

Geologic Setting and Water Quality of Selected Basins in the Active Coal-Mining Areas of Ohio, 1989-91, with a Summary of Water Quality for 1985-91

By Alan C. Sedam and Donna S. Francy

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CONVERSION FACTORS AND VERTICAL DATUM

Multiply	By	To obtain
inch (in.)	25.4	millimeter
foot (ft)	0.3048	meter
mile (mi)	1.609	kilometer
acre	0.4047	hectare
square mile (mi ²)	2.590	square kilometer
foot per mile (ft/mi)	0.1894	meter per kilometer
cubic foot per second (ft ³ /s)	0.02832	cubic meter per second
gallon per minute (gal/min)	0.06309	liter per second
ton (short)	0.90718	metric tons

Chemical concentrations and water temperature are given in metric units. Chemical concentration is given in milligrams per liter (mg/L) or micrograms per liter (µg/L). Milligrams per liter is a unit expressing the concentration of chemical constituents in solution as weight (milligrams) of solute per unit volume (liter) of water. One thousand micrograms per liter is equivalent to one milligram per liter. For concentrations less than 7,000 mg/L, the numerical value is the same as for concentrations in parts per million.

Temperature is given in degrees Celsius (°C), which can be converted to degrees Fahrenheit (°F) by the following equation:

$$F = 1.8(^{\circ}\text{C}) + 32$$

Sea level: In this report “sea level” refers to the National Geodetic Vertical Datum of 1929 (NGVD of 1929)—a geodetic datum derived from a general adjustment of the first-order level nets of both the United States and Canada, formerly called Sea Level Datum of 1929.

IDENTIFICATION OF SAMPLE-COLLECTION SITE

In water-quality data tables, collection sites are identified as follows:

Surface- water site	Station number	Basin code	Station name	Latitude	Longitude
	03116950	G-2	NEWMAN C NR MASSILLON OH	(LAT 40 49 22N	LONG 081 33 06W)

Ground- water site	Station number	Local number	Owner and location	Latitude	Longitude
	403830081220700	TU-53	US POST OFFICE AT SANDYVILLE OH	(LAT 40 38 30N	LONG 081 22 07W)

STANDARD ABBREVIATIONS USED IN STATION NAMES

AB Above	C Creek	L Little	NR Near	TR Tributary
B Branch	E East	LK Lake	R River	W West
BK Brook	F Fork	M Middle	RN Run	
BL Below	G Great	N North	S South	

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ABSTRACT

This report presents streamwater- and ground-water-quality data collected to characterize the baseline water quality for 21 drainage basins in the coal-mining region of eastern Ohio. The study area is mostly within the unglaciated part of eastern Ohio along the western edge of the Appalachian Plateaus Physiographic Province. The data collected from 1989-91 and presented in this report represent the third and final phase of a 7-year study to assess baseline water quality in Ohio's coal region during 1985-91.

During 1989-91, 246 samples from 41 streamwater sites were collected periodically from a long-term site network. Ranges and medians of measurements made at the long-term streamwater sites were the following: specific conductance, 270 to 5,170 and 792 microsiemens per centimeter at 25 degrees Celsius; pH, 2.7 to 9.1 and 7.8; and alkalinity, 1 to 391 and 116 mg/L (milligrams per liter). Ranges and medians of laboratory analyses of the same samples were the following: dissolved sulfate, 13 to 2,100 and 200 mg/L; dissolved aluminum, <10 to 17,000 and 300 µg/L (micrograms per liter); dissolved iron, <10 to 53,000 and 60 µg/L; and dissolved manganese, <10 to 17,000 and 295 µg/L. The ranges for concentrations of total recoverable aluminum, iron, and manganese were similar to the ranges of concentrations found for dissolved constituents. Medians of total recoverable aluminum and iron were about 10 times greater than the medians of dissolved aluminum and iron.

During 1989-91, once-only sample collections were done at 45 streamwater sites in nine basins chosen for synoptic sampling. At several sites in the Middle Hocking River basin and Leading Creek basin, water had low pH and high concentrations of dissolved sulfate and total recoverable and dissolved aluminum, iron, and manganese. These water-quality characteristics are commonly associated with acid mine drainage.

Throughout the entire 7-year study (1985-91), medians for most constituents at the long-term streamwater-sampling sites were fairly consistent, despite the geographic diversity of the study area. Waters from several long-term sites, including several sites in the Moxahala Creek and Middle Hocking River basins, had low pH and high concentrations of several constituents, including dissolved sulfate, iron, aluminum, and manganese; this combination of characteristics is indicative of acid drainage from surface-mining operations. At many of the streamwater sites where concentrations of these constituents were high, pH values in the neutral or alkaline range were indicative of stream buffering by carbonate rock or restoration of mined lands in the drainage system. The basins with sites in this category include Yellow and Cross Creeks and Wheeling Creek basins. Water quality at other sites showed little or no effects from surface mining.

Ground-water samples collected during the last phase of the study (1989-91) were mostly from unconsolidated aquifers. The waters were generally hard to very hard and calcium

bicarbonate in type. During the entire 7-year study period, medians of pH in ground-water samples varied little, and most values were in the alkaline range. Except for a few sites where concentrations of dissolved sulfate exceeded 250 mg/L and concentrations of total recoverable and dissolved iron and manganese exceeded 1,000 µg/L, the quality of ground water at the wells sampled in the study area showed little effect from coal mining.

INTRODUCTION

Coal is Ohio's most important mineral resource. At present, coal production is nearly a \$1 billion industry. In 1989, the value of coal was about equal to that of all of the other mineral commodities in Ohio, including oil and gas (Lopez, 1991, p. 9).

Ohio coal production, which was nearly 48 million tons in 1918, generally declined after World War I until the 1940's, when production increased to fulfill the needs of World War II. The greatest production was 55 million tons in 1970. In 1989, Ohio coal production was 31.4 million tons, about 57 percent of the 1970 peak. According to Carlton (1991), much of the decline can be attributed to environmental constraints imposed by Federal legislation on the coal industry in 1970, 1979, and 1990 to improve air quality and to the relatively high sulfur content of Ohio coals.

Surface mining, which began in Ohio just before World War I, grew rapidly after World War II, and, in 1989, accounted for 65.5 percent of coal mined in the State. Except for some efforts in the 1930's to seal off underground mines, which produced acid drainage (Federal Water Pollution Control Administration, 1968), little environmental control was placed on coal-mining practices until 1948, when the State's first surface-mining law was passed. Overall, the law was ineffective because many mine operators found it cheaper to forfeit their reclamation bonds than to do postmining restoration; the result was a sub-

stantial increase in abandoned surface-mined lands. This situation prompted enactment in 1972 of the Nation's most comprehensive (at the time) surface-mine reclamation law, which required "... extensive preplanning of mining and subsequent reclamation, and compliance with mining rules and regulations to insure restoration of areas affected by strip mining" (Ohio Board on Unreclaimed Strip Mined Lands, 1974, p. 2). The Ohio law was modified to conform to the U.S. Surface Mining Control and Reclamation Act of 1977 (Public Law 95-87), which provides for money collected from Federal fees on surface-mined coal to be used by the State in reclamation of abandoned surface-mined lands (U.S. Department of the Interior, 1979).

The Ohio Department of Natural Resources (ODNR), Division of Reclamation, is responsible for issuing coal-mining permits and enforcing regulations regarding restoration and reclamation of mined lands. When the 1972 law was enacted, the Board on Unreclaimed Strip Mined Lands was set up to assess the State's abandoned surface-mined lands and to make recommendations for remediation of these properties (Ohio Board on Unreclaimed Strip Mined Lands, 1974). Various studies have been done by Federal and State agencies during the past two decades. The Division of Reclamation found that systematic compilation of baseline hydrologic data was needed to facilitate the agencies' processing of coal-mining permits in unmined areas and in areas of active mining. In addition, long-term collection of these baseline data could help in the assessment of cumulative hydrologic effects of coal mining over the years on surface-water and ground-water systems throughout eastern Ohio.

In 1984, preliminary discussions between the U.S. Geological Survey (USGS) and the ODNR, Division of Reclamation, led to development of a 7-year cooperative study that began in 1985 with the establishment of a long-term surface-water site network for

assessment of baseline water quality. The network was periodically sampled from September 1985 through November 1991. For the purpose of this investigation, "baseline data" refers to water-quality data that are collected to describe a hydrologic system at a particular point in time. Periodic collections of such data serve as a base to which future data gathered for a particular area can be compared. The long-term sampling was supplemented by synoptic short-term data collections from surface- and ground-water sites in basins (usually three each year) selected for intensive study.

With this report, results of the 7-year study are complete. Previous reports presented data collected from 1985 through 1986 (Jones, 1988) and data collected from 1987 through 1988 (Sedam, 1991), along with descriptions of geologic settings. This report presents data collected from 1989 through 1991 and includes a discussion of water quality for the entire study period (1985-91).

Purpose and Scope

This report presents streamwater- and ground-water-quality data collected to characterize the baseline water quality of 21 drainage basins in the coal-mining region of eastern Ohio. To determine the baseline water quality from 1989 through 1991, investigators (1) collected streamwater-quality data 6 times at 41 long-term sites¹ in the 21 basins, and (2) selected 10 basins for a synoptic data collection, in which water-quality samples were collected at 45 streamwater sites in 9 basins and 28 ground-water sites in 7 basins.

¹ In this report, "site" refers to a fixed site or position along a stream, or a water well where water samples are collected for chemical analysis.

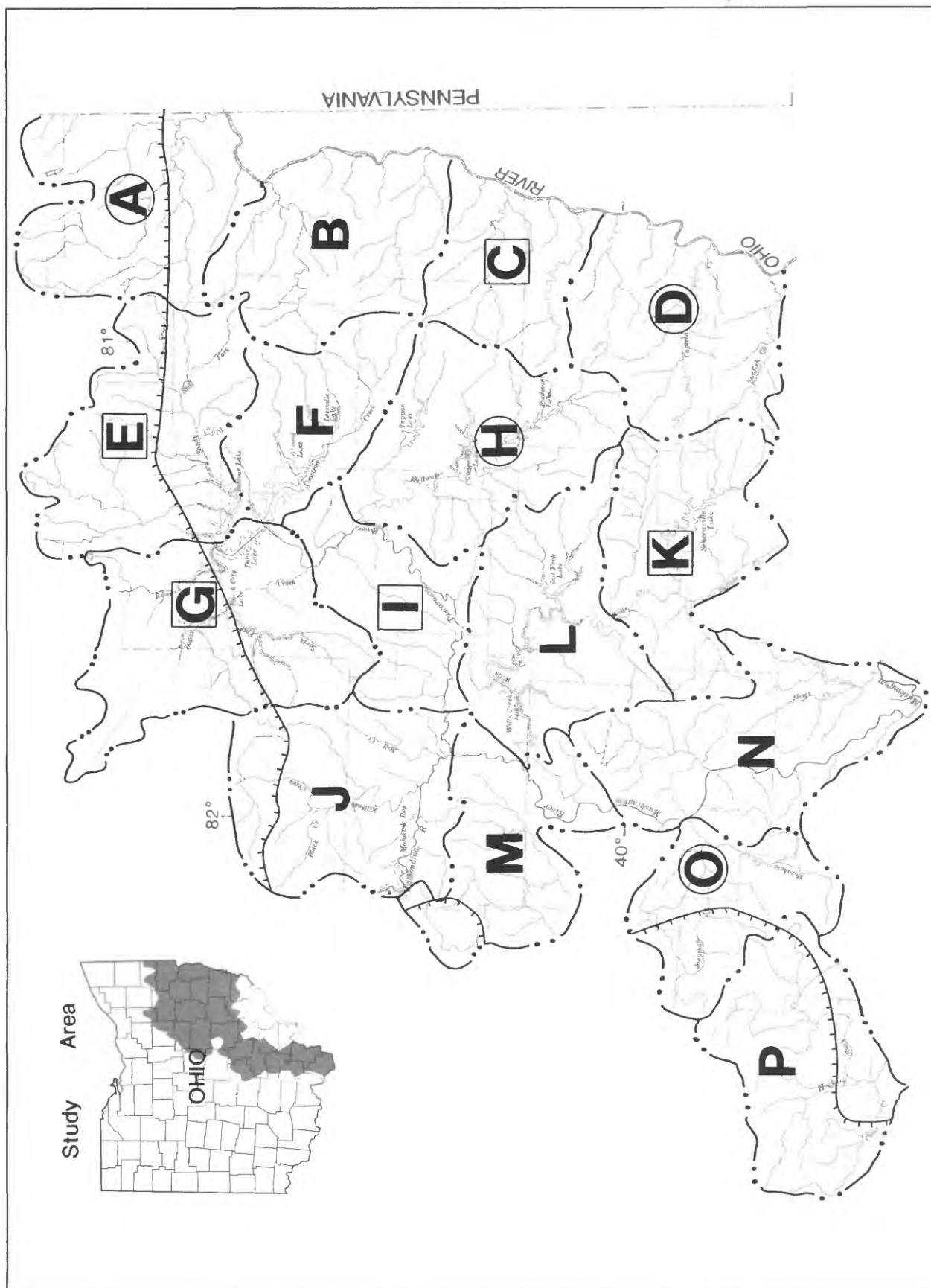
Included in the report are descriptions of the geologic and hydrologic settings of the 10 drainage basins selected for the synoptic data collection.

This report also presents a discussion of water-quality data collected during the entire study period (1985-91). Included is a summary of streamwater-quality data (508 samples collected from the 41-site long-term network in 21 basins) and of ground-water-quality data (69 samples collected in 16 basins).

Description of Study Area

The study area includes all or part of 29 counties in the coal-mining region of eastern Ohio (fig. 1). The 21 drainage basins that make up the study area include the areas of the most intensive surface and subsurface mining activities in the State. The 10 basins are listed in table 1 and are shown by their respective basin identifiers (A, B, and so forth) in figure 1. The basin designations follow the usage of the ODNR. The 10 basins selected for the 1989-91 synoptic data collection were Conotton Creek (basin F), Lower Wills Creek (basin L), Upper Hocking River (basin P), Yellow and Cross Creeks (basin B), Walhonding River (basin J), Upper Muskingum River (basin M), Middle Muskingum River (basin N), Middle Hocking River (basin Q), Leading Creek (basin QQ), and Lower Raccoon Creek (basin S).

The boundary between Pleistocene glaciation on the northwest and the unglaciated terrain on the southeast is shown in figure 1. Most of the study area lies in the unglaciated Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province (Fenneman, 1938, p. 283). The Ohio Department of Natural Resources, Division of Geological Survey, designates the region "Glaciated Plateau" and "Unglaciated Plateau." Local relief is from 100 to 200 ft along the glaciated



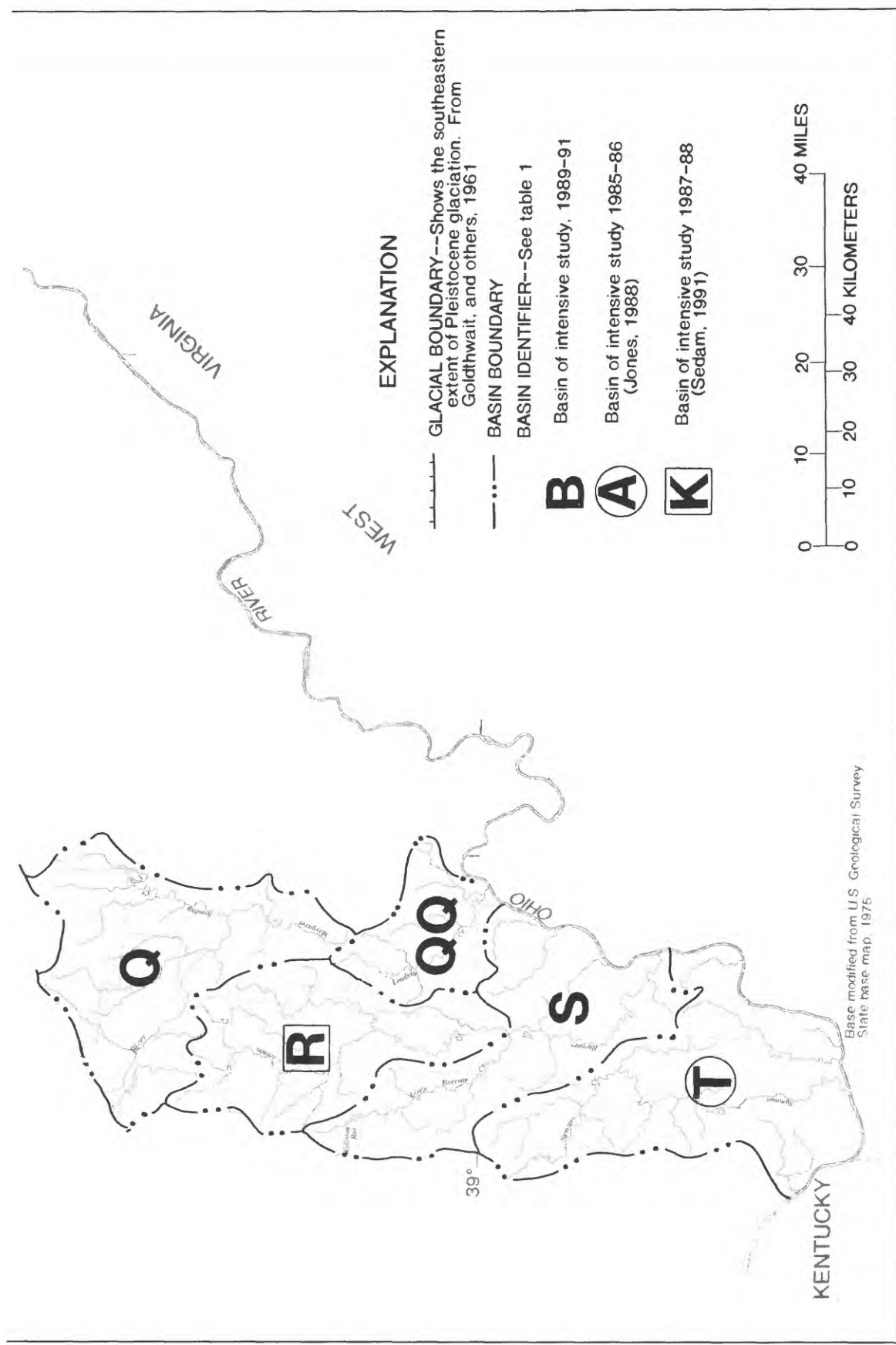


Figure 1.--Locations of study basins.

Table 1. Study basins and assigned identifiers

Basin identifier (fig. 1)	Basin name
A	Little Beaver Creek
B	Yellow and Cross Creeks
C	Short and Wheeling Creeks
D	McMahon, Captina, and Sunfish Creeks
E	Sandy Creek
F	Conotton Creek
G	Middle Tuscarawas River and Sugar Creek
H	Stillwater Creek
I	Lower Tuscarawas River
J	Walhonding River
K	Upper Wills Creek
L	Lower Wills Creek
M	Upper Muskingum River
N	Middle Muskingum River
O	Moxahala Creek
P	Upper Hocking River
Q	Middle Hocking River
QQ	Leading Creek
R	Upper Raccoon Creek
S	Lower Raccoon Creek
T	Symmes, Ice, and Indian Guyan Creeks

western and northern fringes of the study area but is as much as 500 ft in places near the Ohio River Valley. In places, the land surface is flat, such as along the major streams and small areas of the upland surface.

Four generalized land-use categories in eastern Ohio are shown in figure 2. The study area is mostly composed of an uneven distribution of forest and cropland or pasture. The largest urban area (fig. 2) is around the city of Canton in the north. Most of the areas depicted as "mining" within the study area represent surface mining of coal, although sand, gravel, and rock are extracted in some locations.

The bedrock of southeastern Ohio is a sedimentary sequence that consists mainly of shales, sandstones, and limestone ranging from Mississippian to Permian in age. The study area is underlain mostly by a complex repetitive succession of shales, sandstones, limestones, coals, and clays of Pennsylvanian age. The older rocks crop out to the west except where covered by glacial drift, and the younger rocks crop out to the east. The outcrop pattern in the study area trends north-northeast (fig. 3). Regional dip of the rocks is to the southeast, about 30 ft/mi, toward the Appalachian Basin and is modified locally by several low structural features (Lamborn, 1951, p. 13). The lithology and stratigraphic position of the Pennsylvanian formations is shown in figure 4. The designation of the Pottsville, Allegheny, Conemaugh, and Monongahela as "Formations" follows the usage of the USGS. In Ohio, these formations are generally classified as "groups." Collins' review of Ohio geology (1979) explains the criteria used to classify the Mississippian, Pennsylvanian, and Permian Systems in Ohio.

On the basis of formation thickness, Stout and others (1943) noted that the Pottsville contains 42 percent sandstone; the Allegheny, 40 percent; the Conemaugh, less than 30 percent; and the Monongahela, only about 15 percent. Conversely, the carbonate content

increases from older to younger Pennsylvanian rocks; therefore, the Monongahela Formation contains more carbonate than the Allegheny Formation. Based on studies such as Lamborn's (1951, p. 22-26), rocks of the Lower Pennsylvanian in Ohio, except for the coals, are largely of marine origin but are non-marine in the Upper Pennsylvanian. The zone of transition extends from about the middle of the Allegheny into the Conemaugh Formation (Razem and Sedam, 1985, p. 9)

Some 24 beds of mineable coal (thickness, 14 in. or more) in 32 counties made up an original reserve of 46 billion tons of coal in Ohio (Brant and Delong, 1960, p. 3). Most of the coal mined is from the Allegheny and Monongahela Formations. The coals listed in figure 4 make up about 92 percent of the State's original reserves. On the basis of 1989 data, the five leading producers, in order, were the Pittsburgh (No. 8), Meigs Creek (No. 9), Clarion (No. 4A), Middle Kittanning (No. 6), and Lower Kittanning (No. 5) coals, which amounted to about 80 percent of the State's total coal production (Lopez, 1991).

The location of the study area in relation to the principal aquifers in Ohio is shown in figure 5. Much of the northwestern half of the study area is underlain by strata categorized as sandstone aquifers (fig. 5). This area generally coincides with outcrop areas of the Pottsville and Allegheny Formations. Ground-water-resource maps published by ODNR show possible well yields ranging from 5 to 25 gal/min for this area. In general, the various sandstone members in the lower part of the Pennsylvanian system are the aquifers. Similar maps for the rest of the study area show that well yields are from 0 to 5 gal/min, which are typical of the shaly sandstone and carbonate rocks that underlie the area (fig. 5). The area corresponds approximately to the outcrop area of the Conemaugh and Monongahela Formations.

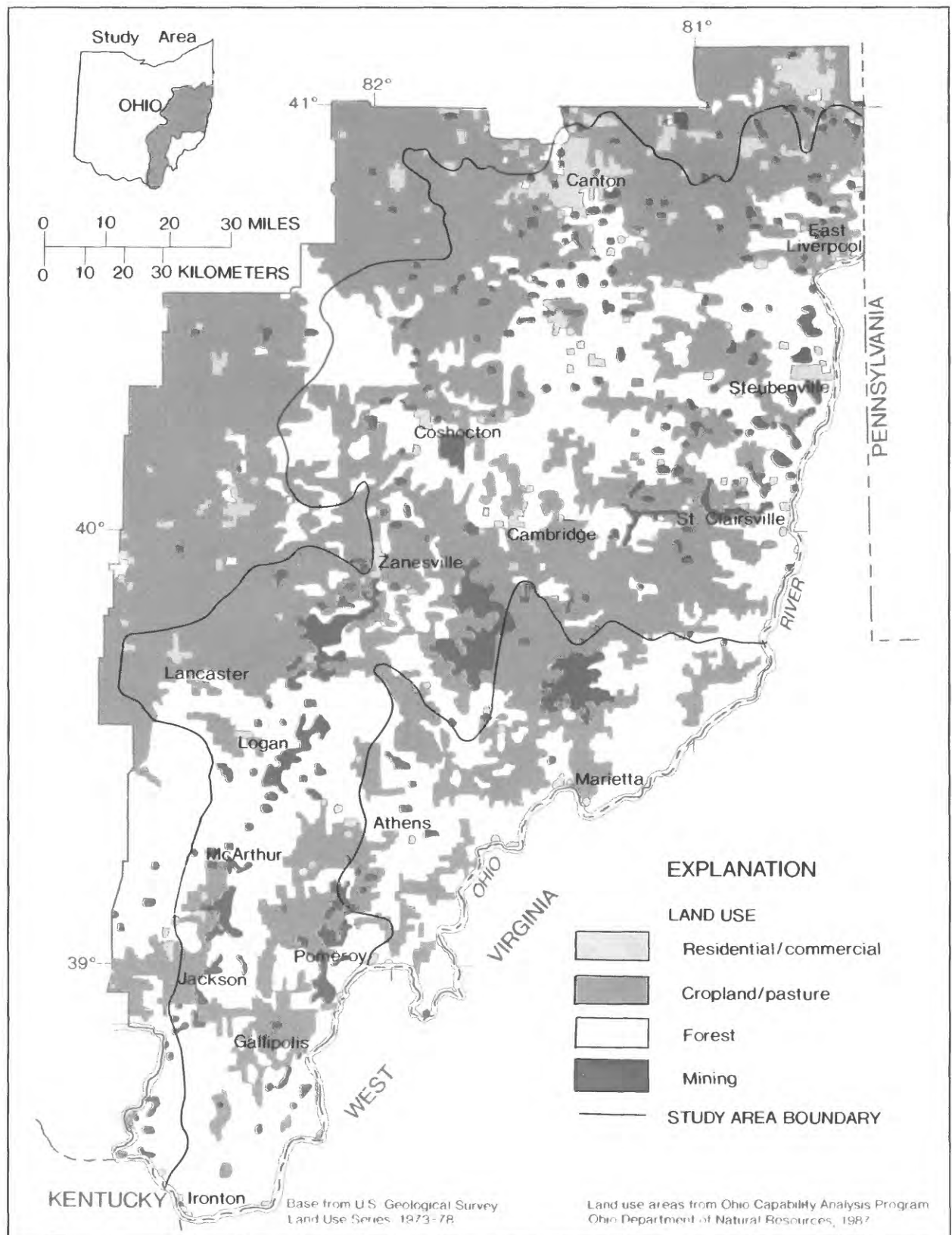


Figure 2.—Land use in eastern Ohio.

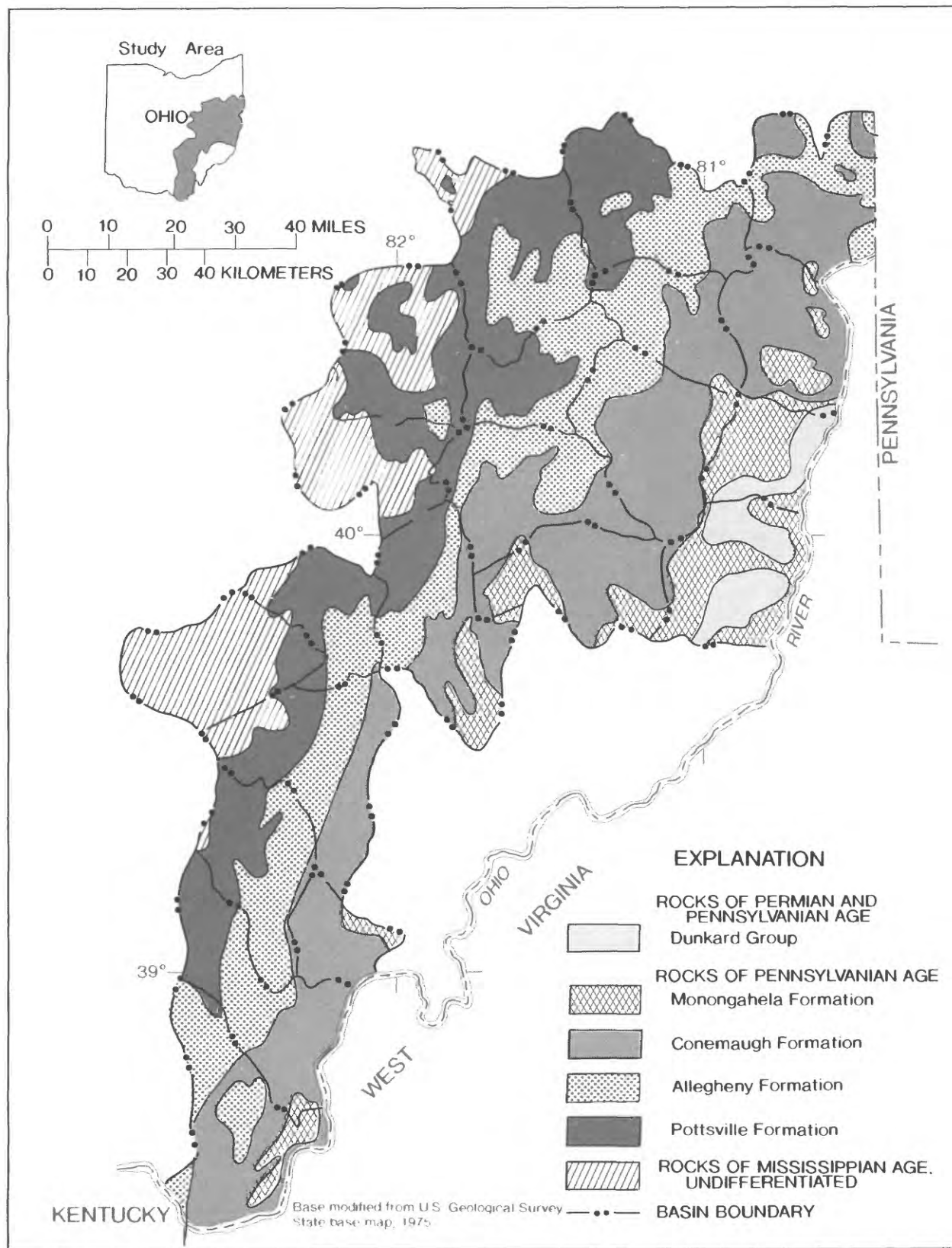


Figure 3.—Generalized geology of study area. (Modified from Collins, 1979, fig. 1.)

System	Group, Formation		Description	Important coal beds
Permian	Dunkard Group	Greene	Mostly red shales and limestones, localized coals, and sandstone bodies. Areas of occurrence are small.	No. 12 Washington
Pennsylvanian and Permian		Washington	Sandstones, shales, and minor coals. Sandstones are typically micaceous, fine to medium grained, and have thin conglomeratic zones. Locally, sandstones may be massively developed.	
Pennsylvanian	Monongahela		Important coal-bearing strata and associated beds of clay, shale, sandstone, and limestone. Sandstones tend to be fine to medium grained, micaceous, and patchy in development. Compared to other Pennsylvanian units, the Monongahela has a smaller proportion of sandstone and a larger proportion of limestone. Limestones tend to be marly, freshwater types. Secondary porosity along fractured surfaces is well developed locally.	No. 11 Waynesburg No. 10 Uniontown No. 9 Meigs Creek No. 8 Pittsburgh
	Conemaugh		Thick repetitious succession of shales and patchy sandstones interspersed with thin, discontinuous coals and clays and widespread limestones. The lower limestones are of marine origin, whereas those in the upper part are marly, freshwater types. Secondary porosity along fractured surfaces is common.	
	Allegheny		Repetitious succession of important coal-bearing strata interspersed with several fine- to coarse-grained, massive, cross-bedded sandstones and thin, persistent limestones. Sandstones, though widespread, have considerable local lateral variation. Solution cavities are developed locally in the limestones.	No. 7 Upper Freeport No. 6A Lower Freeport No. 6 Middle Kittanning No. 5 Lower Kittanning No. 4A Clarion No. 4 Brookville
	Pottsville		Succession of sandstone, shales, clays, coals, and thin limestones. Locally, sandstones are open-textured, conglomeratic, massive, cross-bedded, and commonly are found as deposits filling old channels in eroded terrains. Lateral gradations include shale and coal. Non-depositions during the Early Pennsylvanian precluded the development of the basal conglomerate of the Sharon Member in most of southeastern Ohio.	No. 3 Lower Mercer No. 2 Quakertown No. 1 Sharon
			Thin, discontinuous zone of impure nodular iron ore and ferruginous sandstone. The unit marks the disconformity between Mississippian and Pennsylvanian strata. Age of the deposit is conjectural, but, generally, it is included at the base of the Pottsville Formation.	
Mississippian	Undifferentiated		Variable sequence of sandstones and shales; Maxville Limestone is present in patches at the top. In places, various units (Berea Sandstone, Black Hand Sandstone Member of the Cuyahoga Formation, and Logan Formation) are conglomeratic and sandstones are massive. Lateral and vertical gradation to siltstone and shale is common. Ground-water potential is limited to extreme western areas. Eastward, the section contains saltwater. To the north, post-Mississippian erosion has removed the section.	

Figure 4. Generalized geologic column for southeastern Ohio, including relative position of important coal beds (Razem and Sedam, 1985).

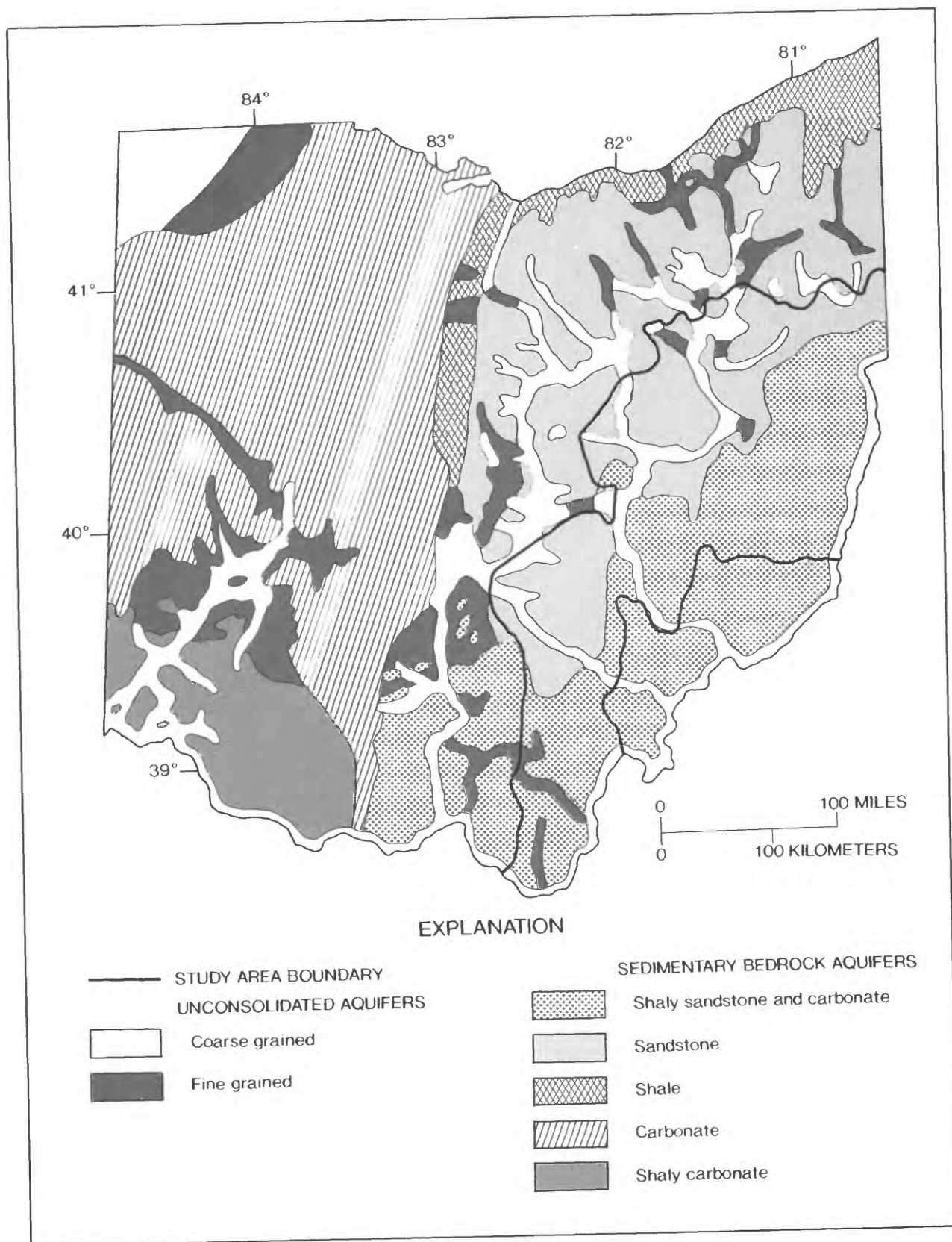
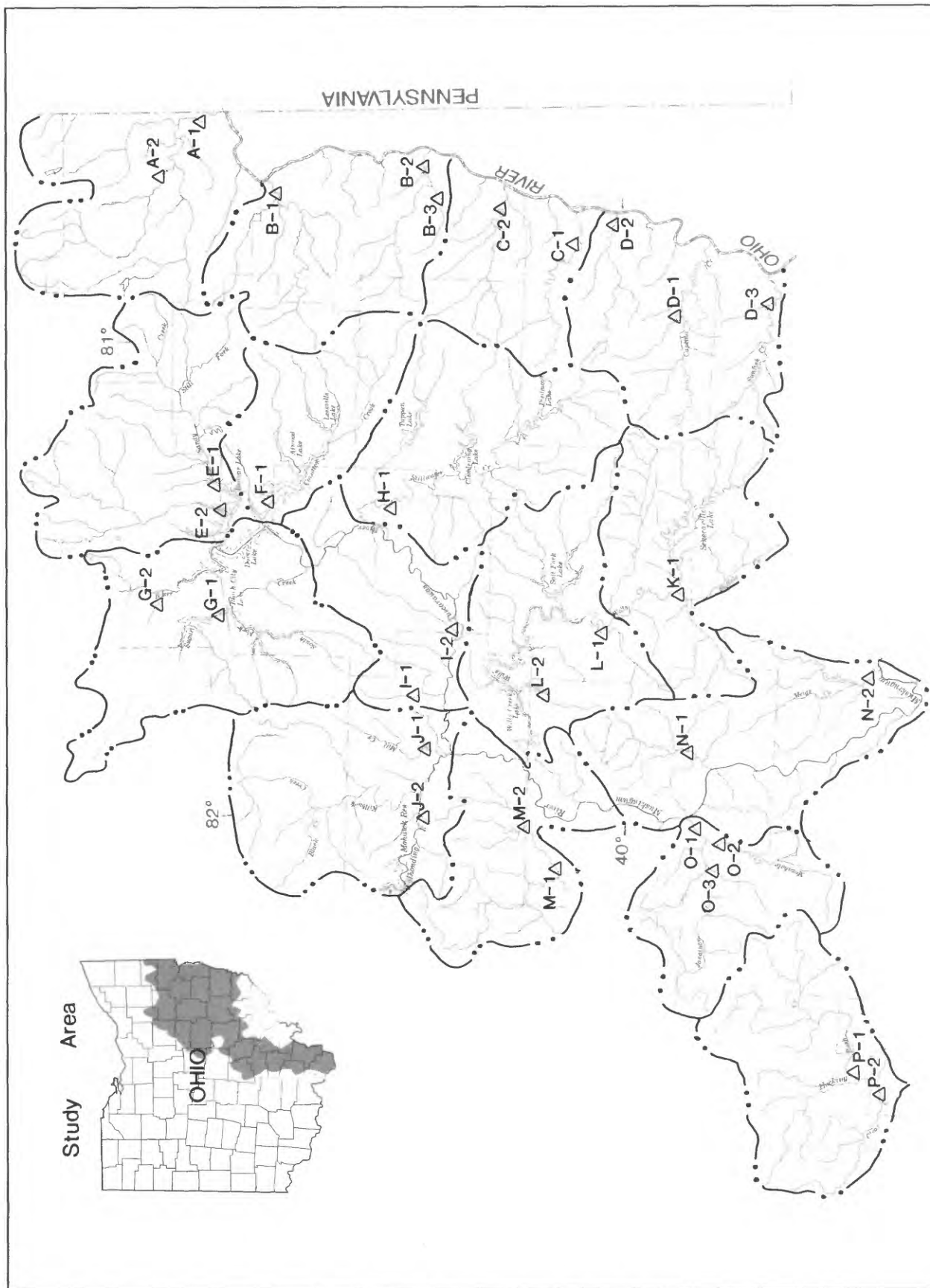


Figure 5.--Principal aquifers in Ohio. (Modified from Sedam and others, 1985, fig.1.)



