

Chemical Quality of Surface Water in the West Branch Susquehanna River Basin Pennsylvania

GEOLOGICAL SURVEY WATER-SUPPLY PAPER 1779-C

*Prepared in cooperation with the
Commonwealth of Pennsylvania
Department of Forests and Waters*



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By E. F. McCARREN

CONTRIBUTIONS TO THE HYDROLOGY OF THE UNITED STATES

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UNITED STATES DEPARTMENT OF THE INTERIOR

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GEOLOGICAL SURVEY

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CONTRIBUTIONS TO THE HYDROLOGY OF THE UNITED STATES

CHEMICAL QUALITY OF SURFACE WATER IN THE WEST BRANCH SUSQUEHANNA RIVER BASIN, PENNSYLVANIA

By E. F. McCARREN

ABSTRACT

The West Branch Susquehanna River is 228 miles long and drains 6,913 square miles of mountainous area in central Pennsylvania. Much of this area is forest-covered wilderness, part of which is reserved as State game land. Wild animals, such as deer, bear, turkey and grouse, are sheltered there, and many streams contain trout and other game fish. This helps to make the region one of the best hunting and fishing areas in Pennsylvania.

The Congress has approved Federal funds for the construction of several reservoirs to prevent flooding of the main river and several of its tributaries. Water stored behind the dams will not be withdrawn below a minimum level designated as conservation pools. These pools will be available for recreation.

Several headwater streams, such as Clearfield, Moshannon, and at times Sinnemahoning Creek, that carry drainage from coal mines are acid and contain high concentrations of dissolved solids, especially sulfates. These streams acidify the West Branch Susquehanna River downstream as far as Jersey Shore.

One of the most influential tributaries affecting the quality of the West Branch Susquehanna River after they merge is Bald Eagle Creek. Bald Eagle Creek enters the main river downstream from Lock Haven which is approximately 100 river miles from the river's source. Because of its alkaline properties, water of Bald Eagle Creek can neutralize acidic water. Many streams draining small areas and several draining large areas such as Pine Creek, Lycoming Creek, and Loyalsock Creek are clear nearly neutral water low in dissolved solids whose pH is about 7.0 most of the time. These streams have a diluting and neutralizing effect on the quality of the West Branch Susquehanna River, so that from Williamsport downstream the river water is rarely acid, and for most of the time it is of good chemical quality.

INTRODUCTION

This report describes the chemical quality of surface streams in the West Branch Susquehanna River basin above Lewisburg, Pa., which is 7 miles from the river mouth. The drainage area of the river at Lewisburg is 6,847 square miles. Quality studies were made of the river at selected locations along its length and of those tributaries that drained areas of 200 square miles or more. The quality of several tributaries draining areas less than 200 square miles was evaluated also because these tributaries affect the quality of the

river significantly when stream and river mix. These tributaries were the principal acidic streams that drained the bituminous coal fields of Clearfield and Cambria Counties.

The chemical quality of streams described is mostly based on data obtained by analyzing daily samples taken from the river near Curwensville from August 1960 to April 1961, at Lock Haven from October 1958 to September 1959, and at Lewisburg from October 1957 to September 1958. Other data based on chemical-quality records of the river and its tributaries obtained as early as 1944 were also used. At many places samples were taken only at high or low streamflow. All samples were analyzed by the U.S. Geological Survey for the most common ionic constituents of water and their individual concentrations. Dissolved solids, specific conductance, pH and color were also determined.

The Commonwealth of Pennsylvania proposes to construct reservoirs in the West Branch Susquehanna River basin on the main river near Curwensville, and on Bald Eagle Creek approximately 1 mile upstream from Blanchard. The Alvin R. Bush Reservoir on Kettle Creek was completed in 1962. It is located approximately 8.5 miles above Westport.

These reservoirs will prevent flooding of streams and will provide areas for storing water of good quality essential for the economical management and distribution of water throughout the river basin. How efficiently the water resources of a river basin are handled by management depends largely on the availability of water-quality information. Such information on acidic water in the basin is useful to engineers designing reservoirs because satellite structures, such as dams and their control gates, spillways and bridges, when made of ordinary materials, are strongly unresistant to acidic-water weathering. The degree and rapidity of deterioration of most structural materials are related directly to the quality of water to which they are exposed.

In the West Branch Susquehanna River basin intangible damages can be expected if streams bearing acidic coal-mine wastes are impounded for recreational use or for controlled distribution to augment streamflow or to improve or stabilize quality in downstream areas. For example, some forms of plankton and certain strains of bacteria may thrive abundantly in impounded acidic water that contains coal-mine wastes because most mine effluents contain heavy concentrations of iron and manganese which nurture their growth. If their growth is unchecked, colonies form to consume oxygen and increase the carbon dioxide content of the water, which may create an unfavorable environment for aquatic life. Water impounded in reservoirs is an unnatural innovation which can cause water temperature

changes in a river system. These water-temperature variations endanger fish life, not only in the impounded water areas but in the stream below. This is an important reason why continuous water temperatures are desirable quality data.

Documenting the changes in the water chemistry of streams that are directly related to reservoirs constructed in the West Branch Susquehanna River basin and where they occur will help to locate economically the facilities needed for storing, treating, and distributing water for public, industrial, and recreational uses.

The acidic character of the West Branch Susquehanna River in its upper part seldom extends into segments of the river below Jersey Shore. Below Jersey Shore the quality of the river supports aquatic life and is more suitable for industrial and domestic uses. However, wastes from some mines may extend into this part of the river occasionally, because of increased mining or because local storms wash more of the waste into receiving streams. The effects of these wastes may be far reaching if the river is at low flow and, therefore, inadequate to dilute them. This stream status, caused by streamflow and waste discharge, is a recognizable but preventable pollution hazard that can cause serious destruction of fish and degrade the West Branch Susquehanna River water generally for domestic, industrial, and recreational uses.

This study and similar investigations are carried out jointly by the Pennsylvania Department of Forests and Waters and the U.S. Geological Survey. The study fulfills part of the objectives of a cooperative program between State and Federal agencies that was begun in 1944.

LOCATION AND ENVIRONMENT OF THE WEST BRANCH SUSQUEHANNA RIVER BASIN

The West Branch Susquehanna River originates in Carrol Township of northwestern Cambria County at an elevation of 1,990 feet. From its source the river flows northward toward Gleason; then at North Bend it takes a southward course toward Lock Haven. The river resumes a northeastward course toward Muncy, changing again to flow southward to Northumberland where it combines with the North Branch to form the Susquehanna River. From its source to its mouth the river falls 578 feet, averaging 3.9 feet of fall per mile. The river from Lock Haven to its mouth falls 110 feet, at an average rate of 1.6 feet per mile.

The river drainage area of 6,913 square miles includes all or parts of Bradford, Cambria, Cameron, Centre, Clearfield, Clinton, Columbia, Elk, Indiana, Lycoming, McKean, Montour, Northumberland, Potter, Sullivan, Tioga and Union Counties (pl. 1). The total popu-

lation of these counties is nearly 1,000,000, according to the 1960 census, and the population of the largest city, Williamsport, is approximately 42,000.

The mean annual air temperature in the basin is 50° F, and the mean annual precipitation is 40 inches.

The West Branch Susquehanna River Basin drains the Allegheny Plateau. The basin is underlain by shale and sandstone containing coal, oil, iron ore, and in some areas, limestone. Acidic wastes from coal mining and associated operations in the upper basin drain into nearby streams. Because of this environment the river is acidic for more than half its length.

In the lower half of the basin, streams draining sandstone and limestone areas acquire alkaline characteristics. The alkaline properties of these tributaries neutralize some of the acid in the main river, but neutralization is not complete above Lock Haven. Not until the river receives the diluting water of Pine Creek and the neutralizing water of Bald Eagle Creek does its acidic character change.

WATER QUALITY IN THE WEST BRANCH SUSQUEHANNA RIVER BASIN

WEST BRANCH SUSQUEHANNA RIVER NEAR CURWENSVILLE

The results of the water-quality study of streams in the river basin are presented in downstream sequence from Curwensville to Lewisburg.

At Curwensville the river drains parts of Clearfield, Cambria, and Indiana Counties where coal deposits are recovered by deep-mining and strip-mining methods. Rocks adjacent to coal seams contain compounds of sulfur, and when strip mining exposes them to air and water, sulfuric acid is formed. Much of the water pumped to the surface from deep mines has been exposed to sulfur-bearing rocks underground for long periods of time, and therefore may be highly acidic and concentrated with solutes (McCarren, Wark, and George, 1961). Mining activities in the vicinity of Curwensville cause acidic wastes to be discharged to nearby streams.

The West Branch Susquehanna River drains an area of 315 square miles at Bower, which is approximately 18 miles upstream from Curwensville (pl. 1). At this location the average discharge for 47 years (1913-60) was 553 cfs (cubic feet per second) (table 1), which is approximately 360 mgd (million gallons per day).

In April 1960, field tests for specific conductance and pH of the river and its tributaries near Curwensville were made and samples taken for chemical analyses. From August 1, 1960, through May 9, 1961, samples were collected daily to determine specific conductance, pH and temperature of the river. The first 10 samples collected

TABLE 1.—Drainage areas and discharge of streams in West Branch Susquehanna River basin

Stream	Stream length (mi)	Drainage basin area (sq mi)	Gaging station and drainage area (sq mi)	Discharge at gage (cfs)		
				Average	Maximum	Minimum
West Branch Susquehanna River at Bower.....		315	At Bower.....	553 47 yrs (1913-60)	31,500 3-18-36	14 8-29-39
Clearfield Creek.....	62	386	At Dimeling.....	576 47 yrs (1913-60)	30,600 3-18-36	6 10-1,9-25
Moshannon Creek.....	46	290	At Osceola Mills.....	114 20 yrs (1914-60)	2,760 3-1-54	6.9 12-5-57
West Branch Susquehanna River at Karthaus.....		1,462	At Karthaus.....	2,499 20 yrs (1940-60)	50,900 4-1-40	109 9-30-43
Sinnemahoning Creek at Sinnemahoning.....	45	685	At Sinnemahoning.....	1,150 22 yrs (1938-60)	59,800 7-18-42	1.2 9-4-39
Kettle Creek.....	41	240	Near Westport.....	381 6 yrs (1964-60)	7,970 3-8-56	8.9 9-9,10-57
West Branch Susquehanna River at Renovo.....		2,975	At Renovo.....	4,875 47 yrs (1908-15, 1919-59)	236,000 3-18-36	80 12-6-08
Bald Eagle Creek.....	51	780	At Blanchard.....	469 6 yrs (1964-60)	5,130 10-15-55	117 8-10-55
Pine Creek.....	72	973	At Cedar Run.....	826 42 yrs (1918-60)	52,000 5-28-46	5.1 9-6-29
Lycoming Creek.....	23	276	Near Trout Run.....	280 46 yrs (1914-60)	21,800 5-27-46	3.2 9-27-36
West Branch Susquehanna River at Williamsport.....		5,682	At Williamsport.....	8,887 65 yrs (1896-1960)	284,000 3-18-36	162 9-17-43
Loyalsock Creek.....	59	493	At Loyalsock.....	762 35 yrs (1925-60)	51,200 11-16-36	12 9-26-39
Muncy Creek.....	33	216	Near Sonestown.....	48.8 20 yrs (1940-60)	7,310 3-11-52	.8 8-31-53
West Branch Susquehanna River at Lewisburg.....		6,847	At Lewisburg.....	10,880 21 yrs (1939-60)	262,000 5-29-46	9-1-33 -----

each month were composited and analyzed to determine quantities of silica, aluminum, iron, manganese, calcium, magnesium, sodium, potassium, bicarbonate, sulfate, chloride, fluoride, nitrate, dissolved solids, color, hardness, and acidity. Concentrations of acidity were used to estimate the quantities of sulfuric acid transported by the river.

Near Curwensville the river contains calcium sulfate water. An average of the analyses completed showed that calcium, magnesium, and sulfate represented approximately 75 percent of the dissolved solids. Calcium and magnesium occurring in rocks and soils in the area are dissolved in water by weathering processes. Although calcium and magnesium cause most of the hardness of the river water, aluminum, iron, and manganese from these rocks and hydrogen ions of the acid mine water also contribute to the hardness. Hardness ranged from 67 to 409 ppm (parts per million) table 2.

Sulfate, which results mostly from oxidation of coal mining wastes, ranged from 74 to 707 ppm from September 1957 to April 1961. Sulfuric acid discharge by the river was approximately 23 tons per day—a value obtained by multiplying the concentration of sulfuric acid (in parts per million) in the river by the discharge at time of sampling, using 0.0027 as a factor for converting to tons per day

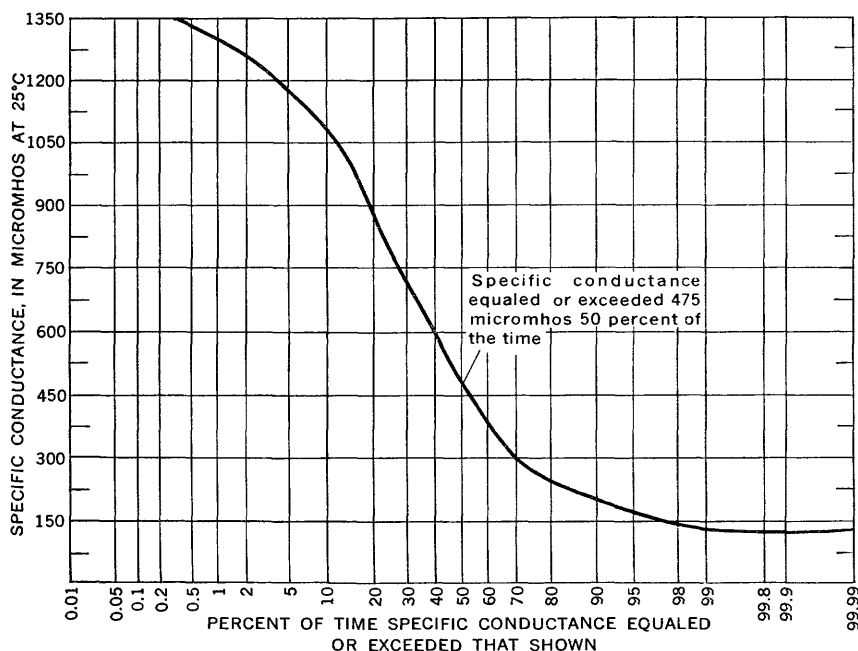


FIGURE 1.—Cumulative frequency curve of specific conductance, West Branch Susquehanna River near Curwensville, Pa., August 1960 to May 1961.

