.5 7.8 11 12.2	::
380 381 382 383 384	227 195 132 129 115	0 0 0 0	186 160 108 106 94	34 62 60 64 59	5 26 4 6 5	.19 .22 .0 .2	::	:-	268 331 233 255 282	8.6 12.5 1.5 2.0 3.4	0.19 .37 .47
385 386 387 388 389	124 124 124 120 117	0 0 0 0	102 102 102 98 96	60 64 63 33 32	3 2 7 5	.1 .2		::	244 258 219 234	2.2 2.2 2.4 1.4 1.5	.45 .57 .94 .97
390 391 392 393 394	117 115 119 122 295	0 0 0 0	96 94 98 100 242	32 33 34 34 163	5 5 8 4 7	.3 .4 .3	::	::	232 220 219 505	1.4 1.3 1.4 1.2 7.9	.96  .93 1.00 .10
395 396 397 398 399	298 293 288 293 293	0 0 0 0	244 240 236 240 240	172 159 162 158 165	4 5 3 8 5	.0	::		541 559 535 540	7.6 7.5 7.6 7.2 6.3	.09 .09 .09
400 401 402 403 404	195 195 195 190 200	0 0 0 0	160 160 160 156 164	31 30 30 32 30	6 5 3 5 4	:1	::	::	250 259 261 254	.0 .0 .1 .4	.13 .04 .10
405 406 407 408 409	200 195 195 195 173	0 0 0 0	164 160 160 160 142	29 31 31 31 26	5 7 7 5 5	.3	::	::	255 276 244 259	1.3 1.4 2.1 1.2 2.0	.67 .61 .69
410 411 412 413 414	171 134 142 149 156	0 0 0 0	140 110 116 122 128	27 14 12 14 16	6 10 10 10 21	.1 .2	::	::	225 223 229  267	1.3 20 23 22 23	.14 .35 .37 .33
415 416 417 418 419	159 156 168 159 163	0 0 0 0	130 128 138 130 134	17 15 45 46 41	13 10 3 5 4	.1 .2 .1 .2	::	::	232 236 234 253 268	23 26 11 12 16	.28 .38 .46 .51
420 421 422 423 424	161 159 159 154 170	0 0 0 0	132 130 130 126	43 54 56 56 47	3 4 5 3 2.4	.2 .2 .1 .2	   35	   239	262 262 276	14 11 11 10 9.7	.31
425 426 427 428 429	149 142 144 142 139	0 0 0 0	122 116 118 116 114	38 42 43 46 43	2 4 2 2 5	.0 .1	::	::	210 228 223  240	2.5 2.4 2.6 2.9 2.9	.24 .27 .26 .28