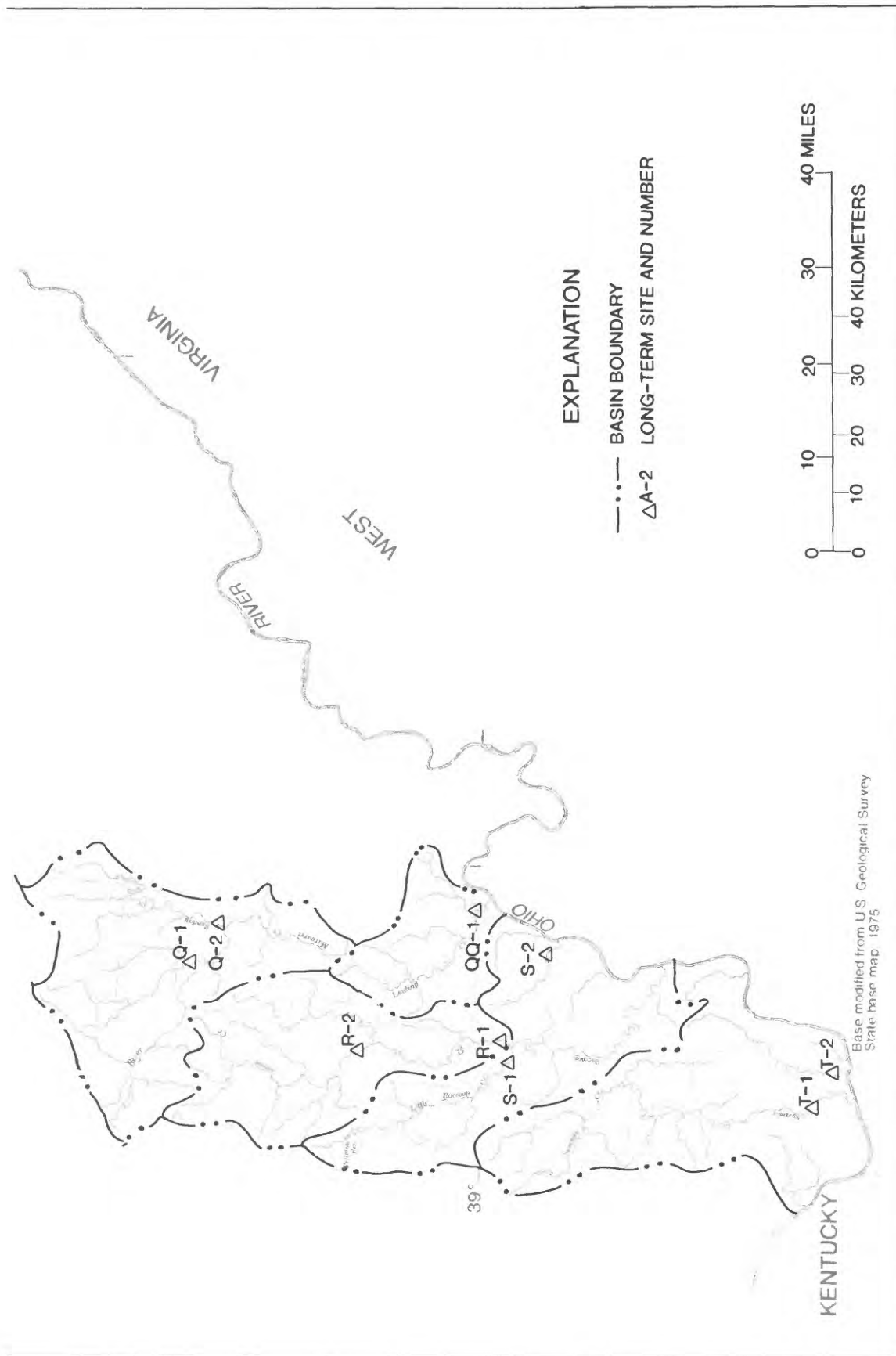


Figure 6.--Locations of long-term streamwater sites.

Superimposed on the northern edges of the study area and extending along alluvial channels cut in the bedrock are the unconsolidated aquifers of glacial and postglacial origin. The coarse-grained aquifers commonly yield 100 to 500 gal/min compared to 25 to 50 gal/min for the fine-grained types (Sedam and others, 1985, p. 342). Some of the coarse-grained aquifers consist of permeable sand and gravel deposits along modern streams; these are sometimes referred to as "watercourse" aquifers, and several such aquifers are present in the study area. Fine-grained aquifers include deposits in buried valleys, alluvium along modern streams, and permeable lenses in glacial drift.

Climatic data for the period 1931-80 have been published in map form for Ohio (Harstine, 1991). This map shows that, although much of the study area receives 40 to 42 in. of precipitation annually, two cell-like areas (one south of Zanesville, the other north of Coshocton) receive only about 36 in. annually. The weather pattern for the period of this report (1989-91) varied considerably. The years 1989 and 1990 were unusually wet, and rainfall amounts were considerably above normal for most of the study area. In contrast, in 1991, rainfall amounts were considerably below normal and were similar to those in the dry years of 1987 and 1988. Based on rainfall summaries compiled by the ODNR, Division of Water (Cashell, 1988-91). The rainfall deficiency for 1988 and 1991 was 2 to 10 in. in eastern Ohio; in the same area in 1990 and 1991, rainfall exceeded normal amounts by 2 to 15 in.

Acknowledgments

The author wishes to thank Vanessa Tolliver and Bill Sterling of the ODNR, Division of Reclamation, for their information concerning locations of coal-mining activity. The author also appreciates the cooperation of various commercial and municipal entities as

well as individual property owners who allowed access to their wells or other data-collection sites.

METHODS OF STUDY

Locations of the long-term streamwater sites are shown in figure 6. In 1987, Leading Creek basin (fig. 1, basin QQ) was added, expanding the study area to 21 basins. With the addition of site QQ-1, the network was expanded to 41 long-term streamwater sites. Long-term streamwater sites were selected to represent general water quality in the basins under baseline conditions.

Streamflow measurements and collection of water samples were generally done for all 41 long-term sites twice annually, in late spring or early summer and in early autumn during low flow from 1989 through 1991. (For this study, low flow was a rate of flow exceeded 70 percent or more of the time.) Because of high flow in autumn 1990, this sampling was postponed until spring at some sites. Stage levels at several USGS continuous-record streamflow-gaging stations in southeastern Ohio were monitored to determine when streamflow would be appropriate for sampling.

During the final phase of the project (1989-91), 10 drainage basins were selected for a one-time synoptic data collection. The short-term streamwater sites for synoptic study were chosen for geographic coverage of the basins. Streamflows were measured and water samples were collected at 45 short-term streamwater sites in 9 of the 10 drainage basins. In autumn 1989, work was done at 15 sites in Conotton Creek, Lower Wills Creek, and Upper Hocking River Basins (fig. 1, basins F, L, and P). Streamflows were measured and water samples were collected in spring 1991 at 15 sites in the Yellow and Cross Creeks, Walhonding River, and Upper Muskingum River basins (fig. 1, basins B, J, and M). In autumn 1991, work was done at 15 sites in the

Middle Muskingum River, Middle Hocking River, and Leading Creek basins (fig. 1, basins N, Q, and QQ). Because of two previous studies in Raccoon Creek basin (Wilson, 1985, 1988), no short-term streamwater sites were chosen in the Lower Raccoon Creek basin (fig. 1, basin S) for synoptic data collection.

In the 10 drainage basins selected for synoptic data collection, short-term ground-water sites were chosen to represent water quality in shallow, productive aquifer(s) in the basin. Ground-water-resource maps and drillers' logs on file at ODNR, Division of Water, were used to locate prospective areas of shallow ground water and well locations that could be used for sampling. Much of the study area is served by water wells supplying small amounts for domestic purposes. The wells range from shallow (less than 50 ft) to deep (more than 50 ft), but most produce water from bedrock. The investigation focused on aquifers that could serve as a public or industrial water supply. Shallow, productive aquifers were found in 7 of the 10 basins. Water levels were measured when possible, and water-quality samples were collected from 28 wells during late summer and autumn 1991 as part of the one-time synoptic sampling.

Measurements of streamflow, specific conductance, pH, temperature, alkalinity, and acidity were made at each streamwater site. Streamflow was measured by the methods described in Rantz and others (1982). Specific conductance, pH, temperature, and alkalinity were determined by the methods discussed in Fishman and Friedman (1989). For water samples having pH greater than 6.5, alkalinity was determined by fixed-endpoint titration. Acidity was determined on water samples having pH less than 5.0 by use of the hot-peroxide-treatment method (American Public Health Association, 1985). If the pH of the stream was 5.0 to 6.5, both alkalinity and acidity were determined. Water samples for chemical analysis were collected according to the equal-transit-rate/equal-width-increment

method (U.S. Geological Survey, Office of Water Data Coordination, 1977) for all streams deeper than 0.5 ft. For streams shallower than 0.5 ft, grab samples were collected. Samples were analyzed for concentrations of dissolved sulfate and for total recoverable and dissolved iron, manganese, and aluminum at the USGS National Water Quality Laboratory in Arvada, Colo.

Ground-water sampling procedures consisted of selecting wells in current use, inspecting the system to avoid treatment devices, and allowing the well to pump long enough to ensure that the water sample was representative of the formation. Ground water was considered to be representative of the formation when water-quality properties (specific conductance, pH, and temperature) stabilized. Field measurements were made in a sample bucket for specific conductance, pH, dissolved-oxygen concentration, and temperature according to the methods discussed in Fishman and Friedman (1989). Alkalinity was determined by fixed-endpoint titration, and bicarbonate concentration was calculated from the alkalinity. Water levels were measured by use of a steel tape when possible.

Laboratory analyses of ground-water samples were done at the USGS National Water Quality Laboratory and included concentrations of the following constituents:

- Total recoverable and dissolved aluminum
- Dissolved sulfate
- Total recoverable and dissolved iron
- Total recoverable and dissolved manganese
- Dissolved silica
- Dissolved calcium
- Dissolved magnesium
- Dissolved sodium
- Dissolved chloride
- Dissolved potassium
- Dissolved organic carbon
- Total dissolved solids (residue on evaporation at 180°C).

Total hardness was calculated from the sums of the concentrations of calcium and magnesium, and noncarbonate hardness was calculated by subtracting the alkalinity from total hardness. In some sample analyses, the amounts of dissolved iron, aluminum, or manganese exceeded the total recoverable concentrations found for the same constituent. This difference was generally less than 10 percent, and is not unusual given the analytical accuracy.

WATER QUALITY OF THE STUDY AREA, 1989-91

Local geology and the extent to which the area has been mined are two important factors that can affect water quality of streams and local aquifers within the study area. Acid mine drainage can cause degradation of streamwater quality in mined areas and can be a potential threat in unmined areas. In addition to acid mine drainage, increased concentrations of constituents that can adversely affect aquatic biota of the stream may be present as a result of mining (Wangness, 1982).

The principal source of acid mine drainage is the widespread presence of sulfur-bearing minerals, such as pyrite and marcasite, in coal-bearing strata. Mining exposes these minerals to air and water. This exposure leads to chemical changes, such as the oxidation of sulfides to sulfates, that commonly result in increased acidity to surface runoff. In some places, limestone is present and serves as a buffer by neutralizing some of the acidic drainage.

Streamwater at Long-Term Sites, 1989-91

The 41 long-term sites were sampled 6 times between August 1989 and October 1991 to obtain a total of 246 samples. Specific conductance, pH, alkalinity, and (or) acidity; concentrations of dissolved sulfate, iron, manganese, and aluminum; and concentrations of total recoverable iron, manganese, and alumi-

num were used to assess the baseline water quality of major streams in the study area. Ranges and medians of these properties and constituents are summarized in table 2 for the 41 long-term surface-water sites. Site identifiers for lower- and upper-range values correspond to the long-term site-sampling locations shown in figure 6. Ranges and percentiles (including medians) of constituents are presented graphically in figure 7, and detailed analyses are given in table 31 at the back of the report.

Specific conductance, the ability of a substance to conduct electricity, increases as ion concentration increases (Hem, 1989). Among the 246 samples from the long-term site network, median specific conductance was 792 $\mu\text{S}/\text{cm}$ and the range was 270 $\mu\text{S}/\text{cm}$ (at site T-1) to 5,170 $\mu\text{S}/\text{cm}$ (at site QQ-1). Specific conductance was greater than 1,000 $\mu\text{S}/\text{cm}$ for all six sampling rounds at nine sites: B-2, B-3, C-1, C-2, K-1, M-2, N-2, O-1, and O-2.

The pH of natural waters is a measure of the acid-base equilibrium achieved by the various dissolved compounds, salts, and gases. Values of pH less than 7 indicate acidic solutions, and those greater than 7 indicate alkaline solutions. Values of pH in the acidic range were found at sites O-1, O-2, Q-1, and Q-2. Of the 24 samples collected in 1989 to 1991 at these sites, the pH ranged from 2.7 to 6.7 and the median was 3.4. For these 24 samples, specific conductances were high, ranging from 809 to 2,180 $\mu\text{S}/\text{cm}$. The high specific conductance and low pH found at these sites are characteristic of water affected by coal mining. In contrast, pH of samples from sites B-2, B-3, C-1, C-2, K-1, M-2, and N-2 was in the alkaline range, although specific conductances at these sites were greater than 1,000 $\mu\text{S}/\text{cm}$. For these sites, acid mine drainage may have been neutralized by interaction with carbonates present as limestone or calcareous shales and sandstones. In the set of 246 samples collected at the long-term streamwater sites, 140 of the pH values were 7.8 or greater.

Table 2. Ranges and medians for selected water-quality characteristics for long-term streamwater sites in active coal-mining areas of Ohio, 1989-91

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 6)	Median
Specific conductance, in µS/cm-----	270 to 5,170	T-1; QQ-1	792
pH-----	2.7 to 9.1	O-2; A-1	7.8
Alkalinity, in mg/L as CaCO ₃ -----	1 to 391	Q-1; K-1	^b 116
Acidity, in mg/L as CaCO ₃ -----	6 to 183	O-1; Q-2	^b 118
Sulfate, dissolved, in mg/L as SO ₄ -----	13 to 2,100	J-1; QQ-1	200
Aluminum, total, in µg/L as Al-----	10 to 16,000	D-3, M-1; Q-1	240
Aluminum, dissolved, in µg/L as Al-----	<10 to 17,000	Several sites; Q-1	30
Iron, total, in µg/L as Fe-----	20 to 54,000	D-3; Q-2	610
Iron, dissolved, in µg/L as Fe-----	<10 to 53,000	Several sites; Q-2	60
Manganese, total, in µg/L as Mn-----	10 to 15,000	D-3; O-2	355
Manganese, dissolved, in µg/L as Mn-----	<10 to 17,000	B-1, D-3; O-2	295

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which acidity or alkalinity could be measured (see "Methods," p. 14).

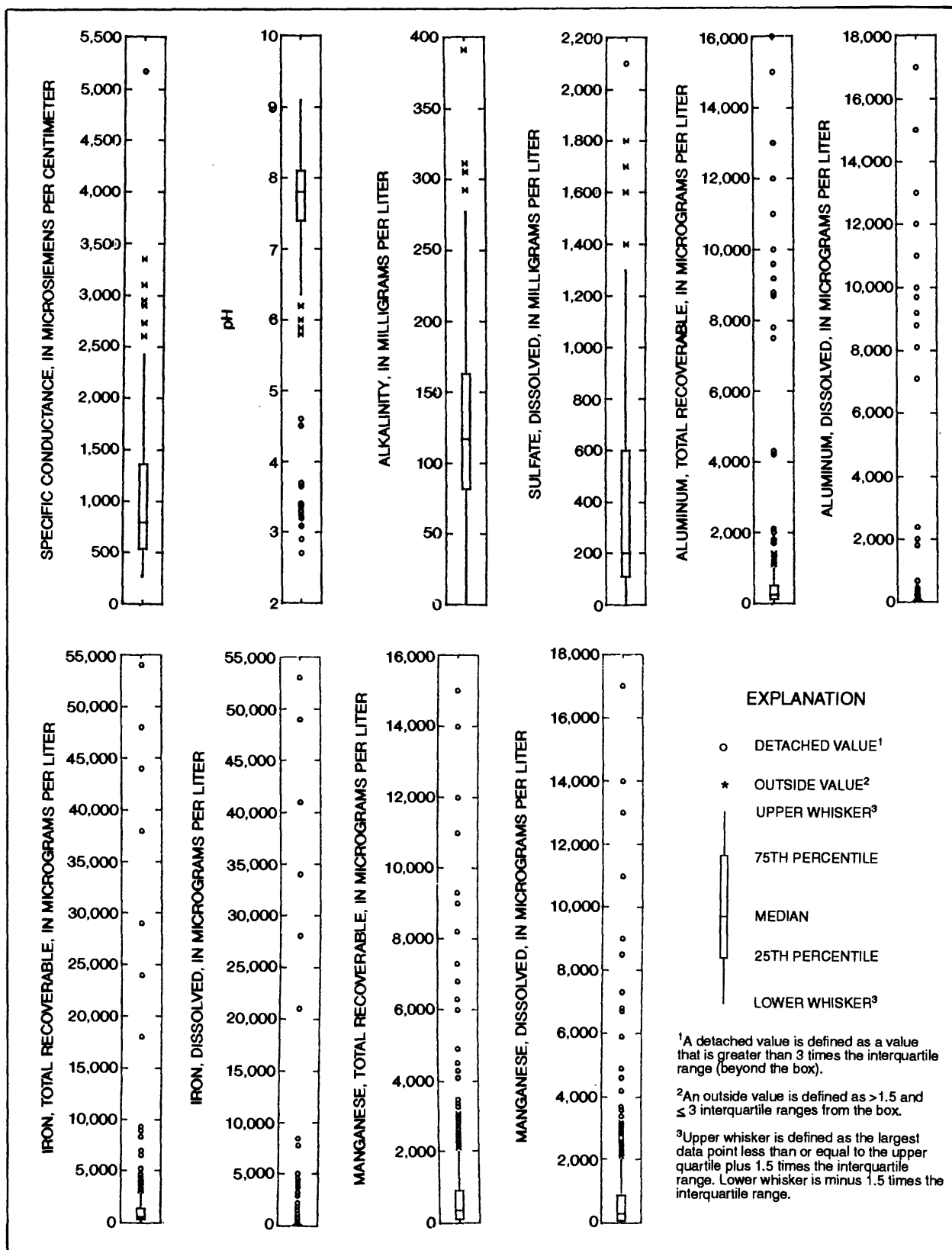


Figure 7.—Ranges, percentiles, and medians of constituents in 246 samples collected at 41 long-term streamwater sites in active coal mining areas of Ohio, 1989-91.

"Buffering capacity," the ability to neutralize additions of acids or bases without a change in pH, is controlled by the concentration of alkalinity and acidity in water (U.S. Environmental Protection Agency, 1986). The median alkalinity in the long-term network was 116 mg/L as CaCO_3 , and the range was 1 mg/L (at site Q-1) to 391 mg/L (at site K-1). Of the 246 samples collected, 21 had measurable acidity in the range of 6 mg/L as CaCO_3 (at site O-1) to 183 mg/L (at site Q-2) (table 31). High acidity is commonly associated with acid mine drainage.

The presence of high (greater than 250 mg/L) dissolved-sulfate concentrations can be an indicator of drainage from coal-mined areas. At four sites where pH was in the acidic range (sites O-1, O-2, Q-1, and Q-2), dissolved-sulfate concentrations also were high. Water from site QQ-1 had a wide range of dissolved-sulfate concentrations over the six samplings and had the highest dissolved-sulfate concentration of water from any site (2,100 mg/L); pH was greater than 7.0 at this site. Neutral or alkaline pH and high dissolved-sulfate concentrations were found at numerous other sites, including sites B-2, B-3, C-1, C-2, D-2, H-1, J-2, K-1, L-2, M-2, N-2, and O-3. Data have shown that sulfate concentrations, once elevated, do not return to their premining levels, even after reclamation (Pfaff and others, 1981; Hren and others, 1984).

Concentrations of dissolved aluminum, iron, and manganese are known to be elevated during and immediately after mining (Dyer and Curtis, 1977). Water from sites O-2 and Q-1 had consistently high concentrations of all three metals, water from site Q-2 had high concentrations of iron and manganese, and water from site O-1 had high concentrations of manganese. At all of these sites, pH was in the acidic range.

Aluminum is present only in small quantities in natural waters; however, waters whose pH is in the acidic range commonly have high

concentrations of aluminum (Ward and Wilmoth, 1968). Concentrations of dissolved aluminum ranged from less than 10 $\mu\text{g/L}$ (at several sites) to 17,000 $\mu\text{g/L}$ (at site Q-1); the median was 30 $\mu\text{g/L}$. The range of concentrations and the median concentration of total recoverable aluminum were similar to those for dissolved aluminum.

Concentrations of iron in water, though generally low, are strongly determined by the chemistry of the water (Hem, 1989). More than 1,000 to 2,000 $\mu\text{g/L}$ of soluble iron in streamwaters generally indicates acidic contamination from mine drainage or other sources (Ward and Wilmoth, 1968). The concentration of dissolved iron ranged from less than 10 $\mu\text{g/L}$ (at several sites) to 53,000 $\mu\text{g/L}$ (at site Q-2). Concentrations of total recoverable iron ranged from 20 $\mu\text{g/L}$ (at site D-3) to 54,000 $\mu\text{g/L}$ (at site Q-2). The median concentration for total recoverable iron (610 $\mu\text{g/L}$) was about 10 times greater than the median concentration for dissolved iron (60 $\mu\text{g/L}$).

Small amounts of manganese are present in dolomite and limestone as a substitute for calcium. More than 1,000 $\mu\text{g/L}$ of dissolved manganese can be present in streams receiving acid mine drainage (Hem, 1989). Concentrations of dissolved manganese ranged from less than 10 $\mu\text{g/L}$ (at sites B-1 and D-3) to 17,000 $\mu\text{g/L}$ (at site O-2); the median was 295 $\mu\text{g/L}$. The range of concentrations and the median concentration of total recoverable manganese were similar to those for dissolved manganese.

In most samples from the Middle Hocking River basin (fig. 1, basin Q) and Moxahala Creek basin (fig. 1, basin O), pH was in the acidic range, and high concentrations of the constituents commonly associated with acid mine drainage were present, including sulfate, iron, aluminum, and manganese. The remnants of past mining activities are clearly visible in the basins.

Ground Water

Coal-mining activity can affect the flow or quality of ground water in an aquifer, especially in shallow aquifers in the same small watershed that is being mined (Eberle and

Razem, 1985). Shallow sources of ground water are present in many places throughout the coal-bearing region of eastern Ohio; however, those that are capable of furnishing enough water for commercial or small industrial purposes are mostly restricted to a few major valleys. The unconsolidated alluvial deposits along the Ohio River valley are not included in the scope of this study.

In this study, the water quality of shallow, productive ground-water sources along Conotton Creek, Walhonding River, and Lower Wills Creek were evaluated (fig. 1, basins F, J, and L, respectively). Farther south, shallow, productive sources of ground water were evaluated in the Upper and Middle Hocking River drainage basins (fig. 1, basins P and Q) and Upper and Middle Muskingum River drainage basins (fig. 1, basins M and N).

Elsewhere, shallow ground-water sources, whether in bedrock or unconsolidated material, are localized and, compared with the types just described, yield relatively small quantities of water to wells. This situation is characteristic of ground water in Yellow and Cross Creeks, Leading Creek, and Lower Raccoon Creek basins (fig. 1, basins B, QQ, and S, respectively); therefore, ground-water quality was not evaluated in these basins.

Analyses and dates of sampling for 28 ground-water sites in 7 basins are included in the discussion of the individual basins that follows this section. Locations of the 28 sites are shown in figure 8, and ranges and medians of selected constituents and physical properties are listed in table 3. Summary statistics and box plots were not done for ground-water samples because only four wells were sampled in each basin.

Hardness of water is defined as its content of metallic ions that react to produce solid boiler scale (Camp, 1963). In freshwater, hardness is derived primarily from calcium and magnesium ions, which are naturally dissolved from carbonate rock by slightly acidic rainwater in contact with carbon dioxide. Sources of hardness related to human activity include abandoned and active mines (Jones, 1988).

Qualitative descriptions of hardness used in this report are based on the following classification (U.S. Environmental Protection Agency, 1986)

	Milligrams per liter as CaCO ₃
Description:	
Soft.....	0-75
Moderately hard.....	75-150
Hard	150-300
Very hard.....	>300

Carbonate hardness is considered equal to alkalinity and is chemically equivalent to the bicarbonates present (U.S. Environmental Protection Agency, 1986). If hardness exceeds alkalinity, the excess is reported as "noncarbonate hardness" (Hem, 1989). Ground water in the seven basins sampled had a median hardness of 292 mg/L as CaCO₃; the range was from 140 mg/L (at Cs-150 and Mu-50) to 730 mg/L (at At-70). Noncarbonate hardness ranged from 0 mg/L (at several sites) to 610 mg/L (at site At-70); the median was 59 mg/L.

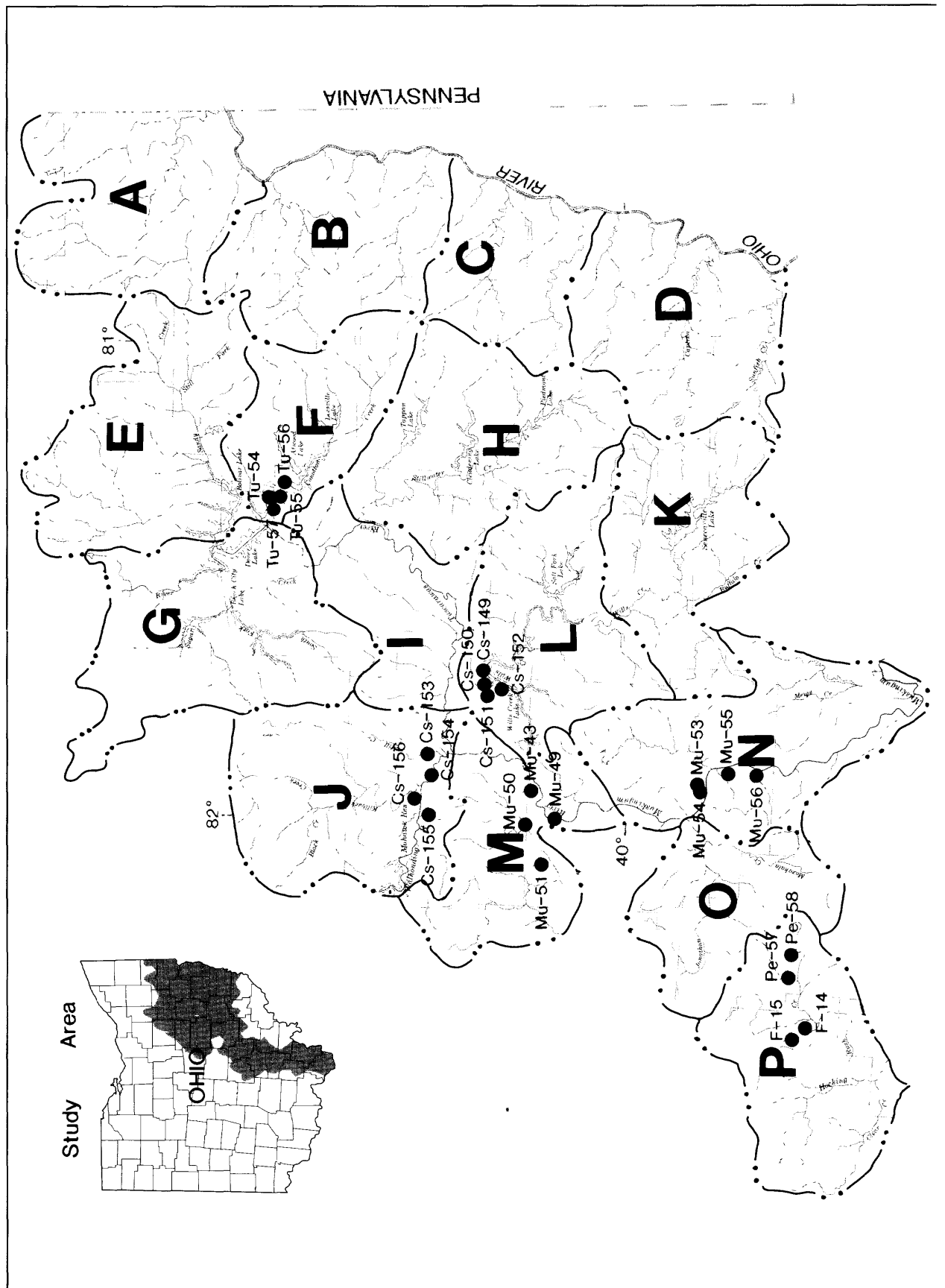
Ground-water samples were analyzed for concentrations of dissolved calcium, magnesium, sodium, and potassium. Calcium is a major constituent of many common rock minerals and is a major component of the solutes in most natural waters (Hem, 1989). For the 28 sites sampled, calcium concentrations ranged from 43 to 190 mg/L, and the median was 78 mg/L. Magnesium is dissolved from carbonate rocks. Magnesium concentrations ranged from 7 to 61 mg/L, and the median was 20 mg/L. In most natural waters, sodium con-

centrations are much greater than potassium concentrations. The U.S. Environmental Protection Agency (1986) recommends a limit of 20 mg/L of sodium in water for very restricted sodium diets and 270 mg/L for moderately restricted diets. The range of sodium concentrations for the 28 sites sampled was 3.5 to 120 mg/L, and the median was 18 mg/L, whereas the range of potassium concentrations was 0.6 to 6.1 mg/L, and the median was 1.8 mg/L.

A Piper diagram (fig. 9) shows the ionic character of the ground water. Water types at most of the 28 sites are similar, except at two wells from the Middle Hocking River basin. Water from one well in this basin stands out in the Piper diagram by having less calcium and magnesium and more sodium plus potassium than the other wells in the study area. Indeed, water from well Hk-54 had the highest concentration of sodium in the study area (120 mg/L) (table 3). Water from another well in the Middle Hocking River basin stands out in the Piper diagram (fig. 9) by having a high concentration of sulfate and a low concentration of carbonate plus bicarbonate. The concentration of dissolved sulfate in water from well At-70 was 470 mg/L, substantially higher than the median of 48 mg/L. At this well, near-maximum concentrations were found for other constituents—hardness (730 mg/L as CaCO_3), noncarbonate hardness (610 mg/L), calcium (190 mg/L), magnesium (61 mg/L), and dissolved solids (822 mg/L).

Dissolved organic carbon (DOC) is a measure of the organic matter present in aqueous solution. Thurman (1985) lists 0.7 mg/L as an approximate concentration for DOC typical of natural ground water. In the seven basins sampled in this study, DOC concentrations ranged from 0.6 to 3.1 mg/L; the median was 0.8 mg/L.

As discussed in the streamwater section, high concentrations of iron, aluminum, and manganese may be associated with waters affected by acid mine drainage. High concentrations of aluminum are typically associated with low pH. Because aluminum is virtually insoluble at the range of pH found for the 28 ground-water samples (7.0 to 7.9), dissolved aluminum concentrations ranged from less than 10 to 70 $\mu\text{g/L}$, and the median was less than 10 $\mu\text{g/L}$. Total recoverable aluminum concentrations were slightly higher. In contrast, under reducing conditions, iron is moderately soluble in water; concentrations of 50 mg/L have been found in water having a pH between 6 and 8 (Hem, 1989). Dissolved-iron concentrations ranged from less than 3 to 8,700 $\mu\text{g/L}$; the median was 275 $\mu\text{g/L}$. Concentrations of total recoverable iron were higher (median of 1,005 $\mu\text{g/L}$). Ground water can contain more than 1.0 mg/L of manganese under some circumstances (Hem, 1989). In the 28 ground-water samples, however, median manganese concentrations for dissolved (70 $\mu\text{g/L}$) and total (59 $\mu\text{g/L}$) manganese were not excessive.



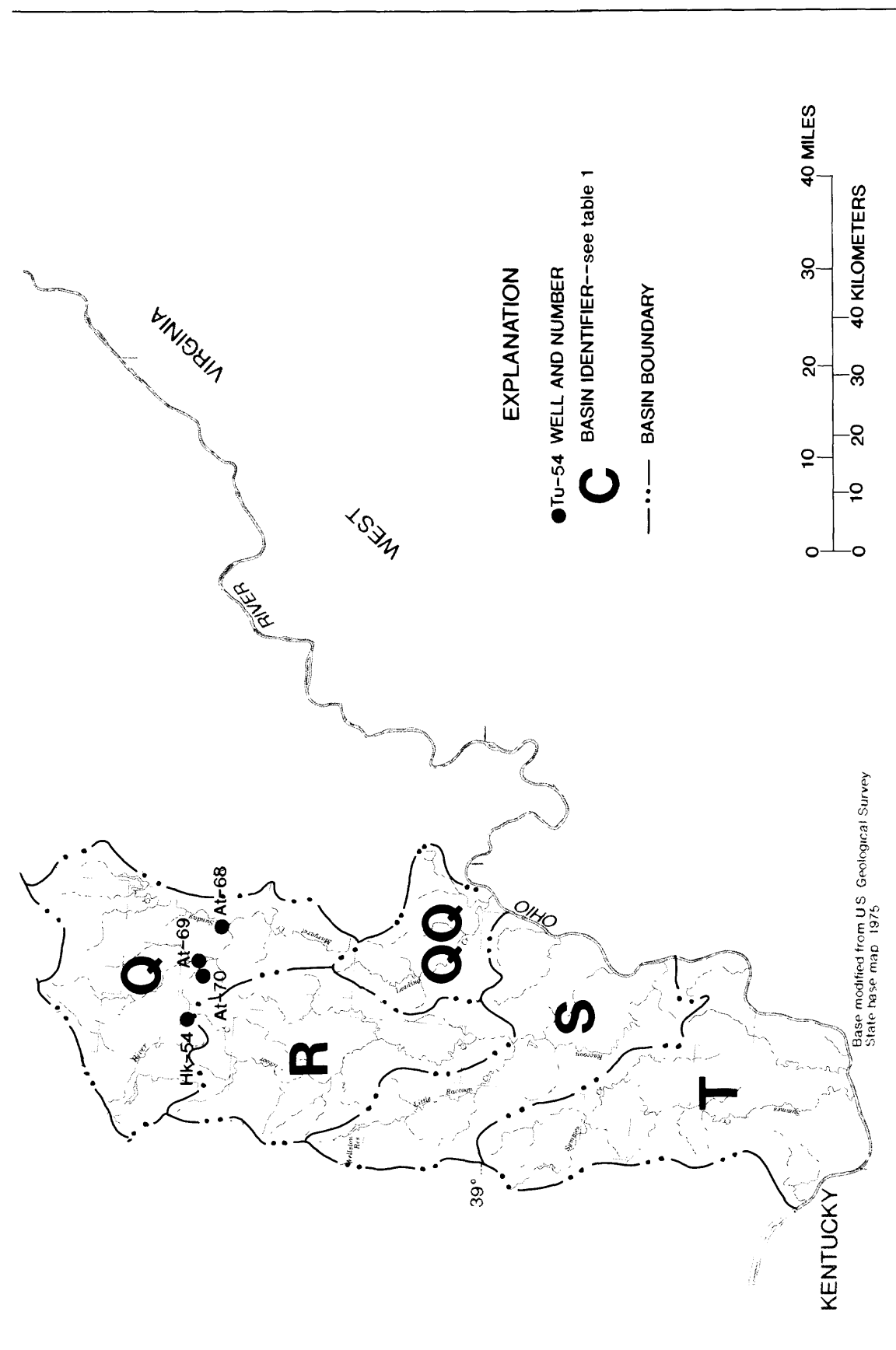


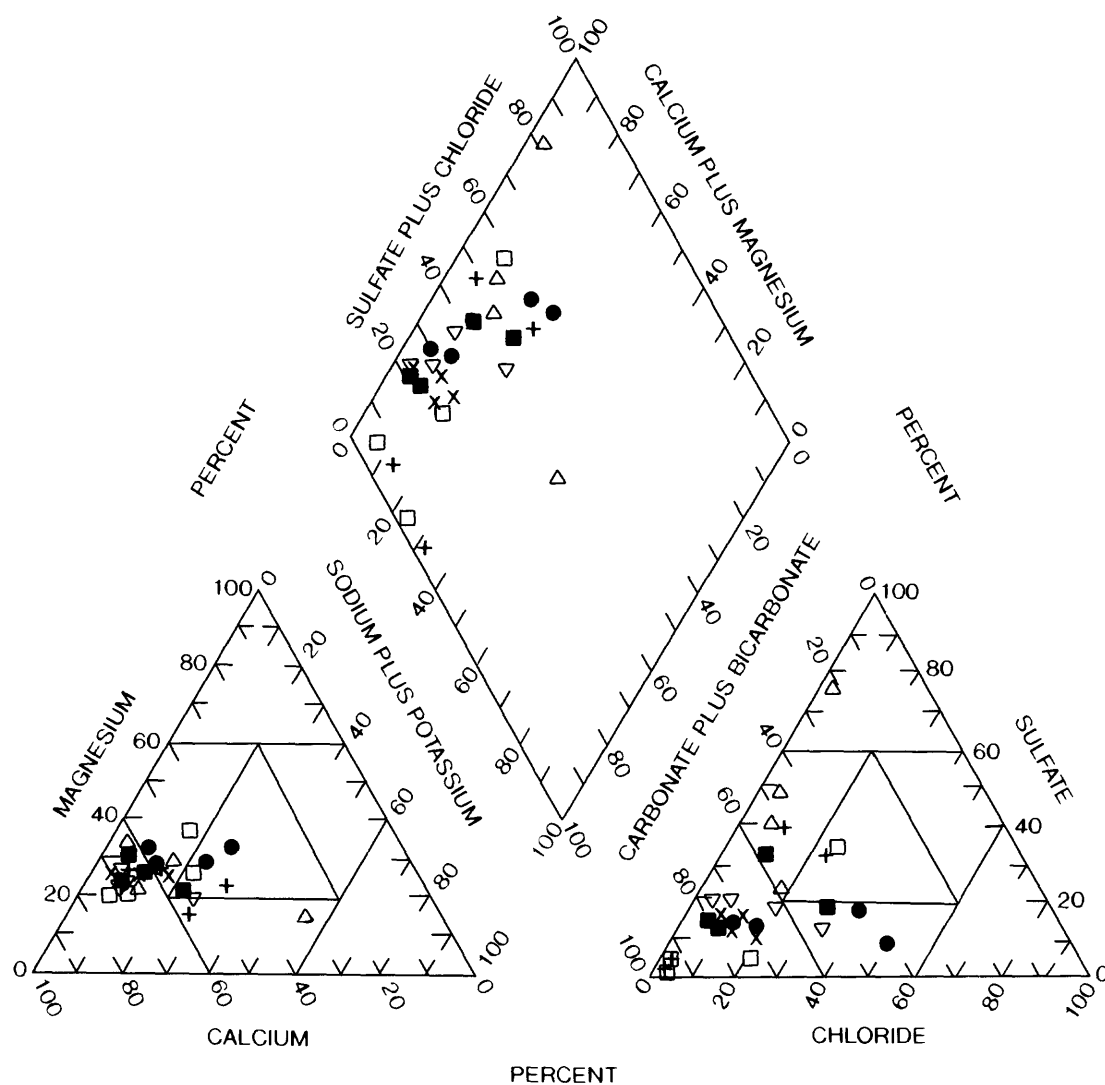
Figure 8.--Locations of ground-water sampling sites, 1989-91.

Table 3. Ranges and medians for selected water-quality characteristics for ground-water sites in active coal-mining areas of Ohio, 1989-1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 8)	Median
Specific conductance, in µS/cm -----	300 to 1,300	Mu-50; Pe-58	650
pH -----	7.0 to 7.9	Cs-152; Mu-50	7.4
Oxygen, dissolved, in mg/L -----	0 to 8.1	Cs-149; Cs-152	3.1
Hardness, in mg/L as CaCO ₃ -----	140 to 730	Cs-150, Mu-50; At-70	292
Noncarbonate hardness, in mg/L as CaCO ₃ -----	0 to 610	Several sites; At-70	59
Calcium, dissolved, in mg/L as Ca -----	43 to 190	Mu-50; At-70	78
Magnesium, dissolved, in mg/L as Mg -----	7 to 61	Cs-150; At-70	20
Sodium, dissolved, in mg/L as Na -----	3.5 to 120	Cs-154; Hk-54	18
Potassium, dissolved, in mg/L as K -----	0.6 to 6.1	Mu-50; Cs-155	1.8
Alkalinity, in mg/L as CaCO ₃ -----	67 to 352	Cs-152; F-14	229
Sulfate, dissolved, in mg/L as SO ₄ -----	<1.0 to 470	Cs-149, Tu-55; At-70	48
Chloride, dissolved, in mg/L as Cl -----	2.1 to 190	Cs-150; Pe-58	26
Silica, dissolved, in mg/L as SiO ₂ -----	6.6 to 15	Tu-55; Pe-58	10
Solids, dissolved, sum of constituents, in mg/L -----	163 to 822	Mu-50; At-70	413
Aluminum, total, in µg/L as Al -----	<10 to 250	Several sites; Tu-55	10
Aluminum, dissolved in µg/L as Al -----	<10 to 70	Most sites; Pe-58	<10
Iron, total, in µg/L as Fe -----	<10 to 9,800	Cs-154; Pe-58	1,005
Iron, dissolved, in µg/L as Fe -----	<3 to 8,700	Several sites; Tu-57	275
Manganese, total, in µg/L as Mn -----	<10 to 790	Several sites; Mu-51	70
Manganese, dissolved, in µg/L as Mn -----	<1 to 830	Cs-154, 155; Mu-51	59
Carbon, organic, dissolved, in mg/L as C -----	0.6 to 3.1	Several sites; Mu-50	.8

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon



EXPLANATION

CATION-ANION PERCENTAGES BASED ON
MILLIEQUIVALENTS PER LITER IN
GROUND-WATER SAMPLES

- + Conotton Creek basin
- Lower Wills Creek basin
- △ Middle Hocking River basin
- ▽ Middle Muskingum River basin
- Upper Hocking River basin
- Upper Muskingum River basin
- x Walhonding River basin

Figure 9.--Piper diagram showing distribution of constituents for ground-water sites sampled in active coal mining areas of Ohio, 1989-91.