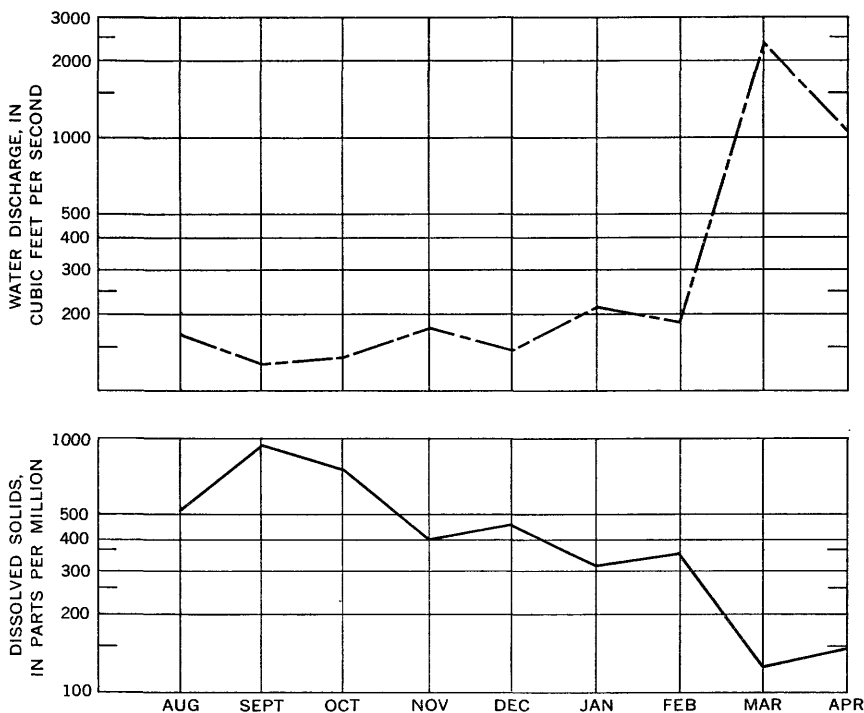


FIGURE 2.—Variations in dissolved-solids content and water discharge, West Branch Susquehanna River near Curwensville, Pa., monthly means, August 1960 to April 1961.

(table 3). The acidity of the river at Curwensville affects the chemical quality of the river downstream as far as Lock Haven where the river flows acidic most of the time (Pennsylvania State Planning Board, 1953).

Specific conductance is a useful characteristic of natural water to indicate the concentration of dissolved minerals. As mineral concentrations increase, the specific conductance also increases. (For the analyses of the river near Curwensville shown in table 2, the Dissolved-solids content (in parts per million) ranges from 61 to 75 percent of the specific conductance in micromhos).

Specific conductance of the river near Curwensville from August 1960 to May 1961 ranged from 132 to 1,390 micromhos (table 4); 50 percent of the time it equaled or exceeded 475 micromhos (fig. 1). The maximum specific conductance for the period of record was 1,400 micromhos on September 11, 1957.

Dissolved-solids content of the river water from 1960 to 1961 ranged from 125 to 934 ppm. As shown in figure 2, the dissolved-solids content was less concentrated during high flow than during low flow.

TABLE 2.—*Chemical analyses, in parts per million, of streams in West Branch Susquehanna River basin, Pa., 1945-61*
 [On dates marked by an asterisk (*) calcium and magnesium were not determined separately]

Date of collection	Mean discharge (cfs)	Temperature (° F)	Silica (SiO ₂)	Aluminum (Al)	Iron (Fe)	Manganese (Mn)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Dissolved solids (residue on evaporation at 180° C)	Hardness as CaCO ₃		Total acidity as H ₂ SO ₄	Specific conduct- ance (microhmhos at 25° C)	pH	Color
West Branch Susquehanna River at McGees Mills																						
Sept. 11, 1945*	-----	72	9.5	6.9	1.0	1.6	108	33	47	0	544	4.5	-----	-----	0.9	780	471	471	82	1,100	3.40	2
Apr. 26, 1960	164	69	-----	-----	-----	-----	-----	-----	85	0	493	1.0	-----	-----	.7	-----	320	320	78	993	3.70	1
Chest Creek at Patton																						
Apr. 26, 1960	-----	66	-----	-----	-----	-----	-----	-----	-----	24	34	1.5	-----	-----	5.0	-----	54	33	-----	133	6.9	1
Chest Creek at Westover																						
Apr. 24, 1960	-----	67	-----	-----	-----	-----	-----	-----	-----	18	94	2.0	-----	-----	3.6	-----	112	97	-----	256	6.6	1
Chest Creek at Mahaffey																						
Sept. 11, 1945*	-----	70	5.4	-----	0.06	-----	35	10	5.8	25	101	10	-----	0.1	1.3	192	128	108	-----	302	6.8	12
Apr. 26, 1960	-----	69	-----	-----	-----	-----	-----	-----	3.9	16	105	2.5	-----	-----	2.8	-----	120	107	-----	280	6.7	1
West Branch Susquehanna River at Bower																						
Aug. 7, 1944*	98.2	74	9.2	5.7	0.14	1.1	54	16	21	0	283	5.9	-----	-----	1.6	410	268	258	68	722	3.30	4
Apr. 16, 1945*	454	54	7.6	2.6	.03	.40	33	11	9.0	1.2	0	153	2.0	.1	1.4	233	146	146	22	364	4.20	1
Sept. 11*	-----	73	6.0	3.7	.45	.80	48	16	12	0	228	6.0	-----	-----	1.0	353	220	220	45	573	3.65	2
Apr. 24, 1959	2	48	6.8	1.9	.08	.38	29	9.9	9.8	1.3	0	136	3.8	.1	.9	210	113	113	29	311	4.30	2

Bell Run at Bells Landing

Apr. 26, 1960	65	3	66	158	6.7
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Anderson Creek at Curwensville

Year	70	4.6	2.1	.24	.80	88	3.1	0	53	1.0	0.6	84	56	56	25	170	3.85
Sept. 11, 1948*	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---
Oct. 7, 1957	56	---	---	.98	---	---	---	0	81	3.0	.2	---	38	38	39	245	3.60
Apr. 26, 1960	---	---	---	---	---	---	---	0	57	---	---	---	---	---	20	177	3.90

West Branch Susquehanna River near Curwensville

Sept. 11, 1957.	65	3.7	0.50	0.94	46	14	22	25	0	707	2.0	1.4	296	173	173	142	1,400	3.20
Jan. 11, 1958.	32	8.6	3.7	0.50	46	14	22	25	0	196	2.0	1.4	296	173	173	142	1,400	3.20
Apr. 26, 1960.	62	7.1	3.7	0.04	93	41	13	20	1.3	0	222	5.0	3.35	340	156	24	442	4.40
June 10.	313	8.6	3.7	0.04	93	41	13	20	1.3	0	213	4.0	2.8	340	156	24	443	4.40
July 28.	241	9.8	7.0	1.0	1.5	70	19	40	1.4	0	380	7.0	1.8	562	253	59	501	4.40
Aug. 1-10.	78	9.8	7.4	0.93	63	20	38	4.4	4.0	0	342	4.0	1.5	514	239	54	457	4.40
Sept. 1-10.	130	13	14	0.66	106	35	76	3.9	0	650	6.0	2.2	934	409	108	312	3.70	
Oct. 1-10.	55	15	11	0.66	86	32	60	3.0	2.4	0	540	4.0	1.2	730	346	83	306	3.35
Nov. 1-10.	168	12	11	0.45	49	16	30	2.0	0	276	3.0	1.4	392	189	29	582	3.35	
Nov. 16.	147	8.4	3.5	0.03	42	13	23	1.8	0	208	2.5	1.1	314	159	24	425	4.00	
Nov. 4-10.	88	9.9	3.0	0.03	36	13	25	1.9	0	305	6.0	1.5	453	210	39	478	4.00	
Dec. 1-10, 1961.	214	11	2.9	0.06	41	13	29	1.8	0	206	4.5	2.2	314	144	24	490	4.35	
Feb. 1-10, 1961.	178	11	3.9	0.06	1.0	4.6	4.2	1.0	0	231	5.0	1.1	351	156	39	538	3.60	
Mar. 1-10.	2,400	6.7	1.2	0.03	55	19	4.6	4.2	0	74	3.0	1.1	125	67	67	103	199	4.40
Apr. 1-10.	1,050	9.9	1.1	0.02	52	21	7.5	7.3	1.5	0	95	2.0	1.3	84	84	10	243	4.50
May 1-9.	5.3	7.9	1.2	0.02	57	24	8.5	4.3	1.8	0	109	1.5	.60	95	95	10	265	4.50

Clearfield Creek at Dimeling

Aug. 8, 1944*	110	80	8.4	5.8	0.35	1.6	37	14	-----	0	212	1.8	0.2	0.0	331	224	224	78	653	3.10	8
Apr. 16, 1945*	598	55	7.5	3.0	.18	.6	19	7.1	2.6	1.1	113	2.0	.2	.8	162	111	111	74	147	3.50	3
Apr. 24, 1969	610	-----	8.6	1.5	.03	.7	23	11	2.7	1.3	0	124	2.2	.1	210	113	113	29	330	3.50	3

Clearfield Creek near Clearfield

[illegible]

TABLE 2.—*Chemical analyses, in parts per million, of streams in West Branch Susquehanna River Basin, Pa., 1945-61—Continued*

[On dates marked by an asterisk (*) calcium and magnesium were not determined separately]

Date of collection	Mean discharge (cfs)	Temperature (° F)	Silica (SiO ₂)	Aluminum (Al)	Iron (Fe)	Manganese (Mn)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Dissolved solids (residue on evaporation at 180° C)	Hardness as CaCO ₃		Total acidity as H ₂ SO ₄	Specific conduct- ance (microhms at 25° C)	pH	Color
																	Calcium, magne- sium	Noncar- bonate				
West Branch Susquehanna River at Clearfield																						
Oct. 7, 1957.....	54	57	-----	-----	-----	-----	-----	-----	-----	-----	0	513	13	-----	1.1	-----	280	280	2.0	1,070	3.20	2
Moshannon Creek at Osceola Mills																						
Aug. 7, 1944*.....	65.9	76	12	18	2.9	2.8	49	21	6.2	-----	0	393	2.2	0.2	0.1	636	399	399	227	1,210	2.80	5
Aug. 10,*.....	18.0	71	20	23	11	3.4	71	29	18	-----	0	593	3.1	-----	1.0	797	584	584	342	1,700	2.60	5
Apr. 16, 1946*.....	116	50	9.8	7.9	2.3	1.6	30	13	1.7	1.4	0	236	1.4	.2	.4	349	245	245	62	788	2.55	4
Apr. 24, 1959.....	129	-----	9.4	2.4	.16	1.5	27	15	2.7	1.5	0	188	2.0	.1	.2	270	129	129	64	505	3.20	3
Moshannon Creek at Phillipsburg																						
Oct. 7, 1957.....	25	54	-----	-----	11	2.6	-----	-----	-----	-----	0	515	5.0	-----	1.3	-----	310	310	3.2	1,210	2.55	3
West Branch Susquehanna River at Karthaus																						
Aug. 8, 1944*.....	850	75	9.0	6.9	0.26	1.6	49	15	15	-----	0	266	7.9	-----	0.6	384	256	256	88	750	3.20	5
Apr. 17, 1945.....	2,490	56	6.8	2.8	.07	.4	22	7.6	4.8	1.0	0	114	1.4	0.1	1.0	167	116	116	35	333	3.60	2
Oct. 10, 1957.....	190	57	-----	-----	5.7	2.0	-----	-----	-----	-----	0	417	10	-----	.5	-----	264	264	2.2	1,020	2.90	8
Sinnemahoning Creek at Cameron																						
July 19, 1957.....	-----	77	-----	-----	-----	-----	-----	-----	8.0	-----	24	19	6.0	-----	0.5	80	34	15	-----	120	7.5	2