REF.	BICAR- BONATE, (MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	ALKA- LINITY (MG/L AS CACO3)	SULFATE, DIS- SOLVED (MG/L AS SO4)	CHLO- RIDE, DIS- SOLVED (MG/L AS CL)	FLUO- RIDE, DIS- SOLVED (MG/L AS F)	SILICA, DIS- SOLVED (MG/L AS SIO2)	SOLIDS, RESIDUE AT 180 DEGREES CELSIUS, DIS- SOLVED (MG/L)	SOLIDS, RESIDUE AT 105 DEGREES CELSIUS, DIS- SOLVED (MG/L)	NITRO- GEN, NITRATE TOTAL, (MG/L AS NO3)	PHOS- PHATE TOTAL, (MG/L AS PO4)
				Antelope	Valley (	106), T.	10 N., F	22 E.			
430 431 432 433 434	146 149 146 200 190	0 0 0 0	120 122 120 164 156	43 38 63 95 99	5 0 4 5 4	0.0 .2 .1 .1	::	::	217 222 280 367 355	2.9 2.6 2.5 9.5	0.21 .21 .14 .31
435 436 437 438 439	188 185 195 190 200	0 0 0 0	154 152 160 156	91 89 86 83 100	3 4 5 5 3.5	.2 .1 .1	   39	350	335 335 347	11 11 11 11 12	.31
440 441 442 443 444	144 146 146 256 237	0 0 0 0	118 120 120 210 194	30 29 31 148 142	2 5 0 3 3	.2	::	::	228 202 203 443 454	2.1 1.8 20 17	.08
445 446 447 448 449	239 237 237 232 220	0 0 0 0	196 194 194 190 180	133 130 131 137 129	5 3 3 4	.1	::	::	455 460 430 432	18 21 14 14 14	.32 .17 .24 
450 451 452 453 454	241 127 134 132 132	0 0 0 0	198 104 110 108 108	117 19 14 13 13	4 4 4 3	.1 .2 .0 .1	::	::	449 200 152 192 178	14 .7 3.8 2.2 1.9	.22
455 456 457 458 459	134 132 134 134 132	0 0 0 0	110 108 110 110 108	11 11 18 13 12	2 1 2 5 0	.1 .0 .2	::	::	176 206 171 177	2.3 2.2 2.0 2.1 1.6	.21 .21 .22 .22
460 461 462 463 464	146 137 127 122 90	0 0 0	120 112 104 100 74	13 13 11 11 7	3 0 2 1 1	.1 .0 .0 .1	::	::	206 171 171 171 171 146	8.6 1.8 1.6 1.4	.17
465 466 467 468 469	127 129 129 132 112	0 0 0 0	104 106 106 108 92	11 11 12 11 12	0 3 4 2 2	.1 .2 .1	::	::	192 175 173 145	1.6 1.6 1.8 1.4	.18
470 471 472 473 474	120 139 127 132 129	0 0 0	100 114 104 108 106	11 26 17 16 17	1.7 5 2 1 2	<.1 .0 .1	48   	152   	182 181 214	4.2 1.4 1.4 1.7	.14 .37 .32 .35
475 476 477 478 479		0 0 0	106 110 110 110 160	22 28 28 25 14	3 2 0 7 14	.2	::	=	179 183 199 178 246	1.5 1.5 1.3 1.5	.24
480 481 482 483 484		0 14 0 0	148 204 206 202 180	24 15 13 14 13	9 13 8 20 14	.0 .0 .1	::	::	240 289 288 289 284	48 13 7.1 4.9 4.1	.06
485 486 487 488 489	163 149 167	0 0 0 0	134 122 138 124 130	14 24 14 14 13	8 3 6 1 8	.1	::	=======================================	223 174  230	3.2 2.0 3.8 8.0	.04

	BICAR-		ALKA-	SULFATE,	CHLO- RIDE,	FLUO- RIDE,	SILICA, DIS-	SOLIDS, RESIDUE AT 180 DEGREES	SOLIDS, RESIDUE AT 105 DEGREES	NITRO- GEN,	PHOS-
REF.	MG/L AS HCO3)	CAR- BONATE, (MG/L AS CO3)	(MG/L AS CACO3)	DIS- SOLVED (MG/L AS SO4)	DIS- SOLVED (MG/L AS CL)	DIS- SOLVED (MG/L AS F)	SOLVED (MG/L AS SIO2)	DIS- SOLVED (MG/L)	DIS- SOLVED (MG/L)	NITRATE TOTAL, (MG/L AS NO3)	TOTAL, (MG/L AS PO4)
			Ar	ntelope Val	ley basi	n (106)	T. 10 N.	, R. 22 E.			
490 491 492 493 494	168 168 159 156 154	0 0 0 0	138 138 130 128 126	14 14 14 13	5 5 4 4 5	0.1	::	::	206 235 212  206	8.7 9.8 9.6 8.2 8.0	0.76 .79 .66 .64
495 496 497 498 499	149 151 146 156 200	0 0 0 0	122 124 120 128	12 15 12 14 16	5 4 5 5 5.9	.2 .14 .16	   50	239	201 200 192 188	7.7 7.6 8.5 8.4 16.4	.59
500 501 502 503 504	181 178 188 185 195	0 0 0 0	148 146 154 152 160	15 14 13 15 13	5 10 8 10 8	.08 .1 .1	::	••	216 241 249 240 258	12.3 17 14 14 15	.24
505 506 507 508 509	154 159 168 154 159	0 0 0 0	126 130 138 126 130	12 10 11 11 10	7 9 8 9 7	.1 .1 .1	••	::	204 217 240 254	15 16 15 18 18	.39 .51 .49 .47
510 511 512 513 514	163 163 159 195 195	0 0 0 0	134 134 130 160 160	10 10 10 8 12	10 10 8 11 10	.0 .1 .1 .1	::	::	233 221 241 246 249	17 16 16 15 9.4	.31
515 516 517 518 519	183 195 198 212 212	0 0 0 0	150 160 162 174 174	11 12 11 11 13	8 8 9 10 12	.0	••		224 255 261 300	9.6 11 14 16 15	.33 .29 .35 .36
520 521 522 523 524	207 207 240 134 134	0 0 0 0	170 170 170 110 110	12 12 13 11 9	10 8 15 1	.3 .1 <.1 .1	51	286	262 260  192 183	11 9.2 19.9 2.0 1.8	.25 .17 .63 .60
525 526 527 528 529	139 134 134 137 127	0 0 0 0	114 110 110 112 104	9 10 12 11 10	4 3 4 2 2 2	.3 .4 .1		••	173 187 188 160	2.1 2.1 2.2 1.5	.57 .56 .44 .34
530 531 532 533 534	122 117 112 117 312	, 0 0 0 0	100 96 92 96 256	11 9 9 5 12	0 0 2 1 8	.1 .1 .1	::	::	168 173 353	1.2 1.4 1.4 1.1 38	.29
535 536 537 538 539	188 115 110 117 122	34 0 0 0 0	222 94 90 96 100	41 10 9.0 8.0	12 1 1 0 2	.2 .2	::		328 160 145 157	7.4 .7 .3 .8	.13 .65 .60 .64
540 541 542 543 544	171 127 156 151 149	0 12 0 0	140 128 128 124 124	16 18 16 16 19	1 3 1 1 2	.0 .12 	::		212 185 227 188	4.0 2.0 3.9 2.7 2.6	.65 .55 .64
545 546	156 159	0	128 130	20 18	4	:1	::	::	199 196	2.1	.59 .58