Because many ground-water sources are used for domestic or public supplies, the following Ohio Environmental Protection Agency (OEPA) (1989) drinking-water-quality standards are listed for reference (all concentrations are expressed as total concentrations unless otherwise noted):

Chloride.....	250 mg/L
Dissolved solid.....	500 mg/L*
Iron (dissolved).....	300 µ/L
Manganese.....	50 µg/L
Sulfate.....	250 mg/L

*Not to exceed 500 mg/L as a monthly average or 750 µg/L at any time (equivalent specific conductance values at 25 degrees Celsius are 800 and 1,200 µS/cm, respectively).

GEOLOGIC SETTING AND WATER QUALITY OF SELECTED BASINS

Each of the 10 drainage basins selected for study in which synoptic measurements were made from 1989 to 1991 is described with respect to its physical setting and geologic framework in the following sections. The physiographic names used in the descriptions are based on designations by Fenneman (1938). Drainage areas mentioned in the text and in tables were derived from the "Gazetteer of Ohio Streams" (Krolczyk, 1954; Cross, 1967), and, in some instances, compiled from drainage-area maps maintained by the USGS.

Results of the synoptic ground-water and streamwater sampling in 10 drainage basins are presented in tables and figures in the discussions for each basin that follow. The synoptic streamwater sampling included collection of water-quality samples at 45 short-term and 17 long-term sites in 9 of the 10 drainage basins. The synoptic ground-water sampling included collection of water-quality samples at 28 sites in 7 of the 10 drainage basins. Additional water-quality analyses

for the long-term streamwater sites sampled during 1989-91 are listed in table 31 at the back of this report.

Conotton Creek Basin

Conotton Creek basin, with a drainage area of about 286 mi², is situated mostly in the southwestern half of Carroll County and extends into the northeastern part of Tuscarawas County and into the northern part of Harrison County (fig. 1, basin F). It lies entirely within the Unglaciaded Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province some 10 to 15 mi south of the limit of glaciation.

Conotton Creek, the principal stream, follows a northwesterly course close to the southern side of the basin toward its confluence with the Tuscarawas River near Zoarville (fig. 10). Several tributaries to Conotton Creek flow generally to the southwest. Two of these tributaries, Indian Fork and McGuire Creek, have been dammed to form, respectively, Atwood and Leesville reservoirs (fig. 10). Much of the terrain along Conotton Creek is relatively flat, but elsewhere, such as northeastward along the upper reaches of the tributaries, the surface is rolling to moderately hilly.

Geologic Setting

The rocks consist of Pennsylvanian strata, which, from west to east and in ascending order, include the Pottsville, Allegheny, Conemaugh, and Monongahela Formations. As shown in figure 4, the stratigraphic section is a cyclic sequence of sandstones and shales interbedded with coal, clay, and limestone. Areally, the Conemaugh of Pennsylvanian age is the most extensive in the Conotton Creek basin.

Streamwater

Map index number	Site type	Site name	Drainage area (square miles)
F-1	Long term	Conotton C nr Somerdale	268
F-2	Short term	Huff Rn at Mineral City	12.1
F-3	do.	Indian F nr Dellroy	33.9
F-4	do.	NF McGuire C at Petersburg	11.2
F-5	do.	Conotton C at Leesville	87.1
F-6	do.	Unnamed Tr Conotton C nr New Cumberland	5.5

Areas of active coal mining are mostly along the western side of the basin, producing coal mainly from the Allegheny Formation. Active areas along the eastern side of the basin southeast of Carrollton also produce from the Allegheny Formation, as well as from the Harlem coal, a minor coal within the Conemaugh Formation. Some remnants of the Monongahela Formation are found in the southeastern corner of the basin in Harrison County. Although an important coal producer elsewhere in Ohio, the Monongahela coals in this area have been mined out.

Aquifers throughout the basin consist mostly of sandstones and thin fractured limestones. Potential yields to wells of as much as 25 gal/min from these units are possible, but, in the eastern half of the basin, yields typically are less than 5 gal/min (Schmidt, 1962a). Along the lower course of Conotton Creek, the valley is filled with unconsolidated sediments that are more than 100 ft thick in places, and potential well yields can range from 5 to 25 gal/min. Many wells in the area are cased into and yield from bedrock beneath the valley-fill sediments.

Water Quality

Synoptic collections of water samples were made at six stream sites and at four ground-water sites in the Conotton Creek basin. Half of the stream sites and all of the ground-water sites were in the western part of the basin (fig. 10), the area with most of the active mining.

The streamwater sites listed in the table above were sampled in mid-October 1989. Their locations are shown in figure 10.

The lower part of Conotton Creek consists mainly of pools of slow-moving water. Because of low water velocities in this area, streamflow measurements cannot be accurately made nor can representative water-quality samples be collected. As an alternative, long-term site F-1 was added for synoptic sampling. Site F-1 represents about 93 percent of the drainage area in the basin, including the lower part of Conotton Creek.

Water-quality data for the six streamwater sites in Conotton Creek basin are given in table 4. Water quality at site F-2 was different from that at the five other sites and was characterized by a lower pH, a higher specific conductance, and higher concentrations of total recoverable and dissolved aluminum, iron, and manganese. At site F-2, constituent concentrations were at the top of the range for the study

area, except for pH and alkalinity, which were at the bottom of the range (table 5). Box plots showing the ranges, percentiles, and medians of constituents for all sites in Conotton Creek basin (fig. 11) show that the concentrations of constituents at site F-2 were well outside the interquartile ranges. To a much lesser extent than site F-2, the water from site F-6 had higher specific conductance and higher concentrations of total recoverable and dissolved manganese, total recoverable iron, and dissolved sulfate than the other four sites.

The water quality at site F-2, and to a much lesser extent at site F-6, is characteristic of water affected by mining. Unrestored coal-mining lands can be found in the northwestern part of Conotton Creek basin.

Ground water

Most wells in the Conotton Creek basin yield water from bedrock except along Conotton Creek, where wells are screened in layers of unconsolidated deposits. The four ground-water sites chosen for sampling were in and near the town of Somerdale (fig. 10). Three of these wells yield water from alluvial deposits along the valley of Conotton Creek (Tu-55, Tu-56, and Tu-57); Tu-54 is finished in the underlying bedrock. In general, the ground water used by the residents in the Somerdale area is derived from both aquifer types.

Water-quality data for four ground-water sites sampled in September 1989 are given in table 6. Values of pH ranged from 7.2 to 7.8, and, except for total and dissolved iron at three sites, concentrations of constituents associated with acid mine drainage were not high.

As shown by the Stiff diagrams (fig. 10), calcium was the principal cation at all four sites; however, the sites differed in anionic character. At sites Tu-56 and Tu-57, sulfate and bicarbonate were the major anions, and, at sites Tu-54 and Tu-55, sulfate concentrations were low and bicarbonate was the dominant anion.

The OEPA water-quality standards for public supplies for dissolved solids, chloride, and sulfate were not exceeded at any of the four sites. Water-quality standards for dissolved iron (300 µg/L) and manganese (50 µg/L), however, were exceeded at sites Tu-54, Tu-55, and Tu-57. The water from site Tu-56 met all OEPA water-supply standards for the constituents analyzed. The waters at all four sites can be classified as hard to very hard.

Lower Wills Creek Basin

The Lower Wills Creek basin, which is about 447 mi² in area, covers most of the northern half of Guernsey County, just touching Belmont County on the east, Tuscarawas County on the north, and extending into the southeastern corner of Coshocton County and the northeastern corner of Muskingum County (fig. 1, basin L). The Lower Wills Creek basin is in the Unglaciated Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province. The northern edge of the basin is about 25 mi south of the glacial limit.

Wills Creek, the principal stream, meanders across the central and northwestern part of the basin before it enters the Tuscarawas River system (fig. 12). Wills Creek is also the principal stream draining upper Wills Creek basin (fig. 1, basin K), described earlier (Sedam, 1991). The two basins are separated at a point upstream from and just south of Cambridge. The largest tributary within the Lower Wills Creek basin, Salt Fork, was dammed near its confluence with Wills Creek to create Salt Fork Reservoir, which extends into the lower reach of several tributaries to Salt Fork in the eastern part of the basin. Other principal tributaries to Wills Creek include Birds Run and Bacon Run in the north and White Eyes and Crooked Creeks in the western half of the basin (fig. 12).

Table 4. Water-quality data for streamwater sites in Conotton Creek basin, October 1989

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)	
October 17, 1989	403426081211900 F-1 CONOTTON C NR SOMERDALE OH (LAT 40 34 26N LONG 081 21 19W)	36	495	7.6	16.0	--	90	130
October 18, 1989	403550081213400 F-2 HUFF RN AT MINERAL CITY OH (LAT 40 35 50N LONG 081 21 34W)	2.6	1,850	4.0	11.5	63	--	980
October 17, 1989	403302081095100 F-3 INDIAN F NR DELLROY OH (LAT 40 33 02N LONG 081 09 51W)	2.9	565	7.5	16.0	--	138	45
October 16, 1989	403047081064700 F-4 N F MCGUIRE C NR CARROLLTON OH (LAT 40 30 47N LONG 081 06 47W)	1.2	320	8.1	18.0	--	113	32
October 16, 1989	03119900 F-5 CONOTTON C AT LEESVILLE OH (LAT 40 26 44N LONG 081 11 49W)	9.6	525	8.0	16.5	--	124	120
October 17, 1989	403304081191600 F-6 UNNAMED TR CONOTTON C NR NW CUMBERLAND OH (LAT 40 33 04N LONG 081 19 16 W)	1.0	855	7.6	16.5	--	87	430

DATE	Aluminum, total recoverable (μg/L as Al)	Aluminum, dissolved (μg/L as Al)	Iron, total recoverable (μg/L as Fe)	Iron, dissolved (μg/L as Fe)	Manganese, total recoverable (μg/L as Mn)	Manganese, dissolved (μg/L as Mn)	
October 17, 1989	403426081211900 F-1 CONOTTON C NR SOMERDALE OH (LAT 40 34 26N LONG 081 21 19W)	540	30	2,200	250	920	990
October 18, 1989	403550081213400 F-2 HUFF RN AT MINERAL CITY OH (LAT 40 35 50N LONG 081 21 34W)	1,300	1,200	10,000	9,000	21,000	24,000
October 17, 1989	403302081095100 F-3 INDIAN F NR DELLROY OH (LAT 40 33 02N LONG 081 09 51W)	180	20	860	330	420	400
October 16, 1989	403047081064700 F-4 N F MCGUIRE C NR CARROLLTON OH (LAT 40 30 47N LONG 081 06 47W)	110	20	330	240	60	50
October 16, 1989	03119900 F-5 CONOTTON C AT LEESVILLE OH (LAT 40 26 44N LONG 081 11 49W)	390	30	760	290	210	180
October 17, 1989	403304081191600 F-6 UNNAMED TR CONOTTON C NR NW CUMBERLAND OH (LAT 40 33 04N LONG 081 19 16 W)	300	40	950	80	1,800	1,900

Table 5. Ranges and medians for selected water-quality characteristics for streamwater sites in Conotton Creek basin, October 1989

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 10)	Median
Specific conductance, in µS/cm-----	320 to 1,850	F-4; F-2	550
pH-----	4.0 to 8.1	F-2; F-4	7.6
Alkalinity, in mg/L as CaCO ₃ -----	87 to 138	F-6; F-3	^b 113
Sulfate, dissolved, in mg/L as SO ₄ -----	32 to 980	F-4; F-2	125
Aluminum, total, in µg/L as Al-----	110 to 1,300	F-4; F-2	345
Aluminum, dissolved, in µg/L as Al-----	20 to 1,200	F-3, 4; F-2	30
Iron, total, in µg/L as Fe-----	330 to 10,000	F-4; F-2	905
Iron, dissolved, in µg/L as Fe-----	80 to 9,000	F-6; F-2	270
Manganese, total, in µg/L as Mn-----	60 to 21,000	F-4; F-2	670
Manganese, dissolved, in µg/L as Mn-----	50 to 24,000	F-4; F-2	695

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity could be measured (see "Methods," p. 14).

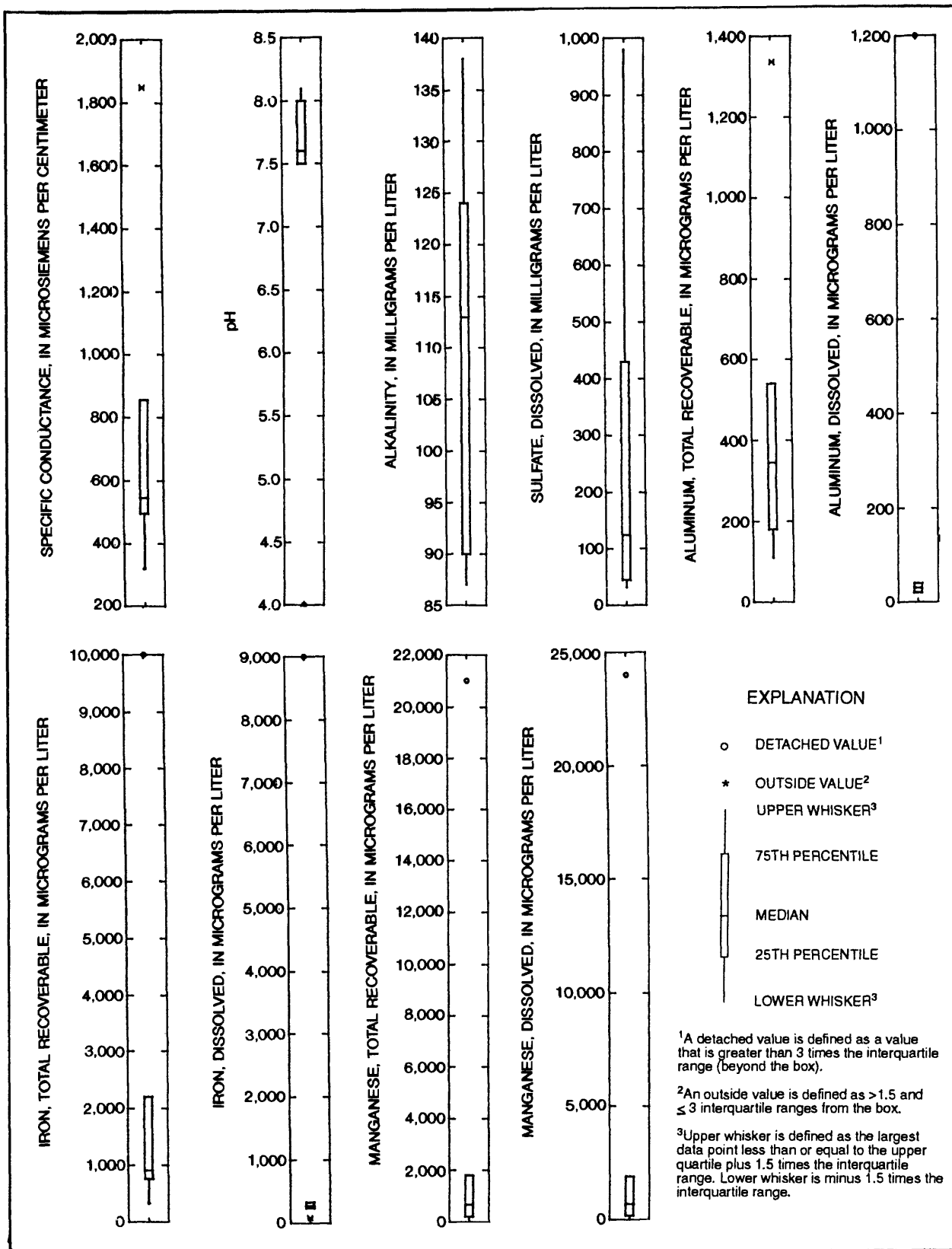


Figure 11.—Ranges, percentiles, and medians of constituents at six streamwater sites in the Conotton Creek basin.

Table 6. Water-quality data for ground-water sites in Conotton Creek basin, September 1989

[°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Depth to water below land-surface datum (feet)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Oxygen, dissolved (mg/L)	Hardness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Magnesium, dissolved (mg/L as Mg)	Sodium, dissolved (mg/L as Na)	Potassium, dissolved (mg/L as K)	Alkalinity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Sept. 5, 1989	40.40	510	7.3	14.5	1.6	240	66	19	15	1.7	288	10
Sept. 5, 1989	29.55	540	7.2	13.5	.6	190	53	15	41	1.8	314	<1.0
Sept. 5, 1989	25.31	620	7.8	14.0	4.2	220	70	11	34	2.0	139	96
Sept. 5, 1989	--	670	7.8	15.0	4.0	320	90	24	9.6	1.3	190	140
403402081213400 TU-54 AT SOMERDALE OH (LAT 40 34 02N LONG 081 21 34W)												
403355081212500 TU-55 AT SOMERDALE OH (LAT 40 33 55N LONG 081 21 25W)												
403356081205000 TU-56 NR SOMERDALE OH (LAT 40 33 56N LONG 081 20 50W)												
403404081214000 TU-57 AT SOMERDALE OH (LAT 40 34 04N LONG 081 21 40W)												
403402081213400 TU-54 AT SOMERDALE OH (LAT 40 34 02N LONG 081 21 34W)												
Sept. 5, 1989	4.6	7.0	226	298	20	<10	2,300	1,600	50	58	0.9	
Sept. 5, 1989	4.4	6.6	264	279	250	<10	3,300	2,800	170	160	1.2	
Sept. 5, 1989	50	13	364	359	<10	20	270	10	20	1	.6	
Sept. 5, 1989	23	14	397	425	<10	20	8,900	8,700	330	340	1.0	
403356081205000 TU-56 NR SOMERDALE OH (LAT 40 33 56N LONG 081 20 50W)												
403404081214000 TU-57 AT SOMERDALE OH (LAT 40 34 04N LONG 081 21 40W)												

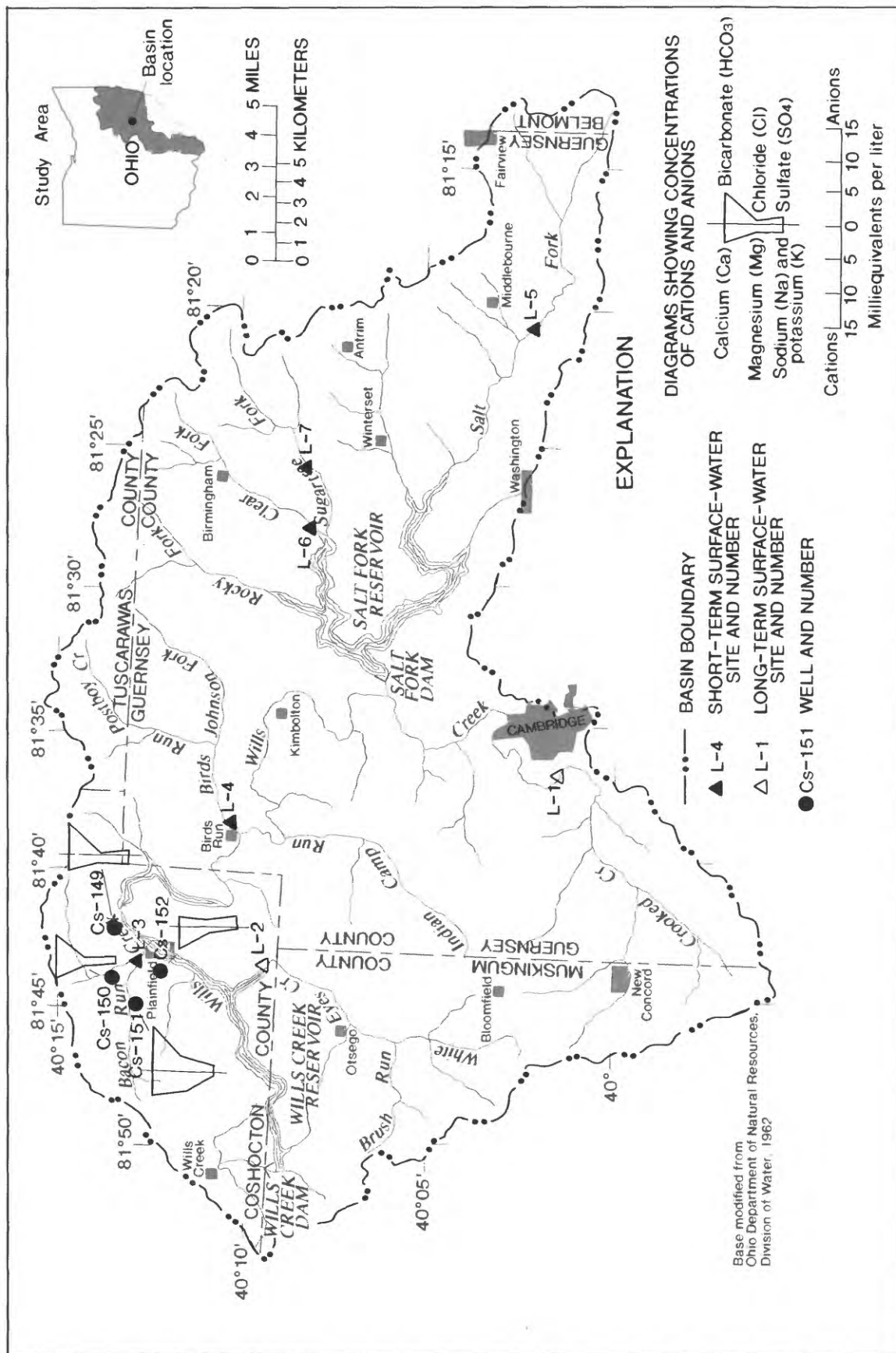


Figure 12.—Lower Wills Creek basin (L), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.

Erosion of the land surface has produced a thoroughly dissected terrain, characterized by wide bottomlands and low hills. Toward the periphery of the basin, near the headwaters of the smaller tributaries, the terrain is more rugged. In contrast, much of the terrain along the lower reach of Wills Creek is wetland.

Geologic Setting

From northwest to southeast across Lower Wills Creek basin, the stratigraphic succession, in ascending order, includes the four major formations of Pennsylvanian age (fig. 4). In the northwest, the Pottsville Formation is present in valley floors in the southeastern corner of Coshocton County and adjacent parts of Guernsey County. In the same area, the uplands consist mostly of the Allegheny Formation and its mineable coals; in places, the uplands are capped by remnants of the Conemaugh Formation. The Conemaugh forms the surface over much of the rest of the basin, and, toward the southeastern corner, scattered remnants of the Monongahela Formation are found at higher elevations. Because of the Cambridge Arch, as described by Lamborn (1951), the rock strata are structurally high in the northwestern to the central part of the basin. This has led to erosion of much of the Conemaugh Formation and exposure of the underlying Allegheny Formation along Wills Creek between Cambridge and Kimbolton (fig. 12). As of 1989, most active permits to mine coal were for the northern part of the basin, where Allegheny coals are common.

In general, ground-water yields from bedrock are meager in the Lower Wills Creek basin. Well yields of 3 to 5 gal/min or less are found in upland areas in the northwestern part of the basin. Similar low yields are common in the eastern part of the basin where the Conemaugh Formation is the principal underlying unit. In the northwestern part of the basin, however, wells tapping bedrock yield as much as 25 gal/min (Walker, 1962b; Sugar,

1988) in some places. Such places are likely to be where permeable units of the Pottsville or Allegheny Formation are present at or below stream level.

Unconsolidated deposits making up the valley fill along Wills Creek can yield as much as 25 gal/min. Many of these potential sources underlie wetland areas, and thus are likely to remain undeveloped. An exception is the area around Plainfield, where the bottomlands along Wills Creek and adjacent tributaries are underlain by sediments that, in places, are 80 to 100 ft thick. The most productive shallow aquifer is found in this area of the basin.

Water Quality

Synoptic collections of water samples were made at two long-term and five short-term streamwater sites in the Lower Wills Creek basin, as listed in the table on the next page. Ground-water samples were collected at four sites near the town of Plainfield, in the northwestern part of the basin (fig. 12).

Wills Creek Dam, which forms Wills Creek Reservoir, is near the downstream end of Wills Creek about 6 mi above the confluence of Wills Creek with the Muskingum River. Elsewhere in the basin, Wills Creek is a rather sluggish stream despite the discharge that it receives from Upper Wills Creek basin. Neither the USGS gaging station at Cambridge nor the gaging station below Wills Creek Dam are suitable as long-term sampling sites for this investigation because Wills Creek is a composite of drainage from various sources. Instead, the drainage carried by Crooked Creek and White Eyes Creek, on which the long-term sites were established (sites L-1 and L-2), is more representative of baseline conditions in the basin than what might be derived from water-quality analyses of the main stem.

The five short-term sites were spaced throughout the basin to assess streamwater quality of active mining areas north of Plainfield and to provide coverage of the remainder of the basin. Small areas of mining

Streamwater

The following streamwater sites were sampled in mid-October 1989. Their locations are shown in figure 12.

Map index number	Site type	Site name	Drainage area (square miles)
L-1	Long term	Crooked C nr Cambridge	42.5
L-2	do.	White Eyes C nr Plainfield	51.7
L-3	Short term	Bacon Rn nr Plainfield	14.5
L-4	do.	Birds Rn at Birds Run	31.0
L-5	do.	Salt F nr Middlebourne	24.5
L-6	do.	Clear F nr Birmingham	11.2
L-7	do.	Sugartree C nr Birmingham	14.9

are located near headwaters of streams that drain to Salt Fork. The forming of Salt Fork Reservoir has reduced the gradient of the downstream part of its tributaries such that it was necessary to go some distance upstream in order to sample at a free-flowing site. Water-quality data for the seven streamwater sites in Lower Wills Creek basin are given in table 7. Water from sites L-2, L-3, and L-5 had dissolved-sulfate concentrations greater than 300 mg/L. Water from these sites had the highest specific conductances, ranging from 880 to 1,050 $\mu\text{S}/\text{cm}$. For all samples, pH was in the alkaline range; the median pH was 7.6 (table 8). Box plots of concentrations for selected water-quality constituents included few outside values and no detached values (fig. 13). In general, the high concentrations of sulfate, manganese, iron, and aluminum and the low pH associated with acid mine drainage were not found in the samples from Lower Wills Creek basin.

Ground water

Four wells that yield from unconsolidated deposits were sampled near the village of Plainfield. Two of the sites, Cs-150 and Cs-151, were along streams that drain the most active coal-producing areas of the basin.

Water-quality data for four ground-water sites sampled in September 1989 are given in table 9. Values for pH ranged from 7.0 to 7.8. Concentrations of total and dissolved iron were high at sites Cs-149, Cs-150, and Cs-151 and ranged from 1,400 to 5,000 $\mu\text{g}/\text{L}$. Concentrations of dissolved sulfate and total recoverable and dissolved manganese and aluminum, commonly associated with acid mine drainage, were not high.

Stiff diagrams (fig. 12) show that calcium was the dominant cation except in well Cs-151, where magnesium and calcium were principal cations. Bicarbonate was the dominant anion except in well Cs-152, where both bicarbonate and sulfate were principal anions.

The OEPA standards for public supplies for dissolved iron (300 µg/L) and manganese (50 µg/L) were exceeded in samples from three of the wells, Cs-149, Cs-151, and Cs-150. The water from well Cs-152 met all OEPA standards for constituents analyzed. The waters were moderately hard (Cs-150), hard (Cs-149, Cs-152), and very hard (Cs-151).

Upper Hocking River Basin

The Upper Hocking River basin, with a drainage area of 465 mi², covers approximately the southeastern two-thirds of Fairfield County, the western third of Perry County, and the northern edge of Hocking County (fig. 1, basin P). The area includes all drainage to the Hocking River at and including Harper Run just west of Logan (fig. 14).

As the principal stream, the Hocking River begins at the northwestern corner of the basin (fig. 14) and flows about 32 mi across the western half of the basin to the point where it becomes the upper end of the Middle Hocking River basin (figs. 1 and 24). Two principal tributaries are Clear Creek, which drains the southwestern end of the basin, and Rush Creek and its tributaries, which drain the eastern two-thirds of the basin.

Pleistocene glacial advances covered about four-fifths of the Upper Hocking River basin (fig. 14). The limit of glacial advance marks the division between the Glaciated and Unglaciated Allegheny Plateau Sections of the Appalachian Plateaus Physiographic Province. Of the two major glacial advances recognized in the area (Goldthwait and others, 1961), the last major advance (Wisconsinan) did not extend as far as the earlier (Illinoian) advance. The Illinoian surface is thus exposed in a zone 2 to 10 mi wide across the Upper Hocking River basin. In the rugged, thoroughly dissected upland that rises above the Hocking River valley in the unglaciated section, local relief is as great as 350 ft.

In contrast, local relief is only minor in Wisconsinan glaciation. Within the zone where Illinoian drift is exposed, modification by post-glacial erosion has left a land surface that is more rugged than the Wisconsinan-drift-covered surface but less rugged than the unglaciated terrain along the southern edge of the basin.

Geologic Setting

From west to east, rock units underlying the Upper Hocking River basin range from undifferentiated Mississippian to Middle Pennsylvanian in age. Lithology of the undifferentiated Mississippian rocks, which underlie most of the basin, is generalized in figure 4. Principal units that are used for water supplies, in ascending order, include the Berea Sandstone and sandstones and shaly sandstones of the Cuyahoga and Logan Formations.

One of the more important of these aquifers is the Black Hand Sandstone Member of the Cuyahoga Formation. Although the Mississippian aquifer units are not closely associated with coal mining, the Black Hand Sandstone has been described as a useful source of water below coal-bearing strata in parts of the subject and adjacent basins where surface mining has become increasingly active (Norris and Mayer, 1982).

Small sections of the Pottsville Formation underlie the Upper Hocking River basin in eastern Fairfield County. Although coal is present, Brant and Delong's assessment study (1960) attaches little importance to Pottsville coals in the area. Equivalents of the conglomeratic sandstone aquifers at the base of the Pottsville Formation in northeastern Ohio are not present in the Upper Hocking River basin. Much of the Pottsville is above the water table as determined from water levels of perennial streams; but, toward the

Table 7. Water-quality data for streamwater sites in Lower Wills Creek basin, October 1989

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)	
October 17, 1989	400117081362600 L-1 CROOKED C NR CAMBRIDGE OH (LAT 40 01 17N LONG 081 36 26W)	5.1	640	7.5	17.0	--	148	140
October 16, 1989	400920081432900 L-2 WHITE EYES C NR PLAINFIELD OH (LAT 40 09 20N LONG 081 43 29W)	5.1	965	7.4	18.0	--	98	390
October 16, 1989	401300081433300 L-3 BACON RN NR PLAINFIELD OH (LAT 40 13 00N LONG 081 43 33W)	2.3	1,050	7.7	18.0	--	96	450
October 16, 1989	401009081385400 L-4 BIRDS RN AT BIRDS RUN OH (LAT 40 10 09N LONG 081 38 54W)	1.2	610	7.7	17.5	--	104	54
October 16, 1989	400130081194900 L-5 SALT FORK NR MIDDLEBOURNE OH (LAT 40 01 30N LONG 081 19 49W)	3.8	880	7.6	16.0	--	131	320
October 16, 1989	400932081265100 L-6 CLEAR F NR BIRMINGHAM OH (LAT 40 09 32N LONG 081 26 51W)	1.2	360	7.9	17.0	--	116	39
October 16, 1989	400910081261000 L-7 SUGARTREE C NR BIRMINGHAM OH (LAT 40 08 10N LONG 081 26 10W)	.82	390	7.5	16.5	--	119	35

DATE	Aluminum, total recoverable (μg/L as Al)	Aluminum, dissolved (μg/L as Al)	Iron, total recoverable (μg/L as Fe)	Iron, dissolved (μg/L as Fe)	Manganese, total recoverable (μg/L as Mn)	Manganese, dissolved (μg/L as Mn)	
October 17, 1989	400117081362600 L-1 CROOKED C NR CAMBRIDGE OH (LAT 40 01 17N LONG 081 36 26W)	590	30	930	90	570	500
October 16, 1989	400920081432900 L-2 WHITE EYES C NR PLAINFIELD OH (LAT 40 09 20N LONG 081 43 29W)	360	20	1,300	190	780	680
October 16, 1989	401300081433300 L-3 BACON RN NR PLAINFIELD OH (LAT 40 13 00N LONG 081 43 33W)	220	20	360	70	360	290
October 16, 1989	401009081385400 L-4 BIRDS RN AT BIRDS RUN OH (LAT 40 10 09N LONG 081 38 54W)	230	20	860	450	190	160
October 16, 1989	400130081194900 L-5 SALT FORK NR MIDDLEBOURNE OH (LAT 40 01 30N LONG 081 19 49W)	760	20	910	30	310	240
October 16, 1989	400932081265100 L-6 CLEAR F NR BIRMINGHAM OH (LAT 40 09 32N LONG 081 26 51W)	320	<10	670	130	170	130
October 16, 1989	400910081261000 L-7 SUGARTREE C NR BIRMINGHAM OH (LAT 40 08 10N LONG 081 26 10W)	350	10	920	190	270	200

Table 8. Ranges and medians for selected water-quality characteristics for streamwater sites in Lower Wills Creek basin, October 1989

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 12)	Median
Specific conductance, in µS/cm-----	360 to 1,050	L-6; L-3	640
pH-----	7.4 to 7.9	L-2; L-6	7.6
Alkalinity, in mg/L as CaCO ₃ -----	96 to 148	L-3; L-1	^b 116
Sulfate, dissolved, in mg/L as SO ₄ ----	35 to 450	L-7; L-3	140
Aluminum, total, in µg/L as Al-----	220 to 760	L-3; L-5	350
Aluminum, dissolved, in µg/L as Al-----	<10 to 30	L-6; L-1	20
Iron, total, in µg/L as Fe-----	360 to 1,300	L-3; L-2	910
Iron, dissolved, in µg/L as Fe-----	30 to 450	L-5; L-4	130
Manganese, total, in µg/L as Mn-----	170 to 780	L-6; L-2	310
Manganese, dissolved, in µg/L as Mn-----	130 to 680	L-6; L-2	240

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity could be measured (see "Methods," p. 14).

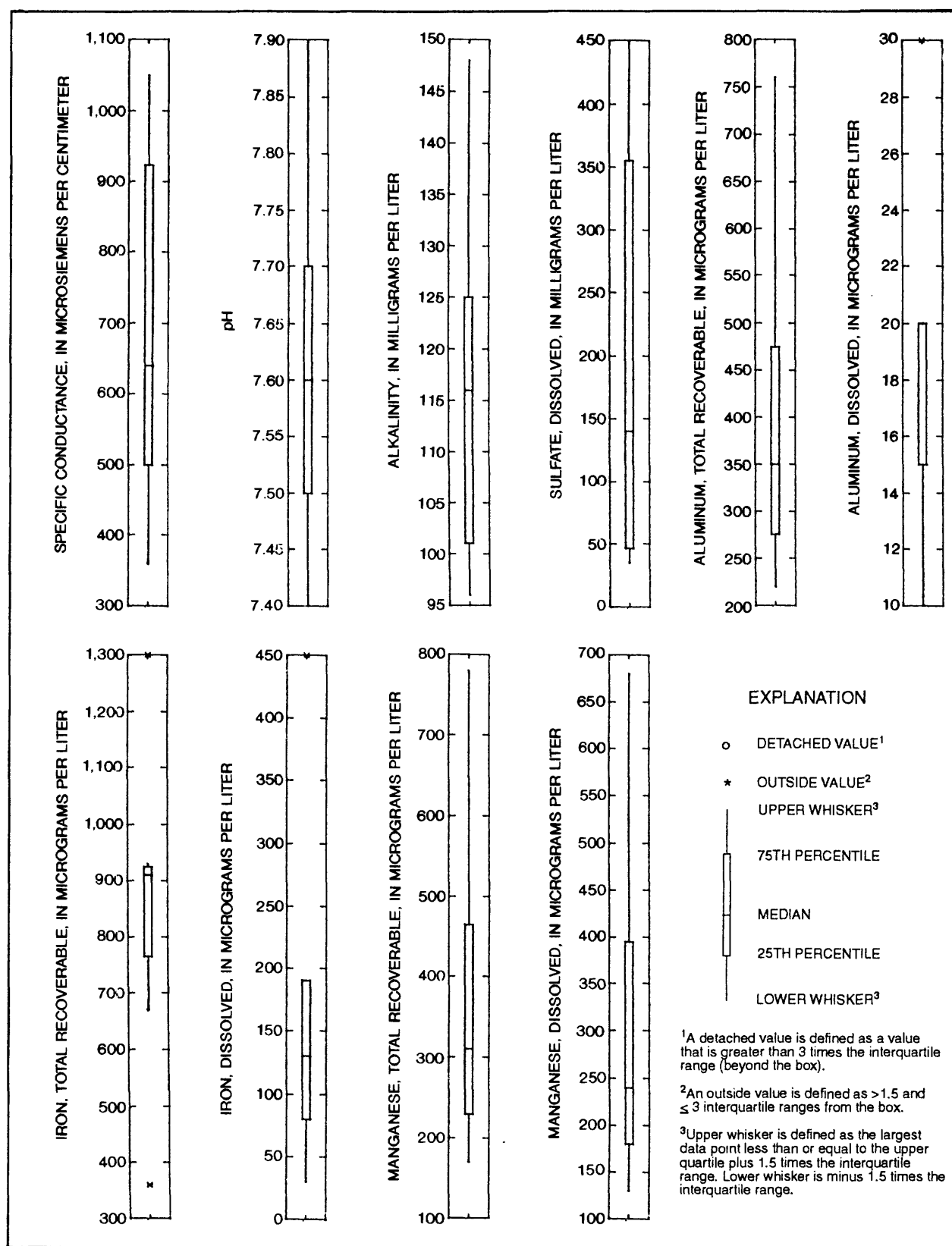


Figure 13.--Ranges, percentiles, and medians of constituents at seven streamwater sites in the Lower Wills Creek basin.