Bennett Branch Sinnemahoning Creek near Driftwood

Oct. 10, 1957	14.1	58						0	228	22	0.2	-----	116	116	3.3	571	3.35	4
Sinnemahoning Creek at Sinnemahoning																		
Aug. 10, 1944*	398	77	11	6.1	0.14	0.9	22	9.2	6.6	0	126	12	0.2	1.0	205	135	59	3.80
May 23, 1945*	1,480	58	3.4	7.1	.05	.00	5.8	1.8	2.3	6	18	1.8	.1	.4	35	22	59	7.3
Sept. 12, 1945*			10	7.1	.20	1.2	26	10		0	141	8.0		1.0	223	159	380	3.70
Oct. 10, 1957	39	61			.32	.89				0	138	18		.2		105	347	4.00
Oct. 30, 1958										10	32	4.0		.9		41	102	6.5
Oct. 30, 1958			6.1		.01	.33	8.4	3.2	2.0	.9	34	4.0		.2		34	101	4.8
Dec. 10, 1958	680								14		55	2.8		.4	69	32	104	5.2
Apr. 16, 1959	980		5.2	0	0	.05	6.5	2.3	3.0	7	30	4.0		.2	99	26	104	4.8
Apr. 23	820		5.6	1.8	.04	.31	8.2	5.7	3.0	1.3	0	60		.2	50	26	72	4.10
May 27	455		6.2	2.6	.03	.72	16	4.5	5.5	1.7	0	78		.3	44	50	201	3.80
July 14	116		11	5.4	.16	1.9	20	10	12	1.8	0	141		.4	136	54	268	3.70
Aug. 17	34			5.0	.04	1.9	21	8.5	14	1.8	0	115		.2	135	91	396	3.80
Oct. 22		48	6.8	5.6	.07	.25	9.0	2.8	4.0	1.5	5	31		.1	64	88	458	3.90
Dec. 17			8.1	1.1	.03	.54	10	4.7	3.0	1.4	0	59		.1	112	45	93	5.5
Jan. 26, 1960			6.9	2.3	.04	.65	11	4.7	4.2	1.0	0	66		.2	112	45	207	3.95
Mar. 10									7.8		2	43		.4		39	207	3.95
Apr. 20											0	145		.2	231	98	123	4.8
Apr. 31			54	6.9	.07	2.4	26	14	13	4.5	0	189		.2	98	98	397	3.70
Oct. 12			46	6.2	.03	1.2	17	8.2	8.2	1.4	0	94		.1	152	123	484	3.75
Nov. 15			32								0	106		.5	196	76	291	4.10
Jan. 6, 1961			32								0	90		.4	161	86	324	3.80
Feb. 9			38	6.8	.1	.00	.17	9.0	4.2	1.3	0	42		.2	75	76	273	4.30
Mar. 22											1	3.0		.4		36	120	4.7

Kettle Creek at Cross Fork

Sept. 12, 1945*		70															52	6.9	5
Oct. 10, 1957	9.0	61							7.8		24	7.9	3.2				64	6.4	8

Kettle Creek near Cross Fork

Aug. 11, 1944*		73	3.1		0.01		6.5	1.2	1.7		20	7.9	1.8	0.0	0.2		60	6.8	10
Apr. 18, 1945*	7.32	47	2.8		.03		4.6	1.3	1.9		14	7.1	.8	.1	1.0	32	41	6.6	9
Sept. 12	142	71									24	7.0	1.0		.4	25	52	6.8	5

Kettle Creek at Dam Site near Westport

Sept. 12, 1957	10.3	68	2.8	.1	0.03	0.02	8.3	1.8	3.2	1.2	20	12	6.2	0.1	0.4	58	84	6.6	2
Oct. 9, 1957	16.5	65	2.6	.1	.01	.01	7.5	1.4	2.9	1.2	18	9.2	7.0	.1	.9	53	78	6.0	5
Nov. 20	103	42	2.7	.1	.01	.00	7.1	1.2	1.7	.9	16	10	3.1	.1	.2	23	61	6.4	2

TABLE 2.—*Chemical analyses, in parts per million, of streams in West Branch Susquehanna River Basin, Pa., 1945-61—Continued*

[On dates marked by an asterisk (*) calcium and magnesium were not determined separately]

Date of collection	Mean discharge (cfs)	Temperature (° F)	Silica (SiO ₂)	Aluminum (Al)	Iron (Fe)	Manganese (Mn)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Dissolved solids (residue on evaporation at 180° C)	Hardness as CaCO ₃		Total acidity as H ₂ SO ₄	Specific conduct- ance (microhmhos at 25° C)	pH	Color	
																	Calcium- magne- sium	Noncar- bonate					
Kettle Creek at Westport																							
July 13, 1945*		77									6	18	1.0		0.1			19			61	6.6	4
Sept. 12*	30 (est)	73									10	37	2.0		.7			31			80	6.5	3
July 19, 1957		83							5.1		6	58	6.5			.0		42			107	5.9	2
Oct. 9.	17.0	66									0										173	4.25	3
Bald Eagle Creek at Blanchard																							
Oct. 11, 1956	246	54	6.1		0.02		42	13	3.6	1.8	1167	19	5.0	0.1	6.1	184	158	22			325	8.5	3
Jan. 2, 1957	425	35	7.6		.02		37	11	2.5	1.5	144	19	3.8	.1	7.0	164	137	20			281	8.4	3
Oct. 28, 1958	191		5.4		.00		38	15	4.7	1.6	165	19	6.5	.2	7.5	185	157	22			321	7.4	3
Dec. 14	185						20		4.6		132	20	6.2		6.7		132	24			272	7.8	3
Apr. 14, 1959	837		5.6		.00		20	6.3	3.5	1.5	132	20	6.2		3.0	116	76	16			170	7.7	3
July 16	154		10		.00		41	14	5.5	1.7	170	21	8.0	.0	4.8	238	160	21			330	8.1	3
Aug. 20	127								3.2		172	21	8.5		5.2		170	29			343	7.7	3
Oct. 21		67									158	21	9.0		7.5	208	158	29			322	7.9	3
Nov. 18		38	5.6		.00		36	10	6.0	1.8	132	22	6.0	.1	7.6	159	131	23			275	7.6	3
Dec. 16											88	22	3.8		4.6	123	96	24			202	7.8	1
Jan. 28, 1960		16			.01	0.03	37	12	6.8	1.4	133	25	8.0	.1	7.8	180	142	32			298	8.2	3
Mar. 7			7.1		.01		37	12	5.0	1.2	135	24	6.0	.1	4.4	166	142	32			288	7.9	2
Apr. 18			6.4		.01		38	12	4.0	1.4	138	23	8.5	.0	8.5	183	145	32			291	7.7	3
June 6			6.6			.02	33	10	2.8	1.3	128	20	5.2	.2	4.8	174	124	19			256	7.8	3
July 11											184	21	6.5		5.8	190	172	21			330	8.0	3
Nov. 16		49	5.9		.00		42	16	6.8	1.7	178	18	7.4	.0	7.4	198	171	25			341	7.9	3
Dec. 29											174	23	9.0		5.6	236	154	12			350	8.0	3
Feb. 17, 1961		38	13		.02	.00	36	8.4	6.0	2.0	112	26	10	.1	6.1	168	125	33			281	6.7	3
Mar. 20		42	17		.00	.00	27	8.0	6.3	1.8	100	19	4.5		4.5	140	101	19			222	7.4	3

Bald Eagle Creek at Beech Creek Station

Aug. 12, 1944*	187	70	2.7	0.02	40	16	2.0	176	14	5.0	0.0	5.0	188	106	21	328	7.8	9
Apr. 18, 1945*	903	55	4.5	.02	0.00	8.2	2.3	196	17	2.8	.0	4.1	112	199	20	100	7.6	8
Oct. 9, 1957	188	67	---	.43	.00	---	17	166	22	6.6	---	8.5	---	140	4	325	7.7	8

Pine Creek at Cedar Run

Aug. 30, 1944*	53.1	68	2.4	0.02	14	3.3	6.1	25	26	7.5	1.2	0.6	75	48	28	186	7.2	4
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Pine Creek at Waterville

July 14, 1945*	---	73	---	---	---	---	---	23	15	4.5	---	0.1	---	33	14	87	7.2	8
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Pine Creek near Jersey Shore

Oct. 13, 1950	---	54	4.5	0.04	7.1	1.7	1.1	1.0	14	1.5	0.2	0.7	42	25	13	64	7.4	7
Nov. 15, 1950	---	---	3.3	.07	6.3	1.7	1.0	.7	12	1.3	.3	.4	37	23	13	57	7.4	4
Dec. 15, 1950	2,840	---	4.0	.02	6.3	1.6	.9	.5	17	1.3	.2	.4	38	22	18	57	6.8	2
Mar. 27, 1951	2,160	40	4.8	.03	6.3	1.6	3.3	.9	14	2.9	.1	1.2	42	22	18	92	6.8	6
May 2, 1951	4,398	---	3.5	.05	7.4	1.6	1.4	.5	13	1.3	.1	.4	43	22	10	53	7.3	3
June 9, 1951	---	71	2.2	.05	7.8	1.6	2.6	1.4	22	2.3	.1	.4	49	28	10	74	7.3	4
July 1, 1951	745	75	3.6	.08	7.8	1.6	2.8	1.0	30	1.3	.1	.3	46	26	11	74	7.0	5
Aug. 11, 1951	398	72	1.9	.05	10	2.8	3.7	1.0	20	3.8	.1	.3	52	36	12	58	7.1	5
Oct. 9, 1957	73	60	---	---	---	---	8.7	---	28	5.0	---	.5	---	32	9	106	6.7	5

Lycoming Creek near Trout Run

Nov. 10, 1952	---	---	---	---	---	---	1.1	1.1	14	2.6	---	0.8	---	27	16	68	7.2	3
Dec. 18, 1952	---	37	---	---	---	---	1.7	.7	11	2.6	---	---	---	18	19	48	7.2	3
Jan. 21, 1953	---	38	---	---	---	---	2.3	8.0	11	3.5	---	1.3	---	19	16	48	7.1	4
Feb. 28, 1953	---	38	---	---	---	---	3.0	8.0	11	2.7	---	1.2	---	18	16	47	7.1	4
Mar. 31, 1953	---	40	---	---	---	---	1.6	8.0	10	2.7	---	1.4	---	18	12	47	6.0	4
May 9, 1953	---	44	---	---	---	---	1.8	8.0	11	1.4	---	1.3	---	19	16	47	7.0	3
July 21, 1953	---	70	---	---	---	---	1.7	13	14	1.8	---	1.3	---	27	16	57	7.2	3
Aug. 18, 1953	---	63	---	---	---	---	1.6	9.0	10	1.8	---	2.7	---	19	12	57	7.2	3
Oct. 9, 1957	72	59	---	---	---	---	2.1	13	14	3.2	---	1.2	---	20	16	50	7.4	3
---	---	---	---	---	---	---	7.6	19	15	2.9	---	1.7	---	20	4	77	6.6	4

¹ Includes equivalents of 2 ppm of carbonate (CO₃).

² Includes equivalents of 4 ppm of carbonate (CO₃).

TABLE 2.—*Chemical analyses, in parts per million, of streams in West Branch Susquehanna River Basin, Pa., 1945-61—Continued*
 [On dates marked by an asterisk (*) calcium and magnesium were not determined separately]

Date of collection	Mean discharge (cfs)	Temperature (° F)	Silica (SiO ₂)	Aluminum (Al)	Iron (Fe)	Manganese (Mn)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Dissolved solids (residue on evaporation at 180° C)	Hardness as CaCO ₃		Total acidity as H ₂ SO ₄	Specific conduct- ance (micromhos at 25° C)	pH	Color
																	Calcium, magne- sium	Noncar- bonate				
Lycoming Creek at Williamsport																						
Sept. 13, 1945*		71									22	10	2.0		1.5		33	15				
West Branch Susquehanna River at Williamsport																						
July 19, 1944*	1,550	76	4.0		0.04	0.0	21	7.0	5.6	14	71	5.1	0.0	.9		130	81	70		204	6.8	3
Mar. 22, 1945*	55,000	44	4.6				5.8	1.8	2.2	7	17	1.1	.1	1.6		39	22	16		58	6.1	5
Sept. 13*		75								32	89	10		1.6			108	82		283	6.6	8
Oct. 9, 1957	708	61			.40	.19			28	25	137	18		3.4			130	110		385	6.5	
Loyalsock Creek at Loyalsock																						
July 20, 1944*	103	68	3.0				6.0	1.4	1.1	13	10	1.2	.0	.5		31	21	10		51	7.0	4
Mar. 26, 1945*	1,360	49	3.2		0.01	0.0	4.5	1.0	1.4	7	10	1.0	.0	.9		28	15	10		41	6.7	5
Nov. 10, 1952	51									12	9.4	2.0		.8			20	10		54	7.0	3
Dec. 18	789	36							1.3	7.0	11	1.7		1.6			18	12		47	7.1	3
Jan. 21, 1953	1,240	38							.3	7.0	10	1.3		.8			18	12		43	7.0	3
Feb. 28	750	39							.4	7.0	13	1.5		.9			33	22		73	6.4	7
Mar. 30	1,760	44							.9	7.0	9.9	1.4		.9			17	11		42	6.9	7
May 5	901	51							1.3	7.0	9.6	1.6		1.1			16	10		43	7.1	4
June 8	751	63							1.0	10	9.0	1.2		1.0			18	10		46	7.1	4
July 13	63	73							1.5	15	9.1	1.8		1.1			20	9		53	7.1	3
Aug. 18	46	68							1.0	16	8.4	1.9		.7			22	10		57	6.9	8
Oct. 9, 1957	313	55							5.5	17	9.4	2.6		.7			16	2		61	6.6	4

Muncy Creek at Sonesdown

Aug. 31, 1944*	2.21	66	3.0	0.02	0.0	5.6	1.0	1.6	13	7.6	1.8	0.1	0.2	28	18	7	45	7.1	7
Mar. 28, 1945*	67.8	50	3.8	0.01	0.0	4.0	.8	1.7	7	8.4	.8	.1	1.4	24	13	8	35	6.4	8

Muncy Creek at Muncy

Oct. 9, 1957	17	56					11	43	21	6.4			6.8		47	12	149	7.1	4
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White Deer Creek at White Deer

Sept. 14, 1945*	26.5	64	3.7	0.02		4.5	1.5		17	3.3	.9	0.1	.4	26	17	3	38	6.7	18
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Buffalo Creek at Lewisburg

July 14, 1945*		70							101	13	1.5		4.3		98	15	203	7.5	7
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TABLE 3.—Sulfuric acid discharge, West Branch Susquehanna River near Curwensville, Pa., April 1960 to April 1961

Sample-collection period	Mean discharge (cfs)	Sulfuric acid		Number of days	Sulfuric acid (tons per interval)
		(ppm)	(tons per day)		
1960					
Apr. 22.....	342	20	18	1	18
Apr. 26.....	313	24	20	1	20
June 10.....	241	24	16	1	16
July 28.....	241	59	38	1	38
Aug. 1-10.....	130	54	19	10	190
Sept. 1-10.....	55	108	16	10	160
Oct. 1-10.....	72	83	16.1	10	161
Nov. 1-10.....	158	29	12.4	10	124
Nov. 16.....	147	24	10	1	10
Dec. 4-10.....	88	39	9	7	63
1961					
Jan. 1-10.....	214	29	16.8	10	168
Feb. 1-10.....	178	39	18.8	10	188
Mar. 1-10.....	2,400	10	64.8	10	648
Apr. 1-10.....	1,050	10	28.4	10	284
Total.....				92	2,088

Aluminum, iron, manganese, and hydrogen ions in equivalents per million (table 5) made up 10 percent of the constituents analyzed; calcium and magnesium 26 percent, sodium and potassium 14 percent, sulfate 48 percent, and chloride, fluoride, and nitrate 2 percent (fig. 3).