TABLE 7.--Water-quality data--Continued

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Lak	ce Tahoe b	pasin (90	), T. 13	N., T. 1	8 E.		
1									
2		::	::		::	0.13			
		Lak	ce Tahoe b	pasin (90	), T. 14	N., R. 1	8 E.		
4									
		Churchi	ill Valley	basin (	(102), T.	14 N., R	. 22 E.		
5			::	••	::		<0.01		::
0	-	Carso	on Valley	basin (	105) T 1	11 N P			
7	0.04	<0.001	0.054	0.02	<0.01	0.005	<0.001	0.036	
,	0.04		on Valley					0.030	
•				Dasin (		0.48	0.04	0.31	
8		0.000 .040	0.020		0.00	.92	.04		
10	::	.000				.19	.05		
11				••		.55	.05		
13		.000			::	.56	.01	::	
14 15		.000			.40	.11	.02	.19	
16		.000	••		.02	.02	.01	.59	
		Carso	on Valley	basin (	105), T.	11 N., R.	22 E.		
17 18	-:-	-:-		::	::	::	<0.01	::	::
		Carso	on Valley	basin (	105). T.	12 N., R.			
19	0.03	<0.001	0.01	0.01	<0.01	<0.003	<0.001	<0.003	
20		.000			.20	.06	.00	2.04	
21		.000	.02	.0	.25	.28	.01	.87	
20 21 22 23		.000			••	.08	.00		
24 25	::	.000		:-	::	.02	.00	::	
26	.05	< .001	.016	.01	< .01	< .003	.001	.007	
27 28	.09	.025	.014	.05	< .01	.01	< .001	< .003	
		.000			.00	.14	.00	.19	
29 30		.000			::	.00	.00		
31 32 33	- ::	.000		- ::		.19	.00		
33		.000					.05		
34	.05	.000	.034	< .01	< .08	.40	< .001	.31	::
36		.000				.29	.01		
34 35 36 37 38 39	::	.000	.00	::	.00	.29 .01 .42 1.15	.01	.03	::
20		.000			.34	1.15	.01	.03	