Table 9. Water-quality data for ground-water sites in Lower Wills Creek basin, September 1989

[°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Specific conductance (µS/cm)	pH (standard units)	Temper- ature, water (°C)	Oxygen, dissolved (mg/L)	Hard- ness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Mag- nesium, dissolved (mg/L as Mg)	Sodi- um, dissolved (mg/L as Na)	Potas- sium, dissolved (mg/L as K)	Alka- linity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Sept. 1, 1989	440	7.8	14.5	0	170	46	14	23	1.4	238	<1.0
Sept. 1, 1989	305	7.7	15.5	--	140	45	7.0	3.8	.80	154	5.0
Sept. 1, 1989	775	7.5	15.5	.4	330	74	35	31	1.2	313	13
Sept. 1, 1989	520	7.0	16.5	8.1	220	67	12	8.7	3.3	67	51

401314081422400	CS-149	NR	PLAINFIELD OH	(LAT 40 13 14N LONG 081 42 24W)
401318081434400	CS-150	NR	PLAINFIELD OH	(LAT 40 13 18N LONG 081 43 44W)
401315081444400	CS-151	NR	PLAINFIELD OH	(LAT 40 13 15N LONG 081 44 44W)
401211081433400	CS-152	NR	PLAINFIELD OH	(LAT 40 12 11N LONG 081 43 34W)

Date	Chloride, dissolved (mg/L as Cl)	Silica, dissolved (mg/L as SiO ₂)	Solids, residue at 180 deg. C, dissolved (mg/L)	Solids, sum of consti- tuents, dissolved (mg/L)	Alumi- num, total, recov- erable (µg/L as Al)	Alumi- num, dissolved (µg/L as Al)	Iron, total, recov- erable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manga- nese, total, recov- erable (µg/L as Mn)	Manga- nese, dissolved (µg/L as Mn)	Carbon, organic, dissolved (mg/L as C)
Sept. 1, 1989	2.5	7.1	237	243	40	<10	5,000	4,100	210	220	1.5
Sept. 1, 1989	2.1	10	151	168	10	<10	3,200	1,400	220	220	1.0
Sept. 1, 1989	62	13	402	421	10	<10	4,800	4,400	60	55	1.1
Sept. 1, 1989	29	9.9	318	221	<10	10	130	16	10	2	.7

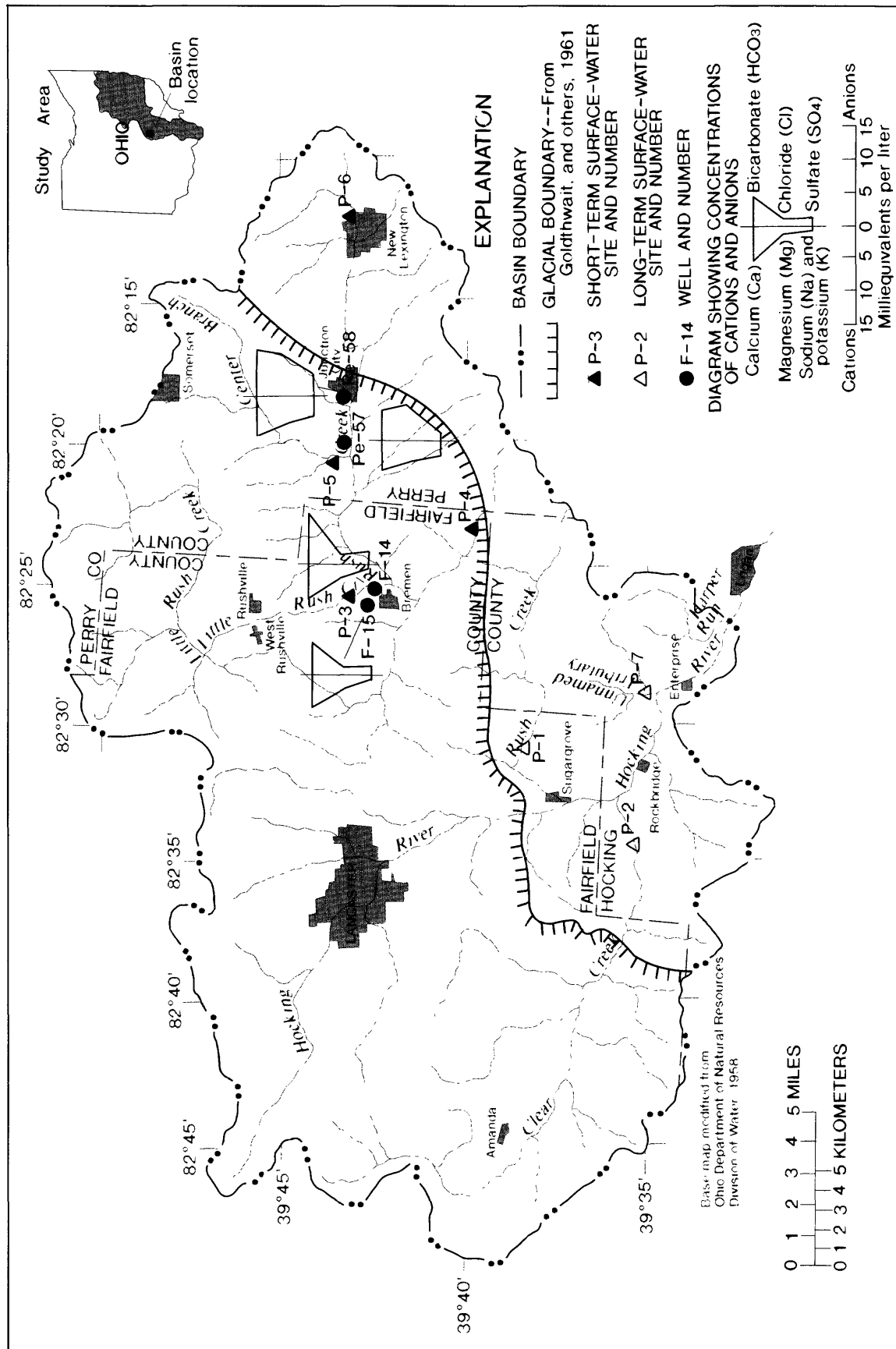


Figure 14.---Upper Hocking River basin (P), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.

eastern side of the basin, the Massillon Sandstone Member (Connoquenessing Sandstone, Member of Pottsville Formation, USGS usage) is available in the subsurface as a water-supply source (Stout and others, 1943).

Scattered remnants of the Allegheny Formation, which underlie the eastern part of the basin, are important as a source of coal (individual coal units are identified in fig. 4). Several permits for surface mining of the Middle Kittanning (No. 6) coal have been issued, but, as of 1989, the No. 6 coal was being mined at only one location along the drainage divide between the Upper and Middle Hocking River basins (fig. 1, basins P and Q). Basal remnants of the Conemaugh Formation are present but only at high elevations along the eastern margin of the basin.

Water Quality

Two long-term sites and five short-term sites were chosen to assess streamwater quality in the Upper Hocking River basin. Four ground-water sites selected for sampling were along the course of Rush Creek.

The streamwater sites in the table on the next page were sampled in mid-October 1989. Their locations are shown in figure 14.

The drainage network of the Upper Hocking River basin is more complex than that of other basins studied in this investigation, at least with respect to the Rush Creek tributary network. Long-term site P-1 is located on the Rush Creek tributary network that drains the eastern half of the basin, the area in which all of the coal deposits and mining activity in the basin are found. In contrast, long-term site P-2 is on a stream draining an area devoid of coal.

Coal-bearing strata are present near the headwaters of each of the five short-term sites that were used in the synoptic sampling. Site P-7 was chosen to measure a small watershed that drains directly to the Hocking River. Sites P-3, P-4, and P-5 were selected to mea-

sure the larger tributaries to Rush Creek. Site P-6 was selected to measure the upper drainage area of Rush Creek.

Results of the analyses of water samples show that water quality at site P-6 was substantially different from water quality at the other sites sampled (table 10). Water at this site was strongly acidic ($\text{pH} = 2.7$), whereas pH at the other six sites was in the alkaline range. Water from site P-6 had high concentrations of dissolved sulfate and total recoverable and dissolved aluminum, iron, and manganese. Concentrations of total recoverable and dissolved iron were especially high (170,000 and 220,000 $\mu\text{g/L}$). The concentration of dissolved manganese in water from site P-1 also was high (1,300 $\mu\text{g/L}$) but was still well below the dissolved-manganese concentration in water from site P-6 (21,000 $\mu\text{g/L}$). At site P-6, constituent concentrations were at or near the maximum, except for pH and alkalinity (table 11). As shown in figure 15, all of the detached and outside values were from data collected at site P-6, except for a low specific conductance at site P-7.

Water from site P-6 had low pH and high concentrations of the constituents associated with acid mine drainage. The Allegheny Formation, in the eastern part of the basin where site P-6 is located, is composed of few carbonates, which can buffer pH. Site P-1 is in an area of active mining; however, the water at this site was alkaline, and, except for total-recoverable and dissolved-manganese concentrations, was not high in constituents associated with acid mine drainage. This absence of symptoms of acid mine drainage at site P-1 may be due to the effects of dilution and the buffering of streams in this area by the Maxville Limestone of Mississippian age.

Streamwater

Map index number	Site type	Site name	Drainage area (square miles)
P-1	Long term	Rush C nr Sugar Grove	229.0
P-2	do.	Clear C nr Rockbridge	89.0
P-3	Short term	L Rush C nr Bremen	97.0
P-4	do.	Turkey Rn nr Bremen	7.3
P-5	do.	Center B Rush C nr Junction City	24.9
P-6	do.	Rush C at New Lexington	9.4
P-7	do.	Unnamed Tr Hocking R nr Enterprise	7.

Ground water

Four ground-water sites along the course of Rush Creek, between Junction City and Bremen, were selected for sampling. The underlying unconsolidated materials in this area constitute a shallow, productive aquifer within which ground-water movement is normally toward Rush Creek. Susceptibility of the aquifer to the effects of mining could result from the spreading of highly degraded stream-water during flood stages or by infiltration induced by a pumping well close to the stream.

Water-quality data for four ground-water sites sampled in September 1989 are given in table 12. Values for pH ranged from 7.3 to 7.7, and concentrations of total recoverable and dissolved iron were greater than 1,000 µg/L at sites F-14, Pe-57, and Pe-58.

Stiff diagrams (fig. 14) show that wells F-14 and F-15 are calcium bicarbonate-type waters. Stiff diagrams for wells Pe-57 and Pe-58 show the highly mineralized character of these waters; calcium and magnesium are major cations, and bicarbonate and chloride are major anions. All four wells contain very hard waters.

The OEPA water-quality standards for chloride and sulfate were not exceeded in waters from any of the wells; however, water-quality standards for dissolved iron (300 µg/L) were exceeded at wells F-14, Pe-57, and Pe-58. Water-quality standards for manganese (50 µg/L) were exceeded at wells F-14, F-15, and Pe-58. Water from well Pe-58, a public-water supply, also exceeded the water-quality standards for dissolved solids (500 mg/L).

Table 10. Water-quality data for streamwater sites in Upper Hocking River basin, October 1989

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, Instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)	
October 16, 1989	03156700 P-1 RUSH C NR SUGAR GROVE OH (LAT 39 38 18N LONG 082 30 42W)	63	555	7.3	15.5	--	91	150
October 16, 1989	03157000 P-2 CLEAR C NR ROCKBRIDGE OH (LAT 39 35 18N LONG 082 34 43W)	43	467	7.9	17.5	--	194	42
October 17, 1989	394323082253100 P-3 LITTLE RUSH C NR BREMEN OH (LAT 39 43 23N LONG 082 25 31W)	11	504	7.7	17.5	--	198	42
October 16, 1989	393950082225900 P-4 TURKEY RN NR BREMEN OH (LAT 39 39 50N LONG 082 22 59W)	2.5	438	7.1	18.5	--	60	41
October 17, 1989	03156549 P-5 CENTER B RUSH C NR JUNCTION CITY OH (LAT 39 43 24N LONG 082 20 36W)	14	375	7.2	17.0	--	78	50
October 17, 1989	394306082121900 P-6 RUSH C AT NEW LEXINGTON OH (LAT 39 43 06N LONG 082 12 19W)	8.8	2,320	2.7	18.0	669	--	1,400
October 16, 1989	393501082290000 P-7 UNNAMED TR TO HOCKING R NR ENTERPRISE OH (LAT 39 35 01N LONG 082 29 00W)	2.7	197	7.1	17.0	--	42	29

DATE	Aluminum, total recoverable (μg/L as Al)	Aluminum, dissolved (μg/L as Al)	Iron, total recoverable (μg/L as Fe)	Iron, dissolved (μg/L as Fe)	Manganese, total recoverable (μg/L as Mn)	Manganese, dissolved (μg/L as Mn)	
October 16, 1989	03156700 P-1 RUSH C NR SUGAR GROVE OH (LAT 39 38 18N LONG 082 30 42W)	420	30	590	10	1,300	1,300
October 16, 1989	03157000 P-2 CLEAR C NR ROCKBRIDGE OH (LAT 39 35 18N LONG 082 34 43W)	330	20	240	40	40	40
October 17, 1989	394323082253100 P-3 LITTLE RUSH C NR BREMEN OH (LAT 39 43 23N LONG 082 25 31W)	220	10	240	20	20	20
October 16, 1989	393950082225900 P-4 TURKEY RN NR BREMEN OH (LAT 39 39 50N LONG 082 22 59W)	270	<10	340	130	230	220
October 17, 1989	03156549 P-5 CENTER B RUSH C NR JUNCTION CITY OH (LAT 39 43 24N LONG 082 20 36W)	490	20	1,000	210	240	160
October 17, 1989	394306082121900 P-6 RUSH C AT NEW LEXINGTON OH (LAT 39 43 06N LONG 082 12 19W)	34,000	34,000	170,000	220,000	20,000	21,000
October 16, 1989	393501082290000 P-7 UNNAMED TR TO HOCKING R NR ENTERPRISE OH (LAT 39 35 01N LONG 082 29 00W)	270	10	300	110	150	140

Table 11. Ranges and medians for selected water-quality characteristics for streamwater sites in Upper Hocking River basin, October 1989

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 14)	Median
Specific conductance, in µS/cm-----	197 to 2,320	P-7; P-6	467
pH-----	2.7 to 7.9	P-6; P-2	7.2
Alkalinity, in mg/L as CaCO ₃ -----	42 to 198	P-7; P-3	^b 84
Sulfate, dissolved, in mg/L as SO ₄ ----	29 to 1,400	P-7; P-6	42
Aluminum, total, in µg/L as Al-----	220 to 34,000	P-3; P-6	270
Aluminum, dissolved, in µg/L as Al-----	<10 to 34,000	P-4; P-6	20
Iron, total, in µg/L as Fe-----	240 to 170,000	P-2, 3; P-6	340
Iron, dissolved, in µg/L as Fe-----	10 to 220,000	P-1; P-6	110
Manganese, total, in µg/L as Mn-----	20 to 20,000	P-3; P-6	230
Manganese, dissolved, in µg/L as Mn-----	20 to 21,000	P-3; P-6	160

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity could be measured (see "Methods," p. 14).

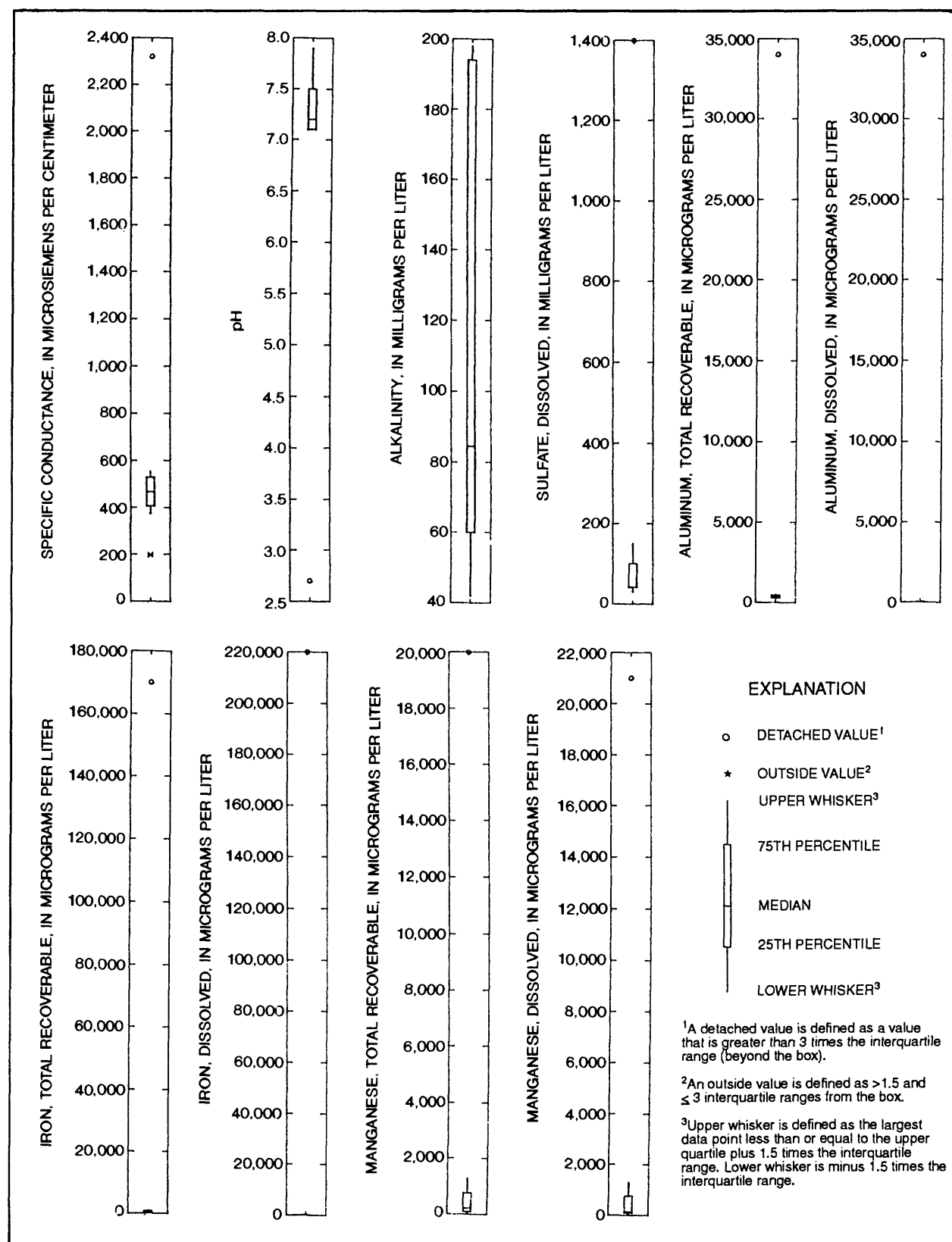


Figure 15.—Ranges, percentiles, and medians of constituents at seven streamwater sites in the Upper Hocking River basin.

Table 12. Water-quality data for ground-water sites in Upper Hocking River basin, August-September 1989

°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Depth to water below land-surface datum (feet)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Oxygen, dissolved (mg/L)	Hardness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Magnesium, dissolved (mg/L as Mg)	Sodium, dissolved (mg/L as Na)	Potassium, dissolved (mg/L as K)	Alkalinity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Aug. 31, 1989	8.73	785	7.4	14.5	0.9	380	100	31	12	1.2	352	70
394231082252100 F-14 NR BREMEN OH (LAT 39 42 31N LONG 082 25 21W)												
Aug. 31, 1989	13.59	660	7.4	16.0	.5	310	83	24	17	1.2	244	49
394257082254000 F-15 AT BREMEN OH (LAT 39 42 57N LONG 082 25 40W)												
Aug. 31, 1989	16.03	965	7.7	12.5	.4	330	74	36	58	1.4	197	46
394326082202700 PE-57 NR JUNCTION CITY OH (LAT 39 43 26N LONG 082 20 27W)												
Sept. 1, 1989	25.63	1,300	7.3	15.0	3.1	480	120	43	65	2.2	304	130
394318082182700 PE-58 WATER WORKS AT JUNCTION CITY OH (LAT 39 43 18N LONG 082 18 27W)												

Date	Chloride, dissolved (mg/L as Cl)	Silica, dissolved (mg/L as SiO ₂)	Solids, residue at 180 deg. C, dissolved (mg/L)	Solids, sum of constituents, dissolved (mg/L)	Aluminum, total, recoverable (µg/L as Al)	Aluminum, dissolved (µg/L as Al)	Iron, total, recoverable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manganese, total, recoverable (µg/L as Mn)	Manganese, dissolved (µg/L as Mn)	Carbon, organic, dissolved (mg/L as C)
Aug. 31, 1989	41	11	353	479	50	<10	6,600	1,200	90	85	0.6
394257082254000 F-15 AT BREMEN OH (LAT 39 42 57N LONG 082 25 40W)											
Aug. 31, 1989	43	12	326	376	<10	<10	190	150	150	170	.9
394326082202700 PE-57 NR JUNCTION CITY OH (LAT 39 43 26N LONG 082 20 27W)											
Aug. 31, 1989	170	9.2	480	516	<10	20	4,100	3,700	70	44	1.3
394318082182700 PE-58 WATER WORKS AT JUNCTION CITY OH (LAT 39 43 18N LONG 082 18 27W)											
Sept. 1, 1989	190	15	784	756	<10	30	9,800	7,900	90	92	1.9

Yellow and Cross Creeks Basin

The Yellow and Cross Creeks basin (fig. 1, basin B), with a drainage area of 439 mi², is a combination of three subbasins that drain to the Ohio River (fig. 16). In the north, the Yellow Creek subbasin includes the northern third of Jefferson County and extends into the eastern part of Carroll County and the southwestern corner of Columbiana County.

Streamwater sites B-1, B-4, and B-5 are in the Yellow Creek subbasin. To the south, the central third of Jefferson County and the northeastern corner of Harrison County form the Short Creek subbasin, which contains streamwater sites B-2 and B-3. The northeastern part of Jefferson County contains a 70-mi² unnamed area between Mingo Junction, Richmond, and Stratton that is drained by several short streams that discharge directly to the Ohio River. Streamwater sites B-6, B-7, and B-8 are in the unnamed subbasin.

The limit of glaciation is less than 10 mi to the north of the Yellow and Cross Creeks basin. Thus, the basin lies entirely within the unglaciated Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province. In general, the terrain is characterized by a thoroughly dissected upland surface; in places, however, the upland consists of broad, relatively flat surfaces that give way abruptly to numerous steep-sided valleys with narrow bottoms. Local relief near the Ohio River valley is as great as 500 ft.

Geologic Setting

Stratigraphic units present, in ascending order, are the Allegheny, Conemaugh, and Monongahela Formations of Pennsylvanian age. The Conemaugh Formation, which is mostly devoid of mineable coal, forms much of the land surface in the study basin.

Structurally, the rocks dip southward to southeastward across the Yellow and Cross Creeks basin. In the northern part of Jefferson County and nearby parts of Columbiana and Carroll Counties, where extensive erosion has produced several deep valleys, the Allegheny Formation is exposed. Elsewhere, the Monongahela Formation is present as scattered remnants along ridgetops that separate the subbasins described earlier and along the drainage divides to the south and west. As of 1989 (Lopez, 1991), coal from the Allegheny Formation was being mined in the northwestern corner of and along the western border of the study basin with Conotton Creek basin (fig. 1, basin F). In the same general area, a small amount of coal from the Conemaugh Formation also was being mined. By far, most of the coal production in the Yellow and Cross Creeks basin is from the Pittsburgh (No. 8) coal at the base of the Monongahela Formation, mostly in the southern part of the study basin and near Knoxville in Jefferson County (fig. 16).

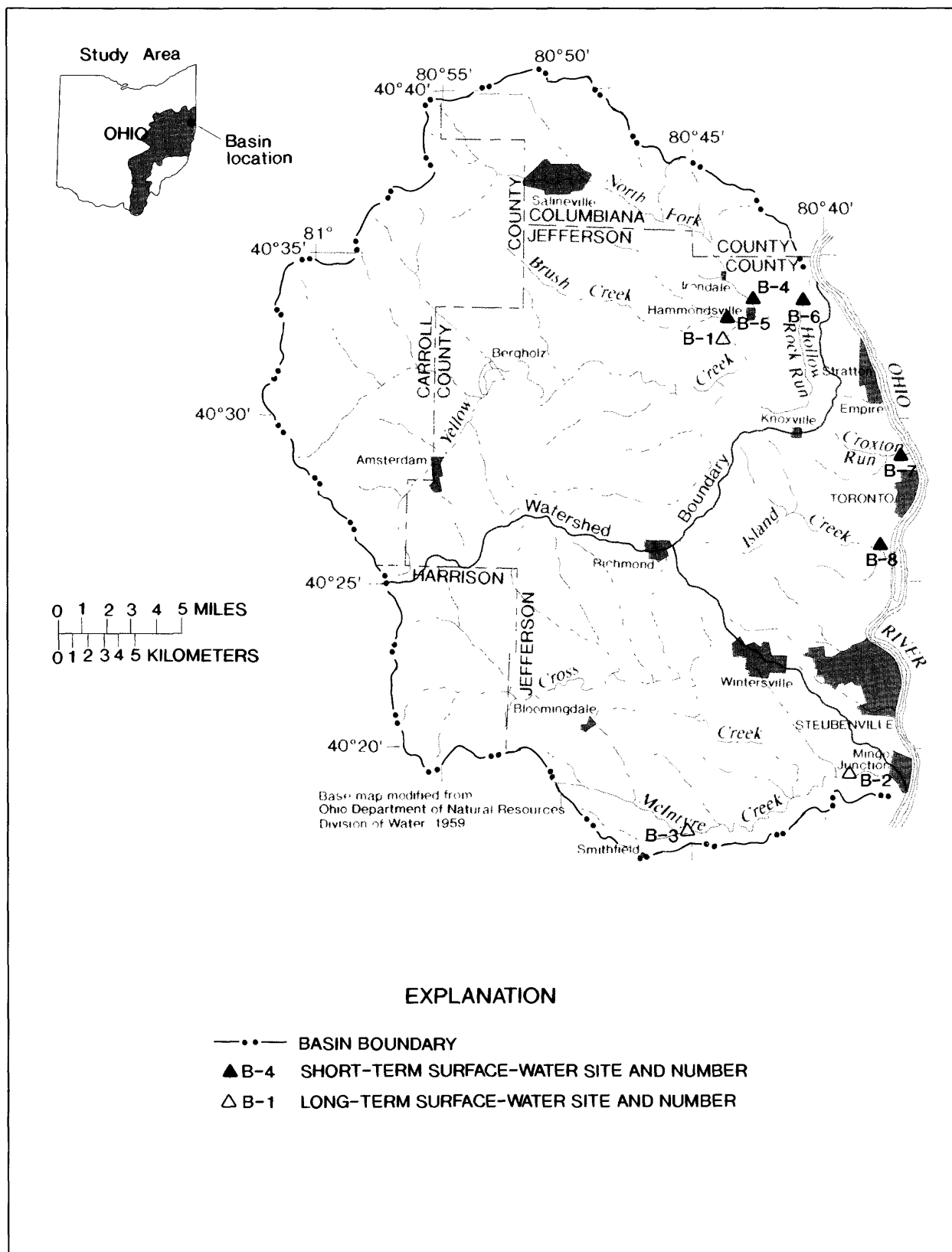


Figure 16.--Yellow and Cross Creeks basin (B) and streamwater sites.

In many places, the Allegheny Formation is not far below the surface, but, currently (1989), there are no underground mines working in the basin (Lopez, 1991). Topographic maps reveal that much of the land surface has been modified by surface mining in the Yellow and Cross Creeks basin. Judging from the length of the list of pending permits to mine coal (Vanessa Tolliver, Ohio Department of Natural Resources, Division of Reclamation, written commun., 1990), interest in the surface mining of coal in the basin remains high. Most of the permits are to mine the No. 8 coal, but, in the northwest and west, there is substantial interest in the Middle Kittanning (No. 6) and Upper Freeport (No. 7) coals in the Allegheny Formation and the Mahoning (7A) and Harlem (7B), both of which are minor coals in the Conemaugh Formation.

Within the stratigraphic sections described above, localized sources of water are present in sandstones; thin, fractured limestones; and even in coal beds. Although the study basin is populated by many users of these sources, according to ground-water-resource maps published by ODNR for the area (Crowell, 1978, 1980; Schmidt, 1959), well yields are generally less than 2 gal/min. In contrast, sands and gravels of the highly productive Ohio Valley aquifer at the eastern edge of the basin can yield more than 1,000 gal/min to wells.

Water Quality

Three long-term and five short-term sites were selected to study streamwater quality in Yellow and Cross Creeks basin. Except for the Ohio River valley aquifer, the Yellow and Cross Creeks basin does not contain a shallow, productive aquifer. Therefore, ground-water quality was not investigated in this basin.

The streamwater sites in the table on the next page were sampled in June 1991. Their locations are shown in figure 16.

Flood-control structures on the Ohio River cause backwater conditions along the downstream reaches of the major tributaries to the

Ohio River. Therefore, the three long-term sites were placed where flow was strong enough for discharge measurements and sampling. Two short-term sites, B-7 and B-8 (fig. 16), were selected to represent water quality in the unnamed area of streams draining directly to the Ohio River. Site B-8 was near the lower end of Island Creek, the largest stream in the unnamed drainage area. Sites B-4, B-5, and B-6 were placed to assess water quality in streams draining the northern part of the study basin.

Water-quality data for surface-water sites show that specific conductance was greater than 1,000 $\mu\text{S}/\text{cm}$ in samples from sites B-2, B-3, B-6, B-7, and B-8 (table 13). The median specific conductance for all sites within the basin was 1,025 $\mu\text{S}/\text{cm}$ (table 14). Water from sites B-2, B-3, B-6, and B-7 had high dissolved-sulfate concentrations (greater than 250 mg/L), and among four sites, water from site B-3 had the highest dissolved-sulfate concentration (1,200 mg/L). None of the concentrations of total recoverable and dissolved iron, aluminum, or manganese were excessively high except at site B-7, where the concentration of total recoverable iron was greater than 1,000 $\mu\text{g}/\text{L}$. Values for pH were in the alkaline range; the median was 8.0. Except for one detached value each for total recoverable and dissolved manganese, all of the values of constituents at surface-water sites were within the interquartile ranges and whiskers (fig. 17).

Therefore, although the concentrations of sulfate and iron were high enough at some sites to be indicative of acid mine drainage, alkaline pH may indicate buffering of streamwaters. The area is underlain by Upper Pennsylvanian units that are more calcareous than those in western reaches of the basin and that may have buffered the acidity of the streams.

Map index number	Site type	Site name	Drainage area (square miles)
B-1	Long term	Yellow C nr Hammondsville	147.0
B-2	do.	Cross C nr Mingo Junction	118.0
B-3	do.	McIntyre C nr Smithfield	14.3
B-4	Short term	NF Yellow C at Hammondsville	59.4
B-5	do.	Brush C at Hammondsville	15.3
B-6	do.	Hollow Rock Rn nr Hammondsville	9.9
B-7	do.	Croxton Rn at Toronto	9.0
B-8	do.	Island C nr Toronto	22.5

Walhonding River Basin

The Walhonding River basin drains an area of 449 mi² that includes the north-central part of Coshocton County and the south-central part of Holmes County (fig. 1, basin J). The basin designated in this study as the Walhonding River basin is considered only part of the Walhonding River basin in the Ohio Water Plan Inventory series (Pree, 1962a). The basin receives drainage from the west, where the Mohican River and Kokosing River join to form the Walhonding River (fig. 18). On the north, drainage from the Upper Killbuck basin enters the Walhonding basin along Killbuck Creek just upstream from Paint Creek, about 5 mi north of Millersburg (fig. 18). Killbuck Creek more or less bisects the basin before it joins the Walhonding River; from that point, the Walhonding River flows eastward across the southern part of the basin. About 6 mi further downstream, at the south-eastern corner of the basin, the Walhonding River joins the Tuscarawas River at Coshocton, where the system becomes the Muskingum River.

About one-quarter or less of the basin was covered by the Killbuck lobe of the last (Wisconsinan) glaciation. The glacial boundary (fig. 18) also separates the Glaciated

Allegheny Plateau and Unglaciaded Allegheny Plateau Sections of the Appalachian Plateaus Physiographic Province.

The nonglaciaded, southern three-quarters of the basin is largely a maturely dissected upland, where local relief is as great as 300 ft. The Walhonding River has cut a fairly wide valley in which the present-day stream follows a meandering path. In its course across the basin, the Walhonding River drops about 80 ft in 21 mi. The valley of the lower part of Killbuck Creek has developed a somewhat incised meander through the unglaciaded upland. Within the flood plain itself, the present course of Killbuck Creek follows a meandering path. The gradient of Killbuck Creek across the basin is about 2 ft/mi, compared to about 4 ft/mi for the Walhonding River. Thus, much of Killbuck Creek is very slow moving, and wetland areas along its course are common. Elsewhere in the nonglaciaded area, the topography in the smaller valleys is more rugged, and streamflow is faster.

In the north, glaciation has subdued the land surface to varying degrees. Data from representative wells (Crowell, 1979) indicate that the drift is only a few feet thick in places, but many of the valleys contain 150 ft or more of fill.

Table 13. Water-quality data for streamwater sites in Yellow and Cross Creeks basin, June 1991

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
June 7, 1991	03110000 B-1 22	YELLOW C NR HAMMONDSVILLE OH (LAT 40 32 16N LONG 080 43 31W) 540	8.4	20.5	--	93	180
June 17, 1991	401857080391700 B-2 35	CROSS C NR MINGO JUNCTION OH (LAT 40 18 57N LONG 080 39 17W) 1,550	8.0	26.0	--	117	850
June 17, 1991	401716080451300 B-3 7.9	MCINTYRE C NR SMITHFIELD OH (LAT 40 17 16N LONG 080 45 13W) 2,100	7.9	26.5	--	204	1,200
June 7, 1991	03110600 B-4 7.2	N F YELLOW C AT HAMMONDSVILLE OH (LAT 40 33 27N LONG 080 42 20W) 510	8.2	20.0	--	66	160
June 7, 1991	403257080430600 B-5 2.3	BRUSH C AT HAMMONDSVILLE OH (LAT 40 32 57N LONG 080 43 06W) 365	7.5	19.5	--	51	120
June 7, 1991	403339080400900 B-6 3.3	HOLLOW ROCK RN NR HAMMONDSVILLE OH (LAT 40 33 39N LONG 080 40 09W) 1,200	8.3	21.5	--	139	590
June 12, 1991	402852080361800 B-7 4.0	CROXTON RN AT TORONTO OH (LAT 40 28 52N LONG 080 36 18W) 1,030	7.9	19.0	--	136	410
June 12, 1991	402610080375700 B-8 18	ISLAND C NR TORONTO OH (LAT 40 26 10N LONG 080 37 57W) 1,020	8.1	22.0	--	139	180

DATE	Aluminum, total recoverable (µg/L as Al)	Aluminum, dissolved (µg/L as Al)	Iron, total recoverable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manganese, total recoverable (µg/L as Mn)	Manganese, dissolved (µg/L as Mn)
June 7, 1991	03110000 B-1 220	YELLOW C NR HAMMONDSVILLE OH (LAT 40 32 16N LONG 080 43 31W) 40	660	20	40	10
June 17, 1991	401857080391700 B-2 560	CROSS C NR MINGO JUNCTION OH (LAT 40 18 57N LONG 080 39 17W) 140	450	140	200	180
June 17, 1991	401716080451300 B-3 120	MCINTYRE C NR SMITHFIELD OH (LAT 40 17 16N LONG 080 45 13W) 60	170	100	40	30
June 7, 1991	03110600 B-4 120	N F YELLOW C AT HAMMONDSVILLE OH (LAT 40 33 27N LONG 080 42 20W) 50	450	120	70	60
June 7, 1991	403257080430600 B-5 40	BRUSH C AT HAMMONDSVILLE OH (LAT 40 32 57N LONG 080 43 06W) 10	80	240	30	40
June 7, 1991	403339080400900 B-6 200	HOLLOW ROCK RN NR HAMMONDSVILLE OH (LAT 40 33 39N LONG 080 40 09W) 120	220	<10	30	30
June 12, 1991	402852080361800 B-7 350	CROXTON RN AT TORONTO OH (LAT 40 28 52N LONG 080 36 18W) 40	1,100	30	80	60
June 12, 1991	402610080375700 B-8 660	ISLAND C NR TORONTO OH (LAT 40 26 10N LONG 080 37 57W) 70	770	120	40	30

Table 14. Ranges and medians for selected water-quality characteristics for streamwater sites in Yellow and Cross Creeks basin, June 1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 16)	Median
Specific conductance, in µS/cm-----	365 to 2,100	B-5; B-3	1,025
pH-----	7.5 to 8.4	B-5; B-1	8.0
Alkalinity, in mg/L as CaCO ₃ -----	51 to 204	B-5; B-3	127
Sulfate, dissolved, in mg/L as SO ₄ -----	120 to 1,200	B-5; B-3	295
Aluminum, total, in µg/L as Al-----	40 to 660	B-5; B-8	210
Aluminum, dissolved, in µg/L as Al-----	10 to 140	B-5; B-2	55
Iron, total, in µg/L as Fe-----	80 to 1,100	B-5; B-7	450
Iron, dissolved, in µg/L as Fe-----	<10 to 240	B-6; B-5	110
Manganese, total, in µg/L as Mn-----	30 to 200	B-5; B-6	40
Manganese, dissolved, in µg/L as Mn-----	10 to 180	B-1; B-2	35

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

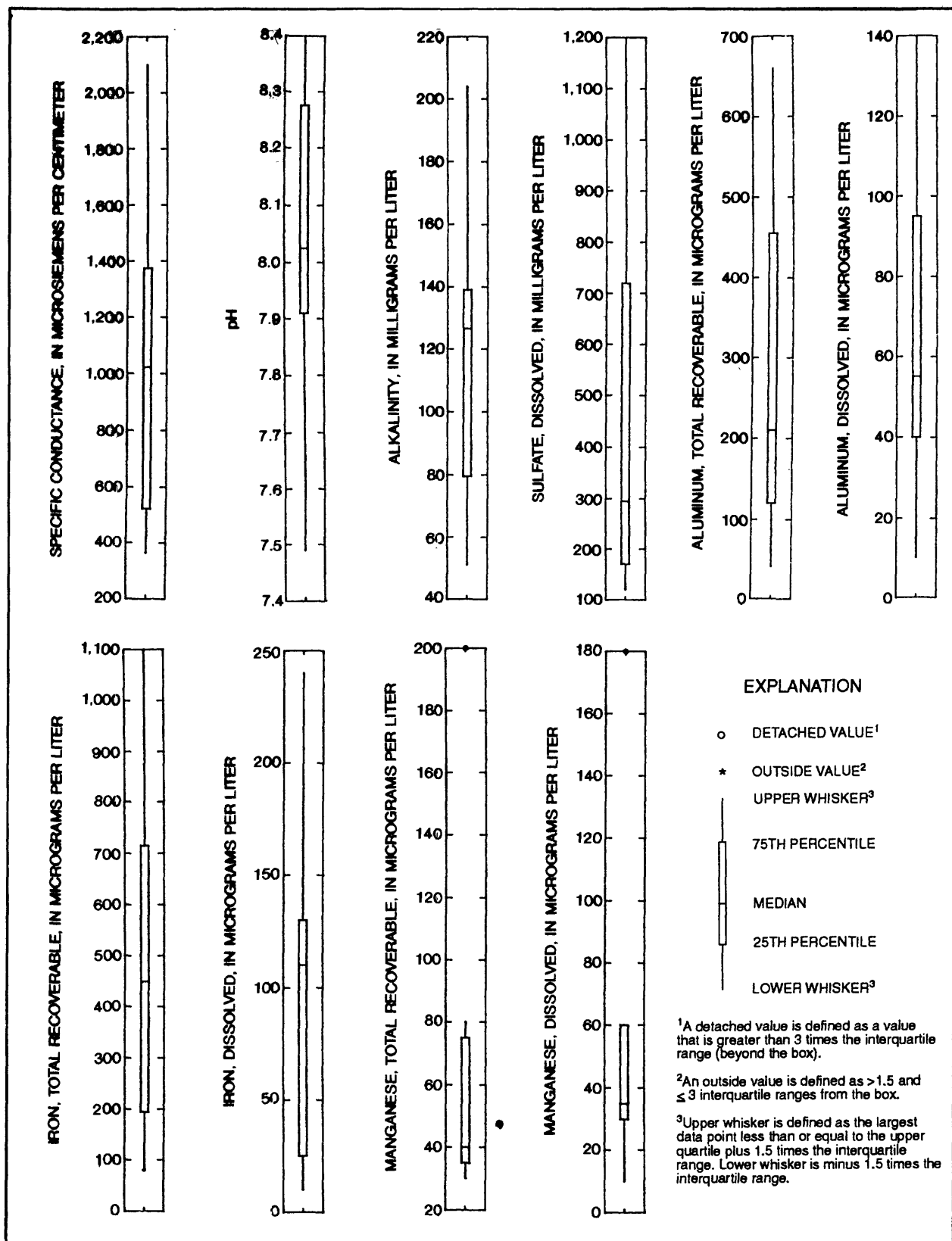


Figure 17.—Ranges, percentiles, and medians of constituents at eight streamwater sites in the Yellow and Cross Creeks basin.

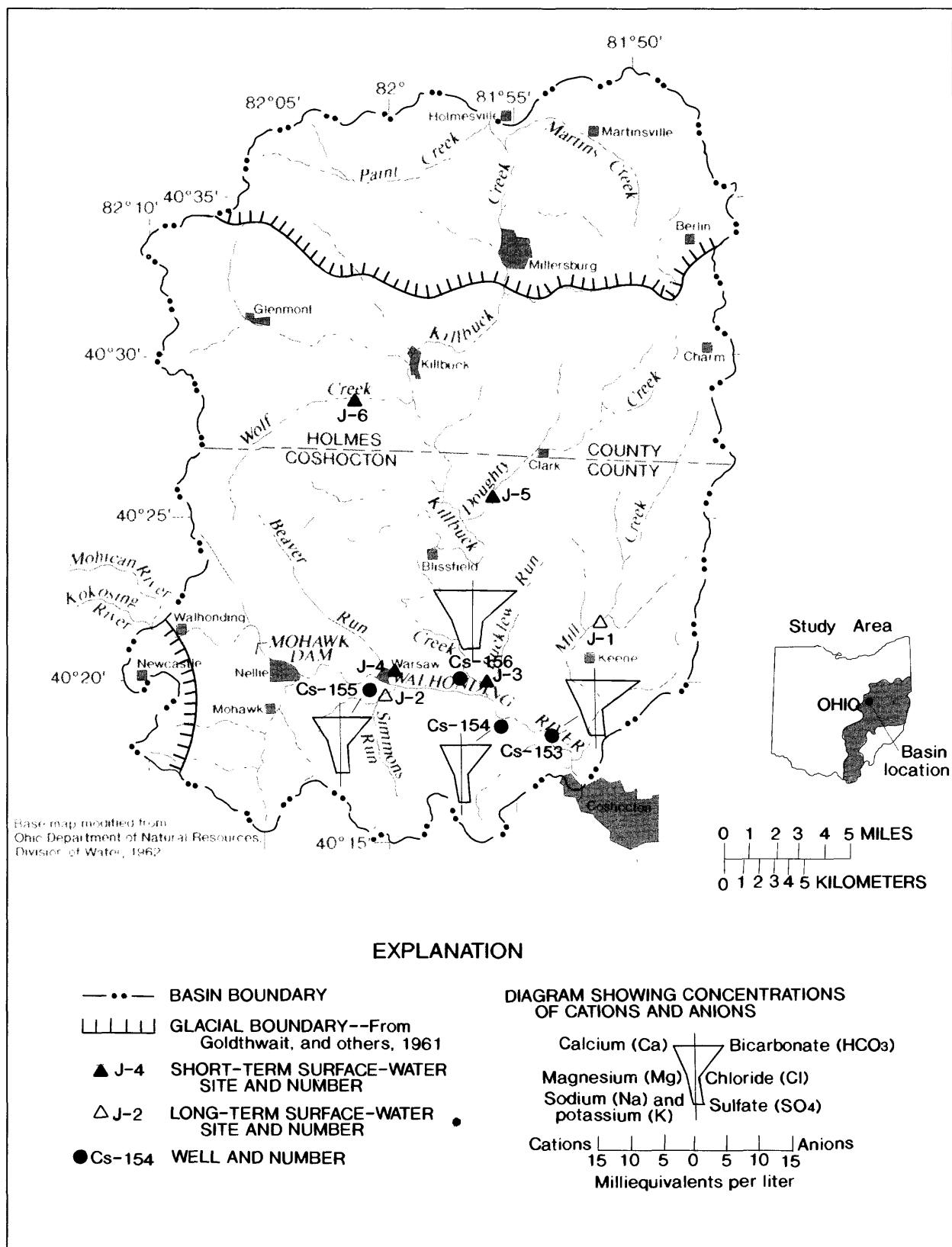


Figure 18.--Walhonding River basin (J), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.