From August 1, 1960, to May 9, 1961, the river was acidic. The pH ranged from 3.25 to 4.8 (table 6). Water temperature during the same period ranged from freezing for many days to 81° F in August (table 7).

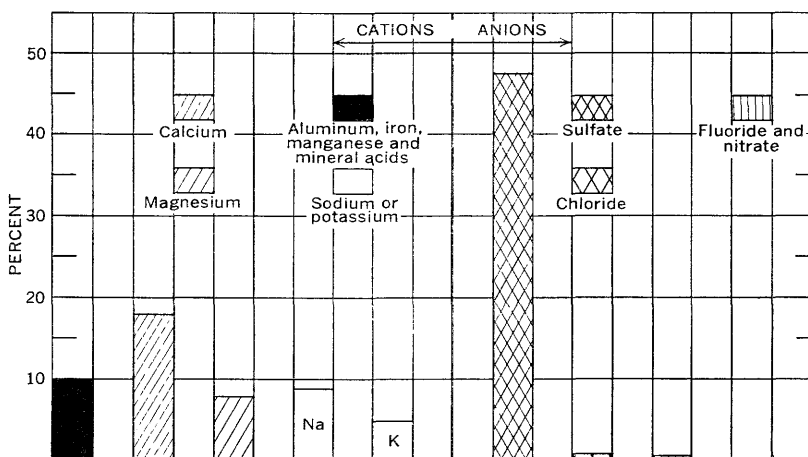


FIGURE 3.—Average percentage composition, based on equivalents per million, West Branch Susquehanna River near Curwensville, Pa., 1960-61.

QUALITY OF SURFACE WATER, SUSQUEHANNA RIVER BASIN C17

TABLE 4.—Specific conductance, in micromhos at 25° C, West Branch Susquehanna River near Curwensville, Pa., August 1, 1960 to May 9, 1961

Day	August	September	October	November	December	January	February	March	April	May
1.....	658	1,250	967	915	-----	496	505	198	257	230
2.....	733	1,250	1,000	738	-----	520	504	198	256	252
3.....	733	1,260	1,020	580	-----	-----	522	218	223	252
4.....	665	1,250	1,030	466	579	481	520	217	234	278
5.....	760	1,220	1,060	540	566	481	521	228	238	303
6.....	723	1,230	1,220	541	692	476	550	229	237	328
7.....	818	1,300	1,190	344	690	504	557	132	237	250
8.....	741	1,330	1,150	641	706	506	575	178	229	250
9.....	771	1,390	1,150	511	679	386	563	184	247	242
10.....	850	1,290	1,110	529	723	382	-----	197	246	-----
11.....	887	1,010	1,110	534	696	348	567	214	213	-----
12.....	829	380	1,070	504	736	313	570	213	190	-----
13.....	792	301	1,080	432	-----	316	-----	226	195	-----
14.....	842	337	1,120	443	756	331	517	232	197	-----
15.....	861	373	1,120	460	867	332	-----	204	172	-----
16.....	933	510	1,130	497	797	358	509	201	153	-----
17.....	978	471	1,190	334	738	412	322	214	153	-----
18.....	1,040	521	1,180	532	-----	413	385	232	189	-----
19.....	996	604	1,150	564	808	379	385	253	190	-----
20.....	1,030	690	1,140	516	797	380	250	255	200	-----
21.....	1,010	721	1,090	585	650	380	147	229	222	-----
22.....	1,030	633	1,010	586	731	385	147	244	242	-----
23.....	1,010	660	996	653	747	380	193	236	196	-----
24.....	1,030	660	989	664	649	382	193	239	166	-----
25.....	1,090	660	813	701	680	395	132	229	189	-----
26.....	1,120	738	800	-----	708	397	132	226	194	-----
27.....	1,080	785	975	689	704	420	154	246	168	-----
28.....	1,080	848	1,020	623	776	446	154	246	195	-----
29.....	1,100	908	914	659	751	446	-----	253	210	-----
30.....	1,130	943	911	-----	659	501	-----	211	230	-----
31.....	1,200	-----	962	-----	540	-----	-----	242	-----	-----

TABLE 5.—Conversion factors—Parts per million to equivalents per million

Ion	Multiply By—
Aluminum (Al).....	0.11119
Bicarbonate (HCO ₃).....	.01639
Calcium (Ca).....	.04990
Carbonate (CO ₃).....	.03333
Chloride (Cl).....	.02820
Fluoride (F).....	.05263
Hydrogen (H).....	.99206
Hydroxide (OH).....	.05880
Iron (Fe ⁺²).....	.03581
Iron (Fe ⁺³).....	.05372
Magnesium (Mg).....	.08224
Manganese (Mn ⁺²).....	.03640
Manganese (Mn ⁺⁴).....	.07281
Nitrate (NO ₃).....	.01613
Potassium (K).....	.02558
Sodium (Na).....	.04350
Sulfate (SO ₄).....	.02082

Chest Creek and Bell Run flow into the river between McGees Mills and Curwensville and usually have a lower dissolved-solids content than the West Branch Susquehanna River. Although there is some strip mining in the Chest Creek drainage basin, neither the water of Chest Creek nor that of Bell Run was acidic at the time of sampling. These tributaries, therefore, dilute and partly neutralize acid water of the main stream. Anderson Creek, which flows into the West Branch Susquehanna River at Curwensville, also drains a strip-mining area and was acidic when sampled (table 2).

The Curwensville dam will create a conservation pool that will be available for recreation, although impounded water in this region is likely to contain acidic coal-mine wastes. If and when water stored in the reservoir has a quality inferior to that of the river below the dam, flow from storage can be withheld until its quality has been improved or it can be released in harmless increments. Furthermore, releases of stored water from above the Curwensville dam can be delayed until discharge and quality of the receiving stream can help to counteract the effects of inferior or acidic water by dilution or neutralization.

TABLE 6.—*pH values, West Branch Susquehanna River near Curwensville, Pa., October 1, 1960 to May 9, 1961*

Day	October	November	December	January	February	March	April	May
1-----	3.60	3.35	-----	4.35	3.85	4.45	4.30	4.55
2-----	3.60	3.55	-----	4.35	3.75	4.45	4.40	4.60
3-----	3.55	3.70	-----	-----	3.80	4.40	4.30	4.40
4-----	3.65	3.80	3.90	4.80	3.90	4.45	4.50	4.50
5-----	3.55	4.00	3.90	4.20	4.05	4.50	4.60	4.50
6-----	3.30	4.00	3.80	4.15	3.85	4.45	4.60	4.55
7-----	3.30	4.10	3.75	4.00	3.60	4.60	4.60	4.60
8-----	3.30	3.85	3.75	3.95	3.40	4.50	4.55	4.70
9-----	3.30	3.85	3.75	4.20	3.30	4.50	4.40	4.60
10-----	3.40	3.85	3.75	4.25	-----	4.55	4.45	-----
11-----	3.35	3.80	3.95	4.00	3.75	4.60	4.55	-----
12-----	3.35	3.85	3.70	4.30	3.60	4.60	4.60	-----
13-----	3.35	3.85	-----	4.35	-----	4.70	4.45	-----
14-----	3.35	3.80	3.60	4.40	3.60	4.60	4.55	-----
15-----	3.40	4.00	3.55	4.40	-----	4.55	4.65	-----
16-----	3.35	3.90	3.70	4.35	3.40	4.60	4.45	-----
17-----	3.30	4.00	3.90	4.10	3.60	4.60	4.55	-----
18-----	3.35	3.75	-----	4.05	3.95	4.60	4.50	-----
19-----	3.45	3.65	3.90	4.25	3.95	4.65	4.60	-----
20-----	3.55	3.65	3.90	4.20	3.95	4.65	4.60	-----
21-----	3.60	3.65	4.10	4.10	4.45	4.70	4.55	-----
22-----	3.55	3.55	4.05	4.00	4.60	4.70	4.50	-----
23-----	3.50	3.45	4.15	4.15	4.40	4.40	4.65	-----
24-----	3.35	3.25	4.20	4.15	4.35	4.45	4.75	-----
25-----	3.55	3.50	4.15	4.30	4.50	4.50	4.60	-----
26-----	3.55	-----	4.10	4.10	4.60	4.60	4.65	-----
27-----	3.35	3.70	4.10	4.05	4.45	4.60	4.70	-----
28-----	3.40	3.90	4.00	4.00	4.50	4.60	4.50	-----
29-----	3.75	3.80	4.15	3.90	-----	4.60	4.55	-----
30-----	3.75	-----	4.20	3.80	-----	4.70	4.60	-----
31-----	3.50	-----	4.15	-----	-----	4.80	-----	-----

TABLE 7.—*Water temperatures (°F), West Branch Susquehanna River near Curwensville, Pa., August 1, 1960 to May 9, 1961*

[Measured at approximately 1:30 p.m. daily]

Day	August	September	October	November	December	January	February	March	April	May
1.....	80	80	62	53	-----	32	32	41	40	50
2.....	79	78	61	51	-----	32	32	41	40	50
3.....	79	76	62	46	-----	32	32	42	41	47
4.....	76	74	61	48	41	32	32	42	40	50
5.....	78	76	61	45	41	32	32	44	43	53
6.....	79	73	60	43	39	33	32	44	45	52
7.....	78	76	57	39	40	34	32	46	45	57
8.....	80	73	59	43	38	34	32	41	43	57
9.....	81	73	59	51	32	32	32	40	41	61
10.....	79	77	61	41	32	33	34	38	41	-----
11.....	79	63	63	41	32	33	34	36	41	-----
12.....	75	63	61	40	32	32	34	36	43	-----
13.....	75	62	61	45	-----	34	34	40	41	-----
14.....	78	62	63	47	33	32	36	40	44	-----
15.....	76	64	65	49	34	32	36	43	49	-----
16.....	79	67	64	50	32	32	36	40	45	-----
17.....	77	66	61	45	32	32	36	37	45	-----
18.....	79	67	58	44	-----	32	33	39	44	-----
19.....	79	66	55	47	32	32	37	39	44	-----
20.....	78	68	54	47	32	32	37	41	46	-----
21.....	75	67	52	43	32	32	35	43	44	-----
22.....	76	68	51	44	32	32	35	40	50	-----
23.....	74	70	50	42	32	32	41	40	55	-----
24.....	75	71	50	42	32	32	41	40	58	-----
25.....	73	72	47	44	34	32	44	45	57	-----
26.....	76	69	59	-----	33	32	44	45	54	-----
27.....	74	67	50	45	33	32	42	50	51	-----
28.....	75	68	52	49	32	32	43	50	51	-----
29.....	80	66	51	50	32	33	-----	49	49	-----
30.....	80	67	50	-----	32	32	-----	46	50	-----
31.....	79	-----	50	-----	32	32	-----	43	-----	-----

The acidic water of the West Branch Susquehanna River near Curwensville will attack ordinary concrete and metals if they are used in the construction of the Curwensville dam. An example of damage done by acidic water to lock and dam structures was illustrated at the Loyahanna Reservoir in western Pennsylvania. After 1 year of service, the control gates had to be replaced because of the corrosive damage done by the impounded water (U.S. Congress, House Committee on Public Works, 1962). Usually streams that are acidic because of mining wastes contain relatively high concentrations of sulfate, which will disintegrate concrete, as shown in the following table (U.S. Bureau of Reclamation, 1949).

Degree	Sulfate in water samples (ppm)
Negligible.....	<150
Positive.....	150-1, 000
Considerable.....	1, 000-2, 000
Severe.....	>2, 000

The sulfate content of samples removed from the river near Curwensville in March to May, 1961 was 74, 95, and 109 ppm. However, the sulfate content of other samples taken since 1957 ranged from

196 to 707 ppm (table 2). Thus, the foregoing table indicates that the water of the West Branch Susquehanna River would attack concrete.

Furthermore, acidic water is not recommended for use in mixing concrete, although some tests in which acid mine water was used in mixing portland-cement concrete indicated an increase in tensile strength or an acceleration in setting time (California Water Pollution Control Board, 1952). Therefore, it may be advisable not to use the water from some surface streams for mixing concrete until samples are tested.

Compared with natural water the West Branch Susquehanna River at Curwensville contains high concentrations of aluminum, iron, and manganese, as do most streams that drain coal mining regions. Manganese concentrations ranged from 0.55 to 3.6 ppm in the 11 samples analyzed (table 2). These concentrations are considered high for most industrial uses (American Society for Testing Materials, 1959). Although the river is acidic, manganese remains in solution; but it precipitates if oxygen is available, and the pH is increased. Neutralization by seepage of alkaline ground water is one way the river pH may be increased.