TABLE 7.--Water-quality data--Continued

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Carso	n Valley	basin (1	05), T. 1	2 N., R.	20 E.		
40 41 42 43 44	::	0.005 .005 .000	0.06   .1 < .25	0.2	0.03	0.25 .00 .05 .04 .00	0.12 .00 .00 .01	0.53	<0.1
45 46 47 48 49	::	.000  .005 .000	.13	::	.00	.00 .10 .00	.00 .00 .00	.04   .01 .00	< .1 < .2 < .1
50 51 52 53 54	::	.000 .000 .000 	.16	.2	.00	.00 .00 .09 .02	.00	.01 .00 .01	< .1 < .1 < .1
55 56 57 58 59	0.04	.000 .000 .004 	< .25 .12 .091	.17	.00 .00 < .01	.02 .10 .014 .15	.00 .00 .001	.04 .07 .071	< .1 < .1 < .2
60 61 62 63 64	::	.000 .005  .005	::	::	.00   	.03 .00 .10 .00	.00	.26	::
65 66 67 68 69	::	.000 .000 .005 .000	.08	:1	1.29 .07  .01	.49 .04 .38 .01 .28	.06 .02 .00 .01	1.66	::
70 71 72 73 74	=======================================	.000 .000 .005	< .25 .1	::	.00	.02 .00 .02 .10	.00 .00 .00	.19	< .1
75 76 77 78 79	::	    .010	::	::	::	.05 .08 .00	.00	::	::
80 81 82 83 84	.09	.010	. <u>:</u>	.3	::	.05 .28 .81	.00	::	::
85 86 87 88 89	::	.005 .005 .000 .000	< .25 .05	::	.00 .02 	.03 2.45 .02 .53	.01 .02 .01	.02	< .1
90 91 92 93 94	::	.000 .005 .025 .015	< .25 .06 .06	.1	.00 .01 .01	.04 .00 .08 .05	.00	.00	< .1

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Carso	on Valley	basin (1	05), T. 1	2 N., R.	20 E.		
95 96 97 98 99	::	0.005	::	::	:: ::	0.00 .02 .04 .15 .03	.00	::	::
100 101 102 103 104	0.07	.000 .005 < .001 .040	0.08	0.1	0.00	.04 .24 .45 < .003 .22	.00 .10 < .001	0.80	::
105 106 107 108 109	::	.005 .015 .000 .015	.08	 	.01 .06 .01	.01 .10 .01 .03	.00 .00 .00	.02 .03 .01	::
110 111 112 113 114	::	.000 .000 .000 .000	.11	::	.05	.26 .09 .02 .03	.02 .01 .00 .01	.42	::
115 116 117 118 119	::	.000 .000 .000 .000	::	::	::	.16 .07 .01 .02 .06	.01 .00 .00 .00	::	::
120 121 122 123 124	::	.000 .000 .000	::	::	::	.00 .05 .27 .00	.00 .00 .00 .00	::	::
125 126 127 128 129	.20	.000 .000 .000 .000	::	::		2.50 .07 .03 .04	.80 .01 .00 .01	::	0.0
130 131 132 133	::	.000 .000 .000	::	::	::	.02 .09 .03 .00	.00 .00 .00	::	::
			on Valley	basin (	105), T.	12 N., R.			
134 135 136 137 138	::	0.000 .010 .010 .005 .015	::	::	0.31	0.02 .01 .12 .04 .04	0.00 .01 .01 .00	0.12	
139 140 141 142 143	::	.005 .005 .005 .000	::	::	::	.01 .30 .00 .05	.01 .00 .00 .00	::	::
144 145 146 147 148	0.03	.005 .010 .010 .015 .005	0.00	0.1	.00	.15 .03 .03 .009	.00 .02 .00 .003	.19 .27 .072	0.04
149 150 151	::	.000 .000 .020	::	::	.01	.75 .71 .03	.06 .03 .01	.30 .88	::

TABLE 7.--Water-quality data--Continued

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Carso	n Valley	basin (1	05), T. 1	2 N., R.	22 E.		
152 153 154 155	::	  	::	::	••	••	<0.01 < .01 < .01 < .01	::	::
		Carso	n Valley	basin (1	05), T. 1	13 N., R.	19 E.		
156 157 158 159 160	::	0.000	0.01	0.00	0.15	0.00 .02 .14 .00	0.00 .01 .04 .00 .04	0.66	::
161 162 163 164 165	::	.005 .000 .010 .000	::	::	.00	1.13 .18 .01 .80 .52	.00 .01 .00 .01	.05	••
166 167 168 169 170	0.09	.000 .000 .000 .000	::	::	.07	.03 .00 .01 .19	.00 .01 .00 .00	.02	<0.2
171 172 173 174 175	.03	.000 .000 .000 .000	.01	.10	.01	.84 .07 .17 .36	.04 .00 .01 .02	.40  .84	< .2
176 177 178 179 180	••	.000	::		••	.04	.08	••	••
181 182 183 184 185	.22	.000	.024	.07	< .01 < .01	.15 .047 .048	.08 .04 .085	.007	::
186 187 188 189 190	:-	::	::	::	::	.00	.00	::	::
191 192 193 194 195	.06	.018 .010	.053	.07	< .01	.97 .01 .01 .029	.11	.008	::
196 197 198	.10	.000	.01	.0	.01	.08 .08 .055	.03 .00 < .001	.04	::