Geologic Setting

Major bedrock units underlying the basin, in ascending order of age, consist of shales of Mississippian age and sandstones and coal-bearing strata of Pennsylvanian age. The oldest unit is the Cuyahoga Formation, of Mississippian age, which lies below a covering of glacial drift in the valley floor of Killbuck Creek at the northern end of the basin. Elsewhere, the overlying Logan Formation (Mississippian) forms many of the valley floors in the nonglaciaded area. To the south and east, the stratigraphic section continues upward with the Pottsville, Allegheny, and Conemaugh Formations of Pennsylvanian age (fig. 4). Only small remnants of the Conemaugh are present in the southeastern part of the basin. Structurally, the stratigraphic section shows a gentle eastward regional dip.

Most of the coal produced in the basin is from the Allegheny Formation, which is present in patches over much of the area. On the basis of information given by Lopez (1991), the Lower Kittanning (No. 5) and the Middle Kittanning (No. 6) coals are the principal units being mined. Other Allegheny units being mined include the Brookville (No. 4) and the Strasburg (No. 5A). The Upper Freeport (No. 7) coal at the top of the Allegheny Formation is missing over much of the area because of erosion. Three Pottsville coals that are present not far below the base of the Allegheny Formation—the Lower Mercer (No. 3), Upper Mercer (No. 3A), and the Tionesta (No. 3B)—are mined locally. Because of their localized presence, not all of the above-mentioned coals are listed in figure 4.

Coal output in 1989 included production from the No. 4 coal near Millersburg and from the No. 5 and No. 6 coals in Coshocton County between Doughty and Mill Creeks (fig. 18).

A number of sites across the basin are under permit to extract either the No. 5 or No. 6 coal. Most of these are in the southeastern part of the basin not far from Coshocton.

Unconsolidated and bedrock aquifers are important water supplies in the basin. The unconsolidated aquifers consist of thick deposits of sand and gravel that are present in much of the glaciaded area and are largely restricted to outwash valleys south of the glaciaded area. Well yields from these deposits can exceed 1,000 gal/min; for bedrock aquifers, however, yields are generally less than 25 gal/min (Crowell, 1979; Sugar, 1988). The Cuyahoga and Logan Formations of Mississippian age are a bedrock source of water along the western parts of the basin, and many wells in this area that yield from these units are more than 400 ft deep. The use of these bedrock sources is limited in places by depth or a change in rock composition from sandstone to shale. In Coshocton County, the Pottsville Formation is a common bedrock source of water supply. Also available are shallow local supplies in sandstone, limestone, and coal units of the Allegheny Formation.

Water Quality

In addition to two long-term streamwater sites, five short-term sites were selected for a synoptic sampling of the basin. Ground-water samples were collected from four wells in the southern third of the basin, the area of most of the surface-mining activity.

The streamwater sites in the table on the next page were sampled in June 1991 because of high flow in fall 1990. Their locations are shown in figure 18.

The Walhonding River and Killbuck Creek receive substantial flow from drainage basins upstream. Sampling sites on these streams would not be diagnostic of conditions within the basin.

Streamwater

Map index number	Site type	Site name	Drainage area (square miles)
J-1	Long term	Mill C nr Coshocton	27.2
J-2	do.	Simmons Rn nr Warsaw	16.4
J-3	Short term	Bucklew Rn nr Warsaw	8.0
J-4	do.	Beaver Rn at Warsaw	13.6
J-5	do.	Doughty C nr Clark	59.7
J-6	do.	Wolf C nr Killbuck	23.1
J-7	do.	Martins C nr Holmesville	22.9

Therefore, the long-term sites, J-1 and J-2, were established on principal tributaries. At site J-1, the USGS has operated a streamflow-gaging station since 1936.

Because backwater along the lower reaches of most tributaries to Killbuck Creek was common, sites J-3, J-5, and J-6 were located upstream at places where a suitable discharge measurement could be made. All short-term sites were on streams that drain areas believed to contain coal-bearing strata. Some of the sites permitted for mining are included in these areas.

Results of analyses of water samples (table 15) show that, of the seven sites sampled, the highest specific conductances were found at sites J-2 (1,290 $\mu\text{S}/\text{cm}$) and J-3 (825 $\mu\text{S}/\text{cm}$). Dissolved-sulfate concentrations were also at or near the maximum in water from these two sites; water from site J-2 had the highest dissolved-sulfate concentration (700 mg/L) (table 16). High concentrations of aluminum, iron, and manganese, often associated with acid mine drainage, were not found in any waters sampled in the basin. Values for pH were in the alkaline range at all sites; the median was 7.8 (fig. 19).

Ground water

Water from four wells opening into the shallow, unconsolidated aquifer underlying the Walhonding Valley was sampled. The town of Warsaw and the city of Coshocton draw their supply from the same aquifer.

Water-quality data for four ground-water sites sampled in August 1990 are shown in table 17. Values of pH ranged from 7.2 to 7.7, and concentrations of constituents commonly associated with acid mine drainage were not high. The Stiff diagrams (fig. 18) show that all four wells contain calcium bicarbonate-type waters.

The water from wells Cs-153, Cs-154, and Cs-155 met all OEPA water-quality standards for public supplies for the constituents reported. The concentration of dissolved solids in water from well Cs-156 only slightly exceeded the water-quality standard (500 mg/L). The waters are classified as hard to very hard.

Table 15. Water-quality data for streamwater sites in Walhonding River basin, June 1991

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conduc- tance (μS/cm)	pH (standard units)	Temper- ature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
June 6, 1991	03140000 J-1 2.5	MILL C NR 380	COSHOCTON OH 7.8	(LAT 40 21 46N LONG 081 51 45W) 19.0	--	101	56
June 10, 1991	401936082001400 J-2 1.1	SIMMONS RN NR 1,290	WARSAW OH 7.6	(LAT 40 19 36N LONG 082 00 14W) 21.5	--	100	700
June 10, 1991	401955081561800 J-3 1.1	BUCKLEW RN NR 825	WARSAW OH 7.8	(LAT 40 19 55N LONG 081 56 18W) 23.0	--	76	380
June 10, 1991	402006082000100 J-4 3.6	BEAVER RN AT 225	WARSAW OH 7.9	(LAT 40 20 06N LONG 082 00 01W) 21.5	--	63	18
June 10, 1991	402500081563200 J-5 11	DOUGHTY C NR 400	CLARK OH 8.0	(LAT 40 25 00N LONG 081 56 32W) 24.5	--	117	48
June 10, 1991	402843082012600 J-6 5.8	WOLF C NR 250	KILLBUCK OH 7.8	(LAT 40 28 43N LONG 082 01 26W) 21.0	--	86	24
June 10, 1991	403655081550200 J-7 2.8	MARTINS C NR 500	HOLMESVILLE OH 7.9	(LAT 40 36 55N LONG 081 55 02W) 22.0	--	180	55

DATE	Aluminum, total recoverable (μg/L as Al)	Aluminum, dissolved (μg/L as Al)	Iron, total recoverable (μg/L as Fe)	Iron, dissolved (μg/L as Fe)	Manganese, total recoverable (μg/L as Mn)	Manganese, dissolved (μg/L as Mn)
June 6, 1991	03140000 J-1 160	MILL C NR 10	COSHOCTON OH 1,300	(LAT 40 21 46N LONG 081 51 45W) 130	220	220
June 10, 1991	401936082001400 J-2 120	SIMMONS RN NR 10	WARSAW OH 420	(LAT 40 19 36N LONG 082 00 14W) 110	230	190
June 10, 1991	401955081561800 J-3 140	BUCKLEW RN NR 20	WARSAW OH 280	(LAT 40 19 55N LONG 081 56 18W) 110	220	240
June 10, 1991	402006082000100 J-4 240	BEAVER RN AT 20	WARSAW OH 630	(LAT 40 20 06N LONG 082 00 01W) 190	40	40
June 10, 1991	402500081563200 J-5 110	DOUGHTY C NR 20	CLARK OH 570	(LAT 40 25 00N LONG 081 56 32W) 60	100	100
June 10, 1991	402843082012600 J-6 110	WOLF C NR 20	KILLBUCK OH 570	(LAT 40 28 43N LONG 082 01 26W) 180	50	70
June 10, 1991	403655081550200 J-7 80	MARTINS C NR <10	HOLMESVILLE OH 130	(LAT 40 36 55N LONG 081 55 02W) 130	10	20

Table 16. Ranges and medians for selected water-quality characteristics for streamwater sites in Walhonding River basin, June 1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 18)	Median
Specific conductance, in µS/cm-----	225 to 1,290	J-4; J-2	400
pH-----	7.6 to 8.0	J-2; J-2	7.8
Alkalinity, in mg/L as CaCO ₃ -----	63 to 180	J-4; J-7	100
Sulfate, dissolved, in mg/L as SO ₄ ----	18 to 700	J-4; J-2	55
Aluminum, total, in µg/L as Al-----	80 to 240	J-7; J-4	120
Aluminum, dissolved, in µg/L as Al-----	<10 to 20	J-7; several sites	20
Iron, total, in µg/L as Fe-----	130 to 1,300	J-2; J-1	570
Iron, dissolved, in µg/L as Fe-----	60 to 190	J-5; J-4	130
Manganese, total, in µg/L as Mn-----	10 to 230	J-7; J-2	100
Manganese, dissolved, in µg/L as Mn-----	20 to 240	J-7; J-2	100

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

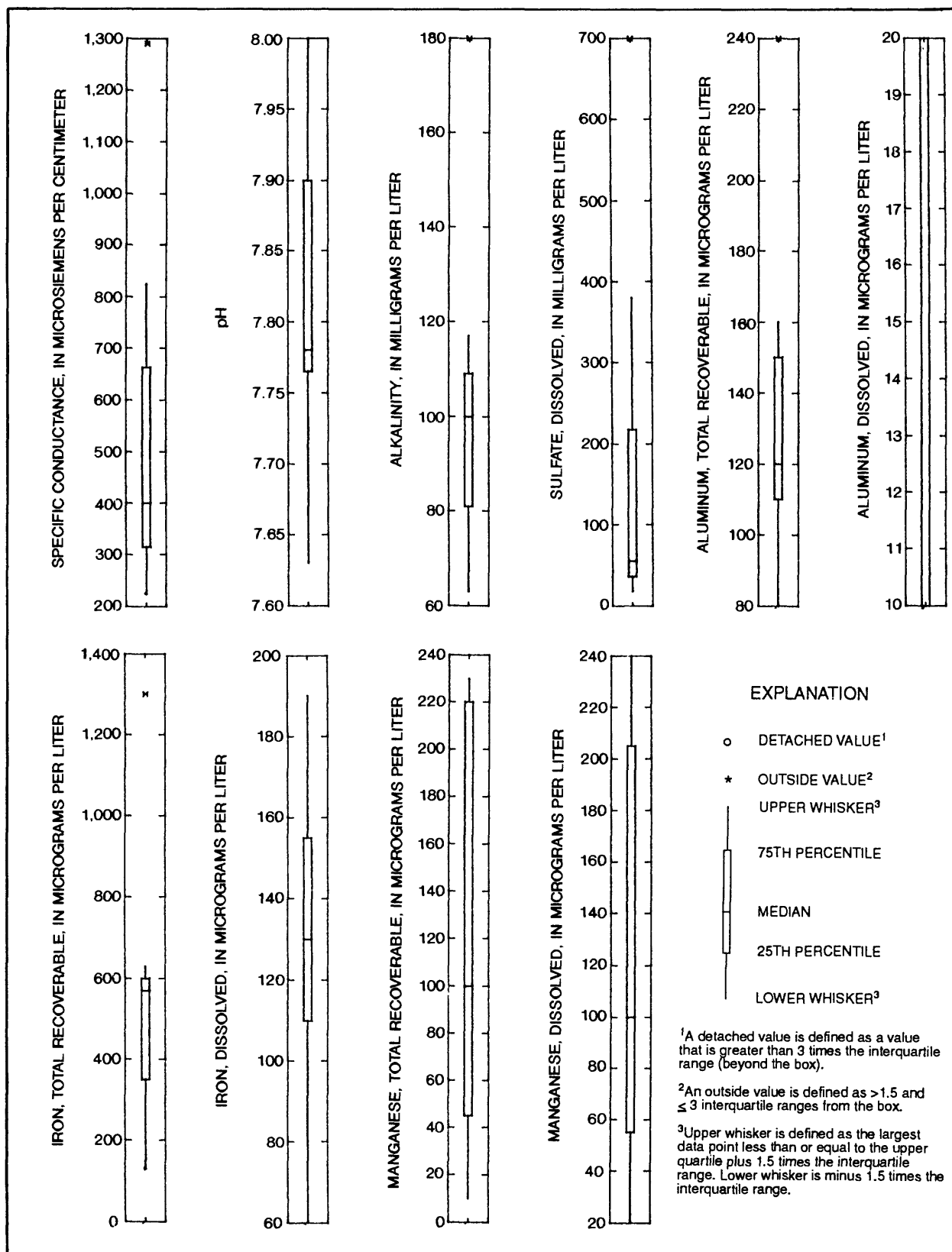


Figure 19.—Ranges, percentiles, and medians of constituents at seven streamwater sites in the Walhonding River basin.

Table 17. Water-quality data for ground-water sites in Walhonding River basin, August 1990

[°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Depth to water below land-surface datum (feet)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Oxygen, dissolved (mg/L)	Hardness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Magnesium, dissolved (mg/L as Mg)	Sodium, dissolved (mg/L as Na)	Potassium, dissolved (mg/L as K)	Alkalinity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Aug. 21, 1990	14.21	635	7.4	14.0	2.2	310	88	21	17	2.1	292	59
	28.61	460	7.7	15.0	4.5	220	64	14	3.5	1.8	163	30
Aug. 21, 1990	--	640	7.4	13.0	5.0	300	83	22	18	6.1	248	35
	26.87	910	7.2	13.0	3.2	390	110	29	36	2.2	330	41
401816081532400 CS-153 NR COSHOCTON OH (LAT 40 18 16N LONG 081 53 24W)												
401830081553900 CS-154 NR COSHOCTON OH (LAT 40 18 30N LONG 081 55 39W)												
401943082003600 CS-155 NR WARSAW OH (LAT 40 19 43N LONG 082 00 36W)												
402007081572100 CS-156 NR WARSAW OH (LAT 40 20 07N LONG 081 57 21W)												
Date	Chloride, dissolved (mg/L as Cl)	Silica, dissolved (mg/L as SiO ₂)	Solids, residue at 180 deg. C, dissolved (mg/L)	Solids, sum of constituents, dissolved (mg/L)	Aluminum, total, recoverable (µg/L as Al)	Aluminum, dissolved (µg/L as Al)	Iron, total, recoverable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manganese, total, recoverable (µg/L as Mn)	Manganese, dissolved (µg/L as Mn)	Carbon, organic, dissolved (mg/L as C)	
Aug. 21, 1990	38	9.4	398	410	<10	<10	270	30	<10	7	0.9	
Aug. 21, 1990	13	8.1	262	232	<10	<10	<10	5	<10	<1	0.7	
Aug. 21, 1990	29	12	382	354	<10	<10	40	<3	<10	<1	0.8	
Aug. 22, 1990	64	12	540	492	<10	<10	120	15	<10	1	0.8	
401816081532400 CS-153 NR COSHOCTON OH (LAT 40 18 16N LONG 081 53 24W)												
401830081553900 CS-154 NR COSHOCTON OH (LAT 40 18 30N LONG 081 55 39W)												
401943082003600 CS-155 NR WARSAW OH (LAT 40 19 43N LONG 082 00 36W)												
402007081572100 CS-156 NR WARSAW OH (LAT 40 20 07N LONG 081 57 21W)												

Upper Muskingum River Basin

The Upper Muskingum River basin is an irregularly shaped drainage area of 366 mi² (fig. 1, basin M). It includes what is approximately the southwestern third of Coshocton County, small parts of Knox and Licking Counties, and an irregularly shaped part of the northwestern corner of Muskingum County (fig. 20).

The Upper Muskingum River basin, as designated in this study, is the same area referred to as "part of Upper Portion of the Muskingum River basin" in the ODNR Ohio Water Plan Inventory series (Schmidt, 1962b). It includes the section of the Muskingum River that begins at the confluence of the Tuscarawas and Walhonding Rivers at Coshocton and extends to, but does not include, the confluence of the Licking River at Zanesville. Two major sources of outside flow to the basin are the Tuscarawas-Walhonding systems on the north and the Wills Creek system, which enters the Muskingum River near where it enters Muskingum County (fig. 20).

The largest interior drainage network is the Wakatomika Creek system, which enters the Muskingum River near Dresden (fig. 20). It drains all but a small part of the northwestern two-thirds of the basin. Except for Symmes Creek, east of Dresden (fig. 20), most of the Muskingum tributaries are less than 3 mi in length.

About 50 mi² of the northwestern extremity of the basin was covered by older (Illinoian) glaciation. Postglacial erosion has left the landscape similar in appearance to the nonglaciated section. The glacial boundary is the basis of dividing the subject basins between the Glaciated and Unglaciated Allegheny Plateau Sections of the Appalachian Plateaus Physiographic Province.

The topographic character for most of the basin is basically that of a thoroughly dissected upland through which the main stream of the Muskingum drainage system has cut a fairly wide path between Coshocton and Dresden. Southward, toward Zanesville, the Muskingum Valley narrows considerably. A wide valley that was formed by glacial or preglacial drainage systems is now largely filled with unconsolidated material through which Wakatomika Creek follows a meandering course for about 15 mi upstream from its confluence with the Muskingum River. In contrast, the smaller tributaries draining upland areas are relatively short with steep-sided valleys. Local relief in the wide bottom valleys is negligible but, along the upland edges, can be as great as 300 ft.

Geologic Setting

Major bedrock underlying the basin, in ascending order of age, consists of sandstones and shales of Mississippian age and coal-bearing strata of Pennsylvanian age. The oldest unit is the Cuyahoga Formation, a prominent member of which is the Black Hand Sandstone. Outcrops of the Black Hand Sandstone are visible in western parts of the basin.

Toward the east, the Cuyahoga Formation and the overlying Logan Formation form the valley bottoms of northwestern Muskingum County. The overlying Pottsville Formation forms much of the upland area north and south of Wakatomika Creek and much of the southwestern corner of Coshocton County. Overlying the Pottsville and capping many of the hills in the western part of the basin are remnants of the Allegheny Formation that contain mineable coals. Near the Muskingum River in Coshocton County and east of the river in Muskingum County, the Conemaugh Formation is present as scattered remnants that thicken to the east.

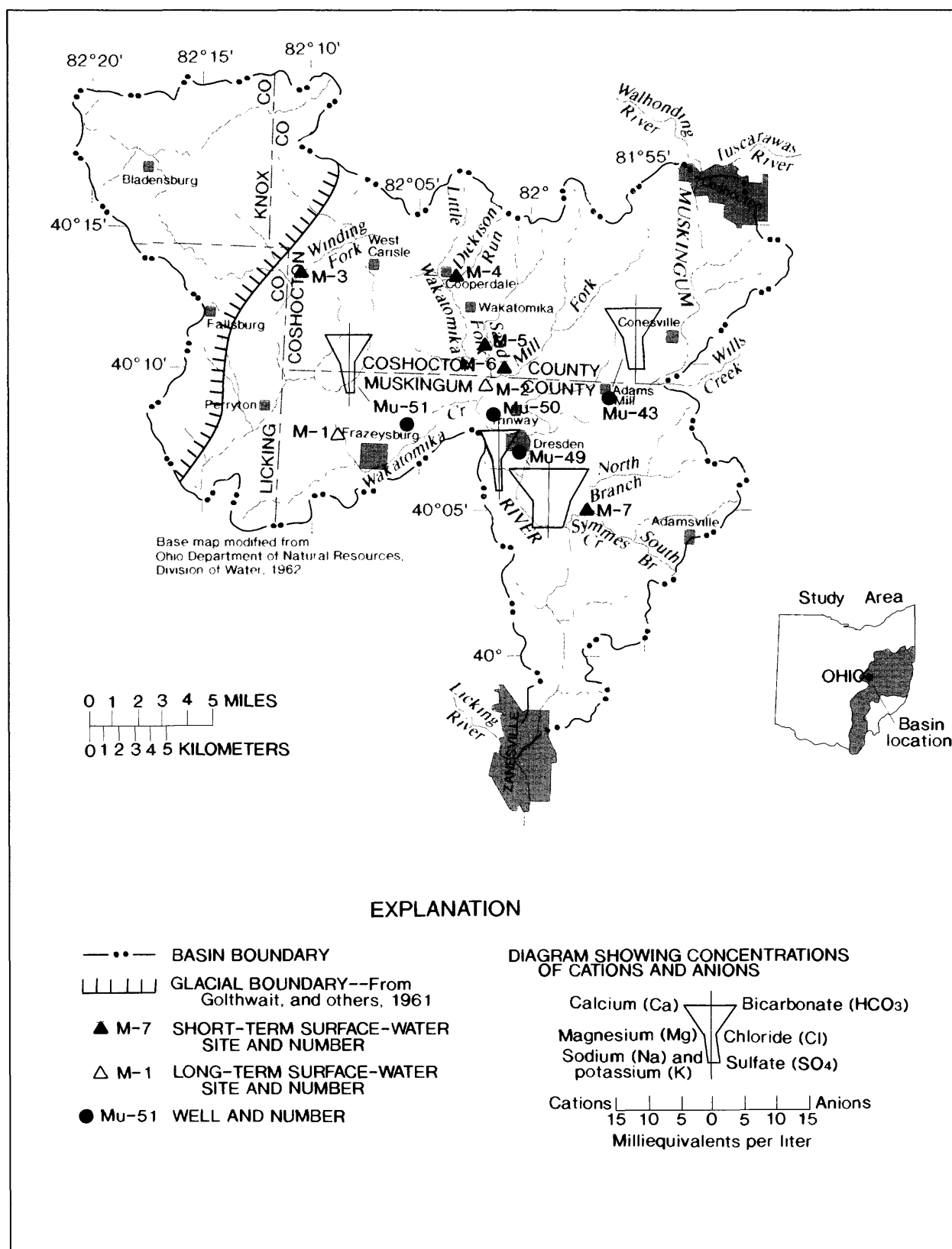


Figure 20.--Upper Muskingum River basin (M), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.

Recent coal production (Lopez, 1991) is mostly confined to the Middle Kittanning (No. 6) coal east of the Muskingum River northeast of Zanesville, including one underground mine. Many areas have been mined out, but numerous areas up and down the eastern part of the basin are under permit to mine the No. 6 coal. The permit list includes as objectives the Lower Kittanning (No. 5) and Upper Freeport (No. 7) coals.

Unconsolidated aquifers are the source of an important water supply in the basin. In places, the valley fill is more than 100 ft thick along the Muskingum River between Coshocton and Dresden (fig. 20). Unconsolidated valley fill can yield as much as 1,000 gal/min in this area, where the Muskingum River is a source of induced recharge (Schmidt, 1962b). Westward along Wakatomika Creek and into Licking County, unconsolidated materials underlie the main valley. Yields to wells in this area are sufficient for small public supplies.

Bedrock aquifers also are important as a water supply in the basin. Maps by Schmidt (1962b, 1980), Hartzell (1982), and Sugar (1988) show possible yields of 10 to 25 gal/min for bedrock areas of the basin that are underlain by sandstones of the Cuyahoga, Logan, and Pottsville Formations. In contrast, through the center of the basin, yields to wells are meager (yields generally less than 3 gal/min). In this upland area, the upper part of the Allegheny Formation and the lower part of the Conemaugh Formation form the surficial terrain. The sandstones that yield more water are more deeply buried.

Water Quality

Two long-term streamwater sites and five short-term sites were selected for a synoptic sampling of the basin. In addition, ground-

water samples were collected from four wells in the northern part of Muskingum County.

Because of high flow in 1990, the streamwater sites in the table on the next page were sampled in early June 1991. Their locations are shown in figure 20.

Two long-term streamwater sites (fig. 20, M-1 and M-2) were placed along the Wakatomika drainage system, an area that was formerly mined. Short-term sites were selected to maximize synoptic coverage of the basin.

Water-quality data for streamwater sites show that specific conductance was greater than 1,000 $\mu\text{S}/\text{cm}$ in samples from sites M-2, M-5, and M-6 (table 18). The median specific conductance for all sites within the basin was 965 $\mu\text{S}/\text{cm}$ (table 19). Dissolved-sulfate concentrations were especially high at site M-6 but were also high at sites M-2, M-5, and M-7. Concentrations of total recoverable and dissolved manganese were greater than 1,000 $\mu\text{g}/\text{L}$ at sites M-2 and M-6. Values for pH were in the slightly alkaline range; the median was 8.0. In figure 21, outside values are shown for total recoverable and dissolved manganese (site M-6), total recoverable aluminum (site M-6), and dissolved iron (site M-3) (fig. 21). Because dissolved-aluminum concentrations were below detection limits at four out of seven sites, box plots for this constituent were not constructed.

Water from site M-6, and, to a lesser extent, from sites M-2, M-5, and M-7, had high specific conductance and high concentrations of some of the constituents commonly associated with acid mine drainage. For all sites in the basin, however, pH in the alkaline range indicates that streamwaters may have been buffered by runoff or diluted by alkaline waters.

Streamwater

Map index number	Site type	Site name	Drainage area (square miles)
M-1	Long term	Wakatomika C nr Frazeyzburg	140.0
M-2	do.	L. Wakatomika C nr Trinway	61.1
M-3	Short term	Winding F Wakatomika C nr W. Carlisle	21.0
M-4	do.	Dickinson Rn at Cooperdale	4.1
M-5	do.	Sand F nr Wakatomika	8.7
M-6	do.	Mill F nr Trinway	24.5
M-7	do.	Symmes C nr Dresden	31.3

Ground water

Ground-water quality was determined by sampling four wells yielding from shallow, unconsolidated sediments that are major industrial and domestic water-supply aquifers in the area. Three of the wells (Mu-43, Mu-49, and Mu-50) were close to the Muskingum River. Well Mu-51, near the town of Frazeyzburg, yields from a buried-valley system not traversed by a large stream at the surface.

Water-quality data for four ground-water sites sampled in August 1990 are given in table 20. Values for pH ranged from 7.4 to 7.9, and concentrations of constituents associated with acid mine drainage were not high.

As shown on the Stiff diagram (fig. 20), well Mu-50 contains calcium bicarbonate-type waters. The ions that dominate the waters from wells Mu-43 and Mu-51 are calcium, bicarbonate, and, to a lesser extent, sulfate (fig. 20). The highly mineralized character of water from well Mu-49 is shown in the Stiff diagram of this well. The ground waters are classified as moderately hard (Mu-50), hard (Mu-51 and Mu-43), and very hard (Mu-49).

The OEPA water-quality standard for dissolved manganese (50 µg/L) was exceeded in water from wells Mu-49, Mu-50, and Mu-51 (table 20). The water-quality standard for iron (300 µg/L) was exceeded in water from wells Mu-50 and Mu-51. Water from well Mu-49, a public-supply well, also exceeded the water-quality standard for dissolved solids (500 mg/L).

Middle Muskingum River Basin

The Middle Muskingum River Basin, which drains an area of 470 mi², covers approximately the south-central third of Muskingum County, the northeastern third of Morgan County, and a small part of the northwestern corner of Noble County (fig. 1, basin N). The basin corresponds to the same area referred to as "a portion of the Middle Muskingum River Basin" (Walker, 1962a) in the ODNr Water Plan Inventory series maps.

Table 18. Water-quality data for streamwater sites in Upper Muskingum River basin, June 1991

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, Instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)	
June 6, 1991	400710082081000 M-1 WAKATOMICA C NR FRAZEYSBURG OH (LAT 41 06 16N LONG 082 07 55W)	22	330	8.2	21.5	--	82	17
June 6, 1991	400912082014700 M-2 LITTLE WAKATOMICA C NR TRINWAY OH (LAT 40 09 12N LONG 082 01 47W)	12	1,550	7.9	18.5	--	78	900
June 6, 1991	401302082103100 M-3 WINDING F WAKATOMICA C NR W CARLISLE OH (LAT 40 13 02N LONG 082 10 31W)	2.8	300	7.8	19.5	--	66	20
June 6, 1991	401305082034900 M-4 DICKINSON RN AT COOPERDALE OH (LAT 40 13 05N LONG 082 03 49W)	0.29	530	8.0	18.0	--	106	170
June 6, 1991	401059082015100 M-5 SAND F (084) NR WAKATOMIKA OH (LAT 40 10 59N LONG 082 01 51W)	1.8	1,390	8.0	19.0	--	104	770
June 6, 1991	400951082013300 M-6 MILL F NR TRINWAY OH (Lat 40 09 51N LONG 082 01 33W)	7.4	1,780	7.7	17.0	--	60	1,400
June 6, 1991	400458081571400 M-7 SYMMES C NR DRESDEN OH (LAT 40 04 58N LONG 081 57 14W)	4.4	965	8.1	17.0	--	107	490

DATE	Aluminum, total recoverable (μg/L as Al)	Aluminum, dissolved (μg/L as Al)	Iron, total recoverable (μg/L as Fe)	Iron, dissolved (μg/L as Fe)	Manganese, total recoverable (μg/L as Mn)	Manganese, dissolved (μg/L as Mn)	
June 6, 1991	400710082081000 M-1 WAKATOMICA C NR FRAZEYSBURG OH (LAT 41 06 16N LONG 082 07 55W)	170	20	580	130	110	100
June 6, 1991	400912082014700 M-2 LITTLE WAKATOMICA C NR TRINWAY OH (LAT 40 09 12N LONG 082 01 47W)	140	<10	600	<10	1,200	1,200
June 6, 1991	401302082103100 M-3 WINDING F WAKATOMICA C NR W CARLISLE OH (LAT 40 13 02N LONG 082 10 31W)	120	10	1,500	390	190	190
June 6, 1991	401305082034900 M-4 DICKINSON RN AT COOPERDALE OH (LAT 40 13 05N LONG 082 03 49W)	110	<10	270	30	90	90
June 6, 1991	401059082015100 M-5 SAND F (084) NR WAKATOMIKA OH (LAT 40 10 59N LONG 082 01 51W)	100	<10	1,300	20	510	470
June 6, 1991	400951082013300 M-6 MILL F NR TRINWAY OH (Lat 40 09 51N LONG 082 01 33W)	290	10	760	90	2,300	2,100
June 6, 1991	400458081571400 M-7 SYMMES C NR DRESDEN OH (LAT 40 04 58N LONG 081 57 14W)	100	<10	350	10	160	160

Table 19. Ranges and medians for selected water-quality characteristics for streamwater sites in Upper Muskingum River basin, June 1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 20)	Median
Specific conductance, in µS/cm-----	300 to 1,780	M-3; M-6	965
pH-----	7.7 to 8.2	M-6; M-1	8.0
Alkalinity, in mg/L as CaCO ₃ -----	60 to 107	M-6; M-7	82
Sulfate, dissolved, in mg/L as SO ₄ ----	17 to 1,400	M-1; M-6	490
Aluminum, total, in µg/L as Al-----	100 to 290	M-5, 7; M-6	120
Aluminum, dissolved, in µg/L as Al-----	<10 to 20	Several sites; M-1	<10
Iron, total, in µg/L as Fe-----	270 to 1,500	M-4; M-3	600
Iron, dissolved, in µg/L as Fe-----	<10 to 390	M-2; M-3	30
Manganese, total, in µg/L as Mn-----	90 to 2,300	M-4; M-6	190
Manganese, dissolved, in µg/L as Mn-----	90 to 2,100	M-4; M-6	190

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

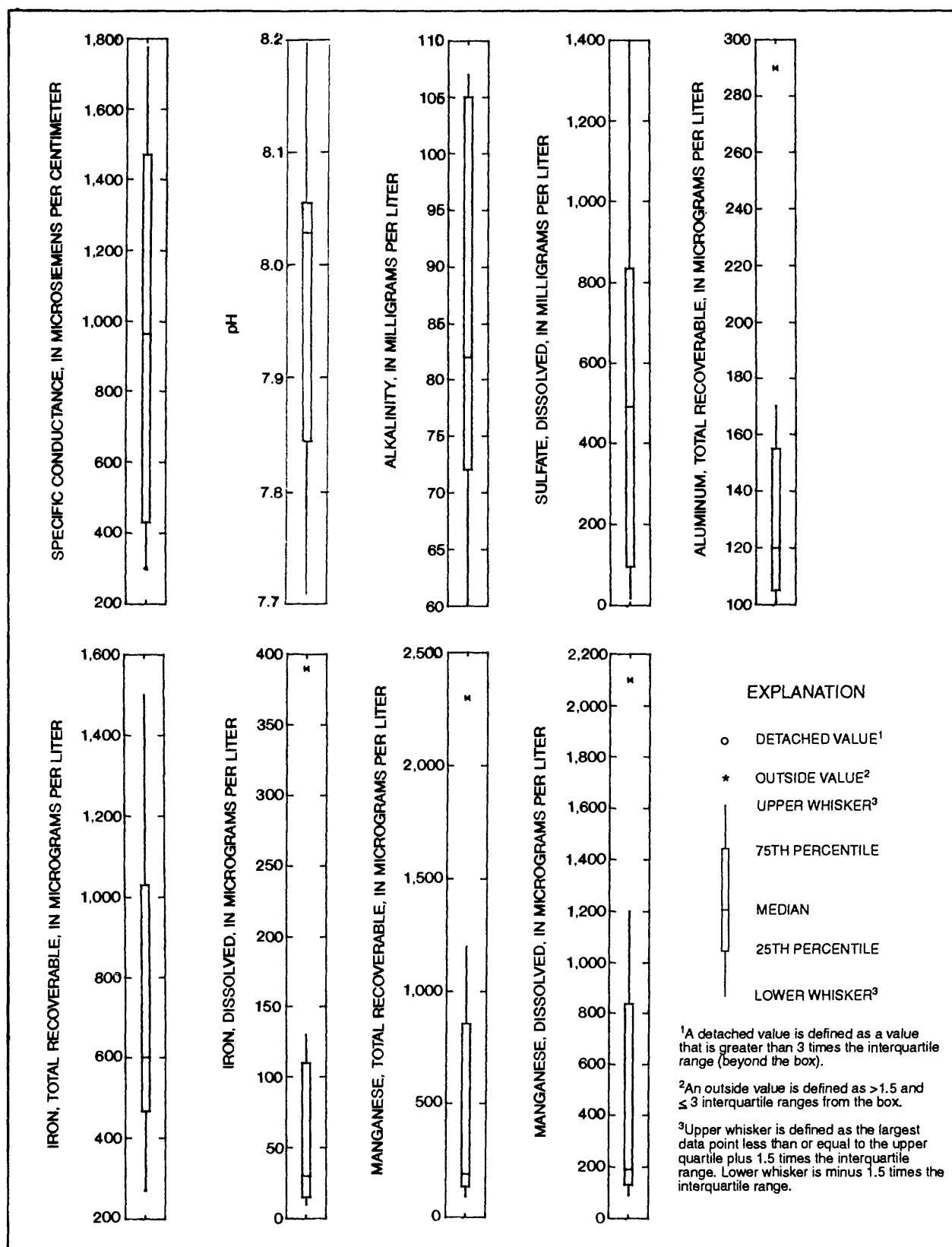


Figure 21.—Ranges, percentiles, and medians of constituents at seven streamwater sites in the Upper Muskingum River basin.

Table 20. Water-quality data for ground-water sites in Upper Muskingum River basin, August 1990

[°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Depth to water below land-surface datum (feet)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Oxygen, dissolved (mg/L)	Hardness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Magnesium, dissolved (mg/L as Mg)	Sodium, dissolved (mg/L as Na)	Potassium, dissolved (mg/L as K)	Alkalinity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Aug. 22, 1990	18.72	615	7.5	13.0	2.0	290	83	19	16	2.7	185	92
Aug. 22, 1990	--	975	7.4	15.0	--	400	120	24	53	5.5	261	88
Aug. 22, 1990	32.46	300	7.9	15.0	3.4	140	43	8.1	4.7	.60	118	17
Aug. 22, 1990	22.16	430	7.7	12.0	.4	230	63	17	4.4	1.0	186	30

Date	Chloride, dissolved (mg/L as Cl)	Silica, dissolved (mg/L as SiO ₂)	Solids, residue at 180 deg. C, dissolved (mg/L)	Solids, sum of constituents, dissolved (mg/L)	Aluminum, total, recoverable (µg/L as Al)	Aluminum, dissolved (µg/L as Al)	Iron, total, recoverable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manganese, total, recoverable (µg/L as Mn)	Manganese, dissolved (µg/L as Mn)	Carbon, organic, dissolved (mg/L as C)
Aug. 22, 1990	23	9.8	367	357	<10	<10	300	150	10	11	0.6
Aug. 22, 1990	110	12	572	570	<10	<10	280	170	400	420	.8
Aug. 22, 1990	8.8	8.9	158	163	30	<10	1,100	380	550	560	3.1
Aug. 22, 1990	9.6	11	230	249	<10	<10	2,100	660	790	830	1.3

The drainage network of the upper end of the Middle Muskingum River basin is somewhat complex. On the eastern bank of the Muskingum River, just northeast of Zanesville (fig. 22), discharge that enters from Mill Run is the most upstream source of drainage from within the Middle Muskingum River basin itself. Upstream from Zanesville, the Muskingum River, which enters from the Upper Muskingum River basin, drains much of the coal-bearing region of Ohio. At the confluence of the Licking River in the center of Zanesville, drainage from the Licking River basin (not part of the study area) enters the Muskingum River. Included in the Middle Muskingum River basin is a small area (3.8 mi²) drained by Chaps Run. Chaps Run drains into the western bank of the Muskingum River and is between the Licking River basin and the Moxahala Creek basin (fig. 1, basin O), which was discussed in an earlier report (Jones, 1988). Thus, the Middle Muskingum River basin receives drainage from three major sources.

Meigs Creek, which enters the Muskingum River at the downstream end of the basin, drains about half of the subject basin. Salt Creek, which enters the Muskingum River at Duncan Falls (fig. 22), drains much of the northern third of the Middle Muskingum River basin. The prevailing direction of streamflow is southward to eastward. The biggest exception is Brush Creek, which flows to the north along the western edge of the basin before entering the Muskingum River midway between Zanesville and Philo (fig. 22).

The Muskingum River valley narrows considerably along its southward approach toward Zanesville. Southward from Duncan Falls and Philo, the Muskingum River flows in a slightly meandering valley that is only a few times the width of the river channel. The valley floor stands in sharp contrast to the adjacent highlands. The same area forms a narrow subbasin along the southwestern side of the Middle Muskingum River basin, within which

numerous short streams that have cut steep-sided valleys in the upland empty directly to the Muskingum River. Downstream from Stockport, the valley widens somewhat. Elsewhere in the basin, the drainage systems assume more of a spread-out, dendritic pattern, and the topography is less rugged.

The nearest glacial limit is about 10 mi to the west in Perry County (fig. 1, basin O). Thus, the Middle Muskingum River basin is in the Unglaciated Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province.

Geologic Setting

The stratigraphic succession, in ascending order and generally from northwest to southeast across the study basin, consists primarily of the Pottsville, Allegheny, Conemaugh, and Monongahela Formations of Pennsylvanian age. Upper units of the Pottsville are present in valley bottoms near Zanesville. In the same area are scattered remnants of coal-bearing units of the Allegheny Formation. The coal-bearing units are more extensive east of the Muskingum River and along Brush Creek to the south. The Middle Kittanning and Upper Freeport coals (No. 6 and No. 7) have been mined in the area. Allegheny outcrops are present along the Muskingum Valley below Philo for several miles.