Iron and manganese impart strong staining characteristics to water when they precipitate as insoluble hydrates. In water, iron and manganese also nurture the growth of some micro-organisms, such as *Crenothrix*, which colonize in reservoirs (American Water Works Association, 1951). When a *Crenothrix* colony forms, it often attaches to the sides of reservoirs, aqueducts, or spillways and appears as a brown or black slimy mass. Unsightly, *Crenothrix* is an alkali-producing colony and, therefore, may promote the precipitation of iron and manganese from water to cause localized staining. Once iron or manganese is precipitated and formed on the reservoir, further precipitation of these ions may continue by catalysis in the area affected.

CLEARFIELD CREEK

Approximately 9 miles east of Curwensville, Clearfield Creek flows into the West Branch Susquehanna River near Clearfield. The Creek is so named because of the clearings along its banks. It is 62 miles long and drains 396 square miles of parts of Cambria and Clearfield Counties. The average discharge of the Creek at Dimeling for 47 years of record (1913-60) was 576 cfs (table 1).

This subbasin is underlain by shale, sandstone, and rich deposits of bituminous coal. As a stream receiving coal mine wastes, Clearfield Creek contains high concentrations of solutes and is acidic most of the time. Little Clearfield Creek, less concentrated with dissolved solids (Pennsylvania State Planning Board, 1947), flows into

Clearfield Creek above Clearfield, diluting somewhat the main creek before it reaches the West Branch Susquehanna River.

Clearfield Creek, which contains a calcium sulfate water (table 2), discharges sulfuric acid into the West Branch Susquehanna River.

MOSHANNON CREEK

Downstream from the confluence of Clearfield Creek with the West Branch Susquehanna River, Moshannon Creek enters the main branch from the south near Karthaus. Moshannon is a corruption of a Delaware Indian word meaning "elk creek." The creek is 46 miles long and drains an area of 290 square miles of parts of Cambria, Blair, Clearfield, and Centre Counties. Discharge records of Moshannon Creek at Osceola Mills appear in table 1.

Underlying this subbasin are sedimentary rocks of shale and sandstone that contain coal deposits. Moshannon Creek also transports wastes from coal mines to the West Branch Susquehanna River. At Osceola Mills Moshannon Creek contains an acidic heavily concentrated calcium and magnesium sulfate water. Calcium, magnesium, and sulfate constitute about 86 percent of the dissolved solids. At times of analysis, the dissolved-solids content ranged from 270 to 800 ppm. The stream contains high concentrations of aluminum, iron, and manganese. For the four samples collected, noncarbonate hardness averaged approximately 350 ppm (table 2).

WEST BRANCH SUSQUEHANNA RIVER AT KARTHAUS

Approximately 42 river miles separate Curwensville from Karthaus. The Drainage area of the West Branch Susquehanna River at Karthaus is 1,462 square miles, including the area drained by Mosquito Creek. The discharge of the river at Karthaus averaged 2,499 cfs, or approximately 1,610 mgd, for the 20-year period (1940-60).

The principal ions in the river at Karthaus are calcium and sulfate. As with most streams, the solute concentration is greater at times of low discharge than at high discharge. Note in table 2 the two samples of August 1944 and April 1945; each was taken at a time of active mining in the headwater area. The April sample was taken when the discharge was average; the August sample was taken when the discharge was approximately one-third as great. In the August sample many solute concentrations are twice as great as in the April sample. The third sample (October 1957) was taken some years after most mining operations had ceased. At that time the discharge was about one-thirteenth of the average. Most concentrations are even greater than in the August 1944 sample, although the concentration of free sulfuric acid is considerably less. Approximately 200 tons of sulfuric acid flowed past Karthaus on August 8, 1944, and 235 tons on April 17,

1945, but only 55 tons on October 10, 1957, which was approximately one-fourth as much as when coal mining was more active in the head-water area.

SINNEMAHONING CREEK

Sinnemahoning Creek drains the largest tributary area in the West Branch Susquehanna River basin. Sinnemahoning is a Delaware Indian word meaning "stony lick." Its basin is underlain by shale and sandstone containing some bituminous coal. The Creek enters the river at Keating, near Karthaus. This stream is approximately 45 miles long and drains an area of 1,030 square miles of parts of Clearfield, Elk, McKean, Cameron, Potter, and Clinton Counties. The average discharge of the stream at Sinnemahoning for a 22-year period (1938-60) was 1,150 cfs (table 1), or approximately 743 mgd.

Sinnemahoning Creek contains a calcium sulfate water, and water in many segments of the stream is usable for industrial, domestic, and recreational purposes. At Costello the stream is used, after treatment, as an industrial supply. However, the stream carries some mining wastes and, therefore, is acidic at times. Waste from paper mills also reaches the stream. For one sample collected during low flow on October 10, 1944, at Sinnemahoning the pH was 3.80, specific conductance 335 micromhos, dissolved-solids content 205 ppm, and hardness 135 ppm. When the flow was a little greater than average on May 23, 1945, the pH was 7.3, specific conductance 59 micromhos, dissolved-solids content 35 ppm, hardness 22 ppm, and bicarbonate concentration 6 ppm (table 2).

KETTLE CREEK

Kettle Creek flows through 41 miles of a picturesque valley of hemlocks and natural wilderness of the Pennsylvania State game land in Clinton County. It is an excellent fishing stream that contains trout, muskellunge, sucker, and fallfish (Pennsylvania Fish Commission, 1961).

The Creek drains an area of 240 square miles of parts of Tioga, Potter, Cameron, and Clinton Counties and flows into the West Branch Susquehanna River at Westport. Underlying the basin are shale and sandstone containing some coal deposits in the upper basin. For a 6-year record near Westport (1954-60) the average discharge was 381 cfs (table 1).

Part of the time Kettle Creek contains a calcium bicarbonate water and part of the time a calcium sulfate water. The bicarbonate and sulfate ion concentrations are predominant at alternate times. Although the stream is acidic on occasion because of coal mine waste (pH 4.25, table 2), its chemical quality is generally good. Most of the time the pH is less than 7.0, and dissolved-solids content is less than 65 ppm.

Completed in 1962 the Alvin R. Bush Dam on Kettle Creek is part of a flood-control project controlling a drainage area of 226 square miles. The dam is located approximately 8.5 miles from the mouth of the Creek. During periods of water surplus the reservoir and recreation areas can be used for storage. Releases of good-quality water from storage could augment flow in the Creek and the West Branch Susquehanna River during dry periods and improve the quality downstream by diluting the concentration of dissolved solids. In addition, the releases of good water from the Kettle Creek Reservoir to the West Branch Susquehanna River could help counteract the deleterious effects of accidental spillages of pollutants into the river.

WEST BRANCH SUSQUEHANNA RIVER AT LOCK HAVEN

There are approximately 100 river miles between Curwensville and Lock Haven. Based on streamflow records of the river at Renovo (table 1), the estimated average discharge at Lock Haven for a 47-year record was 5,460 cfs, or approximately 3,530 mgd.

Although the West Branch Susquehanna River improves in quality between Curwensville and Lock Haven, its water is still acidic at Lock Haven. The improved quality is largely attributable to the river's many small tributaries above Lock Haven that contain bicarbonates and have a lower concentration of dissolved solids than is in the river. Tributary water having these characteristics helps to dilute concentrations of solutes and to neutralize part of the acidity in the main river.

The bicarbonate furnished by some tributaries and that which enters the main stream directly from underground seepage is insufficient, however, to neutralize the river at Lock Haven, where the pH of 4.5 is exceeded only 10 percent of the time (table 8). From October 1, 1958, to September 30, 1959, the estimated discharge of

TABLE 8.—*Given values of physical and chemical properties that were equaled or exceeded for indicated percent of days, West Branch Susquehanna River at Lock Haven, Pa., October 1945 to September 1951*

	Values equaled or exceeded for indicated percent of days				
	1	10	50	90	99
Specific conductance.....micromhos at 25°C..	685	530	225	135	105
pH.....	4.8	4.5	4.05	3.60	3.30
Calcium (Ca).....ppm..	45	36	15	9.8	7.7
Magnesium (Mg).....ppm..	14	5.9	3.5	2.7	1.6
Sodium (Na) and Potassium (K).....ppm..	9.8	7.8	3.4	2.1	1.6
Sulfate (SO ₄).....ppm..	247	191	80	46	35
Chloride (Cl).....ppm..	8.3	6.4	2.7	1.7	1.3
Dissolved solids.....ppm..	288	123	74	58	36
Hardness as CaCO ₃ppm..	222	172	74	45	36

TABLE 9.—*Sulfuric acid discharge, West Branch Susquehanna River at Lock Haven, Pa., October 1958 to September 1959*

Collection period	Mean discharge (cfs)	Sulfuric acid discharge (tons per day)	Number of days	Sulfuric acid discharge (tons per interval)
<i>1958</i>				
Oct. 1-10.....	2,960	192	10	1,920
Oct. 11-18.....	1,480	116	8	928
Oct. 19-31, Nov. 1-9.....	1,680	132	22	2,904
Nov. 10-30.....	3,620	147	11	1,617
Nov. 21-30, Dec. 1-6.....	4,120	111	16	1,776
Dec. 7-8.....	5,230	212	2	424
Dec. 9-10, 12-21.....	2,860	77	12	924
Dec. 22-31.....				
<i>1959</i>				
Jan. 1-4.....	2,080	85	14	1,190
Jan. 5.....	2,700	146	1	146
Jan. 6-17.....	2,060	56	12	672
Jan. 18-21.....	4,610	124	4	496
Jan. 22-26.....	34,800	940	5	4,700
Jan. 27-31, Feb. 1-10.....	6,320	256	15	3,840
Feb. 11-14.....	2,140	87	4	348
Feb. 15-24.....	9,090	245	10	2,450
Feb. 25-28, Mar. 1-6.....	5,550	150	10	1,500
Mar. 7-24.....	10,500	425	18	7,650
Mar. 25-31, Apr. 1-2.....	12,100	327	10	3,270
Apr. 3-8.....	21,900	296	6	1,776
Apr. 9-17.....	9,660	261	9	2,349
Apr. 18-22, 24-27.....	6,630	269	9	2,421
Apr. 23.....	7,020	550	1	550
Apr. 28-30, May 1-3.....	25,100	678	6	4,068
May 4-11.....	9,970	404	8	3,232
May 12-27.....	4,760	257	16	4,112
May 28-31, June 1-6.....	2,350	127	10	1,270
June 7-26.....	1,510	118	20	2,360
June 27-30, July 1-2.....	2,010	81	6	486
July 3-10.....	1,330	86	8	688
July 11-26.....	806	74	16	1,184
July 27-31, Aug. 1.....	716	75	6	450
Aug. 2-6.....	597	87	5	435
Aug. 8-17.....	511	61	10	610
Aug. 18-24.....	626	100	7	700
Aug. 25-28.....	909	83	4	332
Aug. 29-31, Sept. 1-18.....	601	71	21	1,491
Sept. 19-30.....	248	36	12	432
Total.....			364	65,701

sulfuric acid in the river flowing past Lock Haven was 180 tons per day (table 9).

During the 1959 water year the river at Lock Haven contained calcium sulfate water. The specific conductance of the water was less than 230 micromhos 50 percent of the time (fig. 4); the weighted-average noncarbonate hardness was 85 ppm (table 10); and the temperature of the river equaled or exceeded 75°F 26 percent of the time (fig. 5).

BALD EAGLE CREEK

Bald Eagle Creek flows into the West Branch Susquehanna River below Lock Haven. The stream was named after the North American eagle, which frequented the basin area in Colonial days. It is 51 miles long and drains an area of 780 square miles of parts of Centre and Clinton Counties (table 1). Its principal tributaries are Fishing,

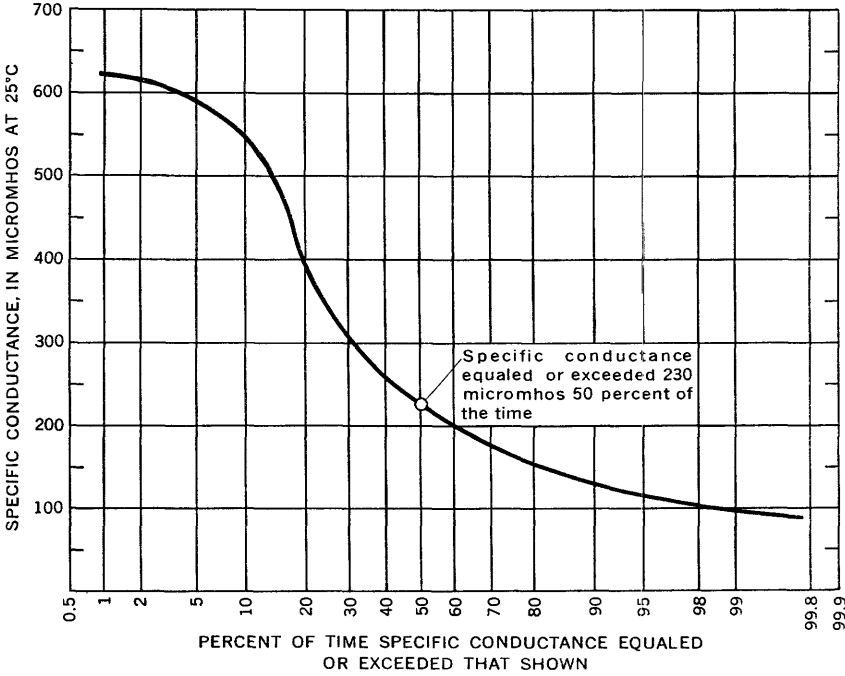


FIGURE 4.—Cumulative frequency curve of specific conductance, West Branch Susquehanna River at Lock Haven, Pa., October 1958 to September 1959.