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Carso	on Valley	basin (1	105), T. 1	13 N., R.	20 E.		
199 200 201 202 203	::	0.005 .010 .005 .010 .005	0.00	::	0.00   	0.00 .11 .09 .44 .41	0.00 .02 .00 .03 .02	0.05	::
204 205 206 207 208	::	.005 .000 .030 .045 .025	::	::	.00	.06 .44 .05 .08	.00 .02 .00 .00	6.56	::
209 210 211 212 213	::	.010 .010 .010 .045 .020	::	::	.00	.37 .04 .38 .03 .27	.04 .00 .00 .13	  -01	::
214 215 216 217 218	::	.035 .013 .025 .035	::	::	.00	.32 .00 .46 .15	.04 .02 .24	.00	::
219 220 221 222 223	::	.030 .020 .030 .015 .035	.04	0.10	.02	.02 .12 .08 .00	.00 .55  .12	.16	
224 225 226 227 228	0.03	.035 .035 .005 .010	  .0 .062	.14	.00 .00  .00 < .01	.02 .04 .00 .04 .005	.00 .00 .00 .00 < .001	.17 .02  .00 .28	::
229 230 231 232 233	.04 .02 .23 .34 .04	.032 .014 .044 .034 .008	.049 .061 .098 .14 .094	.16 .15 .61 .72	< .01 < .01 < .01 < .01 < .01	.031 .01 .003 < .003 .039	.045 < .001 .09 .026 .004	.003 < .003 < .003 < .003 .045	
234 235 236 237 238	::	.005 .010 .010	  < .20 .06	::	   .02	.18 1.40 .07 .03 .01	.01	.00	:-
239 240 241 242 243	.05	.000 .005 .011 	.032	.04	.00 < .01	.59 .00 .003 .06 .66	.00 .00 < .001	.02 < .003	::
244 245 246 247 248	::	.010 .010 .000 	 < .25  .20	::	.00	.01 .05 .30 .30	.03	.02	<0.1
249 250 251 252 253	::	.000 .000 .000	< .25  .09	::	.00	.03 .02 1.76 .00 .24	.00 .01 .05 .00	.10	< .1 < .1 < .1
254 255 256	.04	.007	.082	.11	< .01	.08 .06 < .003	.002	.041	::

	(MG/L AS P)	SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Carso	n Valley	basin (1	05), T. 1	3 N., R.	21 E.		
257 258 259 260 261	0.03	0.020 .020 .010 .005	0.10 .077	0.00	0.00 .01 < .01	0.27 .06 .25 .003	0.01 .00 .08 .001	0.14 .91 .01	
		Carso	n Valley	basin (1	05), T. 1	4 N., R.	19 E.		
262 263 264 265 266	::	0.000 .010 .005 .000	::	::		0.18 .01 .00 .13 .14	0.03 .01 .00 .18 .03	::	::
267 268 269 270 271	::	.000	0.0	0.60	0.02	1.62 .01 .02 1.54 .01	.05 .01 .01 .01	0.04	::
272 273 274 275 276	::	.000 .000 .050 .010 .005	::	::	••	.20 .03 .80 .15	.07 .01 .20 .00	::	::
277 278 279 280 281	::	.000 .005 .010 .000	.02	::	.02   .04	.16 .35 .48 .13 1.03	.02 .02 .00	1.79	::
282 283 284 285 286	0.02	.000 .000 < .001 .005 .000	.052	< .01	< .01 .01	.20 .06 .016 .16	.00 .01 .002 .01	.004	<0.1
287 288 289	.03	.001	.044	< .01 .26	< .01 < .01	.005	.001 < .001	.003	::
		Carso	on Valley	basin (1	105), T.	14 N., R.	20 E.		
290 291 292 293 294	:	0.005 .005 .020 .000	:	::	0.03	0.21 .04 .06 .05	0.00 .01 .02 .00	0.27	::
295 296 297 298 299	::	.015 .005 .000 .000	0.00	::	.02	.04 .42 .04 .56	.02 .03 .00 .03	.20	::
300 301 302 303 304	0.17	.000 .000 .000 .010	.022	0.24	.00	.85 .27 .06 .12 .069	.03 .01 .00 .03 .053	.33	::
305 306 307 308 309	.03	.011	.02	1.0	< .01 < .01	1.20	.110	.092	::