The Conemaugh Formation, with its thick assemblage of shales, shaly sandstones, thin limestones, and minor coals, is present in much of the northern part of the basin and along the western part of the basin south of Zanesville. Farther east and south, the Conemaugh Formation is more extensive. Toward the southeastern corner of Muskingum County and into Morgan County, it is overlain by widespread remnants of the Monongahela Formation.

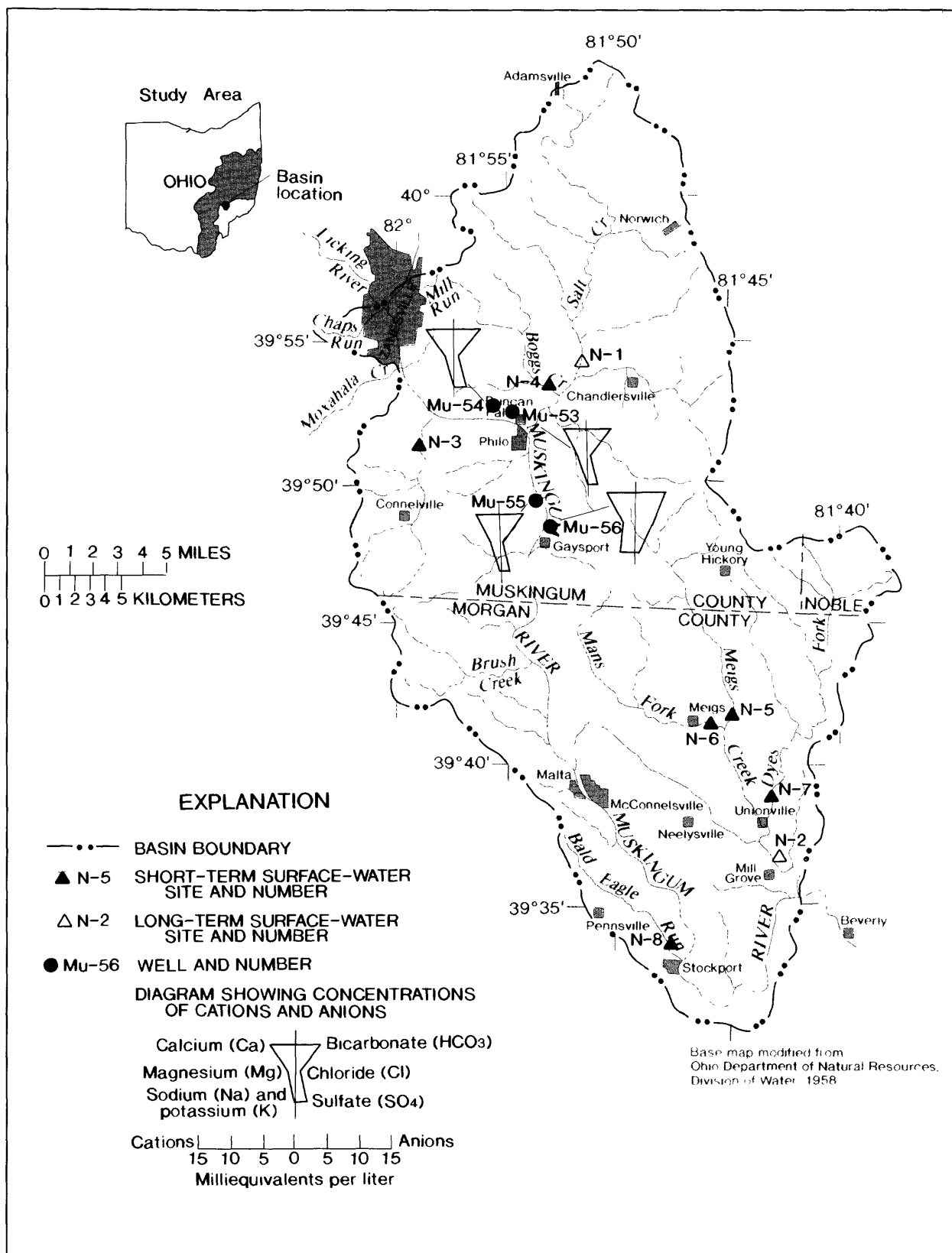


Figure 22.—Middle Muskingum River basin (N), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.

In the basin, the Pittsburgh (No. 8) coal and the Meigs Creek (No. 9) coal in the Monongahela Formation have been heavily surface mined. In places, the Uniontown (No. 10) and the Waynesburg (No. 11) coals are present.

The rock succession exhibits the usual regional east-to-southeast dip toward the Appalachian Basin. A minor structural flexure, the Parkersburg-Lorain Syncline, causes the rocks to increase in dip along the eastern side of the basin such that basal remnants of Permian rocks cap the Pennsylvanian strata. The Parkersburg-Lorain Syncline is a minor structure that is roughly parallel to the Cambridge Arch (Lamborn, 1951), discussed earlier for the Lower Wills Creek basin.

Most of the more valuable coal (No. 9) has been mined out. Recent data (Lopez, 1991) indicate no production for Morgan County in 1989 and minor production of Middle Kittanning (No. 6) coal near Zanesville in Muskingum County. Along the Muskingum-Noble county line, production of the Meigs Creek (No. 9) coal amounted to about 8 percent of Ohio's production for 1989.

Alluvial fill in places along the Muskingum River yields 100 to 500 gal/min and is an important source of supply for several communities in the valley. In contrast, the rock succession for most of the basin yields poorly to wells. Yields of 5 to 25 gal/min are possible in a small area around and south of Zanesville (Walker, 1962a). In this area, Pottsville and lower Allegheny strata are present and are recharged by precipitation and streams. The same units traced southeastward are deeply buried under younger strata, and brines are likely to be present about 250 ft below valley floors (Stout and others, 1943). Water-resource maps from ODNR (Walker, 1962a, 1984) indicate that bedrock yields of 3 to 5 gal/min or less can be expected for most of the basin.

Water Quality

Two long-term streamwater sites and six short-term sites scattered throughout the basin were used to assess streamwater quality in the Middle Muskingum River basin. Four ground-water sites in Muskingum County, between Duncan and Gaysport, were chosen to assess ground-water quality.

The streamwater sites listed in the table on the next page were sampled in late October 1991. Their locations are shown in figure 22.

The Muskingum River carries runoff for a large part of eastern Ohio, and data-collection sites on the main stream would not be representative of sampling and conditions in the basin. Therefore, long-term surface-water sites were established on the two largest tributary systems, Salt Creek and Meigs Creek (fig. 22, N-1 and N-2, respectively). Sites N-4 and N-7 drain areas of active mining, sites N-3, N-5, and N-6 drain areas of former mining, and site N-8 drains an area of little or no mining. Analyses of water samples show that water quality at site N-3 was different from that of the seven other streamwater sites in the basin (table 21). Unlike the other sites, where pH values were in the alkaline range, water at site N-3 was strongly acidic and had high concentrations of aluminum, iron, and manganese. At site N-3, concentrations of total recoverable and dissolved iron, aluminum, and manganese were at the top of the range of values (table 22). Site N-3 and three other sites (N-2, N-5, and N-6) had sulfate concentrations greater than 1,000 mg/L. The medians for specific conductance and dissolved-sulfate concentrations were higher than the medians for most of the other basins examined in this report. Except for site N-3, all of the values of constituents at surface-water sites were within the interquartile ranges and whiskers of box plots in figure 23; for site N-3, detached values are shown for pH and concentrations of aluminum, iron, and manganese.

Streamwater

Map index number	Site type	Site name	Drainage area (square miles)
N-1	Long term	Salt C nr Chandlersville	75.7
N-2	do.	Meigs C nr Beverly	136.0
N-3	Short term	Brush C nr Philo	20.5
N-4	do.	Boggs C nr Duncan Falls	17.9
N-5	do.	Meigs C nr Meigs	35.7
N-6	do.	Mans F nr Meigs	28.0
N-7	do.	Dyes F nr Unionville	87.9
N-8	do.	Bald Eagle Rn nr Stockport	9.5

Water-quality analyses show the effects of acid mine drainage on surface-water quality in the basin. The effects are especially pronounced at site N-3 where acidic conditions exist. This site is in the western part of the basin (fig. 22), where mining of Allegheny coals took place in the past. Water samples from sites N-2, N-5, and N-6 have high concentrations of sulfate and high specific conductances; both characteristics are commonly associated with acid mine drainage. The pH at these three sites remains high, however, probably because of the presence of carbonate strata and better reclamation practices in the eastern part of the basin. Dissolved-sulfate concentrations and specific conductance were lower at site N-8, which is not in a mined area.

Ground Water

The valley fill along the Muskingum River, especially just downstream from Zanesville, meets the criteria for a shallow, productive aquifer. Acid mine drainage from old or abandoned mines may affect ground-water quality in this area. In addition, the Muskingum River a few miles upstream receives the degraded water of Moxahala Creek. In the same area, the communities of Duncan Falls and Philo

draw upon the alluvial source for supply. Therefore, four ground-water sampling sites were chosen in the area (fig. 22).

Water-quality data for four ground-water sites sampled in September 1991 are shown in table 23. Values for pH ranged from 7.1 to 7.5, and concentrations of constituents associated with acid mine drainage were not high.

Stiff diagrams (fig. 22) of wells Mu-53, Mu-54, and Mu-55 are indicative of calcium bicarbonate-type waters. In waters from well Mu-56, calcium was the dominant cation; however, both bicarbonate and chloride were major anions. The ground waters sampled in the basin are classified as hard to very hard (table 23). In well Mu-53, alkalinity was considerably lower than hardness; therefore, non-carbonate hardness was substantial in this well. The only OEPA water-quality standard that was not met in any of the wells sampled for the constituents analyzed was the standard for dissolved manganese (50 µg/L) in water from well Mu-56.

Table 21. Water-quality data for streamwater sites in Middle Muskingum River basin, October 1991

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
October 24, 1991	0.21	03149500 N-1 SALT C NR CHANDLERSVILLE OH (LAT 39 54 31N LONG 081 51 38W) 750	7.6	14.0	--	139	170
October 23, 1991	2.2	03150250 N-2 MEIGS C NR BEVERLY OH (LAT 39 36 00N LONG 081 42 42W) 2,100	7.7	13.5	--	167	1,100
October 31, 1991	.53	395120081590700 N-3 BRUSH C NR PHILO OH (LAT 39 51 20N LONG 081 59 07W) 2,100	3.5	15.0	112	--	1,100
October 31, 1991	.19	395329081530800 N-4 BOGGS C NR DUNCAN FALLS OH (LAT 39 53 29N LONG 081 53 08W) 610	8.1	15.0	--	237	90
October 31, 1991	.01	394130081450700 N-5 MEIGS C NR MEIGS OH (LAT 39 41 30N LONG 081 45 07W) 1,200	8.1	15.0	--	247	470
October 31, 1991	.01	394117081452200 N-6 MANS F NR MEIGS OH (LAT 39 41 17N LONG 081 45 22W) 2,100	7.8	14.5	--	198	1,300
October 31, 1991	1.3	393754081431700 N-7 DYES F NR UNIONVILLE OH (LAT 39 37 54N LONG 081 43 17W) 2,400	8.1	14.5	--	202	1,200
October 31, 1991	.06	393333081475400 N-8 BALD EAGLE RN NR STOCKPORT OH (LAT 39 33 33N LONG 081 47 54W) 690	8.0	13.5	--	292	47

DATE	Aluminum, total recoverable (µg/L as Al)	Aluminum, dissolved (µg/L as Al)	Iron, total recoverable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manganese, total recoverable (µg/L as Mn)	Manganese, dissolved (µg/L as Mn)
October 24, 1991	420	03149500 N-1 SALT C NR CHANDLERSVILLE OH (LAT 39 54 31N LONG 081 51 38W) <10	1,500	100	970	950
October 23, 1991	70	03150250 N-2 MEIGS C NR BEVERLY OH (LAT 39 36 00N LONG 081 42 42W) <10	130	<10	100	100
October 31, 1991	12,000	395120081590700 N-3 BRUSH C NR PHILO OH (LAT 39 51 20N LONG 081 59 07W) 12,000	1,900	1,900	8,700	9,300
October 31, 1991	30	395329081530800 N-4 BOGGS C NR DUNCAN FALLS OH (LAT 39 53 29N LONG 081 53 08W) 30	60	<10	40	50
October 31, 1991	370	394130081450700 N-5 MEIGS C NR MEIGS OH (LAT 39 41 30N LONG 081 45 07W) <10	450	<10	<160	100
October 31, 1991	190	394117081452200 N-6 MANS F NR MEIGS OH (LAT 39 41 17N LONG 081 45 22W) <10	370	<10	330	280
October 31, 1991	60	393754081431700 N-7 DYES F NR UNIONVILLE OH (LAT 39 37 54N LONG 081 43 17W) <10	120	10	60	50
October 31, 1991	100	393333081475400 N-8 BALD EAGLE RN NR STOCKPORT OH (LAT 39 33 33N LONG 081 47 54W) <10	170	<10	40	20

Table 22. Ranges and medians for selected water-quality characteristics for streamwater sites in Middle Muskingum River basin, October 1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 22)	Median
Specific conductance, in µS/cm-----	610 to 2,400	N-4; N-7	1,650
pH-----	3.5 to 8.1	N-3; N-4, 5, 7	7.9
Alkalinity, in mg/L as CaCO ₃ -----	139 to 292	N-2; N-8	^b 202
Sulfate, dissolved, in mg/L as SO ₄ ----	47 to 1,300	N-8; N-6	785
Aluminum, total, in µg/L as Al-----	30 to 12,000	N-4; N-3	155
Aluminum, dissolved, in µg/L as Al-----	<10 to 12,000	Several sites; N-3	<10
Iron, total, in µg/L as Fe-----	60 to 1,900	N-4; N-3	270
Iron, dissolved, in µg/L as Fe-----	<10 to 1,900	Several sites; N-3	<10
Manganese, total, in µg/L as Mn-----	40 to 8,700	N-4, 8; N-3	135
Manganese, dissolved, in µg/L as Mn-----	20 to 9,300	N-8; N-3	100

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity could be measured (see "Methods," p. 14).

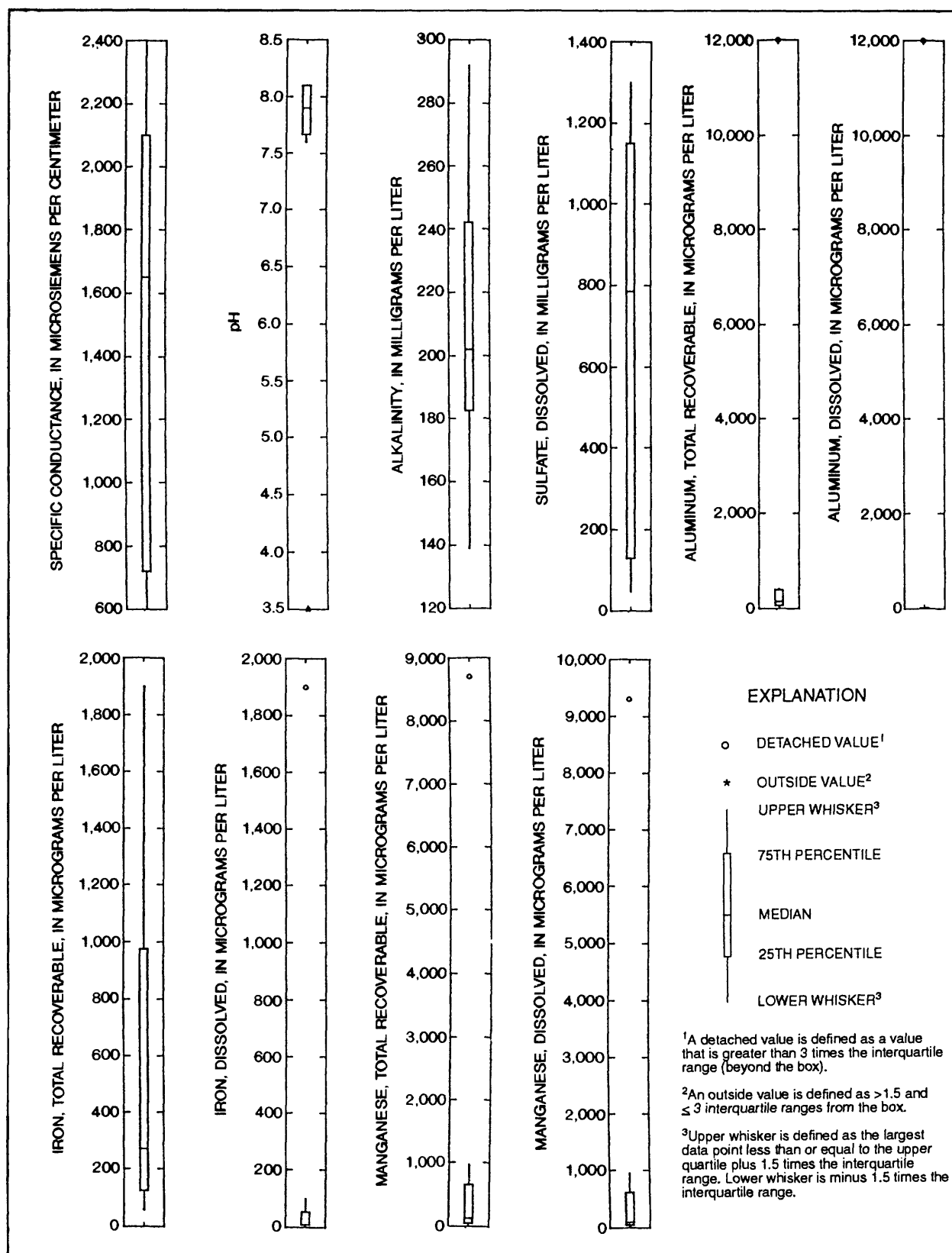


Figure 23.—Ranges, percentiles, and medians of constituents at eight streamwater sites in the Middle Muskingum River basin.

Table 23. Water-quality data for ground-water sites in Middle Muskingum River basin, September 1991

[°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Specific conductance (µS/cm)	pH (standard units)	Temper- ature, water (°C)	Oxygen, dissolved (mg/L)	Hard- ness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Mag- nesium, dissolved (mg/L as Mg)	Sodi- um, dissolved (mg/L as Na)	Potas- sium, dissolved (mg/L as K)	Alka- linity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Sept. 24, 1991	830	7.3	15.5	3.1	410	120	26	19	1.6	250	66
Sept. 24, 1991	480	7.2	12.5	0.9	230	69	14	10	1.4	188	48
Sept. 24, 1991	430	7.5	16.0	3.2	220	63	15	5.9	.70	182	43
Sept. 24, 1991	740	7.1	15.0	3.4	280	82	18	42	3.1	220	43
395225081551600 MU-53 AT DUNCAN FALLS OH (LAT 39 52 25N LONG 081 55 16W)											
395226081555600 MU-54 NR DUNCAN FALLS OH (LAT 39 52 26N LONG 081 55 56W)											
394909081532800 MU-55 NR GAYSPORT OH (LAT 39 49 09N LONG 081 53 28W)											
394825081533000 MU-56 AT GAYSPORT OH (LAT 39 48 25N LONG 081 53 30W)											

Date	Chloride, dissolved (mg/L as Cl)	Silica, dissolved (mg/L as SiO ₂)	Solids, residue at 180 deg. C, dissolved (mg/L)	Solids, sum of constit- uents, dissolved (mg/L)	Alumi- num, total, recov- erable (µg/L as Al)	Iron, total, recov- erable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manga- nese, total, recov- erable (µg/L as Mn)	Manga- nese, dissolved (µg/L as Mn)	Carbon, organic, dissolved (mg/L as C)
Sept. 24, 1991	55	13	484	451	<10	910	<3	10	3	0.6
Sept. 24, 1991	15	12	273	282	20	50	<3	<10	<1	.6
Sept. 24, 1991	4.8	10	245	252	10	110	24	10	10	.6
Sept. 24, 1991	87	9.4	418	416	<10	90	<3	70	60	.8
395225081551600 MU-53 AT DUNCAN FALLS OH (LAT 39 52 25N LONG 081 55 16W)										
395226081555600 MU-54 NR DUNCAN FALLS OH (LAT 39 52 26N LONG 081 55 56W)										
394909081532800 MU-55 NR GAYSPORT OH (LAT 39 49 09N LONG 081 53 28W)										
394825081533000 MU-56 AT GAYSPORT OH (LAT 39 48 25N LONG 081 53 30W)										

Middle Hocking River Basin

The Middle Hocking River basin drains an area of 484 mi² (fig. 1, basin Q). It is the same area referred to as "Hocking River Basin (middle portion)" in the ODNR Water Inventory Plan series (Walker, 1958a). It includes approximately the southern quarter of Perry County, the eastern third of Hocking County, the northwestern half of Athens County, and a small part of Morgan County (fig. 24). The upper end of the basin begins just west of Logan, where drainage from the Upper Hocking River basin (fig. 1, basin P) enters the Middle Hocking River basin. Elsewhere, all drainage to the Hocking River is from within the basin. The Hocking River, as the principal stream, traverses the basin from northwest to southeast. The downstream limit of the basin extends a few miles east of Athens to, but does not include, the tributary basin of Strouds Run.

The Middle Hocking River basin is entirely within the Unglaciaded Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province. The northern border of the basin is only a few miles from the southern limit of Illinoian glaciation, and is further from the last major glacial advance (Wisconsinan). The basin, however, was affected by glaciation to the extent that outwash deposits from the Illinoian and Wisconsinan stages have been mapped in parts of Athens County (Sturgeon and Associates, 1958).

The Middle Hocking River basin is an upland area through which the Hocking River has cut a meandering 34-mile-long path. The valley floor is at least 1/4 mi wide and is several times as wide in some places. The flat to rolling terrain in the main valley extends for several miles upstream in some of the largest tributary valleys in the basin. Local relief, although gentle in the valley bottoms, can be as great as 350 ft between the upland surface and valley floors.

Geologic Setting

The stratigraphic succession begins at the northwestern end of the basin with sandstones and shales of the Cuyahoga and Logan Formations of Mississippian age. The succession continues to the southeast with sandstones, shales, and coal beds of Pennsylvanian age. Strata of the Pottsville Formation are exposed in Perry County and Hocking County and along the Hocking River valley in northwestern Athens County. Outcrops of coal-bearing strata of the Allegheny Formation are extensive in Perry and Hocking Counties. In Athens County, the Allegheny is exposed in stream valleys along the western tier of townships. Just east of Athens and generally along a north-south line, exposed remnants of basal Monongahela Formation are present.

As of 1989, coal was produced at several surface-mining sites in the Middle Hocking basin (Lopez, 1991). The output, which amounted to about 1.25 percent of the State's total, was derived largely from the Middle Kittanning (No. 6) and Lower Freeport (No. 6A) coals, but some was also from the Lower Kittanning (No. 5) and Upper Freeport (No. 7) coals. Much of the basin was formerly mined. Locations of areas covered by mining permits, which are presumably active, are scattered around the northwestern half of the basin. One permit was in effect (as of 1991) for mining the Pittsburgh (No. 8) coal in the Monongahela Formation in what is otherwise an area barren of coal south of Athens.

Bedrock sources of ground water with possible yields to wells of 5 to 25 gal/min had been reported for the northwestern half of the basin (Walker, 1958a). The accumulation of additional well data served to revise this expectation downward to less than 2 gal/min for Athens County (Schmidt, 1985). A recent evaluation of adjacent parts of Hocking and Perry Counties is currently (1993) in preparation at ODNR. According to Norris and Mayer (1982), the Black Hand Sandstone and

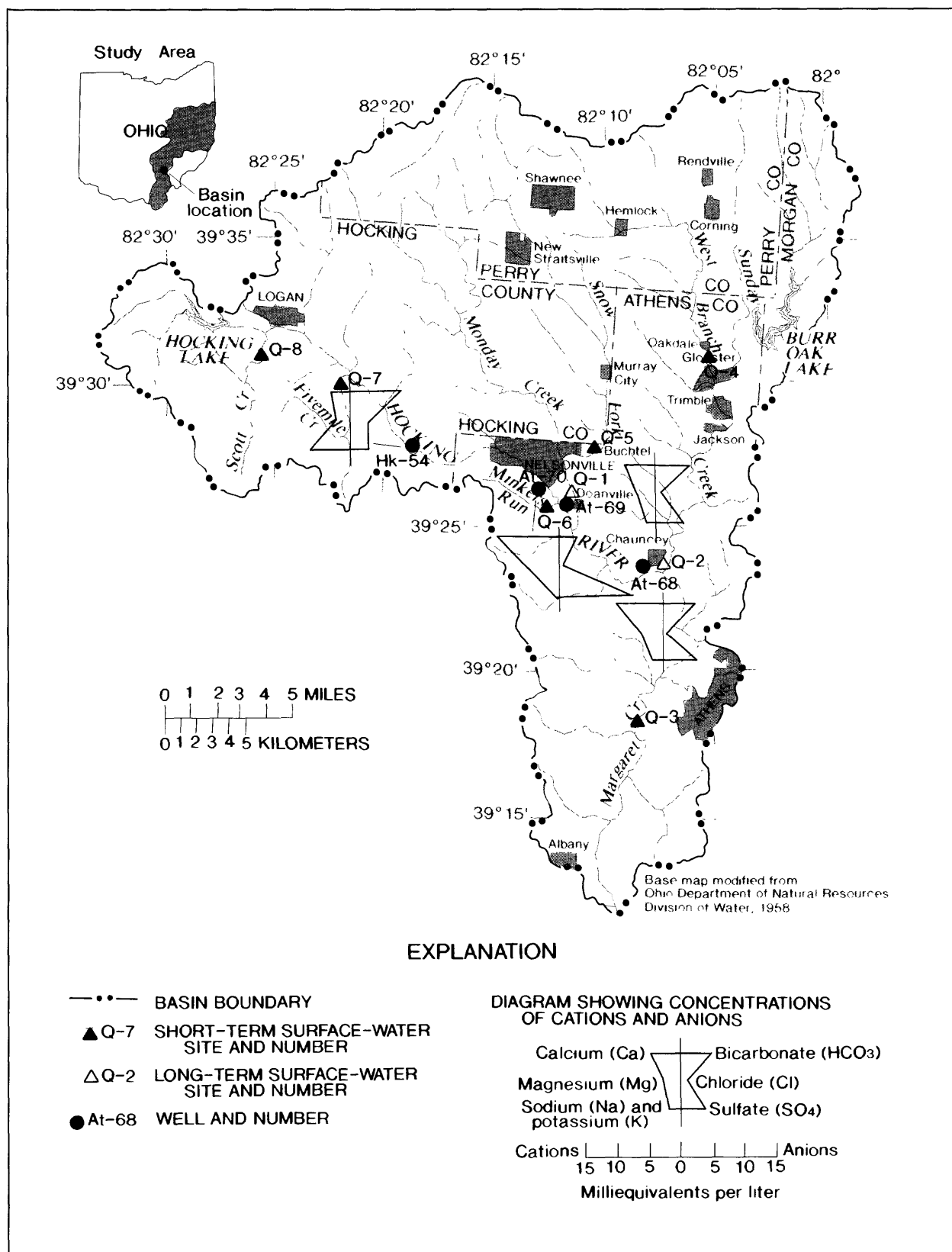


Figure 24.—Middle Hocking River basin (Q), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.

other sandy members of the Mississippian rocks section west of Athens County at the upper end of the basin can be used as a supply of ground water.

In contrast, the unconsolidated sediments forming the valley fill along the Hocking River are ample sources of supply for industrial and municipal purposes. Unconsolidated deposits are present along parts of the tributary systems, but, for the most part, yields to wells are substantially less than those along the Hocking River.

Water Quality

Two long-term sites and six short-term sites were chosen to assess streamwater quality in the Middle Hocking River basin. Four ground-water sites selected for sampling were in the Hocking River valley near Logan and Chauncey.

The streamwater sites listed in the table on the next page were sampled in early November 1991. Their locations are shown in figure 24.

Two long-term streamwater sites were used to assess water quality in two important tributary systems that drain coal-mining areas. These were on Monday Creek at Doanville (fig. 24, site Q-1) and on Sunday Creek at Chauncey (fig. 24, site Q-2).

The results of synoptic sampling of the streams show different degrees of water-quality degradation. Water from sites Q-1, Q-2, Q-5, and Q-6 had low pH and high acidity (table 24), indicative of water affected by acid mine drainage. Degradation of water quality at these sites was further shown by

high concentrations of dissolved sulfate and total recoverable and dissolved aluminum, iron, and manganese. In contrast, at site Q-4, pH was slightly acidic, alkalinity was low, and sulfate concentration was high (440 mg/L). Values of pH at sites Q-3, Q-7, and Q-8 ranged from 7.0 to 7.4. Concentrations of total recoverable or dissolved manganese ranged from 3,400 to 4,700 $\mu\text{g/L}$ at sites Q-3 and Q-8; concentrations of total recoverable iron also were high. At site Q-7, concentrations of constituents associated with acid mine drainage were not high.

The medians for specific conductance and dissolved and total recoverable aluminum, iron, and manganese were high in waters of the Middle Hocking River basin (table 25). The median pH, 5.0, is indicative of the four sites with acidic waters. Box plots show the wide range of values; for example, outside values are shown for aluminum concentration, and detached values are shown for iron concentration (fig. 25). Because there were fewer than five sites for which alkalinity or acidity could be determined, box plots for these characteristics were not constructed.

In general, results of the synoptic sampling of the streams were consistent, considering location of each site with respect to areas of mining. Sites Q-1, Q-2, Q-5, and Q-6 are all downstream from areas of past and current mining of Allegheny coals. Site Q-4 is downstream from coal-bearing areas, and water-quality analyses indicate some effect from acid mine drainage. The three sites where pH is neutral, Q-3, Q-7, and Q-8, are located where little coal has been produced.

Streamwater

Map index number	Site type	Site name	Drainage area (square miles)
Q-1	Long term	Monday C at Doanville	114
Q-2	do.	Sunday C at Chauncey	139
Q-3	Short term	Margaret C nr Athens	44.2
Q-4	do.	WB Sunday C nr Oakdale	34.3
Q-5	do.	Snow F Monday C at Buchtel	24.4
Q-6	do.	Minkers Run nr Nelsonville	5.2
Q-7	do.	Fivemile C nr Logan	12.5
Q-8	do.	Scott C nr Logan	21.5

Ground Water

The unconsolidated deposits along the Hocking River constitute a shallow, productive aquifer. Parts of the Hocking River valley aquifer system are joined by tributary streams with similar unconsolidated deposits that are used for domestic and public supply. Therefore, four wells were sampled in this area.

Water-quality data for four ground-water sites sampled in September 1991 are shown in table 26. Values for pH ranged from 7.3 to 7.5. Concentrations of total recoverable and dissolved iron were greater than 1,000 µg/L at sites At-68, At-70, and Hk-54. The highest dissolved-sulfate concentration was found at site At-70 (470 mg/L).

Stiff diagrams show the heterogeneous ionic character of water from the four wells sampled in the basin (fig. 24). Calcium and magnesium were major cations in all four wells; sodium was dominant in well Hk-54 and minor in well At-70. As for anions, bicarbonate and sulfate were dominant, although, in well Hk-54, chloride was also important. The waters from the four wells are classified as hard to very hard.

The OEPA water-quality standards for public supplies for dissolved solids (500 mg/L), iron (300 µg/L), and manganese (50 µg/L) were not met in samples from wells Hk-54, At-68, and At-70. In addition, the standard for dissolved sulfate was not met in samples from well At-70. Water from well At-69 met all of the OEPA standards among the constituents analyzed for.

Leading Creek Basin

The Leading Creek basin, with a drainage area of 151 mi², includes the western third of Meigs County and small parts of Athens and Gallia Counties (fig. 1, basin QQ). The Leading Creek basin, as defined in this report, is described in the Ohio Water Plan Inventory series (Walker, 1958b) as including part of the Shade River and Leading Creek basins. Leading Creek enters the Ohio River near Middleport (fig. 26).

Table 24. Water-quality data for streamwater sites in Middle Hocking River basin, October-November 1991

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
October 23, 1991	4.9	03158200 Q-1 MONDAY C AT DOANVILLE OH (LAT 39 26 07N LONG 082 11 30W) 1,350	3.3	9.5	143	--	560
October 23, 1991	7.4	392342082072000 Q-2 SUNDAY C AT CHAUNCEY OH (LAT 39 23 42N LONG 082 07 20W) 2,030	2.9	10.5	183	--	770
November 1, 1991	.31	391809082083700 Q-3 MARGARET C NR ATHENS OH (LAT 39 18 09N LONG 082 08 37W) 665	7.2	12.5	--	181	65
November 1, 1991	2.8	393107082051000 Q-4 W B SUNDAY C NR OAKDALE OH (LAT 39 31 07N LONG 082 05 10W) 930	6.7	12.5	--	36	440
November 1, 1991	2.4	03158195 Q-5 SNOW F MONDAY C AT BUCHTEL OH (LAT 39 27 51N LONG 082 10 16W) 1,750	2.7	12.5	332	--	800
November 1, 1991	.52	392540082124800 Q-6 MINKERS RUN NR NELSONVILLE OH (LAT 39 25 40N LONG 082 12 48W) 1,450	3.2	12.5	307	--	780
November 1, 1991	.08	392928082214200 Q-7 FIVE MILE C NR LOGAN OH (LAT 39 29 28N LONG 082 21 42W) 685	7.4	13.0	--	160	140
November 1, 1991	.08	393052082252800 Q-8 SCOTT C NR LOGAN OH (LAT 39 30 52N LONG 082 25 28W) 425	7.0	12.5	--	168	8.5

DATE	Aluminum, total recoverable (µg/L as Al)	Aluminum, dissolved (µg/L as Al)	Iron, total recoverable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manganese, total recoverable (µg/L as Mn)	Manganese, dissolved (µg/L as Mn)
October 23, 1991	16,000	03158200 Q-1 MONDAY C AT DOANVILLE OH (LAT 39 26 07N LONG 082 11 30W) 17,000	4,700	5,000	4,300	4,200
October 23, 991	1,800	392342082072000 Q-2 SUNDAY C AT CHAUNCEY OH (LAT 39 23 42N LONG 082 07 20W) 1,800	48,000	53,000	3,500	3,700
November 1, 1991	100	391809082083700 Q-3 MARGARET C NR ATHENS OH (LAT 39 18 09N LONG 082 08 37W) <10	2,500	950	4,500	4,700
November 1, 1991	30	393107082051000 Q-4 W B SUNDAY C NR OAKDALE OH (LAT 39 31 07N LONG 082 05 10W) <10	120	20	90	100
November 1, 1991	32,000	03158195 Q-5 SNOW F MONDAY C AT BUCHTEL OH (LAT 39 27 51N LONG 082 10 16W) 35,000	15,000	16,000	5,700	6,300
November 1, 1991	5,100	392540082124800 Q-6 MINKERS RUN NR NELSONVILLE OH (LAT 39 25 40N LONG 082 12 48W) 5,100	4,100	3,300	5,300	5,600
November 1, 1991	80	392928082214200 Q-7 FIVE MILE C NR LOGAN OH (LAT 39 29 28N LONG 082 21 42W) <10	720	60	550	490
November 1, 1991	20	393052082252800 Q-8 SCOTT C NR LOGAN OH (LAT 39 30 52N LONG 082 25 28W) <10	1,200	140	3,700	3,400

Table 25. Ranges and medians for selected water-quality characteristics for streamwater sites in Middle Hocking River basin, October-November 1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 24)	Median
Specific conductance, in µS/cm-----	425 to 2,030	Q-8; Q-2	1,140
pH-----	2.7 to 7.4	Q-5; Q-7	5.0
Acidity, in mg/L as CaCO ₃ -----	143 to 332	Q-1; Q-5	^b 245
Alkalinity, in mg/L as CaCO ₃ -----	36 to 181	Q-4; Q-3	^b 164
Sulfate, dissolved, in mg/L as SO ₄ -----	8.5 to 800	Q-8; Q-5	500
Aluminum, total, in µg/L as Al-----	20 to 32,000	Q-8; Q-5	950
Aluminum, dissolved, in µg/L as Al-----	<10 to 35,000	Several sites; Q-5	900
Iron, total, in µg/L as Fe-----	120 to 48,000	Q-4; Q-2	3,300
Iron, dissolved, in µg/L as Fe-----	20 to 53,000	Q-4; Q-2	2,125
Manganese, total, in µg/L as Mn-----	90 to 5,700	Q-4; Q-5	4,000
Manganese, dissolved, in µg/L as Mn-----	100 to 6,300	Q-4; Q-5	3,950

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity and acidity could be measured (see "Methods," p. 14).

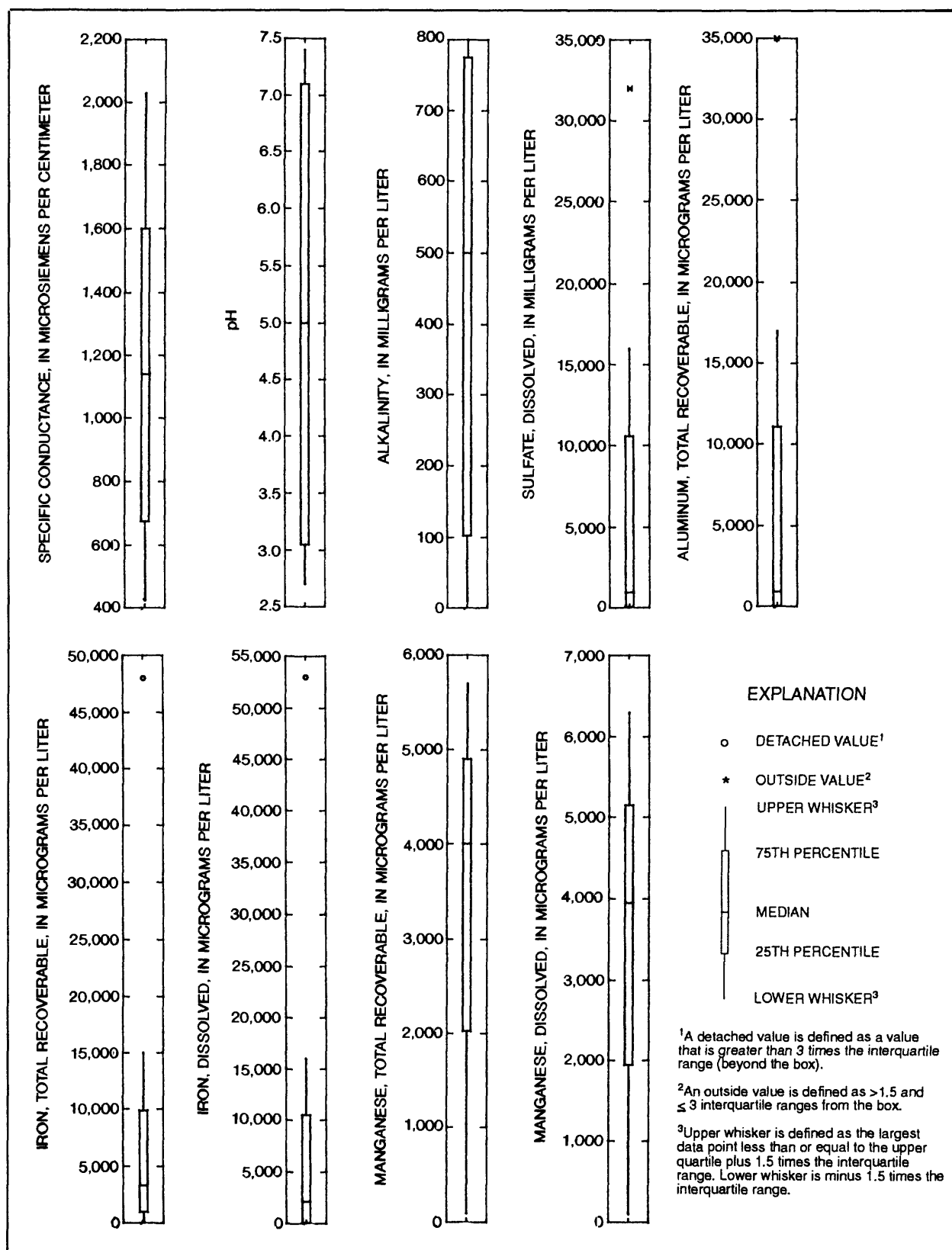


Figure 25.—Ranges, percentiles, and medians of constituents at eight streamwater sites in the Middle Hocking River basin.