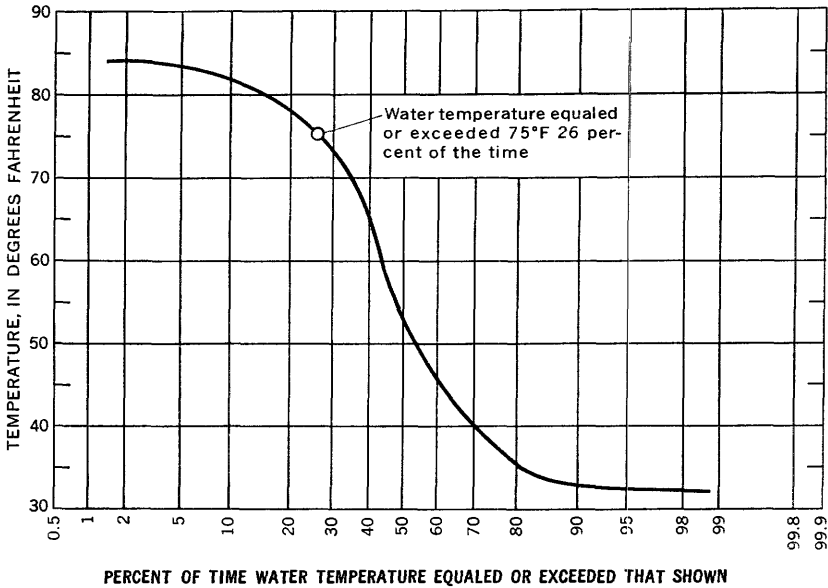


FIGURE 5.—Cumulative frequency curve of water temperature, West Branch Susquehanna River at Lock Haven, Pa., October 1958 to September 1959.

TABLE 10.—*Chemical analyses, in parts per million, West Branch Susquehanna River at Lock Haven, Pa., water year October 1958 to September 1959*

LOCATION—Center of Lockport Bridge which is situated at the termination of North Jay Street, Lock Haven, Clinton County, and 30.1 miles downstream from gaging station at Renovo, Pa.

DRAINAGE AREA—3,337 square miles.

RECORDS AVAILABLE—Chemical analyses: October 1945 to September 1959. Water temperatures: October 1945 to September 1951, October 1958 to September 1959.

EXTREMES 1958-59.—Dissolved solids: Maximum 380 ppm, August 18-24; minimum 68 ppm, January 22-26, March 25-31, April 1-2. Hardness: Maximum, 189 ppm, September 19-30; minimum, 33 ppm, January 22-26. Specific conductance: Maximum daily, 781 micromhos, August 4; minimum daily, 98 micromhos, April 3. Water temperatures: Maximum, 84° F., July 30-31, August 16; minimum, freezing point on several days during winter months.

EXTREMES 1945-51, 1958-59.—Dissolved solids (1945-47, 1958-59): Maximum, 380 ppm, August 18-24, 1959; minimum, 68 ppm, January 22-26, March 25-31, April 1-2, 1959. Specific conductance: Maximum daily, 781 micromhos, September 18, 1951; minimum, 33 ppm, January 22-26, 1958. Water temperatures: Maximum 84° F., July 30-31, August 16, 1959; minimum, freezing point on many days during winter months.

REMARKS.—Records of specific conductance and pH of daily samples available in district office at Philadelphia, Pa. Records of discharge for water year October 1958 to September 1959 based on records for West Branch Susquehanna River at Renovo, Pa., which are given in Water-Supply Paper 1622.

Date of collection	Mean discharge (cfs)	Temperature (° F)	Silica (SiO ₂)	Aluminum (Al)	Iron (Fe)	Manganese (Mn)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Dissolved solids (residue on evap at 180° C)	Hardness as CaCO ₃			Total acidity as (H ⁺)	Specific conduct- ance (micromhos at 25° C)	pH	Color
																	Calcium magne- sium	Noncar- bonate	Calcium magne- sium				
Oct. 1-10, 1958	2,960	—	—	—	—	—	—	—	—	—	0	88	3.2	—	0.0	—	69	69	—	0.5	265	3.85	2
Oct. 11-18	1,480	—	—	—	—	—	—	—	—	—	0	106	3.9	—	—	—	84	84	—	0.6	275	4.15	2
Oct. 19-31, Nov. 1-9	1,680	—	6.7	2.3	0.08	1.3	23	10	6.0	1.3	0	121	4.3	0.1	—	180	99	90	—	6	301	4.20	2
Nov. 10-20	3,620	—	—	—	—	—	—	—	—	—	0	72	3.6	—	2	—	68	68	—	3	220	4.30	2
Nov. 21-30, Dec. 1-6	4,120	—	7.5	1.0	.00	.31	12	5.4	2.2	1.4	0	61	2.6	.1	2	90	52	52	—	3	164	4.5	2
Dec. 7-8, 9-10, 12-21	5,230	—	—	—	—	—	—	—	7.6	—	0	78	3.4	—	—	—	55	55	—	3	213	4.30	2
Dec. 22-31, Jan. 1-4, 1959	2,860	—	—	—	—	—	—	—	—	—	0	98	3.5	—	4.7	—	72	72	—	3	175	4.5	2
Jan. 5	2,090	—	—	—	—	—	—	—	—	—	0	115	—	—	—	—	64	64	—	3	221	4.25	3
Jan. 6-17	2,700	—	—	—	—	—	—	—	—	—	0	77	2.8	—	—	—	64	64	—	4	286	4.15	3
Jan. 18-21	2,060	—	—	—	—	—	—	—	—	—	0	74	—	—	1.3	—	64	64	—	2	198	4.40	3
Jan. 22-26	4,610	—	—	—	—	—	—	—	—	—	0	77	—	—	—	—	33	33	—	2	211	4.30	3
Jan. 27-31, Feb. 1-10	34,800	—	—	—	—	—	—	—	—	—	0	38	1.0	—	—	—	68	68	—	2	116	4.45	3
Feb. 11-14	6,320	—	—	—	—	—	—	—	—	—	0	67	1.9	—	—	—	106	54	54	3	186	4.35	2
Feb. 15-24	2,140	—	—	—	—	—	—	—	—	—	0	43	—	—	—	—	52	52	—	3	134	4.30	2
Feb. 25-28, Mar. 1-6	9,090	—	—	—	—	—	—	—	—	—	0	61	2.3	—	—	—	60	60	—	2	177	4.30	3
Mar. 7-24	5,550	—	—	—	—	—	—	—	—	—	0	74	2.3	—	—	—	48	48	—	2	201	4.20	3
Mar. 25-31, Apr. 1-2	10,500	—	6.9	—	.00	.41	9.6	3.9	2.5	1.0	0	58	2.3	—	—	—	48	48	—	2	161	4.10	3
Apr. 3-8	12,100	—	—	—	—	—	—	—	—	—	0	43	2.0	.1	—	—	37	37	—	2	135	4.5	3
Apr. 9-17	21,900	—	—	—	—	—	—	—	—	—	0	42	1.6	—	—	—	50	50	—	2	121	4.30	3
Apr. 18-22, 24-27	9,660	—	—	—	—	—	—	—	—	—	0	59	2.3	—	—	—	71	71	—	2	169	4.15	2
Apr. 23	6,630	—	—	—	—	—	—	—	—	—	0	67	2.2	—	—	—	107	56	56	3	193	4.05	2
Apr. 28-30, May 1-3	7,020	—	7.1	1.2	.00	.55	12	5.7	2.8	1.3	0	67	3.4	.1	—	—	100	54	54	6	182	4.00	2
	25,100	—	—	—	—	—	—	—	—	—	0	46	2.0	—	—	—	80	41	41	2	142	4.30	2

QUALITY OF SURFACE WATER, SUSQUEHANNA RIVER BASIN C27

[illegible]

Beech and Spring Creeks. This subbasin is underlain by sandstone and limestone.

Bald Eagle Creek contains an alkaline calcium bicarbonate water (table 2). Bicarbonate is the dominant ion. The creek helps to neutralize and to dilute much of the acidity of the West Branch Susquehanna River when the creek and river mix.

Upstream at Blanchard, 14 miles from the mouth of Bald Eagle Creek, the average flow for a 6-year period (1954-60) was 469 cfs. However, the average discharge of this stream into the West Branch Susquehanna River probably is greater because other streams contribute water to Bald Eagle Creek below Blanchard. The first appearance of bicarbonate is in segments of the West Branch Susquehanna River below the mouth of Bald Eagle Creek. Limestone springs that underlie this subbasin give Bald Eagle Creek a dependable dry-season flow. One of these springs at Bellefonte gushes approximately 11,500,000 gallons of pure drinking water daily.

Because all the available supply from the limestone springs is not needed by Bellefonte and surrounding communities, approximately 8,000 gpd are directed to Spring Creek, tributary to Bald Eagle Creek at Milesburg. This spring water is reported to have a year-round temperature of 50° F.

Several small tributaries in the basin are used for domestic supply. Spring Creek and Bald Eagle Creek, downstream as far as Blanchard, are stocked with brook, brown, and rainbow trout. Other species in the stream are smallmouth bass and pickerel. Sportsmen call areas along Spring Creek "fisherman's paradise."

A flood-control reservoir will be constructed on Bald Eagle Creek, approximately 1 mile upstream from Blanchard and 15 miles from the stream mouth. An area of 339 square miles will be drained above the dam site. A conservation pool of about 370 acres will be available for recreation. Because of the year-round availability of water from the many prolific springs in the area, storage of large quantities of good quality water is possible in the impoundment areas. Regulated releases of impounded water could be used to help maintain flow and quality of Bald Eagle Creek and the West Branch Susquehanna River downstream from its confluence with Bald Eagle Creek.

PINE CREEK

Pine Creek, a major tributary of the West Branch Susquehanna River, enters the river near Jersey Shore. The creek lies within a glacial area in its headwaters and flows through Pine Creek Gorge, often referred to as the "Grand Canyon of Pennsylvania". The gorge extends from Mount Tom to Jersey Shore. The stream is 72 miles long and drains 973 square miles of parts of Tioga, Potter, and

Lycoming Counties (table 1). Its main tributaries are Marsh, Babb, and Little Pine Creeks. The subbasin is underlain with shale and sandstone, containing coal and oil deposits.

Pine Creek contains a mixed-type calcium bicarbonate sulfate water that is low in dissolved solids. For the times the stream was analyzed near Jersey Shore the dissolved solids did not exceed 52 ppm, pH ranged from 6.7 to 7.4, and specific conductance did not exceed 105 micromhos (table 2). When the waters of Pine Creek mix with the West Branch Susquehanna River, they help to dilute the concentration of solutes in the main river.

Based on 42 years of record (1918-60), the average discharge of Pine Creek at Cedar Run was 826 cfs, or approximately 520 mgd.

At Waterville the recorded average discharge for 12 years (1908-20) was 1,144 cfs (U.S. Geological Survey, 1960), or approximately 740 mgd. However, at Waterville the discharge does not include the discharge of Little Pine Creek, which drains 172 square miles of Tioga and Lycoming Counties. Therefore, the average discharge of Pine Creek into the West Branch Susquehanna River near Jersey Shore is considerably more than the reported average discharge of Pine Creek at Waterville.

Pine Creek is used as a source of supply to the Galeton-Eldred Water Co. It is also open to fishermen who may catch brown and rainbow trout, small-mouth bass, walleyes, suckers, and bullheads.

LYCOMING CREEK

Lycoming Creek flows into the West Branch Susquehanna River at Williamsport. Lycoming is a Delaware Indian word meaning "sandy stream." The creek is approximately 23 miles long and drains 276 square miles of parts of Bradford, Lycoming, Sullivan, and Tioga Counties (table 1). The subbasin is underlain by shale and sandstone in which are coal deposits.

Lycoming Creek contains a calcium bicarbonate and calcium sulfate water of good quality. The maximum of 10 specific conductances measured on the stream from 1952 to 1957 was 77 micromhos. The range of pH was 6.6 to 7.4 (table 2).

For most of its length, Lycoming Creek provides the environment for the propagation of aquatic life. This stream contains trout, small-mouth bass, suckers, and fallfish.

WEST BRANCH SUSQUEHANNA RIVER AT WILLIAMSPORT

The West Branch Susquehanna River at Williamsport contains a calcium magnesium sulfate water, and commonly has a pH between 6.0 and 7.0. Sulfate is the dominant constituent (table 2).

The acidic characteristic of the river in upstream segments has been changed noticeably in the reach of the river between Lock

Haven and Williamsport. Evidently several tributaries of good-quality water and ground-water seepages from limestone and sandstone aquifers underlying this part of the basin have produced the more diluted and neutral water that flows past Williamsport.

Concentration of dissolved solids decreases as water discharge increases, and conversely the concentration increases when water discharge is low. Low flow usually occurs during the summer; consequently, the concentrations of most solutes are at their highest level at this time.

The average discharge of the river at Williamsport for 65 years (1895-1960) was 8,887 cfs (table 1), or approximately 5,730 mgd.

The Pennsylvania State Health Department has established an automatic monitoring station at Williamsport for detecting acidity and changes in the quality of the West Branch Susquehanna River. Acidity, or other harmful properties of water, can sometimes be counteracted by dilution with good water released from upstream impoundments once they become available. This would tend to lessen the influence of harmful properties of water downstream.

LOYALSOCK CREEK

A short distance to the east of Williamsport, Loyalsock Creek flows into the West Branch Susquehanna River at Montoursville. Loyalsock is a corruption of the Indian Lawi-saquit meaning "middle creek." The stream is 59 miles long and drains an area of 493 square miles of parts of Bradford, Lycoming, Sullivan, and Wyoming Counties. Most of the subbasin lies within the glacial area and is underlain by shale and sandstone containing deposits of soft anthracite.