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Carso	on Valley	basin (1	05), T.	14 N., R.	20 E.		
310 311 312 313 314	::	0.005 .015 .000 .020 .015	::	::	0.02 .00 	0.11 .01 .31 .04	0.00 .00 .39 .00	0.22 .10 	::
315 316 317 318 319	::	.015 .005 .005 .005 .065	0.00	0.80	.00 .01  .04 .02	.04 .06 .03 .29	.02 .01 .01 .01 .03	.23 .01 .16 .01	::
320 321 322 323 324	::	.020 .020 .020 .020 .020	   .02	::	   .04	.08 .19 .01 .02	.01 .02  .00	.11	::
325 326 327 328 329	::	.020 .020 .005 .020 .010	::	::	.07	.04 .19 .22 .27	.00 .04 .01 .03	.61	::
330 331 332 333 334	::	.005 .025 .010 .010 .020	.02	1.00	.02 .01	.02 .00 .46 .27	.00 .02 .02	.03	::
335 336 337 338 339	0.05	.010 .020 .030 .005	.04	.10	.00	.19 .06 .00 1.04 .066	.04 .00 .00 .05 .041	.03	::
340 341 342 343 344	.14	.046 .080 .020 .015 .025	.043	.19   .20	< .01  .10 .00	.033 .43 .07 .00	.01 .01 .00	.003	::
345 346 347 348 349	::	.020 .015 .005 .015	.04	.20	.06 .03 .03	.02 .07 .16 .04	.01 .02 .10 .00	.26 .77 .10	::
350 351 352 353 354	::	.015 .010  .020 .015	::	::	.01	.07 .01  .03 .63	.03 .02 .00	.07	::
355 356 357 358 359	::	.015 .020 .020 .020 .025	::	::	::	.00 .02 .00 .13	.00 .01 .01 .01	::	
360 361 362 363 364	::	.025 .015 .020 .020 .010		::	.02	.13 .02 .06 .00 2.01	.01 .00 .02 .00	.27	
365 366 367 368 369	::	.015 .040 .050 .005	::	::	   .00	.15 .01 .08 .52	.02 .01 .00 .00	.16	::

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Antelo	pe Valley	y basin (	106), T.	10 N., R	. 22 E.		
370 371 372 373 374	::	0.000 .000 .000 .000	::	•••	0.00	0.99 .04 .92 .13 .03	0.11 .01 .04 .00 .01	0.13	::
375 376 377 378 379	::	.000 .000 .000 .005	::	::	.01 .08 .01	.13 .12 .03 .08	.01 .01 .00 .00	.34	<0.1 < .1 < .2
380 381 382 383 384	::	.000 .010 .010 .010	::	::	.01	.00 .01 .87 1.24	.00 .00 .56 .15	.41	< .1
385 386 387 388 389	::	.020 .025 .040	::	::	::	1.49 1.97 .00	.07 .07 .00		0 0 0 0
390 391 392 393 394	::	.050 .055 .065 .020	::	::	::	.01 .04 3.42 .01	.00 .00 .01	::	0 0 0 0
395 396 397 398 399	::	.015  .010 .010		::	::	.03	.00	••	0 0 0 0
400 401 402 403 404	::	.015  .015 .020	::	::	::	1.07	.27  .27 .20	::	0 0 0
405 406 407 408 409	::	.324 .375 .370 .325	=======================================	::		.15 .07 .01 .05	.00 .01 .00	::	0 0 0 0
410 411 412 413 414	::	.015 .005  .005	::	::	::	1.50 .01 	.06		0 0 0 0
415 416 417 418 419	::	.010 .015 .030 .030	::	::	::	.03 .08 2.52 .00	.01 .00 .09 .00		0 0 0 0
420 421 422 423 424	0.13	.020 .025 .035 .025	0.068	0.01	.03	.45 .22 .06 < .003	.01 .01 .00 .002	.36	0 0 0
425 426 427 428 429	::	.035 .030  .030	::	::	::	.38 .02  .27	.00	::	0 0 0 0