Table 26. Water-quality data for ground-water sites in Middle Hocking River basin, September 1991

[°C, degrees Celsius; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available]

Date	Specific conductance (µS/cm)	pH (standard units)	Temper- ature, water (°C)	Oxygen, dissolved (mg/L)	Hard- ness, total (mg/L as CaCO ₃)	Calcium, dissolved (mg/L as Ca)	Mag- nesium, dissolved (mg/L as Mg)	Sodi- um, dissolved (mg/L as Na)	Potas- sium, dissolved (mg/L as K)	Alka- linity, total, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
Sept. 25, 1991	875	7.3	13.0	3.4	440	130	29	23	2.5	244	250
Sept. 25, 1991	760	7.3	17.0	6.8	350	89	30	29	2.4	218	160
Sept. 25, 1991	1,230	7.3	14.5	3.3	730	190	61	12	2.6	115	470
Sept. 25, 1991	970	7.5	14.0	.6	230	60	20	120	3.8	352	130

Date	Chloride, dissolved (mg/L as Cl)	Silica, dissolved (mg/L as SiO ₂)	Solids, residue at 180 deg. C, dissolved (mg/L)	Solids, sum of consti- tuents, dissolved (mg/L)	Alumi- num, total, recov- erable (µg/L as Al)	Iron, total, recov- erable (µg/L as Fe)	Iron, dissolved (µg/L as Fe)	Manga- nese, total, recov- erable (µg/L as Mn)	Manga- nese, dissolved (µg/L as Mn)	Carbon, organic, dissolved (mg/L as C)
Sept. 25, 1991	15	13	615	612	40	2,500	2,500	370	390	0.8
Sept. 25, 1991	24	12	487	477	<10	840	34	50	42	.6
Sept. 25, 1991	8.3	6.7	986	822	<10	2,000	1,600	680	730	.8
Sept. 25, 1991	83	10	562	639	60	1,100	1,200	130	130	.8

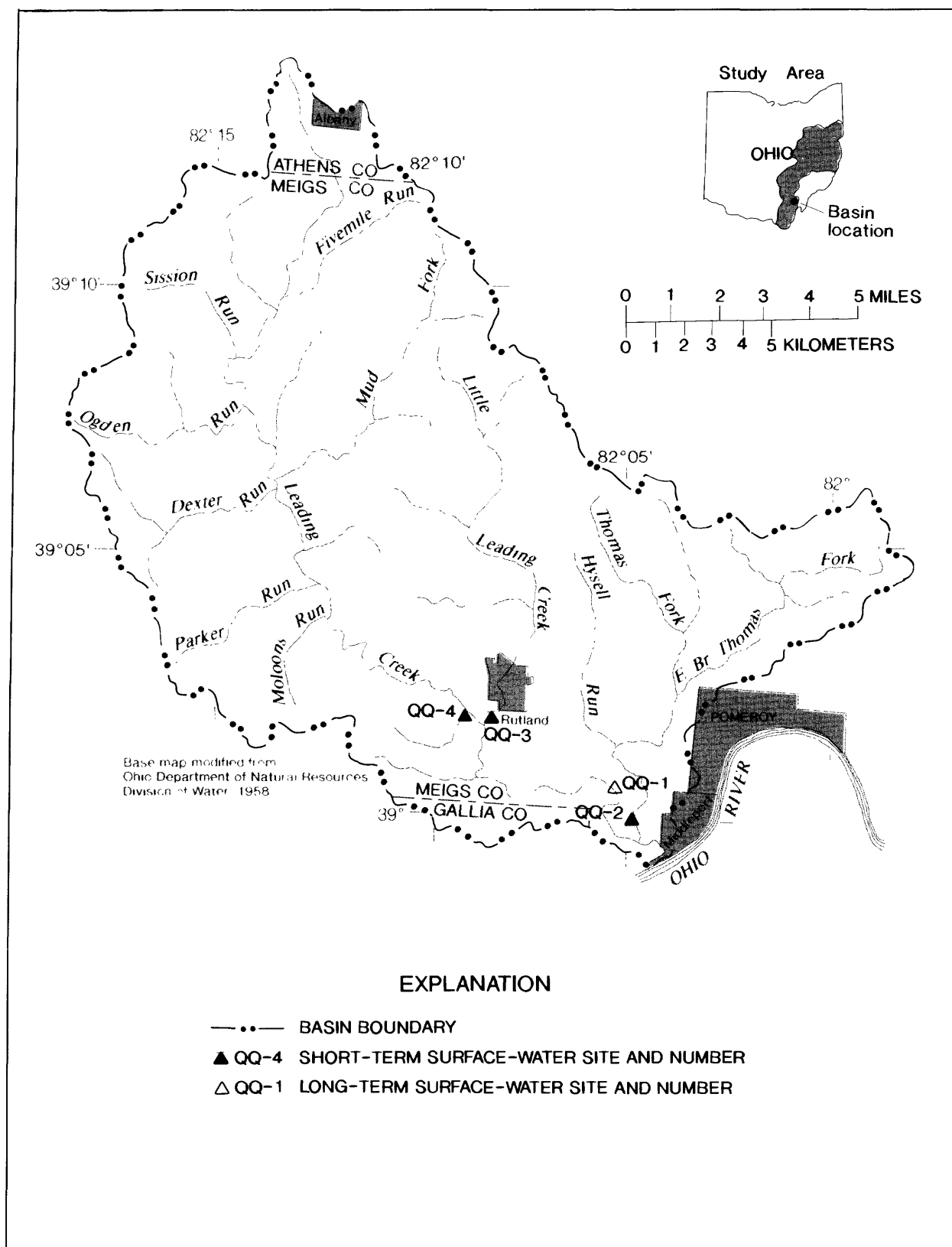


Figure 26.—Leading Creek basin (QQ) and streamwater sites.

Leading Creek basin is beyond the glaciated part of Ohio and is thus entirely within the Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province. The upland is thoroughly dissected, and Leading Creek, the principal stream, has formed a meandering path across the basin. In places, the valley bottoms are a 1/4 mi or more wide. The principal stream has not carved a distinctive flood plain that stands in sharp contrast with upland areas as in, for example, the Middle Muskingum River basin. Many of the tributaries to Leading Creek are in relatively broad-bottomed valleys. Local relief is rarely greater than 300 ft.

Geologic Setting

The stratigraphic succession at the western edge of the basin begins with the Conemaugh Formation, and, toward the middle, is overlain by small remnants of the Monongahela Formation. At the base is the Pittsburgh (No. 8) coal, and the Pomeroy (No. 8A) coal is not far above. Although permits have been issued for mining of the Nos. 8 and 8A in the basin, no production was reported for 1989 (Lopez, 1991). Instead, output from longwall mining of the Clarion (No. 4A) coal underlying the western part of the basin, which amounted to about 2 percent of total output for the State, was the only coal production reported for Meigs County.

Ground-water-resource maps issued by ODNr (Walker, 1958b; Schmidt, 1985) show that, except near the Ohio River, yields to wells from bedrock are very low. Most representative wells yield less than 3 gal/min. The downstream end of Leading Creek meanders through the Ohio River flood plain, which is underlain by sands and gravels that yield as much as 1,000 gal/min; however, this is a very small part of the basin.

Water Quality

Because of the small size of the basin, only one long-term streamwater site and three short-term synoptic sites were used for data collection. The only shallow, productive aquifers in the basin are Ohio Valley aquifers. Because Ohio Valley aquifers are not within the scope of this study, no ground-water-quality data were collected in the Leading Creek basin.

The streamwater sites listed in the table on the next page were sampled in early November 1991. Their locations are shown in figure 26.

At sites QQ-1 and QQ-4, specific conductances were greater than 5,000 $\mu\text{S}/\text{cm}$, concentrations of dissolved sulfate were greater than 2,000 mg/L, and concentrations of total recoverable and dissolved manganese were greater than 2,000 $\mu\text{g}/\text{L}$ (table 27); however, pH values were in the alkaline range. In contrast, the pH at site QQ-2 was 3.5, and dissolved-sulfate concentration was 850 mg/L. Concentrations of total recoverable and dissolved aluminum, manganese, and iron were very high at site QQ-2. Among all the basins in the synoptic sample collections, medians for specific conductance, dissolved sulfate, and total recoverable and dissolved manganese were highest in samples from the Leading Creek basin (table 28). Because only four sites were sampled in the Leading Creek basin, box plots were not constructed.

Concentrations of constituents associated with acid mine drainage were high in waters from sites QQ-1, QQ-2, and QQ-4. Alkaline waters at sites QQ-1 and QQ-4 indicate some buffering of streamwaters. At site QQ-3, none of the concentrations of constituents associated with acid mine drainage were high, and pH was alkaline.

Map index number	Site type	Site name	Drainage area (square miles)
QQ-1	Long term	Leading C nr Middleport	117
QQ-2	Short term	Thomas F nr Middleport	29.2
QQ-3	do.	L Leading C nr Rutland	25.0
QQ-4	do.	Leading C nr Rutland	89.3

Lower Raccoon Creek Basin

The Lower Raccoon Creek Basin, with a drainage area of 434 mi², includes the eastern third of Jackson County, half of Gallia County, and small adjacent parts of Meigs and Vinton Counties (fig. 1, basin S). The Lower Raccoon Creek basin, which corresponds to the same area described by Pree (1962b), consists of two principal subbasin drainage areas (fig. 27). Little Raccoon Creek enters Raccoon Creek just south of the town of Vinton in the north-western part of Gallia County. Upstream from this junction, Raccoon Creek is part of the Upper Raccoon Creek Basin (fig. 1, basin R) described in a previous report (Sedam, 1991). The eastern subbasin includes a 139-mi² area drained by a number of smaller drainage systems and short streams that discharge into the Ohio River. The longest of these streams is Campaign Creek.

All of the Lower Raccoon Creek basin lies in the Unglaciaded Allegheny Plateau Section of the Appalachian Plateaus Physiographic Province. Because of differences in rock types (such as those between massive sandstone and clays, shales, or friable sandstones), several erosional cycles have produced a land surface that ranges from a terrain of long ridges and narrow valleys in the eastern part of the basin to a less rugged terrain toward the center of the basin (Stout, 1927). Local relief is about 300 ft in the hilliest parts of the area.

Geologic Setting

The stratigraphic units that are present in the area include, from west to east in ascending order, (1) basal Pennsylvanian units of the Pottsville and Allegheny Formations, (2) the Conemaugh Formation, which is present in much of the central and eastern part of the basin, and (3) scattered remnants of the Monongahela Formation along the eastern side of the basin. These units dip to the east and south, averaging about 33 ft/mi to the south-east, according to Stout (1927).

The principal coal-bearing units are the Lower Mercer (No. 3) coal in the Pottsville Formation and the Brookville (No. 4), Clarion (No. 4A), Lower Kittanning (No. 5), Middle Kittanning (No. 6), Lower Freeport (No. 6A), and Upper Freeport (No. 7) coals in the Allegheny Formation (fig. 4). Small units of the Pittsburgh (No. 8) and Pomeroy (No. 8A) coals of the Monongahela Formation are found north and south of Gallipolis. About 3 percent of the State's coal production in 1989 was in the Lower Raccoon Creek basin (Lopez, 1991). Most of this was from the Allegheny Formation in the northwestern part of the basin near the drainage divide between the Upper Raccoon Creek basin and the Lower Raccoon Creek basin.

Table 27. Water-quality data for streamwater sites in Leading Creek basin, October-November 1991

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius; --, data not available][

DATE	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)	
October 24, 1991	03160050 QQ-1 LEADING C NR MIDDLEPORT OH (LAT 39 00 31N LONG 082 05 07W)	2.0	5,170	7.4	16.5	--	105	2,100
November 1, 1991	390017082042700 QQ-2 THOMAS F NR MIDDLEPORT OH (LAT 39 00 17N LONG 082 04 27W)	1.2	1,750	3.5	9.5	166	--	850
November 1, 1991	390134082081300 QQ-3 L LEADING C NR RUTLAND OH (LAT 39 01 34N LONG 082 08 13W)	.15	765	7.4	10.0	--	104	210
November 1, 1991	390222082093300 084 QQ-4 LEADING C NR RUTLAND OH (LAT 39 02 22N LONG 082 09 33W)	1.1	6,250	7.2	11.0	--	137	2,700

DATE	Aluminum, total recoverable (μg/L as Al)	Aluminum, dissolved (μg/L as Al)	Iron, total recoverable (μg/L as Fe)	Iron, dissolved (μg/L as Fe)	Manganese, total recoverable (μg/L as Mn)	Manganese, dissolved (μg/L as Mn)	
October 24, 1991	03160050 QQ-1 LEADING C NR MIDDLEPORT OH (LAT 39 00 31N LONG 082 05 07W)	230	20	230	140	2,400	2,500
November 1, 1991	390017082042700 QQ-2 THOMAS F NR MIDDLEPORT OH (LAT 39 00 17N LONG 082 04 27W)	18,000	20,000	2,100	2,100	8,100	8,100
November 1, 1991	390134082081300 QQ-3 L LEADING C NR RUTLAND OH (LAT 39 01 34N LONG 082 08 13W)	30	30	80	20	180	140
November 1, 1991	390222082093300 084 QQ-4 LEADING C NR RUTLAND OH (LAT 39 02 22N LONG 082 09 33W)	50	10	610	350	6,400	6,700

Table 28. Ranges and medians for selected water-quality characteristics for streamwater sites in Leading Creek basin, October-November 1991

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 26)	Median
Specific conductance, in µS/cm-----	765 to 6,250	QQ-3; QQ-4	3,460
pH-----	35 to 7.4	QQ-2; QQ-1,3	7.3
Alkalinity, in mg/L as CaCO ₃ -----	104 to 137	QQ-3; QQ-4	^b 105
Sulfate, dissolved, in mg/L as SO ₄ ----	210 to 2,700	QQ-3; QQ-4	1,475
Aluminum, total, in µg/L as Al-----	30 to 18,000	QQ-3; QQ-4	140
Aluminum, dissolved, in µg/L as Al-----	10 to 20,000	QQ-4; QQ-2	25
Iron, total, in µg/L as Fe-----	80 to 2,100	QQ-3; QQ-2	420
Iron, dissolved, in µg/L as Fe-----	20 to 2,100	QQ-3; QQ-2	245
Manganese, total, in µg/L as Mn-----	180 to 8,100	QQ-3; QQ-2	4,400
Manganese, dissolved, in µg/L as Mn-----	140 to 8,100	QQ-3; QQ-2	4,600

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity could be measured (see "Methods," p. 14).

Principal aquifers are in the surficial bedrock. Of regional importance is the Black Hand Sandstone Member of the Cuyahoga Formation, which yields to many wells in Vinton County and a small part of Jackson County. The southeastern limit of potable water in the Black Hand Sandstone Member is generally along a line about 3 mi east of McArthur that extends through Wellston (fig. 27), according to Norris and Mayer (1982). Locally, productive sources of water are available in sandstone units of the Pottsville and Allegheny Formations.

Differences between modern and preglacial drainage systems have been explained in detail by Stout (1927). Alluvial deposits are present along various streams, but these are mostly of fine-grained sand or clay. Shallow ground-water sources in these deposits are productive only for small domestic supplies. In general, most shallow wells in the area are completed in bedrock.

Water Quality

The Raccoon Creek basin (fig. 1, basins R and S) was studied previously by the USGS (Wilson, 1985, 1988), in cooperation with the ODNR, Division of Reclamation. In the 1985 study, data for use in individual subbasin reclamation projects were collected in synoptic sampling of 41 streamwater sites during base flow in November 1983. Fifteen of those sites, including long-term site S-1 used in the current study, were in the Lower Raccoon Creek basin (fig. 1, basin S). The 1988 study includes chemical and biological data collected from July 1984 through September 1986 at some of the 15 sites in support of reclamation projects in selected subbasins. Because these studies provided considerable streamwater data for the Lower Raccoon Creek basin, no short-term

streamwater sites were selected in this study. Two long-term streamwater sites were sampled six times during 1989-91. The data are included in table 31.

Except for two areas along the Ohio River north of Gallipolis, there are no shallow, productive aquifers in the Lower Raccoon Creek basin. Therefore, no ground-water sampling was done in the basin.

SUMMARY OF WATER QUALITY FOR 1985-91

A principal objective of the investigation was to provide baseline data on the water quality of streams draining Ohio's coal-mining region from 1985-91. The study area included actively mined, unmined, and formerly mined lands. In addition, baseline data were compiled on the quality of ground water in productive, shallow aquifers in the study area.

Streamwater

Streamwater measurements included periodic sampling of 41 long-term sites selected to provide a general assessment of streamwater quality throughout the 21-basin study area. Water-quality data for 246 samples collected during 1989-91 at these sites are summarized in table 2 and figure 7. Water-quality data for 508 samples collected at the same sites during 1985-91 are summarized in table 29 and figure 28. The data for the 1987-88 study period were presented by Sedam (1991, p. 16), and data for 1985-86 were presented by Jones (1988, p. 12-13).

For the long-term site network, the medians for the 1989-91 period were very similar to the medians for the two earlier study periods (1985-86 and 1987-88). Therefore, medians did not change substantially over the 7-year study period.

Table 29. Ranges and medians for selected water-quality characteristics for long-term streamwater sites in active coal-mining areas of Ohio, 1985-91

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius]

Property or constituent	Range	Locations ^a (fig. 5)	Median
Specific conductance, in µS/cm-----	270 to 5,170	T-1; QQ-1	770
pH-----	2.7 to 9.1	O-2, Q-2; A-1	7.8
Alkalinity, in mg/L as CaCO ₃ -----	1 to 391	Q-1; K-1	^b 118
Sulfate, dissolved, in mg/L as SO ₄ -----	13 to 2,100	J-1; QQ-1	200
Aluminum, total, in µg/L as Al-----	<10 to 18,000	P-2, Q-2; Q-1	220
Aluminum, dissolved, in µg/L as Al-----	<10 to 18,000	Several sites; Q-1	30
Iron, total, in µg/L as Fe-----	20 to 62,000	D-3; Q-2	630
Iron, dissolved, in µg/L as Fe-----	<10 to 54,000	Several sites; Q-2	50
Manganese, total, in µg/L as Mn-----	<10 to 50,000	D-3; O-2	350
Manganese, dissolved, in µg/L as Mn-----	<10 to 47,000	Several sites; O-2	290

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

^b Median was calculated on the basis of only those samples for which alkalinity could be measured (see "Methods," p. 14).

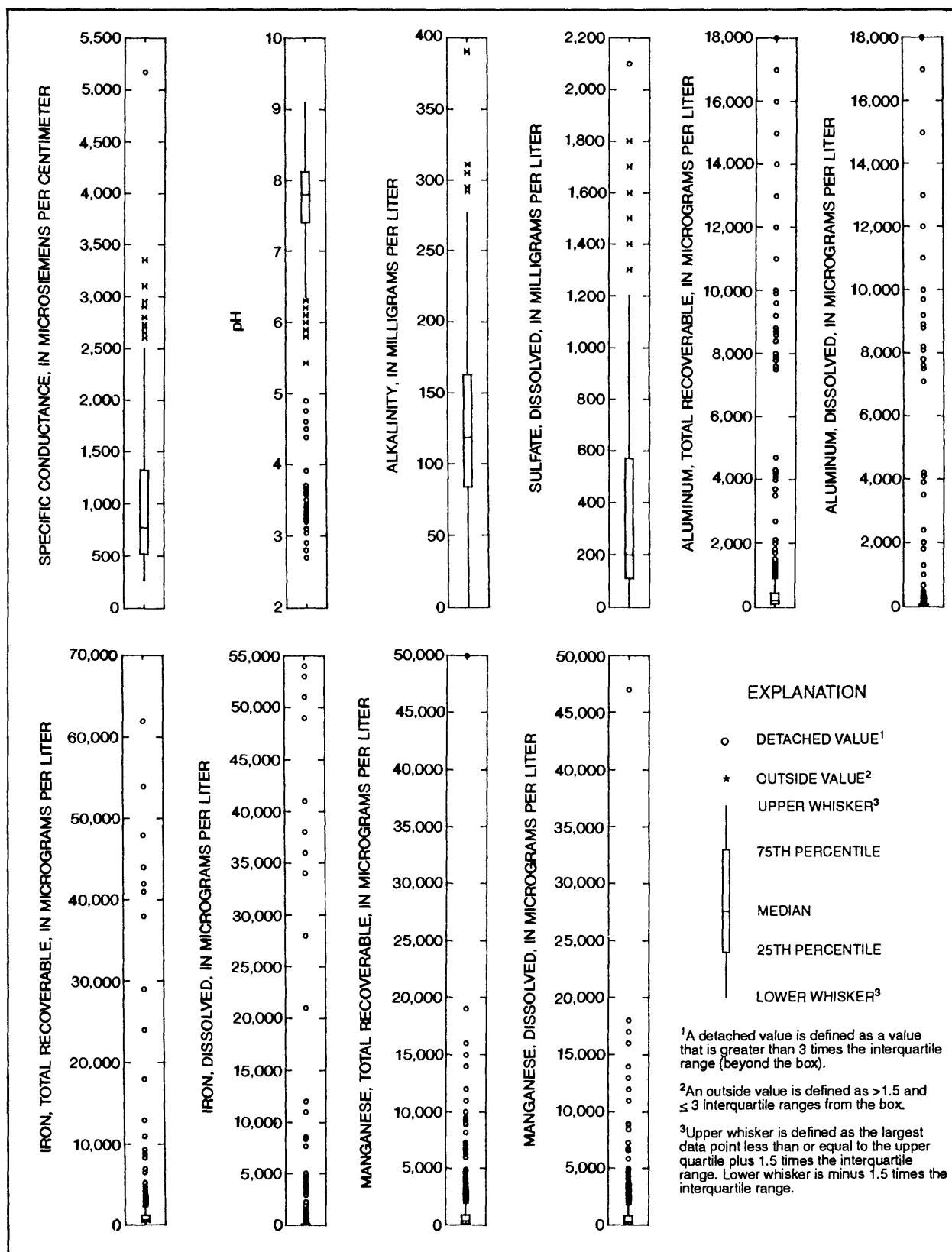


Figure 28.--Ranges, percentiles, and medians of constituents in 508 samples collected at 41 long-term streamwater sites in active coal mining areas of Ohio, 1985-91.

Throughout the entire study period (1985-91), samples from streamwater at sites in the Moxahala Creek and Middle Hocking River basins (basins O and Q, respectively) had high concentrations of constituents and low pH values commonly associated with acid mine drainage. For all three study periods, in streamwater samples from basins O and Q, concentrations of total recoverable and dissolved aluminum, iron, and manganese were at the top of the range of values, and pH was at the bottom of the range of values. Before the 1989-91 study, the highest specific conductances and dissolved-sulfate concentrations were found in the Short and Wheeling Creeks basin (basin C). In 1991, however, the highest specific conductances and dissolved-sulfate concentrations for the long-term site network (1985-91) were found in the Leading Creek basin (basin QQ).

Concentrations of total recoverable and dissolved aluminum, iron, and manganese for a number of sites were near or below minimum detection levels at several sites (table 29). At these same sites, specific conductance and dissolved-sulfate concentration were low, and pH and alkalinity were high; such a combination of characteristics indicates that these waters were probably not affected by mining activities.

Buffering of streamwaters was indicated by samples in which pH was in the neutral to alkaline range but in which concentrations of one or more of the constituents of interest were high. The basins in which samples in this category were collected include the Yellow and Cross Creeks basin (basin B) and Wheeling Creeks basin (basin C). In these areas, there is an abundance of carbonate rock to serve as a buffering agent.

Water from degraded streams is diluted as the streams discharge into larger streams. Two acidic streams, Monday Creek (site Q-1) and Sunday Creek (site Q-2), discharge into the Hocking River. The pH of samples collected at the USGS streamflow-gaging station on the

Hocking River at Athens several miles below the confluence of Monday and Sunday Creeks remains high (pH > 8.0), despite drainage from two acidic streams.

Sites along the same stream can differ considerably in water quality. This is shown by the difference in water quality between two sites, P-6 and P-1, along Rush Creek in the Upper Hocking River basin (basin P). The pH is in the alkaline range for water from site P-1; upstream at site P-6, the pH is in the acidic range. At site P-1, however, moderately high specific conductances and concentrations of dissolved sulfate and manganese indicate a possible source of degradation upstream.

The variability in water quality of different streams in the same drainage basin demonstrates the importance of proper site selection for identifying problem areas in the coal-mining regions. Within individual basins, however, extreme concentrations of certain constituents can be used to identify sources of degradation that might be masked by buffering or downstream dilution.

Ground water

During the 1985-91 data-collection period, 16 of the 21 study basins were sampled for ground water. Of the 69 wells sampled, 54 yielded from unconsolidated aquifers.

For 20 wells, the ground waters were of the calcium bicarbonate type. Except for a few samples with high concentrations of sodium, calcium was the dominant cation, and magnesium was commonly an important secondary cation. Bicarbonate was the principal anion in many samples, although sulfate was often a major anion and was the dominant anion in waters from several wells. A graphical distribution of the principal cations and anions is shown in figure 29.

Analytical results of 69 ground-water samples collected during 1985-91 are summarized in table 30. Similar data collected during 1989-91 only are summarized in table 3. The data for the 1987-88 study period were summarized by Sedam (1991), and 1985-86 ground-water-quality data were summarized by Jones (1988, p. 16-19). Generally, the wells sampled contained hard to very hard waters.

A comparison of water-quality data throughout the 1985-91 study period shows that the medians for most constituents were fairly consistent, considering the geographical diversity of the study area. Concentrations of total recoverable and dissolved iron were exceptions. According to Hem (1989, p. 83), however, the concentration of iron in ground water is determined by pH and dissolved oxygen in the system, which, in turn, is affected by ground-water circulation and mixing. Considering the range of dissolved-oxygen concentration that was observed in the ground-water samples, most of which were derived from unconsolidated sediments, the variations in iron concentration were not unusual.

In an earlier study (Razem and Sedam, 1985), 100 ground-water samples were collected in the same area as for the current study.

The earlier study investigated the ground-water quality of wells drilled into or through a coal bed. Although objectives of the two studies were different, the medians for specific conductance, pH, hardness, calcium, and sulfate were similar in both studies. Slightly different medians were found between the two studies for alkalinity, iron, and manganese.

In general, median specific conductances for ground-water samples were lower than medians for streamwater samples (table 29) for the 1985-91 study period. Median concentrations in ground water were also considerably lower than those in streamwater for dissolved sulfate, and were slightly lower for aluminum and manganese. The pH of ground water varied little, unlike the pH of streamwater. Except for a few ground-water sites where concentrations of dissolved sulfate were greater than 250 mg/L and concentrations of total recoverable and dissolved iron and manganese were greater than 1,000 µg/L, the ground-water quality in the study area shows little, if any, effect from mining of coal.

Table 30. Ranges and medians for selected water-quality characteristics for ground-water sites in active coal-mining areas of Ohio, 1985-91

[mg/L, milligrams per liter; µg/L, micrograms per liter; µS/cm, microsiemens per centimeter at 25 degrees Celsius. Some of the values are from Jones (1988) and Sedam (1991)]

Property or constituent	Range	Locations ^a (fig. 8)	Median
Specific conductance, in µS/cm -----	120 to 1,590	V-90; B-30	640
pH -----	5.4 to 8.9	C-11; B-28	7.4
Oxygen, dissolved, in mg/L -----	0.0 to 8.1	Cs-149; Cs-152	1.6
Hardness, in mg/L as CaCO ₃ -----	13 to 840	Hk-53; Hr-23	288
Noncarbonate hardness, in mg/L as CaCO ₃ -----	0 to 610	Several sites; At-70	65
Calcium, dissolved, in mg/L as Ca -----	3.4 to 250	Hk-53; Hr-23	74
Magnesium, dissolved, in mg/L as Mg -----	1.1 to 64	Hk-53; Co-1	21
Sodium, dissolved, in mg/L as Na -----	3.5 to 280	Cs-154; B-28	19
Potassium, dissolved, in mg/L as K -----	0.6 to 6.8	Mu-50; Tu-50	1.8
Alkalinity, in mg/L as CaCO ₃ -----	8 to 461	V-90; B-28	230
Sulfate, dissolved, in mg/L as SO ₄ -----	1.0 to 630	Cs-149, Tu-55; B-30	59
Chloride, dissolved, in mg/L as Cl -----	2.1 to 200	Cs-150; Tu-53	18
Silica, dissolved, in mg/L as SiO ₂ -----	6.6 to 38	Tu-55; V-90	12
Solids, dissolved, sum of constituents, in mg/L -----	100 to 1,140	V-90; B-30	374
Aluminum, total, in µg/L as Al -----	10 to 600	Several sites; Hr-30	20
Aluminum, dissolved in µg/L as Al -----	<10 to 100	Many sites; Hr-25, 32	<10
Iron, total, in µg/L as Fe -----	10 to 9,800	Cs-154; Pe-58	890
Iron, dissolved, in µg/L as Fe -----	3 to 8,700	Several sites; Tu-57	580
Manganese, total, in µg/L as Mn -----	10 to 3,200	Several sites; Co-1	90
Manganese, dissolved, in µg/L as Mn -----	1 to 3,200	Cs-154, 155; Co-1	85
Carbon, organic, dissolved, in µg/L as C -----	0.5 to 3.1	Several sites; Mu-50	.8

^a Sites at which the minimum values in the ranges were measured are to the left of the semicolon. Sites at which the maximum values in the ranges were measured are to the right of the semicolon.

SUMMARY AND CONCLUSIONS

Coal is Ohio's most economically important mineral resource. Even with the passage of legislation to control mining activities, many abandoned mines are still in need of restoration. Mining activities could affect water quality in the mining regions of southeastern Ohio and remain a potential threat in unmined areas. Therefore, natural-resource managers in Ohio found a need for baseline hydrologic data to facilitate the processing of coal-mining permits in unmined areas and in active areas of mining.

A 7-year study was done in three phases to collect baseline water-quality data in the coal-mining region of eastern Ohio. A long-term network of streamwater sites was established in 21 drainage basins, and periodic sampling was done during 1985-91 to assess baseline streamwater quality. This sampling was supplemented by synoptic data collections at short-term and long-term streamwater and short-term ground-water sites in three basins each year.

This report presents data from the 1989-91 sampling period and includes a summary for the entire study (1985-91). Two previous reports present data collected during 1985-86 (Jones, 1988) and during 1987-88 (Sedam, 1991). All three reports include discussions of geologic setting and physiography in the study basins.

Several physical properties and chemical constituents were measured to assess the water quality of major streams in the study area. These were specific conductance, pH, alkalinity, and (or) acidity, and concentrations of dissolved sulfate and total recoverable and dissolved iron, manganese, and aluminum. Additional constituents and physical properties were measured to assess the ground-water quality. These included hardness and concentrations of dissolved solids, calcium, sodium, potassium, and chloride.

During 1989-91, 246 samples from the long-term streamwater network were analyzed to assess baseline water quality. Water samples from sites in the Middle Hocking River basin and Moxahala Creek basin had low pH, typical of streams affected by mining activities. These samples also had high concentrations of many constituents commonly associated with acid mine drainage, including dissolved sulfate, and total recoverable and dissolved iron, manganese, and aluminum. Water from some other sites in the long-term site network also had high concentrations of these constituents, but pH was in the neutral to alkaline range, indicating buffering of streamwaters.

During the final phase of the study (1989-91), short-term and long-term streamwater sites in nine basins were selected for synoptic sampling. Low pH and high concentrations of certain constituents associated with acid mine drainage were not found in Lower Wills Creek basin. In the Yellow and Cross Creek, Walhonding River, and Upper Muskingum River basins, some constituent concentrations associated with mining activities at some sites were high; however, pH was in the neutral to alkaline range at all sites. This combination of characteristics indicates buffering of streamwaters by carbonate strata or the effects of proper reclamation practices. The Conotton Creek, Upper Hocking River, and Middle Muskingum River basins each had one site where pH was in the acidic range. At these three sites, concentrations of dissolved sulfate and total recoverable and dissolved aluminum, iron, and manganese all were high. The water at the other sites in the Conotton Creek, Upper Hocking River, and Middle Muskingum River basins had pH in the alkaline range and lower concentrations of constituents associated with mining activities.

During the 1989-91 synoptic sampling, the streamwater quality at sites sampled in the Middle Hocking River basin and Leading Creek basin indicated possible effects from

mining activities. In the Middle Hocking River basin, four out of eight sites sampled had acidic waters with very high concentrations of dissolved sulfate and total recoverable and dissolved aluminum, iron, and manganese. In the Leading Creek basin, water from only one site was acidic, but waters from three out of the four sites sampled had high concentrations of some constituents associated with acid mine drainage.

In each of seven of the nine basins selected for synoptic sampling during 1989-91, four ground-water sites representative of water quality in shallow, productive aquifers were sampled. Most ground waters sampled were calcium bicarbonate-type waters, although some waters were fairly heterogeneous in ionic character. Most waters were classified as hard to very hard. Ohio Environmental Protection Agency water-quality standards for public supplies for dissolved solids (500 mg/L), iron (300 µg/L), manganese (50 µg/L), and sulfate (250 mg/L) were exceeded in waters from some wells. Except for high concentrations of total recoverable and dissolved iron at a few sites, the high concentrations of constituents associated with acid mine drainage and low pH were not found in the ground-water samples collected in the seven basins.

More than 500 samples were collected from the long-term streamwater network during 1985-91 to provide a general assessment of streamwater quality throughout the 21-basin study area. Throughout the 7-year period, the medians of water-quality properties

and constituents did not change substantially. Several sites within the Moxahala Creek and Middle Hocking River basins (basins O and Q, respectively) had acidic waters with high concentrations of aluminum, iron, manganese, and sulfate. Buffering of streamwaters in several basins was indicated by samples with pH values in the neutral to alkaline range and high concentrations of one or more of constituents associated with acid mine drainage.

The variability in water quality of different streams in the same drainage basin indicates the importance of proper site selection for identifying problem areas in the coal-mining region. Within individual basins, however, extreme concentrations of certain constituents can be used to identify sources of degradation that might be masked by buffering or downstream dilution.

Of the 21 study basins, 16 were sampled for ground water during 1985-91. Of the 69 wells sampled, 54 yielded from unconsolidated aquifers. In the ground-water samples analyzed, medians of most constituents were fairly consistent over time, except concentrations of total recoverable and dissolved iron. Most pH measurements were in the neutral to alkaline range, and, at a few sites, concentrations of dissolved sulfate or total recoverable and dissolved iron and manganese were high. In general, however, low pH and high concentrations of constituents associated with mining activities were not found in the ground-water samples collected during 1985-91.

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Table 31. Water-quality data for long-term streamwater sites, 1989-91

[°C, degrees Celsius; ft³/s, cubic feet per second; mg/L, milligrams per liter;
 µg/L, micrograms per liter; µS/cm, microsiemens per centimeter
 at 25 degrees Celsius; --, data not available]

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
03109500 A-1 L BEAVER C NR EAST LIVERPOOL OH (LAT 40 40 33N LONG 080 32 27W)							
AUG 1989							
15...	82	928	8.3	21.5	--	124	270
OCT							
17...	67	900	8.2	16.5	--	155	230
SEPT 1990							
04...	135	740	9.1	23.0	--	146	200
JUNE 1991							
07...	106	720	8.9	23.0	--	133	220
AUG							
27...	40	840	8.3	23.5	--	114	240
OCT							
30...	59	832	8.4	10.0	--	155	170
03109100 A-2 M F L BEAVER C NR ROGERS OH (LAT 40 43 22N LONG 080 38 03W)							
AUG 1989							
15...	29	874	8.2	19.5	--	149	200
OCT							
17...	29	890	8.1	16.0	--	149	180
JULY 1990							
10...	47	630	8.1	23.5	--	109	140
JUNE 1991							
07...	35	735	8.4	18.5	--	150	170
AUG							
27...	14	860	8.3	22.0	--	145	170
OCT							
30...	22	1020	8.4	9.0	--	183	180
03110000 B-1 YELLOW C NR HAMMONDSVILLE OH (LAT 40 32 16N LONG 080 43 31W)							
AUG 1989							
15...	17	640	8.3	23.0	--	88	200
OCT							
17...	18	580	8.1	16.5	--	100	180
JULY 1990							
10...	25	585	8.5	27.0	--	93	170
JUNE 1991							
07...	22	540	8.4	20.5	--	93	180
AUG							
27...	5.5	650	7.9	23.5	--	82	200
OCT							
30...	35	598	8.1	9.0	--	123	190

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/l as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
03109500 A-1 L BEAVER C NR EAST LIVERPOOL OH (LAT 40 40 33N LONG 080 32 27W)						
AUG 1989						
15...	90	40	150	20	50	20
OCT						
17...	30	<10	120	10	40	20
SEPT 1990						
04...	260	10	190	10	20	20
JUNE 1991						
07...	110	20	240	10	50	20
AUG						
27...	270	100	610	130	90	50
OCT						
30...	20	10	60	<10	30	20
03109100 A-2 M F L BEAVER C NR ROGERS OH (LAT 40 43 22N LONG 080 38 03W)						
AUG 1989						
15...	140	40	290	10	90	40
OCT						
17...	80	<10	300	10	90	60
JULY 1990						
10...	1100	20	2700	20	300	70
JUNE 1991						
07...	160	30	430	160	110	70
AUG						
27...	260	30	610	20	110	50
OCT						
30...	90	10	230	20	70	40
03110000 B-1 YELLOW C NR HAMMONDSVILLE OH (LAT 40 32 16N LONG 080 43 31W)						
AUG 1989						
15...	170	80	300	<10	30	20
OCT						
17...	40	20	240	20	30	<10
JULY 1990						
10...	120	60	160	<10	30	20
JUNE 1991						
07...	220	40	660	20	40	10
AUG						
27...	140	20	510	20	60	30
OCT						
30...	50	20	100	<10	60	20

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
401857080391700 B-2 CROSS C NR MINGO JUNCTION OH (LAT 40 18 57N LONG 080 39 17W)							
AUG 1989							
15...	23	1790	8.4	23.0	--	134	970
OCT							
17...	21	1640	8.4	17.0	--	156	880
SEPT 1990							
04...	43	1450	8.8	23.0	--	139	630
JUNE 1991							
17...	35	1550	8.0	26.0	--	117	850
AUG							
27...	11	1600	8.4	24.5	--	122	910
OCT							
29...	12	1880	8.3	12.0	--	152	810
401716080451300 B-3 MCINTYRE C NR SMITHFIELD OH (LAT 40 17 16N LONG 080 45 13W)							
AUG 1989							
16...	8.6	2270	8.1	21.0	--	135	1300
OCT							
18...	7.4	2010	8.2	12.5	--	214	1200
SEPT 1990							
05...	12	2100	8.2	20.0	--	214	1200
JUNE 1991							
17...	7.9	2100	7.9	26.5	--	204	1200
AUG							
28...	2.6	1600	8.2	23.0	--	187	1100
OCT							
29...	6.4	2300	8.2	12.0	--	192	1200
03111548 C-1 WHEELING C BL BLAINE OH (LAT 40 04 01N LONG 080 48 31W)							
AUG 1989							
14...	36	2420	8.1	24.5	--	196	1300
OCT							
16...	27	2350	8.4	18.0	--	215	1300
SEPT 1990							
05...	46	2100	8.2	20.0	--	220	1000
JUNE 1991							
18...	48	2080	7.8	20.5	--	196	1100
AUG							
28...	20	2600	7.8	22.0	--	176	610
OCT							
29...	17	2730	7.9	10.0	--	207	1300

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
401857080391700 B-2 CROSS C NR MINGO JUNCTION OH (LAT 40 18 57N LONG 080 39 17W)						
AUG 1989						
15...	340	140	240	10	140	110
OCT						
17...	150	50	180	<10	100	70
SEPT 1990						
04...	1300	80	500	20	210	20
JUNE 1991						
17...	560	140	450	140	200	180
AUG						
27...	330	80	280	<10	150	140
OCT						
29...	500	<10	750	<10	140	40
401716080451300 B-3 MCINTYRE C NR SMITHFIELD OH (LAT 40 17 16N LONG 080 45 13W)						
AUG 1989						
16...	220	150	220	20	120	80
OCT						
18...	40	50	120	20	70	60
SEPT 1990						
05...	290	90	230	60	110	80
JUNE 1991						
17...	120	60	170	100	40	30
AUG						
28...	50	50	260	20	40	50
OCT						
29...	50	50	60	10	50	50
03111548 C-1 WHEELING C BL BLAINE OH (LAT 40 04 01N LONG 080 48 31W)						
AUG 1989						
14...	880	400	1600	10	100	80
OCT						
16...	500	340	860	20	110	80
SEPT 1990						
05...	610	270	300	230	80	80
JUNE 1991						
18...	2000	140	6900	130	230	160
AUG						
28...	2100	190	9300	<10	220	170
OCT						
29...	80	60	3900	<10	170	130

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conduc- tance (μS/cm)	pH (stand- ard units)	Temper- ature, water (°C)	Acidity (mg/L as CaCO ₃)	Alka- linity, field (mg/L as CaCO ₃)	Sulfate, dis- solved (mg/L as SO ₄)
03111500 C-2 SHORT C NR DILLONVALE OH (LAT 40 11 36N LONG 080 44 04W)							
AUG 1989							
16...	48	2310	8.0	19.5	--	121	1300
OCT							
18...	42	2050	8.1	11.5	--	194	1100
SEPT 1990							
05...	72	2050	8.2	19.5	--	200	920
JUNE 1991							
17...	52	2120	8.0	26.0	--	166	1200
AUG							
28...	18	2100	8.0	22.0	--	179	1200
OCT							
29...	17	2280	8.0	11.0	--	212	1200
03114000 D-1 CAPTINA C AT ARMSTRONGS MILLS OH (LAT 39 54 31N LONG 080 55 27W)							
AUG 1989							
14...	8.6	659	8.2	22.5	--	147	160
OCT							
16...	38	515	8.5	17.0	--	158	100
SEPT 1990							
05...	46	450	8.3	24.0	--	141	63
JUNE 1991							
18...	29	430	8.2	25.0	--	124	73
AUG							
26...	2.2	710	8.5	29.5	--	133	170
OCT							
28...	4.7	1100	8.3	17.5	--	183	240
03113550 D-2 MCMAHON C AT BELLAIRE OH (LAT 40 00 39N LONG 080 45 45W)							
AUG 1989							
14...	17	1020	8.2	25.0	--	144	370
OCT							
16...	26	910	8.4	17.5	--	165	300
SEPT 1990							
05...	20	1020	8.1	21.5	--	168	340
JUNE 1991							
18...	39	795	7.9	23.0	--	126	250
AUG							
26...	4.6	1200	8.1	26.5	--	134	480
OCT							
29...	5.2	1430	8.0	11.5	--	174	570