At Loyalsock, 7 miles upstream from the mouth of the creek, the average discharge for 35 years (1925-60) was 762 cfs (table 1). Loyalsock Creek has good-quality calcium bicarbonate sulfate water, low in dissolved solids (table 2). The stream contains trout, small-mouth bass, walleyes, suckers, and bullheads.

MUNCY CREEK

Muncy Creek joins the West Branch Susquehanna River near Muncy, approximately 20 miles upstream from Lewisburg. Muncy is a corruption of the Indian tribal name Minsi, meaning "wolf." The creek is 33 miles long and drains 216 square miles of parts of Columbia, Lycoming, Montour, and Sullivan Counties. Shale, sandstone, and limestone underlie this subbasin.

The average discharge of the stream near Sonestown for 20 years of record (1940-60) was 48.8 cfs (table 1).

Muncy Creek contains good-quality calcium bicarbonate sulfate water, low in dissolved solids (table 2). The species of fish available

in the stream are trout, smallmouth bass, bullheads, suckers, and fallfish.

WEST BRANCH SUSQUEHANNA RIVER AT LEWISBURG

Lewisburg, on the West Branch Susquehanna River, is approximately 7 miles upstream from the river mouth at Northumberland.

During early phases of the cooperative program between the Geological Survey and the Pennsylvania Department of Forests and Waters, the West Branch Susquehanna River at Lewisburg was selected as a site for daily sampling and chemical analyses. Sampling was continuous from 1944 to 1953 and, after a lapse of 3 years, was resumed in 1956. For most of the period of record, daily samples were composited into groups consisting of samples taken in sequence and composited at intervals of 10 days.

The average discharge of the river at Lewisburg for the 21-year period (1939-60) was 10,880 cfs (table 1), or approximately 7,020 mgd. The average discharge of 11,390 cfs for the 1958 water year (table 11) was approximately 4 percent greater than the average discharge for the period of record. Runoff during the 1958 water year was greater during the winter and spring than during the summer and autumn. This pattern of runoff at Lewisburg is characteristic of the entire river basin and causes seasonal variations in dissolved solids and specific conductance. Figure 6 shows variations in specific conductance and water discharge at Lewisburg for the period October 1944 to September 1950.

The specific conductance of the river at Lewisburg from October 1957 to September 1958 was less than 285 micromhos 90 percent of the time and exceeded 205 micromhos only 30 percent of the time (fig. 7). The average for the year was 184 micromhos; a maximum of 406 micromhos occurred October 7 during a period of low flow, and a minimum of 68 micromhos occurred April 8 during a period of high flow.

At Lewisburg the West Branch Susquehanna River contains a calcium sulfate water. Natural water usually contains calcium as a major constituent because compounds of calcium are widespread in soils and sedimentary rock and many of them are readily soluble, both underground and at the surface, through processes of weathering.

Calcium represents approximately 16 percent of the dissolved material in the river at Lewisburg and, combined with magnesium, also common in soils, represents more than 20 percent of the materials in solution (fig. 8). The ratio of calcium to magnesium in the river at Lewisburg, based on parts per million, is usually about 3 to 1. When computed from equivalents per million, the ratio is about 1.8 to 1.

Although the river at Lewisburg is nonacidic most of the time, the acidic character of the upper river accounts for the concentration of

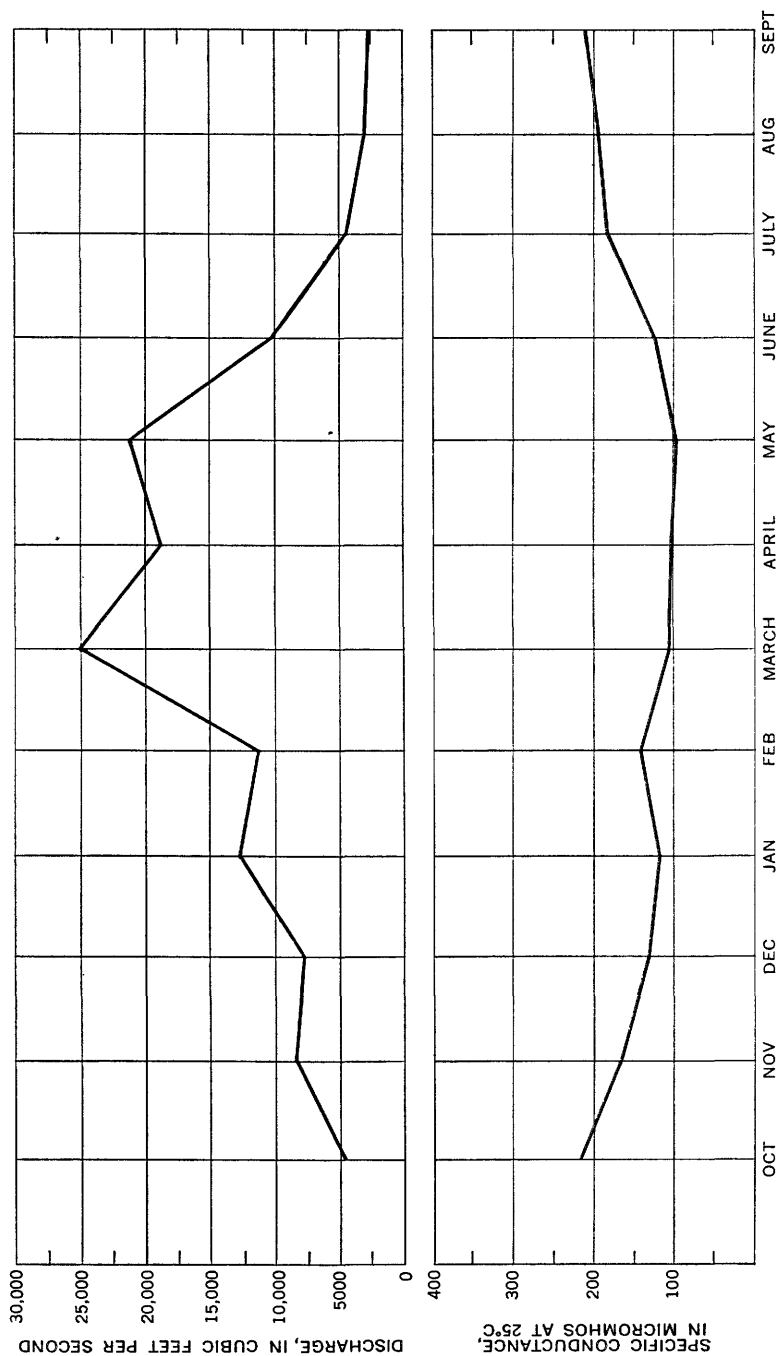


FIGURE 6.—Monthly discharge and specific conductance, West Branch Susquehanna River at Lewisburg, Pa., October 1944 to September 1950.

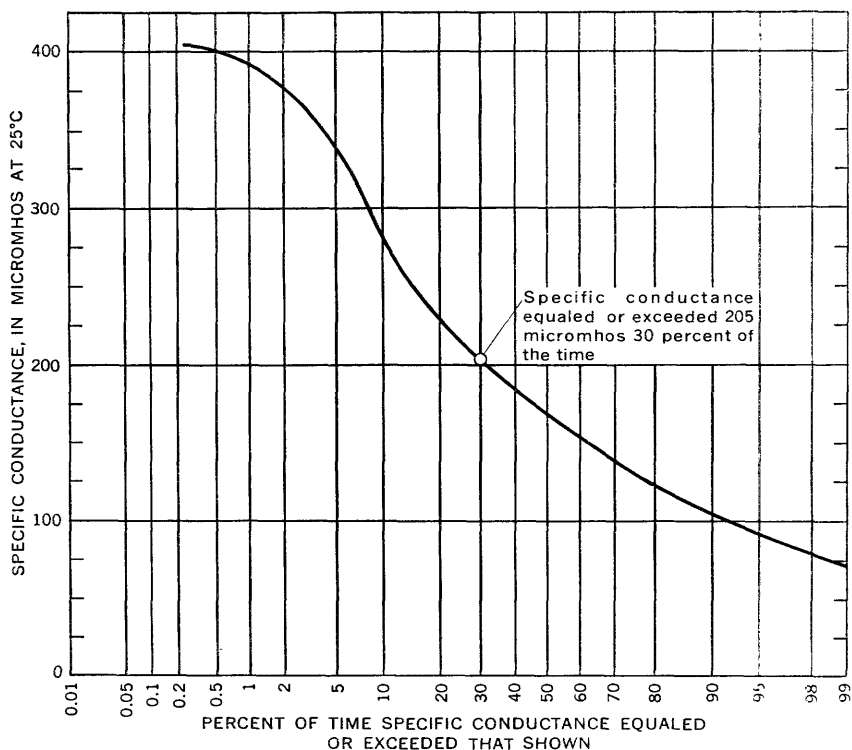


FIGURE 7.—Cumulative frequency curve of specific conductance, West Branch Susquehanna River at Lewisburg, Pa., October 1957 to September 1958.

noncarbonate hardness at Lewisburg. Noncarbonate hardness makes up about 80 percent of the hardness calculated as CaCO_3 . The average hardness (as CaCO_3) for the 1958 water year was 71 ppm (table 11), which was higher than the average reported for other water years. The minimum hardness of water occurred during periods of high flow, commonly during winter and spring. However, much of the time hardness-causing constituents were within tolerance limits prescribed by many industries and other users (Fair and Geyer, 1954).

Approximately 55 percent of the dissolved solids in the river at Lewisburg is sulfate (fig. 9). Waste sulfuric acid draining from coal mines into streams of the headwaters account for the high sulfate. Sulfate does not form insoluble compounds with the concentrations of cations that ordinarily may be dissolved in the river water. Sulfate transported by the river past Lewisburg during the 1958 water year averaged 1,250 tons per day (table 12).

The minimum pH daily samples of the river at Lewisburg during the 1958 water year was 6.2. For a period of 4 years (1944–48) pH of similar samples was 5.0 or more about 97 percent of the time.

TABLE 11.—*Chemical analyses, in parts per million, West Branch Susquehanna River at Lewisburg, Pa., water year October 1957 to September 1958*

LOCATION.—At gaging station at Market Street Bridge at Lewisburg, Union County, 560 feet from east bank of river, 0.2 mile downstream from Buffalo Creek, and 7.4 miles upstream from mouth.

DRAINAGE AREA.—6,847 square miles.

RECORDS AVAILABLE.—Chemical analyses: October 1944 to June 1953, October 1956 to September 1958.

Water temperatures.—October 1944 to June 1953, October 1956 to September 1958.

EXTREMES, 1957-58.—Hardness: Maximum, 150 ppm October 1-11; minimum, 31 ppm April 7-24.

Specific conductance: Maximum daily, 406 micromhos October 7; minimum daily, 68 micromhos April 8.

Water temperatures: Maximum, 79°F July 6, 12; minimum, freezing point on many days during the winter months.

EXTREMES (1944-53, 1956-58).—Dissolved solids: Maximum, 240 ppm October 1-11, 1957; minimum, 46 ppm May 1-10, 1945.

Hardness: Maximum, 156 ppm October 1-10, 1951, August 26-September 6, 1957; minimum, 26 ppm May 21-31, 1946.

Specific conductance: Maximum daily, 423 micromhos August 26, 1957; minimum daily, 63 micromhos November 3, 1956.

Water temperatures: Maximum, 90°F July 28, August 10, 1949; minimum, freezing point on many days during the winter months.

REMARKS.—Records of specific conductance of daily samples available in district office in Philadelphia, Pa. Records of discharge for water year October 1957 to September 1958 given in Water-Supply Paper 1552.