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Antelo	pe Valley	/ basin (	106), T.	10 N., R	. 22 E.		
430 431 432 433 434	::	0.035 .035 .015 .010	::	::	::	0.20 .23 .06 .07	0.00 .00 .01 .01	::	0 0 0 0
435 436 437 438 439	0.09	.015 .015 .020 .020	0.081	0.01	0.02	.32 .39 .41 .011	.01 .01 .01 .001	0.18	0
440 441 442 443 444	::	.005 .050 .015 .015 .005	::	:-		.04 .02 .20 .04	.01 .01 .03 	::	0
445 446 447 448 449	::	.015  .015 .015		::	::	.00  .12 .14	.00  .00 .01	::	0 0 0
450 451 452 453 454	::	.020 .050 .010 .015 .020	::	::	::	.21 .14 .06 .03	.01 .01 .00	::	0
455 456 457 458 459	::	.015 .020 .025	::	::	::	.27 .01	.01 .00	::	0 0 0 0
460 461 462 463 464	::	.010  .015 .015	::	::	.03	.02 .01 .01 .06	.00 .00 .01	.53   	0
465 466 467 468 469	::	.015 .020 .020	::	::	   .00	.00 .01 .04	.01 .00 .00	.06	0
470 471 472 473 474	.06   	.017 .010 .020	.053   	.01   	< .01  	< .003 .01 .01	< .001 .00 .00	.025	0 0 0
475 476 477 478 479	::	.020 .025 .035 .020	::	::	::	.03 .01 .17 .07	.00	::	0
480 481 482 483 484	::	.010 .005 .000	::	::	::	.02 .03 .00 .69	.00	::	0
485 486 487 488 489	::	.015   .035	::	::	::	.59 .08 	.00	::	0

TABLE 7.--Water-quality data--Continued

REF.	PHOS- PHORUS, ORTHO, DIS- SOLVED (MG/L AS P)	ARSENIC, DIS- SOLVED (MG/L AS AS)	BARIUM, DIS- SOLVED (MG/L AS BA)	BORON, DIS- SOLVED (MG/L AS B)	COPPER, DIS- SOLVED (MG/L AS CU)	IRON, DIS- SOLVED (MG/L AS FE)	MANGA- NESE, DIS- SOLVED (MG/L AS MN)	ZINC, DIS- SOLVED (MG/L AS ZN)	METHY- LENE BLUE ACTIVE SUB- STANCE (MG/L)
		Antelo	pe Valley	basin (	106), T.	10 N., R	. 22 E.		
490 491 492 493 494	••	0.045 .040  .035	::	::	::	0.01 .00  .01	0.00	::	0 0 0 0
495 496 497 498 499	0.18	.035 .045 .025 .035 .024	0.092	0.02	0.00 .07 < .01	.00 .13 .06 .01 < .003	.00 .00 .01 .00 < .001	0.15 .05 .009	0
500 501 502 503 504	::	.035 .020 .020 .150 .035	.08	.0   	.04   	.00 .04 .05 .20	.00 .00 .01 .00	.04  	0 0 0
505 506 507 508 509	::	.250 .025 .020	::	::	::	.03 .16 .00	.00	::	0 0 0
510 511 512 513 514	::	.025 .020 .025 .005 .010	::	::	::	.04 .04 .04 .02	.00	::	0 0 0
515 516 517 518 519	::	.020 .015  .015	••	::	::	.01	.01	••	0 0 0 0
520 521 522 523 524	.07	.015 .025 .017 .035	.13	.02	< .01	.00 .18 .007 .00	.00 .01 < .001 .00	.62	0
525 526 527 528 529	::	.045 .040 .040 .025	::	::	' :: ::	.00 .00 .01	.00	::	0 0 0
530 531 532 533 534	::	.020  .020 .030 .040	::	::	::	.00 .02 .04 .34	.00	::	0
535 536 537 538 539	::	.010 .050  .045	::	::	::	.69 .00  .01	.06	::	0 0 0
540 541 542 543 544	::	.075 .075  .095	::	::	::	.00	.00	::	0 0 0
545 546	::	.080	::		::	.25	.01	::	0

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