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
03111500 C-2 SHORT C NR DILLONVALE OH (LAT 40 11 36N LONG 080 44 04W)						
AUG 1989						
16...	530	230	1300	10	80	50
OCT						
18...	330	100	620	20	40	30
SEPT 1990						
05...	410	230	1000	210	50	50
JUNE 1991						
17...	540	230	1400	140	60	50
AUG						
28...	230	170	620	130	40	30
OCT						
29...	180	90	290	<10	30	30
03114000 D-1 CAPTINA C AT ARMSTRONGS MILLS OH (LAT 39 54 31N LONG 080 55 27W)						
AUG 1989						
14...	310	40	430	20	60	30
OCT						
16...	270	<10	250	<10	30	20
SEPT 1990						
05...	470	180	750	260	50	60
JUNE 1991						
18...	700	20	710	100	40	10
AUG						
26...	140	20	230	20	30	20
OCT						
28...	50	<10	90	20	20	10
03113550 D-2 MCMAHON C AT BELLAIRE OH (LAT 40 00 39N LONG 080 45 45W)						
AUG 1989						
14...	350	300	350	<10	30	30
OCT						
16...	1800	230	3100	20	120	40
SEPT 1990						
05...	720	420	1700	800	70	70
JUNE 1991						
18...	1800	100	3100	10	100	20
AUG						
26...	480	120	1000	250	70	50
OCT						
29...	80	20	340	<10	40	30

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
03114250 D-3 SUNFISH C AT CAMERON OH (LAT 39 46 00N LONG 080 56 09W)							
AUG 1989							
14...	3.1	415	8.3	27.0	--	131	42
OCT							
16...	37	350	8.6	16.0	--	165	40
SEPT 1990							
05...	9.4	400	8.5	26.5	--	129	34
JUNE 1991							
18...	6.5	380	8.6	27.0	--	128	44
AUG							
26...	1.1	410	8.2	27.5	--	126	42
OCT							
28...	.81	535	8.1	17.5	--	174	59
03117500 E-1 SANDY C AT WAYNESBURG OH (LAT 40 40 21N LONG 081 15 36W)							
AUG 1989							
16...	50	681	8.0	19.5	--	142	150
OCT							
18...	88	550	7.9	12.5	--	134	130
SEPT 1990							
04...	125	680	8.4	19.0	--	128	130
JUNE 1991							
07...	70	625	8.1	20.5	--	139	160
AUG							
27...	28	725	7.8	21.0	--	163	48
OCT							
29...	23	735	7.6	11.5	--	162	180
403823081213700 E-2 NIMISHILLEN CR AT SANDYVILLE OH (LAT 40 38 23N LONG 081 21 37W)							
AUG 1989							
16...	97	1380	7.8	21.0	--	107	190
OCT							
17...	167	700	7.9	18.0	--	136	110
SEPT 1990							
04...	140	1200	8.3	20.0	--	196	160
JUNE 1991							
07...	99	1360	8.1	17.0	--	292	190
AUG							
27...	53	1550	8.3	21.5	--	222	170
OCT							
29...	62	1480	7.8	13.5	--	274	180

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
03114250 D-3 SUNFISH C AT CAMERON OH (LAT 39 46 00N LONG 080 56 09W)						
AUG 1989						
14...	150	40	170	<10	30	20
OCT						
16...	10	<10	110	10	10	<10
SEPT 1990						
05...	240	30	450	20	20	20
JUNE 1991						
18...	220	30	340	110	30	10
AUG						
26...	300	20	650	100	50	10
OCT						
28...	30	<10	20	<10	<10	20
03117500 E-1 SANDY C AT WAYNESBURG OH (LAT 40 40 21N LONG 081 15 36W)						
AUG 1989						
16...	110	40	290	20	250	240
OCT						
18...	500	20	550	60	340	280
SEPT 1990						
04...	120	30	700	70	270	260
JUNE 1991						
07...	250	30	980	20	490	440
AUG						
27...	80	<10	610	<10	350	290
OCT						
29...	110	20	450	20	360	360
403823081213700 E-2 NIMISHILLEN CR AT SANDYVILLE OH (LAT 40 38 23N LONG 081 21 37W)						
AUG 1989						
16...	140	50	460	40	90	70
OCT						
17...	680	90	1200	290	220	140
SEPT 1990						
04...	170	30	550	60	120	90
JUNE 1991						
07...	230	20	790	40	140	110
AUG						
27...	150	10	450	90	120	80
OCT						
29...	80	10	200	<10	60	60

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
403426081211900 F-1 CONOTTON C NR SOMERDALE OH (LAT 40 34 26N LONG 081 21 19W)							
AUG 1989							
16...	40	510	7.5	20.0	--	81	130
OCT							
17...	36	495	7.6	16.0	--	90	130
JULY 1990							
11...	65	560	7.4	21.0	--	74	160
JUNE 1991							
07...	52	440	7.5	16.5	--	82	140
AUG							
27...	14	540	7.1	21.5	--	95	27
OCT							
29...	16	615	7.1	12.0	--	112	170
03123000 G-1 SUGAR C AB BEACH CITY DAM AT BEACH CITY OH (LAT 40 39 24N LONG 081 34 37W)							
AUG 1989							
16...	20	680	7.9	23.0	--	197	61
OCT							
17...	18	645	7.8	16.5	--	217	71
JULY 1990							
10...	24	700	7.7	22.5	--	193	59
JUNE 1991							
11...	26	660	7.5	21.5	--	195	62
AUG							
27...	5.3	810	9.1	25.0	--	224	61
OCT							
29...	12	885	7.8	14.0	--	268	49
03116950 G-2 NEWMAN C NR MASSILLON OH (LAT 40 49 22N LONG 081 33 06W)							
AUG 1989							
16...	3.8	788	8.0	22.5	--	215	100
OCT							
17...	19	585	7.9	16.0	--	263	94
JULY 1990							
10...	4.2	740	8.2	23.5	--	211	100
JUNE 1991							
11...	5.4	735	7.9	21.5	--	218	100
AUG							
27...	1.2	680	8.7	25.5	--	182	130
OCT							
29...	1.7	845	7.8	13.0	--	311	93

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
403426081211900 F-1 CONOTTON C NR SOMERDALE OH (LAT 40 34 26N LONG 081 21 19W)						
AUG 1989						
16...	290	40	1700	40	1400	1200
OCT						
17...	540	30	2200	250	920	990
JULY 1990						
11...	500	40	2400	60	1500	1500
JUNE 1991						
07...	380	20	1700	50	1200	1100
AUG						
27...	560	30	1700	10	1500	1400
OCT						
29...	330	20	2300	710	3400	3600
03123000 G-1 SUGAR C AB BEACH CITY DAM AT BEACH CITY OH (LAT 40 39 24N LONG 081 34 37W)						
AUG 1989						
16...	1100	30	2400	20	290	230
OCT						
17...	1300	210	2300	520	370	350
JULY 1990						
10...	1300	10	3100	20	420	340
JUNE 1991						
11...	1700	20	2800	80	370	310
AUG						
27...	1100	20	2300	20	350	210
OCT						
29...	490	<10	1500	200	640	640
03116950 G-2 NEWMAN C NR MASSILLON OH (LAT 40 49 22N LONG 081 33 06W)						
AUG 1989						
16...	150	70	360	20	230	200
OCT						
17...	2100	80	3200	120	310	210
JULY 1990						
10...	160	<10	460	20	210	170
JUNE 1991						
11...	1200	20	2100	20	360	300
AUG						
27...	140	40	390	110	240	210
OCT						
29...	150	<10	550	40	270	280

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
03127500 H-1 STILLWATER C AT UHRICHSVILLE OH (LAT 40 23 10N LONG 081 20 50W)							
AUG 1989							
16...	42	1200	8.4	21.5	--	145	520
OCT							
16...	71	930	7.8	15.5	--	113	420
JULY 1990							
09...	93	1150	7.3	22.5	--	131	400
JUNE 1991							
11...	30	1010	7.6	22.0	--	136	450
AUG							
27...	21	860	8.7	25.0	--	104	170
OCT							
28...	19	1150	7.6	14.0	--	277	460
03129100 I-1 WHITE EYES C NR FRESNO OH (LAT 40 18 17N LONG 081 45 01W)							
AUG 1989							
15...	4.1	940	7.4	21.0	--	99	73
OCT							
16...	3.5	510	7.4	15.0	--	159	87
AUG 1990							
17...	8.6	415	7.6	22.5	--	95	69
JUNE 1991							
11...	4.4	420	7.5	23.0	--	119	70
AUG							
26...	.37	380	8.1	28.0	--	57	63
OCT							
28...	1.3	480	7.3	15.0	--	122	80
401624081363400 I-2 BUCKHORN C AT NEWCOMERSTOWN OH (LAT 40 16 24N LONG 081 36 34W)							
AUG 1989							
15...	1.3	535	8.1	19.0	--	89	120
OCT							
16...	1.4	550	7.7	16.0	--	101	120
JULY 1990							
09...	4.7	465	7.5	24.5	--	59	95
JUNE 1991							
11...	1.4	480	7.6	22.0	--	85	110
AUG							
26...	.11	575	8.1	24.5	--	100	150
OCT							
28...	1.1	575	7.4	15.0	--	139	110

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
03127500 H-1 STILLWATER C AT UHRICHSVILLE OH (LAT 40 23 10N LONG 081 20 50W)						
AUG 1989						
16...	250	70	380	10	460	350
OCT						
16...	440	40	670	20	290	220
JULY 1990						
09...	300	10	600	20	380	240
JUNE 1991						
11...	290	30	440	100	610	450
AUG						
27...	90	20	170	130	620	530
OCT						
28...	110	10	320	50	600	620
03129100 I-1 WHITE EYES C NR FRESNO OH (LAT 40 18 17N LONG 081 45 01W)						
AUG 1989						
15...	260	40	2200	70	720	730
OCT						
16...	410	50	2500	450	420	440
AUG 1990						
17...	200	40	1600	40	300	310
JUNE 1991						
11...	230	20	1800	160	500	540
AUG						
26...	80	20	1500	20	460	410
OCT						
28...	100	20	4300	1500	2900	3100
401624081363400 I-2 BUCKHORN C AT NEWCOMERSTOWN OH (LAT 40 16 24N LONG 081 36 34W)						
AUG 1989						
15...	130	50	470	30	370	340
OCT						
16...	280	70	600	360	380	380
JULY 1990						
09...	810	30	2000	80	700	560
JUNE 1991						
11...	160	20	530	200	300	290
AUG						
26...	90	20	410	50	220	150
OCT						
28...	140	10	370	240	2500	2600

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
03140000 J-1 MILL C NR COSHOCTON OH (LAT 40 21 46N LONG 081 51 45W)							
AUG 1989							
15...	2.1	368	7.5	21.5	--	116	50
OCT							
17...	2.2	425	7.5	16.0	--	105	60
AUG 1990							
17...	4.8	370	7.6	19.5	--	100	13
JUNE 1991							
06...	2.5	380	7.8	19.0	--	101	56
AUG							
26...	.19	390	7.8	23.0	--	108	38
OCT							
28...	.62	450	7.4	14.5	--	132	40
401936082001400 J-2 SIMMONS RN NR WARSAW OH (LAT 40 19 36N LONG 082 00 14W)							
AUG 1989							
15...	1.2	1280	7.6	21.5	--	112	630
OCT							
17...	2.2	1180	7.6	15.0	--	102	570
JULY 1990							
09...	2.0	940	7.9	24.5	--	102	400
JUNE 1991							
10...	1.1	1290	7.6	21.5	--	100	700
AUG							
26...	.09	1820	7.4	22.5	--	87	1000
OCT							
28...	.85	1700	7.4	14.0	--	110	1100
395417081323000 K-1 WILLS C AT PLEASANT CITY OH (LAT 39 54 17N LONG 081 32 30W)							
AUG 1989							
15...	12	2200	8.3	21.5	--	236	1300
OCT							
16...	32	1400	8.0	16.0	--	198	600
SEPT 1990							
05...	25	1600	8.0	22.5	--	195	640
JUNE 1991							
11...	9.9	2900	8.2	22.5	--	269	1800
AUG							
26...	7.2	2950	8.2	23.5	--	305	1600
OCT							
23...	5.3	3350	7.9	13.0	--	391	1700

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
03140000 J-1 MILL C NR COSHOCTON OH (LAT 40 21 46N LONG 081 51 45W)						
AUG 1989						
15...	160	40	1400	90	220	200
OCT						
17...	360	<10	1600	290	310	260
AUG 1990						
17...	110	<10	1000	100	180	180
JUNE 1991						
06...	160	10	1300	130	220	220
AUG						
26...	230	10	1800	230	570	520
OCT						
28...	200	<10	1800	140	340	370
401936082001400 J-2 SIMMONS RN NR WARSAW OH (LAT 40 19 36N LONG 082 00 14W)						
AUG 1989						
15...	270	50	590	10	270	200
OCT						
17...	250	20	320	70	260	180
JULY 1990						
09...	130	<10	370	<10	310	10
JUNE 1991						
10...	120	10	420	110	230	190
AUG						
26...	390	20	880	100	810	720
OCT						
28...	180	<10	330	<10	200	190
395417081323000 K-1 WILLS C AT PLEASANT CITY OH (LAT 39 54 17N LONG 081 32 30W)						
AUG 1989						
15...	560	70	680	20	240	190
OCT						
16...	700	30	840	100	270	140
SEPT 1990						
05...	810	30	1400	20	250	130
JUNE 1991						
11...	620	20	830	20	230	130
AUG						
26...	770	20	1200	80	240	100
OCT						
23...	310	10	310	<10	110	60

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (µS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
400117081362600 L-1 CROOKED C NR CAMBRIDGE OH (LAT 40 01 17N LONG 081 36 26W)							
AUG 1989							
15...	1.7	720	8.1	25.0	--	148	180
OCT							
17...	5.1	640	7.5	17.0	--	148	140
AUG 1990							
17...	10	525	7.9	23.5	--	152	93
JUNE 1991							
17...	1.1	690	7.5	23.5	--	151	190
AUG							
26...	.46	710	8.0	22.5	--	116	140
OCT							
23...	.56	1220	7.6	13.5	--	199	350
400920081432900 L-2 WHITE EYES C NR PLAINFIELD OH (LAT 40 09 20N LONG 081 43 29W)							
AUG 1989							
15...	4.2	1000	7.7	24.0	--	99	530
OCT							
16...	5.1	965	7.4	18.0	--	98	390
AUG 1990							
17...	9.0	845	7.7	22.0	--	99	380
JUNE 1991							
17...	4.1	1130	7.5	23.5	--	98	520
AUG							
26...	1.2	1100	8.1	21.5	--	82	600
OCT							
24...	2.1	1350	7.2	12.0	--	89	620
400710082081000 M-1 WAKATOMIKA C NR FRAZEYSBURG OH (LAT 41 06 16N LONG 082 07 55W)							
AUG 1989							
15...	16	377	7.6	22.5	--	94	19
OCT							
17...	18	430	7.6	15.5	--	109	21
JULY 1990							
09...	26	345	7.8	24.5	--	91	20
JUNE 1991							
06...	22	330	8.2	21.5	--	82	17
AUG							
26...	5.8	365	7.7	23.0	--	97	15
OCT							
28...	8.0	415	7.3	14.5	--	122	19

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
400117081362600 L-1 CROOKED C NR CAMBRIDGE OH (LAT 40 01 17N LONG 081 36 26W)						
AUG 1989						
15...	610	20	970	20	270	210
OCT						
17...	590	30	930	90	570	500
AUG 1990						
17...	1400	40	2100	20	300	230
JUNE 1991						
17...	660	<10	1300	260	820	670
AUG						
26...	400	10	870	180	670	580
OCT						
23...	210	<10	640	50	1300	1300
400920081432900 L-2 WHITE EYES C NR PLAINFIELD OH (LAT 40 09 20N LONG 081 43 29W)						
AUG 1989						
15...	250	40	930	20	800	720
OCT						
16...	360	20	1300	190	780	680
AUG 1990						
17...	110	30	560	20	700	780
JUNE 1991						
17...	420	20	1200	30	720	770
AUG						
26...	240	20	940	20	900	700
OCT						
24...	160	<10	920	360	840	890
400710082081000 M-1 WAKATOMIKA C NR FRAZEYSBURG OH (LAT 41 06 16N LONG 082 07 55W)						
AUG 1989						
15...	190	30	660	30	160	140
OCT						
17...	270	10	680	310	170	140
JULY 1990						
09...	130	10	680	120	120	100
JUNE 1991						
06...	170	20	580	130	110	100
AUG						
26...	210	30	990	40	310	230
OCT						
28...	270	40	870	280	270	270

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
400912082014700 M-2 LITTLE WAKATOMIKA C NR TRINWAY OH (LAT 40 09 12N LONG 082 01 47W)							
AUG 1989							
15...	9.2	1250	8.0	22.0	--	79	820
OCT							
17...	9.7	1580	7.5	16.0	--	81	870
JULY 1990							
09...	15	1180	7.8	23.5	--	77	570
JUNE 1991							
06...	12	1550	7.9	18.5	--	78	900
AUG							
26...	2.5	1980	8.0	23.5	--	64	1300
OCT							
28...	5.3	2000	7.2	14.0	--	78	1300
03149500 N-1 SALT C NR CHANDLERSVILLE OH (LAT 39 54 31N LONG 081 51 38W)							
AUG 1989							
14...	4.4	645	7.8	24.0	--	122	120
OCT							
18...	10	585	7.3	11.0	--	138	87
SEPT 1990							
05...	11	480	8.1	24.0	--	152	67
JUNE 1991							
11...	4.7	500	8.0	23.5	--	110	100
AUG							
27...	.62	565	8.0	19.5	--	24	86
OCT							
24...	.21	750	7.6	14.0	--	139	170
03150250 N-2 MEIGS C NR BEVERLY OH (LAT 39 36 00N LONG 081 42 42W)							
AUG 1989							
14...	25	1300	7.8	22.0	--	180	570
OCT							
16...	34	1230	8.1	16.0	--	100	500
SEPT 1990							
05...	--	1100	7.8	22.5	--	175	470
JUNE 1991							
11...	4.3	1600	8.2	21.5	--	166	360
AUG							
26...	3.5	1300	8.3	25.5	--	35	920
OCT							
23...	2.2	2100	7.7	13.5	--	167	1100

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/l as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
400912082014700 M-2 LITTLE WAKATOMIKA C NR TRINWAY OH (LAT 40 09 12N LONG 082 01 47W)						
AUG 1989						
15...	160	50	440	20	600	470
OCT						
17...	40	20	310	100	770	650
JULY 1990						
09...	140	<10	410	20	560	490
JUNE 1991						
06...	140	<10	600	<10	1200	1200
AUG						
26...	60	10	310	100	850	820
OCT						
28...	120	<10	260	70	1200	1300
03149500 N-1 SALT C NR CHANDLERSVILLE OH (LAT 39 54 31N LONG 081 51 38W)						
AUG 1989						
14...	490	30	920	20	360	290
OCT						
18...	390	50	720	140	310	380
SEPT 1990						
05...	150	10	640	60	220	210
JUNE 1991						
11...	160	50	590	100	390	400
AUG						
27...	280	<10	930	110	530	390
OCT						
24...	420	<10	1500	100	970	950
03150250 N-2 MEIGS C NR BEVERLY OH (LAT 39 36 00N LONG 081 42 42W)						
AUG 1989						
14...	360	90	590	120	90	40
OCT						
16...	130	30	270	30	60	30
SEPT 1990						
05...	820	10	1500	20	120	20
JUNE 1991						
11...	320	40	540	110	100	30
AUG						
26...	250	20	220	<10	90	20
OCT						
23...	70	<10	130	<10	100	100

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conduc- tance (μS/cm)	pH (stand- ard units)	Temper- ature, water (°C)	Acidity (mg/L as CaCO ₃)	Alka- linity, field (mg/L as CaCO ₃)	Sulfate, dis- solved (mg/L as SO ₄)
395337082011100 O-1 MOXAHALA C NR DARLINGTON OH (LAT 39 53 37N LONG 082 01 11W)							
AUG 1989							
15...	63	1200	6.7	21.5	--	16	610
OCT							
17...	107	1310	5.8	16.5	21	15	630
SEPT 1990							
05...	39	1320	6.4	24.0	--	12	590
JUNE 1991							
11...	46	1200	6.7	23.0	--	31	630
AUG							
27...	18	1600	5.9	21.0	6.0	4	810
OCT							
24...	12	2000	4.5	14.0	50	--	1100
03148400 O-2 MOXAHALA C AT ROBERTS OH (LAT 39 51 17N LONG 082 03 23W)							
AUG 1989							
15...	27	1600	2.7	19.0	153	--	1000
OCT							
17...	68	1550	3.3	17.0	88	--	820
SEPT 1990							
05...	20	1800	3.2	22.5	137	--	890
JUNE 1991							
12...	37	1700	3.4	20.5	181	--	950
AUG							
27...	11	1900	3.4	21.0	111	--	1100
OCT							
24...	9.7	2180	3.2	15.0	150	--	1400
395214082054700 O-3 JONATHAN C AT WHITE COTTAGE OH (LAT 39 52 14N LONG 082 05 47W)							
AUG 1989							
14...	33	830	8.1	24.0	--	133	240
OCT							
18...	42	900	7.0	13.0	--	111	300
SEPT 1990							
05...	13	975	8.1	24.0	--	114	340
JUNE 1991							
12...	31	1100	8.1	22.5	--	110	360
AUG							
27...	4.7	1800	8.2	25.5	--	72	830
OCT							
24...	4.1	2180	7.7	14.5	--	72	1100

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
395337082011100 O-1 MOXAHALA C NR DARLINGTON OH (LAT 39 53 37N LONG 082 01 11W)						
AUG 1989						
15...	1400	90	1300	40	6800	6800
OCT						
17...	4200	450	2700	260	8200	7300
SEPT 1990						
05...	630	100	280	40	7300	3000
JUNE 1991						
11...	350	50	370	180	6000	5900
AUG						
27...	1800	190	190	90	6300	6700
OCT						
24...	7500	7100	440	270	11,000	11,000
03148400 O-2 MOXAHALA C AT ROBERTS OH (LAT 39 51 17N LONG 082 03 23W)						
AUG 1989						
15...	13,000	13,000	8900	7700	14,000	13,000
OCT						
17...	7800	8100	4200	4500	9300	9000
SEPT 1990						
05...	9600	9700	5200	4200	12,000	13,000
JUNE 1991						
12...	8700	8800	3400	2900	9000	8500
AUG						
27...	10,000	11,000	3600	3700	12,000	14,000
OCT						
24...	11,000	12,000	8300	8400	15,000	17,000
395214082054700 O-3 JONATHAN C AT WHITE COTTAGE OH (LAT 39 52 14N LONG 082 05 47W)						
AUG 1989						
14...	520	140	600	30	2700	2200
OCT						
18...	620	60	230	40	3100	3000
SEPT 1990						
05...	320	120	250	20	3000	3000
JUNE 1991						
12...	550	190	550	250	2500	2400
AUG						
27...	270	200	270	10	2200	2200
OCT						
24...	150	40	70	10	4900	4900

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, instantaneous (ft ³ /s)	Specific conduc- tance (μS/cm)	pH (stand- ard units)	Temper- ature, water (°C)	Acidity (mg/L as CaCO ₃)	Alka- linity, field (mg/L as CaCO ₃)	Sulfate, dis- solved (mg/L as SO ₄)
03156700 P-1 RUSH C NR SUGAR GROVE OH (LAT 39 38 18N LONG 082 30 42W)							
AUG 1989							
14...	58	540	8.0	22.0	--	107	140
OCT							
16...	63	555	7.3	15.5	--	91	150
SEPT 1990							
04...	33	825	7.6	24.0	--	68	350
JUNE 1991							
10...	37	680	7.8	24.0	--	97	180
AUG							
26...	14	599	7.6	24.5	--	92	180
OCT							
25...	11	755	7.2	17.0	--	138	180
03157000 P-2 CLEAR C NR ROCKBRIDGE OH (LAT 39 35 18N LONG 082 34 43W)							
AUG 1989							
14...	37	445	8.6	21.0	--	201	37
OCT							
16...	43	467	7.9	17.5	--	194	42
SEPT 1990							
04...	11	400	8.3	20.5	--	180	26
JUNE 1991							
10...	27	440	8.4	19.5	--	171	33
AUG							
26...	14	332	8.3	23.5	--	153	20
OCT							
25...	18	331	7.8	15.0	--	161	25
03158200 Q-1 MONDAY C AT DOANVILLE OH (LAT 39 26 07N LONG 082 11 30W)							
AUG 1989							
14...	49	960	3.7	19.0	70	--	450
OCT							
16...	41	809	4.6	15.5	36	1	350
SEPT 1990							
05...	14	1300	3.4	20.5	134	--	590
JUNE 1991							
10...	19	1100	3.4	20.5	118	--	620
AUG							
27...	10	1900	3.7	24.0	78	--	360
OCT							
23...	4.9	1350	3.3	9.5	143	--	560

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
03156700 P-1 RUSH C NR SUGAR GROVE OH (LAT 39 38 18N LONG 082 30 42W)						
AUG 1989						
14...	350	40	840	20	880	860
OCT						
16...	420	30	590	10	1300	1300
SEPT 1990						
04...	160	50	470	20	2500	2500
JUNE 1991						
10...	230	50	700	110	410	360
AUG						
26...	260	40	580	120	560	420
OCT						
25...	120	10	320	160	390	360
03157000 P-2 CLEAR C NR ROCKBRIDGE OH (LAT 39 35 18N LONG 082 34 43W)						
AUG 1989						
14...	110	20	310	320	50	60
OCT						
16...	330	20	240	40	40	40
SEPT 1990						
04...	120	30	330	40	30	40
JUNE 1991						
10...	290	30	540	50	60	30
AUG						
26...	240	<10	400	20	50	30
OCT						
25...	130	<10	240	130	90	60
03158200 Q-1 MONDAY C AT DOANVILLE OH (LAT 39 26 07N LONG 082 11 30W)						
AUG 1989						
14...	12,000	10,000	6500	1100	4100	3600
OCT						
16...	4300	2400	1500	420	2900	3100
SEPT 1990						
05...	15,000	15,000	3400	3200	4500	4600
JUNE 1991						
10...	13,000	13,000	4600	3900	3300	3200
AUG						
27...	9200	9200	1800	1800	2600	3100
OCT						
23...	16,000	17,000	4700	5000	4300	4200

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, Instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
392342082072000 Q-2 SUNDAY C AT CHAUNCEY OH (LAT 39 23 42N LONG 082 07 20W)							
AUG 1989							
14...	34	940	6.2	19.0	94	11	450
OCT							
16...	39	821	6.6	15.5	--	27	360
SEPT 1990							
05...	14	1780	3.1	20.5	175	--	780
JUNE 1991							
10...	15	1600	3.4	21.0	148	--	820
AUG							
27...	12	1400	3.3	23.0	122	--	630
OCT							
23...	7.4	2030	2.9	10.5	183	--	770
03160050 QQ-1 LEADING C NR MIDDLEPORT OH (LAT 39 00 31N LONG 082 05 07W)							
AUG 1989							
16...	57	325	7.4	20.5	--	34	91
DEC							
04...	45	1250	7.7	.5	--	75	320
AUG 1990							
31...	7.5	1600	7.7	23.5	--	88	490
JUNE 1991							
13...	2.4	1410	7.7	19.5	--	112	390
AUG							
27...	2.8	3100	7.2	22.5	--	77	1100
OCT							
24...	2.0	5170	7.4	16.5	--	105	2100
385826082201800 R-1 RACCOON C AT VINTON OH (LAT 38 58 26N LONG 082 20 18W)							
AUG 1989							
15...	99	325	7.4	22.5	--	14	120
DEC							
04...	191	326	7.2	1.5	--	18	120
SEPT 1990							
04...	41	430	7.0	23.0	--	25	160
JUNE 1991							
12...	34	431	6.5	21.5	--	68	170
AUG							
27...	10	690	6.8	24.0	--	14	320
OCT							
25...	4.1	498	6.5	12.0	--	30	200

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
392342082072000 Q-2 SUNDAY C AT CHAUNCEY OH (LAT 39 23 42N LONG 082 07 20W)						
AUG 1989						
14...	230	40	38,000	34,000	2200	2000
OCT						
16...	<10	30	24,000	21,000	1500	1300
SEPT 1990						
05...	1800	2000	44,000	41,000	3300	3400
JUNE 1991						
10...	720	670	54,000	49,000	2800	2800
AUG						
27...	660	660	29,000	28,000	2100	2200
OCT						
23...	1800	1800	48,000	53,000	3500	3700
03160050 QQ-1 LEADING C NR MIDDLEPORT OH (LAT 39 00 31N LONG 082 05 07W)						
AUG 1989						
16...	8800	70	18,000	70	600	50
DEC						
04...	360	30	580	40	630	550
AUG 1990						
31...	100	40	350	90	220	210
JUNE 1991						
13...	90	30	150	90	410	420
AUG						
27...	60	20	190	100	950	930
OCT						
24...	230	20	230	140	2400	2500
385826082201800 R-1 RACCOON C AT VINTON OH (LAT 38 58 26N LONG 082 20 18W)						
AUG 1989						
15...	230	10	930	10	1600	1400
DEC						
04...	50	10	590	420	2000	1800
SEPT 1990						
04...	120	10	680	110	950	960
JUNE 1991						
12...	120	10	470	70	2300	2400
AUG						
27...	100	10	360	110	780	760
OCT						
25...	110	20	720	310	630	630

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, Instantaneous (ft ³ /s)	Specific conductance (μS/cm)	pH (standard units)	Temperature, water (°C)	Acidity (mg/L as CaCO ₃)	Alkalinity, field (mg/L as CaCO ₃)	Sulfate, dissolved (mg/L as SO ₄)
390941082212200 R-2 ELK F NR RADCLIFF OH (LAT 39 09 41N LONG 082 21 22W)							
AUG 1989							
14...	4.3	375	7.8	21.0	--	43	120
OCT							
16...	14	377	7.0	16.0	--	26	140
SEPT 1990							
04...	6.7	345	7.1	21.0	--	47	100
JUNE 1991							
12...	2.4	439	7.0	19.5	--	43	140
AUG							
26...	1.0	559	7.1	24.0	--	43	320
OCT							
25...	.64	630	6.6	13.0	--	66	230
03201988 S-1 L RACCOON C NR VINTON OH (LAT 38 57 11N LONG 082 21 56W)							
AUG 1989							
15...	13	545	7.2	22.5	--	36	240
DEC							
04...	76	410	6.9	1.5	--	15	170
SEPT 1990							
04...	28	425	7.0	22.5	--	23	170
JUNE 1991							
12...	13	569	7.0	21.0	--	245	230
AUG							
27...	9.1	620	6.7	23.5	--	10	210
OCT							
25...	8.6	662	6.0	12.5	9.0	13	310
03160105 S-2 CAMPAIGN C NR GALLIPOLIS OH (LAT 38 53 51N LONG 082 11 31W)							
AUG 1989							
16...	.28	905	7.3	23.0	--	65	390
DEC							
05...	11	485	7.6	1.0	--	66	170
AUG 1990							
31...	.62	540	7.7	27.0	--	88	150
JUNE 1991							
12...	.21	648	7.4	26.0	--	85	170
AUG							
27...	.07	720	7.1	29.0	--	46	300
OCT							
24...	.17	830	7.1	19.5	--	57	350

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
390941082212200 R-2 ELK F NR RADCLIFF OH (LAT 39 09 41N LONG 082 21 22W)						
AUG 1989						
14...	170	20	1400	60	870	870
OCT						
16...	30	30	450	170	560	630
SEPT 1990						
04...	50	20	860	160	430	430
JUNE 1991						
12...	120	20	810	130	720	780
AUG						
26...	80	10	450	170	370	370
OCT						
25...	90	30	840	740	2400	2600
03201988 S-1 L RACCOON C NR VINTON OH (LAT 38 57 11N LONG 082 21 56W)						
AUG 1989						
15...	60	20	420	50	2500	2500
DEC						
04...	1200	30	4000	2200	2700	2300
SEPT 1990						
04...	50	30	490	70	2200	2200
JUNE 1991						
12...	70	10	520	210	2100	2100
AUG						
27...	90	10	250	40	2000	2100
OCT						
25...	970	80	2500	180	3100	3000
03160105 S-2 CAMPAIGN C NR GALLIPOLIS OH (LAT 38 53 51N LONG 082 11 31W)						
AUG 1989						
16...	150	20	740	<10	2900	2900
DEC						
05...	320	20	470	150	3000	2800
AUG 1990						
31...	160	20	750	40	850	800
JUNE 1991						
12...	500	20	1100	150	1300	840
AUG						
27...	190	<10	420	<10	2000	1900
OCT						
24...	240	<10	730	50	790	720

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Stream-flow, Instantaneous (ft ³ /s)	Specific con- duc- tance (μS/cm)	pH (stand- ard units)	Temper- ature, water (°C)	Acidity (mg/L as CaCO ₃)	Alka- linity, field (mg/L as CaCO ₃)	Sulfate, dis- solved (mg/L as SO ₄)
383005082280600 T-1 SYMMES C NR GETAWAY OH (LAT 38 30 05N LONG 082 28 06W)							
AUG 1989							
15...	16	445	7.6	22.5	--	65	110
DEC							
05...	184	270	7.6	2.0	--	48	71
AUG 1990							
31...	22	320	7.4	22.0	--	55	91
JUNE 1991							
13...	8.5	490	7.2	24.0	--	65	96
AUG							
28...	12	320	7.2	24.5	--	52	72
OCT							
24...	3.3	395	6.8	12.0	--	66	110
382715082242400 T-2 INDIAN GUYAN C NR BRADRICK OH (LAT 38 27 15N LONG 082 24 24W)							
AUG 1989							
15...	3.5	450	8.1	22.5	--	101	120
DEC							
05...	40	360	7.8	2.5	--	65	97
AUG 1990							
31...	3.8	520	7.9	23.0	--	94	170
JUNE 1991							
13...	2.0	710	7.5	23.0	--	75	240
AUG							
28...	.89	560	7.2	24.5	--	86	200
OCT							
24...	.70	713	7.0	14.5	--	94	270

Table 31. Water-quality data for long-term streamwater sites, 1989-91--Continued

Date	Alumi- num, total recov- erable (µg/L as Al)	Alumi- num, dis- solved (µg/L as Al)	Iron, total recov- erable (µg/L as Fe)	Iron, dis- solved (µg/L as Fe)	Manga- nese, total recov- erable (µg/L as Mn)	Manga- nese, dis- solved (µg/L as Mn)
383005082280600 T-1 SYMMES C NR GETAWAY OH (LAT 38 30 05N LONG 082 28 06W)						
AUG 1989						
15...	330	10	1500	<10	820	770
DEC						
05...	120	<10	870	260	280	270
AUG 1990						
31...	350	10	1200	80	430	400
JUNE 1991						
13...	120	<10	930	70	570	590
AUG						
28...	220	20	760	250	390	390
OCT						
24...	110	10	930	660	360	390
382715082242400 T-2 INDIAN GUYAN C NR BRADRICK OH (LAT 38 27 15N LONG 082 24 24W)						
AUG 1989						
15...	230	20	610	<10	340	290
DEC						
05...	80	<10	410	60	240	220
AUG 1990						
31...	100	30	460	50	430	420
JUNE 1991						
13...	60	<10	510	30	570	570
AUG						
28...	90	10	430	130	800	780
OCT						
24...	70	<10	530	440	630	640

Addendum to "Geologic Setting and Water Quality of Selected Basins in the Active Coal Mining Areas of Ohio, 1989-91, with a Summary of Water Quality for 1985-91," by Alan C. Sedam and Donna S. Francy (U.S. Geological Survey Water-Resources Investigations Report 93-4094)

1. Page 15, Line 27--"1991" should read "1989-91".
2. Page 49, Second paragraph--The sentence, "Streamwater sites B-1, B-4, and B-5 are in the Yellow Creek subbasin," should read "Streamwater sites B-1, B-4, B-5, and B-6 are in the Yellow Creek subbasin." Also, the sentence at the end of the paragraph, "Streamwater sites B-6, B-7, and B-8 are in the unnamed subbasin," should read "Streamwater sites B-7 and B-8 are in the unnamed subbasin."
3. Page 56 (map attached)--The short-term surface-water site J-7 near Holmesville was not indicated on the map in the report.

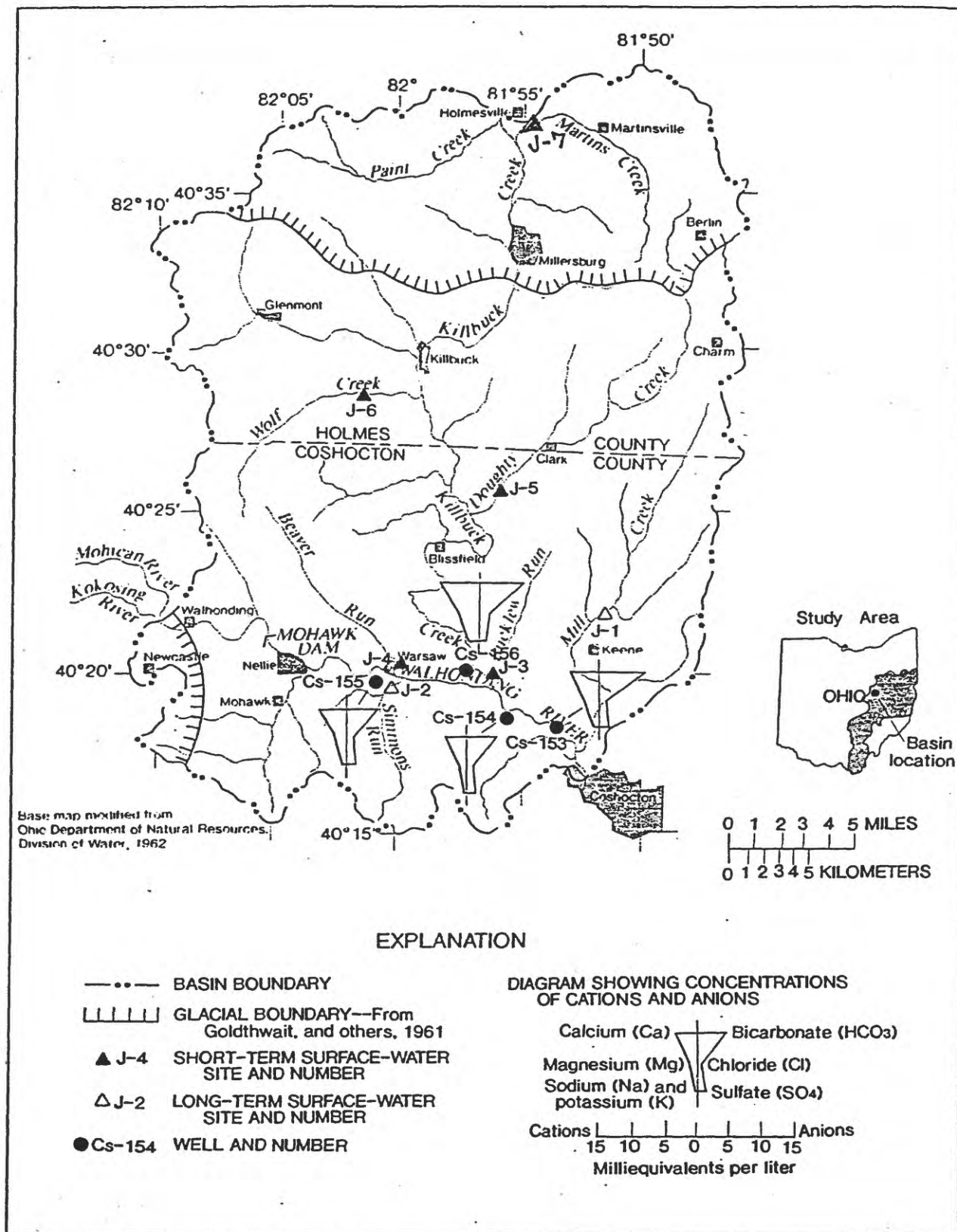


Figure 18.—Walhonding River basin (J), streamwater sites, ground-water sites, and Stiff diagrams for ground-water sites.