Date of collection	Mean dis-charge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal-cium (Ca)	Mag-nesium (Mg)	Sod-ium (Na)	Potas-sium (K)	Bicar-bonate (HCO ₃)	Sulfate (SO ₄)	Chlo-ride (Cl)	Fluo-ride (F)	Ni-trate (NO ₃)	Dis-solved solids (residue on evap-oration at 180°C)	Hardness as CaCO ₃		Specific con-duct-ance (micro-mhos at 25°C)	pH	Color
														Cal-cium, magne-sium	Non-carbon-ate			
Oct. 1-11, 1957	865	4.3	0.01	43	10	14	2.4	27	140	13	0.1	2.0	240	150	128	384	7.1	5
Oct. 12-31-Nov. 1-11	1,280					15		32	104	12		2.5		120	94	313	7.5	8
Nov. 12-17-20	2,620					19		24	105	11		2.0		104	85	281	7.3	5
Nov. 21-22	4,500					9.0		15	52	6.0		2.9		58	46	159	6.9	6
Nov. 23-Dec. 11	2,640					12		20	78	7.0		2.3		84	68	224	6.9	7
Dec. 12-20	4,060					8.3		8	80	6.4		2.3		83	77	218	6.5	5
Dec. 21-31	29,370	6.4	.01	9.1	2.8	1.3	1.1	6	6	2.6	.0	2.1	64	34	29	94	6.8	3
Jan. 6-7, 9, 22, 1958	7,650					5.1		16	45	4.1		2.5		57	44	149	6.8	3
Jan. 23-26, 28-30	10,130					5.5		15	34	4.0		2.8		44	44	121	7.0	7
Jan. 31-Feb. 1, 9, 11, 15	4,710					5.3		19	48	5.0		3.8		64	49	166	7.1	5
Feb. 16-17, 19-27	4,610					7.8		20	57	6.6		3.8		72	56	191	7.0	3
Feb. 28-Mar. 9	25,540					3.2		10	33	2.8		3.1		42	42	112	6.7	5
Mar. 11-30	14,490					4.4		14	40	4.0		3.3		51	34	134	6.9	3
Mar. 31-Apr. 6	36,940	4.9	.05	8.9	2.7	1.6	1.1	6	27	2.2	.1	1.3		33	28	97	6.2	2
Apr. 7-24	39,100					3.0		6	27	2.0		1.8		31	26	84	6.3	3
Apr. 25-May 14	30,320					3.7		8	35	2.1		2.3		38	32	101	6.4	3
May 19-26	9,140					5.8		17	50	3.8		2.7		61	47	154	6.3	3
May 27-June 11	5,590	13	.01	21	6.9	5.7	1.6	20	63	5.2	.1	2.7	134	81	65	203	6.7	3

QUALITY OF SURFACE WATER, SUSQUEHANNA RIVER BASIN C35

Period	9,800	18	5.8	15	44	2.5	1.4	52	40	138	8.6
June 12-22	9,800	---	---	15	44	2.5	---	---	---	---	---
June 24-26, July 1-6	3,900	---	7.1	24	55	3.0	1.8	81	62	203	6.0
July 7-12, 14-18	8,140	9.0	4.0	20	55	4.5	---	69	53	174	6.7
July 19-20	9,800	0.2	4.8	6	53	4.0	1.12	56	31	146	6.7
July 21-22	14,100	---	5.8	14	43	3.2	1.3	50	39	127	6.6
July 28-Aug. 4	9,800	---	---	14	43	3.2	---	---	---	---	---
Aug. 5-31	5,900	---	7.8	14	66	5.2	---	72	61	189	7.0
Sept. 1-13, 15-27	4,130	21	4.5	14	71	5.8	2.4	79	67	203	6.9
Weighted average	11,390	---	---	16	61	5.6	2.2	71	57	184	---

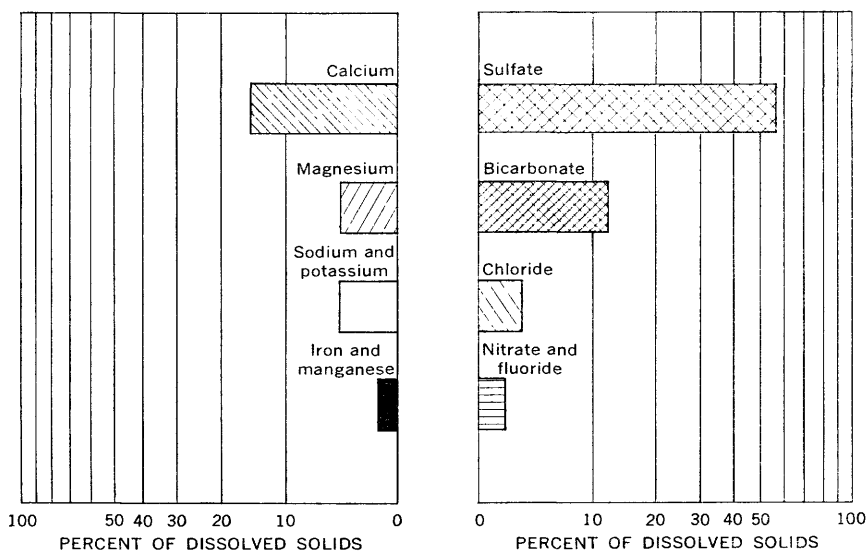


FIGURE 8.—Percentage composition, based on parts per million of average daily analyses, West Branch Susquehanna River at Lewisburg, Pa., 1944-47.

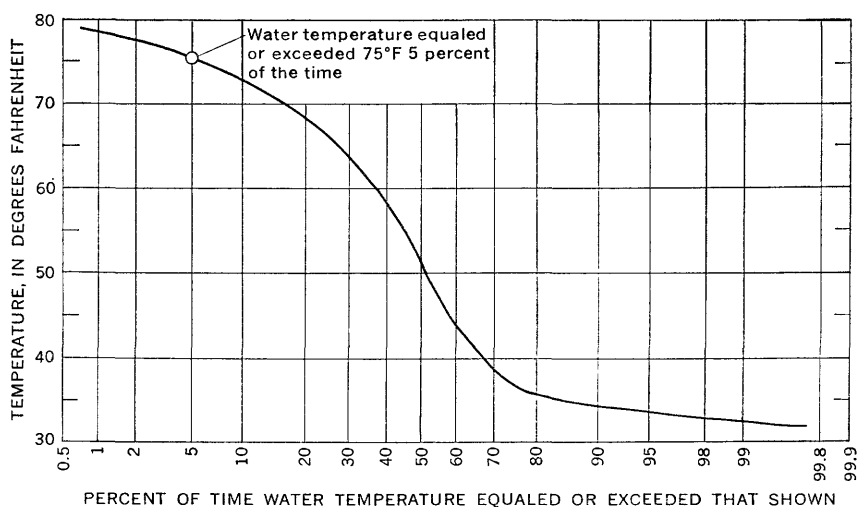


FIGURE 9.—Cumulative frequency curve of water temperature, West Branch Susquehanna River at Lewisburg, Pa., October 1957 to September 1958.

Most of the low values occurred during the war years, 1944-45, when coal mining in the upper basin was at near-peak production.

Temperature of the river during the 1958 water year exceeded 75°F only 5 percent of the time (fig. 9).

Nearly 60 river miles separate Lewisburg from Lock Haven, and this part of the river drains an area of approximately 3,000 square miles. The average discharge at Lewisburg is about twice the average

QUALITY OF SURFACE WATER, SUSQUEHANNA RIVER BASIN C37

TABLE 12.—*Sulfate discharge, West Branch Susquehanna River at Lewisburg, Pa., October 1957 to September 1958*

Collection period	Mean discharge (cfs)	Sulfate discharge (tons per day)	Number of days	Sulfate discharge (tons per interval)
<i>1957</i>				
Oct. 1-11.....	855	323	11	3,553
Oct. 12-31, Nov. 1-11.....	1,280	359	31	11,129
Nov. 12-17, 20.....	2,620	743	7	5,201
Nov. 21, 22.....	4,500	632	2	1,264
Nov. 23-Dec. 11.....	2,640	556	19	10,554
Dec. 12-20.....	4,060	877	9	7,893
Dec. 21-31.....	29,370	2,220	11	24,420
<i>1958</i>				
Jan. 6, 7, 9, 22.....	7,650	929	4	3,716
Jan. 23-26, 29-30.....	10,130	930	6	5,640
Jan. 31, Feb. 1, 9, 11, 15.....	4,710	610	5	3,050
Feb. 16-17, 19-27.....	4,610	709	11	7,799
Feb. 28-Mar. 9.....	25,540	2,276	10	22,760
Mar. 11-30.....	14,490	1,564	20	31,280
Mar. 31-Apr. 6.....	36,940	2,693	7	18,851
Apr. 7-24.....	39,100	2,850	18	51,300
Apr. 25-May 14.....	30,320	2,865	20	57,300
May 16-26.....	9,140	1,234	11	13,574
May 27-June 11.....	5,590	951	16	15,216
June 12-23.....	9,860	1,171	12	14,052
June 24-29, July 1-6.....	3,900	795	12	8,340
July 7-12, 14-18.....	8,140	1,210	11	13,300
July 19-25.....	9,800	1,402	7	9,815
July 26-Aug. 4.....	14,100	1,637	6	9,822
Aug. 5-31.....	5,900	1,041	27	28,107
Sept. 1-13, 15-27.....	4,130	792	26	20,592
Total.....			319	398,528

discharge at Lock Haven for the period of record. Because of this increased streamflow, the river has a lower concentration of solutes at Lewisburg than in upstream segments. At Lewisburg the water is softer, and most of the time the sulfate-ion concentration is less than half that at Lock Haven. However, the concentrations of calcium, magnesium, sodium, and potassium were about the same at both Lock Haven and Lewisburg, whereas the pH and concentration of bicarbonate, chloride, and nitrate were greater at Lewisburg than at Lock Haven (table 13).

A comparison of water quality of the respective branches of the Susquehanna River at Lewisburg and Danville before they merge at Northumberland is shown below:

	<i>Average—1958 water year</i>		
	<i>Specific conductance (micromhos at 25° C)</i>	<i>Hardness</i>	<i>Ratio sulfate to bicarbonate</i>
West Branch At Lewisburg.....	184	71	4 to 1
North Branch At Danville.....	265	110	1.7 to 1

SUMMARY

The principal features of the West Branch Susquehanna River basin affecting water quality are coal mining in the headwaters and limestone formations underlying major tributaries in other parts of the basin. The first of these features causes the river to flow acidic

TABLE 13.—*Chemical analyses, in parts per million, of West Branch Susquehanna River at Lock Haven and Lewisburg, Pa., 1945-59*
 [Time-weighted averages]

	Lock Haven					Lewisburg							
	1946	1947	1948	1949	1950	1959	1945	1946	1947	1948	1949	1950	1958
Discharge (mean annual).....cfs.													
Silica.....	6,416	5,351	5,765	4,822	5,937	5,010	11,530	4.3	10,520	11,029	9,436	11,230	11,390
Iron.....	5.1	5.2					4.5	.03	4.01				
Aluminum.....	1.08	.04					.02						
Manganese.....	1.7	1.5						.11					
Calcium.....	14	14					15	14	13				
Magnesium.....	5.2	5.0					4.7	4.3	3.9				
Sodium + potassium.....	4.4	3.3					5.4	4.1	4.4				
Bicarbonate.....	0	0	0	0	0	0	8	12	10	11	12	8	16
Sulfate.....	72	68	89	109	84	105	52	44	42	52	57	48	61
Chloride.....	2.3	2.4	3.4	4.1	3.0	3.6	3.6	3.2	3.0	4.0	5.0	3.0	5.6
Fluoride.....							1.1						
Nitrate.....	.9	.9	1.0	.9	.9	.9	1.4	1.4	1.4	1.5	1.5	1.3	2.2
Dissolved solids.....	111	109				169	94	84	79				
Hardness, as CaCO ₃	72	68			68	85	57	53	48			54	71
Noncarbonate.....	72	68					50	43	40			47	57
Total acidity as sulfuric acid (H ₂ SO ₄).....	25	22	31	44		22							
Specific conductance.....													
micromhos at 25° C.....	214	200	243	311	263	285	150	136	130	188	173	144	184
Color.....	4	3	2	7	4	2	6	5	3	3	8	3	4

in the upper half of its length, west of Jersey Shore, whereas the second feature tends to neutralize the acidic water as it flows downstream. The change takes place noticeably in the lower half of the river after it mixes with water from Bald Eagle and Pine Creeks.

In the acidic part of the river, particularly the headwater segments and Clearfield, Moshannon, and Sinnemahoning Creeks, the concentrations of aluminum, iron, manganese, and sulfate in water are higher than in downstream segments of the river and in several of the larger tributaries. Concentrations of these and other ionic solutes exceed the tolerances specified by some industries and other water users. However, chemical treatment and(or) dilution can make the water suitable for many users.

The acidic quality of some surface streams whose sulfate concentration also exceeds 150 ppm indicates that ordinary metallic and concrete materials used in bridges and dams in acid-water regions will be subjected to corrosive attack and deterioration.

Aluminum, iron, and manganese in water stain concrete when conditions promote their precipitation from solution as insoluble hydrates. These ions in water also nurture the growth of micro-organisms that often colonize in unsightly masses on the sides of concrete spillways and reservoirs.

Most of the time the West Branch Susquehanna River downstream to Lock Haven contains a calcium sulfate water of an acid pH. However, east of Lock Haven, Bald Eagle Creek contributes calcium bicarbonate water to the river so that below confluence with Bald Eagle Creek the river contains bicarbonate ions for the first time. The high pH and the high alkalinity of Bald Eagle Creek as compared with the river are attributed to the geology of the subbasin which is underlain by limestone. Approximately 9 miles east of the confluence of the Susquehanna River and Bald Eagle Creek, Pine Creek enters the river at Jersey Shore. Pine Creek is a stream of good quality; it is low in dissolved solids and therefore dilutes the more concentrated river water where the creek and river mix. Both Bald Eagle and Pine Creeks contribute water whose quality completes the change of the West Branch Susquehanna River from an acidic to a nonacidic less concentrated river.

This change in quality of the river is noticeable at Williamsport and farther downstream at Lewisburg, although the river still contains a calcium sulfate water, and sulfate is the dominant constituent. At Lewisburg most of the time the pH ranges from 6.0 to 7.0. During the 1958 water year specific conductance was less than 285 micromhos for 90 percent of the time.

Several dams will be constructed on the river and on tributaries throughout the basin. Therefore, much of the basic data presented

in this study will be useful for evaluating any changes in the chemistry of the West Branch Susquehanna River after construction of reservoirs near Curwensville, on Kettle Creek above Westport, and on Bald Eagle Creek above Blanchard. Although designed primarily to control floods, these dams will also provide storage space for surplus water, part of which will be used for recreation. Impounded water can be released in controlled allotments to supplement stream-flow during dry periods and to improve the quality level of the West Branch Susquehanna River, especially the acidic part. At least 100 miles of the river in the upper part of the basin can be made more useful for recreation, domestic, and industrial uses by timely releases of stored water from impoundments when the quality of stored water is suitable for dilution or neutralization purposes.

The West Branch Susquehanna River is less concentrated with dissolved solids than the North Branch where the rivers merge at Northumberland. Despite the influence of coal mining on some tributaries, and on the main river in most of its upper half, many streams throughout the West Branch Susquehanna River basin are clear and relatively pure. A large part of the West Branch Susquehanna River basin is marked for recreational enjoyment, because fish are plentiful, and many wild animals drink from these streams. A man who hunts and fishes in this region may drink from these streams and be refreshed.

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